



Phosphoserine inhibits neighboring arginine methylation in the RKS motif of histone H3

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ABSTRACT

The effects of phosphorylation of histone H3 at serine 10 have been studied in the context of other post-translational modifications such as lysine methylation. We set out to investigate the impact of phosphoserine-10 on arginine-8 methylation. We performed methylation reactions using peptides based on histone H3 that contain a phosphorylated serine and compared the extent of arginine methylation with unmodified peptides. Results obtained via fluorography indicate that peptides containing a phosphorylated serine-10 inhibit deposition of methyl groups to arginine-8 residues. To further explore the effects of phosphoserine on neighboring arginine residues, we physically characterized the non-covalent interactions between histone H3 phosphoserine-10 and arginine-8 using ³¹P NMR spectroscopy. A salt bridge was detected between the negatively charged phosphoserine-10 and the positively charged unmodified arginine-8 residue. This salt bridge was not detected when arginine-8 was symmetrically dimethylated. Finally, molecular simulations not only confirm the presence of a salt bridge but also identify a subset of electrostatic interactions present when arginine is replaced with alanine. Taken together, our work suggests that the negatively charged phosphoserine maximizes its interactions. By limiting its exposure and creating new contacts with neighboring residues, it will inhibit deposition of neighboring methyl groups, not through steric hindrance, but by forming intrapeptide interactions that may mask substrate recognition. Our work provides a mechanistic framework for understanding the role of phosphoserine on nearby amino acid residues and arginine methylation.

1. Introduction

Histone proteins organize DNA, making it accessible in part through modifications on their N-terminal tails [1]. Histone posttranslational modifications (PTMs) can alter protein-protein interactions and are heavily implicated in gene expression [2]. Although single modifications on the N-terminus of histone proteins are established independent epigenetic marks such as the association of acetylation with gene activation [3], information about how these modifications affect each other is now emerging. Multiple studies on the combinatorial crosstalk of histone proteins have revealed synergistic and/or antagonistic effects.

The synergistic coupling of histone H3 phosphorylated serine-10 (H3S10p) and acetylated lysine-14 (H3K14ac) is a canonical example of one PTM enhancing the deposition of another [4–6].

Protein phosphorylation introduces a negatively charged phosphate group onto a serine/threonine/tyrosine residue, and this significantly changes the interactions with other amino acid residues, causing a change in the conformation of this substrate. Its contributions to neighboring PTMs is unclear. For example, it is not clear whether a negatively charged phosphate will provoke a major structural change and mask substrate recognition by the enzymes that deposit neighboring PTMs, or whether it will alter the local hydrogenbond network enough

Abbreviations: H3, histone H3; SDMA, symmetric dimethylarginine; ADMA, asymmetric dimethylarginine; MMA, monomethylated arginine; TLC, thin layer chromatography; REMD, replica exchange MD; PRMT1, protein arginine methyltransferase 1; PRMT5, protein arginine methyltransferase 5; NMR, nuclear magnetic resonance.

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to physically block deposition of neighboring groups. At the molecular level, several investigations have reported that phosphorylation in general reorganizes a salt-bridge network to bring about a localized conformational change and thus effects an exchange of signaling partners [7]. Of these, the phosphoserine arginine contact has been identified as a salient type of salt bridge interaction [8]. Histone H3 harbors an arginine-lysine-serine (RKS) motif that can become phosphorylated at serine-10 [9], and is symmetrically dimethylated at arginine-8 [10,11].

To understand the role of phosphoserine-10 on arginine-8 methylation, we performed *in vitro* methylation reactions using synthetic phosphoserine-10 peptides based on the N-terminal tail of histone H3. Arginine methylation of H3R8 was detected through fluorography and thin layer chromatography. We show that a pre-existing phosphate group on serine 10 (S10p) will inhibit both asymmetric and symmetric dimethylated arginine products (ADMA and SDMA) in neighboring arginine 8 (R8) within an RXS motif of histone H3. We then measured the non-covalent interactions by ^{31}P NMR spectroscopy and confirm a salt bridge between phosphoserine-10 and unmodified arginine-8 of histone H3. Briefly, we estimated the free energy contributions of salt bridges by measuring the change in pK_a of ionizable groups caused by altered electrostatic interactions. This salt bridge was absent in phosphoserine-10 peptides containing symmetrically dimethylated arginine-8 residues. Finally, computational modeling was carried out to further characterize the interactions in these peptides. We conclude that the presence of a phosphate group on serine-10 does not physically prevent methylation of neighboring arginine-8 within the RKS motif of histone H3. Instead, the negatively charged phosphoserine creates interactions that most likely mask PRMT enzyme recognition.

2. Materials and methods

PRMT Expression and Purification of GST fusion proteins of PRMT1 and PRMT4. All methyltransferase constructs were purified from *E. coli* and were obtained from Dr. Steven Clarke (UCLA). Plasmids expressing rat GST-PRMT1 or GST-PRMT4 were expressed in BL21 Star (DE3) cells. Briefly, cells containing GST-PRMT1 or GST-PRMT4 were induced for 4 h at 37 °C with 0.4 mM isopropyl β -D-thiogalactopyranoside (IPTG). They were lysed with seven 20-s sonicator pulses (Fisherbrand Model 120 Sonic Dismembrator). The cell lysates were centrifuged for 50 min at 23,000 $\times$ g at 4 °C. The proteins were then purified from extracts by incubating them with glutathione-sepharose 4B beads (GE Health) according to the manufacturer's instructions. They were next eluted with 30 mM glutathione, 50 mM Tris-HCl, 120 mM NaCl, and 4% glycerol at pH 7.5. Purified protein preparations were quantified by a Lowry assay after trichloroacetic acid precipitation [12].

PRMT5 expressed in bacteria yields inactive enzyme. In addition, and for optimal activity, human PRMT5 acts as part of a larger complex that requires another protein known as methylosome protein 50 (MEP50) [13]. Thus, PRMT5 methylation reactions were carried out using commercially available human recombinant PRMT5 in complex with human recombinant MEP50, produced by co-expression in an insect cell/baculovirus expression system, PRMT5 2–637(C-term), MEP50 2–342(C-term) (HMT-22-148, Reaction Biology Corp.).

Methylation Reactions. Full-length GST-PRMT1 (1–2 μg) or GST-PRMT4 (1–2 μg) was incubated with 1–30 μg of either full-length histone H3 (positive control), unmodified peptide 1 or phosphorylated peptide 2, and [^3H]AdoMet at 37 °C in a 30 μL –90 μL total reaction volume. See figure legends for actual amounts. Negative controls such as an enzyme alone or peptide alone control were carried out in the presence of [^3H]AdoMet and all other reagents. Methylation reactions were carried out as outlined in the literature [14] using full-length histone proteins (New England Biolabs) and unmodified or phosphorylated peptide substrates (VCP Bio, Inc.) corresponding to the histone H3 tail (amino acids 4–16; ~1.4 kDa): NH_2 -TKQTARKSTGGKAP-COOH (unmodified) and NH_2 -TKQTARKSP_TGGKAP-COOH (phosphoserine). The accuracy of the sequence for the peptide substrates was verified by

mass spectrometry by the manufacturer. To the best of our abilities, careful handling and preparation of peptides 1 and 2 was taken to ensure that equal amounts were used in subsequent methylation reactions and analyses by fluorography and TLC. For example, both peptides were dissolved in water to an equal concentration of 1 $\mu\text{g}/\mu\text{L}$.

For SDS-PAGE analysis, methylation reactions were stopped with the addition of an equal volume of SDS gel sample buffer. Briefly, methylation reactions containing peptide substrates were quenched using 2X sample buffer (0.05 M Tris, 20% glycerol, 0.1% bromophenol blue, pH 6.8), heated at 100 °C for 3 min, and loaded on a 20% acrylamide and 0.43% N,N-methylene bisacrylamide gel. Electrophoresis was performed at 80 V at room temperature. The gel was stained with Coomassie Brilliant Blue R-250 for 1 h and destained with 10% methanol and 5% acetic acid for 5 h. The destained gel was prepared for fluorography by incubation in EN3HANCE (PerkinElmer Life Sciences) for 1 h and washed with water for 20 min. The gel was dried at 70 °C *in vacuo* and exposed to Kodak BIOMAX XAR scientific imaging film at –80 °C for different lengths of time ranging from seven days to several weeks using molecular mass standards 1.7–42 kDa (Spectra Multicolor Low Range Protein Ladder, ThermoFisher).

For samples analyzed by thin layer chromatography (TLC), 10 μmol of either monomethylated arginine (ω -MMA), asymmetric dimethylarginine (ADMA), or symmetric dimethylarginine (SDMA) were analyzed separately and also added to approximately 1/3 of the methylation reactions for acid hydrolysis [15]. The samples were resuspended in approximately 5 μL of water and immediately spotted onto 20 $\times$ 20 cm silica-coated plates (Sigma-Aldrich Z122807-25 EA) in 1- μL aliquots. After pre-equilibrating the chamber with the mobile phase: 30% ammonium hydroxide/chloroform/methanol/water (2:0.5:4.5:1), samples were spotted on the plate and run for approximately 5 h. After air drying overnight, the plate was sprayed with a ninhydrin solution [15 mg/mL ninhydrin dissolved in 97% (vol/vol) butanol, 3% (vol/vol) glacial acetic acid] and allowed to air dry. To visualize the ninhydrin spots indicating arginine residues, the plate was heated at 100 °C for 10 min and sectioned into 3-mm-wide slices. The silica from each slice was mixed with 950 μL of water and vortexed. This material was added to 5 mL of scintillation fluid (Research Products International). The samples were loaded on to a Beckmann LS6500 Liquid Scintillation Counter where three 5-min counting cycles were averaged to determine the corresponding counts per minute of each fraction.

Peptides used for ^{31}P NMR. Six peptides (14 amino acids long) corresponding to the histone H3 tail were identical except for specific changes in the amino acid sequences underlined. Note: all peptides used for NMR were phosphorylated. Peptide 1-RKSp: Thr-Lys-Gln-Thr-Ala-Arg-Lys-Ser(PO_3H_2)-Thr-Gly-Gly-Lys-Ala-Pro; peptide 2-RASp: Thr-Lys-Gln-Thr-Ala-Arg-Ala-Ser(PO_3H_2)-Thr-Gly-Gly-Lys-Ala-Pro; peptide 3-AKSp: Thr-Lys-Gln-Thr-Ala-Ala-Lys-Ser(PO_3H_2)-Thr-Gly-Gly-Lys-Ala-Pro; peptide 4-AASp: Thr-Lys-Gln-Thr-Ala-Ala-Ala-Ser(PO_3H_2)-Thr-Gly-Gly-Lys-Ala-Pro; peptide 5-R(SDMA)KSp: Thr-Lys-Gln-Thr-Ala-Arg(Symmetrically Dimethylated Arginine)-Lys-Ser(PO_3H_2)-Thr-Gly-Gly-Lys-Ala-Pro; peptide 6-R(SDMA)ASp: Thr-Lys-Gln-Thr-Ala-Arg(Symmetrically Dimethylated Arginine)-Ala-Ser(PO_3H_2)-Thr-Gly-Gly-Lys-Ala-Pro (1 mg; CPS Scientific, Sunnyvale, CA). In addition, all precautions were taken to ensure the peptides were treated equally and preserved to prevent degradation. The peptides came lyophilized and were stored at –20 °C.

NMR Sample Preparation. A modified protocol optimized for these peptides was based on the work of [16,16]. Briefly, approximately 1 mg of commercially purchased customized peptide was dissolved in 520–550 μL of buffer, final concentration 1 mM. The buffer was made as a stock solution of 25 mL consisting of 30 mM HEPES, 5 mM MgCl_2 , 5 mM KCl, 10% glycerol and 8% D_2O at 22 °C. The peptide-buffer solution was then transferred to a 5 mm diameter Wilmad NMR tube. The pH was adjusted in small increments with 1 M HCl and 1 M NaOH to obtain at least 10 NMR spectra. The pH is first lowered to a value less than 3.00

and then raised to a value above 8.00.

NMR Spectroscopy. ^{31}P NMR data were acquired on a Bruker 400 MHz spectrometer using TopSpin software (version 3.2.6). Final sample volumes were approximately 600 μL . The parameters used for the NMR spectrometer include 1000 scans, 22–25 $^{\circ}\text{C}$, degree of substitution of 4, and a relaxation delay of 3 s (Supplemental Materials Figs. 2–6). ^{31}P spectra referenced indirectly to 85% H_3PO_3 .

pK_a Determination. Data analysis for NMR titration was done under the assumption that the titration sites are noninteracting. Starting with the Henderson-Hasselbalch equation:

$$\text{pH} = \text{p}K_a + \log\left(\frac{[\text{A}^-]}{[\text{HA}]}\right) \quad (1)$$

we replaced protonated and deprotonated concentrations with the chemical shifts using the following relationships:

$$(\delta_{\text{HA}} - \delta_{\text{obs}}) \propto [\text{HA}] \quad (2)$$

$$(\delta_{\text{obs}} - \delta_{\text{A}^-}) \propto [\text{A}^-] \quad (3)$$

to arrive at the modified version of the Henderson-Hasselbalch equation:

$$\text{pH} = \text{p}K_a - \log\left[\frac{(\delta_{\text{HA}} - \delta_{\text{obs}})}{(\delta_{\text{obs}} - \delta_{\text{A}^-})}\right] \quad (4)$$

where δ_{obs} is the observed chemical shift and the δ_{HA} and δ_{A^-} are the chemical shifts of the protonated and deprotonated forms (determined by the fit), respectively. To fit the experimental data of pH versus observed chemical shifts, the following equation has been rewritten as $\delta_{\text{obs}} = f(\text{pH})$.

$$\delta_{\text{obs}} = \frac{\delta_{\text{HA}} + \delta_{\text{A}^-} \times 10^{(\text{p}K_a - \text{pH})}}{1 + 10^{(\text{p}K_a - \text{pH})}} \quad (5)$$

We can arrive at the Hill equation or four-parameter logistic equation by introducing the Hill Slope coefficient and algebraically manipulating equation (4), which results in

$$\delta_{\text{obs}} = \delta_{\text{A}^-} + \frac{\delta_{\text{HA}} - \delta_{\text{A}^-}}{1 + 10^{[(\text{p}K_a - \text{pH}) \times \text{Hill Slope}]}} \quad (6)$$

MATLAB (Release 2019b, Math Works Inc.) scripts and functions were developed in-house to carry out nonlinear regression to model the experimental results. The MATLAB functions were used to determine the values of the δ_{A^-} , δ_{HA} , $\text{p}K_a$, and Hill Slope parameters of the Hill equation. The *nlinfit* MATLAB function used in this step estimates the coefficients using iterative least squares estimation. Confidence intervals at 95% were calculated to determine the uncertainty of the $\text{p}K_a$ curve fitting.

Molecular dynamics simulations. Starting structures for the six peptides used in the experiments described above were built by extracting the appropriate residues from PDB structure 1KX5 [17]. These sequences were then edited to include a phosphate group at serine-10 (all peptides, in keeping with the peptides used for NMR), symmetric dimethylation at arginine-8 (peptides 5-R_{SDMA}KSp and 6-R_{SDMA}ASp), and/or an alanine substitution for lysine-9 (peptides 2-RASp, 4-AASp, and 6-R_{SDMA}ASp).

Solvated peptide systems were built using the program *tleap* in AmberTools 18 [18]. Each system contained the peptide, approximately 4400 water molecules, and the number of chloride (Cl^-) ions necessary to neutralize the system. The peptide was modeled using the AMBER-FB15 force field [19] with new parameters for phosphorylated residues [20]. The water and ions were modeled using TIP3P-FB water model [21] and Joung-Cheatham ion parameters for SPC/E, a water model with similar overall parameters to TIP3P-FB [22].

Molecular dynamics simulations were performed using *pmemd* and *pmemd.cuda* [23] of Amber 18 and all the heavy atom-hydrogen covalent bonds were constrained with the SHAKE algorithm. The real-space cutoff for nonbonded interactions was set at 9.0 $\text{\AA}$. Before running

dynamics, each system was subjected to a two-step energy minimization procedure. In the first step, 1000 steps of minimization were performed with the peptide atoms restrained to their initial positions using a harmonic potential of 10 $\text{kcal mol}^{-1} \text{\AA}^{-2}$. The second round of minimization for 1000 steps was performed without any restraints on the peptide.

After minimization was complete, each system was heated at constant volume from a starting temperature of 100 K to a final temperature of 295.15 K (22 $^{\circ}\text{C}$) over 30 ps with the peptide atoms again restrained using a harmonic potential of 10 $\text{kcal mol}^{-1} \text{\AA}^{-2}$. Next, the density of the system was equilibrated by simulating in the NPT ensemble at 1.0 atm and 295.15 K using a Berendsen barostat with coupling constant 2.0 ps^{-1} and a Langevin thermostat with a collision frequency of 1.0 ps^{-1} for 50 ps. The final structure from this equilibration was used as the starting structure for the replica exchange MD simulations, which were performed in the NVT ensemble.

The replica exchange MD simulations (REMD) were performed at 24 exponentially spaced temperatures ranging from 295.15 K to 370 K. To decorrelate the starting structures, the replicas were first simulated for 10 ns before any exchanges were performed. We then performed 200 or 400 ns (depending on convergence of the conformational ensemble, see below) of replica exchange MD simulation, with exchange attempts every 0.20 ps. Structures were saved every 5.0 or 10 ps, yielding 40,000 structures for each of the six peptides.

MD simulation data analysis. To check the convergence of each simulation, we split each simulation into equal halves, discarded the first 10% of each half, and then computed the Kullback-Leibler (K-L) divergence of the end-to-end distance for the peptide, defined as the distance between the C α atoms of the first and last residues of the peptide. We observed rapid exponential decay of the K-L divergence toward zero for all systems, as would be expected for independent simulations converging to the same ensemble average. Close intramolecular contacts to the phosphate group of phosphorylated Ser10 were detected using the ‘hbond’ function of *cpptraj* in AmberTools 18 [24]. We defined the ‘acceptors’ as the oxygen atoms of the phosphate group; the ‘donors’ were N and O atoms with bonded H atoms (including, for example, Lys and Arg side chains, Thr side chain, the peptide backbone, etc.). We used an O–N/O interatomic distance cutoff of 3.5 $\text{\AA}$ and an O–H–N/O angle cutoff of 135 $^{\circ}$; in other words, interactions had to simultaneously have an interatomic distance of <3.5 $\text{\AA}$ and an angle of >135 $^{\circ}$ in order to be counted. Interactions that occurred in fewer than 1% of the conformational ensemble were ignored.

3. Results

Detection of methylated arginine via fluorography and TLC. To determine whether a phosphorylated serine-10 residue would inhibit methylation of neighboring arginine-8 in histone H3, we carried out *in vitro* methylation reactions using unmodified and phosphorylated peptides corresponding to histone H3. Protein arginine methylation can form three different products: mono-, symmetric- or asymmetric-arginine methylation (MMA, SDMA, ADMA) catalyzed by a family of nine protein arginine methyltransferases (PRMT1-9) [25]. Histone H3 is physiologically asymmetrically dimethylated by PRMT4/CARM1 at arginine-17 [26], at arginine-26 [27], and symmetrically dimethylated by PRMT5 at arginine-8 [11,28]. PRMT4 does not methylate histone H3 at arginine-8; however, since H3 is a physiological substrate, it was used as a control. In addition, Tang et al. have shown that PRMT1, which primarily forms ADMA, catalyzes 85% of the total protein arginine methylation in rat fibroblast and in mouse liver [29]. Thus, in addition to *in vitro* reactions with PRMT4 and PRMT5, we also performed methylation reactions with PRMT1, the most active member of the PRMT family. We expected that only PRMT5 would methylate peptides corresponding to arginine-8 and serine-10 of histone H3.

Fig. 1, panel A shows that PRMT5 will methylate both unmodified peptide 1 (lane 1) and full-length histone H3 (lane 3) but not phosphorylated peptide 2 (lane 2). PRMT5 is the physiological catalyst,

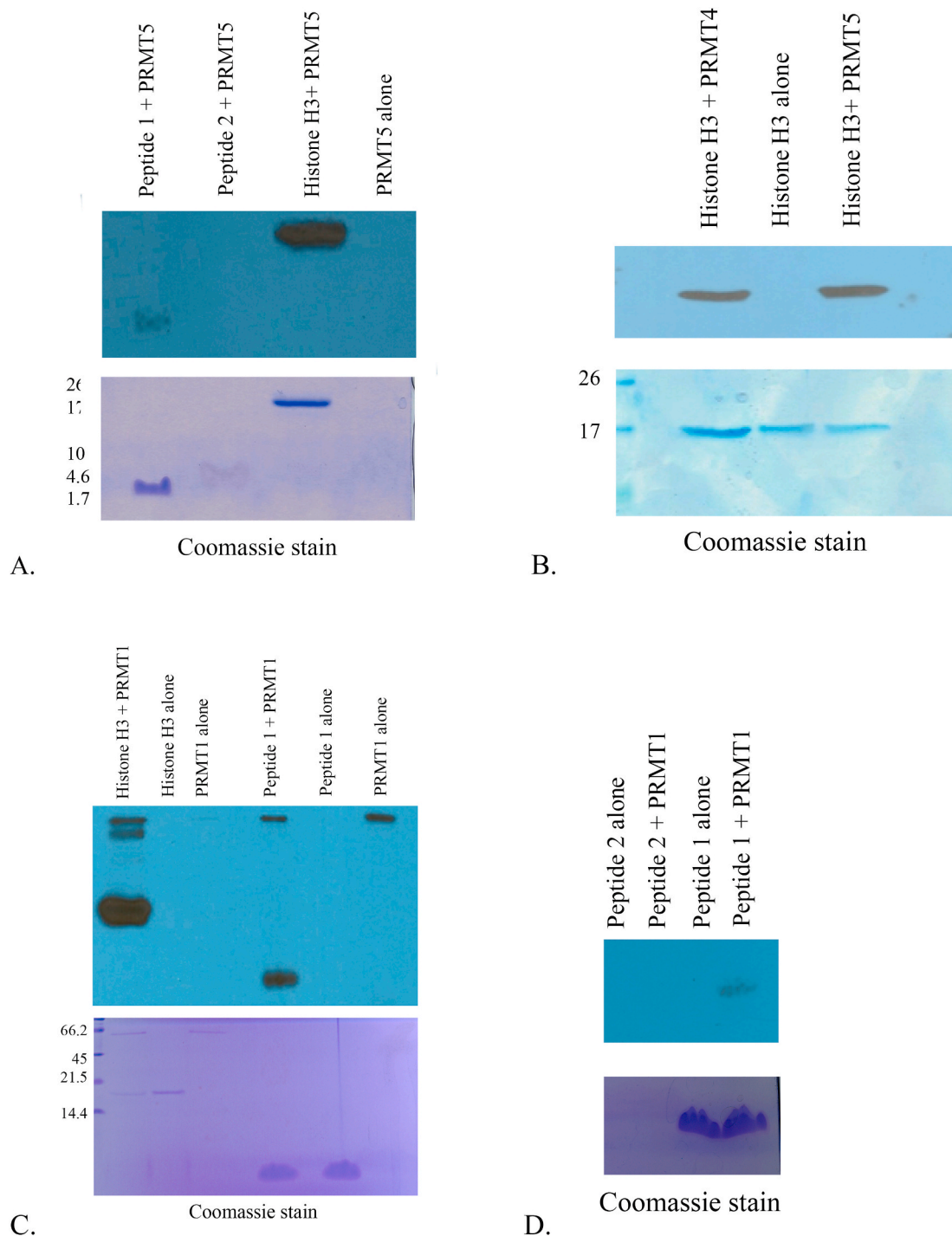


Fig. 1. Methylation of peptide 1 (TKQTARKSTGGKAP) or peptide 2 (TKQTARKSpTGGKAP) corresponding to histone H3 detected via fluorography. A) GST-PRMT5 (1 μ g), B) GST-PRMT4 (1 μ g), C and D) GST-PRMT1 (1 μ g), were incubated with substrates histone H3, peptide 1 (1 μ g) or peptide 2 (1 μ g) in the presence of 0.5 μ M S-adenosyl-L-[methyl- 3 H]methionine for 1 h at 37 $^{\circ}$ C in a final volume of 30 μ l of HEPES buffer as described in Materials and Methods. The samples were then resolved via 20% SDS-PAGE. Full-length histone H3 was used as a positive control, PRMT, and peptide alone reactions lacking enzyme or substrates were used as negative controls. Negative controls such as PRMT1 alone or Peptide 1 alone were carried out in the presence of all reagents, including [3 H]AdoMet. The radioactive methylation reactions were exposed on film as described Materials and Methods for A) 8 days, B) 7 days, C) 3 weeks and D) 9 days.

responsible for producing symmetric dimethylated arginine-8 in histone H3 [11]. Fig. 1 panel B shows that both PRMT4 (lane 1) and PRMT5 (lane 3) catalyze the methylation of full-length histone H3. While PRMT4 does methylate histone H3, it does not methylate it in this region. Thus, PRMT4 was not capable of methylating either peptide 1 or 2 (see supplemental information, Fig. 1). To further explore whether the inhibition is unique to PRMT5, which forms SDMA, or whether the

neighboring phosphate could inhibit other members of the PRMT family, we performed *in vitro* methylation reactions with PRMT1, the most active and predominant member of the PRMT family (Fig. 1, panels C and D and Fig. 2) [29]. PRMT1 forms ADMA, thus this reaction serves to determine whether the phosphorylated serine only inhibits SDMA, catalyzed by PRMT5 or both ADMA and SDMA. Here we show that PRMT1 will methylate full-length histone H3 (lane 1, Fig. 1C) and

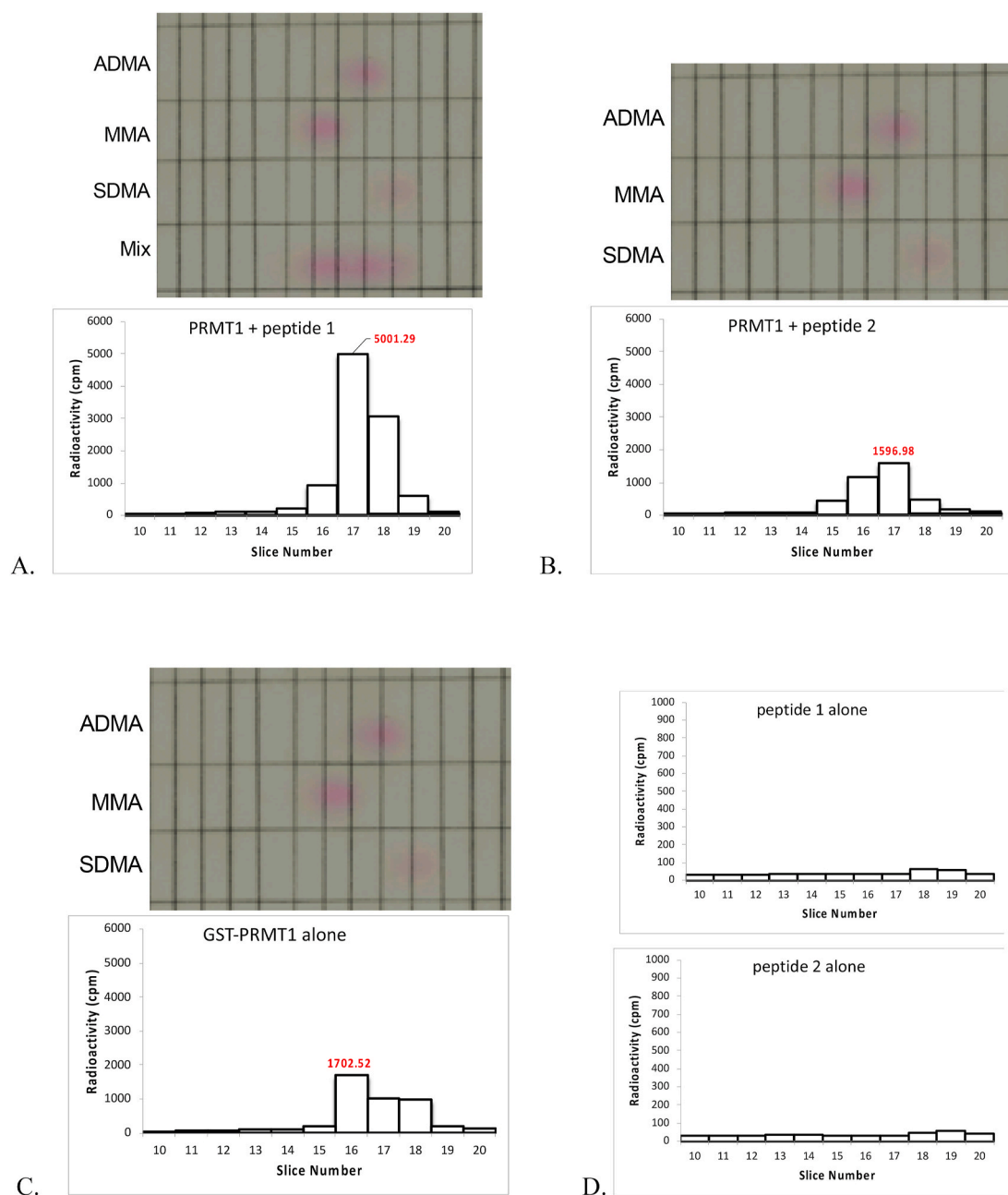


Fig. 2. Hydrolysates of methylation reactions of PRMT1 incubated with either unmodified peptide 1 or phosphorylated peptide 2 and resolved by TLC. TLC for hydrolysates of the reaction mixture and individual and mixed standards of ADMA, MMA, and SDMA. The upper portion shows the ninhydrin staining of the TLC plate containing the arginine derivative standards; the lower portion shows the radioactivity corresponding to the TLC slices of the reaction mixture lane. A) TLC of hydrolysates for PRMT1, 5001 cpm; mix lane contains reaction mixture and standards; B) TLC of hydrolysates for PRMT1 incubated with phosphorylated peptide 2, 1596 cpm; C) TLC of hydrolysates for PRMT1 alone (control), 1702 cpm; and D) TLC of hydrolysates for peptide 1 and peptide 2 alone controls. All controls contained [^3H]AdoMet.

unmodified peptide 1 (lane 4, Fig. 1C and lane 4, Fig. 1D), but not phosphorylated peptide 2 (lane 2, Fig. 1D) (Fig. 2). The two *PRMT1* alone controls show that although one enzyme preparation was more active than the other, both preparations were inhibited by a neighboring phosphate group. The data is representative of several methylation reactions detected by fluorography. Thus, regardless of the methyltransferase used or the methylated mark deposited (ADMA or SDMA), detection of radioactivity (methylation) is greater for the unmodified peptide when compared to the phosphorylated peptide.

Analysis of peptides by fluorography requires optimization because low molecular weight proteins and peptides (<10 kDa) are difficult to resolve by polyacrylamide gel electrophoresis [30]. The SDS-PAGE gels

in Fig. 1A and D of Coomassie blue-stained peptides reveals that peptide 1 stains darker than phosphorylated peptide 2. Coomassie blue is a negatively charged dye that binds to positively charged residues, specifically arginine, lysine, and histidine residues [31]. We hypothesize that the presence of a negatively charged phosphate group is preventing adherence of the dye, especially to neighboring arginine-8 and lysine-9. Unmodified peptide 1 always stained darker than phosphorylated peptide 2. Thus, to further confirm the difference in production of methylated arginine species between peptides 1 and 2, TLC was performed with acid hydrolysates of methylation reactions catalyzed by PRMT1, the most active member of the methyltransferase family (Fig. 2). Three additional methylation reactions served as controls: PRMT1 alone,

peptide 1 alone and peptide 2 alone. These controls were carried out in parallel and in the presence of [^3H]AdoMet. Radioactivity from TLC slices show that when PRMT1 is incubated with unmodified peptide 1, the reaction produces over 5001 cpm (Fig. 2A). However, when PRMT1 is incubated with phosphorylated peptide 2, 1597 cpm are produced (Fig. 2B). Since this amount is not greater than the PRMT1 alone control (1703 cpm) (Fig. 2C) or the peptide alone controls (Fig. 2D), we conclude that a phosphate group on serine-10 inhibits methylation of neighboring arginine-8, even in the presence of the most active member of the PRMT family, PRMT1.

NMR. Methylation does not change the overall positive charge of the arginine guanidinium group; however, depending on the number of methyl groups added can change the number of hydrogen bond donor sites [32–34]. The basic nature of arginine and lysine mean that they add a positive charge to the peptide. Table 1 shows a list of the peptides as well as their net charge at a neutral pH (7.0). To investigate this interaction at a molecular level, we focused on detecting non-covalent interactions between phosphoserine and arginine within the RKS motif of histone H3. This type of experiment is studied by NMR in two ways. The first method involves measuring one or two of the pK_a values of the residues involved and then substituting those potential residues to remove any interactions. The second method involves creating changes in pK_a values by studying the residues involved and comparing changes upon protein unfolding [35]. Since the histone H3 N-terminal tail is inherently unstructured and not folded, the first method (i.e. residue substitution) was used for the detection of the salt bridge between phosphoserine-10 and unmodified arginine-8. Phosphorylated peptides corresponding to the first 14 amino acid residues of histone H3 were purchased. We compared the changes in pK_a value of the original sequence against the pK_a of the peptides with altered amino acids. We used ^{31}P NMR for the observation of a salt bridge because the natural abundance of phosphorus (100%) and gyromagnetic ratio (17.235 MHz T^{-1}) is high enough that we can observe a peak without the need to isotopically label our peptide substrates.

A salt bridge or network of salt bridges form between oppositely charged residues. Two types of salt bridges exist: the surface-exposed salt bridge and the buried salt bridge that can interact in both an inter- and intramolecular manner [36]. Salt bridges buried within the same protein are expected to have stronger electrostatic interactions. For instance, the largest salt bridge observed to date, was detected in T4 lysozyme, whose buried salt bridge between Asp-70 and His-30 contributes favorable stability at about 3–5 kcal/mol, and a pK_a shift of 3 units for Asp-70 [37]. In contrast, a single salt bridge within two oppositely charged residues, an arginine and phosphoserine salt bridge (i, i+4) was calculated as 0.45 kcal/mol using synthetic peptides [38]. Although the pK_a was not reported, it is expected that this value is small. Indeed, a study using ^{31}P NMR reported a downshifted pK_a value of

about 0.7 units in an arginine-phosphoserine salt bridge interaction with a small peptide of 24 amino acid residues [39]. Thus, we developed NMR methods for a 14 amino acid long peptide expecting small changes in pK_a (0.3–0.7 units) as an indication of a salt bridge and performed a ^{31}P NMR titration by varying the pH which allowed for the calculation of the pK_a of the side chain. A list of the experimentally derived pK_a values is shown in Table 1.

A Salt Bridge Interaction in Peptide 1-RKSp (TKQTARKSpTGGKAP). Titrations of peptide 1-RKSp were carried out using the phosphoserine for ^{31}P NMR detection. This allowed for the determination of the acid dissociation constant of the monophosphorylated peptides. Fig. 3 shows representative sigmoidal Henderson-Hasselbalch fitting of the chemical shifts versus pH of all peptides. See supplemental information for additional NMR values and corresponding spectra. As shown in Table 1, the average pK_a value of three trials for peptide 1-RKSp was 5.55. A salt bridge for NMR spectroscopy is detected at a pK_a shift below 0.7 units, as was detected for peptides based on CLASP2 protein [39]. Our value falls within this range when compared to peptide 4-AASp which lacks a positively charged arginine or lysine residue, thus served as a control, and has an average pK_a value of 5.92. The value of the experimentally obtained control was compared to the published pK_a value of solvent exposed phosphoserine model peptide glycylglycylserylalanine (GGSpA) which has a pK_a of 6.1 [40]. Hence, the pK_a value of peptide 1-RKSp shifted 0.37 units (5.92–5.55), indicative of a salt bridge between the neighboring arginine-8 and phosphorylated serine-10. This result also confirms computational simulations described in this study as well as those previously described in the literature [8].

To determine the contribution of the lysine residue, we carried out similar titrations using peptide 2-RASp and peptide 3-AKSp which contains an alanine in place of lysine-9 or arginine-8 respectively. The average pK_a values were 5.72 and 5.83. Since these values fall between both the pK_a of the control peptide 4-AASp and peptide 1-RKSp, with small shifts in delta pK_a (<0.2, Table 1), we postulate that in the absence of an arginine residue, an interaction between the lysine residue and the phosphoserine becomes prominent.

To determine whether methylation of arginine residues affects salt bridge formation between positively charged arginine and negatively charged phosphoserine, we performed titrations using peptide 5-RSDMAKSp and peptide 6-RSDMAASp where arginine is modified with a symmetrically dimethylated group. The delta pK_a values of peptide 5-RSDMAKSp and peptide 6-RSDMAASp shifted 0.18 and –0.15 units respectively, indicating an absence in salt-bridge upon methylation, and underscoring the contribution of the lysine residue when the arginine residue is either methylated or absent.

Molecular Dynamics Simulations. MD simulations of these peptides allow us to visualize and quantify the interactions of phosphoserine-10 in the presence or absence of positively charged residues such as

Table 1

^{31}P NMR titration pK_a derived for the 6 peptides based on the N-terminal sequence of histone H3 amino acid residues 3–16.

#	Peptide Sequence	Net charge (pH 7)	Modification	pK_a Values	Mean pK_a	Std. Dev.	Delta pK_a (AASp)
1	TKQTARKSpTGGKAP	+2	None	5.55 5.60 5.51	5.55	0.05	0.37
2	TKQTARASpTGGKAP	+1	K9 to A	5.78 5.69 5.68	5.72	0.06	0.20
3	TKQTAARKSpTGGKAP	+1	R8 to A	5.68 5.90 5.91	5.83	0.13	0.09
4	TKQTAASpTGGKAP	0	R8 to A and K9 to A	6.04 5.84 5.87	5.92	0.17	–
5	TKQTARSDMAKSpTGGKAP	+2	R8 SDMA	5.70 5.77	5.74	0.05	0.18
6	TKQTARSDMAASpTGGKAP	+1	K9 to A R8 SDMA	6.37 5.76	6.07	0.43	–0.15

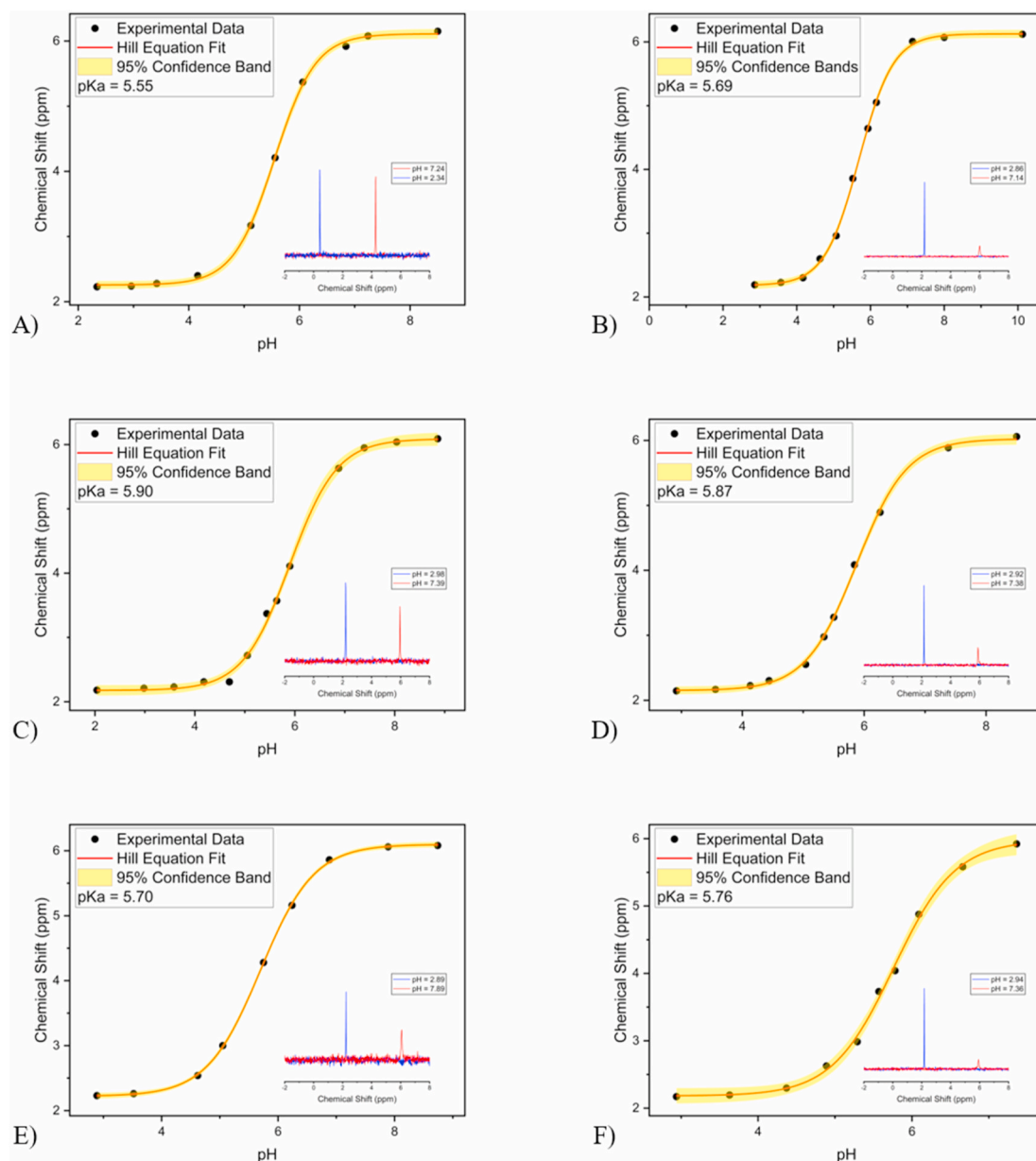


Fig. 3. Dependence of the ^{31}P NMR chemical shifts of 1 mg of A) Peptide 1 (TKQTARKSpTGGKAP), one of three trials, B) Peptide 2 (TKQTARASpTGGKAP), one of three trials, C) Peptide 3 (TKQTAAKSpTGGKAP), one of three trials, D) Peptide 4 (TKQTAASpTGGKAP), one of three trials, E) Peptide 5 (TKQTAR(SDMA)KSpTGGKAP), one of two trials, and F) Peptide 6 (TKQTAR(SDMA)ASpTGGKAP), one of two trials. All peptides were dissolved in approximately 0.5 ml of 30 mM HEPES, 5 mM MgCl_2 , 5 mM KCl, 10% glycerol and 8% D_2O at 22 °C. The titration curves were calculated with the parameters described in Materials and Methods. The x-axis is the experimental pH measured and the y-axis is the chemical shift observed at the respective pH value.

arginine and lysine. In our simulations of peptide 1-RKSp, as well as peptide 2-RASp, we observed a consistently strong interaction (salt bridge) between phosphoserine-10 and the neighboring arginine-8 (Fig. 4A). When both arginine-8 and lysine-9 are replaced by alanine residues, however, other interactions are formed with phosphoserine-10 that at least partially make up for the elimination (Fig. 4B).

To quantify these interactions in a comprehensive way, we computed the presence of intramolecular interactions with the phosphate group of phosphoserine-10 and all of the possible positively charged or partially-positively charged moieties in the peptide. Table 2 lists the phosphoserine-10 interactions (close contacts) that can take place with each constituent group. To more readily understand what is happening, we have separated these interactions into several categories: interactions with the positively charged guanidino group of (unmodified or symmetrically dimethylated) arginine-8, interactions with the

positively charged ammonium groups of lysine residues and the N-terminus of the peptide, interactions with the hydroxyl groups of threonine residues, and interactions with the amide protons of the peptide backbone.

At a glance our results demonstrate that there are *many* intrapeptide interactions formed with the phosphate group of phosphoserine-10, but Table 2 indicates that a few trends stand out. Perhaps most noticeably, we find that the positively charged guanidino group of arginine-8 of peptide 1 is able to form more than one interaction at a time with the phosphate group of phosphoserine-10 (i.e., interactions that are sometimes bidentate), with the average interaction count of 1.53. When lysine-9 is replaced by an alanine in peptide 2-RASp, the average interaction count between these two residues increases further to 1.74. Conversely, when arginine-8 is symmetrically dimethylated, the average interaction count decreases to 0.24–0.26 in peptides 5-R_{SDMA}KSp and 6-

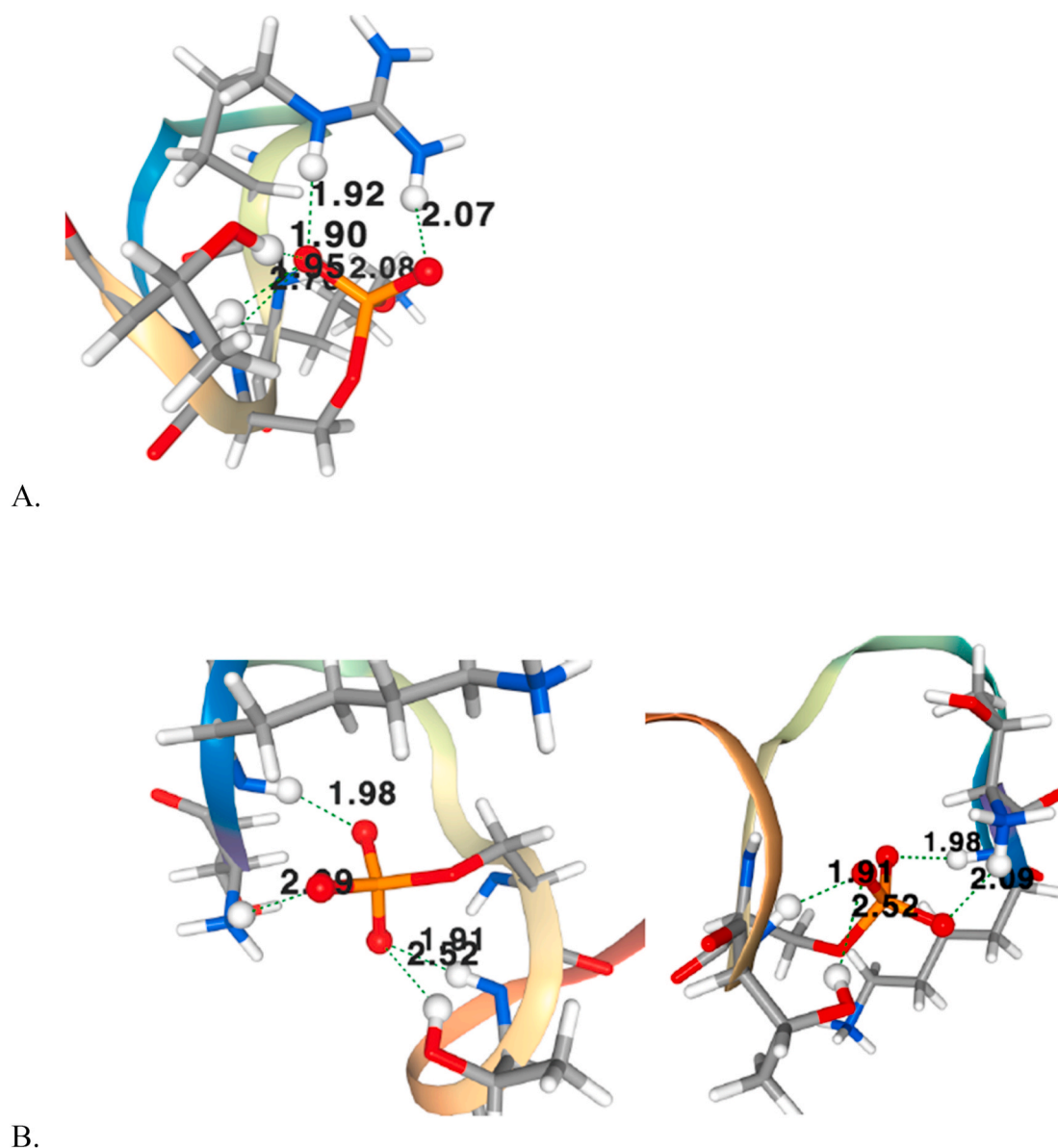


Fig. 4. Comparison of histone H3 peptides 1-RKSp, and 4-AASp, (PDB file 1kx5, Python, chain; ribbon). A) Phosphoserine-10 (red/orange), creates a salt bridge with arginine-8 (blue/white). The salt-bridge is denoted by dashed lines and by the proximity of the atoms (1.92 Å and 2.07 Å, respectively). B) In the absence of positively charged arginine-8 and lysine-9, phosphoserine-10 (red/orange), maintains or creates interactions with other atoms. These include the threonine-11 residue, other lysine residues, and backbone amide protons found both nearby and more distant to phosphoserine-10. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

$R_{SDMA}Sp$. We also observe some interaction with lysine-9, but this is relatively infrequent (0.25) when arginine-8 is present (peptide 1-RKSp) and increases in frequency (0.55) only when arginine-8 is either replaced by an alanine (peptide 3-AKSp) or symmetrically dimethylated (peptide 5- $R_{SDMA}KSp$). We see that the interaction with lysine-9 can be partially compensated for by interactions with other lysine residues or the N-terminus of the peptide when lysine-9 is replaced by an alanine residue. One interaction that remains relatively constant across all peptides, however, is that with the hydroxyl group of the threonine-11 sidechain (0.77–0.93). Finally, we observed considerable interaction between the phosphate group and the amide protons of the nearby backbone (residues 9–11), as well as amide protons that are more distant (all other residues). Most notably, the phosphate group of serine-10 has the greatest interaction with the distant backbone in the absence of arginine or lysine residues. In general, these distant moieties contribute a total of 1.40–2.04 interactions to the phosphate group. Together these data suggest that the phosphate group of phosphoserine-10 is not fully

solvent-exposed and engages in a variety of intrapeptide interactions both when arginine-8 is present (and forms a salt bridge with this residue) and when arginine-8 is not present or is symmetrically demethylated (thereby preventing or hindering the formation of a salt bridge). These intrapeptide interactions may explain why the change in pK_a of the phosphate group is not as large when the salt bridge with arginine-8 is present as compared to other studies (e.g., Kumar et al.).

4. Conclusion

How one PTM can alter another is the focus of intense investigation [41–43]. Moreover, the interplay between arginine methylation and serine phosphorylation has been identified in several proteins [44–46]. Using peptide substrates based on histone H3, we set out to understand the role of phosphoserine-10 on neighboring arginine-8. We focus on the RKS motif because interactions between unmodified arginine-8 and phosphorylated serine-10 affect protein binding [8]. Building on this

Table 2

Probabilities of close contacts^a between various residues/moieties and the phosphate group of serine-10 from REMD simulations.

Peptide	Arg8/Arg (SDMA)8	Lys9	Other Lys/ NH ₃ ⁺	Thr11	Local backbone	Distant backbone
1	1.53	0.25	0.20	0.88	1.79	0.06
2	1.74	n/a	0.36	0.78	1.64	0.18
3	n/a	0.55	0.47	0.80	1.22	0.18
4	n/a	n/a	0.75	0.93	1.02	1.02
5	0.26	0.55	0.27	0.84	1.48	0.26
6	0.24	n/a	0.72	0.77	1.26	0.40

^a Close contacts are defined as those for which the heavy atom distance between an oxygen atom of the phosphate group and the heavy atom “donor” of a hydrogen atom on another residue/moiety are within 3.5 Å and have a bond angle between these two atoms of >135°. Values above 1.00 (100%) can be interpreted as more than one interaction occurring simultaneously. “Arg8/Arg (SDMA)8” represents the charged arginine-8 or symmetrically dimethylated arginine-8 side chain. “Lys9” represents the lysine-9 side chain, while “Other Lys/NH₃⁺” counts interactions with the other Lys residue side chains and the N-terminus of the peptide. “Thr11” represents interactions with the threonine-11 side chain, specifically the hydroxyl group. Finally, “Local backbone” counts interactions with the backbone amide protons of residues 9–11, while “Distant backbone” encompasses the backbone amide protons of the other residues.

evidence, we carried out *in vitro* methylation reactions using peptides based on the RKS motif of histone H3. The fluorography and TLC data show that phosphoserine-10 will inhibit deposition of methylated groups (both ADMA and SDMA) to neighboring arginine-8.

We further investigated why methylation was inhibited in the presence of a phosphoserine through NMR and computational studies. We conclude that a salt bridge forms between phosphoserine-10 and arginine-8 within the RKS motif of histone H3, limiting interactions with other neighboring atoms. This was validated by comparing the pK_a values of several peptides. For example, we expected peptide 2-RASp, which is lacking a lysine residue, to have a similar pK_a value to that of peptide 1-RKSp. However, the average pK_a values for these peptides were 5.55 and 5.72 respectively. This interaction is further confirmed when the arginine residue is completely eliminated (peptide 3-AKSp), with a pK_a value of 5.83. The symmetrically dimethylated arginine peptides 5 and 6 represent the possible interaction between the two post-translational modifications: arginine methylation and serine phosphorylation. The average pK_a values for peptide 5-RSDMAKSp and 6-RSDMAKSp were determined to be 5.74 and 6.07, respectively. When we compare these values to each other, we conclude that the lysine residue is contributing to protein stability. We reason that when the arginine residue is either absent or symmetrically dimethylated, the contribution of the lysine residue becomes significant.

Our molecular dynamics simulation results suggest that the largest observed differences in pK_a of the phosphate group would occur when comparing peptide 1-RKSp to peptides 3-AKSp and 4-AASp which have no arginine residues, or to peptides 5-RSDMAKSp and 6-RSDMAKSp, which have symmetrically dimethylated arginine residues. We would expect to see smaller differences between peptides 1-RKSp and 2-RASp, peptides 3-AKSp and 4-AASp, or peptides 5-RSDMAKSp and 6-RSDMAKSp. Indeed, the experimental data bear these predictions out quite well, with the sole exception of peptide 6-RSDMAKSp, which has a large difference in pK_a from peptide 5-RSDMAKSp. We note, however, that the experimental pK_a for this peptide has the largest uncertainty of all of the peptides examined. Additional computational studies may generate greater insights into the nature of these interactions.

Phosphorylation plays a central role in cellular regulation [47]. With respect to histone H3 serine 10, phosphorylation is associated with cell cycle progression and also increases in response to environmental stimuli [48,49]. Symmetric dimethylation of arginine-8 at histone H3 (H3R8(SDMA)) represses expression of tumor suppressor genes [11,50] and is implicated in germ cell development in zebrafish [51]. We show

that the addition of a charged phosphate group onto a serine residue will significantly change interactions with neighboring amino acid residues and cause a change in the conformational ensemble of the dynamic histone tail (Fig. 4). By limiting its exposure and creating new contacts with neighboring residues, it will inhibit deposition of neighboring methyl groups, not through steric hindrance, but by forming intra-peptide interactions that most likely result in masking enzymatic recognition. We thus conclude that deposition of a methyl group in proximity of a phosphoserine is physically possible, but that the substrate landscape may be altered such that it is unrecognizable by the methyltransferase enzyme. Further work should be done to highlight the interplay between arginine methylation and serine phosphorylation and investigate whether the active site of PRMT5 and PRMTs in general, can accommodate an adjacent phosphate group so that they may recognize and methylate arginine substrates in the presence of phosphoserine.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.abb.2020.108716>.

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