Melt origin across a rifted continental margin: a case for subduction-

related metasomatic agents in the lithospheric source of alkaline

basalt, northwest Ross Sea, Antarctica

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### **ABSTRACT**

Alkaline magmatism associated with the West Antarctic rift system in the northwest Ross Sea (NWRS) includes a north-south chain of shield volcano complexes extending 260 km along the coast of Northern Victoria Land (NVL) as well as numerous small volcanic seamounts located on the continental shelf and hundreds more within a ~35,000 km² area of the oceanic Adare Basin. New <sup>40</sup>Ar/<sup>39</sup>Ar age dating and geochemistry confirm that the seamounts are Pliocene–Pleistocene in age and petrogenetically akin to the mostly middle to late Miocene volcanism on the continent as well as to a much broader region of diffuse alkaline volcanism that altogether encompasses areas of West Antarctica, Zealandia and eastern Australia. All of these continental regions were contiguous prior to the late–stage breakup of Gondwana at ~100 Ma, suggesting that the magmatism is interrelated, yet mantle source and cause of melting remain controversial. The NWRS provides a rare opportunity to study cogenetic volcanism across the transition from continent to ocean and consequently offers a unique perspective from which to evaluate mantle processes and roles of lithospheric and sub-lithospheric sources for mafic alkaline magmas.

Mafic alkaline magmas with > 6 wt.% MgO (alkali basalt, basanite, hawaiite, and tephrite) erupted across the transition from continent to ocean in the NWRS show a remarkable systematic increase in silica-undersaturation,  $P_2O_5$ , Sr, Zr, Nb and light rare earth element (LREE) concentrations as well as LREE/HREE and Nb/Y ratios. Radiogenic isotopes also vary with Nd and Pb ratios increasing and Sr ratios decreasing ocean-ward. These variations cannot be explained by shallow-level crustal contamination or by changes in degree of mantle partial melting but are considered to be a function of the thickness and age of the mantle lithosphere. We propose that the isotopic signature of the most silica-undersaturated and incompatible element enriched basalts best represent the composition of the sub-lithospheric source with low  $^{87}Sr/^{86}Sr$  ( $\leq 0.7030$ ) and  $\delta^{18}O_{olivine}$ 

( $\leq$  5.0 %), high <sup>143</sup>Nd/<sup>144</sup>Nd ( $\sim$  0.5130) and <sup>206</sup>Pb/<sup>204</sup>Pb ( $\geq$  20) ratios. The isotopic 'endmember' signature of the sub-lithospheric source is derived from recycled subducted materials and was transferred to the lithospheric mantle by small degree melts (carbonate-rich silicate liquids) to form amphibole-rich metasomes. Later melting of the metasomes produced silica-undersaturated liquids that reacted with the surrounding peridotite. This reaction occurred to a greater extent as the melt traversed through thicker and older lithosphere continent-ward.

Ancient and/or more recent (~550–100 Ma) subduction along the Pan-Pacific margin of Gondwana supplied the recycled subduction-related residue to the asthenosphere. Melting and carbonate metasomatism were triggered by major episodes of extension beginning in the Late Cretaceous but alkaline magmatism was very limited in its extent. Significant delay of ~30 to 20 Myr between extension and magmatism was likely controlled by conductive heating and the rate of thermal migration at the base of the lithosphere. Heating was facilitated by regional mantle upwelling, possibly driven by slab detachment and sinking into the lower mantle, and/or by edge–driven mantle flow established at the boundary between the thinned lithosphere of the West Antarctic rift and the thick East Antarctic craton.

## **INTRODUCTION**

The source and processes that generate alkaline magmas in intraplate continental settings, particularly in young rifts like East Africa and West Antarctica, remain enigmatic. The debate revolves broadly around whether melt is produced within the asthenosphere or lithospheric mantle, whether the melting and ultimately eruption is triggered by passive extension, lithospheric drip or upwelling plumes, and is further perpetuated by the uncertainty in how melt is modified as it rises through the continental plate. The Late Cenozoic magmatism found within the West Antarctic Rift System (WARS, Fig. 1A) has been explained by a variety of models that attempt to reconcile the petrology, geochemistry and isotopic signatures of mafic alkaline rocks all within the context of the last several hundred million years of Antarctica's tectonic history. Specifically the tectonic events leading up to and including the breakup of the proto-Pacific margin of Gondwana which separated Zealandia from West Antarctica in the Late Cretaceous.

A unifying model for WARS magmatism is warranted given the relative uniformity of mafic alkaline compositions emplaced within a very broad (~1500 x 2500 km) region of thinned continental lithosphere over the last 48 Ma. The mafic compositions are similar to ocean island basalt (OIB) and have HIMU-like isotopic affinities ( $^{206}\text{Pb}/^{204}\text{Pb} > 19.5$ ,  $^{87}\text{Sr}/^{86}\text{Sr} < 0.7035$ ,  $^{143}\text{Nd}/^{144}\text{Nd} > 0.5128$ ). Furthermore, the magmatism has been linked compositionally and tectonically to a much broader region of long-lived diffuse alkaline volcanism that altogether encompasses West Antarctica, Zealandia and eastern Australia (Finn *et al.*, 2005). Existing models for WARS magmatism and beyond, however, are quite varied but can be distilled into two fundamental types; those that involve mantle plumes and those that do not. Both types require mantle sources for alkaline magmas that have been enriched in incompatible minor and trace-elements relative to primitive mantle in order to explain their geochemical and radiogenic isotope compositions. For the mantle plume source, enrichments are considered to be via addition of

oceanic crust and sediment subducted and recycled deep within the mantle (Hofmann and White, 1982; Zindler and Hart, 1986; Weaver, 1991; Chauvel et al., 1992; Willbold and Stracke, 2010). Models proposed for West Antarctica include a mantle plume beneath the active Erebus volcano (Kyle et al., 1992), a large plume that may have helped initiate and control the position of continental break-up in the Late Cretaceous (Lanyon et al., 1993; Weaver *et al.*, 1994; Storey et al., 1999) and older "fossil" plumes proposed to have underplated and accreted to the base of the Gondwana lithosphere prior to break-up (Rocholl et al., 1995; Hart et al., 1997; Panter et al., 2000; Kipf et al., 2014). For non-plume sources the enrichment has been ascribed to mantle metasomatism caused either by small degrees of melting during the early phase of rifting prior to continental break-up (Rocchi et al., 2002; Nardini et al., 2009) or by slab-derived fluids related to a prolonged period of subduction that occurred prior to rifting (Finn et al., 2005; Panter et al., 2006; Martin et al., 2013; Martin et al., 2014; Aviado et al., 2015).

In this study we offer a unique perspective on this debate through analysis of geochemical, isotopic (Sr, Nd, Pb and O), and age (40Ar/39Ar dates) data for basalts erupted across the transition from oceanic lithosphere to thinned continental lithosphere adjacent to the East Antarctic craton. The Neogene alkaline magmatism of the North Western Ross Sea (NWRS) include deposits on land as well as numerous seamounts on the continental shelf and hundreds more located within the oceanic Adare Basin (Fig. 1B). Oceanic basalt from the NWRS have been shown to be of the same age and petrogenetically akin to basalt from the rest of the WARS (Panter and Castillo, 2007). Apart from the Cameroon volcanic line in West Africa (Fitton and Dunlop, 1985; Njome and de Wit, 2014) we know of no other place on Earth where contemporaneous, small-volume, alkaline volcanism occurs across a rifted continental margin, and in particular, a region that has experienced a relatively recent and rapid transition (~110-80 Ma) from subduction to extension to break-up without voluminous magmatism (i.e. no flood basalts; Finn et al., 2005). Thus the geochronology

and petrology of NWRS basalts allows for a novel assessment of lithospheric (continental and oceanic) and sub-lithospheric sources for alkaline magmatism as well as the potential overprint of these domains by subduction-related processes.

#### **TECTONIC & MAGMATIC HISTORY**

The fabric of the pre-Cenozoic history in Northern Victoria Land (NVL), which includes the continental portion of the NWRS (Fig. 1B), is expressed by three fault-bounded lithotectonic blocks (Wilson, Bowers and Robertson Bay terranes) that were amalgamated via subduction during the Ross orogeny in late Neoproterozoic to late Ordovician times (Borg and Stump, 1987; Boger and Miller, 2004; Di Vincenzo et al., 2016). Devonian and Carboniferous calc-alkaline intrusions in NVL (Admiralty suite, 370-350 Ma; Stump, 1995) indicate younger subduction-related magmatism. All of this was part of the more extensive Terra Australis orogen that is characterized by convergent deformation along the entire 18,000 km pre-dispersal length of the Pan-Pacific margin of the Gondwana supercontinent from the Neoproterozoic to the end of the Paleozoic (Cawood, 2005). The Terra Australis orogen terminated with the assembly of Pangaea and heralded the beginning of the late Carboniferous to Triassic Gondwanide orogeny (Cawood, 2005) which is manifested in NVL by deformed siliciclastic strata (Vaughan and Pankhurst, 2008). Following a period of erosion that produced the Kukri peneplain (Isbell, 1999), early to middle Jurassic rifting and magmatism marked the initial stages of Gondwana break-up and the separation of Africa and South America from the Dronning Maud Land margin of Antarctica. Flood basalts and intrusions of the Ferrar large igneous providence were emplaced in a narrow linear band over a distance of 3500 km within a very short time interval at ~183 Ma (Elliot and Fleming, 2004; Burgess et al., 2015; Ivanov et al., 2016). The origin of Ferrar magmatism remains controversial; some authors relate it to mantle plume activity and long-distance transport of magma within the crust (Storey & Kyle, 1997; Elliot et al., 1999; Vaughan and Storey, 2007) while others call for mantle melting fluxed by slab dehydration from subduction along the length of the Gondwana margin (Hergt *et al.*, 1991; Ivanov *et al.*, 2016).

The subduction of the Phoenix oceanic plate beneath the eastern margin of Gondwana ceased, abruptly, in the Cretaceous (Bradshaw, 1989; Mukasa and Dalziel, 2000; Tulloch et al., 2009) and extension that formed the present-day WARS began. The sudden change from a mostly compressional tectonic regime to a tensional one has been explained by the oblique subduction of the Pacific-Phoenix spreading center (Bradshaw, 1989), possibly with the aid of a mantle plume that weaken the lithosphere and controlled the location of rifting between Zealandia and the Marie Byrd Land margin of Antarctica (Weaver et al., 1994). The shutting down of subduction has also been explained by the mechanism of slab capture (Luyendyk, 1995) and also by a collision with a young and thermally buoyant Hikurangi oceanic plateau (Davey et al., 2008). Whichever the cause, rifting led to the final disintegration of Gondwana and the isolation of Antarctica by 83 Ma (magnetic isochron C34; Veevers, 2012 and references therein). It has been estimated that this early rifting phase caused in 500-1000 km of crustal extension across the entire Ross Sea between Marie Byrd Land and western boundary of the East Antarctica craton (DiVenere et al., 1994; Luyendyk et al., 1996). During the Late Cretaceous and early Cenozoic, north-south oriented basins, including the Northern Basin (Fig. 1B) and Victoria Land Basin, and ridges (e.g., Coulman High) were formed (Cooper et al., 1987).

The early phase of broadly distributed extension in the WARS transformed into a more focused phase of extension in the Paleogene (Huerta and Harry, 2007 and references therein). This phase of rifting in the western Ross Sea lead to ~300 km of extension between East and West Antarctica between 80 and 40 Ma (Molnar *et al.*, 1975; Cande *et al.*, 2000), and was accompanied by rapid exhumation and uplift of the adjacent Transantarctic Mountains (Fitzgerald *et al.*, 1986)

and the oldest known igneous activity associated with the WARS. The Meander Intrusive Group consists of Eocene-Oligocene (48–23 Ma) plutonic and subvolcanic alkaline rocks found in NVL (Müller et al., 1991; Tonarini et al., 1997). Rocchi et al. (2002, 2005) and Nardini et al. (2009) propose a multi-stage model for the origin of these rocks that calls upon upper mantle enrichment by small degree melts during Late Cretaceous rifting (amagmatic phase). This was followed more than 30 million years later with melting of this enriched mantle by a combination of warm asthenosphere flow beneath the rift towards the East Antarctic craton and localized decompression in response to the reactivation of NW-SE trans-lithospheric faults by far-field plate tectonic stresses (Rocchi et al., 2002).

By the middle Cenozoic WARS extension was focused in the Victoria Land Basin, which lies along the central and southern margin of the western Ross Sea, and the Adare Basin located in the north western Ross Sea (NWRS; Fig. 1B). The Adare Basin represents a fossil spreading center, the slowest arm (ultraslow; ~12 mm/yr full-spreading rate) of a three-ridge junction between the Australia, East Antarctic and West Antarctic plates, that was active from 43 to 26 Ma (chrons C20 – C8) and resulted in ~140 to 170 km of ENE-WSW plate separation (Cande *et al.*, 2000; Cande and Stock, 2006; Granot *et al.*, 2013). Post-spreading events were minor and occurred at ~24 Ma and ~17 Ma, resulting in an additional 7 km of extension and the creation of the Adare Trough by normal faulting (Granot *et al.*, 2010, 2013). These faults display an en echelon NW-SE structural trend (Fig. 1B). A strong kinematic relationship between sea floor spreading and WARS is supported by magnetic, seismic and structural trends that show continuity across the shelf break into the Northern Basin (Davey *et al.*, 2006; Cande and Stock, 2006; Damaske *et al.*, 2007; Ferraccioli *et al.*, 2009; Selvans *et al.*, 2012). This indicates a continuity in crustal type that is most likely oceanic (Selvans *et al.*, 2012, 2014; Granot *et al.*, 2013). Granot et al. (2013) suggest that

the shallow bathymetry of the Northern Basin (Fig 1B) may be accommodated by thick deposits of glacio-marine sediments on top of oceanic crust that have filled the basin.

Neogene alkaline volcanism along the northern coastline of Victoria Land and off-shore islands comprise the Hallett volcanic province which consists of ~N-S elongated shield volcanoes that range in age from ~14 Ma to ~2 Ma (Kyle, 1990; Müller et al., 1991; Armienti et al., 2003; Nardini et al., 2003; Mortimer et al., 2007; Smellie et al., 2011). Volcanism on the adjacent continental shelf and Adare Basin was first identified through ocean bottom bathymetry (Panter and Castillo, 2007) and consist of hundreds of relatively small volcanic centers distributed over an area of ~35,000 km<sup>2</sup>. The trend of volcanism overlaps with the ~N-S pattern of normal faulting, particularly the major bounding faults of the Adare Trough, and is generally aligned with volcanism on land in the Hallett volcanic province (Fig. 1B). The age of seamount volcanism was first constrained by seismic stratigraphic relationships. Granot et al. (2010) noted that all of the volcanic features are exposed above the seafloor surface, penetrating nearly the entire stratigraphic sequence (to the late Pliocene) and that buried intrusions, which might indicate an older period of magmatism, are not observed along the ~3,200 km length of the multi-channel seismic data collected in the Adare Basin. The spatial distribution, relative age and preliminary petrological data all indicate that the volcanic activity in the oceanic Adare Basin is an extension of WARS volcanism. Based on this it was proposed that both the continental and oceanic portions of the WARS in the NWRS share a common origin (Panter and Castillo, 2007, 2008; Panter et al., 2016).

#### SAMPLE DETAILS AND METHODOLOGY

This study is based on rock samples collected from seafloor dredging, rock and powder samples acquired from personal collections and rock samples provided by the U.S. Polar Rock Repository, Byrd Polar and Climate Research Center (http://research.bpcrc.osu.edu/rr/). Dredge samples were

collected during a 38 day, National Science Foundation funded, marine geophysical survey cruise (NBP0701) from December 2006 to January 2007 onboard the icebreaker RV/IB *Nathaniel B. Palmer*. The dredging was accomplished at a relatively high level of precision by using the *Palmer's* automatic dynamic global positioning system, depth sounding using a pinger attached to the wire at  $\sim$ 300 m above the dredge, and detailed seafloor bathymetry.

The primary objective of this study is to evaluate mantle sources and processes responsible for alkaline volcanism across the continental-oceanic transition in the NWRS. To accomplish this we selected 30 relatively unfractionated basalts (MgO > 6 wt.%). Petrographic data along with mineral chemistry, major and trace elements, Sr–Nd–Pb–O isotopes, and <sup>40</sup>Ar/<sup>39</sup>Ar geochronology were obtained on subsets of these samples (Table 1). The location of the samples is provided in Figure 1B and their whole rock geochemistry and isotopic compositions are presented in Tables 2 and 3. Twelve of the thirty samples are from other studies (see Table 2 for details and references) and of those twelve, four (MA-009a, MA-117, P74794, P74833) were analyzed in this study for mineral chemistry and oxygen isotopes on olivine separates (Table 3) and two of those (MA-009a, MA-117) were also analyzed for Sr-Nd-Pb isotopes on whole rock powders (Table 2). An additional sample, SAX20 (Nardini *et al.*, 2009), is not from the NWRS but from a location to the south within the Melbourne volcanic province (Kyle, 1990) and is included for comparison.

### **Mineral chemistry**

Rocks were crushed and sieved in order to separate phenocrysts between 250 µm and 2 mm in diameter. Six to eight phenocrysts (olivine, clinopyroxene, amphibole and plagioclase) from each sample were mounted and singly polished on 1 cm diameter epoxy rounds. These were then secured into 1-inch brass mounts (three rounds per mount) and carbon coated for analysis by electron microprobe. Major and minor element chemistry was measured on a Cameca SX-100 Electron

Microprobe Analyzer at the University of Michigan EMAL laboratory using a 15kV accelerating voltage beam, 15nA beam current, and 10–12 μm spot size. A total of 307 unknowns were analyzed; 174 olivine, 76 clinopyroxene, 23 amphibole, and 38 plagioclase grains. The following elemental routines were performed for each mineral: olivine = Si, Cr, Fe, Mn, Mg, Ca, P, Ni, clinopyroxene =Si, Al, Ti, Cr, Fe, Mn, Mg, Ca, Na, K, P, Ni, Cl, plagioclase = Mg, Al, K, Ca, Ti, Cr, Mn, Fe, Na, P, Si, Sr, Ba, and amphibole =Mg, Al, K, Ca, Ti, Cr, Mn, Fe, Na, P, Si, Cl, F. Measurements (a total of 475 unknowns) were taken primarily within the cores of minerals (or presumed cores on fragments), and also at rim and intermediate grain locations when compositional or textural variations were observed (Krans, 2013).

## Major and trace elements

Concentrations of major oxides and some of the trace elements of whole rock powders were determined by X-ray fluorescence spectrometry (XRF) at the GeoAnalytical Laboratory of Washington State University (Castillo *et al.*, 2010) and at New Mexico Tech (Hallett and Kyle, 1993). For some samples concentrations of rare earth elements (REE) and other trace elements (Rb, Sr, Y, Ba, Pb, Th, Zr, Nb, and Hf) on whole rocks were determined using a Finnigan Element 2 high resolution inductively-coupled plasma mass spectrometry (ICP-MS) at the Scripps Institution of Oceanography (SIO) Analytical Facility following the method of Janney and Castillo (1996) with some modifications. Prior to ICP analysis, rock samples were crushed in an alumina ceramic jaw crusher. The resultant rock chips were ultrasonically washed in deionized water for 30 minutes, dried in an oven overnight at ~110°C, and, then, fresh-looking pieces were hand-picked under a binocular microscope. The selected chips were powdered in an alumina ceramic grinder. For each sample, about 25 mg of powder was digested with an ultrapure 2:1 concentrated HF:-HNO3 solution in a Teflon beaker, and the mixture was placed on a hot plate (~60°C) and dried

under a heat lamp. About 2 ml of ultrapure 12-N HNO<sub>3</sub> was twice added to the digested sample and evaporated to dryness. After drying, the digested sample was diluted 4000-fold with 2% HNO<sub>3</sub> solution containing 1 ppb In as an internal standard. The instrument drift was monitored and corrected by measuring an in-house rock standard as unknowns within the run. For selected samples some trace elements for whole rocks were analyzed by instrumental neutron activation at New Mexico Tech (Hallett and Kyle (1993).

# Radiogenic isotopes (Sr, Nd, Pb)

Radiogenic isotopes were analyzed at the Scripps Institution of Oceanography (SIO) and the Woods Hole Oceanographic Institution (WHOI). At SIO whole rock powders were dissolved following the dissolution procedure described above for the ICP-MS analyses. Lead was first separated from the dissolved samples using small ion exchange columns in an HBr medium. Residues from Pb extraction were collected and then passed through ion-exchange columns using HCl as eluent to collect Sr and REE. Finally, Nd was separated by passing the REE cuts through small ion exchange columns using alpha hydroxyisobutyric acid as eluent. The Pb, Sr and Nd isotopes were measured using a 9-collector, Micromass Sector 54 thermal ionization mass spectrometer (TIMS) at SIO. Lead isotopes were fractionation corrected using the isotope values of NBS 981 relative to those of Todt et al. (1996). Strontium isotopes were fractionation corrected to  $^{86}$ Sr/ $^{88}$ Sr = 0.1194 and are reported relative to NBS 987  $^{87}$ Sr/ $^{86}$ Sr = 0.70258. Neodymium isotopes were measured in oxide forms, fractionation corrected to <sup>146</sup>NdO/<sup>144</sup>NdO = 0.72225  $(^{146}\text{Nd}/^{144}\text{Nd} = 0.7219)$  and are reported relative to  $^{143}\text{Nd}/^{144}\text{Nd} = 0.511850$  for the La Jolla Nd standard. Analytical uncertainties based on repeated measurements of standards are  $\pm$  0.010 for  $^{206}\text{Pb}/^{204}\text{Pb}$  and  $^{207}\text{Pb}/^{204}\text{Pb}$ ,  $\pm 0.024$  for  $^{208}\text{Pb}/^{204}\text{Pb}$ ,  $\pm 0.000018$  for  $^{87}\text{Sr}/^{86}\text{Sr}$ , and  $\pm 0.000014$  for <sup>143</sup>Nd/<sup>144</sup>Nd. Routine analytical blanks are generally <35 pg for Sr, <10 pg for Nd and <60 pg for

Pb. At WHOI Sr and Nd chemistry from whole rock powders was done with Sr-Spec and Ln-Spec resin respectively. Lead was separated following the HBr-HNO3 procedure of Abouchami *et al.* (1999) using a single column pass. Sr, Nd and Pb analyses were done on the NEPTUNE multicollector ICP-MS. For Sr and Nd, the internal precision is 10-20 ppm ( $2\sigma$ ); external precision, after adjusting to 0.710240 and 0.511847 for the SRM987 and La Jolla Nd standards respectively, is estimated to be 15-25 ppm ( $2\sigma$ ). Pb analyses carry internal precisions on XXX/204 ratios of 15-50 ppm; SRM997 Tl was used as an internal standard, and external reproducibility (including full chemistry) ranges from 20 ppm ( $2\sigma$ ) for 207Pb/206Pb, to 120 ppm ( $2\sigma$ ) for 208Pb/204Pb.

# **Oxygen Isotopes**

Oxygen isotopes on olivine separates were analyzed by *traditional* laser fluorination/gas source mass spectrometry (LF) and also *in situ* by secondary ionization mass spectrometry (SIMS) at the University of Wisconsin-Madison (WiscSIMS) Laboratory (Table 3). In both cases, the composition of olivine was first determined by electron microprobe analysis, specifically forsterite content, which was needed to produce a calibration curve to correct for SIMS bias in measured oxygen values relative to SCOL (see Valley and Kita, 2009). Olivine chosen for analysis by SIMS contain varying percentages of melt and oxide inclusions, which unlike LF, are easily avoided by the SIMS (~10 µm diameter spot size). Furthermore, the small number of olivine grains separated due to the limited amount of material for some of the samples required the use of single crystal SIMS analysis.

For LF, olivine was analyzed in 1.6 to 2.6 mg aliquots (~ 4–5 grains) using a CO<sub>2</sub> laser probe system (BrF<sub>5</sub> reagent) attached to a dual-inlet five-collector Finnigan/MAT 251 mass-spectrometer. In cases where a given sample yielded two size populations of phenocrysts, aliquots were grouped comparatively by size. Measurements for unknowns were corrected by the average

difference between measured and accepted values for UWG-2 Gore Mountain garnet standard  $(\delta^{18}\text{Ovsmow} = 5.80\%, \text{Valley } et \, al., 1995)$  measured on the same day. Daily standard deviation for UWG-2 was  $\pm$  0.08 to 0.10% (2SD, n = 8). Duplicates were attempted for all runs but were only obtained on 5 of the 9 samples analyzed (Table 3) due to sample size and number of highest quality olivine available. Duplicate measurements of unknowns had an average reproducibility of  $\pm$  0.16% for olivine (2SD).

The best olivine grains (i.e. pristine cores with few to no inclusions and no alteration) from each sample were also selected for SIMS. Mineral compositions determined by EMPA are within 15 μm of SIMS analyses but within 5μm of pits. Olivine grains were sorted based on relative size and cast in three 2.5 cm mounts. Unknowns and grains of San Carlos olivine standard were placed within 5 mm from the center of each mount. Mounts were ground and polished to minimize grain topography to less than 1 micron (Kita et al., 2009). Reflected light and back scattered electron images were taken of each grain prior to analysis to assess compositional zoning and select locations on individual grains to be measured in situ by electron microprobe and SIMS. Over 100 olivine phenocrysts from 21 samples (2–3 grains/sample; Table 3) were measured following the procedures of Kita et al. (2009) and Valley and Kita (2009). A CAMECA IMS-1280 large-radius, multi-collection secondary-ion mass-spectrometer was used with a primary <sup>133</sup>Cs<sup>+</sup> beam intensity of 1.5 to 1.8 nA (spot size  $\sim 10 \mu m$ ). The secondary ion beams for  $^{16}O$ ,  $^{16}O^{1}H$ , and  $^{18}O$  were collected with Faraday cup detectors at MRP ~2500 (mass-resolving power). The OH/O ratio is a relative measure of the amount of hydrogen in the volume of olivine analyzed. Ratios of OH/O are background corrected by subtracting the OH/O in bracketing analyses of SCOL, to remove variability due to sample and vacuum conditions and provide a sensitive check against cryptic hydrous alteration or contamination (Wang et al., 2014). Given the wide range of olivine compositions (Fo-70 to Fo-90; Table 3) a calibration curve based on multiple olivine standards

with Mg# from 100 to 1 (HN-OL, SCOL, OR-OL, Fayalite 50278; Table A3) was used to correct for bias in the oxygen isotopic values of the unknowns relative to the San Carlos olivine standard (Valley and Kita, 2009). A bracketed average was calculated from 3–4 analyses of the running standard (San Carlos olivine) before and after analyses of unknowns to yield an average standard precision of  $\pm$  0.28% (2 SD). Eight unknowns representing the most extreme  $\delta^{18}O_{ol}$  values were also selected and an additional 5-6 analysis were taken from cores to test intra-grain reproducibility. The Core\* value presented in Table 3 is the average of multiple spots in the core of a single olivine grain in an individual sample.

# <sup>40</sup>Ar/<sup>39</sup>Ar dating

Selected basalt dredged from seamounts and basalt from the Possession Islands were crushed, sieved, leached with dilute HCl, washed in distilled water and hand-picked to remove phenocrysts and any altered material to produce a groundmass concentrate. - The samples were irradiated in two batches. Samples for batch NM-217 were loaded into a machined Al disc and irradiated for 7 hours in D-3 position, Nuclear Science Center, College Station, Texas. Samples for batch NM-251 were irradiated for 20 hours in the C.T. position, U.S. Geological Survey TRIGA, Denver, Colorado. Fish Canyon Tuff sanidine (FC-2), with an assigned age of 28.02 Ma (Renne et al., 1998), was used as the neutron flux monitor in both batches.

Samples in batch NM-217 were measured using a Mass Analyzer Products 215-50 mass spectrometer on line with automated all-metal extraction system. Groundmass concentrates were step-heated using a Mo double-vacuum resistance furnace. Reactive gases removed during a 12 minute reaction with two SAES GP-50 getters, one operated at ~450°C and the other at 20°C. Gas was also exposed to a W filament operated at ~2000°C and a cold finger operated at -140°C. Samples in batch NM-251 were measured using an Argus VI mass spectrometer on line with

automated all-metal extraction system. Reactive gases were removed during a 2 minute reaction with two SAES GP-50 getters. Getters, W filament and cold finger were operated at the same temperatures as batch NM-217. Electron multiplier sensitivities, blanks and other analytical parameters for both batches are provided in Table 1.

### **RESULTS**

#### Field data – Adare Basin seamounts

The NBP0701 dredging campaign produced a total of ~1000 lbs of rock from seventeen sites within the Ross Sea (Table A1 and Figs. A1–A9). Ice rafted debris (IRD) were recovered from all dredge sites but a total of 48 *in situ* mostly basaltic lava samples were collected from the western flank of the Adare Trough and individual seamounts within the basin and on the continental shelf. The lava samples were initially distinguished shipboard from IRDs by their angularity, absence of manganese coatings and glassy texture. Further examination in thin section reveal a predominance of textures indicative of rapid undercooling (described below) as would be expected for lava erupted on the seafloor. The main strategy was to collect lava from these seamounts with increasing distance from the continent. Individual seamounts are steep sided, conical-shaped edifices that range in height and basal diameter from ~100 m to > 700 m and < 1 km to ~4 km, respectively, with minimum exposed volumes between 0.03 km³ and 3 km³. In some areas, closely spaced seamounts are merged on elevated platforms (Figs. A7 and A8) that were most likely constructed by the coalescence of lava flows. The morphological characteristics of the seamounts are consistent with monogenetic volcanoes.

Four sites were dredged in the central portion of the Adare Trough. Samples from three of the sites (D1, 2 and 3) on the western flank at ~1000 to < 2000 meters below sea level (mbsl) include basaltic boulders and ice rafted debris (IRD) such as dolerite and metamorphic rocks and

all of the samples are coated with a thick layer of manganese oxide. Dredge D3 sampled a seamount exposed in the east-facing upper western scarp that appears to have been cut by the bounding normal fault of the trough (Fig. A2). The other dredge lines on the western flank (D1 and D2) were positioned on the upper scarp facing away from any obvious seamounts in an attempt to sample oceanic crust (Fig. A1). The fourth dredging site (D4) is a volcano duplet located in the middle of the Adare Trough at > 2000 mbsl (Fig. A3) and the samples acquired were mainly angular fragments of vesicular basalt and some pebbles and cobbles of IRD. The manganese oxide coating of these samples is very thin when compared to the samples from the western flank. A cluster of more than 100 seamounts are found in the southern portion of the Adare Basin and on the shelf. Thirteen of these seamounts were dredged. The shallowest ones (D5 to 7) on the continental shelf, ~15 nautical miles away from Cape Adare, are ~400 to 600 mbsl (Figs. A4 and A5) and heavily covered with coralline materials. The farthest seamount (D17), some 65 nautical miles from Cape Adare, is ~2000 to 1400 mbsl (Fig. A9), is free of coralline materials.

The seamounts were erupted through a thick (up to ~2000 m) sequence of sediments that have been accumulating in the basin since its opening at 43 Ma (Cande *et al.*, 2000), however, based on seismic stratigraphy that has been correlated to DSDP sites 273 and 274 (Granot et al., 2010 and references therein) the sediments should be much younger (Early Oligocene to Pliocene). None of the seamounts sampled appear to be volcanically active but the fact that a significant proportion of the lavas are glassy and lack manganese oxide coatings, except for those recovered from the western scarp of the Adare Trough, suggest that the volcanic activity in this region is relatively young. Radial intergrowths (i.e. variolitic texture) of plagioclase and/or pyroxene in the glassy groundmass in many of the lavas indicate rapid undercooling, as would be expected with eruption on the seafloor. In this study, basalt from eight individual seamounts and basalt from the western scarp of the Adare Trough have been dated by the <sup>40</sup>Ar/<sup>39</sup>Ar method to determine the age

of volcanism and to provide further temporal constraints on the Neogene history of the Adare Basin. Furthermore, the age determinations are compared to the age of nearby land deposits in order to resolve the progression of magmatism in the NWRS region of the WARS.

# <sup>40</sup>Ar/<sup>39</sup>Ar chronology

New  $^{40}$ Ar/ $^{39}$ Ar age dating results on four basalt samples dredged from the Adare Trough, five basalts from five different seamounts in the southern Adare Basin, and one basalt sample from a seamount located on the continental shelf, provide the first direct dating of oceanic volcanism in the NWRS. Additionally, five age determinations on basalt collected from the Possession Islands (Possession = 2, Foyn = 2 and McCormick = 1) provide the first age data for this island group. A rhyolite drop stone recovered at D17 was also dated in order to determine the age and provenance of the pyroclastic activity that produced it. The results and analytical details are provided in Table 1 and representative age release and Ca/K spectra are shown in Fig. 2. The remainder of the spectra and isotope correlation ( $^{36}$ Ar/ $^{40}$ Ar vs.  $^{39}$ Ar/ $^{40}$ Ar) diagrams are provided in Fig. A10 and all  $^{40}$ Ar/ $^{39}$ Ar analytical data is provided in Table A2.

The new age data confirm the relatively young history of seamount volcanism in the NWRS. The ages range from  $15.93 \pm 0.76$  Ma to  $0.14 \pm 0.02$  Ma (Table 1) with a median value of 2.88 Ma. All of the seamounts measured are younger than 5 Ma, verifying the Pliocene age of volcanism previously based on seismic stratigraphy (Granot *et al.*, 2010), but further reveal Pleistocene activity ( $\leq 2.6$  Ma) at three seamounts (D4-3, D7-1 and D17-1). The middle Miocene age ( $\sim 16$  Ma) determined on basalt sample D2-1 collected on the western scarp face of the Adare Trough is too young to represent oceanic crust, which in this area is between 33.5 to 30.9 Ma (bounded by magnetic anomalies 120 and 130; Granot et al., 2010). Furthermore, its petrological characteristics are not those of oceanic crust formed at slow to fast spreading centers (*a.k.a.* Mid-

Ocean Ridge Basalt – MORB) but are the same as the seamounts (discussed below). The next oldest age determination at  $4.61 \pm 0.19$  Ma (D3-1) is also from the western flank of the Adare Trough but unlike D2-1 it was collected from a prominent seamount. It is interesting to note that this seamount is cut by the bounding fault, and if the age is representative of the seamount on the whole, then it would suggest fault activity younger than 4.6 Ma. Granot et al. (2010) concludes that uplift of the western flank of the Adare Trough occurred much earlier ( $\sim$ 17 Ma) but that vertical normal faulting may be younger and, in places, still active. The age determined for the western flank seamount is  $\sim$ 1 million years older than the seamount dated in the center of the Trough (D4). The two age determinations from this volcano ( $2.51 \pm 0.09$  Ma and  $3.35 \pm 0.11$  Ma) bracket the range determined for four of the five seamounts dated in the southern Adare Basin (Table 1). The two youngest seamount samples, one from a continental shelf (D7-1,  $0.32 \pm 0.23$  Ma) and the other from the eastern-most seamount dredged within the basin (D17-1,  $0.14 \pm 0.02$  Ma), reveal significantly younger, Pleistocene activity.

The Possession Islands are the closest land deposits to the Adare Basin and are located adjacent to the Adare Peninsula (Fig. 1B). Age determinations for five basalts collected from three islands range from  $1.42 \pm 0.01$  Ma to  $0.053 \pm 0.031$  Ma (Table 1) and have a median value of 0.24 Ma. The new ages represent the youngest volcanic activity yet recorded within the Hallett volcanic province. Smellie et al. (2011) provide  $^{40}$ Ar/ $^{39}$ Ar age data as well as a compilation of ages from previous dating campaigns ( $^{40}$ Ar/ $^{39}$ Ar and K-Ar ages) that range from ~13.8 Ma to ~2.3 Ma. The youngest date,  $2.27 \pm 0.5$  Ma (K-Ar age; McIntosh & Kyle, 1990), is from Cape Roget located at the southern end of the Adare Peninsula and within 15 km of Foyn Island (Possession Islands).

Previous studies indicate that there is no overall migration of activity within the Hallett volcanic province, however, three overlapping volcanic centers on the Daniell Peninsula (Fig. 1B)

show a northward-younging age progression over a period of approximately five million years between  $\sim$ 10 Ma and  $\sim$ 5 Ma (Smellie *et al.*, 2011). New age determinations in this study show that there has been a very broad north-easterly shift in the location and volume of eruptive activity in the NWRS from continental shield volcanism along the coastline in the middle to late Miocene ( $\sim$ 14 to > 5 Ma) to monogenetic island and seamount volcanism on the continental shelf and Adare Basin in the Pliocene to Pleistocene (< 5 Ma to < 100 ka). Evidence for this broad shift in activity is underscored when taking into account the occurrence of Eocene–Oligocene (48 - 23 Ma) plutonic and subvolcanic alkaline rocks (Meander Intrusive Group), which are found within the Melbourne volcanic province (Kyle, 1990) immediately adjacent and to the SW of the Hallett volcanic province, and are considered the earliest expression of WARS magmatism (Rocchi *et al.*, 2002).

## Petrography and mineral chemistry

Twenty of the 30 basalts analyzed for whole rock chemistry and isotopes for the NWRS were examined petrographically and their phenocrysts were analyzed for chemistry by Krans (2013).

Basalts from the NWRS are porphyritic and groundmass ranges from holocrystalline to holohyaline with pilotaxitic to trachytic textures. The glassy groundmasses are most prominent in seamount samples. The samples are variably vesicular and some basalts from seamounts display amygdaloidal texture, having vesicles partially filled with calcite and clay material. Olivine and clinopyroxene are the dominant phenocryst phases with subordinate plagioclase and amphibole. Plagioclase and magnetite are common groundmass phases with lesser amounts of clinopyroxene, olivine and amphibole. In a few samples plagioclase is absent altogether.

Olivine is euhedral to subhedral and often contain inclusions of opaque oxide and melt inclusions that make up to 2% by volume. In seamount samples skeletal crystals of olivine are common and are the result of quench crystallization due to rapid undercooling. Also in seamount samples some olivine grains show minor alteration to iddingsite along cleavage fractures. The forsterite (Fo) content [(Mg/(Mg+Fe<sup>2+</sup>)\*100)] of olivine range from Fo-70 to Fo-90 with an average of Fo-81. Most olivine display weak to strong normal zoning (decrease of 1 to 17 mol.% Fo), with rare reverse (up to 3 mol.% increase in Fo) and oscillatory zoning.

Clinopyroxene is euhedral to subhedral with up to 3 vol. % opaque oxide and melt inclusions. In seamount samples microphenocrysts of clinopyroxene often occur as radial intergrowths (variolitic texture) with plagioclase that, like the occurrence of skeletal olivine, indicate high undercooling. Clinopyroxene phenocrysts are bimodal in their size distribution. The larger phenocrysts ( $\geq 0.5$  mm) often exhibit disequilibrium textures with spongy cores and prominent zoning, while smaller phenocrysts (< 0.5 mm) are euhedral to subhedral and are only occasionally zoned. The clinopyroxenes are classified as diopside and core compositions range from Wo 37–52, En 31–52 and Fs 7–21 with Mg#<sub>cpx</sub> varying from 70 to 90. Diopside phenocrysts generally display weak to strong normal zoning (up to 3 % decrease in Mg#<sub>cpx</sub> from core to rim) and less common, reversely zoned (up to 3 % increase in Mg#<sub>cpx</sub> from core to rim).

Plagioclase phenocrysts are present in only six of the twenty NWRS basalts examined. Euhedral plagioclase phenocrysts have core compositions that range from An-42 to An-69, however, most are subhedral to anhedral with sieve texture and reaction rims. In addition they are reversely zoned with anorthite contents that show an increase from core to rim by up to 22 mol.%. The evidence for disequilibrium in the majority of plagioclase suggests that they are antecrysts (i.e. crystals that formed earlier in the differentiation process and have been reincorporated into the magma).

Amphibole phenocrysts are less common, observed in five of the twenty basalts examined, and typically exhibit mild to extensive reaction rims. The amphibole is classified as kaersutite (Leake et al., 1997) with Mg#<sub>amp</sub> 66 – 72. Kaersutite is often found in alkaline basalts and xenoliths within the NWRS and the adjacent Melbourne Volcanic Province (Kyle, 1986; Rocchi *et al.*, 2002; Coltorti *et al.*, 2004; Perinelli *et al.*, 2006, 2011; Armienti & Perinelli, 2010; Perinelli *et al.*, 2017).

### Major and trace element characteristics

The whole rock composition of the 30 selected NWRS basalts include alkali basalt, basanite, hawaiite and tephrite (Table 2, Fig. 3). All were selected to represent relatively unfractionated compositions (MgO  $\geq$  6 wt.%) and include deposits located on the continent, continental shelf and oceanic Adare Basin (Fig. 1B). The major and trace element composition of NWRS basalts are plotted in Figures 3 to 5 with other basalts (MgO  $\geq$  6 wt.%) from seamounts and islands located in the southern Ross Sea (Terror Rift basin; Aviado et al., 2015) and in the Amundsen Sea off of the Marie Byrd Land coastline (Hubert Miller seamount and Peter I Island: Hart et al., 1995; Kipf et al., 2014) (Fig. 1A). The purpose is to compare basalts derived from melting beneath regions that have minimum crustal thicknesses and from ocean settings that are considered to be related to the WARS. Also shown are a selection of mafic alkaline ocean island basalts (OIB) from the Atlantic and southern Pacific oceans whose geochemical characteristics (and isotopic signatures discussed below) have been repeatedly likened to basalts from the WARS.

The composition of NWRS basalts range from strongly silica-undersaturated (*Ne*-normative > 8 to 21 wt.%) basanite and tephrite to more moderately silica-undersaturated (*Ne*-normative  $\le 11$  wt.%) hawaiite and alkali basalt, two of which are silica-saturated (*Hy*-normative) (Table 2). In addition to their higher silica and lower total alkali ( $K_2O + Na_2O$ ) contents, alkali basalt also have overall lower  $TiO_2$  and  $P_2O_5$  (Fig. 4) and lower concentrations of highly incompatible trace

elements (e.g., La, Ce, Sr and Zr) relative to basanite and tephrite (Fig. 5). The overall compositional spectrum of NWRS basalt matches that of WARS basalt from the southern Ross Sea (Aviado *et al.*, 2015) and Amundsen Sea (Hart et al., 1995; Kipf *et al.*, 2014) as well as OIB, with the notable exception of Yb, which is much lower in concentration for Peter I Island (Fig. 5). There are no distinct compositional trends on plots of major and trace elements against MgO wt.% with the exception Al<sub>2</sub>O<sub>3</sub>, Cr and CaO (Figs. 4 and 5). The variations in concentration of these elements with decreasing MgO content indicates an overall control by clinopyroxene during magma genesis. Nickel contents (not shown) decrease with MgO in samples with Ni < 250 ppm indicating olivine control.

Primitive mantle normalized plots (Fig. 6a-c) of basalts from the NWRS, along with other WARS basalts, display enriched patterns and well-developed K and Pb negative anomalies for nearly all samples. Alkali basalts are again distinguished from basanite (and tephrite) by having overall lower elemental concentrations, less pronounced K and Pb anomalies and small negative P anomalies (Fig. 6b). All of the basalts are enriched in light rare earth elements (LREE) relative to heavy REE, which is typically linked to garnet control during small degrees of mantle partial melting. Alkali basalt have lower  $La_N/Yb_N$  ratios ( $\leq 15$ ) compared to basanite and tephrite ( $La_N/Yb_N \geq 12$  to 27) in NWRS samples, contributing to their slightly flatter patterns. Chondrite normalized REE plots of NWRS and other WARS basalts lack Eu anomalies (Eu/Eu\* between ~0.9 and 1.1) indicating that the magmas were not influenced by the fractionation or accumulation of plagioclase. Once again, all of the WARS basalts fall within the compositional range of OIB and display very similar trace element distribution patterns.

## **Sr-Nd-Pb** isotopes

Radiogenic isotope ratios for NWRS basalts are provided in Table 2 and plotted with other WARS basalts and alkaline OIB in Figure 7. Apart from one sample from the Malta Plateau (MA009a), NWRS basalts encompass a moderate range in measured <sup>87</sup>Sr/<sup>86</sup>Sr (0.70278–0.70379) ratios and narrow range in measured <sup>143</sup>Nd/<sup>144</sup>Nd (0.51282-0.51300) ratios. Isotopes of Pb measured for NWRS basalts also show a restricted range in <sup>207</sup>Pb/<sup>204</sup>Pb (15.52–15.68) and a moderately wide range in <sup>208</sup>Pb/<sup>204</sup>Pb (38.81–39.92) and <sup>206</sup>Pb/<sup>204</sup>Pb (19.22–20.23) ratios. Values for other WARS basalts fall within the same range, except for Peter I Island, having much higher <sup>207</sup>Pb/<sup>204</sup>Pb (>15.7) that is explained by melting of a localized mantle upwelling of an EMII-like source (Hart et al., 1995; Kipf et al., 2014). Seven NWRS basalts have high <sup>206</sup>Pb/<sup>204</sup>Pb ratios (> 20), six of which also have low  ${}^{87}\text{Sr}/{}^{86}\text{Sr}$  (< 0.7030) and high  ${}^{143}\text{Nd}/{}^{144}\text{Nd}$  (> 0.5129) values (Table 2). The isotopic signatures coupled with their OIB compositional characteristics fit what has been referred to as HIMU-like [i.e. derived from melting of a mantle source with high- $\mu = (^{238}\text{U}/^{204}\text{Pb})_{t=0}]$  to characterize mafic alkaline magmatism within the WARS (Rocholl et al., 1995; Hart et al., 1997; Panter et al., 1997, 2000, 2006; Rocchi et al., 2002; Nardini et al., 2009; Kipf et al., 2014; Martin et al., 2013; Aviado et al., 2015). Nevertheless, most of the basalts in this study fall within the range of OIB compositions assigned to a PREMA (PREvalent MAntle; Zindler and Hart, 1986) mantle source; specifically falling within the DM (Depleted Mantle)–PREMA array as defined by Stracke (2012, and references therein). The development of the DM-PREMA array is attributed to the continuous subduction and aging of MORB-type crust during recycling through the mantle (Stracke et al., 2005). Alternatively, PREMA, which is very similar, if not identical, to the socalled focus zone (FOZO - Hart et al., 1992) or the common component ("C" – Hanan and Graham, 1996) in the mantle sources of oceanic basalts is the lithospheric mantle portion of the subducting slab (Castillo, 2015, 2016). Accordingly, the DM-PREMA array may simply be the long-time (~b.y.) integrated record of geochemical depletion of the upper mantle with time, from Bulk Silicate Earth (BSE) to modern DMM.

# Oxygen isotopes

Oxygen isotopic compositions determined by *conventional* laser fluorination (LF) and by secondary ionization mass spectrometry (SIMS) for NWRS basalts are listed in Table 3 and plotted against olivine Fo content and whole rock MgO in Figure 8. Oxygen isotopes measured by LF for nine NWRS basalts range from 4.86 to 5.26‰. Overall, the average LF values for NWRS olivine in this study (5.03 ± 0.16‰, 2SD) are statistically indistinguishable from average olivine LF values ( $\delta^{18}O = 5.29 \pm 0.13\%$ , 2SD) reported for other Northern Victoria Land (NVL) basalts (Nardini *et al.*, 2009) as well as olivine LF values ( $\delta^{18}O = 5.29 \pm 0.18\%$ , 2SD) from xenoliths hosted in NVL basalts, which are considered to represent cumulates from near-primary mantle melts (Perinelli *et al.*, 2011).

Olivine measured by SIMS, including core and rim analyses, and multiple analyses on single grain cores (Core\*), vary from 4.4 to 5.9% (range = 1.5%) and average 5.21  $\pm$  0.26% (2SD,  $n_s$  = 21,  $n_a$  = 176). Intra-crystal variations in  $\delta^{18}O_{ol}$  are as much as 0.6% lighter to 0.5% heavier at rims than cores (average  $\Delta_{core-rim}$  = 0.02%) and are not statistically significant with respect to instrument precision based on the SCOL standard ( $\pm$  0.28% at 2SD). The lowest single spot  $\delta^{18}O_{ol}$  values are from samples NV-4 (4.5%) and NV-4C (4.4%) with average values being slightly higher at 4.82  $\pm$  0.18% ( $n_a$  = 7) and 4.66  $\pm$  0.25% ( $n_a$  = 13), respectively. The highest single spot  $\delta^{18}O_{ol}$  values are from samples MA-009a and MA-117 (5.9%) with averages at 5.54  $\pm$  0.22% ( $n_a$  = 14) and 5.57  $\pm$  0.26% ( $n_a$  = 13), respectively. The difference between single spot analysis of high and low  $\delta^{18}O_{ol}$  values are resolvable beyond analytical precision.

SIMS δ<sup>18</sup>O<sub>ol</sub> average values are not statistically different from LF average values for individual samples (Table 3) or for the data set as a whole (Fig. 8). Multiple core analyses on single grains measured by SIMS (Core\*) are either indistinguishable (e.g., NV-4) or heavier (e.g., D4-3 & MA-117) relative to LF (Table 3). It is interesting to note that out of the seven samples measured using the Core\* technique, five show heavier values (A225A, D4-3, MA-009a, MA-117 and P74833) than *standard* SIMS measurements. Recall that the *standard* measurement is an average of multiple spots on 2 to 3 grains (mostly cores but includes some rims) and that Core\* is the average of multiple spots in the core of one grain only. It is possible that there is some sampling bias contributing to the difference between the two techniques but, if so, the differences fall within analytical precision.

All  $\delta^{18}O_{ol}$  analyses by SIMS and LF (this study and Nardini et al., 2009) are plotted against Fo % and presented in Figure 8A. Also plotted is a range of mantle olivine values from peridotite (LF - Mattey et al., 1994) and olivine data from basalts at several oceanic locations that include the Canary Islands (LF & SIMS - Gurenko *et al.*, 2006, 2011), Reykjanes Peninsula, Iceland (LF - Peate *et al.*, 2009) and the Azores Islands (LF - Grenske et al., 2013), all of which have also been interpreted to represent mantle values. The LF data for NWRS olivine has nearly the same range in values measured for the Canary and Azores islands. Individual SIMS analyses for NWRS olivine have a greater range encompassing these islands and most  $\delta^{18}O_{ol}$  values for OIB measured by LF (Eiler, 2001). However, the average value of all  $\delta^{18}O_{ol}$  analyses in this study (5.20 ± 0.25‰) is similar to average mantle  $\delta^{18}O_{ol}$  (5.18 ± 0.28‰; Mattey et al., 1994). There is no overall correlation between Fo content and  $\delta^{18}O_{ol}$  as well as no correlation with respect to whole rock composition (i.e. basanite, alkali basalt, etc.). There is a very weak positive correlation, however, when considering only LF analyses (this study and Nardini et al., 2009). In Figure 8B, whole rock MgO (wt.%) is plotted against average  $\delta^{18}O_{ol}$  LF and SIMS data for each sample. Again there is a very

weak positive correlation (i.e. lighter  $\delta^{18}O_{ol}$  with decreasing MgO content) mainly exhibited by LF analyses. But when considering both SIMS and LF data for basanite and alkali basalt there is no correlation, in fact, nearly the full range of  $\delta^{18}O_{ol}$  values for NWRS basalt fall between 9 and 10 wt.% MgO (Fig. 8B).

 $\delta^{18}O_{ol}$  analyses by SIMS and LF averaged for each sample are plotted against measured Sr, Nd and Pb isotopic ratios and presented in Figure 9. There is a broad correlation between oxygen and radiogenic isotopes where lower  $\delta^{18}O_{ol}$  correspond with lower  $^{87}Sr/^{86}Sr$  and higher  $^{143}Nd/^{144}Nd$  and  $^{206}Pb/^{204}Pb$  ratios (as well as  $^{207}$  &  $^{208}Pb/^{204}Pb$ , not shown). Significantly, a similar relationship had been generally observed in OIB samples (Eiler et al., 1996).

#### **DISCUSSION**

The main objective of this study is to better understand mantle sources and processes responsible for the production of alkaline magmas within the WARS. The unique contribution of this research relative to past efforts is the location of our basaltic suite, which encompass both oceanic and continental domains (Fig 1). Given their similar composition and roughly equivalent age, the oceanic and continental basalts must share a common origin and therefore provide an effective assessment of the role of continental lithosphere in magma genesis. In addition, the NWRS was located adjacent to the former pan-Pacific margin of Gondwana where subduction dominated the tectonic regime from the Neoproterozoic to the Late Cretaceous (~ 450 Myr); thus, the influence of subduction-related processes on source domains is also examined.

The 30 basalts from the NWRS (Table 2 & Fig. 1B) are plotted against decimal degree longitude in Figure 10 and display systematic variation in major and trace elements as well as isotopic ratios across the transition from land to sea. Ocean-ward, the basalts show an increase in total alkalis, silica-undersaturation (Ne + Lc normative, not shown),  $P_2O_5$  and incompatible trace

elements Sr, Zr, Nb, including LREE (not shown), and LREE/HREE ratios. Also towards the ocean the <sup>87</sup>Sr/<sup>86</sup>Sr ratios decrease and <sup>143</sup>Nd/<sup>144</sup>Nd and <sup>206</sup>Pb/<sup>204</sup>Pb ratios increase. Trends in <sup>207</sup>Pb/<sup>204</sup>Pb and <sup>208</sup>Pb/<sup>204</sup>Pb isotopes (not shown) although poorly defined also generally increase ocean-ward. Variation in oxygen isotopes are less well defined but when samples classified as alkali basalt are considered (Figs. 1B and 3) there is a general decrease (values become lighter) towards the ocean (Fig. 10). The systematic compositional variation requires the evaluation of gradational changes in mantle source, melting processes and/or contamination of melt as it rises through the lithosphere.

# **Contamination by crust?**

The increase in silica-saturation, correlation of higher <sup>87</sup>Sr/<sup>86</sup>Sr and lower <sup>143</sup>Nd/<sup>144</sup>Nd ratios with higher  $\delta^{18}O_{ol}$  values in NWRS basalts erupted on the Antarctic continent suggest that the magmas have assimilated crust. Contrary to this presumption is the fact that our 'continental' samples are also distinguished by having lower incompatible element concentrations relative to their oceanic counterparts (Fig. 10). Progressive assimilation of silicic crust coupled with fractional crystallization (AFC, DePaolo, 1981) should produce the opposite relationship. In Figure 11 we have modeled the effects of the AFC processes using a variety of crustal rock types found within NVL. The plots clearly show that the assimilation of these rocks by basaltic magmas will lead to lower Nd and higher Sr isotopic signatures but will also produce higher concentrations of incompatible elements like Rb (as well as Nb, Zr, Ba, LREE) even if they are significantly lower in the assimilant (e.g., gabbro). Furthermore, it is very difficult to account for the relatively constant concentrations (i.e. no systematic increase or decrease) of major elements (MgO, Al<sub>2</sub>O<sub>3</sub>, CaO) and weakly incompatible to compatible trace elements (Y, Yb, Cr) by AFC if advancing from ocean to continent. Simple mixing of magma with mafic rocks like gabbro or eclogite, however, can predict many of the geochemical and Sr-Nd isotopic variations but require a mixture that consists of  $\geq$  40% of those crust types. Achieving this proportion, especially at shallow depths with relatively small volumes of magma, is unlikely.

Oxygen isotopes have long been used to assess contamination of magma by crust and are a particularly strong tool for detecting the assimilation <sup>18</sup>O-rich continental rocks by mantle derived magma (Harmon & Hoefs, 1995; Eiler, 2001, and references therein). In cases where anomalously light oxygen signatures for olivine (i.e. low  $\delta^{18}O_{ol}$ ) occur in OIB, they have been attributed to the assimilation of hydrothermally altered oceanic lower crust (Hart et al., 1999; Skovgaard et al., 2001; Genske et al., 2013) or cannibalization of hydrothermally altered lavas within the volcanic edifice itself (Wang and Eiler, 2008; Garcia et al., 2008). A rough positive correlation between  $\delta^{18}O_{ol}$  and Fo %, first reported by Nardini et al. (2009) and weakly supported by additional LF analyses in this study (Fig. 8A), lead the authors to conclude that the trend towards lighter  $\delta^{18}$ O<sub>ol</sub> values found in NVL basalts is the result of cannibalization of altered volcanic rocks. To more fully assess the possible effects of assimilation we have selected, in addition to NVL crustal rocks, an altered mugearite lava from Mount Sidley, Marie Byrd Land, which has a very low  $\delta^{18}$ O whole rock value of 1.8%; a result of alteration by highly <sup>18</sup>O-depleted Antarctic meteoric water (Panter et al., 1997). The interaction between basalts and potential contaminants is evaluated on a plot of  $\delta^{18}O_{ol}$  versus whole rock Sr concentration shown in Figure 12. The trend in  $\delta^{18}O_{ol}$  of alkali basalt with longitude (Fig. 10i) and the strong correlation with Sr concentration (Fig. 12) may be explained by 1) AFC involving 'normal' to slightly <sup>18</sup>O-enriched basaltic magma interacting with low  $\delta^{18}$ O volcanic rocks and the contamination increasing ocean-ward, 2) mixing between low  $\delta^{18}$ O (< 5‰) basaltic magma and mafic crust (e.g., gabbro,  $\delta^{18}$ O ~ 7‰) with the proportion of crust increasing continent-ward or 3) reaction between low  $\delta^{18}$ O melt and peridotite (e.g., Opx + Liq<sub>0</sub>  $\leftrightarrow$ Ol + Liq<sub>1</sub>; Yaxley & Green, 1998; Pilet et al., 2008; Lambart et al., 2012) that is enhanced continent-ward as melt migrates though thicker mantle lithosphere. The first scenario

(cannibalization) is difficult to defend given the trend of alkali basalt towards lower  ${}^{87}\text{Sr}/{}^{86}\text{Sr}$  and higher  ${}^{143}\text{Nd}/{}^{144}\text{Nd}$  ratios ocean-ward (Fig. 10g, h). The Sr and Nd isotopic ratios should remain relatively constant if the contaminant was altered by Antarctic meteoric water. If the low  $\delta^{18}\text{O}$  signature of the contaminant was a result of alteration by seawater at high temperatures ( $\geq 250^{\circ}\text{C}$ ; Muehlenbachs, 1986; Gao et al., 2012) it would also possess an elevated Sr isotope signature and therefore assimilation would lead to a magma that has higher  ${}^{87}\text{Sr}/{}^{86}\text{Sr}$  along with a low  $\delta^{18}\text{O}$  value, which is not the case here. Simple mixing provides a better fit for the geochemical and isotopic trends exhibited by alkali basalt (Figs. 11 and 12) but again requires a significant proportion of mafic crust (i.e. gabbro/eclogite) and a source for low  $\delta^{18}\text{O}$  magma. There is a general positive correlation between  $\delta^{18}\text{O}_{\text{ol}}$  and Sr concentration for basanite, hawaiite and tephrite lavas which can be mostly enveloped by AFC model curves constructed based on the interaction between a low– ${}^{18}\text{O}$  magma (e.g., NV-4C) and gabbroic crust but only if the proportion of assimilation to crystal fractionation is allowed to vary significantly ( $r = \sim 1$  to 0.3; Fig. 12). In addition, the lower Sr concentrations in alkali basalt relative to basanite and tephrite cannot be explained by these models.

In summary, processes involving shallow-level contamination by crust do not adequately account for the variation of major and trace elements and isotopes shown by NWRS basalts. Furthermore, none of the scenarios discussed above are able to emulate the well-defined trends in major and trace elements and isotopes between land and sea (Fig. 10). Consequently the third scenario, which discounts the involvement of crust altogether, will be evaluated along with melting processes and mantle source(s) in the following section.

#### Mantle sources

The wide range in  $\delta^{18}$ O<sub>ol</sub> exhibited by NWRS basalt (variation of ~ 1.5%; Fig. 8A) is comparable to the range found in OIB, which has been explained by mantle source heterogeneity; a result of

variably altered, recycled oceanic crust (± sediments) within the convecting mantle (Eiler et al., 1996; Eiler, 2001; Day et al., 2010; Gurenko et al., 2011). A recent comprehensive review of petrological and geochemical constraints on the origin of mafic alkaline rocks (primarily OIB) is provided by Pilet (2015) and details how the direct melting of pyroxenite (recycled upper or lower oceanic crust = MORB or gabbro, respectively), with or without added sediments, cannot reproduce the major element compositions of low-silica alkaline basalt, which include basanite, tephrite and nephelinite (Fig. 3). The melting experiments for silica-deficient pyroxenite in the presence of CO<sub>2</sub> (Dasgupta, 2006; Gerbode and Dasgupta, 2010), however, can produce liquids that are closer in SiO<sub>2</sub> and total alkali content but have Al<sub>2</sub>O<sub>3</sub>/TiO<sub>2</sub> and Na<sub>2</sub>O/K<sub>2</sub>O ratios that vary significantly which is in contrast to the relatively low and constant ratios observed in natural low-silica alkaline basalt (Pilet, 2015), including basalts from the NWRS (5.05  $\pm$  1.06 and 2.75  $\pm$  0.47, respectively). Therefore in order to account for the trace element and radiogenic isotope signatures in mafic alkaline rocks, which are ultimately attributed to recycled materials, a pyroxenitic melt must be further conditioned and homogenized before it reaches the surface. Pilet (2015) describe three possible scenarios for this: 1) pyroxenite melt infiltrates and enriches (metasomatizes) peridotite in the asthenosphere, which is then melted at low degrees ( $\sim 1-5\%$ ) in the presence of CO<sub>2</sub> (Dasgupta & Hirschman, 2007); 2) pyroxenite melt reacts with peridotite (dissolution/precipitation reaction = olivine + liquid ↔ orthopyroxene + liquid; precipitation of either orthopyroxene or olivine may also include clinopyroxene and garnet) under dry conditions (Lambart et al., 2012; Mallik and Dasgupta, 2014) or in the presence of CO<sub>2</sub> (Mallik and Dasgupta, 2014) prior to reaching the surface and 3) a multi-stage process that begins with low degree melt from the asthenosphere that penetrates and forms amphibole-bearing cumulates within the lithospheric mantle. Later melting of these relatively small-volume, trace-element and volatile-rich cumulates (metasomatic veins) at high degrees, and reaction of this liquid (as in 2) with the surrounding mantle on its way to the

surface, can reproduce the major and trace element characteristics of mafic alkaline magma (Pilet et al., 2008).

The third source type, metasomatized lithosphere, has also been proposed for alkaline basalts erupted in continental settings (Stein et al., 1997; Jung et al., 2005; Panter et al., 2006; Ma et al., 2011; Mayer et al., 2014; Rooney et al., 2014, 2017) including those within the WARS (Rocchi et al., 2002; Panter et al., 2003; Nardini et al., 2009; Perinelli et al., 2011; Martin et al., 2013; Aviado et al., 2015). In the continental NVL, Perinelli et al. (2011) consider the variability in olivine oxygen isotope data collected from mantle cumulate xenoliths (5.00–5.72‰), together with olivine from alkaline basalts (4.92–5.53‰, Nardini et al., 2009) and peridotites (4.80–5.81‰, Perinelli et al., 2006) to signify extensive  $^{18}$ O compositional heterogeneity in the lithospheric mantle. Our new  $\delta^{18}$ Ool values collected from continental and oceanic NWRS basalts extend that range (4.44–5.92‰; Fig. 8A). Perinelli et al. (2011) conclude that mantle cumulates were modally and cryptically metasomatized by low  $\delta^{18}$ O melts, which in turn, originated by the melting of earlier formed metasomes emplaced within the continental lithosphere during rifting in the Late Cretaceous. The ultimate origin of the low  $\delta^{18}$ O mantle signature found in NVL and the NWRS will be discussed below.

The major element composition of continental and oceanic basalts from West Antarctica presented in this study are consistent with being derived from mantle sources that are predominately silica-undersaturated and contain amphibole-rich lithologies (Fig. 13) and align with the metasomatized lithospheric hypothesis. The presence of amphibole in the source is also supported by negative K anomalies and high Nb—Ta contents displayed on multi-element, primitive mantle-normalized plots (Fig. 6). Potassium occurs in stoichiometric proportions in mantle minerals amphibole (pargasite, kaersutite) and mica (phlogopite). If these minerals are not completely consumed during mantle partial melting, then K will be retained in the source and the melt will be

depleted in K relative to neighboring trace elements (e.g., Nb, Ta, La and Ce) which will behave incompatibly (LaTourrette et al., 1995; Ionov et al., 2002; Tiepolo et al., 2007 and references therein). High Nb and Ta concentrations ( $\geq 100 \times \text{primitive mantle}$ ; Sun and McDonough, 1989) have been measured on vein amphibole in peridotite (Ionov and Hofmann, 1995; Ionov et al., 2002) and for both vein and disseminated amphibole found in peridotite from NVL (Coltorti et al., 2004; Perinelli et al., 2006). Niobium enrichment of the lithospheric mantle has also be related to metasomatism by carbonatite, which can have concentrations > 200-1000 × primitive mantle (Pfänder et al., 2012 and references therein). Martin et al. (2013) ascribe variations in minor and trace element ratios (e.g., elevated Nb/Ta as established by Pfänder et al., 2012) to signify carbonate metasomatism in the lithospheric mantle sources for alkaline basaltic rocks from Mount Morning in southern Victoria Land (SVL, Fig. 1A). This is also consistent with the Nb/Ta ratio (39) measured on mantle phlogopite from Mount Morning, which is close to the global average for carbonatite (35; Chakhmouradian, 2006), as well as having a Nb concentration (547 ppm) that is mostly higher than phlogopite (and amphibole) from mantle xenoliths that have been attributed to mantle lithosphere enriched by carbonate-rich silicate liquids (Ionov and Hofmann, 1995; Ionov et al., 2002). Experimental results indicate that carbonate phases can coexist with amphibole and phlogopite in spinel and garnet peridotite at high pressures (Brey et al., 1983; Olafsson & Eggler, 1983). Sources with elevated Nb and Ta concentrations may also reside in rutile-bearing peridotite (Kalfoun et al., 2002) and eclogite (Rudnick et al., 2000; John et al., 2004). An eclogitic contribution to the metasomatic enrichment of the sub-continental lithosphere beneath NVL has been proposed by Melchiorre et al. (2011) based on radiogenic values of Os and Hf isotopes measured on sulfide and clinopyroxene (respectively) from amphibole-bearing and dry peridotite xenoliths.

Here we also emphasize that the radiogenic isotope signatures in mafic alkaline rocks are ultimately sourced from materials recycled by subduction. The trace element and Sr. Nd and Pb isotopic signatures of the alkaline rocks fall within the compositional range of OIB. The Sr, Nd and Pb isotopic composition of OIB, however, comes from recycled subducted materials (Hofmann, 2014; White, 2015 and references therein). Specifically, the Pb isotopes of some of the alkaline rocks are highly radiogenic, similar to HIMU OIB, whereas those of the bulk of the rocks are less radiogenic, akin to PREMA (or FOZO) –type OIB, which are also sometimes called young-HIMU (Thirlwall, 1997; Stracke et al., 2005). Although it is widely accepted that the highly radiogenic Pb signature of HIMU is due to recycling of variably altered basaltic crust (e.g., Hofmann, 2014; White, 2015 and references therein), a recycled marine carbonate source for the radiogenic Pb isotopes of HIMU has recently been proposed (Castillo, 2015, 2016). This alternative proposal posits that recycled marine carbonates act as dirty Pb isotopic 'spike' that variably contaminates (i.e., from PREMA to HIMU) the mantle sources of OIB and other intraplate lavas. Additionally, the Nb-Ta enrichment but Pb depletion that are pervasive features of OIB (e.g., Jackson et al., 2008; Peters and Day, 2014; Hart and Gaetani, 2006; Hofmann, 2014) are strongly complemented by the Nb-Ta depletion but Pb enrichment in subduction arc lavas (Castillo, 2015). Notably, the K-depletion of the alkaline rocks is also an additional, distinctive feature of HIMU-type OIB (Castillo, 2015; Weiss et al., 2016). Thus, we propose that a compositional connection exists between the mafic alkaline rocks and previously subducted carbonate-rich materials. This does not conflict with our proposal for metasomatized lithosphere given that the metasomatizing fluids that form the amphibole-rich metasomes are considered to be themselves derived from melting of a sublithospheric source containing recycled subducted materials (discussed below).

### Mantle melting

The variation in major and trace elements and isotopes displayed by NWRS basalts in Figure 10 has been suggested to be the result of the tapping of a heterogeneous, metasomatically enriched, mantle lithosphere at progressively smaller melt fractions ocean-ward (Panter et al., 2011). In Figure 14, basalts from the NWRS along with the other oceanic WARS basalts highlighted in this study are compared to model curves for modal and non-modal batch melting that involve three potential mantle sources that are relevant to what has been discussed above. They include a mantle composition calculated as the source of basaltic volcanism erupted near the Hobbs Coast in Marie Byrd Land (Hart et al., 1997), an average of Zambian eclogites interpreted to represent fossil subducted slab material (John et al., 2004) and an average composition for metasomatized (high-K, phlogopite-bearing) spinel peridotite xenoliths from northwest Germany (H-group; Hartmann and Wedepohl, 1990). Melting of primitive mantle (PM) composition (Hofmann, 1988) is also shown. Additionally for comparison are the results of high-pressure and high-temperature melting experiments on natural hornblendite and clinopyroxene hornblendite from Pilet et al. (2008). Models for PM, Hobbs and H-group sources produce steeper positive slopes with decreasing melt fractions in Nb versus La/Yb<sub>N</sub> space relative to the data arrays for basalt and melting experiments (Fig. 14A) whereas eclogite provides a very good fit to the data. In Nb versus Zr/Nb space (Fig. 14B), melting of metasomatized peridotite (H-group) provides the bestfit, however, eclogite is also reasonable at low melt fractions. All four models produce low Ti/Zr ratios (Fig. 14C) but at lower melt fractions (1 - 3%) values for eclogite are similar to the basalts. In general, the results demonstrate that the overall trend of increasing Nb concentration from alkali basalt to basanite to tephrite could be explained by melting of enriched mantle sources to a smaller degree, which in the case of the NWRS occurs ocean-ward as evidenced by the more common occurrence of strongly silica-undersaturated compositions (i.e. basanite and tephrite) away from the continent (Figs. 1B and 10).

There are two fundamental short-falls with this hypothesis. First is the difficulty in matching compositions for both major and trace elements derived by low-degree melting of eclogitic and peridotitic sources to those of low-Si alkaline basalts (< 45 wt.% SiO<sub>2</sub>) as emphasized by Pilet (2015). Although models of low degrees of melting ( $\leq 3\%$ ) for eclogite produce similar HFSE (e.g., Nb, Zr, Ti) concentrations, they also generate significantly higher LILE (e.g., Rb, Ba, and K) and Pb concentrations and lower HREE (Yb and Lu) and Y concentrations compared to the basalts. The melting of metasomatized peridotite (H-group) yield a better match for LILE and HFSE but Ti concentrations produced are too low (< 1 wt.%) relative to the basalts (2.0 - 4.8 wt. %, Table 2) and thus much lower Ti/Zr ratios (Fig. 14C). Second, is the difficulty in explaining the systematic variation in major and trace elements of NWRS basalts across the continent to ocean transect (Fig. 10) by changes in the degree of partial melting. This is problematic as the lithospheric thickness across the continent to ocean boundary differs substantially. Estimates of crustal thicknesses predict Moho depths beneath the eastern flank of Transantarctic Mountains to be between 25 and 35 km (Behrendt, 1999; Salimbeni et al., 2010) and in the Northern Basin (Fig. 1) between 20 and 15 km (Busetti et al., 1999). A Moho depth of 5 to 6 km is estimated at the shelf break between the Northern and the Adare Basin (Selvans et al., 2014) and that depth extends north into the Adare Trough (Müller et al., 2005; Selvans et al., 2014). The depth to the base of the lithosphere is broadly resolved through regional tomographic studies that indicate thicknesses of > 150 km beneath the East Antarctic craton, 100–70 km within the West Antarctic rift and < 80 km beneath ocean (Danesi and Morelli, 2001; Ritzwoller et al., 2001; Heeszel et al., 2016). If mantle source compositions and mantle potential temperatures near the base of the lithosphere are similar across the region, then partial melting should occur at much shallower depths beneath the Adare

Basin relative to the continent. This scenario would result in higher degrees of partial melting away from the continent, which is contrary to the interpretation based solely on geochemistry.

The most Si-undersaturated and incompatible element enriched basalts (basanite and tephrite) in the NWRS were also interpreted by Panter et al. (2011) to represent small degree melts that scavenged the most easily fusible material (i.e. amphibole). Higher Nb/Y ratios in WARS basalts correlate with higher values that correspond with values anticipated for melting of carbonated peridotite sources (Herzberg and Asimow, 2008), higher Ne + Lc normative contents, and are similar to experimental results from high degree melting of hornblendite (Figs. 13 and 15A and B). As previously mentioned, Nb can reach high concentrations in amphibole and in mantle that has experienced carbonatite metasomatism. Furthermore, the compatibility of Nb in amphibole is mostly less than unity while Y is compatible ( $D^{amph/melt} = 0.53$  and 1.48, respectively; averaged values calculated by Pilet *et al.*, 2011 using data from Tiepolo et al., 2000a, 2000b, and 2007). Thus we interpret the higher Nb/Y ratios in basanite and tephrite as indicating greater contributions from amphibole-rich ( $\pm$  CO<sub>2</sub>) metasomes within the lithospheric mantle. In the case of the NWRS, the signature becomes more prominent ocean-ward (Fig. 15C).

### Melt genesis between ocean and continent

We propose that the geochemical signature of mafic alkaline basalts in the NWRS is a consequence of high degree melting of amphibole-bearing metasomatic veins in both oceanic and continental lithosphere and that the progressive interaction of the melt with the surrounding mantle (*cf.* Pilet et al., 2008), occurring to a greater extent continent-ward, provides the best explanation for the systematic, gradational (i.e. non-stepwise) variation in major and trace element and isotopic data (Figs. 10 and 15C). As briefly outlined above, the multi-step process to produce low-Si alkaline basalts proposed by Pilet *et al.* (2004, 2005, 2008, 2010, 2011; Pilet, 2015) requires lithospheric

enrichment by small degree melts from sub-lithospheric mantle sources followed by high degree melting of the created amphibole-rich metasomes, which in turn, react with the surrounding peridotite during melt migration. There are several important aspects of Pilet et al.'s model that need to be emphasized and aligned with what is being proposed here. First, is that their model demonstrates (theoretically and experimentally) that the major and trace element composition of the amphibole cumulates is not dependent on the compositions of the underlying asthenosphere. The major element composition, including K and Ti of amphibole-bearing cumulate is controlled by the stoichiometry of the minerals, while the trace element content is controlled mostly by the exchange equilibrium between minerals and liquid ( $K_D^{min/melt}$ ). A second vital feature of their model is that unlike major and trace elements the isotopic composition of the cumulate is derived from the underlying asthenosphere. Therefore it is suggested that the isotopic signature of the most Si-undersaturated and incompatible element enriched basalts with the highest Nb/Y ratios may best represent the composition of the sub-lithospheric source with low <sup>87</sup>Sr/<sup>86</sup>Sr (< 0.7030), high  $^{143}$ Nd/ $^{144}$ Nd (~ 0.5130) and high  $^{206}$ Pb/ $^{204}$ Pb ( $\geq$  20) ratios (Figs. 7, 10, 12 and 15D and E). The isotopic 'endmember' is characteristic of sources for Cenozoic alkaline magmas in West Antarctica as well as alkaline basalts erupted on widely scattered continental fragments of East Gondwana (Zealandia and eastern Australia). As noted earlier, the signature has been deemed 'young 'HIMU' or more recently 'CarboHIMU' (McCoy-West et al., 2016) in order to distinguish it from endmember HIMU from oceanic settings (e.g., St. Helena) and plot within the DMM-PREMA field (Fig. 7) as defined by Stracke (2012). Indeed, most researchers agree that the potential mantle sources for this diffuse alkaline magmatic province (DAMP, Finn et al., 2005) reside in metasomatized continental lithospheric domains, although the timing and cause of metasomatism as well as the ultimate source of the isotopic signatures remains controversial (Rocchi et al., 2002; Panter et al., 2006; Timm et al., 2010; Scott et al., 2014; McCoy-West et al., 2010, 2016; van der

Meer *et al.*, 2017). The results presented here and by Kipf et al. (2013) confirm that PREMA (aka. young HIMU or CarboHIMU) isotopic signatures are also present in basalts erupted through oceanic lithosphere.

As proposed earlier, the trace element and radiogenic isotopic signatures of PREMA and HIMU originate from previously subducted slab materials. Specifically, the endmember HIMU may owe its highly radiogenic Pb, low <sup>87</sup>Sr/<sup>86</sup>Sr, and distinctive trace element signatures to Archaean marine carbonates (Castillo, 2015, 2016; Weiss et al., 2016). The variable and less radiogenic Pb plus ~higher <sup>87</sup>Sr/<sup>86</sup>Sr isotopes of PREMA, on the other hand, come from subducted lithospheric mantle that was carbonated, or 'spiked' by a lesser amount of Archaean or possibly younger marine carbonates. Notably, our proposal is highly consistent with recent results indicating that alkaline and carbonatitic rocks from the northern and eastern regions near the edges of the North China Craton acquire some of their distinctive compositional features from subducted marine carbonates (Chen et al., 2016; Li et al., 2017 and references therein). These regions are tectonically similar to WARS in that they were previous convergent margins.

In the NWRS, measured <sup>87</sup>Sr/<sup>86</sup>Sr ratios increase while <sup>143</sup>Nd/<sup>144</sup>Nd and <sup>206</sup>Pb/<sup>204</sup>Pb ratios decrease in basalts erupted with decreasing distance from the continent (Fig. 10). The overall age of the volcanism also increases continent-ward; however, applying an age-correction for radiogenic in-growth cannot explain this variation (i.e. corrections are < 5 × 10<sup>-5</sup> for all three isotopic systems). What is changing is the thickness and age of the lithosphere, both of which increase from the Adare Basin towards the East Antarctic craton. Additionally, as inferred from the correlation between isotopic and Nb/Y ratios (Fig. 15D and E), the basalts erupted closer to the continent are interpreted to have less contribution (i.e. lower Nb/Y ratios) from metasomes and thus their isotopic compositions may be displaced from sub-lithospheric sources by adoption of lithospheric mantle signatures. The longer residence time required for a melt, which is generated from amphibole-

bearing veins near the base of the lithosphere, to migrate upward through progressively thicker and older lithospheric mantle will advance melt–peridotite reactions. Experimental studies show that the reaction between Si-undersaturated (i.e. Ne-normative) liquid will cause dissolution of orthopyroxene in peridotite and will become enriched in silica while crystallizing olivine (Shaw et al., 1998; Shaw, 1999; Lundstrom et al., 2000). Average olivine-liquid equilibrium temperatures calculated for NWRS basalts (Krans, 2013) are  $1244^{\circ} \pm 40^{\circ}$ C for anhydrous conditions (Beattie, 1993),  $1211^{\circ} \pm 37^{\circ}$ C for hydrous conditions (Putirka et al., 2007) and  $1145^{\circ} \pm 29^{\circ}$ C for CO<sub>2</sub>-rich conditions (Sisson and Grove, 1993) and do not vary systematically between continent and ocean. Layered hornblendite and peridotite melting experiments performed at 1.5–2.5 GPa and 1200°– 1325°C on natural samples by Pilet et al. (2008) demonstrate that this reaction will evolve liquids with lower alkalinity, TiO<sub>2</sub>, CaO, FeO and incompatible trace element contents. It will also produce slightly higher MgO and Al<sub>2</sub>O<sub>3</sub> contents, while maintaining relatively constant K<sub>2</sub>O/Na<sub>2</sub>O and Al<sub>2</sub>O<sub>3</sub>/TiO<sub>2</sub> ratios. These features are consistent with the major and trace element trends from nephelinite/basanite to alkali basalt (~41–50 wt.% SiO<sub>2</sub>; Fig. 3) that are documented in this study. Alternatively, increasing degree of partial melting of a common mantle source has been used to explain this compositional continuum found in both oceanic and continental lavas (Frey et al., 1978; Caroff et al., 1997; Beccaluva et al., 2007; Bosch et al., 2014). However, following the arguments that we have presented above, we consider the reaction of Si-undersaturated liquids (i.e. nephelinitic/basanitic) with surrounding peridotite in the lithosphere to be a more plausible explanation for the systematic geochemical and isotopic variation in NWRS basalts from ocean to continent.

Further support for this assertion is provided by olivine oxygen isotope data. Recall that  $\delta^{18}O_{ol}$  values for NWRS basalts are highly variable ( $\Delta \sim 1.5\%$ ; Fig. 8A) and are not adequately explained by contamination from crust (Fig. 12). High  $\delta^{18}O_{ol}$  values correlate with lower Nb/Y

ratios of corresponding whole rock compositions (Fig. 16A) and trend towards mineral values measured on orthopyroxene separated from peridotite xenoliths hosted by NVL basalts (Perinelli *et al.*, 2006). The correlation is consistent with reaction between low  $\delta^{18}$ O liquid generated by melting of amphibole-rich metasomes (i.e. high Nb/Y source) and peridotite, where dissolution of orthopyroxene and crystallization of olivine evolve liquid towards more Si-saturated (Fig. 16B) and  $^{18}$ O-enriched compositions. Moreover, variations in trace elements, including Th and La, with SiO<sub>2</sub> content (Fig. 16C and D) parallel the trends produced by the melting and reaction experiments of Pilet *et al.* (2008), providing additional support for this hypothesis.

An alternative explanation, however, is that the trends may be a result of mixing between Si-undersaturated melt and a liquid generated by the partial melting of recycled pyroxenite (pyroxene + garnet). Mixing proportions require > 20–80% of gabbro/eclogite to account for the compositional range of the basalts (Figs. 11, 12, 16A and B). Based on experimental and thermodynamical constraints, partial melting of volatile-free pyroxenite cannot explain major element compositions of low-SiO<sub>2</sub> alkaline basalts (Lambart et al., 2013; Pilet, 2015 and references therein) though homogenization with volatile-rich Si-undersaturated melt may have some merit. Even so, a model for mixing must be able to emulate the systematic geochemical variation from land to sea in the NWRS, which would require the proportion of pyroxenite to increase towards the continent. Here again we favor the melt-peridotite reaction scenario in that it is better constrained by the evidence and provides a much more straightforward model (cf. Occam's razor) for NWRS volcanism. Yet, some contribution from recycled pyroxenite may be warranted. Melchiorre et al. (2011) call upon the melting of a sub-lithospheric source containing between 15% and 60% eclogite to explain radiogenic Os and Hf isotope values measured in metasomatized peridotite xenoliths from NVL. Evidence from xenoliths, seismic and geodynamic investigations indicate that recycled volatiles and eclogitic material are likely to have been introduced into the sub-lithospheric mantle beneath West Antarctica during Paleozoic–Mesozoic Gondwana subduction (Finn *et al.*, 2005; Sutherland *et al.*, 2010; Melchiorre *et al.*, 2011; Emry *et al.*, 2015; Martin et al., 2015; Broadley *et al.*, 2016).

# Subduction-related agents in metasomes

The lower oxygen isotope values measured on olivine ( $\delta^{18}O \le 5$  ‰) in NWRS basalts is most likely a signature derived from hydrothermally altered subducted materials. Metasomatism of the NVL lithosphere by marine derived volatiles is indicated by heavy halogen (Br, I) and noble gas (Ar, Kr, Xe) compositions liberated from fluid inclusions in olivine and pyroxene in peridotite xenoliths (Broadley *et al.*, 2016). Broadley et al. (2016) suggest that the volatiles were released, possibly at depths up to 200 km (Sumino *et al.*, 2010; Kendrick *et al.*, 2013), during subduction and were incorporated into the overlying mantle wedge and subcontinental lithospheric mantle beneath the Gondwana continental arc during the Paleozoic. Their investigation, along with those of Coltorti et al. (2004), Perinelli et al. (2006, 2011) and Melchiorre et al. (2011), provide strong support for metasomatism of lithosphere by subduction zone processes and contributions from recycled subducted material to the source of alkaline magmas in NVL, including the NWRS.

Mafic eclogite formed by past subduction of oceanic lithosphere have highly variable oxygen isotopic compositions that range from extremely heavy ( $\delta^{18}O > +10$  ‰) to extremely light ( $\delta^{18}O < -10$  ‰) values for mineral and whole rock samples (Zheng *et al.*, 1998, 2003; Putlitz *et al.*, 2000; Früh-Green *et al.*, 2001). The variability of  $\delta^{18}O$  is explained by the exchange between fluids (meteoric or seawater  $\delta^{18}O \le 0$  ‰) and rock ( $\delta^{18}O_{basalt/gabbro/peridotite} > 5-7$  ‰) over a wide range of temperatures and fluid/rock ratios in low to high pressure environments during the subduction cycle. Generally, lower temperature hydrothermal alteration within the upper oceanic crust will enrich rocks in  $^{18}O/^{16}O$  while higher temperature ( $\ge 250$ °C) exchange within the lower

crust/upper mantle will deplete rocks in  $^{18}\text{O}/^{16}\text{O}$  (Muehlenbachs, 1986). Low oxygen isotope values ( $\delta^{18}\text{O} \leq 5$  ‰) measured on whole rock and clinopyroxene in veins and eclogitized metagabbro from the Erro–Tobbio peridotite, Italian Alps, is explained by Früh-Green et al. (2001) to be a consequence of high temperature alteration of lower oceanic crust/upper mantle by seawater that has been preserved (i.e. lack of oxygen isotopic re-equilibration with ambient mantle) through eclogitization and limited fluid circulation within a relatively closed system at high pressures ( $\geq 2$  GPa).

In HIMU ocean island basalts, low oxygen values ( $\delta^{18}O_{ol}$  4.7–5.2 ‰) are also interpreted to be a consequence of long-term recycling of subducted materials (Eiler et al., 1996; Eiler, 2001). Day et al. (2010) call upon partial melting of metasomatized mantle peridotite that contains ~10% recycled pyroxenite/eclogite; itself a product of high-temperature alteration of oceanic crust and lithosphere, in order to account for the correlation between low  $\delta^{18}O_{ol}$  (4.87  $\pm$  0.18 %) and high  $^{206}$ Pb/ $^{204}$ Pb ratios (values up to > 20) in basalt from La Palma, Canary Islands.  $\delta^{18}$ O<sub>ol</sub> values for NWRS basalt display a weak correlation with <sup>206</sup>Pb/<sup>204</sup>Pb, as well as Sr and Nd isotopes (Fig. 9) and indicate that the low  $\delta^{18}O$  signature is associated with the low  $^{87}Sr/^{86}Sr$ , high  $^{143}Nd/^{144}Nd$  and high <sup>206</sup>Pb/<sup>204</sup>Pb ratios. As discussed earlier, the origin of radiogenic Pb signatures (> 19.5) in mafic alkaline rocks found throughout West Antarctica and continental fragments of Gondwana (DAMP, Finn et al., 2005) is controversial and has been related to both lithospheric and sublithospheric sources. The highest <sup>206</sup>Pb/<sup>204</sup>Pb ratios (> 20.7) measured in mafic alkaline rocks are from continental basalts in Marie Byrd Land and Chatham Island (Panter et al., 2000, 2006) and values greater than 21 have been measured on spinel peridotite xenoliths from Chatham Island and the Waitaha domain of southern New Zealand (McCoy-West et al., 2016). For basalts within the NWRS the highest <sup>206</sup>Pb/<sup>204</sup>Pb ratios (> 20) occur ocean-ward (Fig. 10) and, as discussed above, we consider that this signature, along with radiogenic Nd and depleted Sr and O isotopic values, to have been 'originally' derived from recycled crustal materials and transferred by small degree melts to the lithosphere and stored in amphibole-rich cumulates (i.e. metasomes). The higher <sup>206</sup>Pb/<sup>204</sup>Pb ratios found in some continental basalts and xenoliths scattered across the DAMP may be explained by compositional heterogeneity of the recycled material in the sub-lithospheric mantle, melting of higher proportions of the recycled material (*cf.* Day *et al.*, 2010), and/or due to rapid in-growth facilitated by high U/Pb ratios (McCoy-West et al., 2016). The fractionation required to produce high U/Pb has been explained by compatibility differences between these elements during carbonatite metasomatism (McCoy-West *et al.*, 2016) or during post-metasomatic partial dehydration (Panter *et al.*, 2006).

Importantly, the fractionation of oxygen isotopes ( $^{18}\text{O}/^{16}\text{O}$ ) between silicate liquids and silicate minerals at asthenospheric temperatures ( $\geq 1350^{\circ}\text{C}$ ) or temperatures in the lithospheric mantle ( $> 900^{\circ}$ – $1100^{\circ}\text{C}$  at 0.9–1.7 GPa; spinel peridotite stability determined for continental NVL by Armienti and Perinelli, 2010) would be negligible. Therefore if the increase in  $^{206}\text{Pb}/^{204}\text{Pb}$  observed in NWRS basalts ocean-ward (Fig. 10) is a result of radiogenic in-growth, then the correlation with  $\delta^{18}\text{O}$  (Fig. 9) would not be expected (i.e. decoupled systems). Once again we regard the deviation in oxygen and radiogenic isotopes from the ocean 'endmember' of the NWRS to be a consequence of the reaction between silica-undersaturated melts derived from metasomes and surrounding peridotite, increasing as melt migrates through thicker mantle lithosphere continent-ward. The isotopic signatures of the metasomes, on the other hand, were ultimately formed from recycled crustal and lithospheric materials that were transferred to the asthenosphere through carbonatitic silicate melts derived from the down-going slab (Castillo, 2015, 2016).

## Relative timing of extension, metasomatism & magmatism

There are significant time off-sets between major episodes of extension and alkaline magmatism in West Antarctica. The earliest magmatism, the Meander Intrusive Group (48 to 23 Ma) in NVL followed Gondwana break-up and the broad phase of rifting within the WARS (105 to 80 Ma) by  $\sim 30$  Myr. The focused phase of continental extension within the WARS (80 to 40 Ma) occurred  $\sim 25$  Myr prior to the activity that formed the larger shield volcanoes along the continental coastline in the NWRS ( $\sim 14$  to > 5 Ma). Monogenetic island and seamount volcanism on the continental shelf and Adare Basin (< 5 Ma to < 100 ka) followed extension in the Northern Basin and seafloor spreading in the Adare Basin (43 to 26 Ma) by  $\sim 20$  Myr. Although the estimates are loosely constrained, it appears that progressively shorter time periods of off-set between extension and magmatism (i.e.  $\sim 30 \rightarrow 25 \rightarrow 20$  Myr) are also spatially coincident with the broad shift in the location and volume of volcanism as well as with a decrease in post-extensional lithospheric thickness to the northeast within the NWRS.

The timing of metasomatism of the mantle lithosphere in West Antarctica and New Zealand has been estimated based on model ages from isotopes. Lead model ages evaluate the in-growth of high  $^{206}\text{Pb}/^{204}\text{Pb}$  ratios ( $\geq$  20) starting from a PREMA-FOZO composition ( $\sim$  19.5) with variable U/Pb ratios. Estimates range from 500 to 200 Ma (Hart et al., 1997; Panter *et al.*, 2000) and 150 to 50 Ma (Nardini *et al.*, 2009). Model ages calculated for spinel peridotites from southern New Zealand reveal metasomatic enrichments of  $\leq$  180 Ma (McCoy-West *et al.*, 2016), including extremely rapid in-growth of 35 to 13 Ma for some samples with very high  $^{238}\text{U}/^{204}\text{Pb}$  ratios ( $\mu$  = 63 – 466). Melchiorre et al. (2011) present Os model ages (time of Re depletion) for interstitial sulfides in metasomatized peridotite xenoliths from NVL that span the evolutionary history of Antarctica from the Archean–Proterozoic up to the Cretaceous but speculate that the high proportions of eclogite required to explain radiogenic  $^{187}\text{Os}/^{186}\text{Os}$  could have been introduced into the sub-lithospheric mantle by subduction during the Ross Orogeny (550–600 Ma). They conclude

that the ancient eclogitic signatures were likely incorporated into partial melts during the initial phase of WARS extension.

According to Pilet (2015), the difference between how alkaline magma is generated in continental versus oceanic settings is that in continental settings the episode of metasomatic enrichment may be decoupled from the events that cause melting of the metasomes. This scenario has been proposed in continental NVL and is in accordance with the time off-set between rifting and magmatism discussed above. Rocchi et al. (2002, 2005) and Nardini et al. (2009) suggest that during the initial Late Cretaceous phase of WARS extension small volume melts from the asthenosphere produced amphibole-rich metasomatic veins that become a lithospheric mantle source for alkaline magmatism from the middle Eocene to Present. The timing is likely controlled by the thermal gradient in the lithosphere which must reach 1150-1175°C to allow melting of amphibole-rich cumulates (Pilet et al., 2008). Thermobarometric data on peridotite xenoliths (Armienti and Perinelli, 2010; Perinelli et al., 2011) reveal a change in the geothermal gradient from  $0.5^{\circ}$ C/km to  $\sim 3^{\circ}$ C/km in NVL lithosphere. The thermal evolution is estimated to have taken  $\sim 10$  Myr and is credited to the development of edge-driven mantle convection along the boundary between the thinned lithosphere of the WARS and the thick East Antarctic craton established by the Eocene (Faccenna et al., 2008).

Here we extend the model for metasomatism to include the oceanic portion of the NWRS and call upon the influence of younger, mostly amagmatic rifting to generate the low degree sub-lithospheric melts that have enriched the lithosphere. The ultraslow spreading in the Adare Basin (~ 12 mm/yr, Cande et al., 2000) and crustal characteristics, which are consistent with ocean core complexes observed at ultraslow spreading ridges (Selvans *et al.*, 2014), likely facilitated the repose between the melting that formed amphibole-rich veins and the melting of these metasomes to produce alkaline volcanism. Studies of ultraslow seafloor spreading (Dick *et al.*, 2003; Michael *et* 

al., 2003; Cannat  $et\ al.$ , 2006) reveal that mantle melting and the resulting volcanism is not a simple function of spreading rate but that mantle temperatures and mantle lithology/composition are also controlling factors. Ultraslow spreading will lead to cooler temperatures with depth and therefore thicker lithosphere (Reid and Jackson, 1981). Moreover, temperatures for sub-lithospheric mantle beneath ultraslow spreading ridges (70–80 km depth) may be colder by  $\geq 180^{\circ}$ C relative to mantle beneath fast spreading ridges (Husson  $et\ al.$ , 2015). We suggest that the temperature at the base of the oceanic lithosphere beneath the Adare Basin was cool enough to 'freeze-in' low degree melts, and the metasomatic cumulates that they produce, during or soon after seafloor spreading (43–26 Ma). Furthermore, we propose that delay in melting of the metasomes leading to seamount volcanism in the Adare Basin (< 5 Ma) was controlled by the rate of conductive thermal migration from the base of the lithosphere. Basal heating was enhanced by regional mantle upwelling related to subduction death and the sinking of the Pacific slab into the lower mantle (Finn  $et\ al.$ , 2005; Sutherland  $et\ al.$ , 2010) and possibly more recent (Neogene) edge–driven flow (Faccenna  $et\ al.$ , 2008).

## **Integrated model for magmatism**

Fig. 17 illustrates what we envisage to explain the origin of alkaline magmatism in the NWRS. Our model is intimately connected to what is known of the tectonic evolution of the region beginning at a time prior to the break-up of the proto-Pacific margin of Gondwana (Fig. 17A). The protracted history of subduction along this margin is considered to have supplied carbonate-rich slab material directly or by longer-term recycling into the upper asthenosphere, however, our model does not preclude the contribution from more ancient subduction-related sources. A series of extensional phases occurred in the development of the WARS which caused lithospheric thinning and decompression melting of this recycled material in the asthenosphere, probably aided by

heating from passive mantle upwelling and edge-driven flow (Fig. 17B–E). The carbonate-rich silicate liquid produced by these small degrees of melting did not reach the crust but crystallized in the mantle lithosphere to form amphibole-rich cumulates (metasomes) that later became the source of alkaline magmatism. The metasomatic enrichment of the lithosphere is invoked by our model not only to explain some of the geochemical and isotopic characteristics of the basalts and xenoliths described in this and other studies but also to explain why major periods of extension are not concurrent with or immediately followed by magmatism. Other geodynamic evidence for lithospheric mantle sources for alkaline magmatism found on former pieces of Gondwana (DAMP; Finn et al., 2005) is provided by Panter et al. (2006). They document the occurrence of relatively uniform mafic alkaline compositions on Chatham Island that were derived from an amphibolebearing mantle source and erupted intermittently over an 80 Myr period (85 to 5 Ma) as the continental lithosphere of Zealandia rifted away from West Antarctica and drifted north over a distance > 3000 km. Specific to the NWRS portion of the DAMP is that this region represents a recently rifted continental margin across which the lithosphere varies in both thickness and age (Fig. 17). We have shown that this is coincident with gradational changes in basalt geochemistry, isotopes (Figs 10 and 15C) and broadly with eruption age. Our model explains these features by invoking an ocean-ward progression in melting of the metasomes facilitated by the thermal evolution of the lithosphere and coupled to a decrease in the extent of reaction between melt and peridotite prior to eruption which, in turn, is controlled by lithospheric thickness (Fig. 17C-F).

Our multi-stage model provides the most plausible explanation for alkaline volcanism in the northwest Ross Sea and is relevant to interpretations on the origin and cause of alkaline magmatism within the rest of the West Antarctic rift system as well as other continental areas of the southwestern Pacific (i.e. DAMP). Importantly, our model is constructed using contemporaneous, petrogenetically related, alkaline magmas that transition (compositionally) across the boundary between continental and oceanic lithosphere, and is further constrained by the dynamic thermal-tectonic evolution of the region. The contribution of lithospheric mantle, both continental and oceanic, as a source and a contaminant of Si-undersaturated melt is important to understand and should be carefully considered in studies of alkaline basalts worldwide.

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#### **TABLES**

Table 1 – Summary of  $^{40}$ Ar/ $^{39}$ Ar dating results and analytical methods.

Table 2 – Geochemical and radiogenic isotope (Sr, Nd, Pb) analyses of NWRS basalts.

Table 3 – Oxygen isotope data from olivine phenocrysts.

#### **ELECTRONIC APPENDICES - TABLES**

Table A1 – NBP0701 Dredge Log

Table A2 -  $^{40}$ Ar/ $^{39}$ Ar analytical data

Table A3 – SIMS oxygen data

#### **FIGURES**

Fig. 1 – Regional and sample location maps. **A)** subglacial surface elevation map (BEDMAP2, Fretwell et al., 2013) with the approximate boundaries of the West Antarctic rift system delimited by red dashed lines. Also shown are locations of *oceanic* volcanic fields (seamounts and islands) that are compared in this study. MBS = Marie Byrd Seamounts,. **B)** regional map of the north western Ross Sea (NWRS) showing extent of Late Cenozoic volcanism above and below sea level. Faults and volcanic seamounts in the Adare Basin were mapped by Granot et al. (2010) using a combination of seismic and multibeam bathymetry data. The location of the basalts examined in this study (solid color symbols) and from other studies (open symbols) are shown with symbol shapes corresponding to their major element compositional name as defined in Fig. 3. Volcanic

deposits along the northern coastline of Victoria Land and islands are part of the Hallett volcanic province (Kyle, 1990). Here we defined NWRS basalts as those that are included in the Hallett volcanic province, the Malta Plateau of the Melbourne volcanic province (Kyle, 1990), as well as seamounts located on the continental shelf and within the Adare Basin. The bold dashed line delimits the approximate boundary between the East Antarctic craton and the West Antarctic rift. The bold dotted line at 1500 mbsl delimits the approximate boundary between rifted continental lithosphere and oceanic lithosphere. AP = Adare Peninsula, PI = Possession Islands, HP = Hallett Peninsula, DP = Daniel Peninsula, Cl = Coulman Island, MP = Malta Plateau.

Fig. 2 – Representative  $^{40}$ Ar/ $^{39}$ Ar ages for NWRS basalts. Age release, radiogenic argon release ( $^{40}$ Ar\*) and Ca/K spectra on groundmass concentrates are shown. Thickness of boxes for apparent age represent  $2\sigma$  error.

Fig. 3 – Total alkali versus  $SiO_2$  (wt.%) classification of basalts (Le Bas *et al.*, 1986). Thirty basalts from the NWRS (sample locations are shown in Fig. 1B and compositions in Table 2) are marked by solid color symbols (this study, n = 18) and open color symbols (n = 12) are analyses from Nardini et al. (2009), Mortimer et al. (2007), Rocholl et al. (1995) and Aviado et al. (2015). Southern Ross Sea (SRS) seamounts and islands (open black symbols) are from Aviado et al. (2015) and the Hubert Miller seamount (Marie Byrd seamount group) and Peter I Island (blue crosses) are from Hart et al. (1995) and Kipf et al. (2014). One sample (violet triangle, *olivine nephelinite*, SAX20, Nardini et al., 2009) is from the Melbourne volcanic province (MVP) which lies adjacent and to the south of the NWRS area. All basalts plotted have MgO > 6 wt.% and are normalized to 100% volatile free with total Fe as FeO. Over 400 ocean island basalts (OIB) from the Saint Helena, Austral-Cook and Society Islands (GEOROC data base, MgO > 7 wt.%) are

shown for comparison and include those derived from the melting of enriched (EM) and high U/Pb (HIMU) mantle sources.

- Fig. 4 Major elements versus MgO (wt.%). References, symbols and data criteria are given in Fig. 3 and Table 2.
- Fig. 5 Selected trace elements (ppm) versus MgO (wt.%). Symbols and data criteria are given in Fig. 3 and Table 2.
- Fig. 6 Primitive mantle normalized (Sun and McDonough, 1989) trace element diagrams.

  References, symbols and data criteria are the same as in previous figures.
- Fig. 7 Radiogenic isotope (Sr, Nd, Pb) compositions of Antarctic basalts compared to the ocean island basalt data set. Mantle source endmembers PREMA (FOZO), HIMU and EM are taken from Stracke (2012). References, symbols and data criteria are given in Fig. 3 and Table 3. Additionally, basalts MA009a and MA-117 from the Malta Plateau (Rocholl *et al.*, 1995) were measured for Pb isotopes in this study and the results are provided in Table 2.
- Fig. 8 Relationships between  $\delta^{18}O_{ol}$  and (**A**) average Fo % and (**B**) whole rock MgO wt.%. Oxygen isotopes of olivine are measured by SIMS (single spot) and LF (4–5 grains) analysis for NWRS samples (this study) and LF analyses of olivine from other NVL basalts including those in the NWRS and MVP (Mortimer *et al.*, 2007; Nardini *et al.*, 2009). **A**)  $\delta^{18}O_{ol}$  obtained by SIMS and Fo% by EMPA consist of both cores and rims of individual olivine grains comprising a total of 21 samples (Table 3). The line marking the average  $\delta^{18}O$  value of NWRS olivine (5.20 ± 0.25%,

2SD) is based on 190 analyses, including 9 samples by LF (Table 3). The Fo content for samples analyzed by LF is the average of multiple grains determined by EMPA in same sample. The dashed field for mantle olivine represents an average value of  $5.18 \pm 0.28\%$  at 2SD (LF - Mattey et al., 1994) and Fo % of 87–96 (Deer et al., 1992). For comparison values for olivine from the Canary Islands (LF & SIMS - Gurenko *et al.*, 2006, 2011), Reykjanes Peninsula, Iceland (LF - Peate *et al.*, 2009) and the Azores Islands (LF - Genske *et al.*, 2013) are shown. **B**) SIMS  $\delta^{18}$ O<sub>ol</sub> are the average of all spot analyses within a single sample (cores & rims on several grains). Tie-lines connect individual samples that have been analyzed by multiple methods. Core\* is the average  $\delta^{18}$ O<sub>ol</sub> value of 5–6 spots within the core of a single olivine grain.

Fig. 9 –  $\delta^{18}$ O<sub>ol</sub> and Sr, Nd and Pb isotopes. The oxygen isotopic values for olivine phenocrysts from NWRS basalts are LF (open symbols) and SIMS data averaged for multiple spots on 2-3 grains in each sample (Table 3). Error bars are standard error of the mean (SEM). Symbols for the basalts are the same as those given in Fig. 8. Radiogenic isotope ratios are measured values on whole rock. Vertical line marks the average  $\delta^{18}$ O value of NWRS olivine (5.20 ‰) calculated for all analyses (n = 190).

Fig. 10 – Variation in major and trace elements and isotopes with decimal degree longitude for NWRS basalts. References, symbols and data criteria are the same as in Fig. 3, and for oxygen isotopes, Fig. 8. The geographic location of each sample is shown in Fig. 1B. Ocean and continental sectors are defined in Fig. 1B and the ocean to continent transition zone (OCTZ) is demarcated as the region between the base of the continental shelf ( $\sim$ 1500 mbsl) and the inferred boundary between the East Antarctica craton and extended lithosphere comprising the WARS (bold dashed line in Fig. 1B). Lines of regression, which includes all data, have  $r^2$  that range from 0.4 -

0.6 and Spearman rank correlation coefficients that range from 0.6 - 0.8. For  $\delta^{18}O_{ol}$  only alkali basalt is regressed. Oxygen values are SIMS analyses except for three NWRS samples (open symbols) measured by LF from Nardini et al. (2009). Error bars represent standard error of mean (SEM). Sample La/Yb ratios are normalized to primitive mantle (Sun and McDonough, 1989).

Fig. 11 – Sr and Nd isotopic compositions versus Rb concentrations (ppm). A & B) shows selected basalt and crustal rock compositions used for mixing (Langmuir et al, 1977) and assimilationfractional crystallization (AFC; DePaolo, 1981) models. Basalt sample D12-1 was chosen to be representative of a parental magma composition based on its oceanic location and relatively unfractionated composition (MgO = 10.7 wt.%, Cr = 355 ppm, Ni = 205 ppm; Table 2). Also selected is sample SAX20 (Table 2) which is overall the most incompatible element enriched sample (Fig. 6) and has been equated to a primary (metasomatic) melt by Coltorti et al. (2004) and Perinelli et al. (2006). Crustal rocks from NVL consist of gabbro, diorite and granodiorite (Di Vincenzo & Rocchi, 1999; Dallai et al., 2003), metasediment (Henjes-Kunst and Schussler, 2003; Di Vincenzo et al., 2014) and eclogite (Di Vincenzo et al., 1997; Ghiribelli, 2000). Gray fields encompass AFC model curves (shown in detail in C & D) using  $D_{Rb} = 0.01$ ,  $D_{Sr} = 0.1$ ,  $D_{Nd} = 0.4$ and proportion of assimilation to crystallization (r) = 0.8. Mixing curves between basalt, eclogite and gabbro are also shown. Numbered tic marks are the percentage of crust in each mixture. C & **D**) Detail showing AFC model curves using basalts D12-1 (Table 2) and SAX20 (Nardini et al., 2009) and different contaminants (eclogite = sample G3; gabbro = DR13 and AC6; diorite = sample AF16; metasediment = argillitic). Numbers along each curve represents the fraction of liquid remaining (F). Gray dashed arrow shows the trend of increasing SiO<sub>2</sub> (wt.%) in basalt samples. References, symbols and data criteria for the basalts are the same as those given in Fig. 3.

Fig. 12 – Variations in  $\delta^{18}O_{ol}$  with whole rock Sr concentrations for NWRS basalts modeled by AFC using an altered basaltic composition (mugearite MB32.11; (Panter *et al.*, 1997) and mixing and AFC using gabbro from NVL (Di Vincenzo & Rocchi, 1999; Dallai et al., 2003). Sample MA009a was selected to represent a slightly <sup>18</sup>O–enriched basaltic magma and sample NV-4C was selected to represent a low  $\delta^{18}O$  basaltic magma (Tables 2 & 3). AFC models use  $D_{Sr} = 0.1$ , oxygen isotope bulk mineral–melt fractionation = 0.4% (Eiler, 2001), and the proportion of assimilation to crystallization (r) = 0.8, 0.5 and 0.3. Numbered tic marks on AFC curves represent the amount of liquid remaining and tic marks on the mixing curve represents the proportion of gabbro in the mixture. All but two of the basalts have MgO > 6 wt.% (Table 3) and the symbols used are the same as in Figure 8. The methodology for mixing and AFC are the same as in Figure 11.

Fig. 13 – Plot of normalized (CIPW) compositions calculated from whole-rock major element data. Silica undersaturated compositions plot within the Ol – Ne + Lc – Di triangle (truncated at 50 wt.% Ne + Lc) and silica saturated compositions plot within the Ol – Hy – Di triangle (up to 50 wt.% Hy). Symbols and data criteria is the same as in previous figures. Fields for basaltic compositions derived from melting experiments include carbonated pyroxenite (2.9 GPa, 1275-1475°C; Gerbode and Dasgupta, 2010), carbonated eclogite (3.5 GPa, 1250-1400°C; Kiseeva et al., 2012), hornblendite (1.5 GPa, 1150-1350°C; Pilet et al., 2008), clinopyroxene hornblendite (1 GPa, 1250-1400°C; Pilet et al., 2008), amphibole wehrlite (1.0 GPa, 1225-1350°C; Médard et al., 2006), garnet clinopyroxenite (2-2.5 GPa, 1340-1500°C; Hirschmann et al., 2003; Keshav *et al.*, 2004). Also included is melt derived from hornblendite plus peridotite sandwich experiments (1.5 GPa, 1225-1325°C; Pilet et al., 2008). Dashed gray arrow represents increasing proportion of melt fraction (*F*) determined for garnet clinopyroxenite in experiments by Keshav et al. (2004).

Fig. 14 – La/Yb<sub>N</sub> and Zr/Nb versus Nb plots. Basalts are compared to experimental melts by Pilet et al. (2008) on amphibole-bearing lithologies at 1.5 GPa and temperatures 1200-1400°C. La/Yb ratio is normalized to primitive mantle from Sun and McDonough (1989). Symbols and data criteria for the basalts are the same as those given in Fig. 3. Curves represent modal and non-modal batch melting using four different source compositions. Model batch melt of eclogite (cf. Pfänder et al., 2007) use an average eclogite composition (including Group III type) from John et al. (2004), source and melt mode of 0.80 clinopyroxene: 0.20 garnet, and partition coefficients from Pertermann et al. (2004; experiment A343). Non-modal batch melting of garnet lherzolite use the source composition for primitive mantle (PM) from Hofmann (1988), excepting Nb concentration which has been reset to 0.491 ppm (Münker et al., 2003), and a source derived from **Hobbs** Coast basalts in Marie Byrd Land by Hart et al. (1997). Source mode and melt mode reactions follow Salters (1996) where the source mode is 0.53 olivine; 0.04 orthopyroxene; 0.38 clinopyroxene and 0.05 garnet and the melt mode is 0.05 olivine; -0.49 orthopyroxene; 1.31 clinopyroxene and 0.13 garnet. Partition coefficients are from Pilet et al. (2011, and references therein). The fourth source is spinel peridotite that has experienced metasomatic enrichment (average 'H-group' = highpotassium, phlogopite-bearing) from Hartmann and Wedepohl (1990) with source and melt mode reactions and partition coefficients from Pfänder et al. (2012 and references therein). Titanium is compatible in phlogopite ( $K_D^{\text{phl/melt}} = 3$ , Pilet et al., 2011 and references therein) and therefore nonmodal batch melting of phlogopite-bearing peridotite (H-group) in panel C was modeled using a source mode of 0.55 olivine; 0.22 orthopyroxene; 0.15 clinopyroxene; 0.03 spinel; 0.05 phlogopite and melt mode of 0.05 : 0.05 : 0.65 : 0.05 : 0.2, respectively (cf., Ersoy et al., 2010). Numbers along the curves represent the degrees of melting in percent.

Fig. 15 – Compositional parameters and geographic locations (NWRS) of basalts plotted against Nb/Y ratios. Symbols and data criteria for the basalts are the same as those given in Fig. 3. **A**) volatilized peridotite index [= CaO – (2.318\*SiO<sub>2</sub> – 93.626)] defined by Herzberg and Asimow (2008) as a line on a CaO versus SiO<sub>2</sub> diagram that separates magmas derived from carbonated peridotite (Dasgupta et al., 2007), values > 0, and magma derived from carbonate-free peridotite (Walter, 1998; Herzberg, 2004, 2006), values < 0. Experimental melt of amphibole-rich sources (Pilet et al., 2008) are also shown. Line of regression excludes tephrite sample D9-1 (solid green triangle). **B**) Normalized (CIPW) values of basalts and experimental melts. Line of regression excludes sample MA009a. **C**) Longitude of NWRS basalts versus Nb/Y (refer to Fig. 8 for criteria for geographic sectors). Line of regression excludes tephrite D9-1 and a basanite from the continental sector (MP24, Nardini et al., 2009). **D** & **E**) Nd and Sr isotopes versus Nb/Y. Tephrite D9-1 is excluded from regression in both plots and MA009a is also excluded in plot E.

Fig.  $16 - \mathbf{A}$ ) Nb/Y ratio versus  $\delta^{18}$ O (‰). Oxygen isotopes for olivine phenocrysts from NWRS basalts were measured by LF and SIMS analysis (Table 3). Only samples with whole rock MgO values greater than 6 wt.% are plotted. Open symbols represent LF analyses of olivine from Nardini et al. (2009). In this study multiple grains of olivine in the same sample were analyzed by SIMS and averaged (solid colored symbols). Also shown are the average of multiple SIMS spot analyses (Core\*) within the cores of individual olivine grains for several samples (symbols filled with cross). The Nb/Y ratios represent whole rock values of the basalt that host the olivine. Orthopyroxene and vein amphibole  $\delta^{18}$ O and Nb/Y ratios measured on peridotite xenoliths from NVL are from Perinelli et al. (2006). Whole rock gabbro values provided by Dallai et al. (2003) on rocks from NVL. Error bars represent standard error of mean (SEM). B) SiO<sub>2</sub> (wt.%) versus Nb/Y ratios (symbols and data criteria of basalts are the same as in Fig. 3) are compared to the mineral chemistry of

orthopyroxene and amphibole from peridotite xenoliths hosted in alkaline basalts from NVL (Coltorti *et al.*, 2004; Perinelli *et al.*, 2006). Whole rock eclogite from NVL provided by Di Vincenzo et al. (1997) and average eclogite composition for whole rocks from Zambia are from John et al. (2004). Line of regression ( $r^2 = 0.52$ , Spearman rank = 0.68) for Antarctic basalts excludes sample D9-1 (solid green triangle). SiO<sub>2</sub> (wt. %) versus Th (C) and La (D) ppm of basalts are compared to natural peridotite, eclogite and gabbro as in panel B along with results of melting experiments conducted by Pilet et al. (2008). The experimental data is interpreted to represent change in degree of partial melting or source heterogeneity (solid gray arrow) and reaction between basanitic melt and peridotite (dashed black arrow). These trends are emulated by basanite and tephrite/nephelinite (red arrow) and alkali basalt (blue arrow) in this study.

Fig. 17 – Schematic illustration of Late Cretaceous to recent tectonism and the genesis of alkaline magmatism in the NWRS. **A**) pre-breakup continental arc. The subduction tectonic regime was nearly continuous from the late Neoproterozoic (~550 Ma) to Late Cretaceous (~100 Ma) along the Paleo-Pacific margin of Gondwana (Bradshaw, 1989; Cawood, 2005) and provided slab material to the asthenosphere either directly or recycled from depth in mantle upwellings. Deeper mantle upwelling may have also delivered more ancient subducted material to the asthenosphere (Castillo, 2016). **B**) Broad regional extension (~100 to 80 Ma) of Antarctic continental lithosphere (DiVenere *et al.*, 1994; Luyendyk *et al.*, 1996) caused decompression melting of slab material to a small degree. The resultant carbonate-rich silicate liquids rose and froze within the cooler lithospheric mantle to produce metasomes (amphibole-rich veins, *c.f.* Pilet *et al.*, 2008). The isotherms shown in this panel and in what follows are approximations that are in part constrained by the geothermal gradient estimated for NVL continental lithosphere by Armienti & Perinelli (2010). **C**) Focused extension (~80 to 40 Ma) between East and West Antarctica caused incipient lithospheric necking

(Huerta & Harry, 2007) and uplift of the Transantarctic Mountains (Fitzgerald et al., 1986). Heating of the lithosphere at its base was intensified by mantle upwelling and higher temperatures were obtained to melt the metasomes (>1150°C) to a high degree by about 50 Ma, which initiated the oldest know alkaline igneous activity (Meander Intrusive Group, ~48–23 Ma; Rocchi et al., 2002) associated with the WARS. The silica-undersaturated liquid that was produced reacted with the surrounding peridotite as it traversed upward through the lithosphere – see next panel. **D**) (note the change in scale). Ultraslow seafloor spreading formed the oceanic lithosphere of the Adare Basin from 43 to 26 Ma (Cande et al., 2000). The continuity of magnetic, seismic and structural trends from the Adare Basin into the Northern Basin (Davey et al., 2006; Cande and Stock, 2006; Damaske et al., 2007; Ferraccioli et al., 2009; Selvans et al., 2012) indicates that the oceanic crust extends across the continental shelf break (Granot et al., 2013; Selvans et al., 2016) and thus the transition in lithospheric type may be gradational from ocean to continent (OCTZ). Edge-driven convective flow was established at the boundary between thinned lithosphere and the thick East Antarctic craton (Faccenna et al., 2008). Metasomes were formed in the cooler OCTZ and oceanic lithosphere while the earlier formed metasomes in the warmed continental lithosphere continued to melt supplying magma for the Meander Intrusive Group until ~23 Ma. E) Minor post-spreading extensional events in the Adare Basin occurred at ~24 and ~17 Ma and formation of the Adare Trough (Granot et al., 2010, 2013). Alkaline magmatism resumed after ~10 Myr hiatus to form shield volcano complexes ( $\sim$ 14 to > 5 Ma) located along rift-shoulder fault systems. F) Thermal evolution of the lithosphere ocean-ward reached the melting temperature of metasomes by ~5 Ma to produce small volume volcanic seamounts on the continental shelf and in the Adare Basin.

#### **ELECTRONIC APPENDICES - FIGURES**

Figs. A1 - A9 – bathymetry maps of seamounts and dredging transects.

Fig. A10 –  $^{40}$ Ar/ $^{39}$ Ar age spectrum and isochrons.

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Table 1. Summary of <sup>40</sup>Ar/<sup>39</sup>Ar results and analytical methods

Sample	Locality	latitude	longitude	rock type	irrad.#	material	age analysis	integrated age (Ma)	2σ	n	isochron age (Ma)	2σ	n	MSWD	<sup>40</sup> Ar/ <sup>36</sup> Ar		plateau age (Ma)	2σ	n	MSWD	<sup>39</sup> Ar (%)	$^{39}\mathrm{Ar_K}$	K/Ca	2σ	preferred age (Ma)	method
	Adare seamounts																									
D2-1	trough W. scarp	69.861	171.844	hawaite	NM-217	GM conc.	furnace step-heat	15.93	0.76	10												49.4	0.11		$15.93 \pm 0.76$	integraged
D3-1	trough W. scarp	70.000	171.990	hawaiite	NM-217	GM conc.	furnace step-heat	4.93	0.66	10	4.56	0.27	10	1.63	296.2	2.3	4.61	0.19	10	1.52	100.0	100.0	0.28	0.36	$4.61 \pm 0.19$	plateau
D4-1	trough center	70.070	172.316	basanite	NM-251	GM conc.	laser step-heat	3.44	0.08	11	3.76	0.13	11	4.03	293.1	1.5	3.35	0.11	4	0.37	59.9	137.9	0.53	1.02	$3.35 \pm 0.11$	plateau
D4-3	trough center	70.070	172.316	basanite	NM-251	GM conc.	laser step-heat	2.45	0.02	11	2.51	0.09	8	7.94	311.0	16.0	2.61	0.02	5	1.65	50.3	67.1	1.2	0.3	$2.51 \pm 0.09$	isochron
D7-1	continental shelf	71.820	171.900	tephrite	NM-217	GM conc.	furnace step-heat	0.47	0.73	10	0.29	0.15	10	6.14	295.9	3.5	0.32	0.23	7	6.95	91.7	95.7	0.35	0.36	$0.32 \pm 0.23$	plateau
D9-1	S. basin	71.628	172.523	tephrite	NM-217	GM conc.	furnace step-heat	2.91	0.30	10	2.89	0.16	10	15.82	293.1	6.8	2.86	0.14	10	14.68	100.0	167.92	0.925	1.15	$2.86 \pm 0.14$	plateau
D12-1	S. basin	71.631	172.634	basanite	NM-217	GM conc.	furnace step-heat	3.52	0.50	10	2.87	0.07	9	8.05	298.0	10	2.89	0.06	9	6.27	99.8	149.5	0.35	0.39	$2.89 \pm 0.06$	plateau
D15-1	S. basin	71.579	172.733	tephrite	NM-217	GM conc.	furnace step-heat	3.12	0.22	9							2.82	0.04	2	0.82	55.3	187.5	0.30	0.47	$3.12 \pm 0.22$	integrated
D16-1	S. basin	71.597	172.821	tephrite	NM-217	GM conc.	furnace step-heat	2.97	0.56	11	2.66	0.19	11	1.25	296.9	1.9	2.76	0.13	9	1.44	99.2	158.2	0.27	0.31	$2.76 \pm 0.13$	plateau
D17-1	S. basin	71.499	173.486	trachyte	NM-217	GM conc.	furnace step-heat	0.20	0.07	10	0.14	0.01	9	1.81	295.2	4.0	0.14	0.02	9	1.58	99.9	268.2	11.8	14.9	$\boldsymbol{0.14 \pm 0.02}$	plateau
	Possession Islands																									
A227B	McCormick Is.	71.842	170.976	hawaite	NM-251	GM conc.	laser step-heat	0.168	0.031	11	0.058	0.029	8	2.05	295.5	1.6	0.053	0.031	8	2.54	95.6	163.2	0.30	0.34	$0.053 \pm .031$	plateau
*HC-SPI-4 (3873)	Possession	71.885	171.167	tephrite	NM-251	GM conc.	laser step-heat	0.10	0.04	11	0.17	0.02	11	6.98	294.3	2.1	0.16	0.02	5	0.54	77.6	107.5	0.30	0.31	$0.16 \pm 0.02$	plateau
*NV-3C (5166)	Possession	71.885	171.167	benmorite	NM-251	GM conc.	laser step-heat	1.40	0.03	10	1.41	0.01	8	0.64	298.2	6.0	1.42	0.0	8	0.19	85.6	293.4	2.3	2.1	$1.42 \pm 0.01$	plateau
*NV-4C (5171)	Foyn	71.950	171.083	hawaite	NM-251	GM conc.	laser step-heat	0.25	0.14	11	0.49	0.32	10	2.12	294.6	3.9	0.33	0.1	8	0.71	98.3	102.6	0.21	0.22	$0.33 \pm 0.10$	plateau
A225A	Foyn	71.957	171.093	basanite	NM-251	GM conc.	laser step-heat	0.26	0.03	11	0.076	0.042	10	3.00	301.3	2.4	0.24	0.03	3	0.85	66.5	82.8	0.16	0.30	$0.24 \pm .03$	plateau

<sup>\*</sup>Sample provided by the Polar Rock Repository - PRR-# (http://research.bpcrc.osu.edu/rr/)

### **Notes:**

MSWD, mean square weighted deviation.

Integrated age calculated by summing isotopic measurements of all steps. Integrated age error calculated by quadratically combining errors of isotopic measurements of all steps.

Plateau age is inverse-variance-weighted mean of selected steps. Plateau age error is inverse-variance-weighted mean error (Taylor, 1982) times root MSWD where MSWD > 1. Decay constants and isotopic abundances after Steiger and Jäger (1977).

## **Analytical parameters:**

Electron multiplier sensitivity for MAP 215-50 averaged 8.54e-17 moles/pA for furnace analyses and 4.37e-17 for laser analyses. Laser analyses sensitivity on Argus VI averaged 9.84e-17 moles/fA

MAP 215-50 total system blank and background for groundmass concentrate analyses averaged 5850, 7.10, 5.21, 22.4, 20.4 x 10<sup>-18</sup> moles at masses 40, 39, 38, 37 and 36, respectively.

MAP 215-50 total system blank and background for K-feldspar analyses averaged 409, 2.52, 1.10, 2.49, 7.75 x 10<sup>-18</sup> moles at masses 40, 39, 38, 37 and 36, respectively.

Argus VI total system blank and background for groundmass concentrate analyses averaged 358, 2.8, 0.19, 0.50, 1.3 x 10<sup>-18</sup> moles on masses 40, 39, 38, 37 and 36, respectively.

J-factors determined by CO<sub>2</sub> laser-fusion of 6 single crystals from each of 10 radial positions around the irradiation tray.

Correction factors for interfering nuclear reactions were determined using K-glass and  $CaF_2$  and are as follows:  $(40Ar/39Ar)K = 0.0000\pm0.0004$ ;  $(36Ar/37Ar)Ca = 0.00028\pm0.00001$ ; and  $(39Ar/37Ar)Ca = 0.00068\pm0.00005$ .

Table 2: Geochemical and radiogenic isotope (Sr, Nd, Pb) analyses of NWRS basalts

Sample # PRR - #	SAX20	MA009a	MA-117	MP24	A223	MP8	<b>HC-DP-3</b> 3839	MP34	MP32	P74794	P74833	A210B
dec. Latitude S dec. Longitude E	-73.770 165.950	-73.040 167.470	-73.020 167.630	-73.063 167.750	-72.872 168.970	-72.830 169.098	-72.667 169.633	-73.490 169.580	-73.490 169.580	-71.668 170.080	-71.653 170.120	-72.023 170.160
Geographical location	Greene Point, MVP	Malta Plateau	Malta Plateau	Malta Plateau	W of Daniell Peninsula	W of Daniell Pennisula	Daniell Peninsula	Coulman Island	Coulman Island	Robertson Bay	Robertson Bay	SW of Cape Roget
Rock name (TAS)	TEPH	AKB	AKB	BAS	AKB	AKB	AKB	BAS	BAS	AKB	AKB	AKB
Major elements (v	vt.%)											
SiO <sub>2</sub>	41.61	49.89	47.48	43.79	46.31	44.68	45.16	43.20	44.32	47.71	48.95	45.65
TiO <sub>2</sub>	3.69	2.45	2.66	2.83	2.24	2.74	3.19	3.11	3.42	2.49	2.10	
Al <sub>2</sub> O <sub>3</sub>	12.05	13.60	14.38	14.02	14.48	15.06	14.12	14.92	15.87	13.88	16.15	13.42
FeO <sub>t</sub> MnO	14.79 0.29	11.15 0.16	11.58 0.19	10.47 0.18	11.86 0.19	12.91 0.20	11.68 0.19	11.78 0.19	12.06 0.20	11.32 0.17	10.17 0.16	10.60 0.19
MgO	9.52	9.35	8.49	12.64	9.49	9.63	9.11	11.24	8.44	9.41	8.04	
CaO	9.42	9.96	10.98	11.15	11.22	11.03	12.94	11.09	10.52	11.26	10.23	9.83
Na <sub>2</sub> O	5.42	2.23	2.66	3.02	2.75	2.55	2.25	2.83	3.36	2.45	2.83	3.16
K2O	1.57	0.89	1.16	1.32	0.95	0.80	0.85	1.04	1.12	0.94	1.00	
$P_2O_5$	1.63	0.32	0.43	0.57	0.50	0.39	0.52	0.61	0.68	0.38	0.37	0.55
LOI	0.90	0.82	0.44	0.84	00.00	0.73	2.44	0.58	0.60	0.54	-0.48	100.00
Total Mg #	100.00 53.4	100.82 59.9	100.45 56.7	100.00 68.3	99.99 58.8	100.00 57.1	100.01 58.2	100.00 63.0	100.00 55.5	100.55 59.7	100.00 58.5	100.00 68.8
CIPW norm miner		33.3	30.7	06.3	30.0	37.1	30.2	03.0	33.3	33.7	36.3	00.0
Ne	23.7		2.4	12.5	5.0	5.5	5.0	10.3	9.2	1.1		8.6
Lc												
Di	26.5	18.7	22.8	24.9	23.0	20.5	28.6	21.2	18.8	24.0	16.3	21.2
Ну		17.7	40.0					0.7.4			1.5	
Ol	23.7	9.6	19.9	24.8	22.5	24.5	18.4	25.1	21.0	21.0	19.0	27.7
Trace elements (p <sub>i</sub> Sc	pm) 17.4	17			32	33	35	32.8	26.3		21	25
V	130	254	302		32	33	317	32.0	20.5	259	180	
Cr	369	272	349		7	465	389	713	382	477	304	479
Ni		86	93				116			150	99	
Cu		22	42				81			69	28	
Zn		108	108				83			88	74	
Ga	72	21	20	20	20	20	17	24	າາ	20	21	26
Rb Sr	1494	21 461	30 531	38 640	39 608	20 552	21 657	24 667	32 867	20.5 477	22.9 585	
Y	49	24	33	23	40.9	24	26.8	25	30	23.9	21.8	
Zr	500	172	200	194	225	158	217	204	236	167	160	
Nb	183.2	28	52	70.6	42.4	40.3	57.6	67.5	85.3	41.3	40.1	67.7
Cs					0.31					0.15	0.12	
Ba	830	229	337	431	256	251	273	329	399	236	252	
La Ce	125.1	24.36 50.9	32.71 66.81	40.5 86	32.77 69.05	29.3 65	39.2 78.0	39.8 81	51.8 109	29.8 61.1	29 58.3	43.60 82.99
Pr	28	6.39	8	10.2	9.32	8	76.0	10	13	7.54	7.01	10.90
Nd	104.3	25.92	32.01	39	36.09	33	40.0	40.3	51.2	30.3	27.8	
Sm	18.9	5.47	7.08	7.3	7.59	6.9		7.7	9.9	6.6	5.81	
Eu	5.7	1.77	2.13	2.5	2.46	2.4		2.6	3.2	2.23	1.95	
Gd	17.7	5.09	6.76	7.7		7.3		8.3	10	6.16	5.36	
Tb	2.24	0.82	1 5.67	1.01	1.16 6.34	1.01 5.5		1.06 5.5	1.25	0.94	0.82	1.10 5.19
Dy Ho	11.8 1.93	4.54 0.8	5.67 1.1	5.1 0.89	1.16	0.99		1.01	6.5 1.18	5.22 0.98	4.67 0.87	0.90
Er	4.96	2.07	3.06	2.25	3.38	2.58		2.58	3.12	2.43	2.22	
Tm	0.61	0.29	0.4	0.29		0.35		0.33	0.39	0.32	0.29	
Yb	3.76	1.73	2.55	1.87	2.77	2.05		2.1	2.4	1.85	1.74	
Lu	0.57	0.37	0.37	0.26	0.27	0.3		0.29	0.33	0.27	0.27	0.26
Hf T-	12.8			4.62	4.86	4.69		4.99	5.6	4.32	3.99	7.48
Ta	10.25			4.17	2.70	2.57	1 1	3.91	4.78	2.67	2.56	
Pb Th	5 16.65			2.2 5.89	2.01 5.01	1.7 3.43	1.4 5.0	2.1 5.22	2.43 6.05	1.87 3.78	2.13 4.02	2.45 5.07
U	4.93			1.63	0.91	1.02	2.6	1.46	1.82	0.96	0.92	
Radiogenic isotop		lues)					5			2.00	3. <b>3.2</b>	
<sup>87</sup> Sr/ <sup>86</sup> Sr	0.70284	0.705035	0.703786	0.70316	0.703424	0.703357			0.702854			0.703458
<sup>143</sup> Nd/ <sup>144</sup> Nd	0.512979	0.512748	0.512824	0.512876	0.512883	0.512902						0.512891
<sup>206</sup> Pb/ <sup>204</sup> Pb	19.5683	19.222	19.32	19.6465	19.703	19.6407			19.7314			19.363
<sup>207</sup> Pb/ <sup>204</sup> Pb	15.5399	15.625	15.62	15.5665	15.599	15.5169			15.5955			15.615
<sup>208</sup> Pb/ <sup>204</sup> Pb	38.9781	39.327	39.27	39.2801	39.387	39.2011			39.2857			39.202

Samples numbers in large-font/bold = this study. PRR = samples on loan from the Polar Rock Respository (http://research.bpcrc.osu.edu/rr/). Rock name based on total alkali vs. silica diagrame (TAS) shown in Fig. 2.  $FeO_t$  is total iron expressed as FeO, normalized to 100%. LOI = Loss on Ignition. Mg # =  $100*Mg/(Mg + Fe^{2+})$ . Major elements (wt.%) by XRF and trace elements (ppm) by XRF and ICP-MS. CIPW normative (wt.%) were calculated following Irving and Baragar (1971)

Samples SAX20, MP24, MP8, MP34, MP32 are from Nardini et al. (2009). Samples MA009a and MA-117 are from Rocholl et al. (1995) with Sr, Nd and Pb isotopes from this study are italicized. Samples P74794 and P74833 are from Mortimer et al. (2007). Samples NV-6E, NV-21, NV-4B and NV-3B from Aviado et al. (2015). Samples A223, A210B, A240B-1, A232B, A225A, were collected and described by Hamilton (1972) and re-analyzed in this study.

Table 2: Continued

Section   1,00	Sample # PRR - #	A240B-1	A232B	NV-6E	VC-21	A227B	AW82205	<b>HC-FI-2A</b> 3842	<b>NV-4</b> 5169	<b>NV-4C</b> 5171	NV-4B	A225A	HC-SPI-2 3872
Part													
Magnifications (seed as all part of the content o		Cape Adare					Foyn Island	Foyn Island	Foyn Island	Foyn Island	Foyn Island	Foyn Island	
SQ	Rock name (TAS)	BAS	HAW	BAS	BAS	HAW	AKB	AKB	AKB	HAW	AKB	BAS	BAS
Total	Major elements (	wt.%)											
Map													
FeO													
Mor													
Mage	-												
Name													
Mathematical   Math	CaO	11.95	9.84	10.98	11.14	9.92	10.71	10.62	10.65	10.55	10.89	10.58	12.79
P.O.   0.08													
Column   C													
Total			0.73	0.74	0.70	0.60	0.61						
Mg			99.99	100.00	100.00	100.00	100.00						
Ne 18.5													
Control   Cont	CIPW norm miner	rals (wt.%)											
Page		15.9	8.0	15.0	12.7	11.4	9.1	8.9	8.7	11.3	9.0	12.3	
Trace elements   Frame		28.6	19.4	26.5	26.9	21.6	22.7	22.5	22.8	22.7	25.0	22.9	31.9
SC         30         25         L         22         25         25         26         24         27         27         27         28         29         235         209         235         255         209         253         255         209         253         258         255         27         280         28         21         19         91         21         11         11         11         11         11         11         11         11         11         11         12         11			18.4	21.8	16.9	16.8	20.7	21.3	21.0	17.8	24.1	17.4	18.4
V         312         2255         2255         240         278         278         251         250         253         209         263         275           Cr         305         260         225         218         230         431         367         388         269         395         238         332           Ni         115         99         435         373         276         100         55         55         55         401         58         65           Cu         72         275         226         206         95         110         75         373         66         24         171         33         16         36         22         20         21         21         20           Rb         32         32         17         75         375         62         24         171         33         16         36         30         30         30         30         275         262         262         260         73         33         33         33         33         33         33         33         33         33         33         33         33         33         33         33         33<	• •	• •	25			22	25	25	25	2/		23	27
Cr         305         260         225         218         230         431         367         388         269         395         228         323           Cu         772         577         226         206				235	240	22							
Cu         72         75         226         206         ————————————————————————————————————						230							
Zn         95         100         75         80         91         91         97         92         96         78         94         100           Ga         21         21         21         75         37         21         19         19         21         21         21         20           Rb         322         32         21         75         37         26         24         17         33         16         38         44           Sr         800         787         800         695         812         684         672         642         761         592         831         103           Y         350         350         303         30         262         262         290         23         280         331         103         37         260         203         280         331         103         379         263         280         381         311         300         379         269         396         66         68         48         1520         484         1520         484         1820         491         497         481         491         491         491         491         491													
Ga         211         211         211         19         19         21         21         20           Rb         322         232         21         75         37         26         24         17         33         16         36         44           r         800         787         800         695         812         648         672         642         761         592         331         1034           Y         35.0         35.0         35.0         30         30         30         27.5         62.2         29.0         27         33.0         33.7           Nb         83.0         70.0         58         68         71.0         64.1         62.1         77.8         53         82.0         110.6           Cs         0.40         0.69         58         40         22.5         311         300         379         503         38         32.0         10.7           Ba         42.2         61.0         323         449         450         48.2         31.0         30.0         379         50.3         39         54.0         73.0         33.1         32.0         43.2         31.0 <t< td=""><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td><td></td></t<>													
Rb         32         32         32         26         24         17         33         16         36         44           Y         35.0         35.0         30         695         812         684         672         626         29.0         22         33.0         33.3           Y         35.0         326         330         30         223         224         229         29.0         29.0         28.0         33.0           Rb         263         376         225         311         223         224         227         269         290         280         280         331           Cs         0.40         0.69				/5	80								
Sr         800         787         800         695         812         684         672         642         761         592         831         103           Y         350         350         350         330         330         275         262         290         27         330         33.7           Y         263         276         225         311         225         321         269         282         320         170           S         0.40         300         58         68         710         64.1         62.1         77.8         82.0         1016           S         0.40         606         323         449         640         24.2         43.2         37.9         50.3         39         54.20         73.9           C         103.6         15.8         49         640         42.20         43.2         37.9         50.3         39         54.20         73.9           C         103.6         45.9         44         41.0         10.6         40.1         40.8         45.4         46.4         61.3           S         9.95         9.6         44         51.0         12.1         21.1				21	75	37							
Zr         263         276         225         311         223         224         227         269         203         280         331           Nb         83.0         70.0         58         68         77.0         64.1         62.1         77.8         53         82.0         1016           Cs         0.40         0.69													
Nb         83.0         70.0         58         68         −         71.0         64.1         62.1         77.8         53         82.0         103.0           Cs         0.40         0.69         -         7.0         0.29         -         0.33         26.0         33.0         26.9         39.6         35.6         35.0         35.0         51.80         43         49         64.0         42.20         43.2         37.9         50.3         39         54.20         73.9           Ce         103.0         105.20         78         84         125.20         85.0         84.6         80.2         101.4         70         107.80         73.9           Nd         50.00         45.90         44         44         51.10         40.50         40.1         40.8         45.8         45.4         46.0         61.3           Sm         9.95         9.66         9.7         9.1         9.97         8.18         7.0         40.8         45.4         40.8         45.1         40.8         45.2         40.2         40.2         40.2         40.2         40.2         40.2         40.2         40.2         40.2         40.2         40.2													
CS         0.40         0.69													
Ba				58	80	0.47		64.1	62.1	//.8	53		107.6
Ce Pr         103.60 10.3         178.9 10.3         84         125.20 10.6         85.00 4.0         84.6 4.0         88.2 2.0         101.4 2.0         70         107.80 10.8         138.9 2.0           Nd         50.00         45.90 9.95         44.4 9.9         44.5 9.9         40.5 8.18         40.1 2.58         45.4 4.0         45.4 4.0         45.4 9.1         40.8 9.9         45.4 9.9         45.8 8.18         46.4 9.9         46.1 4.0         40.5 9.9         40.8 8.18         45.4 4.0         45.4 9.1         40.8 9.1         45.4 9.1         45.4				323	419			311	300	379	269		564
Pr         10.0         45.90         44         44         51.10         40.50         40.1         40.8         45.9         46         46.40         61.3           Sm         9.95         9.66         9.7         9.1         9.97         8.18         -         -         7.6         9.18         -         -         6.0         9.18         -         -         -         9.18         -         -         -         -         9.18         -         -         -         -         -         9.18         -	La	52.50	51.80	43	49	64.00	42.20	43.2	37.9	50.3	39	54.20	73.9
Ndd         50.00         45.90         44         44         51.10         40.50         40.1         40.8         45.4         36         46.40         61.33           Sm         9.95         9.66         9.7         9.1         9.97         8.18         7.6         9.18         7.6         9.18           Eu         3.04         3.44         2.9         2.8         3.11         2.58         2.5         2.3         2.82         2.82         2.82         2.83         2.82         2.83         2.82         2.83         2.82         2.83         2.82         2.83         2.82         2.83         2.82         2.83         2.82         2.83         2.82         2.83         2.83         2.82         2.83		103.60	105.20			125.20	85.00	84.6	80.2	101.4		107.80	138.9
Sm         9.95         9.66         9.7         9.1         9.97         8.18          7.6         9.18          2.8         3.11         2.58          2.3         2.82          2.8         3.11         2.58          7.6         9.18          2.82         2.82          2.8		50.00	45.90			51.10	40.50	40.1	40.8	45.4		46.40	61.3
Gdd         9         9         1.12         1.13         1.22         1.23         1.24         1.24         1.24         1.24         1.24         1.24         1.26         1.21<													
Tb         1.21         1.13         1.2         1.2         1.16         1.01         1.01         1.08         1.08         1.01         1.00         1		3.04	3.44			3.11	2.58					2.82	
Dy		4.24	4.42			1.16	1.01					1.00	
Ho		1.21	1.13				1.01					1.08	
Fr													
Yb         2.32         2.33         2.4         2.63         2.05         2.1         2.19         2.11         3.1         4.6         6.4         9.1         2.19         2.11         3.1         4.1         2.00         2.20         2.01         2.00         2.1         2.00													
Lu         0.33         0.30         0.3         0.4         0.35         0.26           Hf         6.73         6.44         5.6         5.3         7.64         5.41         4.6         6.47           Ta         5.25         4.24         4.4         5.1         6.03         4.51         5.1         3.8         5.19           Pb         2.00         5.00         1.7         1.9         4.30         2.00         0.9         1.1         3.1         1         2.00         2.9           Th         6.11         5.53         4.6         6.4         7.77         4.37         5.3         5.5         6.6         4.6         6.64         9.1           U         1.70         2.00         1         2.3         2.11         0.90         1.2         0.6         2.2         0.9         2.04         2.6           **Sr/**Sr/**Ssr         0.702944         0.703741         0.702909         0.702900         0.702780         0.512000         0.702780         0.512000         0.512000         0.512000         0.512000         0.512000         0.512000         0.512000         0.512000         0.512000         0.512000         0.512000         0.512000													
Hf 6.73 6.44 5.6 5.3 7.64 5.41  Ta 5.25 4.24 4.4 5.1 6.03 4.51  Pb 2.00 5.00 1.7 1.9 4.30 2.00 0.9 1.1 3.1 1 2.00 2.9  Th 6.11 5.53 4.6 6.4 7.77 4.37 5.3 5.5 6.6 4.6 4.6 6.49  U 1.70 2.00 1 2.0 1 2.3 2.11 0.90 1.2 0.6 2.2 0.9 2.04 2.6  Radiogenic isotopes (measured values)  **S**S***S***S****S*****S*****O.702944 0.703741													
Ta 5.25 4.24 4.4 5.1 6.03 4.51   Pb 2.00 5.00 1.7 1.9 4.30 2.00 0.9 1.1 3.1 1 2.00 2.9   Th 6.11 5.53 4.6 6.4 7.77 4.37 5.3 5.5 6.6 4.6 6.64 9.1   U 1.70 2.00 1 2.3 2.11 0.90 1.2 0.6 2.2 0.9 2.04 2.6   Radiogenic isotopes (measured values)  **S7sr/*6Sr 0.702944 0.703741 0.702900 0.702900 0.702780   **S8T 0.702944 0.512930 0.512874 0.512937 0.512937 0.512945 0.513000   **TH 0.702970 0.702870 0.702870 0.702870 0.702900 0.702800 0.702780   **TH 0.702970 0.702870 0.702870 0.702870 0.512937 0.512937 0.513000   **TH 0.702970 0.702870 0.702870 0.702870 0.512937 0.512937 0.513000 0.512930 0.512874 0.512930 0.512874 0.512937 0.512937 0.513000 0.512930 0.512874 0.512937 0.512937 0.513000 0.512937													
Pb         2.00         5.00         1.7         1.9         4.30         2.00         0.9         1.1         3.1         1         2.00         2.9           Th         6.11         5.53         4.6         6.4         7.77         4.37         5.3         5.5         6.6         4.6         6.64         9.1           U         1.70         2.00         1         2.3         2.11         0.90         1.2         0.6         2.2         0.9         2.04         2.6           Radiogenic isotopes (measured values)           87Sr/8Sr         0.702944         0.703741         0.702900         0.702900         0.702780         0.702780         0.512935         0.512935         0.512935         0.513000           143Nd/144Nd         0.512930         0.512874         0.512937         0.512945         0.513000         0.512935         0.512935         0.513000           206Pb/204Pb         20.026         19.432         20.180         20.184         0.512936         0.512936         0.512937         0.512936         0.512936         0.512936         0.512936         0.512936         0.512936         0.512936         0.512936         0.512936         0.512936         0.512936         0.51													
U       1.70       2.00       1       2.3       2.11       0.90       1.2       0.6       2.2       0.9       2.04       2.6         Radiogenic isotopes (measured values)       87 Sr/86 Sr       0.702944       0.703741       0.702900       0.702900       0.702780       0.702780       0.702970       0.702970       0.702870         143 Nd/144 Nd       0.512930       0.512874       0.512937       0.512945       0.513000       0.512935       0.513000         206 Pb/204 Pb       20.026       19.432       20.180       20.184								0.9	1.1	3.1			2.9
Radiogenic isotopes (measured values)         87 Sr/86 Sr       0.702944       0.703741       0.702900       0.702900       0.702780       0.702970       0.702970       0.702870         143 Nd/144 Nd       0.512930       0.512874       0.512937       0.512945       0.513000       0.512935       0.513000         206 Pb/204 Pb       20.026       19.432       20.180       20.184       20.044       20.044       20.044       15.587       15.587       15.587       15.587       15.680       15.653       15.653       15.680       15.653       15.653       15.680       15.653       15.653       15.680       15.653       15.653       15.680       15.680       15.653       15.680	Th	6.11	5.53	4.6	6.4	7.77	4.37	5.3	5.5	6.6	4.6	6.64	9.1
87 Sr/86 Sr       0.702944       0.703741       0.702900       0.702900       0.702780       0.702970       0.702970       0.702870         143 Nd/144 Nd       0.512930       0.512874       0.512937       0.512945       0.513000       0.512935       0.513000         206 Pb/204 Pb       20.026       19.432       20.180       20.184       20.044       20.044       20.044       15.587         208 204 204       15.633       15.642       15.680       15.653       15.653       15.587				1	2.3	2.11	0.90	1.2	0.6	2.2	0.9	2.04	2.6
143 Nd/144 Nd       0.512930       0.512874       0.512937       0.512945       0.513000       0.512935       0.513000         206 Pb/204 Pb       20.026       19.432       20.180       20.184       20.184       20.044       20.044         207 Pb/204 Pb       15.633       15.642       15.680       15.653       15.653       15.587	•	•	•			0.702000	0.702000	0.702700				0.702070	0.702070
206 Pb/204 Pb       20.026       19.432       20.180       20.184       20.044         207 Pb/204 Pb       15.633       15.642       15.680       15.653       15.587													
<sup>207</sup> Pb/ <sup>204</sup> Pb 15.633 15.642 15.680 15.653 15.587	•							0.513000					0.513000
200 204	•												

Table 2: Continued

Sample # PRR - #	AW82214	NV-3B	D2-1	D4-1	D4-3	D9-1	D12-1
dec. Lat. S dec. Long. E	-71.900 171.167	-71.900 171.200	-69.861 171.844	-70.070 172.316	-70.070 172.316	-71.628 172.523	-71.631 172.634
Geographical location	Possession Island	Possession Island	Adare trough, W. scarp	Adare trough, center	Adare trough, center	S. Adare Basin	S. Adare Basin
Rock name (TAS)	BAS	BAS	HAW	BAS	BAS	TEPH	BAS
Major elements (	wt.%)						
SiO <sub>2</sub>	41.64	44.78	46.75	43.07	41.98	45.26	44.60
TiO <sub>2</sub>	3.47	3.60	2.63	2.99	4.75	3.18	3.06
Al <sub>2</sub> O <sub>3</sub>	13.07 12.23	15.82 12.93	17.18 7.85	12.92 10.20	13.59 13.35	15.18 11.48	13.34 11.47
FeO <sub>t</sub> MnO	0.22	0.22	0.15	0.20	0.27	0.26	0.21
MgO	9.78	6.45	8.83	13.32	9.86	6.39	10.70
CaO	12.60	10.37	10.43	10.84	10.41	9.42	10.02
Na <sub>2</sub> O	4.29	3.65	3.43	3.64	3.15	5.58	4.12
K2O	1.52	1.21	1.73	1.88	1.74	2.16	1.54
P <sub>2</sub> O <sub>5</sub>	1.17	0.96	1.01	0.94	0.90	1.09	0.93
LOI Total	99.99	100.00	3.68 99.99	1.79 100.00	1.22 100.00	1.73 100.00	1.17 99.99
Mg #	58.8	47.1	66.7	70.0	56.8	49.8	62.5
CIPW norm miner				, 6.6	55.5	.5.6	02.0
Ne	19.7	8.5	7.4	16.7	13.0	21.1	14.8
Lc	7.0			4.4			
Di	32.1	18.3	15.2	27.6	22.9	24.5	24.7
Hy Ol	18.8	18.6	18.1	24.6	22.3	14.8	22.6
Trace elements (p							
Sc	27		25	27	30	16	22
V	296	215	197	233	325	185	214
Cr	375	35	277	540	244	135	355
Ni Cu	167 30	296 101	162 40	338 46	139 42	71 25	205 46
Zn	101		71	81	115	136	102
Ga	21		17	16	23	21	19
Rb	47		34	31	37	40	35
Sr	1034	925	901	861	1131	1216	1007
Y Zr	38.0 332	28 217		26.5 292	35.5 394	33.7 553	30.3 347
Nb	118.0	61	75.1	93.6	96.2	155.0	89.4
Cs	0.54	01	, 5.12	33.0	30.2	133.0	03
Ва	626	371	604	467	595	558	401
La	73.80	47		51.01	70.8	88.58	56.48
Ce	142.60	83		95.53	152.4	167.12	111.97
Pr Nd	61.50	10.7 46		10.29 40.58	69.4	17.06 63.93	12.05 48.43
Sm	12.01	9.6		7.23	09.4	10.77	9.11
Eu	3.68	3.1		2.45		3.35	2.90
Gd		9		14.75		24.44	18.03
Tb	1.39	1.2		1.11		1.45	1.30
Dy		5.6	6.10	4.90		6.38	5.74
Ho Er		0.9 2.4		0.90 1.91		1.14 2.50	1.03 2.14
Tm		۷.٦	0.42	0.31		0.41	0.34
Yb	2.37	2.2		1.72		2.35	1.94
Lu	0.31	0.3	0.39	0.28		0.34	0.28
Hf -	7.57	5.2					
Ta	6.89	4.7		2.52	12.0	2.44	2.04
Pb Th	2.00 8.33	1.6 5.5	2.83 7.26	2.52 5.33	13.0 9.0	3.44 6.77	2.01 4.81
U	2.80	3.3 1	1.75	1.41	4.7	2.20	1.33
Radiogenic isotop	es (measured	values)					
<sup>143</sup> Nd/ <sup>144</sup> Nd	0.702884	0.70343 0.51282		0.702870 0.513000		0.702900	0.702779 0.513001
206 Pb/ <sup>204</sup> Pb	0.512945 19.851	20.028		19.237		0.512960 20.234	20.111
<sup>207</sup> Pb/ <sup>204</sup> Pb	15.620	15.658		15.583		15.653	15.625
<sup>208</sup> Pb/ <sup>204</sup> Pb	39.354	39.916		38.812		39.715	39.551

Table 3. Oxygen isotope data from olivine phenocrysts.

2		_ h	SIMS				Core*				LF		MgO (wr)
Sample <sup>a</sup>	Location	Fo <sup>b</sup>	δ <sup>18</sup> O ‰	2SD	SEM	$n_a$	δ <sup>18</sup> O ‰	2SD	SEM	$\mathbf{n}_{a}$	δ <sup>18</sup> O ‰	2SD	wt.%
HC-DP-3 (3839)	Daniel Pen.	84 ± 2	4.89	0.22	0.03	5							9.11
HC-FI-2A (3842)	Foyn Is.	81 ± 5	5.06	0.23	0.08	7					4.88	0.20	9.33
HC-SPI-2 (3872)	Possession Is.	75 ± 9	5.23	0.28	0.07	7					4.90	0.12	9.56
HC-DP-1 (3873)	Daniel Pen.	81 ± 5	5.27	0.24	0.07	6					4.90		4.51
NV-4 (5169)	Foyn Is.	83 ± 2	4.82	0.18	0.07	7	4.87	0.27	0.06	7	4.86	0.22	9.45
NV-4C (5171)	Foyn Is.	82 ± 5	4.71	0.22	0.06	7	4.62	0.21	0.05	6			7.84
A214B	W. of Daniel P.	76 ± 7	5.10	0.16	0.05	4							4.10
A223	W. of Daniel P.	87 ± 3	5.25	0.23	0.12	5					5.03		9.49
A225A	Foyn Is.	81 ± 3	5.15	0.24	0.08	8	5.56	0.15	0.03	6			7.74
A227B	McCormick Is.	82 ± 5	5.08	0.23	0.04	9							7.36
A232B	Adare Pen.	$81 \pm 4$	5.30	0.20	0.08	6					5.06		7.14
A240B-1	Cape Adare	82 ± 5	5.20	0.21	0.07	6							8.72
D12-1	S. Adare Basin	84 ± 2	5.35	0.39	0.07	9							10.70
D2-1	Adare trough	87 ± 1	5.25	0.41	0.05	4							8.83
D4-1	Adare trough	90 ± 1	5.04	0.13	0.10	5					5.25		13.32
D4-3	Adare trough	79 ± 1	5.43	0.23	0.11	4	5.50	0.26	0.05	7	5.17	0.14	9.86
D9-1	S. Adare Basin	79 ± 1	5.05	0.39	0.10	5							6.39
MA-009a	Malta Plateau	78 ± 7	5.44	0.21	0.08	8	5.73	0.26	0.06	6			9.35
MA-117	Malta Plateau	79 ± 7	5.39	0.23	0.11	7	5.78	0.18	0.04	6	5.26	0.12	8.49
P74794	Robertson Bay	80 ± 7	5.33	0.41	0.03	6							9.41
P74833	Robertson Bay	70 ± 7	5.15	0.24	0.11	7	5.40	0.13	0.03	6			8.04

<sup>&</sup>lt;sup>a</sup> numbers in parenthesis are PPR- # of samples on loan from Polar Rock Repository (http://research.bpcrc.osu.edu/rr/).

Core\* is the average calculated for multiple SIMS spot analyses (n<sub>a</sub>) within the core of a single olivine grain.

Laser Fluorination (LF) data are based on 4-5 olivine grains. Standard deviation precision (2SD) based

on repeat analysis. Precision on standard UWG-2 (Core Mountain Garnet) is  $\pm$  0.10% (2 $\sigma$ )

SEM = Standard Error of Mean. (wr) = whole rock MgO content (wt.%).

<sup>&</sup>lt;sup>b</sup> Fo (%) =  $(Mg/(Mg+Fe^{2+}))*100$  with  $\pm 1\sigma$ .

SIMS data are averages of multiple spot analyses ( $n_a$ ) on 2-3 grains. Precision to 2 standard deviations (2SD) on standard San Carlos olivine is  $\pm 0.28\%$ 

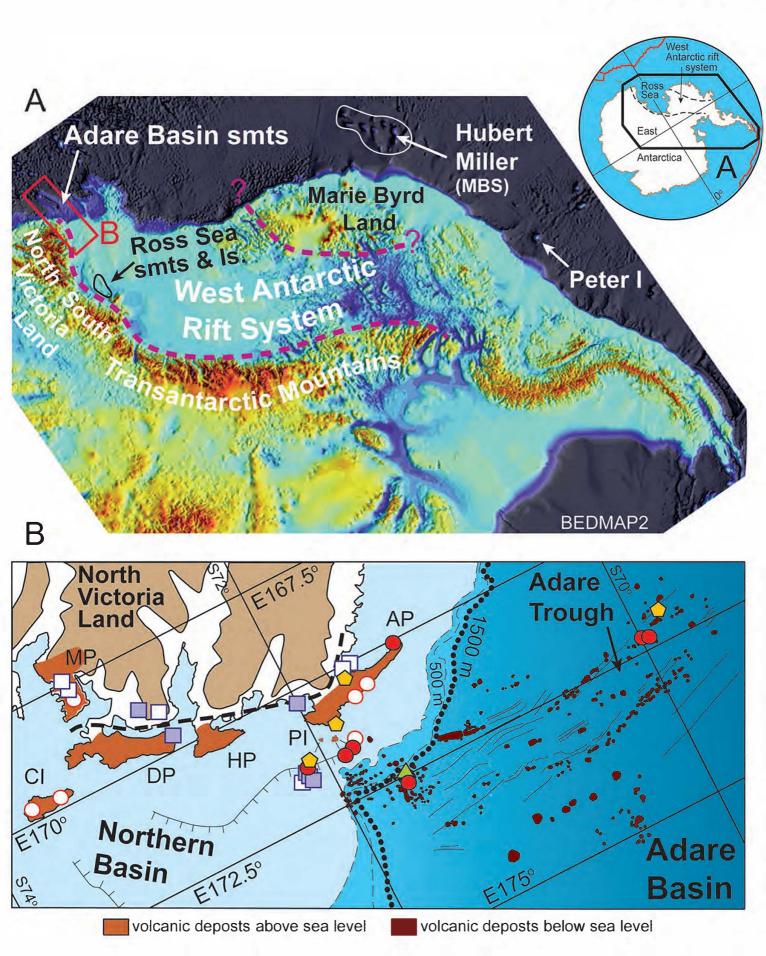


Fig. 1

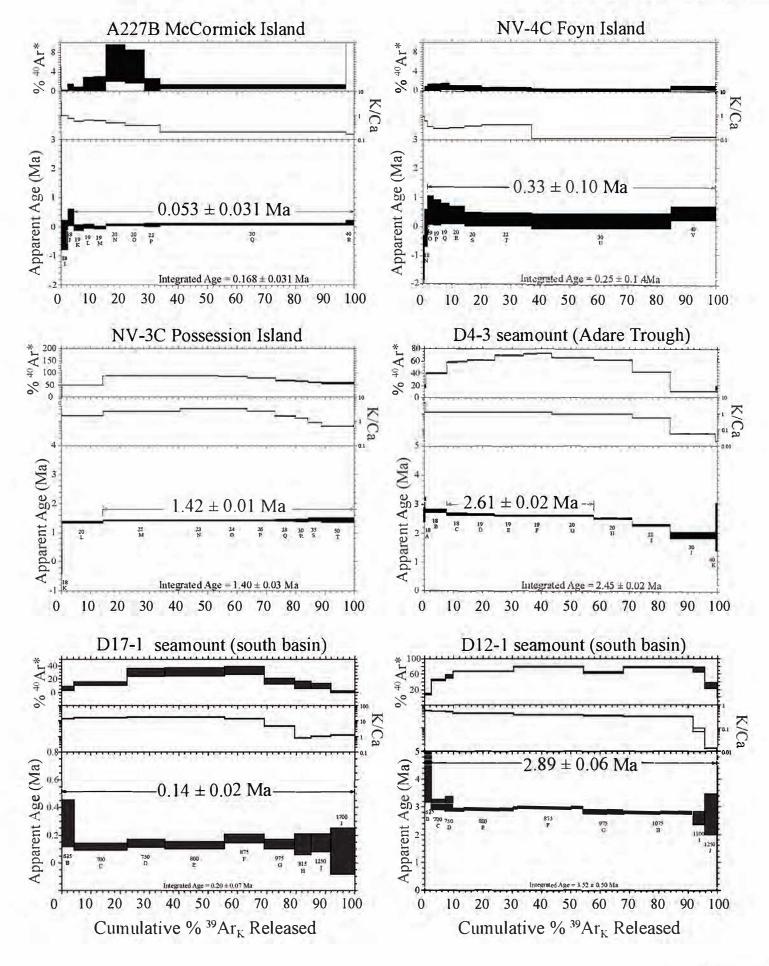
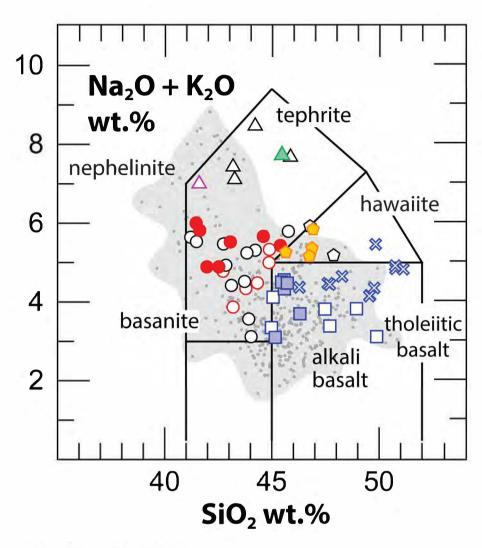


Fig. 2



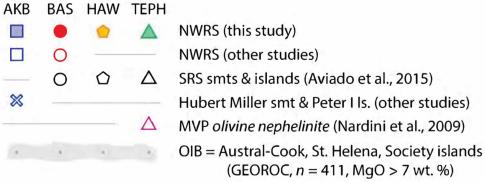


Fig. 3

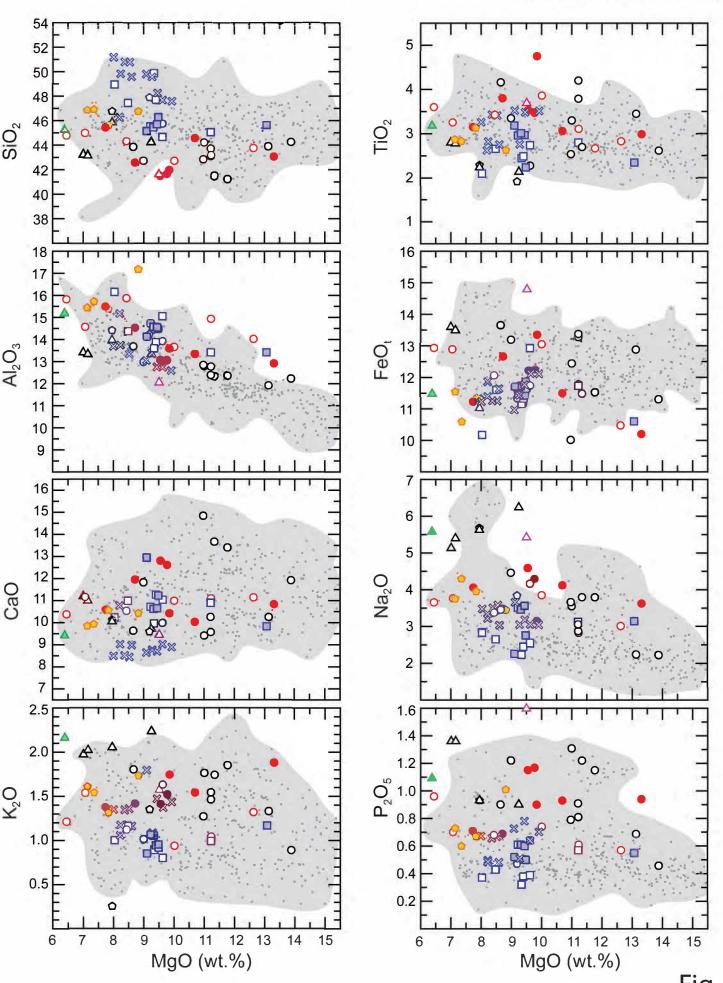


Fig. 4

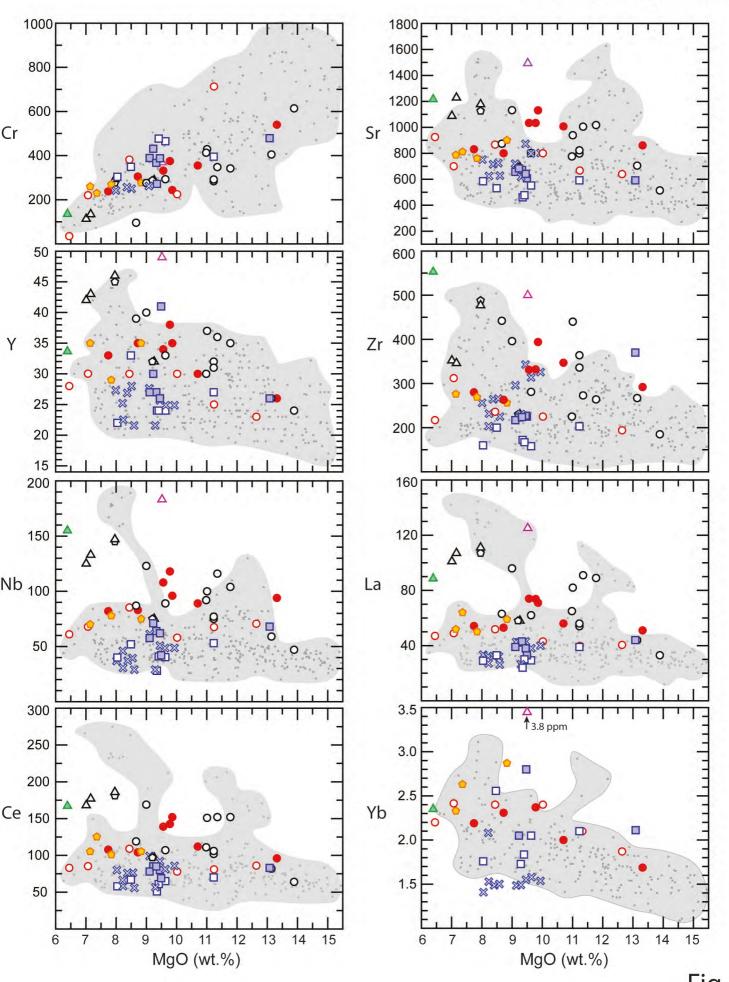


Fig. 5

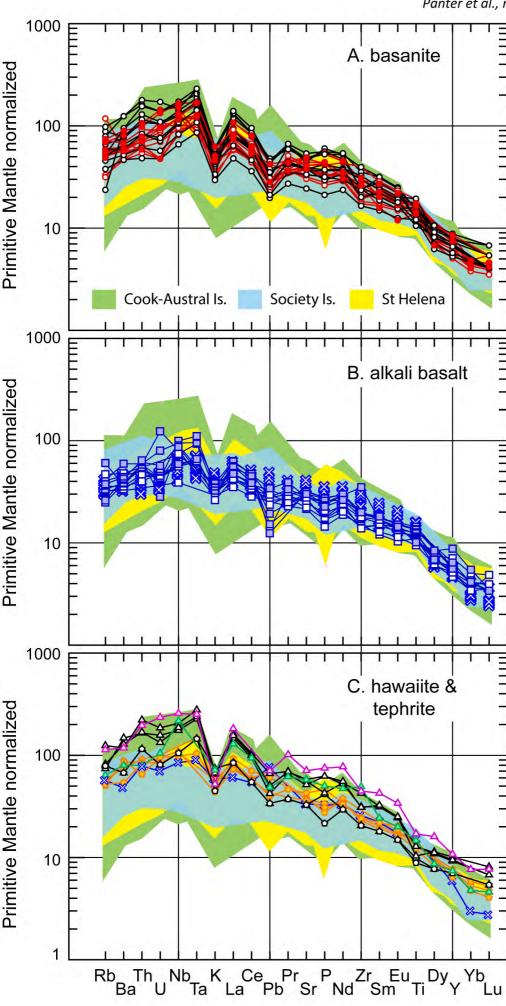


Fig. 6

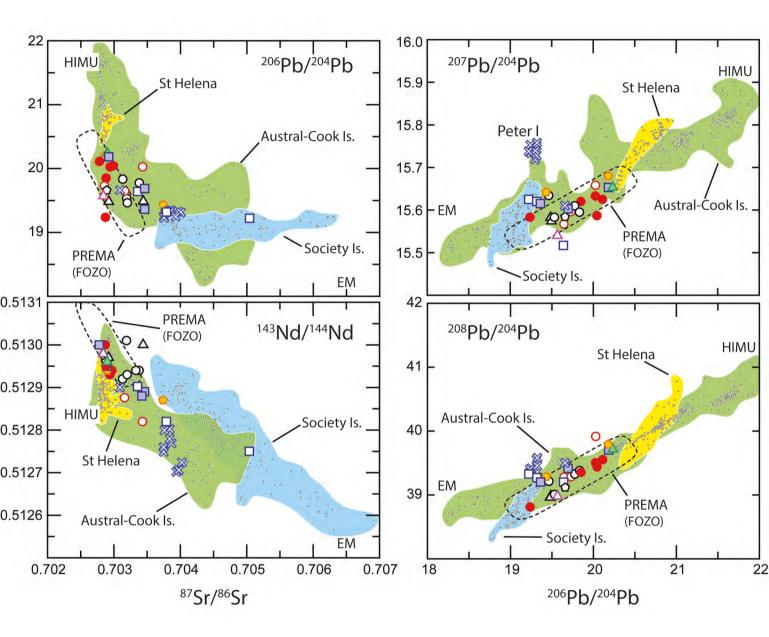


Fig. 7

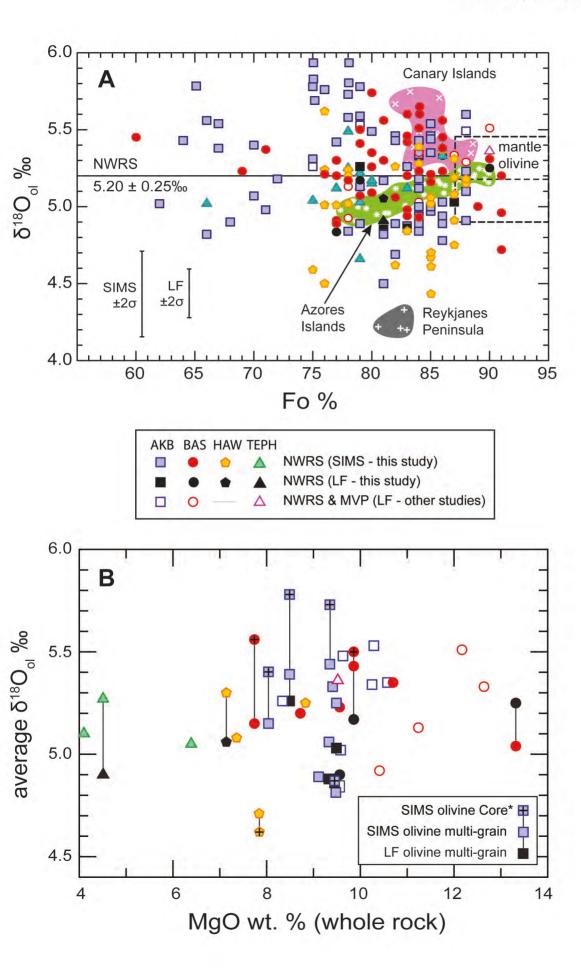


Fig. 8

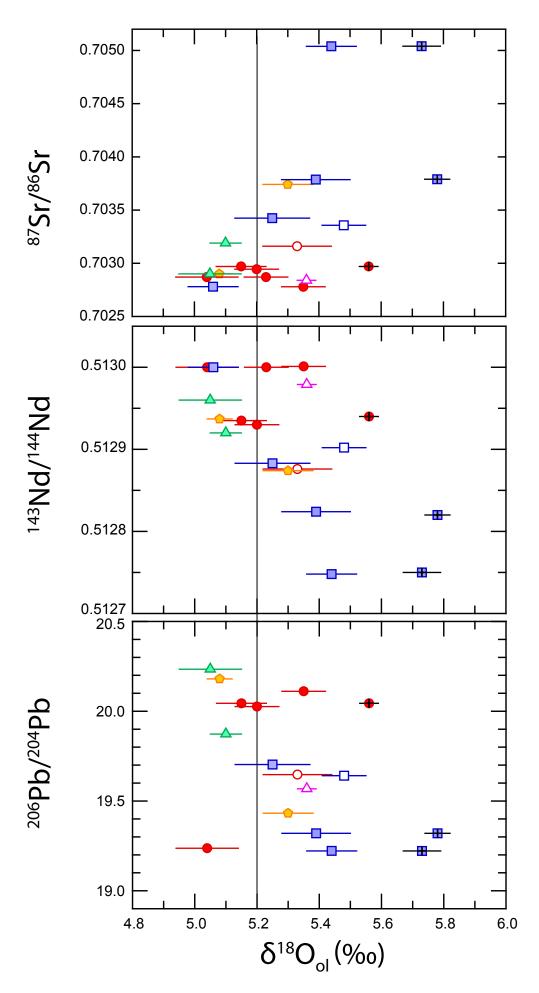


Fig. 9

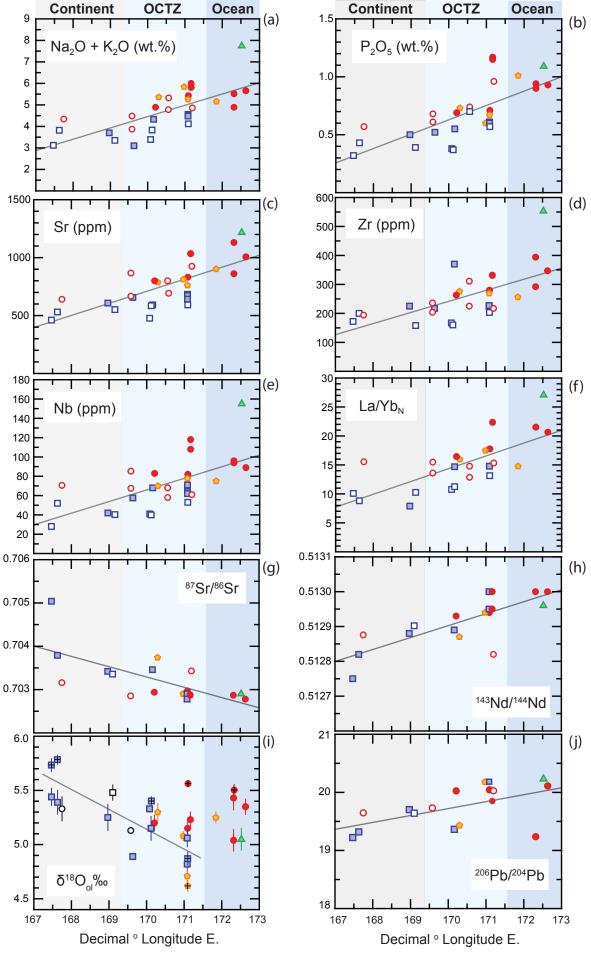


Fig. 10

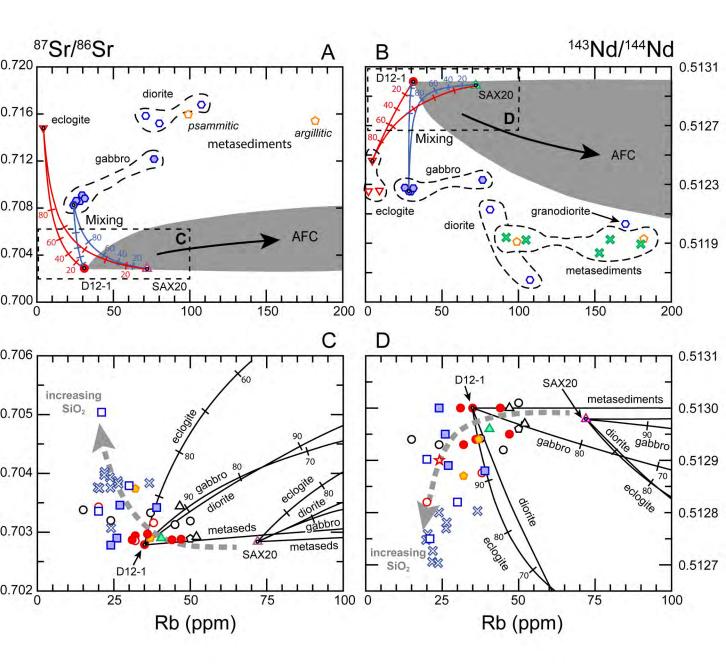


Fig. 11

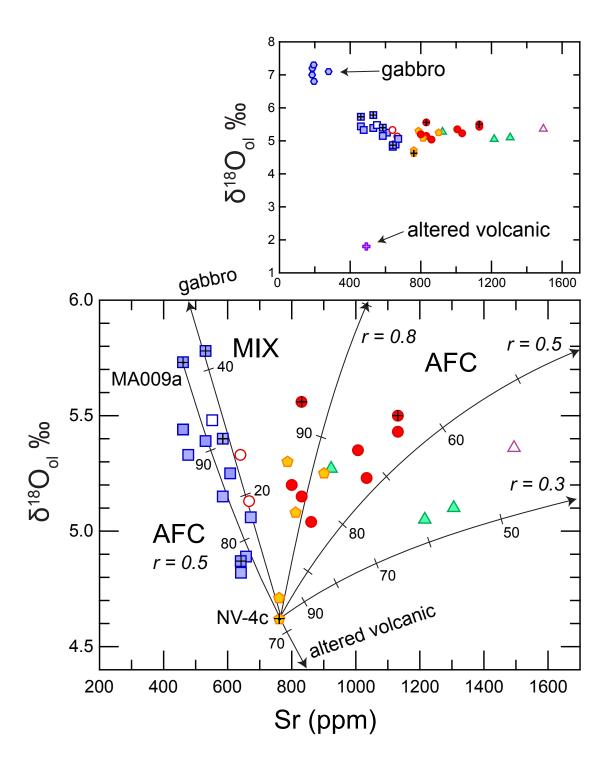


Fig. 12

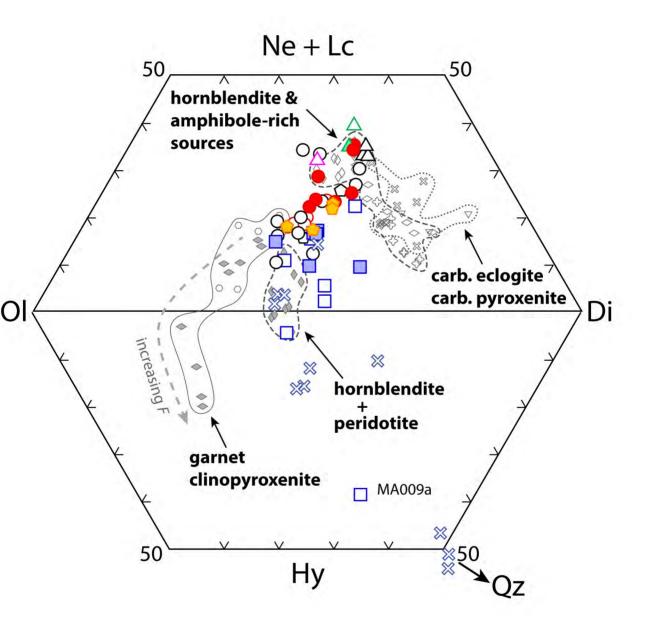


Fig. 13

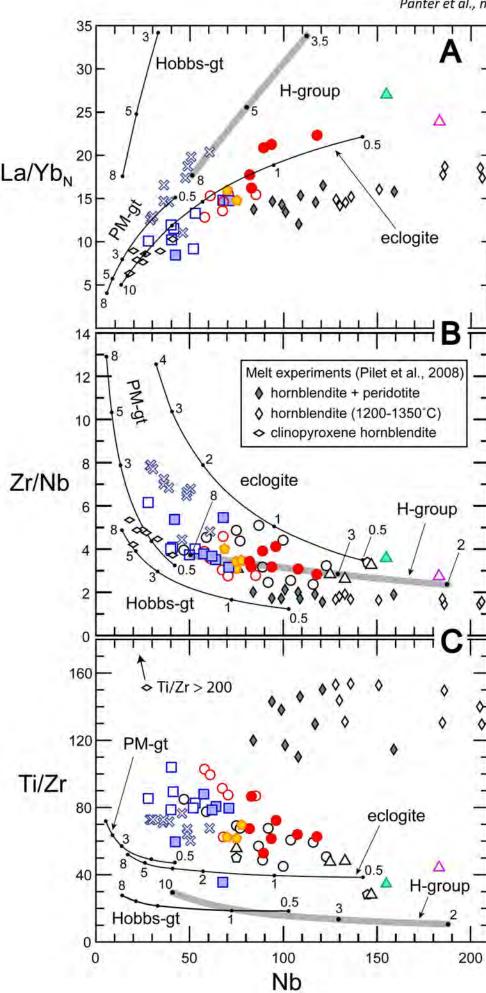


Fig. 14

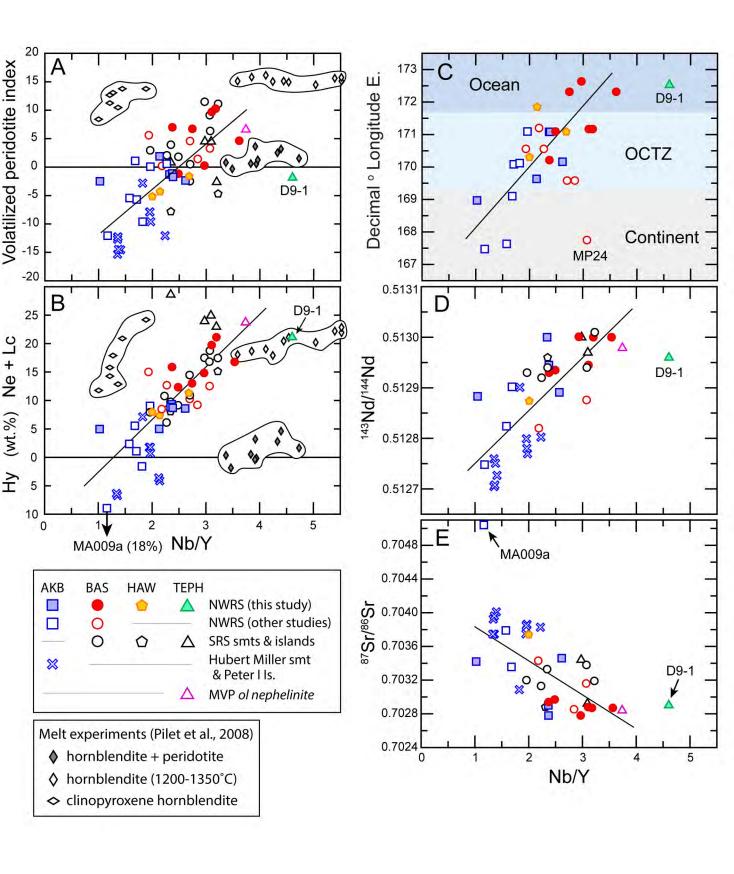


Fig. 15

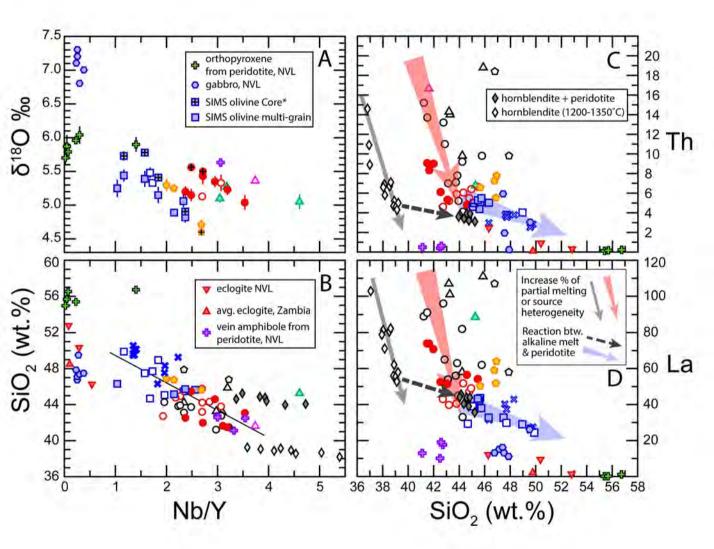


Fig. 16

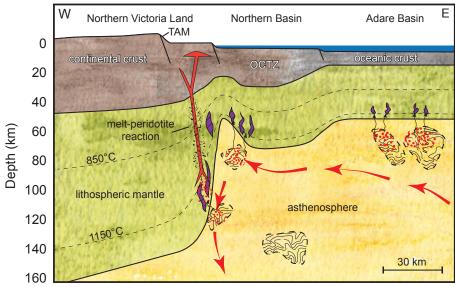
#### A. Pre-break-up subduction (> 100 Ma) W Ε arc volcanism < 0 oceanic crust continental crust Depth (km) lithospheric mantle 50 asthenosphere 100 continental lithospheric mantle flux melting from dehydrating slab 150 change in scale 200 carbonate-rich recycled (G) slab materia detached slab 300 km upwelling / mantle 500 B. Initiation of WARS extension, phase I (100 to 80 Ma) W Ε 0 West Antarctica East Antarctica continental crust 50 Depth (km) metasomes lithospheric mantle 100 850°C 1150°C 150 low degree melting of carbonate-rich asthenosphere recycled slab material 200 300 km 250 C. WARS extension phase II (80 to 40 Ma) Meander Intrusive group 0 East Antarctica continental crust West Antarctica 50 lithospheric mantle Depth (km) high degree melting of metasomes 100 850°C 150 low degree melting of carbonate-rich recycled slab material asthenosphere 200

250

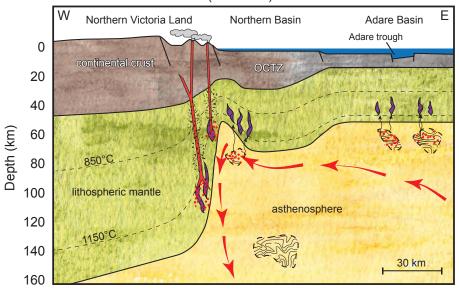
Fig. 17

300 km

### D. Adare/Northern Basin extension (43 to 26 Ma)



#### E. Post Adare Extension I (26 - 5 Ma)



#### F. Post Adare Extension II (< 5 Ma)

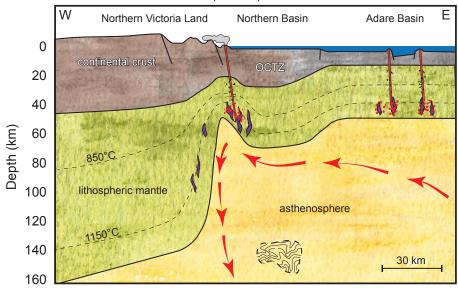


Fig. 17

# NBP0701 Dredge Log

Dredge #	Date	Time on bottom	Start		E	End	Time off bottom	Start Dredge	End Dredge	$\Delta$ (m)	Geographic Location	
Dieage #	Date	(GMT)	Latitude S	Longitude E	Latitude S	Longitude E	(GMT)	Depth (m)	Depth (m)	Δ (III)	Geographic Education	
D01	3-Jan-07	11:26	69 50.39'	171 50.92'	69 50.40'	171 49.85'	13:31	1963	1420	543	Adare Trough (West Scarp)	
D02	3-Jan-07	15:30	69 51.67'	171 50.66'	69 51.68'	171 49.53'	17:31	1566	1132	434	Adare Trough (West Scarp)	
D03	3-Jan-07	22:42	70 00.00'	171 59.40'	70 00.26'	171 58.34'	0:59	1511	1000	511	Adare Trough (West Scarp)	
D04	4-Jan-07	3:30	70 04.20'	172 18.96'	70 04.13'	172 17.93'	5:15	2394	2021	373	Seamount inside Adare Trough	
D05	13-Jan-07	5:55	71 53.49'	171 53.99'	71 54.77'	171 54.01'	7:52	580	473	107	ABS continental shelf	
D06	13-Jan-07	9:55	71 49.29'	171 55. 39'	71 49.57'	171 55.38'	11:20	610	506	104	ABS continental shelf	
D07	13-Jan-07	12:28	71 49.20'	171 54.00'	71 48.99'	171 53.62'	13:33	590	472	118	ABS continental shelf	
D08	13-Jan-07	16:28	71 35.30'	172 24.80'	71 35.58'	172 24.70'	18:00	1610	1381	229	ABS shelf-slope	
D09	13-Jan-07	20:16	71 37.67'	172 31.36'	71 37.88'	172 30.97'	21:35	1739	1545	194	ABS isolated	
D10	14-Jan-07	23:53	71 40.00'	172 31.39'	71 40.27'	172 31.89'	1:24	1630	1411	219	ABS clumps	
D11	14-Jan-07	3:55	71 39.70'	172 43.15'	71 39.91'	172 39.91'	5:34	1760	1481	279	ABS semi isolated; edge of platform	
D12	14-Jan-07	7:44	71 37.86'	172 38.06'	71 38.08'	172 39.07'	9:13	1740	1490	250	ABS isolated; ~edge of platform	
D13	14-Jan-07	11:39	71 37.80'	172 48.20'	71 38.04'	172 48.38'	12:56	1848	1612	236	ABS isolated doublet, N center	
D14	14-Jan-07	14:58	71 36.05'	172 47.60'	71 36.32'	172 47.92'	16:18	1750	1603	147	ABS small cluster, N center	
D15	14-Jan-07	18:27	71 34.75'	172 44.00'	71 35.01'	172 44.28'	20:20	1937	1680	257	ABS isolated, single center	
D16	14-Jan-07	22:12	71 35.80'	172 49.25'	71 35.98'	172 49.26'	23:16	1944	1772	172	ABS satellite cone of D15?	
D17	15-Jan-07	2:37	71 29.95'	173 29.17'	71 30.30'	173 29.19'	4:26	2014	1763	251	ABS large isolated center, central part of basin	