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**A self-assembling peptide nanofiber was developed to sense the microenvironmental pH change associated with bacterial growth. Using a near-infrared probe, a strong correlation was observed between the local pH reduction of bacterial colonies with the degree of peptide disassembly, which led to their enhanced antimicrobial activity against anaerobic bacteria.**

Trigger-responsive nanomaterials have tremendous promise for targeted therapeutic delivery strategies that improve the treatment of a variety of diseases.<sup>1</sup> Among various approaches, self-assembly has been proven as an effective bottom-up approach to construct functional nanomaterials. A wide range of molecular building blocks, including amphiphilic polymers, lipids, proteins and peptides can be custom-designed and assembled into “smart” nanomaterials that can sense various disease-specific micro-environmental conditions.<sup>2</sup> Self-assembled nanomaterials can easily change their physicochemical properties in response to the environmental change and lead to local release of therapeutics with enhanced drug potency and reduced side effects on healthy tissues and cells. In recent years, great levels of success have been achieved for nanomaterials designed for targeted cancer therapy.<sup>3</sup> However, the development of self-assembled nanomaterials for targeted antimicrobial delivery is just getting underway for infectious disease treatment.<sup>4</sup>

Similar to some of the tumor tissues, certain bacteria can reduce the local pH of the infectious tissues through low oxygen triggered anaerobic fermentation.<sup>5</sup> Host immune response can further lower the local pH where bacteria reside through the mechanism of production of lactic acids during phagocytosis.<sup>6</sup> While the acidic pH is considered as an undesirable factor

## Bacterial acidity-triggered antimicrobial activity of self-assembling peptide nanofibers†

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causing the reduction of the antimicrobial activity of several classes of antibiotics,<sup>7</sup> it can be utilized as a natural physiological cue for the design of antimicrobial nanomaterials for targeted antimicrobial delivery. Recent reports have demonstrated advances in the design of acid-sensitive nanoparticles for targeting the bacterial membrane and delivery of small molecule antibiotics to treat bacterial infections in acidic conditions.<sup>4a</sup> Acid-dependent helical polypeptides were also reported to selectively target and eradicate pathogenic *H. pylori* without affecting commensal bacteria in the stomach.<sup>4d</sup> These successful examples provide the inspiration and highlight the feasibility of using pH as a physiological trigger to achieve targeted antimicrobial therapy.

In this work, we sought to develop a novel acid-activatable antimicrobial therapy by capitalizing on our recent development of self-assembling nanofibers (SANs) for bacterial acidity triggered antimicrobial delivery. SANs are supramolecular assemblies of *de novo* designed multidomain peptides (MDPs) that have been explored as highly cytocompatible antimicrobial and cell penetrating nanomaterials.<sup>8</sup> The first generation of MDPs has a general formula of  $K_x(QL)_yK_z$  (amino acid single code letter K: Lysine, Q: Glutamine, L: Leucine) to mimic natural cationic antimicrobial peptides (AMPs). Unlike most conventional AMPs that exist as monomers in solutions, MDPs can form supramolecular  $\beta$ -sheet nanofibers in which the hydrophobic residues and non-polar surface are partially masked between the two sheets, which has been proven as an important factor to minimize the cytotoxicity of MDPs toward mammalian cells.<sup>9</sup> The cytocompatibility of SANs was greatly enhanced compared to that of traditional monomeric AMPs. However, the confinement of the hydrophobic moiety within the assembly can also reduce their antimicrobial activity while monomeric AMPs are more potent to kill bacteria. As such, the current study by developing SANs that undergo pH-responsive disassembly combines the advantages of both self-assembled peptides in terms of their cytocompatibility and monomeric AMPs in terms of their antimicrobial activity to treat acidity-associated bacterial infections. The central hypothesis is SANs under the physiological condition are bio-inert because the membrane-interacting hydrophobic moieties of SANs are buried

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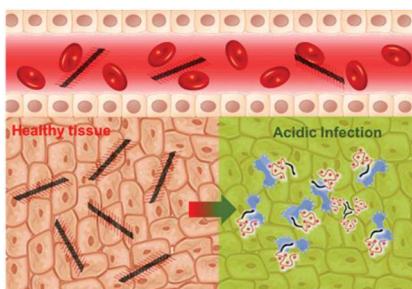


Fig. 1 Cartoon representation of cytocompatible and hemocompatible SANs formed by pH-responsive MDPs and their disassembly triggered by local bacterial acidity for the delivery of activated MDPs to eradicate bacteria.

inside the assembly and not accessible to the cell membrane. At the bacterial colonization site with an acidic pH, MDPs become charged and the increased charge density triggers SANs disassembly and subsequent release of activated peptides to effectively interact with the cell membrane and kill bacteria (Fig. 1).

Three MDPs with the sequences of  $WH_5(QL)_6K_2$ ,  $WH_7(QL)_6K_2$ , and  $WH_9(QL)_6K_2$  abbreviated as  $WH_5$ ,  $WH_7$  and  $WH_9$  (H: Histidine, W: Tryptophan) were initially explored as the building units to fabricate SANs. The sequences were chosen based on the following considerations. First, the central repeating (QL) domain provides the driving force for SANs formation under the neutral physiological condition as discussed in our previous studies.<sup>8b</sup> Second, oligo-histidine with different lengths was incorporated at the N-terminus to endow pH-responsiveness to SANs. At a pH below the  $pK_a$  of histidine, peptides become charged. The electrostatic repulsion among the positively charged MDPs would destabilize SANs and lead to the release of activated MDPs that can effectively eradicate bacteria. Third, the number of histidine residues are varied in order to integrate and achieve a good balance between self-assembly under the neutral condition and disassembly upon acidification. MDPs with different numbers of histidine would render a small library to explore the effect of charge on SANs stability, disassembly efficiency and their resulting biological activities. Lastly, two lysine residues were appended at the C-terminus to ensure sufficient solubility of SANs and minimize lateral fiber aggregation through electrostatic repulsion. All peptides contained a tryptophan residue for accurate determination of the peptide concentration by UV spectroscopy.

Elucidating the physicochemical properties of MDPs as a function of pH is critical to tailor their biological activities. Critical assembly concentration (CAC) measurements were first conducted to investigate their ability to assemble under different pH conditions. As shown in Fig. S2 (ESI<sup>†</sup>), a non-linear relationship was observed for all three MDPs at pH 7.4 (Tris buffer, 20 mM) suggesting the formation of higher ordered assemblies as the concentration increased. At an acidic pH (MES buffer, pH 5.7, 20 mM), a linear correlation was found between the fluorescence intensity and peptide concentrations, suggesting the majority of peptides do not self-assemble and rather remain isolated. It is worth noting that we chose pH 5.7 as the acidic

condition for this study in order to achieve a good balance between the protonation degree of histidine ( $pK_a = \sim 6$ ) and bacterial growth under the acidic condition. The pH-dependent self-assembly and disassembly were further investigated and confirmed by circular dichroism (CD) spectroscopy. At pH 7.4, all three peptides exhibited predominant  $\beta$ -sheet structures as characterized by a minimum peak between 210–220 nm (Fig. S3a, ESI<sup>†</sup>) indicating the formation of SANs. When the pH was reduced to 5.7, which was below the  $pK_a$  point of histidine, the presumed increase in positive charges and electrostatic repulsion triggered disassembly and unfolding of  $\beta$ -sheets to random coils and/or weak helices. As shown in Fig. S3b (ESI<sup>†</sup>), all three MDPs unfolded upon pH reduction, but to different degrees.  $WH_7$  and  $WH_9$  exhibited more disordered structures than  $WH_5$  given the larger blue shifts of the minimum absorption down to  $\sim 203$  nm indicating a greater tendency to disassemble. To quantitatively determine the extent of SANs disassembly, spin dialysis was used to estimate the amounts of disassembled MDPs upon pH reduction. Centrifugal filters with molecular weight (MW) cutoff at 10 kDa and 30 kDa were used to separate the monomeric MDP and any potential non-specific aggregates (up to 9 mers) (due to their amphiphilic nature) from the residual higher ordered assemblies, respectively. As shown in Table 1, no materials were detected in the filtrate for all three MDPs through spin dialysis suggesting the stability and integrity of SANs at the neutral pH. At the acidic pH, disassembly occurred as shown by the increased concentrations of MDPs in the filtrate using both filters. It was estimated that 24.60% of  $WH_5$ , 34.70% of  $WH_7$  and 41.00% of  $WH_9$  were disassembled to monomers based on the dialysis result using the filter with a MW cutoff at 10 kDa. Using a 30 kDa filter, the percentage of MDPs in the filtrate increased to 37.77%, 62.56% and 71.46% for  $WH_5$ ,  $WH_7$  and  $WH_9$ , respectively.

The filtrate of  $WH_9$  at pH 5.7 was analyzed by CD spectroscopy showing highly disordered random coil structures (Fig. S3c, ESI<sup>†</sup>), thus excluding the possibility of the presence of  $\beta$ -sheet oligomers in the filtrate. The oligomeric species present in the filtrate is likely due to the non-specific aggregation between the amphiphilic MDPs. Transmission electron microscopy (TEM) revealed the morphological change of MDPs under different pH treatments. For TEM characterization and the following biological evaluation, we primarily focused on  $WH_9$  because it is the most sensitive to pH change giving a higher extent of disassembly upon solution acidification. At pH 7.4,  $WH_9$  spontaneously self-assembled to form elongated fibers (Fig. 2a). Reducing the pH to 5.7 led to a

Table 1 Quantification of disassembled MDPs

Peptides	10 kDa filter		30 kDa filter	
	pH 7.4	pH 5.7	pH 7.4	pH 5.7
$WH_5$	0	$24.60 \pm 0.08\%$	0	$37.77 \pm 0.56\%$
$WH_7$	0	$34.70 \pm 0.08\%$	0	$62.56 \pm 3.55\%$
$WH_9$	0	$41.00 \pm 0.49\%$	0	$71.46 \pm 0.27\%$

Standard deviation is calculated based on 3 measurements for each sample.

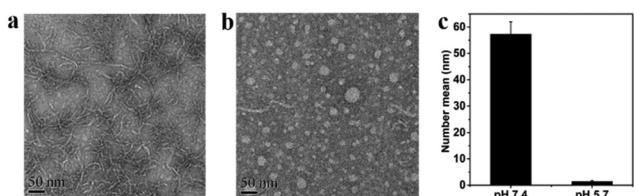


Fig. 2 Negatively stained TEM images of WH<sub>9</sub> at (a) pH 7.4 showing SANS formation and (b) at pH 5.7 showing SANS disassembly. (c) pH-Dependent hydrodynamic size measurement by DLS. Peptide concentration: 100  $\mu$ M in Tris buffer (pH 7.4, 20 mM) and MES buffer (pH 5.7, 20 mM).

significant reduction of the fiber density and the formation of non-specific spherical aggregates (Fig. 2b). We suspect that these spherical aggregates were formed from disassembled MDPs due to the drying effect during TEM sample preparation process. The disassembly of SANS was verified by dynamic light scattering (DLS) measurements showing a dramatically reduced particle size to below 2 nm when pH was reduced (Fig. 2c and Fig. S4, ESI<sup>†</sup>). Although the number mean of the hydrodynamic diameter generated by DLS does not represent the actual size of these nanofibers due to their non-spherical shape, the dramatic size reduction suggests the effectiveness of solution acidity to trigger the disassembly of SANS.

Acidity-triggered SANS disassembly was further studied in the context of bacterial inoculation on an agar plate. We chose *Bacteroides fragilis* as a model bacterium that undergoes anaerobic growth leading to the acidification of the surrounding environment. We first determined whether the growth of *B. fragilis* would influence the pH immediately adjacent to the bacterial colony. Using a pH ratiometric near infrared probe developed recently,<sup>10</sup> we measured the change of fluorescent intensity of the probe with time using an *in vivo* Kodak imager. The results allow us to calculate the pH nearby the colony on an agar plate at 0.5, 2.5, 7, and 22.5 h upon bacterial inoculation. Interestingly, we found that growth of *B. fragilis* released metabolites which can cause the surrounding environment to become acidic (from pH 7.5 to pH 6.3 in less than 24 h) (Fig. S5, ESI<sup>†</sup>). To test whether the *in situ* low pH can induce SANS disassembly, we synthesized and prepared rhodamine (Rho)-labeled WH<sub>9</sub>. Rho-WH<sub>9</sub> had very low fluorescent intensity due to fluorescent quenching upon self-assembly. By reducing the local pH, WH<sub>9</sub> disassembled leading to the recovery of rhodamine fluorescence. Therefore the fluorescence intensity of the peptide reflects the degree of peptide disassembly and can be used to correlate with the microenvironmental pH change associated with bacterial growth. To determine whether the acidic environment nearby bacterial colonies causes SANS to disassemble, Rho-WH<sub>9</sub> was applied on both the bacterial colonies and non-inoculated agars as controls after 24 h of bacterial inoculation. The fluorescence intensity was monitored immediately using an *in vivo* Kodak imager. The results showed an average of 88% increase of the fluorescence intensity for peptides deposited on the bacteria colonies than those on the agar media without bacteria (Fig. 3a), suggesting a local acidic pH can trigger the disassembly of SANS, leading to the recovery of self-quenched fluorescence. The fluorescence intensity of the peptide across the

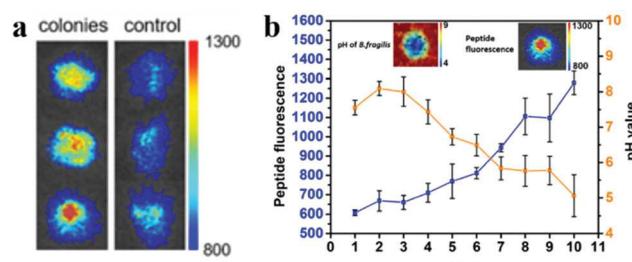


Fig. 3 Local bacterial acidity triggered peptide disassembly as determined by *in situ* fluorescence microscopy. (a) Fluorescence intensity of Rho-labeled WH<sub>9</sub> deposited on 3 bacterial colonies (3 spots on the left panel) compared to those on agar media (right panel) without bacteria showing that the acidic bacterial environment can induce SANS disassembly that led to the recovery of the self-quenched rhodamine fluorescence. (b) Correlation of the local bacterial pH (orange line) with the fluorescence intensity of Rho-labeled WH<sub>9</sub> (blue lines) applied on bacterial colonies. The inset pictures are fluorescent microscopic images of bacterial colonies upon Rho-labeled peptide treatment (right) and colonies containing the near infrared pH probe (left).

bacterial colony was further plotted as a function of imaging pixels ( $\sim 0.1$  mm per pixel) starting from the outermost of a colony (shown as 0 on the x-axis of Fig. 3b) while moving toward the center (shown as 10 on the x-axis of Fig. 3b). The local bacterial pH change across a single colony was measured by the ratiometric fluorescence probe described above and plotted in the same manner. A good correlation was observed between the reduced pH and the increased fluorescence of Rho-WH<sub>9</sub>, further confirming local bacterial acidity can trigger SANS disassembly.

The antimicrobial activities of WH<sub>9</sub> were tested against both Gram-negative bacteria, *Escherichia coli* and *B. fragilis* and Gram-positive bacteria, *Staphylococcus aureus* under the anaerobic condition where bacterial cultures became acidic over time peptides were co-incubated with *E. coli*, *B. fragilis* and *S. aureus* for 48 h and the UV absorbance at 600 nm was measured for the estimation of the minimum inhibitory concentration (MIC) values. As shown in Table 2, WH<sub>9</sub> was effective against all three bacterial strains in the anaerobic condition where the bacterial culture gradually became acidic to pH 6.4. The MIC values of WH<sub>9</sub> were determined at 10  $\mu$ M against *E. coli*, 5  $\mu$ M against *B. fragilis* and 5  $\mu$ M against *S. aureus*. In contrast, the MIC of WH<sub>9</sub> was estimated at 40  $\mu$ M against *E. coli* in the aerobic condition where the culture pH remained neutral (pH was between 7.2 and 7.5 during culture). For aerobic bacterial cultures with a starting pH at 5.7, the MIC was determined as 10  $\mu$ M comparable to that determined in the anaerobic condition. In comparison, the MICs of WH<sub>5</sub> and WH<sub>7</sub> were determined at 40  $\mu$ M and 20  $\mu$ M in the

Table 2 Antimicrobial activity, cytotoxicity and hemolytic activity

E. coli	MIC ( $\mu$ M)			IC <sub>50</sub> ( $\mu$ M)	HC <sub>10</sub> ( $\mu$ M)		
	Anaerobic		Aerobic				
	B. fragilis	S. aureus					
10	5	5	pH 7.4 >40	pH 5.7 10	HC <sub>10</sub> > 160		

acidic aerobic *E. coli* cultures showing less potency to inhibit the growth of bacteria. The antimicrobial activity also correlates well with the peptide's ability to disassemble as detailed in the spin dialysis experiment (Table 1).

The mode of antimicrobial action was investigated by epifluorescence microscopy.  $\text{WH}_9$  was co-incubated with *E. coli* aerobically so that the culture pH can be adjusted and maintained either acidic or neutral during the entire culture. A live-dead assay was performed wherein *E. coli* was incubated with  $\text{WH}_9$  for 3 h, followed by staining with SYTO9 and Propidium Iodide (PI). As shown in (Fig. 4a), a much higher fraction of *E. coli* cells fluoresced red at the acidic pH due to pH-triggered disassembly and release of peptides that can increase the membrane permeability of PI. At the neutral pH (Fig. 4b), the peptide was confined within the SANs and did not have sufficient freedom to access to and further permeate the bacterial cell membrane. The live-dead assay was also performed in Gram-positive *S. aureus* culture showing the same trend of pH-dependent antimicrobial activity (Fig. S6, ESI†).

The physical interaction between  $\text{WH}_9$  and bacteria was also studied by fluorescence microscopy upon incubation of FITC-labelled  $\text{WH}_9$  with *E. coli* followed by PI staining. The binding affinity of  $\text{WH}_9$  toward bacteria was greatly improved upon acidification as demonstrated by the numbers of bacterial cells that were attached by peptides showing green fluorescence on the cell membrane (Fig. S7a, ESI†) while at the neutral condition much less binding occurred between the peptides and bacteria (Fig. S7b, ESI†). Scan electron microscopy (SEM) was used to visualize any morphological change induced in bacteria by exposure to  $\text{WH}_9$  at the acidic condition. As compared to control bacteria without peptide treatment (Fig. S8a, ESI†), *E. coli* incubated with  $\text{WH}_9$  under the acidic culture condition showed significant membrane damage (Fig. S8b, ESI†), suggesting the mode of action is through bacterial membrane disruption by the disassembled peptides.

A critical challenge associated with conventional AMPs is their moderate to severe cytotoxicity and hemolytic activity.<sup>11</sup> We have recently demonstrated that self-assembly can be an

effective approach to reduce the non-polar membrane-contact area of AMPs leading to greatly improved cytocompatibility and bacterial cell selectivity.<sup>9</sup> To evaluate the cytotoxicity of the newly designed  $\text{WH}_9$  toward mammalian cells during blood circulation, NIH/3T3 fibroblasts were incubated with peptides at various concentrations and the cell viability was quantified by the MTT assay. 80  $\mu\text{M}$  was selected as the upper concentration threshold to avoid potential precipitate formation as the peptide concentration increased in the mammalian culture media. As shown in Fig. S9 (ESI†), dose-dependent cell viability was measured showing >80% of cell viability up to 40  $\mu\text{M}$  and 72% cells were still alive upon incubation with peptides at 80  $\mu\text{M}$ . The hemocompatibility was evaluated by incubating human red blood cells (RBCs) with  $\text{WH}_9$  at different concentrations for 1 h and released haemoglobin was measured by UV spectroscopy (Fig. S10, ESI†). Within the tested peptide concentrations up to 160  $\mu\text{M}$  (16 times of the MIC) less than 5% of hemolysis was observed with peptide-treated RBCs compared to the positive control group of RBCs treated with Triton-100. Taken together, the pH-triggered antimicrobial activity and excellent cytocompatibility and hemocompatibility of self-assembled  $\text{WH}_9$  highlight their great potential as a new antimicrobial strategy to effectively treat bacterial infections associated with acidity.

In summary, we have demonstrated a new pH-responsive antimicrobial nanomaterial based on the self-assembly of *de novo* designed MDPs for acid-responsive antimicrobial delivery at the site of infection associated with bacterial acidity. The MDP can be designed to form stable nanofibrous structure in neutral pH with excellent cytocompatibility and hemocompatibility. The pH-triggered disassembly was demonstrated in both the aqueous solution and on a bacteria-inoculated agar plate and shown to be important factors for their antimicrobial activity. This new antimicrobial strategy while awaiting more extensive *in vitro* evaluation and *in vivo* studies holds great promise to treat bacterial infections in which acidity plays an important role in bacterial pathogenesis. For future studies, SANs based on custom-designed non-natural amino acids may offer diverse chemical functionality and broader pH-tunability to suit various clinical needs in the combat of infectious diseases.

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## Conflicts of interest

There are no conflicts to declare.

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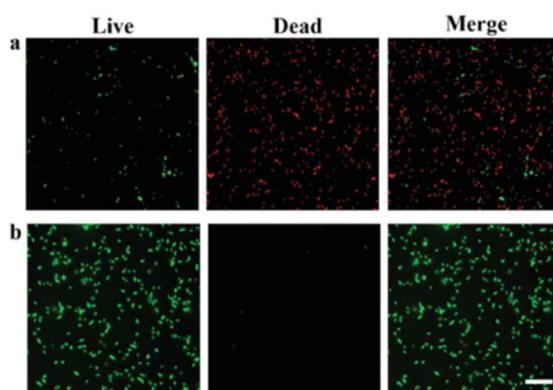


Fig. 4 Fluorescence images of live/dead bacterial assay results. Top panel: *E. coli* treated with 20  $\mu\text{M}$   $\text{WH}_9$  at (a) pH 5.7 and (b) pH 7.4 for 3 h. Live bacteria was stained with SYTO9 (green) and dead bacteria was stained with PI (red). Scale bar: 20  $\mu\text{m}$ .

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