

1 **Micro-distribution of Oxygen Isotopes in Unequilibrated Enstatite Chondrites**

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18 **Abstract**

19 We report petrology and high precision, *in situ* oxygen isotope analyses of silicates in
20 chondrules, fragments, metal-rich nodules, refractory inclusions from the ALH 81189 (EH3), ALH
21 85159 (paired with ALH 81189) and from the MAC 88136 (EL3) chondrite. This is the first report
22 of oxygen isotope ratios for individual objects in an EL3 and for the silicates associated with the
23 metal-rich nodules that are characteristic of unequilibrated enstatite (E3) chondrites. The oxygen
24 isotopic data from the chondrules and other objects form a trend, on a 3-isotope plot, that coincides
25 with the slope~1 primitive chondrule mineral (PCM) line (initially defined by chondrules from the
26 Acfer 094 primitive carbonaceous chondrite), with most objects clustering at the intersection of
27 the PCM line with the terrestrial fractionation (TF) line, near whole rock E3. The data from EH3
28 and EL3 overlap and show a similar distribution, suggesting they formed from a similar pool of
29 precursors or in similar gaseous environments, but their mineral compositions suggest differences
30 in their nebular environments and/or parent bodies. Silicates in the metal-rich nodules we analyzed
31 (in both EH3 and EL3) have oxygen isotope ratios (as well as mineral compositions) similar to the
32 silicate (metal-free) chondrules. This is consistent with formation of the metal-rich nodules prior
33 to chondrite accretion, in an environment and from a process similar to that which formed the
34 coexisting chondrules, but from more metal-rich mixtures of precursors. Olivine in an AOA from
35 ALH 81189 is ^{16}O -rich with $\delta^{18}\text{O} = -46.5\text{\textperthousand}$, $\delta^{17}\text{O} = -48.0\text{\textperthousand}$, similar to the AOAs and refractory
36 inclusions previously reported in E3 and in all other chondrite groups. There is a clear distinction
37 in oxygen isotopic compositions between the chondrules in the E3 chondrites and those in the LL
38 and R as well as those in CV and CM chondrite groups. Chondrules from CR and E chondrites
39 plot closer to the PCM line than all other chondrite groups with E3 chondrules having a different
40 distribution toward more ^{16}O -poor compositions. Chondrules in other chondrite groups form trends

41 above and below the PCM. From the distribution of EC chondrules along the PCM line, we propose
42 that similar pools of chondrule precursors were present in the different (carbonaceous, CR and
43 Acfer 094 and non-carbonaceous, E) chondrule forming regions in the protoplanetary disk but with
44 different amounts of ^{16}O -rich refractory materials, prior to development of the postulated *Jupiter*
45 *divide* that potentially separated inner (non-carbonaceous) from outer (carbonaceous chondrite)
46 Solar System materials or the Jupiter barrier was inefficient in completely separating these
47 materials.

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49

50 **Introduction**

51 Enstatite chondrites are important for constraining conditions and processes in the
52 protoplanetary disk and they may be our closest examples of the material that accreted to form the
53 terrestrial planets (e.g., Javoy 1995, 2010; Dauphas, 2017). They are among the most reduced solar
54 system materials as indicated by their unique mineral assemblages and compositions (e.g., Keil,
55 1968; Weisberg and Kimura, 2012). Their whole rock stable (O, Cr, Ti, Ni and Zn) isotopic
56 compositions are markedly similar to those of the Earth-Moon system (Javoy, 1995; Warren, 2011;
57 Paniello, 2012), but Si isotopes differ, possibly supplying evidence against a direct relationship
58 between E chondrites and the Earth -Moon system (Fitoussi and Bourdon, 2012; Savage and
59 Moynier, 2013; Kadlag et al., 2019). However, metal-silicate Si isotope fractionation in a reduced
60 nebular environment and vapor loss of lighter Si isotopes during planetary volatilization may have
61 contributed to differences in the Si isotopic compositions between Earth and the E chondrites
62 (Sikdar and Rai, 2020). It was recently shown that H, C and N isotopic compositions of E
63 chondrites are also similar to Earth (Piani et al., 2020; Gray et al., 2021) and that enstatite
64 chondrites contain enough H to have delivered to Earth three times the amount of water currently
65 in its oceans. (Piani et al., 2020). The surface chemistry of Mercury suggests reduced E chondrite-
66 like precursors (e.g., Nittler et al., 2011; Burbine et al., 2002; Ebel and Stewart, 2018). Thus, the
67 E chondrites may be similar to the materials that were present in specific feeding zones during
68 accretion of the terrestrial planets.

69 We previously studied the petrology and oxygen isotopic compositions of chondrules in the
70 Sahara 97096, Yamato 691, Kota Kota and LEW 87223 E3 chondrites and showed that they define
71 a slope~1 line, termed the enstatite chondrite mixing (ECM) line (Weisberg et al., 2010, 2011),

72 which is indistinguishable from the primitive chondrule mineral (PCM) line defined by Acfer 094
73 chondrules (Ushikubo et al., 2012). Here we report new, high precision, *in situ* oxygen isotopic
74 analyses of 31 objects (chondrules, refractory inclusions and metal-rich nodules) in the highly
75 primitive Allan Hills (ALH) 81189 (and paired ALH 85159) EH3 and, to our knowledge, the first
76 oxygen isotopic analyses of individual objects in an EL3 chondrite, MacAlpine Hills (MAC)
77 88136. The goals of this work are to study the micro-distributions of oxygen isotopic compositions
78 in chondrules and other objects in the most primitive EH3 and EL3 chondrites, further test the
79 relationship between enstatite chondrite precursors and the precursors of other primitive chondrites
80 (e.g., Acfer 094), explore the relationship between silicates among the different types of objects
81 (including the reduced chondrules, metal-rich nodules, FeO-rich silicates and refractory
82 inclusions) and better document and develop an understanding of the oxygen isotopic differences
83 among the chondrules in primitive chondrites.

84

85 **Background**

86 The distribution of three stable oxygen isotopes in chondrules at the micron scale has provided
87 important insights into the process of isotope exchange between chondrule melt and surrounding
88 gas (e.g., Yu et al., 1995; Boesenberg et al., 2005). *In situ* oxygen isotopic secondary ion mass
89 spectroscopy (SIMS) analyses of chondrules in carbonaceous chondrites has revealed the existence
90 of ^{16}O -rich relict olivine grains (e.g., Yurimoto and Wasson, 2002; Jones et al., 2004; Kunihiro et
91 al., 2004; Rudraswami et al. 2011; Ushikubo et al. 2012; Tenner et al. 2013; Marrocchi et al. 2018;
92 2019). However, excluding the minor occurrence of relict grains, minerals and glass in individual
93 chondrules show indistinguishable oxygen isotope ratios (e.g., Ushikubo et al. 2013; Tenner et al.
94 2015), in contrast to earlier observations by Chaussidon et al. (2008) that reported pyroxene and

95 mesostasis being ^{16}O -depleted in oxygen isotope ratios compared to olivine within the same chon-
96 drules. Reaction of chondrules with SiO molecules during chondrule formation has also been pro-
97 posed to explain petrologic features, such as mineralogical zoning in chondrules (pyroxene-rich
98 shells) (e.g., Tissandier et al., 2002; Libourel et al., 2006), which are likely the result of chondrules
99 forming in open systems with respect to major oxides (Nagahara et al., 2008; Libourel and Portail,
100 2018). Under open-system chondrule formation, oxygen isotope exchange would have occurred
101 efficiently between chondrule melt and the ambient gas (containing SiO , H_2O , and CO molecules)
102 and resulted in internally homogeneous oxygen isotope ratios in each chondrule (Kita et al., 2010;
103 Ushikubo et al., 2012; Tenner et al., 2015; Marrocchi and Chaussidon, 2015). Tenner et al. (2015)
104 presented a mass balance model involving ^{16}O -poor water ice and solid precursors to explain their
105 observation of systematic changes in $\Delta^{17}\text{O}$ ($=\delta^{17}\text{O}-0.52\times\delta^{18}\text{O}$) with redox state of the chondrules
106 in CR chondrites, which was further applied to chondrules in CV (Hertwig et al. 2018). Marrocchi
107 and Chaussidon (2015) presented a model in which SiO molecules played a major roles in control-
108 ling mass independent fractionation of oxygen isotopes among chondrules in carbonaceous chon-
109 drites. In the case of ordinary chondrite chondrules, Kita et al. (2010) suggested that the chondrule-
110 forming region contained two kinds of solid precursors, (1) ^{16}O -poor solid precursors and (2) ^{16}O -
111 rich solid precursors derived from the same oxygen isotope reservoir as carbonaceous chondrites.
112 These two sources were mixed efficiently among chondrules as a result of open system behavior
113 during chondrule formation through interactions between the heated chondrules and the ambient
114 gas.

115 For the enstatite chondrites, Clayton and Mayeda (1985) indicated that individual chon-
116 drules from EH3 chondrites form a distinct cluster on the oxygen 3-isotope diagram, not overlap-
117 ping chondrules from O or C chondrites and scattering about a best-fit line of slope 0.7. In contrast,

118 Tanaka and Nakamura (2017) reported the best-fit line with the slope of 1.27 from chondrules in
119 EH3 and EH4 chondrites that include data from 6 individual chondrules in EH3. However, our
120 previous results on chondrules in four E3 chondrites showed that the oxygen isotope ratios ($\delta^{18}\text{O}$,
121 $\delta^{17}\text{O}$) of olivine and pyroxene in E3 chondrules have a wide range of values spanning as much as
122 10 ‰ in both $\delta^{18}\text{O}$ and $\delta^{17}\text{O}$. Some chondrules from EH3 chondrites cluster around the terrestrial
123 fractionation line near enstatite chondrite whole rock values, some chondrules overlap the ordinary
124 chondrite field on the 3-isotope diagram and some extend toward more ^{16}O -rich compositions sim-
125 ilar to C chondrite chondrules (Weisberg et al., 2011). Olivine in some E3 chondrules shows a
126 wide range of $\Delta^{17}\text{O}$ values (−4‰ to +2‰) that form a mixing line parallel to but distinct from the
127 carbonaceous chondrite anhydrous mixing (CCAM) line (e.g., Clayton and Mayeda, 1984). This
128 E3 chondrule olivine line was defined as the enstatite chondrite mixing (ECM) line (Weisberg et
129 al., 2011) but, as mentioned above, it coincides with the line defined by chondrules from Acfer
130 094 referred to as the PCM (primitive chondrite mineral) line by Ushikubo et al. (2012).

131 Another important issue regarding the E chondrites is the origin of the more oxidized com-
132 ponents in E3 chondrites. FeO-bearing silicates in enstatite chondrites are relatively rare and ap-
133 parently out of equilibrium with the more typical reduced (e.g., near-pure enstatite) mineral com-
134 positions (e.g., Lusby et al. 1987; Weisberg et al., 1994). Weisberg et al. (1994) showed that the
135 FeO-bearing silicates are older than the FeO-poor silicates, marking a more oxidized pre-history
136 for these meteorites. It has been suggested that these minerals were transported from ordinary
137 chondrite source regions to the E chondrite accretion zones (Lin et al., 2002). However, the FeO-
138 bearing pyroxene in E3 chondrites has oxygen isotope values similar to coexisting pure enstatite,
139 suggesting it formed locally and was not transported from the ordinary or other chondrule-forming
140 region (Kimura et al., 2003; Weisberg et al., 2011). Additionally, the petrologic setting of the FeO-

141 bearing pyroxene (enclosed in enstatite) suggests that it predates the reduced, FeO-poor enstatite-
142 rich mineral assemblages and implies that conditions were initially more oxidizing in the E3 chon-
143 drule-forming region and/or that the mixture of chondrule precursors for E chondrites was, at least
144 in part, compositionally similar to chondrule precursors of other chondrite groups. This is sup-
145 ported by the oxidation states of Ti in silicates, which contain a mixture of Ti^{3+} and Ti^{4+} (Simon et
146 al., 2016).

147 Weisberg et al. (2011) reported oxygen isotopic heterogeneity among mineral grains within
148 some of the chondrules in E3 chondrites. They suggested that this implies incomplete melting of
149 the chondrules, survival of minerals from previous generations of chondrules, and chondrule re-
150 cycling or possibly open system behavior during chondrule formation. One chondrule contained a
151 relict grain with an R chondrite-like oxygen isotopic composition. The presence of R chondrite-
152 like oxygen in a large, possibly relict, olivine grain suggests limited mixing of materials from other
153 reservoirs and/or close proximity of E, O and R chondrite precursors.

154 Interestingly, the relatively rare calcium-aluminum-rich inclusions (CAIs) in E3 chondrites
155 have petrologic characteristics and ^{16}O -rich oxygen isotopic compositions similar to CAIs in other
156 chondrite groups and appear to have formed in a different oxygen reservoir from the E3 chondrules
157 (Guan et al., 2000; Fagan et al., 2001; Lin et al., 2003). This suggests that CAIs in all chondrite
158 groups may all have formed in the same region and then been distributed to the accretion zones of
159 the various chondrite groups. However, oxygen isotopes indicate that formation and distribution
160 of chondrules and their precursors was more complicated, with chondrules from each chondrite
161 group forming in their own separate regions, from distinct reservoirs, presumably representing
162 different heliocentric distances or nebular epochs.

163
164 **Methods**

165 We selected 22 objects (chondrules, fragments, metal nodules, refractory inclusions) from thin
166 section ALH 81189, 3 (EH3), 9 objects from ALH 85159, 5 (paired with ALH 81189) and 13
167 objects from MAC 88136, 37 (EL3) for oxygen isotope analysis. Element maps for all sections,
168 back-scattered electron (BSE) images of each object and major mineral compositions were
169 collected using a combination of electron beam instruments.

170 Wavelength dispersive spectroscopic (WDS) maps of Si, Al, Mg, Ca, Fe, Ti, S, P, Na and
171 Ni X-ray emission intensity of each thin section were generated with a Cameca SX100 electron
172 microprobe (at the AMNH). These are “stage maps” (moving stage, stationary electron beam).
173 Operating conditions were 15 kV accelerating voltage and 40 nA beam current, with a dwell time
174 of 12 ms on one micrometer beam spots spaced 4 to 7 microns apart. Modal analysis was based on
175 these elemental X-ray maps. Mineral compositions were determined on nominally 1 μ m spots
176 using the Cameca SX100 and JEOL Super Probe (at Rutgers) electron microprobes. Natural and
177 synthetic standards were chosen based on the compositions of the minerals being analyzed. An
178 accelerating potential of 15 keV and a sample current of 20 nA were used for silicates and 20 keV
179 and 25 nA for metal. Counting times were 20 s on peak, and 10 s on background (off-peak) spec-
180 trometer positions. Relative uncertainties (2 sigma) based on counting statistics for major elements
181 (Si, Fe, Mg) are calculated to be <2% and for Ti, Cr, Mn and Ca they are 10%, 10%, 9% and 5%,
182 respectively. Data reductions were carried out using instrument software based on methods de-
183 scribed by Pouchou and Pichoir (1991).

184 Oxygen isotopes of silicates were analyzed *in situ*, using the Cameca IMS 1280 secondary ion
185 mass spectrometer at the WiscSIMS laboratory. The goal was to measure six grains (olivine and
186 pyroxene) in each object, if grain size allowed, to achieve a representative average and to test for
187 internal homogeneity within each chondrule or other object. The analytical procedure was similar

188 to that described in Kita et al. (2010). A Cs^+ primary ion beam was focused to 12 μm diameter at
189 2 nA. Secondary oxygen ions ($^{16}\text{O}^-$, $^{17}\text{O}^-$, and $^{18}\text{O}^-$) were detected simultaneously using multi-
190 collector Faraday cups with three feedback resistors, 10^{10} ohm, 10^{12} ohm, and 10^{11} ohm for $^{16}\text{O}^-$,
191 $^{17}\text{O}^-$, and $^{18}\text{O}^-$, respectively. As in previous studies, 16-25 unknown analyses were bracketed by 8
192 analyses of a San Carlos olivine reference material (SC-Ol; $\delta^{18}\text{O} = 5.32\text{\textperthousand}$; Kita et al., 2010). Two
193 additional olivine, three pyroxene, and one quartz reference materials were also analyzed to correct
194 instrumental bias (appendix). Average external reproducibility of bracketed SC-Ol analyses (2SD)
195 was $\sim 0.2\text{\textperthousand}$ for $\delta^{18}\text{O}$ and $0.3\text{\textperthousand}$ for $\delta^{17}\text{O}$ and $\Delta^{17}\text{O}$. The external reproducibilities (2SD) for each
196 bracket are assigned as the uncertainties of bracket unknown analyses.

197

198 **Results**

199 *Petrology*

200 The objects we studied from ALH 81189 (Fig. 1a) and paired ALH 85159 (Fig. 1b) included
201 20 porphyritic pyroxene (PP), 1 PP with FeO-rich pyroxene, 2 porphyritic olivine-pyroxene (POP),
202 2 porphyritic olivine (PO), and 1 barred pyroxene chondrule; 1 amoeboid olivine aggregate
203 (AOA); 1 diopside-plagioclase fragment; 2 isolated olivine fragments; and 1 metal-rich nodule
204 containing enstatite (En₉₈). From MAC 88136 (Fig. 1c) we studied an unusually large ($>1\text{mm}$)
205 diopside-rich chondrule, a silica-pyroxene chondrule, a radial pyroxene (RP) and 7 PP chondrules,
206 and 3 metal-rich nodules containing enstatite laths. The latter are the first oxygen isotopic data
207 from silicates in metal-rich nodules in E3 chondrites.

208 As is typical of EH3 and EL3, most chondrules are dominantly FeO-poor enstatite and can be
209 considered type IB (Figs. 2a-d). All textural types of Gooding and Keil (1968) are present in
210 enstatite chondrites but (type I) porphyritic pyroxene (PP, e.g., Fig. 2b) is the most common type.

211 Most of the PP chondrules we studied contain near-endmember enstatite (En_{>98}). One PP (C9) in
212 ALH 81189 was dominated by FeO-rich pyroxene (Wo_{0.1}En₇₃₋₈₃) (Fig. 2c). Unlike most other
213 chondrite groups, olivine is minor in the E3 chondrites. It occurs mainly as poikilitically enclosed
214 grains in enstatite in the pyroxene-rich chondrules, but we also identified rare porphyritic olivine
215 (PO) chondrules and relatively large, isolated olivine mineral fragments (Fo₉₈) up to 300 μm in
216 size in ALH 81189 (Fig. 2a).

217 Several unusual objects were also studied including a diopside–plagioclase chondrule (F9) in
218 ALH 81189 unlike any chondrule or fragment previously described in any E3 chondrite. It is
219 irregular in shape, \sim 300 μm across and composed mainly of laths of diopside (Wo_{44.8}En_{54.3}) having
220 2.5 wt.% Al₂O₃ and 1.5 TiO₂ and plagioclase (An_{79.5}Ab_{20.5}), with minor enstatite (Wo_{0.6}En_{97.1}) and
221 silica (Fig. 3a). The AOA (F17) in ALH 81189 is an irregularly shaped inclusion consisting of
222 olivine surrounding nodules of Ca-pyroxene and anorthite (Fig. 3b) and is texturally and mineral
223 compositionally similar to AOAs in C chondrites (e.g., Grossman and Steele, 1976). The olivine
224 has a composition of Fo₉₉, if calculated based on two end members (Fo and Fa). However, this
225 olivine also contains 0.3 wt. % CaO.

226 The diopside-rich chondrule (C1) in MAC 88136 is \sim 50% coarse diopside (Wo_{46.4}En_{53.2}) in
227 thin section, occurring with enstatite (Wo_{0.7}En_{98.9}), silica and troilite (Fig. 3c-d). Silica apparently
228 occurs mainly along the edges of the diopside grains in the chondrule. The troilite in this chondrule
229 contains 0.4 wt. % Ti and 0.4 Cr, as is typical of troilite in enstatite chondrites (e.g., Keil, 1968).

230 Metal nodules in EL3 chondrites (also called metal-silicate intergrowths) have been a subject
231 of controversy as to whether they are primary, nebula-formed objects or products of impact
232 melting. EH3 chondrites also contain metal-rich objects with assemblages of sulfides, phosphide,
233 graphite and silicates (Weisberg and Prinz, 1998; Lehner et al., 2010) and have been referred to as

234 metal-sulfide nodules (MSN). The EL3 nodules contain laths of enstatite (~En₉₈) intergrown with
235 metal. We studied one metal-rich nodule (M5) in ALH 81189 and three metal-rich nodules (M1-
236 2, M8, and M10) in MAC 88136 (Figs. 4a-d). In the MAC 88136 EL3 chondrite, enstatite occurs
237 mainly as laths, needles and irregular crystals within the metal. In the metal nodule in ALH 81189
238 studied here, the enstatite occurs as anhedral crystals oriented around metal in the center, forming
239 what appears to be a discontinuous enstatite-rich layer (Fig. 4b).

240 Compositions of most silicates we analyzed are reduced enstatite (En>98) and forsterite
241 (Fo>98) with a few exceptions such as C9 in ALH 81189 discussed above. Fig. 5 shows the
242 distribution of silicate compositions in the sections studied.

243

244 *Oxygen isotopes*

245 Our new data for the average oxygen isotope ratios for each object we studied are given in
246 Table 1 (complete dataset is in appendix) and plotted in Fig. 6 along with literature chondrule data
247 from E chondrites. While whole rock E chondrites plot along the terrestrial fractionation (TF) line
248 (e.g., Weisberg and Kimura, 2012), the majority of olivine and pyroxene analyses plot along the
249 slope~1 PCM line, with most data clustering at the intersection of the PCM with the TF line (Fig.
250 6a). The mean oxygen isotope ratios of individual chondrules were determined for most of the
251 chondrules that show internally homogeneous oxygen isotopic ratios (Fig. 6b).

252 Two PP chondrules show a significant range of internal variability (Fig. 6c). ALH 81189 C1
253 contains ¹⁶O-rich relict olivine grains that reach $\delta^{18}\text{O}$ and $\delta^{17}\text{O} \sim -30\text{\textperthousand}$, in contrast to those of
254 pyroxene in the same chondrule that plot above the TF line. ALH 85159 C51 shows the highest
255 $\delta^{18}\text{O}$ and $\delta^{17}\text{O}$ values among all of the chondrule studied and shows significant internal variability
256 from 6‰ to 8‰ and from 4‰ to 6‰, respectively. Olivine in the AOA from ALH 81189 is ¹⁶O-

257 rich with $\delta^{18}\text{O} = -46.5\text{\textperthousand}$, $\delta^{17}\text{O} = -48.0\text{\textperthousand}$, similar to AOAs and refractory inclusions in other
258 chondrite groups (Fig. 6c; e.g., Krot et al., 2004).

259 The individual chondrule data from ALH 81189 and ALH 85159 (EH3) and the more limited
260 data set from MAC 88136 (EL3) overlap and show a similar distribution. The $\Delta^{17}\text{O}$ for ALH 81189
261 and paired ALH 85159 chondrules range from -1.4 to $1.3\text{\textperthousand}$ and for MAC 88136 the range is
262 slightly smaller at -0.6 to $1.3\text{\textperthousand}$. Olivine-rich chondrules (PO, BO, POP) tend to plot on and below
263 the TF line, while more PP chondrules plot above the TF line, as indicated from the distribution of
264 individual olivine and enstatite data in Fig 6a. The PP chondrule (C22) in ALH 81189 has the most
265 ^{16}O -rich mean oxygen isotope ratios ($\delta^{18}\text{O} = 2.3\text{\textperthousand}$, $\delta^{17}\text{O} = -0.2\text{\textperthousand}$). Petrologically, C22 is a
266 porphyritic pyroxene chondrule dominated by enstatite. The only petrologic difference between
267 C22 and other chondrules in ALH 81189 is that C22 has a slightly higher amount of Al-rich
268 mesostasis. The unusual diopside-rich chondrule in MAC 88136, and enstatite in the metal-rich
269 nodules in both EH3 and EL3 all show oxygen isotopic ratios within a similar range of other
270 chondrules. Silica from the silica-rich chondrule (C30) and the average of two analyses of silica
271 in the diopside-rich chondrule (C1; Fig. 3b) in MAC 88136 are also plotted in Fig. 6b. They are
272 compositionally similar, plotting on the TF line at $\delta^{18}\text{O} = 8.7\text{\textperthousand}$ and $8.3\text{\textperthousand}$, respectively toward
273 much more ^{16}O -poor compositions than olivine and pyroxene.

274 The data from most chondrules and other objects in ALH 81189, ALH 85159 and MAC 88136
275 form a regression line $\delta^{17}\text{O} = (0.88 \pm 0.10) \times \delta^{18}\text{O} - (2.1 \pm 0.5)$ that is statistically indistinguishable
276 from the slope~1 PCM line, $\delta^{17}\text{O} = (0.987 \pm 0.013) \times \delta^{18}\text{O} - (2.70 \pm 0.11)$ (Fig. 6d). Four chondrules
277 from ALH 81189 and one from MAC 88136 appear to suggest a resolvable second trend above
278 the PCM line (referred to as “other chondrules” in Fig. 6d). They show indistinguishable $\Delta^{17}\text{O}$
279 values $\sim 0.9\text{\textperthousand}$ (Fig. 6b) and may plot along the slope ~ 0.5 line (Fig. 6b). However, they comprise

280 a linear trend with two heterogeneous chondrules, ALH 85159 C51 and three moderately ^{16}O -rich
281 relict olivine grains from ALH 81189 C1 (Fig. 6d). A regression line through these data form a
282 slightly shallower slope -0.84 ± 0.04 with an intercept of $-1.0\pm0.2\text{\textperthousand}$ (Fig. 6d). This second trend
283 significantly overlaps with the chondrule data from LL chondrites (Kita et al. 2010). To check
284 whether these two groups truly belong to two distributions in the two-dimensional space $\delta^{18}\text{O}$ -
285 $\delta^{17}\text{O}$, we performed a two-dimensional, two samples Kolmogorov-Smirnov test. This test is a non-
286 parametric test that computes the probability that two samples belong to the same distribution by
287 comparing their cumulative distributions. In this case, the test returns a P-value of 7.1×10^{-5} , well
288 below the accepted threshold of 0.05 to reject the hypothesis, allowing us to conclude that these
289 two trends are in fact two separate trends.

290 The data distribute in a similar range to those from our previous study of the Sahara 97096 and
291 Yamato-691 EH3 and the anomalous LEW 87223 chondrite (Weisberg et al., 2011; Fig. 6e). The
292 new dataset is also similar to the individual chondrule data reported by Clayton and Mayeda (1985)
293 and Tanaka and Nakamura (2017), which are shown in Fig. 6e for comparison. In Fig. 6f, a
294 regression line of all E3 chondrules is obtained by using these literature data and all data obtained
295 in this study. The regression line gives $\delta^{17}\text{O} = (0.997\pm0.025)\times\delta^{18}\text{O} - (2.40\pm0.2)$, which is almost
296 identical to the PCM line, but has a large MSWD of 24, reflecting two separate trends at the higher
297 $\delta^{18}\text{O}$ and $\delta^{17}\text{O}$ ends.

298 Type I chondrules in some C chondrite groups show a trend of $\Delta^{17}\text{O}$ increasing with decreasing
299 Mg# for type I chondrules, whereas such a trend is not observed for chondrules in O and R
300 chondrites (Kita et al., 2010, 2015). The data for the E3 chondrules do not show a clear trend (Fig.
301 7). A regression line through the data ($r=0.2$) shows a very slight hint of $\Delta^{17}\text{O}$ decreasing with the
302 average Mg# for the chondrule silicates but it is not statistically significant.

303

304 **Discussion**

305 *Relationship between chondrules, metal-rich nodules and refractory inclusions in EH3 and EL3*
306 *chondrites.*

307 The oxygen isotopic compositions of the silicates within most chondrules analyzed in this
308 study appear to be fairly uniform. Weisberg et al. (2011) specifically selected olivine-bearing
309 chondrules and, as a result discovered oxygen isotopic heterogeneity in minerals from some
310 individual chondrules in EH3 chondrites and suggested incomplete melting of the chondrules,
311 survival of minerals from previous generations of chondrules, chondrule recycling, or possibly
312 open system behavior during chondrule formation, as proposed for chondrules in O and C
313 chondrites. Weisberg et al. (2011) found one chondrule in Sahara 97096 that contained an olivine
314 grain with an R chondrite-like oxygen isotopic composition. They interpreted this to be a relict
315 grain, suggesting limited mixing of materials from other reservoirs and the close proximity of E,
316 O and R chondrite precursors, compared to C chondrites. This is consistent with recent discoveries
317 of whole rock nucleosynthetic isotopic differences between non-carbonaceous (including E, O and
318 R) and carbonaceous chondrites (Trinquier et al., 2009).

319 Overlap in oxygen isotopic compositions between EH3 and EL3 chondrules suggests that
320 they share similar precursors but the mineral assemblages and compositions (e.g., higher Si content
321 in EH metal) in EH3 and EL3 indicate different degrees of reduction (Keil, 1968; Weisberg and
322 Kimura, 2012), suggesting differences in their nebular environments and/or separate, possibly
323 multiple parent bodies (e.g., Weyrauch, 2018). Most of the components that we analyzed show
324 similar oxygen isotopic compositions, including the chondrules and enstatite and forsterite mineral
325 fragments, enstatite in metal-rich nodules and the silicates with more FeO-rich compositions. This

326 is consistent with previous studies that showed that the less common FeO-bearing minerals in E3
327 chondrites share the same oxygen isotopic compositions as the more common, reduced, FeO-poor
328 olivine and pyroxene in E3's (Weisberg et al., 2011, Kimura et al., 2003). Additionally, the unusual
329 diopside-plagioclase fragment (F9; Fig. 3a) in ALH 81189 and the large diopside-enstatite
330 chondrule (C1; Figs. 3c-d) in MAC 88136 also have similar isotopic compositions to the other E3
331 components. Thus, all of these objects likely formed from the same mix of precursors and were
332 likely molten or partially molten and exchanged oxygen in the same, presumably chondrule-
333 forming, region.

334 Derivation of the FeO-bearing silicates from the same oxygen reservoirs as the more
335 reduced silicates and textural and compositional evidence indicating that the FeO-bearing silicates
336 have undergone various degrees of reduction (Rambaldi et al. 1983; Lusby et al. 1987; Weisberg
337 et al. 1994; Kimura et al. 2003; Weisberg et al. 2011), suggest a stage of more oxidizing conditions
338 in the evolution of enstatite chondrites. This is consistent with studies of element valence states
339 that also suggest E chondrites were originally more oxidized at some point in their history (e.g.,
340 Jacquet et al., 2015; Simon et al., 2016).

341 The origin of the metal-rich nodules in E3 chondrites has been controversial. There are
342 four major, conflicting hypotheses for the origin of these nodules in EL3 chondrites: (1) the metal-
343 rich nodules are products of impact melting and formed by mobilization and injection of metal into
344 pores in the EL3 regolith (Van Nierkerk and Keil, 2011), (2) metal-rich nodules are pre-
345 accretionary and resulting from impact disruption of early differentiated bodies (Horstmann et al.,
346 2014), (3) the nodules are chondrule-like melts formed in the early solar system (Weisberg et al.,
347 2013; Ebel and Sack, 2013), and (4) the nodules are aggregates of condensates (El Goresy et al.,
348 2017; Lin et al., 2011). Kadlag et al. (2019) measured Si isotopes in silicates and metal in an EH3

349 chondrite and showed differences in their Si isotopic compositions suggesting that the differences
350 were inherited from different nebular reservoirs. However, Si isotope fractionation between
351 silicate and metal may also be responsible for the differences. Based on major and trace element
352 analyses, Lehner et al. (2014) concluded that the matrix, chondrule, and metal-sulfide nodule
353 compositions are apparently complementary, suggesting all the components of the EH3 chondrites
354 came from the same nebular reservoir.

355 The oxygen isotopic data can help shed light on the origin of the EL3 nodules. The nodules
356 in the EL3 chondrites have irregular to near-spherical shapes and characteristically contain
357 enstatite laths or needles intergrown with the metal (Figs. 4c and d). Their sharp boundaries with
358 surrounding matrix and other objects suggest the nodules formed as free-floating objects similar
359 to the chondrules. They may be aggregates of condensates or were molten or partially molten. For
360 the limited number of metal-rich nodules we were able to analyze, the oxygen isotopic ratios (as
361 well as the mineral compositions) of the enstatite laths in the metal-rich nodules are similar to
362 those of enstatite in the silicate (metal-free) chondrules. Additionally, the metal-rich nodule and
363 chondrule silicates all plot along the same (PCM) mixing line defined by the highly primitive Acfer
364 094 C chondrite. Although we cannot conclusively determine the origin of the metal-rich nodule
365 silicates, the petrologic and oxygen data are more consistent with formation of the metal-rich
366 nodules prior to chondrite accretion, in an environment similar to the coexisting chondrules, and
367 forming from a process similar to that which formed the metal-free chondrules, but from more
368 metal-rich mixtures of precursors.

369 The separation of the metal and silicates in chondrites is not well understood. Metal-rich
370 nodules may have been spatially or temporally separated from the metal-poor chondrules, possibly
371 due to an aerodynamic or a magnetic sorting mechanism that resulted in local metal-rich and metal-

372 depleted regions that contributed to the same parent body (e.g., Hubbard, 2016). Weisberg et al.
373 (2013) suggested a fractional condensation model to account for earlier formed (higher
374 temperature) metal-rich “chondrule” compositions. Metal condenses at higher temperatures (1375
375 K) than silicate (olivine, diopside and enstatite condense at 1100, 1050 and 950 K, respectively)
376 in a nebular reservoir with a C/O ratio greater than 1 at 10^{-3} bar total pressure (Ebel, 2006, Plate
377 9). Thus, it is reasonable to consider the metal-rich nodules in E3 chondrites as essentially "metal-
378 rich chondrules", as suggested by Gooding and Keil (1981) and Alpert et al. (2020) for ordinary
379 chondrites.

380

381 *The Enstatite Chondrite Mixing Line and the PCM line*

382 Although the whole rock oxygen isotopic compositions for E chondrites plot along or close to
383 the TF line on a three isotope diagram (Clayton and Mayeda, 1984, 1999; Weisberg et al., 1995,
384 2011, Weisberg and Kimura, 2012), the individual chondrules, mineral fragments and metal-rich
385 nodules form a trend toward more ^{16}O -rich compositions, (Fig. 6a, b), consistent with our previous
386 studies of chondrules in E3 chondrites (Weisberg et al., 2011). This trend is also supported by the
387 findings of Tanaka and Nakamura (2017), though they suggested their data formed a somewhat
388 steeper slope. Weisberg et al. (2011) referred to the array of E3 chondrite oxygen isotopic ratios
389 as the enstatite chondrite mixing line (ECM) but acknowledged that this mixing line is statistically
390 indistinguishable from the slope-1 trend defined by chondrules from the primitive C chondrite
391 Acfer 094 and referred to as the primitive chondrite mineral line (PCM) by Ushikubo et al. (2012).

392 Our data indicate a clear distinction between chondrules in the E chondrites and those in the LL
393 chondrules which form a slope-0.5 trend mostly above the terrestrial fraction (TF) line with LL
394 chondrules plotting to the left of the PCM (Kita et al., 2010). E chondrite chondrules differ from

395 whole rock R chondrites which form a trend at even higher, ^{17}O -rich values (e.g., Bischoff et al.,
396 2011) and R chondrite chondrules, which overlap with ordinary chondrite chondrules (Kita et al.,
397 2015). The E3 chondrules are also isotopically different from CV and CM chondrite groups, which
398 form a trend apparently parallel to but below the PCM line (Fig. 8). Chondrules from CR and EC
399 chondrites plot closer to the PCM line than other chondrite groups (Fig. 8).

400 The PCM line is a regression line for individual SIMS olivine and pyroxene analyses from the
401 Acfer 094 ungrouped C chondrite, including relict olivine and chondrules with internally
402 heterogeneous oxygen isotopic ratios (Ushikubo et al., 2012). Chondrule data from other C
403 chondrites generally plot on the PCM line (Rudraswami et al. 2011; Tenner et al. 2013; 2015;
404 Marrocchi et al., 2018, 2019). The PCM line is interpreted to represent a mixing trend of two
405 extreme primary oxygen isotope reservoirs of solids in the early solar system that potentially
406 accreted to form the terrestrial planets, including the Earth (Ushikubo et al., 2012). It has been
407 recognized that a small fraction of chondrules in carbonaceous chondrites, including Acfer 094,
408 show OC-like oxygen isotopic ratios, which plot near the TF line ($\Delta^{17}\text{O} \sim 0\text{\textperthousand}$) but left of the PCM
409 line in oxygen three isotope space (Tenner et al. 2017; Hertwig et al. 2018; 2019a; 2019b). On the
410 contrary, chondrule analyses from CV and CM chondrites (Hertwig et al. 2018; 2019b; Chaumard
411 et al. 2018; 2021), which are with slightly better precisions compared to earlier studies, are offset
412 slightly from the PCM line, to the right. Recently, Williams et al. (2020) observed that oxygen
413 isotope ratios of CV chondrite chondrules show two linear trends that are related to their $\varepsilon^{50}\text{Ti}$ and
414 $\varepsilon^{54}\text{Cr}$ abundances. Chondrules with positive $\varepsilon^{50}\text{Ti}$ and $\varepsilon^{54}\text{Cr}$ plot with constant offset below the
415 PCM line and chondrules with negative $\varepsilon^{50}\text{Ti}$ and $\varepsilon^{54}\text{Cr}$ plot towards OC chondrule compositions.
416 These chondrule populations might have formed in the outer and inner protoplanetary disk,
417 respectively. Thus, the PCM line might represent an average of the two trends found by Williams

418 et al. (2020). Schneider et al. (2020) also reported $\varepsilon^{50}\text{Ti}$ and $\varepsilon^{54}\text{Cr}$ values of chondrules in several
419 chondrite groups including CV, but did not find negative $\varepsilon^{54}\text{Cr}$ anomalies from chondrules in CV.
420 However, their CV chondrite chondrule data were limited to the dominant type I chondrules with
421 negative $\Delta^{17}\text{O}$. Kita et al. (2010) argued that condensation of olivine from high temperature gas
422 could have resulted in deviation of oxygen isotopes to the left of the PCM line. Contribution of
423 such fractionated precursors could have been much less abundant among the chondrule precursors
424 for E chondrites than in other chondrite groups..

425 The distribution of chondrules along the PCM line is different between the E3 chondrules and
426 the CR chondrules (Fig. 8). Chondrules from CR and Acfer 094 extend toward more ^{16}O -rich
427 compositions (Ushikubo et al., 2012; Schrader et al., 2013, 2014, 2017; Tenner et al., 2015) while
428 the E3 chondrules mainly form a tighter cluster of data points closer to the TF line (zero in Fig. 8)
429 and E3 whole rock. One possible explanation for this difference is that the amount of ^{16}O -rich
430 refractory material that was mixed into the E3 chondrule-forming region was less than that in the
431 carbonaceous chondrite chondrule forming regions. This may be recorded in the very low
432 abundance of CAIs and AOAs in the E3 chondrites (e.g., Guan et al., 2000; Fagan et al., 2001; Lin
433 et al., 2003). Of the 3 thin sections studied, we found only one AOA. Additionally, this
434 interpretation is supported by the relatively lower Ca/Si and Al/Si ratios of E chondrites relative
435 to other chondrite groups (e.g., Hutchison et al., 2005), and their depletion in refractory lithophile
436 trace elements (Dauphas and Pourmand, 2015). The suggestion of refractory material as chondrule
437 precursors is also consistent with the findings of Jacquet and Marrocchi (2017) that chondrules
438 show recognizable subdued group II REE patterns, presumably inherited from a refractory
439 precursor.

440 Thus, we propose that a similar pool of chondrule precursors were distributed to the different
441 (C, carbonaceous, and E, non-carbonaceous) chondrule forming regions in the protoplanetary disk
442 but with different amounts of ^{16}O -rich refractory materials, prior to development of the postulated
443 *Jupiter divide* (e.g., Warren, 2011; Kruijer et al., 2017). Such an interpretation would be consistent
444 with recent measurements of individual chondrules in C chondrites (Williams et al., 2020).
445 Williams et al. (2020) suggested the Jupiter barrier was inefficient allowing chondrules to move
446 from the inner disk to the outer disk at 2-3 Ma after CAI formation. Additionally, Schrader et al.
447 (2020) showed that dusty olivine chondrules in O and C chondrites have O-isotope compositions
448 that match those of UOC chondrule olivine ($\Delta^{17}\text{O} \sim 0\text{\textperthousand}$), suggesting an origin from a UOC source.
449 They suggested that UOC chondrules and/or chondrule fragments migrated from the inner Solar
450 System outwards to CM chondrite chondrule-forming region, beyond the orbit of Jupiter.

451 Alternatively, the range of E3 oxygen isotopes could reflect addition of heavy isotopes in more
452 volatile precursor material (e.g., cosmic symplectite and/or organic matters, Sakamoto et al., 2007;
453 Hashizume et al. 2011). The $\Delta^{17}\text{O}$ values of non-carbonaceous chondrites correlates with the
454 “oxygen excess” values calculated by the amounts of iron oxide and phosphates (Alexander et al.,
455 2019) and a positive correlation was found for $\Delta^{17}\text{O}$ and Mg # for type I chondrules in CR
456 chondrites (Tenner et al., 2015). These data suggest that the amount of H_2O may also an important
457 factor in controlling the $\Delta^{17}\text{O}$ values of chondrules. Thus, $\Delta^{17}\text{O}$ of chondrules in carbonaceous
458 chondrite groups may be partially controlled by interaction with water. However, E chondrites
459 have relatively low abundances of H_2O and do not show a positive relationship between Mg# and
460 $\Delta^{17}\text{O}$, as in the carbonaceous chondrites.

461 The trend in E3 chondrule oxygen has also been interpreted to result from reactions between
462 ^{16}O -rich, olivine-rich chondrule melts and ^{16}O -poor SiO -rich gas (Tanaka and Nakamura, 2017).

463 Tanaka and Nakamura (2017) suggested that such reactions may explain the oxygen isotopic
464 systematics, by adapting the model presented by Marrocchi and Chaussidon (2015), as well as the
465 reduced and silica-saturated compositions of E chondrite chondrules. They reported a slope of 1.27
466 for their oxygen isotopic compositions. In the Marrocchi and Chaussidon (2015) model, the
467 systematic oxygen isotopic variations in C chondrite chondrules result from open-system gas–melt
468 exchanges during chondrule formation. In their model, the PCM line does not correspond to a
469 mixing line between two discrete reservoirs but instead records the signature of high-temperature
470 interactions between nebular gas and dust (of CI composition and plotting on the carbonaceous
471 chondrite anhydrous mineral [CCAM] line) produced in the accretion disk. Tanaka and Nakamura
472 (2017) used the same model with slightly different endmember oxygen isotope ratios to explain
473 the observed EH3-4 chondrule data that plot along the line with the slope >1 . However, the slope-
474 1.27 line estimated by Tanaka and Nakamura (2017) is based on data that includes enstatite
475 chondrites more equilibrated than type 3. For their unequilibrated chondrites (EH3), they report 6
476 individual chondrules and 13 data points that are from composite chondrules or enstatite separates.
477 A regression line through their unequilibrated chondrite data, the EH3 data from Weisberg et al.,
478 (2011) and our new data here are not distinguishable from a slope-1 line.

479 Gas-melt interactions during chondrule formation has been cited as a major factor in producing
480 the petrologic features of chondrules (Ebel et al., 2018). Marrocchi and Libourel (2013) argued
481 that the sulfur content of chondrules is dependent on the fS_2 and fO_2 of the surrounding gas. Using
482 cathodoluminescence (CL) Libourel and Portail (2018) documented internal zoning of chondrule
483 olivine and interpreted it as evidence for high-temperature gas-assisted near-equilibrium epitaxial
484 growth of the olivine during chondrule formation. Jacquet et al., (2020) identified CL zoning in
485 isolated olivine grains as a signature of decreasing refractory element concentrations toward the

486 margins of grains, and took it to be evidence of interaction of the grains with the surrounding gas.
487 For enstatite chondrites, Piani et al. (2016) suggest that most of the sulfides found in EH3 chondrite
488 chondrules are magmatic minerals that formed after the diffusion of S from a volatile-rich gaseous
489 environment into the molten chondrules. Additionally, sulfidation of silicates has been proposed
490 as the mechanism for formation of Niningerite in EH3 chondrites (e.g., Rubin 1983; Lehner et al.,
491 2013). Thus, open system behavior may have also played a role in establishing the oxygen isotopic
492 compositions of E3 chondrules.

493 In the oxygen isotope exchange model proposed by Marrocchi and Chaussidon (2015) for
494 carbonaceous chondrites, they consider evaporation of CI chondritic dust that contains much more
495 C and H compared to E chondritic materials. If the ambient gas during E chondrite chondrule
496 formation was depleted in H₂O and CO molecules and enriched in SiO, gas-melt oxygen isotope
497 exchange would produce negligibly small mass dependent fractionation in chondrule melts, due to
498 small equilibrium fractionation between silicate and SiO gas (e.g., 0.39‰ at 1600K between
499 pyroxene and SiO gas; Javoy 2012). This is consistent with the observed slope ~1.0 regression line
500 determined from the E3 chondrule data presented in Fig. 6f.

501
502 *A second trend?*

503 Data from seven of the chondrules analyzed here form a separate trend above the PCM line.
504 These oxygen isotopic compositions overlap those of some OC chondrules (Kita et al., 2010) but
505 extend toward more ¹⁶O-poor compositions (Fig. 6d). These chondrules have reduced (Mg-rich)
506 silicate compositions and are petrologically similar to other chondrules in ALH 81189. The trend
507 may be the result of limited mixing between the different oxygen reservoirs. Substantial overlap
508 between the second group chondrule data and LL chondrite chondrules (Fig. 6d) may suggest some

509 mixing of OC-like precursors in E chondrite forming regions. Weisberg et al. (2011) found
510 enstatite in Sahara 97096 with oxygen isotopic ratios that plotted within the field for ordinary
511 chondrite chondrules and a relict grain within a chondrule that had a ^{17}O -rich composition similar
512 to R chondrites. They concluded that there was transport and some mixing of the materials thought
513 to have formed inside of Jupiter, i.e., the O, R and E chondrite reservoirs.

514

515 *Significance of the oxygen isotopic composition of silica*

516 The common occurrence of silica in chondrules and Si in metal in enstatite chondrites, a
517 result of their silica-saturated compositions, is one of the characteristic features that distinguish E
518 chondrites from the other chondrite groups. Silica grains from two different objects (a silica-rich
519 and a diopside-rich chondrule) in MAC 88136 are isotopically similar, plotting on the TF line at
520 $\delta^{18}\text{O} = 8.7\text{\textperthousand}$ and $8.3\text{\textperthousand}$, respectively, and have different oxygen isotope compositions than the
521 olivine and pyroxene in the same chondrules (Fig. 6b). Additionally, the silica plots further up the
522 TF line relative to whole rock compositions (Fig. 6a, b). Thus, they potentially mass balance the
523 more ^{16}O -rich compositions of some chondrules to yield the whole rock composition.

524 Kimura et al. (2005) studied silica polymorphs as a mechanism to understand the thermal
525 histories of E chondrites. They reported cristobalite and tridymite in chondrule mesostases con-
526 sistent with silica as the liquidus phase after crystallization of enstatite, and consistent with melting
527 and rapid crystallization rates for the chondrules. As the last crystallizing phase, silica may have
528 exchanged oxygen with and preserved the composition of the ambient nebular gas. It is also pos-
529 sible that silica condensed from a late stage SiO-rich nebular gas as suggested for silica-rich rims
530 on some chondrules in CR chondrites (Krot et al., 2004). The silica could also form by reduction
531 of Fe from FeO-bearing silicates as observed in some chondrules in unequilibrated E chondrites

532 (e.g., Weisberg et al., 1994). However, FeO-bearing silicates have oxygen isotopic compositions
533 similar to enstatite and forsterite in the chondrules (e.g., Kimura et al., 2003). Thus, reduction
534 reactions would not explain the observed oxygen isotopic compositions of the silica.

535 In the case of MAC 88136 C1, the $\Delta^{17}\text{O}$ of the silica grains are indistinguishable from those of
536 pyroxene in the same chondrule, but the $\delta^{18}\text{O}$ of silica grains are $\sim 3\text{\textperthousand}$ higher than pyroxene, which
537 corresponds to a temperature of equilibrium at 600-700°C using oxygen isotope thermometry
538 (Clayton and Kieffer, 1991; Javoy et al., 2012). However, this temperature seems to be too low if
539 the silica is a high temperature polymorph (as found for chondrule silica by Kimura et. al., 2005).
540 Alternatively, silica acquired higher $\delta^{18}\text{O}$ during low temperature parent body heating because of
541 its fast oxygen isotope diffusion rate compared to olivine and pyroxene (e.g., Cole and
542 Chakraborty, 2001). The difference in $\delta^{18}\text{O}$ may also be the result of applying the instrumental
543 bias of quartz to correct for cristobalite. The difference in bias between quartz and cristobalite has
544 not been evaluated but the $\Delta^{17}\text{O}$ values are not affected (Ushikubo et al., 2012).

545

546 **Conclusions**

547 1) The oxygen isotopic compositions of the silicates in most of the E3 chondrules are fairly uniform
548 within each chondrule, with a few exceptions.

549

550 2) The oxygen isotopic compositions in EH3 and EL3 chondrules overlap, suggesting that they
551 share similar precursors but the mineral assemblages and compositions in EH3 and EL3 indicate
552 different degrees of reduction (e.g., Keil, 1968; Weisberg and Kimura, 2012), suggesting
553 differences in their nebular environments and/or separate, possibly multiple (e.g., Weyrauch, et
554 al., 2018), parent bodies.

555

556 3) The limited metal-rich nodules we analyzed have oxygen isotopic ratios (as well as mineral
557 compositions) similar to the silicate (metal-free) chondrules. This is consistent with formation of
558 the metal-rich nodules prior to chondrite accretion, in an environment similar to the coexisting
559 chondrules, and their forming from a process similar to that which formed the metal-free
560 chondrules, but from more metal-rich mixtures of precursors.

561

562 4) Our data indicate a clear distinction between chondrules in the E chondrites and those in the LL
563 or R chondrules. E chondrite chondrules also differ from CV and CM chondrite groups, which
564 form a trend parallel to but below the PCM line. The chondrules from CR and E3 chondrites plot
565 closer to the PCM line than all other chondrite groups.

566

567 5) The distribution of chondrules along the PCM line is different between the E3 chondrules and
568 the CR chondrules, with CR chondrules extending toward more ^{16}O -rich compositions (Schrader
569 et al., 2013, 2014, 2017; Tenner et al., 2015) while the E3 chondrules mainly form a tighter cluster
570 of data points closer to the TF line near E3 whole rocks. A possible explanation is that a smaller
571 amount of ^{16}O -rich, refractory material was mixed into the E3 chondrule-forming region than that
572 into the carbonaceous chondrite chondrule-forming regions. Thus, we propose that similar pools
573 of chondrule precursors were distributed to the different (C, carbonaceous, and E, non-
574 carbonaceous) chondrule forming regions in the protoplanetary disk but with different amounts of
575 ^{16}O -rich refractory materials, prior to development of the postulated *Jupiter divide* (e.g., Warren,
576 2011; Kruijer et al., 2017) and/or the Jupiter divide was inefficient in separating these materials
577 (e.g., Williams et al., 2020).

578

579 6) Data from seven of the EC chondrules analyzed appear to form a separate trend above the PCM
580 line.

581

582 7) The PCM line might represent an average of the two oxygen isotope trends that Williams et al.
583 (2020) observed in CV chondrules, which they related to their $\varepsilon^{50}\text{Ti}$ and $\varepsilon^{54}\text{Cr}$ abundances. Kita et
584 al. (2010) argued that condensation of olivine from high temperature gas could have resulted in
585 deviation of oxygen isotopes to the left of the PCM line. Contribution of such fractionated
586 precursors could have been much less significant among the chondrule precursors for E chondrites,
587 explaining why the EC chondrules plot closer to the PCM line than some of the other chondrite
588 groups.

589

590

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967

968 Table 1. Oxygen isotope ratios (‰) and errors and average Mg#s for the chondrules studied from
969 ALH 81189, ALH 85159 EH3 and MAC 88136 EL3 chondrites.

970

971 Figure Captions

972 Figure 1. Mg-Ca-Al (red-green-blue) element maps of the three equilibrated enstatite chondrites
973 studied: (a) ALH81189, 3 (b) ALH 85169, 5 and (c) MAC 88136, 37. The circled/numbered areas
974 are the chondrules and other inclusions that were studied. The ones circled in white were analyzed
975 for oxygen isotopes. On these maps the brightest reds are forsterite. The duller reds are enstatite
976 which are the dominant silicate. Metal-sulfide-rich nodules appear mostly black. In (c) the green
977 area marked C1 is the diopside-rich portion of the large chondrule CI in MAC 88136.

978

979 Figure 2. Backscattered electron (BSE) images of representative chondrules and fragments in ALH
980 81189 that were analyzed for oxygen isotopes. (a) Olivine mineral fragment that is over 300 µm
981 across. (b) Enstatite-rich chondrule containing minor forsterite, troilite and interstitial Na-Al-
982 bearing mesostasis. This can be classified as a type IB porphyritic pyroxene chondrule and is
983 representative of the chondrules that are common in E3 chondrites. (c) C9 is an irregular shaped
984 object, possibly a broken chondrule, dominated by FeO-bearing pyroxene and minor silica. Such
985 objects are rare in E3 chondrites. (d) C13 is a barred chondrule with larger grains (up to ~40 µm)
986 of enstatite.

987

988 Figure 3. Images of the more unusual objects identified in the E3 chondrites studied. (a) BSE image
989 of a diopside-plagioclase inclusion in ALH 81189. (b) Mg-Ca-Al (red-green-blue) composite
990 element map of an amoeboid olivine aggregate in ALH 81189 consisting of fine olivine
991 surrounding inclusions (refractory nodules) composed of diopside and anorthite. (c) Mg-Ca-Al
992 (red-green-blue) composite element map of CI in MAC 88136, a large (~ 1mm) chondrule
993 composed of diopside and enstatite with silica and Na-Al-rich plagioclase. (d) Si-Ca-Fe (red-
994 green-blue) composite element map of C1 showing that the silica occurs mainly around the edge
995 of the diopside.

996

997 Figure 4. Images of representative metal-rich nodules that were studied. (a) Fe-Ni-S (red-green-

998 blue) composite element map of a nodule in MAC 88136 composed of schreibersite, graphite,
999 troilite, daubreelite and enstatite surrounded by Si-bearing FeNi metal alloy (red). (b) BSE image
1000 of a metal-rich nodule in ALH 81189 composed of Si-bearing FeNi metal (white), troilite (medium
1001 grey) and enstatite (dark grey). (c) BSE of a metal-rich nodule composed of Is-bearing FeNi metal
1002 intergrown with laths of enstatite. (d) BSE of a cluster of metal-rich nodules in MAC 88136
1003 composed of Si-bearing FeNi (bright white), numerous enstatite laths (dark grey) and graphite
1004 (black). Oxygen isotopes were analyzed in all these metal nodules.

1005

1006 Figure 5. Histograms showing the silicate compositions in the chondrules and other objects in the
1007 three E3 chondrites that were studied. As expected, all three E3 chondrites are dominated by
1008 enstatite with near end-member composition. Most olivine is also near end-member forsterite.
1009 FeO-bearing silicates are much less common.

1010

1011 Fig. 6 Oxygen three-isotope ratios of chondrules in E3 chondrites. (a) Oxygen isotopic ratios of
1012 individual spot analyses. Data plot along primitive chondrule mineral (PCM) line (Ushikubo et al.,
1013 2012), but mostly clustering at its intersection with the terrestrial fractionation line (TF). Other
1014 lines are from Young and Russell (1998, Y-R), and the carbonaceous chondrite anhydrous mineral
1015 line (CCAM, Clayton and Mayeda, 1984). (b) Oxygen 3-isotope diagram showing average oxygen
1016 isotope ratios for each chondrule analyzed in ALH 81189, ALH 85159 and MAC 88136. A blue
1017 dashed line is drawn parallel to the PCM line through seven E chondrite objects that lie above the
1018 PCM, possibly forming a second trend. (c) Oxygen isotope ratios of an AOA in ALH 81189, relict
1019 olivine in ALH 81189 C1, and individual pyroxene analyses from ALH 85159 C51. Inset enlarges
1020 the data from ALH 85159 C51 and ALH 81189 C1 which plot above the TF line. (d) E3 chondrules
1021 are plotted in two groups; majority of chondrules that plot on the PCM line and *other chondrules*
1022 that along with relict olivine in ALH 81189 C1 and heterogeneous chondrule ALH 85159 C51.
1023 The regression lines (dashed lines) and MSWD of the fits of two groups are shown. Chondrule
1024 data from LL3 chondrites are shown as comparison. (e) Data from our previous work on other E3
1025 chondrites (Weisberg et al., 2011) showing a similar trend with some chondrule olivine plotting
1026 toward more ^{16}O -rich compositions. E3 chondrules determined from bulk methods (C&M1985:

1027 Clayton and Mayeda, 1985; T&N2017: Tanaka and Nakamura, 2017) are also show, which gen-
1028 erally agree within the range observed in our new dataset. (f) The regression line of all E3 chon-
1029 drule data. Data used for estimating regression lines is shown in EA4.

1030

1031 Fig 7. (a) Average $\Delta^{17}\text{O}$ vs. average Mg# for the type I (FeO-poor) chondrules, fragments and
1032 metal-rich nodules in ALH 81189 EH3, paired ALH 85159 and MAC 88136 EL3 chondrites. (b)
1033 Average $\Delta^{17}\text{O}$ vs. average Mg# for all chondrules from this study and Weisberg et al. (2011). The
1034 lines are best fit lines through the data ($r=0.2$). The AOA is not included.

1035

1036 Fig. 8. Comparison of the chondrule oxygen isotope data for E chondrite chondrules to those in
1037 Acfer 094 and CR, CV and CM carbonaceous chondrites and LL ordinary chondrites. Data from
1038 this study, Ushikubo et al. (2012), Chaumard et al. (2018), Tenner et al. (2015)' Hertwig et al.
1039 (2018, 2019b), Kita et al. (2010).

1040

1041 **Captions for Supplementary Materials**

1042 Figure S1. BSE images (left) and SE images (right) of the chondrules from ALH 81189,3, ALH
1043 85159, 5 and MAC 88136, 37 that were analyzed in this study. Images on the right show SIMS
1044 spots and images on the left show the pits from the analyses.

1045

1046 Table S1. Mineral compositions (wt. % oxide) of the chondrules studied.

1047

1048 Table S2. Oxygen isotopic compositions of the standards measured using SIMS for calibration of
1049 instrumental mass fractionation.

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1051 Table S3. Oxygen isotopic compositions of the minerals analyzed in the chondrules from ALH
1052 81189, 3, ALH 85159, 5 and MAC 88136, 37.

1053

1054 Table S4. Data from this study and Weisberg et al. (2011) used for estimating the regression in
1055 Fig. 6f.

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Table 1. Oxygen isotope ratios (‰) and errors and average Mg#s for the chondrules studied from ALH 81189, ALH 85159 EH3 and MAC 88136 EL3 chondrites.

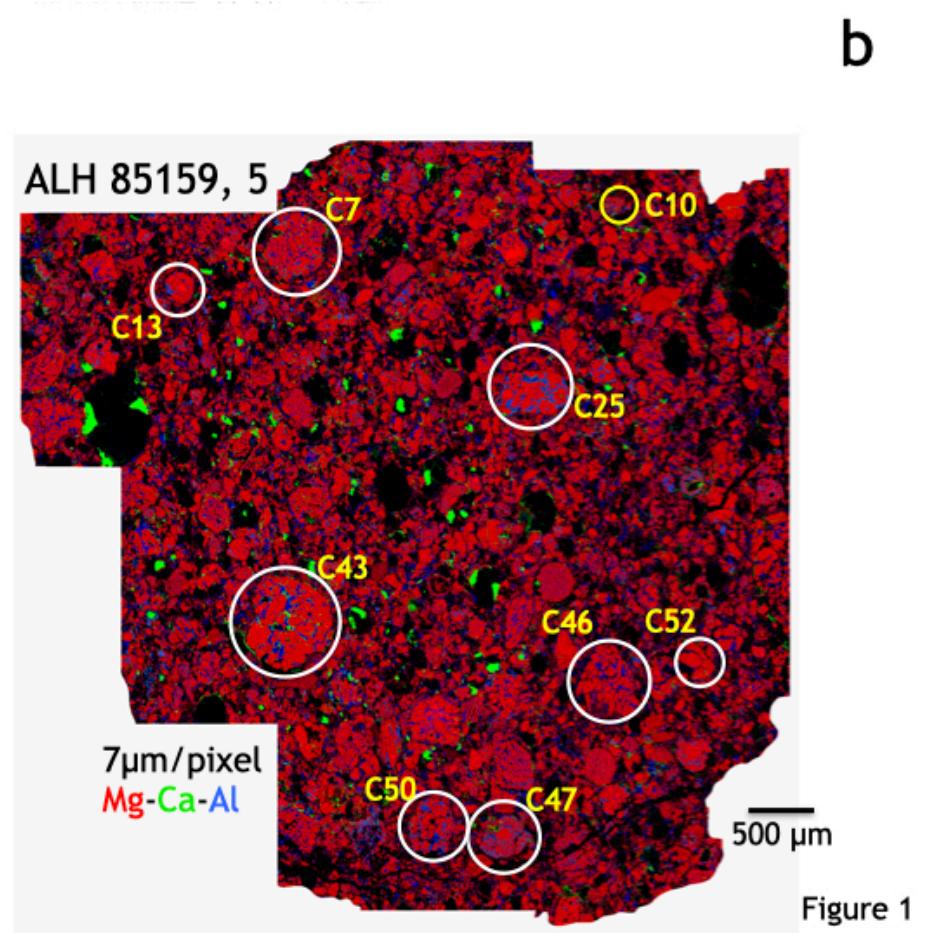
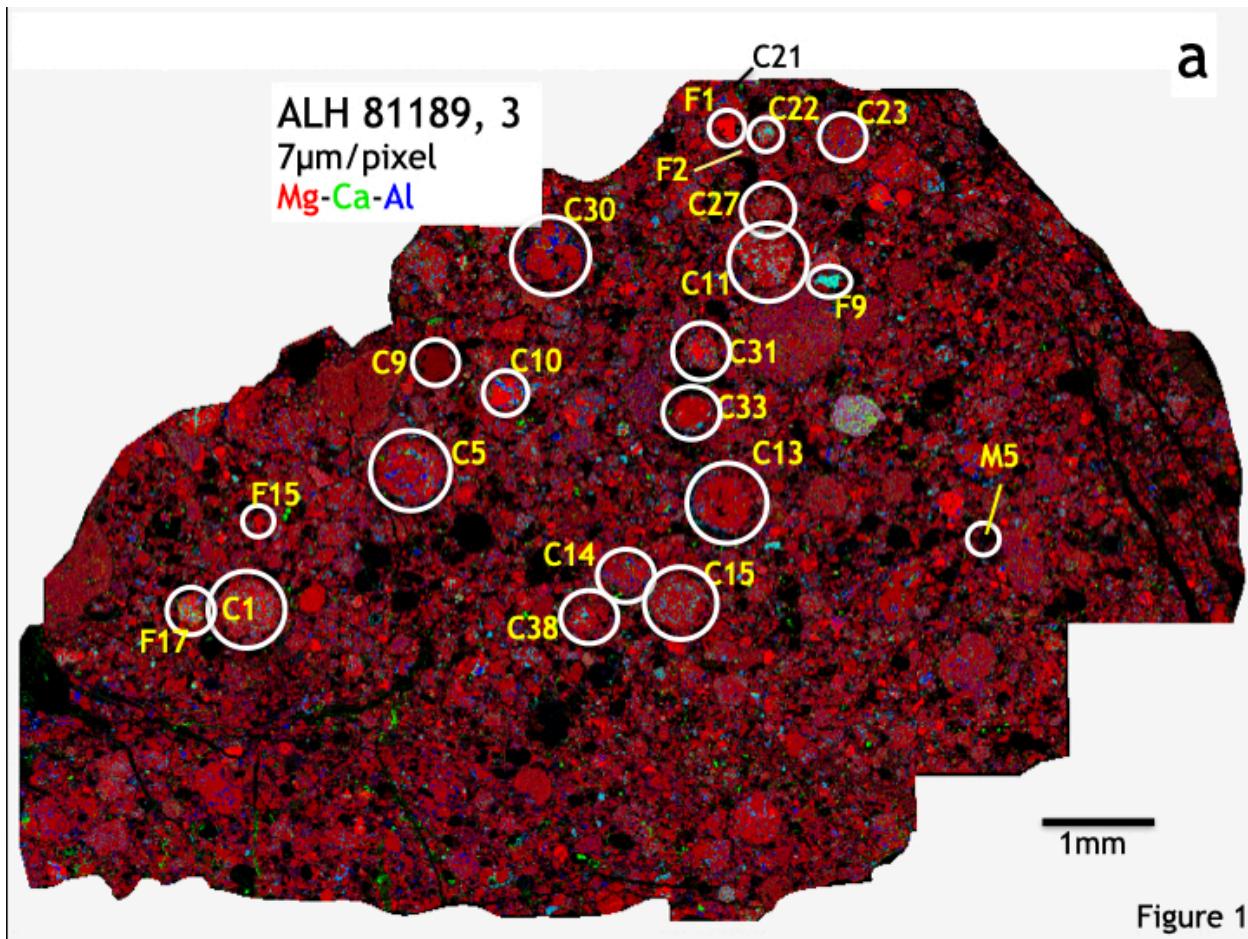
Meteorite	sample	Type	Phase	Avg. Mg#	N for mean*	$\Delta^{18}\text{O}$ (‰)	Unc.	$\Delta^{17}\text{O}$ (‰)	Unc.	$\Delta\Delta^{17}\text{O}$ (‰)	Unc.
ALH 81189	C1	PP	En	99.3	5	6.08	0.46	3.96	0.28	0.80	0.13
ALH 81189	C5	PP	En	98.4	6	5.10	0.33	2.44	0.18	-0.21	0.12
ALH 81189	C9	PP	Pyx	81.0	6	4.34	0.34	1.76	0.23	-0.49	0.17
ALH 81189	C10	PO	Fo, En	98.3	6	5.08	0.32	2.38	0.18	-0.26	0.14
ALH 81189	C11	PP	Ol, En	98.7	6	5.17	0.35	3.53	0.31	0.85	0.22
ALH 81189	C13	Barred	Fo, En	99.0	6	4.41	0.35	1.64	0.27	-0.65	0.16
ALH 81189	C14	PP	En	98.9	6	5.23	0.32	2.45	0.25	-0.27	0.17
ALH 81189	C15	PP	En	98.8	6	5.34	0.32	2.72	0.26	-0.05	0.20
ALH 81189	C21	POP	Fo, En	98.8	4	4.73	0.39	2.35	0.27	-0.11	0.16
ALH 81189	C22	PP	En	99.3	4	2.33	0.50	-0.15	0.23	-1.36	0.16
ALH 81189	C23	PP	En	98.2	6	5.08	0.38	2.70	0.24	0.05	0.22
ALH 81189	C27	PP	En	99.1	1	6.40	0.40	4.13	0.28	0.81	0.21
ALH 81189	C30	PP	En	98.8	6	4.83	0.34	2.31	0.30	-0.20	0.20
ALH 81189	C31	POP	En	99.1	6	5.48	0.31	3.84	0.21	0.99	0.13
ALH 81189	C33	PP	En	98.1	6	5.30	0.36	2.78	0.23	0.03	0.16
ALH 81189	C38	PP	En	99.3	6	4.73	0.43	2.26	0.23	-0.20	0.14
ALH 81189	F1	Min Frag	Ol	98.0	4	4.86	0.32	2.35	0.22	-0.17	0.16
ALH 81189	F2	Min Frag	Ol	98.3	4	4.91	0.34	2.36	0.19	-0.19	0.18
ALH 81189	F5	PO	En	98.7	6	4.42	0.34	1.99	0.19	-0.31	0.12
ALH 81189	F9	Diops-plag	En	97.7	3	4.06	0.85	1.47	0.61	-0.64	0.19
ALH 81189	F17	AOA	Ol	99.7	2	-46.49	0.34	-47.98	0.24	-23.81	0.20
ALH 81189	M5	MN	En	98.5	4	4.44	0.43	2.09	0.30	-0.22	0.18
ALH 85159	C7	PP	En	99.1	6	5.77	0.65	3.25	0.35	0.25	0.12
ALH 85159	C13	PP	En	99.2	6	5.54	0.35	2.70	0.25	-0.18	0.18
ALH 85159	C25	PP	En	99.6	5	6.26	0.36	3.55	0.21	0.30	0.15
ALH 85159	C43	PP	En	99.8	6	5.97	0.35	3.18	0.20	0.08	0.16
ALH 85159	C46	PP	En	98.5	6	6.39	0.39	3.54	0.32	0.22	0.19
ALH 85159	C47	PP	En	99.4	6	6.57	0.40	3.76	0.19	0.34	0.13
ALH 85159	C50	PP	En	99.1	6	6.28	0.39	3.44	0.26	0.17	0.12
ALH 85159	C51	PP	En	99.3	6	6.85	0.78	4.84	0.61	1.27	0.24
ALH 85159	C52	PP	En	98.9	6	5.82	0.37	2.79	0.23	-0.24	0.18
MAC 88136	C1	Diops-rich	Diops/En	99.4	4	5.15	0.32	2.64	0.21	-0.04	0.16
MAC 88136	C1	Diops-rich	silica		1	8.33	0.15	4.22	0.23	-0.12	0.25
MAC 88136	C1	Diops-rich	Silica		1	8.42	0.15	4.26	0.23	-0.12	0.25
MAC 88136	C3	PP	En	99.6	6	5.40	0.40	2.69	0.33	-0.12	0.19
MAC 88136	C6	PP	En	99.6	6	5.51	0.40	2.51	0.31	-0.35	0.20
MAC 88136	C7	PP	En	99.5	5	5.31	0.32	2.39	0.25	-0.38	0.21
MAC 88136	C9	PP	En	99.5	6	6.42	0.34	4.34	0.35	1.00	0.28
MAC 88136	C10	PP	En	99.6	6	5.45	0.35	2.94	0.38	0.11	0.30
MAC 88136	C11	BP	En	99.5	7	5.95	0.34	3.17	0.32	0.07	0.29
MAC 88136	C16	PP	En	99.0-99.9	6	5.93	0.34	3.10	0.25	0.01	0.20
MAC 88136	C17	PP	En	99.4	6	5.65	0.32	2.94	0.28	0.00	0.27
MAC 88136	C30	Silica-rich	Silica		1	8.73	0.15	4.55	0.23	0.01	0.25
MAC 88136	M1	MN	En	NA	4	5.55	0.35	2.81	0.24	-0.08	0.17
MAC 88136	M8	MN	En	98.8	2	5.81	0.38	3.07	0.27	0.05	0.19
MAC 88136	M10	MN	En	99.3	3	5.74	0.45	3.02	0.40	0.04	0.27

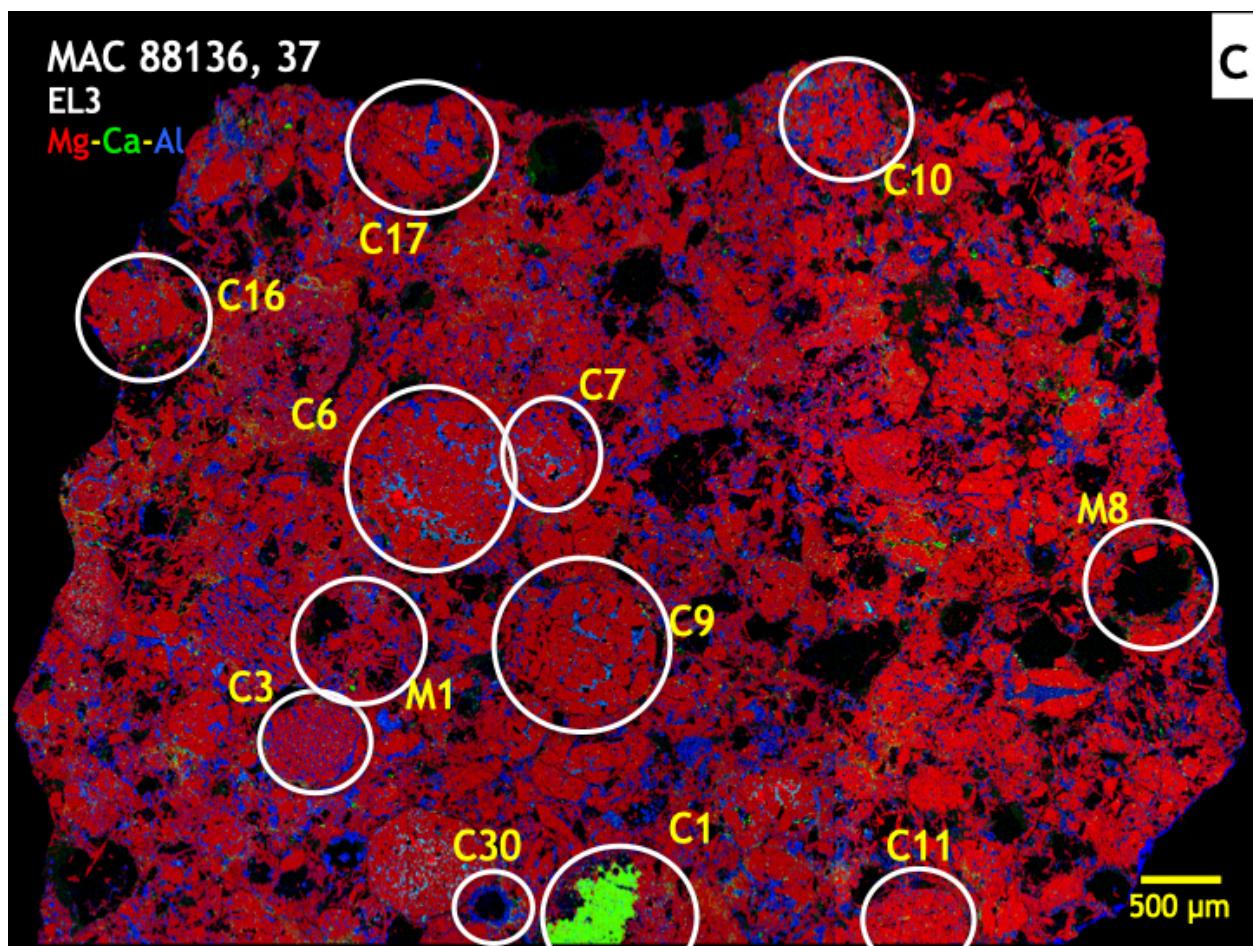
Unc. - Uncertainties of host oxygen isotope ratios are propagated, combining the 2SE of chondrule measurements. Mg# = Mg/(Mg+Fe) X 100 and is the average value for pyroxene and/or olivine for each chondrule.

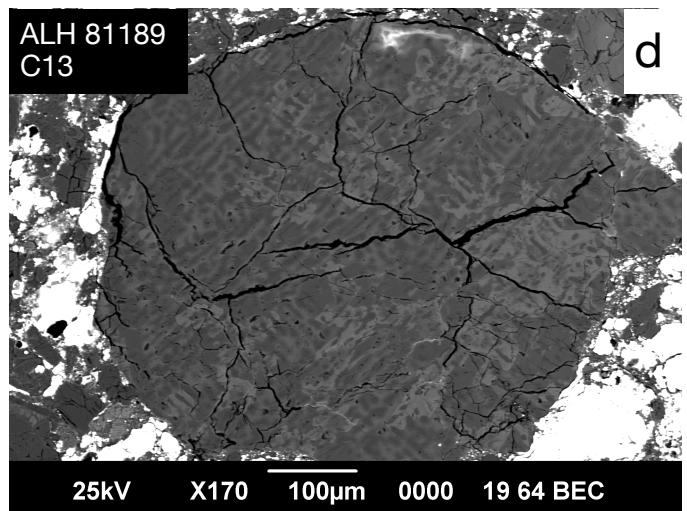
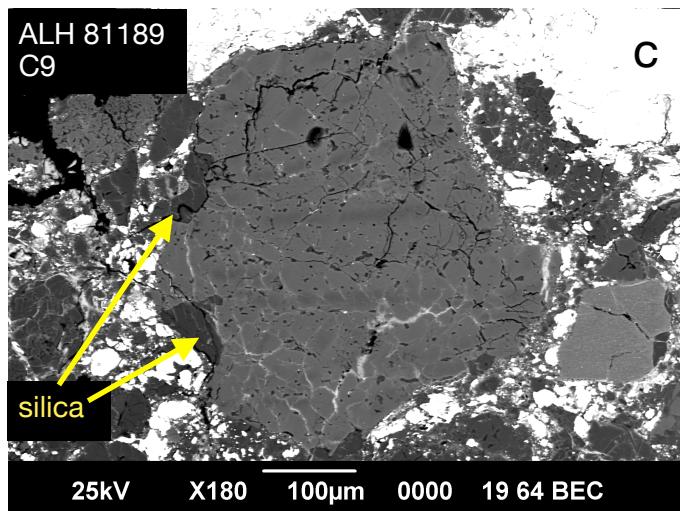
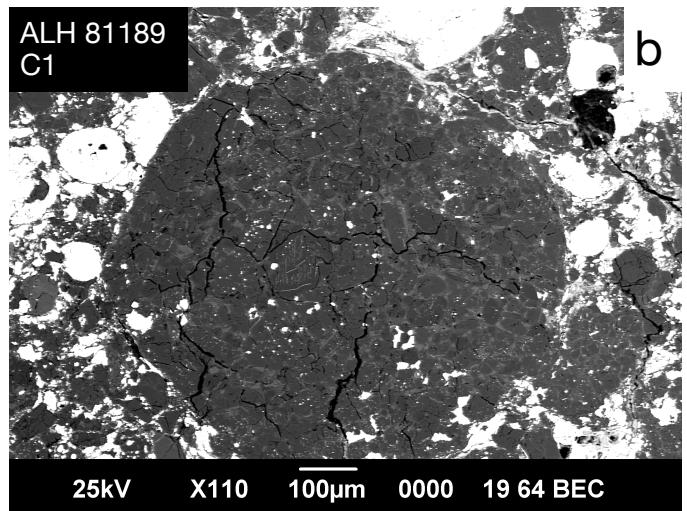
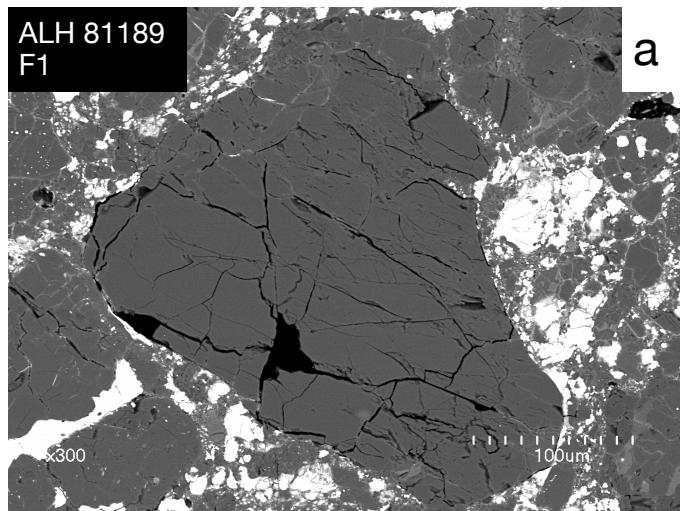
Phases: En-enstatite, Ol- olivine, Pyx - pyroxene, diops-diopside

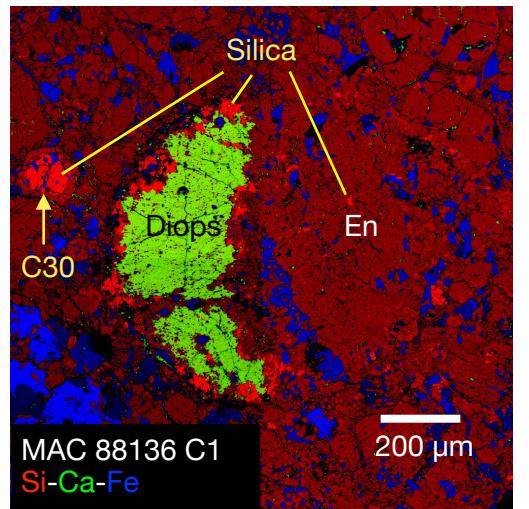
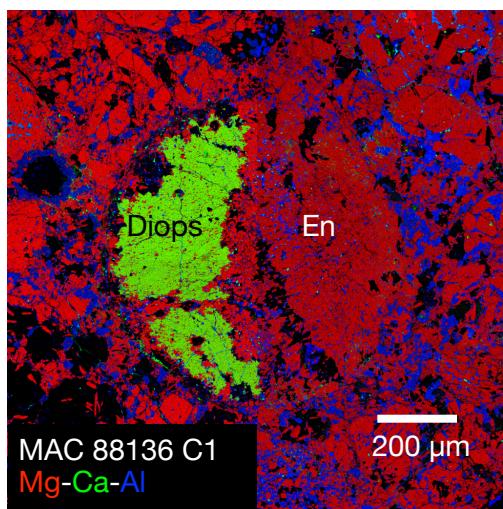
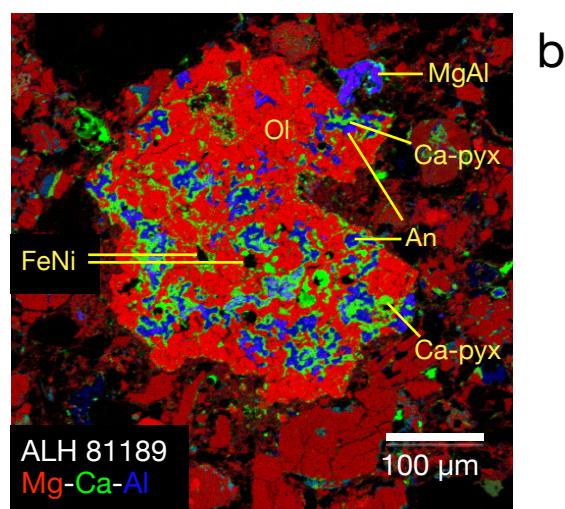
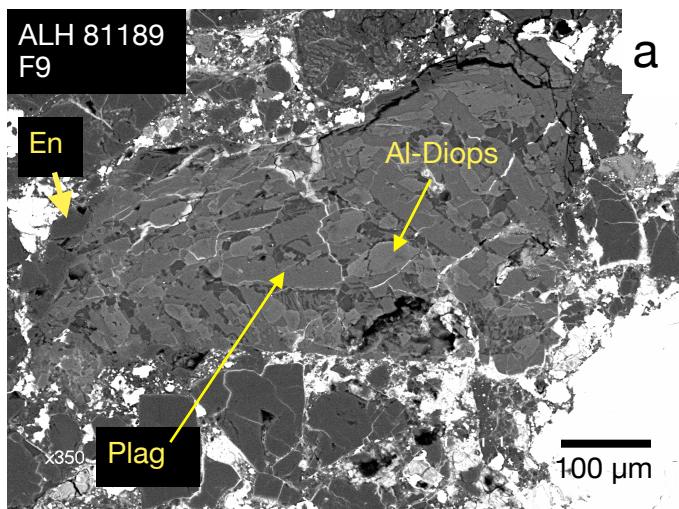
Chondrule types: PP- porphyritic pyroxene, POP- porphyritic olivine and pyroxene, PO - porphyritic olivine, Min Frag - mineral fragment, AOA - amoeboid olivine aggregate, MN - metal-rich nodule.

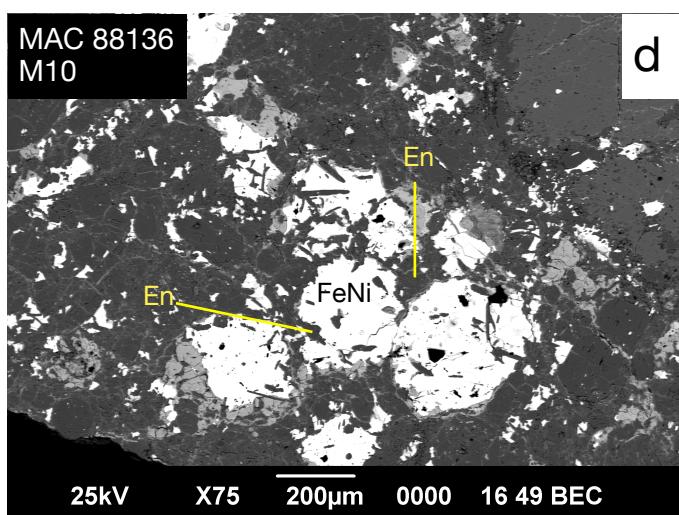
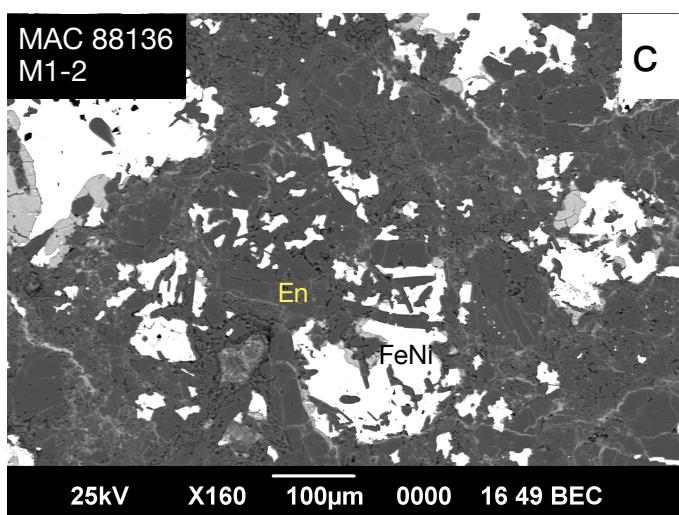
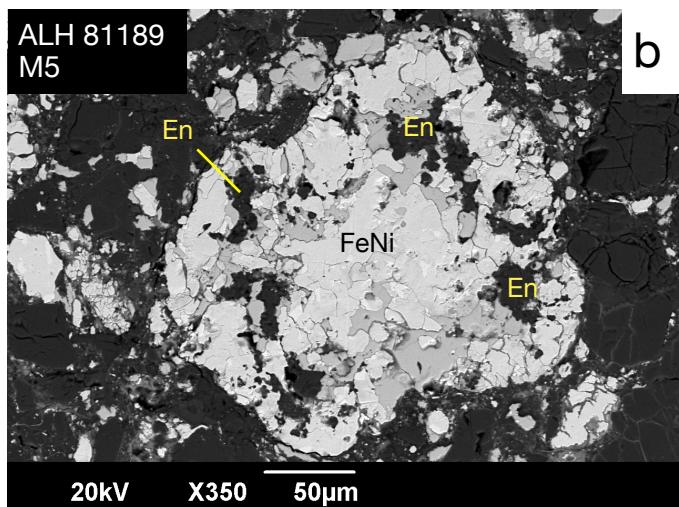
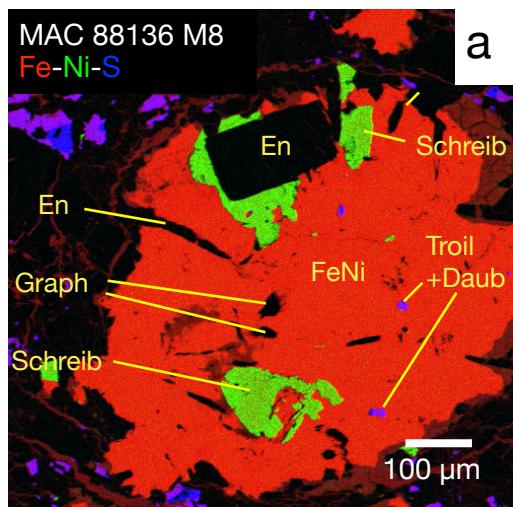
N - the number analyses averaged for the oxygen isotope compositions given.











Average Silicate Compositions

