



## Growth of $\text{NbO}_2$ thin films on $\text{GaN}(0001)$ by molecular beam epitaxy

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### ABSTRACT

We report on the growth of epitaxial  $\text{NbO}_2$  thin films by molecular beam epitaxy on  $\text{GaN}$  (0001). The combination of these two materials is of interest in resistive switching devices that can operate at high temperature. We show that direct deposition of Nb metal on  $\text{GaN}$  under an oxygen environment results in a substantial interfacial reaction layer. We perform detailed spectroscopic and structural analyses of the film and interface and describe their implications for the growth of this materials system.

### 1. Introduction

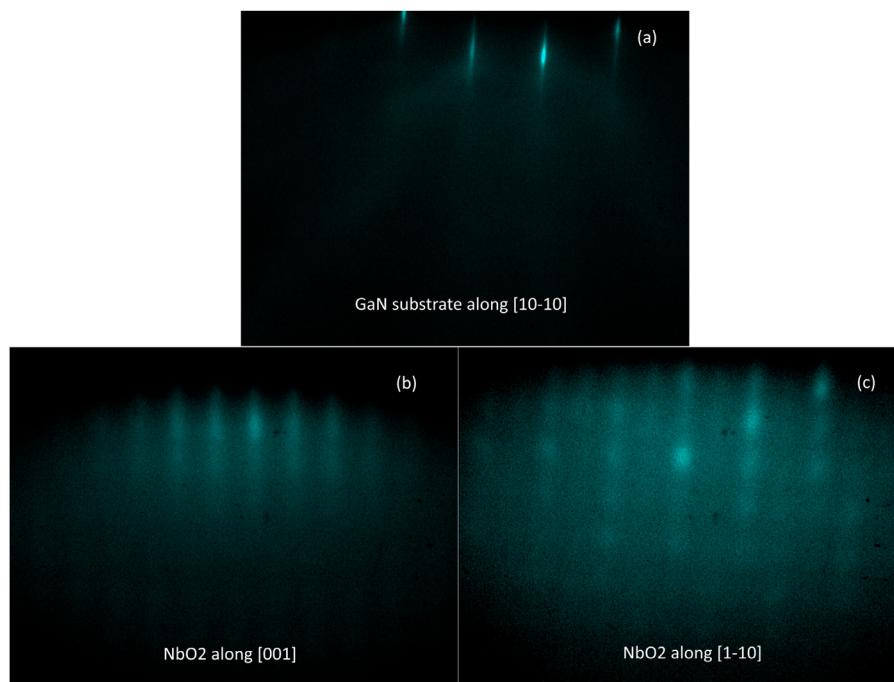
Transition metal oxide-semiconductor heterostructures are of great interest due to their potential for combining the rich electronic functionality of transition metal oxides with the wide use of semiconductors in logic, memory, and optoelectronic devices [1–3]. A number of transition metal oxides exhibit a metal–insulator transition (MIT) that can be useful for a wide variety of applications including electronic and optoelectronic switches, memristors, and tunnel field effect transistors (FETs) [4–6]. For applications in the MIT in electronics, the process for externally modulating the phase transition needs to be highly controllable. Niobium dioxide,  $\text{NbO}_2$ , is a MIT material, whose phase transition temperature is very high (800 °C) [7]. This makes it suitable for use in high temperature and high power electronic devices, where variations in the ambient temperature will not affect the state of the device. For such high temperature applications, Si is not suitable and thus there is a strong drive to use wide band gap semiconductors, such as  $\text{GaN}$  and  $\text{SiC}$  [8,9].  $\text{GaN}$  is currently used in optoelectronic devices and blue LEDs [10,11], but it is also a semiconductor of choice for high temperature, high frequency, and high power applications because of its wide 3.4 eV direct band gap, high thermal and chemical stability and large thermal conductivity [12–14]. A possible application of  $\text{NbO}_2$  on  $\text{GaN}$  is in fabricating hyper-FETs, which inserts a MIT material in the source terminal of a FET as a means to obtain steep switching and high on/off ratio [15]. This new FET design is being explored for reduced power consumption in conventional logic (abrupt switching) as well as for enabling new computational paradigms such as neuromorphic computing [16].  $\text{GaN}$ -based hyper-FETs based on  $\text{NbO}_2$  will be capable of operating at elevated temperatures. In addition,  $\text{GaN}$  is a polar

material with piezoelectric properties that could be utilized for electromechanical devices and thermal sensors [17]. One possible way of inducing the MIT in  $\text{NbO}_2$ , where the MIT occurs due to a Peierls transition [18], is to alter the lattice constant using strain. By growing  $\text{NbO}_2$  on a piezoelectric semiconductor such as  $\text{GaN}$ , one can then use an external electric field to modulate strain in the material. The combination of  $\text{NbO}_2$  and  $\text{GaN}$  would also be potentially useful in resistive switching devices at elevated temperatures.

However, the formation of an abrupt low-defect density interface between a crystalline transition metal oxide layer and a semiconductor material remains a challenging technical problem. There are several reports of epitaxial interfaces between a functional transition metal oxide and  $\text{GaN}$ , including  $(\text{Ba},\text{Sr})\text{TiO}_3$  [19,20],  $\text{YMnO}_3$  [21,22], and  $\text{LiNbO}_3$  [23,24]. Other groups grow rocksalt oxides such as  $\text{CaO}$  and  $\text{MgO}$  [25,26] or rutile  $\text{TiO}_2$  [27,28], which can then be used as buffers for growing other oxides on  $\text{GaN}$  [29]. In this work, we report on the results of growing  $\text{NbO}_2$  by molecular beam epitaxy (MBE) on  $\text{GaN}$  using a simple direct deposition process involving no surface conditioning other than degreasing and annealing of the substrate. We show that for materials system, such a direct deposition process using a metal precursor results in a reaction between the incoming Nb and  $\text{O}_2$  flux and the  $\text{GaN}$  surface. We perform detailed chemical spectroscopic and structural analyses of the film and interface using reflection high energy electron diffraction (RHEED), x-ray photoelectron spectroscopy (XPS), in-plane X-ray diffraction (XRD), and cross-section scanning transmission electron microscopy (STEM) to better understand the resulting structure and propose a method for avoiding this interfacial reaction between  $\text{NbO}_2$  and  $\text{GaN}$ .

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**Fig. 1.** RHEED images of GaN substrate and NbO<sub>2</sub> film. (a) RHEED image of the substrate taken along the [10̄10] azimuth showing relatively flat surface with no reconstruction. (b) RHEED pattern for NbO<sub>2</sub> along the [001] azimuth, which is parallel to the GaN [11̄20] azimuth. (c) RHEED pattern for NbO<sub>2</sub> along the [1̄10] azimuth, which is parallel to the GaN [10̄10] azimuth.

## 2. Experimental details

For this work, a commercially available substrate purchased from MTI Corporation was used, consisting of 0.5  $\mu\text{m}$ -thick GaN (0001) on a 0.1  $\mu\text{m}$  AlN (0001) buffer layer grown on single crystal Si (111). The substrate was cut into 10 mm  $\times$  10 mm pieces and degreased by sonicating for 5 min each in acetone, isopropanol, and deionized water. After degreasing, the substrate was loaded into the load lock chamber where it was first lightly outgassed in vacuum at 150 °C for 30 min prior to transferring into the growth chamber. To clean the GaN surface, the substrate was first exposed to an atomic nitrogen plasma source at a substrate temperature of 300 °C for 30 min. The radio frequency plasma source was operated using 250 W forward power under a background N<sub>2</sub> gas pressure of  $1.3 \times 10^{-3}$  Pa. After the nitrogen plasma treatment, the substrate is then annealed in vacuum at 700 °C for another 30 min. Fig. 1(a) shows the reflection high energy electron diffraction (RHEED) pattern of the cleaned GaN surface just prior to growth taken along the [10̄10] azimuth showing a 1  $\times$  1 pattern with intense diffraction spots along a circular arc, consistent with a clean, flat surface. XPS analysis of the cleaned GaN surface (not shown) shows no detectable carbon and less than 0.2 monolayer of oxygen on the surface.

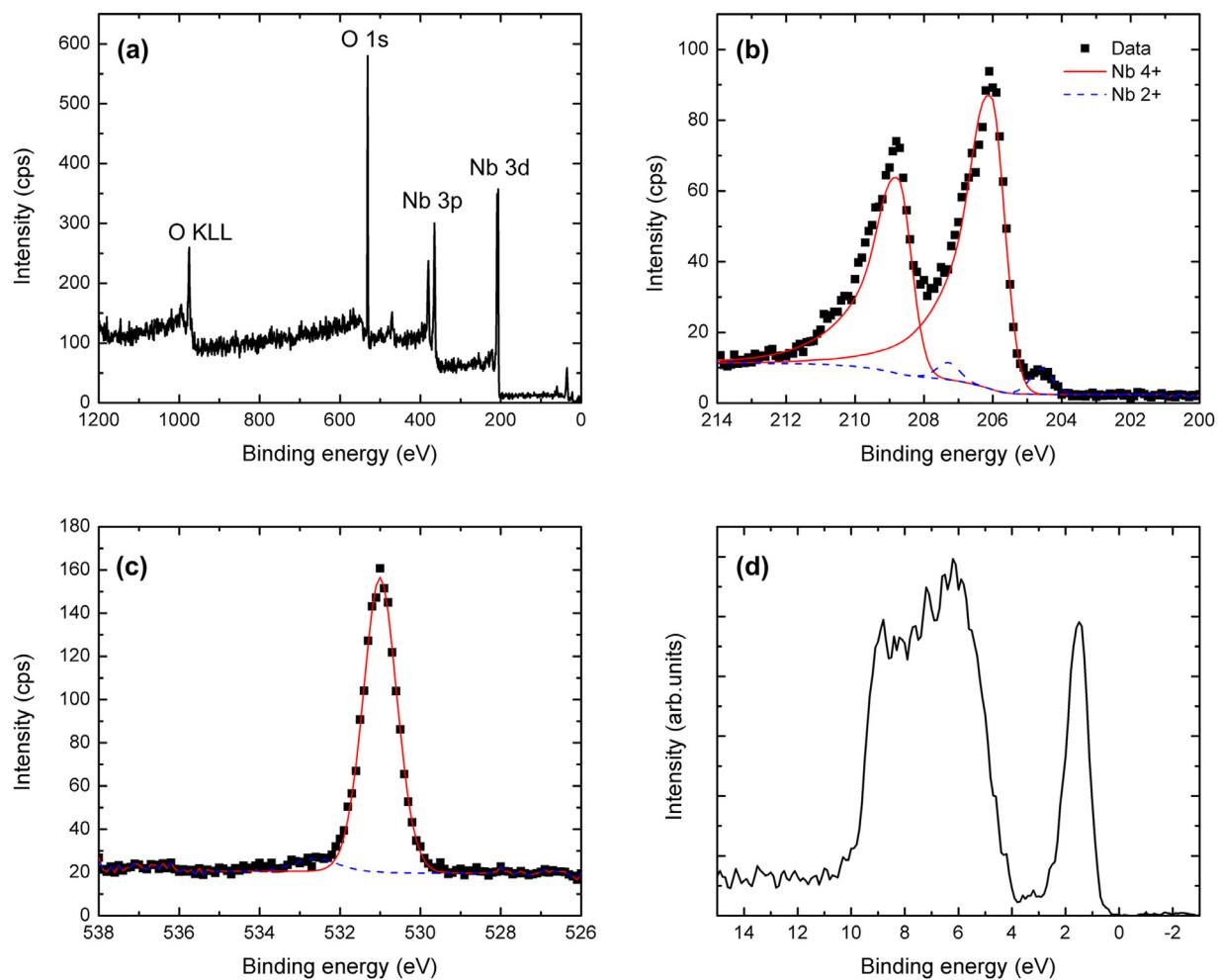
For the NbO<sub>2</sub> film growth, we used electron beam evaporation of Nb metal in the presence of molecular oxygen in a customized DCA Instruments M600 oxide MBE reactor. The growth conditions that we previously found to yield stoichiometric NbO<sub>2</sub> growth on (La,Sr)<sub>2</sub>(Al,Ta)<sub>2</sub>O<sub>6</sub> substrates were used to grow the film on GaN [30]. The growth conditions were Nb metal flux of 2 Å/min with an oxygen pressure of  $8 \times 10^{-4}$  Pa at a substrate temperature of 800 °C. The Nb metal flux is first calibrated under vacuum using an INFICON quartz crystal microbalance placed near the substrate position. We use an electron energy of 7.75 keV and an emission current of 200 mA for the Nb evaporation from the electron beam source. With the Nb shutter closed, the substrate was then heated to the growth temperature at a rate of 30 °C/min after which oxygen was let into the chamber until the target pressure is reached, which took about 5 min. The Nb source shutter and main shutter are then simultaneously opened. The sample was grown for 1 hour (target thickness of 25 nm) with sample rotation.

The sample surface was constantly monitored using RHEED during the growth. Fig. 1(b) and (c) shows the RHEED patterns from the surface of the sample just after growth is finished. After growth, the sample was cooled down at 40 °C/min to 200 °C under the same oxygen pressure used for the growth. After that oxygen flow was stopped and the sample transferred in situ to the XPS analysis chamber. XPS was performed using a monochromatic Al K $\alpha$  source and a VG Scienta R3000 electron energy analyzer. The spectrometer is calibrated such that the measured binding energy of a clean copper sample is 932.68 eV. After XPS analysis, the samples were characterized by both out-of-plane and in-plane x-ray diffraction using Cu K $\alpha$  radiation in a Rigaku Ultima IV diffractometer with an in-plane arm and thin film stage. Cross-section (S)TEM samples were prepared by mechanical wedge polishing with diamond lapping films in the [11̄20] cross-section projection for GaN substrate. The samples were then ion milled in a Fischione 1050 at 5° ion mill angle. TEM conventional and high-resolution images were taken at 300 kV on FEI TF30 field emission microscope. Scanning transmission electron microscopy experiments were performed on a FEI Titan microscope with a CEOS probe aberration-corrector operated at 200 kV. STEM images were collected with a 24.5 mrad probe semi-angle, 28 pA probe current with STEM resolution of 0.8 Å. High angle annular dark field (HAADF) images were acquired in the angle range from 54 to 270 mrad. STEM electron energy loss spectroscopy (EELS) spectrum images were acquired using a 24.5 mrad probe semi-angle, spectrometer collection angle of 82 mrad, 470 pA probe current, and STEM resolution of 2.1 Å.

## 3. Results and discussion

### 3.1. Film composition

The 25 nm-thick NbO<sub>x</sub> film grown on GaN was measured using in situ XPS immediately after growth. Fig. 2 shows XPS data from the sample for the Nb 3d, O 1s and valence band energy regions, as well as a wide range survey scan. The survey scan (Fig. 3a) confirms that only peaks arising from Nb and O are visible. The Nb 3d spectrum (Fig. 3b) shows an asymmetric spin-orbit pair of peaks whose peak maximum is



**Fig. 2.** XPS spectra of  $\text{NbO}_2$  grown on GaN. (a) Survey spectrum showing peaks only from Nb and O. (b) Nb 3d spectrum showing asymmetric spin-orbit pair of peaks at a binding energy of 206.1 eV, consistent with  $\text{Nb}^{4+}$ . A minor component of  $\text{Nb}^{2+}$  at 204.6 eV can also be seen. (c) O 1s spectrum showing main peak with binding energy of 531.0 eV. A small shoulder at higher binding energy is also visible. (d) Valence band spectrum showing strong and sharp Nb 4d feature centered at around 1.5 eV binding energy and broad O 2p band centered around 7.2 eV.

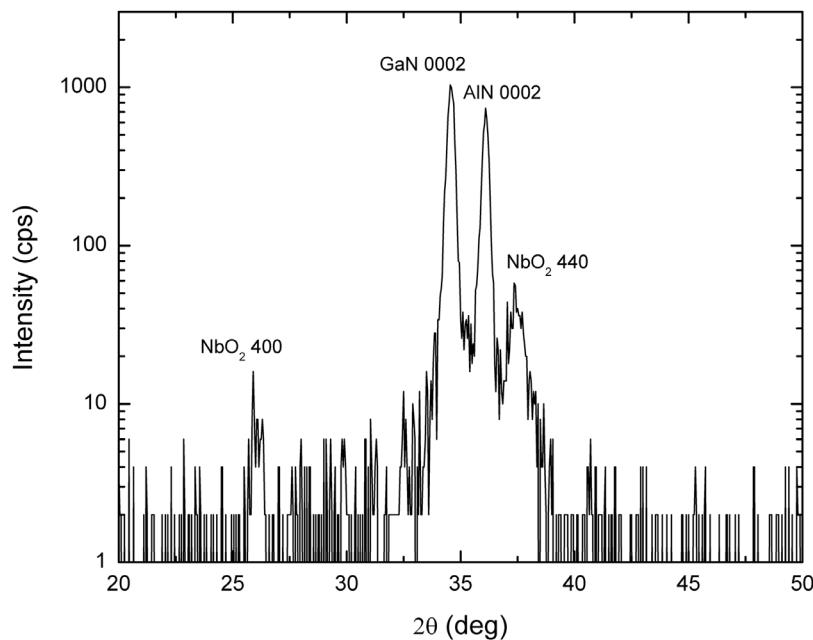
at 206.0 eV. The spectrum can be fit using a Lorentzian peak with an exponential tail to high binding energy. The asymmetric nature of the peaks and its position is consistent with single phase  $\text{NbO}_2$  [31]. There is a very small component at the lower binding energy side of the main peak (~204.6 eV) that is an indication of slight underoxidation of the sample [31]. The O 1s spectrum (Fig. 3c) shows a primary peak with a binding energy of 531.0 eV from the bulk of the film and a much smaller surface peak at 532.5 eV. The calculated O/Nb ratio using Wagner relative sensitivity factors [32] (with the number for Nb modified to give the correct stoichiometry for an  $\text{Nb}_2\text{O}_5$  film) and accounting for sampling depth yields a value of 1.9. Fig. 3d shows the valence band spectrum of the sample where two distinct features are clearly observable: a broad peak centered around 7.2 eV (corresponding to the oxygen 2p-derived band) and a narrow peak centered at 1.5 eV (corresponding to the split-off 4d<sub>xy</sub> Nb state of  $\text{NbO}_2$ ). The shape and relative heights (~0.8) of these two features are consistent with single phase  $\text{NbO}_2$  [31].

### 3.2. Film structure and epitaxy

As observed with RHEED and in-plane XRD, the in-plane orientation of the final  $\text{NbO}_2$  film and the GaN substrate are aligned in a specific way. Fig. 1(b) and (c) shows RHEED patterns for the  $\text{NbO}_2$  film after growth taken along the GaN  $<11\bar{2}0>$  and  $<10\bar{1}0>$  directions, respectively. The average RHEED spacing along the GaN  $<11\bar{2}0>$

azimuth is determined to be about 2.9 Å and is consistent with one-half of the *c* lattice of the low-temperature body-centered tetragonal phase of  $\text{NbO}_2$  (space group  $I4_1/a$ ) [18]. Rotating the sample reveals that this spacing repeats every 60°, showing that the  $\text{NbO}_2$  film has overall a pseudo-six-fold in-plane symmetry. Because the  $\text{NbO}_2$  film has two-fold symmetry in this orientation, the RHEED data indicates that there are three symmetry-related in-plane crystalline domains present in the film. The pseudo-six-fold symmetry is also observed along the GaN  $<10\bar{1}0>$  azimuth with the sample rotation.

XRD measurements were taken on the sample with both symmetric  $2\theta - \omega$  out-of-plane scans, and grazing incidence in-plane  $2\theta_\chi - \phi$  scans along the two major in-plane high symmetry directions of GaN. The XRD measurements are taken using parallel beam geometry with a beam height of 0.5 mm for out-of-plane scans and 0.1 mm for in-plane scans. Fig. 3 shows the out-of-plane XRD scan. In addition to features related to the substrate (GaN and AlN), a peak is found at  $2\theta \sim 37.5^\circ$  ( $d \sim 2.4$  Å) and a second much weaker peak at  $\sim 26.1^\circ$  ( $d \sim 3.4$  Å). The predominant film peak at  $37.5^\circ$  is consistent with the 440 reflection of  $\text{NbO}_2$ . This is the same out-of-plane orientation that is obtained when growing  $\text{NbO}_2$  on  $\text{SrTiO}_3$  (111) [30]. Such an orientation is also consistent with the RHEED patterns in Fig. 2. The minor peak at  $26.1^\circ$  may correspond to some grains of  $\text{NbO}_2$  having 100 out of plane orientation or it may arise from the disordered polycrystalline interfacial reaction layer discussed below. A rocking curve taken about the film 440 peak has a full-width at half-maximum of  $0.66^\circ$ . Fig. 4(a) and (b) shows



**Fig. 3.** Out of plane x-ray diffraction scan of  $\text{NbO}_2$  film grown on GaN. The film is predominantly [110]-oriented with a very small amount of [100] orientation. A rocking curve scan taken about the film 440 peak shows a full width at half maximum of  $0.66^\circ$ .

grazing incidence in-plane  $2\theta_\chi - \phi$  scans aligned along GaN [10 $\bar{1}$ 0] and along GaN [20], respectively. The scans are obtained with a fixed  $\omega$  and  $2\theta$  of  $0.3^\circ$ . For the scan where the GaN 11 $\bar{2}$ 0  $d$ -spacing is observed (Fig. 5a), we see film peaks at  $34.9^\circ$  ( $d = 2.56 \text{ \AA}$ ) and  $37.0^\circ$  ( $d = 2.43 \text{ \AA}$ ). These spacings correspond to  $\text{NbO}_2$  222 and 440 reflections. The reason for the two different film spacings is because of the presence of three symmetry-related domains, two of which are equivalent positive and negative  $60^\circ$  rotations from the other one. In addition to the main film peaks, the corresponding higher order peaks (444 and 880) as well as a weak feature at  $\sim 53.5^\circ$  that may be related to a 100-oriented crystallite are also visible in the XRD scan. For the scan where the GaN 11 $\bar{2}$ 0  $d$ -spacing is observed (Fig. 5b), film peaks at  $62.2^\circ$  ( $d = 1.49 \text{ \AA}$ ) and  $65.3^\circ$  ( $1.43 \text{ \AA}$ ) are observed. These spacings correspond to  $\text{NbO}_2$  004 and 662 reflections. For the same reason described above, two different film in-plane spacings are observed because of the presence of three rotational domains, two of which are identical. Fig. 5 shows the epitaxial relationship between  $\text{NbO}_2$  and GaN that is consistent with the observations in RHEED and XRD. The  $\text{NbO}_2$  440 plane matches the GaN 0001 surface in three possible orientations leading to the pseudo-Six-fold symmetry observed.

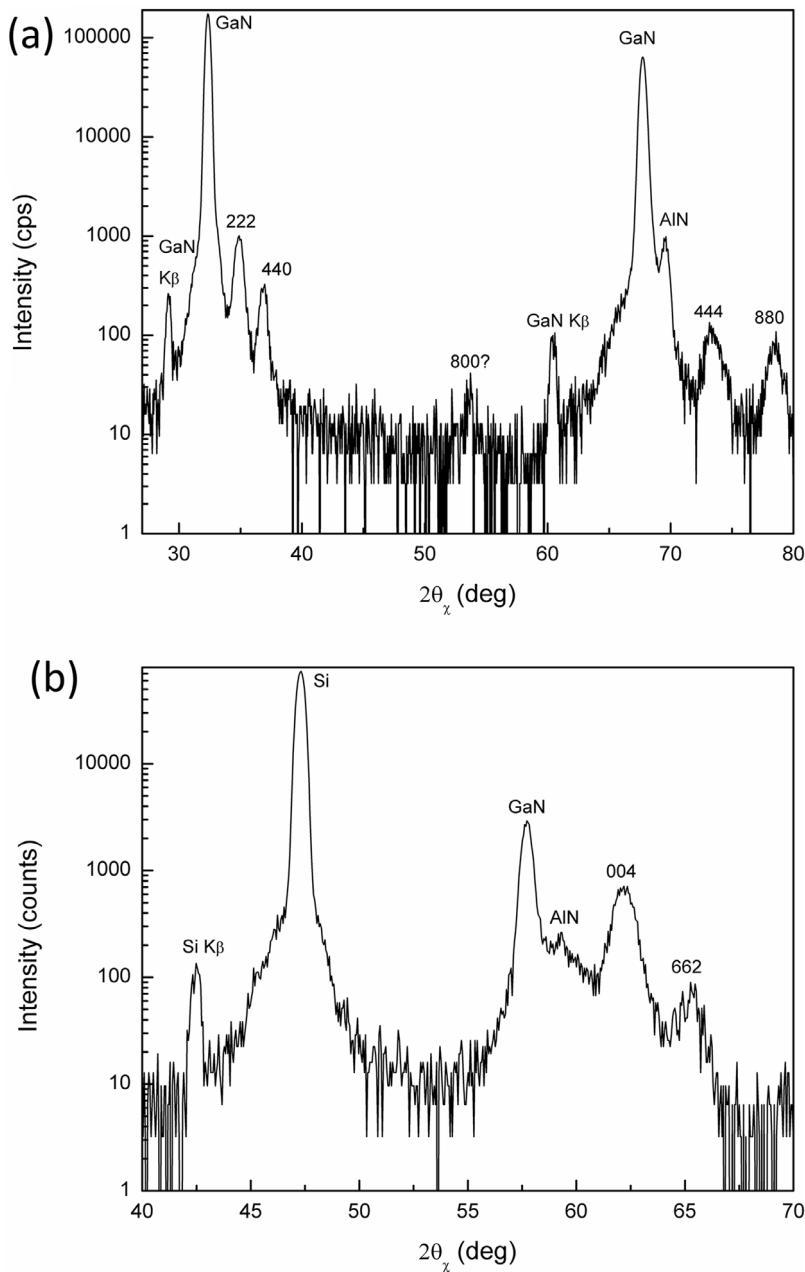
In order to get a more complete picture of the physical structure of the film and interface, cross-section TEM was performed using both standard high resolution lattice imaging, as well as atomically resolved aberration-corrected STEM imaging. Fig. 6(a) shows a high resolution TEM image of the  $\text{NbO}_2$  grown on GaN. The bottom shows the single crystal GaN layer with a rough top surface. In the middle is a micro-crystalline layer that is about 10 nm thick with an amorphous interfacial layer with the substrate about 1–2 nm thick. On the top is the crystalline  $\text{NbO}_2$  layer ( $\sim 12 \text{ nm}$ ). There is clearly a chemical reaction occurring between the  $\text{NbO}_2$  and GaN producing the thick micro-crystalline layer and a thinner amorphous interfacial layer as well as a very rough GaN surface. The nature of this reaction layer is discussed in the next section.

Fig. 6(b–d) shows Fourier transforms taken of different regions of the image. Fig. 6(b) shows the Fourier transform for the top  $\text{NbO}_2$  layer only demonstrating the crystalline nature of this layer. Fig. 6(c) is the Fourier transform for the entire image. Polycrystalline rings from the reaction layer can be seen superimposed on the diffraction spots from both the top layer and the substrate. Fig. 6(d) is the Fourier transform

for the GaN layer only showing significantly sharper reflections. Typical  $\text{NbO}_2$  domain sizes found for the sample was in the range of 15–30 nm. Fig. 7(a) shows a high-resolution HAADF image of the  $\text{NbO}_2$  layer. This image was filtered using the yellow and blue set of reflections shown in Fig. 7(b). The image confirms the 110-orientation of the  $\text{NbO}_2$  epitaxial layer. A zigzag pattern on the left side of the image indicates slightly misoriented grains.

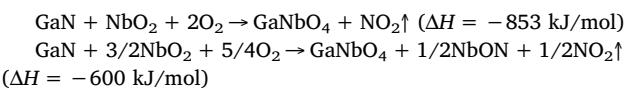
### 3.3. Reaction layer

The reaction layer consists of two regions: a thin ( $< 2 \text{ nm}$ ) amorphous region and a thick ( $\sim 10 \text{ nm}$ ) random polycrystalline region with very small grains. Based on x-ray reflectivity measurements, the polycrystalline region appears to be nearly indistinguishable chemically from the crystalline  $\text{NbO}_2$  top layer (nearly identical electron density) and is likely also composed of  $\text{NbO}_x$  that is highly disordered. To obtain a better understanding of the interface, electron energy loss spectroscopy (EELS) elemental maps are obtained from the STEM image of the polycrystalline region, the amorphous reaction region, and the upper surface of the GaN substrate. Fig. 8 shows elemental maps for Ga, N, Nb, and O obtained from the interface region of the sample. Oranges/yellows indicate large amounts of the element, while greens indicate moderate amounts, and blues/blacks indicate low or insignificant amounts. The upper part of the elemental maps corresponds to the polycrystalline region. This portion consists of Nb and O only with possible residual amounts of Ga. This is consistent with x-ray reflectivity measurements which do not see an electron density difference between the epitaxial layer and the polycrystalline  $\text{NbO}_2$  top layer. The thin middle transition layer indicated by the white horizontal lines in the elemental maps corresponds to the amorphous reaction layer. The layer consists of significant amounts of O, Nb, and Ga plus minor amounts of N. The substrate subsurface as expected consists mainly of Ga and N although some Nb and O appear to have partially diffused into the subsurface region. In order to further elucidate the nature of the amorphous reaction layer, in situ XPS was performed on a sample with nominally 4 nm of  $\text{NbO}_2$  to enable a measurement of the interface chemical composition. For a sample grown using the standard method described in Section 2 where the substrate is heated, then exposed to oxygen gas, and then followed by Nb metal deposition, the interface



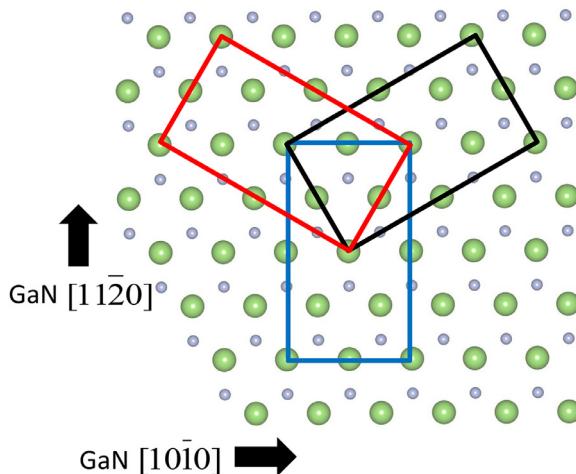
**Fig. 4.** Grazing incidence in-plane x-ray diffraction. Two scans centered on different high symmetry in-plane directions of GaN are shown. (a) Scan showing GaN 10-10 spacing. Film peaks from the 222 and 440 reflections are observed. (b) Scan showing GaN 11-20 spacing. Film peaks from the 004 and 662 reflections are observed.

consists of a combination of NbN, NbO<sub>x</sub>, and GaO<sub>x</sub>. Fig. 9(a) shows the Nb 3d core level for such a sample showing clearly the presence of NbN in addition to oxides of Nb. It is clear from Figs. 8 and 9(a) that there is a reacted region where all four elements (Ga, N, Nb, O) become intermixed. We speculate on the possible compounds present in the reaction layer by looking at the possible reactions between GaN and NbO<sub>2</sub> in the presence of excess O<sub>2</sub> ( $8 \times 10^{-4}$  Pa). We discard the possibility of forming Nb<sub>2</sub>O<sub>5</sub> as based on our and others' experience, the prevailing oxygen gas pressure is not sufficient to form Nb<sub>2</sub>O<sub>5</sub> from elemental Nb [30,33]. We consider the product compounds that can form to be NbGaO<sub>4</sub>, NbN, NbON, NO<sub>x</sub> ( $x = 0.5$  to 2), N<sub>2</sub> and Ga<sub>2</sub>O<sub>3</sub>. From reported values (experimental where available and calculated otherwise) of formation energies [34], we infer that the most likely reaction scenarios are the following because they have the highest (most negative) reaction enthalpies.



The amorphous layer is expected to consist mainly of GaNbO<sub>4</sub> with minor amounts of NbON. This composition would be consistent with what is observed from the STEM-EELS elemental maps and from XPS of thin layers.

It is clear that the crystalline NbO<sub>2</sub> layer formed is not in close contact with the substrate for samples grown using the conditions described above, making coupling between the metal-insulator transition behavior of the NbO<sub>2</sub> and the GaN extremely weak. This strong reactivity between NbO<sub>2</sub> and GaN points to the need to develop a different layer-by-layer approach, where each grown layer must be thermodynamically stable in contact with the layer above and beneath it, similar to the case for SrTiO<sub>3</sub> on Si [35]. As an initial attempt to avoid



**Fig. 5.** Schematic diagram of the epitaxial relationship between  $\text{NbO}_2$  and GaN. The GaN (0001) surface atomic structure is shown with surface Ga atoms in light green and subsurface N atoms in light blue. Three [110]-oriented  $\text{NbO}_2$  surface unit cells are drawn showing the three possible alignments with the GaN surface. The three possible alignments result in pseudo-six-fold symmetry for the  $\text{NbO}_2$  film. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

the reaction layer, we deposit four monolayers of Ga metal on the GaN surface prior to the oxygen exposure. This forms a Ga suboxide on oxygen exposure. Depositing Nb metal in oxygen on this Ga suboxide surface appears to prevent the nitridation of Nb and  $\text{NbO}_2$  forms directly on this surface without structural disruption as observed in RHEED (the  $\text{NbO}_2$  film streaks emerge immediately). Fig. 9(b) shows the Nb 3d core level for a 4 nm  $\text{NbO}_2$  film with the Ga metal insert and confirms no  $\text{NbN}$  is formed. This observation points to the possible need

for inserting a buffer layer such as a Zintl phase transition layer [36] to allow for a smooth chemical environment change from GaN to  $\text{NbO}_2$  in order to avoid the nitridation of Nb. Growing at a much higher rate may also inhibit the reaction as once  $\text{NbO}_2$  is formed as a phase, it is stable against reaction with GaN. However, sufficiently high rates may not be accessible in standard oxide MBE deposition systems. Further studies on the growth process are clearly needed to fully utilize the metal-insulator transition in  $\text{NbO}_2$  in GaN-based hyper-FETs.

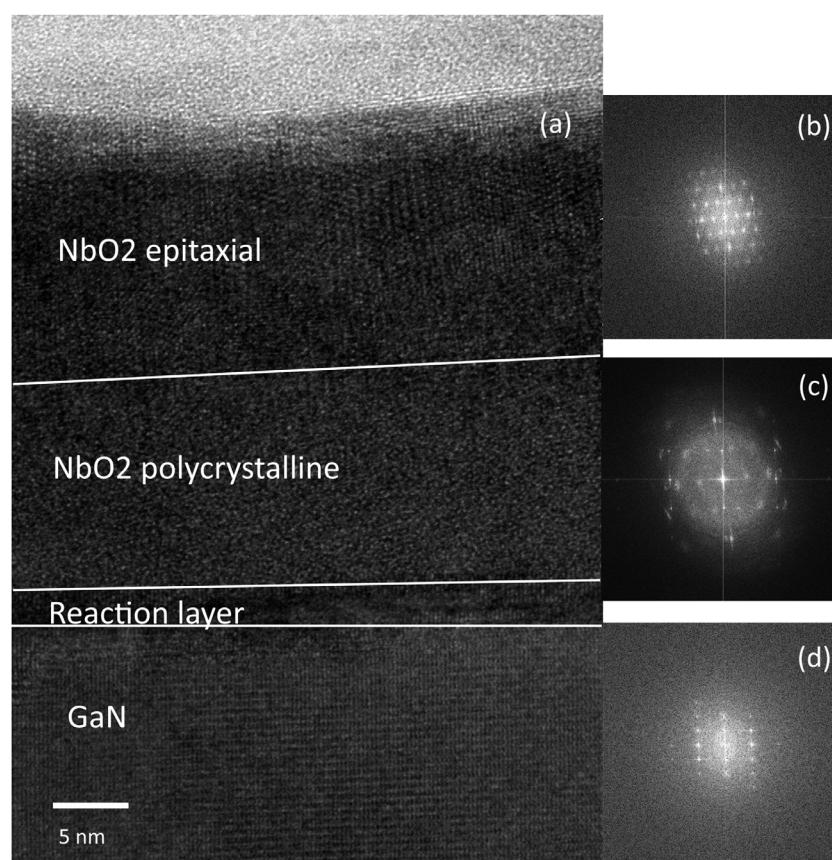
#### 4. Conclusions

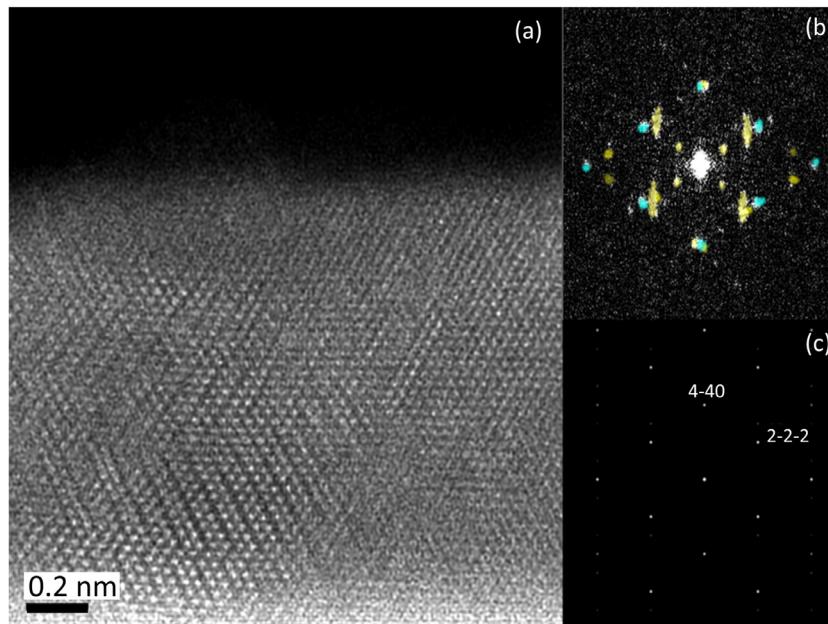
We have grown highly textured crystalline films of  $\text{NbO}_2$  (110) on GaN(0001)/Si(111) substrates. The films have single out of plane orientation but have three in-plane rotational domains with typical size ranging from 15 to 30 nm. We also observe a strong reaction between the incoming Nb and O<sub>2</sub> flux and GaN, resulting in a thin amorphous reaction layer consisting of Ga, Nb, O and N. The reaction also causes randomization of the bottom half of the  $\text{NbO}_2$  film forming a micro-crystalline  $\text{NbO}_x$  layer. A deliberate layer-sequenced approach to growing  $\text{NbO}_2$  on GaN is needed, such as insertion of a Ga metal layer, if one wishes to avoid an interfacial chemical reaction and be able to couple the  $\text{NbO}_2$  film with the GaN substrate.

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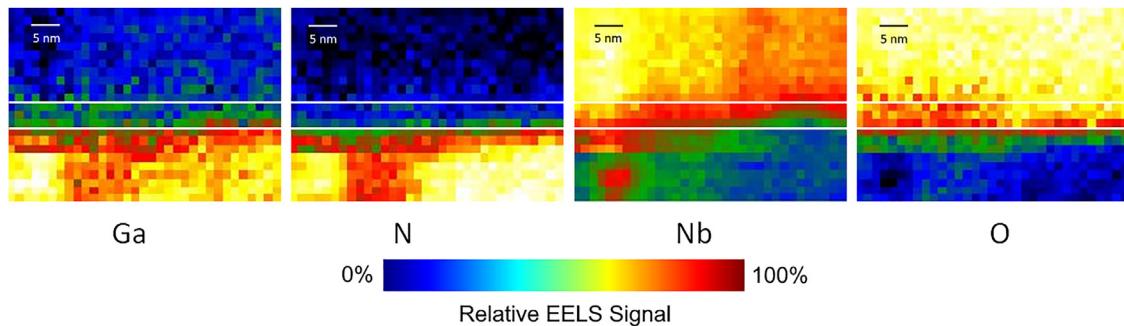
The work in Austin is supported by the National Science Foundation under grant DMR-1507970 and by the Air Force Office of Scientific Research under Grant FA9550-18-1-0053. The authors gratefully acknowledge use of facilities and instrumentation supported by NSF through University of Wisconsin Materials Research Science and Engineering Center (DMR-1720415).

**Fig. 6.** High-resolution TEM image of the layer structure of  $\text{NbO}_2$  grown on GaN. (a) A crystalline  $\text{NbO}_2$  layer is visible as the topmost layer on top of the GaN substrate at the bottom. In between is a polycrystalline  $\text{NbO}_2$  layer with very tiny grains. Between the polycrystalline layer and the substrate is an amorphous reaction layer. (b) Fourier transform of the crystalline  $\text{NbO}_2$  layer. (c) Fourier transform of the entire image showing the polycrystalline arcs from the middle region. (d) Fourier transform of the GaN substrate region.

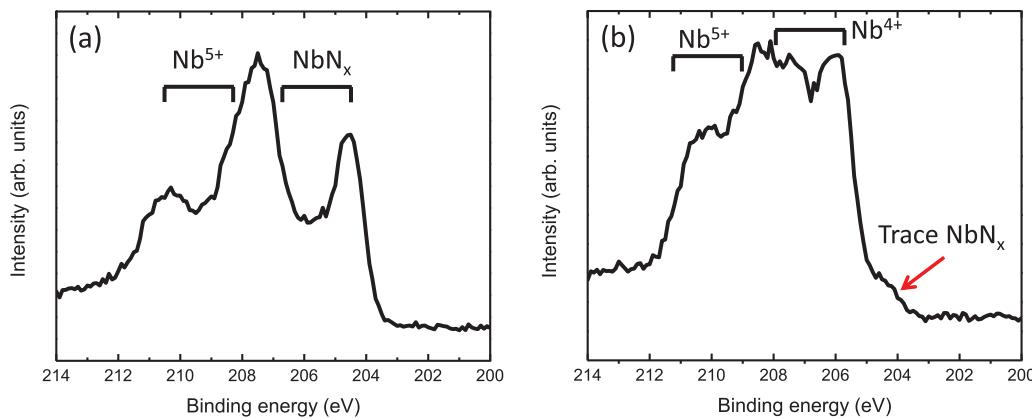




**Fig. 7.** Atomically resolved STEM-HAADF filtered image of  $\text{NbO}_2$  layer. (a) STEM image of  $\text{NbO}_2$  layer showing predominantly 110-oriented grains. A zigzag pattern corresponding to tilted grains is visible on the left side of the image. (b) Fourier transform of the entire image. Reflections marked in blue represent the [110] zone axis of  $\text{NbO}_2$ . Reflections marked in yellow show grains with a slightly different tilt but, on average, align with [110]. (c) Simulated selected area diffraction pattern for the [110] zone axis of  $\text{NbO}_2$ , matching the blue-marked reflections well.



**Fig. 8.** EELS elemental maps of the reaction layer and the substrate. Oranges and yellows indicate high intensity, green indicates medium intensity, and blues and blacks indicate low intensity. The top layer is the polycrystalline portion of the reaction layer. The thin middle layer between the two horizontal white lines is the amorphous reaction layer. The bottom layer is the upper region of the substrate. The polycrystalline portion of the reaction layer is confirmed to be  $\text{NbO}_x$  while the amorphous reaction consists of Nb, O, Ga, and some N. There is also evidence of Nb penetration in the near-surface regions of the substrate. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)



**Fig. 9.** Nb 3d spectra of the interface region of  $\text{NbO}_2$  grown on GaN. (a) Nb 3d spectrum for interface grown using standard deposition process where GaN surface is exposed simultaneously to oxygen gas and Nb metal at high temperature. A significant amount of  $\text{NbN}_x$  and  $\text{Nb}^{5+}$  is observed. (b) Nb 3d spectrum for interface grown with the insertion of four monolayers of Ga metal prior to Nb deposition in oxygen.  $\text{Nb}^{4+}$  is clearly observable and only a small amount of  $\text{NbN}_x$  is formed, but  $\text{Nb}^{5+}$  oxidation state from  $\text{NbGaO}_x$  is still observable.

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