

The role of low-temperature ^{18}O exchange in the isotopic evolution of deep subsurface fluids

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ARTICLE INFO

Editor: Michael E. Boettcher

ABSTRACT

Ca-Na-Cl fluids with high concentrations of dissolved reduced gases reside within fractures in crystalline Precambrian rocks around the world, and have been most intensively studied within South Africa, Fennoscandia and the Canadian Shield. In contrast to surface waters, shallow groundwaters, sedimentary basin brines and metamorphic fluids, the $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values for these Ca-Na-Cl fluids typically plot to the left/above of the Global Meteoric Water Line (GMWL). To date, most interpretive frameworks for these fracture fluids have focused on their production via water-rock alteration reactions that affect both $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values, resulting in co-variation of water isotope values above the GMWL. Such alteration processes include silicate hydration coupled with formation of secondary minerals, radiolytic H_2 formation, isotopic exchange with a H_2 -rich gas, or isotope exchange with O and/or H-bearing minerals. This study presents the first compiled global isotopic dataset for these types of fluids, integrating a large amount of unpublished data with the previously published literature in order to investigate these fracture fluid systems on a global scale. Importantly this global perspective allows differentiation between fluids impacted by late-stage mixing with meteoric waters, from fluids that reflect the most saline end-members stored in the host rocks in hydrogeologic isolation from the surface hydrologic cycle. The most saline fluids are shown to occupy a more restricted range of $\delta^{18}\text{O}$ - $\delta^2\text{H}$ space than previously recognised, with end-member fluids from all of the Precambrian rock settings investigated occupying a range of $\delta^2\text{H}$ $\delta^{18}\text{O}$ isotope space within which there is no co-variation in these values. These findings suggest a set of common processes may define the isotopic signatures of these most saline end-member fluids in Precambrian settings around the world – creating common signatures identifiable in these fluids, despite differences in geologic setting. This study identifies the important role of oxygen isotopic exchange between primary fluids (associated with hydrothermal/metamorphic activity) and the host rocks, taking place under low temperature, low-volume, water-rock ratios over long (Ma) geologic timescales. This process results in progressive ^{18}O depletion in the fluids over time, while $\delta^2\text{H}$ values remain less affected. For each site the specific isotopic signature of the fracture fluid end-member depends on initial hydrothermal/metamorphic fluid composition, rates of isotopic exchange, water-to-rock ratios, and in-situ residence times. We suggest the often-observed co-variation in both $\delta^{18}\text{O}$ - $\delta^2\text{H}$ above the GMWL primarily results from late stage mixing of the fracture fluid end-members with (paleo)-meteoric water, resulting in isotopic regression back towards the GMWL, with end-points defined by the local meteoric-climatic conditions for each site.

1. Introduction

Within crystalline rocks of the Precambrian continental crust around the world, highly saline fracture waters have been identified, and most

intensively studied to date within South Africa, Fennoscandia and the Canadian Shield (e.g. [Frape and Fritz, 1982](#); [Nurmi et al., 1988](#); [Bottomley et al., 2003](#); [Frape et al., 2003](#); [Kieft et al., 2005](#); [Lin et al., 2006](#); [Onstott et al., 2006](#); [Stotler et al., 2012](#); [Kietäväinen et al., 2013](#); [Telling](#)

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<https://doi.org/10.1016/j.chemgeo.2020.120027>

Received 16 October 2020; Received in revised form 10 December 2020; Accepted 11 December 2020

Available online 14 December 2020

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et al., 2018; Heard et al., 2018). Saline fluids in this geologic setting range from Na-Cl-type near the surface, to more Ca-Na-Cl dominated fluids at depth, with salinity ranging from 60 up to 325,000 mg/L (A1 of the accompanying S.I.). Salinity, or Total Dissolved Solids (TDS), gradients typically co-vary with isotopic gradients such that the most saline samples at each locality are depleted in $\delta^{18}\text{O}$ and/or enriched in $\delta^2\text{H}$ relative to the Global Meteoric Water Line (GMWL), resulting in them plotting to the left/above the GMWL (Frape and Fritz, 1982; Frape et al., 1984, 2003; Sherwood Lollar et al., 1993; Kloppmann et al., 2002; Ward et al., 2004; Onstott et al., 2006; Kietäväinen et al., 2013). This contrasts with other saline subsurface fluids such as those found in sedimentary basins, or those interpreted to be metamorphic, which typically lie to the right/below the GMWL (e.g. Sheppard, 1986; Kharaka and Hanor, 2003, and references therein). The processes resulting in the Precambrian hosted saline fluids plotting above the GMWL remain a subject of active debate. Typically, a variety of mechanisms are invoked to account for the isotopic composition of these fluids, including hydrogen isotopic exchange with a H_2 -bearing gas phase out of equilibrium with the water (e.g. H_2S), formation of H_2 and/or O_2 gases from the water, hydration of primary silicates coupled with formation of secondary minerals (e.g. carbonates, clay minerals, iron hydroxides) producing co-variation in $\delta^2\text{H}$ and $\delta^{18}\text{O}$ (Kloppmann et al., 2002 and references therein), and lastly isotope exchange with oxygen- and/or hydrogen-bearing minerals (e.g. Blomqvist, 1990; Clark and Fritz, 1997; Frape et al., 2003, 2014; Kieft et al., 2005; Onstott et al., 2006; Clark, 2015). A general schematic for these proposed mechanisms is displayed in Fig. 1 (along with the typical processes such as evaporation and high temperature exchange that result in the isotopic composition of waters falling below the GMWL).

Despite the identification of relevant mechanisms (Fig. 1), at most localities, the relative importance of the proposed mechanisms and hence the quantitative controls on isotopic signatures that plot above and to the left of the GMWL remain unresolved. Based primarily on bromine and chlorine ratios, some authors had speculated that saline fluids in the Canadian Shield crystalline basement could be due to evaporation of surface water, and/or freezing of seawater, prior to infiltration (e.g. Frape and Fritz, 1982; Herut et al., 1990; Bottomley et al., 1994, 1999; Starinsky and Katz, 2003; Greene et al., 2008). But more recent studies such as Stotler et al. (2012) and Kietäväinen et al. (2013) both independently concluded that saline fluid formation at depth through freezing is inconsistent with the physical and geochemical data available for the Canadian and Fennoscandian Shields. For most localities, a variety of potential mechanisms remain plausible

(Fig. 1), but the relative importance of the proposed mechanisms and hence the quantitative controls on isotopic signatures that plot above and to the left of the GMWL remain a matter of debate.

In particular, the origin of primary fluid components prior to, and post, incorporation into the fracture networks remain uncertain in most cases, with no clear consensus reached on whether the primary source of the dissolved species (here termed salinity) is due to incursion of saline fluids from surface, or is primarily internally-generated in the host rocks (Nurmi et al., 1988; Frape et al., 2003, 2014; Onstott et al., 2006). External origins for the salinity include an incorporated seawater or terrestrial saline component (White, 1965; Bottomley et al., 1994; Frape et al., 2003; Greene et al., 2008; Landes et al., 2015) or dissolution of evaporite deposits (Fontes et al., 1989; Frape et al., 2003), depending on the location, depth and host rock. Proposed internal sources involve in situ water-rock interactions, typically either dissolution and alteration of host minerals (e.g. Fritz and Frape, 1982; Nurmi et al., 1988; Blomqvist, 1990; Kloppmann et al., 2002; Onstott et al., 2006; Kietäväinen et al., 2013), with potential influences from fluid inclusions contained within the host rock matrix (Garrels, 1984; Nordstrom et al., 1989; Lippmann-Pipke et al., 2011). Additionally, the existing salinity may be increased by water loss via reactions with host rock mineralogy or radiolysis (Vovk, 1987; Bucher and Stober, 2000; Kullerud, 2000; Kloppmann et al., 2002; Lin et al., 2005). A coherent theme in all the literature is that extensive water-rock reactions over geologically long timescales have obscured primary geochemical and isotopic features of the fluids, making identification of “source” fluids challenging.

This study integrates previously published data (376 points) with 216 new data points to establish a global dataset of $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values for fracture fluids from the South African Kaapvaal craton, and the Fennoscandian and Canadian Shields – including novel data from some of the deepest and most ancient saline fluids identified to date – from 2.4 and 2.9 km below surface at the Kidd Creek Mine, Timmins, Ontario, Canada. The Kidd Creek brines in particular, shown to be hydrogeologically isolated in the deep subsurface on long geologic timescales with mean residence times of millions to more than a billion years based on noble gases (Holland et al., 2013; Warr et al., 2018), allow the most highly saline and deepest end-members from Precambrian shield settings to be more effectively constrained than in previous studies. As such, this study focuses in some detail on the fracture fluid samples collected from Kidd Creek, since the dedicated subsurface scientific observatory at Kidd Creek, in continuous operation since 2008, means it contains some of the most well characterized geochemical, isotopic and geological context for Precambrian Shield sites in this global dataset. Hosted in Kidd Creek Mine, the deepest base metal mine in North America, this Cu-Ag-Zn deposit is a stratiform Volcanogenic Massive Sulphide (VMS) deposit of 2.7 Ga age (Thurston et al., 2008) located 24 km north of the town of Timmins, Ontario Canada. Part of the Abitibi greenstone belt, the deposit is a massive hydrothermal system analogous to modern day black smoker vent systems on the ocean floor. Originally was laid down horizontally on the ocean floor where rising high temperature silica and metal-rich hydrothermal fluids mixed with infiltrating low temperature seawater, the Kidd-Munro assemblage consists of a series of now steeply dipping interlayered felsic, mafic, ultramafic and metasedimentary units (Bleeker and Parrish, 1996). During the last major regional metamorphic event at 2.67–2.69 Ga metamorphism reached a maximum of greenschist facies, followed by low-grade metamorphic hydrothermal events with peak temperatures of between 300 °C and ~ 220 °C at 2.64 Ga (Smith et al., 1993; Bleeker et al., 1999). The system has been continuously cooling since that time with estimated temperatures below 100 °C for the past two billion years (Li et al., 2016), making this one of the best-preserved sections of Precambrian hydrothermal deposition, with relict “pillow lava” structures still visible in the wall rock.

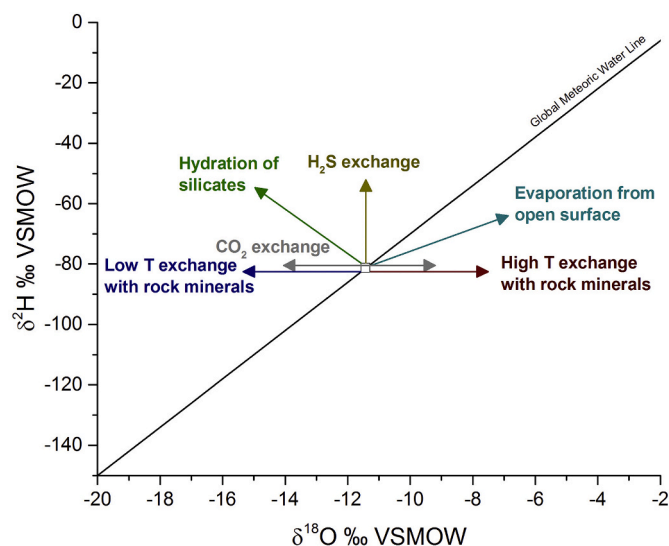


Fig. 1. Subsurface processes that can affect the $\delta^2\text{H}$ and $\delta^{18}\text{O}$ isotopic signature of groundwater. Figure modified from Clark and Fritz (1997).

2. Sample collection and analysis

The $\delta^{18}\text{O}$ and $\delta^2\text{H}$ data for 592 fluid samples in this study from fracture networks in Precambrian crystalline rocks are compiled from studies from the South African Kaapvaal Craton and the Fennoscandian and Canadian Shields (Table A1 of the accompanying S.I.). All samples were collected from boreholes drilled into crystalline rock (which intersect with fluid-bearing fractures) using three techniques, depending on the locality: pumping from surface boreholes, collection from boreholes drilled from surface using samplers actuated at depth, and underground sampling mines from free-flowing and in some cases, valved boreholes. Sampling is described in detail in the respective studies (Frape and Fritz, 1982; Frape et al., 1984; Kelly et al., 1986; Nurmi et al., 1988; Bottomley et al., 1990; Bottomley et al., 1994; Lippmann et al., 2003; Frape et al., 2003; Onstott et al., 2006; Stotler et al., 2009; Stotler et al., 2011; Kietäväinen et al., 2013; Lau et al., 2014; Miettinen et al., 2015; Lau et al., 2016; Li et al., 2016; Simkus et al., 2016; Kietäväinen et al., 2017; Telling et al., 2018; Heard et al., 2018; Warr et al., 2018; Lollar et al., 2019) but described briefly as follows.

For the samples collected by pumping from the boreholes drilled from surface, the methodology is described by Kietäväinen et al. (2013), Purkamo et al. (2013), Bomberg et al. (2015), and others. Target sampling depths were isolated using a series of inflatable packers, and any standing water removed by slow pumping at a rate which matched the flow of water into the borehole from the surrounding fracture network. After a period of 9–63 days (depending on the flow rates), purging was stopped and fluids in the borehole were considered representative of the in situ fracture networks intersecting the boreholes. These fluids were pumped to surface and collected for isotopic and geochemical analyses.

For sampling from boreholes drilled from surface by collecting samples at depth (i.e. without pumping), the methods outlined by Nurmi and Kukkonen (1986) and Sherwood Lollar et al. (1994) were used. Both of these methods are able to collect representative fluid samples at specific depths of a borehole using either polyamide tubing (Nurmi and Kukkonen, 1986) or pressurized sampling vessels (Sherwood Lollar et al., 1994) with shut-off valves at each end. Sampling devices were lowered into the borehole from the surface to the specified sampling depth, either slowly with the valves open to match the filling rate of the tube (Nurmi and Kukkonen, 1986), or at a greater rate with the valves closed until the depth of interest is reached at which point the valves open, pumps circulate the fluids and water entered the sampling vessel (Sherwood Lollar et al., 1994). In both sampling approaches, valves were then closed and the sampled fluids maintained under pressure until extraction at surface.

Finally, samples collected from free-flowing and valved boreholes within mines were taken directly at the borehole collars after the methods described in Sherwood Lollar et al. (2002) and Ward et al. (2004) and Lin et al. (2006). Briefly, a packer was inserted into the borehole below the water level and inflated to isolate the sample. After a period of flushing, samples were collected in Nalgene bottles with no headspace for isotopic analysis of $\delta^{18}\text{O}$ and $\delta^2\text{H}$. Anion and cation samples were filtered using a 0.2 μm filter and collected in a 60 mL Nanopure water washed Nalgene following the methods outlined in Ward et al. (2004), and Lollar et al. (2019). The cation sample bottles contained trace metal grade HNO_3 in vial to prevent precipitation of mineral phases. The specific analytical methods for previously published data are outlined in their respective publications. Where available, water isotope data from service water (waters introduced into the mine for use in drilling and mine operations) is provided in Table A2 to constrain the isotopic composition of any external fluids the mines pump underground, and to identify any potential fluid contamination from ongoing mining activity. Sources of such service waters vary between mines but can range from modern, meteoric water pumped from the surface (Lollar et al., 2019), to groundwaters from shallow aquifers overlying the mining operation (e.g. Onstott et al., 2006). Service waters typically have $\delta^{18}\text{O}$ and $\delta^2\text{H}$ distinct from those of the fracture fluids and

hence can serve as a measure of the degree of mixing and/or contamination, if such is present (Li et al., 2016; Lollar et al., 2019).

2.1. $\delta^{18}\text{O}$ and $\delta^2\text{H}$ sample analysis

For previously unpublished data, the $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values of the fluids were determined at the University of Waterloo using the MP-GVI-IsoPrime and EA/HT-Micromass-IsoPrime systems, respectively. Prior to analysis, the conductivity of groundwater samples was determined, and where conductivity was high ($> 0.1 \text{ S/m}$), samples were pre-treated by azeotropic distillation as per established methods (Fritz et al., 1986; Horita and Gat, 1988). The $\delta^{18}\text{O}$ was determined by CO_2 equilibration using standard procedures based on the principles of Epstein and Mayeda (1953). The $\delta^2\text{H}$ was performed on H_2 produced from water reduced on hot chromium as per Morrison et al. (2001). Accuracy of analysis for both techniques is confirmed by analysing duplicates and comparison to international reference standards sourced from the International Atomic Energy Agency (IAEA). The analytical methodologies are outlined as follows.

The $\delta^{18}\text{O}$ for each fluid sample was determined by the MP-GVI-IsoPrime instrument as previously described in detail Ward et al. (2004), Warr et al. (2018) and others. Initially, septum vials were filled with 200 μL aliquots of each sample and flushed with a 2% CO_2 -helium mixture. All the samples were then heated to 40 °C for a minimum of 3.5 h to allow complete oxygen isotopic equilibration between the CO_2 and the sample H_2O . Following this, the CO_2 in the headspace was automatically extracted using a Gilson Autosampler and GVI Multiprep combined automated sample preparation system. This setup separated CO_2 , N_2 , O_2 and H_2O using a 500 μL sample column. A helium carrier gas then transferred the purified CO_2 to the IsoPrime continuous flow isotope ratio mass spectrometer system (CF-IRMS) where the oxygen isotope ratios are measured to a precision of $\pm 0.2\text{‰}$. Calibrations were done using international reference materials V-SMOW (Vienna Standard Mean Ocean Water) and V-SLAP (Vienna Standard Light Antarctic Precipitation) from the IAEA. All values are expressed with respect to V-SMOW.

For $\delta^2\text{H}$ the EA/HT-Micromass-IsoPrime mass spectrometer was used. For the analytical procedure, 0.2 μL of each fluid sample was first pipetted into 2 mL septum vials which were then placed into a CTC AS200 Auto-sampler (LAS). From here, each sample was sequentially injected into a heated inlet system where a helium carrier gas transported it to a Eurovector Euro 3000 Elemental Analyzer with a 960 °C chromium furnace where the water was reduced to hydrogen gas. The H_2 was again transported via helium carrier gas to the EA/HT-Micromass IsoPrime mass spectrometer, which determined the $\delta^2\text{H}$ value for the sample. The precision for this technique is given as $\pm 0.8\text{‰}$. Calibrations were done using international reference materials V-SMOW (Vienna Standard Mean Ocean Water) and V-SLAP (Vienna Standard Light Antarctic Precipitation) from the IAEA. All values are expressed with respect to V-SMOW.

2.2. Aqueous geochemistry

For salinity data from published data sets, the specific analytical methods of the previously published data are outlined in their respective publications. Where concentrations of dissolved species are given, these were summed to determine total dissolved solids (TDS) in mg/L and are provided alongside the respective water isotope data in Table A1. For previously unpublished samples, the concentrations of dissolved species were analyzed by Activation Laboratories Ltd. (ISO/IEC 17025 certified) as previously outlined in Lollar et al. (2019). The cation content of the samples was measured using Spectro Cirros (ICP-OES) and their counterpart anion samples were analyzed using a DIONEX DX-120 Ion Chromatography System (Ion Chromatography). Analytical accuracy of the measured ionic species was checked by confirming samples had a charge balance error of less than 5% following the approach outlined in

Lollar et al. (2019).

3. Results

The isotopic data from Table A1 is plotted in Fig. 2(A–C) with respect to the GMWL for each region. As mentioned above, these fracture fluids typically fall above the GMWL relative to local meteoric water samples, and compared to service waters derived from local meteoric waters. Data which plot in between the GMWL and the most $\delta^2\text{H}$ -enriched, $\delta^{18}\text{O}$ -depleted values reflect to some extent a greater or lesser degree of mixing with meteoric waters either due to groundwater recharge from surface, or due to anthropogenic activities such as the aforementioned

circulation of mine service waters (e.g. Frapre and Fritz, 1982; Fritz and Frapre, 1982; Bottomley et al., 1994; Clark and Fritz, 1997; Bottomley et al., 1999; Kloppmann et al., 2002; Frapre et al., 2003, 2014; Ward et al., 2004; Onstott et al., 2006; Greene et al., 2008; Kietäväinen et al., 2013; Li et al., 2016). The degree to which the observed ^{18}O – ^2H co-variation at many of these sites (Fig. 2(A–C)) is controlled by such mixing with meteoric waters, versus controlled by water-rock reactions (WRI) has been more difficult to deconvolve however. Kloppmann et al. (2002)'s study of fluids from the Vienne granitic basement complex in France remains one of the few studies to quantitatively resolve the relative contributions of WRI and mixing to the ^{18}O – ^2H correlation trends in their data.

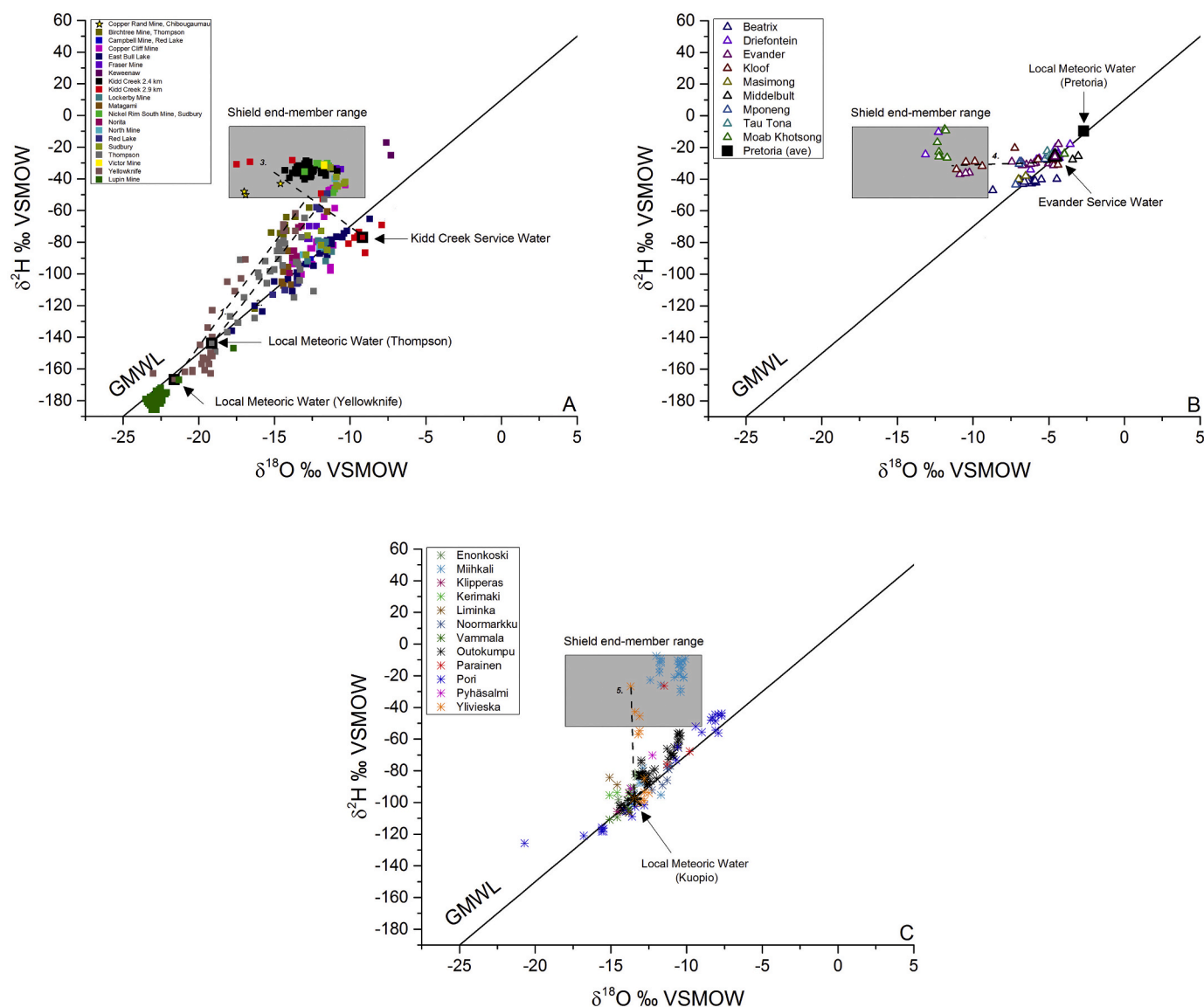


Fig. 2. Oxygen and hydrogen isotope data for fracture fluids from shield environments from Table A1 plotted relative to the Global Meteoric Water Line (GMWL). Data have been divided into the Canadian Shield (A, solid squares), South Africa (B, open triangles) and the Fennoscandian Shield (C, asterisks). Each site is colour-coded as per the key and is available in colour online. Errors for $\delta^2\text{H}$ and $\delta^{18}\text{O}$ are ± 0.8 and ± 0.2 ‰ respectively and are smaller than the plotted symbols. The area shaded in grey is based on the range of $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values for those fluids most elevated above the GMWL in Figures 2ABC, and encompasses end-members from all the shields (see text). For context, samples representative of local meteoric water (enlarged bold symbols) are plotted for two Canadian sites Yellowknife and Thompson, (A - Canada), for Pretoria (B - South Africa), and Kuopio (C - Finland), all from the IAEA WISER database (www.iaea.org). (Paleo)meteoric service water is also plotted from Evander Mine, South Africa (B). Lastly, for Kidd Creek, a sample of mine service water (from a lake on the mine site) is also shown in (A). These examples of local meteoric water allow the figure to demonstrate in dashed lines 1–3 (Fig. 2A), and 4–5 (Fig. 2B and C respectively) how late-stage mixing between the elevated end-member crystalline shield saline fluids, and meteoric end-members recharged from surface can produce the observed spread of data. Line 1 (Yellowknife); Line 2 (Thompson); Line 3 (Kidd Creek); Line 4 (Evander and Kloof); Line 5 (Ylivieska). Also plotted in 2A are brine isotopic data from vugs measured by Guha and Kanwar (1987) from Copper Rand Mine, Canada (yellow stars) which are discussed in the main text. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

In addition to the challenge of identifying the multiple mixing and WRI processes controlling isotopic distribution at the various sites, a major challenge has been identifying (to the degree possible in such highly altered systems), the fracture fluid end-members least affected by mixing with meteoric waters (most elevated over the GMWL). To date this has largely been done simply by extrapolating the linear ^{18}O – ^2H correlation to a hypothetical end-member for a given locality (Frape et al., 1984; Kloppmann et al., 2002). In this study, by integrating the global data from three continents, it is possible to identify a common isotopic range of fracture fluids which lie most elevated above the GMWL and may for the first time provide empirical constraints for the previously hypothetical end-member. This range is highlighted by the shaded box in Fig. 2 which represents a $\delta^2\text{H}$ – $\delta^{18}\text{O}$ range of -7 to -52‰ and -18 to -9‰ respectively, and which includes fluids trapped in vugs formed from hydrothermal/metamorphic activity from Copper Rand Mine in the Canadian Shield (Guha and Kanwar, 1987). This range is comparable to hypothetical end-member ranges proposed by Frape and Fritz, (1982) and Bottomley et al. (1999).

Table 1 represents a global compilation of published noble gas-derived mean fluid residence times for fracture fluids in Precambrian crystalline rocks available to date for South Africa, Fennoscandia and the Canadian Shield for a subset of the samples in Table A1. Although variations exist in the approach and parameters used to calculate fluid residence times, from Table 1 it is clear that fracture fluids in Precambrian host rocks around the world have the documented potential to be geochemically isolated from modern surface hydrogeologic cycle on time scales ranging from hundreds of thousands, to millions and even in extreme cases, more than a billion years (see references in Table 1). In locations where fluids have the longest mean residence times (in the range of 100 Ma and above), such as Kidd Creek Mine, Fraser Mine, Nickel Rim, Yellowknife, Evander, and Kloof, the $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values plot close to, or within the shaded empirical fracture fluid end-member range shown in Fig. 2. These end-member fluids, the oldest and most elevated above the GMWL for each site, also correspond to the most highly saline end-member fluids at each location (Table A1) demonstrating that high salinity and long residence times are associated with the most elevated $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values for each locality.

Figs. 2–3 show that the co-variation in $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values is consistent with mixing from these highly saline ancient fracture fluids, back towards less saline fluids located on the GMWL (dashed lines). While the intersection with the GMWL differs for each site depending on the local meteoric water for each location, at the other end of the trends, the mixing lines from sites across the Canadian Shield as diverse as Yellowknife, Thompson, and Kidd Creek and even from the Fennoscandian sites and South Africa all converge towards the shaded range of $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values in Fig. 2 (Frape et al., 1984; Bottomley et al., 1999; Onstott et al., 2006; Li et al., 2016; Telling et al., 2018; Warr et al., 2018). The identification of a globally similar ^{18}O – ^2H range for these most saline fracture fluid end-members changes the focus for investigation of fractionation effects resulting from the water-rock reactions and provides an insight which we will demonstrate is helpful in deconvolving the effects of mixing versus WRI. While mixing trends show co-variation in $\delta^{18}\text{O}$ – $\delta^2\text{H}$ over a wide range of values, variation within the highest salinity end-members are not characterized by ^{18}O – ^2H correlation – but instead occupy a distribution that is more horizontal. Specifically, the shaded area shows a variation in $\delta^{18}\text{O}$ of $\sim 10\text{‰}$ (relative to total variation for meteoric waters observed globally on the order of $\sim 30\text{‰}$); versus a variation in only 45‰ in $\delta^2\text{H}$ for the shaded area (relative to total variation for meteoric waters globally on the order of $\sim 400\text{‰}$; Craig, 1961). Relative to typical ranges of $\delta^{18}\text{O}$ – $\delta^2\text{H}$ for global waters (Craig, 1961), the highly saline, oldest fluids in the shaded region have a relatively smaller variation in $\delta^2\text{H}$ compared to $\delta^{18}\text{O}$. For the first time this dataset suggests that, once the effects of mixing are recognised based on the ^{18}O – ^2H correlation back towards less saline end-members that intersect the GMWL, the variation within the highest salinity end-members is principally $\delta^{18}\text{O}$ variation. Hence

consideration of the origin of these deep crustal fluids should focus on water-rock reactions that predominantly affect $\delta^{18}\text{O}$ values. This is a novel perspective compared to most previous literature which typically focused more heavily on water-rock reactions and processes that could produce a combined ^{18}O – ^2H correlation in the resulting fluids.

4. Discussion

4.1. A process-based model for global variation in deep saline fracture fluids in Precambrian crust

The most well-known set of processes that strongly affect $\delta^{18}\text{O}$ values while having little or relatively little affect on $\delta^2\text{H}$ isotopic values involve low-temperature ^{18}O isotope exchange with host rocks. Such processes are known to overprint the original fluid isotopic signal and produce significant horizontal shifts (Fig. 1 – this work, Clark and Fritz, 1997; Clark, 2015). If the fluids' position above the GMWL is principally caused by a horizontal shift to the left, then the starting range for these fluids can be considered prior to any ^{18}O isotopic exchange. Given the wide range of values for $\delta^2\text{H}$ of (paleo-)meteoric waters documented on a global scale of more than 400‰ (Craig, 1961), and given the wide range of localities with different climatic and geologic histories sampled in this study, a “common” meteoric-like surface water range is unlikely as a “source” fluid and not further considered. What is common to all sites is they are located in crystalline basement rocks, which with different lithological and geologic histories nonetheless all formed originally from a combination of metamorphic and hydrothermal processes. In cases like Kidd Creek, which formed as a massive subseafloor hydrothermal deposit analogous to modern day “black smokers”, “original” fluids were likely a mixture of hydrothermal/metamorphic fluids and infiltrating Archean seawater. On both the Canadian Shield and the Kaapval Craton, the deepest saline fluids with the longest residence time have in fact been shown to contain noble gases components that are isotopically similar to remnants of past regional metamorphic fluids (Lippmann-Pipke et al., 2011; Holland et al., 2013). While conservative elements like the noble gases may retain information on primary fluids, the multiple processes, extensive WRI and long timescales have, as mentioned previously, altered and over-written geochemical and isotopic information about what the “original or primary” fluid components were. Detailed studies of metabasalts and metasediments by Taylor (1974) and Sheppard (1977, 1986) published $\delta^{18}\text{O}$ and $\delta^2\text{H}$ ranges for metamorphic fluids globally (and magmatic, seawater and meteoric) and suggested, based on thermal history, mineralogy and defined fractionation curves, that a typical $\delta^2\text{H}$ range of ~ -20 to -65‰ , and a $\delta^{18}\text{O}$ range of $\sim +5$ to $+25\text{‰}$ could be estimated for metamorphic fluids since the late Precambrian at least, with few exceptions (e.g. a potential increase in $\delta^2\text{H}$ up to 0‰ if dehydration of oceanic crust is considered – Sheppard, 1986). This relative consistency in global estimates of $\delta^2\text{H}$ values for hydrothermal/metamorphic fluids in Precambrian settings likely reflects the relatively low isotopic variation of $\delta^{18}\text{O}$ and $\delta^2\text{H}$ in host lithologies on a global scale (Taylor, 1974; Taylor, 1978; Longstaffe and Gower, 1983; Clark and Fritz, 1997; Eiler, 2001).

For the Superior Province of the Canadian Shield in particular, from which 226 of the samples in this study were collected, Kerrich and Ludden (2000) estimated the range of $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values for metamorphic fluids (based on bulk rock analyses) of between $+4$ to $+25\text{‰}$ and -20‰ to -65‰ respectively from the Archean to the present day, consistent with the range suggested above by Taylor (1974). Very similar ranges have been estimated for metamorphic fluids for the Fennoscandian Shield (Piribauer et al., 2011; Kietäväinen et al., 2013) and the South African Kaapvaal craton (e.g. Kieft et al., 2005; Lin et al., 2006; Onstott et al., 2006; Zhao et al., 2006; Grové and Harris, 2010), and more generally (e.g. Clark and Fritz, 1997). While $\delta^2\text{H}$ values cannot be considered conservative, it is nonetheless notable that this estimated isotopic range for metamorphic fluids from the literature reviewed

Table 1

Mean fluid residence times in crystalline basement fluids from South Africa, Fennoscandia and the Canadian Shield. Residence times presented here were determined through noble gas analysis of the fluids and are a function of in situ production of radiogenic noble gases over time through radioactive decay of naturally occurring U, Th, and K in the host rock. Residence times represent the estimated time required to accumulate the measured radiogenic noble gas content of the fracture fluids based on the noble gas production parameters (see [Ballentine and Burnard, 2002](#) and references therein). Calculated mean residence times are based on models that integrate porosity estimates, the efficiency of release of noble gases from the rock matrix, and, where highlighted, external fluxes of noble gases into the system. Overall model parameters are briefly described here and the original literature citations are provided for further information regarding site-specific geochemistry, external fluxes, and residence time calculations.

Country	Location	Mean Fluid Residence time (Ma)	Model Parameters	Source
Finland	Outukumpu Deep Drill Hole	17–58	Closed system 100% release from rock matrix (He–Ar)	Kietäväinen et al. (2014)
Canada	Con Mine (Yellowknife)	340	0.5% porosity Closed system 100% release from rock matrix (He)	Bottomley et al. (2005)
		54–425	1.0% porosity Closed system 100% release from rock matrix (He, Ar)	Greene et al. (2008)
Canada	Kidd Creek Mine 2.4 km	1142–1655	1.0% porosity Closed system 100% release from rock matrix (He–Xe)	Holland et al. (2013)
		359–414	1.0% porosity Closed system 100% release from rock matrix (He–Xe)	Warr et al. (2018)
Canada	Kidd Creek Mine 2.9 km	1622–1890	1.0% porosity Closed system 100% release from rock matrix (He–Xe)	Warr et al. (2018)
Canada	Nickel Rim Mine	250–357	0.88% porosity Closed system 100% release from rock matrix (He–Xe)	Warr et al. (2018)
Canada	Fraser Mine	544	1.1% porosity Closed system 100% release from rock matrix (He–Xe)	Warr et al. (2018)
Canada	Whiteshell Nuclear Research Establishment	0.09–20	1.2% porosity Closed system 100% release from rock matrix (He)	Bottomley et al. (1984)
S. Africa	Mponeng Mine	15.8–25	1.0% porosity External flux 100% release from rock matrix (He, Ar) 20–80% release (Xe)	Lin et al. (2006)
S. Africa	Beatrix Mine	16–92	1.0% porosity Closed system 100% release from rock matrix (He, Ar) 20–42% release (Xe)	Lippmann et al. (2003)
		0.6–5.1	1.0% porosity External flux 100% release from rock matrix (He, Ar) 20–44% release (Xe)	Lippmann et al. (2003)
		3–14	1.0% porosity External flux 100% release from rock matrix (He, Ar) 20–44% release (Xe)	Lau et al. (2016)
		13–136	1.0% porosity Closed system 100% release from rock matrix (He–Xe)	Heard et al. (2018)
S. Africa	Evander Mine	13–364	0.7% porosity Closed system 100% release from rock matrix (He, Ar) 40–100% release (Xe)	Lippmann et al. (2003)
		1.0–9.6	1.0% porosity External flux 100% release from rock matrix (He, Ar) 43–80% release (Xe)	Lippmann et al. (2003)
S. Africa	Kloof Mine	21–170	1.0% porosity Closed system 100% release from rock matrix (He, Ar) 22–52% release (Xe)	Lippmann et al. (2003)
		2.7–21.1	1.0% porosity External flux 100% release from rock matrix (He, Ar) 23–55% release (Xe)	Lippmann et al. (2003)
S. Africa	Witwatersrand Basin	40– ≥ 2000 (He–Ar)	1.0% porosity 40 Ma from closed system 100% release from rock matrix (He–Ar)	Lippmann-Pipke et al. (2011)
S. Africa	Masimong Mine	39–97	0.1–0.8% porosity 2000 Ma from fluid inclusion Ne component Closed system 100% release from rock matrix (He–Ar)	Lippmann-Pipke et al. (2011) Heard et al. (2018)

(continued on next page)

Table 1 (continued)

Country	Location	Mean Fluid Residence time (Ma)	Model Parameters	Source
S. Africa	Driefontein Mine	3.4–10.8	47–100% release (Xe) 0.7% porosity External flux 100% release from rock matrix (He–Ar)	Heard et al. (2018)
			47–100% release (Xe) 0.7% porosity Closed system 100% release from rock matrix (He–Xe)	
		1.1–12	2% porosity External flux 100% release from rock matrix (He–Xe)	Heard et al. (2018)
		0.2–0.69	2% porosity Closed system 100% release from rock matrix (He–Ar), 13–100% release (Xe), 0.7% porosity	
S. Africa	Tau Tona Mine	0.77	External flux 100% release from rock matrix (He–Ar), 13–100% release (Xe)	Heard et al. (2018)
		0.1–1.2	0.7% porosity Closed system 100% release from rock matrix (He–Xe)	
			0.05% porosity External flux 100% release from rock matrix (He–Xe)	
			0.05% porosity	
S. Africa	Zondereinde Mine	2.7–12	0.7% porosity Closed system 100% release from rock matrix (He–Xe)	Heard et al. (2018)
		0.006–0.047	0.05% porosity External flux 100% release from rock matrix (He–Xe)	

above is very similar to the range of $\delta^2\text{H}$ values shown in the most saline end-members at the sites in this study (shaded area in Fig. 1). It is well known that $\delta^2\text{H}$ values are less likely to be affected than $\delta^{18}\text{O}$ values, due to the relatively small mineral bound-H reservoir in rocks (Kelly et al., 1986; Guha and Kanwar, 1987; Clark and Fritz, 1997; Onstott et al., 2006). Some small degrees of ^2H -enrichment have been shown. For example, Kloppmann et al. (2002) calculated that 40% loss of water due to hydration of feldspars could feasibly account for a modest ($\sim 15\%$) ^2H enrichment. Alternatively, where hydrous minerals already exist, hydrogen isotopic exchange may result in a small extent of ^2H enrichment ($\sim 20\%$) (e.g. Kieft et al., 2005; Onstott et al., 2006). Concomitant with this isotopic evolution, increasing ionic content can also occur due to water-rock reactions including enrichments in Cl^- , trace metals and cations such as Ca^{2+} through in situ radiolysis of water (e.g. Vovk, 1987; Lin et al., 2005) or through a series of high-low temperature reactions, mineral hydration, gradual contributions from fluid inclusions, and ion exchange with minerals of the host rocks (e.g. Frapre and Fritz, 1982; Nordstrom et al., 1989; Clark and Fritz, 1997; Bucher and Stober, 2000; Kullerud, 2000), Frapre et al., 2003, 2014; Bottomley et al., 2005; Clark, 2015). Presently, the details of the origin and evolution of the salinity component in these fracture fluids remains an active area of research. This study focuses for now solely on the $\delta^{18}\text{O}$ and $\delta^2\text{H}$ isotopic composition.

In contrast to the range of $\delta^2\text{H}$ values, the shaded area (Fig. 2) is significantly more depleted in ^{18}O than the literature estimates for metamorphic/hydrothermal fluids and rocks/metasediments for the Precambrian crust reviewed above. During cooling, mineral growth/precipitation and low temperature equilibration under low water-rock ratios are well known processes that can result in more depleted $\delta^{18}\text{O}$ values in the remaining fluid phase (Fig. 1) due to preferential incorporation of the lighter ^{16}O isotope into the water phase (e.g. Clark and Fritz, 1997). Where water to rock ratios are sufficiently low, the $\delta^{18}\text{O}$ of the fluids at equilibrium with the host rock will depend only on the $\delta^{18}\text{O}$ of the host rock (and the temperature), independent of the initial fluid $\delta^{18}\text{O}$ (Clark and Fritz, 1997; Clark, 2015). Low temperature water-rock reaction rates are sluggish (Clark and Fritz, 1997), hence such processes would require long timescales of water-rock interaction to facilitate sufficient low-temperature isotopic exchange. The long residence times of the fluids (Table 1) associated with the most elevated and highly saline fluids (Figs. 2 and 3) are consistent with the extended residence

times required to drive isotopic exchange between the fluids and host rocks towards equilibrium, even under present-day temperatures. For example, Kieft et al. (2005) and Onstott et al. (2006) applied low temperature equilibration models to fracture fluids from the Kaapval craton to suggest that ~ 70 – 80% isotopic equilibration could account for the $\delta^{18}\text{O}$ (and $\delta^2\text{H}$) values observed in the most saline end-members there (assuming very low water-to-rock ratios). Progressive low temperature equilibrium was proposed by Guha and Kanwar (1987) to explain the $\delta^{18}\text{O}$ and $\delta^2\text{H}$ values of Ca-Na-Cl-rich brines (326–407 g/L TDS) in vugs from Copper Rand Mine on the Canadian Shield (Table A1; Fig. 2A). Given their location in cross fractures within the vein system, the vugs have a clear association with hydrothermal/metamorphic activity, and the vug fluids were considered to be related to the mineralisation of the ore body. From this, Guha and Kanwar (1987) proposed that after fluid emplacement at elevated temperatures ($\sim 250^\circ\text{C}$), mineralisation and cooling resulted in a reduction in the water-rock ratios and progressive depletion in the $\delta^{18}\text{O}$ of the residual saline fluids, resulting in the vug fluids plotting to the left of the GMWL, though no quantification of the extent of equilibration was provided. This study, by integrating a 592-sample dataset from a broad spectrum of Precambrian shield sites around the world, demonstrates that low temperature isotopic equilibration between the host rocks and the fracture fluids is likely the dominant process controlling the isotopic composition of deep cratonic fluids on a global scale. The well-known trends of $\delta^{18}\text{O}$ and $\delta^2\text{H}$ co-variation tend to reflect the effects of later stage mixing with meteoric and paleo-meteoric fluids.

4.2. Quantitative application of the conceptual model to Kidd Creek

By far the greatest density of the new data in Fig. 2 is derived from what are, to date, the oldest fluids discovered in the Precambrian crust – the fracture fluids from 2.4 to 2.9 km depth at Kidd Creek Mine in Timmins, Ontario with mean residence times of more than a billion years (Holland et al., 2013; Warr et al., 2018). This density of samples is due to the Kidd Creek Observatory, which has over the course of 13 years provided the infrastructure and opportunity to monitor a suite of discharging saline fluids, making this the world's most longstanding and deepest location for investigation of subsurface fluids and deep life in the crystalline basement (Lollar et al., 2019). Data from two levels of the observatory can be used to investigate the mechanisms controlling the

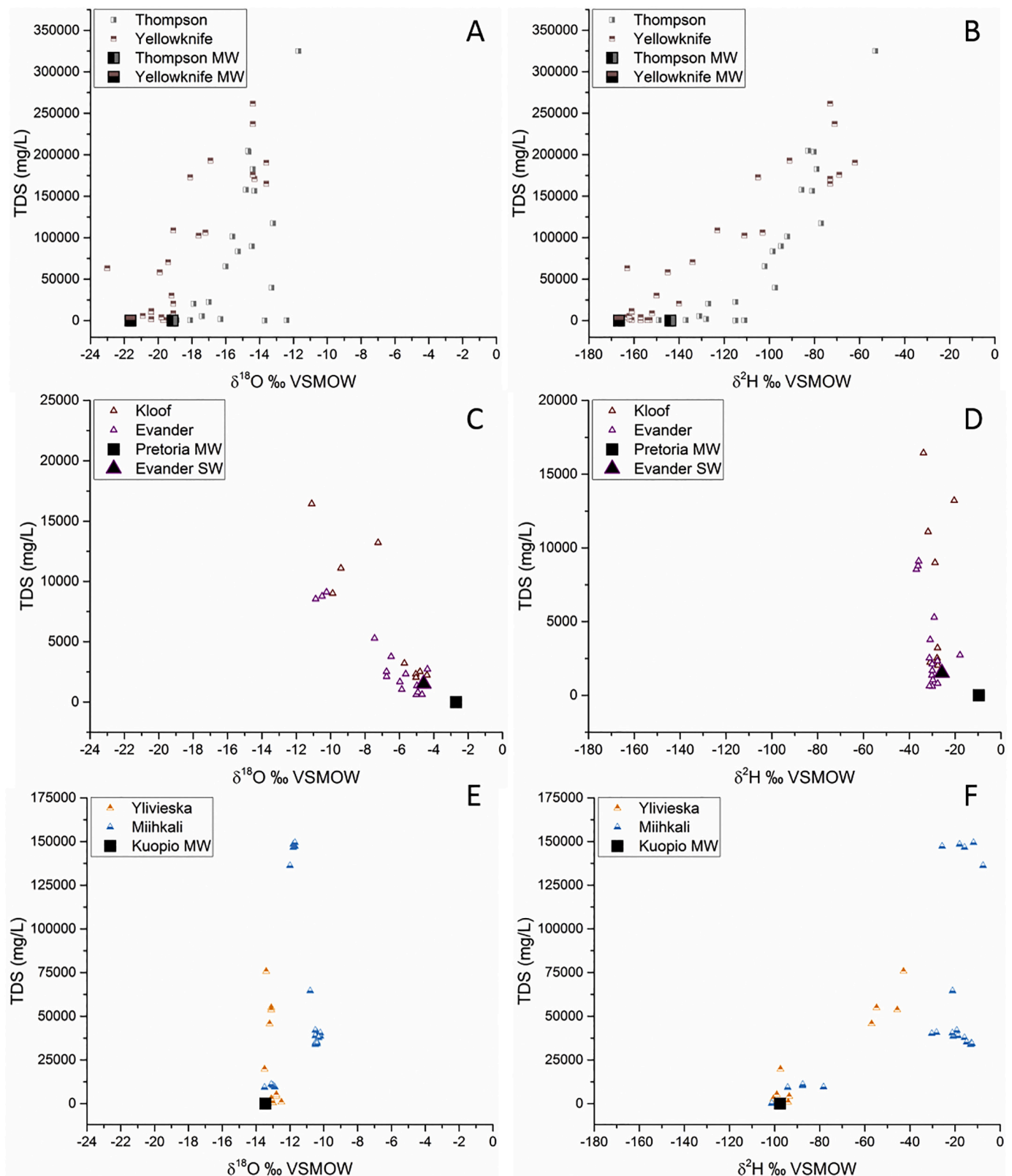


Fig. 3. Variation in total dissolved solid content (TDS) in the fluids with oxygen and hydrogen isotopic data for shield samples environments from Table A1. Data have been divided into the Canadian Shield (A-B), South Africa (C-D), and the Fennoscandian Shield (E-F). Each site is colour-coded as per the key and Fig. 2 and is available in colour online. Errors for $\delta^2\text{H}$ and $\delta^{18}\text{O}$ are $\pm 0.8\text{‰}$ and $\pm 0.2\text{‰}$ respectively and are smaller than the symbols. For context, representative samples of local meteoric water (MW - enlarged bold symbols) are plotted for Yellowknife and Thompson, (Canada) for Pretoria (South Africa), and for Kuopio (Finland), from the IAEA WISER database (www.iaea.org). (Palaeo)meteoric Service water (SW) from Evander Mine, South Africa, is also plotted in C and D for context.

isotopic composition of the fracture waters in further detail. In Fig. 4, the data for Kidd Creek fracture fluids sampled at 2.4 km depth and 2.9 km depth (Table A1) are plotted alongside the service water that represents meteoric water used by the mining company for operations and drilling. This service water is sourced from a local lake, and as such has a slightly shallower gradient than typical GMWL slope, and plots along a line which intersects the GMWL with a classic 'lake water' slope (Lollar et al., 2019). As sampling at the 2.9 km level took place during a major drilling campaign, a significant proportion of the samples initially collected from 2.9 km are dominantly or significantly affected by recent addition of meteoric water during drilling, consistent with the lower salinities observed for these samples (Table A1). For example, the effect of meteoric water addition during mining at this level can be clearly observed in BH13762. The initial sample from this borehole plotted above the GMWL ($\delta^2\text{H}$ of -29.3‰ , $\delta^{18}\text{O}$ of -16.6‰) with a TDS of 215,000 mg/L. However, the two successive samples showed that the TDS decreased by up to 99% and the isotopic composition plot along local meteoric line or else fall in a position intermediate between the GMWL and the average value for 2.9 km (shown by star in Fig. 4). This is consistent with the fact that these 2 samples were collected while the drill rigs were still in place in the holes (Table A1 of the accompanying S. I.). Three other samples from 2.9 km (shown in solid red circles) do not show this dilution with service waters and so have likely flushed out remnants of the drilling water fluid post-drilling completion.

Applying the conceptual model of low-temperature exchange to the specific example of the Kidd Creek data benefits from the very well characterized information available from the literature on the oxygen

isotopic composition of the host rock and in particular, fracture minerals. In this locality, as typically observed elsewhere, calcite and quartz are the two most common and frequently observed fracture minerals and have been in prolonged contact with the water phase for extended time periods. Studies of quartz and calcite fracture minerals show the measured $\delta^{18}\text{O}$ of both primary quartz and calcite fracture minerals associated with metamorphic fluids at Kidd Creek have an average $\delta^{18}\text{O}$ value of $\sim 12.0\text{‰}$ (Kerrick et al., 1987; Huston and Taylor, 1999). Meanwhile, Kerrich and Ludden (2000) used bulk rock analyses to estimate a metamorphic fluid $\delta^{18}\text{O}$ range of $+6$ to $+9\text{‰}$ at Kidd Creek. This range was independently supported by Huston and Taylor (1999) who inferred a similar $\delta^{18}\text{O}$ range for metamorphic fluids of $+6.7$ to $+7.0\text{‰}$ from mineralogic studies. The same samples were also used to propose a tentative metamorphic fluid $\delta^2\text{H}$ range of -9 to -26‰ , which overlaps with the range of -20‰ to -65‰ considered for metamorphic fluids by Kerrich and Ludden (2000). These $\delta^{18}\text{O}$ ranges were also consistent with estimated fluid $\delta^{18}\text{O}$ values calculated from temperatures recorded from fluid inclusions and the $\delta^{18}\text{O}$ of quartz and calcite from metamorphic veins elsewhere in the area (Beatty et al., 1988; Schandl and Wicks, 1991; Barrett et al., 2013) and are used as the basis for the shaded range (metamorphic fluid range) shown in Fig. 5.

Where a fluid-rock system reaches complete isotopic equilibrium under extremely low water-to-rock ratios, the relationship between equilibration temperature, the oxygen isotope fractionation (α) between the $\delta^{18}\text{O}$ of each fracture mineral, and the fluid values ($^{18}\text{O}_{\text{min}}$ and $\delta^{18}\text{O}_{\text{H}_2\text{O}}$ respectively) can be defined by the following equation:

$$\delta^{18}\text{O}_{\text{H}_2\text{O}} = \left(\frac{(1000 + \delta^{18}\text{O}_{\text{min}})}{\alpha_{\text{min-water}}(T)} - 1000 \right) \quad (1)$$

Where the water-to-rock ratio is larger, and/or where isotopic exchange is incomplete, the relationship defined by Eq. 1 will not be valid. Instead,

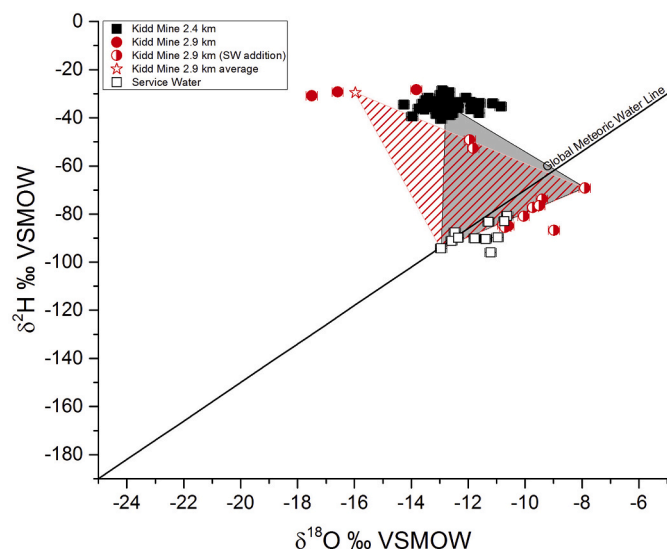


Fig. 4. Kidd Creek water isotope data for 2.9 km (red circles) compared with data from 2.4 km (black squares). While the $\delta^2\text{H}$ is comparable between the two levels, there is a potential minor isotopic depletion in the $\delta^{18}\text{O}$ of the water samples at the deeper level of the mine. This could simply be empirical as there are significantly more samples from the 2.4 km level versus the 2.9 km level to date. For both levels, a shaded region is shown that indicates where samples would fall if mixing were occurring between mine level average (star) and service water used within the mine. The lower slope line on which service waters fall was defined by Lollar et al. (2019). Only some samples from 2.9 km show a mixing effect, consistent with sample collection taking place while active drilling was still underway at this level. All but 5 of the samples from 2.9 km fall along the GMWL with 3 of these having comparable salinities to their 2.4 km counterparts (solid red circles) while 2 plotted within the shaded region. The majority of samples collected from 2.9 km are isotopically closer to the meteoric/service water line and reveal significantly lower salinities, consistent with the dilution effects of service water addition (symbols shown in half-white and half-red circles). The samples from 2.4 km are entirely isotopically distinct from the service waters. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

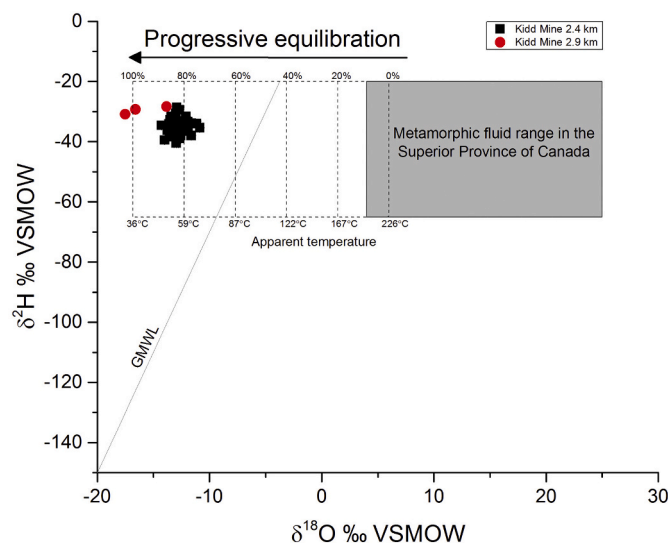


Fig. 5. Modelling isotopic evolution towards low-temperature equilibration between fracture water and quartz at 36 °C for Kidd Creek using isotope data for 2.9 km (red circles) compared with data from 2.4 km (black squares). Estimated metamorphic fluid values for the Canadian Shield are shown in the shaded region based on a $\delta^{18}\text{O}$ value of 6‰ proposed for Kidd Creek by Kerrich and Ludden (2000) and the range estimated for $\delta^{18}\text{O}$ and $\delta^2\text{H}$ in the Superior Province of Canada ($+4$ to $+25\text{‰}$ and -20‰ to -65‰ respectively) from Kerrich and Ludden (2000). This range is consistent with the typical metamorphic range for these types of fluid defined by Taylor (1974) based on metabasalts and metasediment data. The degree of isotopic fractionation between 0 and 1 is given as a percent along the top axis along with the apparent temperature the progressive fractionation corresponds provided for reference along the bottom. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

Eq. 1 can be modified to evaluate the effect of progressive equilibration with the host mineralogy under low water-to-rock ratios after the approach outlined by Karolyt   et al. (2017):

$$\delta^{18}O_{H_2O} = \left(\frac{1000 + \delta^{18}O_{min}}{\alpha_{min-water}(T)} - 1000 \right) \times f + \delta^{18}O_{H_2O(INT)} \times (1-f) \quad (2)$$

where $\delta^{18}O_{H_2O}$ and $\delta^{18}O_{min}$ represent the $\delta^{18}O$ of the fracture fluid and fracture mineral respectively, $\alpha_{min-water}(T)$ is the isotopic fractionation between the fracture mineral and fracture fluid at temperature T , f is the fluid fraction that has equilibrated and $\delta^{18}O_{H_2O(INT)}$ is the initial metamorphic fluid value. In this model $\delta^{18}O_{min}$ is considered constant given the low water to rock ratios (following the method of Clark and Fritz (1997) and Clark (2015)).

Using the above equations, it is possible to evaluate the degree of equilibrium reached between the fracture fluids and the fracture minerals. Taking quartz as an example, at Kidd Creek, this fracture mineral has an average $\delta^{18}O$ value of 12.0‰ which has been estimated based on 10 analyses (Huston and Taylor, 1999), and the corresponding $\alpha_{(qz-water)}$ value can be determined using the temperature-dependent fractionation factors of Knauth and Epstein (1976) and Kawabe (1978). Here, a temperature of 36 °C is used based on a depth of 3 km and a geothermal gradient of 12 °C/km for this region of the Canadian Shield (Artemieva and Mooney, 2001; Warr et al., 2019). The initial metamorphic fluid $\delta^{18}O$ value of 6‰ for Kidd Creek is taken from the literature along with the proposed metamorphic fluid δ^2H range of –20‰ to –65‰ used by Kerrich and Ludden (2000) for the Superior Province. By increasing f from 0 to 1 the progressive isotopic evolution towards equilibrium between the fracture fluids and the associated quartz under extremely low water-to-rock-ratios can be modelled. This is shown in Fig. 5.

Fig. 5 suggests fracture fluids collected from Kidd Creek for this study approach complete low-temperature equilibrium with the surrounding quartz fracture minerals. Though it is possible that the deeper level of the mine (2.9 km) may be slightly closer to complete equilibrium for infinitely low water-rock ratios, this cannot be conclusively confirmed based on the available data since as noted above, there are substantially more samples from 2.4 km than from 2.9 km.

Using the same approach, the relationship between calcite fracture mineralogy and the fracture fluids at Kidd Creek can also be evaluated using the temperature-dependent fractionation factor of O'Neil et al. (1969) and an estimated average fracture calcite $\delta^{18}O$ value of 12‰ ($n = 16$, Kerrich et al., 1987). Incorporating this into Eq. 2, it can be seen that all Kidd Creek fluid samples indicate upwards of 80% equilibrium has also been reached between the fluids and the corresponding calcite fracture minerals. This again, suggests that the system is approaching low temperature equilibrium under low water to rock conditions. For the deeper 2.9 km samples though, equilibrium with calcite alone cannot produce the two most isotopically depleted fluid $\delta^{18}O$ values; this therefore may suggest ^{18}O exchange with quartz is the more dominant exchange mechanism affecting these fluid samples, (as highlighted by Fig. 5) with additional variation potentially due to variability in the local water/rock ratio.

Overall, four lines of evidence support an approach to complete oxygen isotopic equilibration between the fluids and the surrounding host rocks. The first is the good agreement for both carbonate-fluid; and quartz-fluid isotopic systems. It would be rather serendipitous for them to both have comparable isotopic fractionations which both reflect a system approaching isotopic equilibrium. Secondly, literature estimate of the water-to-rock ratio for these crystalline basement settings were summarized by Holland et al. (2013) and further, calculated by Sherwood Lollar et al. (2014), using an adapted model of Bethke (1985), yields ratios of approximately 0.01, which are consistent with the low water to rock ratios required for equilibration between the water phase and the host rock to be approached. Such low water-to-rock ratios are common for these environments as highlighted in Table 1. Third, the thermal history of the Kidd Creek system indicates temperatures have

been below 100 °C for ~2 Ga (Li et al., 2016). Fourth, coupled with the previous point, noble gas dating of these fluids highlights that they have been isolated in situ over Ma to Ga timescales (Holland et al., 2013; Warr et al., 2018). Accordingly, there has likely been sufficient time for the mineral-fluid system to have equilibrated at ≤ 60 °C and resulted in progressive ^{18}O depletion in the fluids.

Expanding on the noble gas aspect further, the Warr et al. (2018) study previously demonstrated, that consistent with hydrogeologic models for draining fracture systems, the fluids sampled at 2.4 km showed decreasing mean residence times compared to the samples taken in 2009 (Holland et al., 2013). Although this decrease in residence times is accompanied by reduced radiogenic noble gas concentrations and a modest decrease in salinity (17% and 14.1% for boreholes 12287a and 12,299 respectively), this reduction in fluid residence times had no corresponding effect on the water isotopes (Table 10 - Warr et al., 2018) which remained constant. Likewise, the same study also identified fluids at the deeper level of the mine with an average mean residence time of 1.7 Ga, significantly older than fluids at 2.4 km. Despite this, the isotopic signature for the water remains comparable between the two levels. This indicates that for the given physical parameters (e.g. temperature, water-to-rock ratios etc.) the fluids in the system are also already close to/at isotopic equilibrium with the host rocks and hence the $\delta^{18}O$ and δ^2H values remaining largely constant for the entire Kidd Creek system (always staying within the grey shaded region characteristic of the oldest most saline deep brines; Fig. 2).

4.3. A global conceptual model

A conceptual model incorporating the three major controlling processes discussed in this study is presented in Fig. 6.

1. High temperature metamorphic/hydrothermal activity controls the initial isotopic composition of the fluid system either due to hydrothermal components directly or due to high temperature equilibration of other fluids (modified seawater or meteoric water) with the metamorphic/hydrothermal fluids. The immediate precursors to the Precambrian Shield fluids hence likely plot to the right of the GMWL and can be estimated from isotopic studies focusing on metabasalts and metasediments (Taylor, 1974; Sheppard, 1977, 1986).

2. Cooling and mineralisation coupled with long-term, low-temperature equilibration in a low water-rock ratio results in ^{18}O depletion in

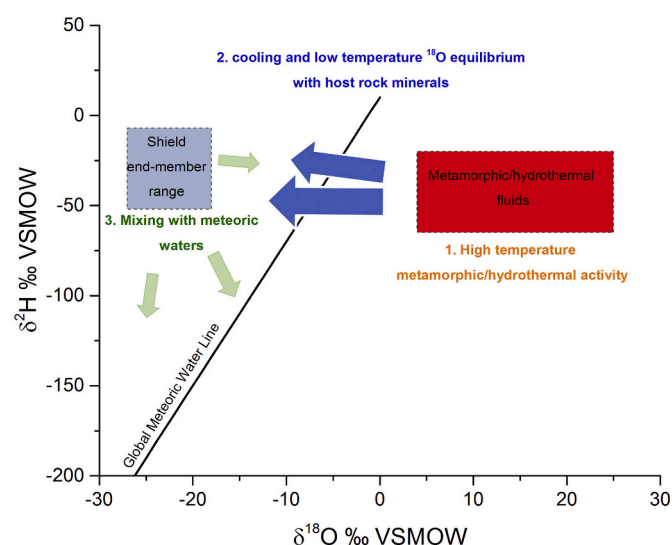


Fig. 6. Conceptual model for formation and evolution of saline crystalline basement fluids after Kelly et al. (1986), Guha and Kanwar (1987) and Bottonley et al. (1994) using the metamorphic/hydrothermal fluid range from Kerrich and Ludden (2000) and the range for fracture fluids identified in this work (Fig. 2).

the fracture fluids and the subsequent position of the fluids above/to the left of the GMWL. The long residence times (Ma-Ga) and high salinities of the samples in this study (shaded area Fig. 2) are consistent with the large degree of equilibration required. Some modest degree of ^2H enrichment may also occur in specific sites due to kaolinite precipitation and/or hydration of feldspars. At this stage the fluids are expected to have developed their characteristic elevated salinities, potentially due to a combination of steps 1 and 2, and/or any site-specific processes. 3. Infiltration of meteoric or paleometeoric water occurs through natural and/or anthropogenic activity and resultant mixing with the shield end-members shifts the observed isotopic signatures towards the GMWL. This also results in the salinities of these fluids being diluted as demonstrated in Fig. 4.

An important component of this model are long residence times of the fracture fluids conducive to promoting high degrees of low-temperature isotopic exchange between the fracture fluids and host rocks. For specific sites, additional factors will control locally the exact degree of equilibration ultimately achieved. Variations in both the water-to-rock ratios within the fracture networks and the distribution of ^{18}O -bearing minerals for instance will play a role in the rate and extent of evolution towards isotopic low-temperature equilibrium between the fracture fluids and the host rock. Considerations of these local factors, coupled with other variables such as geothermal gradients, and/or changes in fracture porosity and geometry alongside the initial fluid origin and composition, the ‘precursor’ metamorphic fluid isotopic composition, and associated mineral isotopic signatures, and temperature are the primary controls affecting the details of the isotopic signature of these deep crustal fluids on a site specific basis.

The scientific literature on the oxygen and hydrogen isotopic composition of saline fluids from the crystalline basement falling above the GMWL has long struggled with the relative importance of “source” or “precursor” fluid signatures, versus fractionation introduced by alteration during water-rock reactions. To date though there is no clear consensus on the mechanism(s) and relative importance of mechanisms at different sites and while it is widely acknowledged that the trends of co-variation in $\delta^{18}\text{O}$ and $\delta^2\text{H}$ results from both mixing of fluids of different provenance and the effects of water-rock reaction, discussion has nonetheless tended to focus on processes capable of causing fractionation in both oxygen and hydrogen. This paper integrates hundreds of data from sites across three continents and in particular, synthesizes the $\delta^{18}\text{O}$ and $\delta^2\text{H}$ signatures for crystalline fracture fluids from the most hydrogeologically isolated, most saline and oldest systems, of which the Kidd Creek Observatory provides the oldest, with residence times estimated to be more than a billion years. These highly saline Kidd Creek brines, hydrogeologically isolated in the deep subsurface on long geologic timescales (Holland et al., 2013; Warr et al., 2018), have allowed more effective constraints to be placed on the potential end-member composition, permitting deconvolution of the effects of mixing. This study constrains the isotopic characteristics of these end-member fluids more effectively than previous work and demonstrates that investigations into the origin of these crystalline fluid end-members

should focus on processes that primarily produce ^{18}O fractionation. Building on this insight, a model is developed that can account for these end-member saline fluids at equilibrium with their host rock, based on low-temperature ^{18}O equilibration of metamorphic fluids on the long geologic time scales documented for these systems.

5. Conclusions

This study presents the first global dataset of water isotope data and salinities compiled from published and unpublished data from S. Africa, Fennoscandia and the Canadian Shield. Careful analysis of this data has allowed differentiation between fluids affected by late-stage mixing with (palaeo)meteoric waters, from fluids with the longest mean residence times that reflect the most saline end-members stored in the host rocks. On a global scale, the most saline Precambrian shield fluids occupy a similar $\delta^{18}\text{O}$ - $\delta^2\text{H}$ space considerably more restricted than previously recognised. This suggests a key set of common processes involving metamorphic/hydrothermal fluids and progressive, low temperature $^{18}\text{O}/^{16}\text{O}$ -isotopic equilibration over long geologic timescales control deep crustal fluids dynamics in Precambrian Shield settings globally. Given that 72% of the continental lithosphere by surface area is Precambrian in age (Goodwin, 1996) this finding potentially has implications for modelling the deep hydro(bio)geosphere over long geologic timescales and understanding planetary processes relevant to the deep carbon cycle and deep subsurface life. Such insight into major processes controlling subsurface fluids is also relevant to planetary science investigations beyond Earth. For example, the subsurface of Mars is largely tectonically quiescent in the present day with the potential for subsurface fracture networks containing low temperature saline fluids preserved on geologic-planetary timescales (e.g. Michalski et al., 2018; Sherwood Lollar et al., 2007; Tarnas et al., 2018, and references therein).

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Acknowledgements

This study was financially supported by the Canada Research Chairs program, Natural Sciences and Engineering Research Council of Canada Discovery and Accelerator grants, and additional funding was provided by the Deep Carbon Observatory Washington, USA, Nuclear Waste Management Organization and Canadian Institute for Advanced Research (CIFAR) Toronto, Canada. Senior author BSL is a CIFAR Fellow. G. Lollar is thanked for his assistance in compiling the fracture fluid isotopic data. Additional thanks are due to colleagues and supporters at the mines whose efforts and support for the sampling program were invaluable.

Appendix A. Appendix

Table A1

The $\delta^{18}\text{O}$ and $\delta^2\text{H}$ data and accompanying salinities and metadata for 592 fluid samples in this study from fracture networks in Precambrian crystalline rocks compiled from studies from the South African Kaapvaal Craton and the Fennoscandian and Canadian Shields. Where data is from a published source the reference is provided. Samples were collected using three techniques: pumping from boreholes drilled from surface (1); sampling from boreholes drilled from surface by collecting samples at depth with downhole devices (2); or sampling from free-flowing and valved boreholes at depth within mines (3). Sampling methods are provided here as numbers for each sample and are discussed in detail in the main text and references provided.

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Canada	Birchtree	Bi	nm	-16.3	-122.0	nm	Bottomley et al. (1994)	3
Canada	Birchtree Mine, Thompson	28.05.2007_BT3950L_BH	1200	-14.3	-69.2	114,840	Unpub.	3
Canada	Birchtree Mine, Thompson	28.05.2007_BT3950L_BH1a	1200	-15.2	-74.1	105,600	Unpub.	3
Canada	Birchtree Mine, Thompson	29.05.2007_BT3950L_BH3	1200	-14.2	-64.3	118,206	Unpub.	3
Canada	Birchtree Mine, Thompson	06.11.2007_BT_3900_9167N_1113850	1190	-13.6	-64.2	127,710	Unpub.	3
Canada	Birchtree Mine, Thompson	06.11.2007_BT_3900_9167N_1113790	1190	-14.2	-72.8	94,314	Unpub.	3
Canada	Birchtree Mine, Thompson	07.11.2007_BT_3900_9167N_1113860	1190	-14.6	-74.4	99,792	Unpub.	3
Canada	Birchtree Mine, Thompson	08.11.2007_BT_3900_9167N_1113740	1190	-14.4	-75.6	105,864	Unpub.	3
Canada	Birchtree Mine, Thompson	27.03.2008_BT_3950_9167N_BT1	1190	-12.7	-58.3	130,680	Unpub.	3
Canada	Campbell	Ca	nm	-12.6	-91.0	nm	Bottomley et al. (1994)	3
Canada	Campbell Mine, Red Lake	Campbell 27-11	1220	-12.2	-58.2	181,744	Unpub.	3
Canada	Campbell Mine, Red Lake	Campbell 27-09	1220	-11.5	-49.3	301,535	Unpub.	3
Canada	Campbell Mine, Red Lake	Campbell 27-09	1220	-12.0	-59.2	231,428	Unpub.	3
Canada	Campbell Mine, Red Lake	Campbell dam	640	-13.4	-105.5	783	Unpub.	3
Canada	Campbell Mine, Red Lake	Campbell shute	640	-13.5	-106.2	715	Unpub.	3
Canada	Campbell Mine, Red Lake	Campbell 14-166	640	-14.3	-110.4	2070	Unpub.	3
Canada	Campbell Mine, Red Lake	Campbell 14-454	640	-15.1	-113.4	2165	Unpub.	3
Canada	CCS		nm	-12.6	-84.1	31,566	Unpub.	3
Canada	CCS		nm	-12.5	-84.0	30,597	Unpub.	3
Canada	CCS		nm	-12.5	-94.2	3848	Unpub.	3
Canada	CCS		nm	-11.3	-95.9	2906	Unpub.	3
Canada	CCS		nm	-13.2	-100.5	3135	Unpub.	3
Canada	CCS		nm	-11.0	-58.6	83,614	Unpub.	3
Canada	CCS		nm	-11.7	-64.2	77,975	Unpub.	3
Canada	CCS		nm	-11.3	-47.0	127,330	Unpub.	3
Canada	CCS		nm	-11.4	-47.5	102,114	Unpub.	3
Canada	Copper Cliff Garson Mine	G-1600	1600	-11.9	-82.0	4279	Frape and Fritz (1982)	3
Canada	Copper Cliff North	CC	nm	-13.8	-94.0	nm	Bottomley et al. (1994)	3
Canada	Copper Cliff South Mine	CCS508421	500	-12.2	-80.0	1888	Frape and Fritz (1982)	3
Canada	Copper Cliff South Mine	CCS508422	500	-11.7	-83.0	1295	Frape and Fritz (1982)	3
Canada	Copper Cliff South Mine	CCS508481	500	-11.8	-80.0	791	Frape and Fritz (1982)	3
Canada	Copper Cliff South Mine	CCS205001	2050	-11.4	-82.0	3544	Frape and Fritz (1982)	3
Canada	Copper Cliff South Mine	CCS205002	2050	-11.1	-82.0	3989	Frape and Fritz (1982)	3
Canada	Copper Cliff South Mine	CCS205003	2050	-11.3	-98.0	3076	Frape and Fritz (1982)	3
Canada	Copper Cliff South Mine	CCS3200	3200	-11.1	-46.0	85,789	Frape and Fritz (1982)	3
Canada	Copper Cliff South Mine	CCS3250	3250	-10.3	-44.0	94,097	Frape and Fritz (1982)	3
Canada	Copper Cliff Wells (Dowling)	1	500	-13.5	-89.0	220	Frape and Fritz (1982)	3
Canada	Copper Cliff Wells (Dowling)	2	500	-12.9	-88.0	187	Frape and Fritz (1982)	3
Canada	Copper Cliff Wells (Dowling)	3	500	-14.1	-98.0	107	Frape and Fritz (1982)	3
Canada	Copper Rand Mine, Chibougaumau	CRV1	nm	-16.9	-50.0	326,053	Guha and Kanwar (1987)	3
Canada	Copper Rand Mine, Chibougaumau	CRV2	nm	-17.0	-48.0	326,294	Guha and Kanwar (1987)	3
Canada	Copper Rand Mine, Chibougaumau	CRV3	nm	-14.6	-43.0	406,550	Guha and Kanwar (1987)	3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Canada	East Bull Lake	P-1	32–53.3	–11.8	–82.2	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-2	24.5–38.1	–11.9	–86.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-3	0–16	–11.9	–85.1	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-3	16–38.1	–12.0	–87.2	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-4	35–55	–11.5	–88.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-5	0–22	–15.8	–124.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-5	88.5–97.6	–17.8	–136.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-6	0–68	–11.7	–80.9	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-7	0–39.6	–10.2	–73.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-8	0–20	–10.5	–76.9	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-9	0–12	–11.6	–80.7	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-9	12–43.2	–11.6	–83.9	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-10	0–34.5	–11.5	–79.1	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-10	34.5–38.5	–11.3	–78.5	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-10	66.3–70.3	–11.2	–79.1	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-10	82–98.7	–11.2	–79.8	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-11	0–49.2	–11.2	–81.3	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-12	0–37	–11.6	–80.2	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-12	38–48.6	–12.0	–83.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-13	0–22	–11.9	–84.2	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	P-14	0–43.3	–11.8	–82.6	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-1	147–298	–13.8	–111.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-1	204–213	–13.5	–99.2	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-1	298–512	–12.4	–95.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-1	504–642	–13.3	–95.0	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-2	0–75	–10.4	–74.6	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-2	75–248	–14.1	–103.4	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-2	111–126	–16.3	–120.4	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-2	248–460	–13.5	–103.2	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-2	460–616	–8.7	–65.4	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-3	0–443	–11.4	–79.8	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-4	0–120	–10.9	–76.7	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-4	105–117	–12.9	–93.9	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-4	120–198	–15.0	–104.9	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-4	198–396	–14.1	–103.8	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL-4	429–456	–11.9	–80.3	nm	Bottomley et al. (1990)	1
Canada	East Bull Lake	EBL4	nm	–12.9	–93.9	nm	Bottomley et al. (1994)	1
Canada	Fraser Mine	Fr33170	3300	–12.7	–78.0	21,689		3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Canada	Fraser Mine	Fr33182	3300	-12.2	-70.0	29,737	Frape and Fritz (1982)	3
Canada	Fraser Mine	Fr33192	3300	-12.8	-78.0	17,713	Frape and Fritz (1982)	3
Canada	Fraser Mine	Fr33300	3300	-12.7	-70.0	22,594	Frape and Fritz (1982)	3
Canada	Fraser Mine, Sudbury	08.01.2014_FML4700_FR47774	1430	-11.0	-33.9	nm	Unpub.	3
Canada	Fraser Mine, Sudbury	06.03.2014_FML4700_FR47774	1430	-10.9	-33.4	39,130	Unpub.	3
Canada	Fraser Mine, Sudbury	06.03.2014_FML4700_FR47774anchor	1430	-10.6	-34.0	99,000	Unpub.	3
USA	Keweenaw		nm	-7.6	-17.1	164,395	Kelly et al. (1986)	3
USA	Keweenaw		nm	-7.3	-25.2	204,463	Kelly et al. (1986)	3
USA	Keweenaw		nm	nm	nm	270,363	Kelly et al. (1986)	3
USA	Keweenaw		nm	nm	-74.0	203,550	Kelly et al. (1986)	3
Canada	Kidd Creek	KC	nm	-15.5	-113.0	nm	Bottomley et al. (1994)	3
Canada	Kidd Creek Mine, Timmins	10.05.2007_KC7850_Station_12262	2390	-14.0	-39.5	173,514	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	27.08.2007_KC_7800_Station_12299	2390	-13.1	-38.5	213,105	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	27.08.2007_KC_7800_Station_12287A	2390	-12.8	-38.7	175,475	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	27.08.2007_KC_7800_Station_12261	2390	-13.0	-35.0	212,432	Unpub.	3
Canada	Kidd Creek Mine, Timmins	11.02.2008_KC_7800_12299	2390	-12.6	-36.5	175,047	Unpub.	3
Canada	Kidd Creek Mine, Timmins	11.02.2008_KC_7800_12261	2390	-12.8	-32.0	nm	Warr et al. (2018)	3
Canada	Kidd Creek Mine, Timmins	19.06.2008_KC_7850_12299	2390	-12.8	-38.6	218,170	Warr et al. (2018)	3
Canada	Kidd Creek Mine, Timmins	20.06.2008_KC_785012287A	2390	-13.0	-40.5	168,550	Warr et al. (2018)	3
Canada	Kidd Creek Mine, Timmins	31.03.2009_KC_7850_12299	2390	-13.3	-35.8	197,833	Warr et al. (2018)	3
Canada	Kidd Creek Mine, Timmins	12.01.2010_KC_7850_12299	2390	-13.5	-36.7	217,405	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	21.10.2010_KC_7850_12299	2390	-13.6	-34.2	213,759	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	21.10.2010_KC_7850_12287A	2390	-13.8	-36.3	167,397	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	21.10.2010_KC_7850_12262	2390	-14.3	-34.6	195,573	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	29.02.2012_KC_7850_12299	2390	-11.6	-38.0	189,117	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	29.02.2012_KC_7850_12287A	2390	-12.6	-38.9	135,367	Unpub.	3
Canada	Kidd Creek Mine, Timmins	29.02.2012_KC_7850_12262	2390	-12.1	-31.7	220,357	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	14.06.2012KCL7850_BH12299	2390	-13.4	-35.7	196,612	Li et al. (2016)	3
Canada	Kidd Creek Mine, Timmins	28.11.2012KCL7850_BH12299	2390	-12.4	-34.7	134,087	Unpub.	3
Canada	Kidd Creek Mine, Timmins	28.11.2012KCL7850_BH12287A	2390	-12.7	-34.1	nm	Unpub.	3
Canada	Kidd Creek Mine, Timmins	17.01.2013KCL7850_BH12299	2390	-13.3	-33.1	137,111	Unpub.	3
Canada	Kidd Creek Mine, Timmins	19.09.2013_KC_7850_BH12299	2390	-13.5	-32.7	178,397	Unpub.	3
Canada	Kidd Creek Mine, Timmins	19.09.2013_KC_7850_BH12287A	2390	-13.5	-35.7	nm	Unpub.	3
Canada	Kidd Creek Mine, Timmins	19.09.2013_KC_7850_BH12262	2390	-13.0	-30.6	nm	Unpub.	3
Canada	Kidd Creek Mine, Timmins	20.09.2013_KC_7850_BH12299	2390	-13.4	-31.7	174,948	Unpub.	3
Canada	Kidd Creek Mine, Timmins	20.09.2013_KC_7850_BH12287A	2390	-13.1	-35.3	139,950	Warr et al. (2018)	3
Canada	Kidd Creek Mine, Timmins	02.04.2014_KC_7850_BH12299	2390	-13.0	-34.4	180,473	Unpub.	3
Canada	Kidd Creek Mine, Timmins	02.04.2014_KC_7850_BH12287A	2390	-12.6	-35.8	140,748	Unpub.	3
Canada	Kidd Creek Mine, Timmins	02.04.2014_KC_7850_BH12262	2390	-12.9	-28.6	167,329	Unpub.	3
Canada	Kidd Creek Mine, Timmins	03.04.2014_KC_7850_BH12299	2390	-12.9	-33.1	181,307	Warr et al. (2018)	3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Canada	Kidd Creek Mine, Timmins	03.04.2014_KC_7850_BH12262	2390	-12.7	-29.4	165,074	Unpub.	3
Canada	Kidd Creek Mine, Timmins	22.10.2015_KCL7850_BH12299	2390	-11.9	-36.5	141,712	Unpub.	3
Canada	Kidd Creek Mine, Timmins	22.10.2015_KCL7850_BH12262	2390	-11.9	-33.4	170,654	Unpub.	3
Canada	Kidd Creek Mine, Timmins	22.10.2015_KCL7850_BH12287A	2390	-12.4	-35.4	nm	Unpub.	3
Canada	Kidd Creek Mine, Timmins	12.07.2016_KCL7850_BH12299	2390	-12.7	-34.0	177,280	Lollar et al. (2019)	3
Canada	Kidd Creek Mine, Timmins	12.07.2016_KCL7850_BH12261	2390	-12.7	-33.5	203,620	Lollar et al. (2019)	3
Canada	Kidd Creek Mine, Timmins	24.01.2017_KCL7850_BH12299	2390	-12.4	-34.2	168,650	Unpub.	3
Canada	Kidd Creek Mine, Timmins	24.01.2017_KCL7850_BH12261	2390	-12.4	-33.6	205,520	Unpub.	3
Canada	Kidd Creek Mine, Timmins	25.01.2017_KCL7850_BH12287A	2390	-12.8	-38.1	137,640	Unpub.	3
Canada	Kidd Creek Mine, Timmins	06.06.2017_KCL7850_BH12299	2390	-11.1	-34.1	172,420	Unpub.	3
Canada	Kidd Creek Mine, Timmins	06.06.2017_KCL7850_BH12261	2390	-11.6	-33.9	192,330	Unpub.	3
Canada	Kidd Creek Mine, Timmins	07.06.2017_KCL7850_BH12287A	2390	-10.9	-35.4	147,880	Unpub.	3
Canada	Kidd Creek Mine, Timmins	29.01.2018_KCL7850_BH12299	2390	-12.4	-35.5	179,800	Unpub.	3
Canada	Kidd Creek Mine, Timmins	14.6.2018-KC7850-12299	2390	-12.6	-38.0	189,220	Unpub.	3
Canada	Kidd Creek Mine, Timmins	14.6.2018-KC7850-12261	2390	-12.7	-35.4	206,700	Unpub.	3
Canada	Kidd Creek Mine, Timmins	05.02.2019_KC7850_12299	2390	-13.1	-33.9	186,570	Unpub.	3
Canada	Kidd Creek Mine, Timmins	05.02.2019_KC7850_12261	2390	-13.0	-33.0	216,290	Unpub.	3
Canada	Kidd Creek Mine, Timmins	10.05.06_KC_8798_980R_11404	2680	-12.0	-59.3	134,109	Unpub.	3
Canada	Kidd Creek Mine, Timmins	01.04.2009_KC_9500_12612A	2740	-9.0	-86.8	1449	Unpub.	3
Canada	Kidd Creek Mine, Timmins	29.02.2012_KC_9500_13677	2900	-10.1	-81.0	3335	Unpub.	3
Canada	Kidd Creek Mine, Timmins	14.06.2012KCL9500_BH13684	2900	-17.5	-30.9	209,042	Unpub.	3
Canada	Kidd Creek Mine, Timmins	14.06.2012KCL9500_BH13762	2900	-16.6	-29.3	215,148	Unpub.	3
Canada	Kidd Creek Mine, Timmins	29.11.2012KCL9500_BH13762	2900	-9.4	-73.8	7527	Unpub.	3
Canada	Kidd Creek Mine, Timmins	16.01.2013KCL9500_BH13762	2900	-9.7	-77.3	2737	Unpub.	3
Canada	Kidd Creek Mine, Timmins	16.01.2013KCL9500_BH2	2900	-13.8	-28.4	nm	Unpub.	3
Canada	Kidd Creek Mine, Timmins	16.01.2013KCL9500_BH4	2900	-11.9	-49.3	nm	Unpub.	3
Canada	Kidd Creek Mine, Timmins	17.01.2013KCL9200_BH1	2805	-9.5	-76.5	4415	Unpub.	3
Canada	Kidd Creek Mine, Timmins	17.01.2013KCL9200_BH2	2805	-7.9	-69.2	nm	Unpub.	3
Canada	Kidd Creek Mine, Timmins	26.09.2013_KC_9500_BH13879	2900	-11.8	-52.8	nm	Unpub.	3
Canada	Lockerby Mine	L40139	4000	-11.7	-84.0	4538	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L40140	4000	-12.1	-79.0	23,170	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L40141	4000	-11.6	-80.0	6862	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L40143	4000	-11.7	-80.0	13,128	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L26000	2600	-11.5	-85.0	128	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L261350	2600	-12.8	-92.0	1964	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L261400	2600	nm	nm	nm	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L261450	2600	-11.2	-86.0	2596	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L261650	2600	-11.6	-92.0	1061		3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Canada	Lockerby Mine	L261750	2600	-11.7	-80.0	nm	Frape and Fritz (1982)	3
Canada	Lockerby Mine	L22450	2600	nm	-81.0	1282	Frape and Fritz (1982)	3
Canada	Lupin Mine	BH 188	890	-22.4	-178.0	3600	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 188	890	-22.1	-175.0	2800	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 188	890	-22.4	-180.0	3100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 188, pressurized	890	-22.8	-176.0	3000	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 188, pressurized	890	-22.9	-179.0	2700	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 188, pressurized	890	-22.5	-182.0	3200	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 188, pressurized	890	-22.8	-178.0	2800	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 188, pressurized	890	-22.7	-179.0	2800	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 64	1130	-23.3	-181.0	2200	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 64	1130	-23.0	-184.0	2800	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 64	1130	-23.4	-181.0	2600	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 64	1130	-23.3	-182.0	nm	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 175, flowing	1130	-22.6	-177.0	26,200	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 175, flowing	1130	-22.8	-174.0	27,500	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 175, flowing	1130	-22.5	-173.0	32,100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 191	1130	-23.2	-179.0	7100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 191	1130	-23.2	-179.0	7700	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 191, pressurized	1130	-23.4	-180.0	5800	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 191, pressurized	1130	-23.0	-179.0	13,700	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 191, pressurized	1130	-23.0	-182.0	7000	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 192	1130	-22.6	-181.0	30,700	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 192	1130	-22.8	-175.0	29,700	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 192, pressurized	1130	-22.7	-175.0	36,300	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 192, pressurized	1130	-22.2	-176.0	40,000	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 192, pressurized	1130	-22.5	-172.0	39,900	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 192, pressurized	1130	-22.6	-173.0	39,600	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 197	1130	-22.8	-186.0	9100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 197, pressurized	1130	-23.0	-178.0	8600	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 197, pressurized	1130	-22.9	-179.0	5900	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 197, pressurized	1130	-22.4	-180.0	9600	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 197, pressurized	1130	-22.8	-178.0	6400	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 197, pressurized	1130	-22.7	-179.0	5900	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 217, pressurized	1130	-23.5	-179.0	8500	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 217, pressurized	1130	-23.2	-180.0	15,500	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 217, pressurized	1130	-22.9	-182.0	7000	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 217, pressurized	1130	-23.1	-179.0	13,200	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 219, pressurized	1130	-23.0	-179.0	10,100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 219, pressurized	1130	-22.9	-179.0	20,600	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 219, pressurized	1130	-22.9	-178.0	8100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 260	1130	-23.0	-177.0	15,900	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 260	1130	-23.0	-183.0	8800	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 260	1130	-23.1	-179.0	12,100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 267, 3–24 m	1130	-23.1	-185.0	4100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 267, 45–66 m	1130	-23.1	-186.0	4700	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 267, 187–269 m	1130	-23.0	-185.0	5600	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 267, pressurized	1130	-23.4	-180.0	6100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 267, leaking	1130	-23.2	-183.0	4800	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 273, pressurized	1130	-22.8	-182.0	11,900	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 273, pressurized	1130	-23.2	-178.0	7900	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 273, pressurized	1130	-23.2	-179.0	10,200	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 273, pressurized	1130	-23.0	-180.0	8300	Stotler et al. (2009)	3
Canada	Lupin Mine	Drips, Bolthole	1130	-23.4	-180.0	3900	Stotler et al. (2009)	3
Canada	Lupin Mine	Drips, Bolthole	1130	-22.9	-184.0	4100	Stotler et al. (2009)	3
Canada	Lupin Mine	BH 105	1300	-22.8	-180.0	9100	Stotler et al. (2009)	3
Canada	Lupin Mine	Purge 1	535	-17.7	-147.0	107,187	Stotler et al. (2011)	2
Canada	Lupin Mine	Purge 4	535	-22.2	-175.0	29,735	Stotler et al. (2011)	2
Canada	Lupin Mine	Purge 6	535	-22.5	-177.0	22,077	Stotler et al. (2011)	2
Canada	Lupin Mine	Purge 7	535	-22.6	-177.0	21,240	Stotler et al. (2011)	2
Canada	Lupin Mine	Lu	nm	-21.3	-167.0	12,469	Bottomley et al. (1994)	3
Canada	Matagami	R28	0	-13.9	-107.0	26,692	Unpub.	3
Canada	Matagami	R36	0	-11.5	-60.8	223,635	Unpub.	3
Canada	Matagami	-25	360	-14.1	-95.9	752	Unpub.	3
Canada	Matagami	-25'	360	-14.4	-107.5	386	Unpub.	3
Canada	Matagami	-35	360	-14.1	-85.6	1329	Unpub.	3
Canada	Matagami	Wall	360	-14.5	-105.0	1035	Unpub.	3
Canada	Matagami	Drip #1	610	-14.4	-98.8	458	Unpub.	3
Canada	Nickel Rim South Mine, Sudbury	29.11.2013_NR_1730m_NR170128	1730	-11.9	-30.9	47,645	Unpub.	3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Canada	Nickel Rim South Mine, Sudbury	29.11.2013_NR_550SumpStation_NR170155	1730	-11.4	-31.9	42,055	Unpub.	3
Canada	Nickel Rim South Mine, Sudbury	05.03.2014_NR_1730m_NR170128	1730	-11.5	-30.3	53,365	Unpub.	3
Canada	Nickel Rim South Mine, Sudbury	22.10.2014_NR_1730m_NR170128	1730	-13.0	-35.6	71,412	Unpub.	3
Canada	Nickel Rim South Mine, Sudbury	22.10.2014_NR_1730m_NR170183	1730	-12.1	-30.2	nm	Unpub.	3
Canada	Nickel Rim South Mine, Sudbury	22.10.2014_NR_1730m_NR170185	1730	-12.2	-30.2	98,934	Unpub.	3
Canada	Nickel Rim South Mine, Sudbury	22.10.2014_NR_1730m_NR170215	1730	-11.0	-46.7	37,960	Unpub.	3
Canada	Nickel Rim South Mine, Sudbury	22.10.2014_NR_1730m_NR170216	1730	-11.1	-48.9	26,585	Unpub.	3
Canada	Norita		nm	-14.0	-99.6	14,756	Unpub.	3
Canada	Norita		nm	-13.8	-90.0	105,320	Unpub.	3
Canada	Norita		61	-13.5	-92.5	121,829	Unpub.	3
Canada	Norita		76	-13.7	-85.5	128,469	Unpub.	3
Canada	Norita		91	-13.5	-89.4	141,184	Unpub.	3
Canada	Norita		230	-13.2	-70.0	213,644	Unpub.	3
Canada	Norita		244	-13.4	-70.8	217,951	Unpub.	3
Canada	North Mine	N3601	3600	-13.0	-88.0	5303	Frape and Fritz (1982)	3
Canada	North Mine	N3602	3600	-13.1	-88.0	6290	Frape and Fritz (1982)	3
Canada	North Mine	N3640	3600	-11.0	-40.0	225,446	Frape and Fritz (1982)	3
Canada	North Mine	N3643	3600	-13.4	-94.0	4113	Frape and Fritz (1982)	3
Canada	North Mine	N3644	3600	-12.0	-83.0	264	Frape and Fritz (1982)	3
Canada	North Mine	N3651	3600	-11.3	-33.0	240,632	Frape and Fritz (1982)	3
Canada	Red Lake, ON, Campbell Mine		1220	-12.2	-58.2	181,744	Unpub.	3
Canada	Red Lake, ON, Campbell Mine		1220	-11.5	-49.3	301,535	Unpub.	3
Canada	Red Lake, ON, Campbell Mine		1220	-12.0	-59.2	231,428	Unpub.	3
Canada	Red Lake, ON, Campbell Mine		640	-13.4	-105.5	783	Unpub.	3
Canada	Red Lake, ON, Campbell Mine		640	-13.5	-106.2	715	Unpub.	3
Canada	Red Lake, ON, Campbell Mine		640	-14.3	-110.4	2070	Unpub.	3
Canada	Red Lake, ON, Campbell Mine		640	-15.1	-113.4	2165	Unpub.	3
Canada	Sudbury	S-138	1600	-10.9	-44.0	252,000	Frape et al. (1984)	3
Canada	Sudbury	S-139	1600	-10.3	-42.0	240,000	Frape et al. (1984)	3
Canada	Sudbury	S-144	1650	-10.8	-45.0	254,000	Frape et al. (1984)	3
Canada	Sudbury	S-147	1600	-10.9	-39.0	249,121	Frape et al. (1984)	3
Canada	Sudbury	S-149	1600	-11.3	-33.0	240,632	Frape et al. (1984)	3
Canada	Sudbury	S-176	488	-11.9	-82.0	4278	Frape et al. (1984)	3
Canada	Sudbury	S-160	50	-11.5	-85.0	127	Frape et al. (1984)	3
Canada	Sudbury	S-178	152	-12.9	-88.0	186	Frape et al. (1984)	3
Canada	Sudbury	S-145	1100	-11.9	-73.0	83,373	Frape et al. (1984)	3
Canada	Sudbury	S-154	1006	-12.8	-76.0	18,812	Frape et al. (1984)	3
Canada	Sudbury	S-175	991	-10.4	-43.0	95,039	Frape et al. (1984)	3
Canada	Sudbury		nm	-11.3	-33.0	240,632	Kelly et al. (1986)	3
Canada	Thompson	T-93	1500	-11.7	-53.0	325,000	Frape et al. (1984)	3
Canada	Thompson	T-94	1220	-14.4	-79.0	182,600	Frape et al. (1984)	3
Canada	Thompson	T-95	1220	-15.6	-92.0	101,433	Frape et al. (1984)	3
Canada	Thompson	T-97	1300	-13.2	-77.0	117,292	Frape et al. (1984)	3
Canada	Thompson	T-86	610	-17.9	-127.0	20,338	Frape et al. (1984)	3
Canada	Thompson	T-87	671	-17.0	-115.0	22,495	Frape et al. (1984)	3
Canada	Thompson	T-92	1220	-16.0	-102.0	65,437	Frape et al. (1984)	3
Canada	Thompson	T-82	305	-16.3	-128.0	1798	Frape et al. (1984)	3
Canada	Thompson	T-83	305	-17.4	-131.0	5371	Frape et al. (1984)	3
Canada	Thompson	T-79	100	-18.9	-149.0	580	Frape et al. (1984)	3
Canada	Thompson	T-80	152	-18.1	-137.0	589	Frape et al. (1984)	3
Canada	Thompson	T-96	50	-13.7	-115.0	117	Frape et al. (1984)	3
Canada	Thompson Mine, Thompson	12.10.2006_TH3600L_Sta7_ "BH#7"	1200	-13.0	-64.8	nm	Unpub.	3
Canada	Thompson Mine, Thompson	12.10.2006_TH3600L_Sta7_ "BH#4"	1200	-14.3	-81.0	156,417	Unpub.	3
Canada	Thompson Mine, Thompson	12.10.2006_TH3600L_Sta7_ "BH#1"	1200	-14.6	-80.2	203,508	Unpub.	3
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta389_BH106583	1200	-15.0	-97.4	nm	Unpub.	3
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta390	1200	-14.5	-85.7	nm	Unpub.	3
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta396	1200	-14.7	-85.6	nm	Unpub.	3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta396_BH116385	1200	-17.3	-91.1	nm	Unpub.	3
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta392	1200	-14.7	-90.7	nm	Unpub.	3
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta396_BH106547	1200	-13.6	-92.8	nm	Unpub.	3
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta391	1200	-14.6	-85.6	nm	Unpub.	3
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta396_BH116386	1200	-16.1	-99.4	nm	Unpub.	3
Canada	Thompson Mine, Thompson	11.10.2006_TH3500L_Sta390_BH116392	1200	-14.8	-85.7	157,753	Unpub.	3
Canada	Thompson Mine, Thompson	1163630 13.06.06	1200	-13.3	-97.4	39,692	Telling et al. (2018)	3
Canada	Thompson Mine, Thompson	1065750 13.06.06	1200	-14.7	-82.8	204,755	Telling et al. (2018)	3
Canada	Thompson Mine, Thompson	1065800 13.06.06	1200	-13.3	-104.2	nm	Telling et al. (2018)	3
Canada	Thompson Mine, Thompson	1065760 14.06.06	1200	-14.5	-94.8	nm	Telling et al., 2018	3
Canada	Thompson Mine, Thompson	1163930 14.06.06	1200	-15.3	-98.5	83,452	Telling et al. (2018)	3
Canada	Thompson No. 1 Shaft	T1	nm	-13.4	-94.0	nm	Bottomley et al. (1994)	3
Canada	Thompson No. 3 Shaft	T3	nm	-15.5	-106.0	nm	Bottomley et al. (1994)	3
Canada	Thompson	777 m: no # (10/97)	777	-12.4	-111	367	Bottomley and Clark (2004)	3
Canada	Victor Mine	1463 m: B-99162 (03/98)	1463	-11.7	-31	98,790	Bottomley and Clark (2004)	3
Canada	Victor Mine	1463 m: B-99160 (03/98)	1463	-11.7	-32	113,309	Bottomley and Clark (2004)	3
Canada	Yellowknife	YK-62	130	-19.6	-161.0	850	Frape et al. (1984)	3
Canada	Yellowknife	YK-65	175	-19.7	-153.0	740	Frape et al. (1984)	3
Canada	Yellowknife	YK-17	1500	-14.4	-71.0	237,107	Frape et al. (1984)	3
Canada	Yellowknife	YK-15	1372	-19.1	-123.0	108,799	Frape et al. (1984)	3
Canada	Yellowknife	YK-32	1372	-16.9	-91.0	193,000	Frape et al. (1984)	3
Canada	Yellowknife	YK-33	1372	-18.1	-105.0	172,900	Frape et al. (1984)	3
Canada	Yellowknife	YK-34	550	-20.4	-162.0	1770	Frape et al. (1984)	3
Canada	Yellowknife	YK-71	335	-19.2	-163.0	4223	Frape et al. (1984)	3
Canada	Yellowknife	YK-73	457	-19.1	-152.0	8831	Frape et al. (1984)	3
Canada	Yellowknife	YK-9	1372	-19.9	-145.0	58,337	Frape et al. (1984)	3
Canada	Yellowknife	YK-10	1372	-23.0	-163.0	63,136	Frape et al. (1984)	3
Canada	Yellowknife	YK-22	1494	-19.1	-140.0	20,481	Frape et al. (1984)	3
Canada	Yellowknife	nm	nm	-14.4	-73.0	261,478	Kelly et al. (1986)	3
Canada	Yellowknife	1067 m: seep (02/95)	1067	-19.3	-157	336.6	Bottomley and Clark (2004)	3
Canada	Yellowknife	1067 m: B-8906 (09/00)	1067	-19.4	-154	538.6	Bottomley and Clark (2004)	3
Canada	Yellowknife	1067 m: B-9452 (09/00)	1067	-19.8	-157	4105.5	Bottomley and Clark (2004)	3
Canada	Yellowknife	1067 m: B-9362 (09/00)	1067	-20.9	-162	5725.6	Bottomley and Clark (2004)	3
Canada	Yellowknife	1372 m: B-3457 (02/95)	1372	-20.4	-161	10,086.6	Bottomley and Clark (2004)	3
Canada	Yellowknife	1372 m: B-5316 (09/00)	1372	-20.4	-161	11,534.5	Bottomley and Clark (2004)	3
Canada	Yellowknife	1372 m: B-5318 (09/00)	1372	-19.2	-150	30,202.4	Bottomley and Clark (2004)	3
Canada	Yellowknife	1372 m: B-3316 (02/95)	1372	-17.2	-103	106,202.5	Bottomley and Clark (2004)	3
Canada	Yellowknife	1494 m: B-5310 (09/00)	1494	-19.4	-134	70,499.2	Bottomley and Clark (2004)	3
Canada	Yellowknife	1494 m: seep (02/95)	1494	-17.6	-111	102,588.4	Bottomley and Clark (2004)	3
Canada	Yellowknife	1616 m: B-7126 (09/00)	1616	-13.6	-73	165,114.7	Bottomley and Clark (2004)	3
Canada	Yellowknife	1616 m: B-6790 (09/00)	1616	-14.3	-73	170,969.3	Bottomley and Clark (2004)	3
Canada	Yellowknife	1616 m: B-5586 (06/96)	1616	-14.4	-69	175,933.7	Bottomley and Clark (2004)	3
Canada	Yellowknife	1616 m: B-6709 (06/96)	1616	-13.6	-62	190,390.7	Bottomley and Clark (2004)	3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
S. Africa	Beatrix	BE224 FW 032601C18W17	768.0	-6.13	-42.7	2276	Lippmann et al. (2003)	3
S. Africa	Beatrix	BE325 H1 032701 CTS	1290.0	-6.71	-43.2	2424	Lippmann et al. (2003)	3
S. Africa	Beatrix	BE327 H3 032701 CTS	1390.0	-6.66	-41.2	2772	Lippmann et al. (2003)	3
S. Africa	Beatrix	BE116 H1 031601 GDW	866.0	-5.85	-42.33	2012	Lippmann et al. (2003)	3
S. Africa	Beatrix	BE116 FW 031401 IDW	866.0	-5.49	-40.1	2244	Lippmann et al. (2003)	3
S. Africa	Beatrix	BE23FW-A4RD	718.0	-6.2	-41.0	nm	Lippmann et al. (2003)	3
S. Africa	Beatrix	BE116 FW 121801	866.0	-6.86	-40.7	2015	Onstott et al. (2006)	3
S. Africa	Beatrix	BE116 GS 031401 EDW	866.0	-6.00	-41.5	1262	Onstott et al. (2006)	3
S. Africa	Beatrix	BE116 H2 031601 GDW	866.0	-4.46	-39.9	nm	Onstott et al. (2006)	3
S. Africa	Beatrix	BE116 H3 031601 GDW	866.0	-6.24	-40.9	nm	Onstott et al. (2006)	3
S. Africa	Beatrix	BE223 FW 031301 A4RD	718.0	-6.37	-39.7	nm	Onstott et al. (2006)	3
S. Africa	Beatrix	BE324 FW 121801	1240.0	-6.88	-41.0	2428	Onstott et al. (2006)	3
S. Africa	Beatrix	BE326 Bh2 2011	1330.0	-5.9	-41.0	4470	Lau et al. (2014)	3
S. Africa	Beatrix	BE326 Bh2 2012	1330.0	-8.7	-47.0	nm	Lau et al. (2014)	3
S. Africa	Beatrix	BE326 Bh2 2012	1330.0	-6.4	-41.4	3590	Simkus et al. (2016)	3
S. Africa	Beatrix	BE326 Bh1	1330.0	-6.6	-41.7	3410	Simkus et al. 2016	3
S. Africa	Driefontein	DR546 BH1 120798	3213.0	-13.14	-24.4	35,041	Lippmann et al. (2003)	3
S. Africa	Driefontein	DR4 IPC H1 090198	945.0	-4.55	-20.3	238	Onstott et al. (2006)	3
S. Africa	Driefontein	DR4 IPC HI 082902	945.0	-3.58	-18.0	176	Onstott et al. (2006)	3
S. Africa	Driefontein	DR548 FW 090901	3300.0	-12.29	-10.3	101,074	Onstott et al. (2006)	3
S. Africa	Driefontein	DR638 FW 111598	2700.0	-6.20	-34.0	885	Onstott et al. (2006)	3
S. Africa	Driefontein	DR938 H1 082001	2716.0	-4.31	-26.7	211	Onstott et al. (2006)	3
S. Africa	Driefontein	DR938 H3 083001	2716.0	-5.01	-27.9	1826	Onstott et al. (2006)	3
S. Africa	Driefontein	DR938 H3 071202	3350.0	-5.64	-26.9	1696	Onstott et al. (2006)	3
S. Africa	Driefontein	DR938 H3 082802	2712.0	-5.80	-27.7	1687	Onstott et al. (2006)	3
S. Africa	Driefontein	DR938 H4 110201	2176.0	-4.68	-26.3	127	Onstott et al. (2006)	3
S. Africa	Driefontein	DR940 FW 092602	2812.0	-4.78	-27.2	264	Onstott et al. (2006)	3
S. Africa	Driefontein	DR51PCFW280711	1046.0	-4.3	-24.6	nm	Lau et al. (2014)	3
S. Africa	Driefontein	DR51PC	1050.0	-4.4	-24.3	190	Simkus et al. (2016)	3
S. Africa	Evander	EV522 FW 030801HWD	1694.0	-5.63	-27.7	2330	Lippmann et al. (2003)	3
S. Africa	Evander	EV818 FW 030601	1950.0	-10.24	-35.9	9106	Lippmann et al. (2003)	3
S. Africa	Evander	EV811 FW 030501	1340.0	-5.02	-30.0	612	Lippmann et al. (2003)	3
S. Africa	Evander	EV818 FW 062101	1950.0	-10.87	-36.9	8544	Lippmann et al. (2003)	3
S. Africa	Evander	EV219 H1 030901 ED	1474.0	-4.98	-30.1	1364	Onstott et al. (2006)	3
S. Africa	Evander	EV219 H5 030901 ED	1474.0	-4.92	-27.6	810	Onstott et al. (2006)	3
S. Africa	Evander	EV221 H1 052102	1624.0	-5.86	-29.7	1048	Onstott et al. (2006)	3
S. Africa	Evander	EV221 H2 111601	1624.0	-4.37	-17.9	2732	Onstott et al. (2006)	3
S. Africa	Evander	EV221 H3 111601	1624.0	-5.97	-30.0	1682	Onstott et al. (2006)	3
S. Africa	Evander	EV522 H1 041801CTS	1694.0	-6.48	-30.9	3776	Onstott et al. (2006)	3
S. Africa	Evander	EV522 H2 041801CTS	1694.0	-6.19	-30.4	nm		3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
S. Africa	Evander	EV813 FW 030601	1410.0	-4.75	-30.3	nm	Onstott et al. (2006)	3
S. Africa	Evander	EV818 FW 030601	1950.0	-10.24	-35.9	9106	Onstott et al. (2006)	3
S. Africa	Evander	EV818 FW 051001	1950.0	-10.50	-36.2	8770	Onstott et al. (2006)	3
S. Africa	Evander	EV820 FW 061802	2005.0	-6.74	-30.0	2115	Onstott et al. (2006)	3
S. Africa	Evander	EV820 FW 121401	2005.0	-6.75	-31.2	2530	Onstott et al. (2006)	3
S. Africa	Evander	EV821 FW 101601	2055.0	-7.44	-29.1	5292	Onstott et al. (2006)	3
S. Africa	Evander	EV914 FW 022801 ED	1400.0	-4.68	-31.2	632	Onstott et al. (2006)	3
S. Africa	Kloof	KL739 FW 062901	3100.0	-7.24	-20.4	13,218	Lippmann et al. (2003)	3
S. Africa	Kloof	KL441 HWDS H1 120198	3082.0	-5.71	-27.7	3221	Lippmann et al. (2003)	3
S. Africa	Kloof	KL441 HWDN FW 022801	3300.0	-9.88	-28.8	9004	Onstott et al. (2006)	3
S. Africa	Kloof	KL441 HWDS H1 050201	3300.0	-4.39	-30.9	2236	Onstott et al. (2006)	3
S. Africa	Kloof	KL441 HWDS H2 050201	3300.0	-4.80	-27.9	2523	Onstott et al. (2006)	3
S. Africa	Kloof	KL441 HWDS H2 100201	3300.0	-5.04	-27.9	2315	Onstott et al. (2006)	3
S. Africa	Kloof	KL441 HWDS H3 111401	3300.0	-5.03	-27.5	2037	Onstott et al. (2006)	3
S. Africa	Kloof	KL443 HWDN FW 030501	3400.0	-10.48	-29.2	nm	Onstott et al. (2006)	3
S. Africa	Kloof	KL443 HWDN FW 050801	3400.0	-11.10	-33.8	16,447	Onstott et al. (2006)	3
S. Africa	Kloof	KL445	3280.0	-9.4	-31.8	11,100	Simkus et al. (2016)	3
S. Africa	Masimong	MM51870 FW 030402	1880.0	-6.52	-38.0	2670	Onstott et al. (2006)	3
S. Africa	Masimong	MM51940 FW 200712	1900.0	-7.0	-40.0	nm	Lau et al. (2014)	3
S. Africa	Masimong	MM51940	1900.0	-7.0	-40.1	nm	Heard et al. (2018)	3
S. Africa	Middelbult	MB FW 040301 Sec33	200.0	-3.43	-27.6	nm	Onstott et al. (2006)	3
S. Africa	Middelbult	MB FW 040301 Sec49	200.0	-3.06	-25.5	776	Onstott et al. (2006)	3
S. Africa	Moab	MK1200-21/06/18 180—2H H2O	nm	-4.0	-23.9	nm	Unpub.	3
S. Africa	Moab	MK101-16/06/18 180—2H H2O	nm	-12.3	-22.8	nm	Unpub.	3
S. Africa	Moab	MK070818 101 LVL WATER ISOTOPES	nm	-11.9	-8.4	nm	Unpub.	3
S. Africa	Moab	MK190618 101 LVL WATER ISOTOPES	nm	-11.8	-9.4	nm	Unpub.	3
S. Africa	Moab	MK210618 1200 level WATER ISOTOPES	nm	-4.0	-24.1	nm	Unpub.	3
S. Africa	Moab	MK231018 95 LVL WATER ISOTOPES	nm	-11.7	-26.4	nm	Unpub.	3
S. Africa	Moab	11.07.19_MKL95_LibA_D0	nm	-12.2	-25.4	nm	Unpub.	3
S. Africa	Moab	17.07.19_MKL95_LibA_D0	nm	-12.3	-25.8	nm	Unpub.	3
S. Africa	Moab	10.07.19_MKL101_Barmenco	nm	-12.4	-16.8	nm	Unpub.	3
S. Africa	Mponeng	MP104 E65XC H1	2825.0	-6.82	-29.1	3146	Onstott et al. (2006)	3
S. Africa	Mponeng	MP104 E65XC H1	2825.0	-6.92	-29.1	4236	Onstott et al. (2006)	3
S. Africa	Mponeng	MP109 BH1 070196	3198.0	-5.25	-27.4	676	Onstott et al. (2006)	3
S. Africa	Mponeng	MP109 FW 101701	3305.0	-6.96	-30.4	11,738	Onstott et al. (2006)	3
S. Africa	Mponeng	MS151 FW 022202	2028.0	-7.19	-43.4	3865	Onstott et al. (2006)	3
S. Africa	Tau Tona	TT100 FW 082702	3025.0	-4.89	-25.7	185	Onstott et al. (2006)	3
S. Africa	Tau Tona	TT107FR240811	3048.0	-5.1	-22.4	nm	Lau et al. (2014)	3
S. Africa	Tau Tona	TT109FW060312BH2	3136.0	-5.0	-25.3	nm	Lau et al. (2014)	3
S. Africa	Tau Tona	TT109 Bh2	3140.0	-5.0	-25.3	300	Simkus et al. (2016)	3
S. Africa	Tau Tona	TT107	3050.0	-4.8	-24.4	200	Simkus et al. (2016)	3
Finland	Enonkoski	17092	0.0	-13.7	-91.4	17,385	Unpub.	3
Finland	Enonkoski	17093	14.0	-13.6	-90.9	18,026	Unpub.	3
Finland	Enonkoski	17094	0.0	-13.5	-90.4	18,916	Unpub.	3
Finland	Enonkoski	17095	14.0	-13.2	-81.2	27,635	Frape et al. (2003)	3

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Finland	Enonkoski	17106	0.0	-13.9	-98.4	13,749	Unpub.	3
Finland	Enonkoski	17107	101.2	-14.5	-98.2	11,708	Frape et al. (2003)	3
Finland	Enonkoski	17108	200.0	-13.2	-83.5	20,119	Unpub.	3
Finland	Enonkoski	17109	301.5	-13.3	-83.6	25,049	Unpub.	3
Finland	Enonkoski	17110	405.8	-13.4	-82.4	23,734	Unpub.	3
Sweden	Klipperas	GTL K1 9	11.0	-11.7	-93.8	400	Unpub.	2
Sweden	Klipperas	GTL K1 9	61.0	-13.5	-105.8	5890	Unpub.	2
Finland	Kerimäki		140.0	-13.4	-94.4	60	Nurmi et al. (1988)	2
Finland	Kerimäki		540.0	-13.5	-93.9	3900	Nurmi et al. (1988)	2
Finland	Kerimäki		700.0	-13.7	-95.4	5300	Nurmi et al. (1988)	2
Finland	Liminka		100.0	-14.5	-106.0	6200	Nurmi et al. (1988)	2
Finland	Liminka		400.0	-12.7	-88.8	16,100	Nurmi et al. (1988)	2
Finland	Liminka		600.0	-12.2	-84.2	37,200	Nurmi et al. (1988)	2
Finland	Liminka		720.0	-12.1	-81.0	41,000	Nurmi et al. (1988)	2
Finland	Miihkali		250.0	-10.4	-30.3	40,077	Unpub.	2
Finland	Miihkali		502.0	-10.4	-28.2	40,680	Frape et al. (2003)	2
Finland	Miihkali		745.0	-10.2	-20.8	38,516	Unpub.	2
Finland	Miihkali		889.0	-11.8	-15.8	146,561	Unpub.	2
Finland	Miihkali		90.0	-10.5	-18.7	38,764	Unpub.	2
Finland	Miihkali		340.0	-10.5	-19.2	41,860	Unpub.	2
Finland	Miihkali		590.0	-10.2	-21.2	40,350	Unpub.	2
Finland	Miihkali		790.0	-10.8	-21.1	64,559	Unpub.	2
Finland	Miihkali		940.0	-11.7	-25.7	147,271	Unpub.	2
Finland	Miihkali		990.0	-11.8	-18.0	148,435	Unpub.	2
Finland	Miihkali		1040.0	-12.4	-22.7	nm	Unpub.	2
Finland	Miihkali		1100.0	-11.7	-11.8	149,380	Unpub.	2
Finland	Miihkali		130–180	-10.1	-9.4	nm	Unpub.	2
Finland	Miihkali		180–230	-10.3	-10.7	nm	Unpub.	2
Finland	Miihkali		330–380	-10.4	-11.5	nm	Unpub.	2
Finland	Miihkali		380–430	-10.4	-11.2	nm	Unpub.	2
Finland	Miihkali		530–580	-10.3	-10.2	nm	Unpub.	2
Finland	Miihkali		580–630	-10.4	-11.2	nm	Unpub.	2
Finland	Miihkali		780–830	-10.6	-10.2	nm	Unpub.	2
Finland	Miihkali		830–880	-11.8	-10.9	nm	Unpub.	2
Finland	Miihkali		930–980	-11.7	-9.1	nm	Unpub.	2
Finland	Miihkali		980–1030	-11.7	-10.7	nm	Unpub.	2
Finland	Miihkali	17071	15.0	-10.3	-15.8	37,676	Unpub.	2
Finland	Miihkali	17072	102.1	-10.4	-14.8	35,139	Unpub.	2
Finland	Miihkali	17073	252.7	-10.5	-12.7	34,662	Unpub.	2
Finland	Miihkali	17074	500.9	-10.5	-13.1	33,791	Unpub.	2
Finland	Miihkali	17075	750.0	-10.4	-12.5	34,275	Frape et al. (2003)	2
Finland	Miihkali	17076	966.7	-12.0	-7.6	136,132	Unpub.	2
Finland	Miihkali	17077	15.0	-12.9	-78.3	9536	Unpub.	2
Finland	Miihkali	17083	10.0	-13.5	-101.3	110	Unpub.	2
Finland	Miihkali	17084	200.6	-13.4	-99.2	132	Unpub.	2
Finland	Miihkali	17085	300.6	-13.0	-87.6	10,288	Unpub.	2
Finland	Miihkali	17086	401.2	-13.1	-87.5	10,956	Unpub.	2
Finland	Miihkali	17079	15.0	-13.5	-94.2	9266	Unpub.	2
Finland	Miihkali	17078	15.0	-13.8	-101.7	nm	Unpub.	2
Finland	Miihkali	17080	15.0	-13.3	-99.1	174	Unpub.	2
Finland	Miihkali	17081	0.0	-11.7	-95.2	nm	Unpub.	2
Finland	Noormarkku	17063	13.0	-11.3	-85.9	1643	Unpub.	2
Finland	Noormarkku	17064	101.5	-11.6	-89.0	7613	Unpub.	2
Finland	Noormarkku	17065	201.2	-11.6	-89.0	10,521	Unpub.	2
Finland	Noormarkku	17066	300.9	-12.3	-92.2	15,572	Unpub.	2
Finland	Noormarkku	17067	404.6	-12.3	-92.2	17,903	Unpub.	2
Finland	Noormarkku	17068	500.0	-11.2	-78.9	37,418	Unpub.	2
Finland	Noormarkku	17069	533.3	-11.1	-76.7	40,769	Unpub.	2
Finland	Noormarkku	17070	559.4	-11.3	-77.8	37,835	Unpub.	2
Finland	Vammala	17089	0.0	-13.8	-104.0	2857	Unpub.	3
Finland	Vammala	17090	0.0	-14.6	-109.3	1782	Unpub.	3
Finland	Vammala	17091	0.0	-15.1	-110.8	1815	Unpub.	3
Finland	Outukumpu	17087	15.0	-13.9	-105.4	80	Unpub.	2
Finland	Outukumpu	17088	450.6	-13.8	-106.3	390	Unpub.	2
Finland	Outukumpu		270.0	-13.9	-98.8	100	Nurmi et al. (1988)	2
Finland	Outukumpu		600.0	-14.4	-102.1	13,400	Nurmi et al. (1988)	2
Finland	Outukumpu		770.0	-14.3	-101.5	13,700	Nurmi et al. (1988)	2
Finland	Outukumpu		1010.0	-13.0	-74.8	26,500	Nurmi et al. (1988)	2
Finland	Outukumpu		1070.0	-13.0	-73.4	27,100	Nurmi et al. (1988)	2
Finland	Outukumpu	S8706701	290.0	-14.4	-105.5	198	Unpub.	2
Finland	Outukumpu	S8706702	606.0	-14.3	-105.1	7570	Unpub.	2
Finland	Outukumpu	S8706703	787.0	-14.2	-105.3	7762	Unpub.	2
Finland	Outukumpu	S8706713	854.0	-13.5	-95.4	17,095	Unpub.	2
Finland	Outukumpu	S8706704	244.0	-13.9	-104.6	116	Unpub.	2
Finland	Outukumpu	S8706705	402.0	-14.1	-102.1	223	Unpub.	2

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Finland	Outukumpu	S8706706a	697.0	-14.4	-102.3	12,870	Frape et al. (2003)	2
Finland	Outukumpu	S8706706b	700.0	-14.4	-102.9	nm	Unpub.	2
Finland	Outukumpu	S8706707	1036.0	-12.9	-80.1	27,908	Unpub.	2
Finland	Outukumpu	S8706708	1178.0	-12.9	-80.1	28,122	Frape et al. (2003)	2
Finland	Outukumpu Deep Drill Hole	OK-1	25.0	-12.1	-79.0	12,900	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-2	100.0	-12.5	-82.0	10,900	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-3	200.0	-12.6	-83.0	11,800	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-4	300.0	-12.7	-83.0	11,200	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-5	400.0	-12.9	-83.0	12,600	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-6	500.0	-12.9	-83.0	12,100	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-7	600.0	-12.9	-83.0	12,200	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-8	700.0	-13.0	-83.0	12,700	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-9	800.0	-13.0	-83.0	13,100	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-10	900.0	-12.9	-83.0	13,100	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-11	1000.0	-12.9	-83.0	12,700	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-12	1100.0	-12.9	-83.0	13,700	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-13	1200.0	-12.8	-83.0	13,000	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-14	1300.0	-12.7	-82.0	12,400	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-15	1400.0	-12.2	-80.0	12,600	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-16	1500.0	-11.2	-73.0	14,700	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-17	1600.0	-10.9	-69.0	21,800	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-18	1700.0	-10.5	-61.0	35,400	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-19	1800.0	-10.4	-58.0	43,900	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-20	1900.0	-10.5	-57.0	48,700	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-21	2000.0	-10.5	-60.0	50,600	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-22	2100.0	-10.6	-62.0	46,200	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-23	2200.0	-10.6	-64.0	41,500	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-24	2300.0	-10.8	-65.0	42,300	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OK-25	2400.0	-11.0	-70.0	64,400	Kietäväinen et al. (2013)	2
Finland	Outukumpu Deep Drill Hole	OKU2260-G1	2260.0	-10.9	-70.7	nm	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU2260-G3	2260.0	-10.9	-70.7	nm	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU2260-G7	2260.0	-11.0	-69.2	nm	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU2260-G10	2260.0	-11.0	-69.2	nm	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU500-G3	500.0	-13.1	-82.8	11,000	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU500-G4	500.0	-13.1	-82.8	Nm	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU500-G8	500.0	-13.1	-82.9	11,000	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU500-G12	500.0	-13.1	-82.7	Nm	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU2300-G1	2300.0	-10.56	-65.4	Nm	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU1820-G1	1820.0	-10.47	-56.0	38,000	Kietäväinen et al. (2017)	1
Finland		OUTO-500 (6)	500.0	-13.11	-82.9	12,000		2

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Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Finland	Outukumpu Deep Drill Hole						Kietäväinen et al. (2017)	
Finland	Outukumpu Deep Drill Hole	OUTO-970 (2)	970.0	-12.94	-82.8	12,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OUTO-1470 (8)	1470.0	-10.84	-70.8	14,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OUTO-1820 (4)	1820.0	-10.59	-56.6	51,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OUTO-2350 (7)	2350.0	-11.06	-69.9	57,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OUTO-2480 (5)	2480.0	-11.30	-66.0	69,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-1	25.0	-12.00	-84.8	7000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-3A	200.0	-12.38	-86.5	7000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-4A	300.0	-12.69	-85.4	7000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-5A	400.0	-12.95	-83.5	11,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-6A	500.0	-13.05	-82.8	11,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-7A	600.0	-12.98	-82.6	12,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-8A	700.0	-13.11	-83.0	12,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-9A	800.0	-13.06	-83.1	12,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-10A	900.0	-13.01	-82.8	12,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OU-11A	1000.0	-13.11	-82.6	12,000	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OKU180-G1	180.0	-12.54	-87.2	8000	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU180-G2	180.0	-12.68	-87.0	7000	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU180-G3	180.0	-12.79	-85.7	8000	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU180-G6	180.0	-12.57	-88.6	6000	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU180-G7	180.0	-12.61	-88.4	5000	Kietäväinen et al. (2017)	1
Finland	Outukumpu Deep Drill Hole	OKU180/25	180.0	-12.55	-88.8	nm	Kietäväinen et al. (2017)	2
Finland	Outukumpu Deep Drill Hole	OKU180/27	180.0	-12.55	-88.8	nm	Kietäväinen et al. (2017)	2
Sweden	Parainen		280.0	-11.5	-26.3	250	Nurmi et al. (1988)	2
Sweden	Parainen		410.0	-11.3	-76.1	1000	Nurmi et al. (1988)	2
Sweden	Parainen		490.0	-9.8	-67.7	6500	Nurmi et al. (1988)	2
Finland	Pori	17096	0.0	-15.6	-115.8	2773	Unpub.	2
Finland	Pori	17097	15.2	-15.5	-118.1	2716	Unpub.	2
Finland	Pori	17098	48.5	-15.6	-118.3	2662	Unpub.	2
Finland	Pori	17099	100.0	-15.5	-116.6	7318	Unpub.	2
Finland	Pori	17100	145.5	-14.0	-105.6	4490	Unpub.	2
Finland	Pori	17101	200.0	-13.4	-102.8	4879	Frape et al. 2003	2
Finland	Pori	17102	250.0	-10.6	-65.1	49,411	Unpub.	2
Finland	Pori	17103	266.7	-9.0	-55.6	69,936	Unpub.	2
Finland	Pori	17104	400.0	-10.7	-73.3	36,583	Unpub.	2
Finland	Pori	17105	401.5	-8.4	-47.9	79,325	Unpub.	2
Finland	Pori-P0-1	S8706722	0.0	-15.6	-117.1	2640	Unpub.	2
Finland	Pori-P0-1	S8706723	52.0	-16.8	-121.0	2780	Unpub.	2
Finland	Pori-P0-1	S8706724	111.0	-20.7	-125.7	2689	Frape et al. 2003	2
Finland	Pori-P0-1	S8706725	151.0	-13.6	-109.0	4554	Unpub.	2
Finland	Pori-P0-1	S8706726	201.0	-12.8	-101.7	5801	Unpub.	2
Finland	Pori-P0-1	S8706727	241.0	-9.4	-52.1	69,072	Unpub.	2
Finland	Pori-P0-1	S8706728	301.0	-8.3	-46.3	91,393	Unpub.	2
Finland	Pori-P0-1	S8706729	351.0	-7.9	-56.1	93,752	Unpub.	2
Finland	Pori-P0-1	S8706730	402.0	-8.1	-54.4	92,030	Unpub.	2
Finland	Pori-P0-1	12930	410.0	-7.9	-44.6	148,554	Unpub.	2
Finland	Pori-P0-1		440.0	-8.1	-44.6	nm	Unpub.	2
Finland	Pori-P0-1	12932	480.0	-7.7	-45.4	119,645	Unpub.	2
Finland	Pori-P0-1	12933	530.0	-8.1	-48.7	24,706	Unpub.	2
Finland	Pori-P0-1	12934	580.0	-7.7	-43.9	127,254	Unpub.	2
Finland	Pyhäsalmi		2400.0	-12.27	-70.2	66,900	Miettinen et al. 2015	3

(continued on next page)

Table A1 (continued)

Country	Location	Sample ID	Depth (m)	$\delta^{18}\text{O}$	$\delta^2\text{H}$	TDS (mg/L)	Source	Sampling Method
Finland	Pyhäsalmi		2400.0	−12.27	−70.2	66,900	Miettinen et al. 2015	3
Finland	Pyhäsalmi		1910.0	−13.70	−91.2	25,700	Miettinen et al. 2015	3
Finland	Ylivieska	S8706714	202.0	−13.0	−97.7	422	Unpub.	2
Finland	Ylivieska	S8706715	302.0	−12.8	−93.5	4174	Unpub.	2
Finland	Ylivieska	S8706716	543.0	−13.2	−57.0	45,670	Unpub.	2
Finland	Ylivieska	S8706717	618.0	−13.4	−42.8	75,759	Unpub.	2
Finland	Ylivieska		640.0	−13.1	−54.8	54,927	Unpub.	2
Finland	Ylivieska	S8706718	101.0	−12.5	−94.0	857	Unpub.	2
Finland	Ylivieska	S8706719	200.0	−13.1	−100.5	2838	Unpub.	2
Finland	Ylivieska	S8706720	300.0	−12.8	−99.0	5184	Unpub.	2
Finland	Ylivieska	S8706721	400.0	−13.5	−97.4	19,753	Unpub.	2
Finland	Ylivieska	17059	240.3	−13.2	−99.4	nm	Unpub.	2
Finland	Ylivieska	17060	357.3	−13.1	−45.6	53,660	Frape et al. 2003	2
Finland	Ylivieska	17061	437.9	−12.8	−84.9	nm	Unpub.	2
Finland	Ylivieska	17062	590.3	−13.7	−26.7	nm	Unpub.	2

Table A2

Water isotope data from service water (waters introduced into the mine for use in drilling and mine operations).

Country	Location	Sample ID	$\delta^{18}\text{O}$	$\delta^2\text{H}$	Source
Canada	Kidd Creek Mine, Timmins	10.05.06_KC_8800_Refuge_Drilling Water	−12.1	−91.6	Unpub.
Canada	Kidd Creek Mine, Timmins	1.04.2009_KC_7850_Service water	−11.2	−96	Unpub.
Canada	Kidd Creek Mine, Timmins	12.01.2010_KC_7850_Service water	−11	−89.8	Unpub.
Canada	Kidd Creek Mine, Timmins	01.03.2012_KC_9500_service water	−11.8	−90.1	Unpub.
Canada	Kidd Creek Mine, Timmins	14.06.2012KCL9500_BHY13762-Service Water	−12.5	−87.6	Unpub.
Canada	Kidd Creek Mine, Timmins	22.10.2015_KCL7850_ServiceWater	−11.4	−90.4	Unpub.
Canada	Kidd Creek Mine, Timmins	12.07.2016_KCL7850_SW (service water)	−12.6	−91.1	Unpub.
Canada	Kidd Creek Mine, Timmins	24.01.2017_KCL7850_SW	−10.6	−80.8	Unpub.
Canada	Kidd Creek Mine, Timmins	06.06.2017_KCL7850_SW	−12.4	−89.8	Unpub.
Canada	Kidd Creek Mine, Timmins	29.01.2018_KCL7850_SW	−10.7	−83.1	Unpub.
Canada	Kidd Creek Mine, Timmins	14.6.2018-KC7850-SW	−13	−94.3	Unpub.
Canada	Kidd Creek Mine, Timmins	05.02.2019_KC7850_SW	−11.3	−83.3	Unpub.
Canada	Nickel Rim South Mine, Sudbury	29.11.2013_NR_1730m_Service Water	−10.3	−72.9	Unpub.
Canada	Nickel Rim South Mine, Sudbury	22.10.2014_NR_1730m_Service water	−10.1	−74.1	Unpub.
Canada	Nickel Rim South Mine, Sudbury	29.11.2014_NR_1730m_Service water	−10.3	−72.9	Unpub.
Canada	Birchtree Mine, Thompson	28.05.2007_BT3950L_Service water	−13.4	−112.1	Unpub.
Canada	Thompson Mine, Thompson		−12.8	−108.4	Unpub.
Canada	Thompson Mine, Thompson		−12.9	−112.2	Unpub.
South Africa	Beatrix	BE223 SW 031301	−6.22	−42.2	Onstott et al. (2006)
South Africa	Driefontein	DR548 SW1 090198	−2.68	−8.6	Onstott et al. (2006)
South Africa	Driefontein	DR548 SW2 120198	−2.17	−8.1	Onstott et al. (2006)
South Africa	Evander	EV219 SW 030901	−4.59	−25.8	Onstott et al. (2006)
South Africa	Evander	EV224 SW1&2081302	−4.65	−20.7	Onstott et al. (2006)
South Africa	Evander	EV818 SW 030601	−3.87	−19.3	Onstott et al. (2006)

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