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2       **Petrogenesis and Provenance of Ungrouped Achondrite**  
3       **Northwest Africa 7325 from Petrology, Trace Elements, Oxygen,**  
4       **Chromium and Titanium Isotopes, and Mid-IR Spectroscopy**

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38 **Abstract**

39  
40 Northwest Africa (NWA) 7325 is an ungrouped achondrite that has recently been recognized  
41 as a sample of ancient differentiated crust from either Mercury or a previously unknown asteroid.  
42 In this work we augment data from previous investigations on petrography and mineral  
43 compositions, mid-IR spectroscopy, and oxygen isotope compositions of NWA 7325, and add  
44 constraints from Cr and Ti isotope compositions on the provenance of its parent body. In  
45 addition, we identify and discuss notable similarities between NWA 7325 and clasts of a rare  
46 xenolithic lithology found in polymict ureilites.

47 NWA 7325 has a medium grained, protogranular to poikilitic texture, and consists of 10-15  
48 vol. % Mg-rich olivine (Fo 98), 25-30 vol. % diopside (Wo 45, Mg# 98), 55-60 vol. % Ca-rich  
49 plagioclase (An 90), and trace Cr-rich sulfide and Fe,Ni metal. We interpret this meteorite to be  
50 a cumulate that crystallized at  $\geq 1200$  °C and very low oxygen fugacity (similar to the most  
51 reduced ureilites) from a refractory, incompatible element-depleted melt. Modeling of trace  
52 elements in plagioclase suggests that this melt formed by fractional melting or multi-stage  
53 igneous evolution. A subsequent event (likely impact) resulted in plagioclase being substantially  
54 remelted, reacting with a small amount of pyroxene, and recrystallizing with a distinctive texture.

55 The bulk oxygen isotope composition of NWA 7325 plots in the range of ureilites on the  
56 CCAM line, and also on a mass-dependent fractionation line extended from acapulcoites. The  
57  $\epsilon^{54}\text{Cr}$  and  $\epsilon^{50}\text{Ti}$  values of NWA 7325 exhibit deficits relative to terrestrial composition, as do  
58 ordinary chondrites and most achondrites. Its  $\epsilon^{54}\text{Cr}$  value is distinct from that of any analyzed  
59 ureilite, but is not resolved from that of acapulcoites (as represented by Acapulco).

60 In terms of all these properties, NWA 7325 is unlike any known achondrite. However, a rare  
61 population of clasts found in polymict ureilites (“the magnesian anorthitic lithology”) are  
62 strikingly similar to NWA 7325 in mineralogy and mineral compositions, oxygen isotope  
63 compositions, and internal textures in plagioclase. These clasts are probably xenolithic in  
64 polymict ureilites, and could be pieces of NWA 7325-like meteorites.

65 Using constraints from chromium, titanium and oxygen isotopes, we discuss two possible  
66 models for the provenance of the NWA 7325 parent body: 1) accretion in the inner solar system  
67 from a reservoir similar to that of acapulcoites in  $\Delta^{17}\text{O}$ ,  $\epsilon^{54}\text{Cr}$  and  $\epsilon^{50}\text{Ti}$ ; or 2) early (< 1 Ma after  
68 CAI formation) accretion in the outer solar system (beyond the snow line), before  $^{54}\text{Cr}$  and  $^{50}\text{Ti}$   
69 anomalies were introduced to this region of the solar system. The mid-IR emission spectrum of

70 NWA 7325 obtained in this work matches its modal mineralogy, and so can be compared with  
71 spectra of new meteorites or asteroids/planets to help identify similar materials and/or the parent  
72 body of NWA 7325.

## 1. INTRODUCTION

75 Northwest Africa (NWA) 7325 was found in Morocco in 2012 as 35 fresh-looking, dark  
76 green stones totaling 345 g in mass. It was reported to be an assemblage of Cr-Al diopside,  
77 calcic plagioclase, and forsterite with a “plutonic igneous” texture, and was classified as an  
78 ungrouped achondrite (Ruzicka et al., 2015). Since then, an additional five meteorites from  
79 northwest Africa have been found to be paired with NWA 7325, making a total mass >1.1 kg  
80 (Meteoritical Bulletin Database).

81 In initial work, Irving et al. (2013) described textures and mineral compositions of NWA  
82 7325, and reported oxygen isotope compositions obtained for several subsamples by laser  
83 fluorination analysis. The oxygen isotope composition reported by Irving et al. (2013) fell  
84 within the broad compositional range shown by ureilites, and also on extensions from the  
85 established trends of acapulcoites and winonaites. Irving et al. (2013) also obtained bulk major  
86 and trace element compositions of cutting dust from NWA 7325 by XRF and ICP-MS. Based on  
87 the highly magnesian mineral compositions (low FeO content) and bulk Al/Si and Mg/Si ratios  
88 of NWA 7325 compared with data from the Messenger mission (Weider et al., 2012), Irving et  
89 al. (2013) suggested that this meteorite could be a plutonic rock excavated from Mercury.

However, the initial age dating of NWA 7325 yielded old ages of  $4562.5 \pm 4.4$  Ma from the Pb-Pb system (Amelin et al., 2013), and  $4562.8 \pm 0.3$  Ma from the  $^{26}\text{Al}$ - $^{26}\text{Mg}$  system (Dunlap et al., 2014). Based on these ages, Dunlap et al. (2014) argued that NWA 7325 was unlikely to be derived from the evolved crust of a planetary sized body, and that an origin on Mercury was therefore doubtful. These preliminary ages have now been refined (Koefoed et al., 2016) to a Pb-Pb age of  $4563.4 \pm 2.6$  Ma and a  $^{26}\text{Al}$ - $^{26}\text{Mg}$  age of  $4563.10 \pm 0.27$  Ma relative to the D'Orbigny angrite anchor, indicating that NWA 7325 crystallized almost contemporaneously with the oldest achondrites such as the quenched angrites (e.g., D'Orbigny; Spivak-Birndorf et al., 2009; Schiller et al., 2010). Koefoed et al. (2016) concluded that the ancient age of NWA 7325 was an argument against formation of NWA 7325 on Mercury, but did not completely rule it out.

101 NWA 7325 has been investigated in many abstracts, and a significant amount of data for this  
102 meteorite is now available in three papers, including petrographic data, bulk chemical analyses,  
103 oxygen, carbon, nitrogen, strontium, xenon and argon isotope compositions, infrared and Raman  
104 spectra, and U-Pb and Al-Mg isotope systematics (Barrat et al., 2015; Koefoed et al., 2016;

105 Weber et al., 2016). These authors have all discussed the petrogenesis and origin of NWA 7325,  
106 and concluded that it is unlike any other known meteorite.

107 In this paper we investigate a new sample of NWA 7325. The results of our work augment  
108 data on petrography and mineral compositions, mid-IR spectroscopy, and oxygen isotope  
109 analyses given in previous publications, and add constraints from Cr and Ti isotope compositions  
110 on the provenance of the NWA 7325 parent body. In addition, we point out and discuss  
111 intriguing mineralogic and oxygen isotope similarities between NWA 7325 and clasts of a rare  
112 xenolithic lithology found in polymict ureilites (Ikeda et al., 2000; Kita et al., 2004).

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## 115 **2. SAMPLES AND METHODS**

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### 117 **2.1. Optical and Electron Microscopy and Electron Microprobe Analyses (EMPA)**

118

119 Two one inch round sections, one standard thin section and one thick section, were prepared  
120 from a 2.197 g sample of NWA 7325 that was purchased by our consortium. Both sections were  
121 first studied for petrography and mineral compositions. Optical and electron microscopy were  
122 performed in the Department of Geosciences at the University of Massachusetts (Amherst).  
123 Back-scattered electron images (BEI) was obtained using the Zeiss EVO50-XVP scanning  
124 electron microscope (SEM). Electron microprobe analysis (EMPA) and wavelength-dispersive  
125 (WDS) X-ray mapping were performed using the Cameca SX-50 electron microprobe. All  
126 analyses utilized natural and synthetic minerals, glasses, oxides and/or metals as standards.  
127 Silicate minerals were analyzed using 15 KeV and 30-60 nA beam current. Olivine grains were  
128 analyzed at 60 nA, with 100-400 second counting times for MnO, Cr<sub>2</sub>O<sub>3</sub>, Al<sub>2</sub>O<sub>3</sub> and CaO, and  
129 40 second counting times for MgO, FeO and SiO<sub>2</sub>. Olivine cores in the Kenna ureilite were  
130 analyzed during every olivine probe run to ensure consistency with previous olivine data of  
131 Goodrich et al. (2013; and references therein) for ureilites and other olivine-rich achondrites.  
132 Plagioclase was analyzed at 10-20 nA with 5-20 second counting times and a slightly de-focused  
133 beam. Na<sub>2</sub>O was always analyzed first to minimize loss during the analyses. Pyroxenes were  
134 analyzed at 30-40 nA with 10-30 second counting times. Sulfides and metal were analyzed using  
135 15 KeV and 50 nA with 10-30 second counting times. Modal mineral abundances were obtained  
136 by point counting collages of BEI of the sections of NWA 7325.

137 Back-scattered electron imaging and EMPA of a magnesian anorthitic clast in polymict  
138 ureilite NWA 10657 (thin section #003) were obtained using the JEOL 8530-FE electron  
139 microprobe at ARES (Astromaterials Research and Exploration Science), Johnson Space Center.  
140 Analytical conditions were similar to those used at U. Mass.

141

## 142 **2.2. Mid-Infrared Emission Spectroscopy**

143 We also used mid-infrared emission spectroscopy to identify the mineralogy of NWA7325  
144 through the use of a Fourier Transform infrared (FTIR) spectrometer. The emissivity technique  
145 uses radiation (heat) emitted from a sample to identify the mineralogy using diagnostic spectral  
146 features that deviate from ideal Planck blackbody radiation at a similar temperature. For this  
147 study, a Thermo Fisher Nicolet 6700 FTIR equipped with a CsI beamsplitter and an uncooled  
148 DTGS detector with a CsI window enabled spectral measurements to be acquired from 2000 to  
149 230  $\text{cm}^{-1}$  (i.e., 5-44  $\mu\text{m}$ ) at 2  $\text{cm}^{-1}$  spectral sampling. This instrument is housed in the Vibrational  
150 Spectroscopy Laboratory at Stony Brook University and was modified for emission  
151 measurements by removing the instrument's internal Globar® (the IR source used for reflectance  
152 measurements) and allowing a heated sample in a glovebox exterior to the spectrometer housing  
153 to act as the infrared source, via a folding mirror that redirects the energy into the ray path of the  
154 spectrometer. The sample was heated to and maintained throughout measurement at  
155 approximately 70 °C and sat in a temperature-regulated environmental sample chamber within  
156 the external Plexiglas® glovebox. The system was purged with nitrogen to drive out other  
157 atmospheric gases (e.g.,  $\text{H}_2\text{O}$ ,  $\text{CO}_2$ ) that otherwise would add spectral features to the data. The  
158 sample studied was the polished thick section of NWA 7325. The exposed surface of the  
159 meteorite on this section was roughly triangular with side lengths of ~12, 12, and 16 mm,  
160 allowing the ~1-cm diameter spot size of the instrument to investigate much of the cross-section  
161 of the meteorite chip.

162 Two blackbody target measurements (at ~70 and 100 °C) were obtained to determine the  
163 instrument response function and instrument temperature used for calibration. The emissivity  
164 spectra of the minerals were derived by reducing the raw wavelength- and temperature-  
165 dependent data by conversion of the sample's raw voltage data measured at the detector into  
166 calibrated sample radiance by dividing the voltage by the instrument response function, then  
167 dividing this sample radiance curve into the temperature-appropriate Planck blackbody curve.

168 The result is (unitless) sample emissivity that ranges from 0 to 1.0 (a blackbody would present an  
169 emissivity of 1.0 across the wavelength range).

170

### 171 **2.3. Oxygen Isotope and Trace Element Analyses**

172 Oxygen three-isotope analyses were performed by secondary ion mass spectrometry (SIMS)  
173 with the Cameca IMS 1280 at the University of Wisconsin (WiscSIMS) using a technique  
174 similar to that of Kita et al. (2010) and Goodrich et al. (2011). The  $\text{Cs}^+$  primary ion beam was  
175 focused to  $\sim 15 \mu\text{m}$  with intensity of  $\sim 4\text{nA}$ . Secondary  $^{16}\text{O}^-$  intensities were typically  $4.5 \times 10^9$   
176 cps and all three isotopes were detected using multicollection Faraday cups (FCs) with external  
177 reproducibility (spot-to-spot) of 0.3-0.4‰ for  $\delta^{18}\text{O}$ ,  $\delta^{17}\text{O}$  and  $\Delta^{17}\text{O}$  ( $= \delta^{17}\text{O} - 0.52 \times \delta^{18}\text{O}$ ).  
178 Several mineral standards (olivine, pyroxene, and plagioclase) were analyzed for calibration of  
179 instrumental biases as a function of the mineral compositions of unknowns (Supporting  
180 Information, Table S1), following the method in Tenner et al. (2013). We used the thick section  
181 of NWA 7325, which also included a San Carlos olivine standard for SIMS analyses. Eight sets  
182 of San Carlos olivine standard analyses bracket 7-12 unknown analyses to monitor instrumental  
183 bias.

184 Trace element analyses of plagioclase were performed after oxygen isotope analyses, similar  
185 to the method described in Kita et al. (2004). We used an  $\text{O}^-$  primary ion beam that is shaped to  
186  $15 \mu\text{m}$  diameter and intensity of  $\sim 3 \text{nA}$ . A total of 17 elements were analyzed by peak switching  
187 magnet field

188 for  $^{23}\text{Na}$ ,  $^{24}\text{Mg}$ ,  $^{27}\text{Al}$ ,  $^{28}\text{Si}$ ,  $^{39}\text{K}$ ,  $^{40}\text{Ca}$ ,  $^{45}\text{Sc}$ ,  $^{47}\text{Ti}$ ,  $^{52}\text{Cr}$ ,  $^{55}\text{Mn}$ ,  $^{57}\text{Fe}$ ,  $^{59}\text{Co}$ ,  $^{60}\text{Ni}$ ,  $^{63}\text{Cu}$ ,  $^{85}\text{Rb}$ ,  $^{88}\text{Sr}$ ,  
189 and  $^{138}\text{Ba}$ . The mass resolving power was set to  $\sim 5,000$  (at 10% height) and no energy offset  
190 was applied. Contributions from molecular and hydride interferences to analyzed atomic ions  
191 are negligibly small. Single analyses took  $\sim 15 \text{ min}$ . We used several plagioclase mineral and  
192 glass standards to estimate relative sensitivity factors (RSF) of trace element peaks to  $^{28}\text{Si}$   
193 (Supporting Information, Table S2), which are used to calculate the trace element concentrations.  
194 Detection limits for trace elements are typically lower than 30 ppb, except for Fe and Ni ( $\sim 0.2$   
195 ppm). Direct ion imaging of analyzed spots were obtained for  $^{23}\text{Na}$ ,  $^{24}\text{Mg}$ ,  $^{39}\text{K}$ ,  $^{40}\text{Ca}$ , and  $^{52}\text{Cr}$ ,  
196 after the trace element analyses to inspect micron-scale zoning of trace elements. The resolution  
197 of the ion image was  $\sim 2 \mu\text{m}$ .

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199  
200

201 **2.4. Chromium and Titanium Isotope Analyses**

202 Analyses of the Cr and Ti isotopic compositions of a bulk sample of NWA 7325 were made  
203 using a 20.94 mg aliquot of homogenized sample powder. A separate 30.31 mg sample of the  
204 acapulcoite type specimen, Acapulco, was also prepared. The samples were dissolved by placing  
205 the powders into a PTFE Parr digestion capsules along with a 2:1 mixture of ultraclean  
206 concentrated HF-HNO<sub>3</sub>. The PTFE capsules were placed into stainless steel jackets and heated  
207 in a 190 °C oven for 96 hours. After digestion in the oven, the samples were treated by re-  
208 dissolving the samples in alternating solutions of 6 N HCl and concentrated HNO<sub>3</sub> to eliminate  
209 fluorides formed during the dissolution process. The Cr was isolated from the bulk matrix of  
210 each sample using a 3-column chromatography procedure previously described by Yamakawa et  
211 al. (2009).

212 Chromium isotopic measurements were made using the Thermo *Triton Plus* thermal  
213 ionization mass spectrometer (TIMS) at the University of California at Davis (UC Davis). The  
214 Cr was loaded onto previously outgassed W filaments by mixing 3 µg of Cr with an Al-silica  
215 gel-boric acid activator with a total load of 12 µg (four filaments in total). The sample filaments  
216 were bracketed by two terrestrial standard filaments before and after loaded with 3 µg of NIST  
217 SRM 979 on each filament. Each filament analysis consisted of 1200 ratios (48 block of 25  
218 ratios) with an 8 second integration time. The intensity of <sup>52</sup>Cr was set to 10 V (±15%) with a  
219 gain calibration completed at the start of each filament and a 60 second baseline measured at the  
220 start of each block. The Faraday cup amplifiers were rotated after every block to eliminate any  
221 issues due to variations in cup efficiencies. Instrumental mass fraction of the Cr isotope ratios  
222 was corrected using an exponential law and a <sup>50</sup>Cr/<sup>52</sup>Cr ratio of 0.051859 (Shields et al., 1966).

223 Titanium was separated from the remaining matrix using a combination of cation and anion  
224 exchange chromatography following the methods of Zhang et al. (2011). Titanium yields after  
225 processing through both cation and anion chromatography were greater than 98%. Titanium  
226 isotope ratios were measured with a Thermo *Neptune Plus* multi-collector inductively coupled  
227 plasma mass spectrometer (MC-ICP-MS) at UC Davis. A standard H-type skimmer cone was  
228 used, while a Jet sample cone was inserted in place of the standard sample cone. Typical  
229 intensity for <sup>48</sup>Ti was 25V (10<sup>11</sup>ohm resistors) for a 1 ppm solution run in high-resolution mode

230 (MSRP ~8000). The isotope ratios were measured in multi-dynamic mode on Faraday cups in  
231 two peak jumping steps, measuring  $^{44}\text{Ca}^+$ ,  $^{46}\text{Ti}^+$ ,  $^{47}\text{Ti}^+$ ,  $^{48}\text{Ti}^+$ ,  $^{49}\text{Ti}^+$ ,  $^{50}\text{Ti}^+$  (in step 1)  
232 and  $^{49}\text{Ti}^+$ ,  $^{51}\text{V}^+$ ,  $^{53}\text{Cr}^+$  (in step 2). The external reproducibilities (2SD) for internally normalized  
233 (to a  $^{49}\text{Ti}/^{47}\text{Ti}$  ratio of 0.749766 [Niederer et al., 1985])  $\varepsilon^{46}\text{Ti}$ ,  $\varepsilon^{48}\text{Ti}$ , and  $\varepsilon^{50}\text{Ti}$ , based on repeated  
234 analyses of pure SPEX Ti solution, are 0.48, 0.23, and 0.53, respectively, consistent with  
235 theoretical expectations.

236

### 237 3. RESULTS

#### 238 3.1. Petrography and Mineral Compositions

239 Both sections show a protogranular to poikilitic texture of high-Ca pyroxene (~0.25–1 mm)  
240 and olivine (~0.1-0.7 mm) grains, surrounded or poikilitically enclosed by plagioclase (Fig. 1).  
241 Olivine grains are rounded and commonly occur as partial or complete mantles around  
242 pyroxenes (Fig. 1, 2a). Modal abundances for the two sections are (by area) ~25-30% pyroxene,  
243 10-15% olivine and 55-60% plagioclase, with trace amounts of sulfide and metal. These values  
244 are similar to modal abundances reported by Irving et al. (2013) and Barrat et al. (2015), but  
245 differ from those (2% olivine and 44% pyroxene) reported by Weber et al. (2016). The sections  
246 also contain several area % voids with shapes similar to those of the olivine (Fig. 1). The voids  
247 may represent former (plucked) olivine grains, but are not included in the modal abundances.

248 Olivine and large pyroxene grains are homogeneous with compositions of  $\text{Fo } 97.5 \pm 0.1$ , and  
249  $\text{Wo } 45.3 \pm 0.2$ ,  $\text{Mg\# } 98.2 \pm 0.2$ , respectively (Table 1, Figure 3,4,5). Major and minor element  
250 compositions are consistent with those of Barrat et al. (2015) and Weber et al. (2016). The large  
251 pyroxene grains show polysynthetic twin lamellae in crossed polarized light and BEI (S4).

252 As noted by Irving et al. (2013) and Weber et al. (2016), plagioclase has a mottled  
253 appearance (Figs. 2b-d). Near contacts with olivine or pyroxene it contains patches of tiny Ca-  
254 rich pyroxene (identity inferred from BEI, x-ray maps, and mixed EMP analyses) and sulfide  
255 grains (Fig. 2b,c,d,f). These inclusions sometimes show elongated shapes and parallel  
256 alignment, suggesting crystallographic control by the plagioclase. Olivine grains have smooth  
257 edges along contacts with plagioclase (Fig. 2d). In contrast, pyroxene grains show resorbed  
258 edges along contacts with plagioclase, with numerous idiomorphic re-entrants of plagioclase  
259 (Fig. 2b,c). Plagioclase also commonly intrudes into pyroxene grains as veins with idiomorphic  
260 side protrusions (Fig. 6a).

261 EMPA profiles in the plagioclase, beginning at contacts with pyroxene or olivine and  
262 extending into the interiors of plagioclase grains show that analyses near the contacts have  
263 excesses of CaO (leading to artificially high An content), MgO (up to ~6 wt.%), and FeO (up to  
264 ~0.5 wt.%), as well as deficits of Si+Al, relative to the “cleaner” analyses from the interior, as a  
265 result of overlap with the small pyroxene inclusions in the analyses (see S5). Based on these  
266 observations, we defined “clean” plagioclase analyses as those having MgO <0.5 wt.%, SO<sub>2</sub>  
267 below detection limit (0.04 wt. %) and molar Si+Al (calculated on the basis of 8 oxygen atoms)  
268 of 3.97-4.02. A compilation of 166 “clean” analyses (Table 1; Fig. 5) showed an average  
269 composition of An 89.7±1.2, with 0.03±0.02 wt.% FeO and 0.3±0.1 wt.% MgO. K<sub>2</sub>O contents  
270 were below detection limit of 0.04 wt. % in all analyses. This composition is consistent with an  
271 average of 71 plagioclase analyses given by Barrat et al. (2015), although these authors do not  
272 report whether their data were selective. In contrast, Weber et al. (2016), reported six  
273 “representative” plagioclase analyses with a wider range of An (79.3-93.5) and higher FeO and  
274 MgO contents. The compositions of Weber et al. (2016) are within the range of all plagioclase  
275 compositions we measured and, by comparison to our observations, may have been contaminated  
276 by tiny pyroxene inclusions.

277 Furthermore, we found that even the “clean” interior areas of plagioclase in NWA 7325 are  
278 not homogeneous (Fig. 6), but are pervaded by fine linear features ( $\leq 2 \mu\text{m}$  wide). In BEI, these  
279 appear as darker zones (lower average Z than surrounding plagioclase) with central areas  
280 consisting of “rosettes” or equigranular crystals (Fig. 6c) of a brighter phase (higher average Z  
281 than surrounding plagioclase). Based on x-ray maps (Figs. 6d-f), the high-Z phase has higher Al  
282 and Mg and much lower Si contents than the surrounding plagioclase, while the darker zones are  
283 similar to the plagioclase but more sodic. From these observations we infer that the veins consist  
284 of Na-enriched plagioclase plus crystals of Mg-Al spinel. SIMS ion images are consistent with  
285 this interpretation (see section 3.4 and S6).

286 The plagioclase also contains small (~10 to 150  $\mu\text{m}$  diameter) “islands” of pyroxene that  
287 show reaction with the plagioclase (Fig. 7). Elemental x-ray maps show that plagioclase  
288 immediately surrounding and intruding into these pyroxene grains is enriched in Na relative to  
289 the average plagioclase composition (Fig. 7a,b). Many of these grains show zonation in BEI,  
290 with darker cores and brighter rims (Fig. 7c; S4). EMPA profiles across such grains show that  
291 the rims are enriched in Ca and Fe, and depleted in Mg, Al and Na, relative to the cores (e.g.,

292 Fig. 7d). Furthermore, some of these grains (e.g., Fig. 7c) have outer rims of dendritic crystals  
293 that appear to be wollastonite, based on x-ray maps and broad beam analyses.

294 Sulfides occur as patches of tiny grains dispersed within plagioclase (Fig. 2b,c). In some  
295 places along plagioclase-boundaries, there are patches of sulfide that appear to have been melted  
296 and dispersed among re-crystallized plagioclase laths (Fig. 2f). In addition, sulfides occur as  
297 ~10-60  $\mu\text{m}$ -sized grains, with rounded to irregular shapes, included in any of the silicates or  
298 along grain boundaries (Fig. 2e). Some contain small blebs of Fe,Ni metal. They commonly  
299 show thin lamellae of a Cr-enriched sulfide (Fig. 2e). These lamellae are too small to analyze,  
300 but are likely daubréelite. The sulfides are troilite with ~3.9 wt.% Cr and ~0.3 wt.% Ni (average  
301 of 13 analyses avoiding Cr-rich lamellae). None of the metal grains were large enough to  
302 analyze cleanly, but minimum values of ~8 to 15 wt.% Ni and ~2.5 to 5.6 wt.% Co are given by  
303 analyses that overlapped silicates.

304

### 305 **3.2. Mid-IR Emission Spectra**

306 An average of 9 emissivity spectra of the polished thick section is shown in Fig. 8. This  
307 spectrum exhibits deep fundamental bands because the polished surface eliminates any volume  
308 scattering features in the spectrum. Using a spectral library of 52 different rock-forming  
309 minerals (Table 2), including a range of feldspar, pyroxene, and olivine compositions, and other  
310 mineral classes, the meteorite spectrum was spectrally unmixed over the spectral range of 2000  
311 to  $300\text{ cm}^{-1}$ , according to the linear-retrieval algorithm (linear least squares) of Ramsey and  
312 Christensen (1998), in order to determine the mineralogic composition of the meteorite chip.  
313 The model fit to the laboratory spectrum indicates that the meteorite sample consists of 57 vol. %  
314 anorthite ( $\text{An}_{89}$ ), 32.8 vol. % diopside ( $\text{Fs}_1\text{Wo}_{49}$ ), and 10.2 vol. % forsterite ( $\text{Fo}_{100}$ ) (Fig. 8).  
315 Although pyrite and troilite were in the spectral endmember library, no Fe sulfide was identified  
316 through mathematical unmixing of the meteorite spectrum, likely because troilite is present in  
317 only trace amounts in the meteorite.

318 These spectral unmixing results coincide well with the mineral compositions and modal  
319 abundances determined from petrographic studies and EMPA in this work and in other  
320 petrologic studies (Irving et al., 2013; Barrat et al., 2015), but vary from results by Weber et al.  
321 (2016) who studied a thin section of NWA 7325 that contains much less olivine (2 vol.%) and  
322 more pyroxene (44%) than our sections. The polished-thin-section reflectance spectrum of

323 Weber et al. (2016) (acquired over an area of 4 x 4 mm and shown in their Figure 6b) was  
324 converted to an emissivity spectrum via Kirchhoff's Law (where emissivity = 1 – reflectivity).  
325 The converted Weber et al. spectrum is shown at the bottom of Figure 8 superposed on our  
326 meteorite emissivity spectrum of the polished thick section. The spectral shapes for the two  
327 meteorite spectra are similar, with the largest difference being the pronounced band at 1105 cm<sup>-1</sup>  
328 (~9 μm) in the Weber et al. (2016) thin section data that likely is due to the higher abundance of  
329 diopside in their section compared with the sections we studied.

330

### 331 **3.3. Oxygen Isotopes**

332 We obtained a total of 19 analyses of NWA 7325 from olivine, pyroxene and plagioclase  
333 (Supporting Information, Table S3). As shown in Fig. 9a, these data plot on the CCAM  
334 (Carbonaceous Chondrite Anhydrous Mineral) line within the region of bulk ureilite data  
335 (Clayton et al., 1977; Clayton and Mayeda, 1996, 1999). Except for one deviant analysis in  
336 plagioclase,  $\delta^{18}\text{O}$  and  $\delta^{17}\text{O}$  values of individual mineral phases are homogeneous within  
337 analytical uncertainties. We intentionally aimed at both the core and the rim of a zoned  
338 pyroxene crystal (similar to Fig. 7c; S4), and found that they do not show any significant  
339 difference (Table S3). The one analysis of plagioclase that is deviant overlapped with a dark  
340 vein (similar to those in Fig. 6). This analysis showed slightly higher  $\delta^{18}\text{O}$  and  $\delta^{17}\text{O}$  values by  
341 0.7 ‰, and higher  $\Delta^{17}\text{O}$  by 0.3‰ (Table S3), compared with the rest of analyses. If this analysis  
342 was a mixture between clean plagioclase and the darker vein, then the vein could have much  
343 higher  $\delta^{18}\text{O}$ ,  $\delta^{17}\text{O}$  and  $\Delta^{17}\text{O}$  values. Shock melting, which we infer to have produced these dark  
344 veins (see below), should not have produced a change in  $\Delta^{17}\text{O}$ . The deviant composition of the  
345 veins could, however, result from preferential weathering of such veins in the terrestrial desert  
346 environment.

347 Excluding this analysis, the average values in each mineral are shown in Table 3. The  $\delta^{18}\text{O}$   
348 values of the three minerals increase slightly from pyroxene (7.3 ‰), to olivine (7.6‰) to  
349 plagioclase (7.9‰), with indistinguishable  $\Delta^{17}\text{O}$  values. Using the modal abundances of the  
350 three minerals, we estimate the bulk oxygen isotope composition of NWA 7253 to be  $\delta^{18}\text{O} =$   
351  $7.7 \pm 0.4\text{‰}$ ,  $\delta^{17}\text{O} = 3.1 \pm 0.3\text{‰}$  (Table 3). The average of 18 spot analyses gives  $\Delta^{17}\text{O} = -$   
352  $0.90 \pm 0.13\text{‰}$ . Figure 9a also shows oxygen isotope compositions determined from bulk samples  
353 of NWA 7325 by Irving et al. (2013), Barrat et al. (2015) and Weber et al. (2016). The bulk

354 sample analyses show some variation within the range of all the SIMS analyses. In addition,  
355 Jabeen et al. (2014) reported oxygen three isotope ratios of plagioclase and pyroxene separates  
356 from NWA 7325 obtained by CO<sub>2</sub> laser fluorination mass spectrometer analyses. The SIMS  
357 plagioclase data match very well with their plagioclase separates, while the SIMS pyroxene data  
358 are ~ 1‰ higher in  $\delta^{18}\text{O}$  than their pyroxene separate. The reason for the discrepancy is not  
359 clear.

360

### 361 **3.4. Trace Elements in Plagioclase**

362 We obtained nine trace element analyses from four plagioclase grains that were also analyzed  
363 for oxygen isotopes. Among them, seven analyses were made on clean plagioclase and two  
364 analyses were on a dark zoned vein. Individual spot data are given in Supporting Information,  
365 Table S2. Concentrations of Ni are near the detection limit (0.2 ppm). The average and 1SD of  
366 seven clean analyses are shown in Table 4 and Figure 10 for selected elements. Most trace  
367 element data show some variability (10-30% in SD) and some analyses show correlated  
368 increases in Mg, Sc, Ti, Cr, Mn, and Fe (Table S2). The enrichment of these elements could be  
369 related to the presence of small pyroxene crystals in the plagioclase (as described above). MgO  
370 and FeO contents of clean plagioclase are calculated to be 0.235% and 0.052%, respectively,  
371 consistent with electron microprobe analyses (Table 1).

372 Analyses on dark zoned veins showed 10-20% lower  $^{28}\text{Si}$  signals and higher Al, consistent  
373 with the inference from electron microprobe analyses that these areas contain  $\mu\text{m}$ -sized spinel  
374 grains (above). Ion images taken after the analyses show  $\mu\text{m}$ -sized inclusions rich in Mg and Cr  
375 (S6), consistent with the electron probe observations (Fig. 6).

376

### 377 **3.5. Chromium and Titanium Isotopes**

378 Chromium and Ti isotopic compositions were obtained in powdered, bulk samples of NWA  
379 7325 and Acapulco (acapulcoite). Deviations of the internally normalized stable  $^{54}\text{Cr}/^{52}\text{Cr}$   
380 and  $^{50}\text{Ti}/^{47}\text{Ti}$  isotope ratios from the terrestrial isotopic composition are presented in Table 5 and  
381 Figure 11. The Cr isotope ratios of both NWA 7325 and Acapulco exhibit deficits relative to the  
382 terrestrial composition with an  $\epsilon^{54}\text{Cr}$  of  $-0.61 \pm 0.11$  and  $-0.70 \pm 0.10$ , respectively. Deficits in  
383 the  $^{50}\text{Ti}$  isotope are also observed in NWA 7325 and Acapulco with  $\epsilon^{50}\text{Ti}$  of  $-1.51 \pm 0.53$  and  
384  $-1.31 \pm 0.27$ , respectively, whereas  $^{46}\text{Ti}$  and  $^{48}\text{Ti}$  are unresolved from terrestrial composition.

385

386

#### 4. DISCUSSION

##### 387 **4.1. Mineralogy and Mineral Compositions of NWA 7325 Compared With Those of Other** 388 **Achondritic Materials**

389 As discussed by Weber et al. (2016), olivine in NWA 7325 is similar to olivine in some  
390 winonaites and acapulcoites/lodranites in its very high Fo content. However, compared with any  
391 known achondrite, its Fe/Mg-Fe/Mn composition is unique, and notable for showing  
392 subchondritic Mn/Mg (Fig. 3a). Its CaO-Cr<sub>2</sub>O<sub>3</sub> composition is also unique, with Cr<sub>2</sub>O<sub>3</sub>  
393 significantly higher than in any olivine-rich achondrites except ureilites, and CaO higher than in  
394 ureilites of similar Cr<sub>2</sub>O<sub>3</sub> content (Fig. 3b). Likewise, pyroxene in NWA 7325 is similar to  
395 pyroxenes in some winonaites and acapulcoite/lodranites in Mg# and Wo, but differs in having  
396 higher Al<sub>2</sub>O<sub>3</sub> (Fig. 4). Plagioclase in NWA 7325 is similar to plagioclase in angrites (Keil,  
397 2012) in its very high An contents (Fig. 5), but has much lower FeO (Fig. 5, 10). It is  
398 distinguished from plagioclase in any other achondrites (Mittlefehldt et al., 1998; Krot et al.,  
399 2013) by higher An, as well as higher MgO (Fig. 5, 10a). Overall, the combination of very  
400 calcic plagioclase and very magnesian mafic silicates in NWA 7325 is not observed in any other  
401 achondrite (Fig. 5). This is further evidenced by the mid-IR emission spectra of the NWA 7325  
402 (Fig. 8), which accurately reflects its mineralogy but does not match the spectrum of any studied  
403 meteorite (e.g., Ashley, 2011). Thus, we conclude that NWA 7325 does not belong to any  
404 known meteorite group, in agreement with Barrat et al. (2015) and Weber et al. (2016).

405 There is, however, another source of achondritic material with which NWA 7325 can be  
406 compared – i.e., achondritic lithologies that have been found only as clasts in meteorite breccias.  
407 Polymict ureilites, for example, contain a significant component of feldspathic clasts (Prinz et  
408 al., 1988; Ikeda et al., 2000; Cohen et al., 2004; Kita et al., 2004; Goodrich and Wilson, 2014).  
409 The majority of these clasts appear to represent two distinct lithologies, the “albitic lithology”  
410 and the “labradoritic lithology”, which have been interpreted as indigenous to the ureilite parent  
411 body based on oxygen isotopes and argued to represent crustal rocks complementary to the  
412 residual ureilites (Ikeda et al., 2000; Cohen et al., 2004; Goodrich et al., 2004; Kita et al., 2004;  
413 Bischoff et al., 2014).

414 Although the plagioclase in the albitic (An 0-32) and labradoritic (An 33-69) lithologies is  
415 more sodic than plagioclase in NWA 7325 (Fig. 5), polymict ureilites also contain a less

416 abundant population of feldspathic clasts that consist of very calcic plagioclase and very  
417 magnesian olivine and/or pyroxene, strikingly similar to the characteristic assemblage of NWA  
418 7325 (Fig. 5). Literature data are available for five of these clasts (Ikeda et al., 2000; Cohen et  
419 al., 2004; Kita et al., 2004), of which one (clast  $\gamma$ -8 in polymict ureilite DaG 319; Ikeda et al.,  
420 2000 and Kita et al., 2004) consists of plagioclase and olivine, two consist of plagioclase and  
421 pyroxene, and two consist only of plagioclase. In addition, we found and analyzed a new  
422 magnesian anorthitic clast in polymict ureilite NWA 10657. The sizes of these clasts (mostly at  
423 the low end of the range 10-500  $\mu\text{m}$ ) compared with grain sizes in NWA 7325 would be  
424 consistent with unrepresentative sampling of their complete mineral assemblages.

425 All six of these clasts have plagioclase compositions in the range An 86-96, similar to NWA  
426 7325 (Fig. 5). Olivine in clast  $\gamma$ -8 has Fo (93), Fe/Mn ratio, and CaO and  $\text{Cr}_2\text{O}_3$  contents similar  
427 to olivine in NWA 7325 (Fig. 3,5). Pyroxene in the pyroxene-bearing clasts have Wo (45-48),  
428 Mg# (95-98), and  $\text{Al}_2\text{O}_3$  contents similar to pyroxene in NWA 7325 (Table 1; Fig. 4). Iron  
429 contents in plagioclase in four of these clasts are very low (<0.03 wt.% FeO), similar to NWA  
430 7325. Moreover, two of these clasts show internal textures in plagioclase that strongly resemble  
431 the internal textures of plagioclase in NWA 7325 (Fig. 12; cf. Figs 6 and 7), including small,  
432 dispersed pyroxene grains, reacted “islands” of high-Ca pyroxene, and “veins” of Na-enriched  
433 plagioclase + (apparently) spinel. Notably, these two clasts both show high FeO contents in  
434 plagioclase (~0.5 wt %), even in areas that appear to be free of inclusions in BEI, similar to  
435 plagioclase in NWA 7325.

436 Oxygen isotope compositions have been determined for two of these clasts (Kita et al., 2004)  
437 and plot on the CCAM line, very similar to the bulk composition of NWA 7325, within the range  
438 of ureilites (Fig. 9b). Despite the ureilite-like oxygen isotope composition of this lithology in  
439 polymict ureilites, it is unlikely to be indigenous to the ureilite parent body because petrologic  
440 modeling indicates that plagioclase this calcic could not have been produced on the ureilite  
441 parent body (Kita et al., 2004; Goodrich et al., 2016a). Therefore, these clasts are probably  
442 xenolithic, and could represent fragments of the NWA 7325 parent body. This is plausible,  
443 because polymict ureilites are known to contain a large variety of chondritic and achondritic  
444 xenoliths (Prinz et al., 1986, 1987a,b; Ikeda et al., 2000; Goodrich et al., 2004, 2016b; Kita et al.,  
445 2004; Downes et al., 2008; Horstmann and Bischoff , 2014).

446

447 **4.2. Equilibration Temperatures for NWA 7325**

448 Based on the distribution of Ca between olivine and high-Ca pyroxene (Köhler and Brey,  
449 1990), the calculated equilibration temperature for NWA 7325 (using the large, unzoned  
450 pyroxene grains) is 1180°C. This is similar to Ca distribution temperatures for ureilites and a  
451 few brachinites, but higher than those of most other olivine-rich achondrites (e.g., Day et al.,  
452 2012; Gardner-Vandy et al., 2013; Goodrich et al., 2015).

453 The distribution of oxygen isotopes between minerals can also provide an estimate of  
454 equilibration temperatures. The oxygen isotope fractionations between minerals in NWA 7325  
455 obtained from SIMS analyses are small (less than 1‰), which suggests formation of these  
456 minerals at igneous temperatures. In contrast, Jabeen et al. (2014) reported 1.6‰ fractionation  
457 between  $\delta^{18}\text{O}$  values of diopside and plagioclase, from which they argue that oxygen isotopes in  
458 NWA 7325 show characteristics of large planetary sized bodies, like Earth and Mars. A  
459 fractionation of 1.6‰ in  $\delta^{18}\text{O}$  between diopside and An<sub>90</sub> translates to an equilibrium  
460 temperature of ~400 °C (Clayton and Kieffer, 1991), which is unreasonably low given the  
461 petrologic characteristics of NWA 7325. An absence of large mass-dependent fractionation of  
462 oxygen isotopes is characteristic of high-temperature igneous differentiation under dry  
463 conditions, such as those among lunar samples (e.g., Spicuzza et al., 2007), but not an indicator  
464 of sizes of planetary bodies.

465

466 **4.3. Petrogenesis of NWA 7325**

467 The subchondritic Mn/Mg ratio of olivine of NWA 7325 (Fig. 3a) indicates that this  
468 meteorite is either a residue of a high degree of partial melting or a cumulate formed at high  
469 degrees of fractional crystallization of a melt (Goodrich and Delaney, 2000). Its high abundance  
470 of plagioclase, which suggest a basaltic composition, as well as its poikilitic texture, suggest that  
471 a cumulate origin is more likely, as interpreted by Irving et al. (2013) and Weber et al. (2016).  
472 In addition, Barrat et al. (2015) discussed constraints from bulk REE abundances that provide  
473 strong support for a cumulate, rather than residue, origin.

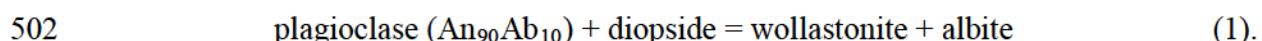
474 Thus, NWA 7325 is a “crustal” rock, a product of a partial melt generated on its parent  
475 asteroid. In contrast, the majority of primitive achondrites (e.g., ureilites, brachinites,  
476 acapulcoites and lodranites, winonaites) are olivine-rich, plagioclase-depleted rocks thought to  
477 be asteroidal residues (Krot et al., 2013). The few known plagioclase-rich achondrites thought to

478 be crustal samples, e.g., GRA 06128/06129 (Shearer et al., 2010; Day et al., 2012), NWA  
479 6704/6693/6926 (Irving et al., 2011; Warren et al., 2013) and NWA 8186 (Srinivasan et al.,  
480 2015), as well as the most abundant types of feldspathic clasts in polymict ureilites, all have  
481 significantly more sodic plagioclase than NWA 7325 (Fig. 5). Thus, not only does NWA 7325  
482 not belong to any known primitive achondrite group, it cannot be related any known group as a  
483 complementary crustal sample. Thus, there is no known meteorite that can provide an exact  
484 analogy for the petrogenesis of NWA 7325.

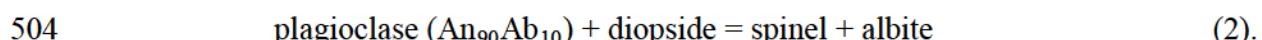
485 NWA 7325 appears to have crystallized from a generally basaltic melt that was depleted in  
486 incompatible elements (see next section), under conditions of very low oxygen fugacity (~IW-  
487 2.9, similar to the most reduced ureilites; Sutton et al., 2016). Pyroxene and olivine were the  
488 earliest phases to crystallize from this melt at temperatures  $\geq 1180$  °C, with plagioclase growing  
489 later around them. The abundance of plagioclase in NWA 7325 (~60%) is too high for the rock  
490 to represent a melt composition (cf. the Ol-Plag-Qtz phase system for Mg-rich systems; Longhi,  
491 1991), so it is mostly likely a cumulate. Its texture (except features resulting from secondary  
492 processes discussed below) is similar to that of heteradcumulates in terrestrial layered igneous  
493 complexes, in which poikilitic crystals (in this case, plagioclase) grow from pore liquids  
494 surrounding cumulus crystals (Wager and Brown, 1967; Hunter, 1996). Considering its probable  
495 asteroidal (rather than planetary) origin, it likely formed in a subvolcanic environment.  
496

497 *4.3.1. Secondary petrologic processes*

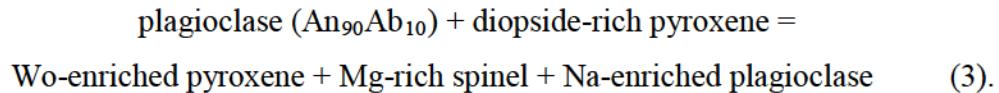
498 At some time after primary crystallization of NWA 7325, plagioclase and sulfides (the  
499 phases with the lowest melting temperatures) were remelted and the melt reacted with small  
500 pyroxene grains that had been included in the plagioclase. The small, zoned pyroxene grains  
501 now observed in plagioclase (Fig. 7, S4) suggest the reaction:



503 The pervasive mottled veins in plagioclase (e.g., Fig. 4) suggest the reaction:



505 However, neither of these proposed reactions can be balanced. In equation [1], Mg is not  
506 accounted for in the products. In equation [2], Ca is not accounted for in the products. This  
507 suggests that the overall reaction that occurred (neglecting the sulfides, which would have melted  
508 and re-crystallized without reaction with silicates) was:



511 This total reaction requires significant mobility of cations and therefore implies a high degree of  
512 melting of the plagioclase, possibly total melting.

513 Based on the presence of mechanical twinning in the pyroxenes (a common effect of shock;  
514 Stöffler et al., 1991), and by analogy to enstatite chondrite impact melt rocks that have some of  
515 the same textural features observed in NWA 7325 plagioclase (Keil, 2007), we suggest that this  
516 remelting of plagioclase and sulfides was caused by impact-related shock. In contrast, Bischoff  
517 et al. (2013) and Weber et al. (2016) argued that it was due to internal reheating, possibly by a  
518 nearby dike-like intrusion (contact metamorphism), followed by rapid cooling. Their arguments  
519 against a shock origin include the paucity of standard shock features in olivine and the absence  
520 of significant Ar loss. However, selective shock melting of low melting-temperature phases can  
521 occur at relatively low degrees of shock (e.g., Warren and Rubin, 2010). Furthermore, it is not  
522 clear what could have caused rapid cooling of a subvolcanic cumulate rock other than excavation  
523 by an impact event. Rapid total melting and only limited reaction with other phases (e.g., small  
524 pyroxene inclusions), such as would occur in a shock event, is also supported by the lack of  
525 evidence for disturbance of the <sup>26</sup>Al-<sup>26</sup>Mg isochron for NWA 7325 (Dunlap et al., 2014; Koefoed  
526 et al., 2016).

527 An important question, given the extremely low FeO contents of “clean” plagioclase areas in  
528 NWA 7325 (potentially a signature feature for NWA 7325-like material) is how FeO might have  
529 been re-distributed during this reaction. One possibility is that FeO from the melted plagioclase  
530 was preferentially partitioned into the newly crystallizing spinel and Wo-rich pyroxene. In this  
531 case, the FeO contents of clean plagioclase in NWA 7325 must be lower than they were in the  
532 primary plagioclase in these rocks. This would suggest that plagioclase in NWA 7325-like  
533 materials that did not experience this re-melting could have higher FeO contents, while only the  
534 “clean” plagioclase in remelted areas has such low FeO. However, this suggestion is not  
535 supported by observations of the clasts of the magnesian anorthitic lithology in polymict ureilites  
536 that have been studied so far, assuming that these clasts represent NWA 7325-like material. The  
537 four clasts that have very low FeO contents in plagioclase were not reported to have textures like  
538 those in NWA 7325 plagioclase (Ikeda et al., 2000; Cohen et al., 2004; Kita et al., 2004),  
539 whereas the two clasts that show internal plagioclase textures like those in NWA 7325 (Fig. 12),

540 showed much higher FeO contents, even in apparently “clean” (inclusion-free) areas. This  
541 suggests the possibility that even these areas are not pure plagioclase, but instead have inclusions  
542 of mafic phases (such as pyroxene and spinel) on a scale much smaller than a micron.  
543 Additional investigations into these textures in NWA 7325 and the magnesian anorthitic clasts in  
544 polymict ureilites could therefore be critical for future attempts to recognize NWA 7325-like  
545 materials among meteoritic materials.

546

#### 547 **4.4. Trace Elements in the Parental Melt of NWA 7325**

548 Concentrations of Mg, Fe, K, Sr, Ti, and Ba in seven analyses of clean plagioclase are  
549 compared in Fig. 10 with those of plagioclase in several other types of achondrites. Plagioclase  
550 in NWA 7325 is highly depleted in Fe, K, Ti, and Ba compared with plagioclase in eucrites,  
551 angrites and the ungrouped basaltic achondrite NWA 011 (Hsu and Crozaz, 1996, 1997; Floss et  
552 al., 2003, 2005). Low Fe contents are obviously related to the high Mg# of NWA 7325  
553 compared with other achondrites. Lower Ti and Ba concentrations in plagioclase in NWA 7325  
554 distinguish it from those in other achondrites. Although partition coefficients for some of these  
555 elements between plagioclase and melt decrease with anorthite contents (Bindemann et al., 1998;  
556 Dohmen and Blundy, 2014), most of the plagioclase shown in Fig. 10 (i.e., angritic and eucritic)  
557 is also anorthite-rich, similar to that of NWA 7325. Thus, the NWA 7325 parental melt must  
558 have been significantly depleted in incompatible element abundances compared with those in  
559 other anorthite-bearing achondrites. Barrat et al. (2015) reported bulk trace element abundances  
560 of NWA 7325, showing depletion in alkali elements (Na, K, Rb) and incompatible trace  
561 elements (REEs), but a high positive Eu anomaly. Our SIMS trace element analyses of  
562 plagioclase are generally consistent with the bulk data, e.g., the high Sr concentrations in  
563 plagioclase are consistent with the Eu anomaly being caused by strong partitioning of Eu<sup>2+</sup> into  
564 plagioclase at low oxygen fugacity. However, in contrast to the high bulk Ba concentrations  
565 reported in Barrat et al. (2015), which were interpreted to be due to terrestrial weathering in a hot  
566 desert environment, we found depleted Ba concentrations in plagioclase, which would be  
567 consistent with the low primary LREE abundances of bulk NWA 7325.

568 Figure 10 also shows data from feldspathic clasts in polymict ureilites, which show a large  
569 range of anorthite compositions and have various trace element abundances (Kita et al., 2004).  
570 The majority of these clasts contain albite-rich plagioclase with high incompatible trace element

571 abundances, and are interpreted to result from fractional crystallization of low-degree ureilitic  
572 melts (Cohen et al., 2004; Kita et al., 2004). However, as mentioned above, a small fraction of  
573 these clasts have anorthite-rich plagioclase and very magnesian mafic minerals (Fig. 5), similar  
574 to NWA 7325. Plagioclase in these magnesian anorthite-rich clasts shows trace element  
575 abundances that are very similar to those of plagioclase in NWA 7325; i.e., enriched in Mg and  
576 Sr, but depleted Fe, K, Ti, and Ba compared with other anorthite-rich plagioclase in achondrites.

577 Using trace element partition coefficients between plagioclase and melt (Bindemann et al.,  
578 1998; Dohmen and Blundy, 2014), we estimated the abundance of Ti, K, Ba, and Sr in the parent  
579 melt of NWA 7325 as was done with plagioclase in felsic clasts in polymict ureilites by Kita et  
580 al. (2004). The estimated trace element abundances in the parent melt of NWA 7325 are very  
581 similar to those calculated for the parent melt of magnesian anorthitic clast  $\gamma$ -8 in polymict  
582 ureilite DaG 319, i.e., high Sr abundance ( $\sim 45 \times$ CI) and sub-CI to CI level abundances ( $0.3\text{--}1 \times$   
583 CI) of Ti, K, and Ba (Table 4). Barrat et al. (2015) also concluded from bulk rock compositional  
584 studies that NWA 7325 crystallized from a melt that was very poor in volatiles (alkali elements)  
585 and incompatible trace elements. They reported a positive Eu anomaly compared to middle  
586 REEs and estimated Eu enrichment of  $14\text{--}23 \times$  CI in the parent melt, which is similar to that of Sr  
587 estimated above.

588 As discussed in Kita et al. (2004), single stage melting of a chondritic precursor would not  
589 produce a melt that is enriched in Sr compared to K and Ba. Sr, K, and Ba are mainly hosted in  
590 plagioclase, in which Sr is compatible ( $D > 1$ ) and others are not ( $D < 1$ ). A partial melt from a  
591 chondritic source would be either more enriched in K and Ba than Sr (at lower degrees of  
592 melting when plagioclase remains in the solid phase), or equally enriched (at high degrees of  
593 melting when plagioclase has been eliminated from the solid). While to some degree the low K  
594 abundance in the estimated parent melt could be due to the volatile-poor nature of the NWA  
595 7325 parent asteroid, the low abundance of Ba still needs to be explained because of its  
596 refractory nature. An incompatible trace element depleted source could be produced either by  
597 repeated extraction of a low-degree partial melt that is enriched in these trace elements (i.e.,  
598 near-fractional melting), or by a multi-stage igneous history .

599 Both Barrat et al. (2015) and Koefoed et al. (2016) discussed possible multi-stage histories  
600 for NWA 7325, based on the positive  $\delta^{26}\text{Mg}^*$  intercept of its Al-Mg isochron (Dunlap et al.,  
601 2014; Koefoed et al., 2016). In these models, the parent melt of NWA 7325 formed by remelting

602 of its source region ~2.5-3 Ma after the initial differentiation of the source (at < 1.8 Ma after CAI;  
603 Koefoed et al., 2016). Barrat et al. (2015) argued that the remelting could have been due to total  
604 impact melting of a gabbroic source. Although a discussion of these models is beyond the scope  
605 of this paper, we do note that the remelting event now evidenced in NWA 7325 by the internal  
606 textures of plagioclase (and argued to be due to impact in section 4.3.1), clearly involved only  
607 partial, not total remelting (principally of plagioclase and sulfide). Therefore, if the total impact  
608 remelting model of Barrat et al. (2015) is correct, then the partial remelting now seen in NWA  
609 7325 must have occurred after the total remelting and would be a tertiary (rather than secondary)  
610 event.

611

#### 612 **4.5. Oxygen, Cr, and Ti Isotope Systematics and the Provenance of NWA 7325**

613 Clayton and Mayeda (1996) first demonstrated that oxygen three-isotope systematics of  
614 achondrites can be used to distinguish groups of achondrites that may be (though are not  
615 necessarily) genetically related. Combining oxygen isotope data with Cr and Ti stable isotope  
616 anomalies (due to nucleosynthetic processes) can provide additional insights into genetic  
617 relationships among planetary materials (e.g., Warren, 2011a,b).

618 NWA 7325 has a negative  $\Delta^{17}\text{O}$  value that is within both the small range of  $\Delta^{17}\text{O}$  of  
619 acapulcoites and the larger range of  $\Delta^{17}\text{O}$  of ureilites (Fig. 9b, 11). Its  $\delta^{18}\text{O}$  value, however, is  
620 significantly higher than that of any acapulcoite (by almost 3‰) and similar to  $\delta^{18}\text{O}$  values of  
621 ureilites of similar  $\Delta^{17}\text{O}$ ; i.e., like ureilites, NWA 7325 plots on the CCAM array on a three  
622 oxygen-isotope diagram (Fig. 9).

623 Both  $\varepsilon^{54}\text{Cr}$  and  $\varepsilon^{50}\text{Ti}$  values of NWA 7325 exhibit deficits relative to terrestrial composition,  
624 as is observed for ordinary chondrites and most achondrites, including acapulcoites and ureilites  
625 (Fig. 11). While  $\Delta^{17}\text{O}$  of NWA 7325 falls within the range observed in ureilites,  $\varepsilon^{54}\text{Cr}$  clearly  
626 sets NWA 7325 apart from all analyzed ureilites (green shaded region in Fig. 11a). The mean  
627 values in  $\varepsilon^{50}\text{Ti}$ - $\Delta^{17}\text{O}$  isotope space (Fig. 11b) show similar offsets from the ureilites, although the  
628 uncertainties associated with  $\varepsilon^{50}\text{Ti}$  permit some overlap with ureilite compositions. The  
629 separation of NWA 7325 from the ureilites is clearly seen in the  $\varepsilon^{54}\text{Cr}$ - $\varepsilon^{50}\text{Ti}$  plot (Fig. 11c). In  
630 contrast, there is no resolvable difference between NWA 7325 and acapulcoites on these  
631 diagrams.

632 Archer et al. (2015) suggested that NWA 7325 experienced late addition of ~0.25% of a  
633 chondritic component, based on abundances of highly siderophile elements (HSE). We have  
634 examined the possibility that the Cr-Ti-O isotope compositions of NWA 7325 were offset from  
635 the ureilite field by addition of a chondritic component. Based on a mixing calculation shown in  
636 Table 7 and Fig. 11, the percentage of various chondritic compositions required to add to  
637 ureilites to bring them to the Cr-Ti-O isotopic compositions of NWA 7325 are: 31-52% H  
638 chondrite; 21-47% L chondrite; 10-34% CI chondrite; 11-29% CM chondrite; 11-20% CV  
639 chondrite; 13-20% CO chondrite; 8-20% CR chondrite; and 20-65% EH chondrite. The  
640 calculation considers the range of ureilite compositions shown in the green square (Fig. 11), as  
641 well as the error bars of the NWA 7325 data point. The calculated fractions are way too high to  
642 be compatible with the potential HSE evidence for chondritic mixing. We can thus rule out the  
643 possibility that mixing with a chondritic component is responsible for the deviation of NWA  
644 7325 from the ureilite field in Fig. 11, and reiterate that NWA 7325 is clearly resolved from  
645 ureilites but not from acapulcoites on this diagram.

646 Based on these combined data for oxygen, chromium and titanium isotopes, as well as  
647 constraints from petrology, we consider two models for the provenance of NWA 7325.  
648

649 *4.5.1. Model 1- Accretion of the NWA 7325 parent body in the inner solar system from a  
650 reservoir similar to that of acapulcoites in  $\Delta^{17}\text{O}$ ,  $\varepsilon^{54}\text{Cr}$  and  $\varepsilon^{50}\text{Ti}$ .*

651 Warren (2011a,b) pointed out that solar system materials fall into two distinct groups on  
652 plots of  $\varepsilon^{54}\text{Cr}$  (or  $\varepsilon^{50}\text{Ti}$ ) vs.  $\Delta^{17}\text{O}$  (e.g., Fig. 11). The carbonaceous chondrites, Eagle Station  
653 pallasites, and an increasing number of CR-, CK- and CV-like ungrouped achondrites (Sanborn  
654 et al., 2013, 2014, 2015; Williams et al., 2016) form one group, which shows positive  $\varepsilon^{54}\text{Cr}$  and  
655  $\varepsilon^{50}\text{Ti}$  values. Earth, Moon, ordinary chondrites, and most of the major differentiated meteorites  
656 (HED, angrites, aubrites, acapulcoites, ureilites, main group pallasites, mesosiderites and IIAB  
657 irons) form the second group, which shows zero or negative  $\varepsilon^{54}\text{Cr}$  and  $\varepsilon^{50}\text{Ti}$  values. A common  
658 interpretation of this bimodality (e.g., Warren, 2011a,b) is that it corresponds to inner solar  
659 system (Earth, Moon, OC, achondrites) vs. outer solar system (CC) materials. If this  
660 interpretation is correct, then NWA 7325 and acapulcoites (as well as ureilites) must have  
661 formed in the inner asteroid belt, and the NWA 7325 parent body may have accreted from the  
662 same reservoir of materials (in terms of  $\varepsilon^{54}\text{Cr}$ ,  $\varepsilon^{50}\text{Ti}$  and  $\Delta^{17}\text{O}$ ) as acapulcoites.

663 However, this model would also have to account for the large difference in  $\delta^{18}\text{O}$  between  
664 NWA 7325 and acapulcoites (Fig. 9b). This difference is too large to be a result of igneous  
665 evolution, because mass dependent oxygen isotope fractionation factors at magmatic  
666 temperatures are very small (Eiler, 2001; Valley et al., 2014). For example, the plagioclase-rich  
667 achondrite GRA 06128/06129 has been suggested to be a crustal cumulate complimentary to  
668 brachinites or brachinitite-like primitive achondrites, and yet its  $\delta^{18}\text{O}$  value is nearly identical to  
669 that of most brachinites (Shearer et al., 2010; Day et al., 2012). Another possibility is that the  
670 difference in  $\delta^{18}\text{O}$  between acapulcoites and NWA 7325 is due to pre-igneous aqueous alteration  
671 on the NWA 7325 parent body, as suggested by Greenwood et al. (2012) to explain mass-  
672 dependent fractionation among winonaites. Although inner solar system bodies are generally  
673 thought to have accreted with little or no water ice, it is possible that the NWA 7325 parent body  
674 accreted later than acapulcoites, after the snow line had moved inward toward the Sun. In this  
675 case, the  $\Delta^{17}\text{O}$  of the water ice must have been very similar to that of the anhydrous rock.

676 Recently, Barrat et al. (2016) reported anomalies observed in the Tm isotopic composition  
677 among Solar System materials. Among the meteorite groups analyzed by Barrat et al. (2016),  
678 objects such as Earth, Mars, and Vesta exhibited a relatively consistent  $\text{Tm}/\text{Tm}^*$  value of less  
679 than 1, while carbonaceous chondrites predominately had  $\text{Tm}/\text{Tm}^*$  values greater than 1. The  
680  $\text{Tm}/\text{Tm}^*$  value for NWA 7325 reported in this same study was the lowest  $\text{Tm}/\text{Tm}^*$  value of any  
681 of the meteorites measured. The observed  $\text{Tm}/\text{Tm}^*$  anomaly, coupled with the negative  $\varepsilon^{54}\text{Cr}$   
682 and  $\varepsilon^{50}\text{Ti}$ , may provide an additional line of evidence for formation in the inner Solar System  
683

684 *4.5.2. Model 2- Accretion of the NWA 7325 parent body in the outer asteroid belt or outer solar  
685 system, beyond the snow line.*

686 Alternatively, the  $\delta^{18}\text{O}$  value of NWA 7325 (i.e., its location on the CCAM array in three-  
687 oxygen isotope space) may indicate that the NWA 7325 parent body accreted in the outer solar  
688 system. This model is suggested by the hypothesis that oxygen isotope compositions on the  
689 CCAM line are a result of parent body aqueous alteration of originally anhydrous primitive  
690 materials (Young and Russell, 1998; Young et al. 1999; Kita et al., 2011; Rudraswami et al.,  
691 2011). We briefly summarize the basis for this interpretation.

692 Most achondrite bulk oxygen isotope ratios plot significantly to the left of the CCAM line on  
693 an oxygen three isotope diagram, with the exception of NWA 7325 and ureilites (Fig. 9). Many

694 of them, especially primitive achondrites that have experienced only low degrees of igneous  
695 processing (e.g., acapulcoites, lodranites, brachinites) plot near the Primitive Chondrule Mineral  
696 (PCM) line, which is a slope  $\sim 1.0$  regression line for the oxygen isotope ratios of olivine and  
697 pyroxene in pristine chondrules from the very primitive (type 3.00) Acfer 094 ungrouped  
698 carbonaceous chondrite (Ushikubo et al., 2012). Oxygen isotope compositions of chondrules in  
699 the least metamorphosed carbonaceous chondrites all plot along the PCM line (Rudraswami et  
700 al., 2011; Tenner et al., 2013, 2015), and therefore this line has been argued to represent the  
701 primordial trend of oxygen isotope compositions in the Solar System (Rudraswami et al., 2011;  
702 Ushikubo et al., 2012; Tenner et al., 2013, 2015). In contrast to materials near the PCM line, the  
703 oxygen isotope compositions of most bulk carbonaceous chondrites are shifted to the right (to  
704 higher  $\delta^{18}\text{O}$ ), and plot along or near the CCAM line. A number of recent studies have shown  
705 that this shift could have resulted from oxygen isotope exchange between  $^{16}\text{O}$ -rich silicates (low  
706  $\Delta^{17}\text{O}$ ) and  $^{16}\text{O}$ -poor fluids (high  $\Delta^{17}\text{O}$ ) originally located along the PCM line in carbonaceous  
707 chondrite parent bodies (e.g., Young and Russell, 1998; Young et al. 1999; Kita et al., 2011;  
708 Rudraswami et al., 2011). Obviously, this interpretation requires that the carbonaceous chondrite  
709 parent bodies accreted with abundant water ice, i.e., in the outer solar system beyond the snow  
710 line (Ciesla and Cuzzi, 2006).

711 The observation that the oxygen isotope composition of NWA 7325 (and also compositions  
712 of ureilites) plots on the CCAM line may indicate that the parent body of NWA 7325 (and that of  
713 ureilites) formed in the outer asteroid belt beyond the snow line, thus accreting with a significant  
714 amount of water-ice and experiencing pre-igneous aqueous alteration. If this is correct, then the  
715 observation that NWA 7325 has  $\Delta^{17}\text{O}$  similar to that of acapulcoites would not be indicative of  
716 any genetic relationship, since acapulcoites would have formed in the inner solar system belt,  
717 inside the snow line.

718 This model for the provenance of NWA 7325 would also have to account for the grouping of  
719 NWA 7325 and ureilites with inner solar system materials (OC and most achondrites), rather  
720 than outer solar system materials (aqueously altered CC), on  $\varepsilon^{54}\text{Cr}$  or  $\varepsilon^{50}\text{Ti}$  vs.  $\Delta^{17}\text{O}$  diagrams  
721 (Fig. 11). This could be explained if the dichotomy between the two groups seen in Fig. 11 is  
722 not exclusively spatial, but has a temporal component as well. Leya et al. (2008) suggested that  
723 carbonaceous chondrites acquired their distinct mix of nucleosynthetic components as a result of  
724 formation later than other meteorites. Dauphas et al. (2010) suggested that planetary materials

725 incorporated different amounts of  $^{54}\text{Cr}$  anomaly-bearing particles due to late injection into  
726 “selective” regions by a supernova. Very early accretion times (<1 Ma after CAI) have been  
727 inferred for the parent bodies of some differentiated meteorites (including ureilites) from Hf-W  
728 age dating and thermal modeling (Bizzarro et al., 2005; Kleine et al., 2005; Hevey and Sanders,  
729 2006; Markowski et al., 2006; Wilson et al., 2008). These accretion times are earlier than the  
730 ages of most analyzed chondrules, and therefore earlier than the accretion of most carbonaceous  
731 chondrite parent bodies (Fujiya et al., 2012; Nagashima et al., 2014). This suggests the  
732 possibility that the parent bodies of NWA 7325 and the ureilites accreted in the solar system  
733 before  $^{54}\text{Cr}$  and  $^{50}\text{Ti}$  anomalies were introduced to this region. In this case, the identical Cr and  
734 Ti isotope compositions of NWA 7325 and acapulcoites would not be indicative of formation  
735 from a spatially common reservoir.

736

737

## 738 5. SUMMARY

739 Ungrouped achondrite NWA 7325 is a protogranular to poikilitic-textured assemblage of 10-  
740 15 vol. % Mg-rich olivine (Fo 98), 25-30 vol. % diopside (Wo 45, Mg# 98), 55-60 vol. % Ca-  
741 rich plagioclase (An 90), and trace amounts of Cr-rich sulfide and Fe,Ni metal. In terms of  
742 modal mineralogy and mineral compositions it is unique compared with all known meteoritic  
743 materials other than a rare feldspathic lithology found as clasts (“the magnesian anorthitic  
744 lithology”) in polymict ureilites. In agreement with previous investigations, we interpret NWA  
745 7325 to be a cumulate rock that crystallized at temperatures  $\geq 1200$  °C and conditions of very low  
746 oxygen fugacity from a generally basaltic, incompatible element-depleted melt. Trace element  
747 abundances in plagioclase indicate that this melt could only have formed by fractional (not  
748 batch) melting of a chondritic source, or a multi-stage igneous history. NWA 7325 experienced  
749 a subsequent event (argued to be impact and excavation), in which plagioclase was substantially  
750 remelted and recrystallized with a distinct texture.

751 The oxygen isotope composition of NWA 7325 plots in the range of ureilites on the CCAM  
752 line and is identical, within error, to the composition of the two analyzed clasts of the magnesian  
753 anorthitic lithology in polymict ureilites. It plots near a slope ~0.5 mass fractionation line  
754 extended to higher  $\delta^{18}\text{O}$  from acapulcoites. In terms of  $\epsilon^{54}\text{Cr}$  and  $\epsilon^{50}\text{Ti}$  values, NWA 7325  
755 exhibits deficits relative to the terrestrial standard, as observed for ordinary chondrites and most

756 achondrites, including ureilites. Its  $\varepsilon^{54}\text{Cr}$  value is distinct from that of any analyzed ureilite, but  
757 is not resolved from that of acapulcoites.

758 NWA 7325 is derived from a parent body that has not previously been sampled by any  
759 known meteorite. However, rare clasts of a magnesian anorthitic lithology in polymict ureilites  
760 show mineral assemblages, textures, and compositions (including oxygen isotope compositions)  
761 that are strikingly similar to NWA 7325. These clasts may be xenoliths derived from the NWA  
762 7325 parent body, and merit further investigation. We discuss two possible models for the  
763 provenance of the NWA 7325 parent asteroid: 1) accretion in the inner solar system from the  
764 same reservoir of Cr, Ti and O isotopes as acapulcoites; 2) early accretion (<1 Ma after CAI) in  
765 the outer solar system (beyond the snow line), before  $^{54}\text{Cr}$  and  $^{50}\text{Ti}$  anomalies were introduced to  
766 this region. The mid-IR emission spectrum obtained in this work from a polished slab of NWA  
767 7325 can be compared in the future with spectra of new meteorites or asteroids/planets to help  
768 identify similar materials and/or the parent body of this meteorite.

769

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Table 1. Compositions of silicates in NWA 7325.

	Olivine (n=64)		Large pyroxenes (n=92)		Plagioclase (n=166)			
	Avg.	SD	Avg.	SD	Avg.	SD	high An	low An
SiO <sub>2</sub>	40.5	0.5	53.0	0.4	45.1	0.8	46.9	45.0
TiO <sub>2</sub>	na		bdl	0.01	bdl		bdl	bdl
Al <sub>2</sub> O <sub>3</sub>	0.05	0.01	2.79	0.14	34.4	0.5	33.5	34.6
Cr <sub>2</sub> O <sub>3</sub>	0.35	0.02	0.92	0.07	bdl		bdl	bdl
FeO	2.50	0.15	0.65	0.05	0.03	0.02	bdl	0.04
MgO	55.7	0.1	19.5	0.1	0.26	0.07	0.28	0.31
MnO	0.09	0.01	0.05	0.02	bdl		bdl	bdl
CaO	0.33	0.02	22.9	0.1	18.5	0.3	17.7	18.7
Na <sub>2</sub> O	na		0.17	0.02	1.17	0.13	1.67	0.99
K <sub>2</sub> O	na		na		bdl		bdl	bdl
SO <sub>2</sub>	na		na		bdl		bdl	bdl
Total	99.6		100.0		99.5		100.2	99.7
Mg#	97.5	0.1	98.2	0.2				
Wo			45.3	0.2				
An					89.7	1.1	85.4	91.2
Or					0.0		0.0	0.0
Ab					10.3	1.0	14.6	8.8

Table 2. Endmember minerals used in spectral unmixing.

Quartz BUR-4120	Silica glass
Microcline BUR-3460	Quenched basalt
Albite WAR-0244	Fo0Fa100 <sup>a</sup>
Oligoclase BUR-060D	Fo10Fa90 <sup>a</sup>
Andesine BUR-240	Fo20Fa80 <sup>a</sup>
Labradorite WAR-4524	Fo30Fa70 <sup>a</sup>
Bytownite WAR-1384	Fo40Fa60 <sup>a</sup>
Anorthite BUR-340	Fo50Fa50 <sup>a</sup>
Actinolite HS-116.4B	Fo55Fa45 <sup>a</sup>
Biotite BIR-840	Fo65Fa35 <sup>a</sup>
Muscovite WAR-5474	Fo70Fa30 <sup>a</sup>
Chlorite WAR-1924	Fo75Fa25 <sup>a</sup>
Enstatite HS-9.4B	Fo80Fa20 <sup>a</sup>
Augite NMNH-9780	Fo89.5Fa10.5 <sup>a</sup>
Augite NMNH-122302	Fo100Fa0"
Serpentine HS-8.4B	Orthoclase WAR-RGSAN01
Serpentine BUR-1690	Oligoclase WAR-5804
Hematite BUR-2600	Pigeonite
Anhydrite ML-S9	Diopside WAR-5780
Gypsum ML-S6	Antigorite NMNH-47108
Calcite ML-C27	Ca-montmorillonite STx-1 solid
Dolomite ML-C28	Magnesiohastingsite HS-115.4B
Nontronite WAR-5108 granular	Magnesiohornblende WAR-0354
Fe-smectite SWa-1 solid	Hypersthene NMNH-B18247
Illite IMt-2 granular	Pyrite ML-SD
K-rich glass	Troilite ML-I9

<sup>a</sup> Synthetic olivine spectra from *Lane et al.*, 2011; otherwise spectra are from the Arizona State University spectral library (Christensen et al., 2000) or co-author Lane's collection.

Table 3. Oxygen isotope ratios of minerals in NWA 7325.

Mineral (mode %)	$\delta^{18}\text{O}$ ‰	$\delta^{17}\text{O}$ ‰	$\Delta^{17}\text{O}$ ‰
Olivine (13%), n=7	7.6 $\pm$ 0.3	3.0 $\pm$ 0.2	-0.97 $\pm$ 0.17
Pyroxene (28%), n=7	7.3 $\pm$ 0.2	3.0 $\pm$ 0.2	-0.84 $\pm$ 0.17
Plagioclase (58%), n=4	7.9 $\pm$ 0.4	3.2 $\pm$ 0.3	-0.90 $\pm$ 0.20
Average*	7.7 $\pm$ 0.4	3.1 $\pm$ 0.3	-0.90 $\pm$ 0.13

Errors quoted are 95% confidence level.

\*Average values of  $\delta^{18}\text{O}$  and  $\delta^{17}\text{O}$  are weighted by modal volume % and the average of  $\Delta^{17}\text{O}$  value is from the mean of 18 spot analyses.

Table 4. Trace element concentrations (ppm) of plagioclase in NWA 7325.

Elements	Mg	K	Sc	Ti	Cr	Mn	Fe	Rb	Sr	Ba
Mean (n=7)	1420	26	0.3	12	6	9	180	0.09	320	0.22
SD %	15	16	52	27	59	37	36	69	0.6	32

Table 5. Chromium and titanium isotopic compositions of NWA 7325 and Acapulco.

Sample	$\varepsilon^{54}\text{Cr}$ ( $\pm 2\text{SE}$ )	$\varepsilon^{46}\text{Ti}$ ( $\pm 2\text{SE}$ )	$\varepsilon^{48}\text{Ti}$ ( $\pm 2\text{SE}$ )	$\varepsilon^{50}\text{Ti}$ ( $\pm 2\text{SE}$ )
NWA 7325	-0.61 $\pm$ 0.11	-0.42 $\pm$ 0.13	-0.06 $\pm$ 0.08	-1.58 $\pm$ 0.33
Acapulco	-0.70 $\pm$ 0.10	-0.40 $\pm$ 0.13	-0.03 $\pm$ 0.09	-1.48 $\pm$ 0.45

Table 6. Trace element abundances in parent melt of NWA 7325 estimated from plagioclase trace element compositions.

	MgO (%)	K (ppm)	Ti (ppm)	Sr (ppm)	Ba (ppm)
Plagioclase	0.23	26	12	320	0.22
D(plagioclase/melt) <sup>*1</sup>	0.037	0.16	0.036	0.92	0.084
Parent melt	6.3	170	320	350	2.6
Abundance/CI <sup>*2</sup>		0.30	0.73	45	1.1

<sup>\*1</sup> Values for Mg, K, Sr and Ba are from Dohmen and Blundy (2014) assuming Na<sub>2</sub>O and CaO wt% of parent melt to be the same as those of bulk 7325 (Barrat et al., 2015). Value for Ti is from Bindemann et al. (1998), assuming T=1500 K and plagioclase composition (X<sub>An</sub>=0.9).

<sup>\*2</sup> CI chondrite abundance is from Anders and Grevesse (1989).

Table 7. Percent mixing (min-max range) of chondrite end-member required to generate NWA 7325 composition from ureilite.

Chondrite Group	$\varepsilon^{54}\text{Cr}-\Delta^{17}\text{O}$		$\varepsilon^{50}\text{Ti}-\Delta^{17}\text{O}$		$\varepsilon^{50}\text{Ti}-\varepsilon^{54}\text{Cr}$	
	Min (%)	Max (%)	Min (%)	Max (%)	Min (%)	Max (%)
CI	10	39	0	49	4	34
CM	11	33	0	29	4	36
CO	13	20	0	20	6	24
CV	11	24	0	22	4	20
CR	8	20	-	-	-	-
H	31	52	0	52	14	86
L	21	49	0	47	14	84
LL	26	45	0	44	16	88
EH	20	65	0	72	10	66
EL	15	68	-	-	-	-

## Figure Captions

**Fig. 1.** Collage of back-scattered electron images (BEI) of studied thick section of NWA 7325. The rock has a protogranular to poikilitic texture of 25-30% high-Ca pyroxene (pyx) and 10-15% olivine (ol) grains, surrounded or poikilitically enclosed by 55-60% plagioclase (plag).

**Fig. 2.** BEI of NWA 7325. (a) Olivine (ol) grains are rounded with concave segments, and commonly occur as partial or complete mantles around pyroxenes (pyx). (b) Plagioclase has a mottled appearance and shows reacted boundaries with both olivine and pyroxene grains. (c) Pyroxene grains in contact with plagioclase show resorbed edges, with idiomorphic reentrants into plagioclase. (d) Plagioclase in contact with olivine grains contains numerous tiny inclusions of Ca-rich pyroxene, often with elongated shapes and parallel alignment, while olivine grains have smooth edges. (e) Grain of troilite showing fine lamellae of Cr-rich phase inferred to be daubreelite. (f) Grain of troilite at pyroxene-olivine-plagioclase junction. Plagioclase shows reaction with both sulfide and pyroxene, but not with olivine.

**Fig. 3.** Compositions of olivine in NWA 7325 compared with olivine in various asteroidal achondrites and rare magnesian anorthitic (Mg-An) clasts in polymict ureilites. (a) Molar Fe/Mg vs. Fe/Mn. (b) Wt.% CaO vs. wt.%  $\text{Cr}_2\text{O}_3$ . The composition of NWA 7325 olivine is not like that of any known achondrites, but is close to compositions of magnesian anorthitic clasts in polymict ureilites. Data for brachinites and brachinite-like achondrites from Nehru et al. (1983), Warren and Kallemeyn (1989), Goodrich and Righter (2000), Mittlefehldt et al. (2003), Goodrich et al. (2006, 2011, 2015), Day et al. (2012), Gardner-Vandy et al. (2013) and Meteoritical Bulletin Database. Data for ungrouped achondrites Zag(b) from Delaney et al. (2000), Divnoe from Petaev et al. (1994), Tafassasset from Gardner-Vandy et al. (2012), LEW 88763 from Gardner-Vandy (2012) and Day et al (2015), and GRA 06128/06129 from Shearer et al. (2010) and Day et al. (2012). For sources of ureilite, lodranite, and winonaite data, see Fig. 5 of Goodrich et al. (2011). Data for pallasites from Mittlefehldt et al. (1998). Data for magnesian anorthitic clasts in polymict ureilites from Ikeda et al. (2000), Kita et al. (2004), Cohen et al. (2004) and Goodrich and Wilson (2014).

**Fig. 4.** Mg# vs. wt.% Al<sub>2</sub>O<sub>3</sub> for high-Ca pyroxene in NWA 7325 compared with various asteroidal achondrites and magnesian anorthitic clasts in polymict ureilites. The composition of NWA 7325 pyroxene is unique on this plot compared with known achondrites, but similar to that in magnesian anorthitic clasts in polymict ureilites. Data for ureilites from Takeda (1989), Takeda et al. (1989), Goodrich et al. (2009, 2014 and references therein), and Weber et al. (2003). Data for brachinites and brachinite-like achondrites from Nehru et al. (1983), Warren and Kallemeyn (1989), Mittlefehldt et al. (2003), Goodrich et al. (2011) and Gardner-Vandy et al. (2013). Data for ungrouped achondrites Zag(b) from Delaney et al. (2000), Divnoe from Petaev et al. (1994), Tafassasset from Gardner-Vandy et al. (2012), LEW 88763 from Gardner-Vandy (2012) and Day et al. (2015), and GRA 06128/06129 from Shearer et al. (2010) and Day et al. (2012). Data for angrites from Mittlefehldt et al. (1998). Data for acapulcoites and lodranites from Takeda et al. (1994), Mittlefehldt et al. (1996) and Burroni and Folco (2008). Data for winonaites from Kimura et al. (1992). Data for magnesian anorthitic clasts in polymict ureilites from Ikeda et al. (2000), Kita et al. (2004), Cohen et al. (2004), Goodrich and Wilson (2014), and this work.

**Fig. 5.** Plot of An# in plagioclase vs. Mg# of olivine and/or pyroxene in NWA 7325 compared with various asteroidal achondrites and feldspathic clasts in polymict ureilites. The combination of very An-rich plagioclase and very magnesian mafic minerals in NWA 7325 is unique compared with any known asteroidal achondrites, but very similar to that of the magnesian anorthitic population of feldspathic clasts in polymict ureilites. ac/lod = acapulcoites and lodranites; brach = brachinites and brachinite-like achondrites. Sources of data as in figures 3 and 4.

**Fig. 6.** (a) BEI showing vein of plagioclase with idiomorphic side protrusions in pyroxene in NWA 7325. (b) BEI showing mottled appearance of plagioclase in NWA 7325. (c) Higher magnification image of plagioclase. Mottled appearance is caused by fine ( $\leq 2 \mu\text{m}$  wide) linear features, which consist of darker zones with central “rosettes” of a brighter phase. Box indicates area of x-ray maps in [d-f]. (d-f) Mg, Al and Si x-ray maps of area in box in [c], showing that darker areas have lower Al and higher Si than surrounding plagioclase, while bright rosettes have

higher Mg and Al and lower Si. Rosettes may be spinel, with the darker zones being Al-depleted (Na-enriched) plagioclase.

**Fig. 7.** Small grains of pyroxene showing reaction textures with surrounding plagioclase in NWA 7325. (a) BEI. Note that plagioclase immediately surrounding and protruding into the pyroxene grain is darker than the bulk of the plagioclase. (b) Na-Ca-Al x-ray map of the area in [a], showing that the reaction zone of plagioclase around the pyroxene grain is more sodic than the bulk of the plagioclase. (c) BEI. Pyroxene grain showing two distinct cores. Tiny crystals of very bright phase in partial halo around the pyroxene were tentatively identified as wollastonite. (d) Profiles of Ca, Mg, Al and Fe along profile 7 marked in [c]. Rims are enriched in Ca, and depleted in Mg, Al and Fe relative to the cores. Explanation for elevated Al between the two cores is unclear.

**Fig. 8.** Mid-infrared emissivity spectrum of NWA 7325 (blue). The spectrum is an average of nine individual spectra (which are each 256 co-added scans acquired during measurement). Also shown is modeled spectrum of NWA 7325 (orange) and the spectra of the identified mineral components (black). The emissivity spectrum of NWA 7325 is repeated at the bottom of the figure (blue) superposed by the reflectance data of Weber et al. (2016) converted to emissivity (green) for comparison.

**Fig. 9.** (a) Oxygen three-isotope analyses of NWA 7325 minerals using SIMS. Individual data points represent single SIMS analyses. Olivine, pyroxene, and anorthite are shown as squares, triangles and diamonds, respectively. One anorthite analysis shown as open symbol hit a vein in the plagioclase and the data deviate beyond analytical uncertainty. Laser fluorination analyses of bulk chips (Irving et al., 2013; Weber et al., 2016), mineral separates (plagioclase “PI” and pyroxene “Px”; Jabeen et al. 2014), and a bulk powder (Barrat et al., 2015) are shown as open circles. The range of ureilite bulk analyses (Clayton and Mayeda, 1996) is enclosed by dotted line. Terrestrial fractionation (TF) and carbonaceous chondrite anhydrous mineral (CCAM) lines are shown for reference. (b) Oxygen isotope ratios of bulk achondrites (Clayton and Mayeda, 1996; Yamaguchi et al., 2002; Greenwood et al. ,2012) and Mg-An-rich clasts in DaG 319 polymict ureilite (Kita et al. 2004) are compared to the average value of three minerals in NWA

7325 (Table 4). Ranges of HED meteorites, Martian meteorites (Mars) and terrestrial mantle and lunar samples (Earth-Moon) are also shown as oval areas. The primitive chondrule mineral (PCM) line (Ushikubo et al., 2012) is shown along with the TFL and CCAM lines as references.

**Fig. 10.** Selected trace element concentrations in plagioclase in NWA 7325, polymict ureilites, and basaltic achondrites. NWA 7325 data points represent individual analysis spots (Online Supporting Information, Table S2), excluding 2 analyses that hit veins in plagioclase. Polymict ureilite data are from Kita et al. (2004). Filled diamonds are for magnesian anorthitic plagioclase clast in polymict ureilite DaG 319. Data for albitic and labradoritic lithology clasts in polymict ureilites shown as grey squares. Data from eucrites, angrites, and ungrouped achondrite NWA 011 are shown as the total range reported in Hsu and Crozaz (1996, 1997) and Floss et al. (2003, 2005).

**Fig. 11.** (a) Comparison of  $\Delta^{17}\text{O}$  and  $\varepsilon^{54}\text{Cr}$  isotopic composition of NWA 7325 with other achondrite and carbonaceous chondrite groups. Dashed lines represent examples of mixing curves between ureilite compositions (range indicated by green box) and various chondrite end-members that pass through the composition of NWA 7325. Mixing lines were calculated by adding 2% increments of the chondrite end-members to the ureilite composition. Literature data for  $\Delta^{17}\text{O}$  are from Clayton and Mayeda (1996, 1999), Clayton et al. (1984, 1991), Scott et al. (2009), Jenniskens et al. (2012, 2014) and Popova et al. (2013). Literature data for  $\varepsilon^{54}\text{Cr}$  are from Ueda et al. (2006), Shukolyukov and Lugmair (2006), Trinquier et al. (2007), Jenniskens et al. (2012, 2014) and Popova et al. (2013). (b) Comparison of  $\Delta^{17}\text{O}$  and  $\varepsilon^{50}\text{Ti}$  of NWA 7325 with other achondrite and carbonaceous chondrite groups. Literature data for  $\Delta^{17}\text{O}$  same as figure (a) and literature data for  $\varepsilon^{50}\text{Ti}$  are from Trinquier et al. (2009), Leya et al. (2008), and Zhang et al. (2011). (c)  $\varepsilon^{54}\text{Cr}$  versus  $\varepsilon^{50}\text{Ti}$  comparison plot of NWA 7325 and other meteorite groups. Each meteorite group is shown as a composite point averaging multiple samples within a group using the references given for (a) and (b). The values for NWA 7325 and Acapulco are given in Table 5.

**Fig. 12.** BEI of magnesian anorthitic clast (clast 9) in polymict ureilite NWA 10657\_003. Clast consists of plagioclase of An 86-89 with an internal texture including small, dispersed pyroxene

grains, reacted “islands” of high-Ca pyroxene, and “veins” of Na-enriched plagioclase + Mg, Al-rich phase inferred to be spinel, similar to plagioclase in NWA 7325 (see figures 6 and 7). Areas outlined by boxes in [a] are shown at higher magnification in [b] and [c]. Linear features seen in plagioclase in [c] are shown at higher magnification in [d].

NWA 7325

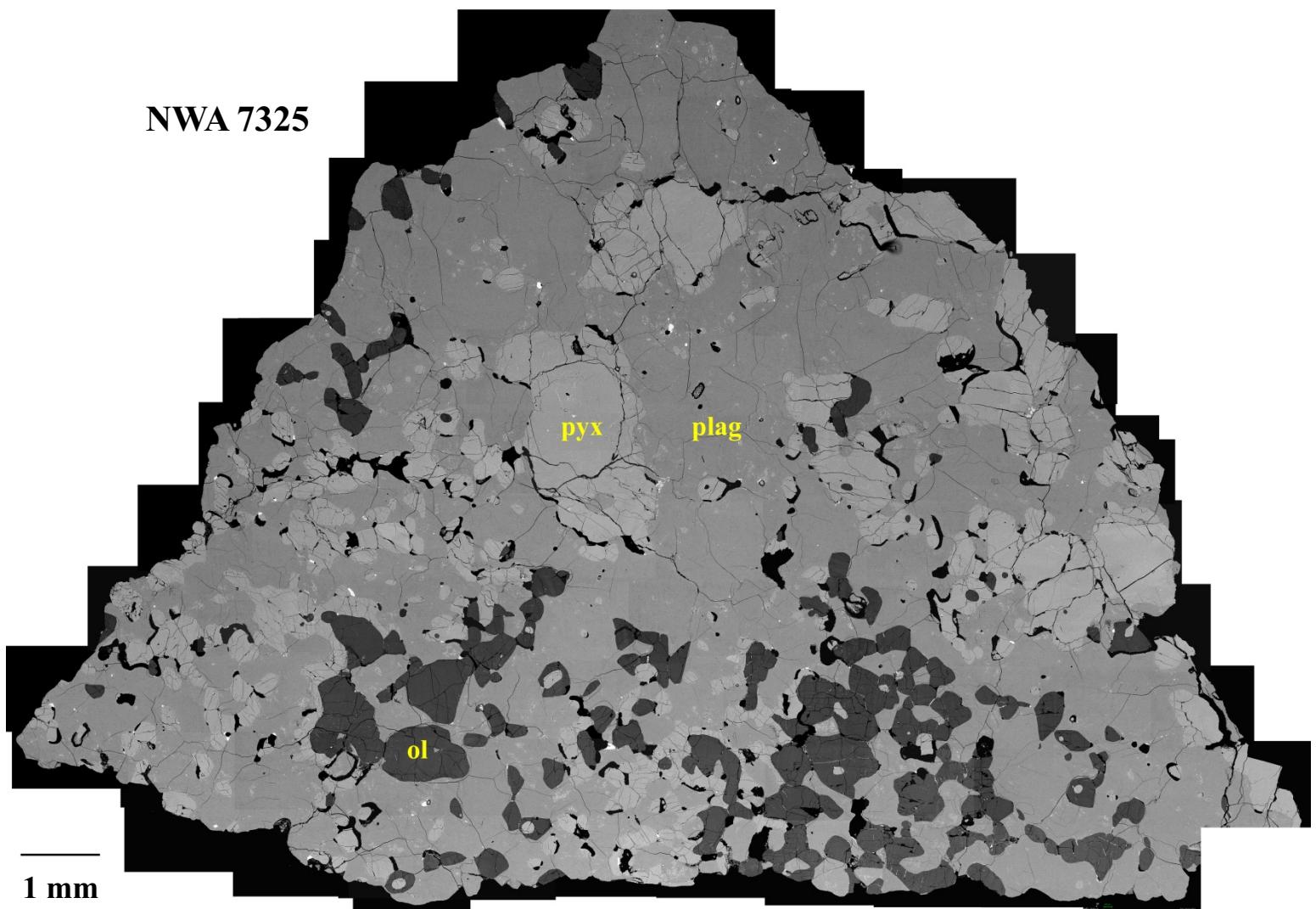


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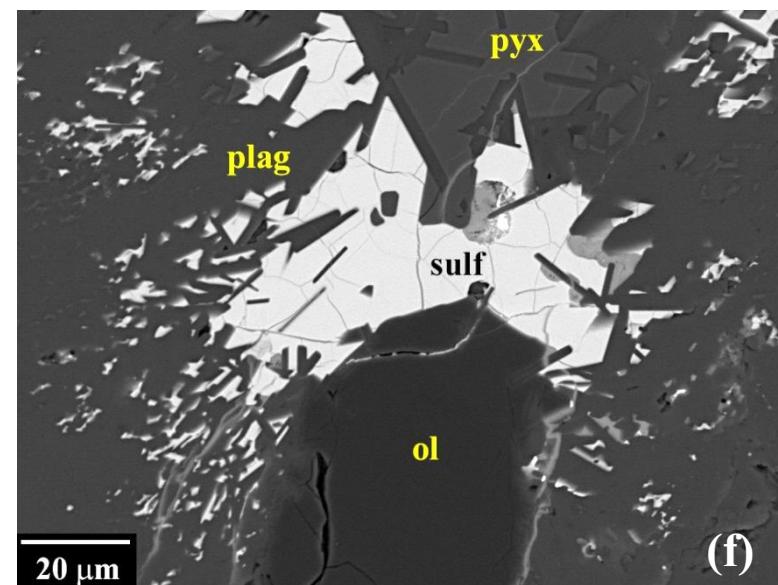
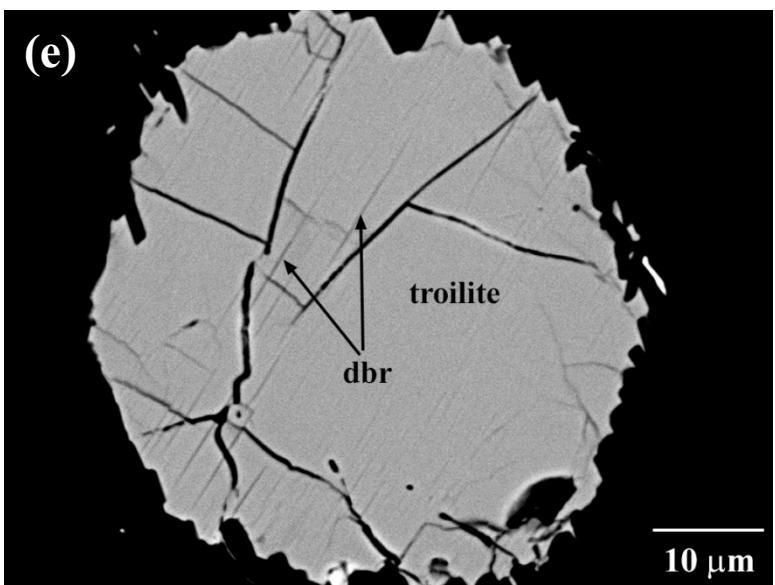
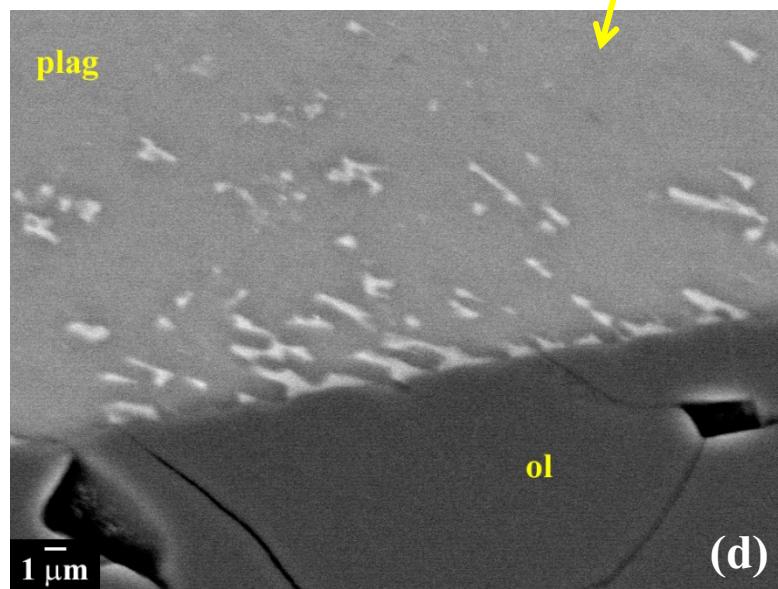
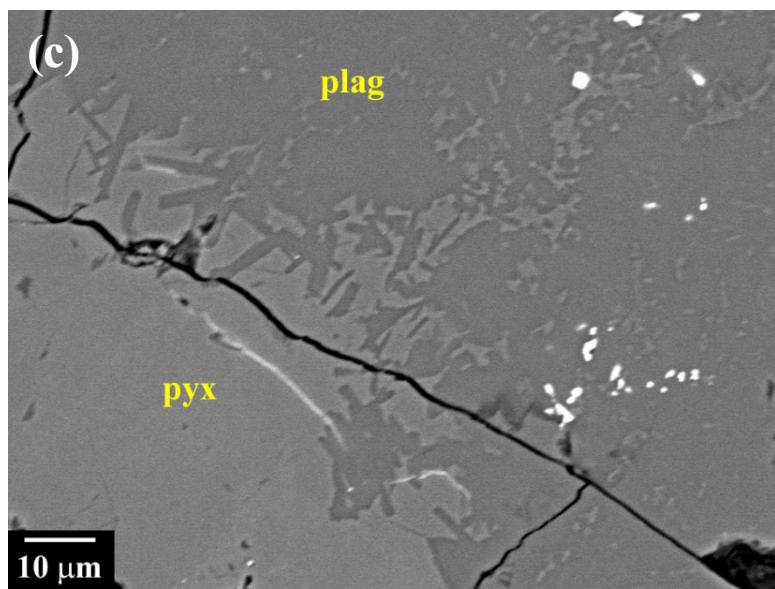
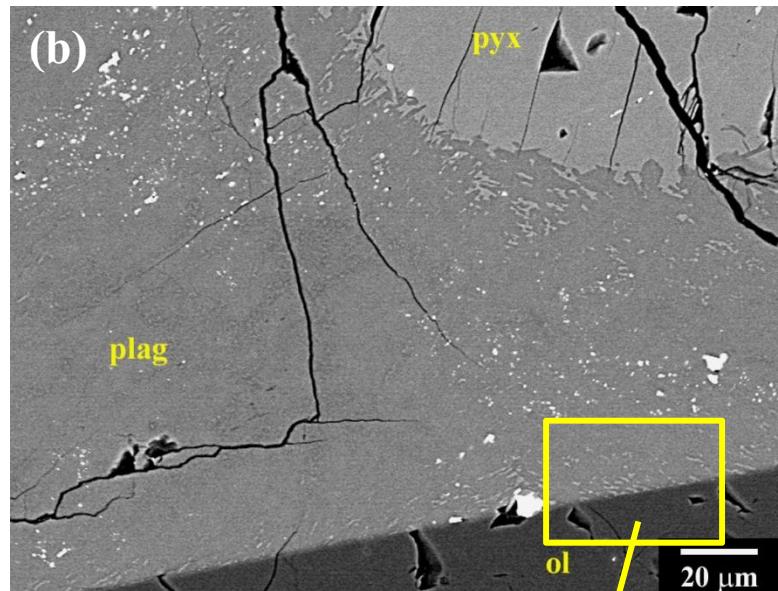
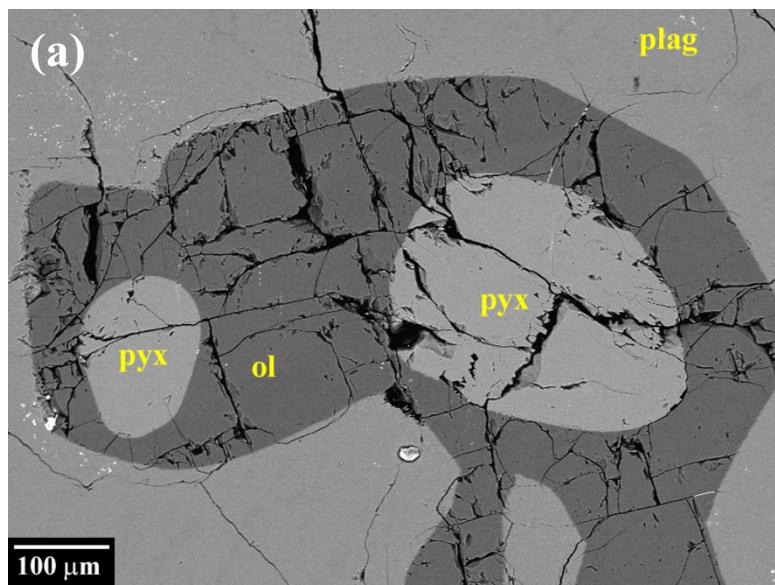


Fig. 2

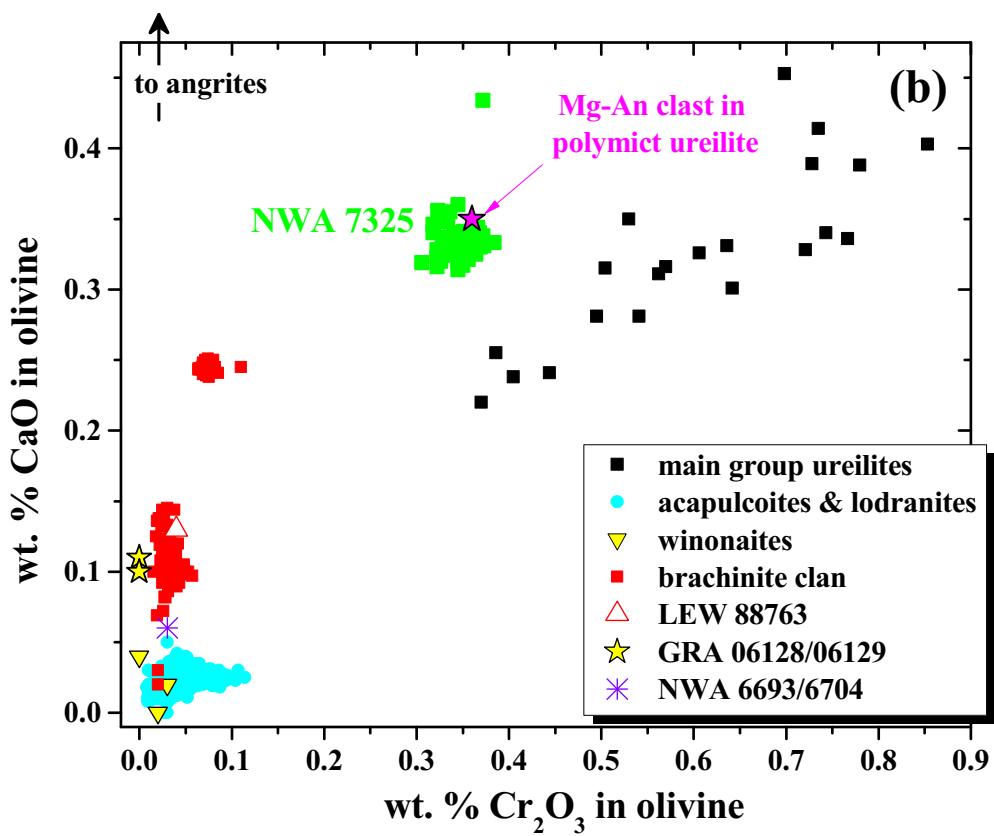
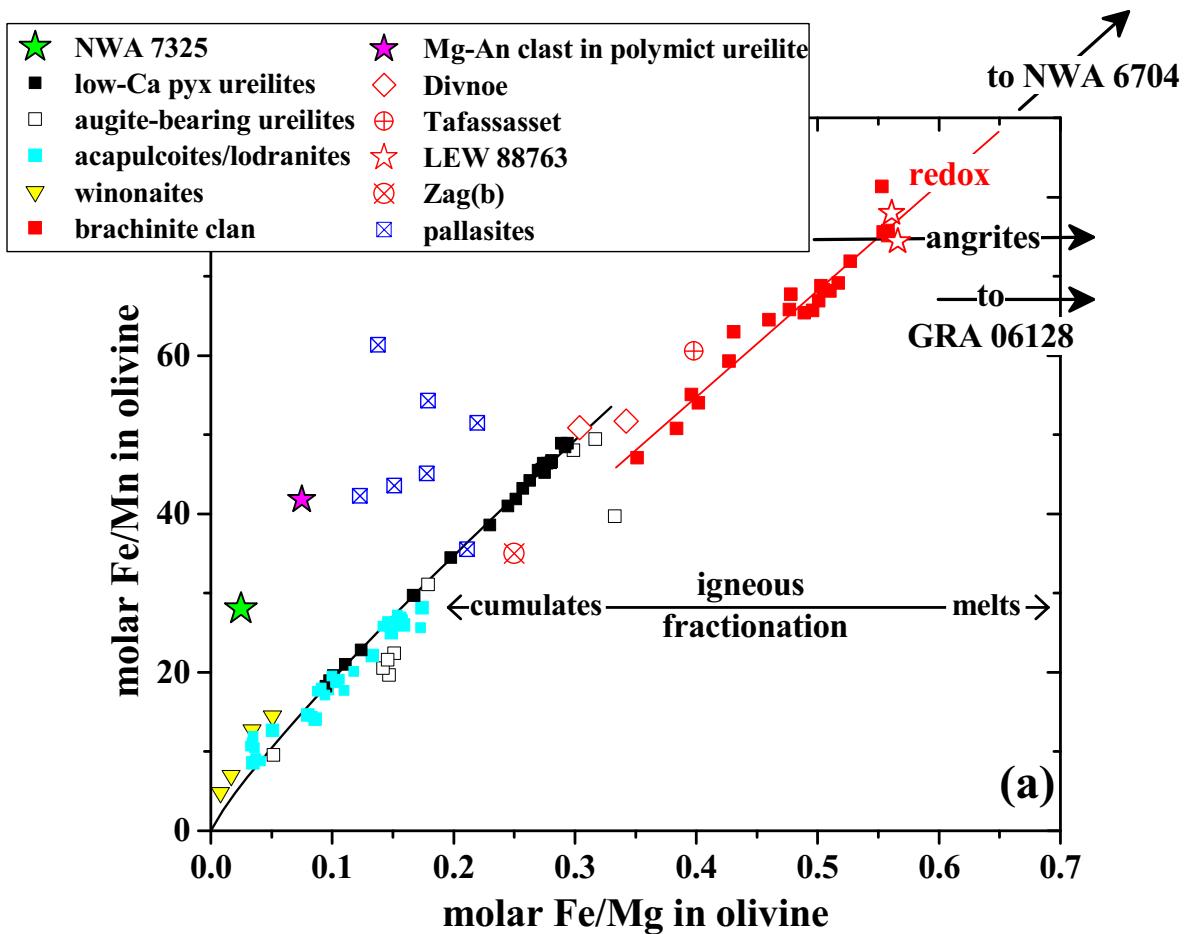


Fig. 3

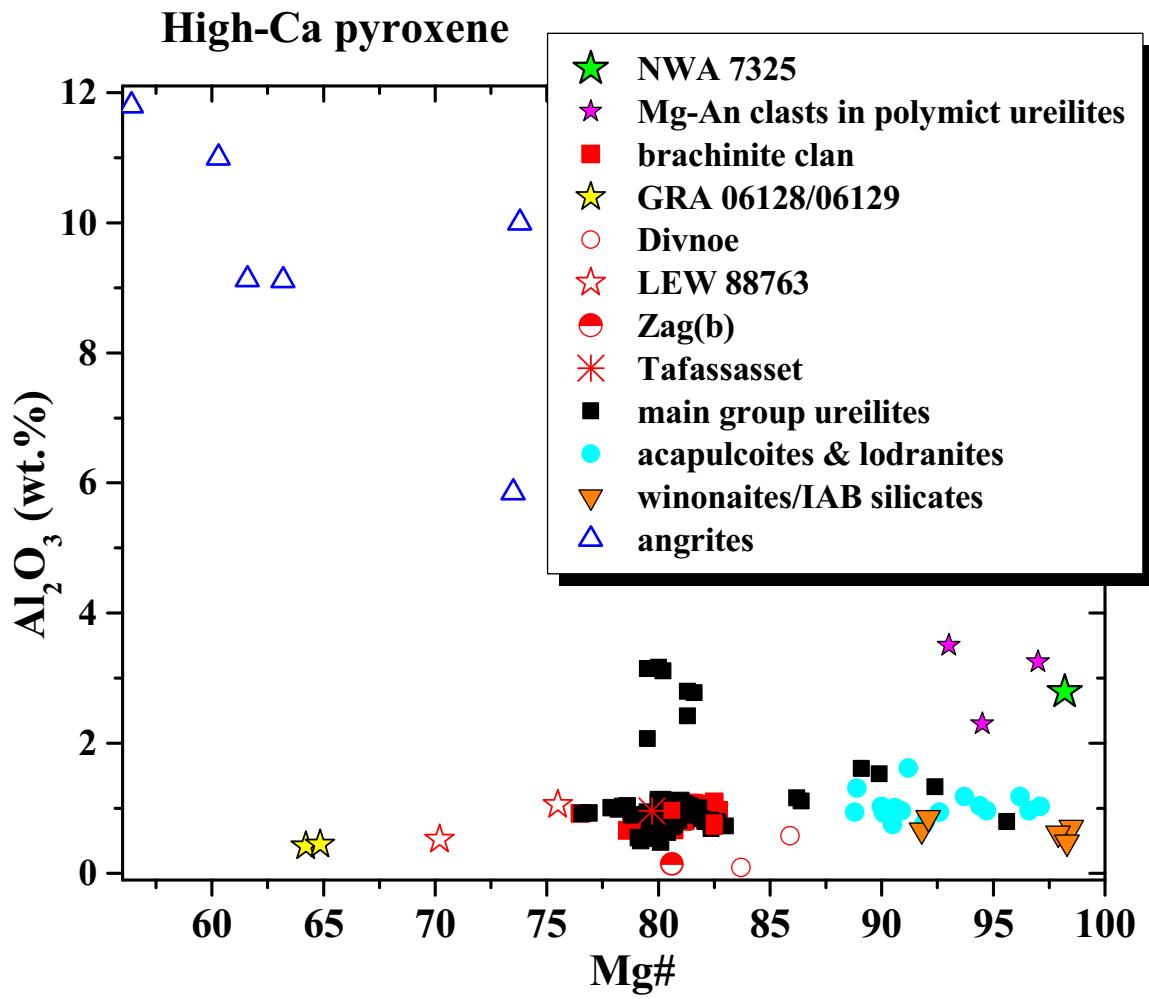


Fig. 4

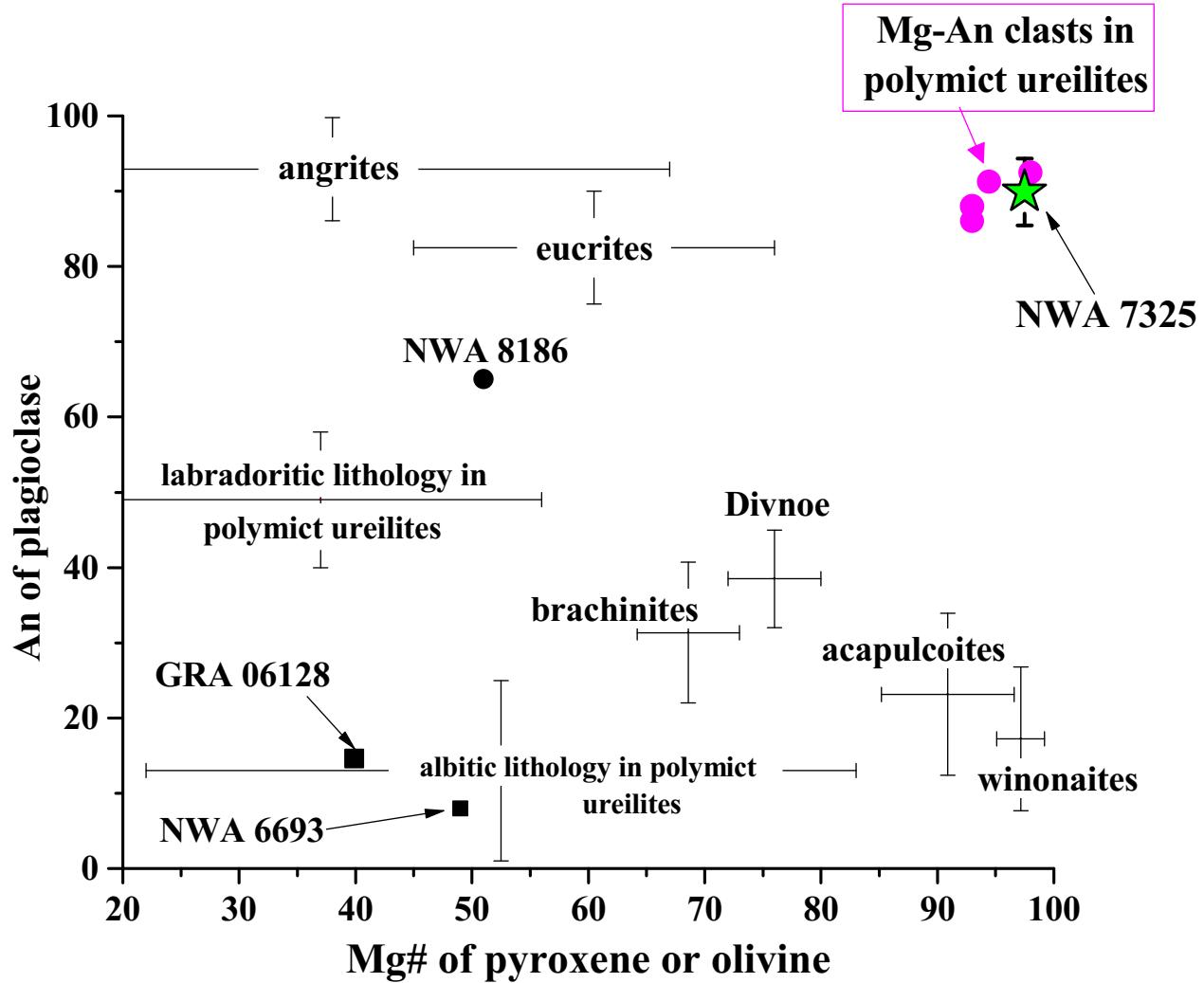


Fig. 5

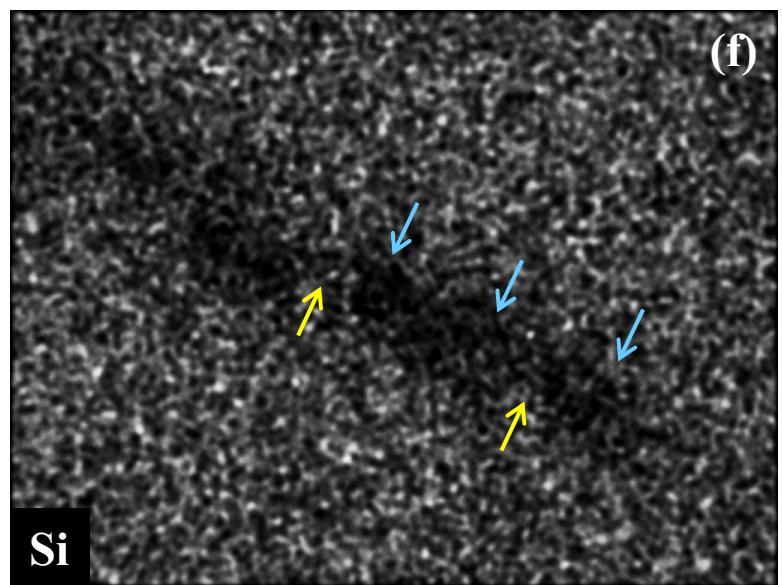
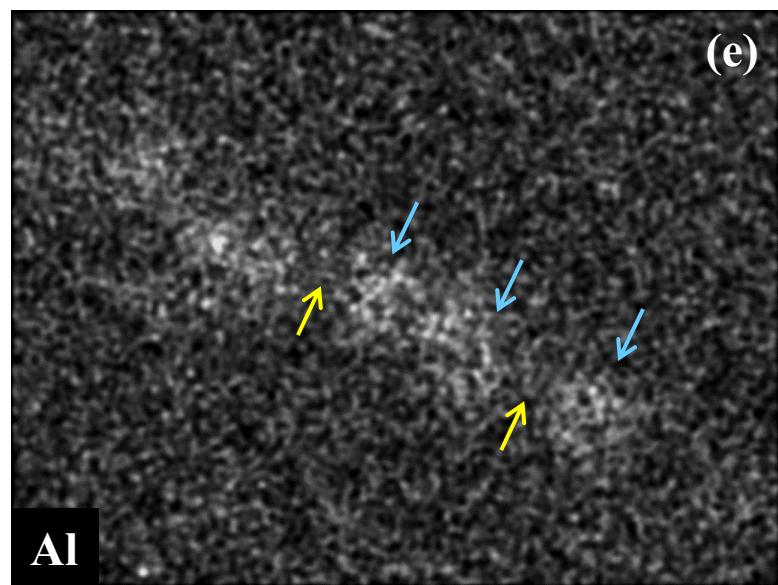
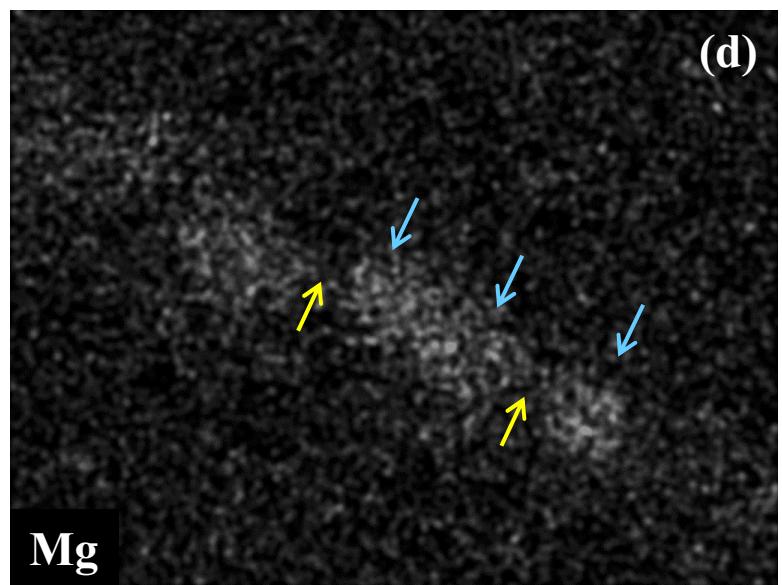
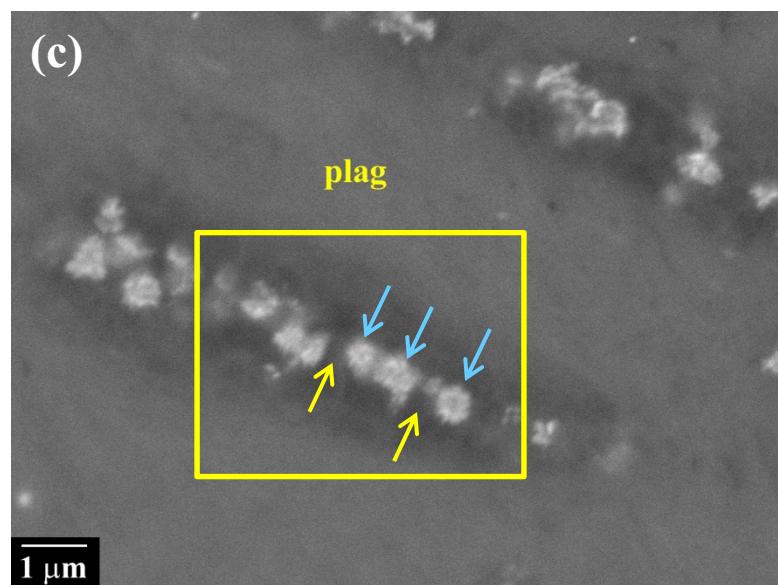
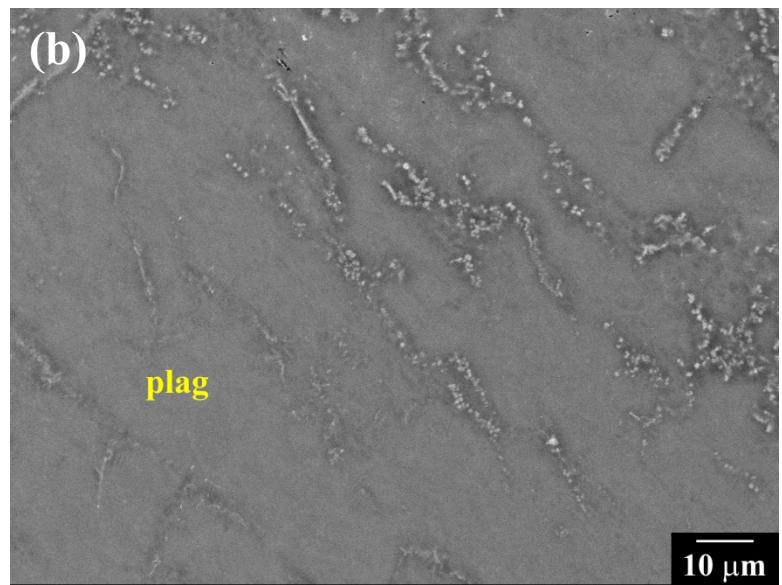
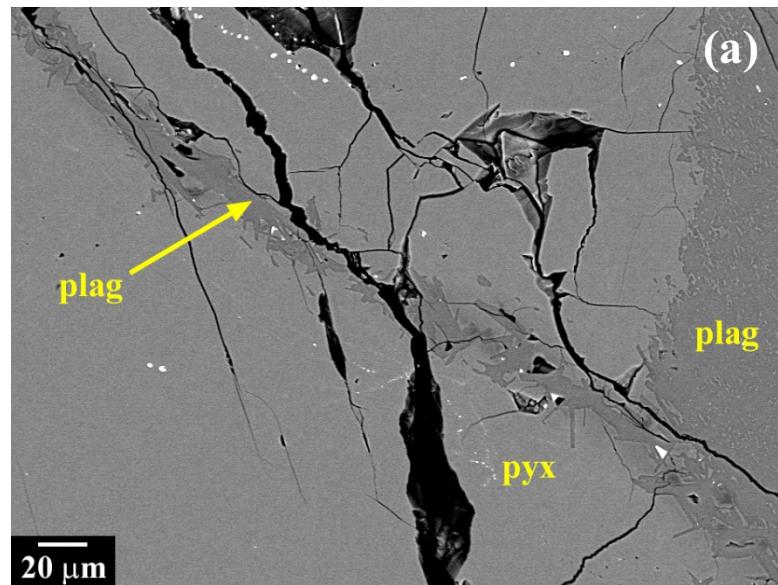


Fig. 6

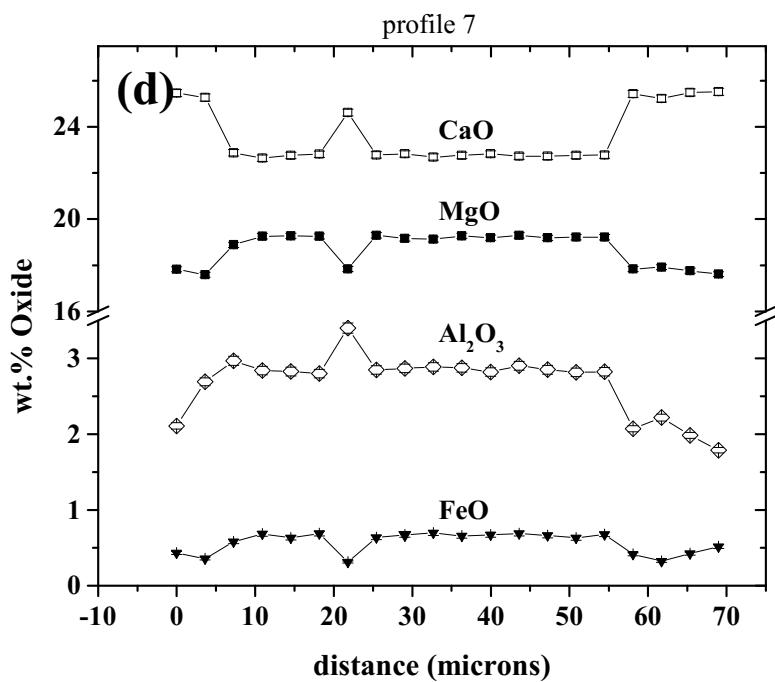
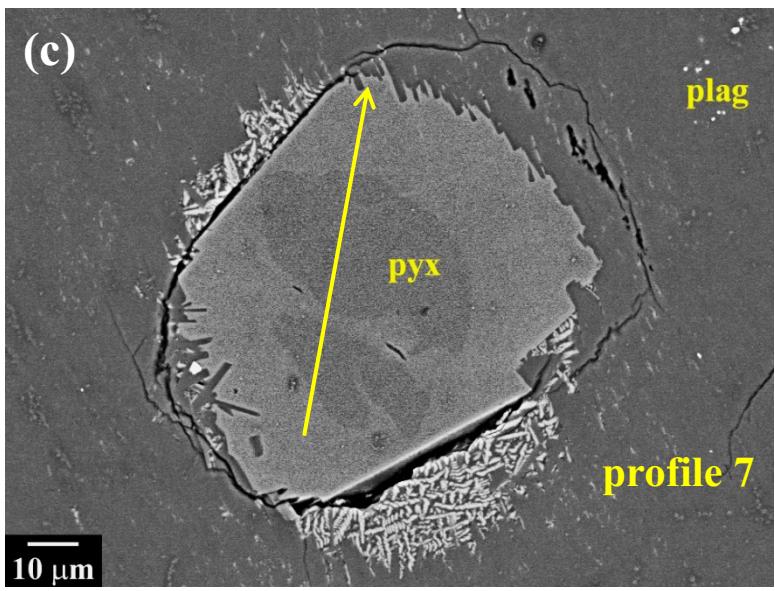
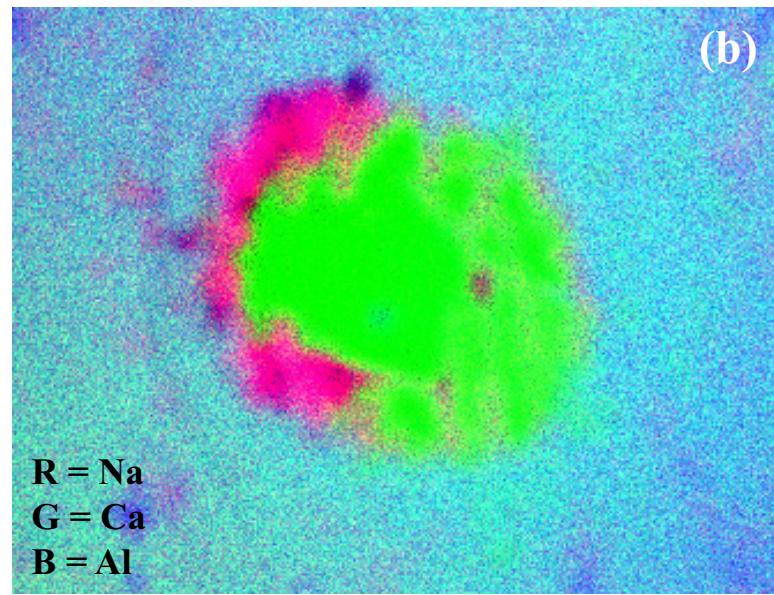
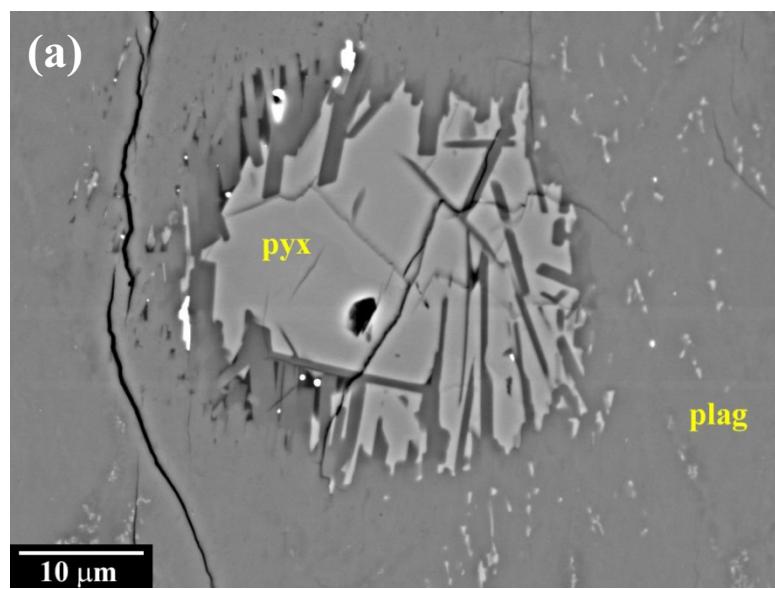


Fig. 7

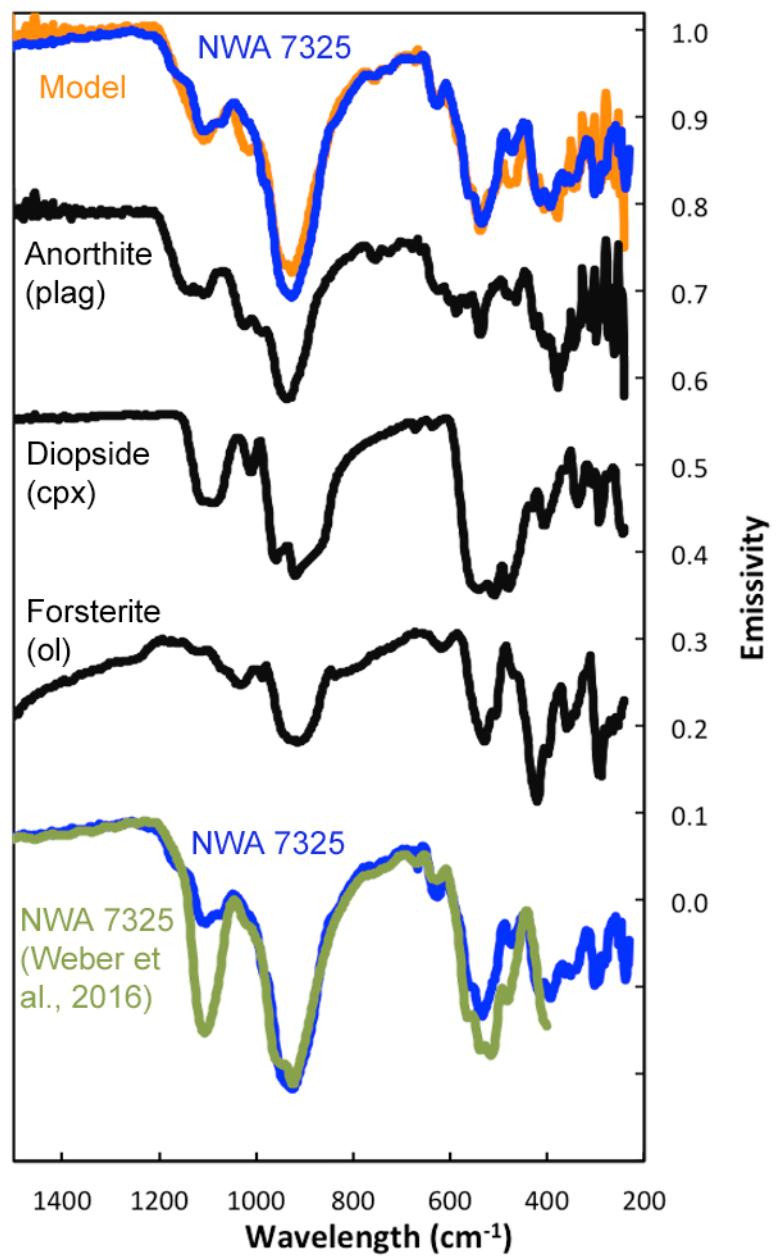


Fig. 8

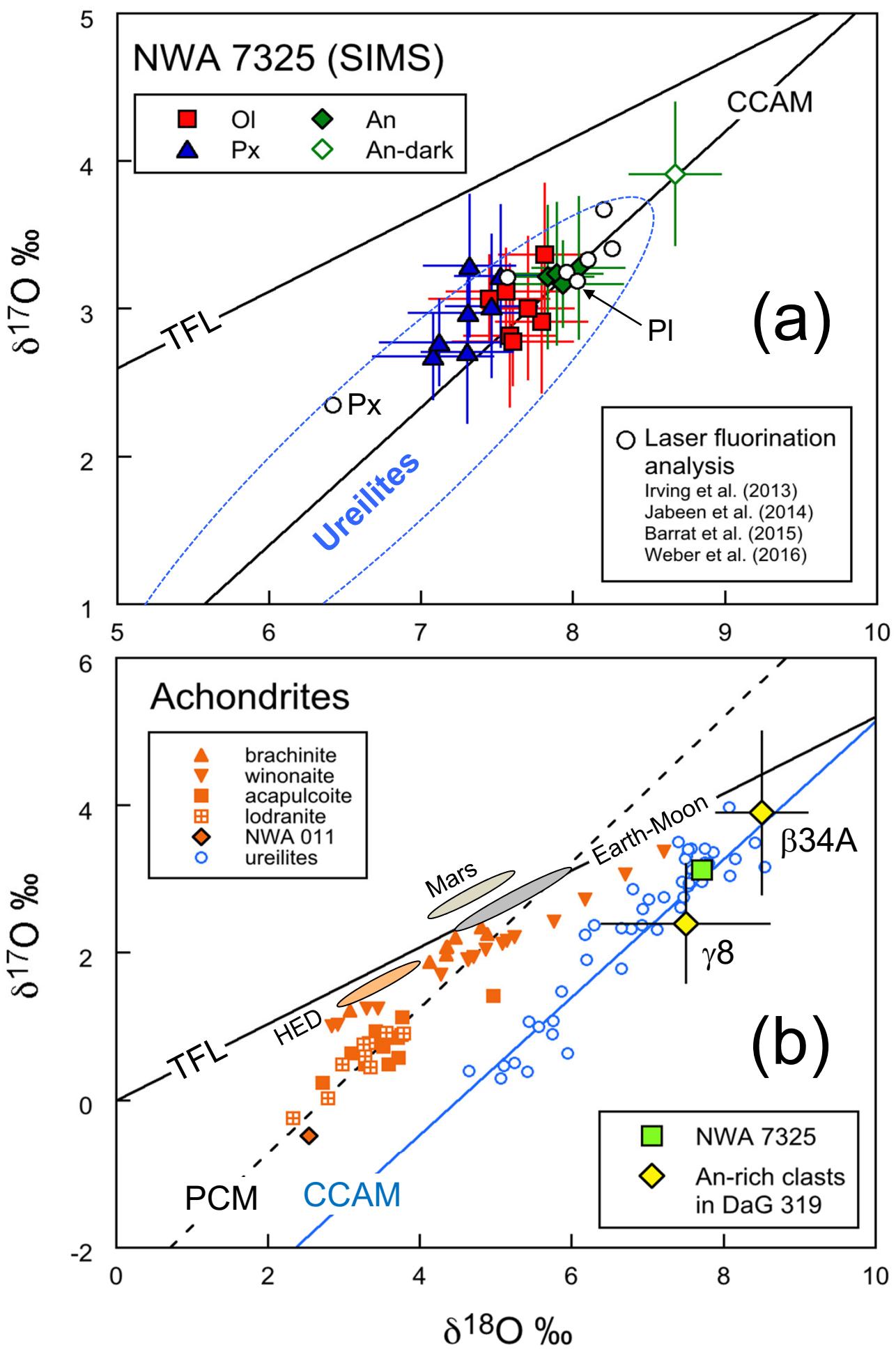


Fig. 9

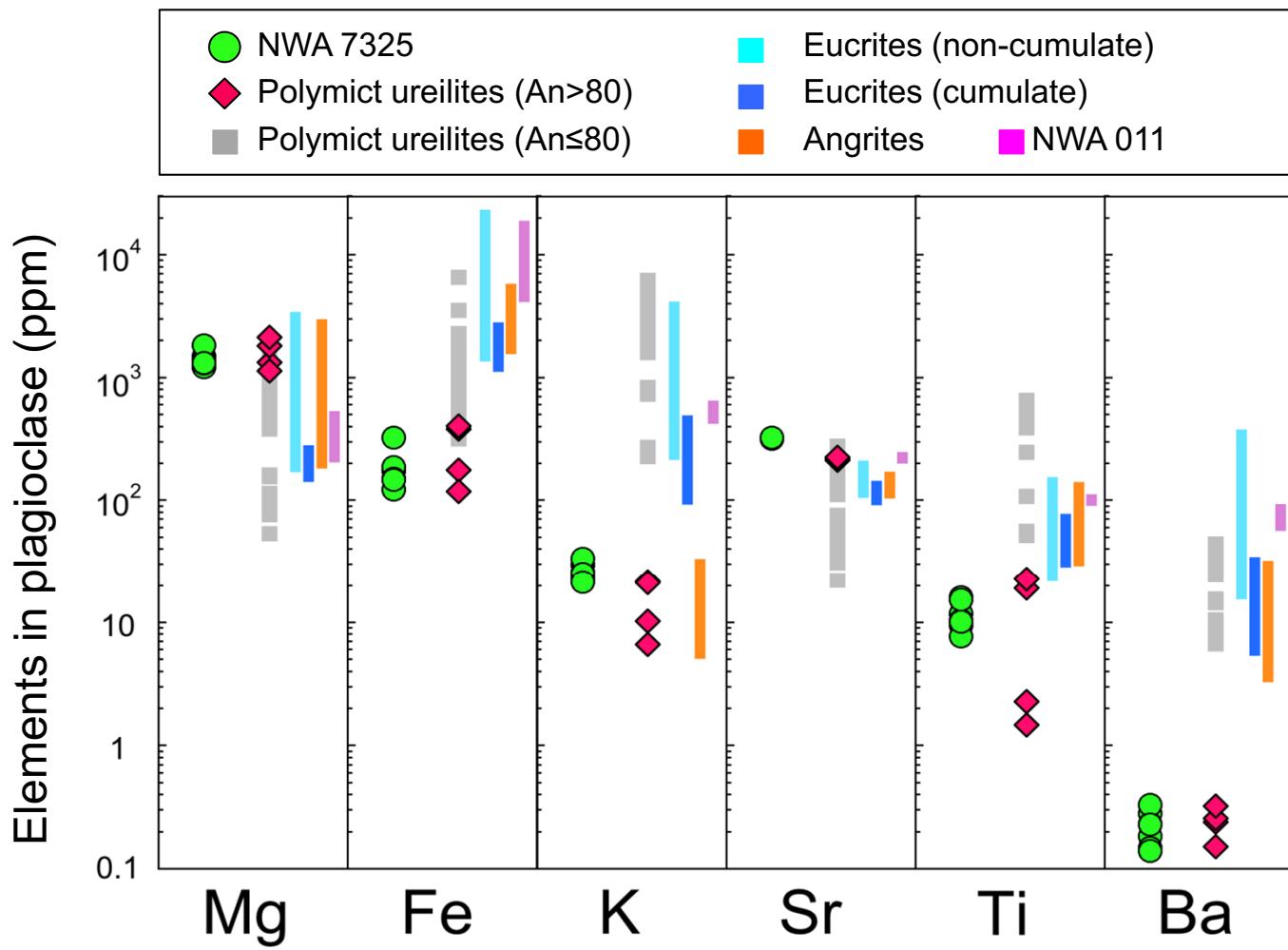


Fig. 10

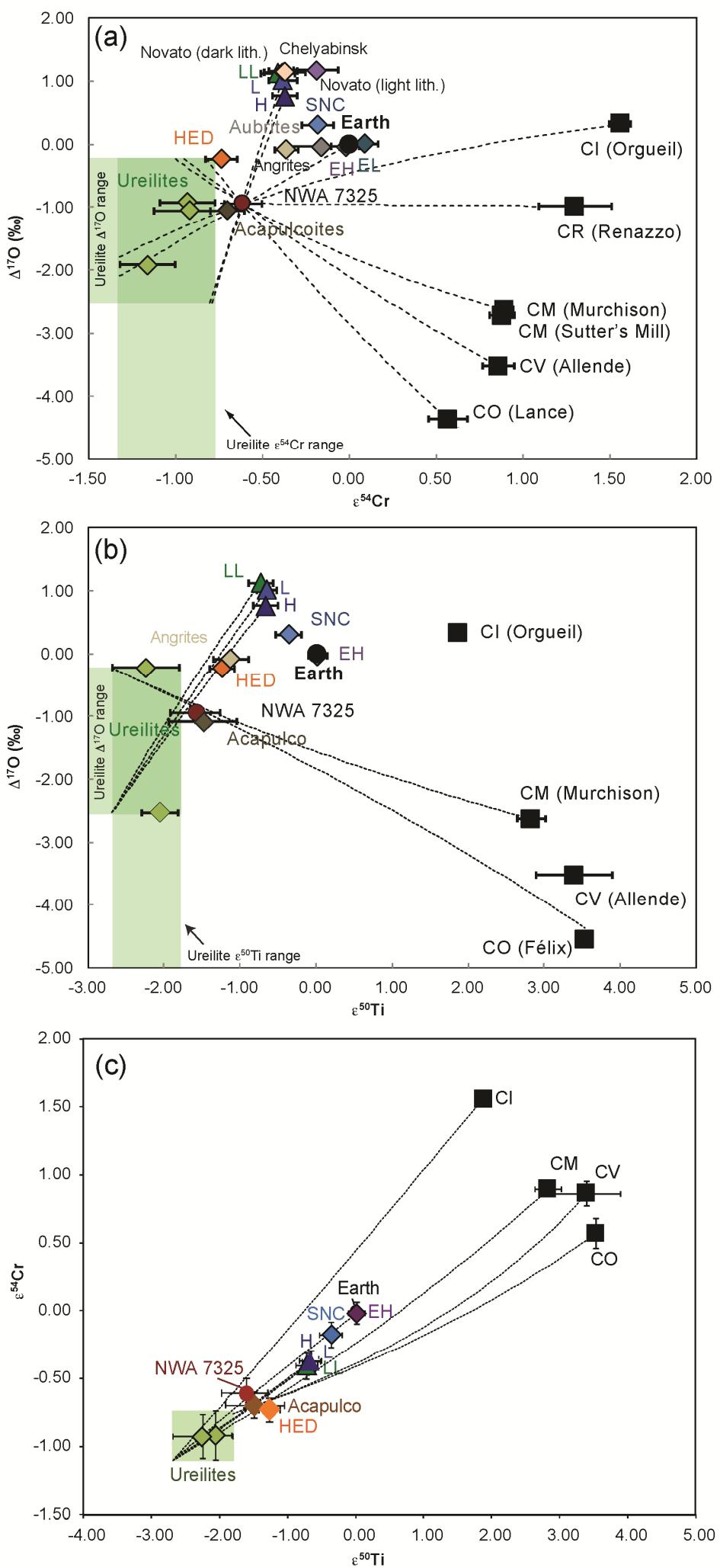


Fig. 11

# Clast 9 in NWA 10657\_003 polymict ureilite

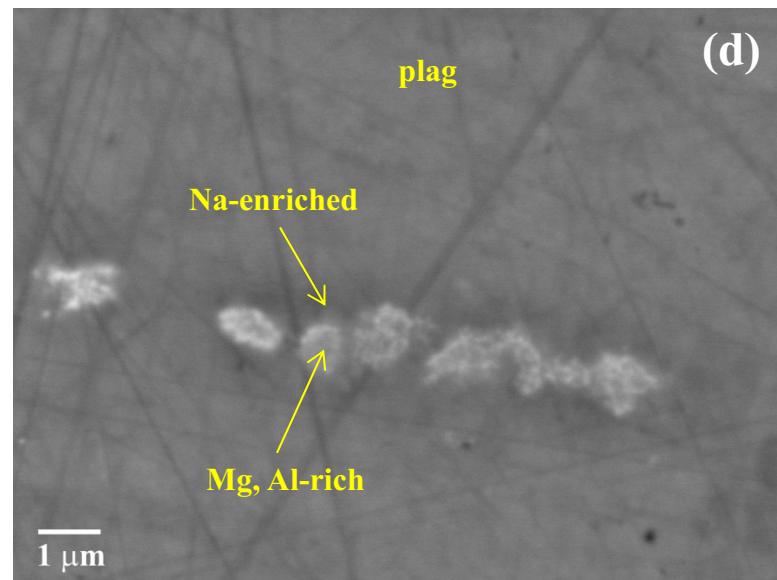
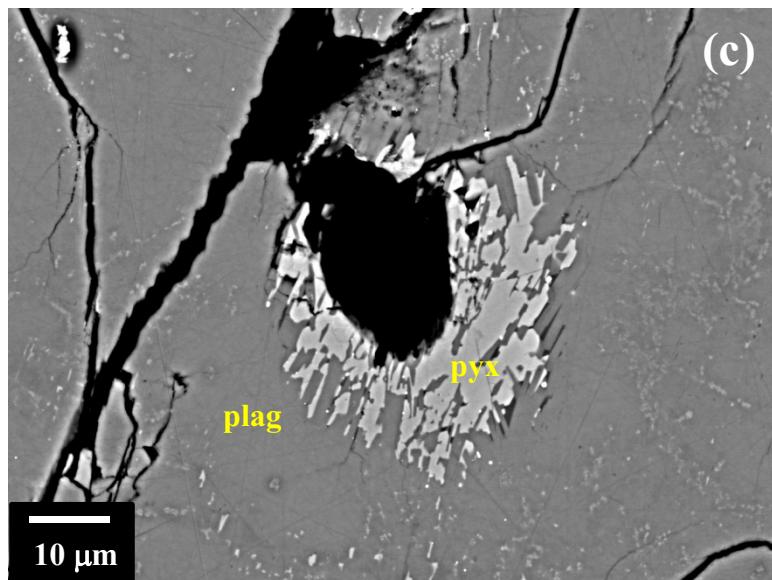
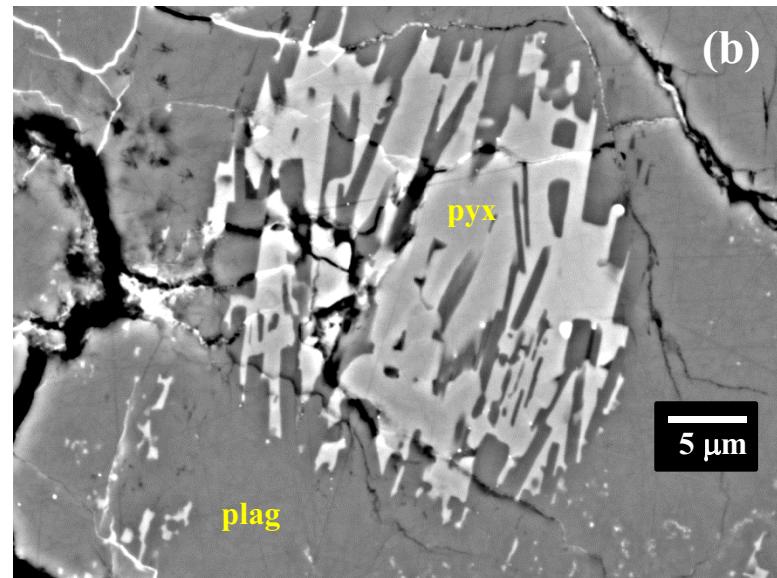
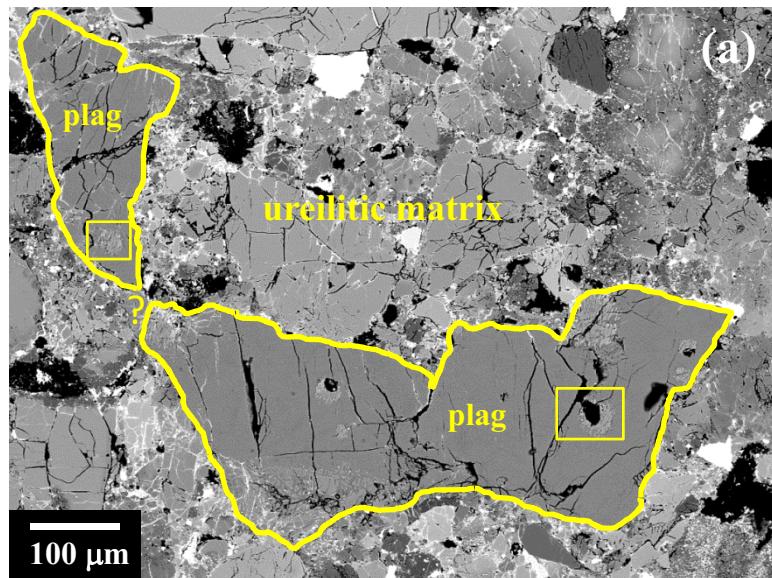


Fig. 12