

Migratory Functionalization of Unactivated Alkyl Bromides for Construction of All-Carbon Quaternary Centers *via* Transposed *tert*-C-Radicals

Chuan Zhu¹⁺, Ze-Yao Liu¹⁺, Luning Tang¹, Heng Zhang¹, Yu-Feng Zhang¹, Patrick J. Walsh^{2*} and Chao Feng^{1*}

Despite remarkable recent advances in transition-metal-catalyzed C(sp³)-C cross-coupling reactions, there remains challenging bond formations. One class of such reactions include formation of *tertiary*-C(sp³)-C bonds, presumably due to unfavorable steric interactions and competing isomerizations of tertiary alkyl metal intermediates. Reported herein is a Ni-catalyzed migratory 3,3-difluoroallylation of unactivated alkyl bromides at remote tertiary centers. This approach enables the facile construction of otherwise difficult to prepare all-carbon quaternary centers. Key to the success of this transformation is an unusual remote functionalization via chain walking to the most sterically hindered tertiary C(sp³) center of the substrate. Preliminary mechanistic and radical trapping studies with primary alkyl bromides suggest a unique mode of tertiary C-radical generation through chain-walking followed by Ni-C bond homolysis. This strategy is complementary to the existing coupling protocols with *tert*-alkyl organometallic or -alkyl halide reagents, and it enables the expedient formation of quaternary centers from easily available starting materials.

¹ Institute of Advanced Synthesis (IAS), School of Chemistry and Molecular Engineering, Nanjing Tech University, 30 South Puzhu Road, Nanjing 211816, P. R. China. ² Roy and Diana Vagelos Laboratories, Department of Chemistry, University of Pennsylvania, 231 South 34th Street, Philadelphia, PA 19104, USA. *e-mail:: pwalsh@sas.upenn.edu; jamcfeng@nitech.edu.cn

+ These authors contribute equally to this work.

Transition-metal-catalyzed construction of all-carbon quaternary centers via *tert*-C(sp³)-C coupling reactions represents a significant synthetic challenge. Not only are severe steric effects encountered around the metal center in such coupling reactions, but competing isomerization pathways of alkylmetal intermediates often have low barriers¹⁻⁴. Nevertheless, the development of cross-coupling protocols that make use of tertiary alkyl-M (M = Mg,⁵⁻⁹ Zn,¹⁰ B¹¹, and Na^{12,13}) reagents has aroused substantial interest from the synthetic community. While some advances have been achieved, restricted substrate scopes, together with the need to synthesize the organometallic reagents, has severely limited application of this strategy (Figure 1a). In this regard, the direct functionalization of *tert*-alkyl electrophiles offers advantages from the perspective of practicality and step economy¹⁴⁻¹⁷. Of note, recent progress in the area of reductive coupling¹⁸⁻²⁰ with organic halides and pseudohalides²¹⁻²⁵ or alkenes^{25,26} have expanded classes of viable coupling partners. These studies provide complementary and efficient avenues to access structurally diverse three-dimensional scaffolds under mild reaction conditions (Figure 1b). For example, the elegant work of Gong's group showcases the generality of this strategy, allowing arylation, alkylation and allylation of tertiary alkyl halides through Ni-catalyzed reductive cross-electrophile couplings.²¹⁻²⁴ In addition to the above mentioned direct coupling manifolds, carbofunctionalization of 1,1-disubstituted or trisubstituted alkenes is gaining momentum (Figure 1c).²⁸⁻³² Representative examples in this vein include Shenvi's hydroarylation/alkylation of unactivated alkenes through Fe/Ni or Mn/Ni co-catalysis^{29,30} and Brown's diarylation and aryloborylation of trisubstituted alkenes^{31,32}. Although these methods enable access to *tert*-C-C linkages, development of strategically different approaches remain in high demand.

By exploiting iterative hydrometallation and β -hydride elimination, chain-walking enables the site-selective cross-coupling at positions remote to the initial metallation site^{33,34}. Owing to the efforts of Sigman³⁵⁻³⁷, Marek³⁸⁻⁴⁰, Mazet⁴¹⁻⁴³, Martin⁴⁴⁻⁴⁷, Zhu⁴⁸⁻⁵⁷, and others⁵⁸⁻⁶⁶, a collection of remote functionalizations, including arylation, alkylation, carboxylation, amination, borylation and thiolation of unactivated alkenes or alkyl halides have been developed. Very recently, our team leveraged the fluorine-effect for a remote fluoro-alkenylation of unactivated alkyl bromides (Figure 1d)⁶⁷. Our system is like other remote functionalization reactions, where the driving force for chain-walking is moving the system lower on the energy landscape by positioning the metal center at a stabilizing position (usually limited to benzylic or α to boron). To expand the scope of remote functionalization reactions, alternative sites must be targeted, such as tertiary centers. With our continuing

interest in remote functionalization, we have uncovered a mechanistically distinct and highly regioselective migratory 3,3-difluoroallylation⁶⁸⁻⁷¹ of unactivated alkyl bromides at tertiary carbon centers. This undirected *tert*-C(sp³)-H functionalization nicely complements existing methods for all-carbon quaternary center construction, especially when tertiary alkyl halide/metal reagents are not readily available or not stable. Notably, during the preparation of this manuscript, Zhu and co-workers reported a relevant Ni-H-catalyzed migratory defluorinative olefin cross-coupling⁵⁷.

Herein, we demonstrate that unactivated primary and secondary alkyl bromides are competent precursors for generating tertiary alkyl coupling partners via Ni-H-mediated chain-walking (Figure 1e).

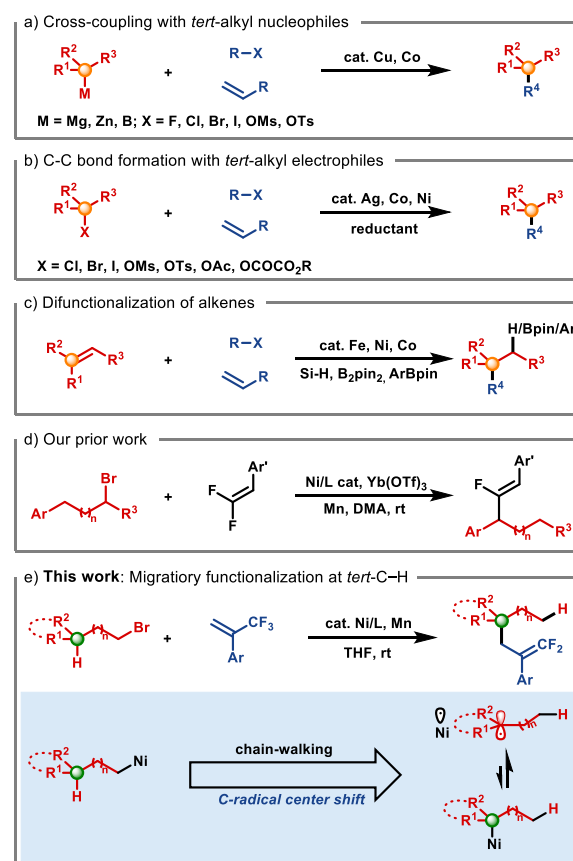


Figure 1. Transition metal catalyzed formation of quaternary centers. **a.** Standard cross-coupling approach, **b.** cross-electrophile coupling, **c.** difunctionalization of alkenes, **d.** our prior work, **e.** difluoroallylation of alkyl bromides with chain walking via transition-metal-catalyzed *tert*-C-C bond-formation.

Results

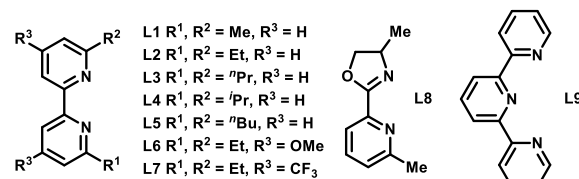
Reaction optimization. A selection of alkyl halides was employed to react with α -trifluoromethylstyrene **2a**. After initial screening, (bromomethyl)cyclohexane was

successfully coupled with **2a** at the *tertiary* position with good regioselectivity (>20:1) 64% assay yield in the presence of NiBr₂·glyme, 6,6'-dimethyl-2,2'-bipyridine (**L1**) and Mn as terminal reductant (Table 1, entry 1, AY determined by integration of the ¹⁹F NMR spectrum against an internal standard). Given the pivotal role of ligands in Ni-catalyzed remote functionalizations, a series of bidentate N-donor ligands were examined to improve the reaction outcome. Substitution next to the nitrogens of the bipy ligands was found indispensable. Without either one or two methyl groups positioned ortho to the nitrogens, no product was observed. This observation is in accordance with previous reports (Supplementary Table 1)^{46,53,54}. We hypothesized that increasing the steric bulk around the metal coordination site would enhance the reaction efficiency. Thus, a series of increasingly bulky substituents, such as ethyl, propyl, and butyl, were subsequently examined (entries 2–7). This study led to 6,6'-diethyl-2,2'-bipyridine (**L2**) as the top candidate, furnishing product **3a** in improved yield and comparable regioselectivity (72% AY and >20:1 regioisomeric ratio, Table 1, entry 2). In addition, with pyrox or terpyridine ligands, essentially no reaction occurred (Table 1, entries 8 and 9). A solvent screen revealed that THF was the optimal choice (Table 1, entries 10–12), allowing the formation of the product in 89% isolated yield with >20:1 regioisomeric ratio. The influence of reductant was also examined. Mn proved superior to Zn, B₂pin₂, diethoxymethylsilane and HCOONa, which are commonly employed in reductive cross-coupling reactions (Supplementary Table 4).

Table 1. Optimization of the reaction conditions.^[a]

Entry	Ligand	Solvent	Yield ^[b]	rr ^[c]
1	L1	DMA	64	>20:1
2	L2	DMA	72	>20:1
3	L3	DMA	70	>20:1
4	L4	DMA	54	>20:1
5	L5	DMA	62	>20:1
6	L6	DMA	62	6:1
7	L7	DMA	n.r.	-
8	L8	DMA	n.r.	-
9	L9	DMA	n.r.	-
10	L2	DMSO	44	1:1

11	L2	1,4-Dioxane	n.r.	-
12	L2	THF	89 ^[d]	>20:1



[a] Unless otherwise noted, reactions were carried out with **1a** (0.6 mmol), **2a** (0.2 mmol), Ni salt (0.01 mmol), ligand (0.012 mmol), Mn powder (0.3 mmol) in DMA (2.0 mL) under a nitrogen atmosphere at room temperature for 12 h. [b] Assay yields determined by ¹⁹F NMR with 1-iodo-4-(trifluoromethyl)benzene as internal standard. [c] Rr refers to the ratio of desired product to the sum of all the other regioisomers, as determined by the ¹⁹F NMR analysis of the crude products. [d] Isolated yields.

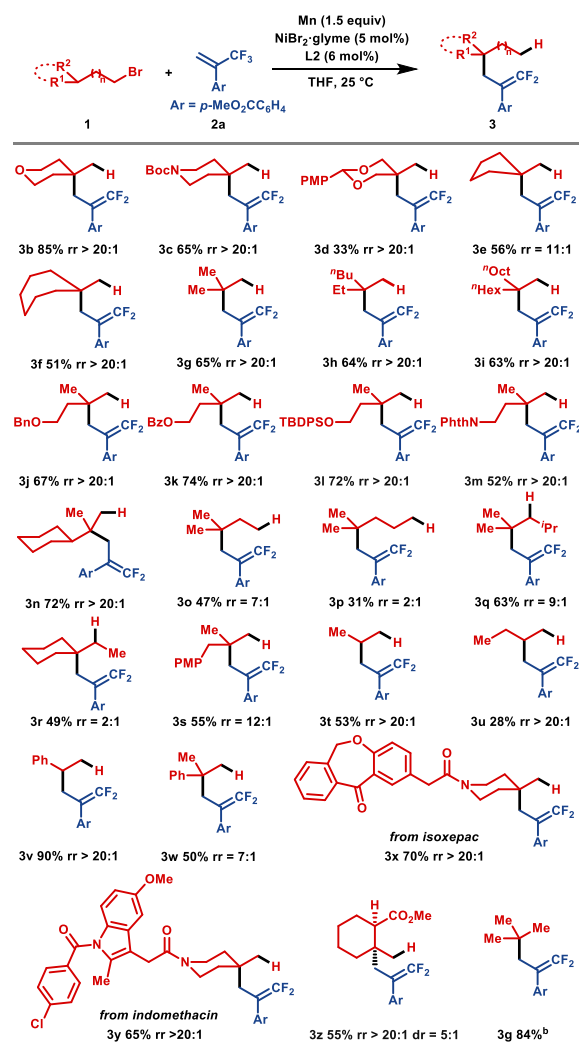
3,3-Difluoroallylation of unactivated alkyl bromides.

With the optimized reaction conditions in hand, the reaction scope of alkyl bromides was examined (Table 2). We found that a broad range of unactivated alkyl bromides were suitable substrates for the difluoroallylation. Cyclic alkyl bromides containing heteroatoms, such as oxygen and *N*-Boc, were well-tolerated, affording the 3,3-difluoroallylated products in 85 and 65% yields with excellent regioselectivities (**3b** and **3c**). Notably, a cyclic acetal was tolerated to afford the desired product with good regioisomeric ratio, albeit in diminished yield (**3d**). Examination of the 5- and 7-membered carbocycles resulted in good regioselectivities (>11:1) with yields of 56 (**3e**) and 51% (**3f**) under the standard conditions. The lowered rr of 11:1 for **3e** may be due to increased strain in the β-H elimination transition state. We were pleased to find that acyclic alkyl bromides provided products containing quaternary centers in 63–65% and high regioselectivities (>20:1, **3g–3i**). Ester, ether, silyl ether and phthalimide moieties were well tolerated, affording the corresponding 3,3-difluoroallylation products in good yields (**3j–3m**). Interestingly, substrates containing two contiguous tertiary carbon centers only led to the migratory product at the proximal site (**3n**). Importantly, it was found that the migration could proceed over more than one C–C bond, albeit with progressively decreased reaction efficiency and regioisomeric ratio (**3o**, 47% yield with 7:1 rr and **3p**, 31% yield with 2:1 rr). Nonetheless, these results highlight the selectivity of the present catalytic system toward tertiary carbon centers over secondary and primary positions. This trend is also observed in the formation of product **3q** and **3r**, where *sec*-alkyl bromides reacted ultimately giving predominantly coupling products at the tertiary site. Intramolecular competition revealed that the tertiary carbon was more favorable than 1° or 2°

and even preferred over benzylic positions (**3s**). These findings stand in contrast to previous disclosures⁵². To further distinguish reactivity between secondary and primary sites, *n*-propyl and *n*-butyl bromide were examined (**1t** and **1u**). It was found that coupling occurred more readily at the more congested secondary position (**3t**, 53% and rr >20:1; **3u**, 28%, rr >20:1). It is noteworthy that chain-walking to more congested positions, in the absence of stabilizing groups, has not been previously realized.

It is interesting that the present reaction system can also be used to functionalize remote benzylic positions

Table 2. Scope with respect to alkyl bromides.^a

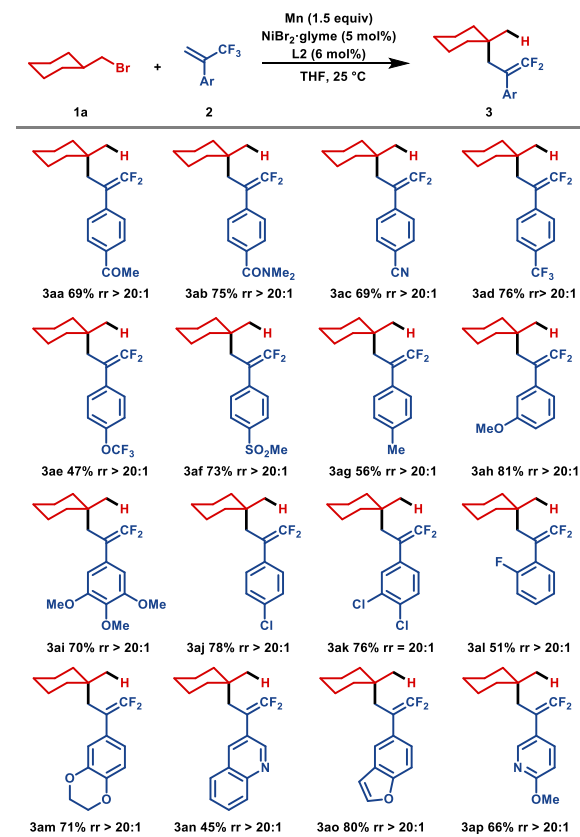


[a] See the Supplementary Information, pages 27–37, for experimental details. Rr refers to the ratio of desired product to the sum of all the other regioisomers, which was determined by the ¹⁹F NMR or GC analysis of the crude products. [b] From *tert*-butyl bromide.

with high efficiency and selectivity (**3v**, 90% yield, >20:1 rr and **3w**, 50% yield with 7:1 rr). Pleasingly, drug derived substrates performed well in the reaction (**3x** and **3y**), demonstrating the synthetic potential of the difluoroallylation in late-stage modification of complex molecules. Furthermore, the diastereoselectivity of this transformation was assessed with enantioenriched substrate **1z**, which delivered the migratory product **3z** with 5:1 dr. Not unexpectedly, the optimized reaction conditions were applicable to the difluoroallylation of tertiary alkylbromide (**3g** from *tert*-BuBr). Finally, 1 mmol scale reaction was accomplished by using commercially available ligand (**L1**) with comparable efficiency, affording **3a** in 65% yield with >20:1 rr.

Reaction scope with trifluoromethylalkenes. We next evaluated different trifluoromethylalkene substrates in this transformation (Table 3). Reactions carried out with α -trifluoromethylstyrenes bearing a wide range of functional groups on the aryl moiety, such as ester, ketone, cyanide, CF₃, OCF₃, sulfone, Me and OMe, all underwent coupling smoothly to afford the desired products in good yields (47–81%) and excellent

Table 3. Scope with trifluoromethyl alkenes.



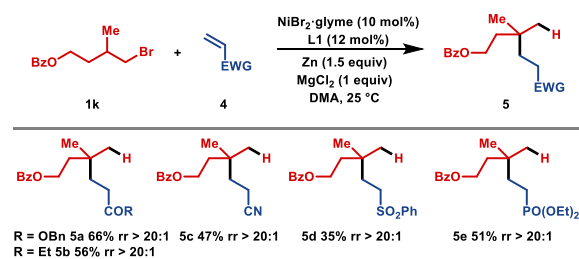
[a] See Supplementary Information, pages 37–44, for experimental details. Rr refers to the ratio of desired product to the sum of all the

other regioisomers, which was determined by the ^{19}F NMR or GC analysis of the crude products.

selectivities (all >20:1, **3aa–3ai**). In addition, substrates containing Cl or F on the aryl were compatible with the transformation (**3aj–3al**, 51–78% yield, all >20:1 rr). Fortunately, heterocyclic trifluoromethyl alkene substrates (**3am–3ap**) were also well tolerated (45–80% yield, >20:1 rr).

Reaction scope with other activated olefins. To expand the scope of this transformation beyond trifluoromethylalkenes, other electron deficient olefins were examined. To our delight, acrylate, vinyl ketone, acrylonitrile, vinyl sulfone and vinyl phosphonate derivatives were amenable under slightly modified reaction conditions. These substrates furnished migratory alkylation products in synthetically useful yields with excellent regioselectivities. These outcomes expand the synthetic reach of this *tert*-carbon-selective remote functionalization strategy, enabling the construction of quaternary carbon centers decorated with diverse functionality.

Table 4. Scope of activated alkenes.

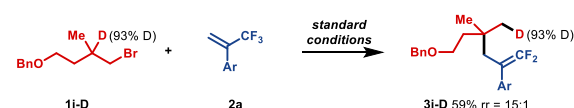


[a] See Supplementary Information, pages 45–47 for experimental details. Rr refers to the ratio of desired product to the sum of all the other regioisomers, which was determined by the GC analysis of the crude products.

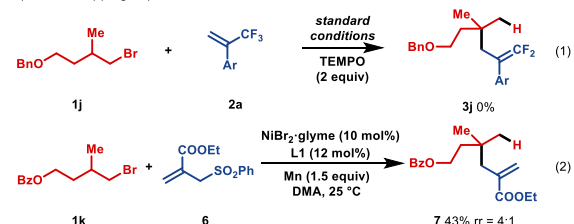
To probe the mechanism of this migratory defluorinative allylation reaction, a set of control experiments were performed. To determine if chain-walking was indeed operating in the present system, the isotope-labelled substrate **1j-D** was examined (Figure 2a). As expected, the deuterium located at the tertiary carbon was selectively transferred to the primary position. This result strongly supports the involvement of chain-walking. It is notable that no deuterium was observed at other positions in the product. We hypothesized that radical intermediates may be involved and, therefore, conducted the reaction in the presence of TEMPO. The radical scavenger TEMPO suppressed the reaction and 97% of **2a** remained, supporting the involvement of radical intermediates (Figure 2b, eq 1). Additionally, when **2a**

was replaced by allylic sulfone **6**, allylation proceeded, suggesting the existence of 3° C-radical intermediate under the catalytic conditions (Figure 2b, eq 2). To further elucidate the mode of C–C bond formation, cyclic β -pinene-derivative (**8**) was subjected to the reaction (Figure 2c). The observation of radical intermediates would be expected to result in ring-opened products, whereas a two-electron process would leave the ring intact. In the event, the resulting ring-opening product (**9**) was exclusively obtained. To explain the results in Figure 2, we propose a tertiary carbon radical is generated and participates in the crucial C–C bond formation step^{72–75}. The oxidative addition of alkyl bromides to low-valent Ni catalysts usually takes place through a cascade of single electron transfer and alkyl radical generating steps⁷⁶. Such transformations, therefore, can be viewed as unusual radical center shifts that are mediated by transition metal catalysts. We believe the steric hindrance encountered at *tert*-C–Ni linkage is conducive to the homolytic rupture of the C–Ni bond, affording tertiary carbon-centered radicals that are a sufficiently long lived to escape the solvent cage and selectively react with trifluoromethylalkene derivatives.

a) D-labelling experiment



b) radical trapping experiments



c) ring-opening experiment

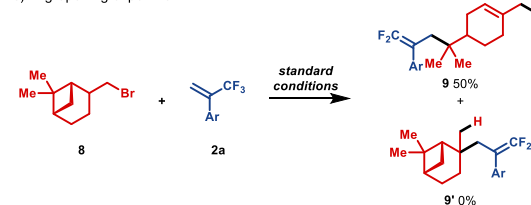


Figure 2. Control experiments (Ar = 4-MeO₂C–C₆H₄). **a.** a deuterium-labelling experiment; **b.** radical trapping experiments; **c.** ring-opening experiment.

Taken together, these findings demonstrate that a radical-engaged chain-walking manifold is involved, which accounts for the unusual selectivity that leads to functionalization at the more congested tertiary position. A proposed mechanism is outlined in Figure 3. The reaction is initiated by the oxidative addition of alkyl bromide **1** to active Ni complex **I** to afford intermediate **II**.

Subsequently, chain-walking of the nickel catalyst from the terminal carbon to the tertiary center via β -hydride elimination and insertion steps allows for the facile generation of *tert*-C-Ni complex **IV**. The Ni-C to the tertiary carbon bond has the lowest BDE and undergoes homolysis, generating a tertiary C-radical **V**. The radical can undergo addition to the trifluoromethylalkene **2** to form a new radical species **VI**. The newly-formed radical intermediate then recombines with the nickel complex to give rise to intermediate **VII**, which undergoes β -fluoride elimination to produce the observed product **3**, accompanied by the generation of F-Ni-complex **VIII**. Finally, reduction of Ni-complex **VIII** with Mn^0 closes the catalytic cycle by regenerating the active catalyst **I**. Whereas the radical manifold is consistent with the control experiments, the possible engagement of a 2-electron pathway relying on the direct addition of alkyl-Ni **IV** across the C-C double bond of **2**, cannot be ruled out.

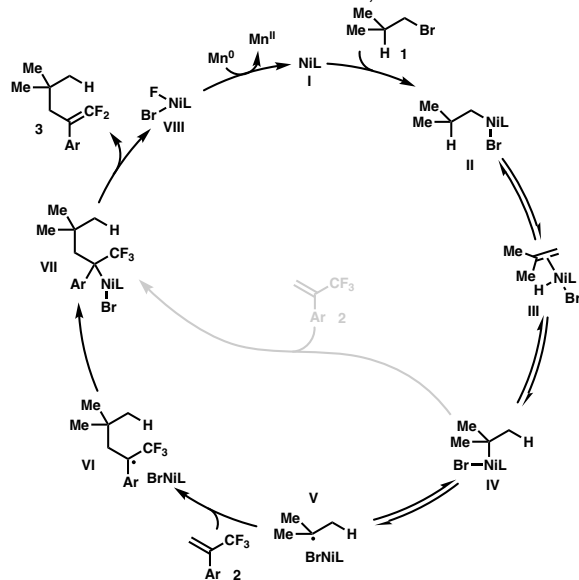


Figure 3. Plausible mechanism. Ni-catalyzed chain-walking process enables a C-radical translocation event, thus allowing migratory functionalization at the sterically more hindered position.

Discussion

In summary, a Ni-catalyzed reductive coupling for the synthesis of 1,1-difluoroalkenes containing quaternary centers is introduced. Difluoroalkenes are known bioisosteres for carbonyl groups in medicinal chemistry. Key to the success of this method is development of a Ni-H initiated remote functionalization of alkyl bromides. This approach enables a defluorinative 3,3-difluoroallylation of unactivated alkyl bromide substrates at sterically congested tertiary positions as an approach to selectively construct all-carbon quaternary centers. It is noteworthy that this transformation represents a rare case of C-radical transposition, which is enabled by a Ni-H chain-walking manifold. The successful

development of this protocol demonstrates that readily available primary and secondary alkyl bromides can be used as progenitors for the construction of quaternary carbon-containing frameworks. In view of the potential impact of this strategy for remote functionalization, efforts to develop additional transformations are underway in these laboratories.

Methods

General procedure for the 3,3-difluoroallylation of unactivated alkyl bromides. To an oven-dried Schlenk tube equipped with a magnetic stir bar was added $NiBr_2 \cdot glyme$ (3.1 mg, 0.01 mmol, 5.0 mol%), L2 (2.5 mg, 0.012 mmol, 6.0 mol%), Mn powder (16.5 mg, 0.3 mmol, 1.5 equiv). After the Schlenk tube was evacuated and filled with nitrogen for three cycles, THF (1.0 mL), compound **1** (0.6 mmol, 3.0 equiv) and compound **2** (0.2 mmol, 1.0 equiv) were added under nitrogen atmosphere. The Schlenk tube was maintained at 25 °C for 12 to 24 h. The reaction mixture was then diluted with ethyl acetate (10 mL) and washed with H_2O (10 mL). The aqueous layer was extracted with ethyl acetate (10 mL $\times$ 2). The combined organic layers were washed with water (10 mL), brine (10 mL) and dried over Na_2SO_4 . After solvent was removed under reduced pressure, the crude residue was analyzed by ^{19}F NMR with 1-iodo-4-(trifluoromethyl)benzene as internal standard, and then the mixture was purified by column chromatography or preparative TLC on silica gel to afford the desired product.

Data availability

The authors declare that all the data supporting the findings of this study are available within the paper and its Supplementary Information files, or from the corresponding authors upon request.

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Author contributions

C.Z. and C.F. conceived and directed the project. Z.-Y.L. performed the experiments. L.T., H.Z., and Y.-F.Z. prepared some of the substrates. C.Z. and C.F. analyzed the results and wrote the manuscript. P.J.W. discussed the chemistry, suggested experiments and revised the manuscript.

Competing interests

The authors declare no competing interests.