

Relay Catalysis To Synthesize β -Substituted Enones: Organocatalytic Substitution of Vinylogous Esters and Amides with Organoboronates

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Cite This: *Org. Lett.* 2020, 22, 1355–1359



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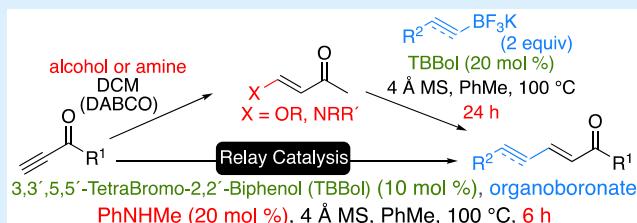
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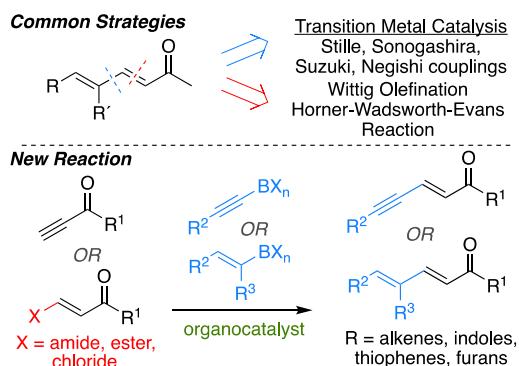
Supporting Information

ABSTRACT: Organocatalysis was shown to facilitate conjugate additions to vinylogous esters and amides for the first time. Subsequent elimination of a β -alcohol or amine provided π -conjugated β -substituted enones. Remarkably, nucleophile addition to the electron-rich vinylogous substrates is more rapid than classical enones, forming monosubstituted products. A doubly organocatalytic (organic diol and methyl aniline) conjugate addition synthesis directly from alkynyl ketones. Both of these catalytic transformations are orthogonal to transition metal catalysis, allowing for good yields, easily accessible or commercially available reagents, high selectivity, reagent recovery and recyclability, facile scalability, and exceptional functional group tolerance.



Polyenes occur in natural products¹ and can be used in Nazarov reactions to build larger, more complex structures.² Transition metal cross-couplings and carbonyl olefinations to form polyenes are powerful and well developed (Scheme 1) but nonetheless suffer from limited functional group

Scheme 1. Strategies for Polyene Structures



tolerance, harsh conditions, difficulty in catalyst recovery, cost, toxicity, and/or environmental issues on scale up in industrial applications.³ Conjugate addition reactions effectively build structural complexity in challenging molecules.⁴ This report details organodiol catalysis with organoboronate nucleophiles for 1,4-additions to vinylogous amides and esters, as well as relay catalysis with this newly discovered reactivity to form polyenes directly from abundant alkynyl ketones. Both catalytic species are readily recycled, and this coupling strategy is highly functional group tolerant and transition metal free.

The conjugate addition of organoboronates to enones was first reported by Suzuki,⁵ followed by enantioselective versions,⁶ and a significant advance to broaden the range of nucleophiles used trifluoroborate salts.⁷ Kinetic studies suggested that the reaction rate is increased by electron-rich β -substituents that stabilize developing cationic charge at the β -carbon.⁸ In light of that observation, we hypothesized that resonance electron donor substituents, ethers and amines, would make the enone even more reactive.^{9,10} This paradigm would reverse that of nearly all conjugate addition reactions, for which vinylogous esters and amides are too electron-rich to be competent substrates.¹¹ Precedence for such conjugate additions is limited to a few reports of palladium catalysis to add aryl boronic acids to vinylogous amides.¹² In a single report of vinylogous substitution, monoaddition did not occur since the enone from the first conjugate addition was more electron-deficient and therefore more reactive than the vinylogous amide. Moreover, limited functional group tolerance was demonstrated.

We tested the hypothesized conjugate addition to vinylogous esters with a 3,3'-modified BINOL catalyst (Table 1, entry 1) that was previously reported to catalyze 1,4-additions.^{7,8} The resulting 2,4-dienyl ketone was confirmed to be the linearly conjugated polyunsaturated ketone 3, the result of the anticipated 1,4-addition of the BF_3K salt followed by phenoxide

Received: December 28, 2019

Published: January 30, 2020

Table 1. Catalyst Screen

entry	catalyst	time	yield
1	(\pm)-C ₇ F ₇ BINOL	24 hours	86 ^b
2	(\pm)-IODO BINOL	24 hours	85 ^a
3	(\pm)-BINOL	24 hours	81 ^b
4	2,2'-biphenol	48 hours	82 ^b
5	TBBol	18 hours	82 ^b
6	(<i>L</i>)-tartaric acid	72 hours	77 ^a
7	(<i>L</i>)-tartramide	48 hours	6 ^a
8	1,4-butanediol	48 hours	42 ^a
9	none	48 hours	61 ^a

Chemical structures of substrates and products:

- Substrate 1: Ph-O-CH=CH-C(=O)-CH₂
- Product 3: Ph-CH=CH-CH=CH-C(=O)-CH₂
- Substrate 2: 2,2'-biphenol
- Product 4: 2,2'-biphenol
- Substrate 5: (±)-BINOL
- Product 6: (±)-BINOL
- Substrate 7: (L)-tartramide
- Product 7: (L)-tartramide
- Substrate 8: 1,4-butanediol
- Product 8: 1,4-butanediol
- Substrate 9: none
- Product 9: none

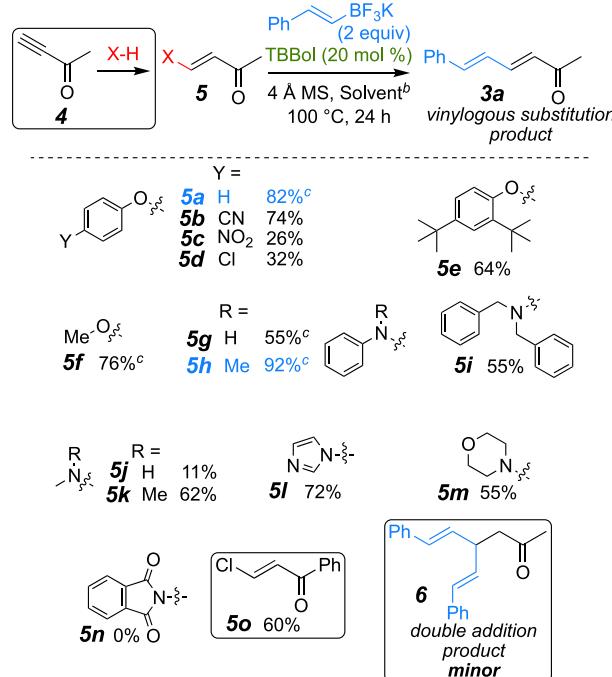
Chemical structures of products:

- Product 1: 2,2'-biphenol
- Product 2: 2,2'-biphenol
- Product 3: 2,2'-biphenol
- Product 4: 2,2'-biphenol
- Product 5: 2,2'-biphenol
- Product 6: 2,2'-biphenol
- Product 7: 2,2'-biphenol
- Product 8: 2,2'-biphenol
- Product 9: 2,2'-biphenol

Chemical structures of products:

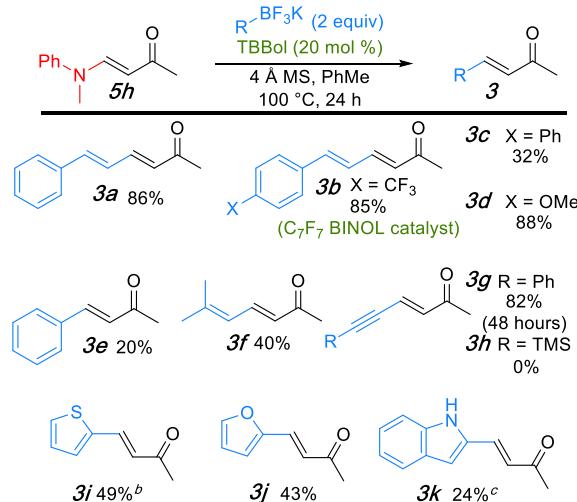
- Product 1: 2,2'-biphenol
- Product 2: 2,2'-biphenol
- Product 3: 2,2'-biphenol
- Product 4: 2,2'-biphenol
- Product 5: 2,2'-biphenol
- Product 6: 2,2'-biphenol
- Product 7: 2,2'-biphenol
- Product 8: 2,2'-biphenol
- Product 9: 2,2'-biphenol

^aBased on quantified peak area in the ¹H NMR of the crude reaction material relative to 4-methylnitrobenzoate as an internal standard.
^bIsolated yields, average of three trials.

Scheme 2. Evaluation of Vinylogous Esters and Amides^a

^aYields of 3a from 5 based on quantified peak area in the ¹H NMR of crude reaction material relative to 4-methylnitrobenzoate as an internal standard. ^bToluene was the solvent for the amides and vinyl chloride. For the esters, the solvent was 1,4-dioxane.¹⁷ ^cAverage isolated yield of 3a from 5 from three experiments.

Diverse organoborionate nucleophiles were useful, including the often-sensitive alkyne salts (Scheme 3). Boronic acids were

Scheme 3. Nucleophiles for the Monocatalytic Reaction^a

^aIsolated yields. ^bBoronic acid used. ^cBoc-indole-2-BF₃K used.

also active, though they exhibited slower reactivity that could be compensated for by increasing reaction time.^{17,18} Substitution of the styrenyl aryl ring did not considerably affect yields except for a *para*-phenyl (3c). Aromatic nucleophiles (3e, 3i, 3j, 3k) gave decreased yields, likely due to the necessity of dearomatization during the key C–C bond formation. Alkyne nucleophiles (3g) were reactive, but required increased reaction times. Unsub-

stituted or TMS alkynyl boronates (**3h**) performed poorly, perhaps due to conflicting reactivity.

Synthesizing vinylogous esters and amides that varied the ketone substituent experienced low yields (see *Supporting Information (SI)*), even with excess amine or catalysts like DABCO. Additionally, alkyne **4** was not reactive to organoboranes under any conjugate addition conditions, which was disappointing since that reaction would have directly generated the doubly unsaturated ketones **3**.

To overcome these limitations, we envisioned a doubly catalytic reaction to both form the vinylogous ester or amide and perform the vinylogous substitution in the same reaction by generating small amounts of the vinylogous intermediate *in situ* (**Scheme 5**). This would avoid purifying potentially unstable vinylogous esters and amides. Mechanistically, methyl aniline would add to alkynyl ketone **4** to form vinylogous amide **5c**, which would then be reactive for the conjugate addition catalyzed by the organodiol. After complexation of the organoboronates **2** to TBBol,⁷ Lewis base coordination provides **II**. Bond formation proceeds to give the boron enolate **IV**,^{6,14} and β -elimination of the aniline would form the polyene **3a**. This C–C bond formation has interesting elements of both conjugate addition and Petasis mechanisms, the latter relationship being apparent from resonance structure **IIb**. A possible reason for only dienyl ketone **3a** being observed could be intramolecular Lewis acid complexation as shown in **IV** followed by concomitant β -elimination/deborylation to generate borate **V**. Catalyst exchange could then occur to regenerate the aniline and the organodiol.

The two best leaving groups (phenol and methyl aniline) were tested, and polyene **3a** was formed in 27% yield with 1 equiv of methyl aniline and 20 mol % TBBol. No **3a** was observed with phenol. Reaction improvement increased the yield of **3a** to 96% in less time than those with pure vinylogous esters and amides (6 h instead of 24 h, **Table 2**, entry 1) using a substoichiometric

(compare **Table 2**, entries 1 and 3). Phosphine decreased the formation of both vinylogous amide **5g** and ketone **3a** (entry 8), and DABCO suppressed organoborate addition entirely. These controls are consistent with the mechanism in **Scheme 4** and exclude Bayless–Hillman-like C–C bond formation.

Scheme 4. Catalytic Cycle

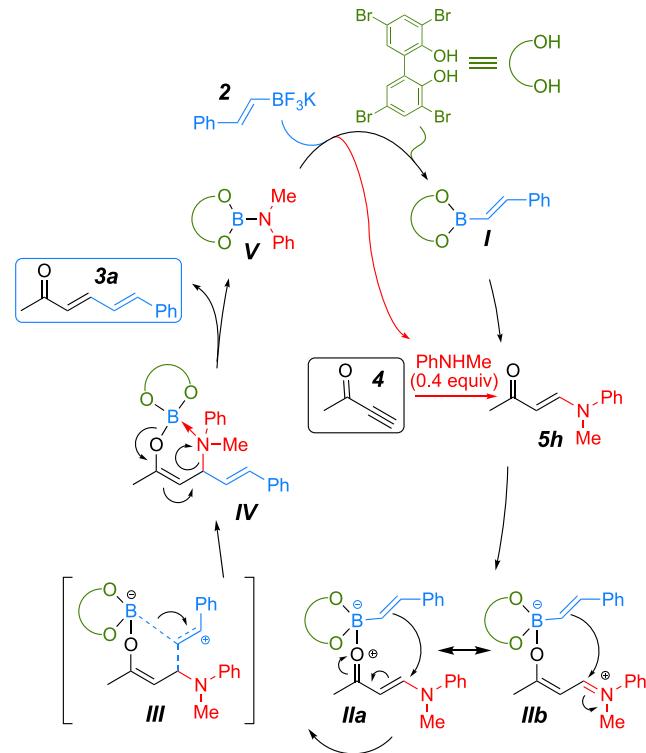
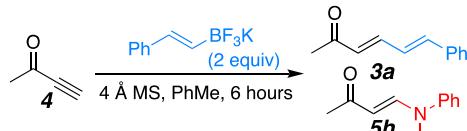


Table 2. Relay Catalysis Control Experiments



entry	TBBol	PhNHMe	Additive (40 mol %)	yield 3a (%) ^a	yield 5g (%) ^a
1	20 mol %	40 mol %	none	96	0
2	20 mol %	none	none	0	0
3	none	40 mol %	none	31	30
4	20 mol %	none	DABCO	0	0
5	none	40 mol %	DABCO	0	31
6	none	none	DABCO	0	0
7	20 mol %	none	PBu ₃	0	0
8	none	40 mol %	PBu ₃	19	13
9	none	none	PBu ₃	0	0

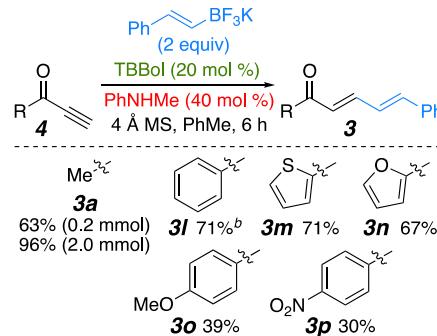
^aBased on quantified peak area in the ¹H NMR of the crude reaction material relative to 4-methylnitrobenzoate as an internal standard.

quantity of methyl aniline.¹⁷ No product was observed without aniline, confirming the necessity of an intermediary vinylogous amide (entries 2, 4, 6, 7, and 9). Without TBBol, slow product formation occurred if aniline were present (entry 3), which is consistent with the observed background reaction (**Table 1**, entry 9). TBBol increased product formation substantially

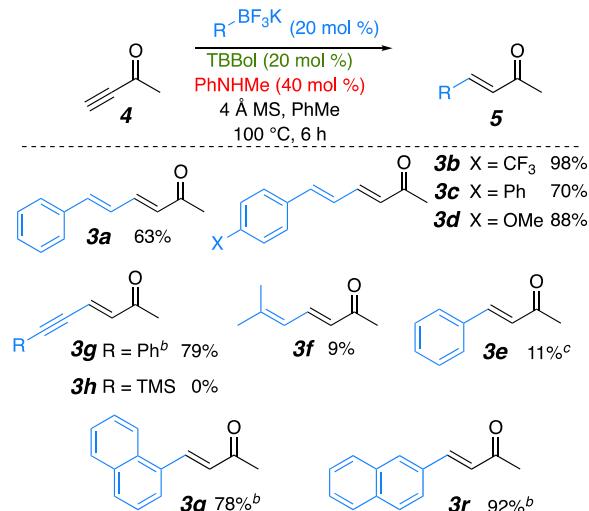
The doubly catalytic reaction also had greater tolerance to ketone modifications. As hypothesized, the vinylogous esters and amides that had been problematic were made productive (**Scheme 5**). Ketone aryl substituents (**3o** and **3p**) gave lower yields than methyl substitution (**3a**), but heterocycle substituents reacted similarly to a methyl substituent (see **3m** and **3n**).

Relay catalysis allowed a significant variety of nucleophiles to add to alkynyl ketones (**Scheme 6**). In this process, boronic acids performed almost as well as trifluoroborate salts (see *SI*). Aryl trifluoroborates were not as reactive, presumably due to the need to break aromaticity in C–C bond formation (i.e., **II** to **III**,

Scheme 5. Doubly Catalytic Trials^a



^aIsolated yields. ^b24 h.

Scheme 6. Nucleophile Variation with Relay Catalysis^a

^aIsolated yields. ^bReaction time 48 h. ^c1 equiv of LiBr for 6 h.

Scheme 4), but almost all vinyl and some alkynyl trifluoroborates gave good yields. This was supported by the observation that naphthyl nucleophiles, which have less aromatic stabilization, produced higher yields.⁷ Aryl substitution for styrenyl nucleophiles did not significantly affect yields (3b–3d). Yields could be increased by individual optimization; for example, increasing the reaction time to 48 h increased the yield for 3g. Lithium bromide, previously used in conjugate additions with aromatic nucleophiles, allowed formation of the β -phenylenone 3e. The only example that performed worse in the relay catalysis was diene 3f. Generally, the strength of this method lies in accommodating heterocycles, alkynes, and other sensitive functional groups that are problematic for other strategies. Functional group tolerance was assessed as described by Glorius,¹⁹ showing high tolerance in nearly all cases.¹⁷ Though a few functional groups were less tolerant of elevated temperatures, this reaction experiences only minor reductions in yield when the temperature is decreased as low as 50 °C, so thermally sensitive groups could be preserved with an increase in reaction time.¹⁷

This transformation excelled on a larger scale, with a nearly quantitative yield on a 2 mmol scale (Table 3, entry 1). We note that the conditions reported in Scheme 6 were broadly applicable for a successful reaction. Table 3 demonstrates that, for an individual substrate reaction, optimization allows for low

Table 3. Scale up and Optimization of a Specific Substrate



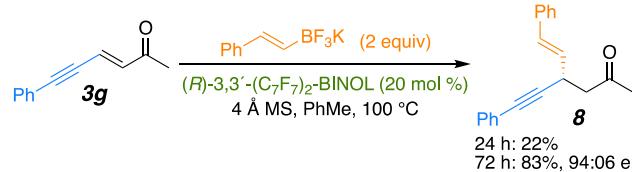
entry	scale	TBBol loading	PhNHMe loading	concen	time	yield ^a
1	2 mmol	20 mol %	40 mol %	0.1 M	6 h	95%
2	2 mmol	10 mol %	10 mol %	0.5 M	14 h	60%
3	2 mmol	5 mol %	5 mol %	0.5 M	14 h	49%
4	5 mmol	10 mol %	20 mol %	0.5 M	6 h	55%
5	5 mmol	10 mol %	20 mol %	0.5 M	42 h	88%

^aIsolated yields.

catalyst loadings while maintaining high yields. With adjustments to concentration and reaction time, lower catalyst loadings could be used with minimal loss in yield (entry 5). TBBol and most of the methyl aniline are easily recoverable upon purification, allowing for recycling.

Products like alkynyl enones 3g can be difficult to reliably access and control, which is likely why there have only been three reports of their use for enantioselective conjugate additions.²⁰ We found that these compounds could be enantioselectively transformed into β -branched ketones by subsequently using a chiral diol catalyst (Scheme 7). Although

Scheme 7. Enantioenriched Double Addition Products



3g was formed in only 6 h from the alkynyl ketone (Scheme 6) or 48 h from the vinylogous amide (Scheme 3), the addition of the second nucleophile to give 8 required 3 days for high conversion with 96:4 er. This comparison clearly demonstrated the greater reaction completion time for enones relative to vinylogous amides. To our knowledge, there are no enantioselective reports for the formation of β -alkynyl/ β -alkenyl ketones.

In conclusion, a previously inaccessible conjugate addition reaction has been formulated, and we present the first report of organocatalyzed vinylogous substitution of vinylogous amides and esters to provide conjugated β -substituted enones. We have exploited this reactivity for relay catalysis to use propargyl ketones directly. These transformations are catalyzed by an easily accessed brominated biphenol organocatalyst, TBBol. The organodiol and trifluoroborate nucleophiles perform the equivalent transformation as transition metal cross-couplings, but with increased functional group tolerance as proven by a Glorius-inspired study. This method allows inclusion of a variety of unsaturated substituents, including heteroaromatics and alkynes. We are now applying this methodology to the synthesis of more complex structures and of cross-conjugated organic polymers.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.9b04584>.

Experimental details and spectra (PDF)

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

The authors are grateful to the Welch Foundation (Grant E-1744) and the NSF (Grant CHE-1800499) for generous financial support.

REFERENCES

- (1) Nguyen, T. N.; May, J. A. Enantioselective Organocatalytic Conjugate Addition of Organoboron Nucleophiles. *Tetrahedron Lett.* **2017**, *58*, 1535–1544.
- (2) Thirsk, C.; Whiting, A. Polyene Natural Products. *J. Chem. Soc. Perkin Trans. 1* **2002**, *8*, 999–1023.
- (3) (a) He, W.; Herrick, I. R.; Atesin, T. A.; Caruana, P. A.; Kellenberger, C. A.; Frontier, A. J. Polarizing the Nazarov Cyclization: The Impact of Dienone Substitution Pattern on Reactivity and Selectivity. *J. Am. Chem. Soc.* **2008**, *130* (3), 1003–1011. (b) Vaidya, T.; Eisenberg, R.; Frontier, A. J. Catalytic Nazarov Cyclization: The State of the Art. *ChemCatChem*; John Wiley & Sons, Ltd.: October 17, 2011; pp 1531–1548.
- (4) Sun, C. L.; Shi, Z. J. Transition-Metal-Free Coupling Reactions. *Chem. Rev.* **2014**, *114*, 9219–9280.
- (5) (a) Sato, M.; Miyaura, N.; Suzuki, A. Rhodium(I)-Catalyzed Asymmetric Hydroboration of Alkenes with 1,3,2-Benzodioxaborole. *Tetrahedron Lett.* **1990**, *31*, 231–234. (b) Hara, S.; Ishimura, S.; Suzuki, A. Cyanuric Fluoride-Induced 1,4-Addition Reaction of Alkenylboronic Acids to α,β -Unsaturated Ketones. Stereoselective Synthesis of γ,δ -Unsaturated Ketones Having Functionalities. *Synlett* **1996**, *1996*, 993–994. (c) Hara, S.; Shudoh, H.; Ishimura, S.; Suzuki, A. The Regioselective 1,4-Addition Reaction of Alkenylboronic Acids to $\alpha,\beta,\alpha',\beta'$ -Unsaturated Ketones. *Bull. Chem. Soc. Jpn.* **1998**, *71*, 2403–2408.
- (6) (a) Turner, H. M.; Patel, J.; Niljianskul, N.; Chong, J. M. Binaphthol-Catalyzed Asymmetric Conjugate Arylboration of Enones. *Org. Lett.* **2011**, *13*, 5796–5799. (b) Okino, T.; Hoashi, Y.; Takemoto, Y. Enantioselective Michael Reaction of Malonates to Nitroolefins Catalyzed by Bifunctional Organocatalysts. *J. Am. Chem. Soc.* **2003**, *125*, 12672–12673. (c) Sugiura, M.; Tokudomi, M.; Nakajima, M. Enantioselective Conjugate Addition of Boronic Acids to Enones Catalyzed by *O*-Monoacyltartaric Acids. *Chem. Commun.* **2010**, *46*, 7799–7800. (d) Sugiura, M.; Kinoshita, R.; Nakajima, M. *O*-Monoacyltartaric Acid Catalyzed Enantioselective Conjugate Addition of a Boronic Acid to Dienones: Application to the Synthesis of Optically Active Cyclopentenones. *Org. Lett.* **2014**, *16*, 5172–5175. (e) Lundy, B. J.; Jansone-Popova, S.; May, J. A. The Enantioselective Conjugate Addition of Alkenylboronic Acids to Indole-Appended Enones. *Org. Lett.* **2011**, *13*, 4958–4961.
- (7) Shih, J.-S.; Nguyen, T. S.; May, J. A. Organocatalyzed Asymmetric Conjugate Addition of Heteroaryl and Aryl Trifluoroborates: a Synthetic Strategy for Discothiophore D. *Angew. Chem., Int. Ed.* **2015**, *54*, 9931–9935.
- (8) Nguyen, T. S.; Yang, M. S.; May, J. A. Experimental Mechanistic Insight into the BINOL-Catalyzed Enantioselective Conjugate Addition of Boronates to Enones. *Tetrahedron Lett.* **2015**, *56*, 3337–3341.
- (9) For additions of organoboronates to *in situ* generated oxocarbeniums, see: (a) Vo, V. T.; Bode, J. W. Expanded Substrate Scope and Improved Reactivity of Ether-Forming Cross-Coupling Reactions of Organotrifluoroborates and Acetals. *J. Am. Chem. Soc.* **2011**, *133*, 14082–14089. (b) Vieira, A. S.; Fiorante, P. F.; Hough, T. L. S.; Ferreira, F. P.; Lüdtke, D. S.; Stefani, H. A. Nucleophilic Addition of Potassium Alkynyltrifluoroborates to D-Glucal Mediated by $\text{BF}_3\text{-OEt}_2$: Highly Stereoselective Synthesis of α -C-Glycosides. *Org. Lett.* **2008**, *10*, 5215–5218. (c) Clausen, D. J.; Floreancig, P. E. Aromatic Cations from Oxidative Carbon-Hydrogen Bond Cleavage in Bimolecular Carbon-Carbon Bond Forming Reactions. *J. Org. Chem.* **2012**, *77*, 6574–6582.
- (10) For additions to iminiums (Petasis reaction), see these reviews: (a) Wu, P.; Givskov, M.; Nielsen, T. E. Reactivity and Synthetic Applications of Multicomponent Petais Reactions. *Chem. Rev.* **2019**, *119*, 11245–11290. (b) Candeias, N. R.; Montalbano, F.; Cal, P. M. S. D.; Gois, P. M. P. Boronic Acids and Esters in the Petais-Borono Mannich Multicomponent Reaction. *Chem. Rev.* **2010**, *110*, 6169–6193.
- (11) Perlmutter, P. *Conjugate Addition Reactions in Organic Synthesis*; *Tetrahedron Organic Chemistry Series* 9, 1st ed.; Elsevier: 1992. Kwon, K. H.; Lee, D. W.; Yi, C. S. Scope and Mechanistic Study of the Coupling Reaction of α,β -Unsaturated Carbonyl Compounds with Alkenes: Uncovering Electronic Effects on Alkene Insertion vs Oxidative Coupling Pathways. *Organometallics* **2012**, *31*, 495–504.
- (12) (a) Liu, Y.; Li, D.; Park, C.-M. Stereoselective Synthesis of Highly Substituted Enamides by an Oxidative Heck Reaction. *Angew. Chem., Int. Ed.* **2011**, *50*, 7333–7336. (b) Zhong, S.; Lu, Y.; Zhang, Y.; Liu, Y.; Wan, J. P. Synthesis of β,β -Diaryl Propiophenones via Palladium-Catalyzed Domino Arylboronation, Elimination and Enone Hydroarylation of Enaminones. *Org. Biomol. Chem.* **2016**, *14*, 6270–6273. (c) Zhi, W.; Li, J.; Zou, D.; Wu, Y.; Wu, Y. Diastereoselective Synthesis of β -Amino Ketone and Acid Derivatives by Palladium-Catalyzed Conjugate Addition. *Tetrahedron Lett.* **2018**, *59*, 2736–2740.
- (13) Diels, O.; Bibergeil, A. Zur Kenntnis Des O-Diphenols. *Ber. Dtsch. Chem. Ges.* **1902**, *35*, 302–313.
- (14) Moquist, P. N.; Kodama, T.; Schaus, S. E. Enantioselective Addition of Boronates to Chromene Acetals Catalyzed by a Chiral Brønsted Acid/Lewis Acid System. *Angew. Chem., Int. Ed.* **2010**, *49*, 7096–7100.
- (15) Occasionally, base catalysis with DABCO is necessary for the formation of the vinylogous ester or amide. DABCO increases the rate of reaction in most cases.
- (16) For substrate synthesis, full experimental details, compound characterization, and control experiments, see *Supporting Information*.
- (17) (a) Pellegrinet, S. C.; Goodman, J. M. Asymmetric Conjugate Addition of Alkynylboronates to Enones: Rationale for the Intriguing Catalysis Exerted by Binaphthols. *J. Am. Chem. Soc.* **2006**, *128*, 3116–3117. (b) Paton, R. S.; Goodman, J. M.; Pellegrinet, S. C. Theoretical Study of the Asymmetric Conjugate Alkenylation of Enones Catalyzed by Binaphthols. *J. Org. Chem.* **2008**, *73*, 5078–5089.
- (18) Boronic acids protodeboronate more readily than the trifluoroborate salts, decreasing their effective concentration.
- (19) (a) Collins, K. D.; Glorius, F. Employing a Robustness Screen: Rapid Assessment of Rhodium(III)-Catalysed C–H Activation Reactions. *Tetrahedron* **2013**, *69*, 7817–7825. Collins, K. D.; Glorius, F. A Robustness Screen for the Rapid Assessment of Chemical Reactions. *Nat. Chem.* **2013**, *5*, 597–601. (b) Collins, K. D.; Rühling, A.; Glorius, F. Application of a Robustness Screen for the Evaluation of Synthetic Organic Methodology. *Nat. Protoc.* **2014**, *9*, 1348–1353.
- (20) (a) Hack, D.; Chauhan, P.; Deckers, K.; Hermann, G. N.; Mertens, L.; Raabe, G.; Enders, D. Combining Silver Catalysis and Organocatalysis: A Sequential Michael Addition/Hydroalkoxylation One-Pot Approach to Annulated Coumarins. *Org. Lett.* **2014**, *16*, 5188–5191. (b) Gao, Z.; Fletcher, S. P. Construction of β to carbonyl stereogenic centres by asymmetric 1,4-addition of alkylzirconocenes to dienones and ynenes. *Chem. Commun.* **2018**, *54*, 3601–3604. (c) Ignatiuk, Z. A.; Janicki, M. J.; Góra, R. W.; Konieczny, K.; Kowalczyk, R. Applications of Thermal Activation, Ball-milling and Aqueous Medium in Stereoselective Michael Addition of Nitromethane to Enynones Catalyzed by Chiral Squaramides. *Adv. Synth. Catal.* **2019**, *361*, 1108–1116.