

# **Salt deliquescence along boulder cracks in the Antarctic Dry Valleys: An overlooked source of moisture for rock weathering processes**

1 M. Ben-Asher<sup>1,2</sup>, A. Mushkin<sup>2,6</sup>, N. Lensky<sup>2,7</sup>, R. Amit<sup>2</sup>, M. C. Eppes<sup>3</sup>, D. D. Ming<sup>4</sup>, E. Shelef<sup>5</sup>, R. S.  
2 Sletten<sup>6</sup>.

3 <sup>1</sup>EDYTEM laboratory, Université Savoie Mont Blanc, CNRS, Le Bourget-du-Lac, 73376, France

4 <sup>2</sup>Geological Survey of Israel, Jerusalem 9550161, Israel

5 <sup>3</sup>Department of Geography and Earth Sciences, University of North Carolina at Charlotte, Charlotte,  
6 North Carolina 28223, USA

7 <sup>4</sup>Johnson Space Center, Houston, TX, USA.

8 <sup>5</sup>Geology and Environmental Science, University of Pittsburgh, Pittsburgh, Pennsylvania, USA.

9 <sup>6</sup>Department of Earth & Space Sciences, University of Washington, Seattle, WA, 98195, USA

10 <sup>7</sup>Institute of Earth Sciences, The Hebrew University of Jerusalem, Jerusalem, Israel.

11 Corresponding author: Matan Ben-Asher (matan.ben-asher@univ-smb.fr)

## **12 Keywords**

13 Salt deliquescence; subcritical cracking; salt shattering; Antarctic Dry Valleys.

## Abstract

Cracking is a primary rock-weathering mechanism in arid environments, where dry conditions typically limit the efficacy of water-driven weathering processes. Here, we present results from a field-based experiment in the hyper-arid and frigid Antarctic Dry Valleys (ADV) that documented recurring periods of transient accumulation of liquid water along rock cracks during otherwise dry conditions. This moisture was likely sourced from the deliquescence of hygroscopic salts during sub-saturated humidity conditions. Analysis of meteorological data from 17 stations scattered throughout the ADV revealed that near-surface atmospheric conditions across one of Earth's driest environments can annually support tens of such deliquescence-efflorescence cycles of hygroscopic salts, e.g.,  $\text{CaCl}_2$ ,  $\text{NaNO}_3$ ,  $\text{NaCl}$ , and  $\text{MgCl}_2$ . This deliquesced moisture may have an important role in the cracking processes of ADV rocks. In a broader context, the results from the ADV suggest that deliquesced atmospheric humidity may be an overlooked source of moisture available for rock weathering processes in otherwise extremely dry deserts on Earth and possibly Mars.

## 27 1 Introduction

### 28 1.1 Rock weathering in hyper-arid environments

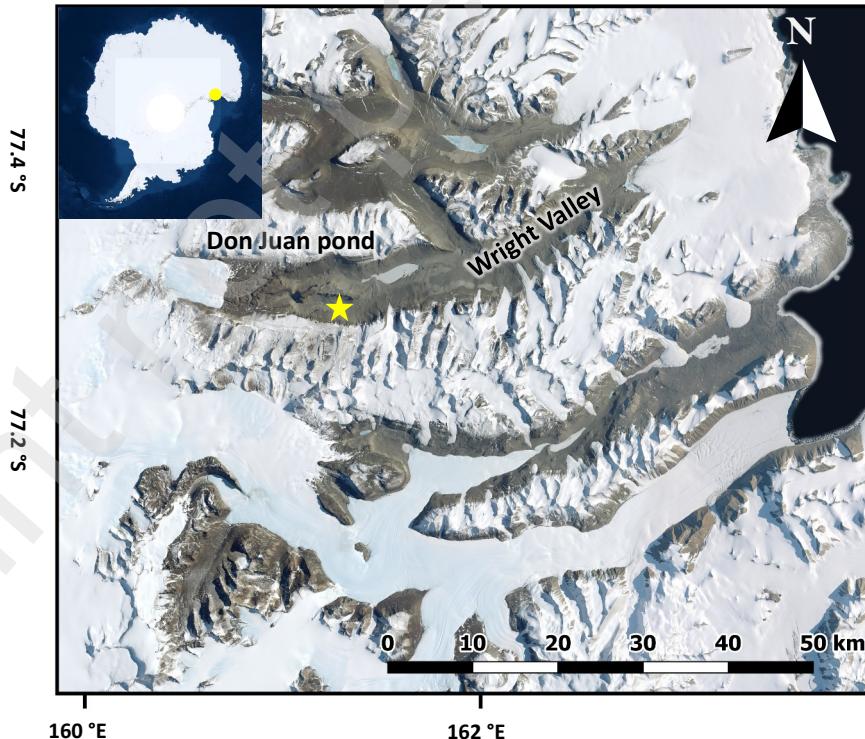
Rock weathering is broadly regarded as a key and often rate-limiting process in the subsequent evolution of terrestrial landscapes. In hyperarid environments, physical disintegration, i.e., the breakup of rocks through cracking, is typically a dominant mode of weathering because the characteristic dry conditions in such settings limit the efficacy of water-dependent chemical, biological, and frost weathering mechanisms (Cooke, 1981; Cooke & Smalley, 1968). When rocks or other brittle materials

34 are subjected to low stresses, cracks can propagate subcritically (Anderson, 2005; Atkinson, 1984).  
35 Natural stresses in arid climates are likely dominantly subcritical in magnitude and commonly attributed  
36 to repeated cycles of thermal expansion/contraction in response to diurnal insolation dynamics (e.g.,  
37 McFadden et al., 2005) or salt weathering (Desarnaud et al., 2016; Sperling & Cooke, 1985; Steiger et  
38 al., 2008; Winkler & Wilhelm, 1970). Nonetheless, laboratory, as well as field-based studies, have  
39 shown that even a slight increase in moisture can significantly increase the rates at which these  
40 otherwise ‘dry’ rock-cracking mechanisms can operate (Eppes et al., 2020; Eppes & Keanini, 2017;  
41 Meredith & Atkinson, 1985; Yoshitaka Nara et al., 2010, 2012; Waza et al., 1980). Here, we present  
42 field-based evidence from one of Earth’s driest and coldest deserts that the deliquescence of  
43 atmospheric humidity by hygroscopic salts is an effective pathway for water delivery to rock cracks.  
44 This moisture delivery pathway to rock cracks may have an important and previously overlooked pace-  
45 setting role in the cracking process of rocks in hyper-arid environments.

## 46 1.2 The Antarctic Dry Valleys

47 The Antarctic Dry Valleys (ADV) (Fig. 1) are amongst the coldest and driest ‘ice-free’ regions  
48 on Earth (Doran et al., 2002; Fountain et al., 2010; Obryk et al., 2020). The mean annual air  
49 temperatures on the valleys floors range between  $-15^{\circ}\text{C}$  and  $-30^{\circ}\text{C}$ , depending on the location (Obryk  
50 et al., 2020), and precipitation is limited to less than 50 mm/yr that occurs primarily as snowfall  
51 (Fountain et al., 2010). These hyperarid and frigid conditions have prevailed in the ADV since the Plio-  
52 Pleistocene (Fielding et al., 2011; Scopelliti et al., 2013), resulting in one of the slowest eroding  
53 landscapes on Earth with estimated bedrock erosion rates below 1 m/m.y. (Balco & Shuster, 2009;  
54 Brook et al., 1995; Margerison et al., 2005; Marrero et al., 2018; Staiger et al., 2006; Sugden et al.,

55 1999; Summerfield et al., 1999). As such, the ADV environment is also regarded as a prime analog site  
56 for the present-day hyperarid and cold surface conditions on Mars (Head & Marchant, 2014; Sletten et  
57 al., 2003; Tamppari et al., 2012). Rock weathering processes in the ADV have been previously  
58 attributed to thermal stress mechanisms (Campbell & Claridge, 1987; Hall, 1999; Lamp et al., 2017) or  
59 to hygroscopic salts (Campbell & Claridge, 1987; Johnston, 1973; Selby & Wilson, 1971; Wellman &  
60 Wilson, 1965), that accumulate at or near the ADV surface due to the extremely dry conditions (Bisson  
61 et al., 2015; Keys, 1979; Keys & Williams, 1981). The presence of up to ~30 salt phases was previously  
62 reported in the ADV soils, including hygroscopic salts, such as NaCl, MgCl<sub>2</sub>, NaNO<sub>3</sub>, and CaCl<sub>2</sub> (Bisson  
63 et al., 2015; Claridge & Campbell, 1977; Goudie & Cooke, 1984; Keys, 1979; Keys & Williams, 1981;  
64 Miotke & von Hodenberg, 1983; Tamppari et al., 2012; Wilson, 1979).



65 **Figure 1: Annotated satellite image of the Antarctic Dry Valleys. Yellow star marks the field site near Don**  
66 **Juan pond in Wright valley. The satellite image was obtained through the QuickMapServices QGIS**  
67 **plugin, from ESRI server (ArcGIS/World\_Imagery).**

68 **1.3 Salt deliquescence in the ADV**

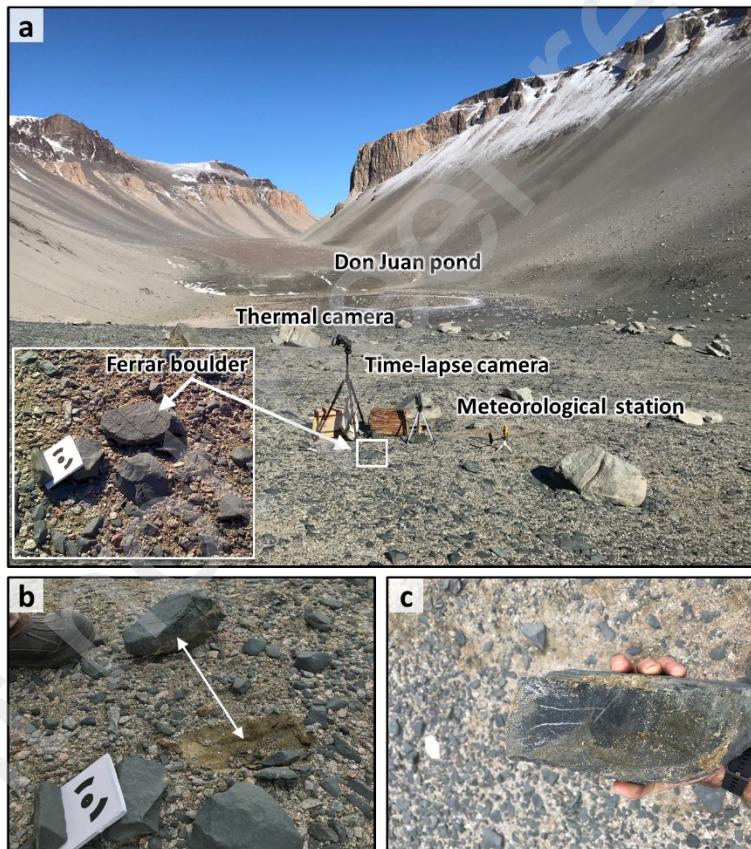
69 Deliquescence occurs when the relative humidity (RH) of the air mass exceeds the  
70 deliquescence relative humidity (DRH) of a specific salt or a salt mixture, and atmospheric water is  
71 absorbed and forms a brine that can further adsorb water. Efflorescence is the reverse process that  
72 occurs when relative humidity is reduced below the efflorescence relative humidity (ERH) and  
73 recrystallization occurs. Salt deliquescence/efflorescence dynamics have been previously documented  
74 in the ADV soils as the appearance of transient ‘wet patches’ (Gough et al., 2016; Harris & Cartwright,  
75 1981; Head et al., 2007; Levy, 2021; Toner et al., 2022) or ‘wet slope streaks’ (Toner et al., 2022)  
76 during events of increased atmospheric humidity. The present study tests whether and how such  
77 deliquescence/efflorescence dynamics can facilitate rock-cracking processes in the ADV.

78 **2 Methods**

79 **2.1 Field experiment and laboratory analyses**

80 A field-based 10-day experiment was performed using local meteorological measurements  
81 (Kestrel 5500), time-lapse photography (Brinno TLC 200), and thermal imaging (FLIR SC430) of a  
82 Ferrar Dolerite boulder with incipient cracks near the Don Juan pond in Wright Valley (Fig. 1, 2). After  
83 the experiment, extraction of the boulders revealed light-toned salts along the rock cracks that were  
84 embedded in the soil during the experiment (Fig. 2b, c). Mineralogical and chemical analyses of the

85 salts were performed at the Geological Survey of Israel. Salts samples taken from a crack in the boulder  
86 were dissolved in distilled water, and chemical analysis for major cations was conducted using  
87 inductively coupled plasma-optical emission spectrometry (Perkin Elmer, Optima 5300) and major  
88 anions using ion chromatography (Dionex ICS-2000). Mineralogic analysis was performed using bulk  
89 X-ray diffraction. Mineral phase identification and semi-quantification were performed using  
90 HighScore Plus® software based on the ICSD database.



91 **Figure 2: a) Field experiment setup near Don Juan pond (in the background, west of the experiment site)**  
92 **in Wright Valley. White arrows mark the imaged boulder. The length of the boulder is 25 cm. Monitoring**  
93 **equipment includes: a time-lapse optical camera used to detect wetting events, a mobile meteorological**

94 station, and a thermal camera. b) Image of the Ferrar dolerite boulder removed from the soil after the  
95 experiment (white arrow). c) The underside of the Ferrar dolerite boulder showing accumulation of salts  
96 in cracks.

97 **2.2 Meteorological data**

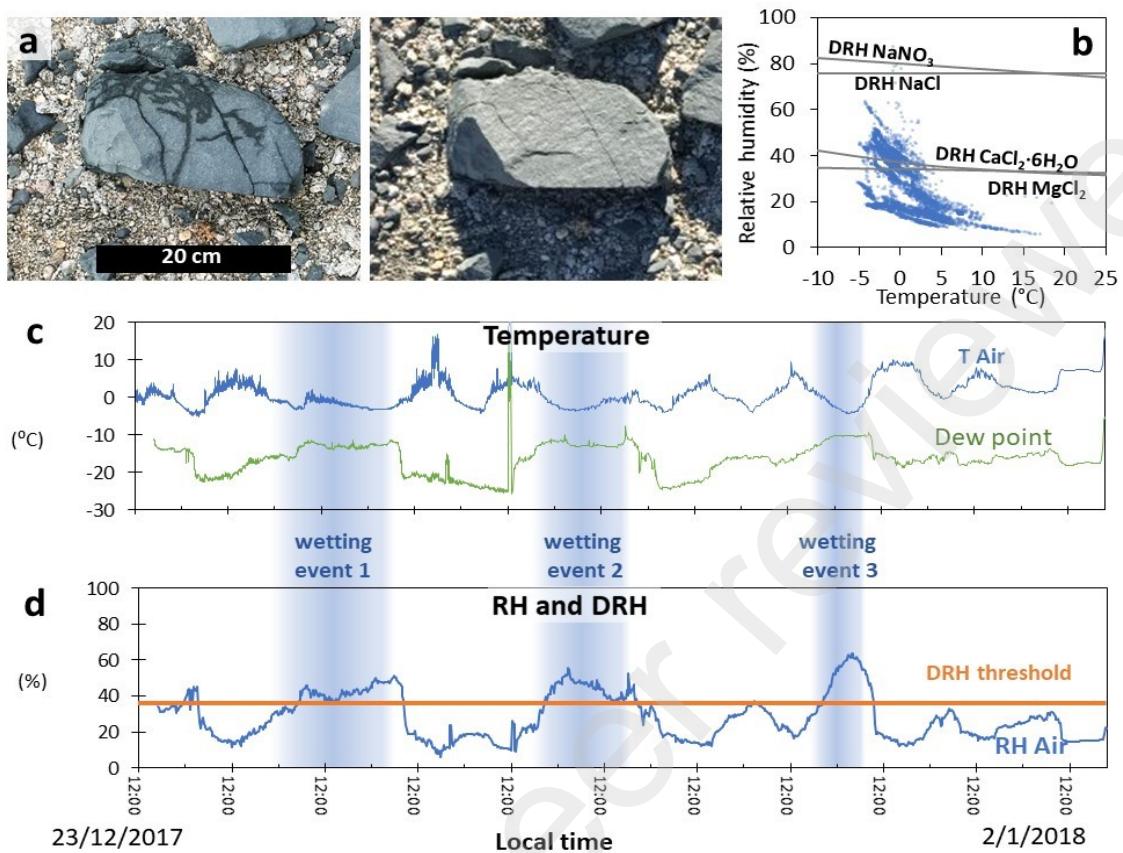
98 Meteorological data from 17 weather stations scattered throughout the ADV were used to  
99 examine the occurrence of deliquescence-efflorescence conditions for  $\text{NaNO}_3$ ,  $\text{CaCl}_2$ ,  $\text{NaCl}$ , and  $\text{MgCl}_2$ .  
100 The weather stations are part of the McMurdo Long Term Ecological Research Project (MCM LTER)  
101 in the ADV. Most stations (11 out of 17) have been operating for over 20 years at 1-hour temporal  
102 recording resolution. Analysis of these data was conducted to quantify the occurrence of supra-DRH  
103 conditions for these salt phases in the ADV environment through time. Deliquescence conditions were  
104 defined as the durations in a year when RH values exceed the DRH of the specific salt phase. A  
105 Deliquescence/efflorescence cycle was defined as a period between the increase of RH above DRH to  
106 when RH decreases below DRH. A minimum duration threshold of 3 hours and a minimum of 5%  
107 excess humidity above DRH were used to filter out short events and those with marginal excess RH.  
108 The DRH value of  $\text{CaCl}_2$  used in the analysis refers to the hexahydrate phase -  $\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$   
109 ('Antarcticite'), which was first described in the ADV (Torii & Ossaka, 1965).

110 **3 Results**

111 **3.1 Salt deliquescence during the field experiment**

112 Time-lapse photography revealed an accumulation of moisture along cracks in the imaged  
113 boulder (identified visually by the darkening of the rock, Fig. 3a) during three discrete periods spanning

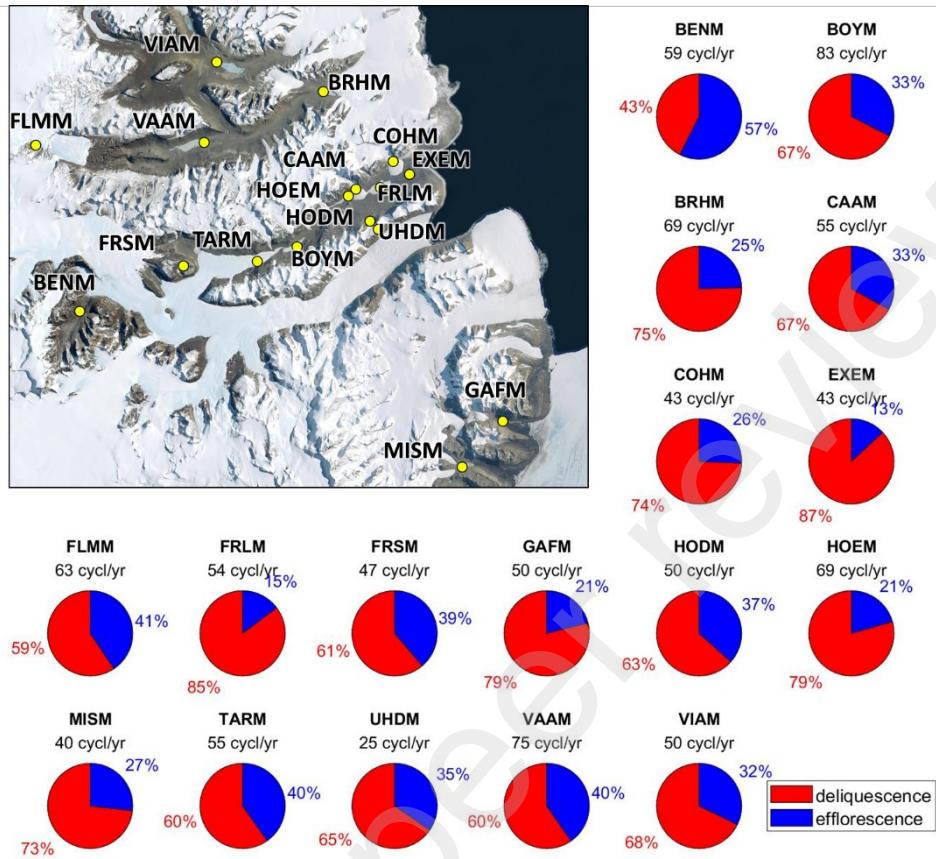
114 between approximately 6-12 hours each (Fig. 3a, supp. Time-lapse video). Since no precipitation was  
115 observed during the experiment, we can overrule snow melt as a potential source, although it is a  
116 recognized source of water for rocks and soils during the austral summers (Hagedorn et al., 2010; Liu et  
117 al., 2015). These periods of moisture stability along the boulder cracks coincided exclusively with RH  
118 values that exceeded 35~40% but did not reach the dew point and included sub-zero air temperatures  
119 (Fig. 3c, d). During these moisture accumulation events, RH values coincided with supra DRH  
120 conditions for chloride salts, such as  $\text{CaCl}_2$  and  $\text{MgCl}_2$  (Fig. 3b, d). Chemical analysis of the salts  
121 samples taken from a crack in the boulder shows that the major anions are  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$  and (55%, 45%,  
122 respectively) and that the major cations are  $\text{Ca}^{+2}$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{+2}$ ,  $\text{SiO}_2$ , and  $\text{Sr}^{+2}$  (71%, 23%, 3%, 2%, 1%,  
123 1% respectively). Excess of  $\text{Cl}^-$  and  $\text{Ca}^+$  after accounting for the complete precipitation of halite and  
124 gypsum, which were the dominant salt phases found in X-ray diffraction, points to the presence of  
125  $\text{CaCl}_2$  and possibly other chlorides, which is not unexpected considering the proximity of the  
126 experiment site to the Don Juan pond – a saline lake rich in  $\text{CaCl}_2$  (Dickson et al., 2013; Toner et al.,  
127 2017). Therefore, the most likely explanation for the observed wetting events in the experiment appears  
128 to be the deliquescence of such salts. Altogether moisture sourced from deliquesced atmospheric  
129 humidity was found to be stable along the boulder cracks for ~25% of the otherwise ‘dry’ 10-day span  
130 of the experiment.



131 **Figure 3: Moisture delivery to rock cracks via salt deliquescence. a)** Images of a Ferrar dolerite boulder  
 132 **during (left) and between (right) wetting events. b)** Data from field measurements of air temperature vs.  
 133 relative humidity during the 10 days experiment. Black lines indicate the DRH of  $\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$ ,  $\text{NaNO}_3$ ,  
 134  $\text{NaCl}$ ,  $\text{MgCl}_2$  salts as a function of temperature. **c)** Ambient air T (blue), and dew point (green) through  
 135 time. Note that air T does not reach the dew point during observed wetting events and that wetting events  
 136 (blue shading) persist through sub-zero temperatures. **d)** Relative humidity (blue) and the DRH threshold  
 137 of 35%~40% (orange). All three wetting events were initiated after supra-DRH conditions were achieved  
 138 and ended when RH declined below the DRH threshold.

139 **3.2 Analysis of meteorological data**

140 Results from analysis of 17 meteorological stations in the ADV show that the conditions for  
141 deliquescence of the salt phases that were examined, i.e.,  $\text{CaCl}_2$ ,  $\text{NaNO}_3$ ,  $\text{NaCl}$ , and  $\text{MgCl}_2$ , prevail on  
142 average for 69% (range 43%-85%), 16% (range 4%-25%), 32% (range 10%-51%) and 83% (range  
143 43%-85%) of the year, respectively (Fig. 4, Fig. supp. S1-3). In addition, the conditions for discrete  
144 deliquescence events happen on average 55 (range 25-83), 30 (range 12-48), 50 (range 20-76), and 39  
145 (range 15-36) times per year, respectively (Fig. 4, Fig. supp. S1-3). For simplicity, deliquescence  
146 conditions for single-phase salt-brine systems were assumed. However, laboratory experiments show  
147 that the DRH of salt mixtures is expected to be even lower than that of the same single salts (e.g. Yang  
148 et al., 2002). Thus, the duration of deliquescence conditions is potentially longer than calculated herein.



149 **Figure 4: Results of deliquescence conditions of  $\text{CaCl}_2$  in 17 meteorological stations in the ADV. The pie**  
 150 **plots show the time fraction of deliquescence conditions (red). The number of estimated deliquescence-**  
 151 **efflorescence cycles per year is marked below the station name. A full description of the stations is found at**  
 152 [\*\*https://mcm.lternet.edu/meteorology-data-sets#met-15.\*\*](https://mcm.lternet.edu/meteorology-data-sets#met-15)

153 **4 Discussion**

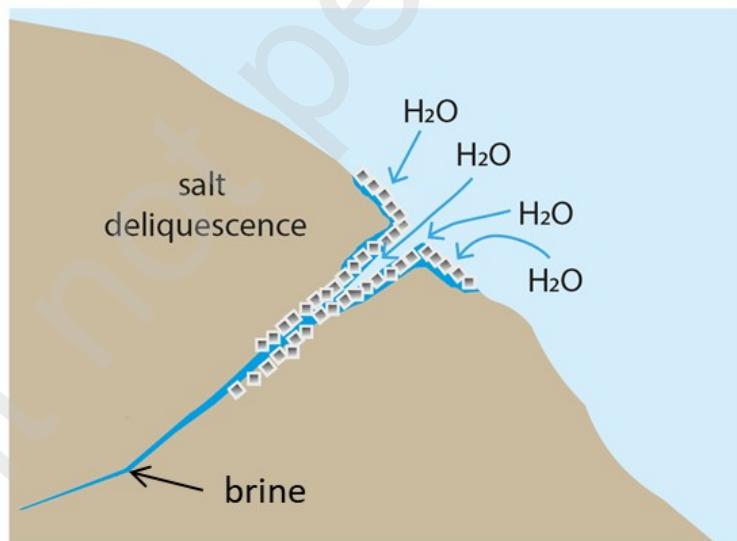
154 The results of the field-based experiment show evidence that salt deliquescence is an effective  
 155 mechanism of moisture delivery to the rock surface and cracks in the hyper-arid and frigid conditions of  
 156 the ADV. Deliquescence/efflorescence cycles such as those observed during the experiment (Fig. 3) are

157 expected to promote rock cracking by ‘salt shattering’, which is regarded as an important weathering  
158 mechanism in arid soils (Amit et al., 1993; A. Goudie & Viles, 1997; Andrew S. Goudie, 2013;  
159 Rodriguez-Navarro & Doehne, 1999), including the ADV (Johnston, 1973) and Mars (Jagoutz, 2006;  
160 Malin, 1974). Salt shattering requires that the amount of water is low enough to limit the leaching of  
161 the salts from the surface/soils, and yet sufficient for salt-water interactions that support cycles of  
162 dissolution and crystallization of salts that can exert local stress when confined within rock pores or  
163 fractures (Amit et al., 1993; Desarnaud et al., 2016; Sperling & Cooke, 1985). In addition, recent studies  
164 have demonstrated that even the rates of ‘dry’ mechanical weathering processes, such as those induced  
165 by salt hydration or cyclic thermal stress-loading (Lamp et al., 2017; McFadden et al., 2005; Richter &  
166 Simmons, 1974; Viles et al., 2010), may accelerate by orders of magnitude in the presence of small  
167 amounts of moisture (Eppes et al., 2020; Eppes & Keanini, 2017). This acceleration is associated with  
168 the weakening of bonds by water molecules at the tip of cracks that propagate slowly in response to  
169 subcritical stresses (Atkinson, 1984; Eppes et al., 2020; Meredith & Atkinson, 1985; Nara & Kaneko,  
170 2006; Voigtländer et al., 2018). The delivery of moisture to the tips of cracks, via salt deliquescence,  
171 can thus also accelerate crack propagation under external subcritical stress. The analysis of data  
172 obtained from permanent ADV meteorological stations (Fig. 4) shows that the atmospheric conditions  
173 that enabled the deliquescence events in our field site near the Don Juan pond are prevalent throughout  
174 the ADV and therefore suggest that deliquesced moisture during otherwise ‘dry’ conditions can be an  
175 important moisture delivery pathway for rock weathering throughout the ADV soil.

176 Laboratory experiments show that ERH can be lower than DRH for a given salt and temperature  
177 due to a kinetic barrier for the nucleation of a crystalline phase (Gough et al., 2016; Martin, 2000). In

178 our experiment, there is no evidence for the reported hysteresis between DRH and ERH. This could be  
179 because of the heterogeneity and impurity of natural brines that can readily facilitate the nucleation of  
180 salt crystals. Furthermore, the onset and termination of the deliquescence events that were documented  
181 during the field experiment were not driven by diurnal oscillations in air temperature and resulting  
182 changes in RH and instead appeared to be more closely associated with pulses of increased vapor  
183 pressure (Fig. Supp. S4).

184 We thus propose that deliquescence/efflorescence cycles may be an important driver of rock  
185 weathering in the ADV and potentially other hyper-arid regions where an accumulation of hygroscopic  
186 salts is observed. This includes Mars, where the presence of deliquescent salts was previously suggested  
187 (Gough et al., 2019; Toner et al., 2015).



188 **Figure 5: illustration of moisture delivery into rock cracks by salt deliquescence. Deliquesced moisture**  
189 **(brine) within rock cracks can accelerate the propagation of cracks by weakening rock chemical bonds by**

190 **water molecules at the tip of cracks that propagate slowly in response to subcritical stresses. Such stresses**  
191 **can originate from cycles of salt crystallization (deliquescence/efflorescence cycles) and/or thermal**  
192 **expansion/contraction in response to diurnal insolation dynamics.**

193 **5 Conclusion**

194       Based on our results from a field experiment that show discrete wetting events of a rock surface  
195   in sub-saturated air and sub-freezing conditions, following an increase in air RH, we conclude that salt  
196   deliquescence may be an effective and overlooked mechanism of water delivery into rock cracks in  
197   hyper-arid and cold conditions on Earth and possibly Mars as well. An analysis of data from 17  
198   permanent meteorological stations shows that conditions for the deliquescence of several salt phases  
199   that are found in the ADV prevail throughout the region and that such moisture delivery may be  
200   widespread in the ADV. We suggest a dual role that salt deliquescence plays in the acceleration of rock  
201   cracking in arid regions. The first is a source of stress load, as cycles of dissolution and crystallization  
202   of salts exert local stress when confined within rock pores or fractures (Sperling and Cooke, 1985;  
203   Desarnaud et al., 2016; Amit et al., 1993). The second is the acceleration of subcritical rock cracking in  
204   the presence of even small amounts of water at the tip of cracks (Eppes et al., 2020; Eppes & Keanini,  
205   2017; Meredith & Atkinson, 1985; Yoshitaka Nara et al., 2010, 2012; Waza et al., 1980).

206 **Acknowledgments**

207       This work was supported by the United States - Israel Binational Science Foundation (BSF),  
208   grant number: 2018610.

210 Amit, R., Gerson, R., & Yaalon, D. H. (1993). Stages and rate of the gravel shattering process by salts  
211 in desert Reg soils. *Geoderma*, 57(3), 295–324. [https://doi.org/10.1016/0016-7061\(93\)90011-9](https://doi.org/10.1016/0016-7061(93)90011-9)

212 Anderson, T. L. (2005). *Fracture Mechanics: Fundamentals and Applications*. CRC Press.

213 Atkinson, B. K. (1984). Subcritical Crack Growth in Geological Materials. *Journal of Geophysical*  
214 *Research*, 89(B6), 4077–4114. <https://doi.org/10.1029/jb089ib06p04077>

215 Balco, G., & Shuster, D. L. (2009). Production rate of cosmogenic  $^{21}\text{Ne}$  in quartz estimated from  $^{10}\text{Be}$ ,  
216  $^{26}\text{Al}$ , and  $^{21}\text{Ne}$  concentrations in slowly eroding Antarctic bedrock surfaces. *Earth and*  
217 *Planetary Science Letters*, 281(1–2), 48–58. <https://doi.org/10.1016/j.epsl.2009.02.006>

218 Bisson, K. M., Welch, K. A., Welch, S. A., Sheets, J. M., Lyons, W. B., Levy, J. S., & Fountain, A. G.  
219 (2015). Patterns and Processes of Salt Efflorescences in the McMurdo region, Antarctica. *Arctic,*  
220 *Antarctic, and Alpine Research*, 47(3), 407–425. <https://doi.org/10.1657/AAAR0014-024>

221 Brook, E. J., Brown, E. T., Kurz, M. D., Ackert, R. P., Raisbeck, G. M., & Yiou, F. (1995). Constraints  
222 on age, erosion, and uplift of Neogene glacial deposits in the Transantarctic Mountains  
223 determined from in situ cosmogenic  $^{10}\text{Be}$  and  $^{26}\text{Al}$ . *Geology*, 23(12), 1063–1066.  
224 [https://doi.org/10.1130/0091-7613\(1995\)023<1063:COAEAU>2.3.CO;2](https://doi.org/10.1130/0091-7613(1995)023<1063:COAEAU>2.3.CO;2)

225 Campbell, I. B., & Claridge, G. G. C. (1987). Antarctica: Soils, Weathering Processes and Environment.  
226 *Develop*, 16.

227 Claridge, G. G. C., & Campbell, I. B. (1977). The salts in Antarctic soils, their distribution and  
228 relationship to soil processes. *Soil Science*, 123(6), 377–384.  
229 <https://doi.org/10.1097/00010694-197706000-00006>

230 Cooke, R. U. (1981). Salt weathering in deserts. *Proceedings of the Geologists' Association*, 92(1), 1–  
231 16. [https://doi.org/10.1016/S0016-7878\(81\)80015-6](https://doi.org/10.1016/S0016-7878(81)80015-6)

232 Cooke, R. U., & Smalley, I. J. (1968). Salt weathering in deserts [10]. *Nature*, 220(5173), 1226–1227.  
233 <https://doi.org/10.1038/2201226a0>

234 Desarnaud, J., Bonn, D., & Shahidzadeh, N. (2016). The Pressure induced by salt crystallization in  
235 confinement. *Scientific Reports*, 6(1), 30856. <https://doi.org/10.1038/srep30856>

236 Dickson, J. L., Head, J. W., Levy, J. S., & Marchant, D. R. (2013). Don Juan Pond, Antarctica: Near-  
237 surface CaCl<sub>2</sub>-brine feeding Earth's most saline lake and implications for Mars. *Scientific  
238 Reports*, 3(4), 1–8. <https://doi.org/10.1038/srep01166>

239 Doran, P. T., McKay, C. P., Clow, G. D., Dana, G. L., Fountain, A. G., Nylen, T., & Lyons, W. B.  
240 (2002). Valley floor climate observations from the McMurdo dry valleys, Antarctica, 1986–  
241 2000. *Journal of Geophysical Research Atmospheres*, 107(24), ACL 13-1-ACL 13-12.  
242 <https://doi.org/10.1029/2001JD002045>

243 Eppes, M. C., & Keanini, R. (2017). Mechanical weathering and rock erosion by climate-dependent  
244 subcritical cracking. *Reviews of Geophysics*, 55(2), 470–508.  
245 <https://doi.org/10.1002/2017RG000557>

246 Eppes, M. C., Magi, B., Scheff, J., Warren, K., Ching, S., & Feng, T. (2020). Warmer, Wetter Climates  
247 Accelerate Mechanical Weathering in Field Data, Independent of Stress-Loading. *Geophysical*  
248 *Research Letters*, 47(24), 1–11. <https://doi.org/10.1029/2020GL089062>

249 Fielding, C. R., Browne, G. H., Field, B., Florindo, F., Harwood, D. M., Krissek, L. A., et al. (2011).  
250 Sequence stratigraphy of the ANDRILL AND-2A drillcore, Antarctica: A long-term, ice-  
251 proximal record of Early to Mid-Miocene climate, sea-level and glacial dynamism.  
252 *Palaeogeography, Palaeoclimatology, Palaeoecology*, 305(1–4), 337–351.  
253 <https://doi.org/10.1016/j.palaeo.2011.03.026>

254 Fountain, A. G., Nylen, T. H., Monaghan, A., Basagic, H. J., & Bromwich, D. (2010). Snow in the  
255 McMurdo Dry Valleys, Antarctica. *International Journal of Climatology*, 30(5), 633–642.  
256 <https://doi.org/10.1002/joc.1933>

257 Goudie, A.S, & Viles, H. A. (1997). *Salt weathering hazard*. wiley.

258 Goudie, A. S., & Cooke, R. U. (1984). Salt efflorescences and Saline lakes; a distributional analysis.  
259 *Geoforum*, 15(4), 563–582. [https://doi.org/10.1016/0016-7185\(84\)90025-3](https://doi.org/10.1016/0016-7185(84)90025-3)

260 Goudie, A. S. (2013). *Arid and semi-arid geomorphology*. Cambridge university press.

261 Gough, R. V., Chevrier, V. F., & Tolbert, M. A. (2016). Formation of liquid water at low temperatures  
262 via the deliquescence of calcium chloride: Implications for Antarctica and Mars. *Planetary and*  
263 *Space Science*, 131, 79–87. <https://doi.org/10.1016/j.pss.2016.07.006>

264 Gough, R. V., Primm, K. M., Rivera-Valentín, E. G., Martínez, G. M., & Tolbert, M. A. (2019). Solid-  
265 solid hydration and dehydration of Mars-relevant chlorine salts: Implications for Gale Crater and  
266 RSL locations. *Icarus*, 321(July 2018), 1–13. <https://doi.org/10.1016/j.icarus.2018.10.034>

267 Hagedorn, B., Sletten, R. S., Hallet, B., McTigue, D. F., & Steig, E. J. (2010). Ground ice recharge via  
268 brine transport in frozen soils of Victoria Valley, Antarctica: Insights from modeling  $\delta^{18}\text{O}$  and  
269  $\delta\text{D}$  profiles. *Geochimica et Cosmochimica Acta*, 74(2), 435–448.  
270 <https://doi.org/10.1016/j.gca.2009.10.021>

271 Hall, K. (1999). The role of thermal stress fatigue in the breakdown of rock in cold regions.  
272 *Geomorphology*, 31(1–4), 47–63. [https://doi.org/10.1016/S0169-555X\(99\)00072-0](https://doi.org/10.1016/S0169-555X(99)00072-0)

273 Harris, H. J. H., & Cartwright, K. (1981). Hydrology of the Don Juan Basin, Wright Valley, Antarctica.  
274 In L. D. McGinnis (Ed.), *Antarctic Research Series* (Vol. 33, pp. 161–184). Washington, D. C.:  
275 American Geophysical Union. <https://doi.org/10.1029/AR033p0161>

276 Head, J. W., Marchant, D., Dickson, J., Levy, J., & Morgan, G. (2007). Slope streaks in the Antarctic  
277 Dry Valleys: Characteristics, candidate formation mechanisms, and implications for slope streak  
278 formation in the Martian environment. *Lunar Planet. Sci.*, XXXVIII.

279 Head, J. W., & Marchant, D. R. (2014). The climate history of early Mars: Insights from the Antarctic  
280 McMurdo Dry Valleys hydrologic system. *Antarctic Science*, 26(6), 774–800.  
281 <https://doi.org/10.1017/S0954102014000686>

282 Jagoutz, E. (2006). Salt-induced rock fragmentation on Mars: The role of salt in the weathering of  
283 Martian rocks. *Advances in Space Research*, 38(4), 696–700.  
284 <https://doi.org/10.1016/j.asr.2005.07.070>

285 Johnston, J. H. (1973). Salt weathering processes in the McMurdo Dry Valley regions of South Victoria  
286 Land, Antarctica. *New Zealand Journal of Geology and Geophysics*, 16(2), 221–224.  
287 <https://doi.org/10.1080/00288306.1973.10431454>

288 Keys, J. R. (1979). Distribution of salts in the McMurdo region, with analyses from the saline discharge  
289 area at the terminus of Taylor Glacier. *Publication of the Geology Department, Victoria  
290 University of Wellington*, 14.

291 Keys, J. R., & Williams, K. (1981). Origin of crystalline, cold desert salts in the McMurdo region,  
292 Antarctica. *Geochimica et Cosmochimica Acta*, 45(12), 2299–2309.  
293 [https://doi.org/10.1016/0016-7037\(81\)90084-3](https://doi.org/10.1016/0016-7037(81)90084-3)

294 Lamp, J. L., Marchant, D. R., Mackay, S. L., & Head, J. W. (2017). Thermal stress weathering and the  
295 spalling of Antarctic rocks. *Journal of Geophysical Research: Earth Surface*, 122(1), 3–24.  
296 <https://doi.org/10.1002/2016JF003992>

297 Levy, J. (2021). Episodic basin-scale soil moisture anomalies associated with high relative humidity  
298 events in the McMurdo Dry Valleys, Antarctica. *Antarctic Science*, 33(5), 533–547.  
299 <https://doi.org/10.1017/S0954102021000341>

300 Liu, L., Sletten, R. S., Hagedorn, B., Hallet, B., McKay, C. P., & Stone, J. O. (2015). An enhanced  
301 model of the contemporary and long-term (200 ka) sublimation of the massive subsurface ice in

302       Beacon Valley, Antarctica. *Journal of Geophysical Research: Earth Surface*, 120(8), 1596–  
303       1610. <https://doi.org/10.1002/2014JF003415>

304       Malin, M. C. (1974). Salt weathering on Mars. *Journal of Geophysical Research*, 79(26), 3888–3894.  
305       <https://doi.org/10.1029/jb079i026p03888>

306       Margerison, H. R., Phillips, W. M., Stuart, F. M., & Sugden, D. E. (2005). Cosmogenic  $^{3}\text{He}$   
307       concentrations in ancient flood deposits from the Coombs Hills, northern Dry Valleys, East  
308       Antarctica: Interpreting exposure ages and erosion rates. *Earth and Planetary Science Letters*,  
309       230(1–2), 163–175. <https://doi.org/10.1016/j.epsl.2004.11.007>

310       Marrero, S. M., Hein, A. S., Naylor, M., Attal, M., Shanks, R., Winter, K., et al. (2018). Controls on  
311       subaerial erosion rates in Antarctica. *Earth and Planetary Science Letters*, 501, 56–66.  
312       <https://doi.org/10.1016/j.epsl.2018.08.018>

313       McFadden, L. D., Eppes, M. C., Gillespie, A. R., & Hallet, B. (2005). Physical weathering in arid  
314       landscapes due to diurnal variation in the direction of solar heating. *Bulletin of the Geological  
315       Society of America*, 117(1–2), 161–173. <https://doi.org/10.1130/B25508.1>

316       Meredith, P. G., & Atkinson, B. K. (1985). Fracture toughness and subcritical crack growth during  
317       high-temperature tensile deformation of Westerly granite and Black gabbro. *Physics of the  
318       Earth and Planetary Interiors*, 39(1), 33–51. [https://doi.org/10.1016/0031-9201\(85\)90113-X](https://doi.org/10.1016/0031-9201(85)90113-X)

319       Miotke, F. D., & von Hodenberg, R. (1983). *Salt fretting and chemical weathering in the darwin  
320       mountains and the dry valleys, victoria land, antarctica. Polar Geography and Geology* (Vol. 7,  
321       p. 122). Taylor & Francis Group. <https://doi.org/10.1080/10889378309377191>

322 Nara, Y., & Kaneko, K. (2006). Sub-critical crack growth in anisotropic rock. *International Journal of*  
323 *Rock Mechanics and Mining Sciences*, 43(3), 437–453.  
324 <https://doi.org/10.1016/j.ijrmms.2005.07.008>

325 Nara, Y., Hiroyoshi, N., Yoneda, T., & Kaneko, K. (2010). Effects of relative humidity and temperature  
326 on subcritical crack growth in igneous rock. *International Journal of Rock Mechanics and*  
327 *Mining Sciences*, 47(4), 640–646. <https://doi.org/10.1016/j.ijrmms.2010.04.009>

328 Nara, Y., Morimoto, K., Hiroyoshi, N., Yoneda, T., Kaneko, K., & Benson, P. M. (2012). Influence of  
329 relative humidity on fracture toughness of rock: Implications for subcritical crack growth.  
330 *International Journal of Solids and Structures*, 49(18), 2471–2481.  
331 <https://doi.org/10.1016/j.ijsolstr.2012.05.009>

332 Obryk, M. K., Doran, P. T., Fountain, A. G., Myers, M., & McKay, C. P. (2020). Climate From the  
333 McMurdo Dry Valleys, Antarctica, 1986–2017: Surface Air Temperature Trends and Redefined  
334 Summer Season. *Journal of Geophysical Research: Atmospheres*, 125(13).  
335 <https://doi.org/10.1029/2019JD032180>

336 Richter, D., & Simmons, G. (1974). Thermal expansion behavior of igneous rocks. *International*  
337 *Journal of Rock Mechanics and Mining Sciences & Geomechanics Abstracts*, 11(10), 403–411.  
338 [https://doi.org/10.1016/0148-9062\(74\)91111-5](https://doi.org/10.1016/0148-9062(74)91111-5)

339 Rodriguez-Navarro, C., & Doehne, E. (1999). Salt weathering: Influence of evaporation rate,  
340 supersaturation and crystallization pattern. *Earth Surface Processes and Landforms*, 24(2–3),  
341 191–209. [https://doi.org/10.1002/\(sici\)1096-9837\(199903\)24:3<191::aid-esp942>3.0.co;2-g](https://doi.org/10.1002/(sici)1096-9837(199903)24:3<191::aid-esp942>3.0.co;2-g)

342 Scopelliti, G., Bellanca, A., Monien, D., & Kuhn, G. (2013). Chemostratigraphy of the early Pliocene  
343 diatomite interval from MIS AND-1B core (Antarctica): Palaeoenvironment implications.  
344 *Global and Planetary Change*, 102, 20–32. <https://doi.org/10.1016/j.gloplacha.2013.01.001>

345 Selby, M. J., & Wilson, A. T. (1971). The origin of the Labyrinth, Wright Valley, Antarctica. *Bulletin  
346 of the Geological Society of America*, 82(2), 471–476. [https://doi.org/10.1130/0016-7606\(1971\)82\[471:TOOTLW\]2.0.CO;2](https://doi.org/10.1130/0016-<br/>347 7606(1971)82[471:TOOTLW]2.0.CO;2)

348 Sletten, R. S., Hallet, B., & Fletcher, R. C. (2003). Resurfacing time of terrestrial surfaces by the  
349 formation and maturation of polygonal patterned ground. *Journal of Geophysical Research E:  
350 Planets*, 108(4), 1–10. <https://doi.org/10.1029/2002je001914>

351 Sperling, C. H. B., & Cooke, R. U. (1985). Laboratory simulation of rock weathering by salt  
352 crystallization and hydration processes in hot, arid environments. *Earth Surface Processes and  
353 Landforms*, 10(6), 541–555. <https://doi.org/10.1002/esp.3290100603>

354 Staiger, J. W., Marchant, D. R., Schaefer, J. M., Oberholzer, P., Johnson, J. V., Lewis, A. R., &  
355 Swanger, K. M. (2006). Plio-Pleistocene history of Ferrar Glacier, Antarctica: Implications for  
356 climate and ice sheet stability. *Earth and Planetary Science Letters*, 243(3–4), 489–503.  
357 <https://doi.org/10.1016/j.epsl.2006.01.037>

358 Steiger, M., Linnow, K., Juling, H., Gölker, G., Jarad, A. E., Brüggerhoff, S., & Kirchner, D. (2008).  
359 Hydration of  $MgSO_4 \cdot H_2O$  and Generation of Stress in Porous Materials. *Crystal Growth &  
360 Design*, 8(1), 336–343. <https://doi.org/10.1021/cg060688c>

361 Sugden, D. E., Summerfield, M. A., Denton, G. H., Wilch, T. I., McIntosh, W. C., Marchant, D. R., &  
362 Rutherford, R. H. (1999). Landscape development in the Royal Society Range, southern Victoria  
363 Land, Antarctica: Stability since the mid-Miocene. *Geomorphology*, 28(3–4), 181–200.  
364 [https://doi.org/10.1016/S0169-555X\(98\)00108-1](https://doi.org/10.1016/S0169-555X(98)00108-1)

365 Summerfield, M. A., Sugden, D. E., Denton, G. H., Marchant, D. R., Cockburn, H. A. P., & Stuart, F.  
366 M. (1999). Cosmogenic isotope data support previous evidence of extremely low rates of  
367 denudation in the Dry Valleys region, southern Victoria Land, Antarctica. *Geological Society  
368 Special Publication*, 162(1), 255–267. <https://doi.org/10.1144/GSL.SP.1999.162.01.20>

369 Tamppari, L. K., Anderson, R. M., Archer, P. D., Douglas, S., Kounaves, S. P., McKay, C. P., et al.  
370 (2012). Effects of extreme cold and aridity on soils and habitability: McMurdo Dry Valleys as  
371 an analogue for the Mars Phoenix landing site. *Antarctic Science*, 24(3), 211–228.  
372 <https://doi.org/10.1017/S0954102011000800>

373 Toner, J. D., Catling, D. C., & Sletten, R. S. (2017). The geochemistry of Don Juan Pond: Evidence for  
374 a deep groundwater flow system in Wright Valley, Antarctica. *Earth and Planetary Science  
375 Letters*, 474, 190–197. <https://doi.org/10.1016/j.epsl.2017.06.039>

376 Toner, J. D., Sletten, R. S., Liu, L., Catling, D. C., Ming, D. W., Mushkin, A., & Lin, P.-C. (2022). Wet  
377 streaks in the McMurdo Dry Valleys, Antarctica: Implications for Recurring Slope Lineae on  
378 Mars. *Earth and Planetary Science Letters*, 589, 117582.  
379 <https://doi.org/10.1016/j.epsl.2022.117582>

380 Toner, J. D., Catling, D. C., & Light, B. (2015). Modeling salt precipitation from brines on Mars:  
381 Evaporation versus freezing origin for soil salts. *Icarus*, 250, 451–461.  
382 <https://doi.org/10.1016/j.icarus.2014.12.013>

383 Torii, T., & Ossaka, J. (1965). Antarcticite: A new mineral, calcium chloride hexahydrate, discovered in  
384 Antarctica. *Science*, 149(3687), 975–977. <https://doi.org/10.1126/science.149.3687.975>

385 Viles, H., Ehlmann, B., Wilson, C. F., Cebula, T., Page, M., & Bourke, M. (2010). Simulating  
386 weathering of basalt on Mars and Earth by thermal cycling. *Geophysical Research Letters*,  
387 37(18), 1–5. <https://doi.org/10.1029/2010GL043522>

388 Voigtländer, A., Leith, K., & Krautblatter, M. (2018). Subcritical Crack Growth and Progressive Failure  
389 in Carrara Marble Under Wet and Dry Conditions. *Journal of Geophysical Research: Solid*  
390 *Earth*, 123(5), 3780–3798. <https://doi.org/10.1029/2017JB014956>

391 Waza, T., Kurita, K., & Mizutai, H. (1980). The effect of water on the subcritical silicate rocks crack  
392 growth in silicate rocks. *Tectonophysics*, 1967, 25–34.

393 Wellman, H. W., & Wilson, A. T. (1965). Salt Weathering, a Neglected Geological Erosive Agent in  
394 Coastal and Arid Environments. *Nature*, 205(4976), 1097–1098.  
395 <https://doi.org/10.1038/2051097a0>

396 Wilson, A. T. (1979). Geochemical problems of the Antarctic dry areas. *Nature*, 280(5719), 205–208.  
397 <https://doi.org/10.1038/280205a0>

398 Winkler, I. M., & Wilhelm, E. J. (1970). Salt Burst by Hydration Pressures in Architectural Stone in  
399 Urban Atmosphere. *Geological Society of America Bulletin*, 81, 567–572.

400 Yang, L., Pabalan, R. T., & Browning, L. (2002). Experimental determination of the deliquescence  
401 relative humidity and conductivity of multicomponent salt mixtures. *Materials Research Society  
402 Symposium - Proceedings*, 713, 135–142. <https://doi.org/10.1557/proc-713-jj11.4>

403