

Programmable Design of Seed Coating Function to Induce Water Stress Tolerance in Semi-Arid Regions

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Abstract

Changes in climate patterns and soil depletion are major challenges that agriculture needs to address to build a sustainable food infrastructure that can feed the growing human population. In semi-arid regions, water is the determining factor for crop production and water stress during seed germination and early seedling growth is the highest cause of crop loss, with dramatic impact on food security for 1 billion people that are threatened by desertification. In nature, some seeds (e.g. chia, basil) produce a mucilage-based hydrogel that creates a germination-promoting microenvironment by retaining water, regulating nutrients entry, and facilitating interactions with beneficial microorganisms. Inspired by this strategy, a two-layered biopolymer-based seed coating has been developed to increase germination and water stress tolerance in semi-arid, sandy soils. Seeds are coated with a silk/trehalose inner layer containing *rhizobacteria* and with a pectin/carboxymethylcellulose (CMC) outer layer that upon sowing reswells and acts as a water jacket. Using *Phaseolus vulgaris* (common bean) cultured under water stress conditions in an experimental farm in Ben Guerir, Morocco, it is demonstrated that the proposed seed coating effectively delivers *rhizobacteria* to form roots nodules, results in plants with better health, and mitigates water stress in drought-prone marginal lands.

Main

The projected growth of the human population together with changes in climate patterns will significantly challenge the food infrastructure¹ and require the development of new technologies²⁻⁸ to enable sustainable food production while minimizing the use of scarce resources (e.g. water, fertile land), reducing the application of energy intensive inputs (e.g. synthesis of nitrogen-based fertilizers consumes 1-2% of the global energy production) and mitigating environmental impact.^{9,10} In semi-arid regions, which constitute circa 15% of the world's land, water is the determining factor for crop production¹¹ and water stress during seed germination and early seedling growth is the highest cause of crop loss,¹² with dramatic impact on food security for 1 billion people that are threatened by desertification and already live in conditions of malnutrition.¹³

For semi-arid soils, water-holding compounds like hydrophilic and superabsorbent polymers can be applied to seeds, mixed into the soil or deposited on roots before planting to increase water retention and usage efficiency.¹⁴⁻¹⁶ However, the application of these polymers is labor and energy intensive and often results in the release of synthetic plastics in the soil.¹⁷ As a complementary approach, plant-growth promoting rhizobacteria (PGPRs) can be used to enhance plant health in conditions of water scarcity.¹⁸⁻²¹ PGPRs are biofertilizers that interact with plant roots to increase availability of nutrients and phytohormones and enhance plant response to heat, saline soil and drought.²²⁻²⁵ Some PGPRs, like rhizobia, can infect legume roots to form symbiotic nodules that fix nitrogen, limiting the use of fertilizers and enhancing plant health in semi-arid regions.²⁶⁻²⁸ Nonetheless, the use of PGPRs is limited by their low resistance to desiccation stress^{29,30} and competition upon resuscitation with a diversity of microorganisms present in the soil environment (i.e. spermosphere)³¹. Together, these limitations make difficult the integration of rhizobia in simple delivery systems – as seed coating technologies – that do not require the use of skills, agricultural practices and resources often not available in semi-arid regions of the world.³²

In nature, polysaccharides present in the seed coat of myxospermous species occupying arid environments (e.g. *Salvia hispanica*, *Ocimum basilicum* and *Plantago ovata*) swells and extrudes, upon sowing, a halo of mucilage that completely envelops the seed.^{33–35} The extruded mucilage generates a growth-promoting spermosphere that retains water, regulate nutrients entry, and facilitates interactions with PGPRs.³³ Seed mucilage is of increasing interest for the pharmaceutical, biomedical and food industry and it is usually a composite of pectic, non-cellulosic, and cellulosic polysaccharides.³⁶ In this study, we have been inspired by the multifunctional coat of myxospermous seed to develop a seed coating technology with programmable function that creates a spermosphere that positively affects the seed niche and promotes water stress-resistance in semi-arid soils.

Results

Seed coating

The coating consists of a two-layered structure (Fig. 1). An SEM image of the dry two layered coating deposited on the surface of a *Phaseolus vulgaris* seed is depicted in Fig. 1a. The inner layer (Layer 1) is made of a 1:3 by weight mixture of silk fibroin and trehalose that contains *Rhizobium tropici* CIAT 899 – referred to as *R. tropici* CIAT 899 onwards. Silk fibroin is a 395kDa structural protein that is extracted from the *Bombyx mori* silk cocoon with a yield of 70-75% and is the main component of the silk textile fibers.^{37,38} Through a water-based process applicable also to silk waste – cocoons unsuitable for reeling, yarn waste and garneted stock – silk fibroin can be regenerated in a water suspension and then easily applied to form coatings through dip-coating and spray-coating techniques.³⁹ The physical, mechanical, biological and biodegradation properties of silk fibroin can be modulated by controlling the protein secondary and tertiary structures (random coil, alpha helix and beta-sheet) at the point of material assembly or with post-processing techniques (silk fibroin polymorphism).^{40–44} Silk fibroin is also known to preserve biological entities and

biomolecules encapsulated in silk-based materials by providing a barrier to oxygen stress and minimize contact with water.⁴⁵ Trehalose is a disaccharide ubiquitously used by natural organisms to impart osmoprotection and that can act as carbon source for rhizobium.^{46–48} The mixture of silk fibroin and trehalose enables Layer 1 adhesion to the seed coat, preservation of *R. tropici* CIAT 899 by mitigating oxidative and osmotic stresses, and release of the biofertilizer upon sowing.^{45,48,49} Given Layer 1 coating thickness (t), an ellipsoidal seed shape (a, b, c), the known concentration of *R. tropici* CIAT 899 (C_a) and assuming a homogeneous dispersion of *R. tropici* CIAT 899 and the formation of a homogeneous coating, it is possible to estimate the number of *R. tropici* CIAT 899 in the inoculum (N) by multiplying C_a with the volume (V) as shown in equation (1).

$$N = C_a \times V = C_a \times \left(\frac{4}{3}\pi ABC - \frac{4}{3}\pi abc \right) \quad (1),$$

where $A=a+t$, $B=b+t$, and $C=c+t$. Using $a = 0.25$ cm, $b = 0.25$ cm, $c = 0.5$ cm, $t = 0.0005$ cm, and $C_a = 10^{10}/\text{cm}^3$, then $N = 6.56 \times 10^6$ *R. tropici* CIAT 899 were encapsulated per seed. The external layer (Layer 2) is a mucilage-like mixture of pectin-carboxymethyl cellulose (P:C 1:1) that contains nutrients and upon sowing forms a hydrogel that acts as a water jacket and provides a suitable environment for rhizobia resuscitation and growth (Fig. 1a-b). Layer 2 was designed as food gel⁵⁰ and contains Ca^{2+} ions that act as crosslinker for pectins' galacturonic acid residues, providing stability to the gel and yielding a gel content of circa 65% (Fig. 1c). CMC molecules present in Layer 2 fill the gaps in the pectin network and confer water superabsorption properties and enhance water retention. In Supplementary Fig.1, we report the investigation of P:C hydrogels volume variation in 154 mM NaCl solution due to water absorption and gel content (GC) as a function of relative pectin and carboxymethyl cellulose (CMC) content and of increasing Ca^{2+} concentrations. Volume variation upon re-hydration and GC of pectin, CMC and their mixtures indicated that P:C 1:1 provides the best performance as a trade-off between volume variation (indication of water uptake) and GC

properties (indication of gel stability over time). The effects of biologically-relevant low, medium and high Ca^{2+} concentrations in P:C gels (5 mM or LCC, 10 mM or MCC and 15 mM or HCC) were then investigated, given the strong beneficial effect of the dication on rhizobia symbiosis with leguminous plants⁵¹. P:C 1:1 volume increased in the first three hours upon rehydration, decreased at the 6 h time point, recovered and plateaued at 12 h (Fig. 1e). The decrease in volume at 6 h was similar for all the materials considered and could be explained with the wash-off of non-crosslinked CMC molecules. Previously reported maximum swelling of pectin hydrogels crosslinked with glutaraldehyde matched the results obtained in this study,⁵² though the use of toxic crosslinkers should be avoided for agricultural applications.^{53,54} Further, studies of biodegradable hydrogels used to modify soil water holding capacity have reported water absorption in the same order of Layer 2.⁵⁵ Water retention is another important parameter to consider in semi-arid soils as the hydrogels can stabilize the humidity around seeds for extended periods of time, acting as a water buffer in-between watering periods⁵⁶. The measurement of volume variation during air-drying indicated that all the P:C gels considered had similar water loss trends, but HCC and MCC gels could retain water longer than LCC, given the higher initial volumes (Fig. 1f). The use of different pectin to CMC ratio in P:C and lower concentrations of Ca^{2+} ions may further be used to regulate water retention (Supplementary Fig. 2). Water absorption studies for gels were also conducted in soil, to simulate how Layer 2 would swell upon sowing. Volume variations of gels were measured after 24h in three soil water content (SWC) conditions (Fig. 1g). Gels' volume significantly increased with SWC ($p < 0.05$), but Ca^{2+} ions did not have statistically significant effect on the swelling for the concentrations considered ($p > 0.05$). Interestingly, an order of magnitude was lost in P:C water uptake, when compared to swelling in saline solution, probably due to the lower water potential of soil and the compressive forces that soil applies on the gel and that limit swelling. However, even in semi-arid soils (SWC=10%), volume variation around 250% indicated the capability of the coating to extract water from

the environment and make it available to the rhizobacteria and the seed. A pressure of circa 412 Pa is applied on the surface of a *P. vulgaris* seed (1.4x0.8x0.7 cm) sowed 3 cm below the surface of a semi-arid soil ($\rho_{\text{avg}} \cong 1.4 \text{ g/cm}^3$). Additionally, soil has a compression modulus of circa 4-7 MPa⁵⁷, indicating that the hydrogels may have complex interactions with the surrounding soil while reswelling occurs as a combination of soil deformation and expansion in air pockets present in the soil.⁵⁸

Mechanical Properties and Gel Microstructure

Rheological measurements help to choose the required settings for material application onto seed surface. Rheological investigation of LCC, MCC and HCC solutions (i.e. before gelling is triggered) showed a shear thinning behavior (Supplementary Fig. 3) and the stability of the solutions' storage and loss modulus (G' and G'' , respectively) (Fig. 2a) over time upon addition of Ca^{2+} ions into the P:C suspension. Exposure of P:C suspension to NaOH, causes a rapid increase in pH that results in immediate gelation of LCC, MCC and HCC. The rapid gelation hinders the application of the Winter-Chambon rule to calculate gelation time. Nonetheless, the evolution of G' and G'' post gelling was monitored over time (Fig. 2b) and showed an increase of G'' that was positively correlated with Ca^{2+} concentration. To further evaluate the suitability of P:C gels to work as seed coating, we measured the mechanical properties of P:C 1:1 hydrogels through unconfined compression tests (Fig. 2c). MCC hydrogels displayed both the highest compressive strength ($5.43 \pm 0.31 \text{ kPa}$) (Fig. 2d) and Young's modulus ($33.86 \pm 4.60 \text{ kPa}$) (Fig. 2e). Nanoindentation tests were also conducted on dry P:C to determine their capability to sustain transportation and storage periods without being damaged. The measured Young's modulus ($\sim 15\text{-}20 \text{ GPa}$, Fig. 2e) and hardness ($\sim 1\text{-}1.5 \text{ GPa}$, Fig. 2f) were of the same order of currently available seed coatings.⁵⁹ Cross-sectional cryo-SEM was used to evaluate the microstructure of the hydrogels. Micrographs showed an interconnected microstructure with an assumed pore size of $\sim 5 \mu\text{m}$ (Supplementary Fig. S4).

However, the pore shape and dimension may have been affected by the formation of ice crystals during the sample preparation, which is known to artificially increase the pore size, particularly in hydrogels with a weak structural integrity such as LCC.⁶⁰

Support of rhizobacteria growth

To investigate the effectiveness of P:C hydrogels as niche to support *R. tropici* CIAT 899 resuscitation and growth, we designed two experimental set-ups. In the first study, *R. tropici* CIAT 899 were released from dissolving silk films into swelling P:C 1:1, as shown in Fig. 3a. In particular, we used *R. tropici* CIAT 899 harboring a green fluorescent protein reporter (*R. tropici* CIAT 899-GFP) to use the fluorescent intensity signal over time as an indication of bacterial colonization and growth. After verifying that *R. tropici* CIAT 899-GFP could use seed mucilage as carbon source (using simulated basil mucilage as an example, Supplementary Fig. 5), we investigated the migration of *R. tropici* CIAT 899-GFP from silk film into P:C 1:1 formed at increasing concentrations of Ca^{2+} (Fig. 3a) and containing nutrients found in seed mucilage (*i.e.* 20.9 mM (D-+)-xylose, 6.28 mM L-(+)-arabinose, 6.28 mM DL-arabinose, L-8.94 mM rhamnose with a ratio of 30:9:9:14)).⁶¹

The release of *R. tropici* CIAT 899-GFP from silk films and the subsequent P:C colonization was investigated using fluorescent microscopy (Fig. 3b)⁶². Silk films were dissolving gradually over time while green spots, corresponding to *R. tropici* CIAT 899-GFP microcolonies, were growing in size and numbers within the depth of the hydrogels. *R. tropici* CIAT 899-GFP microcolonies were more concentrated at the surface of the LCC and HCC gels at day 7, while *R. tropici* CIAT 899 -GFP colonization was more homogenously distributed in MCC samples. Interestingly, a size gradient could be observed in microcolonies present in MCC hydrogels, with larger colonies visible closer to the source of *R. tropici* CIAT 899-GFP. To further investigate hydrogel colonization from the environment, we designed a second study where dry P:C 1:1 were immersed in a solution containing *R. tropici* CIAT 899-

GFP. GFP intensity in the hydrogels increased over time and plateaued at 32h (Fig. 3c). GFP intensity was inversely proportional to the crosslinker content, i.e. the lower the content of Ca^{2+} ions the higher the GFP intensity. To determine the interplay between Layer 2 swelling and bacteria colonization, we immersed dry P:C 1:1 in a 154 mM solution containing *R. tropici* CIAT 899-GFP ($\text{OD}_{600} \sim 0.1$). Histological sections at 6h and 54h post re-hydration showed the presence of gram-negative *R. tropici* CIAT 899-GFP within the gels (Fig. 3d and Supplementary Fig. 6), suggesting that Layer 2 swelling may be able to recruit endogenous microorganism present in the soil and further attract them post-swelling due to the presence of nutrients in the hydrogel. To further support this assumption, diffusion of saccharides in the hydrogel was studied using fluorescein as a working model. Calculated diffusion coefficients were $\sim 10^{-5} \text{ cm}^2/\text{s}$ and not influenced by the amount of crosslinks present in the hydrogels (i.e. Ca^{2+} concentration during material fabrication) (Fig. 3e and 3f).⁶³ This result suggests that the rate of diffusion of nutrients was not the determining factor of *R. tropici* CIAT 899-GFP growth, but other features such as different pore sizes and geometries may have caused an increased colonization in MCC gels.^{64,65}

Coating biodegradation

The biodegradation of the seed coating was evaluated by exposing MCC hydrogels to soil. Biopolymers degradation in the absence of light is mostly catalyzed by microbial activity through enzymatic degradation that accelerate proteolytic and carbohydrolytic processes. Important parameters such as type of indigenous microorganisms, temperature, soil type, composition, pH, salinity, water and carbon contents play prominent roles in determine how quickly biopolymers may be degraded. Fig. 4a depicts the mass loss of MCC hydrogels over time when investigated at 16°C in a soil containing microorganisms. At day 14 and 28, almost 50% and 70% of the dry weight of the gel was lost. When exposed to soil containing sodium azide – a bacteriostat – the mass loss rate of MCC decreased for the first 14 days, when

compared to untreated soil, indicating the critical role of microorganism in biopolymer degradation in soil. For longer time points (up to day 28), the mass loss increased and reached values similar to the one measured for untreated soil, probably indicating the inefficacy of the sodium azide treatment in the long term. Biodegradation studies were also conducted in untreated soil at 25°C and resulted in the complete biodegradation of LCC, MCC and HCC materials in less than 30 days (Supplementary Fig. 7). Biodegradation in sterilized soil only resulted in less than 50% of mass loss over a period of 30 days. The complete biodegradation of the coating within the life cycle of the plant is an important feature to minimize the environmental impact of agriculture and the release of pollutant in the soil. Recently, policymakers have promoted new laws that strictly regulate the biodegradation of polymers released in the environment for agricultural application. For example, in January 2018 the European Chemicals Agency (ECHA) examined the need for an EU-wide restriction on the use of intentionally added microplastic particles (IAMPs) in products placed on the EU market, including food and agriculture, with non-biodegradable microplastics forecasted to be banned in 2025⁶⁶.

Seedling germination and growth in semi-arid conditions

All together, these results suggest that the seed coating properties may be programmed by varying several parameters including relative P:C concentration, amount of Ca²⁺ during materials fabrication and presence of nutrients to design an environment that can support resuscitation and growth of plant growth promoting rhizobacteria. In particular, for the purpose of our study, we selected MCC as the hydrogel of choice to conduct germination studies, given the combination of homogenous *R. tropici* CIAT 899 colonization, water absorption and mechanical properties. In a preliminary study, germination of *P. vulgaris* seeds was tested in a greenhouse setting by applying the following treatments: i) no treatment (negative control), ii) inoculation with 3% poly(vinylpyrrolidone) (PVP) solution with *R.*

256 *tropici* CIAT 899-GFP (positive control), iii) bilayer seed coating with Layer 2 applied via
257 spray-coating, and iv) bilayer coating with Layer 2 applied via dip-coating. Seedlings were
258 checked for nodulations at day 14 after germination, to ensure successful delivery of
259 rhizobacteria and root colonization (Fig. 4b). A 100% nodulation rate was observed for all
260 three procedures where *R. tropici* CIAT 899 were added to the soil. Application of Layer 2
261 both via spray-coating and dip-coating resulted in roots that were more developed when
262 compared to the positive control, suggesting that the seed enhancement technology
263 outperforms current standard materials and can be applied with tools largely available by
264 growers.

265 To further test the efficacy of the two-layer coatings (L2) in mitigating environmental
266 stressors typical of semi-arid regions, we conducted germination studies at the UM6P
267 experimental farm in Ben Guerir, Morocco, using native soil (Fig. 4c). Soil analysis revealed
268 that a composition typical of sandy soils, slightly alkaline and prone to induce drought stress
269 due to limited water retention capacity. The soil was rich in nitrogen and phosphate and
270 concentrations of oligoelements such as manganese and zinc were sufficient, while there was
271 a slight deficiency in copper content (Supplementary Table 1). To investigate the
272 effectiveness of L2 to induce water-stress tolerance, we exposed L2-coated *P. vulgaris* seeds
273 to soils with a decreasing water potential (Ψ_s), by altering the watering conditions to induce
274 water stress regimes (Fig. 4c). Seeds with no coating (C) and seeds coated only with Layer 1
275 (L1) were used as control and one-way ANOVA test with Bonferroni's correction was used to
276 analyze the data collected. When comparing the different watering conditions in 150 g of soil
277 per plant, it was observed that for mild ($\Psi_s = -12$ kPa) and severe ($\Psi_s = -20$ kPa) water-
278 stressed conditions, L2 coating statistically significantly influenced germination and plant
279 health, when compared to the two controls. L2 seeds resulted in shoot dry mass that were
280 higher, respectively, when compared to C and L1 seeds for Ψ_s at -1 kPa, -12 kPa and -20 kPa.

Shoot length was not a factor that seemed to be determined by coating maybe because this was early on in the growth process. No statistical significant difference was found for shoot length between L2, L1 and C seeds for $\Psi_s = -20$ kPa. It was also observed that root architecture was significantly affected by seed and water treatments. Seeds germinated in soil with $\Psi_s = -1$ kPa and $\Psi_s = -5$ kPa had the longest roots in comparison with seeds germinated in water stress conditions, i.e. $\Psi_s = -12$ kPa and $\Psi_s = -20$ kPa. Under water stress regimes, roots in L2 and L1 seeds were statistically significantly longer when compared to C seeds, indicating that the coating treatments provided a better environment for early root establishment. Measurement of chlorophyll content in leaves showed that watering regime but not seed coating affected the production of the green pigment. Measurement of total phenolic compounds (TPC) is correlated to drought stress as plants release phenolic compounds to mitigate water deficit.⁶⁷ TPC was statistically significantly lower for L2 seeds when compared to L1 and C seeds, indicating the contribution of the designed hydrogel to induce water-stress tolerance. Measurement of stomatal conductance corroborated this finding. *P. vulgaris* cultured in water stress regimes lower stomatal conductance to limit water loss.⁶⁸ In our experiments, L2 seeds showed statistically significant higher stomatal conductance for $\Psi_s = -1$ kPa, -5 kPa and -12 kPa, when compared to the L1 and C seeds and a higher stomatal conductance for $\Psi_s = -20$ kPa to C seeds, indicating a better tolerance to water deficiency. Altogether, these results indicate that the L2 coating positively influence plant establishment of *P. vulgaris* in a semi-arid soil and under water stress.

Discussion

In this study we investigated the use of a biomaterials and drug delivery approach to engineer a programmable seed coating technology for effective delivery and growth of rhizobacteria in the spermosphere. Activated upon sowing, the two-layered coating technology enabled the

resuscitation and self-replication of rhizobia within a biopolymer-based hydrogel that resembles seed mucilage and resulted in the formation of microbial colonies that formed symbiotic nodules with plant roots and induced water stress tolerance in semi-arid conditions. Furthermore, the use of biopolymers generally used in food gels provide a largely available, cost-effective and non-toxic solution to mitigate abiotic stress. Together, these findings open the door to the use of enhanced seed coating technologies to address specific weather and soil conditions, to adapt agriculture to changes in climate patterns while also minimizing the use of scarce and energy-intensive inputs.

Methods

Materials

Materials fabrication, hydration and dehydration studies are reported below. Extensive details of the experimental procedures are reported in Supporting Information

Hydrogel fabrication: Four types of gels were prepared with different ratios of low-methoxylated pectin (P) from citrus peel (> 74% galacturonic acid, > 6.7% methoxy groups, Sigma-Aldrich, St. Louis, MO) to NaCMC (C) (molecular weight 90,000 g/mol, degree of substitution 0.7, Sigma-Aldrich, St. Louis, MO). The tested P:C ratios were 1:0, 3:1, 1:1, and 1:3, but the total amount of polysaccharides was 5 wt% for all solutions. For the gel preparation, monomeric sugars (D-(+)-xylose 20.9 mM, L-(+)-arabinose 6.28 mM, DL-arabinose 6.28 mM, L-rhamnose 8.94 mM (Sigma-Aldrich, St. Louis, MO) with the ratio 30:9:9:14) were first added to deionized (DI) water to mimic the ratio found in Basil seeds⁶¹. After complete mixing of the polysaccharides in the sugar solution, three calcium chloride (CaCl₂, Sigma-Aldrich, St. Louis, MO) concentrations were added to each mix from a 1 mM CaCl₂ stock solution in DI water: HCC (15 mM), MCC (10 mM), and LCC (5 mM). After leaving the solution for 24 hours, 70 mol% (mol percentage of polysaccharides) of OH⁻ (from 2 M NaOH stock solution) was added to each gel, to increase the pH and lead to gelation,

according to the egg-box model.⁶⁹ 48 hours later, the samples were prepared and air-dried for 24 hours at RT. From those gels, the best P:C ratio was selected by analyzing both the water content over time and the GC. Only materials with selected P:C ratio were then optimized (pH and CaCl₂ concentrations). Indeed, because the pH is only important for pectin gelation and not NaCMC, the added NaOH amount had to be adjusted: 70 mol% of galacturonic acid (pectin's principal monomer) rather than 70 mol% of all polysaccharides. CaCl₂ concentrations subsequently needed to be increased for the mechanical integrity of the gel. Finally, full characterization was only done for the materials with the selected P:C ratio.

Hydration studies

To understand the swelling kinetics of the selected hydrogels, a hydration study was conducted in solution. Wet 100 mm mesh nylon tea bags (DulytekTM, Seattle, WA) after five flicks and dry hydrogel samples (ten 3.5 mm punches, n = 3) were weighed (Mettler Toledo, Greifensee, Switzerland). To measure the swelling, the gel punches were placed in the tea bags in 154 mM NaCl solution (Sigma-Aldrich, St. Louis, MO) and their weights (punch + bag) were repeatedly measured after 1 h, 3 h, 6 h, 12 h, 24 h, and 36 h. After each time point, the bags with the punches were put back in solution. The swelling ratio (SR) was calculated as follows: $SR = (W_s - W_i) / W_i * 100$, with W_s being the weight of the swollen samples at the different time points and W_i being the initial weight of the dried sample before the experiment. W_s was calculated as the mass of the wet tea bag with the wet sample minus the weight of the wet tea bag. Another hydration study was then conducted in soil (African Violet Potting Mix, Scotts Miracle-Gro, OH) to highlight the effect and interaction of compressional soil forces vs. hydrogel swelling forces. Dried punches of each gel (8 mm diameter, n = 3) were put in 100 mm mesh nylon tea bags. Dried punches were weighed beforehand. Samples in bags were buried in the soil at three different moisture conditions (hydrated with deionized water): 10%, 25%, and 50% over a 24-hour period. Hydration was monitored with a hydrometer (Soil Moisture Sensor and LabQuest mini, Vernier, Beaverton,

OR). Then, tea bags were taken out and cut to retrieve the wet punches, which were weighed. Their SR was finally calculated according to the SR equation previously described.

Dehydration

Dry hydrogel samples (ten 3.5 mm punches per sample, $n = 5$), as well as wet and dry 100 mm mesh nylon tea bags, were weighed. Dried samples (W_i) were put in the tea bags and immersed in 154 mM NaCl solution for 12 hours. Teabags containing the samples were then retrieved from the water and flicked five times before being weighed to determine the SR as described above. Samples were then let to air-dry at RT. From there on, weight was measured every hour (W_s) until the samples were dry to obtain a water content curve (changes in SR over time). W_s was calculated as the mass of the teabag with the sample minus the weight of the teabag (wet weight for the first hour and then dry weight). The pH of the 154 mM NaCl solution in which the bagged samples were swollen was measured after 12 hours with a pH meter (Orion Dual Star, ThermoFisher Scientific, Waltham, MA).

Gel content

The dry weight of 8 mm hydrogel punches ($n = 3$) was measured before immersing them in 154 mM NaCl solution for 12 hours (W_i). The samples were then air-dried for 24 hours at RT and their weight measurement was taken again (W_d). The gel content (GC) was calculated as follows^{70,71} :

$$GC = \frac{W_d}{W_i} \cdot 100 \quad (2)$$

with W_i being the initial dry weight of the gel and W_d corresponding to the weight of the dried sample after swelling and deswelling the initial dry sample.

Macro and micromorphology

To have a visual sense of the swelling volume and integrity of the selected hydrogels, pictures of 8 mm punches were taken in the dry state and after being swelled in 154 mM NaCl solution for 12 hours (camera D3400, Nikon, Tokyo, Japan). Pictures were analyzed with ImageJ

software available from <https://imagej.nih.gov/ij/> (National Institutes of Health, MD). The volumes of the gels were calculated by measuring (1) the diameter of each gel in the dry and wet states at four different angles (0°, 45°, 90°, and 125°) and (2) the height of each gel in the dry and wet states at three spots (on each end and in the middle). The volume variation (VV) was calculated with the following equation:

$$VV = \frac{V_s - V_i}{V_i} \cdot 100 \quad (3),$$

with V_i being the initial dry volume of the hydrogel and V_s being the swollen volume of the gel after 12 hours of immersion in 154 mM NaCl solution.

To visualize the microstructure of the hydrogels, 8 mm punches were swollen in 154 mM NaCl solution for 12 hours. Small pieces 2-3 mm were placed in planchets at room temperature and plunged into liquid nitrogen. The frozen planchets were kept cold and transferred into a ACE900 freeze fracture machine. Samples were fractured, revealing the interior, then shadowed with 2 nm of platinum and coated with 10 nm of carbon. Coated samples were imaged directly in the Zeiss Crossbeam 540 FIB/SEM (Thornwood, NY) fitted with a Leica cryostage while held at below -150°C. The LCC and MCC samples were soaked in 20% glycerol, a cryoprotectant for 30 minutes to prevent ice formation prior to freezing.

Mechanical properties

To evaluate the strength of wet samples, non-confined compression tests were conducted (5943 Instron, Norwood, MA) on round gel samples ($n = 3$, diameter 38 mm, thickness 16 mm). The strain speed rate was set to 100% min⁻¹, and the whole experiment was filmed, until a strain compression of 30%. Tangents to all compressive stress-strain curves were calculated at 5% strain, 10% strain, and 15% strain, to evaluate which tangents were fitting the best the linear (elastic) regions of each gel type.

Nanoindentation: Nanoindentation measurements were performed on a Hysitron TriboIndenter with a nanoDMA transducer (Bruker, Billerica, MA). Samples were indented in load control mode with a peak force of 500 µN and a standard load-peak hold-unload function. Reduced

modulus was calculated by fitting the unloading data (with upper and lower limits being 95% and 20%, respectively) using the Oliver-Pharr method and converted to Young's modulus assuming a Poisson's ratio of 0.33 for all samples. Each type of sample was prepared and indented in triplets to ensure good fabrication repeatability. For each sample, indentation was performed at 3 locations a total of 36 points (6×6 grid with an increment of 20 µm in both directions) at each location to ensure statistical reliability of the modulus measurements.

Bacterial growth

To assess bacterial growth in the presence of the selected hydrogels, two experiments were conducted. For both, a growth media was prepared by mixing 5 g of Bacto™ Peptone (BD Biosciences, Franklin Lakes, NJ), 3 g of yeast extract (Sigma-Aldrich, St. Louis, MO), and 10 mL of 0.7 M CaCl₂ solution to 1L of DI water. The following antibiotics (Sigma-Aldrich, St. Louis, MO) were then added: Rifampicin (25 mg/mL), Nalidixic acid (20 mg/mL), and Tetracycline (10 mg/mL). The media was sterilized by autoclaving for 50 minutes at 120°C. *R. tropici* CIAT 899 was grown in media overnight (28°C, 250 rpm) in 14 mL falcon round-bottom tubes (Corning Inc., Corning, NY). The tubes were then centrifuged at 4200 rpm for 10 minutes (5910 R, Eppendorf, Hauppauge, NY) and the bacterial pellets were resuspended in 5 mL PBS (BupHTM phosphate-buffered saline packs, ThermoFisher Scientific, Waltham, MA). The OD₆₀₀ was finally adjusted to 0.1 before the following two experiments were performed.

Experiment 1: Two MCC gels were fabricated, one with the usual basil-inspired monosaccharides and one with the same amount of sucrose instead. Because it is known that *R. tropici* CIAT 899 can digest sucrose, this experiment would thus give information about the bacteria digestion of basil sugars. 48 hours after the NaOH addition, twelve square hydrogel samples (1 cm³) were cut out and air-dried for 24 hours at RT. The gels were then rehydrated in PBS for 12 hours in 15 mL tubes. For controls (n = 3), the PBS was renewed, and for positive samples (n = 3), the initial PBS was replaced by PBS with *R. tropici* CIAT

899 (OD600 = 0.1). *R. tropici* CIAT 899 synthesizing green fluorescent protein (*R. tropici* CIAT 899-GFP)^{62,72} were used to conduct hydrogel colonization studies. GFP fluorescence intensity of the solution was repeatedly measured at the following time points: 5 h, 8 h, 24 h, 30 h, 36 h, and 48 h (Safire2, Tecan, Switzerland). The excitation light was 491 nm, the emission light was 530 nm, and the gain was set to 70. GFP fluorescent intensity gave information on the amount of the fluorescent reporter expressed by *R. tropici* CIAT 899-GFP and it was used as an indication of living bacteria and their growth.

Experiment 2: The fluorescence experiment was repeated with the three P:C 1:1 gels (LCC, MCC, and HCC). The time points at which fluorescence was repeatedly measured were 2 h, 4 h, 6 h, 9 h, 15 h, 24 h, 27 h, 30 h, 33 h, 39 h, and 48 h.

Histological sections

To determine if bacteria were attracted by the hydrogel and migrated inside them, pieces of hydrogels were retrieved after 6 and 54 hours of incubation with *R. tropici* CIAT 899-GFP, and fixed in 10% formalin (Sigma-Aldrich, St. Louis, MO) for 18 hours, washed twice with PBS, and casted into HistoGel™ (ThermoFisher Scientific, Waltham, MA). Distinct samples were used per each time point considered. After thirty minutes, two slices were cut out from the sample and placed in a cassette in formalin for an additional hour. Finally, all gels were conserved in 70% ethanol before being brought for histology (Gram-negative staining, Hope Babette Tang Histology Facility, Koch Institute, MIT, MA).

Migration studies: 1% (w/v) 45 minute-boiled silk fibroin aqueous solution was prepared from silkworm cocoons (Tajima Shoji Co., LTD., Yokohama, Japan) as described in Rockwood et al.³⁹ and mixed with a 1% (w/v) aqueous solution of trehalose (D-(+)-trehalose anhydrous, TCI Chemicals, Tokyo, Japan). The two solutions were added together to a final silk:trehalose ratio of 1:3. *R. tropici* CIAT 899-GFP were centrifuged at 4200 rpm for 10 minutes (Eppendorf 5910 R, Hamburg, Germany) and the pellet was resuspended in the silk:trehalose solution until an OD600 of 0.1 was reached. 50 mL of this solution was drop-

casted on a PDMS sheet and air-dried for 48 hours at RT to form a film. In the meantime, the three P:C 1:1 gels were prepared as explained earlier. Right after the addition of NaOH, each viscous solution was transferred to the inserts of a 12-transwell plate (Corning Inc., Corning, NY) ($n = 4$) to reach a height of 2 mm. The gels were finally air-dried for 48 hours at RT. On top of the dried gel, a dry film was deposited on Day 0. At the bottom of the transwell plate, PBS was added so that only the bottom of the gel touched the solution through the permeable membrane. One sample of each gel was taken out on day 1, day 3, and day 7, i.e. distinct samples were used per each time point considered. The samples were fixed with 10% formalin solution for 30 minutes (Sigma-Aldrich, St. Louis, MO) and incubated with the TrueVIEW™ kit (Vector Laboratories, CA) for 5 minutes to quench the autofluorescence of the hydrogels. A cross-section of each gel was then imaged with a confocal microscope (inverted Ti Nikon1AR ultra-fast spectral scanning confocal microscope, Nikon Tokyo, Japan) at each time point to observe the migration of the microbes across the hydrogel. All images were taken with the exact same parameters. The fluorescent images were then processed with the ImageJ software to obtain a maximum intensity Z-projection.

Diffusion studies

To compare the diffusion of nutrients through hydrogels, 4-mL solutions of LCC, MCC and HCC hydrogel were prepared and added to 5-mL tubes and solidified for 48 hours. Gels were then hydrated in 154 mM NaCl solution for 24 hours before the diffusion study. 1 mL of a solution of fluorescein (100 mM) in water was pipetted on top of each hydrogel and the gels were photographed every 10 minutes for 150 minutes. The distance the fluorophore had traveled was measured at each time point and these data were used to calculate the diffusion rate of small molecules through the hydrogel⁶³. Samples were measured repeatedly for time points considered.

Material biodegradation

Degradation of hydrogels in soil was evaluated over a one-month period. Dry samples of each gel (five 8 mm punches, $n = 3$) were weighed beforehand (W_i). The samples in 100 mm mesh tea bags were buried in 25% hydrated soil. 50 mL silk films containing *R. tropici* CIAT 899 (OD600 = 0.1) were prepared as explained above and one film was added to each tea bag. The moisture condition of the soil was monitored with a hydrometer and DI water was added when needed to maintain the 25% humidity level (5 mL every three days). A sample of each gel was taken out on days 1, 3, 7, 14, and 28 (measurements were then taken from distinct samples). They were quickly washed in 154 mM NaCl solution and then air-dried for 24 hours at RT before their weight was measured (W_d). Degraded gel (DG) amount was calculated as described in equation (4):

$$DG = \frac{W_i - W_d}{W_i} \cdot 100 \quad (4)$$

where, W_i is the initial dry weight of the sample and W_d the dry weight of the samples after some time in PBS and bacteria or in soil, depending on the experiment. To investigate the effects of a reduced microbial activity on the material biodegradation, 2% sodium azide solution was mixed with the soil (1 ml per 1 g of soil)⁷³. Biodegradation studies were conducted at 16°C and 24°C.

Rheology: Isothermal gelation studies were conducted with a TA Instruments (New Castle, DE) stress-controlled AR-G2 rheometer with a 40 mm, 2° cone-and-plate fixture at 25°C. NaOH was added to the polymer mixture to induce bond formation (time $t = 0$) then immediately transferred onto just rheometer plate, and measurement started at $t = 60-90$ s. For time sweeping tests, storage moduli G' and loss moduli G'' were monitored as a function of time at a 1 Hz frequency and a 2% stress strain under constant temperature (25°C).

Fluorescence Calibration

To convert all fluorescence intensity numbers to OD600 values, a calibration curve was made. *R. tropici* CIAT 899 were diluted in PBS at seven different OD600: 0, 0.1, 0.3, 0.5, 0.7, 1, and

1.1. The corresponding fluorescence intensities were measured and plotted against the OD600. The excitation light was 491 nm, the emission light was 530 nm, and the gain was set to 70. A linear model was fitted with MATLAB (R2018a, MathWorks, Natick, MA).

Statistical Analysis

Statistical analysis was performed on SRs, GCs, Es, and DGs with MATLAB (R2018a, MathWorks, Natick, MA). Normality of the data was verified using the Jarque-Bera test. A One-Way ANOVA test was applied, followed by pairwise comparison testing if the results showed a statistically significant difference between the groups ($p < 0.05$). Bonferroni's correction was applied to counter the effects of multiple comparisons.

Seed coating: *P. vulgaris* seeds were surface sterilized with 10% bleach for 3 minutes, rinsed in H₂O three times, and left to air dry. 80 mL of *R. tropici* CIAT 899-GFP (OD600 = 1) was centrifuged at 4200 rpm in an Eppendorf centrifuge 5910 R (Hamburg, Germany). The supernatant was discarded and 8 mL of dry 6 wt% silk fibroin-trehalose (1:3) solution was added to the spun down *R. tropici* CIAT 899-GFP. Air-dried seeds were then dipped into this solution for 120 seconds, taken out and left to dry (*Layer 1*). After slightly drying, the seeds were either dip-coated into a hydrogel gelation solution just after adding NaOH and left to dry (*Layer 2 dip*) or slightly sprayed with the hydrogel solution (*Layer 2 spray*). After drying, the seeds were planted at the 24-hour mark.

Growth conditions

Initial assessment of seed growth and nodulation was carried out in African Violet soil. Germination of *P. vulgaris* seeds was tested in a greenhouse setting by applying the following treatments: i) no treatment (negative control), ii) inoculation with 3% poly(vinylpyrrolidone) (PVP) solution with *R. tropici* CIAT 899-GFP (positive control), iii) bilayer seed coating with Layer 2 applied via spray-coating, and iv) bilayer coating with Layer 2 applied via dip-coating^{74,75}. Seedlings were checked for nodulations at day 14 after germination, to ensure successful delivery of rhizobacteria and root colonization. N=18 seeds per treatment were

537 used. Additionally, no statistically significant changes in soil electrical conductivity (EC)
538 were measured in soil where coated and uncoated seeds have germinated (2.13 ± 0.03 and
539 2.12 ± 0.02 mS/cm, respectively), indicating that the ions present in the coating do not
540 significantly affect EC.

541 Following this preliminary assessment, seed growth was carried out in semi-arid soil at UM6P
542 experimental farm in Ben Guerir, Morocco. Watering conditions were adjusted to obtain soils
543 with an average water potential (Ψ_s) of -1kPa, -5 kPa, -12 kPa and -20kPa over a 48 hour
544 time interval. Mild and severe water stress growth condition corresponded to $\Psi_s = -12$ kPa
545 and -20kPa, respectively. Soil tensiometers were made from a ceramic cup connected to an
546 acrylic glass tube and has a 136PC15G1 bridge pressure sensor (Micro Switch, Freeport, IL,
547 USA) on the top to measure the pressure inside the tubing caused by water dynamics between
548 the soil and water filled tube.) Tensiometers were used for each treatment in three replicates
549 to monitor the water potential. Lost water through evapotranspiration was replenished to
550 maintain the desired water potential. Healthy seeds from each treatment were evenly
551 germinated on plastic trays of 10 cm containing 150 g of a substrate mixture composed by
552 30% of sieved sand (2 mm) and topsoil, manually shifted from stone. After adding water
553 treatment solution to each replicate, trays were placed in the growth chamber, where relative
554 humidity was 65 to 70%, night temperature 16°C and day temperature 24°C and photoperiod
555 was 16/8 h. All the treatments were laid out in a completely randomized design, replicated
556 five times, and kept for recording physiological attributes. To find out the role of seed
557 coating in alleviating drought stress and to evaluate the effect on plant establishment and
558 behavior along the growth cycle, three random seedlings were transplanted to large pots
559 (30/20 cm), 20 days after sowing. Pots, containing 4 kg of the substrate mixture, were kept
560 under the greenhouse to allow roots and appropriate leaves development. Irrigation was
561 applied following the same water treatments used before. Different measurements from

distinct samples were taken to investigate the treatment general effects: (1) Shoot length over time; (2) Shoot dry weight; (3) Root length measured with WinRhizo (Regent Instruments Inc., Quebec City, Canada) root scanner; (4) Stomatal conductance was measured using a SC-1 Leaf Porometer (Decagon Devices, Inc. USA); (5) Chlorophyll content was measured with a the CL-01 Chlorophyll Content System (Hansatech Instruments Ltd, Norfolk, England); (6) Total Phenolic Compounds (TPC) was measured by grounding in a mortar containing 5 ml of 50% ethanol solution fragments of leaves and roots (0.5 g FM). The extracts were collected in tubes with lids and labelled, then left in the refrigerator overnight. Upon adding 0.5 ml of chloroform in 3 ml of extract, tubes were vortexed and centrifuged for 5 min at 5,000 rpm. TCP assay was performed using the Folin-Ciocalteu reagent.⁷⁶ Briefly, 0.5 ml of extract, 3 ml of distilled water and 0.5 ml of Na₂CO₃ (20%) were mixed in a test tube. After 3 minutes, 0.5 ml of Folin-Ciocalteu reagent was added. The test tubes were left for 30 min at 40°C before measuring absorbance at 760 nm. The content of phenolic compounds was calculated using gallic acid for the standard curve and expressed in milligrams per gram of fresh leaf matter.

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Data availability

All relevant data are included in the paper and/or its Supplementary Information. All raw data are available from the authors on request.

588

589 **Contributions**

590 A.T.Z., J.L., M.M., B.M. and L.K. designed the study. A.T.Z., J.L., M.M., H.S., S.M., D.K.
591 and H.M.E.F. collected and analyzed the data. All authors contributed to the discussion and
592 interpretation of the results. The manuscript was drafted by A.T.Z., L.L., H.S., M.M, L.K and
593 B.M. and reviewed and approved by the other authors.

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598 **Ethics declarations**

599 N/A

600 **Competing interests**

601 B.M and A.T.Z. are co-inventors in a patent application that describes the coating technology
602 reported in this study. BM is co-founder of Mori, Inc, a company that develops silk-based
603 edible coatings to extend the shelf-life of food.

604

605 **Supplementary Information**

606 Supplementary Figs. 1–7

607 Supplementary Table 1

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Fig. 1

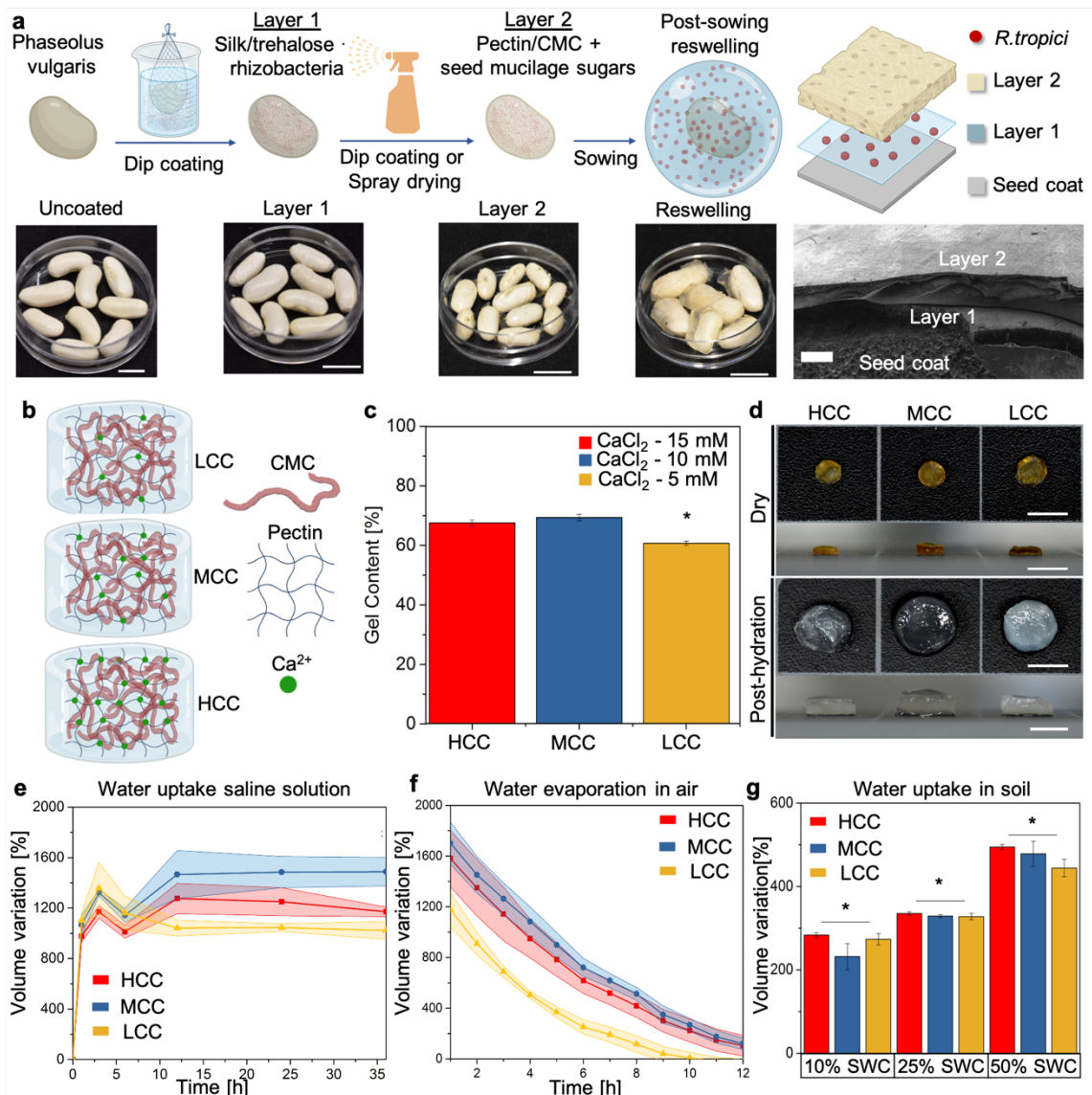


Fig. 1. Material design, fabrication and selection. a) Schematic diagram of the two-layered seed coating fabrication process, relative pictures of *P. vulgaris* coated seeds and crosssection of coated seeds. Layer 1 contains a 1:3 mixture of silk-trehalose that adheres on the silk coat, encapsulates, preserves and releases *R. tropici* CIAT 899. Layer 2 is made of a 1:1 mixture of pectin-CMC (P:C). When the seed is watered, Layer 2 swells into a hydrogel and hydrates Layer 1, which dissolves and releases *R. tropici* CIAT 899. The hydrogel provides an appropriate environment for *R. tropici* CIAT 899 resuscitation and growth. Scale bars for pictures correspond to 10 mm. Scale bar for SEM image of the coating is 10 μm b) Schematic of the pectin-CMC hydrogel structure, where Ca^{2+} are used to crosslink pectin chains while CMC acts as a filler to enhance water uptake. c) Effects of high, medium and low Ca^{2+} concentrations (HCC, MCC and LCC) on gel content. Stars above bars indicate a statistically significant difference in the mean of a material ratio group compared to all other groups (* $p < 0.05$). The black error bars represent the standard error of the mean. d) Representative images of dry and swollen (12h in 154 mM NaCl solution) hydrogels. Scale bars=10 mm. e) Water uptake over time of dried P:C in 154 mM NaCl solution. f) Water evaporation over time of hydrogels after a 12h immersion in 154 mM NaCl solution. Highlighted areas around curves correspond to the standard error of the mean. g) Water uptake of P:C hydrogels in soils of increasing moisture content after 24h. Stars above bars indicate a statistically significant difference in the mean

swelling ratio of a soil humidity group compared to all other groups (* $p < 0.05$). No statistical significant difference was found in the mean swelling ratios between CaCl_2 concentrations at similar soil humidity. The black error bars represent the standard error of the mean.

Fig. 2

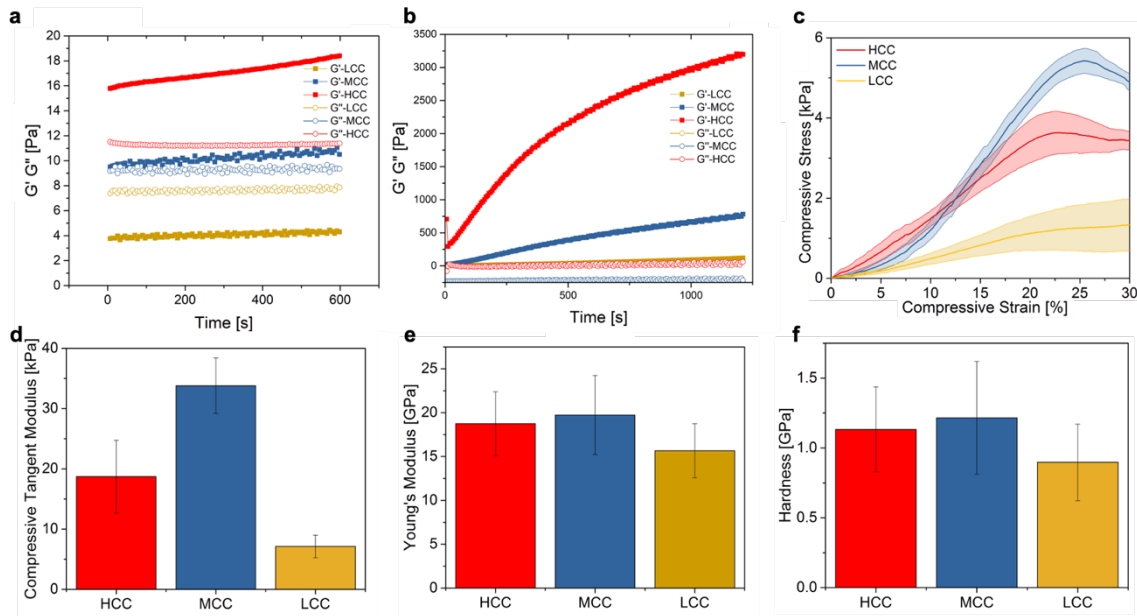


Fig. 2. Mechanical characterization of P:C hydrogels. a) Storage (G') and loss modulus (G'') prior to NaOH addition. b) Storage (B') and loss modulus (G'') after NaOH addition (last step in hydrogel fabrication process). c) Unconfined compression of P:C post hydration for 12h in 154 mM NaCl solution. Highlights around the curves correspond to the standard error of the mean. d) Calculated compressive tangent Young's moduli post hydration for 12h in 154 mM NaCl solution. Statistical significant difference in the mean Young's moduli of MCC compared with LCC (* $p < 0.05$). e) Young's moduli on dry P:C hydrogels measured from nanoindentation studies. No significant statistical difference was measured in the young's moduli of dried materials. f) Hardness of dry P:C 1:1 measured with nanoindentation. No significant statistical difference was measured in the hardness of dried materials. The black error bars represent the standard error of the mean.

Fig. 3

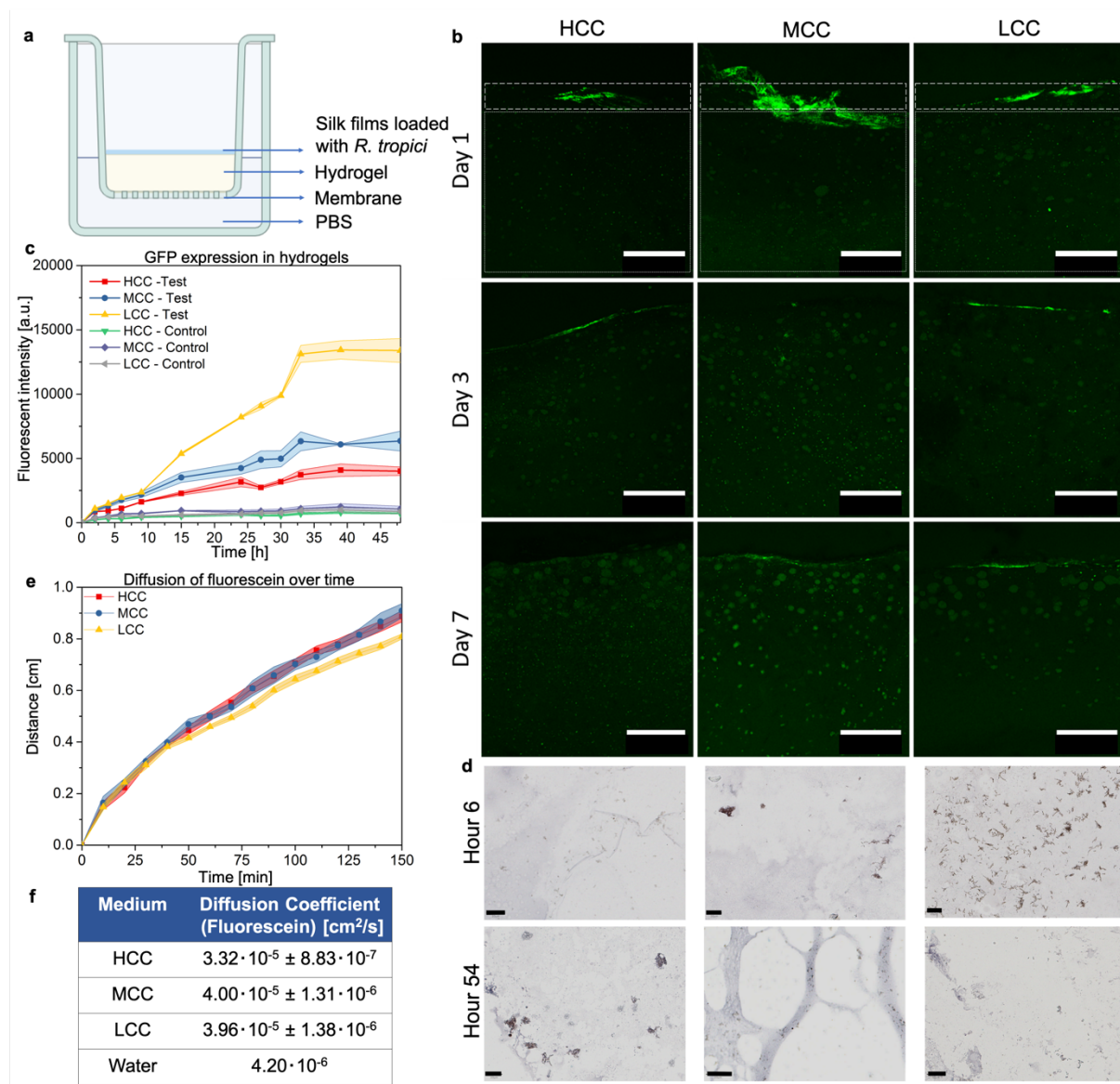


Fig. 3. Use of P:C hydrogels as niche to grow *R. tropici* post rehydration. a) Experimental setup showing *R. tropici* CIAT 899 released from silk films, migrating and growing in P:C hydrogels simulating dry to swollen states. b) Representative confocal cross-sections images showing microbe migration and growth in hydrogels, simulating post-sowing phenomena in soil. Dotted box highlights applied silk film location. Setup of experiment is shown in (A). Scale bars represent 500 μm . c) *R. tropici* CIAT 899-GFP expression in hydrated hydrogels, using the polysaccharides as only energy source. Highlighted areas around curves correspond to the standard error of the mean. d) Representative histology sections of the hydrogels used as only energy source, showing attraction of microbes in the hydrogels. Scale bars represent 20 μm . e) Diffusion of fluorescein in hydrogels to model movement of nutrients/sugars. f) Calculated diffusion coefficients.

Fig. 4

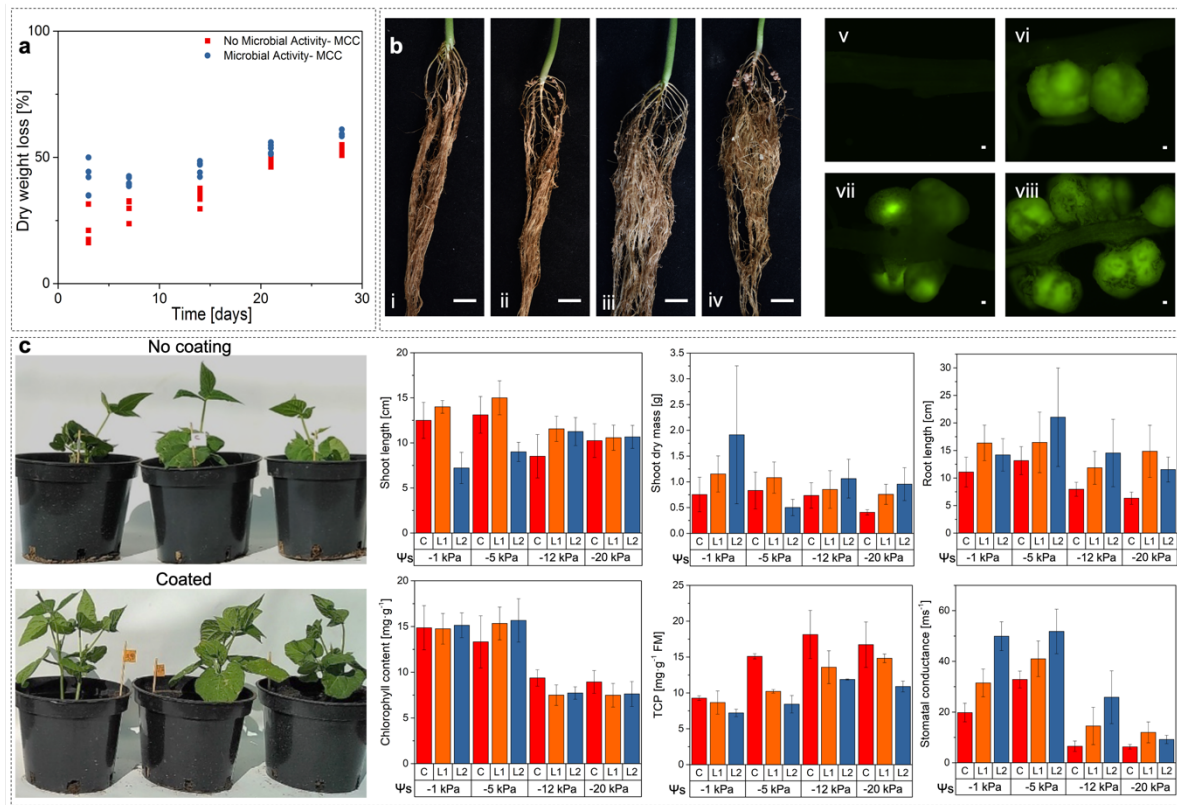


Fig. 4. Degradation in soil of seed coating material and application to *P. vulgaris* to mitigate water stress. a) Degradation of P:C hydrogel with soil microbe activity and without soil microbe activity in 25% humidity soil over a month at 16°C. Dots correspond to collected data. b) Representative root images and corresponding fluorescent microscopy images of nodule formation for plants established from the following treatments: i) and iv) no coating and no inoculation; ii) and vi) inoculation of *R. tropici* CIAT 899-GFP using a 3% PVP solution; iii) and vii) bilayer coating with Layer 2 applied via spray-coating; and iv) and viii) bilayer coating with Layer 2 applied via dip-coating. Scale bars represent 10 mm for roots pictures and 100 μ m for fluorescent microscope images. N=18 plants were tested per treatment type with a 100% nodulation rate for seeds treated with *R. tropici* CIAT 899. No nodulation was visible at day 14 for the negative control. c) Growth of *P. vulgaris* at week 6 in water stress regime for seeds that had no coating and that were coated with L1 and L2 coatings. *P. vulgaris* establishment investigation is shown as a function of coating and water potential (Ψ_s) levels. Ψ_s = -1kPa and -5 kPa correspond to healthy soil moisture content. Ψ_s = -12kPa and -20 kPa represent mild and severe water stress conditions, respectively. *P. vulgaris* plant establishment has been investigating by measuring shoot length, shoot dry mass, root length, chlorophyll content, total phenolic content (TPC) and stomatal conductance. Error bars represent standard deviation; five repeats were used per analysis and condition.