

1 Thermal Properties of Materials under Pressure

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Abstract: High-pressure science and technology have achieved significant progress in the past decades, enabling consistent innovation of knowledge in physical and chemical sciences, materials science, earth science, and multi-disciplinary combinations. However, measurements of thermal properties of materials under pressure and understanding of associated thermal transport mechanisms remain some of the most difficult challenges and complex topics in the high-pressure research. Thermal properties of materials are extremely important for many practical applications such as in energy conversion of devices and thermal management of electronics. However, profound understanding of thermal transport and efficient modulation methods are currently lacking. Recent breakthroughs in high-pressure experimental techniques enable *in situ* measurements of thermal conductivity at extreme pressure-temperature conditions providing a unique insight into thermal transport mechanisms and novel capabilities to realize reversible modulation of materials' thermal properties. In this Review, we discuss recent progress in thermal characterization techniques developed at high pressures, in the determination of the thermal properties of materials ranging from gases, liquids, solids to solid-solid interfaces, as well as in establishing the correlated thermal transport mechanism. Particular attention is paid to applications of high-pressure, high-temperature experimental simulation of materials at Earth's interior and the high-pressure modulations of thermal-physical properties in thermoelectrics. An outlook is also given for future theoretical and experimental investigations of high-pressure thermal properties and potential high-pressure applications on devices' thermal management techniques.

Keywords: High-pressure, thermal-physical properties, thermal conductivity, interfacial thermal conductance

1. Introduction

High pressure science and technology, as a novel and fast progressing research area, coupled with the keep-progressing achievements in characterization techniques, have been mainly exploited to simulate the high pressure environment in planetary bodies and to investigate the physical and chemical properties of materials within them.¹⁻¹⁰ Meanwhile, fruitful opportunities and giant scientific potential have also emerged in multi- and inter-disciplinary research area due to the improvements in versatile high-pressure techniques, which can be achieved through static and dynamic compression approaches.^{3,11-24} Pressure has long been recognized as an important thermodynamic variable (besides the temperature and chemical composition) that can effectively adjust or even reversibly modulate many physical properties of materials. It, however, has also been limited or lagged in applications to relatively low pressure region for quite a long time due to previous unattainability in apparatus and techniques, until the turn of this century when the megabar (100 GPa) pressure diamond anvil cells (DAC) techniques together with a series of in-laboratory integrated characterization techniques were developed.^{11,25-28} Static high-pressure

85 generation techniques primarily consist of Piston-cylinder pressure cell (with a
86 pressure limit <10 GPa), multi-anvils cell (with a maximum pressure of ~90 GPa) and
87 DAC (with a highest pressure up to ~770 GPa) apparatus.²⁹⁻³⁴ Pressure in DAC can be
88 conveniently determined and monitored by the fluorescence peak shift of a ruby
89 sphere crystal (can be widely available until ~100 GPa) placed close to the sample of
90 interest, with an uncertainty of 2-4%, or directly marked by the Raman spectra peaks
91 collected from the near-sample diamond anvil (can be calibrated up to ~410 GPa with
92 an uncertainty of ~15%).^{35,36} Since the invention, due to its convenience and many
93 inherent advantages, DAC has been successfully combined with various physical and
94 chemical characterization techniques, in which a hydrostatic, stable and homogeneous
95 compressive strain (30% or even higher) can be generated without any additional
96 damage to the samples.^{6,8,20,23,37-40} In contrast, traditional strain generation methods
97 such as the mechanical stretching or compression of metals, or the elongation and
98 bending of semiconductors to obtain the strain where samples sit on a flexible
99 substrate, the strain typically can be achieved only no more than 4%,⁴¹⁻⁴⁴ this is far
100 below the requirements of systematic investigation on fundamental mechanism of
101 strain modulation. Moreover, the slippage across the sample-substrate interface or the
102 imperfect strain transfer across different layers can introduce large uncertainties in the
103 strain values, such as a significant scatter or even contradiction in the obtained values
104 derived via strain-induced Raman peak shifts.⁴⁵⁻⁴⁷ Naturally, DAC has become the
105 broadest platform for theoretical and experimental researches from the new dimension
106 of pressure.

107 In the pressure dimension, many unexpected physical properties can emerge as
108 all materials properties will undergo changes in different high-pressure regions, and in
109 this sense the changes normally move from the ordinary condensed-matter category to
110 a new dense-matter physics world.^{5,7-10,15-21,23,24} It is not surprising to see all the
111 electronic, elastic, structural, magnetic, and chemical properties being altered
112 drastically upon compression, mainly due to the interatomic bonding and orbital
113 coupling, interlayer wave function overlap, and valence band splitting being
114 modified.^{8,15,19,20,40,48,49} The phononic or vibrational properties can also be greatly
115 changed in many materials at high pressures, and some of them can be directly probed
116 by *in-situ* Raman spectroscopy and Brillouin scattering techniques.⁵⁰⁻⁵³ More
117 importantly, the materials properties will be tuned by pressure across the various
118 regimes according to conventional classification such as insulators, semiconductors
119 and superconductors,^{5,6,10,23,24,54,55} amorphous and crystalline solids, *etc.*, and many
120 new materials or phases can also be surprisingly created.⁵⁶⁻⁵⁹ To investigate these
121 novel physical and chemical properties under pressure, continued progresses have
122 been achieved in high-pressure characterization techniques, including the structural,
123 optical, electrical, magnetic and thermal characterization of samples within the DAC
124 apparatus.^{1,3,8,19,20,60} For example, the changes of lattice parameters, phase structure as
125 well as the resultant strain induced by pressure can be accurately obtained by the
126 combination of synchrotron X-ray diffraction (synchrotron-XRD) with DAC
127 techniques; these developments result in great scientific progress including the recent
128 breakthrough in exploration of metallic hydrogen and high-temperature

129 superconductors.^{8,16,17,19} Also, the critical phenomena, phase evolution, stress state
130 and magnitude, phonon modes and electronic bandgap which change under pressure
131 or the equivalent strain, can be very sensitively and accurately detected
132 experimentally by combining techniques such as *in-situ* micro-Raman,
133 micro-photoluminescence and absorption spectroscopies with the DAC
134 techniques.^{6,7,18,20,23,40,41,61-63} Despite such tremendous achievements in high-pressure
135 characterization of materials properties, their thermo-physical properties remain one
136 of the most difficult properties to be reliably and accurately measured at high
137 pressures.

138 Thermo-physical properties are important parameters playing crucial roles in
139 thermal transport and thermal management; the latter has been a critical bottleneck in
140 many high-power devices and heat exchange systems, for example, transistors and
141 thermoelectric devices, and electronic systems in mobile-phone and computer
142 terminals.⁶⁴⁻⁶⁷ Thermo-physical properties discussed here are emphasized in
143 non-metallic materials in which phonons are the major heat carriers; these properties
144 mainly involve thermal diffusivity, thermal conductivity, heat capacity, phonon
145 velocity, phonon mean-free-path (MFP), phonon lifetime, interfacial thermal
146 conductance (ITC), and thermal resistance. Phonons have a large variation in their
147 frequencies and MFPs; for bulk materials at room temperature, MFP is normally in
148 the range of 1-100 nm which is within the microstructure scale of the bulk materials.⁶⁸
149 However, in micrometer-scale and even nanometer-scale samples, phonon MFP will
150 be comparable or even limited to the sample length or width or thickness, resulting in
151 stronger boundary scatterings and thus lowering the thermal conductivity. Concerning
152 the aspects of common measurements and applications, thermal conductivity and ITC
153 are the most discussed correlative thermal parameters, where thermal conductivity is
154 significantly affected by sample thickness, width and length,⁶⁹ and also by strong
155 point-defects or dislocation scatterings and isotope scatterings.⁷⁰ While ITC is
156 normally related to the interfacial atomic bonding strength, atomic mixing, interfacial
157 microstructure and interfacial roughness as well as the materials quality near the
158 interface. Both thermal conductivity and ITC are temperature dependent parameters,
159 varying in different temperature regions through different phonon scattering
160 mechanisms; this makes temperature, the most often used external stimuli because of
161 its easier availability, a useful variable to tune, modulate, and control the thermal
162 performance of materials and devices. Besides, strain or pressure can also be used to
163 tune the thermal conductivities of materials and ITCs at interfaces, despite this part of
164 research is still in its early stage due to the challenges and difficulties in high-pressure
165 thermal characterization techniques.^{3,22,38,39,49,60,71-76} Thermal conductivities of some
166 solids and liquids at pressures only up to a few GPa determined with the errors of a
167 few percent were summarized in 1984; this includes the proposed pressure-dependent
168 thermal transport mechanisms for alkali halides, simple metals and non-metallic
169 liquids being theoretically described.⁷⁷ Later in 2007, the pressure dependences of
170 thermal properties of some geophysically relevant minerals up to 10 GPa were
171 summarized, reporting the re-examined experimental values of the thermo-physical
172 parameters with higher accuracy determined using a contact-free laser-flash method.¹⁴

173 Recently, some theoretical and experimental studies have been performed to probe the
 174 strain or high pressure effects on thermal conductivities in a variety of materials
 175 demonstrating rich high-pressure phenomena.^{3,22,38,39,49,60,74-76,78,79} For example, the
 176 thermal conductivity of solid iron at planetary core conditions up to 130 GPa and
 177 3000 K was directly measured using pulsed laser transient heating method.³
 178 Knowledge of thermal-physical properties under pressure can also advance
 179 understanding of heat transport mechanism and provide useful guidelines for the
 180 tunability of thermal properties via pressure or strain method in thermal management,
 181 such as the recent breakthrough in thermal conductivity measurements of MoS₂ under
 182 pressure.⁴⁹ This Review will focus on the discussion of pressure-induced
 183 modifications in thermo-physical properties of materials, thermal transport behaviors
 184 and thermal applications under pressure ranging from ambient pressure up to
 185 hundreds of GPa.

186 2. Basic theory of thermal transport

187 Heat flux generated and transported through a material is normally described by
 188 Fourier's law of heat conduction:

$$189 \quad \dot{Q} = -\kappa \nabla T, \quad (2.1)$$

190 where $\dot{Q}$ is the local heat flux with a unit of Wm^{-2} , κ is the thermal conductivity of
 191 material in a unit of $\text{Wm}^{-1}\text{K}^{-1}$, ∇T is the local temperature gradient through the
 192 material thickness in a unit of Km^{-1} (see schematic in Figure 1a). The Fourier's law
 193 describes how efficiently the heat can be conducted through a material from its high
 194 temperature to low temperature regions. However, in thermal characterization of
 195 various materials and practical device applications, one should take into account that
 196 besides transport through the materials, heat also has to be conducted across the
 197 interfaces of different materials (including dissimilar, epitaxial, bonded, and contact
 198 interfaces). In this case, the thermal resistance of the interface is commonly employed
 199 to evaluate the thermal transport efficiency. The thermal resistance is normally
 200 described as:

$$201 \quad R = \Delta T / Q, \quad (2.2)$$

202 where ΔT is the temperature difference between the two surfaces (in a unit of K), and
 203 Q is the thermal energy conducted through these two surfaces area (in a unit of W).
 204 Therefore, the total thermal resistance R consists of the part along the thermal
 205 conduction path of the materials that is dependent on the thermal conductivity and
 206 thickness of the materials, as well as the thermal boundary resistance at the interface
 207 of two different materials which depends on many interfacial factors including the
 208 micro- and nano-structure, atomic bonding condition, atomic mixing, surface
 209 roughness, surface termination, contamination, and others.

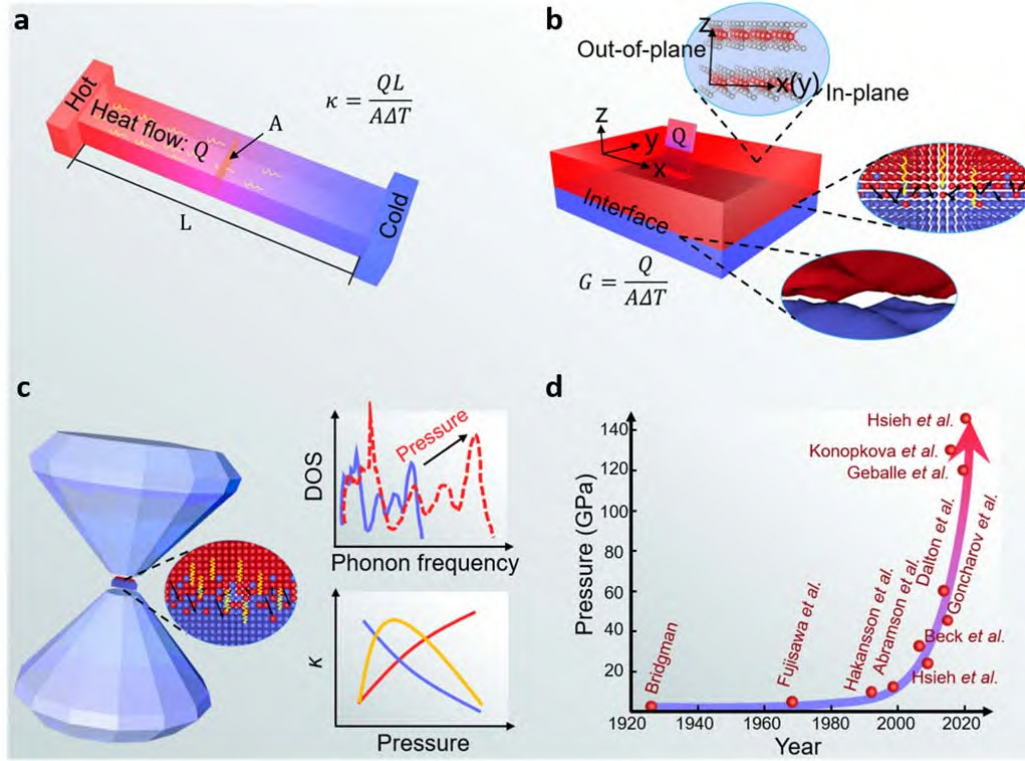


Figure 1 | Thermal transport in materials at ambient and high pressures. **a** | The description of thermal transport and thermal conductivity by Fourier's law of heat conduction. A , Q , and L represent the cross-sectional area, the total thermal energy of heat flow through the cross-sectional area, and the length of thermal transport, respectively. ΔT is the temperature difference from the hot to the cold terminals. The thermal conductivity is expressed as $\kappa = \frac{QL}{A\Delta T}$. **b** | The description of interfacial thermal conductance between two different materials as well as the in-plane and out-of-plane thermal conductivity. Heterointerface contact normally includes conditions of the full contact and the limited contact where some air voids are inevitably introduced during the integration; both of them normally result in an obvious temperature drop ΔT across the interface due to the mismatch of phonon scattering between the two different materials. The interfacial thermal conductance is expressed as $G = \frac{Q}{A\Delta T}$. **c** | The schematic of thermal transport at high pressures generated within a diamond anvil cell as well as the schematic evolution of phonon density of states and thermal conductivity with respect to pressure. In general, the application of pressure compresses the crystal lattice and extends the phonon frequency, thereby can promote the heat carrying ability of electrons and some phonons, bringing about the modification of the thermal conductivity under pressure (e.g., increasing trend, decreasing trend and anomalous trend). **d** | The progress in thermal conductivity measurements at high pressures. The values of pressure and year in panel **d** are taken from Refs. ^{3,38,39,60,115,116,137,150,189,213}.

2.1. Thermal conductivity

Heat transfer ability at the macroscopic scale is characterized using thermo-physical properties such as the thermal conductivity. These thermal properties in turn are correlated to the atomic-level properties and the transport process, mainly through microscale energy carriers like phonon, electron and hole, fluid particle, photon, and so on. In non-metallic crystals such as semiconductors and some thermoelectrics, the main heat carriers are phonons. The total thermal conductivity κ of a crystal involves contributions from a lattice (phonon, κ_l) and an electronic (electron, κ_e) component, which is described by:⁸⁰

$$\kappa = \kappa_l + \kappa_e = \sum_j \frac{1}{3} C_v v_j l_j + L \sigma_e T, \quad (2.3)$$

where C_v , and v_j are the molar heat capacity at constant volume, and the phonon group velocity of the j th phonon mode, respectively; l_j is the MFP of the j th phonon mode which can be further expressed as $l_j = v_j \tau_{p,j}$ where $\tau_{p,j}$ is the effective phonon scattering time of the j th phonon mode, including contributions from all scattering factors; L , σ_e , and T are the Lorenz number, electrical conductivity, and temperature, respectively. In metals, κ is dominated by κ_e owing to the large concentration of free electrons. Both electrical and thermal transport processes involve significant contribution from the free electrons in metals, mainly because the free electrons can be accelerated to a higher drift velocity with a higher energy by the electric field. Hot electrons at higher energy states can carry more thermal energy than cold electrons. The Wiedemann-Franz law ($\kappa_e = L \sigma_e T$) is traditionally used to calculate the contribution of electrons to the thermal conductivity on the basis of σ_e , where $L = (\pi^2/3)(k_B/e)^2$ is normally applied for metals (e and k_B are the electron charge and Boltzmann constant, respectively). In contrast, in semiconductors various doping levels result in much smaller σ_e , and thus smaller κ_e . Therefore, in non-metals such as the semiconductors of GaN, diamond, Si and SiC, and the thermoelectrics of Bi₂Te₃ within the operational temperature range of electronic devices applications, heat is mainly transferred through the lattice vibrations,⁸⁰ making κ_e negligible compared to κ_l . Therefore, κ_l , denoted as κ from now on in the text, will be discussed primarily in this work.

Phonons can be scattered by various mechanisms in an imperfect crystal, shortening the average distance they can travel through the lattice, *i.e.*, lowering their MFPs. Contribution of different mechanisms on phonon scattering are temperature dependent, thus resulting in a temperature-dependent thermal conductivity. Typical phonon scattering mechanisms include contributions from crystalline boundaries, lattice defects and impurities, electrons, and other phonons; below we denote them in terms of the characteristic scattering times.

(1) Phonon-boundary scattering (τ_{p-b} , see the schematic in Figure 2a): It is noted that below 10 K, scattering from the crystal boundaries (crystal size or grain size) dominates because of the finite size of the crystal,⁸⁰ especially in micro- and nanoparticles. Therefore, the thermal conductivity of an ultrathin film may be

270 significantly different to its bulk value due to the phonon transport hindered by
 271 increased boundary scattering. This is especially true when its thickness is close to the
 272 value of the dominant phonon MFP. For example, the dominant phonon MFP range
 273 can be suppressed to a few nanometers by defects, impurity and grain boundaries in
 274 Bi_2Te_3 .⁸¹⁻⁸³ This illustrates why Bi_2Te_3 films with 20-100 nm nanostructures present a
 275 few times smaller thermal conductivity when compared to bulk and pure Bi_2Te_3 due
 276 to the suppression of long MFP phonons.⁸²

277 (2) Phonon-impurity scattering (τ_{p-im} , see the schematic in Figure 2b): The crystal
 278 lattice can be distorted by point defects such as impurities/dopants, vacancies, and
 279 isotopes because of their different mass compared to the host atoms, therefore
 280 resulting in stronger phonon scatterings. In addition, these point defects can change
 281 the bonding between atoms. Linear defects, for example dislocations, also scatter
 282 phonons because the cores and the surrounding strain fields of these dislocations can
 283 change the crystal density and phonon group velocity, respectively.

284 (3) Phonon-phonon scattering (τ_{p-p} , see the schematic in Figure 2c): Inter-phonon
 285 scattering is the most significant scattering mechanism at higher temperature and is
 286 highly temperature dependent. In principle, inter-scattering processes between
 287 phonons can induce a local strain that changes the phonon velocity. Three-phonon
 288 processes are normally exploited to describe the phonon-phonon scattering and can be
 289 further divided into the Normal (N) and Umklapp (U) processes. This scattering
 290 mechanism is more complicated and has different behaviors for low temperature and
 291 high temperature conditions, as well as for longitudinal and transverse polarizations.

292 (4) Phonon-electron scattering (τ_{p-e} , see the schematic shown in Figure 2d):
 293 Contributions from electron scatterings sometimes also need to be considered. In
 294 general, the electron-phonon scattering term is complicated and strongly depends on
 295 the electron carrier concentration, normally resulting in a negligible strength of this
 296 scattering term.

297 (5) Van der Waals (vdW) scattering (τ_{vdW}): In the special case of layered materials,
 298 one crystal orientation bonded via weak vdW force limits heat transport along that
 299 crystal orientation and is partly responsible for the lower thermal conductivity
 300 observed along this direction. Therefore, an additional vdW scattering rate is added in
 301 order to simulate the extra resistance to the heat transport.^{84,85} In this vdW scattering,
 302 the transmissivity between two neighbor layers contacted by vdW bonding is
 303 modelled with a simplified formulation of Prasher transmissivity,⁸⁶ while the
 304 interatomic potential is often described by a short-range one and a long-range one,⁸⁷⁻⁸⁹
 305 where the short-range one is typically Morse potential.

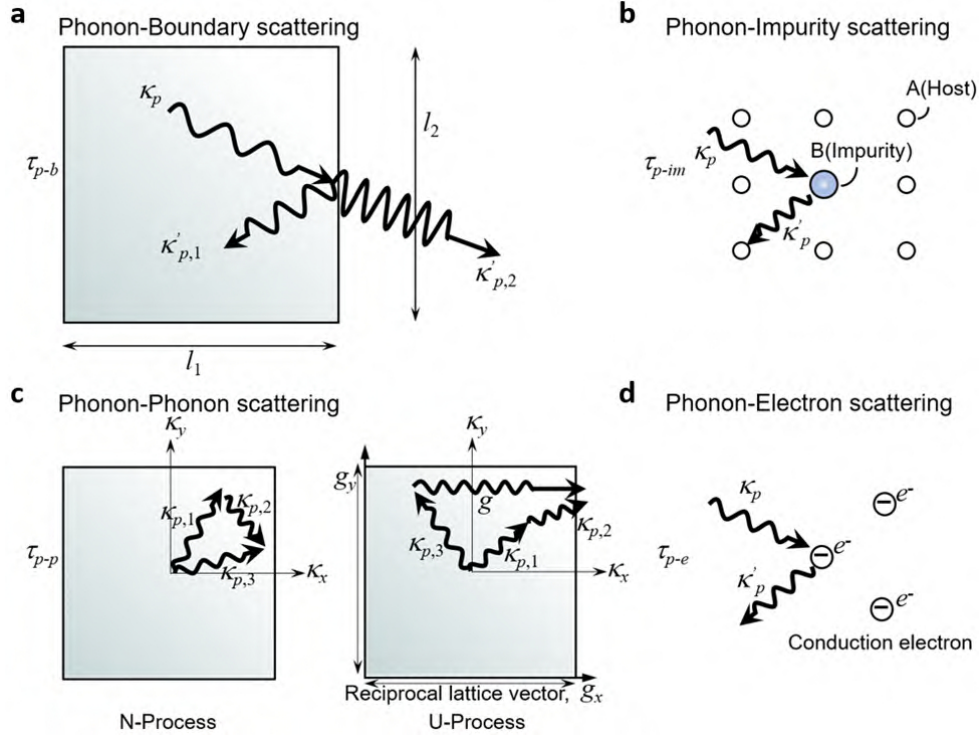


Figure 2 | Phonon scattering mechanisms. **a** | Phonon-boundary scattering, τ_{p-b} , where l_1 and l_2 are the crystal linear dimensions. **b** | Phonon-impurity scattering, τ_{p-im} . **c** | Phonon-phonon scattering, τ_{p-p} , where N-Process represents the Normal process and U-Process represents the Umklapp process. **d** | Phonon-electron scattering, τ_{p-e} . Note that k_p , $k_{p,1}$, $k_{p,2}$, and $k_{p,3}$ represent incident phonons, while $k'_{p,1}$, $k'_{p,2}$, and $k'_{p,3}$ represent scattering phonons, and the different subscripts mean the phonons with different momentums. Panels **a-d** are adapted with permission from Ref.⁶⁹ ©Springer Nature Publishers Limited.

2.2. Interfacial thermal conductance

When heat transfers across heterogeneous material interfaces, an abrupt temperature drop may occur at interfaces, forming a significant bottleneck for the thermal transport. The efficiency of this interfacial heat transfer is quantified by the interfacial thermal conductance (ITC, it is normally expressed by the symbol of G) which physically correlates the heat flux to the temperature drop at the interface via formula of $G = \dot{Q}/\Delta T$, where $\dot{Q}$ is the heat flux (in units of Wm^{-2}) across the interface area and the units of G is $\text{Wm}^{-2}\text{K}^{-1}$ (see the schematic in Figure 1b). The first quantitative experimental report on G was conducted by Kapitza who measured the interfacial thermal conductance between a solid copper and liquid helium in 1941, which is later also termed as “Kapitza conductance”.⁹⁰ Mathematically, G is the inverse of thermal boundary resistance (TBR). On the one hand, ITC is a critical thermal parameter that governs the ability or efficiency of thermal energy dissipation in a wide range of devices for applications and technologies, such as high power and high frequency transistors, high frequency photodiodes, high power light emitting

331 diodes, and phase change memory devices.^{64,65,91-94} On the other hand, low ITC also
332 implies that the equivalent ultrathin interfacial material formed at the heterogeneous
333 interface has a low thermal conductivity which is desirable for applications in
334 thermoelectric materials, thermal barrier coatings, etc.^{67,95-98} All these various
335 applications require fundamental knowledge of ITC related thermal-physical theories.

336 Better understanding of the interfacial thermal transport mechanisms following
337 Kapitza's groundbreaking work have been achieved by focusing investigations on
338 its interplay relationship with interfacial properties and the near-interface material
339 properties.^{64,65,99-107} In terms of solid-solid interfaces, the main controlling factors of
340 ITC are (1) the phonon dispersion and density of states for energy carriers in the
341 respective materials and the near-interface materials region; (2) the interface quality,
342 *i.e.* the interfacial imperfection (ideally perfect smooth interfaces are almost never
343 realized in realistic situations), such as disorder, amorphous, microstructural
344 roughness and atomic mixing which can alter materials vibrational properties and the
345 strength of the interfacial atomic interactions; (3) temperature. In this Review, we first
346 introduce the conceptual theory of ITC at the interfaces, together with the
347 experimental work that has challenged this theoretical knowledge. Then we present
348 some of recent technological and theoretical achievements in understanding of
349 mechanisms of ITC and its modulation explored in a new pressure dimension.

350 **2.3. Thermal transport mechanism under pressure**

351 Application of extreme pressure profoundly affects the phononic and vibrational
352 properties by changing the elastic constants, sound velocities, phonon lifetimes,
353 phonon densities of states, and interface bonding stiffness, etc.; these are closely
354 related to the phonon transport (equivalently, thermal transport) properties of
355 materials. Upon compression, various pressure dependences of thermal conductivities
356 are reported, making the detailed mechanism of thermal transport modulated by
357 pressure particularly complicated (Figure 1c). In general, the pressure dependences of
358 κ are classified into increasing, decreasing, independent and anomalous trends.
359 Normally, on the one hand, the strain generated by pressure will enhance the atomic
360 interaction and compact the intralayer and interlayer bonds, heavily modify the
361 phonon dispersions, thus leading to the phonon velocities being greatly enhanced,
362 which means a drastic increase in κ . Such enhancement in κ with respect to pressure
363 in some cases is nonlinear. This is found to arise from the combined effects of
364 decreased phonon relaxation time coupled with increased phonon group velocity.¹⁰⁸
365 The electronic thermal conductivity discussed in previous section largely depends on
366 the electronic conductivity and thus can have several orders of magnitude increase.
367 However, it is still negligible compared with the value of lattice thermal conductivity
368 in non-metals. On the other hand, pressure-induced phonon anharmonicity and
369 phonon softening are thought to be as the main mechanism for the decreasing trend of
370 κ upon compression.¹⁰⁹ In accord with this, first-principles calculations reveal that the
371 decreased κ under pressure is mainly ascribed to the stronger third anharmonic
372 interaction, the large mass ratio and the significant acoustic-optical frequency gap.¹¹⁰

For the pressure independent trend of κ , the strong electronic correlation effects driven by the electronic topological transition are presently thought as the main reason.¹¹¹ As to the anomalous trend of κ , several different mechanisms have been put forward: (1) In the case of anomalously decreased pressure dependence of κ , the failure of three-phonons scattering process that involving two acoustic phonons and one optical phonon is thought as the fundamental source; this intrinsic scattering process does not occur in materials with a large acoustic-optic frequency gap at high pressure; instead, the scattering between acoustic phonons will dominate.⁷⁴ (2) In another case of non-monotonically decreased pressure dependence of κ , *i.e.*, first increase and then decrease with increasing pressure, the competing scattering processes of three-phonon and four-phonon interactions at high pressures,⁷⁶ or the interplay between group velocity and phonon relaxation time under pressure,¹¹² are the possible mechanisms. (3) In the case of diverse pressure dependence of κ revealed in some rare-earth pyrochlores, the competition between the enhancement of group velocity of phonon modes and the reduction of phonon relaxation time determines the pressure dependence.¹¹³

As pressure can continuously tune materials' lattice dynamics through varying the anharmonicity of atomic/molecular bonding, the interfacial thermal transport behaviors will also be significantly modified. Recent studies show that interfacial stiffness is a crucial parameter for G under pressure, and the interfacial stiffness strongly dominates the thermal transport at weak interfaces but plays a minor role for strong interfaces.¹¹⁴ Moreover, applying pressure can change the phonon scattering processes across the interfaces, for example, from elastic radiation process to inelastic partial transmission processes through modifying the phonon densities.⁷⁹

3. Characterization methods of thermal-physical properties of materials under pressure

Thermal characterization techniques are normally classified as steady-state and transient methods, for both temperature-dependent and pressure-dependent measurements, and for both bulk and thin-film materials. Within these high-pressure thermal characterization methods, through nearly 100-years development (Figure 1d), the former mainly comprises the thermocouple and heater assisted technique such as Ångström method developed in the Piston-cylinder pressure cell or multi-anvil cell, as well as the thermal grating method, the thermocouple method and Raman-based opto-thermal method applied in DAC, while the later includes the transient hot-wire method for Piston-cylinder pressure cell and pulsed laser transient heating method and time-domain thermorefectance method for DAC apparatus. Piston-cylinder pressure cell and multi-anvil cell (the normally used is the first-order multi-anvil cell with eight truncated cubic inner anvils) own large size allowing more space for the thermocouple layout, but are normally limited in pressure range below 10 GPa.^{33,115-118} However, pressure of at least 20 GPa is usually needed to obtain ~50%

increase in parameters such as the elastic constants that are relevant to thermo-physical properties.¹¹⁹⁻¹²³ DAC setup can generate much higher pressure exceeding 100 GPa. However, the very narrow space left between the anvils makes it extremely difficult to spread the wires needed by thermocouples and heaters. Fortunately, recently developed optical contactless methods e.g., Raman opto-thermal method and time-domain thermoreflectance techniques can fulfill the desires to probe the thermal properties at extremely high pressures. These high-pressure thermal characterization methods will be introduced below.

3.1. Steady-state methods

3.1.1. Thermal-couple & heater method in Piston-cylinder pressure cell

The conventional technique to measure the thermal conductivity employs thermocouples and heaters placed together with samples into a symmetric Piston-cylinder pressure cell in the following sequence of *CTSTHTSTC*, where *H* is a heater embedded in the center of the cell, *S* is the sample, *T* is the thermocouple embedded in different regions of the sample for measuring the temperature drop across the sample, and the pressure is applied between two ends of the Piston-cylinder cell (*C*).^{33,124,125} The thermal conductivity can be obtained from the temperature gradient measured from the thermal couples embedded across the sample by solving the Fourier's equation of heat conduction (Eq. 2.1). This method has been widely exploited to measure the pressure-dependent thermal conductivity of various polymers up to 0.4 GPa (the limit of Piston-cylinder pressure cell is only a few GPa).^{33,124-126}

3.1.2. Ångström method in multi-anvil cell

A thousand-tons press applied in a multi-anvil cell or modern gem anvil cell can push the pressure and temperature limit to ~90 GPa and ~3000 K, respectively, providing a higher pressure and a more convenient steady-state environment for thermal conductivity measurements. Using Ångström method in such a multi-anvil cell, pressure-dependent thermal diffusivities of silica glass and olivine, CaGeO₃ perovskite, *etc.*, have been successfully measured.^{117,118,127} In the Ångström method (see the schematic cross-section in Figure 3a¹²⁷), the sample with a diameter of ~3-4 mm is encapsulated into a cylindrical tube where a graphite- or LaCeO₃-like heater is mounted to heat the attached samples up to high temperatures; meanwhile, the thermocouple wires need to be embedded into the sample or fixed next to the sample for temperature measurements. This fabricated sample-containing cylinder is then sealed into a MgO octahedron (with an edge length of ~14-25 mm), and compressed together by 6 outer anvil wedges and 8 tungsten carbide made cubic inner anvils (with ~3-10 mm truncated edge length).¹²⁵ Under this apparatus configuration, for instance, using a loading of 900-tons press and inner anvil with truncated length of 3 mm, a

high pressure environment of ~ 25 GPa can be achieved.¹²⁸ Following the above design, applying a sinusoidal temperature wave with a variable frequency in a central wire heater creates a heat wave propagating radially through the surrounding sample of the cylindrical shape. Note that the amplitude and phase of the varying temperature wave are relevant to the thermal diffusivity of the compressed sample. By measuring the amplitude ratio and phase difference of the temperature between the center and outer surface of the sample, the thermal diffusivity at different pressures up to ~ 20 GPa can be extracted.^{115,117,118,129}

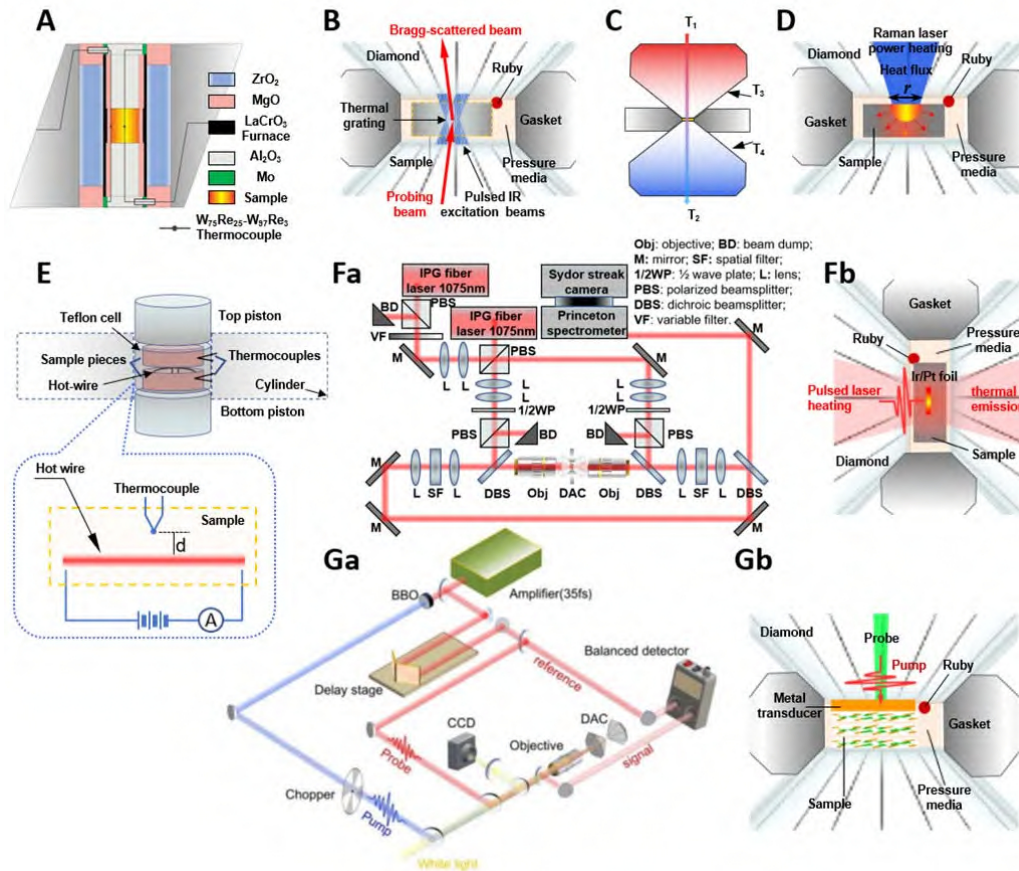


Figure 3 | The schematic of thermal characterization methodologies developed under pressure. **A** | The Ångström method developed in the multi-anvil cell, **B** | the thermal grating method applied in DAC, **C** | the thermocouple method used in DAC, **D** | the Raman-based opto-thermal method used in DAC, **E** | the transient hot-wire method developed in the Piston-cylinder pressure cell, **Fa-Fb** | the pulsed laser transient heating method applied in DAC, **Ga-Gb** | the time-domain thermorefectance method and picosecond transient thermorefectance method applied in DAC. Panels **A-D** are steady-state characterization methods while **E-G** are transient characterization methods. Panels **A-G** are taken or adapted with permission from Refs.^{49,109,116,130-135}, respectively. ©Elsevier Publishers Limited. ©AIP Publishers Limited. ©Springer Nature Publishers Limited. ©APS Publishers Limited.

472 3.1.3. Optical thermal grating method in DAC

473 An optical thermal grating method^{131,136,137} compatible with the DAC technique
474 (see Figure 3b) is developed for measuring thermal diffusivity at high pressures up to
475 12 GPa.¹³⁷ In this method, two mode-locked pulse lasers, with the pulses duration of
476 ~80 ps and the laser wavelengths among the sample absorbing range, are incident at
477 an angle of 2θ onto the sample to form a constructive interference with the laser
478 intensity distributed periodically. After absorption, the incident laser produces a
479 periodic distribution of both temperature and refraction index in the sample. Here the
480 period of the thermal grating is defined as $d=\lambda/(2\sin\theta)$, where λ is the laser
481 wavelength. Then a Bragg-diffracting probe pulse is incident onto the same position
482 to monitor the temperature evolution as a function of time. At certain conditions,
483 when the heat diffusion from pulse laser heating can be approximated as
484 one-dimensional, the decay rate of the periodic temperature distribution R_d then can
485 be used to extract the thermal diffusivity of the sample according to $R_d=4\pi^2D/d^2$,
486 where D is the thermal diffusivity. This method is a contactless, optical method which
487 can allow to avoid previous complex and low-efficient fabrication of thermocouple
488 and heater embedded samples.

489 3.1.4. Thermocouple method in DAC

490 A thermocouple based steady-state method adapted to the DAC apparatus was
491 recently put forward in 2018 (Figure 3c).¹³² Four ultra-tiny thermocouples have been
492 carefully mounted onto the culets/tables parts and the near-culet side-surface parts of
493 diamond anvils, respectively. Then combining a steady-state thermal finite element
494 model (FEM) with the above four-points' temperatures measured from the
495 thermocouples at each pressure, an actual temperature field distribution within the
496 pressured DAC can be resolved in the FEM based thermal model. Therefore, by
497 adjusting the thermal parameters of the samples until a consistency is achieved
498 between the measured temperature and the model calculated temperature at the exact
499 same positions of the DAC, the thermal conductivity of both diamond anvils and the
500 sample at each pressure and their pressure dependences can be obtained.

501 3.1.5. Raman-based opto-thermal method in DAC

502 Phonon frequencies of semiconductors normally display a negative dependence
503 with temperature due to anharmonic effects. Increasing temperature causes the lattice
504 of a material to expand which decreases the frequency of the lattice vibrations
505 (phonons), resulting in a temperature dependent phonon frequency shift; this
506 temperature induced phonon shifts can be directly measured from the Raman spectra
507 thus contactlessly probing the local sample temperature in the laser spot. This
508 principle was later developed into a Raman-based opto-thermal method to measure
509 the thermal conductivity of materials. The principle can be simply described as the

510 following:

$$511 \quad \kappa = \left(\frac{2\alpha \Delta P}{\pi r \Delta T} \right) = \frac{2\alpha \chi_T}{\pi r \chi_P}, \quad (3.1)$$

512 where α is the absorption coefficient, ΔP is the incident laser power variation, r is the
513 laser spot radius, ΔT is the temperature variation from spot center induced at the
514 corresponding power variation, while χ_T and χ_P are the coefficients of temperature
515 dependence and power dependence of Raman mode shifts, respectively.

516 This method uses the excitation laser from the Raman spectroscopy
517 simultaneously as a local heating source and a thermometer sensor; the Fourier heat
518 transport equation is further solved analytically to extract the thermal conductivity. In
519 detail, first, using an external heat source and conventional thermometry, the
520 temperature dependence of the Raman mode of the sample is measured using low
521 power laser excitation to avoid local heating by laser on the sample; this temperature
522 dependence serves as a calibration which can be used for local temperature
523 determination during the laser heating. This method also requires knowledge of the
524 laser power absorbed in the sample, which can be determined from the absorbance
525 measurements.¹⁰⁹ Then the Raman shift is measured as a function of the laser power,
526 from which the local heating is determined by applying the formerly determined
527 temperature dependence of the Raman frequency. Based on the above principles, a
528 Raman-based opto-thermal method combined with the DAC technique is developed to
529 measure the thermal conductivity of materials at high pressures (Figure 3d).
530 Application of these techniques has made it possible to measure the thermal
531 conductivities of several thermoelectric materials under pressure as well as their
532 pressure dependences.^{22,109,138,139}

533 3.2. Transient methods

534 3.2.1. Transient hot-wire method in Piston-cylinder pressure cell

535 The transient technique used to measure the thermal conductivity of materials
536 under pressure generated in a Piston-cylinder pressure cell is the hot-wire method,
537 alternatively termed as Ni-wire assisted electrical-bridge method (see the schematic in
538 Figure 3e). In detail, a Ni-wire was installed into a Piston-cylinder cell together with
539 the target sample and then was electrically heated at a constant power.^{116,140-147} At the
540 same time, an electrical-bridge was used to monitor the Ni-wire resistance for
541 recording the temperature change as a function of time. By comparing these data with
542 the temperature evolution as a function of time calculated using a theoretical model,
543 the thermal conductivities of the sample at different pressures can be extracted.
544 Through this method, the room temperature thermal conductivities of CsCl and NaCl
545 at pressures up to 2 GPa, as well as the temperature dependent thermal conductivities
546 of C_{60} and C_{70} at pressures up to 1 GPa were measured.^{116,143,148}

547 **3.2.2. Pulsed laser transient heating method in DAC**

548 In 2007, an optical pulsed laser transient heating (TH) method was developed in
549 combination with the DAC techniques to measure the high-pressure high-temperature
550 thermal diffusivity of several minerals at pressures and temperatures as high as 125
551 GPa and 2600 K.^{3,38,60,134,149} In such a TH thermal characterization method (see Figure
552 3f), a thin metallic coupler (typically, Ir or Pt) needs to be embedded into the target
553 sample; when performing measurements, this metallic coupler is heated by a pulsed
554 laser, resulting in a temperature variation as a function of time generated at its surface;
555 then the surface thermal emission spectrum from the metallic coupler is collected by
556 an all-reflective microscope and analyzed based on the black-body radiation theory.
557 Fitting the measured temperature evolution to the calculated results from a FEM
558 based thermal model, one can extract the thermal diffusivity of the sample at different
559 pressures and temperatures.

560 More recently, this technique has been modified to enable measurements of
561 metallic samples and to improve the accuracy of determination of thermal
562 conductivity of non-metallic samples. In the modified TH technique,¹³⁵ the sample is
563 continuously heated by double-sided lasers to a desired bulk temperature and then a
564 pulsed laser heated from one side generating a heat wave propagating across the
565 sample. The radiative temperatures are then monitored from both sides and the
566 thermal conductivity is further extracted from FEM calculations. It is noted that the
567 TH method is presently one of the limited methods appropriate for high-temperature
568 high-pressure thermal properties measurements, especially for sample temperatures
569 higher than 1400 K. This method has been employed to measure the thermal
570 conductivity of solid Fe at planetary core conditions of ~130 GPa and 3000 K,³ and
571 later the thermal conductivity of solid Fe and Fe-Si alloys up to 144 GPa and 3300
572 K⁶⁰ and the assemblage of lower mantle minerals crystallized from pyrolite glass at
573 temperatures up to 2500 K at 120 GPa.¹⁵⁰

574 **3.2.3. Time-domain thermoreflectance method in DAC**

575 Time-domain thermoreflectance (TDTR) is also a contactless optical method that
576 presently being widely applied to measure the thermal conductivity^{95,151} and
577 interfacial thermal conductance^{105,152} of thin films. The TDTR method can provide a
578 satisfied accuracy of better than 10% and a very fast measurement time; moreover, the
579 technique is readily compatible with DAC operation as the anvils are transparent for
580 visible and near infrared lasers, which are normally used in TDTR. To perform TDTR
581 measurements, a very thin metal film (typically Al, Au, Pt, or Cu with a thickness
582 <100 nm) is usually deposited on one smooth side of the sample, serving as a
583 transducer to absorb thermal energy from the pump laser and transfer the heat to the
584 sample. The fundamental principle of the metal film transducer is that the reflectivity
585 of a material is temperature dependent. For example, for the gold film, the change in
586 its surface reflectivity with temperature is approximately linear to the variations in its

587 surface temperature within the range of 300-800 K,¹⁵³ given by:

588
$$\frac{\Delta R}{R} = \left(\frac{1}{R} \frac{\partial R}{\partial T}\right) \Delta T = C_{TR} \Delta T, \quad (3.2)$$

589 where C_{TR} is the thermorefectance coefficient. Notably, the C_{TR} is dependent on the
590 type of metal and the wavelength of the probed reflected light. The thin metal
591 transducer film irradiated by a pulsed pump laser beam and absorbing the laser pulse
592 energy, serves as a periodic heating source. The temperature of the metal transducer
593 film rapidly rises due to the photoexcitation of electrons by the laser pulses. These hot
594 electrons thermalize the optical and acoustic phonons quickly (normally at the
595 picosecond scale). Then the generated heat gradually diffuses into the underlying
596 sample layers at different time scales; this process is governed by the thermal
597 conductivity, heat capacity and density of the materials and by the thermal boundary
598 resistance at the interfaces. The process of heat diffusion in the metal film transducer
599 layer and the sample layers leads to the decay of surface temperature, resulting in the
600 temperature-dependent change of surface reflectivity of the metal film transducer.
601 These reflectivity variations can be monitored by using another pulsed or continuous
602 wave probe laser focused on the same spot as the pump laser. Then, by combining the
603 measured thermorefectance curve with an analytical or numerical thermal model
604 calculation, the thermal conductivity and interfacial thermal conductance of the
605 sample can be extracted. This method requires that the pump and probe laser spots
606 must be well concentric while the sample surface should be as clean and smooth as
607 possible. The significant advantage of this method is that the sample size can be as
608 small as the laser spot (*i.e.*, $\sim 10 \mu\text{m}$ diameter), and the characterization temperature
609 can be ranged from liquid nitrogen temperature to few-thousands of Kelvins when
610 depositing the appropriate metal film transducers.

611 In 2009, the TDTR technique was successfully combined with the DAC
612 technique (*i.e.*, TDTR-DAC technique, see schematic in Figure 3g) to measure the
613 pressure-dependent thermal conductivity of muscovite mica up to 24 GPa.³⁹ Later, the
614 pressure evolutions of thermal conductivities of various materials from crystals to
615 amorphous polymers were investigated up to 60 GPa using TDTR-DAC technique. To
616 study the effects of variations in thermo-physical parameters of the metal film
617 transducers at high pressures on the extracted thermal conductivities of target samples,
618 the pressure evolutions of critical thermal parameters (*i.e.*, thermorefectance,
619 piezo-optical coefficient and physical stability) for several metal film transducers (*i.e.*,
620 Ta, Al and Au(Pd) thin films) were further investigated. It was found that the acoustic
621 strengths for Ta and Au(Pd) films are essentially independent of pressure, while it
622 drops abruptly and remains small for the commonly used Al film upon initial loading
623 and up to 12 GPa.¹⁵⁴ Thermal conductivities of liquid pressure transmission mediums
624 under pressure, such as silicone oil and methanol-ethanol mixture, were also
625 measured using TDTR-DAC technique.¹⁵⁵ Furthermore, the ITC (e.g. Al/SiC,
626 Al/graphene/SiO_x/SiC, Al/SiO_x/SiC¹¹⁴) as a function of pressure can also be extracted
627 using this method. Based on the TDTR-DAC technique, in 2019, this method was
628 extended to combine with a picosecond pulsed laser based transient thermorefectance
629 technique to measure the pressure-dependent cross-plane thermal conductivity of

630 MoS₂ up to 25 GPa.^{49,156} Moreover, an *in-situ* method was developed for measuring
631 the high-pressure high-temperature thermal diffusivities of Pt and Fe up to ~60 GPa
632 and 2000 K by combining the thermoreflectance measurements (albeit with ns laser
633 pulses) and laser heated diamond anvil cell techniques.¹⁵⁷

634 3.3. Calculation methods

635 Thermal transport properties of various materials can be theoretically computed
636 as a function of pressure and temperature using a variety of methods including
637 first-principles based phonon Boltzmann transport equation (BTE), density functional
638 theory (DFT) and MD simulations. For example, theoretical calculations of
639 pressure-dependences of lattice thermal conductivities have been performed in iron,
640 argon, diamond, silicon, cubic boron nitride, and so on.¹⁵⁸⁻¹⁶³

641 A combined DFT and Peierls-Boltzmann transport equation (PBTE-DFT)
642 method was developed and benchmarked against the measured pressure-dependent κ
643 values for a number of materials;¹⁶⁴ in this method, a general quantitative accuracy
644 has been demonstrated without adjustable empirical parameters. The PBTE-DFT
645 method has shown a capability to reliably predict κ of a variety of materials from
646 one-dimensional to bulk under conditions of varying temperature and pressure,
647 including the considerations of defects and higher order anharmonic process. Yet, the
648 PBTE-DFT method has the limitations related to neglecting the effects of temperature
649 oscillating or varying on the material surface with time, and also not accounting for
650 the variations in the phonon distributions near a surface or interface of a
651 material.^{164,165} These limitations make PBTE-DFT method questionable in
652 applications (e.g., predicting κ) of materials with extreme disorder or crystals with
653 strong anharmonicity.

654 MD simulations are another tool for calculating κ , particularly for applications
655 where PBTE-DFT calculations are not effective. To explore the evolution of κ under
656 pressure, an *ab initio* MD was combined with phonon normal mode analysis and the
657 PBTE to predict, for instance, κ of MgO at high temperatures and high pressures.¹⁶⁶
658 However, only small simulation domains can be handled by *ab initio* MD due to the
659 large computational expense; this makes it most suitable for small nano-structures or
660 bulk materials at high temperatures and high pressures. Another shortage is that the
661 empirical potentials MD simulations needed are often not available for materials of
662 interest. For thermal transport calculations, the potentials need to include essential
663 harmonic and anharmonic phonon properties at the temperatures and pressures of
664 interest, while most empirical potentials were not developed with these characteristics
665 yet.

666 Recently, a computationally efficient thermal snapshot interatomic force
667 constants (IFCs) fitting technique was developed to obtain the cubic and quartic IFCs
668 that used as important inputs for κ calculations.^{76,167,168} In this method, the harmonic
669 forces are subtracted out from the force-displacement dataset using the short and
670 long-range harmonic IFCs within the framework of density functional perturbation
671 theory implemented in the Quantum ESPRESSO, with only the anharmonic IFCs

672 fitted to the remaining forces. Then both harmonic (second-order) and anharmonic
673 (third and fourth-order) IFCs are used as inputs to determine phonon modes and
674 phonon-phonon scattering rates. The scattering rates are obtained by iteratively
675 solving the PBTE for phonon transport including three-phonon, four-phonon and
676 phonon-isotope scattering terms. The pressure in the calculations was obtained by
677 taking the derivative of the fourth-order anharmonic Helmholtz free energy with
678 respect to volume at each temperature.
679

680 4. Thermal properties of materials under pressure

681 4.1. Gases

682 Upon compression, gas can undergo gas-liquid-solid transformation. The
683 discussion in this section only refers to the gas naturally existed at ambient conditions.
684 In general, the thermal conductivities of gases under pressure are obviously smaller
685 than those of solids. Figure 4a and Table 1 summarize the κ of several typical gases
686 with respect to pressure. Around 1 GPa, most gases transform into liquids or solids,
687 accompanied by a rapid increase in κ . Limited by experimental techniques, the
688 majority of the thermal studies of gases in the last century was the pressure range
689 within 1 GPa.⁷⁷ For instance, κ measurements of hydrogen were carried out using a
690 coaxial cylindrical cell for pressure up to 66 MPa in 1966, and the effects of
691 temperature on κ of hydrogen under pressure were established.¹⁶⁹ Furthermore,
692 pressure-dependent κ parameters of oxygen, methane, nitrogen, neon, argon and other
693 gases, were also investigated mainly via the transient hot-wire method.¹⁷⁰⁻¹⁷⁴ Recently,
694 with the help of DAC and pulsed laser TH techniques, the pressure range of thermal
695 characterizations of gases has been extended over 10 GPa. In 2012, the κ of argon was
696 measured in pulsed laser transiently heated DAC at pressures up to 50 GPa and
697 temperatures up to 2500 K,¹⁷⁵ paving the way for thermal studies of gases at higher
698 pressures. For argon, some theoretical calculations were also performed on predicting
699 its κ at high pressures,^{176,177} which provide valuable guidance for experimental
700 studies.

701

702 **Table 1 | Summary of thermal conductivities for various materials from gases, liquids to**
703 **solids investigated under pressure.**

704

	Materials	Strain type	Methods	Maximum pressure	Thermal conductivity ($\text{Wm}^{-1}\text{K}^{-1}$)	Ref.
Gases	Ar	comp.	parallel plate method	240 MPa	0.018-0.107 (1 atm-196 MPa, 298 K)	178
	Ar	comp.	transient hot-wire method	11 MPa	0.018-0.021 (0.58-6.31 MPa, 308 K)	174
	Ar	comp.	transient heating	50 GPa	10.66-78.28 (10-50 GPa,	175

	technique			300 K)		
	Ar	comp.+ tension	MD+BTE calculations	6%comp.+6 %tension	0.1-10 (isotropic strains from -0.06 to 0.06, 20 K)	179
	Ar	comp.	DFT+BTE calculations	150 GPa	3.32-56.71 (10-90 GPa, 400 K)	177
	Ar	comp.	MD calculations	50 GPa	0.72-25.93 (0-50 GPa, 293 K)	176
	N ₂	comp.	transient hot-wire method	27.7 MPa	0.038-0.046 (1-27.7 MPa, 470 K)	180
	N ₂ +CO ₂	comp.	transient hot-wire method	30 MPa	change with proportion	180
	H ₂	comp.	transient short hot-wire method.	99 MPa	0.197-0.28 (0.26-97.76 MPa, 323 K)	181
	H ₂	comp.	-	66 MPa	0.19-0.24 (3.54-65.6 MPa, 298 K)	169
	Ne	comp.	parallel plate method	260 MPa	0.05-0.12 (1 atm-250 MPa, 298 K)	173
	Methane	comp.	transient hot-wire method	70 MPa	0.0337-0.1169 (0.99-67.3 MPa, 295 K)	171
	O ₂	comp.	transient hot-wire method	70 MPa	0.028-0.068 (2-65.3 MPa, 310 K)	170
Liquids	Toluene	comp.	ac-heated wire method	1000 MPa	0.128-0.25 (0-902.6 MPa, 300 K)	182
	H ₂ O	comp.	line heat source probe	700 MPa	0.65-0.82 (108-700 MPa, 298 K)	183
	H ₂ O	comp.	TDTR+DAC	22 GPa	0.54-24.69 (0-22 GPa, 300 K)	184
	H ₂ O	comp.	transient hot-wire method	0.8 GPa	0.56-1.54 (0-0.65 GPa, 273-243 K)	185
	Silicone oil	comp.	TDTR+DAC	23 GPa	0.16-1.53 (0-23 GPa, 300 K)	155
	4:1 Methanol-etha nol	comp.	TDTR+DAC	23 GPa	0.21-1.98 (0-23 GPa, 300 K)	155
Earth materials	Muscovite	comp.	TDTR+DAC	24 GPa	0.46-6.6 (0-24 GPa, 300 K)	39
	(Mg, Fe)O (Mg, Fe)SiO ₃	comp.	TH+DAC	133 GPa	-	134
	(Mg, Fe)O	comp.	TDTR+DAC	120 GPa	2.8-50 (0-120 GPa, 300 K)	186
	(Mg, Fe)SiO ₃	comp.	TDTR+DAC	120 GPa	5-30 (0-120 GPa, 300 K)	187
	(Mg, Fe)CO ₃	comp.	TDTR+DAC	67 GPa	2.5-45 (0-67 GPa, 300 K)	188

	MgO	comp.	TDTR+DAC	60 GPa	53-161 (0-60 GPa, 300 K)	189
	MgO	comp.	DFT+P-BTE calculations	150 GPa	66-341 (0-150 GPa, 300 K) 2-46 (0-150 GPa, 3000 K)	190
	MgSiO ₃	comp.	TDTR (two-side) +DAC	144 GPa	8-37.1 (11-144 GPa, 300 K)	191
	Pyrolite	comp.	TH+DAC	124 GPa	3.9 (0-80 GPa, 2000-2500 K) 5.9 (0-124 GPa, 2000-3000 K)	150
	Ringwoodite	comp.	TDTR+DAC	25 GPa	3-16 (0-25 GPa, 300 K)	192
	δ -(Al,Fe)OOH	comp.	TDTR+DAC	110 GPa	5-60 (0-110 GPa, 300 K)	193
	Iron	comp.	TH+DAC	130 GPa	20-40 (35-130 GPa, 2000-3000 K)	3
	Iron	comp.	TDTR+DAC	120 GPa	76-120 (0-120 GPa, 300 K)	60
	Fe _{0.96} Si _{0.04}	comp.	TDTR+DAC	125 GPa	16.5-60 (0-125 GPa, 300-3300 K)	60
	Fe _{0.85} Si _{0.15}	comp.	TDTR+DAC	144 GPa	11.5-40 (0-144 GPa, 300-3300 K)	60
	<i>hcp</i> Iron	comp.	TH+DAC	134 GPa	70-80 (constant above 46 GPa)	111
	Iron	comp.	DFT calculations	340 GPa	150-250 (120-340 GPa, 4500-6500 K)	2
Thermoelectric materials	Pb _{0.99} Cr _{0.01} Se	comp.	opto-thermal Raman	6 GPa	2.1-8.2 (0-6 GPa, 300 K)	22
	PdS	comp.	opto-thermal Raman	10 GPa	25-9 (0-10 GPa, 300 K)	139
	CuInTe ₂	comp.	opto-thermal Raman	8 GPa	2.1-8.2 (0-8 GPa, 300 K)	109
	CuInTe ₂	comp.	BTE calculations	5 GPa	7.5-4.1 (0-5 GPa, 300 K)	194
Electronic materials	MoS ₂	comp.	ps-TTR+DAC	19 GPa	3.5-25 (0-19 GPa, 300 K)	49
	Si	comp.	TDTR+DAC	45 GPa	73-300 (0-36 GPa)	195
	Si _{0.991} Ge _{0.009}	comp.	TDTR+DAC	45 GPa	24-360 (0-36 GPa)	195
	Si	torsion	TDTR+DAC	24 GPa	142-7.6 (0-24 GPa, 300 K)	196
	Si	comp.+ tension	MD+BTE calculations	3%comp.+3%tension	135-155 (isotropic strains from -0.03 to 0.03, 300 K)	179
	Si	comp.+ tension	MD calculations	4%comp.+4%tension	100-450 (Stillinger and Weber Si potentials, 300 K)	197
	Monolayer silicene	comp.	DFT+BTE calculations	10%	25-170 (0-10% strain, 300 K)	198
	Monolayer h-BAs	tension	MD calculations	3-7%	180.2-375 (3% strain along armchair direction)	199

					180.2-406.2 (3% strain along zigzag direction)	
	BAs	comp.	DFT+BTE calculations	80 GPa	1331-823 (0-80 GPa, 300 K)	76
	GaAs	comp.	DFT+BTE calculations	20 GPa	49-70 (0-16 GPa, 300 K)	108

* “-” in the table means that the original papers did not mention the item. The acronyms “comp.”, “MD”, “BTE”, “DFT”, “TH”, “TDTR”, “TTR”, and “DAC” represent “compression”, “molecular dynamics”, “Boltzmann transport equation”, “density-functional theory”, “pulsed laser transient heating”, “time-domain thermoreflectance”, “transient thermoreflectance”, and “diamond anvil cell”, respectively.

4.2. Liquids

Most studies on the κ parameters of liquids under pressure, on the one hand, were previously focused on organic liquids such as oil, ethanol, toluene to explore their implications on food and industrial products.^{182,183,200} On the other hand, interests on the pressure-dependent κ of pressure transmitting medium used in DAC were attracted,¹⁵⁵ such as the methanol-ethanol mixture and silicone oil, which can provide the fundamental parameters for the accurate study of thermal conductivity or interfacial thermal conductance of other materials under pressure. In addition, H₂O, as one of the most important substances in life, was also systematically studied on its κ under pressure.^{184,201} The evolution of the κ of the above-mentioned liquids as a function of pressure is plotted in Figure 4b and summarized in Table 1, showing typical changes of the orders of 0.1 to 10 Wm⁻¹K⁻¹ within the transformation range from liquids to solids.

4.3. Solids

Pressure dependence of the thermal transport in solids and amorphous materials has been studied for approximately 40 years. Here, we discuss the progress in pressure dependent thermal conductivities of Earth materials, thermoelectric materials, two-dimensional layered materials, semiconducting electronic materials, and the interfacial thermal conductance at solid-solid interfaces (Figure 4-6 and Table 1).

4.3.1. Earth materials

Understanding the heat transport and thermal evolution of Earth’s interior requires knowledge of κ of minerals and melts at relevant extreme temperatures and pressures. In Earth’s core, the thermal evolution and the energies driving the geomagnetic field are highly sensitive to the κ of core materials at high pressure and high temperature conditions, because these κ values will define the adiabatic heat flux, thermal and compositional energy that are available to support the operation of Earth’s magnetic field. Meanwhile, κ of the Earth’s lower mantle minerals greatly

impacts the mantle convection style and affects the heat conduction from the core to the mantle. Note that conduction is the major heat transfer mechanism at the core-mantle boundary, in which κ value of the bottom boundary layer of the mantle determines the magnitude of heat flux from the core. More importantly, this κ value is intimately related to the instability of the boundary layer, the formation of mantle plumes, the long-term thermal evolution of both mantle and core, and the driving force for the generation of the geomagnetic field.

Earth's core is composed of iron (Fe) alloyed with some light elements. A wide range of κ of Fe and its alloys have been theoretically predicted at core conditions. Experimentally, the κ of solid Fe which is about $18\text{--}44\text{ Wm}^{-1}\text{K}^{-1}$ at pressures (130 GPa) and temperatures (3000 K) of the core condition was directly measured, using a dynamically double-sided infrared continuous-wave lasers heated DAC with one-side additional thermal disturbance created by another infrared pulsed laser.^{3,135,202} Recently, an increased κ of γ -Fe ($90\text{--}125\text{ Wm}^{-1}\text{K}^{-1}$) and in a mixed phase of Fe ($45\text{--}65\text{ Wm}^{-1}\text{K}^{-1}$) with increasing pressure up to about 46 GPa was measured, through a single-sided laser heated DAC technique where the power absorbed by a Fe metal foil at hotspot was calculated from a thermodynamic simulation in the COMSOL software.¹¹¹ While κ of ε -Fe lies in the range of $70\text{--}80\text{ Wm}^{-1}\text{K}^{-1}$ at 1600–2100 K (with an error of $\sim 40\%$) and is almost independent of pressure up to 134 GPa; this independent κ of ε -Fe was attributed to the strong electronic correlation effects driven by the electronic topological transition.¹¹¹ In addition, the high pressure κ of solid Fe and Fe-Si alloys was measured at both room temperature and high temperature up to 144 GPa and 3300 K by the combination of TDTR³⁹ and TH method.^{3,38} It was found that at room temperature, κ of solid Fe increases from $\sim 76\text{ Wm}^{-1}\text{K}^{-1}$ to $\sim 90\text{ Wm}^{-1}\text{K}^{-1}$ at ~ 13 GPa and then decreases to a minimum of $\sim 40\text{ Wm}^{-1}\text{K}^{-1}$ at around 40 GPa, followed by a further increase again reaching $\sim 130\text{ Wm}^{-1}\text{K}^{-1}$ when the pressure is near the core-mantle boundary (CMB) condition (120 GPa); this minimum may be due to an electronic topological transition occurring at ~ 40 GPa. Addition of Si impurity significantly lowers κ of pure Fe, for example, $16.5\text{ Wm}^{-1}\text{K}^{-1}$ for $\text{Fe}_{0.96}\text{Si}_{0.04}$ and $11.5\text{ Wm}^{-1}\text{K}^{-1}$ for $\text{Fe}_{0.85}\text{Si}_{0.15}$, respectively, at ambient conditions. With increasing pressure, thermal conductivities of both $\text{Fe}_{0.96}\text{Si}_{0.04}$ and $\text{Fe}_{0.85}\text{Si}_{0.15}$ increase monotonically up to 40 GPa, and then saturate at $\sim 40\text{ Wm}^{-1}\text{K}^{-1}$ and $19\text{ Wm}^{-1}\text{K}^{-1}$, respectively (Figure 4c). However, at high temperature of ~ 3300 K, κ of $\text{Fe}_{0.85}\text{Si}_{0.15}$ firstly increases to $\sim 40\text{ Wm}^{-1}\text{K}^{-1}$ (80 GPa) and then decreases to $\sim 20\text{ Wm}^{-1}\text{K}^{-1}$ (144 GPa), while κ of $\text{Fe}_{0.96}\text{Si}_{0.04}$ increases to $\sim 60\text{ Wm}^{-1}\text{K}^{-1}$ (125 GPa, Figure 4d).⁶⁰ All of these κ values measured by experimental methods are several times lower than those obtained by calculation and electrical resistivity measurements.^{2,4,203–205}

In the earlier recognition, MgO is thought as a typical mineral of lower mantle materials. The κ of MgO has been calculated through a numerical technique combined with the PBTE-DFT method, providing a model for the P-T dependence of the κ of MgO at conditions from ambient to the CMB.^{190,206} It was found that κ increases from $15\text{--}20\text{ Wm}^{-1}\text{K}^{-1}$ at the pressure of the 670 km seismic discontinuity to $40\text{--}50\text{ Wm}^{-1}\text{K}^{-1}$ at the CMB pressure. Furthermore, it was found that at 2000 K, κ of MgO measured via the TH technique has about 50% enhancement when pressure increases

from ambient to 32 GPa; it was also revealed that the radiative part of κ of the lower mantle increases with depth until saturate to $\sim 0.54 \text{ Wm}^{-1}\text{K}^{-1}$ at the depth of CMB.^{38,134,207-210} However, at 300 K, through TDTR measurements in DAC, κ of MgO was found increasing from $50 \text{ Wm}^{-1}\text{K}^{-1}$ at ambient conditions to about $160 \text{ Wm}^{-1}\text{K}^{-1}$ at 60 GPa, agreeing well with the previous model and first-principles predictions.^{190,211,212} Concerning other Earth minerals, recent investigations on the pressure dependences of κ of other lower mantle minerals include $(\text{Mg,Fe})\text{O}$,¹⁸⁶ $(\text{Fe,Mg})\text{SiO}_3$,¹⁸⁷ $(\text{Fe}_{0.78}\text{Mg}_{0.22})\text{CO}_3$,¹⁸⁸ $\delta\text{-(Al,Fe)OOH}$,¹⁹³ $\text{Mg}_{0.94}\text{Fe}_{0.06}\text{SiO}_3$,²¹³ $(\text{Mg}_{0.9}\text{Fe}_{0.1})\text{O}$,²¹³ $\text{Mg}(\text{OH})_2$,²¹⁴ CaGeO_3 ,¹²⁷ $(\text{Mg}_{0.9}\text{Fe}_{0.1})_2\text{SiO}_4$,¹⁴ olivine $(\text{Mg}_{0.9}\text{Fe}_{0.1})_2\text{SiO}_4(\text{Fo}_{90})$,²¹⁵ ringwoodite,¹⁹² etc. Among them, thermal anomalies were found in an important water-carrying mineral $\delta\text{-(Al,Fe)OOH}$, of which the κ varies drastically by 2-3 fold when compressed across the spin transition of iron, thus leading to an abnormally low κ value at the lowermost mantle pressure.¹⁹³ Similar anomalous pressure dependence of κ was also observed in the mineral of $(\text{Fe}_{0.78}\text{Mg}_{0.22})\text{CO}_3$ when pressurized across the spin transition zone.¹⁸⁸

As an important mineral in Earth's lower mantle near the CMB, the κ of MgSiO_3 was found to be about $11 \text{ Wm}^{-1}\text{K}^{-1}$ at 144 GPa, increasing nearly 6 times than that at ambient condition, when measured by a modified TDTR technique in which the pump and probe lasers illuminate the opposite sides of the sample of the DAC.¹⁹¹ Through non-equilibrium MD simulations, the κ values of MgSiO_3 at 1000 K show enhancements from $8.5 \text{ Wm}^{-1}\text{K}^{-1}$ to $14\text{-}20 \text{ Wm}^{-1}\text{K}^{-1}$ in the range of 20-130 GPa with an obvious anisotropy. Note that these recently measured and calculated values are within the range of previous estimates of κ values at the CMB ($4\text{-}29 \text{ Wm}^{-1}\text{K}^{-1}$).^{216-218,150,186,187,191,219-224}

4.3.2. Thermoelectric materials

Thermoelectric (TE) materials with a high-efficient figure-of-merit (zT , which is normally expressed as $zT = \sigma S^2 T / \kappa$, where S is the Seebeck coefficient) are an ideal candidate to overcome the future energy crisis by transforming waste heat into power generation. The current bottleneck is that the zT is still too low to achieve a value of >3 for practical applications, according to the present modulation methods such as doping, alloying, temperature, and nanostructure engineering. Pressure as a fundamental thermodynamic variable can modify physical (such as crystal structure, electrical and magnetic) properties and chemical properties even at room temperature through the lattice compression, thus enables pressure-tuning as a promising and reliable approach to modulate the thermoelectric properties, including electrical conductivity and thermal conductivity, ultimately achieving high zT values.

Through the aforementioned DAC combined Raman opto-thermal method, the pressure dependence of κ was measured in a strained TE material of CuInTe_2 ,¹⁰⁹ and it is found that the κ significantly reduces from $11.7 \text{ Wm}^{-1}\text{K}^{-1}$ to a minimum of $4.1 \text{ Wm}^{-1}\text{K}^{-1}$ at 2.2 GPa. When the pressure increases to above 6 GPa, where the material undergoes a transition into a new phase, the κ jumps to $\sim 30 \text{ Wm}^{-1}\text{K}^{-1}$ and keeps increasing to about $40 \text{ Wm}^{-1}\text{K}^{-1}$ until 7.9 GPa where all the Raman modes almost

vanish (Figure 4e). Pressure-induced phonon anharmonicity and phonon softening are thought as the main mechanism for such a reduction of κ under pressure. It is interesting that the κ of another TE material PbS continuously decreases from about 25 $\text{Wm}^{-1}\text{K}^{-1}$ at ambient to about 9 $\text{Wm}^{-1}\text{K}^{-1}$ at near 4 GPa and then saturates until about 11 GPa.¹³⁹ A continuous decrease of κ is also witnessed in a half-Heusler TE material FeNbSb, which decreases from $\sim 5 \text{ Wm}^{-1}\text{K}^{-1}$ at ambient to $\sim 2 \text{ Wm}^{-1}\text{K}^{-1}$ at the pressure of 18 GPa.¹³⁸ Recently, a significantly enhanced room temperature zT value of ~ 1.7 (~ 4 times enhancement) was reported at about 2.8 GPa in a Cr-doped PbSe within the DAC, and this enhancement was ascribed to a pressure driven topological phase transition at 2.6 GPa.²² In this work, via a similar Raman opto-thermal method, an increase of κ from $\sim 2 \text{ Wm}^{-1}\text{K}^{-1}$ at ambient condition to $\sim 6 \text{ Wm}^{-1}\text{K}^{-1}$ at 4.8 GPa was observed, followed by an immediate jump to a larger value of $\sim 8 \text{ Wm}^{-1}\text{K}^{-1}$ due to the transition into a *Pnma* phase (Figure 4e).

Many new results have been also achieved through first-principles calculations, for instance, the strain engineering effect on the κ of the commercial low-temperature TE material Bi_2Te_3 was simulated with a 50% reduction shown in κ when under a 6% tensile strain, while a 61% increment when under a 4% compressive strain.²²⁵ Via similar calculation methods, the κ of skutterudites CoSb_3 and IrSb_3 (medium-temperature TE materials) both exhibit an approximate parabolic trend with respect to pressure at the same temperature.²²⁶ CoSb_3 has an increased κ from $\sim 11.3 \text{ Wm}^{-1}\text{K}^{-1}$ at ambient condition to $\sim 13.3 \text{ Wm}^{-1}\text{K}^{-1}$ at the maximum zT value's ($zT=1.09$) pressure of 58 GPa, despite the largest κ of $\sim 18.6 \text{ Wm}^{-1}\text{K}^{-1}$ appearing at ~ 30 GPa. Meanwhile, IrSb_3 shows a decreased κ from $\sim 14.2 \text{ Wm}^{-1}\text{K}^{-1}$ at ambient condition to $\sim 8.8 \text{ Wm}^{-1}\text{K}^{-1}$ at its maximum zT value's ($zT=1.4$) pressure of 54 GPa, with its largest value of $\sim 25.4 \text{ Wm}^{-1}\text{K}^{-1}$ turning up at ~ 20 GPa.

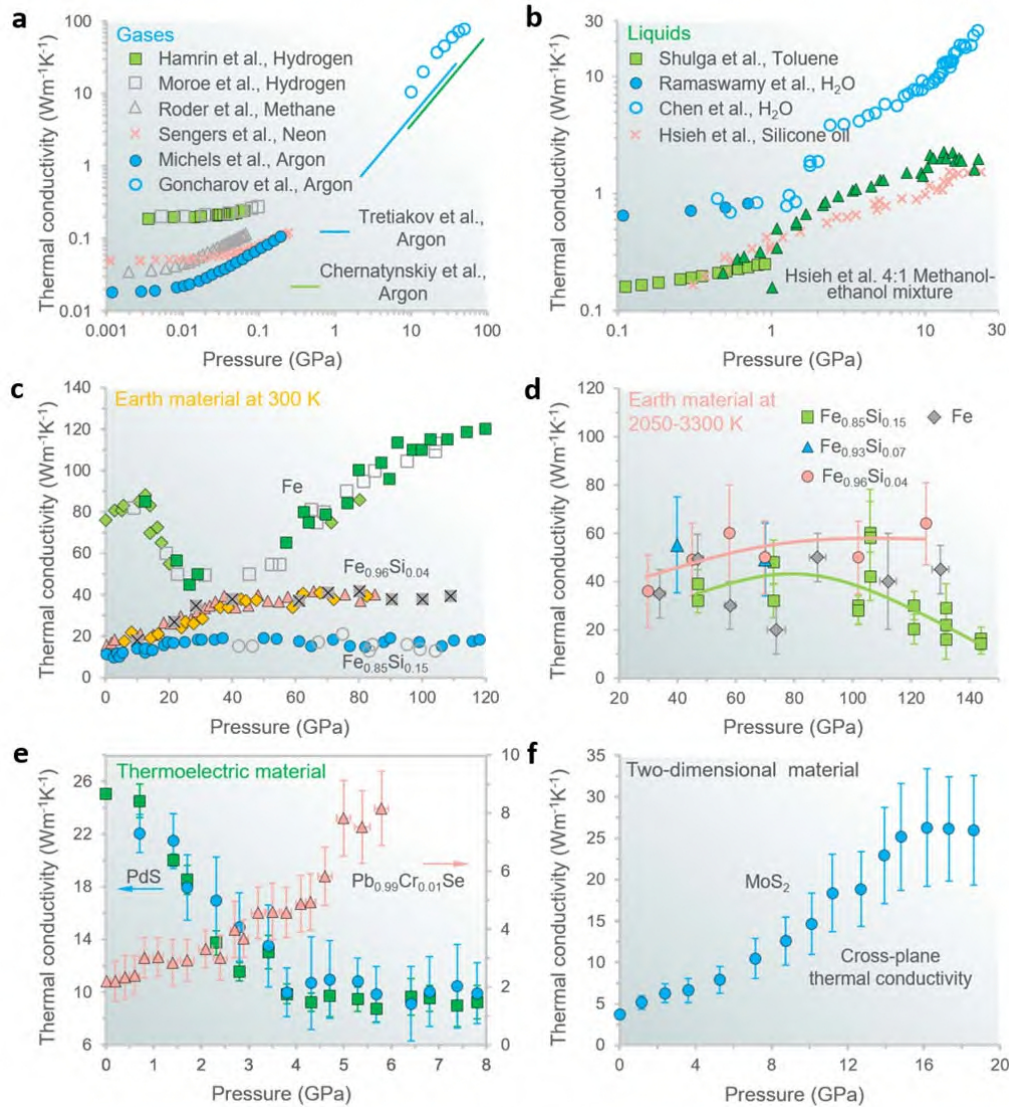


Figure 4 | Thermal conductivities of gases, liquids, Earth, thermoelectric and two-dimensional materials under pressure. **a** | Pressure dependence of the thermal conductivity of hydrogen, neon, argon, and methane gases. The values are taken from Refs. ^{169,171,173,175-178,181}. **b** | Pressure dependence of the thermal conductivity for liquids of H_2O , silicone oil, methanol-ethanol mixture, and toluene. The values are taken from Refs. ^{155,182-184}. **c** | Thermal conductivities of Earth materials as a function of pressure for iron and iron-silicon alloys up to 120 GPa at 300 K. **d** | Thermal conductivities of Earth materials as a function of pressure for iron and iron-silicon alloys up to 144 GPa and 3300 K. The values are taken from Refs. ^{3,60}. **e** | Thermal conductivities of thermoelectric materials with respect to pressure for $\text{Pb}_{0.99}\text{Cr}_{0.01}\text{Se}$ and PdS at 300K. The values are taken from Ref. ^{22,139}. **f** | Pressure dependence of the thermal conductivity for two-dimensional material MoS_2 . The values are taken from Ref. ⁴⁹. Lines in panels **a** and **d** are simulation results. Panels **c-f** are adapted with permission from Refs. ^{3,22,49,60,139}. ©Springer Nature Publishers Limited, ©APS Publishers Limited. ©Elsevier Publishers Limited.

862 4.3.3. Two-dimensional layered materials

863 Extreme strain has a very sensitive and profound impact on phonon transport
864 properties that are correlated to the κ modulation of two-dimensional layered
865 materials. These materials normally possess highly anisotropic thermal conductivities
866 along the in-plane and cross-plane directions due to their weak interlayer bonding by
867 van der Waals force. For example, the reported in-plane thermal conductivity ($\kappa_{\parallel}$,
868 ranging from 35-85 $\text{Wm}^{-1}\text{K}^{-1}$ at ambient condition) of MoS_2 ,²²⁷⁻²²⁹ an archetypal
869 transition-metal-dichalcogenide (TMDs) with a layered crystal structure, is more than
870 10 times larger than the cross-plane thermal conductivity ($\kappa_{\perp}$, about 2–4.5
871 $\text{Wm}^{-1}\text{K}^{-1}$).^{227,230,231} Small $\kappa_{\perp}$ could jeopardize heat dissipation of TMDs-based
872 electronics, while strain engineering or high pressure techniques are promising to
873 enhance and modulate the $\kappa_{\perp}$, which can also revolutionize the tunability and thermal
874 management engineering techniques in all TMDs-based electronic devices. Despite
875 extensive theoretical studies of strain's effect on κ in TMDs, no consistent
876 conclusions have been drawn.^{72,78,232,233} The pressure dependence of κ of bulk MoS_2
877 up to 25 GPa was experimentally obtained for the first time in 2019.⁴⁹ An increase of
878 $\kappa_{\perp}$ from 3.5 $\text{Wm}^{-1}\text{K}^{-1}$ at ambient condition to about 25 $\text{Wm}^{-1}\text{K}^{-1}$ at ~25 GPa was
879 observed using a picosecond transient thermoreflectance technique, as shown in
880 Figure 4f. Combined with coherent phonon spectroscopy and first-principles
881 calculations, it was further revealed that the drastic change in $\kappa_{\perp}$ arises from the
882 strain enhanced interlayer interaction, heavily modifying the phonon dispersion
883 curves, and decreasing the phonon lifetime. In general, this enhancement in $\kappa_{\perp}$ is
884 mainly attributed to the unbundling effect along the cross-plane direction, while the
885 change of electronic thermal conductivity under pressure has negligible contribution.
886 The strain effects on many other two-dimensional materials are still experimentally
887 unexplored.

888 An abnormal strain-dependence is recently shown in a novel hexagonal phase
889 monolayer BAs. Through DFT+BTE calculations, the $\kappa_{\parallel}$ was predicted to be
890 enhanced to 375.0 $\text{Wm}^{-1}\text{K}^{-1}$ and 406.2 $\text{Wm}^{-1}\text{K}^{-1}$ from 180.2 $\text{Wm}^{-1}\text{K}^{-1}$ along the
891 armchair and zigzag directions, respectively, under only 3% stretching.¹⁹⁹ This
892 enhancement is correlated to the fact that stretching makes the flexural out-of-plane
893 mode being the dominant heat carrier.

894 4.3.4. Semiconductor electronic materials

895 High κ of semiconducting electronic materials is the key for efficient device
896 thermal dissipation and thermal management. Many methods have been tried to
897 improve their κ values, however applications of pressure remains underexplored even
898 though the available in the literature data suggest enhanced materials thermal
899 performance. Up to date, the majority of works on the pressure dependence of κ in
900 semiconducting electronic materials are performed by first-principles calculations. By
901 means of separate calculations of the harmonic and anharmonic effects of strain on

materials stiffness and phonon properties, the κ of the widely used Si semiconductor is found constant within a 3% compression (equivalent to ~ 4 GPa) and only has about 10% decrease when it is under a 3% tension; this was mainly ascribed to the anomalous behavior of its phonon lifetime which increases with the strain when moving from compression to tension and allows greater root-mean-square displacement under compression. Experimentally, the κ of Si as a function of pressure up to 45 GPa was measured via TDTR (see Figure 5a). In contrast to the above predictions, a κ of 115-130 $\text{Wm}^{-1}\text{K}^{-1}$ within the entire diamond cubic phase range of 0-13 GPa was observed in Si without obvious pressure dependence, while a continuously increased κ to $\sim 300 \text{ Wm}^{-1}\text{K}^{-1}$ was witnessed within the metallic phase in the range of 16-36 GPa.¹⁹⁵ Meanwhile, the pressure dependence of κ of $\text{Si}_{0.991}\text{Ge}_{0.009}$ was also measured revealing an increasing trend in both pressure regions of the semiconducting phase and the metallic phase, but accompanied with a jump between the diamond cubic phase and the primitive hexagonal phase at ~ 13 GPa as well as a sharp drop between the primitive hexagonal phase and the intermediate *Cmca/hcp* phase at ~ 36 GPa.¹⁹⁵ Moreover, via the combined density functional perturbation theory with the phonon Boltzmann equation, a dramatically increased κ of $\sim 12000 \text{ Wm}^{-1}\text{K}^{-1}$ for the ultra-wide bandgap natural diamond ($\sim 17000 \text{ Wm}^{-1}\text{K}^{-1}$ for isotope-pure diamond, Figure 5b) at room temperature and 400 GPa was predicted; the overall increased frequency scale with pressure that results in the higher acoustic velocities and lower phonon-phonon scattering rates is regarded as the primary mechanism for this enhancement, *i.e.*, an increase in the optical mode frequencies with pressure weakens the acoustic-optical coupling thus driving the κ of diamond to far higher values (Figure 5b).²³⁴

Interestingly, many anomalous pressure dependences of κ were reported in some binary compound semiconductor materials especially those of large mass ratio. For example, the pressure dependences of κ of several binary compounds with a set of widely varying mass ratio are examined using a first-principles approach, *i.e.*, GaAs, SiC, BN, BP, BSb, BAs, BeTe and BeSe.⁷⁴ The calculations revealed that those compounds with similar mass ratio, such as GaAs, SiC, BN and BP, show an increased κ with pressure, while compounds with large mass ratio (e.g., BSb, BAs, BeTe, BeSe) that have significant frequency gaps between the acoustic and optical phonons exhibit a decreasing κ with pressure (Figure 5c). These anomalously decreased pressure dependence were found arising from the fundamentally different nature of the intrinsic scattering processes for heat-carrying acoustic phonons, *i.e.*, the common three-phonon scattering process that involve two acoustic phonons and one optical phonon does not occur due to the large acoustic-optical frequency gap; instead, the scattering between acoustic phonons dominate at high pressures thus limiting κ . Some more interesting calculations were recently reported on BAs, for example, a different non-monotonic pressure dependence of κ was calculated and shown to firstly increase as most material and then decrease at high temperatures. This was further revealed due to the competing responses of three-phonon and four-phonon interactions to pressure.⁷⁶ Based on the first-principles calculations combined with phonon Boltzmann transport equation,¹¹³ the κ of wide-bandgap bulk GaN in the

normal-pressure range of wurtzite phase and high-pressure rocksalt phase (Figure 5d) was calculated, showing that at the wurtzite-to-rocksalt phase-transition pressure, the κ of GaN has a sharp reduction of 91%, then increases almost linearly within the rocksalt phase range until 68 GPa; in addition, the κ of wurtzite GaN exhibits a nonmonotonic dependence on pressure with an $\sim 66\%$ increase first up to ~ 20 GPa then a slow decrease until the phase transition pressure.¹¹² According to phonon mode analysis, the sharply reduced κ of the rocksalt phase was ascribed to the enhanced lattice anharmonicity, while the nonmonotonic pressure-dependence of κ in wurtzite GaN was attributed to the interplay between group velocity and phonon relaxation time.¹¹²

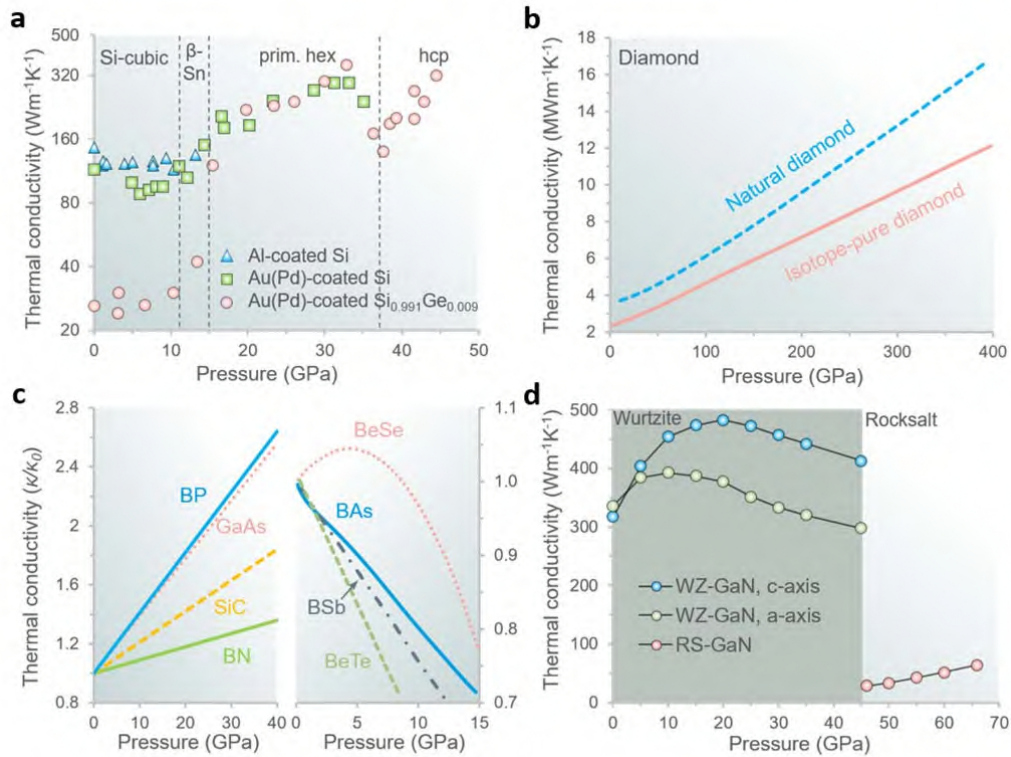


Figure 5 | Thermal conductivities of semiconducting electronic materials under pressure. The thermal conductivity as a function of pressure for **a** | electronic materials of Si and Si_{0.991}Ge_{0.009} measured using TDTR near room temperature within the range of 0-45 GPa. The predicted pressure dependence of the thermal conductivity from first-principles calculations for **b** | the natural and isotope-purified ultra-wide bandgap diamond at 300 K up to 400 GPa, **c** | binary compound semiconductors of GaAs, SiC, BP and BN with an increasing trend, and BAs, BeTe and BSb with a decreasing trend, and BeSe with a non-monotonic trend. **d** | Predicted nonmonotonic pressure dependence of the thermal conductivity for wide bandgap semiconductors of GaN across their phase transition pressures. Panels **a-d** are adapted with permission from Refs.^{74,112,195,234}, respectively. ©APS Publishers Limited.

969 When heat transfers across an interface, an unavoidable temperature drop exists
 970 at the interface, making the original interfacial thermal conductance G (or inversely,
 971 thermal boundary resistance, TBR) critical in controlling heat conduction through
 972 interfaces, especially for the cases of solid-solid interfaces that are widely involved in
 973 nanostructures, nanoscale composites, and superlattices.^{64,235-239} For instance, the
 974 effective thermal conductivity of materials can be reduced to be lower than that of the
 975 amorphous limit by introducing a high density of interfaces through nanostructure
 976 engineering; in this way, the thermoelectric energy conversion can be improved.^{95,151}
 977 According to the present theoretical understanding, the more dissimilar the densities
 978 of phonon vibrational states of the two materials of the interface, the lower the G ;
 979 however, a variety of anomalous interfacial thermal transport behaviors contradicting
 980 the theoretical predictions were experimentally observed especially in examples of
 981 metal/diamond interfaces.¹⁰² The effect of interfacial bonding on interfacial thermal
 982 transport has been extensively discussed. Lattice-dynamic theory illustrates that weak
 983 interfacial bonding can significantly lower G .²⁴⁰ The MD simulations also showed
 984 that the low interfacial stiffness can suppress G .^{241,242} In the case of extremely weak
 985 interfacial bonding, G scales with the square of the interfacial force constant, while in
 986 the strong interfacial stiffness condition, G saturates at the AMM predicted value.

987 The interfacial stiffness can be modulated by an applied stress through a
 988 generated discontinuity in displacement. The applied pressure can enable continuous
 989 tuning of materials lattice dynamics through varying the anharmonicity of atomic or
 990 molecular bonds,⁷⁷ and allow the direct observation of the role of the interfacial
 991 bonding on the suppression of G (Figure 6a-d). For example, on the ITC comparison
 992 of three types of interfaces loaded in DAC with the interfacial bonding from strong to
 993 weak, *i.e.*, the pressure dependent G values of clean Al/SiC, Al/SiO_x/SiC and
 994 Al/graphene/SiO_x/SiC interfaces measured by the TDTR method, it was found that G
 995 of the strong Al/SiC interface (~ 200 MW/m²K) weakly depends on the pressure,
 996 while G of the weak Al/SiO_x/SiC and Al/graphene/SiO_x/SiC interfaces (~ 30 MW/m²K
 997 at ambient, Figure 6c) increases rapidly with pressure and approaches the similar
 998 value of the strong Al/SiC interface at >8 GPa. These results demonstrate that the
 999 interfacial stiffness dominates the thermal transport at weak interfaces but plays a
 1000 minor role for strong interfaces.¹¹⁴ The alloying of semiconductors will also affect the
 1001 behaviors of G at their interfaces, for instance, on the examples of the
 1002 pressure-dependent G between the Al and Au(Pd) transducers and the Si and Si(Ge)
 1003 substrates measured by the regular and beam-offset TDTR up to 45 GPa, a nearly 50%
 1004 enhanced G was observed at the Al/Si interface with a 2.2 nm native oxide layer
 1005 which increases from 260 to 380 MW/m²K between 0 and 10 GPa.¹⁹⁵ It was also
 1006 found that the difference in the initial G between Au(Pd)/Si and Au(Pd)/Si(Ge) due to
 1007 the different interfacial bonding is suppressed at 10 GPa. After crossing the
 1008 semiconductor-metal phase transition at ~ 15 GPa, G of Au(Pd)/Si and Au(Pd)/Si(Ge)
 1009 increases to a discontinued higher average values of 750 and 470 MW/m²K with weak
 1010 pressure dependence, respectively (Figure 6d).¹⁹⁵ For much strongly bonded

interfaces, G of Al/MgO interface increasing from ~ 0.5 to $1.1 \text{ GW/m}^2\text{K}$ and G of SrRuO₃/SrTiO₃ interface increasing to $\sim 0.8 \text{ GW/m}^2\text{K}$ within the 0-60 GPa range were observed using a similar experimental method (Figure 6b).²⁴³ All these comparisons under pressure indicate that interfacial stiffness is an influential parameter for G , even when the interface is relatively clean and strongly bonded.

Many metal-diamond interfaces show a G anomalously higher (such as Au and Pb, 5-10 times) than those values predicted by the elastic phonon radiation limit model;^{102,105} this means significant inelastic phonon scattering processes are probably involved. To enable the study of inelastic processes and understand the hidden mechanism, it would be beneficial to investigate an extreme situation of weakly bonded metal-diamond interfaces. Also, applying high pressure can extend the metal phonon density of states to significant higher frequencies and stiffen the interfacial bonding greatly to suppress the extrinsic effects. By applying pressure in a DAC up to 50 GPa via TDTR to measure G of metal-diamond interfaces for Pb, Au(Pd), Pt and Al films deposited on diamond anvils, it was observed that in all cases G increases weakly or saturates to similar values at high pressures (Figure 6a-b).⁷⁹ To understand this tendency, it was proposed that the anharmonic conductance at metal-diamond interfaces is controlled by partial transmission processes, which means that a diamond phonon inelastically scatters at the interface and absorbs or emits a metal phonon.⁷⁹

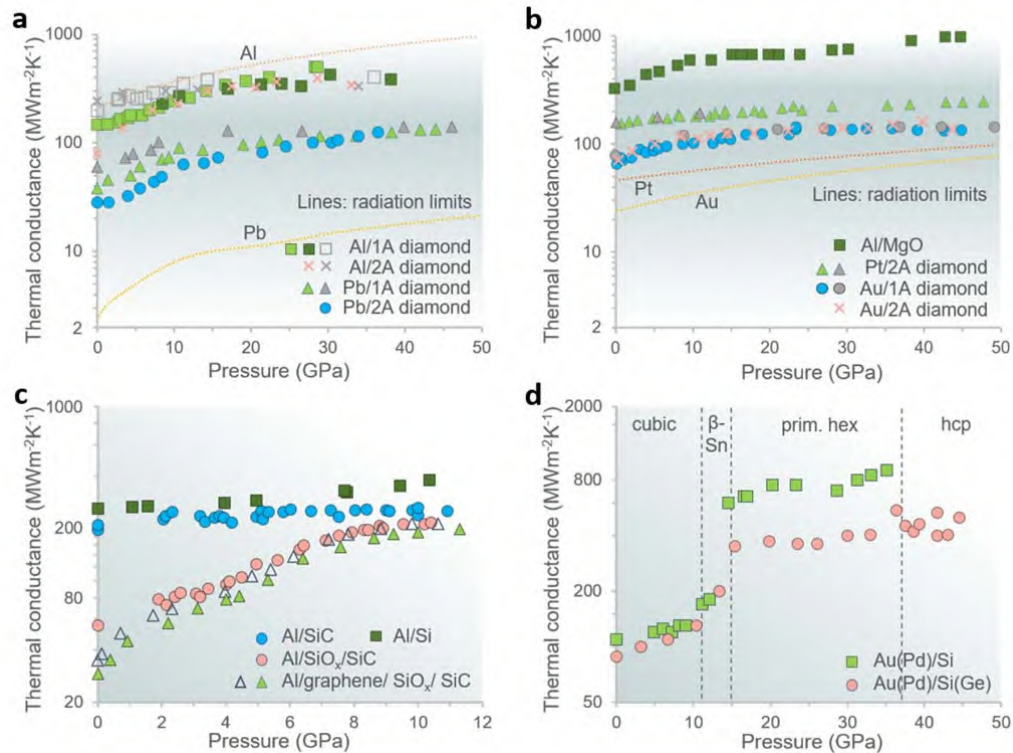


Figure 6 | Pressure dependence of the interfacial thermal conductance at solid-solid interfaces. A various range of interfacial thermal conductance with different interfacial stiffness at solid-solid interfaces were measured under pressure at examples of diamond based **a** | Al/diamond and Pb/diamond interfaces, and **b** | Au/diamond and Pt/diamond interfaces up to 50 GPa (lines are

1036 predictions by the elastic phonon radiation limit model), as well as at examples of Al film
 1037 transducer based strongly bonded Al/MgO interfaces up to 45 GPa, and **c** | strongly bonded Al/Si
 1038 and Al/SiC interfaces, and weakly bonded Al/SiO_x and Al/graphene interfaces up to 12 GPa. The
 1039 interfacial thermal conductance measured under pressure at examples of Au(Pd) film transducer
 1040 based **d** | weakly bonded Au(Pd)/Si and Au(Pd)/Si(Ge) interfaces across the phase transition
 1041 pressures up to 45 GPa. All data were measured at room temperature. Panels **a-d** are adapted with
 1042 permission from Refs.^{79,114,195,243}, respectively. ©Springer Nature Publishers Limited. ©APS
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 1044

1045 **5. Current applications**

1046 **5.1 High-pressure high-temperature DAC simulations of materials in Earth's** 1047 **interior**

1048 Earth's core is under extreme conditions of high temperature about 6000 K and
 1049 high pressure over 360 GPa at its center. Laser-heated DAC techniques provide the
 1050 necessary and effective approaches for the Earth's interior exploration and simulation,
 1051 by providing the relevant pressure-temperature conditions. A variety of coupled to
 1052 DAC *in situ* probes such as XRD, X-ray absorption and Raman spectroscopy, TH, and
 1053 TDTR²⁴⁴⁻²⁴⁶ enable multiple experiments on the geoscience materials, such as the core
 1054 and the mantle minerals, to study their physical and chemical properties, whereby
 1055 contributing to understanding of the Earth's interior structure, composition, and
 1056 evolution. Heat transport is crucial to clarify the thermal evolution and dynamics of
 1057 the Earth's interior. A wealth of exploration works have been reported on the κ
 1058 investigations of Earth's core, mantle, and core-mantle boundary materials through
 1059 DAC experiments obtained at the corresponding high-pressure high-temperature
 1060 conditions of Earth and first-principles calculations.^{2,3,149,150,203,205,213} It is found that
 1061 the adiabatic heat flux is 8-16 terawatts at the core-mantle boundary by computing
 1062 the thermal and electrical conductivity in liquid mixtures at Earth's core conditions
 1063 using first-principles calculations, higher than the previous estimates which are
 1064 obtained based on the mantle convection.^{2,247,248} To experimentally resolve this puzzle,
 1065 the κ of iron-silicon alloys at Earth's core conditions was measured to be as low as 20
 1066 Wm⁻¹K⁻¹ through TDTR and TH experiments in DAC, suggesting a lower minimum
 1067 heat flow of about 3 terawatts actually across the core-mantle boundary than
 1068 previously expected.⁶⁰ In addition, the κ of solid iron at the Earth's core conditions via
 1069 laser-heated DAC was measured within 18-44 Wm⁻¹K⁻¹; these values are near the low
 1070 end of previous estimates and in agreement with palaeomagnetic measurements,
 1071 indicating that the solid inner core has persisted since the beginning of Earth's history
 1072 and as old as the dynamo.³ Moreover, the κ of mantle minerals measured near the
 1073 environment of lowermost mantle in DAC up to 120 GPa and 2500 K, revealed a
 1074 value of ~3.9 Wm⁻¹K⁻¹ at 80 GPa and 2000-2500 K while ~5.9 Wm⁻¹K⁻¹ at 124 GPa
 1075 and 2000-3000 K.¹⁵⁰ Therefore, these results further indicate that high-pressure

high-temperature experiments and simulations of Earth materials are important in the exploration of Earth's geodynamo and the prediction of Earth's evolution.

5.2 Strain modulation of thermoelectric performance

Thermoelectric materials not only can directly convert heat into electricity to fabricate generators according to Seebeck effect but also can utilize electricity for cooling to produce refrigerators based on Peltier effect. Since the discovery of the Seebeck effect in 1821, thermoelectric materials have witnessed hundreds of year's development and unveiled many high-performance systems such as bismuth telluride, lead chalcogenides, silicon-germanium, skutterudites, Zintl phases, and half-Heusler compounds.^{67,249} Despite of these achievements, current thermoelectric materials are still too inefficient to achieve the goal of low-cost and large-scale applications in energy conversion. Thermoelectric performance, which can be assessed by the aforementioned zT formula, is overall determined by Seebeck coefficient, electrical conductivity, and thermal conductivity of materials. However, these physical quantities are mutually related, making them difficult to be modulated individually. To enhance the zT value, many approaches were adopted and some remarkable results were also obtained, such as band engineering,²⁵⁰ nanostructuring strategies,²⁵¹ and strain modulation.^{22,109} Particularly, strain modulation via pressure is found as a promising tool to improve the thermoelectric performance, which was recently confirmed by continued breakthroughs in examples of bismuth telluride,²⁵² lead chalcogenides,^{22,253} ternary chalcopyrite,¹⁰⁹ half-Heusler compounds,¹³⁸ and so on. There are also many theoretical studies starting to focus on the enhancement of thermoelectric performance by pressure, which detailly illustrate and predict the evolution of the aforementioned three key parameters in thermoelectrics with respect to pressure.²⁵⁴⁻²⁵⁶ In spite of these exciting efforts, there still lacks a comprehensive understanding and a universal mechanism on how the thermoelectric performance can be enhanced or modulated by pressure. One recent important finding is the pressure-induced electronic topological phase transition (TPT) interplayed with the closing and reopening of the band gap. The TPT is always accompanied by the anomalies of Seebeck coefficient and giant enhancement of electrical conductivity, resulting in the improvement of thermoelectric performance. This mechanism has been discovered well applied in bismuth telluride, lead selenide, and so on.^{22,257,258} However, some materials like PdS, FeNbSb, CuInTe₂ are found that without undergoing such TPT, their thermoelectric performances were still improved under pressure; these phenomena were ascribed to the optimization of the carrier concentration and the band structure. In addition, the low-frequency Raman modes are found to be softened as the pressures are increased in PdS and CuInTe₂; this phonon softening can be correlated to a decrease in thermal conductivity.^{109,138,139} Generally speaking, high-pressure technique has demonstrated its giant potential as a novel clean and effective strategy to optimize the thermoelectric performance.

1116 6. Concluding remarks and outlook

1117 Progressive achievements in high-pressure κ characterization techniques have
1118 allowed the in-depth understanding of heat transport, and uncovered many abnormal
1119 thermal phenomena and mechanisms that are hard to be probed at ambient conditions,
1120 as well as realized the reversible modulations of thermal conductivities of various
1121 materials. Guided by thermal transport theory, first-principles aided calculations have
1122 also predicted many unusual pressure-dependent κ behaviors that challenge the
1123 conventional thermal transport principles and provide promising chances for their
1124 practical thermal applications, however, these remain to be experimentally examined
1125 and physically illustrated.

1126 Up to date, one of the greatest challenges for the κ characterization under
1127 pressure is that the size of samples loaded in the high-pressure apparatus is normally
1128 too small, making this the main difficulty faced in the measurements based on the
1129 present available methods. For example, in contact method using thermal couple and
1130 heater, the sample size needs to be larger than the thermal couple and the heater size,
1131 pushing the limited space within the DAC very challenging for this layout. Similar in
1132 the contactless method via optical approach, the sample size needs to be larger than
1133 the spot radius of the pump laser used in TDTR and TH methods, and the spot radius
1134 of the excitation laser used in Raman optothermal method; besides, the optical method
1135 further requires the samples' surfaces to be as smooth as possible, otherwise the
1136 optical signal would be seriously degraded by scattering. Another challenge is the
1137 measurement of in-plane thermal conductivity under pressure, or the anisotropic
1138 evolution of κ under pressure. Despite many reports to study the pressure-dependent κ ,
1139 the outcomes are either isotropic or cross-plane thermal conductivity, while the
1140 interesting pressure-dependent in-plane thermal conductivity of abundant anisotropic
1141 materials remains unsolved. The in-plane thermal conductivity has broad theoretical
1142 utilities and practical applications in many anisotropic materials such as
1143 two-dimension materials and ultrathin films that are urgently needed for lateral heat
1144 spreading in electronics. Significant interests will be further raised if in-plane thermal
1145 conductivity modulation can be definitively realized and theoretically understood
1146 through methods of the strain modulation.

1147 Furthermore, the κ modulation via strain method may steer new directions in
1148 development of electronics and devices for thermal management. Especially but not
1149 limited in electronic systems, intensity of heat density generation and the associated
1150 heat dissipation need to be fully understood and carefully conducted away, or the
1151 electronics will face significant temperature rise, likely leading to reliability
1152 challenges and possible failure. Take an example of the transistors, which is in the
1153 heart of the electronics industry. The tendency described by the Moore's law is related
1154 to successive shrinking of the device size to integrate a greater number of transistors
1155 to a more compacted chip, thus leading to dramatic power consumption increase in
1156 the integrated circuits; this brings a huge challenge in heat dissipation to ensure the
1157 device high performance and reliability. For instance, the heat flux generated at the
1158 local hotspot due to self-heating during the operation can exceed 1000 W cm^{-2} in high

1159 electron mobility transistors; this is a big threat to the device reliability and lifetime.
1160 From the view of thermal management, the total thermal resistance of devices needs
1161 to be greatly minimized. This consists of contribution from the κ and thickness of
1162 materials as well as the major hinder of ITC. It should be noted that in high power
1163 transistors, we expect larger κ , while in thermoelectrics lower κ values are desired.
1164 These contradictory details should be taken into account in thermal management
1165 design and this reminds us about the necessity to make careful considerations of the
1166 device purposes and requirements. In any case, both types of device applications
1167 would benefit from higher ITC to fast and efficiently conduct the heat across the
1168 interfaces. Presently, enhancing κ and ITC of materials are achieved mainly by
1169 improving the single-crystal materials quality with fewer defects, increasing the grain
1170 size of polycrystalline materials to reduce the grain boundary scattering, introducing
1171 high thermal conducting interfacial materials or integrating with diamond like high κ
1172 materials. Besides of the above ambient methods, strain or pressure can also be used
1173 to tune the κ of materials and the ITC at interfaces, despite this part of research is still
1174 in its infant stage due to the challenges of high-pressure thermal characterization
1175 techniques; however, this method may inform novel thermal management techniques
1176 in electronic devices.

1177 In this Review, we have provided a systematic discussion of the progress related
1178 to the thermal properties' evolution under pressure, summarizing the behaviors of a
1179 variety of materials from gases, liquids to solids and solid-solid interfaces that have
1180 been investigated, as well as the thermal characterization methodological
1181 developments and some interesting theoretical predictions. Exciting application
1182 examples of thermo-physical properties characterization of minerals at high pressure
1183 and high temperature that are important in modelling thermal evolution of the deep
1184 Earth's mantle/core conditions, and thermal conductivity modulations in
1185 thermoelectrics by pressure are also introduced. Knowledge about the pressure
1186 dependence of thermal conductivity has opened new physical insights into the nature
1187 of phonon scattering and thermal transport from a new dimension, and extended the
1188 thermal modulation ability, especially when in combination with the temperature.
1189 Furthermore, a promising outlook in theoretical and experimental investigations of
1190 various high-pressure thermal properties and related thermal management techniques
1191 is also discussed. An additional viewpoint, which is worth to note, is that improving
1192 the reliability and precision in determining the pressure dependent thermo-physical
1193 properties require further implementation of advanced characterization techniques.
1194 Presently, defying the small size and very narrow space of diamond anvil cell tools,
1195 all-optical laser thermorefectance, transient heating, and opto-thermal fast contactless
1196 experimental techniques have attracted more and more interests, and displayed great
1197 potential in future developments, although there are still lots of unknowns need to be
1198 explored and improved in these methods.

1199 **References**

1200 1. Ohta, K. *et al.* The electrical conductivity of post-perovskite in Earth's D'' layer. *Science* **320**,

- 1201 89-91 (2008).
- 1202 2. Pozzo, M., Davies, C., Gubbins, D. & Alfè, D. Thermal and electrical conductivity of iron at
1203 Earth's core conditions. *Nature* **485**, 355-358 (2012).
- 1204 3. Konôpková, Z., McWilliams, R. S., Gómez-Pérez, N. & Goncharov, A. F. Direct measurement
1205 of thermal conductivity in solid iron at planetary core conditions. *Nature* **534**, 99-101
1206 (2016).
- 1207 4. Ohta, K., Kuwayama, Y., Hirose, K., Shimizu, K. & Ohishi, Y. Experimental determination of
1208 the electrical resistivity of iron at Earth's core conditions. *Nature* **534**, 95-98 (2016).
- 1209 5. Knudson, M. D. *et al.* Direct observation of an abrupt insulator-to-metal transition in dense
1210 liquid deuterium. *Science* **348**, 1455-1460 (2015).
- 1211 6. Celliers, P. M. *et al.* Insulator-metal transition in dense fluid deuterium. *Science* **361**,
1212 677-682 (2018).
- 1213 7. Eremets, M. I., Drozdov, A. P., Kong, P. & Wang, H. Semimetallic molecular hydrogen at
1214 pressure above 350 GPa. *Nat. Phys.* **15**, 1246-1249 (2019).
- 1215 8. Ji, C. *et al.* Ultrahigh-pressure isostructural electronic transitions in hydrogen. *Nature* **573**,
1216 558-562 (2019).
- 1217 9. Cheng, B., Mazzola, G., Pickard, C. J. & Ceriotti, M. Evidence for supercritical behaviour of
1218 high-pressure liquid hydrogen. *Nature* **585**, 217-220 (2020).
- 1219 10. Jiang, S. *et al.* A spectroscopic study of the insulator-metal transition in liquid hydrogen
1220 and deuterium. *Adv. Sci.* **7**, 1901668 (2020).
- 1221 11. Mao, H.-K., Chen, X.-J., Ding, Y., Li, B. & Wang, L. Solids, liquids, and gases under high
1222 pressure. *Rev. Mod. Phys.* **90**, 015007 (2018).
- 1223 12. Zhang, L., Wang, Y., Lv, J. & Ma, Y. Materials discovery at high pressures. *Nat. Rev. Mater.*
1224 **2**, 17005 (2017).
- 1225 13. Miao, M., Sun, Y., Zurek, E. & Lin, H. Chemistry under high pressure. *Nat. Rev. Chem.* **4**,
1226 508-527 (2020).
- 1227 14. Hofmeister, A. M. Pressure dependence of thermal transport properties. *Proc. Natl. Acad.*
1228 *Sci. USA* **104**, 9192-9197 (2007).
- 1229 15. Drozdov, A., Eremets, M., Troyan, I., Ksenofontov, V. & Shylin, S. I. Conventional
1230 superconductivity at 203 kelvin at high pressures in the sulfur hydride system. *Nature* **525**,
1231 73-76 (2015).
- 1232 16. Drozdov, A. *et al.* Superconductivity at 250 K in lanthanum hydride under high pressures.
1233 *Nature* **569**, 528-531 (2019).
- 1234 17. Somayazulu, M. *et al.* Evidence for superconductivity above 260 K in lanthanum
1235 superhydride at megabar pressures. *Phys. Rev. Lett.* **122**, 027001 (2019).
- 1236 18. Snider, E. *et al.* Room-temperature superconductivity in a carbonaceous sulfur hydride.
1237 *Nature* **586**, 373-377 (2020).
- 1238 19. Medvedev, S. *et al.* Electronic and magnetic phase diagram of β -Fe_{1.01}Se with
1239 superconductivity at 36.7 K under pressure. *Nat. Mater.* **8**, 630-633 (2009).
- 1240 20. Chen, X.-J. *et al.* Enhancement of superconductivity by pressure-driven competition in
1241 electronic order. *Nature* **466**, 950-953 (2010).
- 1242 21. Sun, L. *et al.* Re-emerging superconductivity at 48 kelvin in iron chalcogenides. *Nature*
1243 **483**, 67-69 (2012).
- 1244 22. Chen, L.-C. *et al.* Enhancement of thermoelectric performance across the topological

- 1245 phase transition in dense lead selenide. *Nat. Mater.* **18**, 1321-1326 (2019).
- 1246 23. Chi, Z.-H. *et al.* Pressure-induced metallization of molybdenum disulfide. *Phys. Rev. Lett.*
- 1247 **113**, 036802 (2014).
- 1248 24. Zhao, X.-M. *et al.* Pressure tuning of the charge density wave and superconductivity in
- 1249 2H-TaS₂. *Phys. Rev. B* **101**, 134506 (2020).
- 1250 25. Mao, H. K. High-pressure physics: Sustained static generation of 1.36 to 1.72 megabars.
- 1251 *Science* **200**, 1145-1147 (1978).
- 1252 26. Mao, H. K. & Bell, P. M. High-pressure physics: The 1-megabar mark on the ruby R₁ static
- 1253 pressure scale. *Science* **191**, 851-852 (1976).
- 1254 27. Jayaraman, A. Diamond anvil cell and high-pressure physical investigations. *Rev. Mod.*
- 1255 *Phys.* **55**, 65-108 (1983).
- 1256 28. Wang, L. *et al.* Nanoprobe measurements of materials at megabar pressures. *Proc. Natl.*
- 1257 *Acad. Sci. USA* **107**, 6140-6145 (2010).
- 1258 29. Dubrovinsky, L., Dubrovinskaia, N., Prakapenka, V. B. & Abakumov, A. M. Implementation
- 1259 of micro-ball nanodiamond anvils for high-pressure studies above 6 Mbar. *Nat. Commun.*
- 1260 **3**, 1163 (2012).
- 1261 30. Irifune, T., Kunimoto, T., Shinmei, T. & Tange, Y. High pressure generation in Kawai-type
- 1262 multianvil apparatus using nano-polycrystalline diamond anvils. *Comptes Rendus*
- 1263 *Geoscience* **351**, 260-268 (2019).
- 1264 31. Liebermann, R. C. Multi-anvil, high pressure apparatus: a half-century of development
- 1265 and progress. *High Press. Res.* **31**, 493-532 (2011).
- 1266 32. Dubrovinsky, L. *et al.* The most incompressible metal osmium at static pressures above
- 1267 750 gigapascals. *Nature* **525**, 226-229 (2015).
- 1268 33. Barker Jr, R. & Chen, R. Grüneisen parameter from thermal conductivity measurements
- 1269 under pressure. *J. Chem. Phys.* **53**, 2616-2620 (1970).
- 1270 34. Walker, I. Nonmagnetic piston-cylinder pressure cell for use at 35 kbar and above. *Rev.*
- 1271 *Sci. Instrum.* **70**, 3402-3412 (1999).
- 1272 35. Mao, H. K., Xu, J. & Bell, P. M. Calibration of the ruby pressure gauge to 800 kbar under
- 1273 quasi-hydrostatic conditions. *J. Geophys. Res.* **91**, 4673-4676 (1986).
- 1274 36. Akahama, Y. & Kawamura, H. Pressure calibration of diamond anvil Raman gauge to 410
- 1275 GPa. *J. Phys. Conf. Ser.* **215**, 012195 (2010).
- 1276 37. Livneh, T. & Sterer, E. Resonant Raman scattering at exciton states tuned by pressure and
- 1277 temperature in 2H-MoS₂. *Phys. Rev. B* **81**, 195209 (2010).
- 1278 38. Beck, P. *et al.* Measurement of thermal diffusivity at high pressure using a transient
- 1279 heating technique. *Appl. Phys. Lett.* **91**, 181914 (2007).
- 1280 39. Hsieh, W.-P., Chen, B., Li, J., Keblinski, P. & Cahill, D. G. Pressure tuning of the thermal
- 1281 conductivity of the layered muscovite crystal. *Phys. Rev. B* **80**, 180302 (2009).
- 1282 40. Wang, Y. *et al.* Pressure-induced phase transformation, reversible amorphization, and
- 1283 anomalous visible light response in organolead bromide perovskite. *J. Am. Chem. Soc.* **137**,
- 1284 11144-11149 (2015).
- 1285 41. Conley, H. J. *et al.* Bandgap engineering of strained monolayer and bilayer MoS₂. *Nano*
- 1286 *Lett.* **13**, 3626-3630 (2013).
- 1287 42. Bertolazzi, S., Brivio, J. & Kis, A. Stretching and breaking of ultrathin MoS₂. *ACS Nano* **5**,
- 1288 9703-9709 (2011).

- 1289 43. Wang, Y. *et al.* Strain-induced direct–indirect bandgap transition and phonon modulation
1290 in monolayer WS₂. *Nano Res.* **8**, 2562-2572 (2015).
- 1291 44. Wu, W. *et al.* Giant mechano-optoelectronic effect in an atomically thin semiconductor.
1292 *Nano Lett.* **18**, 2351-2357 (2018).
- 1293 45. He, K., Poole, C., Mak, K. F. & Shan, J. Experimental demonstration of continuous
1294 electronic structure tuning via strain in atomically thin MoS₂. *Nano Lett.* **13**, 2931-2936
1295 (2013).
- 1296 46. Rice, C. *et al.* Raman-scattering measurements and first-principles calculations of
1297 strain-induced phonon shifts in monolayer MoS₂. *Phys. Rev. B* **87**, 081307 (2013).
- 1298 47. Wang, Y., Cong, C., Qiu, C. & Yu, T. Raman spectroscopy study of lattice vibration and
1299 crystallographic orientation of monolayer MoS₂ under uniaxial strain. *Small* **9**, 2857-2861
1300 (2013).
- 1301 48. Ci, P. *et al.* Quantifying van der Waals interactions in layered transition metal
1302 dichalcogenides from pressure-enhanced valence band splitting. *Nano Lett.* **17**,
1303 4982-4988 (2017).
- 1304 49. Meng, X. *et al.* Thermal conductivity enhancement in MoS₂ under extreme strain. *Phys.*
1305 *Rev. Lett.* **122**, 155901 (2019).
- 1306 50. Howie, R. T., Magdău, I. B., Goncharov, A. F., Ackland, G. J. & Gregoryanz, E. Phonon
1307 localization by mass disorder in dense hydrogen-deuterium binary alloy. *Phys. Rev. Lett.*
1308 **113**, 175501 (2014).
- 1309 51. Howie, R. T., Dalladay-Simpson, P. & Gregoryanz, E. Raman spectroscopy of hot hydrogen
1310 above 200 GPa. *Nat. Mater.* **14**, 495-499 (2015).
- 1311 52. Dalladay-Simpson, P., Howie, R. T. & Gregoryanz, E. Evidence for a new phase of dense
1312 hydrogen above 325 gigapascals. *Nature* **529**, 63-67 (2016).
- 1313 53. Li, F. *et al.* Brillouin scattering spectroscopy for a laser heated diamond anvil cell. *Appl.*
1314 *Phys. Lett.* **88**, 203507 (2006).
- 1315 54. Nayak, A. P. *et al.* Pressure-induced semiconducting to metallic transition in multilayered
1316 molybdenum disulphide. *Nat. Commun.* **5**, 3731 (2014).
- 1317 55. Chi, Z. *et al.* Superconductivity in pristine 2H_a-MoS₂ at ultrahigh pressure. *Phys. Rev. Lett.*
1318 **120**, 037002 (2018).
- 1319 56. Mao, W. L. *et al.* Bonding changes in compressed superhard graphite. *Science* **302**,
1320 425-427 (2003).
- 1321 57. Li, Q. *et al.* Superhard monoclinic polymorph of carbon. *Phys. Rev. Lett.* **102**, 175506
1322 (2009).
- 1323 58. Tian, Y. *et al.* Ultrahard nanotwinned cubic boron nitride. *Nature* **493**, 385-388 (2013).
- 1324 59. Huang, Q. *et al.* Nanotwinned diamond with unprecedented hardness and stability.
1325 *Nature* **510**, 250-253 (2014).
- 1326 60. Hsieh, W.-P. *et al.* Low thermal conductivity of iron-silicon alloys at Earth's core conditions
1327 with implications for the geodynamo. *Nat. Commun.* **11**, 3332 (2020).
- 1328 61. Chen, X.-J. *et al.* Pressure-induced phonon frequency shifts in transition-metal nitrides.
1329 *Phys. Rev. B* **70**, 014501 (2004).
- 1330 62. Ma, Z. *et al.* Pressure-induced emission of cesium lead halide perovskite nanocrystals.
1331 *Nat. Commun.* **9**, 4506 (2018).
- 1332 63. Li, M., Liu, T., Wang, Y., Yang, W. & Lü, X. Pressure responses of halide perovskites with

1333 various compositions, dimensionalities, and morphologies. *Matter Radiat. Extremes* **5**,
1334 018201 (2020).

1335 64. Cahill, D. G. *et al.* Nanoscale thermal transport. *J. Appl. Phys.* **93**, 793-818 (2003).

1336 65. Cahill, D. G. *et al.* Nanoscale thermal transport. II. 2003–2012. *Appl. Phys. Rev.* **1**, 011305
1337 (2014).

1338 66. Giri, A. & Hopkins, P. E. A review of experimental and computational advances in thermal
1339 boundary conductance and nanoscale thermal transport across solid interfaces. *Adv.*
1340 *Funct. Mater.* **30**, 1903857 (2020).

1341 67. Snyder, G. J. & Toberer, E. S. Complex thermoelectric materials. *Nat. Mater.* **7**, 105-114
1342 (2008).

1343 68. Huxtable, S. T., Cahill, D. G. & Phinney, L. M. Thermal contact conductance of adhered
1344 microcantilevers. *J. Appl. Phys.* **95**, 2102-2108 (2004).

1345 69. Kaviani, M. *Heat transfer physics* (Cambridge University Press, 2014).

1346 70. Chen, S. *et al.* Thermal conductivity of isotopically modified graphene. *Nat. Mater.* **11**,
1347 203-207 (2012).

1348 71. Li, X., Maute, K., Dunn, M. L. & Yang, R. Strain effects on the thermal conductivity of
1349 nanostructures. *Phys. Rev. B* **81**, 245318 (2010).

1350 72. Ding, Z., Pei, Q.-X., Jiang, J.-W. & Zhang, Y.-W. Manipulating the thermal conductivity of
1351 monolayer MoS₂ via lattice defect and strain engineering. *J. Phys. Chem. C* **119**,
1352 16358-16365 (2015).

1353 73. Wei, Z., Chen, Y. & Dames, C. Negative correlation between in-plane bonding strength
1354 and cross-plane thermal conductivity in a model layered material. *Appl. Phys. Lett.* **102**,
1355 011901 (2013).

1356 74. Lindsay, L., Broido, D. A., Carrete, J., Mingo, N. & Reinecke, T. L. Anomalous pressure
1357 dependence of thermal conductivities of large mass ratio compounds. *Phys. Rev. B* **91**,
1358 121202 (2015).

1359 75. Ouyang, T. & Hu, M. Competing mechanism driving diverse pressure dependence of
1360 thermal conductivity of XTe (X=Hg, Cd, and Zn). *Phys. Rev. B* **92**, 235204 (2015).

1361 76. Ravichandran, N. K. & Broido, D. Non-monotonic pressure dependence of the thermal
1362 conductivity of boron arsenide. *Nat. Commun.* **10**, 827 (2019).

1363 77. Ross, R. G., Andersson, P., Sundqvist, B. & Backstrom, G. Thermal conductivity of solids
1364 and liquids under pressure. *Rep. Prog. Phys.* **47**, 1347-1402 (1984).

1365 78. Chen, J., Walther, J. H. & Koumoutsakos, P. Strain engineering of Kapitza resistance in
1366 few-layer graphene. *Nano Lett.* **14**, 819-825 (2014).

1367 79. Hohensee, G. T., Wilson, R. & Cahill, D. G. Thermal conductance of metal–diamond
1368 interfaces at high pressure. *Nat. Commun.* **6**, 6578 (2015).

1369 80. Vandersande, J. & Wood, C. The thermal conductivity of insulators and semiconductors.
1370 *Contemp. Phys.* **27**, 117-144 (1986).

1371 81. Wang, Y., Qiu, B., J. H. McGaughey, A., Ruan, X. & Xu, X. Mode-wise thermal conductivity
1372 of bismuth telluride. *J. Heat Transfer* **135**, 091102 (2013).

1373 82. Borca-Tasciuc, D.-A. *et al.* Thermal properties of electrodeposited bismuth telluride
1374 nanowires embedded in amorphous alumina. *Appl. Phys. Lett.* **85**, 6001-6003 (2004).

1375 83. Mavrokefalos, A. *et al.* Thermoelectric and structural characterizations of individual
1376 electrodeposited bismuth telluride nanowires. *J. Appl. Phys.* **105**, 104318 (2009).

1377 84. Park, K. H., Mohamed, M., Aksamija, Z. & Ravaoli, U. Phonon scattering due to van der
1378 Waals forces in the lattice thermal conductivity of Bi₂Te₃ thin films. *J. Appl. Phys.* **117**,
1379 015103 (2015).

1380 85. Al-Alam, P. *et al.* Lattice thermal conductivity of Bi₂Te₃ and SnSe using Debye-Callaway
1381 and Monte Carlo phonon transport modeling: Application to nanofilms and nanowires.
1382 *Phys. Rev. B* **100**, 115304 (2019).

1383 86. Prasher, R. Acoustic mismatch model for thermal contact resistance of van der Waals
1384 contacts. *Appl. Phys. Lett.* **94**, 041905 (2009).

1385 87. Huang, B.-L. & Kaviani, M. Ab initio and molecular dynamics predictions for electron and
1386 phonon transport in bismuth telluride. *Phys. Rev. B* **77**, 125209 (2008).

1387 88. Qiu, B. & Ruan, X. Molecular dynamics simulations of lattice thermal conductivity of
1388 bismuth telluride using two-body interatomic potentials. *Phys. Rev. B* **80**, 165203 (2009).

1389 89. Termentzidis, K. *et al.* Large thermal conductivity decrease in point defective Bi₂Te₃ bulk
1390 materials and superlattices. *J. Appl. Phys.* **113**, 013506 (2013).

1391 90. Kapitza, P. L. Heat transfer and superfluidity of helium II. *Phys. Rev.* **60**, 354-355 (1941).

1392 91. Pop, E. Energy dissipation and transport in nanoscale devices. *Nano Res.* **3**, 147-169
1393 (2010).

1394 92. Kuball, M. & Pomeroy, J. W. A review of raman thermography for electronic and
1395 opto-electronic device measurement with submicron spatial and nanosecond temporal
1396 resolution. *IEEE Trans. Device Mater. Reliab.* **16**, 667-684 (2016).

1397 93. Siegrist, T., Merkelbach, P. & Wuttig, M. Phase change materials: Challenges on the path
1398 to a universal storage device. *Annu. Rev. Condens. Matter Phys.* **3**, 215-237 (2012).

1399 94. Wuttig, M., Bhaskaran, H. & Taubner, T. Phase-change materials for non-volatile photonic
1400 applications. *Nat. Photonics* **11**, 465-476 (2017).

1401 95. Costescu, R. M., Cahill, D. G., Fabreguette, F. H., Sechrist, Z. A. & George, S. M. Ultra-low
1402 thermal conductivity in W/Al₂O₃ nanolaminates. *Science* **303**, 989-990 (2004).

1403 96. Losego, M. D., Blitz, I. P., Vaia, R. A., Cahill, D. G. & Braun, P. V. Ultralow thermal
1404 conductivity in organoclay nanolaminates synthesized via simple self-assembly. *Nano Lett.*
1405 **13**, 2215-2219 (2013).

1406 97. Giri, A., Donovan, B. F. & Hopkins, P. E. Localization of vibrational modes leads to reduced
1407 thermal conductivity of amorphous heterostructures. *Phys. Rev. Mater.* **2**, 056002 (2018).

1408 98. Merabia, S., Shenogin, S., Joly, L., Keblinski, P. & Barrat, J.-L. Heat transfer from
1409 nanoparticles: A corresponding state analysis. *Proc. Natl. Acad. Sci. USA* **106**,
1410 15113-15118 (2009).

1411 99. Little, W. The transport of heat between dissimilar solids at low temperatures. *Can. J.*
1412 *Phys.* **37**, 334-349 (1959).

1413 100. Swartz, E. & Pohl, R. Thermal resistance at interfaces. *Appl. Phys. Lett.* **51**, 2200-2202
1414 (1987).

1415 101. Swartz, E. T. & Pohl, R. O. Thermal boundary resistance. *Rev. Mod. Phys.* **61**, 605-668
1416 (1989).

1417 102. Stoner, R. & Maris, H. Kapitza conductance and heat flow between solids at
1418 temperatures from 50 to 300 K. *Phys. Rev. B* **48**, 16373-16387 (1993).

1419 103. Stevens, R. J., Smith, A. N. & Norris, P. M. Measurement of thermal boundary
1420 conductance of a series of metal-dielectric interfaces by the transient thermoreflectance

technique. *J. Heat Transfer* **127**, 315-322 (2005).

104. Snyder, N. Heat transport through helium II: Kapitza conductance. *Cryogenics* **10**, 89-95 (1970).

105. Lyo, H.-K. & Cahill, D. G. Thermal conductance of interfaces between highly dissimilar materials. *Phys. Rev. B* **73**, 144301 (2006).

106. Hopkins, P. E., Norris, P. M. & Stevens, R. J. Influence of inelastic scattering at metal-dielectric interfaces. *J. Heat Transfer* **130**, 022401 (2008).

107. Stevens, R. J., Zhigilei, L. V. & Norris, P. M. Effects of temperature and disorder on thermal boundary conductance at solid–solid interfaces: Nonequilibrium molecular dynamics simulations. *International Journal of Heat and Mass Transfer* **50**, 3977-3989 (2007).

108. Sun, Z., Yuan, K., Zhang, X. & Tang, D. Pressure tuning of the thermal conductivity of gallium arsenide from first-principles calculations. *Phys. Chem. Chem. Phys.* **20**, 30331-30339 (2018).

109. Yu, H. *et al.* Large enhancement of thermoelectric performance in CuInTe₂ upon compression. *Mater. Today Phys.* **5**, 1-6 (2018).

110. Wang, L. *et al.* High-pressure phases of boron arsenide with potential high thermal conductivity. *Phys. Rev. B* **99**, 174104 (2019).

111. Saha, P., Mazumder, A. & Mukherjee, G. D. Thermal conductivity of dense hcp iron: Direct measurements using laser heated diamond anvil cell. *Geosci. Front.* **11**, 1755-1761 (2020).

112. Yuan, K., Zhang, X., Tang, D. & Hu, M. Anomalous pressure effect on the thermal conductivity of ZnO, GaN, and AlN from first-principles calculations. *Phys. Rev. B* **98**, 144303 (2018).

113. Lan, G., Ouyang, B. & Song, J. The role of low-lying optical phonons in lattice thermal conductance of rare-earth pyrochlores: A first-principle study. *Acta Mater.* **91**, 304-317 (2015).

114. Hsieh, W.-P., Lyons, A. S., Pop, E., Keblinski, P. & Cahill, D. G. Pressure tuning of the thermal conductance of weak interfaces. *Phys. Rev. B* **84**, 184107 (2011).

115. Fujisawa, H., Fujii, N., Mizutani, H., Kanamori, H. & Akimoto, S. i. Thermal diffusivity of Mg₂SiO₄, Fe₂SiO₄, and NaCl at high pressures and temperatures. *J. Geophys. Res.* **73**, 4727-4733 (1968).

116. Håkansson, B., Andersson, P. & Bäckström, G. Improved hot-wire procedure for thermophysical measurements under pressure. *Rev. Sci. Instrum.* **59**, 2269-2275 (1988).

117. Katsura, T. Thermal diffusivity of silica glass at pressures up to 9 GPa. *Phys. Chem. Miner.* **20**, 201-208 (1993).

118. Xu, Y. *et al.* Thermal diffusivity and conductivity of olivine, wadsleyite and ringwoodite to 20 GPa and 1373 K. *Phys. Earth Planet. Inter.* **143-144**, 321-336 (2004).

119. Bercegeay, C. & Bernard, S. First-principles equations of state and elastic properties of seven metals. *Phys. Rev. B* **72**, 214101 (2005).

120. Hsieh, W.-P. *et al.* Testing the minimum thermal conductivity model for amorphous polymers using high pressure. *Phys. Rev. B* **83**, 174205 (2011).

121. Park, C. H., Cheong, B.-H., Lee, K.-H. & Chang, K. J. Structural and electronic properties of cubic, 2H, 4H, and 6H SiC. *Phys. Rev. B* **49**, 4485-4493 (1994).

- 1465 122. Zha, C.-S., Mao, H.-k. & Hemley, R. J. Elasticity of MgO and a primary pressure scale to
1466 55 GPa. *Proc. Natl. Acad. Sci. USA* **97**, 13494-13499 (2000).
- 1467 123. Xie, J., Chen, S. P., Tse, J. S., Gironcoli, S. d. & Baroni, S. High-pressure thermal expansion,
1468 bulk modulus, and phonon structure of diamond. *Phys. Rev. B* **60**, 9444-9449 (1999).
- 1469 124. Barker Jr, R., Chen, R. & Frost, R. Influence of pressure and chemical structure on the
1470 thermal conductivity of vitreous poly (alkyl methacrylates). I. *J. Polym. Sci: Polym. Phys.* **15**,
1471 1199-1210 (1977).
- 1472 125. Hsieh, W.-P. *Testing theories for thermal transport using high pressure*, University of
1473 Illinois at Urbana-Champaign (2012).
- 1474 126. Frost, R. S., Chen, R. Y. S. & Barker Jr., R. E. Pressure dependence of thermal conductivity
1475 in polyethylene. *J. Appl. Phys.* **46**, 4506-4509 (1975).
- 1476 127. Manthilake, M. A. G. M., de Koker, N. & Frost, D. J. Thermal conductivity of CaGeO₃
1477 perovskite at high pressure. *Geophys. Res. Lett.* **38**, L038301 (2011).
- 1478 128. Frost, D. J. *et al.* A new large-volume multianvil system. *Phys. Earth Planet. Inter.*
1479 **143-144**, 507-514 (2004).
- 1480 129. Katsura, T. Thermal diffusivity of olivine under upper mantle conditions. *Geophys. J. Int.*
1481 **122**, 63-69 (1995).
- 1482 130. Manthilake, G. M., de Koker, N., Frost, D. J. & McCammon, C. A. Lattice thermal
1483 conductivity of lower mantle minerals and heat flux from Earth's core. *Proc. Natl. Acad.*
1484 *Sci. USA* **108**, 17901 (2011).
- 1485 131. Chai, M., Brown, J. M. & Slutsky, L. J. Thermal diffusivity of mantle minerals. *Phys. Chem.*
1486 *Miner.* **23**, 470-475 (1996).
- 1487 132. Yue, D. *et al.* Accurate temperature measurement by temperature field analysis in
1488 diamond anvil cell for thermal transport study of matter under high pressures. *Appl. Phys.*
1489 *Lett.* **112**, 081901 (2018).
- 1490 133. Franco, A. An apparatus for the routine measurement of thermal conductivity of
1491 materials for building application based on a transient hot-wire method. *Appl. Therm. Eng.*
1492 **27**, 2495-2504 (2007).
- 1493 134. Goncharov, A. F., Beck, P., Struzhkin, V. V., Haugen, B. D. & Jacobsen, S. D. Thermal
1494 conductivity of lower-mantle minerals. *Phys. Earth Planet. Inter.* **174**, 24-32 (2009).
- 1495 135. McWilliams, R. S., Konôpková, Z. & Goncharov, A. F. A flash heating method for
1496 measuring thermal conductivity at high pressure and temperature: Application to Pt. *Phys.*
1497 *Earth Planet. Inter.* **247**, 17-26 (2015).
- 1498 136. Abramson, E. H., Brown, J. M. & Slutsky, L. J. The thermal diffusivity of water at high
1499 pressures and temperatures. *J. Chem. Phys.* **115**, 10461-10463 (2001).
- 1500 137. Abramson, E. H., Slutsky, L. J. & Brown, J. M. Thermal diffusivity of fluid oxygen to 12
1501 GPa and 300 °C. *J. Chem. Phys.* **111**, 9357-9360 (1999).
- 1502 138. Pang, H.-J. *et al.* Pressure tuning of thermoelectric performance in FeNbSb. *J. Alloys*
1503 *Compd.* **805**, 1224-1230 (2019).
- 1504 139. Chen, L.-C. *et al.* Pressure-induced enhancement of thermoelectric performance in
1505 palladium sulfide. *Mater. Today Phys.* **5**, 64-71 (2018).
- 1506 140. Håkansson, B. & Ross, R. G. Effective thermal conductivity of binary dispersed
1507 composites over wide ranges of volume fraction, temperature, and pressure. *J. Appl. Phys.*
1508 **68**, 3285-3292 (1990).

- 1509 141. Andersson, O. & Suga, H. Thermal conductivity of the Ih and XI phases of ice. *Phys. Rev.*
1510 *B* **50**, 6583-6588 (1994).
- 1511 142. Andersson, O. & Suga, H. Thermal conductivity of normal and deuterated
1512 tetrahydrofuran clathrate hydrates. *J. Phys. Chem. Solids* **57**, 125-132 (1996).
- 1513 143. Andersson, O., Soldatov, A. & Sundqvist, B. Thermal conductivity of C₆₀ at pressures up
1514 to 1 GPa and temperatures in the 50-300 K range. *Phys. Rev. B* **54**, 3093-3100 (1996).
- 1515 144. Larsson, R. & Andersson, O. Lubricant thermal conductivity and heat capacity under
1516 high pressure. *Proceedings of the Institution of Mechanical Engineers, Part J: Journal of*
1517 *Engineering Tribology* **214**, 337-342 (2000).
- 1518 145. Andersson, O., Chobal, O., Rizak, I., Rizak, V. & Sabadosh, V. Effects of pressure and
1519 temperature on the thermal conductivity of Sn₂P₂S₆. *Phys. Rev. B* **83**, 134121 (2011).
- 1520 146. Yu, J., Tonpheng, B., Gröbner, G. & Andersson, O. Thermal properties and transition
1521 studies of multi-wall carbon nanotube/nylon-6 composites. *Carbon* **49**, 4858-4866 (2011).
- 1522 147. Andersson, O. Thermal conductivity of normal and deuterated water, crystalline ice, and
1523 amorphous ices. *J. Chem. Phys.* **149**, 124506 (2018).
- 1524 148. Soldatov, A. & Sundqvist, B. Molecular rotation in C₇₀ at high pressures: A thermal
1525 conductivity study. *J. Phys. Chem. Solids* **57**, 1371-1375 (1996).
- 1526 149. Goncharov, A. F. *et al.* Effect of composition, structure, and spin state on the thermal
1527 conductivity of the Earth's lower mantle. *Phys. Earth Planet. Inter.* **180**, 148-153 (2010).
- 1528 150. Geballe, Z. M., Sime, N., Badro, J., van Keken, P. E. & Goncharov, A. F. Thermal
1529 conductivity near the bottom of the Earth's lower mantle: Measurements of pyrolite up
1530 to 120 GPa and 2500 K. *Earth Planet. Sci. Lett.* **536**, 116161 (2020).
- 1531 151. Chiritescu, C. *et al.* Ultralow thermal conductivity in disordered, layered WSe₂ crystals.
1532 *Science* **315**, 351-353 (2007).
- 1533 152. Ge, Z., Cahill, D. G. & Braun, P. V. Thermal conductance of hydrophilic and hydrophobic
1534 interfaces. *Phys. Rev. Lett.* **96**, 186101 (2006).
- 1535 153. Beran, A. The reflectance behaviour of gold at temperatures up to 500 °C. *Tschermaks*
1536 *Mineralogische und Petrographische Mitteilungen* **34**, 211-215 (1985).
- 1537 154. Hsieh, W.-P. & Cahill, D. G. Ta and Au(Pd) alloy metal film transducers for time-domain
1538 thermoreflectance at high pressures. *J. Appl. Phys.* **109**, 113520 (2011).
- 1539 155. Hsieh, W.-P. Thermal conductivity of methanol-ethanol mixture and silicone oil at high
1540 pressures. *J. Appl. Phys.* **117**, 235901 (2015).
- 1541 156. Jeong, J. *et al.* Picosecond transient thermoreflectance for thermal conductivity
1542 characterization. *Nanoscale Microscale Thermophys. Eng.* **23**, 211-221 (2019).
- 1543 157. Hasegawa, A., Yagi, T. & Ohta, K. Combination of pulsed light heating thermoreflectance
1544 and laser-heated diamond anvil cell for in-situ high pressure-temperature thermal
1545 diffusivity measurements. *Rev. Sci. Instrum.* **90**, 074901 (2019).
- 1546 158. Omini, M. & Sparavigna, A. An iterative approach to the phonon Boltzmann equation in
1547 the theory of thermal conductivity. *Physica B: Condens. Matter* **212**, 101-112 (1995).
- 1548 159. Fugallo, G., Lazzeri, M., Paulatto, L. & Mauri, F. Ab initio variational approach for
1549 evaluating lattice thermal conductivity. *Phys. Rev. B* **88**, 045430 (2013).
- 1550 160. Stillinger, F. H. & Weber, T. A. Computer simulation of local order in condensed phases of
1551 silicon. *Phys. Rev. B* **31**, 5262-5271 (1985).
- 1552 161. Tersoff, J. Empirical interatomic potential for carbon, with applications to amorphous

- 1553 carbon. *Phys. Rev. Lett.* **61**, 2879-2882 (1988).
- 1554 162. Brenner, D. W. Empirical potential for hydrocarbons for use in simulating the chemical
1555 vapor deposition of diamond films. *Phys. Rev. B* **42**, 9458-9471 (1990).
- 1556 163. Weber, W. Adiabatic bond charge model for the phonons in diamond, Si, Ge, and α -Sn.
1557 *Phys. Rev. B* **15**, 4789-4803 (1977).
- 1558 164. Broido, D. A., Malorny, M., Birner, G., Mingo, N. & Stewart, D. A. Intrinsic lattice thermal
1559 conductivity of semiconductors from first principles. *Appl. Phys. Lett.* **91**, 231922 (2007).
- 1560 165. Lindsay, L. First principles Peierls-Boltzmann phonon thermal transport: A topical review.
1561 *Nanoscale Microscale Thermophys. Eng.* **20**, 67-84 (2016).
- 1562 166. de Koker, N. Thermal conductivity of MgO periclase from equilibrium first principles
1563 molecular dynamics. *Phys. Rev. Lett.* **103**, 125902 (2009).
- 1564 167. Tian, F. *et al.* Unusual high thermal conductivity in boron arsenide bulk crystals. *Science*
1565 **361**, 582-585 (2018).
- 1566 168. Ravichandran, N. K. & Broido, D. Unified first-principles theory of thermal properties of
1567 insulators. *Phys. Rev. B* **98**, 085205 (2018).
- 1568 169. Hamrin, C. E. & Thodos, G. The thermal conductivity of hydrogen for pressures up to
1569 660 atm and temperatures between 1.6 and 74.6 °C. *Physica* **32**, 918-932 (1966).
- 1570 170. Roder, H. M. The thermal conductivity of oxygen. *J. Res.* **87**, 279-310 (1982).
- 1571 171. Roder, H. M. Thermal conductivity of methane for temperatures between 110 and 310 K
1572 with pressures to 70 MPa. *Int. J. Thermophys.* **6**, 119-142 (1985).
- 1573 172. Gilmore, T. F. & Comings, E. W. Thermal conductivity of binary mixtures of carbon
1574 dioxide, nitrogen, and ethane at high pressures: Comparison with correlation and theory.
1575 *AIChE Journal* **12**, 1172-1178 (1966).
- 1576 173. Sengers, J. V., Bolk, W. T. & Stigter, C. J. The thermal conductivity of neon between 25 °C
1577 and 75 °C at pressures up to 2600 atmospheres. *Physica* **30**, 1018-1026 (1964).
- 1578 174. Millat, J., Mustafa, M., Ross, M., Wakeham, W. A. & Zalaf, M. The thermal conductivity
1579 of argon, carbon dioxide and nitrous oxide. *Physica A: Statistical Mechanics and its*
1580 *Applications* **145**, 461-497 (1987).
- 1581 175. Goncharov, A. F. *et al.* Thermal conductivity of argon at high pressures and high
1582 temperatures. *J. Appl. Phys.* **111**, 112609 (2012).
- 1583 176. Tretiakov, K. V. & Scandolo, S. Thermal conductivity of solid argon at high pressure and
1584 high temperature: A molecular dynamics study. *J. Chem. Phys.* **121**, 11177-11182 (2004).
- 1585 177. Chernatynskiy, A. & Phillpot, S. R. Thermal conductivity of argon at high pressure from
1586 first principles calculations. *J. Appl. Phys.* **114**, 064902 (2013).
- 1587 178. Michels, A., Sengers, J. V. & Van De Klundert, L. J. M. The thermal conductivity of argon
1588 at elevated densities. *Physica* **29**, 149-160 (1963).
- 1589 179. Parrish, K. D., Jain, A., Larkin, J. M., Saidi, W. A. & McGaughey, A. J. H. Origins of thermal
1590 conductivity changes in strained crystals. *Phys. Rev. B* **90**, 235201 (2014).
- 1591 180. Johns, A. I., Rashid, S., Rowan, L., Watson, J. T. R. & Clifford, A. A. The thermal
1592 conductivity of pure nitrogen and of mixtures of nitrogen and carbon dioxide at elevated
1593 temperatures and pressures. *Int. J. Thermophys.* **9**, 3-19 (1988).
- 1594 181. Moroe, S. *et al.* Measurements of hydrogen thermal conductivity at high pressure and
1595 high temperature. *Int. J. Thermophys.* **32**, 1887-1917 (2011).
- 1596 182. Shulga, V. M., Eldarov, F. G., Atanov, Y. A. & Kuyumchev, A. A. Thermal conductivity and

1597 heat capacity of liquid toluene at temperatures between 255 and 400 K and at pressures
1598 up to 1000 MPa. *Int. J. Thermophys.* **7**, 1147-1161 (1986).

1599 183. Ramaswamy, R., Balasubramaniam, V. M. & Sastry, S. K. Thermal conductivity of
1600 selected liquid foods at elevated pressures up to 700 MPa. *J. Food Eng.* **83**, 444-451
1601 (2007).

1602 184. Chen, B., Hsieh, W.-P., Cahill, D. G., Trinkle, D. R. & Li, J. Thermal conductivity of
1603 compressed H₂O to 22 GPa: A test of the Leibfried-Schlomann equation. *Phys. Rev. B* **83**,
1604 132301 (2011).

1605 185. Ross, R. G., Andersson, P. & Bäckström, G. Thermal conductivity of allotropic
1606 modifications of ice. *Nature* **259**, 553-554 (1976).

1607 186. Hsieh, W.-P., Deschamps, F., Okuchi, T. & Lin, J.-F. Effects of iron on the lattice thermal
1608 conductivity of Earth's deep mantle and implications for mantle dynamics. *Proc. Natl.*
1609 *Acad. Sci. USA* **115**, 4099-4104 (2018).

1610 187. Hsieh, W. P., Deschamps, F., Okuchi, T. & Lin, J. F. Reduced lattice thermal conductivity of
1611 Fe-bearing bridgmanite in Earth's deep mantle. *J. Geophys. Res.: Solid Earth* **122**,
1612 4900-4917 (2017).

1613 188. Chao, K.-H. & Hsieh, W.-P. Thermal conductivity anomaly in (Fe_{0.78}Mg_{0.22})CO₃ siderite
1614 across spin transition of iron. *J. Geophys. Res.: Solid Earth* **124**, 1388-1396 (2019).

1615 189. Dalton, D. A., Hsieh, W.-P., Hohensee, G. T., Cahill, D. G. & Goncharov, A. F. Effect of mass
1616 disorder on the lattice thermal conductivity of MgO periclase under pressure. *Sci. Rep.* **3**,
1617 2400 (2013).

1618 190. Tang, X. & Dong, J.J. Lattice thermal conductivity of MgO at conditions of Earth's interior.
1619 *Proc. Natl. Acad. Sci. USA* **107**, 4539-4543 (2010).

1620 191. Ohta, K. *et al.* Lattice thermal conductivity of MgSiO₃ perovskite and post-perovskite at
1621 the core-mantle boundary. *Earth Planet. Sci. Lett.* **349-350**, 109-115 (2012).

1622 192. Marzotto, E. *et al.* Effect of water on lattice thermal conductivity of ringwoodite and its
1623 implications for the thermal evolution of descending slabs. *Geophys. Res. Lett.* **47**, e87607
1624 (2020).

1625 193. Hsieh, W.-P. *et al.* Spin transition of iron in δ -(Al,Fe)OOH induces thermal anomalies in
1626 Earth's lower mantle. *Geophys. Res. Lett.* **47**, e87036 (2020).

1627 194. Elalfy, L., Music, D. & Hu, M. First principles investigation of anomalous
1628 pressure-dependent thermal conductivity of chalcopyrites. *Materials* **12**, 3491 (2019).

1629 195. Hohensee, G. T., Fellinger, M. R., Trinkle, D. R. & Cahill, D. G. Thermal transport across
1630 high-pressure semiconductor-metal transition in Si and Si_{0.991}Ge_{0.009}. *Phys. Rev. B* **91**,
1631 205104 (2015).

1632 196. Harish, S. *et al.* Thermal conductivity reduction of crystalline silicon by high-pressure
1633 torsion. *Nanoscale Res. Lett.* **9**, 326 (2014).

1634 197. Kuryliuk, V., Nepochaty, O., Chantrenne, P., Lacroix, D. & Isaiev, M. Thermal conductivity
1635 of strained silicon: Molecular dynamics insight and kinetic theory approach. *J. Appl. Phys.*
1636 **126**, 055109 (2019).

1637 198. Xie, H. *et al.* Large tunability of lattice thermal conductivity of monolayer silicene via
1638 mechanical strain. *Phys. Rev. B* **93**, 075404 (2016).

1639 199. Raeisi, M., Ahmadi, S. & Rajabpour, A. Modulated thermal conductivity of 2D hexagonal
1640 boron arsenide: A strain engineering study. *Nanoscale* **11**, 21799-21810 (2019).

- 1641 200. Yebra, F., Troncoso, J. & Romani, L. Thermal conductivity measurements for organic
1642 liquids at high pressure. *J. Chem. Thermodyn.* **142**, 106005 (2020).
- 1643 201. Ross, R. G. & Sandberg, O. The thermal conductivity of four solid phases of NH_4F , and a
1644 comparison with H_2O . *J. Phys. C: Solid State Phys.* **11**, 667-672 (1978).
- 1645 202. McWilliams, R. S., Dalton, D. A., Konôpková, Z., Mahmood, M. F. & Goncharov, A. F.
1646 Opacity and conductivity measurements in noble gases at conditions of planetary and
1647 stellar interiors. *Proc. Natl. Acad. Sci. USA* **112**, 7925-7930 (2015).
- 1648 203. Xu, J. *et al.* Thermal conductivity and electrical resistivity of solid iron at Earth's core
1649 conditions from first principles. *Phys. Rev. Lett.* **121**, 096601 (2018).
- 1650 204. de Koker, N., Steinle-Neumann, G. & Vlček, V. Electrical resistivity and thermal
1651 conductivity of liquid Fe alloys at high P and T, and heat flux in Earth's core. *Proc. Natl.*
1652 *Acad. Sci. USA* **109**, 4070-4073 (2012).
- 1653 205. Seagle, C. T., Cottrell, E., Fei, Y., Hummer, D. R. & Prakapenka, V. B. Electrical and thermal
1654 transport properties of iron and iron-silicon alloy at high pressure. *Geophys. Res. Lett.* **40**,
1655 5377-5381 (2013).
- 1656 206. Tang, X. & Dong, J. J. Pressure dependence of harmonic and anharmonic lattice
1657 dynamics in MgO: A first-principles calculation and implications for lattice thermal
1658 conductivity. *Phys. Earth Planet. Inter.* **174**, 33-38 (2009).
- 1659 207. Goncharov, A. F., Haugen, B. D., Struzhkin, V. V., Beck, P. & Jacobsen, S. D. Radiative
1660 conductivity in the Earth's lower mantle. *Nature* **456**, 231-234 (2008).
- 1661 208. Goncharov, A. F., Struzhkin, V. V. & Jacobsen, S. D. Reduced radiative conductivity of
1662 low-spin (Mg,Fe)O in the lower mantle. *Science* **312**, 1205-1208 (2006).
- 1663 209. Lobanov, S. S., Holtgrewe, N., Lin, J.-F. & Goncharov, A. F. Radiative conductivity and
1664 abundance of post-perovskite in the lowermost mantle. *Earth Planet. Sci. Lett.* **479**, 43-49
1665 (2017).
- 1666 210. Lobanov, S. S. *et al.* Blocked radiative heat transport in the hot pyrolytic lower mantle.
1667 *Earth Planet. Sci. Lett.* **537**, 116176 (2020).
- 1668 211. Stackhouse, S., Stixrude, L. & Karki, B. B. Thermal conductivity of periclase (MgO) from
1669 first principles. *Phys. Rev. Lett.* **104**, 208501 (2010).
- 1670 212. de Koker, N. Thermal conductivity of MgO periclase at high pressure: Implications for
1671 the D'' region. *Earth Planet. Sci. Lett.* **292**, 392-398 (2010).
- 1672 213. Goncharov, A. F. *et al.* Experimental study of thermal conductivity at high pressures:
1673 Implications for the deep Earth's interior. *Phys. Earth Planet. Inter.* **247**, 11-16 (2015).
- 1674 214. Yagi, T. *et al.* Thermal diffusivity measurement in a diamond anvil cell using a light pulse
1675 thermorefectance technique. *Meas. Sci. Technol.* **22**, 024011, (2010).
- 1676 215. Chang, Y.-Y., Hsieh, W.-P., Tan, E. & Chen, J. Hydration-reduced lattice thermal
1677 conductivity of olivine in Earth's upper mantle. *Proc. Natl. Acad. Sci. USA* **114**, 4078-4081
1678 (2017).
- 1679 216. Osako, M. & Ito, E. Thermal diffusivity of MgSiO_3 perovskite. *Geophys. Res. Lett.* **18**,
1680 239-242 (1991).
- 1681 217. Hofmeister, A. M. Mantle values of thermal conductivity and the geotherm from
1682 phonon lifetimes. *Science* **283**, 1699-1706 (1999).
- 1683 218. Hofmeister, A. M. Inference of high thermal transport in the lower mantle from
1684 laser-flash experiments and the damped harmonic oscillator model. *Phys. Earth Planet.*

- 1685 *Inter.* **170**, 201-206 (2008).
- 1686 219. Okuda, Y. *et al.* The effect of iron and aluminum incorporation on lattice thermal
1687 conductivity of bridgmanite at the Earth's lower mantle. *Earth Planet. Sci. Lett.* **474**, 25-31
1688 (2017).
- 1689 220. Ohta, K., Yagi, T., Hirose, K. & Ohishi, Y. Thermal conductivity of ferropericlase in the
1690 Earth's lower mantle. *Earth Planet. Sci. Lett.* **465**, 29-37 (2017).
- 1691 221. Haigis, V., Salanne, M. & Jahn, S. Thermal conductivity of MgO, MgSiO₃ perovskite and
1692 post-perovskite in the Earth's deep mantle. *Earth Planet. Sci. Lett.* **355**, 102-108 (2012).
- 1693 222. Ammann, M. W. *et al.* Variation of thermal conductivity and heat flux at the Earth's core
1694 mantle boundary. *Earth Planet. Sci. Lett.* **390**, 175-185 (2014).
- 1695 223. Tang, X., Ntam, M. C., Dong, J., Rainey, E. S. G. & Kavner, A. The thermal conductivity of
1696 Earth's lower mantle. *Geophys. Res. Lett.* **41**, 2746-2752 (2014).
- 1697 224. Stackhouse, S., Stixrude, L. & Karki, B. B. First-principles calculations of the lattice
1698 thermal conductivity of the lower mantle. *Earth Planet. Sci. Lett.* **427**, 11-17 (2015).
- 1699 225. Yu, C., Zhang, G., Zhang, Y.-W. & Peng, L.-M. Strain engineering on the thermal
1700 conductivity and heat flux of thermoelectric Bi₂Te₃ nanofilm. *Nano Energy* **17**, 104-110
1701 (2015).
- 1702 226. Yang, X. *et al.* Pressure induced excellent thermoelectric behavior in skutterudites CoSb₃
1703 and IrSb₃. *Phys. Chem. Chem. Phys.* **21**, 851-858 (2019).
- 1704 227. Liu, J., Choi, G.-M. & Cahill, D. G. Measurement of the anisotropic thermal conductivity
1705 of molybdenum disulfide by the time-resolved magneto-optic Kerr effect. *J. Appl. Phys.*
1706 **116**, 233107 (2014).
- 1707 228. Jo, I., Pettes, M. T., Ou, E., Wu, W. & Shi, L. Basal-plane thermal conductivity of few-layer
1708 molybdenum disulfide. *Appl. Phys. Lett.* **104**, 201902 (2014).
- 1709 229. Sahoo, S., Gaur, A. P. S., Ahmadi, M., Guinel, M. J. F. & Katiyar, R. S.
1710 Temperature-dependent Raman studies and thermal conductivity of few-layer MoS₂. *J.*
1711 *Phys. Chem. C* **117**, 9042-9047 (2013).
- 1712 230. Muratore, C. *et al.* Cross-plane thermal properties of transition metal dichalcogenides.
1713 *Appl. Phys. Lett.* **102**, 081604 (2013).
- 1714 231. Jiang, P., Qian, X., Gu, X. & Yang, R. Probing anisotropic thermal conductivity of
1715 transition metal dichalcogenides MX₂ (M = Mo, W and X = S, Se) using time-domain
1716 thermoreflectance. *Adv. Mater.* **29**, 1701068 (2017).
- 1717 232. Wang, X. & Tabarraei, A. Phonon thermal conductivity of monolayer MoS₂. *Appl. Phys.*
1718 *Lett.* **108**, 191905 (2016).
- 1719 233. Yuan, K., Zhang, X., Li, L. & Tang, D. Effects of tensile strain and finite size on thermal
1720 conductivity in monolayer WSe₂. *Phys. Chem. Chem. Phys.* **21**, 468-477 (2019).
- 1721 234. Broido, D. A., Lindsay, L. & Ward, A. Thermal conductivity of diamond under extreme
1722 pressure: A first-principles study. *Phys. Rev. B* **86**, 115203 (2012).
- 1723 235. Nan, C.-W., Birringer, R., Clarke, D. R. & Gleiter, H. Effective thermal conductivity of
1724 particulate composites with interfacial thermal resistance. *J. Appl. Phys.* **81**, 6692-6699
1725 (1997).
- 1726 236. Shenogin, S., Xue, L., Ozisik, R., Keblinski, P. & Cahill, D. G. Role of thermal boundary
1727 resistance on the heat flow in carbon-nanotube composites. *J. Appl. Phys.* **95**, 8136-8144
1728 (2004).

- 1729 237. Koh, Y. K., Cao, Y., Cahill, D. G. & Jena, D. Heat-transport mechanisms in superlattices.
1730 *Adv. Funct. Mater.* **19**, 610-615 (2009).
- 1731 238. Zhou, Y. *et al.* Thermal characterization of polycrystalline diamond thin film heat
1732 spreaders grown on GaN HEMTs. *Appl. Phys. Lett.* **111**, 041901 (2017).
- 1733 239. Zhou, Y. *et al.* Barrier-layer optimization for enhanced GaN-on-diamond device cooling.
1734 *ACS Appl. Mater. Interfaces* **9**, 34416-34422 (2017).
- 1735 240. Young, D. A. & Maris, H. J. Lattice-dynamical calculation of the Kapitza resistance
1736 between fcc lattices. *Phys. Rev. B* **40**, 3685-3693 (1989).
- 1737 241. Hu, M., Keblinski, P. & Schelling, P. K. Kapitza conductance of silicon-amorphous
1738 polyethylene interfaces by molecular dynamics simulations. *Phys. Rev. B* **79**, 104305
1739 (2009).
- 1740 242. Ong, Z.-Y. & Pop, E. Molecular dynamics simulation of thermal boundary conductance
1741 between carbon nanotubes and SiO₂. *Phys. Rev. B* **81**, 155408 (2010).
- 1742 243. Wilson, R. B., Apgar, B. A., Hsieh, W.-P., Martin, L. W. & Cahill, D. G. Thermal
1743 conductance of strongly bonded metal-oxide interfaces. *Phys. Rev. B* **91**, 115414 (2015).
- 1744 244. Duffy, T. S. Synchrotron facilities and the study of the Earth's deep interior. *Rep. Prog.*
1745 *Phys.* **68**, 1811-1859 (2005).
- 1746 245. Shen, G. Y. & Mao, H. K. High-pressure studies with x-rays using diamond anvil cells. *Rep.*
1747 *Prog. Phys.* **80**, 016101 (2016).
- 1748 246. Lin, J. F., Alp, E. E. & Goncharov, A. F. in *Treatise on Geochemistry (Second Edition)* Vol.
1749 **15** (eds Heinrich D. Holland & Karl K. Turekian) 195-211 (Elsevier, 2014).
- 1750 247. Lay, T., Hernlund, J. & Buffett, B. A. Core–mantle boundary heat flow. *Nat. Geosci.* **1**,
1751 25-32 (2008).
- 1752 248. Nimmo, F. in *Treatise on Geophysics (Second Edition)* (ed Gerald Schubert) 27-55
1753 (Elsevier, 2015).
- 1754 249. Mao, J. *et al.* Advances in thermoelectrics. *Adv. Phys.* **67**, 69-147 (2018).
- 1755 250. Pei, Y., Wang, H. & Snyder, G. J. Band engineering of thermoelectric materials. *Adv.*
1756 *Mater.* **24**, 6125-6135 (2012).
- 1757 251. Novak, T. G., Kim, K. & Jeon, S. 2D and 3D nanostructuring strategies for thermoelectric
1758 materials. *Nanoscale* **11**, 19684-19699 (2019).
- 1759 252. Ovsyannikov, S. V. *et al.* Giant improvement of thermoelectric power factor of Bi₂Te₃
1760 under pressure. *J. Appl. Phys.* **104**, 053713 (2008).
- 1761 253. Ovsyannikov, S. V. & Shchennikov, V. V. Pressure-tuned colossal improvement of
1762 thermoelectric efficiency of PbTe. *Appl. Phys. Lett.* **90**, 122103 (2007).
- 1763 254. Ibarra-Hernández, W., Verstraete, M. J. & Raty, J.-Y. Effect of hydrostatic pressure on the
1764 thermoelectric properties of Bi₂Te₃. *Phys. Rev. B* **90**, 245204 (2014).
- 1765 255. Zhang, Y., Hao, S., Zhao, L.-D., Wolverton, C. & Zeng, Z. Pressure induced thermoelectric
1766 enhancement in SnSe crystals. *J. Mater. Chem. A* **4**, 12073-12079 (2016).
- 1767 256. Alsaleh, N. M., Shoko, E. & Schwingenschlögl, U. Pressure-induced conduction band
1768 convergence in the thermoelectric ternary chalcogenide CuBiS₂. *Phys. Chem. Chem. Phys.*
1769 **21**, 662-673 (2019).
- 1770 257. Chandra Shekar, N. V., Polvani, D. A., Meng, J. F. & Badding, J. V. Improved
1771 thermoelectric properties due to electronic topological transition under high pressure.
1772 *Physica B: Condens. Matter* **358**, 14-18 (2005).

1773 258. Polvani, D. A., Meng, J. F., Chandra Shekar, N. V., Sharp, J. & Badding, J. V. Large
1774 improvement in thermoelectric properties in pressure-tuned *p*-type $\text{Sb}_{1.5}\text{Bi}_{0.5}\text{Te}_3$. *Chem.*
1775 *Mater.* **13**, 2068-2071 (2001).
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1791 X.J.C. developed the outline of this work. Y.Z., Z.Y.D. and X.J.C. compiled all data in tables.
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1800 The authors declare no competing interests.
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