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Quantum Effects in Chemical Reactions under Polaritonic Vibrational Strong Coupling

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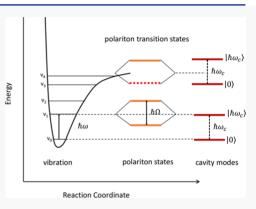
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ABSTRACT: The electromagnetic field in an optical cavity can dramatically modify and even control chemical reactivity via vibrational strong coupling (VSC). Since the typical vibration and cavity frequencies are considerably larger than thermal energy, it is essential to adopt a quantum description of cavity-catalyzed adiabatic chemical reactions. Using quantum transition state theory (TST), we examine the coherent nature of adiabatic reactions in cavities and derive the cavity-induced changes in eigenfrequencies, zero-point energy, and quantum tunneling. The resulting quantum TST calculation allows us to explain and predict the resonance effect (i.e., maximal kinetic modification via tuning the cavity frequency), collective effect (i.e., linear scaling with the molecular density), and selectivity (i.e., cavity-induced control of the branching ratio). The TST calculation is further supported by perturbative analysis of polariton normal modes, which not only provides physical insights to cavity-catalyzed chemical reactions but also presents a general approach to treat other VSC phenomena.



he electromagnetic field of an optical cavity can mix with quantum states of molecular systems to form polaritons and thus modify chemical kinetics in the vibrational strong coupling (VSC) regime. In particular, recent experiments have clearly demonstrated dramatic effects on the reactivity and selectivity of chemical kinetics when the vibrational and infrared (IR) cavity modes are strongly coupled. 1-8 Specifically, we highlight three effects: (i) resonance, i.e., maximal cavityinduced suppression or enhancement of the reaction rate under the vibrational resonance condition; (ii) collectivity, i.e., collective enhancement of the cavity-induced correction with the increase of molecule density in a cavity; (iii) selectivity, i.e., regulation of the branching ratio of reaction channels via tuning the cavity frequency and other parameters. These observations suggest a coherent light-matter interaction in terms of vibrational polaritons and support the physical picture where the IR mode catalyzes chemical reactions on the ground electronic state potential surface. This intriguing picture has stimulated theoretical and numerical studies; yet, the underlying mechanism of VSC-catalyzed chemical reactivity remains elusive.

Recent theoretical analysis adopts the quantum transition state theory (TST) for thermal rate calculations but has not fully elucidated the intriguing phenomena in VSC-catalyzed reactions. With the TST framework, the time scale separation between electronic and nuclear degrees of freedom ensures that the reactions occur adiabatically on the ground-state potential energy surface, and the adiabatic reaction rate is predominantly determined by the activation energy on the ground state energy surface. ^{9,10} Many intriguing molecular mechanisms such as

dipolar interactions, anharmonicity, and anisotropic alignments have been proposed. ^{11–15} In this paper, we examine the quantum nature of adiabatic reactions in cavities and analyze the resonant, collective, and selective effects of VSC-catalyzed reactions within the framework of multidimensional quantum TST.

VSC Model and Partition Functions. For simplicity, we begin with a single reactive molecule in a single-mode cavity, where the vibrational strong coupling (VSC) is described by the dipole interaction. Then, the Pauli–Fierz Hamiltonian of the cavity quantum electrodynamics (QED) is given as

$$H = H_{sys} + H_{VSC} = \left[\frac{1}{2}p^2 + U(q)\right] + \left[\frac{1}{2}p_c^2 + \frac{1}{2}(\omega_c q_c + \mu(q)A_0)^2\right]$$
(1)

where $\{p, q\}$ are the mass-scaled phase space variables of the reactive system, i.e., $q = \sqrt{mx}$ and $p = \sqrt{mx}$ with mass m, U(q) is the reactive potential, $\{p_o, q_c\}$ are the effective phase space variables of the cavity field, ω_c is the cavity frequency, $\mu(q)$ is the dipole moment, and A_0 is the cavity potential strength. The first term in the square brackets is the system Hamiltonian, H_{sys} , and the second term is the VSC Hamiltonian, H_{VSC} , which includes the dipole self-energy (DSE) term, $(\mu A_0)^2/2$. Here, we assume a

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constant vector potential in the cavity such that, $A(q)=qA_0=q/\sqrt{V\epsilon}$, where V is the cavity volume and ϵ is the permittivity. The QED potential A(q) is a vector and carries the polarization implicitly. In this paper, we adopt a scalar notation, which can be easily translated to the vector format if needed. Further, all phase space variables and Hamiltonians are understood as quantum operators unless otherwise specified.

Interestingly, the form of light—matter interaction in eq 1 follows the same functional form as the Zwanzig Hamiltonian for a system embedded in a Gaussian bath, where the classical thermodynamics of the open system is not affected by the bath. Explicitly, the reduced thermal distribution or, equivalently, the partition function of the open system is given in the classical limit as

$$Z_{sys}^{cl} = \frac{1}{Z_{cav}^{cl}} \int dq \int dp \int dq_c \int dp_c e^{-\beta H^{cl}} = \int dq \int dp e^{-\beta H_{sys}^{cl}}$$
(2)

where the superscript cl denotes the classical limit and Z^{cl}_{cav} is the classical partition function of an empty cavity. Precisely because of the DSE term, the thermal distribution function of the system remains unperturbed by the cavity. Thus, within this classical description, all equilibrium properties, including bond length, dissociation energy, and activation energy, are not altered by VSC. 9,10,19 Yet, this argument becomes more subtle in the quantum version, also known as the Caldeira–Leggett Hamiltonian. In the path integral formalism, the reduced quantum partition function becomes

$$Z_{sys} = \frac{1}{Z_{cav}} \int \mathcal{D}[q_c(\tau)] \int \mathcal{D}[q(\tau)] e^{-\beta H} = \int \mathcal{D}[q(\tau)] e^{-\int_0^\beta H_{sys} d\tau + I[q(\tau)]}$$
(3)

which $I[q(\tau)]$ is the influence functional.²⁰ Physically, the influence functional arises from the quantized cavity field, which dresses the system under VSC, and introduces corrections to equilibrium properties up to second order of light-matter coupling strength, i.e, $(\mu A_0)^2$, which is on the same order of the DSE term. 19 These cavity-induced corrections can be evaluated perturbatively in the VSC regime, as illustrated in this paper. A direct effect of the influence functional is to introduce quantum fluctuations into the correlation function, which manifest in vibrational spectra. These simple observations are consistent with a recent analysis of vibrational polaritons, in particular, regarding the role of the DSE term. 19 For chemical reactions in a cavity, the influence functional modifies the zero-point-energy (ZPE) fluctuations and quantum tunneling effect, which will be calculated in this paper within the framework of quantum transition state theory (TST).

To proceed, we expand the dipole moment around the equilibrium of the molecule potential (i.e., q=0, in the current notation), giving $\mu(q)=\mu_{eq}+\mu'q+\cdots$ where μ_{eq} is the permanent dipole at equilibrium and $\mu'=(\partial\mu/\partial q)_{eq}$ is the gradient of the dipole, i.e., vibrational transition dipole moment. Due to the quadratic form of H_{VSC} , the permanent dipole μ_{eq} term can be removed by shifting the cavity field according to $\omega_c q_c + \mu_{eq} A_0 \rightarrow \omega_c q_c$. Thus, the cavity VSC Hamiltonian becomes independent of the permanent dipole and is rewritten as

$$H_{VSC} = \frac{1}{2}p_c^2 + \frac{1}{2}\omega_c^2 \left(q_c + \frac{\mu'A_0}{\omega_c}q\right)^2 = \frac{1}{2}p_c^2 + \frac{1}{2}\omega_c^2(q_c + gq)^2$$
(4)

where $g = \mu' A_0/\omega_c$ is introduced as a dimensionless parameter to characterize the VSC strength. Since experimental measurements are often reported in terms of the Rabi frequency Ω_R , we

establish the following relation, $\hbar\Omega_R = g\omega_c^2\hbar/\sqrt{\omega\omega_c} = 2\hbar\omega_l\eta$. Here, η is used to quantify the light—matter interaction strength. In the Fabry–Perot cavity, the cavity frequency ω_c is tuned by changing the cavity length while keeping η constant, so we define the coupling constant g as a function of ω_c

$$g = 2\eta \sqrt{\frac{\omega}{\omega_c}} \tag{5}$$

where η and ω are fixed in the current setting.

Quantum TST. A general starting point of the quantum reaction rate is the stationary-phase approximation to the partition function, ^{21,22} giving

$$k_{TST} = \frac{1}{\beta h} \frac{Z_{\ddagger}}{Z_{eq}} \tag{6}$$

where Z_{\ddagger} is the transition state (TS) partition function excluding the unstable mode associated with the reaction coordinate in the reactive barrier region and Z_{eq} is the reactant partition function in the equilibrium well region. We use \ddagger to denote the unstable mode in the reactive barrier region (i.e., TS) and associated quantities. As explained in a comprehensive study, ²² various rate expressions including transition state theory, centroid rate theory, and the instanton solution can be unified under the conceptual framework of eq 6. In this paper, we adopt the standard quantum transition state theory, as presented below.

We begin with the harmonic approximation for a single reaction coordinate and arrive at the quantum transition state theory (TST) expression

$$k_{TST} = \frac{2}{\beta h} \sinh\left(\frac{\omega \beta \hbar}{2}\right) e^{-\beta E_a} \tag{7}$$

where ω is the vibrational frequency in the reactant well and E_a is the activation energy. This is the case for single molecules outside the cavity without the light—matter interaction. Here, quantum tunneling is excluded and will be considered later in the context of centroid TST. The single-mode rate expression is generalized to the multidimensional TST rate k_{TST}^g by introducing the correction factor κ , defined via $k_{TST}^g = \kappa k_{TST}$. The correction factor can be written explicitly as $k_{TST}^g = \kappa k_{TST}^g = \kappa k_{TST}^$

$$\kappa = \frac{\prod_{i} \sinh(\lambda_{i} \beta \hbar/2)}{\sinh(\omega \beta \hbar/2) \prod_{i}^{\ddagger} \sinh(\lambda_{b,i} \beta \hbar/2)}$$
(8)

where λ_i denotes the vibration eigenfrequency in the reactant well, $\lambda_{b,i}$ denotes the vibrational eigenfrequency in the reactive barrier, and \prod^{\ddagger} denotes all these modes excluding the unstable mode. Specifically, in a cavity, the coherent QED field couples individual molecules, modifies the vibrational frequencies, and thus changes the rates, which is quantified by the cavity-induced correction factor κ .

To examine the cavity-induced effects on transition state rate, we rewrite the correction factor in eq 8 as

$$\kappa = \frac{\prod_{i} [1 - \exp(-\lambda_{i} \beta \hbar)]}{[1 - \exp(-\omega \beta \hbar)] \prod_{i}^{\frac{1}{\epsilon}} [1 - \exp(\lambda_{b,i} \beta \hbar)]} \exp(\beta \hbar S/2)$$

$$= \kappa^{*} \exp(\beta \hbar S/2) \tag{9}$$

where S is the frequency shift corresponding to the zero-pointenergy (ZPE) contribution of vibrational modes to the activation energy, given explicitly as

$$S = \sum_{i} \lambda_{i} - \sum_{i}^{\ddagger} \lambda_{b,i} - \omega \tag{10}$$

and κ^* is the modified correction factor excluding the ZPE contribution. In the high-temperature limit, the ZPE contribution can be ignored, and the correction factor reduces to the classical limit

$$\kappa^* \approx \frac{1}{\omega} \frac{\Pi \lambda_i}{\Pi^{\ddagger} \lambda_{b,i}} = \frac{\lambda_{\ddagger}}{\omega_{\ddagger}} = \kappa_{GH}$$
(11)

where ω_{\ddagger} is the unstable frequency without VSC, λ_{\ddagger} is the unstable eigenfrequency under VSC, and their ratio defines the Grote–Hynes (GH) factor κ_{GH} . The GH correction factor can be understood either as a multidimensional effect on canonical TST or as a kinetic caging effect in generalized Langevin dynamics of the reaction coordinate. These two pictures are equivalent, as demonstrated in Pollak's derivation of the Grote–Hynes rate based on the Zwanzig formalism of dissipative dynamics. In the low-temperature limit, the multidimensional connection is dominated by the ZPE contribution and reduces to

$$\kappa_{\text{ZDF}} \approx \exp(\beta \hbar S/2)$$
(12)

In between these two limits, a reasonable approximation is to combine the two limiting expressions, giving $\kappa \approx \kappa_{GH} \exp(\beta \hbar S/2)$, which interpolates between the high- and low-temperature limits. As shown in Figure 1, the VSC-induced correction factor

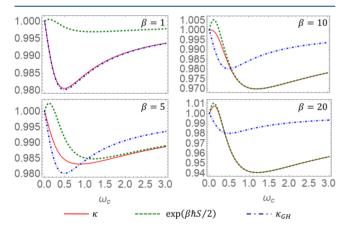


Figure 1. Cavity-induced correction factor κ (solid lines) as a function of ω_c at different inverse temperatures β . The high-temperature limit κ_{GH} (dot-dashed line) and the low-temperature limit κ_{ZPE} (dashed line) are provided for comparison. Relevant parameters are $\eta = \eta_{\frac{1}{2}} = 0.1$ and $\omega_{\frac{1}{2}} = 0.5$. All physical quantities are in units of the vibrational frequency ω .

defined in eq 8 changes from the GH form in eq 11 to the ZPE shift form in eq 12 as the temperature decreases. Finally, to connect with thermodynamics, we can write eq 9 as

$$\kappa \equiv \underbrace{\kappa^*}_{entropy} \exp(\beta \underbrace{\hbar S/2}_{enthalpy}) \equiv \exp(-\beta \Delta G_{\ddagger})$$
(13)

which allows us to identify κ^* as the entropy correction and the ZPE shift as the enthalpy correction and thus defines the cavity-induced free energy change $\Delta G_{\dot{z}}$

Single-Molecule Reaction in Cavity. In the framework of TST, the potential surface U(q) in eq 1 is approximated by a harmonic oscillator in the equilibrium reactant well, giving

$$H = H_s + H_{VSC} = \left[\frac{1}{2}p^2 + \frac{1}{2}\omega^2q^2\right] + \left[\frac{1}{2}p_c^2 + \frac{1}{2}\omega_c^2(q_c + gq)^2\right]$$
(14)

where ω is the vibrational frequency at equilibrium. The quadratic Hamiltonian defines the Hessian matrix

$$\begin{bmatrix} \omega^2 + g^2 \omega_c^2 & g \omega_c^2 \\ g \omega_c^2 & \omega_c^2 \end{bmatrix}$$

where $\omega^2 + g^2\omega_c^2$ is the effective frequency of the reaction coordinate. Diagonalization of the Hessian matrix yields a pair of eigenvalues, λ_+^2 and λ_-^2

$$\lambda_{\pm}^{2} = \frac{1}{2}(\omega^{2} + g^{2}\omega_{c}^{2} + \omega_{c}^{2}) \pm \frac{1}{2}\sqrt{(\omega^{2} + g^{2}\omega_{c}^{2} - \omega_{c}^{2})^{2} + 4g^{2}\omega_{c}^{4}}$$
(15)

In the perturbative regime, the two eigenfrequencies become

$$\lambda_{+} pprox \omega + rac{(g\omega_{c})^{2}\omega}{2(\omega^{2} - \omega_{c}^{2})}$$

$$\lambda_{-} pprox \omega_{c} - \frac{(g\omega_{c})^{2}\omega_{c}}{2(\omega^{2} - \omega_{c}^{2})}$$

which are nearly identical to the exact eigensolution except at resonance $\omega = \omega_c$. The divergence of the above perturbative expansion at the resonance is particularly interesting, as it suggests the largest perturbation due to VSC and thus the maximal cavity-induced correction to the reaction rate [i.e., κ in eq 8], which has been observed experimentally as the resonant effect. Adding these two frequencies, we have

$$\lambda_{+} + \lambda_{-} = \omega + \omega_{c} + \frac{g^{2}\omega_{c}^{2}}{2(\omega + \omega_{c})} + O(g^{4})$$
(16)

such that the overall frequency shift in the well is positive, indicating the increase of the zero-point energy in the reactant due to VSC. The detailed derivation and calibration of the perturbative solution can be found in Sec. I of the Supporting Information (SI).

At the transition state, the potential energy surface can be approximated by a parabolic barrier, giving $U(q) = E_a - 1/2\omega_{\bar{q}}^2q_{\bar{q}}^2$, where $q_{\bar{z}}$ is the mass-scaled barrier coordinate defined as $q_{\bar{z}} = \sqrt{m(x-x_{TS})}$. The Hessian matrix at the barrier can now be written as

$$\begin{bmatrix} -\omega_{\ddagger}^{2} + g_{\ddagger}^{2}\omega_{c}^{2} & g_{\ddagger}\omega_{c}^{2} \\ g_{\ddagger}\omega_{c}^{2} & \omega_{c}^{2} \end{bmatrix}$$

where $\omega_{\ddagger}^2 - g_{\ddagger}^2 \omega$ is the effective barrier frequency. Again, diagonalization of the above matrix yields a pair of eigenvalues, λ_{b+}^2 and $-\lambda_{b-}^2$, which take the same form as eq 16, except for the replacement of ω^2 with $-\omega_{\pm}^2$. i.e.,

$$\pm \lambda_{b\pm}^2 = \frac{1}{2} (\omega_{\epsilon}^2 - \omega_{\ddagger}^2 + g_{\ddagger}^2 \omega_{\epsilon}^2) \pm \frac{1}{2} \sqrt{(\omega_{\ddagger}^2 - g_{\ddagger}^2 \omega_{\epsilon}^2 + \omega_{\epsilon}^2)^2 + 4g_{\ddagger}^2 \omega_{\epsilon}^4}$$
(17)

The corresponding perturbative solutions are given as

$$\lambda_b \equiv \lambda_{b+} \approx \omega_c + \frac{(g_{\ddagger}\omega_c)^2 \omega_c}{2(\omega_{\ddagger}^2 + \omega_c^2)}$$

$$\lambda_{\ddagger} \equiv \lambda_{b-} \approx \omega_{\ddagger} - \frac{(g_{\ddagger}\omega_c)^2 \omega_{\ddagger}}{2(\omega_{\ddagger}^2 + \omega_c^2)}$$

which agree almost perfectly with the exact solution, as shown in Sec. I of the SI. In the barrier region, the stable frequency λ_b increases, whereas the unstable frequency ω_{\ddagger} decreases; both change quadratically with the cavity coupling strength g_{\ddagger} .

For a single reactive molecule in a cavity, the correction factor in eq 8 becomes

$$\kappa = \frac{\sinh\left(\frac{\lambda_{+}\beta}{2}\right)\sinh\left(\frac{\lambda_{-}\beta}{2}\right)}{\sinh\left(\frac{\omega\beta}{2}\right)\sinh\left(\frac{\lambda_{+}\beta}{2}\right)}$$
(18)

The typical frequencies reported experimentally are higher than thermal energy, $\beta\hbar\omega>1$, so the ZPE shift is the dominant contribution. To leading order in g and g_{\ddagger} , the ZPE shift is given as

$$S = \lambda_{+} + \lambda_{-} - \lambda_{b} - \omega = \frac{\omega_{c}^{3}}{2} \left[\frac{g^{2}}{\omega_{c}^{2} + \omega_{c}\omega} - \frac{g_{\pm}^{2}}{\omega_{c}^{2} + \omega_{\pm}^{2}} \right] + O(g^{4})$$
(19)

which is derived in Sec. I of the SI along with other perturbative results. The perturbation expression determines the sign of the frequency shift and consequently the cavity-induced change in the reaction rate

$$S < 0$$
 $e^{\beta \hbar S/2} < 1$ if $\omega \omega_c > \omega_{\ddagger}^2$

$$S > 0$$
 $e^{\beta \hbar S/2} > 1$ if $\omega \omega_c < \omega_{\dot{\tau}}^2$

where $g=g_{\ddagger}$ is assumed. Evidently, for a small cavity frequency, $\omega_c < \omega_{\ddagger}^2/\omega$, the ZPE contribution suppresses the rate; otherwise, for a large cavity frequency, $\omega_c < \omega_{\ddagger}^2/\omega$, the ZPE contribution enhances the rate. This is demonstrated in the shift S plotted in Figure 1, which is positive at small ω_c and negative at large ω_c . For the low cavity frequency, κ is not completely determined by the ZPE shift except at low-temperature $\beta\omega\hbar=20$, so the rate is not necessarily suppressed even when $\omega_c < \omega_{\ddagger}^2/\omega$, as shown in Figure 1. For typical reactive systems, we have $\omega_{\ddagger} < \omega$ and $g_{\ddagger} > g$ such that the correction to the reaction rate is further suppressed in comparison with the special case of $g=g_{\ddagger}$ considered above.

Figure 1 compares the VSC-induced correction factor κ evaluated at different temperatures. We note the change of κ from the high-temperature limit $\kappa \approx \kappa_{GH}$ to the low-temperature limit $\kappa \approx \exp(-\beta \hbar S/2)$ as β increases. Accompanying this change, Figure 1 exhibits the shift of the resonance from the barrier frequency $\omega_c = \omega_{\ddagger}$ at high temperature to near the vibrational resonance $\omega_c \approx \omega$ at the room temperature.

- At high temperature, i.e., $\omega\hbar\beta\ll 1$, the reaction rate approaches the classical limit. Then, according to eqs 9 and 11, the cavity-induced correction reduces to the Grote—Hynes factor, i.e., $\kappa\approx k_{GH}=\lambda_{\ddagger}/\omega_{\ddagger}$. Using the leading order expression of eq 17, we can easily identify the barrier resonant condition $\omega_c=\omega_{\ddagger}^{31}$ where λ_{\ddagger} or κ_{GH} reaches the minimal as a function of cavity frequency (see Sec. I of SI). In addition to the barrier resonance, the classical correction factor is always smaller than unity ($\kappa_{GH}<1$) and is temperature-independent. Evidently, at room temperature, the typical vibrational energy gap is considerably larger than thermal energy; thus, classical TST is not applicable, and a quantum mechanical treatment is essential.
- At intermediate to low temperature, i.e., $\omega\hbar\beta \ge 1$, which is experimentally relevant, we use the exact expression in eq 8 and find the resonance condition $\omega_c \approx \omega$. The

vibrational resonance coincides with the curve crossing in Figure 2(a), where the VSC has the maximal effect on the

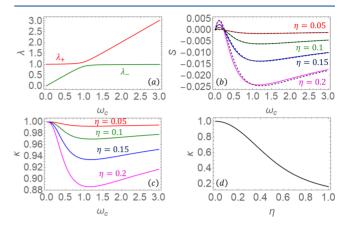


Figure 2. (a) The two eigenfrequencies in the reactant well, λ_+ and λ_- , as a function of cavity frequency ω_c at the VSC strength of $\eta=0.1$. (b) ZPE shift S as a function of ω_c for different values of VSC strength η , along with the perturbation result (dashed lines). (c) κ as a function of ω_c for different values of VSC strength η . (d) κ as a function of η at $\omega_c=1$. All physical quantities are in units of the vibrational frequency ω . Relevant parameters are $\eta_{\pm}=\eta$, $\beta=10$, and $\omega_{\pm}=0.5$.

normal modes. At low temperature, κ is dominated by the ZPE shift in eq 12, so the minimal in κ can be approximately determined by eq 19. In comparison, the experiment measurement of the cavity-induced correction factor is larger, and the measured resonance width is narrower. These discrepancies can arise from the collectivity and quantum tunneling effects, which are

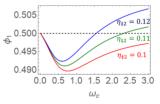


Figure 3. Branching ratio ϕ_1 as a function of ω_ϵ for three different values of $\eta_{\ddagger 2}$. Relevant parameters are $\eta_1 = \eta_{\ddagger 1} = 0.1$, $\omega_{\ddagger 1} = 0.5$, $\omega_{\ddagger 2} = 1.2$, and $\beta = 5$. All physical quantities are in units of the vibrational frequency ω .

not considered in Figures 1-3 but will be addressed in the later part of the paper.

Further, as the VSC strength η increases, the ZPE shift in Figure 2(b) and the correction factor κ in Figure 2(c) become amplified, while its resonance remains close to the vibrational frequency $\omega_c \approx \omega$. In Figure 2(b), the perturbational expression of S in eq 19 is shown in good agreement with the exact solution. Finally, in Figure 2(d), the η -dependence of κ follows the perturbation analysis in eq 19, which predicts $\ln(\kappa) \propto -\eta^2 \beta$.

Mode Selectivity. Quantum control of reaction kinetics has been the holy grail of chemistry. In a remarkable experiment, Ebbesen and his co-workers⁴ have demonstrated that the IR cavity can suppress or enhance a reaction channel relative to another channel and thus achieve chemical selectivity via tuning the cavity frequency. In other words, the VSC of a molecular system can lead to the preferential breaking or formation of one

chemical bond among multiple IR active bonds through the VSC resonance.

Based on the normal-mode analysis of single-mode VSC in this paper, we consider a reaction coordinate with two reactive barriers, one to the left and another to the right of the reactant equilibrium. Then, the branching ratio of the two reactive channels is $\phi_1 = k_1/(k_1 + k_2) = 1/(1 + k_2/k_1)$, which is determined by the ratio of rate constants. Using eq 9, the ratio of the two TST rate constants is given as

$$\frac{k_2}{k_1} = \frac{\kappa_2^*}{\kappa_1^*} \exp(\beta \hbar \Delta S/2 - \beta \Delta E_a) \tag{20}$$

where $\Delta S = S_2 - S_1$ and $\Delta E_a = E_{a2} - E_{a1}$. Since the reactant well is the same for both channels, the branching ratio depends on the nature of the barriers, characterized by three parameters $\{\omega_{\ddagger}, g_{\ddagger}, E_a\}$. Given the large resonant frequency around $\omega \approx 10^3$ cm⁻¹, the cavity-induced correction factor is qualitatively described by the ZPE shift. Then, using the perturbation result in eq 19, the selectivity can be simply estimated by the ZPE difference

$$\hbar \frac{\Delta S}{2} - \Delta E_a = \frac{\hbar \omega_c^3}{4} \left[\frac{g_{\pm 1}^2}{\omega_c^2 + \omega_{\pm 1}^2} - \frac{g_{\pm 2}^2}{\omega_c^2 + \omega_{\pm 2}^2} \right] + (E_{a1} - E_{a2})$$
(21)

which is a function of cavity frequency ω_c .

For simplicity, we assume $E_{a1}=E_{a2}$; then, the limiting values of selectivity as shown in Figure 3 are determined by two ratios: $\omega_{1\ddagger}/\omega_{2\ddagger}$ and $\eta_{1\ddagger}/\eta_{2\ddagger}$. In the limit of small cavity frequency $\omega_c \rightarrow 0$, the sign of eq 21 is determined by $\eta_{\ddagger\,1}/\omega_{\ddagger 1} - \eta_{\ddagger 2}/\omega_{\ddagger 2}$, which is negative for the parameters in Figure 3, so the branching ratio favors channel 2, i.e., $\phi_1 < 50\%$. For a large cavity frequency, the sign of eq 21 is determined by $\eta_{1\ddagger} - \eta_{2\ddagger}$. Thus, in Figure 3, for $\eta_{2\ddagger} = 0.12 > \eta_{1\ddagger} = 0.1$ and $\eta_{2\ddagger} = 0.11 > \eta_{1\ddagger} = 0.1$, the branching ratio favors channel 1, i.e., $\phi_1 > 50\%$ at a large cavity frequency, and exhibits a switch of selectivity as a function of ω_c . In comparison, the curve with $\eta_{2\ddagger} = \eta_{1\ddagger} = 0.1$ always stays below 50% as the carrier frequency is tuned. In between the two limits, the branching ratio exhibits a maximal reduction between the two barrier frequencies, $[\omega_{1\ddagger}, \omega_{2\ddagger}]$.

Despite its simplicity, the single-mode model presented above confirms the possibility of cavity-enabled selectivity of chemical reactions. To go beyond the single-mode picture, one encounters a more interesting problem of the nonlinear coupling between molecular modes in a cavity, which bears similarity to the solvated ABA model.³² The correlated mode couplings in the ABA model may shed light on the VSC-induced cooperativity in intramolecular vibrational relaxation (IVR), energy transfer, and reaction kinetics and will be a subject for future study.

Collectivity and N-Dependence. Experimental measurements demonstrate that both the Rabi frequency and cavity-induced correction increase with the molecular density. This collective effect has inspired much theoretical interest. ^{33,34} Within the TST framework, individual molecules are thermally activated without any coherence among molecules at the transition state such that the cavity correction κ does not exhibit strong dependence on the molecular density. In contrast, in the coherent picture, the cavity creates coherent polariton states, which are thermally activated to react. Thus, the experimentally observed scaling with molecular density can be easily explained

using the *N*-scaled VSC coupling strength. Here, we first demonstrate the *N*-independence in the incoherent TST and then contrast it with the *N*-scaling in the coherent version.

To begin, we consider the N-particle Hamiltonian in a cavity

$$\begin{split} H &= H_{sys} + H_{VSC} \\ &= \left[\sum_{n=1}^{N} \frac{1}{2} p_{n}^{2} + U_{N}(q_{1}, \, \cdots, \, q_{N}) \right] + \left[\frac{1}{2} p_{c}^{2} + \frac{1}{2} \omega_{c}^{2} \left(q_{c} + g \sum_{n=1}^{N} q_{n} \right)^{2} \right] \end{split} \tag{22}$$

where index n is the particle index in the cavity and U_N is the N-particle potential. Here, we assume N noninteracting, identical molecules in the cavity. In the reactant well, we adopt the harmonic approximation, $U_N = \sum_{n=1}^N \omega^2 q_n^2/2$, so that the VSC polariton is described by the Hessian matrix

$$\begin{bmatrix} \omega^2 + g_N^2 \omega_c^2 & g_N \omega_c^2 \\ g_N \omega_c^2 & \omega_c^2 \end{bmatrix}$$

where $g_N = \sqrt{Ng}$ is the collective VSC constant for homogeneous coupling and can be easily generalized to the case of inhomogeneous coupling. The Hessian matrix yields a pair of polariton frequencies, λ_{\pm} , and the remaining N-1 modes are dark states with the unperturbed frequency ω .

In the incoherent TST picture, one reactive molecule (e.g., n=1) is thermally activated to the transition state, while the other N-1 molecules remain in equilibrium, so the TS potential can be approximated as $U_N=E_a-1/2\omega_{\frac{3}{2}}^2q_{\frac{3}{2}}^2+\sum_{n=2}^N 1/2\omega^2q_n^2$. The corresponding Hessian matrix becomes

$$\begin{bmatrix} -\omega_{\ddagger}^{2} + g_{\ddagger}^{2}\omega_{c}^{2} & g_{N-1}g_{\ddagger}\omega_{c}^{2} & g_{\ddagger}\omega_{c}^{2} \\ g_{N-1}g_{\ddagger}\omega_{c}^{2} & \omega^{2} + g_{N-1}^{2}\omega_{c}^{2} & g_{N-1}\omega_{c}^{2} \\ g_{\ddagger}\omega_{c}^{2} & g_{N-1}\omega_{c}^{2} & \omega_{c}^{2} \end{bmatrix}$$

where the collective coupling is $g_{N-1}=g\sqrt{N-1}$. The eigensolution of the above Hessian matrix yields three eigenvalues: two real frequencies, $\lambda_{b\pm}^2$, and one imaginary frequency, $-\lambda_{\pm}^2$. The remaining N-2 modes are dark states with the equilibrium frequency ω . These eigenvalues are evaluated in Sec. II of the SI and yield the cavity-induced correction

$$\kappa = \frac{\sinh\left(\frac{\lambda_{+}\beta}{2}\right)\sinh\left(\frac{\lambda_{-}\beta}{2}\right)}{\sinh\left(\frac{\lambda_{b+}\beta}{2}\right)\sinh\left(\frac{\lambda_{b-}\beta}{2}\right)}$$
(23)

Surprisingly, as shown in the SI, to the leading order of g^2 , the energy shift S(N) and barrier frequency $\lambda_{\ddagger}(N)$ of the N-particle system are identical to their corresponding values in the single-molecule case, giving explicitly

$$S(N) = \frac{\omega_c^3}{2} \left[\frac{g^2}{\omega_c^2 + \omega_c \omega} - \frac{g_{\ddagger}^2}{\omega_c^2 + \omega_{\ddagger}^2} \right] + O(Ng^4)$$

$$\lambda_{\ddagger}(N) = \omega_{\ddagger} \left[1 - \frac{(g_{\ddagger}\omega_c)^2}{2(\omega_{\ddagger}^2 + \omega_c^2)} \right] + O(Ng^4)$$

where the first terms are exactly the simple molecule results and the next order corrections are in terms of Ng^4 . Since there is no N-dependence in both the high-T and low-T limits of κ , we

expect weak N-dependence in incoherent TST. This prediction is confirmed in Sec. II of the SI and in Figure 4(a), where κ exhibits almost no N-dependence.

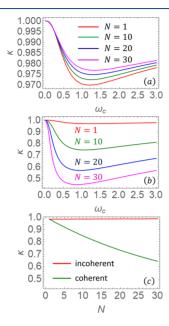


Figure 4. (a) κ calculated in incoherent TST as a function of ω_c for different N with $\eta=\eta_{\ddagger}=0.1$. (b) κ calculated in coherent TST as a function of ω_c for different N with corresponding VSC strength $\eta=\eta_{\ddagger}=0.1\sqrt{N}$. (c) Comparison of coherent and incoherent κ as a function of N at $\omega_c=1$. All physical quantities are in units of the vibrational frequency ω_t , and the other relevant parameters are $\omega_{\ddagger}=0.5$ and $\beta=10$.

In the coherent TST picture, the polariton is activated collectively to the transition state. The resulting TS potential is

$$U_N = E_a - \frac{1}{2}\omega_{\ddagger}^2 q_{\ddagger s}^2 + \sum_{i=2}^N \frac{1}{2}\omega^2 q_i^2$$
 (24)

where q_i is the *i*-th dark state and $q_{\frac{1}{2}s}$ is the bright state at the barrier $q_{\frac{1}{2}s} = \sum_n \sqrt{m_n} (x_n - x_{TS}) / \sqrt{N}$. Within the harmonic approximation, the above potential and the original molecular potential in eq 22 are related via a linear transformation and are thus equivalent. The Hessian matrix at the TS is then given as

$$\begin{bmatrix} -\omega_{\ddagger}^{2} + g_{\ddagger N}^{2}\omega_{c}^{2} & g_{\ddagger N}\omega_{c}^{2} \\ g_{\ddagger N}\omega_{c}^{2} & \omega_{c}^{2} \end{bmatrix}$$

with $g_{\ddagger N} = g_{\ddagger} \sqrt{N}$. Therefore, all the results can be carried over from single-molecule reactions to N-particle coherent reactions if the VSC constant g is scaled up by a factor of \sqrt{N} , $g \rightarrow g \sqrt{N}$, which is exactly the same N-dependence for the Rabi frequency. To illustrate the coherent scaling, Figure 4(b) shows the dramatic enhancement of the cavity effect as N increases. Figure 4(c) compares the coherent and incoherent TST correction factors as a function of N and clearly shows the linear scaling with N (equivalently, Ω_R^2) in the coherent rate constant.

In coherent polariton theory, the collective VSC strength and cavity-induced correction can be enhanced as N increases. This type of cooperativity has been studied for nonadiabatic reactions in a cavity $^{33-36}$ but not for adiabatic reactions on the ground-state surface. In reality, N should be interpreted as the number of coherently coupled molecules, which is limited by dynamic and

static disorder. $^{37-39}$ Thus, the N-dependence in coherent reactions and other collective dynamics should be renormalized by localization and polaron effects. 40,41

Quantum Tunneling. As explained in eqs 8 and 11, the Grote—Hynes correction is a classical effect, and the ZPE shift is a quantum effect. Another contribution is quantum tunneling at the reactive barrier, which further enhances the cavity effect. A simple way to account for the tunneling contribution is centroid TST, ⁴² which is deduced from the stationary-phase evaluation of the barrier partition function in eq 6 using the centroid variable, ²² giving

$$k_{c-TST} = \frac{(\omega_{\ddagger}\hbar\beta/2)}{\sin(\omega_{\ddagger}\hbar\beta/2)} k_{TST} = \frac{\omega_{\ddagger}}{2\pi} \frac{\sinh(\omega\hbar\beta/2)}{\sin(\omega_{\ddagger}\hbar\beta/2)} e^{-\beta E_a}$$
(25)

where ω_{\ddagger} and ω are the barrier and reactant frequencies, respectively. The centroid TST rate is the same as quantum TST with the tunneling correction for a parabolic barrier. ^{22,24} With cavity VSC, the centroid rate becomes $k_{c-TST}^g = \kappa_{centroid} k_{c-TST}$, where the cavity correction factor in eq 8 becomes

$$\kappa_{centroid} = \frac{\lambda_{\ddagger}}{\omega_{\ddagger}} \frac{\sin(\omega_{\ddagger}\hbar\beta/2)}{\sin(\lambda_{\ddagger}\hbar\beta/2)} \kappa \tag{26}$$

where κ has been evaluated previously. Figure 5 shows $\kappa_{centroid}$ calculated from eq 26 at several temperatures. In comparison

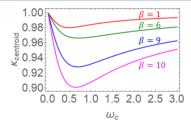


Figure 5. Cavity-induced correction factor $\kappa_{centroid}$ defined in centroid TST as a function of ω_c at different temperatures. All physical quantities are in units of the vibrational frequency ω , and other relevant parameters are $\eta = \eta_{\dot{\tau}} = 0.1$ and $\omega_{\dot{\tau}} = 0.5$.

with eq 8 and Figure 1, quantum tunneling dramatically enhances the cavity-induced correction and can be combined with the collective effect to improve the quantitative agreement with experimental measurements.

Interestingly, the prefactor of the centroid TST diverges as $\omega_{\pm}\hbar\beta_{c} = 2\pi$, which suggests the failure of TST at or below the crossover temerature T_c . An easy fix is to replace the anharmonic well and barrier potentials with the optimal quadratic free energy potentials determined by the variational principle. 28,43 More generally, in the deep tunneling regime (i.e., $T < T_c$), quantum fluctuations dominate and the concept of "transition state" based on a local reaction coordinate is no longer valid. Instead, we should adopt the picture of "coherent tunneling", which is delocalized and involves the full anharmonic potential. One approach is based on the concept of the "instanton", which is a nontrivial stationary solution to the barrier partition function at $T < T_c$ and predicts the quantum tunneling rate. ^{22,44} As stated earlier, the cavity field can be treated as a single-mode harmonic bath and thus introduces an influence functional in the path integral expression of the system partition function [see eq 3]. In a previous study, the harmonic bath effect on the tunneling rate was found to reduce the instanton rate as the system-cavity coupling increases, 22,45 consistent with the experimental measurements in optical cavities. Further, quantum tunneling

can permit a natural justification of the coherence and collectivity as described in eq 24.

In summary, as the typical vibration and cavity frequencies (10³ cm⁻¹) are considerably larger than thermal energy, it is essential to adopt a quantum description of ground-state chemical reactions in a cavity. Within the framework of quantum TST, we are able to account for zero-point-energy shift and quantum tunneling induced by VSC and demonstrate the resonant effect, collective effect, and selectivity as well as various parametric dependences as summarized below. Based on the relatively simple TST calculations, our predictions are consistent with experiments. Much of the reported study is built on perturbative normal-mode analysis, which not only provides physical insights into cavity-catalyzed chemical reactions but also presents a useful tool to treat other VSC effects.

- (1) Dipole self-energy term and thermal equilibrium: The DSE term introduces additional restoring forces along the reaction coordinate and effectively increases the vibrational frequency. According to the perturbation analysis, the DSE contribution is on the same order as the linear dipole term in the PF Hamiltonian and thus cannot be ignored in the VSC regime. As a result of the Zwanzig form of the VSC Hamiltonian, the equilibrium thermodynamics of the cavity-dressed system exhibits no VSC perturbation in the classical regime (see eq 2) and a temporal correlation in the quantum regime (see eq 3). This explains the weak cavity effects on equilibrium quantities including the bond length, disassociation energy, and activation energy. For the same reason, the permanent dipole does not contribute to the thermal reaction rate, and the leading order is the linear transition dipole moment (see eq 4).
- (2) Cavity-catalyzed single-molecule reactions: (i) The cavity-induced correction can be attributed to the ZPE shift quantified by S and the kinetic effect quantified by κ^* , which can be related to enthalpy and entropy changes, respectively see eq 13). In the experimentally relevant regime, the dominant contribution arises from the ZPE shift except for small cavity frequency. (ii) The sign of ZPE shift in eq 19 predicts the suppression or enhancement of the reaction rate, and the amplitude of the shift depends on the VSC strength or the Rabi frequency quadratically. (iii) With typical parameters, the cavity-induced correction exhibits a maximal suppression when the cavity frequency is near the vibrational resonance frequency.
- (3) Mode selectivity: VSC can either enhance or suppress the branching ratio in a multiple-channel reaction, thus achieving mode selectivity in a cavity. Within a single-mode two-barrier model, the branching ratio of the two competing reactions is determined by the relative free energy difference of the two reactive barriers, i.e., eq 21, characterized by three parameters $\{\omega_{\ddagger}, g_{\ddagger}, E_a\}$. Perturbation analysis establishes the functional-dependence of the branching ratio on these parameters and suggests the reordering of preferential selectivity as the cavity frequency ω_c is tuned.
- (4) Collectivity and N-scaling: In the TST framework, individual molecules are thermally activated without any coherence between molecules at the transition state. Within this incoherent picture, to leading order of VSC strength, the frequency shift and correction factor are

- identical to the single-molecule results and are thus N-independent. In contrast, in the coherent picture, the polariton state is thermally activated to yield collective barrier crossing. Thus, the experimentally observed collectivity can be easily explained by rescaling the single-molecule VSC strength, $g(N) = \sqrt{Ng}$.
- (5) Quantum tunneling: The centroid TST is used to account for quantum tunneling at the reactive barrier for $T > T_c$ and is shown to enhance the VSC-induced correction considerably. Yet, the centroid correction diverges at the crossover temperature T_c , which indicates the failure of TST at $T \le T_c$ and the emergence of coherent tunneling as described in the instanton picture, where under-barrier tunneling dominates over thermal activation.

Chemical reactions are complex, involving cohesive motion of many degrees of freedom in a thermal environment. Theoretical models based on normal-mode analysis allow for analytical solutions and provide useful insight into this complex process. Part of the simplicity arises from the harmonic form of the cavity field, which introduces a tunable bosonic bath mode. Therefore, many emerging phenomena in polariton chemistry to an be understood with the combination of quantum simulations and analytical solutions.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.jpclett.1c02210.

S1. Perturbation Analysis: Single-Molecule VSC; S2. Perturbation Analysis: N-Molecule VSC (PDF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) Hutchison, J. A.; Schwartz, T.; Genet, C.; Devaux, E.; Ebbesen, T. W. Modifying chemical landscapes by coupling to vacuum fields. *Angew. Chem., Int. Ed.* **2012**, *51*, 1592.
- (2) Thomas, A.; George, J.; Shalabney, A.; Dryzhakov, M.; Varma, S. J.; Moran, J.; Chervy, T.; Zhong, X.; Devaux, E.; Genet, C.; et al. Ground-state chemical reactivity under vibrational coupling to the vacuum electromagnetic field. *Angew. Chem.* **2016**, *128*, 11634–11638. (3) Ebbesen, T. W. Hybrid light—matter states in a molecular and
- (3) Ebbesen, T. W. Hybrid light-matter states in a molecular ar material science perspective. *Acc. Chem. Res.* **2016**, *49*, 2403.

- (4) Thomas, A.; Lethuillier-Karl, L.; Nagarajan, K.; Vergauwe, R. M.; George, J.; Chervy, T.; Shalabney, A.; Devaux, E.; Genet, C.; Moran, J.; et al. Tilting a ground-state reactivity landscape by vibrational strong coupling. *Science* **2019**, *363*, *615*.
- (5) Lather, J.; Bhatt, P.; Thomas, A.; Ebbesen, T. W.; George, J. Cavity catalysis by cooperative vibrational strong coupling of reactant and solvent molecules. *Angew. Chem., Int. Ed.* **2019**, *58*, 10635.
- (6) Lather, J.; George, J. Improving enzyme catalytic efficiency by cooperative vibrational strong coupling of water. *J. Phys. Chem. Lett.* **2021**, *12*, 379–384.
- (7) Hirai, K.; Hutchison, J. A.; Uji-i, H. Recent progress of vibropolaritonic chemistry. *ChemPlusChem* **2020**, *85*, 1981–1988.
- (8) Hirai, K.; Takeda, R.; Hutchison, J. A.; Uji-i, H. Modulation of Prins cyclization by vibrational strong coupling. *Angew. Chem., Int. Ed.* **2020**, *59*, 5332–5335.
- (9) Li, T. E.; Nitzan, A.; Subotnik, J. E. On the origin of ground-state vacuum-field catalysis: Equilibrium consideration. *J. Chem. Phys.* **2020**, *152*, 234107.
- (10) Campos-Gonzalez-Angulo, J. A.; Yuen-Zhou, J. Polaritonic normal modes in transition state theory. *J. Chem. Phys.* **2020**, *152*, 161101.
- (11) Galego, J.; Climent, C.; Garcia-Vidal, F. J.; Feist, J. Cavity Casimir-Polder Forces and Their Effects in Ground-State Chemical Reactivity. *Phys. Rev. X* **2019**, *9*, 021057.
- (12) Climent, C.; Galego, J.; Garcia-Vidal, F. J.; Feist, J. Plasmonic Nanocavities Enable Self-Induced Electrostatic Catalysis. *Angew. Chem., Int. Ed.* **2019**, *58*, 8698–8702.
- (13) Triana, J. F.; Hernández, F. J.; Herrera, F. The shape of the electric dipole function determines the sub-picosecond dynamics of anharmonic vibrational polaritons. *J. Chem. Phys.* **2020**, *152*, 234111.
- (14) Vurgaftman, I.; Simpkins, B. S.; Dunkelberger, A. D.; Owrutsky, J. C. Negligible effect of vibrational polaritons on chemical reaction rates via the density of states pathway. *J. Phys. Chem. Lett.* **2020**, *11*, 3557–3562.
- (15) Schäfer, C.; Flick, J.; Ronca, E.; Narang, P.; Rubio, A. Shining Light on the Microscopic Resonant Mechanism Responsible for Cavity-Mediated Chemical Reactivity; 2021, arXiv preprint, arXiv:2104.12429.
- (16) Flick, J.; Ruggenthaler, M.; Appel, H.; Rubio, A. Atoms and molecules in cavities, from weak to strong coupling in quantum-electrodynamics (qed) chemistry. *Proc. Natl. Acad. Sci. U. S. A.* **2017**, *114*, 3026–3034.
- (17) Power, E. A.; Zienau, S. Coulomb gauge in non-relativistic quantum electro-dynamics and the shape of spectral lines. *Philos. Trans. R. Soc. London A* **1959**, 251, 427.
- (18) Zwanzig, R. Nonlinear generalized Langevin equations. *J. Stat. Phys.* **1973**, *9*, 215–220.
- (19) Fischer, E. W.; Saalfrank, P. Ground state properties and infrared spectra of anharmonic vibrational polaritons of small molecules in cavities. *J. Chem. Phys.* **2021**, *154*, 104311.
- (20) Feynman, R. P.; Vernon, F. L. The theory of a general quantum system interacting with a linear dissipative system. *Ann. Phys.* **2000**, *281*, 547–607.
- (21) Hanggi, P.; Talkner, P.; Borkovec, M. Reaction-rate theory: Fifty years after Kramers. *Rev. Mod. Phys.* **1990**, *62*, 251.
- (22) Cao, J.; Voth, G. A. A unified framework for quantum activated rate processes: I. General theory. J. Chem. Phys. 1996, 105, 6856.
- (23) Wolynes, P. G. Quantum theory of activated events in condensed phases. *Phys. Rev. Lett.* **1981**, 47, 968.
- (24) Pollak, E. Transition state theory for quantum decay rates in dissipative systems: the high-temperature limit. *Chem. Phys. Lett.* **1986**, 127, 178.
- (25) Grote, R. F.; Hynes, J. T. The stable states picture of chemical reactions. II. rate constants for condensed and gas phase reaction models. *J. Chem. Phys.* **1980**, *73*, 2715.
- (26) Truhlar, D. G.; Garrett, B. C. Multidimensional transition state theory and the validity of Grote-Hynes theory. *J. Phys. Chem. B* **2000**, *104*, 1069.

- (27) Nitzan, A. Chemical Dynamics in Condensed Phases: Relaxation, Transfer and Reactions in Condensed Molecular Systems; Oxford University Press: Oxford, U.K., 2006.
- (28) Cao, J.; Voth, G. A. A theory for quantum activated rate constants in dissipative systems. *Chem. Phys. Lett.* **1996**, *261*, 111.
- (29) Pollak, E. Theory of activated rate processes: A new derivation of Kramers' expression. *J. Chem. Phys.* **1986**, *85*, 865.
- (30) Pollak, E.; Tucker, S. C.; Berne, B. J. Variational transition-state theory for reaction rates in dissipative systems. *Phys. Rev. Lett.* **1990**, *65*, 1399
- (31) Li, X.; Mandal, A.; Huo, P. Cavity frequency-dependent theory for vibrational polariton chemistry. *Nat. Commun.* **2021**, *12*, 1315.
- (32) Kryvohuz, M.; Cao, J. Noise-induced dynamics symmetry breaking and stochastic transitions in ABA molecules: II. Symmetricantisymmetic normal mode switching. *Chem. Phys.* **2010**, *370*, 258.
- (33) Gu, B.; Mukamel, S. Cooperative conical intersection dynamics of two pyrazine molecules in an optical cavity. *J. Phys. Chem. Lett.* **2020**, 11. 5555.
- (34) Campos-Gonzalez-Angulo, J. A.; Ribeiro, R. F.; Yuen-Zhou, J. Resonant catalysis of thermally activated chemical reactions with vibrational polaritons. *Nat. Commun.* **2019**, *10*, 4685.
- (35) Galego, J.; Garcia-Vidal, F. J.; Feist, J. Many molecule reaction triggered by a single photon in polaritonic chemistry. *Phys. Rev. Lett.* **2017**, *119*, 136001.
- (36) Phuc, N. T.; Ishizaki, A.; Trung, P. Q. Controlling the electron-transfer reaction rate through molecular-vibration polaritons in the ultrastrong coupling regime. *Sci. Rep.* **2020**, *10*, 7318.
- (37) Moix, J.; Zhao, Y.; Cao, J. Equilibrium-reduced density matrix formulation: Influence of noise, disorder, and temperature on localization in excitonic systems. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2012**, 85, 115412.
- (38) Herrera, F.; Spano, F. C. Cavity-controlled chemistry in molecular ensembles. *Phys. Rev. Lett.* **2016**, *116*, 238301.
- (39) Scholes, G. D.; DelPo, C. A.; Kudisch, B. Entropy reorders polariton states. J. Phys. Chem. Lett. 2020, T11, 6389.
- (40) Lee, C. K.; Moix, J. M.; Cao, J. Coherent quantum transport in disordered systems: A unified polaron treatment of hopping and bandlike transport. *J. Chem. Phys.* **2015**, *142*, 164103.
- (41) Wersäll, M.; Munkhbat, B.; Baranov, D.; Herrera, F.; Cao, J.; Antosiewicz, T. J.; Shegai, T. Correlative Dark-Field and Photoluminescence Spectroscopy of Individual Plasmon-Molecule Hybrid Nanostructures in a Strong Coupling Regime. *ACS Photonics* **2019**, *6*, 2570–2576.
- (42) Voth, G. A.; Chandler, D.; Miller, W. H. Rigorous formulation of quantum transition state theory and its dynamical corrections. *J. Chem. Phys.* **1989**, *91*, 7749–7760.
- (43) Cao, J.; Voth, G. A. Modeling physical systems by effective harmonic oscillators: The optimized quadratic approximation. *J. Chem. Phys.* **1995**, *102*, 3337.
- (44) Miller, W. H. Semiclassical limit of quantum mechanical transition state theory for non-separable systems. *J. Chem. Phys.* **1975**, 62, 1899–1906.
- (45) Cao, J.; Minichino, C.; Voth, G. A. The computation of electron transfer rates: The nonadiabatic instanton solution. *J. Chem. Phys.* **1995**, *103*, 1391.
- (46) Garcia-Vidal, F.; Ciuti, C.; Ebbesen, T. Manipulating matter by strong coupling to vacuum fields. *Science* **2021**, *373* (6551), eabd0336.