

## Hydrophobe Containing Polypeptoids Complex with Lipids and Induce Fusogenesis of Lipid Vesicles

*Marzhana Omarova<sup>1</sup>, Yueheng Zhang<sup>1</sup>, Igor Kevin Mkam Tsengam<sup>2</sup>, Jibao He<sup>3</sup>, Tianyi Yu<sup>2</sup>, Donghui Zhang<sup>2\*</sup> and Vijay John<sup>1\*</sup>*

1. Department of Chemical and Biomolecular Engineering  
Tulane University, New Orleans, LA 70118, United States
2. Department of Chemistry  
Louisiana State University, Baton Rouge, LA 70803, United States
3. Coordinated Instrumentation Facility  
Tulane University, New Orleans, LA 70118, United States

Submitted to the *Journal of Physical Chemistry B* for the Lawrence Pratt Festschrift issue.

\*Corresponding authors. Vijay T. John, phone: 504-865-5883 e-mail: [vj@tulane.edu](mailto:vj@tulane.edu). Donghui Zhang, phone: 225-578-4893 email: [dhzhang@lsu.edu](mailto:dhzhang@lsu.edu)

## ABSTRACT

The hydrophobic effect of alkyl group insertion into phospholipid bilayers is exploited in modifying and modulating vesicle structure. We show that amphiphilic polypeptoids (peptide mimics) with N-decyl sidechains, a.k.a. hydrophobically modified polypeptoids (HMPs), can insert into the membrane bilayer of phospholipid-based vesicles. Such insertion leads to disruption of the liposomes and the formation of hydrophobically modified polypeptoid (HMP)-lipid complexes that are colloidally stable in aqueous solution. Interestingly, when these complexes are added to fresh liposomes, remnant dynamically free hydrophobes (the N-decyl groups) bridge liposomes and fuse them. The fusion leads to the engulfing of liposomes and the formation of multilayered vesicles. The morphology of the liposome system can be changed from arresting fusion and forming clustered vesicles to the continued formation of multilayered liposomes simply by controlling the amount of the HMP-lipid complex added. The entire procedure occurs in aqueous systems without the addition of any other solvents. There are several implications to these observations including the biological relevance of mimicking fusogenic proteins such as the SNARE proteins, and the development of new drug delivery technologies to impact delivery to cell organelles. The biocompatible nature of polypeptoids also leads to further useful aspects of the study.

## INTRODUCTION

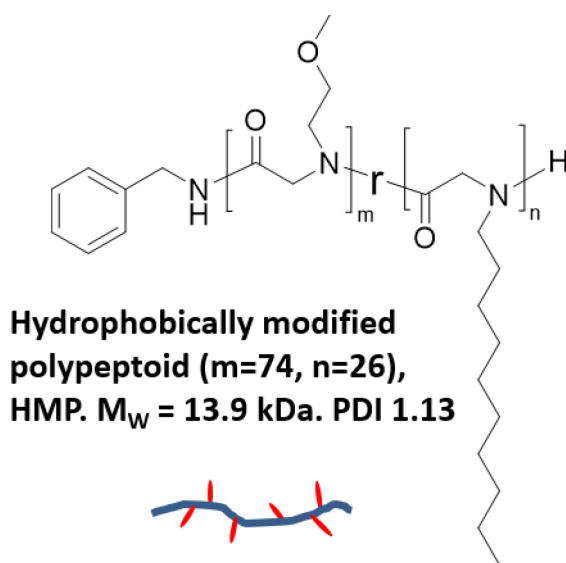
Synthetic lipid vesicles or liposomes are a useful and convenient platform for research on model cell membrane ever since their first discovery by Bangham et al<sup>1</sup> with the self-assembly of such closed systems having implications to understanding the origin of life.<sup>2</sup> Liposomes offer a number of advantages in drug delivery applications, such as a simple and scalable method of preparation and designability suitable for encapsulation of small molecules<sup>3</sup> and nucleic acids.<sup>4</sup> A number of drug formulations uses liposomes or lipid based nanoparticles in the clinical trials currently<sup>4-5</sup> with a few approved for medical use in drug delivery.<sup>6</sup>

The mechanistic understanding of the formation and transformation of single bilayer-based vesicles to multilamellar vesicles is a continuing area of active research with implications to the fundamental knowledge of biological systems and to applied aspects of drug delivery. Multilamellar lipid vesicles formed through high energy shear are well documented<sup>7</sup> but the formation of such multilamellar vesicles is imprecise and it is likely that fragile biomolecules in the vesicles are degraded through such shear effects. Recent interesting work describes the use of dendrimersomes or multilamellar vesicles formed using amphiphilic Janus dendrimers by injecting THF solutions of the dendrimer into water or a buffer.<sup>8</sup> These are multi-step synthetic processes involving the use of organic solvents and their biocompatibility remain to be determined. Other recent examples include the use of rod-like oligofluorenes in a acetonitrile-water mixed solvent.<sup>9</sup>

Our work is based on a specific manifestation of the hydrophobic effect where alkyl hydrophobes on the backbone of a water-soluble biopolymer insert into membrane lipid bilayers. The concepts of such insertion are well established and is the reason why detergents lyse cell

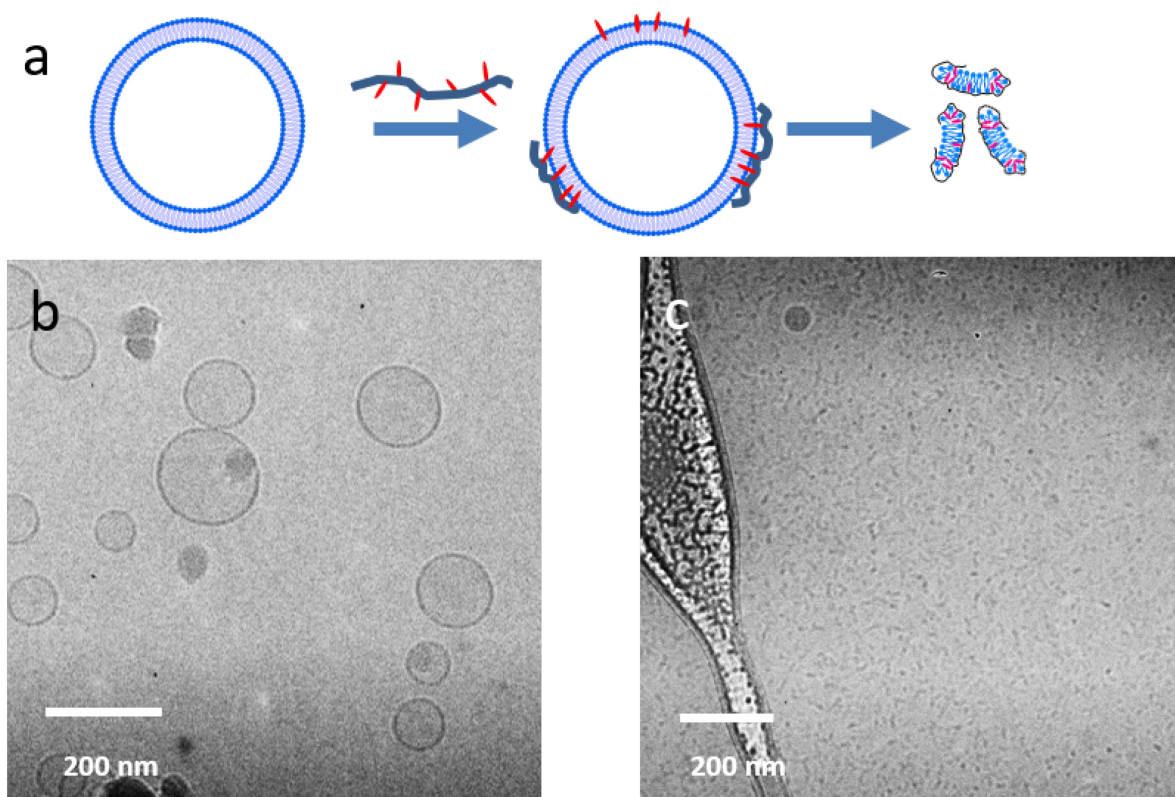
membranes. In the specific system studied here, the biopolymer is a polypeptoid. Peptoids are a class of peptide mimics where the substituents are on the nitrogen rather than the carbon atoms.<sup>10</sup> These polymers are therefore structurally similar to peptides<sup>11</sup> and without the hydrogen bonding and stereogenic centers along the backbone have a more flexible conformation leading to resistance against degradation by proteases.<sup>12</sup> The biocompatibility of peptoids renders them useful in biomedical applications.<sup>13</sup>

We specifically focus on the use of a hydrophobically modified polypeptoid (HMP) with approximately 100 monomer units where the hydrophobic modification involves the random replacement of about 25% of the *N*-2-methoxyethyl group with *n*-decyl groups (C10) which form the hydrophobes attached to the backbone of the water-soluble polymer, thus conferring a degree of amphiphilicity to the polymer. The detailed synthesis of the polymer is reported in our earlier publications (and additionally summarized in the materials and methods section).<sup>14</sup> Figure 1 illustrates the structure of the polymer.



**Figure 1.** Molecular weight and structure of hydrophobically modified polypeptoid.

In our earlier work,<sup>14-15</sup> we found that adding small amounts of HMP to phosphatidylcholine-based liposomes led to disruption of the liposomes with the observation of the formation of 2 and 3 layered liposomes. With sufficient addition of HMP at a composition of 0.25 wt% lipid to 0.5 wt% HMP, all liposomes become fragmented with the formation of HMP-lipid complexes. Figure 2 illustrates the concept and shows a cryogenic transmission electron micrograph (cryo-TEM) of such complexes that are formed from liposomes with a diameter on the order of 100 nm. Figure 2a is a schematic of the mechanism of hydrophobe insertion into the liposome and the resulting disruption into HMP-lipid complexes with an approximate 9:1 lipid:HMP molar ratio. We posit that the localized insertion of hydrophobes damages the membrane



**Figure 2.** (a) Schematic to describe liposome disruption by HMP (b) typical cryo-TEM of unilamellar liposomes used in this study (c) cryo-TEM of the HMP-lipid complex formed by the addition of 0.5 wt% HMP to liposomes containing 0.25 wt% lipid.

integrity leading to the disruption of the liposome. These nanoscale HMP-lipid complexes were also characterized by small-angle neutron scattering (SANS) in addition to cryo-TEM to indicate elongated small wormlike entities with a 5.1 nm radius of gyration, 2 nm radius and a 38 nm contour length as calculated from the flexible cylinder model fitting.<sup>14</sup> We have also shown that such HMP-lipid complexes are able to sustain hydrophobic drug moieties and are easily able to enter mammalian cells, leading to potential applications in drug delivery systems.<sup>15</sup>

The focus of this paper is on a second manifestation of the concept of hydrophobe insertion into lipid bilayers that is equally interesting with the potential to further fundamental understanding of the hydrophobic effect and lead to new applications. The concept relates to the addition of the HMP-lipid complex to new liposomes. In this instance, all hydrophobes of the HMP do not have the ability to insert into the liposome bilayer as they interact with lipids in the complex. We have found that these HMP-lipid complexes are therefore unable to disrupt the liposomes. Rather they remodel liposomes to build multi-layered vesicles. We also show that the formation of these multi-layered liposomes is the result of the fusogenic properties of the HMP-lipid complexes, which can also arrest intermediate structures, opening up possibilities to further modulating vesicle structure. The details of these finding are described in the following sections of the paper.

## EXPERIMENTAL METHODS

### *Materials*

L- $\alpha$ -phosphatidylcholine (PC, >95%, from soy) was purchased from Avanti Polar Lipids. Fluorescein isothiocyanate–dextran (FITC–dextran, Mw 3–5 kDa) was purchased from Sigma Aldrich. All other chemicals and solvents were purchased from Sigma Aldrich and used as received unless otherwise noted. The solvents used for polymerization were further purified by using alumina columns under argon protection. CD<sub>2</sub>Cl<sub>2</sub> and CDCl<sub>3</sub> were purchased from Cambridge Isotope laboratories. <sup>1</sup>H NMR was collected by Bruker AV-400 III spectrometer at 298K and analyzed using Topspin software. Chemical shifts ( $\delta$ ) given in parts per million (ppm) were referenced to protio impurities.

### *Hydrophobically modified polypeptoid (HMP) synthesis*

*N*-decyl glycine derived *N*-carboxyanhydride (De-NCA) and *N*-methoxyethyl glycine derived *N*-carboxyanhydride (MeOEt-NCA) monomers were synthesized by published procedure.<sup>16</sup> The HMPs were synthesized through primary amine-initiated ring-opening polymerization of the corresponding R-NCA monomers as reported previously.<sup>14</sup> Copolymerization of *N*-methoxyethyl NCA and *N*-decyl NCA yields a random copolymer of *N*-methoxyethyl glycine units and *N*-decyl glycine units. In a typical synthesis, in glove box, stock solutions of MeOEt-NCA (1.3 mL, 0.52 mmol, 0.4 M) and De-NCA (433  $\mu$ L, 0.17 mmol, 0.4 M) in THF were pre-mixed before the addition of benzyl amine stock solution (74.8  $\mu$ L, 6.9  $\mu$ mol, 92.7 mM) in THF. The mixture was stirred at 50 °C under nitrogen atmosphere for 72 h to reach complete conversion. The polymerization conversion was tracked by monitoring the disappearance of -C=O peak at 1780 cm<sup>-1</sup> and 1740 cm<sup>-1</sup> in the reaction aliquots taken over time using FT-IR spectroscopy. The volatiles were removed under vacuum using Schlenk line. The crude polymer was further purified by re-

dissolved in DCM and precipitated with ample hexanes twice to obtain the final product as a white solid (61.6 mg, 65.6% yield). Polymer composition was determined by end group analysis using  $^1\text{H}$ -NMR and the polymer polydispersity index (PDI) was obtained using size-exclusion chromatography (SEC).

#### *Size-exclusion chromatography (SEC)*

SEC experiments were performed in DMF with 0.1 M LiBr at 25 °C with a flow rate of 0.5 ml/min. 3.0 mg HMP polymer was dissolved in DMF solution (0.6 ml) containing LiBr (0.1M) and left to stand overnight. The polymer solutions were filtered with 0.45  $\mu\text{m}$  PTFE filters before injecting into the SEC system. SEC analysis of the hydrophobically modified polypeptoids was performed using an Agilent 1200 system equipped with three Phenomenex 5  $\mu\text{m}$ , 300  $\times$  7.8 mm columns, a Wyatt DAWN EOS multiangle light scattering (MALS) detector (GaAs 30mW laser at  $\lambda = 690\text{ nm}$ ) and Wyatt OptilabrEX differential refractive index (DRI) detector. The data analysis was performed using Wyatt Astra V 5.3 software. The PDI were obtained using polystyrene standards.

#### *Liposome preparation*

The liposomes were prepared by the thin-film hydration technique where the lipids are first dissolved in an organic solvent and then evaporated to form a lipid thin film. Typically, 100 mg PC lipid was dissolved in 15 mL chloroform and methanol mixture (2/1, v/v) in a round bottom flask. The solvent was then evaporated on a rotavapor (Buchi R-205) at room temperature at 100 mbar for 3 hours to form a thin lipid film. The film was further treated in vacuum at 6 mbar for 30 min to remove residual solvent. The formed thin lipid film was then hydrated using DI water at 50 °C which yielded a suspension of large lipid vesicles. FITC-dextran loaded vesicles were prepared in the similar way with the exception of using FITC-dextran solution in DI water for the hydration

step. The lipid film was hydrated by using 1 mg/mL of FITC–dextran solution at 50 °C for 30 min. The vesicle suspension was extruded 21 times through polycarbonate membrane with a pore size of 100 nm to downsize the unextruded vesicles into small unilamellar vesicles with an average diameter of 100 nm.

#### *Cryo-SEM*

A Hitachi S-4800 field emission Scanning Electron Microscope with the operating voltage of 3 kV was used to obtain cryogenic SEM images of emulsions and bacterial biofilm. Samples were loaded into rivets mounted onto the cryo-SEM sample holder. The samples were then plunged into slushed liquid nitrogen to freeze the sample. This was followed by fracturing at –130 °C using a flat-edge cold knife and sublimation of the solvent at –95 °C for 15 min to etch the sample. The temperature was lowered back to -130°C and the sample was then sputtered with a gold–palladium composite at 10 mA for 132 s before imaging.

#### *Cryo-TEM*

The morphology of the complexes was characterized by FEI Tecnai G2 F30 twin transmission electron microscope operated at 300 kV equipped with SDD EDS for elemental mapping. Cryo-TEM imaging was done on an FEI G2 F30 Tecnai TEM operated at 150 kV. To prepare the sample, a 200-mesh lacey carbon grid (Electron Microscopy Sciences) was picked up with tweezers and mounted on the plunging station of an FEI Vitrobot. Four microliters of the solution were applied to the grid. The excess liquid was blotted by filter paper attached to arms of the Vitrobot for 2 s to form a thin film. The sample was then vitrified by plunging into liquid ethane. The vitrified sample was finally transferred onto a single-tilt cryo specimen holder for imaging.

## SAXS

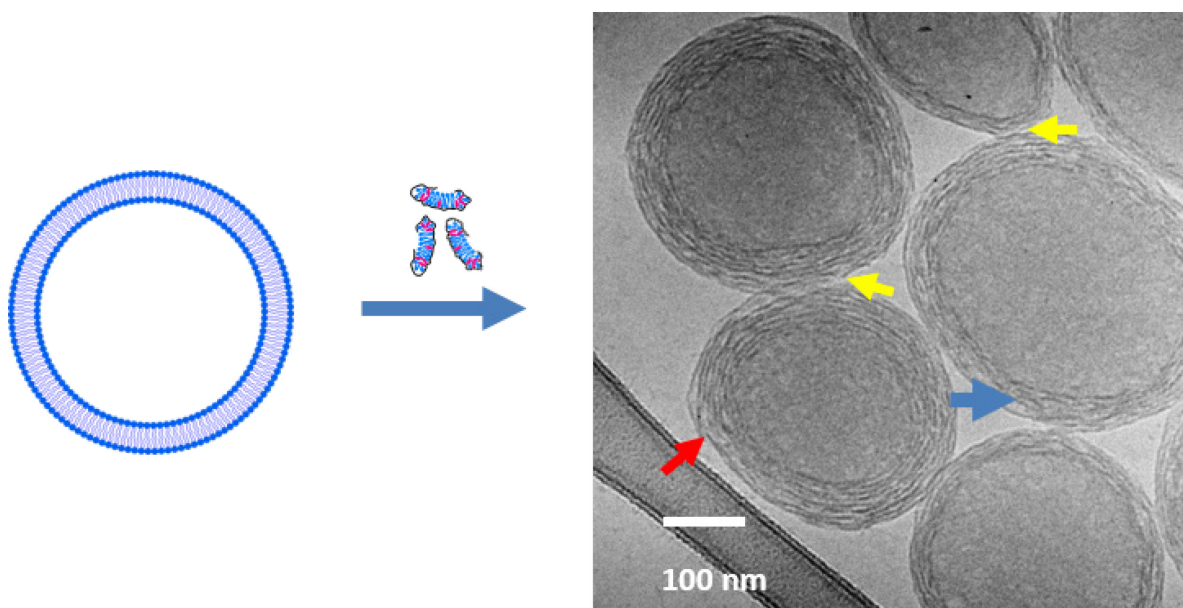
Small-angle X-ray scattering (SAXS) experiments were performed at the Advance Photon Source on beamline 12-BM. All measurements were conducted with the 12 keV beam. All measurements were conducted at 25 °C. The samples were loaded in 1.5 mm quartz capillaries and placed on a sample holder at a sample-to-detector distance of 2 m. The data are presented as absolute intensity versus the wave vector  $q = 4\pi \sin(\theta/2)/\lambda$ , where  $\lambda$  is the wavelength and  $\theta$  is the scattering angle. The reduction of SAXS data and background subtraction were performed by using Irena SAS macros on IGOR pro software.

## *Fluorescent microscopy*

FITC-dextran was encapsulated within liposomes by hydrating lipid film with aqueous solution of FITC-dextran. The loaded liposome suspension was transferred to a syringe and extruded 21 times through an 800 nm polycarbonate membrane. Unencapsulated FITC-dextran was removed by dialyzing through a dialysis bag (MW cutoff: 30 kDa) against a 100:1 deionized water bath volume at 25 °C for 1 h. Fluorescent microscopy images were taken with a Nikon A1 confocal microscope. 20  $\mu$ L sample was pipetted onto a standard microscope slide. 488 nm laser was used to excite the fluorescence-tagged samples, the emission wavelength is 525 nm.

## RESULTS AND DISCUSSION

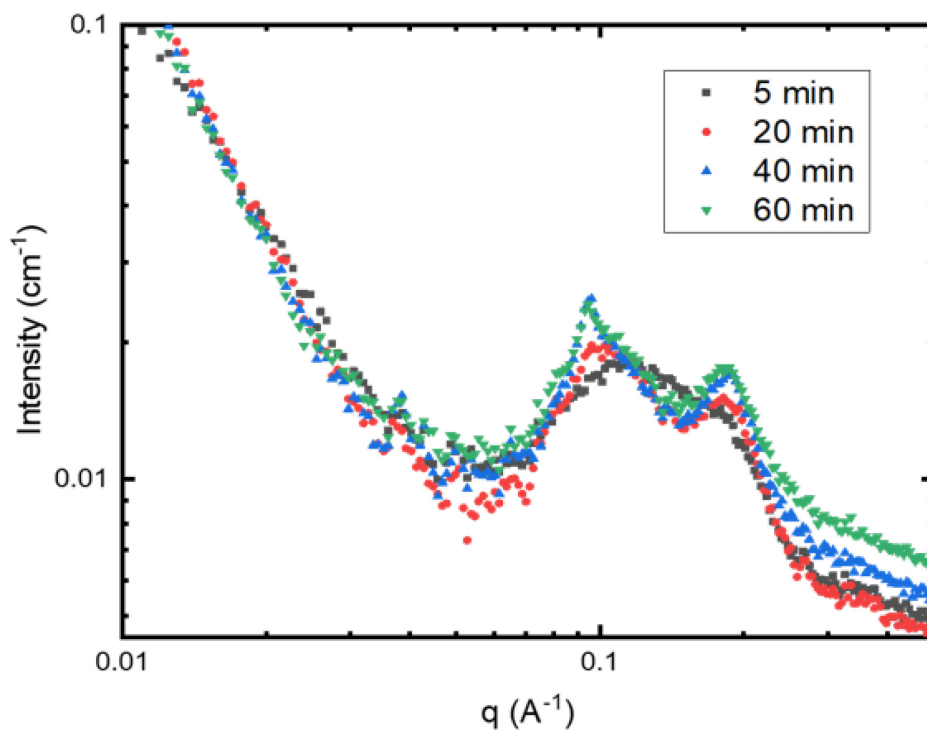
Figure 3 illustrates the primary finding that we seek to understand in this paper. The concise description of the phenomenon is as follows: 1) upon mixing lipid vesicles (Figure 2) with HMP at a lipid to HMP weight ratio of 1:2 the vesicles rupture and equilibrate into approximately 10 nm fragments (Figure 2); 2) the fragments when mixed with fresh liposomes (2:1 volume ratio of liposomes (0.25 wt% lipid) to HMP-lipid complexes (0.25 wt% lipid, 0.5 wt% HMP)) lead to the formation of a fascinating structure of multilayered vesicles as shown in Figure 3. On close examination of the vesicles, we observe the following: (1) The layered vesicles are usually larger than the original PC liposomes, (2) the layers do not appear to be continuous, (3) the individual layers appear to be more loose and flexible rather than the tight curvatures formed in liposomes or traditional multilamellar vesicles (MLVs). Since the layers do not appear to be continuous lamelli,



**Figure 3.** Multilayered vesicle formation: liposomal templates (0.25 wt% lipid) and HMP-lipid (0.25 wt% lipid and 0.5 wt % HMP) complexes are mixed to generate these multilayered vesicles (final concentrations: 0.25wt% lipid, 0.17wt% HMP). The waviness of the layers and the lack of full continuity (blue arrows) are shown. The yellow arrows point to potential connections between these vesicles and the red arrow is a free ending of a layer.

we do not refer to these structures as multilamellar vesicles but rather as multilayered vesicles. We also see what appears to be connections between these structures, but it is not entirely clear if these are actual connections or if these result from small overlaps between adjacent vesicles.

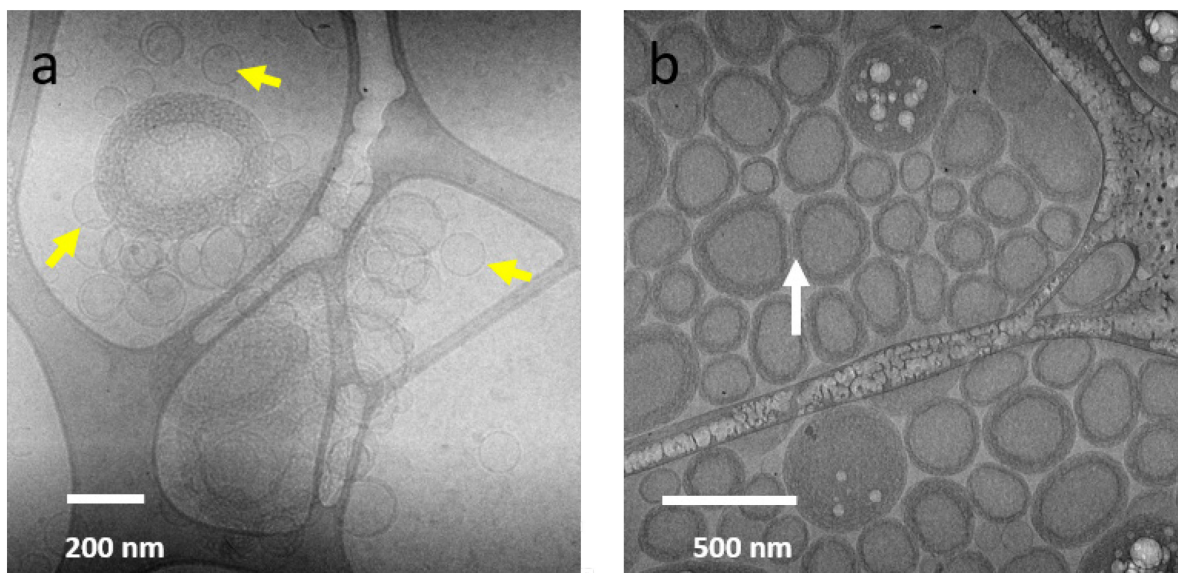
We conducted a time dependent SAXS analysis of the process of the multi-layer formation at the Advanced Light Source at Argonne National Laboratory, with the results shown in Figure 4.



**Figure 4.** The transition to multilayered vesicles upon HMP-lipid complex addition from 1 min to 60 min. Diffraction peaks begin appearing at 40 min post mixing indicating the presence of multilayered vesicles.

The sample was kept stationary in a capillary and the high flux of the synchrotron x-ray radiation allows sufficient data acquisition in 10 seconds, providing an opportunity to capture the scattering curves as the sample undergoes transitions in real time. The  $q^{-2}$  decay at low  $q$  is indicative of the presence of bilayer structures.<sup>17</sup> Broad diffraction peaks emerge as early as 20 mins and sharpen and stabilize around 40 mins. As the incubation time increases, the signal reveals the diffraction peaks at  $q=0.095 \text{ \AA}^{-1}$  and  $q=0.19 \text{ \AA}^{-1}$ , where the first peak indicates a repeat distance  $d$  of 6.6 nm and the second peak is the higher order peak verifying a lamellar structure. The broadness of the peaks is perhaps correlated to the fact that these are multilayered structures with wavy sheets and with rather imprecise spacings.

We have also examined the evolution of the multilayered structure through cryo-TEM as shown in Figure 5.

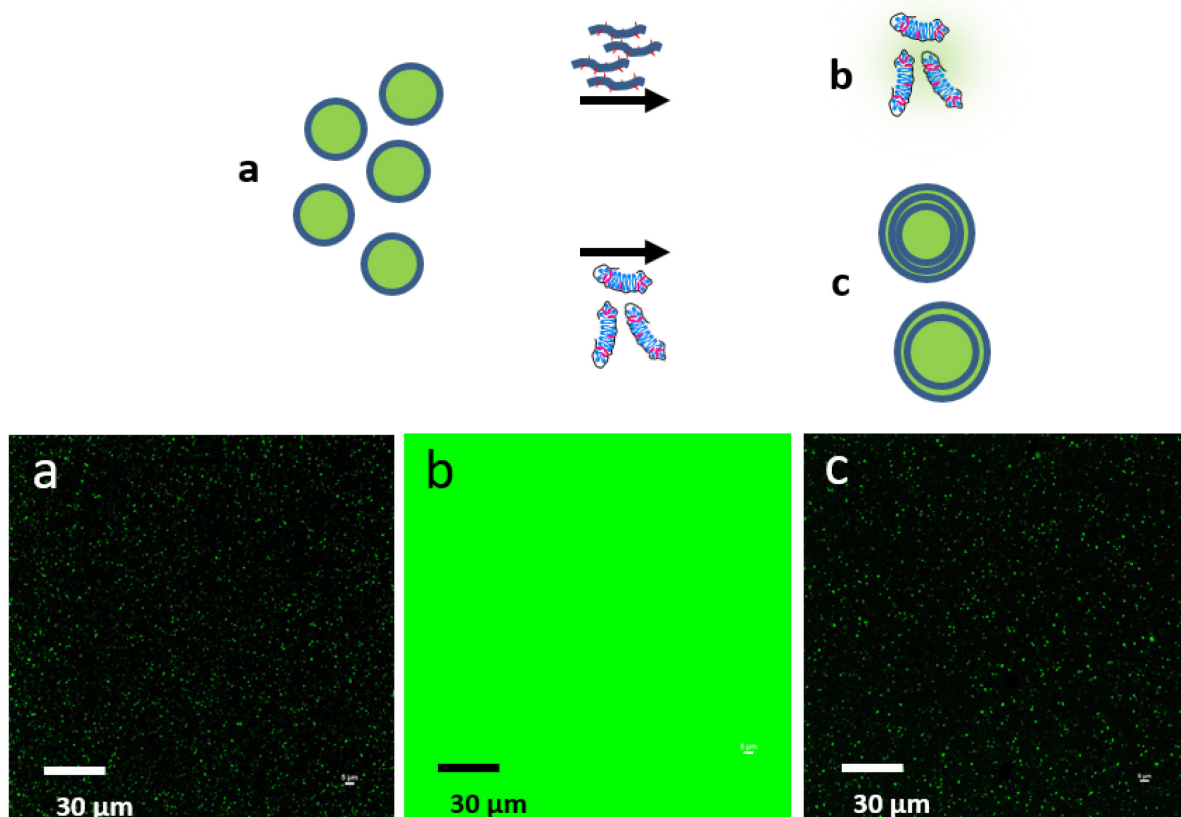


**Figure 5.** Structural transitions as a function of time (a) aggregation of vesicles is observed 20 min post mixing of complexes and liposomes together with remaining unilamellar vesicles are present (yellow arrows), (b) multilayer vesicles are formed in 4 hours post mixing. The flexibility of the layers in conforming curvature to adjacent vesicles is noted by the white arrow. The system is made up with (0.5% HMP25+ 0.25% lipid) that form the HMP-lipid complexes added to 0.25% lipid containing vesicles to obtain a final concentration of 0.17% HMP25, 0.25% lipid.

Vitrifying the sample after 20 mins shows a transition from the essentially unilamellar structures of Figure 5 (left) to a system containing a mixture of remnant unilamellar liposomes with the emergence of bilayered and multilayered vesicles. It is also important to note the observation of vesicle clustering. The clustering could be the initial step of multilayered vesicle formation where the HMP-lipid complexes attach to unilamellar vesicles and bring vesicles together. It is also possible although somewhat speculative, that the depletion effect of adding small colloids (the HMP-lipid complexes) to the much larger liposomes leads to the clustering of the larger liposomes following which growth into multilayered structures occurs. We note however, that we have not seen the growth into layered structures with unmodified polypeptoids (UMP) where all the nitrogen substituents are the methoxy ethyl moieties (Figure 1), clearly indicating that it is the alkyl hydrophobes on the backbone responsible for this transition.<sup>8a</sup>

When the sample is incubated at room temperature for 4 hours, large areas of the grid contain the multilayered vesicles as shown in Figure 5 on the right. We see a flexibility in the curvature of the multilayered vesicles and a tendency for two adjacent vesicles to flatten. In the 4 hour time period, we also see some extremely large multilayered vesicles as shown in Supporting Information S1 which also illustrates additional cryo TEMs of the flexible vesicles. Again, we note the novelty of this transition from unilamellar liposomes to multilayered vesicles through the addition of such HMP-lipid fragments. The literature cites several examples of polymers being able to break liposomes into fragments and this is indeed the basis for membrane protein extraction through the use of styrene-maleic acid (SMA) amphiphilic polymers (amphipols)<sup>18</sup> and dendrimers.<sup>19</sup> However, we are unaware of any studies where such polymer-lipid fragments have been reported to bridge bilayers to form multilayered vesicular structures.

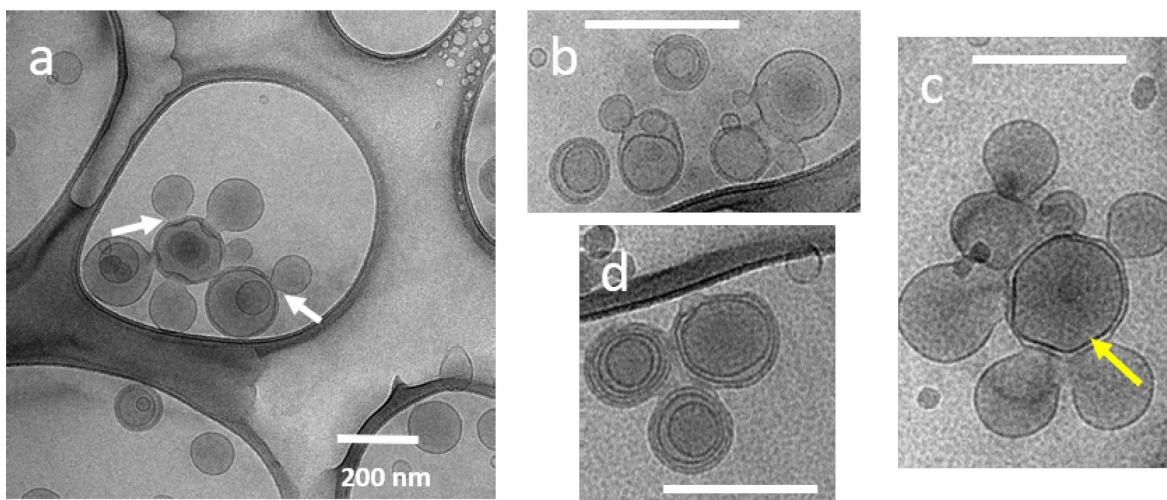
A crucial part of understanding the mechanism of multilayer formation is in assessing the structural stability of the added lipid vesicles acting as templates for growth of the layers. In other words, it is necessary to understand if liposomes originally loaded with a water-soluble drug lose their cargo when they transition to the multilayer structure. Accordingly, we loaded liposomes with fluorescent FITC-dextran and then added the HMP complex to this system as shown in the pathway from a to b in Figure 6.



**Figure 6.** The FITC-dextran leak test to show retention of liposomal cargo upon transformation from unilamellar to multilayer vesicles. (a) represents 0.25 wt% liposomes loaded with 0.02 wt% FITC-dextran (4kD). (b) represents the system when 0.5 wt% HMP is added to disrupt the liposomes and release the dye into solution, (c) represents the case where the HMP-lipid complexes (final lipid concentration 0.25 wt%, HMP 0.17 wt%) is added to system (a) to generate multilayer vesicles without significant release of the dye. The circle represents a liposome bilayer.

The HMP complexly disrupts the liposomes releasing the fluorescent dye as seen in the transition from the bright pinprick type fluorescent pattern in system a to the broad background fluorescence in system b. The pathway from a to c is one where liposomes loaded with FITC-dextran is contacted with the HMP-lipid complexes. The retention of the bright dot pattern in system c is an indication that there is negligible dye leakage in this pathway. Thus, the observation indicates that there must be clear fusion or bridging between liposomes in the creation of the multilayers to allow retention of the cargo in the multilayered structure.

To try to arrest the formation of the multilayered liposomes we conducted an experiment where we added just a small aliquot of the HMP-lipid complexes (10 v% of the level used to rapidly form the multilayered vesicles) to fresh liposomes and incubated the system for 24 hours prior to vitrification and imaging. Interestingly, as Figure 7 illustrates, there is clear evidence of

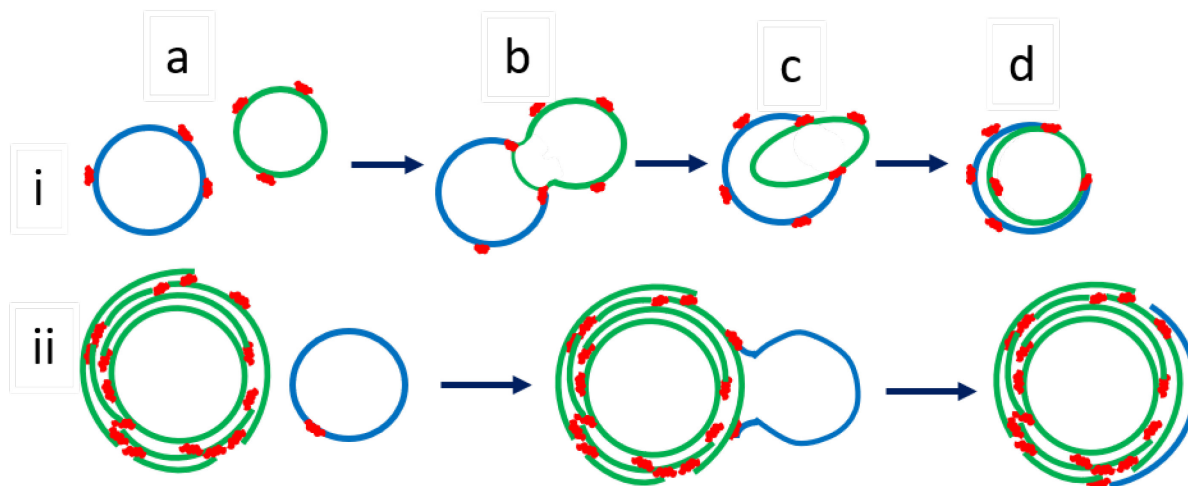


**Figure 7.** Vesicle clustering and fusion as an initial step, upon addition of a small quantity of the HMP-lipid complexes. 10v% of HMP-lipid complexes are added to fresh liposomes to understand the transition to multilayered vesicles (0.25wt% lipid, 0.05wt% HMP final composition, 24 h incubation). (a) Five outer vesicles attempting to fuse into the center vesicle results in a clustered state where fusion necks are observed (white arrow). (b), (c) (d) are additional images at higher magnifications that show the clustering, and the flattening of inner layers (yellow arrow) that we attribute to the incompleteness of the layers and internal pressure gradients from the fusion process. All scale bars are 200 nm.

vesicle fusion that is arrested. We have taken images from various parts of the TEM grid to show regions of multiple fused vesicles some containing multiple layers.

We also observe fusion “necks” where bilayers join (yellow arrows), and in vesicles with just a couple of layers we often see a flattening of layers (blue arrows) again indicating flexibility in the layers that may be made up of bilayer strands rather than a complete bilayer. In a sense these cryo-TEM images provide a rationale for the fact that large molecule contents of vesicles do not leak out during fusion which may be the initial step to multilayer vesicle formation. The observation is very similar to the vesicle fusion that is done by SNARE proteins (snap receptor proteins) that mediate neurotransmitter release,<sup>20</sup> although we note that the literature on SNARE proteins does not address the formation of multilayered vesicles. Thus, we see HMP-lipid complexes as being able to mimic SNARE protein behavior by being fusogenic to vesicles, and at high concentrations being able to form multilayered vesicles. The literature cites other examples of systems that induce fusion. In a fascinating example, carbon nanotubes have been shown to induce vesicle fusion through insertion of the nanotube into the bilayers of adjacent vesicles and allowing a sliding of lipid molecules along the hydrophobic surface of the nanotubes.<sup>21</sup> Metal ion binding to amphiphilic ligands consisting of synthetic bipyridine lipoligands has been reported to induce fusion of vesicles leading to giant vesicles.<sup>22</sup> These results indicate that bridging vesicles could be a general phenomenon to induce fusogenesis. We note that addition of the HMP-lipid complexes to 40% of the level required to rapidly form the multilayered vesicles, also leads to multilayered vesicles albeit seemingly with a reduced number of layers. The results are shown in Supporting Information S2, and perhaps points to variations in the rate of formation of the multilayered vesicles as a function of the concentration of the complexes. We also note the flexibility of the curvatures of the multilayered vesicles shown in Supporting Information S2.

Based on these observations we propose a mechanistic model of multilayered vesicle formation as shown in Figure 8.



**Figure 8.** Potential transition mechanisms of unilamellar vesicles to multilayered vesicles: The top row (i) shows the process of fusion of two unilamellar vesicles (a) to a two-vesicle cluster (b) followed by engulfment (c) and the formation of a bilayered vesicle (d). The bottom row (ii) shows the continuation of the process to multilayered vesicles with layers that are not continuous.

First, it is recognized that the hydrophobic interaction is responsible for the tendency of the alkyl chains of HMP to shield themselves from water and embed into the lipid bilayer. Functionalization of the polypeptoid yields randomly distributed alkyl chains throughout the backbone. The HMP-lipid fragments are prepared from mixing lipid and HMP at a 9:1 molar ratio (0.25 wt% lipid with Mw 775 g/mol and 0.5 wt% HMP25 with Mw 13900 g/mol). Every molecule of HMP contains on average 25 randomly distributed decyl groups as the hydrophobes. If there are 9 lipid molecules attached to each HMP, the number is translated to approximately 1 lipid for every 3 hydrophobes. The observation that HMP on its own disrupts lipid bilayers implies that the 25 hydrophobes on each HMP molecule are available to insert into lipid bilayers and disrupt the bilayers. On the other hand, in the HMP-lipid complex, some of the hydrophobes are noncovalently

attached to lipid species and there are fewer hydrophobes available to create membrane disruption. We therefore propose that the HMP-lipid complexes attach to liposomes without disrupting them as a first step as shown in (a) of Figure 8.

The self-assembly to multilayered vesicles may begin with vesicle clustering through the depletion interactions brought about by the HMP-lipid complexes that are initially in solution prior to interaction with the vesicles. After attachment to a vesicle some of the free hydrophobes then insert into lipid bilayers of an adjacent vesicle forming the fusogenic cluster as shown in (b) of Figure 8 which also indicates the formation of bridging necks of two vesicles undergoing fusion. (c) and (d) show the process of an outer vesicle (in green) engulfing an inner vesicle (in blue) to form a bilayered system. The sequence in the second row essentially shows addition of layers where a vesicle (blue) fuses to a layered vesicle (green) and then opens up to engulf the larger layered vesicle and thus add a portion of an additional layer. Arresting the process with insufficient HMP-lipid complexes may lead to the formation of multilayered vesicles with attached single layer vesicles that are unable to fully fuse into another layer as shown in the cryo-electron micrographs of Figure 7. A somewhat similar transition has been proposed to understand the fusion when anionic DNA is attached to cationic vesicles and therefore bridges between vesicles.<sup>23</sup> The mechanism of such induced fusion is electrostatics, while this work describes the bridging of vesicles using the hydrophobic interaction. A fascinating glimpse of such hydrophobic interactions inducing bilamellar vesicle formation has been shown by Raghavan's group using vesicles of a mixture of cationic and anionic surfactants, and hydrophobically modified chitosan, a 200 kDa cationic biopolymer.<sup>24</sup> These researchers have clearly pointed out the role of the hydrophobically modified chitosan in bridging liposome layers in the final conformation of a mixture of unilamellar and bilamellar vesicles.

Thus, as shown in the second row (ii) of Figure 8, we hypothesize the layering of vesicles occurs by building around existing layers. A new layer cannot assemble in between existing layers but can only become the outermost one. The multilayered vesicle formation schematic in Figure 8 is an attempt to explain the existence of open bilayers wrapped around vesicles through HMP-lipid complex bridging. The resulting structures can possibly undergo transitions described in Figure 8 until all the HMP-lipid fragments are used. The mechanisms outlined in Figure 8 serve to describe the processes occurring during the transition in a sequence but potentially take place very quickly to the stable conformation of multilayers as shown in Figure 3.

## CONCLUSIONS

To summarize the key aspects of this work, the addition of HMP-lipid complexes to lipid vesicles initiates a transformation to multilayered vesicles through an initial clustering and an engulfment of vesicles. It is remarkable that the clustering and engulfment retains large molecule intra-vesicular cargo without spillage into the bulk aqueous medium. The process can be controlled through the amount of the complexes that are added, leading to an arresting of intermediate structures en route to the formation of multilayered vesicles. The layers of the multilayered vesicles appear to be made of incomplete sheets of lipid bilayers which may be connected or bridged by the HMP-lipid complexes. The entire process is a consequence of hydrophobe insertion into lipid bilayers through the hydrophobic effect.

The consequences of these observations are significant. The biocompatibility of polypeptoids implies that these materials may find applications in therapeutics. While our earlier work has shown that the HMP-lipid complex can enter cells,<sup>15</sup> the observations shown here indicate that the complexes could be used in targeting intracellular organelles. Additionally,

multilayered vesicles can be designed with specific cargo in the core and multiple hydrophobic cargoes in the lipid bilayers opening new avenues in delivering therapeutics. The ability to fuse vesicles with small amounts of the complexes additionally points to analogies with SNARE proteins and the ability to influence neurotransmission. These are aspects that will be the focus of our continued work.

## ASSOCIATED CONTENT

### **Supporting Information**

Supporting Information Section S1 shows a cryo-TEM of a 4-hour sample. Section S2 shows cryo-TEM images of multilayered vesicles formed with a smaller aliquot of HMP-lipid complexes.

## AUTHOR INFORMATION

### **Corresponding Authors**

\*Vijay T. John, phone: 504-865-5883 e-mail: [vj@tulane.edu](mailto:vj@tulane.edu).

\*Donghui Zhang, phone: 225-578-4893 email: [dhzhang@lsu.edu](mailto:dhzhang@lsu.edu)

## ACKNOWLEDGMENTS

Support from the National Science Foundation grant NSF – 1805608 is gratefully acknowledged. Small angle x-ray scattering done at the Advanced Photon Source at Argonne National Laboratories (ANL) and was supported by the U.S. Department of Energy under EPSCoR

grant no. DE-SC0012432. We are grateful to Dr. Sungsik Lee at the Advanced Photon Source at ANL for assistance with the SAXS measurements.

It has been a privilege to be Lawrence Pratt's colleague for 14 years and for being able to learn from his knowledge of science and his wisdom.

## REFERENCES

1. Bangham, A., The 1<sup>st</sup> description of liposomes - A citation classic commentary on diffusion of univalent ions across the lamellae of swollen phospholipids by Bangham, A.D., Standish, M.M., and Watkins, J.C. *Current Contents/Life Sciences* **1989**, (13), 14-14.
2. (a) de Souza, T. P.; Fahr, A.; Luisi, P. L.; Stano, P., Spontaneous Encapsulation and Concentration of Biological Macromolecules in Liposomes: An Intriguing Phenomenon and Its Relevance in Origins of Life. *Journal of Molecular Evolution* **2014**, 79 (5-6), 179-192; (b) Monnard, P. A.; Deamer, D. W., Membrane self-assembly processes: Steps toward the first cellular life. *Anatomical Record* **2002**, 268 (3), 196-207.
3. Mamot, C.; Drummond, D. C.; Hong, K.; Kirpotin, D. B.; Park, J. W., Liposome-based approaches to overcome anticancer drug resistance. *Drug Resistance Updates* **2003**, 6 (5), 271-279.
4. Barba, A. A.; Bochicchio, S.; Dalmoro, A.; Lamberti, G., Lipid Delivery Systems for Nucleic-Acid-Based-Drugs: From Production to Clinical Applications. *Pharmaceutics* **2019**, 11 (8).
5. Kaczmarek, J. C.; Kowalski, P. S.; Anderson, D. G., Advances in the delivery of RNA therapeutics: from concept to clinical reality. *Genome Medicine* **2017**, 9.
6. (a) Lamichhane, N.; Udayakumar, T. S.; D'Souza, W. D.; Simone, C. B.; Raghavan, S. R.; Polf, J.; Mahmood, J., Liposomes: Clinical Applications and Potential for Image-Guided Drug Delivery. *Molecules* **2018**, 23 (2); (b) Bulbake, U.; Doppalapudi, S.; Kommineni, N.; Khan, W., Liposomal Formulations in Clinical Use: An Updated Review. *Pharmaceutics* **2017**, 9 (2).
7. (a) Diat, O.; Roux, D.; Nallet, F., EFFECT OF SHEAR ON A LYOTROPIC LAMELLAR PHASE. *Journal De Physique II* **1993**, 3 (9), 1427-1452; (b) Zipfel, J.; Berghausen, J.; Lindner, P.; Richtering, W., Influence of shear on lyotropic lamellar phases with different membrane defects. *Journal of Physical Chemistry B* **1999**, 103 (15), 2841-2849.
8. (a) Zhang, S. D.; Sun, H. J.; Hughes, A. D.; Moussodia, R. O.; Bertin, A.; Chen, Y. C.; Pochan, D. J.; Heiney, P. A.; Klein, M. L.; Percec, V., Self-assembly of amphiphilic Janus dendrimers into uniform onion-like dendrimersomes with predictable size and number of bilayers. *Proceedings of the National Academy of Sciences of the United States of America* **2014**, 111 (25), 9058-9063; (b) Hao, T. F.; Tan, H. N.; Li, S. L.; Wang, Y. L.; Zhou, Z. P.; Yu, C. Y.; Zhou, Y. F.; Yan, D. Y., Multilayer onion-like vesicles self-assembled from amphiphilic hyperbranched multiarm copolymers via simulation. *Journal of Polymer Science* **2020**, 58 (5), 704-715.
9. Luo, J. C.; Liu, T.; Qian, K.; Wei, B. Q.; Hu, Y. H.; Gao, M.; Sun, X. Y.; Lin, Z. W.; Chen, J. H.; Bera, M. K.; Chen, Y. H.; Zhang, R. M.; Mao, J. L.; Wesdemiotis, C.; Tsige, M.; Cheng, S. Z. D.; Liu, T. B., Continuous Curvature Change into Controllable and Responsive Onion-like Vesicles by Rigid Sphere-Rod Amphiphiles. *Acs Nano* **2020**, 14 (2), 1811-1822.
10. (a) Lahasky, S. H.; Hu, X.; Zhang, D., Thermoresponsive Poly(alpha-peptoid)s: Tuning the Cloud Point Temperatures by Composition and Architecture. *ACS Macro Letters* **2012**, 1 (5), 580-584; (b) Gangloff, N.; Ulbricht, J.; Lorson, T.; Schlaad, H.; Luxenhofer, R., Peptoids and Polypeptoids at the Frontier of Supra- and Macromolecular Engineering. *Chemical Reviews* **2016**, 116 (4), 1753-1802.
11. (a) Sun, J.; Zuckermann, R. N., Peptoid Polymers: A Highly Designable Bioinspired Material. *ACS Nano* **2013**, 7 (6), 4715-4732; (b) Zhang, D.; Lahasky, S. H.; Guo, L.; Lee, C.-U.; Lavan, M., Polypeptoid Materials: Current Status and Future Perspectives. *Macromolecules* **2012**, 45 (15), 5833-5841.

12. (a) Murnen, H. K.; Rosales, A. M.; Dobrynin, A. V.; Zuckermann, R. N.; Segalman, R. A., Persistence Length of Polyelectrolytes with Precisely Located Charges. *Soft Matter* **2013**, *9* (1), 90-98; (b) Rosales, A. M.; Murnen, H. K.; Kline, S. R.; Zuckermann, R. N.; Segalman, R. A., Determination of the Persistence Length of Helical and Non-helical Polypeptoids in Solution. *Soft Matter* **2012**, *8* (Copyright (C) 2016 American Chemical Society (ACS). All Rights Reserved.), 3673-3680; (c) Miller, S. M.; Simon, R. J.; Ng, S.; Zuckermann, R. N.; Kerr, J. M.; Moos, W. H., Proteolytic Studies of Homologous Peptide and N-Substituted Glycine Peptoid Oligomers. *Bioorganic & Medicinal Chemistry Letters* **1994**, *4* (22), 2657-2662.
13. (a) Xuan, S.; Lee, C.-U.; Chen, C.; Doyle, A. B.; Zhang, Y.; Guo, L.; John, V. T.; Hayes, D.; Zhang, D., Thermoreversible and Injectable ABC Polypeptoid Hydrogels: Controlling the Hydrogel Properties through Molecular Design. *Chemistry of Materials* **2016**, *28* (3), 727-737; (b) Lu, L.; Lahasky, S. H.; Zhang, D.; Garino, J. C., Directed Growth of Polymer Nanorods Using Surface-Initiated Ring-Opening Polymerization of N-Allyl N-Carboxyanhydride. *ACS Applied Materials & Interfaces* **2016**, *8* (6), 4014-4022; (c) Hörtz, C.; Birke, A.; Kaps, L.; Decker, S.; Wächtersbach, E.; Fischer, K.; Schuppan, D.; Barz, M.; Schmidt, M., Cylindrical Brush Polymers with Polysarcosine Side Chains: A Novel Biocompatible Carrier for Biomedical Applications. *Macromolecules* **2015**, *48* (7), 2074-2086.
14. Zhang, Y. H.; Xuan, S. T.; Owoseni, O.; Omarova, M.; Li, X.; Saito, M. E.; He, J. B.; McPherson, G. L.; Raghavan, S. R.; Zhang, D. H.; John, V. T., Amphiphilic Polypeptoids Serve as the Connective Glue to Transform Liposomes into Multilamellar Structures with Closely Spaced Bilayers. *Langmuir* **2017**, *33* (11), 2780-2789.
15. Zhang, Y. H.; Heidari, Z.; Su, Y.; Yu, T. Y.; Xuan, S. T.; Omarova, M.; Aydin, Y.; Dash, S.; Zhang, D. H.; John, V., Amphiphilic Polypeptoids Rupture Vesicle Bilayers To Form Peptoid-Lipid Fragments Effective in Enhancing Hydrophobic Drug Delivery. *Langmuir* **2019**, *35* (47), 15335-15343.
16. Xuan, S. T.; Gupta, S.; Li, X.; Bleuel, M.; Schneider, G. J.; Zhang, D. H., Synthesis and Characterization of Well-Defined PEGylated Polypeptoids as Protein-Resistant Polymers. *Biomacromolecules* **2017**, *18* (3), 951-964.
17. Lee, H. Y.; Hashizaki, K.; Diehn, K.; Raghavan, S. R., Reverse self-assembly of lipid onions induced by gadolinium and calcium ions. *Soft Matter* **2013**, *9* (1), 200-207.
18. (a) Burrige, K. M.; Harding, B. D.; Sahu, I. D.; Kearns, M. M.; Stowe, R. B.; Dolan, M. T.; Edelmann, R. E.; Dabney-Smith, C.; Page, R. C.; Konkolewicz, D.; Lorigan, G. A., Simple Derivatization of RAFT-Synthesized Styrene-Maleic Anhydride Copolymers for Lipid Disk Formulations. *Biomacromolecules* **2020**, *21* (3), 1274-1284; (b) Grethen, A.; Olusegun, A. O.; Danielczak, B.; Vargas, C.; Keller, S., Formation of lipid-bilayer nanodiscs by styrene/maleic acid (2:1) copolymer. *European Biophysics Journal with Biophysics Letters* **2017**, *46*, S303-S303; (c) Jamshad, M.; Lin, Y. P.; Knowles, T. J.; Parslow, R. A.; Harris, C.; Wheatley, M.; Poyner, D. R.; Bill, R. M.; Thomas, O. R. T.; Overduin, M.; Dafforn, T. R., Surfactant-free purification of membrane proteins with intact native membrane environment. *Biochemical Society Transactions* **2011**, *39*, 813-818.
19. Berenyi, S.; Mihaly, J.; Wacha, A.; Toke, O.; Bota, A., A mechanistic view of lipid membrane disrupting effect of PAMAM dendrimers. *Colloids and Surfaces B-Biointerfaces* **2014**, *118*, 164-171.
20. (a) Yang, Z. W.; Gou, L.; Chen, S. Y.; Li, N.; Zhang, S. L.; Zhang, L., Membrane Fusion Involved in Neurotransmission: Glimpse from Electron Microscope and Molecular Simulation. *Frontiers in Molecular Neuroscience* **2017**, *10*; (b) Marsden, H. R.; Korobko, A. V.; Zheng, T. T.;

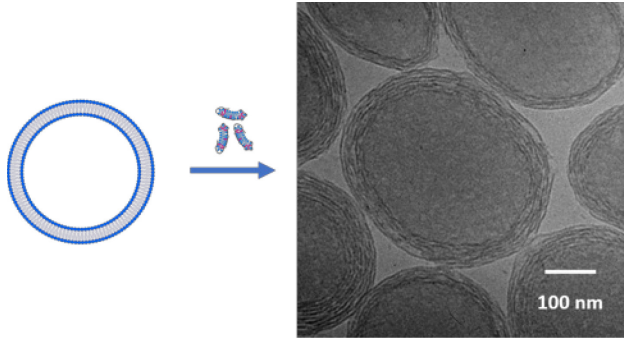
Voskuhl, J.; Kros, A., Controlled liposome fusion mediated by SNARE protein mimics. *Biomaterials Science* **2013**, *1* (10), 1046-1054.

21. Bhaskara, R. M.; Linker, S. M.; Vogele, M.; Kofinger, J.; Hummer, G., Carbon Nanotubes Mediate Fusion of Lipid Vesicles. *Acs Nano* **2017**, *11* (2), 1273-1280.

22. Richard, A.; Marchi-Artzner, V.; Lalloz, M. N.; Brienne, M. J.; Artzner, F.; Gulik-Krzywicki, T.; Guedeau-Boudeville, M. A.; Lehn, J. M., Fusogenic supramolecular vesicle systems induced by metal ion binding to amphiphilic ligands. *Proceedings of the National Academy of Sciences of the United States of America* **2004**, *101* (43), 15279-15284.

23. Huebner, S.; Battersby, B. J.; Grimm, R.; Cevc, G., Lipid-DNA complex formation: reorganization and rupture of lipid vesicles in the presence of DNA as observed by cryoelectron microscopy. *Biophys J* **1999**, *76* (6), 3158-66.

24. Lee, J. H.; Agarwal, V.; Bose, A.; Payne, G. F.; Raghavan, S. R., Transition from unilamellar to bilamellar vesicles induced by an amphiphilic biopolymer. *Physical Review Letters* **2006**, *96* (4).



multilayered lipid vesicles are formed via simple mixing of unilamellar liposomes and lipid-polypeptoid complexes