



## Full-length article

# Conformationally flexible dendronized cyclotetrameratrylenes (CTTV)s self-organize a large diversity of chiral columnar, Frank-Kasper and quasicrystal phases

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## ABSTRACT

Primary structure endows the tertiary structure while the tertiary structure determines the function of hierarchical self-organizations. This is a well-established fundamental principle both for biological and non-biological synthetic assemblies. The role of the conformational flexibility-rigidity in this process is less understood. Here we select the conformationally flexible monodisperse tetramer of veratrole, 2,3,7,8,12,13,17,18-octamethoxy-5,10,15,20-tetrahydrotrabenzo[a,d,g,j]-cyclododecatetraene, known as cyclotetrameratrylene (CTTV) and use it as an apex model to compare with the less flexible corresponding trimer known as cyclotrimeratrylenes (CTV) and with the rigid triphenylene (Tp) and 1,3,5-trihydroxybenzene (THB) when dendronized with libraries of self-assembling dendrons and with n-alkyl groups. Unexpectedly, a large diversity of chiral helical assemblies including supramolecular columns assembled from chiral spheres and crowns, 12-fold quasi liquid crystals (QLC), Frank-Kasper A15 (space group  $Pm\bar{3}n$ ) but not  $\sigma$  phases (space group  $P4_2/mnm$ ), and supramolecular orientational memory effect (SOM) were observed in the case of the dendronized CTTV. The more rigid structures at the apex provide thermally more stable helical chiral periodic and quasiperiodic self-organizations with low dynamics which may facilitate the freezing of metastable rather than equilibrium structures. Conformational flexibility also changes the structure of the self-organization generated from supramolecular spheres. These experiments indicate that additional studies on the topic of conformational flexibility are desirable.

## Introduction

Self-assembling dendrons, dendrimers and covalent and supramolecular dendronized polymers [1–56] self-organize a large diversity of chiral structures regardless of whether they are hierarchically assembled from chiral or achiral building blocks. Helical chirality and deracemization of a racemic helix [57–61] in crystal state are two of the most fundamental principles that mediate this process both in biological and in synthetic systems. The primary structure of the self-assembling dendrons and dendrimers including their generation number, multiplicity of their branching point, the dynamics and the flexibility of the supramolecular and covalent polymers and oligomers building their apex are key constructs of these helical assemblies. Libraries of self-assembling dendrons and dendrimers with different primary structure but identical apex functionality have been employed to construct a Periodic Table [29] of self-assembling dendrons correlating their primary

structure to their tertiary structure. Different functional groups at the apex have been used to assemble supramolecular dendronized polymers via different H-bonding functional groups, donor-acceptor interactions, ionic-interactions including their solvation by crown-ethers and oligoethylenes [42,47,49–51,56,65] and their correlation has been compared with that of specific covalent polymers. Less is known about the role of the precise degrees of polymerization of covalent dendronized polymers on hierarchical self-organization although living polymerizations have been used [62–67] to correlate the degree of polymerization with tertiary structure. Living polymerizations provide only an average estimate of the degree of polymerization and its correlation to tertiary structure is limited by the very small molar fraction of chains with precise degree of polymerization [68–72]. Triphenylene (Tp) [34] represent a planar trimer of 1,2-dihydroxy benzene while CTV [33] and CTTV [73] represent perfectly monodisperse crown trimers and more conformationally flexible tetramers of veratrole. Therefore, a

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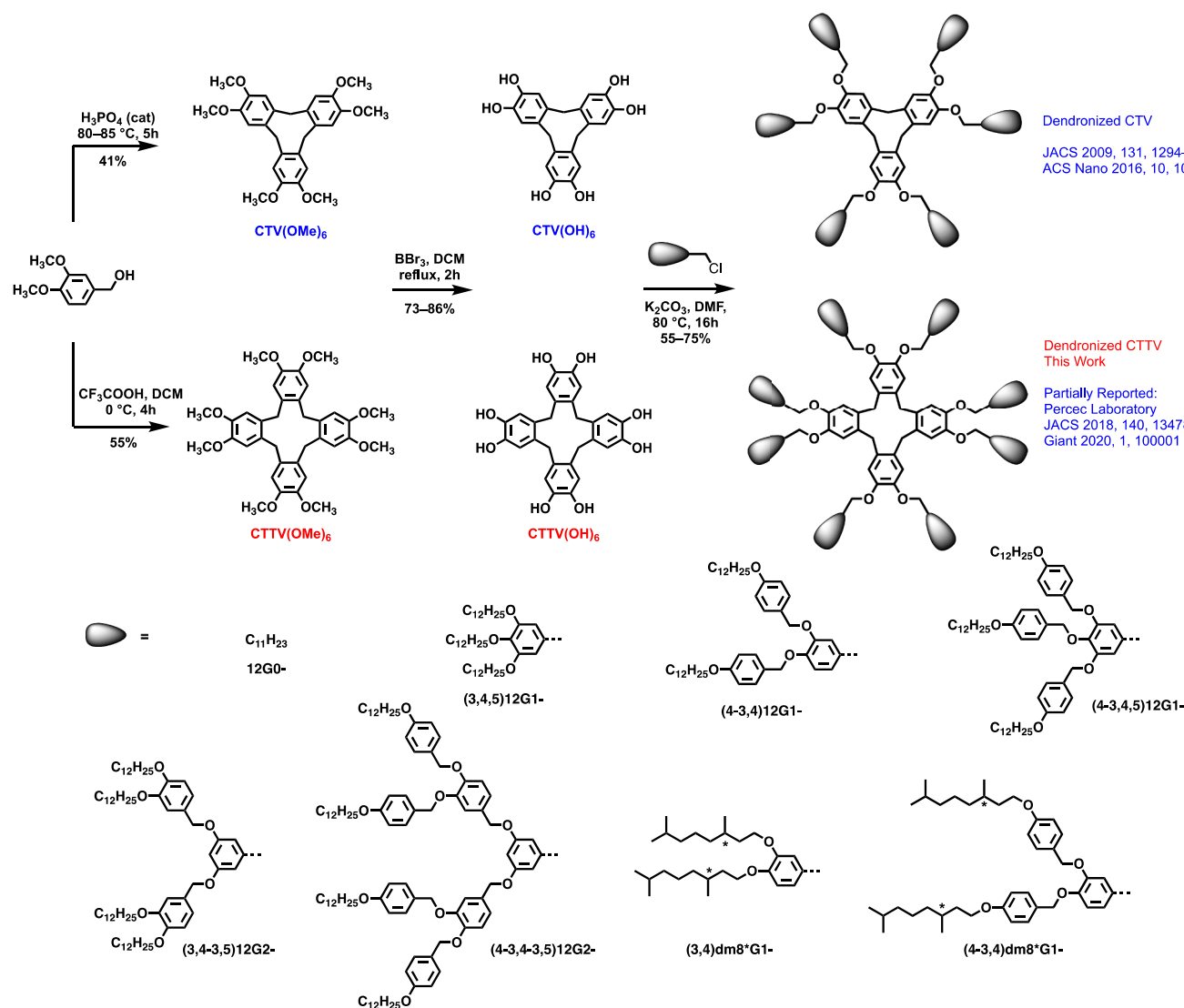


Fig. 1. Synthesis of dendronized CTV and CTTV.

comparison of the conformationally flexible dendronized CTTV with that of the less flexible CTV and Tp would bring some light on the role of conformational flexibility of a dendronized covalent apex on the corresponding tertiary structure. This publication reports the synthesis of *n*-alkyl-substituted and dendronized CTTV, provides the structural analysis of the corresponding self-organizations and compares them with those resulted from the more rigid CTV, triphenylenes and other assemblies to demonstrate the role of the conformational flexibility on their hierarchical self-organization. Two previous discoveries with dendronized CTTV prompted this series of experiments. The first chiral helical columnar hexagonal periodic array self-organized from chiral supramolecular spheres [73] and one of the few supramolecular orientational memory effects (SOM) observed at the transition between columnar hexagonal and Frank-Kasper A15 phase [74–79] prompted the current series of investigations.

## Methods

### Synthesis

The synthesis of the molecules including 2,3,7,8,12,13,17,18-octamethoxy-5,10,15,20-tetrahydrotetrabenzo[a,d,g,j]-cyclododecatetraene, known as cyclotetrameratrylene (CTTV) [80,81],

*n*-dodecyl-substituted CTTV 12G0-CTTV [82], dendronized CTTV including (3,4,5)12G1-CTTV, (4-3,4,5)12G1-CTTV, (4-3,4)12G1-CTTV, (3,4-3,5)12G2-CTTV, (4-3,4-3,5)12G2-CTTV, (3,4)dm8\*G1-CTTV, and (4-3,4)dm8\*G1-CTTV was reported previously by Percec laboratory but with the exception of (3,4)dm8\*G1-CTTV were not characterized [73]. All of them were resynthesized by similar methods with even higher purities than previously reported.

**NMR.** <sup>1</sup>H NMR (500 MHz) and <sup>13</sup>C NMR (126 MHz) spectra were recorded on a Bruker DRX 500 instrument at 300 K using the indicated solvent.

**High-performance liquid chromatography (HPLC).** The purity of all CTTV final compounds was determined by a combination of thin-layer chromatography (TLC) on silica gel coated aluminum plates (with F254 indicator; layer thickness, 200 μm; particle size, 2–25 μm; pore size 60 Å, SIGMA-Aldrich) and high pressure liquid chromatography (HPLC) using THF as mobile phase at 1 mL/min, on a Shimadzu LC-10AT high pressure liquid chromatograph equipped with a Perkin Elmer LC-100 oven (40 °C), containing two Perkin-Elmer PL gel columns of 5 × 10<sup>2</sup> and 1 × 10<sup>4</sup> Å, a Shimadzu SPD-10A UV detector (λ = 254 nm), a Shimadzu RID-10A RI-detector, and a PE Nelson Analytical 900 Series integrator data station.

**Matrix-assisted laser desorption/ionization time of flight (MALDI-TOF) mass spectrometry.** MALDI-TOF mass spectrometry was

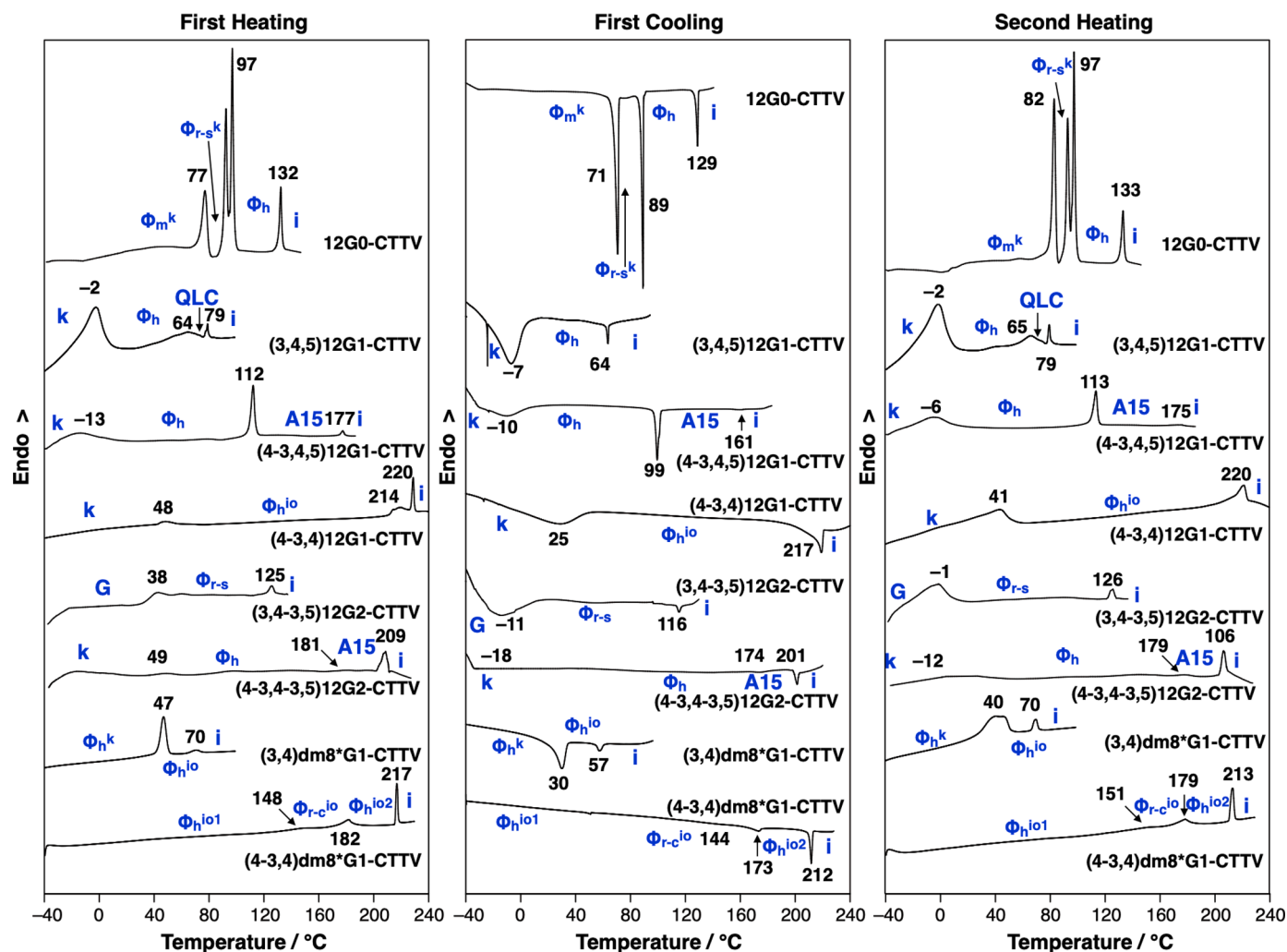


Fig. 2. DSC traces of *n*-alkylated and dendronized CTTV.

performed on PerSeptive Biosystems-Voyager-DE (Framingham, MA) mass spectrometer equipped with a nitrogen laser (337 nm) and operating in linear mode. Internal calibration was performed using Angiotensin II, and Bombesin as standards. The analytical samples were obtained by mixing THF solution of the sample (5–10 mg/mL), and the matrix (2,5-dihydroxybenzoic acid) (10 mg/mL) in a 1:1 to 1:5 v/v ratio. The prepared solution (0.5  $\mu$ L) was loaded on the MALDI plate and allowed to dry at 25  $^{\circ}$ C before the plate was inserted into the vacuum chamber of the MALDI instrument.

**Differential scanning calorimetry (DSC).** Thermal transitions and their associated enthalpy changes were determined on a TA Instruments Q100 differential scanning calorimeter (DSC) equipped with a refrigerated cooling system at 10  $^{\circ}$ C min $^{-1}$  heating and cooling rates. Indium was used as calibration standard. The transition temperatures were calculated as the maxima and minima of their endothermic and exothermic peaks.

**Melting points.** Melting points were measured using a uni-melt capillary melting point apparatus (Arthur H. Thomas Company) and were uncorrected.

**Thermal optical polarized microscopy (TOPM).** The texture was recorded by an Olympus BX51 optical microscope (100 $\times$  magnification) equipped with a Mettler FP82HT hot stage and a Mettler Toledo FP90 Central Processor was used to verify thermal transitions.

**X-ray diffraction (XRD).** X-ray diffraction (XRD) measurements were performed using Cu-K $\alpha$ 1 radiation ( $\lambda$  = 1.542  $\text{\AA}$ ) from a Bruker-Noniun FR-591 rotating anode X-ray source equipped with a 0.2  $\times$

0.2 mm $^2$  filament and operated at 3.4 kW. Osmic Max-Flux optics, and triple pinhole collimation were used to obtain a highly collimated beam with a 0.3  $\times$  0.3 mm $^2$  spot on a Bruker-AXS Hi-Star multiwire area detector. To minimize attenuation, and background scattering, an integral vacuum was maintained along the length of the flight tube, and within the sample chamber. Samples were held in glass capillaries (1.0 mm in diameter), mounted in a temperature-controlled oven (temperature precision:  $\pm$  0.1  $^{\circ}$ C, temperature range from  $-10$  to 210  $^{\circ}$ C). Aligned samples for fiber XRD experiments were prepared using a custom-made extrusion device [57]. The powdered sample ( $\sim$ 10 mg) was heated inside the extrusion device. After slow cooling, the fiber was extruded in the liquid crystal phase, and cooled to 23  $^{\circ}$ C. Typically, the aligned samples have a thickness of 0.3–0.7 mm, and a length of 3–7 mm. All XRD measurements were performed with the aligned sample axis perpendicular to the beam direction. Primary XRD analysis was performed using Datasqueeze (version 3.0.5) [83].

**Molecular modeling.** Molecular modeling, and simulation experiments were performed using Material Studio Modeling (version 5) software from Accelrys. The Forcite module was used to perform the energy minimizations on the supramolecular structures. Correct models must match the experimental density, the lattice dimensions and the original XRD data.

## Results and discussion

Fig. 1 outlines the synthesis of CTV(OMe) $_6$  and of CTTV(OMe) $_8$  by

**Table 1**

Thermal analysis of selected dendronized CTTV, CTV, triphenylene (Tp), and 1,3,5-trihydroxybenzene (THB) molecules, and their phase transition temperatures and enthalpy changes.

Compound	n <sup>a</sup>	Phase transition	T (heat) <sup>b</sup> (°C)	T (cool) <sup>c</sup> (°C)	ΔT <sup>d</sup> (°C)	ΔH <sup>e</sup> (kcal·mol <sup>-1</sup> )
(3,4,5)12G1-CTTV	8	Φ <sub>h</sub> – QLC – i	64, 79	N/A, 64	N/A, 15	N/A, 0.65
(3,4,5)12G1-CTV <sup>f</sup>	6	Φ <sub>h</sub> – A15 – i	69, 93	60, 76	9, 17	18.65, 0.67
(3,4,5)12G1-Tp <sup>g</sup>	6	Φ <sub>h</sub> <sup>io</sup> – σ – A15 – i	40, 96, 108	26, 86, 98	14, 10, 10	2.48, 0.43, 1.42
(4-3,4,5)12G1-CTTV	8	Φ <sub>h</sub> – A15 – i	112, 177	99, 161	13, 16	12.82, 0.88
(4-3,4,5)12G1-CTV <sup>f</sup>	6	Φ <sub>h</sub> <sup>io</sup> – A15 – i	105, 187	N/A, 166	N/A, 21	N/A, 1.04
(4-3,4,5)12G1-THB <sup>i</sup>	3	Φ <sub>h</sub> – i	127	122	5	3.05
(4-3,4)12G1-CTTV	8	Φ <sub>h</sub> <sup>io</sup> – i	220	217	3	6.60
(4-3,4)12G1-CTV <sup>f</sup>	6	Φ <sub>h</sub> – i	224	209	15	19.65
(4-3,4-3,5)12G2-CTTV	8	A15 – i	209	201	8	6.94
(4-3,4-3,5)12G2-Tp <sup>h</sup>	6	QLC – i	214	206	8	4.71
(3,4)dm8*G1-CTTV	8	Φ <sub>h</sub> <sup>io</sup> – i	70	57	13	1.20
(3,4)dm8*G1-CTV <sup>h</sup>	6	Φ <sub>h</sub> – i	76	58	8	0.87
(4-3,4)dm8*G1-CTTV	8	Φ <sub>h</sub> <sup>io</sup> – i	217	212	5	3.62
(4-3,4)dm8*G1-CTV <sup>h</sup>	6	Φ <sub>h</sub> – i	233	218	15	17.85

<sup>a</sup> Number of dendrons in the structure.

<sup>b</sup> Phase transition temperature upon heating.

<sup>c</sup> Phase transition temperature upon cooling.

<sup>d</sup> Differences between T (heat) and T (cool).

<sup>e</sup> Enthalpy of the phase transition.

<sup>f</sup> Data was published in Table 1 of Ref [33].

<sup>g</sup> Data was published in Table S1 of Ref [77].

<sup>h</sup> Data was published in Table 1 of Ref [34].

<sup>i</sup> Data was published in Table 1 of Ref [22]; Φ<sub>h</sub>: *P6mm* hexagonal columnar liquid crystalline phase; io: intracolumnar order; A15: *Pm3n* simple cubic lattice; σ: *P4<sub>2</sub>/mmn* tetragonal lattice; QLC: Quasi-liquid crystals (12-fold); i: isotropic phase.

methods elaborated in our laboratory to generate predominantly the veratrole trimer CTV(OMe)<sub>6</sub> or veratrole tetramer CTTV(OMe)<sub>8</sub> respectively [80,81].

The different acids used in the electrophilic cyclization of veratrole, H<sub>3</sub>PO<sub>4</sub> at 80–85 °C for trimerization and CF<sub>3</sub>COOH at 0 °C for cyclo-tetramerization were responsible for the regiospecificity of this reaction [80]. The synthesis of CTV(OH)<sub>6</sub> and CTTV(OH)<sub>8</sub> was accomplished by demethylation of the corresponding methoxy precursors with BBr<sub>3</sub> in methylenechloride. The synthesis of *n*-alkylated and dendronized CTTV was accomplished by the etherification of CTTV(OH)<sub>8</sub> with the corresponding *n*-alkyl bromide or with the self-assembling dendron containing a benzyl chloride at the apex (Fig. 1). All CTTV derivatives were analyzed by DSC.

Their DSC traces are shown in Fig. 2 while transition temperatures

collected from Fig. 2 together with their enthalpy changes and phases assigned by XRD are summarized in Tables 1, 2, and S1. Nomenclature of the dendrons and dendrimers was adapted from the previous publications in our laboratory [39] and is consistent with other publications, for example, CTV [33], Tp [34], and CTTV [73]. More specifically, (3,4,5)12G1-CTTV refers to the first generation (G1) dendron 3,4,5-tridodecyloxy benzyl [(3,4,5)12G1] on the periphery of CTTV.

A comparison of the transition temperature between the highest temperature self-organized phase and the isotropic liquid, known as isotropization temperature, T<sub>i</sub>, on heating and cooling provides a qualitative indication between the flexibility of the system and its ability to reach an equilibrium state. Table 1 compares the T<sub>i</sub> data available for all dendronized CTTV, THB [22], Tp [34] and CTV [33]. For the case of 12G0-CTTV this difference in T<sub>i</sub> is 3 or 4 °C, depending if the data are collected from the first or second heating scan (Fig. 2). This indicates that this compound undergoes a very fast transition that correlates with the fast dynamic of its CTTV core. Dendronized CTTVs exhibit a similarly small difference between T<sub>i</sub> on heating and cooling for similar degrees of order of their assemblies. This is the case of (4-3,4)12G1-CTTV (3 °C), (4-3,4)dm8\*G1-CTTV (5 °C), (3,4)dm8\*G1-CTTV (13 °C), (3,4)dm8\*G1-CTV (8 °C), (4-3,4-3,5)12G2-CTTV (8 °C), (4-3,4-3,5)12G2-Tp (8 °C), (4-3,4,5)12G1-CTTV (16 °C), and (4-3,4,5)12G1-CTV (21 °C) (Table 1).

The situation for (3,4,5)12G1CTTV (15 °C), (3,4,5)12G1CTV (75 °C), and (3,4,5)12G1Tp (10 °C) is more complex since for the case of CTTV, the highest temperature phase changes from A15 for CTV and Tp, to QLC for CTTV. This means the highest flexibility of CTTV changes not only the T<sub>i</sub> but also the nature of the phase specially when the assembly is generated from spherical objects like in the case of (3,4,5)12G1CTTV and (3,4,5)12G1CTV, as well as in the case of (4-3,4-3,5)12G2-CTTV and (4-3,4-3,5)12G2-Tp. The elucidation of this sensible arrangement in which supramolecular spheres are self-organized, requires additional experiments. On increasing the order of the phase from the highest temperature from Φ<sub>h</sub> or Φ<sub>h</sub><sup>io</sup> to A15 or to QLC a small increase in this supercooling process was observed (Fig. 2 and Tables 1, S1). It is instructive to compare the T<sub>i</sub> values of the more rigid but also monodisperse structures at the apex such as THB [22], Tp [34] and CTV [33]. In all cases more rigid structures at the apex provide higher T<sub>i</sub> values and higher degrees of supercooling. This trend is also valid for H-bonding mediated structures at the apex (all libraries including dendritic dipeptides [1,23,24,84,86–90], crown ethers [47,50], DA interactions [46,91,92], salts [16], solvated salts [42,47,49–51,56,65]). It is also important to observe that dendronized CTV, CTTV and Tp undergo a transition from column to sphere forming assemblies (Fig. 2, Table 1) while dendronized CE and oligoethylenes including their complexes as well as polymers undergo a transition from column to elongated columns [42,47,49–51,56,65]. This is not a completely elucidated event considering that some lower molar mass dendronized polymers [62,63,70,71] undergo the column to sphere transition process. It is remarkable to observe that this small library (Fig. 1) provided a rare example of QLC (for (3,4,5)12G1-CTTV) and a Frank-Kasper A15 phase (for (4-3,4,5)-12G1-CTTV) but no σ phase. We are certain that additional experiments will also uncover the Frank-Kasper σ phase in CTTV derivatives.

A similar trend as that observed from DSC (Fig. 2) was obtained from the thermal optical polarized microscopy (TOPM) micrographs shown in Fig. 3. Equilibrium textures were observed regardless of if the side groups of CTTV are *n*-alkyl or self-assembling dendrons as long as the same phase is investigated by TOPM.

Tables 2 and S1 summarize the range of temperatures, the d-spacings, their assignments and lattice dimensions of all phases as obtained by using oriented-fiber XRD experiments. Some representative examples of oriented fiber XRDs are shown in Fig. 4 with the corresponding models in Figure S1. DSC traces of 12G0-CTTV (Fig. 2, Table 1) indicate three phases before isotropization at 132 °C. Below 77 °C, there is a crystalline columnar monoclinic phase (Φ<sub>m</sub><sup>k</sup>: P2). From 77 to 97 °C, a



**Table 2**

Structural analysis of the supramolecular dendrimers self-assembled from dendronized CTTV.

Supramolecular Dendrimers	T (°C)	Phase <sup>a</sup>	$a, b, c$ (Å) <sup>b</sup>	$\rho$ (g•cm <sup>-3</sup> ) <sup>c</sup>	$M_{wt}$ (g•mol <sup>-1</sup> ) <sup>d</sup>	$t$ (Å) <sup>e</sup>	$\mu$ <sup>f</sup>	$D_{col}$ (Å) <sup>g</sup> / $D_{sph}$ (Å) <sup>g1</sup>	$d_{100}, d_{010}, d_{110}, d_{200}, d_{020}, d_{220}, d_{2-20}, d_{230}, d_{400}, d_{420}$ (Å) <sup>h</sup> $d_{100}, d_{010}, d_{110}, d_{210}, d_{310}, d_{320}, d_{410}, d_{500}$ (Å) <sup>i</sup> $d_{00002}, d_{12100}, d_{10102}, d_{12101}, d_{12103}$ (Å) <sup>j</sup> $d_{100}, d_{110}, d_{200}, d_{210}, d_{300}$ (Å) <sup>k</sup> $d_{200}, d_{210}, d_{211}, d_{310}, d_{320}, d_{222}, d_{321}, d_{400}$ (Å) <sup>l</sup> $d_{200}, d_{100}, d_{010}, d_{110}, d_{300}$ (Å) <sup>m</sup> $d_{110}, d_{020}, d_{200}, d_{130}, d_{220}, d_{040}, d_{001}$ (Å) <sup>n</sup>
12G0-CTTV	25	$\Phi_m^k$	25.0, 22.0, 4.8 <sup>j</sup>	1.04	1835.1	4.8	0.9 ~ 1	22.0	24.9, 21.9, 17.6, 12.4, 10.9, 9.3, 7.7, 6.9, 6.1, 5.8 <sup>h</sup>
	85	$\Phi_{r-s}^k$	26.5, 23.8, 4.8 <sup>j</sup>			4.8	1.0 ~ 1	23.8	27.0, 23.3, 18.8, 11.7, 8.3, 7.2, 6.4, 5.4 <sup>i</sup>
	105	$\Phi_h^{io}$	28.9, 28.9, 4.8 <sup>j</sup>			4.8	1.1 ~ 1	28.9	25.1, 14.5, 12.6 <sup>k</sup>
(3,4,5)12G1-CTTV	30	$\Phi_h$	30.1, 30.1, – <sup>j</sup>			4.9	1.1 ~ 1	30.1	26.1, –, –, – <sup>k</sup>
	70	QLC	68.5, 68.5, 68.5	0.97	5633.2	–	5.6 ~ 6	46.8 <sup>g1</sup>	34.4, 31.6, 29.9, 28.7, 18.5 <sup>j</sup>
(4–3,4,5)12G1-CTTV	70	$\Phi_h$	46.3, 46.3, – <sup>j</sup>	1.01	8180.2	4.8	1.0 ~ 1	46.3	40.2, 23.1, 20.1, –, –, – <sup>k</sup>
	125	A15	84.9, 84.9, 84.9			–	5.7 ~ 6	52.7 <sup>g1</sup>	42.7, 37.9, 34.6, –, 23.5, – 22.6, 21.1 <sup>l</sup>
(4–3,4)12G1-CTTV	80	$\Phi_h^{io}$	43.0, 43.0, 4.9 <sup>j</sup>	1.04	5856.5	4.9	0.8 ~ 1	43.0	37.4, 21.5, 18.7, –, – <sup>k</sup>
(3,4–3,5)12G2-CTTV	50	$\Phi_h$	39.9, 23.0, – <sup>j</sup>	0.98	8806.8	–	0.8 ~ 1	46.1	39.9, 23.0, 20.1, –, – <sup>k</sup>
(4–3,4–3,5)12G2-CTTV	100	$\Phi_h$	41.0, 41.0, – <sup>j</sup>	1.07	12,202	–	0.9 ~ 1	41.0	35.3, 20.6, 17.7, 13.4, 11.8 <sup>k</sup>
	190	A15	73.2, 73.2, 73.2	1.07		–	6.7 ~ 7	45.4 <sup>g1</sup>	51.2, 36.5, 32.7, 29.9, –, 26.0, –, – <sup>l</sup>
(3,4)dm8*G1-CTTV	15	$\Phi_h^k$	36.0, 36.0, 44.0 <sup>j</sup>	1.01	3709.8	5.5	1.0 ~ 1	36.0	31.2, 18.0, 15.6, 11.8, 10.4 <sup>k</sup>
	60	$\Phi_h^{io}$	34.4, 34.4, 34.4 <sup>j</sup>			–	5.8 ~ 6	34.4	29.8, 17.2, 14.9, 11.3, 9.5 <sup>k</sup>
(4–3,4)dm8*G1-CTTV	100	$\Phi_h^{io1}$	43.0, 43.0, 4.9 <sup>j</sup>	1.04	5407.8	4.9	0.9 ~ 1	43.0	37.3, 21.1, 18.6, –, – <sup>k</sup>
	85	$\Phi_{r-c}^{io}$	70.4, 46.0, 4.9 <sup>j</sup>			4.9	1.8 ~ 2	40.4	38.6, 35.2, 23.0, 20.9, 19.3, 17.6 <sup>m</sup>
	105	$\Phi_h^{io2}$	41.3, 41.3, 4.9 <sup>j</sup>			4.9	0.8 ~ 1	41.3	35.8, 20.7, 17.9, –, – <sup>k</sup>

<sup>a</sup>  $\Phi_m^k$ : P2 crystalline monoclinic phase,  $\Phi_{r-s}^k$ : P2mm crystalline simple rectangular columnar phase,  $\Phi_h^{io}$ : P6mm Liquid crystalline columnar hexagonal phase with intracolumnar order. QLC: Quasi-liquid crystals (12-fold);  $\Phi_h$ : P6mm hexagonal columnar lattice; A15: Pm3n simple cubic lattice;  $\Phi_h^k$ : P6mm crystalline hexagonal phase;  $\Phi_{r-c}^{io}$ : C2mm Simple rectangular columnar liquid crystalline phase.

<sup>b</sup> Lattice parameters (with uncertainty of ~1%).

<sup>c</sup> Experimental density ( $\rho$ ) measured at room temperature (24 °C).

<sup>d</sup> Molecular weight ( $M_{wt}$ ) of the compound.

<sup>e</sup> Stratum thickness calculated from the VWAXS pattern.

<sup>f</sup> Average number of molecules in the unit cell of columnar phases, calculated using  $\mu = N_A A t \rho / M_{wt}$  where  $N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$  = Avogadro's number, A is the area of the column cross-section calculated from the lattice parameters and  $\rho$  is the density. Average number of dendrimers forming a sphere of the unit cell of a QLC phase  $\mu = (a^3 N_A \rho) / 6 M_{wt}$  and A15 phase is  $\mu = (a^3 N_A \rho) / 8 M_{wt}$  where a is the lattice parameter,  $M_{wt}$  is the molecular weight,  $N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$  = Avogadro's number,  $\rho$  is the density of the compound.

<sup>g</sup> Column diameter: for  $\Phi_h$  phase  $D_{col} = a$ , for  $\Phi_{r-s}$ ,  $\Phi_{r-c}$  and  $\Phi_m$  phase, lattice dimensions a and b are given; for the Cubic A15 phase and QLC phase  $a = b = c$ . <sup>g1</sup> Sphere diameter calculated for the A15 phase using  $D_{sphere} = 2(3a^3/32\pi)^{1/3}$  and for the QLC12-fold phase  $[D_{sph} = 2a/(8\pi)^{1/3}]$ .

<sup>h</sup> Experimental diffraction peaks d-spacing for  $\Phi_m$  phase.

- <sup>i</sup> Experimental diffraction peaks *d*-spacing for  $\Phi_{r-s}^k$  phase.  
<sup>j</sup> Experimental diffraction peaks *d*-spacing for  $\Phi_h$  phase.  
<sup>k</sup> Experimental diffraction peaks *d*-spacing for A15 phase.  
<sup>l</sup> Experimental diffraction peaks *d*-spacing for  $P2mm$  phase.  
<sup>m</sup> Experimental diffraction peaks *d*-spacing for  $\Phi_{r-c}$  phase.  
<sup>§</sup> This is fiber direction.

simple rectangular crystalline columnar phase ( $\Phi_{r-s}^k$ ;  $P2mm$ ) is observed. From 97 °C to its isotropic point (132 °C), a hexagonal columnar liquid crystalline phase with intracolumnar order ( $\Phi_h^{io}$ ;  $P6mm$ ) is seen.

Characteristic very wide-angle powder X-ray diffraction patterns (VWAXS) at 25 °C for 12G0-CTTV and their corresponding radial plots at 25 °C are shown in Fig. 4a-c and have been summarized in Table 2. XRD patterns (Fig. 4a) and corresponding radial plot (Fig. 4b) at 25 °C indicate that it is a  $\Phi_m^k$  phase. The lattice parameters at 25 °C are determined to be  $a = 25.0$ ,  $b = 22.0$  Å,  $c = 4.8$  Å, and  $\gamma = 82.7^\circ$ . Measured *d*-spacings at 25 °C are  $d_{100} = 24.9$  Å,  $d_{010} = 21.9$  Å,  $d_{110} = 17.6$  Å,  $d_{200} = 12.4$  Å,  $d_{020} = 10.9$  Å,  $d_{220} = 9.3$  Å,  $d_{2-20} = 7.7$  Å,  $d_{230} = 6.9$  Å,  $d_{400} = 6.1$  Å,  $d_{420} = 5.8$  Å, and  $d_{001} = 4.8$  Å (Table 2). VWAXS at 85 °C for 12G0-CTTV and their corresponding radial plots at 85 °C is shown in Fig. 4d and have been summarized in Table 2. XRD patterns (Fig. 4d) and corresponding radial plot (Fig. 4e) at 85 °C indicate that it is  $\Phi_{r-s}^k$ . The lattice parameters at 85 °C are determined to be  $a = 26.5$ ,  $b = 23.8$  Å, and  $c = 4.8$  Å. Measured *d*-spacings at 85 °C are  $d_{100} = 27.0$  Å,

$d_{010} = 23.3$  Å,  $d_{110} = 18.8$  Å,  $d_{210} = 11.7$  Å,  $d_{310} = 8.3$  Å,  $d_{320} = 7.2$  Å,  $d_{410} = 6.4$  Å,  $d_{500} = 5.4$  Å (Table 2). VWAXS at 105 °C for 12G0-CTTV and their corresponding radial plots at 105 °C are shown in Fig. 4g and have been summarized in Table 2. XRD patterns (Fig. 4g) at 105 °C indicate that this is a  $\Phi_h^{io}$  phase. The lattice parameters at 105 °C are  $a = 28.9$ ,  $b = 28.9$  Å, and  $c = 4.8$  Å. Measured *d*-spacings at 105 °C are  $d_{100} = 25.1$  Å,  $d_{110} = 14.5$  Å,  $d_{200} = 12.6$  Å,  $d_{001} = 4.8$  Å (Table 2).

From the known molecular weight ( $M_w$ ) 1835.1 g.mol<sup>-1</sup>, we determined the average number of dendrimers forming the supramolecular column stratum to be equal to one molecule per stratum ( $\mu = 0.9 \sim 1$  for  $\Phi_m^k$  phase,  $\mu = 1.0 \sim 1$  for  $\Phi_{r-s}^k$  phase,  $\mu = 1.1 \sim 1$  for  $\Phi_h^{io}$  phase). The column diameter ( $D_{col}$ ) is 22.0 Å for  $\Phi_m^k$  phase, 23.8 Å for  $\Phi_{r-s}^k$  phase, and 28.9 Å for  $\Phi_h^{io}$  phase (Table 2). Figure S1 shows the molecular models for  $\Phi_m^k$ ,  $\Phi_{r-s}^k$ , and  $\Phi_h^{io}$  phases. The repeat unit is formed by one molecule for all the phases (Fig. S1a, b) and the column is formed in a similar way with an average strata distance of 4.8 Å. In the first step, one molecule is placed with a distorted CTTV core (Fig. S1a, b). In the second step, another molecule is placed at an average distance 4.8 Å (Fig. S1d).

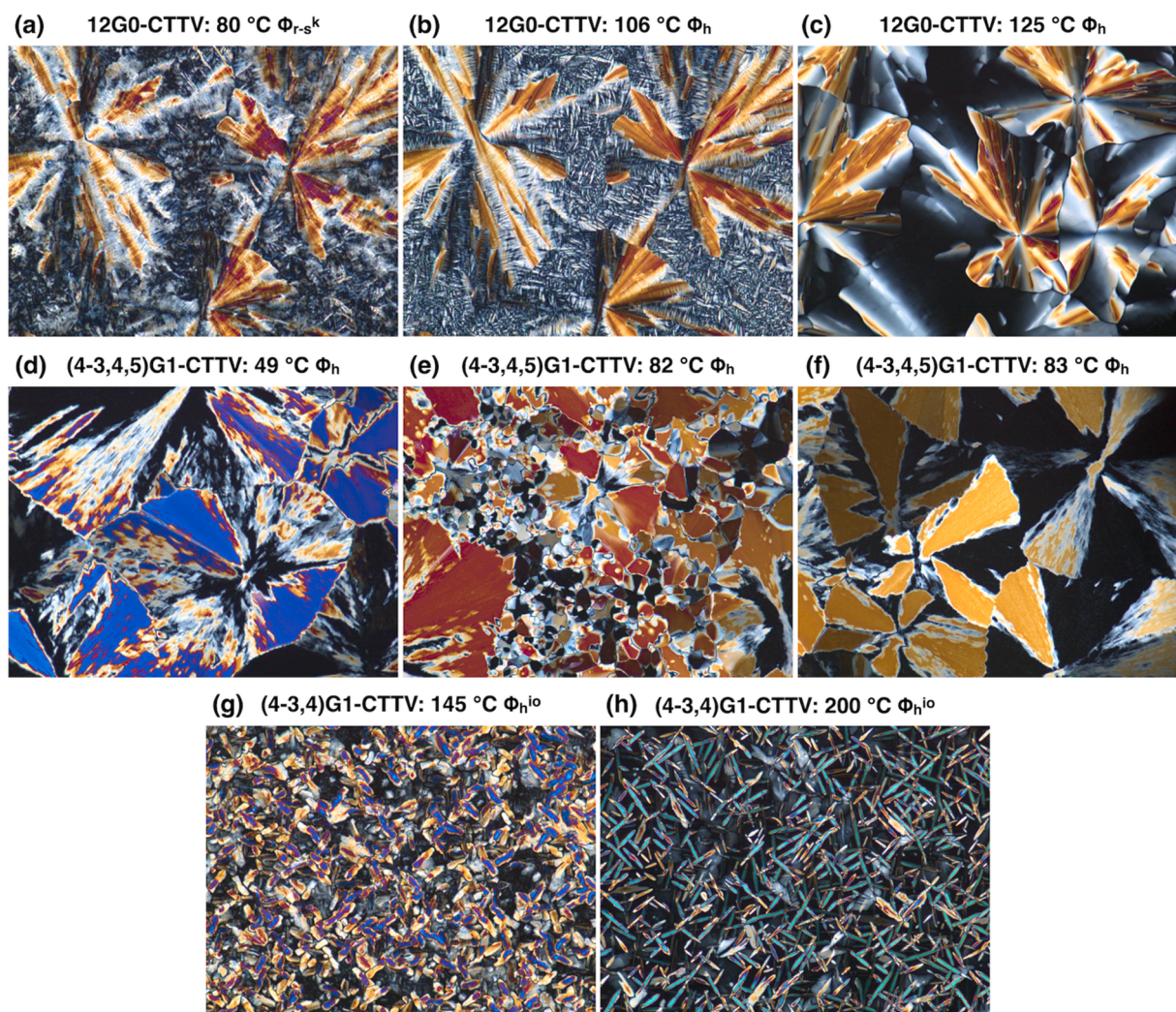
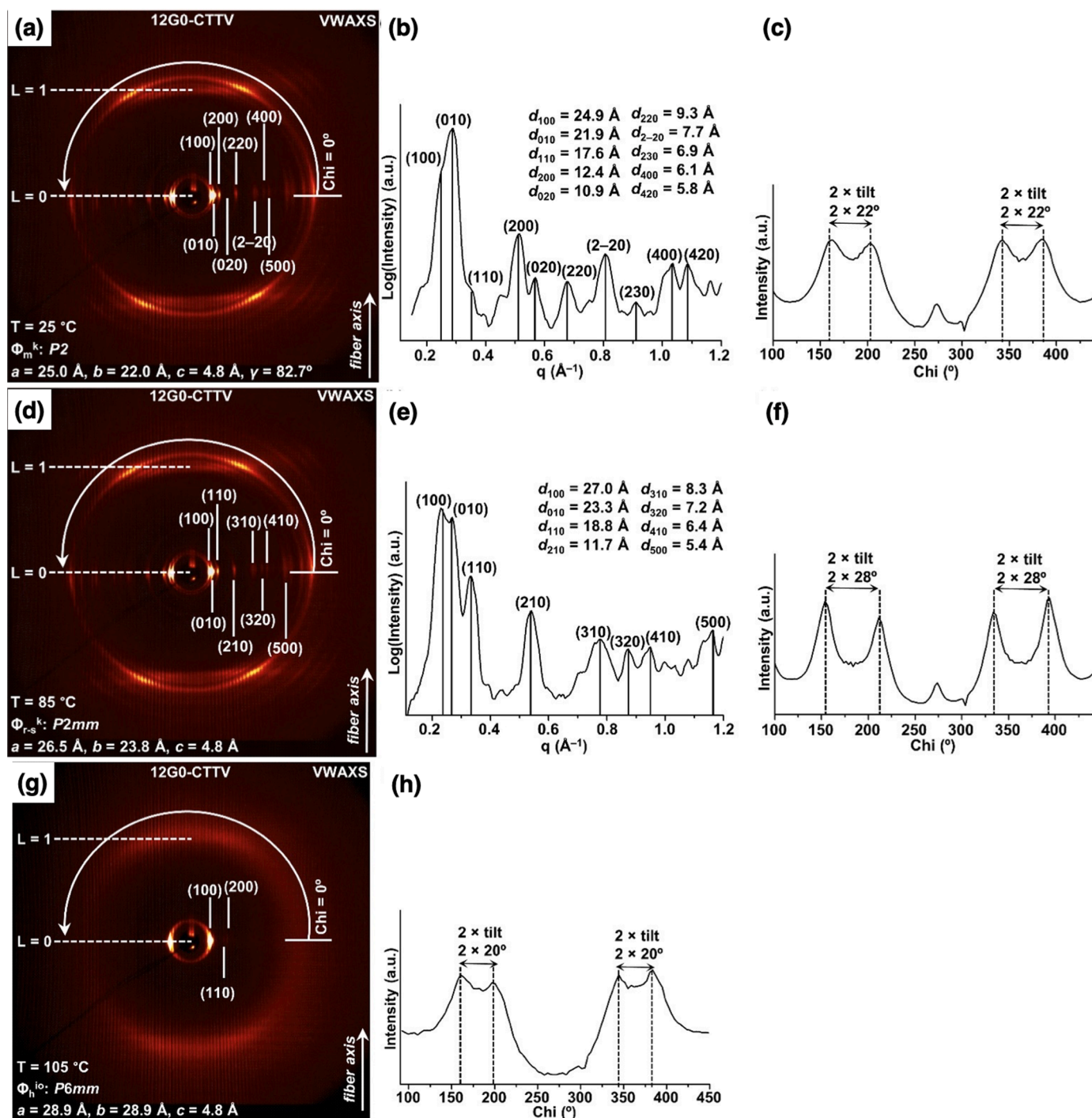


Fig. 3. Thermal optical polarized micrographs (TOPMs) of dendronized CTTV of 12G0-CTTV (a–c), (4-3,4,5)12G1-CTTV (d–f), and (4-3,4)12G1-CTTV (g, h) at indicated temperatures.





**Fig. 4.** (a) Very wide-angle fiber XRD (VWAXS) pattern ( $\Phi_m^k: P2$  phase) of 12G0-CTTV collected at  $25^\circ\text{C}$  during first heating. Diffraction indexes, fiber axis and lattice parameters are indicated. (b) Corresponding radial plot indicating the  $\Phi_m^k: P2$  phase. d-Spacings are shown. (c) Azimuthal plot [Intensity (a.u.) vs. Chi ( $^\circ$ )] from the XRD pattern at  $25^\circ\text{C}$  illustrating the tilting feature. (d) VWAX pattern ( $\Phi_{r-s}^k: P2mm$  phase) of 12G0-CTTV collected at  $85^\circ\text{C}$  during first heating. Diffraction indexes, fiber axis and lattice parameters are indicated. (e) Corresponding radial plot indicating the  $\Phi_{r-s}^k: P2mm$  phase. d-Spacings are shown. (f) Azimuthal plot [Intensity (a.u.) vs. Chi ( $^\circ$ )] from the XRD pattern at  $85^\circ\text{C}$  illustrating the tilting feature. (g) VWAX pattern ( $\Phi_h^{i0}: P6mm$  phase) of 12G0-CTTV collected at  $105^\circ\text{C}$  during first heating. Diffraction indexes, fiber axis and lattice parameters are indicated. (h) Azimuthal plot [Intensity (a.u.) vs. Chi ( $^\circ$ )] from the XRD pattern at  $105^\circ\text{C}$  illustrating the tilting feature.

Thus, a column structure is formed (Fig. S1d). These columns will arrange themselves in a monoclinic array (Fig. S1e), in a simple rectangular array (Fig. S1f), and in a hexagonal array (Fig. S1g). VWAXS (Fig. 4) indicates the tilting angle of the alkyl groups with the columnar axis [ $22^\circ$  for  $\Phi_m^k$  phase (Fig. 4c),  $28^\circ$   $\Phi_{r-s}^k$  phase (Fig. 4f), and  $20^\circ$   $\Phi_h^{i0}$  phase (Fig. 4h)].

Fig. 5 shows a representative XRD of the QLC phase of (3,4,5)12G1-

CTTV. In contrast, the corresponding dendronized molecules with CTV and Tp at the apex showed A15 or  $\sigma$  phases (Table 1) [30,31]. DSC traces (Fig. 2, Table S1) indicate that (3,4,5)12G1-CTTV displays three phases. XRD patterns were used to determine them. Below  $-2^\circ\text{C}$ , there is an unknown crystalline phase (k). From  $-2^\circ\text{C}$  to  $64^\circ\text{C}$ , a  $\Phi_h$  phase is found. From  $64^\circ\text{C}$  to  $79^\circ\text{C}$ , a QLC phase is observed. At  $79^\circ\text{C}$ , this phase reaches the isotropization temperature (i).

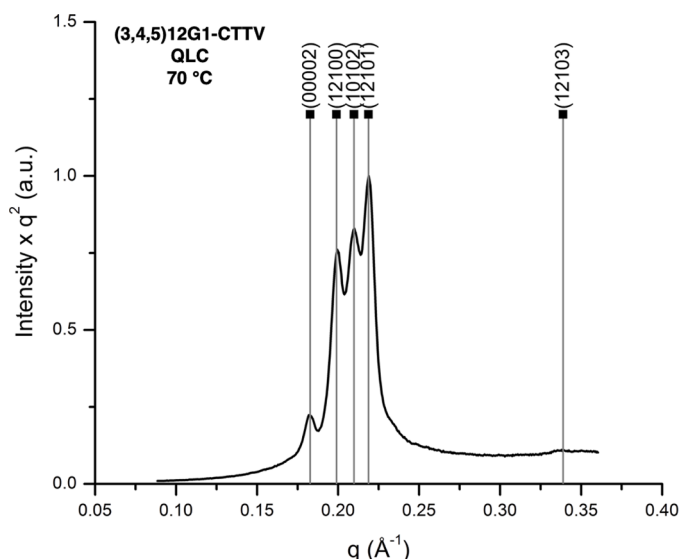


Fig. 5. Small angle powder diffraction pattern collected at 70 °C from (3,4,5) 12G1-CTTV self-organized into the QLC.

XRD patterns at 30 °C for (3,4,5)12G1-CTTV have been summarized in Table 2. XRD patterns at 30 °C indicates a  $\Phi_h$  phase. Lack of off axis diffraction pattern suggests that this phase is a 2D hexagonal lattice. The lattice parameters at 30 °C are determined to be  $a = 30.1$  and  $b = 30.1$  Å. The measured d-spacings at 30 °C are  $d_{100} = 26.1$  Å (Table 2). From the known molecular weight ( $M_{wt}$ ) 5633.2 g.mol<sup>-1</sup>, we calculated the average number of dendrimers forming the supramolecular column stratum. We find that in both  $\Phi_h$  phases, there must be one molecule ( $\mu = 1.1 \sim 1$  for  $\Phi_h$ ) per column cross-section. The column diameter is 30.1 Å for  $\Phi_h$  phase [ $D_{col} = a$ ]. The c-axis of the  $\Phi_h$  phase is along the fiber axis. Figure S2a–e displayed the molecular models of the  $\Phi_h$  phase. For the  $\Phi_h$  phase, the repeat unit is formed by one molecule per stratum (Fig. S2a–c). In the first step, one molecule is placed with highly distorted CTTV core (Fig. S2a, b). In the second step, another molecule is placed on top of the first molecule at 4.9 Å (Fig. S2c, d). Thus, each molecule will stack over one another to assemble the supramolecular column (Fig. S2c, d) and thus these columns will arrange themselves in a hexagonal array (Fig. S2e). Small angle XRD patterns (SAXS) (Fig. 5) at 70 °C indicate that this phase is QLC. The lattice parameters at 30 °C are

determined to be  $a = b = c = 68.5$  Å. Measured d-spacings at 30 °C are  $d_{00002} = 34.4$  Å,  $d_{12100} = 31.6$  Å,  $d_{10102} = 29.9$  Å,  $d_{12101} = 28.7$  Å, and  $d_{12103} = 18.5$  Å (Table 2). From the known molecular weight ( $M_{wt}$ ) 5633.2 g.mol<sup>-1</sup>, we calculate the average number of dendrimers forming the unit cell of the QLC phase to be about 34 molecules in a unit cell. Therefore, each sphere in the QLC phase is formed by six molecules. Hence, we calculate the number of molecules ( $\mu$ ) constructing the sphere which is 6 [ $\mu = \mu'/6$ ,  $\mu = 5.6 \sim 6$ ] in a sphere. The sphere diameter is 46.8 Å for the QLC phase [ $D_{sph} = 2a/(8\pi)^{1/3}$ ] (Table 2). The models of the sphere which form the QLC lattice are shown in Fig. S2f, g. The sphere is formed by six molecules. The CTTV cores of the six molecules placed in a similar way as in the columns of the hexagonal phase. But the phenyl rings and the alkyl groups are much more disordered to form a sphere. Fig. S2f and S2g represent the top view and the side view respectively. A schematic representation of the QLC lattice is provided in Fig. S2h.

DSC traces (Fig. 2, Table S1) of the supramolecular dendrimer of (4-3,4)12G1-CTTV indicate that it exhibits two phases before its isotropization at 220 °C. During the first heating (Fig. 2), a small hump is found before the isotropization (from 214 to 220 °C), but because of very small temperature range ( $\sim 6$  °C) the phase could not be determined. Interesting the hump does not appear during the cooling or subsequent heating and cooling. Below 48 °C, there is an unknown crystalline phase (k). A  $\Phi_h^{io}$  phase is observed between 48 and 220 °C during the first heating. VWAXS and their corresponding radial plots at 80 °C are shown in Fig. 6 and have been summarized in Table 2. The lattice parameters at 80 °C are determined to be  $a = b = 43.0$  Å,  $c = 4.9$  Å. Measured d-spacings are  $d_{100} = 37.4$  Å,  $d_{110} = 21.5$  Å,  $d_{200} = 18.7$  Å,  $d_{001} = 4.9$  Å (Table 2). The absence of off-axis features (Fig. 6a) in VWAX XRD pattern indicates that it is a liquid crystalline phase. From the known molecular weight ( $M_{wt}$ ) 5856.5 g.mol<sup>-1</sup>, we infer an average number of dendrimers forming the supramolecular column stratum to be one molecule ( $\mu = 0.8 \sim 1$ ) per stratum. The column diameter is 43.0 Å [ $D_{col} = a$ ]. The c-axis of this hexagonal phase is along the fiber axis. The molecular model for the  $\Phi_h^{io}$  phase is shown in Fig. S3. The repeat unit is formed by one molecule (Fig. 3c). In the first step, one molecule is placed with a distorted CTTV core (Fig. S3a–c). In the second step, another molecule is placed on top of the first molecule at 4.9 Å (Fig. S3d–g). Thus, each molecule will stack over one another to form the supramolecular column of the  $\Phi_h^{io}$  phase (Fig. S3d–g). These columns, therefore, arrange themselves in a hexagonal array (Fig. S3f). The Intensity (a.u.) vs. Chi (°) plot (Fig. 6d) indicates that the alkyl groups are tilted by 27° with the column axis.

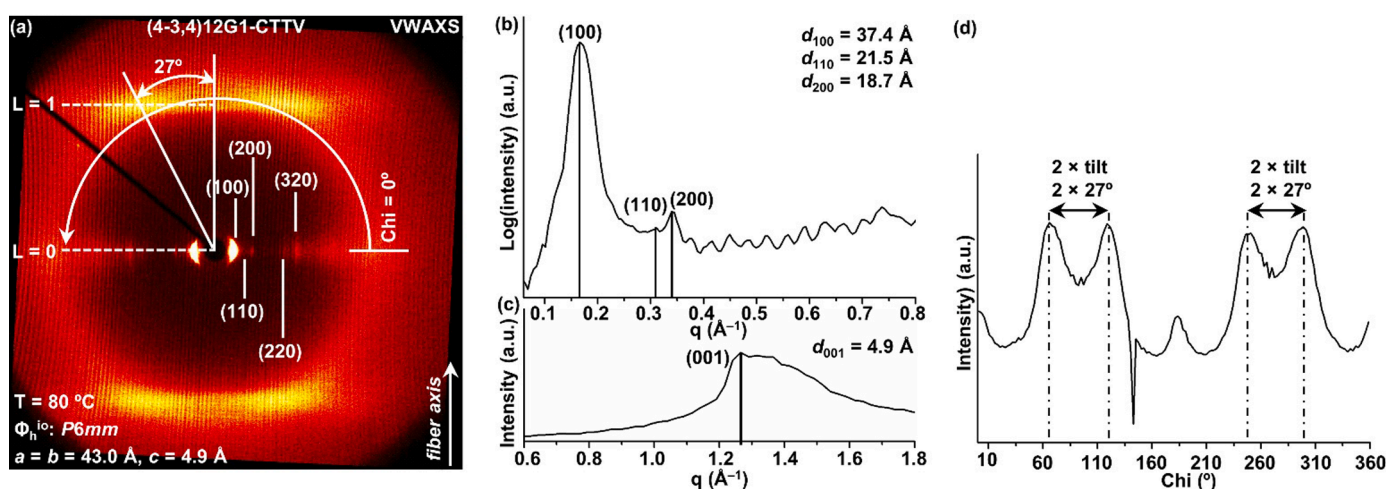


Fig. 6. (a) Very wide-angle XRD pattern (VWAXS) of (4-3,4)12G1-CTTV at 80 °C. (100), (110), (200), (220), and (320) diffractions indicate a  $\Phi_h^{io}$  phase. Diffraction indexes, fiber axis, tilting feature, and lattice parameters are indicated. (b) Radial plot along the equatorial plane from the XRD pattern. d-Spacings are shown. (c) Radial plot along the meridional axis from the XRD. d-Spacing is shown. (d) Intensity (a.u.) vs. Chi (°) plot from the XRD pattern illustration the tilting features of the alkyl groups ( $-C_{12}H_{25}$ ).



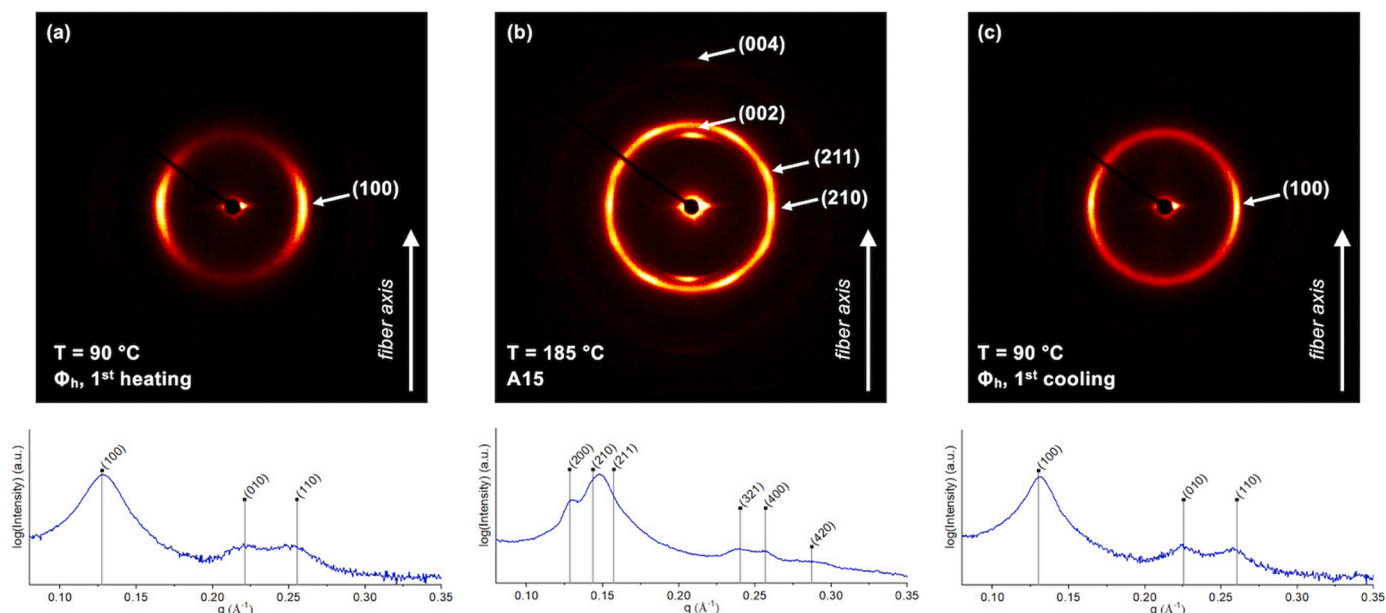


Fig. 7. Small angle fiber diffraction pattern from (4-3,4-3,5)12G2-CTTV self-organized into  $\Phi_h$  upon first heating (a), A15 phase (b), and into  $\Phi_h$  upon first cooling (c).

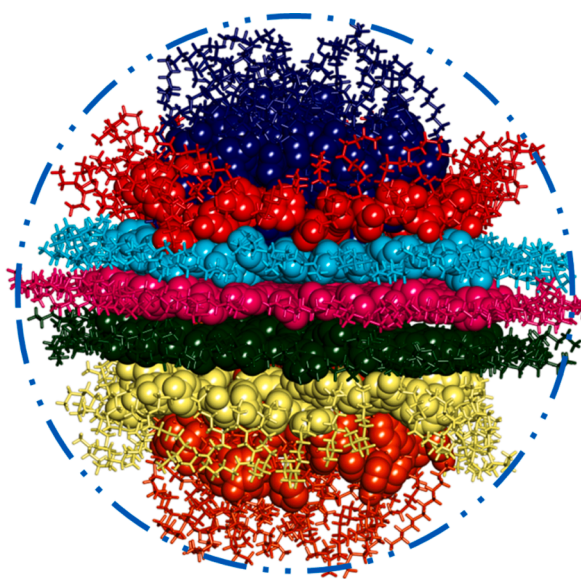


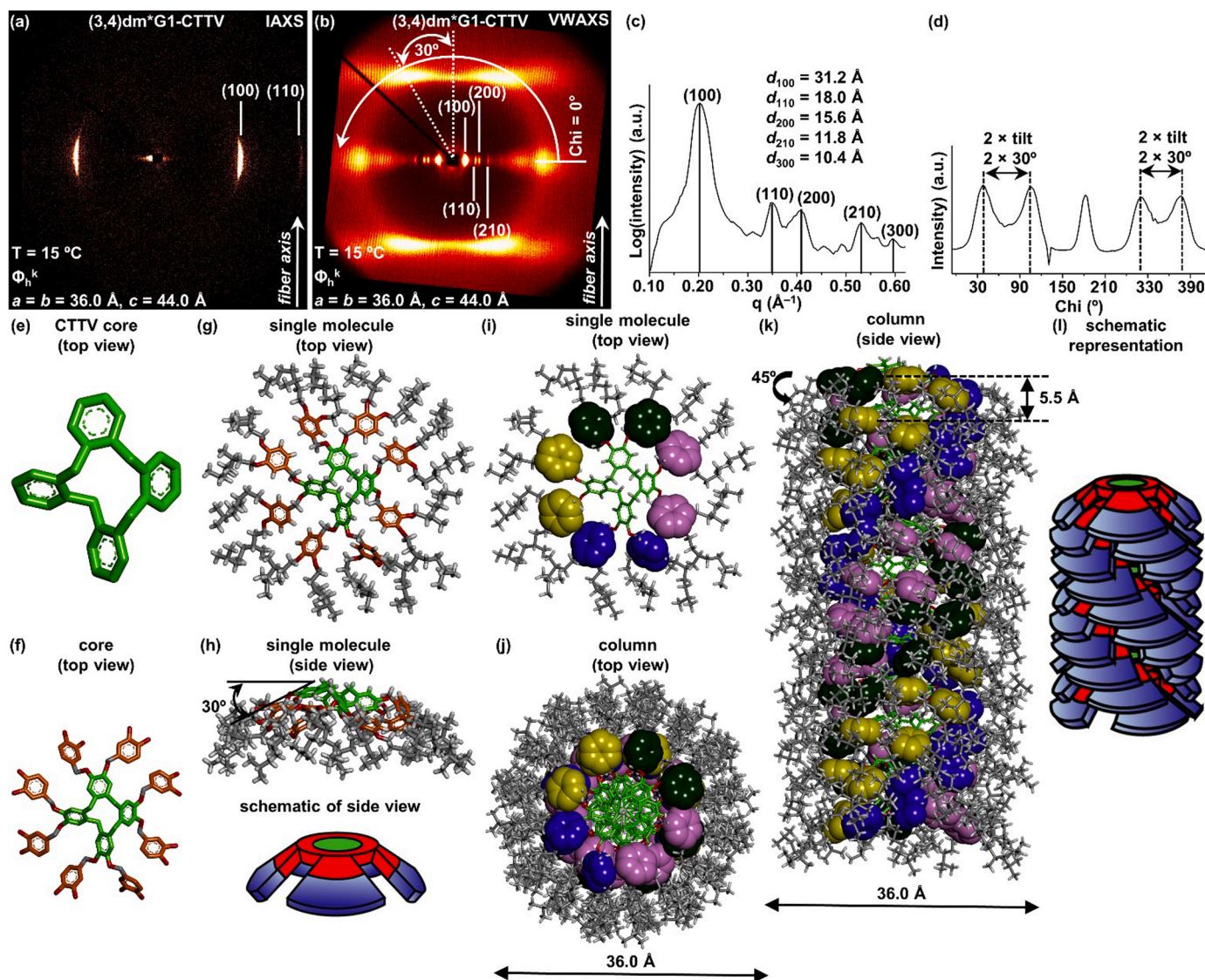
Fig. 8. Mechanism of formation of the supramolecular sphere of the A15 phase of (4-3,4-3,5)12G2-CTTV.

The DSC traces (Fig. 2, Table S1) indicate that (3,4-3,5)12G2-CTTV exhibits two phases before isotropization at 125 °C. Below 38 °C, there is a glassy (G) phase. From 38 to 125 °C, a  $\Phi_h$  phase is found. At 125 °C, the compound reaches the isotropic point (i). Characteristic X-ray diffraction patterns at 50 °C for (3,4-3,5)12G2-CTTV have been summarized in Table 2. XRD patterns 50 °C and the radial plots indicates that it is a  $\Phi_h$  phase. Lack of off axis diffraction pattern indicates that it is a 2D hexagonal phase. The lattice parameters at 50 °C are determined to be  $a = b = 46.1$  Å. Measured d-spacings at 50 °C are  $d_{100} = 39.9$  Å,  $d_{110} = 23.0$  Å, and  $d_{200} = 20.1$  Å (Table 2). From the known molecular weight ( $M_{wt}$ ) 8806.8 g.mol<sup>-1</sup>, we determined an average number of dendrimers forming the supramolecular column stratum to be one molecule ( $\mu = 0.8 \sim 1$ ) per stratum. The column diameter is 46.1 Å [ $D_{col} = a$ ]. The c-axis of the phase is along the fiber axis. Figure S4a–d displays the molecular models for  $\Phi_h$  phase. For  $\Phi_h$  phase the repeat unit is formed by

one molecule (Fig. S4a, b). In the first step, one molecule is placed with highly distorted CTTV core (Fig. S4a, b). In the second step, another molecule is placed on top of the first molecule at 4.9 Å (Fig. S4c, d). Thus, each molecule will stack over one another to form the column (Fig. S4c, d) and thus these supramolecular columns will arrange themselves in a hexagonal array.

The oriented fiber XRD from Fig. 7 demonstrated that (4-3,4-3,5) 12G12-CTTV forms  $\Phi_h$  and A15 phases but did not show a supramolecular orientational memory (SOM) effect [74–79]. In contrast, (4-3,4, 5)12G12-CTTV was reported to show an orthogonal SOM [77] effect when cooled from A15 to  $\Phi_h$  phase. SOM was discovered in our laboratory [74–79]. It represents a special case of epitaxial orientation when oriented fibers of self-assembling dendrimers generated in the  $\Phi_h$  phase maintain the orientation of an A15, body-central cubic (BCC) or  $\sigma$  phase upon cooling back to the  $\Phi_h$  phase from the sphere forming supramolecular assemblies. This event generates a large diversity of novel morphologies that are not accessible by any other mechanism [74–79].

DSC traces (Fig. 2, Table S1) indicate that (4-3,4,5)12G1-CTTV exhibits three distinguished phases before isotropization at 177 °C. Below –13 °C, there is an unknown crystalline phase. From –13 to 112 °C, a  $\Phi_h$  phase is found. From 112 to 177 °C, a A15 phase is observed. At 177 °C, the A15 phase undergoes isotropization (i). Below –13 °C, the phase seems to be crystalline to be denoted as k. Characteristic X-ray diffraction patterns at 70 °C for (4-3,4,5)12G1-CTTV have been summarized in Table 2. XRD patterns 70 °C indicate that it is a  $\Phi_h$  phase. The lattice parameters at 30 °C are determined to be  $a = 46.3$  Å and  $b = 46.3$  Å. Measured d-spacings at 70 °C are  $d_{100} = 40.2$  Å,  $d_{110} = 23.1$  Å, and  $d_{200} = 20.1$  Å (Table 2). From the known molecular weight ( $M_{wt}$ ) 8180.2 g.mol<sup>-1</sup>, we determine the average number of dendrimers forming the supramolecular column stratum that is about one molecule ( $\mu = 1.1 \sim 1$  for  $\Phi_h$ ). The column diameter is 46.3 Å for  $\Phi_h$  phase [ $D_{col} = a$ ]. The c-axis of the hexagonal phase is along the fiber axis. Fig. S5a–e shows the molecular models for the  $\Phi_h$  phase. The repeat unit is formed by one molecule (Fig. S5a–c). For the  $\Phi_h$  phases, in the first step, one molecule is placed with highly distorted CTTV core (Fig. S5a, b). In the second step, another molecule is placed on top of the first molecule at 4.9 Å (Fig. S5c, d). Thus, each molecule will stack over one another to form the column (Fig. S5c, d) and thus these columns will arrange themselves in a hexagonal array (Fig. S5e). WAXS at 125 °C indicate that the phase is A15. The lattice parameters at 125 °C are determined to be a



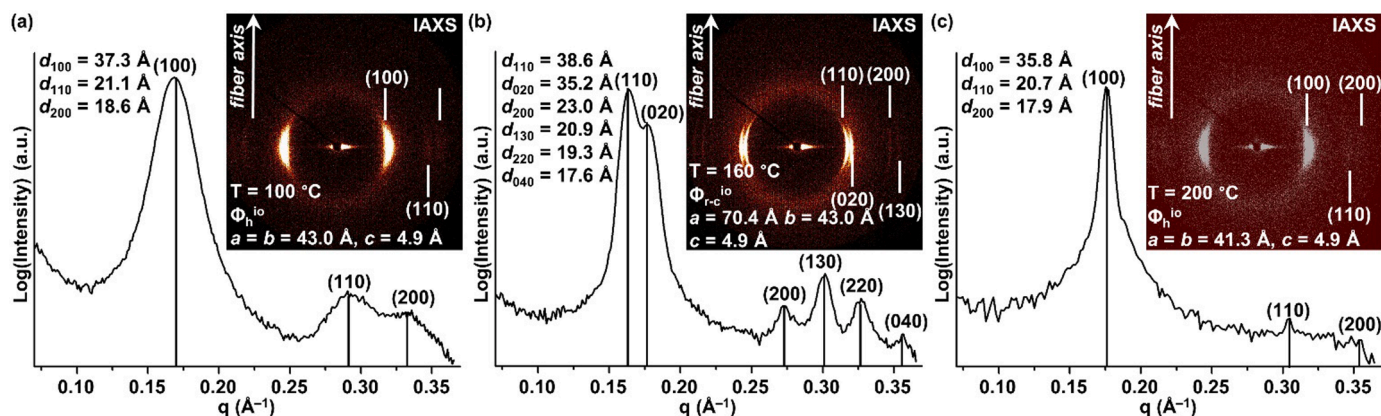
**Fig. 9.** (a) Intermediate-angle fiber XRD (IAXS) patterns ( $\Phi_h^k$  phase) of (3,4)dm8\*G1-CTTV at 15 °C. (b) VWAX patterns ( $\Phi_h^k$  phase) of (3,4)dm8\*G1-CTTV at 15 °C. Diffraction indexes, fiber axis and lattice parameters are indicated. (c) Radial plot along the equatorial plane showing the (100), (110), (200), (210), and (300) diffractions indicating a hexagonal phase. d-spacings are shown. (d) Azimuthal plot [Intensity (a.u.) vs.  $\chi$  (°)] from the VWAX XRD pattern illustrating the tilting feature. (e–k) Molecular models for the ( $\Phi_h^k$  phase) of (3,4)dm8\*G1-CTTV at 15 °C; (e) Twisted and distorted CTTV core (top view); (f) CTTV core with the phenyl rings (top view); H-atoms have been omitted for clarity in e–f. (g) Single molecule (top view); (h) Single molecule showing the tilting feature (side view); (i–k) Rainbow color model; (i) Single molecule showing the color code (top view); (j) Column (colored, top view); (k) Column (colored, side view). (l) Schematic representation of the columns in the hexagonal array (side view). Color code used in the model (e–h): O atoms, red; H atoms, white; C atoms in the CTTV core, green; C atoms in the other phenyl rings, orange; all other C atoms, gray.

$= b = c = 84.9$  Å. Measured d-spacings at 125 °C are  $d_{200} = 42.7$  Å,  $d_{210} = 37.9$  Å,  $d_{211} = 34.6$  Å,  $d_{320} = 23.5$  Å,  $d_{321} = 22.6$  Å, and  $d_{400} = 21.1$  Å (Table 2). From the known molecular weight ( $M_w$ ) 8180.2 g.mol<sup>-1</sup>, we infer that the average number of dendrimers forming a sphere of the unit cell should be 6 molecules (5.7 ~ 6) in a sphere. The sphere diameter is 52.7 Å for the A15 phase [ $D_{sph} = 2(3a^3/(32\pi)^{1/3})$ ] (Table 2). The models of the sphere which form the A15 lattice is shown in Fig. S5f, g. The sphere is formed by six molecules. The CTTV cores of the six molecules placed in a similar way as in the columns of the hexagonal phase. But the phenyl rings and the alkyl groups are highly disordered to form a sphere. Fig. S5f and S5g represent the top view and the side view respectively. A schematic representation of the A15 lattice is provided in Fig. S5h.

DSC traces (Fig. 2, Table S1) indicate that (4-3,4-3,5)12G2-CTTV exhibits three phases before isotropization at 209 °C. Below 49 °C, there is an unknown crystalline phase (k). From 49 to 181 °C, a  $\Phi_h$  phase is observed. From 181 to 209 °C, a A15 phase is available. At 209 °C, this

phase undergoes isotropization (i). Characteristic X-ray diffraction patterns at 100 °C for (4-3,4-3,5)12G1-CTTV have been summarized in Table 2 and are shown in Fig. 7a. XRD patterns at 100 °C and the radial plots (Fig. 7) indicate that it is columnar liquid crystalline hexagonal phase ( $\Phi_h$  phase). The lattice parameters at 100 °C are determined to be  $a = b = 41.0$ . Measured d-spacings at 100 °C are  $d_{100} = 35.3$  Å,  $d_{110} = 20.6$  Å,  $d_{200} = 17.7$  Å,  $d_{210} = 13.4$  Å and  $d_{200} = 11.8$  Å (Table 2). From the known molecular weight ( $M_w$ ) 12,202.0 g.mol<sup>-1</sup>, we determined the average number of dendrimers forming the supramolecular column stratum to be one molecule ( $\mu = 0.9 \sim 1$ ) in the repeat unit of  $\Phi_h$  phase. The column diameter is 41.0 Å [ $D_{col} = a$ ]. The c-axis of the phase is along the fiber axis. Figure S6a–e consists of the molecular models for  $\Phi_h$  phase. The repeat unit is formed by one molecule (Fig. S6a, b). In the first step, one molecule is placed with highly distorted CTTV core (Fig. S6a, b). In the second step, another molecule is placed on top of the first molecule at 4.9 Å (Fig. S6c, d). Thus, each molecule will stack over





**Fig. 10.** (a) Radial plot along the equatorial plane from the IAXS of (4-3,4)dm8\*G1-CTTV at 100 °C. (100), (110), and (200) diffractions indicate a columnar hexagonal phase. d-spacings are shown. IAXS at 100 °C is shown in the inset. Diffraction indexes, fiber axis and lattice parameters are indicated. (b) Radial plot along the equatorial plane from IAXS of (4-3,4)dm8\*G1-CTTV at 160 °C. (110), (020), (200), (130), (220) and (040) diffractions indicate a centered rectangular phase. d-spacings are shown. Intermediate-angle XRD pattern (IAXS) at 160 °C is shown in the inset. Diffraction indexes, fiber axis and lattice parameters are indicated. (c) Radial plot along the equatorial plane from IAXS of (4-3,4)dm8\*G1-CTTV at 200 °C. (100), (110), and (200) diffractions indicate a columnar hexagonal phase. d-spacings are shown. Intermediate-angle XRD pattern (IAXS) at 200 °C is shown in the inset. Diffraction indexes, fiber axis, and lattice parameters are indicated.

one another to form the column (Fig. S6c, d) and the columns will arrange themselves in a hexagonal array (Fig. S6e). SAXS (Fig. 7b) at 190 °C indicate that the phase is A15 phase. The lattice parameters are determined to be  $a = b = c = 73.2$  Å. Measured d-spacings at 125 °C are  $d_{200} = 51.2$  Å,  $d_{210} = 36.5$  Å,  $d_{211} = 32.7$  Å,  $d_{310} = 29.9$  Å, and  $d_{222} = 26.0$  Å (Table 2). From the known molecular weight ( $M_w$ ) 12,202 g. mol<sup>-1</sup>, we calculate an average number of dendrimers forming a sphere of the A15 phase. This value is seven molecules (6.7 ~ 7) in a sphere of the A15 lattice. The sphere diameter is 45.4 Å for the A15 phase [ $D_{sph} = 2(3a^3/(32\pi))^{1/3}$ ] (Table 2). The models of the sphere which form the A15 lattice is shown in Fig. S6f, g. The sphere is formed by six molecules. The CTTV cores of the six molecules placed in a similar way as in the columns of the hexagonal phase. But the phenyl rings and the alkyl groups are highly disordered to form a sphere. Fig. S6f and S6g represent the top view and the side view respectively. A schematic representation of the A15 lattice is provided in Fig. S6h.

The mechanism of formation of the spheres in case of the A15 phase of (4-3,4-3,5)12G2-CTTV is shown in Fig. 8. It can be noticed that the seven CTTV dendrimers are stacked on one another in a similar way as in the  $\Phi_h$  phase but their outer side phenyl rings and the alkyl chains are flexibly distorted to take a spherical shape for the formation of the QLC phase. The mechanism of formation of spheres for QLC phase in case of (3,4,5)12G-CTTV or for A15 phase of (4-3,4,5)12G-CTTV are similar.

Fig. 9 illustrates the oriented fiber of (3,4)dm8\*G1-CTTV together with the helical model of its chiral column, which was published previously [73].

DSC traces (Fig. 2, Table S1) indicate that (4-3,4)dm8\*G1-CTTV exhibits three distinguished phases. Below 148 °C, there is a  $\Phi_h^{io1}$  phase ( $\Phi_h^{io1}$ ). From 148 to 182 °C, a center rectangular crystalline columnar ( $\Phi_{r-c}^{io}$ ) phase with intracolumnar order is found. From 182 °C to the melting point 217 °C there is another  $\Phi_h^{io}$  phase ( $\Phi_h^{io2}$ ). IAXS for (4-3,4)dm8\*G1-CTTV and their corresponding radial plots at 100 °C is shown in Figure 10 and have been summarized in Table 2. The lattice parameters at 100 °C ( $\Phi_h^{io1}$ ) are determined to be  $a = 43.0$  and  $b = 43.0$  Å. Measured d-spacings at 25 °C are  $d_{100} = 37.3$  Å,  $d_{110} = 21.1$  Å, and  $d_{200} = 18.6$  Å (Table 2). The lattice parameters at 160 °C ( $\Phi_{r-c}^{io}$ ) are determined to be  $a = 70.4$  Å and  $b = 43.0$  Å. Measured d-spacings are  $d_{110} = 38.6$  Å,  $d_{020} = 35.2$  Å,  $d_{200} = 23.0$  Å,  $d_{130} = 20.9$  Å,  $d_{220} = 19.3$  Å, and  $d_{040} = 17.6$  Å (Table 2). The lattice parameters at 200 °C ( $\Phi_h^{io2}$ ) are determined to be  $a = 41.3$  and  $b = 41.3$  Å. Measured d-spacings at 200 °C are  $d_{100} = 35.8$  Å,  $d_{110} = 20.7$  Å, and  $d_{200} = 17.9$  Å (Table 2). From the known molecular weight ( $M_w$ ) 5407.8 g. mol<sup>-1</sup>, we determine the average number of dendrimers forming the supramolecular column

stratum to be one molecule ( $\mu = 0.9 \sim 1$  for  $\Phi_h^{io1}$  and  $\Phi_h^{io2}$ ). The column diameter is 43.0 Å for  $\Phi_h^{io1}$  phase while the column diameter for the  $\Phi_h^{io2}$  phase is 41.3 Å [ $D_{col} = a$ ]. For the  $\Phi_{r-c}^{io}$  phase, there are two molecules ( $\mu = 1.8 \sim 2$ ) for  $\Phi_{r-c}^{io}$  phase per stratum. The c-axis of this monoclinic phase is along the fiber axis. Figure S7 consists of the molecular models for  $\Phi_h^{io1}$ ,  $\Phi_{r-c}^{io}$ , and  $\Phi_h^{io2}$  phases. For  $\Phi_h^{io1}$  and  $\Phi_h^{io2}$  the repeat unit is formed by one molecule (Fig. S7a–c). For the  $\Phi_h^{io}$  phases, in the first step, one molecule is placed with highly distorted CTTV core (Fig. S7a–c). In the second step, another molecule is placed on top of the first molecule at 4.9 Å (Fig. S7e–g). Thus, each molecule will stack over one another to form the column (Fig. S7f and S7h) and the columns will arrange in a hexagonal array (Fig. S7h). For  $\Phi_{r-c}^{io}$  phase, the repeat unit is formed by two molecules (Fig. S7d). For the  $\Phi_{r-c}^{io}$  phase, in the first step, two molecules are placed in front of each other with distorted CTTV core (Figure S7d) to form an oval shape repeat unit. In the second step, another dimer is placed on top of the first dimer at 4.9 Å. Thus, these dimers will stack over one another to form the column of the center rectangular phase ( $\Phi_{r-c}^{io}$ ) (Fig. S7g) and thus these columns will arrange themselves in a center rectangular array (Fig. S7i).

## Conclusions

The primary-tertiary structure-function is one of the most fundamental concepts responsible for biological and synthetic self-organizations [39,42,43,84,85]. The less understood part of this concept refers to the role of the conformational flexibility during this process. In this publication we compare the dynamics of an *n*-alkylated with dendronized conformationally flexible CTTV with its less flexible CTV, Tp and THB. These experiments demonstrated that conformational flexibility may select unusual equilibrium structures while more rigid structures may freeze non-equilibrium structures since more rigid apex molecules increase phase transition temperatures while decreasing the dynamics of the self-organized system. Helical columnar structures generated from chiral spheres and from crowns, QLC and Frank-Kasper A15 phases dominate the hierarchical self-organizations observed so far with dendronized CTTV. However, the structural requirements that mediate the transition from a supramolecular column to supramolecular spheres or elongated columns and their accompanied self-organizations require additional experiments in order to be elucidated. Additional experiments are required to explain how conformational flexibility changes the mechanism of self-organization of supramolecular spheres in different Frank-Kasper and quasicrystal phases. The concept raised in this report could inspire additional discoveries in other forms of

self-organized soft matter including lipids, block co-polymers, surfactants, dendron-like silsesquioxane-cage molecules, [93–125], which were reported to forms complex architectures including Frank-Kasper phases (Fig. 10).

### Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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### Supplementary materials

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