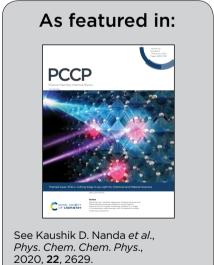


Showcasing research from the groups of Prof. Anna I. Krylov at USC and Prof. Sonia Coriani at DTU.

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This work introduces a new theoretical approach for computing resonant inelastic X-ray scattering (RIXS) cross sections within the equation-of-motion coupled-cluster (EOM-CC) framework. The key idea is to combine core-valence separation (CVS) with damped response theory. By removing the ionization continuum from the Fock space, CVS affords robust convergence of the EOM-CC response equations in the X-ray regime. The capabilities of the new approach are illustrated by spectral calculations of benzene and the benzene radical cation.







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How to stay out of trouble in RIXS calculations within equation-of-motion coupled-cluster damped response theory? Safe hitchhiking in the excitation manifold by means of core-valence separation†

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We present a novel approach for computing resonant inelastic X-ray scattering (RIXS) cross sections within the equation-of-motion coupled-cluster (EOM-CC) framework. The approach is based on recasting the sum-over-states expressions for RIXS moments into closed-form expressions by using damped response theory. Damped response formalism allows one to circumvent problems of divergent behavior of response equations in the resonant regime. However, the convergence of response equations in the X-ray frequency range is often erratic due to the electronically metastable (i.e., resonant) nature of the virtual core-excited states embedded in the valence ionization continuum. We circumvent this problematic behavior by extending the core-valence separation (CVS) scheme, which decouples the valence-excited and core-excited configurations of the excitation manifold, into the response domain. The accuracy of the CVS-enabled damped response theory, implemented within the EOM-EE-CCSD (EOM-CC for excitation energies with single and double excitations) framework, is assessed by comparison against standard damped EOM-EE-CCSD response calculations. The capabilities of the new approach are illustrated by calculations of RIXS cross sections for benzene and benzene radical cation.

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1 Introduction

Resonant inelastic X-ray scattering¹⁻⁵ (RIXS) is a non-linear twophoton spectroscopy based on transitions involving core-level states. As illustrated in Fig. 1, RIXS is a one-photon-in/onephoton-out process. The energy difference between the incident (absorbed) and emitted photons (ω_i and ω_e , respectively) characterizes the difference between the valence electronic states:

$$\omega_i - \omega_e = E_f - E_0. \tag{1}$$

By virtue of using X-ray photons, RIXS exploits the highly localized and element-specific nature of the core orbitals.⁶ Although RIXS is a coherent scattering process, it can be thought of as an initial excitation of a core-level electron into a valence orbital followed by radiative decay of another valence electron to fill the core hole. This molecular orbital picture is shown in Fig. 2 using water as an example: the rightmost panel shows the electronic configuration of the virtual state S_{virtual} and the red and blue arrows illustrate absorption and emission transitions. In this respect, one can (crudely) describe RIXS as simultaneous union of the X-ray absorption and X-ray emission events. This connection between XAS (X-ray absorption spectroscopy), XES (X-ray emission spectroscopy), and RIXS is illustrated in Fig. 2. Thus, RIXS spectroscopy probes the unoccupied valence orbitals as in XAS, the occupied valence orbitals as in XES, and their correlation. Owing to its non-linear nature, the selection rules in RIXS differ from those in linear spectroscopies. For example, RIXS can provide the information about the transitions between valence excited states that are optically dark in onephoton spectroscopies. The spatial localization of core orbitals and their sensitivity to the chemical environment endows RIXS

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[†] Electronic supplementary information (ESI) available: Programmable equations for iterative solutions of damped response equations; LiH data for validation of code with numerical sum-over-states RIXS moments; relevant Cartesian coordinates; raw data for XAS, XES, and RIXS spectra including NTO analysis; basis set used for water; comparison of the experimental results for benzene. See DOI: 10.1039/c9cp03688a

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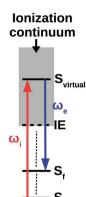


Fig. 1 RIXS is a non-linear process involving two photons. The excitation photon promotes the system from the ground state S_0 into the virtual state S_{virtual} and the system emits a photon by relaxing into an excited state S_f . The difference in the incident and emitted photon frequencies equals the excitation energy of final state (eqn (1)). The system can also relax back to the ground state giving rise to the REXS (resonant elastic X-ray scattering) signal. Whereas the final excited states that give rise to the dominant spectral features lie below the ionization energy and are, therefore, bound with respect to electron ejection, the high-lying core-excited intermediate state S_{virtual} is embedded inside the ionization continuum shown by the gray box (valence resonances above the ionization onset can also contribute to the RIXS spectra).

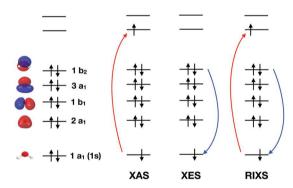


Fig. 2 Molecular orbital picture of the XAS, XES, and RIXS processes illustrated by using water. XAS entails excitation of core electrons into virtual molecular orbitals. The XES signal is produced by the relaxation of valence electrons into the core hole. RIXS process can be thought of as a two-step process coherently combining XAS and XES events.

with the sensitivity to probe the local environment of an atom, its oxidation state, and the bonding pattern around it. Initial applications of RIXS were limited to studying charge-transfer and crystal-field transitions in metal oxides, 4,5 however, advances in X-ray light sources and instrumentation have afforded studying ultrafast nuclear dynamics with RIXS.⁷

The increasing popularity of X-ray spectroscopies⁸⁻¹¹ such as RIXS, XAS, and XES have stimulated the development of quantum-chemical theory and electronic structure tools for modeling these spectra^{12,13} As in the case of VUV-based techniques,¹⁴ theoretical modeling is required to unambiguously relate experimental measurements to molecular structures. However, as nicely described in a recent review¹² of the theoretical modeling of X-ray spectroscopies, accurate description of core states is much more challenging than the description of

valence states. Despite significant progress, the theoretical tools are lagging behind the experimental capabilities, creating a bottleneck for maximizing the scientific impact of multibillion advanced light sources facilities.

The challenges for electronic structure methods in modeling molecular processes in the X-ray regime include the prevalence of open-shell electronic structure patterns, the difficulty in computing high-lying core-excited states with standard solvers. 15-17 orbital relaxation effects due to strongly perturbed atomic core, the sensitivity of the signal to environmental effects, and the metastable character of the core-excited states embedded inside the valence ionization continuum. 18,19 Furthermore, computational modeling of multiphoton X-ray processes, such as RIXS, faces additional challenges. As in the case of other non-linear optical phenomena, the RIXS transition moments formally depend on all states of the system by virtue of the sum-over-states expressions that appear in the second and higher orders of perturbation theory. This has two important consequences. On the fundamental level, the sum over states means that the non-linear properties are more sensitive to the approximations to the many-body problem than the first-order properties—the latter depend only on the wave functions of the two states involved in a transition whereas the former are sensitive to the entire spectrum of a model Hamiltonian. On the practical level, the sum-over-states expressions need to be either truncated (by retaining several, presumably dominant, terms) or recast into a closed form using response theory. Given the resonant nature of RIXS, the former (approximate) approach may provide a qualitatively correct answer, which is often exploited in computational studies using multi-reference wave functions; see e.g., ref. 20-22. However, it is difficult to evaluate a priori the loss of accuracy due to the truncation. Therefore, the response theory approach is more desirable. 12,23,24 Unfortunately, the more rigorous response formulation leads to increased complexity of the equations and computer codes. Finally, one faces the non-convergence of response solutions due to the coupling of response (or virtual) states with the valence ionization

Analytic implementations of the state-of-the-art algebraic diagrammatic construction²³ (ADC) and coupled-cluster²⁴ methods for computing RIXS cross sections for small to medium-size systems within the damped response theory^{25–28} framework have been reported. There is a continuing interest in less expensive approaches based on time-dependent density functional theory^{29,30} (TD-DFT) and DFT-parameterized wave-function methods,³¹ however, the performance of TD-DFT methods in the X-ray domain is often negatively affected by the approximate treatment of the exchange–correlation.^{32,33} Whereas numeric performance of analytic implementations based on damped response theory and correlated wave functions has been discussed and illustrated by examples,^{23,25,26,28} the issue of diverging response solutions has remained hitherto unaddressed.

The equation-of-motion coupled-cluster (EOM-CC) methods^{34–39} provide a robust single-reference multi-state strategy for computing multiconfigurational wave functions, excitation energies, electron affinities, and ionization potentials of closed- and open-shell

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systems. Being a multi-state method, EOM-CC is an excellent platform for modeling spectroscopy because it describes multiple electronic states of different characters in a single computational framework; this leads to a balanced description of the states involved in transitions and also facilitates the calculations of the respective interstate properties. In contrast to multi-reference approaches, EOM-CC does not involve system-specific parameterization (e.g., active-space selection), thus satisfying Pople's requirements for theoretical model chemistry 40 that can be used for systematic studies and comparisons between different systems. The EOM-CC framework yields reliable lower-order properties such as solvatochromic shifts, 41 transition dipole moments, 35 spin-orbit⁴²⁻⁴⁵ and non-adiabatic couplings, 46-48 as well as higher-order properties⁴⁹ such as two-photon absorption cross sections^{50–55} and static and dynamical polarizabilities.^{56–59} Whereas the bulk of prior developments and applications of the EOM-CC methods as well as of the closely related coupled-cluster response theory 60-62 targeted the VUV regime, these methods are now being extended to the X-ray regime and their performance is being explored for computing, for example, XAS, 15-17,19,63-65 XES, 24,66 and RIXS 24,66 spectra.

Building on the (EOM-)CC damped response implementation reported by Faber and Coriani, 24 here we present a novel damped response implementation for computing RIXS cross sections with a variant of the EOM-CC singles and doubles method for electronic excitation^{35–37,67,68} (EOM-EE-CCSD) that provides converged response solutions by judicious exploitation of the excitation manifold. Our strategy to achieve robust convergence of the response equations in RIXS calculations is to employ the core-valence separation (CVS) scheme originally proposed by Cederbaum et al.69 This approach has been employed to compute resolved core-excited and core-ionized states within the framework of EOM-CC and CC-LR response theory, 19,64,70,71 but has not yet been employed for multiphoton X-ray processes such as RIXS. The CVS scheme projects out the valence ionization continuum and enables the resolution of core-excited (real or response) states. Specifically, we extend the recently reported CVS-EOM-EE-CCSD method with frozen-core (fc) approximation (fc-CVS-EOM-EE-CCSD)¹⁹ to the calculation of RIXS cross sections. The same idea is also explored within the coupled-cluster singles and doubles complex polarization propagator approach by Faber and Coriani.66

Here, we present the theory of RIXS cross sections within the EOM-EE-CCSD and fc-CVS-EOM-EE-CCSD methods. We validate the implementation on a small test system (LiH) and then compare the results of EOM-EE-CCSD and fc-CVS-EOM-EE-CCSD RIXS spectra for the water molecule. The method is implemented in the O-Chem electronic structure package. 72,73 Our productionlevel implementation is capable of treating both closed- and openshell references. As an illustrative example, we compute and discuss RIXS spectra of benzene and its cation, highlighting the capabilities of the fc-CVS-EOM-EE-CCSD method. In the present study, we focus exclusively on the electronic factors entering RIXS cross sections; the extension of the theory to include vibronic effects, which are important for quantitative modeling of the spectra, will be carried out in future work.

2 Theory

2.1 CCSD and EOM-EE-CCSD theory

The ground-state coupled-cluster wave function is given by

$$\Psi_0 = e^T |\Phi_0\rangle, \tag{2}$$

where T is the cluster operator given in terms of creation $(a^{\dagger}, b^{\dagger})$ and annihilation (i, j) operators as follows:

$$T = T_1 + T_2; \quad T_1 = \sum_{ia} t_i^a a^{\dagger} i; \quad T_2 = \frac{1}{4} \sum_{iiab} t_{ij}^{ab} a^{\dagger} b^{\dagger} ji.$$
 (3)

The T operator satisfies the following coupled-cluster equations:

$$\langle \Phi_{\nu} | \bar{H} | \Phi_0 \rangle = 0, \tag{4}$$

where ν spans the singles and doubles excitation manifold for CCSD and $\bar{H} = e^{-T}He^{T}$ is the similarity-transformed Hamiltonian. As usual, labels a, b refer to virtual and i, j to occupied molecular (spin-)orbitals.

In the EOM-CCSD approach for excitation energies, the target excited states are described as excitations from the CCSD wave function as follows:

$$L^k \bar{H} = L^k E_k \tag{5}$$

and

$$\bar{H}R^k = E_k R^k, \tag{6}$$

where L and R are the EOM-CCSD left (de-excitation), and EOM-CCSD right (excitation) operators

$$L = l_0 + L_1 + L_2;$$
 $L_1 = \sum_{ia} l_i^a i^{\dagger} a;$ $L_2 = \frac{1}{4} \sum_{ijab} l_{ij}^{ab} i^{\dagger} j^{\dagger} ba$ (7)

and

$$\hat{R} = r_0 + R_1 + R_2; \quad R_1 = \sum_{ia} r_i^a a^{\dagger} i;$$

$$R_2 = \frac{1}{4} \sum_{iiab} r_{ij}^{ab} a^{\dagger} b^{\dagger} j i.$$
(8)

For the CCSD reference state (k = 0), $R = r_0$ and $L = 1 + \Lambda$, where $\Lambda = \Lambda_1 + \Lambda_2$ is the CCSD Λ operator. For EOM-CCSD states, $l_0 = 0$.

2.2 RIXS within EOM-EE-CCSD damped response theory

The RIXS (and REXS) cross section as a function of scattering angle θ (the angle between the polarization vector of the incoming photon and the propagation vector of the outgoing photon) is given in terms of RIXS transition strengths $S_{ab,cd}^{if}$ as follows:

$$\sigma^{\text{RIXS}}(\theta) = \frac{1}{15} \frac{\omega_e}{\omega_i} \sum_{xy} \left[\left(2 - \frac{1}{2} \sin^2 \theta \right) S_{xy,xy}^{if} + \left(\frac{3}{4} \sin^2 \theta - \frac{1}{2} \right) \left(S_{xy,yx}^{if} + S_{xx,yy}^{if} \right) \right],$$

where ω_e and ω_i are frequencies of emitted and incident photons and the indices x, y denote the Cartesian components. 23,24 If θ is defined as the angle between the polarization vector of the

outgoing photon and the propagation vector of the incoming photon, $\cos^2 \theta$ should replace $0.5 \sin^2 \theta$ in the above expression.⁷⁴

Within the non-Hermitian EOM-CC theory, $S_{ab,cd}^{if}$ are given by

$$S_{ab,cd}^{if} = \frac{1}{2} \left(M_{ab}^{i \leftarrow f} M_{cd}^{f \leftarrow i} + (M_{cd}^{i \leftarrow f})^* (M_{ab}^{f \leftarrow i})^* \right), \tag{9}$$

where * denotes complex conjugation. Here, $M_{ab}^{f \leftarrow i}$ and $M_{ab}^{i \leftarrow f}$ are the right and left RIXS moments given by the sum-over-states expressions, which within the EOM-CC framework assume the following form:

$$\begin{split} &M_{xy}^{f-i} \left(\omega_{i,x} + i\gamma, -\omega_{e,y} - i\gamma\right) \\ &= -\sum_{n\geq 0} \left(\frac{\left\langle \Phi_0 L^f e^{-T} \middle| \mu^y \middle| e^T R^n \Phi_0 \right\rangle \left\langle \Phi_0 L^n e^{-T} \middle| \mu^x \middle| e^T R^i \Phi_0 \right\rangle}{\Omega_{ni} - \omega_{i,x} - i\gamma} \right. \\ &+ \frac{\left\langle \Phi_0 L^f e^{-T} \middle| \mu^x \middle| e^T R^n \Phi_0 \right\rangle \left\langle \Phi_0 L^n e^{-T} \middle| \mu^y \middle| e^T R^i \Phi_0 \right\rangle}{\Omega_{ni} + \omega_{e,y} + i\gamma} \right) \end{split}$$

$$(10)$$

and

$$\begin{split} M_{xy}^{i \leftarrow f} \left(-\omega_{i,x} + i\gamma, \omega_{e,y} - i\gamma \right) \\ &= -\sum_{n \geq 0} \left(\frac{\langle \Phi_0 L^i e^{-T} | \mu^x | e^T R^n \Phi_0 \rangle \langle \Phi_0 L^n e^{-T} | \mu^y | e^T R^f \Phi_0 \rangle}{\Omega_{ni} - \omega_{i,x} + i\gamma} \right. \\ &+ \frac{\langle \Phi_0 L^i e^{-T} | \mu^y | e^T R^n \Phi_0 \rangle \langle \Phi_0 L^n e^{-T} | \mu^x | e^T R^f \Phi_0 \rangle}{\Omega_{ni} + \omega_{e,y} - i\gamma} \end{split}$$

$$(11)$$

Here, $\Omega_{nm} = E_n - E_m$ is the energy difference between states n and m and the photon frequencies (ω 's) are augmented with the phenomenological damping term⁷⁷ $i\gamma$.

Introducing the identity operator, $\hat{\bf l}=\sum_{\rho}|\Phi_{\rho}\rangle\langle\Phi_{\rho}|$, in eqn (10) and (11), we obtain

$$M_{xy}^{f \leftarrow i} (\omega_{i,x} + i\gamma, -\omega_{e,y} - i\gamma) = -\sum_{\rho\nu} \left(\langle \Phi_0 L^f | \bar{\mu}^y | \Phi_\rho \rangle \sum_{n \geq 0} \frac{\langle \Phi_\rho | R^n \Phi_0 \rangle \langle \Phi_0 L^n | \Phi_\nu \rangle}{\Omega_{ni} - \omega_{i,x} - i\gamma} \langle \Phi_\nu | \bar{\mu}^x | R^i \Phi_0 \rangle \right.$$

$$\left. + \langle \Phi_0 L^f | \bar{\mu}^x | \Phi_\rho \rangle \sum_{n \geq 0} \frac{\langle \Phi_\rho | R^n \Phi_0 \rangle \langle \Phi_0 L^n | \Phi_\nu \rangle}{\Omega_{ni} + \omega_{e,y} + i\gamma} \langle \Phi_\nu | \bar{\mu}^y | R^i \Phi_0 \rangle \right)$$

$$(12)$$

and

$$M_{xy}^{i \leftarrow f} \left(-\omega_{i,x} + i\gamma, \omega_{e,y} - i\gamma \right) = -\sum_{\rho\nu} \left(\langle \Phi_0 L^i | \bar{\mu}^x | \Phi_\rho \rangle \sum_{n \geq 0} \frac{\langle \Phi_\rho | R^n \Phi_0 \rangle \langle \Phi_0 L^n | \Phi_\nu \rangle}{\Omega_{ni} - \omega_{i,x} + i\gamma} \langle \Phi_\nu | \bar{\mu}^y | R^f \Phi_0 \rangle \right.$$

$$\left. + \langle \Phi_0 L^i | \bar{\mu}^y | \Phi_\rho \rangle \sum_{n \geq 0} \frac{\langle \Phi_\rho | R^n \Phi_0 \rangle \langle \Phi_0 L^n | \Phi_\nu \rangle}{\Omega_{ni} + \omega_{e,y} - i\gamma} \langle \Phi_\nu | \bar{\mu}^x | R^f \Phi_0 \rangle \right).$$

$$(13)$$

Here, $\bar{\mu} = e^{-T}\mu e^{T}$ is the similarity-transformed dipole moment operator and ρ spans the full excitation manifold, however,

only the reference, singly, and doubly excited determinants survive for EOM-CCSD. Using the response intermediates defined as 50

$$[D^x]_{\nu}^n = \langle \Phi_{\nu} | \bar{\mu}^x | R^n \Phi_0 \rangle \tag{14}$$

and

$$[\tilde{D}^x]^n_\rho = \langle \Phi_0 L^n | \bar{\mu}^x | \Phi_\rho \rangle \tag{15}$$

along with solutions, X and \tilde{X} , of response equations given by

$$\sum_{\nu} (\bar{H} - E_0 - (\omega + i\gamma))_{\rho\nu} (X_{Re} + iX_{Im})_{\nu} = D_{\rho}$$
 (16)

and

$$\sum_{\rho} (\tilde{X}_{Re} + i\tilde{X}_{Im})_{\rho} (\bar{H} - E_0 - (\omega + i\gamma))_{\rho\nu} = \tilde{D}_{\nu}, \tag{17}$$

eqn (12) and (13) become

$$M_{xy}^{f \leftarrow i} \left(\omega_{i,x} + i\gamma, -\omega_{e,y} - i\gamma \right)$$

$$= -\sum_{\rho} \left(\left[\tilde{D}^{y} \right]_{\rho}^{f} \left[X^{x} (\omega_{i,x} + i\gamma) \right]_{\rho}^{i} + \left[\tilde{D}^{x} \right]_{\rho}^{f} \left[X^{y} (-\omega_{e,y} - i\gamma) \right]_{\rho}^{i} \right)$$

$$= -\sum_{\rho} \left(\left[\tilde{X}^{y} (-\omega_{e,y} - i\gamma) \right]_{\rho}^{f} \left[D^{x} \right]_{\rho}^{i} + \left[\tilde{X}^{x} (\omega_{i,x} + i\gamma) \right]_{\rho}^{f} \left[D^{y} \right]_{\rho}^{i} \right)$$

$$(18)$$

and

$$M_{xy}^{i \leftarrow f} \left(-\omega_{i,x} + i\gamma, \omega_{e,y} - i\gamma \right)$$

$$= -\sum_{\rho} \left(\left[\tilde{D}^{x} \right]_{\rho}^{i} \left[X^{y} (\omega_{e,y} - i\gamma) \right]_{\rho}^{f} + \left[\tilde{D}^{y} \right]_{\rho}^{i} \left[X^{x} (-\omega_{i,x} + i\gamma) \right]_{\rho}^{f} \right)$$

$$= -\sum_{\rho} \left(\left[\tilde{X}^{x} (-\omega_{i,x} + i\gamma) \right]_{\rho}^{i} \left[D^{y} \right]_{\rho}^{f} + \left[\tilde{X}^{y} (\omega_{e,y} - i\gamma) \right]_{\rho}^{i} \left[D^{x} \right]_{\rho}^{f} \right),$$

$$(19)$$

where $X = X_{\text{Re}} + iX_{\text{Im}}$ and $\tilde{X} = \tilde{X}_{\text{Re}} + i\tilde{X}_{\text{Im}}$. We solve the damped response eqn (16) and (17) by modifying our standard DIIS procedure^{78,79} as described in the ESI.†

We note that the above damped response equations implicitly employ the following resolvent equation:

$$\sum_{n\geq 0} \frac{\langle \Phi_{\rho} | R^n \Phi_0 \rangle \langle \Phi_0 L^n | \Phi_{\nu} \rangle}{\Omega_{ni} - \omega - i\gamma} = \langle \Phi_{\rho} | (\bar{H} - E_0 - \omega - i\gamma)^{-1} | \Phi_{\nu} \rangle, \tag{20}$$

which parallels our formal derivation of the 2PA response equations. 50 Here, the sum runs over all states of the system.

As a two-photon process, RIXS involves a virtual state (Fig. 1). Owing to the resonant nature of RIXS, the corresponding virtual state often has dominant contributions from real core-excited high-lying states, which are metastable with respect to electron ejection. Specifically, they are Feshbach resonances embedded in the valence ionization continuum. This presence of the continuum cripples the convergence of the response solutions, X and \tilde{X} , in eqn (16) and (17). Specifically, the problem manifests itself by oscillatory behavior of doubly excited amplitudes and by their large magnitude. We note that this problem does not arise in low-level theories without explicit double excitations^{18,66} such as CIS/TDDFT, ADC(2), or CC2 because the decay channels of core-level

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states involve two-electron transitions. The damped response theory framework only mitigates the convergence issues due to resonant photons (i.e., singularities in the resolvent) but does not address the impact of the coupling with the continuum. To circumvent this problem, we employ the CVS scheme within the response manifold.

2.3 RIXS within fc-CVS-EOM-EE-CCSD damped response theory

Within the fc-CVS-EOM-CCSD framework, we first impose the frozen-core (fc) approximation and create distinct core occupied (C), valence occupied (O), and unoccupied (V) subspaces. As reported in ref. 19, the reference, CV, COVV, and CCVV excitation manifolds are then used for computing the fc-CVS-EOM-CCSD states (below we refer to this configurational space as the CVS manifold). Note that in the fc-CVS-EOM-CCSD scheme, the OV and OOVV subspaces and, consequently, the valence continuum have been projected out, which blocks the decay channels and makes the fc-CVS-EOM-CCSD states bound. This separation of the bound and continuum configurations within CVS is akin to diabatization. It is also related to the Feshbach-Fano treatment of resonances. 80,81

In the context of calculating RIXS moments, we employ the CVS scheme to approximate the sum-over-states expressions in egn (10) and (11) such that only the fc-CVS-EOM-EE-CCSD states are included in the sum. The fc-CVS-EOM-CCSD resolvent equation is given by

$$\frac{\langle \Phi_{\rho} | \Phi_{0} \rangle \langle \Phi_{0} | \Phi_{\nu} \rangle}{-\omega - i\gamma} + \sum_{n}^{\text{CVS}} \frac{\langle \Phi_{\rho} | R^{n} \Phi_{0} \rangle \langle \Phi_{0} L^{n} | \Phi_{\nu} \rangle}{\Omega_{ni} - \omega - i\gamma} =$$

$$\langle \Phi_{\rho} | (\bar{H}^{\text{cvs}} - E_{0} - \omega - i\gamma)^{-1} | \Phi_{\nu} \rangle,$$
(21)

where the \bar{H}^{cvs} is written in the basis of the reference and the CVS excitation manifold (CV, COVV, and CCVV configurations). Eqn (21) reduces to

$$\begin{split} &\sum_{n}^{\text{CVS}} \frac{\langle \Phi_{\rho} | R^{n} \Phi_{0} \rangle \langle \Phi_{0} L^{n} | \Phi_{\nu} \rangle}{\Omega_{ni} - \omega - i \gamma} = \\ &\langle \Phi_{\rho} | (\bar{H}^{\text{cvs}} - E_{0} - \omega - i \gamma)^{-1} | \Phi_{\nu} \rangle - \frac{\delta_{0\rho} \delta_{0\nu}}{-\omega - i \gamma} \end{split}$$

The fc-CVS-EOM-EE-CCSD response equations are thus given by

$$\sum_{\nu} (\bar{H}^{\text{cvs}} - E_0 - (\omega + i\gamma))_{\rho\nu} (X_{\text{Re}} + iX_{\text{Im}})_{\nu}$$

$$= (1 - |\Phi_0\rangle\langle\Phi_0|)_{\rho\lambda} D_{\lambda}$$
(22)

and

$$\sum_{\rho} (\tilde{X}_{Re} + i\tilde{X}_{Im})_{\rho} (\bar{H}^{cvs} - E_0 - (\omega + i\gamma))_{\rho\nu}$$

$$= \tilde{D}_{\lambda} (1 - |\Phi_0\rangle \langle \Phi_0|)_{\lambda\nu}, \qquad (23)$$

where ρ , λ , and ν span the reference, CV, COVV, and CCVV excitation manifolds.

3 Computational details

The validation calculations for LiH were carried out using the STO-3G basis and r_{LiH} = 1.6 Å. Water calculations were carried out using the same setup as in ref. 24. Specifically, we computed RIXS cross sections including the lowest 20 excited states, using the geometry ($r_{OH} = 0.95421 \text{ Å}, \angle_{HOH} = 104.78^{\circ}$) and basis set from ref. 24 (also given in the ESI†). We use the EOM-EE-CCSD results reported at the incident photon frequency of 535.74 eV that is resonant with the $1s \rightarrow 4a_1$ core excitation in water. The corresponding fc-CVS-EOM-EE-CCSD incident photon frequency used in our calculations is 535.23 eV.

The calculations for neutral benzene were carried out at the geometry optimized with RI-MP2/cc-pVTZ. The calculations for the cation were carried out at two different geometries: one of the neutral (referred to as Franck-Condon (FC) structure) and one of the cation (optimized with EOM-IP-CCSD/cc-pVTZ). We considered the optimized geometry of the ²B_{3g} state, which is the lowest Jahn-Teller state (this structure can be described as "elongated" in terms of ref. 82 or "acute D_{2h} " in terms of ref. 83). The optimized structure of the second Jahn-Teller state, ${}^{2}B_{2g}$, is nearly degenerate with the 2B3g structure and the barrier separating the two minima is well below zero-point energy.83 Consequently, benzene cation's structure can be described by free pseudo-rotation.83 We used the 6-311(2+,+)G** basis with the uncontracted carbon core (denoted as 6-311(2+,+)G**(uC)).

Depending on molecular orientation, symmetry labels corresponding to the same orbital or vibrational mode may be different. Q-Chem's standard molecular orientation is different from that of Mulliken.84 For example, Q-Chem places water molecule in the xz plane instead of the yz. Consequently, for C_{2y} symmetry, b_1 and b_2 labels are flipped. Q-Chem places benzene in the xy plane (with the x-axis passing through the carbon atoms, see ESI†) instead of the yz plane. Consequently, the b_{1u} and b_{3u} labels are flipped in Q-Chem relative to the Mulliken convention. More details can be found at http:// iopenshell.usc.edu/resources/howto/symmetry/. To avoid confusion with different molecular orientations and relabeling the states, here we report the structures and symmetry labels following Q-Chem's notations.

The calculations of XAS were carried out using fc-CVS-EOM-EE-CCSD. For the cation, we tested restricted and unrestricted open-shell Hartree-Fock (ROHF and UHF, respectively) references and the difference between the two calculations was found to be small. Below we report the ROHF results. The XES transitions between different core and valence states were computed as the transitions between valence- and core-ionized states of the neutral reference, as suggested in ref. 24. The valence states were described by EOM-IP-CCSD with frozen core and the core states were described by fc-CVS-EOM-IP-CCSD. The orbital character of the XAS and XES transitions was analyzed using Natural Transition Orbitals computed with the libwfa module⁸⁵ and Dyson orbitals.^{86,87}

The RIXS spectra were computed with RHF references for closedshell systems and UHF references for open-shell systems using the new fc-CVS-EOM-EE-CCSD scheme. All RIXS calculations for Paper PCCP

benzene were performed with γ = 0.01 a.u. = 0.272 eV. The cation's RIXS calculations were performed with γ = 0.005 a.u. = 0.136 eV.

The stick spectra were convoluted with normalized Gaussian functions:

$$g(x) = \frac{1}{\sigma\sqrt{2\pi}} \exp\left(-\frac{(x-x_0)^2}{2\sigma^2}\right),\tag{24}$$

$$FWHM = 2\sqrt{2\ln(2)}\sigma \approx 2.35482\sigma. \tag{25}$$

4 Results and discussion

4.1 Validation and benchmarks

The calculation of RIXS cross sections using damped response theory combined with CVS is implemented in the Q-Chem electronic structure package. Our production-level implementation builds upon the fc-CVS-EOM-CCSD suite of codes and the non-linear response module. The programmable equations are implemented using the general library for tensor computations, libtensor. The implementation enables calculation of RIXS cross sections for both closed- and open-shell references. We validated our analytic damped response theory approach against numeric sum-over-states calculations for the first excited state of LiH using the STO-3G basis; the results are given in the ESI.† The standard (*i.e.*, full-valence) EOM-EE-CCSD implementation was also validated against ref. 24.

To quantify the effect of using CVS within the response equations on the RIXS spectra, we compare the fc-CVS-EOM-EE-CCSD and standard EOM-EE-CCSD results for the water molecule. The incident photon frequency in each calculation is resonant with the $1s_O \rightarrow 4a_1$ core excitation. For this incident photon, the dominant RIXS transitions correspond to final

states $1B_2$, $2A_1$, and $3A_1$ with orbital characters of $1b_2 \rightarrow 4a_1$, $3a_1 \rightarrow 4a_1$, and $1b_2 \rightarrow 2b_2$ (in Q-Chem's symmetry notation), respectively, as presented in Table 1. Table 1 reveals that using the fc-CVS scheme has a minor impact on the RIXS cross sections for these dominant transitions in the spectrum. Although the effect on the cross sections of several minor bands is larger, the appearance of the overall computed spectrum is not affected by invoking CVS, as illustrated in Fig. 3 and by the two-dimensional RIXS map in the ESI,† which looks very similar to the one reported in ref. 24. The changes in relative intensities of the major peaks due to different scattering angles (θs) are also captured correctly.

Here we limit the comparison of the computed RIXS spectra to the full valence EOM-CCSD damped response calculations, ²⁴

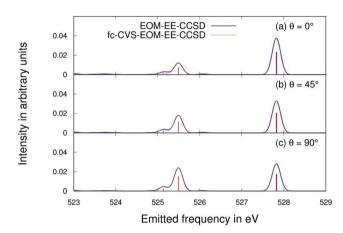


Fig. 3 Comparison of EOM-EE-CCSD and fc-CVS-EOM-EE-CCSD RIXS spectra for the incident photon resonant with the 1s \rightarrow 4a₁ core excitation in water (FWHM = 0.33 eV). (a) θ = 0°, (b) θ = 45°, (c) θ = 90°.

Table 1 Comparison of EOM-EE-CCSD and fc-CVS-EOM-EE-CCSD excitation energies and RIXS cross sections for H_2O ($\gamma=0.0045563$ a.u.). Incident photon frequency for EOM-EE-CCSD calculations is 19.6881298 a.u. (535.74 eV). Incident photon frequency for fc-CVS-EOM-EE-CCSD calculations is 19.6692301 a.u. (535.23 eV). The columns marked 'ratio' give the ratio of computed RIXS cross sections with and without CVS

	EOM-EE-CCSD				fc-CVS-EOM-EE-CCSD						
Ex. st. ^a	$E_{\rm ex}$ (eV)	$\sigma_{0^{\circ}}^{\rm RIXS} ({\rm a.u.})$	$\sigma_{45^{\circ}}^{RIXS}\left(\mathrm{a.u.}\right)$	$\sigma_{90^{\circ}}^{\rm RIXS} (\rm a.u.)$	$E_{\rm ex}$ (eV)	$\sigma_{0^{\circ}}^{\rm RIXS} ({\rm a.u.})$	Ratio	$\sigma_{45^{\circ}}^{RIXS}\left(\mathrm{a.u.}\right)$	Ratio	$\sigma_{90^{\circ}}^{\rm RIXS} (\rm a.u.)$	Ratio
$\overline{1}B_2$	7.41	0.023556	0.020601	0.017646	7.40	0.023646	1.00	0.020690	1.00	0.017734	1.00
$1A_2$	9.16	0.000491	0.000391	0.000292	9.15	0.000180	0.37	0.000157	0.40	0.000135	0.46
$2A_1$	9.74	0.007563	0.011340	0.015116	9.74	0.007645	1.01	0.011462	1.01	0.015280	1.01
$2B_2$	10.01	0.000759	0.000576	0.000393	10.01	0.000053	0.07	0.000047	0.08	0.000040	0.10
$3A_1$	10.10	0.001288	0.001930	0.002573	10.10	0.001314	1.02	0.001970	1.02	0.002626	1.02
$3B_2$	10.39	0.000253	0.000222	0.000190	10.38	0.000257	1.02	0.000225	1.01	0.000193	1.01
$2A_2$	10.80	0.000015	0.000015	0.000015	10.79	0.000029	1.90	0.000025	1.70	0.000021	1.47
$4B_2$	11.21	0.000159	0.000120	0.000082	11.21	0.000009	0.05	0.000007	0.06	0.000006	0.08
$4A_1$	11.22	0.000011	0.000015	0.000020	11.21	0.000011	0.99	0.000016	1.04	0.000020	1.04
$5B_2$	11.30	0.000062	0.000054	0.000046	11.30	0.000060	0.96	0.000052	0.97	0.000045	0.98
$3A_2$	11.43	0.000038	0.000030	0.000021	11.42	0.000010	0.25	0.000008	0.28	0.000007	0.34
$1B_1$	11.50	0.000430	0.000339	0.000248	11.50	0.000130	0.30	0.000114	0.34	0.000097	0.39
$6B_2$	11.68	0.000092	0.000072	0.000051	11.67	0.000016	0.18	0.000014	0.20	0.000012	0.24
$5A_1$	11.69	0.000002	0.000003	0.000004	11.69	0.000002	1.21	0.000003	1.10	0.000004	1.10
$7B_2$	11.72	0.000076	0.000060	0.000043	11.71	0.000016	0.21	0.000014	0.23	0.000012	0.27
$4A_2$	11.83	0.000155	0.000117	0.000079	11.81	0.000005	0.03	0.000004	0.04	0.000004	0.05
$8B_2$	12.20	0.000641	0.000483	0.000324	12.20	0.000013	0.02	0.000012	0.02	0.000010	0.03
$6A_1$	12.26	0.000046	0.000067	0.000088	12.26	0.000045	0.98	0.000066	0.98	0.000087	0.98
$7A_1$	12.66	0.000113	0.000169	0.000224	12.66	0.000118	1.04	0.000175	1.04	0.000233	1.04
$9B_2$	13.07	0.000298	0.000247	0.000195	13.06	0.000191	0.64	0.000167	0.68	0.000144	0.74

^a Q-Chem's symmetry notations.

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because, although water is a popular benchmark molecule for theory, 23,24 it is not well suited for making comparisons with the experiments. The complications arise due to an anomalously strong effect of nuclear motions on its RIXS spectra.²² Similarly to valence ionization, 89 core-level excitation of water leads to ultrafast dissociation. 22 Because stretching of the OH bond strongly affects valence electronic states, 22,33 this ultrafast motion has strong effect on the RIXS and XES lineshapes. Consequently, RIXS spectra for water can only be quantitatively understood when nuclear degrees of freedom are properly included in the simulation, as was done in ref. 22 and 90. We also note that in order to reproduce features at higher energy loss, more excited states need to be included in the calculations. For example, to reproduce the third dominant peak in the spectrum, which is not captured by the present calculations with 20 excited states, Faber and Coriani included 40 excited states.66

4.2 Illustrative calculation: RIXS of benzene and benzene cation

4.2.1 Benzene. Modeling X-ray spectra of benzene is challenging due to its high symmetry, multiple core orbitals, and the need for very diffuse functions to describe Rydberg excited states. Fig. 4 shows occupied molecular orbitals of benzene; the respective vertical ionization energies (IEs) computed with EOM-IP-CCSD at the geometry of the neutral are collected in

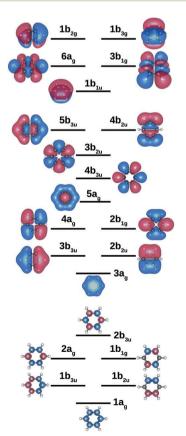


Fig. 4 Occupied molecular orbital diagram for benzene.

Table S2 of the ESI.† In this case, the correlated Dyson orbitals are visually indistinguishable from the canonical Hartree-Fock molecular orbitals and the correlated EOM-IP states can be mapped onto the latter. In the discussion below, we use molecular orbitals from Fig. 4 to refer to many-body EOM-IP states and vice versa.

Due to symmetry, all six 1s_C core orbitals of benzene are delocalized, however, the splittings of the respective IEs are small (all six lie within 0.08 eV) because of their compactness. Table S2 (ESI†) also lists the IEs of the valence orbitals; the lowest IE of 9.2 eV corresponds to the degenerate pair of π orbitals, which give rise to the Jahn-Teller distortion. These values of IEs provide the maximum range of energy loss expected in XES and RIXS calculations from valence excited states; a higher energy loss may result from valence resonances.

The XAS spectrum of neutral benzene is shown in Fig. 5; the respective transition energies and strengths are collected in Table S3 of the ESI.† The positions and relative intensities of the main peaks agree well with the experiment 91,92 after a systematic shift of -0.8 eV is applied. Fig. 5 also shows the leading NTOs for the first two peaks; the NTOs for other transitions are collected in the ESI.† The NTOs reveal that peak A is derived from the $1s_C \to \pi^*(B_{1u})$ transition, whereas doubly degenerate peak B corresponds to the $1s_C \rightarrow Ry(B_{2u}/B_{3u})$ transition. The NTO analysis shows that excitation of peak A creates a hole in the $2a_g$ and $1b_{1g}$ core orbitals, whereas the excitation of peak B creates holes in $1b_{2u}/1b_{3u}$ and in an orbital, which can be described as a linear combination of $1a_g$ and $2a_g$. The weights of the NTOs (i.e., respective σ^2 values) quantify the relative contribution of each hole. Although the energies of the core orbitals giving rise to peak A and peak B are very close, their different symmetries have a profound effect on the shape of the RIXS emission spectra corresponding to pumping these two peaks.

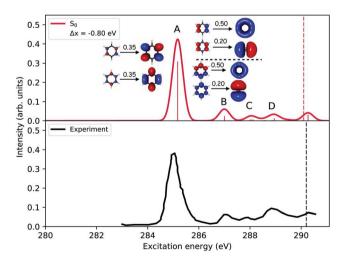


Fig. 5 XAS spectrum of benzene computed with fc-CVS-EOM-CCSD/6- $311(2+,+)G^{**}(uC)$; FWHM = 0.8 eV. Dashed vertical lines correspond to the IEs. The energy shift required to align the NEXAFS profiles with the experimental one is 0.8 eV. The computed IE has been shifted by the same amount as used to align the NEXAFS profiles. NTOs and their weights are shown for peaks A and B. The experimental spectrum is from ref. 91.

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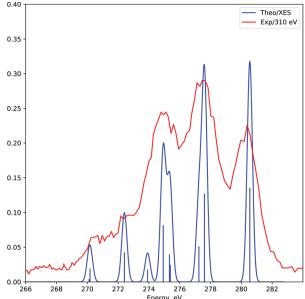


Fig. 6 XES transitions in the neutral benzene corresponding to ionization from all core orbitals. Blue: Theoretical spectrum computed with fc-CVS-EOM-IP-CCSD and fc-EOM-IP-CCSD using $6-311(2+,+)G^{**}(uC)$ and shifted by 1.2 eV (FWHM = 0.4 eV). Red: experimental spectrum⁹³ corresponding to 310 eV excitation. Intensities are in arbitrary units.

The non-resonant X-ray emission of benzene could result from ionization of any of the six core orbitals. Thus, the XES spectrum can be computed as the sum of all possible transitions between the valence- and six lowest core-ionized states. The result is shown in Fig. 6. Apart from the overall shift of -1.2 eV, the computed spectrum agrees very well with the experimental non-resonant emission spectrum. ⁹³ As pointed out in ref. 93, the structure of the XES spectrum can only be explained in terms of transitions to the delocalized core orbitals of proper symmetries (the localized 1s holes would yield an entirely different pattern); our calculations provide a numeric confirmation of this conclusion.

In contrast to XES, the RIXS emission spectrum is determined by the shape and symmetry of the specific core hole created in the excitation step as well as the (spectator) excited electron; we discuss these effects below. As suggested by Fig. 2, the RIXS process can be crudely described as a union of XAS excitation followed by the emission via XES-like transitions. For a specific core hole created in the excitation step (say, core hole A created by pumping peak A), assuming that the removal of the excited electron does not strongly perturb the (N-1)-electron core, the RIXS emission spectra of the neutral benzene can be approximated by computing the XES transitions between the specific core- and all valence-ionized states. In other words, we synthesize an approximate RIXS emission spectrum by ignoring the effect of the spectator electron and by choosing the specific core hole that modulates the intensities of the XES peaks from Fig. 6. Below, we refer to the so-computed emission spectra as "poor man's RIXS emission". The purpose of these calculations is pedagogical: they illustrate the origin of the dependence of the RIXS emission spectra on the core hole (or pumping frequency) and provide simple molecular orbital picture of RIXS. Of course, we anticipate that not all features of the proper RIXS spectra will be captured by such calculations; peak intensities and positions may be affected by the spectator electron. The comparison of the properly computed RIXS spectra with poor man's RIXS will highlight fundamental aspects of the RIXS process: its coherent two-photon nature and the effect of the spectator electron.

The NTO analysis of the XAS spectrum of benzene (Fig. 5) and Table S4 in the ESI†) shows the core hole orbitals corresponding to peaks A and B. Although the shapes of NTOs are slightly different from the Hartree-Fock orbitals (for example, the hole orbitals from the second NTO pairs of the two degenerate transitions giving rise to peak B show mixing of $1a_{\rm g}$ and $2a_{\rm g}$ canonical molecular orbitals from Fig. 4), one can still map the NTOs into the canonical and/or Dyson orbitals from Fig. 4. We use these orbitals (and the respective NTO weights) to compute the corresponding poor man's RIXS emission; the resulting emission spectra are shown in Fig. 7. The NTO analysis of the corresponding transitions is given in Table S5 of the ESI;† it confirms that the molecular orbitals from Fig. 4 provide qualitatively correct description of the transitions. The two poor man's RIXS emission spectra show significant differences: the relaxation to core hole A yields only three features, whereas the relaxation to core hole B shows more features dominated by the one due to the relaxation of the HOMO to the core hole B around 282 eV. This is not surprising as the symmetry of the core hole dictates the symmetry of the valence orbitals that are active in poor man's RIXS (and proper RIXS) emission. For example, the presence of the b_{3u} core hole

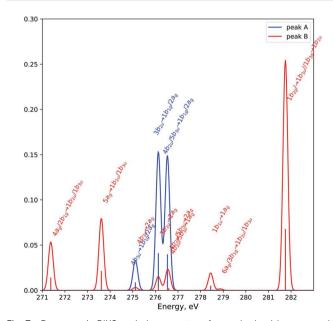


Fig. 7 Poor man's RIXS emission spectra of core-ionized benzene with the core holes of XAS peaks A (blue) and B (red) with FWHM = 0.25 eV. Intensities are in arbitrary units. The oscillator strengths are computed between fc-CVS-EOM-IP-CCSD and fc-EOM-IP-CCSD states with 6-311(2+,+)G**(uC) basis set. Note that the actual computed XES spectrum of benzene in Fig. 6 has the same peak positions (before the overall shift of -1.2 eV) as in poor man's RIXS emission spectra but the intensities are different.

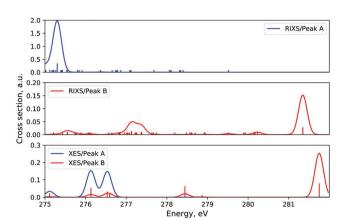


Fig. 8 Benzene: computed RIXS spectra corresponding to pumping peak A (285.97 eV) and peak B (287.80 eV) with θ = 0 and the corresponding poor man's RIXS emission spectra (FWHM = 0.25 eV). The positions of the excited states included in RIXS calculations are shown by sticks (sticks that give zero intensity are given some small finite height)

allows only the relaxation from the $a_{\rm g}$, $b_{\rm 1g}$, and $b_{\rm 2g}$ valence levels (in D_{2h} group, only B_{1u} , B_{2u} , and B_{3u} transitions are optically allowed). These calculations suggest that the properly computed RIXS emission spectra should also show strong dependence on the pumping frequency.

The properly computed RIXS emission spectra corresponding to pumping peak A and peak B computed with fc-CVS-EOM-EE-CCSD are shown in Fig. 8. The selection rules for the two-photon RIXS transitions in benzene (D_{2h}) allow only gerade transitions within the dipole approximation. The RIXS spectrum for the two pumping frequencies is computed using the same set of valence excited states (20 excited states per irrep); their positions and contributions are shown by sticks in the top two panels of Fig. 8. Note that the positions of the sticks in the two panels are offset by 1.83 eV by virtue of eqn (1). The difference between the two RIXS spectra, corresponding to the resonant $1s_C \rightarrow \pi^*$ core excitation (peak A) and $1s_C \rightarrow Ry$ core excitation (peak B), is stark. As one can see, the relative intensities of the RIXS lines corresponding to the same excited states are vastly different; compare, for example, the first line from the right corresponding to the doubly degenerate RIXS transitions to the $1B_{2g}$ and $1B_{3g}$ excited states at 6.45 eV (HOMO-LUMO $\pi \to \pi^*$ excitation). Because of the different symmetry of the core hole created by pumping peak A and peak B, these transitions have nearly zero intensity when peak A is pumped and become a dominant feature of the spectrum when peak B is pumped. This RIXS peak at 281.35 eV emission energy for pumping peak B also appears in the corresponding poor man's RIXS emission spectrum at 281.75 eV, illustrating a spectator shift of -0.40 eV. Similarly, the dominant RIXS transition for pumping peak A shows up at 275.30 eV emission energy (10.67 eV energy loss) and corresponds to doubly degenerate transitions to $13B_{2g}$ and $12B_{3g}$ states. These excited-state RIXS transitions are the second-most intense for the pumping frequency B and show up at 277.13 eV emission energy. The poor man's RIXS emission spectra reproduces this feature, albeit at 276.53 eV. Thus, the spectator shifts for pumping frequencies A and B are -1.23 eV and 0.60 eV,

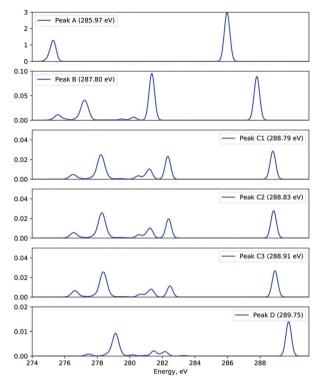


Fig. 9 RIXS spectra corresponding to excitation of peaks A-D computed with fc-CVS-EOM-CCSD/6-311(2+,+) $G^{**}(uC)$: FWHM = 0.4 eV. The y-axes shows absolute RIXS cross sections in atomic units.

respectively. The poor man's RIXS emission spectra for pumping frequency A shows a dominant feature at 276.13 eV (labeled $3b_{2u} \rightarrow 1b_{1g}/2a_g$). In the properly computed RIXS spectrum, this feature due to the net transitions to the $16B_{2g}$ and $16B_{3g}$ states at 10.85 eV energy loss is no longer as dominant; the primary reason being the smaller dipole coupling of the response state with the final state due to the presence of the spectator electron.

Fig. 9 shows the (properly computed) RIXS spectra corresponding to pumping peaks A-D; the full two-dimensional RIXS energy-loss maps are shown in Fig. 10 and in Fig. S3 in the ESI.† The computed RIXS cross sections at selected pumping frequencies are given in Tables S6-S8 of the ESI.† As expected,

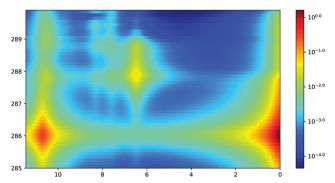


Fig. 10 Computed RIXS/REXS two-dimensional energy-loss (X axis) spectra with the intensities shown on the logarithmic scale; fc-CVS-EOM-CCSD/6-

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the brightest RIXS signal is obtained by pumping peak A, which dominates the XAS spectrum. Scanning the pumping frequency leads to pronounced changes in peak positions and relative intensities, as anticipated above. The RIXS spectrum obtained for pumping frequency A, in particular, shows stark differences compared to the spectra for higher energy pumping frequencies. For example, the RIXS peak around the 6.45 eV energy loss does not appear for pumping frequency A. Also, the elastic scattering peak (at zero energy loss) and the inelastic peak at 10.67 eV $(1A_g \rightarrow 13B_{2g}/12B_{3g})$ are the most intense at this incident frequency and are 1 to 2 orders of magnitude more intense than the brightest emissions for higher energy pumping frequencies, in agreement with the experimental findings.⁹³ The spectra corresponding to pumping peaks B, C1, C2, C3, and D appear very similar. The energy-loss peaks at 6.45 eV $\left(1A_{\rm g} \rightarrow 1B_{\rm 2g}/1B_{\rm 3g}\right)$ and 10.67 eV $\left(1A_{\rm g} \rightarrow 13B_{\rm 2g}/12B_{\rm 3g}\right)$ are important in the RIXS spectra for pumping frequencies B and C1/C2/C3. The energy-loss peak at 6.45 eV, however, is not significant for pumping frequency D. For the B and C1/C2/C3 pumping frequencies, the energy-loss peak at 7.57 eV ($1A_g \rightarrow 11B_{2g}/11B_{3g}$) is also noticeable.

We conclude this section by comparing the computed RIXS spectra with the available experimental data. 93-95 Fig. S3 in the ESI† shows the RIXS spectra from the gas-phase 93 and condensed-phase studies. 4 As one can see, the spectra show significant changes as a function of pumping frequency. We also note that, despite general agreement between the two experiments, there are some discrepancies, especially in relative peak intensities. A close inspection of the available data shows that the scan at 285 eV should correspond to pumping peak A and the other scans correspond to pumping peaks C-D. Unfortunately, the reported data 93,94 do not include excitation frequency corresponding to peak B. Thus, we limit our comparisons to the two sets from ref. 93. Fig. 11 compares the computed RIXS spectra corresponding to pumping peaks A and C1 with the experimental energy loss spectra for 285 and 288.2 eV. As one

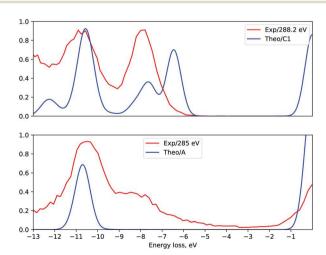


Fig. 11 Benzene RIXS energy-loss spectra (arbitrary intensities). Theory (blue): energy loss corresponding to pumping peaks A and C1 (FWHM = 0.8 eV). The experimental spectra⁹³ are shown in red.

can see, the agreement between theory and experiment is reasonably good—the positions of major features are reproduced well. In the low-energy spectrum, theory does not capture the shoulder at -8.5 eV, whereas for the spectra at 288.2 eV pump, there is a discrepancy for the peak at -8.0 eV. One possible explanation of these (relatively small) discrepancies is a small mismatch between the reported experimental pumping frequencies and the XAS peak positions; given the strong dependence of the emission spectra on the pumping frequency, this can explain the disagreement. Alternatively, these difference may be due to nuclear motions, which are not included in the present theoretical framework and are believed to be important in RIXS spectra of benzene.⁹⁴

4.2.2 Benzene cation. XAS and RIXS spectra of open-shell species derived by the ionization of closed-shell molecules acquire new, distinguishing features due to the presence of the valence hole. The excitation of the core electrons can populate this valence hole, giving rise to a characteristic lower energy peak in the XAS spectra of the open-shell species. The resonant excitation of this peak in RIXS provides a means to obtain the energy-loss spectrum characteristic of the open-shell species, which can be thus probed in the presence of the respective neutral parent molecules, even when the latter are present in great excess. In particular, the characteristic features of open-shell RIXS energy-loss spectra corresponding to excitation to the valence singly occupied orbital and the XES-like spectrum of the parent closed-shell molecule with the same core hole (poor man's RIXS for the cation) are expected to be similar, if computed at the same geometry. Note that this similarity between XES-like spectrum of the neutral and RIXS of the cation only holds for the excitation frequency that corresponds to the transition filling the valence hole; the RIXS emission spectra corresponding to excitations into higher lying unoccupied orbitals would be more similar to the emission spectra of the dication. Here, we illustrate these essential relationships between the closed-shell molecules and the respective ionized species by considering XAS and RIXS of benzene and its cation.

Benzene cation is a classic Jahn–Teller system⁸³ with doubly degenerate (at the Franck–Condon geometry) ground state. Jahn–Teller distortions (of 0.2 eV) lead to two nearly degenerate minima, with the barrier between them well below the zero-point energy.⁸³

Fig. 12 shows the XAS spectra of the cation (computed for the ${}^2\mathrm{B}_{3\mathrm{g}}$ state) and the respective NTOs for the lowest spectral feature. At the geometry of the neutral, peak A corresponds to the excitation to the valence hole (transitions between $1b_{2\mathrm{u}}/1b_{3\mathrm{u}}$ core and $1b_{3\mathrm{g}}/1b_{2\mathrm{g}}$ HOMO), while peaks C and D resemble the transitions in the neutral species (*i.e.*, peaks A and B of the XAS of the neutral). Structural relaxation has a rather small effect on the spectral shape: peak A is blue-shifted by 0.3 eV. The results for all spectral lines are given in Tables S9 and S10 of the ESI.†

Fig. 13 compares the RIXS emission spectrum of the benzene cation computed for pumping transition corresponding to the excitation into the valence hole (*i.e.*, with incident photon resonant with peak A of the cation) with the corresponding poor

0.0 278

280

282

0.5 Bz + (B_{3g}) @ B_{3g} min Bz+ (B_{3g}) @ FC 0.4 ntensity (arb. units)

Fig. 12 XAS spectra of neutral benzene and its cation computed with fc-CVS-EOMEE-CCSD/6-311(2+,+)G**(uC) at the Franck-Condon geometry and at the optimized geometry of the ${}^{2}B_{3q}$ ionized state; FWHM = 0.8 eV.

286

Excitation energy (eV)

288

290

292

284

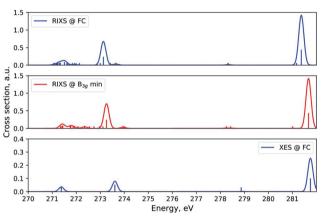


Fig. 13 Benzene cation (${}^{2}B_{3q}$). RIXS spectra corresponding to the valence pump computed at the FC geometry (281.37 eV pump) and at the optimized geometry of the cation (²B_{3g} minimum, 281.67 eV pump) at θ = 0 and the corresponding poor man's RIXS spectrum (marked as 'XES'). FWHM = 0.25 eV.

man's RIXS for the cation (i.e., the XES-like spectrum of benzene with the core hole in $1b_{2u}/1b_{3u}$ orbitals). The RIXS spectra were computed by including the lowest 22 excited states below 10.35 eV. As expected, the two dominant features are similar—the elastic peak and the transition to the 2²A_g state are slightly blue shifted by about 0.3 eV. Geometric relaxation has relatively small effect on the position and intensities of these two major features. The computed RIXS cross sections for benzene cation are presented in Table S11 of the ESI.†

Conclusions

In this contribution, we presented an extension of damped response theory for calculating RIXS cross sections using CVS within the response equations. Our primary aim was to address problematic divergence and erratic behavior of the response solutions in the X-ray frequency range, which is needed for RIXS calculations. This problematic behavior arises due to the coupling between virtual states and the decay channels into the valence ionization continuum. We have extended a newly developed fc-CVS-EOM-EE-CCSD variant of the EOM-EE-CCSD theory to RIXS calculations by combining the damped response theory and the CVS truncation of the response space. This method exploits the ability of CVS to decouple the valenceexcited and core-excited configurations of the excitation manifold, thereby blocking the autoionization decay channels for the virtual (i.e., response) states. The numeric results demonstrate that this approach solves the problematic divergence of the response solutions, while not affecting the quality of the computed RIXS spectrum. Our fc-CVS-EOM-EE-CCSD implementation can treat both closed- and open-shell species, which we illustrated by calculating the RIXS emission spectra of (closed-shell) benzene and its (open-shell) cation. We analyzed the XAS, XES, and RIXS spectra of these species in terms of molecular orbitals and demonstrated the qualitative differences and similarities between their spectra. For neutral benzene, the computed spectra can be compared to the experimental ones, illustrating the robust performance of the fc-CVS-EOM-EE-CCSD approach. By providing an accurate and robust tool for computing electronic RIXS transition moments, fc-CVS-EOM-EE-CCSD serves as an excellent platform for further theoretical developments, such as the inclusion of nuclear motions, which is needed for a complete description of these phenomena.

Conflicts of interest

The authors declare the following competing financial interest(s): A. I. K. is a member of the Board of Directors and a part-owner of Q-Chem, Inc.

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References

- 1 F. Gel'mukhanov and H. Ågren, Phys. Rep., 1999, 312,
- 2 S. Butorin, J. Electron Spectrosc. Relat. Phenom., 2000, **110-111**, 213-233.
- 3 F. de Groot, Chem. Rev., 2001, 101, 1779-1808.

- 4 A. Kotani and S. Shin, Rev. Mod. Phys., 2001, 73, 203-246.
- 5 L. Ament, M. van Veenendaal, T. P. Devereaux, J. Hill and J. van den Brink, *Rev. Mod. Phys.*, 2011, **83**, 705–767.
- 6 T. Schmitt, F. de Groot and J. Rubensson, *J. Synchrotron Radiat.*, 2014, 21, 1065–1076.
- 7 S. Eckert, J. Norell, P. Miedema, M. Beye, M. Fondell, W. Quevedo, B. Kennedy, M. Hantschmann, A. Pietzsch, B. Van Kuiken, M. Ross, M. Minitti, S. Moeller, W. Schlotter, M. Khalil, M. Odelius and A. Föhlisch, *Phys. Rev. Lett.*, 2017, 56, 6088–6092.
- 8 Synchrotron Radiation: Basics, Methods and Applications, ed. S. Mobilio, F. Boscherini and C. Meneghini, Springer, 2014.
- 9 *X-Ray Absorption and X-ray Emission Spectroscopy; Theory and Applications*, ed. J. van Bokhoven and C. Lamberti, Wiley & Sons, 2016.
- 10 X-Ray Free Electron Lasers: Applications in Materials, Chemistry and Biology, ed. U. Bergmann, V. Yachandra and J. Yano, Royal Society of Chemistry, 2017.
- 11 M. Nisoli, P. Decleva, F. Calegari, A. Palacios and F. Martín, Chem. Rev., 2017, 117, 10760–10825.
- 12 P. Norman and A. Dreuw, Chem. Rev., 2018, 118, 7208-7248.
- 13 T. Fransson, Y. Harada, N. Kosugi, N. A. Besley, B. Winter, J. J. Rehr, L. G. M. Pettersson and A. Nilsson, *Chem. Rev.*, 2016, 116, 7551–7569.
- 14 O. Kostko, B. Bandyopadhyay and M. Ahmed, *Annu. Rev. Phys. Chem.*, 2016, **67**, 19–40.
- D. Zuev, E. Vecharynski, C. Yang, N. Orms and A. I. Krylov, J. Comput. Chem., 2015, 36, 273–284.
- 16 S. Coriani, O. Christiansen, T. Fransson and P. Norman, *Phys. Rev. A: At., Mol., Opt. Phys.*, 2012, **85**, 022507.
- 17 S. Coriani, T. Fransson, O. Christiansen and P. Norman, *J. Chem. Theory Comput.*, 2012, **8**, 1616–1628.
- 18 A. Sadybekov and A. I. Krylov, J. Chem. Phys., 2017, 147, 014107.
- 19 M. L. Vidal, X. Feng, E. Epifanovski, A. I. Krylov and S. Coriani, *J. Chem. Theory Comput.*, 2019, 15, 3117–3133.
- 20 M. Lundberg, T. Kroll, S. DeBeer, U. Bergmann, S. A. Wilson, P. Glatzel, D. Nordlund, B. Hedman, K. O. Hodgson and E. I. Solomon, J. Am. Chem. Soc., 2013, 135, 17121–17134.
- 21 D. Maganas, P. Kristiansen, L.-C. Duda, A. Knop-Gericke, S. DeBeer, R. Schlögl and F. Neese, *J. Phys. Chem. C*, 2014, **118**, 20163–20175.
- 22 E. Ertan, V. Savchenko, N. Ignatova, V. Vaz da Cruz, R. C. Couto, S. Eckert, M. Fondell, M. Dantz, B. Kennedy, T. Schmitt, A. Pietzsch, A. Föhlisch, F. Gel'mukhanov, M. Odelius and V. Kimberg, *Phys. Chem. Chem. Phys.*, 2018, 20, 14384–14397.
- 23 D. Rehn, A. Dreuw and P. Norman, *J. Chem. Theory Comput.*, 2017, **13**, 5552–5559.
- 24 R. Faber and S. Coriani, *J. Chem. Theory Comput.*, 2019, **15**, 520–528.
- 25 A. Buckingham and P. Fischer, *Phys. Rev. A: At., Mol., Opt. Phys.*, 2000, **61**, 035801–035804.
- 26 K. Kristensen, J. Kauczor, A. J. Thorvaldsen, P. Jørgensen, T. Kjaergaard and A. Rizzo, *J. Chem. Phys.*, 2011, **134**, 214104–214120.

- 27 P. Norman, Phys. Chem. Chem. Phys., 2011, 12, 20519-20535.
- 28 J. Kauczor, P. Norman, O. Christiansen and S. Coriani, J. Chem. Phys., 2013, 139, 211102.
- 29 L. Jensen, J. Autschbach and G. C. Schatz, J. Chem. Phys., 2005, 122, 224115.
- 30 J. Kauczor and P. Norman, *J. Chem. Theory Comput.*, 2014, **10**, 2449–2455.
- 31 D. Maganas, S. DeBeer and F. Neese, *Inorg. Chem.*, 2017, **56**, 11819–11836.
- 32 N. A. Besley, A. T. B. Gilbert and P. M. W. Gill, *J. Chem. Phys.*, 2009, **130**, 124308.
- 33 N. A. Besley, Chem. Phys. Lett., 2012, 542, 42-46.
- 34 K. Emrich, Nucl. Phys., 1981, A351, 379-396.
- 35 J. F. Stanton and R. J. Bartlett, J. Chem. Phys., 1993, 98, 7029-7039.
- 36 A. I. Krylov, Annu. Rev. Phys. Chem., 2008, 59, 433-462.
- 37 R. J. Bartlett, Mol. Phys., 2010, 108, 2905-2920.
- 38 K. Sneskov and O. Christiansen, Wiley Interdiscip. Rev.: Comput. Mol. Sci., 2012, 2, 566-584.
- 39 R. J. Bartlett, Wiley Interdiscip. Rev.: Comput. Mol. Sci., 2012, 2, 126–138.
- 40 J. A. Pople, Energy, Structure and Reactivity: Proceedings of the 1972 Boulder Summer Research Conference on Theoretical Chemistry, Wiley, New York, 1973, pp. 51–61.
- 41 L. V. Slipchenko, J. Phys. Chem. A, 2010, 114, 8824-8830.
- 42 O. Christiansen, J. Gauss and B. Schimmelpfennig, *Phys. Chem. Chem. Phys.*, 2000, 2, 965–971.
- 43 K. Klein and J. Gauss, J. Chem. Phys., 2008, 129, 194106.
- 44 E. Epifanovsky, K. Klein, S. Stopkowicz, J. Gauss and A. I. Krylov, *J. Chem. Phys.*, 2015, **143**, 064102.
- 45 P. Pokhilko, E. Epifanovsky and A. I. Krylov, *J. Chem. Phys.*, 2019, **151**, 034106.
- 46 A. Tajti and P. Szalay, J. Chem. Phys., 2009, 131, 124104.
- 47 T. Ichino, J. Gauss and J. F. Stanton, *J. Chem. Phys.*, 2009, **130**, 174105.
- 48 S. Faraji, S. Matsika and A. I. Krylov, J. Chem. Phys., 2018, 148, 044103.
- 49 T. Helgaker, S. Coriani, P. Jørgensen, K. Kristensen, J. Olsen and K. Ruud, *Chem. Rev.*, 2012, **112**, 543–631.
- 50 K. D. Nanda and A. I. Krylov, J. Chem. Phys., 2015, 142, 064118.
- 51 K. D. Nanda and A. I. Krylov, J. Chem. Phys., 2017, 146, 224103.
- 52 K. Nanda and A. I. Krylov, *J. Phys. Chem. Lett.*, 2017, 8, 3256–3265.
- 53 K. Nanda and A. I. Krylov, J. Chem. Phys., 2018, 149, 164109.
- 54 M. de Wergifosse, C. G. Elles and A. I. Krylov, *J. Chem. Phys.*, 2017, **146**, 174102.
- 55 M. de Wergifosse, A. L. Houk, A. I. Krylov and C. G. Elles, *J. Chem. Phys.*, 2017, **146**, 144305.
- 56 P. B. Rozyczko, S. A. Perera, M. Nooijen and R. J. Bartlett, J. Chem. Phys., 1997, 107, 6736–6747.
- 57 C. Hättig, O. Christiansen, S. Coriani and P. Jørgensen, *J. Chem. Phys.*, 1998, **109**, 9237.
- 58 K. D. Nanda and A. I. Krylov, *J. Chem. Phys.*, 2016, 145, 204116.
- 59 K. Nanda, A. I. Krylov and J. Gauss, *J. Chem. Phys.*, 2018, **149**, 141101.

- 60 H. Koch and P. Jørgensen, J. Chem. Phys., 1990, 93, 3333-3344.
- 61 T. B. Pedersen and H. Koch, J. Chem. Phys., 1997, 106, 8059-8072.
- 62 O. Christiansen, P. Jørgensen and C. Hättig, Int. J. Quantum Chem., 1998, 68, 1-52.
- 63 B. Peng, P. Lestrange, J. Goings, M. Caricato and X. Li, J. Chem. Theory Comput., 2015, 11, 4146-4153.
- 64 S. Coriani and H. Koch, J. Chem. Phys., 2015, 143, 181103.
- 65 S. Coriani and H. Koch, J. Chem. Phys., 2016, 145, 149901.
- 66 R. Faber and S. Coriani, Phys. Chem. Chem. Phys., 2019, DOI: 10.1039/c9cp03696b.
- 67 H. Sekino and R. J. Bartlett, Int. J. Quantum Chem., 1984, 26, 255-265.
- 68 H. Koch, H. Jensen, P. Jørgensen and T. Helgaker, J. Chem. Phys., 1990, 93, 3345-3350.
- 69 L. Cederbaum, W. Domcke and J. Schirmer, Phys. Rev. A: At., Mol., Opt. Phys., 1980, 22, 206.
- 70 R. Myhre, S. Coriani and H. Koch, J. Chem. Theory Comput., 2016, 12, 2633-2643.
- 71 B. Tenorio, T. Moitra, M. Nascimento, A. Rocha and S. Coriani, J. Chem. Phys., 2019, 150, 224104.
- 72 A. I. Krylov and P. M. W. Gill, Wiley Interdiscip. Rev.: Comput. Mol. Sci., 2013, 3, 317-326.
- 73 Y. Shao, Z. Gan, E. Epifanovsky, A. T. B. Gilbert, M. Wormit, J. Kussmann, A. W. Lange, A. Behn, J. Deng and X. Feng, et al., Mol. Phys., 2015, 113, 184-215.
- 74 F. Gel'mukhanov and H. Ågren, Phys. Rev. A: At., Mol., Opt. Phys., 1994, 49, 4378-4389.
- 75 C. Hättig, O. Christiansen and P. Jørgensen, J. Chem. Phys., 1998, 108, 8331-8354.
- 76 M. Paterson, O. Christiansen, F. Pawlowski, P. Jørgensen, C. Hättig, T. Helgaker and P. Salek, J. Chem. Phys., 2006, **124**, 054332.
- 77 In the exact expression of the RIXS scattering moment, the damping factor in each sum-over-states term corresponds to the inverse lifetime of the intermediate state. In damped response theory, however, a phenomenological non-statespecific value for the damping factor is usually used.
- 78 P. Pulay, Chem. Phys. Lett., 1980, 73, 393.

- 79 G. E. Scuseria, T. J. Lee and H. F. Schaefer, Chem. Phys. Lett., 1986, 130, 236-239.
- 80 H. Feshbach, Ann. Phys., 1962, 19, 287-313.
- 81 V. Averbukh and L. Cederbaum, J. Chem. Phys., 2005, 123, 204107.
- 82 H. Tachikawa, J. Phys. Chem. A, 2018, 122, 4121-4129.
- 83 R. Lindner, K. Müller-Dethlefs, E. Wedum, K. Haber and E. Grant, Science, 1996, 271, 1698-1702.
- 84 R. Mulliken, J. Chem. Phys., 1955, 23, 1997-2011.
- 85 F. Plasser, M. Wormit and A. Dreuw, J. Chem. Phys., 2014, 141, 024106.
- 86 C. M. Oana and A. I. Krylov, J. Chem. Phys., 2007, 127, 234106.
- 87 M. L. Vidal, A. I. Krylov and S. Coriani, Phys. Chem. Chem. Phys., 2019, DOI: 10.26434/chemrxiv.9616991.v2.
- 88 E. Epifanovsky, M. Wormit, T. Kuś, A. Landau, D. Zuev, K. Khistyaev, P. U. Manohar, I. Kaliman, A. Dreuw and A. I. Krylov, J. Comput. Chem., 2013, 34, 2293-2309.
- 89 E. Kamarchik, O. Kostko, J. M. Bowman, M. Ahmed and A. I. Krylov, J. Chem. Phys., 2010, 132, 194311.
- 90 V. Vaz da Cruz, F. Gel'mukhanov, S. Eckert, M. Iannuzzi, E. Ertan, A. Pietzsch, R. C. Couto, J. Niskanen, M. Fondell, M. Dantz, T. Schmitt, X. Lu, D. McNally, R. M. Jay, V. Kimberg, A. Föhlisch and M. Odelius, Nat. Commun., 2019, 10, 1013.
- 91 D. Menzel, G. Rocker, H.-P. Steinrück, D. C. P. A. Heimann, W. Huber, P. Zebisch and D. R. Lloyd, J. Chem. Phys., 1992, 96, 1724.
- 92 E. E. Rennie, B. Kempgens, H. M. Köppe, U. Hergenhahn, J. Feldhaus, B. S. Itchkawitz, A. L. D. Kilcoyne, A. Kivimäki, K. Maier, M. N. Piancastelli, M. Polcik, A. Rüdel and A. M. Bradshaw, J. Chem. Phys., 2000, 113, 7362-7375.
- 93 P. Skytt, J. Guo, N. Wassdahl, J. Nordgren, Y. Luo and H. Ågren, Phys. Rev. A: At., Mol., Opt. Phys., 1995, 52, 3572-3576.
- 94 F. Hennies, S. Polyutov, I. Minkov, A. Pietzsch, M. Nagasono, H. Ågren, L. Triguero, M.-N. Piancastelli, W. Wurth, F. Gel'mukhanov and A. Föhlisch, Phys. Rev. A: At., Mol., Opt. Phys., 2007, 76, 032505.
- 95 J.-H. Guo, M. Magnuson, C. Såthe, J. Nordgren, L. Yang, Y. Luo, H. Ågren, K. Z. Xing, N. Johansson, W. R. Salaneck, R. Daik and W. J. Feast, J. Chem. Phys., 1998, 108, 5990-5996.