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7 Detrital chromite from Jack Hills, Western Australia: signatures of metamorphism and
8 constraints on chromite provenance

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26 ABSTRACT

27 Detrital chromites are commonly reported within Archean metasedimentary rocks, but have
28 thus far garnered little attention for use within provenance studies. Yet, systematic variations
29 of Cr-Fe spinel mineral chemistry with changing tectonic setting means chromite has been
30 extensively used as a petrogenetic indicator, and so detrital chromites represent good candidates
31 to investigate the petrogenesis of eroded Archean mafic and ultramafic crust. Here, we report
32 the compositions of detrital chromites within fuchsite (Cr-muscovite rich) metasedimentary
33 rocks from the Jack Hills, geologically renowned for hosting Hadean (>4000 Ma) zircons,
34 situated within the Narryer Terrane, Yilgarn Craton, Western Australia. We highlight signatures
35 of metamorphism, including highly elevated ZnO and MnO, coupled with lowered Mg# in
36 comparison to magmatic chromites, development of pitted domains, and replacement of
37 primary inclusions by phases abundant as metamorphic assemblages within host
38 metasedimentary rocks. Oxygen isotope compositions of detrital chromites indicate partial
39 exchange to complete equilibration with host metasedimentary rocks. Variability of
40 metamorphic signatures between chromites sampled only meters apart further indicates
41 modification occurred *in-situ* by interaction detrital chromites with of metamorphic fluids and
42 secondary mineral assemblages. Alteration likely occurred during upper greenschist to lower
43 amphibolite facies metamorphism and deformation of host metasedimentary rocks at ~2650
44 Ma. Regardless of metamorphic signatures, sampling location or grain shape, chromite cores
45 yield a consistent range in Cr#. While other key petrogenetic indices, such as Fe₂O₃ and TiO₂
46 contents, are complicated in Jack Hills chromites by mineral non-stoichiometry and secondary
47 mobility within metasedimentary rocks, we demonstrate that the Cr# of chromite yields
48 significant insights into their provenance. Importantly, moderate Cr# preclude a komatiitic
49 origin for the bulk of chromites. This reflects a dearth of komatiites and associated intrusives
50 within the erosional catchment of the Jack Hills metasedimentary units. We suggest Cr# fit well

51 with chromites derived from layered intrusions, and that a single layered intrusion may account
52 for the observed chemical compositions of Jack Hills detrital chromites. Thus, we show that,
53 where careful characterisation of key metamorphic signatures is undertaken, detrital chromites
54 may yield valuable information on the petrogenesis and geodynamic setting of poorly preserved
55 mafic and ultramafic crust.

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60 INTRODUCTION

61

62 The detrital record of crustal evolution within the early Earth is dominated by analysis of the
63 mineral zircon. This is particularly evident in the Narryer Terrane, within the Yilgarn Craton of
64 Western Australia (Fig. 1). Here, Proterozoic to late Archean (Cavosie *et al.*, 2004; Crowley *et*
65 *al.*, 2005; Rasmussen *et al.*, 2010; Wang & Wilde, 2018) metasedimentary rocks in the Jack
66 Hills (Fig. 2) yield the oldest known fragments of terrestrial crust; individual grains of detrital
67 zircon yield ^{207}Pb - ^{206}Pb ages of up to 4374 ± 6 Ma (Compston & Pidgeon, 1986; Wilde *et al.*,
68 2001; Valley *et al.*, 2014). Despite isolated occurrences of Hadean detrital zircon elsewhere
69 (e.g., Byerly *et al.*, 2018 and references therein), including Mt. Narryer to the SW (Fig. 1;
70 Froude *et al.*, 1983; Pidgeon & Nemchin, 2006), and rare examples of Hadean xenocrystic
71 zircon (Nelson *et al.*, 2000; Wyche *et al.*, 2004; Iizuka *et al.*, 2006; Chaudhuri *et al.*, 2018),
72 Jack Hills detrital zircons represent critically important remnants of Hadean (>4000 Ma) crust.

73 These zircons therefore provide a unique window into the Hadean Earth, and have
74 subsequently been rigorously interrogated using numerous geochemical and isotopic techniques
75 (e.g., Cavosie *et al.*, 2018). Previous investigations have broached a wide range of subjects,
76 including Earth's ancient dynamo (Borlina *et al.*, 2020; Tarduno *et al.*, 2020), the origins of life
77 (Bell *et al.*, 2015) and the role of impacts within the early Earth (Bell & Harrison, 2013; Cox
78 *et al.*, 2017). Despite intense study, what Jack Hills zircons tell us about the composition,
79 evolution and subsequent destruction of their original host rocks is controversial, with
80 conflicting hypotheses inferring vastly disparate geodynamic conditions within the early Earth.
81 Such hypotheses include a 'cool early Earth' (e.g., Valley *et al.*, 2002), with the putative
82 operation of plate tectonics (Harrison *et al.*, 2008; 2017; Bell *et al.*, 2014), production of zircon-
83 bearing crust by internal reworking of mafic protocrust (Amelin *et al.*, 1999; Kemp *et al.*, 2010),
84 or compositionally diverse Hadean protoliths (Wang & Wilde, 2018).

85 While the Jack Hills zircon record yields valuable, if controversial, constraints on the
86 evolution of felsic Hadean and Archean crust, the zircon record provides little information on
87 the evolution of contemporaneous mafic and ultramafic crust. This is of particular importance,
88 as emerging evidence suggests mafic to ultramafic crust was the dominant compositional
89 component of the Archean (e.g., Dhuime *et al.*, 2015; Kamber *et al.*, 2015; Tang *et al.*, 2016).
90 Evolved, zircon-bearing crust may therefore be over represented within detrital records, in part
91 due to the poorer preservation potential of its mafic to ultramafic counterparts. While the wider
92 Yilgarn Craton is renowned for the presence of economically significant komatiites (Arndt *et*
93 *al.*, 2008), the generation and evolution of mafic and ultramafic crust within the Narryer Terrane
94 is poorly constrained. Indeed, the only >3100 Ma mafic and ultramafic crust described within
95 the entire Yilgarn Craton is the Eoarchean Manfred Complex (Wyche, 2007), a 3730 Ma
96 disseminated layered intrusion within the Narryer Terrane (Fletcher *et al.*, 1988; Kinny *et al.*,
97 1988; Myers, 1988b; Rowe & Kemp, 2020).

98 However, the largely unexplored eroded remnants of mafic and ultramafic crust are
99 ubiquitous within Jack Hills metasedimentary rocks, in the form of detrital chromite. Unlike
100 zircon and other detrital phases observed at Jack Hills, including minor monazite and xenotime
101 (Rasmussen *et al.*, 2010; Iizuka *et al.*, 2010), chromite has a magmatic provenance restricted
102 solely to mafic and ultramafic crust (Barnes & Roeder, 2001). Furthermore, chromites within
103 their mafic or ultramafic protoliths are frequently used as a petrogenetic indicator, owing to the
104 presence of systematic chemical variations in chromite formed under different tectonic
105 conditions (Irvine, 1965; 1967; Dick & Bullen., 1984; Roeder, 1994; Barnes & Roeder, 2001;
106 Kamenetsky *et al.*, 2001). Critically, the sequestration of platinum group elements (PGEs) into
107 spinel means that chromite is amenable to geochronology using Re-Os and Pt-Os decay systems
108 (Shirey & Walker, 1998), potentially allowing a temporal framework of the generation of
109 chromite-bearing mafic and ultramafic crust to be constrained from the detrital record. The

110 value of detrital chromites for elucidating the provenance of magmatic protoliths has previously
111 been demonstrated (e.g., Barnes & Roeder, 2001; Lenaz & Princivalle, 2005; Barkov *et al.*,
112 2013), though additional care is required to quantify the effects of metamorphism on detrital
113 chromite found within ancient terranes (Barnes, 2000; Colás *et al.*, 2014). Understanding how
114 and when the Jack Hills detrital chromite formed will therefore provide valuable information
115 on the composition and petrogenesis of poorly described mafic and ultramafic crust that was
116 potentially exposed at the time the Jack Hills sedimentary succession was deposited

117 Here, we report major and minor element geochemistry of detrital chromites from
118 thirteen samples of metasedimentary rocks collected from within the Jack Hills (Fig. 2). This
119 study combines data collected by two groups (University of Bristol; 14WA and 16WA samples
120 and University of Wisconsin-Madison; 01JH samples), representing the first systematic study
121 of detrital chromite from Archean sedimentary rocks, and signifying a new direction compared
122 to traditional provenance studies undertaken by analysis of detrital zircon. We highlight the
123 effects of metamorphism on chromite mineral chemistry, discuss the retention of primary
124 signatures, and propose a plausible provenance for detrital chromites. Finally, we discuss the
125 significance of these grains for enhancing our understanding of the metamorphic history of the
126 Jack Hills metasedimentary rocks, the distribution of mafic and ultramafic crust within the
127 Archean upper crust at the time of deposition, and the wider potential of detrital chromites for
128 provenance studies of ancient mafic and ultramafic crust.

129

130 GEOLOGICAL SETTING

131

132 The Narryer Terrane

133 The Narryer Terrane is the most north-westerly terrane within the Yilgarn Craton, in Western
134 Australia (Myers, 1988a; Kemp *et al.*, 2018; Fig. 1), and has been interpreted as a deep crustal

135 allochthon thrust above the Youanmi Terrane (Nutman *et al.*, 1993), prior to or coincident with
136 cratonic amalgamation (Kemp *et al.*, 2018). The terrane is dominantly composed of granitic
137 lithologies, now largely preserved as quartzofeldspathic orthogneisses with >3 Ga protolith
138 ages, with minor ultramafic and mafic intrusives, and metasedimentary rocks (Myers &
139 Williams, 1985; Williams & Myers, 1987; Myers, 1988a). Neoarchean (~2700-2650 Ma)
140 granitic rocks are also abundant (Kemp *et al.*, 2018). Much of the Narryer Terrane has
141 undergone high-grade, polyphase deformation, with amphibolite to granulite facies events at
142 ~2700-2650 Ma forming the observed gneissic fabric (Myers, 1988a; Kinny, 1990; Nutman *et*
143 *al.*, 1991). There is evidence for previous high-grade thermal events, particularly at ca. 3300
144 Ma (Nutman *et al.*, 1991; Kinny & Nutman, 1996), and it has been postulated orthogneisses
145 within the Narryer Terrane underwent multiple episodes of deformation and anatexis during the
146 Archean (Kinny & Nutman, 1996).

147 Despite complicated zircon geochronology (Pidgeon & Wilde, 1998), three dominant
148 quartzofeldspathic orthogneiss units are identified within the Narryer Terrane (Kemp *et al.*,
149 2018). The Meeberrie gneiss is a biotite-rich migmatite that consists of 3670-3600 Ma
150 monzogranitic and 3730 Ma tonalitic protoliths (Nutman *et al.*, 1991; Kinny & Nutman, 1996;
151 Pidgeon & Wilde, 1998). The Eurada gneiss is a series of ~3480 Ma tonalitic gneisses that
152 comprise a fault bound lozenge west of Mount Narryer (Nutman *et al.*, 1991). The Dugel gneiss
153 yields a well constrained age of 3375 ± 26 Ma (Nutman *et al.*, 1991) and it is thought that its
154 syenogranitic protoliths intruded the Meeberrie Gneiss as a series of sheet-like and pegmatitic
155 bodies (Myers, 1988a; Kemp *et al.*, 2018).

156 The Meeberrie and Dugel gneiss host dispersed fragments of the Manfred Complex, a
157 magmatically and tectonically dismembered and variably metamorphosed layered intrusion.
158 The complex is dominantly amphibolitic (after gabbro and leucogabbro), with pyroxenite,
159 metaperidotite and anorthosite (Williams & Myers, 1987; Fletcher *et al.*, 1988; Myers, 1988b;

160 Rowe & Kemp, 2020). Relict igneous textures and layering are locally preserved (Kemp *et al.*,
161 2018). Zircon within Manfred Complex anorthosite and leucogabbro yield $^{207}\text{Pb}/^{206}\text{Pb}$ ages of
162 3730 ± 6 Ma (Kinny *et al.*, 1988; Kemp *et al.*, 2018), with other lithologies yielding Sm-Nd and
163 Pb-Pb whole rock (WR) ages of 3680 ± 70 Ma and 3689 ± 146 Ma, respectively (Fletcher *et al.*,
164 1988). Spinel (spinel *sensu-stricto* to picotite) and olivine chemical compositions are consistent
165 with formation of the Manfred Complex within a thickened oceanic plateau; reduced, high Al-
166 Ca-Fe picritic to tholeiitic basaltic parental melt compositions are thought to be derived from
167 shallow partial melting of spinel lherzolite (Rowe & Kemp, 2020).

168

169 Jack Hills

170 The Jack Hills are located at the southern margin of Narryer Terrane (Figs. 1 & 2) and comprise
171 a thin, ~ 70 km-long belt, with a distinctly curvilinear morphology produced by dextral shearing
172 (Spaggiari, 2007a). The Jack Hills belt is tectonically juxtaposed with the surrounding gneisses,
173 except for localised intrusion of 2654 Ma monzogranite, also known as ‘The Blob’ (Pidgeon &
174 Wilde, 1998; Spaggiari *et al.*, 2007b), and has been interchangeably referred to as a greenstone
175 (e.g., Spaggiari *et al.*, 2007a/b) or metasedimentary/supracrustal belt (e.g., Wang & Wilde,
176 2018). It largely consists of siliciclastic units, including metaconglomerate, quartzite and
177 quartz-mica schist, with minor intercalated mafic and ultramafic rocks, banded iron formation
178 (BIF) and chert. The presence of grunerite within BIF and hornblende within mafic schist
179 indicates at least portions of the belt have reached amphibolite facies metamorphism; later
180 greenschist facies metamorphism defines the dominant metamorphic signature of the belt
181 (Spaggiari, 2007a).

182 Deformation has tectonically disturbed and juxtaposed lithological associations, making
183 an original stratigraphy difficult to discern. Spaggiari (2007a) divided the belt into four
184 associations determined by lithological variability (Fig. 2). Briefly, association 1 consists of

185 interbedded BIF, chert and quartzite, mafic and ultramafic rocks, and black and white banded
186 quartzites. Association 2 yields pelitic to semi-pelitic associations, now present as quartz-mica
187 and andalusite schists, with accompanying mafic schist and quartzite. The presence of an S1
188 cleavage and recumbent folding absent from other lithological units suggests associations 1 and
189 2 were deformed prior to the deposition of units 3 and 4 (Spaggiari, 2007a). Association 3 is
190 restricted to the central region of the belt at Eranondoo Hill (Fig. 2), and was derived from
191 mature, siliciclastic sediments interpreted to have been deposited within a deltaic alluvial fan
192 (Spaggiari *et al.*, 2007a/b) between ~3050 Ma and 2650 Ma (Crowley *et al.*, 2005; Rasmussen
193 *et al.*, 2010). The discovery of Proterozoic detrital zircons (Cavosie *et al.*, 2004; Dunn *et al.*,
194 2005; Grange *et al.*, 2010) within the Jack Hills belt led to the recognition of Unit 4, which
195 hosts metasedimentary rocks deposited during the Proterozoic. Wang & Wilde (2018) observed
196 interbedded siliciclastic units at the same apparent metamorphic grade but with both Archean
197 and Proterozoic depositional ages, highlighting that the depositional and/or tectonic
198 relationships of units 3 and 4 may be more complex than previously postulated.

199 Metasedimentary rocks within Jack Hills show clear indications of deformation; intense
200 shearing is particularly evident within the anastomosing micaceous matrix, where a strong
201 foliation is coincident with flattening and recrystallisation of quartzite cobbles (Spaggiari,
202 2007a). Thermal or fluid events within the Jack Hills belt occurred at ~3080 Ma, 2650 Ma,
203 ~1850-1800 Ma and 800 Ma (Spaggiari, *et al.*, 2007b; Rasmussen *et al.*, 2010; 2011). Monazite-
204 xenotime thermometry of secondary inclusions within detrital zircon yield temperatures of 420-
205 475 °C (Rasmussen *et al.*, 2011), argued to represent peak upper greenschist to lower
206 amphibolite facies metamorphism within unit 3 metasedimentary rocks at ~2650 Ma
207 (Rasmussen *et al.*, 2010). However, like the depositional ages of metasedimentary units, the
208 exact P-T conditions and timing of metamorphism within unit 3 and 4 metasedimentary units
209 are ambiguous (Kemp *et al.*, 2018). A Proterozoic overprint at ~1800 Ma (Spaggiari *et al.*,

210 2007b; Rasmussen *et al.*, 2010) and a discrete event at 800 Ma (Rasmussen *et al.*, 2010; 2011)
211 are of unknown metamorphic grade, but also coincide with formation of monazite and xenotime
212 (Rasmussen *et al.*, 2010; 2011).

213 Metasedimentary rocks from Eranondoo Hill, chiefly oligomict pebble to cobble
214 metaconglomerate and quartzite, are renowned for hosting Hadean detrital zircon (Compston
215 & Pidgeon, 1986; Wilde *et al.*, 2001), with most studies focused on metasedimentary rocks at
216 the W-74 ‘discovery site’ (Fig. 2). Unsurprisingly, Jack Hills detrital zircons have been the
217 subject of numerous publications and reviews (e.g., Harrison *et al.*, 2017; Cavosie *et al.*, 2018).

218 While a detailed description of the zircons is beyond the remit of this publication, it is important
219 to note there is some debate surrounding the source of detrital zircon ^{207}Pb - ^{206}Pb age distribution
220 peaks, and thus the source of detritus. Whilst ^{207}Pb - ^{206}Pb age peaks correspond with major units
221 of the Narryer Terrane and the granitic lithologies surrounding Jack Hills (Nutman *et al.*, 1991;
222 Pidgeon & Wilde, 1998), distal sources of more intermediate composition have also been
223 suggested (Crowley *et al.*, 2005). Additionally, the protoliths of >3800 Ma zircons are
224 unknown, and may derive from a source exogenous to the Narryer Terrane. Sources external to
225 the Narryer Terrane therefore cannot be discounted for detrital chromites. If detrital phases are
226 derived from the Narryer Terrane, a further consideration is that both detrital zircon and
227 chromite are present within quartzite clasts of the metaconglomerates (e.g., Grange *et al.*, 2010;
228 Dare *et al.*, 2016). This demonstrates that the detrital phases have undergone multiple
229 sedimentary cycling events. It is therefore plausible that the source of detrital chromite had
230 already been completely or partially eroded at the time of deposition of the Jack Hills sediments
231 (3050 Ma to 2650 Ma), and therefore may not represent the distribution of mafic and ultramafic
232 crust during the late Archean (Crowley *et al.*, 2005; Rasmussen *et al.*, 2010).

233 Despite concerted interest in detrital zircon at Jack Hills, detrital chromites within the
234 same metasedimentary rocks have garnered less attention. Detrital chromites with low Mg#

235 (100x molar Mg/(Mg+Fe²⁺)) and elevated ZnO and MnO were reported by Cavosie *et al.* (2002)
236 for grains from the W-74 site; comparable major element compositions were described by Dare
237 *et al.* (2016) in analyses of detrital chromites ~1 km to the NW of this locality. Dare *et al.*
238 (2016) proposed that low MgO contents of chromites precluded derivation of Jack Hills detrital
239 chromites from komatiites, and that both chromite and Fe-Ni-sulphides observed within
240 quartzite clasts are the erosional products of at least one layered intrusion. Unpublished Re-Os
241 model ages (T_{MAs}) of 3500 Ma to 3200 Ma were reported by Valley *et al.* (2005) for the same
242 population of chromites described within Cavosie *et al.* (2002). These data suggest the analysed
243 detrital chromites are at least Palaeoarchean in age, and highlight the potential preservation of
244 robust Re-Os systematics through peak metamorphism within the Jack Hills.

245

246 METHODS AND MATERIALS

247

248 Sample collection and preparation

249 Sampling locations of metasedimentary rocks are shown in Fig. 2. Seven samples of pebble
250 metaconglomerates (14WA1-4, 16WA5-6 and 01JH54) were collected from within 10 m of the
251 W-74 site (Wilde *et al.*, 2001) at Jack Hills (Fig. 2b). A further sample of pebble
252 metaconglomerate (16WA8) and a quartzite (16WA7) were collected along strike
253 approximately 35m to the WSW. Two pebble metaconglomerates (16WA9-10) were sampled
254 from a prominent ridge across the valley to the NE. 01JH35 is a metaconglomerate sampled
255 ~800 m to the NE of the W-74 site. 01JH36, collected approximately 200m to the ENE of
256 01JH35, is a quartzite with discrete, mm-scale heavy mineral bands (Cavosie *et al.*, 2004).
257 Petrographic sample descriptions of metasedimentary rocks and their detrital mineral
258 assemblages are provided in supplementary material. While fuchsite-rich (muscovite with >1
259 wt. % Cr₂O₃; Challis *et al.*, 1995) metasedimentary rocks were largely collected to yield the

260 highest concentrations of chromite grains, 01JH36, 16WA6 and 16WA10 lack significant
261 fuchsite. Two chromite-bearing ultramafic rocks from within the Narryer Terrane are also
262 included, with descriptions and sampling locations provided within Supplementary Material.
263 Sample 13TKN22 is a metaperidotite from the 3730 Ma Manfred Complex, collected from NE
264 of Mt. Narryer, ~60 km SW of the Jack Hills. Sample 16WA13 is a heavily recrystallised
265 ultramafic rock of unknown age sampled from the SW limb of Jack Hills (Fig. 2a); this sample
266 is part of association 1 of Spaggiari *et al.* (2007a) and so is considered >3000 Ma.

267 Chips of Narryer Terrane ultramafic rocks were mounted in epoxy for petrographic
268 study and identification of chromite. 14WA, 16WA and 01JH35 chromites were separated
269 using standard crushing and separation procedures, then sieved to yield size fractions of ≤ 500
270 μm for analysis. 01JH36 and 01JH54 chromites were separated by electric pulse disaggregation
271 (EPD). 14WA1-4 heavy minerals were concentrated using the Wilfey table, followed by
272 magnetic and heavy liquid separation, while 16WA5-10 separates were concentrated via
273 panning. 01JH chromites were concentrated using heavy liquids and magnetic separation.
274 Chromites were picked by hand, separated per sample location, grain size and rounding shape,
275 and mounted in epoxy. Detrital grains within a thin section of 14WA2 were also analysed, as
276 were chromite liberated via HF leaching of quartzite cobbles from 14WA2 and 16WA5. No
277 chemical differences between chromites present within quartzite cobbles or matrix were
278 observed, so they are discussed together below.

279

280 Chromite mineral chemistry

281 *14WA, 16WA and 13TKN22*

282 Chromites were imaged using reflected light microscopy and/or back scattered electron imaging
283 using a Hitatchi S-3500N scanning electron microscope at the University of Bristol. Chromite
284 major elements were determined from individual spots and line scans using the Cameca SX100

285 electron microprobe at the University of Bristol. Chromium, Al, Fe, Mg, Zn, Mn, Ti, V, and Ni
286 abundances of chromite were determined using a 20kV accelerating voltage, a 10nA beam
287 current, 1 μm beam diameter, and a PAP matrix correction. Silicon, Na and Ca were included
288 within the set up to monitor any silicate contamination; analyses with >0.15 wt.% oxide of these
289 elements were omitted. Counting times were 30s for Cr, Fe, Al and Zn and 60s for Ti, Ca, Mg,
290 Na, Mn, V, Si and Ni, with the following standards used for instrument calibration: Cr₂O₃ (Cr),
291 albite (Al and Na), ilmenite (Fe and Ti), St. John's olivine (Mg and Si), Zn metal (Zn), Mn
292 metal (Mn), V metal (V), Ni metal (Ni) and wollastonite (Ca). The overlap of Ti K β on V K α
293 was corrected either via analysis of V-free SrTiO₃, or using high-resolution slits. No resolvable
294 variability in the V₂O₃ contents of chromite or chromite secondary standards was observed
295 between the two set-ups. Typical detection limits, expressed in ppm, were: Cr(260), Al(250),
296 Na(450), Fe(320), Ti(120), Mg(250), Si(180), Zn(400-500), Mn(300), V(200), Ni(350) and
297 Ca(140).

298 Four spinel standards (8316, 8311, 79-1 and 8315), well characterised for Fe³⁺/ Σ Fe
299 using both EPMA and Mössbauer from Wood & Virgo (1991), were analysed *in-run* to monitor
300 the integrity of EPMA measurements. Long term reproducibility of major elements (Al, Cr, Fe,
301 Mg) within 8316 and 8311, which possess the most comparable mineral chemistry to samples,
302 are <1.2 wt.% (2 σ). The method of Droop (1987) was employed to calculate the ferric iron
303 content of chromite by stoichiometry. The determined Fe³⁺/ Σ Fe (molar 100x Fe³⁺/(Fe³⁺+Fe²⁺))
304 of 8316 (0.21 \pm 0.07; 2 σ) and 8311 (0.17 \pm 0.06; 2 σ), measured across multiple analytical
305 sessions, are in good agreement with reported values. After correction for ferric iron, 14WA4,
306 16WA6 and 16WA9-10 yielded chromite with systematically low totals, typically between 97%
307 and 98%, despite other analyses within the same run yielding good totals. The cause of low
308 totals within these samples is unknown but may reflect non-stoichiometry, likely due to site
309 vacancies. Fe₂O₃ contents of chromite should therefore be considered minimum abundances for

310 14WA and 16WA samples. Chromites with totals of 97.5-102% from 14WA4, 16WA6 and
311 16WA9-10 are included, while measurements from other samples are restricted to those that
312 yield totals of 98-102%.

313

314 *01JH samples*

315 EPMA analyses were undertaken at the University of Wisconsin-Madison Department of
316 Geoscience using a CAMECA SX51 electron microprobe equipped with Probe for EPMA
317 software. Analyses were conducted at 20 kV, with a 20 nA beam current and a focused electron
318 beam. Count times for all elements were 10 seconds on peak and a total of 10 seconds on off-
319 peak background positions. The matrix correction was the PAP procedure. PHA differential
320 was used for Al and O, other elements were in integral mode. Oxygen was determined with a
321 60° 2d PC0 diffractor crystal, while Si, Al and Mg were acquired on TAP, Ti and Cr on PET,
322 and V, Mn, Fe, Ni, Cu and Zn on LIF. Overlap corrections were made on Al for Cr, Ni and Ti,
323 on V for Ti, on Fe for Mn, and on O for Cr. The following standards were used for instrument
324 calibration: USNM Chromite (Cr, Al, Mg, O), Harvard University hematite (Fe), synthetic
325 tephroite (Mn, Si), V₂O₅ (V), rutile (Ti), Ni₂SiO₄ (Ni), ZnAl₂O₄ (Zn) and Cu₂O (Cu).
326 Reproducibility, Detection limits.

327 EPMA analyses were processed using two methods. In the first, oxygen measurements
328 were omitted and analyses were processed using the same techniques as 14WA and 16WA;
329 namely, calculation of ferric iron by stoichiometry using the method of Droop (1987), and
330 exclusion of analyses with >0.15 wt.% SiO₂ and totals of <98 %. In the second method, cations
331 and measured oxygen were converted into molar proportions, and the ferrous/ferric ratio of iron
332 adjusted to balance the negative charge from measured O. This technique assumes Fe is the
333 only multivalent cation present (*e.g.*, Cr is present as Cr³⁺, V as V³⁺ and Mn as Mn²⁺) and that
334 site vacancies are negligible. The latter scenario is unlikely to be true within altered Cr-rich

335 spinel (see Kamperman *et al.*, 1996), and so ferric iron compositions determined by the second
336 method represent maximum values. For consistency, data calculated using stoichiometry are
337 presented within the main text, while analyses processed using measured O are discussed in
338 supplementary material.

339

340 Oxygen isotope analyses

341 Chromite grains from samples 01JH36, 01JH42 and 01JH54 were analysed for oxygen isotope
342 ratio by laser fluorination at the University of Wisconsin-Madison Department of Geoscience.
343 A total of 35 analyses of oxygen isotope ratio in chromite were made using gas source mass
344 spectrometry with BrF₅ and a 32 W CO₂ laser (Valley *et al.*, 1995). Chromite data were
345 corrected for accuracy with the UWG-2 garnet oxygen standard, which was analysed multiple
346 times at the beginning of the session (Valley *et al.*, 1995). The reproducibility of UWG-2 per
347 session ranged from ± 0.06 to ± 0.24 (2 σ). Samples of chromite were first sieved into seven
348 size fractions, ranging from 105-149 μm to $>500 \mu\text{m}$ (Supplementary Table 1). At least three
349 size fractions were analysed from each sample, with two to three aliquots analysed of each size
350 fraction. Aliquots of chromite were prepared by soaking detrital grains in concentrated HF at
351 room temperature overnight. Each ~ 2 mg aliquot of chromite thus consisted of hundreds of
352 clean detrital grains. Values are reported in standard per mil notation relative to V-SMOW
353 (Vienna Standard Mean Ocean Water).

354

355 RESULTS

356

357 Chromite morphology and inclusion assemblages

358 Jack Hills detrital chromites are $\leq 500 \mu\text{m}$ grains of variable morphology that are enclosed by,
359 or closely associated with, fuchsite within quartzite or metaconglomerate matrix (Fig. 3a). Finer

360 ($\leq 100 \mu\text{m}$) chromite grains have previously been reported within metaconglomerate quartzite
361 cobbles (Dare *et al.*, 2016), but were uncommon in samples of this study (Fig. 3b), though HF
362 leaching of quartzite cobbles liberated a small number of grains (n= 12). Many chromites show
363 textural evidence of sedimentary transport, including abrasion and rounding of broken surfaces.
364 Chromite derived from Jack Hills metasedimentary rocks occur as euhedral octahedra (EO),
365 rounded octahedra (RO) or rounded grains (RC: rounded chromite) (Fig. 3a). EO yield minimal
366 rounding on two or fewer faces, with many euhedral grains showing little or no evidence of
367 sedimentary transport. RC grains demonstrate limited evidence of original habit on two or less
368 faces, and are often present as highly spherical morphologies. RO are chromites with
369 morphologies intermediate to EO and RC, and are the most abundant morphology. More
370 quartzitic lithologies possess a greater proportion of euhedral grains. 16WA9 and 16WA10,
371 sampled furthest from the W-74 site, yield a more bimodal distribution with chromite
372 dominantly RC and minor EO.

373 A range of internal textures are present (Fig. 4), with grains often heavily cracked and
374 displaying ragged or lobate boundaries with surrounding fuchsite (Fig. 3c). Fractures
375 occasionally display distinct polygonal morphologies (Fig. 4b), and larger fractures may be
376 filled with quartz, fuchsite, and Fe-oxide. Distinct ‘pitted’ textural domains within chromite
377 reveal the presence of typically $\leq 10 \mu\text{m}$ inclusions of dominantly quartz and fuchsite (Figs.
378 4&5). Rare porous textures with $\geq 30 \mu\text{m}$ inclusion assemblages may occasionally account for
379 the majority of the grain (Fig. 4e-f). Both pitted and porous domains are conspicuous at
380 chromite rims or adjacent to cracks, and particularly where chromite is enclosed by fuchsite.
381 Fine laths of rutile, often showing apparent alignment to chromite crystallographic directions,
382 are also commonly associated with pitted and porous domains (Fig. 4e-f & 5a). Isolated
383 monomineralic and polyphase silicate and oxide inclusions present within chromites are
384 typically globular or anhedral in habit (Fig. 5), but may more rarely possess subhedral to

385 euhedral morphologies (Fig. 5a&d). Inclusions associated with fractures and crack filling
386 assemblages of quartz, fuchsite and rutile (Fig. 5a-b&e) are often accompanied by other fine-
387 grained (typically ≤ 10 μm) phases, including Fe-oxide (Fig. 5b), Fe-sulphide (Fig. 5c), and
388 rarer monazite (Fig. 5f).

389

390 Chromite mineral chemistry

391 Representative analyses of detrital chromites from each metasedimentary sample location are
392 presented in Table 1, and the full EPMA data set is available in Supplementary Material 2.
393 Chromite major element abundances show no variation with grain size or rounding shape
394 (Supplementary Figs.) but differ systematically between sampling locations (Figs. 6 & 7).

395

396 *Sample variability (Ti⁴⁺ and divalent cations; Fe²⁺, Mg²⁺, Zn²⁺, Mn²⁺, and Ni²⁺)*

397 Chromite from all sampled metasedimentary rocks yield elevated ZnO (up to 13 wt.%) and
398 MnO (up to 4.4 wt.%), coupled with low Mg# (typically ≤ 30 ; Fig. 6). Largely, samples with
399 increasingly elevated ZnO yield lower and more homogeneous Mg# (Fig. 6a). 14WA2
400 chromites possess the lowest wt.% ZnO (average ZnO 1.23 wt.%) and largest range of Mg#,
401 with a single grain (14WA2-PB-46) yielding Mg# of ~ 30 . Unusually elevated MnO contents
402 are present in 01JH35, where chromites yield > 3.5 wt.% MnO (Fig. 6b). No variation of MnO
403 with Mg# is apparent (Fig. 6b), though some 16WA samples show variance between ZnO and
404 MnO (Fig. 6d).

405 TiO₂ is typically < 1 wt.% (Figs. 6c). Individual analyses where TiO₂ is above 1 wt.%
406 correlate with areas where laths of rutile are present, and are omitted from consideration.
407 Elevated TiO₂ of ≥ 1 wt.% is most apparent within lower Cr# (100x molar Cr/(Cr+Al)) grains
408 (Fig. 7). As the bulk of analyses possess ≤ 0.3 wt.% TiO₂, this suggests analyses yielding TiO₂
409 > 0.3 wt.% may also overlap with rutile laths. However, many chromites with ~ 0.3 -0.8 wt.%

410 TiO₂ show no evidence of rutile laths in BSE images. NiO within grains is largely below the
411 quantification limit of ~0.04 wt.%.

412

413 *Sample variability (trivalent cations: Cr³⁺, Al³⁺, V³⁺, Fe³⁺)*

414 Despite considerable variability of divalent cation contents between sampling locations, detrital
415 chromite show a similar range in Cr# between samples (Fig. 7). Chromites yield moderate to
416 high, and variable Cr# of 48-82, with the bulk of chromites displaying Cr# of 54-66 (Fig. 7a).

417 The apparent smaller range of Cr# in chromites with higher wt.% ZnO is likely a sampling bias
418 due to fewer analyses. V₂O₃ was not measured for 16WA5-7, but where measured is present at

419 0.05 wt.%, to 0.37 wt.% and shows no inter-sample variability. Jack Hills detrital chromites
420 yield low calculated Fe₂O₃ contents; the bulk of the population contain ≤2 wt.% Fe₂O₃, though

421 analyses with up to 8 wt.% Fe₂O₃ are present. This results in highly variable Fe³⁺/ΣFe (100x
422 molar Fe³⁺/(Fe³⁺+Fe²⁺)) ratios of 0-20 (Fig. 7d). Many chromites yield non-stoichiometric
423 compositions (*e.g.*, deviation from A²⁺B³⁺₂O₄), shown by cation totals of less than 3 when

424 normalised to 4 oxygens. This likely reflects cation vacancies (*e.g.*, Kamperman *et al.*, 1996),
425 and results in an under-estimation of ferric iron contents by stoichiometric calculations (Droop,
426 1987). Non-stoichiometry also affects analyses conducted by direct measurement of oxygen

427 (Supplementary Material); analyses with significant cation deficiency results in substantial
428 over-estimation (>10 wt.%; Supplementary Material) of ferric iron contents to balance the
429 negative charge of oxygen. Thus, stoichiometric and charge balance of 01JH analyses represent
430 minimum and maximum ferric iron contents of Jack Hills detrital chromites, respectively.

431

432 *Within grain variability*

433 *Zoning*; While the largest systematic variations in chromite chemistry are between sample
434 location, line scans of 14WA and 16WA detrital chromites reveal distinct zonation trends (Fig.

435 8). Chromites do not exhibit morphologically or microstructurally distinct cores and rims, but
436 we use those terms here to describe the centre and outer portions of grains. Most chromites have
437 homogeneous ZnO contents (Fig. 8a) or else slightly lower wt.% ZnO towards rims (Fig. 8b-
438 c). MnO is largely uniform across chromites, although slightly lower abundances towards the
439 rim may be observed. Homogeneous Mg# appears to be restricted to chromites within high ZnO
440 16WA7 (Fig. 8c), with all other samples yielding chromites that display distinct variability in
441 Mg#. Low ZnO samples, such as 14WA2 and 16WA5, yield clear decreases in Mg# from the
442 core to rim of the grain (Fig. 8a). High ZnO samples, such as 14WA3 and many examples
443 within 14WA4, often yield rims with higher Mg# (Fig. 8b) though lowering of Mg# from core
444 to rim is also observed. Grains with elevated Mg# at the rims show increases in both MgO and
445 FeO from core to rim, compensated by decreases in ZnO and MnO.

446 While grains yield zoning profiles of divalent cations, intra-grain changes in Cr# are
447 largely absent (Fig. 8). Core-to-rim trends of marginally higher or lower Cr# (e.g., ± 2) are
448 sometimes observed (Fig. 8 a&c), but there is no systematic behaviour in samples from different
449 locations. Like Cr#, Fe₂O₃ contents are generally homogeneous across chromite grains, with
450 variability in some zonation profiles likely reflecting low calculated Fe₂O₃ contents. However,
451 individual grains do yield distinct variations in Fe₂O₃ that appear systematic in origin, present
452 as both decreases and enrichments in Fe₂O₃ from core to rim, with increasing Fe₂O₃ often
453 coupled with a small increase in Cr# (Fig. 8a).

454

455 *Coupled elemental and textural variation;* Distinct zones of elevated Cr# within detrital
456 chromite are rarely present (Fig. 9). These features are closely associated with pitted domains
457 and fractures, and exhibit both diffuse and more commonly sharp (Fig. 9a-e) boundaries with
458 surrounding chromite. High Cr# domains, often distinguished by an absence of fractures and
459 inclusions in comparison to surrounding chromite (Fig. 9a-b), mostly occur at the edge of grains

460 adjacent to enclosing fuchsite or associated with fractures (Fig. 9a-c). These domains may also
461 be localised, occurring as distinct polygonal areas defined by crystallographic directions (Fig.
462 9d-e). These domains are characterised by high Cr# (>70), low totals (<98 wt.%), the apparent
463 absence of Fe₂O₃, lower ZnO and V₂O₃, and occasionally slightly elevated MnO. High Cr#
464 domains within detrital chromite are chemically indistinguishable from low Fe₂O₃, high Cr#
465 grains within the same samples. Only a single domain of low Cr# has been observed within
466 chromite, bounding a monomineralic quartz inclusion alongside iron oxide (Fig. 9f).

467

468 Oxygen Isotope Compositions

469 Analysis of oxygen isotope ratio in various chromite size fractions was conducted to determine
470 if detrital chromite grains preserve primary (magmatic) $\delta^{18}\text{O}$ values, or if there is evidence of
471 oxygen exchange with the metasedimentary host rocks. Exchange of oxygen isotopes between
472 chromite and surrounding quartz at low temperature (e.g., <600 °C) would result in lowered
473 $\delta^{18}\text{O}$ values for chromite (e.g., Lowry *et al.*, 2003). Each sample showed >1 ‰ variability in
474 $\delta^{18}\text{O}$, with $\delta^{18}\text{O}$ chromite values ranging from 0.03 to 2.19 ‰ (Fig. 10; Supplementary Table
475 1). For samples 01JH36 and 01JH42, the lowest chromite $\delta^{18}\text{O}$ values (0.69 and 0.03 ‰,
476 respectively) were detected in the smallest size fraction, and there was a systematic increase of
477 $\delta^{18}\text{O}$ with grain size (Fig. 10a-b). In contrast, 01JH54 showed no systematic variation of
478 chromite $\delta^{18}\text{O}$ with grain size, and also contained the highest $\delta^{18}\text{O}$ measured for chromite at
479 2.19 ‰ (Fig. 10; Supplementary Table 1).

480

481

482 DISCUSSION

483

484 Chromite morphology and the origin of inclusion assemblages

485 The variably rounded morphologies of chromite, from RC to EO, attests to a protracted and
486 complex reworking history of numerous detrital grains. Further to this, the presence of chromite
487 within quartzite cobbles of metaconglomerates (Fig. 3b; Dare *et al.*, 2016) suggests cycling of
488 some grains in at least two sedimentation events. Categorisation of grains by shape may indicate
489 provenance from multiple geographic sources or differing modes of sedimentary transport
490 (Dott, 2003). The extensive sedimentary cycling history of some chromites is consistent with
491 the mature nature of host metasedimentary rocks (>98 wt.% SiO₂; Cavosie *et al.*, 2004), but at
492 odds with the highly fractured nature of many grains (Fig. 4b-e), which would not survive
493 extensive sedimentary transport. This strongly suggests that fracturing of grains occurred *in-*
494 *situ*, likely alongside deformation of the host metasedimentary rocks.

495 Jack Hills detrital chromites contain inclusion assemblages of quartz, fuchsite, rutile,
496 Fe-oxide, monazite, and Fe-sulphide (Fig. 5). These phases, particularly silicates quartz and
497 fuchsite, typically form under distinctly different conditions to magmatic chromite. However,
498 quartz and fuchsite are common components of Jack Hills quartzites and metaconglomerates,
499 alongside minor secondary rutile, monazite and xenotime (Harrison *et al.*, 2007; Rasmussen *et*
500 *al.*, 2010; Iizuka *et al.*, 2010). Inclusions of quartz and Cr-poor muscovite have also been
501 reported within detrital zircon (Cavosie *et al.*, 2004; Hopkins *et al.*, 2008; 2010; Rasmussen *et*
502 *al.*, 2011; Bell *et al.*, 2015), both as monomineralic and polyphase inclusions alongside
503 monazite and xenotime (Rasmussen *et al.*, 2011). Monazite and xenotime inclusions within
504 detrital zircons yield ²⁰⁷Pb-²⁰⁶Pb ages of ~2650 Ma and 800 Ma that are demonstrably younger
505 than the host grains (Rasmussen *et al.*, 2011), attesting to a secondary origin of many zircon
506 inclusion assemblages. Cameron *et al.* (2016) additionally showed that ~60 % of quartz
507 inclusions within zircon exhibited evidence of $\delta^{18}\text{O}$ exchange with Jack Hills metasedimentary
508 rocks. It is therefore hypothesised that many quartz inclusions within detrital zircons were
509 significantly altered during metamorphism of host metasedimentary rocks or, alongside

510 muscovite, rutile, monazite and xenotime, were precipitated from post-depositional fluids that
511 filled voids left from the dissolution of primary apatite (Rasmussen *et al.*, 2012; Bell *et al.*,
512 2015; Cavosie *et al.*, 2018; *c.f.* Hopkins *et al.*, 2010; 2012).

513 The occurrence of assemblages of the same phases as observed within detrital zircon,
514 coupled with the unlikelihood of quartz and muscovite stability within the mafic-ultramafic
515 melt that crystallised chromite, provides evidence that inclusion assemblages within chromite
516 are secondary in origin. Interestingly, many secondary inclusions of quartz and fuchsite are
517 apparently isolated from visible fractures within chromites (*e.g.*, Fig. 5a & 8a). Isolation of
518 inclusions has been used as a line of evidence for a primary origin of many quartz and muscovite
519 inclusions within detrital zircon (Bell *et al.*, 2015b). The detection of clearly secondary,
520 apparently isolated inclusion assemblages within detrital chromite suggests caution should be
521 applied when using this as a line of evidence for the presence of primary inclusions, and instead
522 points to the presence of sub-micron or annealed fractures within grains, or fracturing below
523 the 2D surface shown by electron imaging (*e.g.*, Cavosie *et al.*, 2018).

524 Our data therefore indicate that inclusion assemblages formed from interaction of
525 detrital chromite and metamorphic fluids within the metasedimentary host. Pitted and porous
526 domains, which are analogous to spongy textures described elsewhere (Gervilla *et al.*, 2012;
527 Colás *et al.*, 2014) and host the same secondary mineral assemblages, were also likely formed
528 by this process. Coarser, euhedral inclusions may reflect direct dissolution of primary inclusion
529 phases, akin to aforementioned replacement of primary inclusion assemblages within detrital
530 zircon (Rasmussen *et al.*, 2011). It is more problematic to prescribe an origin to inclusions of
531 Fe-sulphide and rutile. Sulphide and chromite are commonly co-liquidus phases under iron-
532 sulphide saturated magmatic conditions; however, secondary pyrite is often observed within
533 Jack Hills metasedimentary rocks (*e.g.*, Cavosie *et al.*, 2004; Supplementary Material) and iron-
534 sulphides (typically ≤ 40 μm pyrite, pyrrhotite, intergrowths of pyrite and pyrrhotite, and rarer

535 pentlandite) are ubiquitous within isolated quartzite cobbles (Dare *et al.*, 2016). Pyrrhotite,
536 pyrite and rare chalcopyrite inclusions within Jack Hills chromites appear to be closely linked
537 to pitted domains (Fig. 5c); we therefore propose a secondary origin for Fe-sulphide inclusions
538 within chromite.

539 Inclusions of rutile may be anhedral, globular or preserved as laths aligned to chromite
540 crystallographic axes and are particularly abundant within pitted and porous domains. Observed
541 alignment of alteration phases within chromite (Figs. 4e & 5a) is noted elsewhere, including in
542 chlorite (Fleet *et al.*, 1993; Gerville *et al.*, 2012) and phlogopite (Rollinson *et al.*, 2002).
543 Whether rutile was precipitated from metamorphic fluids, or was exsolved from chromite,
544 perhaps via changes in the oxidisation state of grains (e.g., Cameron *et al.*, 1979), is unclear.
545 High-Ti domains associated with cracks within zircon (Harrison & Schmitt, 2007; Hofmann *et*
546 *al.*, 2009), and the secondary growth of rutile within metasedimentary matrix (Harrison *et al.*,
547 2007) attests to Ti mobility within Jack Hills metasedimentary rocks. While we consider a
548 secondary origin of rutile most likely, we note direct exsolution from chromite that previously
549 had higher TiO₂ contents cannot be discounted.

550

551 Chromite mineral chemistry

552 *Signatures of metamorphism*

553 Jack Hills detrital chromites typically yield high ZnO and MnO coupled with low Mg#
554 (typically ≤ 30 ; Fig. 6). Chromites with such distinctive chemical compositions are uncommon
555 (e.g., Wylie *et al.*, 1987; Santti *et al.*, 2010; Fanlo *et al.*, 2015) and largely attributed to
556 secondary processes (e.g., Barnes, 2000). A number of studies have investigated the effects of
557 secondary modification of chromite major, minor and trace elements (e.g., Barnes, 2000;
558 González-Jiménez *et al.*, 2009; Mukherjee *et al.*, 2010; 2015; Gerville *et al.*, 2012; Colás *et al.*,
559 2014). Hypotheses as to the various processes that modify chromite include: elemental mobility

560 via interaction with aqueous fluids during the breakdown of surrounding Mg-Fe silicates via
561 serpentinisation (Burkhard, 1993; Marques *et al.*, 2007; Hodel *et al.*, 2017) and/or
562 metamorphism (Barnes, 2000; González-Jiménez *et al.*, 2009; Gerville *et al.*, 2012; Colás *et*
563 *al.*, 2014; Fanlo *et al.*, 2015; Ahmed & Surour, 2016), Cu-Zn-Ni-(Co) sulphide mineralisation
564 of the host rock (Wylie, 1987; Marques *et al.*, 2007; Fanlo *et al.*, 2015), and magmatic sulphide
565 mineralisation (Groves *et al.*, 1977). The effects of these processes on chromite chemistry are
566 dependent on the temperature and longevity of alteration, the nature and $f\text{O}_2$ of the aqueous
567 medium, the composition of the host rock and subsequent chromite/silicate ratio, and the
568 fluid/rock ratio during modification (Colás *et al.*, 2014; Ahmed & Surour, 2016). The most
569 commonly observed divalent cation mobility during thermal events is a decrease in Mg#, as
570 Fe^{2+} diffusively enters the chromite lattice at the expense of Mg^{2+} (Barnes, 2000; Colás *et al.*,
571 2014). ZnO , MnO and to a lesser extent CoO , also diffuse into the chromite lattice in exchange
572 for MgO , NiO and often TiO_2 (Barnes, 2000; Colás *et al.*, 2014).

573 Within Jack Hills detrital chromites, coupled growth in ZnO content with increasingly
574 homogenised and lowered Mg# likely signifies greater exchange with the modifying medium
575 (Fig. 6a). This may indicate variable modification within their protolith; however, analogous
576 signatures of alteration have been previously reported by Challis *et al.* (1995), where detrital
577 chromites within fuchsite sedimentary rocks yielded elevated ZnO (up to 13.7 wt.%) and MnO
578 (up to 3.5 wt.%), coupled with low Mg#. Critically, the detrital grains described by Challis *et*
579 *al.* (1995) possessed a well constrained provenance, demonstrating the observed signatures
580 could only have formed during post-depositional metamorphism.

581 We propose that the dominant signatures of secondary modification within Jack Hills
582 detrital chromites occurred from exchange of grains with metamorphic fluids and/or
583 assemblages during metamorphism of their metasedimentary hosts. Firstly, as established
584 above, interaction of chromite with metamorphic fluids is shown by replacement of inclusion

585 assemblages (Fig. 5) and the development of pitted and porous domains. Secondly, large
586 differences in chromite chemistry between samples does not relate to the physical
587 characteristics of grains, such as shape or grain size (Supplementary Figs.), but are a function
588 of sampling site and therefore sedimentary horizon. Homogeneity of Mg# (e.g., 16WA7 &
589 16WA9; Fig. 6) requires modification under the same physiochemical conditions, including
590 temperature, fO_2 and Mg# of modifying medium, despite the probable variation in sedimentary
591 transport characteristics. Observed Mg# uniformity is therefore difficult to reconcile with
592 metamorphic modification with an ultramafic or mafic protolith.

593 Finally, and most significantly, chromites yield substantial inter-sample chemical
594 variability despite the proximity (often <5 m) of sampling locations (Fig. 2), with the meter-
595 scale variability of ZnO and Mg# of detrital chromites reconciled by changes in grain size and
596 modal proportions of chromite within the host metasedimentary rocks (14WA1-4;
597 Supplementary Material). Chromites are the sole ZnO bearing phase, but are present at low
598 modal proportions of <1 %. As such, minor changes in the modal proportion and grain size of
599 detrital chromites can account for large variations of wt.% ZnO within detrital chromites during
600 interaction and equilibration with secondary fluids. However, 01JH36 detrital chromites,
601 sampled from a heavy mineral layer within which chromites represent a significant component,
602 yield elevated ZnO (2.02 to 4.74 wt.%) within 01JH36 detrital chromites. Simple changes in
603 the modal proportion of chromite therefore cannot account for the ZnO content of grains within
604 this sample. Instead, 01JH36 chromite mineral chemistry may be reconciled by metamorphic
605 equilibration of ZnO at length scales greater than heavy mineral bands (\geq mm-scale) and, given
606 the distance of this sample from the W-74 site (Fig. 2), variable physiochemical compositions
607 of metamorphic fluids across the Jack Hills. Thus, variability in modification signatures may
608 also reflect changing parameters of metamorphic fluids, as also suggested by the often poor
609 correlation between Mg# and MnO with the modal proportion of detrital chromite.

610

611 *Oxygen isotope compositions*

612 *In-situ* modification of chromites is also suggested by oxygen isotope compositions. The
613 variation of $\delta^{18}\text{O}$ with grain size in chromites from samples 01JH36 and 01JH42 (Fig. 10a-b)
614 indicates they experienced partial exchange of oxygen isotopes with the metasedimentary host
615 rocks. If exchange was by diffusion inwards from the grain boundary, then greater modification
616 of smaller grains is expected given their greater surface area to volume ratio. The lower $\delta^{18}\text{O}$
617 values of the smallest chromite grains from 01JH36 and 01JH42 may therefore approach
618 equilibration with host quartz during metamorphism or retrogression. In contrast, chromite
619 grains in sample 01JH54 show no grain size dependence on $\delta^{18}\text{O}$ (Fig. 10c). Given the
620 conspicuously lower chromite $\delta^{18}\text{O}$ values recorded in 01JH35 and 01JH36, we tentatively
621 interpret chromite $\delta^{18}\text{O}$ values in sample 01JH54 as pre-depositonal. Low $\delta^{18}\text{O}$ ($\leq 2.5\text{--}3.0\text{ ‰}$)
622 are reported within chromitite from ophiolites and layered intrusions due to exchange with late-
623 stage magmatic and sub-solidus fluids (Lowry *et al.*, 2003; Schannor *et al.*, 2018).

624 The oxygen isotope ratios for smallest chromite size fractions in samples 01JH36 and
625 01JH42 were compared with previously published $\delta^{18}\text{O}$ values for quartz in the same samples
626 (Supplementary Table 2; Cavosie *et al.*, 2005). Cavosie *et al.* (2005) found that values of
627 $\delta^{18}\text{O}(\text{Qz})$ are homogeneous at hand sample scale in these rocks, indicating metamorphic
628 equilibration of quartz in pebbles and matrix. If attainment of oxygen isotope equilibrium
629 between quartz and chromite is assumed, the $\Delta^{18}\text{O}$ value ($\Delta^{18}\text{O}_{\text{Quartz-Chromite}} = \delta^{18}\text{O}_{\text{Q}} - \delta^{18}\text{O}_{\text{C}}$) for
630 samples 01JH36 ($\Delta^{18}\text{O}_{\text{Q-C}} = 12.02\text{ ‰}$) and 01JH42 ($\Delta^{18}\text{O}_{\text{Q-C}} = 12.70\text{ ‰}$), can be applied to
631 calculate the temperature at which the observed fractionations were established (Zheng *et al.*,
632 1991). Calculated temperatures are 412 °C and 390 °C for samples 01JH36 and 01JH42,
633 respectively. Sample 01JH54 yields a distinctly smaller fractionation ($\Delta^{18}\text{O}_{\text{Q-C}} = 9.11\text{ ‰}$) that
634 would correspond to a temperature of 530 °C if it represented metamorphic equilibration.

635 However, as discussed above, the analysis of different sized grains suggests that chromite grains
636 in this sample preserve pre-depositonal $\delta^{18}\text{O}$ from and may not be appropriate for metamorphic
637 thermometry. While it cannot be demonstrated that oxygen isotope equilibrium was established
638 between quartz and chromite based on $\delta^{18}\text{O}$ analysis of 2 mg aliquots in the other two samples,
639 we note that the derived temperatures based on measured fractionations are consistent with
640 greenschist facies metamorphic conditions previously documented within the Jack Hills
641 (Cavosie *et al.*, 2004; Spaggiari *et al.*, 2007b).

642 Quartz in the metasedimentary rocks analysed has been shown to have recrystallized
643 during metamorphism, and yields homogeneous $\delta^{18}\text{O}$ values at >cm scales (Cavosie *et al.*,
644 2005). In contrast, detrital chromites in the same samples preserve detrital morphologies, likely
645 preserve gradients in $\delta^{18}\text{O}$ at sub-mm scale (Fig. 10), and are not equilibrated in $\delta^{18}\text{O}$. We
646 therefore use the $\delta^{18}\text{O}$ values determined for quartz (Cavosie *et al.*, 2005) and temperatures
647 determined from oxygen isotope thermometry to estimate the $\delta^{18}\text{O}$ value of fluids present
648 during metamorphism (Supplementary Table 3). Calculated water $\delta^{18}\text{O}$ values using quartz-
649 water fractionation factors of Sharp *et al.* (2016) and the published quartz $\delta^{18}\text{O}$ values yields
650 water $\delta^{18}\text{O}$ values of 8.58 ‰ (at 412 °C) for samples 01JH42 and 8.16 ‰ (at 390 °C) for sample
651 01JH36; a value of water $\delta^{18}\text{O}$ of 8.50 ‰ (at 530 °C) for sample 01JH54 is comparable to the
652 other two samples, but may be fortuitous given the lack of evidence for equilibration in sample
653 01JH54 (Fig. 10). Regardless, the range of all calculated water $\delta^{18}\text{O}$ values, from 8.2-8.6 ‰,
654 are typical for magmatic or metamorphic fluids (Sheppard, 1986).

655

656 *Trivalent cations: retained primary signatures?*

657 While divalent cations exchange more readily during metamorphism, trivalent cations appear
658 largely immobile during low temperature modification (e.g., Barnes, 2000). Chromite
659 interacting with acidic fluids during sea-floor hydrothermal metasomatism may rarely induce

660 trivalent cation exchange (Wylie *et al.*, 1987; Marques *et al.*, 2007; Hodel *et al.*, 2017); more
661 generally, however, nascent trivalent mobility occurs at greenschist facies metamorphism
662 (Kimball *et al.*, 1990; González-Jiménez *et al.*, 2009) and becomes increasingly pervasive at
663 and above amphibolite facies (Barnes, 2000; González-Jiménez *et al.*, 2009; Colás *et al.*, 2014).
664 Under oxidising conditions, trivalent cation exchange most commonly facilitates the transition
665 of chromite to ferritchromit $[Fe^{2+}(Fe^{3+},Cr)_2O_4]$, where Al_2O_3 is lost to Fe_2O_3 either diffusively
666 (Wylie *et al.*, 1987; Gerville *et al.*, 2012; Colás *et al.*, 2014) or via reaction with magnetite rims
667 (Evans & Frost, 1975; Barnes, 2000). Extensive formation of ferritchromit and Cr-magnetite,
668 and thus modification of chromite core compositions, appears to be largely restricted to
669 amphibolite facies metamorphism and above (Barnes, 2000).

670 Regardless of the degree of secondary modification of divalent cations, Jack Hills
671 detrital chromites possess a broad but consistent range in Cr# across all metasedimentary
672 sample locations (Fig. 7). The range of Cr# may be explained by either the absence of
673 significant trivalent mobility, or the equilibration of chromite during secondary modification
674 with a Cr-Al bearing equilibrator (*e.g.*, fuchsite or metamorphic fluid). If secondary exchange
675 was significant for trivalent cations, the latter scenario should manifest as homogenisation to
676 lower or higher Cr# in increasingly Zn-rich samples. This is not observed. Additionally,
677 zonation profiles of variable Mg# but homogeneous Cr# within low ZnO chromites (Fig. 8a)
678 would require more exchange, and thus faster diffusion, of trivalent than divalent cations. Such
679 an order of relative diffusivities is in strong contrast to previous literature (*e.g.*, Barnes, 2000),
680 and considered extremely unlikely.

681 Unmodified Cr# within chromites is at apparent odds with the presence of surrounding
682 fuchsite, whose elevated Cr content is almost certainly derived from detrital chromites. Fuchsite
683 yields Cr_2O_3 concentrations of ~2-3 wt.% (Cr# < 5; Supplementary Table 4), that decrease with
684 increasing distance from detrital chromite grains (*e.g.*, Rasmussen *et al.*, 2011). Such low

685 concentrations within fuchsite require only minor mobilisation of Cr from chromite, and
686 suggest that the Cr required was scavenged from more heavily altered areas of chromite during
687 metamorphic precipitation of fuchsite, or from dissolution of strongly modified grains in acidic
688 fluids. This is supported by lobate boundaries of detrital chromites with surrounding fuchsite
689 (Fig. 9a-b). Selection of grains with well-defined shapes has likely filtered our dataset for such
690 heavily modified compositions; the process of fuchsite formation causes ragged grain
691 boundaries and extensive fracturing of chromites, which significantly modifies grain
692 morphology (Fig. 3).

693 Localised zones of Cr-Al mobility may, however, shown by high Cr# domains (Fig. 9),
694 with their generation reconciled via preferential loss of Al during localised exchange between
695 chromite and fuchsite and/or the metamorphic fluid that precipitated fuchsite. High Cr#, low
696 Fe³⁺# and inclusion free zones of alteration have previously been documented by Arai *et al.*
697 (2006), who suggested such domains represent the loss of Al³⁺ without concurrent addition of
698 Fe³⁺. Alternatively, high Cr# domains may reflect compositions transitional to ferritchromit or
699 fluid-mediated recrystallisation, although further microstructural analysis would be required to
700 show this. The distinct boundary between the bulk chromite and high Cr# domains (Fig. 9) was
701 suggested by Arai *et al.* (2006) to represent a spinel miscibility gap at high-Cr, Low-Fe³⁺
702 conditions. Importantly, the sharp gradients indicate negligible diffusion between high Cr#
703 domains and bulk chromites. Therefore, while high Cr# domains clearly represent areas of
704 elevated alteration during metamorphism of host metasedimentary rocks, the retention of sharp
705 boundaries implies an absence of trivalent cation mobility within bulk chromites.

706 High Cr# domains yield strong compositional similarities to high Cr# chromite, which
707 creates increased uncertainty as to the primary origin of high Cr# grains. However, inter-grain
708 variability in Mg# is still apparent within 14WA2, and the zonation profile of high Cr# 14WA2-
709 PB-12 (Cr# ~75) shows Mg# variability (Supplementary Material Fig. XX); if elevated Cr#

710 were a product of enhanced trivalent mobility, Mg# heterogeneity should not be observed. This,
711 coupled with the sharp boundaries of high Cr# domains, provides strong evidence that chromite
712 with consistently high and homogeneous in Cr# are pre-depositional in origin.

713

714 Timing and cause of chromite modification

715 It is apparent from the mineral chemistry and inclusion assemblages of Jack Hills detrital
716 chromites that grains were modified within host metasedimentary rocks. However, the timing
717 and magnitude of metamorphic events within the Jack Hills are poorly understood (*e.g.*, Kemp
718 *et al.*, 2018). Previous work has attested to three thermal or fluid flow events within Jack Hills
719 metasedimentary rocks at ca. 2650 Ma, ~1850-1800 Ma and 800 Ma (Spaggiari, 2007a;
720 Rasmussen *et al.*, 2010; 2011; Kemp *et al.*, 2018). Below, we deliberate the timing and
721 magnitude of metamorphic events experienced by Jack Hills detrital chromites.

722

723 *Dominant metamorphic signatures; a ~2650 Ma event*

724 Metamorphism at ~2650 Ma has been proposed to represent peak metamorphic conditions at
725 the W-74 site (EARLIER PAPER; Rasmussen *et al.*, 2010; 2011). While inclusion assemblages
726 within Jack Hills chromites have not been dated, and so cannot be definitively linked to this
727 event, previous investigation of secondary monazite and xenotime within detrital zircons from
728 the W-74 site observed a large population of this age (Rasmussen *et al.*, 2011). Monazite-
729 xenotime Gd exchange and Ti-in-quartz thermometry indicates metamorphic temperatures of
730 <487 °C at this time, typical of upper greenschist facies metamorphism (Rasmussen *et al.*,
731 2011). Importantly, temperatures calculated for oxygen isotope equilibrium between chromite
732 and quartz fall within the temperatures anticipated for greenschist facies metamorphism, and
733 also argue for exchange at this time. Given that the ~2650 Ma event represents the highest grade
734 of metamorphism experienced by metasedimentary rocks local to the W-74 site, this event

735 likely imposed the observed chromite mineral chemistry. The chemical compositions of Jack
736 Hills chromites can therefore be explained by partial to complete exchange with Fe-, Zn- and
737 Mn-bearing metamorphic fluids and mineral assemblages during upper greenschist to lower
738 amphibolite facies metamorphism at ca. 2650 Ma. Metamorphic fluids were likely acidic to
739 facilitate the mobility of Ti^{4+} (e.g., van Baalen, 1993), and enable localised dissolution of
740 chromite to form fuchsite.

741 This event also coincides with amphibolite to granulite metamorphism throughout the
742 Narryer Terrane, including gneisses adjacent to the belt, and ubiquitous granitic emplacement
743 within the entire Yilgarn Craton (Kemp *et al.*, 2018). This includes the ca. 2650 Ma
744 monzogranite ('The Blob granite'; Pidgeon & Wilde, 1998), which intrudes the Jack Hills belt
745 to the SW of Eranondoo Hill (Fig. 2). Speculatively, this intrusion, or other granitic intrusions
746 adjacent to the Jack Hills belt, may have supplied heat and a metasomatic component
747 (particularly volatile Zn) to metamorphic fluids; similarly modified detrital chromites reported
748 by Challis *et al.* (1995) were postulated to have gained their chemical compositions by
749 interaction with fluids derived from local granitic plutons. An alternative and more local source
750 of ZnO within Jack Hills metasedimentary rocks may come from dissolution of sphalerite
751 (ZnS), which may be a potential detrital phase given the abundance of Fe-sulphides within
752 quartzite cobbles.

753 Studies have shown that metasedimentary rocks with Proterozoic depositional ages are
754 complexly intercalated with, and at the same apparent metamorphic grade, as metasedimentary
755 associations with inferred Archean depositional ages (Cavosie *et al.*, 2004; Dunn *et al.*, 2005;
756 Grange *et al.*, 2010; Wang & Wilde, 2018). This, coupled with ~1850-1800 Ma and ~800 Ma
757 ages within the Jack Hills (Spaggiari *et al.*, 2007b; Rasmussen *et al.*, 2010; 2011), indicates
758 Proterozoic overprinting at a lower metamorphic grade than peak metamorphic conditions.

759 Subtle zonation trends within some chromites, namely increased Mg# within 14WA4 (Fig. 8b),
760 may be reconciled by minor elemental mobility (Mg, Fe, Zn) within the Proterozoic.

761

762 *Pre-depositional modification?*

763 Given the susceptibility of chromite to sub-solidus exchange, signatures of modification prior
764 to deposition must also be considered. Divalent cations have undergone significant exchange
765 within host metasedimentary rocks; it is unclear if preservation of high Mg# within the cores
766 of some low ZnO grains (e.g., 14WA2-PB-46; Fig. 6b) represent pre-depositional Mg#, or have
767 still undergone partial exchange. However, pre-depositional metamorphism of accessory
768 chromites (e.g., within their protolith), should result in trivalent cation mobility namely the
769 development of ferritchromit rims (Barnes, 2000; Colás *et al.*, 2014). While the generation of
770 ferritchromit is limited within chromitites, where the higher modal abundance of chromite in
771 chromitite has been shown to inhibit significant metamorphic exchange and thus the generation
772 of ferritchromit (e.g., Mukherjee *et al.*, 2010), chromitite typically represent relatively minor
773 components of igneous bodies (Barnes & Roeder, 2001). Given the presence of EO, which have
774 undergone little abrasion and rounding during sedimentary reworking, ferritchromit or
775 magnetite rims should be clearly visible if present prior to deposition. No ferritchromit has been
776 observed within this study, providing evidence chromites have not undergone resolvable
777 trivalent mobility prior to erosion from their protoliths. While we cannot definitively exclude
778 trivalent cation mobility prior to deposition, we suggest that Cr# of Jack Hills detrital chromites
779 represent the most robust chemical signature for investigation of provenance.

780

781 Provenance

782 While detrital chromites exhibit protracted metamorphic histories, some chemical
783 signatures place constraints on the magmatic protolith. Consistent variability of Cr# regardless

of shape may indicate a common mafic or ultramafic protolith. While we base this interpretations on least altered chromites within samples 14WA2 and 01JH54, the consistent range of chromite Cr# from all 14WA, 16WA and 01JH metasedimentary rocks analysed potentially suggests that a single, large-scale magmatic source potentially dominated the mafic and ultramafic crust within the catchment of the Jack Hills sedimentary rocks. Unfortunately, important chemical tools for understanding the petrogenesis of chromite and consequently their protolith, namely Mg# and TiO₂, are not appropriate to apply to Jack Hills detrital chromites due to modification within host metasedimentary rocks. Therefore, commonly used plots for chromite provenance, such as TiO₂ vs Cr₂O₃ or Al₂O₃ (Kamenetsky *et al.*, 2001), are unlikely to be reliable. We propose trivalent cation mobility was limited within Jack Hills detrital chromites, but cannot fully exclude metamorphic re-equilibration of Fe³⁺/ΣR³⁺ prior to deposition. We therefore suggest careful use of Fe³⁺/ΣR³⁺ may provide better insights into the provenance of Jack Hills chromites; unmistakably non-stoichiometric grains (*i.e.*, Fe³⁺ = <0) from 14WA2 are excluded Fig. 13 for clarity. The retention of unmodified Cr# within cores of Jack Hills chromite will therefore yield the most information on provenance of detrital grains, alongside careful and conservative use of Fe³⁺/ΣR³⁺ and, where rutile laths are demonstrably absent, TiO₂ contents.

To gain understanding of the petrogenesis of Jack Hills detrital chromites, and potentially of mafic and ultramafic crust within the Narryer Terrane, we use the fields of Barnes & Roeder (2001), a comparative study of over 26,000 spinels from different tectonic settings. We compare these fields to grains from 14WA2 and 01JH54, which, due to low ZnO coupled with partial exchange of Mg#, we interpret to represent the least altered compositions of detrital chromite. We do not include the chemical compositions of chromitites within the fields of Figs. 10-13; chromitites comprise a low volumetric component of igneous bodies, particularly layered intrusions, and have been suggested to induce bias within provenance studies (Power

809 *et al.*, 2000; Barnes & Roeder, 2001). While 14WA2 appear conform to a Cr-Al trend (Fig. 11;
810 Irvine, 1967; Barnes & Roeder, 2001), where no enrichment of Fe^{3+} is observed with changing
811 Cr#, this likely reflects non-stoichiometry. The consistently low ferric iron contents of 01JH54
812 chromites are not suggestive of Fe-Ti or Rum trends (Fig. 11), which form via the reaction of
813 plutonic chromites with intercumulus liquid enriched in both Fe^{3+} and Ti.

814 As expected, in Cr# vs. $\text{Fe}^{2+}\#$ (inverse Mg#) compositional space, Jack Hills detrital
815 chromites fall within the 90th percentile of spinel within high grade metamorphic rocks (Fig.
816 12a). The decrease of Mg# within Jack Hills detrital chromites due to metamorphic exchange
817 results in comparable Mg# to chromites modified within mafic-ultramafic protoliths; this
818 process is also shown in Fig. 12d, where an increase in metamorphic grade from greenschist to
819 amphibolite facies results in significantly lowered and homogenised Mg# within komatiitic
820 chromites. While superficially Jack Hills detrital chromites yield a similar Cr# variability to
821 those modified by metamorphism, the range in Cr# shown within the metamorphic field (Fig.
822 12a) represents compositions transitional from magmatic Cr# to ferritchromit and magnetite.
823 As such alteration phases are absent within Jack Hills chromites, and zonation profiles show
824 limited Cr# variability, this process cannot explain the fit of grains to the metamorphic field of
825 Barnes & Roeder (2001). A different provenance is therefore required for the observed Cr#.

826 The variable and moderate to high Cr# of Jack Hills detrital chromites (typically 55-70)
827 are comparable to ophiolitic chromites (Fig. 12b), continental layered intrusions (Fig. 12c), and
828 intra-plate settings such as ocean island tholeiites and oceanic plateaus (Dick & Bullen, 1984;
829 Barnes & Roeder, 2001; Arai *et al.*, 2011). Critically, the Cr# of Jack Hills chromites indicate
830 the majority of detrital grains cannot be derived from komatiites (Fig. 12d). As a product of
831 high temperature and high degree mantle melts, komatiites have high Cr/Al ratios and therefore
832 yield chromite with high Cr# (typically >70; Barnes & Roeder, 2001) that yield tightly clustered
833 compositions due to the inhibition of magmatic fractionation or subsolidus equilibration. In

834 comparison to both Al-undepleted komatiites (AUDK; Fig. 12d) and Al-depleted komatiites
835 (ADK), Jack Hills detrital chromites yield more variable and moderate Cr# (typically 55-70).
836 ADK are not shown in Fig. 12 as they yield chromite with too high Cr# (>85) to represent a
837 source of Jack Hills detrital chromite. Some portions of komatiitic lavas, such as olivine-rich
838 dunitic channels and sheets, have been shown to yield chromite with lower Cr# (60-70) than
839 bulk ADK and AUDK (Barnes, 1998; Barnes & Roeder, 2001). However, chromite is not
840 abundant in these reduced cumulates, and would be unlikely to dominate the detrital record
841 over their bulk komatiitic counterparts.

842 We suggest that, on the basis of Cr#, the most likely provenance of Jack Hills detrital
843 chromites is a layered intrusion. Notably, though chromitite fields are not included in Fig. 12
844 and Fig. 13, Jack Hills detrital chromites yield comparable Cr# to Stillwater 'G' and Bushveld
845 chromitites (e.g., Campbell & Murck, 1993; Langa et al., 2020 and references therein). While
846 variable chromite morphologies may reflect different transport mechanisms, an absence of
847 chemical variation with rounding shape of grains may be further qualitative evidence of a
848 potential layered intrusion source; layered intrusions may yield consistent lithologies and
849 mineral chemistry laterally for many kilometres. In the absence of definitive Fe_2O_3 or TiO_2
850 concentrations, tectonic settings with comparable Cr#, including podiform and other mantle
851 derived Cr-spinel, cannot be excluded (as shown by ophiolites; Fig. 13a). Thus, while we
852 consider a layered intrusive source the most likely source of Jack Hills detrital chromites,
853 further investigation of additional chemical signatures may provide evidence for or against this
854 interpretation.

855 However, chromite from early Archean anorthositic layered intrusions yield marked
856 physiochemical differences to those from stratiform complexes, which dominate the layered
857 intrusion fields of Fig. 12 and Fig. 13. Archean anorthositic layered intrusions typically form
858 sill-like bodies that are significantly less volumetric than Proterozoic massif-type anorthositic

859 intrusions, and are thought to represent sub-volcanic intrusions coeval to overlying greenstone
860 belts (e.g., Ashwal & Bybee, 2017). Archean sill-like anorthositic intrusions yield spinel that
861 are typically Fe rich, with populations of high Cr# (>70) or low Cr# (<50) (Rollinson *et al.*,
862 2002; 2010; Mondal *et al.*, 2006; Dharma Rao *et al.*, 2013; Mukherjee *et al.*, 2015; Rowe &
863 Kemp, 2020), thought to reflect derivation from parental melts of komatiitic, boninitic or
864 hydrous basaltic compositions (Ashwal & Bybee, 2017). Thus, typical mineral chemistry of
865 anorthositic layered intrusive chromites are not comparable to Jack Hills chromites. While
866 chromites extending to Cr# comparable to Jack Hills detrital chromites are documented (e.g.,
867 Mukherjee *et al.*, 2010), they typically represent a minor component of bulk chromite
868 compositions. As with chromites from komatiitic dunitic channels, such chromites are unlikely
869 to dominate detrital records over higher Cr# counterparts.

870 While Jack Hills detrital chromites do not yield Cr# equivalent to Cr-spinel from sill-
871 like Archean anorthositic layered intrusions, chromite from late Archean layered intrusions
872 with comparable chemical compositions are known (e.g., Rollinson *et al.*, 2010; Berger *et al.*,
873 2013; Szilas *et al.*, 2017). The development of massif-type anorthositic intrusions at the
874 beginning of the Proterozoic are thought to coincide with the emergence of increasingly
875 thickened and strengthened lithosphere, a consequence of secular cooling and potentially
876 shifting geodynamic regimes (Ashwal & Bybee, 2017). This therefore does not exclude the
877 presence of more massif-type layered intrusions during the early Archean; the preservation
878 potential of larger, and thus higher density, stratiform intrusions would be significantly lower
879 in the presence of hotter Archean mantle.

880

881 Potential sources within the Narryer Terrane

882 As the mineral chemistry of Jack Hills detrital chromites allude to a layered intrusion origin,
883 and unpublished Re-Os isotopic data yield model ages that are at least Palaeoarchean in age

884 (Valley *et al.*, 2005), the sole described layered intrusion within the Narryer Terrane, the 3730
885 Ma Manfred Complex (Kinny *et al.*, 1988; Fletcher *et al.*, 1988; Myers, 1988b; Rowe & Kemp,
886 2020), can be considered as a potential source. Spinel within the Manfred Complex is
887 dominantly spinel *sensu stricto* or picotite, and occurs closely associated with magmatic olivine
888 (Rowe & Kemp, 2020) or exsolved from cumulus orthopyroxene or clinopyroxene. Chromite
889 is completely absent in the Manfred Complex harzburgites and websterites studied by Rowe &
890 Kemp (2020); however, metaperidotites within the Manfred Complex contain chromite
891 (Fletcher *et al.*, 1988; Myers, 1988b; Kemp *et al.*, 2018). Here, we compare the least altered
892 14WA2 and 01JH54 detrital grains with chromites from Manfred Complex sample 13TKN22
893 (Supplementary Material), a pyroxene-phyric peridotite metamorphically modified to
894 hornblende and altered to serpentine.

895 13TKN22 chromites (n=13 analyses) occur as disseminated crystals with variably
896 formed magnetite rims (Fig. 14a-b), and yield consistent Cr# of 76-80 (Table 2), coupled with
897 low Mg# of <20, likely reflecting metamorphism of the host metaperidotite. 13TKN22
898 chromites yield limited ZnO (~0.6 wt.%) and heterogeneous MnO (0.4-1.4 wt.%). NiO is
899 present at magmatic concentrations of >0.1 wt.%, but this may reflect the high magnetite
900 component of 13TKN22 chromites (Barnes, 1998). While divalent cation mineral chemistry
901 suggests metamorphic alteration of 13TKN22 chromites, the sharpness of the boundary
902 between chromite cores and magnetite rims (Fig. 14a-b) and the tightly clustered compositions
903 of cores (Fig. 14d-e) indicate restricted trivalent mobility (*e.g.*, Barnes, 2000). While the highest
904 Cr# Jack Hills detrital grains overlap with the Cr# of 13TKN22 chromites (Fig. 14), detrital
905 chromites possess significantly lower ferric iron contents and TiO₂, though given the
906 uncertainties introduced by Ti mobility within host metasedimentary rocks comparisons should
907 be taken tentatively. 13TKN22 chromites therefore cannot account for the bulk population of
908 Jack Hills chromite detrital grains, which possess lower Cr#, and these lithologies did not

909 represent the dominant component of mafic to ultramafic crust within the erosional catchment
910 of Jack Hills metasedimentary rocks.

911 While mafic and ultramafic crust within the Narryer Terrane tends to be collectively
912 grouped as the Manfred Complex, more recent geochronological investigations have shown
913 many of these lithologies actually have Paleoarchean igneous crystallisation ages (Kemp *et al.*,
914 2018). Continued investigation of these units may reveal further insights into the provenance
915 of Jack Hills detrital chromites. Mafic and ultramafic rocks that occur within the Jack Hills belt
916 may also represent the source for Jack Hills detrital chromites, particularly given the proximal
917 source suggested for EO. Jack Hills mafic and ultramafic rocks are heavily sheared and
918 extensively recrystallised (Spaggiari, 2007a), however analysis of one sample (16WA13;
919 Supplementary Material) found relict chromite cores within texturally distinct chromite
920 stringers. Whether this texture is igneous or deformational in origin is unclear, but the
921 development of thick and indistinct magnetite and ferritchromit rims (Fig. 14c-d) yields
922 evidence for at least amphibolite facies metamorphism (*e.g.*, Barnes, 2000), likely representing
923 amphibolites facies metamorphism of the Jack Hills belt prior to deposition of Unit 3 and 4
924 metasedimentary rocks (Spaggiari *et al.*, 2007a/b). While 16WA13 relict cores possess
925 comparable Cr# to some Jack Hills detrital grains, gradational and lobate boundaries between
926 relict cores and ferritchromit-magnetite rims, and their compositional trend towards magnetite
927 (Fig. 14) are evidence for mobility of trivalent cations. This indicates that the Cr# of 16WA13
928 chromite cores represent maximum values, and that comparisons should be taken cautiously.
929 However, further analyses of Jack Hills mafic and ultramafic rocks, particularly those preserved
930 within lower strain areas, may prove more fruitful.

931

932 CONCLUSIONS

933 We have conducted a detailed textural and chemical investigation of detrital chromite from the
934 Jack Hills, Western Australia, in a bid to expand beyond the intensively studied zircon record
935 and provide additional perspectives on sources of sedimentary detritus (Fig. 15). Detrital
936 chromites have undergone modification during metamorphism of host metasedimentary rocks,
937 with significant mobility of divalent cations. This is shown by increasingly homogenised and
938 lowered Mg#, elevated ZnO and MnO, and partial to complete exchange of oxygen isotopes.
939 Interaction with metamorphic fluids also resulted in the generation of pitted domains and
940 replacement of primary mineral assemblages with metamorphic phases present in the matrix of
941 host metasedimentary rocks, dominantly quartz and fuchsite. We propose metamorphic
942 modification coincided with peak upper greenschist to lower amphibolite facies metamorphism
943 of Jack Hills metasedimentary rocks at ~2650 Ma.

944 Despite the development of fuchsite and high Cr# domains, Jack Hills chromites yield
945 a consistent range in Cr# across all samples and have limited intra-grain variability, indicating
946 the Cr# of chromites cores are robust. An absence of ferritchromit rims on euhedral grains
947 further argues against significant trivalent mobility prior to deposition. Critically, the moderate
948 and variable Cr# of most detrital chromites suggest that komatiites or associated intrusives were
949 not a significant component of mafic-ultramafic crust within the erosional catchment of Jack
950 Hills metasedimentary rocks. We propose a layered intrusive origin for Jack Hills detrital
951 chromites, though note that in the absence of further constraints from ferric iron and TiO₂
952 contents other sources cannot be excluded. It is currently unclear whether detrital chromites
953 represent the remnants of a hitherto unknown intrusion, or were sourced from the 3730 Ma
954 Manfred Complex or from ultramafic horizons within the Jack Hills belt. The lack of significant
955 komatiite contribution may have geodynamic implications for eruption efficiencies within the
956 catchment of Jack Hills sediments (e.g., heat-pipe vs. ‘plutonic squishy lid’; Moore & Webb,

957 2013; Rozel *et al.*, 2017), and more broadly, the tectonic distribution of mafic and ultramafic
958 crust within the Archean.

959 While this study focused on Jack Hills detrital chromites, we have demonstrated that
960 with a careful and detailed approach, characteristics of the igneous precursors of detrital
961 chromites may be determined. However, we show that identification of key physiochemical
962 signatures of chromite modification are critical to assess the veracity of interpretations derived
963 from detrital chromites preserved within Archean metasedimentary rocks. Future studies of
964 Archean detrital chromites should ideally be restricted to mature metasedimentary units that
965 have undergone at most lower amphibolite facies metamorphism to minimise trivalent cation
966 mobility. Where alteration of key physiochemical signatures can be shown to be absent, Cr#,
967 Fe₂O₃ and TiO₂ contents may be used to provide significant insights into the provenance of
968 mafic or ultramafic protoliths. Fuchsite metasedimentary rocks are commonly reported within
969 Archean terranes (e.g., Randive *et al.*, 2015), indicating that the eroded remnants of mafic and
970 ultramafic crust are abundant. Detrital chromites within fuchsite metasedimentary rocks may
971 therefore represent an innovative tool in understanding the distribution of eroded Archean mafic
972 and ultramafic crust, and yield important insights into geodynamic regimes of the early Earth.

973

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985

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1325

1326 Detrital chromite from Jack Hills, Western Australia: signatures of metamorphism and
1327 constraints on chromite provenance

1328

1329 Figure captions

1330

1331 Fig. 1: Map of the Yilgarn Craton, Western Australia, showing the position of the Narryer
1332 Terrane, Jack Hills and Mount Narryer. Modified from Kemp et al., 2018 with Terrane
1333 boundaries after Cassidy *et al.* (2006).

1334

1335 Fig. 2: (a) Simplified geological map of Jack Hills and surrounding quartzofeldspathic gneiss
1336 and granitoids with sampling locations, after Spaggiari *et al.* (2007b) and Pidgeon & Wilde
1337 (1998). (b) Higher resolution geological map of sampling locations at and around the W-74
1338 discovery site (see Fig. 2a). Modified after Spaggiari *et al.* (2007b). See Methods and Materials
1339 for relative positions of samples from the W-74 site.

1340

1341 Fig. 3: Jack Hills detrital chromite. (a) Optical microscope image illustrating *ex-situ* variably
1342 rounded morphologies of chromite grains. Note the green-brown fuchsite. (b) BSE image of
1343 chromite observed *in-situ* within quartzite cobble. Despite isolation from metaconglomerate
1344 matrix the cobble has been infiltrated by secondary Cr-muscovite. (c) BSE image of complex
1345 and ragged boundaries between chromite and fuchsite within metaconglomerate matrix. Grain
1346 heavily fractured on its southern margin. Chr – chromite, Qz – quartz, Fch – fuchsite. Iron
1347 oxides are likely magnetite, but have not been analysed by EPMA.

1348

1349 Fig. 4: BSE images showing chromite textural variation. All scale bars 50 μm . (a) Smooth with
1350 minimal fracturing, commonly associated with high Cr#. (b) Polygonal fracturing, no pitted
1351 domains. (c) Fractured with very fine pitted domains. Note coarser pores, with pitted domains
1352 localised to edges and near fractures. (d) Fractured with increasingly pitted textures, often away
1353 from obvious fractures. (e) and (f) porous chromite, fractures apparently lost or in-filled by
1354 secondary material. Secondary material often aligned to crystallographic axes. Qz – quartz, Fch
1355 – fuchsite and Rt- rutile. Iron oxides are likely magnetite, but have not been analysed by EPMA.

1356

1357 Fig. 5: BSE images of inclusion assemblages. (a) Isolated euh-subhedral quartz (Qz). Brighter
1358 crystallites in quartz are rutile, an Mg-silicate, and phosphate. Note the surrounding pitted
1359 domain, including darker quartz and rutile (Rt) laths, but absence of cracks. (b) Isolated
1360 polyphase assemblage of quartz and iron oxide. (c) Subhedral Fe-sulphide within pitted domain
1361 of chromite; pyrrhotite (Po) and chalcopyrite (Ccp) (outlined in red). (d) Euhedral rutile within
1362 quartz; note the ragged boundaries of quartz and surrounding pitted domain. (e) Anhedral
1363 inclusion of fuchsite. (f) Globular inclusion of quartz and fuchsite with monazite. Note the
1364 expansion cracks radiating from inclusion.

1365

1366 Fig. 6: Variation of Mg#, ZnO, MnO and TiO₂. Each data point is separated by sample and
1367 represents a core or near-core composition of chromite. Slightly elevated Mg# of 01JH
1368 chromites in comparison to those from 14WA and 16WA is a function of higher ferric iron
1369 contents. (a) ZnO vs Mg#. (b) MnO vs Mg#. (c) TiO₂ vs Mg#. Most grains yield low TiO₂ of
1370 <0.25 wt.%. It is likely that some analyses above this value overlapped with subsurface rutile
1371 laths. However, some grains with >0.25 wt.% TiO₂ show no evidence of rutile laths. TiO₂ >1
1372 wt.% clearly represent overlap with rutile laths, so are omitted. (d) MnO vs ZnO.

1373

1374 Fig. 7: Variation of Cr#. (a) ZnO vs Cr#. No variation in the range of Cr# despite apparent
1375 increasing ZnO. (b) Cr# vs $Fe^{2+}/(100x molar Fe^{2+}/(Fe^{2+}+Fe^{3+}))$. (c) TiO₂ concentrations vs
1376 Cr#. Scatter towards 1 wt.% a mixture of high TiO₂ grains and likely overlap of rutile laths.
1377 High Cr# grains appear to be slightly elevated in TiO₂ in comparison to bulk lower Cr# grains.
1378 (d) $Fe^{3+}/\Sigma Fe$ vs Cr#. Significant non-stoichiometry of 14WA and 16WA chromites has resulted
1379 in markedly lower $Fe^{3+}/\Sigma Fe$ in comparison to 01JH chromites.

1380

1381 Fig. 8: Examples of zonation patterns observed within 14WA and 16WA Jack Hills detrital
1382 chromites, including BSE images of line path taken. Isolated, globular inclusions of fuchsite
1383 (fch) also highlighted in 14WA2-PB-46. (a) Decreasing Mg# towards rims: most commonly
1384 shown by low ZnO samples 14WA2 and 16WA5. 14WA2-PB-46 yields the most variable
1385 zoning patterns observed in all Jack Hills grains and is likely the least altered grain analysed.
1386 This example also includes elevated Fe₂O₃ at its rims, resulting in an area of slightly elevated
1387 Cr#. (b) Elevated Mg# towards rims: commonly observed in high ZnO samples such as
1388 14WA3-4 and 16WA7. ZnO and Mn are lost from chromite at the expense of FeO and MgO,
1389 increasing the Mg# of rims relative to the core of the grain. (c) Largely homogeneous Mg#. No
1390 change in mineral chemistry across the grain in divalent or trivalent cations; this example has a
1391 slight elevation in Mg# on one edge after a minor decrease from the core.

1392

1393 Fig. 9: Variation in mineral chemistry: BSE images with elevated contrast to variable high Cr#
1394 domains. (a) and (b) High Cr#, pit-free domains at the edge of grains, and (c) along or associated
1395 with cracks. (d) and (e) High Cr# domains bound by chromite crystallographic axes, closely
1396 associated with laths of fuchsite (fch) and anhedral quartz (qz) along margins of high Cr# zones

1397 and rutile (rt). (f) The only observed example of a domain of lowered Cr#, encircling an
1398 inclusion of quartz with iron oxide rims. Note the radial fractures associated with the secondary
1399 inclusion. Except where stated, all scale bars are 50 μm .

1400

1401 Fig. 10: Oxygen isotope ratios ($\delta^{18}\text{O}$ reported at ‰ variation to V-SMOW) for different size
1402 fractions of Jack Hills detrital chromite grains measured by laser fluorination. $\delta^{18}\text{O}$
1403 uncertainties (2σ) are ± 0.08 for 01JH36, ± 0.12 for 01JJ42 and ± 0.06 for 01JH54. (a) and (b)
1404 Samples 01JH36 and 01JJ42, which show strong evidence of grain size dependency on $\delta^{18}\text{O}$
1405 values. (c) Sample 01JH54, showing no evidence of grain size dependency on $\delta^{18}\text{O}$ values.

1406

1407 Fig. 12: Cr^{3+} - Al^{3+} - Fe^{3+} triangular plot, modified from Barnes & Roeder (2001), showing three
1408 magmatic compositional pathways for spinel: Cr-Al, Fe-Ti, and Rum trends. The compositional
1409 space of ferrichromit, Cr-magnetite, and spinel miscibility gaps shown for reference. Core or
1410 near core compositions of low ZnO 14WA2 and 01JH54 chromites are shown.

1411

1412 Fig. 12: Provenance: Cr# vs $\text{Fe}^{2+}\#$ ($100x\text{Fe}^{2+}/(\text{Fe}^{2+}+\text{Mg}^{2+})$), diagrams modified from Barnes &
1413 Roeder (2001). (a) Strong fit of 14WA2 detrital chromite with chromite that have undergone
1414 high-grade metamorphism due to exchange of Mg^{2+} for Fe^{2+} , and thus increased $\text{Fe}^{2+}\#$. (b) Fit
1415 of bulk field for chromite from ophiolites, although Jack Hills chromites plot at considerably
1416 higher $\text{Fe}^{2+}\#$ due to metamorphic exchange of Mg^{2+} for Fe^{2+} . (c) Strong fit of chromite Cr# in
1417 Jack Hills chromite with chromite from continental layered mafic intrusions. Jack Hills detrital
1418 chromite $\text{Fe}^{2+}\#$ are again largely higher than 50th percentile due to metamorphic exchange. (d)

1419 Distinct absence of fit with chromite from Al-undepleted (AUD) komatiites. GSF; Greenschist
1420 facies AUD komatiites, AF; Amphibolite facies AUD komatiites.

1421

1422 Fig. 13: Provenance: TiO_2 vs. $\text{Fe}^{3+}/\Sigma\text{R}^{3+}$, diagrams modified from Barnes & Roeder (2001).
1423 Chromites that ostensibly yield no Fe^{3+} from 14WA2 are excluded for clarity. Fit of 14WA2
1424 and 01JH54 chromite to all fields for and (a) ophiolites, (b) layered intrusions and (c) Al-
1425 undepleted komatiites. GSF; Greenschist facies komatiites, AF; Amphibolite facies komatiites.

1426

1427 Fig. 14: Cr-spinel compositions from the Narryer Terrane. BSE images of Manfred Complex
1428 13TKN22 chromite; (a) finer chromite (chr) within 13TKN22, with a strong magnetite rim
1429 (mag) and an inclusion of olivine (ol), and (b) coarser chromite within hornblende (hbl).
1430 Serpentine (srp) and magnetite (mag) also shown. (c) and (d) Relict, porous chromite with thick
1431 ferrichromit and magnetite rims within 16WA13, an ultramafic from the SW limb of the Jack
1432 Hills belt. Note the softening of the boundary between core and rim in comparison to 13TKN22.
1433 (e)-(g) Compositional variability of 13TKN22 and 16WA13 chromite in comparison to Jack
1434 Hills detrital chromites. Spinel compositional fields from Manfred Complex harzburgites and
1435 websterites reported by Rowe & Kemp (2020) included for comparison. Layered intrusion
1436 fields from Barnes & Roeder (2001). (e) Cr^{3+} - Al^{3+} - Fe^{3+} triangular plot, (f) $\text{Cr}^{\#}$ vs $\text{Fe}^{2+}\#$ and (g)
1437 TiO_2 vs $\text{Fe}^{3+}/\Sigma\text{R}^{3+}$.

1438

1439 Fig. 15: The inferred petrogenesis of Jack Hills chromites. (a) Original igneous crystallisation
1440 in a single, spatially extensive igneous protolith, postulated to represent a layered intrusion. (b)
1441 Exhumation of layered intrusion and first sedimentary cycling event. Continued erosion

1442 resulted in loss of contemporaneous silicates; some chromites preserved isolated within
1443 quartzite. Chromites (\pm zircon) now observed within the matrix of Jack Hills metasedimentary
1444 rocks present previously in heavy mineral layers or disseminated throughout quartzite. (c)
1445 Further sedimentary reworking in second cycling event deposits chromites in Jack Hills
1446 conglomerates and quartzites. Dispersion of chromites and zircons within both quartzite
1447 cobbles and matrix. (d) Metamorphism of Jack Hills metasedimentary rocks at \sim 2650 Ma, likely
1448 representing peak metamorphism at upper greenschist to lower amphibolite facies. This event
1449 also formed fuchsite and other metamorphic assemblages within the host metasedimentary
1450 rock, replaced chromite primary inclusion assemblages, and led to the partial to complete
1451 exchange of chromite divalent cations with metamorphic fluids. (e) Proterozoic metamorphic
1452 overprint of metasediments within the Jack Hills. Minor metamorphic exchange potentially
1453 indicated by zonation profiles.

Fig.1

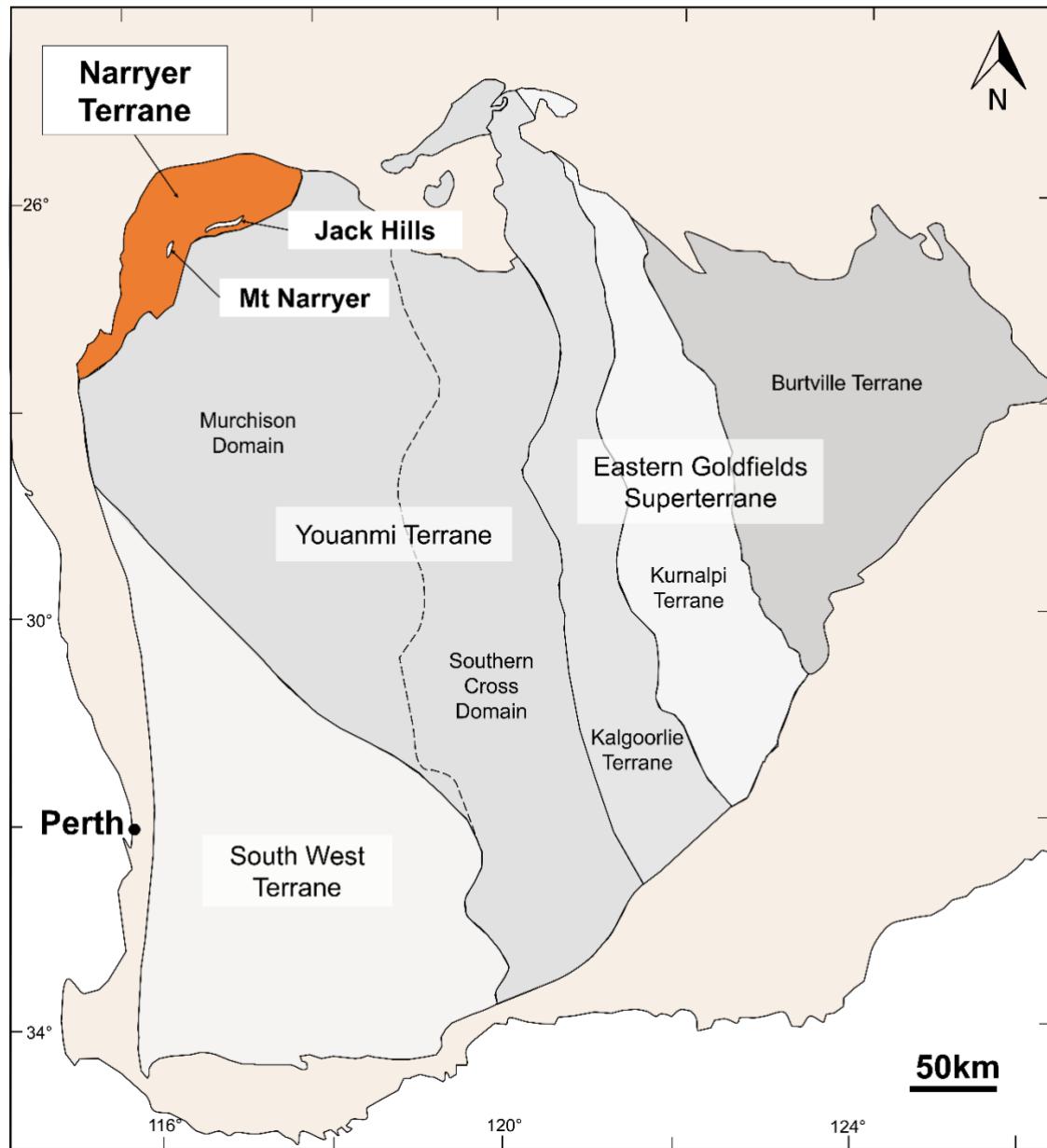
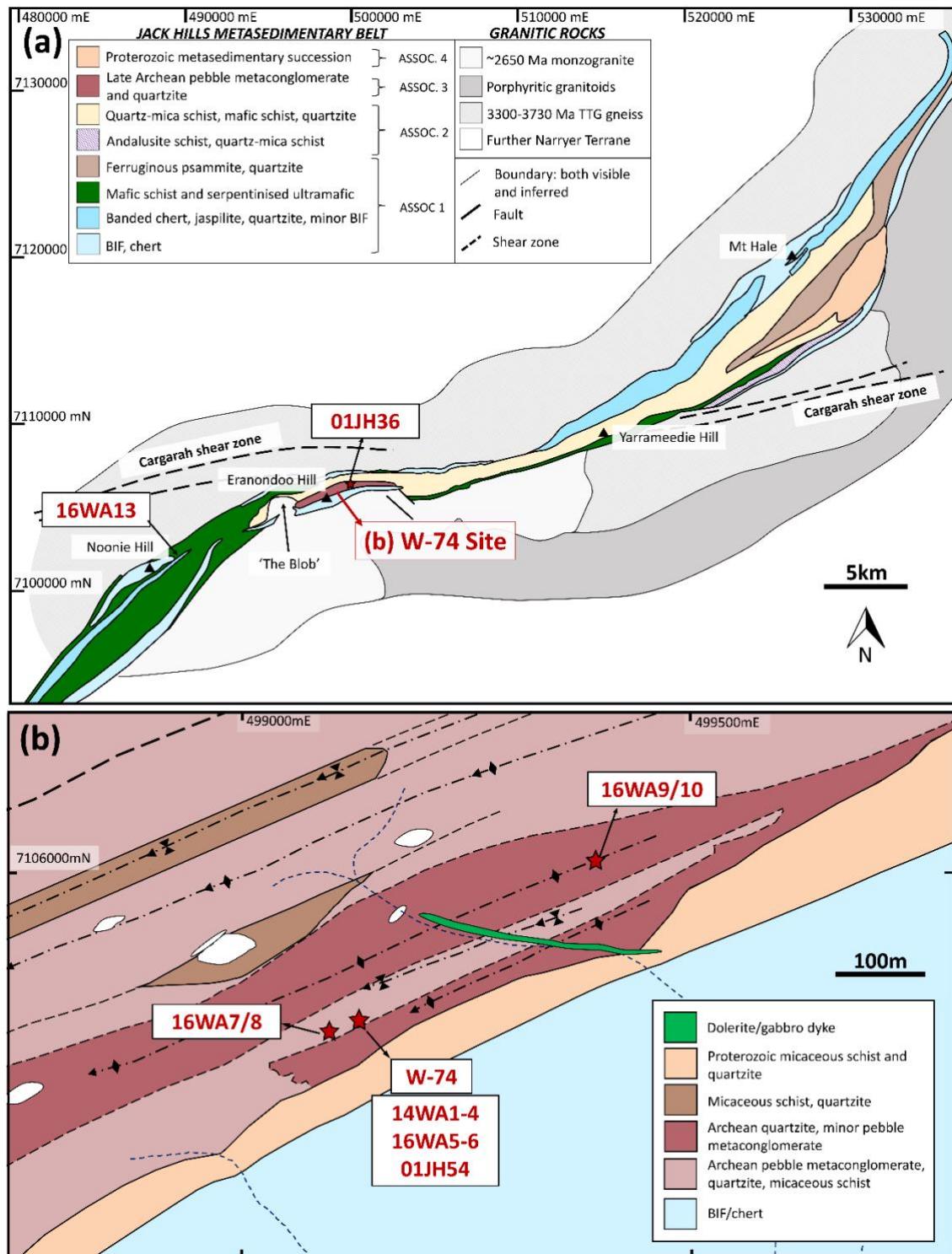


Fig.2



To add: 01JH35 sampling location

Fig. 3

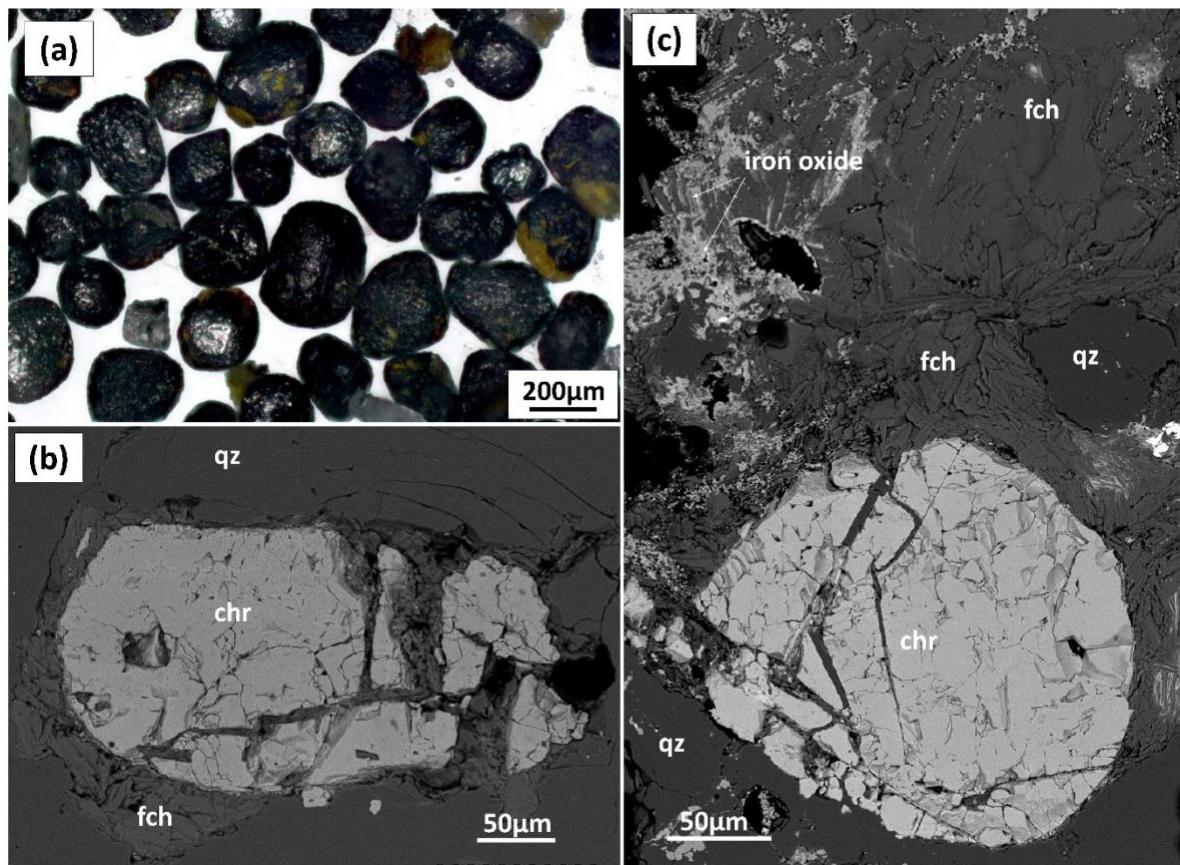


Fig. 4

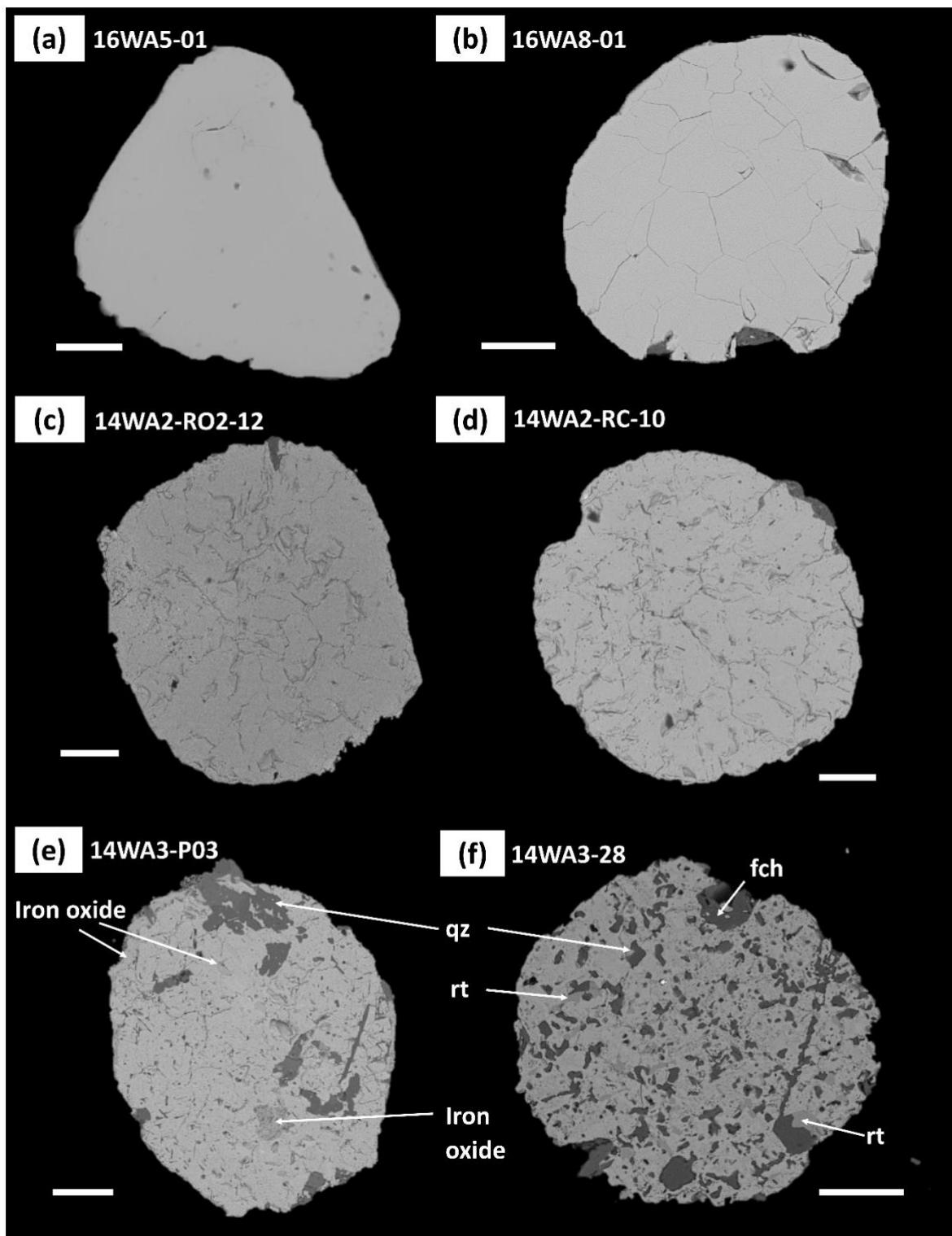


Fig. 5

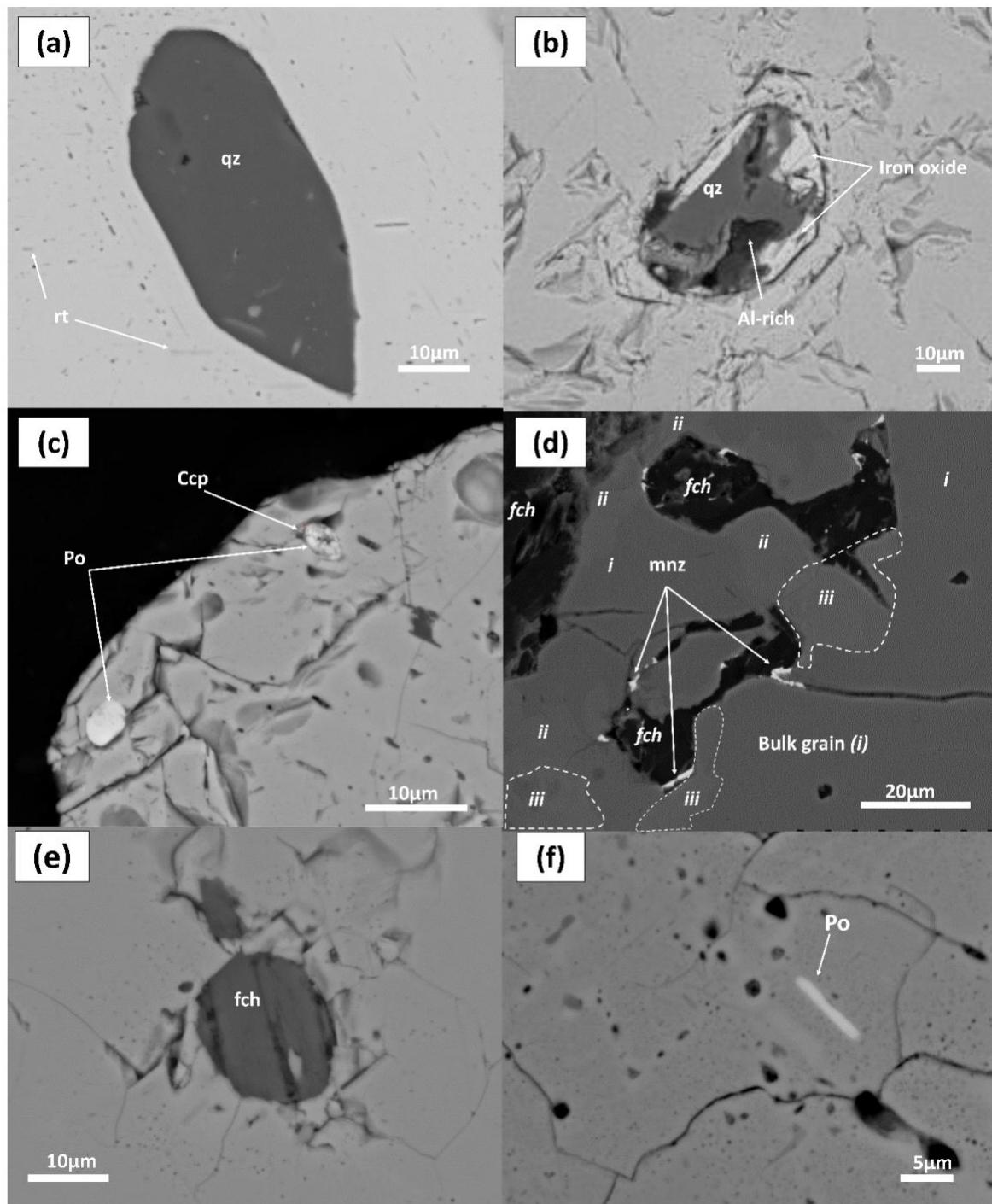


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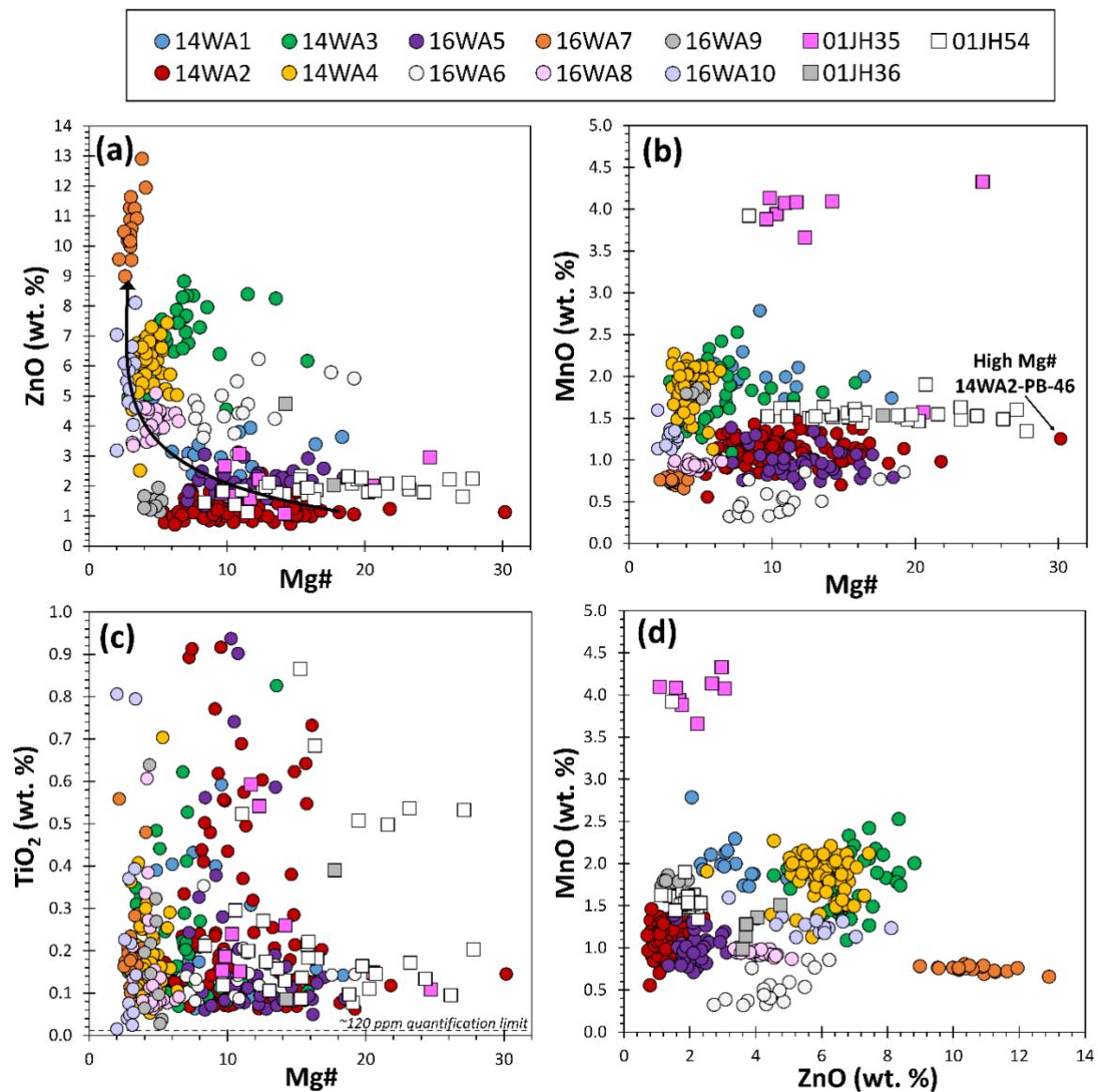


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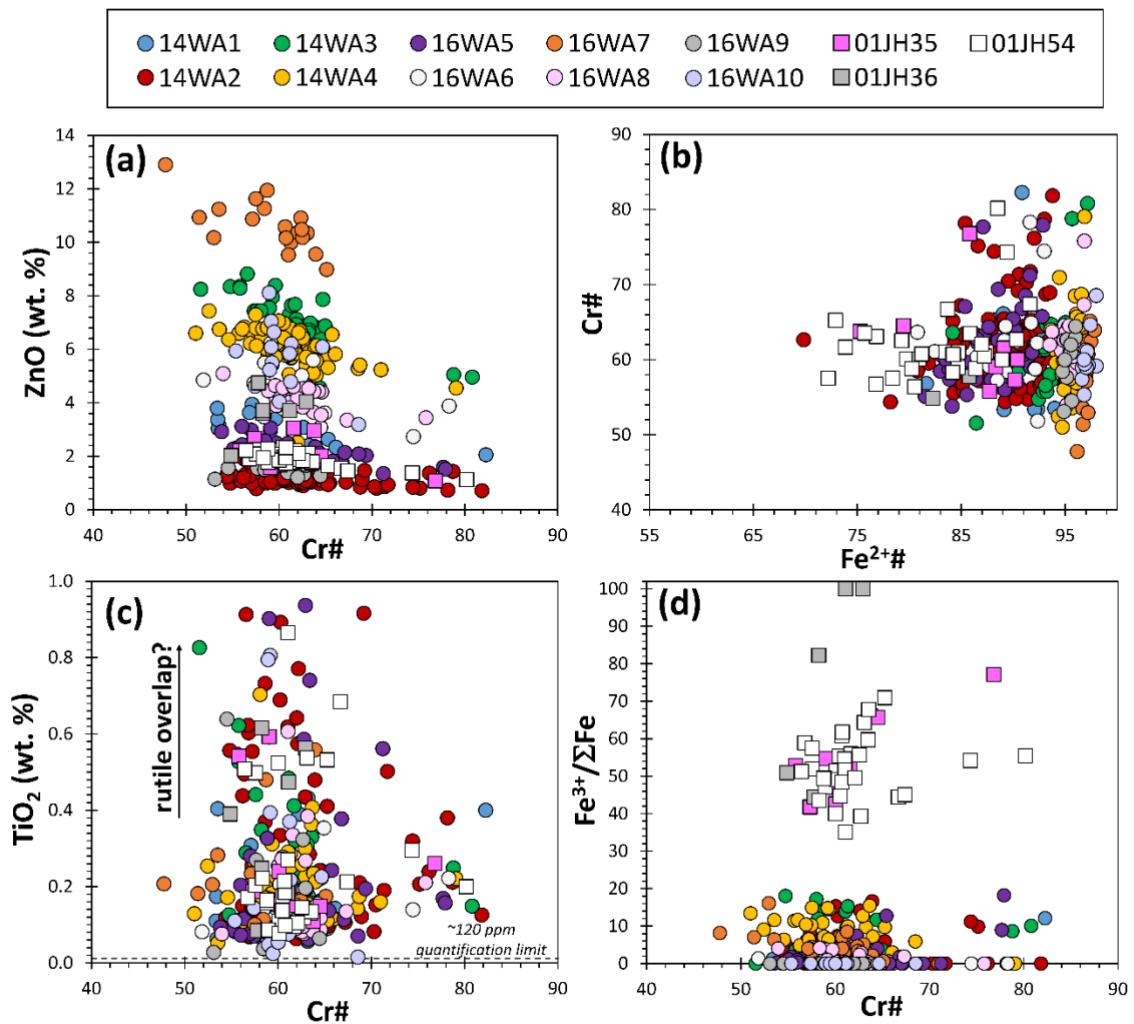


Fig. 8

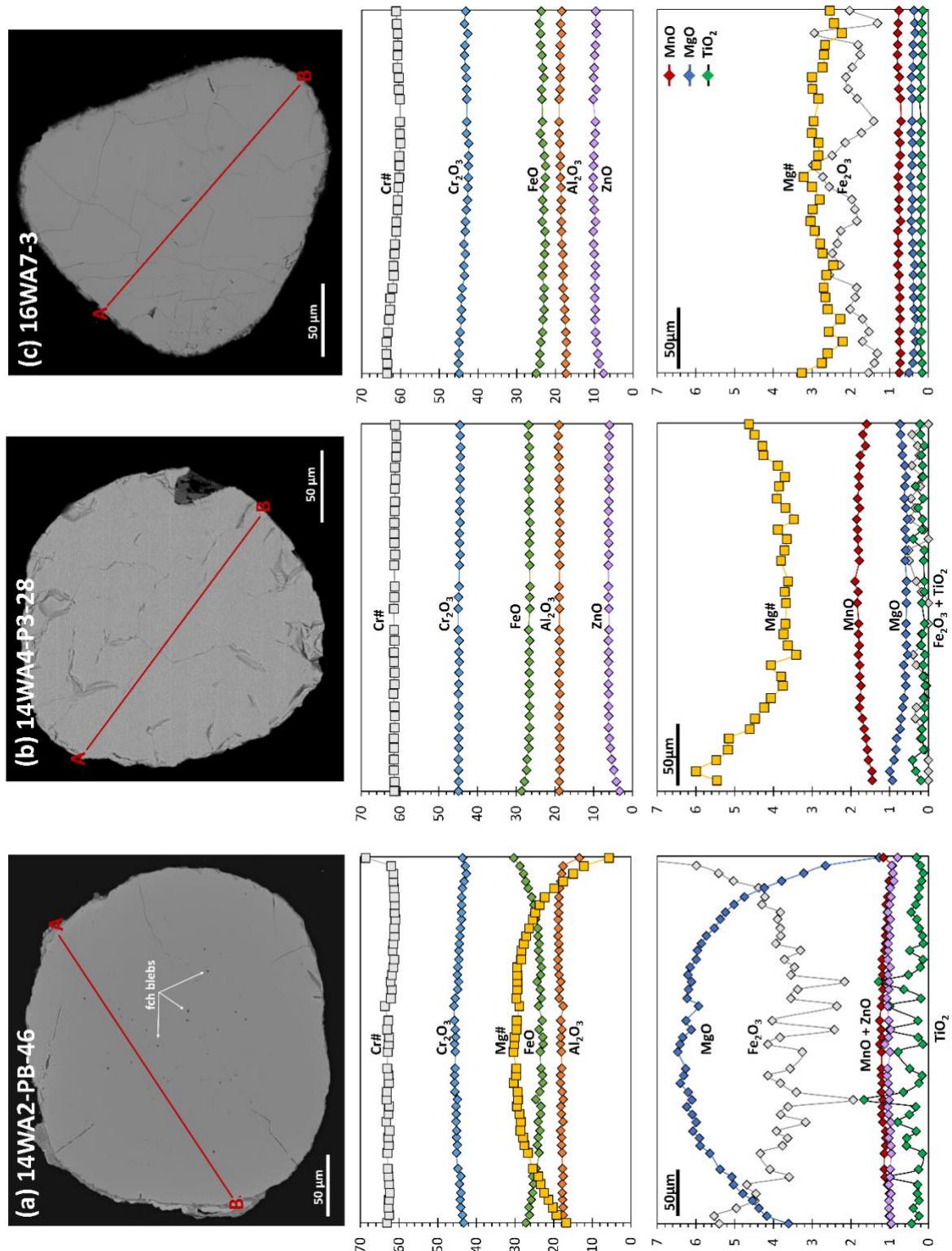


Fig. 9

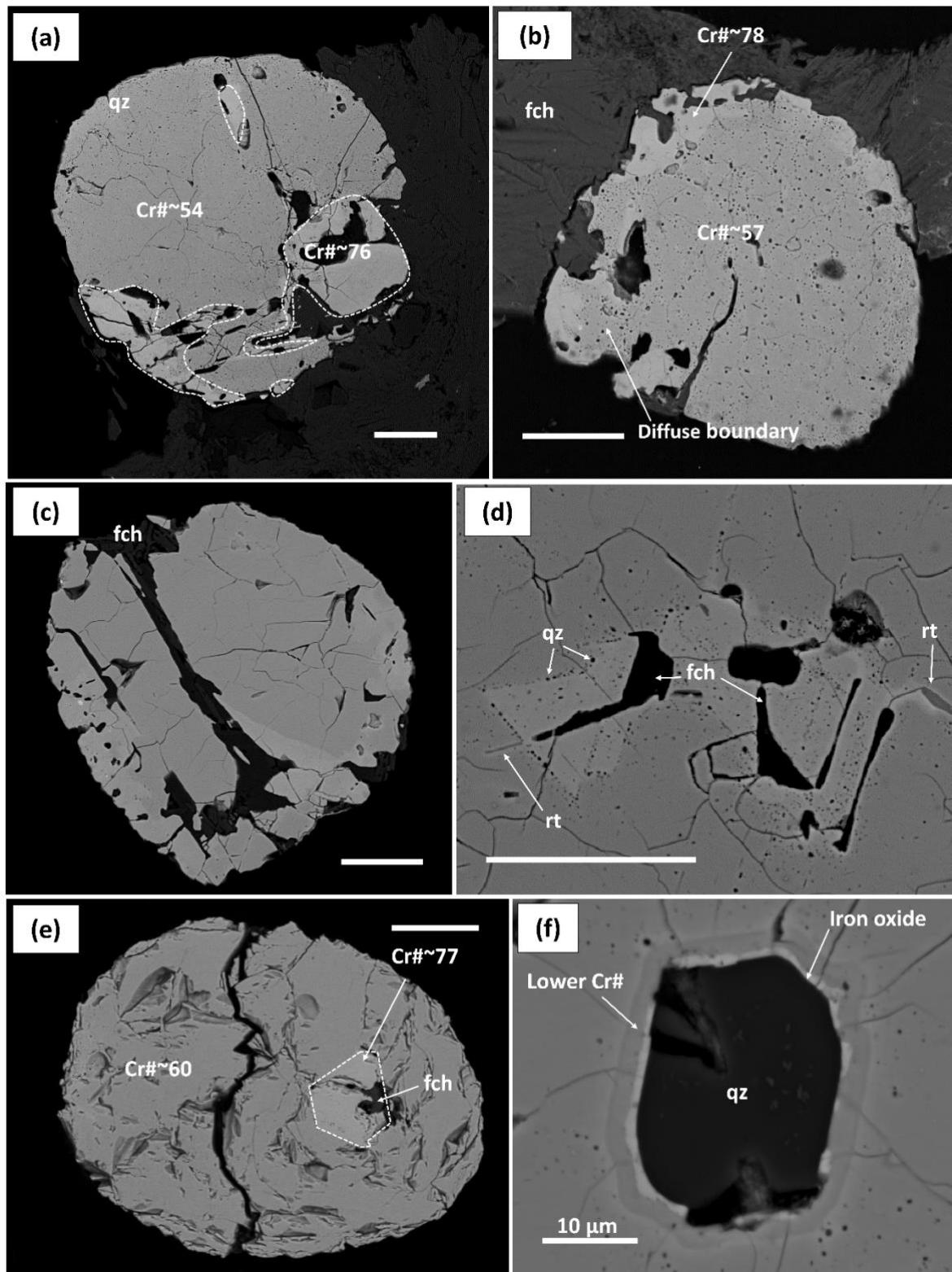


Fig. 10

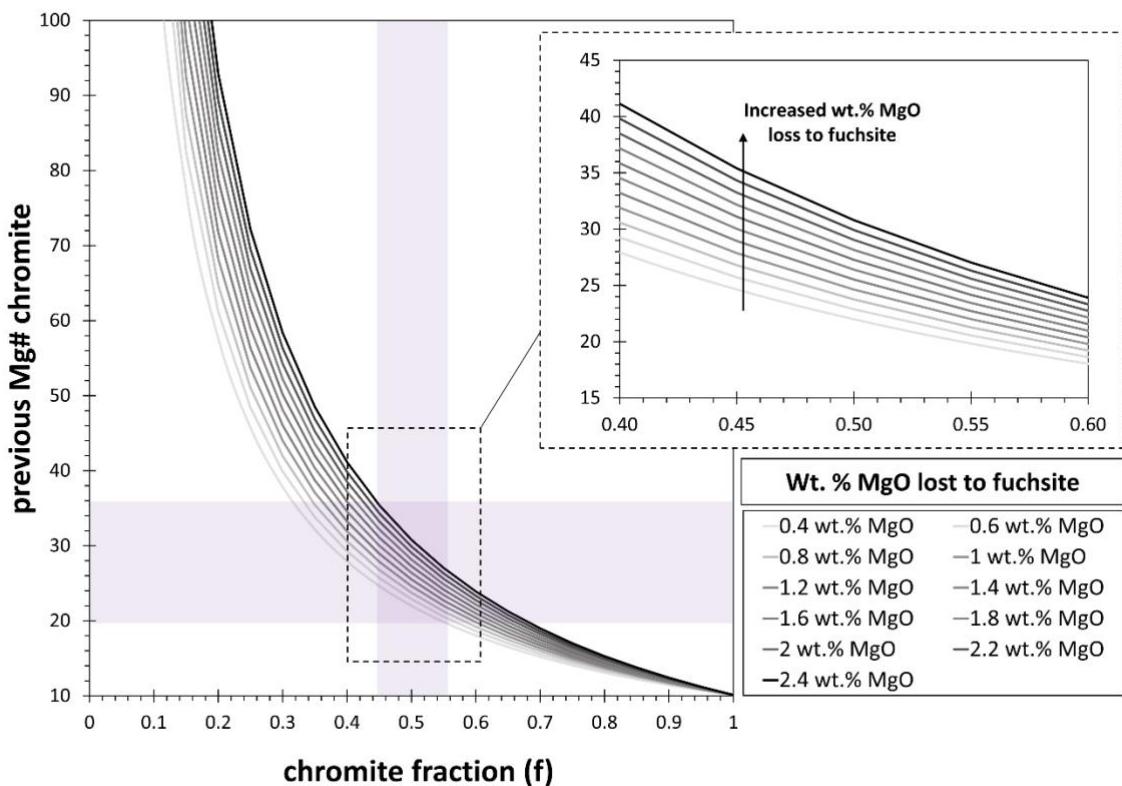


Fig. 11

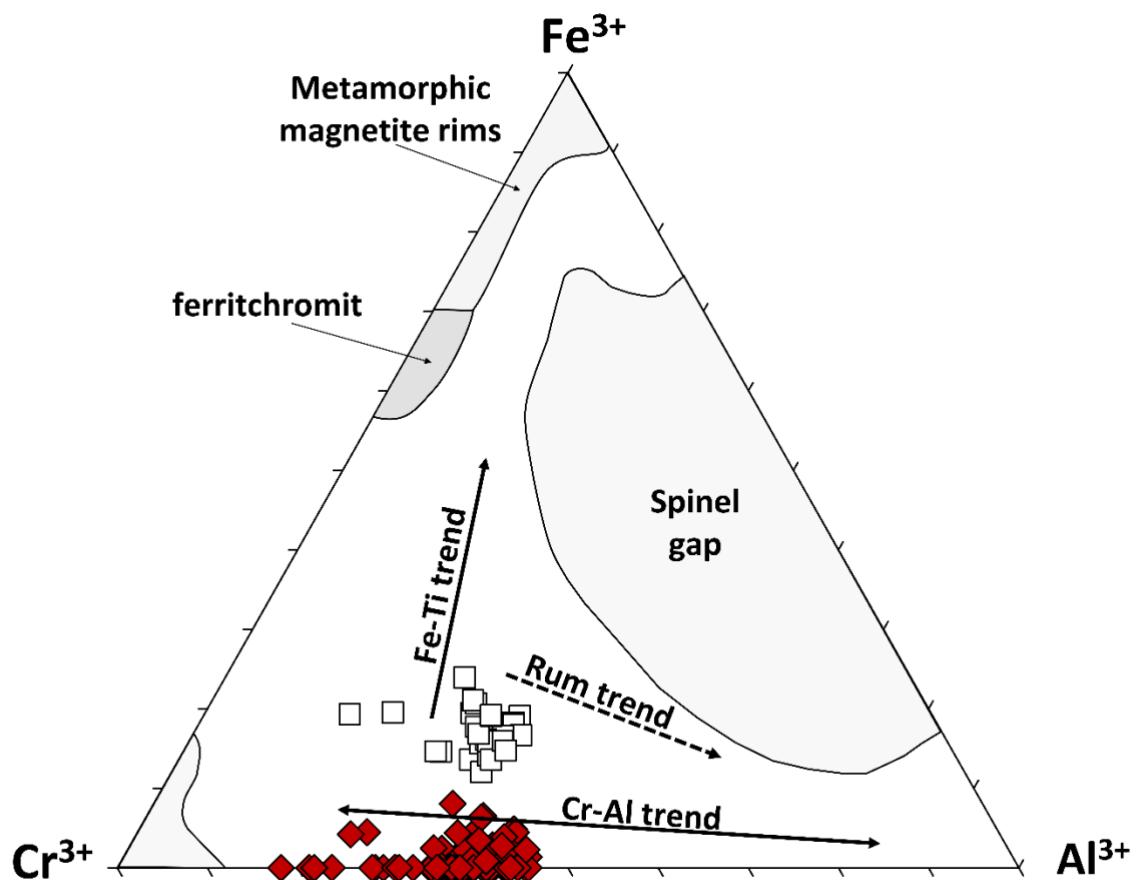


Fig. 12

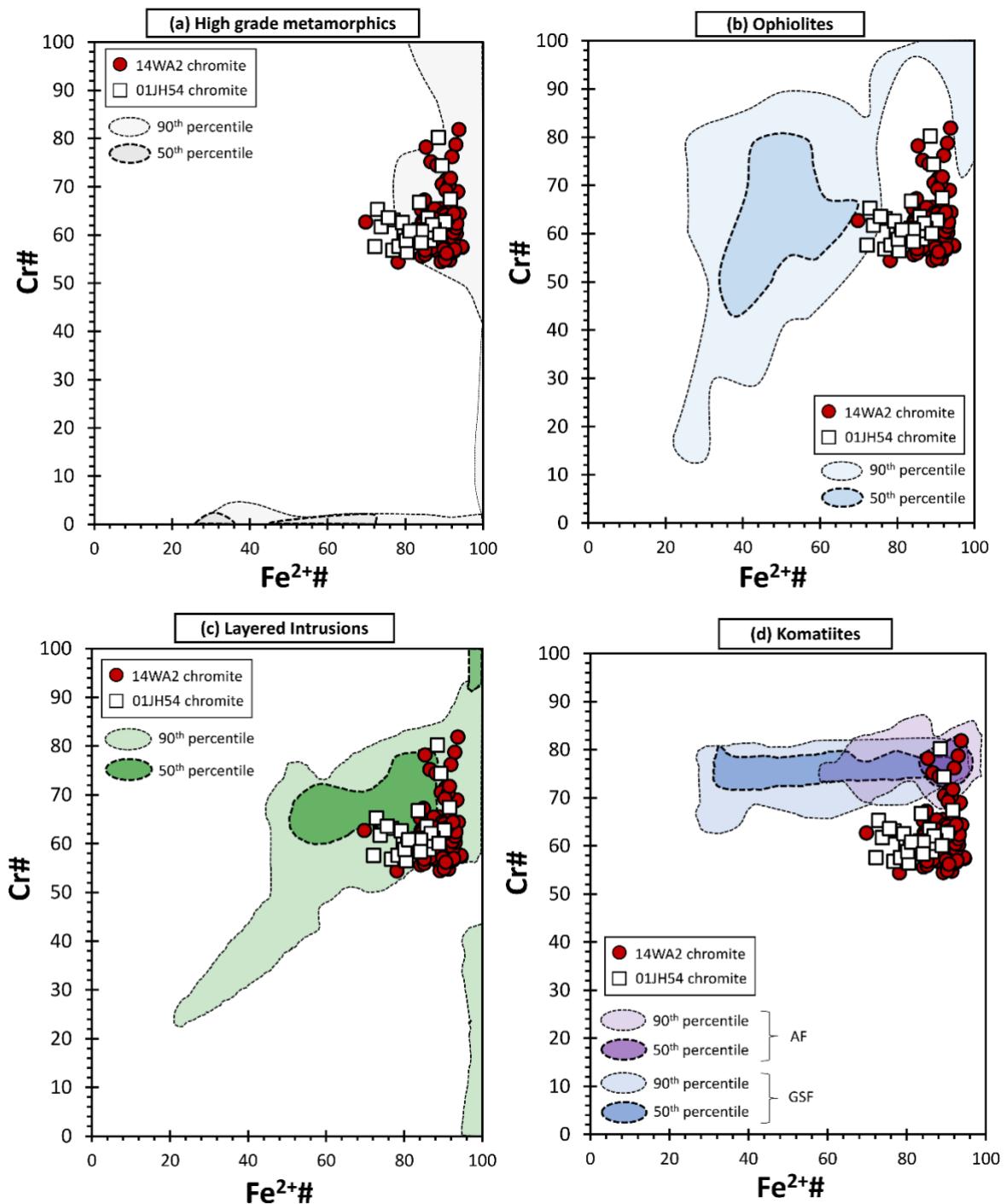


Fig. 13

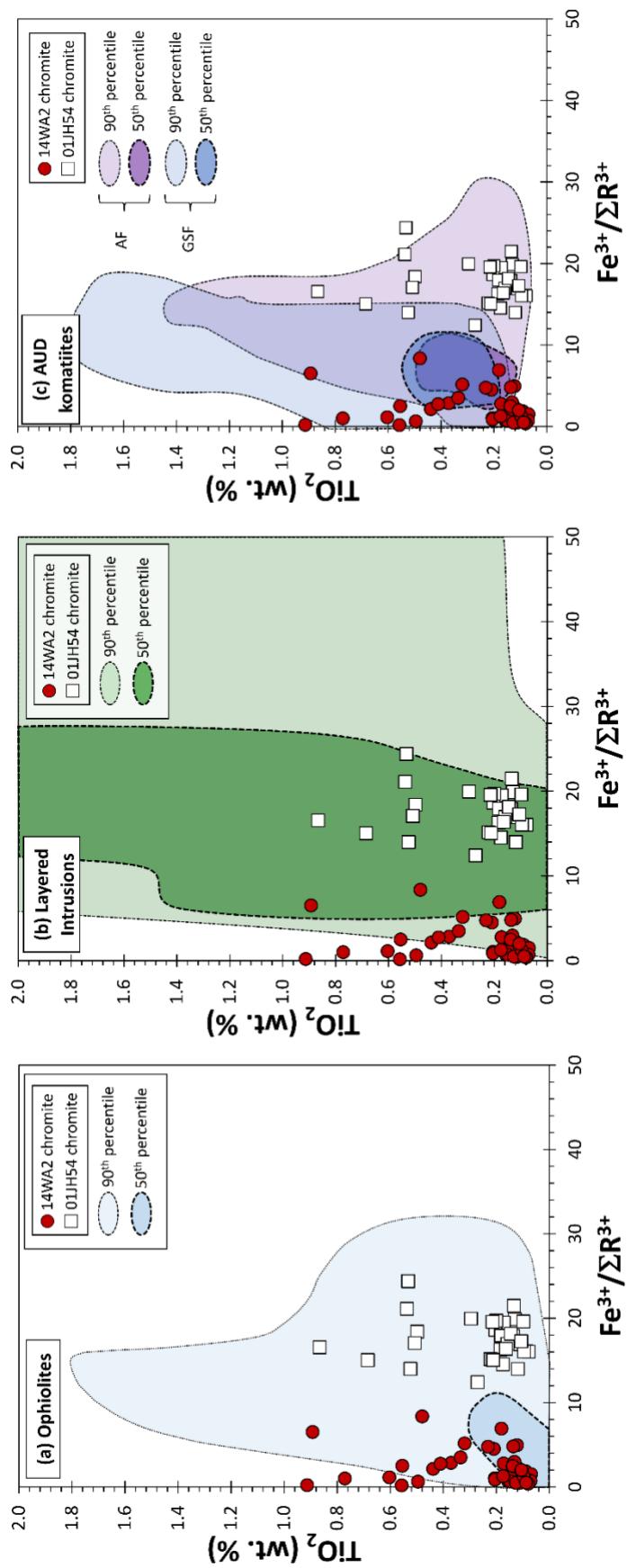


Fig. 14

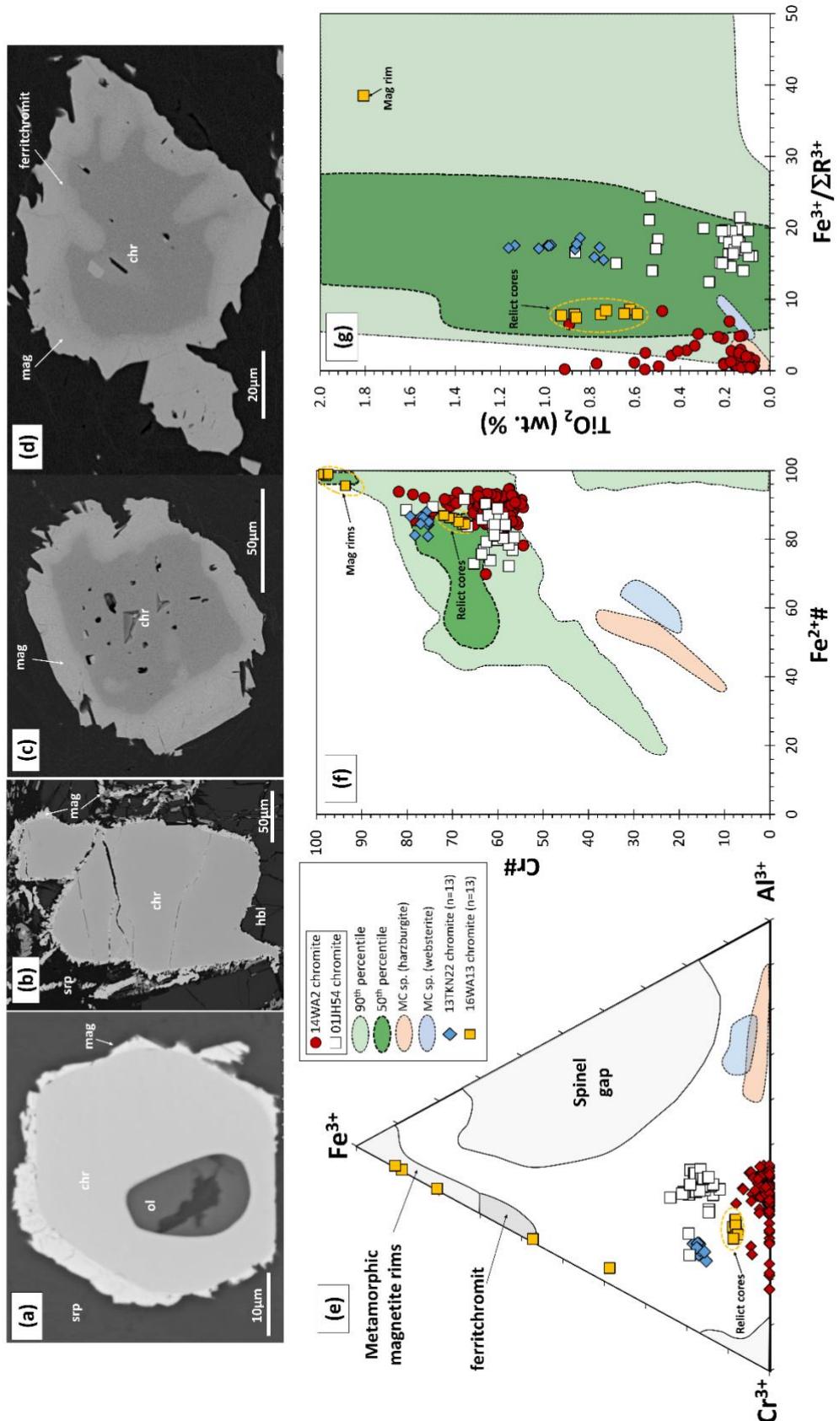


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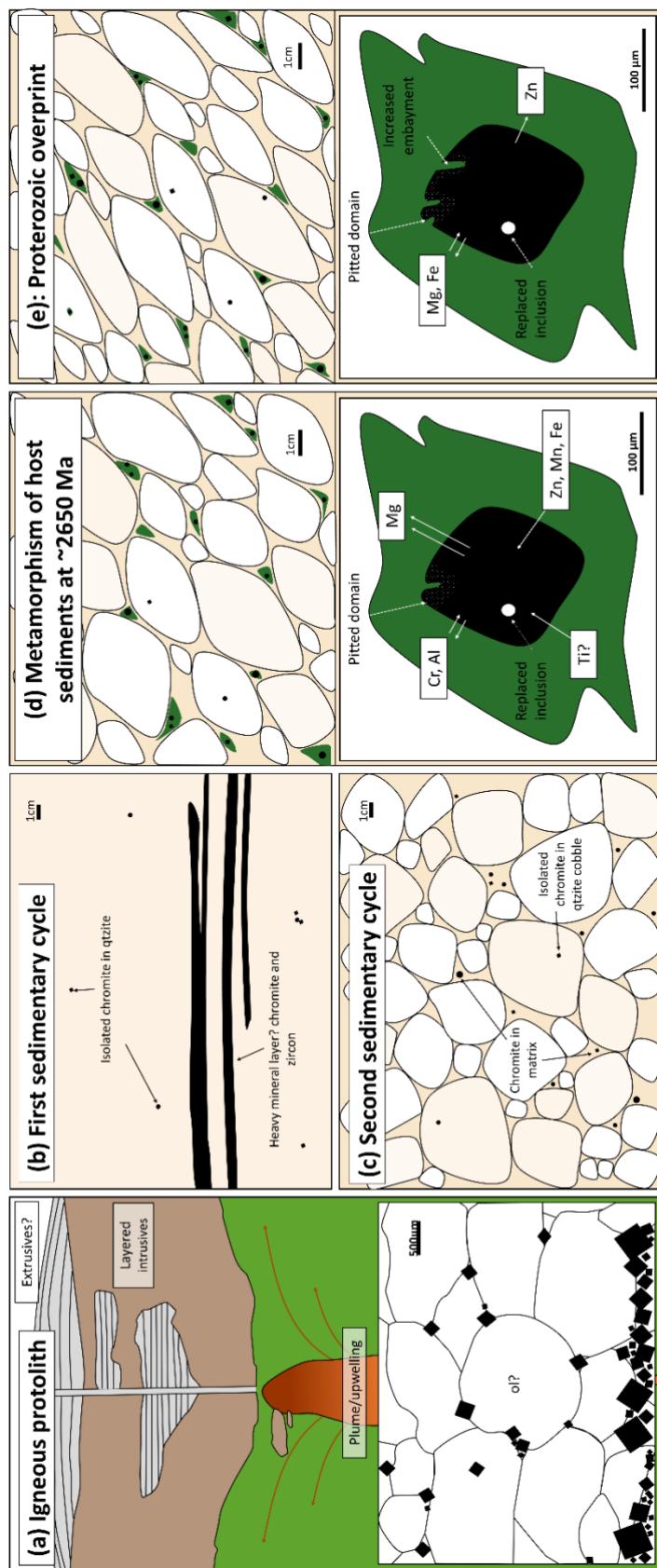


Table 1: Representative EPMA analysis of Jack Hills chromites.

Sample	14WA1	14WA2	14WA3	14WA4	16WA5	16WA6	16WA7	16WA8	16WA9	16WA10	01JH35	01JH36	01JH54
Grid Ref. (UTM)	499135E 7105846N	499135E 7105846N	499135E 7105846N	499135E 7105846N	499135E 7105846N	499141E 7105855N	499102E 7105837N	499096E 7105837N	499395E 7106007N	499393E 7106004N		499947E 7106431N	499137E 7105849N
Grain #	4	17	44	28	19	12	7	5 core	2	20	4	5	5
Rounding	RO	RO (2)	RO	RO/RC	RO	EO	RO/EO	RO/RC	EO	RC			
SiO₂	0.02	0.08	0.00	0.00	0.01	0.01	0.00	0.01	0.03	0.03	0.03	0.01	0.03
TiO₂	0.18	0.32	0.27	0.09	0.12	0.19	0.28	0.08	0.22	0.04	0.54	0.08	0.11
Al₂O₃	18.52	11.89	18.13	18.53	20.28	18.26	22.43	19.02	19.48	20.35	21.16	21.14	19.86
Cr₂O₃	46.20	51.52	42.43	44.42	44.18	45.82	38.52	44.69	45.26	44.11	39.79	43.14	44.59
V₂O₃	0.12	0.19	0.16	0.11	n/a	n/a	n/a	n/a	0.14	0.13	0.23	0.12	0.14
Fe₂O₃	0.00*	3.97*	4.37*	0.82*	0.00*	0.00*	1.81*	0.05*	0.00*	0.00*	16.82 [¥]	12.26 [¥]	15.89 [¥]
FeO	27.66	28.26	24.39	26.17	28.76	27.50	23.79	28.63	29.19	26.65	13.52	14.15	13.49
MgO	1.20	2.40	1.39	0.62	2.18	1.67	0.49	0.91	0.76	0.42	1.06	1.32	1.92
MnO	2.13	1.46	2.03	1.84	0.71	0.33	0.73	0.95	1.80	1.28	3.66	1.51	1.46
CaO	0.00	0.01	0.00	0.01	0.00	0.01	0.01	0.00	0.00	0.00	n/a	n/a	n/a
Na₂O	0.00	0.10	0.00	0.00	n/a	n/a	n/a	n/a	n/a	n/a	n/a	n/a	n/a
NiO	0.00	0.00	0.00	0.01	0.02	0.00	0.00	0.01	0.00	0.00	0.01	0.00	0.00
ZnO	3.08	0.84	7.29	6.19	2.11	4.32	11.24	3.95	1.39	5.50	2.22	4.74	1.80
CuO	n/a	0.01	0.02	0.00									
Total	99.10	101.03	100.46	98.80	98.38	98.11	99.31	98.31	98.27	98.52	99.05	98.87	99.29
Mg#	7.20	11.83	8.03	3.95	11.90	9.77	3.30	5.36	4.44	2.76	12.19	14.24	20.24
Cr#	62.60	74.41	61.09	61.66	59.37	62.74	53.53	61.19	60.92	59.25	60.04	57.78	60.10
Fe³⁺/ΣFe	0.00	11.22	13.87	2.74	0.00	0.00	6.42	0.00	0.00	0.00	52.81	44.51	51.46
Fe³⁺/ΣR³⁺	0.00	5.17	5.65	1.07	0.00	0.00	2.34	0.00	0.00	0.00	18.32	13.86	16.93

Table 1: Representative EPMA analysis of chromite from each sample location. V₂O₃ not measured for 16WA5-7 and partial 16WA8. *Fe₂O₃ calculated from AB₂O₃ stoichiometry using the equations of Droop (1987). [¥]Fe₂O₃ calculated by charge balance from direct EPMA measurement of oxygen.

Table 2: Oxygen isotopic composition ($\delta^{18}\text{O}_{\text{VSMOW}}$) of Jack Hills detrital chromites.

Size fraction (μm)	105-149	149-210	210-300	300-350	350-425	Uncertainty (2sd)
01JH36	0.65		1.15	1.16		0.08
01JH36	0.73		1.41	1.71		
01JH36			1.41	1.77		
01JH36 Average	0.69		1.32	1.55		
01JH54		1.62	2.17	1.02	1.65	0.06
01JH54		1.69	2.2	1.17	2.11	
01JH54		2.07				
01JH54 Average		1.79	2.19	1.10	1.88	

Table 2: Oxygen isotopic composition ($\delta^{18}\text{O}_{\text{SMOW}}$) of detrital chromites from 01JH36 and 01JH54 determined by laser fluorination. Values represent deviation from standard mean ocean water (V-SMOW) in per mil (‰). Each analysis consisted of a ~2 mg aliquot of chromite separated by grain size.

Any separation by rounding shape?

Table 3: Representative EPMA compositions of 13TKN22 chromites and 16WA13 relict chromite cores.

Sample Grid Ref. (UTM) Crystal # Classification	13TKN22 40 chromite	13TKN22 42 chromite	13TKN22 47 chromite	16WA13 489515E 7102336E	16WA13 489515E 7102336E	16WA13 489515E 7102336E
SiO₂	0.00	0.00	0.02	0.03	0.02	0.03
TiO₂	1.13	1.03	0.97	0.62	0.92	0.86
Al₂O₃	9.23	9.26	9.18	13.31	12.80	12.52
Cr₂O₃	43.66	44.18	43.87	45.96	46.48	46.77
V₂O₃	n/a	n/a	0.25	n/a	n/a	n/a
Fe₂O₃	12.88	12.63	12.95	6.47	5.76	5.87
FeO	28.10	27.72	27.20	27.89	28.25	28.17
MgO	2.72	3.41	3.72	2.83	2.68	2.65
MnO	1.32	0.45	0.43	1.21	1.20	1.19
CaO	0.01	0.03	0.03	0.01	0.00	0.03
NiO	0.14	0.12	0.13	0.08	0.12	0.10
ZnO	0.60	0.59	0.58	1.08	1.06	1.08
Total	99.80	99.43	99.33	99.10	99.61	99.30
Mg #	14.73	17.99	14.59	13.00	12.49	12.37
Cr #	76.03	76.20	76.22	69.84	70.89	71.47
Fe³⁺/ΣFe	29.20	29.08	30.00	17.28	15.51	15.78
Fe³⁺/ΣR³⁺	17.59	17.18	17.64	8.56	7.72	7.86

Table 3: Representative EPMA analysis of chromite from Manfred Complex metaperidotite 13TKN22 and Jack Hills recrystallised ultramafic 16WA13. V₂O₃ not measured for some chromites within 13TKN22. Fe₂O₃ calculated from AB₂O₃ stoichiometry using the equations of Droop (1987).