

# Electronic Structure and Anisotropic Compression of $Os_2B_3$ to 358 GPa

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## Abstract

High pressure study on ultra-hard transition-metal boride  $Os_2B_3$  was carried out in a diamond anvil cell under isothermal and non-hydrostatic compression with platinum as an X-ray pressure standard. The ambient-pressure hexagonal phase of  $Os_2B_3$  is found to be stable with a volume compression  $V/V_0 = 0.670 \pm 0.009$  at the maximum pressure of  $358 \pm 7$  GPa. Anisotropic compression behavior is observed in  $Os_2B_3$  to the highest pressure, with the  $c$ -axis being the least compressible. The measured equation of state using the 3rd-order Birch-Murnaghan fit reveals a bulk modulus  $K_0 = 397$  GPa and its first pressure derivative  $K_0' = 4.0$ . The experimental lattice parameters and bulk modulus at ambient conditions also agree well with our density-functional-theory (DFT) calculations within an error margin of  $\sim 1\%$ . DFT results indicate that  $Os_2B_3$  becomes more ductile under compression, with a strong anisotropy in the axial bulk modulus persisting to the highest pressure. DFT further enables the studies of charge distribution and electronic structure at high pressure. The pressure-enhanced electron density and repulsion along the  $Os$  and  $B$  bonds result in a high incompressibility along the crystal  $c$ -axis. Our work helps to elucidate the fundamental properties of  $Os_2B_3$  under ultrahigh pressure for potential applications in extreme environments.

Keywords: transition-metal borides, ultra-hard materials, high pressure, diamond anvil cell, synchrotron x-ray diffraction, density functional theory, mechanical properties, crystal anisotropy

## 30 1. Introduction

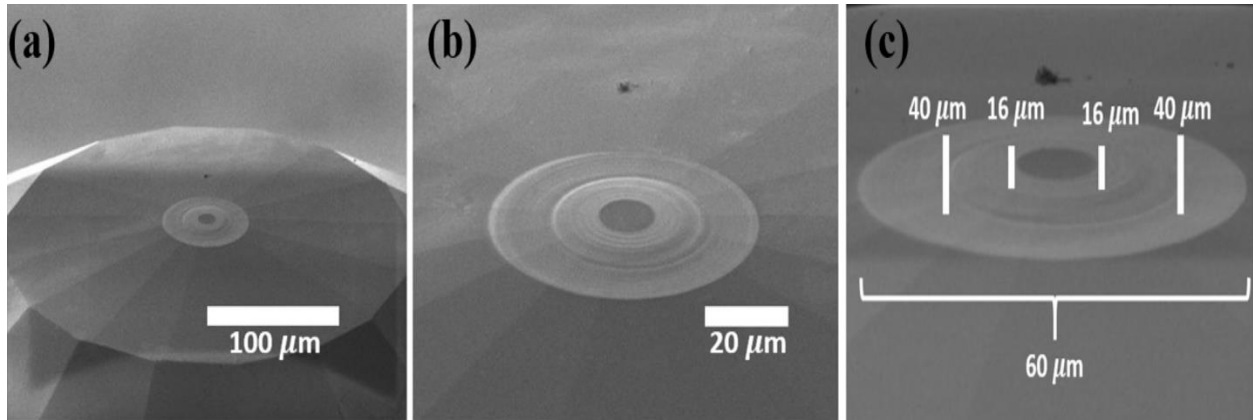
31 The high-pressure behavior of ultra-hard and highly incompressible materials such as  
 32 covalently bonded light elements (*C*, *N*, *O*, and *B*), refractory metals (*Re*, *Os*, etc.), and ceramic  
 33 materials like transition-metal carbides and borides (*WC*, *ReB<sub>2</sub>*, *OsB<sub>2</sub>*, etc.) are of interest because  
 34 of their superior mechanical properties and industrial applications as super abrasives [1-3]. The  
 35 premium example of ultra-hard materials is diamond, and its mechanical and optical properties  
 36 under extreme conditions have been studied in detail [4]. However, diamonds and other carbon-  
 37 based materials are prone to oxidation in air at moderately high temperatures and reaction with  
 38 ferrous alloys, which make them less desirable in many industrial applications. Transition-metal  
 39 borides (TMBs) are proposed alternatives to diamond for abrasives and cutting tools, because some  
 40 TMBs like *OsB<sub>2</sub>* have high hardness of 25-35 GPa [5-7] and thermal stability at high temperatures.  
 41 The light-element boron forms covalent bonds with the transition metal once incorporated  
 42 interstitially in the metal lattice. The directional covalent bonds combined with the high valence  
 43 electron density gives the material its high incompressibility and hardness value. To date, micro-  
 44 hardness at ambient pressure and high-pressure studies have been carried out on to a maximum  
 45 pressures of 25-30 GPa, yet none has been conducted to pressures comparable to the bulk modulus  
 46 of the material other than a static compression of *ReB<sub>2</sub>* to 241 GPa [8]. Gu et al. have showed that  
 47 many TMBs, such as *OsB<sub>2</sub>*, *OsB*, and *Os<sub>2</sub>B<sub>3</sub>*, have similar compression curves as diamond up to  
 48 30 GPa [9]. While many of the osmium-based TMBs were studied in detail, *Os<sub>2</sub>B<sub>3</sub>* has received  
 49 little attention. The boron atom positions and crystal structure within *Os<sub>2</sub>B<sub>3</sub>* were not accurately  
 50 determined until neutron diffraction experiments revealed its *B-B* distances in 2012 [10].

51 Recent advancements in diamond anvil cell (DAC) devices have allowed compression  
 52 experiments to reach multi-megabar pressures using focused ion beam (FIB) machined diamond  
 53 anvils [11-14]. These toroidal designs produced in Refs. [11] and [12] help reduce elastic  
 54 deformation known as cupping in diamonds under extreme pressures. These developments have  
 55 allowed diamond anvils to sustain static pressures as high as 5 megabars (500 GPa) in DACs,  
 56 replicating pressures of inner-planetary cores [12]. In this study, we utilize the toroidal diamond  
 57 anvil technology combined with X-ray diffraction (XRD) to compress the *Os<sub>2</sub>B<sub>3</sub>* hexagonal phase.  
 58 For the first time, ultra-incompressible *Os<sub>2</sub>B<sub>3</sub>* is studied under static pressures up to  $358 \pm 7$  GPa.

The experimental data show good agreements with corresponding first-principles simulations. This combined experiment-theory work provides insights to elucidate the fundamental properties of  $Os_2B_3$  under ultrahigh pressure for extreme-environment applications.

## 2. Experimental and Computational Methods

Compression experiment on  $Os_2B_3$  was carried out with specially fabricated diamond anvils that were machined using a  $Ga^+$  focused ion beam (FIB). A toroidal design with a 16 micron culet and 40 micron outer region was made with a TESCAN LYRA 3 FIB parameters of 30keV accelerating voltage, 4.5 nA beam current, and a 50 nm beam size at the University of Alabama's Central Analytical Facility. The total depth of the toroid design was milled to 3 microns and was surrounded by an outer flat region extending to 60 microns shown in Figure 1. Two anvils were fabricated in this fashion and placed within a membrane driven diamond anvil cell (DAC) in an opposed anvil configuration. Compression of the sample material was conducted without using a pressure medium and was non-hydrostatic in nature. A rhenium gasket was indented to 25 micron thickness with a laser drilled sample hole of 8 micron diameter for sample placement. The gasket aids in the prevention of radial displacement of the sample material as well as minimizing diamond-diamond contact. The hexagonal phase of  $Os_2B_3$  sample material (with space group  $P6_3/mmc$ ) was bought from American Elements of purity 99.9% and mixed with platinum powder (99.97% purity) from Alfa-Aesar for pressure calibration.



**Figure 1.** Top view of machined toroidal anvil atop beveled diamond anvil culet. (b) Close-up image of toroidal design on the central culet (c) Close-up side view of the toroidal anvil with design parameters of 16, 40, and 60 microns labeled.

X-ray diffraction (XRD) experiments were performed utilizing the X-ray diffraction setup at the HPCAT Beamline 16 BM-D at the Advanced Photon Source, Argonne National Laboratory [15]. The DAC was positioned so that an X-ray beam of wavelength  $\lambda = 0.4133 \text{ \AA}$  and beam size  $3.7 \text{ \mu m}$  (vertical)  $\times$   $3.8 \text{ \mu m}$  (horizontal) FWHM (full width half maximum) along the compression axis and incident on the sample. Pressure gradients across the beam width were not measured, however, they are expected to be minimal due to the small size of the x-ray beam (3-4  $\text{\mu m}$ ) compared to the diameter of the diamond culet (16  $\text{\mu m}$ ). XRD measurements were taken in approximately 2 GPa pressure steps totaling 157 data points. Scattered X-rays were collected on a Pilatus 1M detector situated downstream of beam incidence with sample to detector distance calibrated to 344.63 mm using  $\text{CeO}_2$  calibration file analyzed in the Dioptas software. A schematic experimental setup was reported in Ref. [8]. The integrated sample XRD profiles were processed using the GSAS-II software [16] to extract crystal lattice parameters for the  $\text{Os}_2\text{B}_3$  hexagonal phase, where the Os atom positions were held at (1/3, 2/3, 1/4) and the B atoms at (2/3, 1/3, 3/4). The  $\text{Os}_2\text{B}_3$  data were fit to the 3rd-order Birch-Murnaghan equation of state (EOS) to determine the bulk modulus  $K_0$  and its pressure derivative  $K_0'$  using the sample volumetric compression  $x = V_0/V$ :

$$P(V) = \frac{3}{2} K_0 \left[ x^{\frac{7}{3}} - x^{\frac{5}{3}} \right] \left[ 1 + \frac{3}{4} (K_0' - 4) \left( x^{\frac{2}{3}} - 1 \right) \right]. \quad (1)$$

Ambient measurements of lattice parameters were taken at the University of Alabama at Birmingham via XRD analysis, and they were determined to be  $a_0 = 2.915 \text{ \AA}$  and  $c_0 = 12.92 \text{ \AA}$  for  $\text{Os}_2\text{B}_3$ . The ambient pressure platinum lattice parameter was also measured to be  $a_0 = 3.924 \text{ \AA}$  and LeBail refinement of platinum experimental XRD peaks showed error within pressure measurements to be under 2.0%.

The density functional theory (DFT) [17, 18] calculations were performed using a plane-wave basis and pseudopotential method as implemented in VASP (the Vienna ab initio simulation package, version 5.4.4) [19, 20]. We adopted the projector augmented wave (PAW) method [21, 22], and considered both the Ceperley–Alder–Perdew–Zunger local density approximation (LDA) functional [23] and the Perdew–Burke–Ernzerhof generalized gradient approximation (GGA) functional [24]. The Os  $5d^6 6s^2$  and B  $2s^2 2p^1$  states were treated as valence electrons, and a kinetic energy cutoff ENCUT of 420 eV was used in the expansion for the valence wave functions. Our

choice of ENCUT, which is 30% larger than the recommended value in the VASP pseudopotential files, suffices to converge the DFT total energy with a difference  $< 10^{-4}$  eV/atom. The Monkhorst–Pack sampling integration over the Brillouin zone [25] was chosen by a  $\Gamma$ -centered k-point mesh with a grid size of  $31 \times 31 \times 7$  (resolution  $= 0.01 \times 2\pi/\text{\AA}$ ). The convergence criteria for self-consistent and structure relaxation calculations were set as  $10^{-6}$  eV/unit cell and  $10^{-3}$  eV/ $\text{\AA}$ , respectively. The simulations are performed at zero temperature and various external pressures. At each given external pressure point, we first computed the fully relaxed lattice constants and atomic positions. Our calculated ambient lattice constants with the GGA functional are  $a_0 = 2.943$   $\text{\AA}$  and  $c_0 = 12.939$   $\text{\AA}$ , while those with the LDA functional are  $a_0 = 2.906$   $\text{\AA}$  and  $c_0 = 12.814$   $\text{\AA}$ . Both GGA and LDA results are similar to the experimental values, within an error margin of 1.0%.

After the structure relaxation, we then computed the elastic tensor by using the strain–stress method [26] as embedded in VASP. The calculations included contributions from distortions of rigid ions and ionic relaxations. The ionic contributions can be obtained by inverting the ionic Hessian matrix, which is the matrix of second energy derivatives with respect to atomic positions, and multiplied by the internal strain tensor [27]. Base on the stress–strain relationship, we can derive the elastic constants  $C_{ij}$  and determine various mechanical properties. For example, in the Vogit–Reuss–Hill approximations [28–30], the bulk modulus ( $K$ ) and shear modulus ( $G$ ) are obtained as follows:

$$K = (K_V + K_R)/2, \quad G = (G_V + G_R)/2, \quad (2)$$

$$K_V = [(C_{11} + C_{22} + C_{33}) + 2(C_{12} + C_{13} + C_{23})]/9, \quad (3)$$

$$K_R = 1/[S_{11} + S_{22} + S_{33} + 2(S_{12} + S_{23} + S_{31})], \quad (4)$$

$$G_V = [(C_{11} + C_{22} + C_{33}) - 2(C_{12} + C_{23} + C_{31}) + 3(C_{44} + C_{55} + C_{66})]/15, \quad (5)$$

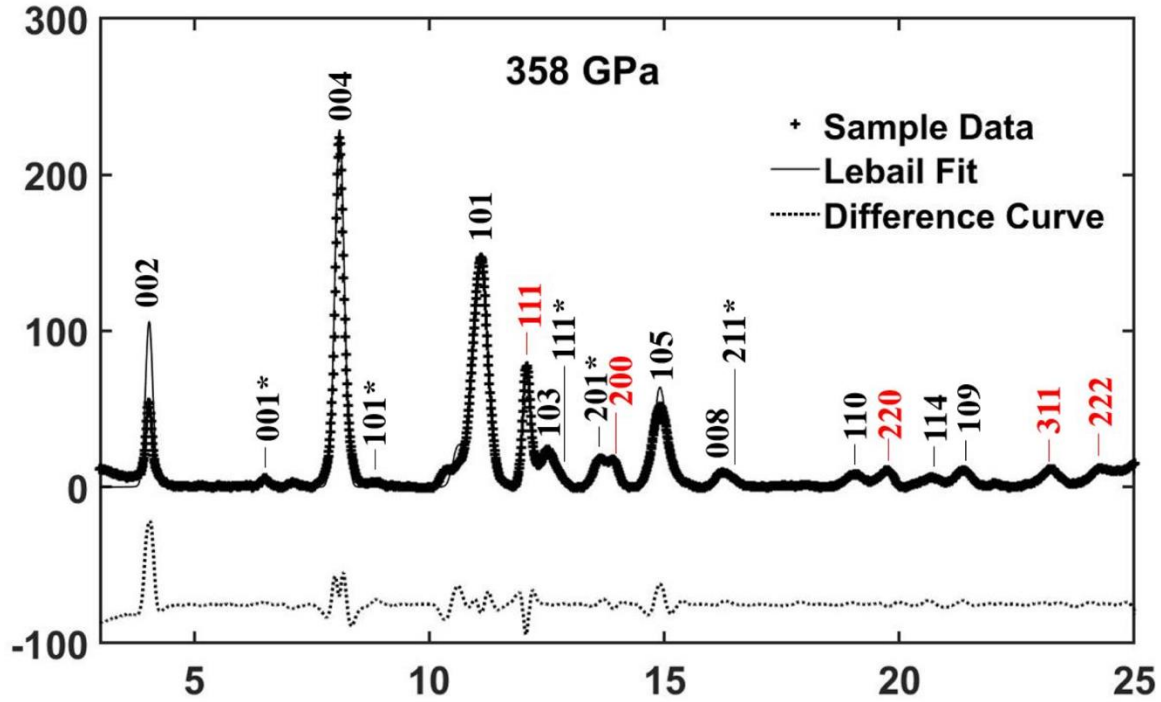
$$G_R = 15/[4(S_{11} + S_{22} + S_{33}) - 4(S_{12} + S_{23} + S_{31}) + 3(S_{44} + S_{55} + S_{66})]. \quad (6)$$

Here,  $C_{ij}$  and  $S_{ij}$  are elastic constants and elastic compliances, respectively. The computed bulk modulus values  $K_0$  at ambient conditions are 358.9 GPa with GGA, and 391.7 GPa with LDA. The LDA bulk modulus is consistent with previous theoretical values [9, 31], and it also agrees better (within an error margin of 1.5%) with the experimental value  $K_0 = 397$  GPa obtained by fitting the experimental  $P$ - $V$  curve to equation (1). In our GGA and LDA calculations of the mechanical

properties, their overall trends with pressure are comparable. Since the results from the LDA functional agree better with the corresponding experiments, below we will mainly present and discuss the LDA results. Finally, the theoretical structural visualization and charge distribution were plotted by the VESTA software (version 3.4.8) [32].

### 3. Results

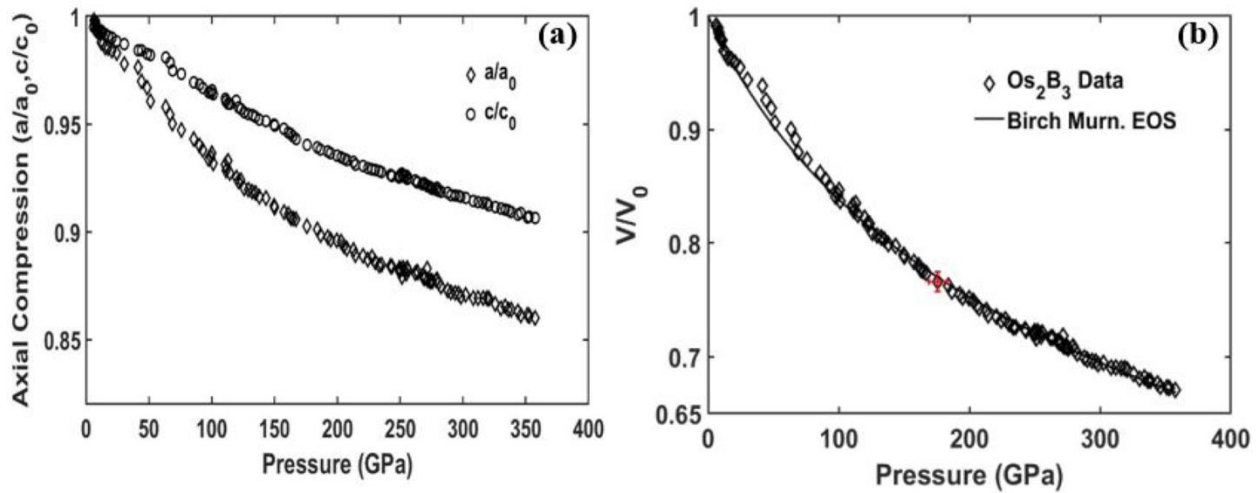
Figure 2 shows the integrated XRD pattern with Leball fit to the XRD data at the maximum pressure of  $358 \pm 7$  GPa. The dotted curve shown below the XRD peaks is the difference curve between the sample data and the Leball fit to the data. Strong  $Os_2B_3$   $hkl$  peaks are labeled and coexist with weaker  $OsB_2$  impurity peaks labeled with asterisks. Platinum peaks used for pressure calibration are labeled in red in Figure 2.



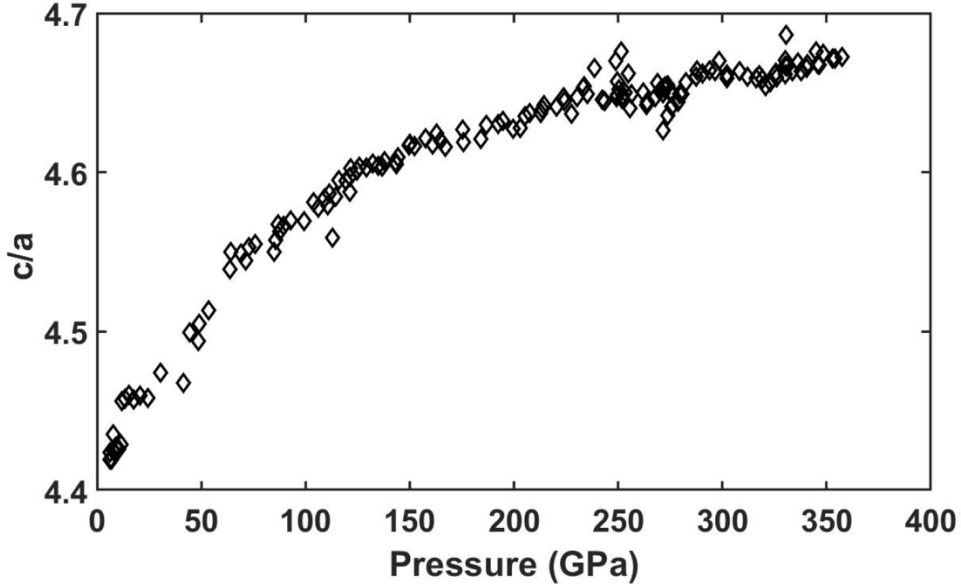
**Figure 2.** Integrated x-ray diffraction profile along with Leball fit for the weak orthorhombic  $OsB_2$  (marked by an asterisk) and the hexagonal  $Os_2B_3$  phases at the highest pressure of  $358 \pm 7.04$  GPa. The platinum pressure marker peaks are indicated in red. The dotted curve at the bottom is the difference curve between the observed intensities and the Leball fit.

Figure 3(a) displays the relative axial compression of lattice parameters for the hexagonal phase of  $Os_2B_3$ . The lattice  $a$  parameter was shown to be the most compressible, while the lattice

$c$  parameter is the least compressible, with the values of  $a = 2.507 \pm 0.004 \text{ \AA}$  and  $c = 11.71 \pm 0.021 \text{ \AA}$  at the maximum pressure. It is to be noted that the anisotropy between compression of the  $a$ -axis and  $c$ -axis increases with pressure to the highest pressure of  $358 \pm 7 \text{ GPa}$ . The non-hydrostatic component of pressure may impact the observed anisotropy at low pressures, however, at higher pressures when pressure is much greater than the shear strength of material we expect these effects to be minimal. The effects of anisotropy also can be seen in the compression ratio of  $c$  and  $a$  lattice parameters in Figure 4. The  $a$ -axis is compressed more rapidly, resulting in an increasing  $c/a$  ratio to the maximum value of  $c/a = 4.672 \pm 0.008$ . The ratio begins to flatten out with higher pressure as the axes are compressed more equally. Due to enhanced interatomic interactions with increasing pressure, the material becomes stiffer. At higher enough pressure, the  $a$  and  $c$  lattice parameters will eventually tend to saturate and become weakly pressure-dependent, resulting in a slower increase of the  $c/a$  ratio. Figure 3(b) is the corresponding volumetric compression data fitted with the 3rd-order Birch-Murnaghan EOS. No phase change was observed, and the hexagonal  $\text{Os}_2\text{B}_3$  phase remained stable to 358 GPa, with a maximum volume compression  $V/V_0 = 0.670 \pm 0.009$ . As mentioned above, the ambient bulk modulus  $K_0 = 397 \text{ GPa}$  and pressure derivative  $K_0' = 4.0$  were obtained by fitting with equation (1).



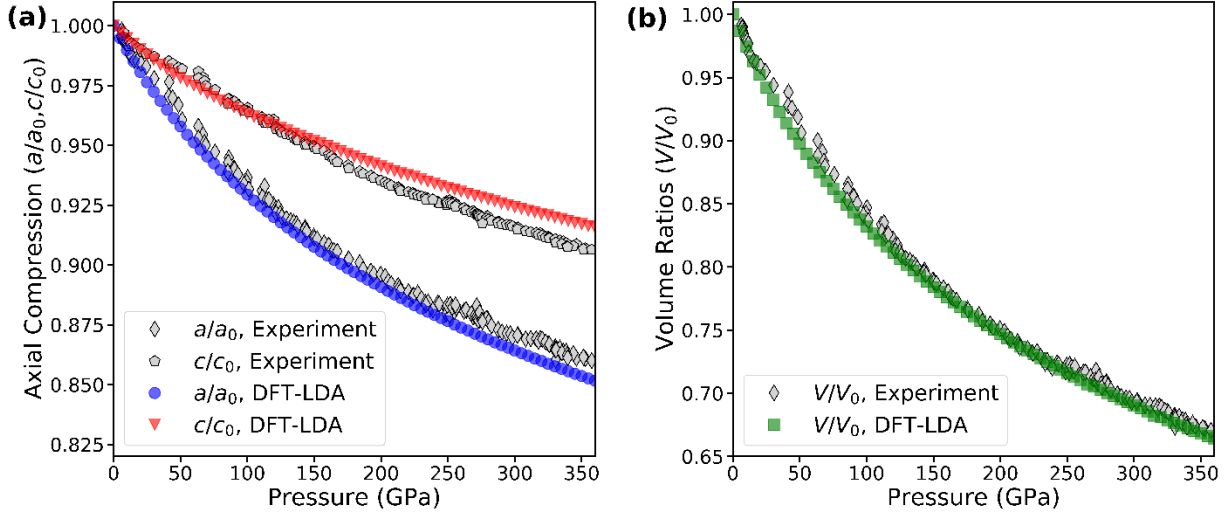
**Figure 3.** (a) Axial compression of  $\text{Os}_2\text{B}_3$  as a function of pressure up to  $358 \pm 7.04 \text{ GPa}$ . (b) Volumetric compression of  $\text{Os}_2\text{B}_3$  fitted to the 3rd-Order Birch Murnaghan equation of state. Sample compression remains stable and no phase change was observed.



**Figure 4.** Compression ratio of the lattice  $c$  and  $a$  parameters as a function of pressure up to  $358 \pm 7$  GPa. The  $c/a$  ratio is seen to increase and flatten with increasing pressure.

Figure 5 shows our DFT calculations of lattice parameters with the LDA functional for hexagonal  $Os_2B_3$  under hydrostatic compression up to 360 GPa. The DFT-LDA ambient lattice parameters  $a_0 = 2.906$  Å and  $c_0 = 12.814$  Å are within an error margin of 1% compared to the experiments. At the highest pressure of 360 GPa, the DFT-LDA value of  $a/a_0 = 0.852$  slightly underestimates the experimental value of 0.860, while the DFT-LDA value of  $c/c_0 = 0.916$  slightly overestimates the experimental value of 0.906. On the other hand, the DFT-LDA value of  $V/V_0 = 0.664$  agree well with the experimental value of 0.670 at maximum compression. As mentioned in the previous section, the computed DFT-LDA bulk modulus value of  $K_0 = 391.7$  GPa also matches well with the experimental fitted value of  $K_0 = 397$  GPa. Therefore, for hexagonal  $Os_2B_3$ , DFT calculations utilizing the LDA functional have led to good theory-experiment agreements. We note that our theoretical volume versus pressure curve is a concave up function, which suggests that the curve protrusion observed experimentally between the pressure range 40 – 70 GPa in Figure 3 is potentially caused by a non-hydrostatic condition. The theoretical results also indicated a distinct anisotropic behavior, with the  $c$ -axis being more incompressible than the  $a$ -axis. Below we will discuss the anisotropic behavior in more detail by further examining the electronic structures and pressure evolutions of elastic constants.

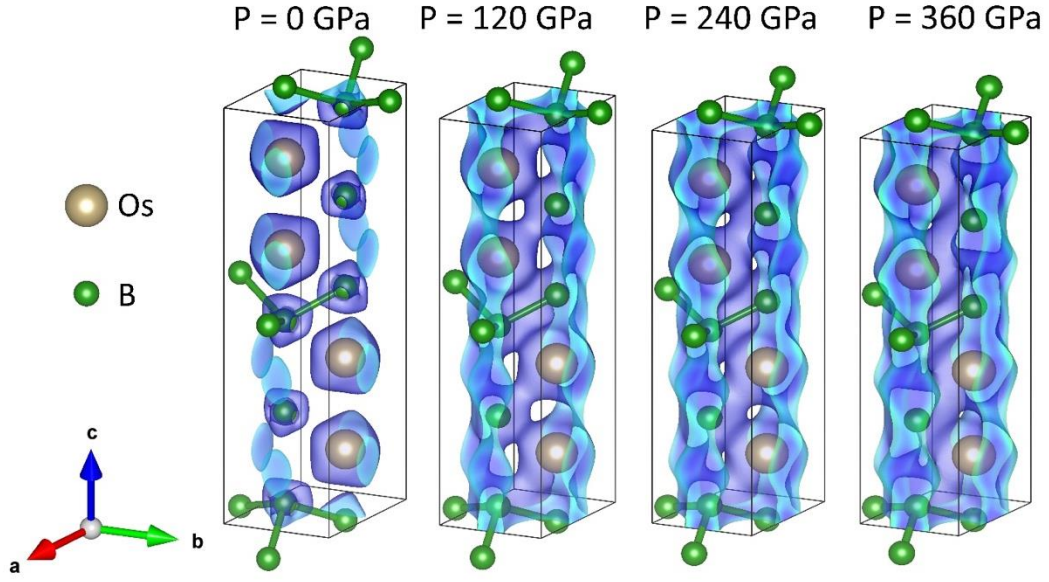




**Figure 5.** (a) Theoretical hydrostatic pressure dependence of the axial compression for lattice parameters  $a/a_0$  and  $c/c_0$  up to 360 GPa for hexagonal  $Os_2B_3$ . The density functional theory (DFT) calculations were performed with the local density approximation (LDA) functional. (b) Corresponding ratio of unit-cell volume  $V/V_0$  versus pressure curve. The experimental values are plotted accordingly.

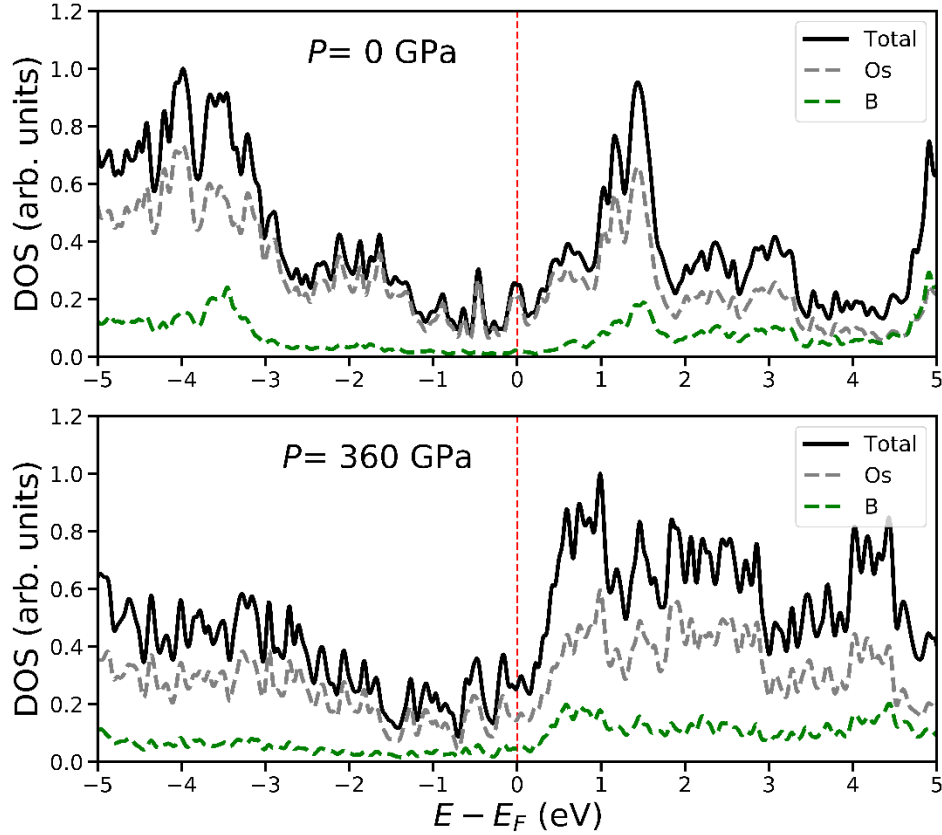
#### 4. Discussion

Figure 6 shows the  $Os_2B_3$  hexagonal structures (with space group  $P6_3/mmc$ ) and iso-surfaces of charge density under different external pressures. The crystal structure of  $Os_2B_3$  can be depicted as stackings of  $Os$  and  $B$  layers along the  $c$ -direction. The  $B$ - $B$  bond length is near 1.9 Å. There is an offset between the  $B$  and  $Os$  atoms in the  $a$ -axis, suggesting that their electrostatic repulsive forces do not directly push each other. On the other hand, the  $B$  and  $Os$  atoms are directly aligned along the  $c$ -axis, which leads to highly oriented repulsive electron–electron interactions. When the applied pressure increases, in addition to  $B$ - $B$  covalent bonds, a substantial amount of electron density is centered along the  $Os$ - $B$  bond direction, indicating strong directional bonding behavior along the  $c$ -axis. Since materials with a large hardness are likely to have highly directional bonds [7, 8, 33], our theoretical results suggest that the highest hardness of  $Os_2B_3$  should occur along the  $c$ -axis.



**Figure 6.**  $Os_2B_3$  hexagonal crystal structures and iso-surfaces of charge density at different pressures. The iso-surface levels were set to be  $0.1 a_0^{-3}$  (with  $a_0$  being the Bohr radius). The input data for the VESTA plotting software are based on the VASP output file CHGCAR, which contains the lattice vectors, atomic coordinates, the total charge density multiplied by the volume on the fine fast Fourier transform-grid, and the projector augmented wave one-center occupancies.

In addition to charge density, we have studied the electronic density of states (DOS) of  $Os_2B_3$ . Figure 7 shows the computed DOS at 0 GPa (top) and 360 GPa (bottom), respectively. We have adjusted the Wigner-Seitz radius for each atom type at different pressure point, in order to reduce the overlap between atomic spheres when integrating the DOS. At 0 GPa, the DOS has a predominant  $Os$  character at the Fermi level ( $E_F$ ). At high pressure, the partial DOS at  $E_F$  from  $Os$  and  $B$  atoms both increase, implying that metallic bonding is increased or covalent bonding is reduced. The behavior is consistent with the iso-surface charge density plot in Figure 6 and the pressure-enhanced ductility to be discussed later in Figure 9. On the other hand, pressure also causes a stronger hybridization between  $Os$  and  $B$  atoms, as well as between  $B$  atoms themselves. Therefore, strong covalent  $Os-B$  and  $B-B$  bonds are still present. The strong  $Os-B$  bonding is also responsible for the  $c$ -axis incompressibility of  $Os_2B_3$  at high pressure.



**Figure 7.** Theoretical  $Os_2B_3$  density of states (DOS) calculated by density functional theory (DFT) using the local density approximation (LDA) functional at pressure  $P = 0$  GPa (top) and  $P = 360$  GPa (bottom). The vertical dashed line indicates the Fermi level ( $E_F$ ).

To further study the mechanical properties, we resort to the elastic constants  $C_{ij}$  obtained directly in our DFT calculations. Due to the hexagonal symmetry of  $Os_2B_3$  (with a space group  $P6_3/mmc$ ), there are only five independent elastic coefficients:  $C_{11}$ ,  $C_{12}$ ,  $C_{13}$ ,  $C_{33}$ ,  $C_{44}$ . These five elastic coefficients together with  $C_{66} = (C_{11} - C_{12})/2$  are shown in Figure 8(a).  $C_{11}$  and  $C_{33}$  represent resistances to compression along the  $a$ -axis and  $c$ -axis, respectively. It is noted that  $C_{11}$  and  $C_{33}$  are largely increased under pressure, indicating a strong incompressibility along both axes. Furthermore,  $C_{33}$  is larger than  $C_{11}$ , which suggests that the  $c$ -axis is the least compressible. As discussed before, this strong  $c$ -axis incompressibility is related to the strong electron charge density and repulsion in the  $Os$ - $B$  bonds along the  $c$ -direction.  $C_{44}$  and  $C_{66}$  are the parameters that directly determine the indentation hardness of a solid [34]. The large  $C_{44}$  and  $C_{66}$  indicate a strong resistance to uniaxial shear strain in the (100) and (001) planes, respectively. Here,  $C_{66} < C_{44}$ , hence shear distortion along the (100) plan is more difficult. Moreover,  $C_{11}$  and  $C_{33}$  are higher than  $C_{44}$

and  $C_{66}$ , which indicate that unidirectional compression resistances along the main crystal axes are higher than the shear deformation resistances.

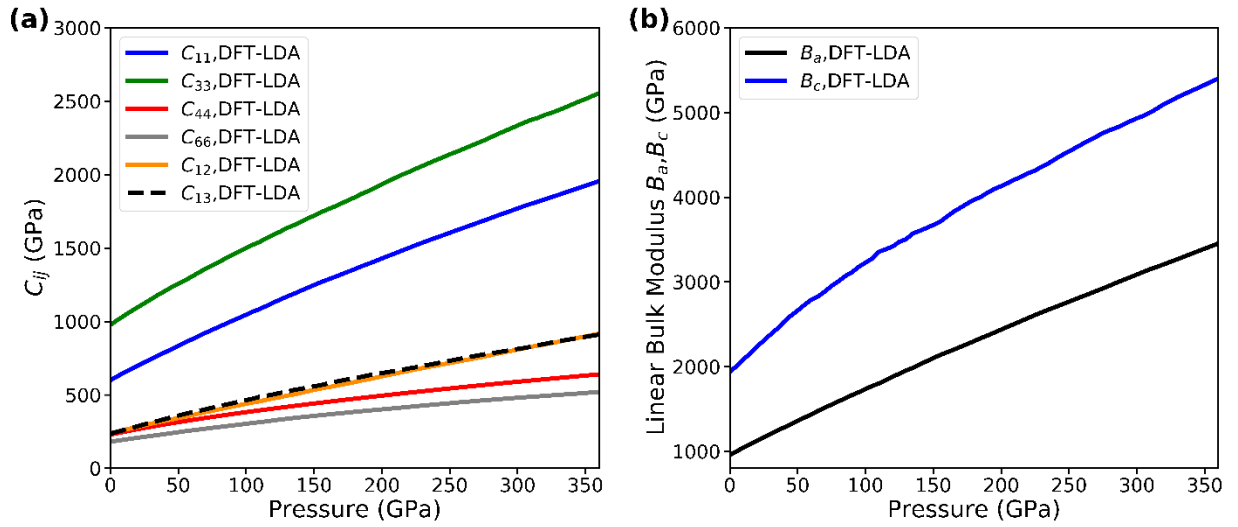
We next consider the axial bulk moduli along the  $a$ -axis ( $B_a$ ) and  $c$ -axis ( $B_c$ ) [35] to describe the anisotropic elasticity of  $Os_2B_3$ :

$$B_a = \alpha / (2 + \beta) \quad , B_c = B_a / \beta, \quad (7)$$

$$\alpha = 2(C_{11} + C_{12}) + 4C_{13}\alpha + C_{33}\alpha^2, \quad (8)$$

$$\beta = (C_{11} + C_{12} - 2C_{13}) / (C_{33} - C_{13}). \quad (9)$$

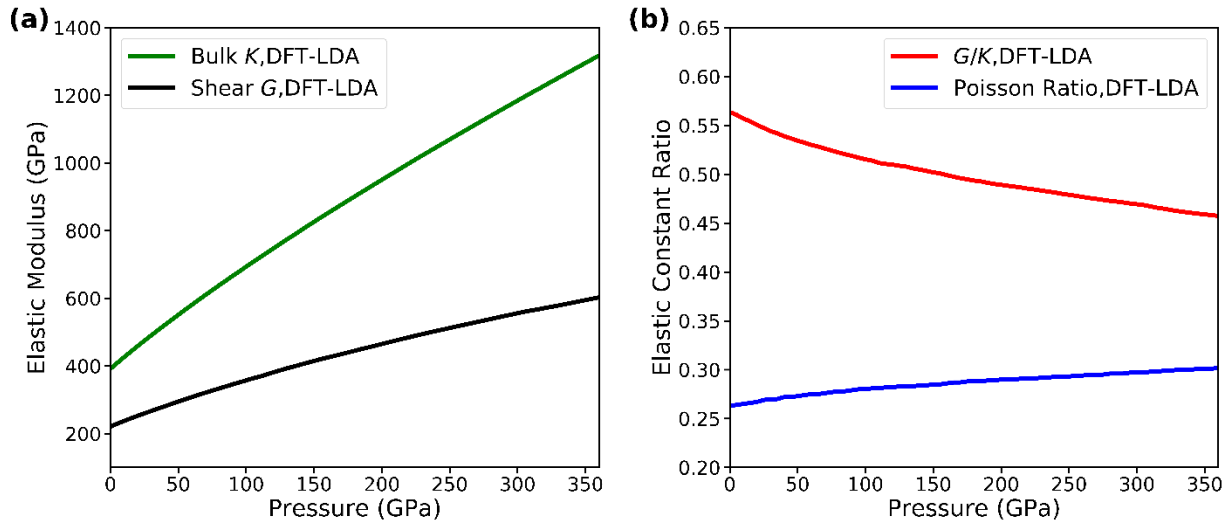
The pressure evolutions of  $B_a$  and  $B_c$  are given in Figure 8(b). It is noted that  $B_c$  is always larger than  $B_a$  in the pressure range under study, suggesting a strong anisotropy of the elasticity in the hexagonal phase of  $Os_2B_3$ . This result is consistent with the behaviors of  $C_{11}$  and  $C_{33}$  as a function of pressure, showing that compression along the  $c$ -axis is more difficult.



**Figure 8.** (a) Selected elastic constants versus pressure for hexagonal  $Os_2B_3$ . The density functional theory (DFT) calculations were performed with the local density approximation (LDA) functional. (b) Axial bulk moduli (defined in the main text) along the  $a$ -axis ( $B_a$ ) and  $c$ -axis ( $B_c$ ) as a function of pressure. The calculations were based on the elastic constants in (a).

Finally, it is noted that the shear modulus ( $G$ ) and bulk modulus ( $K$ ) can help predict a material's brittle or ductile behavior, which is related to its reversible compressive deformation and fracture ability. In particular, according Pugh's criterion [36], a material behaves in a ductile

manner if  $G/K < 0.571$ , or otherwise it should be brittle in nature. Figure 9(a) shows the calculated values of  $G$  and  $K$  as a function of pressure. While both  $G$  and  $K$  are increased by compression,  $K$  grows faster than  $G$ . This indicates that  $G/K$  is decreased with pressure, as shown in Figure 9(b). The computed  $G/K$  ratio at 0 GPa is 0.564, and at 360 GPa is 0.457. Since a low (high)  $G/K$  value is associate with ductility (brittleness), the ductile behavior of  $Os_2B_3$  is enhanced with pressure. Poisson's ratio is another good indicator of the brittle or ductile behavior, and it can be obtained by  $(3K - 2G)/[2(3K + G)]$ . According to Frantsevich's rule [37], a material with a larger Poisson's ratio ( $>1/3$ ) is ductile, or otherwise it is brittle. Figure 9(b) shows that the Poisson's ratio of  $Os_2B_3$  is increased under compression, again implying a pressure-enhanced ductility.



**Figure 9.** (a) Theoretical bulk ( $K$ ) and shear ( $G$ ) moduli of  $Os_2B_3$  as a function of pressure. The density functional theory (DFT) calculations were performed with the local density approximation (LDA) functional.  $G$  and  $K$  were obtained by using the Voigt–Reuss–Hill approximation [Equations (2) – (6)] and the DFT-LDA elastic constants. (b) The  $G/K$  and Poisson ratios as a function of pressure.

The study on transition-metal boride under high pressure can offer unique insights in the changes in electronic structure induced by compression. Three dimensional covalent  $sp^3$  bonds of carbon atoms gives diamond its ultra-hardness and incompressibility, indicating the necessary role of covalent bonds if one wishes to replicate diamond's supreme physical properties. As discussed above, the pressure-enhanced  $Os-B$  covalent bonds in  $Os_2B_3$  can cause a strong resistance to compression and a highly anisotropic behavior. Although the bulk modulus value of  $Os_2B_3$  is shown to be comparable to that of diamond, this alone does not necessarily guarantee a high strength, as the latter corresponds to resistance to plastic deformation. There are also  $Os-Os$

metallic bonds that would indicate weak resistance to deformation along certain axis. In addition,  $Os_2B_3$  has a similar hexagonal structure to  $ReB_2$  but differs by having alternating planar boron layers with puckered boron layers [8]. This puckering of boron atoms may act as a potential shield to atomic dislocation under applied loads. Ref. [9] has shown a micro-hardness value of 15 GPa for  $Os_2B_3$ , which is thereby not a super-hard material. However, this does not rule out practical mechanical applications to utilize  $Os_2B_3$  or other transition metal borides, because of their impressive bulk modulus values and stability at high temperatures.

## 5. Conclusion

Isothermal and non-hydrostatic compression experiments utilizing toroidal diamond anvil cell technology has allowed for the first-time a unique look at  $Os_2B_3$  under pressure conditions replicating planetary interiors, with a maximum volume compression value of  $V/V_0 = 0.670 \pm 0.009$  for  $Os_2B_3$  at  $358 \pm 7.04$  GPa. Increasing anisotropy between the lattice  $c$  and  $a$  parameters was also shown to persist to the highest pressure. Bulk modulus and its first pressure derivatives were calculated from a fit to the 3rd-order Birch Murnaghan equation of state for  $Os_2B_3$ , with  $K_0=397$  GPa and  $K_0' = 4.0$ , which are comparable to the values in other ultra-hard materials. The results from density functional theory simulations were in good agreement with the experiments. The high bulk modulus, anisotropic behavior, and ultra-incompressible features of  $Os_2B_3$  thereby make it a potential material for extreme-environment applications.

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