

Controlling neutral and charged excitons in MoS₂ with defects

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Abstract

In this contribution, we use heavy ion irradiation and photoluminescence spectroscopy to demonstrate that defects can be used to tailor the optical properties of two-dimensional molybdenum disulfide (MoS₂). Ion irradiation introduces defects that can control optical excitations in the inner core shell of MoS₂ by binding A_{1s}- and B_{1s}- excitons and A⁻ trions at various defect densities. We show that up to the fluences of $1 \times 10^{14} \text{ cm}^{-2}$, the MoS₂ lattice remains crystalline, and defect densities can be controlled, while at higher fluences ($\geq 1 \times 10^{15} \text{ cm}^{-2}$) the large number of introduced defects distorts the excitonic structure of the material. Insights gained from this study will aid in understanding the many-body interactions in low dimensional materials and may ultimately be used to develop novel materials for optoelectronic applications.

Keywords

Ion Irradiation, Defects, Optical properties

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I. INTRODUCTION

Optoelectronic properties of transition metal dichalcogenides (TMDs) can be manipulated by taking the system out of equilibrium, which has been done using temperature¹⁻⁴, pressure^{5,6}, electrical bias^{7,8}, magnetic fields⁹, and light ion¹⁰⁻¹² and electron irradiations^{12,13}. Molybdenum disulfide (MoS₂) remains the most widely studied TMD material system because of its remarkable flexibility¹⁴ and enhanced photoresponsivity¹⁵. Fabrication of MoS₂ by chemical vapor deposition, mechanical exfoliation, and physical vapor deposition introduces intrinsic defects, which has been shown to degrade material's properties¹⁶. Carbon and oxygen impurities introduced at various stages of fabrication can occupy interstitial sites in MoS₂¹⁷, act as traps for charge carriers¹⁸, and degrade carrier concentrations¹⁹.

Theory, on the other hand, predicts that defects can be used to control properties of single and multi-layered materials. According to first principles calculations, native defects such as S and Mo vacancies create impurity levels in bulk MoS₂ within the bandgap and in the valence band and lead to pronounced photon absorption in visible light region²⁰. These impurity levels can be used to manipulate photovoltaic energy conversion at longer wavelength²⁰. Several groups investigated the role defects (mostly vacancies and grain boundaries) play in determining the electronic, optical, and optoelectronic properties of low-dimensional MoS₂ systems, but no explicit understanding of the phenomena has been attained^{16,20-24}. Most studies focused on doping MoS₂ to achieve control of the optical properties, which included studies on chemical doping²⁵ and low energy ion implantation²⁶.

Instead of concentrating on low energy ion implantation, we explore intermediate energy ion irradiation to control defect densities in MoS₂. While low energy (typically <1 keV) ion implantation provides an exciting route to alter material properties, nuclear energy deposition

induces defects that can dominate the material response. In the case of intermediate energies (range from a few hundred keV to several MeV), electronic energy deposition starts to play an important role in the modification of optical properties of these low dimensional materials. We utilize ion irradiation to controllably introduce defects and investigate the change in photoluminescence (PL) spectra of MoS₂ and elucidate the role defects play in controlling the optical response of the material.

II. MATERIALS AND METHODS

Experimental Details

Molybdenum disulfide ultrafine powder with average particle size of ~90 nm and specific surface area of ~35 m²/g was purchased from the Graphene Supermarket²⁷. Prior to irradiation studies, the quality of MoS₂ was ascertained in FEI Tecnai F20 scanning/transmission electron microscope (S/TEM) operated at an accelerating voltage of 80 kV. Figure 1(a) provides a representative overview of an individual MoS₂ flake, with an insert showing its crystallographic structure, consistent with 2H-MoS₂. Figure 1(b) shows a high-resolution micrograph of the MoS₂ atomic structure, which is in good agreement with the structure predicted by density-functional theory (DFT) calculations provided in Figure 1(d).

For irradiation studies, MoS₂ powder was placed into a 20 mL vial, submerged in an anhydrous N, N-dimethylformamide (99.8% purity) from Sigma Aldrich²⁸ at a concentration of 20 mg/mL, and sonicated for 3.5 hours. After completion of the sonication, flakes were transferred to a Si (001) substrate with 300 nm thickness of SiO₂. Figure 1(c) is a representative example of the flakes that were obtained using this methodology and shows an optical micrograph of large area

pristine MoS₂ flake on Si/SiO₂ substrate. Note that the flakes obtained using dispersion/sonication techniques varied in thickness and contained mono-, bi-, and tri-layers of MoS₂.

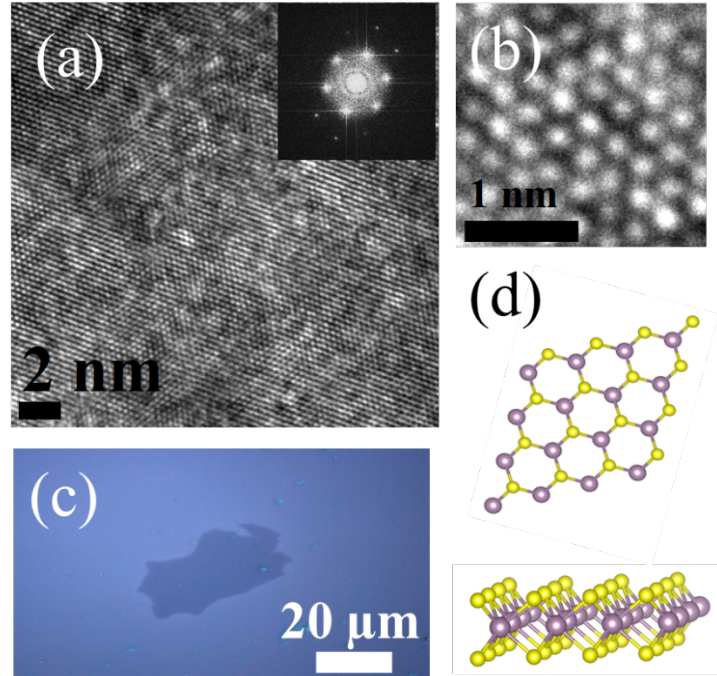


FIG. 1. Overview of the MoS₂ samples used in this study: (a) TEM micrograph of the MoS₂ flake, where the inset reflects the 2H phase of the material; (b) high-resolution TEM micrograph of the atomic structure; (c) optical micrograph of large area pristine MoS₂ on Si/SiO₂ substrate; and (d) top view and side view of the atomic structure of monolayer MoS₂, where Mo atoms are depicted in purple and S atoms in yellow.

Specimens were subjected to 3 MeV Au²⁺ ion irradiation at room temperature using a 1.7 MV General Ionex tandem accelerator. The ion fluence ranged from 1×10^{12} to 1×10^{16} cm⁻². The target chamber pressure during the irradiation was at about 6×10^{-8} torr. Photoluminescence spectroscopy was performed on a Horiba Jobin-Yvon LabRam ARAMIS microRaman with 1800 g/mm grating and a cooled CCD detector. The excitation laser used was 632.8 nm (1.96 eV) HeNe, processed at 298 K (+/- 2 K). The laser power on the sample was about 130 μW, which has been

proven to have a good signal-to-noise ratio on monolayer materials²⁹. To avoid the local heating effect that may be induced by the laser, we used a large density filter (50% transparency), analyzed our sample for any visible damage, and checked the intensities of the peaks in the spectra, as decreasing intensity peaks signify heating damage to the material.

Computational Details

We computed defect formation energies using density functional theory (DFT) as implemented in the Vienna ab initio simulation package (VASP)³⁰. We performed calculations using projector-augmented wave potentials^{31,32} and treat the exchange-correlation using the Perdew-Burke-Ernzerhof (PBE)³³ generalized gradient approximation functional. A plane wave cutoff energy of 520 eV, and Methfessel-Paxton smearing³⁴ with smearing energy width of 0.10 eV and Γ -centered Monkhorst-Pack k -point meshes³⁵ were employed for Brillion zone integration. Defect calculations were performed in a 4×4 supercell, with 20 Å vacuum spacing between layers, and a 3×3 k -point mesh.

The formation energy $E^f[X^q]$ of a point defect X with charge q is determined by DFT calculations using a supercell approach following:

$$E^f[X^q] = E_{\text{tot}}[X^q] - E_{\text{tot}}[\text{pristine}] - \sum_i n_i \mu_i + qE_F + E_{\text{corr}},$$

where $E_{\text{tot}}[X^q]$ and $E_{\text{tot}}[\text{pristine}]$ are the total DFT-derived energies of the supercell containing the defect X and the pristine supercell respectively, n_i is the number of atoms of species i added/removed, μ_i is the corresponding chemical potential of the species, and E_F is the Fermi energy. The final term E_{corr} corrects for the artefacts introduced by treating charged defects in the periodic supercell approach, which was evaluated using the method developed by Freysoldt and Neugebauer³⁶.

III. RESULTS AND DISCUSSION

Photoexcitation in semiconducting MoS₂ is expected to produce pairs of bound electrons and holes known as excitons, the recombination of which emits a photon observed as PL. Photogenerated electron-hole pairs can have strong Coulomb interaction leading to the formation of bound excitons and can alter the optical transitions in these systems. The spatial confinement combined with the reduced dielectric screening as compared to bulk crystals suggests that the optical response of low-dimensional MoS₂ flakes will be dominated by neutral excitons.

Previous studies³⁷⁻⁴¹ indicated that the spectrum of bare MoS₂ should display two distinctive peaks: A_{1s} exciton peak between 1.86-1.92 eV and B_{1s} exciton peak within the range of 2.0-2.1 eV, which are associated with direct optical transitions to the lowest conduction bands from the highest spin-orbit valence bands⁴²⁻⁴⁴. It should be noted that both A_{1s} and B_{1s} exciton peaks are formed by an electron and a hole in valence band. The ground state A_{1s} exciton has a smaller PL energy than the B_{1s} exciton, which is associated with the spin orbit split valence band. Other possible spectral features in MoS₂ include the A⁻ trion peak arising from interactions between two electrons and a hole that can occur when excess electrons bind to photoexcited electron-hole pairs and create positively charged states³⁸⁻⁴⁰.

As seen from Figure 2, the PL spectrum of the unirradiated sample exhibits an A_{1s} exciton peak at 1.91 eV and a B_{1s} exciton peak at 2.09 eV. The inset in the right bottom corner is a schematic representation of the electron and hole with respect to Fermi level (E_F) in an exciton. The spectral position of the peak shifts upon irradiation and opposing trends were noted for the A_{1s} and B_{1s} exciton peaks: The A_{1s} exciton peak shifts towards higher energies, while B_{1s} towards

lower energies with increasing ion fluence. Note that Figure 2 discusses only exciton peaks and data corresponding to the trion peak will be provided later in the manuscript.

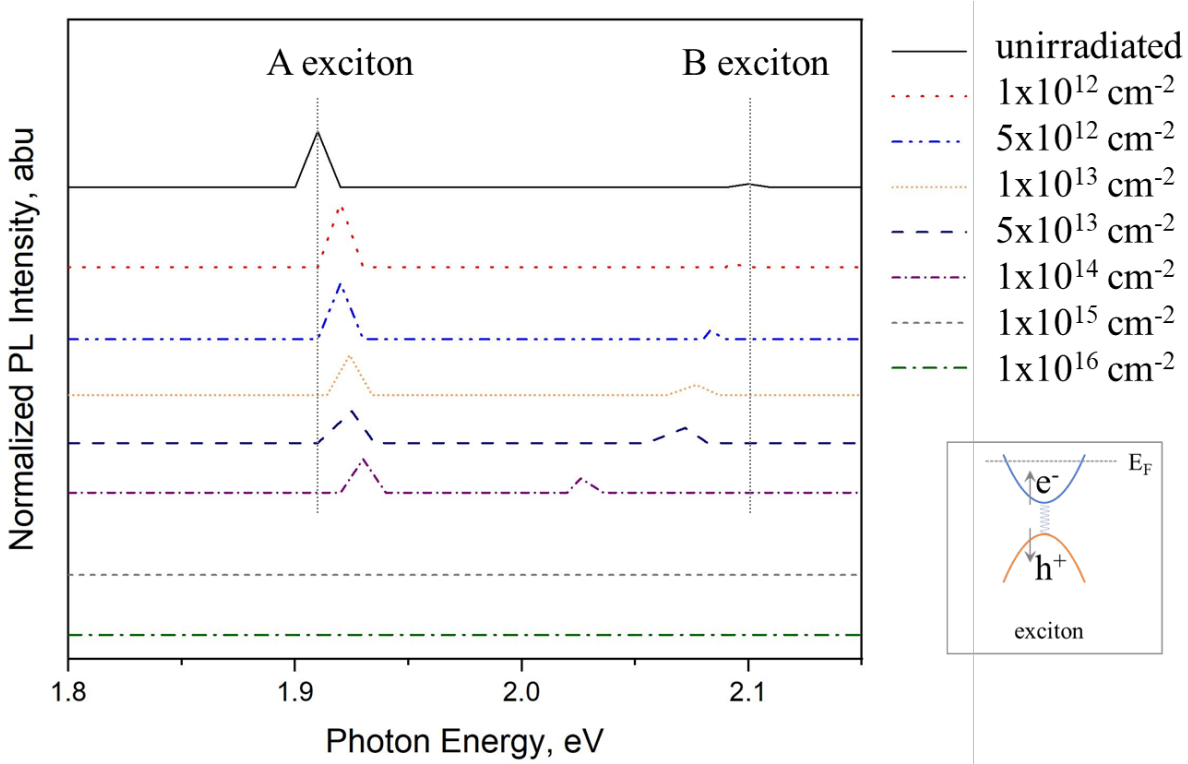


FIG. 2. Photoluminescence of unirradiated MoS_2 specimens and those irradiated to various fluences with 3 MeV Au ions. Here the peak appearing around 1.91 eV corresponds to A excitons and peak around 2.09 eV to B excitons. The vertical lines are provided for reader's convenience to illustrate the shift in peaks as a function of increasing ion fluence. The inset is a schematic representation of the location of electron and hole in respect to Fermi energy level in exciton.

In the past, optical response of TMDs has been changed using applied bias and femtosecond light pulses that introduced large number of carriers and increased screening of the Coulomb interaction^{7,45}. Generally, using a p-type dopants caused a shift towards higher energies in excitonic peaks while n-type dopant in a shift towards lower energies in excitonic peaks^{46,47}. Note that the underlying mechanisms governing doping and irradiation are different. Energetic Au

ions transfer energy to the material and initiate the displacement of atoms from their lattice sites. Ballistic collisions between Au ions and target atoms (Mo and S) lead to the formation of damage cascades that break the periodicity of the lattice. Furthermore, inelastic collisions can excite electrons, ionize target atoms, and result in short-lived electronic excitations. Irradiation can produce various defect configurations in low-dimensional MoS₂, including vacancies, adatoms, vacancy clusters, grain boundaries with complex dislocation cores, and interfaces with lattice mismatch. DFT calculations indicate that vacancies and vacancy clusters are the most abundant point defect type in MoS₂, which can include single sulfur vacancy (V_s) and double sulfur vacancy (V_{s2}), alongside less energetically favorable antisite defects where Mo occupies V_{s2} (Mo_{s2}), and a pair of S atoms occupying Mo position (S_{2Mo})⁴⁸.

The number of introduced defects is expected to progressively increase with increasing ion fluence. At higher ion fluences ($\geq 1 \times 10^{15} \text{ cm}^{-2}$), the accumulated damage eliminates both exciton peaks in MoS₂ as can be seen in Figure 2(a). To further illustrate this trend, Figure 3(a) plots both A_{1s} and B_{1s} exciton peaks as a function of increasing ion fluence. As can be seen from Figure 3(a), exciton peaks are getting closer, which generally indicates a reduction in the spin-orbit coupling in MoS₂, but additional work is needed to confirm this observation. Figure 3(b) plots the B/A exciton ratios as a function of increasing fluence.

In the past, the B/A exciton ratio has been used to estimate the defect density of pristine MoS₂ flakes. High-quality MoS₂ monolayers grown using chemical vapor deposition method (CVD) were shown to have B/A ratios between 0.01-0.03⁴³. The B/A ratio of unirradiated MoS₂ used in this study was 0.03, which is comparable to that of CVD grown MoS₂ and indicates good initial sample quality. Upon irradiation to the lowest fluence of $1 \times 10^{12} \text{ cm}^{-2}$, the B/A ratio doubles (0.06), and reaches 0.52 at the intermediate fluence of $1 \times 10^{14} \text{ cm}^{-2}$. Thus, up to the fluences of

$1 \times 10^{14} \text{ cm}^{-2}$, the lattice remains crystalline and defect densities in MoS₂ can be controlled, while at higher fluences large number of introduced defects distorts excitonic structure of the material.

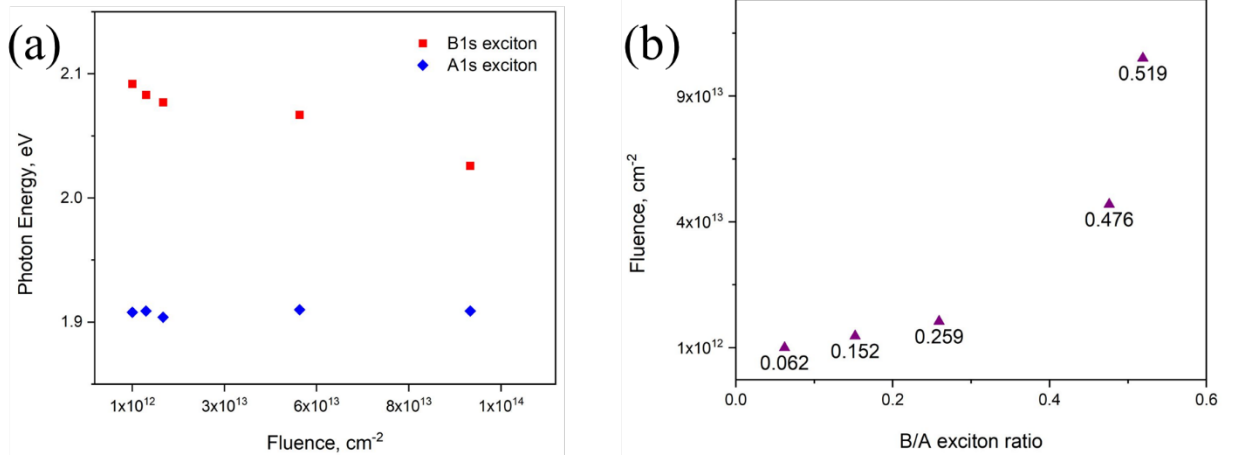


FIG. 3. Relationship between A and B exciton peaks and ion fluence in irradiated MoS₂ samples, where a) plots A1s and B1s exciton peak energies as a function of increasing ion fluence and b) illustrates how B/A exciton ratios change with ion fluence.

Preliminary DFT calculations were conducted in parallel to our experimental work. Theory indicates that point defects, such as sulfur vacancies, are the most common defect type in MoS₂. The relaxation of the structure is minimal upon formation of vacancies, and neutrally charged defects retain trigonal symmetry. DFT calculations of the formation energies of the S vacancy (illustrated in Figure 4(a)) and S divacancy (shown in Figure 4(b)) in monolayer MoS₂ in the Mo-rich/S-poor limit are shown in Figure 4(c) as a function of the Fermi level position. The valence and conduction band edge positions calculated with respect to the vacuum level are indicated by the vertical dotted lines. The slope of the formation energy plot corresponds to the most stable charge state for the defect over that range of Fermi energies.

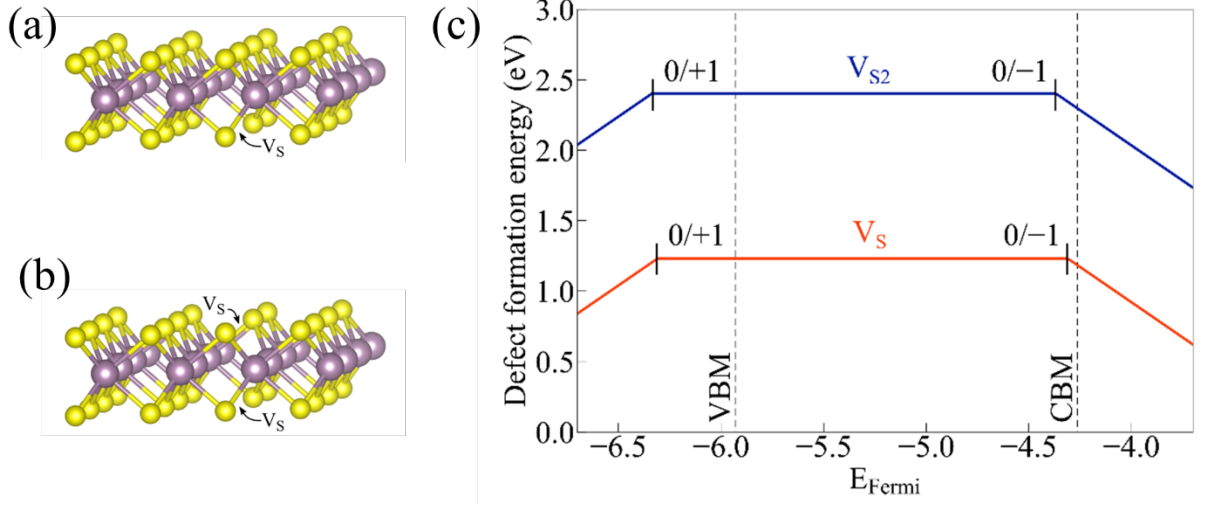


FIG. 4. DFT simulation results showing schematics of (a) S vacancy V_s and (b) S divacancy V_{s2} , where Mo atoms are shown in purple and S in yellow, and (c) formation energies of V_s and V_{s2} in monolayer MoS₂ in the Mo-rich/S-poor limit, as a function of Fermi level position. For both defects, the 0/-1 charge transition level is predicted to be within the band gap and to be associated with a defect level close to the conduction band minimum.

The calculated formation energy of the single S vacancy in the Mo-rich/S-poor limit is about 1.25 eV, which is in good agreement with the values reported in the literature^{4,24,49,50}. The S vacancy formation energy in the S-rich limit is higher, about 3 eV, but even under those conditions it is still predicted to be one of the lowest energy intrinsic point defects. The formation energy of the S divacancy in the Mo-rich/S-poor limit is calculated to be about 2.4 eV, indicating only a slight binding energy between the S vacancies. Our calculations predict that both 0/-1 charge transition levels (CTLs) fall within the band gap and are about 1.55-1.65 eV above the valence band minimum (VBM) in monolayer MoS₂, again in good agreement with other computational studies^{4,24,49,50}. The CTLs indicate the presence of (deep acceptor) defect states in the gap which could trap excitons.

Experimentally, we observe a sharp peak at ~ 1.82 eV in the PL spectra of the sample irradiated to $1 \times 10^{16} \text{ cm}^{-2}$ (see Figure 5(a)) and ~ 1.70 eV in the sample irradiated to $1 \times 10^{15} \text{ cm}^{-2}$ (see Figure 5(b)). These peaks were not observed in samples irradiated to lower fluences and are believed to arise from the valley excitons migrating to defects such as sulfur vacancies, which are formed in higher concentrations under higher irradiation fluences. Excitons have been known to migrate to defect bound states in monolayer MoS_2 , and the PL peaks (at 1.70 and 1.82 eV) have been previously verified to be reflective of a defect bound state^{51,52}. With DFT predicting that V_s and V_{s2} are the most favorable defect types that form in MoS_2 , it is likely that the observed PL peaks are associated with these defects. DFT predicts that the defect states introduced by S vacancies and divacancies are towards the top of the band gap, which is consistent with the energies of the observed PL peaks. However, more detailed calculations at higher level of theory are needed to obtain a better quantitative comparison between experiments and theory.

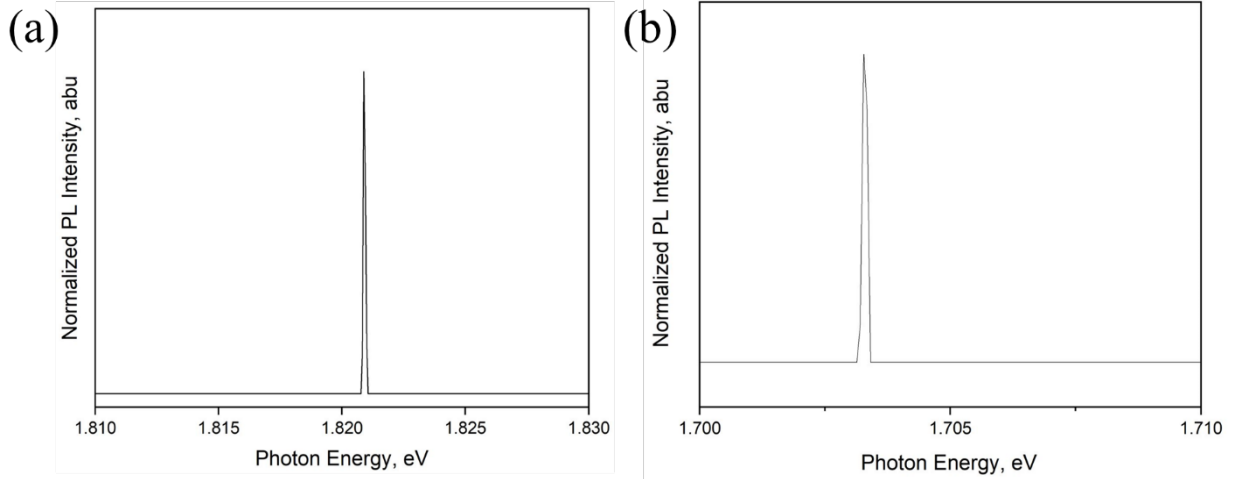


FIG. 5. PL peaks observed in samples irradiated with 3 MeV Au ions to fluences of a) $1 \times 10^{16} \text{ cm}^{-2}$ and b) $1 \times 10^{15} \text{ cm}^{-2}$.

However, it is possible that the sample irradiated to the highest fluence has amorphized and the observed line could be reflective of the crystalline-to-amorphous transition. While the free-standing specimens irradiated to the same fluence were determined to be crystalline in nature through transmission electron microscopy imaging, specimens on the substrate cannot be examined in the same fashion. Thus, we cannot rule out the possibility of the amorphization of MoS₂ flakes at the highest irradiation fluence due to extensive collision cascades in the substrate that introduce additional damage to MoS₂. Further studies will be conducted to ascertain the crystalline state of the specimen and confirm that the observed peaks indeed correspond to V_s and V_{s2}.

While initial assessment is promising, more work needs to be done to match experimental results with theoretical predictions. Theory agrees that the A_{1s} and B_{1s} excitons should be within the energy range determined experimentally in this work. Nonetheless, DFT has limitations such as underestimation of the bandgap and not considering exciton binding in these systems. DFT calculations can predict defect formation energies and the effect these defects can have on optical response of the material; they are difficult to match experimentally. For instance, due to the limitations of standard ground state DFT with semi-local functionals, we are unable to make quantitative predictions of the energy of such transitions for direct comparison with the experimental results. Thus, further work needs to be done to achieve better integration between experiments and modeling.

Beyond controlling neutral excitons, defects can bind photoexcited electron-hole pairs and transfer the exciton spectral weight to trions. We observe the photoemission splitting between bound and free trions, as can be seen from Figure 6(a), where bound trions are illustrated in blue and the free trion in red. The inset in Figure 6(a) depicts the position of electrons and holes with

respect to E_F in a trion. Figure 6(b) highlights the difference between bound and free trions in MoS₂. In the sample irradiated to the fluence of $1 \times 10^{15} \text{ cm}^{-2}$, both free and bound trions are present within the same PL spectrum. Note that the lifetimes of the states are much longer than the trion/exciton lifetimes, as PL spectra of irradiated MoS₂ samples remains unchanged weeks after irradiation experiments.

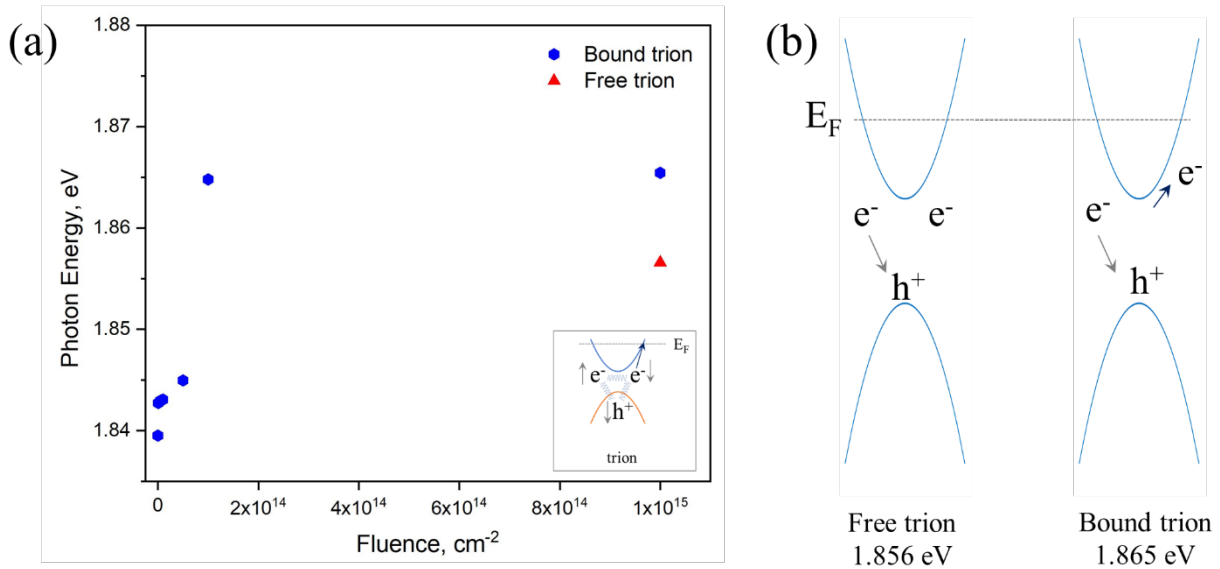


FIG. 6. (a) Dependence on ion fluence of the trion peaks and (b) corresponding schematic illustration of the bound and free trions in MoS₂. The inset in (a) is a schematic representation of the location of electrons and a hole in respect to Fermi energy level in trion.

It is known that 2H-MoS₂ breaks inversion symmetry in singular layer but restores it in even number of layers^{40, 53}. The lack of inversion symmetry in monolayer MoS₂ (that arises due to the coupling of valley and spin degrees) allows manipulation of the single valley using defects. Furthermore, bulk MoS₂ is an indirect bandgap material with a bandgap of ~1.2 eV, which transitions to a direct bandgap semiconductor with an optical bandgap of ~1.87 eV when thinned

to its monolayer form⁵⁴. In this study, we did not fabricate monolayer MoS₂ and presented data averages over the response of mono-, bi-, and tri-layered flakes, which highlights the versatility of our approach. In the future, the response of mono- and bi-layer flakes to irradiation will be investigated individually to determine if there is a correlation between defect-bound states and number of layers.

Since atomic structure of MoS₂ is trigonal prismatic with hexagonal lattice occupied by one Mo and two S atoms, both highest valence bands and lowest conduction bands of MoS₂ primarily form from Mo d-orbitals^{55,56}. Previous theoretical calculations indicated that strongly localized Mo d-orbitals are primary contributors to the conduction band states at the K point, which have minimal interlayer coupling³⁷. However, linear combination of Mo d-orbitals with antibonding S p_z-orbitals near the Γ point can lead to strong interlayer coupling³⁷. The defects introduced in the course of irradiation can reside between layers and alter d-electron physics. Thus, our work demonstrates the feasibility of engineering novel optical behaviors in these materials using defects, which were previously thought to be detrimental to device performance.

IV. CONCLUSION

In this study, the value of defect engineering in tailoring optical response of low dimensional MoS₂ flakes was demonstrated. We conducted a systematic study of the scientific phenomena that define optical properties of MoS₂ and attempted to establish a correlation between optical properties and defects. While previous studies indicate that intrinsic defects and other structural imperfections are not necessarily detrimental to the properties of the material, the role of extrinsic defects in defining the properties of MoS₂ was not clear. We illustrate that irradiation-induced defects provide benefits to optical properties of these two-dimensional materials and can

be used to bind excitons and trions to enhance device performance and enable unprecedented functionalities. With the study of structural defects in two-dimensional materials still in its infancy, additional work should be conducted in the future to ensure that the gap between experiments and computational work is bridged and has a sound foundation.

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