

## EVOLUTIONARY BIOLOGY

## Carbon isotope evidence for the global physiology of Proterozoic cyanobacteria

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Ancestral cyanobacteria are assumed to be prominent primary producers after the Great Oxidation Event [ $\approx 2.4$  to 2.0 billion years (Ga) ago], but carbon isotope fractionation by extant marine cyanobacteria ( $\alpha$ -cyanobacteria) is inconsistent with isotopic records of carbon fixation by primary producers in the mid-Proterozoic eon (1.8 to 1.0 Ga ago). To resolve this disagreement, we quantified carbon isotope fractionation by a wild-type planktic  $\beta$ -cyanobacterium (*Synechococcus* sp. PCC 7002), an engineered Proterozoic analog lacking a CO<sub>2</sub>-concentrating mechanism, and cyanobacterial mats. At mid-Proterozoic pH and pCO<sub>2</sub> values, carbon isotope fractionation by the wild-type  $\beta$ -cyanobacterium is fully consistent with the Proterozoic carbon isotope record, suggesting that cyanobacteria with CO<sub>2</sub>-concentrating mechanisms were apparently the major primary producers in the pelagic Proterozoic ocean, despite atmospheric CO<sub>2</sub> levels up to 100 times modern. The selectively permeable microcompartments central to cyanobacterial CO<sub>2</sub>-concentrating mechanisms ("carboxysomes") likely emerged to shield rubisco from O<sub>2</sub> during the Great Oxidation Event.

## INTRODUCTION

Members of the phylum Cyanobacteria are the only extant bacteria capable of oxygenic photosynthesis, leading to the inference that ancestral cyanobacteria were responsible for the Paleoproterozoic accumulation of atmospheric O<sub>2</sub> known as the Great Oxidation Event [GOE; 2.4 to 2.0 billion years (Ga) ago] (1). Although estimates of when oxygenic photosynthesis originated span a billion years—from sometime in the Paleoarchean eon (3.6 to 3.2 Ga ago) to immediately preceding the GOE [Fig. 1 and the Supplementary Materials (SM)] [e.g., (2, 3)]—the oxidative impact of this metabolism across the GOE was profound. Atmospheric O<sub>2</sub> concentrations increased by up to 100 million-fold (1, 4) relative to CO<sub>2</sub> concentrations (Fig. 1), while primary productivity rose to potentially modern levels (5). Following the GOE, the trajectories of both atmospheric O<sub>2</sub> concentrations and primary productivity appear to have stalled, with atmospheric oxygen falling to somewhere between 0.1 and 10% of present atmospheric levels [1 PAL = 210,000 parts per million (ppm) O<sub>2</sub>; Fig. 1] (1, 6) and oxygenic primary production decreasing to less than 10% of modern values (5). Stabilization of the Earth system at this intermediate state of oxygenic primary production characterized much of the Proterozoic eon (7, 8). There are a variety of hypotheses for why this stasis defined the Proterozoic Earth system [e.g., (9–12)] and the physiology of ancestral cyanobacteria features prominently in all of them.

While ancestral cyanobacteria are assumed to play a central role in Proterozoic biogeochemistry, there is limited direct evidence of the ecological niches that they occupied. The oldest unambiguous cyanobacterial microfossils are found in 2.018- to 2.015-Ga peritidal black cherts of the Orosirian Belcher Group (13, 14). When similarly preserved fossil cyanobacteria are found in younger Proterozoic rocks, they are also interpreted as ancient analogs of benthic cyanobacteria in littoral environments (15). If the paleontological record

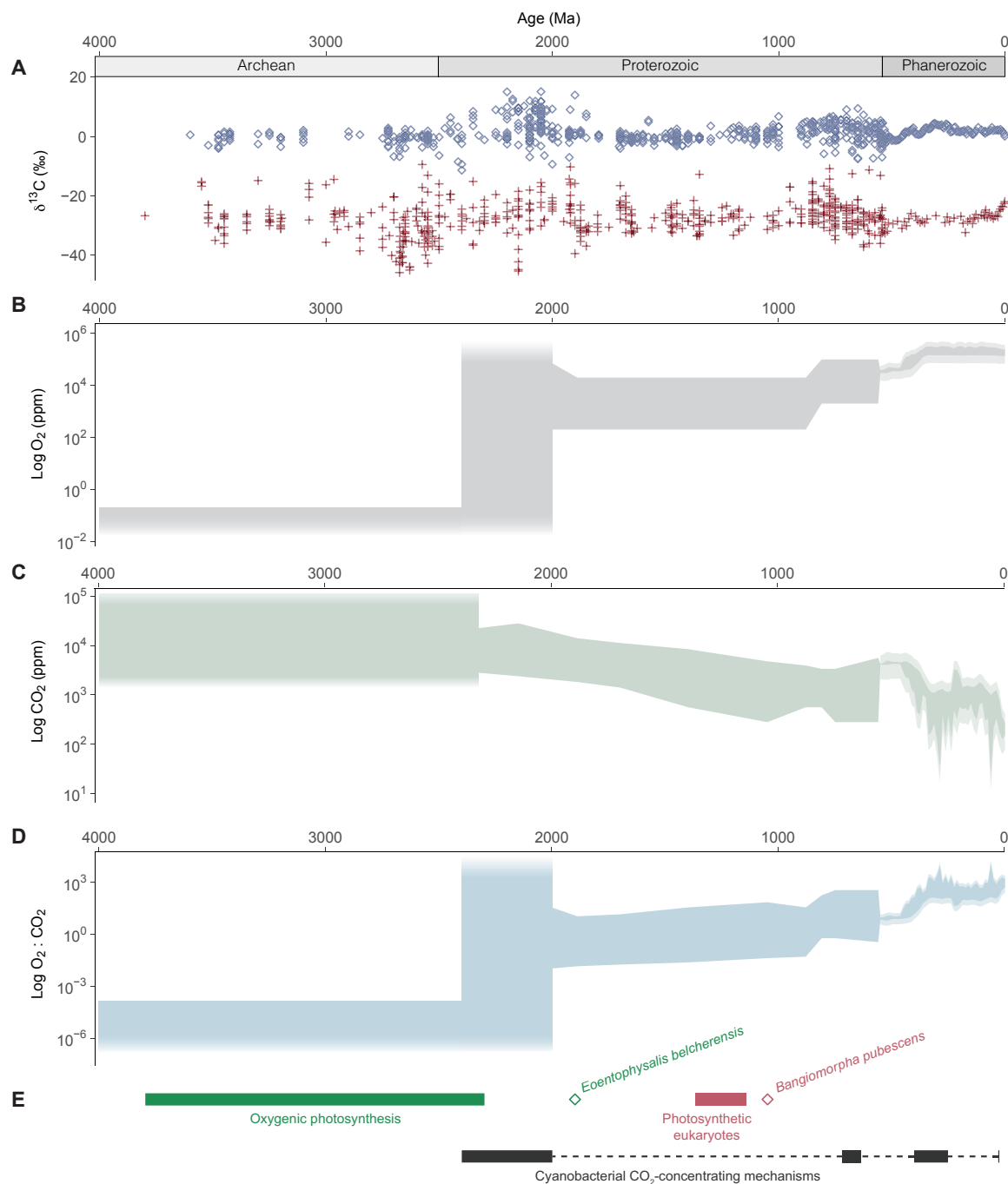
is expanded to include all possible microfossils with cyanobacterial affinities, then benthic forms still dominate, with rare and contentious interpretations of cyanobacteria in planktic habitats (16, 17). The lack of fossil indicators for planktic cyanobacteria may reflect an absence of these cyanobacterial lineages at this time (18) or the improbable preservation of cyanobacterial microfossils in pelagic environments (19). Paired biomarker and nitrogen isotope measurements identify the presence of pelagic cyanobacteria by 1.1 Ga ago (20), but earlier documentation of a pelagic habitat would help evaluate hypotheses for the global influence of cyanobacteria in the Proterozoic Earth system.

If Proterozoic cyanobacteria inhabited a globally important ecological niche, the productivity of the biosphere would be largely dependent on their ability to fix carbon. At the level of the global marine ecosystem, the most continuous evidence of carbon fixation by the dominant primary producers is preserved in sedimentary marine carbon isotope records. The carbon isotopic difference between carbonate minerals and total organic carbon (TOC) ( $\epsilon_{\text{TOC}}$ ; eq. S1) in sedimentary rocks has well-resolved coverage between the GOE, the origin of photosynthetic eukaryotes (21), and the ultimate ecological dominance of photosynthetic eukaryotes in the pelagic marine environment (22). Although the isotopic difference summarized by  $\epsilon_{\text{TOC}}$  is imparted initially by the net carbon isotope effect associated with carbon fixation by primary producers ( $\epsilon_p$ ; eq. S2), carbon isotope fractionations associated with geologic preservation do not allow for  $\epsilon_{\text{TOC}}$  to be directly substituted for  $\epsilon_p$  (23).

We used bootstrap resampling and Monte Carlo simulations to produce a new record of  $\epsilon_p$  in the middle of the Proterozoic eon (1.8 to 1.0 Ga ago), taking into account isotopic fractionations that occur as the primary substrates and products of carbon fixation (e.g., dissolved CO<sub>2</sub> and photoautotrophic biomass) are transformed into their final geological states (e.g., carbonate rocks and TOC). This new  $\epsilon_p$  record was derived from a curated dataset of carbon isotope measurements from sedimentary rocks from a variety of depositional settings, including open and shallow marine environments (24). The middle Proterozoic shows limited variation in the sedimentary carbon isotope record [e.g., (8, 24)] spanning the proposed "Age of Cyanobacteria" (25). As a result, it represents a favorable target for isolation of any cyanobacterial component of the Proterozoic  $\epsilon_p$  record.

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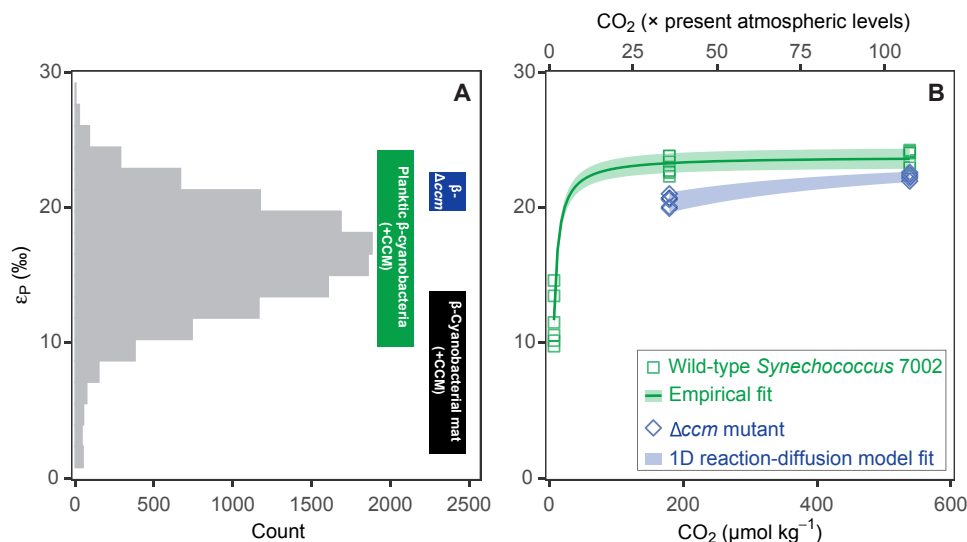
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**Fig. 1. Isotopic, atmospheric, and biologic context for the Proterozoic “Age of Cyanobacteria” (25).** (A) Carbonate  $\delta^{13}\text{C}$  values shown in blue diamonds and total organic carbon  $\delta^{13}\text{C}$  values shown in red crosses (24). Ma, million years. (B) Mass-independent sulfur isotope fractionation restricts Archean  $p\text{O}_2$  estimates to  $<10^{-6}$  PAL or 2 ppm (4). Proterozoic and Phanerozoic  $p\text{O}_2$  estimates come from proxies and modeling (8, 59). (C) Archean, Proterozoic, and Phanerozoic  $\text{CO}_2$  estimates come from proxies and modeling (8, 39, 59). (D) Estimated range of  $\text{O}_2$ -to- $\text{CO}_2$  ratios (each expressed in ppm) from the Archean through the Phanerozoic eons. (E) Range of time estimates for the origin of oxygenic photosynthesis (e.g., 2, 3) shown as a green bar and the earliest unambiguous cyanobacterial microfossils (*Eoentophysalis belcherensis*) shown as a green diamond (13, 14). Age of earliest unambiguous photosynthetic eukaryote (*Bangiomorpha pubescens*) shown as red diamond with corresponding molecular clock estimates for the primary plastid endosymbiosis shown as a red bar (21). Proposed dates for the emergence of a cyanobacterial CCM shown as black bars [e.g., (34)].

Our statistical simulation of middle Proterozoic  $\epsilon_p$  values yielded a distribution in which 95% of the values fall between 8 and 24 per mil (‰) (95th percentile) with a median value of 16‰ (Fig. 2A and the SM). This  $\epsilon_p$  distribution provides a benchmark to compare different autotrophic contributions to global Proterozoic primary production.

Benthic cyanobacteria have, for example, been proposed as ecologically important contributors to Proterozoic primary production (18). In modern cyanobacterial mats, benthic photoautotrophic biomass is commonly enriched in  $^{13}\text{C}$  relative to biomass from planktic environments [e.g., (26, 27)]. We used our statistical simulation



**Fig. 2. Middle Proterozoic  $\epsilon_P$  estimates as compared to empirical cyanobacterial  $\epsilon_P$  values.** (A) Histogram of estimated  $\epsilon_P$  values between 1.0 and 1.8 Ga. Boxed vertical ranges represents  $\epsilon_P$  values from a cyanobacterial mat system [black (26)] and  $\epsilon_P$  values reported here in cultures of WT (green) and Δccm mutant (blue) *Synechococcus* sp. PCC 7002 strains. (B) Measured values of  $\epsilon_P$  increase at higher dissolved  $\text{CO}_2$  levels in cultures of *Synechococcus* sp. PCC 7002. In the WT strain (green squares),  $\epsilon_P$  values covary with  $[\text{CO}_{2(\text{aq})}]^{-1}$  (green line, fig. S5;  $R^2$ , 0.96). Blue diamonds are experimental results for the Δccm mutant, which requires  $\geq 36 \times \text{PAL}$   $\text{CO}_2$  to grow under our experimental conditions. The shaded blue band represents calculations from a 1D reaction-diffusion model trained on physiological observations of the Δccm mutant. Horizontal axes refer to  $\text{CO}_2$  in the culture headspace relative to PAL (1 PAL = 280 ppm  $\text{CO}_2$ ; upper axis) and the corresponding dissolved  $\text{CO}_2$  in the culture medium (micromole per kilogram; lower axis). Data points represent biological replicates ( $n = 6$  for each condition).

to quantify the distribution of  $\epsilon_P$  values in a well-characterized modern mat system on the basis of previously published values of  $\delta^{13}\text{C}_{\text{carb}}$  and  $\delta^{13}\text{C}_{\text{org}}$  (26). In this system, the predicted distribution of  $\epsilon_P$  values has a median value of 8.5‰ and a range of 4 to 13‰ (95th percentile; fig. S3). This exercise suggests that the dynamics of carbon supply in cyanobacterial mats appears to limit the overall  $\epsilon_P$  range that they can preserve, especially in hypersaline environments (26). The  $\epsilon_P$  distribution for this system covers less than 25% of the middle Proterozoic  $\epsilon_P$  record, with the overlap restricted to a small tail in the Proterozoic distribution that extends to  $\epsilon_P$  values less than 10‰ (Fig. 2A). Detailed datasets do not exist that can similarly constrain how  $\epsilon_P$  distributions for cyanobacterial mats might change if  $\text{CO}_2$  levels approached those proposed for middle Proterozoic (8). Proof-of-concept experiments, however, indicate that mat  $\epsilon_P$  values average  $\approx 11\%$  when overlying  $\text{CO}_2$  levels are  $< 36 \times \text{PAL}$  (1 PAL = 280 ppm  $\text{CO}_2$ ) and approach  $\approx 25\%$  only at  $\text{CO}_2$  levels of  $\approx 320$  to  $420 \times \text{PAL}$  [table 2 in (27)]. Although benthic cyanobacterial microfossils are common in the Proterozoic eon,  $\epsilon_P$  values associated with cyanobacterial mats appear to be much less than those seen in the middle Proterozoic  $\epsilon_P$  distribution unless  $\text{CO}_2$  levels were much greater than proposed for the middle Proterozoic (Fig. 2A).

The middle Proterozoic  $\epsilon_P$  distribution also differs from  $\epsilon_P$  values characteristic of planktic cyanobacteria dominant in open ocean ecosystems today (28). Values of  $\epsilon_P$  cluster from  $\approx 15$  to 19‰ in physiologically controlled experiments with a planktic member of the monophyletic marine *Synechococcus/Prochlorococcus* (Syn/Pro) group (28), *Synechococcus* sp. CCMP838. This tight range spans less than 33% of the middle Proterozoic  $\epsilon_P$  distribution. Experimental Syn/Pro  $\epsilon_P$  values lack sensitivity to  $\text{CO}_2$  levels [between 6 and 18 μmol kg<sup>-1</sup> (28)] or specific growth rate (28), which suggests that variations in these factors cannot be called on to explain the full middle Proterozoic  $\epsilon_P$  distribution. A complete interpretation of the Proterozoic carbon isotope record thus seems to require major

contributions by noncyanobacterial primary producers or a shift in our understanding of carbon fixation by Proterozoic cyanobacteria.

It is possible that extant marine cyanobacteria from the Syn/Pro clade may not represent apt physiological analogs for Proterozoic cyanobacteria. All extant cyanobacteria use at least one  $\text{CO}_2$ -concentrating mechanism (CCM) (29) to increase the supply of  $\text{CO}_2$  to rubisco (ribulose 1,5-bisphosphate carboxylase/oxygenase), the key  $\text{CO}_2$ -fixing enzyme in the Calvin-Benson cycle (30). Cyanobacterial rubisco is partitioned into a selectively permeable protein microcompartment known as a carboxysome along with carbonic anhydrase. Inside the carboxysome, actively accumulated intracellular  $\text{HCO}_3^-$  is rapidly interconverted into  $\text{CO}_{2(\text{aq})}$  through the activity of carbonic anhydrase (29, 31, 32). Examination of cyanobacterial CCMs reveals a clear division within the phylum (29). The marine Syn/Pro clade (α-cyanobacteria) contain α-carboxysomes and Form 1A rubisco that are evolutionarily (29) and structurally distinct (29, 33) from the β-carboxysomes and form 1B rubisco shared by the freshwater, estuarine, and marine species (the β-cyanobacteria) in the remainder of the phylum.

As α-cyanobacteria diverged from cyanobacterial lineages of β-cyanobacteria at the end of the Proterozoic eon, between 1.0 and 0.5 Ga ago (18), β-carboxysomes appear to be the more ancient basis for a cyanobacterial CCM. Estimates for the initial emergence of CCMs in β-cyanobacteria span over 2 Ga of earth history (34) and are often associated with drops in global  $\text{CO}_2$  associated with glacial episodes at ca. 2.4 to 2.0 Ga, ca. 0.7 to 0.6 Ga, and, potentially, 0.4 to 0.3 Ga ago (Fig. 1). It is possible that either biochemical differences between α- and β-cyanobacteria or the absence of β-carboxysomes in Proterozoic cyanobacteria could account for the mismatch between  $\epsilon_P$  values from α-cyanobacteria and the middle Proterozoic  $\epsilon_P$  distribution. Potential biochemical differences between α- and β-cyanobacteria include the influx and efflux of rubisco substrates

and products from the carboxysome (33) as well the kinetics of rubisco and carbonic anhydrase within the carboxysome (35). These differences would likely alter how whole-cell carbon fixation rates respond to changing environmental conditions (e.g.,  $\text{CO}_2$  concentrations), potentially expanding or contracting the accessible range of cyanobacterial  $\epsilon_P$  values. The possible absence of a  $\beta$ -carboxysome in Proterozoic cyanobacteria would allow freer access of substrates to and from rubisco and carbonic anhydrase, potentially affecting cyanobacterial  $\epsilon_P$  values over a wide range of  $\text{CO}_2$  concentrations as well.

We propose that primary production by cyanobacteria in the middle Proterozoic might resemble either carbon fixation by extant cyanobacteria with  $\beta$ -carboxysome-based CCMs or a physiologically distinct mode of carbon fixation by ancestral  $\beta$ -cyanobacteria lacking a CCM. To evaluate these possibilities, we determined  $\epsilon_P$  values for a model cyanobacterium containing  $\beta$ -carboxysomes, wild-type (WT) *Synechococcus* sp. PCC 7002 (*Synechococcus* 7002), and an engineered mutant of this strain lacking carboxysomes ( $\Delta ccm$ ) (31, 36, 37) across a range of  $\text{CO}_2$  concentrations. Net carbon isotope fractionation by WT *Synechococcus* 7002 allows us to compare  $\epsilon_P$  relationships in  $\beta$ -cyanobacteria to previously published  $\epsilon_P$  values from  $\alpha$ -cyanobacteria (fig. S6) (28). The  $\Delta ccm$  mutant, which is high  $\text{CO}_2$  requiring, represents a potential physiological analog for pre-CCM-bearing Proterozoic cyanobacteria.

## RESULTS

WT *Synechococcus* 7002 grew at dissolved  $\text{CO}_2$  concentrations of 7 to 538  $\mu\text{mol l}^{-1}$ , corresponding to headspace  $\text{CO}_2$  of 1 to 107  $\times$  PAL at pH 6.7 to 8.1. The  $\Delta ccm$  mutant failed to grow at  $\text{CO}_2$  levels of 1, 18, and 30  $\times$  PAL but was able to grow at 36 and 107  $\times$  PAL at pH 7.3 to 8.1 (fig. S4). These experimental conditions are consistent with both  $p\text{CO}_2$  [1 to 100 PAL (8)] and pH [6.8 to 8.2 (38, 39)] estimates relevant to the middle Proterozoic marine biosphere (fig. S10). The  $\epsilon_P$  values from acclimated WT batch cultures range from  $11.7 \pm 2.0\text{‰}$  to  $23.8 \pm 0.5\text{‰}$  over 1 to 107  $\times$  PAL, while for  $\Delta ccm$  batch cultures,  $\epsilon_P$  values range from  $20.5 \pm 0.4$  to  $22.3 \pm 0.2$  over 36 to 107  $\times$  PAL (Fig. 2B). In both the WT and  $\Delta ccm$  experiments, values of  $\epsilon_P$  increase with higher concentrations of  $\text{CO}_{2(\text{aq})}$ , in contrast to the insensitivity of  $\epsilon_P$  to  $\text{CO}_{2(\text{aq})}$  in cyanobacteria with  $\alpha$ -carboxysomes (fig. S6A) (28). The positive response of  $\epsilon_P$  to increasing  $\text{CO}_2$  concentrations indicates that transport limitation is a controlling factor in  $\beta$ -cyanobacterial carbon isotope fractionation, as has been well established for photosynthetic eukaryotes (fig. S6B) (28, 40).

In WT *Synechococcus* 7002,  $\epsilon_P$  values show a negative covariation with the inverse of dissolved  $\text{CO}_2$  concentrations ( $R^2 = 0.96$ ; figs. S5 and S6), further confirming similarities between cyanobacterial and algal net carbon isotope fractionation. Although the  $\Delta ccm$  mutant did not grow over the full range of experimental  $\text{CO}_2$  concentrations, it exhibits a 2.5-fold larger decrease in  $\epsilon_P$  values over the same drop in  $\text{CO}_2$  concentrations when compared to WT ( $\approx 1.8\text{‰}$  versus  $\approx 0.7\text{‰}$  from 107 to 36  $\times$  PAL; Fig. 2B and fig. S5). These different  $\text{CO}_2$  responses suggest that different mechanisms control  $\text{CO}_2$  transport to rubisco in the  $\Delta ccm$  mutant and WT strains.

To explore the isotopic response of the  $\Delta ccm$  mutant to varying  $\text{CO}_2$  concentrations, we used a one-dimensional (1D) reaction-diffusion model, in which rubisco is uniformly distributed throughout the cytosol (31). This model quantifies the isotopic consequences of the competition between a purely diffusional supply of  $\text{CO}_{2(\text{aq})}$  to the site of carbon fixation and  $\text{CO}_2$  fixation into biomass, using three

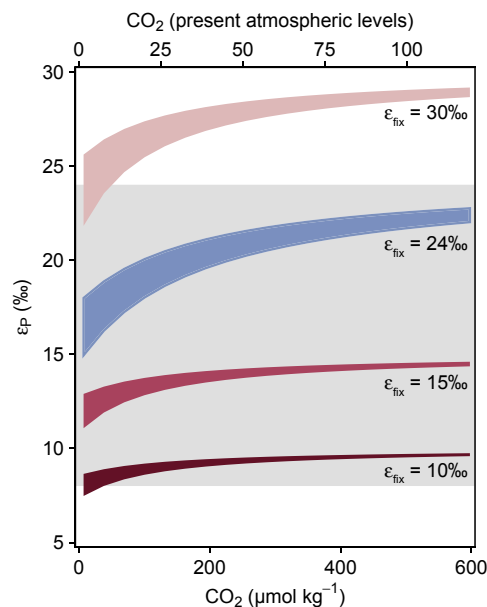
interdependent parameters: (i) the proportion of cellular surface area available for diffusion, (ii) the diffusion coefficient for  $\text{CO}_{2(\text{aq})}$  into the cell, and (iii) the distance over which  $\text{CO}_{2(\text{aq})}$  diffuses into the cell until it meets a free rubisco and is fixed (see the SM for detailed model description).

Modeled  $\epsilon_P$  values for the  $\Delta ccm$  mutant increase with respect to  $\text{CO}_2$  concentrations (from  $\sim 20\text{‰}$  at 36  $\times$  PAL to  $\sim 22\text{‰}$  at 107  $\times$  PAL) with a slightly nonlinear functional dependence (Fig. 2B). Training the model on measured physiological parameters for the  $\Delta ccm$  mutant illustrates the inefficiency of carbon fixation by rubisco relative to a purely diffusional supply of  $\text{CO}_{2(\text{aq})}$ . To reproduce our  $\epsilon_P$ - $\text{CO}_2$  observations,  $\approx 70$  to 90% of the carbon brought into a cyanobacterium without a carboxysome must be lost through back diffusion. This “leakiness” is calculated as the difference between the gross diffusive flux of  $\text{CO}_2$  into the cell and the net rate of  $\text{CO}_2$  fixation into biomass. The inability of the  $\Delta ccm$  mutant to grow at  $\text{CO}_2$  levels below 36 PAL during our experiments (Fig. 2B) was likely due to a combination of factors limiting the intracellular accumulation of  $\text{CO}_2$ , including the leakiness of the cell and the lack of an encapsulated carbonic anhydrase to convert accumulated  $\text{HCO}_3^-$  into  $\text{CO}_2$  at the site of carbon fixation.

## DISCUSSION

The distribution of  $\epsilon_P$  values extracted from the middle Proterozoic sedimentary record span a range of 8 to 24 $\text{‰}$  (95th percentile; Fig. 2A). If cyanobacteria accounted for the majority of primary production at this time, as is commonly asserted, then they should be able to produce a similar range of  $\epsilon_P$  values. Our simulations of a previously characterized mat system (26) suggest that net carbon isotope fractionations by cyanobacteria in benthic settings may only account for the lower 25% of the middle Proterozoic  $\epsilon_P$  distribution (Fig. 2A). Here, we show that net carbon isotope fractionation by  $\beta$ -cyanobacteria without carboxysomes only covers 13% of the middle Proterozoic  $\epsilon_P$  distribution (Fig. 2A). In contrast, the  $\epsilon_P$  range that we determined for planktic cyanobacteria with  $\beta$ -carboxysomes covers  $>90\%$  of the middle Proterozoic distribution, suggesting that this physiology, in the appropriate ecological niches, could be responsible for a large proportion of Proterozoic primary production (Fig. 2A).

To understand whether evolutionary differences between extant and ancestral rubiscos might allow for  $\beta$ -cyanobacteria without carboxysomes to produce the full middle Proterozoic  $\epsilon_P$  range, we used the  $\Delta ccm$  model to calculate the  $\epsilon_P$  relationships that might characterize  $\beta$ -cyanobacteria lacking carboxysomes with ancestral rubisco under middle Proterozoic  $\text{CO}_2$  levels. We incorporated middle Proterozoic estimates of  $\text{O}_2$  concentrations [0.1 to 10% PAL, compiled in (8)] in these model simulations as well. Although the timing of evolutionary changes within the rubisco phylogeny remains unconstrained (41), maximum carboxylation rates for ancestral variants of form 1B rubisco are  $\sim 50$  to 70% of their modern equivalents, while the corresponding Menten constants for  $\text{CO}_{2(\text{aq})}$  are  $\sim 40$  to 80% of their modern equivalents (42). Over a wide range of dissolved  $\text{CO}_2$  and  $\text{O}_2$  concentrations relevant to the Proterozoic ocean, our calculations suggest that a cyanobacterium without carboxysomes will exhibit a limited range of whole-cell  $\epsilon_P$  values ( $< \sim 10\text{‰}$ ; Fig. 3 and the SM). While lower  $\text{O}_2$  concentrations slightly contract the range of  $\epsilon_P$  values (by  $\sim 3\text{‰}$ ) relative to those accessible at higher  $\text{O}_2$  concentrations, the primary control seems to be the mismatch between a fast rate of



**Fig. 3. Modeled relationships between  $\epsilon_P$  and  $\text{CO}_2$  concentration for  $\beta$ -cyanobacteria without a CCM incorporating estimated middle Proterozoic  $\text{O}_2$  levels [0.1 to 10% PAL (8)].** The gray band represents the estimated middle Proterozoic distribution of  $\epsilon_P$  values (95th percentile; 8 to 24‰). The blue field represents calculations extending the observed fractionation by the  $\Delta\text{ccm}$  mutant across possible Proterozoic  $\text{CO}_2$  and  $\text{O}_2$  levels. The red fields represent calculations incorporating the measured kinetics of ancestral form 1B rubisco (table S3) (42) and the full range of known intrinsic isotope effects for rubisco ( $\epsilon_{\text{fix}} = 10, 15, \text{ and } 30\text{‰}$ ; the SM).

$\text{CO}_2$  supply by diffusion and a slower rate of  $\text{CO}_2$  fixation, which restricts the accessible range of net carbon isotope fractionation across all  $\text{CO}_2$  levels in the modeled environment (Fig. 3). In this model, the absolute value of each  $\epsilon_P$  range is set by the intrinsic carbon isotope fractionation factor assumed for rubisco ( $\epsilon_{\text{fix}}$ ; Fig. 3 and the SM). We note that resurrected forms of ancient rubisco have not yet been isotopically characterized. However, it appears that the lack of carboxysomes, rather than how reconstructed rubiscos ultimately fractionate carbon isotopes, restricts any one example of this physiological state from producing the full middle Proterozoic  $\epsilon_P$  distribution.

The middle Proterozoic  $\epsilon_P$  distribution ultimately reflects the interaction between the mode of carbon fixation and  $\text{CO}_2$  supply for middle Proterozoic autotrophs. Estimates of middle Proterozoic atmospheric  $p\text{CO}_2$  values range from 1 to 100 PAL (8), but the temporal and spatial resolution of these estimates is extremely coarse. The middle Proterozoic  $\epsilon_P$  distribution encompasses a variety of marine environments and atmospheric conditions over the course of 800 million years, and therefore,  $p\text{CO}_2$  and dissolved  $\text{CO}_2$  could have exhibited wide variation in time and space over this interval. Estimates of  $p\text{CO}_2$  over the past  $\approx 70$  million years, for example, span a relative range of  $\approx 150$ -fold (60 to 8900 ppm by volume; <https://www.paleo-co2.org>), while dissolved  $\text{CO}_2$  in the modern ocean varies over a relative range of  $\approx 370$ -fold [8 to 2900  $\mu\text{mol kg}^{-1}$  (43)]. If atmospheric or marine  $\text{CO}_2$  in the middle Proterozoic varied similarly then planktic cyanobacteria with  $\beta$ -carboxysomes could produce the full range of middle Proterozoic  $\epsilon_P$  values because of the strong dependence of their net carbon isotope fractionation on  $\text{CO}_2$  concentrations (Fig. 2).

Although this inference does not rule out alternate forms of carbon fixation, the ranges of  $\epsilon_P$  values produced by other plausible middle

Proterozoic primary producers appear to be more restricted even when large variations in middle Proterozoic  $\text{CO}_2$  concentrations are considered. In the case of  $\beta$ -cyanobacteria lacking CCMs, this is due to the slow rate of  $\text{CO}_2$  fixation relative to the fast supply of  $\text{CO}_2$  by diffusion, which restricts the  $\epsilon_P$  response across different  $\text{CO}_2$  concentrations (Fig. 3). Anoxygenic phototrophs lack carboxysomes (44), suggesting that their isotopic fractionation may show a similar lack of sensitivity to  $\text{CO}_2$  concentrations as the  $\Delta\text{ccm}$  mutant investigated here. In cyanobacterial mats, limited  $\text{CO}_2$  supply appears to restrict  $\epsilon_P$  to low values except, perhaps, when  $\text{CO}_2$  levels are  $>300$  PAL (27). Hypotheses that call on different carbon fixation modes to explain the middle Proterozoic  $\epsilon_P$  distribution would therefore require the fortuitous preservation of the products of carbon fixation by a diversity of different primary producers.

We recognize that we cannot exclusively rule out these diversity hypotheses, but the genetic, biochemical, environmental, and physiological evidence discussed here points toward a prominent role for ancestral cyanobacteria with  $\beta$ -carboxysome-based CCMs in the middle Proterozoic biosphere. A Paleoproterozoic (or earlier) origin for the CCM in cyanobacteria is consistent with taphonomic inferences of late Mesoproterozoic biomineralization by CCM-bearing cyanobacteria (45). Cyanobacterial CCMs increase the access of rubisco to  $\text{CO}_2$  to mitigate the enzyme's dual-substrate specificity for both  $\text{CO}_2$  and  $\text{O}_2$  [e.g., (46)]. Under the  $\text{O}_2$ -to- $\text{CO}_2$  ratios found in modern environments, competition between carboxylation and oxygenation reactions is metabolically expensive and imposes a wasteful loss of fixed carbon (47). Although Proterozoic  $p\text{CO}_2$  estimates are higher than modern, spanning  $\sim 1$  to 100 PAL [compiled in (8)], the jump in atmospheric  $\text{O}_2$  across the GOE (1, 48) increased the ratio of  $\text{O}_2$  to  $\text{CO}_2$  up to 100 million-fold (Fig. 1). These enhanced ratios were sustained throughout the Proterozoic at values at least four orders of magnitude greater than at the end of the Archean.

The transition to higher  $\text{O}_2$ -to- $\text{CO}_2$  ratios in the Proterozoic marine environment would have increased  $\text{O}_2$ -to- $\text{CO}_2$  ratios within Proterozoic cyanobacteria (49). The carboxysome may therefore have been an evolutionary innovation in response to extreme environmental oxygenation across the GOE. Despite being the principal component of the CCM in all cyanobacteria today, the carboxysome's original function may have been to shield rubisco from  $\text{O}_2$  (50), after which it was repurposed as a CCM. This proposed function is consistent with predictions of limited  $\text{CO}_2$  and  $\text{O}_2$  permeation through the central pores of carboxysomal shell proteins (33). Early encapsulation inside of a dysoxic carboxysome could further explain why the specificity for  $\text{CO}_2$  versus  $\text{O}_2$  is lower in cyanobacterial form 1B rubisco than in form 1B rubisco from Archaeplastida (51), despite a common lineage [e.g., (19)] and over a billion years of shared environmental history (21).

Whether or not the carboxysome originated as an  $\text{O}_2$ -exclusion mechanism, its carbon isotope consequences appear to reach back at least 1.8 Ga (Fig. 2). Paleontological interpretations of ancestral cyanobacteria have long been rationalized in terms of morphological and local ecological stasis on geological time scales [e.g., (52)]. The observations reported here extend this working hypothesis of stasis to levels of biological organization—from the global marine ecosystem down to the organellar and, perhaps, biochemical realms—that have not been previously accessible to paleontological insight (17, 19). When viewed in terms of the comprehensive nature of the Proterozoic carbon isotope record, this suggests that, like in the modern ocean, pelagic cyanobacteria were an important component

of Proterozoic marine primary productivity. If Proterozoic cyanobacteria were not strictly benthic forms restricted to littoral environments, then a range of hypotheses for limited primary productivity can be ruled out, from environmental hypotheses that rely on an inaccessible pelagic photic zone (53, 54) to evolutionary hypotheses that posit a planktic lifestyle as a derived trait (18, 55). The possibility that Proterozoic cyanobacteria so closely resembled an extant model cyanobacterium opens the door to direct testing of other hypotheses for limiting primary productivity [e.g., (9–12)] through new experiments in comparative physiology and competition under proposed Proterozoic environmental regimes. Cyanobacterial stasis in terms of ecology, morphology, cytology, and biochemistry may have been the foundation behind low Proterozoic productivity (7). The progressive increase of productivity through time could represent a stepwise scaling (56) away from this continuously maintained cyanobacterial state through the introduction of new avenues of primary production in the oceans (19) and, eventually, on land.

## MATERIALS AND METHODS

### Middle Proterozoic $\epsilon_P$ values

Our statistical simulations were based on bootstrap resampling of a curated dataset of  $\delta^{13}\text{C}$  values of carbonate minerals and TOC in 1.0- to 1.8-Ga-old sedimentary rocks (24). We sampled uniform distributions representing possible C isotope fractionation during the conversion and preservation of dissolved  $\text{CO}_2$  as carbonate minerals and primary biomass as TOC. The distribution of equilibrium isotope effects between  $\text{CO}_2(\text{aq})$  and  $\text{HCO}_3^-$  ( $\epsilon_{\text{HCO}_3^- - \text{CO}_2(\text{aq})}$ ) ranged from 8.9 to 11.7‰ (57) assuming photic zone temperatures of 3° to 30°C [e.g., (38)]. Experimentally determined kinetic isotope effects associated with the precipitation of calcite and aragonite relative to  $\text{HCO}_3^-$  ( $\epsilon_{\text{cc-HCO}_3^-}$ ) ranged from 0.8 to 3.3‰ (58). Carbon isotope fractionations associated with secondary biological processes such as heterotrophic consumption of primary organic matter ( $\epsilon_{\text{reworking}}$ ) ranged from 0 to 1.5‰ (23). Full simulations are detailed in the SM.

### Culturing and isotope assays

*Synechococcus* sp. strain PCC 7002 (*Synechococcus* 7002) and a previously engineered  $\Delta\text{ccm}$  mutant strain lacking a carboxysome were grown in A+ media, at 37°C under saturating light levels of  $\sim 227 \pm 5 \mu\text{mol photons m}^{-2} \text{s}^{-1}$  provided by cool-white fluorescence lamps. Cultures were grown in 125-ml conical flasks with foam stoppers (Jaece Industries Identi-plug), continuously shaking, in an incubator that kept headspace  $\text{CO}_2$  constant by continuous replacement with a mixture of  $\text{CO}_2$  and air during each experiment. Headspace  $\text{CO}_2$  varied across three experimental conditions: 0.04% (v/v)  $\text{CO}_2$  (air), 1% (v/v)  $\text{CO}_2$ , and 3% (v/v)  $\text{CO}_2$ , corresponding to  $\text{CO}_2(\text{aq})$  concentrations of 7, 180, and 538  $\mu\text{mol kg}^{-1}$ , respectively. At each  $\text{CO}_2$  condition, strains were acclimated through the serial inoculation of four consecutive cultures. Each culture grew to an optical density at 730 nm of  $\sim 0.2$  before inoculating the next culture with 1 to 3% of the final cell density and harvesting biomass. Harvested biomass was kept at  $-70^\circ\text{C}$ , then centrifuged, and washed twice with ultrapurified water before isotopic analysis. Carbon isotope compositions of biomass were determined by first combusting samples in a Thermo Fisher Scientific FlashEA under a flow of He gas. The resultant  $\text{CO}_2$  was analyzed with a Thermo Fisher Scientific Delta V Isotope Ratio Mass Spectrometer in continuous-flow mode. Carbon isotope compositions are expressed as the relative per mil difference

between the ratio of  $^{13}\text{C}/^{12}\text{C}$  in the sample ( $^{13}\text{C}/^{12}\text{C}_{\text{sample}}$ ) and a standard of Vienna Pee Dee Belemnite ( $^{13}\text{C}/^{12}\text{C}_{\text{VPDB}}$ ). Headspace  $\text{CO}_2$  gas was purified and analyzed with a Thermo Fisher Scientific 253+ Isotope Ratio Mass Spectrometer in dual-inlet mode.

### One-dimensional reaction-diffusion model

A full model description is in the SM. We used a 1D model of steady-state diffusion of  $\text{CO}_2$  between an infinite extracellular source and an intracellular sink to represent rubisco-catalyzed entry of  $\text{CO}_2$  into the Calvin-Benson cycle. A fixed distance separates the  $\text{CO}_2$  source and enzymatic sink. Both the diffusive transport and the sink reaction are isotopically selective. Independent model inputs include the carbon fixation rates observed for the  $\Delta\text{ccm}$  mutant grown under 1 and 3%  $\text{CO}_2$  headspace, the calculated concentration of dissolved  $\text{CO}_2$ , and fractionation factors for form 1B rubisco ( $\epsilon_{\text{fix}}$ ) and diffusion of  $\text{CO}_2$  in solution ( $\epsilon_{\text{diff}}$ ). The model has three free parameters: (i) the intracellular distance over which  $\text{CO}_2(\text{aq})$  diffuses, (ii) the intracellular diffusion coefficient for  $\text{CO}_2(\text{aq})$ , and (iii) the proportion of cellular surface area available for diffusion. We “trained” the model by selecting interdependent sets of these three parameters that could reproduce experimental  $\epsilon_P$  values at the observed carbon fixation rates in the  $\Delta\text{ccm}$  mutant. In the trained model, we additionally used previously characterized kinetics of extant rubiscos and reconstructed ancestral rubiscos to determine possible  $\epsilon_P$  values in  $\beta$ -cyanobacteria without a CCM over a range of environmental conditions.

## SUPPLEMENTARY MATERIALS

Supplementary material for this article is available at <http://advances.sciencemag.org/cgi/content/full/7/2/eabc8998/DC1>

[View/request a protocol for this paper from Bio-protocol.](#)

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## Carbon isotope evidence for the global physiology of Proterozoic cyanobacteria

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## Supplementary Materials for

### Carbon isotope evidence for the global physiology of Proterozoic cyanobacteria

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## Supplementary Materials

### 1. Estimates of O<sub>2</sub>, CO<sub>2</sub>, and oxygenic productivity through Earth history

In broad brush, the concentration of atmospheric oxygen has increased through Earth history in two steps (Fig. S1). Atmospheric oxygen was negligible in the Archean Eon at  $<10^{-6}$  preindustrial atmospheric levels (PAL = 280 ppm CO<sub>2</sub> and 200,000 ppm O<sub>2</sub>) (4). The Paleoproterozoic accumulation of atmospheric oxygen, the Great Oxidation Event (GOE), was underway by 2.43 Ga, and continued for another  $\approx 200$  million years (60). Atmospheric oxygen concentrations may have reached near modern levels early in the Paleoproterozoic Era at  $\sim 10^{-3}$  PAL before returning to values between  $\sim 10^{-3}$  to  $\sim 10^{-1}$  PAL in the Mesoproterozoic Era (e.g., 1, 4, 8, 61). After land plants arose early in the Paleozoic Era, atmospheric oxygen concentrations increased again from  $10^{-1}$  PAL to approximately present levels with some minor variation afterward (59). In contrast, atmospheric CO<sub>2</sub> concentrations have decreased, perhaps less dramatically, through time. Archean atmospheric CO<sub>2</sub> concentrations of  $\sim 40$ –400 PAL are consistent with atmospheric and general circulation models (7, 8, 39). CO<sub>2</sub> levels declined from potentially  $\sim 10$ –100 PAL at the beginning of the Proterozoic Eon to  $\sim 1$ –20 PAL at the end of the Proterozoic (8). During the Phanerozoic Eon, modeling predicts a general negative covariation in CO<sub>2</sub> concentrations with O<sub>2</sub> concentrations (59).

Environmental, ecological and evolutionary mechanisms have been hypothesized to limit the oxygenic productivity of the marine biosphere on geologic timescales. Some environmental explanations propose nutrient limitation either by phosphorus (9, 62), nitrogen (10, 63), or trace metals (11, 64). Others call on cyanobacterial exclusion from a toxic marine photic zone (53, 54). In ecological scenarios, ecosystem structure is proposed to control oxygen production. Competition between populations of oxygenic and anoxygenic photosynthesizers may have limited oxygenic productivity in the water column

(12, 65). Self-shading within a turbid community of cyanobacteria could have a similar effect on net primary productivity (66). Evolutionary hypotheses ascribe limited oxygenic productivity in the Proterozoic to cyanobacteria with enzymatic (67), metabolic (68), cytological (69), or lifestyle (18, 55) traits of their modern descendants. Although ancestral cyanobacteria play a central role in all potential hypotheses for restricting Proterozoic primary productivity, the geologic record contains limited direct evidence for either their global ecological niche or their intracellular physiology at this time.

## 2. Estimates of Mid-Proterozoic $\epsilon_P$ values by statistical simulation

Statistical simulations provide a way to extract estimated values of  $\epsilon_P$ , the isotopic fractionation between dissolved  $\text{CO}_2$  and biomass produced via photoautotrophic carbon fixation, from the geologic carbon isotope record. We restricted our analysis to the non-transitional middle Proterozoic interval (1.0-1.8 Ga), identified previously with a mean difference method (24), to reduce the confounding effects of non-steady state variation in the carbon isotope record. The entire middle Proterozoic interval was treated as a single time bin as finer time-binning did not affect the results of the statistical analysis (24).

We resampled  $\delta^{13}\text{C}$  values from a previously published, curated dataset (24). In this dataset, middle Proterozoic carbonates from a variety of depositional settings, including open and shallow marine environments, have homogeneous  $\delta^{13}\text{C}$  values with a mean and standard deviation of  $0 \pm 1.5$  ‰ (Fig. S2). This homogeneity likely reflects the geographically uniform nature of the marine bicarbonate reservoir (70). Total organic carbon measurements in this dataset similarly reflect a broad range of marine environments and were filtered to exclude demonstrably authigenic and, importantly (70), thermally altered sediments (24).

Values of  $\epsilon_{\text{TOC}}$ , the carbon isotopic fractionation between carbonate minerals ( $\delta^{13}\text{C}_{\text{carb}}$ ) and total organic carbon ( $\delta^{13}\text{C}_{\text{org}}$ ) preserved in the sedimentary rocks (24) is defined as:

$$(1) \quad \epsilon_{\text{TOC}} \equiv \delta^{13}\text{C}_{\text{carb}} - \delta^{13}\text{C}_{\text{org}},$$

where  $\delta^{13}\text{C}_{\text{carb}} = [({}^{13}\text{C}/{}^{12}\text{C}_{\text{carb}}/{}^{13}\text{C}/{}^{12}\text{C}_{\text{standard}}) - 1] * 1000$  and  $\delta^{13}\text{C}_{\text{org}} = [({}^{13}\text{C}/{}^{12}\text{C}_{\text{org}}/{}^{13}\text{C}/{}^{12}\text{C}_{\text{standard}}) - 1] * 1000$ , both expressed as permil (‰). Bootstrap resampling of middle Proterozoic carbon isotope records yields an  $\epsilon_{\text{TOC}}$  distribution with a mean of 28‰ and a range of 20-35‰ (95<sup>th</sup> percentile; Fig. S2).  $\epsilon_{\text{TOC}}$  values represent the combined effects of a number of processes including *i*) the fractionation between dissolved  $\text{CO}_2$  and the biomass of primary producers ( $\epsilon_P$ ), *ii*) the fractionation between dissolved  $\text{CO}_2$  and carbonate minerals ( $\epsilon_{\text{HCO}_3^- - \text{CO}_2(\text{aq})}$ ,  $\epsilon_{\text{cal/ara} - \text{HCO}_3^-}$ ), *iii*) and the fractionation associated with biological reworking of

organic carbon during export and burial ( $\epsilon_{\text{reworking}}$ ) (23). The isotopic difference between dissolved  $\text{CO}_2(\text{aq})$  and biomass produced via photoautotrophic carbon fixation ( $\epsilon_P$ ) is typically defined by:

$$(2) \quad \epsilon_P = \delta^{13}\text{C}_{\text{CO}_2(\text{aq})} - \delta^{13}\text{C}_{\text{biomass}}.$$

We randomly sampled uniform distributions representing fractionation during the conversion and preservation of dissolved  $\text{CO}_2$  as carbonate rocks and primary biomass as TOC. The distribution of equilibrium isotope effects between  $\text{CO}_2(\text{aq})$  and  $\text{HCO}_3^-$  ( $\epsilon_{\text{HCO}_3^- - \text{CO}_2(\text{aq})}$ ) ranged from 8.9-11.7‰ (57) assuming photic zone temperatures of 3-30°C (e.g., 38). Experimentally determined kinetic isotope effects associated with the precipitation of calcite and aragonite from  $\text{HCO}_3^-$  ( $\epsilon_{\text{cal/ara} - \text{HCO}_3^-}$ ) ranged from 0.8-3.3‰ (58). Fractionations associated with secondary biological processes such as heterotrophic consumption of primary organic matter ( $\epsilon_{\text{reworking}}$ ) ranged from 0-1.5‰ (23) (Fig. S2). Values of  $\delta^{13}\text{C}$  for dissolved  $\text{CO}_2(\text{aq})$  and biomass were calculated according to the following approximations:

$$(3) \quad \delta^{13}\text{C}_{\text{CO}_2(\text{aq})} = \delta^{13}\text{C}_{\text{carb}} - \epsilon_{\text{cal/ara} - \text{HCO}_3^-} - \epsilon_{\text{HCO}_3^- - \text{CO}_2(\text{aq})}$$

$$(4) \quad \delta^{13}\text{C}_{\text{biomass}} = \delta^{13}\text{C}_{\text{org}} - \epsilon_{\text{reworking}}$$

The resulting distribution of  $\epsilon_P$  values, calculated according to Eq. 2, are offset from the  $\epsilon_{\text{TOC}}$  distribution by  $\approx 12\text{‰}$ , with a mean of 16‰ and a  $2\sigma$  range of 8-24‰ (Fig. S2). The statistical simulations were run over 10,000 bootstrap trials.

### 3. Estimates of modern mat $\epsilon_P$ values by statistical simulation

We additionally used these techniques to estimate  $\epsilon_P$  values from a published characterization of a modern cyanobacterial mat system (26). This mat system was grown and sampled in an experimental pond with a depth of <1 m. The pond was artificially stratified with an anoxic bottom layer of highly concentrated brine and an upper layer diluted with freshwater. We used the same conversion to calculate  $\delta^{13}\text{C}_{\text{CO}_2(\text{aq})}$  values from observed  $\delta^{13}\text{C}_{\text{carb}}$  values, drawing from uniform distributions representing  $\epsilon_{\text{HCO}_3^- - \text{CO}_2(\text{aq})}$  at observed pond temperatures of 23-57°C and  $\epsilon_{\text{cal/ara} - \text{HCO}_3^-}$  (58) (Eq. 3; see previous section for full simulation description). We did not include fractionations associated with secondary biological processes ( $\epsilon_{\text{reworking}}$ ) due to the young age of the mat, the anoxic bottom layer, and absence of burial in the system. However, due the small range of  $\epsilon_{\text{reworking}}$  (0-1.5‰), the resulting distribution is relatively insensitive to this

parameter. Statistical simulations of carbon isotope fractionation in this previously characterized mat system yielded an  $\varepsilon_P$  distribution with a range of 4-13‰ (95<sup>th</sup> percentile) and a median value of 8.5‰ (Fig. S3), likely reflecting the impact of limited carbon transport into this hypersaline environment.

#### 4. Culturing Description

Culturing experiments used wild-type (WT) *Synechococcus* sp. strain PCC 7002 (*Synechococcus* 7002) and a  $\Delta ccm$  mutant strain lacking a carboxysome. The  $\Delta ccm$  strain was generated by transforming WT cells in exponential growth phase with 0.5 ng/mL of plasmid DNA, containing a kanamycin resistance cassette flanked by 750 bp homology arms for recombination into the *ccm* locus (36). After incubation at 30°C in constant illumination ( $\sim 150 \mu\text{mol photons m}^{-2} \text{ s}^{-1}$ ) for 24 hours, transformed cells were selected for with 100  $\mu\text{g/mL}$  kanamycin on A+ media solidified with Bacto Agar (1%; w/v) in 3% (v/v) CO<sub>2</sub>. Individual colonies were patched onto new plates and tested for segregation by PCR, using primer pairs specific to either the transformed (primers KAMo0113 and EBJp0048) or WT (primers EBJp0048 and EBJp0006) genome. Presence of the PCR product specific to the transformed genome and absence of the PCR product specific to the WT genome was used as an indicator of full segregation. The 5'-3' sequences for primers are

KAMo0113: CGACTGAATCCGGTGAGAAT,

EBJp0048: CGGTGGAGACGATGATCCG,

and EBJp0006: ATAGGTTCTGAATTGTTCTACTTCTTCGGTGT.

For experiments on carbon isotope fractionation, all cultures grew in A+ media (71) at 37°C under saturating light levels of  $\sim 227 \pm 5 \mu\text{mol photons m}^{-2} \text{ s}^{-1}$  provided by cool-white fluorescence lamps.

Cultures were grown in 125 ml conical flasks with foam stoppers (Jaece Industries Identi-plug<sup>TM</sup>), continuously shaking, in an incubator that kept headspace CO<sub>2</sub> constant by continuous replacement with a mixture of CO<sub>2</sub> and air during each experiment. Headspace CO<sub>2</sub> varied across three experimental conditions: 0.04% (v/v) CO<sub>2</sub>, 1% (v/v) CO<sub>2</sub>, and 3% (v/v) CO<sub>2</sub>, corresponding to CO<sub>2(aq)</sub> concentrations of 7, 180, and 538  $\mu\text{mol kg}^{-1}$ , respectively. The measured pH of the cultures (8.1 in air, 7.3 in 1% CO<sub>2</sub>, and 6.7 in 3% CO<sub>2</sub>) and the headspace *p*CO<sub>2</sub> were used to calculate dissolved CO<sub>2</sub> via the csys program (72) adapted for the R statistical computing environment. Dissolved CO<sub>2</sub> concentrations calculated for the growth medium ranged from 7  $\mu\text{mol kg}^{-1}$  in air (1 × PAL) to 538  $\mu\text{mol kg}^{-1}$  in 3% CO<sub>2</sub> (107 × PAL). We plot both headspace CO<sub>2</sub> levels and dissolved CO<sub>2</sub> concentrations due to the influence of temperature, salinity, and pH on dissolved CO<sub>2</sub> concentrations.

We attempted to grow the  $\Delta ccm$  mutant lacking carboxysomes at both 0.5% (v/v) and 0.8% (v/v) CO<sub>2</sub>, corresponding to 18 × PAL and 30 × PAL. However, under our experimental conditions, the mutant strain was only able to grow at 1% CO<sub>2</sub> (36 × PAL) or greater, consistent with the previously observed

inability of this high-CO<sub>2</sub>-requiring phenotype to grow in air (31). Strains were acclimated to each CO<sub>2</sub> condition over the course of four serially-transferred cultures. Cultures grew to an OD<sub>730 nm</sub> of ~0.2 before inoculating the next acclimation with 1-3% of the final cell density and harvesting biomass (Fig. S4). Harvested biomass was kept at -70°C. Prior to isotopic analysis, biomass was centrifuged and washed twice with ultrapurified water.

For cultures, values of  $\epsilon_P$  were calculated relative to external  $\delta^{13}\text{C}$  values for CO<sub>2(aq)</sub> according to:

$$(5) \quad \epsilon_P = 1000 \left[ \frac{(\delta^{13}\text{C}_{\text{CO}_2(\text{aq})} + 1000)}{(\delta^{13}\text{C}_{\text{biomass}} + 1000)} - 1 \right]$$

The net specific growth rate of each culture,  $\mu$  (h<sup>-1</sup>), was calculated according to:

$$(6) \quad \mu = \frac{\ln(\text{cell density}_{\text{final}}) - \ln(\text{cell density}_{\text{initial}})}{(t_{\text{final}} - t_{\text{initial}})}$$

Initial or final cell densities are in cells ml<sup>-1</sup> and  $t_{\text{final}} - t_{\text{initial}}$  is the duration of each batch culture in hours. Carbon fixation rates ( $\mu\text{g C cell}^{-1} \text{ s}^{-1}$ ) were calculated by multiplying the carbon content per cell ( $\mu\text{g C cell}^{-1}$ ) by the net specific growth rate in s<sup>-1</sup>:

$$(7) \quad C_{\text{fix}} = C_{\text{cell}} \cdot \mu$$

Growth parameters for all cultures are plotted in Fig. S5 and reported in Table S1. Doubling times for the wild-type strain ranged from  $5.1 \pm 0.7$  h grown under air,  $3.6 \pm 0.6$  h grown under 1% CO<sub>2</sub> and  $4.0 \pm 0.3$  h grown under 3% CO<sub>2</sub> on average. Doubling times for the  $\Delta\text{ccm}$  mutant grown under equivalent headspaces were slightly greater (slower growth rates) under 1% CO<sub>2</sub> ( $5.5 \pm 1.0$  h) and slightly smaller (faster growth rates) under 3% CO<sub>2</sub> ( $3.5 \pm 0.7$  h). Carbon fixation rates for the wild-type strain were  $67 \pm 21$ ,  $98 \pm 36$ , and  $106 \pm 30$  fg C cell<sup>-1</sup> h<sup>-1</sup> grown under air, 1%, and 3% CO<sub>2</sub>, respectively (Table S1). Carbon fixation rates for the  $\Delta\text{ccm}$  mutant were  $105 \pm 33$  and  $166 \pm 44$  fg C cell<sup>-1</sup> h<sup>-1</sup> grown under 1% and 3% CO<sub>2</sub>, respectively. The carbon fixation rates for wild-type strain in air ( $67 \pm 21$  fg C cell<sup>-1</sup> h<sup>-1</sup>) are ~3× previously published carbon fixation rates for *Synechococcus* 7002 (73).

## 5. Carbon isotope measurements

The carbon isotope composition of cyanobacterial biomass was measured in the Earth Systems Stable Isotope Lab at the University of Colorado Boulder. Samples were oxidized and combusted with a

Thermo Scientific FlashEA and the resultant CO<sub>2</sub> was analyzed with a Thermo Scientific Delta V Isotope Ratio Mass Spectrometer. Ratios are expressed as <sup>13</sup>C/<sup>12</sup>C in units of relative per mil (‰) difference between the sample (R<sub>sample</sub>) and a standard (R<sub>standard</sub>) of Vienna Pee Dee Belemnite (VPDB) according to:

$$(8) \quad \delta^{13}\text{C} = [({}^{13}\text{C}/{}^{12}\text{C}_{\text{sample}}/{}^{13}\text{C}/{}^{12}\text{C}_{\text{standard}}) - 1] * 1000 (\text{‰})$$

Acetanilide (University of Indiana;  $\delta^{13}\text{C}$  of -29.53‰; weight % C of 71.09) was used to correct for linearity and drift. Acetanilide was additionally used as a monitoring standard and a bovine gelatin (pugel), from University of California Santa Cruz ( $\delta^{13}\text{C}$  of -12.62‰; weight % C 44.02) as a discrimination standard.

Isotopic analysis of CO<sub>2</sub> gas was conducted in the Earth Systems Stable Isotope Lab at the University of Colorado Boulder. The CO<sub>2</sub> was analyzed with a Thermo Scientific 253+ Isotope Ratio Mass Spectrometer. The gas was analyzed for four runs with seven cycles at 12.5 volts on 44 m/z with an integration time of 26 s per cycle. Ratios are expressed as <sup>13</sup>C/<sup>12</sup>C in units of relative per mil (‰) difference between the sample (R<sub>sample</sub>) and a standard (R<sub>standard</sub>) of VPBD. The gas was run against an in-house standard that has been corrected against several commercially purchased Oztec bottles of varying compositions.

## 6. 1D reaction-diffusion model

We used a one-dimensional (1D) model of steady-state diffusion of CO<sub>2</sub> between an infinite source and a sink to represent rubisco-catalyzed entry of CO<sub>2</sub> into the Calvin-Benson cycle (Fig. S7). A fixed distance, L, separates the CO<sub>2</sub> source and enzymatic sink. Both diffusion and the sink reaction are isotopically selective according to  $\epsilon_{\text{diffusion}}$  and  $\epsilon_{\text{fix}}$ , respectively (Table S2). Here,  $\epsilon_{\text{fix}}$  refers to the kinetic fractionation factor associated with rubisco (=  $\epsilon_{\text{rubisco}}$ ). Steady-state diffusion in 1D is described by:

$$(9) \quad D \times \frac{d^2C}{dz^2} = 0,$$

where D is the diffusion coefficient, C is the concentration of dissolved CO<sub>2</sub> (= <sup>13</sup>CO<sub>2</sub> + <sup>12</sup>CO<sub>2</sub>), and z is the distance coordinate between source and sink. Eq. 9 is subject to the boundary conditions at the edges of the domain  $0 \leq z \leq L$ :

$$(10) \quad C(z = 0) = C_0$$

$$(11) \quad -D \times \frac{dC}{dz}(z = L) = J_{\text{fix}}$$

$C_0$  represents the initial concentration of  $\text{CO}_{2(aq)}$ , and  $J_{\text{fix}}$  ( $=C_{\text{fix}}/[\text{SA} \times f_{\text{SA}}]$ ; Table S2) represents the carbon fixation rate normalized to the fraction ( $f_{\text{SA}}$ ) of the surface area (SA) of the cell available to  $\text{CO}_2$  transport. The final solution for  $C$  as a function of distance has the form:

$$(12) \quad C(z) = C_0 - \frac{J_{\text{fix}}}{D} \times z$$

In this expression,  $C_0$  is in units of  $\mu\text{g } \mu\text{m}^{-3}$ ,  $D$  is in units of  $\mu\text{m}^2 \text{ s}^{-1}$ ,  $z$  is in units of  $\mu\text{m}$  and  $J_{\text{fix}}$  is in units of  $\mu\text{g C } \mu\text{m}^{-2} \text{ s}^{-1}$ . The per-cell carbon fixation rates reflected in  $J_{\text{fix}}$  are calculated by dividing the measured carbon fixation rates in  $\mu\text{g C cell}^{-1} \text{ s}^{-1}$  by the surface area of a cell, which we define as a rod with a radius of  $0.5 \mu\text{m}$  and a length of  $2 \mu\text{m}$ , as well as the fraction of the surface area of the cell available to  $\text{CO}_2$  transport. (Table S2).

Similar equations apply for the heavy isotope ( $^{13}\text{C}$ ):

$$(13) \quad {}^{13}\text{D} \times \frac{d^2}{dz^2} ({}^{13}\text{C}) = 0$$

Eq. 13 is subject to the boundary conditions at the edges of the domain  $0 \leq z \leq L$ :

$$(14) \quad {}^{13}\text{C}(z = 0) = {}^{13}\text{C}_0$$

$$(15) \quad -{}^{13}\text{D} \times \frac{d^{13}\text{C}}{dz} (z = L) = {}^{13}J_{\text{fix}}$$

The final solution for  $^{13}\text{C}$  as a function of distance has the form:

$$(16) \quad {}^{13}\text{C}(z) = {}^{13}\text{A} + {}^{13}\text{B} \times z$$

$^{13}\text{A}$  and  $^{13}\text{B}$  are integration constants defined by the boundary conditions such that

$$(17) \quad {}^{13}\text{A} = {}^{13}\text{C}_0$$

$$(18) \quad -{}^{13}\text{D} \times {}^{13}\text{B} = {}^{13}J_{\text{fix}}.$$

We defined  $^{13}J_{fix}$  through:

$$(19) \quad \alpha_{fix} = \frac{^{13}J_{fix}}{^{12}J_{fix}} \bigg/ \frac{^{13}C_L}{^{12}C_L}$$

Here  $\alpha_{fix}$  is the fractionation factor associated with CO<sub>2</sub> fixation by rubisco (Table S2), while  $^{13}C_L$  and  $^{12}C_L$  represent the concentrations of each isotopologue of CO<sub>2</sub> at  $z = L$ . Applying the trace isotope abundance approximation ( $C_L \approx ^{12}C_L$ ,  $J_{fix} \approx ^{12}J_{fix}$ ) leads to:

$$(20) \quad ^{13}J_{fix} = \alpha_{fix} \times J_{fix} \times \frac{^{13}C_L}{C_L}.$$

We combined Eqs. 18 and 20

$$(21) \quad -^{13}D \times ^{13}B = \alpha_{fix} \times J_{fix} \times \frac{^{13}C_L}{C_L}$$

to solve for  $^{13}B$ :

$$(22) \quad ^{13}B = - \frac{\alpha_{fix} \frac{J_{fix}}{D} ^{13}C_0}{\alpha_{diff} C_L + \alpha_{fix} \frac{J_{fix}}{D} L}$$

Here  $\alpha_{diff}$  is the fractionation factor associated with CO<sub>2</sub> diffusion to the site of fixation, which is defined as  $\alpha_{diff} = ^{13}D / ^{12}D$  (Table S2). The final solution for  $^{13}C$  is then:

$$(23) \quad ^{13}C(z) = ^{13}C_0 \left( 1 - \frac{\alpha_{fix} \frac{J_{fix}}{D}}{\alpha_{diff} C_L + \alpha_{fix} \frac{J_{fix}}{D} L} z \right).$$

We then calculated  $\varepsilon_P$  as the difference between the carbon isotopic composition of fixed carbon and dissolved CO<sub>2</sub> using:

$$(24) \quad \varepsilon_P = \frac{\left( \frac{^{13}C_0}{^{12}C_0} - \alpha_{fix} \frac{^{13}C_L}{^{12}C_L} \right)}{\frac{^{13}C_0}{^{12}C_0}} \times 1000.$$

Independent model inputs are: the carbon fixation rates observed for the  $\Delta ccm$  mutant grown under 1% and 3% CO<sub>2</sub> headspace, the concentration of dissolved CO<sub>2</sub> calculated from  $pCO_2$  and the culture pH using the csys program (72), and the fractionation factor for diffusion of CO<sub>2</sub> in water ( $\varepsilon_{diff} = [\alpha_{diff} - 1] \times 1000$ ; Table S2). Two previously published values of the fractionation factor for form IB

rubisco ( $\epsilon_{fix} = [\alpha_{fix} - 1] \times 1000$ ) in cyanobacteria are  $22 \pm 0.2\text{‰}$  (74) and  $20.9 \pm 0.8$  (75), both measured on rubisco from in *Synechococcus* PCC 6301. Here, we set  $\epsilon_{fix}$  at 24.3‰, the maximum whole-cell  $\epsilon_P$  value we observed for *Synechococcus* 7002.

The model has three free physiological parameters: i) the intracellular distance over which  $\text{CO}_{2(aq)}$  diffuses ( $z$ ); ii) the intracellular diffusion coefficient for  $\text{CO}_{2(aq)}$  ( $D$ ); and iii) the proportion of cellular surface area available for diffusion ( $f_{SA}$ ). Free parameters covary over the constrained ranges in Table S2. We trained the model by selecting the interdependent parameter sets that were able reproduce experimental  $\epsilon_P$  values observed in the *Synechococcus* 7002  $\Delta ccm$  mutant (within the  $\approx 0.65\text{‰}$  data spread for a given experimental condition) at carbon fixation rates observed for both 1% and 3% headspace  $\text{CO}_2$  (36 and 107 PAL, respectively). We then continued calculations and model predictions incorporating all interdependent parameter sets [ $z$ ,  $D$ ,  $f_{SA}$ ] that could reproduce experimental  $\epsilon_P$  values.

In order to apply the 1D reaction-diffusion model across varying  $\text{CO}_2$  levels, we incorporated a previously published model of carbon fixation rate (76):

$$(25) \quad C_{fix} = \frac{k_{cat} N_{sites} C}{C + K_{m,CO_2} \left( 1 + \frac{O}{K_{m,O_2}} \right)}$$

Here,  $C_{fix}$  represents the carbon fixation rate as a function of  $\text{CO}_2$  ( $C$ ) and  $\text{O}_2$  ( $O$ ) concentrations and rubisco kinetic parameters (values listed in Table S2). We fit this theoretical relationship through the observed range of carbon fixation rates using the kinetic parameters for Form IB rubisco in *Synechococcus* 7002 and number of rubisco sites per cell as  $N_{sites} = 2.3 \times 10^5$ – $3.4 \times 10^5$  (Fig. S8). However, this fit could be similarly achieved by scaling the rubisco kinetics to account for differences between calculated and measured carbon fixation rates. The trained dependence of carbon fixation rate on  $\text{CO}_2$  concentration is then used as a model input to predict  $\epsilon_P$  values at varying  $\text{CO}_2$  and  $\text{O}_2$  levels (Fig 2B). Modeled  $\epsilon_P$  values increase with  $\text{CO}_2$  concentration (Fig. S8). Values of  $\epsilon_P$  show a greater dependence on carbon fixation at low  $\text{CO}_2$  concentrations (Fig. S8B).

The trained model and corresponding carbon fixation rate calculation can then be used to predict  $\epsilon_P$  values for a cyanobacterium without a carboxysome grown under varying  $\text{CO}_2$  and  $\text{O}_2$  conditions with independently derived kinetic parameters for rubisco (see next section).

## 7. Evolutionary changes in rubisco kinetics

The  $\Delta ccm$  model exercise suggests that, without a CCM, fractionation by extant cyanobacteria with Form IB rubisco can only explain a limited portion of estimated middle Proterozoic  $\epsilon_P$  values. The  $\Delta ccm$

model incorporates the kinetics and fractionation for modern Form 1B rubisco (Table S2). Here, we explore how evolutionary changes in rubisco kinetics could impact these modeling results.

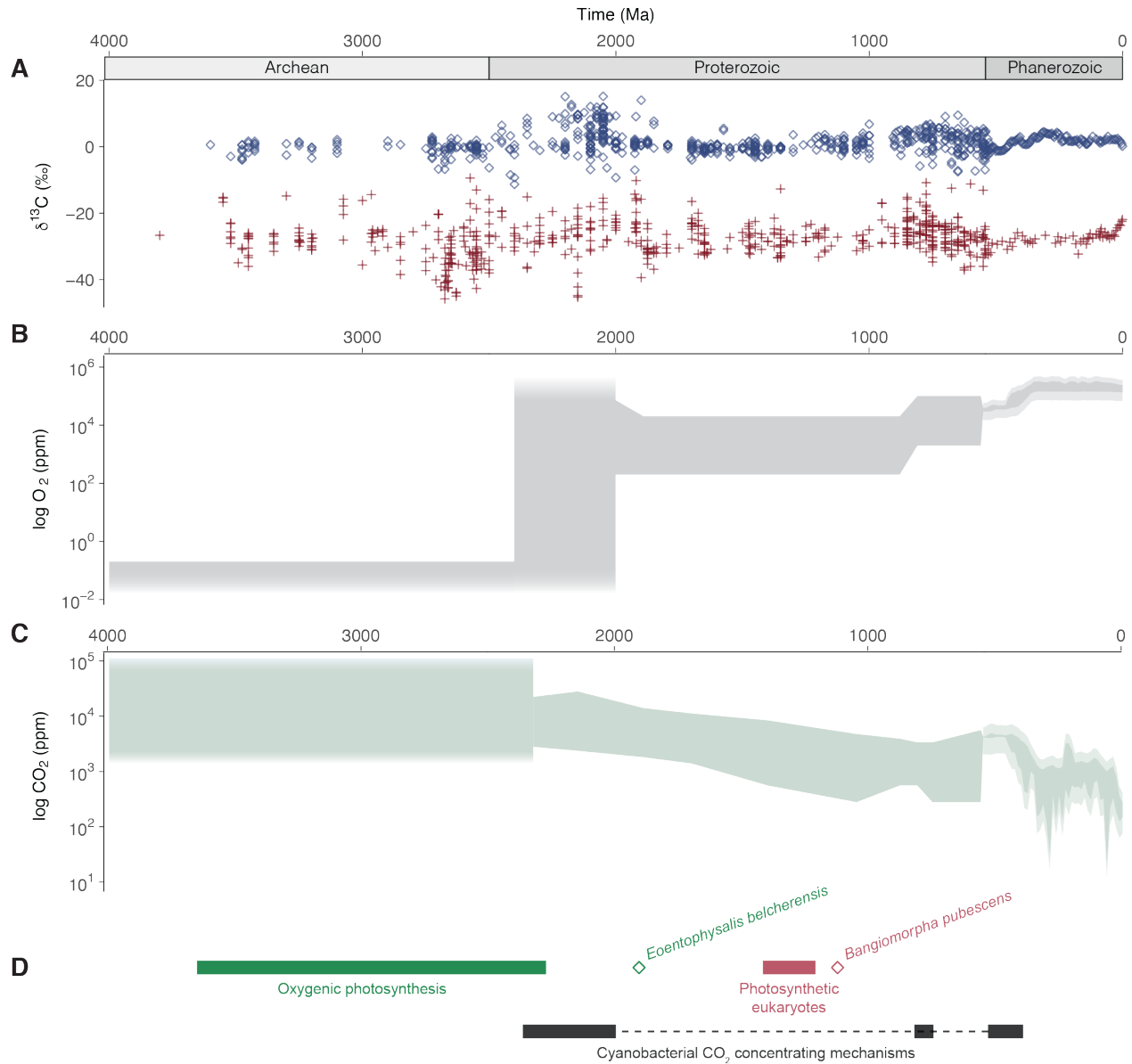
Ancestral sequence reconstruction of Form IA and Form IB rubisco suggest that the kinetics of ancestral Form I rubiscos differed from modern rubisco, although the timing of these evolutionary changes remain unconstrained (42). For example,  $k_{cat}$  and  $K_{m, CO_2}$  values for ancestral rubiscos are ~40-80% of modern forms (42). While fractionation factors for these reconstructed rubisco ( $\epsilon_{fix}$ ) forms have not yet been characterized, the kinetics constrain how far  $\epsilon_P$  values can deviate from a maximum  $\epsilon_{fix}$  value and thus constrain the overall range of  $\epsilon_P$  values accessible for a given  $\epsilon_{fix}$  value. Substituting the kinetics associated with ancestral Form 1A and Form 1B rubisco, while keeping the rest of the model inputs constant, results in a 7-8‰ range in  $\epsilon_P$  values for any given value of  $\epsilon_{fix}$  for rubisco (Fig. S9). This range is less than half the 95<sup>th</sup> percentile range (~16‰) from our resampling of the mid-Proterozoic carbon isotope record suggesting that, regardless of the ancestral rubisco  $\epsilon_{fix}$  value, net carbon isotope fractionation by cyanobacteria with these ancestral rubisco forms could not be responsible for the entire range of mid-Proterozoic  $\epsilon_P$  estimates.

We additionally used the measured characteristics of different extant rubisco forms to determine the possible constraints imposed on  $\epsilon_P$  by the relationship between kinetics and fractionation factors in cyanobacteria without a CCM. Model sensitivity runs based on paired inputs of rubisco kinetics compiled in (51) and fractionation factors compiled in (77) show the  $\epsilon_P$  ranges accessed by Form 1A, Form 1B, and Form II rubiscos in cyanobacteria without a CCM (Fig. S9, Table S3). Similar to the results with the  $\Delta ccm$  mutant reported here, Form 1B rubiscos in extant cyanobacteria without a CCM can only reproduce a limited range of mid-Proterozoic  $\epsilon_P$  values (Fig. S9). If Form II rubisco operated in cyanobacteria without a CCM it would be consistent with a similar range of estimates, while Form 1A rubisco would produce a much smaller range of  $\epsilon_P$  values (Fig. S9). Only the kinetics and fractionation factors of Form 1B rubisco from higher plants (*Spinacia oleracea* and *Nicotiana tabacum*) produce  $\epsilon_P$  ranges larger than the middle Proterozoic estimates, if they operated in cyanobacteria without a CCM (Fig. S9).

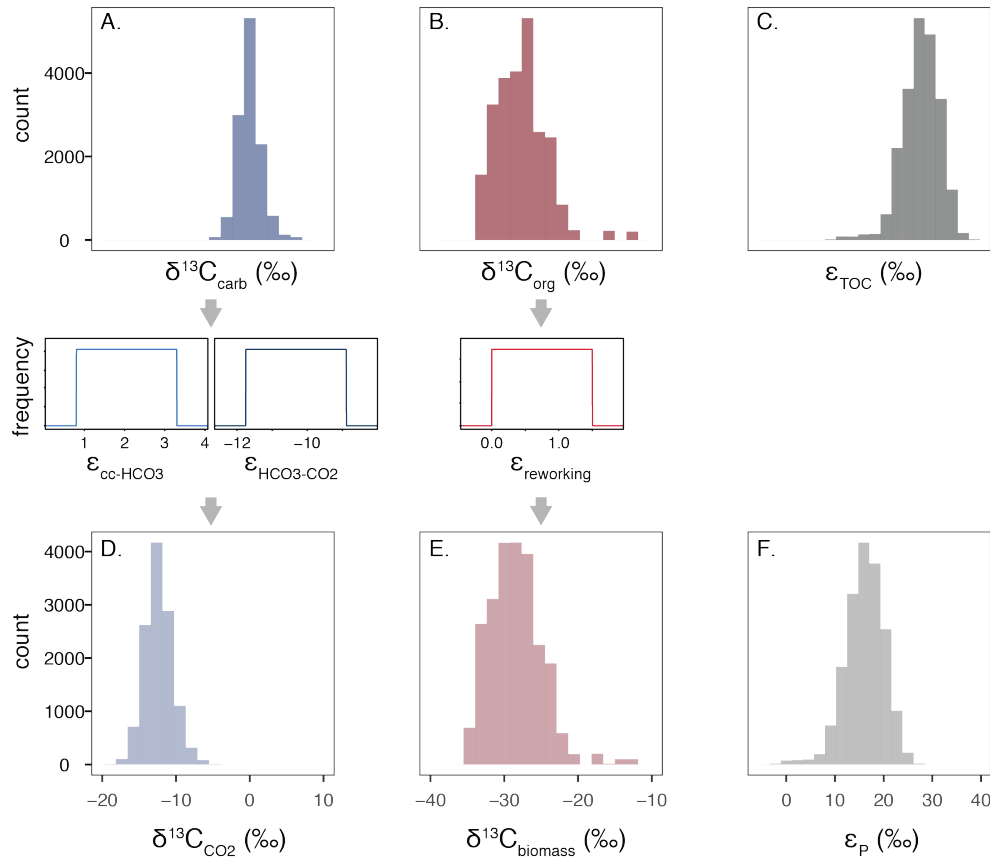
Form 1B rubiscos within extant higher green plants, such as the rubiscos found in *Spinacia oleracea* and *Nicotiana tabacum*, have kinetic parameters and  $\epsilon_{rubisco}$  values that yield substantially larger ranges of  $\epsilon_P$  values when incorporated in the 1-D reaction-diffusion model developed for the  $\Delta ccm$  mutant of *Synechococcus* sp. 7002. This suggests that, if middle Proterozoic cyanobacteria contained rubiscos resembling Form 1B in higher green plants, a CCM would not be required to produce the full distribution of middle Proterozoic  $\epsilon_P$  values. Archaeplastida are extant representatives of a primary endosymbiotic event, in which an original plastid, derived from an ancestral cyanobacterium, was incorporated into a eukaryotic host cell at  $\approx 1.3$  Ga (21). Although it is possible that the kinetics of Form 1B rubiscos from Archaeplastida are a vestigial remnant of this ancient endosymbiotic event, it seems unlikely that the

selective environment provided by the early Proterozoic surface ocean was similar enough to that within green plant chloroplasts to maintain enzymatic stasis over the ensuing billion years. Altogether, this consideration of potential ancestral rubisco kinetics suggests that carbon isotope fractionation by cyanobacteria lacking a CCM is inconsistent with the estimated distribution of Proterozoic  $\epsilon_p$  values.

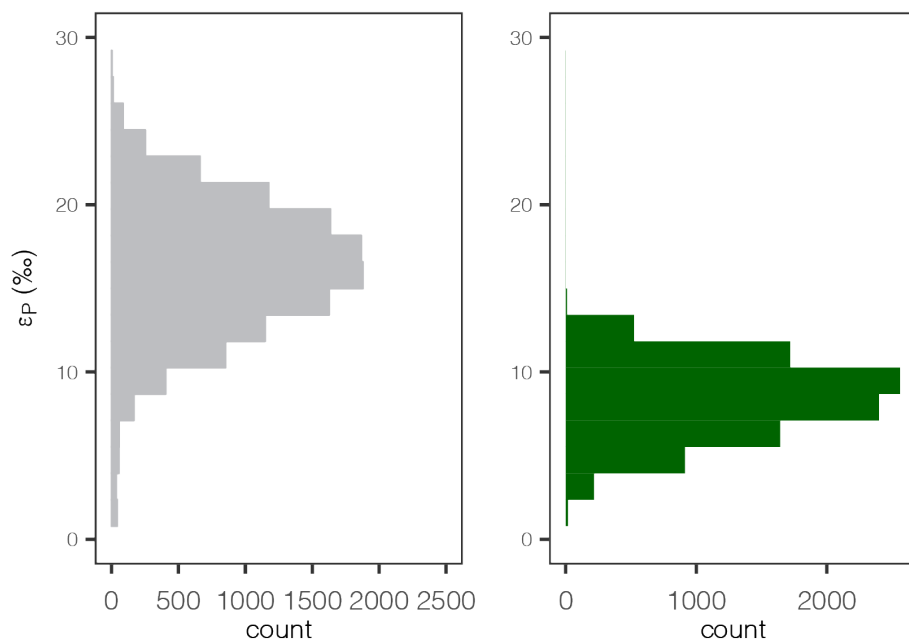
## Supplementary Figures and Tables



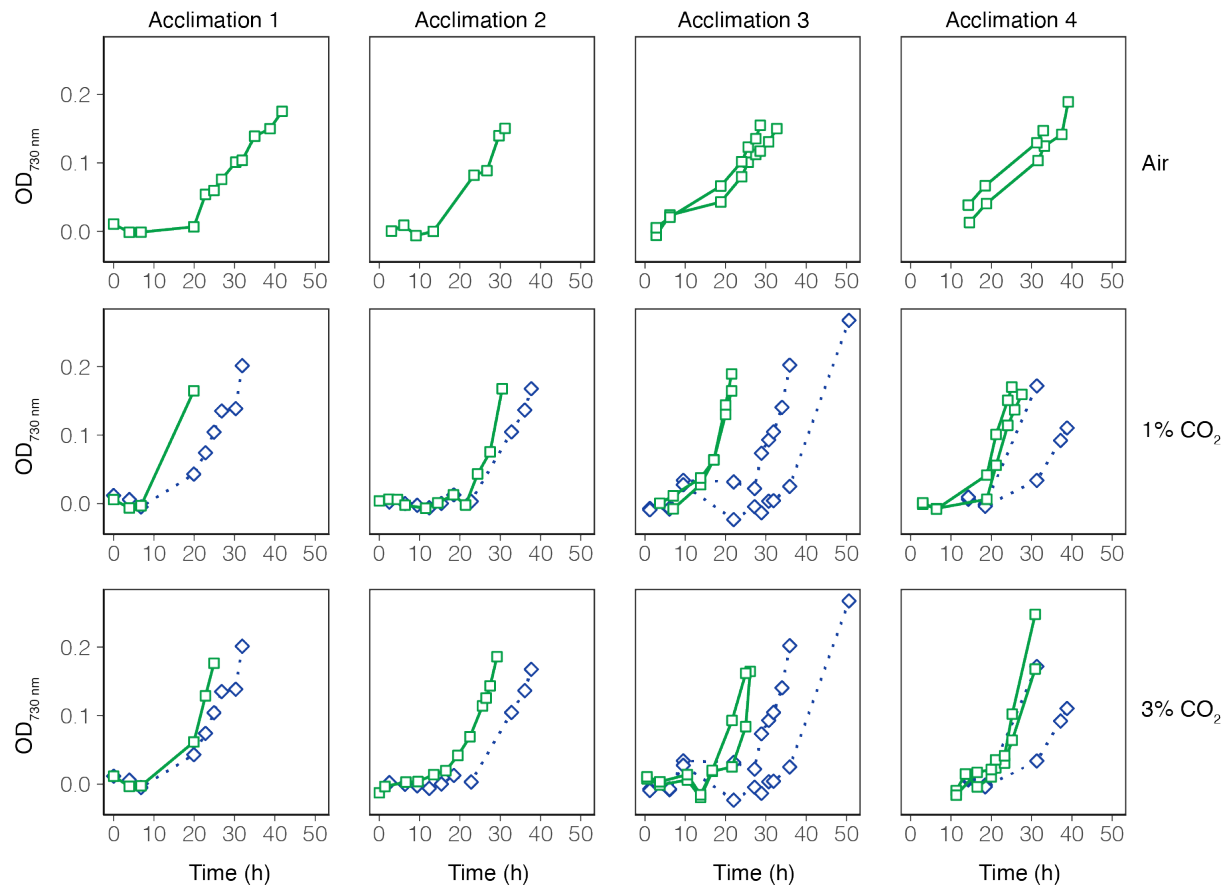
**Fig. S1.** The carbon isotope record and estimated atmospheric and biologic changes through Earth history. **A.** Values of  $\delta^{13}\text{C}$  from carbonate minerals (blue) and  $\delta^{13}\text{C}$  values of total organic carbon (red) are compiled in Krissansen-Totton et al., (24). **B.** Archean  $\text{O}_2$  estimates restrict  $p\text{O}_2$  to  $< 10^{-6}$  present atmospheric levels or 2 ppm (4). Proterozoic  $\text{O}_2$  estimates are compiled in Crockford et al., (8). Phanerozoic  $\text{O}_2$  estimates come from proxies and modeling (59). **C.** Archean  $\text{CO}_2$  estimates are from Halevy and Bachan (39). Proterozoic  $\text{CO}_2$  estimates are compiled in Crockford et al., (8). Phanerozoic  $\text{CO}_2$  estimates come from proxies and modeling (59). **D.** Range of estimates for the origin of oxygenic photosynthesis shown as a green bar (2, 3) and the earliest unambiguous cyanobacterial microfossils (*Eoentophysalis belcherensis*) shown as a green diamond (13, 14). Earliest unambiguous fossil photosynthetic eukaryote (*Bangiomorpha pubescens*) shown as red diamond with corresponding molecular clock estimates for the primary plastid endosymbiosis shown as a red bar (21). Proposed dates for the emergence of a cyanobacterial  $\text{CO}_2$  Concentrating Mechanism shown as black bars (e.g., 34).



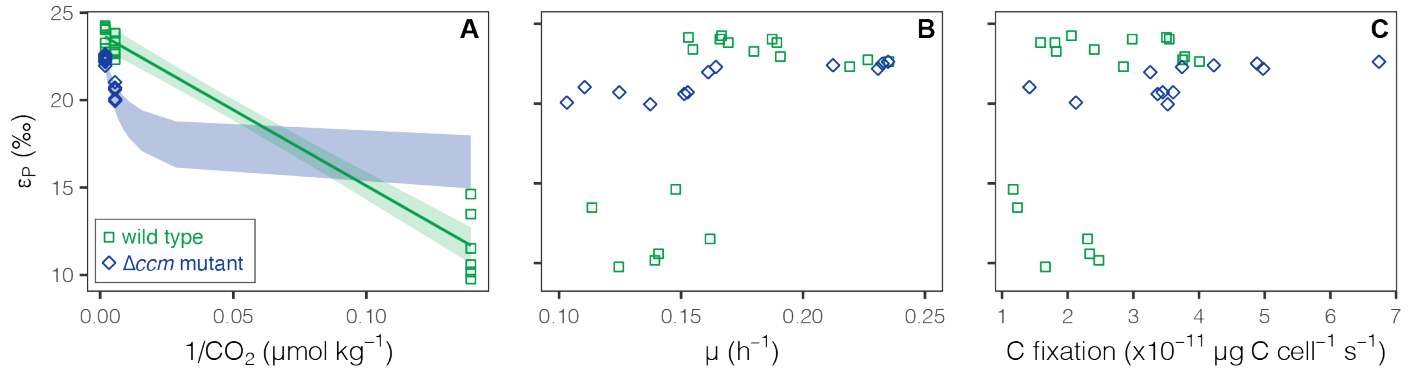
**Fig. S2.** Representation of bootstrap resampling and Monte Carlo simulations. **A.** Resampled  $\delta^{13}\text{C}$  values of carbonates and **B.** organic carbon during the non-transitional period between 1.0-1.8 Ga from Krissansen-Totton et al. (24). **C.** The distribution of  $\epsilon_{\text{TOC}}$  values from the bootstrap resampling, calculated according to Eq. 1. **D.** The distribution of predicted  $\delta^{13}\text{C}$  values of dissolved  $\text{CO}_2$ , calculated by incorporating the isotope effects associated with precipitation of calcite or aragonite from  $\text{HCO}_3^-$  and the conversion between  $\text{HCO}_3^-$  and dissolved  $\text{CO}_2$  (inset uniform distributions  $\epsilon_{\text{cc-HCO}_3}$  and  $\epsilon_{\text{HCO}_3\text{-CO}_2}$ ). **E.** The distribution of predicted  $\delta^{13}\text{C}$  values of primary biomass calculated by incorporating an isotope effect associated with secondary reworking in the water column and sediments (inset uniform distribution  $\epsilon_{\text{reworking}}$ ). **F.** The distribution of predicted  $\epsilon_p$  values calculated according to Eq. 2. See text for simulation details.



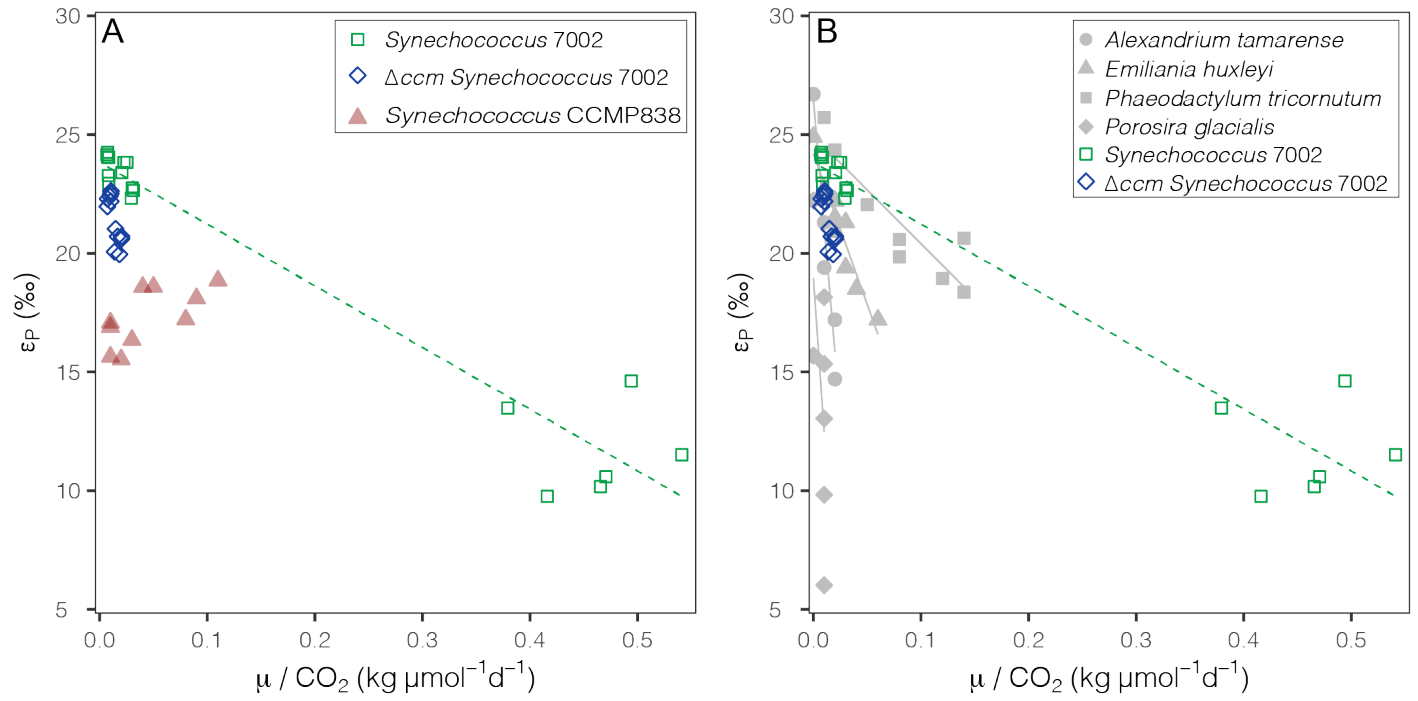
**Fig. S3.** Comparing the  $\epsilon_P$  distribution between the middle Proterozoic (left) and a modern cyanobacterial mat (right). Both distributions are the result of Monte Carlo simulations and bootstrap resampling of sedimentary carbon isotope records (left) and a previously published isotopic characterization of a cyanobacterial mat (right; 26).



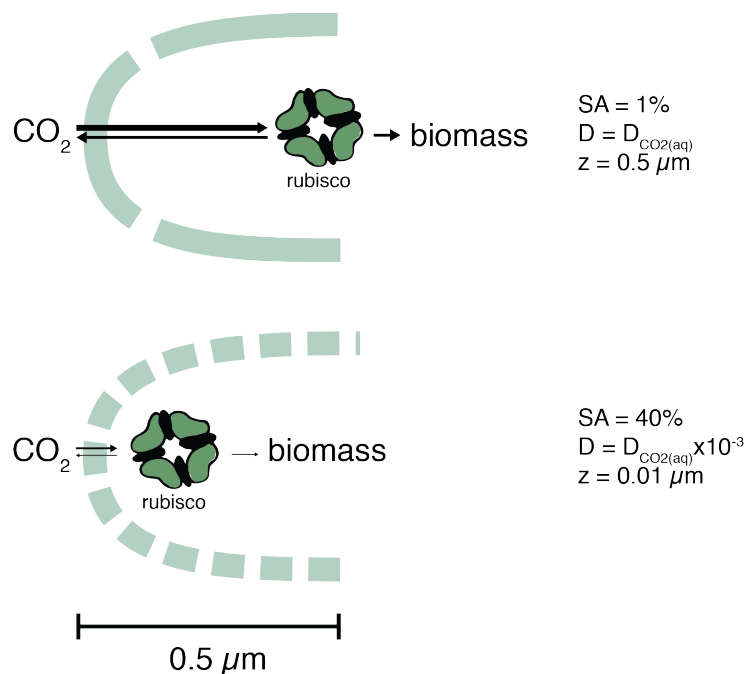
**Fig. S4.** Growth curves for cultures of *Synechococcus* 7002 in air, 1% CO<sub>2</sub>, and 3% CO<sub>2</sub> corresponding to 1, 36, and 107 PAL CO<sub>2</sub>. Green squares and solid lines represent wild-type *Synechococcus* 7002. Blue diamonds and dashed lines represent the  $\Delta ccm$  mutant.



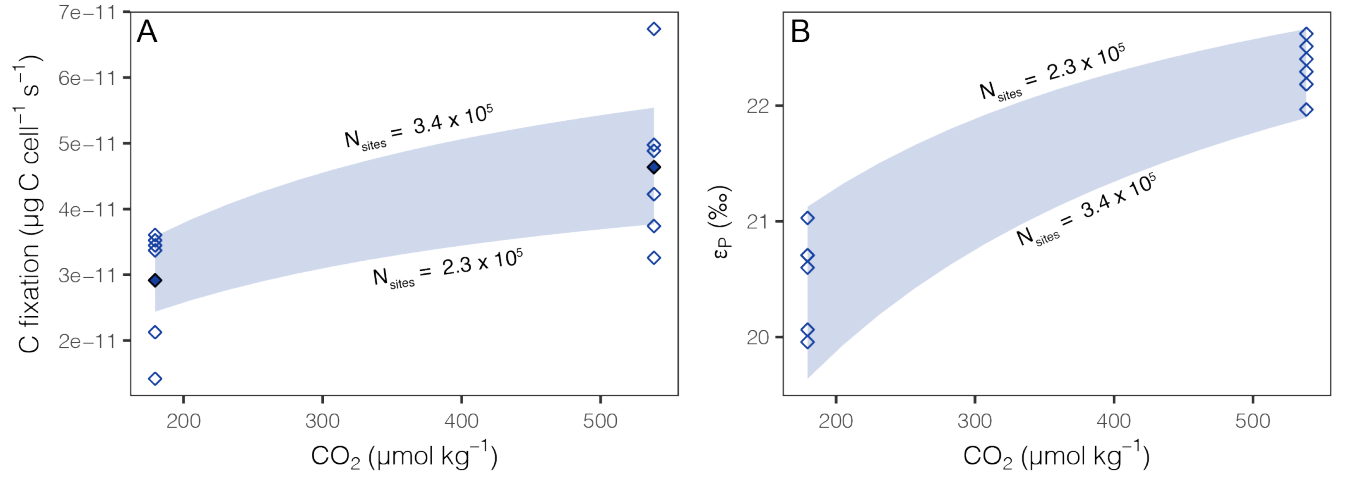
**Fig. S5.** The dependence of  $\epsilon_p$  values observed in *Synechococcus* 7002 cultures on  $\text{CO}_2$  and growth parameters. **A.** Values of  $\epsilon_p$  in the wild-type strain (green symbols) show a linear dependence on  $1/\text{CO}_2$  (green line;  $R^2$ : 0.96, shading represents 95% confidence interval). Values of  $\epsilon_p$  in the  $\Delta ccm$  strain (blue symbols) were used to model the dependence of  $\epsilon_p$  on  $\text{CO}_2$  concentration (blue shading). **B.** Values of  $\epsilon_p$  do not vary systematically with specific growth rate ( $\mu$ ). **C.** The relationship between carbon fixation rate and  $\epsilon_p$  values used to train the  $\Delta ccm$  strain model.



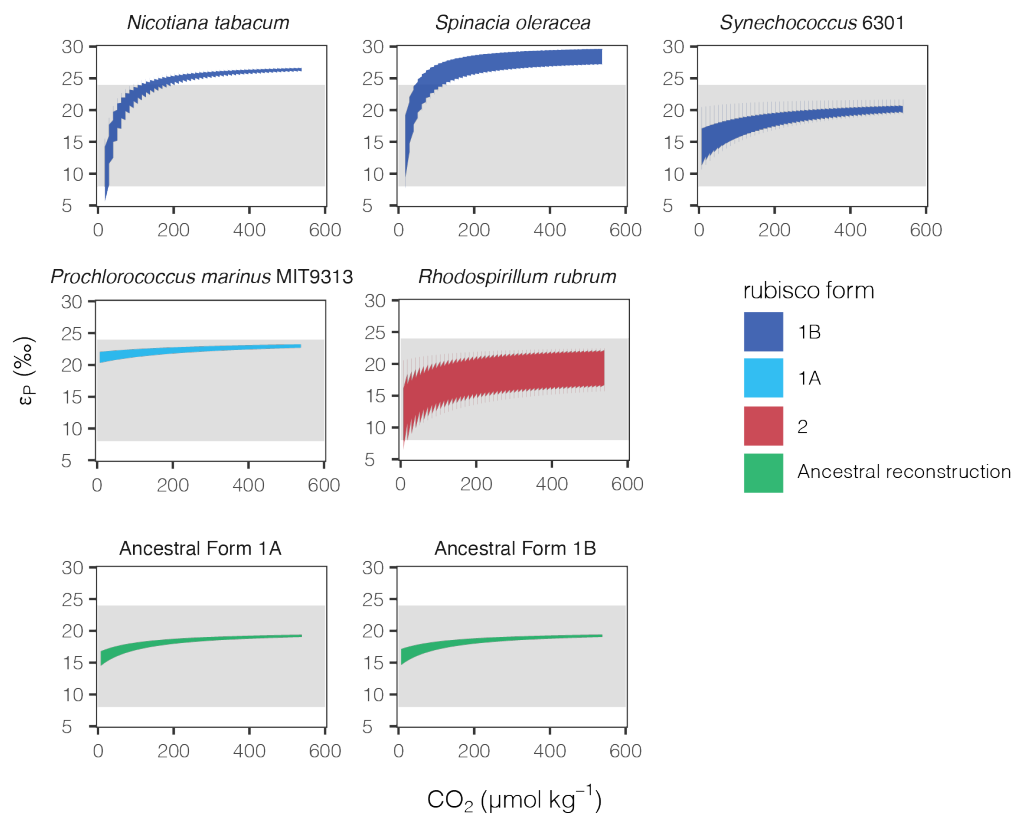
**Fig. S6.** Cyanobacterial and algal  $\epsilon_P$  relationships. **A.** Cyanobacterial  $\epsilon_P$  relationships observed in wild-type *Synechococcus* 7002 and  $\Delta\text{ccm}$  mutant (this study) differ from the  $\epsilon_P$  relationship found in alpha cyanobacterium *Synechococcus* CCMP838 (28). **B.**  $\epsilon_P$  relationships found in wild type *Synechococcus* 7002 and  $\Delta\text{ccm}$  mutant resemble  $\epsilon_P$  relationships in eukaryotic algae (78).



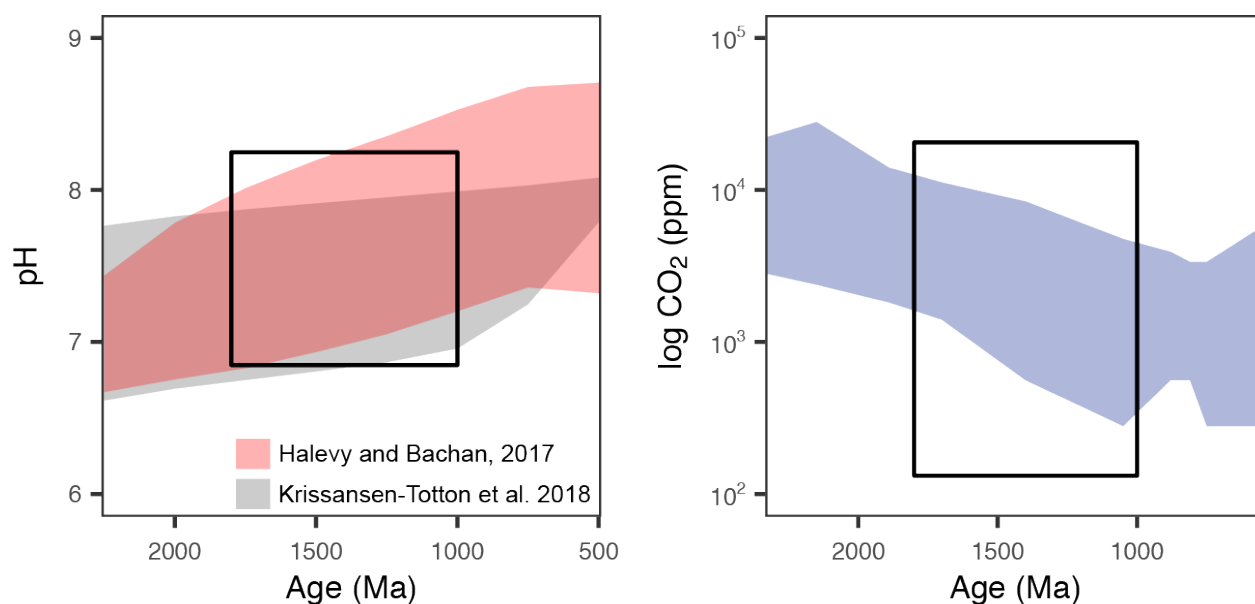
**Fig. S7.** 1D reaction-diffusion model. Fluxes include diffusion of CO<sub>2</sub> into the cell, autotrophic carbon fixation through the Calvin-Benson cycle, and diffusion of CO<sub>2</sub> out of the cell. The three interdependent model parameters are the surface area available for diffusion of CO<sub>2</sub> (SA), the diffusion coefficient for CO<sub>2</sub> at 37°C (D), and the distance to the rubisco active site (z). Example values of these parameters are shown here to illustrate the range of values possible for each parameter.



**Fig. S8.** Carbon fixation rate and  $\epsilon_P$  values as a function of dissolved  $\text{CO}_2$  concentration in the  $\Delta ccm$  mutant. **A:** Open symbols represent measured carbon fixation rate for cultures of the  $\Delta ccm$  mutant grown under a 1% and 3%  $\text{CO}_2$  headspace (36 and 107 PAL  $\text{CO}_2$ ). The shaded area represents the theoretical relationship between carbon fixation rate and  $\text{CO}_2$  concentration from Clark et al. (76) fit to the observed culture data using the number of rubisco sites ( $N_{\text{sites}}$ ). This fit could be similarly achieved by scaling the rubisco kinetics or incorporating an unknown factor to account for differences between the theoretical model and the measured carbon fixation rates. **B:** The carbon fixation rate as a function of dissolved  $\text{CO}_2$  concentration is then used as a model input to predict corresponding  $\epsilon_P$  values.



**Fig. S9.** The modeled relationship between  $\epsilon_P$  values and  $\text{CO}_2$  for both extant rubisco and ancestral reconstructions. The grey band represents the estimated middle Proterozoic distribution of  $\epsilon_P$  (95<sup>th</sup> percentile; 8-24‰). For extant rubisco forms, colored fields represent model results with  $\epsilon_{\text{fix}}$  values from (77) and rubisco kinetics from (51) listed in Table S3. For the ancestral forms, colored fields represent model results with an example  $\epsilon_{\text{fix}}$  value of 20‰ and kinetic parameters from (42). In all cases the model was run over  $\text{O}_2$  concentrations of 0.14-14  $\mu\text{mol l}^{-1}$  (0.1-10% PAL).



**Fig. S10.** Experimental culture conditions projected onto Proterozoic  $p\text{CO}_2$  and pH estimates. The height of the boxes represents the experimental range of pH (left) and dissolved  $\text{CO}_2$  translated into  $p\text{CO}_2$  levels using the culture pH between assumed Proterozoic ocean temperatures of 3–30°C (right). The width of the box represents the middle Proterozoic time period resampled for the statistical simulations of the carbon isotope record. Estimates of Proterozoic pH are 95% confidence intervals from (39) and (38). Estimates of Proterozoic  $p\text{CO}_2$  levels are from (8).

**Supplementary Table 1.** Growth parameters for wild type *Synechococcus* 7002 and  $\Delta ccm$  mutant. Net specific growth rate was calculated according to Eq. S6. Carbon fixation rates ( $C_{fix}$ ) was calculated according to Eq. S7. Analytical error for  $\delta^{13}C_{biomass}$  values is  $\pm 0.1\%$  and  $0.01\%$  for  $\delta^{13}C_{CO_2}$  values. Values of  $\delta^{13}C_{CO_2}$  in air are from <https://www.esrl.noaa.gov/gmd/dv/iadv/index.php> (Niwt Ridge site).

Headspace CO <sub>2</sub> (PAL)	CO <sub>2</sub> ( $\mu\text{mol kg}^{-1}$ )	Strain	Acclimation	Doubling time (h)	$C_{fix}$ (fg C cell <sup>-1</sup> h <sup>-1</sup> )	$\delta^{13}C_{biomass}$ (‰)	$\delta^{13}C_{CO_2}$ (‰)	$\epsilon_P$ (‰)
1	7	WT	1	6.1	44	-22.5	-8.0	13.5 $\pm$ 0.1
1	7	WT	2	4.7	42	-23.5	-8.0	14.6 $\pm$ 0.1
1	7	WT	3	4.9	84	-19.6	-8.0	10.6 $\pm$ 0.1
1	7	WT	3	4.3	83	-20.4	-8.0	11.5 $\pm$ 0.1
1	7	WT	4	5.6	60	-18.7	-8.0	9.8 $\pm$ 0.1
1	7	WT	4	5.0	89	-19.1	-8.0	10.2 $\pm$ 0.1
36	180	WT	1	3.0	144	-50.2	-27.7	22.6 $\pm$ 0.1
36	180	WT	2	4.5	87	-50.8	-27.7	23.4 $\pm$ 0.1
36	180	WT	3	3.1	135	-50.1	-27.7	22.7 $\pm$ 0.1
36	180	WT	3	3.2	103	-49.8	-27.7	22.3 $\pm$ 0.1
36	180	WT	4	4.1	65	-51.1	-27.7	23.8 $\pm$ 0.1
36	180	WT	4	3.7	57	-51.1	-27.7	23.8 $\pm$ 0.1
107	538	WT	1	3.6	136	-63.0	-40.6	22.9 $\pm$ 0.1
107	538	WT	2	4.2	107	-63.9	-40.6	24.0 $\pm$ 0.1
107	538	WT	3	3.9	66	-63.1	-40.6	23.3 $\pm$ 0.1
107	538	WT	3	3.7	128	-63.8	-40.6	24.0 $\pm$ 0.1
107	538	WT	4	4.5	126	-63.8	-40.6	24.1 $\pm$ 0.1
107	538	WT	4	4.2	74	-63.9	-40.6	24.3 $\pm$ 0.1
36	180	$\Delta ccm$	1	4.5	130	-48.3	-27.7	20.7 $\pm$ 0.1
36	180	$\Delta ccm$	2	5.6	124	-48.3	-27.7	20.7 $\pm$ 0.1
36	180	$\Delta ccm$	3	5.1	127	-47.5	-27.7	20.0 $\pm$ 0.1
36	180	$\Delta ccm$	3	6.7	77	-47.7	-27.7	20.1 $\pm$ 0.1
36	180	$\Delta ccm$	4	4.6	121	-48.0	-27.7	20.6 $\pm$ 0.1
36	180	$\Delta ccm$	4	6.3	51	-48.4	-27.7	21.0 $\pm$ 0.1
107	538	$\Delta ccm$	1	4.2	135	-62.4	-40.6	22.3 $\pm$ 0.1
107	538	$\Delta ccm$	2	4.3	117	-61.9	-40.6	22.0 $\pm$ 0.1
107	538	$\Delta ccm$	3	3.3	152	-62.2	-40.6	22.4 $\pm$ 0.1
107	538	$\Delta ccm$	3	3.0	179	-62.1	-40.6	22.2 $\pm$ 0.1
107	538	$\Delta ccm$	4	3.0	176	-62.3	-40.6	22.5 $\pm$ 0.1
107	538	$\Delta ccm$	4	3.0	243	-62.3	-40.6	22.6 $\pm$ 0.1

**Supplementary Table 2.** 1D reaction-diffusion model parameters.

Parameter	Description	Values	Units	Notes
$C_{fix}$	Measured C fixation rate	$1.4 \times 10^{-11}$ - $6.7 \times 10^{-11}$	$\mu\text{g C cell}^{-1} \text{ s}^{-1}$	Grown under 25-75 PAL $\text{CO}_2$
$J_{fix}$	Converted C fixation rate		$\mu\text{g C } \mu\text{m}^{-2} \text{ s}^{-1}$	$C_{fix}/(SA \times fSA)$
$C_0$	Experimental dissolved $[\text{CO}_2]$	$8.8 \times 10^{-14}$ - $6.6 \times 10^{-12}$	$\mu\text{g C } \mu\text{m}^{-3}$	Calculated from $p\text{CO}_2$ and pH using csys (72)
$z$	Distance to the sink	0.01-0.5	$\mu\text{m}$	Based on a cell with a radius of $0.5 \mu\text{m}$
$SA$	Surface area of a rod	6.3	$\mu\text{m}^2$	Based on a rod with a radius of $0.5 \mu\text{m}$ and a length of $2 \mu\text{m}$ representing a cyanobacterial cell
$fSA$	Effective surface area	0.01-0.4	fraction	Proportion of surface area available for diffusion of $\text{CO}_2$
$D$	Diffusion coefficient $\text{CO}_2$	26.9-2690	$\mu\text{m}^2 \text{ s}^{-1}$	Constrained by the diffusion coefficient of $\text{CO}_{2(aq)}$ in water ( $2.69 \times 10^3 \mu\text{m}^2 \text{ s}^{-1}$ at $37^\circ\text{C}$ ) (79)
$\alpha_{diff}$	Fractionation associated with diffusion of $\text{CO}_2$	1.00087		O’Leary et al. (80)
$\alpha_{fix}$	Fractionation associated with rubisco	1.0243		Note 1.
$k_{cat}$	form IB rubisco	11.4	molecules $\text{site}^{-1} \text{ s}^{-1}$	adapted from Clark et al. (76)
$K_{m, \text{CO}_2}$	form IB rubisco	185	$\mu\text{mol l}^{-1}$	adapted from Clark et al. (76)
$K_{m, \text{O}_2}$	form IB rubisco	1300	$\mu\text{mol l}^{-1}$	adapted from Clark et al. (76)
$\text{O}_2$	Concentration of dissolved $\text{O}_2$	143	$\mu\text{mol l}^{-1}$	Calculated. Note 2.
$N_{sites}$	rubisco active sites	$2.3 \times 10^5$ - $3.4 \times 10^5$	number	Note 3.

<sup>1</sup> Previously published fractionation for form IB rubisco in *Synechococcus* PCC 6301 are  $22 \pm 0.2\text{‰}$  (74) and  $20.9 \pm 0.8$  (75). Here, we set  $\epsilon_{fix}$  at  $24.3\text{‰}$ , the maximum whole-cell  $\epsilon_P$  value we observed for *Synechococcus* 7002.

<sup>2</sup> The concentration of dissolved oxygen was calculated from DOTABLES from the USGS (<https://www.usgs.gov/software/dotables>) using growth medium conditions of  $37^\circ\text{C}$ , 0.821 atm, and 32‰ salinity.

<sup>3</sup> The number of rubisco active sites was used to fit the theoretical relationship (76) through the observed range of carbon fixation rates at 36 and 107 PAL. This fit could be similarly achieved by scaling the rubisco kinetics.

**Supplementary Table 3.** Model inputs for rubisco characteristics from extant organisms and ancestral state reconstructions (ASR). Values for  $\varepsilon_{\text{fix}}$  from (77), kinetic parameters for extant rubiscos from (51), kinetic parameters for ancestral rubiscos from (42)

Source	Form	$\varepsilon_{\text{fix}}$	$k_{\text{cat}}$ ( $\text{s}^{-1}$ )			$K_{\text{m CO}_2}$ ( $\mu\text{M}$ )			$K_{\text{m O}_2}$ ( $\mu\text{M}$ )			Taxonomy
ASR	Anc. 1B	--	4.77	±	0.1	113	±	6	2010	±	571	Ancestral
ASR	Anc. 1A	--	4.72	±	0.1	120	±	10	641	±	49	Ancestral
Spinacia oleracea	1B	29.3	2.9	±	0.2	10	±	1	250	±	33	C3 plants
Spinacia oleracea	1B	29.3	3.2	±	0.1	12.1	±	1	574	±	19	C3 plants
Spinacia oleracea	1B	29.3	3.2	±	0.1	12.1	±	1	574	±	19	C3 plants
Spinacia oleracea	1B	29.3	2.7	±	0.1	11	±	2	520	±	25	C3 plants
Nicotiana tabacum	1B	27.4	3.9	±	0.2	9	±	0	292	±	20	C3 plants
Nicotiana tabacum	1B	27.4	3.4	±	0.1	10.7	±	1	295	±	71	C3 plants
Nicotiana tabacum	1B	27.4	3.1	±	0.3	9.7	±	0	283	±	15	C3 plants
Nicotiana tabacum	1B	27.4	3.4	±	0.1	10.7	±	1	295	±	71	C3 plants
Synechococcus 6301	1B	22.0	11.4	±	0.7	185	±	17	1300	±	172	Cyanobacteria
Synechococcus 6301	1B	22.0	12	±	0.7	167	±	15	529	±	69.8	Cyanobacteria
Synechococcus 6301	1B	22.0	11.8	±	0.2	200	±	9	199	±	42	Cyanobacteria
Synechococcus 6301	1B	22.0	2.57	±	0.2	142	±	1	664	±	82	Cyanobacteria
Synechococcus 6301	1B	22.0	3.71	±	0.2	167	±	2	529	±	14	Cyanobacteria
Synechococcus 6301	1B	22.0	11.6	±	0.4	340	±	12	972	±	26	Cyanobacteria
Synechococcus 6301	1B	22.0	9.78	±	0.5	152	±	23	1230	±	135	Cyanobacteria
Prochlorococcus marinus MIT9313	1A	24.0	6.58	±	0.3	309	±	24	1400	±	300	Cyanobacteria
Rhodospirillum rubrum	2	20.4	5.7	±	0.3	89	±	8	406	±	53.6	Alphaproteobacteria
Rhodospirillum rubrum	2	20.4	12.3	±	0.3	149	±	8	159	±	25	Alphaproteobacteria
Rhodospirillum rubrum	2	20.4	4.3	±	0.5	125	±	12	143	±	3	Alphaproteobacteria
Rhodospirillum rubrum	2	20.4	7.3	±	0.3	80	±	11	406	±	48	Alphaproteobacteria

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