Copper-Catalyzed Conjugate Additions to Isocyanoalkenes

John-Paul R. Marrazzo,^a Allen Chao,^b Yajun Li,^c and Fraser F. Fleming^{a*}

- ^a Department of Chemistry, Drexel University, 3401 Chestnut St., Philadelphia, PA 19104-2875, USA.
- ^b Abzena, 360 George Patterson Blvd, Bristol, PA 19007, USA.
- ^c Fujian Institute of Research on the Structure of Matter, University of Chinese Academy of Sciences, 155 Yangqiao Road West, Fuzhou, Fujian, 350002, China

Supporting Information Placeholder

$$\begin{array}{c} R^2 \text{ Cat} \\ R^1 = NC \\ R_1 = H, \text{ Pr} \end{array} \qquad \begin{array}{c} R^2 \\ R^1 = R^2 \\ R^2 = AC \\ R_2 = AC \\ R_3 = R \end{array} \qquad \begin{array}{c} Nu \\ R_1 = R^2 \\ R_2 = R^2 \\ R_3 = R^2 \\ R_4 = R^2 \\ R_4 = R^2 \\ R_5 = R^2 \\ R_5 = R^2 \\ R_7 = R$$

ABSTRACT: A copper iodide-Pyox complex catalyzes the first conjugate addition of diverse sulfur, nitrogen, and carbon nucleophiles to isocyanoalkenes. The anionic addition generates metalated isocyanoalkanes capable of S_{Ni} displacements, providing a rapid route to a series of functionalized, cyclic isocyanoalkanes. The Cu(I)I-Pyox complex efficiently catalyzes a first-in-class conjugate addition affording a range of complex, functionalized isocyanoalkanes that are otherwise challenging to synthesize while laying a foundation for catalytic reactions that maintain the isocyanide group.

Introduction Isocyanides are chemical chameleons by virtue of the terminal, ambiphilic carbon that reacts with nucleophiles, lelectrophiles, radicals, and transition metals. The exceptional reactivity of isocyanides stems from the unusual bonding of the divalent, terminal carbonoid carbon whose evolution to a more stable tetravalent configuration is the driving force for many reactions: insertions (Scheme 1, $1 \rightarrow 2$), additions ($1 \rightarrow 3$), multicomponent reactions ($1 \rightarrow 4$), and radical cyclizations ($1 \rightarrow 5$). The exceptionally diverse reactivity profile provides rapid access to heterocycles such as 2 and 5, nucleosides, peptidomimetics (cf. 4), and natural product analogs.

Scheme 1. Representative Isocyanides Reactions.

In contrast to the diverse suite of reactions that deploy the terminal isocyanide carbon as a lynchpin, there are relatively few methods for synthesizing the types of substituted isocyanides required for pharmaceutical discovery. The fruitful use of conjugate additions to access complex scaffolds is, for additions to isocyanoalkenes, limited to a few stoichiometric, anionic conjugate additions (Scheme 2, $6 \rightarrow 7 \rightarrow 8$). The absence

of any catalytic conjugate addition to isocyanoalkenes reflects the extremely strong binding of most potential transition metal complexes that might serve as catalysts; isocyanides tenaciously complex transition metals making them very effective scavengers for removing trace metals. Not surprisingly, catalytic manipulations of isocyanides that maintain the isocyanide functionality are extremely rare. The difficulty reflects the strong σ -bond donation from the isocyanide carbon to the transition metal d-orbitals coupled with a π -donation from the metal to the isocyanide π^* orbital which favor insertion and addition reactions over reversable complexation.

Catalyzing conjugate additions to isocyanoalkenes requires overcoming two main challenges: irreversible complexation of the catalyst to the isocyanide and selective delivery of the nucleophile to the alkene rather than the inherently more reactive carbenoid isocyanide carbon. ¹² A highly efficacious copper(I) catalyst is described that reversibly complexes isocyanoalkenes to generate nitrilium-like intermediates 11 that are activated toward anionic conjugate addition (Scheme 2, $10 \rightarrow 11 \rightarrow 12 \rightarrow 9$). In contrast to previous work, the transformation is catalytic and does not require isocyanoalkenes bearing an additional electron withdrawing group (cf. 10 with 6).

Scheme 2. Conjugate Additions to Isocyanoalkenes.

Results and Discussion. An initial screening in which the isocyanoalkene 13a was treated with a series of nucleophiles 13 in the presence or absence of metal complexes 14 afforded mainly products resulting from addition to the terminal isocyanide carbon. The strategy for identifying a potential catalyst was therefore redesigned by first developing an uncatalyzed addition with highly nucleophilic PhSK and then adding metal salts as potential catalysts to accelerate the conjugate additions with the goal of ultimately identifying a complex capable of catalyzing the addition of carbon and nitrogen nucleophiles.

Sporadic, reversible, complexation of isocyanides to copper (I) salts hinted at the viability of copper complexes as catalysts for conjugate additions to isocyanoalkenes. 15 Copper(I)-isocyanide complexes are unusual in reversibly forming complexes because although the σ -donation from the isocyanide is strong, there is minimal π -donation from copper to the isocyanide. ¹⁶ In the event, addition of PhSK generated from PhSH and K2CO3 (10 mol%), provided the phenylsulfanylisocyanide 14a (Table 1, entry 1). A control experiment with no base gave only trace amounts of the phenylsulfanylisocyanide 14a (Table 1, entry 2). Screening a series of copper salt-ligand combinations (see Figure S1 in the SI for details) identified the combination of catalytic CuI¹⁷ and commercially available Pyox (L1)¹⁸ with K₂CO₃ as an efficacious promoter for the addition of PhSH to 13a (Table 1, entry 3). Cs₂CO₃ was found to promote a faster reaction than K2CO3, though the soluble bases KHMDS and Et3N dramatically accelerated the addition allowing complete conversion to 14a at -78 °C within 30 and 20 min, respectively (Table 1, entries 5 and 6). NaH promoted the fastest conversion of 13a to 14a requiring only 8 min at -78 °C; no conjugate addition was observed in a control reaction in the absence of CuI/CF₃Pyox (Table 1, cf. entries 7 and 8). Varying the structure of the PyOx catalyst in reactions with CuI and L2 and L3 gave virtually the same reaction profile as with L1 (Table 1, cf. entry 7 with entries 9, and 10).19

Table 1. Addition of PhSH to Isocyanoalkene 9a.

Entry	Base (equiv)	CuI/Pyox*a	T	Time	Yield ^b
		(mol %)	(°C)	(h)	
1	K ₂ CO ₃ (0.1)	-	25	18 h	75%
2	-	-	25	72 h	trace
3	$K_2CO_3(0.1)$	2.5	25	2 h	64%

4	$Cs_2CO_3(0.3)$	2.5	25	1 h	70%
5	KHMDS (1.0)	2.5	-78	30 min	78%
6	Et ₃ N (1.0)	2.5	-78	20 min	79%
7	NaH (0.1)	2.5	-78	8 min	85%
8	NaH (0.1)	-	-78	40 min	-
9	NaH (0.1)	2.5 L2	-78	8 min	67%
10	NaH (0.1)	2.5 L3	-78	8 min	87%

^a Ligand **L1** (5 mol %) was used unless stated otherwise. ^b Isolated yield after chromatography on C-2 silica. ²⁰

Identifying Cu(I)-Pyox as an efficient catalyst provided the foundation for expanding the conjugate addition to diverse sulfur, nitrogen, and carbon nucleophiles (Scheme 3). Addition of allylsulfide to 13a readily afforded 14b under conditions identical to that with thiophenol. In contrast, the conjugate addition of thiophenol to the o-substituted isocyanoalkene 13b (13, Ar = o-MeOC₆H₄) to afford 14c required elevating the reaction from -78 °C to rt; presumably, the syn-pentane-like interaction between the o-methoxy substituent and the alkene in 13b prevents planarity which retards the conjugate addition. Ethyl 3-mercaptopropanoate reacted efficiently with 13a to afford 14d while the addition of chloropropylthiol or bromopropylthiol to 13a afforded 14e and 14f, respectively (Scheme 3).

The successful conjugate addition of sulfenylates prompted an evaluation of anionic carbon nucleophiles (Scheme 3). Using the same Cu(I)-Pyox combination with diethyl methylmalonate smoothly provided the diester 14g, a class of ester-isocyanides that are otherwise challenging to synthesize.²² Addition of methylmalononitrile to 13a with Cu(I)-Pyox and NaH provided a very efficient route to the isocyanonitrile 14h; analogous conjugate additions with the significantly less acidic diphenylacetonitrile provided isocyanonitriles 14i and 14j. The nitrogen nucleophiles imidazole and phthalimide were equally as effective in the conjugate addition, affording the corresponding heterocycle-containing isocyanides 14k and 14l, respectively.

Scheme 3. Cu-Catalyzed Additions to Isocyanoalkenes.

^a Cs₂CO₃ was used in place of NaH.

The successful addition of carbon nucleophiles suggested generating carbocyclic isocyanides by harnessing the intermediate metalated isocyanide in an S_Ni displacement.²³ The conjugate addition-cyclization of bromomalonates was particularly appealing because the resulting isocyanocyclopropanes are precursors of functionalized cyclopropylamines,²⁴ valuable γ-turn mimetics that feature as pharmacophores²⁵ in cancer treatment.²⁶ Cu(I)-Pyox proved optimal²⁷ in catalyzing the addition of diethyl bromomalonate to **13a** with Cs₂CO₃ in THF²⁸ to afford **16a** (Scheme 4). A brief survey of bases revealed that NaH formed the isocyanocyclopropane **16a** more rapidly (20 min at rt), but less efficiently than Cs₂CO₃ (67% and 83%, respectively).²⁹

Catalytic Cu(I)/Pyox in combination with bromomalonates efficiently generated a range of isocyanocyclopropanes (Scheme 4). Modulating the steric demand of the malonate alkyl ester substituent by changing from ethyl to methyl had minimal effect, (cf. **16a** with **16b**) whereas the *t*-butyl malonate required 95 h for full conversion to the *t*-butyl ester-isocyanide **16c**. Isocyanoalkenes **13** with aromatic substituents, aromatics with electron withdrawing or electron donating substituents, or cinnamyl substituents, were equally well tolerated in forming the corresponding ester-isocyanides (**16a-e**, **16f-i**, and **16j**, respectively). The addition of diethyl bromomalonate to the β -substituted isocyanoalkene **15** (R¹ = p-CF₃C₆H₄) afforded the penta-substituted cyclopropane **16i** as the only detectable diastereomer. Extending the addition-cyclization to the allyl malonate **17**³¹ afforded the six-membered isocyanide **16k**.

Scheme 4. Cu-Catalyzed Conjugate Addition-Cyclizations.

Reactions were performed on a 0.1-1.0 mmol scale with Nuc (1.1-1.5 equiv), CuI (2.5 mol%), L1 (2.5 mol%), Cs₂CO₃ (1.2 equiv), THF (0.1 M), rt, 0.5-95 h. ^a The yield on a 1.0 mmol scale was 70%. ^b K₂CO₃ was employed rather than Cs₂CO₃. ^c Performed at 50 °C for 20 h for complete conversion.

Mechanistically, the conjugate additions likely proceed via complexation of the isocyanide to a tetrahedral Cu(I)I-Pyox complex 18.32 13C NMR of the isocyanoalkene 13a with Cu(I)I in d₈-THF showed signal suppression of the ¹³C NMR isocyanide carbon, the olefinic carbon, the proximal quarternary carbon, and the ortho-methine carbons (see Figure S2 in the SI for details), consistent with a fast, reversible complexation of copper to the terminal isocyanide carbon (Scheme 5, $18 \leftrightarrows 19$). Association of iodide in the copper-isocyanide complex 18 appears likely because addition of iodide scavengers (AgBF₄, AgOTf, NaBPh₄) afforded a much less efficient catalyst. Complex 19 activates the alkene toward anionic conjugate addition by creating a nitrilium-like intermediate while preventing nucleophilic attack on the terminal carbon.³³ Nucleophilic addition to 19 generates 20 whose protonation (Scheme 3) or cyclization (Scheme 4) sets the configuration at the chiral center; an inability to coax asymmetry in the conjugate addition is consistent with location of the chiral ligand distal to the newly formed chiral center. The resulting, neutral copper-isocyanide complex releases the isocyanoalkane **14** and frees the catalyst for further turnover.

Scheme 5. Proposed Catalytic Cycle.

Isocyanocyclopropanes provide a direct route to cyclopropylamides that are potent pharmacophores and serve as precursors to nucleosides³⁴ and heterocycles.³⁵ As an illustration of the potential for isocyanocyclopropanes in multi-component cascades, **16b** was treated with imidazole **21** and paraformaldehyde to efficiently generate imidazopyrazine **22** (Equation 1).

Ph
$$CO_2Me$$
 CO_2Me CO_2Me

ConclusionsCu(I)-Pyox catalyzes therst-in-kind conjugate addition of diverse nucleophiles isocyanoalkensewhile maintaining the isocyanide group. The delicate activation omotes conjugate addition with a range of sulfur, nitrogen, and carbon nucleophiles through activation of the isocyanoalkene by reversible binding of the catalyst. The intermediate metalated isocyanides can be protonated or trapped to generate isocyanides with complex molecular architectures that are otherwise challenging to access. Identification of the key mechanistic features in the active copper catalyst provide an extremely promising foundation for developing additional copper-catalyzed reactions while providing a valuable synthetic route to a diverse array of functionalized isocyanides.

Experimental Section

General Experimental Conditions: Tetrahydrofuran (THF) was freshly distilled from Na/benzophenone ketyl prior to use. All reactions were performed in dry glassware under an atmosphere of dry nitrogen. Other reagents were purchased as analytical or ACS grade and used without further purification unless stated otherwise. Saturated, aqueous ammonium chloride was buffered to pH = 7.0 by addition of ammonium hydroxide. Reactions that required a microwave heat source were performed in sealed 2 mL microwave reaction vials (Biotage #355629) equipped with a Teflon-coated stir bar and were maintained at a constant temperature using a Biotage® Initiator+ microwave reactor, equipped with an IR-sensor. Thin layer chromatography (TLC) was performed with glassbacked, 250 μm thickness, F254 hard layer SiliaPlate TLC Plates purchased from Silicycle. TLC plates were visualized by exposure to short wavelength UV light (254 nm). Flash chromatography was

performed using the Buchi Reveleris X2 Automated Flash Chromatography System with commercial or self-prepared cartridges filled with SilicaFlash ® silica gel P60 (30-400 mesh) purchased from Silicycle or with "C-2 silica" prepared as previously described.²⁰ ¹H NMR and ¹³C NMR high resolution nuclear magnetic resonance spectra were recorded at room temperature on a Varian Mercury Plus 400 (400 MHz/101 MHz) or a Varian Unity Inova 500 (500 MHz/126 MHz) spectrometer. Chemical shifts were referenced to CDCl₃ (\delta 7.26) for 1H NMR and CDCl₃ (\delta 77.16) for 13C NMR. ¹H NMR data are reported as follows: chemical shift, multiplicity (s =singlet, d = doublet, t = triplet, q = quartet, dd = doublet of doublet, m = multiplet, br = broad resonance, etc.), integration, and coupling constant (Hz). ¹H and ¹³C NMR data are reported in parts per million (ppm) on the δ scale and referenced to tetramethylsilane or the proton residual of CHCl₃ for ¹H NMR and the carbon signal of CDCl₃ for ¹³C NMR. High-resolution mass spectra were obtained on a Thermo-Electron LTQ-FT 7T Fourier transform ion cyclotron resonance (FT-ICR) spectrometer with an atmospheric pressure chemical ionization (APCI) source with direct infusion run in positive ion mode at 5 kV. Enantiomeric ratios (er's) were determined by performing chiral high pressure liquid chromatography (HPLC) analysis on a Shimadzu Nexera X2 instrument with a Daicel Chiral Technologies Inc. CHIRALPAK® IE, amylose-based (Amylose tris(3,5-dichlorophenylcarbamate) immobilized on 5 µm silica-gel) column (4.6 mm x 250 mm).

General IsocyanoalkeneSynthesis Procedure: Isocyanoalkenes were prepared following a modification of the published procedure. ³⁶ A solution of alkyllithium (1.2 equiv) was added to a -60 °C, THF (0.08 M) solution of the nitrile (1.0 equiv) and CuCN (0.02 equiv). After 15-30 min, neat isopropyl formate (5.0 equiv) was added and then the mixture was heated to 50 °C. After 12-20 h, when the tautomerization was complete as judged by TLC analysis, the reaction was allowed to cool to rt and then cooled to -60 °C. Neat triethylamine (9.0 equiv) was added followed by the dropwise addition of neat phosphoryl chloride (3.0 equiv). Upon completion, as monitored by TLC, the reaction mixture was poured into saturated, aqueous sodium carbonate, the phases were separated and then the aqueous phase was extracted with EtOAc (3 x 15 mL). The combined organic extract was washed with water (1 x 20 mL) and brine (1 x 20 mL), dried (Na₂SO₄), filtered, and then concentrated under reduced pressure. The crude product was purified by flash chromatography.

1-(1-isocyanovinyl-)2-methoxybenzen(43b). Prepared following the generalisocyanoalkenesynthesis procedure with 2methoxybenzonitrile (500 mg, 3.76 mmol), CuCN (6.7 mg, 75 µmol), THF (37 mL, 80 mM), MeLi (2.83 mL, 4.50 mmol) except that isopropyl formate (1.88 mL, 18.8 mmol) was added to a -30 °C solution for 40 min before the mixture was heated to 50 °C. After 48 h, the reaction mixture was allowed to cool and then diisopropylamine (4.61 mL, 33.8 mmol) and POCl₃ (1.05 mL, 11.3 mmol) were added. Purification (dry-loaded with Celite onto a 12 g silica gel cartridge, 10% EtOAc/Hexanes) afforded 357 mg (60%) of the isocyanoalkene 13b as a yellow liquid: IR (ATR): 3007, 2839, 2112, 1601 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.53 (dd, J= 7.8, 1.7 Hz, 1H), 7.39-7.33 (m, 1H), 7.02 (tdd, J= 7.8, 1.0, 0.5 Hz, 1H), 6.96 (br d, J = 8 Hz, 1H), 6.01 (t, J = 5.2 Hz, 1H), 5.79-5.76 (m, 1H), 3.90 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 172.5, 161.9, 157.2, 130.8, 129.2, 121.1, 120.7, 119.5, 111.4, 55.6; HRMS (+APCI) m/z: [M + H]⁺ Calcd. for C₁₀H₁₀NO 160.0762; Found 160.0758.

(Z)-1-(1-isocyanopentl-en-1-yl)-4-(trifluoromethyl)benzene (15a). Prepared following the general isocyanoalkenes synthesis procedure with 4-(trifluoromethyl)benzonitrile (500 mg, 2.92 mmol), CuCN (5.2 mg, 0.060 mmol), THF (30 mL, 0.10 M), BuLi (2.19 mL, 3.50 mmol). After 2 h at -60 °C, isopropyl formate (1.46

mL, 14.6 mmol) was added, the mixture was heated to 50 °C for 25 h, cooled and then diisopropylamine (3.59 mL, 26.3 mmol) and POCl₃ (0.81 mL, 8.8 mmol) were added to afford, after purification (dry-loaded with Celite onto a 12 g silica gel cartridge, 100% Hexanes eluent), 106 mg (23%) of **tise**cyanoalken**45a**as a yellow oil: IR (ATR): 2966, 2877, 211,81686 cm¹; ¹H NMR (400 MHz, CDCl₃) δ 7.737.54 (m, 4H), 6.39 (t, J= 7.4 Hz, 1H), 2.48 (q, J= 7.4 Hz, 2H), 1.58 (sext, J= 7.4 Hz, 2H), 1.02 (t, J= 7.4 Hz, 3H); 13 C { 1 H} NMR (100 MHz, CDCl₃) δ 167.4, 135.8 (q, ^{4}J = 1.4 Hz), 133.5, 130.8 (q, ^{2}J = 32 Hz), 125.8 (q, ^{3}J = 3.8 Hz), 125.08, 125.05, 123.8 (q, ^{1}J = 270 Hz), 31.3, 21.7, 13.8; HRMS (+APCI) m/z: [M + H] $^{+}$ Calcd. for C₁₃H₁₃F₃N 240.1000; Found 240.0994.

Diethyl 2-(2-(chloromethyl)allyl)malonat (17). Prepared following a modification of the published procedure.³⁷ Neat diethylmalonate (0.476 mL, 3.12 mmol) was added dropwise to a 0 °C, DMF suspension (15 mL, 0.20 M) of NaH (43.0 mg, 1.78 mmol) and then the reaction mixture was allowed to warmed to rt. After 55 min, the reaction was transferred by syringe to a DMF solution (15 mL, 0.31 M) of methallyl dichloride (0.541 mL, 4.68 mmol). After 14 h, aqueous 1.0 M HCl (20 mL) was added and then the mixture was extracted with diethyl ether (4 x 20 mL). The combined organic extract was washed with water (1 x 30 mL) and brine (1 x 30 mL), dried (MgSO₄), filtered, and then concentrated under reduced pressure. The crude product was purified by flash chromatography (12 g silica gel cartridge, 5% EtOAc/Hexanes eluent) to afford 210 mg (27%) of i as a colorless liquid: IR (ATR): 2983, 1729, 1646 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.25-5.19 (m, 1H), 5.02 (td, J= 1.4, 1 Hz, 1H), 4.20 (q, J= 7.0 Hz, 2H), 4.19 (q, J= 7.0 Hz, 2H) 4.06 (d, J=1 Hz, 2H), 3.62 (t, J=7.9 Hz, 1H), 2.79 (br d, 7.9 Hz, 2H), 1.26 (t, J=7 Hz, 6H); ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) δ 168.7, 141.5, 116.6, 61.6, 50.4, 47.8, 32.0, 14.0; HRMS (+APCI) m/z: [M + H]⁺ Calcd. for C₁₁H₁₈O₄Cl 249.0894; Found 249.0891.

General IsocyanoalkeneConjugate Addition Procedure: Neat nucleophile (1.5 equiv) was added to a rt, THF solution (0.10 M) of isocyanoalkene (1.0 equiv), base (0.30 – 1.0 equiv), copper iodide (0.025 equiv), and (*S*-4-tert-butyl-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (0.025 equiv). The reaction was monitored by TLC (1 – 10 h for cesium carbonate or 5 min – 24 h for NaH) and when complete, saturated, aqueous ammonium chloride was added. The phases were separated, the aqueous phase was extracted with EtOAc (3 x 2 mL), and then the combined organic phase was washed with water (1 x 10 mL) and brine (1 x 10 mL), dried (Na₂SO₄), and filtered. The crude isocyanide was concentrated under reduced pressure and was then purified by silica gel chromatography.

(2-([1,1'-Biphenyl]-4-yl)-2-isocyanoethyl)(phenysulfane (14a). Prepared following the general isocyanoalkene conjugate addition procedure with 4-(1-isocyanovinyl)-1,1'-biphenyl (13a)³⁶ (30.0 mg, 146 mmol), Cs₂CO₃ (14.3 mg, 43.0 mmol), Cu(I)I (0.70 mg, 4.0 mmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (1.0 mg, 4.0 mmol), and thiophenol (20 mL, 200 mmol) for 1 h at rt to afford, after purification (dry-loaded with Celite onto C-2 silica, 10% EtOAc/Hexanes eluent), 39 mg (85%) of the isocyanosulfide **14a**as a colorless oil: IR (ATR): 3032, 2138 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.62-7.55 (m, 4H), 7.48-7.41 (m, 4H), 7.40-7.26 (m, 6H), 4.76 (dd, J = 7.9, 5.9 Hz, 1H), 3.43-3.31 (m, 2H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 158.6, 141.9, 140.2, 134.7, 134.0, 131.2, 129.3, 128.9, 127.7, 127.7, 127.5, 127.1, 126.7, 57.9, 42.8; HRMS (+APCI) m/z: [M + H]⁺ Calcd. for C₂₁H₁₈NS 316.1160; Found 316.1154.

1-Phenyl2-(phenylthio)ethanl-amine (i). Prepared following a modification of the published procedure. Four drops of concentrated, aqueous HCl (12 M) was added to a rt, ethanolic solution (0.2 M) of **14a** (30.0 mg, 95 μ mol). After 18 h, the ethanol was

removed under reduced pressure and then the residue was dissolved in CH₂Cl₂ (5 mL) that was then washed with water (5 mL). The aqueous phase was separated, neutralized with 2M NaOH until pH = 7 as determined by monitoring with pH paper, and then extracted with CH₂Cl₂ (3 x 5 mL), dried (Na₂SO₄), and concentrated under reduced pressure to afford, after purification (dry-loaded with Celite onto 4g silica gel cartridge, 0 to 100% EtOAc/Hexanes gradient eluent), 29 mg (99%) of the primary amine i as a colorless oil: IR (ATR): 3366, 3288, 3028, 2921, 2860 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.59-7.54 (m, 4H), 7.46-7.38 (m, 6H), 7.36-7.27 (m, 3H), 7.23-7.18 (m, 1H), 4.14 (dd, J= 9.4, 4.0 Hz, 1H), 3.34 (dd, J= 13.4, 4.0 Hz, 1H), 3.06 (dd, J=13.4, 9.4 Hz, 1H), 1.87 (s, 2H); ${}^{13}C\{{}^{1}H\}$ NMR (400 MHz, CDCl₃) δ 143.2, 140.8, 140.5, 135.8, 129.8, 129.0, 128.8, 127.34, 127.27, 127.1, 126.8, 126.4, 54.4, 43.8; HRMS (+APCI) m/z: [M - NH₂] Calcd. for C₂₀H₁₇S 289.1051; Found 289.1049.

(E)-N-(1-([1,1'-biphenyl]-4-yl)-2-(phenylthio)ethyl)1-(2-((S)-dinaphtho[2,1d:1',2'-f][1,3,2]dioxaborepin4-yl)phe-nyl)methanimine(ii). Prepared following the published general procedure: 39 2-Formylphenylboronic acid (7.8 mg, 52 µmol), and (S-BINOL (17 mg, 57µmol) were added to a CDCl₃ (1.0 mL) suspension of the amine i (10 mg, 52 µmol), and 4 Å molecular sieves. After 5 min at rt, an aliquot was removed and the 1 H NMR spectrum of ii was obtained to determine the enantiomeric ratio (1:1.0) of the staring primary amine i by comparing the imine C-H for the (R) and (S) complexes: 1 H NMR (400 MHz, CDCl₃) δ 8.48 (s, 1H), 8.29 (s, 1H) Δ δ = 0.19 ppm.

(2-([1,1'-biphenyl]-4-yl)-2-isocyanoethyl)(allyl)sulfane (14b) Prepared following the general isocyanoalkene conjugate addition procedure with 4-(1-isocyanovinyl)-1,1'-biphenyl (13a)³⁶ (30.0 mg, 146 μmol), NaH (5.26 mg, 219 μmol), Cu(I)I (0.7 mg, 3.7 μmol), (*S*)-4-(*tert*-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5dihydrooxazole (1.0 mg, 3.7 µmol), and allylmercaptan (16.3 mg, 219 µmol) for 30 min at -78 °C to afford, after purification (dryloaded with Celite onto a C-2 silica gel, 20 3% EtOAc/Hexanes eluent), 35 mg (86%) of the isocyanoallylsulfane 14b as a clear oil: IR (ATR): 3030, 2919, 2139, 1633 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.67-7.56 (m, 4H), 7.50-7.42 (m, 4H), 7.41-7.35 (m, 1H), 5.84-5.71 (m, 1H), 5.18-5.01 (m, 2H), 4.85 (dd, J = 7.7, 5.7 Hz, 1H), 3.20-3.06 (m, 2H), 3.00 (dd, J= 14.1, 7.7 Hz, 1H), 2.93 (dd, J= 14.1, 5.7 Hz, 1H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 158.4, 141.9, 140.2, 135.0, 133.9, 128.9, 127.7, 127.6, 127.1, 126.7, 118.1, 58.8, 38.2, 35.4; HRMS (+APCI) m/z: [M - NC] C₁₇H₁₇S 253.1051; Found 253.1048.

(2-Isocyane2-(2-methoxyphenyl)ethyl)(phenyl)sulfa(14c) Prepared following the general isocyanoalkene conjugate addition procedure with 1-(1-isocyanovinyl)-2-methoxybenzene (13b) (30.0 mg, 188 μmol), Cs₂CO₃ (18.4 mg, 56.5 μmol), Cu(I)I (0.9 mg, 5 μmol), (*S*)-4-(*tert*-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5dihydrooxazole (1.3 mg, 5.0 µmol), and thiophenol (29 µL, 280 μmol), for 14 h at 50 °C to afford, after purification (dry-loaded with Celite onto a 4g gel silica cartridge, 20% EtOAc/Hexanes eluent), 29 mg (76%) of the isocyanosulfide **14c** as a yellow oil: IR (ATR): 3067, 2140 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.51 (ddt, J = 7.6, 1.7, 0.5 Hz, 1H), 7.46-7.42 (m, 2H), 7.34-7.29 (m, 3H), 7.27-7.22 (m, 1H), 7.02 (td, J=7.6, 1.0 Hz, 1H), 6.84 (dd, J=8.2, 1.0 Hz, 1H), 5.16 (dd, J= 9.3, 3.7 Hz, 1H), 3.76 (s, 3H), 3.42 (dd, J = 14.1, 3.7 Hz, 1H), 3.12 (dd, J = 14.1, 9.3 Hz, 1H); ${}^{13}\text{C}\{{}^{1}\text{H}\}$ NMR (100 MHz, CDCl₃) δ 157.7, 155.4, 134.6, 130.7, 130.0, 129.0, 127.0, 126.9, 123.9, 121.1, 110.5, 55.3, 53.7, 40.4; HRMS (+APCI) m/z: [M + H]⁺ Calcd for C₁₆H₁₆NOS 270.0953; Found 270.0949.

Ethyl 3 -((2-([1,1'-biphenyl]4-yl)-2-isocyanoethyl)thio)propanoate(14d) Prepared following the general isocyanoalkene conjugate

addition procedure with 41-isocyanovinyl)1,1'-biphenyl (13a)36 (30.0 mg, 146 µmol), NaH (3.5 mg, 150 µmol), Cu(I)I (0.7 mg, 4 µmol), (S-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin2-yl)-4,5dihydrooxazole (1.0 mg, 3.7 µmol), and eth%-mercaptopropionate (29.4 mg, 219 µmol), for 10 min at rt to afford, after purification (dry-loaded with Celite onto a 4g silica gel cartridge, 10% EtOAc/Hexaneseluent), 35 mg (82%) of the isocyanosulfanylester **14d** as a yellow oil: IR (ATR): 3059, 2138, 1728 cm, ¹H NMR (400 MHz, CDC_b) δ 7.67-7.56 (m, 4H), 7.597.42 (m, 4H), 7.41 7.35 (m, 1H), 4.89 (ddJ= 7.7, 5.8 Hz, 1H), 4.16 (gJ= 7.1 Hz, 2H), 3.09 (dd, J = 14.0, 7.7 Hz, 1H), 3.03 (dd, J = 14.0, 5.8 Hz, 1H), 2.82 (t, J = 7.1 Hz, 2H), 2.59 (tJ = 7.1 Hz, 2H), 1.26 (tJ7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 171.6, 158.6, 141.9, 140.1, 134.9, 128.9, 127.7, 127.7, 127.1, 126.6, 60.8, 58.8, EtOAc/Hexanes eluent), 27 mg (97%) of the isocyanonitrile **14h** as 40.6, 34.9 27.9, 14.2; HRMS (+APCI)m/z: [M - NC] Calcd for C₁₉H₂₁O₂S 313.1262Found313.1272.

(2-([1,1'-biphenyl]-4-yl)-2-isocyanoethyl)(&hloropropyl)sulfane (14e) Prepared following the geneissocyanoalkeneoniugate addition procedure with 4-(1-isocyanovinyl)1,1'-biphenyl (13a)36 (50.0 mg, 244 μmol), NaH (8.8 mg, 365 μmol), Cu(I)I (1.2 mg, 6.1 µmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5dihydrooxazole (1.7 mg, 6.1 µmol), and &hloro-1-propanethiol (29.6 mg, 268 µmol) for 28 min at 8 °C to afford, after purification (dry -loaded with Celite onto a C -2 silica gel, 20 0-3% EtOAc/Hexanes gradient eluent), 59.8 mg (91%) of the isocyano- dition procedure with 4-(1-isocyanovinyl)-1,1'-biphenyl (13a)³⁶ sulfanylchloride 14e as an oil: IR (ATR): 3030, 2137 cm⁻¹; ¹H NMR (400 MHz, CDC₄) δ 7.687.56 (m, 4H), 7.507.43 (m, 4H), 7.41-7.35 (m, 1H), 4.88 (dd, J = 7.6, 5.7 Hz, 1H), 3.63 (t, J = 6.2Hz, 2H), 3.07 (dd,/= 14.0, 7.6 Hz, 1H), 3.01 (dd,= 14.0, 5.7 Hz, 1H), , 2.70 (t, J = 7.0 Hz, 2H), 2.051.96 (m, 2H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 158.7, 142.0, 140.1, 134.8, 128.9, 127.8, 127.7, 127.1, 126.7, 58.9, 43.2, 43.4, 32.1, 30.0; HRMS (+APCI) m/z: [M + H]⁺ Calcd for G₈H₁₉NCIS 316.09**Z**; Found316.0926.

(2-([1,1'-Biphenyl]-4-yl)-2-isocyanoethyl)(&promopropyl)sulfane (14f). Prepared following the generatiocyanoalkeneonjugate addition procedure with 4 -(1-isocyanovinyl)1,1'-biphenyl (13a)³⁶ (30.0 mg, 146 µmol), NaH (8.8 mg, 365 pll); Cu(I)I (0.7 mg, 3.7 µmol), (S-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2yl)-4,5-dihydrooxazole (1.0 mg, 3.7 µmol), and Bromopropane 1-thiol (24.9 mg, 161 µmol), 30 min at 8 °C to afford, after purification (dry-loaded with Celite onto a C -2 silica gel,²⁰ 10% EtOAc/Hexanes eluent), 40.1 mg (76%) of the isocyanosulfanylbromide 14f as a colorless oil: IR (ATR): 3030, 2921, 2137 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.677.56 (m, 4H), 7.507.42 (m, 4H), 7.417.35 (m, 1H), 4.87 (dd/= 7.6, 5.8 Hz, 1H), 3.49 (t, J= 6.3 Hz, 2H), 3.07 (dd/= 14.0, 7.6 Hz, 1H), 3.01 (dd/= 14.0, 5.8 Hz, 1H), 2.69 (t_y = 7.0 Hz, 2H), 2.142.04 (m, 2H); 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 158.7, 142.0, 140.1, 134.8, 128.9, 127.8, 127.7, 127.1, 126.7, 58.9, 40.41, 32.2, 31.8, 31.2; HRMS (+APCI) m/z: [M - NC] Calcd for C₁₇H₁₈BrS 333.0313 Found 333.0307.

Diethyl 2 -(2-([1,1'-biphenyl]-4-yl)-2-isocyanoethyl-2-methylmalonate (14g) Prepared following the generalocyanoalkeneonjugate addition procedure with 4(1-isocyanovinyl)1,1'-biphenyl (13a)³⁶ (30.0 mg, 146 µmol), Q€O₃ (14.3 mg, 43.9 µmol), Cu(I)I (0.7 mg, 4 µmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin 2-yl)-4,5-dihydrooxazole (1.0 mg, 4 µmol), and diethyl 2 methylmalonate⁴⁰ (38.2 mg, 219 µmol) for 29 h at rt to afford, after purification (dry-loaded with Celite onto a C-2 silica gel column, 20 5% EtOAc/Hexanes eluent), 30 mg (55%) of the isocyanoester 14g as a yellow oil: IR (ATR): 2982, 2136, 1726 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.64-7.56 (m, 4H), 7.51-7.42 (m, 4H), 7.39-7.34 (m, 1H), 5.00 (dd, J=9.2, 3.5 Hz, 1H), 4.30-4.16 (m, 2H), 4.24 (q, 1H)J=7.1 Hz, 2H), 2.51 (dd, J=14.9, 9.2 Hz, 1H), 2.44 (dd, J=14.9, 3.5 Hz, 1H), 1.60 (s, 3H), 1.29 (t, J = 7.1 Hz, 3H), 1.29 (t, J = 7.1

Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) 171.3, 171.1, 158.6, 141.5, 140.3, 137.0, 128.9, 127.7, 127.6, 127.1, 126.3, 61.9, 61.8, 54.8, 52.5, 44.6, 20.6, 14.0, 13.9; HRMS (+APCI) m/z: [M - NC] Calcd for C₂₂H₂₅O₄ 353.1753; Found 353.1751.

2-(2-([1,1'-biphenyl]-4-yl)-2-isocyanoethyl-2-methylmalononitrile (14h). Prepared following the general isocyanoalkene conjugate addition procedure with 4-(1-isocyanovinyl)-1,1'-biphenyl (13a)³⁶ (20.0 mg, 97 μmol), Cs₂CO₃ (9.5 mg, 29 μmol), Cu(I)I (0.5 mg, 2 μmol), (S-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2yl)-4,5-dihydrooxazole (0.7 mg, 2 µmol), and 2-methylmalononitrile⁴¹ (12.0 mg, 146 μmol) for 10 min at rt to afford, after purification (dry-loaded with Celite onto a 4g silica gel cartridge, 10% a white oil: IR (ATR): 3027, 2155, 2136 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.71-7.66 (m, 2H), 7.61-7.57 (m, 2H), 7.52-7.43 (m, 4H), 7.41-7.36 (m, 1H), 5.08 (dd, J= 8.9, 4.9 Hz, 1H), 2.66 (dd, J= 14.6, 8.9 Hz, 1H), 2.44 (dd, J=14.6, 4.9 Hz, 1H), 1.95 (s, 3H); ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) 162.3, 142.9, 139.8, 133.7, 129.0, 128.3, 128.0, 127.2, 126.6, 114.7, 114.6, 55.1, 45.8, 29.7, 25.8; HRMS (+APCI) m/z: [M - NC] Calcd. for C₁₈H₁₅N₂ 259.1235; Found 259.1233.

4-([1,1'-biphenyl]-4-yl)-4-isocyane2,2-diphenylbutanenitrile (14i). Prepared following the general isocyanoalkene conjugate ad-(30.0 mg, 146 μmol), NaH (3.5 mg, 146 μmol), Cu(I)I (0.7 mg, 3.7 μmol), (S-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5dihydrooxazole (1.0 mg, 3.7 µmol), and diphenylacetonitrile (226 mg, 1.17 mmol), 1.5 h at 0 °C to afford, after purification (dryloaded with Celite onto a C-2 silica gel, 20 0-5% EtOAc/Hexanes gradient eluent), 43.5 mg (75%) of the isocyanonitrile **14i** as a colorless oil: IR (ATR): 3031, 2922, 2240, 2137 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.60-7.53 (m, 4H), 7.48-7.32 (m, 15H), 4.83 (dd, *J* = 8.7, 3.9 Hz, 1H), 3.20 (dd, J = 14.4, 8.7 Hz, 1H), 2.91 (dd, J = 14.4, 8.7 Hz, 1H)14.4, 3.9 Hz, 1H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 159.9, 141.9, 140.1, 138.8, 138.7, 136.0, 129.3, 129.2, 128.9, 128.7, 128.5, 127.9, 127.7, 127.1, 127.0, 126.9, 126.5, 120.9, 55.3, 50.0, 47.4; HRMS (+APCI) m/z: $[M + H]^+$ Calcd. for $C_{29}H_{23}N_2$ 399.1861; Found 399.1863.

4-Isocyane4-(2-methoxyphenyl2,2-diphenylbutanenitrile (14i). Prepared following the general isocyanoalkene conjugate addition procedure with 1-(1-isocyanovinyl)-2-methoxybenzene (13b) (20.0 mg, 126 μmol), NaH (3.0 mg, 130 μmol), Cu(I)I (0.6 mg, 3.1 μmol), (*S*)-4-(*tert*-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5dihydrooxazole (0.9 mg, 3.1 µmol), and diphenylacetonitrile (36.4 mg, 188 µmol) for 5 h at 0 °C to afford, after purification (dryloaded with Celite onto a C-2 silica gel,²⁰ 0-5% EtOAc/Hexanes gradient eluent), 40.8 mg (99%) of the isocyanonitrile **14i** as a colorless oil: IR (ATR): 3073, 2925, 2853, 2138 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.47-7.29 (m, 12H), 7.01 (td, J= 7.6, 1.1 Hz, 1H), 6.83 (dd, J= 8.3, 1.1 Hz, 1H), 5.12 (dd, J= 8.9, 3.6 Hz, 1H), 3.77 (s, 3H), 2.96 (dd, J= 14.2, 8.9 Hz, 1H), 2.87 (dd, J= 14.2, 3.6 Hz, 1H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 159.2, 155.2, 139.3, 139.0, 130.0, 129.1, 128.9, 128.4, 128.3, 127.07, 127.06, 126.9, 125.2, 121.0, 120.95, 110.7, 55.2, 50.7, 49.8, 44.8; HRMS (+APCI) m/z: [M - NC] Calcd. for C₂₃H₂₀ON 326.1545; Found 326.1544.

1-(2-([1,1'-biphenyl]-4-yl)-2-isocyanoethyl)IH-imidazole (14k). Prepared following the general isocyanoalkene conjugate addition procedure with 4-(1-isocyanovinyl)-1,1'-biphenyl (13a)³⁶ (30.0 mg, 146 µmol), NaH (3.5 mg, 146 µmol), Cu(I)I (0.7 mg, 3.7 μmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5dihydrooxazole (1.0 mg, 3.7 µmol), and imidazole (79.6 mg, 1.17 mmol) for 5 h at 0 °C to afford, after purification (dry-loaded with Celite onto a 4 g silica gel cartridge, 0-6% MeOH/CH₂Cl₂ gradient eluent), 35.0 mg (88%) of the isocyanoimidazole **14k** as a yellow

oil: IR (ATR): 3111, 3031, 2934, 2139, 1670m1; 1H NMR (400 MHz, CDCl₃) δ 7.66-7.55 (m, 4H), 7.50-7.42 (m, 2H), 7.42-7.36 (m, 1H), 7.34 (t/= 1.2 Hz, 1H), 7.37.27 (m, 2H), 7.07 (t/= 1.2 Hz, 1H), 6.90 (t_y = 1.2 Hz, 1H), 5.02 (t_y = 5.9 Hz, 1H), 4.37 (d, $J = 5.9 \text{ Hz}, 2\text{H};^{13}\text{C}\{^{1}\text{H}\} \text{ NMR (100 MHz, CDC}\}) \delta 161.1, 142.6,$ 139.8, 137.6, 131.9, 130.1, 129.0, 128.0, 127.96, 127.1, 126.5, 119.4, 59.2, 53.1; HRMS (+APCI) m/z: [M + H] + Calcd. for C₁₈H₁₆N₃ 274.1344; Found274.1339.

2-(2-([1,1'-biphenyl]-4-yl)-2-isocyanoethyl)isoindolin4,3-dione (141). Prepared following the general cyanoalkeneonjugate addition procedure with 4-(1-isocyanovinyl)1,1-biphenyl (13a)36 (30.0 mg, 146 µmol), NaH (7.0 mg, 2920ol), Cu(I)I (0.7 mg, 3.7 µmol), (S-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5dihydrooxazole (1.0 mg, 3.7 µmol), and phthamilidine (53.8 mg, 365 µmol) for 97 h at 50 °C to afford, after purification-(dagled with Celite onto a 4 g silicael cartridge, 55% MeOH/CHCl2 gradient eluent), 38.0 mg (75%) of the isocyanophthamilidiaes a yellow solid (mp 138.4139.4 °C): IR (ATR): 3031, 2947, 2139, 1776, 1714 cn⁴; ¹H NMR (400 MHz, CDQ) δ 7.937.88 (m, 2H), 7.80-7.74 (m, 2H), 7.67.62 (m, 2H), 7.647.56 (m, 2H), 7.567.51 (m, 2H), 7.487.42 (m, 2H), 7.497.35 (m, 1H), 5.28 (dd J = 9.7)5.4 Hz, 1H), 4.28 (dd/= 13.9, 9.7 Hz, 1H), 3.91 (dd/= 13.9, 5.4 Hz, 1H); $^{13}C(^{1}H)$ NMR (100 MHz, CDC)) δ 167.7, 159.8, 142.2, 140.1, 134.4, 132.7, 131.7, 128.9, 127.9, 127.8, 127.1, 126.8, 123.8, 56.3, 44.1; HRMS (+APCI) m/z: [M - NC] Calcd. for C₂₂H₁₆O₂N 326.1181; Found326.1179.

General IsocyanocyclopropaneSynthesisProcedura Neat ester (1.1 equiv) wa added to a rt, THF solution (0.10 M)sofcyanoalkene(1.0 equiv), C₂CO₃ (1.2 equiv), Cu(I)I (0.025), and ligand (0.025 equiv). Upon completion, as determined by monitoring by TLC, saturated, aqueous ammonium chloride was added. The phases were separated and then the aqueous phase was extracted 63.9, 163.5, 157.8, 142.2, 140.1, 131.3, 128.9, 128.6, 127.8, with EtOAc (3 x 2 mL). The combined organic extract was washed127.2, 127.1, 83.4, 82.8, 46.0, 44.4, 28.0, 27.3, 22.7; HRMS with water (1 x 10 mL) and brine (1 x 10 mL), dried (Na₂SO₄), filtered, and then concentrated under reduced pressure. The crude 420.2184. product was purified by flash chromatography.

Diethyl 2 -([1,1'-biphenyl]-4-yl)-2-isocyanocyclopropane, 1-dicarboxvlate 16a). Prepared following the gerral isocvanocyclopropane synthesis procedure with 4-(1-isocyanovinyl)1,1'-biphenyl(13a)³⁶ (20.0 mg, 97.4 μmol), QSO₃ (34.9 mg, 107 μmol), Cu(I)I (0.5 mg, 2. 4 µmoI), (S)-4-(tert-butyI)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (0.7 mg, 2.4 µmol), and diethyl bromomalonate (25.6 mg, 107 µmol) for 2 h at rt to afford, after purification (dryloaded with Celite onto 2 silica gel²⁰ 5% EtOAc/Hexanes eluent), 29.0 mg (83%) of the cyanocyclopropane**16a** as a colorless oil: IR (ATR) 2981, 2129, 1736 cm¹H NMR (400 MHz, CDC) δ 7.647.54 (m, 4H), 7.537.34 (m, 5H), 4.40 (q, J = 7.1 Hz, 2H), 3.943.81 (m, 2H), 2.50 (ABq, Δv = 37.6 Hz, J = 6.8, 2H), 1.40 (tJ = 7.1 Hz, 3H), 0.91 (tJ = 7.1 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 164.8, 164.4, 158.6, 142.4, 140.0, 130.7, 128.9, 128.5, 129, 127.3, 127.1, 62.9, 62.2, 46.6, 43.1, 23.3, 14.1, 13.5; HRMS (+APCIn/z: [M + H] + Calcd. for C₂₂H₂₂O₄N 364.1549 Found364.1546.

Dimethyl 2-([1,1'-biphenyl]-4-yl)-2-isocyanocyclopropane, 1-dicarboxylate (16b). Prepared following the generalocyanocyclopropane synthesis procedure with 4-(1-isocyanovinyl)1,1'-biphenyl(13a)³⁶ (30.0 mg, 150 μmol), GCO₃ (57.1 mg, 175 μmol), Cu(I)I (0.7 mg, 3.7 µmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (1.0 mg, 3.7 µmol), and dimethyl bromomalonate (33.9 mg, 161 µmol) for 4 h at rt to afford, after purification (dryloaded with Celite onto **Q** silica gel²⁰ 5% EtOAc/Hexanes eluent), 36.2 mg (74%) of the cyanocyclopropane 16b as a yellow solid (mp 114-815.8 °C): IR (ATR) 2953, 2130, 1740 cn²; ¹H NMR (400 MHz, CDG) δ 7.647.56 (m, 4H),

7.527.34 (m, 5H), 3.94 (s, 1H), 3.42 (s, 1H), 2.53 (ABq, $\Delta v = 34.1$ Hz, J = 6.8 Hz, 2H); ¹³C(¹H) NMR (400 MHz, CDCb) δ 165.2, 164.8, 158.9, 142.5, 139.9, 130.5, 128.9, 128.5, 127.9, 127.3, 127.2, 53.8, 53.1, 46.8, 43.3, 23.7; HRMS (+APIGIA): [M + H]+ Calcd. for C₂₀H₁₈O₄N 336.1236 Found 336.1231 HPLC (Daicel Chiral Technologies Inc. CHIRALPARIE, ethyl acetate/hexaes = 10/90, flow rate = 2.0 mL/min, I = 238 nm) tR = 8.9 min (major), 10.5 min (minor) indicated er = 1.05:1. Meparation of Dimethyl 2-([1,1'-biphenyl]-4-yl)-2-isocyanocyclopropan&,1-dicarboxvlate (16b) on a One Millimole Scale epeated the prepairs of 16b on a one mmol scale with 4(1-isocyanovinyl)1,1'-biphenyl (13a)³⁶ (206.0 mg, 1.00 mmol), C₂CO₃ (392.4 mg, 1.20 mmol), Cu(I)I (4.8 mg, 0.025 mmol), §-4-(tert-butyI)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (6.8 mg, 0.025 mmol), and dimethyl bromomalonate (233.0 mg, 1.10 mmol) for 105 h at rt to afford, after purification (dry-loadedwith Celite onto C-2 silica gel,²⁰ 5% EtOAc/Hexanes eluent), 235.0 mg (70%) of thecyanocyclopropan46b as a yellow solid spectrally identical to material previously isolated.

Di-tert-butyl 2 -([1,1'-biphenyl]-4-yl)-2-isocyanocyclopropane 1,1-dicarboxylate [16c]. Prepared following the generislocyanocyclopropanesynthesis procedure with 4-(1-isocyanovinyl)1,1biphenyl (13a)³⁶ (30.0 mg, 146 μmol), Cs₂CO₃ (57.1 mg, 175 μmol), Cu(I)I (0.7 mg, 3.7 μmol)*S*(-4-(*tert*-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (1.0 mg3.7 µmol), and di-tert-butyl bromomalonate (47.5 mg, 161 µmol) for 2 h at rt to afford, after purification (dry-loaded with Celite onto G2 silica gel,²⁰ 5% EtOAc/Hexanes eluent), 51.4 mg (84%) of istree yanocyclopropane 6cas a yellow oil: IR (ATR) 2981, 2133, 1733 cm ¹; ¹H NMR (400 MHz, CDC₈) δ 7.627.55 (m, 4H), 7.547.43 (m, 4H), 7.407.35 (m, 1H), 2.35 (ABq, $\Delta v = 65.3 \text{ Hz}$, J = 6.7, 2H), 1.60 (s, 9H), 1.11 (s, 9H); ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCl₃) δ (+APCI) m/z: [M + H] + Calcd. for C₂₆H₃₀NO₄ 420.2175 Found

Diethyl 2 -isocyane2-(o-tolyl)cyclopropane1, 1-dicarboxylate (16d). Prepared following the generatocyanocyclopropansynthesis procedure with 1-(1-isocvanovinyl)2-methylbenzen (30.0 mg, 210z µmol), C₂CO₃ (81.9 mg, 251 µmol), Cu(I)I (1.0 ma. 5.2 µmol). (S-4-(tert-butvl)-2-(5-(trifluoromethyl)pyridin-2yl)-4,5-dihydrooxazole (1.4 mg, 5.2 µmol), and diethyl bromomalonate (65.1 mg, 272 µmol) for 44 h at rt to afford, after purification (dry-loaded with Celite onto-2 silica gef,0 0-5% EtOAdHexanes gradient eluent), 38.8 mg (62%) of tliseocyanocyclopropant6d as a yellow oil: IR (ATR) 2926, 2126, 1731 cfp. 1H NMR (400 MHz, CDCl₃) δ 7.31-7.25 (m, 2H), 7.237.16 (m, 2H), 4.464.36 (m, 2H), 3.88 (q, J = 7.1 Hz, 2H), 2.54 (s, 3H), 2.41 (ABq $\Delta v =$ 16.3 Hz, J = 6.4 Hz, 2H), 1.40 $I_{\rm t}$ J = 7.1 Hz, 3H), 0.91 (tJ = 7.1Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDC₈) δ 164.9, 164.6, 157.9 138.8, 131.1, 130.4, 129.8, 129.2, 126.0, 62.8, 62.1, 53.2, 42.0, 24.2, 19.1, 14.1, 13.5; HRMS (+APCIm/z: [M + H] + Calcd. for C₁₇H₂₀NO₄ 302.1393 Found302.1385.

Diethyl 2 -isocyane2-(naphthalen2-yl)cyclopropane1, 1-dicarboxylate (16e). Prepared following the generatocyanocyclopropane synthesis procedure with 2-(1-isocyanovinyl) naphthale new pane synthesis procedure (30.0 mg, 167 µmol), QSO₃ (65.5 mg, 201 µmol), Cu(I)I (0.8 mg, 4.2 µmol), (S-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (1.1 mg, 4.2 µmol), and diethyl bromomalonate (52.0 mg, 218 µmol) for 46 h at rt to afford, after purification (dry-loaded with Celite onto 2 silica gel?0 10% EtOAc/Hexanes eluent), 42.8 mg (76%) of thisocyanocyclopropantes a colorless oil: IR (ATR) 2923, 2129, 1732 cm 1H NMR (400 MHz, CDC₃) δ 7.88-7.81 (m, 4H), 7.697.50 (m, 3H), 4.41 (q, J = 7.2 Hz, 2H), 3.833.68 (m, 2H), 2.58 (ABq, $\Delta v = 70.4$ Hz,= 6.8 Hz,

2H), 1.42 (t, J = 7.1 Hz, 3H), 0.78 (t, J = 7.1 Hz, 3H); ${}^{13}C\{{}^{1}H\}$ NMR (100 MHz, CDCI₃) δ 165.0, 164.5, 162.0, 133.6, 132.8, 129.3, 128.8, 128.4, 127.9, 127.7, 127.3, 127.0, 125.3, 63.0, 62.2, 47.1, 43.2, 23.5, 14.3, 13.6; HRMS (+APCI) m/z: [M + H]+ Calcd. for C₂₀H₂₀O₄N 338.1393; Found 338.1392.

Diethyl 2 -isocyane2-(4-methoxyphenyl)cyclopropanle1-dicarboxylate (16f). Prepared following the generasocyanocyclopropanesynthesis procedure with (1-isocyanovinyl) 4-methoxybenzene⁶ (30.0 mg, 188 μmol), C₆CO₃ (92.1 mg, 283 μmol), Cu(I)I (0.9 mg, 4.7 μmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (1.3 mg, 4.7 µmol), and diethyl bromomalonate (67.6 mg, 283 μmol) for 2.5 h at rt to afford, after pu- ¹⁹F NMR (400 MHz, CDCl) δ -62.9; HRMS (+APCl)m/z: [M + rification (dry-loaded with Celite onto a 4 g silica gel cartridge, 10% EtOAc/Hexanes eluent), 45.0 mg (76%) oisthreyanocyclopropane**16f** as a yellow oil: IR (ATR) 2963, 2129, 1735 cfn¹H NMR (400 MHz, CDC) δ 7.377.31 (m, 2H), 6.956.80 (m, 2H), 4.36 (q, J = 7.1 Hz, 2H), 3.91-3.81 (m, 2H), 3.80 (s, 3H), 2.41(ABq, $\Delta v = 42.7 \text{ Hz} / = 6.7 \text{ Hz}$, 2H), 1.37 (t/= 7.1 Hz, 3H), 0.94 (t, J = 7.1, 3H); ¹³C{¹H} NMR (100 MHz, CDC₈) δ 165.1, 164.6, 14.3, 13.8; HRMS (+APCI)m/z: [M - NC] Calcd. for C₁₆H₁₉O₅ 291.1233 Found291.1235

Diethyl 2 -isocyane2-(2-methoxyphenyl)cyclopropanle1-dicarboxylate (16a). Prepared following the generalocyanocyclopropanesynthesis procedure with (1-isocyanovinyl) 2-methoxybenzene(13b) (30.0 mg, 188 μmol), C₂CO₃ (73.7 mg, 226 μmol), Cu(I)I (0.9 mg, 4.7 µmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (1.3 mg, 4.7 µmol), and diethyl bromomalonate (58.6 mg, 245 µmol) for 19 h at rt to afford, after purification (dryloaded with Celite onto 62 silica gel,20 0-5% EtOAc/Hexanes gradient eluent), 48.0 mg (80%) ofisbeyanocyclopropan46g as a yellow oil: IR (ATR) 2982, 2129, 1726 cm¹; ¹H NMR (400 MHz, CDC_b) δ 7.36 (td, J = 8, 1.7 Hz, 1H), 7.30 (dd, J = 7.6, 1.7 Hz, 1H), 6.94 (t/= 7.6 Hz, 1H), 6.89 (d/= 8 Hz, 1H), 4.40 (q/= 7.1 Hz, 2H), 3.98.78 (m, 2H), 3.88 (s, 3H), 2.33 (ABq, $\Delta v = 35.9$ Hz/= 6.4 Hz, 2H), 1.41 (t/= 7.1 Hz, 3H), 0.96 (t,J= 7.1 Hz, 3H),¹³C(¹H) NMR (100 MHz, CDC) δ 165.4, 165.2, 158.2, 131.2, 130.1, 121.1, 120.6, 120.4, 111.0, 62.5, 61.7 55.5, 43.9, 41.4, 24.7, 14.2, 13.6; HRMS (+APOnlyz: [M + H] + Calcd. for G₇H₂₀NO₅ 318.1342 Found318.1331.

Diethyl 2 -(4-chlorophenyl-)2-isocyanocyclopropane, 1-dicarboxylate (16h). Following the general procedure for the synthesis of isocyanocycloproparsewith 1-chloro-4-(1-isocyanovinyl)benzene³⁶ (42.0 mg, 257 μmol), GCO₃ (100.7 mg, **3**8 μmol), Cu(I)I (1.2 mg, 6.4 µmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (1.7 mg, 6.4 µmol), and diethyl bromomalonate (67.5 mg, 282 µmol) for 19 h at rt to afford, after purification (dry-loaded with Celite onto C -2 silica gel. 20 0-5% EtOAc/Hexanes gradient eluent), 57.0 mg (70%) potyanocyclopropane16h as a colorless oil: IR (ATR) 2984, 2129, 1730 cm ¹H NMR (400 MHz, CDC_δ) δ 7.407.33 (m, 4H), 4.38 (q,/= 7.1) Hz, 1H), 4.37 (q,J= 7.1 Hz, 1H), 3.959.82 (m, 2H), 2.43 (ABq, $\Delta v = 27.0$, J = 6.8 Hz, 2H), 1.38 (t/= 7.1 Hz, 3H), 0.96 (t/= 7.1Hz, 3H); $^{13}C\{^{1}H\}$ NMR (100 MHz, CDC $\}$) δ 164.6, 164.2, 159.0, 135.7, 130.4, 129.5, 128.9, 62.9, 62.3, 46.1, 42.9, 23.3, 14.1, 13.6 ished procedure. 42 An ethanolic solution (0.2 M) of N-((1 H-imid-HRMS (+APCI)m/z: [M + H]+ Calcd. for G₆H₁₇NO₄CI 322.0846 Found322.0838.

Diethyl 2-isocyane3-propyl-2-(4-(trifluoromethyl)phenyl)cyclopropane1,1-dicarboxylate (16i). Prepared following the general isocyanocyclopropansynthesis procedure with 2)-1-(1-isocyanopenŧ1-en-1-yl)-4-(trifluoromethyl)benzene **15a** 20.0 mg, 83 μmol), C₂CO₃ (32.7 mg, 100 μmol), Cu(I)I (0.4 mg, 2 μmol)β-(

4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-dihydrooxazole (0.6 mg, 2 µmol), and diethyl bromomalonate 12µL, 130 µmol) for 20 h at 50 °C to afford, after purification (4g silica gel cartridge, 10% EtOAc/Hexanes eluent), 29 mg (86%) confyanocyclopropane16i as a yellow oil: IR (ATR): 2964, 2876, 2126, 1735 cm¹; ¹H NMR (400 MHz, CDC₈) δ 7.60 (dd, J = 33.9, 8.1 Hz, 4H), 4.37 (q,/= 7.1 Hz, 2H), 3.9-3.84 (m, 2H), 2.6-2.57 (m, 1H), 2.051.90 (m, 1H), 1.74.61 (m, 3H), 1.38 (t/= 7.1 Hz, 3H), 1.08 (t, J = 7.1 Hz, 3H), 0.98 (t, J = 7.1 Hz, 3H); ${}^{13}C\{{}^{1}H\}$ NMR $(100 \text{ MHz}, \text{CDC}_b) \delta 164.9, 163.3, 161.1, 137.0 (dJ = 1.6 \text{ Hz}),$ 131.5 (q_1^2J = 32.5 Hz), 128.9, 125.7 (q_1^2J = 3.8 Hz), 123.6 (q_1^2J = 273.7 Hz), 62.5, 62.3, 49.4, 44.7, 32.1, 27.1, 21.6 14.0, 13.8, 13.6; H]⁺ Calcd C₂₀H₂₃F₃NO₄ 398.1579 Found398.1579.

Diethyl (E) -2-isocyane2-styrylcyclopropane1, 1-dicarboxylate (16j). Prepared following the general socyanocyclopropansynthesis procedure with E1-(3-isocyanobutal, 3-dien-1-yl) benzen & (7.0 mg, 45 μmol), C₂CO₃ (17.3 mg, 54.0 μmol), Cu(I)I (0.2 mg, 1 µmol), (S)-4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin-2-yl)-4,5-160.5, 158.1, 129,724.0, 114.1, 62.9, 62.2, 55.5, 43.0, 26.1, 23.5, dihydrooxazole (0.3 mg, 1 μmol), and diethyl bromomalonate (80 μL, 50 μmol) for 25 min at rt to afford, after purification (4g silica gel cartridge, 5% EtOAc/Hexanes eluent), 10 mg (711/kg)ccfyanocyclopropan 46j as a colorless oil: IR (ATR3031, 2983, 2132, 1731 cm¹; ¹H NMR (400 MHz, CDC₈) δ 7.417.27 (m, 5H), 6.94 (d, J=15.7 Hz, 1H), 6.10 (d/=15.7 Hz, 1H), 4.394.31 (m, 2H),4.23 (q, J = 7.1 Hz, 2H), 2.26 (ABq, $\Delta v = 121.8$ Hz, J = 6.6 Hz, 2H), 1.36 (t, J = 7.1 Hz, 3H), 1.26 (t, J = 7.1 Hz, 3H); ${}^{13}C\{{}^{1}H\}$ NMR (400 MHz, CDCl₃) δ 165.5, 164.7, 159.8, 135.2, 133.8, 128.8, 128.5, 126.7, 119.7, 77.2, 62.8, 62.6, 42.7, 26.5, 14.1, 14.0; HRMS (+APCI) m/z: [M + H] + Calcd. for C₁₈H₂₀O₄N 314.1393; Found314.1391.

> Diethyl 3-([1,1'-biphenyl]-4-yl)-3-isocyane5-methylenecyclohexane 1, 1-dicarboxylate 16k). Prepared following the generatoryanocyclopropanesynthesis procedure with 4-(1-isocyanovinyl) 1,1'-biphenyl (**13a**)³⁶ (20.0 mg, 97.4 µmol), C₂CO₃ (38 mg, 120 μmol), Cu(I)I (0.5 mg, 2 μmol), (S)4-(tert-butyl)-2-(5-(trifluoromethyl)pyridin2-yl)-4,5-dihydrooxazole (0.7 mg, p2mol), and di-,ethyl 2(2-(chloromethyl)allyl)malonate(36.4 mg, 146 µmol), 24 h at room temperature to afford, after purification (4g silica gel cartridge, 20%EtOAc/Hexanes eluent), 36 mg (88%) ofistoecyanocyclopropane 6k as a yellow oil: IR (ATR): 3036, 2128, 1731 cm ¹; ¹H NMR (400 MHz, CDC₃) δ 7.687.58 (m, 6H), 7.507.44 (m, 2H), 7.417.36 (m, 1H), 5.29 (qJ= 1.6 Hz, 1H), 5.13 (qJ= 1.7 Hz, 1H), 4.394.32 (m, 1H), 4.291.14 (m, 3H), 3.32 (dt/= 13.7, 2 Hz, 1H), 2.94 (dt, J = 14.7, 2 Hz, 1H), 2.722.56 (m, 3H), 2.30 (dq, J = 13.7, 2 Hz, 1H), 1.32 (t, J = 7.1 Hz, 3H), 1.26 (t, J = 7.1 Hz, 3H)Hz, 3H); ¹³C{¹H} NMR (400 MHz, CDC_b) δ 171.0, 169.1, 159.8, 141.4, 140.1, 139.5, 137.4, 128.9, 127.7, 127.5, 127.1, 125.2, 116.8, 64.7, 62.1, 61.8, 54.4, 47.3, 40.7, 37.9, 14.0, 13.8; HRMS (+APCI) m/z: [M + H]⁺ Calcd. for C₂₆H₂₈O₄N 418.2019; Found 418.2022.

Dimethyl (E)-2-([1,1'-biphenyl]-4-yl)-2-((7-benzyl7,8-dihydroimidazo[1,5a]pyrazin-5(6H)-ylidene)amino)cyclopropant, 1-dicarboxylate 22): Prepared following a modification of the pubazol-5-yl)methyl)-1-phenylmethanamine (12 mg, 60 μmol), paraformaldehyde (1.8 mg, 50 µmol), and isocyanide **16b** (20 mg, 60 umol) were stirred at rt. After 90 h, the reaction mixture was concentrated under reduced pressure and then purified (silica gel column 2 cm x 10 cm, 3% MeOH/CH2Cl2 eluent) to afford 26 mg (82%) of the benzoimadazopyrazine 22 as a yellow oil: IR (ATR): 3029, 2952, 2822, 1731 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ 8.06-8.01 (m, 2H), 7.73-7.68 (m, 2H), 7.67-7.62 (m, 2H), 7.51-7.44 (m, 2H), 7.43-7.32 (m, 5H), 7.32-7.26 (m, 1H), 7.12 (d, J = 1.0 Hz, 1H), 6.72 (d, J=1.0 Hz, 1H), 4.25 (d, J=18.2 Hz, 1H), 3.92 (d, J=18.2

= 13.3 Hz, 1H), 3.76 (d/= 14.3 Hz, 1H), 3.728.63 (m, 3H), 3.69 (s, 3H), 3.48 (ABq, Δv = 70.7, J= 12.6 Hz, 2H), 3.46 (s, 3H); $^{13}C\{^{1}H\}$ NMR (400 MHz, CDCl₃): δ 173.3, 170.1, 168.0, 145.1, 140.0, 137.3, 133.1, 131.0, 129.0, 128.95, 128.4, 128.1, 127.4, 127.2, 126.5, 122.9, 92.2, 66.9, 61.0, 57.9, 53.2, 52.8, 47.5, 44.9; HRMS (+APCI) m/z: [M + H]⁺ Calcd. for $C_{32}H_{31}N_{4}O_{4}$ 535.2346; Found 535.2337.

ASSOCIATED CONTENT

Supporting Information

¹H and ¹³C spectra and an HPLC trace of racemate **16a** are provided in the supporting information.

The Supporting Information is available free of charge on the ACS Publications website.

FAIR Data is available as Supporting Information for publication which includes the primary NMR FID files for compounds: 13b, 15, 17, 14a, i, ii, 14b, 14c, 14d, 14e, 14f, 14g, 14h, 14i, 14j, 14k, 14l, 16a, 16b, 16c, 16d, 16e, 16f, 16g, 16h, 16i, 16j, 16k, and 22.

AUTHOR INFORMATION

Corresponding Author

- (1) Suginome, M.; Ito, Y. Product Class 7: Isocyanides and Related Compounds in Science of Synthesis, Vol. 19 (Ed.:S.-I. Murahashi), Thieme Stuttgart, **2004**, pp. 445-530.
- (2) (a) Zhu, J.; Wang, Q.; Wang, M.-X. Multicomponent Reactions in Organic Synthesis, Wiley-VCH, Weinheim, **2014**; (b) Dömling, A. Recent Developments in Isocyanide Based Multicomponent Reactions in Applied Chemistry. *Chem. Rev.* **2006**, *106*, 17-89; c) Zhu, J. Recent Developments in the Isonitrile-Based Multicomponent Synthesis of Heterocycles. *Eur. J. Org. Chem.* **2003**, 1133-1144.
- (3) Zhang, B.; Studer, A. Recent advances in the synthesis of nitrogen heterocycles via radical cascade reactions using isonitriles as radical acceptors. *Chem. Soc. Rev.* 2015, *44*, 3505-3521.
- (4) Chakrabarty, S.; Choudhary, S.; Doshi, A.; Liu, F.-Q.; Mohan, R.; Ravindra, M. P.; Shah, D.; Yang, X.; Fleming, F. F. Catalytic Isonitrile Insertions and Condensations Initiated by RNC–X Complexation. *Adv. Synth. Catal.* **2014**, *356*, 2135-2196.
- (5) Giustiniano, M.; Basso, A.; Mercalli, V.; Massarotti, A.; Novellino, E.; Tron, G. C.; Zhu, J. To each his own: isonitriles for all flavors. Functionalized isocyanides as valuable tools in organic synthesis. *Chem. Soc. Rev.* **2017**, *46*, 1295-1357.
- (6) Knapp, J. M.; Kurth, M. J.; Shaw, J. T.; Younai, A. Strategic applications of multicomponent reactions in diversity-oriented synthesis in Diversity-Oriented Synthesis: Basics and Applications in Organic Synthesis. Drug Discovery, and Chemical Biology, First Edition. Edited by Andrea Trabocchi, **2013** John Wiley & Sons, Inc. Published 2013 by John Wiley & Sons, Inc.
- (7) Slobbe, P.; Ruijter, E.; Orru, R. V. A. Recent applications of multicomponent reactions in medicinal chemistry. *MedChemComm*. **2012**, *3*, 1189–1218.
- (8) (a) Chepyshev, S. V.; Lujan-Montelongo, J. A.; Chao, A.; Fleming, F. F. Alkenyl Isocyanide Conjugate Additions: A Rapid Route to γ-Carbolines. *Angew. Chem., Int. Ed.* **2017**, *56*, 4310-4313. (b) Bloom, J. D.; DiGrandi, M. J.; Dushin, R. G.; Curran, K. J.; Ross, A. A.; Norton, E. B.; Terefenko, E.; Jones, T. R.; Feld, B.; Langy, S. A. Thiourea inhibitors of herpes viruses. part 1: bis-(aryl)thiourea inhibitors of CMV. *Bioorg. Med. Chem. Lett.* **2003**, *13*, 2929–2932. (c) Helal, C. J.; Lucas, J. C. A Concise and Regioselective Synthesis of I-Alkyl-4-imidazolecarboxylates. *Org. Lett.* **2002**, *4*, 4133-4134. (d) King, R. B.; Efraty, A. Addition of phosphorus–hydrogen and arsenic–hydrogen bonds to vinyl isocyanide. *J. Chem. Soc., Perkin Trans.* **1974**, *1*, 1371-1373.

*Fraser F. Fleming – Department of Chemistry, Drexel University, 3401 Chestnut St., Philadelphia, PA 19104-2875, USA; Email: fleming@drexel.edu

Authors

John-Paul R. Marrazzo – Department of Chemistry, Drexel University, 32 S. 32nd Street, Philadelphia, PA 19104-2875, USA **Allen Chao** – Abzena, 360 George Patterson Blvd, Bristol, PA 19007, USA

Yajun Li – Fujian Institute of Research on the Structure of Matter, University of Chinese Academy of Sciences, 155 Yangqiao Road West, Fuzhou, Fujian, 350002, China

Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

ACKNOWLEDGMENT

Financial support for this research from NSF (1953128) is gratefully acknowledged. HRMS analysis conducted by Timothy P. Wade, Andrew Greene, and Hannah Palmer are gratefully acknowledged.

REFERENCES

- (9) (a) Szczepaniak, G.; Ruszczyńska, A.; Kosiński, K.; Bulska, E.; Grela, K. Highly efficient and time economical purification of olefin metathesis products from metal residues using an isocyanide scavenger. *Green Chem.* **2018**, *20*, 1280-1289. (b) Galan, B. R.; Kalbarczyk, K. P.; Szczepankiewicz, S.; Keister, J. B.; Diver, S. T. A Rapid and Simple Cleanup Procedure for Metathesis Reactions. *Org. Lett.* **2007**, *9*, 1203-1206.
- (10) For some of the few transition metal-catalyzed reactions of isocyanides see: (a) Kazmaier, U.; Ackermann, S. An Improved Protocol for the Synthesis of Allylated Isonitriles. *Synlett.* **2004**, 2576-2578. (b) Ito, Y.; Sawamura, M.; Matsuoka, M.; Matsumoto, Y.; Hayashi, T. Palladium-catalyzed allylation of α -isocyanocarboxylates. *Tetrahedron Lett.* **1987**, *28*, 4849-4852.
- (11) Lazar, M.; Angelici, R. J. in Modern Surface Organometallic Chemistry (Eds.: J.-M. Basset, R. Psaro, D. Roberto, R. Ugo), Wiley-VCH, Weinheim, **2009**, chap. 13, pp. 513-556.
- (12) Aumann, R.; Kuckert, E.; Krueger, C.; Angermund, K. Nitrogen Heterocycles from Carbene Complexes and Alkenyl Isocyanides. *Angew. Chem. Int. Ed.* **1987**, *26*, 563-564.
- (13) MeNO₂/ K_2 CO₃, diethyl malonate/NaH, malononitrile/ K_2 CO₃, (E)-2-(benzylideneamino)acetic acid/Et₃N, PhLi, and Et₂Zn.
- (14) The palladium complexes Pd(Ph₃P)₄, Pd₂dba₃, [PdCl(allyl)]₂ gave complex reaction mixtures with minimal addition.
- (15) (a) Nenajdenko, V. G.; Gulevich, A. V.; Sokolova, N. V.; Mironov, A. V.; Balenkova, E. S.; Chiral Isocyanoazides: Efficient Bifunctional Reagents for Bioconjugation. *Eur. J. Org. Chem.* **2010**, *8*, 1445-1449. (b) Knol, D.; Koole, N. J.; de Bie, M. J. A. A 13 C NMR Investigation of some Tetrakisisocyanocopper(1) Tetrafluoroborate Complexes: Chemical Shift of, and Coupling to, the Isocyano Carbon. *Organ. Magnet. Res.* **1976**, *8*, 213-218. (c) Saegusa, T.; Murase, I.; Ito, Y. Synthetic Reactions by Complex Catalysts. XXVI. The Copper-catalyzed Reaction of α,β-Unsaturated Isocyanide with Active Methylene Compounds. *Bull. Chem. Soc. Jpn.* **1972**, *45*, 1884-1888.
- (16) (a) Toth, A.; Floriani, C.; Chiesi-Villa, A.; Guastini, C. Structurally diversified neutral copper(I) isocyanide complexes: monoand bi-nuclear complexes from the reaction of copper(I) halides with p-tolyl isocyanide. *J. Chem. Soc. Dalton Trans.* 1988, 1599-1605. (b) Cotton, F. A.; Zingales, F. The Donor-Acceptor Properties of Isonitriles as Estimated by Infrared Study. *J. Am. Chem. Soc.* 1961, 83, 351-355.

- (17) Several copper complexes were examined as potential catalysts for the thiophenol addition: CuI, CuBr, CuCN, Cu₂O, (CuOTf)₂.toluene, copper 3-methylsalicylate, and copper thiophene-carboxylate.
- (18) (a) Yang, G.; Zhang, W. Renaissance of pyridine-oxazolines as chiral ligands for asymmetric catalysis. *Chem. Soc. Rev.* **2018**, *47*, 1783-1810. (b) Werner, E. W.; Mei, T.-S.; Burckle, A. J.; Sigman, M. S. Enantioselective Heck Arylations of Acyclic Alkenyl Alcohols Using a Redox-Relay Strategy. *Science* **2012**, *338*, 1455-1458.
- (19) McCammant, M. S.; Sigman, M. S. Development and investigation of a site selective palladium-catalyzed 1,4-diffunctionalization of isoprene using pyridine–oxazoline ligands *Chem. Sci.* **2015**, *6*, 1355–1361.
- (20) Chao, A.; Alwedi, E.; Fleming, F. F. Isocyanide Purification: C-2 Silica Cleans Up a Dirty Little Secret. *Synthesis* **2019**, *51*, 2122-2127.
- (21) Addition of chloropropylsulfenylate to **13a** was anticipated to trigger chloride displacement via $S_{\rm Ni}$ of the putative anion. Monitoring the reaction revealed the isocyanide "anion" to have a weak nucle-ophilicity at -78 °C whereas warming caused competitive ejection of sulfenylate rather than cyclization.
- (22) Schölkopf, U.; Meyer, R. Syntheses with 2-metalated isocyanides, XXXVIII. Trialkylmethyl substituted glycines and pyrrole-2, 4-dicarboxylic esters from 2-isocyanoacrylic esters and carbanions. *Liebigs Ann. Chem.* **1977**, 1174-1182.
- (23) Altundas, B.; Marrazzo, J.-P. R.; Fleming, F. F. Metalated isocyanides: formation, structure, and reactivity. *Org. Biomol. Chem.* **2020**, *18*, 6467-6482.
- (24) Znabet, A.; Polak, M. M.; Janssen, E.; de Kanter, F. J. J.; Turner, N. J.; Orrua, R. V. A.; Ruijter, E. A highly efficient synthesis of telaprevir by strategic use of biocatalysis and multicomponent reactions *Chem. Commun.* 2010, **46**, 7918–7920.
- (25) (a) Miyamura, S.; Itami, K.; Yamaguchi, J. Syntheses of Biologically Active 2-Aryleyclopropylamines. *Synthesis* **2017**, *49*, 1131–1149. (b) Talele, T. T. J. The "Cyclopropyl Fragment" is a Versatile Player that Frequently Appears in Preclinical/Clinical Drug Molecules. *Med. Chem.* **2016**, *59*, 8712-8756. (c) Chen, H.; Li, Y.; Sheng, C.; Lv, Z.; Dong, G.; Wang, T.; Liu, J.; Zhang, M.; Li, L.; Zhang, T.; Geng, D.; Niu, C.; Li, K. Design and Synthesis of Cyclopropylamide Analogues of Combretastatin-A4 as Novel Microtubule-Stabilizing Agents. *J. Med. Chem.* **2013**, *56*, 685–699.
- (26) Ota, Y.; Miyamura, S.; Araki, M.; Itoh, Y.; Yasuda, S.; Masuda, M.; Taniguchi, T.; Sowa, Y.; Sakai, T.; Itami, K.; Yamaguchi, J.; Suzuki T. Design, synthesis and evaluation of γ-turn mimetics as LSD1-selective inhibitors. *Bioorg. Med. Chem.* **2018**, *26*, 775-785.
- (27) CuI was the most efficient catalyst among CuBr, CuCl, Cu(3-methylsalicylate), CuTc, CuCN, and Cu(OTf)₂ or other metal salts [AgBF₄, AgOTf, ZnBr₂, Yt(OTf)₃, Sc(OTf)₃, and LaCl₃]. CF₃Pyox was the best ligand among a series of diamine, bisphosphine, nacnac, and IMES ligands.
- (28) THF provided higher yields than MTBE, Et₂O, and PhMe whereas no reaction was observed in CH₂Cl₂, DMF, DMSO, 2-MeTHF, 1,4-dioxane, MeNO₂, C₆H₆, CCl₄, or MeCN.
- (29) NaH and Cs₂CO₃ were significantly more effective than Et₃N, t-BuOK, K₂HPO₄, or KHCO₃.
- (30) Attempts to perform the conjugate addition to isocyanoalkenes with aliphatic substituents have not been successful.
- (31) Lu, Y.-Q.; Li, C.-J. Novel [3+2] annulation via a trimethylenemethane zwitterion equivalent in water. *Tetrahedron Lett.* **1996**, *37*, 471-474.

- (32) Control experiments with diethyl bromomalonate, 13a, and catalytic Cu(I) in the absence of ligand L1 afforded 16a, but significantly less efficiently. Careful analysis revealed that the initial Cu(I)I suspension slowly formed a pale-yellow solution implying that a soluble copper isocyanide complex was the true catalytic species. Hong, X.; Tan, Q.; Liu, B.; Xu, B. Isocyanide-Induced Activation of Copper Sulfate: Direct Access to Functionalized Heteroarene Sulfonic Esters. *Angew. Chem. Int. Ed.* 2017, 56, 3961-3965.
- (33) TEMPO (2.0 equiv) suppressed the conjugate addition forming a complex reaction mixture. The inconclusive result reflects the high reactivity of isocyanoalkenes toward radicals: Lenoir, I.; Smith, M. L. Vinyl isonitriles in radical cascade reactions: formation of cyclopenta-fused pyridines and pyrazines. *J. Chem. Soc. Perkin Trans. I* **2000**, 641-643.
- (34) (a) Wang, H.-X.; Li, W.-P.; Zhang, M.-M.; Xie, M.-S.; Qu, G.-R.; Guo, H.-M. Synthesis of chiral pyrimidine-substituted diester D–A cyclopropanes via asymmetric cyclopropanation of phenyliodonium ylides. *Chem. Commun.* **2020**, *56*, 11649-11652. (b) Hao, E.-J.; Fu, D.-D.; Wang, D.-C.; Zhang, T.; Qu, G.-R.; Li, G.-X.; Lan, Y.; Guo, H.-M. Chemoselective asymmetric dearomative [3 + 2] cycloaddition reactions of purines with aminocyclopropanes. *Org. Chem. Front.* **2019**, *6*, 863–867.
- (35) (a) Dalling, A. G.; Yamauchi, T.; McCreanor, N. G.; Cox, L.; Bower, J. F. Carbonylative C–C Bond Activation of Electron-Poor Cyclopropanes: Rhodium-Catalyzed (3+1+2) Cycloadditions of Cyclopropylamides. *Angew. Chem. Int. Ed.* **2019**, *58*, 221 –225. (b) Zhang, M.-C.; Wang, D.-C.; Xie, M.-S.; Qu, G.-R.; Guo, H.-M.; You, S.-L. Cu-catalyzed Asymmetric Dearomative [3+2] Cycloaddition Reaction of Benzazoles with Aminocyclopropanes. *Chem.* **2019**, *5*, 156–167. (c) Garve, L. K. B.; Kreft, A.; Jones, P. G.; Werz, D. B. Synthesis of 2-Unsubstituted Pyrrolidines and Piperidines from Donor–Acceptor Cyclopropanes and Cyclobutanes: 1,3,5-Triazinanes as Surrogates for Formylimines. *J. Org. Chem.* **2017**, *82*, 9235–9242.
- (36) Li, Y.; Fleming, F. F. Direct Conversion of Nitriles to Alkene "Isonitriles" *Angew. Chem. Int. Ed.* **2016**, 55, 47, 14770-14773.
- (37) Lu, Y-Q.; Chao-Jun, L. Novel [3+2] Annulation via a Trimethylenemethane Zwitterion Equivalent in Water *Tetrahedron Lett.* **1996**, 37, 471-474.
- (38) Ramesh, R.; Reddy, D. S. Zinc mediated allylations of chlorosilanes promoted by ultrasound: Synthesis of novel constrained sila amino acids *Org. Biomol. Chem.***2014**, *12*, 4093-4097.
- (39) Perez-Fuertes, Y.; Kelly, A. M.; Johnson, A. L.; Arimori, S.; Bull, S. D.; James, T. D. Simple Protocol for NMR Analysis of the Enantiomeric Purity of Primary Amines *Org. Lett.* **2006**, *8*, 609-612.
- (40) Ogawa, K.; Sasaki, M.; Nozaki, T. Malonic ester and acetoacetic ester synthesis of 2-[11,14C]methyl-fatty acids *Appl. Radiat. Isot.* **1997**, *48*, 623-630.
- (41) Ghorai, M. K.; Talukdar, R.; Tiwari, D. P. A Route to Highly Functionalized β-Enaminoesters via a Domino Ring-Opening Cyclization/Decarboxylative Tautomerization Sequence of Donor-Acceptor Cyclopropanes with Substituted Malononitriles *Org. Lett.* **2014**, *16*, 2204-2207.
- (42) Galli, U.; Hysenlika, R.; Meneghetti, F.; Grosso, E. D.; Pelliccia, S.; Novellino, E.; Giustiniano, M.; Tron, G. C. Exploiting the Nucleophilicity of the Nitrogen Atom of Imidazoles: One-Pot Three-Component Synthesis of Imidazo-Pyrazines *Molecules* **2019**, *24*, 1959.