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Time-Resolved Changes in Dielectric Constant of Metal Halide Perovskites under Illumination

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Time-Resolved Changes in Dielectric Constant of Metal Halide Perovskites under Illumination

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ABSTRACT: Despite their impressive performance as a solar absorber, much remains unknown on the fundamental properties of metal halide perovskites (MHPs). Their polar nature in particular is an intense area of study, and the relative permittivity (ϵ_r) is a parameter widely used to quantify polarization over a range of different timescales. In this report, we have exploited frequency-dependent time-resolved microwave conductivity (TRMC) to study how ϵ_r of a range of MHPs changes as a function of time, upon optical illumination. Further characterization of charge carriers and polarizability are conducted by femtosecond transient absorption and stimulated Raman spectroscopy. We find that changes in ϵ_r are roughly proportional to photogenerated carrier density, but decay with a shorter time constant than conductance, suggesting that the presence of charge carriers alone do not determine polarization.

Despite being processable from solution at low temperature ($\leq 100^\circ\text{C}$),¹ lab-scale solar cells based on metal halide perovskites (MHPs) exhibit certified power conversion efficiencies (PCEs) comparable to those of monocrystalline silicon.² While MHPs offer industry and the academic community many exciting opportunities, consistent mechanistic understanding and consensus still lag behind more established systems such as organic semiconductors³ or silicon,⁴ with many unanswered questions remaining.⁵

Ionic semiconductors have long been known to exhibit properties distinct from covalent semiconductors,⁶ and the polar nature of MHPs is an intense area of study.^{7–10} The relative permittivity (ϵ_r) is a parameter widely-used to encapsulate polarization over a range of different timescales.^{10,11} As expected for an ionic system, the relative permittivity in MHPs is comparatively large ($\epsilon_r \sim 5 - 50$)¹⁰ when likened to a covalent system, and depends heavily on temperature.¹² A so-called giant dielectric constant has been reported in MHPs, where ϵ_r increases by three orders of magnitude under illumination from 1 sun conditions, at low probe frequencies.¹³ This large dielectric constant has been invoked to explain some of the prominent unresolved issues in the theory of MHPs, such as large polarons, enhanced screening, and defect tolerance.^{14–17}

In this report we use the contactless semiconductor characterization technique, time-resolved microwave conductivity (TRMC),^{18,19} to evaluate photoinduced changes in the real component of relative permittivity ($\Delta\epsilon_r$) as a function of time. TRMC is traditionally employed to evaluate photoinduced changes in the conductance (ΔG) of thin films,¹⁸ crystals,²⁰ fluids,²¹ suspensions,²² powders,²³ or discontinuous films.²⁴ Cavity-based TRMC additionally enables one to probe photoinduced changes in the dielectric constant ($\Delta\epsilon_r$), through changes in resonant frequency.^{25–30} Typically the community casts changes in the real component of dielectric constant into the imaginary component ($\Delta G''$) of complex conductance,²⁸ and interpret it in this context. For simplicity we here just evaluate and consider $\Delta\epsilon_r$. While impedance spectroscopy is the standard technique for evaluating ϵ_r in photovoltaic materials for frequencies up to $\sim\text{MHz}$,³¹ cavity-based TRMC enables one to

probe ϵ_r in the traditionally challenging¹¹ regime of $\sim 10\text{ GHz}$.³² Because TRMC is designed to study the electrical properties as a function of time, t , with close to nanosecond resolution, it allows one to evaluate $\Delta\epsilon_r(t)$, in response to optical stimulation, with similar resolution. TRMC does not require electrical contacts, and hence can eliminate potential ambiguity arising from interfacial phenomena.³³

In this work, we examine two mixed-cation, mixed-halide perovskite compositions: a double-cation composition: $\text{FA}_{0.83}\text{Cs}_{0.17}\text{Pb}(\text{I}_{0.9}\text{Br}_{0.1})_3$ (FACs) and a triple-cation composition: $(\text{FA}_{0.83}\text{MA}_{0.17})_{0.95}\text{Cs}_{0.05}\text{Pb}(\text{I}_{0.9}\text{Br}_{0.1})_3$ (FAMACs), where FA is formamidinium and MA is methylammonium. These compounds are chosen as they are known to lead to high PCE when employed in solar cells,^{34,35} and have been shown to be highly stable in air.^{36,37} **Figure 1** shows as-measured changes in the real components of conductance (ΔG) and relative permittivity ($\Delta\epsilon_r$) of these compounds as a function of time before, during, and after illumination by a nanosecond pulsed laser. Similar data are shown for methylammonium lead iodide (MAPbI_3), formamidinium lead iodide (FAPbI_3), lead iodide (PbI_2), and the organic semiconductor blend of poly(3-hexylthiophene-2,5-diyl) and phenyl- C_{61} -butyric acid methyl ester ($\text{P3HT:PC}_{61}\text{BM}$), in the Supporting Information Figure S1.

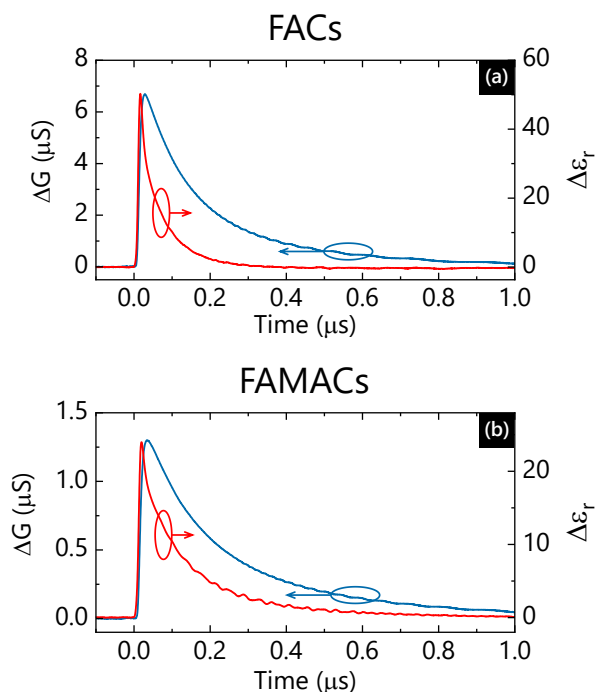


Figure 1 Change in the real component of conductance (left axis, blue line) and relative permittivity (right axis, red line) of (a) FACS and (b) FAMACs, measured as a function of time, before, during, and after illumination by a 532 nm, nanosecond pulsed laser. Incident laser fluence was $\sim 6 \times 10^{13} \text{ cm}^{-2}$ in all cases.

A detailed description of how these parameters were extracted is provided in Supporting Information Section S2, but the technique relies on evaluating changes in the resonant frequency of microwave cavity as a result of optical stimulation. Signals where the peak width is comparable to the response time of our cavity, which is roughly 5 ns, can be interpreted as artefacts of the technique (see Supporting Information Section S4 for more details). We hence conclude that MAPbI₃, FAPbI₃, PbI₂, and P3HT:PC₆₁BM, exhibit no measurable $\Delta \epsilon_r$. The mixed-halide, mixed-cation MHP compounds, however, exhibit a non-negligible $\Delta \epsilon_r$.

While the peak values of ΔG reported here are lower than those reported elsewhere in the literature,³⁸ relative values are generally unambiguous. A number of instrumental and analytical factors are known to affect these extracted values.¹⁹ It is also important to point out that, while all samples were fabricated, transported, and stored in an inert environment, all TRMC measurements were carried out in air. In Section S5 of the Supporting Information we measured ΔG as we flowed N₂ over MAPbI₃ and FAPbI₃ samples, and under atmospheric-pressure air. For these short-duration experiments we did not observe appreciable differences in transport behavior in N₂ compared to air. We however are aware that over long periods of time, exposure to air can significantly affect the properties of MHPs.³⁹

By carrying out a similar analysis as a function of incident laser fluence, we can approximate how $\Delta \epsilon_r$ depends on the carrier concentration (see Supporting Information Section S6 for more details). **Figure 2** shows the peak value of $\Delta \epsilon_r$ as a function of the approximate carrier density in FACS and FAMACs.

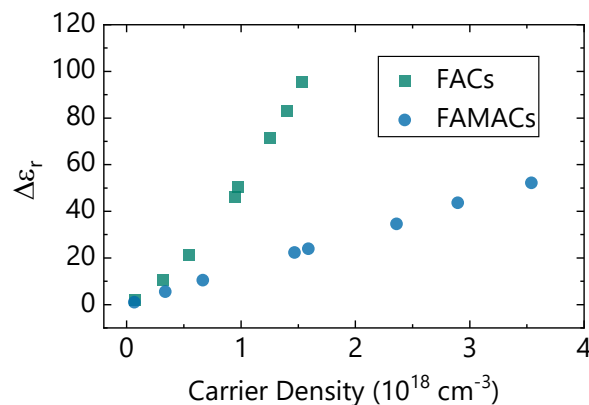


Figure 2 Photoinduced change in the real component of relative permittivity as a function of approximate charge carrier density of FACS and FAMACs.

The origin of the imaginary component of conductance, and the relative magnitude of $\Delta G''$ to ΔG , remains under debate. Field-induced time resolved microwave conductivity (FI-TRMC)^{40,41} has been used to study the nature of charge transport in a range of systems including organic semiconductors²⁹ and graphene⁴² by studying complex conductance. Changes in $\Delta G''$ (and implicitly $\Delta \epsilon_r$) observed in thin films of TiO₂ upon illumination have been associated with trap states,²⁸ invoking Drude⁴³ and Drude-Smith⁴⁴ models to describe the motion of charge carriers in this system. Similar studies into MHPs, however, present a more complicated picture where the molecular motions of A-site cations and polaronic charges contribute to $\Delta G''$.²⁵

The frequency regime we probe in these experiments (~ 10 GHz) is associated with molecular motions of A-site cations.^{32,45} In a previous report we illustrated how the addition of a piperidinium-salt 1-butyl-1-methylpiperidinium tetrafluoroborate ([BMP]⁺[BF₄]⁻) can reduce trap density in FACS films, without significantly modifying the long-range charge carrier mobility.⁴⁶ Here we measured $\Delta \epsilon_r$ against carrier concentration for thin films of FACS with and without the [BMP]⁺[BF₄]⁻ additive (see Supporting Information Figure S8). While we observed a small change in $\Delta \epsilon_r$ as a function of carrier density with and without the additive, we do not consider this difference in $\Delta \epsilon_r$ large enough to be explained by the reduction in the concentration of trap states alone. The difference is much smaller than that exhibited between FACS and FAMACs in **Figure 2**, which are anticipated to have similar trap density. For this reason, $\Delta \epsilon_r$ is here primarily attributed to photoinduced changes in the A-site cation motions, although we acknowledge a number of sources for $\Delta G''$ are possible.

To ensure that there were no systematic changes in measurement or sample conditions over the duration of the experiment, pristine FACS was measured in opposite sweeping directions. The extracted change in $\Delta \epsilon_r$ versus carrier density (see Supporting Information Figure S9) shows only small differences between forward and reverse sweeps.

From our transient data we can evaluate the half-life $\tau_{1/2}$, defined as the time taken for ΔG or $\Delta \epsilon_r$ to reach half of its peak value.^{28,47} **Figure 3** shows $\tau_{1/2}$ extracted as a function of laser fluence for ΔG and $\Delta \epsilon_r$, for FACS and FAMACs. The values of $\tau_{1/2}$ were evaluated from measured transient data, which was

deconvolved with the instrument response function¹⁸ of the apparatus.

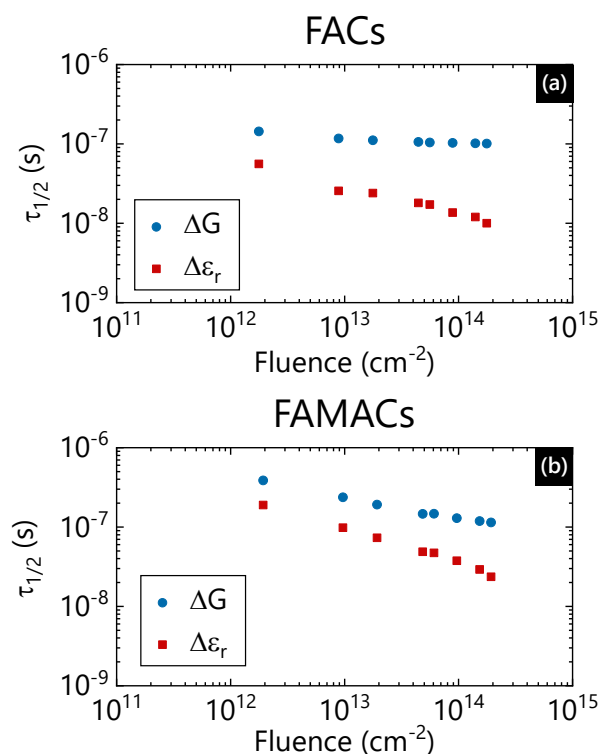


Figure 3 Half-life $\tau_{1/2}$ of photoconductance (ΔG) and change in relative permittivity ($\Delta \epsilon_r$) as a function of fluence for FACs and FAMACs.

Figure 3 shows that $\Delta \epsilon_r$ decays over a different timescale from ΔG for both the FACs and FAMACs samples. This suggests that, while they are clearly correlated changes (see **Figure 2**), the relative permittivity is not simply due to the presence of charge carriers alone. It is important to emphasize here that ΔG and $\Delta \epsilon_r$ have been evaluated from different fit parameters of the same data, meaning that differences in measurement or sample conditions are not responsible.

To help probe the underlying electronic and vibrational features on molecular timescales we employed femtosecond transient absorption (fs-TA) and ground-state femtosecond stimulated Raman spectroscopy (GS-FSRS)^{48,49} on FACs and FAMACs thin-film samples (see Supporting Information Section S9 for details). We tuned the 530 nm photoexcitation pulse energy from 0.01 to 0.15 μJ and collected the fs-TA spectra. Higher pump power corresponds to higher carrier density in a similar range of **Figure 2**, leading to the lengthening of the photobleaching band recovery time due to the commonly invoked “phonon bottleneck” to cool down hot charge carriers up to hundreds of picoseconds timescale (see Supporting Information Section S9 for further discussions).^{50,51} Using global analysis of the fs-TA spectra (see the retrieved lifetimes in Table S3 and evolution-associated difference spectra in Figure S10), we uncovered the time constant (k^{-1}) that is most sensitive to the laser fluence, and plotted the corresponding rate constants for FACs and FAMACs in **Figure 4(a)**. By assuming that carrier thermalization is primarily due to electron-longitudinal optical (LO) phonon scattering we would expect k to decrease with increasing ϵ_r , as described by the electron-phonon scattering

equation (see Section S9).^{50,52} This is consistent with our TRMC data.

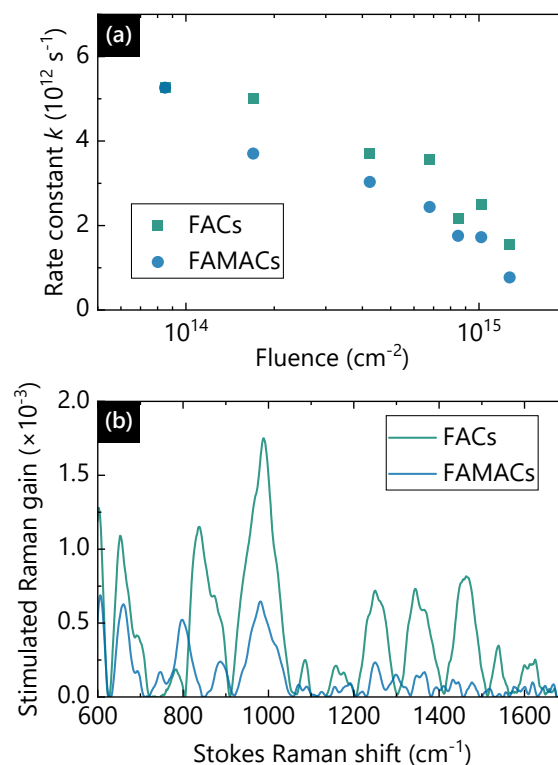


Figure 4 Electronic and vibrational spectroscopic characterization of FACs and FAMACs with ultrafast laser pulses. (a) The second decay rate constant from global analysis of the femtosecond transient absorption spectra of FACs and FAMACs as a function of fluence. The femtosecond pump wavelength is 530 nm. (b) Ground-state femtosecond stimulated Raman spectroscopy (FSRS) of FACs and FAMACs. The picosecond Raman pump is at 800 nm and the femtosecond Raman probe is on the Stokes side.

Notably, GS-FSRS in **Figure 4(b)** manifests larger peak intensities for FACs than FAMACs under similar pre-resonance conditions (Raman pump at 800 nm with 0.4 μJ pulse energy, to the red side of the electronic absorption cutoff wavelength around 750 nm). This result reveals that the A-site organic cations likely have larger electric polarizabilities (hence the Raman peak intensities,^{9,53} e.g., $\sim 1000 \text{ cm}^{-1}$ peak attributed to the C–N stretch) in FACs than FAMACs. In other words, it can be concluded that the presence of a small amount of MA effectively reduces the polarizability of MHPs.

In conclusion, using frequency-dependent TRMC, we have studied how the $\sim 10 \text{ GHz}$ dielectric constant (ϵ_r) of mixed-halide, mixed-cation MHPs exhibit notable changes under illumination. The changes in ϵ_r are roughly proportional to photogenerated carrier density, corroborated by the pump-dependent fs-TA experiments. Interestingly, the timescales involved in such changes in conductance are distinct from those associated with changes in ϵ_r . The dynamics of molecular cations were further characterized by vibration-sensitive FSRS measurements, revealing a larger polarizability of FACs than FAMACs, and suggesting that a small amount of MA effectively reduces the polarizability of MHPs. The correlated TRMC and ultrafast spec-

troscopic results reported in this work could help the community better understand the relationships between optical illumination and polarizability in this class of materials.

ASSOCIATED CONTENT

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/xxx>.

Experimental methods, analysis technique for frequency-dependent time-resolved microwave conductivity (TRMC), cavity response time, deconvolution of TRMC signals, measurements under nitrogen, evaluation of carrier density, TRMC data of the additive-treated sample, analysis of measurement-to-measurement variation, and fs-TA spectroscopy with global analysis.

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Notes

The authors declare no competing interests.

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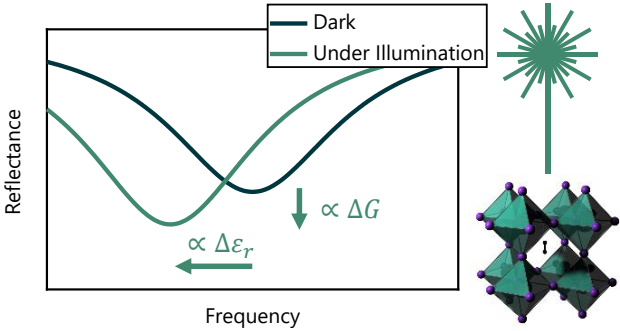
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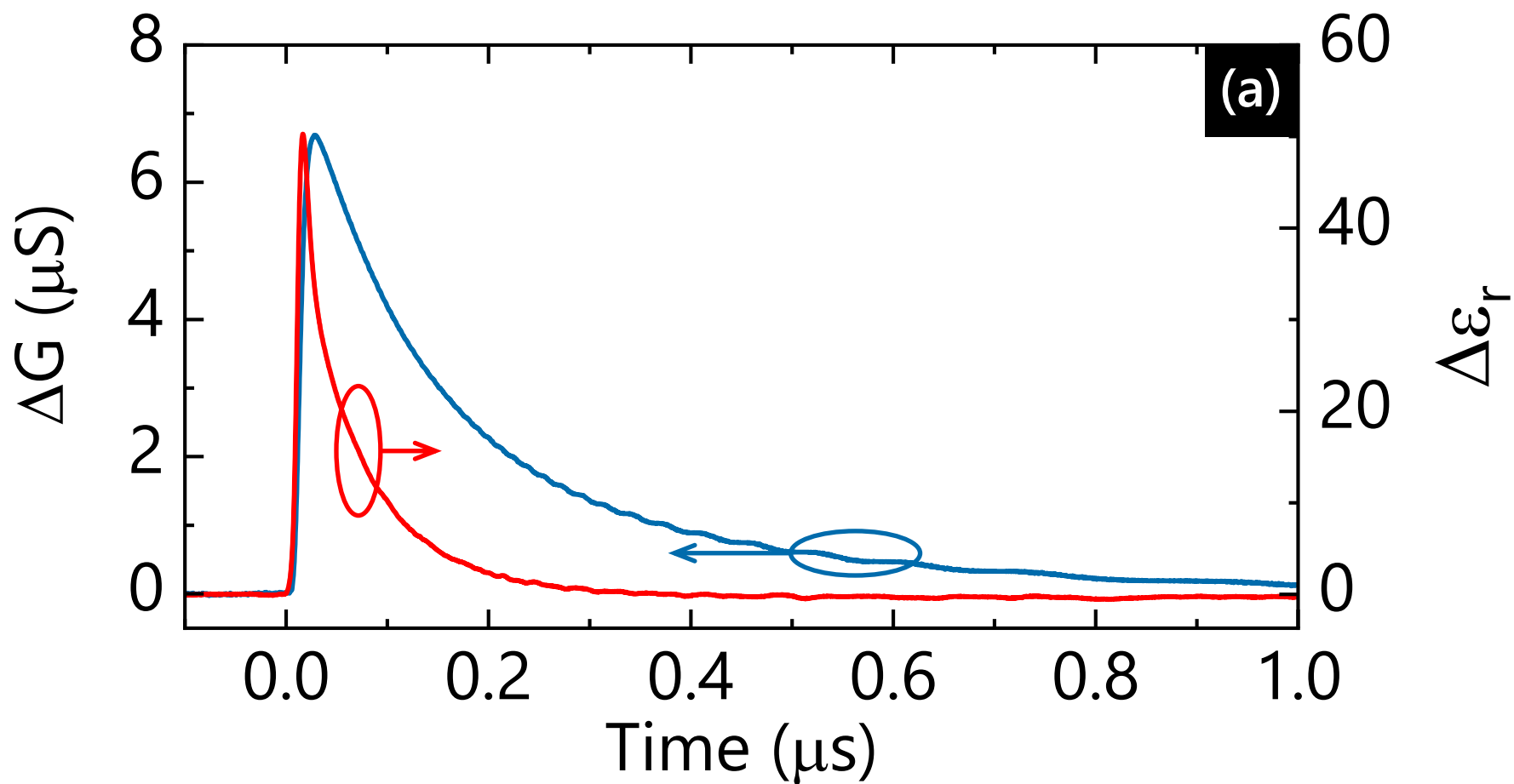
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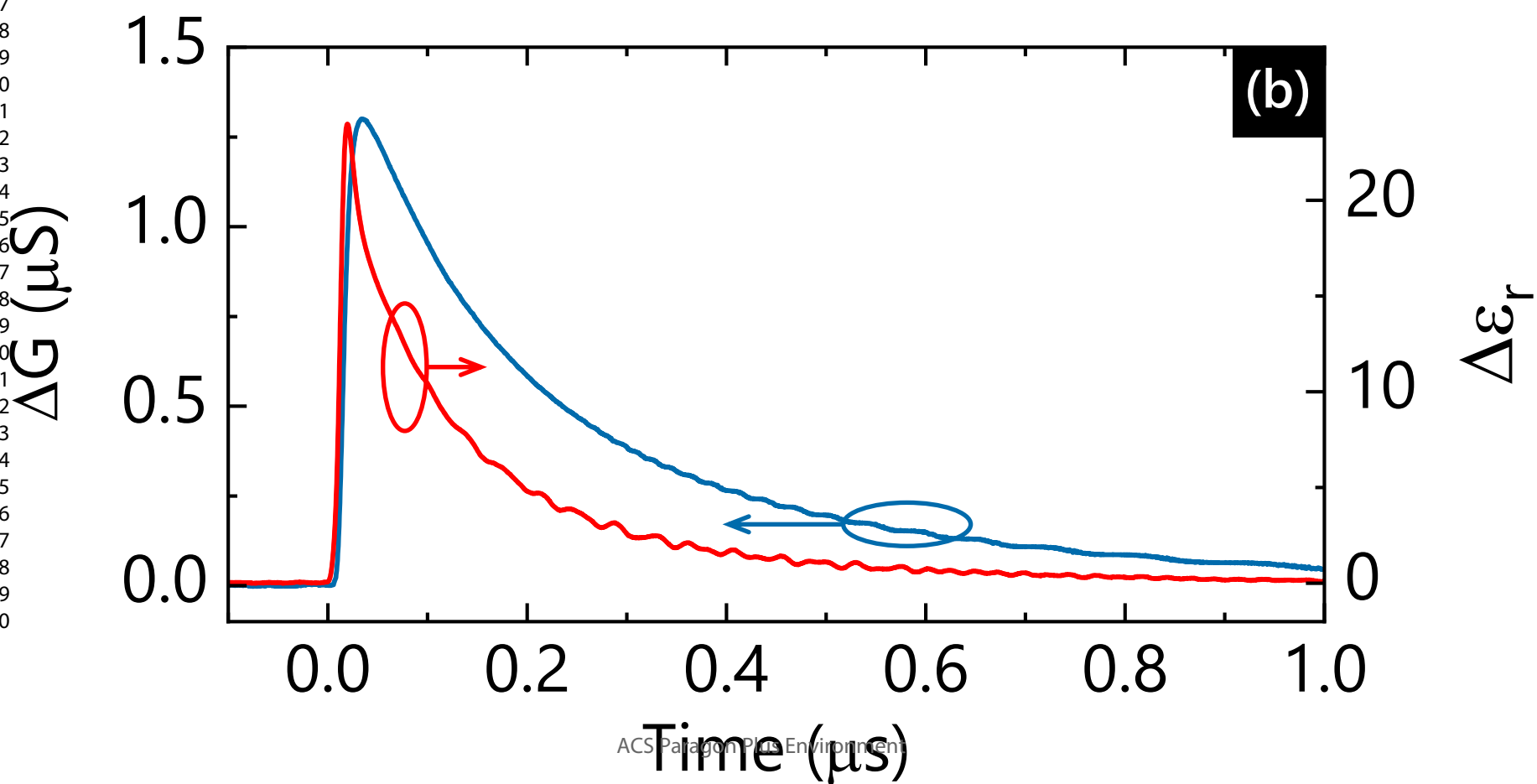
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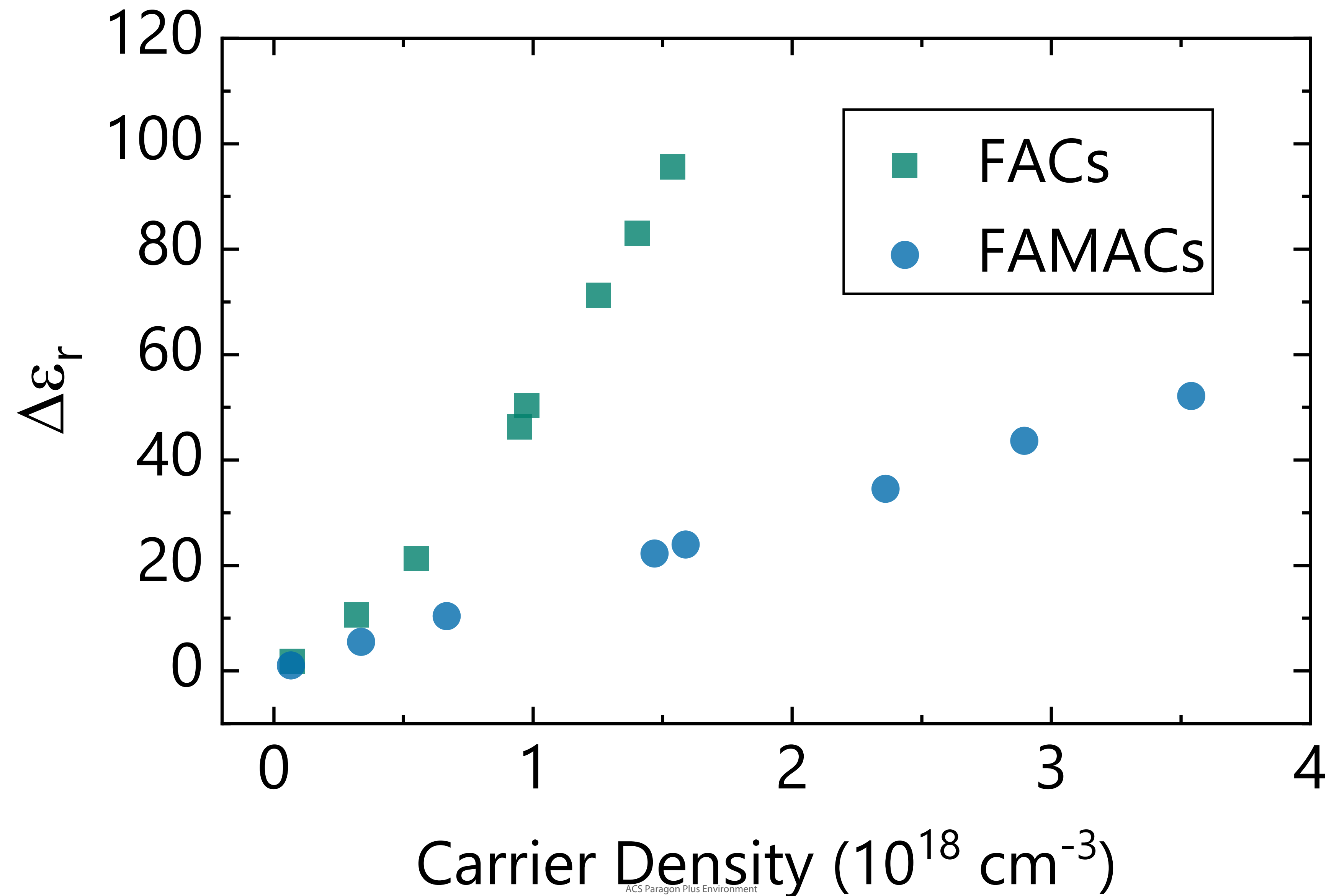


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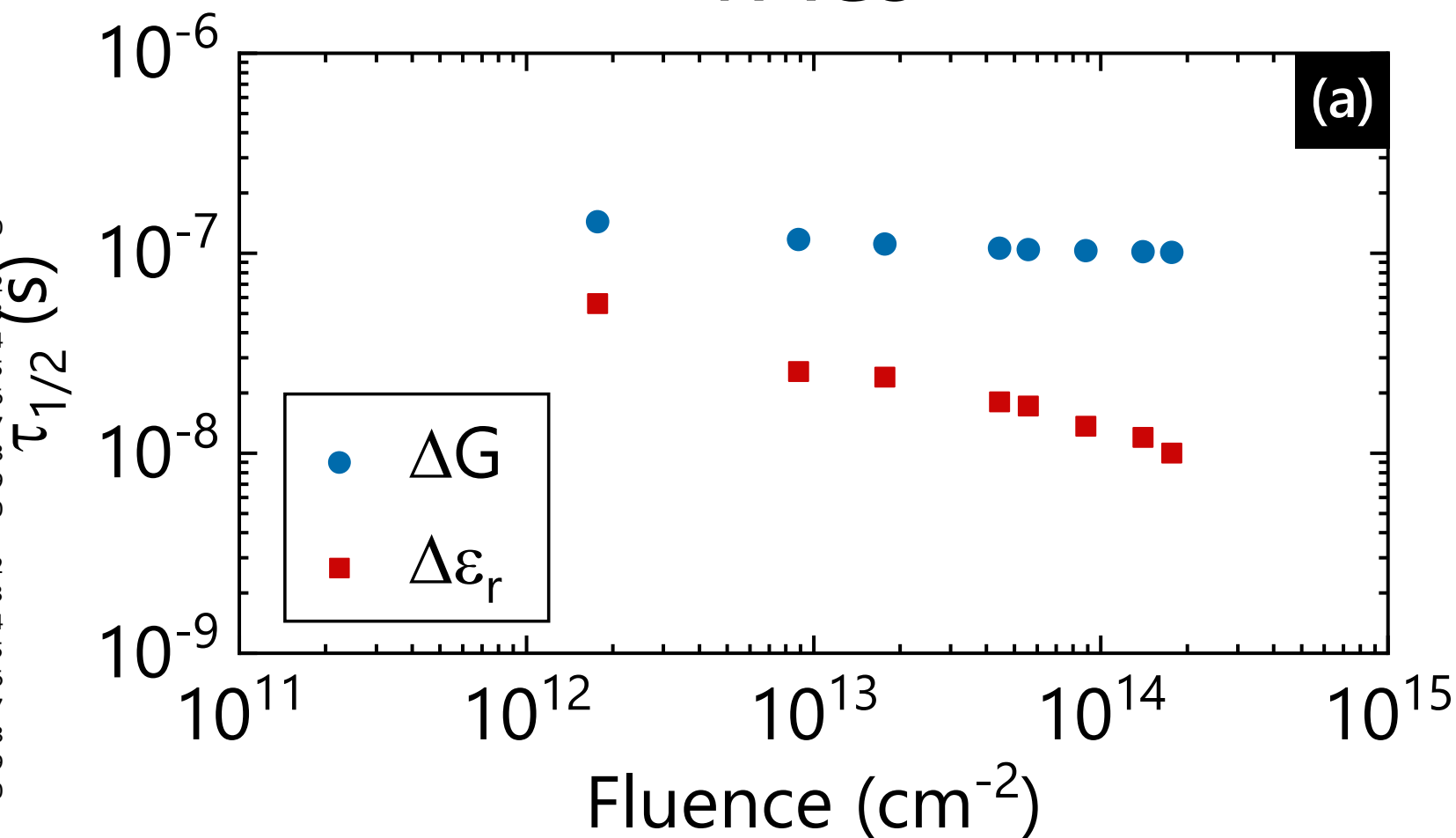
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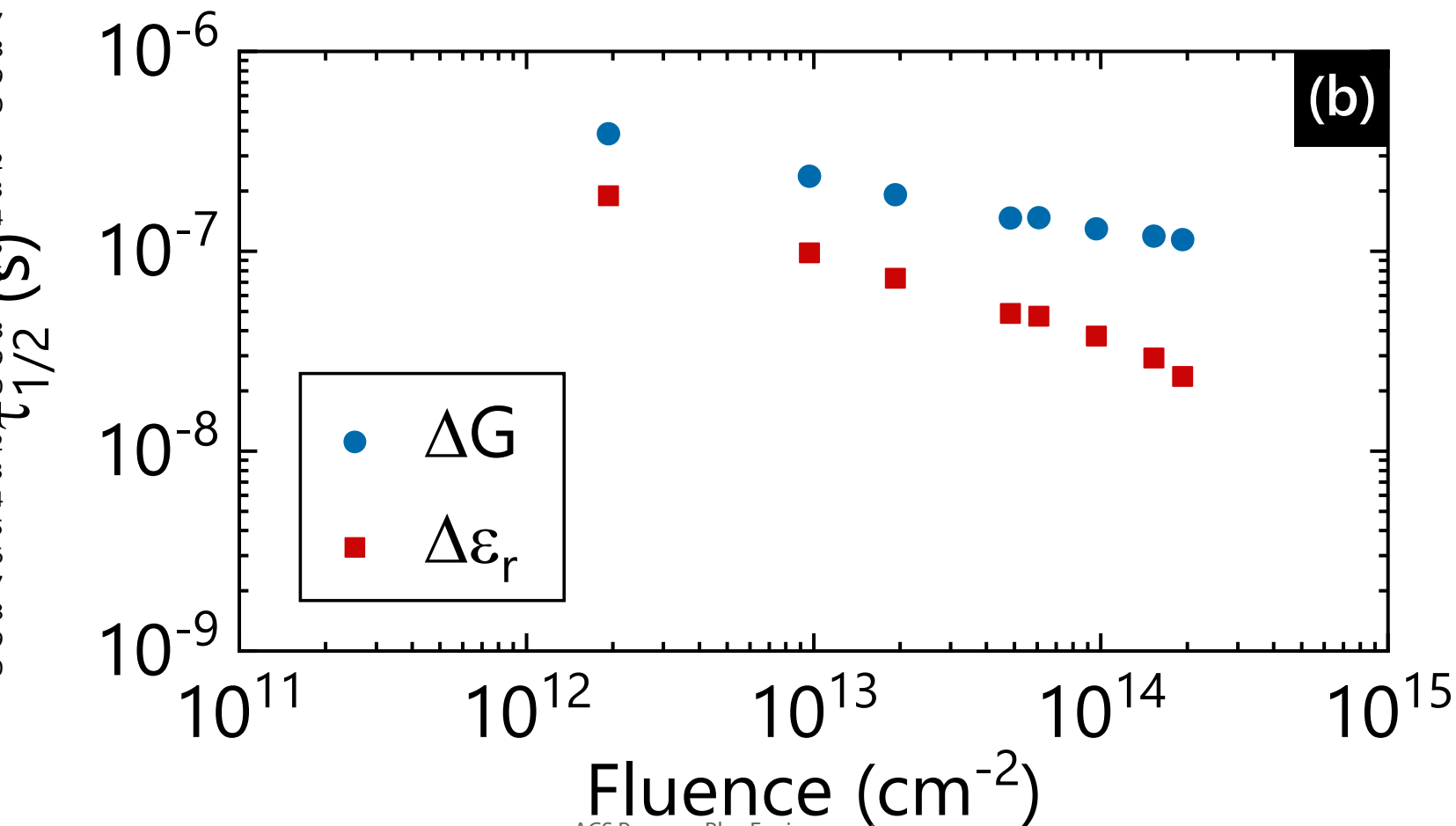
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