

Effect of Band Symmetry on Photocurrent Production in Quasi-One-Dimensional Transition-Metal Trichalcogenides

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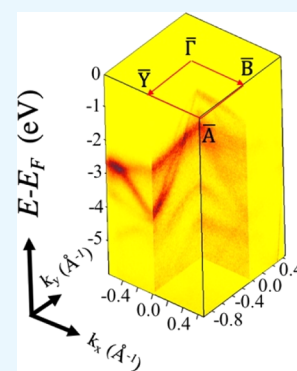
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ABSTRACT: Photocurrent production in quasi-one-dimensional (1D) transition-metal trichalcogenides, $\text{TiS}_3(001)$ and $\text{ZrS}_3(001)$, was examined using polarization-dependent scanning photocurrent microscopy. The photocurrent intensity was the strongest when the excitation source was polarized along the 1D chains with dichroic ratios of 4:1 and 1.2:1 for ZrS_3 and TiS_3 , respectively. This behavior is explained by symmetry selection rules applicable to both valence and conduction band states. Symmetry selection rules are seen to be applicable to the experimental band structure, as is observed in polarization-dependent nanospot angle-resolved photoemission spectroscopy. Based on these band symmetry assignments, it is expected that the dichroic ratios for both materials will be maximized using excitation energies within 1 eV of their band gaps, providing versatile polarization sensitive photodetection across the visible spectrum and into the near-infrared.



KEYWORDS: trichalcogenides, phototransistors, 2D materials, band symmetries, polarization dependence

INTRODUCTION

Transition-metal trichalcogenides (TMTs) are a unique class of two-dimensional (2D) materials that are set apart by their strongly anisotropic optical and electronic properties within their 2D layers.^{1–6} This anisotropy is the result of their quasi-one-dimensional (1D) crystal structure in which MX_3 ($\text{M} = \text{Ti}, \text{Zr}, \text{Hf}, \text{Ta}, \text{Nb}$; $\text{X} = \text{S}, \text{Se}, \text{Te}$) trigonal prisms are formed into 1D chains by strong covalent bonds along the crystallographic b axis.⁷ These 1D chains then form 2D layers in the $\bar{a}$ – b plane through van der Waals-like interactions.⁷ Because of their 1D nature, the TMTs have been garnering increasing attention for possible applications in electronics,^{5,8–13} optoelectronics,^{3–5,14–22} thermoelectrics,^{23–26} and energy storage.^{27–30} Because of the nature of the chains, the semimetallic and metallic MX_3 trichalcogenides can be scaled to nanometer widths and retain exceptionally high current densities,^{31–34} suggesting that these materials may be suitable for devices on the subnanometer scale.

Of particular interest are the strongly polarization-dependent optical properties, which have been demonstrated for both $\text{TiS}_3(001)$ and $\text{ZrS}_3(001)$.^{3–6,20,22,35,36} The polarization-dependent optical properties are promising for the creation of polarization-sensitive photodetectors. Photocurrent measurements, in previous studies, have shown dichroic ratios as high as 4:1 for TiS_3 ³ and 2.55:1 for ZrS_3 ,⁴ and the dichroic ratio has been reported to depend on both the excitation energy^{3–5} and the sample thickness.^{4,6} Although established

experimentally, the underlying mechanisms at play in this photocurrent production have not been intensely studied. Here, we show that these effects are the result of electronic orbital symmetry by comparing polarization-dependent scanning photocurrent microscopy (SPCM), along wires 45–85 nm in height by 0.5–0.9 μm wide, to polarization-dependent nanospot angle-resolved photoemission spectroscopy (nano-ARPES) for both $\text{TiS}_3(001)$ and $\text{ZrS}_3(001)$.

EXPERIMENTAL DETAILS

Nanowhiskers of TiS_3 and ZrS_3 were synthesized through direct reactions of metallic titanium or zirconium with sulfur vapor in vacuum-sealed quartz ampules, as described elsewhere.^{1,2,7,10,12,16,37,38} Figure S1 in the Supporting Information shows optical photographs of TiS_3 and ZrS_3 crystals, as well as their Raman spectra. These spectra are in perfect agreement with the previously published reports^{7,37,38} and confirm the high crystallinity of the TiS_3 and ZrS_3 whiskers. Phototransistors were fabricated by mechanically exfoliating TiS_3 and ZrS_3 crystals onto SiO_2 substrates, with the transistor channel parallel to the crystallographic b axis. Contacts 2 μm wide and $\sim 3 \mu\text{m}$ apart were defined using a Heidelberg BWL 66FS laser lithography system;

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then, 5 nm of Cr and 45 nm of Au were deposited with an AJA electron-beam evaporation system, as indicated in Figure 1. I – V

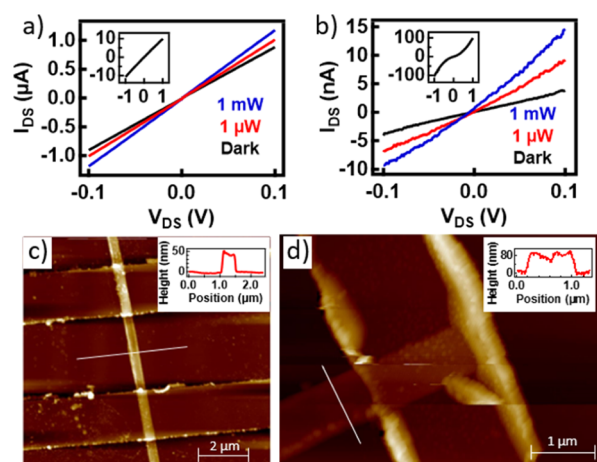


Figure 1. I – V curves for the (a) TiS₃ and (b) ZrS₃ phototransistors under different illumination intensities. The insets show expanded ranges to emphasize the difference in linearity. AFM of the (c) TiS₃ and (d) ZrS₃ phototransistors with insets showing the height profiles taken along the white lines.

curves indicate that this process forms largely Ohmic contacts on TiS₃ (Figure 1a), which is consistent with previous studies;^{9,10,12} however, the contacts appear to form Schottky barriers on ZrS₃ (Figure 1b). It has been shown that pure Au forms superior contacts with TiS₃ because of strong bonding with the surface sulfur.⁹ Au contacts may also reduce Schottky barrier formation in ZrS₃,^{9,12} nonetheless, Au is a large work function metal and ZrS₃ is an n-type semiconductor, so Schottky barrier formation is generally expected. A recognized cause of low mobility in these systems is phonon scattering,¹² which appears to be suppressed by an overcoat of Al₂O₃.¹⁰

Both TiS₃ and ZrS₃ phototransistors show a considerable change in resistance under illumination, as shown in Figure 1, consistent with previous studies,^{14,21} but this effect is reduced at higher voltages for ZrS₃ because of the nonlinearity of the I – V curves. Atomic force microscopy (AFM) was performed using a Bruker dimension ICON system, which shows that the TiS₃ phototransistor channel was

roughly 45 nm tall and 500 nm wide (Figure 1c) and the ZrS₃ channel was approximately 85 nm tall and 900 nm wide (Figure 1d).

SPCM was performed with a 488 nm solid-state laser (Coherent Sapphire 100) passed through a neutral density filter followed by a linear polarizer, before being reflected off a dichroic mirror and focused onto the phototransistors using an inverted Olympus IX 81 microscope. The neutral density filter was used to maintain a light intensity of 100 μW for all measurements. Measurements were performed at normal incidence, so that the incident light was polarized perpendicular to the $\vec{c}$ axis. For more information on the SPCM setup, see the studies by Li et al.³⁹ and Chen et al.⁴⁰

NanoARPES measurements were performed on the ANTARES beamline at the synchrotron SOLEIL⁴¹ on nanowhiskers of TiS₃(001) and ZrS₃(001) that were exfoliated *in situ* under vacuum better than 10^{−10} Torr. Precise alignment and positioning on the nanowhiskers were achieved with a spot size of ~500 nm.^{1,2,42,43} Measurements were taken at an incidence angle of 45° with respect to the surface normal. The polarization dependence of nanoARPES was observed by comparing measurements where the plane of incidence coincides with either $\vec{a}$ – $\vec{c}$ or $\vec{b}$ – $\vec{c}$ crystallographic planes.

Thus far, all directions have been in terms of the crystallographic coordinate system; however, throughout this paper, we will also discuss the directions along the surface Brillouin zone (BZ) as well as the rectangular representation of the electron orbitals. It is convenient to clarify before going further that the quasi-1D chains lie along the crystallographic $\vec{b}$ axis, which corresponds to the $\vec{y}$ axis of the electron orbitals and the $\vec{\Gamma}$ to $\vec{Y}$ direction along the BZ. Similarly, the crystallographic $\vec{a}$ axis maps to the $\vec{x}$ electron orbital axis and the $\vec{\Gamma}$ to $\vec{B}$ direction of the BZ. To avoid confusion, the BZ directions will always be noted as either $\vec{\Gamma}$ to $\vec{Y}$ or $\vec{\Gamma}$ to $\vec{B}$.

RESULTS AND DISCUSSION

The SPCM images of TiS₃(001) and ZrS₃(001) are shown in Figure 2 with an excitation energy of 2.5 eV (488 nm) and the incident light polarized along either the $\vec{b}$ ($\theta = 0^\circ$) or $\vec{a}$ ($\theta = 90^\circ$) axis. For both materials, the photocurrent intensity shows a notable polarization dependence, which is consistent with other polarization studies on TiS₃(001)^{3,5,6,35,36} and ZrS₃(001).^{4,6,20} However, the change in photocurrent intensity as a result of changing light polarization is far more significant for ZrS₃(001) than for TiS₃(001), as seen by the photocurrent images in Figure 2, where the dichroic ratio for ZrS₃(001) is ~4:1, but for TiS₃(001), the ratio is only ~1.2:1.

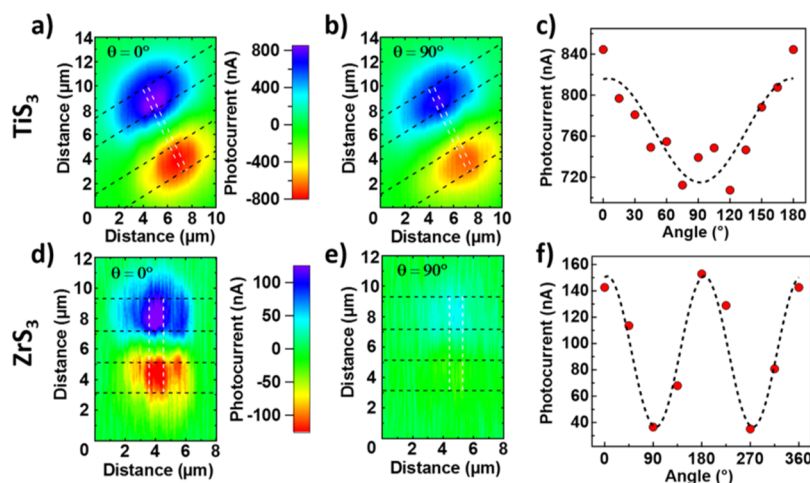


Figure 2. SPCM images of TiS₃(001) (a,b) and ZrS₃(001) (d,e) using light polarized along the $\vec{b}$ (a,d) and $\vec{a}$ (b,e) axes. The dotted black and white lines indicate the contacts and transistor channel, respectively. Angle dependence of the photocurrent intensity in TiS₃ (c) and ZrS₃ (f), where the angle is measured with respect to the $\vec{b}$ -axis. The photocurrent intensities in (c) and (f) are the average of the photocurrent magnitudes at the source and drain.

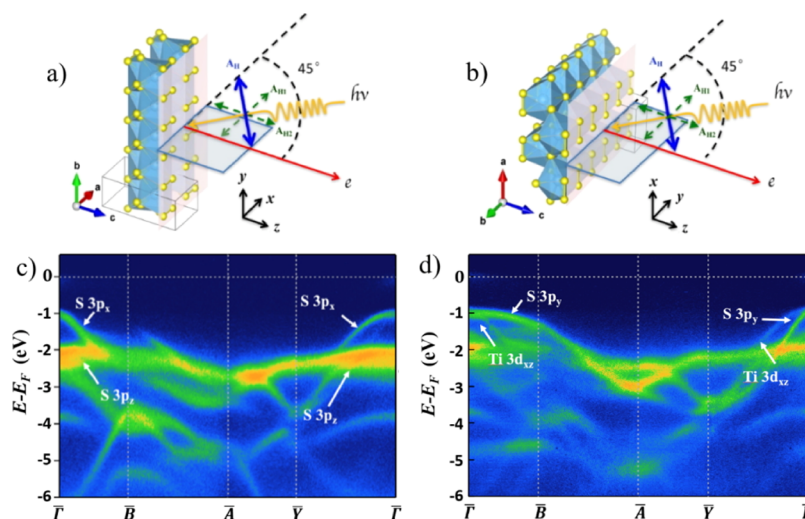


Figure 3. Experimental setups illustrating polarizations (a) $\epsilon_x + \epsilon_z$ and (b) $\epsilon_z + \epsilon_y$ used in the ARPES measurements of $\text{TiS}_3(001)$ shown in (c,d), respectively, taken at a photon energy of 100 eV.

The polarization-dependent behavior can be explained by examining the average electron transition rate dictated by Fermi's golden rule,^{44–50} which is shown in eq 1

$$\Gamma = \frac{2\pi}{\hbar} |\langle \varphi_f | H' | \varphi_i \rangle|^2 \delta(E_f - E_i - \hbar\omega) \quad (1)$$

where φ_i , φ_f , E_i , and E_f are the initial and final state wave functions and energies, respectively, and H' is the perturbation Hamiltonian. For photoexcitation measurements, H' can be approximated as $\frac{e}{m_e c} \vec{A} \cdot \vec{P}$ where $\vec{A}$ is the vector potential and $\vec{P}$ is the momentum operator, under the assumptions that the field is sufficiently weak so that $|\vec{A}|^2$ is negligible and the momentum of the photon is negligible compared to the electron's momentum.^{49,50} Using this approximation, $H'|\varphi_i\rangle$ is proportional to $\vec{\epsilon} \cdot \vec{r}$ where $\vec{\epsilon}$ is the light polarization vector and $\vec{r}$ is the dipole of the electron orbital. Furthermore, group theory considerations dictate that the transition will only occur if $\vec{\epsilon} \cdot \vec{r}$ belongs to the same irreducible representation as φ_f . The TMTs belong to the $P2_1/m$ space group^{1,2,51} where the rotation axis is along the crystallographic b axis. This space group corresponds to the C_{2h} group symmetry in Schönflies notation. However, at the surface of the material, both rotational and inversion symmetries are lost such that the group symmetry reduces to C_s . This loss of symmetry and the corresponding change in the selection rules are likely responsible for the dependence of the dichroic ratio on the sample thickness.^{4,6} Based on density functional theory (DFT) calculations,⁵² the bottom of the conduction band in TiS_3 is dominated by the $\text{Ti } 3d_{z^2}$ and $3d_{x^2-y^2}$ electron orbitals. Similarly, the bottom of the conduction band in ZrS_3 should be largely $\text{Zr } 4d_{z^2}$ and $4d_{x^2-y^2}$.⁵³ All of these orbitals belong to the highest symmetry irreducible representations in the C_{2h} and C_s symmetry groups. If the final states are limited to d_{z^2} and $d_{x^2-y^2}$, then, in the bulk with C_{2h} symmetry, light polarized along the orbital y axis (ϵ_y = along the chains) will exclusively excite electrons from the p_y orbitals and light polarized along the x axis (ϵ_x) will excite the p_x orbitals. Because of the reduction of symmetry at the surface, ϵ_y can also excite the d_{yz} and d_{xy} orbitals and $\epsilon_x + \epsilon_z$ can also excite the d_{xz} , d_{z^2} , and $d_{x^2-y^2}$ orbitals, though the latter two should have negligible contributions to the valence band.^{52,53} These additional

excitations will only occur if there is p–d hybridization since $d \rightarrow d$ transitions are optically forbidden.^{49,50} The stronger photocurrent signal using ϵ_y (Figure 2) indicates that the p_y orbitals contribute most strongly to photocurrent production.

The cause of the p_y dominance, in the dichroism of photoconductance, can be examined through symmetry assignments of the experimentally determined band structure using polarization-dependent nanoARPES, which is shown for $\text{TiS}_3(001)$ and $\text{ZrS}_3(001)$ in Figures 3 and 4, respectively.

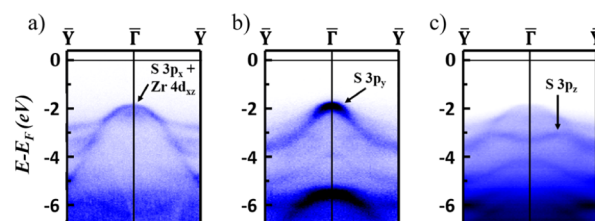


Figure 4. Angle-resolved photoemission derived band structure of $\text{ZrS}_3(001)$ using incident light polarizations (a) ϵ_x , (b) ϵ_y , and (c) $\epsilon_x + \epsilon_z$. The photon energy is 130 eV.

Photoemission is extremely surface sensitive with a final state that is a fully symmetric free electron such that the selection rules are the same as the photocurrent measurements. Electron orbitals within the same irreducible representations were distinguished with the assistance of previously reported DFT calculations.^{52,53}

Figure 3a schematically shows the light polarization geometry for the nanoARPES measurements on $\text{TiS}_3(001)$ shown in Figure 3c where $\vec{\epsilon} = \epsilon_x + \epsilon_z$. Our previous work^{1,2} utilized this polarization because it emphasizes the top of the valence band, which may be attributed to the $\text{S } 3p_x$ orbitals.^{52,53} These p_x orbitals are highly anisotropic with a hole effective mass that is twice as large along $\bar{\Gamma}$ to $\bar{Y}$ than along $\bar{\Gamma}$ to $\bar{B}$, as has been noted elsewhere.^{1,2} The $\text{S } 3p_z$ orbitals contribute most strongly to the band structure 2 eV below the Fermi level (E_F) or ~ 1 eV below the valence band maximum, and these bands have minimal dispersion. The $\text{S } 3p_z$ weighted bands are very evident with light polarized along $\epsilon_x + \epsilon_z$ (Figure 3c), and the symmetry selection rules allow photoemission from this initial state. The features of the $\text{S } 3p_z$ bands

can be more easily discerned after a second derivative treatment of the data, which was reported in our previous work.² Both of these assignments are in agreement with DFT calculations.⁵²

When the light polarization is switched to $\vec{e} = e_z + e_y$ (Figure 3b), additional bands are observed near the top of the valence band (Figure 3d) which can be assigned to the S 3p_y orbitals. The S 3p_x and 3p_y orbitals at the top of the valence band are nearly degenerate at $\bar{\Gamma}$, contrary to DFT calculations, which place the 3p_y orbital weighted band at a binding energy well below the 3p_x orbital.^{3,20,52–54} Unlike the 3p_x orbitals, the 3p_y bands have a significantly larger hole effective mass along $\bar{\Gamma}$ to $\bar{B}$ than along $\bar{\Gamma}$ to $\bar{Y}$. The 3p_x hybridized bands remain visible, though significantly weaker, with the light polarization perpendicular to the $\vec{x}$ orbital axis. This indicates that although these bands are mostly S 3p_x weighted, they also have other contributions which DFT calculations⁵² place as largely the Ti 3d_{xz} orbitals.

The nanoARPES measurements on ZrS₃(001) with polarization vectors e_x , e_y , and $e_x + e_z$ are shown in Figure 4a–c, respectively. As with TiS₃(001), the S 3p_x and 3p_y orbitals in ZrS₃(001) are both present near the top of the valence band and nearly degenerate at $\bar{\Gamma}$. Previous ARPES measurements on ZrS₃(001) attributed these two bands to spin–orbit coupling.⁵⁵ The orbital symmetries of these two bands, near the top of the valence band, are observed to be significantly different in Figure 4. Furthermore, the spectral weight assignments to sulfur weight p bands, as suggested by theory, and the absence of wave vector displacement that is typical of spin–orbit coupling indicate that this band splitting at the top of the valence band at $\bar{\Gamma}$ is not dominated by spin–orbit splitting and has no exchange splitting contribution.

The S 3p_x orbital weighted band is almost entirely absent in Figure 4c which is consistent with S 3p_x combined with a minor Zr 4d_{xz} component dominating the top of the valence band. As with TiS₃, the S 3p_z character states in ZrS₃ contribute to the valence band structure at binding energies ~ 1 eV below the valence band maximum. These symmetry assignments are the same as seen in Figure 3 for TiS₃(001) and agree with DFT for TiS₃(001)⁵² and ZrS₃(001).⁵³

Based on Figures 3 and 4, the S p_y orbitals extend from close to the valence band maximum to binding energies ~ 2 eV below the valence band maximum. The S p_y orbitals, nonetheless, have a strong spectral weight density within 0.5 eV of the valence band maximum. Because the top of the valence band occurs at the center of the BZ,^{1,2} the near degeneracy of the p_y and p_x bands at the center of the BZ, $\bar{\Gamma}$, affects light polarization sensitivity. If the trichalcogenide S p_x orbital weighted band occurred at significantly higher binding energies than the S p_y orbital weight bands, as suggested by theory,^{3,20,52–54} then excitation energies close to the band gap would generate larger photocurrents with light polarized perpendicular to the 1D chains, resulting in a region of photon energy where the dichroic ratio could switch.

Larger Z value TMTs, specifically ZrSe₃ and HfSe₃, do exhibit an appreciable splitting in binding energy between the trichalcogenide S p_x and S p_y bands at $\bar{\Gamma}$,⁵⁵ but this splitting is not resolved here at $\bar{\Gamma}$, the center of the surface BZ, for TiS₃(001) and ZrS₃(001), as seen in Figures 3 and 4, respectively. Yet, the existing band structure calculations,^{52,53} that include symmetry assignments, are seen to be in agreement with the experiment. The band structure is also

seen to be consistent with many of the device characteristics as noted elsewhere.¹²

Since the d_{z²} and d_{x²–y²} orbitals dominate the bottom 0.5 eV of the conduction band for both ZrS₃ and TiS₃,^{52,53} it is expected that the dichroic ratios will be maximized with excitation energies within 1 eV of the band gap and will begin to diminish for larger excitation energies. The band gaps for TiS₃ and ZrS₃ are ~ 1 ^{2,14,15,17,54,56,57} and ~ 2 eV,^{1,4,20,53,54,58–61} respectively, so that the dichroic ratios should be maximized with excitation energies of 1.0–2.0 eV for TiS₃ and 2.0–3.0 eV for ZrS₃. This conclusion is supported by comparison of the presently reported SPCM data with other experimental measurements.^{3–6} Here, the polarization dependence of the photoconductance is far more significant for ZrS₃(001) than for TiS₃(001), but the photon energy of 2.5 eV (488 nm) used here is much closer to the band gap of ZrS₃ than TiS₃. For TiS₃, the dichroic ratio has been observed to increase as the photon excitation energy is decreased from 2.3 to 1.5 eV.³ Although prior absorption calculations vary significantly,^{3–5,62} they tend to place the maximum dichroic ratio at slightly higher excitation energies, which is likely caused by the discrepancies between calculated^{3,20,52–54} and experimental^{1,2,55} band structures mentioned earlier. Indeed, the optical excitation wavelength used here (488 nm) is very close to the observed optical resonance seen for ZrS₃(001) (see Figure S2 in the Supporting Information). Despite possible discrepancies, it is clear that when combined, TiS₃ and ZrS₃ have promise as polarization sensitive photodetectors over a wide range of wavelengths.

CONCLUSIONS

Both TiS₃(001) and ZrS₃(001) exhibit strongly polarization sensitive photocurrent production because of their electronic bands with distinct in-plane symmetries. As we have noted, existing band structure calculations that include symmetry assignments, as noted elsewhere,^{52,53} are seen to be in agreement with the experiment. The S 3p_y orbital weighted band is strongest within 0.5 eV of valence band maximum and dominates photocurrent production. Because the bottom 0.5 eV of the conduction band for both materials is comprised primarily of d_{z²} and d_{x²–y²} orbitals, the dichroism in the photocurrent production will be most significant for excitation energies between 1.0 and 2.0 eV for TiS₃(001) and 2.0 and 3.0 eV for ZrS₃(001). Thus, TiS₃ and ZrS₃ show considerable promise as versatile polarization sensitive photodetectors across the visible spectrum and into the near-infrared.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsami.0c11892>.

Raman spectra and optical absorption spectra of suspensions of TiS₃ and ZrS₃ (PDF)

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Author Contributions

P.A.D., A.J.Y., M.C.A., and S.J.G. developed the experimental concepts. H.Y., Jose Avila, and M.C.A. conducted the nanoARPES measurements. H.Y., Jose Avila, M.C.A., S.J.G., and T.K. analyzed the corresponding nanoARPES data. S.J.G., J.-S.C., and A.D. conducted the SPCM and *I*–*V* measurements and analyzed the corresponding data. S.J.G. fabricated the transistor devices and performed AFM. A.L., Jehad Abourahma, and A.S. grew and characterized the TiS₃ and ZrS₃ crystals. The conclusions were formed through discussions by all authors. The manuscript was written by S.J.G. and P.A.D. with contributions and suggestions from all authors.

Notes

The authors declare no competing financial interest.

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