Aryl Phosphonate-Directed Ortho C-H Borylation: Rapid Entry into **Highly-Substituted Phosphoarenes**

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ABSTRACT: Phosphonate-directed *ortho* C–H borylation of aromatic phosphonates is reported. Using simple starting materials and commercially accessible catalysts, this method provides steady access to ortho-phosphonate aryl boronic esters bearing pendant functionality and flexible substitution patterns. These products serve as flexible precursors for a variety of highly-substituted phosphoarenes, and in situ downstream functionalization of the products is described.

Aryl phosphonates are useful motifs in organic synthesis as their versatility allows for transformation into a number of functional groups, including phosphines, phosphoramidates, and phosphorus oxides.¹⁻² Additionally, some aryl phosphonates and related compounds have also shown interesting bioactivities, 3-5 including in the clinical treatment of various cancers. 6

Due to their value, several strategies have been reported to prepare aryl phosphonates. However, current methods to generate functionalized aryl phosphonates typically involve the construction of the C-P bond via pre-functionalized starting materials such as organolithium or Grignard reagents, 7 (pseudo)halides, 8-12 boronic acids, 13-15 carboxylic acids, 16 or hypervalent iodine compounds. 17-18 These methods are often limited by the availability of the starting materials, and thus only aryl phosphonates with simple structures can be synthesized readily. More complex aryl phosphonates, such as the ones bearing ortho-substitutions, are often very challenging to access for this reason. One attractive route to the synthesis of highly-substituted phosphonates is via *ortho*-boryl aryl phosphonates, as these products are readily diversified and are able to be converted into a range of phosphoarenes. However, like the other methods mentioned above, current routes to these intermediates require multiple steps to access (Figure 1).¹⁹

Previous Ortho-Boryl Aryl Phosphonate Synthesis: multi-step route required

- simple catalytic components & easily acessible starting materials
 broad functional group tolerance & flexible substitution pattern
 versatible functionalization of products

Figure 1. Previous Synthesis of Ortho-Boryl Aryl Phosphonates and Our Design.

Over the last couple decades, directed C-H borylation has emerged as a powerful tool to prepare ortho-substituted aryl boronic esters from simple starting materials.²⁰⁻²² A variety of directing groups have been demonstrated in this field, including a number of carbonyls, ²³⁻²⁷ Lewis basic nitrogens, ^{23, 26-32} phenols, ³³ silanes, ³⁴⁻³⁵ thianes, ³⁶⁻³⁹ ethers, ^{23, 26, 40} and halides. ²³ We envisioned phosphonate-directed ortho-borylation as an attractive platform for the preparation of a variety of ortho-functionalized aryl phosphonates (Figure 1). A general method to convert widely accessible simple aryl phosphonates into orthoboryl phosphonates, when combined with the vast chemistry of aryl boronates, would allow for the facile preparation of a wide variety of ortho-functionalized aryl phosphonates and their downstream derivatives.

Previously, we have shown that benzyl phosphines can be used as directing groups for C-H borylation, which provides access to ortho-borylated benzylic phosphines. 41-42 Later, Shi and Takaya independently reported C-H borylation conditions to prepare *ortho*-borylated phosphines via the use of an aryl phosphine as a directing group. 43-44 However, these latter methods are limited to phosphines bearing a single type of aromatic group. Moreover, no reactivity was observed with substrates bearing pre-existing *ortho*-substitutions. Thus, there is a clear need to have an alternative way to prepare these ortho-borylated phosphoryl aromatic compounds that can overcome these limitations. In this paper, we describe the successful realization of that goal, and report conditions for the preparation of orthophosphonate aryl boronic esters via a C-H borylation strategy using phosphonate as a directing group. Using easily accessed starting materials and commercially available catalytic components, we show that the reaction enjoys broad functional group tolerance and is highly flexible in regard to substitution patterns.

Our studies began with the borylation of commercially available diethyl phenylphosphonate, and the preliminary screen of iridium catalysts revealed (COD)Ir(acac) as the optimal iridium source (see Supporting Information). The use of [(COD)₂Ir]BF₄ provided comparable results, but the selection of (COD)Ir(acac) is advantageous because it is bench stable and less expensive.

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We then examined the effects of ligand with the model substrate 1, an aryl phosphonate substituted at the *ortho* position (Table 1). The *ortho*-borylation of 1 would produce the highly-substituted phosphonate 2, whose substitution pattern would be difficult to access using other methods.

Table 1. Reaction Optimization.

entry	ligand	yield of 2	yield of 3
1	no ligand	0%	7%
2	2-picolylamine	0%	4%
3	dtbpy	0%	41%
4	tmphen	0%	38%
5	$P[2-Si-(^{i}Pr)_{2}H-C_{6}H_{4}](o-tol)_{2}$	31%	0%
6	PPh_3	0%	7%
7	$P(4-F-C_6H_4)_3$	4%	0%
8	$P(4-CF_3-C_6H_4)_3$	4%	0%
9	$P[3,5-(CF_3)_2-C_6H_3]_3$	84%	0%

^a Yield determined by ¹H NMR analysis with 1,3,5-trimethoxybenzene as internal standard.

Various ligands were examined in conjunction with (COD)Ir(acac) as catalyst. We began by investigating conditions without added external ligand, such as those that had shown success in our previous work on benzyl phosphine-directed C–H borylation. However, with this catalyst system we did not observe any *ortho*-borylation; instead, the borylation took place at the *meta* position to give 3 (Table 1, entry 1).

Moving forward, very little reactivity was observed with 2-picolylamine as ligand, the ligand used in our benzyl amine-directed C–H borylation (entry 2).³⁰ Although we found that the borylation was more efficient with bipyridine ligands, such as dtbpy and tmphen that have often been used in aryl borylation, there was still no formation of the desired *ortho*-borylated product (entry 3-4). The selective formation of the *meta*-borylation

isomer is in itself notable. This result suggests that the primary selectivity is steric in nature, but the fact that borylation *para* to the phosphonate is not observed also suggests a strong electronic differentiation between the methyl and phosphonate groups.

Excitingly, however, when the phosphine/silane-based bidentate ligand previously reported by Smith, Maleczka, and Krska was used, we observed 31% yield of **2** (entry 5). Despite the encouraging result using this ligand, we continued to explore other options as we recognized that use of a commercially available ligand would make the method more broadly useful. Therefore, we explored simple phosphine ligands as an alternative. Although PPh₃ did not provide any of the *ortho*-borylation product (entry 6), with more electron-deficient triaryl phosphines we started to observe the desired *ortho* selectivity (entry 7-8). Following this trend, we finally identified the use of P[3,5-(CF₃)₂-C₆H₃]₃ in this reaction, which has the optimal balance of electronic and steric properties, affording **2** in 84% yield (entry 9). Page 14.

With optimized reaction conditions developed, the scope of this reaction was then investigated (Scheme 1). The borylated phosphonate 2 was isolated in 78% yield on 1 mmol scale. During purification, we observed some loss of product due to the unstable nature of boronic esters, but such an effect can be minimized by running the reaction at larger scales. For example, scaling the same reaction up to gram scale provided 87% isolated yield of 2.

Substrates without *ortho*-substitution next to the phosphonate require somewhat more stringent conditions, including use of a higher catalyst loading, use of iridium pre-catalyst with a less coordinating counter-ion [(COD)₂Ir]BF₄, and elevated reaction temperature (4). Under these conditions, however, we observed a significant amount of bis-borylation at both *ortho* positions.⁴⁶ To avoid the bis-borylation, we hypothesized that the use of larger directing group could restrict the conformational rotation of the phosphonate group in the mono-borylated product and prevent the second C–H activation at the other *ortho* position. This idea is supported by the use of diisopropyl phosphonate, only mono-borylated product is produced (5).

a Run with [(COD)zlr]BF4. b 120 °C. °7.5 mol % catalyst loading. d Parenthetical yields calculated by 1H NMR with 1,3,5-trimethoxybenzene as internal standard. e 110 °C.

Scheme 1. Scope of Phosphonate-Directed C-H Borylation

In contrast, substrates bearing varied substitutions at the preexisting *ortho* position participate in this reaction under standard conditions and result in borylated products efficiently (613), including ethers (6, 9), trifluoromethyl (7), aniline (8), and
halogens (10-13). With a bulky substitution at the *ortho* position, such as bromine, we found that the reaction requires higher
catalyst loading and temperature to proceed (11), presumably
due to the limitation of conformational rotation of phosphonate
by steric congestion. We found, however, that simply switching
to the smaller dimethyl phosphonate group overcomes this limitation and higher yield is observed under milder conditions
(12). Phosphonates with aromatic substitutions were also easily
borylated under the standard conditions (13); however, product
instability precluded isolation.

With *meta*-substituted aryl phosphonates, the borylation only occurs at the less sterically hindered site (14). These reactions are most efficient with weakly donating or electron-withdrawing groups at the *meta* position (15). With a strong electron-donating group at the *meta* position, yield of the borylated product is lower (16).⁴⁷ However, these electronic effects appear to be very sensitive. Simple inclusion of a weak electron-with-drawing group in the substrate mitigates the effects of the *meta* donor group (17-18).

Aryl phosphonates bearing di-substitutions at other positions can also participate in this reaction (19-20), and even highly-substituted substrates could be converted to the borylated products with good efficiencies (21-22). As with other directed C—H borylation reactions, this reaction is sensitive to steric effects. We did not observe reactivity with alkyl substitution next to the reactive C—H bond (23). However, with a smaller substitution, such as fluorine, high yields of the desired product are observed (24). A naphthyl substrate also works well under these reaction conditions and can be borylated in 90% isolated yield (25). One limitation that we have found with this reaction is that Lewis basic heterocycles, such as pyridine, were not tolerated and resulted in only minimal conversion (26). However, less-basic heterocycles are tolerated (27-28).

^a NaOH, H₂O₂, 0 °C to rt. ^b CuCl₂, acetone/H₂O, 90 °C. ^c CuBr₂, acetone/H₂O, 90 °C. ^d Zn(CN)₂, Cu(NO₃)₂·3H₂O, CsF, MeOH/H₂O, 100 °C.

Scheme 2. In Situ Downstream Functionalization

As one of the most versatile functional groups in organic synthesis, boronic esters can be readily converted into a range of other structures. For example, we were rapidly able to prepare aryl phosphonates containing *ortho* phenols (29), 48 chlorides

(30),⁴⁹ bromides (31),⁴⁹ and nitriles (32)⁵⁰ by applying known methods from the literature (Scheme 2). It is noteworthy that the purification of the boronic esters is not required prior to many downstream functionalizations. This is particularly important because the purification of some borylated products can cause significant material loss due to their instabilities.

In addition, aryl boronic esters are reliable precursors for substituted biaryl compounds. Previously, Westcott and co-workers reported conditions for Suzuki coupling to transform *ortho*boryl aryl phosphonates into biarylphosphonate substrates, such as **33**.¹⁹ The further functionalization of **33** led to the synthesis of a dialkyl biarylphosphine ligand **34**, showing the utility of *ortho*-borylated aryl phosphonate products in ligand synthesis (Scheme 3).⁵¹

Scheme 3. Biaryl Phosphonate Synthesis and Their Utility in Ligand Synthesis

Finally, we wished to compare the relative directing group ability of phosphonates to esters (Scheme 4).²⁴ Under the standard and related conditions (see Supporting Information), borylation *ortho* to the ester (36) is preferred and mono-borylation *ortho* to the phosphonate was not observed. Interestingly, however, subsequent borylation at that site appears to be competitive (37).

 a 5 mol % (COD)Ir(acac), 5 mol % P[3,5-(CF₃)₂-C₆H₃]₃, 5 equiv B₂pin₂, THF/dioxane, 100 °C, 24 h; yields calculated by ^1H NMR with 1,3,5-trimethoxybenzene as internal standard.

Scheme 4. Comparison Between Phosphonate and Ester

In conclusion, we developed conditions for C–H borylation using phosphonate as a directing group, which allows the preparation of *ortho*-phosphonate aryl boronic esters from simple starting materials. This reaction uses commercially available catalytic components and has a broad functional group tolerance. Using this method, highly-substituted aryl phosphonates can be prepared, which is a known synthetic challenge in this field.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Experimental procedures and spectra data (PDF)

Crystallographic data for compound 25 (CIF)

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TOC Graphic:

simple catalytic components & easily acessible starting materials
 broad functional group tolerance & flexible substitution pattern
 versatible product functionalization