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# Surface and dynamical properties of Gel<sub>2</sub>

# Archit Dhingra<sup>1,\*</sup>, Alexey Lipatov<sup>2</sup>, Haidong Lu<sup>1</sup>, Katerina Chagoya<sup>3,4</sup>, Joseph Dalton<sup>3</sup>, Alexei Gruverman<sup>1</sup>, Alexander Sinitskii<sup>2</sup>, Richard G Blair<sup>5</sup> and Peter A Dowben<sup>1</sup>

- <sup>1</sup> Department of Physics and Astronomy, University of Nebraska–Lincoln, 855 North 16th Street, Lincoln, NE 68588-0299, United States of America
- <sup>2</sup> Department of Chemistry, University of Nebraska–Lincoln, 639 North 12th Street, Lincoln, NE 68588-0304, United States of America
- <sup>3</sup> Florida Space Institute, University of Central Florida, Orlando, FL 32826, United States of America
- <sup>4</sup> Department of Mechanical and Aerospace Engineering, University of Central Florida, Orlando, FL 32816, United States of America
  - Department of Physics, University of Central Florida, Orlando, FL 32816-2385, United States of America
- \* Author to whom any correspondence should be addressed.

E-mail: archit.dhingra@huskers.unl.edu

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#### Abstract

GeI<sub>2</sub> is an interesting two-dimensional wide-band gap semiconductor because of diminished edge scattering due to an absence of dangling bonds. Angle-resolved x-ray photoemission spectroscopy indicates a germanium rich surface, and a surface to bulk core-level shift of 1.8 eV in binding energy, between the surface and bulk components of the Ge  $2p_{3/2}$  core-level, making clear that the surface is different from the bulk. Temperature dependent studies indicate an effective Debye temperature ( $\theta_D$ ) of 186 ± 18 K for the germanium x-ray photoemission spectroscopy feature associated with the surface. These measurements also suggest an unusually high effective Debye temperature for iodine (587 ± 31 K), implying that iodine is present in the bulk of the material, and not the surface. From optical absorbance, GeI<sub>2</sub> is seen to have an indirect (direct) optical band gap of 2.60 (2.8) ± 0.02 (0.1) eV, consistent with the expectations. Temperature dependent magnetometry indicates that GeI<sub>2</sub> is moment paramagnetic at low temperatures (close to 4 K) and shows a diminishing saturation moment at high temperatures (close to 300 K and above).

### 1. Introduction

Two-dimensional (2D) semiconductors that are scalable down to the nanometer range are few and far between because edge scattering [1-5] and edge states [6,7] dominate as transistor channel widths approach 10 nm. The trichalcogenides [8-11], and the 2D layered germanium (II) iodide (GeI<sub>2</sub>) [12-14] are among the few 2D semiconductors where edge states and edge disorder are suppressed.

Monolayer GeI<sub>2</sub> is predicted to be a semiconductor with a modestly large band gap ( $E_g \sim 1.72-$ 3.05 eV) [14–20] that is thought to be thermally stable at high temperatures (~600 K) [14, 17] with electron mobility values similar to that of single-layer MoS<sub>2</sub> [14]. GeI<sub>2</sub> also offers the possibility of diminished edge scattering and can exist without dangling bonds [14, 21, 22] as iodine passivates the germanium bonds [21, 22]. This means that, at least in principle, GeI<sub>2</sub> is a possible 2D semiconductor-channel material without persistent dangling bonds, edge scattering, and catalytically active deleterious edge states of the transition metal dichalcogenides. A GeI<sub>2</sub>-based heterostructure has been recently theorized to exhibit an interfacial Hall effect, which may lead to development of lowpower spintronic devices [23]. In spite of all of these potential advantages, it still remains an open question as to whether the promise of GeI<sub>2</sub> for electronic [14, 23] and thermoelectric [17] applications can be realized. Utility and electronic devices are only worth pursuing if the material and its surface are stable.

Here, GeI<sub>2</sub>, an unusual 2D van der Waals (vdW) material, has been characterized using x-ray diffraction (XRD), Raman spectroscopy, high-resolution transmission electron microscopy (HRTEM), Kelvin probe force microscopy (KPFM), angleresolved x-ray photoemission spectroscopy (ARXPS), and temperature dependent x-ray photoemission **OP** Publishing

spectroscopy (XPS). Experimental indirect band gap and direct band gaps of  $GeI_2$  are extracted from the optical absorbance spectra. The effects of temperature on  $GeI_2$  (dynamical properties) are evident in both temperature dependent XPS and magnetometry. This is among the first experimental studies conducted to test the theoretically models of  $GeI_2$ .

## 2. Experimental section

#### 2.1. Synthesis of layered GeI<sub>2</sub>

Germanium (II) iodide (GeI<sub>2</sub>) was synthesized, as previously reported in the literature [24]. A mixture of powdered Ge and CuI was heated *in vacuo* at 400 °C for 12 h in a sublimation tube. The sublimate was removed from the sublimator and placed in a clean sublimation tube and held *in vacuo* at 120 °C to remove any GeI<sub>4</sub>. Large cm-sized plate-like crystals were recovered.

#### 2.2. Characterization of GeI<sub>2</sub>

X-ray fluorescence (XRF), XRD, Raman spectroscopy, HRTEM, KPFM, ARXPS, and temperature dependent XPS were used to characterize the layered GeI<sub>2</sub> crystals. The XRF analysis was performed using a PANalytical Epsilon XRF spectrometer with a 50 kV silver anode. The product showed no copper contamination with XRF. Powder XRD was performed using a PANalytical Empyrean x-ray diffractometer with a 1.8 kW copper K<sub> $\alpha$ </sub> source. The powder diffraction patterns were collected from 5° to 80° 2 $\theta$  using a step size of 0.03° 2 $\theta$  and a dwell time of 10.8 s. Raman spectra were collected using a DXR Raman microscope with a 532 nm laser operated at the power of 2 mW to prevent sample damage.

High crystallinity of  $GeI_2$  was confirmed by HRTEM. The images were acquired using a FEI Tecnai Osiris scanning transmission electron microscope equipped with a HAADF detector and a X-FEG high brightness Schottky field emission gun. The accelerating voltage was set to 200 kV. Selected area electron diffraction (SAED) pattern was also acquired in the TEM.

A commercial atomic force microscope (AFM) system (MFP3D, Asylum Research) was used to obtain AFM and KPFM images of the GeI<sub>2</sub> flakes. Pt coated Si tips (PPP-EFM, Nanosensors) were used for imaging the surface potential of the GeI<sub>2</sub> flakes, where Au(111)/mica was used as the reference sample for calibration.

All the core-level XPS measurements were performed in an ultra-high vacuum chamber using a SPECS x-ray Al anode (hv = 1486.6 eV) as the source and a hemispherical electron analyzer (PHI Model: 10–360) that has an angular acceptance of  $\pm 10^{\circ}$ . The ARXPS data were collected by changing the photoemission take-off angle between 0° and 60°, with respect to the surface normal, as described elsewhere [9]. The temperature dependent XPS was performed by using a liquid nitrogen cryostat to cool the samples between 240 K and 300 K, as described in a recent study [11].

#### 2.3. Optical absorbance spectroscopy of GeI<sub>2</sub>

Absorbance measurements were taken in transmission mode using a StellarNet BLUE-Wave UVIS-50 spectrometer with deuterium–halogen light source (SL1+SL3).

#### 2.4. Magnetometry measurements

The magnetic properties were measured using Quantum Design MPMS XL superconducting quantum interference device magnetometer, which offers a sensitivity of  $1 \times 10^{-8}$  emu. The magnetization (M) versus temperature (T) curve was measured between 4 K and 340 K while the sample was cooled in an applied magnetic field (H = 500 Oe), while the magnetization (M) versus applied magnetic field (H) curves were recorded at two different temperatures: 4 K and 300 K.

#### 3. Structural properties

GeI<sub>2</sub> crystallizes with a CdI<sub>2</sub>-type structure (see inset of figure 1(a)) in the trigonal space group  $P\bar{3}m1(164)$ with a = 4.251 Å and c = 6.828 Å, consistent with the literature [12, 13, 20]. The powder XRD pattern of synthesized GeI<sub>2</sub> (figure 1(a)) matches the literature pattern (JC-PDS 52–1295) with no detectable oxide phases. Williamson-Hall [25] analysis of the integral breadths of the sample's peaks indicated the maximum crystalline regime was 116.7 nm and a lattice strain of 0.156%. Analysis of the 00 l peaks alone resulted in a *c*-direction size of 110.8 nm and a strain of 0.140%. This indicates that the stacking and the inplane order of the sample are on par.

Figure 1(b) shows representative Raman spectrum of a GeI<sub>2</sub> crystal, which reveals two main peaks at ~84 cm<sup>-1</sup> and 121 cm<sup>-1</sup>. These peaks appear at frequencies that are lower than what was recently observed for other van der Waals materials (like TiS<sub>3</sub> and ZrS<sub>3</sub>) [11], which could be due to higher atomic masses of both Ge and I (in comparison with the atomic masses of S, Ti and Zr). Despite the instability of GeI<sub>2</sub> in ambient conditions (see figure S1 of the supplementary file available online at stacks.iop.org/2DM/9/025001/mmedia), due to the oxidation of its surface, the crystals show a layered structure similar to the well-distinguished steps observed for other 2D materials, like graphene and MoS<sub>2</sub> [26].

The high quality of these crystals is further confirmed by the HRTEM image shown in figure 1(c), which demonstrates the long-range order of the GeI<sub>2</sub> atoms with a characteristic  $d_{(110)}$  spacing of 0.209 nm. Moreover, a SAED pattern recorded on the same crystal (see figure 1(d)) indicates a hexagonal arrangement of GeI<sub>2</sub> atoms. This hexagonal arrangement



**Figure 1.** (a) The XRD pattern of GeI<sub>2</sub>, along with the inset showing the CdI<sub>2</sub>-type crystallographic structure of layered GeI<sub>2</sub>. (b) Raman spectrum of a GeI<sub>2</sub> crystal right after micromechanical exfoliation. (c) HRTEM image of a GeI<sub>2</sub> crystal elucidating long-range order of its atoms that have a characteristic  $d_{(110)}$  spacing of 0.209 nm. (d) SAED pattern corresponding to the HRTEM image of the GeI<sub>2</sub> crystal (shown in (c)), which confirms hexagonal arrangement of its atoms.

of atoms implies that the GeI<sub>2</sub> lattice belongs to the point group D<sub>3d</sub>, which is consistent with the existence of the  $P\bar{3}m1(164)$  trigonal space group as revealed by our XRD measurements (figure 1(a)), and also with the existing literature [12, 13, 20]. And according to the indexing of the diffraction spots in this SAED pattern, the observed view corresponds to the *ab* plane of GeI<sub>2</sub> with the lattice parameter  $a = b \approx 0.42$  nm (which is in line with a = 4.251 Å, mentioned above and elsewhere [12, 13]).

#### 4. Surface properties

The topography measurements show that the surface is made of relatively flat flakes with large terraces, as shown in figure 2 (top panels). From KPFM, it is evident that the surface potential of the GeI<sub>2</sub> crystals has some local variations, as revealed by bottom panels of figure 2 (as well as figure S2). The surface work function of GeI<sub>2</sub> is slightly less than that of Au(111) (which is ~5.33 eV [27]), but the possibility of surface oxidation raising the measured work function of GeI<sub>2</sub> cannot be excluded.

The photoemission spectra of the Ge  $2p_{3/2}$  corelevel collected at 0° and 45°, shown in figure 3(a), reveal that the Ge  $2p_{3/2}$  core-level feature contains two components separated by 1.8 eV. This is indicative of a surface-to-bulk core-level shift. From the result of the ARXPS measurements shown in the inset of figure 3(a), it is clear that the P<sub>1</sub> feature observed at the higher binding energy value of 1220.1  $\pm$  0.1 eV is closer to the surface than the  $P_2$  feature observed at the lower binding energy of 1218.3  $\pm$  0.1 eV. Binding energy of P<sub>2</sub> (1218.3  $\pm$  0.1 eV) matches the binding energy of Ge  $2p_{3/2}$  core-level for GeI<sub>2</sub> reported elsewhere [28]. That said, no such phenomenon is observed in the case of iodine as only a single I  $3d_{5/2}$ core-level feature is recorded at the binding energy value of 619.2 eV (see figure 3(b)), which is in line with the binding energy value of the I 3d<sub>5/2</sub> corelevel in CdI<sub>2</sub> [29]. A wider XPS spectrum (survey XPS spectrum) with high signal-to-noise ratio was also recorded to confirm the high quality of GeI<sub>2</sub> crystals (see figure S3 of the supplementary file).

The angle-resolved photoemission intensity ratio of the I  $3d_{5/2}$  core-level to Ge  $2p_{3/2}$  core-level spectra (shown in the inset of figure 3(b)) hints that this system terminates in germanium or, at least, that the surface is iodine deficient. And, thus, the large surface to bulk core-level shift observed for germanium is consistent with a surface that differs substantially from the bulk, which should be expected for a material that can be easily exfoliated from bulk. Therefore, a low cleavage energy of 0.16 J m<sup>-2</sup> for GeI<sub>2</sub> [14], which



**Figure 2.** Topography (AFM) and surface potential (KPFM) of GeI<sub>2</sub> crystals. KPFM measurements on the GeI<sub>2</sub> single crystals show that the surface potential is typically in the range of 0.25-0.30 V relative to Au(111) with small variations at different locations.



**Figure 3.** ARXPS measurements. (a) The raw XPS spectra of the Ge  $2p_{3/2}$  core-level, and (b) the raw XPS spectra of the I  $3d_{5/2}$  core-level collected at 0° (solid red triangles and line) and 45° (solid blue triangles and line). The inset in (a) reveals that P<sub>1</sub> (the Ge  $2p_{3/2}$  core-level feature at higher binding energy) is closer to the surface than P<sub>2</sub> (the Ge  $2p_{3/2}$  core-level feature at lower binding energy). The hollow triangles and circles in (a) represent the P<sub>1</sub> and P<sub>2</sub> core-level components, respectively, and the total fit to the raw Ge  $2p_{3/2}$  core-level spectra is shown by the solid black lines. The inset in (b) shows the ratio of photoemission peak intensities of I  $3d_{5/2}$  core-level to Ge  $2p_{3/2}$  core-level as a function of photoemission take-off angle with respect to the surface normal.

is lower than that of graphite [30], makes it more akin to a vdW structure than many other layered materials.

To understand the dynamical behavior of  $GeI_2$  surface, we performed temperature-dependent XPS measurements on this material (figure 4). Figure 4(a)

shows representative XPS spectra of the Ge  $2p_{3/2}$  core-level collected between 270 K and 300 K, while figure 4(b) shows the representative XPS spectra of the I  $3d_{5/2}$  core-level for temperatures ranging from 240 K to 300 K. The photoemission intensities of the



**Figure 4.** Temperature dependent XPS measurements. (a) Raw XPS of the Ge  $2p_{3/2}$  core-level (solid markers) collected at temperatures ranging from 270 K to 300 K, along with the total fit (black line). The hollow markers show the P<sub>1</sub> and P<sub>2</sub> features. (b) XPS spectra of the I  $3d_{5/2}$  core-level recorded at temperatures ranging from 240 K to 300 K. (c) Debye–Waller factor plot for the P<sub>1</sub> feature of the Ge  $2p_{3/2}$  core-level, as well as the temperature dependence of the XPS intensities of the P<sub>1</sub> and P<sub>2</sub> components (inset). (d) Debye–Waller plot for the I  $3d_{5/2}$  core-level.

I  $3d_{5/2}$  core-level and the P<sub>1</sub> feature of the Ge  $2p_{3/2}$ core-level are found to decrease with increasing temperature, which is unlike the temperature dependence of the photoemission intensity of the P2 component (as indicated in the inset of figure 4(c)). An inverse relationship between the XPS intensity and temperature is usually expected since photoemission is a scattering process, which is hindered by the dynamical motion of the scattering centers. However, since the XPS intensity of the P<sub>2</sub> feature of the Ge 2p<sub>3/2</sub> corelevel increases with increase in temperature, it indicates an increase in germanium concentration. Since the XPS intensity of the  $P_1$  component (the surface XPS component) does not increase with temperature, segregation of germanium to the subsurface region is highly probable. This suggests that the surface or near surface region of GeI<sub>2</sub> tends to become germanium rich, and the germanium rich surface will be a factor to consider as it could affect the contact potentials in device applications.

A quantitative relationship between the XPS intensities and thermal motion of atoms is given by the Debye–Waller model. According to this model, the XPS intensity is an exponentially decaying function of temperature, and this function depends

on the core-level's Debye–Waller factor (W(T)) as [11, 31–36]:

$$I = I_0 e^{-2W(T)}.$$

And in case of isotropic vibrations, W(T) is [11, 31–36]:

$$W(T) = \frac{3(\hbar\Delta k)^2 T}{2mk_B \theta_D^2},$$

here *m* is the scatterer's mass (which is ~72.6 u for germanium and ~126.9 u for iodine),  $k_B$  is Boltzmann constant,  $(\hbar \Delta k)$  is electron momentum transfer for the given core-level, *T* is the absolute temperature of the scatterer, and  $\theta_D$  is its effective Debye temperature. The electron momentum transfer,  $\hbar \Delta k$ , is calculated from the kinetic energy ( $E_{kin}$ ) associated with the photoelectron of a given core-level, as  $\hbar \Delta k = \sqrt{2m_e E_{kin}}$  [32, 36]. The value of  $\theta_D$  depends on the slope ( $\alpha$ ) of the Debye–Waller plots (figures 4(c) and (d)), and is given as:

$$-\alpha = \frac{3(\hbar\Delta k)^2}{mk_B\theta_D^2}.$$



Owing to this experiment's geometry, the photoemission intensity in this experiment is dominated by vibrational modes along the surface normal, and not anharmonic or in-plane modes. So, the value of Debye temperatures of the P<sub>1</sub> feature of Ge 2p<sub>3/2</sub> corelevel ( $\theta_D = 186 \pm 18$  K) and the I  $3d_{5/2}$  core-level ( $\theta_D = 587 \pm 31$  K), extracted from the linear fits in figures 4(c) and (d), have to be their respective effective Debye temperatures. Nevertheless, this is still comparable to other measures of Debye temperature [37].

The low effective Debye temperature of the Ge XPS feature, associated with the surface component of the Ge  $2p_{3/2}$  core-level (P<sub>1</sub>), can be understood to be a result of the partial covalent nature of the Ge-I bond and germanium surface termination. The covalent nature of the Ge-I bond is in agreement with the Raman features of GeI<sub>2</sub>, noted above, which appear at the small wavenumbers of 84  $cm^{-1}$  and 121  $cm^{-1}$ . It is noteworthy that the effective Debye temperature of iodine (587  $\pm$  31 K) is way higher than that of germanium. The significantly higher Debye temperature seen for iodine is indicative of the iodine residing in the bulk not the surface, and this, in turn, is consistent with the ARXPS measurements, which indicates that the surface terminates in germanium and not iodine.

### 5. Electronic and magnetic properties

The Tauc method of optical absorption analysis [38–47] was applied to optical absorbance spectrum of pristine GeI<sub>2</sub> (figure S4, supporting information) to determine its indirect and direct band gap energies.

Extrapolating the linear part of the Tauc plot in figure 5(a) to the x-axis, i.e. the square root of the absorbance with energy, provides an indirect band gap of 2.60  $\pm$  0.02 eV GeI<sub>2</sub>, which is in agreement with the 2.59 eV band gap predicted by theory [14]. The linear extrapolation of the square of the absorbance with energy, plotted in figure 5(b), indicates that the direct band gap, of GeI<sub>2</sub>, is  $2.8 \pm 0.1$  eV. The agreement between the theoretical band gap of monolayer GeI<sub>2</sub> and the experimentally obtained optical band gap of GeI<sub>2</sub> crystal implies that the interlayer interactions in this material are, indeed, weak. This, in turn, is consistent with its low cleavage energy [14] and proneness to facile intercalation [13]. It is worth mentioning that even though numerous theoretical efforts on GeI<sub>2</sub> predict a range of band gaps (1.72-3.05 eV [14–20]), density functional theory (DFT) is generally notorious for underestimating the ground state band gap of materials [48–51]. On the other hand, while DFT will tend to underestimate the band gap, DFT will frequently agree with the optical gap, which can often be much smaller than the ground state band gap because of Coulombic interactions.

The magnetic properties of GeI<sub>2</sub> are temperature dependent, as shown in figure 6. GeI<sub>2</sub> is moment paramagnetic at extremely low temperatures, i.e. at T = 4 K (see inset of figure 6), but increasingly behaves as a more conventional paramagnet at room temperature (T = 300 K). This lack of magnetic ordering, which may be attributed to the absence of magnetic anisotropy, is consistent with the existing theory on GeI<sub>2</sub> [14], and the Mermin–Wagner theorem [52]. Nonetheless, weak exchange coupling cannot be excluded.



**Figure 6.** The magnetization (M) versus temperature (T) curve for pristine layered GeI<sub>2</sub> measured between 4 and 340 K while the sample was cooled in a magnetic field (H = 500 Oe). Inset shows the magnetization (M) versus applied magnetic field (H) curves obtained at T = 4 K (blue) and T = 300 K (red).

# 6. Conclusion

In conclusion, GeI<sub>2</sub> is a moderately wide-band gap semiconductor with an indirect band gap of  $2.60 \pm 0.02$  eV (consistent with theory) and a direct band gap of 2.8  $\pm$  0.1 eV. The surface terminates in germanium, and the surface is quite susceptible to dynamical motion, i.e. temperature-dependent XPS reveals a low effective Debye temperature of 186  $\pm$  18 K for the surface component of the Ge 2p<sub>3/2</sub> core-level. Our KPFM results indicate that, in spite of some local potential variations, fabrication of n-type GeI<sub>2</sub>-based electronic Schottky devices is possible [16] since work function of GeI<sub>2</sub> is slightly less than that of Au [9, 53, 54]. Nevertheless, the germanium rich surface will be a factor to consider from the standpoint of device fabrication and applications as facile oxidation could affect contact potentials.

Regardless of the lack of magnetic ordering, confirmed by our temperature dependent magnetometry measurements, the possibility that the magnetic properties can be modified seems likely. The ease of exfolilation [14] and intercalation [13] make GeI<sub>2</sub> capable of accomodating extrinsic species. These foreign species can, then, induce magnetic ordering [55–58] and enhance the spin–orbit coupling [59–67]. As the inversion symmetry is already broken at the surface of GeI<sub>2</sub>, some intrinsic spin–orbit coupling is expected at the surface and may be present at an inhomogeneous interface. With intercalation of some species, it may be possible for GeI<sub>2</sub>-based heterostructures to show proximity effects as has been reported for graphene-based heterostructures [68-71]. Besides, introduction of exotic species may also allow for tuning of the band gap of GeI<sub>2</sub> as needed [72, 73].

# Data availability statement

The data that support the findings of this study are available upon reasonable request from the authors.

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## ORCID iDs

2198-4710

Archit Dhingra © https://orcid.org/0000-0001-9352-4361 Alexey Lipatov © https://orcid.org/0000-0001-5043-1616 Alexander Sinitskii © https://orcid.org/0000-0002-8688-3451 Peter A Dowben © https://orcid.org/0000-0002-

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