

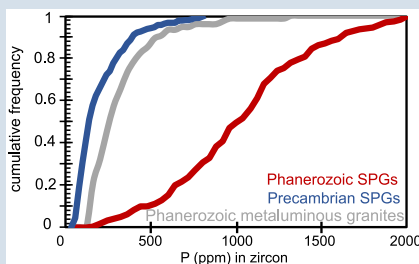
Secular variability in zircon phosphorus concentrations prevents simple petrogenetic classification

C.E. Bucholz^{1*}, J. Liebmann², C.J. Spencer³



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Abstract



Phosphorus (P) concentrations in zircon have been used to discriminate their derivation from metaluminous *versus* strongly peraluminous granites (SPGs) based on the empirical observation of lower P concentrations in zircon from Phanerozoic metaluminous *versus* peraluminous granites. Higher P concentrations in zircon from Phanerozoic SPGs reflect enhanced apatite solubility in peraluminous melts and overall higher P concentrations in peraluminous granites. However, SPGs derived from partial melting of Precambrian sedimentary rocks have lower P concentrations compared to Phanerozoic metaluminous granites, reflecting lower P concentrations in Precambrian *versus* Phanerozoic sedimentary sources. We demonstrate that zircons from Precambrian SPGs also have lower P concentrations compared to Phanerozoic counterparts, likely reflecting lower P concentrations in their parental melts. Applying the P-in-zircon proxy to the detrital zircon record does not effectively discriminate between metaluminous and peraluminous sources and underestimates contributions from peraluminous granites. Although detrital zircons are an important early Earth archive, a uniformitarian perspective cannot always be applied when using trace element proxies developed on Phanerozoic samples.

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Introduction

Trace element concentrations in zircon are used to “fingerprint” the magma from which they crystallised (e.g., Grimes *et al.*, 2015). Of particular interest is identification of zircon from “I-type” or metaluminous granites *versus* those from “S-type” or strongly peraluminous granites (SPGs¹; Burnham and Berry, 2017; Trail *et al.*, 2017). Understanding the relative contribution of zircon from metaluminous *versus* peraluminous granites throughout Earth’s history yields information on lithologic diversity and tectonic environments through time. SPGs dominantly form in collisional orogenic environments through the partial melting of sedimentary rocks (Nabelek, 2020). Thus, their inferred presence (or absence) through time yields information on both tectonic regimes and the availability of sedimentary rocks to be recycled into magmas during orogenic cycles.

One proposed proxy to discriminate zircon from metaluminous and peraluminous granites is phosphorus (P) concentrations in zircon (Burnham and Berry, 2017). As the solubility of apatite increases with melt aluminium saturation index (Pichavant *et al.*, 1992; Wolf and London, 1994), P concentrates in peraluminous melts during differentiation due to the lack of apatite precipitation (Bea *et al.*, 1994), a trend contrary to that observed in metaluminous melts (Lee and Bachmann, 2014).

Indeed, Phanerozoic SPGs have significantly higher P₂O₅ (on average ~0.25 wt. % and up to ~1.5 wt. % P₂O₅; Bea *et al.*, 1992; Bucholz, 2022; Fig. 1a,b) than metaluminous granites (which generally have <0.20 wt. % at bulk rock SiO₂ > 65 wt. %). Further, upon apatite saturation, P concentrations are buffered at higher levels in peraluminous than metaluminous melts.

Consequently, assuming equivalent zircon-melt P partition coefficients for metaluminous and peraluminous granitic systems, zircons from SPGs should have higher P contents than those from metaluminous granitic melts (Burnham and Berry, 2017). Indeed, P in zircons from Phanerozoic peraluminous *versus* metaluminous granites have higher P concentrations (means of 1017 ± 53 *versus* 312 ± 48 ppm (2 s.e.); Burnham and Berry, 2017; Zhu *et al.*, 2020; Fig. 1c,d). Further, P and REE + Y concentrations are strongly correlated in zircon from Palaeozoic SPGs, suggesting the incorporation of P, REE, and Y into the zircon lattice *via* the coupled xenotime-type mechanism [(Y, REE)³⁺ + P⁵⁺ = Zr⁴⁺ + Si⁴⁺]. Based on these observations it has been proposed that P > 750 ppm and (REE + Y) < 1.15*P are robust criteria for identifying zircon from peraluminous granites (Burnham and Berry, 2017).

However, P₂O₅ concentrations in SPGs are not elevated above those of metaluminous granites throughout Earth history

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1 We prefer the use of strongly peraluminous granite over “S-type” granites as the former is purely a geochemical definition that avoids confusion associated with the “alphabet” granite classification scheme (c.f., Frost *et al.*, 2001). However, we emphasise that the SPGs discussed here have all been previously demonstrated to have formed *via* the partial melting of metasedimentary rocks.



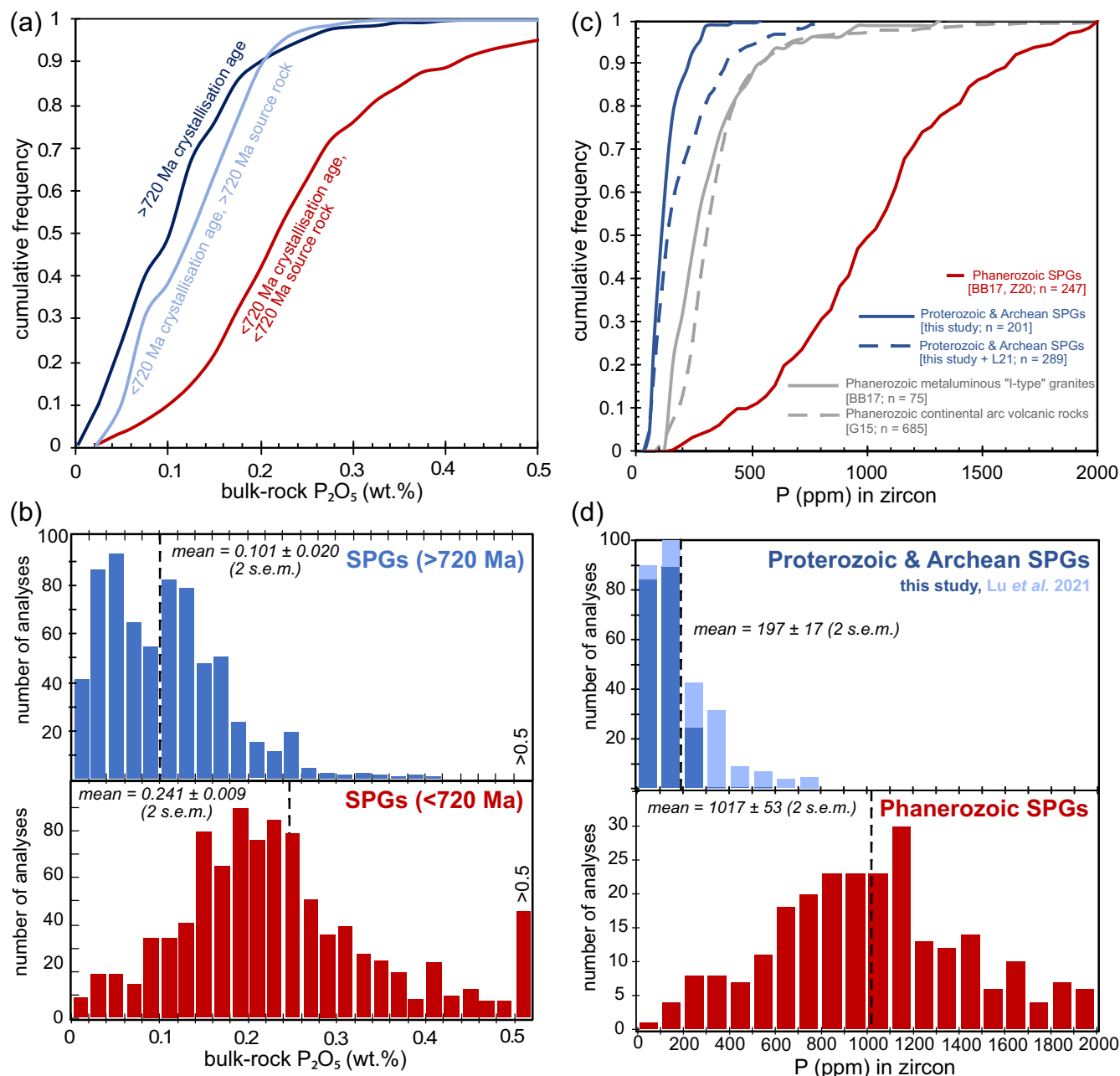


Figure 1 (a) Cumulate frequency distributions of SPG bulk rock P_2O_5 . (b) Histograms of SPG bulk rock P_2O_5 with source rock and crystallisation ages >720 Ma (top) and <720 Ma (bottom). Data in (a) and (b) from Bucholz (2022). (c) Cumulative frequency distributions of P in zircon for Proterozoic and Archean SPGs, as well as Phanerozoic SPGs, "I-type" granites, and continental arc volcanic rocks. (d) Histograms of zircon P concentrations for Proterozoic/Archean SPGs and Phanerozoic SPGs. Phanerozoic SPG and I-type granite data in (c) and (d) are from Burnham and Berry (2017) and Zhu et al. (2020) and continental arc volcanic zircon from Grimes et al. (2015).

(Bucholz, 2022). Maximum and average P_2O_5 concentrations in SPGs derived from sedimentary rocks deposited after 720 Ma are elevated above those derived from older source rocks (Fig. 1a,b). This temporal variation mirrors a similar increase in P concentrations in marine siliciclastic sediments deposited after 720 Ma, related to increased atmospheric and marine O_2 levels and enhanced authigenic P burial (Reinhard et al., 2017). After considering the factors affecting P concentrations in SPGs, Bucholz (2022) concluded that the temporal change in P concentrations is most likely explained as reflecting higher P concentrations in younger sedimentary source rocks. Elevated P concentrations in SPGs derived from sedimentary rocks deposited after 720 Ma could reflect both higher melt P concentrations

and/or inherited P-rich restitic minerals (e.g., apatite, monazite, or P-bearing silicates).

If P concentrations in Archean and most Proterozoic SPGs are relatively low and if this, to some degree, represents lower melt P concentrations, zircon crystallising from such melts may record these low P concentrations. To test this hypothesis, we analysed trace element concentrations (including P) in zircon from Meso- to Paleoproterozoic and Archean SPGs.

Studied Samples and Methods

We analysed trace elements in zircon (n = 201 analyses) from 16 previously characterised SPGs (Liebmann et al., 2021a,b) from

the southwestern USA (ca. 1.4 Ga), the North China Craton (ca. 1.9 and 2.5 Ga), Finland (ca. 1.8–1.9 Ga), and Ghana (ca. 2.2 Ga) (see [Tables S-1 and S-2](#) for locality, lithologic, and bulk rock major elements). The average bulk rock P_2O_5 contents of these SPGs are 0.088 ± 0.036 wt. % (2 s.e.). U-Pb isotopes and trace elements (Ca, P, Y, Hf, and REEs) were collected simultaneously using split stream laser ablation inductively coupled plasma mass spectrometry at UC Santa Barbara. Full analytical details are given in the [Supplementary Information \(SI\)](#) and sample and standard analyses are given in [Tables S-4 and S-5](#), respectively. Cathodoluminescence imaging and U-Pb ages were used to ensure analysis of magmatic zircon (*i.e.* related to the crystallisation of the SPG) and not inherited cores/grains ([Fig. S-1](#)). Any analyses of inherited cores (with older U-Pb ages) or with >5 % discordance of U-Pb ages were excluded. No apatite inclusions were observed. We are aware of only one other locality of >720 Ma SPGs where trace elements in magmatic zircon were analysed ([Lu et al., 2021](#)) and include these data in our discussion.

Results

The P concentrations in Archean and Proterozoic SPGs are positively skewed (average = 132 ± 10 (2 s.e.) ppm; median = 110 ppm; and including [Lu et al. \(2021\)](#) data, average = 196 ± 17 ppm; median = 138 ppm). In comparison, P concentrations in zircon from Phanerozoic SPGs are normally distributed with an average of 1017 ± 53 (2 s.e.) ([Fig. 1c,d](#)). Cumulative frequency distributions of P concentrations highlight this difference ([Fig. 1c](#)) and the mean P concentration in zircon from Precambrian and Phanerozoic SPGs are statistically distinguishable ($P < 0.00001$, Wilcoxon rank sum test). Zircon REE + Y concentrations from Precambrian SPGs, although weakly correlated with P ($R^2 = 0.63$), are defined by a slope of >1.15, distinct from the Phanerozoic SPG trend ([Fig. 2](#)). Neither P nor total REE concentrations correlate with Ca concentrations, suggesting that P concentrations are not reflecting the incorporation of apatite inclusions in the ablation volume.

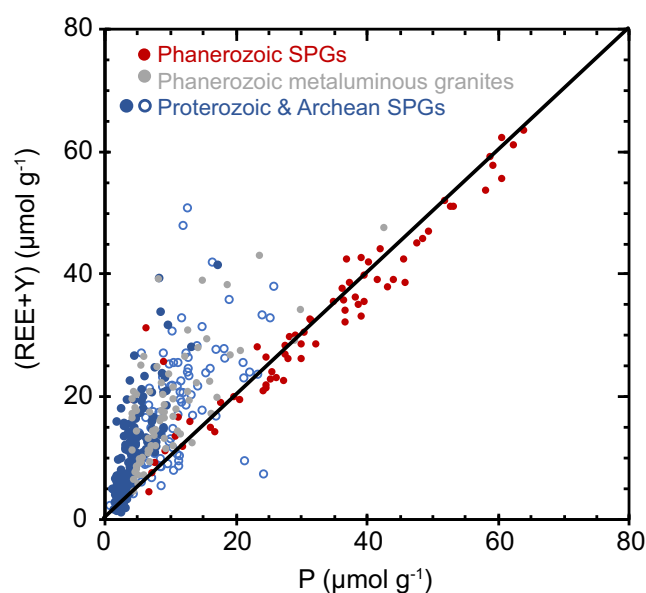


Figure 2 REE + Y versus P (in $\mu\text{mol/g}$) in zircon. Phanerozoic granite data are the same as in [Figure 1](#). Proterozoic and Archean SPGs are from this study (filled symbols) and [Lu et al. \(2021\)](#) (open symbols).

What Do Zircon P Concentrations Represent?

The P concentration of zircon reflects equilibrium and/or disequilibrium processes during crystallisation from melt, including (1) equilibrium partitioning dependent on temperature, pressure, and melt composition, (2) partitioning reflecting melt-crystal boundary layer concentrations controlled by diffusion of P in the melt, or (3) surface controlled non-equilibrium growth (*e.g.*, [Watson and Liang, 1995](#); [Hofmann et al., 2009](#)). In the first scenario, assuming that P partitioning is both Henrian and that equilibrium between zircon and melt occurred, then lower P concentrations in zircon from Archean/Proterozoic *versus* Phanerozoic SPGs could reflect either lower P concentrations in their parental melts or different conditions of crystallisation (*e.g.*, melt composition, pressure, and temperature) which resulted in variable zircon-melt P partition coefficients. Phosphorus partitioning between melt and zircon as a function of melt composition, temperature, and pressure is poorly understood with limited experimental studies available ([Rubatto and Hermann, 2007](#); [Taylor et al., 2015](#)). However, Archean/Proterozoic SPGs are thought to have formed under similar conditions (*e.g.*, pressure, temperature) to those in the Phanerozoic ([Bucholz and Spencer, 2019](#)). Thus, in the first scenario of equilibrium partitioning, the simplest explanation is that lower P concentrations in zircon from Archean/Proterozoic SPGs, as compared to Phanerozoic ones, reflect lower melt P concentrations.

In the second scenario, the growth rate of zircon approaches (or exceeds) the diffusivity of P (and/or REE) in the adjacent melt. For hydrous rhyolites (~6 wt. % H_2O), experimentally calibrated P and Zr diffusivities (D) are similar with $\log(D_P)$ and $\log(D_{Zr})$ between -13.5 to -14.5 m^2/s at 750 – 850 °C (see review of [Zhang and Gan, 2022](#)). However, when zircon growth is sufficiently fast, P due to its general incompatibility is excluded from the zircon and builds up in a melt boundary layer. Thus, the growth of new zircon in equilibrium with this boundary layer acquires higher P concentrations elevated above that predicted by equilibrium partitioning with the bulk (*i.e.* far field) melt. However, all else being equal (*e.g.*, P diffusivity and partition coefficients), melts with higher bulk P concentrations would produce boundary layers with higher P concentrations which would be reflected in higher zircon P concentrations.

Finally, in the third scenario, unequivocal evidence for surface controlled, non-equilibrium growth manifests as sector zoning ([Watson and Liang, 1995](#)), however (sub-) μm scale oscillatory variations P concentrations have also been attributed to non-equilibrium growth ([Hofmann et al., 2009](#)). We generally avoided analysing portions of zircon that exhibited sector zoning ([Fig. S-1](#)). However, when analysed, sector zoned zircon had similar P concentrations to other zircon from the sample (*c.f.*, 18IM15b; [Fig. S-1](#)). Further, the $25 \mu\text{m}$ laser spot used homogenised any visible oscillatory zoning ([Fig. S-1](#)), providing an integrated P concentration of the ablation volume. In zircon where multiple (2–3) spots were analysed, P concentrations agree within 25 ppm for 70 % and within 50 ppm for 80 % of zircon ([Fig. S-3](#)), suggesting that potential local variability induced by non-equilibrium growth was mostly homogenised.

No matter the P incorporation mechanism, our study demonstrates unequivocally that zircons from Archean and Proterozoic SPGs have lower P concentrations on average than their Phanerozoic SPGs ([Figs. 1, 2](#)). A parsimonious explanation for this observation is that lower P concentrations in zircons from Precambrian SPGs reflect lower P concentrations of their granitic parental melts. To explore this idea, we modelled melt P_2O_5 and

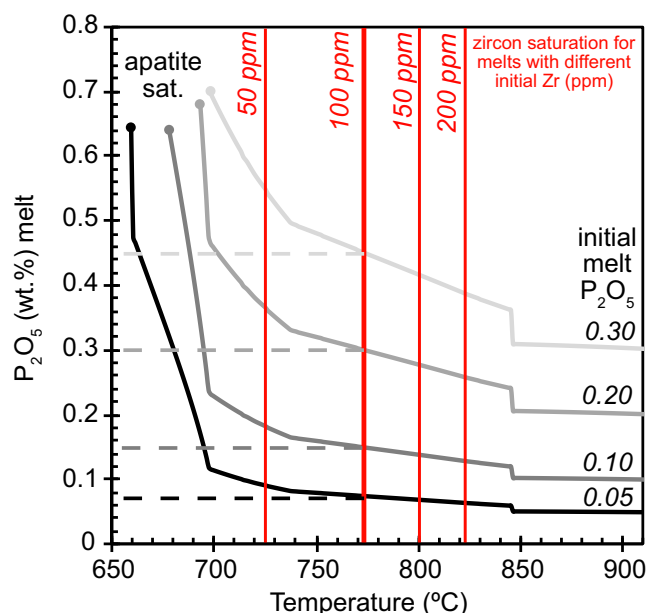


Figure 3 Model results for melt P_2O_5 (wt. %) versus temperature during crystallisation of a SPG melt for different initial melt P_2O_5 (wt. %). Melt trajectories end once apatite saturation is reached. Zircon saturation temperatures are indicated with vertical red lines.

zircon and apatite saturation during crystallisation of a SPG melt using alphaMELTS (Pichavant *et al.*, 1992; Smith and Asimow, 2005; Boehnke *et al.*, 2013). Full modelling details are given in the SI. Calculations were performed for initial melt P_2O_5 contents typical of SPGs across Earth history (0.05, 0.10, 0.20, and 0.30 wt. %; Fig. 1a,b).

Due to compositional proximity to the granitic eutectic, zircon saturation temperatures were primarily controlled by increasing Zr concentrations in the melt with progressive differentiation. For initial Zr concentrations in the melt of 50, 100, 150, and 200 ppm, calculated zircon saturation temperatures are ~727, 772, 800, and 822 °C, respectively. [Average Zr concentrations of pre-720 Ma SPGs from Bucholz (2022) are 120 ± 7 ppm (2 s.e.).] The calculated saturation temperatures agree with calculated Ti-in-zircon temperatures which on average are 810 ± 119 °C and 741 ± 105 °C (2 s.d.) assuming a melt TiO_2 activity of 0.5 and 1, respectively (see SI for calculation details). For 100 ppm Zr in the initial melt, modelled melt P_2O_5 concentrations at zircon saturation are 0.08, 0.15, 0.30, and 0.45 wt. % for initial melt P_2O_5 contents of 0.05, 0.10, 0.20, and 0.30 wt. %, respectively (Fig. 3). Importantly, at zircon saturation, the melt P_2O_5 concentration has not significantly increased from its initial value as the melt fraction is still high (~75 %). For a melt with initially low P_2O_5 (0.05–0.10 wt. %) typical for Precambrian SPGs, melt P_2O_5 increases significantly only within 20–30 °C of the granitic eutectic when the melt fraction decreases dramatically. Calculated melts ultimately reach apatite saturation 70–110 °C below that of the zircon saturation for a melt with 100 ppm initial Zr (Fig. 3). Therefore, although melt P_2O_5 increases due to delayed apatite saturation in these peraluminous melts, it remains low at zircon saturation in SPG melts with initially low P contents.

Implications and Conclusion

As SPGs predominantly form during collisional orogenesis, an accurate understanding of their temporal distribution provides information on the tectonic evolution and crustal recycling

through time (Bucholz and Spencer, 2019; Frost and Da Prat, 2021). Although detrital zircon can help characterise igneous rocks throughout Earth history, a uniformitarian approach cannot always be applied when using trace element proxies developed on Phanerozoic samples. Specifically, this study provides an example of how temporal variations in sedimentary rocks affect our interpretation of the igneous rock (and mineral) record. For example, the use of zircon P concentrations to identify metaluminous *versus* peraluminous source rocks has been applied to Archean and Hadean detrital zircon (Burnham and Berry, 2017; Zhu *et al.*, 2020). In particular, P and the correlation between P and REE + Y in Jack Hills zircon are more similar to Phanerozoic metaluminous granites and have been used to infer that the Jack Hills zircon crystallised from “TTG-like” magmas (Burnham and Berry, 2017). However, the nature of the source rocks for the Jack Hills zircon is debated with elevated $^{18}O/^{16}O$ ratios (Cavosie *et al.*, 2005; Trail *et al.*, 2007) and Al concentrations in Jack Hills zircon (Ackerson *et al.*, 2021), potentially indicating that some Jack Hills zircons are from granites with a sedimentary source. Our results demonstrate that low P contents and $P/(Y + REE)$ values <1 in Precambrian zircons cannot rule out crystallisation from a peraluminous granitic melt. Therefore, low P contents in Jack Hills zircon is consistent with crystallisation from both metaluminous and peraluminous melts, supporting inferences that the Jack Hills zircon could be sourced from a variety of granitic magmas (Bell, 2017; Ackerson *et al.*, 2021).

Similarly, both cumulative distributions of P contents and P correlations with REE + Y in detrital zircons have been used to infer that SPG formation in the Archean was negligible and limited until the Neoproterozoic to Palaeozoic (Zhu *et al.*, 2020). However, we again urge caution about extrapolating these metrics to Precambrian zircon when sedimentary source regions of SPGs had low P concentrations. Further, Phanerozoic SPGs with sedimentary source rocks deposited before 720 Ma also have low P concentrations (Fig. 1a) and should also contain magmatic zircon with low P. However, P concentrations of detrital zircon may be useful in local studies of young terranes to identify contributions from Phanerozoic P-rich SPGs. Identification of detrital zircon from SPGs with Precambrian source rocks is perhaps best done through a combination of O isotopes with new developing zircon proxies such as Al concentrations (Ackerson *et al.*, 2021) or Si isotopes (Trail *et al.*, 2018).

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Additional Information

Supplementary Information accompanies this letter at <https://www.geochemicalperspectivesletters.org/article2240>.



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Supplementary Information

The Supplementary Information includes:

- Analytical Methods
- Ti-in-zircon Thermometry
- Zircon and Apatite Saturation Modelling
- Tables S-1 to S-5
- Figures S-1 to S-4
- Supplementary Information References

Analytical Methods

Sample details including name, location, mineralogy, age, and major element composition are given in Tables S-1 and S-2. Cathodoluminescence imaging was previously done by Liebmann *et al.* (2021a, b) and provided in Figure S-1. Analyses were focused on sites with no obvious inclusions or cracks. U-Pb isotopes and trace elements were collected simultaneously on the same zircon volume using split-stream (SS) laser ablation (LA) inductively coupled plasma mass spectrometry (ICPMS) at the University of California, Santa Barbara (Kylander-Clark *et al.*, 2013). Samples were ablated with a Photon Machines 193 nm excimer laser with a HelEx ablation cell coupled to a Nu Instruments Plasma 3D multicollector ICPMS for U-Pb measurements and an Agilent 7700X quadrupole ICPMS for trace element measurements. One analytical session was undertaken in September 2021. A 25 µm spot size was used. The laser fluence at the sample surface was ~1.5 J/cm². Each analysis consisted of the laser firing twice followed by a 20 s wash out to remove surface contamination. Zircon were then continuously ablated for 60 shots at a 5 Hz repetition rate, yielding a total ablation time of 12 s.

Every 10 unknown analyses were bracketed by standard zircon 91500 (1062.4 ± 0.4 ID-TIMS $^{206}\text{Pb}/^{238}\text{U}$ date; Wiedenbeck *et al.*, 1995), which was used as the primary standard for U–Pb isotopic analyses. NIST612 standard reference glass was measured every 30 analyses and used as a primary standard for trace element analyses. Secondary zircon reference materials analysed throughout the session included GJ-1 (601.7 ± 1.3 ID-TIMS $^{206}\text{Pb}/^{238}\text{U}$ date; Jackson *et al.*, 2004; Kylander-Clark *et al.*, 2013), Plešovice (337.13 ± 0.37 ID-TIMS $^{206}\text{Pb}/^{238}\text{U}$ date; Sláma *et al.*, 2008), OG-1 (3467.1 ± 0.6 Ma $^{206}\text{Pb}/^{238}\text{U}$ date; Stern *et al.*, 2009), FC1 (1099.3 ± 0.7 Ma $^{206}\text{Pb}/^{238}\text{U}$ date; Paces and Miller, 1993), and R33 (419.26 ± 0.39 Ma $^{206}\text{Pb}/^{238}\text{U}$ date; Black *et al.*, 2004). BHVO standard glasses were also analysed at the beginning, end, and throughout the analytical session. We obtained $^{207}\text{Pb}/^{206}\text{Pb}$ -corrected $^{206}\text{Pb}/^{238}\text{U}$ concordia dates of 599.5 ± 28.0 Ma (2 s.d., $n = 13$) for GJ-1, 331.6 ± 9.8 Ma (2 s.d., $n = 7$) for Plešovice, 3462.7 ± 52.9 Ma (2 s.d., $n = 7$) for OG, 1081.4 ± 31.1 Ma (2 s.d., $n = 6$) for FC1, and 418.6 ± 19.4 Ma (2 s.d., $n = 7$) for R33, which are accurate to -0.36 %, -1.64 %, -0.13 %, -1.62 %, and -0.16 % of their reference values, respectively. Standard analyses are given in Table S-5.

For trace-element analyses, ^{90}Zr (assuming ~43.14 wt. % Zr) was used as an internal standard and peaks were measured at ^{27}Al , ^{44}Ca , ^{28}Si , ^{31}P , ^{49}Ti , ^{89}Y , ^{90}Zr , ^{93}Nb , ^{139}La , ^{140}Ce , ^{141}Pr , ^{146}Nd , ^{147}Sm , ^{153}Eu , ^{157}Gd , ^{159}Tb , ^{163}Dy , ^{165}Ho , ^{166}Er , ^{169}Tm , ^{172}Yb , ^{175}Lu , and ^{178}Hf . The zircon standards with the most completely characterised trace element concentrations are 91500 (Wiedenbeck *et al.*, 2004) and GJ-1 (Piazolo *et al.*, 2017). Reference P concentrations in these standards are 24 ± 1 ppm (Jochum *et al.*, 2005) and 30.2 ± 12.8 (Piazolo *et al.*, 2017), which compare favourably with our average of 31.5 ± 4.5 ppm and 35.4 ± 4.5 ppm (2 s.d.). For 91500, measured Y and REE concentrations are within ~10 % of reference values (Wiedenbeck *et al.*, 2004) except for Pr (-52 %), Nd (-17 %), and Lu (-15 %) (Table S-3; Figure S-2).

Data reduction follows that of Garber *et al.*, (2020). Briefly, Iolite plug-in version 2.21 (Paton *et al.*, 2011) for the Wavemetrics Igor Pro software was used to correct measured isotopic ratios and elemental intensities for baselines, laser- and plasma-induced fractionation, and drift. Both errors associated with laser stability throughout each analysis and run, as well as detector counting statistics were considered in the reported uncertainties. Only zircon analyses with >95 % concordance were included in the final data set. U–Th–Pb and trace element data are given in Table S-4.

Ti-in-zircon Thermometry

Temperatures for each zircon analysis were calculated via Ti-in-zircon thermometry (Ferry and Watson, 2007):

$$\log(\text{ppm Ti}) = 5.711 \pm 0.072 - 4800(\pm 86)/T(\text{K}) - \log(a_{\text{SiO}_2}) + \log(a_{\text{TiO}_2})$$

a_{SiO_2} was assumed to be 1 (although it may be below unity during early zircon crystallisation when a granitic melt is not quartz saturated near their liquidus). The Ti-in-zircon expression was evaluated for both a_{TiO_2} values of 1 and 0.5. A recent thermodynamic study by Schiller and Finger (2019) demonstrated that a_{TiO_2} in strongly peraluminous granitic melts is often well below unity near 0.5. Calculated temperatures are given in Table S-4.

Zircon and Apatite Saturation Modelling

We modelled the equilibrium crystallisation of sample CO-17-8, a representative Proterozoic strongly peraluminous granite using alphaMELTS (Smith and Asimow, 2005). The sample modelled composition in wt. % was 71.74 SiO₂, 0.39 TiO₂, 14.61 Al₂O₃, 2.70 FeO(total), 0.50 MgO, 1.26 CaO, 2.60 Na₂O, 5.88 K₂O, 3.00 H₂O. Crystallisation modelling was conducted at 0.3 GPa from 930 to 650 °C at an oxygen fugacity one log unit below the fayalite-magnetite-quartz buffer. P₂O₅ and Zr were not modelled using alphaMELTS, but rather assumed to behave incompatibly until apatite saturation was reached. Although other major silicate phases such as plagioclase and alkali feldspar can be important reservoirs for P in some strongly peraluminous granites (London, 1992), P incorporation is not considered in these phases for simplicity. We modelled 4 different starting P₂O₅ contents of 0.05, 0.1, 0.2, and 0.3 wt. % and assumed a starting concentration of 100 ppm for Zr, in the range of that typical for SPGs (Bucholz and Spencer, 2019). We used the apatite saturation model of Pichavant *et al.* (1992) and the zircon saturation model of Boehnke *et al.* (2013). Although other phosphates (*e.g.*, monazite and xenotime) will influence the P budget of a crystallising peraluminous melt, we do not consider it in our models because its effect (as compared to apatite) is expected to be negligible. Monazite is indeed found in many strongly peraluminous granites and, like xenotime, its solubility is much lower than apatite (generally saturating at several hundredths of a wt. % P₂O₅). However, its solubility is also strongly controlled by the availability of REE (Stepanov *et al.*, 2012) which limits the amount of monazite that can crystallise to trace amounts. Due in part to this fact and that it has a lower P₂O₅ concentration (~29 wt. %) as compared to apatite (~42 wt. %) it does not contribute significantly influence the P budget of a crystallising melt (*e.g.*, Wolf and London, 1995). Alkali feldspars can also incorporate several tenths up to ~1 wt. % P₂O₅ (London, 1992) however the partitioning of P between granitic melts and alkali feldspar is not well enough understood to model quantitatively. alphaMELTS has been shown to model biotite Fe/Mg ratios and saturation temperatures well in strongly peraluminous granites (Bucholz *et al.*, 2018), however it has more difficulty predicting saturation of other aluminous phases typically found in strongly peraluminous granites

(e.g., muscovite and garnet). However, for the purposes of zircon saturation, the “M” factor [or normalised molar $(\text{Na}+\text{K}+2\text{Ca})/(\text{Al}*\text{Si})$] used in calculating saturation temperatures (c.f. Boehnke *et al.*, 2013) does not vary significantly (between 1.25 to 0.94 over the temperature interval of interest (broadly 700-900 °C) (see Fig. S-4) and is primarily a function of increasing ASI during melt differentiation. ASI is expected to increase during differentiation of strongly peraluminous granites (e.g., Bea *et al.*, 1994) and alphaMELTS predicts this well. Similarly, ASI is the primary control on apatite solubility (other than temperature) and our modelling results should reproduce apatite saturation behaviour well.

The vertically zoned Pedrobernardo pluton (central Spain) is a natural example of closed system, *in-situ* differentiation of a strongly peraluminous granitic melt (Bea *et al.*, 1994) and is a useful comparator dataset for our alphaMELTS modelling. Magmatic differentiation in the Pedrobernardo pluton results in increases in bulk-rock SiO_2 from ~71 to 76 wt. % and P_2O_5 from 0.28 to 0.5-0.6 wt. % from biotite+muscovite granites in the lower zone of the pluton to muscovite-bearing aplites and pegmatites of the upper zone. Bulk-rock ASI increases from ~1.18-1.20 to 1.30 and M decreases from 1.16-1.36 to 0.88-0.99. Although the alphaMELTS calculations were undertaken for a different bulk-rock composition and we also recognise that bulk-rock granite chemistry is likely not representative of melt compositions, the broad variations in geochemistry for the Pedrobernardo pluton are similar the calculated melt compositions using alphaMELTS (see Fig. S-4). Thus, we suggest that alphaMELTS models apatite and zircon saturation satisfactorily for the purposes of this study, though a more detailed future study is warranted. Last, we do recognise that the temperatures of apatite and zircon saturation may not be accurate if the alphaMELTS is not predicting the correct temperatures of crystallisation for different mineral assemblages. However, the important point is that zircon saturation will precede that of apatite and for granitic melts with low initial P_2O_5 concentrations and P_2O_5 concentrations will still be low at the onset of zircon saturation.

Supplementary Tables

Table S-1 List of strongly peraluminous granite samples from which zircon was separated and analysed with locality, mineralogy, and magmatic crystallisation ages. References are given for geochronology of samples, bulk-rock geochemistry, and detailed petrologic descriptions.

Sample ID	Locality	Mineralogy	Age $\pm 2\sigma$ (Ma)	Reference
17FIN04A	Turku Area, Finland	Grt+Bt+Afs+Pl+Bt	1850	Liebmann <i>et al.</i> , 2021b
17FIN06A	Turku Area, Finland	Grt+Bt+Afs+Pl+Bt	1815 \pm 16	Liebmann <i>et al.</i> , 2021b
17FIN06B	Turku Area, Finland	Grt+Bt+Afs+Pl+Bt	1815 \pm 16	Liebmann <i>et al.</i> , 2021b
18IM-3	Huaian Complex, North China Craton	Grt+Ms+Qz+Afs+Pl \pm Bt	1917 \pm 70	Liebmann <i>et al.</i> , 2021a
18IM-12B	Jining Complex, North China Craton	Grt+Qz+Afs+Pl	1929 \pm 29	Liebmann <i>et al.</i> , 2021a
18IM-13C	Xiwulanbulang Area, North China Craton	Grt+Bt+Qz+Afs+Pl \pm Ms	2536 \pm 13	Liebmann <i>et al.</i> , 2021a
18IM-15B	Xiwulanbulang Area, North China Craton	Grt+Qz+Afs+Pl	2530 \pm 60	Liebmann <i>et al.</i> , 2021a
18IM-19A	Daqingshan-Wulashan Complex, North China Craton	Grt+Bt+Qz+Afs+Pl	2478 \pm 18	Liebmann <i>et al.</i> , 2021a
18IM-19C	Daqingshan-Wulashan Complex, North China Craton	Grt+Bt+Qz+Afs+Pl	2478 \pm 18	Liebmann <i>et al.</i> , 2021a
18IM-20	Daqingshan-Wulashan Complex, North China Craton	Grt+Bt+Qz+Afs+Pl	2374 \pm 48	Liebmann <i>et al.</i> , 2021a
18IM-21C	Daqingshan-Wulashan Complex, North China Craton	Grt+Qz+Afs+Pl	1855 \pm 16	Liebmann <i>et al.</i> , 2021a
18IM-23D	Daqingshan-Wulashan Complex, North China Craton	Grt+Qz+Afs+Pl	2453 \pm 11	Liebmann <i>et al.</i> , 2021a
18IM-25C	Daqingshan-Wulashan Complex, North China Craton	Ms+Qz+Afs+Pl \pm Bt	2493 \pm 28	Liebmann <i>et al.</i> , 2021a
CO-17-8	Silver Plume granite, Colorado, USA	Ms+Bt+Qz+Afs+Pl	1447 \pm 50	(Bucholz and Spencer, 2019; Liebmann <i>et al.</i> , 2021b)
19GH-9	Baoulé-Mossi domain, West African Craton	Ms+Bt+Qz+Afs+Pl	2183 \pm 12	Liebmann <i>et al.</i> , 2021c
19GH-11B	Baoulé-Mossi domain, West African Craton	Ms+Bt+Qz+Afs+Pl	2188 \pm 20	Liebmann <i>et al.</i> , 2021c

Mineral abbreviations: Qz, quartz; Afs, alkali feldspar; Pl, plagioclase; Grt, garnet; Ms, muscovite, Bt, biotite

Table S-2 Sample bulk-rock major element geochemistry (normalised to 100 %) (originally reported in Liebmann *et al.*, 2021a, b).

Sample	SiO ₂	TiO ₂	Al ₂ O ₃	FeO ^T	MnO	MgO	CaO	Na ₂ O	K ₂ O	P ₂ O ₅	Total
18IM3	78.77	0.38	11.05	2.75	0.05	0.88	1.05	2.26	2.76	0.05	100
18IM21C	83.28	0.04	8.91	2.60	0.18	0.66	1.13	2.23	0.78	0.19	100
17FIN06B	71.40	0.06	15.16	3.53	0.04	0.73	1.49	3.17	4.32	0.10	100
17FIN04A	71.61	0.23	14.70	3.23	0.07	0.96	1.63	3.05	4.44	0.08	100
17FIN06A	66.14	0.67	16.31	6.35	0.05	2.25	1.65	2.98	3.52	0.07	100
19GH11B	74.12	0.08	14.81	0.50	0.01	0.15	1.26	3.41	5.66	0.01	100
19GH9	73.70	0.11	15.77	0.96	0.02	0.28	2.93	5.05	1.15	0.03	100
CO-17-8	71.74	0.39	14.61	2.70	0.03	0.50	1.26	2.60	5.88	0.28	100
18IM-13C	72.39	0.34	15.19	2.82	0.01	0.89	3.11	3.37	1.77	0.10	100
18IM-19	73.14	0.32	14.25	3.20	0.05	1.43	1.29	2.52	3.73	0.06	100
18IM-15B	67.10	0.60	15.92	6.49	0.09	2.18	3.62	2.92	0.99	0.09	100
18IM-25C	75.81	0.06	13.41	0.87	0.00	0.17	0.97	2.76	5.91	0.03	100
18IM-23D	72.83	0.11	14.26	2.58	0.04	0.80	0.61	2.04	6.61	0.12	100
18IM20	64.68	0.27	18.10	5.61	0.07	1.84	2.68	3.57	3.11	0.07	100
18IM12B	70.78	0.14	15.10	5.50	0.13	1.95	2.92	2.38	1.06	0.04	100

Table S-3 Phosphorus values of zircon standards compared to reference values.

zircon standard	P (ppm) reference value	2 s.d.	P (ppm) measured	2 s.d.*	# of analyses
91500	24 ³	1	31.5	4.5	42
GJ-1	30.2 ⁴	12.8	35.4	4.4	13

*propagated error including internal precision and standard deviation of analyses

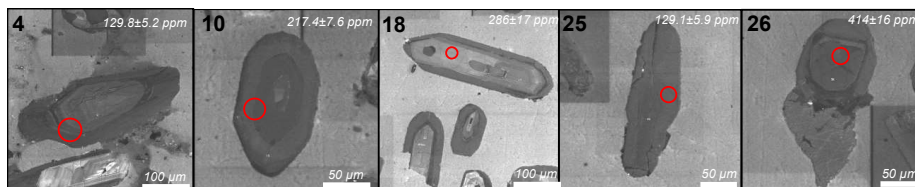
Table S-4 Zircon LA-ICPMS U-Th-Pb isotopes and trace element data. Table S-4 (.xlsx) can be downloaded from the online version of this article at <https://doi.org/10.7185/geochemlet.2240>

Table S-5 Standard LA-ICPMS U-Th-Pb isotopes and trace element data. Table S-5 (.xlsx) can be downloaded from the online version of this article at <https://doi.org/10.7185/geochemlet.2240>

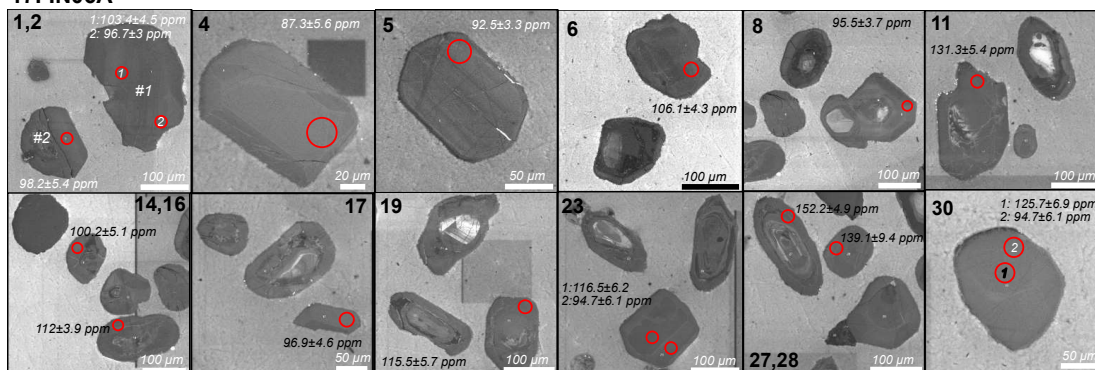
Supplementary Figures

Figure S-1 Cathodoluminescence images of analysed zircon with sample number in the heading and grain number labelled on each image, showing LA-ICPMS spots (red circles) and P concentrations (ppm) listed.

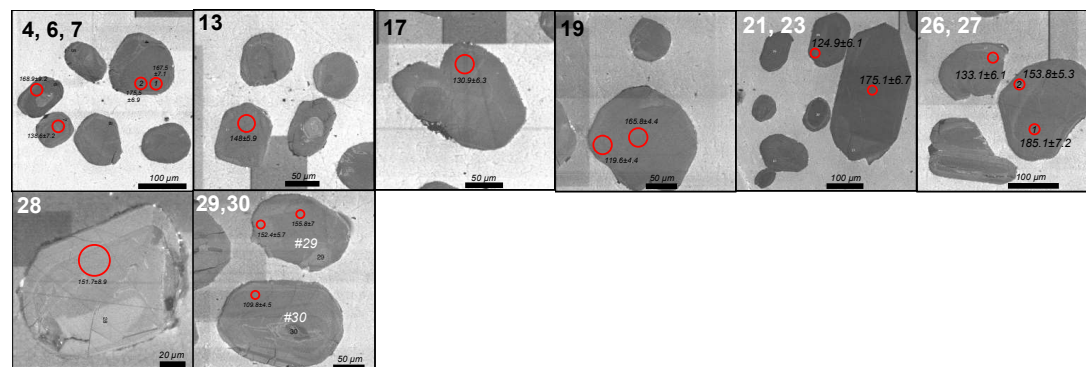
17FIN04A



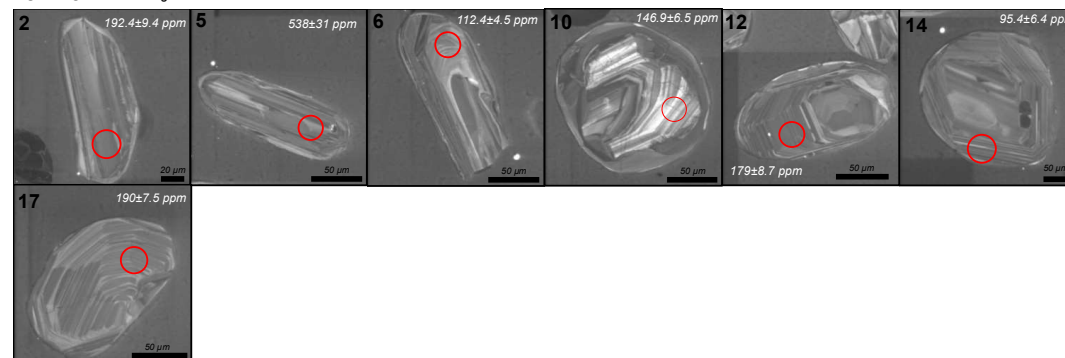
17FIN06A



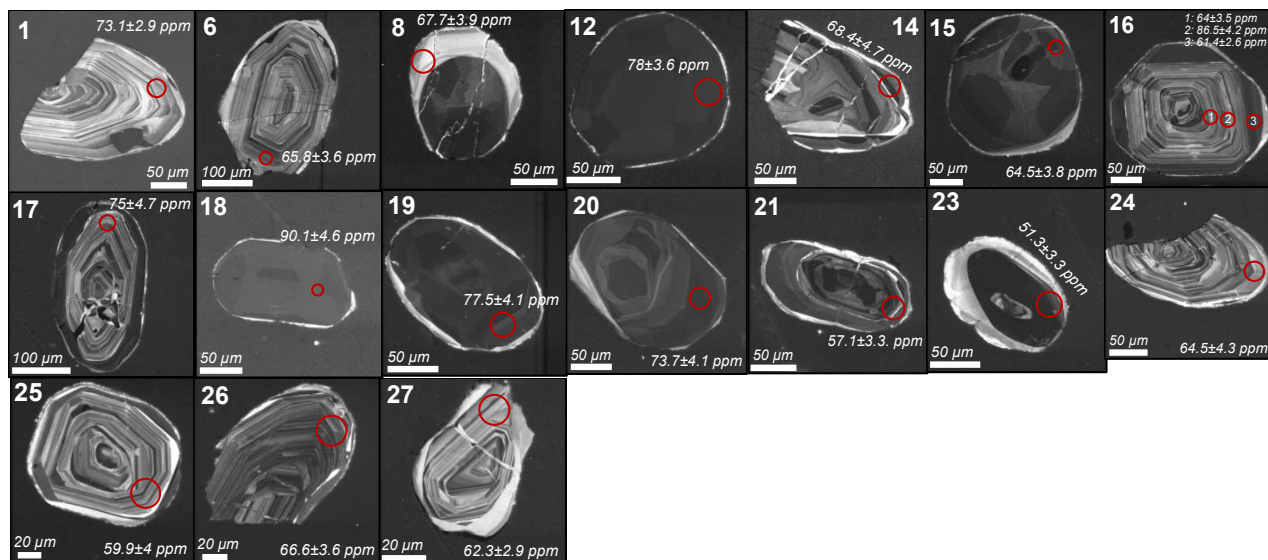
17FIN06B



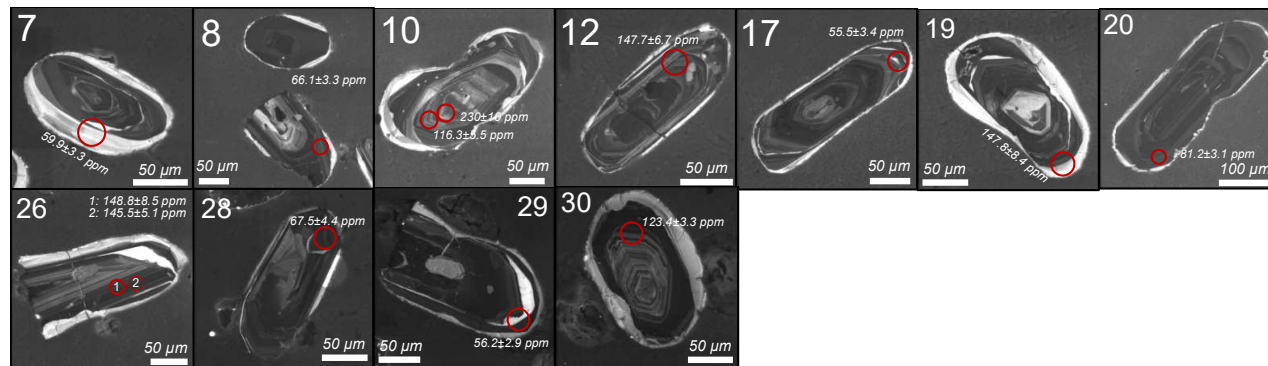
18IM-3 no CL image available for zircon 19



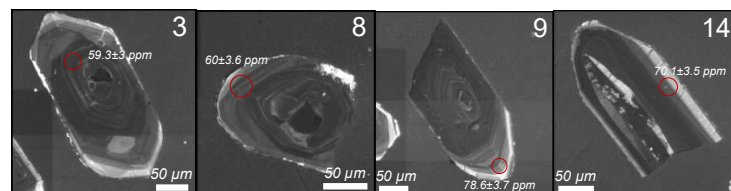
18IM15b



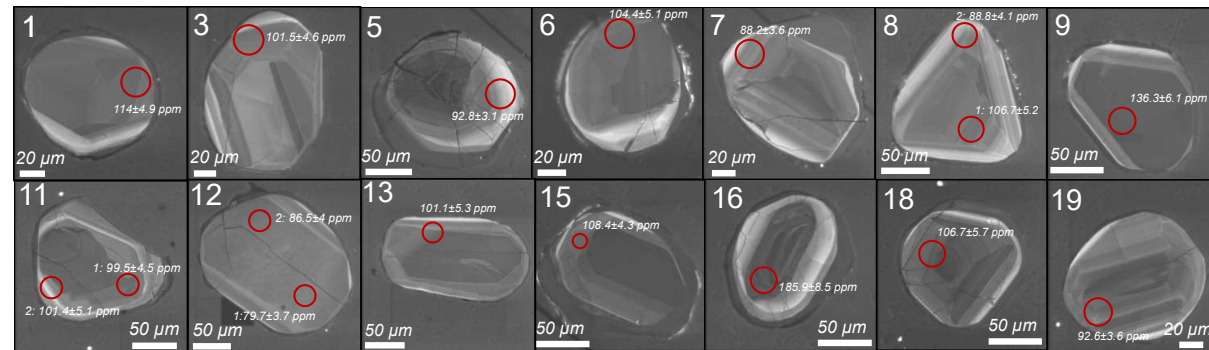
18IM13c



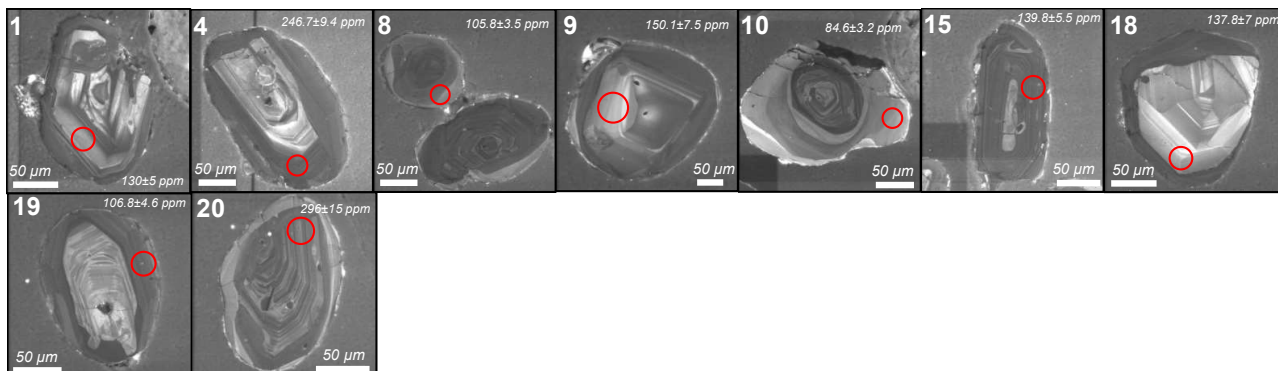
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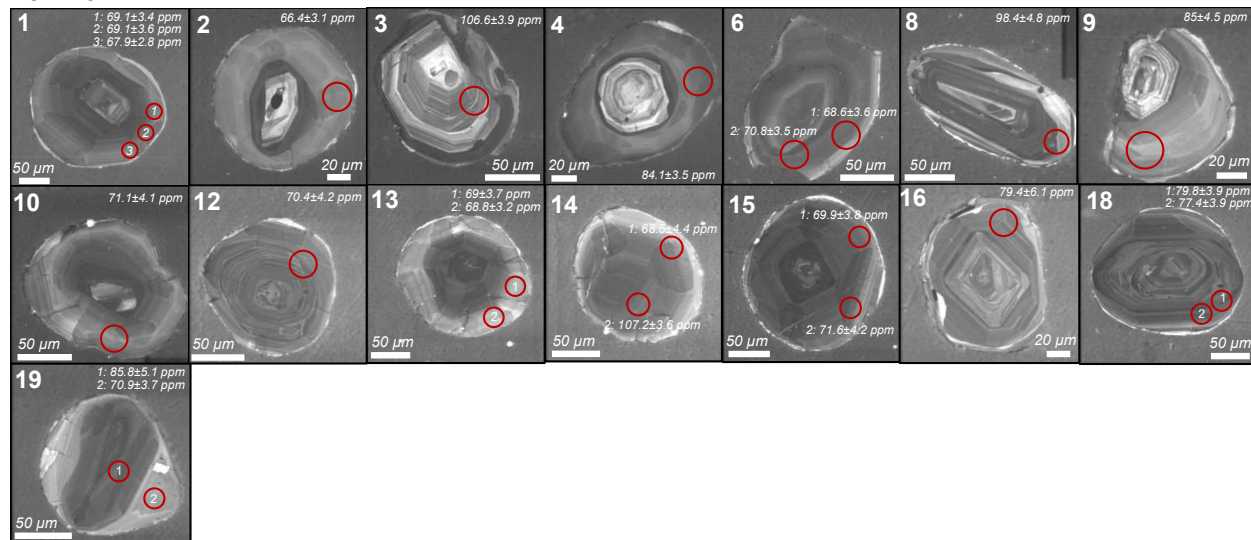
18IM12b



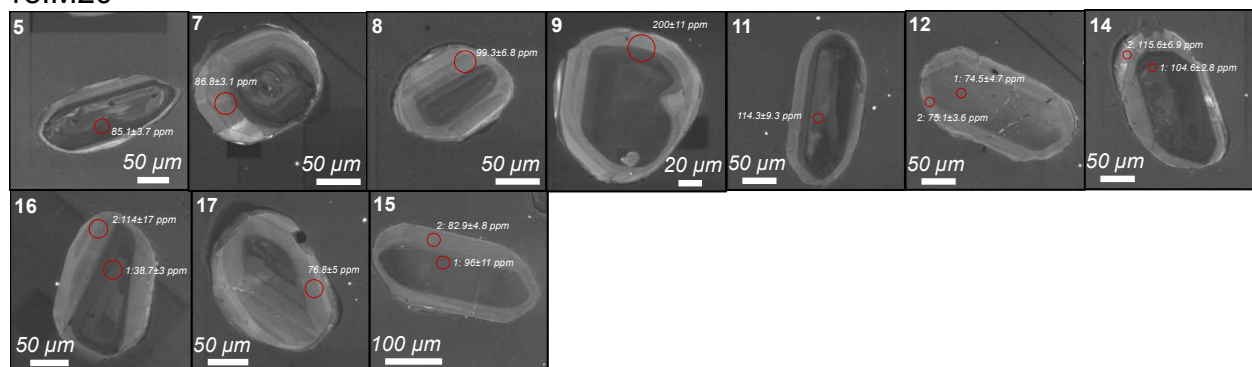
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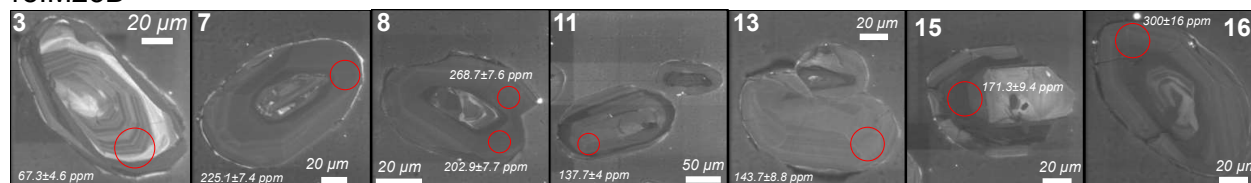
18IM19c



18IM20

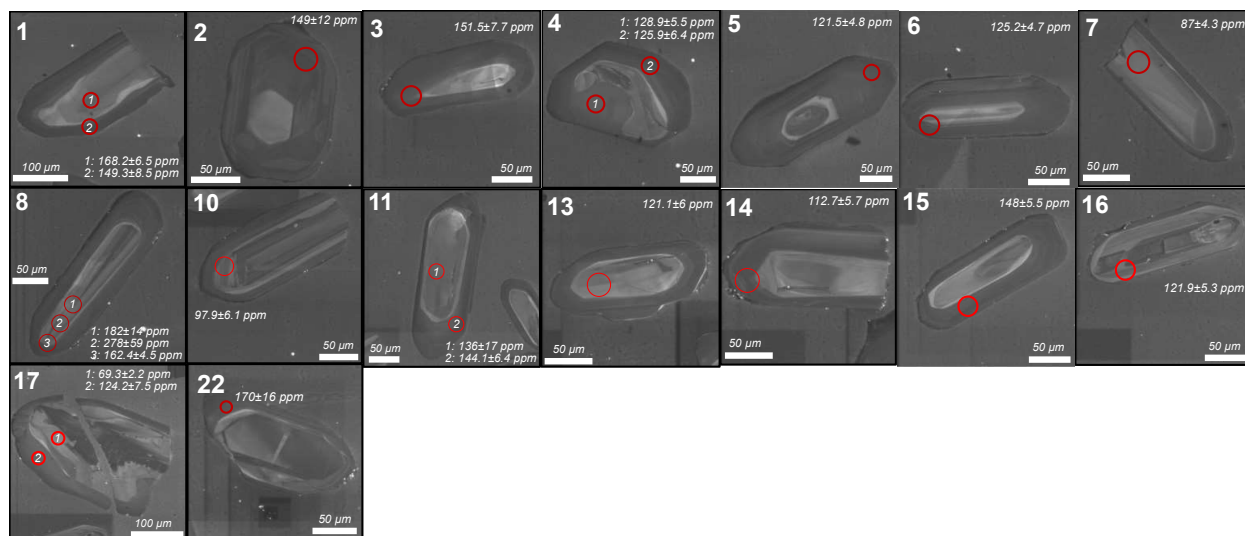


18IM23D

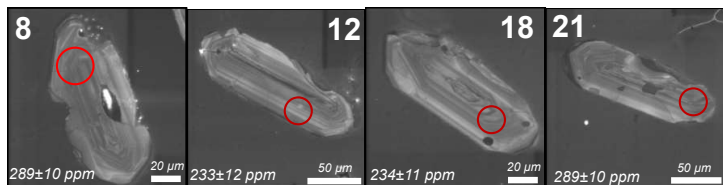


no CL images for 18, 19, 20, 21, 22

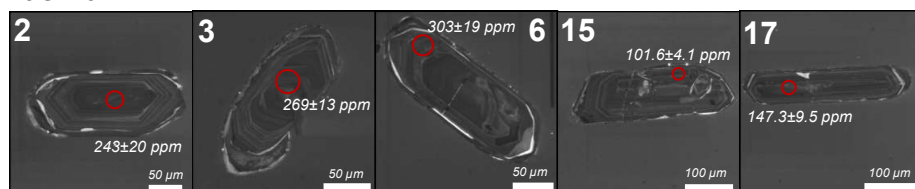
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CO-17-8



19GH-9



19GH-11B

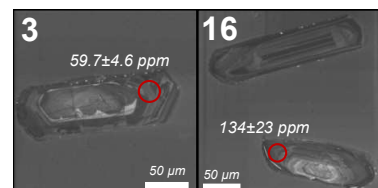


Figure S-2 Measured Y and REE concentrations in reference zircons normalised to reference values. Reference values are from GeoREM (Jochum *et al.*, 2005) for 91500 (**a**) and Piazzolo *et al.* (2017) for GJ-1 (**b**).

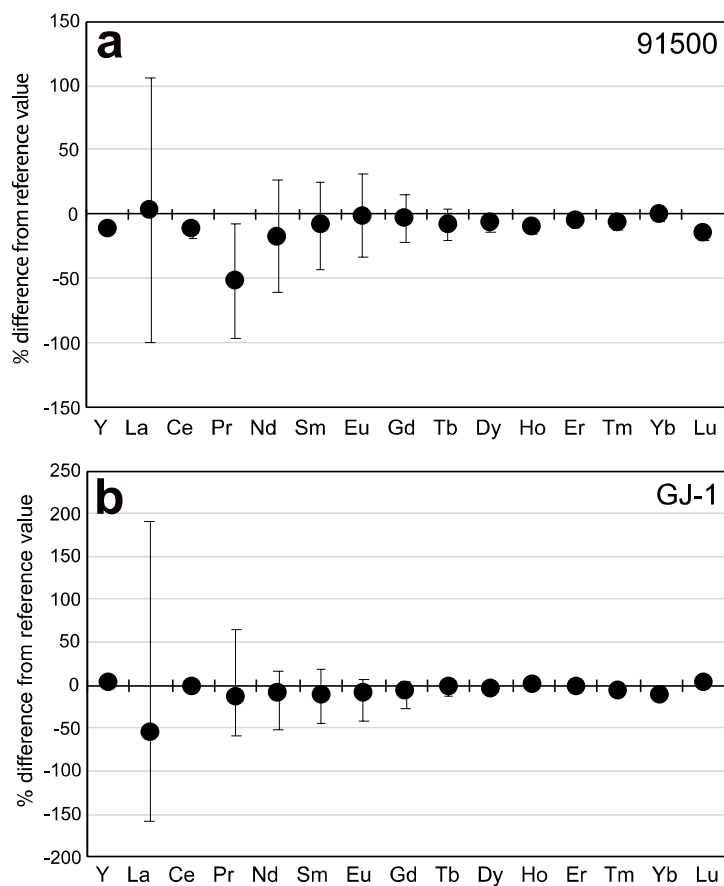


Figure S-3 Comparison of measured intra-zircon phosphorus concentrations for zircon where more than one spot was analysed. Numbers in symbols indicates the analysed grain number. Dashed diagonal lines indicate constant ppm offset (± 25 , 50, and 100 ppm) for reference.

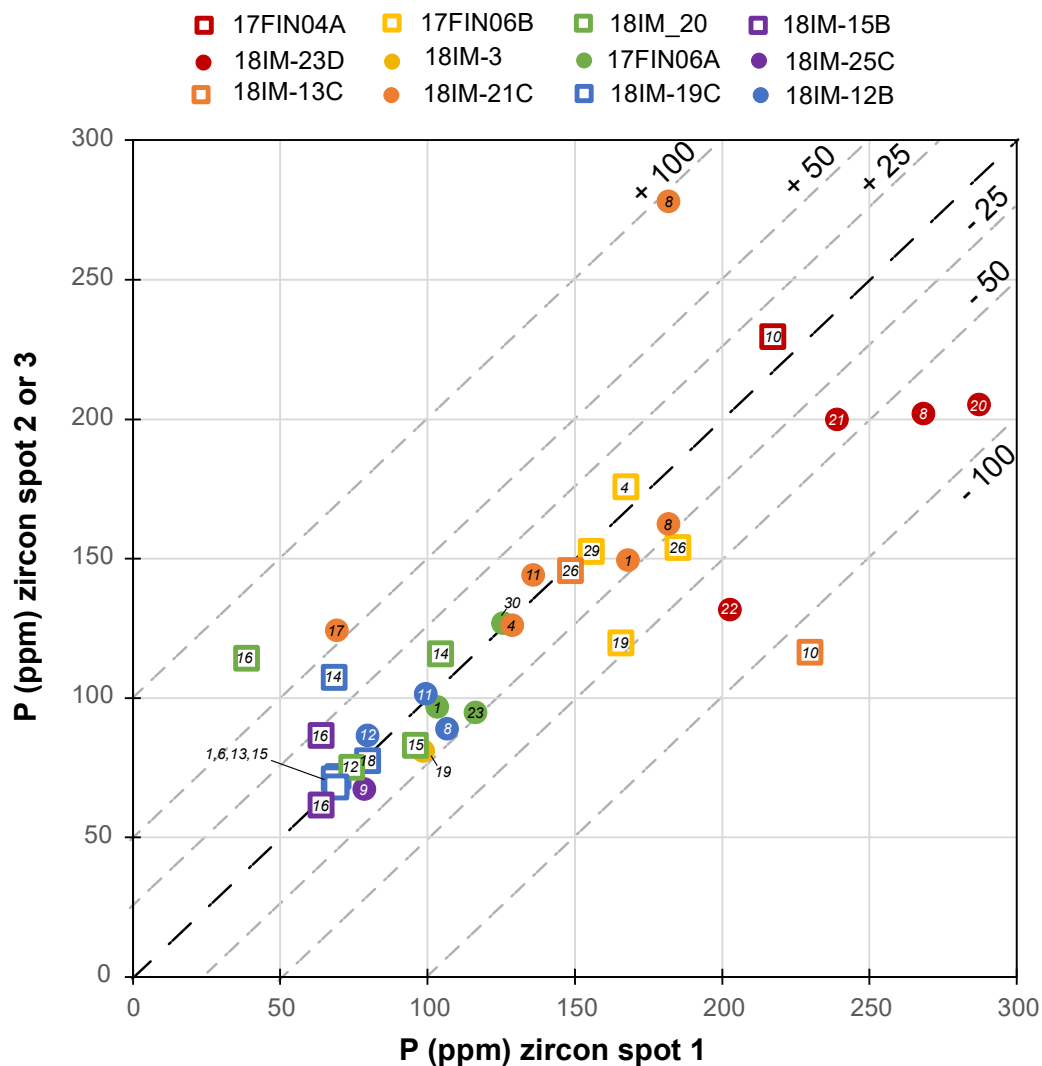
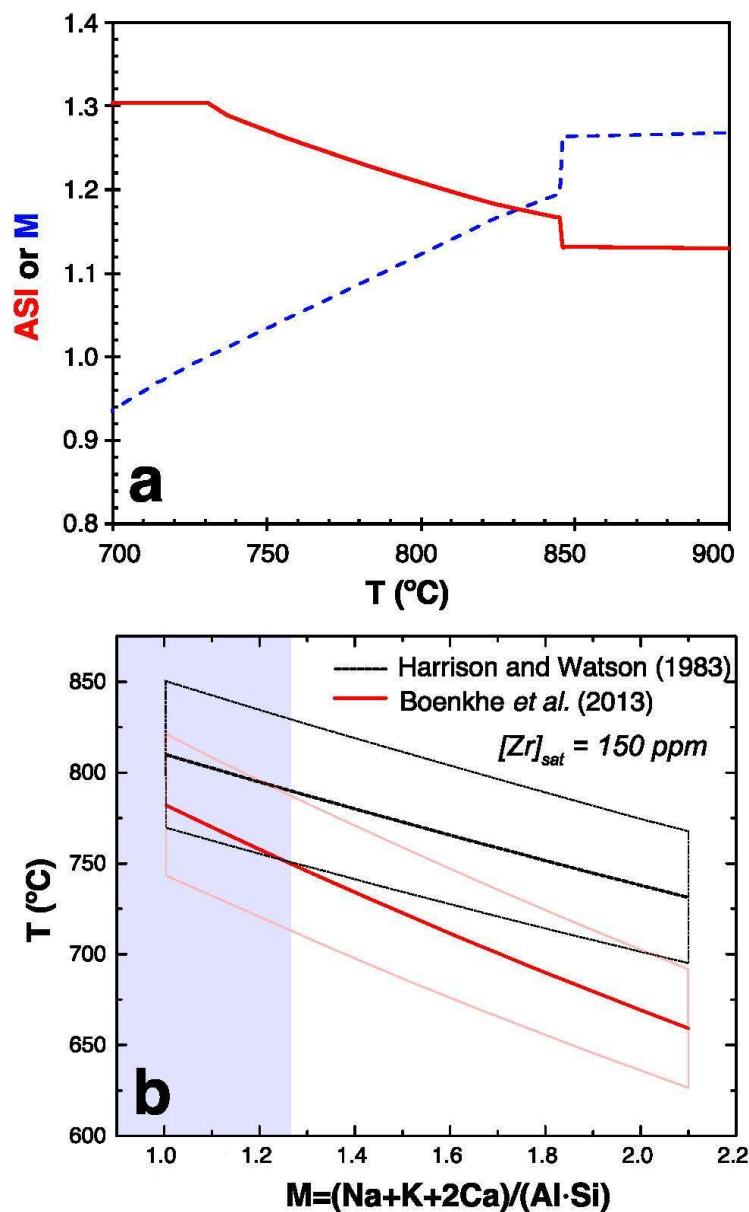


Figure S-4 (a) alphaMELTS modelling results showing variations in melt ASI (molar $\text{Al}/(\text{Ca}+\text{Na}+\text{K})$ and M [molar $(\text{Na}+\text{K}+2\text{Ca})/(\text{Al}*\text{Si})$] with temperature. ASI and M are strongly correlated due to the similarities in the molar components involved in their calculation. At low temperatures M continues to decrease due to an increase in molar Si, which is not reflected in the ASI. **(b)** Calculated zircon saturation temperatures versus M for both (Harrison and Watson, 1983; Boehnke *et al.*, 2013) for zircon saturation at 150 ppm Zr in the melt (figure modified after Boehnke *et al.*, 2013). Blue shaded region indicates range of M values in alphaMELTS modelling (0.9-1.27)



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