- Quantitative investigation of gas flow, powder-gas interaction, and powder behavior under
 different ambient pressure levels in laser powder bed fusion
- 3

4 Abstract

5 The powder motion in laser powder bed fusion (LPBF) processes cause defect and variability 6 issues in the built products. It has been reported that the ambient pressure has a significant 7 influence on the powder motion, but the physical effects of the ambient pressure on the gas flow, powder-gas interaction, and powder behavior are not quantitatively understood. In this work, we 8 9 have developed the first three-dimensional multiphysics model for LPBF to simulate the molten 10 pool dynamics, depression zone evolution, gas flow structure, and powder motion in a fully 11 coupled manner. The model enables the first quantitative investigation of the gas flow, powder-12 gas interaction, and powder behavior in LPBF with different ambient pressure levels, all of which 13 are difficult to measure by experiments. The simulation results show a consistent gas flow structure 14 for all different pressure levels, but the gas flow parameters (temperature, velocity, Reynolds 15 number, and Knudsen number) vary significantly with the ambient pressure. Four powder-gas 16 interaction modes are defined by the gas flow around the particle and the gas-induced forces on 17 the particle, and the interaction modes, individually or collectively, control the motion of each 18 particle. With the changes in the ambient pressure and the gas flow parameters, the significance of 19 the four modes to the powder motion varies, and the powder behavior (temperature, force, velocity, 20 and ejection angle) becomes different. A new strategy is proposed to mitigate the powder motion 21 based on the modeling results.

23 Keywords

- 24 Laser powder bed fusion; Gas flow; Powder motion; Multiphysics modeling; High-speed X-ray
- 25 imaging.
- 26

27 Nomenclature

- ρ : Density
- \vec{V} : Velocity
- t: Time
- *p*: Pressure
- τ : Viscous stress
- μ : Viscosity
- \vec{g} : Gravity
- \vec{S}_D : Darcy's term
- e: Total energy
- k: Thermal conductivity
- T: Temperature
- Y: Mass fraction of metal vapor
- D: Diffusivity of metal vapor
- K_D : Resistance function in Darcy's term
- ϕ : Levelset function
- λ : Mean free path
- R_s : Specific gas constant
- V_{ct} : Tangential velocity in the condensed phase
- V_{gt} : Tangential velocity in the gaseous phase
- σ_v : Tangential momentum accommodation coefficient
- m_p : Particle mass
- \vec{V}_{pc} : Velocity at particle center
- \vec{F}_f : Fluid-induced force on particle
- \vec{F}_{col} : Collision force on particle

 I_p : Moment of inertia of particle $\vec{\omega}_p$: Angular velocity of particle α_k : Spring coefficient for particle collision α_d : Damping coefficient for particle collision E: Young's modulus ν : Poisson ratio ζ : Damping coefficient

28

29 1. Introduction

30 Laser powder bed fusion (LPBF) is a mainstream additive manufacturing technology in which 31 a laser selectively fuses metal powder particles layer-by-layer to print three-dimensional builds. 32 Although LPBF has shown promising capabilities of achieving complex and flexible geometries 33 to a designer's needs, there are defects and variability issues that inhibit further adaptation of this 34 emerging manufacturing method. One major cause of the defects and print variability is the gas-35 flow-induced powder motion in LPBF. It has been reported in the literature that laser heating 36 triggers strong evaporation on the metal surface. The vapor flux forms a vapor jet and entrains the surrounding gas, creating a complex gas flow, and eventually, the powder-gas interaction drives 37 38 the powder motion [1]. It is observed that the powder particles have a tendency of being entrained 39 towards and then ejected from the laser-illuminated zone (LIZ) [2, 3]. The entrainment dislocates the powder particles originally deposited around the LIZ [4]. The uniformity of the powder 40 41 distribution is disrupted, which leads to the variability of builds from print to print. More 42 importantly, the powder ejection gives rise to airborne spatters in solid or liquid form [5]. The 43 spattering leads to a series of deleterious effects, including degraded surface roughness [6], lack-44 of-fusion [7], increased oxidation [8], and inhomogeneity of the feedstock [9], all of which 45 contribute to poor mechanical properties in the final build [10].

46 Research efforts have been motivated to study the effects of process conditions on the powder 47 motion in LPBF, in the hope that the powder motion can be mitigated, if not fully controlled. The 48 investigated process conditions include varying the laser power and scanning speed [2, 11], the 49 ambient gas pressure [12-14] and composition [15, 16], as well as applying a cross-flow in the 50 build chamber [17]. Among these conditions, varying the ambient gas pressure is found to 51 influence the powder motion dramatically. It is reported in [13] that the numbers of the entrained 52 and ejected powder particles increase as the ambient pressure is decreased below the atmospheric 53 pressure (1 bar). Under extremely low ambient pressure (smaller than 1 mbar), few particles can 54 be entrained near the LIZ, and the particles surrounding the LIZ are mostly ejected. Similar 55 observations are reported in [12] where the authors correlate the ambient pressure with the vapor jet divergent angle and the powder motion. Apparently, extremely low ambient pressure leads to 56 57 a large amount of ejected powder particles and therefore should be avoided. The authors in [18] 58 explored the window of ambient pressure between 10 mbar to 1 bar and found that 200 mbar 59 achieves a similar amount but less variability of spattered particles. For the ambient pressure higher 60 than the atmospheric pressure, the only study found in the literature is [14]. This rareness might 61 be explained by the difficulty of building a high-pressure chamber. According to [14], the number 62 of entrained particles is decreased as the ambient pressure is increased above 1 bar, but the number 63 of ejected particles is increased. The authors suggested that high ambient pressure does not have 64 an advantage in reducing the powder motion.

There is a large parameter window (laser power, scanning speed, ambient pressure, etc.) from which one can identify the optimal parameter combinations to mitigate the powder motion. In that regard, the underlying physics of the powder-gas interaction needs to be fully understood to minimize the costly trial-and-error tests to identify the optimal conditions. The high-speed X-ray and optical imaging are capable of visualizing and tracking the powder motion with high spatial (up to 1 μ m) and temporal resolution (up to 1 μ s). However, quantitative information regarding the powder-gas interaction, i.e., the gas flow surrounding the powder particles and the forces exerted on the particle surfaces, is difficult to measure experimentally. In this regard, multiphysics simulations can be a complement to the imaging techniques to determine these quantities.

Multiphysics simulations have been extensively utilized to study the molten pool behavior in 74 75 LPBF. The simulated phenomena include the laser-metal interaction [7], molten pool flow [19], 76 pore formation [20], keyhole instability [21], and lack-of-fusion development [22, 23]. There are 77 relatively fewer efforts for gas flow simulations in LPBF. The vapor jet expansion and the 78 entrainment flow were simulated in [2] but the thermal-fluid coupling between the gas and the 79 molten pool was ignored. This coupling was added in a later study [24], but the simulation did not 80 track the powder motion. Powder particles are typically either ignored or assumed to be sintered 81 to the substrate. In a recent publication [25], the gas flow and powder motion were simulated, but 82 the gas flow was decoupled with the molten pool dynamics. The vapor jet was artificially defined, 83 and the effects of the jetting angle on the powder motion were investigated. In a previous 84 publication [26], we developed a two-dimensional (2D) model that fully coupled the molten pool 85 dynamics, gas flow, and powder motion. However, the phenomena in the third dimension were 86 ignored, which can compromise the fidelity of the simulation results.

In an attempt to further improve the fidelity of multiphysics simulations, we extend our previous 2D model to a three-dimensional (3D) model in this work. Particularly, the 3D model is used to investigate the effects of the ambient pressure on the gas flow, powder-gas interaction, and powder behavior. The simulation results are validated against high-speed X-ray imaging experiments. The model predictions of gas pressure, velocity, and temperature, as well as the 92 powder quantities such as force, velocity, and temperature, are identified as functions of the 93 ambient pressure, which are used to quantitatively understand the underlying physics under 94 different ambient pressure levels.

95 **2. Methods**

96 2.1 Multiphysics simulation

97 The calculation domain for the multiphysics simulation consists of the condensed phase region 98 and the gaseous phase region. The condensed region includes the substrate, molten pool, and the 99 individual powder particles. The gaseous region includes the metal vapor due to evaporation and 100 the ambient protection gas (typically argon). The interface between the condensed and gaseous 101 region is referred to as the C-G interface. The calculation domain is discretized into a Cartesian 102 mesh, and variables are computed at each cell of the mesh using computational fluid dynamics 103 (CFD) methods. Both the condensed and gaseous regions are subject to CFD computation. The C-104 G interface is captured by the level-set method. The individual particles are tracked by the discrete 105 element method (DEM). The CFD and DEM computations are fully-coupled and carefully 106 coordinated to model the powder-gas interaction. The modeling methodology of the current model 107 is consistent with our previous model, with the dimension extended from 2D to 3D. The methods 108 are briefly described here, and more details can be referred to [26-29].

109 The governing equations for the CFD computation are the conservation equations of mass,110 momentum, energy, and the vapor species:

$$\frac{\partial \rho}{\partial t} + \nabla \cdot \left(\rho \vec{V} \right) = 0 \tag{1}$$

$$\frac{\partial(\rho V)}{\partial t} + \nabla \cdot \left(\rho \vec{V} \vec{V}\right) = -\nabla p + \nabla \cdot \tau + \rho \vec{g} + \vec{S}_D$$
⁽²⁾

$$\frac{\partial(\rho e)}{\partial t} + \nabla \cdot \left(\rho e \vec{V}\right) = -\nabla \cdot \left(p \vec{V}\right) + \nabla \cdot \left(\tau \cdot \vec{V}\right) + \rho \vec{g} \cdot \vec{V} + \nabla \cdot \left(k \nabla T\right)$$
(3)

$$\frac{\partial(\rho Y)}{\partial t} + \nabla \cdot \left(\rho Y \vec{V}\right) = \nabla \cdot \left(\rho D \nabla Y\right) \tag{4}$$

Here, Eq. (1) is the mass conservation equation where ρ is the density, \vec{V} is the velocity, and t is 111 112 the time. Eq. (2) is the momentum conservation equation where p is the pressure, τ is the viscous stress tensor, \vec{g} is the gradational vector, and \vec{S}_D is the Darcy's term. Eq. (3) is the energy 113 114 conservation equation where *e* is the total energy (internal and kinetic energy), and *k* is the thermal 115 conductivity. Eq. (4) is the vapor species conservation equation where Y is the mass fraction of the 116 metal vapor in the gaseous region, and D is the diffusion coefficient of the vapor. It is assumed 117 that the condensed region is a continuum with only one species (metal), and the gaseous region 118 has two species, metal vapor and argon gas. Eq. (4) is only solved in the gaseous region. The 119 Darcy's term is written as:

$$S_D = -K_D(T)(\vec{V} - \vec{V}_s) \tag{5}$$

In Eq. (5), $\vec{V_s}$ is a solid velocity. It is assigned to be the particle velocity if the cell is inside a powder particle, and is assigned to be zero otherwise. K_D is a function of temperature T. It is a large number if T is below the solidus temperature, is equal to zero if T is above the liquidus temperature, and is smoothed connected if T is in between. With Eqs. (1)-(4), the pressure, velocity, temperature, and vapor mass fraction can be obtained at each cell center by the CFD computation.

125 The evolution of the C-G interface is governed by the level-set advection equation:

$$\frac{\partial \phi}{\partial t} + \vec{V} \cdot \nabla \phi = 0 \tag{6}$$

In Eq. (6), ϕ is the level-set function. ϕ is defined to be the signed distance to the interface, and the zero-level-set isosurface represents the C-G interface. Once the C-G interface is identified, the ghost fluid method is applied to separate the calculation domain into the condensed and gaseous region. The boundary conditions are carefully defined at the C-G interface to capture the interface
phenomena, including the surface tension, thermal-capillary (Marangoni) effect, recoil pressure,
and evaporation vapor flux.

Different from our previous model, a slip boundary condition is implemented in the current model to adapt for the larger Knudsen number in the gas that may occur in the low ambient pressure condition. The Knudsen number is defined as the ratio of the mean free path λ to a physical length scale *L*: $Kn = \lambda/L$. Here, *L* can be a typical powder diameter in LPBF. The mean free path λ is calculated as:

$$\lambda = \frac{\mu}{\rho} \sqrt{\frac{\pi}{2R_s T}} \tag{7}$$

137 In Eq. (7), μ is the gas viscosity, and R_s is the specific gas constant. The Knudsen number can be 138 used as the criterion to classify the "flow regimes" as follows [30, 31]:

• $0 \le Kn \le 0.01$: continuum flow (Navier-Stokes equation with no-slip boundary 140 condition).

• $0.01 \le Kn \le 0.1$: slip flow (Navier-Stokes equation with slip boundary condition).

• $Kn \gtrsim 0.2$: Navier-Stokes equation should be replaced with the Boltzmann equation (statistical mechanics).

As will be presented later, the Knudsen number in the gaseous region is typically > 0.01 when the ambient pressure is lower than 1 bar. Under this Knudsen number, the flow can enter the "slip flow" regime, and the slip boundary condition needs to be applied. Under extremely low pressure, the Knudsen number is > 0.2. This flow regime is beyond the capability of the current "continuum flow" model described by Eqs. (1)-(4). Therefore, only ambient pressure larger than 10 mbar is simulated (discussed later). The slip boundary condition can be written as:

$$V_{ct} - V_{gt} = \frac{2 - \sigma_v}{\sigma_v} \lambda \frac{\partial V_{gt}}{\partial n}$$
(8)

In Eq. (8), V_{ct} and V_{gt} are the tangential velocity on the condensed and gaseous side of the C-G interface, and their difference is the velocity slip. σ_v is the tangential momentum accommodation coefficient. $(\partial V_{gt}/\partial n)$ is the gradient of V_{gt} along the normal direction *n* of the C-G interface.

In the DEM computation, the initial powder bed configuration is randomly generated by a rain-dropping algorithm [32]. Then, the motion of each powder particle is tracked according to the Newton's equations of motion:

$$m_p \frac{d\vec{V}_{pc}}{dt} = \oint_{S_p} \vec{F}_f dS + m_p \vec{g} + \vec{F}_{col}$$
⁽⁹⁾

$$I_p \frac{d\vec{\omega}_p}{dt} = \oint_{S_p} \vec{r} \times \vec{F}_f dS \tag{10}$$

In Eqs. (9) and (10), m_p is the mass of the particle, \vec{V}_{pc} is the velocity at the particle center, I_p is the momentum of inertia of the particle, and $\vec{\omega}_p$ is the angular velocity of the particle. The integrations on the right-hand-side of Eqs. (9) and (10) are carried out on the particle surface S_p . \vec{F}_f is the force exerted by the surrounding flow on S_p , which can be obtained from the CFD computation. \vec{r} is a vector pointing from the particle center to the particle surface. \vec{F}_{col} is the force resulted from the collision with surrounding particles. The force exerted on the *i*th particle by its collision with the *j*th particle can be calculated as:

$$F_{col,ij} = -(\alpha_{k,ij}\delta_{ij} + \alpha_{d,ij}V_{n,ij})n_{ij}$$
⁽¹¹⁾

In Eq. (11), n_{ij} is a unit vector pointing from the i^{th} particle to the j^{th} particle. δ_{ij} is the gap between the two particles calculated by $\delta_{ij} = R_i + R_j - d_{ij}$, where R_i and R_j are the radii of the i^{th} and j^{th} particles, and d_{ij} is the distance between the two particles. The collision force in Eq. (11) is only activated when the gap δ_{ij} is negative, i.e., there is an overlap between the two particles. Otherwise, $F_{col,ij}$ is set to be zero. $V_{n,ij}$ is the normal relative velocity of the i^{th} particle with respect to the j^{th} particle, calculated as $V_{n,ij} = (V_i - V_j) \cdot n_{ij}$, where V_i and V_j are the velocities at the center of the i^{th} and the j^{th} particles. $\alpha_{k,ij}$ and $\alpha_{d,ij}$ are the spring and damping coefficient for the collision, calculated by:

$$\alpha_{k,ij} = \left(\frac{8}{15}\right)^{0.8} \left[\left(\frac{R_i R_j}{R_i + R_j}\right)^2 \frac{m_i m_j}{m_i + m_j} \left(\frac{E}{1 - \nu^2}\right)^4 \nu^2 \right]^{0.2}$$
(12)

$$\alpha_{d,ij} = 2m_i \zeta \sqrt{\frac{\alpha_{k,ij}}{m_i}}$$
(13)

171 In Eqs. (12) and (13), m_i and m_j are the mass of the i^{th} and the j^{th} particle. *E* is the Young's 172 modulus of the metal, v is the Poisson ratio, and ζ is the damping ratio.

173 In all the simulations and experiments of this work, the powder and substrate material is 174 stainless steel 316L, and the ambient gas material is argon. Their thermophysical properties used 175 in the CFD and DEM computation are listed in Table 1. The calculation domain and mesh of the 176 simulation are shown in Fig. 1. A fined mesh region (Fig. 1(a)) is defined to capture the major physical phenomena, e.g., the molten pool dynamics and the powder-gas interaction. Coarse mesh 177 178 is extended from the fine mesh region along all $\pm X$, $\pm Y$, and $\pm Z$ directions, as shown in Fig. 1(b). 179 The purpose of the coarse mesh is to allow the gas flow to be fully developed within the calculation 180 domain.

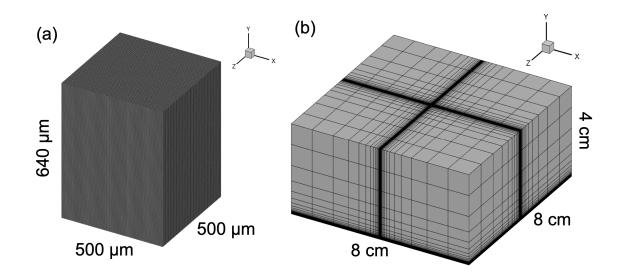


Fig. 1. Simulation domain and mesh. (a) Fine mesh region. (b) Entire calculation domain
 showing the coarse mesh extended from the fine mesh.

186 2.2 High-speed X-ray imaging

187 High-speed synchrotron X-ray imaging experiment is performed for in-situ observation of the 188 powder motion during LPBF under various ambient pressure levels. The synchrotron X-ray beam 189 is provided by the 32-ID-B beamline of the Advanced Photon Source (APS), Argonne National 190 Laboratory (ANL). The detailed information about the setup for in-situ X-ray imaging of LPBF 191 has been documented in [12]. In brief, a miniature stainless steel 316L powder bed is assembled 192 with a metal plate (substrate) sandwiched by two glassy carbon plates, which are used to contain 193 the powder layer on top of the substrate. The powder layer is spread manually with $100 \pm 10 \,\mu m$ 194 thickness. The width of the powder bed (the distance between two glassy carbon plates) is 500 µm.

195 Table 1. Thermophysical properties of stainless steel 316 L and argon gas [33-35].

Property	Stainless Steel 316L			Argon
	Solid	Liquid	Vapor	
Density (kg m ⁻³)	T dependent	T dependent	Ideal gas	Ideal gas
Phase change temperature (K)	1675 (solidus)	1708 (liquidus)	3135 (boiling)	

Specific heat (J kg ⁻¹ K ⁻¹)	847	714	495	520
Phase change latent heat (J kg ⁻¹)		2.9×10^5 (melting)	7.45×10 ⁶ (vaporization)	
Thermal conductivity (W m ⁻¹ K ⁻¹)	T dependent	T dependent	T dependent	T dependent
Viscosity (kg m ⁻¹ s ⁻¹)		0.015	T dependent	T dependent
Surface tension (N m ⁻¹)		1.7 − 1.5×10 ⁻⁴ T		
Young's Modulus (MPa)	2.2×10 ³			
Poisson's ratio	0.3			
Damping ratio	0.5			
Tangential momentum accommodation coefficient				1.0

197 We note that a larger sample width will reduce the X-ray imaging visibility of the powder bed, 198 while a smaller sample width makes the experiment deviate too much from a real LPBF process. 199 The visibility of DZ is sacrificed in the current experiment, as a clear observation of DZ requires 200 a much smaller sample width. A laser beam, with 1070 ± 10 nm wavelength and $\sim 200 \ \mu m \ 1/e^2$ 201 beam diameter, is used to melt the powder bed from above. The laser output is 312 W in power 202 and 800 µs in duration in the present work. The laser beam is kept stationary to realize a "spot 203 welding" type of laser melting. The powder particles have a mean diameter of 40 µm. The 204 synchrotron X-ray beam penetrates and illuminates the laser-matter interaction region from the 205 horizontal direction. The transmitted X-ray is converted into visible light by a LuAG:Ce 206 scintillator, and then recorded by a high-speed visible-light camera (Photron FastCam SA-Z). The 207 powder bed is enclosed in a vacuum chamber, refilled with high purity argon gas to adjust the 208 inner chamber pressure.

210 **3. Result and Discussion**

211 Simulations and experiments are utilized simultaneously to investigate the effects of ambient 212 pressure. Different ambient pressure levels are used in the investigation, as summarized in Table 213 2. Four ambient pressure levels in the atmospheric and sub-atmospheric (sub-atm) conditions (1 214 bar, 400 mbar, 50 mbar, 10 mbar) are investigated in both simulations and experiments. Hyper-215 atmospheric (hyper-atm) conditions are not investigated with experiments due to the difficulties 216 of building a high-pressure chamber, but two hyper-atm pressure levels (3 bar, 5 bar) are 217 investigated using simulations. Extremely low pressure levels (< 10 mbar) are not investigated in 218 the current work because the Knudsen number is typically > 0.2, as discussed earlier. The current 219 work will focus on the pressure range 10 mbar - 5 bar. For each pressure level in Table 2, at least 220 three experiments and simulations are conducted with random initial configurations of the powder 221 bed to ensure the statistical reliability of the results.

222 Table 2. Ambient pressure levels investigated.

Pressure group	Pressure Level	Investigation approach
Low pressure	10 mbar, 50 mbar	Experiment & Simulation
Medium pressure	400 mbar, 1 bar	Experiment & Simulation
High pressure	3 bar, 5 bar	Simulation

223

224 3.1 Overview of Simulation and Experimental Results

Exemplary results from the 3D simulation are shown in Fig. 2. The corresponding animations are given in Supplementary Video 1. A time sequence is presented in Fig. 2(a-e) to demonstrate the dynamic process. A variety of quantitative information, not directly measurable from the experiment, can be extracted from the simulation. Fig. 2(a, b) shows the laser absorption distribution and temperature field on the C-G interface. A vertical slice is extracted to show the 230 gas flow in Fig. 2 (c-e), where the temperature, velocity, (gauge) pressure, and vapor mass fraction 231 field can be quantified. The gauge pressure is defined as the difference between absolute and ambient pressure, $p - p_{amb}$. The laser heating of the powder bed, shown by Fig. 2(a), results in the 232 233 powder melting, formation of the depression zone (DZ), and the powder motion (Fig. 2(b)). It can 234 be seen in Fig. 2(c, e) that the evaporation occurs at the molten pool surface, causing the formation 235 of the DZ and vapor jet. The vapor jet induces the entrainment of the surrounding gas, as shown 236 by the streamlines in Fig. 2(d). Both the vapor jet and the entrainment flow contribute to the 237 powder motion. These pieces of quantitative information will be examined thoroughly in later 238 discussions.

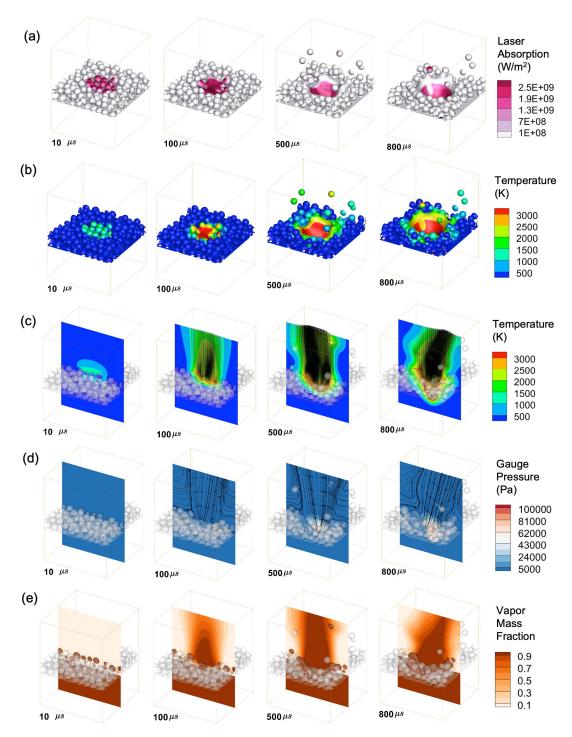


Fig. 2. Demonstration of the 3D model. A time sequence is selected from the simulation for p_{amb} = 400 mbar. The timestamp indicates the time period since the laser is turned on. (a) Laser absorption on the C-G interface. (b) The temperature on the C-G interface. (c) Vertical slice showing the temperature and flow field in the gaseous region. (d) Vertical slice showing the gauge pressure and streamlines of the flow in the gaseous region. (e) Vertical slice showing the vapor mass fraction field in the gaseous region.

247	To validate the current model, we compare between experiment and simulation results at the
248	pressure levels (1 bar, 50 mbar, 10 mbar) in Fig. 3(a-c). The animations for Fig. 3(a), Fig. 3(b),
249	and Fig. 3(c) are given as Supplementary Video 2, 3, and 4, respectively. As the ambient pressure
250	decreases, there are more spattered powder particles and the diverging angle of ejection increases.
251	These observations agree with the results in [12]. At each pressure level, we compared the velocity
252	of the ejected particles between experiment and simulation results, as shown in Fig. 4. Here, we
253	extracted the maximum velocity during the lifespan of each ejected particle and present the average
254	and extreme values over all the ejected particles with the error bars in Fig. 4. The magnitude of the
255	particle velocity matches reasonably well between simulation and experiment.
256	It is noticed that the ejected particles can be in a liquid form. These liquid particles can deform
257	and attach on other particles to form clusters, as seen in the X-ray imaging results in Fig. 3.
258	Unfortunately, the current model cannot capture the deformation and clustering of the liquid
259	particles. In the current model, the temperature distribution in the particle is calculated based on
260	the energy conservation (Eq. (3)). However, the fluid flow inside the particle is ignored, and the
261	particle is assumed to be a rigid body. Therefore, the particles in the current model cannot deform
262	or cluster with other particles (even if the temperature inside the particle is above the liquidus
263	temperature).
264	

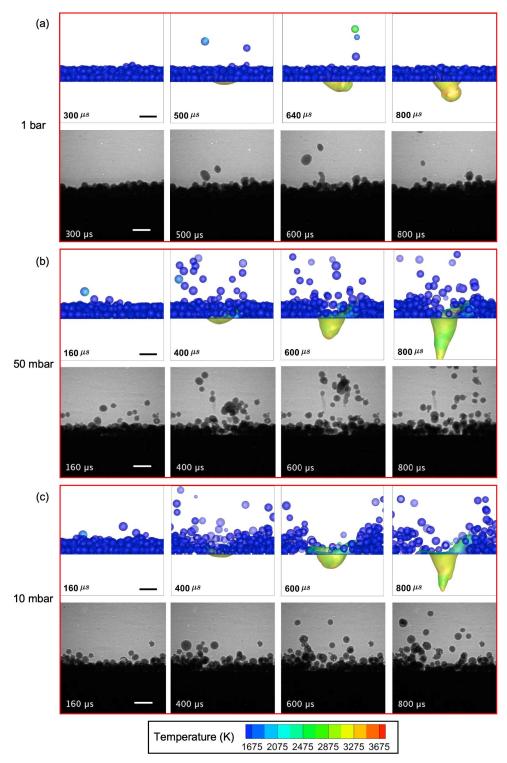


Fig. 3. Powder motion from the simulation and X-ray imaging. (a) $p_{amb} = 1$ bar. (b) $p_{amb} = 50$ mbar. (c) $p_{amb} = 10$ mbar. The timestamp indicates the time period since the laser is turned on. All the scale bars are 100 µm. The simulation results are shown from a projected view similar to that in X-ray imaging experiments. The temperature is shown on the C-G interface for the simulation.

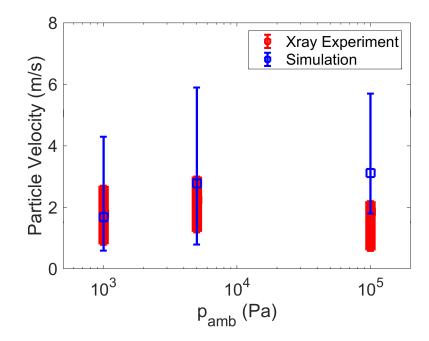


Fig. 4. Comparison of velocity for the ejected particles between X-ray experiment and simulation.

In the following discussion, the gas flow as a function of ambient pressure is first analyzed. After that, the powder motion driven by the powder-gas interaction are analyzed under different ambient pressure levels. Finally, we track the powder particles in all the simulations and statistically quantify the powder behavior, such as the velocity, forces, and temperature of the particles.

281

282 3.2 Vapor Jet and Induced Entrainment Flow

Significant gas flow occurs when the vapor jet initiates from the molten pool surface. The vapor jet then induces the entrainment of the ambient gas. The formation and development of the vapor jet at different ambient pressure levels are examined in Fig. 5. (5 bar, 1 bar, 10 mbar) are selected as the representative levels for the high-, medium-, and low-pressure groups in Table 2. In Fig. 5, a vertical slice with the temperature and flow field is extracted (as in Fig. 2(c)) at four moments to characterize the key physical events.

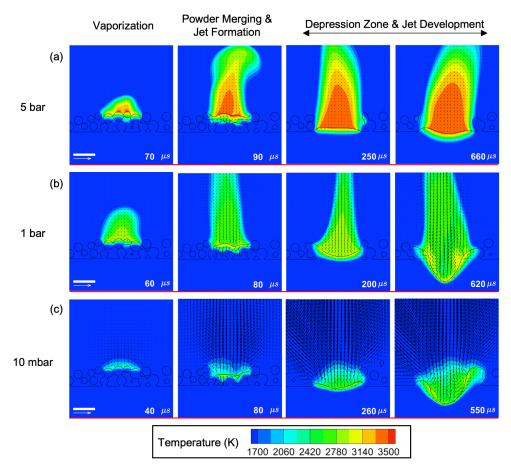


Fig. 5. Vapor jet formation and development under different ambient pressure levels. (a) p_{amb} = 5 bar. (b) p_{amb} = 1 bar. (c) p_{amb} = 10 mbar. For (a-c), a time sequence is given to characterize the key physical events. The vertical slice is at the same location shown in Fig. 2(c, d). All scale bars are 100 µm. All arrow keys are 2000 m/s in the gaseous region and 8 m/s for the condensed region.

It is observed in Fig. 5(a-c) that the evaporation initiates about $40 - 70 \ \mu s$ after the laser is turned on, which agrees with the observation in [12]. With the decrease of ambient pressure, the evaporation starts earlier because the boiling temperature is lower under lower pressure. According to the Clausius-Clapeyron equation, the boiling temperature is about 2350 K, 3150 K, and 3550 K under ambient pressure levels 10 mbar, 1 bar, and 5 bar, respectively. At about $80 - 90 \ \mu s$, the powder particles merge into a molten pool, and a vapor jet emerges from the molten pool surface. After that, the vapor jet develops as the recoil pressure on the molten pool surface creates the DZ. At the last moment in Fig. 5(a-c), the vapor jet has reached a "stable" state, where the temperatureand flow field do not significantly change with time.

Fig. 5 shows that the temperature in the vapor jet is significantly decreased as the ambient pressure decreases. The stable vapor jet temperature at the opening of DZ is about 3400 K, 2700 K, and 2100 K, for $p_{amb} = 5$ bar, 1 bar, and 10 mbar, respectively. This can be again explained by the lower boiling temperature under lower pressure. For the flow field, it can be observed that the vapor jet is confined and has a small velocity under high ambient pressure (Fig. 5(a)). As the ambient pressure decreases, the vapor jet expands more freely and has an increasingly larger velocity (Fig. 5(b, c)).

311 To further analyze the structure of the vapor jet and induced entrainment flow, we extract 312 more quantitative information at the last moment in Fig. 5(a-c) when the vapor jet has reached the 313 stable state, as detailed in Fig. 6. The velocity magnitude in a vertical slice is shown in Fig. 6(a), 314 which is at the same slicing location as in Fig. 5. The pressure field and streamlines for the vertical 315 slice are shown in Fig. 6(b). To characterize the three-dimensional flow field, we extract a 316 horizontal slice cutting through the powder bed. The projected (horizontal) velocity magnitude is 317 shown in Fig. 6(c). The pressure field and streamlines for the horizontal slice is shown in Fig. 6(d). 318 The dashed circle in Fig. 6(c, d) marks the LIZ. We will use Fig. 6(a, b) to characterize the vapor 319 jet expansion and Fig. 6(c, d) to characterize the induced entrainment flow.

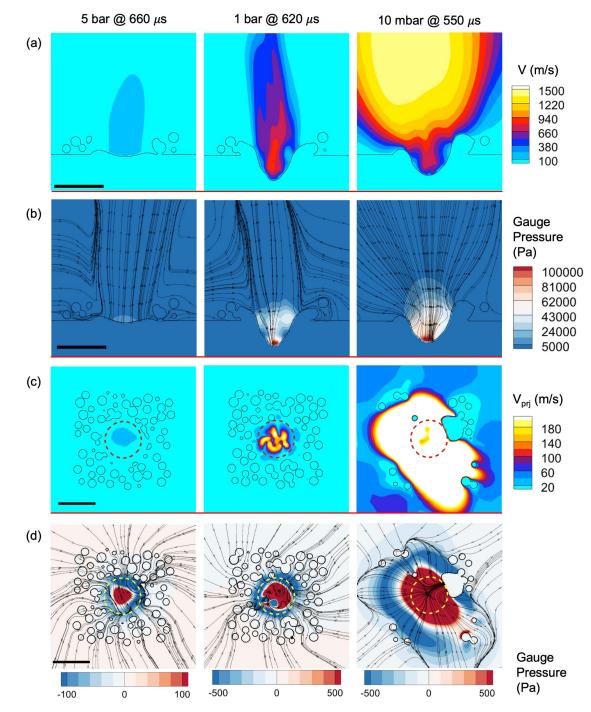


Fig. 6. Gas flow structure at the stable state is analyzed for $p_{amb} = 5$ bar, 1 bar, and 10 mbar. (a) Velocity magnitude at a vertical slice. The slicing location is the same as that shown in Fig. 2(c, d) and Fig. 5. (b) Gauge pressure and streamlines at the same vertical slice. (c) Projected (horizontal) velocity magnitude at a horizontal slice cutting through the powder bed. (d) Gauge pressure and streamlines at the same horizontal slice. The red and yellow dashed circle in (c, d) indicates the LIZ. All the scale bars are 200 µm.

328 For all the ambient pressure levels, the vapor jet expansion is driven by the pressure decrease 329 from the DZ bottom towards the DZ opening, as shown in Fig. 6(b). The pressure decrease 330 corresponds to an increase of velocity magnitude from the DZ bottom towards the DZ opening 331 (Fig. 6(a)). The velocity reaches its maximum at a location above the DZ where the pressure does 332 not notably change anymore along the streamlines. Then, the velocity magnitude decreases radially 333 towards the ambient as the vapor jet momentum dissipates into its surrounding. We denote the 334 pressure difference between the DZ bottom and DZ opening in Fig. 6(b) as Δp_{exp} , which can be viewed as the driving force for the vapor jet expansion. Δp_{exp} is relatively small (~ 10⁴ Pa) under 335 336 high ambient pressure (5 bar), which corresponds to a weak vapor jet expansion. Under 1 bar, Δp_{exp} is increased to ~ 10⁵ Pa. As the ambient pressure further decreases to 10³ Pa, Δp_{exp} does not 337 338 significantly change, but the vapor jet expands with a larger diverging angle due to the less 339 confinement imposed by the small ambient pressure. The Bernoulli equation for the compressible flow can be written as $\frac{1}{2}\rho v^2 + \frac{\gamma}{\gamma-1}p = \text{constant}$, where ρ is the density, v is the streamline-wise 340 341 velocity, γ is the specific heat ratio, and p is the pressure. As the pressure decreases from the DZ bottom to the DZ opening (by Δp_{exp}), the momentum ρv^2 increases according to the Bernoulli 342 equation. With the pressure decrease Δp_{exp} being similar (~ 10⁵ Pa) for the cases of $p_{amb} = 1$ bar 343 and $p_{amb} = 10$ mbar, the increase of momentum ρv^2 should also be comparable in these two cases. 344 Since ρ is significantly smaller for $p_{amb} = 10$ mbar, the velocity increase is significantly larger. To 345 346 summarize, the vapor jet velocity is expected to increase as the ambient pressure decreases, which 347 is confirmed in Fig. 6(a). The vapor jet expansion velocity, V_{exp} , can be characterized by the maximum velocity in the vapor jet. V_{exp} is plotted as a function of p_{amb} in Fig. 5(a) where V_{exp} is 348 about 200 m/s, 1000 m/s, and 1500 m/s for $p_{amb} = 5$ bar (5×10⁵ Pa), 1 bar (1×10⁵ Pa), and 10 mbar 349 350 $(1 \times 10^3 \text{ Pa})$, respectively.

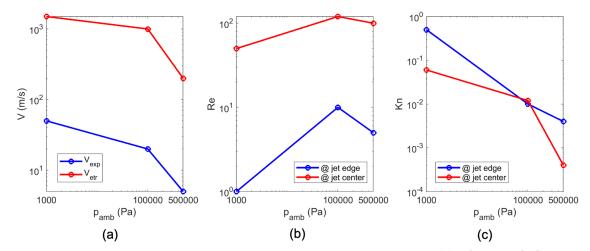


Fig. 7. Gas flow characteristics as a function of ambient pressure. (a) Characteristic vapor jet expansion velocity V_{exp} and induced entrainment velocity V_{etr} . (b) Reynolds number at the jet center and jet edge. (c) Knudsen number at the jet center and jet edge. All the quantities are extracted based on the vertical and horizontal slices shown in Fig. 6.

357 The induced entrainment flow is visualized by the streamlines in Fig. 6(d). For all ambient 358 pressure levels, a high-pressure region is formed in the LIZ (dashed circle), where the vapor jet is 359 initiated. Right outside the high-pressure region, there is a "ring" of low-pressure region beyond 360 which the pressure is close to the ambient pressure. The low-pressure ring acts as a separation 361 between the entrained gas and the expanding vapor. Outside the "ring," the gas is driven towards 362 the LIZ by the pressure difference between the "ring" and ambient. Inside the ring, the gas radially 363 expands from the LIZ outwards. We denote the pressure difference between the ambient and the 364 low-pressure ring as Δp_{etr} , which can be viewed as the driving force for the entrainment flow. Δp_{etr} 365 is relatively small (~ 100 Pa) under high ambient pressure (5 bar), indicating weak entrainment. 366 At 1 bar, Δp_{etr} is increased to ~ 500 Pa. Δp_{etr} does not significantly change as the ambient pressure 367 further decreases to 10 mbar, but the size of the low-pressure ring is significantly increased due to the enhanced vapor jet expansion. Again, according to the Bernoulli equation, Δp_{etr} can cause 368 369 larger entrainment velocity under low ambient pressure. As confirmed in Fig. 6(c), the entrainment 370 velocity is increased as the ambient pressure decreases. The entrainment velocity, V_{etr} , can be

characterized by the projected velocity at the outer edge of the low-pressure ring. V_{etr} is plotted as a function of p_{amb} in Fig. 5(a) where V_{etr} is ~ 5 m/s, 20 m/s, and 50 m/s for $p_{amb} = 5$ bar (5×10⁵ Pa),

373 1 bar $(1 \times 10^5 \text{ Pa})$, and 10 mbar $(1 \times 10^3 \text{ Pa})$, respectively.

374 Finally, we quantify the Reynolds number (Re) and Knudsen number (Kn) for the gas flow as 375 functions of ambient pressure, as shown in Fig. 7(b, c). The Reynolds number is defined as Re =376 $\rho UL/\mu$, where ρ is the gas density, U is the velocity magnitude, μ is the gas viscosity, and L is a 377 physical length scale chosen to be 50 µm as the typical diameter of a large powder particle. The Knudsen number is again defined as $Kn = \lambda/L$. The Reynolds number is considered as the ratio 378 379 between the inertia and viscous force and has been used to estimate the drag force exerted by the 380 gas flow on powder particles [1, 25]. The Knudsen number, as discussed earlier, can be used to 381 judge whether statistical mechanics or (continuum) fluid dynamics governs the flow behavior. We 382 select two locations in the horizontal slice in Fig. 6(c, d) to characterize Re and Kn of the gas flow. 383 The first location is the center of the LIZ, which is considered the typical location where the 384 powder particles are ejected. The second location is a typical location on the outer edge of the low-385 pressure ring, where powder particles are subjected to the entrainment flow. These two locations 386 are denoted as "jet center" and "jet edge," and their (Re, Kn) values are calculated from simulations 387 and shown in Fig. 7(b, c).

It is observed in Fig. 7(b) that the Reynolds number does not vary notably for the high and medium pressure levels (5 bar and 1 bar). Re is ~ 100 at the jet center and ~ 10 at the jet edge. Under low ambient pressure (10 mbar), Re decreases to ~ 50 at the jet center and ~ 1 at the jet edge. The decrease of Re can be attributed to the small gas density due to the low ambient pressure. It is also noted that the Reynolds number is typically 1 - 2 magnitudes smaller at the jet edge than the jet center. The drag force on powder particles can be estimated using the "Stokes limit" assumption [1], which assumes that the Reynolds number approaches zero. We point out that the Stokes limit assumption can be justified for the entrainment flow around the powder particles as Re \leq 10. However, the Reynolds number is ~ 100 for the surrounding gas flow when powder particles are ejected. In such a case, the Stokes limit assumption should be applied with caution.

398 It can be observed in Fig. 7(c) that the Knudsen number increases remarkably as the ambient 399 pressure decreases. It is safe to assume that a continuum gas flow for high ambient pressure (5 400 bar), as Kn is 0.0004 - 0.004. The Knudsen number for the medium pressure (1 bar) is about 0.01, 401 indicating the necessity of employing the slip boundary conditions even at the atmospheric 402 pressure. The no-slip boundary condition should be carefully employed in gas flow simulations 403 for LPBF, and it should be replaced by the slip boundary condition when Kn > 0.01. Under low 404 ambient pressure (10 mbar), Kn is 0.05 - 0.5, which suggests that the continuum flow assumption 405 will be invalid should the ambient pressure be further decreased. We stress that caution needs to 406 be exercised when carrying out continuum fluid dynamics simulations under extremely low 407 ambient pressure levels for LPBF.

408

409 3.3 Powder Motion Driven by Powder-Gas Interaction

The vapor jet and induced entrainment flow determine the powder-gas interaction, which drives the powder motion. The typical powder particle trajectories at 1 bar are shown in Fig. 8. An isometric view is given in Fig. 8(a), where five typical particle trajectories are shown in lines with circle markers (the triangle markers indicate the initial location). The red cylinder indicates the laser, and the box in light blue indicates the powder bed. The Y direction is the vertical direction, and (X, Z) are the two horizontal directions. Fig. 8(b-d) shows three projection views from the X-Z, X-Y, and Z-Y plane.

417	These results confirmed the four characteristic models of powder-gas interaction that we
418	identified with our previous 2D model. The four modes are briefly introduced below, with more
419	details given in [26]:
420	• Recoil mode: It is defined when significant evaporation occurs on the surface of the powder
421	particle, and the recoil pressure dominates over the drag force (gas-flow-induced force) on the
422	particle surface. The powder particle is driven toward the opposite direction of the vapor jet by
423	the recoil pressure.
424	• Entrainment mode: It is defined when the surroundings of the particle are dominated by the
425	entrainment flow, and the particle is driven by the drag force towards the LIZ.
426	• Elevation mode: It is defined when the particle is simultaneously subject to the entrainment
427	flow and the vapor jet expansion, and the particle is vertically ejected by the drag force with a
428	relatively small divergence angle.
429	• Expulsion mode: It is defined when the surroundings of the particle are dominated by the
430	expanding vapor jet, and the particle is ejected by the drag force with a relatively large
431	divergence angle.
432	The powder motion can be governed by one powder-gas interaction mode or a sequence of modes.
433	Specifically, for the simulation result in Fig. 8, Particles 1 and 5 are sequentially governed by the
434	entrainment, elevation, and expulsion mode. They are entrained towards and subsequently ejected
435	from the LIZ. Particle 2 is dominated by the elevation mode. It moves vertically upwards with
436	neither entrainment nor ejection. Particle 3 is dominated by the expulsion mode. It is ejected by a
437	tilted vapor jet with a large divergence angle. Particle 4 sequentially experiences the entrainment,
438	elevation, and recoil mode. It is first entrained into the LIZ, and then evaporation occurs on its top

surface due to laser heating. Finally, the recoil mode drives the particle to dive into the moltenpool.

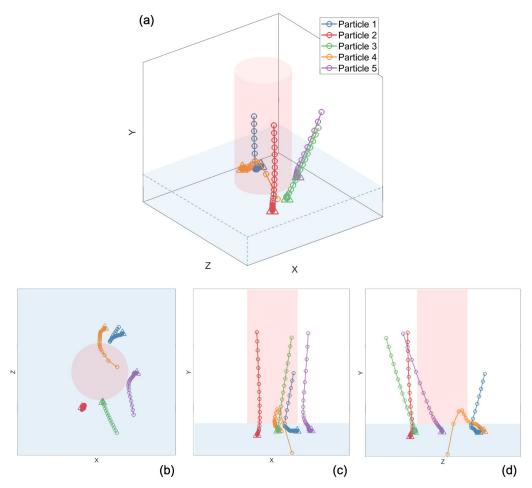


Fig. 8. Typical particle trajectories at 1 bar (atmospheric pressure). (a) Isometric view where the particle trajectories are lines with circle markers. The triangle marker indicates the initial locations of the particles. The red cylinder indicates the laser. The box in light blue indicates the powder bed. The Y direction is the horizontal direction, and (X, Z) are the two vertical directions. (b-d) Projected views of (a) from X-Z, X-Y, and Y-Z plane.

A new powder ejection mechanism, denoted as the "recoil-mode" powder ejection, is identified in the current simulations. Two typical powder particles governed by the recoil mode ejection are shown as Particles 6 and 7 in Fig. 9. Again, an isometric view is given in Fig. 9(a), and three projection views from the X-Z, X-Y, and Y-Z plane are given in Fig. 9(b-d). Particles 6 and 7 are first entrained and then ejected at a large ejecting angle (for Particle 7), or even

453 downwards to the powder bed (for Particle 7). To quantify the transient situation for Particle 6, 454 vertical planes can be determined by the laser centerline and the center of Particle 6 at selected 455 moments, and the temperature and flow field on the planes are shown in Fig. 10. The entrainment-456 ejection particle motion can be observed in Fig. 10(a) where Particle 6 is circled in red. Although 457 the entrainment-ejection motion is similar to Particles 1 and 5 in Fig. 8, the driving force for the 458 ejection is different. The temperature and flow field surrounding Particle 6 is zoomed in at two 459 key moments (450 µs and 530 µs), as shown in Fig. 10(b, c). At 450 µs, the upper left corner of 460 Particle 6 cuts into the LIZ, and the laser heating causes significant local evaporation. As a result, 461 the recoil pressure pushes the particle away from the LIZ and downwards to the powder bed. At 462 530 µs, Particle 6 has departed from the LIZ but is heated at the lower left corner by the reflected 463 laser beam. The reflected beam can be traced back to the asymmetric DZ clearly seen in Fig. 10(a). 464 Therefore, Particle 6 is again subject to recoil pressure, and the latter subsequently ejects Particle 465 6 at a large ejecting angle. Particle 7 also cuts into the LIZ (Fig. 9(b)) when it is heated by the laser 466 on its top (similar to the scenario in Fig. 10(b)). But Particle 7 is not subsequently heated by the 467 reflected laser at its bottom position. Therefore, recoil pressure is only generated on the top of the 468 particle to push it downwards to the powder bed. It is noted that the recoil-mode powder ejection 469 is also inferred from experimental observations [1, 5].

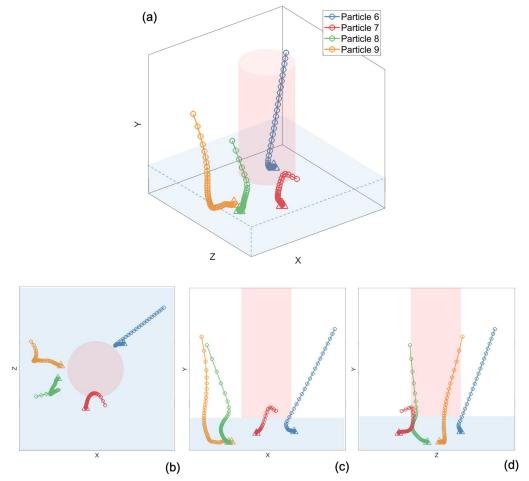


Fig. 9. Miscellaneous powder particle trajectories. Particle 6 is under $p_{amb} = 3$ bar. Particle 7 is under $p_{amb} = 5$ bar. Particles 8 and 9 are under $p_{amb} = 50$ mbar. (a) Isometric view where the powder trajectories are lines with circle markers. The triangle marker indicates the initial locations of powder particles. The red cylinder indicates the laser. The box in light blue indicates the powder bed. The Y direction is the horizontal direction, and (X, Z) are the two vertical directions. (b-d) Projected views of (a) from X-Z, X-Y, and Y-Z plane.

The recoil-mode powder ejection is found to be active only in the cases with medium- and high-pressure groups in Table 2. In the cases of medium pressure, the contribution of the recoil mode to the powder ejection is minor, and the powder ejection is primarily driven by the expulsion mode, where the expanding vapor jet exerts drag force on and ejects the powder particles. In the high-pressure conditions, the drag force becomes significantly smaller due to the reduced vapor jet velocity (Fig. 6). The expulsion mode weakens, and the recoil mode becomes the dominating mechanism for the powder ejection. As a result, the recoil-mode ejection is mostly distinguishable in the high-pressure simulations. As for the low-pressure cases, no recoil-mode ejection can be
identified because the powder particles are typically ejected by the enhanced vapor jet expansion
before they can be significantly heated by the laser.

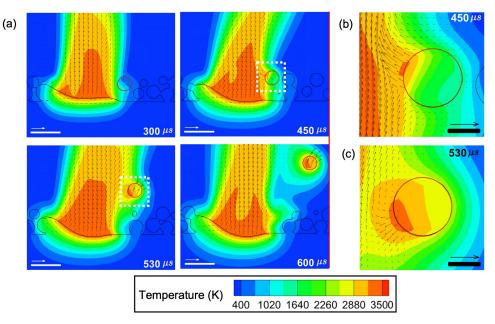


Fig. 10. Recoil-mode ejection under $p_{amb} = 3$ bar. (a) Time sequence from a vertical slice formed by the laser centerline and the center of Particle 6 (red-circled). The slice shows the temperature and flow field. The scale bars are 100 µm. The arrow keys are 500 m/s for the gaseous region and 2 m/s for the condensed region. (b, c) The temperature and flow field surrounding Particle 6 is zoomed in at 450 µs and 530 µs. The scale bars are 25 µm. The arrow keys are 250 m/s for the gaseous region. The flow field inside the particle is not shown.

495

496 Two new sequences of powder-gas interaction modes have also been identified that can cause 497 complicated powder motions, as demonstrated by Particles 8 and 9 in Fig. 9. Fig. 9(c) shows a 498 meandering trajectory for Particle 8: it is first ejected, then entrained, and finally ejected again. 499 Particle 8 can be considered sequentially governed by the expulsion mode, entrainment mode, and 500 again expulsion mode, which suggests a competition between the vapor jet expansion and the 501 entrainment flow surrounding Particle 8. To visualize the competition, we extract the horizontal 502 slices cutting through the center of Particle 8 at four moments, as shown in Fig. 11(a). As Particle 503 8 is elevating during this meandering motion, the slicing locations for the four moments in Fig.

504 11(a) are Y = 28.5 μ m, 56.8 μ m, 90.7 μ m, and 150.5 μ m, respectively, with Y = 0 being the 505 substrate surface location. At 230 µs in Fig. 11(a), Particle 8 (solid red circle) is ejected from the 506 LIZ (dashed red circle) by the expanding vapor jet, governed by the expulsion mode. At 420 - 580507 us, Particle 8 is away from the vapor jet expansion, and its surroundings are dominated by the 508 entrainment flow, as shown by the zoomed-in flow field in Fig. 11(b, c). Therefore, Particle 8 is 509 governed by the entrainment mode and is driven towards the LIZ. Finally, Particle 8 meets the 510 expanding vapor jet again at 680 µs. Following this, Particle 8 is again governed by the expulsion 511 mode and is ejected from the LIZ. It is noted that the ejection-entrainment motion of Particle 8 has 512 also been observed in the experiments [12].

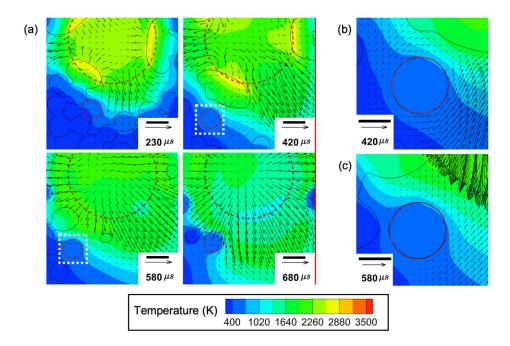


Fig. 11. Meandering powder motion under $p_{amb} = 5000$ mbar. (a) Time sequence from a horizontal slice cutting through the center of Particle 8 (solid circle in red). The slices show the temperature and flow field. The red-dashed circle indicates the laser diameter (LIZ). The scale bars are 50 µm. The arrow keys are 1000 m/s for the gaseous region and 4 m/s for the condensed region. (b, c) The temperature and flow field surrounding Particle 8 is zoomed in at 420 µs and 580 µs. The scale bars are 25 µm. The arrow keys are 50 m/s for the gaseous region. The flow field inside the particle is not shown.

522 The second new sequence of powder-gas interaction modes manifests in Particle 9 (Fig. 9), 523 which is sequentially governed by the ejection and elevation mode. There is a similar competition 524 between the vapor jet expansion and the entrainment flow surrounding Particle 9. As can be seen 525 in Fig. 9(c), Particle 9 is first ejected from the LIZ governed by the expulsion mode. As Particle 9 526 moves away from the LIZ, the vapor jet expansion weakens while the entrainment flow is 527 strengthened. As the effects of these two flows are balanced, Particle 9 is governed by the elevation 528 mode, which drives Particle 9 to move vertically upwards (Fig. 9(c, d)) with little horizontal 529 movement (Fig. 9(b)).

530 As exemplified by Particles 8 and 9, the particles can be governed by new sequences of 531 powder-gas interaction modes following their initial ejection from the LIZ, which leads to their 532 complex motions. These new sequences of powder-gas interaction modes are caused by the 533 balance (or imbalance) between the two driving forces stemming from the vapor jet expansion and 534 the entrainment flow. It is noted that these sequences of modes and the consequent powder motions 535 are almost exclusively observed in the low-pressure group (Table 2). This can be explained by the 536 wide low-pressure ring observed under low ambient pressure (Fig. 6(d)), which creates more 537 opportunities for the entrainment flow to balance or even overcome the vapor jet expansion, 538 resulting in the new mode sequences and the complex powder motions. In contrast, the low-539 pressure ring observed for higher ambient pressure levels is narrow (Fig. 6(d)). Therefore, once 540 the ejected powder particles pass the low-pressure ring, the weak entrainment flow is not adequate 541 to alter the particle trajectory.

543 *3.4 Statistical Quantification of Powder Behavior*

We have repeated the simulation four times at each ambient pressure level in Table 2 with random initial powder bed configurations. For each simulation, quantities such as velocity and average temperature can be tracked for each powder particle (examples are given in Appendix A). Statistical quantification is carried out to show the powder-related quantities as functions of ambient pressure, as shown in Fig. 12. The detailed definitions of the terms in Fig. 12 and the methods of extracting the values for these terms are given in Appendix B.

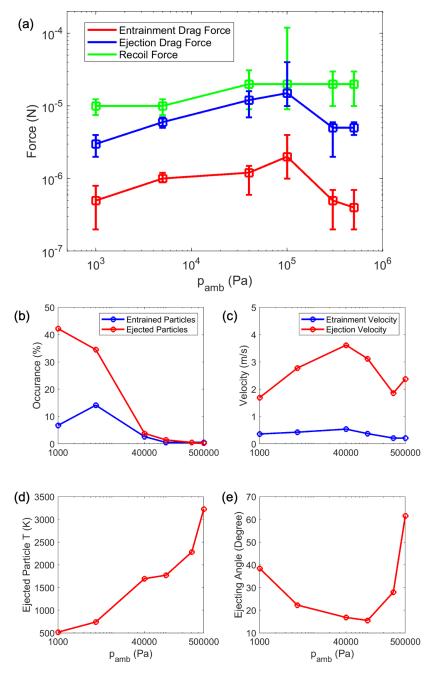


Fig. 12. Statistical quantification of powder behavior as a function of the ambient pressure.
The definitions of the terms in (a-e) and the method of extracting these pieces of information are
detailed in Appendix B. (a) Magnitudes of the entrainment drag, ejection drag, and recoil forces.
(b) Occurrences of the entrained and ejected particles. (c) Entrainment and ejection velocities. (d)
The temperature of the ejected particles. (e) The ejecting angle of the ejected particles.

557 In Fig. 12(a), the magnitudes of the drag and recoil forces (defined in Appendix B) on powder 558 particles are shown as functions of the ambient pressure. The drag forces (both for powder 559 entrainment and ejection) are the maximum at the medium-pressure levels. In the cases of the high-560 pressure levels, the drag forces are reduced because the gas flow velocity (both vapor jet expansion 561 and induced entrainment flow) is small, as shown in Fig. 6(a, c). In the cases of the low-pressure 562 levels, although the gas flow velocity is increased, the flow enters the "slip flow" regime, and the drag force decreases with the increasing Knudsen number [30]. The ejection drag force $(10^{-6} - 10^{-1})^{-6}$ 563 ⁵ N) is typically one magnitude larger than the entrainment drag force $(10^{-7} - 10^{-6} \text{ N})$ for all ambient 564 565 pressure levels. The recoil force, in most cases, only briefly occurs on the top surface of powder 566 particles when the vapor jet begins to form. The particles subject to the recoil force then rapidly merge into the molten pool. In this scenario, the magnitude of recoil force (10^{-5} N) does not 567 568 significantly change with the ambient pressure. At medium and high ambient pressure levels, however, a few particles can have a larger recoil force (up to $\sim 10^{-4}$ N) when they are ejected (e.g., 569 570 Particles 6 and 7 in Fig. 9) or dives into the molten pool (e.g., Particle 2 in Fig. 8) driven by the 571 recoil mode. This explains the larger error bar for medium and high ambient pressure levels in Fig. 572 12(a).

573 Next, the occurrences of entrained and ejected particles (defined in Appendix B) are counted 574 in Fig. 12(b) as functions of the ambient pressure. It is observed that the occurrence of ejected 575 particles increases with decreasing ambient pressure, which is due to the enhanced vapor jet 576 expansion. The occurrence of entrained particles first increases with decreasing ambient pressure but decreases at the lowest pressure level (10 mbar, i.e., 10^3 Pa). The increase of entrained particles 577 578 can be attributed to the wider low-pressure ring under lower ambient pressure (Fig. 6(d)). However, 579 the entrainment drag is reduced significantly at the lowest pressure level, as discussed in Fig. 12(a), 580 which explains the decrease of entrained particles.

The entrainment and ejection particle velocity (defined in Appendix B) are plotted as functions of the ambient pressure in Fig. 12(c). The particle velocities are the maximum at the medium pressure levels and decrease for lower or higher pressure levels, which is consistent with the trend for the entrainment and ejection drags in Fig. 12(a). For high ambient pressure levels, the ejection velocity is typically smaller due to the smaller ejection drag. However, when the recoilmode ejection occurs, the ejection velocity can be large despite the smaller drag force. Therefore, there is an increase in ejection velocity for the highest pressure level (5 bar, or 5×10^5 Pa).

It is noticed in Fig. 12(c) that the entrainment velocity (~ 0.5 m/s) is typically smaller than the ejection velocity (~ 3 m/s). The velocity magnitude can only be qualitatively compared with Xray imaging results, as the particle velocity observed in X-ray imaging is only the projected velocity. Typical particle trajectories are manually extracted from the X-ray imaging to calculated the projected particle velocity (examples given in Appendix A). The measured maximum velocity in a particle's lifespan agrees with the range 0.5 - 3 m/s in Fig. 12(c).

Finally, the temperature and ejecting angle (defined in Appendix B) of the ejected particles are quantified as functions of the ambient pressure, as shown in Fig. 12(d, e). As can be seen in Fig. 12(d), the temperature of the ejected particles increases from 500 K to almost 3500 K as the ambient pressure increases from 10 mbar (1×10^3 Pa) to 5 bar (5×10^5 Pa). This trend agrees with the observation in [14] and can be explained as follows. Under high ambient pressure, the boiling temperature of the metal is increased, and therefore, the powder particles can be further heated before being ejected by the vapor jet.

In Fig. 12(e), it is observed that the ejecting angle is increased from ~ 15 degrees to ~ 40 degrees as the ambient pressure decreases from 1 bar $(1 \times 10^5 \text{ Pa})$ to 10 mbar $(1 \times 10^3 \text{ Pa})$. This is because the vapor jet expansion is subject to less confinement at lower pressure levels (Fig. 5(c)). As a result, the vapor jet expands at a larger diverging angle and causes a larger ejecting angle for the ejected particles. This trend also agrees with the findings in [12]. As the ambient pressure increases from 1 bar $(1 \times 10^5 \text{ Pa})$ to 5 bar $(5 \times 10^5 \text{ Pa})$, the ejecting angle dramatically increases due to the increasingly frequent occurrence of recoil-mode ejection. The recoil-mode ejection can cause a large ejecting angle, even larger than 90 degrees, when the particle is ejected downwards to the powder bed (e.g., Particle 7 in Fig. 9).

610

611 **4. Discussion on Powder Motion Mitigation**

612 The simulation results show that a high ambient pressure can significantly suppress the gas 613 flow (Fig. 6), and therefore, reduce the drag forces (Fig. 12(a)) and the occurrences of powder 614 entrainment and ejection (Fig. 12(b)). This suggests that applying of a high ambient pressure can 615 be an effective way to mitigate powder motion. The effects of hyper-atmospheric conditions (3 bar 616 and 5 bar) have been experimentally investigated in [14] with a moving laser. The authors found 617 that the number of entrained particles decreases with the ambient pressure, which agrees with the 618 current simulation results. However, the authors also found that the number of ejected particles 619 increases with the ambient pressure, which is different from the current simulation results for a 620 stationary laser. This discrepancy can be explained as follows. In a stationary-laser process under 621 high ambient pressure, the particles illuminated by the laser beam center are melted and merged 622 into the molten pool, and the particles outside the laser illumination will not have significant 623 motion because the gas flow is suppressed under high ambient pressure. Only the particles 624 illuminated by the rim of the laser can potentially be ejected by the recoil-mode ejection (Fig. 10). 625 As the number of particles in the rim of the laser is low for a stationary laser, few particle ejection 626 events are possible. And if those particles are successfully ejected, no more particles will be

627 entrained to the rim of the laser and then be ejected, so the powder ejection will not continue. The 628 situation is very different with a moving laser. As the laser moves along a given contour, it 629 constantly catches new particles in its moving rim. All the particles on the laser scanning contour 630 will be caught by the rim of the laser at certain moments and likely be ejected by the recoil-mode 631 ejection. Therefore, the high ambient pressure may not reduce the powder ejection in a moving-632 laser case, and under the conditions in [14], the high ambient pressure even increases the powder 633 ejection. Moreover, the ejected particles under high ambient pressure are typically in a molten state 634 (Fig. 12(d)) and have a large divergence angle (Fig. 12(e)). They can stick with other particles to 635 form agglomerates, which can grow through a "snow-balling" mechanism [5], that the 636 agglomerates can be later ejected again and stick with more particles. This is confirmed in [14], 637 where the authors found more powder spatter with larger size and temperature under high ambient 638 pressure conditions.

639 Based on the above reasoning, we propose a new strategy that can potentially mitigate the 640 powder motion in a moving-laser process: applying a high ambient pressure along with a pulsed 641 laser beam. The pulsed laser beam moves in consecutive points and remains at each point for a 642 certain amount of time (exposure time) before moving to the next point. We expect two beneficial 643 effects from this strategy: (1) only the particles illuminated by the rim of the laser beam of each 644 pulse can be ejected by the recoil-mode ejection. Therefore, the number of potential ejected 645 particles is reduced compared to the continuous moving laser. (2) the point distance between two 646 consecutive points and the exposure time can be optimized to further reduce powder ejection. For 647 example, the point distance and exposure time can be adjusted so that the particles at the rim of 648 the previous pulse will fall in the center of the next pulse before they can be ejected.

649 It is noted that the use of pulsed laser beam in LPBF has been reported in the literature [36-650 39], but the laser pulsating is used to control the molten pool stability, the cooling rate, and the 651 microstructure. To the best of our knowledge, the effects of pulsed laser on the powder motion 652 have not been investigated. It is also emphasized that a pulsed laser can only effectively mitigate 653 powder motion in a high ambient pressure. Under medium and low ambient pressure levels, 654 surrounding particles will continuously move toward the rim of the laser through the entrainment 655 mode and then be ejected through the elevation or expulsion modes. In the future, we will perform 656 X-ray imaging experiments and numerical simulations to examine the effectiveness of this 657 proposed strategy.

658

659 **5.** Conclusion

660 In this work, a 3D multiphysics model is developed that fully couples the molten pool 661 dynamics, gas flow, and powder motion in LPBF. The model capabilities highlight in resolving 662 the gas flow surrounding the powder particles, as well as quantifying the forces exerted on the 663 particle surfaces. In a "spot-welding" setup, simulations are carried out to investigate the effects 664 of the ambient pressure (in the range of 10 mbar - 5 bar) on the gas flow, powder-gas interaction, 665 and powder behavior. The temperature, flow velocity, and pressure field in the gas under different 666 ambient pressure levels are first quantified. Then, characteristic powder motions are examined and 667 explained by the powder-gas interaction mode(s). Finally, the powder behavior (e.g., velocity, 668 temperature, and ejecting angle) is statistically analyzed as a function of the ambient pressure. The 669 model predictions of powder motion agree well with the high-speed X-ray imaging experiments. 670 A consistent gas flow pattern is observed in simulations for all ambient pressure levels

671 investigated. The vapor jet initiates from the DZ and expands into the ambient, while the ambient

672	gas is entrained towards the LIZ. The vapor jet expansion is driven by a pressure decrease from
673	the DZ bottom towards the DZ opening. The entrainment flow is driven by a pressure decrease
674	from the ambient towards the LIZ. As the ambient pressure decreases from 5 bar to 10 mbar, the
675	major effects on the gas flow are summarized as follows:
676	• The characteristic velocity of vapor jet expansion increases from 200 m/s to 1500 m/s. The
677	characteristic velocity of the entrainment flow increases from 5 m/s to 50 m/s.
678	• The divergent angle of the vapor jet expansion significantly increases.
679	• The characteristic temperature of the vapor jet decreases from 3400 K to 2100 K.
680	• The Knudsen number of the gas flow increases from the range $(0.004 - 0.004)$ to the range
681	(0.05 - 0.5). The gas flow enters the slip flow regime (Kn > 0.01) when the ambient
682	pressure is about 1 bar. It is expected that the continuum assumption of the gas flow fails
683	$(Kn \gtrsim 0.2)$ when the ambient pressure is below 10 mbar.
684	The four powder-gas interaction modes previously identified in the 2D simulations (i.e., recoil,
685	entrainment, elevation, and expulsion modes) are confirmed in the current 3D simulations. The
686	motion of a specific powder particle can be attributed to that the particle being subject to a single
687	powder gas interaction mode or a sequence of modes. More knowledge regarding the interaction

687 powder-gas interaction mode or a sequence of modes. More knowledge regarding the interaction 688 modes and the resultant powder motion have been revealed with the 3D simulation results. 689 Specifically, the recoil mode is found as another cause for the powder ejection besides the expulsion mode. Such recoil-mode ejection is mostly distinguished at ambient pressure levels 690 691 above 1 bar. New sequences of powder-gas interaction modes are also found at ambient pressure 692 levels below 400 mbar. The expulsion-elevation mode sequence causes the powder particle to 693 elevate after being ejected from the LIZ. The expulsion-entrainment-expulsion mode sequence 694 causes the particle to have a meandering motion.

From a statistical analysis, the major effects of the ambient pressure on the powder-relatedquantities are summarized as follows:

- 697 The number of ejected particles monotonically increases as the ambient pressure is
 698 decreased.
- The ejection drag force on the particle has a magnitude of $10^{-6} 10^{-5}$ N. It is the maximum when the ambient pressure is at a medium level (~ 1 bar). The entrainment drag force has a magnitude of $10^{-7} - 10^{-6}$ N and also peaks at ~ 1 bar. The recoil force has a magnitude of $\sim 10^{-5}$ N and does not significantly change with the ambient pressure.
- The ejection velocity of the particle is about 2 4 m/s. The entrainment velocity is about
 0.5 m/s. For both velocities, the maximum velocity is achieved at ~ 400 mbar.
- The temperature of the ejected particles monotonically increases from ~ 500 K to ~ 3500
 K, as the ambient pressure is increased from 10 mbar to 5 bar.
- The current 3D model provides quantitative information regarding the gas flow, powder-gas interaction, and powder behavior. Based on the above information, we propose that implementing a pulsed laser under a high ambient pressure can reduce the powder motion in LPBF.
- 710

711 Appendix A: Tracking powder-related quantities for simulations and experiments

For each powder particle in the simulation, the angular velocity, torque, velocity, forces (F_p and F_v), and average temperature can be tracked with respect to time. The tracking of Particles 1 -5 in Fig. 8 is given in Fig. A1, and the tracking of Particles 6 – 9 in Fig. 9 is given in Fig. A2, as examples. F_p is the force (magnitude) integrated by the pressure distribution (including recoil pressure if evaporation occurs) on the particle surface. F_v is the force (magnitude) integrated by the viscous stress distribution on the particle surface. These integrations are performed numerically using the "Lagrangian points" [26] on the particle surface. The average temperature of the particle
is calculated by averaging the temperature of the cells inside a particle from the CFD computation.
The particle tracking is terminated once the particle moves out of the calculation domain or merges
into the molten pool.

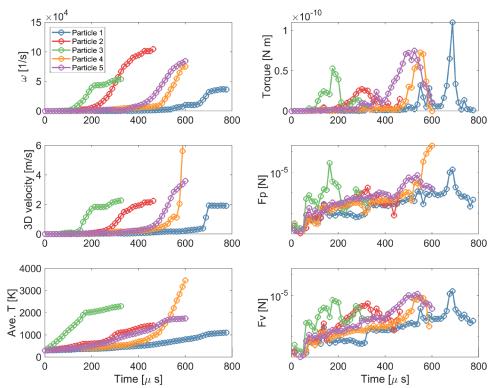


Fig. A1. Tracking of the angular velocity, torque, velocity, forces, and average temperature

- 724 of Particles 1 5 (in Fig. 8) with respect to time.
- 725

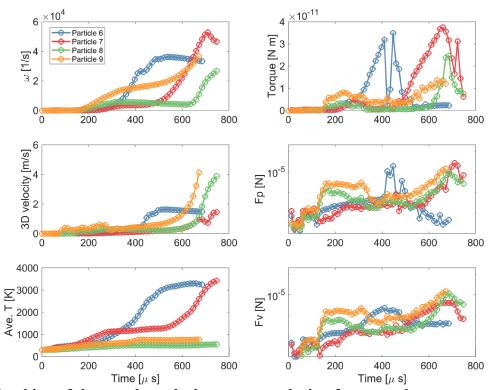
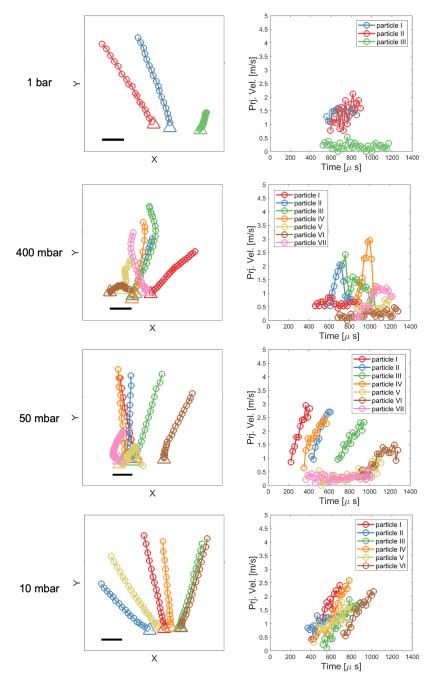


Fig. A2. Tracking of the angular velocity, torque, velocity, forces, and average temperature
 of Particles 6 – 9 in (Fig. 9) with respect to time.

The projected particle trajectories are manually extracted from the high-speed X-ray imaging, and the projected particle velocities can be calculated from the particle trajectories. Examples are shown in Fig. A3. The maximum projected velocities in a particle's lifespan are comparable to the velocity magnitude (0.5 - 3 m/s) shown in Fig. 12(c).



735 Fig. A3. Typical projected particle trajectories and velocities manually tracked from high-

- 736 speed X-ray imaging results. All scale bars are 100 μ m.
- 737

739 Appendix B: Details on statistical quantification of powder behavior

740 For each simulation, The "entrained particle" and the "ejected particle" are defined as follows. 741 If a powder particle moves towards the LIZ from the X-Z projection view, e.g., Fig. 8(b) and Fig. 742 9(b), for a projected distance larger than 50 µm, it is defined as an entrained particle. If a particle 743 moves away from the LIZ for a distance larger than 50 µm (not projected distance), it is defined 744 as an ejected particle. It is noted a particle can be both an entrained particle and an ejected particle. 745 For each entrained particle, the projected (X-Z plane) drag force on the particle when it is 746 entrained (i.e., moving towards the LIZ) is defined as the "entrainment drag." For each ejected 747 particle, the drag force (not projected) on the particle when it is ejected (i.e., moving away from 748 the LIZ) is defined as the "ejection drag." For both the entrained and ejected particle, the recoil 749 force is defined as the force (not projected) exerted on the particle by the recoil pressure if there is 750 any evaporation occurs on the particle surface. Based on these definitions, we can identify the 751 maximum entrainment drag, ejection drag, and recoil force for each entrained/ejected particle 752 during its lifespan. Finally, we average these maximum forces over all entrained/ejected particles 753 in the simulations and plot the averaged value in Fig. 12(a). The range of these maximum forces 754 is indicated by the error bars in Fig. 12(a). The occurrences of the entrained and ejected particles 755 are normalized by the total amount of particles in the powder bed, as shown in Fig. 12(b).

In Fig. 12(c), the entrainment and ejection velocities are defined as follows. For each entrained particle, the projected (X-Z plane) velocity when the particle is entrained is defined as the "entrainment velocity." For each ejected particle, the velocity (not projected) when the particle is ejected is defined as the "ejection velocity." Based on these definitions, we can identify the maximum entrainment and ejection velocity for each entrained/ejected particle during its lifespan. Finally, we average these maximum velocities over all entrained/ejected particles in the simulations and plot the averaged values in Fig. 12(c).

For each ejected particle, we can identify the maximum particle (average) temperature when it is ejected. We average the maximum particle temperature over all ejected particles and plot the averaged values in Fig. 12(d). Also, for each ejected particle, the last simulated moment when the particle is ejected is used to calculate the ejecting angle. The ejecting angle is calculated by acos (v_y/v) , where v is the particle velocity magnitude and v_y is the vertical component of velocity. We average the ejecting angle over all the ejected particles and plot the averaged values in Fig. 12(e).

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