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Influence of intrinsic defects on the structure and dynamics of the mixed Pb–Sn perovskite: first-principles DFT and NAMD simulations†

Qi Liu,^{ab} Akang Li,^{ab} Weibin Chu,^{cd} Oleg V. Prezhdo^{cd} and WanZhen Liang^{ab}

The mixed tin (Sn) and lead (Pb) perovskite compositions have shown great potential in perovskite photovoltaic devices due to the significantly enhanced material stability and prolonged carrier lifetime, compared to the pure Sn halide perovskites. In spite of the increasing interest, the behaviors of photo-generated charges and of the intrinsic point defects, such as the metal cation vacancies (V_{Sn} and V_{Pb}) and the interstitial halogen (I_i), have not been well understood in this class of materials. We report first-principles density functional theory (DFT) calculations combined with *ab initio* non-adiabatic molecular dynamics (NAMD) simulations on the static and dynamic structures of $\text{MA}_2\text{SnPbI}_6$ with and without these intrinsic defects. We discuss the nature of the defect states and unveil the influence of the intrinsic point defects on the structure, optoelectronic properties, and charge carrier dynamics of $\text{MA}_2\text{SnPbI}_6$. The I_i defect significantly shortens the carrier lifetime by creating mid-gap states that provide new recombination pathways. In comparison, the vacancy defects have much weaker influence on the carrier lifetime. Both V_{Sn} and V_{Pb} produce the defect states just below the valence band maxima (VBMs), and do not alter the band gap. They affect the carrier lifetime through changing the energy dispersions of VBMs and the conduction band minima (CBMs). We suggest that excess cations should be used in the synthesis of perovskites to avoid the appearance of interstitial halogen defects.

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1 Introduction

Perovskite solar cells (PSCs) have been intensively studied in the past decade because of their high power-conversion-efficiency (PCE), low cost and easy synthesis.^{1–7} The first PSC was designed by Miyasaka *et al.* with a PCE of only 3.8%.³ Subject to the rapid advances over the past ten years, the certified PCE increased quickly to 29.5%.⁸ These developments have demonstrated the great potential of PSCs as the next-generation low-cost solar technology. Unfortunately, the most efficient PSCs all contain Pb. The Pb-induced pollution and poisoning are serious environmental concerns, which have hampered the commercialization of the technologies based on these

materials. Developing alternative strategies to minimize the impact of toxic Pb on the environment under the premise of maintaining the photoelectric activity of perovskite materials has become an important topic in perovskite photovoltaics.^{2,7,9–11}

To reduce or eliminate the content of Pb, scientists have suggested to develop PSCs containing less Pb and Pb-free PSCs.^{12–16} One idea is to develop Pb-free PSCs by replacing Pb with other elements. The earliest suggestion was to replace Pb with Sn and Ge, group 14 elements with comparable ionic radii.^{17,18} Later, transition metals such as Cu, Mn, Fe, Co, and Ni were identified as promising alternatives to Pb, owing to their rich chemistries and multiple oxidation states. However, these Pb-free compounds suffer from poor efficiency and stability.

Another idea is to replace two divalent Pb(II) ions with monovalent and trivalent ions. Experimentalists have successfully synthesised double perovskites by using heterovalent substitution with monovalent Cu, Ag, Au or even alkali-metals, and trivalent In, Sb, and Bi cations.^{19–32} However, these kinds of Pb-free perovskite materials possess serious flaws in optoelectronic properties, which make them less likely candidates for PSCs. For example, the most easily synthesized Ag–Bi and Ag–Sb double perovskites^{19,20,23,33} have an indirect band gap, which will lead to a large amount of solar energy dissipated into heat after the carrier excitation.³⁴

^aState Key Laboratory of Physical Chemistry of Solid Surfaces, Collaborative Innovation Center of Chemistry for Energy Materials, Fujian Provincial Key Laboratory of Theoretical and Computational Chemistry, People's Republic of China. E-mail: liangwz@xmu.edu.cn

^bDepartment of Chemistry, College of Chemistry and Chemical Engineering, Xiamen University, Xiamen, Fujian 361005, People's Republic of China

^cDepartment of Chemistry, University of Southern California, Los Angeles, California 90089, USA. E-mail: prezhdo@usc.edu

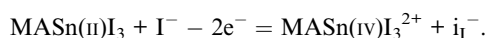
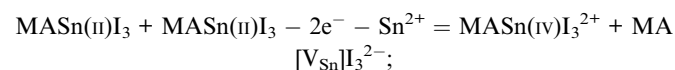
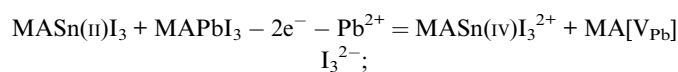
^dDepartment of Physics and Astronomy, University of Southern California, Los Angeles, California 90089, USA

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The PCE of Ag–Bi double perovskite-based PSCs is much lower than that of Pb-based, and even Sn-based PSCs. It is about 1.25% in its maximum.³³ In addition, because AgI is highly sensitive to light, this kind of material cannot be filled with iodide cations, resulting in a wide band gap. The bulk phase of the In(I)–Bi(III) based perovskite has been predicted theoretically.³⁵ It has a direct band gap and a carrier effective mass similar to that of the traditional Pb-bearing materials. However, the stability of monovalent In⁺ is lower than that of Ag⁺ and Tl⁺, and it has been reported experimentally that Cs₂In(I)In(III)X₆ perovskites are stable only at high temperature, while low temperature can cause a rotation between In(I) octahedra and In(III) octahedra, which leads to further degradation.³⁶

An increasing number of studies have shown that it is difficult to eliminate Pb from high-performance PSCs. The more stable and better performing photoelectric materials are those still containing Pb, but in reduced amounts.^{18,37–46} In such compounds, Pb is partially replaced by other elements, including Sn, Ge, Sr, or other alkaline-earth-metals. At present, the practical scheme is to adopt a mixed Sn–Pb composition, and to decrease the Pb content as much as possible, in order to reduce the environmental toxicity of Pb while maintaining the working stability of the device. For instance, Ramirez *et al.* adopted a 1 : 1.5 stoichiometric Sn : Pb ratio and reported a relatively high PCE of 10.5%.⁴⁷ Kapil *et al.* produced a spike structure of the Pb–Sn mixed perovskite and reached a 17.6% PCE.⁴⁸ Lanzetta *et al.* explained the surface-oxidation mechanism of Pb–Sn based perovskites.⁴⁹ Long *et al.*⁵⁰ discovered that the presence of Sn enhances the carrier lifetime of the traditional Pb-bearing perovskites.

When Sn(II) is exposed to air, it oxidizes spontaneously to form Sn(IV). Such a process leads to formation of various intrinsic defects,^{42,51,52} which have a severe impact on the optoelectronic properties of the material.^{53–63} The most common intrinsic defects generated in Pb–Sn perovskites include the Pb and Sn vacancies (V_{Pb} and V_{Sn}), and the interstitial iodide (i_I), which can be formed by the following chemical processes.



In this work, we study the structure, and the static and dynamic properties of the mixed Pb–Sn perovskite, and elucidate how the structure, and the optoelectronic and photoelectric properties are influenced by the intrinsic defects that are formed by the oxidation of Sn(II). We perform both first-principles density functional theory (DFT) calculations and non-adiabatic molecular dynamics (NAMD) simulations to obtain the fundamental insights, guiding the construction of high performance PSCs. We generate a microscopic

understanding of the characteristics of the materials with and without the intrinsic defects, and study the nature of the defect states, which usually play a decisive role in determining carrier recombination and diffusion lengths in solar cells, and explain why the photophysical properties of mixed Sn–Pb perovskites are enhanced considerably compared to those of Sn-based crystals.

2 Computational details

The first-principles DFT simulations were performed with the Vienna *Ab Initio* Simulation Package (VASP)^{64–66} to study the geometric and electronic structures of the pristine and defective MA₂SnPbI₆ perovskite series. Unless otherwise specified, the generalized gradient approximation exchange-correlation functional of Perdew–Burke–Ernzerhof (PBE)⁶⁶ was adopted in the calculations. Since PBE tends to underestimate bandgaps, we employed the HSE06 functional^{67,68} for benchmark calculations. HSE06 includes a fraction of screened Hartree–Fock exchange, $\alpha = 0.25$, improving the discontinuity in the Kohn–Sham potential derivative for integer numbers of electrons.

The spin-orbit coupling (SOC) effect is taken into account in the electronic structure calculations.

The electronic constituents are 4d 5s 5p for Sn, 5d 6s 6p for Pb, 5p 6s for I, 2s 2p for C and N, and 1s for H. For all the bulk-like systems, we adopted the Γ -point only for the HSE06 + SOC calculations, $4 \times 4 \times 4$ Γ -centered k -point grid generated by the Monkhorst–Pack scheme for detailed properties obtained with PBE + SOC, and $12 \times 12 \times 12$ linear scheme for the band-structure calculation with PBE + SOC. The projector augmented wave pseudopotentials with the cutoff energy of 500 eV were employed.

Considering the interaction between the hydrogen atoms and high-electronegativity groups, the PBE with the DFT-D3 dispersion correction of Grimme with zero-damping^{69–71} was applied to optimize the geometric structures. During the optimization of the geometries, all structures were allowed to relax to ensure that each atom was in mechanical equilibrium without any residual force larger than 10^{-4} eV Å^{−1}.

Once the geometry optimization was completed, the repeated velocity rescaling was performed to equilibrate the structures at 300 K for 5 ps. Then, 8 ps adiabatic molecular dynamics trajectories were obtained within the microcanonical ensemble with the molecular dynamics time step set to 1 fs. To perform the NAMD simulations, we computed the non-adiabatic couplings d_{ij} using the time-domain numerical method⁷² via the overlap between orbitals i and j at sequential time steps $d_{ij} = -i\hbar \langle \phi_i | \nabla_R | \phi_j \dot{R} \rangle = -i\hbar \left\langle \phi_i \left| \frac{\partial}{\partial t} \right| \phi_j \right\rangle = -i\hbar \frac{\langle \phi_i(t) | \phi_j(t + \Delta t) \rangle - \langle \phi_i(t + \Delta t) | \phi_j(t) \rangle}{2\Delta t}$.

The non-adiabatic coupling is proportional to the overlap of the wavefunctions between the initial (i) and final (j) states, $-i\hbar \langle \phi_i | \nabla_R | \phi_j \rangle$, and the nuclear velocity, dR/dt . The final 6000 configurations of the microcanonical molecular dynamics trajectories were selected as initial geometries for the following NAMD simulations of the nonradiative charge relaxation and

electron-hole recombination with the Python extension for *Ab Initio* Dynamics code.^{73,74} Since the structures had direct bandgaps located at the Γ -point, the NA couplings were computed for the Γ -point only.

3 Results and discussion

3.1 Geometric structures

At first, we optimize all the possible structures of pristine $\text{MA}_2\text{SnPbI}_6$ to determine which kind of topological arrangement of the metal cations gives the highest stability. For this reason, we build a few cubic perovskite-phase $\text{MA}_2\text{SnPbI}_6$ structures with different arrangements of Pb and Sn atoms, as shown in Fig. 1. Then, we optimize them and calculate their free energies. The results in Table 1 prove that the structure in Fig. 1(c) with the face-to-face arrangement of Pb and Sn atoms possesses the best thermal stability. This minimum unit cell is able to generate all internal point defects. Thus, this configuration is adopted in the later study at an affordable computational cost.

The influence of V_{Pb} , V_{Sn} , and i_{I} on the structure of $\text{MA}_2\text{SnPbI}_6$ is then explored. All the defective structures are relaxed with the lattice parameters fixed at the defect-free values to simulate point defects in a bulk crystal. To describe the influence of the defects on the system's geometry, we show the topological structure parameters of the defect-adjacent atoms in the flawed structures in Fig. 2 and Table 2, which demonstrate that when V_{Pb} or V_{Sn} appears, the average Sn–I bond lengths are longer than those without any defect. Thus, the atoms around the vacancies tend to move backwards the defect sites, so that the $[\text{SnI}_6]^{4-}$ octahedra are strongly distorted to accommodate these inert defects. However, it has weaker effect towards the structure of $[\text{PbI}_6]^{4-}$ octahedra. In addition, the vacancy defects flatten the metal-halide framework, as manifested in the increase of the average $\angle \text{M-X-M}$ angles. For the i_{I} structure, the atoms around i_{I} are pulled away from the point defect by the adjacent framework, as reflected by the shorter

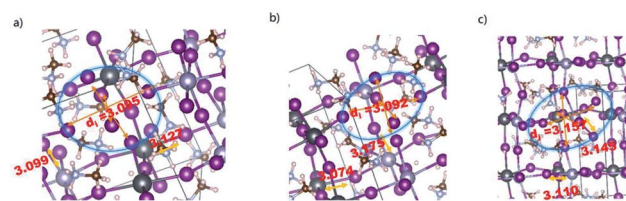


Fig. 2 Optimized defect-bearing structures of the $\text{MA}_2\text{SnPbI}_6$ perovskites: (a) V_{Pb} , (b) V_{Sn} and (c) i_{I} .

bond lengths, smaller bond angles and distortion of M–X octahedral coordination.

The formalism of DFE allows calculation of defect and impurity equilibrium structures and concentrations, and the relative stability of the different charge states of a given defect in the crystals.⁷⁵ We also calculated the relative DFEs of those point defects (see Table 2). The DFEs of charged defects are affected by the Fermi energy and the valence band level. The DFEs are expressed as⁷⁶ $\text{DFE} = E_{\text{defected}} - (E_{\text{perfect}} + \sum \epsilon_{i,\text{dopants}}) + q(E_{\text{f}} + \text{VBM} + \Delta V_{\text{corr}}) + V_{\text{corr}}$, in which, E_{f} represents the Fermi energy, and ΔV_{corr} means the VBM potential correction with the bulk flawless phase reference. q denotes the charge state of a defect: for a neutral defect, $q = 0$, and if the defect structure has a positive charge, $q = 1$. V_{corr} is set for the electrostatic interaction correction between charged point defects in the periodically repeated supercells. In this work, the Makov–Payne

approach⁷⁷ with the monopole approximation, $V_{\text{corr}} = \frac{\alpha q^2}{\epsilon L}$, is adopted to calculate this term, in which the constant α represents the Madelung factor,^{78–80} and ϵ is the dielectric constant obtained from the density functional perturbation theory.^{81,82} The calculated DFEs demonstrate that i_{I} is generated most easily, followed by V_{Pb} and V_{Sn} . When a Sn(II) cation gets oxidized to Sn(IV) , vacancy formation in the adjacent Pb(II) atoms is more likely than that in the Sn(II) atoms.

Considering that the intrinsic defects in actual system are often scattered in the lattice, we build a structure expressed as $\text{MASn}_{0.03125}\text{Pb}_{0.96875}\text{I}_3$, which possesses a large unit cell constructed by setting the minimum unit cell of $\text{MA}_2\text{SnPbI}_6$ shown in Fig. 1(c) in the center and surrounding it by the high-stability MAPbI_3 (see Fig. S1†). Then we investigated the influence of three kinds of point defects on the topological structure and calculate DFEs. The data shown in Table 2 and S1† tell that the

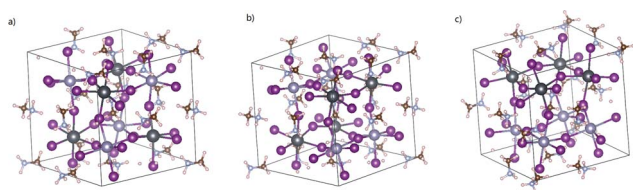


Fig. 1 Optimized structure of mixed Pb–Sn perovskite $\text{MA}_2\text{SnPbI}_6$.

Table 1 Geometric parameters and free energies for the $\text{MA}_2\text{SnPbI}_6$ structures without defects

Structures	(a)	(b)	(c)
$a/\text{\AA}$	12.25	11.74	11.16
$b/\text{\AA}$	12.23	12.25	12.20
$c/\text{\AA}$	12.28	12.28	12.21
Free energy/eV	−417.302	−417.303	−417.360

Table 2 The main geometric parameters and defect formation energy (DFE) of $\text{MA}_2\text{SnPbI}_6$ with and without defects. \bar{d}_{I} denotes the average distance for the vacancy-sharing atoms in the flawed crystal

Structures	Pristine	V_{Sn}	V_{Pb}	i_{I}
$d_{\text{Sn-I}}/\text{\AA}$	3.134	3.175	3.099	3.110
$d_{\text{Pb-I}}/\text{\AA}$	3.185	3.074	3.127	3.145
$\angle \text{Sn-I-Sn}/^\circ$	154.2	161.1	156.4	147.2
$\angle \text{Pb-I-Pb}/^\circ$	153.5	165.6	167.8	153.2
$\min(\angle \text{M-X-M})^\circ$	148.6	146.6	148.2	108.2
$\bar{d}_{\text{I}}/\text{\AA}$	—	3.092	3.095	3.157
DFE/(eV per cell)	—	+0.294	+0.128	+0.045

influence of internal defects on $\text{MA}_2\text{SnPbI}_6$ and $\text{MASn}_{0.03125}\text{Pb}_{0.96875}\text{I}_3$ is consistent. However, the defect-induced deformation on the $\text{MASn}_{0.03125}\text{Pb}_{0.96875}\text{I}_3$ lattice is slightly weaker than that on the $\text{MA}_2\text{SnPbI}_6$ lattice because of the tolerance of a large amount of stable MAPbI_3 in the former. The tolerance effect of the surrounding protective layer on the bond length of the defect-sharing atoms is less than 3%, indicating that the deformation on the defect-sharing structure is not affected by the lattice size. Therefore, to achieve a better balance between the computational cost and accuracy, we adopt the $\text{MA}_2\text{SnPbI}_6$ structure for later study.

3.2 Electronic and optoelectronic properties

Before NAMD calculations and carrier recombination analysis, we first make a static energy band analysis of the electronic structure. The band structure and projected density-of-states (pDOSs) of $\text{MA}_2\text{SnPbI}_6$ with and without the defects are plotted in Fig. 4 and S3–S5.†

The calculated band gap of $\text{MA}_2\text{SnPbI}_6$ is between those of MAPbI_3 (1.57 eV) and MASnI_3 (1.28 eV) by HSE06 + SOC, attributed to the hybridization of inert Pb-6s orbitals (with a minor 6p contribution) and active Sn-5s orbitals (with a minor 5p contribution) in the VBM. The CBM of $\text{MA}_2\text{SnPbI}_6$ is also composed of Pb-6p, Sn-5p and I-5p. i_I creates a characteristic deep defect state (DDS) lying in the band gap near the Fermi level, which is 0.43 eV above the VBM, and 0.26 eV below the CBM. This DDS is composed of the I-5p orbital of the interstitial I^- , and its energy dispersion $E(k)$ is nearly flat.

The characteristic defect states created by V_{Pb} and V_{Sn} are located below the VBMs, and they thus do not change the width of the band gap. However, the vacancies create several impurity states under the VBM and change the energy dispersions, indicating that they will play a role in the carrier migration and recombination. The pDOS in Fig. S3† shows that V_{Sn} weakens the contribution of Sn-5s to the VBM and CB edge states, while V_{Pb} had no such severe influence. The contribution of Pb-6s orbitals to the pDOS starts at lower energies compared to Sn-5s, and therefore, the influence of V_{Pb} on the VBM near the Fermi level is weaker than that of V_{Sn} . The charge distributions of the pristine and flawed $\text{MA}_2\text{SnPbI}_6$ are shown in Fig. S4.†

The relative mobility of the charge carriers can be characterized by the carrier effective masses.^{83,84} The effective masses are calculated by fitting the energy dispersions of the VBM and the CBM to quadratic functions as $\frac{1}{m_{ii}^*} = \frac{1}{\hbar^2} \frac{\partial^2 E_n(k_i)}{\partial k_i^2}$ ($i = x, y, z$), where m_{xx} , m_{yy} , and m_{zz} correspond to the components in the x , y , and z directions, respectively. The results calculated for $\text{MA}_2\text{SnPbI}_6$ with and without defects are shown in Fig. 3(b).

When Sn atoms are present in the structure, the effective electron mass of the CBM (\bar{m}_c) is very close to or even slightly lower than that of the pure Pb-based perovskite (approx. 5%⁸⁵), indicating higher electron carrier mobility. Introduction of V_{Pb} and i_I can inhibit the electron mobility to a certain extent, and V_{Sn} has no obvious effect on it. In addition, these intrinsic defects have a significant impact on the hole mass in the VBM (\bar{m}_v), especially i_I .

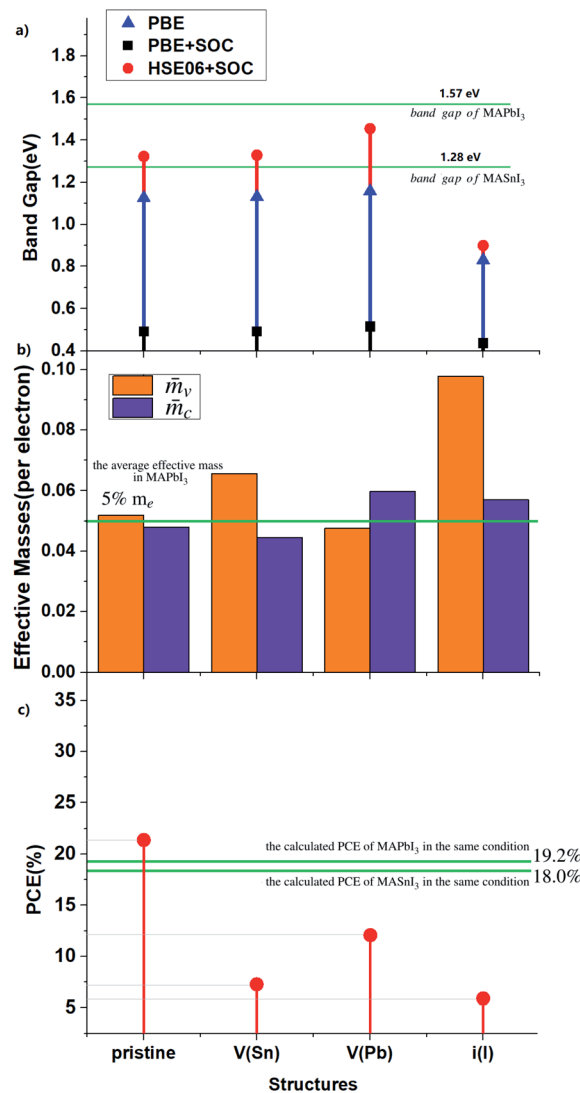


Fig. 3 (a) Band gaps calculated by PBE, PBE + SOC and HSE06 + SOC. (b) Effective masses and (c) PCEs calculated by PBE + SOC for the perfect and defective $\text{MA}_2\text{SnPbI}_6$. Comparison is made with the pure Pb-based and Sn-based perovskites.

V_{Sn} and V_{Pb} insert defect states below the VBM, and increase the energy dispersion of the VBM without changing the band gap magnitude. Since the VBMs arise mainly from the Sn-5s orbitals, V_{Sn} reduces the VBM dispersion more significantly. In other words, vacancy defects, such as V_{Sn} and V_{Pb} , make the VBM more localized.

The band gap is a key property of a light-to-energy conversion device, putting limits on its efficiency. Photons with a lower energy than the band gap cannot be absorbed efficiently, while electron-hole pairs generated above the band gap usually relax to the band edges, so that only part of the absorbed energy can be turned into work. In this sense, the absorption spectrum directly determines the PCE value. Hereafter, we adopt the spectral-limited maximum efficiency method as the PCE descriptor, going beyond the band gap, as suggested by Yu and Zunger^{86,87} to calculate the number of photons absorbed by

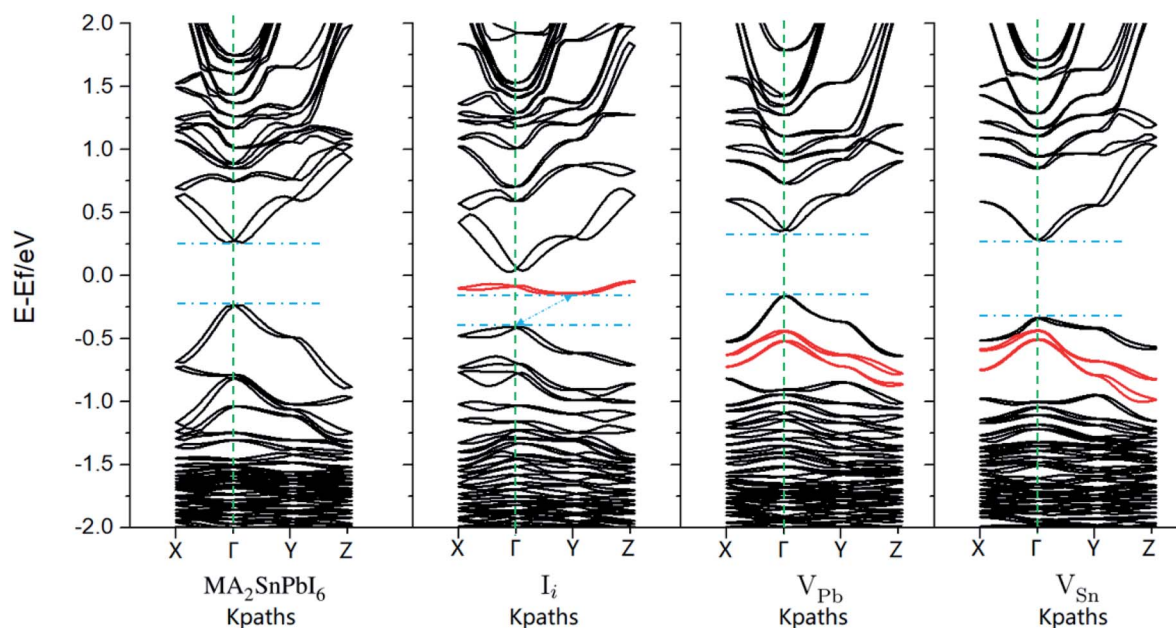


Fig. 4 Band structure of $\text{MA}_2\text{SnPbI}_6$ (a) without and (b–d) with intrinsic defects calculated by PBE + SOC. The red lines highlight the bands created by the intrinsic defects.

each structure. The efficiency, η , is obtained as $\eta = \frac{1}{n_{\text{tot}}} \int_{E_{\text{g}}^{\text{dir}}}^{\infty} \text{ph}_{\text{abs}}(E) n_{\text{ph}}(E) dE$. Here, n_{tot} is the total number of photons emitted by the sun at AM1.5,⁸⁸ $\text{ph}_{\text{abs}}(E)$ is the photon absorptivity characteristic of the material, and $n_{\text{ph}}(E)$ represents the number of sun photons at the energy, E , in the unit of eV. In this model, no absorption occurs below the direct bandgap, $E_{\text{g}}^{\text{dir}}$, since absorption through an indirect transition requires a change in the electron momentum assisted by phonons.

The calculated PCEs of the materials under investigation corresponding to the same optoelectronic device thickness (shown in Fig. S6†) are shown in Fig. 3(c) and S5 and S6.† The calculated results show that the band gap of pristine $\text{MA}_2\text{SnPbI}_6$ is closer to the Shockley–Queisser limit value of 1.37 eV (ref. 89) than the MAPbI_3 and MASnI_3 band gaps. Therefore, it exhibits higher PCE. All three point defects decrease the PCE. V_{pb} has the weakest impact on the electronic structure and PCE. V_{sn} decreases the PCE more than V_{pb} . The DDS induced by I_i greatly changes the band structure and has the most serious impact on the PCE.

3.3 Electron–hole recombination and hot electron relaxation

In order to quantitatively describe the influence of the intrinsic defects on the carrier lifetimes, we have performed a series of NAMD simulations on the photexcited carrier relaxation and electron–hole recombination processes for the perfect and defective structures. Considering the computational effort, the NAMD simulations are based on the PBE functional without SOC since the band gaps calculated by PBE without SOC are much closer to those obtained by HSE06 + SOC than the band gaps obtained by PBE + SOC, and NAMD calculations based on HSE06 + SOC are too computationally demanding. The

calculated non-adiabatic coupling (NAC) matrices are shown in Fig. S7.†

Fig. 5 shows the pure-dephasing functions calculated using the second order cumulant expansion as $D(t) = \exp\left(-\frac{1}{\hbar^2} \int_0^t dt' \int_0^{t'} dt'' C(t'')\right)$. This optical response function characterizes the vibrationally induced pure-dephasing process for a pair of states entangled in a coherent superposition. It can be obtained directly or *via* the second-order cumulant expansion,⁹⁰ which allows notably better convergence. Pure-dephasing is associated with fluctuations of electronic energy levels due to coupling to phonons. The fluctuations are characterized by the energy gap autocorrelation function (ACF). The un-normalized ACF is defined as $C(t) = \langle \Delta E(t) \Delta E(0) \rangle$. It represents an ensemble average of the correlation of the energy gap fluctuation. A rapid decay of the ACF implies fast pure-dephasing. The ACFs characterize memory of the energy gap fluctuation.

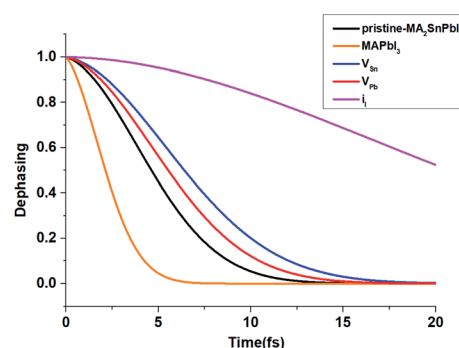


Fig. 5 Calculated pure-dephasing function of $\text{MA}_2\text{SnPbI}_6$ with and without defects. The result comparison is made with MAPbI_3 .

Generally, shorter pure-dephasing time leads to slower dynamics, as exemplified by the quantum Zeno effect.⁹¹ The pure-dephasing times depend most strongly on the magnitude of the phonon-induced fluctuation of the corresponding electronic energy gap.⁹² Fig. S8(a)† shows the un-normalized ACF of the energy gap fluctuation, and the ACF initial values characterize the fluctuation magnitude. The fluctuation is the largest for the pristine system, resulting in the fastest decay of the pure-dephasing function. The i_I defect exhibits slower pure-dephasing, because the corresponding energy gap and its fluctuation are the smallest.

Fourier transform of the ACF produces the spectral density $I(\omega) = \left| \frac{1}{2\pi} \int_{-\infty}^{\infty} dt \exp(-i\omega t) C(t) \right|^2$. As shown in Fig. S8(b),† the calculated spectral densities characterize the phonon modes that couple to the electronic subsystem. Low-frequency modes play a predominant role, corresponding to twisting and stretching motion of the $[\text{PbI}_6]^{4-}$ and $[\text{SnI}_6]^{4-}$ octahedra, according to the previous research.^{93,94}

To determine the time scales τ of the non-radiative electron-hole recombination, we obtain 30 ps of NAMD data, and fit the state populations to the short-time, linear approximation of exponential decay, *i.e.*, $P(t) = \exp(-t/\tau) \approx 1 - t/\tau$. Table 3 demonstrates that the carrier lifetime of the $\text{MA}_2\text{SnPbI}_6$ mixed perovskite is between those of MAPbI_3 and MASnI_3 . Due to the higher VBM energy, pristine $\text{MA}_2\text{SnPbI}_6$ possesses a smaller band gap, and therefore, exhibits a shorter carrier lifetime than MAPbI_3 . On the other hand, the incorporation of Pb into the Sn-based perovskite increases the carrier lifetime compared with that of the pure Sn-based perovskite. All the defective $\text{MA}_2\text{-SnPbI}_6$ perovskites exhibit shorter carrier lifetimes than the pristine system. This can be traced to smaller initial values of the unnormalized ACF and longer pure-dephasing times. The carrier lifetimes of the perovskites with the cation vacancies are shorter because the VBM state is mixed with the defect states formed in the VB, Fig. 6(a). The V_{Sn} defect influences the electronic states near the VBM more strongly than the V_{Pb} defect, because it is energetically closer to the VBM, and notably hybridizes the VBM and deeper VB state, Fig. S3.† As a result, V_{Sn} changes the nonradiative charge recombination between the VBM and CBM more significantly than V_{Pb} . Changes in the spatial distributions of charges near the VBM and the CBM, Fig. S4,† are an important reason for variation of the carrier lifetime in different structures.⁹⁶

The i_I -bearing system exhibits much faster nonradiative carrier recombination than the other systems studied in this work, Fig. 6. The DDS plays a decisive role here. The i_I creates DDS, Fig. 4(a), splitting the band gap into two smaller gaps,

Table 3 Calculated time scales of the carrier recombination for $\text{MA}_2\text{SnPbI}_6$ with and without defects. Comparison is made with MAPbI_3 and MASnI_3

Structure	SnPb	V_{Pb}	V_{Sn}	i_I	all-Pb	all-Sn ⁹⁵
τ/ns	54.88	20.11	5.72	0.441	159.6	8.37

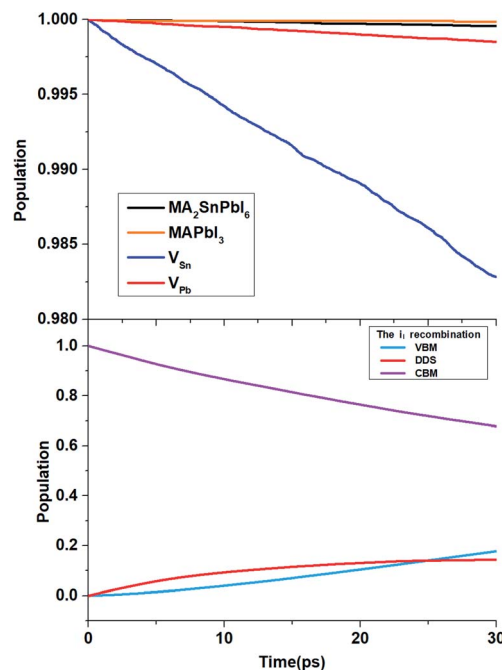


Fig. 6 Electron-hole recombination process from the CBM to the VBM for $\text{MA}_2\text{SnPbI}_6$ with and without defects.

0.43 eV + 0.26 eV by PBE + SOC. Nonradiative transitions across small energy gaps are fast, resulting in quick carrier quenching. 32% of active carriers recombine within the 30 ps NAMD simulation and 14% of them are captured by the interstitial DDS, Fig. 6(b).

To unveil the effect of these point defects on the carrier relaxation process, we simulate the hot electron relaxation dynamics from the low-lying CB edge states to the CBM, as shown in Fig. 7. Considering hot charge carriers at energies away from the band edges can provide advantages in solar energy applications, such as faster charge separation and transport, generation of additional carriers, and reduced

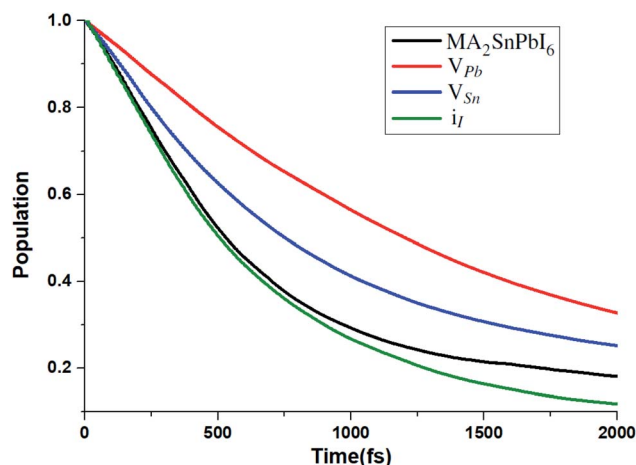


Fig. 7 Hot electron relaxation process from CBM + 1, CBM + 2 and CBM + 3 to the CBM for $\text{MA}_2\text{SnPbI}_6$ with and without defects.

trapping,⁹⁷ here we include all the relaxation effects from CBM + 1, CBM + 2 and CBM + 3 to the CBM.

The CB edges are mainly composed of Sn-5s 5p and Pb-6s6p orbitals (see Fig. S3†). Therefore, V_{Pb} and V_{Sn} significantly weaken the contribution of Pb and Sn to CB edges, leading to the increase of the relaxation time. The calculated carrier relaxation times τ for the V_{Pb} - and V_{Sn} -bearing structures are 2.02 and 1.29 ps, respectively. Because the i_{I} defect creates the mid-gap states and hardly affects the deep CB states, the carrier relaxation time of the i_{I} -bearing structure ($\tau = 0.97$ ps) is nearly the same as that of the defect-free structure ($\tau = 1.01$ ps).

4 Concluding remarks

We have performed first-principles DFT calculations combined with NAMD simulations on the static and dynamic characteristics of $\text{MA}_2\text{SnPbI}_6$ with and without intrinsic point defects, and compared its properties with those of its pure counterparts MASnI_3 and MAPbI_3 . Meantime, in order to discuss the stability of defects in the realistic system as accurately as possible, we also calculated the geometric structures and defect formation energies of $\text{MASn}_{0.03125}\text{Pb}_{0.96875}\text{I}_3$ with a large supercell. This study unveils the nature of defect states and reveals how the intrinsic defects in the mixed Pb–Sn perovskite frameworks affect the electronic and geometric structures, the electron-vibrational coupling, the hot electron relaxation times, and the times of the nonradiative electron–hole recombination process at the ambient temperature. The following conclusions can be extracted from this study.

(1) The oxidation of Sn(II) results in the generation of many intrinsic defects such as V_{Sn} , V_{Pb} and i_{I} in the $\text{MA}_2\text{SnPbI}_6$ material. i_{I} can be most easily generated, followed by V_{Pb} and V_{Sn} . When one Sn(II) atom is oxidized, its adjacent Pb(II) atoms are prone to forming a defective vacancy, prior to Sn(II) atoms, and once a defect appears nearby, the $[\text{SnI}_6]^{4-}$ octahedra are always distorted more severely than $[\text{PbI}_6]^{4-}$ octahedra. It has been noted that Sn–Pb mixed perovskite materials possess an important feature: when Sn(II) is oxidized to Sn(IV), an adjacent Pb-vacancy forms, which will partially tolerate the negative impact of Sn(IV) on the photoelectric properties of Sn–Pb mixed perovskite materials.

(2) V_{Pb} , V_{Sn} and i_{I} affect the crystal structure differently. V_{Pb} and V_{Sn} cause elongation and distortion of the MX octahedra around the defects. This is reflected in a displacement of the octagon geometric centers. The M–X bond lengths of the center ion increase to offset somewhat the distortion of the octahedral structure. i_{I} makes the two adjacent MX octahedra repel each other, manifested in the reduction of the M–X bond lengths. Additionally, the vacancy defects flatten the M–X framework, while i_{I} distorts it severely.

(3) The band gap width and the carrier lifetime of $\text{MA}_2\text{SnPbI}_6$ lie between those of MAPbI_3 and MASnI_3 because the orbital hybridization of Sn and Pb affects the energy level of its VBMs, which are composed of the Sn-5s5p and Pb-6s6p orbitals. $\text{MA}_2\text{-SnPbI}_6$ exhibits higher PCE than MAPbI_3 and MASnI_3 because its band gap width is much closer to the Shockley–Queisser limit. The tolerance of Pb-vacancy defects by the mixed Pb–Sn perovskite is also reflected in the electronic structure. The CBMs of

$\text{MA}_2\text{SnPbI}_6$ are predominantly composed of the Sn-5p orbitals so that removing a Pb atom has a nearly negligible impact on its CB-related properties. Therefore, we can conclude that the existence of Pb atoms can inhibit the degradation arising from Sn oxidation in $\text{MA}_2\text{SnPbI}_6$. This is of great significance for maintaining the photoelectric activity of Sn-bearing perovskites.

(4) The point defects shorten the carrier lifetime of the material in the following order, V_{Pb} , V_{Sn} and i_{I} , with i_{I} giving the fastest carrier losses. The i_{I} defect creates deep defect states in the middle region of the band gap, which introduce new carrier recombination pathways, leading to a significant decrease of the carrier lifetime and the PCE. 32% of the carriers are lost within 30 ps due to the nonradiative electron–hole recombination in the presence of i_{I} and 14% of them are captured by the deep defect states created by the i_{I} defect. In order to avoid formation of the i_{I} defects, we suggest to decrease the anion content, or increase the cation content, to prevent the generation of i_{I} during nucleation of the crystal. Both V_{Sn} and V_{Pb} do not create defect levels in the band gap. However, they alter the energy dispersions of both the VBM and the CBM, changing the carriers' mobility and lifetime. The V_{Sn} defect weakens hybridization of the Pb and Sn orbitals, such that the Sn-5s and 5p orbitals contribute to the VBM very weakly around the defect. V_{Pb} has less impact on the material than V_{Sn} , because Pb atoms contribute less to the VBM edge states.

(5) Contrary to their influences on the electron–hole recombination process, the three kinds of point defects exhibit an impact on the hot carrier relaxation process with the order of i_{I} , V_{Sn} and V_{Pb} . i_{I} hardly affects the relaxation and both V_{Pb} and V_{Sn} slow down the relaxation, attributed to the defect-induced different changes on the energy dispersions of CB edge states.

This study establishes the microscopic mechanisms underlying the simultaneous improvement of the perovskite stability and the charge carrier lifetime by the hybrid cation strategy. It reveals the nature of the defect states and the influence of the intrinsic defects on the structures and optoelectronic properties, as well as the charge carriers' mobility and lifetimes. The general and fundamental principles established by analysis of the simulation results are useful for design of advanced materials for solar energy and optoelectronic applications.

Conflicts of interest

There are no conflicts to declare.

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