

Structural Characterization of Human Milk Oligosaccharides (HMOs) using Ultra-High Performance Liquid Chromatography-Helium Charge Transfer Dissociation Mass Spectrometry (UHPLC-He-CTD-MS)

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Running header: Milk oligosaccharides by UHPLC-He-CTD-MS

Abstract

The combination of helium charge transfer dissociation mass spectrometry (He-CTD-MS) with ultra-high performance liquid chromatography (UHPLC) is presented for the analysis of a complex mixture of acidic and neutral human milk oligosaccharides (HMOs). The research focuses on the identification of the monosaccharide sequence, the branching patterns, sialylation/fucosylation arrangements, and the differentiation of isomeric oligosaccharides in the mixture. Initial studies first optimized the conditions for the UHPLC separation and the He-CTD-MS conditions. Results demonstrate that He-CTD is compatible with UHPLC timescales and provides unambiguous glycosidic and cross-ring cleavages from both the reducing and the non-reducing ends, which is not typically possible using collision-induced dissociation (CID). He-CTD produces informative fragments, including $^{0,3}A_n$ and $^{0,4}A_n$ ions, which have been observed with electron transfer dissociation (ETD), electron detachment dissociation (EDD), and ultraviolet photodissociation (UVPD) and are crucial for differentiating the α -2,3- vs α -2,6-linked sialic acid (Neu5Ac) residues present among sialyllacto-*N*-tetraose HMOs. In addition to the linkage positions, He-CTD is able to differentiate structural isomers for both sialyllacto-*N*-tetraoses and lacto-*N*-fucopentaoses structures by providing unique, unambiguous cross ring cleavages of types $^{0,2}A_n$, $^{0,2}X_n$ and $^{1,5}A_n$ while preserving most of the labile Neu5Ac and fucose groups.

Introduction

Human breast milk is nature's gold standard of nutrition for newborns (Remoroza *et al.*, 2018). Breast milk contains 6-20 g/L of non-conjugated carbohydrates that are known as human milk oligosaccharides (HMOs) (Ninonuevo and Lebrilla, 2009; Bao *et al.*, 2013; Auer *et al.*, 2021). HMOs comprise more than 130 distinct structures and are the third most abundant component in breast milk after lactose and lipids (Bao *et al.*, 2013; Nijman *et al.*, 2018; Auer *et al.*, 2021). HMOs can be categorized as neutral or acidic in nature depending on their monosaccharide composition (Plaza-Díaz *et al.*, 2018). Neutral HMOs are further categorized into fucosylated and *N*-containing HMOs, the latter of which contains *N*-acetylglucosamine (GlcNAc) at the terminal positions (Plaza-Díaz *et al.*, 2018). Acidic HMOs contain sialic acid (Neu5Ac) at one or both termini and represent 12%-14% of the total HMO content (Plaza-Díaz *et al.*, 2018). HMOs help strengthen the immune system in infants, and they can prevent the adhesion of pathogens to epithelial cells, which blocks the initial step of infection (Ninonuevo and Lebrilla, 2009; Remoroza *et al.*, 2018). HMOs also serve as prebiotics that promote the growth of beneficial gut bacteria (Ninonuevo and Lebrilla, 2009). Specific HMOs, including 2'-fucosyllactose (2'FL) and 3/6'-sialyllactose (3'SL; 6'SL), are beneficial to the brain development of infants (Ruhaak and Lebrilla, 2012a; Berger *et al.*, 2020; Gu *et al.*, 2021).

HMOs can be seen with linear or branched structures of three to ten monosaccharide constituents (Bao *et al.*, 2013), and have a core structure at the reducing end commonly made with lactose (Gal β 1-4Glc) or *N*-acetyllactosamine (Gal β 1-4GlcNAc) (Remoroza *et al.*, 2018; Auer *et al.*, 2021). Elongation occurs through the addition of β -1,3- or β -1,6-linked lacto-*N*-biose (Gal β 1-3GlcNAc, Type I) or *N*-acetyllactosamine (Gal β 1-4GlcNAc, Type II) disaccharide component to the core structure. HMOs with linear and branched structures are named as *para*-HMOs and *iso*-

HMOs, respectively (Bode, 2012; Plaza-Díaz *et al.*, 2018). Chain branching is mainly initiated with the introduction of disaccharide units to the main chain via β -1,6 linkages (Plaza-Díaz *et al.*, 2018). The core and the elongated structures can be further modified with α -1,2-, α -1,3- or α -1,4-linked fucose (Fuc) or α -2,3- and/or α -2,6-linked Neu5Ac units, which results in a collection of both structural and linkage isomers. (Bode, 2012; Plaza-Díaz *et al.*, 2018; Auer *et al.*, 2021)

The biological activity and health benefits of HMOs depend on the details of the linkage patterns, the branching positions, and sites of modification, so a wide range of analytical techniques must usually be incorporated to fully characterize and quantify the different HMOs. Some of the key techniques that are widely used for HMOs analysis include size-exclusion chromatography (SEC) (Marino *et al.*, 2011; Grabarics *et al.*, 2017), liquid chromatography (LC) (Austin and Bénét, 2018; Remoroza *et al.*, 2018), reversed-phase high-performance chromatography (RP-HPLC) (Leo *et al.*, 2010), high-pH anion-exchange chromatography (HPAEC) (Gu *et al.*, 2021), porous graphitized carbon (PGC) (Hong *et al.*, 2014; Gu *et al.*, 2021) and hydrophilic interaction liquid chromatography (HILIC) (Marino *et al.*, 2011; Remoroza *et al.*, 2018), capillary electrophoresis (CE) (Bao and Newburg, 2008; Galeotti *et al.*, 2014), micellar electrokinetic chromatography (MEKC) (Porfirio *et al.*, 2020), nuclear magnetic resonance (NMR) spectroscopy (Van Leeuwen *et al.*, 2014) and mass spectrometry (MS) (Ninonuevo and Lebrilla, 2009; Ruhaak and Lebrilla, 2012b; Remoroza *et al.*, 2018).

Offline and online MS profiling of HMOs have been considered as key techniques in characterizing HMOs, mainly due to the speed and ability to identify molecular masses and structural information (Adamson and Håkansson, 2007). Offline-MS techniques such as matrix-assisted laser desorption/ionization time-of-flight mass spectrometry (MALDI-TOF-MS) are widely used for HMO characterization and are effective in distinguishing α -2,3 and α -2,6-

sialylated isomers via post-source decay (PSD) fragmentation (Von Seggern *et al.*, 2003; Nie *et al.*, 2012). Generally speaking, MALDI can tolerate high salt concentrations in the samples, but MALDI ionization tends to cause the unwanted loss of labile Neu5Ac groups, so is sometimes limited in its application to HMOs (Von Seggern *et al.*, 2003). These drawbacks can be overcome by performing suitable purification procedures and using optimized matrices, which add complexity to the experimental workflow (Selman *et al.*, 2012).

Electrospray ionization (ESI) and nano-ESI are useful in performing online-MS analysis for both derivatized and underivatized HMOs (De Leoz *et al.*, 2019). Online separation techniques coupled with tandem MS (MS/MS) detection enable deeper insight into the structural characteristics of complex samples (Han and Costello, 2011; Oursel *et al.*, 2017). Glycosidic fragments (B/Y and C/Z) are produced when C-O glycosidic bonds are broken between two neighboring monosaccharide residues (Domon and Costello, 1988). Cross-ring fragments (A/X) are produced when C-C or C-O bonds are broken within a monosaccharide ring (Domon and Costello, 1988). Cross-ring cleavages tend to be less common because they require breaking two covalent bonds to be observed. Glycosidic fragments provide details about the composition and sequence of an oligosaccharide, but cross-ring fragments can provide details about the location of sugar modifications—like methylation, acetylation, sulfation—and linkage positions between sugar residues (Domon and Costello, 1988; Han and Costello, 2011). Different MS/MS techniques have been involved in HMOs analysis and they have demonstrated characteristic features to each technique.

Collision-induced dissociation (CID) produces glycosidic cleavages and a few cross-ring cleavages of native HMOs (Mank *et al.*, 2019). Even though CID can provide valuable diagnostic fragments (De Leoz *et al.*, 2019; Mank *et al.*, 2019), CID also enables rearrangements,

consecutive, internal fragments and multiple neutral losses—such as H₂O and CO₂—which complicate the product ion spectra (Ernst *et al.*, 1997; Schaller-Duke *et al.*, 2018; De Leoz *et al.*, 2019; Mank *et al.*, 2019). CID with metalated HMOs provides more structurally informative fragments relative to their protonated equivalents (Adamson and Håkansson, 2007; Han and Costello, 2011; Schaller-Duke *et al.*, 2018), and high-energy collision-induced dissociation (HCD) can result in more cross-ring cleavages compared to conventional CID (Han and Costello, 2011).

Electron-based ion activation methods such as electron transfer dissociation (ETD) (Han and Costello, 2011), negative electron transfer dissociation (NETD) (Wolff *et al.*, 2010), electron capture dissociation (ECD) (Liu and Hakansson, 2011), electron-induced dissociation (EID) (Wolff *et al.*, 2008), electronic excitation dissociation (EED) (Tang *et al.*, 2018), and electron detachment dissociation (EDD) (Adamson and Håkansson, 2007) have all been applied in MS/MS analysis of oligosaccharides. These techniques have generally reduced the loss of labile modifications relative to CID, and they generally provide better sequence coverage of oligosaccharides relative to CID (Schaller-Duke *et al.*, 2018). Even though the electron-based ion-activation techniques are effective in the structural characterization of HMOs, most of these techniques are limited to Fourier-transform ion cyclotron resonance (FT-ICR) mass analyzers, whose expense limits widespread adoption (Adamson and Håkansson, 2007; Han and Costello, 2011).

Photon-based MS/MS dissociation methods, including infrared multiple photon dissociation (IRMPD) and ultraviolet photodissociation (UVPD), have created interest for the characterization of HMOs due to their ability to produce an array of informative cross-ring cleavages in a shorter time scale than most electron-based activation techniques (Ko and Brodbelt, 2011). IRMPD provides similar results to CID and can be used as a complementary method to

CID to gather structural details for HMOs (Schaller-Duke *et al.*, 2018). UVPD is capable of producing diverse fragment ions, including cleavages within Neu5Ac, and the formation of UVPD-specific $^{4,5}X$ ions—even in the absence of derivatization—is highly beneficial in studies involving Neu5Ac (Devakumar *et al.*, 2007; Wilson and Brodbelt, 2008; Ko and Brodbelt, 2011). Derivatization plays a significant role in UVPD analysis, which can directly affect the native structure analysis of oligosaccharides (Devakumar *et al.*, 2007; Wilson and Brodbelt, 2008; Ko and Brodbelt, 2011; Ropartz *et al.*, 2014).

Helium charge transfer dissociation (He-CTD) is an emerging ion-activation technique that shows promising results for oligosaccharides (Ropartz *et al.*, 2016, 2017; Buck-Wiese *et al.*, 2020; Pepi *et al.*, 2020), peptides (Hoffmann and Jackson, 2014), proteins (Li *et al.*, 2018), and lipids (Li and Jackson, 2017). He-CTD builds on the work of Schlathölter's group and Zubarev's group (Bari *et al.*, 2010, 2011; Chingin *et al.*, 2014) and uses a kiloelectronvolt beam of helium cations to ionize and fragment the precursor ions (Hoffmann and Jackson, 2014; Sasiene *et al.*, 2021a). Like UVPD and some electron-based methods, He-CTD produces a variety of structurally informative A/X cross-ring cleavages (Ropartz *et al.*, 2016, 2017; Buck-Wiese *et al.*, 2020; Pepi *et al.*, 2020; Sasiene *et al.*, 2021a, 2021b, 2021c). Buck-Weise *et al.* showed that He-CTD can be used to determine the connectivity patterns of β -1,3- and β -1,4-linked native glycans which is beneficial in linkage isomer differentiation (Buck-Wiese *et al.*, 2020). More recently, UHPLC-He-CTD-MS has been successfully used to analyze a range of complex oligosaccharides, including highly methylated pectins and highly sulfated carrageenan mixtures (Mendis *et al.*, 2021a, 2021b). These studies have shown that UHPLC-He-CTD-MS is capable of providing a series of unambiguous fragments from both the reducing and non-reducing ends, which has aided the identification of

site-specific modifications and isomeric structures of oligosaccharides in complex mixtures (Mendis *et al.*, 2021a, 2021b).

The current work employs UHPLC-He-CTD-MS for the structural determination of a series of neutral and acidic HMOs, including LNFP I, LNFP II, LSTa, LSTb, LSTc, and DSLNT. The structures are shown in **Fig. 1**. Results show that He-CTD produced more structurally informative cross-ring fragments than CID and that spectral acquisition rates are fast enough to handle complex mixtures of HMOs. The work demonstrates that He-CTD is a viable approach to high energy, radical-induced fragmentation of oligosaccharides, even when they are in the 1+ charge state and contain labile and acidic *N*-acetylneuraminic acid/sialic acid groups.

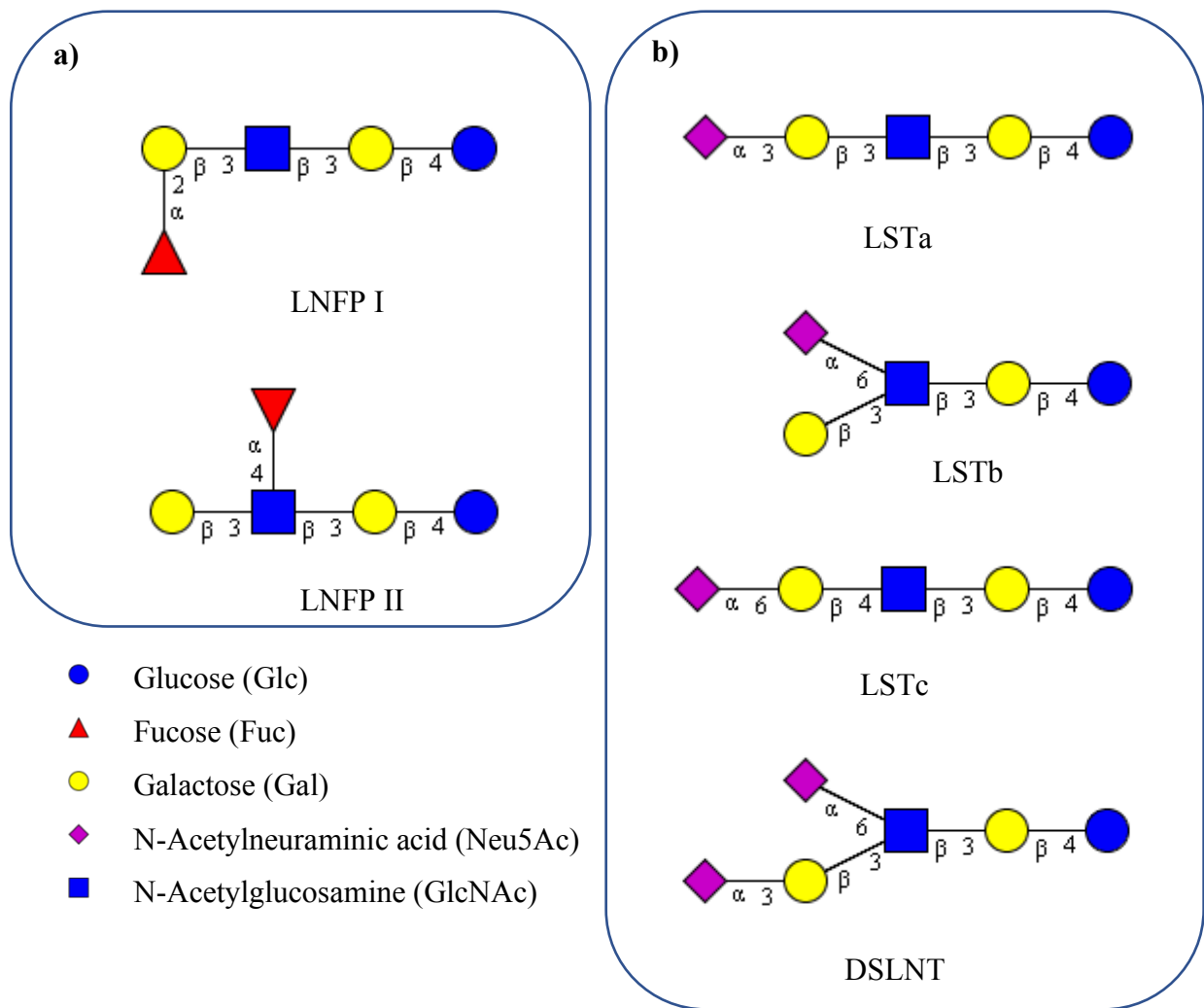


Fig. 1. Brief expansion of human milk oligosaccharides composition investigated in this work: a) neutral oligosaccharides b) acidic oligosaccharides

Results and discussion

First, we developed an optimized HILIC separation condition for a relatively simple synthetic mixture of HMOs. The mixture contained two neutral lacto-*N*-fucopentaoses (LNFP I and LNFP II, three acidic sialyllacto-*N*-tetraoses (LSTa, LSTb, LSTc), and an acidic disialyllacto-*N*-tetraose (DSLNT). The standard mixture was selected because the components have been relatively well characterized by other novel methods of tandem mass spectrometry and because they contains a variety of linkage patterns and modification sites of the labile sialic and fucose groups. Chromatographic separation was performed using the HILIC technique and the extracted ion chromatogram in **Fig. 2** illustrates the elution profile of the HMO mixture.

Both the neutral and acidic HMOs show almost baseline separation under the HILIC conditions, which is consistent with the work of others (Marino *et al.*, 2011; Grabarics *et al.*, 2017; Remoroza *et al.*, 2018). Neutral HMOs such as LNFP I and II elute first at 15.0 and 16.0 minutes (Marino *et al.*, 2011; Remoroza *et al.*, 2018). Neutral HMOs are non-sialylated compounds and are less polar than the acidic HMOs used in this study. Acidic HMOs such as LSTa, LSTb, and LSTc elute respectively at 16.2, 16.8, and 17.4 minutes. Acidic HMOs contain sialylated structures with additional carboxylic acid group(s), which display higher polarity compared to neutral oligosaccharides and elute later under HILIC conditions. Finally, DSLNT, which contains two Neu5Ac groups with a total of two carboxylic acid groups, is the largest in the size and has the highest polarity in the HMOs mixture; DSLNT therefore elutes last at 19.6 min.

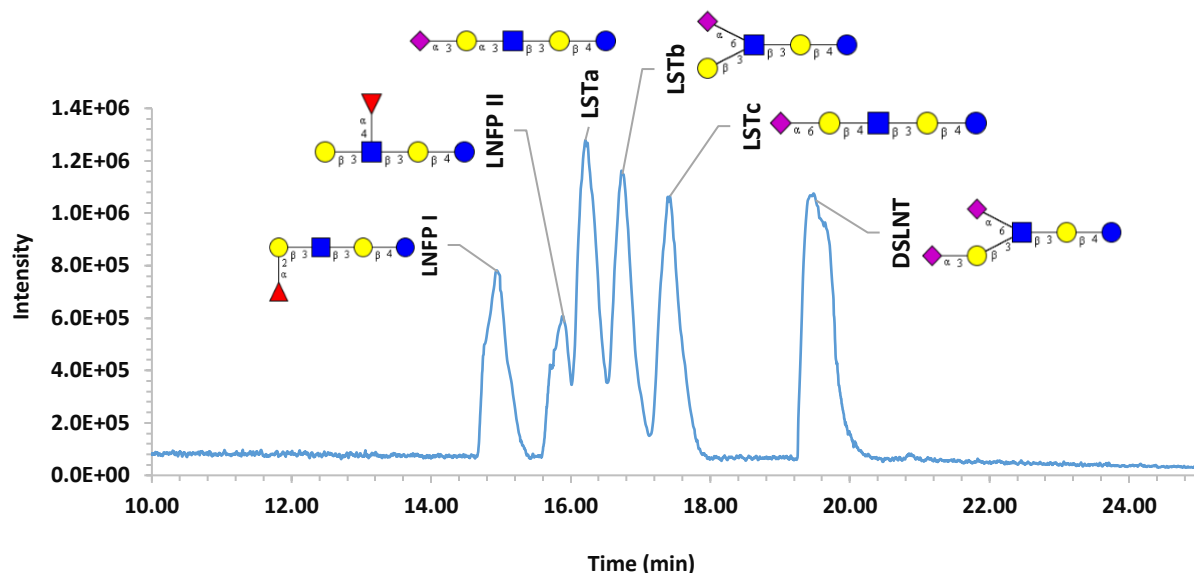


Fig. 2. Extracted ion chromatogram of HILIC separation of the complex mixture of milk oligosaccharides.

After the HILIC separation, HMO compounds were analyzed with either He-CTD or CID. The observed product ions were annotated according to Domon & Costello, and the largest branch in the oligosaccharide structure represents by the symbol α (Domon and Costello, 1988). The two neutral oligosaccharides, LNFP I and LNFP II, are structural isomers with different fucosylation positions (Pfenninger *et al.*, 2002a, 2002b; Mank *et al.*, 2019). For both LNFP I and LNFP II, the sodium adducted ion provided the most abundant precursor at m/z 876.3, so was selected for MS/MS analysis. Sodium was not intentionally added to any of the mobile phases.

Figs. 3a and 4a show the CID spectra for LNFP I and II, respectively. **Figs. 3b and 4b** show the He-CTD spectra of the same precursors under otherwise-indential conditions. CID produced an array of glycosidic fragments in LNFP I, including many unambiguous B/Y and C/Z cleavages. The observed fragments are spread throughout the oligosaccharide backbone. Compared to LNFP I, LNFP II provided fewer unambiguous fragments with CID, and they are

localized in between GlcNAc2 and Gal3 units. CID is able to produce diagnostic glycosidic fragment ions for each neutral HMO, including the C₂ fragment at m/z 349.0 (with LNFP I), and the C₂-Z_{3α} fragment at m/z 372.1 (with LNFP II), both of which comply with previous experiments using CID (Pfenninger *et al.*, 2002b, 2002a; Mank *et al.*, 2019). Both the diagnostic fragment ions provide valuable information on the composition of the non-reducing ends and are helpful in the preliminary differentiation of LNFP I and LNFP II (Mank *et al.*, 2019). In addition, CID is able to produce unambiguous cross-ring fragments on the GlcNAc unit in both LNFP I and II (^{2,4}X₂ and ^{0,2}X₂ respectively), but these fragments provide limited additional structural information.

In contrast to CID, He-CTD spectra for LNFP I and LNFP II in **Figs. 3b and 4b**, respectively, contain rich glycosidic and cross-ring fragments. The fragment ion maps inset in the figures shows that He-CTD produces several unambiguous fragments throughout the structures. Both LNFP I and II produced the diagnostic glycosidic products C₂ (m/z 349.1) and B₂-Y_{3α}/C₂-Z_{3α} at (m/z 372.0) respectively, which help to distinguish each isomeric structure, as described by Mank *et al* using CID. (Mank *et al.*, 2019). He-CTD exhibits unambiguous cross-ring fragment ions such as ^{1,5}A_n, ^{1,5}X_n and ^{0,2}X_n with both LNFP I and II, and these informative fragments provide information on the location of the labile fucose unit and provide information on the linkage position between the monosaccharide units. ^{1,5}A_n, ^{1,5}X_n and ^{0,2}X_n fragments have been previously observed with UHPLC-He-CTD-MS for carrageenan oligosaccharides, and these cross-ring fragments were useful for localizing the labile sulfate modification positions on the oligosaccharide structures (Mendis *et al.*, 2021a).

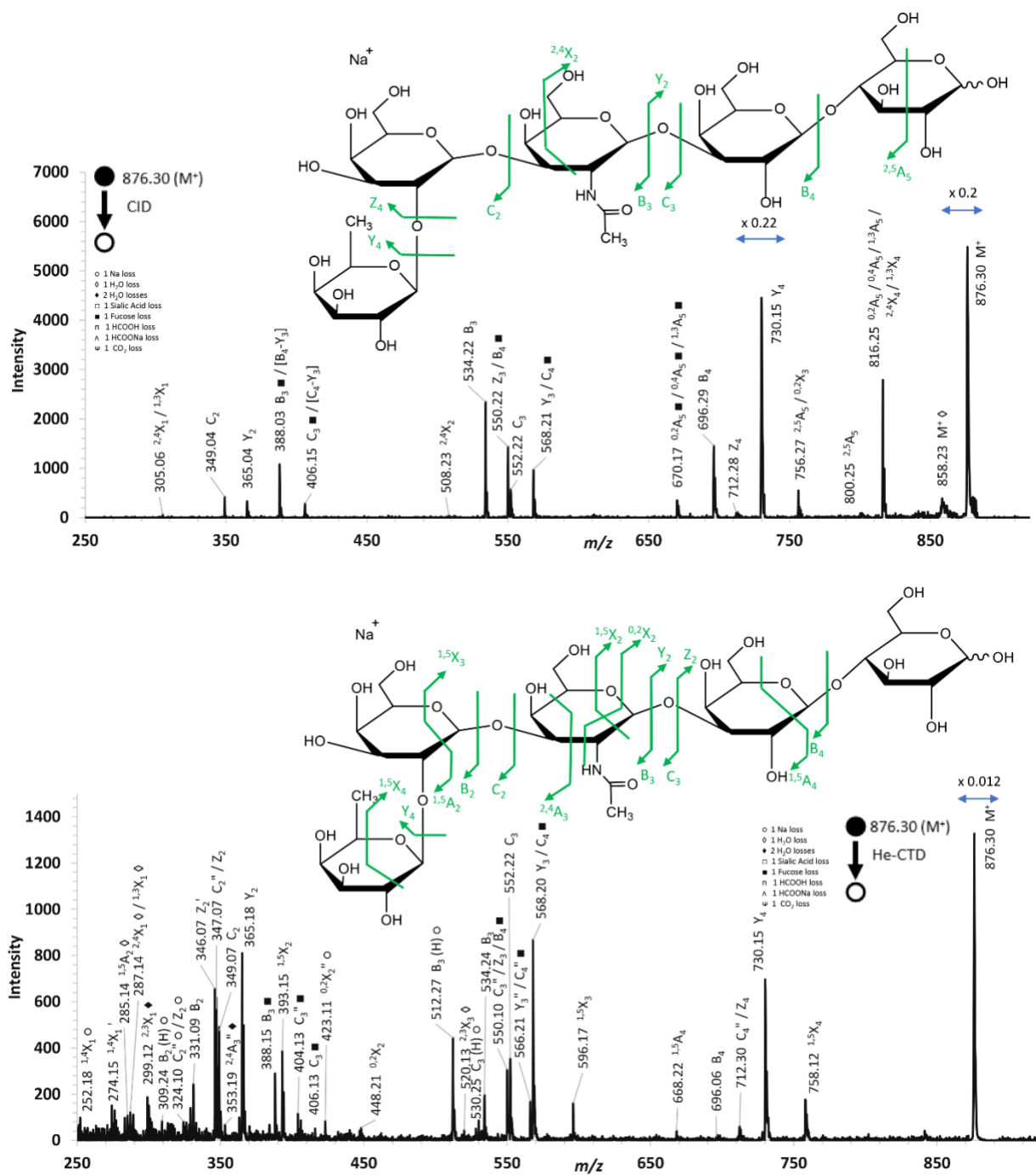


Fig. 3. HILIC-UHPLC-MS/MS spectra of LNFP I at 15.0 min collected in positive ion mode using a) CID and b) He-CTD. The insets show only the unambiguous annotated product ions.

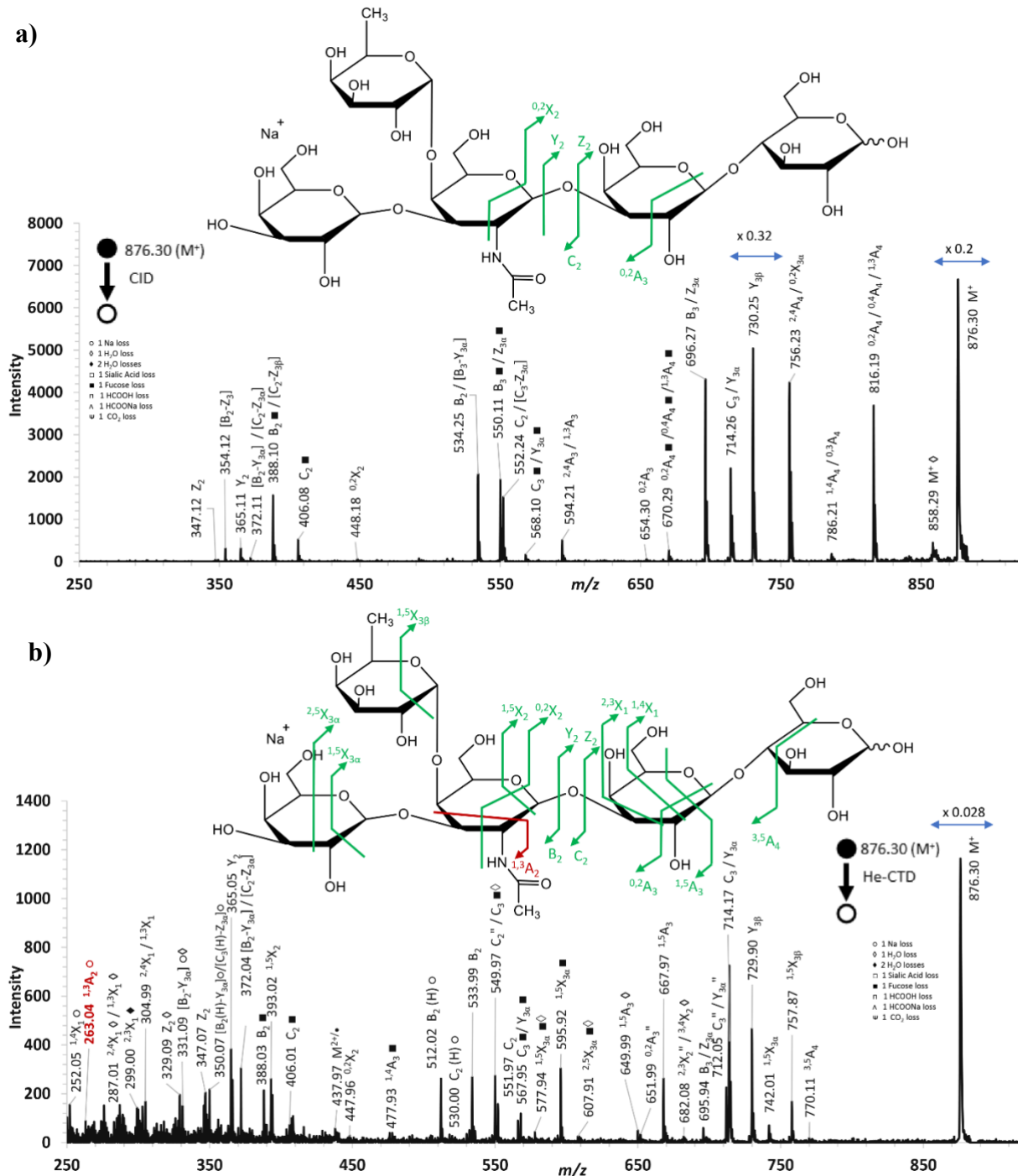


Fig. 4. HILIC-UHPLC-MS/MS spectra of LNFP II at 16.0 min collected in positive ion mode using a) CID and b) He-CTD. The insets show only the unambiguous annotated product ions

The presence of a Fuc1-Gal2-disaccharide unit at the non-reducing end in LNFP I is explained by the C₂ ion at m/z 349.1. The Y₄ and ^{1,5}X₄ fragment ions also help narrow the fucose location to the terminal position. In LNFP II, the B₂-Y_{3α}/C₂-Z_{3α} fragment at m/z 372.0 indicates the presence of a Gal-GlcNAc-Fuc unit at the non-reducing end, and the structural details obtain from ^{1,3}A₂, ^{0,2}X₂ and ^{1,5}X_{3α} fragment ions, suggest that the terminal Gal1 is attached to GlcNAc2 via a 1,3-linkage. The ^{1,3}A₂, ^{1,5}X₂ and ^{1,5}X_{3β} fragment ions suggest that the branched fucose unit is attached to GlcNAc via a 1,4- or 1,6-linkage (Adamson and Håkansson, 2007). The remaining product ions from LNFP I and II show that He-CTD is capable of preserving the labile fucose residue on a majority of the product ions and that He-CTD is able to evade the fucose migration that has plagued earlier CID studies (Wuhrer *et al.*, 2006; Aldredge *et al.*, 2013; Mank *et al.*, 2019).

The acidic HMOs—LSTa, LSTb, and LSTc—provide a challenging family of sialylated isomers to resolve. The unambiguous fragment maps and spectra for each sialylated isomer with both CID and He-CTD are shown in **Figs. 5-7** and **Figs. S1-S3**. Among these oligosaccharides, LSTa and LSTc display linkage isomerism to each other, and LSTb shows structural isomerism to LSTa and LSTc (De Leoz *et al.*, 2019). As observed in the unambiguous fragmentation maps for CID of three isomers in **Figs. 5a-7a**, glycosidic cleavages are dominant in CID and cross-ring cleavages are infrequent and/or unhelpful. CID resulted in a series of unambiguous glycosidic cleavages from both the reducing and non-reducing termini with both LSTa and LSTc.

The CID fragments B₄ and Y₄ indicate the presence of terminal sialic acid and glucose units at non-reducing and reducing ends, respectively. Fragment pairs of B₂-Y₄, B₂-B₃ and B₃-B₄ provide the Gal-GlcNAc-Gal monosaccharide sequence in LSTa and LSTc. For LSTb, CID resulted in fewer unambiguous glycosidic cleavages, which makes the monosaccharide composition determination more difficult to assign. The glycosidic fragments C₂ and Y_{3α}, (**Figs.**

S2a) indicate that the labile sialic acid unit tends to be absent with CID, and this neutral loss increases the uncertainty in the structural determination process (De Leoz *et al.*, 2019).

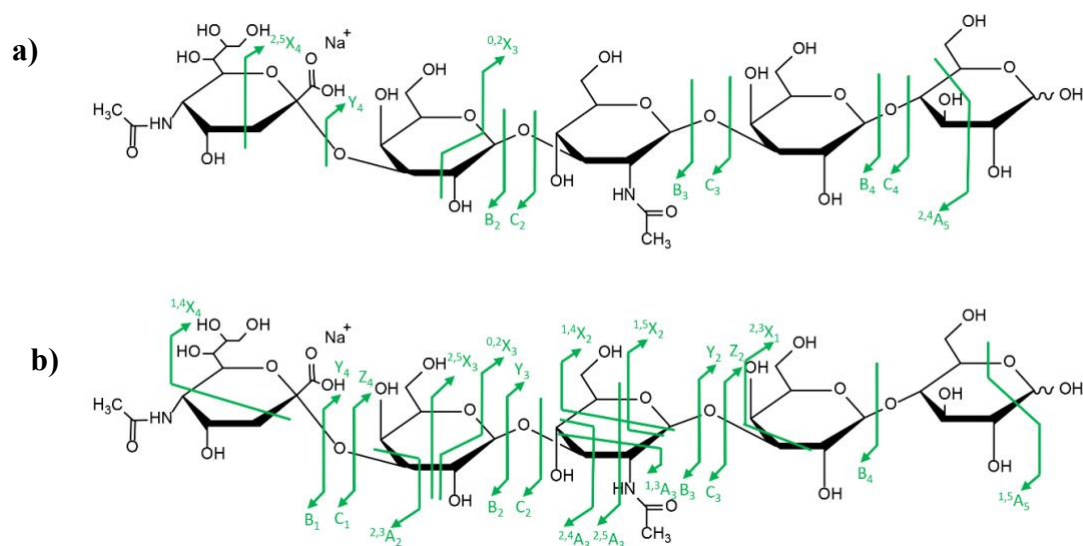


Fig. 5. HILIC-UHPLC-MS/MS product ion maps of LSTa at 16.2 min collected in positive ion mode using a) CID and b) He-CTD. The product maps show only the unambiguous annotated product ions

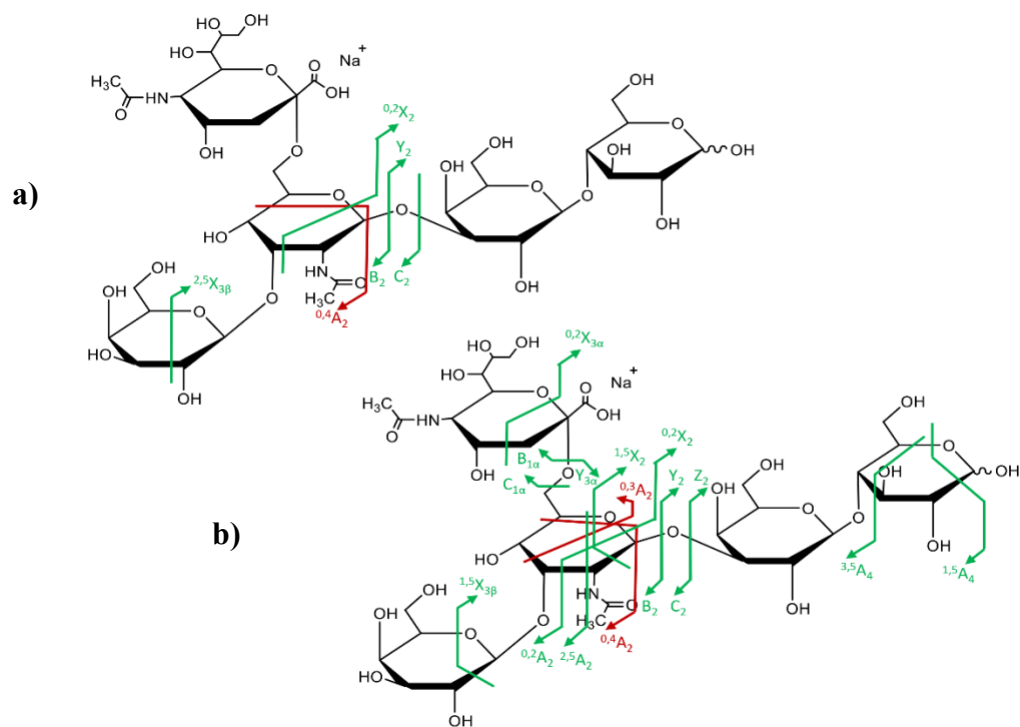


Fig. 6. HILIC-UHPLC-MS/MS product ion maps of LSTb at 16.8 min collected in positive ion mode using a) CID and b) He-CTD. The product maps show only the unambiguous annotated product ions

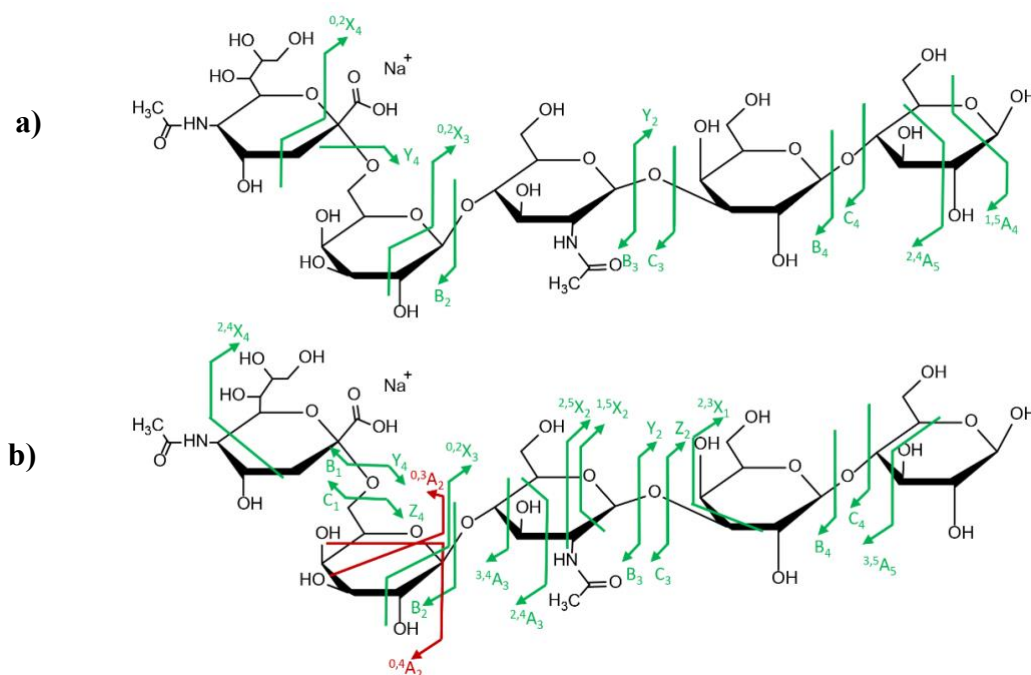


Fig. 7. HILIC-UHPLC-MS/MS product ion maps of LSTc at 17.4 min collected in positive ion mode using a) CID and b) He-CTD. The product maps show only the unambiguous annotated product ions.

In addition to the glycosidic cleavages, the CID spectra in **Figs. 5a-7a** demonstrate that there are several unambiguous cross-ring cleavages ($^{2,4}A_n$, $^{2,5}X_n$, and $^{0,2}X_n$) at the terminals of LSTa-c (Adamson and Håkansson, 2007; De Leoz *et al.*, 2019). The unambiguous $^{2,4}A_5$ and B_4 fragments, which are observed in LSTa and LSTc, provide details about the reducing end (De Leoz *et al.*, 2019). For LSTb, the $^{0,4}A_2$ fragment at m/z 374.0 is the only unambiguous cross-ring fragment that is observed with CID. The presence of the $^{0,4}A_2$ cross-ring fragment ion is advantageous because it indicates the attachment of Neu5Ac to C-6 on the GlcNAc unit (Adamson and Håkansson, 2007; Ko and Brodbelt, 2011). In short, the level of detail provided by CID for these three isomers is not sufficient to differentiate them with confidence. Whereas differentiation

can be accomplished by considering their elution times, the elution times of standards may not always be feasible when elucidating structures in natural samples.

The He-CTD MS/MS fragmentation maps for LSTa, LSTb, and LSTc are shown in **Figs. 5b-7b**, respectively. As shown in the fragmentation maps, He-CTD produces more glycosidic fragments for all three HMOs, and the unambiguous glycosidic cleavages help to understand the monosaccharide composition of the oligosaccharides in a similar fashion in CID. In addition to the glycosidic fragments, He-CTD provides many cross-ring fragments, including, $^{0,2}A_n$, $^{0,3}A_n$, $^{0,4}A_n$, $^{1,3}A_n$, $^{1,5}A_n$, $^{2,4}A_n$, $^{2,5}A_n$, $^{3,5}A_n$, $^{0,2}X_3$, $^{1,4}X_n$, $^{1,5}X_n$ and $^{2,5}X_n$. These structurally informative cross-ring fragments are beneficial in determining the linkage and branching patterns of the three isomers. As an example, He-CTD produced $^{2,4}A_3$, $^{1,3}A_3$, and $^{0,2}X_3$ product ions, and these ions are useful in identifying the 1,3-linkage between Gal2 and GlcNAc3. Similarly, the fragments $^{1,5}X_2$ and $^{2,5}A_3$ help localize the -NHCOCH₃ active group on GlcNAc in LSTa to the C-2 position. Such details are not possible with CID because of the limited cross-ring cleavages (De Leoz *et al.*, 2019). The linkage-related information has been previously observed with UHPLC-He-CTD-MS studies performed with highly methylated pectin and carrageenan oligosaccharides, which helps to identify the 1,4-linkage patterns between monosaccharide units and localize the 6-*O*- methyl-esterification and 4-*O*- sulfation sites on the oligosaccharides in the complex mixtures (Mendis *et al.*, 2021a, 2021b).

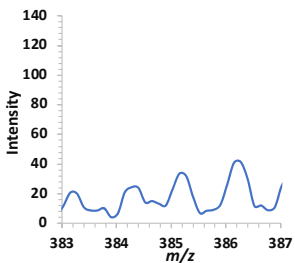
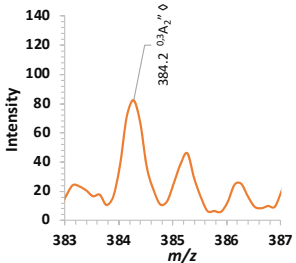
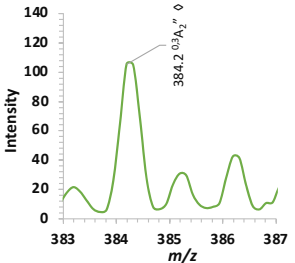
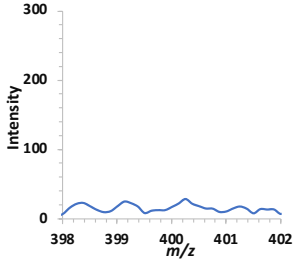
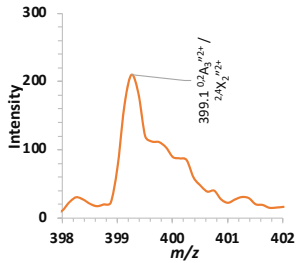
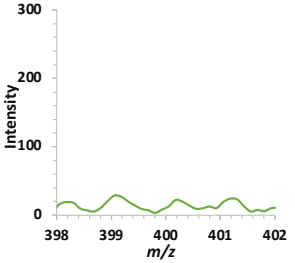
The He-CTD product ion spectra for acidic HMOs also provide valuable information on the sialic acid linkage position, which is advantageous for isomer differentiation. The He-CTD spectrum for LSTa (**Fig. S1**) indicates the presence of possible $^{1,3}A_2/^{2,4}A_2$ cross-ring ions at m/z 374.2, which have previously been observed with EDD for doubly deprotonated LSTa (Adamson and Håkansson, 2007). The $^{1,3}A_2/^{2,4}A_2$ cross-ring ions suggest that Neu5Ac is attached to Gal2 at

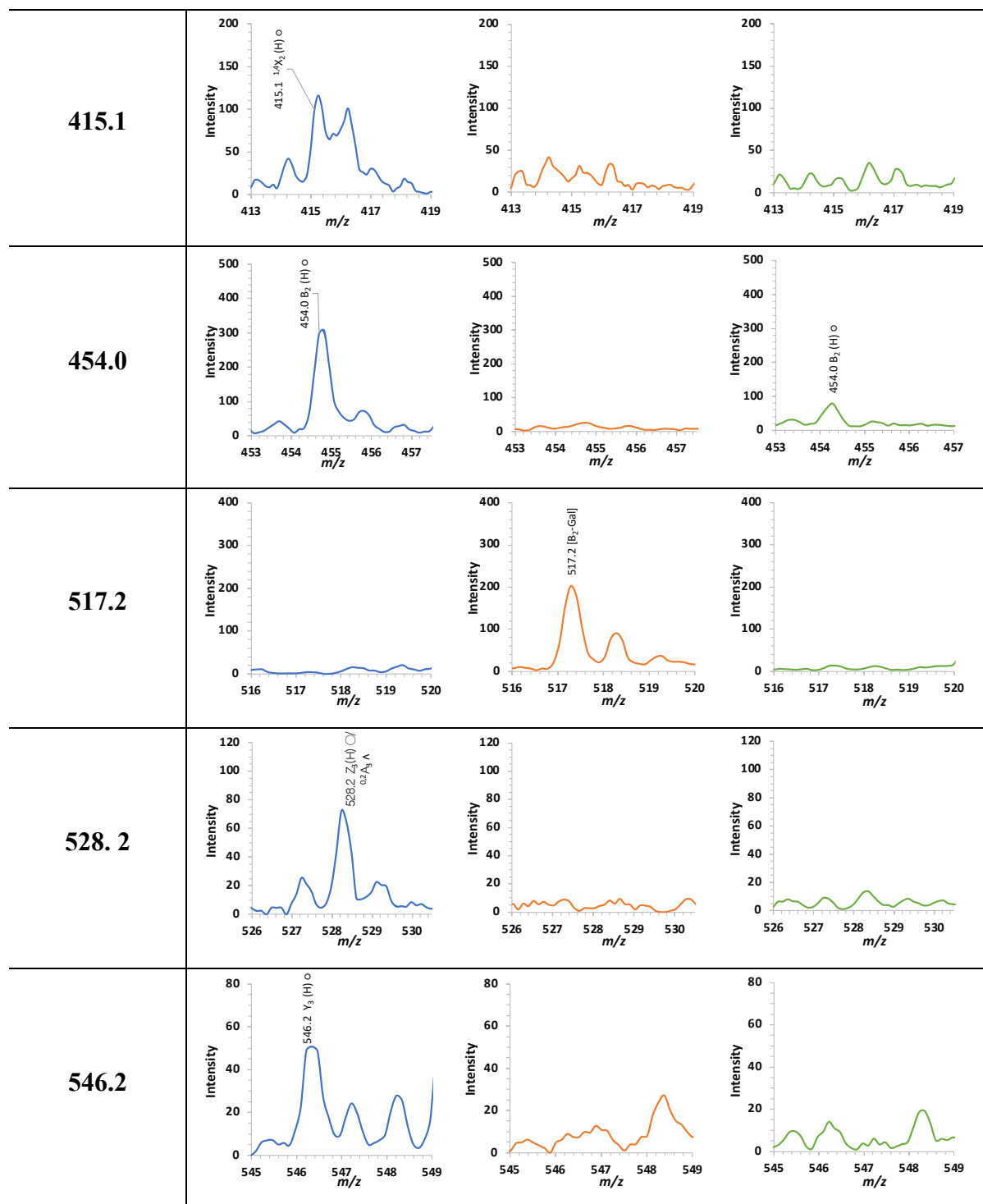
the C-3 or C-4 positions. In the He-CTD product ion spectrum for LSTb (**Fig. S2**), the $^{0,3}A_2$ and $^{0,4}A_2$ fragments observed at m/z 374.2 and m/z 384.2, respectively, have been reported with ETD of Mg^{2+} -adducted LSTb, with UVPD of deprotonated LSTb, and with EDD of doubly deprotonated LSTb (Adamson and Håkansson, 2007; Han and Costello, 2011; Ko and Brodbelt, 2011). These fragments provide details of the C-6 linked Neu5Ac unit. For LSTb, He-CTD provides sufficient detail to localize the Neu5Ac attachment, and the presence of $^{1,5}X_{3\beta}$, $^{0,2}A_2$, and $^{0,3}A_2$ fragments confirm the presence of 1,3-linkage among the Gal-GlcNAc2 unit on the reducing terminus. When comparing the He-CTD spectra of LSTc with LSTb, LSTc follows a similar fragmentation pattern at the Neu5Ac location (**Fig. S3**). The fragments $^{0,3}A_2$ and $^{0,4}A_2$ are also observed with LSTc, which again confirm the C-6 position of the Neu5Ac residue to the Gal2 unit.

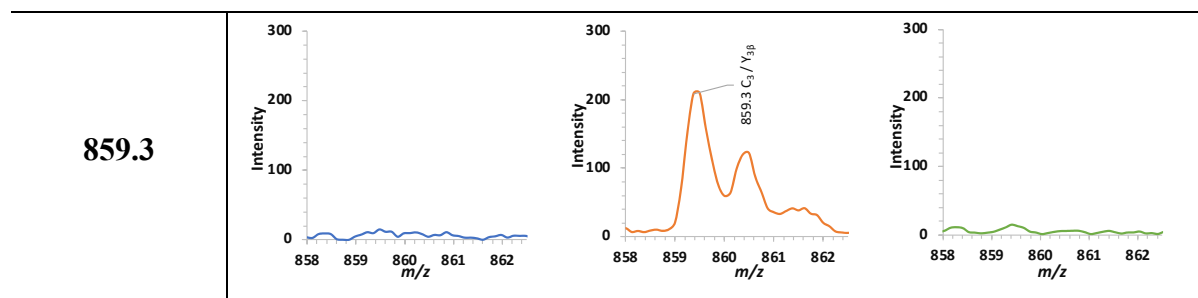
Some of these linkage-related fragments show product ions that are isobaric with different product ions from other isomers. Therefore, to be confident with the structural assignments, additional scrutiny is required. **Table I** shows a comparison of additional diagnostic peaks that correspond to each isomer. The peaks observed at m/z 415.1 [$^{1,4}X_2(H)-Na$], m/z 528.2 [$Z_2(H)-Na$ / $^{0,2}A_3-HCOONa$] and m/z 546.2 [$Y_3(H)-Na$] are present above the minimal S/N level only in LSTa. Among those unique product ions, $Y_3(H)-Na$ and $^{1,4}X_2(H)-Na$ help to differentiate the unbranched structure of LSTa relative to LSTb. For LSTb, the peaks at m/z 517.2 [B_2-Gal] inform the presence of GlcNAc and Neu5Ac units in the non-reducing end, and the peak at m/z 292.1 (B_{1a}) (**Fig. S2b**) locates the Neu5Ac unit to the terminal position. Peaks observed at m/z 859.3 and m/z 399.1 are dominant only in LSTb, but the further investigation needs to be performed with heavy ^{18}O labeling to break the ambiguity of the peak assignments [39]. Labeling with heavy ^{18}O has been used successfully for He-CTD experiments for oligosaccharides in the past to differentiate ions originated from reducing and non-reducing termini (Ropartz *et al.*, 2016). The

dominant peak at m/z 384.2 [$^{0,3}A_2''$ -H₂O] in LSTb and LSTc gives the information that LSTb and LSTc have a Neu5Ac attached at either C-4 or C-6 position, and this fragment is absent in LSTa. Peaks observed at m/z 454.0 [$B_2(H)$ -Na] in both LSTa and LSTc, indicate that LSTa and LSTc have unbranched non-reducing terminals compared to LSTb. With He-CTD, the Gal and GlcNAc residues at the non-reducing end in each isomer undergo multiple cross-ring cleavages. These extensive fragmentation patterns may be related to the elevated electron density in the acidic Neu5Ac residue and the increased propensity for ionization during cation-cation activation (Bari *et al.*, 2010, 2011; Sasiene *et al.*, 2021a). Enhanced activation on the acidic, non-reducing terminus might also be related to the propensity for the adducting Na⁺ ion to bind with Neu5Ac residue (Han and Costello, 2011).

Table I. Comparison of He-CTD product ions for the isomers LSTa, LSTb and LSTc.

m/z	LSTa	LSTb	LSTc
384.2			
399.1			





CID and He-CTD of di-sialylated branched oligosaccharide (DSLNT) are investigated presented in **Fig. S4**. In CID fewer unambiguous glycosidic and cross-ring cleavages were observed, as expected. Similar to the mono-sialylated HMOs, the majority of the cross-ring cleavages—E.g, ${}^{0,2}\text{X}_{3\alpha}$ and ${}^{2,5}\text{X}_{3\alpha}$ —occurred only on Gal2 units next to the Neu5Ac group. The presence of a $\text{B}_2/\text{Y}_{3\alpha}$ fragment provides informs us that one of the Neu5Ac groups is located at the non-reducing end. The presence of a B_3 ion and $\text{B}_2/\text{Y}_{3\alpha}$ fragment helps to locate the second Neu5Ac group on the GlcNAc3 unit. The lack of unambiguous cross-ring and glycosidic cleavages at the terminal sites limits the structural information on the reducing and non-reducing ends.

In contrast to the CID results summarized above, He-CTD shows unambiguous glycosidic cleavages throughout the main branch, which helps to identify the monosaccharide sequence of the DSLNT. Unlike CID, He-CTD produces a majority of its unambiguous cross-ring cleavages on the GlcNAc at the center region. The unambiguous ${}^{2,4}\text{A}_3$ and ${}^{1,5}\text{X}_2$ cross-ring fragments enable the identification of the possible C-3 or C-6 attachment of the attached Neu5Ac group on the GlcNAc unit, which is a major advantage for He-CTD compared to CID. ${}^{2,4}\text{A}$ and ${}^{1,5}\text{X}$ cross-ring cleavages have been previously reported with He-CTD for branched xyloglucans and, with the help of additional cross-ring fragments such as ${}^{2,5}\text{A}$ and/or ${}^{2,5}\text{X}$, 1,4- and 1,6-linked branching patterns have been identified (Sasiene *et al.*, 2021c).

These results demonstrate the ability to perform HILIC-UHPLC-He-CTD-MS for effective analysis of a complex mixture of human milk oligosaccharides (HMOs). He-CTD is capable of

providing important cross-ring cleavages that help to differentiate linkage isomers and structural isomers in both neutral and acidic milk oligosaccharides. The beneficial cross-ring fragments observed with He-CTD show commonality with other radical and high energy activation techniques like ETD, EDD and UVPD. Unlike some of the previously cited work, all the product ion spectra collected with He-CTD in this study are based on underivatized HMOs. He-CTD has a modest advantage of being applicable to the analysis of native HMOs. In addition, He-CTD is able to preserve the majority of the Neu5Ac and fucose residues. These findings give the impression that He-CTD is a valuable tool that can be used in the native structure analysis of oligosaccharides.

Materials and methods

Reagents and oligosaccharides

Ammonium formate was purchased from Oakwood Chemical (Estill, SC, USA). Formic acid (FA) and LC/MS grade acetonitrile (ACN) were purchased from ThermoFisher Scientific (Fair Lawn, NJ, USA). Ultra-pure 18 M Ω water was obtained from a Milli-Q apparatus from Millipore (Burlington, MA, USA). Individual samples of sialylacto-*N*-tetraose a (LSTa), sialylacto-*N*-tetraose b (LSTb), sialylacto-*N*-tetraose c (LSTc), disialyllacto-*N*-tetraose (DSLNT), lacto-*N*-fucopentaose I (LNFP I), and lacto-*N*-fucopentaose II (LNFP II) were purchased from Biosynth Carbosynth (San Diego, CA, USA). The complex mixture of HMOs was prepared from equimolar solutions of individual HMO samples.

Hydrophilic Interaction Liquid Chromatography (HILIC) separation

Chromatographic separation was performed on a Shimadzu Nexera X2 UHPLC system (Kyoto, Japan) using an Accucore 150-Amide-HILIC column (2.6 μ m, 2.1 mm x 100 mm) from

ThermoFisher Scientific. The composition of the two mobile phases was: A) 99.9% ACN with 0.1% (v/v) formic acid, and B) 10 mmol/L ammonium formate with 0.1% (v/v) formic acid. A binary gradient was used for the separation with a flow rate of 0.4 mL/min and column oven temperature of 50 °C. The gradient was changed in a linear mode from 15% to 35% of mobile phase B from 0-20 min, then increased to 75% from 20-25 min and a column re-equilibration at 85% of mobile phase A for 5 min.

On-line He-CTD

On-line He-CTD was performed on a modified Bruker amaZon ETD quadrupole ion trap from Bruker Daltonics (Bremen, Germany), as described previously (Mendis *et al.*, 2021a, 2021b). Briefly, a saddle field fast ion source from VSW/Atomtech, (Macclesfield, UK) was fixed above a 3-mm hole in the ring electrode of the 3D ion trap and connected to a variable leak valve (Model 203, Granville-Phillips) to control the flow of helium gas through the source. The flow of helium was indirectly measured from the ion gauge in the main vacuum chamber and kept at about $\sim 1.2 \times 10^{-5}$ mbar. The ion source was connected to an economical home-built system that employs a 10 kV Ultravolt HVA series high voltage power supply (Advanced Energy, Denver, CO, USA). The +5.5 kV high voltage from the Ultravolt UHA was pulsed from ground to high with rise time as fast as 150 ns by using a Behlke 101-03 switch (Behlke, Billerica, MA, USA). A TTL signal was taken from the MS² event of the Bruker amaZon from pin 28 of the auxiliary control port and sent to an Agilent 33250A arbitrary function generator (AFG, Keysight Technologies, Santa Rosa, CA, USA) to provide an independently variable delay and pulse width. A DS1054 digital oscilloscope (Rigol, Beaverton, OR USA) compared the trigger waveform from the AFG with the scan function of the Bruker amaZon to ensure that the high voltage pulses coincided with the desired storage period of the scan function. The saddle field fast ion source has an 85% conversion efficiency, so

the 5.5 kV pulse provided helium cations with a kinetic energy of approximately 4.7 keV. The saddle field fast ion source pulse was matched with the fragmentation portion of the scan function of the instrument with the CID amplitude set to 0 V.

UHPLC-HILIC-He-CTD-MS

The effluent from the UHPLC was connected to the standard Bruker Apollo electrospray ionization source (Billerica, MA). Capillary voltage and end cap voltage were set to +4500 V and -700 V. Nebulizer pressure was set to 30 psi, the dry gas flow rate was set to 10 L/min and the dry temperature was at 340 °C. Experiments were conducted in the data-dependent acquisition (DDA) mode from m/z 250–1500. The most abundant charge state of each analyte was chosen as the precursor ion with a window of 8 Da. The CID collision energy was set at 0 V when He-CTD was activated. The ion gun was pulsed on for 50 ms, and the MS² event, which includes the He-CTD activation time followed by a storage time, was set to 200 ms and a low mass cut-off of 27%. As a comparison, data of the same HMO mixture using the same UHPLC conditions was also collected using traditional CID-MS/MS on the same instrument with CID collision energy at 0.40 V and 50 ms activation time. For CID-MS/MS, smart fragmentation was on and set to 80-120%.

Data Analysis

Raw data were transformed in Bruker Compass DataAnalysis 4.0 SP4 software and further processed using Microsoft Excel (Microsoft, Redmond, WA, USA). The peaks in the deconvoluted spectra were chosen manually based on their signal-to-noise ratio, isotope envelope distribution and fragmentation patterns. Product ion assignments for CID and He-CTD were achieved using an in-house analysis of an HMO database and Glycoworkbench software. ChemDraw 19.1 (PerkinElmer, Waltham, MA, USA) was used for chemical structure illustrations.

Conclusions

The utility of UHPLC-He-CTD-MS has been demonstrated for the characterization of a complex mixture of neutral and acidic human milk oligosaccharides (HMOs). Analytes in the mixture consisted of DP5 and DP6 linear and branched structures with labile Neu5Ac (sialic acid) and Fucose residues. Our results demonstrate that He-CTD can perform effectively with the fast timescales and low sample loads required for HILIC-UHPLC separation and provides spectra with adequate signal-to-noise to determine structural characteristics for HMOs. He-CTD product ion spectra contained a series of cross-ring and glycosidic cleavages, which allow unambiguous identification of structural and linkage isomers in the HMOs mixture. The presence of $^{0,3}A_2$ and $^{0,4}A_2$ fragments in He-CTD are also observed with ETD, EDD and UVPD and enable differentiation of the α -2,3- vs α -2,6-linked sialic acid (Neu5Ac) residues present among sialyllacto-*N*-tetraoses. The use of heavy ^{18}O labeling could resolve the identification of some ambiguous peak assignments caused by the modest mass resolution of the 3D ion trap in this study. Finally, the results indicate that UHPLC-He-CTD-MS is a potential tool for the online characterization of biologically important oligosaccharides with diverse chemical properties. Expectations are that He-CTD will work equally well for complex mixtures of *N*- and *O*-linked glycans.

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Conflict of Interest

None declared

Abbreviations

2-AB, 2-aminobenzamide; ACN, Acetonitrile; CE, Capillary electrophoresis; CID, Collision induced dissociation; He-CTD, Helium charge transfer dissociation; He-CTD-MS, He-CTD mass spectrometry; Da, Dalton; DDA, Data-dependent acquisition; DP, Degree of polymerization; DSLNT, Disialyllacto-*N*-tetraose; ECD, Electron capture dissociation; EDD, Electron detachment dissociation; EED, Electron excitation dissociation; ESI, Electrospray ionization; ETD, Electron transfer dissociation; FTICR, Fourier-transform ion cyclotron resonance; Fuc, L fucose; Gal, D-galactose; Glc, D-glucose; GlcNAc, *N*-acetylglucosamine; HCD, Higher-energy collisional dissociation; HILIC, Hydrophilic interaction liquid chromatography; HMOs, Human milk oligosaccharides; HPAEC, High-pH anion-exchange chromatography; ICR, Ion cyclotron resonance; IRMPD, Infrared multiphoton dissociation; LC, Liquid chromatography; LMCO, Low mass cutoff; LNFP, Lacto-*N*-fucopentaose; LST, Sialyllacto-*N*-tetraose; *m/z*, mass to charge ratio; MALDI, Matrix-assisted laser desorption/ionization; MS/MS, Tandem mass spectrometry; MSⁿ, Multi-stage fragmentation; NETD, Negative electron transfer dissociation; Neu5Ac, Sialic acid; NMR, Nuclear magnetic resonance; PGC, Porous graphitized carbon; PSD, Post-source decay; QIT, Quadrupole ion traps; RP-LC, Reversed-phase LC; SEC, Size-exclusion chromatography;

TOF, Time-of-flight; UHPLC, Ultrahigh performance liquid chromatography; UVPD, Ultraviolet photodissociation;

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