

Pentacene and Tetracene Molecules and Films on H/Si(111): Level Alignment from Hybrid Density Functional Theory

Svenja M. Janke^{1,2}, Mariana Rossi^{2,†}, Sergey V. Levchenko^{3,2},
Sebastian Kokott², Matthias Scheffler², Volker Blum^{1,4}

¹ Duke University, Mechanical Engineering and Materials Science, Duke University, Durham, NC, USA

² Fritz Haber Institute of the Max Planck Society, Berlin, Germany

³Skolkovo Institute of Science and Technology, Moscow, Russia

⁴Duke University, Department of Chemistry, Duke University, Durham, NC, USA

Abstract. The electronic properties of hybrid organic-inorganic semiconductor interfaces depend strongly on the alignment of the electronic carrier levels in the organic/inorganic components. In the present work, we address this energy level alignment from first principles theory for two paradigmatic organic-inorganic semiconductor interfaces, the singlet fission materials tetracene and pentacene on H/Si(111), using all-electron hybrid density functional theory. For isolated tetracene on H/Si(111), a type I-like heterojunction (lowest-energy electron and hole states on Si) is found. For isolated pentacene, the molecular and semiconductor valence band edges are degenerate. For monolayer films, we show how to construct supercell geometries with up to 1,192 atoms, which minimize the strain between the inorganic surface and an organic monolayer film. Based on these models, we predict the formation of type II heterojunctions (electron states on Si, hole-like states on the organic species) for both acenes, indicating that charge separation at the interface between the organic and inorganic components is favored. The paper discusses the steps needed to find appropriate low-energy interface geometries for weakly bonded organic molecules and films on inorganic substrates from first principles, a necessary prerequisite for any computational level alignment prediction.

Keywords: tetracene, pentacene, silicon, level alignment, singlet fission, hybrid organic-inorganic materials, surfaces, thin films, monolayer

[†] Present address: Max Planck Institute for Structure and Dynamics of Matter, Hamburg, Germany

1. Introduction

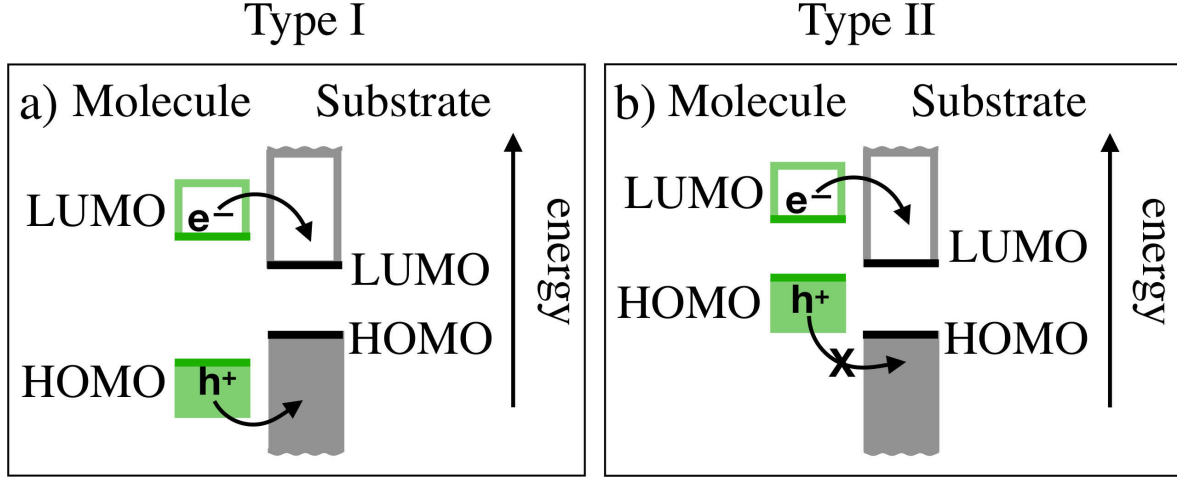


Figure 1. Two possible schematic energy level alignment diagrams for hybrid organic-inorganic systems. a) Type I heterojunction, where the organic film has a wider band gap than the inorganic substrate. b) Type II heterojunction, where the energy levels of the organic component are staggered with respect to the inorganic substrate.

The electronic level alignment between two different semiconductors in contact with each other is of significant technological and physical importance, determining phenomena such as internal charge separation, quantum confinement, or charge recombination in semiconductor heterostructures. A process of particular interest is the generation of excitons in one part of a heterostructure, which may dissociate into individual charged carriers that can be separated at an internal interface. In single-junction solar cells, charge carriers are collected at the band edges of semiconductors, i.e., the conduction band minimum (CBM) and the valence band maximum (VBM). As a result, the energy fraction of each absorbed photon that surpasses the energy of the band gap is usually lost as heat. This energy loss defines the Shockley-Queisser limit[1] that limits the theoretical maximum efficiency of single-junction solar cells to below approximately 30 %. A possible way to overcome the thermalization energy loss is singlet fission[2, 3, 4, 5, 6]. In singlet fission, a high-energy photon generates a singlet exciton, which can dissociate into two lower-energy triplet excitons prior to separation into individual carriers, resulting in four instead of two carriers and preserving a significant fraction of the energy that would otherwise be lost as heat. Two prominent examples for fission materials are pentacene[7, 8] (Pc) and tetracene[9, 10, 11, 12] (Tc), whose triplet exciton energies roughly match the band gap of silicon[3, 13, 8, 14].

In principle, the Shockley-Queisser limit could be overcome by augmenting conventional solar cells with layers of singlet fission materials that enable charge carrier insertion from the triplet excitons into Si[15, 3, 2, 16]. To transfer or split excitons at an organic/inorganic interface, such as Tc and Pc at H/Si(111), the energy level alignment between the “highest occupied molecular orbital” and “lowest unoccupied molecular

orbital” (HOMO and LUMO, respectively, here used synonymously with the VBM and CBM) of the components at the interface is a defining quantity[17, 18, 19]. Figure 1 schematically illustrates two possible energy level alignments between an organic film and an inorganic substrate. Assuming sufficiently small exciton binding energies, a photogenerated triplet exciton in the organic film could either dissociate into two carriers that cross into the substrate at the heterojunction (type I, Figure 1 a) or split into a pair of carriers, of which only one enters the substrate (type II, Figure 1 b)[2, 15]. A type I level alignment has been proposed for Tc on H/Si(111) although hole extraction from H/Si(111) to Tc was still observed in the same study[20]. Direct triplet insertion from Tc into passivated Si appears to be at least hindered [21, 20, 16]. For Pc, based on electron affinity and ionization potential measurements, type II heterojunction behavior was suggested[22].

In this paper, we predict the electronic level alignment of Pc and Tc molecules and monolayer films on intrinsic, i.e., undoped H/Si(111) using first principles theory, specifically using hybrid density-functional theory. For the correct description of this alignment[17, 19, 23, 24, 25] and consequently the singlet fission properties[26], building a model unit cell that reflects the geometric structure at the interface is crucial. However, the resulting unit cells are large and computationally demanding. On the one hand, the interaction between Tc or Pc and H/Si(111) is weak, as evidenced[27] by the observed lack of a wetting layer[28] and the standing in-plane (herringbone) orientation of Pc[29] and Tc[30, 31] molecules around room temperature in thin films and islands on surfaces like H/Si(111). On the other hand, the unit cell lattice parameters for islands, monolayers and thin films of both Pc on H/Si(111)[29, 32, 28, 33] and Pc and Tc at other weakly interacting surfaces[34, 35, 31, 36, 37, 38, 39, 40, 30, 41, 42, 43, 44] do not match the unit cell lattice parameter expected[45] for H/Si(111) well. As a result[46], large low-strain commensurate approximate unit cells are necessary, leading to slab models of the interface with hundreds of atoms or more. Indeed, Pc molecules on H/Si(111) grow in at least two distinct orientations[29], which require large unit cells to approximate them, as shown in Figure 2 and further discussed in Section 3.

The need for large commensurate supercells aggravates the already challenging determination of level alignments from first principles. It is well established that the electronic delocalization errors[47, 48] associated with the relatively affordable level of density-functional theory (DFT) in the generalized gradient approximation (GGA) lead to fundamental band gaps that are too small, sometimes incorrect ordering of electronic levels and hence potentially qualitatively wrong level alignments[49, 50, 51, 52, 53]. Methods that offer significant improvement over DFT-GGA, like the quasi-particle *GW* approximation[54, 49, 55] or hybrid functionals in DFT[56, 57, 58], are associated with high computational cost that either make them very demanding or, in the case of *GW*, essentially prohibit[49, 56, 57] their application to systems of the size required here.

A recent, deeper discussion of the double challenge of potentially large structure model sizes on the one hand and of achieving a sufficiently high level of theory to capture all relevant effects that affect energy level alignments on the other hand can be found,

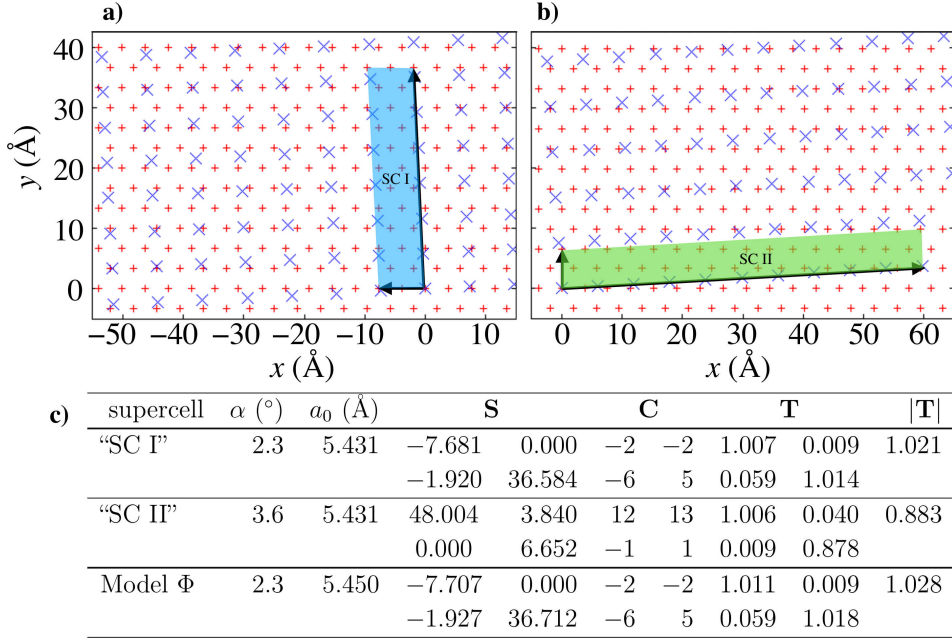


Figure 2. Overlay of the substrate and film lattices based on the experimental Si lattice parameter (5.431 \AA [45], red crosses) and a) the Pc film unit cell alignment I (blue crosses) and b) the Pc film unit cell alignment II (blue crosses) proposed by Nishikata *et al.*[29]. Experimental lattice parameters for the unstrained film unit cells I and II are reproduced in Table 1 further below. Hypothetical film supercells (SCs) that have been strained to match the substrate lattice are shown as blue (SC I) and green (SC II) areas, showing that this can be accomplished by a relatively small strain for SC I vs. a rather large strain for SC II. c) Numerical representation of the strained film SCs shown in subfigures a) and b) that lead to exact coincidence of the supercell lattice vectors with the underlying substrate lattice. α is the rotational angle of the Pc film on the substrate and a_0 is the assumed lattice parameter of the substrate. Eqs. 2, 3 and 4 in Sec. "Computational Approach" define the in-plane lattice parameter matrix of the combined film-substrate supercell, **S**, the supercell matrix **C** with respect to the substrate's (111) plane and the strain transformation **T** from the unstrained (incommensurate) to the strained (commensurate) film. **|T|** summarizes the area strain between the unstrained[29] and the strained film supercell. **S** and **T** for "SC I" and "SC II" are based on the experimental substrate lattice parameter of Si and the experimental film lattice parameters of Ref.[29]. "Model Φ " is the same model as SC I but its **S** is defined using the computational lattice parameter for bulk Si (DFT-PBE+TS level of theory, see below).

e.g., in Ref. [59]. In that work, a nearly strain-free interface (poly(para-phenylene) at the rock-salt ZnO(100) surface) was investigated in detail, allowing for small model sizes and thus an analysis of the electronic level alignment up to the G_0W_0 level of theory. In contrast, the present work focuses on analyzing the construction of nearly strain-free computational structure models in a system where a small-cell, low-strain approximate structure is not available. Electronic effects are analyzed at the already rather challenging level of hybrid DFT, which offers the appropriate mathematical form to capture at least highest occupied and lowest unoccupied levels in principle[58] and

which can be extended to a form that accounts for simple consequences of screening as well.[60]

In the following, we address the task of creating suitable low-strain supercells for interfaces between Pc or Tc and H/Si(111). Two types of organic-inorganic system geometries were investigated for this work: (i) molecules adsorbed at H/Si(111) in isolation from one another, referred to as the “dilute limit”, and (ii) molecules forming a closed monolayer-like film on H/Si(111), referred to as the “monolayer limit” below. By combining these models with hybrid density functional theory calculations, we arrive at a fully computational approach for predicting the structure and level alignment between these acene films and the H/Si(111) substrate.

2. Computational Approach

Computational Details

All DFT calculations were carried out using the all-electron electronic structure code FHI-aims[61, 62, 63, 64] with large scale calculations facilitated by the ELSI[65] infrastructure, the ELPA eigenvalue solver[66], and a linear-scaling implementation of hybrid functionals in periodic DFT[67, 68]. For structure prediction, we used DFT-GGA in the Perdew-Burke-Ernzerhof (PBE) exchange-correlation functional[69] together with the Tkatchenko-Scheffler (TS) pairwise dispersion scheme[70]. In combination with PBE, the TS scheme has been shown to reproduce lattice vectors and volumes of organic crystals closely[71, 72, 73] and predicts lattice parameters and internal geometries within 2 % of experimental results for acene bulk materials[74]. In the present work, we find similarly good agreement for crystal polymorphs of Pc and Tc, as well as for the lattice constant for bulk Si, as shown in the supplementary material, Table S1, Table S2 and Table S3 (these tables also include a comparison to a more recent many-body dispersion (MBD) scheme[75]).

Hybrid density functionals include a fraction of non-local exact exchange that partially corrects the delocalization error. Compared to *GW*, hybrid DFT provides a more affordable balance between accuracy and computational cost, while retaining the appropriate mathematical form[58] to yield acceptable fundamental gaps for typical semiconductors[56, 57]. We investigated the electronic level alignments using the Heyd-Scuseria-Ernzerhof (HSE06) functional with $\alpha = 25$ % Hartree-Fock exchange and a screening parameter of $\omega = 0.11$ (Bohr radii) $^{-1}$ [76, 77, 78] by single-point calculations using DFT-PBE+TS predicted geometries. Full and projected electronic densities of states (PDOS) were computed using a Gaussian broadening function with a width of 0.1 eV. The energetic positions of the frontier levels of the organic and inorganic subsystems were extracted from a Mulliken analysis[79], on which the PDOS computations are also based. Visualizations of atomic configurations were obtained using the Jmol[80] and VMD[81] computer programs.

Adsorption energies of acene molecules on the substrate are calculated as

$$\Delta E = \frac{E_{\text{Sub+Mol}}^{n \times m} - n \times m \cdot E_{\text{Sub}}^{1 \times 1} - N_{\text{Mol}} \cdot E_{\text{Mol}}}{N_{\text{Mol}}}. \quad (1)$$

$E_{\text{Sub+Mol}}^{n \times m}$ is the total energy of the combined molecules-on-substrate model, N_{Mol} is the number of molecules, and $n \times m$ is the number of atoms per layer in the H/Si(111) substrate. E_{mol} and $E_{\text{Sub}}^{1 \times 1}$ are the total energy of the isolated molecule and of a (1×1) unit cell of the H/Si(111) substrate in vacuum, respectively. If multiple molecules are present in a cell, we calculate the average of descriptors of the structure (the absorption distance d_z , angle with the surface normal θ and the herringbone angle ω between the molecules).

H/Si(111)

The lateral unit cell parameter of all final system geometries involving H/Si(111) slabs was set to $a_{111} = 3.854 \text{ \AA}$, derived from the predicted lattice parameter of bulk Si, $a_0 = 5.450 \text{ \AA}$, using DFT-PBE+TS and “tight” settings (see Table S3 in the supplementary material for the variation of a_0 with different density functionals). All slab geometries involving H/Si(111) were hydrogenated on both sides. Only one side was decorated with molecules for models of the organic films. A dipole correction[82] was used to minimize any residual interaction between slab surfaces across the vacuum.

To determine acceptable but affordable slab thicknesses of H/Si(111), we investigated the convergence of the substrate electronic frontier levels as a function of slab thickness without adsorbates. These simulations were carried out using vacuum layers of 200 \AA between the slabs, $(12 \times 12 \times 1)$ k-point grids, and FHI-aims’ “tight” computational defaults, i.e., benchmark-quality settings[83, 84]. The H positions and outermost two Si double layer atomic positions were fully relaxed at the DFT-PBE+TS level of theory. As shown in Figure S1 in the SI, the energetic positions of the VBM and CBM converge slowly with the number of layers in the slab, similar to other findings in the literature[85, 86, 87, 88] and attributed to quantum well behavior due to confinement of the electronic eigenstates in the thin slab[88, 87, 89]. For the larger DFT-HSE06 supercell calculations including organic films, we use ten- and six-double-layer slabs. As shown in Figure S1, for six double-layers, the CBM calculated by DFT-HSE06 is still approximately 0.2 eV higher than for thick slabs. Similarly, the calculated band gap of the six-double-layer slab is 1.465 eV, approximately 0.3 eV higher than the calculated bulk band gap. The slow convergence of the gap with slab thickness and the computational cost for DFT-HSE06 calculations of structures above 1,000 atoms (the largest film models considered in this work using six-double-layer slabs) make it impossible to systematically consider much thicker slab models. We return to this point below, concluding that the expected remaining CBM shift for a thicker slab would not be large enough to alter the qualitative film-substrate level alignments resulting from our calculations.

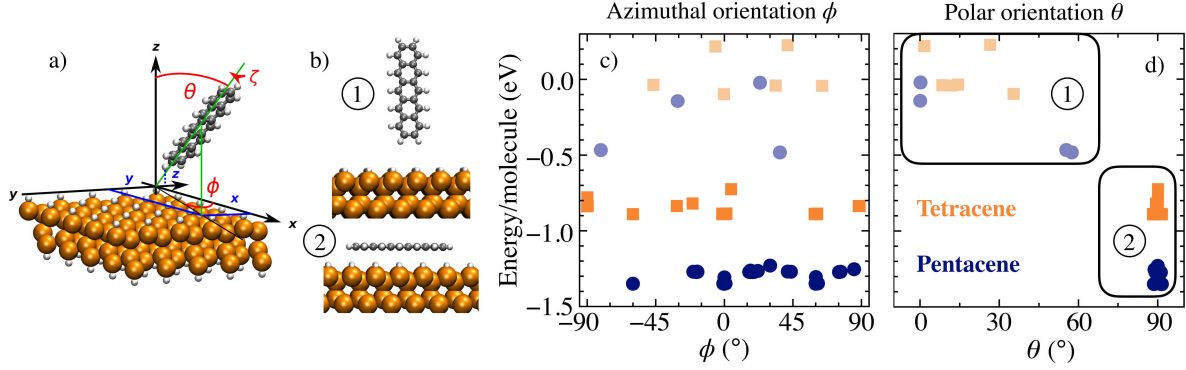


Figure 3. a) Degrees of freedom of a rigid Tc molecule on H/Si(111); the center of mass is at x , y and z . The molecule's orientation is given by the azimuth angle ϕ of the molecule's long axis with the x -axis ($[1\bar{1}0]$ -direction), the polar angle θ of the molecule's long axis with the z -axis and the rotation of the molecule around its long axis by the angle ζ . b) Example of a molecule in a standing (1) and in a lying (2) geometry. After geometry relaxation with DFT-PBE+TS, the calculated energy per molecule (Eq. 1) for different geometries of Pc (blue circles) and Tc (orange squares) is given as a function of c) the azimuth angle and d) the polar angle. Lighter shades indicate a standing orientation (1), darker shades a lying orientation (2).

The Dilute Limit

For the dilute limit of acene adsorption, we placed single Tc molecules in (4×4) supercells of the H/Si(111) substrate, whereas Pc molecules were placed in (5×5) supercells. As shown in Figure S2 in the supplementary material, the size of the supercells is sufficient to isolate the molecules from their periodic images.

As illustrated in Figure 3a, adsorption geometries can be characterized by the molecular orientation with respect to the surface and by the molecule's lateral placement. Different local minima of the potential energy surface (lying vs. standing geometries, see Figure 3b) were determined from a total of 145 initial geometry starting points each for Pc and Tc. 45 starting geometries were selected according to a grid of $\theta = 0^\circ, 45^\circ$ and 90° with $\phi = 30^\circ, 60^\circ$ and 90° . For $\theta = 90^\circ$, rotation around the molecule's long axis $\zeta = 0^\circ, 45^\circ$ and 90° were sampled. The molecules' centers of mass were placed at arbitrary x - and y -positions at 2 \AA above the plane of the H atoms. The remaining 100 starting geometries were selected by orienting the molecules randomly in the ranges $\theta = 0^\circ - 90^\circ$, $\phi = -30^\circ - 90^\circ$ and placing the molecules' centers of mass at randomly chosen x - and y -positions within the supercell, at vertical positions between $z = 1.6 - 4.0 \text{ \AA}$.

Structure optimization was initially carried out using FHI-aims' "light" settings and the DFT-PBE+many-body dispersion (MBD)[75] approach on two double layered H-terminated Si slabs. This initial set of pre-optimized geometries was next refined using the PBE+TS scheme for consistency with other simulations in this work. We chose a subset of 50 configurations for refined relaxations, consisting of the thirty lowest-energy configurations plus twenty picked from the rest of the initial pool. The latter

also included configurations where the acene molecule was found to be standing on the substrate. For these configurations, the slab thicknesses were increased to ten double-layers. During post-relaxation using FHI-aims’ “intermediate” settings and DFT-PBE+TS, the slabs were separated by 65 Å of vacuum and the top four double layers were allowed to move. After post-relaxation, all residual forces were below 0.005 eV/Å. DFT-HSE06 follow-up calculations of electronic total and projected densities of states were carried out using “intermediate” settings. The k-point meshes employed for relaxation and electronic structure investigations of the different structures are detailed in Table S4 in the supplementary material.

The Monolayer Limit

We pursued two different approaches to obtain suitable low-strain models for Pc/Tc monolayer films on the H/Si(111) substrate:

- Based on experimentally determined[29] film lattice parameters and the periodicity at room temperature (see Figure 2), we built a computational model for Pc monolayer films on H/Si(111), called “Model Φ ”.
- As an alternative approach, we used a protocol to obtain combined film-on-substrate supercells, independent of whether experimental lattice parameters and periodicity are known. In this protocol, geometries for the “monolayer limit” were obtained by fitting freestanding monolayer Tc and Pc films onto the H/Si(111) substrate. Details of this construction methodology are reported in the results section below.

For the second construction strategy, initial two-dimensional lattice parameters for free-standing monolayer film models (no substrate) were found by fully relaxing their lateral unit cells using DFT-PBE+TS, $(10 \times 10 \times 1)$ k-point meshes, and FHI-aims’ “intermediate” settings. Adjacent monolayers were separated by at least 75 Å of vacuum. The free-standing monolayer geometries consist of “standing” molecules arranged side by side in a herringbone pattern. For Pc, this herringbone structure is modeled after the experimentally known geometries[29] of Pc films on H/Si(111) at room temperature. For Tc, they are modeled from a separately constructed unit cell with two Tc molecules arranged in a herringbone pattern (see Figure 4 as well as Figure S3 and Table S5 in the SI). The final relaxed DFT-PBE+TS lattice parameters of Pc and Tc films are almost identical but slightly shorter than the experimentally reported lattice parameters for Pc monolayers on weakly interacting surfaces (see discussion in the results section).

Commensurate models of combined film/substrate supercells are described as follows. For a weak film/substrate interaction, the energetic disadvantage of straining the film (needed to form an exact coincidence lattice) will outweigh the advantage of adsorbing molecules at energetically preferred adsorption sites on the substrate[46]. Thus, we consider film-substrate geometries based on adsorbed films that are only minimally strained compared to a free-standing model. In a mathematical representation, the lattice vectors of the combined film/substrate supercell used in simulations are multiples of the substrate lattice vectors \mathbf{b}_1 and \mathbf{b}_2 (the lattice vectors

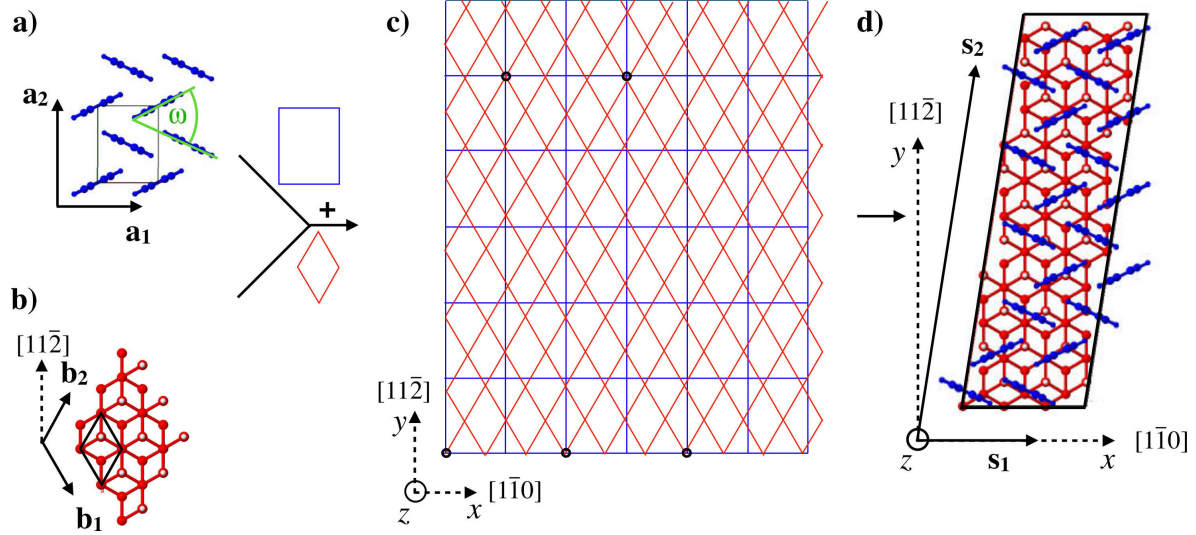


Figure 4. a) The in-(001)-plane unit cell of the acene film (blue) with the lattice parameters \mathbf{a}_1 and \mathbf{a}_2 and the herringbone angle ω between the molecules. b) The H/Si(111) surface unit cell with lattice parameters \mathbf{b}_1 and \mathbf{b}_2 as well as the $[11\bar{2}]$ direction. Si atoms are depicted in red, H atoms in white. c) Extended unit cell representations of film and substrate, superimposed to find approximate points of coincidence (black). d) Example coincidence supercell constructed from points of approximate coincidence (model C). The x , y and z -axes of the coordinate system as well as two surface directions of the H/Si(111) substrate and the lattice parameters of the combined system \mathbf{s}_1 and \mathbf{s}_2 are shown.

defining the grid of red crosses in Figures 2a and b). The supercell lattice vectors \mathbf{s}_1 with their individual components (s_{11}, s_{12}) and \mathbf{s}_2 with (s_{21}, s_{22}) can be expressed in matrix notation as follows[46]:

$$\begin{pmatrix} \mathbf{s}_1 \\ \mathbf{s}_2 \end{pmatrix} = \mathbf{C} \cdot \begin{pmatrix} \mathbf{b}_1 \\ \mathbf{b}_2 \end{pmatrix} = \begin{pmatrix} c_{11} & c_{12} \\ c_{21} & c_{22} \end{pmatrix} \cdot \begin{pmatrix} \mathbf{b}_1 \\ \mathbf{b}_2 \end{pmatrix} \quad (2)$$

The coefficients c_{11} , c_{12} , c_{21} and c_{22} are integers. Similarly, for primitive film lattice vectors \mathbf{a}_1 and \mathbf{a}_2 (the lattice vectors defining the grid of blue crosses in Figures 2a and b), we can define superlattice vectors \mathbf{f}_1 and \mathbf{f}_2 of the film, using different sets of integer coefficients \tilde{c}_{11} , \tilde{c}_{12} , \tilde{c}_{21} and \tilde{c}_{22} :

$$\begin{pmatrix} \mathbf{f}_1 \\ \mathbf{f}_2 \end{pmatrix} = \tilde{\mathbf{C}} \cdot \begin{pmatrix} \mathbf{a}_1 \\ \mathbf{a}_2 \end{pmatrix} = \begin{pmatrix} \tilde{c}_{11} & \tilde{c}_{12} \\ \tilde{c}_{21} & \tilde{c}_{22} \end{pmatrix} \cdot \begin{pmatrix} \mathbf{a}_1 \\ \mathbf{a}_2 \end{pmatrix} \quad (3)$$

A supercell of the low-strained film will have unit vectors \mathbf{f}_1 and \mathbf{f}_2 that are close to two substrate superlattice vectors \mathbf{s}_1 and \mathbf{s}_2 . Accordingly, a strain transformation \mathbf{T} of \mathbf{f}_1 and \mathbf{f}_2 can be introduced so that the resulting strained film supercell matches the substrate supercell exactly. Defining \mathbf{s}_1 , \mathbf{s}_2 and \mathbf{f}_1 , \mathbf{f}_2 as rows of two matrices \mathbf{S} and \mathbf{F} , respectively, we can write:

$$\mathbf{S} = \mathbf{T}\mathbf{F}. \quad (4)$$

Here, the film supercell vectors \mathbf{F} can correspond to rotated or unrotated versions of the overall film with respect to the substrate. In either case, for low-strain approximant supercells, the matrix \mathbf{T} should be as close as possible to the identity matrix. The determinant $|\mathbf{T}|$ is a measure of the area strain on the film. A value larger than unity corresponds to a stretched film, and a value lower than unity corresponds to compressive strain.

For atomic position optimization within large supercell models combining Pc or Tc monolayers with the H/Si(111) substrate, we employed the k-point meshes as shown in Table S4 in the SI. Of the H/Si(111) slabs containing six double layers of Si, the top four Si double layers, the surface H atoms and the molecular adsorbates were allowed to relax using DFT-PBE+TS and FHI-aims’ “intermediate” settings until the residual forces on all optimized atoms were smaller than 0.005 eV/Å. The resulting geometries were then used to calculate the electronic structure (total and projected densities of states) of Tc and Pc monolayers on H/Si(111) using DFT-HSE06, FHI-aims’ “intermediate” settings and the k-space meshes detailed in Table S4 in the SI.

3. Results

3.1. Dilute Limit

In Figure 3c and Figure 3d, we show the adsorption energies ΔE (Eq. 1) of individual Tc and Pc molecules on H/Si(111) as a function of the azimuthal angle ϕ of the molecule’s long axis with the x -axis (Figure 3c) and θ of the molecule’s long axis with the surface normal (Figure 3d). Each data point corresponds to a specific, fully relaxed geometry obtained from a different starting geometry. Neither Pc nor Tc show a distinct preference for a particular azimuthal orientation (Figure 3c). In contrast, both acene molecules prefer a lying orientation (i.e. $\theta \approx 90^\circ$, $\zeta = 90^\circ$ in Figure 3d and conformation (2) in Figure 3b) over a standing orientation (i.e. $\theta < 60^\circ$ in Figure 3d and conformation (1) in Figure 3b). This agrees with previous experimental[28] and computational[90, 91, 92] observations of Pc on weakly interacting surfaces and with the general observation that a lying, rod-like, aromatic molecule should have a stronger interaction with a substrate than a standing one[27]. The minimum energy per molecule is $\Delta E = -0.890$ eV for lying Tc and $\Delta E = -1.350$ eV for lying Pc. Among the cases we investigated, only four of the Pc and six of the Tc cases assume a standing orientation. The minimum energies per molecule found for standing Tc (-0.097 eV) and for standing Pc (-0.483 eV) are much less favorable than for the lying case. Two standing Tc molecules display $\Delta E > 0.0$ eV, indicating that, if the molecules are initially placed too far from the substrate ($d_z > 11.3$ Å), the interaction between molecules and substrate may not be sufficiently large to relax into local minima based on the minimum force criteria chosen here.

In Figure 5, we visualize the DFT-HSE06 predicted electronic densities of states of the energetically most favorable “dilute” Tc and Pc adsorption geometries (for both

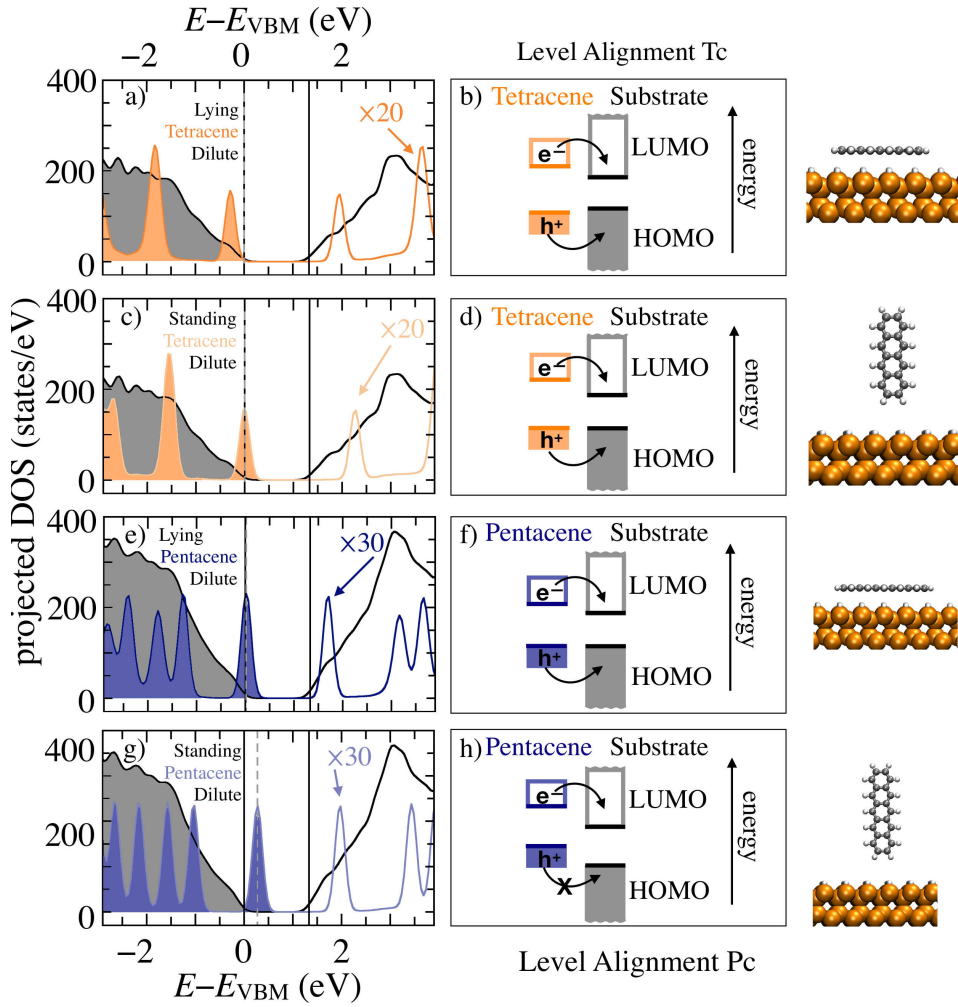


Figure 5. DFT-HSE06 projected densities of state (PDOS) for single Tc and Pc molecules adsorbed on H/Si(111) aligned to the substrate's VBM, decomposed into PDOS for the H/Si(111) substrate atoms (black curves) and the adsorbed Tc (orange curves) and Pc (blue curves) molecules. Filled areas indicated occupied levels, empty areas unoccupied levels. Vertical solid lines indicate the positions of inorganic HOMO and LUMO, the dashed grey line marks the position of the electron chemical potential. Due to the artificial Gaussian broadening (0.1 eV) applied to compute the PDOS, the peaks of the overall HOMO appear broadened beyond the actual position of the chemical potential (the structures shown are in fact insulating, not metallic). a) PDOS for the lying Tc case. b) Schematic energy level diagram of the HOMOs and LUMOs of the lying Tc case, c) PDOS for the standing Tc case. d) Schematic energy level diagram of the standing Tc case, e) PDOS for the lying Pc case. f) Schematic energy level diagram of the lying Pc case, g) PDOS for the standing Pc case, h) Schematic energy level diagram of the standing Pc case. The orientation of the organic molecule on the substrate is illustrated schematically on the right. Because the molecule contains much fewer atoms than the underlying substrate, the DOS projected onto the acene films were scaled by a factor of 20 (Tc) and 30 (Pc) to make them comparable to the DOS projected onto the substrate atoms. Note that the supercells chosen for Pc (5×5) contain significantly more Si atoms per molecule than those chosen for Tc (4×4).

lying and standing cases), projected onto the isolated molecules and onto the substrate, as well as schematic depictions of the frontier level alignment for each case. For lying Tc (Figure 5a and b), the organic HOMO and LUMO fall into the substrate’s valence and conduction band (type I alignment). This means that in principle, charge transfer from the molecule to the substrate should be possible for both hole- and electron-like carriers. For standing Tc (Figure 5c and d), the molecular HOMO and substrate VBM are practically degenerate. Lying Pc (Figure 5e and f) also shows a degenerate molecular HOMO and substrate VBM, whereas for standing Pc (Figure 5g and h) the molecular HOMO would be in the substrate gap (type II alignment). A transfer of hole carriers from the molecule to the substrate would be hindered.

From Figure 5 we observe that, from the lying to the standing orientation, the molecular frontier levels shift almost uniformly upwards by ~ 0.25 eV compared to the substrate bands. The shift is due to different electrostatic interactions between the standing or lying molecule and the surface[23]. In the lying case, the molecule’s π -system interacts more strongly with the underlying substrate. The HOMO and LUMO are hence shifted downward in energy compared to the standing case. Note that varying the hybrid functional[93] or considering dynamical screening effects[94, 95, 59] might alter the energy levels further. Nevertheless, the results indicate that hole transfer from isolated molecules to the substrate should be more difficult for a standing molecule than for a lying one.

3.2. Dense Monolayer Limit

For the monolayer films, we first consider theoretical models based on the experimentally observed coincidence pattern for Pc on H/Si(111)[29]. In a second step, we test a protocol to obtain approximate combined film-on-substrate supercells, independent of whether experimental lattice parameters and periodicity are known.

3.2.1. Interface Model Based on Experimental Lattice Coincidence (“Model Φ ”)

Nishikata *et al.*[29] identified point-on-line coincidence of superlattices of Pc-dendrites with the H/Si(111) substrate lattice at room temperature from low-energy electron microscopy (LEEM) and low-energy electron diffraction (LEED). Pc was observed to grow in two orientations on the H/Si(111) substrate with slightly different unit cells, labelled “T” and “IP” in Figure 2 and Table 1. Nishikata *et al.* characterized supercells based on the film lattice parameters and their orientation. For supercells “T”, a periodicity of (36.0 ± 0.8) Å (six unit vectors of the Pc film) was identified in the \mathbf{a}_1 -direction. For supercells “IP”, the identified periodicity is (47.84 ± 0.01) Å (eight unit vectors of the Pc film) in the \mathbf{a}_1 -direction. Both supercells show a periodicity of one unit vector of the Pc film in the \mathbf{a}_2 -direction[29]. Using Eqs. 2, 3 and 4, we can evaluate the experimentally reported supercells. The results are tabulated in Figure 2c. For supercell “T”, we find $|\mathbf{T}| = 1.021$. This amounts to a stretch of the film on the substrate with a resulting area strain on the film of 2 %. For supercell “IP”,

Table 1. Experimental lattice parameters \mathbf{a}_1 and \mathbf{a}_2 for acene films on weakly interacting surfaces from selected literature sources, compared to free-standing monolayer film models in vacuum (i.e., without including a substrate) after full relaxation by DFT-PBE+TS in this work. γ is the angle between the lateral lattice vectors. RT is room temperature, GIXRD is grazing-incidence x-ray diffraction, XRD is x-ray diffraction, LEED is low-energy electron diffraction, RHEED is reflection high energy electron diffraction, and STM is scanning tunneling microscopy.

Molecule	Coverage	\mathbf{a}_1 (Å)	\mathbf{a}_2 (Å)	$\gamma(^{\circ})$	Substrate	Method	Temperature	Ref.
Tc	1 ML	5.85	7.39	90	free-standing			this work
Tc	3 ML	5.5 ± 0.6	7.3 ± 0.6		H/Si(001)	STM	RT	[31]
Tc	1.5-12 ML	5.93 ± 0.02	7.70 ± 0.02	90.0 ± 0.2	H/Si(001)	GIXRD	RT	[36]
Pc	1 ML	5.85	7.36	90	free-standing			this work
Pc	1 ML	6.32 ± 0.06	7.73 ± 0.07	84 ± 1	H/Si(111)	RHEED	65°C	[28]
Pc	< 1 ML	6.02 ± 0.01	7.62 ± 0.02	90.0 ± 0.4	H/Si(111)	LEED	RT	unit cell I[29]
Pc	< 1 ML	5.98 ± 0.01	7.56 ± 0.01	90.3 ± 0.1	H/Si(111)	LEED	RT	unit cell II[29]
Pc	1 ML	6.1 ± 0.2	7.89	86 ± 0.5	Bi(001)	LEED	RT	[44]
Pc	1 ML	5.916	7.588	89.95	SiO ₂	GIXRD	RT	[34]
Pc	< 1 ML	5.90	7.62	90 ± 0.2	SiO ₂	GIXRD	0, 22, 45°C	[43]
Pc	0.6-100 nm	5.94	7.54	89.5	SiO ₂	GIXRD	20°C, 50°C	[96]
Pc	2 ML	5.90 ± 0.01	7.51 ± 0.01	89.92 ± 0.01	SAMs	GIXRD	RT	[35]
Pc	< 30 nm	5.9	7.5	90	KCl(001)	XRD	RT	[97]

$|\mathbf{T}| = 0.883$. A 12 % area compression is large for a weakly interacting system. We conclude that supercell “II” cannot easily be modeled as commensurate structure with the underlying H/Si(111) substrate, unless a much larger commensurate supercell in the \mathbf{a}_2 direction is considered. Additionally, Nishikata *et al.*[29] reported that supercell “T” is more common than supercell “II”. We therefore focus our comparison on supercell “T” with lattice vectors based on the experimentally suggested periodicity (see Figure 2c).

We first investigate the experimentally observed coincidence pattern for Pc on H/Si(111) for supercell “T”. Based on the film lattice parameters determined by Nishikata *et al.*[29] and the DFT-PBE+TS silicon lattice parameter of $a_0 = 5.450$ Å (to ensure a strain-free H/Si(111) substrate in the computations) the area strain of the computational model is characterized by $|\mathbf{T}|=1.028$. This combination is referred to as “Model Φ ” in Figure 2c. After geometry relaxation with DFT-PBE+TS, the calculated adsorption energy per Pc molecule is $\Delta E = -2.222$ eV. As expected, the film is energetically more favorable than the adsorption of an isolated, lying Pc molecule on the substrate due to more favorable molecule-molecule interactions within the film (see also Table S5 in the supplementary material). The molecules form a herringbone pattern with a computed “edge-to-face” angle of $\omega = 49^\circ$ between the planes of neighboring molecules, and with a tilt of $\theta = 22^\circ$ to the surface normal (see Figure 7a).

Regarding experimental values for ω and θ in monolayer films, we are not aware of an “apples-to-apples” comparison for a Pc monolayer on H/Si(111). However, a value of $\omega = 52.7^\circ$ was reported in a grazing-incidence X-ray diffraction (GIXRD) study of Pc monolayer films on amorphous silicon oxide[98]. Several different phases of Tc and Pc are known to be stable at room temperature, with reported herringbone angles both computationally and experimentally in a broadly similar range[34, 42, 38, 40, 98, 99, 100]. The same GIXRD study[98] as well as a GIXRD study of two layers of Pc on self-assembling membranes[35] place θ at $\sim 0-4^\circ$. This is similar to the tilt angle proposed for the so-called thin-film phase of Pc[42, 40, 101] whose lattice parameters are similar to those of the monolayer[42]. In contrast, our value of $\theta = 22^\circ$ is closely in line with tilt angles in the range of 20° to 28° observed in bulk-like Pc polymorphs, both experimentally[42, 102, 103] and computationally[38, 104]. Refs.[98] and [105] discuss the discrepancy of the tilt angle in the monolayer in terms of reduced film lattice parameters compared to the bulk phase and, therefore, a higher in-plane molecular density with corresponding upright molecules. Another mechanism that could lead to more upright molecules in experiment than those in our fully relaxed geometries is thermal motion. However, while past experimental studies of Pc polymorphs show some change of the tilt angle with temperature, the magnitude of the effect is in the range of a few degrees[105] and does not support a thermally driven transition from $\theta = 22^\circ$ to fully upright molecules. Instead, we show below that computational film models using the DFT-PBE+TS density functional are denser and that nearly upright molecular geometries in the films, in line with GIXRD, would result for these denser computational film models as well.

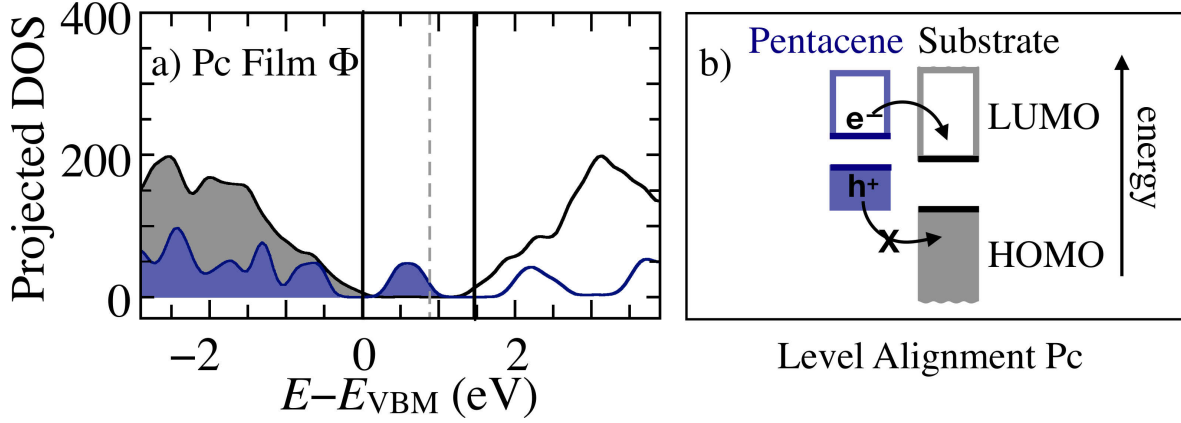


Figure 6. a) DFT-HSE06 projected densities of states for the combined film-on-substrate model Φ based on the supercell “T” experimental lattice parameters and periodicity[29]. Separate contributions are shown for the H/Si(111) substrate atoms (black) and the monolayer Pc molecules (blue). The solid black lines mark the positions of the substrate slab’s HOMO (chosen as 0 eV) and LUMO and the dashed line that of the chemical potential. b) Schematic energy level diagram of the HOMO and LUMO of the H/Si(111) substrate (black) as well as the monolayer Pc film (blue) for the combined film-on-substrate model Φ . In both panels, filled areas indicate occupied levels, empty areas unoccupied levels.

Electronic Properties of Model Φ Figure 6a shows the DFT-HSE06 densities of states projected onto the substrate and Pc monolayer for the combined film-on-substrate model Φ . A schematic energy level diagram is shown in Figure 6b. The LUMO of the adsorbed Pc monolayer is found within the substrate’s conduction band. In contrast, the HOMO of the adsorbed Pc film is within the substrate’s band gap. This type II level alignment is similar to Pc/Si alignment suggested based on separately measured ionization potentials and electron affinities[22]. Based on this alignment, hole transfer from the film to the substrate would be hindered (holes would transfer from substrate to film instead), while electrons could still transfer into the substrate.

Regarding the absolute values of the predicted gaps, the bulk band gap of Si (~ 1.17 eV at $T=0$ K[45]) is well reproduced with the present flavor of DFT-HSE06 (calculated bulk band gap for “tight” settings, $\alpha = 25\%$, $\omega = 0.11$ (Bohr radii) $^{-1}$: 1.165 eV). As discussed above and shown in Figure S2 in the supporting material, the band gap of a bare H/Si(111) slab of six Si double layers is still about 0.3 eV greater than the bulk band gap. The comparison to the vacuum level in Figure S2 shows that the remaining change of the gap with increasing slab thickness is mainly due to the CBM, i.e., the qualitative band offsets between film and substrate reported based on six Si double layer slabs will likely remain unchanged since the substrate CBM would move down for a thicker slab. Importantly, in Figure 6a the difference between substrate CBM (upper black solid line) and molecular HOMO (dashed grey line) is still significantly larger than 0.3 eV. Thus, the remaining overestimation of the substrate band gap due to the finite thickness of the slab is not expected to change the observed type-II level

alignment concluded here for model Φ .

For the pentacene film, the molecular orbitals are broadened compared to the dilute limit (Figure 5). They are shifted up in energy compared to the isolated standing orientation, and the film’s HOMO-LUMO gap is reduced to 1.05 eV. Qualitatively, both the broadening and the upwards shift are consistent with the increased confinement of the π -systems of the molecules in the more closely packed arrangement of the monolayer film. In comparison, the G_0W_0 band gap (using DFT in the local-density approximation as the starting point for G_0W_0) of a Pc solid is found to be 2.1 eV in Ref. [106], i.e., significantly larger than the DFT-HSE06 film gap predicted in the present work. While part of this difference may be attributable to structural differences in the film, the bulk of the discrepancy most likely stems from the different dielectric properties and environment of the Pc film compared to the Si bulk. For the Pc film, a different and higher α parameter in the HSE06 functional than for Si would be appropriate[93].

The difference between the predicted and the likely actual band gap of the Pc film is important when comparing qualitatively to the experimentally known excitonic properties of Pc. Experimentally, the emission energy of the first triplet state in Pc was measured to be 0.81–0.90 eV[107, 7, 108, 109], a little less than half of the energy of the first singlet state[14] of 1.83 eV[110]. The G_0W_0 gap for Pc[106] is correctly expected to be higher than the experimental singlet energy, whereas the DFT-HSE06 predicted gap in our film model is too low. However, given the computed considerable shift of the Pc HOMO into the substrate gap and given the expected downward shift of the actual substrate CBM compared to the slab model used in Figure 6, it is entirely conceivable that the experimentally expected triplet energy is still sufficient to allow for charge separation with holes remaining on the Pc film and electrons injected into the substrate.

3.2.2. Computational Procedure for Finding Lattice Coincidence Patterns We now turn to the determination of geometry and electronic level alignment between a film and a substrate by a purely computational approach, i.e., without relying on experimental input regarding the coincidence pattern. For Pc on H/Si(111), we can compare predictions to the results obtained above for “Model Φ ”. For Tc on H/Si(111), the experimental coincidence pattern is unknown and the results presented below are thus our best available predictions for the level alignment in this system.

As described in Section 2, we first relax free-standing monolayer films (i.e., *in vacuo*) of acene molecules arranged in a herringbone pattern to obtain the lattice parameters that such films would assume without any interaction with the substrate. Table 1 includes the predicted lateral lattice parameters and unit cell angles for the Tc and Pc model films in comparison to experimental unit cell parameters of Tc and Pc submonolayer films[29, 43], monolayers[28, 44, 34] or thin films[31, 36, 35, 97, 96] observed on different weakly interacting substrates. The experimentally reported unit cell angles γ for Pc[29] and Tc[31, 36] at room temperature on hydrogenated silicon are in good agreement with our simulated free-standing films. For the experimentally reported

lattice parameters, there is noticeable scatter for Tc and Pc films on different substrates. Our theoretically determined Pc and Tc lattice parameters are within < 0.5 % of one another, i.e., essentially identical. Compared to the Pc monolayer films on H/Si(111)[29] and to Tc thin films on H/Si(100)[36, 31], the experimentally observed \mathbf{a}_2 parameters are larger by about 4 %. The fully DFT-PBE+TS relaxed films are thus likely slightly denser than actual experimental Tc and Pc monolayer films.

Figure 4 illustrates the process of determining approximate coincidence lattices between the substrate and film superlattices for the case where the film \mathbf{a}_1 lattice vector is aligned with the $[1\bar{1}0]$ surface direction. The unit cells are shown for the acene film in Figure 4a and for the substrate unit cell in Figure 4b. Figure 4c shows the coincidence pattern between both lattices. Approximate points of close coincidence between lattice vectors of the acene and the H/Si(111) lattices are identified by black circles. A particular resulting combined cell, model C described below, is shown in Figure 4d.

We followed two different routes to construct combined film-on-substrate supercells. In the first route, we identified approximate coincidence points between substrate and film superlattices visually, by overlaying substrate and supercell lattice points. In the second route, described in more detail below, we obtained the points by a script written for this purpose. The script compares the agreement between possible multiples of unit cells for films and substrate. Because of the very similar lattice parameters predicted for Pc and Tc free-standing films, the same set of resulting commensurate supercells is used for Tc and for Pc in our simulations. In the following, we will discuss the lattice coincidence at the example of Pc. Similar conclusions apply to Tc. Figure 7 shows unrotated and rotated commensurate approximant supercells between the substrate and the films, labelled as Models A-E, that were selected for further study in this work. Models A-C in Figure 7a illustrate different models from the first (visual) strategy, whereas models D and E in Figure 7b and c) were derived from the script-based strategy. The table in Figure 7d) summarizes their 2D lattice vectors and area strains as defined in Eq. 2 and Eq. 4. Finally, Figure 7 lists their unit cell areas, number of atoms included in the full (film plus substrate) structure model, and DFT-PBE+TS calculated film adsorption energies per molecule for both Pc and Tc.

Model A, shown as the red area in Figure 7a, is the smallest possible commensurate unit cell that corresponds to reasonably close coincidence points of the lattices. While the coincidence in the x direction is close, in the y direction the molecular supercell is more extended than the corresponding underlying substrate supercell. In Figure 7d, both \mathbf{T} and $|\mathbf{T}|$ reveal quantitatively that the film-substrate mismatch in the small-cell commensurate approximant model A corresponds to a compression of the film of ~ 10 % in y direction. This is a significant compression that, as we show below, leads to a noticeable change of the electronic structure of the combined film-substrate model compared to larger coincidence cells. Lower strain is achieved for the larger commensurate unit cells B and C, shown as the blue and grey areas in Figure 7a. The values of \mathbf{T} and $|\mathbf{T}|$ associated with them are much closer to unity. However,

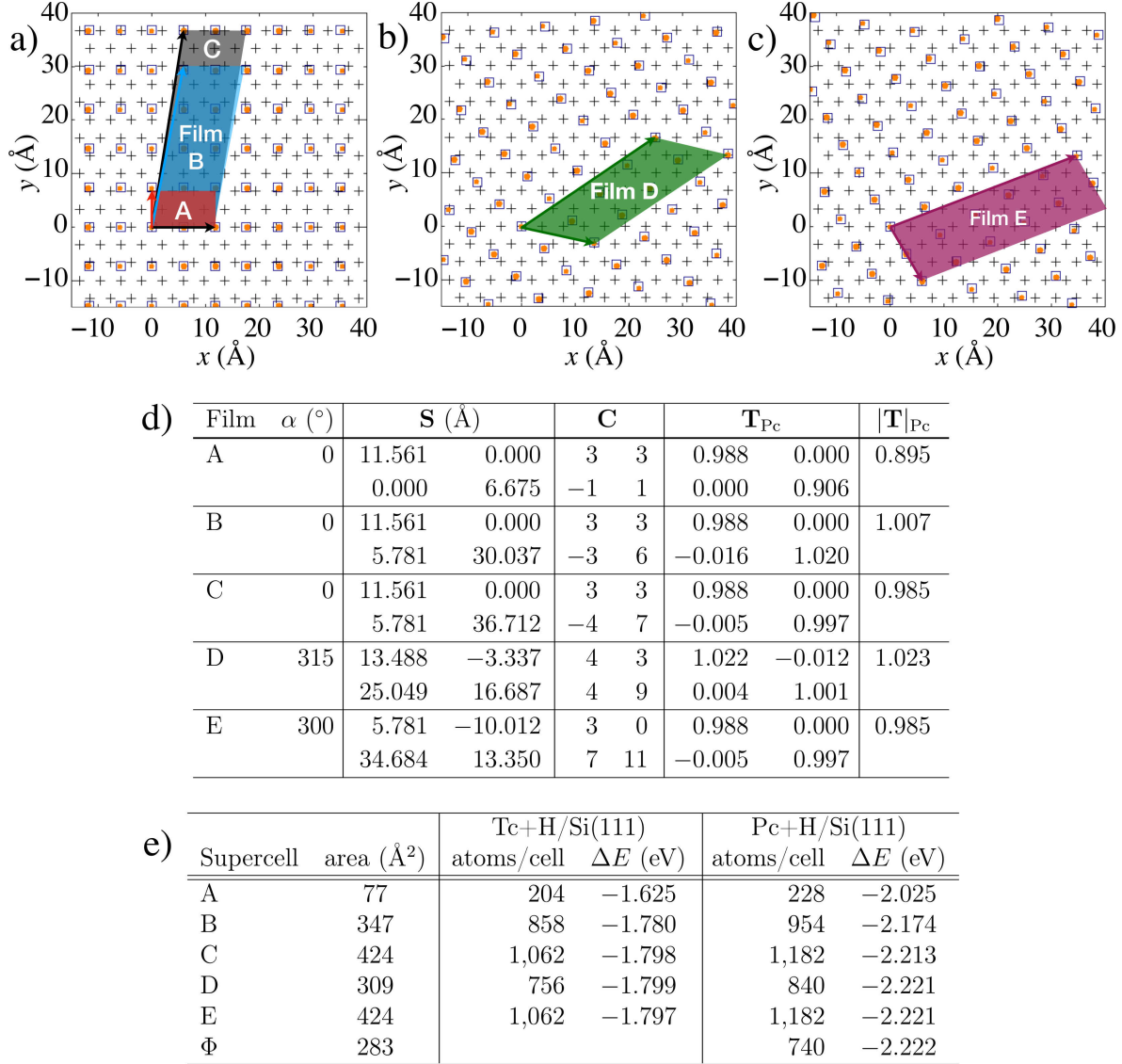


Figure 7. The coincidence pattern between the H/Si(111) substrate (black crosses), the tetracene film (orange circles) and the pentacene film (blue squares). Panels a–c) show different rotations α of the acene film with respect to the $[1\bar{1}0]$ -direction of H/Si(111). a) $\alpha = 0^\circ$ for cells A (red), C (blue) and D (black) b) $\alpha = 315^\circ$ for unit cell B and c) $\alpha = 300^\circ$ for unit cell E. Although cell C (a, black) and E (c, purple) have the same area, they are not symmetry equivalent: H/Si(111) exhibits three-fold symmetry due to the ABC stacking of the Si double layers. d) The matrix of the in-(001)-plane lattice vectors of the combined film-substrate supercell \mathbf{S} is a multiple (\mathbf{C}) of the substrate lattice vectors \mathbf{b}_1 and \mathbf{b}_2 . The strain transformation of the supercell parameters of the film \mathbf{F} when the film is combined with the substrate is expressed in terms of the matrix \mathbf{T} . $|\mathbf{T}|$ describes the agreement between the area of the substrate and monolayer film supercell. e) The combined (film plus substrate) unit cell areas, the number of atoms in the computational structure models, and the film adsorption energy per molecule calculated with DFT-PBE+TS for a Pc or Tc monolayer on H/Si(111).

these models necessitate much larger overall numbers of atoms than A, and are thus a challenge for electronic structure predictions beyond the level of hybrid functional DFT.

To obtain models D and E (Figure 7b and c), a script was used that explores the search space of points of coincidence systematically in two steps.

- The first step tests the difference in area of possible substrate and film supercells \mathbf{S} and \mathbf{F} by evaluating $|\mathbf{T}|$. \mathbf{S} was sampled from $\mathbf{C} = \begin{pmatrix} -10 & -10 \\ -10 & -10 \end{pmatrix}$ to $\mathbf{C} = \begin{pmatrix} 10 & 10 \\ 10 & 10 \end{pmatrix}$, covering 190,612 supercells and a maximum unit cell area of 1286 \AA^2 . Because the lattice vectors \mathbf{a}_1 and \mathbf{a}_2 are almost twice as long as \mathbf{b}_1 and \mathbf{b}_2 , we sampled corresponding film supercells \mathbf{F} from $\tilde{\mathbf{C}} = \begin{pmatrix} -5 & -5 \\ -5 & -5 \end{pmatrix}$ to $\tilde{\mathbf{C}} = \begin{pmatrix} 5 & 5 \\ 5 & 5 \end{pmatrix}$, covering 13,808 supercells and a maximum unit cell area of 1077 \AA^2 .
- The second step rotates the film's \mathbf{a}_1 axis compared to the substrate's \mathbf{b}_1 axis by an angle α . If, in the first step, $|1 - |\mathbf{T}|| < 1.1$, i.e., if the areas of \mathbf{S} and \mathbf{F} agree within 10 %, \mathbf{F} is rotated by α in steps of 1° with respect to \mathbf{S} , resulting in a new transformation matrix $\mathbf{T}^{(\alpha)}$. After the second step, a total of 159,890,112 possible supercells were compared for Pc.

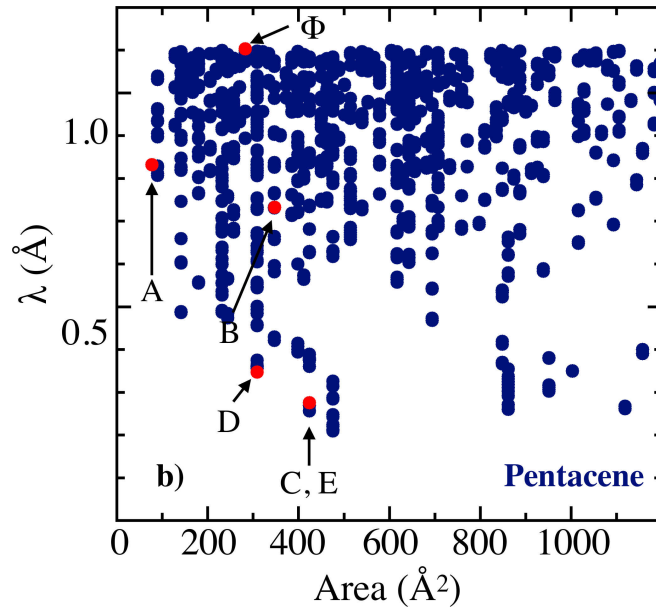


Figure 8. Least-squares difference λ between the lattice vectors of the H/Si(111) substrate and Pc (blue circles) as a function of the H/Si(111) supercell area for different commensurate coincidence lattice models determined by a script as described in the text. Supercell models A-E, as well as the supercell model Φ determined according to the experimental study of Nishikata *et al.*[29] for Pc, are marked in red and indicated by arrows.

While $|\mathbf{T}|$ provides a good measure of the agreement of the area of the film and substrate supercell, this does not mean that the supercell lattice parameters of film and substrate agree well or match in shape. To find the best match between the supercell lattice parameters for a rotation α , the agreement between the actual 2D lattice parameters of the film and substrate supercell for a rotation α can be quantified by the combination of substrate and film supercell lattice parameters that show the

smallest disagreement:

$$\lambda = \sqrt{\sum_n \min_{i=1,2} \left(\sum_k \left(f_{n,k}^{(\alpha)} - s_{i,k} \right)^2 \right)} \quad (5)$$

Figure 8 categorizes different film-substrate models for Pc in terms of their coincidence supercell area (x -axis) as a measure of computational cost, and in terms of their lattice parameter mismatch λ (y -axis) as a measure of degree of lattice coincidence for $\lambda < 1.1$ Å. The script-optimized models D and E display the lowest or close to lowest λ values for their respective unit cell areas in Figure 8. The visually adjusted model C shows a low λ value, whereas the differently stretched model B does not agree so well. Likewise the small-cell approximant A shows a rather high λ , in addition to its considerable area strain $|\mathbf{T}|$ in Figure 7d. As a final point of comparison, the experimental supercell I for Pc[29], Model Φ in its commensurate form is also shown in Figure 8b. Due to its size and the noticeable difference in lattice parameters between the freestanding Pc film model and the supercell I deduced from experiment, the λ value of this model is considerably higher than for the specifically constructed models.

3.2.3. Geometric and Total-Energy Properties of Film Models A-E We next investigate the geometries that the molecules assume in the predicted monolayer films A-E after relaxation. A full list of all geometry parameters discussed below is provided for each film in Table S6 and Table S7 in the supplementary material, where we define the adsorption distance d_z of a molecule to the substrate as the distance between the z -component of the center of mass of the molecule atoms and the plane to the average z -position of the top layer of Si atoms (see Figure S4). The Pc molecules in models A-E assume tilt angles θ of the molecules' long axes to the substrate normal within 2° . The same is true for Tc films, except for model D where the angles are on average $\theta = 8^\circ$. Interestingly, these tilt angles value are in line with the GIXRD result for Pc monolayer films on amorphous silicon oxide[98], two layered Pc films on self-assembled membranes[35] and with the tilt angle suggested for the thin-film phase of Pc[42, 40, 101]. The average edge-to-face or herringbone angle ω for models B, C, D and E lies between 51° and 54° , again in close agreement to the experimental value for Pc on amorphous silicon oxide[98]. In contrast, the much more compressed small-cell model A shows a slightly smaller average herringbone angle of $\omega = 46^\circ$ for both Pc and Tc.

While amorphous silicon oxide is of course a different substrate than H/Si(111), the agreement observed here regarding tilt angles is consistent with the idea that the films interact only weakly with these substrates and the detailed film structure is largely determined by intermolecular interactions within the films. The small value of θ indicates that the higher-density monolayer film model reflects the experimentally suggested geometry better than Model Φ , which was built based on the experimental supercell lattice parameters for Pc[29]. While speculative, it seems possible that the PBE+TS density functional used to construct the free-standing film model is slightly

too attractive, necessitating a smaller lattice parameter than the experimental film. In this scenario, the correct molecular tilt would be obtained for the denser model. In contrast, when using the experimental lattice parameter (Model Φ), the molecules in the film would tilt too much, in order to improve their overall packing as dictated by DFT-PBE+TS, as we observe in Model Φ .

A comparison of the energy per molecule ΔE for the combined film-on-substrates models after geometry optimization is given in Figure 7e. The adsorption energies for the low-strain unit cells B-E are within 0.05 eV per molecule of each other. Film A, where the film is more compressed, is less favorable by about 0.15 eV per molecule. Model Φ that is based on experimental lattice parameters for Pc[29], where the film is stretched, shows the same (even very slightly more favorable) overall adsorption energy compared to models B-E. This indicates that the unknown, hypothetical exact film-substrate interface with minimum total energy at the Born-Oppenheimer surface (no finite- or zero- T vibrational effects) might be somewhere in between, i.e., slightly stretched compared to the free-standing monolayer film model.

Electronic Structure of Models A-E Figure 9 shows the densities of states projected onto the substrate and organic monolayer for Tc (a-e) and Pc (f-j) for the combined film-on-substrate models A-E. The same general observations hold for supercell models B-E. Similar to Model Φ (Fig. 6), the LUMOs of both acene monolayer films are within the substrate’s conduction band. Likewise, the films’ HOMO levels (dashed grey lines in Fig. 9) are in the substrate’s band gap. The HOMO-LUMO gap of the isolated standing monolayers is on average 1.75 eV for Tc models B-E and 1.12 eV for Pc models B-E, slightly higher than the 1.05 eV found for Model Φ . The level alignment is qualitatively the same type II heterojunction as found for Φ . Importantly, as noted above for model Φ , the remaining uncertainty of the substrate band gap of ≈ 0.3 eV due to the finite thickness of the Si slab is not large enough to alter this qualitative conclusion for models B-E. While the densities of states shown are artificially broadened using a Gaussian broadening of 0.1 eV to obtain smooth curves, the actual predicted differences between the film HOMO (dashed grey lines) and the substrate CBM (upper black lines) are visibly larger in these cases in Fig. 9.

In contrast, for the much more compressed film of the small-cell approximant unit cell A, more significant changes of the electronic structure are observed. First, its molecular HOMO seems broadened. As a consequence, the HOMO-LUMO bandgap for both Tc (1.10 eV for model A) and Pc (0.44 eV for model A) has closed considerably compared to the less strained models. More significantly, the overall band gap of the combined film-substrate system – which is still present for Tc for Pc in models B-E – is more than half for model A for both Tc (0.33 eV) and for Pc (0.14 eV). For Pc, this means that the band gap closes almost entirely (see also Table S8 in the SI). In view of these changes, the large-strain approximant cell A is not a safe model to predict the electronic structure of this organic-inorganic hybrid system.

In Figure 10, we visualize the frontier orbitals, i.e., the HOMO (purple) and

the LUMO (green), at the Γ -point for the example of Tc film B in Figure 9c for a superposition of three and two degenerate eigenstates (within 10 meV of one another) for the HOMO (Figure 10a) and LUMO (Figure 10b), respectively. The localization of the eigenstates does not change for the following five eigenstates. In agreement with the PDOS analysis above, we find that the nearly degenerate eigenstates that form the HOMO are delocalized on the molecular film. The eigenstates contributing to the LUMO are similarly delocalized on the more bulk-like atoms in the middle of the Si-slab, in agreement with the observation that the onset of the band gap is formed by contributions from bulk layers of the slab (see Figure S5 in the SI). The HOMO and LUMO are hence indeed spatially removed from one another, with what appears to amount to a small barrier formed by the interface itself. This supports the idea that a charge separation of electrons and holes could occur in a type II heterojunction-like manner at the interface between Tc and Pc monolayer films and H/Si(111).

In section 3.2.1, we concluded that a transfer of electrons generated by split triplet excitons to the substrate CBM seemed possible in view of our computed DOS for Model Φ . This conclusion was based on the expected CBM level for a slab of converged thickness and literature values of singlet and triplet energies in Pc. Within the uncertainties of both the HSE06 density functional and our model (restricted slab thickness), the same observations remain true for the Pc models B–E. For Tc, the experimentally measured triplet energy is ~ 1.25 eV[111, 112, 10] and that of the singlet 2.32 eV[111]. With these values, the approximate triplet level in the Tc films in our models is again slightly above the inorganic LUMO (CBM). Thus, singlet and triplet excitons in both Pc and Tc could potentially dissociate at the heterojunction and pass electrons into the inorganic substrate, while holes would remain on the film.

Finally, based on UV photoelectron spectroscopy measurements of *n*-doped, bare H/Si(111) samples and on H/Si(111) samples coated with 10 nm or more Tc, MacQueen *et al.*[20] suggest a type I level alignment with the HOMO of the Tc film, lying ≈ 150 meV below the VBM of the Si substrate. These observations are at variance with our computational observations for much thinner films, which account for the levels in a single system, i.e., including the electrostatics at the interface.[59] It seems possible that differences in film thickness and substrate doping might lead to different level alignment between our computational results and the experimental observation by MacQueen *et al.* On the other hand, our calculations are performed for the combined film-substrate system, not for separate film and substrate systems that were used to deduce alignments in the experimental study. Encouragingly, for the combined film-substrate system, MacQueen *et al.*[20] observe a hole transfer from the substrate to the Tc film that is consistent with the conclusions drawn from our model system. Finally, their suggestions of a triplet exciton level roughly degenerate with the LUMO of the H/Si(111) is also consistent with our qualitative picture.

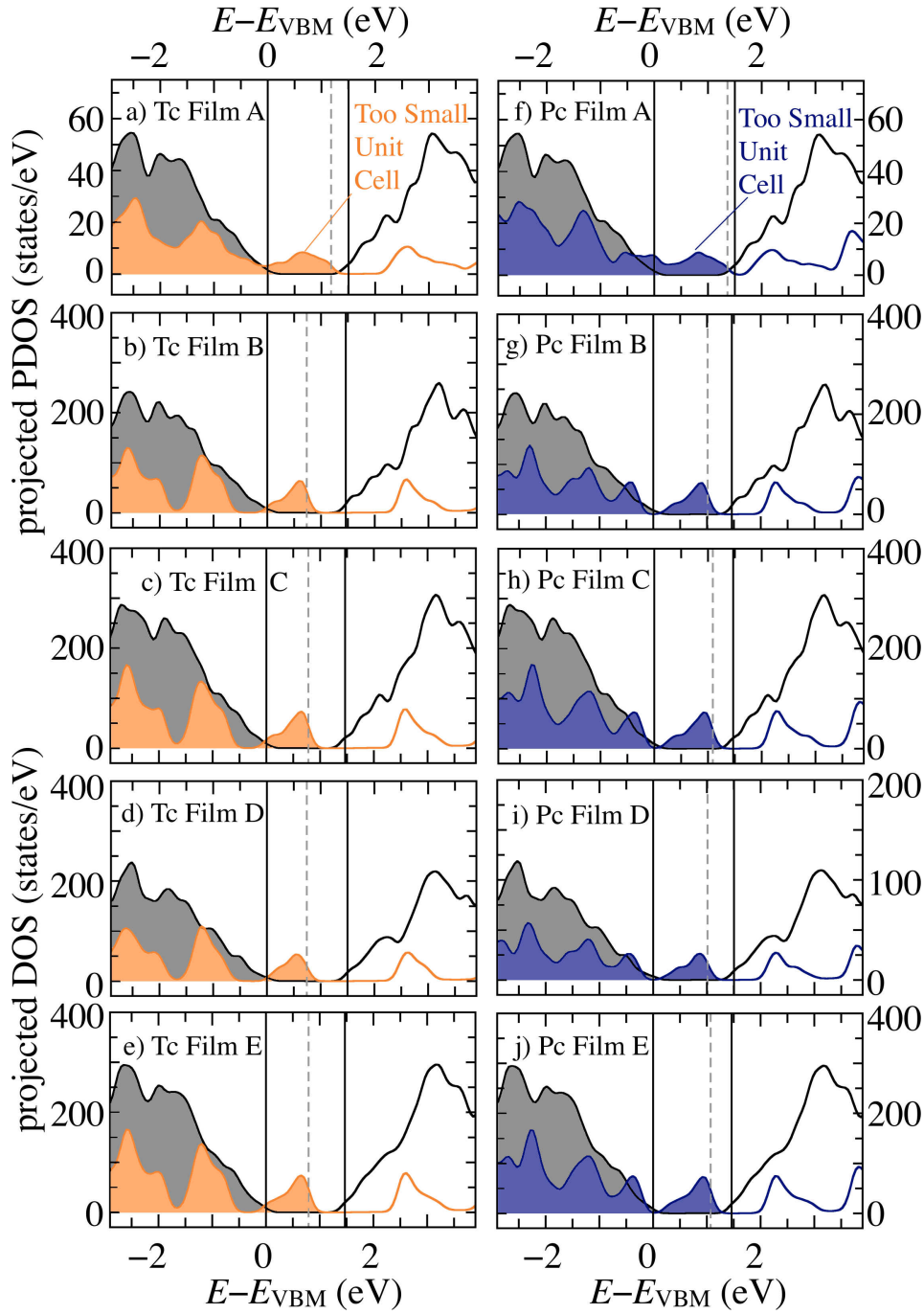


Figure 9. The DFT-HSE06 DOS projected onto the substrate atoms (black) the Tc (orange) and Pc (blue) film atoms for a) Tc film A b) Tc film B c) Tc film C, d) Tc film D, e) Tc film E, f) Pc film A, g) Pc film B, h) Pc film C, i) Pc film D, j) Pc film E. Filled areas indicate occupied levels, empty areas unoccupied levels. The black solid lines mark the position of the substrate's HOMO (set as the energy zero in each case) and LUMO. The dashed grey lines mark the position of the electron chemical potential. Due to the artificial Gaussian broadening used for the PDOS (0.1 eV), the peaks of the overall HOMO are visually broadened above the electronic chemical potential level. Actual fractional occupations in the self-consistent DFT calculations are only found in films A, not in B-E.

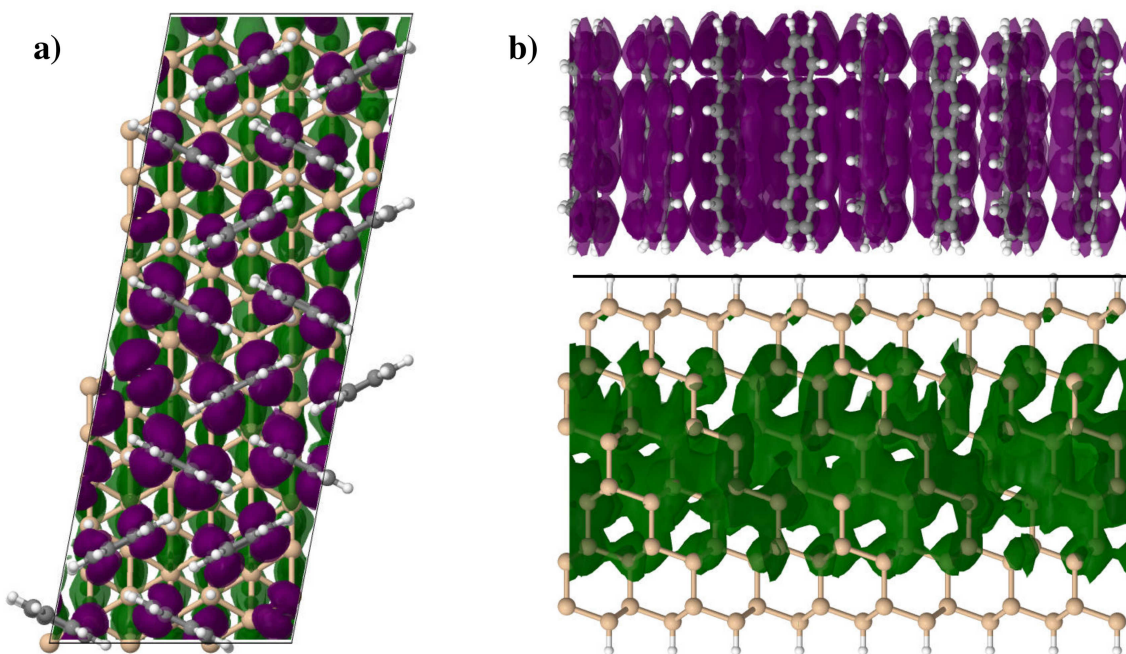


Figure 10. Visualization of the superposition of the eigenstates that contribute to the frontier orbitals of the HOMO (purple, three eigenstates) and LUMO (green, two eigenstates) at the Γ -point in tetracene film B at HSE06 level of theory, viewed along the z -axis (a) and the x -axis (b) of the computational supercell. The HOMO and the LUMO orbitals were visualized with Jmol[80] using a cutoff of $0.0004 \text{ e}/\text{\AA}^3$.

4. Conclusion

In conclusion, we present a complete computational study, using hybrid density-functional theory, of the electronic level alignment of Pc and Tc molecules and monolayer films at the intrinsic, i.e., undoped, H/Si(111) interface. As part of our study, we describe the impact of two different approaches to construct appropriate, commensurate supercell models for weakly bonded crystalline thin films on a crystalline substrate with a different lattice parameter:

- The first approach, shown for Pc, relies on available experimental input data as far as possible, particularly regarding the overall film lattice parameter and the molecular orientation (herringbone) inside the film. The remaining (unknown) geometric parameters of the film are obtained by structure optimization using standard van der Waals corrected semilocal density-functional theory. The resulting model (“Model Φ ” in the text above) achieves a plausible description of the electronic levels at the surface, but yields intra-film geometry parameters that differ from experimentally known molecular orientations and tilt angles in Pc monolayer films on other substrates[98]. Conceivably, the experimental lattice parameter is larger than the optimum lattice parameter that would be achieved by the density functional, causing other structural distortions (molecular tilts) in the theoretical result to compensate for the small systematic errors of the overall film lattice parameter in the theory.
- The second approach relies on using fully computationally obtained film and substrate lattice parameters, inferred from a free-standing model for the organic film, which are then joined to form commensurate supercells of varying size and strain. This approach yields molecular orientations in Pc films that are quite similar to available experimental data. A consistent description of electronic levels is achieved based on all commensurate film models considered that are large enough (up to 1,192 atoms) to exhibit low internal strain. In contrast, the simplest and smallest commensurate supercell model, with $\approx 10\%$ compressive strain, leads to noticeable distortions of the electronic structure.

Our investigation further reveals type-I like level alignments for the isolated Pc and Tc molecules, which prefer lying orientations on the substrate. Films of Pc and Tc exhibit standing molecular geometries and type-II like level alignments with the substrate. According to these results, it should be possible to separate carriers at the organic-inorganic interface, with electrons passing into the substrate and holes remaining on the films. In particular, based on our results and on known exciton energies in the literature[107, 7, 108, 109, 111, 112, 10], triplet excitons should also be able to split at the interface and inject electrons into the substrate. We finally note that our computationally predicted type-II level alignment in Tc films on intrinsic H/Si(111) substrates is not in agreement with a recent UV photoelectron study of thicker Tc films on n-doped H/Si(111).[20] The latter study concludes type-I level alignment between the

transport levels of the Tc film and the Si substrate, albeit from separate photoemission spectroscopy measurements of the clean substrate and the film. Nevertheless, in the same study, hole extraction from the substrate to the organic films is still possible in the combined film-substrate system, i.e., in principle in line with our result of type-II level alignment. While the origins of the observed differences are not clear, changes due to substrate doping are one possibility. It would be interesting to extend our methodology to consider semiconductor substrates with controlled doping densities, i.e., Fermi level, in future work.

Acknowledgments

S.M.J. thanks the Deutsche Forschungsgemeinschaft (DFG, German Research Foundation) for a postdoctoral fellowship, grant number 393196393. In addition, this work was funded by the Deutsche Forschungsgemeinschaft (DFG) - Projektnummer 182087777 - SFB 951. V.B. acknowledges support from N.S.F. Award DMR-1729297.

References

- [1] Shockley W and Queisser H J 1961 *Journal of Applied Physics* **32** 510–519
- [2] Rao A and Friend R H 2017 *Nature Reviews Materials* **2** 17063
- [3] Smith M B and Michl J 2010 *Chemical Reviews* **110** 6891–6936
- [4] Congreve D N, Lee J, Thompson N J, Hontz E, Yost S R, Reuswig P D, Bahlke M E, Reineke S, Van Voorhis T and Baldo M A 2013 *Science* **340** 334–337
- [5] Allardice J R, Thampi A, Dowland S, Xiao J, Gray V, Zhang Z, Budden P, Petty A J, Davis N J L K, Greenham N C, Anthony J E and Rao A 2019 *Journal of the American Chemical Society* **141** 12907–12915
- [6] Einzinger M, Wu T, Kompalla J F, Smith H L, Perkinson C F, Nienhaus L, Wieghold S, Congreve D N, Kahn A, Bawendi M G and Baldo M A 2019 *Nature* 90–94
- [7] Burgos J, Pope M, Swenberg C E and Alfano R R 1977 *Physica Status Solidi (b)* **83** 249–256
- [8] Jundt C, Klein G, Sipp B, Moigne J L, Joucla M and Villaeys A 1995 *Chemical Physics Letters* **241** 84–88
- [9] Merrifield R, Avakian P and Groff R 1969 *Chemical Physics Letters* **3** 386–388
- [10] Geacintov N, Pope M and Vogel F 1969 *Physical Review Letters* **22**(12) 593–596
- [11] Swenberg C and Stacy W 1968 *Chemical Physics Letters* **2** 327–328
- [12] Groff R P, Avakian P and Merrifield R E 1970 *Physical Review B* **1**(2) 815–817
- [13] Ehrler B, Walker B J, Boehm M L, Wilson M W B, Vaynzof Y, Friend R H and Greenham N C 2012 *Nature Communications* **3**
- [14] Wilson M W B, Rao A, Ehrler B and Friend R H 2013 *Accounts of Chemical Research* **46** 1330–1338
- [15] Dexter D 1979 *Journal of Luminescence* **18–19** 779–784
- [16] Futscher M H, Rao A and Ehrler B 2018 *ACS Energy Letters* **3** 2587–2592
- [17] Tkatchenko A, Romaner L, Hofmann O T, Zojer E, Ambrosch-Draxl C and Scheffler M 2010 *MRS Bulletin* **35** 435–442
- [18] Tabachnyk M, Ehrler B, Gelinass S, Boehm M L, Walker B J, Musselman K P, Greenham N C, Friend R H and Rao A 2014 *Nature Materials* **13** 1033–1038
- [19] Zhu X and Kahn A 2010 *MRS Bulletin* **35** 443–448
- [20] MacQueen R W, Liebhaber M, Niederhausen J, Mews M, Gersmann C, Jäckle S, Jäger K, Tayebjee M J Y, Schmidt T W, Rech B and Lips K 2018 *Mater. Horiz.* **5**(6) 1065–1075

- [21] Piland G B, Burdett J J, Hung T Y, Chen P H, Lin C F, Chiu T L, Lee J H and Bardeen C J 2014 *Chemical Physics Letters* **601** 33–38
- [22] Campbell I H and Crone B K 2009 *Journal of Applied Physics* **106** 113704
- [23] Heimel G, Salzmann I, Duhm S, Rabe J P and Koch N 2009 *Advanced Functional Materials* **19** 3874–3879
- [24] Wang H, Levchenko S V, Schultz T, Koch N, Scheffler M and Rossi M 2019 *Advanced Electronic Materials* **0** 1800891
- [25] Akaike K 2018 *Japanese Journal of Applied Physics* **57** 03EA03
- [26] Gish M K, Pace N A, Rumbles G and Johnson J C 2019 *The Journal of Physical Chemistry C* **123** 3923–3934
- [27] Hlawacek G and Teichert C 2013 *Journal of Physics-Condensed Matter* **25**
- [28] Shimada T, Nogawa H, Hasegawa T, Okada R, Ichikawa H, Ueno K and Saiki K 2005 *Applied Physics Letters* **87** 061917
- [29] Nishikata S, Sazaki G, Sadowski J T, Al-Mahboob A, Nishihara T, Fujikawa Y, Suto S, Sakurai T and Nakajima K 2007 *Physical Review B* **76**(16) 165424
- [30] Shi J and Qin X R 2006 *Physical Review B* **73**(12) 121303
- [31] Tersigni A, Shi J, Jiang D T and Qin X R 2006 *Physical Review B* **74**(20) 205326
- [32] Ruiz R, Nickel B, Koch N, Feldman L C, Haglund R F, Kahn A and Scoles G 2003 *Physical Review B* **67**(12) 125406
- [33] Sadowski J T, Al-Mahboob A, Fujikawa Y and Sakurai T 2008 *Nucleation and Growth of Thin Pentacene Films Studied by LEEM and STM* (Berlin, Heidelberg: Springer Berlin Heidelberg) pp 257–279 ISBN 978-3-540-77968-1
- [34] Fritz S E, Martin S M, Frisbie C D, Ward M D and Toney M F 2004 *Journal of the American Chemical Society* **126** 4084–4085
- [35] Yang H, Shin T J, Ling M M, Cho K, Ryu C Y and Bao Z 2005 *Journal of the American Chemical Society* **127** 11542–11543
- [36] Tersigni A, Qin X R, Kim C Y, Gordon R A and Jiang D T 2011 *Physical Review B* **84**(3) 035303
- [37] Matsubara R, Sakai M, Kudo K, Yoshimoto N, Hirosawa I and Nakamura M 2011 *Organic Electronics* **12** 195–201
- [38] Nabok D, Puschnig P, Ambrosch-Draxl C, Werzer O, Resel R and Smilgies D M 2007 *Physical Review B* **76**(23) 235322
- [39] Ruiz R, Nickel B, Koch N, Feldman L C, Haglund R F, Kahn A, Family F and Scoles G 2003 *Physical Review Letters* **91**(13) 136102
- [40] Schiefer S, Huth M, Dobrinevski A and Nickel B 2007 *Journal of the American Chemical Society* **129** 10316–10317
- [41] Verlaak S, Steudel S, Heremans P, Janssen D and Deleuze M S 2003 *Physical Review B* **68**(19) 195409
- [42] Yoshida H, Inaba K and Sato N 2007 *Applied Physics Letters* **90** 181930
- [43] Ruiz R, Mayer A C, Malliaras G G, Nickel B, Scoles G, Kazimirov A, Kim H, Headrick R L and Islam Z 2004 *Applied Physics Letters* **85** 4926–4928
- [44] Sadowski J T, Nagao T, Yaginuma S, Fujikawa Y, Al-Mahboob A, Nakajima K, Sakurai T, Thayer G E and Tromp R M 2005 *Applied Physics Letters* **86** 073109
- [45] Kittel C 2005 *Introduction to Solid State Physics* (John Wiley & Sons)
- [46] Hooks D E, Fritz T and Ward M D 2001 *Advanced Materials* **13** 227–241
- [47] Cohen A J, Mori-Sanchez P and Yang W 2012 *Chemical Reviews* **112** 289–320
- [48] Mori-Sánchez P, Cohen A J and Yang W 2008 *Physical Review Letters* **100**(14) 146401
- [49] Marom N 2017 *Journal of Physics: Condensed Matter* **29** 103003
- [50] Marom N, Hod O, Scuseria G E and Kronik L 2008 *The Journal of Chemical Physics* **128** 164107
- [51] Körzdörfer T, Kümmel S, Marom N and Kronik L 2009 *Physical Review B* **79**(20) 201205
- [52] Körzdörfer T, Kümmel S, Marom N and Kronik L 2010 *Physical Review B* **82**(12) 129903
- [53] Marom N, Caruso F, Ren X, Hofmann O T, Körzdörfer T, Chelikowsky J R, Rubio A, Scheffler

- M and Rinke P 2012 *Physical Review B* **86**(24) 245127
- [54] Tamblyn I, Darancet P, Quek S Y, Bonev S A and Neaton J B 2011 *Physical Review B* **84**(20) 201402
- [55] Draxl C, Nabok D and Hannewald K 2014 *Accounts of Chemical Research* **47** 3225–3232
- [56] Garza A J and Scuseria G E 2016 *The Journal of Physical Chemistry Letters* **7** 4165–4170
- [57] Hofmann O T, Deinert J C, Xu Y, Rinke P, Stähler J, Wolf M and Scheffler M 2013 *The Journal of Chemical Physics* **139** 174701
- [58] Perdew J P, Yang W, Burke K, Yang Z, Gross E K U, Scheffler M, Scuseria G E, Henderson T M, Zhang I Y, Ruzsinszky A, Peng H, Sun J, Trushin E and Görling A 2017 *Proceedings of the National Academy of Sciences* **114** 2801–2806
- [59] Nabok D, Höfling B and Draxl C 2019 *Chemistry of Materials* DOI:10.1021/acs.chemmater.9b01802
- [60] Kronik L and Kümmel S 2018 *Advanced Materials* **30** 1706560
- [61] Blum V, Gehrke R, Hanke F, Havu P, Havu V, Ren X, Reuter K and Scheffler M 2009 *Computer Physics Communications* **180** 2175–2196
- [62] Havu V, Blum V, Havu P and Scheffler M 2009 *Journal of Computational Physics* **228** 8367–8379
- [63] Ren X, Rinke P, Blum V, Wieferink J, Tkatchenko A, Sanfilippo A, Reuter K and Scheffler M 2012 *New Journal of Physics* **14** 053020
- [64] Knuth F, Carbogno C, Atalla V, Blum V and Scheffler M 2015 *Computer Physics Communications* **190** 33–50
- [65] Yu V W Z, Corsetti F, García A, Huhn W P, Jacquelin M, Jia W, Lange B, Lin L, Lu J, Mi W, Seifitokaldani A, Vaázquez-Mayagoitia A, Yang C, Yang H and V B 2018 *Computer Physics Communications* **222** 267–285
- [66] Marek A, Blum V, Johanni R, Havu V, Lang B, Auckenthaler T, Heinecke A, Bungartz H J and Lederer H 2014 *Journal of Physics: Condensed Matter* **26** 213201
- [67] Ihrig A C, Wieferink J, Zhang I Y, Ropo M, Ren X, Rinke P, Scheffler M and Blum V 2015 *New Journal of Physics* **17** 093020
- [68] Levchenko S V, Ren X, Wieferink J, Johanni R, Rinke P, Blum V and Scheffler M 2015 *Computer Physics Communications* **192** 60–69
- [69] Perdew J P, Burke K and Ernzerhof M 1996 *Physical Review Letters* **77**(18) 3865–3868
- [70] Tkatchenko A and Scheffler M 2009 *Physical Review Letters* **102**(7) 073005
- [71] Hoja J, Reilly A M and Tkatchenko A 2017 *Wiley Interdisciplinary Reviews: Computational Molecular Science* **7** e1294
- [72] Al-Saidi W A, Voora V K and Jordan K D 2012 *Journal of Chemical Theory and Computation* **8** 1503–1513
- [73] Bedoya-Martinez N, Giunchi A, Salzillo T, Venuti E, Della Valle R G and Zojer E 2018 *Journal of Chemical Theory and Computation* **14** 4380–4390
- [74] Schatschneider B, Liang J J, Reilly A M, Marom N, Zhang G X and Tkatchenko A 2013 *Physical Review B* **87**(6) 060104
- [75] Ambrosetti A, Reilly A M, Jr R A D and Tkatchenko A 2014 *The Journal of Chemical Physics* **140** 18A508
- [76] Heyd J, Scuseria G E and Ernzerhof M 2003 *The Journal of Chemical Physics* **118** 8207–8215
- [77] Heyd J, Scuseria G E and Ernzerhof M 2006 *The Journal of Chemical Physics* **124** 219906
- [78] Krukau A V, Vydrov O A, Izmaylov A F and Scuseria G E 2006 *The Journal of Chemical Physics* **125** 224106
- [79] Mulliken R S 1955 *The Journal of Chemical Physics* **23** 1833–1840
- [80] Jmol: an open-source Java viewer for chemical structures in 3D. <http://www.jmol.org/>
- [81] Humphrey W, Dalke A and Schulten K 1996 *Journal of Molecular Graphics* **14** 33–38
- [82] Neugebauer J and Scheffler M 1992 *Physical Review B* **46**(24) 16067–16080
- [83] Lejaeghere K, Bihlmayer G, Björkman T, Blaha P, Blügel S, Blum V, Caliste D, Castelli I E, Clark S J, Dal Corso A, de Gironcoli S, Deutsch T, Dewhurst J K, Di Marco I, Draxl C,

- Dulak M, Eriksson O, Flores-Livas J A, Garrity K F, Genovese L, Giannozzi P, Giantomassi M, Goedecker S, Gonze X, Grånäs O, Gross E K U, Gulans A, Gygi F, Hamann D R, Hasnip P J, Holzwarth N A W, Iuşan D, Jochym D B, Jollet F, Jones D, Kresse G, Koepnik K, Küçükbenli E, Kvashnin Y O, Loch I L M, Lubeck S, Marsman M, Marzari N, Nitzsche U, Nordström L, Ozaki T, Paulatto L, Pickard C J, Poelmans W, Probert M I J, Refson K, Richter M, Rignanese G M, Saha S, Scheffler M, Schlipf M, Schwarz K, Sharma S, Tavazza F, Thunström P, Tkatchenko A, Torrent M, Vanderbilt D, van Setten M J, Van Speybroeck V, Wills J M, Yates J R, Zhang G X and Cottenier S 2016 *Science* **351**
- [84] Huhn W P and Blum V 2017 *Physical Review Materials* **1**(3) 033803
- [85] Li Y and Galli G 2010 *Physical Review B* **82**(4) 045321
- [86] Sagisaka K, Nara J and Bowler D 2017 *Journal of Physics: Condensed Matter* **29** 145502
- [87] Delley B and Steigmeier E F 1995 *Applied Physics Letters* **67** 2370–2372
- [88] Scherpelz P and Galli G 2017 *Physical Review Materials* **1**(2) 021602
- [89] Fischetti M V, Fu B, Narayanan S and Kim J 2011 *Semiclassical and Quantum Electronic Transport in Nanometer-Scale Structures: Empirical Pseudopotential Band Structure, Monte Carlo Simulations and Pauli Master Equation* (New York, NY: Springer New York) pp 183–247
- [90] Tsetseris L and Pantelides S T 2005 *Applied Physics Letters* **87** 233109
- [91] Ample F and Joachim C 2008 *Surface Science* **602** 1563–1571
- [92] DiLabio G A, Johnson E R and Pitters J 2009 *The Journal of Physical Chemistry C* **113** 9969–9973
- [93] Skone J H, Govoni M and Galli G 2016 *Phys. Rev. B* **93**(23) 235106
- [94] Neaton J B, Hybertsen M S and Louie S G 2006 *Physical Review Letters* **97**(21) 216405
- [95] Refaely-Abramson S, Sharifzadeh S, Jain M, Baer R, Neaton J B and Kronik L 2013 *Phys. Rev. B* **88**(8) 081204
- [96] Kakudate T, Yoshimoto N and Saito Y 2007 *Applied Physics Letters* **90** 081903
- [97] Kiyomura T, Nemoto T, Ogawa T, Minari T, Yoshida K, Kurata H and Isoda S 2006 *Japanese Journal of Applied Physics* **45** 401–404
- [98] Mannsfeld S C B, Virkar A, Reese C, Toney M F and Bao Z 2009 *Advanced Materials* **21** 2294–2298
- [99] Meyenburg I, Breuer T, Karthäuser A, Chatterjee S, Witte G and Heimbrodt W 2016 *Physical Chemistry Chemical Physics* **18**(5) 3825–3831
- [100] Siegrist T, Kloc C, Schön J H, Batlogg B, Haddon R C, Berg S and Thomas G A 2001 *Angewandte Chemie International Edition* **40** 1732–1736
- [101] Ambrosch-Draxl C, Nabok D, Puschnig P and Meisenbichler C 2009 *New Journal of Physics* **11** 125010
- [102] Holmes D, Kumaraswamy S, Matzger A J and Vollhardt K P C 1999 *Chemistry – A European Journal* **5** 3399–3412
- [103] Campbell R B, Robertson J M and Trotter J 1961 *Acta Crystallographica* **14** 705–711
- [104] Della Valle R G, Venuti E, Brillante A and Girlando A 2009 *ChemPhysChem* **10** 1783–1788
- [105] Haas S, Batlogg B, Besnard C, Schiltz M, Kloc C and Siegrist T 2007 *Physical Review B* **76**(20) 205203
- [106] Refaely-Abramson S, Jain M, Sharifzadeh S, Neaton J B and Kronik L 2015 *Phys. Rev. B* 081204(R)
- [107] Geacintov N, Burgos J, Pope M and Strom C 1971 *Chemical Physics Letters* **11** 504 – 508
- [108] Vilar M R, Heyman M and Schott M 1983 *Chemical Physics Letters* **94** 522 – 526
- [109] de Cheveigné S, Klein J, Léger A, Belin M and Défourneau D 1977 *Phys. Rev. B* **15**(2) 750–754
- [110] Sebastian L, Weiser G and Bässler H 1981 *Chemical Physics* **61** 125 – 135
- [111] Tomkiewicz Y, Groff R P and Avakian P 1971 *The Journal of Chemical Physics* **54** 4504–4507
- [112] Vaubel G and Baessler H 1971 *Molecular Crystals and Liquid Crystals* **15** 15–25

Supplementary Information

Pentacene and Tetracene Molecules and Films on H/Si(111): Level Alignment from Hybrid Density Functional Theory

Svenja M. Janke^{1,2}, Mariana Rossi^{2,3}, Sergey Levchenko^{4,2}, Sebastian Kokott², Matthias Scheffler² and Volker Blum^{1,5}

¹ *Department of Mechanical Engineering and Materials Science, Duke University, Durham, NC, USA*

² *Fritz Haber Institute of the Max Planck Society, Berlin, Germany*

³ *Max Planck Institute for Structure and Dynamics of Matter, Hamburg, Germany*

⁴ *Skolkovo Institute of Science and Technology, Moscow, Russia*

⁵ *Department of Chemistry, Duke University, Durham, NC, USA*

Bulk Pentacene

Table S1: Comparison of the lattice parameters and angles of the crystal polymorph of pentacene obtained at 90 K and 300 K from x-ray diffraction [1] to values obtained after relaxation with DFT-PBE and Tkatchenko-Scheffler (TS) [2] as well as with many-body dispersion (MBD) [3] van der Waals (vdW) correction with “intermediate” settings. All residual forces on the atoms and lattice vectors in the computational structures are below 0.005 eV/Å. The computational energy minimization of the structure used the lattice parameters and atomic positions as supplied in the $T=90$ K cif-file of the experimental publication [1] as its initial guess. All atomic positions and lattice parameters were allowed to relax. The values obtained with DFT-PBE+TS are in good agreement with the DFT-PBE+MBD and experimental values.

	\mathbf{a}_1 (Å)	\mathbf{a}_2 (Å)	\mathbf{c} (Å)	α (°)	β (°)	γ (°)
DFT-PBE+TS	6.11	7.63	14.59	78	87	85
DFT-PBE+MBD	6.25	7.67	14.38	77	89	84
90 K [1]	6.239(1)	7.636(1)	14.330 (2)	76.978(3)	88.136(3)	84.415(3)
300 K [1]	6.266(1)	7.775(1)	14.530(1)	76.475(4)	87.682(4)	84.684 (4)

Bulk Tetracene

Table S2: Comparison of the lattice parameters and angles of a tetracene crystal measured by x-ray diffraction [4, 5] to values obtained after relaxation with DFT-PBE+TS and DFT-PBE+MBD at “intermediate” settings. All residual forces on atoms and lattice vectors were below 0.005 eV/Å. The relaxation used the lattice parameters and atomic positions as supplied in the cif-file of the experimental publication [4], CSD 114446 [6] as initial guess. All atomic positions and lattice parameters were allowed to relax.

	\mathbf{a}_1 (Å)	\mathbf{a}_2 (Å)	\mathbf{c} (Å)	α (°)	β (°)	γ (°)
DFT-PBE+TS	5.97	7.76	12.97	77	73	86
DFT-PBE+MBD	5.98	7.82	12.95	77	72	86
175 K [4]	6.0565(9)	7.8376(11)	13.0104(18)	77.127(2)	72.118(2)	85.792(2)
140 [5]	5.99(2)	7.77(4)	13.2(10)	102.4(5)	113.8(9)	86.0(4)
293 K [5]	6.065(2)	7.915(5)	13.445(12)	101.10(6)	113.31(9)	85.91(4)

Bulk Si

Table S3: Predicted lattice parameters for bulk Si using different density functionals and van der Waals corrections with a $(12 \times 12 \times 12)$ k-point mesh and “tight settings”. Minimum-energy lattice parameters at the Born-Oppenheimer surface were extracted by fitting total energies for different volumes to the Birch-Murnaghan formula [7, 8].

Functional	a_0 (Å)
DFT-PBE	5.472
DFT-PBE+TS	5.450
DFT-PBE+MBD	5.436
HSE06+MBD	5.414
HSE06+TS	5.426
Lit. (RT) [9, 10]	5.431
Lit. (6.4 K) [11]	5.419

H/Si(111) Band Gap Convergence with Slab Thickness

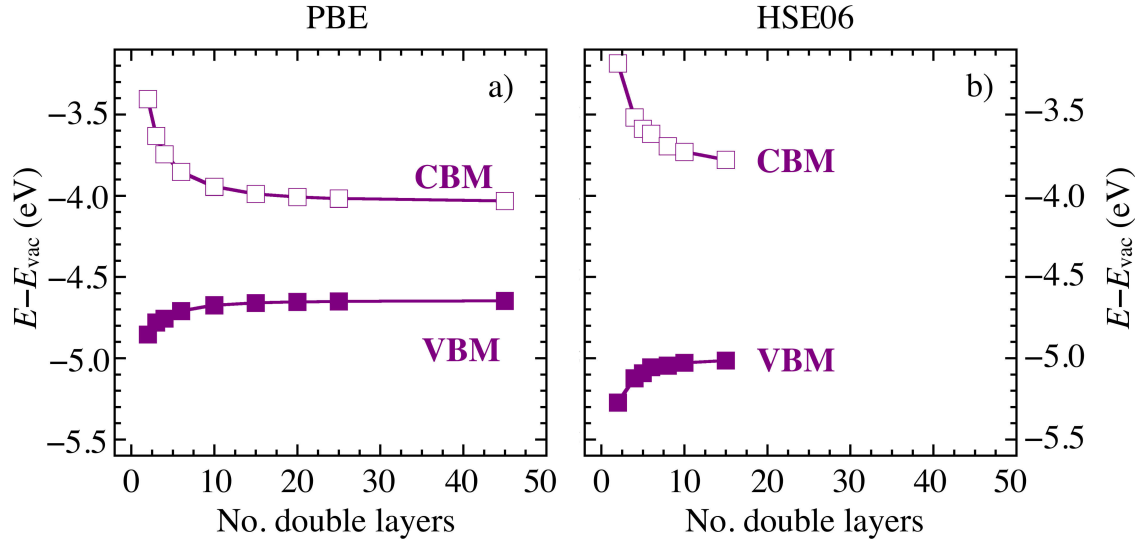


Figure S1: Variation of the valence band maximum (VBM, filled purple squares) and the conduction band minimum (CBM, open purple squares) of the H/Si(111) slab models as a function of the number of double layers. The energy axis values are referenced to the vacuum level ($E - E_{vac}$). a) DFT-PBE results. b) DFT-HSE06 results from surfaces relaxed with DFT-PBE+TS and HSE06+TS. We relaxed $(1 \times 1 \times n)$ H/Si(111) slabs as a function of the number of Si double layers n in z -direction, using a $(12 \times 12 \times 1)$ k-point mesh and a vacuum layer thickness of 200 Å, until the minimum force was below 0.01 eV/Å with “tight” numerical settings of FHI-aims. The atomic positions of the top and bottom two Si double layers and the H atoms were relaxed while all others were kept fixed during the relaxation. The slow convergence of the band gap with the thickness of the slab has been observed previously for various Si surfaces [12–15] and is explained by a quantum well behavior due to confinement of the electronic eigenstates in thin slabs [14–16]. The DFT-HSE06 bulk band gap of 1.165 eV (“tight” settings) is close to the experimental value of 1.17 eV at $T=0$ K [10].

Dilute Limit of Acene Adsorption on H/Si(111)

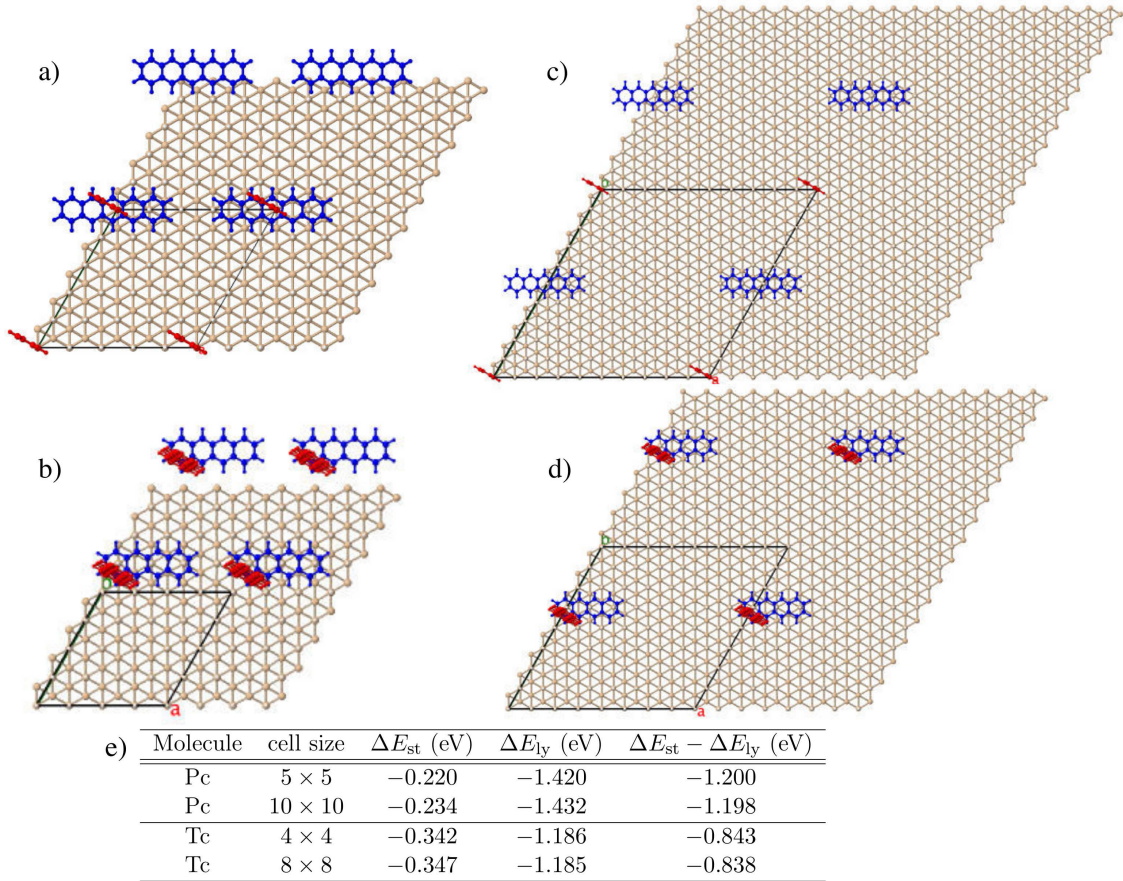


Figure S2: Images of lying (blue) and standing (red) acenes adsorbed on H/Si(111) in the unit cells considered for the dilute limit in this work. We chose a cell size in xy -plane of (5×5) for Pc (a) and (4×4) for Tc (b) to isolate each molecule from its periodic images. If the cell size is doubled (c, d), the energy per molecule remains within a few meV of that of the smaller cell size (e, Table), both for a lying molecule (ΔE_{ly}) and a standing molecule (ΔE_{st}). Total energy calculations were performed using DFT-PBE+TS and “intermediate” settings. For the geometries in a–e, a previously relaxed $(1 \times 1 \times 10)$ double layer unit cell was repeated to form the different underlying slabs. For the positions of the molecules, we selected one molecule in a lying and one in a standing orientation from the structure search in Figure 3 of the main text. We then transferred the molecular positions to the slab models shown in this figure, keeping the distance to the average plane through the surface hydrogen atoms in the original structure.

Computational Parameters for Interface Models of the Dilute and Monolayer Limit

Table S4: For different models considered in the text, the table shows the unit cell area, total number of atoms (film and substrate) per unit cell, and k-point meshes used for structure relaxation (DFT-PBE+TS) and electronic structure calculation (HSE06). The HSE06 calculations were performed using the structures obtained with DFT-PBE+TS. “Intermediate” settings were used.

Model	area (\AA^2)	No. Tc +H/Si(111) atoms per cell	No. Pc +H/Si(111) atoms per cell	DFT-PBE+TS k-point mesh	HSE06 k-point mesh
Dilute limit					
Tc	206	382		$(2 \times 2 \times 1)$	$(5 \times 5 \times 1)$
Pc	322		586	$(2 \times 2 \times 1)$	$(5 \times 5 \times 1)$
Monolayer limit					
A	77	204	228	$(3 \times 5 \times 1)$	$(4 \times 7 \times 1)$
B	347	858	954	$(6 \times 2 \times 1)$	$(3 \times 1 \times 1)$
C	424	1062	1182	$(6 \times 2 \times 1)$	$(3 \times 1 \times 1)$
D	309	756	840	$(6 \times 2 \times 1)$	$(3 \times 1 \times 1)$
E	424	1062	1182	$(6 \times 3 \times 1)$	$(4 \times 2 \times 1)$
Φ	283		740	$(6 \times 2 \times 1)$	$(6 \times 2 \times 1)$

Free-Standing Monolayer Relaxation Geometries

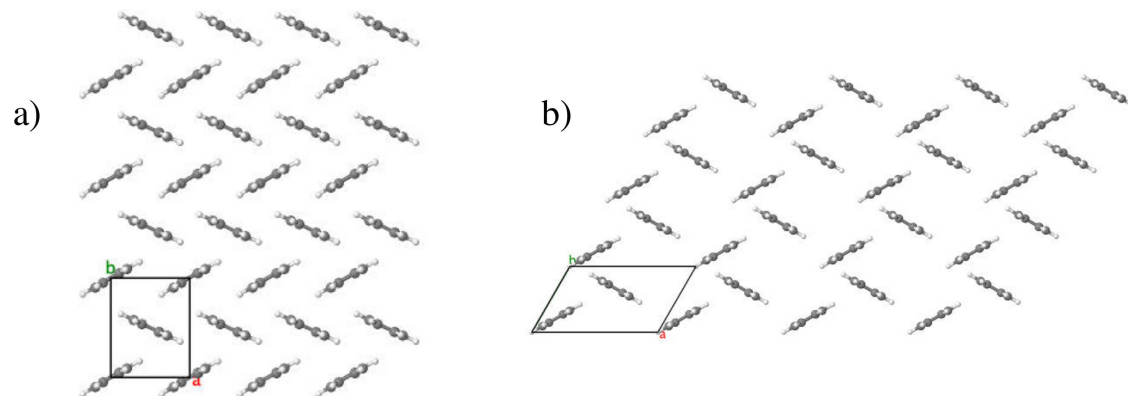


Figure S3: Initial geometry before relaxation of a Pc (a) and Tc (b) monolayer in vacuum viewed along **c**-direction. The unit cell used in the relaxation is indicated by a black frame. For Pc, the initial unit cell was based on the unit cell I structure suggested by Nishikata *et al.* [17]. For Tc, the initial unit cell was based on the observation that Tc molecules assume a standing orientation and a herringbone pattern on weakly interacting surfaces [18]. Relaxations were performed with PBE+TS and “intermediate” settings until all residual forces on atoms and lattice vectors were below 0.005 eV/Å.

Free-Standing Monolayer Lattice Parameters and Energies

Table S5: In-plane lattice parameters of the Pc and Tc unit cells after relaxation, calculated using “intermediate” settings with DFT-PBE and TS [2] as well as with MBD [3] van der Waals (vdW) corrections. All cells were rotated to have one of their lattice parameters parallel to the x -axis. Within the cells, all molecules assumed a herringbone pattern after relaxation. For Pc, unit cells were based on unit cell I determined for Pc dendrites on H/Si(111) [17]. For Tc, a unit cell was modeled with a herringbone pattern between two standing molecules based on the experimentally described geometry[18, 19]. For reference, relaxations for Tc were also performed based on the unit cell I used for Pc as a starting point. The resulting lattice vectors are almost identical. The energy is given with reference to the isolated molecule in vacuum as energy per molecule ΔE_{mol} . The two bottom lines show the lattice parameters from the experimental study of Pc dendrites on H/Si(111) at room temperature [17]. Their energy marked (^a) corresponds to the energy after relaxation of the molecular positions with DFT-PBE+TS while the lattice parameters remained fixed to the experimental lattice parameters. The experimental error bars for the unit cell parameters are therefore included in the table and the notation “exp/TS” indicates the fact that experimental geometries and computed energies are combined in the lowest two lines.

	vdW	c_{11} (Å)	c_{12} (Å)	c_{21} (Å)	c_{22} (Å)	ΔE_{mol} (eV)
Pc (unit cell I based)	TS	5.85	0.00	0.00	7.36	− 1.974
Tc (unit cell I based)	TS	5.83	0.00	0.02	7.41	− 1.556
Tc (model)	TS	5.85	0.00	0.00	7.39	− 1.556
Pc (unit cell I based)	MBD	5.89	0.00	0.00	7.40	− 1.588
Tc (unit cell I based)	MBD	5.89	0.00	0.00	7.44	− 1.255
Pc supercell I	exp/TS	6.02±0.01	0.00	0.00	7.62±0.01	−1.909 ^a
Pc supercell II	exp/TS	5.98±0.01	0.00	0.00	7.56±0.01	−1.932 ^a

Models of Monolayer Pentacene Interfaces with H/Si(111): Geometries and Energies

Table S6: For pentacene, the lattice parameters and angle γ between the lattice vectors, the energy per molecule ΔE , the average angle θ of the molecules to the surface normal, the average herringbone angle ω and the average distance d_z of the z -component of the molecules' center of mass to the plane through the average height of the first layer of Si atoms (see Figure S4) are shown.

Film	a (Å)	b (Å)	$\gamma(^{\circ})$	ΔE (eV)	$\theta (^{\circ})$	$\omega (^{\circ})$	d_z (Å)
free-standing	5.85	7.36	90				
A	5.78	6.68	90	−2.022	1	46	10.0
B	5.78	30.59	79	−2.213	1	54	10.2
C	5.78	37.16	81	−2.221	1	52	10.1
D	13.90	30.10	48	−2.175	1	51	10.2
E	5.78	37.16	81	−2.221	1	52	10.1
Φ	6.13	7.71	87	−2.222	22	49	9.5

Models of Monolayer Tetracene Interfaces with H/Si(111): Geometries and Energies

Table S7: For tetracene, the lattice parameters and angle γ between the lattice vectors, the energy per molecule ΔE , the average angle θ of the molecules to the surface normal, the average herringbone angle ω and the average distance d_z of the z -component of the molecules' center of mass to the plane through the average height of the first layer of Si atoms (see Figure S4) are shown.

Film	a (Å)	b (Å)	$\gamma(^{\circ})$	ΔE (eV)	$\theta (^{\circ})$	$\omega (^{\circ})$	d_z (Å)
free-standing	5.85	7.39	90				
A	5.78	6.68	90	−1.625	1	46	8.8
B	5.78	30.59	79	−1.798	1	54	8.9
C	5.78	37.16	81	−1.762	1	53	8.9
D	13.90	30.10	48	−1.780	8	51	8.9
E	5.78	37.16	81	−1.802	1	53	8.9

Definition of Monolayer Film Adsorption Distance

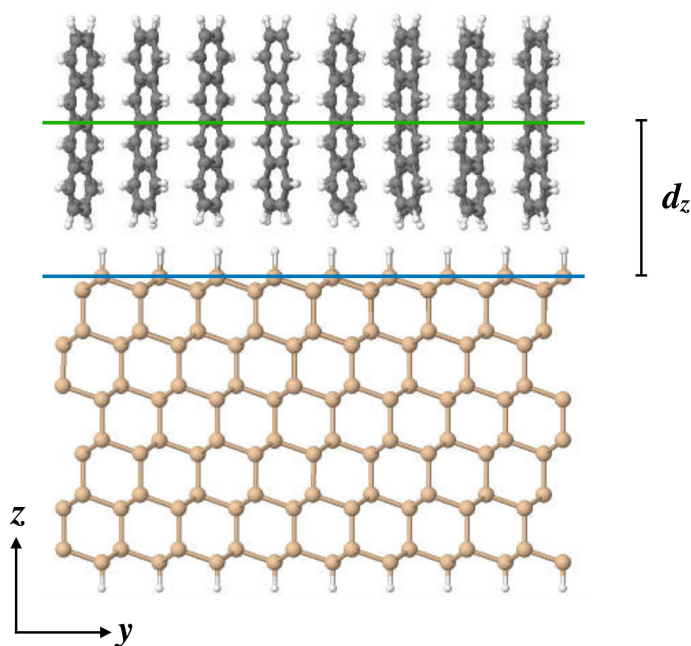


Figure S4: Side view of the slab model of a tetracene film on H/Si(111), model B. The adsorption distance d_z of a molecule to the substrate is defined as the distance between the z -component of the center-of-mass of the molecular atoms (green line) and the plane defined by the average z -position of the top layer of Si atoms (blue line).

Frontier Orbital Energies in the Dilute and Monolayer Limits

Table S8: The DFT-HSE06 energy and the gaps of the VBM and CBM of the substrate and HOMO and LUMO of the film extracted from HSE06 Mulliken analysis, counting all contributions above 0.0025 total occupation. The band gap of the isolated substrate (PBE+TS geometry, calculated with HSE06, “intermediate” settings) is 1.465 eV, the gaps given for “Tetracene” and “Pentacene” correspond to those of the isolated molecule in vacuum.

Limit	Model	substrate			film			overall band gap (eV)
		VBM (eV)	CBM (eV)	gap (eV)	HOMO (eV)	LUMO (eV)	gap (eV)	
Isolated	Tetracene				2.269			
Monolayer Limit	A	-5.59	-4.08	1.51	-4.41	-3.31	1.10	0.33
	B	-5.53	-4.07	1.46	-4.86	-3.03	1.83	0.79
	C	-5.54	-4.07	1.47	-4.76	-3.07	1.68	0.69
	D	-5.52	-4.02	1.50	-4.78	-3.07	1.71	0.76
	E	-5.54	-4.07	1.47	-4.80	-3.06	1.74	0.73
Dilute Limit	Tc lying	-5.40	-4.07	1.33	-5.68	-3.51	2.17	1.61
	Tc standing	-5.42	-4.08	1.33	-5.42	-3.21	2.21	1.33
Isolated	Pentacene				1.696			
Monolayer Limit	A	-5.67	-4.16	-1.50	-4.30	-3.86	-0.44	0.14
	B	-5.59	-4.13	1.46	-4.58	-3.42	1.16	0.45
	C	-5.60	-4.13	1.47	-4.53	-3.44	1.09	0.40
	D	-5.66	-4.16	1.50	-4.66	-3.52	1.14	0.50
	E	-5.61	-4.14	1.46	-4.53	-3.44	1.09	0.39
	Φ	-5.53	-4.06	1.46	-4.64	-3.59	1.05	0.58
Dilute Limit	Pc lying	-5.40	-4.07	1.33	-5.38	-3.74	1.63	1.31
	Pc standing	-5.42	-4.09	1.33	-5.14	-3.45	1.69	1.06

H/Si(111) Surface Density of States Decomposition

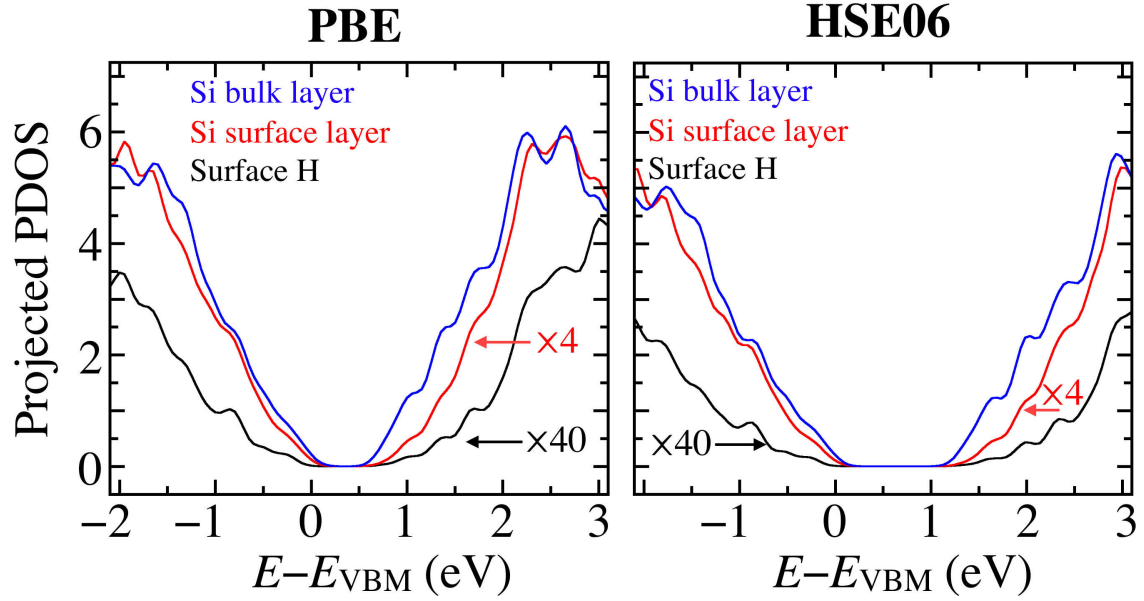


Figure S5: Projected densities of states onto surface H (black), Si bulk layers (blue) and Si surface layers (red) for 10 Si double layers in H/Si(111) calculated with PBE (left) and HSE06 (right) from surfaces relaxed with DFT-PBE+TS and HSE06+TS at “intermediate” settings of FHI-aims. The curves of the surface H and surface Si are scaled by a factor of 40 and 4, respectively. The onset of the band gap is formed by contributions from the bulk layers (blue).

References

- [1] C. C. Mattheus, A. B. Dros, J. Baas, A. Meetsma, J. L. d. Boer, and T. T. M. Palstra, “Polymorphism in pentacene,” *Acta Crystallographica Section C*, vol. 57, no. 8, pp. 939–941, 2001.
- [2] A. Tkatchenko and M. Scheffler, “Accurate molecular van der waals interactions from ground-state electron density and free-atom reference data,” *Physical Review Letters*, vol. 102, p. 073005, 2009.
- [3] A. Ambrosetti, A. M. Reilly, R. A. D. Jr., and A. Tkatchenko, “Long-range correlation energy calculated from coupled atomic response functions,” *The Journal of Chemical Physics*, vol. 140, no. 18, p. 18A508, 2014.
- [4] D. Holmes, S. Kumaraswamy, A. J. Matzger, and K. P. C. Vollhardt, “On the nature of nonplanarity in the [n]phenylenes,” *Chemistry – A European Journal*, vol. 5, no. 11, pp. 3399–3412, 1999.
- [5] U. Sondermann, A. Kutoglu, and H. Bassler, “X-ray diffraction study of the phase transition in crystalline tetracene,” *The Journal of Physical Chemistry*, vol. 89, no. 9, pp. 1735–1741, 1985.
- [6] C. R. Groom, I. J. Bruno, M. P. Lightfoot, and S. C. Ward, “The Cambridge Structural Database,” *Acta Crystallographica Section B*, vol. 72, no. 2, pp. 171–179, 2016.
- [7] F. D. Murnaghan, “The compressibility of media under extreme pressures,” *Proceedings of the National Academy of Sciences*, vol. 30, no. 9, pp. 244–247, 1944.
- [8] F. Birch, “Finite elastic strain of cubic crystals,” *Physical Review*, vol. 71, pp. 809–824, 1947.
- [9] CRC Handbook of Chemistry and Physics, Internet Version, 2005. p.9–44.
- [10] C. Kittel, *Introduction to Solid State Physics*. John Wiley & Sons, 2005.
- [11] D. N. Batchelder and R. O. Simmons, “Lattice constants and thermal expansivities of silicon and of calcium fluoride between 6 and 322 k,” *The Journal of Chemical Physics*, vol. 41, no. 8, pp. 2324–2329, 1964.
- [12] Y. Li and G. Galli, “Electronic and spectroscopic properties of the hydrogen-terminated si(111) surface from ab initio calculations,” *Physical Review B*, vol. 82, p. 045321, 2010.
- [13] K. Sagisaka, J. Nara, and D. Bowler, “Importance of bulk states for the electronic structure of semiconductor surfaces: implications for finite slabs,” *Journal of Physics: Condensed Matter*, vol. 29, no. 14, p. 145502, 2017.
- [14] B. Delley and E. F. Steigmeier, “Size dependence of band gaps in silicon nanostructures,” *Applied Physics Letters*, vol. 67, no. 16, pp. 2370–2372, 1995.
- [15] P. Scherpelz and G. Galli, “Optimizing surface defects for atomic-scale electronics: Si dangling bonds,” *Physical Review Materials*, vol. 1, p. 021602, 2017.
- [16] M. V. Fischetti, B. Fu, S. Narayanan, and J. Kim, *Semiclassical and Quantum Electronic Transport in Nanometer-Scale Structures: Empirical Pseudopotential Band Structure, Monte Carlo Simulations and Pauli Master Equation*, pp. 183–247. New York, NY: Springer New York, 2011.

- [17] S. Nishikata, G. Sazaki, J. T. Sadowski, A. Al-Mahboob, T. Nishihara, Y. Fujikawa, S. Suto, T. Sakurai, and K. Nakajima, "Polycrystalline domain structure of pentacene thin films epitaxially grown on a hydrogen-terminated si(111) surface," *Physical Review B*, vol. 76, p. 165424, 2007.
- [18] J. Shi and X. R. Qin, "Flux dependence of the morphology of a tetracene film on hydrogen-passivated si(100)," *Physical Review B*, vol. 73, p. 121303, 2006.
- [19] A. Tersigni, J. Shi, D. T. Jiang, and X. R. Qin, "Structure of tetracene films on hydrogen-passivated si(001) studied via stm, afm, and nexafs," *Physical Review B*, vol. 74, p. 205326, 2006.