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Main Manuscript for

Bulk and Grain-Scale Minor Sulfur Isotope Data Reveal Complexities in the Dynamics of Earth's Oxygenation

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This PDF file includes Main Text, Figures 1–5 and their captions.

ABSTRACT:

The disappearance of mass-independent sulfur isotope fractionation (S-MIF) within the ~2.3-billion-year-old (Ga) Rooihooft Formation has been heralded as a chemostratigraphic marker of permanent atmospheric oxygenation. Reports of younger S-MIF, however, question this

narrative, leaving significant uncertainties surrounding the timing, tempo and trajectory of Earth's oxygenation. Leveraging a new bulk quadruple S-isotope record, we return to the South African Transvaal Basin in search of support for supposed oscillations in atmospheric oxygen beyond 2.3 Ga. Here, as expected, within the Rooihoochte Formation, our data capture a collapse in $\Delta^{34}\text{S}$ values and a shift from Archean-like $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ slopes to their mass-dependent counterparts. Importantly, interrogation of a $\Delta^{33}\text{S}$ -exotic grain reveals extreme spatial variability, whereby atypically large $\Delta^{33}\text{S}$ values are separated from more typical Paleoproterozoic values by a subtle grain-housed siderophile-enriched band. This isotopic juxtaposition signals the co-existence of two sulfur pools that were able to escape diagenetic homogenization. These large $\Delta^{33}\text{S}$ values *require* an active photochemical sulfur-source, fingerprinting atmospheric S-MIF production after its documented cessation elsewhere at ~ 2.4 Ga. By contrast, the $\Delta^{33}\text{S}$ monotony observed in overlying Timeball Hill Formation, with muted $\Delta^{33}\text{S}$ values ($<0.3\text{‰}$) and predominantly mass-dependent $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ systematics, remains in stark contrast to recent reports of pronounced S-MIF within proximal formational-equivalents. If reflective of atmospheric processes, these observed kilometer-scale discrepancies disclose heterogeneous S-MIF delivery to the Transvaal Basin and/or poorly resolved fleeting returns to S-MIF production. Rigorous bulk and grain-scale analytical campaigns remain paramount to refine our understanding of Earth's oxygenation and substantiate claims of post-2.3 Ga oscillations in atmospheric oxygen.

SIGNIFICANCE STATEMENT

The permanent disappearance of mass-independent sulfur isotope fractionation (S-MIF) from the sedimentary record has become a widely accepted proxy for atmospheric oxygenation. This framework, however, neglects inheritance derived from oxidative weathering of pre-existing S-MIF-bearing sedimentary sulfide minerals (i.e., crustal memory), which has recently been invoked to explain apparent discrepancies within the sulfur isotope record. Herein, we demonstrate that such a crustal memory effect does not confound the Carletonville S-isotope record; rather, the pronounced $\Delta^{33}\text{S}$ values identified within Rooihoochte Formation represent the youngest known unequivocal oxygen-free photochemical products. Previously observed ^{33}S -enrichments within the succeeding Timeball Hill Formation, however, contrasts with our record, revealing kilometer-scale heterogeneities that highlight significant uncertainties in our understanding of the dynamics of Earth's oxygenation.

MAIN TEXT

The transition from an anoxic and biologically limited planet towards today's familiar oxygenated world is thought to have occurred in a stepwise fashion, proceeding via two pronounced increases in $p\text{O}_2$ that bound the Proterozoic Eon (1-3). Of these, Earth's initial oxygenation—commonly referred to as the Great Oxidation Event (GOE; 4)—represents perhaps the most fundamental transformation experienced by the Earth System, with deep-seated climatic and biogeochemical ramifications (5-8). Despite its undeniable geobiological importance, deciphering the intricacies of planetary oxygenation, and its broader consequences, has historically been hindered by a paucity of direct atmospheric proxies, leaving our understanding largely extrapolated from strictly surficial observations.

The recognition of mass-independent sulfur isotope anomalies (S-MIF; $\Delta^{33}\text{S} \gtrsim |0.4\text{‰}|$, [Equ. 2–3, Methods](#)) within the sedimentary minor sulfur isotope record, however, overcame this deadlock and, until recently (9), was regarded as the only direct means of tracking the evolution of

atmospheric chemistry (10-13). Scrutiny of this now heavily populated record reveals that $\Delta^{34}\text{S}^*$ values carried by the majority of sulfur-bearing phases deposited prior to ~2.3 billion-years-ago (Ga) are conspicuously pronounced (SI Appendix, Fig. S1), featuring $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ ratios (–1.5 to –0.9; 14, 15) that deviate significantly from those that characterize processes active on the contemporary planet[†] (2, 22, 23). In detail, the generation and preservation of S-MIF *approximating that entombed within the geological record* is directly linked to low atmospheric oxygen in multiple ways: Firstly, there are only two *experimentally verified* mechanisms—SO₂ photolysis (10, 12, 24) and SO₂ photo(de)excitation (25-27)—capable of generating the large $\Delta^{33}\text{S}$ and $\Delta^{36}\text{S}$ values archived in the geological record (SI Appendix, Fig. S1). These candidate mechanisms rely on photons that are readily absorbed by oxygen and ozone, in turn, confining S-MIF genesis to atmospheric domains largely devoid of these UV-shielding gases (28, 29). More subtly, reduced atmospheres permit chemically discrete exit channels, presenting vectors that shuttle S-MIF to surficial settings that are otherwise homogenized when atmospheric oxygen exceeds 10^{–5} of the present atmospheric level (PAL; 13). Finally, oxygen-impoverished oceans increase the preservation potential of S-MIF, ensuring the rapid burial of discrete sulfur phases before their isotopic composition is lost via homogenization (30, 31). Assimilating these links with vanishingly low oxygen availability, several authors have proposed that the geological loss of S-MIF can be used as a robust indicator of a rise in atmospheric oxygen above 10^{–5} PAL (13), thereby providing a convenient stratigraphic marker defining the GOE (10, 11, 32-37).

Despite its simple elegance, capturing the geological demise of S-MIF has proven surprisingly difficult and, rather than clarifying matters, emerging data has served to polarize the debate. For instance, after over a decade of searching (33, 36), the demise of S-MIF was finally captured in continuous section within several proximal expressions of the Rooihooogte Formation cored within the South African Transvaal Basin (34; Figures 1–2). Here, the seemingly logical collapse of $\Delta^{34}\text{S}$ values to near-zero, coupled with the transition from Archean-like $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ slopes to their mass-dependent (S-MDF) counterparts, was used to conclude that planetary oxygenation occurred quickly (< 10⁶ years), culminating in an oxygenated atmosphere by ~2.33 Ga (34). This narrative, however, proved short-lived and has since been contested by reports of more-or-less continuous small-magnitude S-MIF extending beyond 2.31 Ga within the Western Australian Turee Creek Group (38, 39). Here, acknowledging the marked diachrony between the available minor sulfur isotope records, the persistence of S-MIF within the supposedly younger Kazput Formation (38, 40-43) was interpreted to reflect inheritance from a crustally adulterated (i.e., S-MIF-bearing) seawater sulfate reservoir sustained by oxidative sulfide weathering in response to a much earlier rise in oxygen, beginning at, or before, 2.45 Ga (38, 39). Taking a contrarian viewpoint, recent identification of S-MIF within the Transvaal Basin in the overlying Timeball Hill and Boshhoek formations has been interpreted to capture primary syndepositional atmospheric S-MIF genesis (8). Here, leveraging sporadic ³³S enrichments ($\Delta^{33}\text{S}$ values up to ~3‰) associated with Archean-like $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ systematics, Poulton et al. envisage a protracted and climate-driven oscillatory

* $\Delta^{34}\text{S} = \Delta^{33}\text{S}$ or $\Delta^{36}\text{S}$ (Methods Summary)

[†]Non-zero $\Delta^{34}\text{S}$ values have been documented in contemporary atmospheric sulfate aerosols and their environmental counterparts housed within polar cryogenic record (16-18). The generation of these S-MIF signals has been ascribed to active stratospheric sulfur photochemistry and, indeed, has been readily applied to fingerprint explosive volcanic episodes whose plumes penetrate the photochemically conducive part of the mid-stratosphere (e.g., 18, 19). When compared with its geological counterpart, the recent cryogenic S-MIF record features more muted $\Delta^{33}\text{S}$ values (<|–4.75|; 18) and different slopes $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ (≈ –4 vs. –1; 16). Combining these quadruple sulfur isotope differences with their sub-decal persistence (18, 19) implies different photochemistries, acting against vastly different chemical backdrops. Cognate with the uncertainties associated with Archean-aged S-MIF, the genesis of S-MIF within the contemporary atmosphere is still a subject of debate (16, 20, 21).

atmospheric evolution, whereby permanent atmospheric oxygenation was delayed by up to 100 million years beyond the deposition of the Rooihooft Formation (8).

Conceptually, while not new, the influence of oxidative sedimentary recycling, via the so-called crustal memory effect (CME; 44, 45), has remained firmly rooted within the theoretical domain owing to the lack of demonstrably coeval datasets necessary to validate its operation. Accordingly, the cratonic dichotomy emerging from the ever-growing Palaeoproterozoic database (8, 34, 36-39, 41), has served to realize some of these theoretically grounded concerns, prompting some workers to question the overall significance of the geological departure of S-MIF (3, 38, 39, 46). Accepting that CME-induced inertia has the potential to mask atmospheric dynamics and their wider relationship with other geobiological/climatic events (3, 38, 45), there is now a pressing need to understand how, if at all, these weathering signals have permeated the minor sulfur isotope record and, ultimately, how we reconcile these conflicting signals to reconstruct the operation of the Earth system. Consequently, to simultaneously explore the existence, longevity and influence of the CME within South Africa's minor sulfur isotope record, and test hypotheses surrounding a delayed and dynamic route to planetary oxygenation, we revisit the Rooihooft–Timeball Hill sedimentary succession deposited in the Carletonville area of the Transvaal Basin (Figures 1–2; SI Appendix). Here, focusing on a proprietary diamond drill-core, KEA-4, we first extended the available bulk quadruple sulfur isotope chemostratigraphy through the majority of the Timeball Hill Formation, leveraging traditional fluorination dual-inlet gas-source isotope ratio mass spectrometry (GS-IRMS, Methods Summary). Pyrites from selected samples were then subjected to *in-situ* isotopic analysis via secondary-ion mass spectrometry (SIMS; Methods Summary) to reveal how the identified S-isotope variability became entrained within the geological record. Contextualizing these new data within available intra- (8) and inter-basinal datasets (37-39) hints toward a complex and dynamic route to planetary oxygenation that we are only beginning to resolve.

RESULTS

STRATIGRAPHIC QUADRUPLE SULFUR ISOTOPE VARIABILITY WITHIN CORE KEA-4. Based on the stratigraphic evolution of $\Delta^{34}\text{S}$ values and related changes in the $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ ratios carried by chemically isolated pyrites within three cores (EBA-2, EBA-4 & KEA-4; Figure 1; Methods Summary), the Carletonville sedimentary succession was subdivided into three distinct chemostratigraphic intervals (34). Subsequent work, albeit generally lacking $\Delta^{36}\text{S}$ data, has since broadly corroborated these observations, extending them to an additional core (EBA-1; Figure 1; 8). Within this framework, the initial quadruple sulfur isotope data from the S-MIF Interval (>563.68 m) of core KEA-4 documented $\Delta^{33}\text{S}$ values as high as 7.4‰ and $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ systematics that populated the Archean Reference Array (ARA; Figure 3; 14, 15, 47). By contrast, pyrites from the S-MDF Interval (<558.56 m) were found to possess near-zero $\Delta^{33}\text{S}$ values ($|\Delta^{33}\text{S}| < 0.3\text{‰}$) and $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ ratios that approximated mass-dependent expectations (~ -7 ; Figure 3; 22). Separating these distinctive sulfur isotope systematics, the so-called Transitional Interval displayed variable, yet muted, $\Delta^{33}\text{S}$ values associated with atypically shallow $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ slopes (34).

Although most of our attention has been directed toward extending KEA-4's bulk quadruple sulfur isotope record into the overlying Timeball Hill Formation, six additional samples were analyzed from the S-MIF Interval within the Rooihooft Formation (Figure 2–4; SI Dataset S1). Here, akin to the previously reported enveloping samples (34), these samples also display a large range of $\Delta^{33}\text{S}$ values (0.77–7.59‰) that, irrespective of their magnitude, conform to the Archean Reference Array (Figure 3; 14, 15, 47). By contrast, the remainder of the newly presented samples ($n = 121$) display a surprising lack of variability, with average (± 1 SD, hereafter 1σ) $\Delta^{33}\text{S}$ and $\Delta^{36}\text{S}$ values of

0.04 ± 0.05‰ and −0.6 ± 0.33‰, respectively. These observations remain in stark contrast to stratigraphically isolated reports of elevated $\Delta^{33}\text{S}$ values (up to ~3‰) within the Timeball Hill Formation captured by Poulton and colleagues' lower resolution, and principally triple sulfur isotope, EBA dataset (8), which is separated from our own by perhaps 5 km (Figure 1; SI Appendix Fig. S2–S4). Here, manifest principally as small positive $\Delta^{33}\text{S}$ values (< 0.3‰), these new data broadly resemble those previously reported from the lowermost few meters of the lower Timeball Hill Formation (8, 34). Within the context of the entire dataset, the Timeball Hill data clearly diverge from those from the Rooihogte-housed S-MIF Interval, clustering around the theoretically forecast S-MDF slope (Figure 3; 22). Interestingly, however, an enlargement around the origin reveals a previously unrecognized spread in data (Figure 3), with some samples plotting toward, or even upon, the ARA (Figure 3B). Lithostratigraphic dissection of this data implies that the incidence of potentially post-Rooihogte-aged mass-independent behavior decreases up section, with its disappearance prior to the deposition of the upper Timeball Hill Formation (Figure 3B, SI Appendix Fig. S2). Again, these observations conflict with the limited Archean-like quadruple sulfur isotope data (n = 5) presented from cores EBA-1 and -2 (8).

The new $\delta^{34}\text{S}$ data span a greater range (−29 to +54‰) than their published counterparts (34), capturing a range equivalent to that seen over Earth's entire history (2). That said, within this dataset, given the general scarcity and stratigraphic isolation of $\delta^{34}\text{S}$ values exceeding +20‰ (n = 6), the samples that define the upper end of this range could be considered outliers that perhaps reflect a high-order variability that is not fully captured by our sampling resolution (Figure 2). Stratigraphic consideration of these $\delta^{34}\text{S}$ data illustrate that the previously documented transition to low $\delta^{34}\text{S}$ values beginning within the Transitional Interval was geologically short-lived and, rather than reflecting a broadscale evolution of $\delta^{34}\text{S}$ values {Luo, 2016 #1025}, captured the falling limb of a now broadly resolved large negative $\delta^{34}\text{S}$ excursion that recovers within the lowermost few meters of the Timeball Hill Formation. Excluding a stratigraphically isolated ^{34}S -depleted sample at 535.93 m and four ^{34}S -enriched samples at 548.23, 515.68, 453.30 and 441.20 m, compared with the remainder of the record, the lower Timeball Hill dataset features a relatively restricted range of $\delta^{34}\text{S}$ values. Here, $\delta^{34}\text{S}$ values hover around 0–10‰ until ~440m, whereafter they appear to define a 50-m-thick positive excursion and recovery, which is terminated by a diabase intrusion beneath the overlying Gatsrand Quartzite Member (Figure 2). Immediately above the Gatsrand Quartzite Member, within the upper Timeball Hill Formation, $\delta^{34}\text{S}$ values descend from around 0‰ to −24‰ over ~50 m before making an oscillatory recovery before, again, being truncated by a diabase intrusion (Figure 2). Thereafter, the $\delta^{34}\text{S}$ values waver between 0 and 20‰ before descending toward negative values in the upper part of our record. Comparison of the $\delta^{34}\text{S}$ data between our record and those of EBA-1 and EBA-2 shows similar trends, with similarly centered bimodal distributions within the Rooihogte and lower Timeball Hill Formations ceding to an approximately unimodal distribution within the upper Timeball Hill Formation. Given this broad approximation in $\delta^{34}\text{S}$ data it is surprising, therefore, that we record vastly discrepant $\Delta^{33}\text{S}$ systematics (SI Appendix Fig. S3).

CORE KEA-4 UNDER THE MICROSCOPE. *In-situ* sulfur isotope analysis typically reveals striking inter- and intra-grain isotopic heterogeneities (30, 35, 38, 48) that are homogenized via conventional fluorination GS-IRMS approaches (34). Consequently, to understand how the atypically large $\Delta^{33}\text{S}$ values seen in the Rooihogte bulk-SF₆ record are distributed at the micro-scale, samples were selected for SIMS analysis from two discrete depths (565.70 & 564.74 m) and a single depth (521.51 m) within the overlying lower Timeball Hill Formation (Figure 2, SI Appendix Fig. S5). Discriminating texturally, sample KEA-4 564.74 m was subdivided for SIMS analysis, with

subsamples A and B featuring visible layered and granular pyrites, respectively (SI Appendix Fig. S5).

Backscattered electron (BSE) images of these SIMS targets show that the pyrites can be loosely described as either euhedral or amorphous agglomerates; however, subtle contrast differences apparent within some of the BSE images hint towards more complex growth/recrystallization histories as should be expected in 2.3-billion-year-old rocks (SI Appendix Dataset S2). Backscattered electron images demonstrate that most pyrite grains, irrespective of their stratigraphic level, feature more inclusion-rich interiors and inclusion-poor exteriors, implying at least a common two-stage growth history throughout the succession (SI Appendix Dataset S2). Indeed, similar textures are apparent in core EBA-2 (8, 34), implying that analogous multi-stage growth histories maybe common across the entire Carletonville area. Importantly, in certain instances, pyrites from core KEA-4 have been observed to disturb primary sedimentary laminae, signaling that these pyrites were precipitated, and subsequently overgrown, prior to compaction (34). Pyrite grain-size variability was found to be independent of the host lithology (i.e., mudstone–fine siltstone) and the grain size of the examined pyrites (10s–100s μm) universally exceeded those of host matrix (34). Given that pyrite is more dense than most common rock-forming minerals, hydraulic equivalence should exclude detrital pyrites within normal sedimentary régimes (34, 49, 50), leaving the pyrite inventory dominated by authigenic pyrites. This inference is reinforced by the observed absence of rounded and/or pitted pyrite grains (or cores) that would be indicative of prolonged transport (48, 50). Interestingly, however, we concede that the brecciated mudstone encountered at 213.7 m within the upper Timeball Hill Formation (Figures 2–4, SI Appendix Fig. S6) presents a curious exception. Here, entrainment of older pyrites within mudstone clasts provides a hypothetical mechanism whereby S-MIF-bearing pyrites could escape oxidative destruction via shielding during transport beneath an oxygenated atmosphere. In this instance, despite returning mass-dependent quadruple sulfur isotope systematics at the bulk-scale ($\delta^{34}\text{S} = 12.45$; $\Delta^{33}\text{S} = 0.04$; $\Delta^{36}\text{S} = 0.34$), it is important to stress that these data are sample-specific, and thus, the dominance of S-MDF shown in a 5-cm thick sample taken at 213.7 m in core KEA-4, may not be representative of this widespread, and often overlooked, facies (51). Equally, as recently argued by Warke et al. (52), lithological control on S-MIF delivery may be more common in glaciogenic or high-energy lithotypes and, as such, may explain some of the large $\Delta^{33}\text{S}$ values seen in Rietfontein Member and its reworked equivalent found in the basal Boshhoek Formation (8; SI Appendix). Ongoing grain-scale sulfur isotope work will explore this possibility.

At a cursory glance, the SIMS dataset broadly echoes the main features identified from the bulk dataset, whereby the large $\Delta^{33}\text{S}$ values that typify the S-MIF Interval are lost in the overlying S-MDF Interval as $\Delta^{33}\text{S}$ values collapse to 0‰ (0.02 ± 0.06 ‰, $n = 18$). Here, bulk-SF₆, texture-specific-SF₆ and average SIMS-derived $\Delta^{33}\text{S}$ values coalesce at 0‰, signaling a representative sampling strategy and, thus, true isotopic homogeneity in this part of the record (Figure 2; SI Appendix Dataset S3). Closer scrutiny of the SIMS data from the S-MIF Interval (>563.68 m), however, reveals a startling heterogeneity, with SIMS-derived $\Delta^{33}\text{S}$ values spanning the full spectrum encompassed by the Carletonville succession (0.34–7.4‰). Importantly, each sample showcases different $\Delta^{33}\text{S}$ ranges and distributions. For example, each of the three samples from the S-MIF Interval feature a population of pyrites with $\Delta^{33}\text{S}$ values around 1–2‰, while only two samples exhibit an additional, less well constrained, population with $\Delta^{33}\text{S}$ values exceeding 6‰ (i.e., KEA-4 565.70 m & KEA-4 564.74 m A). When average GS-IRMS- and SIMS-derived values are compared a varying degree of agreement is observed. For instance, while close agreement between analytical approaches is seen in sample KEA-4 565.70 m, the SIMS datasets from samples KEA-4 564.74 m A and B seem to be

biased toward lower $\Delta^{33}\text{S}$ values. This analytical mismatch implies that the large $\Delta^{33}\text{S}$ values maybe more common and volumetrically more important than our SIMS data portrays; an analytical oversight particularly apparent in sample KEA-4 564.74 m B where values never exceed 4‰ (Figure 2).

Grain-scale $\delta^{34}\text{S}$ analysis reveals large spatial variability throughout (Figure 2). In the S-MDF Interval, the $\delta^{34}\text{S}$ data are roughly centered on the relevant mean and texture-specific $\delta^{34}\text{S}$ values, defining a slightly positively skewed population with an average $\delta^{34}\text{S}$ value of $4.68 \pm 2.17\text{‰}$ ($\pm 1\sigma$). In contrast, while $\delta^{34}\text{S}$ values are generally higher in the S-MIF Interval, again, each sample displays different $\delta^{34}\text{S}$ ranges and distributions. For instance, sample KEA-4 564.74 m A features the most dispersed population, with a $\delta^{34}\text{S}$ range of 3.74–17.74‰ that forms a bi-, or possibly tri-, modal distribution. The remaining samples define tighter populations (i.e., less dispersed), albeit with outliers, with $\delta^{34}\text{S}$ values ranging from 10.85–16.50‰ in sample KEA-4 564.74 m B and 8.02–27.29‰ in sample KEA-4 565.70 m, respectively. Similar discrepancies between the SIMS- and GS-IRMS-derived datasets imply a bias towards higher $\delta^{34}\text{S}$ values, and thus an under-sampling of a pyrite phase carrying lower $\delta^{34}\text{S}$ values within the SIMS-derived dataset.

When the data are considered together in $\Delta^{33}\text{S}$ – $\delta^{34}\text{S}$ three-isotope-space (Figure 4, SI Appendix Fig. S3), the samples from the S-MIF Interval define a steep array with a negative slope. This slope is defined by an endmember (ES1) with elevated $\Delta^{33}\text{S}$ values ($\sim 7\text{‰}$) and less positive $\delta^{34}\text{S}$ values ($\sim 5\text{‰}$) and another (ES2) with muted $\Delta^{33}\text{S}$ values ($\sim 1\text{‰}$) and more positive $\delta^{34}\text{S}$ values ($\sim 17\text{‰}$). While it is tempting to relate these systematics to morphology, in practice this is not easy in core KEA-4, and no unifying model could be applied. Tantalizingly, in sample KEA-4 564.74m B, more euhedral pyrites seem to feature elevated $\delta^{34}\text{S}$ values ($15.1 \pm 0.92\text{‰}$, $n = 33$) and more subdued $\Delta^{33}\text{S}$ values ($0.8 \pm 0.20\text{‰}$, $n = 33$), whereas amorphous pyrites possess lower $\delta^{34}\text{S}$ values ($11.9 \pm 0.92\text{‰}$, $n = 9$) and higher $\Delta^{33}\text{S}$ values ($2.1 \pm 0.74\text{‰}$, $n = 9$). This relationship, however, is not as distinct in the remaining samples and, in fact, may be reversed (e.g., KEA-4 564.74 m A). Given that pyrite genesis is inherently governed by the immediate diagenetic realm, as we go on to discuss, there is no *a priori* reason why pyrite morphology should easily translate to inter-grain sulfur isotope systematics. There is perhaps no better example to illustrate this oversimplification than Grain 9 from sample KEA-4 564.74 m A (Figure 5, SI Appendix Fig. S7; SI Appendix Dataset S2). Here, this amalgamation of three morphologically and chemically similar pyrite grains chronical a broadly analogous range of $\Delta^{33}\text{S}$ variability over 300 μm (Figure 5; SI Appendix Fig. S7) as has been reported from rocks spanning the first two billion years of Earth History (SI Appendix Fig. S1)—Clearly, morphology is *not* the key discriminator here.

To understand how such profound isotopic variability becomes transposed and incorporated into the geological record, the chemistry of Grain 9 and an additional control from the S-MDF Interval (KEA-4 521.51m, Domain #1a, Grain 1, SI Appendix Dataset S2) were mapped by wavelength-dispersive X-ray spectroscopy (WDS; Methods Summary). Besides the core–rim textural differences revealed via SEM-derived BSE imagery (SI Appendix Dataset S2), the resulting WDS-maps unveiled complex and important growth histories that would have otherwise remained opaque (Figure 5, SI Appendix Fig. S7). Focusing on Grain 9 obtained from within the S-MIF Interval, an irregular embayed growth zone enriched in Co and Ni, perhaps reflecting partial dissolution of the earliest generation of pyrite, was found to separate two generations of chemically analogous and otherwise Co- and Ni-impoverished pyrites (Figure 5E–H, SI Appendix Fig. S7). The recognition of this subtle Co- and Ni-rich aureole is of fundamental importance because it acts as an intra-grain stratigraphic marker, providing temporal context for the observed sub-grain sulfur isotope variability (Figure 5E, SI Appendix Fig. S7). In detail, the two SIMS analyses outside of the Co- and

Ni-rich rim returned atypically large $\Delta^{33}\text{S}$ values centered around 7‰ (Domain 4b #3–4, [SI Appendix Dataset S2](#)), while the three analyses appearing within the rim's interior were associated with much lower magnitude $\Delta^{33}\text{S}$ values (Domain 4b #1–2, 6, $\Delta^{33}\text{S} = 0.35\text{--}0.99\text{‰}$; [SI Appendix Dataset S2](#)). A similar spatial separation is mirrored within the $\delta^{34}\text{S}$ data, with larger positive $\delta^{34}\text{S}$ values ($\sim 16\text{--}17\text{‰}$) confined to the inside of the Co- and Ni-rich rim, while lower $\delta^{34}\text{S}$ ($\sim 4\text{‰}$) values are confined to its exterior. The single analysis that appeared to straddle the siderophile-defined rim (Domain 4b #5, [SI Appendix Dataset S2](#)) features intermediate $\Delta^{33}\text{S}$ and $\delta^{34}\text{S}$ values that we attribute to roughly equal mixing between the pre- and post-rim endmembers ([Figure 5E](#), [SI Appendix Fig. S7](#)). Excluding the analysis associated with intermediary sulfur isotope systematics as an analytically-induced mixing artifact (48), we again identify two isotopically-distinct end-members: (ES1) pyrites located within, and therefore pre-dating, the metal-rich rim that possess muted $\Delta^{33}\text{S}$ ($< 1\text{‰}$) and more positive $\delta^{34}\text{S}$ values ($\sim 17\text{‰}$) and (ES2) pyrites post-dating the Co- and Ni-rich rim that feature elevated $\Delta^{33}\text{S}$ ($\sim 7\text{‰}$) values coincident with less positive $\delta^{34}\text{S}$ values ($\sim 4\text{‰}$). Within the wider SIMS dataset from the S-MIF Interval, it is these endmembers that define the steep negative $\Delta^{33}\text{S}\text{--}\delta^{34}\text{S}$ arrays apparent in [Figure 4](#); thus, verifying the extension of these observations to the wider dataset.

DISCUSSION

TWO CO-EXISTING ISOTOPICALLY DISTINCT SULFUR POOLS FINGERPRINT ANOXIC PHOTOCHEMISTRY. Grain 9, from 565.70 meters core depth within the S-MIF Interval, shows unprecedented $\Delta^{33}\text{S}$ variability ([Figure 5](#)), recording a similar spread in $\Delta^{33}\text{S}$ values as encountered over the entirety of Earth History ([SI Appendix Fig. S1](#)). Surprisingly, after mapping Grain 9, we noticed that analyses returning elevated $\Delta^{33}\text{S}$ values and more subdued $\delta^{34}\text{S}$ values (ES1) were separated from those with apposing sulfur isotope systematics (ES2) by a subtle embayed dissolution/growth zone enriched in Co and Ni ([Figures 4–5](#), [SI Appendix Fig. S7](#)). This observation implies that the elevated $\Delta^{33}\text{S}$ values were entombed within the pyrite record relatively late compared to the more inclusion-rich core.

Given that S-MIF behaves as a conservative tracer within Earth's exogenic sulfur cycle (i.e., it cannot be created nor destroyed, only diluted), the restriction of large magnitude $\Delta^{33}\text{S}$ values towards the periphery of Grain 9 can be hypothetically rationalized by post-depositional ingress of photolytic sulfur remobilized from juxtaposed Archean-aged sediments. That said, it is imperative to note that fewer than 2% of minor sulfur isotope measurements made on appropriately aged materials return $\Delta^{33}\text{S}$ values that exceed those seen in the Rooihogte Formation (i.e., 7.6‰ ; [Figure 2](#), [SI Appendix Fig. S1–S4](#), [S7–S9](#)) and, those that do, are separated from the Rooihogte Formation by hundreds of kilometers and some 200-million-years (30, 47). Consequently, invoking late-stage sulfur ingress not only requires a mechanism capable of selectively remobilizing sulfur sourced from the most extreme positive-tail of the geological $\Delta^{33}\text{S}$ record ([SI Appendix Fig. S1](#)) but, more locally, commands an unidentified source of sulfur with $\Delta^{33}\text{S}$ values of sufficient size to dominate the bulk data obtained from the S-MIF interval ([Figure 2](#)). Furthermore, even if we disregard these improbable corollaries, we consider an explanation reliant on late-stage sulfur ingress incongruous with the stratigraphic $\delta^{34}\text{S}$, $\Delta^{33}\text{S}$ and $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ heterogeneity captured within the wider Carletonville minor sulfur isotope record ([Figures 2–3](#); 8, 34), which would presumably be erased via wholesale late-stage sulfur addition. This viewpoint is reinforced at the grain-scale, where SIMS analyses from the Rooihogte Formation capture a remarkable range of $\Delta^{33}\text{S}$ values whose distributions appear to be highly sample specific yet disappear as the bulk $\Delta^{33}\text{S}$ values collapse to zero within the overlying Timeball Hill Formation ([Figure 2](#)). Finally, besides the subtle intra-grain siderophile-rich aureole, the trace element (As, Ni, Co) systematics of the pre- and post-rim pyrites in Grain 9 are remarkably similar. This observation is in strong conflict with the injection of an

externally sourced fluid, which, reasonably, would be expected to be chemically distinct, culminating in the growth of chemically distant pyrites and metal monosulfides. Indeed, the infrequent detection of acid-volatile sulfide throughout the examined succession and, where measurable, its mass-dependent sulfur isotope systematics, provide strong empirical evidence against hydrothermal sulfur remobilization and retention (Methods Summary; Dataset S1). Moreover, the lack of any strong correlation between $\Delta^{33}\text{S}$ (or $\delta^{34}\text{S}$) and EPMA-derived trace-metal concentrations (SI Appendix Fig. S8; SI Dataset S4) bolster this stance and, when combined with the preceding arguments, effectively preclude post-depositional sulfur ingress as a viable explanation of the observed spatial sulfur isotope heterogeneity.

Eliminating post-depositional explanations, a syn-depositional model must be sought to explain the spatial $\Delta^{33}\text{S}$ variability observed within Grain 9 (Figure 5E) and, via extrapolation, the entire SIMS dataset (Figure 4). Here, drawing conceptually from the ideas developed by Farquhar et al. (30) we argue for the co-existence of two isotopically distinct diagenetically active sulfur pools. Although alternate views exist (31, 53-56), against an oxygen-free backdrop, most workers envisage the operation of two Archean atmospheric exit channels, supplying elemental sulfur (S_8) and sulfate (SO_4^{2-}) to the Earth's surficial sulfur cycle with positive and negative $\Delta^{33}\text{S}$ values, respectively[†] (12, 13, 47, 57). Indeed, it is the relative proportions of each sulfur source that is thought to underpin the $\Delta^{33}\text{S}$ – $\delta^{34}\text{S}$ array that typifies Neoproterozoic-aged sulfides ($\Delta^{33}\text{S} \approx 0.9 \times \delta^{34}\text{S}$; (47). Consequently, the divergence from these purely photochemical expectations toward a steep linear relationship constrained by ES1 ($\uparrow\Delta^{33}\text{S}$ & $\downarrow\delta^{34}\text{S}$) and ES2 ($\downarrow\Delta^{33}\text{S}$ & $\uparrow\delta^{34}\text{S}$; Figure 4) within our dataset is curious and presents the opportunity to track pyrite formation 2.3 billion years ago.

The extreme isotopic heterogeneity juxtaposed at the μm -scale (Figure 5E) requires that the sulfur sources for ES1- and ES2-type pyrites were unable to undergo full isotopic exchange and managed to evade homogenization during early diagenesis. Pyrite formation is a complex process that can be conceptually reduced to two steps: namely, the conversion of aqueous sulfide to metastable iron monosulfides (FeS) and its conversion to pyrite (58, 59). While the latter involves numerous complex pathways, broadly speaking, the process is simplified within the literature to the H_2S and the polysulfide pathways (59, 60). Against an oxygen-free backdrop, polysulfides would have perhaps formed via the reaction between MSR-derived sulfide and atmospherically sourced S_8 . Accordingly, the $\Delta^{33}\text{S}$ value of pyrite produced via the polysulfide pathway should reflect the relative proportion of sulfate-derived sulfur to S_8 -derived sulfur (30). Negative $\Delta^{33}\text{S}$ values are conspicuously absent from our dataset (Figure 4), which would be typically invoked to fingerprint derivation from a photolytic sulfate source (23, 28, 30). Nonetheless, given the expectation that sulfate should be readily available within the porewaters during early diagenesis, we hypothesize that the initially formed ES2-type pyrites acquired the bulk of their sulfur via microbial sulfate reduction (MSR) of porewater-sulfate, with a potentially small contribution from elemental sulfur resulting in the intermediate $\Delta^{33}\text{S}$ values we observe. By contrast, we imagine the later-stage pyrites, featuring atypically large $\Delta^{33}\text{S}$ values, derived their sulfur principally from a photolytic S_8 source as MSR-derived sulfide became scarcer; a notion supported by the relatively restricted range of $\delta^{34}\text{S}$ values displayed by ES2 pyrites housed within relatively TOC- and Fe-replete rocks (8,

[†] While SO_2 photolysis is the favoured S-MIF source reaction of the majority (e.g., 14, 28, 47), others argue that sulfur recombination reactions are the origin of S-MIF (53-55). Besides difficulties in reproducing the $\Delta^{33}\text{S}$ magnitude and $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ systematics preserved in the geological record that are common to models reliant on SO_2 photolysis (14), theory predicts that sulfur recombination reactions require a sign reversal, whereby the sulfate and sulfide endmembers carry positive and negative $\Delta^{33}\text{S}$ values, respectively (54). Mechanistic uncertainties notwithstanding, we attest that (i) the inter-grain-scale distribution of $\Delta^{33}\text{S}$ values observed in Figure 5 and (ii) the larger spread of $\delta^{34}\text{S}$ values in association with low $\Delta^{33}\text{S}$ values, both within our dataset (Figure 4) and the wider geological record, implicate sulfate as the low- $\Delta^{33}\text{S}$ endmember and, thus, favor the canonical sign arrangement.

46) that imply the operation of MSR under generally sulfate-limited conditions (Figure 4). Combining our observations with similar rationalizations derived from the 2.7-billion-year-old Joy Lake sequence (61) and the ~2.65–2.5-billion-year-old Campbellrand–Malmani carbonate platform (30) provides evidence for the existence for multiple diagenetically active and isotopically distinct sulfur pools outside of the Neoproterozoic.

SEARCHING FOR THE CRUSTAL MEMORY EFFECT (CME). The prevalence of muted $\Delta^{33}\text{S}$ values within the Paleoproterozoic minor sulfur isotope record (35–39, 48, 62) has been proposed to be a consequence of the oxidative weathering of continentally-hosted S-MIF-bearing sulfides fueled, in turn, by a rise in atmospheric oxygen in proximity to the Archean–Proterozoic boundary (30–33). If true, such a memory effect serves to decouple the sedimentary S-MIF record from contemporaneous atmospheric chemistry (3, 38, 44, 45). Perhaps unjustly, considering the magnitude of the $\Delta^{33}\text{S}$ values (Figures 2–3), the atmospheric significance of the observations made by Luo et al. (34) were questioned via the recognition of sub-2‰ $\Delta^{33}\text{S}$ values with Archean-like $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ ratios preserved within the Australian Kazput Formation, purportedly deposited well after 2.3 Ga (38, 39). This apparent diachronous disappearance of S-MIF was used to argue for regional-scale expressions of the CME, resulting from a rise in atmospheric oxygen before 2.45 Ga (38, 39). While we remain unconvinced that the clearly facies-controlled ephemeral decreases in $\delta^{34}\text{S}$ and $\Delta^{33}\text{S}$ values seen within the siltstones of the upper Boolgeeda Iron Formation record an important oxidation event (38), the existence of the CME outside of Western Australia remains largely a theoretical construct (45). As such, we look to our expanded bulk and grain-scale minor sulfur isotope records for evidence of such an effect within the Transvaal Basin (Figures 2–5).

The complete disassociation of the Carletonville minor sulfur isotope record from atmospheric chemistry via the CME, is contested by recollection of the computational weathering experiments conducted by Reinhard and colleagues (45). Here, by applying geologically reasonable boundary conditions, Reinhard et al. placed an extreme upper limit of ~3‰ on the weathering-derived $\Delta^{33}\text{S}$ signal that could be conveyed to a hypothetical global seawater-sulfate inventory (SI Appendix Fig. S1). Given that the S-MIF interval in the Carletonville area features $\Delta^{33}\text{S}$ values encroaching on 8‰ (Figure 2; 34), we can effectively preclude crustal inheritance via a crustally adulterated seawater-sulfate reservoir as the primary control on the bulk isotope systematics. In fact, our grain-scale fingerprinting of the S_8 photolytic endmember strongly argues for more direct photochemical modulation (34). Candidly, there is no known way of divorcing the high magnitude $\Delta^{33}\text{S}$ values seen at the bulk-, and now at the grain-, scale within the Rooihoogte Formation from active photochemistry (Figures 2–3). Accordingly, we *stress* that these data demand an oxygen-free atmosphere and should be considered a robust marker of such.

Moving forward, understanding how the Rooihoogte minor sulfur isotope record is dominated by photolytic sulfur (8, 34) while other successions have seemingly been compromised remains crucial to decoding the meaning and utility of the sedimentary S-MIF record. Forgoing the large magnitude $\Delta^{33}\text{S}$ values seen in the S-MIF interval (Figures 2–5), the CME remains an unexplored concern, obstructing the interpretation of the wider Paleoproterozoic minor sulfur isotope record (35, 36, 38, 52). Immediately above the proposed Transitional Interval, $\Delta^{33}\text{S}$ values collapse to near-zero coincident with a transition toward steeper $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ slopes (34). Concerned by the restricted stratigraphic range of the quadruple sulfur isotope data presented by Luo et al. (34), and the presence of large stratigraphic gaps separating the discrete cores analyzed by Philippot et al. (38), we consider it extremely unlikely that the original records (34, 38, 39) would have displayed much overlap. Consequently, to make a fair comparison and search for the CME above the loss of S-MIF in the Carletonville area, we extended the bulk-SF₆ record into the upper Timeball Hill Formation

(SI Appendix Fig. S2–S4). Despite these efforts, however, we failed to capture any $\Delta^{33}\text{S}$ values outside of the conservative threshold of $|\text{0.3‰}|$ that has been used to discriminate between mass-independent and mass-dependent sulfur isotope fractionation (8; Figures 2–3, SI Appendix Fig. S2). Grain-scale examination of a sample from 521.51 m is equally monotonous, with SIMS data defining a tight population of $\Delta^{33}\text{S}$ values centered at 0‰ (Figures 2, 5). Constrained by U–Pb ages of two closely spaced micro-tuffs from the upper Timeball Hill Formation (63), these new data encompass upwards of 50 million years of sedimentation, extending the Carletonville quadruple sulfur isotope record to ~ 2.26 Ga (63). Within this extended framework, the observed isotopic monotony demonstrates that, at least in core KEA-4, there is no strong evidence for the reappearance of S-MIF within the resolution of our dataset (Figures 2–5; SI Appendix Fig. S2–S4). The apparent predominance of largely mass-dependent sulfur isotope behavior within core KEA-4 after 2.3 Ga (34) remains in stark contrast to the sulfur isotope systematics of the supposedly crustally-influenced Australian Turee Creek succession (38, 39, 48) and, indeed, the computational forecasts of 10–100 million-year survival times of S-MIF within the Earth’s exogenic sulfur cycle from which these claims were grounded (45).

At the most simple level, the effectively instantaneous loss of S-MIF in the Carletonville area of the Transvaal Basin (Figure 2) can be reconciled with its persistence within the Turee Creek Group by invoking either (i) basin-controlled heterogeneities in the $\Delta^{33}\text{S}$ values carried by seawater-sulfate (38, 39, 42) or (ii) that the ages linking the two records are misinterpreted (41, 43). The Turee Creek Group is linked to the studied succession by a Re–Os isochron age obtained from glaciogenic Meteorite Bore Member (38) and supporting U–Pb detrital zircon ages obtained from its underlying formational equivalent (40). Rejecting the depositional significance of these ages, Bekker et al. (41, 43) recently presented a series of contested stratigraphic arguments (42, 43), concluding that the Turee Creek succession was much older (i.e., 2.45–2.42 Ga) than previously envisaged (i.e., 2.45–2.2 Ga) and hence predates the examined succession (Figure 2). While (41)reinterpretation is more convenient, resulting in quadruple sulfur isotope systematics that are more reminiscent of those reported from older rocks (SI Appendix Fig. S9), it is pertinent to recall the dichotomy we identify within proximal and indisputably age-equivalent records from within the same basin (SI Appendix Fig. S2–S3). While Archean-aged catchments apparently fail to yield non-zero $\Delta^{33}\text{S}$ values under contemporary boundary conditions (64), we know that modern catchments do, in fact, yield sulfur isotope heterogeneities (65). Consequently, it remains plausible that these heterogeneities might have been communicated to the presumably much smaller early Paleoproterozoic seawater-sulfate reservoir (38, 39), culminating in basinal heterogeneities that were perhaps amplified in the Turee Creek Basin. Clearly further geochronological constraints are needed to discriminate between these competing explanations of the (38, 39, 41–43); for now, however, we caution against the use of sulfur isotope chemostratigraphy to either construct or test hypothesized correlation schemes.

Returning to the Transvaal (Figure 2), we see no evidence to support a long-lived and/or pronounced CME above the loss of S-MIF in the Carletonville quadruple sulfur isotope record. That said, within our record, the samples plotting between the expected S-MIF and S-MDF $\Delta^{36}\text{S}/\Delta^{33}\text{S}$ arrays (SI Appendix Fig. S4) may record subtle and short-lived expressions of crustal recycling. Furthermore, the muted $\Delta^{33}\text{S}$ values captured by ES2-type pyrites within our SIMS dataset may signal the influence of the CME beneath an anoxic atmosphere capable of S-MIF export. While full exploration of these hypotheses awaits detailed isotopic and elemental grain-scale analyses, the notion of prolonged low-level sulfide oxidation is supported by recent $p\text{O}_2$ estimates derived from the Archean–Paleoproterozoic increase in shale-hosted molybdenum abundances and $\delta^{98/95}\text{Mo}$ values (66).

A DYNAMIC ROUTE TO ATMOSPHERIC OXYGENATION? Concerning the timing and trajectory of planetary oxygenation, and its role within the evolution of the climate system, contrasting opinions abound (5, 34, 36, 38, 39, 41). Initially the term Great Oxidation Event was coined to describe the broad evolution from a reducing atmospheric state, comprised principally of N_2 and CO_2 , to a more oxidizing equivalent, where O_2 rose above ultra-trace levels, occurring on hundred-million-year timescales (4). Subsequently, however, perhaps incited by its unintended linguistic connotations, the GOE has been publicized by some as a singular event (33, 34, 36), while others perceive a multi-stage or dynamic transition, occurring on a range of timescales (5, 8).

Alongside the blueprints of planetary oxygenation, early Paleoproterozoic sedimentary successions contain evidence for up to four glacial episodes between ~ 2.45 and 2.22 Ga (5, 7, 11). Recording the oldest and perhaps best known of these, the Makganyene Formation, housed within Griqualand West Basin, comprises a series of tropical glaciomarine diamictites that are interpreted to reflect the oldest known Snowball Earth event (7). Recent U–Pb dating of cross-cutting mafic volcanics genetically related to the Ongeluk Formation constrains this glacial episode to before 2.43 Ga, making it potentially correlative with glacial episodes preserved globally (5). Indeed, in Fennoscandia, Warke et al. (37) documented a loss of S-MIF prior to $2,434 \pm 6.6$ Ma, reporting negative $\Delta^{33}S$ values with $\Delta^{36}S/\Delta^{33}S$ systematics that statistically resemble the ARA within the Seidorechka Sedimentary Formation, that surrender to near-zero $\Delta^{33}S$ values that define a steep S-MDF-like $\Delta^{36}S/\Delta^{33}S$ slope in the overlying volcanically influenced Polisarka Sedimentary Formation. Placing this Fennoscandian shift in quadruple sulfur isotope systematics within the relevant chronostratigraphic frameworks, these authors present a case for one broadly synchronous S-MIF to S-MDF transition, arguing that the GOE preceded a Paleoproterozoic Snowball Earth (37). Although this is geochronologically permissible (37), numerous stratigraphic arguments invalidate the implied correlation between the Fennoscandian loss of S-MIF and its supposed equivalent within the Rooihogte Formation (Figures 2–5, S7; 5, 34, 41, 47, 67). Consequently, excluding an unrecognized volcanically derived overprint, these Fennoscandian data combined with those presented herein demand the diachronous loss of S-MIF. Put differently, these records require at least two discrete intervals whereby S-MIF production was terminated as atmospheric oxygen rose above 10^{-5} PAL (13). In turn, these observations imply a dynamic route to planetary oxygenation (5, 8), precluding conceptual models reliant on a unidirectional evolutionary driver of planetary oxygenation (i.e., the emergence of oxygenic photosynthesis; 68). Rather, the apparent oscillatory nature of atmospheric oxygen during the early Paleoproterozoic is better reconciled with an Earth system vulnerable to tipping points encountered through biological and/or tectonic feedbacks (5, 8).

Compiling available redox indicators from the Kaapvaal-housed Griqualand West and Transvaal basins, Gumsley et al. (5) envisaged at least three oscillations in atmospheric pO_2 (SI Appendix Fig. S9). In their model, the authors foretold a short-lived rise in pO_2 in the wake of the ~ 2.43 Ga Makganyene-aged Snowball Earth, which, following from the previous discussion, has since been substantiated by the apparent Fennoscandian loss of S-MIF in the prelude to the Paleoproterozoic's first glacial episode (37). The second and third oxygenation episodes pictured by Gumsley et al. (5) are constrained by the loss of relatively high magnitude $\Delta^{33}S$ values and the shift from $\Delta^{36}S/\Delta^{33}S$ slopes that populate the ARA toward those that conform to theoretical MDF predictions within the lower portions of the Duitschland and Rooihogte formations, respectively (33, 34, 36). While the paucity of high-precision age constraints currently precludes unequivocal distinction between the conflicting stratigraphic arguments (41, 51, 52, 67) that underpin the separation of these hypothesized oxygenation episodes, in their absence, we can confidently assert that the

Rooihogte-hosted loss of S-MIF (Figure 2, SI Appendix Fig. S2–3) represents the youngest texturally constrained evidence for an oxygen-free atmosphere some 2.3-billion-years-ago (69). Moreover, its separation from the apparent Fennoscandian loss of S-MIF by perhaps 100-million-years supports a dynamic path toward a fully oxygenated atmosphere punctuated by rises and falls in atmospheric pO_2 above and below the 10^{-5} PAL (13) threshold necessary to attenuate atmospheric S-MIF export (5, 8).

More recently, Poulton and colleagues (8) have reported stratigraphically isolated $\Delta^{33}S$ values encroaching on 3‰ to argue for repeated returns to reduced atmospheric states capable of S-MIF production beyond 2.3 Ga. Within our quadruple sulfur isotope dataset, however, we fail to capture this variability in a core separated by less than 5 km (Figure 1–5, S2–S4). This kilometer-scale discrepancy has two potential provisional explanations: (i) The dichotomy between the records speaks to spatially variable S-MIF delivery within the Transvaal Basin; (ii) or extremely rapid atmospheric shifts that are not captured within our higher resolution record. Discrimination between these two competing explanations requires targeted grain-scale sulfur isotope analyses to texturally fingerprint the vector, and therefore significance, of these clearly rare ^{33}S -enrichments seen beyond 2.3 Ga. For now, however, these imagined younger oscillations in atmospheric oxygen (8) appear to have operated at a higher temporal frequency than those envisaged by Gumsley et al. (5), implicating different drivers that are difficult to reconcile with rapid and repeated relapses within the global climate system. Mechanistic uncertainties aside, the apparent oscillatory nature of S-MIF within the geological minor sulfur isotope record suggests that the disappearance of S-MIF would *not* make a definitive chemostratigraphic marker, nor can it be readily employed to define the GOE itself (38), which, upon appropriate investigation, maybe subdivisible into a series of oxygenation episodes.

CONCLUSIONS AND OUTLOOK. In search of supporting evidence to substantiate claims of post-2.3 Ga oxygen dynamics, we produced the first high-resolution quadruple sulfur isotope record extending up into the upper Timeball Hill Formation yet failed to capture $\Delta^{33}S$ values that exceed 0.3‰. Contrastingly, within the S-MIF Interval of the Rooihogte Formation, SIMS analyses disclose pronounced inter- and intra-sample $\Delta^{33}S$ variability, featuring some of the most prominent $\Delta^{33}S$ values ever reported from the geological record. The preservation and differential stratigraphic distribution of these elevated $\Delta^{33}S$ values effectively precludes post-depositional overprinting while implicating complex incorporation of isotopically distinct sulfur pools into pyrite. Importantly, given the preponderance of small magnitude $\Delta^{33}S$ values within the geological minor sulfur isotope record, the larger magnitude $\Delta^{33}S$ values observed in the Rooihogte Formation cannot be reconciled with inheritance from crustally-adulterated seawater (45) and, instead, are most parsimoniously explained as primary atmospheric derivatives. The presence of large magnitude $\Delta^{33}S$ values in association with Archean-like $\Delta^{36}S$ – $\Delta^{33}S$ slopes, and their subsequent disappearance within the Rooihogte Formation, provides direct evidence for the accumulation of atmospheric oxygen above 10^{-5} PAL (13) by ~ 2.3 Ga (69). Persistently low $\Delta^{33}S$ values coupled with generally mass-dependent $\Delta^{36}S/\Delta^{33}S$ systematics within the Timeball Hill Formation, deemphasize the wider significance of the crustal memory effect within the Transvaal Basin as it was originally proposed (45), although a more muted influence may have been identified. Chronological and stratigraphic arguments (5, 41, 52, 67) separate the S-MIF to S-MDF transition documented herein from a supposed equivalent in the Imandra–Varzuga Greenstone Belt, northwest Russia (37), by in excess of 100 million-years. This apparent repeated loss of S-MIF at ~ 2.4 and ~ 2.3 Ga lends credence to models that invoke feedback-driven dynamics in the prelude to irreversible atmospheric

oxygenation (5). Discrepancies between the Transvaal-housed minor sulfur isotope records separated by a few kilometers, however, highlights difficulties in reconstructing the narrative of Earth's oxygenation and, at present, offers little concrete support for inferred oxygen dynamics beyond 2.3 Ga.

MATERIALS AND METHODS

ISOTOPIC NOMENCLATURE: Following convention (2, 23), both bulk and grain-scale sulfur isotope data are reported in delta (δ) notation, reflecting permil (‰) deviations of the least abundant isotope ($^{33,34,36}\text{S}$) normalized to its most abundant counterpart (^{32}S), relative to the same ratio in the international reference standard, Vienna Canyon Diablo Troilite, V-CDT):

$$\delta^X\text{S} = [({}^X\text{S}/{}^{32}\text{S})_{\text{sample}}/({}^X\text{S}/{}^{32}\text{S})_{\text{V-CDT}} - 1] * 1000 \text{ [Equ. 1]}$$

Most processes fractionate S-isotopes mass-dependently, whereby $\delta^{33}\text{S} \approx 0.515 * \delta^{34}\text{S}$ and $\delta^{36}\text{S} \approx 1.91 * \delta^{34}\text{S}$. Departure from this mass-dependent behavior, termed mass-independent fractionation (MIF), is expressed in capital-delta (Δ) notation, where

$$\Delta^{33}\text{S} = 1000 * [\ln(\delta^{33}\text{S}/1000 + 1) - 0.515 * \ln(\delta^{34}\text{S}/1000 + 1)] \text{ [Equ. 2]},$$

and

$$\Delta^{36}\text{S} = 1000 * [\ln(\delta^{36}\text{S}/1000 + 1) - 1.91 * \ln(\delta^{34}\text{S}/1000 + 1)] \text{ [Equ. 3]}.$$

BULK SULFUR ISOLATION AND ISOTOPE ANALYSIS: The bulk quadruple sulfur isotope data reported herein are the product of two analytical campaigns separated by several years. Akin to the approach outlined in Luo et al. (34), the initial analytical campaign (2014–2015) removed the carbonate-associated sulfate and acid-volatile sulfide via a 6 M HCl pretreatment. The resultant carbonate-free residues were then dried below 60°C, allowing the extraction of pyrite-housed sulfur via a 2-hour reflux with acidified chromous chloride (CrCl_2) (70). Here, as is convention, the evolved H_2S was directed to a Zn-acetate trap within a N_2 stream where it was captured via precipitation as zinc sulfide (ZnS). Subsequent dropwise addition of silver nitrate (AgNO_3) then displaced the Zn, culminating in a silver sulfide precipitate (Ag_2S), which was recovered, cleaned and reserved for mass spectrometry. More recently (Aug–Sep 2021), although the same apparatus and reagents were employed, we opted to pursue the two-step sequential reductive approach used by Poulton et al. (8). Here, bulk powders were subject to an initial two-hour reflux with ethanoic 6M HCl prior to exchanging the Zn-acetate traps and a second reflux with acidified 2M CrCl_2 . Finally, as above, AgNO_3 was used to convert the resultant ZnS into Ag_2S . While this methodological modification is more time consuming, it is advantageous because it allowed the isolation of acid-volatile sulfide (AVS) prior to the capture of its chromium-reducible counterpart (CRS), thereby offering a convenient chemical screen for the presence of disseminated hydrothermally derived exotic metal monosulfides (8).

The purified Ag_2S was reacted with a 10-fold excess of fluorine gas (F_2) at 300 °C for in excess of 6 hours in Monel® reaction vessels. The reaction product, sulfur hexafluoride (SF_6), was cryogenically separated from residual F_2 and from HF and other trace contaminants at liquid nitrogen temperatures. Finally, the SF_6 was purified via gas chromatography using a composite column composed of 5Å molecular sieve (3.18-mm diameter, 1.8 m long), followed by a Hayesp-Q column (3.18-mm diameter, 3.6 m long). Under these conditions SF_6 was eluted after ~10 min using an He carrier gas (20 mL·min⁻¹ at 50 °C). The S-isotope composition of the pure SF_6 was measured by dual-inlet gas-source isotope ratio mass spectrometry using a Thermo-Finnigan MAT 253 equipped with four collectors arranged to measure the intensity of SF_5^+ ion beams at mass/charge ratios (m/z) of 127, 128, 129, and 131 ($^{32}\text{SF}_5^+$, $^{33}\text{SF}_5^+$, $^{34}\text{SF}_5^+$, and $^{36}\text{SF}_5^+$). Analytical precision was estimated

from the long-term reproducibility of Ag_2S fluorinations ($n = 28$), yielding 2σ uncertainties of 0.26, 0.014, and 0.19‰ for $\delta^{34}\text{S}$, $\Delta^{33}\text{S}$, and $\Delta^{36}\text{S}$, respectively (34). Again, while more time intensive, combination with monoisotopic elemental fluorine and measurement as sulfur pentafluoride cations removes the need for isobaric corrections that either hinder (^{33}S), or completely prevent (^{36}S), precise minor sulfur isotope determination via conventional combustion approaches (i.e., as SO_2).

SUB-GRAIN-SCALE S-ISOTOPE ANALYSIS: Sample selection for micro-analysis was based on the existing SF_6 -derived sulfur isotope chemostratigraphy presented by Luo et al. (34). In hand-specimen, each candidate sample contained macroscopic pyrite grains, layers or nodules. Initial petrographic analysis was conducted at the State Key Laboratory of Biogeology and Environmental Geology (BGEG) at China University of Geosciences (CUG), Wuhan, where pyrite morphology, texture and grain-size were determined using both reflected and transmitted light. Based on these observations SIMS targets were selected, prepared and carbon coated for observation via scanning electron microscopy using a Hitachi SU8000 coupled with an energy-dispersive X-ray spectrometer. Both secondary electron and back-scattered electron modes were used to investigate the textural association of the pyrite grains.

In-situ triple sulfur isotope measurements were conducted using the CAMECA IMS 1280 housed at the Wisconsin Secondary Ion Mass Spectrometry Laboratory (WiscSIMS), University of Wisconsin–Madison, USA. Detailed descriptions of the instrument and the employed measurement protocol are provided elsewhere (35, 48, 71–73) and, thus, are only summarized here. To monitor instrumental behavior and normalize the data, an in-house standard, UWPY-1 ($\delta^{34}\text{S} = 16.04 \pm 0.18\text{‰}$, $\Delta^{33}\text{S} = -0.003 \pm 0.009\text{‰}$ (73), was imbedded within each mount and polished with $0.25\text{ }\mu\text{m}$ Al_2O_3 powder. Mounts were coated with 60 nm of gold prior to analysis. Combining a normal-incidence electron flood gun, for charge compensation, with a $^{133}\text{Cs}^+$ beam ($\sim 5\text{ nA}$, $15\text{ }\mu\text{m}$ spot size) accelerated at 10 kV (impact energy = 20 kV), allowed simultaneous detection of secondary $^{32}\text{S}^-$, $^{33}\text{S}^-$ and $^{34}\text{S}^-$ ions via designated Faraday cup detectors with modified slits (73). Each measurement consisted of 10s of pre-sputtering, 80s for peak centering and alignment, and 80s of measurement time. Count rates of $\sim 5 \times 10^9\text{ s}^{-1}$ for $^{32}\text{S}^-$ beams typified pyrite analyses and confirmed the stability of individual analyses. The significance of hydride interferences ($^{32}\text{SH}^-$) on the $^{33}\text{S}^-$ ion beam, were determined by beam deflection upon culmination of each individual analysis (48). Here, the ratio of the $^{32}\text{SH}^-$ tail at the $^{33}\text{S}^-$ peak position relative to the $^{32}\text{SH}^-$ peak ($^{32}\text{SH}^-_{\text{tail}}/^{32}\text{SH}^-_{\text{peak}}$) can be used to remove the $^{32}\text{SH}^-$ contribution to the $^{33}\text{S}^-$ peak. In this instance, however, the $^{32}\text{SH}^-$ contribution to the $^{33}\text{S}^-$ peaks was negligible, resulting in irresolvable corrections ($0.1\text{--}1 \times 10^{-3}\text{‰}$). Measurement routines began with four analyses of UWPY-1, which were repeated after every 7–22 sample analyses. Replicate analyses of UWPY-1 were used for calibration, yielding an average 2σ reproducibility of 0.08‰ and 0.31‰ for $\Delta^{33}\text{S}$ and $\delta^{34}\text{S}$, respectively.

IN-SITU ELEMENTAL ANALYSIS: The *in-situ* elemental composition of pyrites was determined using a JEOL JXA-8100 Electron Probe Micro Analyzer equipped with four wavelength dispersive spectrometers (WDS) housed within the State Key Laboratory of Geological Processes and Mineral Resources, CUG, Wuhan. Here, prior to analysis, samples were coated with a conductive $\sim 20\text{ nm}$ carbon film, following the precautions outlined by Zhang and Yang (74). Spot analyses were conducted in proximity to the SIMS pits, using an accelerating voltage of 20 kV, a beam current of 20 nA and a $1\text{ }\mu\text{m}$ spot size. Ten-second peak counting times were used for As, S, Fe, Co, Ni and Cr quantification, while 5 s background counting times were employed at both the high and low energy background positions. Data were corrected online using a ZAF (atomic number, absorption, fluorescence) correction procedure (75). For spatial maps, an accelerating voltage of 20 kV, a beam

current of 100 nA, and a 1 μm spot size with a 25 ms dwell time were used. Elemental abundances were standardized against gallium arsenide (As), pyrite (S, Fe), chromium (Cr), cobalt (Co) and nickel (Ni). The respective 1σ precision associated with major (i.e., Fe & S) and trace (i.e., As, Cr, Co and Ni) metal(oid) determinations was found to be better than 0.4% and 10% when elemental concentrations exceeded 1000 ppm.

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FIGURES:



Figure 1: Geographic and geologic maps locating core KEA-4 and the EBA (-1, -2 and -4) cores within the Carletonville area of the Transvaal Basin, South Africa. The geologic map providing the broad spatial context has been simplified from Coetzee et al. (51), while the precise locations of core KEA-4 (34) and those obtained within the eastern boundary area (EBA-1, -2 & -4; 8, 34) of the Kloof Goldfields mining district (grey shading) are taken from Sibanye Stillwater's open access documentation. The redline within the lower right panel signifies the less than five kilometers of separation between core KEA-4 and EBA-1. The EBA cores themselves are conservatively separated by less than 500 m. Les., Es., and Mozamb. abbreviate Lesotho, Eswatini and Mozambique, respectively.

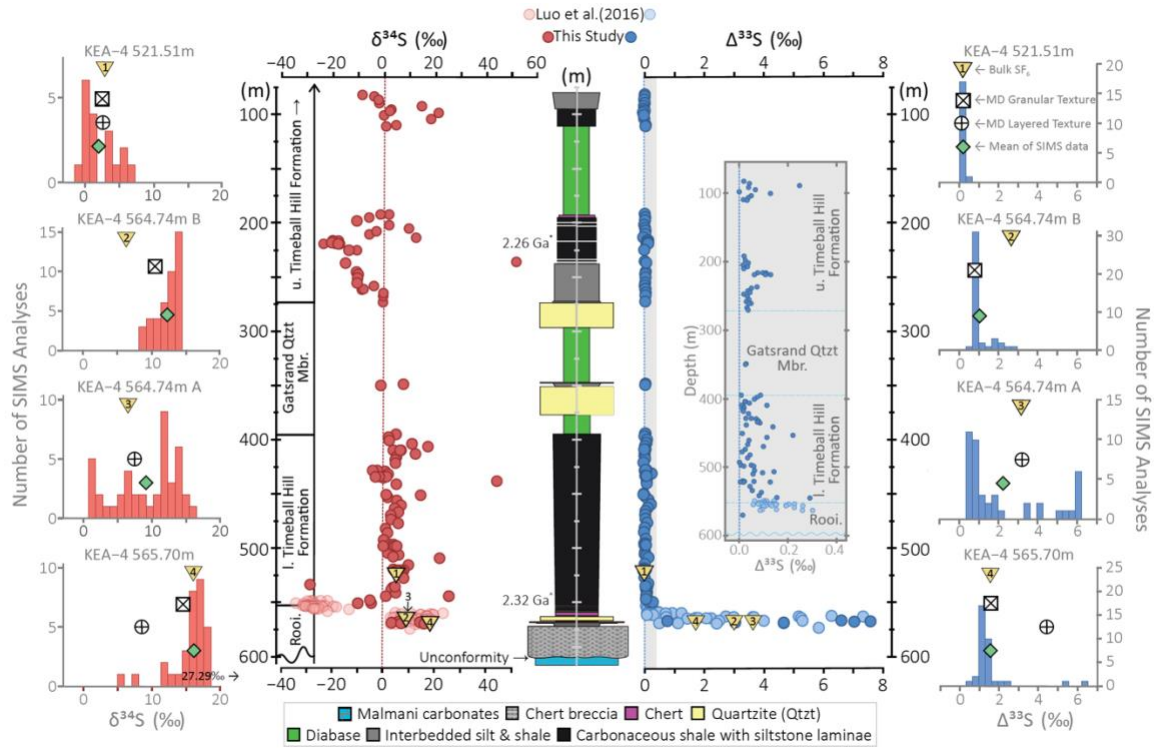


Figure 2: Bulk and grain-scale triple sulfur isotope data from core KEA-4. Here, the bulk SF_6 -derived $\delta^{34}\text{S}$ (red) and $\Delta^{33}\text{S}$ (blue) data are plotted stratigraphically, while the intra-sample $\delta^{34}\text{S}$ and $\Delta^{33}\text{S}$ ranges are illustrated as color-coded, sample-specific, histograms. The stratigraphic position of the samples studied by SIMS are located numerically as yellow triangles. For clarity, the grey insert illustrates the stratigraphic spread in $\Delta^{33}\text{S}$ values between 0 and 0.4‰ that was otherwise obscured by the large $\Delta^{33}\text{S}$ values that typify the Rooihooigte Formation (Rooi). Within the stratigraphic plots, the majority of the data below 548.85 m have been published previously (34), and are separated from those reported herein via the use of lighter shaded datapoints. The $\delta^{34}\text{S}$ and $\Delta^{33}\text{S}$ histograms are constructed using respective 1‰ and 0.4‰ bin widths, each centered at 0‰. Superimposed on these histograms, the respective green and yellow symbols depict the average SIMS-derived values and their corresponding SF_6 bulk values. Ornamented open symbols reflect micro-drilled (MD), texture-specific, SF_6 -derived values (34). Where applicable, analytical uncertainties are encompassed by individual data-points. Age constraints are provided by a Re–Os isochron obtained from diagenetic pyrites straddling the Rooihooigte–Timeball Hill formational boundary (2316 ± 7 Ma, 33, 69), and upper-intercept $^{207}\text{Pb}/^{206}\text{Pb}$ SHRIMP ages of tuff-hosted zircons from the lower (2310 ± 9 Ma; 63) and upper Timeball Hill Formation (2256 ± 6 and 2266 ± 4 Ma; 63). The lower (l.) and upper (u.) Timeball Hill formations are separated by the Gatsrand Quartzite Member (Gatsrand Qtzt Mbr)/Klapperkop Member that houses the oolitic ironstones within the Timeball Hill Formation (51; SI Appendix).

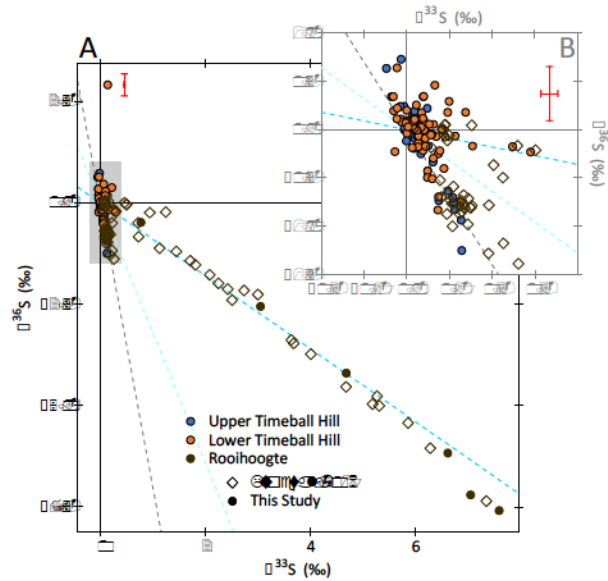


Figure 3: Bulk-derived $\Delta^{36}\text{S}$ – $\Delta^{33}\text{S}$ systematics of chemically isolated sulfides from core KEA-4. Insert B expands the data condensed around the origin, conforming to the grey box in A. In both plots, color is used to separate samples from the Rooihogte (black) and the lower (orange) and upper (blue) Timeball Hill formations, respectively. Filled symbols differentiate new data from those presented previously (34), which are given as open symbols. Dashed lines in each plot correspond to the so-called Archean reference array (blue, $\Delta^{36}\text{S}/\Delta^{33}\text{S} \approx -0.9$ (14, 15, 47), the inferred mixing S-MIF–S-MDF array (green, $\Delta^{36}\text{S}/\Delta^{33}\text{S} = -3.2$) identified by Poulton et al. (8) and the much steeper slope resulting from biologically mediated mass-dependent processes (gray, $\Delta^{36}\text{S}/\Delta^{33}\text{S} \approx -7$; (14, 22). Common external 3σ uncertainties are shown for $\Delta^{33}\text{S}$ ($\pm 0.021\text{‰}$) and $\Delta^{36}\text{S}$ ($\pm 0.285\text{‰}$). Figure S4 presents a lithostratigraphic dissection of the insert for clarity.

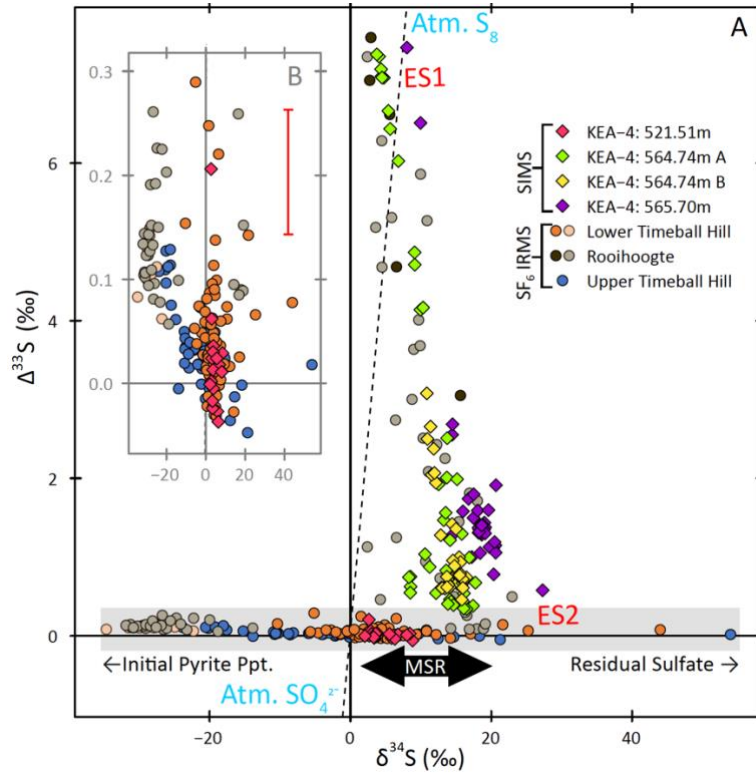


Figure 4: Bulk- and SIMS-derived $\Delta^{33}\text{S}$ – $\delta^{34}\text{S}$ systematics of sulfides from core KEA-4. Insert **B** expands the data condensed around the origin, conforming to the grey box in **A**. In both plots, bulk and spatially resolved data are differentiated by shape and color. Here, colored circles place the bulk GS-IRMS data within a lithostratigraphic context, while colored diamonds discriminate between samples targeted by SIMS analysis (see legend). The dashed line represents the positive relationship between $\Delta^{33}\text{S}$ and $\delta^{34}\text{S}$ ($\Delta^{33}\text{S} \approx 0.9 \times \delta^{34}\text{S}$) seen in Neoarchean sulfides, whose origin is thought by the majority to record mixing between atmospherically derived elemental sulfur (S_8) and sulfate (SO_4^{2-}), carrying positive and negative $\Delta^{33}\text{S}$ values, respectively (23, 47). Around this array, microbial sulfate reduction (MSR) causes lateral movement in $\delta^{34}\text{S}$ (arrows) via the preferential reduction, conversion and precipitation of low- $\delta^{34}\text{S}$ pyrite, concurrently driving the $\delta^{34}\text{S}$ of seawater-sulfate to higher values. The endmembers discussed in the text (ES1 and ES2) are approximately located in red. Precision varies by analytical approach, with larger uncertainties associated with SIMS analysis relative to more traditional GS-IRMS analysis. Adopting the larger of these uncertainties, only within the inset (**B**) do the SIMS $\Delta^{33}\text{S}$ uncertainties become resolvable beyond the datapoints (red vertical error bar).

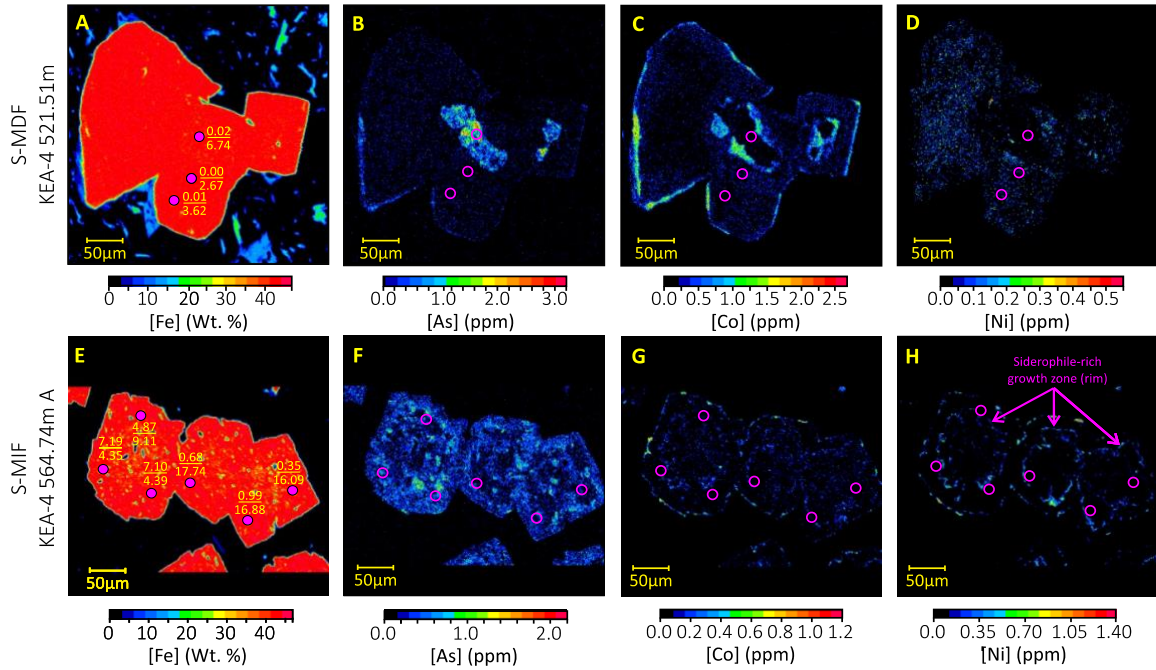


Figure 5: Elemental maps (Fe, As, Co, Ni) of representative pyrite agglomerates selected from the S-MDF (KEA-4 521.51m, Domain #1a, Grain 1; A–D) and the S-MIF (KEA-4 564.74m A, Domain #4b, Grain 9; E–H) intervals. The pink circles locate the SIMS pits within each WDS map, with the resultant $\Delta^{33}\text{S}$ and $\delta^{34}\text{S}$ data disclosed as the numerator and denominator, respectively. The arrows in panel H emphasize the subtle siderophile-rich zone discussed within the text. See [SI Appendix Fig. S7](#) for an annotated interpretation of panel H.

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