

The effects of substitutional Fe-doping on magnetism in MoS₂ and WS₂ monolayers

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Abstract

Doping of two-dimensional (2D) semiconductors has been intensively studied toward modulating their electrical, optical, and magnetic properties. While ferromagnetic 2D semiconductors hold promise for future spintronics and valleytronics, the origin of ferromagnetism in 2D materials remains unclear. Here, we show that substitutional Fe-doping of MoS₂ and WS₂ monolayers induce different magnetic properties. The Fe-doped monolayers are directly synthesized via chemical vapor deposition. In both cases, Fe substitutional doping is successfully achieved, as confirmed using scanning transmission electron microscopy. While both Fe:MoS₂ and Fe:WS₂ show PL quenching and n-type doping, Fe dopants in WS₂ monolayers are found to assume deep-level trap states, in contrast to the case of Fe:MoS₂, where the states are found to be shallow. Using μm - and mm-precision local NV⁻ magnetometry and superconducting quantum interference device, we discover that, unlike MoS₂ monolayers, WS₂ monolayers do not show a magnetic phase transition to ferromagnetism upon Fe-doping. The absence of ferromagnetism in Fe:WS₂ is corroborated using density functional theory calculations.

Supplementary material for this article is available [online](#)

Keywords: ferromagnetism, dilute magnetic semiconductor, substitutional doping, trap states, chemical vapor deposition, n-type doping

(Some figures may appear in colour only in the online journal)

1. Introduction

Spin-based technologies have emerged as a viable path to overcome the increasing challenges of conventional CMOS transistor scaling. The materials for spintronics are expected to show ferromagnetism in semiconductors, air-stability, and a Curie temperature at or above room temperature (RT). The first generation of such materials has been realized by doping transition metals in conventional semiconductors known as a dilute magnetic semiconductor (DMS). Since the observation of ferromagnetism in p-type (In,Mn)As and (Ga,Mn)As at cryogenic temperature, realizing RT ferromagnetism in semiconductors has been a pursuit for a few decades to realize practical applications [1, 2]. While the RT ferromagnetism has been hypothesized in nitrides and oxides, such as (Ga,Mn)N and (Zn,Mn)O [3, 4], it has neither been confirmed broadly nor demonstrated for a device structure [5]. The highest recorded Curie temperature in bulk DMS crystals is still widely considered as 110 K in (Ga,Mn)As [6], showing the limitations of conventional DMS materials for practical spintronics device applications.

Recent excitement in the field of two-dimensional (2D) materials has demonstrated ferromagnetism in atomically thin layers of chromium-based and iron-based alloys such as chromium triiodide (CrI₃) [7], chromium tribromide (CrBr₃) [8], chromium germanium telluride (Cr₂Ge₂Te₆) [9], and iron germanium telluride (Fe₃GeTe₂) [10]. While these findings indicate the possibility of realizing magnetism in a 2D limit, those 2D ferromagnets remain either metallic or insulating [7, 11, 12]. Several theoretical studies have predicted that DMS based on transition metal dichalcogenide (TMD) monolayers would exhibit ferromagnetic behaviors even at RT, which is a critical requirement for practical applications. Fan *et al* used first-principles calculations to define the magnetic properties of MoS₂, showing that doping of V, Mn, Fe, Co, and Cu into MoS₂ monolayers with low concentration could yield DMSs [13]. Mishra *et al* reported long-range ferromagnetism in Mn-doped MoS₂, MoSe₂, MoTe₂, and WS₂ monolayers using density functional theory (DFT) [14]. Lin *et al* used first-principles calculations to show the magnetic properties in Fe, Co, and Mn-doped MoS₂ monolayers [15]. Experiential observations of magnetism have been made in bulk TMDs [16] or transition metal-doped few-layer TMDs, using dopants, including V [17], Mn [18], Co [19–21], Ni [21, 22], Cu [23], Nb [20] and Re [24]. RT ferromagnetism was shown in 5% Cu-doped MoS₂ nanosheets [23], 10% Co-doped MoS₂ nanocrystals [21] and 10% Ni-doped nanocrystals [21]. Recently, the RT ferromagnetism was successfully demonstrated in *in situ* Fe-doped MoS₂ (Fe:MoS₂) and V-doped WSe₂ (V:WSe₂) via chemical vapor deposition [25, 26]. However, the origin of ferromagnetism in such materials remains uncertain.

Here, we report the effects of Fe-doping on the magnetism in WS₂ monolayers by studying their optical, electrical, and magnetic properties of *in situ* synthesized Fe:WS₂ monolayers, and comparing with the case of Fe:MoS₂. We confirm the substitutional doping of Fe atoms into transition metal sites using the high-resolution scanning transmission

electron microscopy (STEM). We show that Fe dopants produce two different trap states in Fe:MoS₂ and Fe:WS₂ by studying their photoluminescence signatures. Employing nitrogen-vacancy (NV[−]) center magnetometry, and magnetization measurements using a superconducting quantum interference device (SQUID), we find that ferromagnetism appears only in Fe:MoS₂ monolayers, while Fe:WS₂ monolayers show a paramagnetic response. Using DFT calculations, we show that the magnetic properties in Fe:MoS₂ and Fe:WS₂ monolayers arise from the charge transfer between the Fe dopants and neighboring S atoms, and the magnetic moments of paired Fe atoms vanish only in Fe:WS₂.

2. Methods

2.1. Synthesis of Fe:WS₂ and Fe:MoS₂

The synthesis procedures of Fe:WS₂ and Fe:MoS₂ monolayers are shown in figure 1. Thin-film of transition metal oxide (WO₃ and MoO₃, Kurt J. Lesker, 99.99%) (5 nm thick) was deposited on a SiO₂/Si substrate prepared as the transition metal source. Iron oxide (Fe₃O₄, Alfa Chemicals) and sulfur powder (Alfa Aesar, 99.5%) were used for Fe and S sources, respectively. For the growth of Fe:WS₂ monolayers, a Fe₃O₄ powder (0.01 g) was sandwiched between a bare SiO₂/Si substrate and the WO₃-deposited SiO₂/Si substrate. The prepared substrate was loaded in the middle of the furnace tube. A sulfur powder (0.6 g) was loaded upstream. After loading the substrates and precursors, Ar gas (Praxair, UHP) was supplied while the mechanical pump was running to purge ambient air. The base and growth pressures were 150 mTorr and 3 Torr, respectively. Then the furnace temperature was increased up to 900 °C, which was held for 5 min. After completing the growth, the furnace lid was opened to cool down the sample to RT before the samples were taken out. The growth recipe of Fe:MoS₂ monolayers was similar to that of Fe:WS₂ monolayers, except for using MoO₃ instead of WO₃. The growth process of undoped WS₂ monolayers was also similar to that of the growth process of Fe:WS₂ monolayers, except for its growth without Fe₃O₄.

2.2. Characterization of Fe:MoS₂ and Fe:WS₂

STEM was carried out on an FEI Titan³ G2 S/TEM operated at 80 kV to investigate the atomic structure of WS₂ and MoS₂ triangles. A high-angle annular dark-field (HAADF) detector was used for Z contrast imaging. To reduce noise and increase the visibility of atoms, a Gaussian Blur filter with a blurring width of 0.03 nm was applied with an image processing program called ImageJ.

Raman and RT PL spectroscopies were performed using a Horiba XPLORE spectrometer equipped with a 532 nm laser at the atmospheric pressure. Low-temperature micro-photoluminescence (μ -PL) measurements were taken inside a closed-cycle cryostat (attoDRY1100) at 4 K base temperature. For a laser excitation, we utilized nonresonant laser pumping at 532 nm in continuous wave mode. A laser spot size of

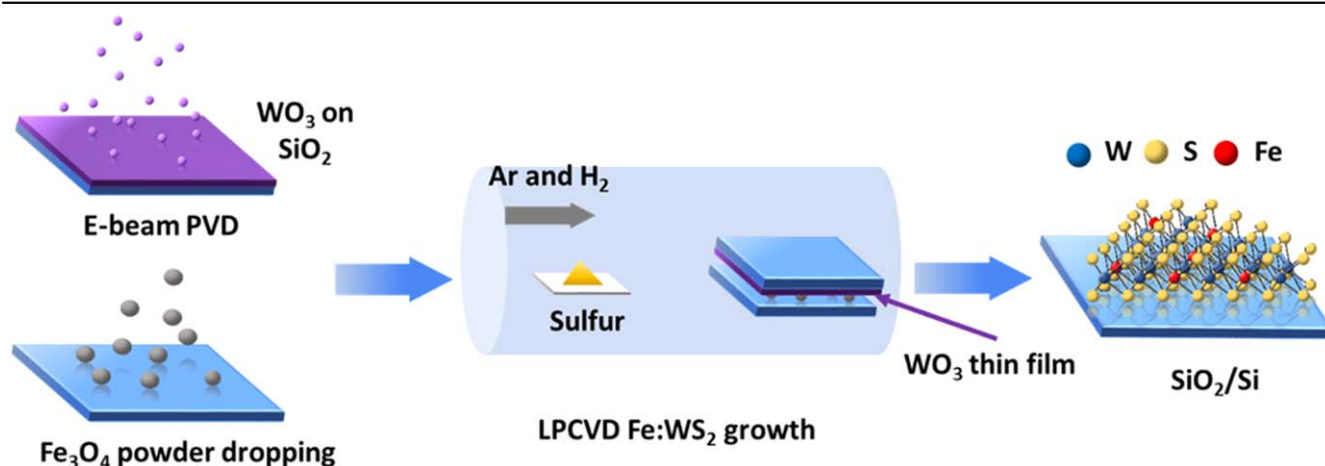


Figure 1. Schematic diagram of Fe:WS₂ growth process.

approximately 0.85 microns was achieved using a cryogenic microscope objective lens with a numerical aperture of 0.82. The reflected PL signal was collected by the same objective and guided by a multi-mode fiber to a spectrometer with an attached liquid-nitrogen-cooled silicon charge-coupled device camera.

The XPS measurement was performed using a PHI VersaProbe III x-ray photoelectron spectroscopy (XPS) equipped with a monochromatic Al K α source ($\lambda = 1486.7$ eV). For the calibration, the binding energy of the C1s peak was fitted to 284.8 eV. The spectral analysis software Multipeak was employed for XPS peak deconvolution, where Voigt line shape and an active Shirley background were used for peak fitting.

Magnetic measurements were performed using a Superconducting Quantum interference Device Magnetometer (Quantum Design Model 2000) with the magnetic field applied parallel to the substrate plane. The magnetic field range was scanned from -2 to 2 T, while the temperature is maintained at 300 K. The large magnetic field range allowed the subtraction of the diamagnetic background signal of the substrate. Moreover, the temperature dependence measurement was performed from 300 to 400 K, while a magnetic field of 0.1 T was applied.

The PL emission of NV⁻ center depends on the spin state of its ground state electron before excitation and is 20%–30% higher for $m_s = 0$ ('bright') sublevel than for the $m_s = \pm 1$ ('dark') sublevels [27]. This is because after excitation, decay through the intersystem crossing from the excited $m_s = \pm 1$ states make the relaxation partially non-radiative, decreasing the emitted fluorescence. Thus, the electron spin resonance (ESR) between the $m_s = 0$ and $m_s = \pm 1$ sublevels of the NV⁻ center's ground state can be detected under optical excitation at 532 nm and a simultaneous microwave (MW) radiation at ~ 2.87 GHz polarizing these spin states. This resonance appears as a reduction of the detected fluorescence when a coupling (spin flipping) occurs between the bright state and the dark states by scanning the microwave frequency across the $m_s = 0 \rightarrow m_s = \pm 1$ transitions. Our NV⁻ magnetometry experiments were conducted in a confocal microscopy setup. The samples were illuminated on the

surface using a frequency-doubled Nd:YAG laser (532 nm, $300 \mu\text{W}$) focused with a NA = 0.8 objective giving a ~ 425 nm diffraction-limited spot size. Simultaneously, a sequence of 100 ms alternative on/off MW pulses (30 dBm before the antenna) was applied using a loop antenna centered about the focus, and their frequency was scanned across the resonance to polarize the electron spin of NV⁻ centers. The PL signal of the NV⁻ centers was then collected by the same objective and directed through a 550 nm long-pass filter into an avalanche photodiode. The relative PL is the ratio of total PL counts with and without MW radiation, where the resonance peaks appear at the frequencies of minimum ratio.

Projector augmented wave [28, 29] spin-polarized DFT [30, 31] calculations were carried out with the Vienna *ab initio* simulation package [32]. For this, the Perdew–Burke–Ernserhof generalized gradient approximation for the exchange-correlation functional [33] was used with a basis set including plane waves with energies less than or equal to 600 eV. The Brillouin zone of the pristine WS₂ unit cell was sampled with a $6 \times 6 \times 1$ Γ -centered Monkhorst-Pack grid [34]. Ionic relaxation iterations persisted until the Hellmann–Feynman forces on all atoms settled below 1 meV per Å, while electron field optimization iterations continued until changes in both the total energy and Kohn–Sham eigenvalues fell below 10^{-7} eV. For all structures, 20 Å of vacuum was inserted in the z -direction (out-of-plane direction) to negate interactions of each system with its periodic images. Larger Fe:WS₂ systems were simulated by substituting W atoms with Fe in a 5×5 supercell, and were sampled with a single k -point at Γ . To investigate whether or not these systems exhibit ferromagnetism, spin-polarized DFT calculations for each system were initialized with each atom possessing a magnetic moment of zero.

3. Results

3.1. Synthesis and characterizations of Fe:MoS₂ and Fe:WS₂ monolayers

Fe-doping of monolayer WS₂ (MoS₂) was realized by growing WS₂ (MoS₂) with Fe₃O₄ via LPCVD contact-growth

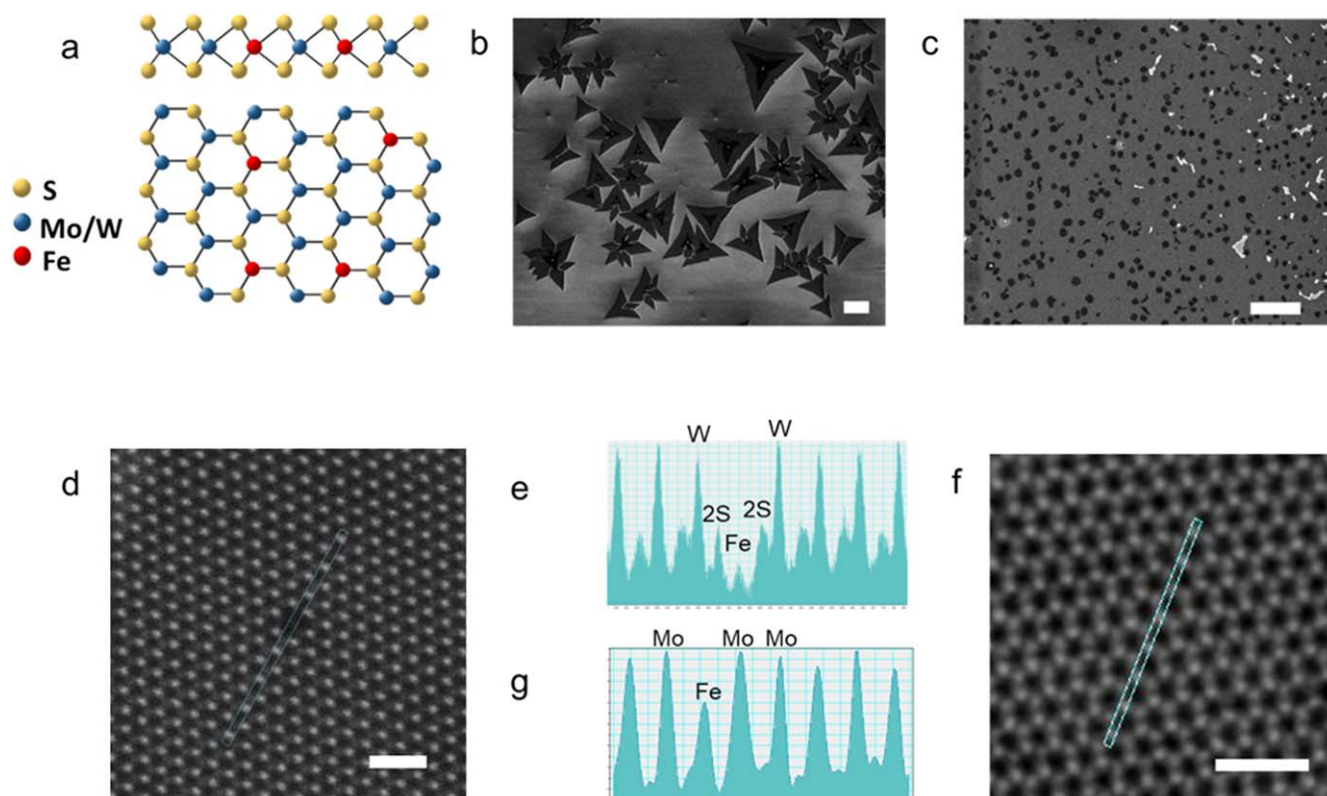


Figure 2. Schematic diagram, and SEM, and TEM images of Fe-doped MoS₂ and WS₂ monolayers. (a) Schematic of Fe:MoS₂ or Fe:WS₂ monolayers, (b) and (c) are SEM images of Fe:WS₂ and Fe:MoS₂. The scale bar is 5 μ m. (d) and (f) are HAADF-STEM images of Fe:WS₂ and Fe:MoS₂ monolayers, respectively. The scale bar is 1 nm. (e) and (g) are STEM intensity profiles of Fe:WS₂ and Fe:MoS₂, respectively.

(see Methods section) [35]. Figure 2(a) depicts schematic top and side views of the atomic structures of Fe:WS₂ monolayers, where Fe atoms replace transition metal sites in the host lattice. Figures 2(b) and (c) show the scanning electron microscope (SEM) image, where the dark area represents a bilayer region and the bright spot is a Fe₃O₄ particle.

Figure 2(d) shows a HAADF-STEM image of a Fe:WS₂ monolayer. The brightness of atoms in STEM image is directly proportional to the square of atomic number ($\sim Z^2$), providing evidence to classify the atoms in 2D materials. The STEM intensity profile (figure 2(e)) clearly distinguishes atoms of W ($Z = 74$), 2S ($Z = 32$), and Fe ($Z = 26$). According to the intensity profile, the position of the Fe peak coincides with a W site, confirming the replacement of the transition metal (W) sites. Figures 2(f) and (g) show the STEM image and intensity profile of a Fe:MoS₂ monolayer confirming the substitution of Fe atoms in Mo sites. XPS was used to estimate the concentration of Fe atoms in Fe:WS₂ and Fe:MoS₂ monolayers, as shown in figures S11(a) and (b) (available online at stacks.iop.org/NANO/32/095708/mmedia), which represent the XPS spectra of undoped/Fe-doped WS₂ monolayers and undoped/Fe-doped MoS₂ monolayers, respectively. The deconvolution of the spectra from Fe-doped samples yielded two peaks arising from Fe–O (blue) and Fe–S bonds (green) compared to the undoped cases. While the Fe–O peak is from unreacted Fe₃O₄ (powder) used as Fe source during the Fe-doping process, the Fe–S peak is correlated to substituted Fe atoms at transition metal sites of the monolayers

[36]. Using the Fe–S peak intensity, the Fe concentration was estimated to be approximately 0.6% for both Fe-doped samples. The AFM image of Fe:WS₂ (figure SI2) shows that the doped sample is a monolayer.

3.2. Optical and Raman characterization

Figure 3(a) shows the PL spectra of Fe:MoS₂ (red curve) and undoped MoS₂ monolayers (black curve). The doped MoS₂ monolayers exhibit PL quenching by 35% and a redshift of 29 ± 0.5 meV when compared to the undoped case. This PL quenching arises from non-radiative channels activated by shallow traps within the MoS₂ bandgap when Fe dopants are introduced [37, 38]. Given that the charges are highly mobile and have access to a large amount of material in shallow trap states, it is expected that the MoS₂ sees more electron concentration when Fe dopants are introduced. Since Fe acts as an n-type dopant, the electron concentration in Fe:MoS₂ monolayers is expected to increase upon Fe-doping. Consequently, neutral excitons would be reduced along with an increase in negative trion emissions, causing a redshift in the PL peak position, which is consistent with previous studies [39, 40]. Figure 3(b) shows the PL spectra of Fe:WS₂ (red curve) and undoped WS₂ monolayers (black curve). Interestingly, we find that Fe:WS₂ shows the blue shift of the PL peak position by 13 ± 0.5 meV (and 40% quenching) compared to the undoped case. This blue-shift can be attributed to Fe dopants in WS₂ monolayers, which produce deep-level defects (rather than shallow trap states) [41]. Given that deep

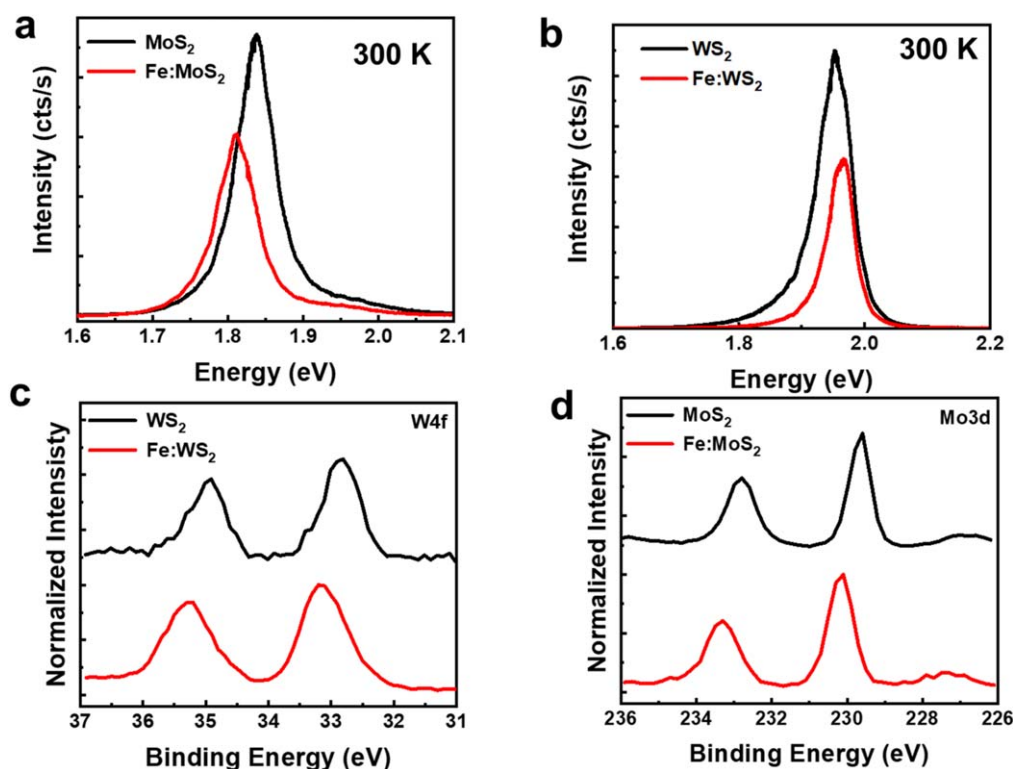


Figure 3. PL and XPS data of Fe-doped and undoped MoS₂ and WS₂ monolayers. (a) PL spectra of Fe:MoS₂ and MoS₂ monolayers at 300 K. The Fe:MoS₂ peak is red-shifted. (b) PL spectra of Fe:WS₂ and WS₂ monolayers at 300 K. The Fe:WS₂ peak is slightly blue-shifted. (c) XPS spectra of Fe:WS₂ and WS₂ monolayers. The blue-shift of binding energies of W in Fe:MoS₂ indicates that the Fe is an n-type dopant. (D) XPS spectra of Fe:MoS₂ and MoS₂ monolayers. The blue-shift of W peaks in Fe:MoS₂ also shows the Fe is an n-type dopant.

trap states are deeply localized charge states in the bandgap of the material, and the charges are immobile in them, the charge concentrations are only available within a small region of the material, resulting in a decrease in the mobile electron concentration. Therefore, the negative trion intensity would be decreased, causing a blue-shifted in the PL peak position. Figures SI3(a) and (b) show the PL spectra at 4 K, where the A-exciton and the sulfur vacancy-induced emission become resolvable upon doping in MoS₂ with the intensity of the former dominating the intensity of the latter, unlike the case of Fe:WS₂. In addition, we find that while no doping-related linewidth broadening appears in MoS₂ case, WS₂ shows drastic PL broadening from 72 meV (WS₂) to 92 meV (Fe:WS₂) monolayers, providing another evidence that Fe:MoS₂ monolayers possess different dopant related trap states than Fe:WS₂ [42].

It is generally recognized that in semiconductors, Raman modes are sensitive to a perturbation such as doping due to the changes in material's electronic band structure caused by doping [27]. Substitutional doping of TMDs affects the A₁ vibrational mode (the out-of-plane vibration of chalcogen atoms) due to the change in electron density, consequently altering the strength of electron-phonon couplings in TMDs [43]. Since changes in electron density of shallow and deep trap state are different, we examined the changes in phonon frequency of both undoped and Fe-doped WS₂ and MoS₂ monolayers using Raman spectroscopy. Figure SI4(a) shows the Raman spectra of both Fe:MoS₂ (red curve) and pristine

MoS₂ (black curve) monolayers. The A₁ mode of the Fe:MoS₂ monolayers shifts to higher wavenumbers compared to the undoped monolayers, indicating that the A₁ mode is sensitive to doping. The redshift of A₁ mode is attributed to the increased electron density of the layer, supporting that the Fe impurity is an n-type dopant, which is consistent with an earlier study describing Raman spectra of n-type (Rhenium) Re:MoS₂ [24]. Substituting Fe dopants into TMDs causes A₁ mode linewidth broadening from 5.61 ± 0.3 to 6.82 ± 0.2 cm⁻¹, commensurate with recent studies of Fe:MoS₂ monolayers [25]. Figure SI4(b) shows the Raman spectra of Fe:WS₂ and undoped WS₂ monolayers, where no discernible A₁ mode peak shift and broadening were observed. However, we find an additional Raman vibrational mode at ~ 383 cm⁻¹ in the Fe:WS₂ monolayers (figure SI4(c)), which is attributed to the substituted Fe atoms (i.e. FeS₂). The emergence of an additional Raman vibrational mode at 383 cm⁻¹ with no apparent changes in the A₁ vibrational mode in Fe:WS₂ monolayers gives another indication that Fe:WS₂ monolayers have different Fe dopants-related trap states from Fe:MoS₂ monolayers.

3.3. Binding energy

Since the Fermi level can shift due to the concentration of charge and type of dopant carriers in doped semiconductors, an analysis of the Fermi level shift of TMDs can be used to investigate the change in electron density concentrations of

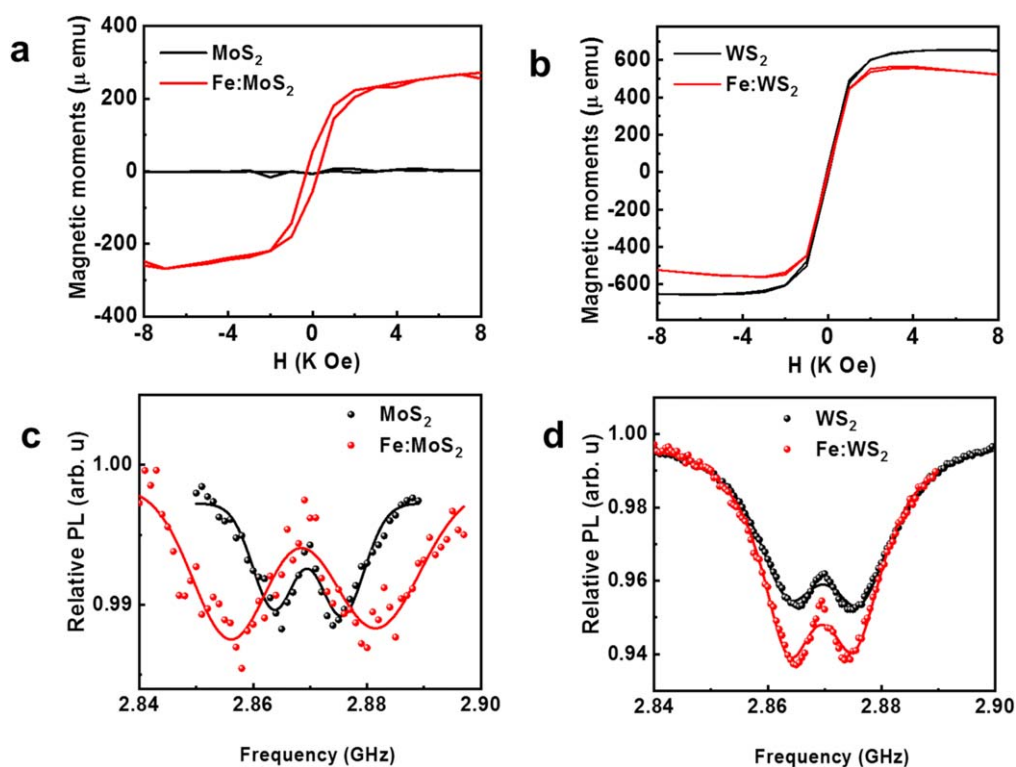


Figure 4. Magnetic characteristics of Fe-doped and undoped MoS_2 and WS_2 monolayers. (a)–(b) Magnetization M – H curves of doped and undoped MoS_2 and WS_2 monolayers, respectively. A pronounced hysteresis loop is observed in MoS_2 monolayers upon Fe-doping, suggesting a magnetic phase transition to ferromagnetism. (c)–(d) Comparison of ODMR spectra of NV^- centers coated on Fe-doped and undoped MoS_2 and WS_2 monolayers, respectively. The resonance dips for the case of Fe:MoS_2 display apparent Zeeman shifts revealing the local magnetic field. No discernible shifts were observed for the case of Fe:WS_2 . All data were recorded at RT.

TMDs upon Fe substitutional doping. Figure 3(c) shows the XPS spectra of W in the Fe:WS_2 monolayers (red curve). For comparison, we also measured the XPS spectra of undoped WS_2 monolayers (black curve). The binding energy of $\text{W}4f_{5/2}$ (at 35.2 eV) and $\text{W}4f_{7/2}$ (at 33.2 eV) from the Fe:WS_2 monolayers are ~ 0.25 eV higher than those from undoped WS_2 monolayers. This result implies that the Fermi energy of Fe:WS_2 monolayers is higher than that of the undoped ones, indicating that Fe is an n-type dopant, which is consistent with an earlier study [44]. Similarly, figure 3(d) shows the XPS spectra of Mo in the Fe:MoS_2 (red line) and undoped MoS_2 monolayers (black line), respectively. The binding energies of $\text{Mo}3d_{3/2}$ (233.3 eV) and $\text{Mo}3d_{5/2}$ (230.1 eV) in the Fe:MoS_2 is ~ 0.5 eV higher than those of undoped MoS_2 monolayers. Since, in shallow trap states, more electrons are available than deep trap states, this result explains the twice as much blue-shifted binding energy of Fe:MoS_2 compared to that of Fe:WS_2 as more electrons are available in Fe:MoS_2 than Fe:WS_2 monolayers.

3.4. Magnetic characterization

To study the impact of Fe dopants on the magnetic properties of WS_2 and MoS_2 , we measured the magnetization M – H curves of undoped and Fe-doped monolayers via a SQUID magnetometer. Figure 4(a) shows that Fe:MoS_2 monolayers exhibit a magnetic phase transition to ferromagnetism, as evidenced by observing a M – H hysteresis loop. It is worth

mentioning that the magnetization M – H curves are obtained after subtraction of a well-known linear diamagnetic background, which is common in the TMDs semiconductors [16]. One should note that the background affects only the y-axis and does not produce a shift on the x-axis. Figure 4(b) shows magnetization data of WS_2 and Fe:WS_2 monolayers, which does not show a measurable magnetic phase transition. The absence of ferromagnetism in Fe:WS_2 may arise from inadequate concentrations of Fe dopants or the presence of paired Fe atoms with the zero-net magnetic moments in Fe:WS_2 .

To further investigate the magnetic features of our Fe-doped monolayers, we also performed spatially resolved magnetometry using NV^- centers in nanodiamonds. Given that the electron spin energy levels of NV^- centers are ultrasensitive to local magnetic fields induced by ferromagnet atoms, performing magnetometry based on the optically detected magnetic resonance (ODMR) in these spin states can give insight into local magnetic properties of our Fe-doped monolayers. The defect (color) center in diamond is generally recognized as a promising platform for various applications in quantum sensing [45], nanoscale metrology [46, 47], and quantum information processing [48]. While the ODMR-based magnetometry is extendible to cryogenic temperatures by conducting the experiments in a cryostat, we performed our measurements at RT only. The ESR signal of NV^- centers can split into a pair of dips when the degeneracy of the $m_s = \pm 1$ sublevels is lifted by an external magnetic field (see Methods section). We analyzed the net Zeeman shifts in the ESR dips of NV^- centers in

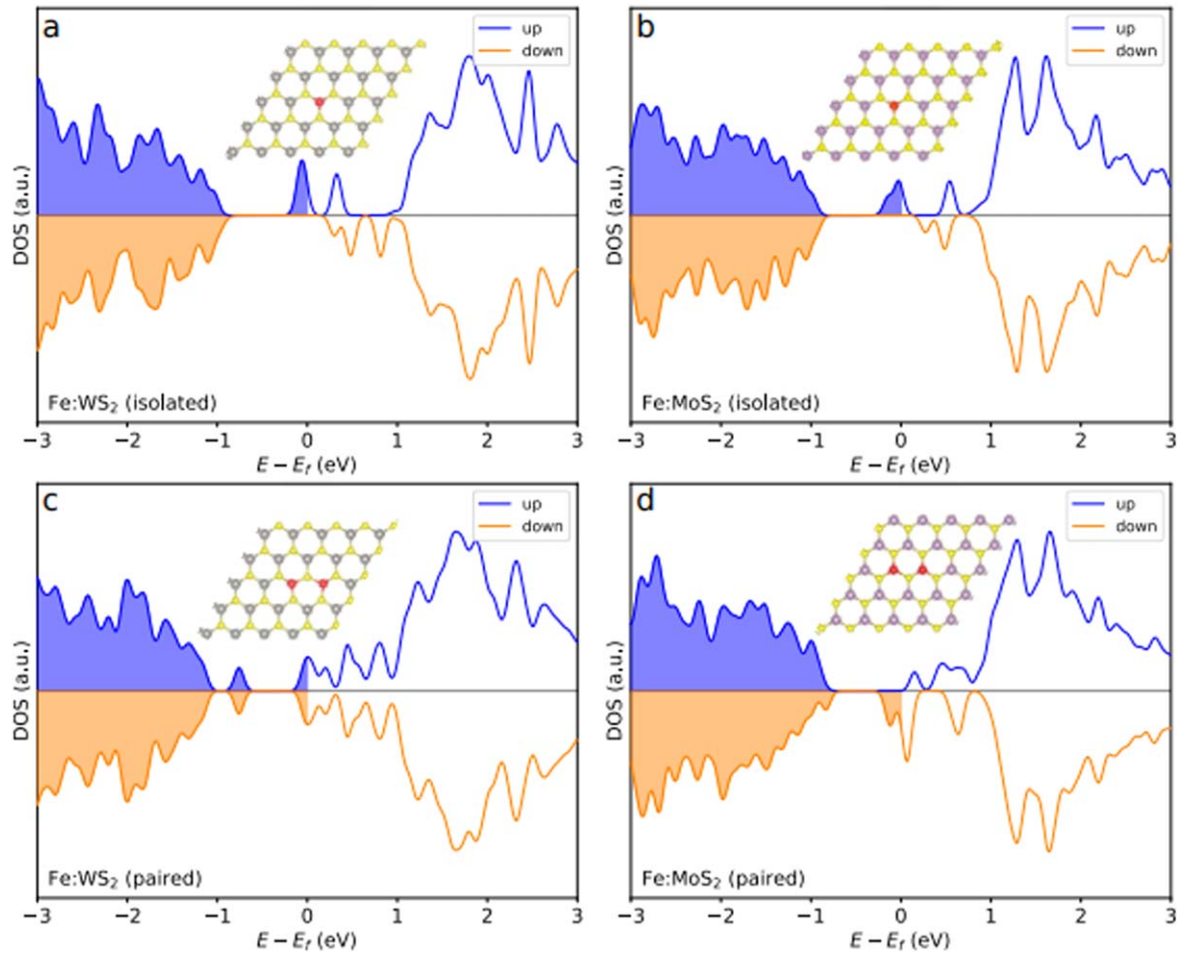


Figure 5. DFT calculations of Fe:MoS₂ and Fe:WS₂ monolayers. Spin-polarized density of states for isolated Fe dopants in (a) Fe:WS₂ and (b) Fe:MoS₂ and paired Fe dopants in (c) Fe:WS₂ and (d) Fe:MoS₂. The filled regions denote occupied states.

a spin-coat proximity [49] of our materials to measure any local magnetic field induced by the Fe dopants. Figures 4(c) and (d) show a comparison of exemplary ODMR signals of NV⁻ centers in the vicinity of Fe-doped and undoped MoS₂ and WS₂ monolayers. The ~ 10 MHz energy splitting between the ESR dips for the undoped cases in both panels shows a readily broken degeneracy of $m_s = \pm 1$ sublevels due to a pseudo-magnetic field caused by non-axial strain in the nanodiamond crystals. We subtracted this splitting from the total to find out the net Zeeman shifts due to any local magnetic field induced by the Fe dopants in the monolayers. Figure 4(c) shows an apparent net energy splitting for the case of Fe-doped MoS₂ compared to the undoped area of the same sample. We performed a statistical analysis (not shown) on the net shifts at 4×5 spots of a $20 \mu\text{m} \times 25 \mu\text{m}$ grid on each area which resulted in a mean net splitting (~ 11 MHz) only at the Fe:MoS₂ area. According to the Zeeman splitting term in the NV⁻ center's spin Hamiltonian [46] $\Delta E = g_e \mu_B \mathbf{B} \cdot \hat{\mathbf{S}}$, with g_e and μ_B being the Landé g -factor and Bohr magneton respectively, this energy shift yields an average projection of a 0.5 ± 0.1 mT local magnetic field on the NV⁻ centers' spin. This reveals an apparent preserved magnetization induced on this sample upon Fe-doping and is an indication of RT ferromagnetism of Fe:MoS₂ monolayers. In contrast, performing

similar experimental procedure on Fe-doped and undoped areas of WS₂ showed no discernible shifts in the ESR dips (see figure 4(d) for an illustrative comparison). The fact that Fe-doped area of WS₂ shows no local magnetic strength long after loading indicates that this monolayer does not possess a preserved magnetization. In order to examine the ability of Fe:WS₂ to maintain magnetization in short-term, we performed ODMR measurements immediately after applying a short (100 ms) pulse of magnetic field (100 mT) at each step of the MW frequency scan. The results still showed no noticeable shifts in the ESR dips, ruling out the ferromagnetism of Fe:WS₂ and suggesting a zero net magnetic moment caused by an equal number of opposite Fe electron spins.

3.5. DFT modeling

To understand why ferromagnetism is not observed in Fe:WS₂, spin-polarized DFT was used to simulate WS₂ with substitutional Fe dopants. These calculations can predict ferromagnetism in the material if the ground state occupation of one spin projection is greater than the other. First, isolated Fe dopants were simulated by replacing a single W atom with Fe in a 5×5 WS₂ supercell. The resulting density of spin states, shown in figure 5(a), depicts that there are more

Table 1. Charge transfers to and magnetic moments of Fe atoms in Fe:WS₂ and Fe:MoS₂. The paired row shows the charge transfers to and magnetic moments of both Fe atoms in their respective systems.

	Charge-transfer (e^-)		Magnetic moment (μ_B)	
	Fe:WS ₂	Fe:MoS ₂	Fe:WS ₂	Fe:MoS ₂
Isolated	0.733	0.716	1.179	1.176
Paired	0.667, 0.668	0.713, 0.797	0.000, 0.000	1.165, 2.071

spin-up than spin-down states below the Fermi-level, signifying that the system is ferromagnetic. However, when Fe dopant atoms are paired, so that two adjacent W sites are replaced with Fe, the density of spin-up and spin-down states are identical (figure 5(c)), suggesting that Fe:WS₂ is paramagnetic when Fe atoms occupy adjacent sites. This behavior differs from that of Fe:MoS₂, which exhibits ferromagnetic behavior for both isolated and paired Fe dopant configurations, as shown in figures 5(b) and (d). Paired Fe atoms induce more structural distortion in Fe:WS₂ than in Fe:MoS₂. This distortion pulls the Fe atoms apart, giving them more room to accumulate electric charge (see figure SI5). This could explain why more charge is transferred away from paired Fe atoms in Fe:WS₂ than in Fe:MoS₂. Meanwhile, isolated Fe atoms do not distort the TMD structures, so the Fe atoms host similar amounts of charge in both TMDs (figure SI5).

We furthermore investigated the origin of the difference in these magnetic behaviors by examining the amount of charge transfer between the Fe dopants and neighboring S atoms. As S is more electronegative than Fe, S is expected to pull negative charge from the valence states of neighboring Fe. Doing so can unpair a d-orbital electron bound to a Fe atom, granting it a net magnetic moment. We used a Bader charge analysis [50–52] to investigate the degree of this charge transfer in all Fe-doped systems considered. Table 1 shows that the charge transfer to isolated Fe is very similar in Fe:WS₂ and Fe:MoS₂. Correspondingly, the isolated Fe atoms have a similar nonzero magnetic moment. However, when Fe is paired, the charge transfer to the Fe atoms in Fe:WS₂ is notably smaller than those in Fe:MoS₂. This phenomenon could explain why the Fe atoms in Fe:WS₂ lack a magnetic moment. Furthermore, the charge transfer to paired Fe atoms in Fe:MoS₂ is not equal, and the Fe atom that has gained more charge also nets a larger magnetic moment.

Recalculating the magnetic moments using LDA + U yielded results similar to those of PBE (see table SI1). The most notable difference is that LDA + U predicts that paired Fe atoms have small but nonzero magnetic moments in Fe:WS₂, while PBE predicts that their magnetic moments vanish. However, these magnetic moments are sensitive to the difference between the Hubbard term U and the exchange term J [53], and determining the appropriate values of U and J is beyond the scope of this work. Here, we used $U = 1$ and $J = 1$. Regardless, both PBE and LDA + U predict that the magnetic moments of paired Fe atoms are weaker in Fe:WS₂

than in Fe:MoS₂, which is consistent with our experimental findings (table SI1).

4. Conclusions

We have studied the effects of Fe-doping on the magnetic and optical properties of *in situ* synthesized Fe:WS₂ and Fe:MoS₂ monolayers. The presence of Fe atoms in transition metal sites (Mo or W) was confirmed using the HAADF-STEM intensity profile. While both Fe:MoS₂ and Fe:WS₂ were found to be n-type materials, a new Raman peak around 383 cm⁻¹ was found only in Fe:WS₂ monolayers, which is another evidence of Fe-doping. The PL spectra of Fe:MoS₂ (Fe:WS₂) is red-shifted (blue-shifted), while they are all quenched. Importantly, The SQUID result indicates that Fe:WS₂ monolayers are paramagnetic, while Fe:MoS₂ monolayers remain ferromagnetic at RT. The ODMR-based magnetometer with NV-center confirmed that Fe:MoS₂ is ferromagnetic while Fe:WS₂ is paramagnetic. The ODMR spectrum showed the energy splitting between the two typical ESR dips of the NV centers to be ~21 MHz in the vicinity of Fe:MoS₂ monolayers. This splitting was significantly larger than the splitting for undoped MoS₂, demonstrating the presence of a local magnetic field in Fe:MoS₂ monolayers. However, no enhancement in the energy splitting between two ESR dips of Fe:WS₂ and WS₂ monolayers was observed, indicating that there was no long-range order ferromagnetic states in Fe:WS₂. The DFT calculations were consistent with the experimental observations. The results from DFT calculation considering PBE approximation indicate that the observed properties are likely to arise from charge transfer between the Fe dopants and neighboring S atoms, depending on the structure of the doped system. While isolated Fe atoms induce similar ferromagnetic behaviors in both MoS₂ and Fe:WS₂, the magnetic moments of paired Fe atoms vanish only in Fe:WS₂, which explains different magnetic properties between Fe:WS₂ (paramagnetic) and Fe:MoS₂ (ferromagnetic). Furthermore, LDA + U considered DFT calculation showed weaker magnetic moments of paired Fe atoms in Fe:WS₂ than in Fe:MoS₂. Given that the magnetic moments are depending on the difference between the Hubbard term and the exchange term, all the experimental and theoretical results are consistent. These contrasting magnetic behaviors caused by paired Fe atoms can be attributed to paired Fe atoms generating more accumulated electric charge in Fe:WS₂ by pulling the Fe atoms apart. These findings allow us to control the material's magnetic properties layer-by-layer and can be applied for spintronics.

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