



# A convenient synthesis of short-chain $\alpha$ -(1 $\rightarrow$ 2) mannopyranosyl oligosaccharides



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## ABSTRACT

Sugar 1,2-orthoesters are by-products of chemical glycosylation reactions that can be subsequently rearranged *in situ* to give *trans* glycosides. They have been used as donors in the synthesis of the latter glycosides with good regio- and stereo-selectivity. Alkyl  $\alpha$ -(1  $\rightarrow$  2) linked mannopyranosyl disaccharides have been reported as the major products from the rearrangement of mannopyranosyl orthoesters. Recent studies in this laboratory have shown that  $\alpha$ -(1  $\rightarrow$  2) linked mannopyranosyl di-, tri- and tetrasaccharides can be obtained in one step from mannopyranosyl allyl orthoester under optimized reaction conditions. In addition to the expected mono- and disaccharides (56%), allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1  $\rightarrow$  2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1  $\rightarrow$  2)-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside and allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1  $\rightarrow$  2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1  $\rightarrow$  2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside were obtained in 23% and 6% isolated yields, respectively, from the oligomerization of a  $\beta$ -D-mannopyranosyl allyl 1,2-orthoester, along with small amounts of higher DP oligomers. Possible mechanisms for the oligomerization and side reactions are proposed based on NMR and mass spectrometric data.

## 1. Introduction

The structure of the protecting group on O2 of aldopyranosyl ring donors can greatly affect the stereochemical outcome of chemical glycosylation. For example, *O*-acetyl groups stabilize oxocarbenium ion intermediates by forming stable cyclic acyloxonium ion intermediates [1,2]. The acyloxonium ion intermediate provides two sites for nucleophilic attack by an acceptor, resulting in two different products (Scheme 1), one being the desired 1,2-*trans* glycoside (route a in Scheme 1). The second product is a sugar 1,2-orthoester, which is an undesired by-product of the reaction (route b in Scheme 1) [3–5].

Although 1,2-orthoesters are undesired side products of chemical glycosylation, recent studies have revealed their value as protecting groups and as glycosyl donors [6]. Orthoesters are stable under neutral and basic conditions and can serve as effective transient protecting groups. Glycosyl 1,2-orthoesters hydrolyze to the corresponding 2-*O*-acyl reducing sugars, which serve as precursors in the construction of glycosylation donors. They can also be transformed into 1-*O*-acetates with trifluoroacetic acid. One such application involves the use of 1,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranoside to prepare 2-deoxy-2-[<sup>19</sup>F]

fluoro-D-glucose [7]. Orthoesters also react with alcohols under mildly acidic conditions [8], an example of which is the well-known Johnson-Claisen rearrangement. Sugar 1,2-orthoesters show similar reactivities [9,10] and utilities in the formation of oligosaccharides. Kong utilized them to regioselectively synthesize oligosaccharides with minimally-protected acceptors, and Fraser-Reid utilized them in the synthesis of numerous targets [2,11]. Lindhorst reported more than 20 years ago that an allyl disaccharide formed in an unusual side-reaction during orthoester rearrangement [12]. Kong later found that a methyl  $\alpha$ -mannopyranosyl disaccharide formed from the rearrangement of the corresponding methyl orthoester, and successfully used the latter as the building block in the synthesis of large oligosaccharides [1,6]. A mechanism was proposed [6a,10], and mono- and disaccharides were reported as the only products other than those generated by hydrolysis. Recent studies in this laboratory have shown that additional short-chain *trans*-(1  $\rightarrow$  2) linked oligosaccharides can be synthesized in one step during the self-condensation of sugar allyl orthoesters, and can serve as building blocks in the synthesis of high-mannose *N*-glycans or used directly to prepare mannose oligosaccharides. Self-condensation gave a mixture of short-chain oligosaccharides ranging from mono- to octa-

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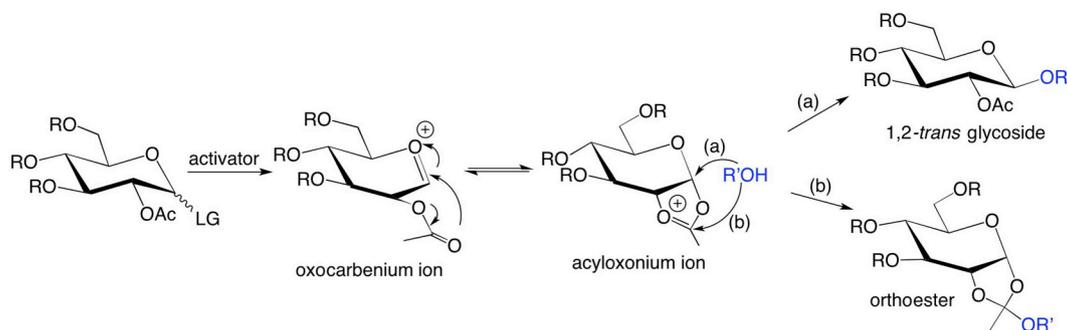
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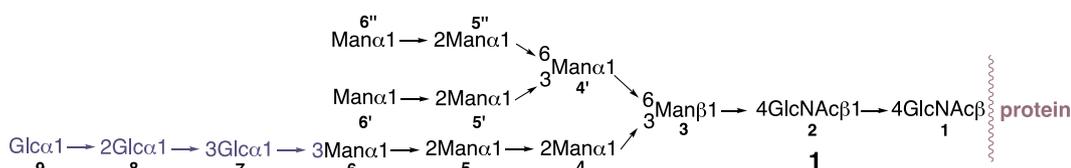
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**Scheme 1.** Formation of (a) a 1,2-*trans*-glycoside via intramolecular rearrangement of an oxocarbenium ion intermediate involving the acyl oxygens at C2, followed by  $S_N2$  ring-opening of an acyloxonium ion intermediate involving R'OH, and (b) an orthoester via nucleophilic attack of the acyloxonium ion intermediate by R'OH.

saccharides, with di- and trisaccharides obtained as the major products (~56% and ~23% isolated yields, respectively).

spectrum confirmed the *O*-glycosidic linkage to be  $\alpha$ -(1  $\rightarrow$  2) based on the H2' signal for the O2-acetylated  $\alpha$ Man residue being significantly



## 2. Results and discussion

### 2.1. Characterization of reaction products

*Trans* (1  $\rightarrow$  2) *O*-glycosidic linkages involving  $\alpha$ -D-mannopyranosyl residues are abundant in high-mannose *N*-glycans such as **1** and other biologically important glycoconjugates. These sub-fragments have been synthesized from D-mannose via transglycosylation using  $\alpha$ -mannosidases [13]. For example, Matsuo and coworkers obtained ~14 mg of  $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ ManOH from 10 g of D-mannose using this approach. Subsequent chemical syntheses of *trans* (1  $\rightarrow$  2) linked  $\alpha$ -mannopyranosyl di- and trisaccharides have been reported as precursors to the  $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man arm attached to  $\beta$ Man (residue 3) in high-mannose *N*-glycan **1** in improved yield over enzymatic synthesis [14]. The latter chemical methods involved stepwise construction using a mannose monosaccharide as the starting material. For example, Matsuo and coworkers prepared a mono-glucosylated high-mannose dodecasaccharide using a mannose trisaccharide precursor [14c]. This route involved more than ten steps to give an orthogonally protected mannosyl trisaccharide precursor in an overall yield of <10%.

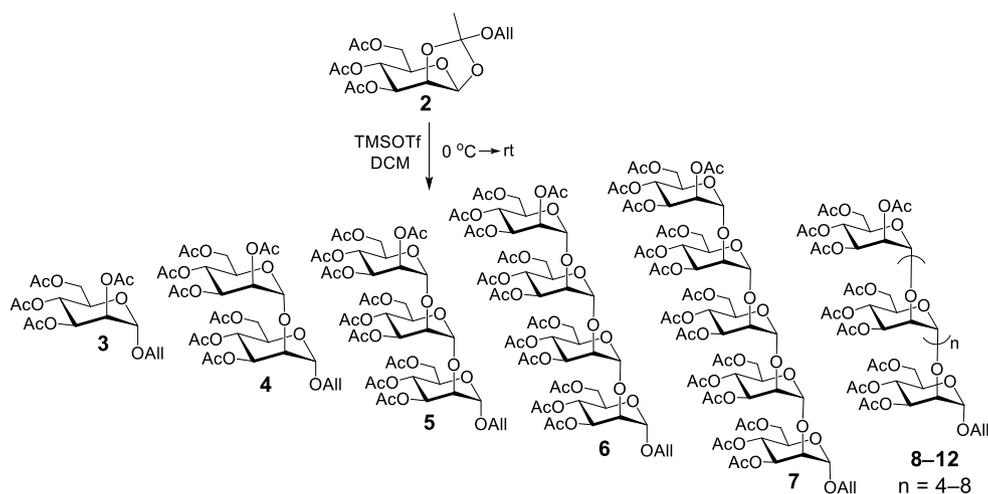
During recent chemical syntheses [15] of nested fragments of **1**, the  $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man arm appended to  $\beta$ Man (residue 3) was to be constructed from a mannose monosaccharide acceptor and a  $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man disaccharide donor. The latter was to be derived from an  $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man product obtained from the self-condensation of an acylated mannosyl orthoester [10,12]. It was found that the self-condensation of 3,4,6-tri-*O*-acetyl-1,2-*O*-(allyloxyethylidene)- $\beta$ -D-mannopyranose (**2**) produced polar products in addition to the expected monosaccharide **3** and disaccharide **4**, all having lower  $R_f$  values on TLC than **2** (Scheme 2). After column chromatography, the two major reaction products were isolated in high purity and were identified as [ $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man] $_x$  congeners. The least polar and most abundant congener was found to be  $\alpha$ -D-Man-(1  $\rightarrow$  2)- $\alpha$ -D-Man-OAll (**4**) as determined by NMR and MS analyses, which is consistent with previous reports [12]. 1D  $^1\text{H}$  (Fig. 1A) and  $^{13}\text{C}\{^1\text{H}\}$  NMR spectra of **4** (see Supplementary Information) gave chemical shifts identical to those reported previously [6,12]. Two H1–H2 cross peaks (Fig. 2A) observed at 4.89 ppm/5.22 ppm and 4.93 ppm/4.02 ppm in the 2D  $^1\text{H}$ – $^1\text{H}$  gCOSY

downfield (5.22 ppm) of the signal arising from H2 of the O2-glycosylated  $\alpha$ Man residue (4.02 ppm).

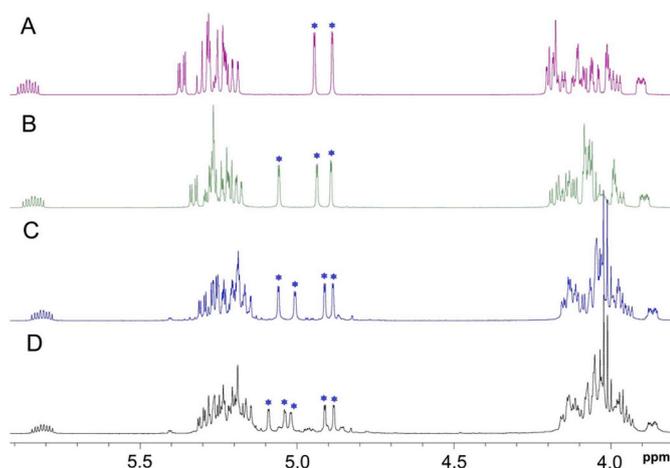
The 1D  $^1\text{H}$  NMR spectrum of the more polar compound contained three signals at 4.9–5.1 ppm (anomeric hydrogen region) (Fig. 1B), suggesting trisaccharide (**5**) as the likely product. Mass spectrometric analysis gave a molecular ion at  $m/z$  987.2938, which matches the calculated mass of **5** (see Supplementary Information). The 2D  $^1\text{H}$ – $^1\text{H}$  gCOSY spectrum (Fig. 2B) confirmed the linkage to be  $\alpha$ -(1  $\rightarrow$  2) in that the three H1–H2 cross peaks at 4.88 ppm/5.22 ppm, 4.92 ppm/3.98 ppm and 5.06 ppm/4.06 ppm indicated that only one H2 resided on a O2-acetylated  $\alpha$ Man residue. The isolated yield of **5** was 23% or less depending on the reaction conditions.

Other products isolated from the reaction mixture included the expected monosaccharide, allyl tetra-*O*-acetyl- $\alpha$ -D-mannopyranoside (**3**), which migrated most rapidly on TLC and gave analytical data matching that reported previously. More polar products, which migrated similarly on TLC, were isolated in reduced purity. 1D  $^1\text{H}$  NMR analysis of the latter (Fig. 1C and D) indicated that two of these products were  $\alpha$ Man-(1  $\rightarrow$  2)-[ $\alpha$ Man-(1  $\rightarrow$  2)] $_2$ - $\alpha$ Man (**6**) and  $\alpha$ Man-(1  $\rightarrow$  2)-[ $\alpha$ Man-(1  $\rightarrow$  2)] $_3$ - $\alpha$ Man (**7**) based on the number of anomeric hydrogen signals observed in their  $^1\text{H}$  NMR spectra (anomeric region 4.8–5.1 ppm), and their molecular masses determined from MS analysis (see Supplementary Information). The 2D  $^1\text{H}$ – $^1\text{H}$  gCOSY spectrum of **6** (see Supplementary Information) is similar to those of **4** and **5** and confirmed its constituent *O*-glycosidic linkages to be  $\alpha$ -(1  $\rightarrow$  2).

Additional five products **8**–**12** migrated on TLC very closely to **6** and **7**, but they could not be fully characterized due to their small quantities and similar polarities. ESI-MS analyses indicated that they are higher molecular weight [ $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man] $_x$  congeners (see Supplementary Information). An attempt was made to use ESI-MS to determine the relative abundances of these products, but the data were complicated by multiple signals caused by molecular fragmentation. LC-MS suggested the presence of minor products having structures different from the expected allyl acylated  $\alpha$ -(1  $\rightarrow$  2)-mannopyranosyl oligosaccharides (see Supplementary Information). Some gave mass fragments consistent with a loss of 40 or 42 amu, indicating the loss of allyl or acetyl groups, respectively. Based on the available analytical data, the  $\alpha$ -(1  $\rightarrow$  2)-mannopyranosyl oligosaccharides shown in Scheme 2 are believed to be the major products generated from the self-



**Scheme 2.** Monosaccharide **3** and allyl  $\alpha$ -(1  $\rightarrow$  2)-mannopyranosyl oligosaccharides **4–12** produced from the self-condensation of 3,4,6-tri-*O*-acetyl-1,2-*O*-(allyloxyethylidene)- $\beta$ -D-mannopyranose (**2**).



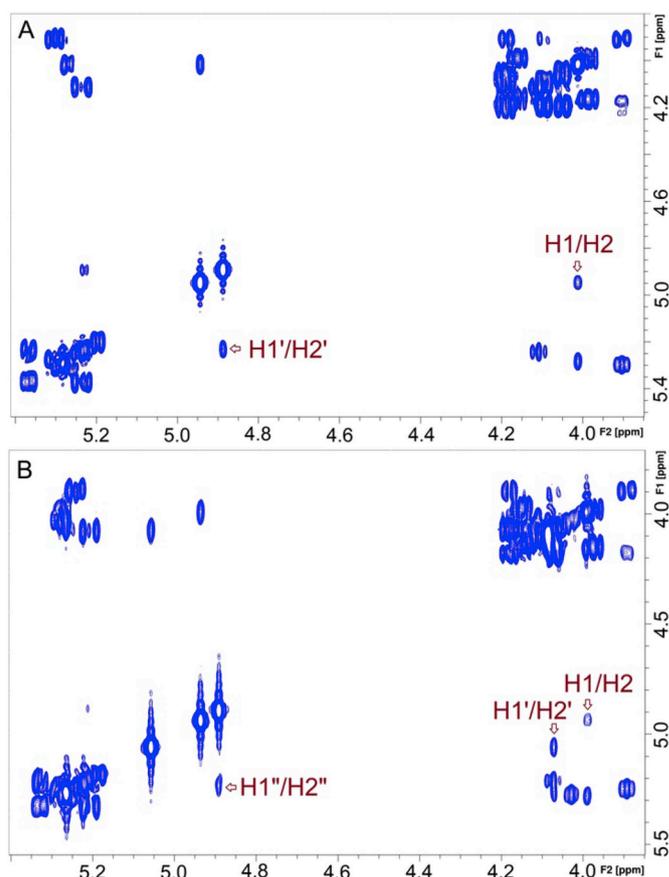
**Fig. 1.** Partial 1D  $^1\text{H}$  NMR spectra of isolated  $\alpha$ Man-(1  $\rightarrow$  2)- $\alpha$ Man linked oligosaccharides **4** (A), **5** (B), **6** (C) and **7** (D) obtained from the self-polymerization of orthoester **2**, showing the anomeric H1 signals labeled with blue stars. (For interpretation of the colored regions in this figure, the reader is referred to the Web version of this article.)

condensation of 3,4,6-tri-*O*-acetyl-1,2-*O*-(allyloxyethylidene)- $\beta$ -D-mannopyranose (**2**).

## 2.2. Mechanistic considerations

Chemical mechanisms proposed by Kong [6] and Linhorst [12] that were invoked to explain the formation of disaccharide **4** do not explain the formation of oligomers **5–12** and the formation of reaction by-products observed during the self-polymerization of **2**. Four oxygen atoms of orthoester **2** [16] can potentially be activated by TMS during rearrangement, leading to different intermediates and products. Based on our results and those in prior reports (Table 1), the activation of the orthoester oxygen and O2 in **2** leads to reaction products **3** and **4**, respectively (Scheme 3). Intermediate **14**, produced via route R1, serves as an electrophile in reaction with TMSOAll to give **3**, and in reaction with intermediate **17**, produced from O2-activated **15** via route R2, to give disaccharide **4**. Routes R1 and R2 are catalytic in that the net consumption of TMSOTf is zero, and if both processes are concurrent, only AcOAll accumulates as a non-carbohydrate end-product.

Route 2 in Scheme 3 is believed to play a key role in the production of oligosaccharides **5–7** as shown in Scheme 4. The condensation of **15**



**Fig. 2.** Partial  $^1\text{H}$ - $^1\text{H}$  gCOSY spectra of two  $\alpha$ Man-containing homo-oligosaccharides containing  $\alpha$ -(1  $\rightarrow$  2) *O*-glycosidic linkages obtained from the self-condensation of **2**. (A) Disaccharide **4**. (B) Trisaccharide **5**.

and **17** gives disaccharide **18**, which is then subject to glycosylation by donor **16** to give trisaccharide **19**. A similar glycosylation of **19** by **16** gives tetrasaccharide **20**. Compounds **18–20** share a common TMS-activated O2 in their non-reducing terminal  $\alpha$ Man residues, rendering them potent nucleophiles in glycosylation reactions with donor **14** (see routes R3–R5 in Scheme 4) generated from route 1 (Scheme 3). In principle, the iterative process described in Scheme 4 can continue beyond **20** to give higher-order oligomers **8–12** (Scheme 2).

**Table 1**  
Product distributions observed from the self-condensation of **2** under different reaction conditions.

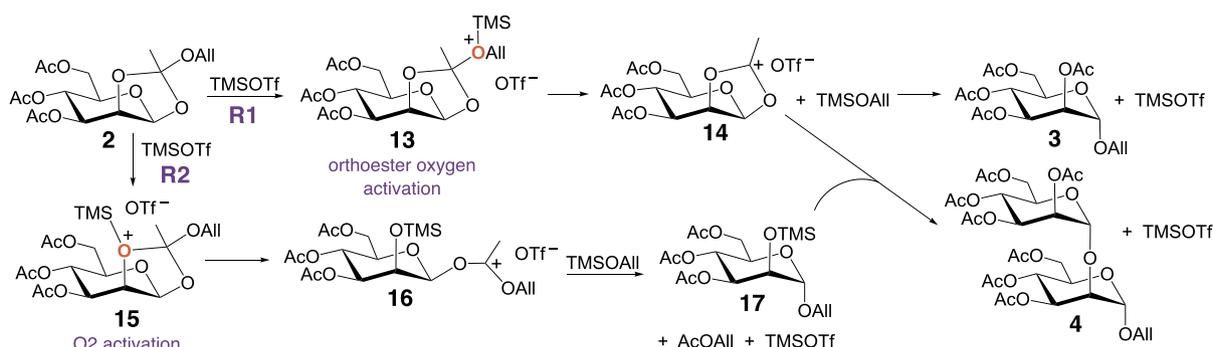
aglycone group	protecting group	conc. (mM)	reaction T (°C)	catalyst (eq.)	product distributions (%)						source
					3	4	5	6	7	other products	
allyl	Ac	350	0	TMSOTf (0.03)	12	51	20	5	<i>np</i> <sup>a</sup>	minor	<b>b</b>
allyl	Ac	160	0	TMSOTf (0.1)	25	47	13	2	1	minor	<b>b</b>
allyl	Ac	322	-30	TMSOTf (0.01)	6	56	23	6	<i>np</i>	minor	<b>b</b>
allyl	Ac	210	-30	TMSOTf (0.03)	6	53	22	6	2	minor	<b>b</b>
allyl	Ac	35	0	TMSOTf (0.03)	12	~38	<i>np</i>	<i>np</i>	<i>np</i>	<i>obs</i> <sup>c</sup>	<b>b</b>
allyl	Ac	130	0	TMSOTf (0.2)	15	29	<i>nr</i> <sup>d</sup>	<i>nr</i>	<i>nr</i>	<i>nr</i>	ref. <b>12</b>
allyl	Bz	300	-40	TMSOTf (0.03)	20	66	<i>nr</i>	<i>nr</i>	<i>nr</i>	<i>nr</i>	ref. <b>6(g)</b>
Me	Ac	NA	-30	BF <sub>3</sub> ·Et <sub>2</sub> O (0.3)	<5	<i>nr</i>	<i>nr</i>	<i>nr</i>	<i>nr</i>	86% ( <b>25</b> )	ref. <b>11(e)</b>

<sup>a</sup> *np* = product detected but not isolated in pure form, so its percentage could not be determined.

<sup>b</sup> *This study.*

<sup>c</sup> *obs* = a significant amount of these products co-eluted with 4–7 but their percentages could not be determined.

<sup>d</sup> *nr* = none reported.



**Scheme 3.** The partitioning of orthoester **2** between two routes, R1 and R2, to give monosaccharide **3** and disaccharide **4**, respectively.

Based on **Scheme 4**, larger oligomers 7–12 should be favored at high reactant concentrations, whereas oligomers 4–6 and monosaccharide **3** should be the major products at low reactant concentrations. However, NMR and MS analyses of partially purified products generated at dilute reactant concentration (35 mM **2**) showed the presence of by-products that co-eluted with  $\alpha$ -(1  $\rightarrow$  2) oligomers. NMR and mass spectrometric (see Supplementary Information) data indicated the existence of two reaction pathways shown in **Scheme 5**. One of these pathways (A; **Scheme 5**) produces monosaccharide **23**, the 3-*O*-deacetylated derivative of **3**. Experimental evidence of the formation of **23** derives from MS data showing a molecular ion having a *m/z* value that is 42 amu smaller than that observed for **3** (loss of -COCH<sub>3</sub>), while the 1D <sup>1</sup>H NMR spectrum of **23** contains an H3 signal shifted upfield, relative to that in **3**, due to O3 deacetylation.

MS analysis of **25** gave a molecular ion having an *m/z* of 659.1816, and NMR showed no evidence of an allyl group in the structure. 1D <sup>1</sup>H and <sup>13</sup>C NMR spectra of **25** gave signals that matched those reported for an  $\alpha,\beta$ -(1  $\rightarrow$  1) linked disaccharide (see Supplementary Information) [11e]. These results indicate that the type(s) of oxygen activation of orthoester **2** by TMS may depend on reactant concentrations and on the Lewis acid used in the activation (TMSOTf vs BF<sub>3</sub>·Et<sub>2</sub>O) (**Table 1**) (**Scheme 6**). For example, O5 activation provides another route to **3**, whereas O1 activation may lead to monosaccharide **31** and  $\alpha,\beta$ -(1  $\rightarrow$  1) disaccharide **32**. The activity of the latter pathway remains uncertain because it is difficult to detect **32** in the presence of **25**. Related side-reactions appear to occur during the rearrangement of galactose orthoester **33**, but a complex mixture of products precluded their purification and characterization (data not shown). Based on LC-MS analysis of the reaction mixture (see Supplementary Information), minor intermediates **21** and **24** may be extended further by reaction with donors **16** and **14**.

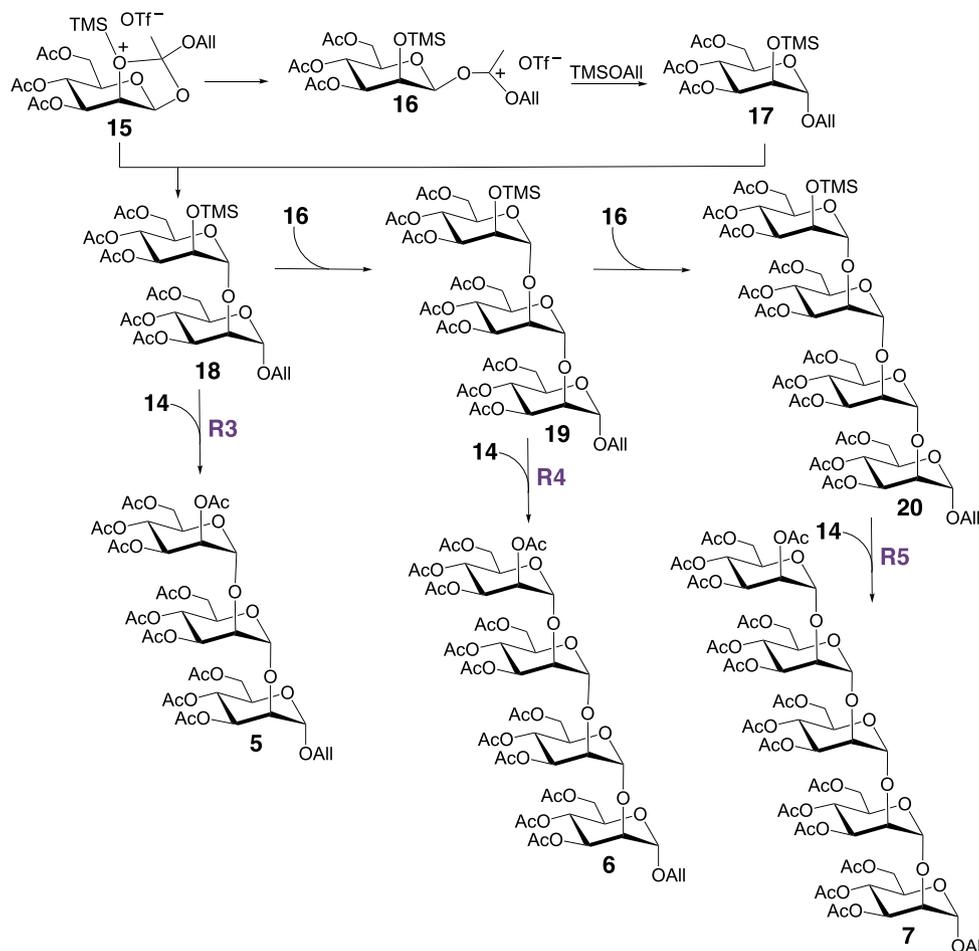
### 3. Conclusions

A rapid and efficient chemical method has been developed to prepare short-chain *trans*-(1  $\rightarrow$  2) linked mannopyranosyl oligosaccharides in one step from the self-oligomerization of a mannopyranosyl allyl orthoester (**2**). The trisaccharide, allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1  $\rightarrow$  2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside (**5**) and disaccharide, allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1  $\rightarrow$  2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside (**4**), were obtained in 23% and 56% isolated yields, respectively, and both are useful precursors in chemical syntheses of nested fragments of high-mannose *N*-glycan **1** [15]. Larger  $\alpha$ Man-(1  $\rightarrow$  2) linked oligosaccharides (tetra- to hexasaccharides) were also obtained in high purity but lower yields for other synthetic applications.

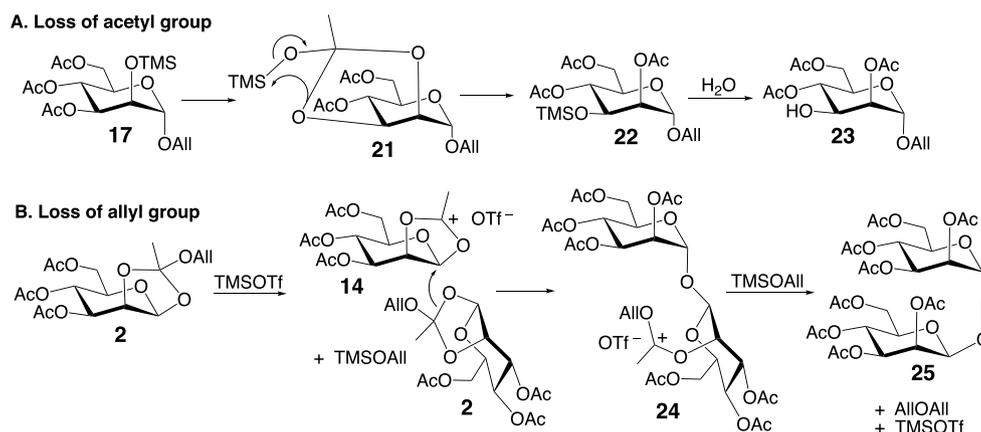
### 4. Experimental section

#### 4.1. General methods

All chemicals were purchased as anhydrous reagent grade and were used without further purification, and all reactions were performed under anhydrous conditions unless otherwise noted. Compound **2** was purchased from PracticaChem without further purification. Reactions were monitored by thin-layer chromatography (TLC) on silica gel pre-coated aluminum plates. Zones were detected by heat/charring with a *p*-anisaldehyde-sulfuric acid visualization reagent [15]. Flash column chromatography on silica gel (preparative scale) was performed on the Reveleris® X2 flash chromatography system. <sup>1</sup>H and <sup>13</sup>C{<sup>1</sup>H} NMR spectra were recorded at 22 °C on a Bruker Avance III HD 500-MHz FT-NMR spectrometer or a Varian DirectDrive 600-MHz FT-NMR



**Scheme 4.** Proposed mechanism for the formation of oligosaccharides 5–7 during the self-condensation of orthoester 2. Intermediates 15–17 arise from route R2 shown in Scheme 3. Both 14 and 16 serve as glycosyl donors in  $S_N2$ -like processes that result in inversion of configuration at C1 of the donor.

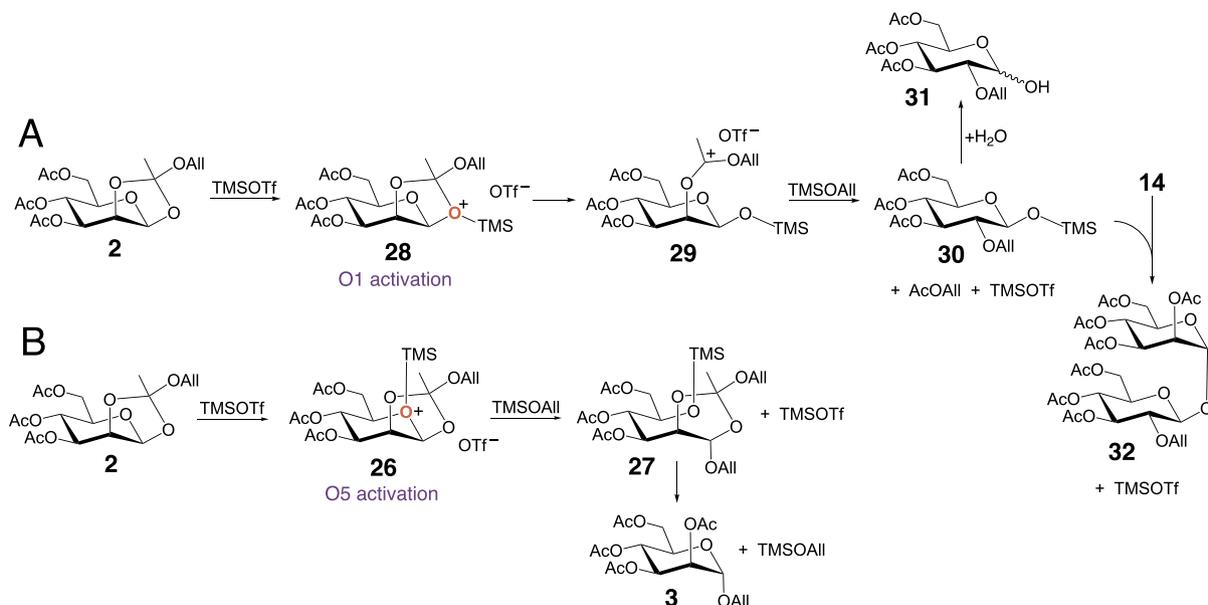


**Scheme 5.** Two reaction pathways detected during the self-condensation of 2, giving by-products 23 and 25.

spectrometer. Chemical shifts are reported in ppm ( $\delta$ ) relative to the  $^1\text{H}$  signal of residual  $\text{CHCl}_3$  at  $\delta$  7.26 ppm and the  $^{13}\text{C}$  signal at  $\delta$  77.23 ppm. Abbreviations for NMR signal multiplicities are: s = singlet; dd = doublet of doublets; d = doublet; dt = doublet of triplets; t = triplet; td = triplet of doublets; q = quartet; m = multiplet. Two-dimensional NMR spectra were recorded on the same instruments using Bruker or Varian data processing software. Mass spectrometric analyses were performed on a Bruker microTOF-Q II quadrupole time-of-flight (QTOF) mass spectrometer with an ESI source.

#### 4.2. General oligomerization reaction conditions

Under a  $\text{N}_2$  atmosphere, 3,4,6-tri-*O*-acetyl-1,2-*O*-(allyloxyethylidene)- $\beta$ -D-mannopyranose (2) (5.43 g, 14.0 mmol) was dissolved in 40 mL of DCM to give a 0.35 M solution, and 4 Å molecular sieves (5.0 g) were added to the solution. The reaction solution was cooled to 0 °C and 0.03 eq. of TMSOTf was added. The reaction solution was stirred at room temperature overnight and then quenched with a few drops of  $\text{Et}_3\text{N}$ . The reaction mixture was vacuum-filtered through a Celite pad, the filtrate was



**Scheme 6.** Other modes of oxygen activation of **2**. (A) O1 activation leading to **31** and **32**. (B) O5 activation leading to **3**.

collected and concentrated at 30 °C *in vacuo*, and the residue was purified by column chromatography to give disaccharide **4** (2.40 g, 3.55 mmol, 51.0%), trisaccharide **5** (0.90 g, 0.93 mmol, 20.1%), tetrasaccharide **6** (0.22 g, 0.18 mmol, 5.0%), and larger oligomers (0.28 g) along with monomer **3** (640 mg, 1.65 mmol, 11.9%) (Scheme 2).

#### 4.3. Analytical data on isolated products

##### 4.3.1. Allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside (**4**)

$^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  5.90 (m, 1H,  $\text{OCH}_2\text{-CH=CH}_2$ ), 5.41 (dd,  $J = 3.6, 10.2$  Hz, 1H), 5.27–5.37 (m, 5H), 5.19 (dd,  $J = 10.1, 3.3$  Hz, 1H), 5.16 (t,  $J = 9.9$  Hz, 1H), 4.98–4.94 (m, 3H), 4.98 (d,  $J = 1.8$  Hz, 1H), 4.94 (d,  $J = 1.8$  Hz, 1H), 4.07–4.27 (m, 8H), 3.95 (ddd,  $J = 9.8, 6.9, 3.6$  Hz, 1H), 2.15 (s, 3H), 2.14 (s, 3H), 2.09 (s, 3H), 2.08 (s, 3H), 2.04 (s, 3H), 2.03 (s, 3H), 2.01 (s, 3H).  $^{13}\text{C}\{^1\text{H}\}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  170.8, 170.5, 170.4, 169.8, 169.8, 169.7, 169.4 ( $\times 2$ ), 133.2, 118.1, 99.5, 99.2, 77.0, 70.4, 69.8, 69.2, 68.6, 68.4, 62.5, 62.2, 21.0 ( $\times 7$ ). HRMS: ( $m/z$ ) calcd for  $\text{C}_{29}\text{H}_{40}\text{O}_{18}\text{Na}^+$  ( $\text{M} + \text{Na}$ ) $^+$  699.2113; found 699.2117.

##### 4.3.2. Allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside (**5**)

$^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  5.88 (m, 1H,  $\text{OCH}_2\text{-CH=CH}_2$ ), 5.38 (dd,  $J = 3.6, 10.2$  Hz, 1H), 5.22–5.35 (m, 8H), 5.11 (d,  $J = 1.8$  Hz, 1H), 4.98 (d,  $J = 1.8$  Hz, 1H), 4.94 (d,  $J = 1.8$  Hz, 1H), 4.15–4.25 (m, 4H), 4.00–4.18 (m, 10H), 3.94 (ddd,  $J = 9.8, 6.9, 3.6$  Hz, 1H), 2.17 (s, 3H), 2.14 (s, 3H), 2.11 (s, 3H), 2.07 (s, 3H), 2.05 (2s, 6H), 2.03 (s, 3H), 2.02 (s, 3H), 2.01 (s, 3H), 1.99 (s, 3H).  $^{13}\text{C}\{^1\text{H}\}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  171.1, 170.8, 170.5, 170.1 ( $\times 2$ ), 169.8 ( $\times 2$ ), 169.5 ( $\times 2$ ), 169.4, 138.1, 118.2, 99.8, 99.4, 97.5, 77.4, 76.7, 70.5, 69.7, 69.6, 69.5, 69.3, 68.7 ( $\times 3$ ), 68.4, 66.3, 66.2, 62.6, 62.2, 62.1, 21.0 ( $\times 10$ ). HRMS: ( $m/z$ ) calcd for  $\text{C}_{41}\text{H}_{56}\text{O}_{26}\text{Na}^+$  ( $\text{M} + \text{Na}$ ) $^+$  987.2959; found 987.2938.

##### 4.3.3. Allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside (**6**)

$^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  5.88 (m, 1H,  $\text{OCH}_2\text{-CH=CH}_2$ ), 5.30 (dd,  $J = 3.6, 10.2$  Hz, 1H), 5.14–5.28 (m, 11H), 5.06 (d,  $J = 1.8$  Hz, 1H), 5.00 (d,  $J = 1.8$  Hz, 1H), 4.91 (d,  $J = 1.8$  Hz, 1H), 4.88 (d,

$J = 1.8$  Hz, 1H), 3.94–4.15 (m, 14H), 3.97 (ddd,  $J = 9.8, 6.9, 3.6$  Hz, 1H), 2.06 (s, 3H), 2.03 (s, 3H), 2.02 (2s, 6H), 1.99 (2s, 6H), 1.98 (s, 6H), 1.95 (2s, 6H), 1.94 (s, 6H), 1.93 (2s, 6H), 1.91 (s, 3H).  $^{13}\text{C}\{^1\text{H}\}$  NMR (150 MHz,  $\text{CDCl}_3$ )  $\delta$  171.1, 170.8, 170.7 ( $\times 2$ ), 170.5, 170.0 ( $\times 2$ ), 169.5 ( $\times 2$ ), 169.4 ( $\times 2$ ), 169.3, 133.1, 118.1, 99.8, 99.6, 99.2, 97.4, 77.0, 76.4, 70.4, 69.8, 69.6, 69.5 ( $\times 3$ ), 69.2, 68.7, 68.6 ( $\times 3$ ), 68.4, 66.4, 66.3, 66.2, 66.1, 62.5, 62.1 ( $\times 3$ ), 21.0 ( $\times 10$ ). HRMS: ( $m/z$ ) calcd for  $\text{C}_{53}\text{H}_{72}\text{O}_{34}\text{Na}^+$  ( $\text{M} + \text{Na}$ ) $^+$  1275.3804; found 1275.3792.

##### 4.3.4. Allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside (**7**)

$^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  5.82 (m, 1H,  $\text{OCH}_2\text{-CH=CH}_2$ ), 5.31 (dd,  $J = 3.6, 10.2$  Hz, 1H), 5.14–5.28 (m, 14H), 5.09 (d,  $J = 1.8$  Hz, 1H), 5.06 (d,  $J = 1.8$  Hz, 1H), 5.05 (d,  $J = 1.8$  Hz, 1H), 4.91 (d,  $J = 1.8$  Hz, 1H), 4.88 (d,  $J = 1.8$  Hz, 1H), 3.92–4.15 (m, 17H), 3.87 (ddd,  $J = 9.8, 6.9, 3.6$  Hz, 1H), 2.06 (s, 3H), 2.04 (s, 6H), 2.01 (2s, 6H), 1.99 (s, 3H), 1.98(3s, 15H), 1.97 (s, 3H), 1.95 (s, 3H), 1.94 (3s, 15H), 1.93 (2s, 6H), 1.91 (s, 3H). HRMS: ( $m/z$ ) calcd for  $\text{C}_{65}\text{H}_{88}\text{O}_{42}\text{Na}^+$  ( $\text{M} + \text{Na}$ ) $^+$  1563.4650; found 1563.4684.

##### 4.3.5. Allyl 2,3,4,6-tetra-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranosyl-(1 $\rightarrow$ 2)-3,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside (**8**)

HRMS: ( $m/z$ ) calcd for  $\text{C}_{77}\text{H}_{104}\text{O}_{50}\text{Na}^+$  ( $\text{M} + \text{Na}$ ) $^+$  1851.5496; found 1851.5466.

##### 4.3.6. Allyl 2,4,6-tri-*O*-acetyl- $\alpha$ -D-mannopyranoside (**23**)

$^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  5.81 (m, 1H,  $\text{OCH}_2\text{-CH=CH}_2$ ), 5.21, 5.15 (m, 2H,  $\text{OCH}_2\text{-CH=CH}_2$ ), 5.00 (dd,  $J = 3.6, 1.7$  Hz, H-2), 4.99 (dd,  $J = 10.1, 9.8$  Hz, H-4), 4.81 (d,  $J = 1.7$  Hz, H-1), 4.19 (dd,  $J = 12.2, 5.5$  Hz, H-6a), 4.08 (m, 1H,  $\text{OCH}_2\text{-CH=CH}_2$ ), 4.02 (dd,  $J = 12.2, 2.3$  Hz, H-6b), 4.01 (dd,  $J = 9.8, 3.6$  Hz, H-3), 3.93 (m, 1H,  $\text{OCH}_2\text{-CH=CH}_2$ ), 3.84 (ddd,  $J = 10.1, 5.5, 2.3$  Hz, H-5), 2.07 (s, 3H), 2.03 (s, 3H), 2.01 (s, 3H). HRESIMS: ( $m/z$ ) calcd for  $\text{C}_{15}\text{H}_{22}\text{O}_9\text{Na}^+$  ( $\text{M} + \text{Na}$ ) $^+$  369.1192; found 369.1228.

## Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.carres.2019.107897>.

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