

1 Review

2

Ag–NHC Complexes in π -Activation of Alkynes

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7 **Abstract:** Silver–NHC (NHC = N-heterocyclic carbene) complexes play a special role in the field of 8 transition-metal-complexes due to (1) their prominent biological activity, and (2) their critical role 9 as transfer reagents for the synthesis of metal–NHC complexes by transmetalation. However, the 10 application of silver–NHCs in catalysis is underdeveloped, in particular, when compared to their 11 group 11 counterparts, gold–NHCs (Au–NHC) and copper–NHCs (Cu–NHC). In this Special Issue 12 on *Featured Reviews in Organometallic Chemistry*, we present a comprehensive overview of the 13 application of silver–NHC complexes in π -activation of alkynes. Functionalization of alkynes is one 14 of the most important processes in chemistry that is at the bedrock of organic synthesis. The recent 15 studies on silver–NHC complexes show significant promise as unique and highly selective catalysts 16 in this class of reactions. The review covers π -activation reactions catalyzed by Ag–NHCs since 2005 17 (the first example of π -activation in catalysis by Ag–NHCs) through December 2022. The review 18 focuses on the structure of NHC ligands and π -functionalization methods, covering the following 19 broadly defined topics: (1) intramolecular cyclizations; (2) CO₂ fixation; (3) hydrofunctionalization 20 reactions. By discussing the role of Ag–NHC complexes in π -functionalization of alkynes, the 21

22 Reader is provided with an overview of this important area of research and the role of Ag–NHCs to 23 promote reactions that are beyond other group 11 metal–NHC complexes.

24 **Keywords:** silver; Ag; N-heterocyclic carbenes; NHCs; alkynes; π -functionalization; π -activation; Ag– 25 NHCs, group 11 metals, gold–NHCs, copper–NHCs

26 1. Introduction

27 After the first synthesis of a silver–NHC complex in 1993 by Arduengo [1], the breakthrough 28 was achieved in 1998 by Lin and co-workers who found that Ag(I)–NHCs serve as efficient transfer 29 reagents for the synthesis of other NHC–metal complexes [2]. Owing to the ease of synthesis of Ag– 30 NHCs, their stability, and avoidance of free-carbenes, this reaction now represents one of the most 31 straightforward approaches to the synthesis metal–NHCs that has enormously contributed to the 32 widespread application of metal–NHC complexes in catalysis and biology [3,4].

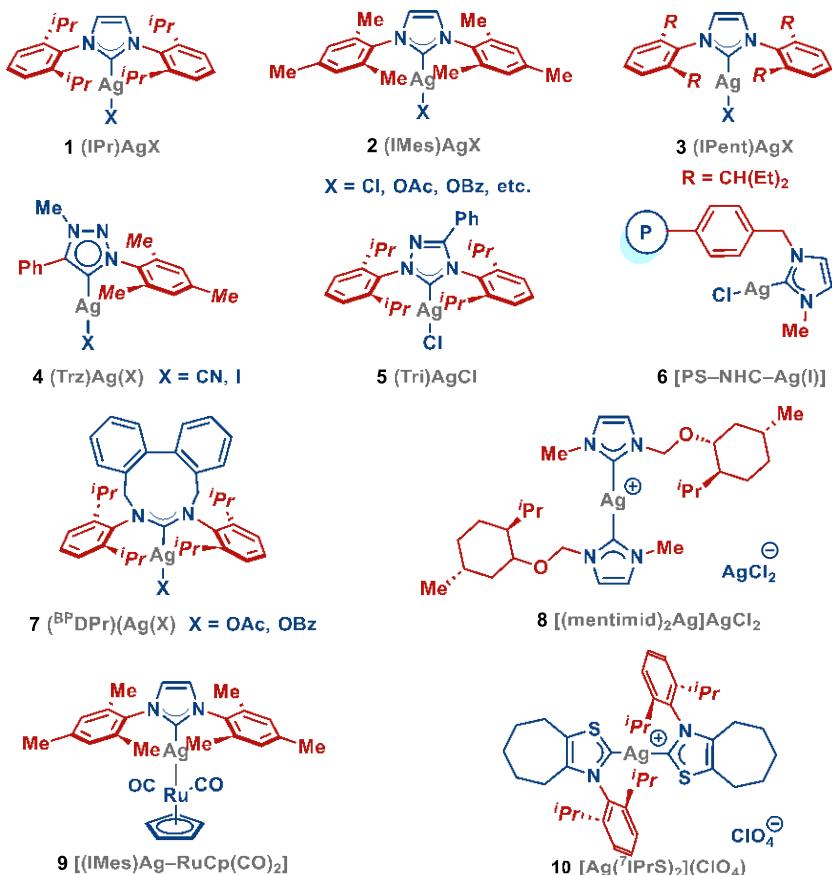
33 Simultaneously, another major direction in the area of Ag–NHCs was their investigation as 34 potential antimicrobial and anticancer agents, where Ag–NHC complexes have been proposed to 35 serve as slow releasing agents [5]. The area of silver–NHCs in medicinal chemistry is a vibrant area 36 of research with numerous reviews published in the last years [5,6].

37 Electronically, silver is a [Kr] 4d¹⁰s¹ coinage metal [7] with the current market price [8] 38 significantly lower than gold (Ag, \$24.00, 1 oz vs. Au, \$1,815, 1 oz), but higher than copper (Cu, \$3.73, 39 1 oz), which is mirrored by the relative abundance of group 11 metals in the Earth's crust (Ag, 0.07 40 pm; Cu, 50 ppm; Au, 0.0011 ppm) [9]. Complexation of strongly σ -donating NHC ligands to silver 41 enhances stability of silver, which can be exploited in catalysis [4,10]. Studies have found that 42 compared with other group 11 metals, Ag–NHC bond is longer (e.g., IPr–AgCl, 2.056 Å, Ag–Cl, 2.313 43 Å; IPr–AuCl, 1.941 Å, Au–Cl, 2.270 Å; IPr–CuCl, 1.881 Å, Cu–Cl, 2.106 Å) owing to the weaker π -44 donation of silver [11]. Furthermore, studies on π -activation of alkynes [12] established that π to σ 45 metal donation as well as metal to π^* back-bonding is in the order of Au > Cu > Ag [13]. Overall, these

46 mechanistic studies highlight that (1) silver–NHC complexes are well-suited as transmetalating 47 reagents, and (2) silver–NHC complexes are suitable for electrophilic activation of alkynes by π -48 coordination of cationic silver(I)–NHC to alkynes with electronic features complementary to other 49 coinage metals.

50 In this Special Issue on *Featured Reviews in Organometallic Chemistry*, we present a comprehensive 51 overview of the application of silver–NHC complexes in π -activation of alkynes. Several excellent 52 reviews on silver and silver–NHC complexes have been published [10]. These reviews have 53 addressed general aspects and applications of silver in organic synthesis [6,10]. A review addressing 54 specifically π -functionalization of alkynes by silver–NHC complexes has not been published thus far. 55 π -Functionalization of alkynes [12] is one of the most important processes in chemistry that is used 56 for the synthesis of a wide range of compounds in the areas ranging from drug discovery, through 57 agrochemistry, biochemistry and natural product synthesis to materials science [14]. The recent 58 studies on silver–NHC complexes show significant promise as unique and highly selective catalysts 59 in this class of reactions. The present review covers π -activation reactions catalyzed by Ag–NHCs 60 since 2005 (the first example of π -activation in catalysis by Ag–NHCs) through December 2022. The 61 review focuses on the structure of NHC ligands and π -functionalization methods. The review is 62 divided into the following sections: (1) intramolecular cyclizations; (2) CO_2 fixation; (3) 63 hydrofunctionalization reactions. Where relevant π -functionalization of olefins by silver–NHCs is 64 also discussed for comparison purposes or to introduce the topic from a historical perspective. We 65 hope that by discussing the role of Ag–NHC complexes in π -functionalization of alkynes, the Reader 66 will be provided with an overview of this important area of research and the role of Ag–NHC to 67 promote reactions that are beyond other group 11 metal–NHC complexes [7].

68 Structures of the most common Ag–NHC complexes used in π -functionalization of alkynes are 69 presented in Figure 1. Relevant bond lengths are presented in Table 1. For studies on electronic and 71 70 properties of NHC ligands, the Reader is encouraged to consult the following reviews [3,4].



72

73 **Figure 1.** Structures of the most common Ag–NHC complexes used in π -functionalization of alkynes.

74 **Table 1.** Ag–C_(carbene) bond lengths of the most common Ag–NHC complexes used in π -
 75 functionalization of alkynes. Ag–X bond lengths are shown for comparison.

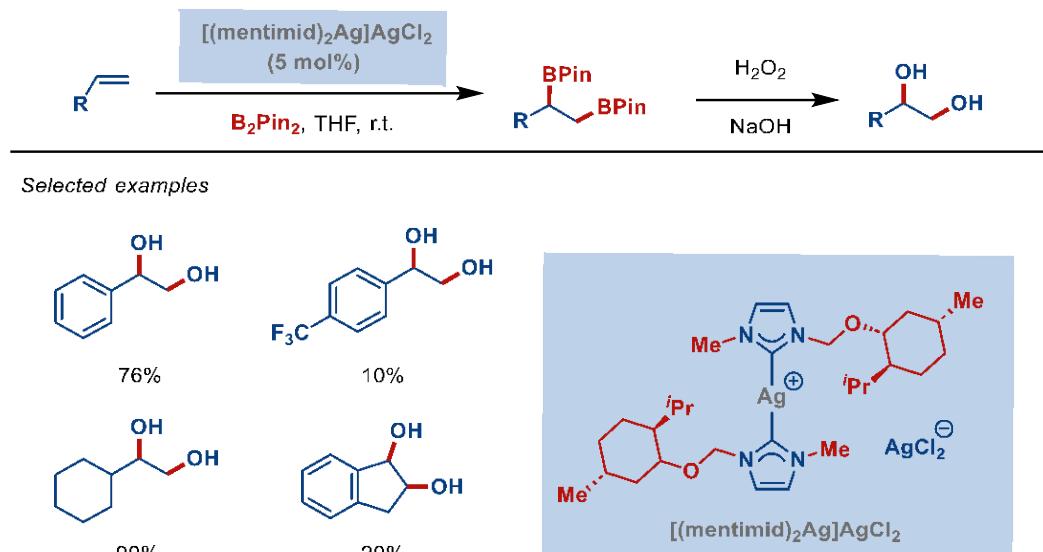
Entry	Complex No.	[(NHC)Ag(X)]	Ag–C bond length (Å)	Ag–X bond length (Å)	Reference
1	1	[(IPr)Ag(Cl)]	2.056	2.313 (X = Cl)	11d
2	2	[(IMes)Ag(Cl)]	2.056	2.314 (X = Cl)	11e
3	3a	[(IPent)Ag(OAc)]	2.067	2.111 (X = OAc)	17
4	3b	[(IPent)Ag(OBz)]	2.059	2.100 (X = OBz)	17
5	3c	[(IPent)Ag(4-CIOBz)]	2.064	2.100 (X = 4-CIOBz)	17
6	4a	[(Trz)Ag(CN)]	2.087	2.073 (X = CN)	24
7	4b	[(Trz)Ag(I)]	2.091	2.636 (X = I)	24
8	7a	[^{B^P} DPr)Ag(OAc)]	2.089	2.112 (X = OAc)	28
9	7b	[^{B^P} DPr)Ag(OBz)]	2.089	2.122 (X = OBz)	28
10	8	[(mentimid) ₂ Ag(AgCl ₂)]	2.102	2.952 (X = AgCl ₂)	15
11	9	[(IMes)Ag(RuCp(CO) ₂)]	2.111	2.617 (X = RuCp(CO) ₂)	32
12	10	[Ag(⁷ IPrS) ₂] ⁺ [ClO ₄] ⁻	2.081	-	19

76

77 It should be noted that Ag–C_(carbene) distances of the complexes shown in Figure 1 range between
 78 2.056–2.091 Å for monomeric complexes and between 2.102–2.111 Å for bimetallic complexes. As 79
 expected, there is a much greater variation in the Ag–X bond lengths, with values ranging from 2.073–80
 2.636 Å for monomeric complexes and 2.617–2.952 Å for bimetallic complexes. These distances in the 81
 most common NHC–Ag complexes used in π -activation of alkynes are consistent with the major role 82 of
 the counterion on their reactivity. Likewise, it should be noted that the three orbital contributions 83 to the
 Ag–NHC bond involve σ -donor, $d \rightarrow \pi^*$ (Ag to NHC π^* -backdonation) and $\pi \rightarrow d$ (NHC to Ag 84 π -
 donation). All three contributions should be considered in understanding the properties of Ag–85 NHC
 complexes in catalysis, which depend on the nature of the NHC scaffolds. The most effective 86 Ag–NHC
 complexes discovered to date in π -activation of alkynes are sterically-hindered IPent, half-87 umbrella
 shaped thiazol-2-ylidene ⁷IPrS and heteroatom-substituted 1,2,4-triazolylidene Tri NHC 88 ligands.
 Future studies should carefully address the role of orbital contributions in elucidating the 89 reactivity of
 Ag–NHCs in catalysis.

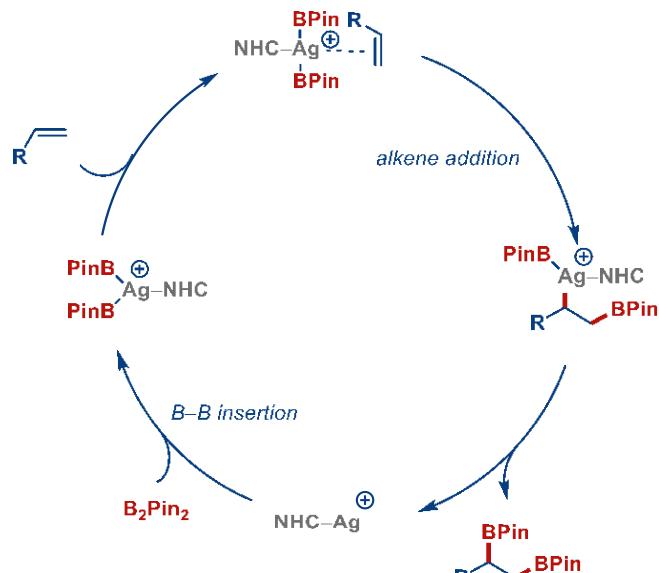
90 2. Intramolecular Cyclizations

91 The first application of silver–NHC complexes in catalysis was reported by Fernandez and Peris
 92 in 2005 in catalytic diboration of alkenes (Scheme 1) [15]. The authors developed a menthol-based
 93 bis-metallic Ag(I)–NHC complex, [(mentimid)₂Ag]AgCl₂, which provided relatively high reactivity
 94 in diboration of terminal and activated internal alkenes using B₂Pin₂. Although no asymmetric
 95 induction was observed, the authors demonstrated the beneficia effect of Ag(I)–NHC in that the
 96 analogous Ag(I)–phosphine and cationic Ag⁺ salts were completely unreactive. The high reactivity
 97 was ascribed to the combination of strong σ -donation allowing to break the B–B bond and low
 98 propensity of Ag(I)–NHC in β -hydride elimination of alkyl–boryl intermediate due to low π -back-99
 bonding from Ag(I)–NHC. The proposed catalytic cycle is presented in Scheme 2. The key step 100
 involves insertion of the Ag–NHC complexes into the B–B bond to afford diboryl species, which 101
 undergoes alkene diborylation. An improved Ag(I)–NHC catalyst system was subsequently reported 102
 by the same authors (Scheme 3) [16]. This 2005 study set the stage for the exploration of Ag(I)–NHCs
 103 as efficient catalysts for electrophilic activation of alkynes.
 104



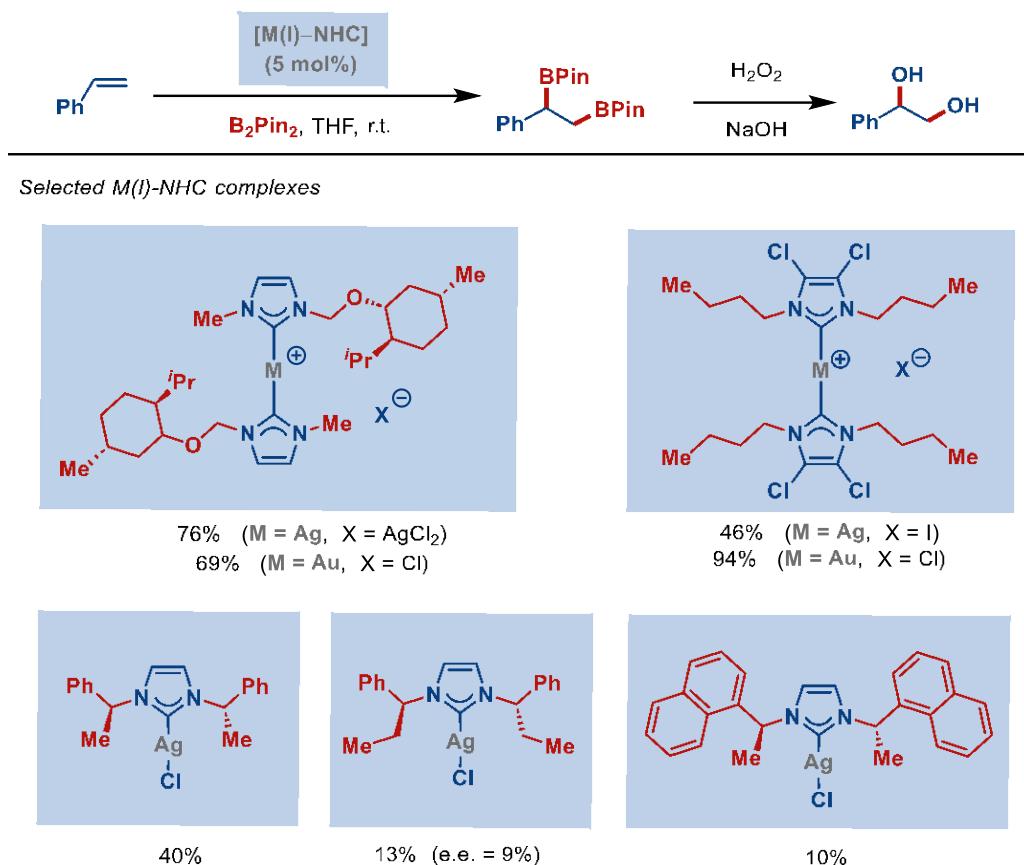
105

106 **Scheme 1.** Ag(I)-NHC-Catalyzed Diboration of Terminal and Activated Internal Alkenes Using
 107 B_2Pin_2 Reported by Fernandez and Peris.



108

109 **Scheme 2.** The Catalytic Cycle of Ag(I)-NHC-Catalyzed Diborylation of Alkenes Reported by
 110 Fernandez and Peris.

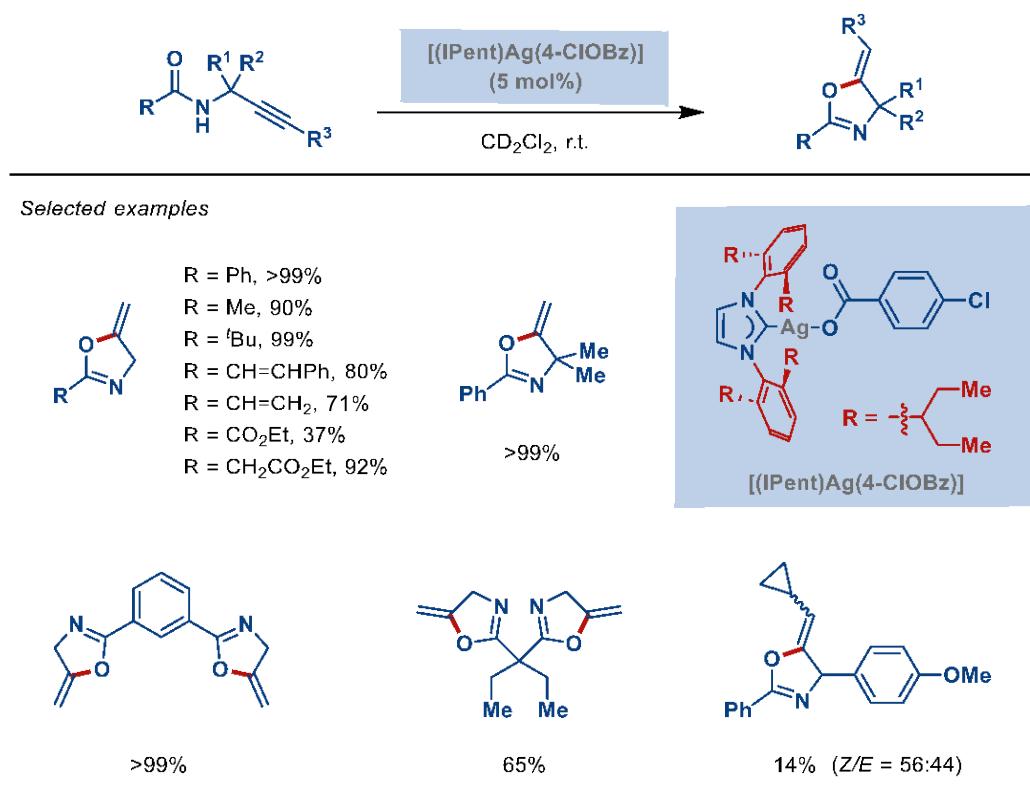


111 **Scheme 3.** Metal-NHC-Catalyzed Diboration of Internal Alkenes Using B_2Pin_2 Reported by
112 Fernandez and Peris.

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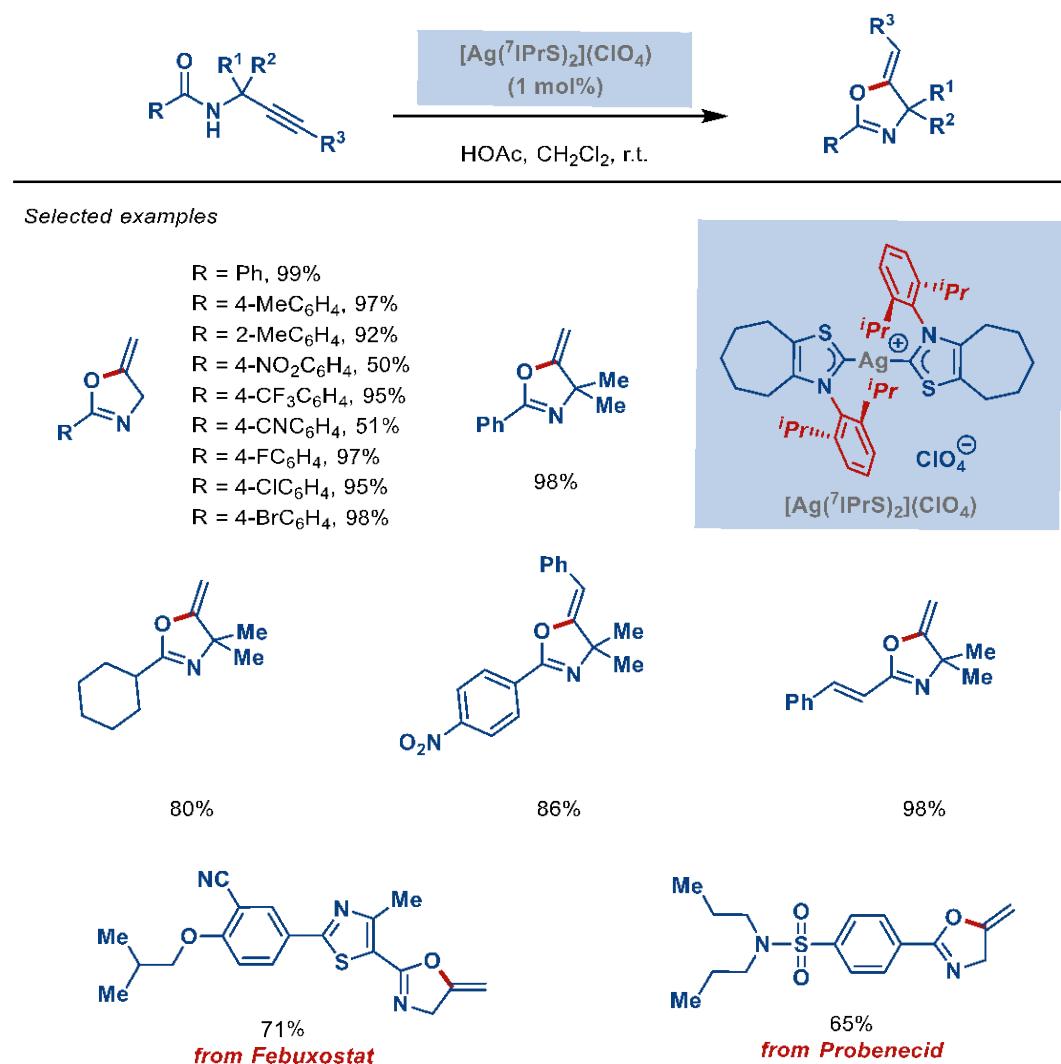
114 In 2016, Hii, Nolan and co-workers reported an impressive study on the effect of $Ag(I)$ -NHC
115 carboxylates, $[(NHC)Ag(O_2CR)]$, in intramolecular cyclization of propargylic amides to afford 116
117 oxazolidines (Scheme 4) [17]. The authors synthesized a series of $[(NHC)Ag(O_2CR)]$ complexes with 118
the goal of tuning electronic and steric properties of $Ag(I)$ -NHC complexes by the NHC ligand, and 119
their stability by the carboxylate ligand. The balance between the stability and activity of $Ag(I)$ -NHCs 120
is a major consideration in catalysis. The authors identified $[(IPent)Ag(4-ClOBz)]$ bearing a bulky 121
IPent 122 ligand and electron-deficient 4-Cl-OBz throw-away ligand as the most effective combination 123
for 124 catalysis. A range of propargylic amides was cyclized to oxazolidines under very mild room 125
temperature conditions at 5 mol% catalyst loading. The substrate scope of this intramolecular 126
cyclization was found to be complementary to Au-NHC catalysis [18], showcasing the synthetic 127
utility of $Ag(I)$ -NHC complexes in catalysis. An important finding of this study is the capacity to 128
independently tune sterics of the NHC ligand together with stability of the $Ag(I)$ -NHC complex by 129
carboxylate ancillary ligand.

130



128
 129 **Scheme 4.** Ag(I)-NHC-Catalyzed Intramolecular Cyclization of Propargylic Amides to Afford
 130 Oxazolidines Reported by Hii, Nolan and co-workers.

131 In 2022, we reported Ag(I)-thiazol-2-ylidene complexes and their application in the 132 intramolecular cyclization to afford oxazolidines (Scheme 5) [19]. The non-classical framework of 133 thiazol-2-ylidene offers new opportunities in catalysis due to differentiated half-umbrella-shaped 134 ligand structure and enhanced π -electrophilicity [20]. We found that these Ag(I)-thiazol-2-ylidene 135 complexes are highly active in the electrophilic cyclization of propargylic amides. These reactions 136 proceeded in excellent yields at room temperature in the presence of 1 mol% of the bis-NHC-Ag(I) 137 complex. The most reactive was $[\text{Ag}(\text{IPrS}_2)(\text{ClO}_4)]$ bearing a cycloheptyl thiazol-2-ylidene and 138 perchlorate anion. The reaction was applied to late-stage functionalization of pharmaceuticals, 139 showcasing the mild reaction conditions and potential applications in medicinal chemistry.
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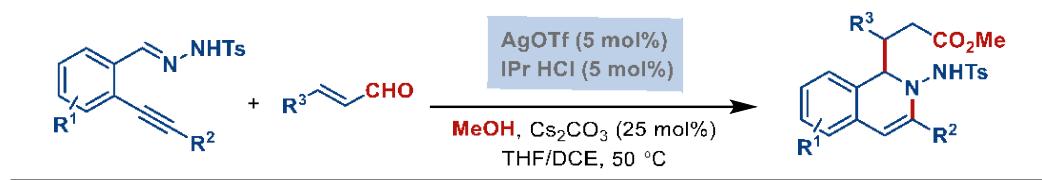


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142 **Scheme 5.** Ag(I)-NHC-Catalyzed Intramolecular Cyclization of Propargylic Amides to Afford
 143 Oxazolidines Reported by Zhang, Fang, Szostak and co-workers.

144 Another intramolecular cyclization involving Ag(I)-NHCs was reported by Wu and co-workers
 145 in 2010 (Scheme 6) [21]. In this work, the authors found that the combination of AgOTf and IPrHCl 146
 in the presence of Cs_2CO_3 as a base enabled a tandem tricomponent cyclization of N' -(2-147
 alkynylbenzylidene)-hydrazides with α,β -unsaturated aldehydes and methanol to afford 148
 functionalized 1,2-dihydroisoquinolines. The reaction proceeds via intramolecular cyclization of 149
 hydrazide onto the π -activated alkyne, followed by nucleophilic addition of homoenolate and 150
 methanol. The scope of the synthesis of the 2-amino-1,2-dihydroisoquinoline products is significant, 151
 permitting for the synthesis of medicinally-relevant heterocycles.

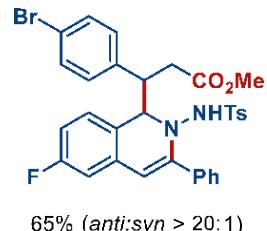
152 In 2019, Hashmi and co-workers reported a related approach based on Ag(I)-NHC-catalyzed
 153 intramolecular *6-endo-dig* cyclization to form 6-membered benzofused spirocycles (Scheme 7) [22].



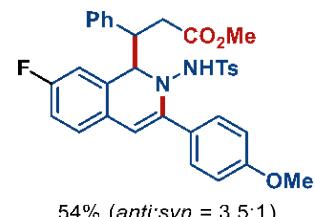
Selected examples



$R^3 = \text{Ph}$, 94%, *anti:syn* > 20:1
 $R^3 = 4\text{-BrC}_6\text{H}_4$, 78%, *anti:syn* > 20:1
 $R^3 = 4\text{-MeC}_6\text{H}_4$, 66%, *anti:syn* > 20:1
 $R^3 = 4\text{-MeOC}_6\text{H}_4$, 70%, *anti:syn* > 20:1
 $R^3 = \text{pyridin-3-yl}$, 83%, *anti:syn* > 13:1



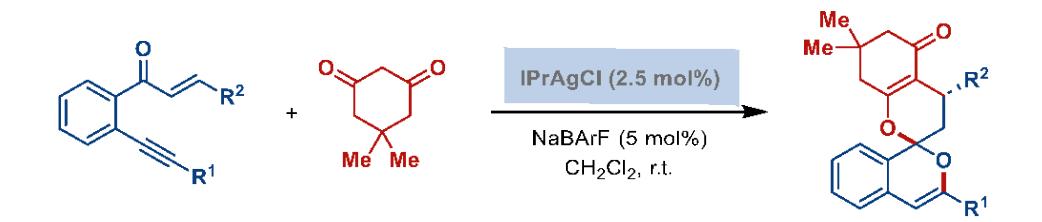
$R^3 = \text{Ph}$, 65% (*anti:syn* > 20:1)
 $R^3 = 4\text{-BrC}_6\text{H}_4$, 92% (*anti:syn* > 20:1)
 $R^3 = 4\text{-MeC}_6\text{H}_4$, 75% (*anti:syn* > 20:1)
 $R^3 = \text{pyridin-3-yl}$, 77% (*anti:syn* > 8:1)



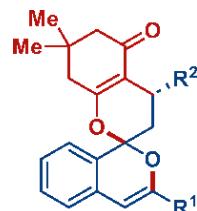
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155 Scheme 6. Ag(I)-NHC-Catalyzed Tandem TricOMPONENT Cyclization to Afford 2-Amino-1,2-Dihydroisoquinolines Reported by Wu and co-workers.

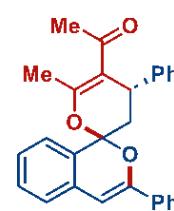
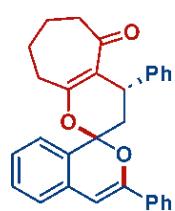
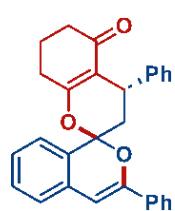
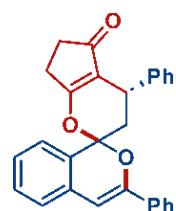
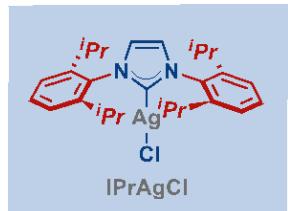
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Selected examples



$R^1 = \text{Ph}$, $R^2 = \text{Ph}$, 92%
 $R^1 = \text{Ph}$, $R^2 = 4\text{-MeOC}_6\text{H}_4$, 55%
 $R^1 = \text{H}$, $R^2 = \text{Ph}$, 64%
 $R^1 = \text{Ph}$, $R^2 = 4\text{-BrC}_6\text{H}_4$, 83%
 $R^1 = 4\text{-MeOC}_6\text{H}_4$, $R^2 = \text{Ph}$, 87%
 $R^1 = 4\text{-CF}_3\text{C}_6\text{H}_4$, $R^2 = \text{Ph}$, 93%
 $R^1 = \text{Ph}$, $R^2 = \text{H}$, 15%
 $R^1 = \text{Ph}$, $R^2 = 2,4,6\text{-Tri-MeC}_6\text{H}_2$, 74%



157

83%

158

159 Scheme 7. Ag(I)-NHC-Catalyzed Intramolecular 6-*endo-dig* Cyclization to Form 6-Membered Benzofused Spirocycles Reported by Hashmi and co-workers.

160

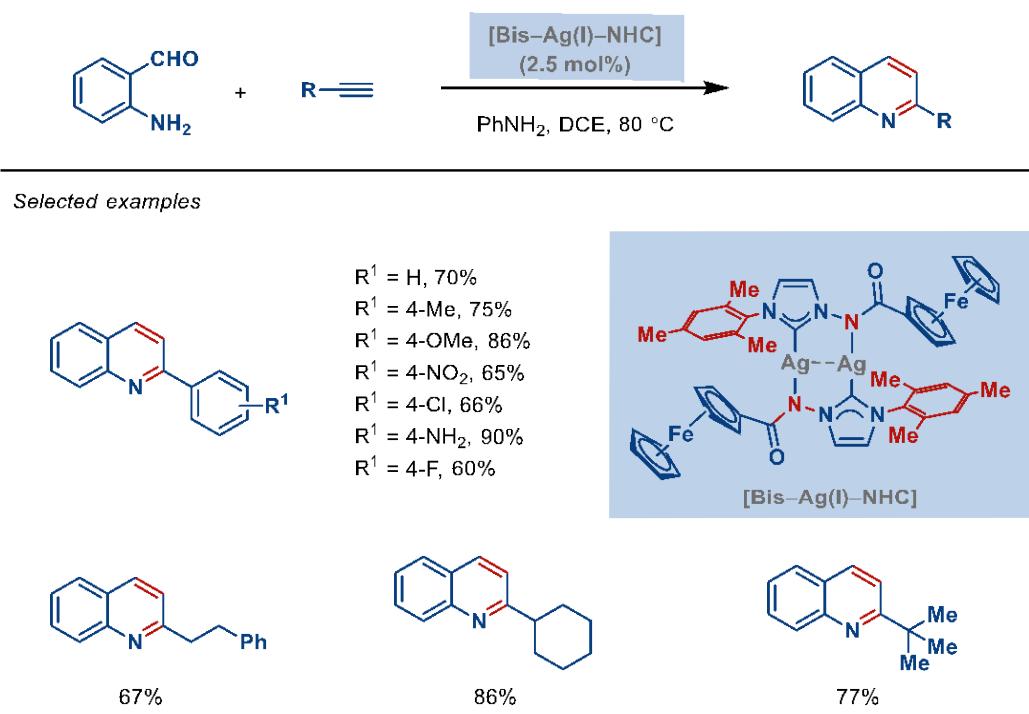
161 Mechanistically, the key step involves intramolecular cyclization of the carbonyl group onto π -activated alkyne, followed by intermolecular Michael addition and spirocyclization. The role of Ag-

162

163 NHC is two-fold: as an alkyne and enol π -activator to facilitate intramolecular cyclizations. The
 164 authors screened various group 11 metal catalysts for this intriguing spriocyclization and found that
 165 $[(\text{IPr})\text{AgCl}]/\text{NaBAr}^{\text{F}}$ is the most effective catalyst. This system outperformed simple Ag salts, such as
 166 AgNTf_2 , AgBF_4 or AgOTf as well as various Au and Cu catalysts, such as AuBr_3 , $[(\text{IPr})\text{AuCl}_3]$ or
 167 $[(\text{IPr})\text{CuCl}]/\text{NaBAr}^{\text{F}}$. The methodology is particularly notable for broad substrate scope and rapid,
 168 convergent approach to biologically privileged 6,6-spiroketals, highlighting the utility of Ag(I)-
 169 NHCs catalysis in the synthesis of O-heterocycles.

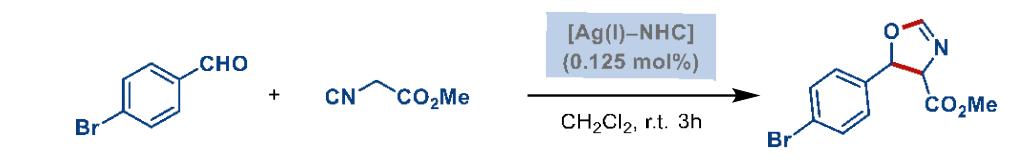
170 In 2013, Bera and co-workers reported an interesting synthetic approach to quinolines by Ag(I)-
 171 NHC catalysis (Scheme 8) [23]. In contrast to the approaches described in Schemes 3–6, this reaction
 172 involves Ag(I)-NHC-catalyzed alkyne hydroamination, followed by condensation with 2-
 173 aminobenzaldehyde. The catalyst used in this case is a bimetallic Ag(I)-NHC bridged by two anionic
 174 N-Mes/N-ferrocenoyl amide ligands. Although no information was provided on the comparative
 175 activity of other complexes, the scope of the method appears to be quite broad. The reaction delivers
 176 important 2-functionalized quinoline heterocycles in high yields by a three-component coupling. 177

178 In light of the reactions described above, it is important to mention Ag(I)-NHC-catalyzed
 179 synthesis of oxazolines from benzaldehydes and isocyanates reported by Albrecht in 2015 (Scheme
 180 9) [24]. This reaction features non-classical silver triazolylidene complexes prepared readily by the
 181 Lin method from Ag_2O and the corresponding triazolium salts. In the complex synthesis, the use of
 182 CH_3CN as a solvent resulted in C–C bond activation and the formation of $[(\text{trz})\text{Ag}(\text{CN})]$ complexes,
 183 while $[(\text{trz})\text{Ag}(\text{X})]$ complexes were formed in CH_2Cl_2 . These Ag(I)-trz complexes showed slightly
 184 higher reactivity than the analogous imidazol-2-ylidene Ag(I) complexes in the synthesis of
 185 oxazolines. The reaction was highly effective even at 0.10 mol% catalyst loading, showcasing the
 186 powerful role of Ag(I)-NHC in promoting intramolecular cyclizations.

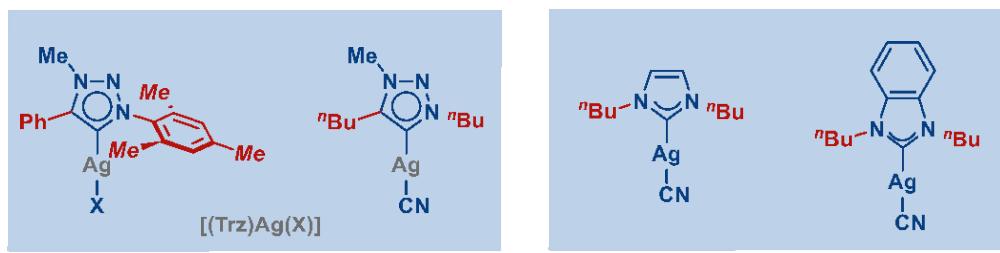


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Scheme 8. Bimetallic Ag(I)-NHC-Catalyzed Tandem Tricomponent Cyclization to Afford 2-Functionalized Quinolines Reported by Bera and co-workers.



Selected Ag(I) -NHC complexes



191 94% ($\text{X} = \text{CN}$)
92% ($\text{X} = \text{I}$)

192 95%

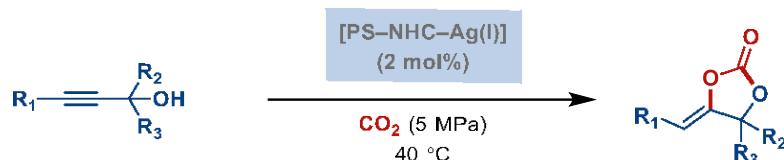
193 86%

194 84%

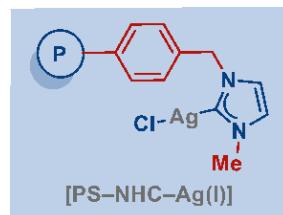
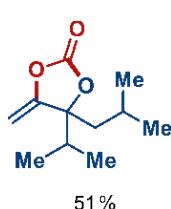
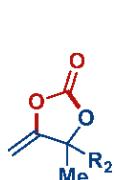
191 **Scheme 9.** Ag(I) -NHC-Catalyzed Synthesis of Oxazolines from Benzaldehydes and Isocyanates
192 Reported by Albrecht and co-workers.

194 **3. CO₂ Fixation**

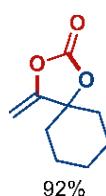
195 The ability to incorporate CO₂ in carbon–carbon bond forming reactions as a renewable C1
196 synthon is of great interest in organic synthesis [25]. In 2013, Jiang and co-workers reported 197
polystyrene supported Ag(I)-NHC complexes, [PS–NHC–Ag(I)], for CO₂ fixation into propargylic 198
alcohols (Scheme 10) [26]. The [PS–NHC–Ag(I)] complexes were readily prepared by the reaction of 199
appropriately substituted N-alkyl-imidazoles with polystyrene-supported benzyl chloride. The most 200
active was N-Me substituted complex. Interestingly, analogous Cu–NHC complex, [PS–NHC–Cu(I)], 201
showed no activity under the reaction conditions. These [PS–NHC–Ag(I)] complexes promoted the 202
carboxylative cyclization of a range of propargylic alcohols to terminal alkylidene cyclic carbonates 203
in generally excellent yields under 5 MPa pressure of CO₂ at 40 °C. This approach by Jiang and co-204
workers has several benefits, including high catalytic activity, ease of catalyst separation and catalyst 205
recyclability.



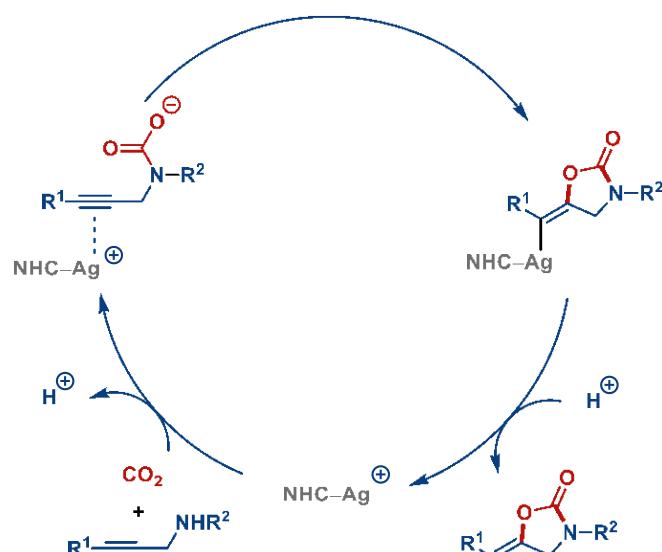
Selected examples



206 $\text{R}^2 = \text{Ph}$, 81%
 $\text{R}^2 = \text{Me}$, 96%
 $\text{R}^2 = \text{Et}$, 99%
 $\text{R}^2 = \text{iPr}$, 99%
 $\text{R}^2 = \text{iBu}$, 51%
 $\text{R}^2 = \text{nHex}$, 65%



207 **Scheme 10.** Polystyrene Supported Ag(I)-NHC-Catalyzed CO₂ Fixation into Propargylic Alcohols
208 Reported by Jiang and co-workers.



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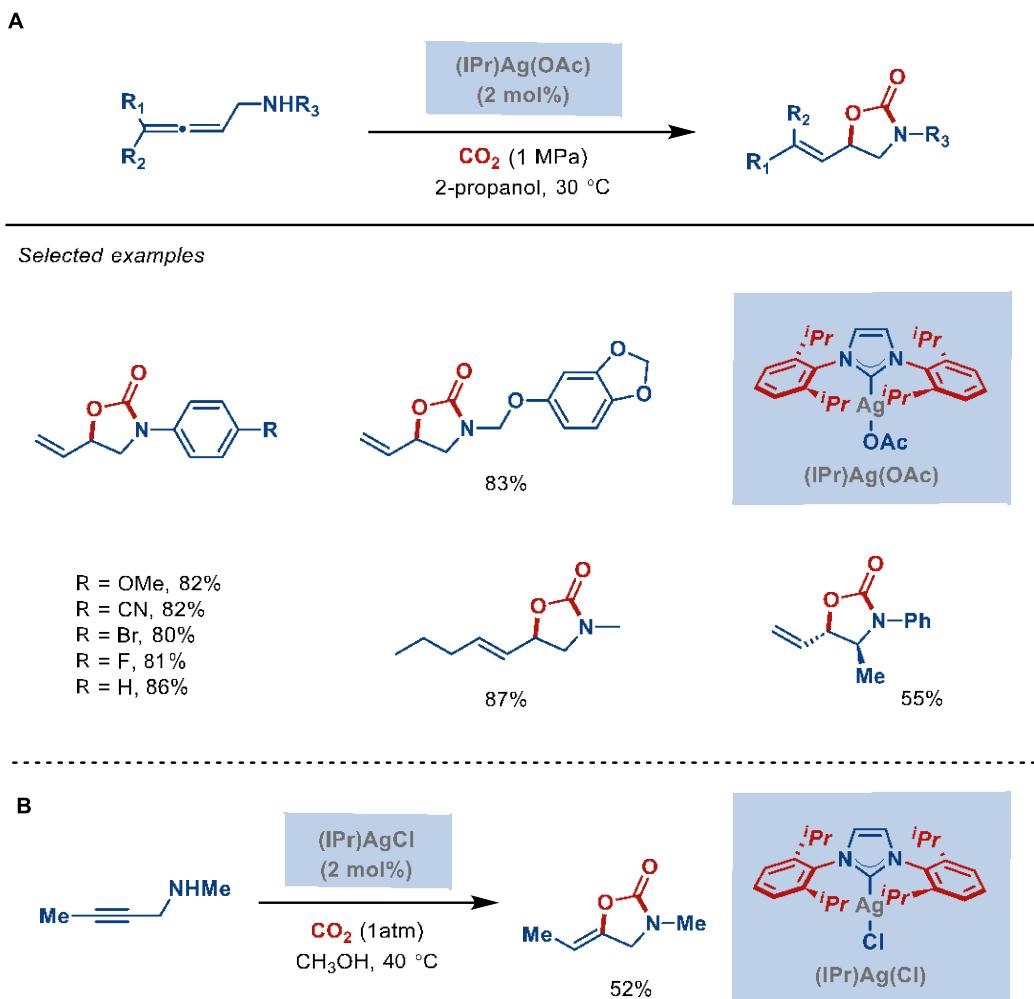
210 **Scheme 11.** The General Catalytic Cycle of Ag(I)-NHC-Catalyzed CO_2 Fixation to Propargylamines.

211

212 The proposed catalytic cycle is presented in Scheme 11. The reaction involves π -activation of the
 213 alkyne by cationic Ag-NHC species, followed by nucleophilic attack of the carbamate anion. 214
 Protonolysis regenerates the active Ag-NHC species. This represents a general mechanism for 215
 cyclization reactions mediated by Ag-NHC complexes.

216 In 2015, Ikariya and co-workers reported Ag(I)-NHC-catalyzed fixation of CO_2 into 217
 allenylmethylamines (Scheme 12A) [27]. The authors identified $[(\text{IPr})\text{Ag}(\text{OAc})]$ as the most effective 218
 catalyst to afford allenyl-1,3-oxazolidin-2-ones. The choice of metal, ancillary ligand and counterion 219
 was critical for this process. The analogous Au and Cu complexes, $[(\text{IPr})\text{Au}(\text{OAc})]$ and 220
 $[(\text{IPr})\text{Cu}(\text{OAc})]$, were completely ineffective, while $[(\text{IPr})\text{Ag}(\text{Cl})]$ showed minimal activity (<10%). 221
 The reaction showed good scope of allenylmethylamines at atmospheric pressure of CO_2 at 30 °C. 222
 Mechanistically, two competing pathways were proposed, carboxylative cyclization leading to 223
 allenyl-1,3-oxazolidin-2-ones and intramolecular hydroamination resulting in 2,5-dihydropyrroles, 224
 initiated by π -coordination to the internal or external allene double bond. Interestingly, the same 225
 group reported intramolecular carboxylative cyclization of propargylamines to alkylidene-1,3-226
 oxazolidin-2-ones mediated by $[(\text{IPr})\text{AgCl}]$ in modest yield (Scheme 12B).

227



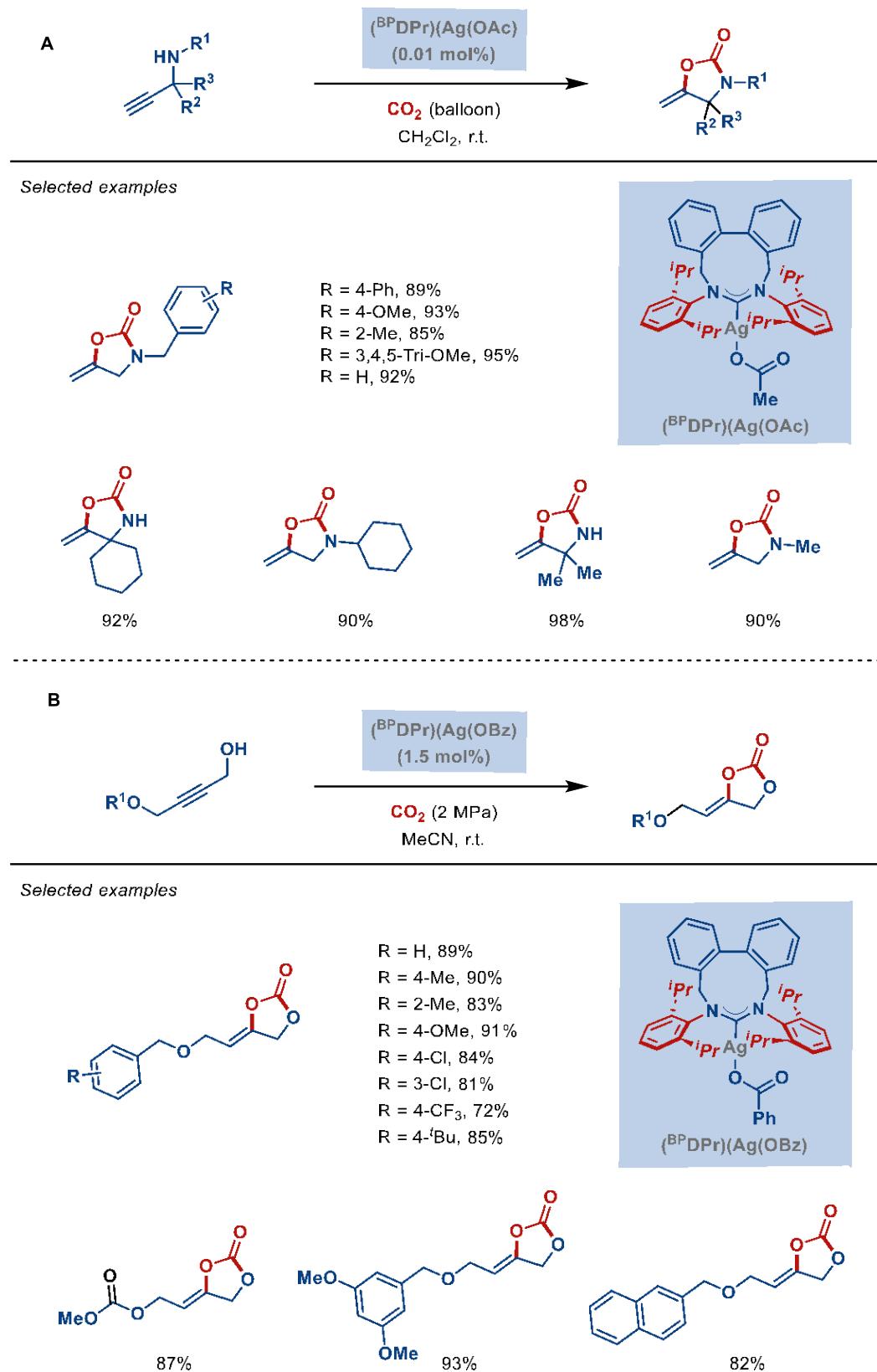
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229 **Scheme 12.** Ag(I)-NHC-Catalyzed CO₂ Fixation into Allenylmethylamines Reported by Ikariya and
230 co-workers.

231

232 In 2021, an important breakthrough was reported by Cervantes-Reyes, Hashmi and co-workers
233 in identifying ring expanded Ag(I)-NHC complexes as efficient catalysts for carboxylative cyclization
234 of propargylic alcohols and amines (Scheme 13) [28]. The most active complexes were 235
[(^{BP}DPr)(Ag(OAc)] and [(^{BP}DPr)(Ag(OBz)] featuring a nine-membered, bulky NHC ligand (^{BP}DPr = 236 1,3-
237 bis(2,6-diisopropylphenyl)-1,3-diazonine-2-ylidene) and carboxylate counterions. These
238 complexes are characterized by some of the largest buried volumes reported for [(NHC)AgX] 239
complexes to date, [(^{BP}DPr)(Ag(OAc)]: %V_{bur} = 52.9%; [(^{BP}DPr)(Ag(OBz)]: %V_{bur} = 54.5%). The scope 240
of the carboxylative cyclization mediated by these ring-expanded NHCs is particularly broad, which 241
has been ascribed to the steric distribution of the ligand on the metal center. The highlight is the 242
ability to promote carboxylative cyclization of unsubstituted propargylic alcohols and amines to 243
afford terminal and internal unsubstituted oxazolidinones and cyclic α -methylene carbonates in 244
excellent yields.

244



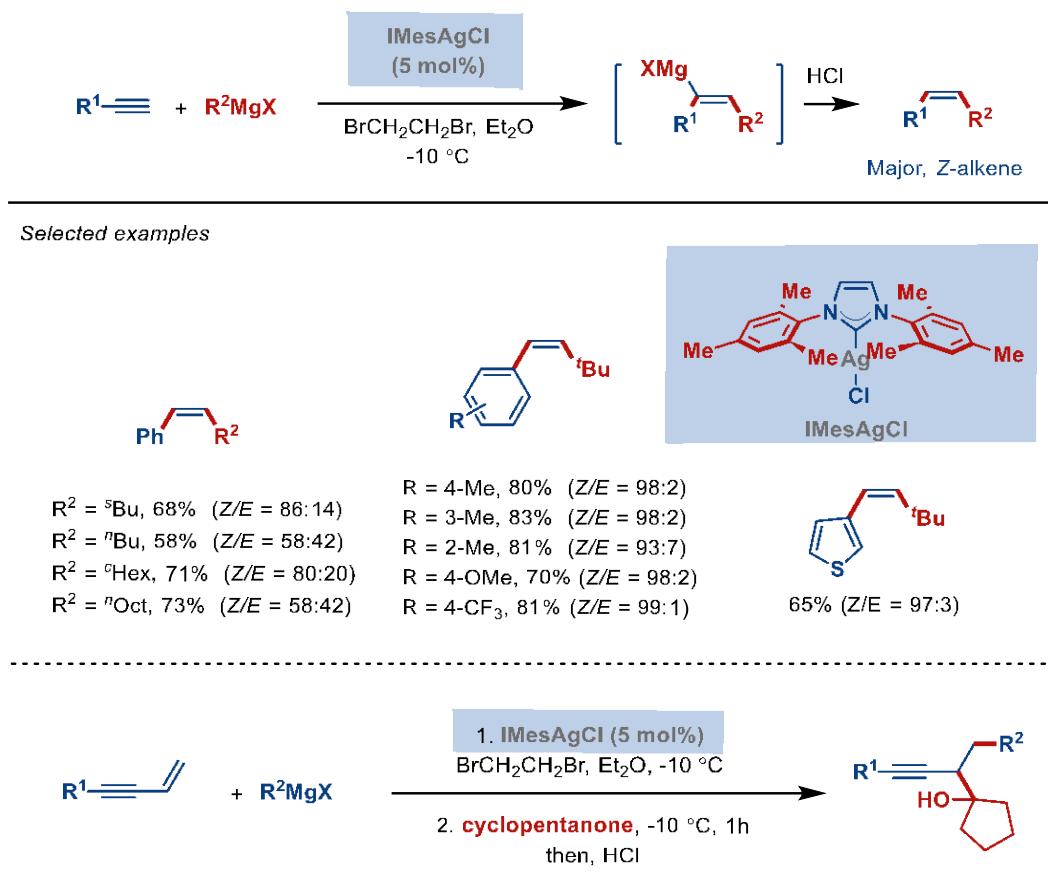
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246 **Scheme 13.** Ag(I)-NHC-Catalyzed Carboxylative Cyclization of Propargylic Alcohols and Amines
 247 Reported by Cervantes-Reyes, Hashmi and co-workers.

248 4. Hydrofunctionalization Reactions

249 The catalytic hydrofunctionalization of alkynes is among the most useful transformations in
 250 organic synthesis. [29] In 2009, Kambe and co-workers reported carbomagnesiation of alkynes

251 catalyzed by Ag(I)-NHC complexes (Scheme 14) [30]. This reaction proceeds in the presence of
 252 [(IMes)AgCl] as a catalyst, alkyl Grignard reagent as a nucleophile and BrCH₂CH₂Br as a 253
 254 stoichiometric additive. Ag-NHCs are the preferred catalyst over simple silver salts, such as AgOTs,
 255 and Ag-phosphine systems, such as AgOTs/PPh₃, affording higher yields and Z:E selectivity up to
 256 99:1. Mechanistically, the reaction involves the formation of an alkyl silver complex, followed by *anti*-
 257 alkyne insertion and transmetalation. The scope of this process is broad with respect to aryl alkynes
 258 using *t*-BuMgCl as a nucleophile. However, lower selectivity was observed with less sterically-259
 260 hindered Grignard reagents. The authors extended the utility of this process to
 carbofunctionalization of enynes and trapping with carbonyl electrophiles.



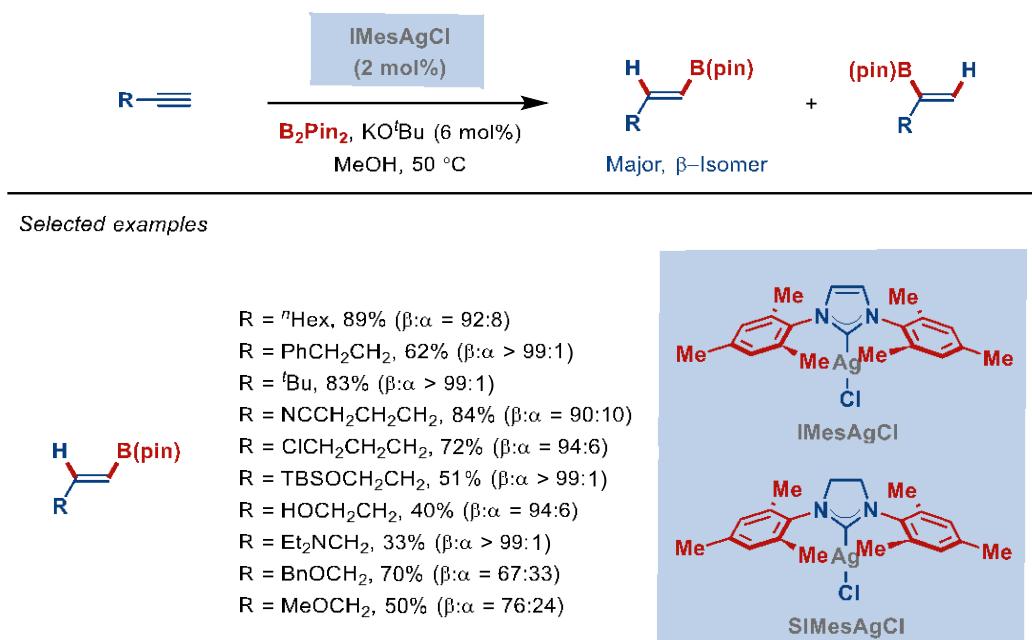
261

262 **Scheme 14.** Ag(I)-NHC-Catalyzed Carbomagnesiation of Alkynes Reported by Kambe and co-
 263 workers.

264

265 In 2014, a significant method for hydroboration of alkynes catalyzed by Ag(I)-NHC complexes
 266 was reported by Yoshida and co-workers (Scheme 15) [31]. [(IMes)AgCl] in the presence of catalytic
 267 KOtBu and B₂Pin₂ (1 equiv) in MeOH at 50 °C was identified as the optimal system for this 268
 transformation. Interestingly, the imidazolin-2-ylidene analogue, [(SIMes)AgCl], showed almost 269
 identical reactivity, while the imidazol-2-ylidene counterpart, [(IPr)AgCl], was completely unreactive 270
 under the tested conditions. The scope of the reaction is very broad and involves terminal aliphatic 271
 alkynes and internal aromatic alkynes. The yields and selectivity for the formation of β-272
 hydroboration products are generally high to excellent. Mechanistically, the key step is the formation

273 of [(IMes)Ag-BPin] species by σ -metathesis between [(IMes)Ag-OtBu] and B₂Pin₂. This [(IMes)Ag-BPin] species adds across the alkyne bond to generate β -boryl-organosilver, which undergoes 274 protonolysis. 275
276



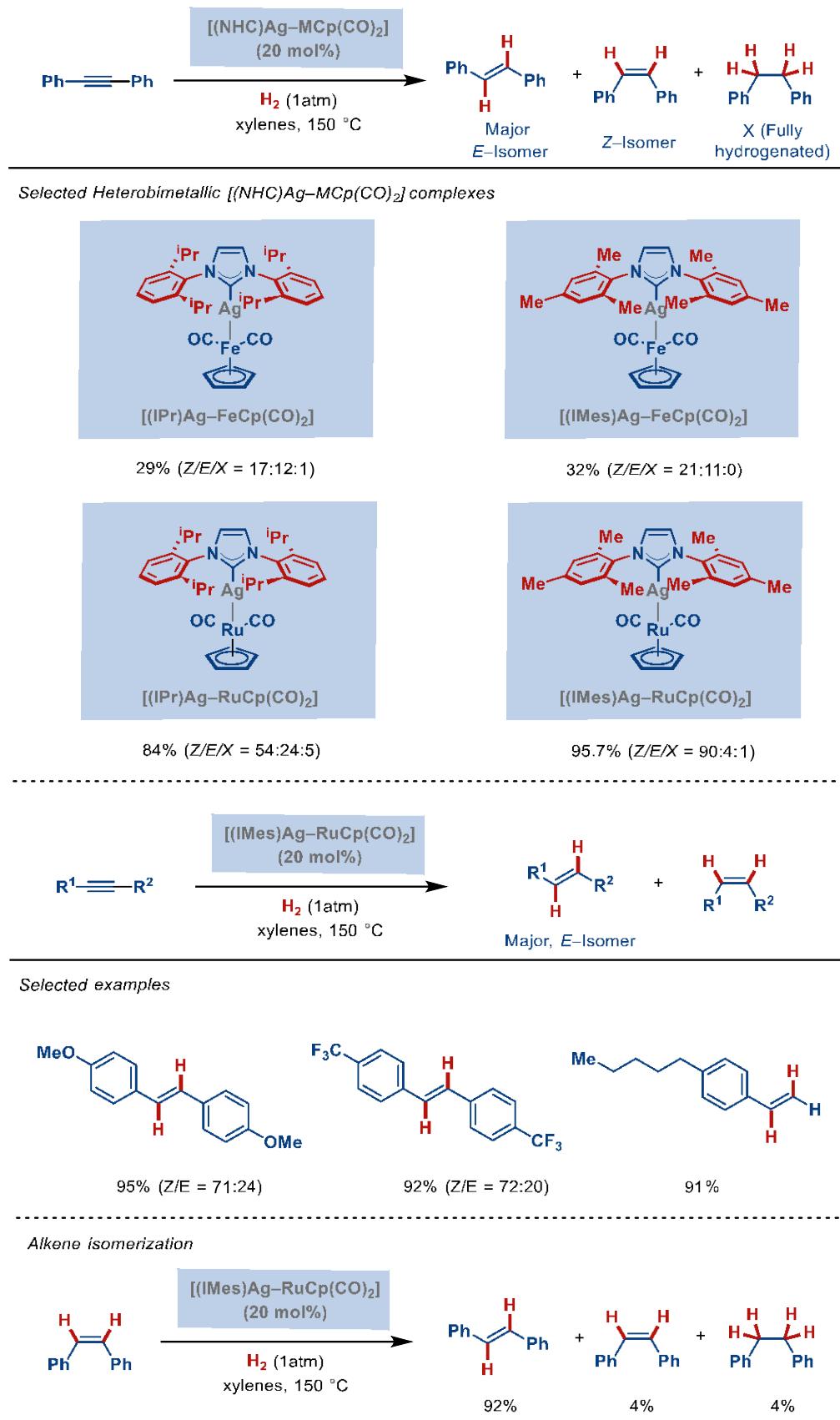
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278 **Scheme 15.** Ag(I)-NHC-Catalyzed Hydroboration of Alkynes Reported by Yoshida and co-workers.

279

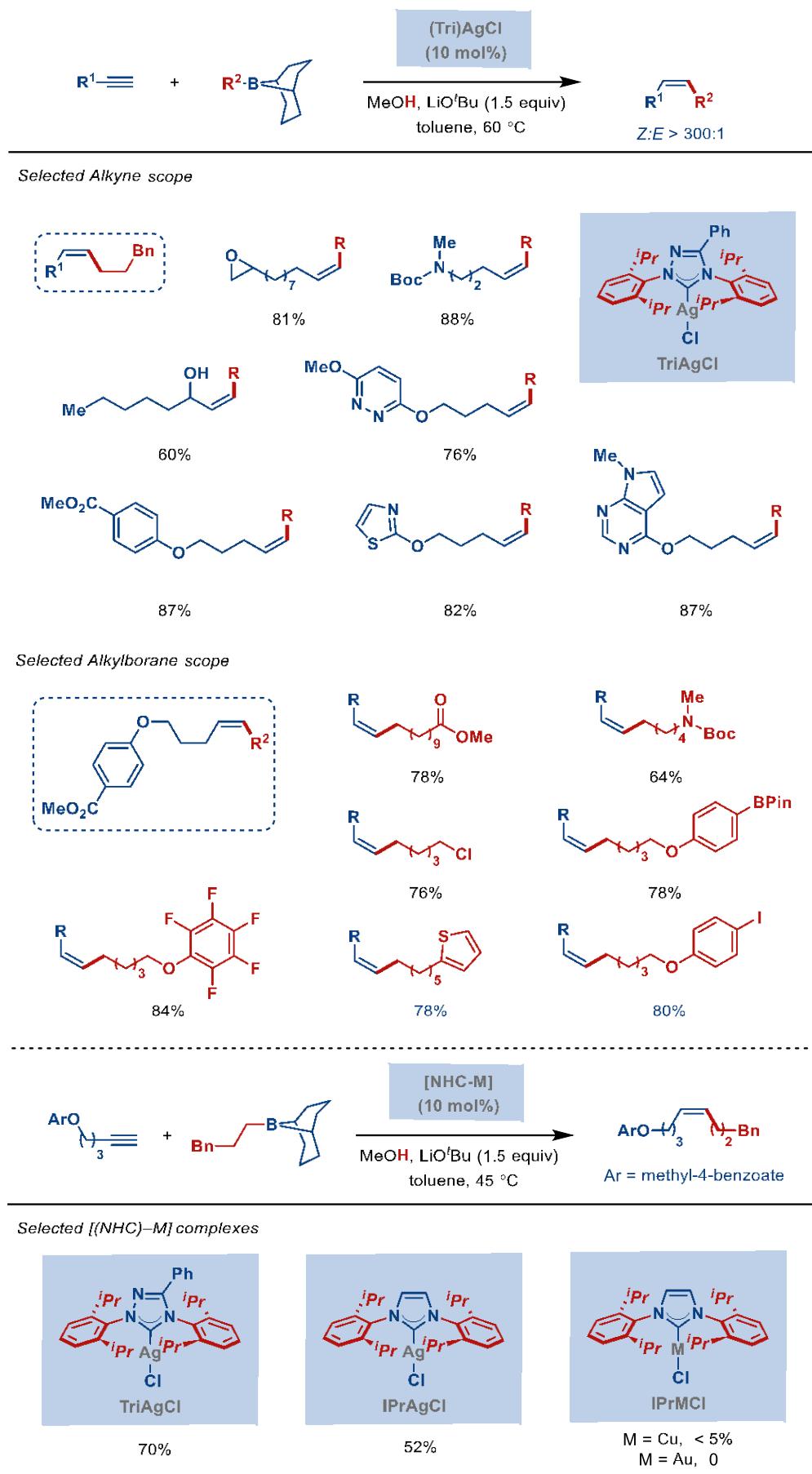
280 In 2014, Mankad and co-workers reported an intriguing *E*-selective hydrogenation of alkynes by 281 [Ag–Ru] bimetallic catalysis (Scheme 16) [32]. The most effective catalyst system is [(IMes)Ag– 282 RuCp(CO)₂] under atmospheric pressure of H₂ in xylenes at 150 °C. The bimetallic cooperation is 283 critical to this process as no reaction is observed by each of the catalyst systems alone. IPr and Cu as 284 well as FeCp(CO)₂ can be used, however, the yields and selectivity are lower than with [(IMes)Ag– 285 RuCp(CO)₂] complex. Mechanistically, bimetallic H₂ activation is followed by *syn*-alkyne insertion 286 into [(IMes)–Ag–H] to afford α -alkenyl–Ag(NHC) intermediate and protonolysis by [RuCp(CO₂)–H]. 287 The authors demonstrated that *Z/E* alkene isomerization takes place under the reaction conditions. 288 The 289 functional group tolerance of this method is broad as demonstrated by the Glorius robustness 290 test, 291 where only aldehydes were found to inhibit the reaction rate. This study demonstrates the 292 potential 293 of Ag–NHCs as an effective class of ligands in the emerging area of bimetallic catalysis [33]. 294 In 2019, 295 Lalic group reported hydroalkylation of alkynes catalyzed by Ag(I)-NHC complexes 296 (Scheme 17) 297 [34]. The most intriguing feature of this report is the use of 1,2,4-triazolylidene NHC, 298 [(Tri)AgCl] (Tri 299 = 1-phenyl-2,4-Dipp-1,2,4-triazolylidene) as the more effective ligand than the 299 [(IPr)AgCl]. Furthermore, the analogous Cu complex, [(IPr)CuCl] was 295 completely unreactive. 296 297 298 299 299 calculated values, DMSO) [35], which may contribute to the higher reactivity of 297 [(Tri)AgCl] vs. 298 299 299 functional group tolerance.

299



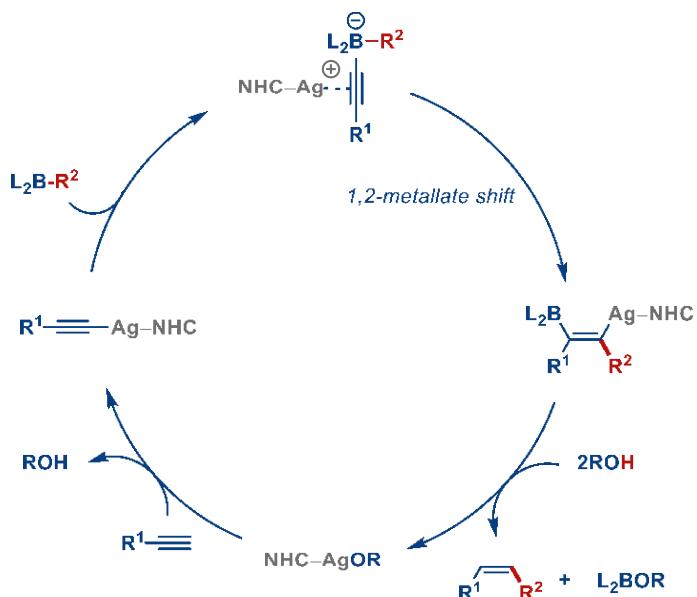
300

301 **Scheme 16.** Bimetallic [Ru-Ag(I)-NHC]-Catalyzed *E*-Selective Hydrogenation of Alkynes Reported
302 by Mankad and co-workers.



306 Mechanistically, the reaction involves the combination of σ - and π -activation of alkynes, which
 307 may further explain better reactivity of Ag–NHCs vs. Cu–NHCs. The authors proposed that silver 308
 309 acetylide reacts with alkyl borane, followed by 1,2-metalate shift after π -activation. The catalytic cycle 310
 311 is completed by protodemetalation and protodeboronation (Scheme 18). This report provides a clear 312
 inherently favored by silver.

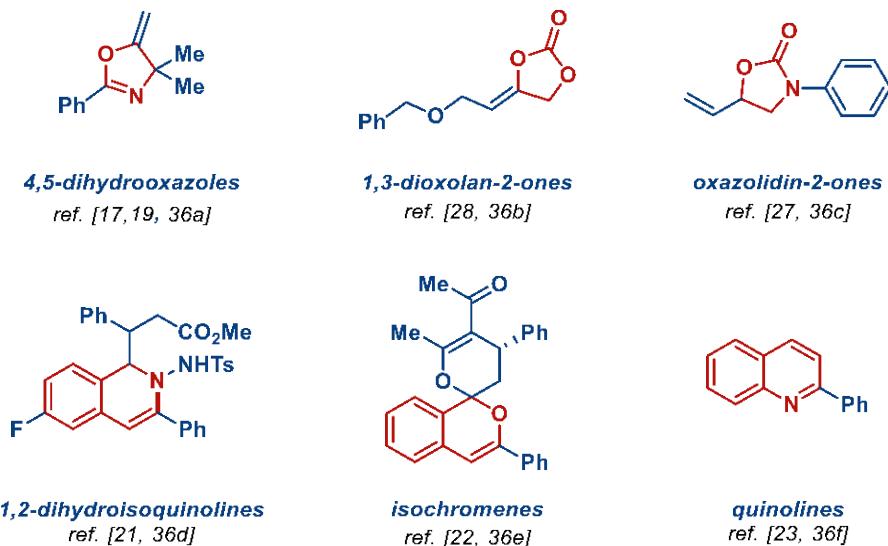
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313

314 **Scheme 18.** The Catalytic Cycle of Ag(I)-NHC-Catalyzed Hydroalkylation of Alkynes Reported by
 315 Lalic and co-workers.

316



317

318 **Figure 2.** Heterocyclic Scaffolds Synthesized by Ag(I)-NHC Complexes.

319 5. Conclusions and Outlook

320 In summary, in the past 15 years significant advances have been made in using Ag–NHC 321
 322 complexes for the synthetically important functionalization of alkynes. Among the major advantages 323
 324 of Ag–NHCs is the enhanced stability of silver rendered possible by strongly σ -donating NHC 325
 generic classes of reactions: intramolecular cyclizations, CO_2 fixation, and hydrofunctionalization

326 reactions. These reactions provide heterocyclic products important for medicinal chemistry research
327 and functionalized building blocks for organic synthesis. Among the reported reactions, the most
328 noteworthy are processes that specifically demonstrate the beneficial role of Ag–NHCs, such as
329 electrophilic cyclization of propargylic amides, CO₂ fixation, bimetallic Ag–Ru hydrogenation of
330 alkynes, and hydroalkylation of alkynes.

331 An interesting consideration is the fact that Ag–NHC complexes appear to be particularly well-
332 suited for the synthesis of heterocycles that are part of potential therapeutic agents (Figure 2) [36]. 333 This reactivity of Ag–NHCs in π -activation of alkynes bodes well for the broad practical application 334 of
this class of M–NHCs in medicinal chemistry research.

335 Despite significant progress, there are several areas that should be addressed in the future to
336 render this Ag–NHC manifold even of more general utility in organic synthesis: (1) mechanistic 337 studies are urgently needed to elucidate the role of Ag–NHCs in comparison with other group 11 338 metal–NHC complexes; (2) the role of NHC ligands has been rarely explored in Ag–NHC catalysis, 339 with majority of reactions limited to testing only IPr and IMes ligands; (3) although the mechanistic 340 basis for several alkyne functionalization manifolds using Ag–NHCs has been established, few 341 reactions have been explored using this catalysis manifold; (4) the role of counterion has not been 342 fully elucidated, with simple carboxylate anions typically preferred for activation of alkynes; (5) the 343 development of asymmetric processes using chiral Ag–NHCs has not been accomplished.

344 In the group of coinage metals, silver has several major advantages over Au and Cu, including
345 low price and availability (vs. Au), capacity to promote various activation modes, ease of 346 transmetallation, complementary electronic properties, such as π to σ metal donation and metal to π^* 347 back-bonding. The reported studies clearly demonstrate that researchers using group 11 metal–NHC 348 complexes should always consider Ag(I)–NHCs in the development of catalytic processes.

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