



Magnesium Isotopes as a Tracer of Crustal Materials in Volcanic Arc Magmas in the Northern Cascade Arc

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Fifteen North Cascade Arc basalts and andesites were analyzed for Mg isotopes to investigate the extent and manner of crustal contributions to this magmatic system. The $\delta^{26}\text{Mg}$ of these samples vary from within the range of ocean island basalts (the lightest being $-0.33 \pm 0.07\text{‰}$) to heavier compositions (as heavy as $-0.15 \pm 0.06\text{‰}$). The observed range in chemical and isotopic composition is similar to that of other volcanic arcs that have been assessed to date in the circum-pacific subduction zones and in the Caribbean. The heavy Mg isotope compositions are best explained by assimilation and fractional crystallization within the deep continental crust with a possible minor contribution from the addition of subducting slab-derived fluids to the primitive magma. The bulk mixing of sediment into the primitive magma or mantle source and the partial melting of garnet-rich peridotite are unlikely to have produced the observed range of Mg isotope compositions. The results show that Mg isotopes may be a useful tracer of crustal input into a magma, supplementing traditional methods such as radiogenic isotopic and trace element data, particularly in cases in which a high fraction of crustal material has been added.

Keywords: magnesium isotopes, assimilation and fractional crystallization, North Cascade Arc, arc volcanism, subduction zone

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INTRODUCTION

Volcanic arcs involve significant crust-mantle interactions, particularly through assimilation and fractional crystallization as magma rises to the surface and/or the addition of subducted sediment, altered oceanic crust, and slab-derived fluids to the mantle source or the primitive magma (Kelemen et al., 2007; Plank, 2014). This varied crustal material can be a major determinant of the final composition of these igneous rocks. Quantifying the crustal contribution to these magmas, and the mechanism(s) by which these materials were added, informs our understanding of the petrogenetic history of volcanic arcs, motivating the development of chemical tracers of these processes.

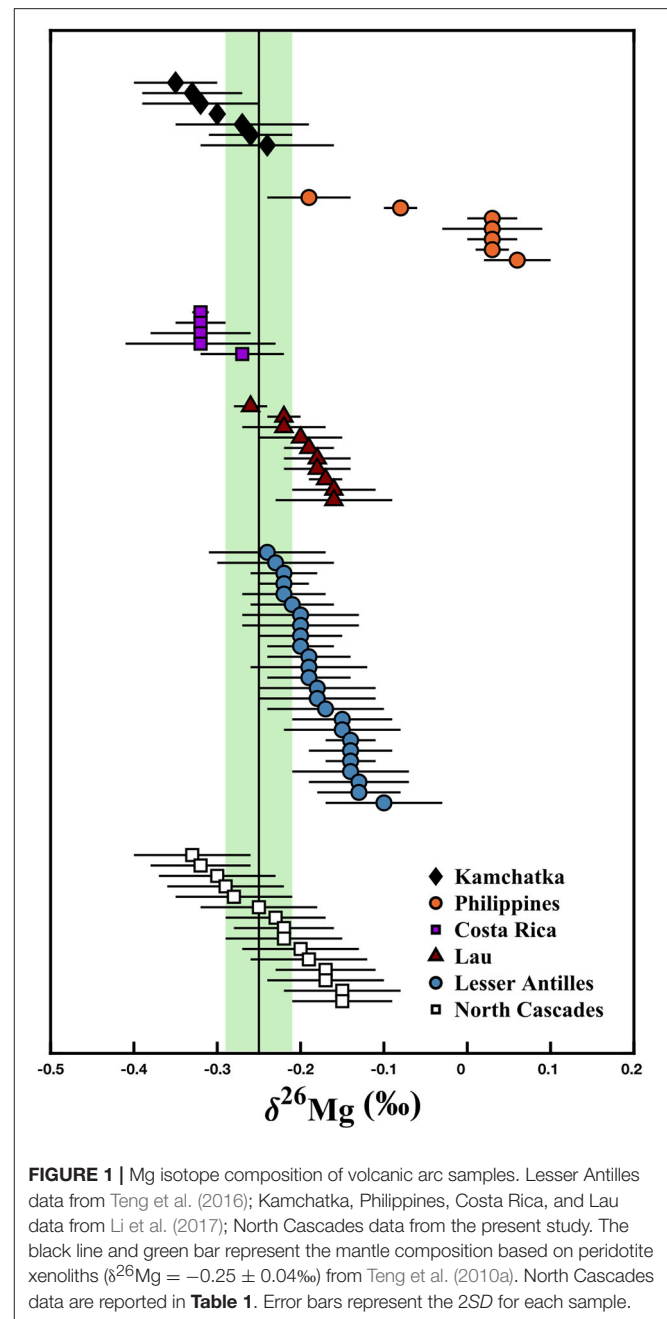
The unique systematics of Mg isotopes in mantle and crustal materials offer a new approach for tracing crustal additions to arc magmas. With a few exceptions, most significant Mg isotope fractionation occurs at the low temperatures of Earth's surface, while little fractionation occurs at high temperatures and pressures (Teng, 2017 and references therein). The Mg isotope compositions of uncontaminated mid-ocean ridge basalts and of the mantle are $\delta^{26}\text{Mg} = -0.25 \pm 0.07$ and $-0.25 \pm 0.04\text{‰}$, respectively (Teng et al., 2010a). Large deviations from the well-constrained mantle Mg isotope composition in an unweathered mafic rock may be indicative of crustal input

to the magma. Surface material displays a wide range of $\delta^{26}\text{Mg}$ values, from -5.57 to $+1.81\text{‰}$, reflecting the variety of fractionation mechanisms possible at low temperatures (Teng, 2017). During chemical weathering, for example, light isotopes are preferentially removed from a rock, which in combination with variations in watershed lithology, produces isotopically light river (Pogge von Strandmann et al., 2008; Tipper et al., 2008; Teng et al., 2010b; Huang et al., 2012) and ocean water ($-0.83 \pm 0.07\text{‰}$; Foster et al., 2010; Ling et al., 2011). The residual silicate rock and sediment is typically isotopically heavy due to these same processes (as heavy as $+1.81\text{‰}$; Tipper et al., 2008, 2010; Li et al., 2010; Teng et al., 2010b; Liu et al., 2014). This chemical weathering and other associated processes, such as carbonate precipitation, have produced an extremely heterogeneous upper crust in regard to Mg isotopes, and the subduction of this material has also resulted in isotopically heterogeneous lower crust and mantle wedge material (Li et al., 2010; Teng et al., 2013; Yang et al., 2016; Wang et al., 2017).

Recent investigations into the Mg isotope systematics of volcanic arc systems in the circum-Pacific subduction zones and in the Caribbean have found samples with a wide range of Mg isotope compositions ($-0.35 \pm 0.05\text{‰}$ to $+0.06 \pm 0.04\text{‰}$), which has been attributed to variable slab-derived fluid additions (Figure 1; Teng et al., 2016; Li et al., 2017). To explore the effects of crustal contamination on Mg isotopes in volcanic arcs, we measured 15 samples from the North Cascade Volcanic Arc, where both uncontaminated primary magma and contaminated, evolved magmas have been previously observed. These samples are geochemically well-characterized and thought to record a range of crustal content due the addition of subducted material and/or crustal assimilation during magma transport (Figure 2; Mullen and Weis, 2013, 2015; Mullen and McCallum, 2014; Mullen et al., 2017). Our results show that the Mg isotope compositions of these arc magmas vary from within the established values for uncontaminated mantle-derived mafic magmas to relatively heavy compositions, likely due to crustal contamination (Figure 1). The use of Mg isotopes as a tracer of crustal material in magma is not limited to specific sites, but could be applicable to a variety of volcanic arc settings (Teng et al., 2016; Li et al., 2017).

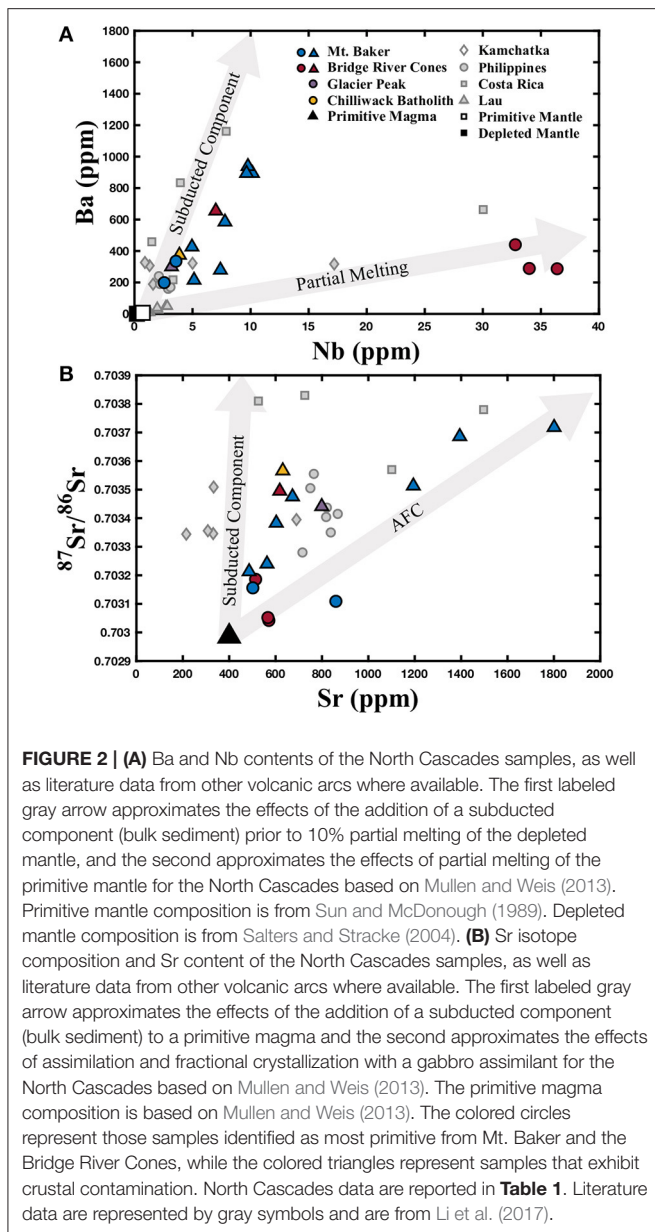
SAMPLES

The Cascade Arc, located on the western margin of North America, is the result of the subduction of the Juan de Fuca plate beneath the North American plate. Most of the associated magmatic rocks contain geochemical evidence for the addition of subducted material, derived from both oceanic crust and sediment, to their mantle sources (Mullen et al., 2017). In addition, some magmas assimilated continental crust during transit to the surface (Mullen et al., 2017). We choose 15 samples from the North Cascade Volcanic Arc for Mg isotope analysis in an effort to cover samples exhibiting a wide range of known crustal content, from negligible to considerable, based on trace element compositions and Sr-Nd-Hf-Pb isotope data. The analyzed samples represent a range of locations and



rock types within the northern segment of the Cascade Arc, including the Mt. Baker volcanic field, the Bridge River Cones, Glacier Peak, and the Chilliwack batholith (Tepper, 1996; Mullen and Weis, 2013, 2015; Mullen and McCallum, 2014; Mullen et al., 2017). To ensure that no weathering has occurred, all samples were collected from the interior of lava flows, were confirmed to have low LOI contents ($<1\%$), and were examined for signs of weathering in thin section (Mullen and Weis, 2013).

The North Cascade samples fall into three petrologic groups: (1) alkali basalts from the Bridge River Cones; (2) calc-alkaline arc basalts from Mt. Baker, Glacier Peak, and the Chilliwack



batholith; and (3) andesites from Mt. Baker and the Bridge River Cones. The alkali basalts from the Bridge River Cones display trace element and Sr-Nd-Hf-Pb isotope compositions that indicate an enriched mantle source and the absence of a subducted component (Figure 2; Mullen and Weis, 2013). A slab gap between the stagnant Explorer and Juan de Fuca plates near the northern termination of the arc likely facilitated upwelling of enriched asthenosphere that produced these alkali basalts nearly free of slab input (Mullen and Weis, 2013, 2015; Mullen et al., 2017). One andesite from the Bridge River Cones does show signs of crustal assimilation, like other andesites from the North Cascades (Figure 2; Mullen and Weis, 2013). Previous work has found that the calc-alkaline basalts from Mt. Baker, Glacier Peak, and the Chilliwack batholith, as well

as andesites from Mt. Baker and the Bridge River Cones, demonstrate a clear subducted component, marked by elevated abundances of lead (Pb), large ion lithophile elements (LILE), and light rare earth elements (LREE) relative to high field strength elements (HSFE) (Figure 2; Mullen and Weis, 2013, 2015; Mullen and McCallum, 2014). Furthermore, the $^{87}\text{Sr}/^{86}\text{Sr}$ vs. Sr systematics indicate the involvement of assimilation and fractional crystallization in the deep crust in some samples (Figure 2; Mullen and Weis, 2013, 2015; Mullen and McCallum, 2014).

METHODS

Magnesium isotope analyses were performed at the Isotope Laboratory in the Department of Earth and Space Sciences at the University of Washington. All procedures are similar to those reported in previous publications (Teng et al., 2007, 2010a; Yang et al., 2009; Li et al., 2010; Teng and Yang, 2014).

The powdered rock samples were dissolved in Savillex screw-top Teflon beakers on a hotplate at $\sim 125^\circ\text{C}$ using a 3:1 HF-HNO₃ acid mixture followed by a 3:1 HCl-HNO₃ acid mixture. The samples were kept in each acid mixture for several days to ensure complete dissolution. To achieve cation separation, the samples were passed twice through a column containing Bio-Rad 200–400 mesh AG50W-X8 cation exchange resin in a 1N HNO₃ media, and were eluted using 1N HNO₃. The resulting Mg cut contains >99% of the Mg in the sample to prevent isotope fractionation within the column, and limits the concentrations of the other elements to <5% of the concentration of Mg (Teng et al., 2007).

The purified Mg samples were then analyzed on a Nu Plasma MC-ICP-MS using the standard-sample bracketing method. The Mg concentrations of the sample and standard solutions were matched to within 5% to prevent mass bias caused by a concentration mismatch (Teng and Yang, 2014). The solutions contained ~ 300 ppb Mg in 3% nitric acid. The samples were introduced using the “wet” plasma method and ^{24}Mg , ^{25}Mg , and ^{26}Mg were analyzed simultaneously using three Faraday cups (H5, Ax, and L4). Results are presented in delta notation:

$$\delta^x\text{Mg} (\text{‰}) = \left[\frac{\left(\frac{x\text{Mg}}{^{24}\text{Mg}} \right)_{\text{Sample}}}{\left(\frac{x\text{Mg}}{^{24}\text{Mg}} \right)_{\text{Standard}}} - 1 \right] * 1000 \quad (1)$$

where x refers to mass 25 or 26.

The precision of the measured $^{26}\text{Mg}/^{24}\text{Mg}$ ratio for one sample solution at the 2SD level, based on repeat standard analyses during a single analytical session, is $< \pm 0.07\text{‰}$, comparable to previous Mg isotope studies (Li et al., 2010; Teng et al., 2010a; Ling et al., 2011; Teng and Yang, 2014). The reference materials San Carlos Olivine ($n = 4$) and Seawater ($n = 3$) were each analyzed at least once during an analytical session, each time yielding a $\delta^{26}\text{Mg}$ value within the established 2SD of -0.25 and -0.83‰ , respectively. These values agree

with previously published data (Teng et al., 2015; Hu et al., 2016a).

RESULTS

Magnesium isotopic compositions, MgO concentrations, relevant trace element data, and Sr isotope compositions for the North Cascade Arc are reported in **Table 1**. The Mt. Baker samples range from $\delta^{26}\text{Mg} = -0.32$ to $-0.15 \pm 0.06\text{‰}$; the Bridge River Cones samples range from $\delta^{26}\text{Mg} = -0.33$ to $-0.22 \pm 0.07\text{‰}$; the Mg isotope composition of the Chilliwack Batholith and Glacier Peak samples is $\delta^{26}\text{Mg} = -0.23$ and $-0.22 \pm 0.07\text{‰}$, respectively. Weak correlations between $\delta^{26}\text{Mg}$ and MgO content, $^{87}\text{Sr}/^{86}\text{Sr}$, Th/Yb, and Pb/Ce may be evident in these samples, but given the number of processes that can affect the chemical and isotopic composition of arc magmas, clear correlations are not expected (**Figures 3, 5**). No correlation is observed between Mg isotope composition and Ba/Th, Dy/Yb, or Sm/Yb (**Figures 4, 5**). The increase in $\delta^{26}\text{Mg}$ from the

MORB-like composition is small; however, it is resolvable with the current precision ($\leq 0.07\text{‰}$).

DISCUSSION

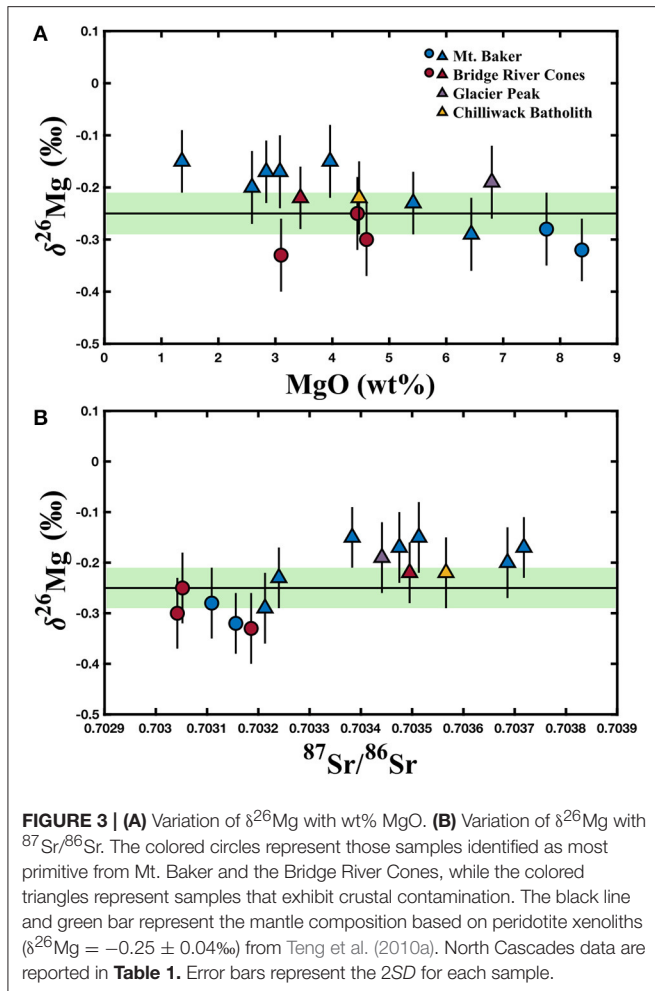
Mg Isotope Variations in the North Cascade Volcanic Arc

The alkali basalts from the Bridge River Cones are essentially free of crustal contamination based on trace element and Sr-Nd-Hf-Pb isotope data; a conclusion which is supported by Mg isotope systematics (**Figure 2**; Mullen and Weis, 2013). The samples, with $\delta^{26}\text{Mg}$ between -0.33 and $-0.25 \pm 0.07\text{‰}$, fall within the range of Mg isotope compositions exhibited by MORBs ($\delta^{26}\text{Mg} = -0.31$ to -0.19‰) and OIBs ($\delta^{26}\text{Mg} = -0.35$ to -0.18‰) (Teng et al., 2010a). They do not reach the heavier compositions found elsewhere in the North Cascades, for example at Mt. Baker (**Figure 3**). Thus, the Mg isotope composition of these basalts supports the conclusion that the Bridge River Cones likely reflect the melting of a primitive mantle

TABLE 1 | Magnesium isotope and trace element compositions of samples and standards^a.

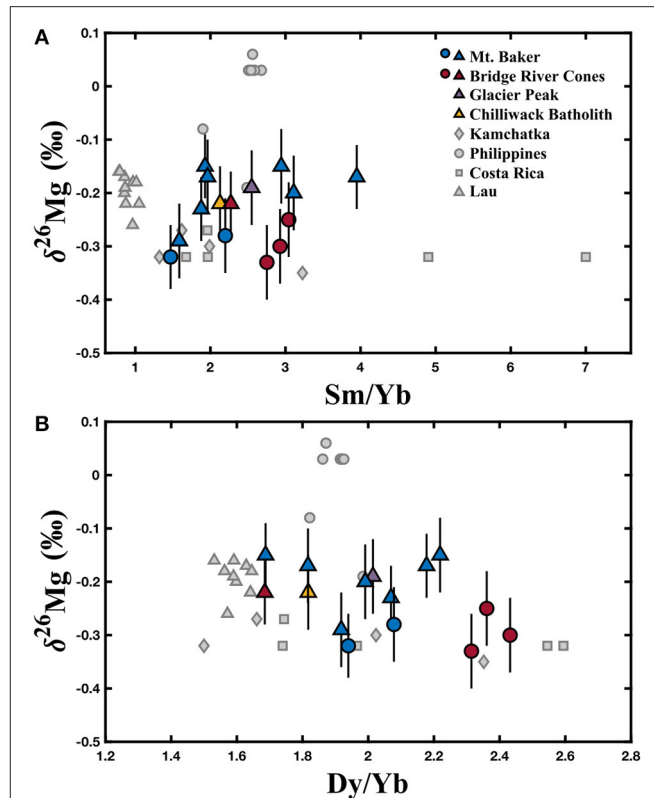
Sample	$\delta^{26}\text{Mg}$ (‰)	2SD	$\delta^{25}\text{Mg}$ (‰)	2SD	MgO (wt%)	Sm/Yb	Dy/Yb	Th/Yb	Pb/Ce	Ba/Th	Ba (ppm)	Nb (ppm)	Sr (ppm)	$^{87}\text{Sr}/^{86}\text{Sr}$
NORTH CASCADE ARC														
Mt. BAKER														
Tarn Plateau	-0.28	0.07	-0.19	0.06	7.76	2.199	2.078	1.295	0.122	155.8	335	3.57	860	0.703109
Sulfur Creek	-0.23	0.06	-0.13	0.04	5.42	1.878	2.069	0.446	0.084	206.7	279	7.40	563	0.703240
Lake Shannon	-0.29	0.07	-0.19	0.06	6.44	1.586	1.918	0.410	0.100	195.5	215	5.13	486	0.703213
Park Butte	-0.32	0.06	-0.18	0.04	8.38	1.470	1.939	0.303	0.088	331.7	199	2.56	502	0.703156
Cathedral Crag	-0.15	0.07	-0.11	0.06	3.96	2.944	2.219	1.163	0.097	229.0	426	4.95	1194	0.703513
Coleman Pinnacle 101B	-0.15	0.06	-0.09	0.04	1.36	1.929	1.688	3.326	0.192	120.0	894	10.16	603	0.703383
Coleman Pinnacle 105	-0.17	0.06	-0.10	0.04	2.84	3.949	2.178	3.010	0.185	158.2	938	9.77	1801	0.703718
Coleman Pinnacle 86	-0.20	0.07	-0.08	0.06	2.59	3.110	1.991	2.749	0.187	148.5	894	9.65	1394	0.703686
Table Mountain	-0.17	0.07	-0.12	0.06	3.08	1.963	1.817	1.846	0.169	132.1	585	7.79	673	0.703475
BRIDGE RIVER CONES														
Tuber Hill East Dot	-0.22	0.06	-0.08	0.04	3.44	2.272	1.685	1.568	0.173	257.9	655	7.0	618	0.703495
Tuber Hill East Cap	-0.30	0.07	-0.16	0.06	4.60	2.928	2.432	0.808	0.043	143.1	289	34.0	571	0.703042
Tuber Hill East Plateau	-0.33	0.07	-0.16	0.06	3.10	2.752	2.314	0.625	0.077	212.6	440	32.8	514	0.703186
Nichols Valley	-0.25	0.07	-0.14	0.06	4.44	3.043	2.361	0.816	0.037	140.0	287	36.4	567	0.703052
CHILLIWACK BATHOLITH														
Mount Sefrit	-0.23	0.07	-0.09	0.06	6.80	2.550	2.015	1.329	0.164	161.3	300	3.19	799	0.703441
GLACIER PEAK														
Dishpan Gap	-0.22	0.07	-0.14	0.06	4.47	2.129	1.817	1.378	0.139	112.7	374	3.87	631	0.703566
STANDARDS														
Seawater	-0.81	0.07	-0.43	0.06										
Replicate	-0.88	0.07	-0.54	0.06										
Replicate	-0.83	0.07	-0.50	0.06										
SC Olivine	-0.23	0.07	-0.05	0.06										
Duplicate	-0.25	0.07	-0.20	0.06										
Duplicate	-0.31	0.06	-0.15	0.04										
Replicate	-0.24	0.06	-0.11	0.04										

^aNorth Cascades major and trace element data from Mullen and McCallum (2013, 2014); Mullen and Weis (2015); Mullen et al. (2017). 2SD = Two standard deviation of the population of n ($15 < n < 22$) repeat measurements of the standards during an analytical session.



source with little to no contribution from subducted material or assimilated continental crust (**Figure 3**; Mullen and Weis, 2013).

The calc-alkaline arc basalts from Mt. Baker, Glacier Peak, and the Chilliwack batholith and the andesites from Mt. Baker and the Bridge River Cones do show the influence of subducted material and/or assimilation and fractional crystallization of the continental crust (Mullen and McCallum, 2014). While all of these samples are thought to exhibit some crustal contamination, we consider the Tarn Plateau and Park Butte samples to be the most primitive, given their high MgO contents (>7 wt%), high Mg# (>0.6), and relatively high compatible trace element content (e.g., Cr > 200 ppm; Mullen and Weis, 2013, 2015; Mullen and McCallum, 2014). The Mg isotope compositions of these samples are also quite light (-0.28 ± 0.07 and $-0.32 \pm 0.07\text{‰}$, respectively), similar to the primitive alkali basalts from the Bridge River Cones, which are known to contain little crustal contamination (**Figure 3**). The remainder of the calc-alkaline basalts and andesites display variable crustal involvement from subducted material and/or assimilated continental crust (Tepper, 1996; Mullen and Weis, 2013, 2015; Mullen and McCallum, 2014). The relatively heavy Mg isotope compositions of these samples likely reflects the addition of isotopically heavy crustal

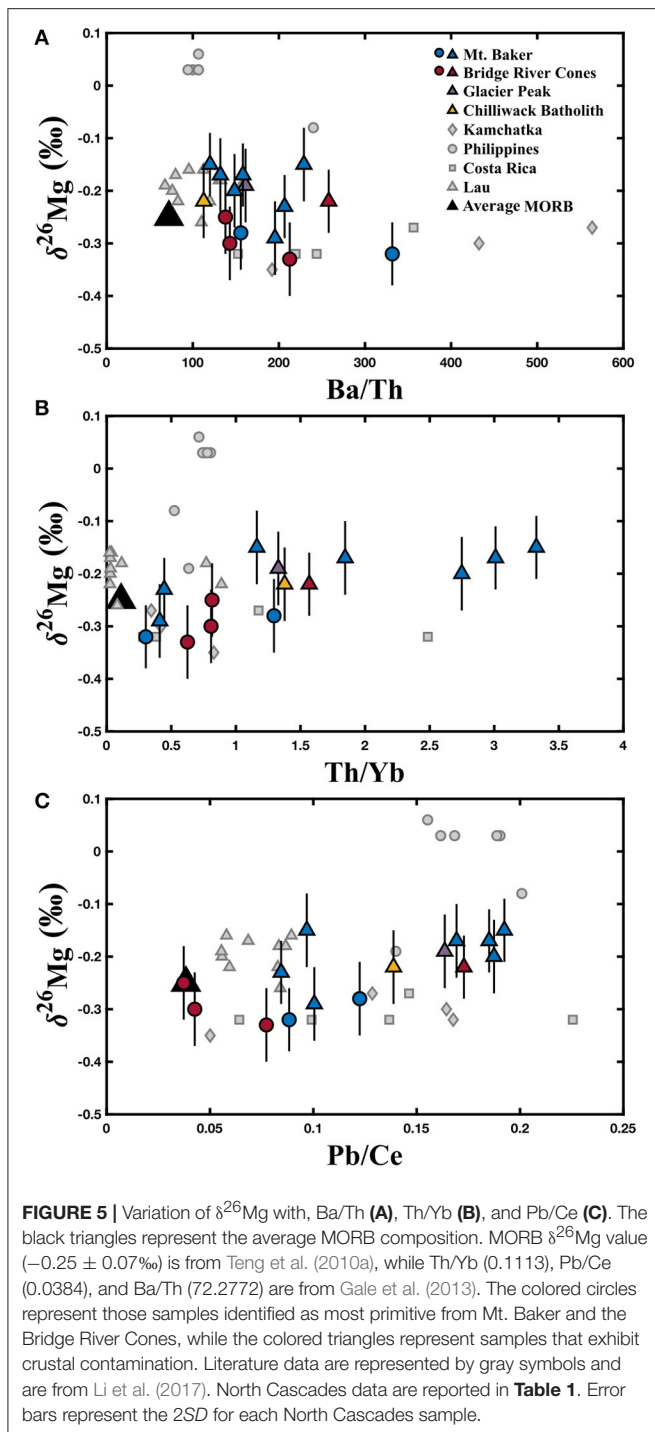


material, such as subducted sediment ($\delta^{26}\text{Mg} = -3.65$ to $+0.52\text{‰}$), altered oceanic crust ($\delta^{26}\text{Mg} = -2.76$ to $+0.21\text{‰}$), and deep continental crust ($\delta^{26}\text{Mg} = -0.76$ to $+0.19\text{‰}$) (**Figure 3**; Huang, 2013; Teng et al., 2013; Yang et al., 2016; Hu et al., 2017). The following sections will examine the possible causes of the elevated Mg isotope compositions of these arc samples.

Hypotheses for Mg Isotope Variations

The observed increase in Mg isotope composition may be the result of one of three processes, or a combination thereof: the primary melt was derived from partial melting of garnet-rich peridotite, isotopically heavy subducted material was added to the primitive magmas or the mantle source, or the primitive magmas underwent assimilation and fractional crystallization during transport within the continental crust.

Garnet has a light Mg isotope composition compared to coexisting silicates (Li et al., 2011, 2016; Wang et al., 2012, 2014; Huang et al., 2013). The difference in Mg coordination between garnet (eight-fold) and most silicate minerals (six-fold) causes ^{24}Mg to be preferentially incorporated into garnet, leaving the surrounding minerals enriched in ^{26}Mg (Li et al.,



2011, 2016; Wang et al., 2012, 2014; Huang et al., 2013). Both equilibrium and disequilibrium inter-mineral fractionation can occur between garnet and coexisting silicates (Li et al., 2016). Therefore, partial melting that leaves a garnet-rich residue could produce a melt with a relatively heavy Mg isotope composition. However, the trace element data do not support this explanation for the origin of the isotopically heavy magmas. Because of the affinity of the HREE for garnet, partial melts leaving a

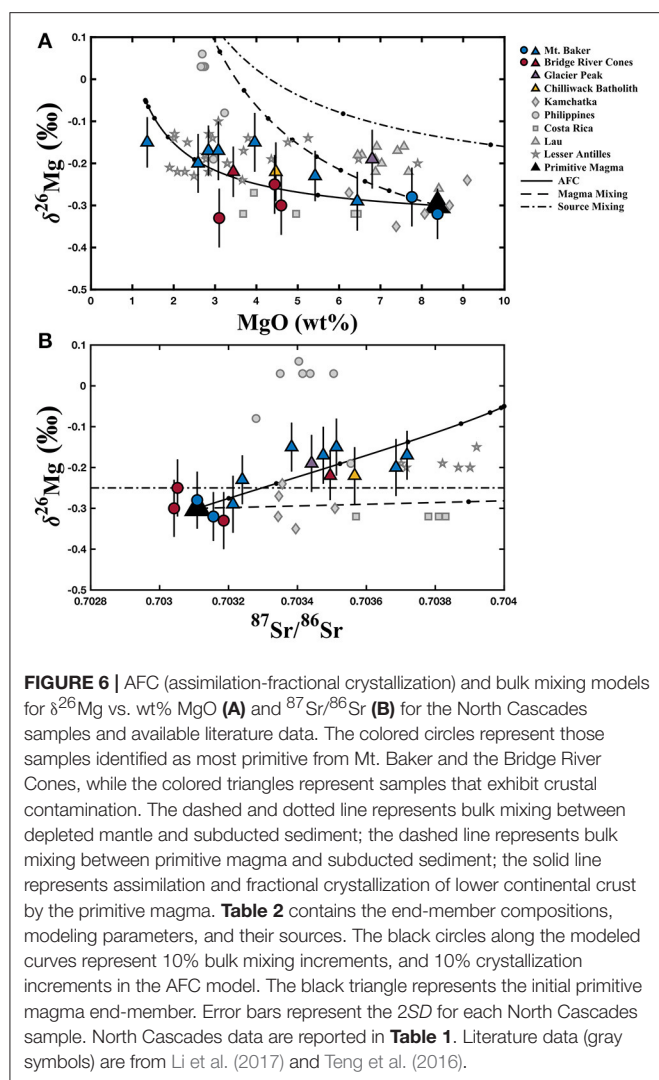
garnet-rich residue would also display relatively high Sm/Yb and Dy/Yb ratios (Lassiter and DePaolo, 1997). Therefore, $\delta^{26}\text{Mg}$ should be positively correlated with Sm/Yb and Dy/Yb, which is not observed in the North Cascades or any other volcanic arc assessed to date (Figure 4). Furthermore, previous work has ruled out residual garnet for Mt. Baker and Glacier Peak based on trace element modeling (Mullen and Weis, 2013, 2015). Residual garnet associated with the Bridge River Cones is likely, however, those samples do not exhibit heavy Mg isotope compositions (Mullen and Weis, 2013). We therefore conclude that partial melting in the presence of garnet is unlikely to have produced the observed Mg isotope compositions.

The addition of a crustal component to the magmas is more consistent with the Mg isotope data. Because crustal material, such as subducted sediment, generally has lower MgO contents than the mantle source and primitive magmas (McDonough and Sun, 1995; Rudnick and Gao, 2003; Plank, 2014), crustal input into a magma is generally associated with a decrease in MgO. Fractional crystallization can produce similar changes in chemical composition; however, fractional crystallization does not fractionate Mg isotopes, with the exception of processes associated with garnet (Teng, 2017 and references therein). A heavy Mg isotope composition in these samples is also associated with increased $^{87}\text{Sr}/^{86}\text{Sr}$, which is expected given that crustal materials have elevated $^{87}\text{Sr}/^{86}\text{Sr}$ compared to most mantle-derived melts (Kelemen et al., 2007; Plank, 2014). Slab-derived fluid additions to the mantle wedge are typically associated with increases in Pb/Ce and Ba/Th ratios, while subducted sediment melts are typically associated with increased Th/Yb ratios. Weak correlations between $\delta^{26}\text{Mg}$ and Pb/Ce and Th/Yb may be present in these samples, but no correlation with Ba/Th is observed (Figure 3). Taken together, the North Cascade Volcanic Arc samples exhibit a range of Mg isotope compositions that is best explained by crustal additions, rather than garnet involvement.

Modeling Crustal Input

To investigate the origin of crustal contributions to the North Cascade Arc, an AFC model (DePaolo, 1981) combining primitive magma and lower continental crust and two different two-component bulk-mixing models (primitive magma + subducted sediment and depleted mantle + subducted sediment) were constructed for the North Cascade Arc (Figure 6). The models are designed to be representative of the calc-alkaline arc basalts from Mt. Baker, Glacier Peak, and the Chilliwaack batholith and the andesites from Mt. Baker and the Bridge River Cones, but not the alkali basalts from the Bridge River Cones since they are known to be petrogenetically distinct from the other samples. All end-member compositions and other model parameters are listed in Table 2.

The primitive magma composition was selected based on the samples judged to be the most primitive in the Mt. Baker dataset (Tarn Plateau and Park Butte). The depleted mantle composition is from Salters and Stracke (2004), with the exception of the Mg isotope composition (-0.25‰), which is the canonical mantle value described in Teng et al. (2010a) and elsewhere (Huang et al., 2011; Hu et al., 2016b; Wang et al., 2016). The lower



continental crust composition is based on the North Cascades lower crust end-member from Mullen and Weis (2013). The Mg isotope composition of the lower crust in this region is unknown, so a reasonable composition (-0.05‰) was selected based on the range exhibited by lower crustal material in Yang et al. (2016). Finally, the subducted sediment end-member is based on the subducted sediment at the Cascades from Plank (2014). However, the Mg isotope composition of the subducted sediment component affecting the mantle source or primitive magma is unknown, so again a reasonable value ($+0.20\text{‰}$) was selected based on Hu et al. (2017). For the AFC models, the ratio of assimilation to crystallization was fixed at 0.5, and the bulk partition coefficients for MgO and Sr were estimated and fixed at 3.0 and 3.2, respectively, after the volcanic arc modeling of Teng et al. (2016). Minor changes in the bulk partition coefficients have little impact on the model. Although some of the relevant modeling parameters can only be estimated, these models do demonstrate that AFC is a feasible explanation for the observed data, while bulk mixing alone is not.

TABLE 2 | Modeling parameters and end-member compositions.

Sample	$\delta^{26}\text{Mg}$ (‰)	MgO (wt%)	Sr (ppm)	$^{87}\text{Sr}/^{86}\text{Sr}$
END-MEMBER COMPOSITIONS				
Depleted mantle	-0.25^a	38.2 ^d	9.8 ^d	0.70260 ^d
Primitive magma	-0.30	8.38	502	0.70311
Lower crust	-0.05^b	6.57 ^e	401 ^e	0.70400 ^e
Subducted sediment	$+0.20^c$	2.53 ^f	323 ^f	0.71490 ^f
MODELING PARAMETERS				
D_{Sr}	3.2 ^g			
D_{MgO}	3.0 ^g			
r	0.5 ^g			

^aTeng et al. (2010a).

^bYang et al. (2016).

^cHu et al. (2017).

^dSalter and Stracke (2004).

^eMullen and Weis (2013).

^fPlank (2014).

^gTeng et al. (2016).

The North Cascade Arc two-component bulk mixing models are unable to reproduce the observed trends in the data, particularly the Sr isotope compositions. To approximate the MgO content and Mg isotope composition of some North Cascades samples with the primitive magma mixing model, the sediment must contribute more than 50% of the final magma, which is unreasonably high (**Figure 6A**). The Sr and Mg isotope primitive magma mixing model predicts very little change in Mg isotope composition across the whole range of Sr isotope compositions observed in the samples, which does not agree with the observed data (**Figure 6B**). The two-component mantle source mixing model requires an even higher fraction of crustal material (more than 80%) to produce the observed Mg isotope compositions given the high MgO content of the mantle, and again no change in Mg isotope composition is predicted for the observed range in Sr isotopes (**Figure 6**). Therefore, bulk sediment additions to the primitive magma or mantle source are unlikely to be the cause of the elevated Mg isotope compositions found in the North Cascade Arc samples.

Assimilation and fractional crystallization in the deep continental crust, on the other hand, can match the observed trends without an excessively high crustal contribution. The observed major element and isotopic trends can be reproduced with between 100 and 60% liquid remaining in the system (**Figure 6**). Unlike the bulk mixing models, AFC can approximate the observed increase in Mg isotope composition within the range of Sr isotope compositions exhibited by the samples (**Figure 6**). Given the elevated Pb/Ce and Ba/Th ratios, and the apparent correlation between $\delta^{26}\text{Mg}$ and Pb/Ce in these samples, fluid additions may also have made a minor contribution to the final Mg isotope compositions. However, quantifying the slab-derived fluid effect on Mg isotopes is not currently possible given our limited understanding of Mg isotope behavior during slab dehydration. Assimilation and fractional crystallization is the favored explanation for the increase in Mg isotope

composition, and while minor effects from the addition of slab-derived fluids to the mantle source and primitive may have occurred, those processes are unlikely to be the primary cause of the heavy Mg isotope compositions observed in the North Cascade Volcanic Arc.

The Mg isotope compositions exhibited by the North Cascade Volcanic Arc samples are similar to those from previously analyzed volcanic arcs (**Figure 1**). Li et al. (2017) analyzed arc samples from Kamchatka ($\delta^{26}\text{Mg} = -0.35 \pm 0.05$ to $-0.24 \pm 0.08\text{‰}$), the Philippines (-0.19 ± 0.05 to $+0.06 \pm 0.04\text{‰}$), Costa Rica (-0.32 ± 0.01 to $-0.27 \pm 0.05\text{‰}$), and Lau (-0.26 ± 0.02 to $-0.16 \pm 0.05\text{‰}$), while Teng et al. (2016) analyzed samples from the Lesser Antilles (-0.24 ± 0.07 to $-0.10 \pm 0.07\text{‰}$). The Cascade samples (-0.33 ± 0.07 to $-0.15 \pm 0.06\text{‰}$) have similar Mg isotope compositions to the samples from all of those locations except the Philippines, which have unusually heavy compositions (Li et al., 2017). The present dataset fills in a gap in the Mg isotope data for arc volcanics from the circum-Pacific subduction zones, representing the margin along the northwestern coast of North America. The similarity to these other arc samples, including those from the Lesser Antilles in the Caribbean, demonstrates that the processes affecting Mg isotopes are likely consistent from margin to margin. Much of the combined data for the volcanic arcs worldwide can be approximated by the same assimilation and fractional crystallization model used with the North Cascades samples (**Figure 6**). AFC processes are, therefore, the most likely explanation for the range in Mg isotope data observed in volcanic arcs. Small differences in the Mg isotope compositions of the primitive magma, subducted sediment, and assimilated lower crust between and within different arcs likely do impact the final magma composition, producing the observed variety between the arcs.

Although crustal input to the North Cascades magmas produced only small, just beyond two-sigma analytical uncertainty, variations from normal mantle values, this likely required a significant crustal addition. Nonetheless, improvements in analytical precision may make possible the use of Mg isotopes as a valuable tracer of crustal recycling that is complementary to established methods such as Sr-Nd-Hf-Pb isotopic and trace element data. The crustal materials that may be involved in arc volcanism, including subducted sediment and sediment melt, altered oceanic crust, slab-derived fluids, and assimilated continental crust, have extremely varied Mg isotope compositions. This variability, along with the lack of fractionation during most high temperature processes, may permit Mg isotopes to be a valuable tool in tracing arc volcanic processes. With a thorough understanding of the composition of these different reservoirs, Mg isotope systematics, combined with existing geochemical indicators, would elucidate the nature

of crustal input in a given arc. Furthermore, Mg isotopes may provide a solution for detecting crustal input in settings where commonly-used radiogenic isotopes are ambiguous due to lack of isotopic contrast between the crust and primary magmas (e.g., Mullen et al., 2017). These findings also represent a significant step in our understanding of the Mg cycle by demonstrating that silicate material from the crust can alter the isotopic composition of a magma.

SUMMARY

The main conclusions from this study are:

1. The Mg isotope composition of samples from the North Cascade Arc range from -0.33 ± 0.07 to $-0.15 \pm 0.06\text{‰}$.
2. The alkali basalts from the Bridge River Cones reflect the partial melting of the mantle source with minimal crustal contamination, while the calc-alkaline basalts and andesites from Mt. Baker, Glacier Peak, Chilliwack Batholith, and the Bridge River Cones do have Mg isotope compositions indicative of variable crustal contamination.
3. The samples with high $\delta^{26}\text{Mg}$ are best explained by the addition of isotopically heavy deep continental crust to primitive magmas through assimilation and fractional crystallization with a possible minor contribution from slab-derived fluids.
4. With further study and analytical improvements, Mg isotopes will become a useful tool in understanding the generation and evolution of magmatic rocks.

AUTHOR CONTRIBUTIONS

AB, F-ZT, and EM: Contributed to the conception of the presented idea; EM: Provided the analyzed samples; AB: Performed the relevant experiments; AB: Wrote the manuscript with support from EM and F-ZT.

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Conflict of Interest Statement: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

The reviewer, PS, declared a past co-authorship with one of the authors, F-ZT, to the handling editor.

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