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Detrital chromite from Jack Hills, Western Australia: signatures of metamorphism and
constraints on chromite provenance

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ABSTRACT

Detrital chromites are commonly reported within Archean metasedimentary rocks, but have thus far garnered little attention for use within provenance studies. Yet, systematic variations of Cr-Fe spinel mineral chemistry with changing tectonic setting means chromite has been extensively used as a petrogenetic indicator, and so detrital chromites represent good candidates to investigate the petrogenesis of eroded Archean mafic and ultramafic crust. Here, we report the compositions of detrital chromites within fuchsitic (Cr-muscovite rich) metasedimentary rocks from the Jack Hills, geologically renowned for hosting Hadean (>4000 Ma) zircons, situated within the Narryer Terrane, Yilgarn Craton, Western Australia. We highlight signatures of metamorphism, including highly elevated ZnO and MnO, coupled with lowered Mg# in comparison to magmatic chromites, development of pitted domains, and replacement of primary inclusions by phases abundant as metamorphic assemblages within host metasedimentary rocks. Oxygen isotope compositions of detrital chromites indicate partial exchange to complete equilibration with host metasedimentary rocks. Variability of metamorphic signatures between chromites sampled only meters apart further indicates modification occurred *in-situ* by interaction detrital chromites with of metamorphic fluids and secondary mineral assemblages. Alteration likely occurred during upper greenschist to lower amphibolite facies metamorphism and deformation of host metasedimentary rocks at ~2650 Ma. Regardless of metamorphic signatures, sampling location or grain shape, chromite cores yield a consistent range in Cr#. While other key petrogenetic indices, such as Fe₂O₃ and TiO₂ contents, are complicated in Jack Hills chromites by mineral non-stoichiometry and secondary mobility within metasedimentary rocks, we demonstrate that the Cr# of chromite yields significant insights into their provenance. Importantly, moderate Cr# preclude a komatiitic origin for the bulk of chromites. This reflects a dearth of komatiites and associated intrusives within the erosional catchment of the Jack Hills metasedimentary units. We suggest Cr# fit well

with chromites derived from layered intrusions, and that a single layered intrusion may account for the observed chemical compositions of Jack Hills detrital chromites. Thus, we show that, where careful characterisation of key metamorphic signatures is undertaken, detrital chromites may yield valuable information on the petrogenesis and geodynamic setting of poorly preserved mafic and ultramafic crust.

INTRODUCTION

The detrital record of crustal evolution within the early Earth is dominated by analysis of the mineral zircon. This is particularly evident in the Narryer Terrane, within the Yilgarn Craton of Western Australia (Fig. 1). Here, Proterozoic to late Archean (Cavosie *et al.*, 2004; Crowley *et al.*, 2005; Rasmussen *et al.*, 2010; Wang & Wilde, 2018) metasedimentary rocks in the Jack Hills (Fig. 2) yield the oldest known fragments of terrestrial crust; individual grains of detrital zircon yield ^{207}Pb - ^{206}Pb ages of up to 4374 ± 6 Ma (Compston & Pidgeon, 1986; Wilde *et al.*, 2001; Valley *et al.*, 2014). Despite isolated occurrences of Hadean detrital zircon elsewhere (*e.g.*, Byerly *et al.*, 2018 and references therein), including Mt. Narryer to the SW (Fig. 1; Froude *et al.*, 1983; Pidgeon & Nemchin, 2006), and rare examples of Hadean xenocrystic zircon (Nelson *et al.*, 2000; Wyche *et al.*, 2004; Iizuka *et al.*, 2006; Chaudhuri *et al.*, 2018), Jack Hills detrital zircons represent critically important remnants of Hadean (>4000 Ma) crust.

These zircons therefore provide a unique window into the Hadean Earth, and have subsequently been rigorously interrogated using numerous geochemical and isotopic techniques (*e.g.*, Cavosie *et al.*, 2018). Previous investigations have broached a wide range of subjects, including Earth's ancient dynamo (Borlina *et al.*, 2020; Tarduno *et al.*, 2020), the origins of life (Bell *et al.*, 2015) and the role of impacts within the early Earth (Bell & Harrison, 2013; Cox *et al.*, 2017). Despite intense study, what Jack Hills zircons tell us about the composition, evolution and subsequent destruction of their original host rocks is controversial, with conflicting hypotheses inferring vastly disparate geodynamic conditions within the early Earth. Such hypotheses include a 'cool early Earth' (*e.g.*, Valley *et al.*, 2002), with the putative operation of plate tectonics (Harrison *et al.*, 2008; 2017; Bell *et al.*, 2014), production of zircon-bearing crust by internal reworking of mafic protocrust (Amelin *et al.*, 1999; Kemp *et al.*, 2010), or compositionally diverse Hadean protoliths (Wang & Wilde, 2018).

While the Jack Hills zircon record yields valuable, if controversial, constraints on the evolution of felsic Hadean and Archean crust, the zircon record provides little information on the evolution of contemporaneous mafic and ultramafic crust. This is of particular importance, as emerging evidence suggests mafic to ultramafic crust was the dominant compositional component of the Archean (*e.g.*, Dhuime *et al.*, 2015; Kamber *et al.*, 2015; Tang *et al.*, 2016). Evolved, zircon-bearing crust may therefore be over represented within detrital records, in part due to the poorer preservation potential of its mafic to ultramafic counterparts. While the wider Yilgarn Craton is renowned for the presence of economically significant komatiites (Arndt *et al.*, 2008), the generation and evolution of mafic and ultramafic crust within the Narryer Terrane is poorly constrained. Indeed, the only >3100 Ma mafic and ultramafic crust described within the entire Yilgarn Craton is the Eoarchean Manfred Complex (Wyche, 2007), a 3730 Ma disseminated layered intrusion within the Narryer Terrane (Fletcher *et al.*, 1988; Kinny *et al.*, 1988; Myers, 1988b; Rowe & Kemp, 2020).

However, the largely unexplored eroded remnants of mafic and ultramafic crust are ubiquitous within Jack Hills metasedimentary rocks, in the form of detrital chromite. Unlike zircon and other detrital phases observed at Jack Hills, including minor monazite and xenotime (Rasmussen *et al.*, 2010; Iizuka *et al.*, 2010), chromite has a magmatic provenance restricted solely to mafic and ultramafic crust (Barnes & Roeder, 2001). Furthermore, chromites within their mafic or ultramafic protoliths are frequently used as a petrogenetic indicator, owing to the presence of systematic chemical variations in chromite formed under different tectonic conditions (Irvine, 1965; 1967; Dick & Bullen., 1984; Roeder, 1994; Barnes & Roeder, 2001; Kamenetsky *et al.*, 2001). Critically, the sequestration of platinum group elements (PGEs) into spinel means that chromite is amenable to geochronology using Re-Os and Pt-Os decay systems (Shirey & Walker, 1998), potentially allowing a temporal framework of the generation of chromite-bearing mafic and ultramafic crust to be constrained from the detrital record. The

value of detrital chromites for elucidating the provenance of magmatic protoliths has previously been demonstrated (*e.g.*, Barnes & Roeder, 2001; Lenaz & Princivalle, 2005; Barkov *et al.*, 2013), though additional care is required to quantify the effects of metamorphism on detrital chromite found within ancient terranes (Barnes, 2000; Colás *et al.*, 2014). Understanding how and when the Jack Hills detrital chromite formed will therefore provide valuable information on the composition and petrogenesis of poorly described mafic and ultramafic crust that was potentially exposed at the time the Jack Hills sedimentary succession was deposited

Here, we report major and minor element geochemistry of detrital chromites from thirteen samples of metasedimentary rocks collected from within the Jack Hills (Fig. 2). This study combines data collected by two groups (University of Bristol; 14WA and 16WA samples and University of Wisconsin-Madison; 01JH samples), representing the first systematic study of detrital chromite from Archean sedimentary rocks, and signifying a new direction compared to traditional provenance studies undertaken by analysis of detrital zircon. We highlight the effects of metamorphism on chromite mineral chemistry, discuss the retention of primary signatures, and propose a plausible provenance for detrital chromites. Finally, we discuss the significance of these grains for enhancing our understanding of the metamorphic history of the Jack Hills metasedimentary rocks, the distribution of mafic and ultramafic crust within the Archean upper crust at the time of deposition, and the wider potential of detrital chromites for provenance studies of ancient mafic and ultramafic crust.

GEOLOGICAL SETTING

The Narryer Terrane

The Narryer Terrane is the most north-westerly terrane within the Yilgarn Craton, in Western Australia (Myers, 1988a; Kemp *et al.*, 2018; Fig. 1), and has been interpreted as a deep crustal

allochthon thrust above the Youanmi Terrane (Nutman *et al.*, 1993), prior to or coincident with cratonic amalgamation (Kemp *et al.*, 2018). The terrane is dominantly composed of granitic lithologies, now largely preserved as quartzofeldspathic orthogneisses with >3 Ga protolith ages, with minor ultramafic and mafic intrusives, and metasedimentary rocks (Myers & Williams, 1985; Williams & Myers, 1987; Myers, 1988a). Neoproterozoic (~2700-2650 Ma) granitic rocks are also abundant (Kemp *et al.*, 2018). Much of the Narryer Terrane has undergone high-grade, polyphase deformation, with amphibolite to granulite facies events at ~2700-2650 Ma forming the observed gneissic fabric (Myers, 1988a; Kinny, 1990; Nutman *et al.*, 1991). There is evidence for previous high-grade thermal events, particularly at ca. 3300 Ma (Nutman *et al.*, 1991; Kinny & Nutman, 1996), and it has been postulated orthogneisses within the Narryer Terrane underwent multiple episodes of deformation and anatexis during the Archean (Kinny & Nutman, 1996).

Despite complicated zircon geochronology (Pidgeon & Wilde, 1998), three dominant quartzofeldspathic orthogneiss units are identified within the Narryer Terrane (Kemp *et al.*, 2018). The Meeberrie gneiss is a biotite-rich migmatite that consists of 3670-3600 Ma monzogranitic and 3730 Ma tonalitic protoliths (Nutman *et al.*, 1991; Kinny & Nutman, 1996; Pidgeon & Wilde, 1998). The Eurada gneiss is a series of ~3480 Ma tonalitic gneisses that comprise a fault bound lozenge west of Mount Narryer (Nutman *et al.*, 1991). The Dugel gneiss yields a well constrained age of 3375 ± 26 Ma (Nutman *et al.*, 1991) and it is thought that its syenogranitic protoliths intruded the Meeberrie Gneiss as a series of sheet-like and pegmatitic bodies (Myers, 1988a; Kemp *et al.*, 2018).

The Meeberrie and Dugel gneiss host dispersed fragments of the Manfred Complex, a magmatically and tectonically dismembered and variably metamorphosed layered intrusion. The complex is dominantly amphibolitic (after gabbro and leucogabbro), with pyroxenite, metaperidotite and anorthosite (Williams & Myers, 1987; Fletcher *et al.*, 1988; Myers, 1988b;

Rowe & Kemp, 2020). Relict igneous textures and layering are locally preserved (Kemp *et al.*, 2018). Zircon within Manfred Complex anorthosite and leucogabbro yield $^{207}\text{Pb}/^{206}\text{Pb}$ ages of 3730 ± 6 Ma (Kinny *et al.*, 1988; Kemp *et al.*, 2018), with other lithologies yielding Sm-Nd and Pb-Pb whole rock (WR) ages of 3680 ± 70 Ma and 3689 ± 146 Ma, respectively (Fletcher *et al.*, 1988). Spinel (spinel *sensu-stricto* to picotite) and olivine chemical compositions are consistent with formation of the Manfred Complex within a thickened oceanic plateau; reduced, high Al-Ca-Fe picritic to tholeiitic basaltic parental melt compositions are thought to be derived from shallow partial melting of spinel lherzolite (Rowe & Kemp, 2020).

Jack Hills

The Jack Hills are located at the southern margin of Narryer Terrane (Figs. 1 & 2) and comprise a thin, ~70km-long belt, with a distinctly curvilinear morphology produced by dextral shearing (Spaggiari, 2007a). The Jack Hills belt is tectonically juxtaposed with the surrounding gneisses, except for localised intrusion of 2654 Ma monzogranite, also known as ‘The Blob’ (Pidgeon & Wilde, 1998; Spaggiari *et al.*, 2007b), and has been interchangeably referred to as a greenstone (*e.g.*, Spaggiari *et al.*, 2007a/b) or metasedimentary/supracrustal belt (*e.g.*, Wang & Wilde, 2018). It largely consists of siliciclastic units, including metaconglomerate, quartzite and quartz-mica schist, with minor intercalated mafic and ultramafic rocks, banded iron formation (BIF) and chert. The presence of grunerite within BIF and hornblende within mafic schist indicates at least portions of the belt have reached amphibolite facies metamorphism; later greenschist facies metamorphism defines the dominant metamorphic signature of the belt (Spaggiari, 2007a).

Deformation has tectonically disturbed and juxtaposed lithological associations, making an original stratigraphy difficult to discern. Spaggiari (2007a) divided the belt into four associations determined by lithological variability (Fig. 2). Briefly, association 1 consists of

interbedded BIF, chert and quartzite, mafic and ultramafic rocks, and black and white banded quartzites. Association 2 yields pelitic to semi-pelitic associations, now present as quartz-mica and andalusite schists, with accompanying mafic schist and quartzite. The presence of an S1 cleavage and recumbent folding absent from other lithological units suggests associations 1 and 2 were deformed prior to the deposition of units 3 and 4 (Spaggiari, 2007a). Association 3 is restricted to the central region of the belt at Eranondoo Hill (Fig. 2), and was derived from mature, siliciclastic sediments interpreted to have been deposited within a deltaic alluvial fan (Spaggiari *et al.*, 2007a/b) between ~3050 Ma and 2650 Ma (Crowley *et al.*, 2005; Rasmussen *et al.*, 2010). The discovery of Proterozoic detrital zircons (Cavosie *et al.*, 2004; Dunn *et al.*, 2005; Grange *et al.*, 2010) within the Jack Hills belt led to the recognition of Unit 4, which hosts metasedimentary rocks deposited during the Proterozoic. Wang & Wilde (2018) observed interbedded siliciclastic units at the same apparent metamorphic grade but with both Archean and Proterozoic depositional ages, highlighting that the depositional and/or tectonic relationships of units 3 and 4 may be more complex than previously postulated.

Metasedimentary rocks within Jack Hills show clear indications of deformation; intense shearing is particularly evident within the anastomosing micaceous matrix, where a strong foliation is coincident with flattening and recrystallisation of quartzite cobbles (Spaggiari, 2007a). Thermal or fluid events within the Jack Hills belt occurred at ~3080 Ma, 2650 Ma, ~1850-1800 Ma and 800 Ma (Spaggiari, *et al.*, 2007b; Rasmussen *et al.*, 2010; 2011). Monazite-xenotime thermometry of secondary inclusions within detrital zircon yield temperatures of 420-475 °C (Rasmussen *et al.*, 2011), argued to represent peak upper greenschist to lower amphibolite facies metamorphism within unit 3 metasedimentary rocks at ~2650 Ma (Rasmussen *et al.*, 2010). However, like the depositional ages of metasedimentary units, the exact P-T conditions and timing of metamorphism within unit 3 and 4 metasedimentary units are ambiguous (Kemp *et al.*, 2018). A Proterozoic overprint at ~1800 Ma (Spaggiari *et al.*,

2007b; Rasmussen *et al.*, 2010) and a discrete event at 800 Ma (Rasmussen *et al.*, 2010; 2011) are of unknown metamorphic grade, but also coincide with formation of monazite and xenotime (Rasmussen *et al.*, 2010; 2011).

Metasedimentary rocks from Eranondoo Hill, chiefly oligomict pebble to cobble metaconglomerate and quartzite, are renowned for hosting Hadean detrital zircon (Compston & Pidgeon, 1986; Wilde *et al.*, 2001), with most studies focused on metasedimentary rocks at the W-74 ‘discovery site’ (Fig. 2). Unsurprisingly, Jack Hills detrital zircons have been the subject of numerous publications and reviews (*e.g.*, Harrison *et al.*, 2017; Cavosie *et al.*, 2018). While a detailed description of the zircons is beyond the remit of this publication, it is important to note there is some debate surrounding the source of detrital zircon ^{207}Pb - ^{206}Pb age distribution peaks, and thus the source of detritus. Whilst ^{207}Pb - ^{206}Pb age peaks correspond with major units of the Narryer Terrane and the granitic lithologies surrounding Jack Hills (Nutman *et al.*, 1991; Pidgeon & Wilde, 1998), distal sources of more intermediate composition have also been suggested (Crowley *et al.*, 2005). Additionally, the protoliths of >3800 Ma zircons are unknown, and may derive from a source exogenous to the Narryer Terrane. Sources external to the Narryer Terrane therefore cannot be discounted for detrital chromites. If detrital phases are derived from the Narryer Terrane, a further consideration is that both detrital zircon and chromite are present within quartzite clasts of the metaconglomerates (*e.g.*, Grange *et al.*, 2010; Dare *et al.*, 2016). This demonstrates that the detrital phases have undergone multiple sedimentary cycling events. It is therefore plausible that the source of detrital chromite had already been completely or partially eroded at the time of deposition of the Jack Hills sediments (3050 Ma to 2650 Ma), and therefore may not represent the distribution of mafic and ultramafic crust during the late Archean (Crowley *et al.*, 2005; Rasmussen *et al.*, 2010).

Despite concerted interest in detrital zircon at Jack Hills, detrital chromites within the same metasedimentary rocks have garnered less attention. Detrital chromites with low Mg#

(100x molar $\text{Mg}/(\text{Mg}+\text{Fe}^{2+})$) and elevated ZnO and MnO were reported by Cavosie *et al.* (2002) for grains from the W-74 site; comparable major element compositions were described by Dare *et al.* (2016) in analyses of detrital chromites ~1 km to the NW of this locality. Dare *et al.* (2016) proposed that low MgO contents of chromites precluded derivation of Jack Hills detrital chromites from komatiites, and that both chromite and Fe-Ni-sulphides observed within quartzite clasts are the erosional products of at least one layered intrusion. Unpublished Re-Os model ages ($T_{\text{MA}s}$) of 3500 Ma to 3200 Ma were reported by Valley *et al.* (2005) for the same population of chromites described within Cavosie *et al.* (2002). These data suggest the analysed detrital chromites are at least Palaeoarchean in age, and highlight the potential preservation of robust Re-Os systematics through peak metamorphism within the Jack Hills.

METHODS AND MATERIALS

Sample collection and preparation

Sampling locations of metasedimentary rocks are shown in Fig. 2. Seven samples of pebble metaconglomerates (14WA1-4, 16WA5-6 and 01JH54) were collected from within 10 m of the W-74 site (Wilde *et al.*, 2001) at Jack Hills (Fig. 2b). A further sample of pebble metaconglomerate (16WA8) and a quartzite (16WA7) were collected along strike approximately 35m to the WSW. Two pebble metaconglomerates (16WA9-10) were sampled from a prominent ridge across the valley to the NE. 01JH35 is a metaconglomerate sampled ~800 m to the NE of the W-74 site. 01JH36, collected approximately 200m to the ENE of 01JH35, is a quartzite with discrete, mm-scale heavy mineral bands (Cavosie *et al.*, 2004). Petrographic sample descriptions of metasedimentary rocks and their detrital mineral assemblages are provided in supplementary material. While fuchsite-rich (muscovite with >1 wt. % Cr_2O_3 ; Challis *et al.*, 1995) metasedimentary rocks were largely collected to yield the

highest concentrations of chromite grains, 01JH36, 16WA6 and 16WA10 lack significant fuchsite. Two chromite-bearing ultramafic rocks from within the Narryer Terrane are also included, with descriptions and sampling locations provided within Supplementary Material. Sample 13TKN22 is a metaperidotite from the 3730 Ma Manfred Complex, collected from NE of Mt. Narryer, ~60 km SW of the Jack Hills. Sample 16WA13 is a heavily recrystallised ultramafic rock of unknown age sampled from the SW limb of Jack Hills (Fig. 2a); this sample is part of association 1 of Spaggiari *et al.* (2007a) and so is considered >3000 Ma.

Chips of Narryer Terrane ultramafic rocks were mounted in epoxy for petrographic study and identification of chromite. 14WA, 16WA and 01JH35 chromites were separated using standard crushing and separation procedures, then sieved to yield size fractions of ≤ 500 μm for analysis. 01JH36 and 01JH54 chromites were separated by electric pulse disaggregation (EPD). 14WA1-4 heavy minerals were concentrated using the Wilfey table, followed by magnetic and heavy liquid separation, while 16WA5-10 separates were concentrated via panning. 01JH chromites were concentrated using heavy liquids and magnetic separation. Chromites were picked by hand, separated per sample location, grain size and rounding shape, and mounted in epoxy. Detrital grains within a thin section of 14WA2 were also analysed, as were chromite liberated via HF leaching of quartzite cobbles from 14WA2 and 16WA5. No chemical differences between chromites present within quartzite cobbles or matrix were observed, so they are discussed together below.

Chromite mineral chemistry

14WA, 16WA and 13TKN22

Chromites were imaged using reflected light microscopy and/or back scattered electron imaging using a Hitachi S-3500N scanning electron microscope at the University of Bristol. Chromite major elements were determined from individual spots and line scans using the Cameca SX100

electron microprobe at the University of Bristol. Chromium, Al, Fe, Mg, Zn, Mn, Ti, V, and Ni abundances of chromite were determined using a 20kV accelerating voltage, a 10nA beam current, 1 μm beam diameter, and a PAP matrix correction. Silicon, Na and Ca were included within the set up to monitor any silicate contamination; analyses with >0.15 wt.% oxide of these elements were omitted. Counting times were 30s for Cr, Fe, Al and Zn and 60s for Ti, Ca, Mg, Na, Mn, V, Si and Ni, with the following standards used for instrument calibration: Cr_2O_3 (Cr), albite (Al and Na), ilmenite (Fe and Ti), St. John's olivine (Mg and Si), Zn metal (Zn), Mn metal (Mn), V metal (V), Ni metal (Ni) and wollastonite (Ca). The overlap of Ti $\text{K}\beta$ on V $\text{K}\alpha$ was corrected either via analysis of V-free SrTiO_3 , or using high-resolution slits. No resolvable variability in the V_2O_3 contents of chromite or chromite secondary standards was observed between the two set-ups. Typical detection limits, expressed in ppm, were: Cr(260), Al(250), Na(450), Fe(320), Ti(120), Mg(250), Si(180), Zn(400-500), Mn(300), V(200), Ni(350) and Ca(140).

Four spinel standards (8316, 8311, 79-1 and 8315), well characterised for $\text{Fe}^{3+}/\Sigma\text{Fe}$ using both EPMA and Mössbauer from Wood & Virgo (1991), were analysed *in-run* to monitor the integrity of EPMA measurements. Long term reproducibility of major elements (Al, Cr, Fe, Mg) within 8316 and 8311, which possess the most comparable mineral chemistry to samples, are <1.2 wt.% (2σ). The method of Droop (1987) was employed to calculate the ferric iron content of chromite by stoichiometry. The determined $\text{Fe}^{3+}/\Sigma\text{Fe}$ (molar $100\times \text{Fe}^{3+}/(\text{Fe}^{3+}+\text{Fe}^{2+})$) of 8316 (0.21 ± 0.07 ; 2σ) and 8311 (0.17 ± 0.06 ; 2σ), measured across multiple analytical sessions, are in good agreement with reported values. After correction for ferric iron, 14WA4, 16WA6 and 16WA9-10 yielded chromite with systematically low totals, typically between 97% and 98%, despite other analyses within the same run yielding good totals. The cause of low totals within these samples is unknown but may reflect non-stoichiometry, likely due to site vacancies. Fe_2O_3 contents of chromite should therefore be considered minimum abundances for

14WA and 16WA samples. Chromites with totals of 97.5-102% from 14WA4, 16WA6 and 16WA9-10 are included, while measurements from other samples are restricted to those that yield totals of 98-102%.

01JH samples

EPMA analyses were undertaken at the University of Wisconsin-Madison Department of Geoscience using a CAMECA SX51 electron microprobe equipped with Probe for EPMA software. Analyses were conducted at 20 kV, with a 20 nA beam current and a focused electron beam. Count times for all elements were 10 seconds on peak and a total of 10 seconds on off-peak background positions. The matrix correction was the PAP procedure. PHA differential was used for Al and O, other elements were in integral mode. Oxygen was determined with a 60° 2d PC0 diffractor crystal, while Si, Al and Mg were acquired on TAP, Ti and Cr on PET, and V, Mn, Fe, Ni, Cu and Zn on LIF. Overlap corrections were made on Al for Cr, Ni and Ti, on V for Ti, on Fe for Mn, and on O for Cr. The following standards were used for instrument calibration: USNM Chromite (Cr, Al, Mg, O), Harvard University hematite (Fe), synthetic tephroite (Mn, Si), V₂O₅ (V), rutile (Ti), Ni₂SiO₄ (Ni), ZnAl₂O₄ (Zn) and Cu₂O (Cu). Reproducibility, Detection limits.

EPMA analyses were processed using two methods. In the first, oxygen measurements were omitted and analyses were processed using the same techniques as 14WA and 16WA; namely, calculation of ferric iron by stoichiometry using the method of Droop (1987), and exclusion of analyses with >0.15 wt.% SiO₂ and totals of <98 %. In the second method, cations and measured oxygen were converted into molar proportions, and the ferrous/ferric ratio of iron adjusted to balance the negative charge from measured O. This technique assumes Fe is the only multivalent cation present (*e.g.*, Cr is present as Cr³⁺, V as V³⁺ and Mn as Mn²⁺) and that site vacancies are negligible. The latter scenario is unlikely to be true within altered Cr-rich

spinel (see Kamperman *et al.*, 1996), and so ferric iron compositions determined by the second method represent maximum values. For consistency, data calculated using stoichiometry are presented within the main text, while analyses processed using measured O are discussed in supplementary material.

Oxygen isotope analyses

Chromite grains from samples 01JH36, 01JH42 and 01JH54 were analysed for oxygen isotope ratio by laser fluorination at the University of Wisconsin-Madison Department of Geoscience. A total of 35 analyses of oxygen isotope ratio in chromite were made using gas source mass spectrometry with BrF₅ and a 32 W CO₂ laser (Valley *et al.*, 1995). Chromite data were corrected for accuracy with the UWG-2 garnet oxygen standard, which was analysed multiple times at the beginning of the session (Valley *et al.*, 1995). The reproducibility of UWG-2 per session ranged from ± 0.06 to ± 0.24 (2 σ). Samples of chromite were first sieved into seven size fractions, ranging from 105-149 μm to >500 μm (Supplementary Table 1). At least three size fractions were analysed from each sample, with two to three aliquots analysed of each size fraction. Aliquots of chromite were prepared by soaking detrital grains in concentrated HF at room temperature overnight. Each ~ 2 mg aliquot of chromite thus consisted of hundreds of clean detrital grains. Values are reported in standard per mil notation relative to V-SMOW (Vienna Standard Mean Ocean Water).

RESULTS

Chromite morphology and inclusion assemblages

Jack Hills detrital chromites are ≤ 500 μm grains of variable morphology that are enclosed by, or closely associated with, fuchsite within quartzite or metaconglomerate matrix (Fig. 3a). Finer

($\leq 100\ \mu\text{m}$) chromite grains have previously been reported within metaconglomerate quartzite cobbles (Dare *et al.*, 2016), but were uncommon in samples of this study (Fig. 3b), though HF leaching of quartzite cobbles liberated a small number of grains ($n = 12$). Many chromites show textural evidence of sedimentary transport, including abrasion and rounding of broken surfaces. Chromite derived from Jack Hills metasedimentary rocks occur as euhedral octahedra (EO), rounded octahedra (RO) or rounded grains (RC: rounded chromite) (Fig. 3a). EO yield minimal rounding on two or fewer faces, with many euhedral grains showing little or no evidence of sedimentary transport. RC grains demonstrate limited evidence of original habit on two or less faces, and are often present as highly spherical morphologies. RO are chromites with morphologies intermediate to EO and RC, and are the most abundant morphology. More quartzitic lithologies possess a greater proportion of euhedral grains. 16WA9 and 16WA10, sampled furthest from the W-74 site, yield a more bimodal distribution with chromite dominantly RC and minor EO.

A range of internal textures are present (Fig. 4), with grains often heavily cracked and displaying ragged or lobate boundaries with surrounding fuchsite (Fig. 3c). Fractures occasionally display distinct polygonal morphologies (Fig. 4b), and larger fractures may be filled with quartz, fuchsite, and Fe-oxide. Distinct ‘pitted’ textural domains within chromite reveal the presence of typically $\leq 10\ \mu\text{m}$ inclusions of dominantly quartz and fuchsite (Figs. 4&5). Rare porous textures with $\geq 30\ \mu\text{m}$ inclusion assemblages may occasionally account for the majority of the grain (Fig. 4e-f). Both pitted and porous domains are conspicuous at chromite rims or adjacent to cracks, and particularly where chromite is enclosed by fuchsite. Fine laths of rutile, often showing apparent alignment to chromite crystallographic directions, are also commonly associated with pitted and porous domains (Fig. 4e-f & 5a). Isolated monomineralic and polyphase silicate and oxide inclusions present within chromites are typically globular or anhedral in habit (Fig. 5), but may more rarely possess subhedral to

385 euhedral morphologies (Fig. 5a&d). Inclusions associated with fractures and crack filling
386 assemblages of quartz, fuchsite and rutile (Fig. 5a-b&e) are often accompanied by other fine-
387 grained (typically $\leq 10\ \mu\text{m}$) phases, including Fe-oxide (Fig. 5b), Fe-sulphide (Fig. 5c), and
388 rarer monazite (Fig. 5f).

389 Chromite mineral chemistry

390 Representative analyses of detrital chromites from each metasedimentary sample location are
391 presented in Table 1, and the full EPMA data set is available in Supplementary Material 2.
392 Chromite major element abundances show no variation with grain size or rounding shape
393 (Supplementary Figs.) but differ systematically between sampling locations (Figs. 6 & 7).

394 *Sample variability (Ti^{4+} and divalent cations; Fe^{2+} , Mg^{2+} , Zn^{2+} , Mn^{2+} , and Ni^{2+})*

395
396 Chromite from all sampled metasedimentary rocks yield elevated ZnO (up to 13 wt.%) and
397 MnO (up to 4.4 wt.%), coupled with low Mg# (typically ≤ 30 ; Fig. 6). Largely, samples with
398 increasingly elevated ZnO yield lower and more homogeneous Mg# (Fig. 6a). 14WA2
399 chromites possess the lowest wt.% ZnO (average ZnO 1.23 wt.%) and largest range of Mg#,
400 with a single grain (14WA2-PB-46) yielding Mg# of ~ 30 . Unusually elevated MnO contents
401 are present in 01JH35, where chromites yield > 3.5 wt.% MnO (Fig. 6b). No variation of MnO
402 with Mg# is apparent (Fig. 6b), though some 16WA samples show variance between ZnO and
403 MnO (Fig. 6d).

404
405 TiO_2 is typically < 1 wt.% (Figs. 6c). Individual analyses where TiO_2 is above 1 wt.%
406 correlate with areas where laths of rutile are present, and are omitted from consideration.
407 Elevated TiO_2 of ≥ 1 wt.% is most apparent within lower Cr# (100x molar $\text{Cr}/(\text{Cr}+\text{Al})$) grains
408 (Fig. 7). As the bulk of analyses possess ≤ 0.3 wt.% TiO_2 , this suggests analyses yielding TiO_2
409 > 0.3 wt.% may also overlap with rutile laths. However, many chromites with ~ 0.3 - 0.8 wt.%

TiO₂ show no evidence of rutile laths in BSE images. NiO within grains is largely below the quantification limit of ~0.04 wt.%.

Sample variability (trivalent cations: Cr³⁺, Al³⁺, V³⁺, Fe³⁺)

Despite considerable variability of divalent cation contents between sampling locations, detrital chromite show a similar range in Cr# between samples (Fig. 7). Chromites yield moderate to high, and variable Cr# of 48-82, with the bulk of chromites displaying Cr# of 54-66 (Fig. 7a). The apparent smaller range of Cr# in chromites with higher wt.% ZnO is likely a sampling bias due to fewer analyses. V₂O₃ was not measured for 16WA5-7, but where measured is present at 0.05 wt.%, to 0.37 wt.% and shows no inter-sample variability. Jack Hills detrital chromites yield low calculated Fe₂O₃ contents; the bulk of the population contain ≤2 wt.% Fe₂O₃, though analyses with up to 8 wt.% Fe₂O₃ are present. This results in highly variable Fe³⁺/ΣFe (100x molar Fe³⁺/(Fe³⁺+Fe²⁺)) ratios of 0-20 (Fig. 7d). Many chromites yield non-stoichiometric compositions (e.g., deviation from A²⁺B³⁺₂O₄), shown by cation totals of less than 3 when normalised to 4 oxygens. This likely reflects cation vacancies (e.g., Kamperman *et al.*, 1996), and results in an under-estimation of ferric iron contents by stoichiometric calculations (Droop, 1987). Non-stoichiometry also affects analyses conducted by direct measurement of oxygen (Supplementary Material); analyses with significant cation deficiency results in substantial over-estimation (>10 wt. %; Supplementary Material) of ferric iron contents to balance the negative charge of oxygen. Thus, stoichiometric and charge balance of 01JH analyses represent minimum and maximum ferric iron contents of Jack Hills detrital chromites, respectively.

Within grain variability

Zoning; While the largest systematic variations in chromite chemistry are between sample location, line scans of 14WA and 16WA detrital chromites reveal distinct zonation trends (Fig.

8). Chromites do not exhibit morphologically or microstructurally distinct cores and rims, but we use those terms here to describe the centre and outer portions of grains. Most chromites have homogeneous ZnO contents (Fig. 8a) or else slightly lower wt.% ZnO towards rims (Fig. 8b-c). MnO is largely uniform across chromites, although slightly lower abundances towards the rim may be observed. Homogeneous Mg# appears to be restricted to chromites within high ZnO 16WA7 (Fig. 8c), with all other samples yielding chromites that display distinct variability in Mg#. Low ZnO samples, such as 14WA2 and 16WA5, yield clear decreases in Mg# from the core to rim of the grain (Fig. 8a). High ZnO samples, such as 14WA3 and many examples within 14WA4, often yield rims with higher Mg# (Fig. 8b) though lowering of Mg# from core to rim is also observed. Grains with elevated Mg# at the rims show increases in both MgO and FeO from core to rim, compensated by decreases in ZnO and MnO.

While grains yield zoning profiles of divalent cations, intra-grain changes in Cr# are largely absent (Fig. 8). Core-to-rim trends of marginally higher or lower Cr# (*e.g.*, ± 2) are sometimes observed (Fig. 8 a&c), but there is no systematic behaviour in samples from different locations. Like Cr#, Fe₂O₃ contents are generally homogeneous across chromite grains, with variability in some zonation profiles likely reflecting low calculated Fe₂O₃ contents. However, individual grains do yield distinct variations in Fe₂O₃ that appear systematic in origin, present as both decreases and enrichments in Fe₂O₃ from core to rim, with increasing Fe₂O₃ often coupled with a small increase in Cr# (Fig. 8a).

Coupled elemental and textural variation; Distinct zones of elevated Cr# within detrital chromite are rarely present (Fig. 9). These features are closely associated with pitted domains and fractures, and exhibit both diffuse and more commonly sharp (Fig. 9a-e) boundaries with surrounding chromite. High Cr# domains, often distinguished by an absence of fractures and inclusions in comparison to surrounding chromite (Fig. 9a-b), mostly occur at the edge of grains

adjacent to enclosing fuchsite or associated with fractures (Fig. 9a-c). These domains may also be localised, occurring as distinct polygonal areas defined by crystallographic directions (Fig. 9d-e). These domains are characterised by high Cr# (>70), low totals (<98 wt.%), the apparent absence of Fe₂O₃, lower ZnO and V₂O₃, and occasionally slightly elevated MnO. High Cr# domains within detrital chromite are chemically indistinguishable from low Fe₂O₃, high Cr# grains within the same samples. Only a single domain of low Cr# has been observed within chromite, bounding a monomineralic quartz inclusion alongside iron oxide (Fig. 9f).

Oxygen Isotope Compositions

Analysis of oxygen isotope ratio in various chromite size fractions was conducted to determine if detrital chromite grains preserve primary (magmatic) $\delta^{18}\text{O}$ values, or if there is evidence of oxygen exchange with the metasedimentary host rocks. Exchange of oxygen isotopes between chromite and surrounding quartz at low temperature (*e.g.*, <600 °C) would result in lowered $\delta^{18}\text{O}$ values for chromite (*e.g.*, Lowry *et al.*, 2003). Each sample showed >1 ‰ variability in $\delta^{18}\text{O}$, with $\delta^{18}\text{O}$ chromite values ranging from 0.03 to 2.19 ‰ (Fig. 10; Supplementary Table 1). For samples 01JH36 and 01JH42, the lowest chromite $\delta^{18}\text{O}$ values (0.69 and 0.03 ‰, respectively) were detected in the smallest size fraction, and there was a systematic increase of $\delta^{18}\text{O}$ with grain size (Fig. 10a-b). In contrast, 01JH54 showed no systematic variation of chromite $\delta^{18}\text{O}$ with grain size, and also contained the highest $\delta^{18}\text{O}$ measured for chromite at 2.19 ‰ (Fig. 10; Supplementary Table 1).

DISCUSSION

Chromite morphology and the origin of inclusion assemblages

The variably rounded morphologies of chromite, from RC to EO, attests to a protracted and complex reworking history of numerous detrital grains. Further to this, the presence of chromite within quartzite cobbles of metaconglomerates (Fig. 3b; Dare *et al.*, 2016) suggests cycling of some grains in at least two sedimentation events. Categorisation of grains by shape may indicate provenance from multiple geographic sources or differing modes of sedimentary transport (Dott, 2003). The extensive sedimentary cycling history of some chromites is consistent with the mature nature of host metasedimentary rocks (>98 wt.% SiO₂; Cavosie *et al.*, 2004), but at odds with the highly fractured nature of many grains (Fig. 4b-e), which would not survive extensive sedimentary transport. This strongly suggests that fracturing of grains occurred *in-situ*, likely alongside deformation of the host metasedimentary rocks.

Jack Hills detrital chromites contain inclusion assemblages of quartz, fuchsite, rutile, Fe-oxide, monazite, and Fe-sulphide (Fig. 5). These phases, particularly silicates quartz and fuchsite, typically form under distinctly different conditions to magmatic chromite. However, quartz and fuchsite are common components of Jack Hills quartzites and metaconglomerates, alongside minor secondary rutile, monazite and xenotime (Harrison *et al.*, 2007; Rasmussen *et al.*, 2010; Iizuka *et al.*, 2010). Inclusions of quartz and Cr-poor muscovite have also been reported within detrital zircon (Cavosie *et al.*, 2004; Hopkins *et al.*, 2008; 2010; Rasmussen *et al.*, 2011; Bell *et al.*, 2015), both as monomineralic and polyphase inclusions alongside monazite and xenotime (Rasmussen *et al.*, 2011). Monazite and xenotime inclusions within detrital zircons yield ²⁰⁷Pb-²⁰⁶Pb ages of ~2650 Ma and 800 Ma that are demonstrably younger than the host grains (Rasmussen *et al.*, 2011), attesting to a secondary origin of many zircon inclusion assemblages. Cameron *et al.* (2016) additionally showed that ~60 % of quartz inclusions within zircon exhibited evidence of $\delta^{18}\text{O}$ exchange with Jack Hills metasedimentary rocks. It is therefore hypothesised that many quartz inclusions within detrital zircons were significantly altered during metamorphism of host metasedimentary rocks or, alongside

510 muscovite, rutile, monazite and xenotime, were precipitated from post-depositional fluids that
511 filled voids left from the dissolution of primary apatite (Rasmussen *et al.*, 2012; Bell *et al.*,
512 2015; Cavosie *et al.*, 2018; *c.f.* Hopkins *et al.*, 2010; 2012).

513 The occurrence of assemblages of the same phases as observed within detrital zircon,
514 coupled with the unlikelihood of quartz and muscovite stability within the mafic-ultramafic
515 melt that crystallised chromite, provides evidence that inclusion assemblages within chromite
516 are secondary in origin. Interestingly, many secondary inclusions of quartz and fuchsite are
517 apparently isolated from visible fractures within chromites (*e.g.*, Fig. 5a & 8a). Isolation of
518 inclusions has been used as a line of evidence for a primary origin of many quartz and muscovite
519 inclusions within detrital zircon (Bell *et al.*, 2015b). The detection of clearly secondary,
520 apparently isolated inclusion assemblages within detrital chromite suggests caution should be
521 applied when using this as a line of evidence for the presence of primary inclusions, and instead
522 points to the presence of sub-micron or annealed fractures within grains, or fracturing below
523 the 2D surface shown by electron imaging (*e.g.*, Cavosie *et al.*, 2018).

524 Our data therefore indicate that inclusion assemblages formed from interaction of
525 detrital chromite and metamorphic fluids within the metasedimentary host. Pitted and porous
526 domains, which are analogous to spongy textures described elsewhere (Gervilla *et al.*, 2012;
527 Colás *et al.*, 2014) and host the same secondary mineral assemblages, were also likely formed
528 by this process. Coarser, euhedral inclusions may reflect direct dissolution of primary inclusion
529 phases, akin to aforementioned replacement of primary inclusion assemblages within detrital
530 zircon (Rasmussen *et al.*, 2011). It is more problematic to prescribe an origin to inclusions of
531 Fe-sulphide and rutile. Sulphide and chromite are commonly co-liquidus phases under iron-
532 sulphide saturated magmatic conditions; however, secondary pyrite is often observed within
533 Jack Hills metasedimentary rocks (*e.g.*, Cavosie *et al.*, 2004; Supplementary Material) and iron-
534 sulphides (typically $\leq 40\ \mu\text{m}$ pyrite, pyrrhotite, intergrowths of pyrite and pyrrhotite, and rarer

pentlandite) are ubiquitous within isolated quartzite cobbles (Dare *et al.*, 2016). Pyrrhotite, pyrite and rare chalcopyrite inclusions within Jack Hills chromites appear to be closely linked to pitted domains (Fig. 5c); we therefore propose a secondary origin for Fe-sulphide inclusions within chromite.

Inclusions of rutile may be anhedral, globular or preserved as laths aligned to chromite crystallographic axes and are particularly abundant within pitted and porous domains. Observed alignment of alteration phases within chromite (Figs. 4e & 5a) is noted elsewhere, including in chlorite (Fleet *et al.*, 1993; Gervilla *et al.*, 2012) and phlogopite (Rollinson *et al.*, 2002). Whether rutile was precipitated from metamorphic fluids, or was exsolved from chromite, perhaps via changes in the oxidation state of grains (*e.g.*, Cameron *et al.*, 1979), is unclear. High-Ti domains associated with cracks within zircon (Harrison & Schmitt, 2007; Hofmann *et al.*, 2009), and the secondary growth of rutile within metasedimentary matrix (Harrison *et al.*, 2007) attests to Ti mobility within Jack Hills metasedimentary rocks. While we consider a secondary origin of rutile most likely, we note direct exsolution from chromite that previously had higher TiO₂ contents cannot be discounted.

Chromite mineral chemistry

Signatures of metamorphism

Jack Hills detrital chromites typically yield high ZnO and MnO coupled with low Mg# (typically ≤ 30 ; Fig. 6). Chromites with such distinctive chemical compositions are uncommon (*e.g.*, Wylie *et al.*, 1987; Santti *et al.*, 2010; Fanlo *et al.*, 2015) and largely attributed to secondary processes (*e.g.*, Barnes, 2000). A number of studies have investigated the effects of secondary modification of chromite major, minor and trace elements (*e.g.*, Barnes, 2000; González-Jiménez *et al.*, 2009; Mukherjee *et al.*, 2010; 2015; Gervilla *et al.*, 2012; Colás *et al.*, 2014). Hypotheses as to the various processes that modify chromite include: elemental mobility

via interaction with aqueous fluids during the breakdown of surrounding Mg-Fe silicates via serpentinisation (Burkhard, 1993; Marques *et al.*, 2007; Hodel *et al.*, 2017) and/or metamorphism (Barnes, 2000; González-Jiménez *et al.*, 2009; Gervilla *et al.*, 2012; Colás *et al.*, 2014; Fanlo *et al.*, 2015; Ahmed & Surour, 2016), Cu-Zn-Ni-(Co) sulphide mineralisation of the host rock (Wylie, 1987; Marques *et al.*, 2007; Fanlo *et al.*, 2015), and magmatic sulphide mineralisation (Groves *et al.*, 1977). The effects of these processes on chromite chemistry are dependent on the temperature and longevity of alteration, the nature and fO_2 of the aqueous medium, the composition of the host rock and subsequent chromite/silicate ratio, and the fluid/rock ratio during modification (Colás *et al.*, 2014; Ahmed & Surour, 2016). The most commonly observed divalent cation mobility during thermal events is a decrease in Mg#, as Fe^{2+} diffusively enters the chromite lattice at the expense of Mg^{2+} (Barnes, 2000; Colás *et al.*, 2014). ZnO, MnO and to a lesser extent CoO, also diffuse into the chromite lattice in exchange for MgO, NiO and often TiO_2 (Barnes, 2000; Colás *et al.*, 2014).

Within Jack Hills detrital chromites, coupled growth in ZnO content with increasingly homogenised and lowered Mg# likely signifies greater exchange with the modifying medium (Fig. 6a). This may indicate variable modification within their protolith; however, analogous signatures of alteration have been previously reported by Challis *et al.* (1995), where detrital chromites within fuchsitic sedimentary rocks yielded elevated ZnO (up to 13.7 wt.%) and MnO (up to 3.5 wt.%), coupled with low Mg#. Critically, the detrital grains described by Challis *et al.* (1995) possessed a well constrained provenance, demonstrating the observed signatures could only have formed during post-depositional metamorphism.

We propose that the dominant signatures of secondary modification within Jack Hills detrital chromites occurred from exchange of grains with metamorphic fluids and/or assemblages during metamorphism of their metasedimentary hosts. Firstly, as established above, interaction of chromite with metamorphic fluids is shown by replacement of inclusion

assemblages (Fig. 5) and the development of pitted and porous domains. Secondly, large differences in chromite chemistry between samples does not relate to the physical characteristics of grains, such as shape or grain size (Supplementary Figs.), but are a function of sampling site and therefore sedimentary horizon. Homogeneity of Mg# (*e.g.*, 16WA7 & 16WA9; Fig. 6) requires modification under the same physiochemical conditions, including temperature, fO_2 and Mg# of modifying medium, despite the probable variation in sedimentary transport characteristics. Observed Mg# uniformity is therefore difficult to reconcile with metamorphic modification with an ultramafic or mafic protolith.

Finally, and most significantly, chromites yield substantial inter-sample chemical variability despite the proximity (often <5 m) of sampling locations (Fig. 2), with the meter-scale variability of ZnO and Mg# of detrital chromites reconciled by changes in grain size and modal proportions of chromite within the host metasedimentary rocks (14WA1-4; Supplementary Material). Chromites are the sole ZnO bearing phase, but are present at low modal proportions of <1 %. As such, minor changes in the modal proportion and grain size of detrital chromites can account for large variations of wt.% ZnO within detrital chromites during interaction and equilibration with secondary fluids. However, 01JH36 detrital chromites, sampled from a heavy mineral layer within which chromites represent a significant component, yield elevated ZnO (2.02 to 4.74 wt.%) within 01JH36 detrital chromites. Simple changes in the modal proportion of chromite therefore cannot account for the ZnO content of grains within this sample. Instead, 01JH36 chromite mineral chemistry may be reconciled by metamorphic equilibration of ZnO at length scales greater than heavy mineral bands (\geq mm-scale) and, given the distance of this sample from the W-74 site (Fig. 2), variable physiochemical compositions of metamorphic fluids across the Jack Hills. Thus, variability in modification signatures may also reflect changing parameters of metamorphic fluids, as also suggested by the often poor correlation between Mg# and MnO with the modal proportion of detrital chromite.

Oxygen isotope compositions

In-situ modification of chromites is also suggested by oxygen isotope compositions. The variation of $\delta^{18}\text{O}$ with grain size in chromites from samples 01JH36 and 01JH42 (Fig. 10a-b) indicates they experienced partial exchange of oxygen isotopes with the metasedimentary host rocks. If exchange was by diffusion inwards from the grain boundary, then greater modification of smaller grains is expected given their greater surface area to volume ratio. The lower $\delta^{18}\text{O}$ values of the smallest chromite grains from 01JH36 and 01JH42 may therefore approach equilibration with host quartz during metamorphism or retrogression. In contrast, chromite grains in sample 01JH54 show no grain size dependence on $\delta^{18}\text{O}$ (Fig. 10c). Given the conspicuously lower chromite $\delta^{18}\text{O}$ values recorded in 01JH35 and 01JH36, we tentatively interpret chromite $\delta^{18}\text{O}$ values in sample 01JH54 as pre-depositional. Low $\delta^{18}\text{O}$ ($\leq 2.5\text{--}3.0\text{‰}$) are reported within chromitite from ophiolites and layered intrusions due to exchange with late-stage magmatic and sub-solidus fluids (Lowry *et al.*, 2003; Schannor *et al.*, 2018).

The oxygen isotope ratios for smallest chromite size fractions in samples 01JH36 and 01JH42 were compared with previously published $\delta^{18}\text{O}$ values for quartz in the same samples (Supplementary Table 2; Cavosie *et al.*, 2005). Cavosie *et al.* (2005) found that values of $\delta^{18}\text{O}(\text{Qz})$ are homogeneous at hand sample scale in these rocks, indicating metamorphic equilibration of quartz in pebbles and matrix. If attainment of oxygen isotope equilibrium between quartz and chromite is assumed, the $\Delta^{18}\text{O}$ value ($\Delta^{18}\text{O}_{\text{Quartz-Chromite}} = \delta^{18}\text{O}_{\text{Q}} - \delta^{18}\text{O}_{\text{C}}$) for samples 01JH36 ($\Delta^{18}\text{O}_{\text{Q-C}} = 12.02\text{‰}$) and 01JH42 ($\Delta^{18}\text{O}_{\text{Q-C}} = 12.70\text{‰}$), can be applied to calculate the temperature at which the observed fractionations were established (Zheng *et al.*, 1991). Calculated temperatures are 412 °C and 390 °C for samples 01JH36 and 01JH42, respectively. Sample 01JH54 yields a distinctly smaller fractionation ($\Delta^{18}\text{O}_{\text{Q-C}} = 9.11\text{‰}$) that would correspond to a temperature of 530 °C if it represented metamorphic equilibration.

However, as discussed above, the analysis of different sized grains suggests that chromite grains in this sample preserve pre-depositional $\delta^{18}\text{O}$ from and may not be appropriate for metamorphic thermometry. While it cannot be demonstrated that oxygen isotope equilibrium was established between quartz and chromite based on $\delta^{18}\text{O}$ analysis of 2 mg aliquots in the other two samples, we note that the derived temperatures based on measured fractionations are consistent with greenschist facies metamorphic conditions previously documented within the Jack Hills (Cavosie *et al.*, 2004; Spaggiari *et al.*, 2007b).

Quartz in the metasedimentary rocks analysed has been shown to have recrystallized during metamorphism, and yields homogeneous $\delta^{18}\text{O}$ values at >cm scales (Cavosie *et al.*, 2005). In contrast, detrital chromites in the same samples preserve detrital morphologies, likely preserve gradients in $\delta^{18}\text{O}$ at sub-mm scale (Fig. 10), and are not equilibrated in $\delta^{18}\text{O}$. We therefore use the $\delta^{18}\text{O}$ values determined for quartz (Cavosie *et al.*, 2005) and temperatures determined from oxygen isotope thermometry to estimate the $\delta^{18}\text{O}$ value of fluids present during metamorphism (Supplementary Table 3). Calculated water $\delta^{18}\text{O}$ values using quartz-water fractionation factors of Sharp *et al.* (2016) and the published quartz $\delta^{18}\text{O}$ values yields water $\delta^{18}\text{O}$ values of 8.58 ‰ (at 412 °C) for samples 01JH42 and 8.16 ‰ (at 390 °C) for sample 01JH36; a value of water $\delta^{18}\text{O}$ of 8.50 ‰ (at 530 °C) for sample 01JH54 is comparable to the other two samples, but may be fortuitous given the lack of evidence for equilibration in sample 01JH54 (Fig. 10). Regardless, the range of all calculated water $\delta^{18}\text{O}$ values, from 8.2-8.6 ‰, are typical for magmatic or metamorphic fluids (Sheppard, 1986).

Trivalent cations: retained primary signatures?

While divalent cations exchange more readily during metamorphism, trivalent cations appear largely immobile during low temperature modification (*e.g.*, Barnes, 2000). Chromite interacting with acidic fluids during sea-floor hydrothermal metasomatism may rarely induce

trivalent cation exchange (Wylie *et al.*, 1987; Marques *et al.*, 2007; Hodel *et al.*, 2017); more generally, however, nascent trivalent mobility occurs at greenschist facies metamorphism (Kimball *et al.*, 1990; González-Jiménez *et al.*, 2009) and becomes increasingly pervasive at and above amphibolite facies (Barnes, 2000; González-Jiménez *et al.*, 2009; Colás *et al.*, 2014). Under oxidising conditions, trivalent cation exchange most commonly facilitates the transition of chromite to ferritchromit [$\text{Fe}^{2+}(\text{Fe}^{3+}, \text{Cr})_2\text{O}_4$], where Al_2O_3 is lost to Fe_2O_3 either diffusively (Wylie *et al.*, 1987; Gervilla *et al.*, 2012; Colás *et al.*, 2014) or via reaction with magnetite rims (Evans & Frost, 1975; Barnes, 2000). Extensive formation of ferritchromit and Cr-magnetite, and thus modification of chromite core compositions, appears to be largely restricted to amphibolite facies metamorphism and above (Barnes, 2000).

Regardless of the degree of secondary modification of divalent cations, Jack Hills detrital chromites possess a broad but consistent range in Cr# across all metasedimentary sample locations (Fig. 7). The range of Cr# may be explained by either the absence of significant trivalent mobility, or the equilibration of chromite during secondary modification with a Cr-Al bearing equilibrant (*e.g.*, fuchsite or metamorphic fluid). If secondary exchange was significant for trivalent cations, the latter scenario should manifest as homogenisation to lower or higher Cr# in increasingly Zn-rich samples. This is not observed. Additionally, zonation profiles of variable Mg# but homogeneous Cr# within low ZnO chromites (Fig. 8a) would require more exchange, and thus faster diffusion, of trivalent than divalent cations. Such an order of relative diffusivities is in strong contrast to previous literature (*e.g.*, Barnes, 2000), and considered extremely unlikely.

Unmodified Cr# within chromites is at apparent odds with the presence of surrounding fuchsite, whose elevated Cr content is almost certainly derived from detrital chromites. Fuchsite yields Cr_2O_3 concentrations of ~2-3 wt.% (Cr# <5; Supplementary Table 4), that decrease with increasing distance from detrital chromite grains (*e.g.*, Rasmussen *et al.*, 2011). Such low

concentrations within fuchsite require only minor mobilisation of Cr from chromite, and suggest that the Cr required was scavenged from more heavily altered areas of chromite during metamorphic precipitation of fuchsite, or from dissolution of strongly modified grains in acidic fluids. This is supported by lobate boundaries of detrital chromites with surrounding fuchsite (Fig. 9a-b). Selection of grains with well-defined shapes has likely filtered our dataset for such heavily modified compositions; the process of fuchsite formation causes ragged grain boundaries and extensive fracturing of chromites, which significantly modifies grain morphology (Fig. 3).

Localised zones of Cr-Al mobility may, however, shown by high Cr# domains (Fig. 9), with their generation reconciled via preferential loss of Al during localised exchange between chromite and fuchsite and/or the metamorphic fluid that precipitated fuchsite. High Cr#, low Fe³⁺# and inclusion free zones of alteration have previously been documented by Arai *et al.* (2006), who suggested such domains represent the loss of Al³⁺ without concurrent addition of Fe³⁺. Alternatively, high Cr# domains may reflect compositions transitional to ferritchromit or fluid-mediated recrystallisation, although further microstructural analysis would be required to show this. The distinct boundary between the bulk chromite and high Cr# domains (Fig. 9) was suggested by Arai *et al.* (2006) to represent a spinel miscibility gap at high-Cr, Low-Fe³⁺ conditions. Importantly, the sharp gradients indicate negligible diffusion between high Cr# domains and bulk chromites. Therefore, while high Cr# domains clearly represent areas of elevated alteration during metamorphism of host metasedimentary rocks, the retention of sharp boundaries implies an absence of trivalent cation mobility within bulk chromites.

High Cr# domains yield strong compositional similarities to high Cr# chromite, which creates increased uncertainty as to the primary origin of high Cr# grains. However, inter-grain variability in Mg# is still apparent within 14WA2, and the zonation profile of high Cr# 14WA2-PB-12 (Cr# ~75) shows Mg# variability (Supplementary Material Fig. XX); if elevated Cr#

were a product of enhanced trivalent mobility, Mg# heterogeneity should not be observed. This, coupled with the sharp boundaries of high Cr# domains, provides strong evidence that chromite with consistently high and homogeneous in Cr# are pre-depositional in origin.

Timing and cause of chromite modification

It is apparent from the mineral chemistry and inclusion assemblages of Jack Hills detrital chromites that grains were modified within host metasedimentary rocks. However, the timing and magnitude of metamorphic events within the Jack Hills are poorly understood (*e.g.*, Kemp *et al.*, 2018). Previous work has attested to three thermal or fluid flow events within Jack Hills metasedimentary rocks at ca. 2650 Ma, ~1850-1800 Ma and 800 Ma (Spaggiari, 2007a; Rasmussen *et al.*, 2010; 2011; Kemp *et al.*, 2018). Below, we deliberate the timing and magnitude of metamorphic events experienced by Jack Hills detrital chromites.

Dominant metamorphic signatures; a ~2650 Ma event

Metamorphism at ~2650 Ma has been proposed to represent peak metamorphic conditions at the W-74 site (EARLIER PAPER; Rasmussen *et al.*, 2010; 2011). While inclusion assemblages within Jack Hills chromites have not been dated, and so cannot be definitively linked to this event, previous investigation of secondary monazite and xenotime within detrital zircons from the W-74 site observed a large population of this age (Rasmussen *et al.*, 2011). Monazite-xenotime Gd exchange and Ti-in-quartz thermometry indicates metamorphic temperatures of <487 °C at this time, typical of upper greenschist facies metamorphism (Rasmussen *et al.*, 2011). Importantly, temperatures calculated for oxygen isotope equilibrium between chromite and quartz fall within the temperatures anticipated for greenschist facies metamorphism, and also argue for exchange at this time. Given that the ~2650 Ma event represents the highest grade of metamorphism experienced by metasedimentary rocks local to the W-74 site, this event

likely imposed the observed chromite mineral chemistry. The chemical compositions of Jack Hills chromites can therefore be explained by partial to complete exchange with Fe-, Zn- and Mn-bearing metamorphic fluids and mineral assemblages during upper greenschist to lower amphibolite facies metamorphism at ca. 2650 Ma. Metamorphic fluids were likely acidic to facilitate the mobility of Ti^{4+} (e.g., van Baalen, 1993), and enable localised dissolution of chromite to form fuchsite.

This event also coincides with amphibolite to granulite metamorphism throughout the Narryer Terrane, including gneisses adjacent to the belt, and ubiquitous granitic emplacement within the entire Yilgarn Craton (Kemp *et al.*, 2018). This includes the ca. 2650 Ma monzogranite ('The Blob granite'; Pidgeon & Wilde, 1998), which intrudes the Jack Hills belt to the SW of Eranondoo Hill (Fig. 2). Speculatively, this intrusion, or other granitic intrusions adjacent to the Jack Hills belt, may have supplied heat and a metasomatic component (particularly volatile Zn) to metamorphic fluids; similarly modified detrital chromites reported by Challis *et al.* (1995) were postulated to have gained their chemical compositions by interaction with fluids derived from local granitic plutons. An alternative and more local source of ZnO within Jack Hills metasedimentary rocks may come from dissolution of sphalerite (ZnS), which may be a potential detrital phase given the abundance of Fe-sulphides within quartzite cobbles.

Studies have shown that metasedimentary rocks with Proterozoic depositional ages are complexly intercalated with, and at the same apparent metamorphic grade, as metasedimentary associations with inferred Archean depositional ages (Cavosie *et al.*, 2004; Dunn *et al.*, 2005; Grange *et al.*, 2010; Wang & Wilde, 2018). This, coupled with ~1850-1800 Ma and ~800 Ma ages within the Jack Hills (Spaggiari *et al.*, 2007b; Rasmussen *et al.*, 2010; 2011), indicates Proterozoic overprinting at a lower metamorphic grade than peak metamorphic conditions.

Subtle zonation trends within some chromites, namely increased Mg# within 14WA4 (Fig. 8b), may be reconciled by minor elemental mobility (Mg, Fe, Zn) within the Proterozoic.

Pre-depositional modification?

Given the susceptibility of chromite to sub-solidus exchange, signatures of modification prior to deposition must also be considered. Divalent cations have undergone significant exchange within host metasedimentary rocks; it is unclear if preservation of high Mg# within the cores of some low ZnO grains (e.g., 14WA2-PB-46; Fig. 6b) represent pre-depositional Mg#, or have still undergone partial exchange. However, pre-depositional metamorphism of accessory chromites (e.g., within their protolith), should result in trivalent cation mobility namely the development of ferritchromit rims (Barnes, 2000; Colás *et al.*, 2014). While the generation of ferritchromit is limited within chromitites, where the higher modal abundance of chromite in chromitite has been shown to inhibit significant metamorphic exchange and thus the generation of ferritchromit (e.g., Mukherjee *et al.*, 2010), chromitite typically represent relatively minor components of igneous bodies (Barnes & Roeder, 2001). Given the presence of EO, which have undergone little abrasion and rounding during sedimentary reworking, ferritchromit or magnetite rims should be clearly visible if present prior to deposition. No ferritchromit has been observed within this study, providing evidence chromites have not undergone resolvable trivalent mobility prior to erosion from their protoliths. While we cannot definitively exclude trivalent cation mobility prior to deposition, we suggest that Cr# of Jack Hills detrital chromites represent the most robust chemical signature for investigation of provenance.

Provenance

While detrital chromites exhibit protracted metamorphic histories, some chemical signatures place constraints on the magmatic protolith. Consistent variability of Cr# regardless

of shape may indicate a common mafic or ultramafic protolith. While we base this interpretations on least altered chromites within samples 14WA2 and 01JH54, the consistent range of chromite Cr# from all 14WA, 16WA and 01JH metasedimentary rocks analysed potentially suggests that a single, large-scale magmatic source potentially dominated the mafic and ultramafic crust within the catchment of the Jack Hills sedimentary rocks. Unfortunately, important chemical tools for understanding the petrogenesis of chromite and consequently their protolith, namely Mg# and TiO₂, are not appropriate to apply to Jack Hills detrital chromites due to modification within host metasedimentary rocks. Therefore, commonly used plots for chromite provenance, such as TiO₂ vs Cr₂O₃ or Al₂O₃ (Kamenetsky *et al.*, 2001), are unlikely to be reliable. We propose trivalent cation mobility was limited within Jack Hills detrital chromites, but cannot fully exclude metamorphic re-equilibration of Fe³⁺/ΣR³⁺ prior to deposition. We therefore suggest careful use of Fe³⁺/ΣR³⁺ may provide better insights into the provenance of Jack Hills chromites; unmistakably non-stoichiometric grains (*i.e.*, Fe³⁺ = <0) from 14WA2 are excluded Fig. 13 for clarity. The retention of unmodified Cr# within cores of Jack Hills chromite will therefore yield the most information on provenance of detrital grains, alongside careful and conservative use of Fe³⁺/ΣR³⁺ and, where rutile laths are demonstrably absent, TiO₂ contents.

To gain understanding of the petrogenesis of Jack Hills detrital chromites, and potentially of mafic and ultramafic crust within the Narryer Terrane, we use the fields of Barnes & Roeder (2001), a comparative study of over 26,000 spinels from different tectonic settings. We compare these fields to grains from 14WA2 and 01JH54, which, due to low ZnO coupled with partial exchange of Mg#, we interpret to represent the least altered compositions of detrital chromite. We do not include the chemical compositions of chromitites within the fields of Figs. 10-13; chromitites compromise a low volumetric component of igneous bodies, particularly layered intrusions, and have been suggested to induce bias within provenance studies (Power

809 *et al.*, 2000; Barnes & Roeder, 2001). While 14WA2 appear conform to a Cr-Al trend (Fig. 11;
810 Irvine, 1967; Barnes & Roeder, 2001), where no enrichment of Fe^{3+} is observed with changing
811 Cr#, this likely reflects non-stoichiometry. The consistently low ferric iron contents of 01JH54
812 chromites are not suggestive of Fe-Ti or Rum trends (Fig. 11), which form via the reaction of
813 plutonic chromites with intercumulus liquid enriched in both Fe^{3+} and Ti.

814 As expected, in Cr# vs. $\text{Fe}^{2+}\#$ (inverse Mg#) compositional space, Jack Hills detrital
815 chromites fall within the 90th percentile of spinel within high grade metamorphic rocks (Fig.
816 12a). The decrease of Mg# within Jack Hills detrital chromites due to metamorphic exchange
817 results in comparable Mg# to chromites modified within mafic-ultramafic protoliths; this
818 process is also shown in Fig. 12d, where an increase in metamorphic grade from greenschist to
819 amphibolite facies results in significantly lowered and homogenised Mg# within komatiitic
820 chromites. While superficially Jack Hills detrital chromites yield a similar Cr# variability to
821 those modified by metamorphism, the range in Cr# shown within the metamorphic field (Fig.
822 12a) represents compositions transitional from magmatic Cr# to ferritchromit and magnetite.
823 As such alteration phases are absent within Jack Hills chromites, and zonation profiles show
824 limited Cr# variability, this process cannot explain the fit of grains to the metamorphic field of
825 Barnes & Roeder (2001). A different provenance is therefore required for the observed Cr#.

826 The variable and moderate to high Cr# of Jack Hills detrital chromites (typically 55-70)
827 are comparable to ophiolitic chromites (Fig. 12b), continental layered intrusions (Fig. 12c), and
828 intra-plate settings such as ocean island tholeiites and oceanic plateaus (Dick & Bullen, 1984;
829 Barnes & Roeder, 2001; Arai *et al.*, 2011). Critically, the Cr# of Jack Hills chromites indicate
830 the majority of detrital grains cannot be derived from komatiites (Fig. 12d). As a product of
831 high temperature and high degree mantle melts, komatiites have high Cr/Al ratios and therefore
832 yield chromite with high Cr# (typically >70; Barnes & Roeder, 2001) that yield tightly clustered
833 compositions due to the inhibition of magmatic fractionation or subsolidus equilibration. In

comparison to both Al-undepleted komatiites (AUDK; Fig. 12d) and Al-depleted komatiites (ADK), Jack Hills detrital chromites yield more variable and moderate Cr# (typically 55-70). ADK are not shown in Fig. 12 as they yield chromite with too high Cr# (>85) to represent a source of Jack Hills detrital chromite. Some portions of komatiitic lavas, such as olivine-rich dunitic channels and sheets, have been shown to yield chromite with lower Cr# (60-70) than bulk ADK and AUDK (Barnes, 1998; Barnes & Roeder, 2001). However, chromite is not abundant in these reduced cumulates, and would be unlikely to dominate the detrital record over their bulk komatiitic counterparts.

We suggest that, on the basis of Cr#, the most likely provenance of Jack Hills detrital chromites is a layered intrusion. Notably, though chromitite fields are not included in Fig. 12 and Fig. 13, Jack Hills detrital chromites yield comparable Cr# to Stillwater 'G' and Bushveld chromitites (*e.g.*, Campbell & Murck, 1993; Langa et al., 2020 and references therein). While variable chromite morphologies may reflect different transport mechanisms, an absence of chemical variation with rounding shape of grains may be further qualitative evidence of a potential layered intrusion source; layered intrusions may yield consistent lithologies and mineral chemistry laterally for many kilometres. In the absence of definitive Fe₂O₃ or TiO₂ concentrations, tectonic settings with comparable Cr#, including podiform and other mantle derived Cr-spinel, cannot be excluded (as shown by ophiolites; Fig. 13a). Thus, while we consider a layered intrusive source the most likely source of Jack Hills detrital chromites, further investigation of additional chemical signatures may provide evidence for or against this interpretation.

However, chromite from early Archean anorthositic layered intrusions yield marked physiochemical differences to those from stratiform complexes, which dominate the layered intrusion fields of Fig. 12 and Fig. 13. Archean anorthositic layered intrusions typically form sill-like bodies that are significantly less volumetric than Proterozoic massif-type anorthositic

intrusions, and are thought to represent sub-volcanic intrusions coeval to overlying greenstone belts (*e.g.*, Ashwal & Bybee, 2017). Archean sill-like anorthositic intrusions yield spinel that are typically Fe rich, with populations of high Cr# (>70) or low Cr# (<50) (Rollinson *et al.*, 2002; 2010; Mondal *et al.*, 2006; Dharma Rao *et al.*, 2013; Mukherjee *et al.*, 2015; Rowe & Kemp, 2020), thought to reflect derivation from parental melts of komatiitic, boninitic or hydrous basaltic compositions (Ashwal & Bybee, 2017). Thus, typical mineral chemistry of anorthositic layered intrusive chromites are not comparable to Jack Hills chromites. While chromites extending to Cr# comparable to Jack Hills detrital chromites are documented (*e.g.*, Mukherjee *et al.*, 2010), they typically represent a minor component of bulk chromite compositions. As with chromites from komatiitic dunitic channels, such chromites are unlikely to dominate detrital records over higher Cr# counterparts.

While Jack Hills detrital chromites do not yield Cr# equivalent to Cr-spinel from sill-like Archean anorthositic layered intrusions, chromite from late Archean layered intrusions with comparable chemical compositions are known (*e.g.*, Rollinson *et al.*, 2010; Berger *et al.*, 2013; Szilas *et al.*, 2017). The development of massif-type anorthositic intrusions at the beginning of the Proterozoic are thought to coincide with the emergence of increasingly thickened and strengthened lithosphere, a consequence of secular cooling and potentially shifting geodynamic regimes (Ashwal & Bybee, 2017). This therefore does not exclude the presence of more massif-type layered intrusions during the early Archean; the preservation potential of larger, and thus higher density, stratiform intrusions would be significantly lower in the presence of hotter Archean mantle.

Potential sources within the Narryer Terrane

As the mineral chemistry of Jack Hills detrital chromites allude to a layered intrusion origin, and unpublished Re-Os isotopic data yield model ages that are at least Palaeoarchean in age

(Valley *et al.*, 2005), the sole described layered intrusion within the Narryer Terrane, the 3730 Ma Manfred Complex (Kinny *et al.*, 1988; Fletcher *et al.*, 1988; Myers, 1988b; Rowe & Kemp, 2020), can be considered as a potential source. Spinel within the Manfred Complex is dominantly spinel *sensu stricto* or picotite, and occurs closely associated with magmatic olivine (Rowe & Kemp, 2020) or exsolved from cumulus orthopyroxene or clinopyroxene. Chromite is completely absent in the Manfred Complex harzburgites and websterites studied by Rowe & Kemp (2020); however, metaperidotites within the Manfred Complex contain chromite (Fletcher *et al.*, 1988; Myers, 1988b; Kemp *et al.*, 2018). Here, we compare the least altered 14WA2 and 01JH54 detrital grains with chromites from Manfred Complex sample 13TKN22 (Supplementary Material), a pyroxene-phyrlic peridotite metamorphically modified to hornblende and altered to serpentine.

13TKN22 chromites (n=13 analyses) occur as disseminated crystals with variably formed magnetite rims (Fig. 14a-b), and yield consistent Cr# of 76-80 (Table 2), coupled with low Mg# of <20, likely reflecting metamorphism of the host metaperidotite. 13TKN22 chromites yield limited ZnO (~0.6 wt.%) and heterogeneous MnO (0.4-1.4 wt.%). NiO is present at magmatic concentrations of >0.1 wt.%, but this may reflect the high magnetite component of 13TKN22 chromites (Barnes, 1998). While divalent cation mineral chemistry suggests metamorphic alteration of 13TKN22 chromites, the sharpness of the boundary between chromite cores and magnetite rims (Fig. 14a-b) and the tightly clustered compositions of cores (Fig. 14d-e) indicate restricted trivalent mobility (*e.g.*, Barnes, 2000). While the highest Cr# Jack Hills detrital grains overlap with the Cr# of 13TKN22 chromites (Fig. 14), detrital chromites possess significantly lower ferric iron contents and TiO₂, though given the uncertainties introduced by Ti mobility within host metasedimentary rocks comparisons should be taken tentatively. 13TKN22 chromites therefore cannot account for the bulk population of Jack Hills chromite detrital grains, which possess lower Cr#, and these lithologies did not

represent the dominant component of mafic to ultramafic crust within the erosional catchment of Jack Hills metasedimentary rocks.

While mafic and ultramafic crust within the Narryer Terrane tends to be collectively grouped as the Manfred Complex, more recent geochronological investigations have shown many of these lithologies actually have Paleoproterozoic igneous crystallisation ages (Kemp *et al.*, 2018). Continued investigation of these units may reveal further insights into the provenance of Jack Hills detrital chromites. Mafic and ultramafic rocks that occur within the Jack Hills belt may also represent the source for Jack Hills detrital chromites, particularly given the proximal source suggested for EO. Jack Hills mafic and ultramafic rocks are heavily sheared and extensively recrystallised (Spaggiari, 2007a), however analysis of one sample (16WA13; Supplementary Material) found relict chromite cores within texturally distinct chromite stringers. Whether this texture is igneous or deformational in origin is unclear, but the development of thick and indistinct magnetite and ferritchromite rims (Fig. 14c-d) yields evidence for at least amphibolite facies metamorphism (*e.g.*, Barnes, 2000), likely representing amphibolite facies metamorphism of the Jack Hills belt prior to deposition of Unit 3 and 4 metasedimentary rocks (Spaggiari *et al.*, 2007a/b). While 16WA13 relict cores possess comparable Cr# to some Jack Hills detrital grains, gradational and lobate boundaries between relict cores and ferritchromite-magnetite rims, and their compositional trend towards magnetite (Fig. 14) are evidence for mobility of trivalent cations. This indicates that the Cr# of 16WA13 chromite cores represent maximum values, and that comparisons should be taken cautiously. However, further analyses of Jack Hills mafic and ultramafic rocks, particularly those preserved within lower strain areas, may prove more fruitful.

CONCLUSIONS

We have conducted a detailed textural and chemical investigation of detrital chromite from the Jack Hills, Western Australia, in a bid to expand beyond the intensively studied zircon record and provide additional perspectives on sources of sedimentary detritus (Fig. 15). Detrital chromites have undergone modification during metamorphism of host metasedimentary rocks, with significant mobility of divalent cations. This is shown by increasingly homogenised and lowered Mg#, elevated ZnO and MnO, and partial to complete exchange of oxygen isotopes. Interaction with metamorphic fluids also resulted in the generation of pitted domains and replacement of primary mineral assemblages with metamorphic phases present in the matrix of host metasedimentary rocks, dominantly quartz and fuchsite. We propose metamorphic modification coincided with peak upper greenschist to lower amphibolite facies metamorphism of Jack Hills metasedimentary rocks at ~2650 Ma.

Despite the development of fuchsite and high Cr# domains, Jack Hills chromites yield a consistent range in Cr# across all samples and have limited intra-grain variability, indicating the Cr# of chromites cores are robust. An absence of ferritchromite rims on euhedral grains further argues against significant trivalent mobility prior to deposition. Critically, the moderate and variable Cr# of most detrital chromites suggest that komatiites or associated intrusives were not a significant component of mafic-ultramafic crust within the erosional catchment of Jack Hills metasedimentary rocks. We propose a layered intrusive origin for Jack Hills detrital chromites, though note that in the absence of further constraints from ferric iron and TiO₂ contents other sources cannot be excluded. It is currently unclear whether detrital chromites represent the remnants of a hitherto unknown intrusion, or were sourced from the 3730 Ma Manfred Complex or from ultramafic horizons within the Jack Hills belt. The lack of significant komatiite contribution may have geodynamic implications for eruption efficiencies within the catchment of Jack Hills sediments (*e.g.*, heat-pipe vs. ‘plutonic squishy lid’; Moore & Webb,

2013; Rozel *et al.*, 2017), and more broadly, the tectonic distribution of mafic and ultramafic crust within the Archean.

While this study focused on Jack Hills detrital chromites, we have demonstrated that with a careful and detailed approach, characteristics of the igneous precursors of detrital chromites may be determined. However, we show that identification of key physiochemical signatures of chromite modification are critical to assess the veracity of interpretations derived from detrital chromites preserved within Archean metasedimentary rocks. Future studies of Archean detrital chromites should ideally be restricted to mature metasedimentary units that have undergone at most lower amphibolite facies metamorphism to minimise trivalent cation mobility. Where alteration of key physiochemical signatures can be shown to be absent, Cr#, Fe₂O₃ and TiO₂ contents may be used to provide significant insights into the provenance of mafic or ultramafic protoliths. Fuchsitic metasedimentary rocks are commonly reported within Archean terranes (*e.g.*, Randive *et al.*, 2015), indicating that the eroded remnants of mafic and ultramafic crust are abundant. Detrital chromites within fuchsitic metasedimentary rocks may therefore represent an innovative tool in understanding the distribution of eroded Archean mafic and ultramafic crust, and yield important insights into geodynamic regimes of the early Earth.

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1325

Detrital chromite from Jack Hills, Western Australia: signatures of metamorphism and constraints on chromite provenance

Figure captions

Fig. 1: Map of the Yilgarn Craton, Western Australia, showing the position of the Narryer Terrane, Jack Hills and Mount Narryer. Modified from Kemp *et al.*, 2018 with Terrane boundaries after Cassidy *et al.* (2006).

Fig. 2: (a) Simplified geological map of Jack Hills and surrounding quartzofeldspathic gneiss and granitoids with sampling locations, after Spaggiari *et al.* (2007b) and Pidgeon & Wilde (1998). (b) Higher resolution geological map of sampling locations at and around the W-74 discovery site (see Fig. 2a). Modified after Spaggiari *et al.* (2007b). See Methods and Materials for relative positions of samples from the W-74 site.

Fig. 3: Jack Hills detrital chromite. (a) Optical microscope image illustrating *ex-situ* variably rounded morphologies of chromite grains. Note the green-brown fuchsite. (b) BSE image of chromite observed *in-situ* within quartzite cobble. Despite isolation from metaconglomerate matrix the cobble has been infiltrated by secondary Cr-muscovite. (c) BSE image of complex and ragged boundaries between chromite and fuchsite within metaconglomerate matrix. Grain heavily fractured on its southern margin. Chr – chromite, Qz – quartz, Fch – fuchsite. Iron oxides are likely magnetite, but have not been analysed by EPMA.

Fig. 4: BSE images showing chromite textural variation. All scale bars 50 μm . (a) Smooth with minimal fracturing, commonly associated with high Cr#. (b) Polygonal fracturing, no pitted domains. (c) Fractured with very fine pitted domains. Note coarser pores, with pitted domains localised to edges and near fractures. (d) Fractured with increasingly pitted textures, often away from obvious fractures. (e) and (f) porous chromite, fractures apparently lost or in-filled by secondary material. Secondary material often aligned to crystallographic axes. Qz – quartz, Fch – fuchsite and Rt- rutile. Iron oxides are likely magnetite, but have not been analysed by EPMA.

Fig. 5: BSE images of inclusion assemblages. (a) Isolated euh-subhedral quartz (Qz). Brighter crystallites in quartz are rutile, an Mg-silicate, and phosphate. Note the surrounding pitted domain, including darker quartz and rutile (Rt) laths, but absence of cracks. (b) Isolated polyphase assemblage of quartz and iron oxide. (c) Subhedral Fe-sulphide within pitted domain of chromite; pyrrhotite (Po) and chalcopyrite (Ccp) (outlined in red). (d) Euhedral rutile within quartz; note the ragged boundaries of quartz and surrounding pitted domain. (e) Anhedral inclusion of fuchsite. (f) Globular inclusion of quartz and fuchsite with monazite. Note the expansion cracks radiating from inclusion.

Fig. 6: Variation of Mg#, ZnO, MnO and TiO₂. Each data point is separated by sample and represents a core or near-core composition of chromite. Slightly elevated Mg# of 01JH chromites in comparison to those from 14WA and 16WA is a function of higher ferric iron contents. (a) ZnO vs Mg#. (b) MnO vs Mg#. (c) TiO₂ vs Mg#. Most grains yield low TiO₂ of <0.25 wt.%. It is likely that some analyses above this value overlapped with subsurface rutile laths. However, some grains with >0.25 wt.% TiO₂ show no evidence of rutile laths. TiO₂ >1 wt.% clearly represent overlap with rutile laths, so are omitted. (d) MnO vs ZnO.

1373

1374 Fig. 7: Variation of Cr#. (a) ZnO vs Cr#. No variation in the range of Cr# despite apparent
1375 increasing ZnO. (b) Cr# vs $\text{Fe}^{2+}\#$ (100x molar $\text{Fe}^{2+}/(\text{Fe}^{2+}+\text{Fe}^{3+})$). (c) TiO_2 concentrations vs
1376 Cr#. Scatter towards 1 wt.% a mixture of high TiO_2 grains and likely overlap of rutile laths.
1377 High Cr# grains appear to be slightly elevated in TiO_2 in comparison to bulk lower Cr# grains.
1378 (d) $\text{Fe}^{3+}/\Sigma\text{Fe}$ vs Cr#. Significant non-stoichiometry of 14WA and 16WA chromites has resulted
1379 in markedly lower $\text{Fe}^{3+}/\Sigma\text{Fe}$ in comparison to 01JH chromites.

1380

1381 Fig. 8: Examples of zonation patterns observed within 14WA and 16WA Jack Hills detrital
1382 chromites, including BSE images of line path taken. Isolated, globular inclusions of fuchsite
1383 (fch) also highlighted in 14WA2-PB-46. (a) Decreasing Mg# towards rims: most commonly
1384 shown by low ZnO samples 14WA2 and 16WA5. 14WA2-PB-46 yields the most variable
1385 zoning patterns observed in all Jack Hills grains and is likely the least altered grain analysed.
1386 This example also includes elevated Fe_2O_3 at its rims, resulting in an area of slightly elevated
1387 Cr#. (b) Elevated Mg# towards rims: commonly observed in high ZnO samples such as
1388 14WA3-4 and 16WA7. ZnO and Mn are lost from chromite at the expense of FeO and MgO,
1389 increasing the Mg# of rims relative to the core of the grain. (c) Largely homogeneous Mg#. No
1390 change in mineral chemistry across the grain in divalent or trivalent cations; this example has a
1391 slight elevation in Mg# on one edge after a minor decrease from the core.

1392

1393 Fig. 9: Variation in mineral chemistry: BSE images with elevated contrast to variable high Cr#
1394 domains. (a) and (b) High Cr#, pit-free domains at the edge of grains, and (c) along or associated
1395 with cracks. (d) and (e) High Cr# domains bound by chromite crystallographic axes, closely
1396 associated with laths of fuchsite (fch) and anhedral quartz (qz) along margins of high Cr# zones

and rutile (rt). (f) The only observed example of a domain of lowered Cr#, encircling an inclusion of quartz with iron oxide rims. Note the radial fractures associated with the secondary inclusion. Except where stated, all scale bars are 50 μm .

Fig.10: Oxygen isotope ratios ($\delta^{18}\text{O}$ reported at ‰ variation to V-SMOW) for different size fractions of Jack Hills detrital chromite grains measured by laser fluorination. $\delta^{18}\text{O}$ uncertainties (2σ) are ± 0.08 for 01JH36, ± 0.12 for 01JJ42 and ± 0.06 for 01JH54. (a) and (b) Samples 01JH36 and 01JH42, which show strong evidence of grain size dependency on $\delta^{18}\text{O}$ values. (c) Sample 01JH54, showing no evidence of grain size dependency on $\delta^{18}\text{O}$ values.

Fig. 12: Cr^{3+} - Al^{3+} - Fe^{3+} triangular plot, modified from Barnes & Roeder (2001), showing three magmatic compositional pathways for spinel: Cr-Al, Fe-Ti, and Rum trends. The compositional space of ferritchromite, Cr-magnetite, and spinel miscibility gaps shown for reference. Core or near core compositions of low ZnO 14WA2 and 01JH54 chromites are shown.

Fig. 12: Provenance: Cr# vs $\text{Fe}^{2+}\#$ ($100 \times \text{Fe}^{2+} / (\text{Fe}^{2+} + \text{Mg}^{2+})$), diagrams modified from Barnes & Roeder (2001). (a) Strong fit of 14WA2 detrital chromite with chromite that have undergone high-grade metamorphism due to exchange of Mg^{2+} for Fe^{2+} , and thus increased $\text{Fe}^{2+}\#$. (b) Fit of bulk field for chromite from ophiolites, although Jack Hills chromites plot at considerably higher $\text{Fe}^{2+}\#$ due to metamorphic exchange of Mg^{2+} for Fe^{2+} . (c) Strong fit of chromite Cr# in Jack Hills chromite with chromite from continental layered mafic intrusions. Jack Hills detrital chromite $\text{Fe}^{2+}\#$ are again largely higher than 50th percentile due to metamorphic exchange. (d)

1419 Distinct absence of fit with chromite from Al-undepleted (AUD) komatiites. GSF; Greenschist
1420 facies AUD komatiites, AF; Amphibolite facies AUD komatiites.

1421

1422 Fig. 13: Provenance: TiO_2 vs. $\text{Fe}^{3+}/\Sigma\text{R}^{3+}$, diagrams modified from Barnes & Roeder (2001).
1423 Chromites that ostensibly yield no Fe^{3+} from 14WA2 are excluded for clarity. Fit of 14WA2
1424 and 01JH54 chromite to all fields for and (a) ophiolites, (b) layered intrusions and (c) Al-
1425 undepleted komatiites. GSF; Greenschist facies komatiites, AF; Amphibolite facies komatiites.

1426

1427 Fig. 14: Cr-spinel compositions from the Narryer Terrane. BSE images of Manfred Complex
1428 13TKN22 chromite; (a) finer chromite (chr) within 13TKN22, with a strong magnetite rim
1429 (mag) and an inclusion of olivine (ol), and (b) coarser chromite within hornblende (hbl).
1430 Serpentine (srp) and magnetite (mag) also shown. (c) and (d) Relict, porous chromite with thick
1431 ferritchromit and magnetite rims within 16WA13, an ultramafic from the SW limb of the Jack
1432 Hills belt. Note the softening of the boundary between core and rim in comparison to 13TKN22.
1433 (e)-(g) Compositional variability of 13TKN22 and 16WA13 chromite in comparison to Jack
1434 Hills detrital chromites. Spinel compositional fields from Manfred Complex harzburgites and
1435 websterites reported by Rowe & Kemp (2020) included for comparison. Layered intrusion
1436 fields from Barnes & Roeder (2001). (e) Cr^{3+} - Al^{3+} - Fe^{3+} triangular plot, (f) Cr# vs $\text{Fe}^{2+}\#$ and (g)
1437 TiO_2 vs $\text{Fe}^{3+}/\Sigma\text{R}^{3+}$.

1438

1439 Fig. 15: The inferred petrogenesis of Jack Hills chromites. (a) Original igneous crystallisation
1440 in a single, spatially extensive igneous protolith, postulated to represent a layered intrusion. (b)
1441 Exhumation of layered intrusion and first sedimentary cycling event. Continued erosion

1442 resulted in loss of contemporaneous silicates; some chromites preserved isolated within
1443 quartzite. Chromites (\pm zircon) now observed within the matrix of Jack Hills metasedimentary
1444 rocks present previously in heavy mineral layers or disseminated throughout quartzite. (c)
1445 Further sedimentary reworking in second cycling event deposits chromites in Jack Hills
1446 conglomerates and quartzites. Dispersion of chromites and zircons within both quartzite
1447 cobbles and matrix. (d) Metamorphism of Jack Hills metasedimentary rocks at \sim 2650 Ma, likely
1448 representing peak metamorphism at upper greenschist to lower amphibolite facies. This event
1449 also formed fuchsite and other metamorphic assemblages within the host metasedimentary
1450 rock, replaced chromite primary inclusion assemblages, and led to the partial to complete
1451 exchange of chromite divalent cations with metamorphic fluids. (e) Proterozoic metamorphic
1452 overprint of metasediments within the Jack Hills. Minor metamorphic exchange potentially
1453 indicated by zonation profiles.

Fig.1

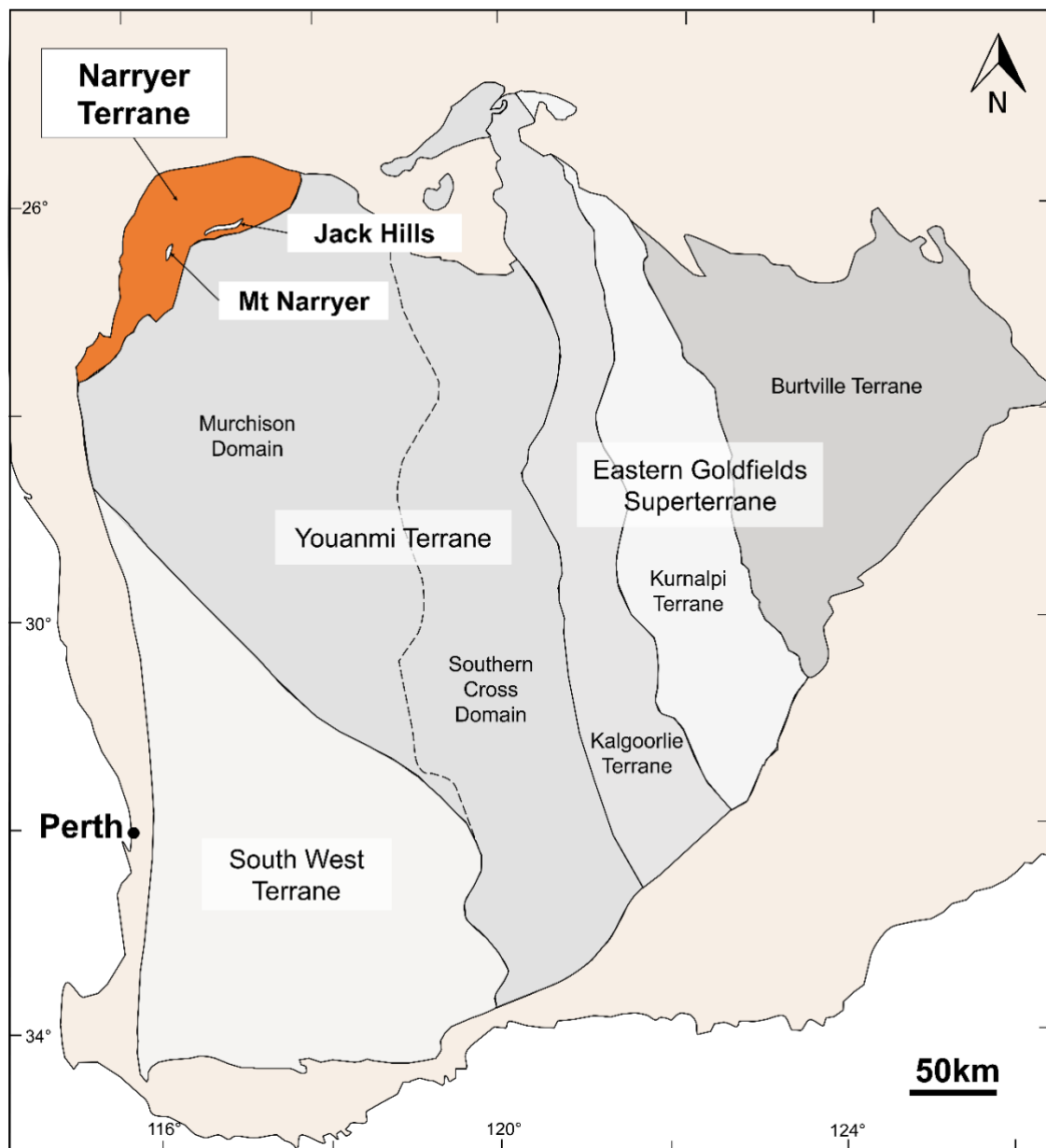
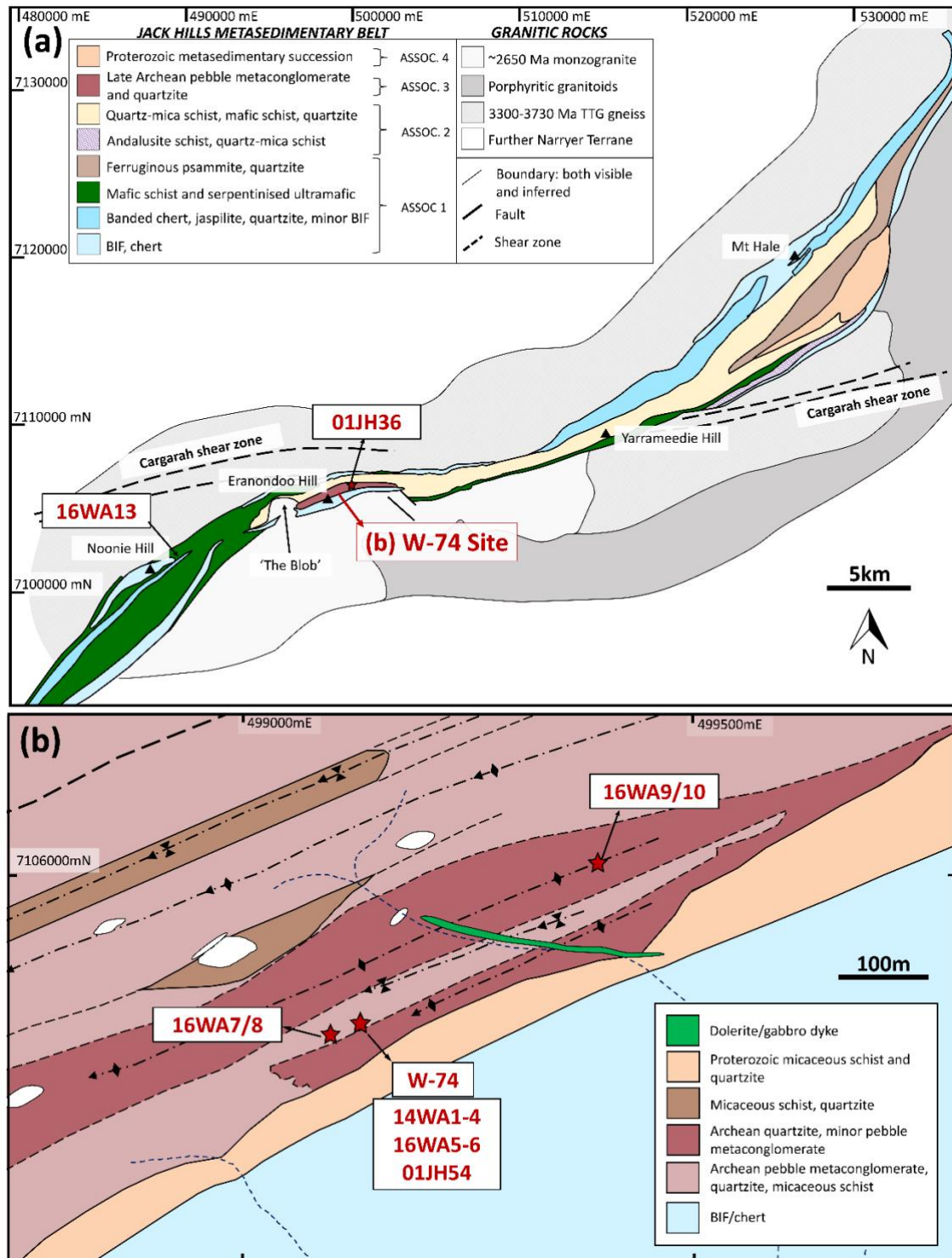


Fig.2



To add: 01JH35 sampling location

Fig. 3

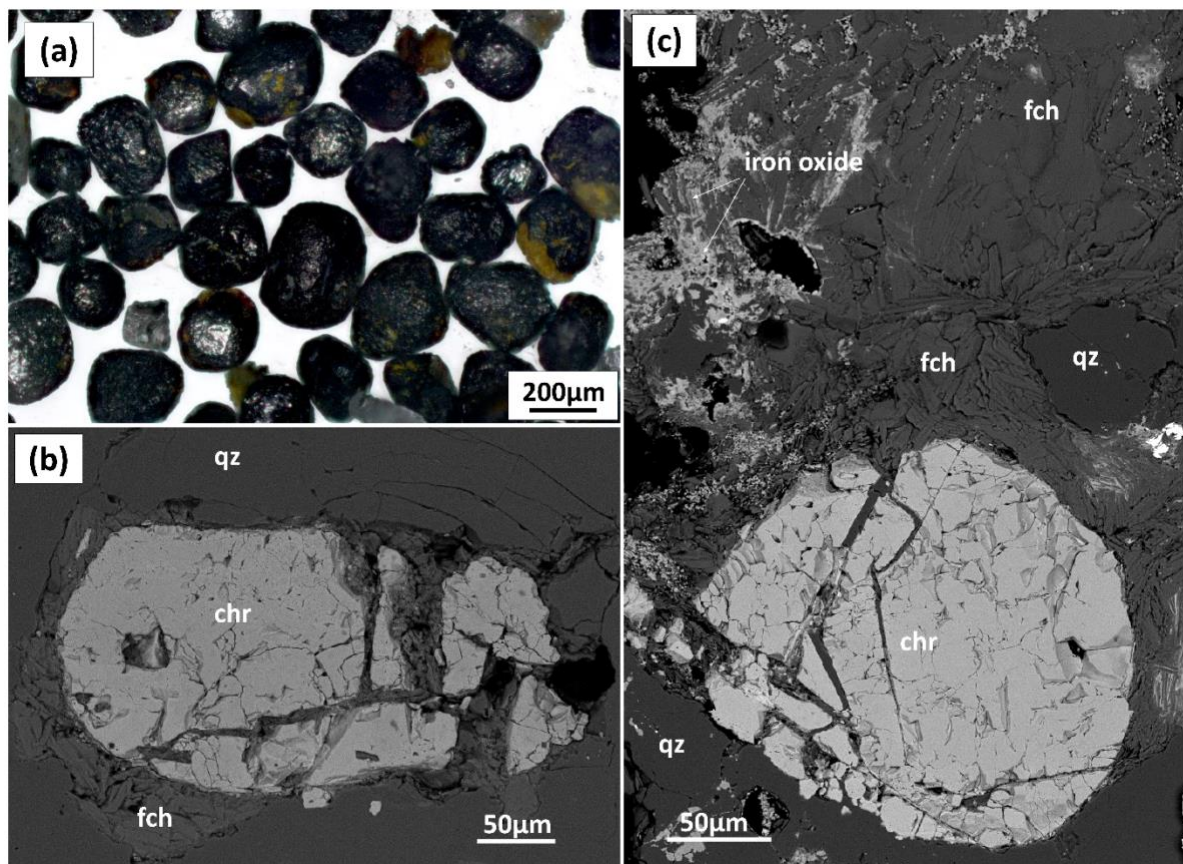


Fig. 4

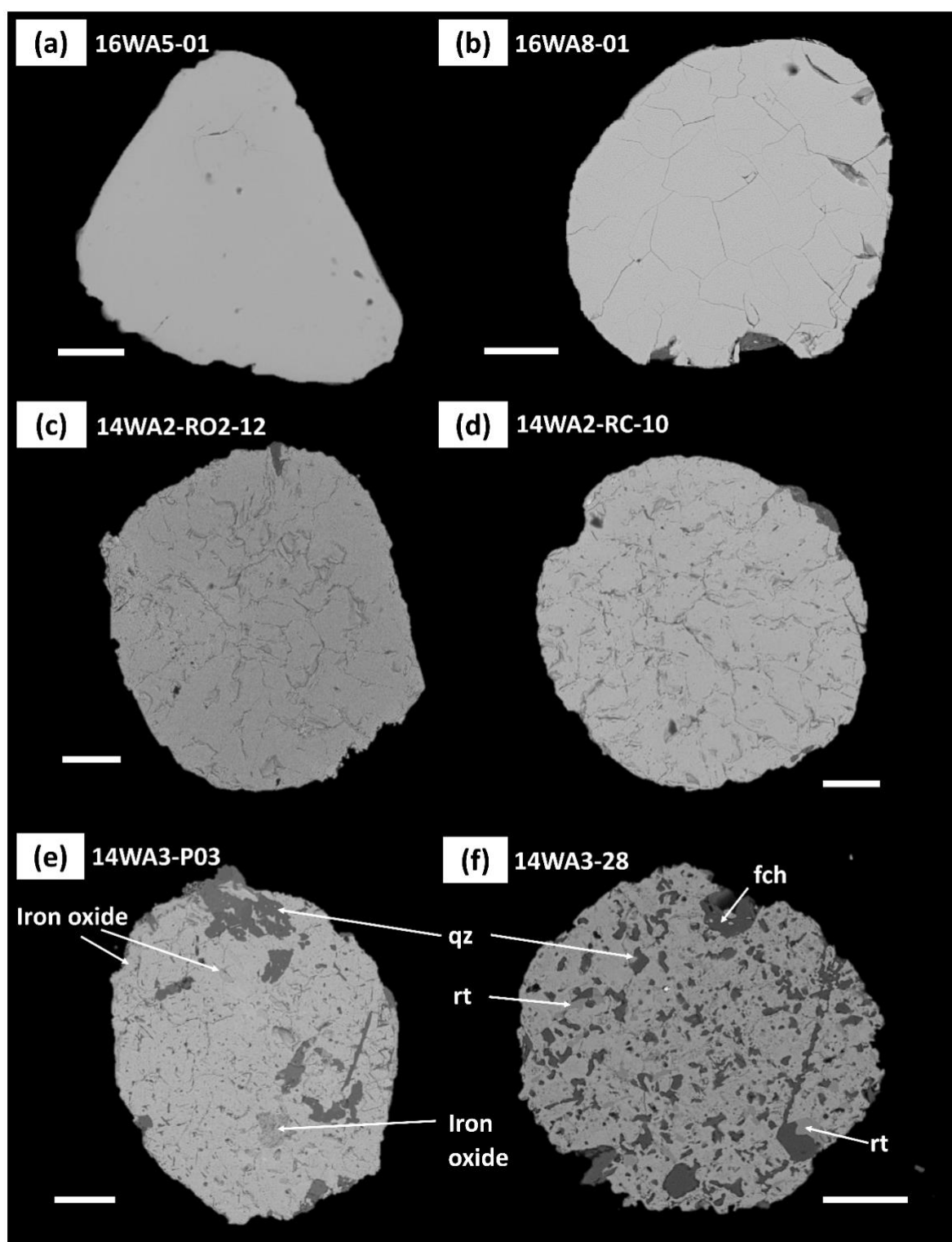


Fig. 5

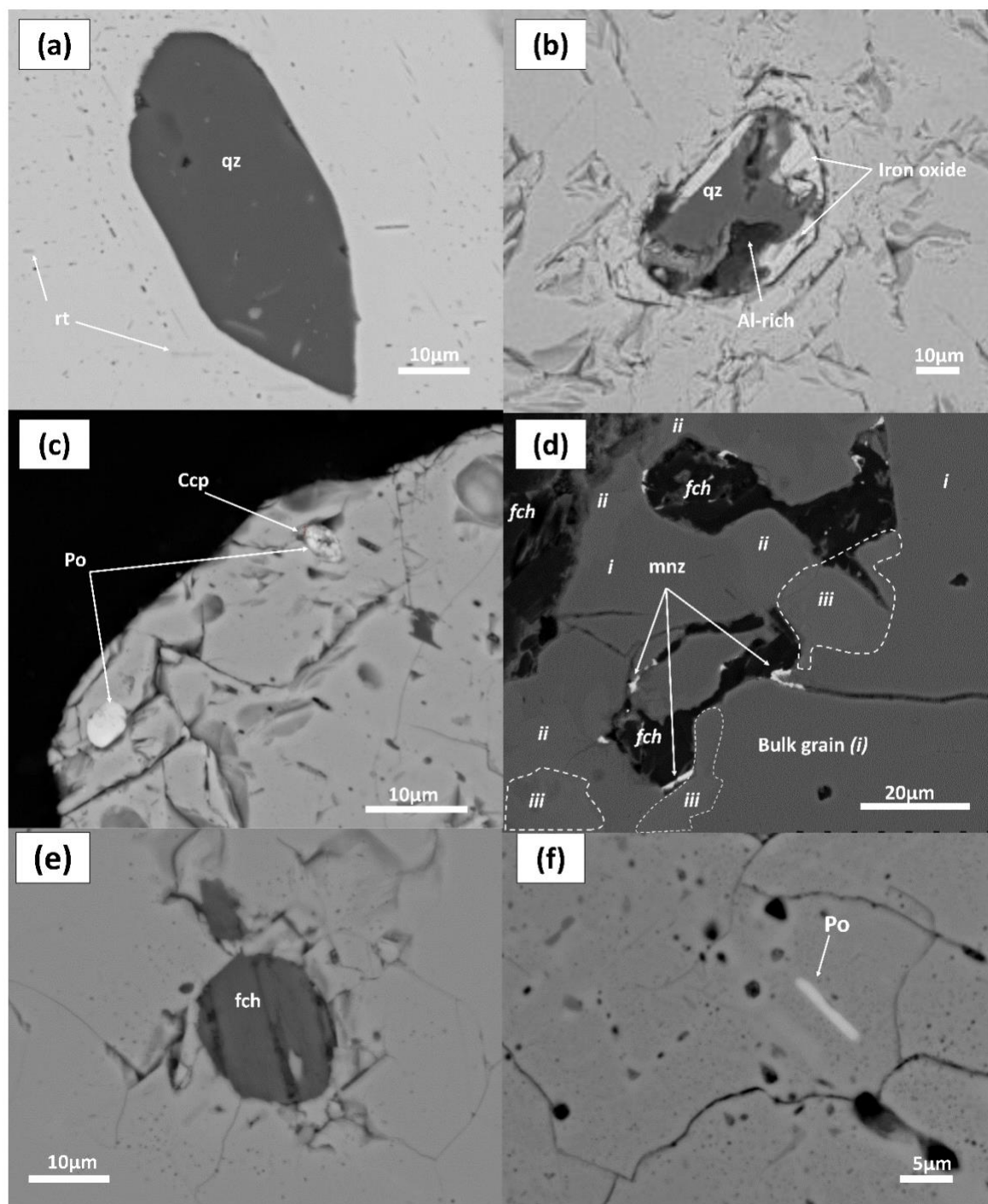


Fig. 6

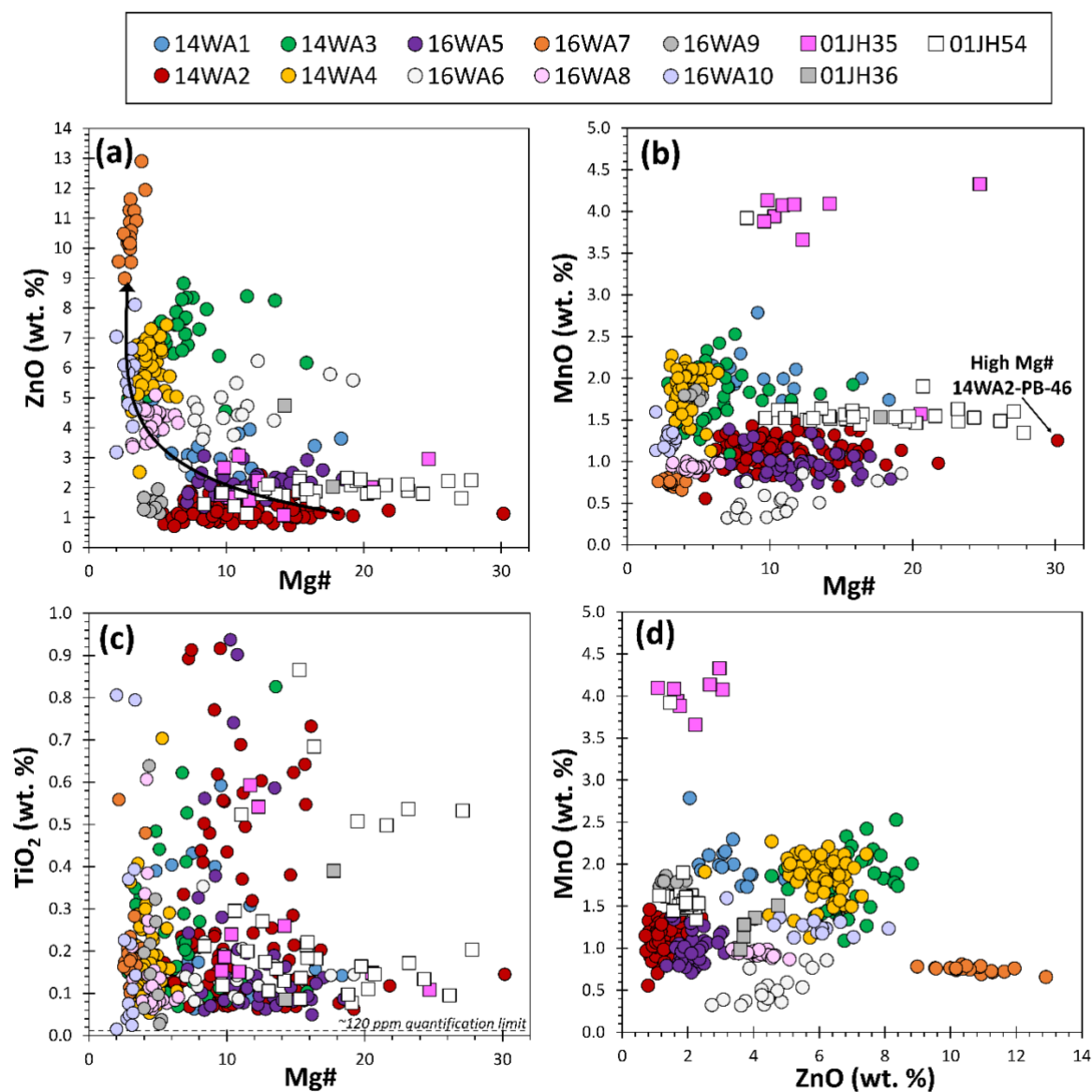


Fig. 7

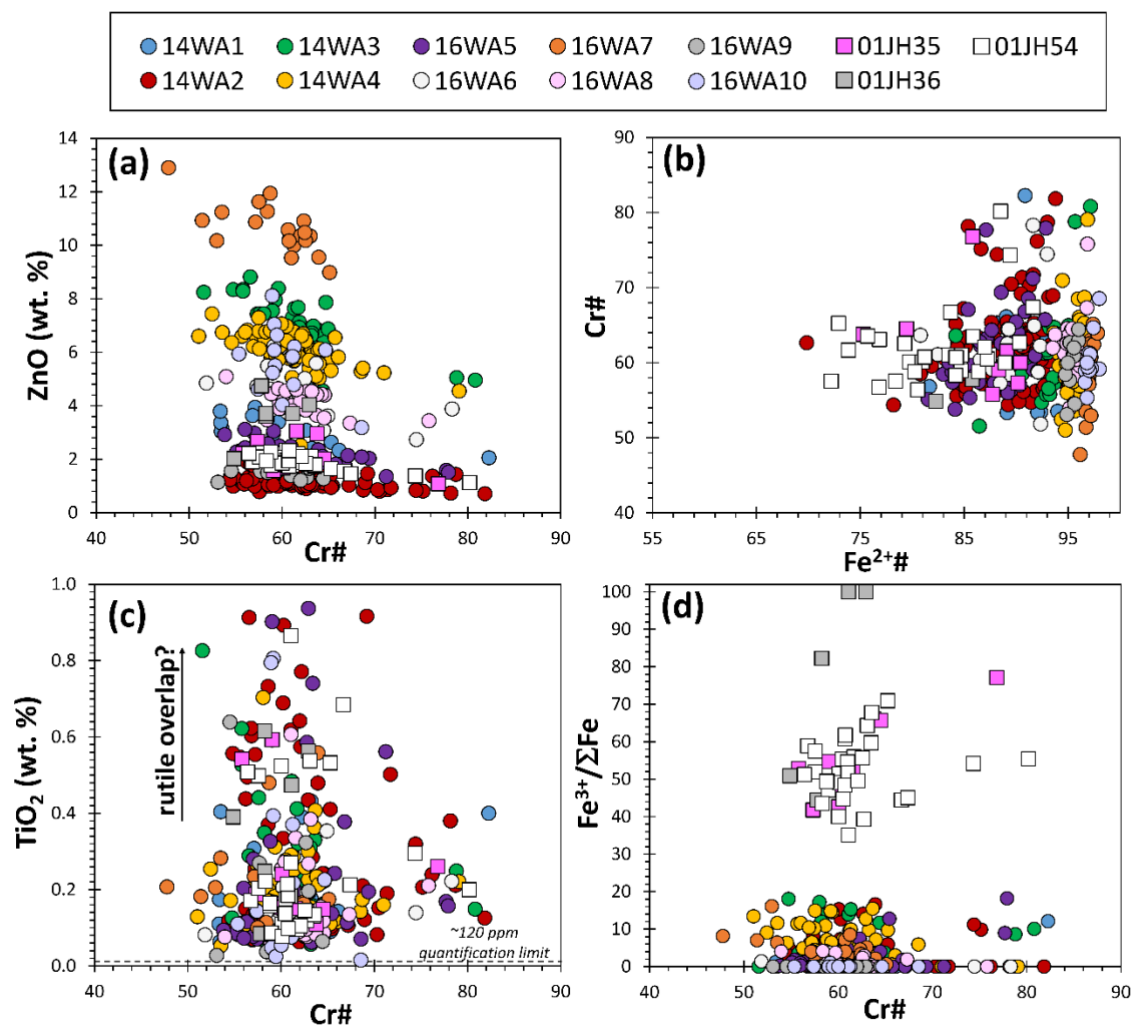


Fig. 8

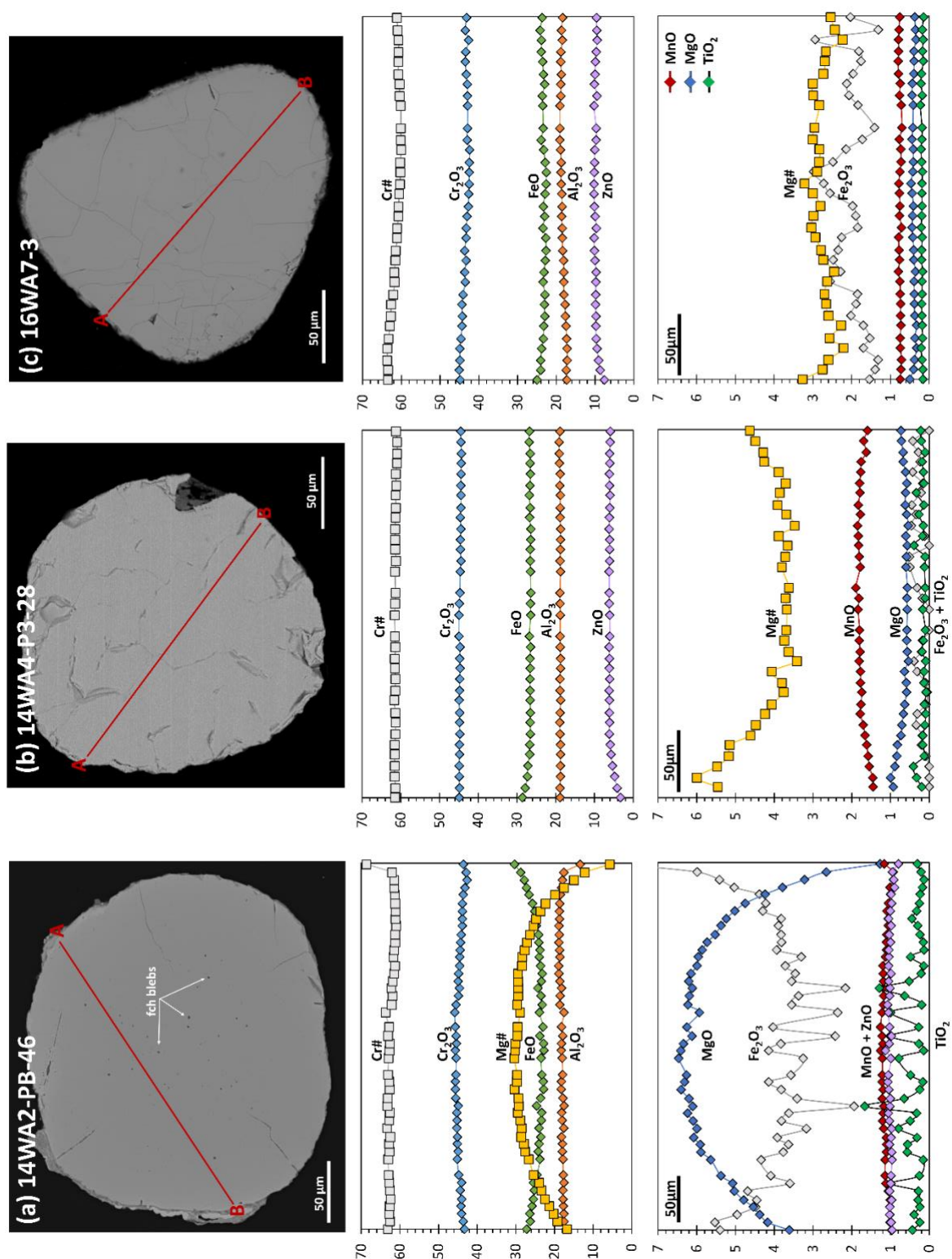


Fig. 9

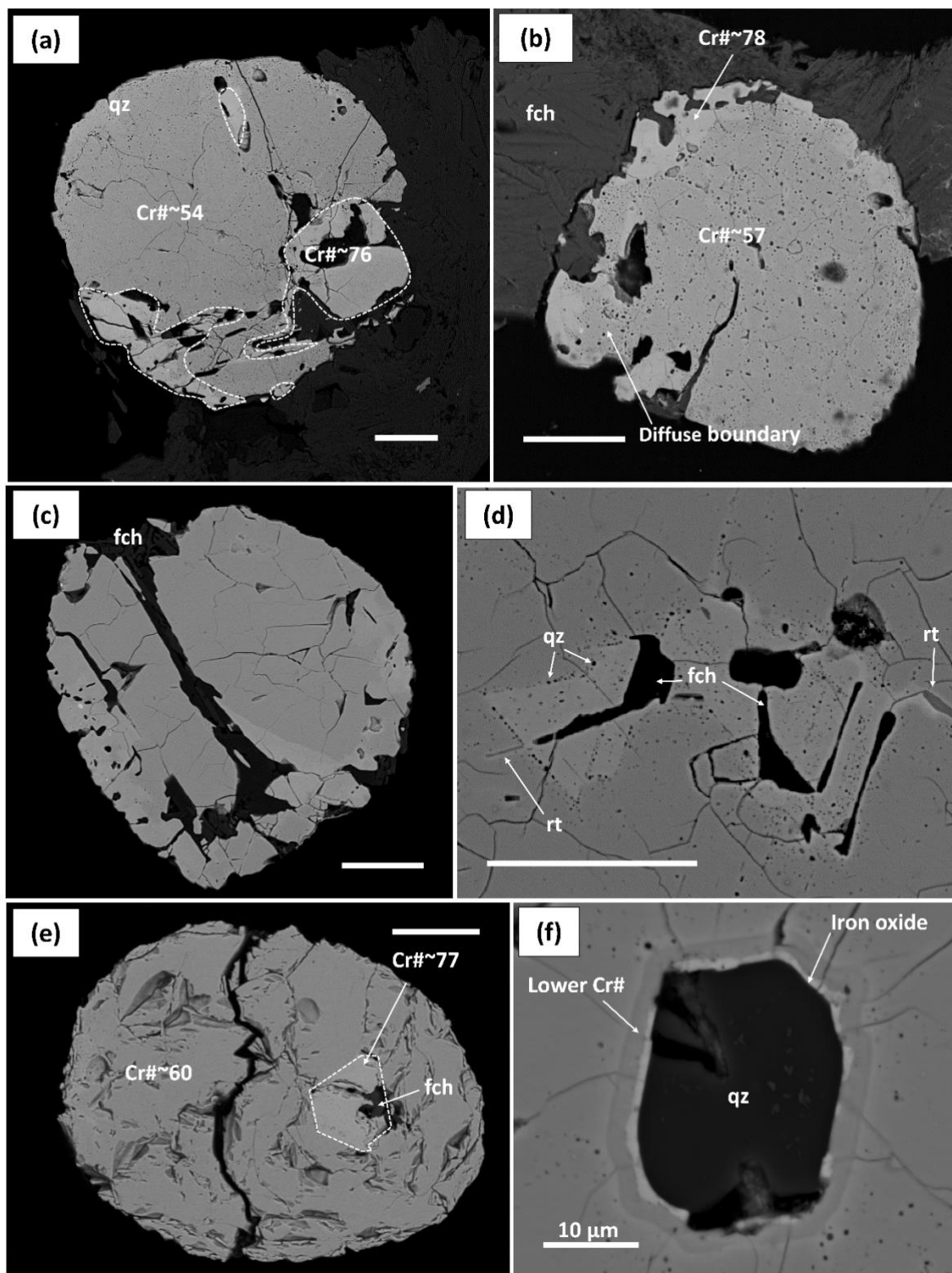


Fig. 10

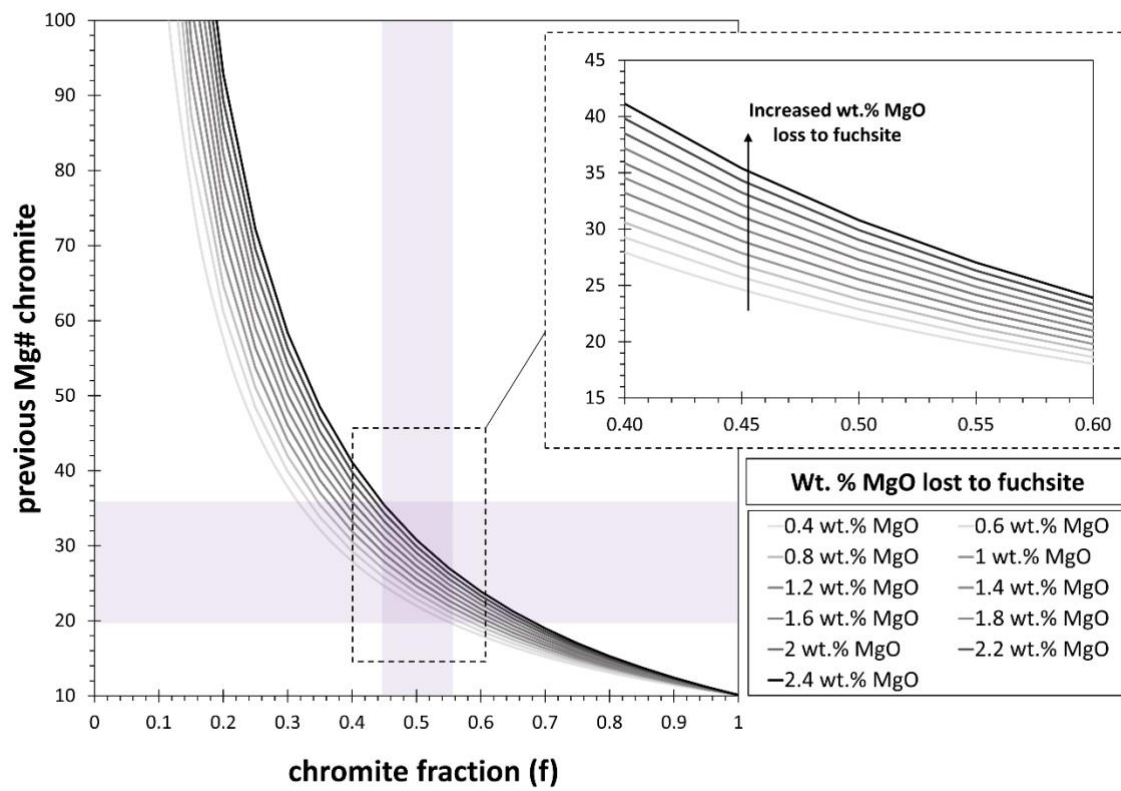


Fig. 11

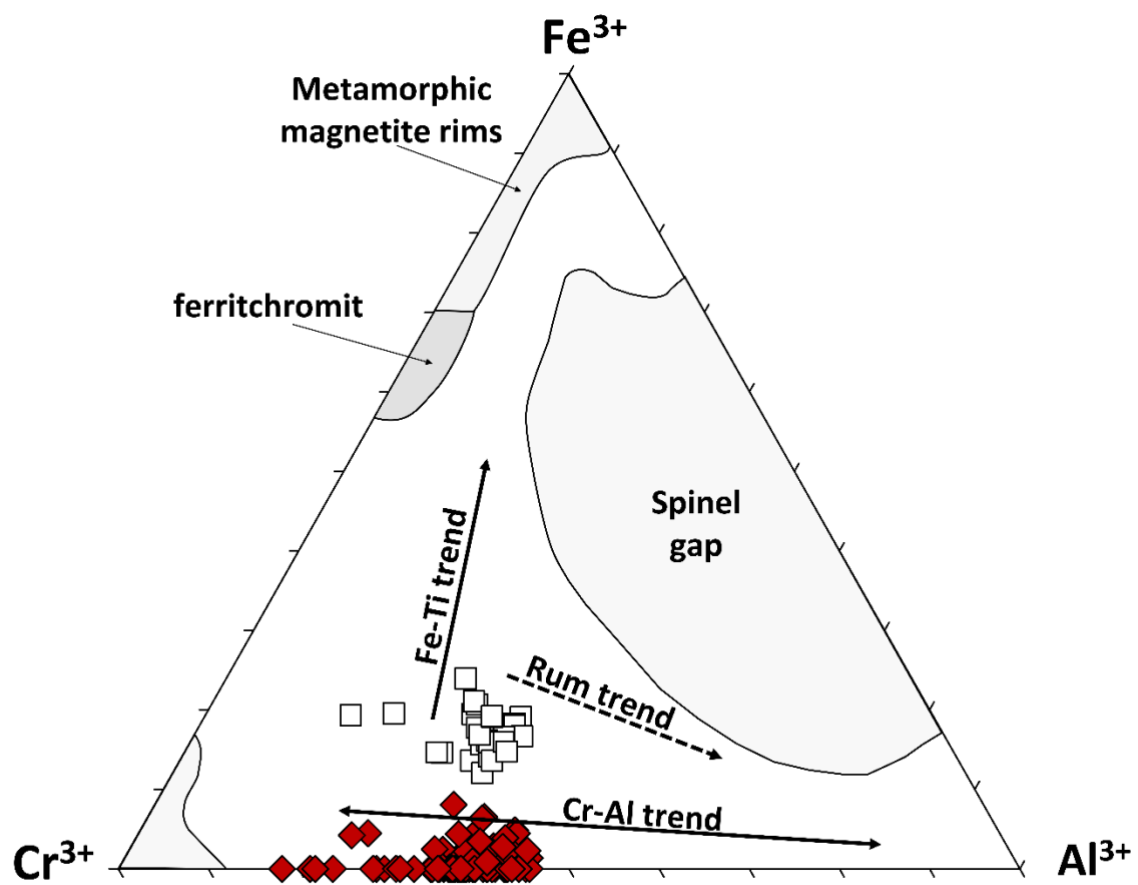


Fig. 12

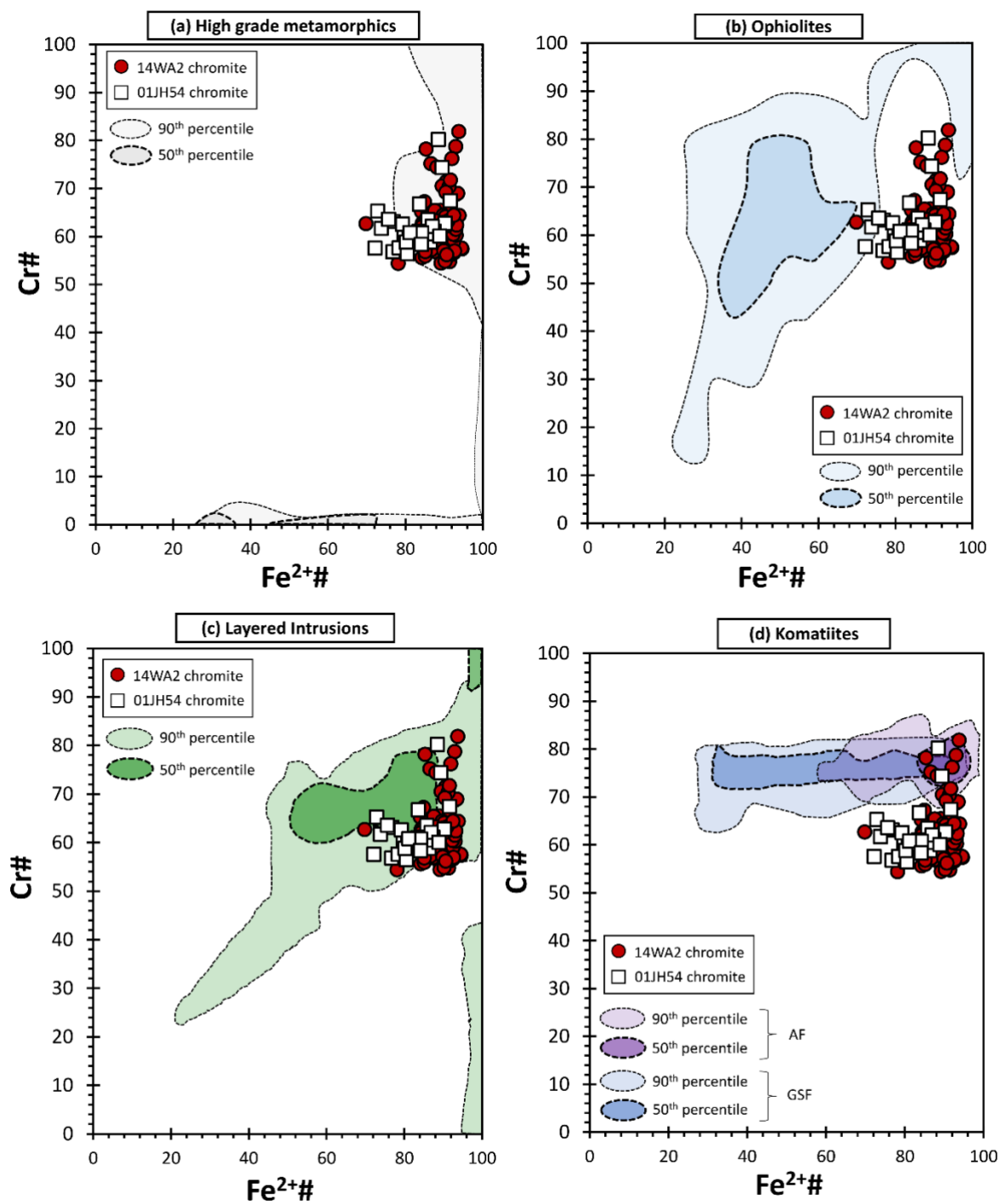


Fig. 13

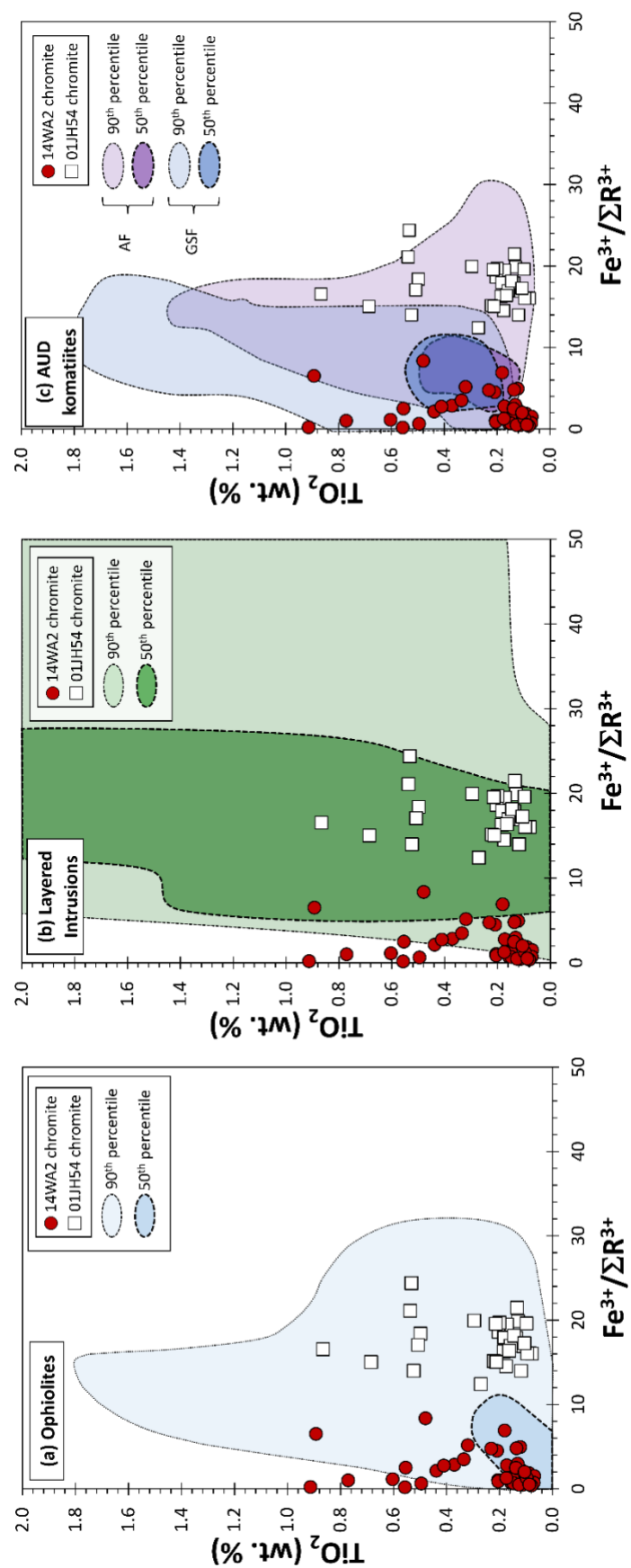


Fig. 14

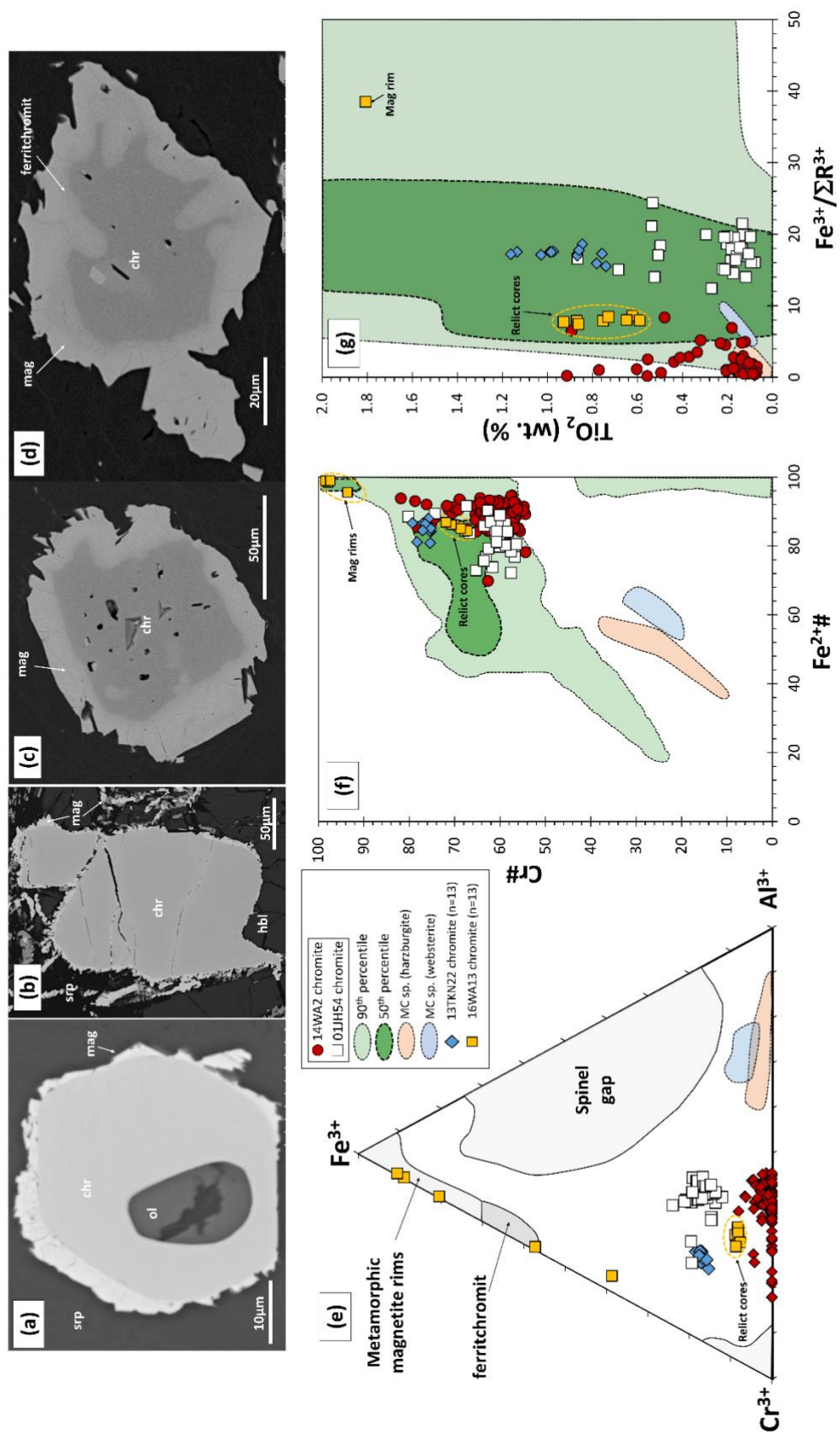


Fig. 15

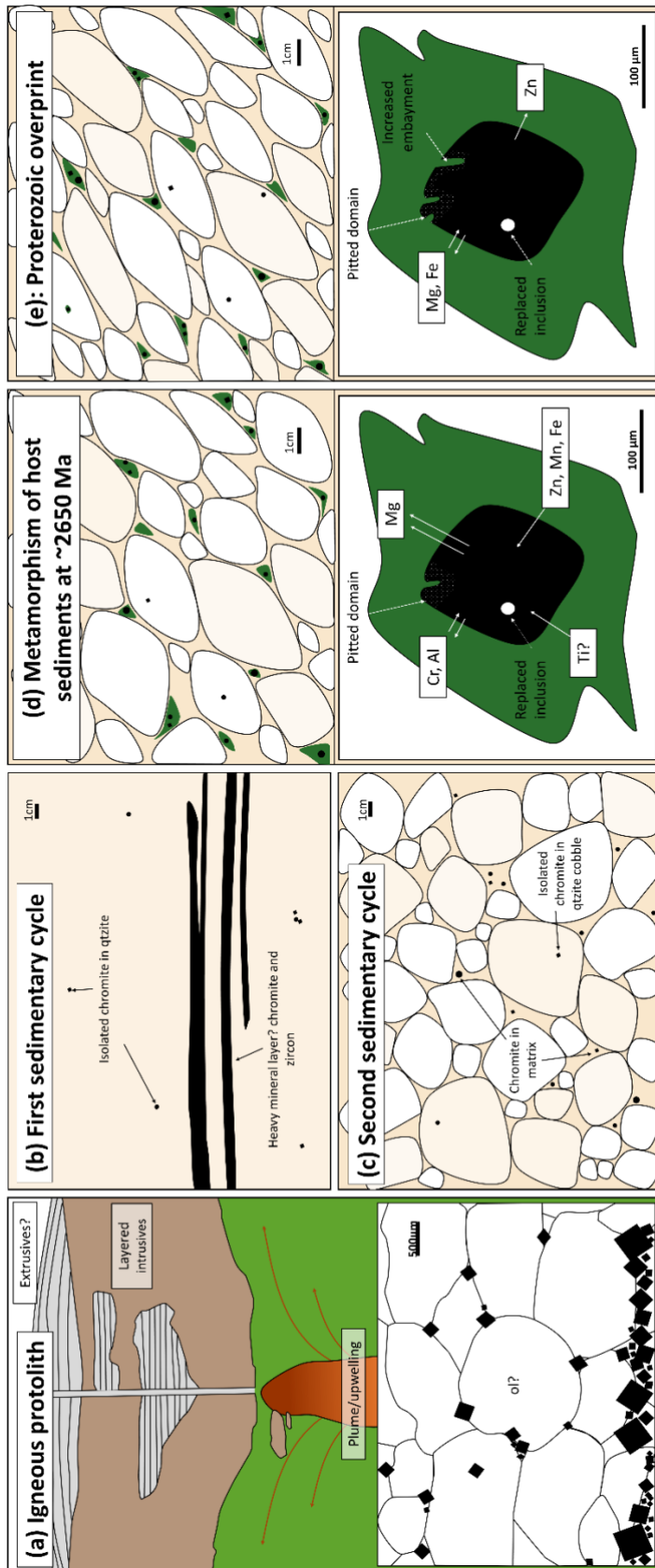


Table 1: Representative EPMA analysis of Jack Hills chromites.													
Sample	14WA1	14WA2	14WA3	14WA4	16WA5	16WA6	16WA7	16WA8	16WA9	16WA10	01JH35	01JH36	01JH54
Grid Ref.	499135E	499135E	499135E	499135E	499135E	499141E	499102E	499096E	499395E	499393E		499947E	499137E
(UTM)	7105846N	7105846N	7105846N	7105846N	7105846N	7105855N	7105837N	7105837N	7106007N	7106004N		7106431N	7105849N
Grain #	4	17	44	28	19	12	7	5 core	2	20	4	5	5
Rounding	RO	RO (2)	RO	RO/RC	RO	EO	RO/EO	RO/RC	EO	RC			
SiO ₂	0.02	0.08	0.00	0.00	0.01	0.01	0.00	0.01	0.03	0.03	0.03	0.01	0.03
TiO ₂	0.18	0.32	0.27	0.09	0.12	0.19	0.28	0.08	0.22	0.04	0.54	0.08	0.11
Al ₂ O ₃	18.52	11.89	18.13	18.53	20.28	18.26	22.43	19.02	19.48	20.35	21.16	21.14	19.86
Cr ₂ O ₃	46.20	51.52	42.43	44.42	44.18	45.82	38.52	44.69	45.26	44.11	39.79	43.14	44.59
V ₂ O ₃	0.12	0.19	0.16	0.11	n/a	n/a	n/a	n/a	0.14	0.13	0.23	0.12	0.14
Fe ₂ O ₃	0.00*	3.97*	4.37*	0.82*	0.00*	0.00*	1.81*	0.05*	0.00*	0.00*	16.82 [‡]	12.26 [‡]	15.89 [‡]
FeO	27.66	28.26	24.39	26.17	28.76	27.50	23.79	28.63	29.19	26.65	13.52	14.15	13.49
MgO	1.20	2.40	1.39	0.62	2.18	1.67	0.49	0.91	0.76	0.42	1.06	1.32	1.92
MnO	2.13	1.46	2.03	1.84	0.71	0.33	0.73	0.95	1.80	1.28	3.66	1.51	1.46
CaO	0.00	0.01	0.00	0.01	0.00	0.01	0.01	0.00	0.00	0.00	n/a	n/a	n/a
Na ₂ O	0.00	0.10	0.00	0.00	n/a	n/a	n/a	n/a	n/a	n/a	n/a	n/a	n/a
NiO	0.00	0.00	0.00	0.01	0.02	0.00	0.00	0.01	0.00	0.00	0.01	0.00	0.00
ZnO	3.08	0.84	7.29	6.19	2.11	4.32	11.24	3.95	1.39	5.50	2.22	4.74	1.80
CuO	n/a	n/a	n/a	n/a	n/a	n/a	n/a	n/a	n/a	n/a	0.01	0.02	0.00
Total	99.10	101.03	100.46	98.80	98.38	98.11	99.31	98.31	98.27	98.52	99.05	98.87	99.29
Mg#	7.20	11.83	8.03	3.95	11.90	9.77	3.30	5.36	4.44	2.76	12.19	14.24	20.24
Cr#	62.60	74.41	61.09	61.66	59.37	62.74	53.53	61.19	60.92	59.25	60.04	57.78	60.10
Fe ³⁺ /ΣFe	0.00	11.22	13.87	2.74	0.00	0.00	6.42	0.00	0.00	0.00	52.81	44.51	51.46
Fe ³⁺ /ΣR ³⁺	0.00	5.17	5.65	1.07	0.00	0.00	2.34	0.00	0.00	0.00	18.32	13.86	16.93

Table 1: Representative EPMA analysis of chromite from each sample location. V₂O₃ not measured for 16WA5-7 and partial 16WA8. *Fe₂O₃ calculated from AB₂O₃ stoichiometry using the equations of Droop (1987). [‡]Fe₂O₃ calculated by charge balance from direct EPMA measurement of oxygen.

Table 2: Oxygen isotopic composition ($\delta^{18}\text{O}_{\text{VSMOW}}$) of Jack Hills detrital chromites.						
Size fraction (μm)	105-149	149-210	210-300	300-350	350-425	Uncertainty (2sd)
01JH36	0.65		1.15	1.16		0.08
01JH36	0.73		1.41	1.71		
01JH36			1.41	1.77		
01JH36 Average	0.69		1.32	1.55		
01JH54		1.62	2.17	1.02	1.65	0.06
01JH54		1.69	2.2	1.17	2.11	
01JH54		2.07				
01JH54 Average		1.79	2.19	1.10	1.88	

Table 2: Oxygen isotopic composition ($\delta^{18}\text{O}_{\text{SMOW}}$) of detrital chromites from 01JH36 and 01JH54 determined by laser fluorination. Values represent deviation from standard mean ocean water (V-SMOW) in per mil (‰). Each analysis consisted of a ~2 mg aliquot of chromite separated by grain size.

Any separation by rounding shape?

Table 3: Representative EPMA compositions of 13TKN22 chromites and 16WA13 relict chromite cores.						
Sample Grid Ref. (UTM) Crystal # Classification	13TKN22 40 chromite	13TKN22 42 chromite	13TKN22 47 chromite	16WA13 489515E 7102336E relict chromite	16WA13 489515E 7102336E relict chromite	16WA13 489515E 7102336E relict chromite
SiO ₂	0.00	0.00	0.02	0.03	0.02	0.03
TiO ₂	1.13	1.03	0.97	0.62	0.92	0.86
Al ₂ O ₃	9.23	9.26	9.18	13.31	12.80	12.52
Cr ₂ O ₃	43.66	44.18	43.87	45.96	46.48	46.77
V ₂ O ₃	n/a	n/a	0.25	n/a	n/a	n/a
Fe ₂ O ₃	12.88	12.63	12.95	6.47	5.76	5.87
FeO	28.10	27.72	27.20	27.89	28.25	28.17
MgO	2.72	3.41	3.72	2.83	2.68	2.65
MnO	1.32	0.45	0.43	1.21	1.20	1.19
CaO	0.01	0.03	0.03	0.01	0.00	0.03
NiO	0.14	0.12	0.13	0.08	0.12	0.10
ZnO	0.60	0.59	0.58	1.08	1.06	1.08
Total	99.80	99.43	99.33	99.10	99.61	99.30
Mg #	14.73	17.99	14.59	13.00	12.49	12.37
Cr #	76.03	76.20	76.22	69.84	70.89	71.47
Fe ³⁺ /ΣFe	29.20	29.08	30.00	17.28	15.51	15.78
Fe ³⁺ /ΣR ³⁺	17.59	17.18	17.64	8.56	7.72	7.86

Table 3: Representative EPMA analysis of chromite from Manfred Complex metaperidotite 13TKN22 and Jack Hills recrystallised ultramafic 16WA13. V₂O₃ not measured for some chromites within 13TKN22. Fe₂O₃ calculated from AB₂O₃ stoichiometry using the equations of Droop (1987).