1 **Integrated flux and pool size analysis in plant central metabolism reveals unique roles of glycine**

2 **and serine during photorespiration**

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9

10 **Abstract**

11 Photorespiration is an essential process juxtaposed between plant carbon and nitrogen metabolism that

12 responds to dynamic environments. Photorespiration recycles inhibitory intermediates arising from

13 oxygenation reactions catalyzed by rubisco back into the C_3 cycle, but it is unclear what proportions of its

14 nitrogen-containing intermediates (glycine and serine) are exported into other metabolisms *in vivo* and

15 how these pool sizes affect net CO_2 gas exchange during photorespiratory transients. To address this

16 uncertainty, we measured rates of amino acid export from photorespiration using isotopically

17 nonstationary metabolic flux analysis. This analysis revealed that ~23-41% of the photorespiratory carbon

18 was exported from the pathway as serine under various photorespiratory conditions. Furthermore, we

19 determined that the build-up and relaxation of glycine pools constrained a large portion of photosynthetic

20 acclimation during photorespiratory transients. These results reveal the unique and important roles of

21 glycine and serine in successfully maintaining various photorespiratory fluxes that occur under

22 environmental fluctuations in nature and providing carbon and nitrogen for metabolism.

Main

In C_3 plants, photorespiration is a high-flux metabolic pathway that recycles the inhibitory 2-phosphoglycolate (2PG) produced from the oxygenation reaction catalyzed by ribulose-1,5-bisphosphate carboxylase-oxygenase (rubisco). The recycling of 2PG to the C_3 intermediate, 3-phosphoglycerate (3PGA) involves a series of energy-consuming reactions across multiple organelles [1,2]. Upon cursory consideration, photorespiration is often viewed as a wasteful process, because the conversion from glycine to serine releases a large amount of CO_2 and ammonia, the former is released from photorespiration and the latter is re-assimilated with high-energy cost. However, besides the beneficial removal of 2PG, photorespiration also plays a vital role in central metabolism due to its close interaction with many other pathways, especially nitrate assimilation [3–6]. Although the interaction between photorespiration and nitrate assimilation is not mechanistically clear, nitrate assimilation depends to some degree on photorespiration [5,7]. For example, plants grown under elevated CO_2 partial pressures with decreased photorespiration have decreased nitrate assimilation into organic nitrogen (N) compounds and total N concentration of plant tissues [8,9]. Additionally, the major route of ammonia assimilation, glutamate synthesis, increases as photorespiration increases possibly not only facilitating increased rates of ammonia recapture from photorespiration but also increased rates of *de novo* nitrate assimilation [10,11].

The mechanism linking photorespiration to nitrate assimilation could be related to the export of N-rich amino acids from photorespiration. Due to the high metabolic flux channeled through photorespiration, even small changes in photorespiratory flux could affect the flux distribution through central carbon (C) and N metabolism. The extent to which photorespiration operates as a closed cycle to regenerate 3-PGA for the C_3 cycle or as an open cycle allowing metabolite export to interacting metabolisms remains unknown [12]. Exporting glycine and serine from photorespiration into non-photorespiratory metabolism (*e.g.,* protein synthesis, providing amino groups to alternative metabolic processes or transport) instead of returning their C to the chloroplast as 3PGA additionally affects the CO_2 loss and the energy costs associated with recycling of 2PG into 3PGA.

Evidence for a fraction of photorespiratory C exported from photorespiration as glycine and/or serine is revealed in biochemical modeling of leaf photosynthesis [13]. The export of photorespiratory C as amino acids can explain the decreased rates of net CO_2 assimilation (*A*) in response to CO_2 and O_2 at high CO_2 partial pressures, typically 2-4 times ambient conditions and saturating illumination [13]. In this paper we define glycine and serine export as amino acids that leave photorespiratory metabolism, this includes amino acids that leave the leaf entirely, as in phloem exudate, but also to amino acids that leave photorespiration to participate in non-photorespiratory metabolism in the same leaf tissue. The fraction of photorespiratory C exported as amino acids estimated using photosynthetic models ranges between 0.3-

56   0.77 under the high CO_2 partial pressures where this approach can be used, with serine being the primary

57   amino acid exported [14,15]. Exporting glycine and serine from photorespiration increases photosynthesis at

58   high CO_2 partial pressures when carbon assimilation is limited by the availability of inorganic phosphate

59   in the chloroplast [16].

60   Exporting glycine and serine from photorespiration is also coupled to the N flux to these amino acids.

61   Specifically, this N flux out of photorespiration in the form of amino acids must be supplemented by

62   increasing the *de novo* assimilation of N to replace the amino groups required for glyoxylate amination to

63   glycine, potentially explaining why the estimated fraction of amino acid export increases with increasing

64   NO_3^- availability [15]. Despite this compelling evidence for serine and glycine export under high CO_2

65   partial pressures when photorespiration is limited, it is unclear if high rates of amino acid export happen

66   under more physiologically relevant CO_2 partial pressures and forms a regular component of central

67   metabolism under more typical growing conditions. C export from photorespiration also raises interesting

68   questions concerning the functional importance of photorespiration, since this export would not replenish

69   C lost from the C_3 cycle following rubisco oxygenation.

70   Another uncertainty of photorespiration is the speed of its metabolic response and subsequent impact on

71   *A*. Photorespiration is a dynamic process, responding to *A*, stomatal conductance, and light regimes.

72   However, it is mostly investigated under steady-state conditions by comparing the metabolomic profile

73   and physiology following different environmental and genetic perturbations [17–19]. Under non-steady-state

74   conditions, amounts of photorespiratory intermediates are sensitive to photorespiratory conditions both in

75   the long- and short-term [17–19]. Photorespiratory intermediates additionally accumulate during transient

76   changes in rubisco oxygenation rate or when downstream photorespiratory reactions are not

77   stoichiometrically balanced [20–24]. When photorespiratory intermediates upstream of the glycine

78   decarboxylase (GDC) reaction like glycine accumulate, the CO_2 release per rubisco oxygenation should

79   decrease proportionally, increasing subsequent *A* during this accumulation. Despite this theoretical

80   underpinning, it is not clear if the change in glycine pools is large enough to affect *A* to a measurable

81   degree during transients in photorespiratory rates.

82   The use of isotopic tracers and metabolic flux analysis (MFA) are powerful tools to resolve how

83   metabolism responds to environmental changes [25,26]. For example, isotopic labeling with $^{13}CO_2$ in

84   combination with varied CO_2 and/or O_2 mole fraction has been commonly used to infer the impact of

85   photorespiration on leaf C metabolism. Under different CO_2/O_2 conditions, distinct ^{13}C labeling

86   enrichment has been observed in many primary and specialized metabolites [20,27], as well as proteins [28,29].

87   These studies were usually performed at isotopic pseudo-steady state with a single time-point, which does

88   not allow flux quantification in autotrophic metabolism [30]. Recent development of isotopically

89 nonstationary metabolic flux analysis (INST-MFA) more fully leverages the benefits of [13]CO_2 labeling to

90 provide a global assessment of C fluxes through autotrophic leaf metabolism [26,31–34]. INST-MFA is

91 therefore a promising approach for estimating fluxes and metabolic responses to changes in

92 photorespiratory rates *in vivo*.

93 Here, INST-MFA was applied to examine the C fluxes through and out of the photorespiratory pathway

94 under different photorespiratory conditions manipulated by the O_2 mole fraction. This INST-MFA

95 analysis gave a quantitative estimate of serine and glycine exported from photorespiration under steady-

96 state conditions at ambient CO_2 partial pressures, which agreed with estimates made under high CO_2

97 partial pressures using estimates from CO_2 response curves. Metabolite pool size measurements of 38

98 central metabolites showed that glycine had the strongest changes in response to photorespiratory shifts.

99 We also determined that changes in glycine concentration were large enough to explain 20-30% of the

100 gain in net CO_2 assimilation when transitioning from low to high photorespiratory conditions.

101

Results

Isotopically Nonstationary Metabolic Flux Analysis

104 To estimate C flux distribution on central metabolism, transient ^{13}C labeling experiments were performed

105 on fully expanded *Nicotiana tabacum* (tobacco) leaves after short-term acclimation to three O_2 conditions

106 (2%, 21%, and 40%). The labeling experiments were carried out at the same irradiance as the growth

107 chamber (100 µmol m^{-2} s^{-1}) to minimize changes in metabolism due to light acclimation. Metabolic

108 fluxes through central metabolism were determined from a metabolic network model (Supplementary

109 Data 1), mass isotopologue distributions (MIDs) during the transient labeling period (Supplementary Data

110 2), and measured input and output fluxes to constrain the flux solutions. The independently determined

111 input and output fluxes used to constrain the flux solution included the *A*, rubisco oxygenation rate (v_o),

112 rubisco carboxylation rate (v_c), starch and sucrose production rates, and output flux ratios of amino acids

113 and sucrose measured in vascular exudate (Supplementary Table 1 and 2 and Supplementary Method 1-

114 4). INST-MFA was performed under the assumption that the metabolism is steady-state (Supplementary

115 Method 4 with steady-state assumption validated below). The selection of labeling data for INST-MFA

116 modeling and tests for several network models are described in Supplementary Method 4. A subset of key

117 measured labeling data and the labeling kinetics simulated by the best-fit INST-MFA flux values are

118 shown in Supplementary Figure 1 and the detailed fits of the best-fit models are shown in Supplementary

119 Data 3.

120   Manipulating O_2 condition had a large and significant effect on v_o (2.60-26.37 µmol g^{-1} FW hr^{-1}) and a

121   smaller but significant effect on v_c (53.36-79.70 µmol g^{-1} FW hr^{-1}, Fig. 1, 95% confidence intervals of the

122   best-fit fluxes are shown in Supplementary Figure 2 and 3 and Supplementary Data 4). The rate of starch

123   and sucrose synthesis decreased significantly with increasing rates of photorespiration (Fig. 1 and

124   Supplementary Data 4). Direct measurement of the allocation of photo-assimilated C using $^{14}CO_2$

125   labelling showed that rates of starch and sucrose synthesis are strongly correlated with A (Supplementary

126   Figure 4 and Supplementary Method 2).

127   The INST-MFA revealed the export rates of photorespiratory serine and glycine from the

128   photorespiratory pathway at ambient CO_2 partial pressures (Fig. 1). The serine export rate to alternative

129   sinks increased from 3.61 to 7.02 µmol g^{-1} FW hr^{-1} at high rates of photorespiration, but it decreased to

130   0.02 µmol g^{-1} FW hr^{-1} at low photorespiration rates. The export of glycine to alternative sinks was less

131   than 0.01 µmol g^{-1} FW hr^{-1} at three conditions, much lower than that of serine. A portion of the export of

132   amino acids from photorespiration, the removal of glycine or serine through phloem, was estimated based

133   on the output flux to phloem and the output flux ratios of amino acids and sucrose measured in vascular

134   exudate (Supplementary Table 2 and Supplementary Method 4). The removal rates of glycine or serine

135   through the phloem were small relative to the total export, between 0.03 to 0.38 µmol g^{-1} FW hr^{-1} at

136   various photorespiratory conditions (Fig. 1). The total amount of C leaving the photorespiratory pathway

137   as serine scaled with the rate of photorespiration. At high photorespiration (40% O_2), 41% of the

138   photorespiratory C influx (52.8 µmol C g^{-1} FW hr^{-1}) was exported as serine (at a rate of 21.5 µmol C g^{-1}

139   FW hr^{-1}). The percentage of photorespiratory C leaving as serine was 32% at normal photorespiration

140   (21% O_2) and 23% at low photorespiration (2% O_2). These data indicate that 23-41% of the

141   photorespiratory C could be exported from the pathway as serine under physiologically relevant CO_2

142   partial pressures. The INST-MFA model was also run with glycerate labelling data and allowing glycerate

143   export from the photorespiratory pathway. This extended model predicted that glycerate and glycine

144   export rate was slow (< 0.1 µmol g^{-1} FW hr^{-1} in all oxygen conditions), while the serine export rate was

145   slow at low photorespiration and rose at normal and high photorespiration (<0.1, 3.7 and 7.0 µmol g^{-1} FW

146   hr^{-1} in 2%, 21%, and 40% O_2 respectively) (see Supplemental Method 4).

Estimating photorespiratory C export using CO_2 response of A

148   The export of photorespiratory C to amino acid synthesis or alternative metabolism accounts for the

149   behavior of A at high CO_2 partial pressures in leaf photosynthesis models [13,15]. Using this approach, we

150   estimated the fraction of C that leaves the photorespiratory pathway as glycine (α_G) and serine (α_S) using

151   the measured CO_2 response of A at ambient O_2 and high light ($\sim$1000 µmol m^{-2} s^{-1}) fitted with the

152   photosynthetic model over the triose phosphate utilization (TPU)-limited range to compare to the INST-

153   MFA results (Fig. 2). The high-CO_2 portion of the curve in orange, where TPU limits CO_2 assimilation,

154   showed the characteristic curve that is only explained by amino acid removal from photorespiration [13,35].

155   This curvature was used to fit a resultant α_G and α_S of 0.01 and 0.45, indicating that the C flow out of

156   photorespiration was primarily in the form of serine, consistent with the estimates ($\alpha_G < 0.01$ and $\alpha_S =$

157   0.32) calculated from the INST-MFA (Fig. 1).

Metabolic responses to low and high photorespiration

159   The absolute pool sizes of 38 central metabolic intermediates were measured in tobacco leaves exposed to

160   varied photorespiratory conditions. Compared to the ambient O_2 condition, the metabolic changes at high

161   photorespiration (40% O_2) and low photorespiration (2% O_2) were modest (Fig. 3). Most metabolites had

162   insignificant changes in pool size during the short-term photorespiratory changes (Fig. 3 shows the mean

163   fold-changes of the pool size between 2% and 21% O_2 or between 40% and 21% O_2; Supplementary Data

164   5 shows the absolute concentrations under the three oxygen conditions). The most significantly altered

165   metabolites were photorespiratory intermediates (glycine, glycolate and glyoxylate), alanine, and ADPG

166   (Supplementary Data 5). Many photorespiratory intermediates (glycine, serine, glycerate, glycolate and

167   glyoxylate) were far from fully labelled even at high photorespiratory condition (Supplementary Figure

168   5). The average ^{13}C enrichment of these metabolites reached between 10% and 75% following 30 min of

169    $^{13}CO_2$ labeling (Supplementary Figure 5), indicating that these intermediates exist in multiple subcellular

170   pools with an active pool being involved in photorespiration and other pools (presumably in the vacuole

171   [25]) not involved in photorespiration. The active and inactive pools of glycine were estimated using the

172   measured total pool size and the dilution parameters from the INST-MFA (Supplementary Data 4 and 5).

173   The extent of changes in glycine was much greater in the active pool (15-fold) than in the total pool (4-

174   fold) (Fig. 4). These data support the steady-state assumption of INST-MFA (Supplementary Figure 6).

175   These data highlight the value of combining metabolite abundance with labeling and demonstrate the

176   principle that in many cases changes in metabolite pool size do not directly translate into flux changes.

177   For example, the pool size of 2PG was not significantly different between different oxygen conditions

178   (Fig. 3 and Supplementary Data 5), but the ^{13}C labeling kinetics of 2PG were faster in high oxygen than in

179   low oxygen (Supplementary Figure 1 and Supplementary Data 2), suggesting higher flux through rubisco

180   oxygenation. This highlights that time-dependent labeling profiles are more informative indicators for

181   flux rather than changes in pool sizes and that inferring flux from pool sizes alone can lead to inaccurate

182   interpretations.

Photorespiration under non-steady state

184   Quantifying the steady-state active pool size of glycine at low and high photorespiratory conditions

185   helped determine how much the size of glycine pools constrain net gas exchange during photorespiratory

186   transients. To do this, A during O_2 transients was measured when tobacco leaves were exposed to 2% O_2

187   until the steady-state and then switched to 40% O_2. During the low-to-high O_2 transient, A did not

188   instantly reach a new steady state at 40% O_2 but gradually declined at an exponential rate for ~2 min until

189   the new steady-state was reached (Fig. 5). The extra C gained or lost during the transient in the decrease

190   (or increase) in A as compared to how much C would be fixed with an instant transition to the new steady

191   state was defined as the "C lag", which can be quantified by integrating the area between the fitted curve

192   of A and a baseline of A at the new steady state (Fig. 5a). During the transition from 2% O_2 to 40% O_2, the

193   total glycine pool increased by 30.8 ± 2.4 µmol m^{-2} based on the differences in steady-state levels (after

194   30 min acclimation to high or low oxygen) between high and low oxygen levels (Fig 4) or 20.8 ± 6.2

195   µmol m^{-2} based on the initial build-up of glycine pool after switching oxygen for 60 s when the gas

196   exchange transient is near complete (Supplementary Figure 7), accounting for 20-30% of the total C lag

197   $(50.4 \pm 1.7$ µmol m$^{-2})$ as the accumulated glycine would otherwise be metabolized by the downstream

198   glycine decarboxylase reaction (2 mol of glycine yielding 1 mol of CO_2). The build-up of the active

199   glycine pool based on the steady-state measurement $(25.7 \pm 3.4$ µmol m^{-2}, Fig 4) accounted for ~25% of

200   the total C lag. When looking at the transition from 40% O_2 to 2% O_2, the lag in the photosynthetic

201   increase resulted in loss of C represented by the shaded area (Fig. 5b). The reduction of the active glycine

202   pool based on the steady-state measurement explained ~15% of this total C lag $(83.9 \pm 3.8$ µmol m$^{-2})$.

203   Note that the change in concentrations in response to O_2 treatment is negligible and small in other

204   metabolites upstream of glycine (i.e. 2PG, glycolate and glyoxylate, Fig. 5), further indicating glycine is

205   the primary metabolic pool impacting this transient response.

206

207   **Discussion**

208   **Impact of photorespiratory flux changes on C and N metabolism**

209   Our INST-MFA indicates that C was exported from photorespiration under ambient CO_2 partial pressures

210   as serine at high rates. The INST-MFA-estimated fraction of serine export is 0.27-0.39 (95% confidence

211   interval using the parameter continuation method) under ambient O_2 and CO_2 partial pressures

212   (Supplementary Figure 2 and Supplementary Data 4), which is close to the 95% confidence interval of the

213   value estimated using photosynthetic models fitted at high CO_2 partial pressures (0.38-0.51) (Fig. 2).

214   Given these large export rates, where do these amino acids go? Prior studies have shown that glycine is

215   incorporated into leaf protein when photorespiration is induced by low CO_2 [28,36], and almost all

216 photorespiratory glycine is routed to protein synthesis in water-stressed soybean leaves under high light
217 and low CO_2 partial pressure [29]. Protein synthesis is not likely to explain all the amino acid export from
218 photorespiration, since glycine and serine are present in similar proportions in total plant protein, but
219 serine is exported at a much higher rate [37]. Past work also suggests that at least some photorespiratory
220 serine and glycine is exported out of source leaves under photorespiratory conditions as is evident by an
221 increased labeling in serine and glycine within the petiole when source leaves were labeled with $^{14}CO_2$
222 under elevated O_2 [38], but our vascular exudates indicated that removal from the leaf only accounted for a
223 small percentage of the total amino acids exported from photorespiration. Photorespiratory glycine and
224 serine are also required for the synthesis of various essential metabolites. For example. serine provides the
225 carbon backbone for the synthesis of cysteine and tryptophan, which are further used to produce sulphur-
226 containing compounds (*e.g.* S-adenosylmethionine) and indolic compounds (*e.g.* auxin and
227 glucosinolates) [39,40]. Serine also is an important substrate for the biosynthesis of phosphatidylserine and
228 sphingolipid that are critical to plant development and survival [41]. Glycine enhances the synthesis of
229 glutathione (γ-Glu–Cys–Gly) as suggested by previous leaf feeding experiments [42]. Taken together, these
230 findings indicate that photorespiration is an important source of amino acids for protein and metabolite
231 synthesis and future work should work to resolve where specifically this serine goes within metabolism.

232 The increased fraction of photorespiratory C exported as amino acids with increased photorespiration in
233 this study also helps explain the mechanistic link between photorespiration and N metabolism [5,43]. Large
234 serine export fluxes diverted from photorespiration would need to be supplemented by equivalent *de novo*
235 N assimilation. The *de novo* N assimilation rate would need to increase to support a higher total amount
236 of photorespiratory C leaving as amino acids under higher photorespiratory condition. Supporting this
237 hypothesis, plants under ambient photorespiration had ~10-30% higher N assimilation rates than those
238 under low photorespiration (high CO_2 or low O_2) [8]. The coupling of N assimilation with photorespiration-
239 derived glycine and serine has also been suggested from a dual labeling study with $^{13}CO_2$ and ^{15}N-
240 ammonium nitrate [44]. In this 6-h labeling study, up to 50% of ^{15}N glycine and ^{15}N serine molecules were
241 also ^{13}C enriched, probably due to large turnover glycine and serine associated with high flux rates of
242 photorespiration, whereas many other amino acids were only ^{15}N-labelled because their ^{15}N-amino groups
243 were not fixed onto ^{13}C-enriched C skeletons. Exporting glycine and serine from the photorespiratory
244 pathway could enable plants to efficiently assimilate N onto C skeletons produced during
245 photorespiration, also stimulating CO_2 uptake at high CO_2 [15].

246 The finding that large amounts of C leave photorespiration for other metabolisms and do not re-enter the
247 C_3 cycle suggests some interesting hypotheses concerning the central purpose of photorespiration.
248 Photorespiration is thought to be essential for returning C back into the C_3 cycle, but our flux values

249   indicate that only 43% returns following serine export, instead of the theoretically maximal 75% under

250   21% O_2. This rate of carbon removal is (large) relative to other fluxes in central metabolism, comprising

251   50% and 62% the C flux of starch and sucrose syntheses at 21% O_2. This indicates that the C_3 cycle is

252   resilient against carbon removal and can replenish its own intermediates even under high rates of removal

253   following rubisco oxygenation. This is not surprising given the ability for the C_3 cycle to operate even

254   under high rates of rubisco oxygenation, like what occurs under 40% O_2 or elevated temperatures.

255   Conceptually this is similar to carbon removal from the C3 cycle to other metabolisms like sucrose and

256   starch. These findings favor the hypothesis that the major roles of photorespiration are to provide C and N

257   for downstream metabolism in addition to processing inhibitory byproducts of rubisco oxygenation. This

258   would help explain why it interacts with so many other aspects of metabolism. It may also explain why

259   photorespiration is maintained in all oxygenic phototrophs, and is even essential (albeit in reduced

260   amounts) in plant species with an efficient carbon concentrating mechanism in addition to its role in

261   processing 2PG [45].

262

263   **Role of glycine in photorespiration and *A* during transients**

264   Our metabolomics analyses identified a significant correlation between photorespiratory rate and glycine

265   pool size, which is large enough to explain much of the transient response of *A* to changing

266   photorespiratory conditions. The metabolically active glycine pools inferred from the dilution parameters

267   in our INST-MFA changed 15-fold when comparing low and high photorespiration (Fig. 4). Our gas

268   exchange data showed that during the low-to-high O_2 transient, the concomitant accumulation of active

269   glycine pools that are not decarboxylated accounted for ~25% of the excess CO_2 fixed (Fig. 5a).

270   Interestingly, the relaxation of active glycine pools accounts for a slightly smaller ~15% of the excess

271   CO_2 lost during the high-low O_2 transient (Fig. 5b). These findings indicate a unique connection between

272   the accumulation of a single metabolite and the subsequent effects to *A*. Further research is needed to

273   determine what the remaining determinants of the acclimation of net assimilation between

274   photorespiratory conditions are and how well these findings translate to other changes in photorespiratory

275   rate, like under fluctuating light [46].

276   These findings support a hypothesis as to the role of glycine in photorespiration as a "safe" buffering

277   metabolite in a dynamic pathway. The biochemical structure and kinetics of photorespiration may be

278   configured to accumulate glycine so that other upstream or downstream intermediates do not change too

279   rapidly or build up to inhibitory or cytotoxic concentrations. Photorespiratory intermediates downstream

280   of glycine were not significantly changed at varying photorespiratory conditions (Fig. 3). Many other

281 studies have shown that glycine is the most drastically changed photorespiratory intermediates upon

282 photorespiratory pressures [17,18,20,21,47], suggesting that glycine is a well-tolerated metabolite in plant cells.

283 The build-up of glycine may also suggest that photorespiration has been optimized to delay the release of

284 ammonium until there is sufficient glutamine synthetase/glutamate synthase activity to recapture the

285 ammonium released from the GDC reaction. In contrast, other photorespiratory intermediates are riskier

286 to accumulate. For example, the cytotoxicity of 2-PG, glycolate, glyoxylate, and serine suggests tight

287 regulation of their intracellular concentrations [45,48–52]. The toxicity of glycolate and glyoxylate may be

288 more nuanced however, since transformants expressing novel photorespiratory genes had ~3 and ~8 fold

289 higher concentrations of glycolate and glyoxylate with no obvious cytotoxic effects [53]. In addition to these

290 cytotoxic intermediates, serine may act as a signal for the transcriptional regulation of photorespiration-

291 related genes, as external supply of serine induces transcriptional deregulation so pool sizes would need to

292 be relatively buffered to prevent a hyper-sensitivity to short-term changes in photorespiratory flux [54].

293 Furthermore, accumulated glycerate could exert a feed-back effect on photosynthesis by blocking the

294 stromal fructose-1,6-bisphosphatase [55]. Of all the photorespiratory intermediates, glycine therefore

295 appears to be the most inert to metabolism and therefore a waypoint to help safely navigate dynamic

296 changes in flux.

297 The transient build-up in glycine pools when switching from low to high photorespiratory condition

298 indicates that the capacity of the downstream photorespiratory enzyme GDC may not be sufficient to

299 process large changes in glycine influx under non-steady state conditions. Limiting GDC activity even by

300 ~50% leads to a 100-fold accumulation of glycine during the day, while stimulating GDC activity leads to

301 a ~40% reduction in steady-state level of glycine [56,57]. Moreover, upregulating GDC activity improves

302 plant performance both in the laboratory and field conditions [57,58], suggesting its key role in controlling

303 photorespiratory flux. While GDC is normally not limited by its substrate glycine and NAD^+, the efficient

304 operation of GDC requires maintaining low levels of its product NADH, which may accumulate during

305 rapid changes in flux through GDC [59]. The mitochondrial TRX system has been proposed to be important

306 in the redox regulation of GDC under fluctuating light, adding to its potential role under transient

307 conditions[60,61].

308 An additional explanation for the increase in glycine pools relative to serine during increasing

309 photorespiration is the need to maintain appropriate concentration gradients to drive transport of these

310 metabolites between the mitochondrion and peroxisome. While these transporters are not yet identified,

311 they must facilitate large fluxes and are most likely driven by concentration gradients that are

312 proportional to their fluxes [62]. Glycine flux into the mitochondria is twice as large as serine flux out since

313 the GDC/serine hydroxymethyltransferase reactions require two glycine per serine produced and

314 assuming a "closed cycle" of photorespiration. Since higher transport gradients would translate to higher

315 leaf-level pool sizes of metabolites, the differences in flux may also help explain the high build-up of

316 glycine relative to serine.

317

Methods

319 **Plant material.** Tobacco (*Nicotiana tabacum*) plants were grown in a growth chamber under an 18-h

320 day/6-h night photoperiod at a light intensity of 100 μmol m^{-2} s^{-1}, day/night temperatures of 23°C/18°C,

321 and relative humidity of 60%. All experiments were performed on the fully expanded source leaves of 8-

322 week-old plants.

323 **Gas exchange and $^{13}CO_2$ labeling system.** Leaf gas exchange measurements under controlled $O_2/CO_2/N_2$

324 conditions were performed using a LI-6800 (LI-COR Biosciences, USA) with a fast atmospheric

325 switching and quenching system as described previously [33]. Leaves were equilibrated and monitored in

326 the cuvette at 40 Pa $^{12}CO_2$ levels with three O_2/N_2 conditions (2% O_2 / 98% N_2, 21% O_2/ 79% N_2, and

327 40% O_2/ 60 % N_2), light intensity of 100 μmol photon m^{-2} s^{-1}, leaf temperature of 25 °C, and relative

328 humidity of 70%. $^{13}CO_2$ labeling was performed after 20-30 min of acclimation to reach a photosynthetic

329 steady-state. Illuminated leaves were incubated at 0, 0.5, 1, 2, 2.5, 3, 5, 7, 10, 15, and 30 min with $^{13}CO_2$-

330 containing gas mixture ([40 Pa] $^{13}CO_2$ with each O_2/N_2 condition) before quenching. Pseudo-steady state

331 metabolism was assumed during the labeling period at each oxygen condition (Supplementary Figure 6).

332 Detailed descriptions of tissue harvesting and quenching for metabolic analyses are shown in

333 Supplementary Method 1.

334 **Metabolite labeling and concentration.** The extraction and analytical procedures for quantifying

335 metabolite labeling and concentration were similar to the method described previously [33]. Gas

336 chromatography-mass spectrometry (GC-MS) was used to analyze amino acids, organic acids, and sugars.

337 Ion-pair chromatography-tandem mass spectrometry (IPC-MS/MS) was used to analyze the

338 phosphorylated intermediates in C_3 cycle. Anion exchange chromatography-tandem mass spectrometry

339 (AEC-MS/MS) was used to analyze nucleotide sugars and additional phosphorylated intermediates.

340 Detailed descriptions of the extraction and mass spectrometry analyses are shown in Supplementary

341 Method 1 and Supplementary Data 6. Abbreviations of metabolites are shown in Supplementary Data 7.

342 **Measurements of external fluxes.** Rates of starch and sucrose synthesis were measured by the steady

343 state pulse-chase labeling of $^{14}CO_2$ as described previously [35] with slight modifications (Supplementary

344 Method 2). Rates of rubisco carboxylation and oxygenation were calculated as described earlier using gas

345   exchange measurements [2], with the required biochemical and diffusional parameters experimentally

346   estimated by gas exchange and isotope analysis (see Supplementary Method 3 for details). The output

347   flux ratios of amino acids and sucrose were estimated by measuring steady-state levels of sucrose and

348   amino acids in vascular exudate as described earlier [26,33]. Detailed descriptions of exudate collection were

349   described in Supplementary Method 1.

350   **Metabolic network and flux determination.** The network model used for flux determination was

351   modified from published reaction models [26,33,34] (Supplementary Data 1). INST-MFA was performed

352   using the elementary metabolite unit (EMU)-based software INCA (version 1.8) [63] ran within MATLAB

353   2018b (Mathworks, USA). Model formulation details are described in Supplementary Method 4 and the

354   INCA model files are included in Supplementary Data 8. Metabolic fluxes were estimated by iteratively

355   minimizing the sum of squared residuals (SSR) between the experimentally measured and simulated

356   isotopic labeling patterns and measured external rates, using a Levenberg-Marquardt optimization

357   algorithm [64]. Each flux estimation was repeated a minimum of 100 times from random initial values to

358   find best-fit estimates. All results were subjected to a χ^2-statistical test for goodness-of-fit and the 95%

359   confidence intervals were computed for all estimated fluxes using two independent methods. A parameter

360   continuation method was used to evaluate the sensitivity of SSR to flux variations [65]. A Monte Carlo

361   method was used to estimate flux distributions based on 3,000 sets of external flux measurements and

362   MID data perturbed by normally distributed noise with the measured SD. The parameter continuation and

363   Monte Carlo simulations were computed in parallel using a high-performance computing environment

364   (Institute for Cyber-Enabled Research at Michigan State University) and a cloud computing environment

365   (Microsoft Azure). The cloud-based workflow to run the Monte Carlo simulations can be found at

366   https://github.com/codexf/MFA_Cloud_Computing.

367   **Estimating photorespiratory C export using CO_2 response of A.** A/C_c curves were measured with a LI-

368   6800 at a light intensity of 1000 μmol photon $m^{-2} s^{-1}$ and a leaf temperature of 25 °C. The sequence of

369   reference CO_2 partial pressures was 40, 5, 7.5, 10, 15, 20, 25, 30, 35, 40, 45, 50, 60, 80, 100, and 120 Pa.

370   Photosynthetic parameters, including the C exported from photorespiration as glycine (α_G) or serine (α_S),

371   were estimated from the A/C_i curves using the modified Farquhar, von Caemmerer and Berry (FvCB)

372   biochemical model of photosynthesis fitted against rubisco, RuBP-regeneration, triose-phosphate use

373   limitations, α_G and (α_S) [15,66,67]. Data and custom scripts can be found at

374   https://github.com/codexf/Modeling_Photosynthesis.

375   **Transient response of A to changes in O_2 mole fraction.** Dynamic photosynthesis was recorded using a

376   LI-6800 with 2-s intervals. The leaves were illuminated (100 μmol $m^{-2} s^{-1}$) in 40 Pa CO_2 levels (CO_2

377   injector manually controlled) with either high O_2 (40% O_2 / 60% N_2) or low O_2 (2% O_2 / 98% N_2). When

378 the steady-state rate of A had been achieved, the cuvette was switched to low or high O_2 and leaves were

379 left to reach a new steady-state rate of A. The time course data were fitted using one-phase exponential

380 functions and the integrated area between the fitted curve and the baseline of the new steady-state A was

381 calculated to estimate the total C lag during the transients. Data and custom scripts can be found at

382 https://github.com/codexf/Modeling_Photosynthesis.

383

Data availability

385 All data used in this study are provided as Supporting Information of the article or available upon request.

386 Gas exchange data for modeling the photosynthetic responses to carbon dioxide concentration and

387 photosynthetic responses during oxygen transients can be found on GitHub at

388 https://github.com/codexf/Modeling_Photosynthesis.

389

Code availability

391 Custom scripts for modeling the photosynthetic responses to carbon dioxide concentration and

392 photosynthetic responses during oxygen transients can be found on GitHub at

393 https://github.com/codexf/Modeling_Photosynthesis. A cloud-based workflow to run the Monte Carlo

394 simulations for computing confidence intervals of metabolic fluxes can be found on GitHub at

395 https://github.com/codexf/MFA_Cloud_Computing.

396

417

Author contributions

419   BJW and XF designed the experiments. XF carried out the bulk of the experiments and analyzed the

420   results. SEW advised and assisted the $^{14}CO_2$ experiments to measure starch and sucrose partitioning. LG

421   performed the isotope analysis to measure gas exchange parameters. XF wrote the manuscript with

422   contributions from all the authors. BJW serves as the author responsible for contact and ensures

423   communication.

424

Competing interests

426   The authors declare no competing interests.

427

Figure legends

429   Fig. 1: Central carbon assimilatory metabolic fluxes in tobacco leaves exposed to varying O_2 conditions.

430   Fluxes were estimated by ^{13}C INST-MFA using the metabolic network model and experimental inputs

431   including mass isotopologue distributions of measured metabolites, the net CO_2 assimilation rate, rates of

432   rubisco carboxylation and oxygenation, rates of starch and sucrose synthesis, and output flux ratios of

433   amino acids and sucrose. Best-fit flux estimates are shown in numbers and corresponding variable width

434   of arrows. Arrows in blue indicate reactions constrained (not fixed) based on the experimental

435   measurements (Supplementary Table 1). Flux units are expressed as μmol molecule g^{-1} FW hr^{-1} (for the

436   conversion to μmol carbon g^{-1} FW hr^{-1}, multiply the presented values by the number of carbons in the

437   molecule). Metabolites compartmentalized to the plastid and cytosol are denoted by (".p") and (".c")

438   respectively. Abbreviations are shown in Supplementary Data 7.

439 Fig. 2: The response of A to the chloroplastic CO_2 concentration (Cc) in tobacco leaves exposed to 21%

440 O_2 under high light (1000 μmol m^{-2} s^{-1}). Measured A are presented as dots with error bars (mean ± SD, n

441 = 9 biologically independent leaves) fitted with the modified Farquhar, von Caemmerer and Berry

442 (FvCB) biochemical model of photosynthesis with two terms α_G and α_S, where α_G is the proportion of the

443 carbon exported from the photorespiratory pathway as glycine, and α_S the proportion of the carbon

444 exported from photorespiration as serine (see Method section: *Estimating photorespiratory C export*

445 *using CO_2 response of A*). The parameter values obtained from the individual fits are shown as mean ±

446 SD (n = 9 biologically independent leaves), including the maximum rubisco carboxylation rate (V_{cmax}),

447 the maximum rate of electron transport (J), the rate of triose phosphate export from the chloroplast (T_p),

448 α_G and α_S. The curves represent Rubisco (red), RuBP (green), and TPU (orange) limitation.

449 Fig. 3: Metabolic pool size alterations in response to photorespiratory conditions. Tobacco leaves were

450 exposed to the three O_2/N_2 conditions (all at 40 Pa $^{12}CO_2$ levels) in the LI-6800 cuvette for 30 min until

451 quenching with liquid nitrogen under illumination. Leaves acclimated to 2% O_2 and 40% O_2 were

452 compared to the ambient O_2 condition respectively and the steady-state pool sizes of the identified

453 metabolites (plotted on the y-axis) were determined using GC-MS and LC-MS/MS. The x-axis plots the

454 log2-transformed relative ratio of abundance of each metabolite in the leaves exposed to 2% O_2 (blue) or

455 40% O_2 (red) normalized to the same metabolite in the leaves exposed to 21% O_2. Results are presented as

456 dots with error bars (mean ± SEM, n = 7 or 8 biologically independent leaves). Asterisk represents a

457 significant difference between two conditions (Benjamini-Hochberg procedure was used to control the

458 false discovery rate at 0.05 for multiple *t*-tests including all metabolites). Supplementary Data 5 presents

459 the absolute concentration of each metabolite under different oxygen conditions with the results of one-

460 way ANOVA followed by Tukey's test for multiple pairwise comparisons among the three oxygen

461 conditions per metabolite. Abbreviations of metabolites are shown in Supplementary Data 7.

462 Fig. 4: Active and inactive pools of glycine at varying O_2 levels. Tobacco leaves were exposed to the

463 three O_2/N_2 conditions (all at 40 Pa $^{12}CO_2$ levels) in the LI-6800 cuvette for 30 min until quenching with

464 liquid nitrogen under illumination. Steady-state pool sizes of glycine were determined using GC-MS. The

465 proportion of active (gray) and inactive (white) pools of glycine were estimated using the dilution

466 parameters in INST-MFA modeling. Data are presented as box plots (center line at the median, upper

467 bound at 75th percentile, lower bound at 25th percentile) with whiskers extended to the extreme data

468 point ≤ 1.5 × interquartile range from the edge of the box (n = 7 or 8 biologically independent leaves).

469 Different letters above the data boxes denote statistically significant differences among the three

470 conditions for active glycine pools (uppercase letters) and inactive glycine pools (lowercase letters) at $P <$

471 0.05 (one-way ANOVA followed by Tukey's test for multiple pairwise comparisons). For active glycine

472   pools, the P values for Tukey multiple comparisons of means are 0.0006 (2% O2 - 21% O2), <0.0001

473   (2% O2 - 40% O2), and 0.0042 (21% O2 - 40% O2). For inactive glycine pools, the P values for Tukey

474   multiple comparisons of means are 0.0260 (2% O2 - 21% O2), 0.0138 (2% O2 - 40% O2), and 0.9511

475   (21% O2 - 40% O2).

476   Fig. 5: Dynamics of net CO_2 assimilation rate (A) during oxygen transients from 2% O_2 to 40% O_2 (a) and

477   from 40% O_2 to 2% O_2 (b). The leaves were acclimated in a LI-6800 cuvette at low O_2 or high O_2 for 30

478   min to reach the photosynthetic steady-state. Upon switching to the opposite oxygen condition, A changed

479   drastically due to slow gas mixing and returned to the previous steady state within ~10 s. The time when

480   the previous steady state was reached after switching oxygen was set to zero for the convenience of curve

481   fitting. Data points of different colors correspond to independent time-course measurements on ten

482   different plants. Lines of different colors represent the fitted curves of a one-phase exponential function

483   for individual replicates. Integrated area between the fitted curve and the baseline of new steady state for

484   a represented replicate is presented in the inserts. The integrated area obtained from the ten individual fits

485   are shown as mean ± SD (n = 10 biologically independent leaves).

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643

1 **Integrated flux and pool size analysis in plant central metabolism reveals unique roles of glycine**

2 **and serine during photorespiration**

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9

10 **Abstract**

11 Photorespiration is an essential process juxtaposed between plant carbon and nitrogen metabolism that

12 responds to dynamic environments. Photorespiration recycles inhibitory intermediates arising from

13 oxygenation reactions catalyzed by rubisco back into the C_3 cycle, but it is unclear what proportions of its

14 nitrogen-containing intermediates (glycine and serine) are exported into other metabolisms *in vivo* and

15 how these pool sizes affect net CO_2 gas exchange during photorespiratory transients. To address this

16 uncertainty, we measured rates of amino acid export from photorespiration using isotopically

17 nonstationary metabolic flux analysis. This analysis revealed that ~23-41% of the photorespiratory carbon

18 was exported from the pathway as serine under various photorespiratory conditions. Furthermore, we

19 determined that the build-up and relaxation of glycine pools constrained a large portion of photosynthetic

20 acclimation during photorespiratory transients. These results reveal the unique and important roles of

21 glycine and serine in successfully maintaining various photorespiratory fluxes that occur under

22 environmental fluctuations in nature and providing carbon and nitrogen for metabolism.

Main

In C_3 plants, photorespiration is a high-flux metabolic pathway that recycles the inhibitory 2-phosphoglycolate (2PG) produced from the oxygenation reaction catalyzed by ribulose-1,5-bisphosphate carboxylase-oxygenase (rubisco). The recycling of 2PG to the C_3 intermediate, 3-phosphoglycerate (3PGA) involves a series of energy-consuming reactions across multiple organelles [1,2]. Upon cursory consideration, photorespiration is often viewed as a wasteful process, because the conversion from glycine to serine releases a large amount of CO_2 and ammonia, the former is released from photorespiration and the latter is re-assimilated with high-energy cost. However, besides the beneficial removal of 2PG, photorespiration also plays a vital role in central metabolism due to its close interaction with many other pathways, especially nitrate assimilation [3–6]. Although the interaction between photorespiration and nitrate assimilation is not mechanistically clear, nitrate assimilation depends to some degree on photorespiration [5,7]. For example, plants grown under elevated CO_2 partial pressures with decreased photorespiration have decreased nitrate assimilation into organic nitrogen (N) compounds and total N concentration of plant tissues [8,9]. Additionally, the major route of ammonia assimilation, glutamate synthesis, increases as photorespiration increases possibly not only facilitating increased rates of ammonia recapture from photorespiration but also increased rates of *de novo* nitrate assimilation [10,11].

The mechanism linking photorespiration to nitrate assimilation could be related to the export of N-rich amino acids from photorespiration. Due to the high metabolic flux channeled through photorespiration, even small changes in photorespiratory flux could affect the flux distribution through central carbon (C) and N metabolism. The extent to which photorespiration operates as a closed cycle to regenerate 3-PGA for the C_3 cycle or as an open cycle allowing metabolite export to interacting metabolisms remains unknown [12]. Exporting glycine and serine from photorespiration into non-photorespiratory metabolism (*e.g.,* protein synthesis, providing amino groups to alternative metabolic processes or transport) instead of returning their C to the chloroplast as 3PGA additionally affects the CO_2 loss and the energy costs associated with recycling of 2PG into 3PGA.

Evidence for a fraction of photorespiratory C exported from photorespiration as glycine and/or serine is revealed in biochemical modeling of leaf photosynthesis [13]. The export of photorespiratory C as amino acids can explain the decreased rates of net CO_2 assimilation (*A*) in response to CO_2 and O_2 at high CO_2 partial pressures, typically 2-4 times ambient conditions and saturating illumination [13]. In this paper we define glycine and serine export as amino acids that leave photorespiratory metabolism, this includes amino acids that leave the leaf entirely, as in phloem exudate, but also to amino acids that leave photorespiration to participate in non-photorespiratory metabolism in the same leaf tissue. The fraction of photorespiratory C exported as amino acids estimated using photosynthetic models ranges between 0.3-

56   0.77 under the high CO_2 partial pressures where this approach can be used, with serine being the primary

57   amino acid exported [14,15]. Exporting glycine and serine from photorespiration increases photosynthesis at

58   high CO_2 partial pressures when carbon assimilation is limited by the availability of inorganic phosphate

59   in the chloroplast [16].

60   Exporting glycine and serine from photorespiration is also coupled to the N flux to these amino acids.

61   Specifically, this N flux out of photorespiration in the form of amino acids must be supplemented by

62   increasing the *de novo* assimilation of N to replace the amino groups required for glyoxylate amination to

63   glycine, potentially explaining why the estimated fraction of amino acid export increases with increasing

64   NO_3^- availability [15]. Despite this compelling evidence for serine and glycine export under high CO_2

65   partial pressures when photorespiration is limited, it is unclear if high rates of amino acid export happen

66   under more physiologically relevant CO_2 partial pressures and forms a regular component of central

67   metabolism under more typical growing conditions. C export from photorespiration also raises interesting

68   questions concerning the functional importance of photorespiration, since this export would not replenish

69   C lost from the C_3 cycle following rubisco oxygenation.

70   Another uncertainty of photorespiration is the speed of its metabolic response and subsequent impact on

71   A. Photorespiration is a dynamic process, responding to A, stomatal conductance, and light regimes.

72   However, it is mostly investigated under steady-state conditions by comparing the metabolomic profile

73   and physiology following different environmental and genetic perturbations [17–19]. Under non-steady-state

74   conditions, amounts of photorespiratory intermediates are sensitive to photorespiratory conditions both in

75   the long- and short-term [17–19]. Photorespiratory intermediates additionally accumulate during transient

76   changes in rubisco oxygenation rate or when downstream photorespiratory reactions are not

77   stoichiometrically balanced [20–24]. When photorespiratory intermediates upstream of the glycine

78   decarboxylase (GDC) reaction like glycine accumulate, the CO_2 release per rubisco oxygenation should

79   decrease proportionally, increasing subsequent A during this accumulation. Despite this theoretical

80   underpinning, it is not clear if the change in glycine pools is large enough to affect A to a measurable

81   degree during transients in photorespiratory rates.

82   The use of isotopic tracers and metabolic flux analysis (MFA) are powerful tools to resolve how

83   metabolism responds to environmental changes [25,26]. For example, isotopic labeling with $^{13}CO_2$ in

84   combination with varied CO_2 and/or O_2 mole fraction has been commonly used to infer the impact of

85   photorespiration on leaf C metabolism. Under different CO_2/O_2 conditions, distinct ^{13}C labeling

86   enrichment has been observed in many primary and specialized metabolites [20,27], as well as proteins [28,29].

87   These studies were usually performed at isotopic pseudo-steady state with a single time-point, which does

88   not allow flux quantification in autotrophic metabolism [30]. Recent development of isotopically

89   nonstationary metabolic flux analysis (INST-MFA) more fully leverages the benefits of $^{13}CO_2$ labeling to

90   provide a global assessment of C fluxes through autotrophic leaf metabolism [26,31–34]. INST-MFA is

91   therefore a promising approach for estimating fluxes and metabolic responses to changes in

92   photorespiratory rates *in vivo*.

93   Here, INST-MFA was applied to examine the C fluxes through and out of the photorespiratory pathway

94   under different photorespiratory conditions manipulated by the O_2 mole fraction. This INST-MFA

95   analysis gave a quantitative estimate of serine and glycine exported from photorespiration under steady-

96   state conditions at ambient CO_2 partial pressures, which agreed with estimates made under high CO_2

97   partial pressures using estimates from CO_2 response curves. Metabolite pool size measurements of 38

98   central metabolites showed that glycine had the strongest changes in response to photorespiratory shifts.

99   We also determined that changes in glycine concentration were large enough to explain 20-30% of the

100   gain in net CO_2 assimilation when transitioning from low to high photorespiratory conditions.

101

Results

Isotopically Nonstationary Metabolic Flux Analysis

104   To estimate C flux distribution on central metabolism, transient ^{13}C labeling experiments were performed

105   on fully expanded *Nicotiana tabacum* (tobacco) leaves after short-term acclimation to three O_2 conditions

106   (2%, 21%, and 40%). The labeling experiments were carried out at the same irradiance as the growth

107   chamber (100 µmol m^{-2} s^{-1}) to minimize changes in metabolism due to light acclimation. Metabolic

108   fluxes through central metabolism were determined from a metabolic network model (Supplementary

109   Data 1), mass isotopologue distributions (MIDs) during the transient labeling period (Supplementary Data

110   2), and measured input and output fluxes to constrain the flux solutions. The independently determined

111   input and output fluxes used to constrain the flux solution included the *A*, rubisco oxygenation rate (v_o),

112   rubisco carboxylation rate (v_c), starch and sucrose production rates, and output flux ratios of amino acids

113   and sucrose measured in vascular exudate (Supplementary Table 1 and 2 and Supplementary Method 1-

114   4). INST-MFA was performed under the assumption that the metabolism is steady-state (Supplementary

115   Method 4 with steady-state assumption validated below). The selection of labeling data for INST-MFA

116   modeling and tests for several network models are described in Supplementary Method 4. A subset of key

117   measured labeling data and the labeling kinetics simulated by the best-fit INST-MFA flux values are

118   shown in Supplementary Figure 1 and the detailed fits of the best-fit models are shown in Supplementary

119   Data 3.

120   Manipulating O_2 condition had a large and significant effect on v_o (2.60-26.37 µmol g^{-1} FW hr^{-1}) and a

121   smaller but significant effect on v_c (53.36-79.70 µmol g^{-1} FW hr^{-1}, Fig. 1, 95% confidence intervals of the

122   best-fit fluxes are shown in Supplementary Figure 2 and 3 and Supplementary Data 4). The rate of starch

123   and sucrose synthesis decreased significantly with increasing rates of photorespiration (Fig. 1 and

124   Supplementary Data 4). Direct measurement of the allocation of photo-assimilated C using $^{14}CO_2$

125   labelling showed that rates of starch and sucrose synthesis are strongly correlated with A (Supplementary

126   Figure 4 and Supplementary Method 2).

127   The INST-MFA revealed the export rates of photorespiratory serine and glycine from the

128   photorespiratory pathway at ambient CO_2 partial pressures (Fig. 1). The serine export rate to alternative

129   sinks increased from 3.61 to 7.02 µmol g^{-1} FW hr^{-1} at high rates of photorespiration, but it decreased to

130   0.02 µmol g^{-1} FW hr^{-1} at low photorespiration rates. The export of glycine to alternative sinks was less

131   than 0.01 µmol g^{-1} FW hr^{-1} at three conditions, much lower than that of serine. A portion of the export of

132   amino acids from photorespiration, the removal of glycine or serine through phloem, was estimated based

133   on the output flux to phloem and the output flux ratios of amino acids and sucrose measured in vascular

134   exudate (Supplementary Table 2 and Supplementary Method 4). The removal rates of glycine or serine

135   through the phloem were small relative to the total export, between 0.03 to 0.38 µmol g^{-1} FW hr^{-1} at

136   various photorespiratory conditions (Fig. 1). The total amount of C leaving the photorespiratory pathway

137   as serine scaled with the rate of photorespiration. At high photorespiration (40% O_2), 41% of the

138   photorespiratory C influx (52.8 µmol C g^{-1} FW hr^{-1}) was exported as serine (at a rate of 21.5 µmol C g^{-1}

139   FW hr^{-1}). The percentage of photorespiratory C leaving as serine was 32% at normal photorespiration

140   (21% O_2) and 23% at low photorespiration (2% O_2). These data indicate that 23-41% of the

141   photorespiratory C could be exported from the pathway as serine under physiologically relevant CO_2

142   partial pressures. The INST-MFA model was also run with glycerate labelling data and allowing glycerate

143   export from the photorespiratory pathway. This extended model predicted that glycerate and glycine

144   export rate was slow (< 0.1 µmol g^{-1} FW hr^{-1} in all oxygen conditions), while the serine export rate was

145   slow at low photorespiration and rose at normal and high photorespiration (<0.1, 3.7 and 7.0 µmol g^{-1} FW

146   hr^{-1} in 2%, 21%, and 40% O_2 respectively) (see Supplemental Method 4).

147   **Estimating photorespiratory C export using CO_2 response of A**

148   The export of photorespiratory C to amino acid synthesis or alternative metabolism accounts for the

149   behavior of A at high CO_2 partial pressures in leaf photosynthesis models [13,15]. Using this approach, we

150   estimated the fraction of C that leaves the photorespiratory pathway as glycine (α_G) and serine (α_S) using

151   the measured CO_2 response of A at ambient O_2 and high light (~1000 µmol $m^{-2}\,s^{-1}$) fitted with the

152   photosynthetic model over the triose phosphate utilization (TPU)-limited range to compare to the INST-

153   MFA results (Fig. 2). The high-CO_2 portion of the curve in orange, where TPU limits CO_2 assimilation,

154   showed the characteristic curve that is only explained by amino acid removal from photorespiration [13,35].

155   This curvature was used to fit a resultant α_G and α_S of 0.01 and 0.45, indicating that the C flow out of

156   photorespiration was primarily in the form of serine, consistent with the estimates ($\alpha_G < 0.01$ and $\alpha_S =$

157   0.32) calculated from the INST-MFA (Fig. 1).

Metabolic responses to low and high photorespiration

159   The absolute pool sizes of 38 central metabolic intermediates were measured in tobacco leaves exposed to

160   varied photorespiratory conditions. Compared to the ambient O_2 condition, the metabolic changes at high

161   photorespiration (40% O_2) and low photorespiration (2% O_2) were modest (Fig. 3). Most metabolites had

162   insignificant changes in pool size during the short-term photorespiratory changes (Fig. 3 shows the mean

163   fold-changes of the pool size between 2% and 21% O_2 or between 40% and 21% O_2; Supplementary Data

164   5 shows the absolute concentrations under the three oxygen conditions). The most significantly altered

165   metabolites were photorespiratory intermediates (glycine, glycolate and glyoxylate), alanine, and ADPG

166   (Supplementary Data 5). Many photorespiratory intermediates (glycine, serine, glycerate, glycolate and

167   glyoxylate) were far from fully labelled even at high photorespiratory condition (Supplementary Figure

168   5). The average ^{13}C enrichment of these metabolites reached between 10% and 75% following 30 min of

169     $^{13}CO_2$ labeling (Supplementary Figure 5), indicating that these intermediates exist in multiple subcellular

170   pools with an active pool being involved in photorespiration and other pools (presumably in the vacuole

171   [25]) not involved in photorespiration. The active and inactive pools of glycine were estimated using the

172   measured total pool size and the dilution parameters from the INST-MFA (Supplementary Data 4 and 5).

173   The extent of changes in glycine was much greater in the active pool (15-fold) than in the total pool (4-

174   fold) (Fig. 4). These data support the steady-state assumption of INST-MFA (Supplementary Figure 6).

175   These data highlight the value of combining metabolite abundance with labeling and demonstrate the

176   principle that in many cases changes in metabolite pool size do not directly translate into flux changes.

177   For example, the pool size of 2PG was not significantly different between different oxygen conditions

178   (Fig. 3 and Supplementary Data 5), but the ^{13}C labeling kinetics of 2PG were faster in high oxygen than in

179   low oxygen (Supplementary Figure 1 and Supplementary Data 2), suggesting higher flux through rubisco

180   oxygenation. This highlights that time-dependent labeling profiles are more informative indicators for

181   flux rather than changes in pool sizes and that inferring flux from pool sizes alone can lead to inaccurate

182   interpretations.

Photorespiration under non-steady state

184   Quantifying the steady-state active pool size of glycine at low and high photorespiratory conditions

185   helped determine how much the size of glycine pools constrain net gas exchange during photorespiratory

186   transients. To do this, A during O_2 transients was measured when tobacco leaves were exposed to 2% O_2

187   until the steady-state and then switched to 40% O_2. During the low-to-high O_2 transient, A did not

188   instantly reach a new steady state at 40% O_2 but gradually declined at an exponential rate for ~2 min until

189   the new steady-state was reached (Fig. 5). The extra C gained or lost during the transient in the decrease

190   (or increase) in A as compared to how much C would be fixed with an instant transition to the new steady

191   state was defined as the "C lag", which can be quantified by integrating the area between the fitted curve

192   of A and a baseline of A at the new steady state (Fig. 5a). During the transition from 2% O_2 to 40% O_2, the

193   total glycine pool increased by 30.8 ± 2.4 µmol m^{-2} based on the differences in steady-state levels (after

194   30 min acclimation to high or low oxygen) between high and low oxygen levels (Fig 4) or 20.8 ± 6.2

195   µmol m^{-2} based on the initial build-up of glycine pool after switching oxygen for 60 s when the gas

196   exchange transient is near complete (Supplementary Figure 7), accounting for 20-30% of the total C lag

197   (50.4 ± 1.7 µmol m^{-2}) as the accumulated glycine would otherwise be metabolized by the downstream

198   glycine decarboxylase reaction (2 mol of glycine yielding 1 mol of CO_2). The build-up of the active

199   glycine pool based on the steady-state measurement (25.7 ± 3.4 µmol m^{-2}, Fig 4) accounted for ~25% of

200   the total C lag. When looking at the transition from 40% O_2 to 2% O_2, the lag in the photosynthetic

201   increase resulted in loss of C represented by the shaded area (Fig. 5b). The reduction of the active glycine

202   pool based on the steady-state measurement explained ~15% of this total C lag (83.9 ± 3.8 µmol m^{-2}).

203   Note that the change in concentrations in response to O_2 treatment is negligible and small in other

204   metabolites upstream of glycine (i.e. 2PG, glycolate and glyoxylate, Fig. 5), further indicating glycine is

205   the primary metabolic pool impacting this transient response.

206

207   **Discussion**

208   **Impact of photorespiratory flux changes on C and N metabolism**

209   Our INST-MFA indicates that C was exported from photorespiration under ambient CO_2 partial pressures

210   as serine at high rates. The INST-MFA-estimated fraction of serine export is 0.27-0.39 (95% confidence

211   interval using the parameter continuation method) under ambient O_2 and CO_2 partial pressures

212   (Supplementary Figure 2 and Supplementary Data 4), which is close to the 95% confidence interval of the

213   value estimated using photosynthetic models fitted at high CO_2 partial pressures (0.38-0.51) (Fig. 2).

214   Given these large export rates, where do these amino acids go? Prior studies have shown that glycine is

215   incorporated into leaf protein when photorespiration is induced by low CO_2 [28,36], and almost all

216   photorespiratory glycine is routed to protein synthesis in water-stressed soybean leaves under high light

217   and low CO_2 partial pressure [29]. Protein synthesis is not likely to explain all the amino acid export from

218   photorespiration, since glycine and serine are present in similar proportions in total plant protein, but

219   serine is exported at a much higher rate [37]. Past work also suggests that at least some photorespiratory

220   serine and glycine is exported out of source leaves under photorespiratory conditions as is evident by an

221   increased labeling in serine and glycine within the petiole when source leaves were labeled with $^{14}CO_2$

222   under elevated O_2 [38], but our vascular exudates indicated that removal from the leaf only accounted for a

223   small percentage of the total amino acids exported from photorespiration. Photorespiratory glycine and

224   serine are also required for the synthesis of various essential metabolites. For example. serine provides the

225   carbon backbone for the synthesis of cysteine and tryptophan, which are further used to produce sulphur-

226   containing compounds (*e.g.* S-adenosylmethionine) and indolic compounds (*e.g.* auxin and

227   glucosinolates) [39,40]. Serine also is an important substrate for the biosynthesis of phosphatidylserine and

228   sphingolipid that are critical to plant development and survival [41]. Glycine enhances the synthesis of

229   glutathione (γ-Glu–Cys–Gly) as suggested by previous leaf feeding experiments [42]. Taken together, these

230   findings indicate that photorespiration is an important source of amino acids for protein and metabolite

231   synthesis and future work should work to resolve where specifically this serine goes within metabolism.

232   The increased fraction of photorespiratory C exported as amino acids with increased photorespiration in

233   this study also helps explain the mechanistic link between photorespiration and N metabolism [5,43]. Large

234   serine export fluxes diverted from photorespiration would need to be supplemented by equivalent *de novo*

235   N assimilation. The *de novo* N assimilation rate would need to increase to support a higher total amount

236   of photorespiratory C leaving as amino acids under higher photorespiratory condition. Supporting this

237   hypothesis, plants under ambient photorespiration had ~10-30% higher N assimilation rates than those

238   under low photorespiration (high CO_2 or low O_2) [8]. The coupling of N assimilation with photorespiration-

239   derived glycine and serine has also been suggested from a dual labeling study with $^{13}CO_2$ and ^{15}N-

240   ammonium nitrate [44]. In this 6-h labeling study, up to 50% of ^{15}N glycine and ^{15}N serine molecules were

241   also ^{13}C enriched, probably due to large turnover glycine and serine associated with high flux rates of

242   photorespiration, whereas many other amino acids were only ^{15}N-labelled because their ^{15}N-amino groups

243   were not fixed onto ^{13}C-enriched C skeletons. Exporting glycine and serine from the photorespiratory

244   pathway could enable plants to efficiently assimilate N onto C skeletons produced during

245   photorespiration, also stimulating CO_2 uptake at high CO_2 [15].

246   The finding that large amounts of C leave photorespiration for other metabolisms and do not re-enter the

247   C_3 cycle suggests some interesting hypotheses concerning the central purpose of photorespiration.

248   Photorespiration is thought to be essential for returning C back into the C_3 cycle, but our flux values

249   indicate that only 43% returns following serine export, instead of the theoretically maximal 75% under

250   21% O_2. This rate of carbon removal is (large) relative to other fluxes in central metabolism, comprising

251   50% and 62% the C flux of starch and sucrose syntheses at 21% O_2. This indicates that the C_3 cycle is

252   resilient against carbon removal and can replenish its own intermediates even under high rates of removal

253   following rubisco oxygenation. This is not surprising given the ability for the C_3 cycle to operate even

254   under high rates of rubisco oxygenation, like what occurs under 40% O_2 or elevated temperatures.

255   Conceptually this is similar to carbon removal from the C3 cycle to other metabolisms like sucrose and

256   starch. These findings favor the hypothesis that the major roles of photorespiration are to provide C and N

257   for downstream metabolism in addition to processing inhibitory byproducts of rubisco oxygenation. This

258   would help explain why it interacts with so many other aspects of metabolism. It may also explain why

259   photorespiration is maintained in all oxygenic phototrophs, and is even essential (albeit in reduced

260   amounts) in plant species with an efficient carbon concentrating mechanism in addition to its role in

261   processing 2PG [45].

262

263   **Role of glycine in photorespiration and A during transients**

264   Our metabolomics analyses identified a significant correlation between photorespiratory rate and glycine

265   pool size, which is large enough to explain much of the transient response of A to changing

266   photorespiratory conditions. The metabolically active glycine pools inferred from the dilution parameters

267   in our INST-MFA changed 15-fold when comparing low and high photorespiration (Fig. 4). Our gas

268   exchange data showed that during the low-to-high O_2 transient, the concomitant accumulation of active

269   glycine pools that are not decarboxylated accounted for ~25% of the excess CO_2 fixed (Fig. 5a).

270   Interestingly, the relaxation of active glycine pools accounts for a slightly smaller ~15% of the excess

271   CO_2 lost during the high-low O_2 transient (Fig. 5b). These findings indicate a unique connection between

272   the accumulation of a single metabolite and the subsequent effects to A. Further research is needed to

273   determine what the remaining determinants of the acclimation of net assimilation between

274   photorespiratory conditions are and how well these findings translate to other changes in photorespiratory

275   rate, like under fluctuating light [46].

276   These findings support a hypothesis as to the role of glycine in photorespiration as a "safe" buffering

277   metabolite in a dynamic pathway. The biochemical structure and kinetics of photorespiration may be

278   configured to accumulate glycine so that other upstream or downstream intermediates do not change too

279   rapidly or build up to inhibitory or cytotoxic concentrations. Photorespiratory intermediates downstream

280   of glycine were not significantly changed at varying photorespiratory conditions (Fig. 3). Many other

281 studies have shown that glycine is the most drastically changed photorespiratory intermediates upon
282 photorespiratory pressures [17,18,20,21,47], suggesting that glycine is a well-tolerated metabolite in plant cells.
283 The build-up of glycine may also suggest that photorespiration has been optimized to delay the release of
284 ammonium until there is sufficient glutamine synthetase/glutamate synthase activity to recapture the
285 ammonium released from the GDC reaction. In contrast, other photorespiratory intermediates are riskier
286 to accumulate. For example, the cytotoxicity of 2-PG, glycolate, glyoxylate, and serine suggests tight
287 regulation of their intracellular concentrations [45,48–52]. The toxicity of glycolate and glyoxylate may be
288 more nuanced however, since transformants expressing novel photorespiratory genes had ~3 and ~8 fold
289 higher concentrations of glycolate and glyoxylate with no obvious cytotoxic effects [53]. In addition to these
290 cytotoxic intermediates, serine may act as a signal for the transcriptional regulation of photorespiration-
291 related genes, as external supply of serine induces transcriptional deregulation so pool sizes would need to
292 be relatively buffered to prevent a hyper-sensitivity to short-term changes in photorespiratory flux [54].
293 Furthermore, accumulated glycerate could exert a feed-back effect on photosynthesis by blocking the
294 stromal fructose-1,6-bisphosphatase [55]. Of all the photorespiratory intermediates, glycine therefore
295 appears to be the most inert to metabolism and therefore a waypoint to help safely navigate dynamic
296 changes in flux.

297 The transient build-up in glycine pools when switching from low to high photorespiratory condition
298 indicates that the capacity of the downstream photorespiratory enzyme GDC may not be sufficient to
299 process large changes in glycine influx under non-steady state conditions. Limiting GDC activity even by
300 ~50% leads to a 100-fold accumulation of glycine during the day, while stimulating GDC activity leads to
301 a ~40% reduction in steady-state level of glycine [56,57]. Moreover, upregulating GDC activity improves
302 plant performance both in the laboratory and field conditions [57,58], suggesting its key role in controlling
303 photorespiratory flux. While GDC is normally not limited by its substrate glycine and NAD^+, the efficient
304 operation of GDC requires maintaining low levels of its product NADH, which may accumulate during
305 rapid changes in flux through GDC [59]. The mitochondrial TRX system has been proposed to be important
306 in the redox regulation of GDC under fluctuating light, adding to its potential role under transient
307 conditions[60,61].

308 An additional explanation for the increase in glycine pools relative to serine during increasing
309 photorespiration is the need to maintain appropriate concentration gradients to drive transport of these
310 metabolites between the mitochondrion and peroxisome. While these transporters are not yet identified,
311 they must facilitate large fluxes and are most likely driven by concentration gradients that are
312 proportional to their fluxes [62]. Glycine flux into the mitochondria is twice as large as serine flux out since
313 the GDC/serine hydroxymethyltransferase reactions require two glycine per serine produced and

assuming a "closed cycle" of photorespiration. Since higher transport gradients would translate to higher leaf-level pool sizes of metabolites, the differences in flux may also help explain the high build-up of glycine relative to serine.

Methods

Plant material. Tobacco (*Nicotiana tabacum*) plants were grown in a growth chamber under an 18-h day/6-h night photoperiod at a light intensity of 100 μmol m^{-2} s^{-1}, day/night temperatures of 23°C/18°C, and relative humidity of 60%. All experiments were performed on the fully expanded source leaves of 8-week-old plants.

Gas exchange and $^{13}CO_2$ labeling system. Leaf gas exchange measurements under controlled $O_2/CO_2/N_2$ conditions were performed using a LI-6800 (LI-COR Biosciences, USA) with a fast atmospheric switching and quenching system as described previously [33]. Leaves were equilibrated and monitored in the cuvette at 40 Pa $^{12}CO_2$ levels with three O_2/N_2 conditions (2% O_2 / 98% N_2, 21% O_2/ 79% N_2, and 40% O_2/ 60 % N_2), light intensity of 100 μmol photon m^{-2} s^{-1}, leaf temperature of 25 °C, and relative humidity of 70%. $^{13}CO_2$ labeling was performed after 20-30 min of acclimation to reach a photosynthetic steady-state. Illuminated leaves were incubated at 0, 0.5, 1, 2, 2.5, 3, 5, 7, 10, 15, and 30 min with $^{13}CO_2$-containing gas mixture ([40 Pa] $^{13}CO_2$ with each O_2/N_2 condition) before quenching. Pseudo-steady state metabolism was assumed during the labeling period at each oxygen condition (Supplementary Figure 6). Detailed descriptions of tissue harvesting and quenching for metabolic analyses are shown in Supplementary Method 1.

Metabolite labeling and concentration. The extraction and analytical procedures for quantifying metabolite labeling and concentration were similar to the method described previously [33]. Gas chromatography-mass spectrometry (GC-MS) was used to analyze amino acids, organic acids, and sugars. Ion-pair chromatography-tandem mass spectrometry (IPC-MS/MS) was used to analyze the phosphorylated intermediates in C_3 cycle. Anion exchange chromatography-tandem mass spectrometry (AEC-MS/MS) was used to analyze nucleotide sugars and additional phosphorylated intermediates. Detailed descriptions of the extraction and mass spectrometry analyses are shown in Supplementary Method 1 and Supplementary Data 6. Abbreviations of metabolites are shown in Supplementary Data 7.

Measurements of external fluxes. Rates of starch and sucrose synthesis were measured by the steady state pulse-chase labeling of $^{14}CO_2$ as described previously [35] with slight modifications (Supplementary Method 2). Rates of rubisco carboxylation and oxygenation were calculated as described earlier using gas

345 exchange measurements [2], with the required biochemical and diffusional parameters experimentally

346 estimated by gas exchange and isotope analysis (see Supplementary Method 3 for details). The output

347 flux ratios of amino acids and sucrose were estimated by measuring steady-state levels of sucrose and

348 amino acids in vascular exudate as described earlier [26,33]. Detailed descriptions of exudate collection were

349 described in Supplementary Method 1.

350 **Metabolic network and flux determination.** The network model used for flux determination was

351 modified from published reaction models [26,33,34] (Supplementary Data 1). INST-MFA was performed

352 using the elementary metabolite unit (EMU)-based software INCA (version 1.8) [63] ran within MATLAB

353 2018b (Mathworks, USA). Model formulation details are described in Supplementary Method 4 and the

354 INCA model files are included in Supplementary Data 8. Metabolic fluxes were estimated by iteratively

355 minimizing the sum of squared residuals (SSR) between the experimentally measured and simulated

356 isotopic labeling patterns and measured external rates, using a Levenberg-Marquardt optimization

357 algorithm [64]. Each flux estimation was repeated a minimum of 100 times from random initial values to

358 find best-fit estimates. All results were subjected to a χ^2-statistical test for goodness-of-fit and the 95%

359 confidence intervals were computed for all estimated fluxes using two independent methods. A parameter

360 continuation method was used to evaluate the sensitivity of SSR to flux variations [65]. A Monte Carlo

361 method was used to estimate flux distributions based on 3,000 sets of external flux measurements and

362 MID data perturbed by normally distributed noise with the measured SD. The parameter continuation and

363 Monte Carlo simulations were computed in parallel using a high-performance computing environment

364 (Institute for Cyber-Enabled Research at Michigan State University) and a cloud computing environment

365 (Microsoft Azure). The cloud-based workflow to run the Monte Carlo simulations can be found at

366 https://github.com/codexf/MFA_Cloud_Computing.

367 **Estimating photorespiratory C export using CO_2 response of A.** A/C_c curves were measured with a LI-

368 6800 at a light intensity of 1000 μmol photon $m^{-2} s^{-1}$ and a leaf temperature of 25 °C. The sequence of

369 reference CO_2 partial pressures was 40, 5, 7.5, 10, 15, 20, 25, 30, 35, 40, 45, 50, 60, 80, 100, and 120 Pa.

370 Photosynthetic parameters, including the C exported from photorespiration as glycine (α_G) or serine (α_S),

371 were estimated from the A/C_i curves using the modified Farquhar, von Caemmerer and Berry (FvCB)

372 biochemical model of photosynthesis fitted against rubisco, RuBP-regeneration, triose-phosphate use

373 limitations, α_G and (α_S) [15,66,67]. Data and custom scripts can be found at

374 https://github.com/codexf/Modeling_Photosynthesis.

375 **Transient response of A to changes in O_2 mole fraction.** Dynamic photosynthesis was recorded using a

376 LI-6800 with 2-s intervals. The leaves were illuminated (100 μmol $m^{-2} s^{-1}$) in 40 Pa CO_2 levels (CO_2

377 injector manually controlled) with either high O_2 (40% O_2 / 60% N_2) or low O_2 (2% O_2 / 98% N_2). When

12

378 the steady-state rate of A had been achieved, the cuvette was switched to low or high O_2 and leaves were

379 left to reach a new steady-state rate of A. The time course data were fitted using one-phase exponential

380 functions and the integrated area between the fitted curve and the baseline of the new steady-state A was

381 calculated to estimate the total C lag during the transients. Data and custom scripts can be found at

382 https://github.com/codexf/Modeling_Photosynthesis.

383

Data availability

385 All data used in this study are provided as Supporting Information of the article or available upon request.

386 Gas exchange data for modeling the photosynthetic responses to carbon dioxide concentration and

387 photosynthetic responses during oxygen transients can be found on GitHub at

388 https://github.com/codexf/Modeling_Photosynthesis.

389

Code availability

391 Custom scripts for modeling the photosynthetic responses to carbon dioxide concentration and

392 photosynthetic responses during oxygen transients can be found on GitHub at

393 https://github.com/codexf/Modeling_Photosynthesis. A cloud-based workflow to run the Monte Carlo

394 simulations for computing confidence intervals of metabolic fluxes can be found on GitHub at

395 https://github.com/codexf/MFA_Cloud_Computing.

396

Acknowledgements

417

Author contributions

419   BJW and XF designed the experiments. XF carried out the bulk of the experiments and analyzed the

420   results. SEW advised and assisted the $^{14}CO_2$ experiments to measure starch and sucrose partitioning. LG

421   performed the isotope analysis to measure gas exchange parameters. XF wrote the manuscript with

422   contributions from all the authors. BJW serves as the author responsible for contact and ensures

423   communication.

424

Competing interests

426   The authors declare no competing interests.

427

Figure legends

429   Fig. 1: Central carbon assimilatory metabolic fluxes in tobacco leaves exposed to varying O_2 conditions.

430   Fluxes were estimated by ^{13}C INST-MFA using the metabolic network model and experimental inputs

431   including mass isotopologue distributions of measured metabolites, the net CO_2 assimilation rate, rates of

432   rubisco carboxylation and oxygenation, rates of starch and sucrose synthesis, and output flux ratios of

433   amino acids and sucrose. Best-fit flux estimates are shown in numbers and corresponding variable width

434   of arrows. Arrows in blue indicate reactions constrained (not fixed) based on the experimental

435   measurements (Supplementary Table 1). Flux units are expressed as μmol molecule g^{-1} FW hr^{-1} (for the

436   conversion to μmol carbon g^{-1} FW hr^{-1}, multiply the presented values by the number of carbons in the

437   molecule). Metabolites compartmentalized to the plastid and cytosol are denoted by (".p") and (".c")

438   respectively. Abbreviations are shown in Supplementary Data 7.

439   Fig. 2: The response of A to the chloroplastic CO_2 concentration (Cc) in tobacco leaves exposed to 21%

440   O_2 under high light (1000 μmol m^{-2} s^{-1}). Measured A are presented as dots with error bars (mean ± SD, n

441   = 9 biologically independent leaves) fitted with the modified Farquhar, von Caemmerer and Berry

442   (FvCB) biochemical model of photosynthesis with two terms $α_G$ and $α_S$, where $α_G$ is the proportion of the

443   carbon exported from the photorespiratory pathway as glycine, and $α_S$ the proportion of the carbon

444   exported from photorespiration as serine (see Method section: *Estimating photorespiratory C export*

445   *using CO_2 response of A*). The parameter values obtained from the individual fits are shown as mean ±

446   SD (n = 9 biologically independent leaves), including the maximum rubisco carboxylation rate (V_{cmax}),

447   the maximum rate of electron transport (J), the rate of triose phosphate export from the chloroplast (T_p),

448   $α_G$ and $α_S$. The curves represent Rubisco (red), RuBP (green), and TPU (orange) limitation.

449   Fig. 3: Metabolic pool size alterations in response to photorespiratory conditions. Tobacco leaves were

450   exposed to the three O_2/N_2 conditions (all at 40 Pa $^{12}CO_2$ levels) in the LI-6800 cuvette for 30 min until

451   quenching with liquid nitrogen under illumination. Leaves acclimated to 2% O_2 and 40% O_2 were

452   compared to the ambient O_2 condition respectively and the steady-state pool sizes of the identified

453   metabolites (plotted on the y-axis) were determined using GC-MS and LC-MS/MS. The x-axis plots the

454   log2-transformed relative ratio of abundance of each metabolite in the leaves exposed to 2% O_2 (blue) or

455   40% O_2 (red) normalized to the same metabolite in the leaves exposed to 21% O_2. Results are presented as

456   dots with error bars (mean ± SEM, n = 7 or 8 biologically independent leaves). Asterisk represents a

457   significant difference between two conditions (Benjamini-Hochberg procedure was used to control the

458   false discovery rate at 0.05 for multiple t-tests including all metabolites). Supplementary Data 5 presents

459   the absolute concentration of each metabolite under different oxygen conditions with the results of one-

460   way ANOVA followed by Tukey's test for multiple pairwise comparisons among the three oxygen

461   conditions per metabolite. Abbreviations of metabolites are shown in Supplementary Data 7.

462   Fig. 4: Active and inactive pools of glycine at varying O_2 levels. Tobacco leaves were exposed to the

463   three O_2/N_2 conditions (all at 40 Pa $^{12}CO_2$ levels) in the LI-6800 cuvette for 30 min until quenching with

464   liquid nitrogen under illumination. Steady-state pool sizes of glycine were determined using GC-MS. The

465   proportion of active (gray) and inactive (white) pools of glycine were estimated using the dilution

466   parameters in INST-MFA modeling. Data are presented as box plots (center line at the median, upper

467   bound at 75th percentile, lower bound at 25th percentile) with whiskers extended to the extreme data

468   point ≤ 1.5 × interquartile range from the edge of the box (n = 7 or 8 biologically independent leaves).

469   Different letters above the data boxes denote statistically significant differences among the three

470   conditions for active glycine pools (uppercase letters) and inactive glycine pools (lowercase letters) at $P <$

471   0.05 (one-way ANOVA followed by Tukey's test for multiple pairwise comparisons). For active glycine

472   pools, the P values for Tukey multiple comparisons of means are 0.0006 (2% O2 - 21% O2), <0.0001

473   (2% O2 - 40% O2), and 0.0042 (21% O2 - 40% O2). For inactive glycine pools, the P values for Tukey

474   multiple comparisons of means are 0.0260 (2% O2 - 21% O2), 0.0138 (2% O2 - 40% O2), and 0.9511

475   (21% O2 - 40% O2).

476   Fig. 5: Dynamics of net CO_2 assimilation rate (A) during oxygen transients from 2% O_2 to 40% O_2 (a) and

477   from 40% O_2 to 2% O_2 (b). The leaves were acclimated in a LI-6800 cuvette at low O_2 or high O_2 for 30

478   min to reach the photosynthetic steady-state. Upon switching to the opposite oxygen condition, A changed

479   drastically due to slow gas mixing and returned to the previous steady state within ~10 s. The time when

480   the previous steady state was reached after switching oxygen was set to zero for the convenience of curve

481   fitting. Data points of different colors correspond to independent time-course measurements on ten

482   different plants. Lines of different colors represent the fitted curves of a one-phase exponential function

483   for individual replicates. Integrated area between the fitted curve and the baseline of new steady state for

484   a represented replicate is presented in the inserts. The integrated area obtained from the ten individual fits

485   are shown as mean ± SD (n = 10 biologically independent leaves).

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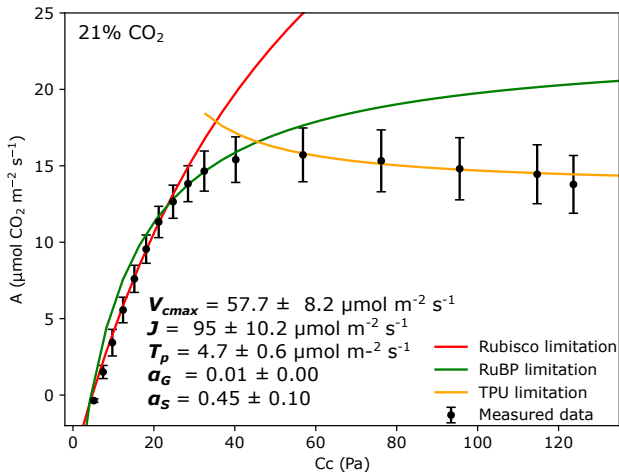554   *Plant J.* **45**, 490–511 (2006).

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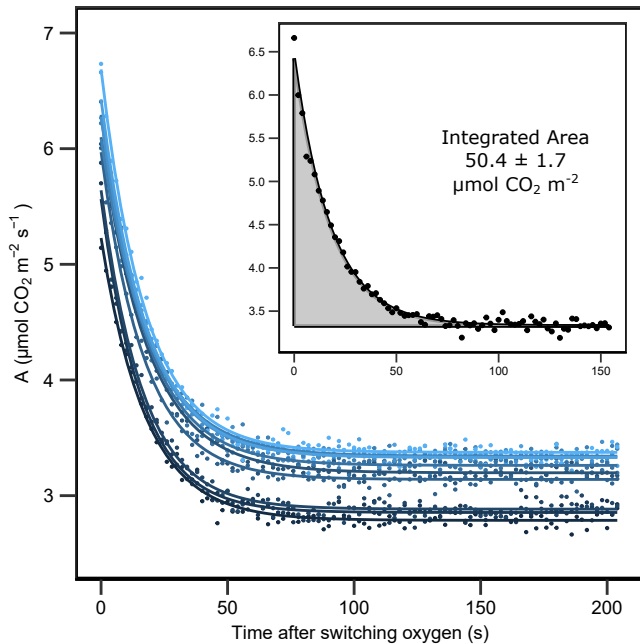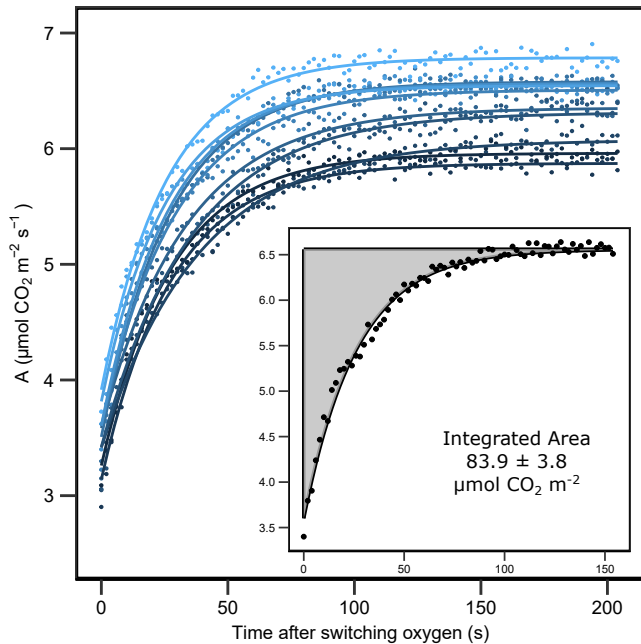
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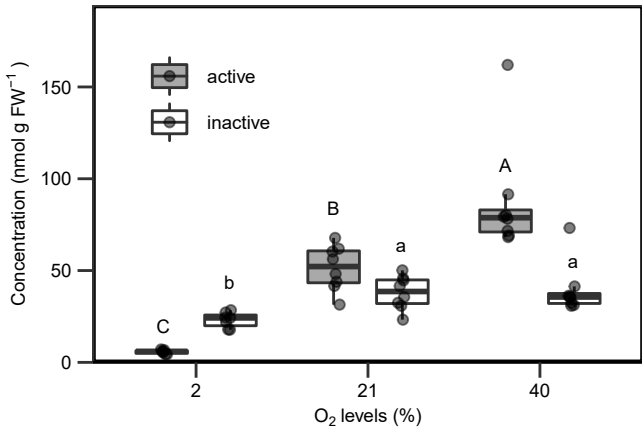
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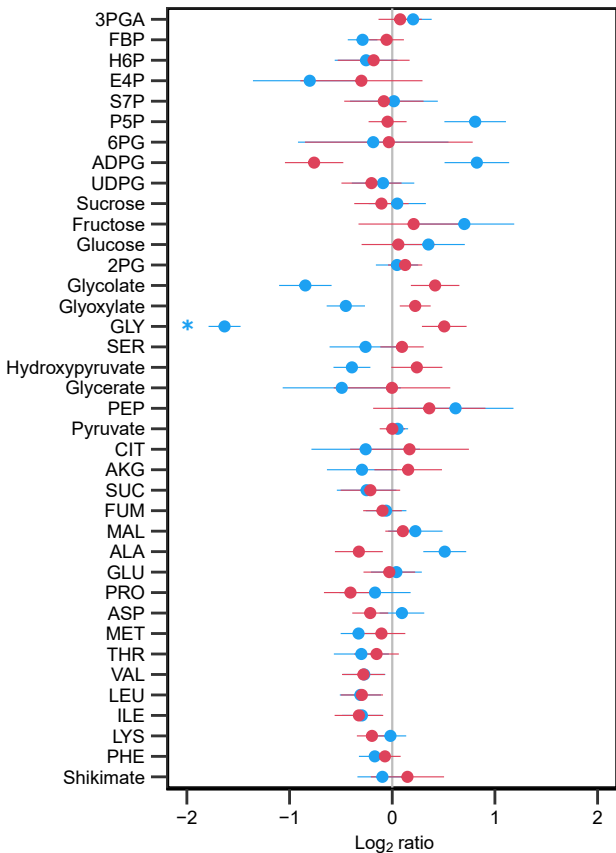
643

21% CO_2

V_{cmax} = 57.7 ± 8.2 µmol m^{-2} s^{-1}
J = 95 ± 10.2 µmol m^{-2} s^{-1}
T_p = 4.7 ± 0.6 µmol m^{-2} s^{-1}
a_G = 0.01 ± 0.00
a_S = 0.45 ± 0.10

— Rubisco limitation
— RuBP limitation
— TPU limitation
● Measured data

A (µmol CO_2 m^{-2} s^{-1})

Cc (Pa)

a Transients from 2% O_2 to 40% O_2

A (μmol CO_2 m^{-2} s^{-1})

Time after switching oxygen (s)

Integrated Area
50.4 ± 1.7
μmol CO_2 m^{-2}

b Transients from 40% O_2 to 2% O_2

A (μmol CO_2 m^{-2} s^{-1})

Time after switching oxygen (s)

Integrated Area
83.9 ± 3.8
μmol CO_2 m^{-2}

Legend: ● 2% O_2 / 21% O_2 ● 40% O_2 / 21% O_2

Y-axis (top to bottom): 3PGA, FBP, H6P, E4P, S7P, P5P, 6PG, ADPG, UDPG, Sucrose, Fructose, Glucose, 2PG, Glycolate, Glyoxylate, GLY, SER, Hydroxypyruvate, Glycerate, PEP, Pyruvate, CIT, AKG, SUC, FUM, MAL, ALA, GLU, PRO, ASP, MET, THR, VAL, LEU, ILE, LYS, PHE, Shikimate

X-axis: Log$_2$ ratio

Supplementary Figure 1. Transient $^{13}CO_2$ labeling in selected metabolites. Experimental measurements are shown as dots with error bars (mean ± SD, n = 3 biologically independent leaves). Solid lines represent the simulated labeling kinetics obtained with our best-fit set of fluxes from INST-MFA. The nominal mass isotopologues are represented as M0 and labeled isotopologues are distinguished by their mass differences from M0 (M1, M2, M3, etc.).
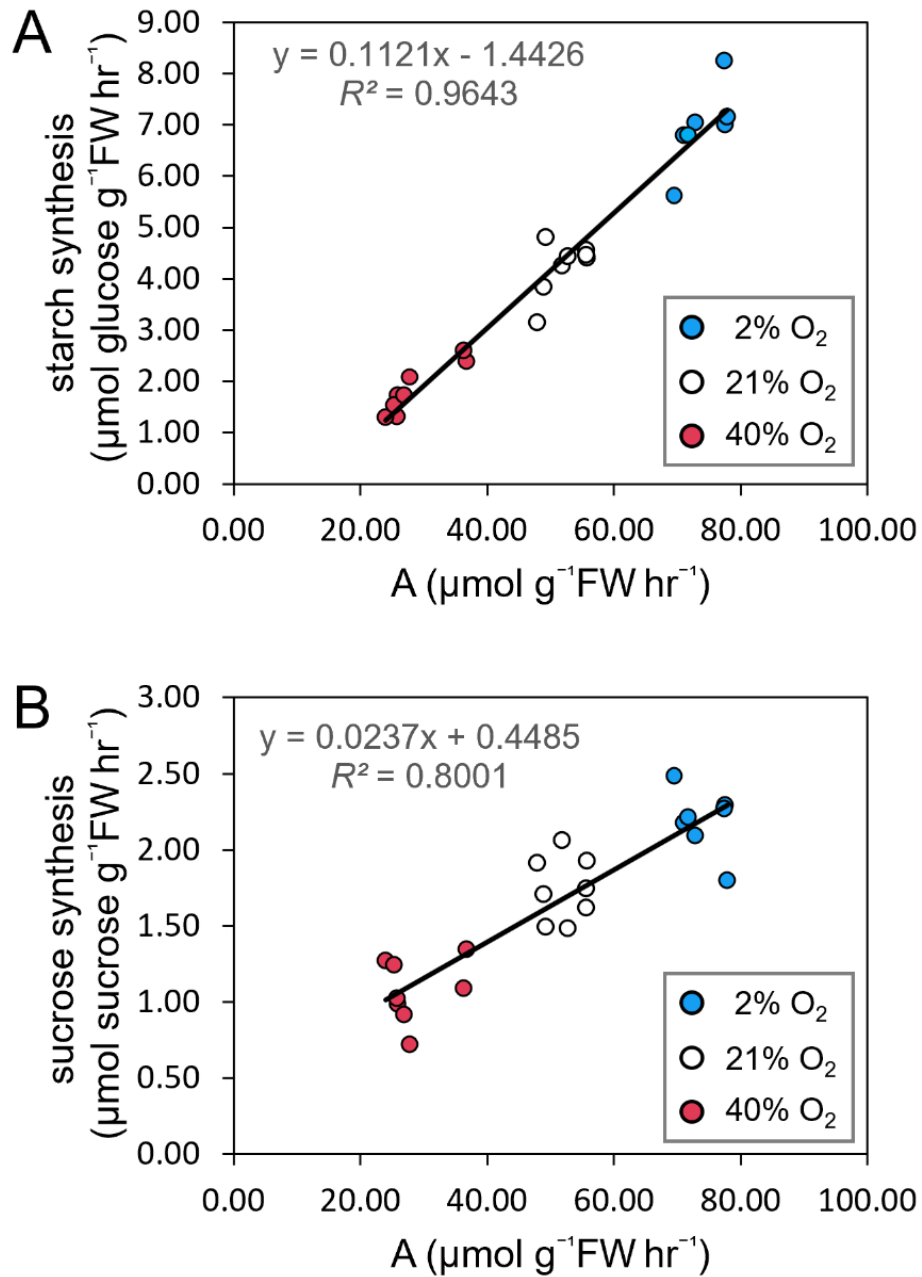
Supplementary Figure 2. Estimated fluxes and 95% confidence intervals by parameter continuation for reactions in the C3 cycle and photorespiration. Dots represent the best-fit values and horizontal lines represent the upper and lower bounds on the 95% confidence intervals calculated by the parameter continuation method. Results of all reactions are shown in a tabular form in Supplementary Data 4 and abbreviations of reactions are shown in Supplementary Data 7.
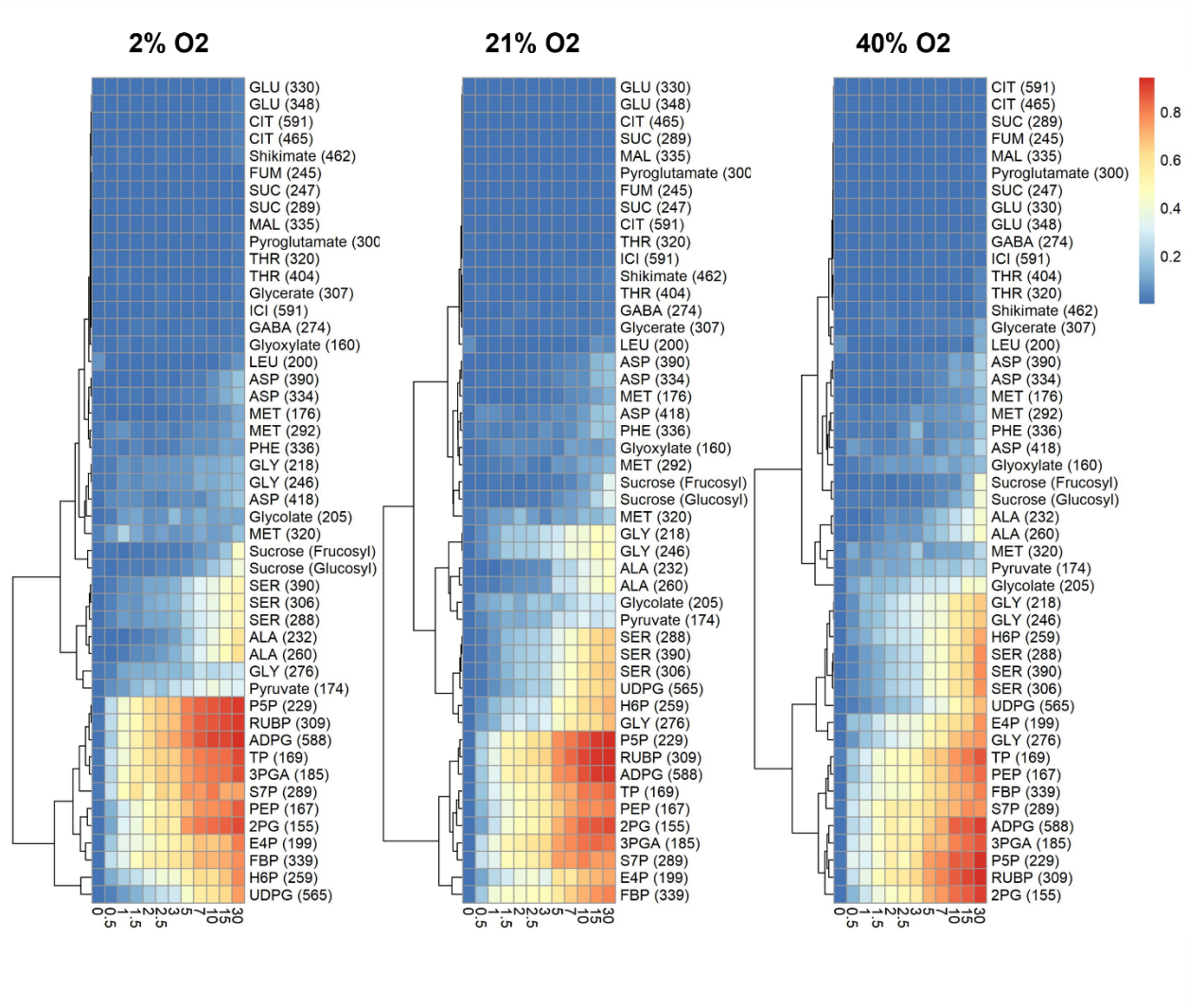
Supplementary Figure 3. Estimated flux values and 95% confidence intervals by parameter continuation for reactions in starch and sucrose synthesis and the TCA cycle. Dots represent the best-fit values and horizontal lines represent the upper and lower bounds on the 95% confidence intervals calculated by the parameter continuation method. Results of all reactions are shown in a tabular form in Supplementary Data 4 and abbreviations of reactions are shown in Supplementary Data 7.
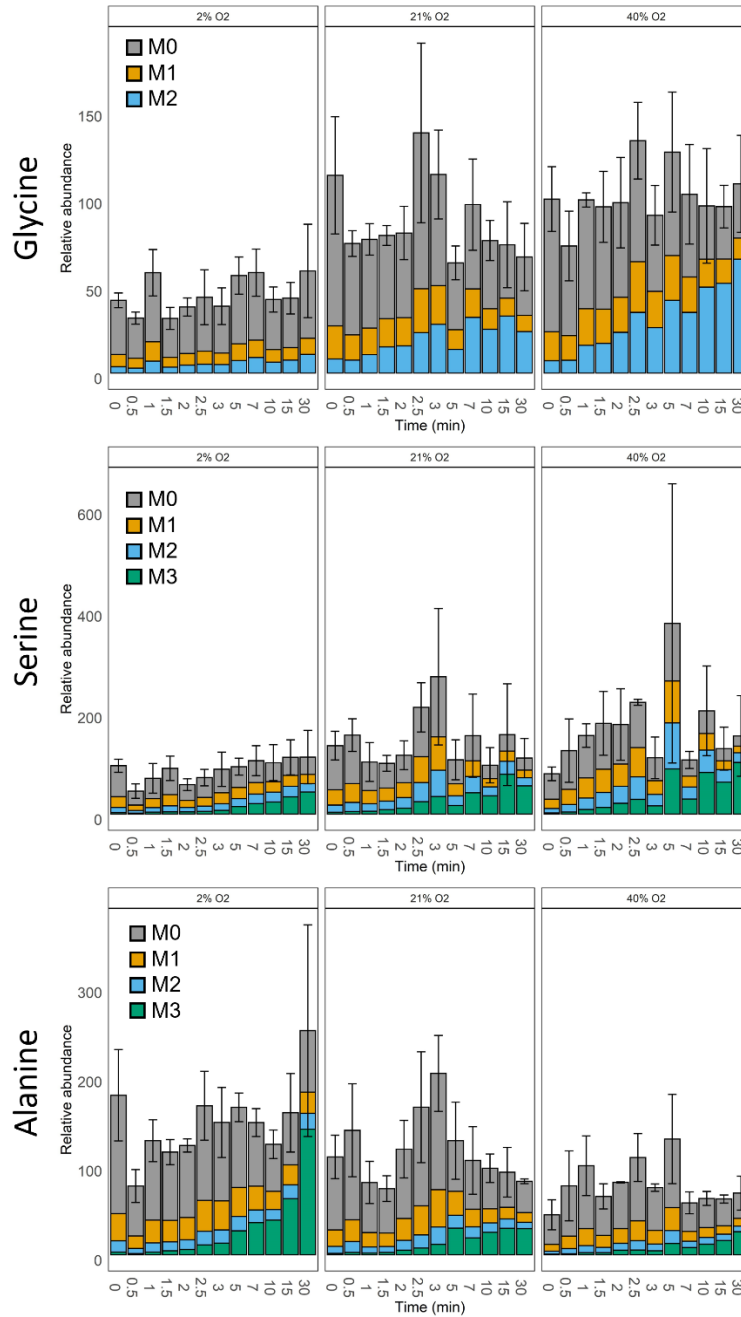
Supplementary Figure 4. The relationships between the net CO₂ assimilation rate (*A*) and synthesis rates of starch and sucrose under varied O₂ conditions. Rates of starch (A) and sucrose synthesis (B) were measured by the steady state pulse-chase labeling of $^{14}CO_2$ at each of the three O_2 conditions (see Supplementary method 2 for the estimation of starch and sucrose synthesis rates).
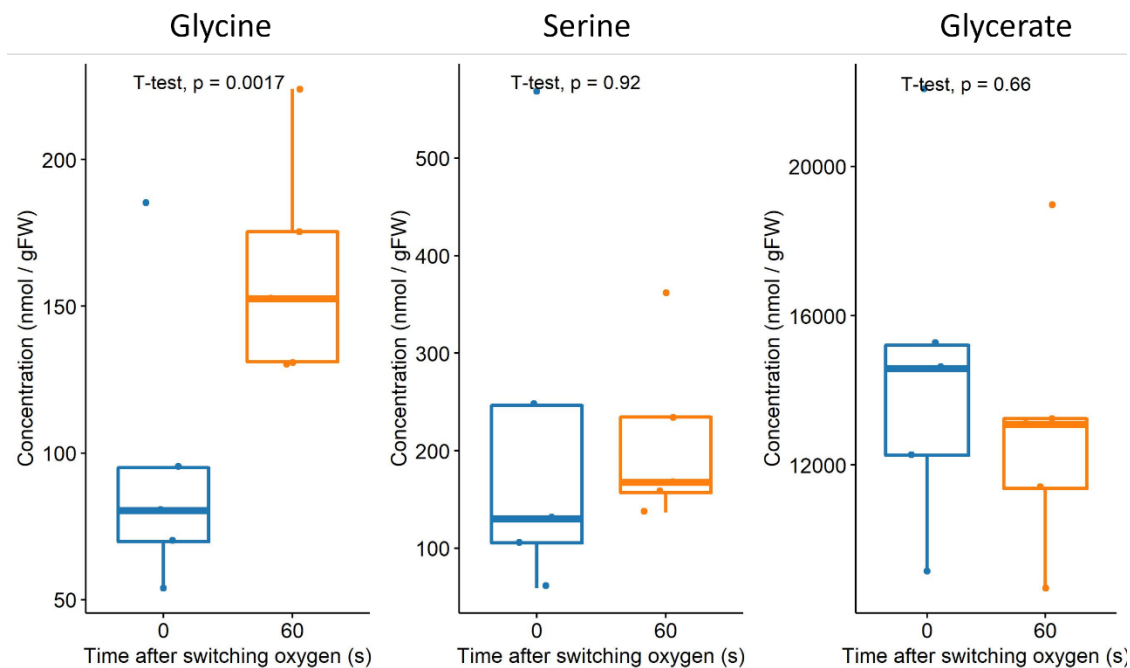
Supplementary Figure 5. Dendrogram and heatmap of the average ^{13}C-enrichment for identified metabolites over the 30 min labeling period under varying oxygen conditions. The scale of the heat map indicates blue as lowest and red as highest in the average ^{13}C-enrichment. The average ^{13}C enrichment of a metabolite was calculated by the formula of $(\frac{1}{N}) \sum_{i=1}^{N} Mi \times i$, where M$_i$ represents the fraction of the mass isotopologue distribution (MID) after natural isotope abundance correction and N represents the number of carbon atoms in the metabolite. Nominal masses of M0 mass isotopologues are shown in parentheses after metabolite names.

Supplementary Figure 6. Relative abundance of key metabolites (glycine, serine, and alanine) was relatively constant during the labeling period at each oxygen condition. Tobacco leaves were exposed to the three O_2 conditions in the LI-6800 cuvette for 20-30 min before incubating at 0, 0.5, 1, 2, 2.5, 3, 5, 7, 10, 15, and 30 min with $^{13}CO_2$-containing gas mixture at each oxygen condition. To assess differences in relative pool sizes of each metabolite between time-points, relative amounts of all isotopologues of a given metabolite were normalized to the internal standard and fresh weight. Data are presented as mean ± SEM (n = 3 biologically independent leaves). Total bar height represents the relative abundance of the total pool, and the shaded sections represent the fractions of isotopologues.

Supplementary Figure 7. Transient changes in photorespiratory intermediates (glycine, serine, and glycerate) after a switch from low to high photorespiratory condition. Leaves were acclimated under 2% oxygen for 30 min to reach the photosynthetic steady state. Samples were collected before (time = 0 s) and after (time = 60 s) switching to 40% oxygen. Data are presented as box plots (center line at the median, upper bound at 75th percentile, lower bound at 25th percentile) with whiskers extended to the extreme data point ≤ 1.5 × interquartile range from the edge of the box. Statistical significance between treatments (0 s and 60 s after switching oxygen) in each metabolite was assessed using two-tailed *t*-tests ($P < 0.05$, n = 5 biologically independent leaves). While serine and glycerate did not change during the transition, glycine increased significantly. The build-up of glycine pool after switching oxygen for 60 s was 20.6 ± 6.2 (mean ± SD) $\mu mol\ m^{-2}$ on a leaf area basis or 65.6 ± 19.6 (mean ± SD) $nmol\ g^{-1}$ FW on a fresh-weight basis.

Supplementary Table 1. External flux measurements used in INST-MFA as constraints to determine intracellular fluxes. The net CO_2 assimilation rate was measured in the ^{13}C-labeled leaves using a LI-6800. Rates of rubisco carboxylation and oxygenation were calculated based on the biochemical model of photosynthesis using gas exchange data. Rates of starch and sucrose synthesis were measured by the steady state pulse-chase labeling of [14]CO_2 at each of the three O_2 conditions (see Supplementary method 2 for the estimation of starch and sucrose synthesis rates). Flux values are presented as mean ± SEM (n = 7 biologically independent leaves) in units of µmol molecule g^{-1} FW hr^{-1} (for the conversion to µmol carbon g^{-1} FW hr^{-1}, multiply the presented values by the number of carbons in the molecule).

External rates	2% O_2	21% O_2	40% O_2
net CO_2 assimilation rate (µmol CO_2 g^{-1} FW hr^{-1})	74.34 ± 0.77	48.82 ± 0.76	34.11 ± 0.78
rubisco oxygenation rate[1] (µmol O_2 g^{-1} FW hr^{-1})	2.59 ± 0.06	18.8 ± 0.44	27.63 ± 0.44
rubisco carboxylation rate[1] (µmol CO_2 g^{-1} FW hr^{-1})	83.33 ± 0.77	65.92 ± 0.83	55.62 ± 0.86
starch synthesis rate (µmol glucose g^{-1} FW hr^{-1})	6.95 ± 0.29	4.22 ± 0.18	1.85 ± 0.17
sucrose synthesis rate (µmol sucrose g^{-1} FW hr^{-1})	2.19 ± 0.08	1.72 ± 0.07	1.09 ± 0.07

[1] The calculation of rubisco carboxylation and oxygenation rates requires the determination of three gas exchange parameters. The mesophyll conductance (g_m), respiration in the light (R_L) and the CO_2 compensation point in the absence of mitochondrial respiration (Γ^*) were determined to be 2.00 µmol m^{-2} s^{-1} Pa^{-1}, 0.67 µmol m^{-2} s^{-1}, and 4.04 Pa (see Supplementary Method 4 for details of calculations).

Supplementary Table 2. Sucrose and amino acid levels in vascular exudates. Vascular exudates were collected using a pressure bomb chamber to force release of liquid from the petiole of a fresh cut tobacco leaf previously exposed to ambient oxygen (21%) and analyzed by GC-MS (see Supplementary Method 1). Values of the levels are mean ± SEM (n = 9 biologically independent leaves). The ratios of several amino acids to sucrose (Glutamate (GLU) / Sucrose = 0.159, Aspartate (ASP) / Sucrose = 0.270, Alanine (ALA) / Sucrose = 0.128, Glycine (GLY) / Sucrose = 0.042, Serine (SER) / Sucrose = 0.168) were used to constrain the output flux reaction (S6P + 0.159*GLU + 0.27*ASP + 0.129*ALA.c + 0.042*dummyGLY + 0.169*dummySER -> Sink) in all INST-MFA models. The amino acid to sucrose ratios were comparable to those measured in the aphid stylet-collected phloem sap[1].

Metabolite	Level (nmol/µL)
Sucrose	0.403 ± 0.063
ASP	0.1 ± 0.021
SER	0.069 ± 0.017
GLU	0.064 ± 0.013
ALA	0.048 ± 0.008
GLY	0.014 ± 0.001

[1] Winter, H., Lohaus, G. & Heldt, H. W. Phloem transport of amino acids in relation to their cytosolic levels in barley leaves. Plant Physiol. 99, 996–1004 (1992).

Supplementary Method 1. The extraction and analytical procedures for quantifying metabolite labeling and concentration.

1.1 Harvesting and quenching

For the $^{13}CO_2$ labeling experiment at each O_2/N_2 condition (2% O_2 / 98% N_2, 21% O_2/ 79% N_2, or 40% O_2/ 60 % N_2), labeled tobacco leaf samples were collected at time points of 0, 0.5, 1, 2, 2.5, 3, 5, 7, 10, 15, 30 min inside the 9 cm^2 cuvette of a LI-COR 6800 (LI-COR Biosciences, USA) with at least three biological replicates. To quench metabolism rapidly under illumination at a given time point, liquid nitrogen was directly sprayed on the leaf surface by inserting a liquid nitrogen cryospray nozzle (Wallach UltraFreeze, Cooper Surgical, USA) via an injection port on the LI-COR cuvette. The frozen leaf potions were rapidly removed from the cuvette, then cut and placed in a tube with liquid nitrogen and stored at -80°C until analysis before they were thawed. This sampling procedure has been extensively described earlier [1] with pictures of the setup and measurements of the $^{13}CO_2$ switching and quenching speed.

To determine the steady-state pool sizes of central metabolites in leaves exposed to the three O_2/N_2 conditions (all at 40 Pa $^{12}CO_2$ levels), leaves were put in the LI-COR cuvette for 30 min to acclimate until they reached a photosynthetic steady-state. At least seven biological replicates for each condition were harvested and quenched as described earlier.

To measure the vascular sucrose and amino acids, exudates were collected using a pressure bomb chamber on plants harvested under ambient conditions. A fresh cut tobacco leaf was placed inside the chamber, with the cut end of the petiole protruding from the rubber seal of the chamber. Pressure was gradually applied from a gas cylinder until droplets of liquid appeared at the cut surface. Exudates were collected from nine randomly chosen mature leaves in different plants. The exudate collection was carried out in the same condition for each replicate within 3 min to avoid dehydration and lack of light. Samples were subsequently frozen in liquid nitrogen and stored at −80°C for further analysis.

To determine the transient changes in metabolite pool sizes after a switch from low to high photorespiratory condition, leaves were first put in the LI-COR cuvette under low oxygen (40 Pa $^{12}CO_2$, 2% O_2, and 98% N_2) for 30 min. After the leaves reached the photosynthetic steady-state, five low-oxygen adapted leaves were quenched and collected immediately. The other five leaves were quenched and collected after the gas in the cuvette was switched to high oxygen (40 Pa $^{12}CO_2$, 40% O_2, and 98% N_2) for 1 min. In all experiments, leaves were randomly assigned to one of the treatments or time points.

1.2 Metabolite extraction of polar metabolites

Frozen leaf tissues were pulverized under cryogenic conditions using a FastPrep-24 tissue homogenizer with a CoolPrep Adapter (MP biomedicals, USA). Metabolites were extracted using a methanol/chloroform/water protocol as described earlier [2]. The extracts were separated into upper (methanol/water) phase and bottom (chloroform) phase after centrifugation. Several aliquots of polar extracts were taken from the upper layer, lyophilized to dryness, and frozen at -80°C before the GC-MS and LC-MS/MS analyses. For pool size quantification of unlabeled samples, known concentrations of internal standards (nor-Leucine, 98 atom % ^{15}N, 98 atom % ^{13}C free amino acid mixture, D-[UL-$^{13}C_6$] fructose 1, 6-bisphosphate, and [UL-$^{13}C_{12}$] sucrose) were spiked onto the tissue before the extraction.

1.3 Gas chromatography- mass Spectrometry (GC-MS) analyses

Organic acids, amino acids, and sugars recovered in the polar fractions were first derivatized by adding methoxyamine hydrochloride dissolved in dry pyridine. The reaction mixture was shaken at 90 rpm at 40°C for 12 h. An aliquot of the reaction mixture was silylated by adding N, O-Bis (trimethylsilyl)

trifluoroacetamide with 1% (w/v) trimethylchlorosilane and incubated at 60°C for 60 min, resulting in trimethylsilyl (TMS) derivatives. Another aliquot of the reaction mixture was silylated by adding N-(tert-butyldimethylsilyl)-N-methyltrifluoroacetamide with 1% (w/v) tert-butyldimethylchlorosilane and incubated at 60°C for 60 min, resulting in trimethylsilyl (TBDMS) derivatives. The TMS- and TBDMS-derivatives were analyzed by an Agilent 7890A GC (Agilent, USA) equipped with a 30 m VF-5ms column and interfaced to an Agilent 5975C inert XL Mass Selective Detector (Agilent, Santa Clara, CA, USA). Aliquots of the derivatized sample (1 μL) were injected in splitless mode with a helium carrier gas flow set at 1 mL/min. For some highly abundant metabolites (e.g., malate, fumarate, glycerate, and sucrose, etc.), a split mode with a 20:1 split ratio was applied. The inlet temperature and MS transfer line temperature were set at 230°C and 300°C, respectively. The oven temperature was initially held at 40°C for 1 min, then raised at 40°C/min to 80°C, ramped at 10°C/min to 240°C, then 20°C/min to 320°C, and finally kept at this temperature for 5 min. The mass spectrometer was operated on scan mode (electron impact at 70 eV, ion source temperature at 230°C, the quadrupole temperature at 150°C), monitoring ions at m/z values of 50–600.

GC-MS data files were deconvoluted and searched against the NIST 14 Mass Spectral Library and the Golm Metabolome Database using NIST software AMDIS V2.72 for initial identification. Chromatographic peaks were further identified based on the retention time of authentic standard peaks and the corresponding MS spectrum. The characteristic fragment ions that were used for the isotopologue analysis are shown in Supplementary Data 6. The mass isotopologue distribution (MID) for each metabolite was correct for natural abundance of isotopes and mean ^{13}C enrichment was calculated as described earlier [3]. The EI spectrum of the TMS derivative of sucrose cannot be directly used for the isotopologue analysis, as each of the two major fragment ions m/z 361 and 451 represents a mixture of glucosyl and fructosyl moieties of sucrose. The fragmentation patterns of [UL-$^{13}C_6$fru] sucrose was used to deduce the proportion of the glucosyl and fructosyl moieties in each fragment ion. Fragment m/z 361 corresponds to a mixture of 54% fructosyl and 46% glucosyl unit. Fragment m/z 451 is a mixture of 96% fructosyl and 4% glucosyl. The actual MIDs of glucosyl and fructosyl moieties of sucrose were solved using the m/z 361-367 and 451-457 as described earlier[3].

1.4 Ion-pair chromatography – tandem mass spectrometry (IPC-MS/MS) analyses

Phosphorylated intermediates in C3 cycle were analyzed by an IPC-MS/MS method as described earlier[1]. Aliquots of freshly reconstituted lyophilized extract (10 μL) were injected into a Waters ACQUITY UPLC pump system (Waters, USA) coupled to a Waters Quattro Premier tandem quadrupole mass spectrometer in negative-mode electrospray ionization (ESI). Chromatographic separation was performed using a Waters UPLC BEH C18 column (2.1× 50 mm) at 40°C using a multi-step gradient with eluent A (10 mM tributylamine in 5%(v/v) methanol, adjusted to pH 6 with 0.5 M acetic acid) and eluent B (methanol): 0-1 min, 95-85% A; 1-6 min, 65-40% A; 6-7 min, 40-0% A; 7-8 min, 0% A; 8-9 min, 100% A. The flow rate was 0.3 mL min^{-1}, source temperature 120°C, and desolvation temperature 350°C. The parent-product ion pairs for each metabolite and the specific metabolite-dependent MS parameters are shown in Supplementary Data 6.

1.5 Anion exchange chromatography – tandem mass spectrometry (AEC-MS/MS) analyses

Nucleotide sugars (ADPG and UDPG) and additional phosphorylated intermediates (i.e., 2PG, PEP) were analyzed by AEC-MS/MS as described earlier [1]. Aliquots of freshly reconstituted lyophilized extract (10 μL) were analyzed by a Waters ACQUITY UPLC pump system coupled to a Xevo ACQUITY TQ Triple Quadrupole Detector (Waters, USA) in negative-ESI mode. Chromatographic separation was performed using an IonPac AS11 analytical column (Dionex, 2 × 250 mm) coupled to an IonPac guard column AG11

(2 × 50 mm, Dionex, Thermo Scientific, USA) and an IonPac ATC-3 Anion Trap Column (4 × 35 mm, Dionex, Thermo Scientific, USA) using a multi-step gradient with eluent A (0.5 mM KOH) and eluent B (75 mM KOH): 0-2 min, 100% A; 2-4 min, 100-93% A; 4-13 min, 93-60% A; 13-15 min, 0% A; 15-17 min, 100% A, at a flow rate of 0.35 mL min^{-1}. Salt suppression was carried out using a post-column anion self-regenerating suppressor (Dionex ADRS 600, Thermo Scientific, USA), operated in the AutoSuppression External Water mode with a flow rate of 3.5 mL min^{-1} and current of 50 mA. The parent-product ion pairs for each metabolite and the specific metabolite-dependent MS parameters are shown in Supplementary Data 6.

1.6 Data analysis for metabolite labeling and profiling data

Chromatographic peaks for LC-MS/MS and GC-MS were integrated using the Waters software MassLynx version 4.2 and DExSI version 1.11 (https://github.com/DExSI/DExSI/). For natural isotope abundance correction, the measured mass isotopologue distributions determined from the GC-MS and LC-MS/MS data were subsequently processed using Fluxfix version 0.1 (https://fluxfix.science/). Statistical analysis of metabolite labeling and profiling data was performed in R version 4.0.2.

Supplementary Method 2. Steady state pulse-chase labeling of $^{14}CO_2$ to estimate rates of starch and sucrose accumulation at varying O_2 conditions

Fully expanded tobacco leaves were pulsed with $^{14}CO_2$ at each of the three O_2/N_2 conditions (2% O_2/ 98%N_2, 21% O_2/ 79% N_2, and 40% O_2/ 60 % N_2) through the sample port on the back of a head of a LI-6800 (LI-COR Biosciences, USA) for 10 min. The assimilated ^{14}C was then chased with $^{12}CO_2$ for 5 min, under a light intensity of 100 μmol photon m^{-2} s^{-1} and temperature of 25°C. The leaves were exposed to CO_2 concentrations of 400 μL L^{-1} (v/v) during the entire pulse-chase labeling process. At the end of the chase period, the labeled leaf tissue was immediately frozen in liquid nitrogen and stored at −80°C. Frozen tissue was extracted with 0.5 mL formic acid solution (formic acid/ethanol 4:75, v/v). The suspension was centrifuged at 12000 g, 4°C for 10 min. Half of the supernatant was directly counted for radioactivity (total soluble fraction) using a 1450 Microbeta Trilux scintillation counter (Perkin Elmer, USA). The other half the supernatant was passed through a Dowex 1 (Cl$^-$ form) column, followed by a Dowex 50 (H$^+$ form) column. The first 1 mL flow through was discarded and the next five 1 mL fractions were separately collected and counted for radioactivity (neutral soluble fraction). The ionic fraction was calculated as the difference in radioactivity between total soluble fraction and the neutral soluble fraction. The pellet was washed once with water and twice with 80 % ethanol and then resuspended in 0.5 mL of 0.2 M KOH. Starch in the pellet was gelatinized at 95 °C for 30 min, then digested with 50 U alpha -amylase and 6.5 U amyloglucosidase in an acetate buffer (pH 5) for 2 days. After centrifugation, the supernatant was removed and counted for radioactivity. Proportion of counts in the starch and neutral soluble fraction (presumably mostly sucrose) relative to total counts were multiplied by the photosynthetic rate of each leaf to estimate rates of starch and sucrose accumulation. Due to the relatively large sucrose pool and slow turnover rate, we assume that the loss of sucrose exported from the labeled leaf region during the short labeling period is negligible.

Supplementary Method 3. Determining diffusional and biochemical parameters for estimating rates of rubisco oxygenation (v_o) and rubisco carboxylation (v_c).

3.1 Describing the Gas-Exchange and Carbon Isotope Discrimination System

To estimate g_m, which is required to calculate the intercellular CO_2 concentrations needed to estimate v_o and v_c, *in vivo* gas exchange measurements of steady-state photosynthesis were combined with on-line measurements of carbon isotope discrimination, Δ , as described earlier[4-6] using a LI-6800 (LI-COR

Biosciences, USA) coupled to a tunable infrared laser differential absorption spectrometer (TILDAS-CS, Aerodyne Research, USA).

A computer-controlled front-end was provided for the system, as an integrated hardware and software system "Vinland" created by H. Stuart-Williams at the Australian National University (Contact information available upon request). In brief, software running on a PC/Arduino pair reads and displays output from the laser and controls hardware devices including valves (VICI Valco Instruments Co. Inc., USA) and mass flow controllers (Alicat Scientific, Inc., USA) to manage a gas mixing system to make synthetic air and dope it with isotopically characterized reference CO_2 as required, using pneumatic injection valves with capillaries (SGE MOVPT, SGE International, Ringwood, Australia).

The fully expanded leaf was placed across the leaf chamber avoiding the midrib. The CO_2 in the leaf chamber, flow rate, and irradiance were set to 400 µmol mol^{-1}, 300 µmol s^{-1}, 1750 µmol m^{-2} s^{-1}, respectively. The LI-6800 was located inside a heating cabinet (Percival Scientific, USA) to maintain a 25°C leaf temperature and a 1.2 kPa vapor pressure deficit throughout the measurement. One synthetic air line was bubbled through a flask of water and fed into the LI-6800 air inlet port. A separate isotope reference line was doped with a specified concentration of CO_2 ($\delta^{13}C$ vs. VPDB: -4.6 ‰, (Airgas Specialty Gases, USA)) to produce a calibration reference line to correct the raw $\delta^{13}C$ signals. To determine Δ, the reference and sample exhaust lines of the LI-6800 were subsampled into the TILDAS-CS on an alternating basis. To accomplish this, two lines connected the LI-6800 exhaust ports A and B carried sample and reference airstreams through Nafion tubing and a dry-ice ethanol trap to remove all water vapor from the airstream. These lines and the calibration reference line from the mixing system, were connected to separate ports on the VICI multiposition valve to switch airstreams entering the TILDAS-CS. The TILDAS-CS measured molar ratios of the different carbon dioxide isotopologues ($^{16}O^{12}C^{16}O$, $^{16}O^{12}C^{17}O$, $^{16}O^{12}C^{18}O$ & $^{16}O^{13}C^{16}O$). Scripts running in the Vinland software controlled the valves and mixing and calculated/averaged raw $\delta^{13}C$ values for each measurement period.

$\delta^{13}C$ was calculated as:

$$\delta^{13}C = \left(\frac{R_{sample}}{R_{standard}} - 1 \right) * 1000 \tag{1}$$

where R_{sample} is the molar ratio of the fractional abundance of the heavy isotope over the fractional abundance of the light isotope from the airstream that is measured by the TILDAS-CS, $\frac{^{16}O^{13}C^{16}O}{^{16}O^{12}C^{16}O + {}^{16}O^{12}C^{17}O + {}^{16}O^{12}C^{18}O}$. While the $R_{standard}$ is the molar ratio of Vienna Pee Dee Belemnite, the international reference standard for $^{13}C/^{12}C$ and is 0.0111797 [4].

The TILDAS-CS was calibrated using an offset calibration approach as described earlier [4] to correct the raw $\delta^{13}C$. With the corrected $\delta^{13}C$, the observed ^{13}C photosynthetic discrimination, Δ_{obs}, was calculated from an equation presented in [4,7]:

$$\Delta_{obs} = \frac{\zeta \left(\delta^{13}C_{sample} - \delta^{13}C_{reference} \right)}{1 + \delta^{13}C_{sample} - \zeta \left(\delta^{13}C_{sample} - \delta^{13}C_{reference} \right)} \tag{2}$$

$$\zeta = \frac{^{12}CO_{2;\,reference}}{^{12}CO_{2;\,reference} - {^{12}CO_{2;\,sample}}} \tag{3}$$

where $\delta^{13}C_{sample}$ and $\delta^{13}C_{reference}$ are the carbon isotope compositions of the sample and reference airstreams coming from the LI-6800. ζ is the ratio of the $^{12}CO_2$ mole fraction in the dry air coming into the gas-exchange cuvette over the difference in $^{12}CO_2$ mole fractions of air in and out of the cuvette [4].

3.2 Measurement Cycle

To find accurate Δ_{obs} values, the system needed to cycle the calibration reference line and the LI-6800 sample and reference lines to determine the $\delta^{13}C$ of the gases. A Vinland script automated the calibration and sampling analysis. The calibration reference line was measured at two different concentrations (39 and 46.5 Pa) at the beginning and end of the measurement cycle. In-between the calibration measurements were the LI-6800 reference and sample measurements. These lines were subsampled five times to produce five technical replicates within one measurement cycle. An autolog program was written on the LI-6800 GUI to match the timing of the Vinland script so that gas-exchange measurements could be taken simultaneously with the carbon isotope discrimination measurements.

Measurements were corrected using an offset calibration. Offsets were calculated from the true (-4.6 ‰) and measured calibration reference $\delta^{13}C$ values at each CO_2 concentration. With these offsets, a linear regression was modeled between the CO_2 concentrations and offsets to create an offset calibration line. This offset calibration line was applied to correct the raw $\delta^{13}C$ measurements [4].

3.3 Final Estimation of g_m

Estimation of g_m was performed using the online isotope data according to [4,5]

$$g_m = \frac{1+t}{1-t}\frac{b'_3 - a_m - \frac{\alpha_b}{\alpha_e}e\frac{R_L}{A-R_L}}{\frac{\bar{a}}{1-t} + \frac{1}{1-t}[(1+t)*b'_3 - \bar{a}]\frac{C_i}{C_a} - \frac{1+t}{1-t}\frac{\alpha_b}{\alpha_e}e\frac{R_L}{A-R_L}\frac{C_i - \Gamma^*}{C_a} - \frac{1+t}{1-t}\frac{\alpha_b}{\alpha_f}f\frac{\Gamma^*}{C_a} - \Delta_{obs}}\frac{A}{C_a} \tag{4}$$

where, $t = \frac{\alpha_{ac}E}{2g_{ac}}$, and $\bar{a} = \frac{a_b(C_a - C_s) + a_s(C_s - C_i)}{C_a - C_i}$. e was calculated with the assumption that the plant used old photosynthate as the substrate for respiration, $e = \delta^{13}C_{tank} - \delta^{13}C_{atmosphere}$, where $\delta^{13}C_{tank}$ was the measured $\delta^{13}C$ of the sample airstream, and the $\delta^{13}C_{atmosphere}$ was assumed to be -8‰. R_L and Γ^* values were used from the common intersection method (described below). Other variables and fractionation factors used in the equation 4 are defined in the table below as described earlier [4].

Variable (units)	Definition
A (µmol m^{-2} s^{-1})	Net CO_2 assimilation rate
a_b (‰)	$^{12}C/^{13}C$ fractionation for CO_2 diffusion in the boundary layer ($a_b = 2.9$‰)
a_m (‰)	Summed $^{12}C/^{13}C$ fractionation during the liquid-phase dissolution and dissolution of CO_2 diffusion in the boundary layer ($a_m = 1.8$‰)
a_s (‰)	$^{12}C/^{13}C$ fractionation for CO_2 diffusion in air ($a_s = 4.4$‰)
$\bar{a}$ (‰)	The weighted fractionation for CO_2 diffusion across boundary layer and stomata in series
b'_3 (‰)	Overall *in vivo* $^{12}C/^{13}C$ fractionation during carboxylation by rubisco and PEPC (b'_3 = 30‰)

C_a (Pa)	The CO_2 partial pressure in the ambient air
C_i (Pa)	The CO_2 partial pressure in the intercellular airspace of the leaf
C_i^* (Pa)	The CO_2 partial pressure in the intercellular airspace of the leaf at the photorespiratory CO_2 compensation point
C_s (Pa)	The CO_2 partial pressure at the leaf surface
E (mol m^{-2} s^{-1})	Transpiration rate
e (‰)	$^{12}C/^{13}C$ fractionation during decarboxylation
f (‰)	$^{12}C/^{13}C$ fractionation during photorespiration (f = 11.8‰)
g_{ac} (mol m^{-2} s^{-1})	Conductance to diffusion of CO_2 in air
R_L (µmol m^{-2} s^{-1})	Non-photorespiratory CO_2 release in the light
t	Ternary correction coefficient
α_{ac}	$\alpha_{ac} = 1 + \bar{a}$
α_b	$\alpha_b = 1 + b'_3$
α_e	$\alpha_e = 1 + e$
α_f	$\alpha_f = 1 + f$
Γ^* (Pa)	The CO_2 partial pressure in the chloroplast at the photorespiratory CO_2 compensation point

3.4 Measurement of C_i^* and R_L

To measure C_i^* and R_L, we used the common intersection method [8,9]. As described above, measurements were performed using a LI-6800. The LI-6800 maintain a 25°C leaf temperature, a 1.5 kPa vapor pressure deficit, and a flow rate of 500 µmol s^{-1} during the measurement. The common intersection method measurements were linear fits of CO_2 response curves measured at 3, 5, 7, 9, 11, 40 (Pa) at irradiances of 250, 165, 120, 80, 50 (µmol m^{-2} s^{-1}). C_i^* and R_L values were determined from the linear regression of the slopes and y-intercepts from each irradiance as described earlier [9,10].

3.5 Calculating Γ^*

Γ^* was determined from $\Gamma^* = C_i^* + \frac{R_L}{g_m}$, where g_m, C_i^*, and R_L were determined from the gas exchange, and carbon isotope discrimination approaches described earlier [11]. g_m and Γ^* were resolved iteratively using previous g_m, Γ^*, and R_L values to improve solutions accuracy; iterations were stopped when unity was reached. Gas exchange and stable carbon isotope data were analyzed using R; scripts will be available upon request.

3.6 Calculating v_o and v_c
The key equation for estimating v_o and v_c using measured rates of A is

$$A = v_c - 0.5v_o - R_L \tag{5}$$

Based on the following relationships from the previous work [12]

$$\Phi = \frac{v_o}{v_c} \tag{6}$$

$$\Phi = \frac{2\Gamma^*}{C_c} \tag{7}$$

where Γ^* was determined from the section 3.5 above, C_c is the partial pressure of CO_2 at the site of rubisco catalysis and was determined by

$$C_c = C_i - \frac{A}{g_m} \qquad (8)$$

where A and C_i were determined from gas exchange measurements and g_m was determined from the section 3.3 above.

Rearranging equations (5) – (7), we can estimate v_o and v_c as

$$v_c = \frac{A + R_L}{1 - \Gamma^* / C_c} \qquad (9)$$

$$v_o = \frac{v_c - A - R_L}{0.5} \qquad (10)$$

Supplementary Method 4. Isotopically non-stationary metabolic flux analysis (INST-MFA) model formulation

Pseudo-steady state metabolism was assumed for INST-MFA because no significant changes in the relative pool sizes of most metabolites was observed throughout the 30 min labeling period at each oxygen condition (Supplementary Figure 7). A network model of central metabolism was constructed using the INCA package (version 1.8) to represent the interactions between the C_3 cycle, photorespiration, starch and sucrose synthesis, the TCA cycle, and amino acid metabolism. The stoichiometry of reactions and atom transitions for each reaction is listed in Supplementary Data 1. The INCA model files (in MATLAB readable format) are included in Supplementary Data 8. Metabolic fluxes were estimated by iteratively minimizing the sum of squared residuals (SSR) between the measured and simulated isotopic labeling patterns and external rate measurements. At each iteration, the network model was used to simulate isotopic labeling kinetics for a given set of fluxes. The fluxes were iteratively adjusted until the local optimum was achieved. The optimization was restarted from 100 random initial values of fluxes to find the global optimum (best-fit solution) (Supplementary Data 4).

The INST-MFA modeling reconciles as well as possible all experimental measurements for each oxygen condition, including the time-course mass isotopologue distribution data, rubisco oxygenation rate (v_o), rubisco carboxylation rate (v_c), starch and sucrose production rates, and output flux ratios of amino acids and sucrose measured in vascular exudate. Due to the technical difficulties of collecting exudates at 2% or 40% oxygen, we used exudates collected at 21% oxygen to constrain the output flux ratios of amino acids and sucrose in all models (Supplementary Table 2). Because the leaf export of amino acids and sucrose were stoichiometrically constrained to each other, the leaf export of glycine or serine was calculated by multiplying the output flux by the ratio of glycine or serine to sucrose.

The labeling data of glycolate, glyoxylate, and glycerate were not included in the INST-MFA modeling for the following reasons: Glycolate and glyoxylate were in low abundance, resulting in relatively large variability in the labeling data (Supplementary Data 2). The labeling of glyoxylate and glycerate was extremely slow, with the average ^{13}C-enrichment reaching only 12% and 16% at 30 min of labeling under high oxygen (Supplementary Figure 5). The lower-than-expected pool size and labeling plateau of glyoxylate was also found in another $^{13}CO_2$ labeling study in Arabidopsis [13]. The slow labeling and large pool size of glycerate indicates the presence of large metabolically inactive pools due to cellular

heterogeneity. To accurately describe the labeling kinetics of metabolites such as glycine, serine, and sugar phosphates, dilution parameters were used in the INCA model to introduce metabolic inactive pools that are not significantly ^{13}C-enriched within the 30 min labeling period (Supplementary Data 1).

To test for the presence of the export of photorespiratory glycine and serine to alternative sinks besides phloem, we removed these two equations (dummyGLY -> outGLY, dummySER -> outSER) from the model. This resulted in slightly increased degrees of freedom (1,714 to 1,716), but worse model fitting for 21% oxygen and 40 % oxygen. The SSR increased from 786 to 882 in the 21% oxygen model and increased from 791 to 1,162 in the 40 % oxygen model. However, the SSR for the 2% oxygen model was not changed when these two reactions were removed, consistent with the very little flux through these export reactions under low photorespiratory conditions. Including the export reactions of photorespiratory glycine and serine to alternative sinks improved the goodness of fit between the flux solution and measured data when modeling photorespiratory scenarios.

Although there is limited evidence for the export of glycerate from the photorespiratory pathway, we included this reaction in the model and added the glycerate labeling data. The addition of the glycerate export reaction only slightly changed the model fitting. The glycine and serine export rates were very close to the values estimated from the original model without the glycerate export reaction. The resulting glycerate and glycine export rate was < 0.1 µmol g^{-1} FW hr^{-1} at all oxygen conditions, while the serine export rate was <0.1 µmol g^{-1} FW hr^{-1} at 2% oxygen, 3.7 µmol g^{-1} FW hr^{-1} at 21% oxygen, and 7.0 µmol g^{-1} FW hr^{-1} at 40% oxygen.

Removing all dilution parameters from the model resulted in worse model fitting but predicted the similar trend of export rates of photorespiratory glycine and serine to alternative sinks besides phloem as compared to the model with dilution parameters (Fig. 1). Without dilution parameters, the glycine export rate was <0.1 µmol g^{-1} FW hr^{-1} at all oxygen conditions; the serine export rate was <0.1 µmol g^{-1} FW hr^{-1} at 2% oxygen, 5.0 µmol g^{-1} FW hr^{-1} at 21% oxygen, and 8.1 µmol g^{-1} FW hr^{-1} at 40% oxygen.

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