

Impact of D1-V185 on the Water Molecules that facilitate O₂ Formation by the Catalytic Mn₄CaO₅ Cluster in Photosystem II

Christopher J Kim, Han Bao, Robert L. Burnap, and Richard J. Debus

Biochemistry, Just Accepted Manuscript • DOI: 10.1021/acs.biochem.8b00630 • Publication Date (Web): 26 Jun 2018

Downloaded from <http://pubs.acs.org> on July 9, 2018

Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.

Impact of D1-V185 on the Water Molecules that facilitate O₂ Formation by the Catalytic Mn₄CaO₅ Cluster in Photosystem II

*Christopher J. Kim[‡], Han Bao^{§, ⊥}, Robert L. Burnap[§], and Richard J. Debus^{‡, *}*

[‡]Department of Biochemistry, University of California, Riverside CA 92521,

[§]Department of Microbiology and Molecular Genetics, Oklahoma State University, Stillwater, OK 74078

Abstract

The oxidations of the O₂-evolving Mn₄CaO₅ cluster in Photosystem II are coupled to the release of protons to the thylakoid lumen via one or more proton egress pathways. These pathways are comprised of extensive networks of hydrogen-bonded water molecules and amino acid side chains. The hydrophobic residue, D1-V185, is adjacent to numerous water molecules in one of these networks. The D1-V185N mutation dramatically slows O-O bond formation. This impairment has been attributed to a disruption of the hydrogen bonded water molecules that are crucial for proton egress or whose rearrangement is required for catalysis. In this study, FTIR spectroscopy was employed to characterize the impact of the D1-V185N mutation on the carboxylate groups and water molecules that form a network of hydrogen bonds in this putative proton egress pathway. By analyzing carboxylate stretching modes, carbonyl stretching modes

1
2
3 of hydrogen-bonded carboxylic acids, O–H stretching modes of hydrogen-bonded water
4 molecules, and D–O–D bending modes, we obtain evidence that the D1-V185N mutation
5
6
7 perturbs the extensive network of hydrogen bonds that extends from Y_Z to D1-D61 to a greater
8
9
10 extent than any mutation yet examined, but does not alter the water molecules that interact
11
12 directly with D1-D61. The mutation also alters the environments of the carboxylate groups
13
14 whose pK_a values change in response to the S_1 to S_2 and S_2 to S_3 transitions. Finally, the
15
16 mutation alters the environment of the water molecule whose bending mode vanishes during the
17
18 S_2 to S_3 transition, consistent with assigning the Ca^{2+} -bound W3 as the water molecule that
19
20 deprotonates and joins oxo bridge O5 during the S_2 to S_3 transition, possibly as the second
21
22 substrate water molecule for O_2 formation.
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 Nearly all the molecular oxygen in the biosphere is produced by the catalytic Mn_4CaO_5
4 cluster in Photosystem II (PSII). Molecular oxygen is released as the by-product of oxidizing
5 water to provide the electrons and protons needed by plants, algae, and cyanobacteria to drive
6 ATP formation and CO_2 fixation. Our understanding of water oxidation by the Mn_4CaO_5 cluster
7 has advanced rapidly in the last five years because of new developments in X-ray
8 crystallography and the synergism between recent structural, computational, and advanced
9 biophysical studies (for review, see refs 1-7). PSII is a large protein complex that is integral to
10 the thylakoid membrane. The structure of PII has been determined to 1.85 – 1.95 Å in
11 cyanobacteria (8-10), 2.78 Å in a red alga (11), 2.7 Å in pea (12, 13), 3.2 Å in spinach (14), and
12 5.2 Å in *Arabidopsis thaliana* (15). PSII is dimeric *in vivo*, with each monomer containing 20 –
13 22 subunits and having a molecular weight of approximately 350 kDa. At the core of each
14 monomer is a heterodimer of two subunits known as D1 and D2. Light-induced separations of
15 charge within the D1/D2 heterodimer drive the oxidation of the Mn_4CaO_5 cluster, with tyrosine
16 Y_Z serving to transfer electrons from the Mn_4CaO_5 cluster to P_{680}^{*+} . Four electrons are removed
17 from the Mn_4CaO_5 cluster during each catalytic cycle, advancing the cluster through five
18 oxidation states. These states are termed S_n , where “n” denotes the number of oxidizing
19 equivalents stored ($n = 0 - 4$). The S_1 state predominates in dark-adapted samples. The S_4 state
20 is a transient intermediate whose formation triggers the utilization of the four stored oxidizing
21 equivalents to form O_2 from two substrate-water-derived metal ligands, the regeneration of the S_0
22 state, and the binding of at least one of two new substrate water molecules. The cluster’s metal
23 ions are ligated by six carboxylate groups, one histidine side chain, and four water molecules.
24 All but one of the amino acid ligands are provided by the D1 subunit.
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 In the highest resolution structures, the Mn_4CaO_5 cluster consists of a distorted Mn_3CaO_4
4 cube that is linked to a fourth Mn ion (denoted Mn4) by two oxo bridges (denoted O4 and O5,
5 see Figure 1). Considerable evidence supports identifying O5 as the deprotonated form of one of
6 the two substrate water molecules (16-19). It is present as a μ -hydroxo bridge in the S_0 state (19)
7 and as a fully deprotonated μ -oxo bridge in the S_2 state (16). The Mn_4CaO_5 cluster is structurally
8 flexible. In the S_2 state it interconverts between two nearly isoenergetic conformations (20). In
9 one conformation, O5 coordinates to Mn4, leaving Mn1 five-coordinate. In the second
10 conformation, O5 coordinates to Mn1, leaving Mn4 five-coordinate. During the S_2 to S_3
11 transition, an additional oxygen joins O5 between Mn4 and Mn1 (21-25). This additional
12 oxygen [denoted O6 (25)] has been proposed to correspond to a water molecule that deprotonates
13 and relocates from the Mn4-bound W2 position (26-32) or the Ca^{2+} -bound W3 position (33-38).
14 It is thought to correspond to a deprotonated form of the second substrate water molecule (22,
15 23, 25, 27, 29-31, 33-37, 39-42). Its relocation to a position between Mn4 and Mn1 and adjacent
16 to O5 during the S_2 to S_3 transition would ensure that the two substrate oxygen atoms have
17 opposite spins as the O-O bond is formed in the S_4 state, substantially decreasing the activation
18 energy for bond formation (22, 23, 39, 43). The relocation also would ensure that the two
19 oxygen atoms that will form the O-O bond are held in close proximity only after the S_3 state is
20 fully formed, thereby preventing deleterious catalase-like activity in the lower S states (23, 41).
21 The subsequent formation of the $Y_Z^*S_3$ state locks the two oxygen atoms in position for O-O
22 bond formation in the S_4 state (44).
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

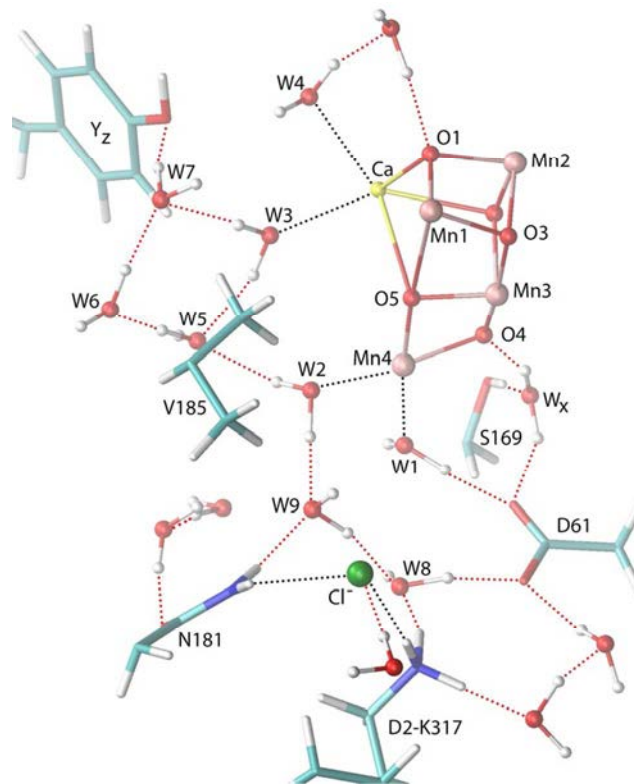


FIGURE 1. Water molecules and selected residues in the vicinity of the Mn₄CaO₅ cluster.

Residues are from the D1 subunit unless noted otherwise. Salmon-colored spheres, manganese ions; yellow sphere, calcium; green sphere, chloride; red spheres, oxygen atoms of μ -oxo bridges and water molecules. The coordinates for this figure were constructed with QM/MM methods (45) based on the coordinates in the 1.9 Å X-ray crystallographic model (8) and were kindly provided by V. S. Batista and coworkers.

The oxidations of the Mn₄CaO₅ cluster are coupled to the release of protons, with electrons and protons being removed alternatively as the S state cycle proceeds (46-48). The alternating deprotonations and oxidations prevent the redox potential of the Mn₄CaO₅ cluster from rising to levels that prevent its subsequent oxidation by the Y_Z[•] radical anion. Deprotonation of the Mn₄CaO₅ cluster takes place via one or more proton egress pathways that

1
2
3 link the Mn_4CaO_5 with the thylakoid lumen. These pathways are comprised of networks of
4 hydrogen-bonded water molecules and protonatable amino acid side chains [e.g., refs 49-57].
5
6 Elements of these networks are undoubtedly dynamic in nature (51, 58). One such network
7
8 includes a series of hydrogen-bonded water molecules that stretches from Y_Z to D1-D61 (see
9
10 Figure 1). D1-D61 is the initial residue of a proton egress pathway that facilitates the S_3 to S_4
11
12 transition (59-65). Adjacent to numerous water molecules in this network is the hydrophobic
13
14 residue, D1-V185. The side-chain of this residue lies 3.7 Å from O5 and 3-5 Å from waters W2,
15
16 W3, W5, W6, and W7. The D1-V185N mutation has been shown to dramatically slow the rate
17
18 of O_2 release and extend the kinetic lag phase of the S_3 to S_4 transition (66). This lag phase
19
20 occurs prior to Mn reduction/ O_2 formation has been attributed to proton movement (46-48, 67,
21
22 68) and/or a structural rearrangement (69). The alterations caused by the D1-V185N mutation
23
24 have been attributed to a disruption of the hydrogen bonded water molecules in the proton egress
25
26 pathway that facilitates the S_3 to S_4 transition (66). Indeed, the D1-V185N mutation was
27
28 constructed to disrupt the orientation and dynamics of these water molecules (66).
29
30
31
32
33
34

35 FTIR difference spectroscopy is an ideal method for characterizing structural changes
36
37 (70-73) and has been employed extensively to study the S state cycle in PSII (74-80). FTIR is
38
39 especially suited for probing changes in hydrogen bonding. In PSII, the C=O stretching modes
40
41 of hydrogen-bonded carboxylic acids have been monitored in the 1760 – 1730 cm^{-1} region (63,
42
43 65, 81-85), the N-H stretching mode of a protonated histidyl group has been monitored in the
44
45 3000 – 2400 cm^{-1} region (86), the O-H stretching modes of weakly hydrogen-bonded water
46
47 molecules have been monitored in the 3700 – 3500 cm^{-1} region (65, 84, 85, 87-94), the O-H
48
49 stretching modes of strongly hydrogen-bonded water molecules have been monitored in the
50
51 3200 – 2500 cm^{-1} region (65, 84, 85, 88, 93-95), and D-O-D bending modes have been
52
53
54
55
56
57
58
59
60

1
2
3 monitored in the 1250 – 1150 cm⁻¹ region (65, 94, 96). In this study, we employed FTIR
4 spectroscopy to characterize the impact of the D1-V185N mutation on the carboxylate groups
5 and water molecules that form part of a putative proton egress pathway leading from the
6 Mn₄CaO₅ cluster to the thylakoid lumen. We find that the D1-V185N mutation perturbs this
7 network to a greater extent than any other mutation yet examined, alters the environments of the
8 carboxylate groups whose pK_a values change in response to the S₁ to S₂ and S₂ to S₃ transitions,
9 and alters the environment of the water molecule whose D-O-D bending mode disappears during
10 the S₂ to S₃ transition. The latter observation is consistent with identifying W3 as the water
11 molecule that deprotonates and joins O5 during the S₂ to S₃ transition.
12
13
14
15
16
17
18
19
20
21
22
23
24
25

26 MATERIALS AND METHODS

27
28 *Construction of Mutant and Propagation of Cultures.* Cells and thylakoid membranes of
29 the D1-V185N mutation-bearing strain of *Synechocystis* sp. PCC 6803 have been characterized
30 previously (66). The mutation was constructed in the *psbA-2* gene of *Synechocystis* sp. PCC
31 6803 and transformed into a strain of *Synechocystis* that lacks all three *psbA* genes and contains a
32 hexahistidine-tag (His-tag) fused to the C-terminus of CP47 (97). The wild-type strain described
33 in this manuscript was constructed in an identical manner as the D1-V185N strain but with a
34 transforming plasmid that carried no mutation. Large-scale cultures were propagated as
35 described previously (98).
36
37
38
39
40
41
42
43
44
45
46

47 *Purification of PSII core complexes.* Oxygen-evolving PSII core complexes were purified
48 as described previously in buffer containing 1.2 M betaine, 10% (v/v) glycerol, 50 mM MES-
49 NaOH (pH 6.0), 20 mM CaCl₂, 5 mM MgCl₂, 50 mM histidine, 1 mM EDTA, and 0.03% (w/v)
50 *n*-dodecyl β-D-maltoside (65). These were concentrated to approx. 1 mg of Chl/mL, flash-
51
52
53
54
55
56
57
58
59
60

1
2
3 frozen, and stored at $-80\text{ }^{\circ}\text{C}$.
4

5 *Preparation of FTIR samples.* PSII core complexes were transferred into 40 mM
6 sucrose, 10 mM MES-NaOH (pH 6.0), 5 mM CaCl_2 , 5 mM NaCl, 0.06% (w/v) *n*-dodecyl β -D-
7 maltoside, concentrated, mixed with 1/10 volume of fresh 100 mM potassium ferricyanide, then
8 dried lightly with dry N_2 gas (94). Samples were then rehydrated and maintained at a relative
9 humidity of 95% by spotting six 1 μL droplets of 40% (v/v) glycerol in water around the
10 window's periphery before a second window was placed on the first with a thin o-ring spacer in
11 between. For samples having natural abundance H_2^{16}O exchanged for D_2^{16}O or D_2^{18}O , the
12 lightly dried samples were rehydrated with 40% (v/v) glycerol(OD)₃ (98% D, Cambridge Isotope
13 Laboratories, Inc., Andover, MA) in D_2^{16}O (99.9% D, Cambridge Isotope Laboratories, Inc.,
14 Andover, MA) or D_2^{18}O (98% D, 97% ^{18}O , Cambridge Isotope Laboratories, Inc., Andover,
15 MA), respectively (65, 94, 96). Sealed samples were equilibrated in the FTIR sample
16 compartment at 0°C in darkness for 1.5 h, illuminated with 6 pre-flashes, then dark-adapted for
17 30 min (65, 94). For each sample, the absorbance at 1657 cm^{-1} (amide I band) was 0.6 – 1.1.
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35

36 *FTIR Spectroscopy.* Spectra were recorded with a Bruker Vertex 70 spectrometer
37 (Bruker Optics, Billerica, MA) outfitted with a pre-amplified, midrange D317 photovoltaic MCT
38 detector (Kolmar Technologies, Inc., Newburyport, MA), as described in ref. (94). After dark
39 adaptation, samples were given six flashes at 13 s intervals. Two transmission spectra were
40 recorded before the first flash and one transmission spectrum was recorded starting 0.33 s after
41 the first and subsequent flashes (each transmission spectrum consisted of 100 scans). The 0.33 s
42 delay was included to allow the oxidation of $\text{Q}_\text{A}^{\bullet-}$ by the ferricyanide. Difference spectra of the
43 successive S-state transitions (e.g., $\text{S}_{\text{n}+1}$ -minus- S_n difference spectra, henceforth written
44 $\text{S}_{\text{n}+1}$ - S_n), were obtained by dividing the transmission spectrum obtained after the n^{th} flash by the
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 transmission spectrum obtained before the n^{th} flash, then converting the ratio to units of
4
5 absorption. The background noise level and the stability of the baseline were obtained by
6
7 dividing the second pre-flash transmission spectrum by the first and converting the ratio to units
8
9 of absorption (these spectra are labeled dark–dark in each figure – these are control difference
10
11 spectra obtained *without* a flash being given). The sample was then dark-adapted for 30 min and
12
13 the cycle was repeated. For each sample, the illumination cycle was repeated 15 times. The
14
15 spectra of 20-61 samples were averaged (see figure legends). The amplitudes of the D1-V185N
16
17 difference spectra were multiplied by factors of 1.2 to 2.4 to normalize the peaks corresponding
18
19 to the reduction of ferricyanide to ferrocyanide by Q_A^- (at 2115 and 2037 cm^{-1} , respectively) to
20
21 those in the corresponding wild-type spectra (this procedure normalizes the spectra to the extent
22
23 of flash-induced charge-separation).
24
25
26
27

28 *Other Procedures.* Chlorophyll concentrations were determined as described in ref. (99).
29
30
31
32

33 RESULTS

34
35 **Midfrequency Region.** The mid-frequency difference spectra produced by four flashes
36
37 given to wild-type and D1-V185N PSII core complexes are compared in Figure 2 (black and red
38
39 traces, respectively). The spectra produced by the first, second, third, and fourth flashes of the
40
41 wild-type samples closely resemble spectra presented previously for PSII preparations from
42
43 *Synechocystis* sp. PCC 6803, *Thermosynechococcus elongatus*, and spinach, and correspond
44
45 predominantly to S_2 - S_1 , S_3 - S_2 , S_0 - S_3 , and S_1 - S_0 difference spectra, respectively [e.g., refs. (74-76,
46
47 78-80, 100-102)]. The S_2 - S_1 spectrum of D1-V185N PSII core complexes (upper red trace in
48
49 Figure 2) showed substantial changes in throughout the mid-frequency region. In the amide I
50
51 region, the 1707(-) cm^{-1} feature was enhanced and a positive feature appeared at 1677 cm^{-1} in
52
53
54
55
56
57
58
59
60

1
2
3 place of the 1681(-) cm^{-1} and 1672(+) cm^{-1} features of the wild-type. In the asymmetric
4
5 carboxylate stretching [$\nu_{\text{asym}}(\text{COO}^-)$] and amide II region, the 1629(-) cm^{-1} feature was
6
7 enhanced, a positive feature appeared at 1603 cm^{-1} , the 1587(+) cm^{-1} , 1560(-) cm^{-1} , and 1544(-)
8
9 cm^{-1} features were diminished substantially, a negative feature appeared at 1573 cm^{-1} , and the
10
11 1531(+)/1523(-) cm^{-1} derivative feature was eliminated. The features in the symmetric
12
13 carboxylate stretching [$\nu_{\text{sym}}(\text{COO}^-)$] region were diminished: the 1416(-)/1410 (+) cm^{-1}
14
15 derivative feature shifted to 1423(-)/1416(+) cm^{-1} , the 1400 (-) cm^{-1} feature shifted to 1388(-)
16
17 cm^{-1} , the 1364(+) cm^{-1} and 1354(-) cm^{-1} features shifted to 1368(+) cm^{-1} and 1360(-) cm^{-1} ,
18
19 respectively, and a small positive feature appeared at 1288 cm^{-1} . Also, the 1260(+)/1250(-) cm^{-1}
20
21 feature was diminished. In the carbonyl stretching [$\nu(\text{C=O})$] region, the 1746(-) cm^{-1} feature was
22
23 diminished and replaced by a negative feature at 1737 cm^{-1} . An expanded view of this region is
24
25 shown in Figure 3 (top left panel).
26
27
28
29
30

31 The $\text{S}_3\text{-S}_2$ spectrum of D1-V185N also showed changes throughout the mid-frequency
32
33 region (second set of traces in Figure 2). In the amide I region, a negative feature appeared at
34
35 1704 cm^{-1} , the 1697(+) cm^{-1} feature was enhanced, the 1687(-) cm^{-1} and 1675(-) cm^{-1} features
36
37 were diminished, and the 1675(-)/1666(+)/1659(-) cm^{-1} feature was replaced with a
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

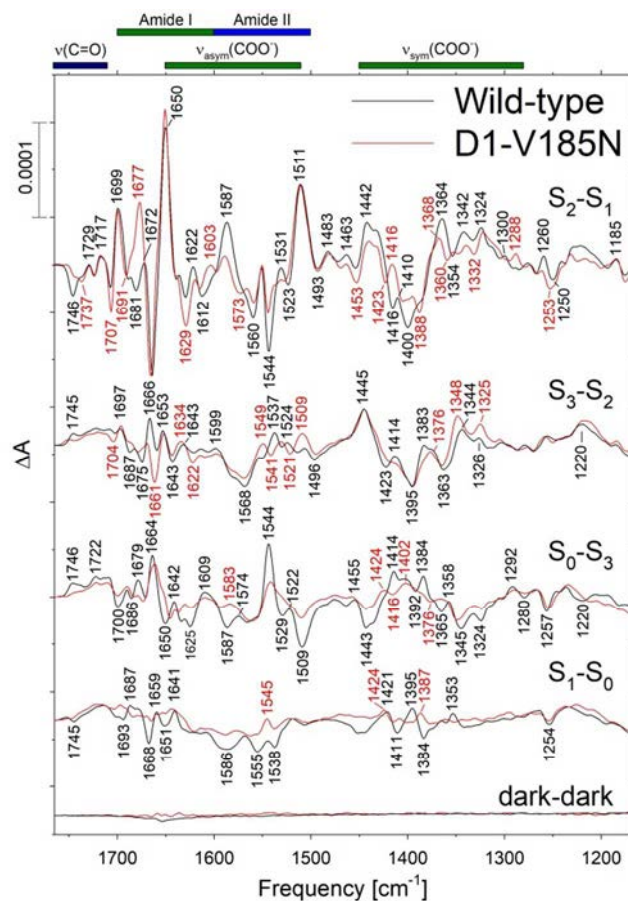


FIGURE 2. Mid-frequency FTIR difference spectra of wild-type (black) and D1-V185N (red) PSII core complexes in response to four successive flash illuminations applied at 0°C. The data (plotted from 1770 cm^{-1} to 1170 cm^{-1}) represent the averages of 22 wild-type and 20 D1-V185N samples (33,000 and 30,000 scans, respectively). The D1-V185N spectra were multiplied vertically by factors of 1.2 to 1.4 to approximately normalize the wild-type and mutant spectra to the extent of flash-induced charge separation. The dark-dark control trace of the mutant (lower red trace) also multiplied by a factor of 1.4. The dark-dark control traces show the noise level and the stability of the baseline.

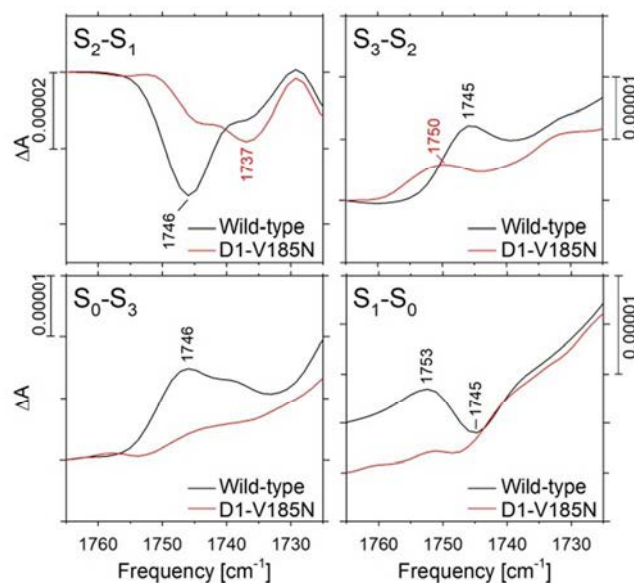


FIGURE 3. FTIR difference spectra of wild-type (black) and D1-V185N (red) PSII core complexes in the $\nu(\text{C}=\text{O})$ region. The data are expanded from those shown in Figure 2.

negative feature at 1661 cm^{-1} . In the $\nu_{\text{asym}}(\text{COO}^-)$ /amide II region, a negative feature appeared at 1622 cm^{-1} , and the $1537(+)\text{ cm}^{-1}$ and $1524(+)\text{ cm}^{-1}$ features were replaced with a $1549(+)/1541(-)/1531(+)/1521(-)/1509(+)\text{ cm}^{-1}$ feature. In the $\nu_{\text{sym}}(\text{COO}^-)$ region, the $1423(-)/1414(+)\text{ cm}^{-1}$ derivative feature was eliminated, the $1383(+)\text{ cm}^{-1}$ feature shifted to 1376 cm^{-1} , and the $1344(+)\text{ cm}^{-1}$ feature shifted to 1348 cm^{-1} . In the $\nu(\text{C}=\text{O})$ region, the $1746(+)\text{ cm}^{-1}$ feature was diminished and shifted to 1750 cm^{-1} . An expanded view of this region is shown in Figure 3 (top right panel).

1
2
3 The S_0 - S_3 spectrum of D1-V185N was diminished substantially in amplitude and showed
4 substantial changes throughout the mid-frequency region (Figure 2, 3rd set of traces). In the
5
6 amide I region, the $1700(-)$ cm^{-1} and $1679(+)$ cm^{-1} features were nearly eliminated. In the
7
8 $\nu_{\text{asym}}(\text{COO}^-)$ /amide II region, the $1650(-)$ cm^{-1} feature shifted to 1646 cm^{-1} , the $1642(+)$ cm^{-1} and
9
10 $1625(-)$ cm^{-1} features were eliminated, the $1587(-)$ cm^{-1} feature was replaced with a $1583(+)$ cm^{-1}
11
12 feature, the $1544(+)$ cm^{-1} and $1509(-)$ cm^{-1} features were sharply diminished, and the $1529(-)$
13
14 $/1522(+)$ cm^{-1} feature was eliminated. In the $\nu_{\text{sym}}(\text{COO}^-)$ region, the $1443(-)$ cm^{-1} feature was
15
16 sharply diminished, and the $1414(+)/1392(-)/1384(+)/1365(-)/1358(+)$ cm^{-1} feature was replaced
17
18 with a $1424(+)/1416(-)/1402(+)/1376(-)/1365(+)$ cm^{-1} feature, the $1345(-)$ cm^{-1} feature shifted to
19
20 $1349(-)$ cm^{-1} , and the $1324(-)$ and $1280(-)$ cm^{-1} features were diminished. In the $\nu(\text{C=O})$ region,
21
22 the $1746(+)$ cm^{-1} feature was eliminated (see expanded view in Figure 3, lower left panel).
23
24
25
26
27
28

29 The S_1 - S_0 spectrum of D1-V185N also was diminished substantially in amplitude and
30 showed substantial changes throughout the mid-frequency region (Figure 2, 4th set of traces). In
31
32 the $\nu_{\text{sym}}(\text{COO}^-)$ region, the $1421(+)$ cm^{-1} and $1395(+)$ cm^{-1} features shifted to $1424(+)$ cm^{-1} and
33
34 $1387(-)$ cm^{-1} , respectively. Most other features throughout in the mid-frequency region were
35
36 sharply diminished or eliminated. In the $\nu(\text{C=O})$ region, the $1753(+)/1745$ cm^{-1} derivative
37
38 feature was eliminated (see expanded view in Figure 3, lower right panel).
39
40
41
42

43 **Strongly H-bonded O–H stretching region.** The O–H stretching vibrations of strongly
44 H-bonded OH groups are observed as very broad positive features between 3200 and 2500 cm^{-1}
45
46 ($65, 84, 85, 88, 93-95$). These features have been observed in PSII core complexes from
47
48 *Synechocystis* sp. PCC 6803 and *Thermosynechococcus elongatus* and in PSII membranes from
49
50 spinach and are diminished or eliminated (presumably downshifted) in the presence of D_2O ($65,$
51
52 88). These regions of wild-type and D1-V185N PSII core complexes are compared in Figure 4.
53
54
55
56
57
58
59
60

1
2
3 In the S_2 - S_1 spectrum (Figure 4, upper traces), the broad feature is overlain with numerous
4 positive features that correspond to C–H stretching vibrations from aliphatic groups and N–H
5 stretching vibrations and their Fermi resonance overtones from the imidazole group(s) of one or
6 more histidine residues (87, 88, 103). Our data show that the D1-V185N mutation causes only
7 slight perturbations in this region of any of the difference spectra.
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

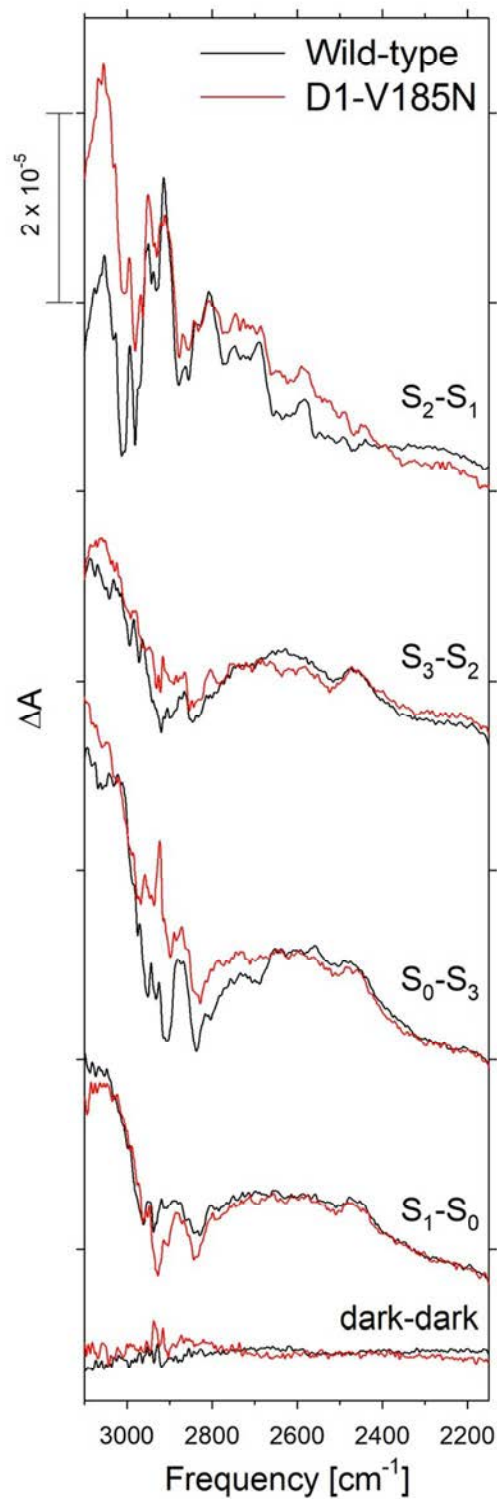


FIGURE 4. FTIR difference spectra of wild-type (black) and D1-V185N (red) PSII core complexes between 3100 and 2150 cm⁻¹ in response to four successive flash illuminations applied at 0°C. The data were collected simultaneously with that shown in Figure 2. The spectra

1
2
3 of the D1-V185N samples were multiplied vertically as in Figure 2, and were shifted vertically to
4 coincide approximately at 3700 cm^{-1} . Dark-dark control traces show the noise level and the
5 stability of the baseline (lower traces).
6
7
8
9

10
11
12 **Weakly H-bonded O–H stretching region.** The O–H stretching vibrations of weakly
13 hydrogen bonded OH groups of water molecules appear between 3700 and 3500 cm^{-1} . These
14 features are shifted 930 - 960 cm^{-1} to lower frequencies in D_2^{16}O and approximately 10 cm^{-1} to
15 lower frequencies in H_2^{18}O (65, 87, 88, 104). These features have been examined in PSII from
16 *Thermosynechococcus elongatus* (87-89, 96), *Synechocystis* sp. PCC 6803 (65, 84, 85, 90, 92,
17 94), and spinach (91, 93). In our wild-type samples, this region of the S_2 - S_1 spectrum exhibited a
18 weak negative feature at 3663 cm^{-1} , a weak positive feature at 3617 cm^{-1} , and a large negative
19 feature at 3584 cm^{-1} (Figure 5, upper black trace). Corresponding features in the O–D region
20 were observed at $2710(-)\text{ cm}^{-1}$, $2682(+)\text{ cm}^{-1}$, and $2650(-)\text{ cm}^{-1}$ (Figure 6, upper black trace).
21 The D1-V185N mutation diminished the $3663(-)\text{ cm}^{-1}$ feature and replaced the broad positive
22 feature centered at 3617 cm^{-1} with a broad negative feature centered at 3619 cm^{-1} (Figure 5,
23 upper black trace). Corresponding D1-V185N-induced changes were observed in the O–D
24 region of the S_2 - S_1 spectrum (Figure 6, upper red trace): the negative feature at 2710 cm^{-1} was
25 diminished and the broad positive feature centered at 2682 cm^{-1} was replaced with a broad
26 negative feature centered at 2683 cm^{-1} .
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46

47 The weakly hydrogen bonded O–H regions of the S_3 - S_2 , S_0 - S_3 , and S_1 - S_0 spectra in our
48 wild-type samples showed broad negative features centered at approximately 3600 , 3618 , and
49 3611 cm^{-1} , respectively (Figure 5, lower black traces). The corresponding negative features in
50 the O–D regions showed minima at 2634 , 2675 , and 2663 cm^{-1} , respectively (Figure 6, lower
51
52
53
54
55
56
57
58
59
60

1
2
3 black traces). The apparent D1-V185N-induced decrease in the intensity of the 3600 cm^{-1}
4
5 feature in the S_3 - S_2 spectrum (Figure 5, second red trace) may be the result of a shift in the
6
7 baseline because similar mutation-induced decrease was not observed in the O–D region (Figure
8
9
10 6, second red trace). The D1-V185N mutation produced little significant change in the O–H or
11
12 O–D regions of the S_0 - S_3 or S_1 - S_0 spectra (Figures 5 and 6, third and fourth sets of traces).
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

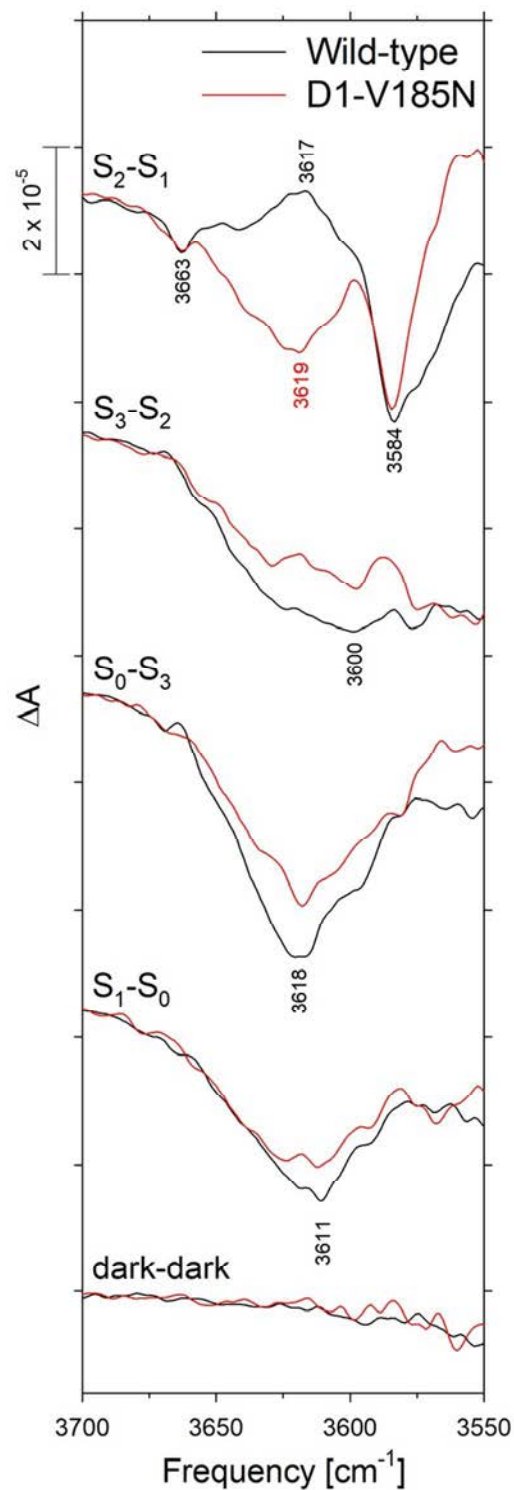


FIGURE 5. FTIR difference spectra of wild-type (black) and D1-V185N (red) PSII core complexes in the weakly hydrogen bonded O–H stretching region in response to four successive

1
2
3 flash illuminations applied at 0°C. The data were collected simultaneously with that shown in
4
5 Figure 2, were multiplied vertically as in Figure 2, and were shifted vertically to coincide
6
7 approximately at 3700 cm⁻¹. Dark-dark control traces show the noise level and the stability of
8
9 the baseline (lower traces).
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

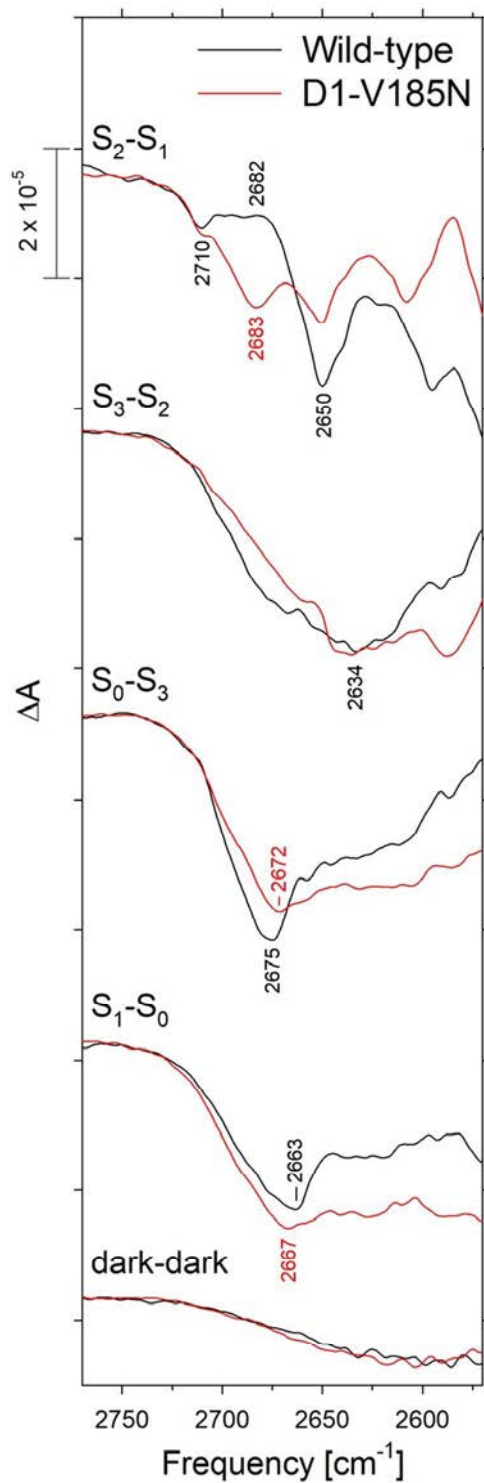


FIGURE 6. FTIR difference spectra of wild-type (black) and D1-V185N (red) PSII core complexes in the O–D stretching region in response to four successive flash illuminations

1
2
3 applied at 0°C. The data represent the averages of 32 wild-type and 61 D1-V185N samples
4 (48,000 and 90,300 scans, respectively) after hydration with D₂¹⁶O and were recorded
5 simultaneously with the D₂¹⁶O data shown in Figures 7 and 8, respectively. The D1-V185N
6 spectra were multiplied vertically by factors of 1.2 as in Figure 8 and were shifted vertically to
7 coincide approximately at 2750 cm⁻¹. Dark-dark control traces show the noise level and the
8 stability of the baseline (lower traces).
9
10
11
12
13
14
15
16
17
18

19 **D-O-D Bending Region.** D-O-D vibrational modes appear near 1210 cm⁻¹ (96, 105) and
20 are inherently weak. However, they can be observed in D₂¹⁶O-D₂¹⁸O double difference spectra
21 (65, 94, 96). The mid-frequency FTIR difference spectra of wild-type PSII hydrated with D₂¹⁶O
22 or D₂¹⁸O are shown in Figure 7. The corresponding spectra of D1-V185N are shown in Figure 8.
23 In Figure 8, the difference spectra of the mutant samples were multiplied vertically by factors of
24 1.2 normalize the spectra to the amplitudes of the negative peak of ferricyanide at 2115 cm⁻¹ and
25 the positive peak of ferrocyanide at 2037 cm⁻¹ in the corresponding wild-type spectra shown in
26 Figure 7. To calculate the D₂¹⁶O-D₂¹⁸O double difference spectra, the difference spectra shown
27 in Figures 7 and 8 were subtracted directly. The resulting S_{n+1}-S_n D₂¹⁶O-D₂¹⁸O double difference
28 spectra of wild-type and D1-V185N in the D-O-D bending [$\delta(\text{DOD})$] region are compared in
29 Figure 9 (black and red traces, respectively). The double difference spectra for the wild-type
30 PSII core complexes resembled those reported previously for *Thermosynechococcus elongatus*
31 (96) and *Synechocystis* sp. PCC 6803 (65, 94). The negative feature at 1263 cm⁻¹ in the S₂-S₁
32 double difference spectrum was reported previously in spectra from *Thermosynechococcus*
33 *elongatus* and was not assigned to a $\delta(\text{DOD})$ mode (96). It has not been reported previously in
34 spectra from *Synechocystis* sp. PCC 6803 (65, 94). It may be relevant that samples exhibiting
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 this feature were equilibrated at a relative humidity of 95% (Ref. (96) and this study), whereas
4
5 samples not exhibiting this feature were equilibrated at a relative humidity of 99% (65, 94).
6

7
8 In the S_2-S_1 $D_2^{16}O$ -*minus*- $D_2^{18}O$ double difference spectrum, the D1-V185N mutation
9
10 shifted the intense 1221(-) and 1210(+) cm^{-1} to 1224(-) and 1212(+) cm^{-1} , respectively, shifted
11
12 the less intense 1188(-) and 1174(+) cm^{-1} features to 1196(-) and 1184(+) cm^{-1} , respectively, and
13
14 eliminated the negative feature at 1263 cm^{-1} (Figure 9, upper pair of traces). In the S_3-S_2 $D_2^{16}O$ -
15
16 $D_2^{18}O$ double difference spectrum, the D1-V185N mutation shifted the weak 1200(+) and 1192(-
17
18) cm^{-1} features to 1194(+) and 1184(-) cm^{-1} , respectively, diminished the 1238(-) cm^{-1} feature and
19
20 shifted it to 1232 cm^{-1} , and eliminated the 1265(-) and 1171(+) cm^{-1} features. There were few
21
22 recognizable features in the S_0-S_3 and S_1-S_0 $D_2^{16}O$ - $D_2^{18}O$ double difference spectra (Figure 9,
23
24 lower two pairs of traces), possibly because of the lower amplitudes of the S_0-S_3 and S_1-S_0
25
26 difference spectra, as was observed in spectra equilibrated with H_2O (Figure 2).
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

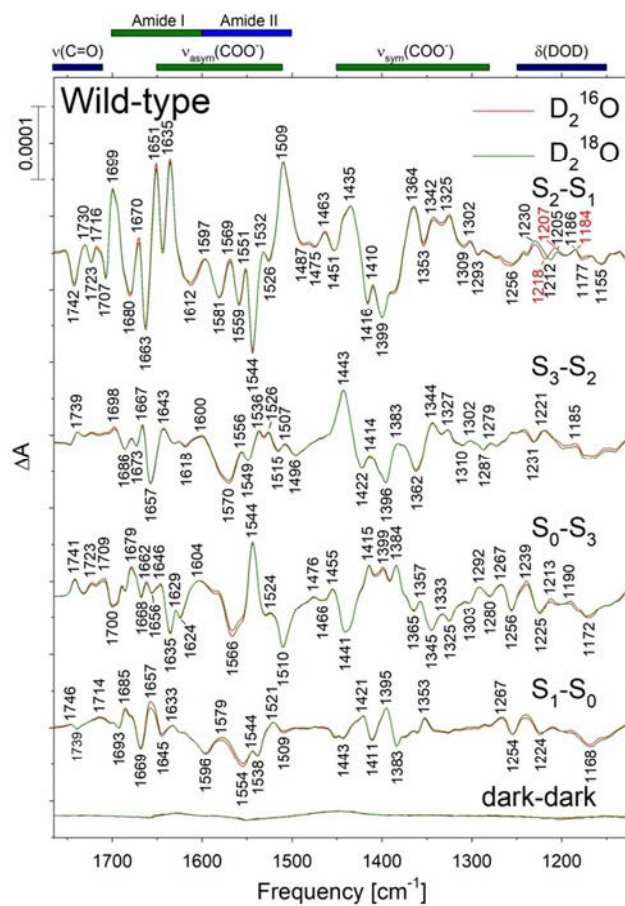


FIGURE 7. Mid-frequency FTIR difference spectra of wild-type PSII core complexes in response to four successive flash illuminations applied at 0°C after hydration with $D_2^{16}O$ (red) or $D_2^{18}O$ (green). The $D_2^{16}O$ and $D_2^{18}O$ data represent the averages of 32 samples (48,000 scans for each trace) and 30 samples (44,400 scans for each trace), respectively. The dark-dark control traces show the noise level and the stability of the baseline (lower traces).

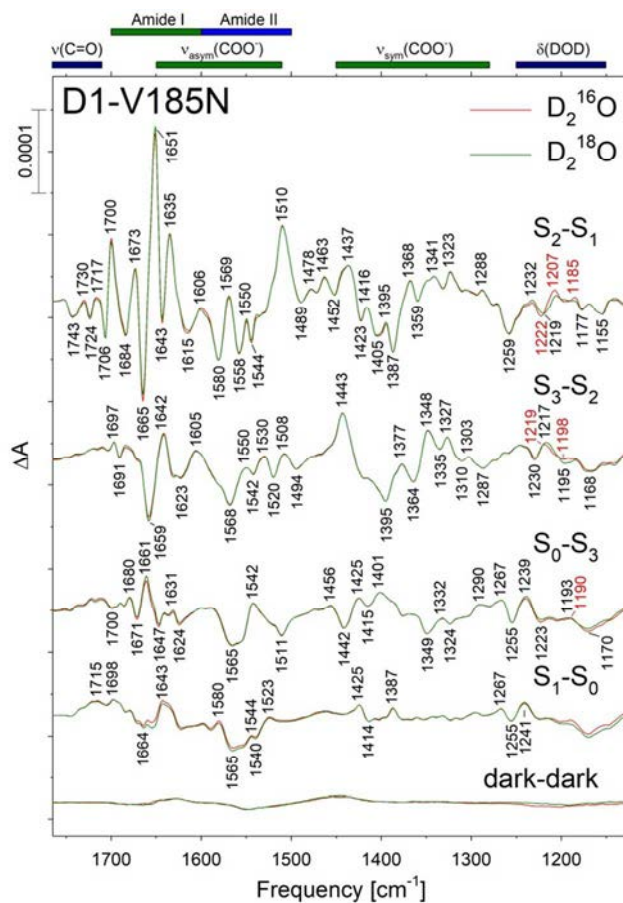


FIGURE 8. Mid-frequency FTIR difference spectra of D1-V185N PSII core complexes in response to four successive flash illuminations applied at 0°C after hydration with D₂¹⁶O (red) or D₂¹⁸O (green). The D₂¹⁶O and D₂¹⁸O data represent the averages of 61 samples (90,300 scans for each trace) and 60 samples (89,900 scans for each trace), respectively. The D1-V185N spectra have been multiplied vertically by factors of 1.2 to approximately normalize the spectra to the extent of flash-induced charge separation in the corresponding wild-type spectra of Figure 7. The dark-dark control traces were also multiplied by factors of 1.2 and show the noise level and the stability of the baseline (lower traces).

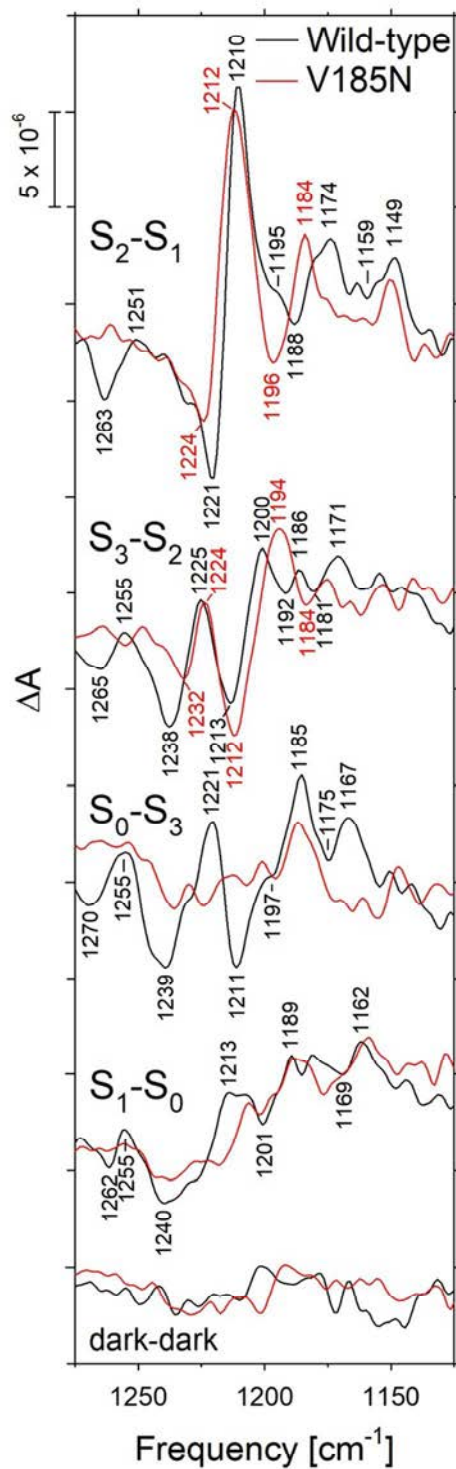


FIGURE 9. The $S_{n+1}-S_n$ $D_2^{16}O-D_2^{18}O$ double difference spectra of wild-type (black) and D1-V185N (red) PSII core complexes in the D-O-D bending region. The data of Figures 7 and 8

1
2
3 were subtracted directly but were offset vertically to maximize overlap. The lower traces show
4
5 the noise levels and were obtained by calculating the $D_2^{16}O$ - $D_2^{18}O$ difference spectra of the dark-
6
7 dark traces shown in Figures 7 and 8.
8
9

11 DISCUSSION

12
13
14 **Carboxylate Residues.** The oxidations of the Mn_4CaO_5 cluster during the S state cycle
15 alter $\nu_{sym}(COO^-)$ and $\nu_{asym}(COO^-)$ modes of carboxylate groups and amide I and amide II modes
16
17 in the polypeptide backbone (101, 102, 106). On the basis of a recent QM/MM study, it was
18
19 concluded that multiple carboxylate ligands of the Mn_4CaO_5 cluster contribute to most of the
20
21 features in the $\nu_{sym}(COO^-)$ region of the S_2 - S_1 spectrum (107). However, the individual mutation
22
23 of four of the Mn_4CaO_5 clusters carboxylate ligands, D1-D170H (108, 109), D1-E189Q (110,
24
25 111), D1-E189R (111), D1-E333Q (92), and D1-D342N (98) produced little or no changes to
26
27 any of the mid-frequency S_{n+1} - S_n FTIR difference spectra (also see ref. (112)). Experimentally,
28
29 mutations constructed at residues involved in the network of hydrogen bonds that extend from
30
31 Y_Z to D1-D61 and the D1-E65/D1-R334/D2-E312 triad and located 5 – 11 Å from the nearest
32
33 Mn ion (e.g., D1-D61A, D1-E65A, D1-N87A, D1-Q165E, D1-N181A, D1-R334A, D2-E312A,
34
35 and D2-K317A), cause numerous changes throughout the mid-frequency FTIR difference
36
37 spectra, including in the $\nu_{sym}(COO^-)$ region of the S_2 - S_1 spectrum (63, 65, 81, 83-85).
38
39
40
41
42
43
44

45 Because D1-V185 is located adjacent to numerous water molecules in the network of
46
47 hydrogen bonds that links Y_Z with D1-D61, the D1-V185N mutation would be expected to
48
49 substantially perturb this network. Indeed, the D1-V185N mutation alters features in the
50
51 $\nu_{sym}(COO^-)$ and $\nu_{asym}(COO^-)$ /amide II regions of the S_2 - S_1 spectrum to a greater extent than any
52
53 mutation yet examined. In the $\nu_{sym}(COO^-)$ region, the substantial decrease of the $1400(-) \text{ cm}^{-1}$
54
55
56
57
58
59
60

1
2
3 feature is also produced by the D1-Q165E and D1-R334Q mutations (83). The substantial
4 decrease of the 1364(+) cm^{-1} feature is also produced by the D1-D61A (65), D1-Q165E (83),
5 D1-R334A (83), and D2-E312A (63) mutations, by the substitution of Sr^{2+} for Ca^{2+} (93, 94, 113-
6 116), and to lesser extents by the D1-N181A (84) and D2-K317A (81) mutations. Finally,
7 alteration of the 1416(-)/1410(+) cm^{-1} feature is also produced by D1-D61A (65), D1-N87A (85),
8 D1-N181A (84), D1-R334A (83), and D2-K317A (81) mutations. In the $\nu_{\text{asym}}(\text{COO}^-)$ /amide II
9 region, the substantial decrease of the 1587(+) cm^{-1} feature is also produced by the D1-D61A
10 (63, 65), D1-N181 (84), D1-Q165E (83), D1-R334A (83), and D2-K317A (81) mutations. This
11 feature corresponds to a $\nu_{\text{asym}}(\text{COO}^-)$ mode because it shifts in globally ^{13}C -labeled samples
12 (101, 102, 106, 117) but not in globally ^{15}N -labeled samples (81, 101, 102, 106, 118). The
13 substantial decrease of the 1544(-) cm^{-1} feature is also produced by the D1-D61A (63, 65), D1-
14 E65A (63), D1-R334A (83), and D2-E312A (63) mutations, and to a lesser extent by the D1-
15 E329Q mutation (63). The elimination of the 1531(+)/1523(-) cm^{-1} feature is also produced by
16 the D1-D61A (63, 65), D1-R334A (83), D2-E312 (63), and D2-K317A (81) mutations and to a
17 lesser extent in some preparations having Sr^{2+} substituted for Ca^{2+} (94, 113, 115). The 1544(-)
18 and 1531(+)/1523(-) cm^{-1} features correspond to amide II modes because they shift in both ^{13}C
19 labeled (101, 102, 106, 117) and ^{15}N labeled (81, 101, 102, 106, 118) samples.
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41

42 These similarities of the D1-V185N-induced changes in the $\text{S}_2\text{-S}_1$ spectrum to those
43 produced by the other mutations imply that the D1-V185N mutation perturbs the same extensive
44 network of hydrogen bonds that is perturbed by the other mutations, as expected from the recent
45 X-ray crystallographic structures (8-10). However, the overall extents of the changes to the
46 $\nu_{\text{sym}}(\text{COO}^-)$ and $\nu_{\text{asym}}(\text{COO}^-)$ /amide II regions are greater than those that are caused by the other
47 mutations, implying that the D1-V185N mutation perturbs the networks of hydrogen bonds
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 located between Y_Z, D1-D61, and the D1-E65/D1-R334/D2-E312 triad to a greater extent than
4
5 the other mutations.
6

7
8 In addition to altering features throughout the $\nu_{\text{sym}}(\text{COO}^-)$ and $\nu_{\text{asym}}(\text{COO}^-)$ /amide II
9
10 regions of the S₂-S₁ spectrum, the D1-V185N mutation also alters features throughout the S₃-S₂
11
12 spectrum. In wild-type PSII, the features in the S₀-S₃ and S₁-S₀ spectra reverse those in the S₂-S₁
13
14 and S₃-S₂ spectra (74-76, 78-80). Because of the alterations produced by the D1-V185N
15
16 mutation to the S₂-S₁ and S₃-S₂ spectra, it is not surprising that the mutation produces substantial
17
18 changes in these regions of the S₀-S₃ and S₁-S₀ spectra. The lower amplitudes of these spectra
19
20 and the relative absence of features in the S₁-S₀ spectrum may reflect the lower efficiency of the
21
22 S state transitions in the D1-V185N mutant (66).
23
24
25

26 **Networks of H-bonds.** The extensive networks of hydrogen bonds in the vicinity of the
27
28 Mn₄CaO₅ cluster have been studied on the basis of changes to the $\nu(\text{C}=\text{O})$ modes of protonated
29
30 carboxylatic acid groups (63, 65, 81-85). In the S₂-S₁ spectrum of wild-type, the 1746(-) cm⁻¹
31
32 feature has been assigned to a carboxylate group whose pK_a value decreases in response to the
33
34 charge that develops on the Mn₄CaO₅ cluster during the S₁ to S₂ transition. This feature is
35
36 diminished or eliminated by the D1-D61A, D1-E65A, D1-E329Q, D1-R334A, and D2-E312A
37
38 mutations, and by overly dehydrating samples (63, 65, 83). That the same $\nu(\text{C}=\text{O})$ mode is
39
40 altered by all of these mutations and treatments implies the existence of an extensive network of
41
42 hydrogen bonds that extends at least 20 Å across the Mn₄CaO₅ cluster (63, 65, 79, 83). In the S₃-
43
44 S₂ spectrum of wild-type, the small 1745(+) cm⁻¹ feature has been assigned to a carboxylate
45
46 group whose pK_a value increases in response to the structural changes that occur during the S₂ to
47
48 S₃ transition. This feature is altered by the D1-D61A mutation and eliminated by the D1-Q165E
49
50 and D1-E329Q mutations. That the same $\nu(\text{C}=\text{O})$ mode is altered by all three mutations implies
51
52
53
54
55
56
57
58
59
60

1
2
3 the existence of an additional network of hydrogen bonds that extends at least 13 Å across the
4 Mn₄CaO₅ cluster (65, 79, 83). Elements of these networks of hydrogen bonds may exist only
5 transiently (58, 119). During the S₃ to S₀ transition, the 1746(-) cm⁻¹ and 1745(+) cm⁻¹ features
6 appear to be reversed, resulting in a positive feature at 1746 cm⁻¹ in the S₀-S₃ spectrum (79, 83).
7
8 The D1-V185N mutation alters the 1746(-) cm⁻¹ feature in the S₂-S₁ spectrum, resulting in a
9 feature with a maximum at 1737 cm⁻¹ and a lower frequency shoulder. The mutation also alters
10 the 1745(+) cm⁻¹ feature in the S₃-S₂ spectrum, diminishing its intensity and shifting it to 1750
11 cm⁻¹. These alterations, in particular the 5-9 cm⁻¹ shifts, have been observed in no other mutant.
12
13 The shifts imply that the D1-V185N mutation alters the environments of the carboxylate groups
14 whose pK_a values change in response to the S₁ to S₂ and S₂ to S₃ transitions.
15
16
17
18
19
20
21
22
23
24
25

26 **Hydrogen-bonded Water Molecules.** The broad positive feature observed between
27 3200 and 2500 cm⁻¹ in the S₂-S₁ spectrum has been studied with QM/MM methods and attributed
28 to the coupled O-H stretching vibrations of strongly hydrogen-bonded water molecules in the
29 network that links D1-D61 with the Ca²⁺ ion and Y_Z (93). In the QM/MM analyses, the broad
30 feature is dominated by the O-H stretching vibrations of W1 and W2 (93). This feature is
31 eliminated by the D1-D61A mutation (65), consistent with the hydrogen bond that exists
32 between this residue and W1 in the recent X-ray crystallographic structural models (8-10). The
33 broad positive features in the other S_{n+1}-S_n spectra have been assigned to the strongly hydrogen
34 bonded O-H stretching vibrations of water molecules in networks of hydrogen bonds that are
35 highly polarizable (88, 95). That the D1-V185N mutation only slightly perturbed these regions
36 of the S_{n+1}-S_n spectra implies that D1-V185 has limited influence on the networks of hydrogen
37 bonds in the vicinity of W1 and W2.
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52

53 The features observed between 3700 and 3500 cm⁻¹ in the S₂-S₁ spectrum also have been
54
55
56
57
58
59
60

1
2
3 studied with QM/MM methods. They have been attributed to the coupled O-H stretching
4
5 vibrations of weakly hydrogen-bonded water molecules in the extended network of hydrogen
6
7 bonds that links D1-D61 with the Ca^{2+} ion and Y_Z (93). The broad negative features observed in
8
9 the other S_{n+1} - S_n spectra in this region are presumed to have the same origin (88, 89).
10
11 Consequently, these features contain contributions from multiple water molecules. The 3663(-)
12
13 cm^{-1} and 3617(+) cm^{-1} features in the S_2 - S_1 spectrum are diminished by the D1-N181A mutation
14
15 and eliminated by the D1-D61A and D1-E333Q mutations (65, 84, 92). The 3617(+) cm^{-1}
16
17 feature is also eliminated by the D1-N87A and D1-N87D mutations (85). These features are also
18
19 diminished sharply by the substitution of Sr^{2+} for Ca^{2+} (94). The broad negative feature centered
20
21 at 3603 cm^{-1} in the S_3 - S_2 spectrum is also altered by the D1-D61A mutation. Consequently, the
22
23 Ca^{2+} ion and the D1-D61, D1-N87, D1-N181, and D1-E333 residues all influence the network of
24
25 hydrogen bonds that links D1-D61 with the Ca^{2+} ion and Y_Z . In the S_2 - S_1 spectrum of D1-
26
27 V185N PSII core complexes, the decrease of the 3663(-) cm^{-1} feature and the replacement of the
28
29 broad positive 3617 cm^{-1} feature with a broad negative feature centered at 3619 cm^{-1} implies that
30
31 D1-V185 participates in the same network of hydrogen bonds. The replacement of the 3617(+)
32
33 cm^{-1} feature with a negative 3619(-) cm^{-1} feature further implies that the perturbation of this
34
35 extended network by the D1-V185N mutation is at least as substantial as the perturbations
36
37 introduced by the D1-D61A mutation.
38
39
40
41
42
43

44 **The D-O-D bending region.** The H-O-H bending mode [$\delta(\text{HOH})$] is sensitive to
45
46 hydrogen bond interactions and disappears when a water molecule deprotonates (96).
47
48

49 Unfortunately, $\delta(\text{HOH})$ modes appear near 1640 cm^{-1} , a congested region of the spectrum. In
50
51 contrast, $\delta(\text{DOD})$ modes appear near 1210 cm^{-1} , a region nearly devoid of overlapping protein
52
53 modes (96, 105). Because these modes are very weak, they are detected in D_2^{16}O - D_2^{18}O double
54
55
56
57
58
59
60

1
2
3 difference spectra (65, 94, 96). In such a double difference spectrum, a shift of a single $\delta(\text{DOD})$
4 mode would show as up to four peaks, two from D_2^{16}O and two from D_2^{18}O . Fewer peaks may
5 be observed because of spectral overlap. In wild-type samples, the number of features in the
6 $\text{D}_2^{16}\text{O}-\text{D}_2^{18}\text{O}$ double difference spectrum of the S_1 to S_2 transition (Figure 9, upper pair of traces)
7 implies that at least two $\delta(\text{DOD})$ modes are shifted during this transition (65, 94, 96). In wild-
8 type samples, the amplitudes of most of the features oscillate during the S state cycle. For
9 example, the $1210(+)$ cm^{-1} feature in the S_2-S_1 double difference spectrum becomes negative (at
10 $1211-1213$ cm^{-1}) in the S_3-S_2 and S_0-S_3 double difference spectra and positive again (at 1213 cm^{-1})
11 in the S_1-S_0 double difference spectrum. Also, the $1221(-)$ cm^{-1} feature in the S_2-S_1 double
12 difference spectrum becomes positive (at 1225 cm^{-1}) in the S_3-S_2 double difference spectrum and
13 positive (at 1221 cm^{-1}) in the S_0-S_3 double difference spectrum. The oscillations imply that the
14 $\delta(\text{DOD})$ modes that are altered during the S_1 to S_2 transition are altered reversibly during the S
15 state cycle (65, 94, 96). One of these DOD molecules must form a hydrogen bond with D1-D61
16 because the D1-D61A mutation eliminates the intense $1221(-)$ cm^{-1} and $1210(+)$ cm^{-1} features
17 from the S_2-S_1 $\text{D}_2^{16}\text{O}-\text{minus}-\text{D}_2^{18}\text{O}$ double difference spectrum (65). Because the D1-V185N
18 mutation shifted the $1221(-)$ and $1210(+)$ cm^{-1} features in this spectrum (and the corresponding
19 $1225(+)$ and $1213(-)$ features in the S_3-S_2 double difference spectrum) by only $2-3$ cm^{-1} [no more
20 than is observed between different wild-type sample preparations (65, 94, 96)], we conclude that
21 the mutation has limited influence on the water molecules that interact with D1-D61. The same
22 conclusion was reached on the basis of the limited perturbations caused by the D1-V185N
23 mutation in the $3200 - 2500$ cm^{-1} region of the S_2-S_1 spectrum (see above).
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50

51 The larger $8-10$ cm^{-1} mutation-induced shifts of the $1188(-)$ and $1174(+)$ cm^{-1} features in
52 the S_2-S_1 double difference spectrum, in addition to other changes observed between 1200 and
53
54
55
56
57
58
59
60

1
2
3 1140 cm^{-1} in the S_2 - S_1 and S_3 - S_2 double difference spectra, shows that the D1-V185N mutation
4
5 perturbs at least one water molecule whose bending mode changes reversibly during the S state
6
7 cycle. This conclusion is consistent with the changes observed between 3700 and 3500 cm^{-1} in
8
9 the S_2 - S_1 spectrum that were attributed to D1-V185N-induced alterations in the extended
10
11 network of hydrogen bonds that links D1-D61 with the Ca^{2+} ion and Y_Z (see above).
12
13

14
15 Another D1-V185N-induced alteration to the S_3 - S_2 double difference spectrum (Figure 9,
16
17 second pair of traces) is the decreased intensity of the 1238(-) cm^{-1} feature and its 6 cm^{-1} shift to
18
19 1232 cm^{-1} . In wild-type PSII, the 1238(-) cm^{-1} feature in the S_3 - S_2 double difference spectra has
20
21 no positive counterpart in any of the other double difference spectra, implying that the bending
22
23 mode of one water molecule is eliminated during the S_2 to S_3 transition (65, 94, 96). Therefore,
24
25 the 1238(-) cm^{-1} feature must correspond to a water molecule that deprotonates during the S_2 to
26
27 S_3 transition (or to a water molecule that physically replaces the water molecule that
28
29 deprotonates, see below). The 1238(-) cm^{-1} feature in the S_3 - S_2 double difference spectrum is
30
31 eliminated or shifted substantially by the substitution of Sr^{2+} for Ca^{2+} (94). Because
32
33 crystallographic (120) and computational (45, 121, 122) studies show that the substitution of Sr^{2+}
34
35 for Ca^{2+} perturbs only W3, W4, and W5, it has been proposed that (1) W3 is the water molecule
36
37 that deprotonates and joins O5 between Mn4 and Mn1 during the S_2 to S_3 transition and (2) W5
38
39 moves to the coordination position on Ca^{2+} vacated by W3 (94). Equating W5 with the 1238(-)
40
41 cm^{-1} feature (94) would be consistent with this feature's alteration by the D1-V185N mutation:
42
43 W5 is within 4.7 Å of D1-V185 in the recent X-ray crystallographic structures of PSII (8-10).
44
45 Consequently, the D1-V185N-induced shift of the 1238(-) cm^{-1} feature in the S_3 - S_2 double
46
47 difference spectrum would be consistent with identifying W3 as the water molecule that
48
49 deprotonates and joins O5 during the S_2 to S_3 transition.
50
51
52
53
54
55
56
57
58
59
60

SUMMARY AND CONCLUSIONS

The D1-V185N mutation perturbs the extensive network of hydrogen bonds that extends from Y_Z to D1-D61 to a greater extent than any other mutation yet examined, yet does not alter water molecules that interact directly with D1-D61. These perturbations are likely the reason that the mutation dramatically slows the structural rearrangement or proton release event that is the rate-limiting step for the S₃ to S₄ transition. The mutation also alters the environments of the carboxylate groups whose pK_a values change in response to the S₁ to S₂ and S₂ to S₃ transitions. Finally, the mutation alters the environment of the water molecule whose bending mode disappears during the S₂ to S₃ transition, consistent with assigning the Ca²⁺-bound W3 as the water molecule that deprotonates and joins O5 during the S₂ to S₃ transition, possibly as the second substrate water molecule for O₂ formation.

ACKNOWLEDGEMENTS

The authors are grateful to Anh P. Nguyen for maintaining the mutant and wild-type cultures of *Synechocystis* sp. PCC 6803 and for purifying the thylakoid membranes that were used for the isolation of PSII core complexes and to V. S. Batista and coworkers for providing the QM/MM-optimized coordinates for the Mn₄CaO₅ cluster and its environments.

AUTHOR INFORMATION

Corresponding Author

1
2
3 *Email: richard.debus@ucr.edu, Tel.: +1 (951) 827-3483. Fax.: +1 (951) 827-4294. Department
4
5 of Biochemistry, University of California, Riverside, CA 92521-0129 USA
6
7

8 **ORCID**

9
10
11 Richard J. Debus: 0000-0003-1321-8730
12
13

14 **Present Addresses**

15
16
17 [†] MSU-DOE Plant Research Laboratory, Michigan State University, East Lansing, MI 48824
18
19

20 **Funding Sources**

21
22
23 Work supported by the Department of Energy, Office of Basic Energy Sciences, Division of
24
25 Chemical Sciences (grant DE-SC0005291 to R. J. D. for FTIR studies) and by the National
26
27 Science Foundation (MCB-1716408 to R. L. B. for mutant construction).
28
29
30

31 **Notes**

32
33
34
35 The authors declare no competing financial interests.
36
37

38 **ACKNOWLEDGMENT**

39
40
41 The authors are grateful to Anh P. Nguyen for maintaining the mutant and wild-type cultures of
42
43 *Synechocystis* sp. PCC 6803 and for purifying the thylakoid membranes that were used for the
44
45 isolation of PSII core complexes and to V. S. Batista and coworkers for providing the QM/MM-
46
47 optimized coordinates for the Mn₄CaO₅ cluster and its environments.
48
49

50 **ABBREVIATIONS**

51
52
53 Chl, chlorophyll; EDTA, ethylenediaminetetraacetic acid; FTIR, Fourier transform infrared;
54
55 MES, 2-(N-morpholino)-ethanesulfonic acid; P₆₈₀, chlorophyll multimer that serves as the light-
56
57
58

1
2
3 induced electron donor in PSII; PSII, photosystem II; Q_A, primary plastoquinone electron
4 acceptor; XFEL, X-ray free electron laser; Y_Z, tyrosine residue that mediates electron transfer
5
6 between the Mn₄O₅Ca cluster and P₆₈₀⁺.
7
8
9

11 12 REFERENCES

- 13
14 (1) Cox, N. and Messinger, J. (2013) Reflections on substrate water and dioxygen
15 formation, *Biochim. Biophys. Acta* 1827, 1020-1030.
16
17 (2) Yano, J. and Yachandra, V. K. (2014) Mn₄Ca Cluster in Photosynthesis: Where and
18 How Water is Oxidized to Dioxygen, *Chem. Rev.* 114, 4175-4250.
19
20 (3) Shen, J.-R. (2015) The Structure of Photosystem II and the Mechanism of Water
21 Oxidation in Photosynthesis, *Annu. Rev. Plant Biol.* 66, 23-48.
22
23 (4) Krewald, V., Retegan, M., and Pantazis, D. A. (2016) Principles of Natural
24 Photosynthesis, *Top. Curr. Chem.* 371, 23-48.
25
26 (5) Pérez-Navarro, M., Neese, F., Lubitz, W., Pantazis, D. A., and Cox, N. (2016)
27 Recent developments in biological water oxidation, *Curr. Opin. Chem. Biol.* 31, 113-119.
28
29 (6) Askerka, M., Brudvig, G. W., and Batista, V. S. (2017) The O₂-Evolving Complex of
30 Photosystem II: Recent Insights from Quantum Mechanics/Molecular Mechanics (QM/MM),
31 Extended X-ray Absorption Fine Structure (EXAFS), and Femtosecond X-ray Crystallography
32 Data, *Acc. Chem. Res.* 50, 41-48.
33
34 (7) Vinyard, D. J. and Brudvig, G. W. (2017) Progress Toward a Molecular Mechanism
35 of Water Oxidation in Photosystem II, *Annu. Rev. Phys. Chem.* 68, 101-116.
36
37 (8) Umena, Y., Kawakami, K., Shen, J.-R., and Kamiya, N. (2011) Crystal Structure of
38 Oxygen-Evolving Photosystem II at a Resolution of 1.9 Å, *Nature* 473, 55-60.
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 (9) Suga, M., Akita, F., Hirata, K., Ueno, G., Murakami, H., Nakajima, Y., Shimizu, T.,
4 Yamashita, K., Yamamoto, M., Ago, H., and Shen, J.-R. (2015) Native structure of photosystem
5 II at 1.95 Å resolution viewed by femtosecond X-ray pulses, *Nature* 517, 99-103.
6
7

8
9
10 (10) Tanaka, A., Fukushi, Y., and Kamiya, N. (2017) Two Different Structures of the
11 Oxygen-Evolving Complex in the Same Polypeptide Frameworks of Photosystem II, *J. Am.*
12 *Chem. Soc.* 139, 1718-1721.
13
14

15
16
17 (11) Ago, H., Adachi, H., Umena, Y., Tashiro, T., Kawakami, K., Kamiya, N., Tian, L.,
18 Han, G., Kuang, T., Liu, Z., Wang, F., Zou, H., Enami, I., Miyano, M., and Shen, J.-R. (2016)
19 Novel Features of Eukaryotic Photosystem II Revealed by Its Crystal Structure Analysis from a
20 Red Alga, *J. Biol. Chem.* 291, 5676-5687.
21
22

23
24
25 (12) Su, X., Ma, J., Wei, X., Cao, P., Zhu, D., Chang, W., Liu, Z., and Li, M. (2017)
26 Structure and assembly mechanism of plant C₂S₃M₂-type PSII-LHCII supercomplex, *Science*
27 357, 815-820.
28
29

30
31
32 (13) Cao, P., Su, X., Pan, X., Liu, Z., Chang, W., and Li, M. (2018) Structure, assembly
33 and energy transfer of plant photosystem II supercomplex, *Biochim. Biophys. Acta* -(in press).
34
35

36
37 (14) Wei, X., Su, X., Cao, P., Liu, X., Chang, W., Li, M., Zhang, X., and Liu, Z. (2016)
38 Structure of spinach photosystem II-LHII supercomplex at 3.2 Å resolution, *Nature* 534, 69-74.
39
40

41
42 (15) van Bezouwen, L. S., Caffarri, S., Kale, R. S., Kouril, R., Thunnissen, A.-M. W. H.,
43 Oostergetal, G. T., and Boekem, E. J. (2017) Subunit and chlorophyll organization of the plant
44 photosystem II supercomplex, *Nature Plants* 3, 17080.
45
46

47
48 (16) Rapatskiy, L., Cox, N., Savitsky, A., Ames, W. M., Sander, J., Nowaczyk, M. M.,
49 Rögner, M., Boussac, A., Neese, F., Messinger, J., and Lubitz, W. (2012) Detection of Water-
50
51

1
2
3 Binding Sites of the Oxygen-Evolving Complex of Photosystem II Using W-band ^{17}O Electron-
4 Electron Double Resonance-Detected NMR Spectroscopy, *J. Am. Chem. Soc.* *134*, 16619-16634.

5
6
7 (17) Pérez Navarro, M., Ames, W. M., Nilsson, H., Lohmiller, T., Pantazis, D. A.,
8 Rapatskiy, L., Nowaczyk, M. M., Neese, F., Boussac, A., Messinger, J., Lubitz, W., and Cox, N.
9
10 (2013) Ammonia binding to the oxygen-evolving complex of photosystem II identified the
11 solvent-exchangeable oxygen bridge (μ -oxo) of the manganese tetramer, *Proc. Natl. Acad. Sci.*
12 *USA* *110*, 15561-15566.
13
14

15
16
17 (18) Lohmiller, T., Krewald, V., Pérez Navarro, M., Retegan, M., Rapatskiy, L.,
18 Nowaczyk, M. M., Boussac, A., Neese, F., Lubitz, W., Pantazis, D. A., and Cox, N. (2014)
19 Structure, ligands and substrate coordination of the oxygen-evolving complex of photosystem II
20 in the S_2 state: a combined EPR and DFT study, *Phys. Chem. Chem. Phys.* *16*, 11877-11892.
21
22

23
24 (19) Lohmiller, T., Krewald, V., Sedoud, A., Rutherford, A. W., Neese, F., Lubitz, W.,
25 Pantazis, D. A., and Cox, N. (2017) The First State in the Catalytic Cycle of the Water-Oxidizing
26 Enzyme: Identification of a Water-Derived μ -Hydroxo Bridge, *J. Am. Chem. Soc.* *139*, 14412-
27 14424.
28
29

30
31 (20) Pantazis, D. A., Ames, W., Cox, N., Lubitz, W., and Neese, F. (2012) Two
32 Interconvertible Structures that Explain the Spectroscopic Properties of the Oxygen-Evolving
33 Complex of Photosystem II in the S_2 State, *Angew. Chem. Int. Ed.* *51*, 9935-9940.
34
35

36
37 (21) Siegbahn, P. E. M. (2008) A Structure-Consistent Mechanism for Dioxygen
38 Formation in Photosystem II, *Chem. Eur. J.* *14*, 8290-8302.
39
40

41
42 (22) Siegbahn, P. E. M. (2013) Water oxidation mechanism in photosystem II, including
43 oxidations, proton release pathways, O-O bond formation and O_2 release, *Biochim. Biophys. Acta*
44 *1827*, 1003-1019.
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 (23) Cox, N., Retegan, M., Neese, F., Pantazis, D. A., Boussac, A., and Lubitz, W. (2014)
4
5 Electronic structure of the oxygen-evolving complex in photosystem II prior to O-O bond
6
7 formation, *Science* 345, 804-808.
8

9
10 (24) Zaharieva, I., Chernev, P., Berggren, G., Anderlund, M., Styring, S., Dau, H., and
11
12 Haumann, M. (2016) Room-Temperature Energy-Sampling K β X-ray Emission Spectroscopy of
13
14 the Mn₄Ca Complex of Photosystem II Reveals Three Manganese-Centered Oxidation Steps and
15
16 Suggests Coordination Change Prior to O₂ Formation, *Biochemistry* 55, 4197-4211.
17
18

19 (25) Suga, M., Akita, F., Sugahara, M., Kubo, M., Nakajima, Y., Nakane, T., Yamashita,
20
21 K., Umena, Y., Nakabayashi, M., Yamane, T., Nakano, T., Suzuki, M., Masuda, T., Inoue, S.,
22
23 Kimura, T., Nomura, T., Yonekura, S., Yu, L.-J., Sakamoto, T., Motomura, T., Chen, J.-H.,
24
25 Kato, Y., Noguchi, T., Tono, K., Joti, Y., Kameshima, T., Hatsui, T., Nango, E., Tanaka, R.,
26
27 Naitow, H., Matsuura, Y., Yamashita, A., Yamamoto, M., Nureki, O., Yabashi, M., Ishikawa, T.,
28
29 Iwata, S., and Shen, J.-R. (2017) LIght-induced structural changes and the site of O=O bond
30
31 formation in PSII caught by XFEL, *Nature* 543, 131-135.
32
33

34
35 (26) Askerka, M., Vinyard, D. J., Brudvig, G. W., and Batista, V. S. (2015) NH₃ Binding
36
37 to the S₂ State of the O₂-Evolving Complex of Photosystem II: Analogue to H₂O Binding during
38
39 the S₂ to S₃ Transition, *Biochemistry* 54, 5783-5786.
40
41

42 (27) Capone, M., Bovi, D., Narzi, D., and Guidoni, L. (2016) Reorganization of Substrate
43
44 Waters between the Closed and Open Cubane Conformers during the S₂ to S₃ Transition in the
45
46 Oxygen Evolving Complex, *Biochemistry* 54, 6439-6442.
47
48

49 (28) Askerka, M., Wang, J., Vinyard, D. J., Brudvig, G. W., and Batista, V. S. (2016) S₃
50
51 State of the O₂-Evolving Complex of Photosystem II: Insights from QM/MM, EXAFS, and
52
53 Femtosecond X-ray Diffraction, *Biochemistry* 55, 984.
54
55
56
57
58
59
60

1
2
3 (29) Retegan, M., Krewald, V., Mamedov, F., Neese, F., Lubitz, W., Cox, N., and
4
5 Pantazis, D. A. (2016) A five-coordinate Mn(IV) intermediate in biological water oxidation:
6
7 spectroscopic signature and a pivot mechanism for water binding, *Chem. Sci.* 7, 72-84.

8
9
10 (30) Capone, M., Narzi, D., Bovi, D., and Guidoni, L. (2016) Mechanism of Water
11
12 Delivery to the Active Site of Photosystem II along the S₂ to S₃ Transition, *J. Phys. Chem. Lett.*
13
14 7, 592-596.

15
16
17 (31) Retegan, M. and Pantazis, D. A. (2016) Interaction of methanol with the oxygen-
18
19 evolving complex: atomistic models, channel identification, species dependence, and
20
21 mechanistic implications, *Chem. Sci.* 7, 6463-6476.

22
23
24 (32) Wang, J., Askerka, M., Brudvig, G. W., and Batista, V. S. (2017) Crystallographic
25
26 Data Support the Carousel Mechanism of Water Supply to the Oxygen-Evolving Complex of
27
28 Photosystem II, *ACS Energy Lett.* 2, 2299-2306.

29
30
31 (33) Ames, W., Pantazis, D. A., Krewald, V., Cox, N., Messinger, J., Lubitz, W., and
32
33 Neese, F. (2011) Theoretical Evaluation of Structural Models of the S₂ State in the Oxygen
34
35 Evolving Complex of Photosystem II: Protonation States and Magnetic Interactions, *J. Am.*
36
37 *Chem. Soc.* 133, 19743-19757.

38
39
40 (34) Bovi, D., Narzi, D., and Guidoni, L. (2013) The S₂ State of the Oxygen-Evolving
41
42 Complex of Photosystem II Explored by QM/MM Dynamics: Spin Surfaces and Metastable
43
44 States Suggest a Reaction Path Towards the S₃ State, *Angew. Chem. Int. Ed.* 52, 11744-11749.

45
46
47 (35) Shoji, M., Isobe, H., and Yamaguchi, K. (2015) QM/MM study of the S₂ to S₃
48
49 transition reaction in the oxygen-evolving complex of photosystem II, *Chem. Phys. Lett.* 636,
50
51 172-179.

1
2
3 (36) Isobe, H., Shoji, M., Shen, J.-R., and Yamaguchi, K. (2015) Strong Coupling
4 between the Hydrogen Bonding Environment and Redox Chemistry during the S₂ to S₃
5 Transition in the Oxygen-Evolving Complex of Photosystem II, *J. Phys. Chem. B* 119, 13922-
6 13933.
7

8
9
10
11
12 (37) Ugur, I., Rutherford, A. W., and Kaila, V. R. I. (2016) Redox-coupled substrate
13 water reorganization in the active site of Photosystem II - The role of calcium in substrate water
14 delivery, *Biochim. Biophys. Acta* 1857, 740-748.
15
16

17
18
19 (38) Beal, N. J., Corry, T. A., and O'Malley, P. J. (2018) A Comparison of Experimental
20 and Broken Symmetry Density Functional (BS-DFT) Calculated Electron Paramagnetic
21 Resonance (EPR) Parameters for Intermediates Involved in the S₂ to S₃ State Transition of
22 Nature's Oxygen Evolving Complex, *J. Phys. Chem. B* 122, 1394-1407.
23
24
25

26
27
28 (39) Siegbahn, P. E. M. (2009) Structures and Energetics for O₂ Formation in
29 Photosystem II, *Acc. Chem. Res.* 42, 1871-1880.
30
31

32
33 (40) Li, X. and Siegbahn, P. E. M. (2015) Alternative mechanisms for O₂ release and O-O
34 bond formation in the oxygen evolving complex of photosystem II, *Phys. Chem. Chem. Phys.* 17,
35 12168-12174.
36
37

38
39
40 (41) Krewald, V., Retegan, M., Neese, F., Lubitz, W., Pantazis, D. A., and Cox, N. (2016)
41 Spin State as a Marker for the Structural Evolution of Nature's Water-Splitting Catalyst, *Inorg.*
42 *Chem.* 55, 488-501.
43
44

45
46
47 (42) Siegbahn, P. E. M. (2018) Water Oxidation by PSII: A Quantum Chemical
48 Approach, in *Mechanisms of Primary Energy Transduction in Biology* (Wikström, M., Ed.) pp
49 273-295, Royal Society of Chemistry, London.
50
51
52
53
54
55
56
57
58
59
60

- 1
2
3 (43) Siegbahn, P. E. M. (2006) O-O Bond Formation in the S₄ State of the Oxygen-
4 Evolving Complex in Photosystem II, *Chem. Eur. J.* 12, 9217-9227.
5
6
7 (44) Nilsson, H., Rappaport, F., Boussac, A., and Messinger, J. (2014) Substrate-water
8 exchange in photosystem II is arrested before dioxygen formation, *Nature Commun.* 5, 4305-
9 doi:10.1038/ncomms5305.
10
11
12 (45) Vogt, L., Ertem, M. Z., Pal, R., Brudvig, G. W., and Batista, V. S. (2015)
13 Computational Insights on Crystal Structures of the Oxygen-Evolving Complex of Photosystem
14 II with Either Ca²⁺ or Ca²⁺ Substituted by Sr²⁺, *Biochemistry* 54, 820-825.
15
16
17 (46) Klauss, A., Haumann, M., and Dau, H. (2012) Alternating electron and proton
18 transfer steps in photosynthetic water oxidation, *Proc. Natl. Acad. Sci. USA* 109, 16035-16040.
19
20
21 (47) Klauss, A., Haumann, M., and Dau, H. (2015) Seven Steps of Alternating Electron
22 and Proton Transfer in Photosystem II Water Oxidation Traced by Time-Resolved Photothermal
23 Beam Deflection at Improved Sensitivity, *J. Phys. Chem. B* 119, 2677-2689.
24
25
26 (48) Zaharieva, I., Dau, H., and Haumann, M. (2017) Sequential and Coupled Proton and
27 Electron Transfer Events in the S₂ to S₃ Transition of Photosynthetic Water Oxidation Revealed
28 by Time-Resolved X-ray Absorption Spectroscopy, *Biochemistry* 55, 6996-7004.
29
30
31 (49) Murray, J. W. and Barber, J. (2007) Structural Characteristics of Channels and
32 Pathways in Photosystem II Including the Identification of an Oxygen Channel, *J. Struct. Biol.*
33 159, 228-237.
34
35
36 (50) Gabdulkhakov, A., Guskov, A., Broser, M., Kern, J., Müh, F., Saenger, W., and
37 Zouni, A. (2009) Probing the Accessibility of the Mn₄Ca Cluster in Photosystem II: Channels
38 Calculation, Noble Gas Derivatization, and Cocrystallization with DMSO, *Structure* 17, 1223-
39 1234.
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 (51) Bondar, A.-N. and Dau, H. (2012) Extended protein/water H-bond networks in
4 photosynthetic water oxidation, *Biochim. Biophys. Acta 1817*, 1177-1190.

5
6
7 (52) Vassiliev, S., Zaraiskaya, T., and Bruce, D. (2012) Exploring the energetics of water
8 permeation in photosystem II by multiple steered molecular dynamics simulations, *Biochim.*
9
10
11
12 *Biophys. Acta 1817*, 1671-1678.

13
14 (53) Ogata, K., Yuki, T., Hatakeyama, M., Uchida, W., and Nakamura, S. (2013) All-
15 Atom Molecular Dynamics Simulation of Photosystem II Embedded in Thylakoid Membranes,
16
17
18
19 *J. Am. Chem. Soc. 135*, 15670-15673.

20
21 (54) Bao, H., Dilbeck, D. L., and Burnap, R. L. (2013) Proton transport facilitating water-
22 oxidation: the role of second sphere ligands surrounding the catalytic metal cluster, *Photosynth.*
23
24
25
26 *Res. 116*, 215-219.

27
28 (55) Linke, K. and Ho, F. M. (2014) Water in Photosystem II: Structural, functional, and
29 mechanistic considerations, *Biochim. Biophys. Acta 1837*, 14-32.

30
31 (56) Vogt, L., Vinyard, D. J., Khan, S., and Brudvig, G. W. (2015) Oxygen-evolving
32 complex of Photosystem II: an analysis of second-shell residues and hydrogen-bonding
33
34
35
36
37
38 networks, *Curr. Opin. Chem. Biol. 25*, 152-158.

39
40 (57) Takoaka, T., Sakashita, N., Saito, K., and Ishikita, H. (2016) pK_a of a Proton-
41 Conducting Water Chain in Photosystem II, *J. Phys. Chem. Lett. 7*, 1925-1932.

42
43 (58) Guerra, F., Siemers, M., Mielack, C., and Bondar, A.-N. (2018) Dynamics of Long-
44 Distance Hydrogen-Bond Networks in Photosystem II, *J. Phys. Chem. B* 4625-4641.

45
46 (59) Hundelt, M., Hays, A.-M. A., Debus, R. J., and Junge, W. (1998) Oxygenic
47 Photosystem II: The Mutation D1-D61N in *Synechocystis* sp. PCC 6803 Retards S-State
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 Transitions without Affecting Electron Transfer from Y_Z to P_{680}^+ , *Biochemistry* 37, 14450-
4
5 14456.

6
7 (60) Clausen, J., Debus, R. J., and Junge, W. (2004) Time-Resolved Oxygen Production
8
9 by PSII: Chasing Chemical Intermediates, *Biochim. Biophys. Acta* 1655, 184-194.

10
11 (61) Ferreira, K. N., Iverson, T. M., Maghlaoui, K., Barber, J., and Iwata, S. (2004)
12
13 Architecture of the Photosynthetic Oxygen-Evolving Center, *Science* 303, 1831-1838.

14
15 (62) Ishikita, H., Saenger, W., Loll, B., Biesiadka, J., and Knapp, E.-W. (2006) Energetics
16
17 of a Possible Proton Exit Pathway for Water Oxidation in Photosystem II, *Biochemistry* 45,
18
19 2063-2071.

20
21 (63) Service, R. J., Hillier, W., and Debus, R. J. (2010) Evidence from FTIR Difference
22
23 Spectroscopy of an Extensive Network of Hydrogen Bonds near the Oxygen-Evolving Mn_4Ca
24
25 Cluster of Photosystem II Involving D1-Glu65, D2-Glu312, and D1-Glu329, *Biochemistry* 49,
26
27 6655-6669.

28
29 (64) Dilbeck, D. L., Hwang, H. J., Zaharieva, I., Gerencser, L., Dau, H., and Burnap, R.
30
31 L. (2012) The D1-D61N Mutation in *Synechocystis* sp. PCC 6803 Allows the Observation of pH-
32
33 Sensitive Intermediates in the Formation and Release of O_2 from Photosystem II, *Biochemistry*
34
35 51, 1079-1091.

36
37 (65) Debus, R. J. (2014) Evidence from FTIR Difference Spectroscopy That D1-Asp61
38
39 Influences the Water Reactions of the Oxygen-Evolving Mn_4CaO_5 Cluster of Photosystem II,
40
41 *Biochemistry* 53, 2941-2955.

42
43 (66) Dilbeck, D. L., Bao, H., Neveu, C. L., and Burnap, R. L. (2013) Perturbing the Water
44
45 Cavity Surrounding the Manganese Cluster by Mutating the Residue D1-Valine Has a Strong
46
47 Effect on the Water Oxidation Mechanism of Photosystem II, *Biochemistry* 52, 6824-6833.
48
49
50
51
52
53
54
55
56
57
58
59
60

- 1
2
3 (67) Haumann, M., Liebisch, P., Müller, C., Barra, M., Grabolle, M., and Dau, H. (2005)
4 Photosynthetic O₂ Formation Tracked by Time-Resolved X-ray Experiments, *Science* 310, 1019-
5 1021.
6
7
8
9
10 (68) Gerencsér, L. and Dau, H. (2010) Water Oxidation by Photosystem II: H₂O-D₂O
11 Exchange and the Influence of pH Support Formation of an Intermediate by Removal of a Proton
12 before Dioxygen Creation, *Biochemistry* 49, 10098-10106.
13
14
15
16
17 (69) Bao, H. and Burnap, R. L. (2015) Structural rearrangements preceding dioxygen
18 formation by the water oxidation complex of photosystem II, *Proc. Natl. Acad. Sci. USA* 112,
19 E6139-E6147.
20
21
22
23
24 (70) Zscherp, C. and Barth, A. (2001) Reaction-Induced Infrared Difference Spectroscopy
25 for the Study of Protein Reaction Mechanisms, *Biochemistry* 40, 1875-1883.
26
27
28
29 (71) Barth, A. and Zscherp, C. (2002) What Vibrations Tell Us About Proteins, *Q. Rev.*
30 *Biophys.* 35, 369-430.
31
32
33 (72) Rich, P. R. and Iwaki, M. (2005) Infrared Protein Spectroscopy as a Tool to Study
34 Protonation Reactions Within Proteins, in *Biophysical and Structural Aspects of Bioenergetics*
35 (Wikström, M., Ed.) pp 314-333, Royal Society of Chemistry, Cambridge, U.K.
36
37
38
39 (73) Barth, A. (2007) Infrared Spectroscopy of Proteins, *Biochim. Biophys. Acta* 1767,
40 1073-1101.
41
42
43
44 (74) Noguchi, T. (2007) Light-Induced FTIR Difference Spectroscopy as a Powerful Tool
45 Toward Understanding the Molecular Mechanism of Photosynthetic Oxygen Evolution,
46 *Photosyn. Res.* 91, 59-69.
47
48
49
50 (75) Noguchi, T. (2008) Fourier Transform Infrared Analysis of the Photosynthetic
51 Oxygen-Evolving Center, *Coord. Chem. Rev.* 251, 336-346.
52
53
54
55
56
57
58
59
60

1
2
3 (76) Debus, R. J. (2008) Protein Ligation of the Photosynthetic Oxygen-Evolving Center,
4
5 *Coord. Chem. Rev.* 252, 244-258.

6
7 (77) Chu, H.-A. (2013) Fourier transform infrared difference spectroscopy for studying
8
9 the molecular mechanism of photosynthetic water oxidation, *Frontiers Plant Sci.* 4, 146.

10
11 (78) Noguchi, T. (2013) Monitoring the reactions of photosynthetic water oxidation using
12
13 infrared spectroscopy, *Biomedical Spectroscopy and Imaging* 2, 115-128.

14
15 (79) Debus, R. J. (2015) FTIR Studies of Metal Ligands, Networks of Hydrogen Bonds,
16
17 and Water Molecules near the Active Site Mn_4CaO_5 Cluster in Photosystem II, *Biochim.*
18
19 *Biophys. Acta* 1847, 19-34.

20
21 (80) Noguchi, T. (2015) Fourier transform infrared difference and time-resolved infrared
22
23 detection of the electron and proton transfer dynamics in photosynthetic water oxidation,
24
25 *Biochim. Biophys. Acta* 1847, 35-45.

26
27 (81) Pokhrel, R., Service, R. J., Debus, R. J., and Brudvig, G. W. (2013) Mutation of
28
29 Lysine 317 in the D2 Subunit of Photosystem II Alters Chloride Binding and Proton Transport,
30
31 *Biochemistry* 52, 4758-4773.

32
33 (82) Suzuki, H., Yu, J., Kobayashi, T., Nakanishi, H., Nixon, P. J., and Noguchi, T.
34
35 (2013) Functional Roles of D2-Lys317 and the Interacting Chloride Ion in the Water Oxidation
36
37 Reaction of Photosystem II As Revealed by Fourier Transform Infrared Analysis, *Biochemistry*
38
39 52, 4748-4757.

40
41 (83) Service, R. J., Hillier, W., and Debus, R. J. (2014) A Network of Hydrogen Bonds
42
43 near the Oxygen-Evolving Mn_4CaO_5 Cluster of Photosystem II Probed with FTIR Difference
44
45 Spectroscopy, *Biochemistry* 53, 1001-1017.

1
2
3 (84) Pokhrel, R., Debus, R. J., and Brudvig, G. W. (2015) Probing the Effect of Mutations
4 of Asparagine 181 in the D1 Subunit of Photosystem II, *Biochemistry* 54, 1663-1672.
5

6
7 (85) Banerjee, G., Ghosh, I., Kim, C. J., Debus, R. J., and Brudvig, G. W. (2017)
8 Substitution of the D1-N87 site in photosystem II of cyanobacteria mimics the chloride-binding
9 characteristics of spinach photosystem II, *J. Biol. Chem.* 293, 2487-2497.
10
11

12 (86) Nakamura, S. and Noguchi, T. (2017) Infrared Determination of the Protonation
13 State of a Key Histidine Residue in the Photosynthetic Water Oxidizing Center, *J. Am. Chem.*
14 *Soc.* 139, 9364-9375.
15
16

17 (87) Noguchi, T. and Sugiura, M. (2000) Structure of an Active Water Molecule in the
18 Water-Oxidizing Complex of Photosystem II as Studied by FTIR Spectroscopy, *Biochemistry*
19 39, 10943-10949.
20
21

22 (88) Noguchi, T. and Sugiura, M. (2002) FTIR Detection of Water Reactions During the
23 Flash-Induced S-State Cycle of the Photosynthetic Water-Oxidizing Complex, *Biochemistry* 41,
24 15706-15712.
25
26

27 (89) Noguchi, T. (2007) FTIR Detection of Water Reactions in the Oxygen-Evolving
28 Center of Photosystem II, *Phil. Trans. R. Soc. Lond. B* 363, 1189-1195.
29
30

31 (90) Shimada, Y., Suzuki, H., Tsuchiya, T., Tomo, T., Noguchi, T., and Mimuro, M.
32 (2009) Effect of a Single-Amino Acid Substitution of the 43 kDa Chlorophyll Protein on the
33 Oxygen-Evolving Reaction of the Cyanobacterium *Synechocystis* sp. PCC 6803: Analysis of the
34 Glu354Gln Mutation., *Biochemistry* 48, 6095-6103.
35
36

37 (91) Hou, L.-H., Wu, C.-M., Huang, H.-H., and Chu, H.-A. (2011) Effects of Ammonia
38 on the Structure of the Oxygen-Evolving Complex in Photosystem II As Revealed by Light-
39 Induced FTIR Difference Spectroscopy, *Biochemistry* 50, 9248-9254.
40
41
42
43
44
45
46
47
48

1
2
3 (92) Service, R. J., Yano, J., Dilbeck, D. L., Burnap, R. L., Hillier, W., and Debus, R. J.
4
5 (2013) Participation of Glutamate-333 of the D1 Polypeptide in the Ligation of the Mn₄CaO₅
6
7 Cluster in Photosystem II, *Biochemistry* 52, 8452-8464.
8

9
10 (93) Nakamura, S., Ota, K., Shibuya, Y., and Noguchi, T. (2016) Role of a Water
11
12 Network around the Mn₄CaO₅ Cluster in Photosynthetic Water Oxidation: A Fourier Transform
13
14 Infrared Spectroscopy and Quantum Mechanics/Molecular Mechanics Calculation Study,
15
16 *Biochemistry* 55, 597-607.
17

18
19 (94) Kim, C. J. and Debus, R. J. (2017) Evidence from FTIR Difference Spectroscopy
20
21 That a Substrate H₂O Molecule for O₂ Formation in Photosystem II is Provided by the Ca Ion of
22
23 the Catalytic Mn₄CaO₅ Cluster, *Biochemistry* 56, 2558-2570.
24

25
26 (95) Noguchi, T., Suzuki, H., Tsuno, M., Sugiura, M., and Kato, C. (2012) Time-
27
28 Resolved Infrared Detection of the Proton and Protein Dynamics during Photosynthetic Oxygen
29
30 Evolution, *Biochemistry* 51, 3205-3214.
31

32
33 (96) Suzuki, H., Sugiura, M., and Noguchi, T. (2008) Monitoring Water Reactions during
34
35 the S-State Cycle of the Photosynthetic Water-Oxidizing Center: Detection of the DOD Bending
36
37 Vibrations by Means of Fourier Transform Infrared Spectroscopy, *Biochemistry* 47, 11024-
38
39 11030.
40

41
42 (97) Debus, R. J., Campbell, K. A., Gregor, W., Li, Z.-L., Burnap, R. L., and Britt, R. D.
43
44 (2001) Does Histidine 332 of the D1 Polypeptide Ligate the Manganese Cluster in Photosystem
45
46 II? An Electron Spin Echo Envelope Modulation Study, *Biochemistry* 40, 3690-3699.
47

48
49 (98) Strickler, M. A., Walker, L. M., Hillier, W., Britt, R. D., and Debus, R. J. (2007) No
50
51 Evidence from FTIR Difference Spectroscopy That Aspartate-342 of the D1 Polypeptide Ligates
52
53

1
2
3 a Mn Ion That Undergoes Oxidation during the S₀ to S₁, S₁ to S₂, or S₂ to S₃ Transitions in
4
5 Photosystem II, *Biochemistry* 46, 3151-3160.
6

7 (99) Hays, A.-M. A., Vassiliev, I. R., Golbeck, J. H., and Debus, R. J. (1998) Role of D1-
8 His190 in Proton-Coupled Electron Transfer Reactions in Photosystem II: A Chemical
9
10 Complementation Study, *Biochemistry* 37, 11352-11365.
11
12

13 (100) Noguchi, T. and Sugiura, M. (2002) Flash-Induced FTIR Difference Spectra of the
14
15 Water Oxidizing Complex in Moderately Hydrated Photosystem II Core Films: Effect of
16
17 Hydration Extent on S-State Transitions, *Biochemistry* 41, 2322-2330.
18
19

20 (101) Noguchi, T. and Sugiura, M. (2003) Analysis of Flash-Induced FTIR Difference
21
22 Spectra of the S-State Cycle in the Photosynthetic Water-Oxidizing Complex by Uniform ¹⁵N
23
24 and ¹³C Isotope Labeling, *Biochemistry* 42, 6035-6042.
25
26

27 (102) Yamanari, T., Kimura, Y., Mizusawa, N., Ishii, A., and Ono, T.-A. (2004) Mid- to
28
29 Low-Frequency Fourier Transform Infrared Spectra of S-State Cycle for Photosynthetic Water
30
31 Oxidation in *Synechocystis* sp. PCC 6803, *Biochemistry* 43, 7479-7490.
32
33

34 (103) Noguchi, T., Inoue, Y., and Tang, X.-S. (1999) Structure of a Histidine Ligand in the
35
36 Photosynthetic Oxygen-Evolving Complex as Studied by Light-Induced Fourier Transform
37
38 Infrared Spectroscopy, *Biochemistry* 38, 10187-10195.
39
40

41 (104) Kandori, H. and Shichida, Y. (2000) Direct Observation of the Bridged Water
42
43 Stretching Vibrations Inside a Protein, *J. Am. Chem. Soc.* 122, 11745-11746.
44
45

46 (105) Venyaminov, S. Yu. and Prendergast, F. G. (1997) Water (H₂O and D₂O) Molar
47
48 Aborptivity in the 1000-4000 cm⁻¹ Range and Quantitative Infrared Spectroscopy of Aqueous
49
50 Solutions, *Anal. Biochem.* 248, 234-245.
51
52
53
54
55
56
57
58
59
60

1
2
3 (106) Kimura, Y., Mizusawa, N., Ishii, A., Yamanari, T., and Ono, T.-A. (2003) Changes
4 of Low-Frequency Vibrational Modes Induced by Universal ^{15}N - and ^{13}C -Isotope Labeling in
5 S_2/S_1 FTIR Difference Spectrum of Oxygen-Evolving Complex, *Biochemistry* 42, 13170-13177.
6
7

8
9 (107) Nakamura, S. and Noguchi, T. (2016) Quantum mechanics/molecular mechanics
10 simulation of the ligand vibrations of the water-oxidizing Mn_4CaO_5 cluster in photosystem II,
11 *Proc. Natl. Acad. Sci. USA* 113, 12727-12732.
12
13

14 (108) Chu, H.-A., Debus, R. J., and Babcock, G. T. (2001) D1-Asp170 is Structurally
15 Coupled to the Oxygen Evolving Complex in Photosystem II as Revealed by Light-Induced
16 Fourier Transform Infrared Difference Spectroscopy, *Biochemistry* 40, 2312-2316.
17
18

19 (109) Debus, R. J., Strickler, M. A., Walker, L. M., and Hillier, W. (2005) No Evidence
20 from FTIR Difference Spectroscopy That Aspartate-170 of the D1 Polypeptide Ligates a
21 Manganese Ion That Undergoes Oxidation during the S_0 to S_1 , S_1 to S_2 , or S_2 to S_3 Transitions in
22 Photosystem II, *Biochemistry* 44, 1367-1374.
23
24

25 (110) Kimura, Y., Mizusawa, N., Ishii, A., Nakazawa, S., and Ono, T.-A. (2005) Changes
26 in Structural and Functional Properties of Oxygen-Evolving Complex Induced by Replacement
27 of D1-Glutamate 189 with Glutamine in Photosystem II: Ligation of Glutamate 189 Carboxylate
28 to the Manganese Cluster, *J. Biol. Chem.* 280, 37895-37900.
29
30

31 (111) Strickler, M. A., Hillier, W., and Debus, R. J. (2006) No Evidence from FTIR
32 Difference Spectroscopy that Glutamate-189 of the D1 Polypeptide Ligates a Mn Ion that
33 Undergoes Oxidation During the S_0 to S_1 , S_1 to S_2 , or S_2 to S_3 Transitions in Photosystem II,
34 *Biochemistry* 45, 8801-8811.
35
36

37 (112) Debus, R. J. (2016) Identifying carboxylate ligand vibrational modes in photosystem
38 II with QM/MM methods, *Proc. Natl. Acad. Sci. USA* 113, 12613-12615.
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 (113) Strickler, M. A., Walker, L. M., Hillier, W., and Debus, R. J. (2005) Evidence from
4 Biosynthetically Incorporated Strontium and FTIR Difference Spectroscopy that the C-Terminus
5 of the D1 Polypeptide of Photosystem II Does Not Ligate Calcium, *Biochemistry* 44, 8571-8577.
6
7

8
9
10 (114) De Riso, A., Jenson, D. L., and Barry, B. A. (2010) Calcium Exchange and
11 Structural Changes during the Photosynthetic Oxygen Evolving Cycle, *Biophys. J.* 91, 1999-
12 2008.
13
14

15
16
17 (115) Suzuki, H., Taguchi, Y., Sugiura, M., Boussac, A., and Noguchi, T. (2006) Structural
18 Perturbations of the Carboxylate Ligands to the Mn Cluster upon $\text{Ca}^{2+}/\text{Sr}^{2+}$ Exchange in the S-
19 state Cycle of Photosynthetic Oxygen Evolution as Studied by Flash-Induced FTIR Difference
20 Spectroscopy, *Biochemistry* 45, 13454-13464.
21
22
23

24
25
26 (116) Polander, B. C. and Barry, B. A. (2013) Calcium, Strontium, and Protein Dynamics
27 during the S_2 to S_3 Transition in the Photosynthetic Oxygen-Evolving Cycle, *J. Phys. Chem. Lett.*
28 4, 3356-3362.
29
30
31

32
33 (117) Noguchi, T., Sugiura, M., and Inoue, Y. (1999) FTIR Studies on the Amino-Acid
34 Ligands of the Photosynthetic Oxygen-Evolving Mn-Cluster, in *Fourier Transform*
35 *Spectroscopy: Twelfth International Conference* (Itoh, K. and Tasumi, M., Eds.) pp 459-460,
36 Waseda University Press, Tokyo, Japan.
37
38
39
40

41
42 (118) Service, R. J., Yano, J., McConnell, I., Hwang, H. J., Nicks, D., Hille, R.,
43 Wydrzynski, T., Burnap, R. L., Hillier, W., and Debus, R. J. (2011) Participation of Glutamate-
44 354 of the CP43 Polypeptide in the Ligation of Manganese and the Binding of Substrate Water in
45 Photosystem II, *Biochemistry* 50, 63-81.
46
47
48
49

50
51 (119) del Val, C., Bondar, L., and Bondar, A.-N. (2014) Coupling between inter-helical
52 hydrogen bonding and water dynamics in a proton transporter, *J. Struct. Biol.* 186, 95-111.
53
54
55
56
57
58

1
2
3 (120) Koua, F. H. M., Umena, Y., Kawakami, K., and Shen, J.-R. (2013) Structure of Sr-
4 substituted photosystem II at 2.1 Å resolution and its implications in the mechanism of water
5 oxidation, *Proc. Natl. Acad. Sci. USA* 110, 3889-3894.
6
7

8
9
10 (121) Terrett, R., Petrie, S., Pace, R. J., and Stranger, R. (2014) What does the Sr-
11 substituted 2.1 Å resolution crystal structure of photosystem II reveal about the water oxidation
12 mechanism?, *Chem. Commun.* 50, 3187-3190.
13
14

15
16 (122) Pitari, F., Bovi, D., Narzi, D., and Guidoni, L. (2015) Characterization of the Sr²⁺-
17 and Cd²⁺-Substituted Oxygen-Evolving Complex of Photosystem II by Quantum
18 Mechanics/Molecular Mechanics Calculations, *Biochemistry* 54, 5959-5968.
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

For Table of Contents Use Only

Impact of D1-V185 on the Water Molecules that facilitate O₂ Formation by the Catalytic Mn₄CaO₅ Cluster in Photosystem II[†]

Christopher J. Kim, Han Bao, Robert L. Burnap, and Richard J. Debus*

