

High-Speed Direct Writing of MoSe₂ by Maskless and Gas-Free Laser-Assisted Selenization Process

Yoonsoo Rho,^{\$} Healin Im,^{\$} Letian Wang, Matthew Eliceiri, Brian Blankenship, Sunkook Kim,* and Costas P. Grigoropoulos*

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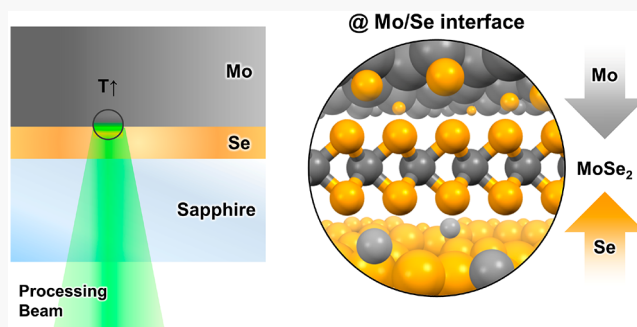
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ABSTRACT: Transition metal dichalcogenides (TMDCs) have shown exceptional optoelectronic properties that can potentially substitute conventional silicon-based devices and be utilized in sensors and energy devices. To exploit their wide array of potential applications, it is necessary to develop methods capable of on demand, location selective, and tunable formation of structures of arbitrary shape. Here, we demonstrated high-speed direct writing of MoSe₂ by laser-induced selenization process in vacuum or ambient environment. Laser irradiation on predeposited Se/Mo multilayer promptly forms polycrystalline MoSe₂ in a site-selective manner without use of photomask and toxic gas. *In situ* reflectance measurement and temperature simulation identified distinct characteristic processing stages, revealing that the time required for synthesizing ~20 nm-thick MoSe₂ polycrystal film is of the order of 10⁻³ s, which is significantly faster than the conventional selenization process using furnace annealing. We believe that the laser synthesis of MoSe₂ demonstrated in this study is of general applicability and therefore offers a simple and straightforward route for obtaining arbitrary patterns in TMDCs.



INTRODUCTION

TMDCs have attracted significant interest due to their outstanding optoelectronic properties, such as relatively large band gap (1.2–1.9 eV) with moderate mobility at room temperature (up to ~200 cm²/V·s),^{1–3} as well as rich exciton dynamics when reduced to mono- or few-layers.⁴ In addition, their mechanical flexibility and highly tunable properties are advantageous for high-performance flexible electronics.^{5,6} Such promising properties are attributed to the multiple van der Waals stacked atomically thin layers that are terminated by covalently bonded atoms and relatively free of surface defects. In turn, TMDCs are often recalled as alternatives to silicon-based nanoscale electronics that suffer from intrinsic surface defects that significantly impair charge transport performances.⁷

Although TMDCs are highly promising, technological developments in defining their shape and location still experience bottlenecks that hinder efforts to substitute conventional bulk materials. As a bottom-up approach, it has been shown that chemical vapor deposition (CVD) can form mono- or few- layer TMDCs in large area, followed by shape definition using a mask-based CMOS process.^{8,9} However, this approach generated random seeding, as the initial nucleation site affected the continuity and imposed time-consuming lateral growth processes. In this regard, sulfurization or selenization of predefined metal films (i.e., molybdenum

(Mo) or Tungsten (W)) may offer a rather straightforward route to form TMDCs by controlling solely chalcogen gas (i.e., Sulfur (S) or Selenium (Se)).^{10–17} On the basis of this method, synthesis of mono- or few- layer MoS₂,^{11,13–15} WS₂,^{12,15,16} and WSe₂¹⁷ have been demonstrated. However, those methods require prolonged heating and cooling time under toxic gas flow, which is not applicable to sensitive and flexible substrates. Furthermore, it requires photolithography steps to define the shape and location of the TMDCs.

Laser has been considered as the most feasible means to supply local energy for materials modification by providing photons in temporally and spatially controlled manner.^{18,19} Thus, without using wet photolithography processes, arbitrary shapes can be readily produced rapidly and site-selectively. In this regard, laser has been implemented as a clean and precise processing tool for the local thinning,^{20,21} patterning, and doping²² of TMDCs. Despite its advantages, synthesis of TMDCs by laser has not been introduced yet, which is partly ascribed to the difficulty in controlling sensitive stoichiometry

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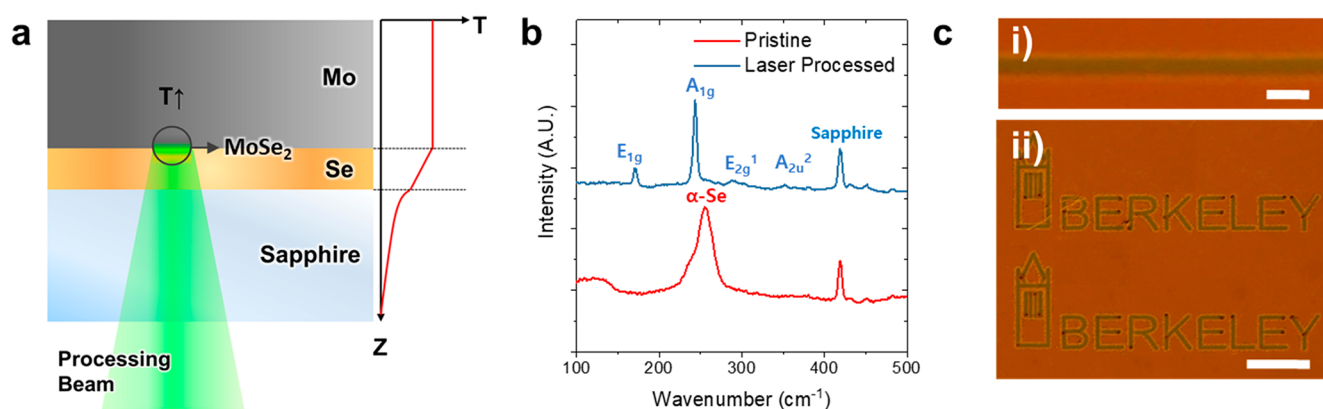


Figure 1. (a) Schematic of the laser-induced MoSe₂ synthesis in sapphire/Se/Mo multilayer. The laser beam was irradiated through the transparent sapphire substrate, and MoSe₂ was formed at Se/Mo interface. The temperature profile shows linear drops in the Se layer, exponential drops in sapphire, and almost no change in the Mo layer. (b) Raman spectra obtained from the pristine and laser-processed area. (c) Optical microscope images obtained through the sapphire substrate show (i) simple line and (ii) arbitrary patterns. The scale bars in panels (i) and (ii) indicate 10 and 100 μm , respectively.

in case of CVD growth by the locally varying laser-induced temperature.

Here, we demonstrate direct writing of MoSe₂ based on Mo and Se multilayers by laser-induced selenization process. Raman, X-ray photoelectron spectroscopy (XPS), and transmission electron microscopy (TEM) analyses confirmed the formation of polycrystalline MoSe₂ at the interface between Mo and Se layers. We interrogated the laser processing parameters for MoSe₂ synthesis and identified distinct characteristic synthesis stages by applying *in situ* reflectance measurement and temperature simulation. The time required for the formation of a 20 nm-thick layer of MoSe₂ was found to be of the order of 10^{-3} s, which is much shorter than that of CVD and sulfurization or selenization by furnace annealing. Thus, our new approach allows fast formation of arbitrary TMDC patterns in microscale resolutions for various future potential applications.

METHODS

Preparation of Se-Mo Films on Sapphire Substrate. A Se layer (25 nm) was deposited on a sapphire substrate using RF sputtering under 100 sccm of Ar flow, and the RF power was set as 15 W. The pressure was 10 mTorr during the deposition, whereas the base pressure was 4.0×10^{-6} Torr. During the deposition of the Mo layer (100 nm), the RF power was 150 W under the same Ar flow rate. The pressure during the deposition was 10 mTorr.

Laser-Induced Selenization Process. A CW 532 nm laser (Sprout, 5W) was focused through a 10X objective lens (NA $\sim$ 0.28). The sample was placed in a stainless-steel chamber equipped with a quartz window for optical access. A dichroic mirror placed between the objective lens and zoom lens allowed optical monitoring during the processing. For vacuum processing, the pressure was maintained below 1 mTorr. For preparation of the sample for the Raman, XPS, and TEM analyses and *ex-situ* reflectance measurement, the laser-induced selenization process was conducted under a vacuum, while we conducted the *in situ* reflectance measurement at ambient environment conditions.

Characterization. A CW 532 nm laser focused by a 50X objective lens and 1800 cm^{-1} grating was used for Raman measurement (inVia, Renishaw), and the power was maintained below 0.5 mW to avoid any damage to the sample.

XPS depth profiling was performed using an XPS spectrometer (K-Alpha+, ThermoFisher Scientific. Co.). The XPS spectra were calibrated at the C 1s peak of 285.0 eV. Cross-sectional TEM and STEM images were obtained using analytic STEM (JEM-2100F, JEOL Co.).

***In Situ* reflectance measurement.** *In situ* reflectance measurement was performed at ambient condition. A He–Ne laser (wavelength: 633 nm) was coupled as a probing beam. A polarized beam splitter (PBS) coaxially combines the probing beam to the processing beam (532 nm) before the objective lens. A beam expander was used to expand the probing beam to adjust the focal spot size of the 633 nm beam to be smaller than the focal spot of the processing beam. A quarter wave plate located between the PBS and the objective lens performs half-wave retardation when the probing beam passes through it twice, allowing it to pass through the PBS directed to the avalanche photo diode (APD) after color filters that block the probing beam. The processing beam was modulated using an acoustic optical modulator (AOM) such that the rise and decay times of laser beam irradiation are controlled within 2 μs . The beam splitter split a fraction of the processing beam directly to a photodiode to trigger an oscilloscope that recorded the *in situ* change in the 633 nm power registered at the APD. The schematic diagram is included in the [Supporting Information](#).

Heat Transfer Simulation. The details are described in the [Supporting Information](#).

RESULTS AND DISCUSSION

Amorphous Se (a-Se) (25 nm) and Mo (100 nm) layers were deposited on a sapphire substrate by radio frequency (RF) sputtering. The thicknesses of deposited a-Se and Mo layers were determined by AFM and SEM cross-sectional analysis. An as-prepared Se/Mo multilayer sample was placed in a processing chamber equipped with quartz window allowing laser-assisted synthesis of MoSe₂ under vacuum or ambient conditions. A continuous wave (CW) laser beam (λ = 532 nm) was irradiated through the transparent sapphire substrate (Figure 1a) and focused via an objective lens (numerical aperture (NA) $\sim$ 0.28) resulting in a spot size of $\sim$ 5.4 μm ($1/e^2$) at Se/Mo interface. This processing scheme allowed the *in situ* monitoring of MoSe₂ synthesis, whereas the opaque Mo layer prevented optical inspection. The laser fluence impinging

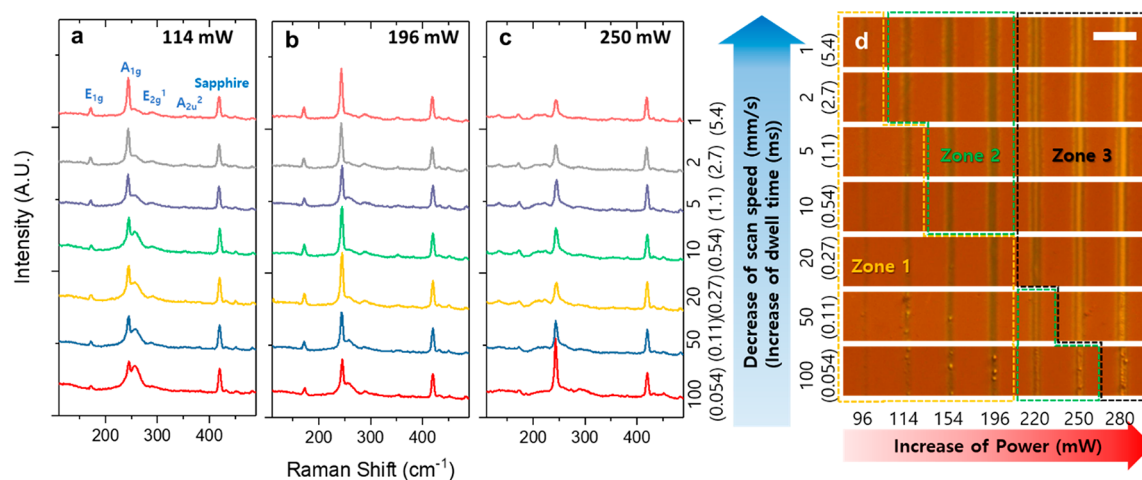


Figure 2. Raman spectra and optical microscope images obtained by varying laser power and scan speed at vacuum condition show distinct characteristic processing stages. Raman spectra at a fixed laser power of (a) 114 mW, (b) 196 mW, and (c) 250 mW for varying laser scan speeds (mm/s) (dwell time (ms)). (d) Laser fluence (mW) with respect to scan speed (mm/s) (dwell time (ms)) diagram consisting of optical microscope images of synthesized MoSe₂ lines. The dashed boxes explicitly show the distinct processing regions, zone 1 (yellow), zone 2 (green), and zone 3 (black). The scale bar indicates 20 μ m.

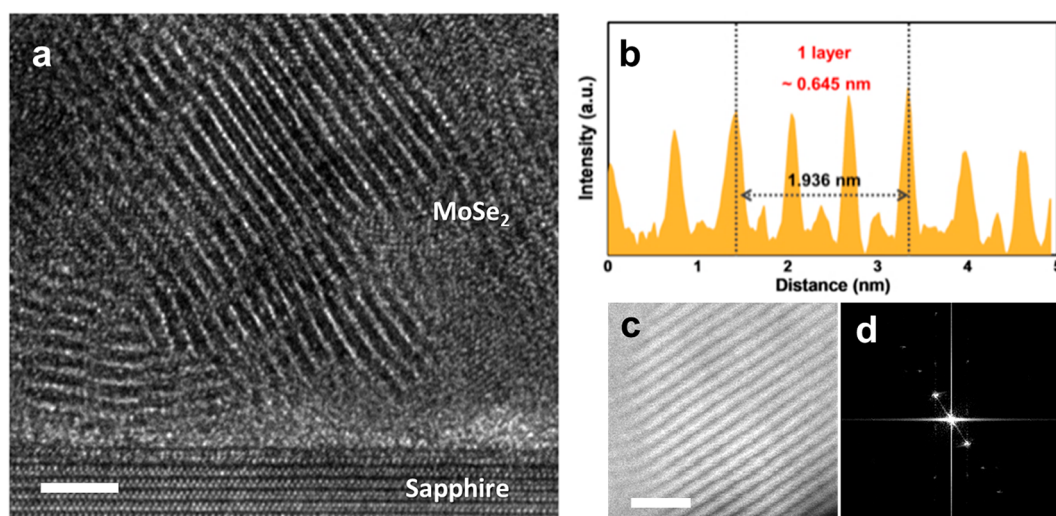


Figure 3. (a) Cross-sectional TEM image taken from the interface between sapphire and MoSe₂ after laser processing using a laser fluence of 196 mW at a scan speed (dwell time) of 2 mm/s (2.7 ms), corresponding to zone 2. (b) Line profiles of interlayer spacing show 6.45 Å between layers. (c) STEM and (d) FFT images obtained from the MoSe₂ region. The scale bars indicate 3 nm.

on the sample was mostly absorbed by the Mo-layer due to its large extinction coefficient ($k \sim 7.18$), while Se and sapphire exhibit orders of magnitude smaller values ($k \sim 0.106$ and $k \sim 0$, respectively). Despite its thinness, the Se layer possesses a low thermal conductance of 21 MW/m²·K ($\sim k/L$, where k and L indicate thermal conductivity (0.519 W/m·K) and thickness (25 nm) of Se layer, respectively). The inefficient heat conduction through the Se layer induced a sharp drop in temperature. Further, we note that the Se layer was sandwiched between the substrate and the Mo layer to mitigate the evaporation of Se during laser processing. In contrast, the Se layer deposited on top of the Mo layer easily evaporated at relatively low laser fluence owing to its low melting and boiling temperatures (494 and 958 K, respectively). After processing using a laser fluence of 196 mW at 2 mm/s of scan speed, the Raman spectra clearly changed as shown in Figure 1b. Raman spectra from pristine MoSe₂ sample showed a broad a-Se peak at 260 cm⁻¹ while

sharp peaks at 170 cm⁻¹, 242 cm⁻¹, 288 cm⁻¹, and 352 cm⁻¹ indicated formation of 2H-MoSe₂, which corresponded to phonon vibration modes at E_{1g}, A_{1g}, E_{2g}¹, and A_{2u}², respectively. Note that all the optical measurements, including Raman spectra and optical microscopy, were collected through the transparent sapphire substrate. Distinct color changes indicating the formation of MoSe₂ were observed in the optical microscope images. The interference in multithinlayer film structure sensitively changed the reflectance spectra depending on the material thickness. Such a site-selective synthesis scheme allowed the formation of MoSe₂ patterns of arbitrary shape (Figure 1c-ii).

Raman spectra were obtained by varying both laser power and scan speed and are shown in Figure 2a–c. Note that the scan speed is connected to the dwell time of the laser beam (v/w , where v and w are the laser spot size and scan speed, respectively). At a laser fluence of 114 mW (Figure 2a), which is close to the threshold of initiating the synthesis of MoSe₂,

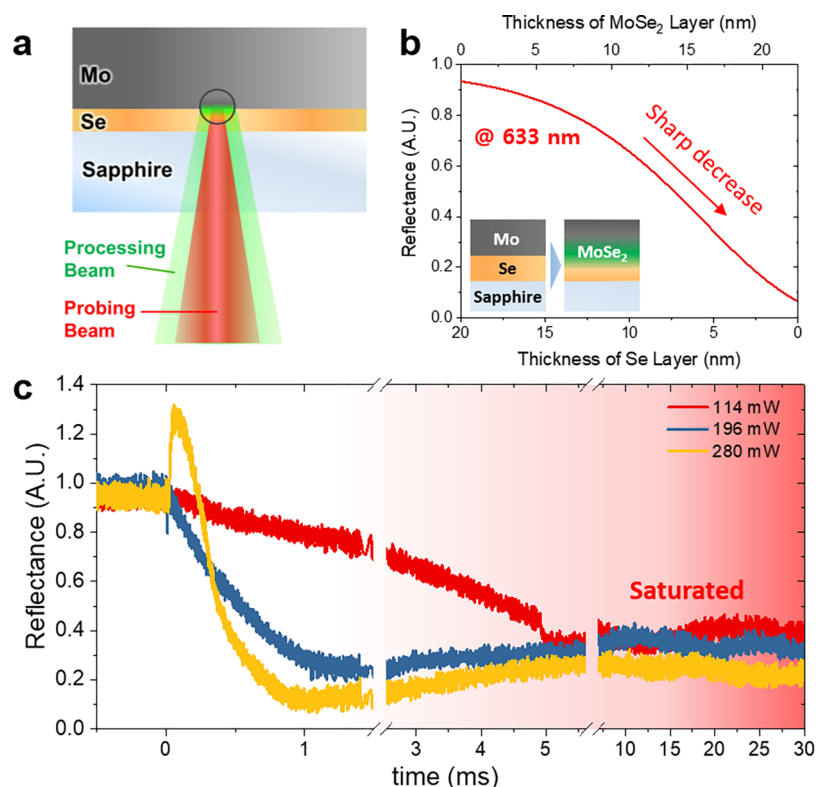


Figure 4. *In situ* reflectance measurement showing the processing time for the saturation of MoSe₂ synthesis. (a) Schematic illustration of the *in situ* reflectance measurement utilizing coaxially coupled probing and processing beams. (b) Calculated reflectance of the Se/MoSe₂/Mo multilayer at a wavelength of 633 nm incident from the sapphire side as a function of the thicknesses of the Se layer and the MoSe₂ layer formed. (c) Measured *in situ* reflectance data depending on the incident processing laser power. The plots correspond to incident laser powers of 114, 196, and 280 mW, indicated by the red, blue, and yellow lines, respectively.

A_{1g} and E_{1g} peaks start to appear at a scan speed (dwell time) of 100 mm/s (0.054 ms), while a-Se peak near 260 cm^{-1} remained prominent. Upon lowering the scan speed (leading to an increased dwell time), the intensity of a-Se peak gradually diminished and was dominated by MoSe₂ peaks to finally disappear at 1 mm/s (5.4 ms). This result indicates that the time required for saturation of the interdiffusion of Mo and Se layers to form MoSe₂ was ~ 5 ms at this processing parameter. Figure 2d compiles the optical microscope images depicting the laser fluence versus scan speed dependence. In this figure, we have marked the characteristic processing regions by dashed lines. Zone 1 indicates the mixture of a-Se and MoSe₂, whereas zone 2 shows saturation of MoSe₂ formation. The gradual color change in the optical microscope images is a manifestation of the decrease in the thickness of Se-layer, accompanying the formation of the MoSe₂ layer. As we increased the laser fluence to 196 mW, the transition from zone 1 to zone 2 occurred faster at ~ 0.5 ms (Figure 2b). At higher laser fluence (230 mW), the a-Se peak immediately disappeared at 100 mm/s (0.054 ms) and the intensity of the MoSe₂ peaks decreased abruptly (Figure 2c). The optical microscope images in zone 3 showed scan lines of bright metallic color, which are ascribed to the Mo surface being exposed by the fast interdiffusion of Mo and Se without the formation of high-quality MoSe₂.

The TEM images show that polycrystalline 2H-MoSe₂ was formed at the interface between Mo and Se layers, as shown in Figure 3. The TEM sample was prepared using a scan speed (dwell time) 2 mm/s (5.4 ms) that induced saturation of the process (zone 2) at 196 mW. We occasionally observed voids

formed between the sapphire substrate and MoSe₂ (Figure S1). We attribute this void formation to the volume shrinkage (12.7%) upon formation of MoSe₂ in case of ideal stoichiometry (i.e., molar ratio; Mo/Se = 1:2) and/or organic contamination introduced during the sample preparation. Nevertheless, a clear 2H phase MoSe₂ structure was observed from the zone 2 sample (Figure 3a) with an interlayer distance (along *c*-axis) of 0.645 Å. These findings were confirmed by TEM image and fast Fourier transform (FFT) pattern (Figure 3b,c). The diffraction pattern from a discrete and ordered dot shows the crystalline orientation along the *c*-axis (Figure 3d). In addition, the grain size was ~ 20 nm (Figure S1), which is in a range similar to the previously reported MoS₂, WS₂, or WSe₂ synthesized via gas phase sulfurization or selenization of transition metal films.^{10–14} Meanwhile, the results obtained from the laser-processed region corresponding to zone 1 (using a laser fluence of 114 mW at a scan speed of 2 mm/s) show smaller grain sizes mixed with a-Se (Figure S2), which can be directly correlated with the Raman spectra of zone 1. In addition, the clear shifts in elemental bonding status can be observed using XPS depth profiling obtained from zone 2 sample (Figure S3). As we collected XPS spectra from the top Mo layer to the sapphire substrate, the Mo 3d_{5/2} and Mo 3d_{3/2} peaks abruptly blue-shifted at the interface between Mo and MoSe₂ layers, whereas the Se 3d peak appeared simultaneously. The blue shift in Mo 3d peaks implies the increased oxidation number of the elements, indicating the formation of Mo–Se bonding.²³ The peak widths were broadened compared to those of pristine 2H-MoSe₂, which were affected

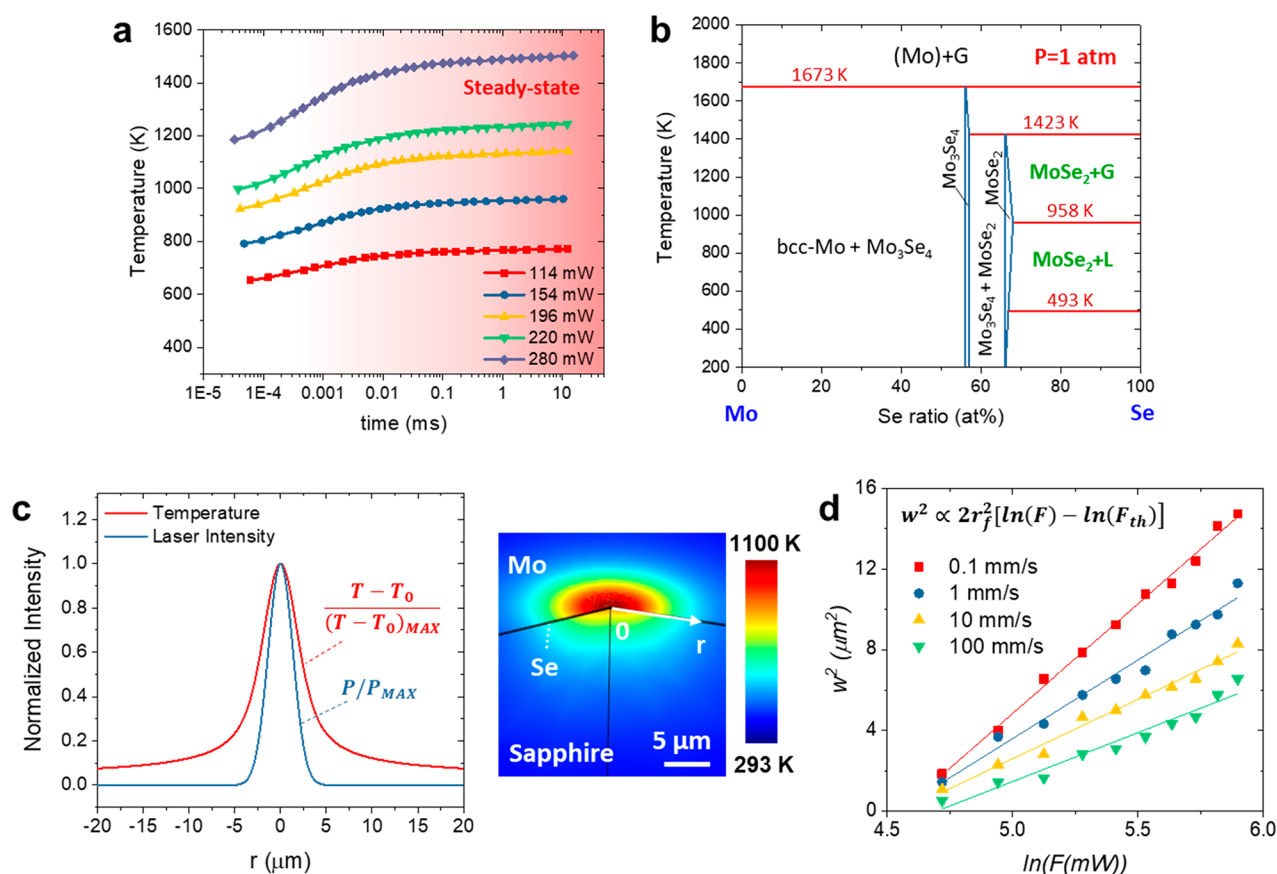


Figure 5. (a) Simulated peak temperatures at the interface between Mo and Se layers for incident laser powers of 112, 154, 196, 220, and 280 mW. (b) Mo–Se temperature composition (T-x) phase diagram at a pressure of 1 atm.²⁷ (c) Normalized temperature and laser power profiles along the *r* direction, as indicated in the right panel. Right panel: Simulated temperature field of the Se/Mo multilayer at a laser fluence of 196 mW after 10 ms. (d) Square of width of the MoSe₂ pattern (w^2) versus the logarithmic value of the laser fluence ($\ln(F)$) obtained from various scan speeds.

by presence of defects and multi grain boundaries of polycrystalline structure.^{24,25}

The processing time required for MoSe₂ synthesis was experimentally probed by *in situ* reflectance measurements at ambient environment conditions. A He–Ne laser beam ($\lambda \sim 633$ nm) coaxially aligned with the processing laser ($\lambda \sim 532$ nm) allowed the real-time probing of the change in reflectance at 633 nm (Figure 4a). Using thin film optics,²⁶ the reflectance of multilayer structure (sapphire/Se/MoSe₂/Mo) at 633 nm can be deduced, depending on the Se thickness and corresponding synthesized MoSe₂ layer as shown in Figure 4b (the details of the derivation are provided in Supporting Information). The reflectance decreased sharply as Se was consumed into MoSe₂, and the ratio of changes in thicknesses of Se and MoSe₂ was calculated as 1.12 (Supporting Information). The *in situ* measured reflectance data at various laser fluences are depicted in Figure 4c. At a low laser fluence (114 mW), the reflectance gradually decreased until ~ 6 ms and saturated. At moderate laser fluence (196 mW), it decreased faster during the earlier stage and until ~ 1 ms. At high fluence (280 mW), the reflectance rose sharply and dropped suddenly. The fast interlayer diffusion exposed the Mo surface without the formation of MoSe₂, thereby increasing the reflectance during the early stage. The subsequent fast oxidation of Mo layer starting from the top surface produced volatile MoO_x, that could be easily released. This process resulted in a sharp drop in the reflectance. The experimental *ex situ* reflectance measurement from the sample prepared at

vacuum condition (Figure S4) showed a similar trend, except for the high laser fluence (280 mW) that was marked by increased and sustained reflectance due to the absence of oxidation of Mo film. On the basis of this result, we deduced that the time scale required for MoSe₂ synthesis was in the range of 1–6 ms, depending on the laser fluence, which was significantly faster than the conventional selenization or sulfurization process using the furnace annealing method ($\sim$ hours).^{10–14}

The high-speed and spatially confined processing was primarily enabled by the rapid temperature change induced by focused laser power. We modeled the temporal and spatial temperature distribution using finite element method (FEM) based on the sapphire/Se/Mo multilayer structure as shown in Figure 5 (details are provided in the Supporting Information). Figure 5a shows the transient temperature at the center of the irradiated spot during laser heating of a nonmoving sample. The temperature profile developed rapidly within 10^{-2} ms, which is much shorter than the time for the saturation of processing (1–6 ms). At the threshold fluence (~ 114 mW) that corresponds to the zone 1, the temperature increased up to ~ 600 K, while at highest fluence (280 mW) (zone 3), it reached up to 1500 K. Figure 5b shows a Mo–Se–temperature–composition (T-x) phase diagram at a pressure of 1 atm.²⁷ The MoSe₂ synthesis is expected to occur within the temperature range of 493–1423 K, where a mixture of MoSe₂ and Se in either gaseous or liquid phase is present atomic ratio of Se exceeding 0.66. In addition to the fast temporal temperature

profile, the lateral profile shows favorable microscale confinement (Figure 5c). We note that the simulation was based on laser power of 196 mW and dwell time of 10 ms. The full width half-maximum (fwhm) of normalized temperature profile ($\frac{T-T_0}{(T-T_0)_{\max}}$ where T_0 is the room temperature) along the r direction in Figure 5a was estimated to be $5.3 \mu\text{m}$, which is slightly larger than the fwhm of laser beam intensity profile of $3.4 \mu\text{m}$ ($\sim \sqrt{\ln 2/2} \cdot w$ where w is $1/e^2$ width, $5.7 \mu\text{m}$). The pattern widths shown in Figure 2d are in a range similar to the fwhm of the temperature profile. Also, for Gaussian-like temperature profile, the width of the processed area (w) is related to the laser fluence (F) by $w^2 \propto 2r_f^2 [\ln(F) - \ln(F_{th})]$, where r_f and F_{th} are the radii of $1/e^2$ laser focal spot ($5.4 \mu\text{m}$) and threshold fluence that can initiate the process, respectively. The equation can correlate the measured pattern width with the laser fluence as shown in Figure 5d. In addition, the intercept of the extrapolated line at $w = 0$ should indicate the F_{th} . As anticipated, this procedure yields identical value of $\sim 95 \text{ mW}$ for various scan speeds.

Despite the fast processing scheme of the order of 10^{-3} s, the kinetics of the laser-assisted selenization can be quantified by the mass diffusion equation using the estimated temperature and measured processing time. The corresponding formula and constants were introduced by Shin et al.²⁸ based on the selenization of thick Mo film ($\sim 3 \mu\text{m}$) by furnace annealing under Se vapor atmosphere. The MoSe_2 thickness (L_{MoSe_2}) can

be formulated by $L_{\text{MoSe}_2} = \sqrt{2 \frac{C_{\text{Se},0}}{C_{\text{Se},1}} D_{\text{MoSe}_2} t}$ where $C_{\text{Se},0}$ and $C_{\text{Se},1}$ are Se atomic densities in fully covered Se layer on Mo surface ($3.7 \times 10^{22} \text{ cm}^{-3}$) and in MoSe_2 crystal ($1.1 \times 10^{22} \text{ cm}^{-3}$), respectively. D_{MoSe_2} , which is the diffusion coefficient of Se in

MoSe_2 , is given by $D_{\text{MoSe}_2} = D_{\text{MoSe}_2,0} \exp\left(-\frac{E_{D_{\text{MoSe}_2}}}{KT}\right)$ where $D_{\text{MoSe}_2,0}$ and $E_{D_{\text{MoSe}_2}}$ are a prefactor ($5.5 \times 10^{-7} \text{ cm}^2/\text{s}$) and the activation energy of MoSe_2 formation (0.68 eV), respectively. In our sample configuration, Se was provided by solid Se layer, which can satisfy the assumption of fully covered Se layer on Mo and MoSe_2 surface. With the substitution of 900 K and 2.7 ms (corresponding to the saturation of processing at zone 2), L_{MoSe_2} was estimated as 12.5 nm, which is similar to the synthesized MoSe_2 as shown in the TEM image. As the diffusion of Se occurs mainly through grain boundaries of MoSe_2 , the nonuniformity of the polycrystals in our case can affect the local D_{MoSe_2} , resulting in uneven thickness of the produced MoSe_2 . Thus, the results indicate that the high-speed laser processing induced MoSe_2 formation, in our study, is consistent with the diffusion kinetics based on bulk films.

CONCLUSIONS

We demonstrated high-speed direct writing MoSe_2 by laser-assisted selenization. Arbitrary patterns of MoSe_2 could be written from a Se/Mo multilayer at vacuum or ambient conditions. Raman, XPS, and TEM analyses revealed the formation of polycrystalline 2H- MoSe_2 at the interface between Mo and Se. On the basis of the *in situ* reflectance measurement and the temperature simulation, we revealed that the high-speed processing was enabled by the rapid temperature increase induced by laser. The diffusion model is shown to be well suited for the interpretation of our results. We believe that our investigation provides useful insight into the

utilization of laser-assisted selenization or sulfurization processes for the location selective synthesis and arbitrary shape patterning of TMDC materials.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jpcc.0c04914>.

Cross-sectional TEM images taken from zone 1 and zone 2; XPS depth profiles of Mo 3d and Se 3d core peaks from zone 2; ex-situ reflectance data; optical microscope images and Raman spectra of MoSe_2 obtained at ambient condition; methods to calculate the reflectance of multilayer thin films; methods to calculate the volume change of films as a result of laser-induced selenization process; details of heat transfer simulation; schematics of *in situ* reflectance measurement set up (PDF)

AUTHOR INFORMATION

Corresponding Authors

Costas P. Grigoropoulos – Laser Thermal Laboratory, Department of Mechanical Engineering, University of California, Berkeley, California 94720, United States; orcid.org/0000-0002-8505-4037; Email: cgrigoro@berkeley.edu

Sunkook Kim – School of Advanced Materials Science and Engineering, Sungkyunkwan University, Suwon-si, Gyeonggi-do 16419, Republic of Korea; orcid.org/0000-0003-1747-4539; Email: seonkuk@skku.edu

Authors

Yoonsoo Rho – Laser Thermal Laboratory, Department of Mechanical Engineering, University of California, Berkeley, California 94720, United States

Healin Im – School of Advanced Materials Science and Engineering, Sungkyunkwan University, Suwon-si, Gyeonggi-do 16419, Republic of Korea

Letian Wang – Laser Thermal Laboratory, Department of Mechanical Engineering, University of California, Berkeley, California 94720, United States

Matthew Eliceiri – Laser Thermal Laboratory, Department of Mechanical Engineering, University of California, Berkeley, California 94720, United States

Brian Blankenship – Laser Thermal Laboratory, Department of Mechanical Engineering, University of California, Berkeley, California 94720, United States

Complete contact information is available at:

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Author Contributions

[§]Both authors contributed equally.

Notes

The authors declare no competing financial interest.

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