

# Hollow Octadecameric Self-Assembly of Collagen-like Peptides

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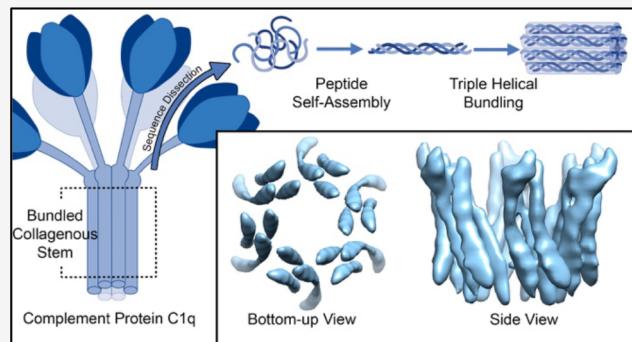
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**ABSTRACT:** The folding of collagen is a hierarchical process that starts with three peptides associating into the characteristic triple helical fold. Depending on the specific collagen in question, these triple helices then assemble into bundles reminiscent of  $\alpha$ -helical coiled-coils. Unlike  $\alpha$ -helices, however, the bundling of collagen triple helices is very poorly understood with almost no direct experimental data available. In order to shed light on this critical step of collagen hierarchical assembly, we have examined the collagenous region of complement component 1q. Thirteen synthetic peptides were prepared to dissect the critical regions allowing for its octadecameric self-assembly. We find that short peptides (under 40 amino acids) are able to self-assemble into specific  $(ABC)_6$  octadecamers. This requires the ABC heterotrimeric composition as the self-assembly subunit, but does not require disulfide bonds. Self-assembly into this octadecamer is aided by short noncollagenous sequences at the N-terminus, although they are not entirely required. The mechanism of self-assembly appears to begin with the very slow formation of the ABC heterotrimeric helix, followed by rapid bundling of triple helices into progressively larger oligomers, terminating in the formation of the  $(ABC)_6$  octadecamer. Cryo-electron microscopy reveals the  $(ABC)_6$  assembly as a remarkable, hollow, crown-like structure with an open channel approximately 18 Å at the narrow end and 30 Å at the wide end. This work helps to illuminate the structure and assembly mechanism of a critical protein in the innate immune system and lays the groundwork for the *de novo* design of higher order collagen mimetic peptide assemblies.



## INTRODUCTION

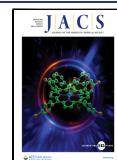
The structure of collagen and collagen-like proteins is defined by a triple helical folding motif in which three peptide strands, adopting a left-handed polyproline type II (PP2) helix, intertwine in a right-handed superhelix. Three chains associate with a one amino acid stagger as the natural register.<sup>1,2</sup> This triple helix is stabilized by a repetitive primary sequence in each peptide conforming to an  $(Xaa-Yaa-Gly)_n$  repeat, where glycine is required due to steric constraints and also forms stabilizing interstrand hydrogen bonds. Positions Xaa and Yaa can be any amino acid, but are frequently proline and hydroxyproline, respectively, which help to stabilize the helix by preorganization.<sup>3–8</sup> Additionally, it has been found that pairwise interactions between peptide strands within a helix,<sup>9–14</sup> synthetic amino acid side chains,<sup>15–17</sup> or covalent linkages<sup>18–21</sup> can be either stabilizing or destabilizing. These interactions can be used to bias helix assembly into specific compositions as well as to control registration between peptide strands (Figure 1a).

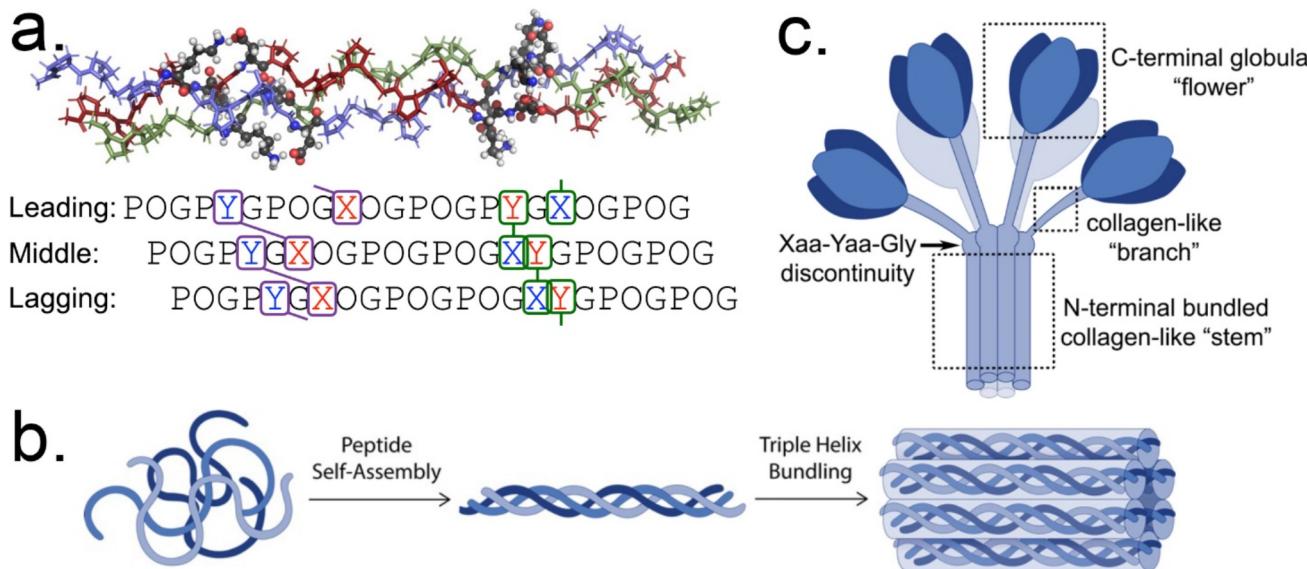
Collagen structural hierarchy, however, does not end at triple helix formation, as most collagens undergo a multistep assembly in which triple helices associate to form higher order structures. For example, type I collagen triple helices bundle together to form fibrils.<sup>22,23</sup> These fibrils further associate and

eventually create macroscopic fibers. Despite the nearly ubiquitous higher order assembly of triple helices in nature, synthetic collagen mimetic peptides (CMPs) have very rarely been observed to assemble beyond a triple helix, and the ones that do undergo such assembly form extremely large, polydisperse assemblies.<sup>24–32</sup> Therefore, unlike the bundling of  $\alpha$ -helices into coiled coils for which interhelix interactions are reasonably well understood and good predictive models exist,<sup>33–36</sup> understanding the mechanism of triple helix bundling (Figure 1b) and the sequence constraints of this process for the collagen triple helix is almost entirely unknown. This gap in knowledge results from (1) the challenges associated with the study of suitable natural systems (extremely large size, extensive posttranslational modification and cross-linking, difficult expression, and poor solubility) and (2) the lack of small, synthetically accessible model systems which undergo controlled self-assembly beyond the triple helix.

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**Figure 1.** (a) Molecular model and sequence of a collagen triple helix highlighting possible axial (purple lassos) and lateral (green lassos) pairwise interactions. “X” and “Y” indicate amino acids in the Xaa or Yaa position of the Xaa-Yaa-Gly collagen repeat. Letters shown in blue are frequently positively charged amino acids, while letters in red are frequently negatively charged amino acids. (b) Depiction of peptide self-assembly to triple helices followed by triple helical bundling. (c) Cartoon of the protein C1q, which is one of several structurally related “defense collagens” that play important roles in innate immunity. The stem of C1q is composed of sequences that closely match the requirement of a collagen triple helix.

Complement component 1q (C1q)<sup>37,38</sup> is a critical component of the innate immune system and is one example of a family of “defense collagens” which include MBL (mannose binding lectin),<sup>39,40</sup> SP-A (surfactant protein A),<sup>41–43</sup> and adiponectin<sup>44–46</sup> among others. C1q is a “bouquet”-shaped structure self-assembled from 18 peptide strands. Peptides C1q-A, C1q-B, and C1q-C associate to form an ABC heterotrimer, and six of these heterotrimers further assemble to generate the full (ABC)<sub>6</sub> octadecamer.<sup>47–49</sup> The C-terminal half of each peptide is composed of a globular domain involved in antigen recognition.<sup>50</sup> The N-terminal half contains a long collagen-like region (CLR) with the characteristic (Xaa-Yaa-Gly)<sub>n</sub> repeat. This repeat is disrupted once in the middle of this domain, creating a kink in the overall bouquet structure that allows the large globular “flowers” to spread out.<sup>51–54</sup> This kink also delineates two different portions of the CLR, which we define here as the “stem” region (where all six triple helices are tightly packed) and the “branch” region (where the helices splay out and away from one another) (Figure 1c). Finally, the last several amino acids of the N-termini do not follow the Xaa-Yaa-Gly motif and also contain cysteine residues involved in interstrand disulfide bonds connecting C1q-A and C1q-B into a heterodimer and two copies of C1q-C into a homodimer.<sup>55,56</sup>

Our hypothesis is that the stem region of C1q can self-assemble into a controlled octadecamer based entirely on interactions between triple helices and is independent of any other portion of the molecule. Additionally, we hypothesize that this assembly does not require posttranslational modifications (other than hydroxyproline) and is not dependent on cysteine-mediated covalent cross-links. If true, this fragment of C1q would serve as a molecularly well-defined starting point to understand helix-helix interactions in collagen. This could then provide insights into the higher-level assembly and structure of this critical family of defense collagens as well as traditional fibrillar collagens in their hierarchical assembly. Additionally, we hope that advances

provided here will lead to a new generation of *de novo* designed peptides similar to the robust design criteria available to  $\alpha$ -helical coiled coils.

In the current study we describe a small (less than 40 amino acid) synthetic peptide derived from the collagen-like protein C1q, which is able to self-assemble into an  $(ABC)_6$  octadecamer with a remarkable hollow, crown-like structure revealed by cryo-electron microscopy. To our knowledge, this is the first report of a collagen-like triple helical system that undergoes a second step of self-assembly to form a discrete (nonpolydisperse) oligomer.

## RESULTS AND DISCUSSION

**C1q Dissection.** The CLR of C1q has previously been demonstrated to have an octadecameric structure based on enzymatic degradation of the globular domains off from the full native C1q structure.<sup>53</sup> More recently, an expression system was developed that contains the full CLR and a designed trimeric coiled coil at the C-termini, which also maintains the octadecameric assembly.<sup>57</sup> To dissect out the minimum domain necessary for oligomerization, we prepared 13 peptides derived from human C1q<sup>58</sup> by standard Fmoc solid phase peptide synthesis<sup>59</sup> (Table 1). See Supporting Information for details of synthesis, purification by high-performance liquid chromatography (HPLC), and characterization by mass spectrometry (Table S1 and Figures S1–S16). Peptides were tested for two degrees of self-assembly: triple helix formation and oligomerization of triple helices into octadecamers. Triple helix formation and thermal stability (melting temperature,  $T_m$ ) were determined by circular dichroism spectroscopy (CD), while octadecamer assembly was determined by size exclusion chromatography (SEC), analytical ultracentrifugation (AUC), and/or cryo-EM. Methods and conditions used in all cases are described in the Supporting Information.

The fundamental question we wanted to answer was, what is the minimum structure required to allow self-assembly into an octadecamer? We started by preparing “stem” region peptides

**Table 1. Names and Sequences of Peptides Studied<sup>a</sup>**

"Stem" Region Peptides:	
A	<b>EDLCRAPD</b> GKKGEAGROGRRGROGLKGEQGEPGAOGIR
B	QLS <b>CTG</b> POAOI <b>GO</b> GIOTGOGP <b>D</b> GQOGTOGI <b>K</b> GEKGLOGL
C	NTG <b>Y</b> GIOMGLOGAO <b>G</b> KD <b>Y</b> D <b>G</b> LOGPK <b>E</b> PGIO
Cysteine to Alanine Replacement:	
A-Ala	<b>EDL</b> CRAPD <b>GKK</b> GEAGROGRRGROGLK <b>GEQG</b> EPGAOGIR
B-Ala	QLS <b>ATG</b> POAOI <b>GO</b> GIOTGOGP <b>D</b> GQOGTOGI <b>K</b> GEKGLOGL
C-Ala	NTG <b>Y</b> GIOMGLOGAO <b>G</b> KD <b>Y</b> D <b>G</b> LOGPK <b>E</b> PGIO
Collagen Exclusive Peptides:	
A-cx	<b>PD</b> GKKGEAGROGRRGROGLK <b>GEQG</b> EPGAOGIR
B-cx	<b>I</b> O <b>G</b> IOGIOTGOGP <b>D</b> GQOGTOGI <b>K</b> GEKGLOGL
N-terminal deletion series:	
A-1	<b>DL</b> CRAPD <b>GKK</b> GEAGROGRRGROGLK <b>GEQG</b> EPGAOGIR
A-2	<b>L</b> CRAPD <b>GKK</b> GEAGROGRRGROGLK <b>GEQG</b> EPGAOGIR
A-3	<b>RA</b> CRAPD <b>GKK</b> GEAGROGRRGROGLK <b>GEQG</b> EPGAOGIR
A-4	<b>R</b> CRAPD <b>GKK</b> GEAGROGRRGROGLK <b>GEQG</b> EPGAOGIR
A-5	<b>AP</b> D <b>GKK</b> GEAGROGRRGROGLK <b>GEQG</b> EPGAOGIR

<sup>a</sup>"O" is (2S,4R)-hydroxyproline. Glycines in the putative triple helical region are shown in bold. Cysteine (or alanine mutant) is highlighted in orange and underlined. Positively charged amino acids K and R are shown in blue; negatively charged amino acids D and E are shown in red. Sequences not conforming to the canonical (Xaa-Yaa-Gly), repeat are highlighted in gray.

A, B, and C (see Table 1). These peptides contain the full amino acid sequence of C1q starting from the N-termini and extending to the Xaa-Yaa-Gly discontinuity. They are believed to be tightly packed with one another in the full native C1q structure and also contain N-terminal cysteines for covalent cross-linking between peptides. While we include hydroxyproline, other known post-translational modifications such as hydroxylysine were not included. All peptides were synthesized as N-acetylated and C-amidated derivatives for self-assembly stabilization.<sup>60–62</sup> After solid phase synthesis, these peptides were cross-linked by iodine oxidation and HPLC purified to generate the C1q-A–C1q-B (A–B) disulfide dimer and C1q-C–C1q-C (C–C) disulfide dimer. These dimers were then allowed to fold and assemble independently or mixed in a 2:1 ratio and similarly allowed to fold and assemble for over one month. Figure 2 shows the CD and SEC data from this study. Circular dichroism shows that while A–B and C–C dimers independently are only suggestive of weak triple helices (melting at or below 15 °C), the 2:1 mixture of A–B with C–C results in a much more robust triple helical signal with a melting temperature of approximately 42 °C. Furthermore, SEC demonstrated that the 2:1 (A–B):(C–C) mixture forms a species much higher in mass than a simple triple helix. Based on SEC calibration curves (see Figure S19A) this high mass peak is consistent with an (A–B)<sub>6</sub>(C–C)<sub>3</sub> assembly mimicking native C1q. In contrast, for C–C alone there was no trace of a higher mass species, while for A–B alone a small high mass peak was observed. While in previous work, Brodsky and Kajava studied the thermal stability of the full CLR domain of C1q and found that the melting temperature was near 37 °C, here we observe a much shorter region, yet with higher thermal stability.<sup>63,64</sup> This is the first demonstration of the C1q region assembling from such a short N-terminal fragment.

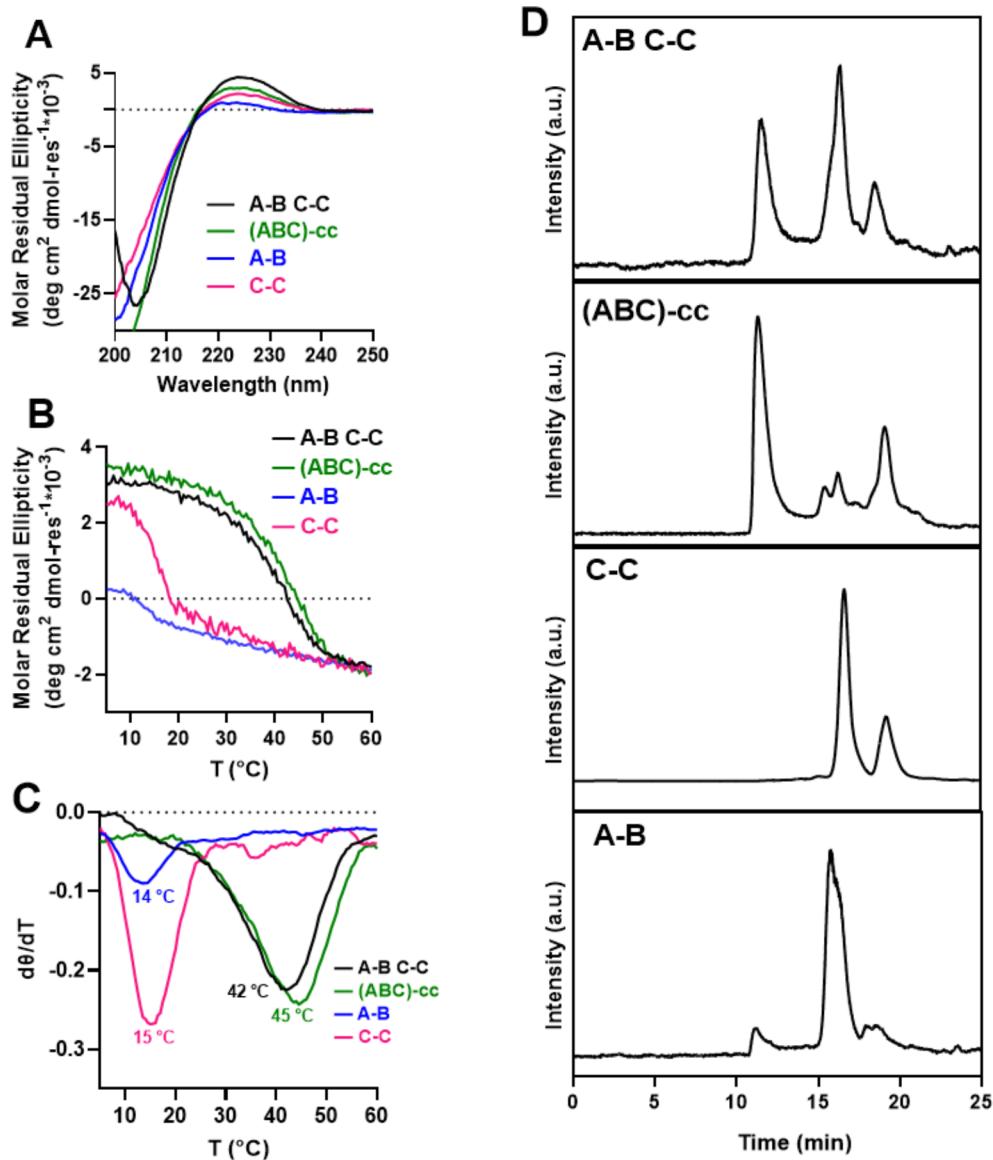
#### Disulfide Bonds Formed through Covalent Capture.

As an alternative approach to disulfide bond formation, instead of mixing preformed and purified disulfide bonded peptides, we instead equilibrated the three stem region peptides A, B, and C together under reducing conditions and subsequently oxidized the solution to form the disulfide bonds *in situ* after oligomerization was confirmed by SEC. This allows us to test if

the bonds form with specificity through covalent capture or would result in a statistical production of all possible disulfide bonds. Figure 2 ("ABC-cc", green) shows the data for this covalently captured assembly, which has nearly identical CD spectra and thermal unfolding characteristics to the presynthesized mixture of dimeric peptides. HPLC and mass spectral analysis of the covalently captured solutions revealed A–B, A–C, and C–C dimeric species in addition to some remaining monomers, but no other dimers were observed (no A–A, B–B, nor B–C, nor C–C dimers, Figure S11). This nonstatistical distribution of disulfide bonds strongly suggests the oligomeric self-assembly preorganizes the formation of suitable disulfide bonds. Interestingly, however, A–C disulfide bonded pairs have not previously been shown in the literature for the native C1q assembly but were observed in our synthetic system and suggest that the cysteines in peptides A and C may be near enough to one another after assembly to covalently cross-link, while this is apparently not favored for A–A, B–B, nor B–C.

**Disulfide Bonds Are Not Required for Stabilization or Self-Assembly.** Next, we wanted to determine the requirement of disulfide bonding in our short C1q mimic peptides. To do this we prepared peptides A-Ala, B-Ala, and C-Ala (Table 1) in which the cysteine from each peptide was replaced with alanine. From these peptides and the previously described cysteine-containing versions we prepared three additional series of self-assembling systems: one in which the A–B disulfide linkage was left intact but the C–C dimer was replaced with C-Ala, a second in which the C–C disulfide linkage was left intact but the A–B dimer was replaced with A-Ala and B-Ala, and third a system in which all the disulfide bonds were eliminated using a mixture of A-Ala, B-Ala, and C-Ala. The four possible self-assembling systems are shown schematically in Figure 3A. In all cases we observed triple helix formation by CD (Figure S18A and Figure 3B and C) as well as higher order assembly consistent with octadecamer formation by SEC (Figure 3D). Impressively, the thermal stability of the self-assembling system with no cysteine (A-Ala, B-Ala, C-Ala) showed the highest thermal stability (47 °C) and an SEC trace that showed a higher fraction of oligomer formation than any system containing disulfide linkages (Figure 3D). The C–C disulfide bond actually appears to moderately destabilize the triple helix, as observed by thermal unfolding, while the A–B disulfide bond does not seem to impact stability positively or negatively. These results are consistent with our study of the ABC assembly under reduced conditions (Figure S20) in which peptides A, B, and C assembled in the presence of DTT (dithiothreitol) formed a heterotrimer with a thermal stability of 46 °C, comparable to (ABC)-Ala and (ABC)-cc (Figure S20A,B). This sample under reduced conditions also assembled into an octadecameric structure according to the SEC characterization (Figure S20C). From these data, we can conclude that triple helix formation and oligomerization of C1q require neither the C-terminal amino acids beyond the Xaa-Yaa-Gly disruption nor covalent stabilization from disulfide bond formation. In fact, the thermal stability of this C1q mimic is improved when cysteine is entirely replaced with alanine.

**An ABC-Type Heterotrimer Is Required.** To understand the requirements of self-assembly of the stem region of C1q in more detail, we investigated the self-assembly of the supramolecular peptides C1q A-Ala, B-Ala, and C-Ala individually and in pairs. Figure 4A and Figure S17 show that none of these peptides individually displays significant PP2 character by CD.

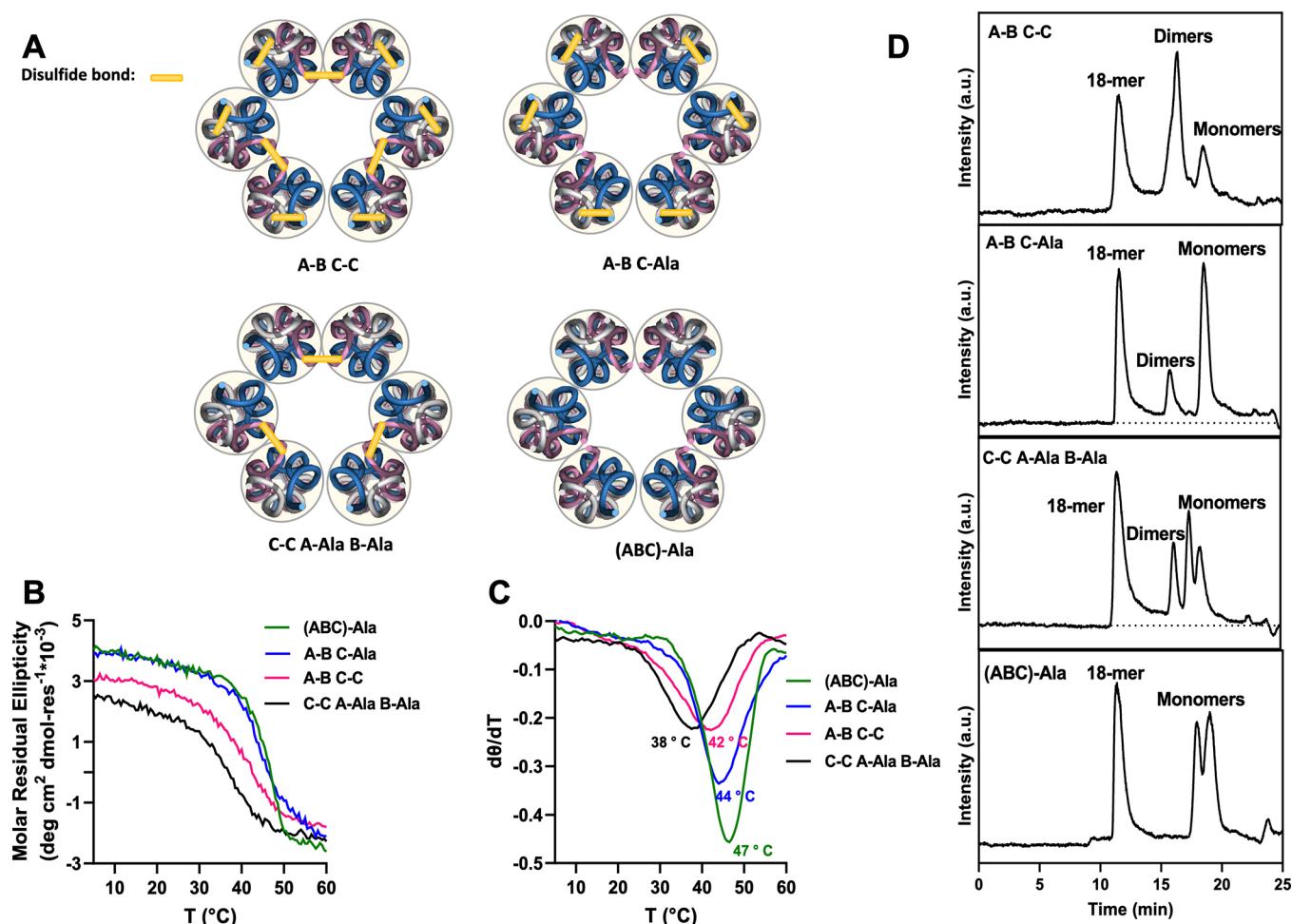


**Figure 2.** Circular dichroism (CD) and size exclusion chromatography (SEC) of the stem region peptides. The CD spectra measurement was performed with 0.3 mM peptide in 10 mM Tris-HCl buffer, with a wavelength scanning from 200 to 250 nm at 5 °C. The melting curves were collected from 5 to 60 °C with a heating rate of 10 °C/hour at the wavelength of the maximum MRE value of each sample. (A) CD spectra of disulfide bonded dimeric peptides and mixtures of these dimers. (B) CD melting curve of the same samples. (C) First-order derivative of the melting curves shown in (B). (D) Size-exclusion chromatography of disulfide bonded peptide dimers and their mixtures.

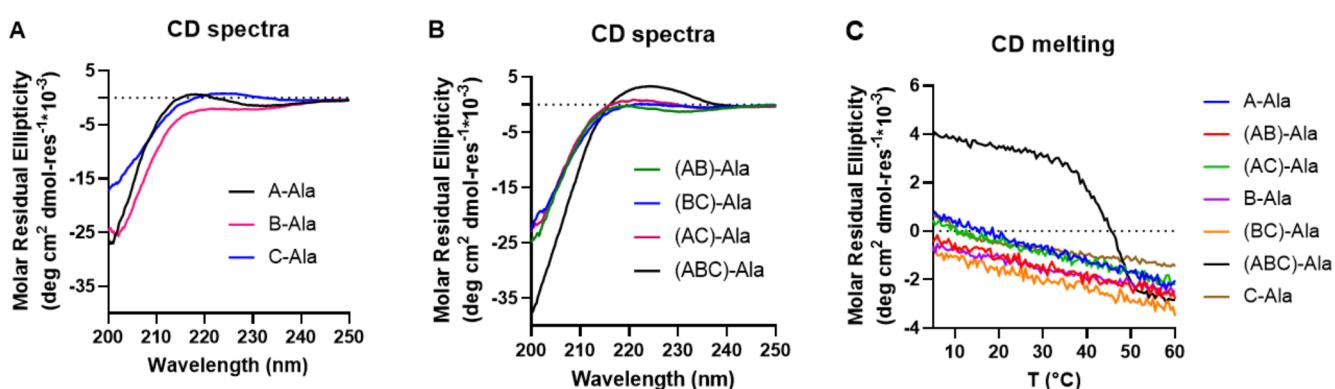
Pairwise mixtures also lack significant triple helical character, as thermal analysis showed the stabilities of the helices to be below 10 °C. In dramatic contrast, the combination of all three peptides forms a strong PP2 helix with a melting temperature of 47 °C (Figure 4C). This demonstrates that triple helix formation in C1q is strongly dependent on the combination of all three peptides into a unique heterotrimer, as only the ABC combination shows strong PP2 character and good thermal stability.

Additionally, we carried out a detailed AUC study of the combined ABC system as well as each peptide alone and in pairwise mixtures. The results for the sedimentation velocity experiments are shown in Figure 5. Measurements of peptides A-Ala, B-Ala, and C-Ala alone, and in pairs, resulted in virtually identical, near-vertical sedimentation patterns, reflecting a homogeneous composition with a weight-average sedimentation coefficient of 0.54 s and a weight-average molar mass of

3620 Da, indicative of a monomeric composition; neither triple helix nor oligomeric assembly is observed (Figure 5A and Figure S19C). On the other hand, mixtures containing all three peptides resulted in one additional species sedimenting at 4.12 s (Figure 5B,C), with an apparent molar mass of 71.4 kDa (Figure 5D). Molar masses are approximate since the precise partial specific volume is an estimate from the amino acid sequence. However, this mass is in good agreement with what is expected from an 18-peptide assembly composed of six units each of A-Ala, B-Ala, and C-Ala,  $(ABC\text{-Ala})_6$ , which has an expected mass of 67.9 kDa. The measurements of the three-peptide mixture at multiple concentrations did not produce a significant change in composition, suggesting that the portion of the sample co-sedimenting at the same speed as the peptide monomeric controls is a result of nonstoichiometric mixing, with one or two of the peptides being present in excess. The remainder of the material is fully assembled into the species



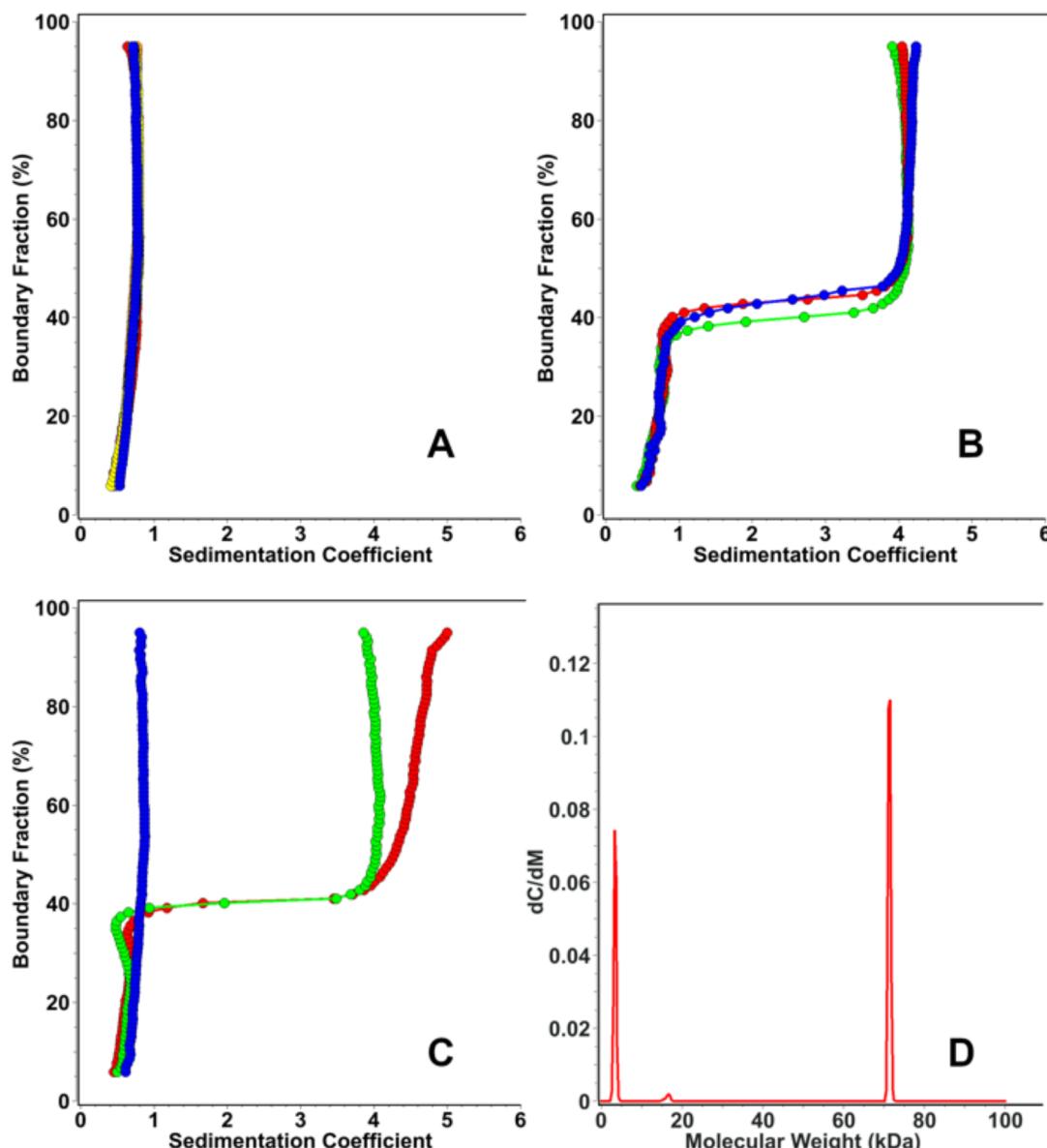
**Figure 3.** (A) Schematic depiction of covalent and supramolecular assemblies of the octadecameric assembly of C1q stem peptides indicating the four different disulfide bonding patterns generated. (B) CD melting data. The melting curves were collected from 5 to 60 °C with a heating rate of 10 °C/hour at the wavelength that gives the maximum MRE value of each sample; sample concentration is 0.3 mM peptide in 10 mM Tris-HCl buffer. (C) The first-order derivatives of the CD melting results. (D) SEC of assemblies with different disulfide bonding patterns. All CD and SEC data were collected after 60 days of equilibration. All SEC chromatographs were monitored at 220 nm.



**Figure 4.** Characterization of the supramolecular (ABC)-Ala assembly composition. (A) CD spectra of monomeric peptide solutions. (B) CD spectra of peptide mixture solutions. (C) CD melting curves of monomeric peptides and peptide mixtures. The melting curves were collected from 5 to 60 °C with a heating rate of 10 °C/hour at the wavelength that gives the maximum MRE value of each sample; sample concentration is 0.3 mM peptide in 10 mM Tris-HCl buffer. Only the complete ABC system shows significant triple helical character.

with a sedimentation coefficient of 4.12 s and does not change composition when the concentration is changed. This suggests that the  $K_d$  (the dissociation constant) is lower than the lowest concentration accessed in this experiment (14  $\mu$ M), and the assembly has strong binding, but only when all three peptides

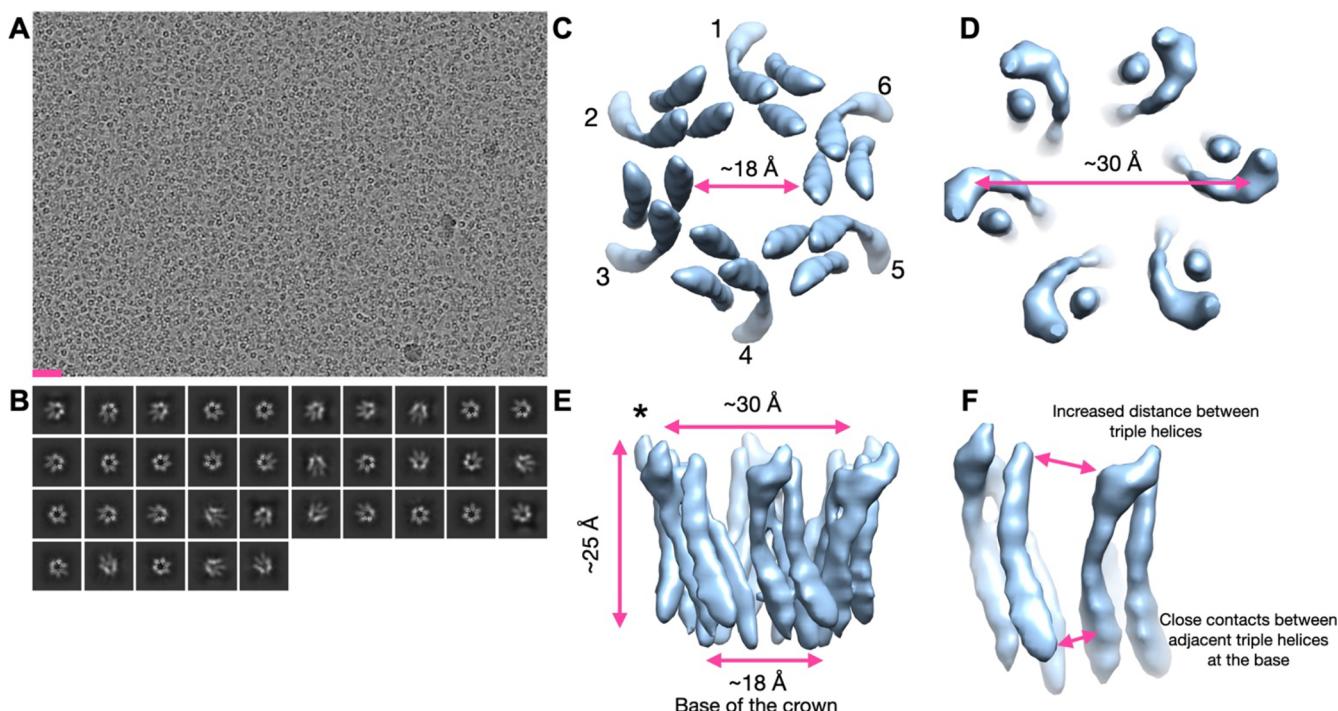
are present in the mixture. This is consistent with the CD and SEC data presented above. When the three-peptide mixture was diluted and heated at 65 °C for 20 min, then cooled at room temperature for 2 h before measurements, only monomeric species were observed, which did not reassemble



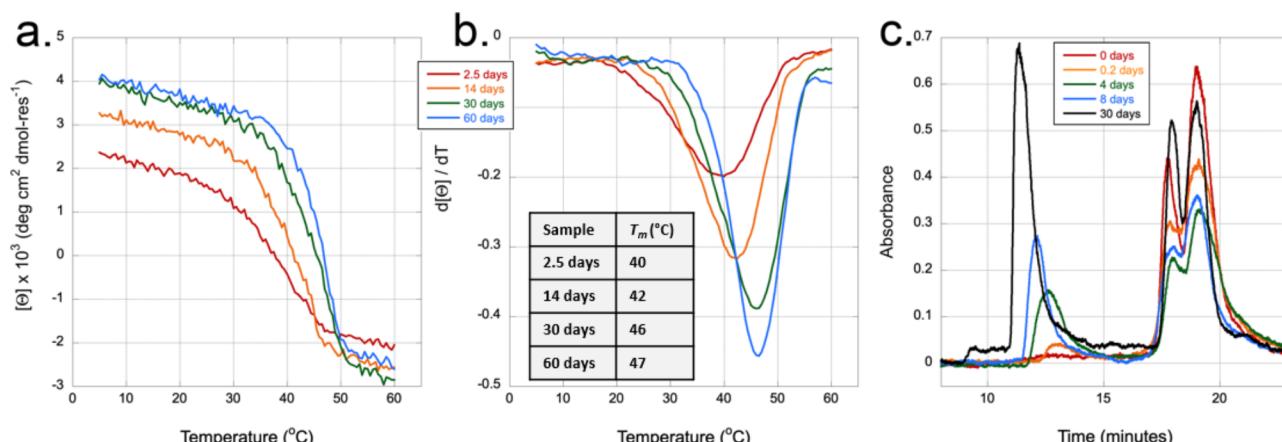
**Figure 5.** Sedimentation velocity analysis of A-Ala, B-Ala, and C-Ala peptides. (A) Individual peptides and mixtures of each pair of peptides in 150 mM NaCl (red: peptide A-Ala alone, orange: peptide B-Ala alone, magenta: peptide C-Ala alone, cyan: peptide mixture of A-Ala and B-Ala, yellow: peptide mixture of A-Ala and C-Ala, blue: peptide mixture of B-Ala and C-Ala). All peptides have approximately the same molar mass and sedimentation behavior and therefore are indistinguishable. Peptide pairs do not associate to form higher order structures. (B) Three different concentrations of all three peptides mixed together in 150 mM NaCl (187  $\mu$ M at 276 nm (green), 35  $\mu$ M at 230 nm (red), and 14  $\mu$ M at 220 nm (blue)). (C) A mixture of all three peptides in 150 mM NaCl (green), 0 mM NaCl (red), and 150 mM NaCl after heat treatment (blue). (In detail, the three-peptide mixture was diluted and heated in a 65  $^{\circ}$ C water bath for 20 min to fully denature the assembly and then cooled to room temperature for 2 h before measurements.) (D) Representative molar mass distribution for *s*-value distribution corresponding to the red trace (35  $\mu$ M concentration) in panel B. Molar mass data were determined by genetic algorithm–Monte Carlo analysis.

within 2 h. Finally, we examined the effect of ionic strength on the three-peptide mixture. The sample measured in pure water exhibited a slight heterogeneity of the larger species (Figure 5C). Genetic algorithm analysis detected a shift of the major peak to 4.38 s and additional minor species sedimenting at 4.87 s and 5.49 s. A similar shift was observed at each peptide concentration (data not shown). Due to the significant charge of the peptide, a measurement without the concentration of screening salts present in buffered samples may result in nonideal results; hence we did not investigate these changes in sedimentation further.

While our CD, SEC, and AUC mixing experiments demonstrate that an ABC composition is required for assembly, we are unable to determine the register of the three peptides with respect to one another in a triple helix. Predictions of thermal stability from our previously published algorithm “SCEPTTr”,<sup>9,10</sup> which are usually quite accurate, are extremely poor for this system of peptides ( $-2$   $^{\circ}$ C  $T_m$  predicted vs 47  $^{\circ}$ C  $T_m$  experimentally observed), which suggests that interhelical contacts are a major contributor in the overall stability of the system and also means that prediction of register cannot be made. This is a stumbling



**Figure 6.** Cryo-EM characterization of  $(ABC\text{-Ala})_6$ . (A) Raw micrograph of  $(ABC\text{-Ala})_6$  particles. Scale bar is 20 nm. (B) 2D class averages of the  $(ABC\text{-Ala})_6$  particles. (C–F) Reconstructed images of the assembly. (B, C) The assembly from the “bottom” and “top”, respectively illustrating the change in pore diameter, which increases from approximately 18 Å to 30 Å. (E) Side view of the assembly with approximate dimensions indicated. (F) Zoomed side view showing the interface between adjacent triple helices. Strands in the same triple helix are always in close proximity to each other, while adjacent triple helices make close contacts at the base of the crown and move further apart from each other near the top.

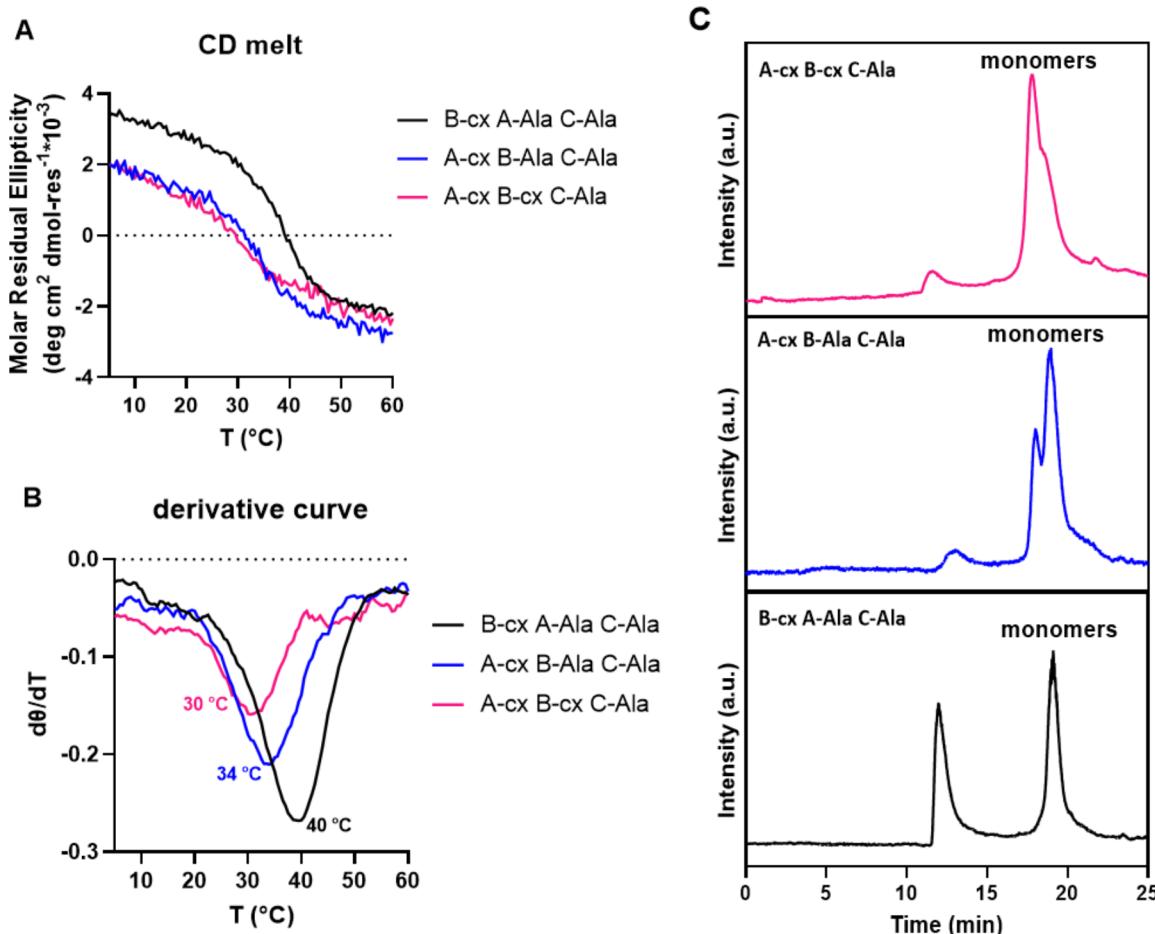


**Figure 7.** Extended time course of  $(ABC\text{-Ala})_6$  folding. (a) CD melting data: signal was monitored of samples with a concentration of 0.3 mM peptide in 10 mM Tris·HCl buffer at 225 nm versus time with a heating rate of 10 °C/hour. (b) Derivative of data presented in (a). Over time MRE and thermal stability increase, reaching a plateau after 30 days. (c) SEC from 0 to 30 days shows the slow increase in the proportion of oligomers as well as their increase in apparent mass over time.

block for future designed systems based on C1q, which we hope to overcome with future molecular level characterization.

**Structure of the  $(ABC\text{-Ala})_6$  Assembly.** We used cryo-electron microscopy (cryo-EM) to directly investigate the structure of the  $(ABC\text{-Ala})_6$  peptide assembly. Electron micrographs of the sample revealed small ring-like particles approximately 50 Å in diameter (Figure 6A). The ring-like structure was further evident from 2D class averages containing top views, which revealed 6-fold symmetry (Figure 6B). This structure was reconstructed with  $C_6$  symmetry imposed and filtered to  $\sim 5$  Å. Attempts to reconstruct with  $C_1$ ,  $C_2$ , and  $C_3$  symmetries yielded lower quality density maps. In density

maps viewed from the “bottom” (Figure 6C) or “top” (Figure 6D) six triple helical structures are evident, which form a supercoil around a hollow center. When looking at the three-dimensional density map from a “side view”, it is clear that the triple helices interact to form a crown-like structure (Figure 6E) which has a pore diameter of about 18 Å at the base and opens to approximately 30 Å at the top. While the strands in each triple helix are observed to be in close proximity with each other, adjacent triple helices only made close contacts at the narrower base of the crown and then moved further apart. The clear triple helical density that we observed (Figure 6C–F) represents less than half of the amino acids from each



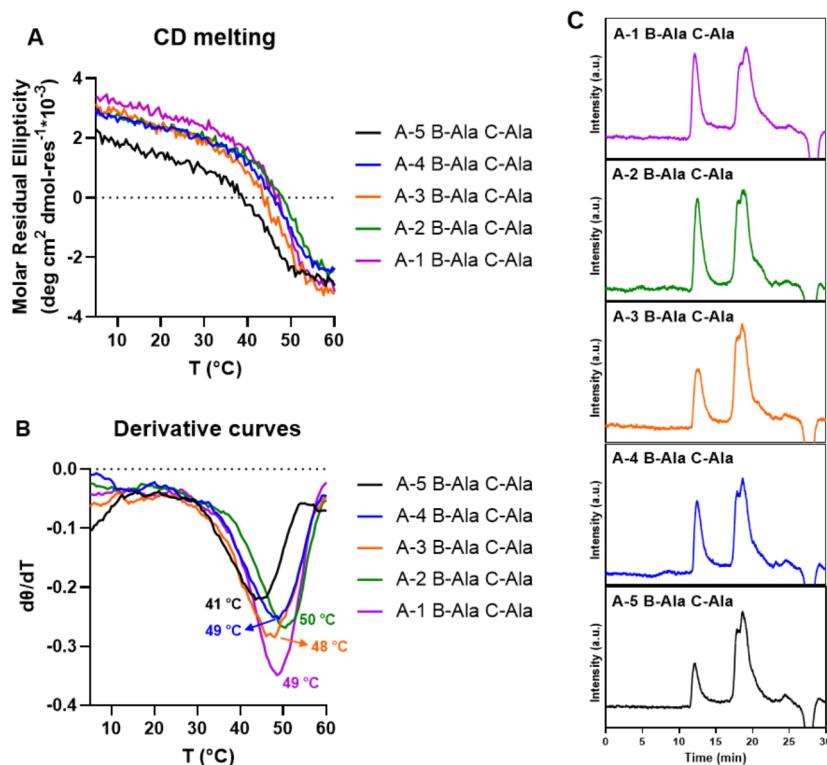
**Figure 8.** CD and SEC characterizations of self-assembled samples with varied N-terminal domains. (A) CD melting curves of supramolecular heterotrimers with or without the N-terminal fragment of A-Ala and B-Ala peptides; the melting curves were collected from 5 to 60 °C with a heating rate of 10 °C/hour at the wavelength that gives the maximum MRE value of each sample, sample concentration is 0.3 mM in 10 mM. (B) First-order derivative of the melting curves in (A). (C) SEC of assemblies with or without the N-terminal domains of A-Ala and B-Ala peptides. (Red) neither N-terminal domain, (blue) only B-Ala N-terminal domain, (black) only A-Ala N-terminal domain.

strand. As a result of the relatively modest resolution of the structure, as well as the small sequence coverage in the map, we are unable to definitely determine the handedness observed in the images. The hand imposed on the images in Figure 6 is based on the assumption of the right-handed helicity of the component triple helices, which are well known. Assuming this chirality, the crown itself appears as a left-handed super helix of right-handed triple helices. Strang et al. and Shelton et al. have previously performed electron microscopic studies of whole C1q that contains the full length of the collagenous and globular domains.<sup>48,65</sup> Their results suggest the complete C1q is a bouquet structure containing a stalk-like central region 3–6 nm in diameter with a length of 10–12.5 nm. Our results show a crown-like structure, which is consistent with their low-resolution rotary shadowing and negative stain TEM results.

**Rate and Mechanism of Folding.** The rate of assembly of the (ABC-Ala)<sub>6</sub> was monitored by CD over time (Figure S18B and Figure 7). The maximum near 225 nm, which corresponds to PP2 character, slowly increased over a very long time (one month), where it reached a plateau, demonstrating its very slow folding rate. Interestingly, the thermal unfolding profile of the system at different time points also changed with the  $T_m$  slowly increasing from just below 40 °C to its final value of 47 °C. SEC over a similar time frame initially shows only very small quantities of assembled material.

At no time do we observe assemblies corresponding to a simple triple helix by SEC (which elute near 15 min). Instead, the first observable oligomers correspond to at least hexamers with an elution time of 13 min. With extended equilibration time, both the quantity and the size of the oligomers increase, ultimately eluting just after 11 min, which matches expectations for the ultimate octadecamer assembly of (ABC-Ala)<sub>6</sub> (Figure 7c). This suggests that the rate-limiting step of assembly is triple helix formation. Once formed, triple helices are rapidly consumed into a higher order oligomer, which may be as small as (ABC-Ala)<sub>2</sub>. This explains the lack of observable ABC-Ala triple helix at any time point by SEC.

**Minimum Required Sequence.** Finally we set about reducing the length of the peptide to further isolate the minimum sequence necessary to form a stable triple helix and octadecamer. The N-terminal portion of peptides A-Ala and B-Ala contains a short domain that is not consistent with the Xaa-Yaa-Gly requirement of a triple helix (see Table 1, gray highlighted area), while C-Ala's sequence is entirely compatible with triple helix formation. In order to determine if this noncollagenous region is required for triple helix or oligomer assembly, two new peptides were prepared that eliminated the noncollagenous sequence, named A-cx and B-cx (A and B “collagen exclusive”). Mixtures of these peptides that are fully consistent with the sequence of a triple helix and those that



**Figure 9.** CD and SEC of the self-assembled samples from the deletion mutants of peptide A and B-Ala and C-Ala. The deletion mutants have different lengths of the N-terminal NC domain compared to peptide A. (A) CD melting curves. The samples assembled from the deletion mutants and peptides B-Ala and C-Ala all assembled into collagen triple helices and show thermal transitions. (B) Derivative curves of the melting curves in (A). (C) SEC of the deletion mutants' assemblies. Oligomers shown in all the assemblies.

contain additional N-terminal amino acids were made, thereby eliminating one or both N-terminal regions for a total of three new mixtures: A-cx/B-cx/C-Ala, A-cx/B-Ala/C-Ala, and A-Ala/B-cx/C-Ala. In all cases CD showed the formation of stable triple helices. However, melting data demonstrate that elimination of either or both noncollagenous regions results in moderate to significant thermal destabilization (Figure 8A,B). Of the two N-terminal noncollagenous regions, the one associated with C1q-A appears to be substantially more important than that associated with C1q-B. Elimination of the noncollagenous domain of C1q-B left the triple helix with only moderately reduced melting temperature (40 °C vs 47 °C). Furthermore, the same system retains strong oligomerization, as demonstrated by SEC (Figure 8C). In contrast, eliminating the noncollagenous domain of A or both noncollagenous domains results in a melting temperature of 34 and 30 °C, respectively, and elimination of most, but not all, high-mass oligomers. These results suggest that the noncollagenous domain of C1q-A plays an important, but not absolutely required, structural role in triple helix oligomerization.

To further investigate the importance of C1q-A's noncollagenous domain, we synthesized five additional mutants with different lengths of the noncollagenous domain (Table 1, N-terminal deletion series), mixed them for self-assembly with peptides B-Ala and C-Ala, and conducted similar CD and SEC experiments (Figure 9). Peptide mutant A-5 has the shortest length of the noncollagenous domain, including only one alanine residue. According to the CD melting results, the heterotrimer assembled from A-5, B-Ala, and C-Ala has a melting temperature of 41 °C, which is 7 °C more stable than

the heterotrimer from A-cx, B-Ala, and C-Ala. The A-5 assembly also oligomerized more effectively than the A-cx, B-Ala, and C-Ala assembly (Figure 9 C), indicating that just one additional amino acid results in a substantial improvement of self-assembly. Further increasing the length of the noncollagenous domain moderately increases the thermal stability of the assembly.

## CONCLUSION

In this work we have begun a dissection of the self-assembly mechanism of the defense collagen protein C1q. We found that well-controlled assembly of C1q can be maintained despite elimination of the majority of the native protein. Specifically, the stem region of C1q stretching from the N-terminus to the Xaa-Yaa-Gly discontinuity is sufficient to allow well-controlled self-assembly of an octadecamer. Self-assembly does not occur to an appreciable extent in the absence of any of the three peptides, supporting the requirement of the formation of an ABC heterotrimeric helix. Disulfide bonds are not required for the formation or stabilization of this system, nor are posttranslational modifications other than hydroxyproline. Cryo-EM reveals a beautiful crown-like, hollow, hexameric assembly that perfectly matches expectations from SEC and AUC. Individual peptide strands are able to be visualized that form the characteristic triple helix, and six of these helices form a supercoil around a large hollow pore. The overall structure resembles a crown in that the triple helices are tightly packed on one end and open up on the other, changing the pore diameter from approximately 18 Å to 30 Å. In two of the three peptides there is a short sequence that does not conform to the Xaa-Yaa-Gly pattern at the N-terminus. While oligomerization

is maintained when both of these regions are deleted, the stability and fraction of oligomer formed are substantially reduced without them. This appears to be primarily driven by the N-terminal sequence associated with C1q-A. Self-assembly of the C1q stem peptides into the final octadecamer is a very slow process requiring approximately one month of folding. The rate-limiting step appears to be triple helix formation, which then leads to a series of higher order oligomers, with increasing stability that ultimately terminates with the final formation of the  $(ABC)_6$  octadecamer.

This study helps to dissect the mechanisms of self-assembly of this critical protein family involved in innate immunity and is also the first example of a triple helical peptide that can self-assemble into a discrete higher order structure without significant polydispersity. Many of the previous structural studies looking at collagen and collagen-like higher order assemblies have been based upon very low resolution TEM studies,<sup>66</sup> which provide few insights into the molecular packing of triple helices. Additionally, most of the structural models for collagen packing are based on X-ray fiber diffraction data combined with X-ray crystallography.<sup>67,68</sup> The crystallographic studies do not necessarily reflect the solution state of these macromolecules, and the fiber diffraction data can be consistent with many different models.<sup>69</sup> The structure described in this paper is a critical first step in understanding the molecular interactions between collagen triple helices. We expect that these results will lay the groundwork for the design and synthesis of *de novo* designed triple helical bundles.

## ■ ASSOCIATED CONTENT

### Data Availability Statement

Cryo-EM density map for the  $(ABC\text{-Ala})_6$  assembly is available from the Electron Microscopy Data Bank with accession code EMD-28960. The UltraScan software used to analyze the AUC data is open source and freely available from Github repository (<https://github.com/ehb54/ultrascan3>), the AUC data itself is available upon request from the UltraScan LIMS server at the Canadian Center for Hydrodynamics.

### ■ Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/jacs.2c12931>.

Methods and further experimental details ([PDF](#))

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### Notes

The authors declare no competing financial interest.

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