

Ir(III)-based Agents for Monitoring Cytochrome P450 3A4 Active Site Occupancy

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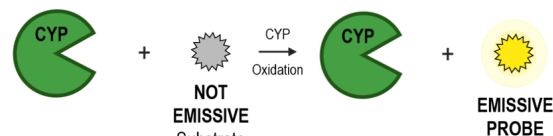
Supporting Information Placeholder

ABSTRACT: Cytochromes P450 (CYPs) are a superfamily of enzymes responsible for biosynthesis and drug metabolism. Monitoring the activity of CYP3A4, the major human drug-metabolizing enzyme, is vital for assessing metabolism of pharmaceuticals and identifying harmful drug-drug interactions. Existing probes for CYP3A4 are irreversible turn-on substrates that monitor activity at specific time points in end-point assays. To provide a more dynamic approach, we designed, synthesized and characterized emissive Ir(III) and Ru(II) complexes that allow monitoring of CYP3A4 active site occupancy in real time. In the bound state, probe emission is quenched by the active site heme. Upon displacement from the active site by CYP3A4-specific inhibitors or substrates, these probes show high emission turn-on. Direct probe binding to the CYP3A4 active site was confirmed by X-ray crystallography. The lead Ir(III)-based probe has nanomolar K_d and high selectivity for CYP3A4, efficient cellular uptake, and low toxicity in CYP3A4-overexpressing HepG2 cells.

Cytochromes P450 are crucial enzymes responsible for biomolecule synthesis and drug metabolism. Among 57 human CYPs, CYP3A4 is the major drug-metabolizing enzyme responsible for oxidizing the majority of pharmaceuticals.¹ Due to high substrate promiscuity and plasticity of the active site, CYP3A4 is implicated in many drug-drug interactions that can cause drug toxicity.^{2–5} Additionally, CYP3A4 displays genetic polymorphism where mutations facilitate or slow down drug metabolism, thereby affecting therapeutic efficiency.^{6–8} These attributes make CYP3A4 an important target for activity monitoring, especially in complex systems such as liver microsomes and hepatocytes that model human drug metabolism *in vitro*. Current methods for monitoring CYP3A4 activity involve marker substrates, which require cumbersome

and costly HPLC analyses conducted over multiple time points, or irreversible turn-on reagents that make it difficult to monitor

A. Irreversible monitoring of CYP3A4 due to oxidation by CYP.



B. Novel and reversible monitoring of CYP3A4 due to quenching by CYP.

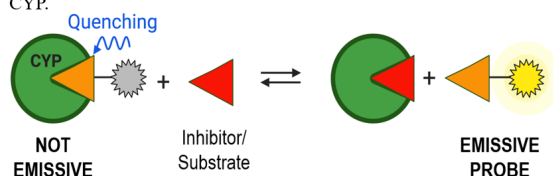


Figure 1. Emissive probes for monitoring metabolism by CYP3A4.

changes to CYP3A4 activity over time (Figure 1A).^{9–14} As an alternative approach to these classical methods, in this communication we report emissive Ir(III) and Ru(II) complexes that allow sensing of occupancy of the CYP3A4 active site (Figure 1B).

We chose to examine Ir(III) and Ru(II) complexes as probes for CYP3A4 because they are powerful tools for monitoring biological activity.^{15–24} Probes of this class have long luminescence lifetimes, ranging from hundreds of nanoseconds up to ~100 μ s,^{15,16,25} which allows for time-resolved gating that can be used to exclude background emission from biomolecules and fluorogenic substrates.

Thus, these compounds were expected to provide a distinct advantage over previous CYP3A4 probes containing organic-based fluorescent groups,²⁶ whose low nanosecond lifetimes preclude the measurement of CYP activity in human liver microsomes, the gold standard in drug metabolism.

Transition metal-based probes were designed to interact with a hydrophobic surface within the substrate access channel of CYP3A4,²⁷ and included a pyridyl side chain (see R_1 , Figure 2) to anchor the complex to the enzyme through direct heme iron coordination. Emissive sensors **2–5** were synthesized as racemic mixtures of Δ and Λ isomers (Figure 2A). Ligand **1** was heated with Ru(II) precursors *cis*-[Ru(L₁)₂Cl₂] (L₁ = 2,2'-bipyridine or 1,10-phenanthroline), which gave compounds **2** and **3**. Alternatively, treating **1** with [Ir(μ -Cl)(C[^]N)₂]₂ [C[^]N = 2-phenylpyridine (ppy) or 2-phenylquinoline (pq)] gave complexes **4** and **5**. Complexes **2–5** were characterized by ¹H NMR, IR and electronic absorption spectroscopies and electrospray mass spectrometry. All data were consistent with the structures shown in Figure 2. Importantly, electronic absorption and emission spectra for **2–5** were in good agreement with data for the parent Ru(II) or Ir(III) complexes devoid of the R_1 side chain.^{28–30} All complexes emit brightly when excited with 435 nm light (Figure S7), with **4** having the highest emission quantum yield of 0.086(9) and a remarkable lifetime of 1.6 μ s, over twice as long as **2** and **3** (Table 1).

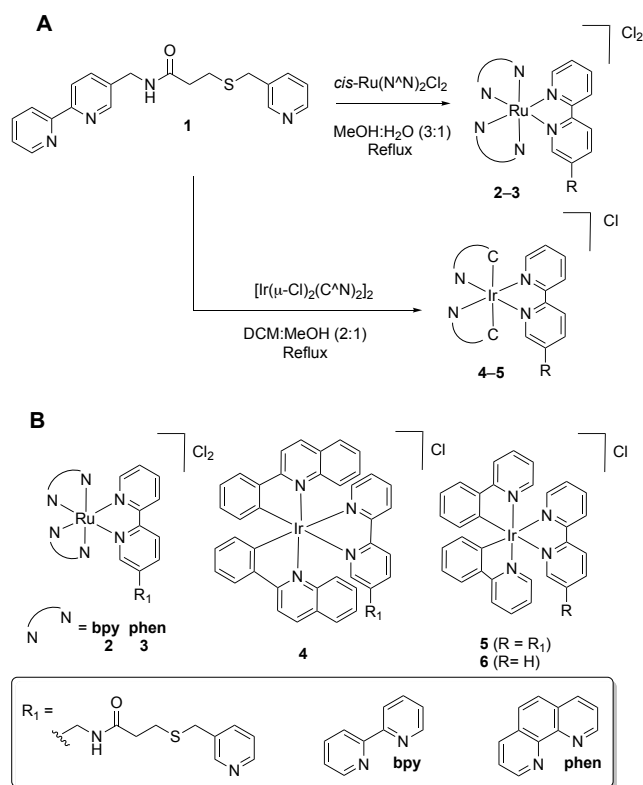


Figure 2. Synthesis (A) and structures (B) of Ir(III) and Ru(II) CYP3A4 photosensors **2–5**.

Equilibrium titration of CYP3A4 with **2–5** showed that all complexes exhibit type II binding, indicative of strong pyridine nitrogen coordination to the heme (Figures 3A, C–F). Spectral dissociation constants for **2–5** are listed in Table 1. Complexes **4** and **5** are far more potent than Ru(II) inhibitors **2** and **3**, indicating that CYP3A4 preferably binds monocationic over dicationic complexes. Importantly, attachment of the R_1 side chain dramatically increases the inhibitory potency, by nearly 100-fold. Control compound **6** shows type I binding (blue shift in the Soret band) and is a weak

inhibitor with a K_d of 11.2 ± 0.08 μ M, whereas analog **5** with the pyridyl containing R_1 chain exhibits type II binding, with a stronger affinity of 130 ± 11 nM. Both the binding affinity determined from the equilibrium titrations and the IC₅₀ data indicated that Ir(III) sensors bind tighter and inhibit CYP3A4 more potently than Ru(II) compounds, with tunable K_d values as low as 70 ± 2 nM for **4**.

Table 1. Dissociation constants (K_d), IC₅₀ values for CYP3A4 (μ M) and emission quantum yields for sensors **2–5**.

Compound	K_d^a (μ M)	IC ₅₀ ^b (μ M)	Φ_{em}^c (H ₂ O)	τ / μ s (H ₂ O)
2	53 $\pm$ 4	6.0 $\pm$ 0.5	0.046(3)	0.66
3	23 $\pm$ 2	3.1 $\pm$ 0.4	0.042(9)	0.75
4	0.070 $\pm$ 0.002	0.25 $\pm$ 0.02	0.086(9)	1.6
5	0.130 $\pm$ 0.011	0.20 $\pm$ 0.01	0.007(1)	0.062
6	11.2 $\pm$ 0.8	1.02 $\pm$ 0.02	ND	ND

^aDetermined by spectrophotometric titration assay. ^bCYP3A4 activity assay with BFC, 293 $\pm$ 3K, 0.2 μ M CYP3A4, 0.3 μ M cytochrome P450 reductase, vs. DMSO control (100% activity), standard error <10%. ^cEmission spectra of absorption matched solutions in argon sparged H₂O ($A_{435} \sim 0.07$), λ_{ex} = 435 nm, 455 nm longpass filter, referenced to Ru(bpy)₃ Φ_{em} = 0.042.

Next, Ir(III) complexes **4** and **5** were co-crystallized with CYP3A4 (Figure 3C–F, S8). In both structures, the inhibitor's R_1 side chain curls above the heme and the terminal pyridine nitrogen ligates to the heme iron (Fe–N distance of 2.20–2.23 Å). Hydrophobic residues Phe108, Phe220, Phe57, and Leu482 are in close contact with the ppy and pq groups of **4** and **5**, respectively. Electron density was well defined for the heme-ligating pyridine, part of the tether, and the Ir(III) cores. The Ir ligands were poorly defined, which suggests that both the Δ and Λ isomers of **4** and **5** were bound to the active site. Stereochemistry was not specified during structural refinement, but the Λ and Δ isomers (shown in Figure 3C–F) were preferably selected for **4** and **5**, respectively, and fit into electron density maps by the refining program. Importantly, **4** and **5** are the first iridium complexes characterized to bind to a CYP enzyme.^{31–38}

To ensure that **4** binds to CYP3A4 more selectively than to other CYP isoforms, IC₅₀s of **4** against CYP3A4, CYP1A2, and CYP2C9 were determined using commercially available inhibitor screening kits (BioVision). Data from these kits vs. the soluble reconstituted system in Table 1 can not be compared directly because they were acquired under different conditions.²⁷ Derived IC₅₀ values were 2.8 ± 1.0 μ M, >100 μ M and 79 ± 6 μ M for CYP3A4, CYP2C9 and CYP1A2, respectively (Figure 3G). The 28- and >36 -fold difference in IC₅₀ demonstrates the high selectivity and preferential binding of **4** to a larger and expandable active site of CYP3A4 (Figure S8). For comparison, the volume of the active site cavity in ligand-free CYP3A4 is 1400 Å³ as compared to 375 Å³ and 470 Å³ in CYP1A2 and CYP2C9, respectively.^{7,39,40}

With compound **4** identified as a lead, we evaluated its ability to act as an active site photosensor by measuring changes in emission upon addition of ligand-free or substrate/inhibitor-bound CYP3A4 (Figure 3H). Strong luminescence quenching was observed when **4** (5 μ M) was mixed with ligand-free CYP3A4 (3 μ M), consistent with other emissive probes for P450 enzymes.^{41–45} The quenching was partial when CYP3A4 was bound to a substrate or inhibitor prior to addition of **4**. Importantly, the emission levels were ligand dependent and correlated with the ligands' binding

affinity: the strongest CYP3A4 binder, ritonavir ($K_d = 19$ nM), was the most difficult to displace, whereas the weakly bound substrate, testosterone (K_d of 1.5 μ M and 30 μ M for two binding sites), was expelled by the probe more easily.

recovers in a manner proportional to the binding affinity of CYP3A4 substrates and inhibitors. Furthermore, photosensor **4** penetrates and inhibits CYP3A4 in hepatic cells, and emits brightly in the intracellular environment. This new class of photosensors is

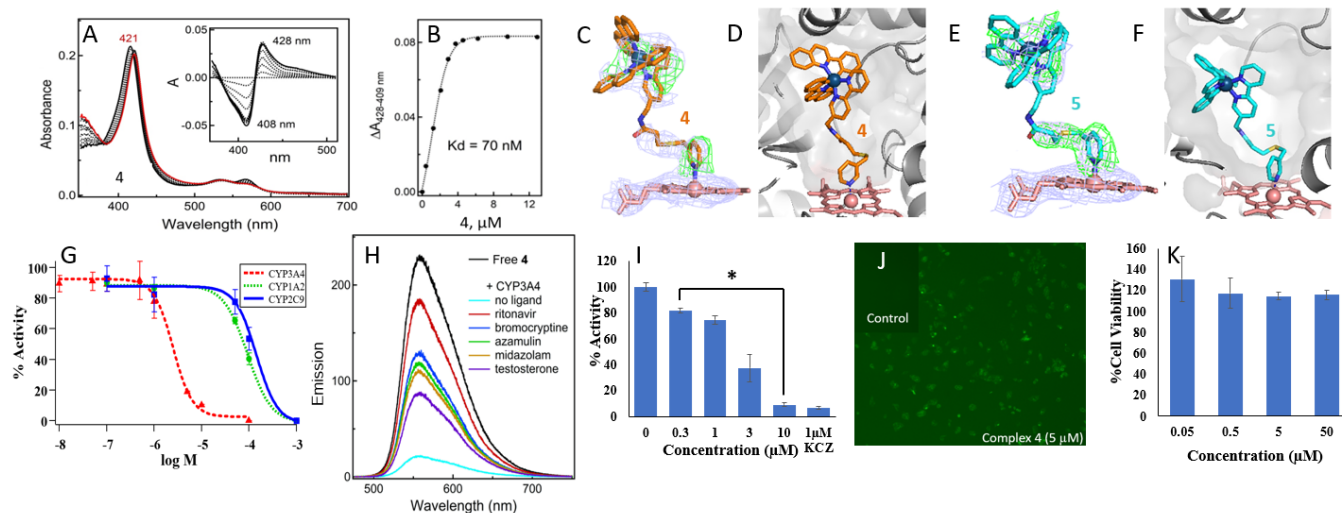


Figure 3. (A) Spectral changes observed during equilibrium titration of CYP3A4 with **4**, inset is difference spectra; (B) Titration plot with derived K_d value; (C, D) Crystal structure of the **4**-CYP3A4 complex at 2.78 Å resolution (PDB ID 7UAY). (E, F) Crystal structure of the **5**-CYP3A4 complex at 2.65 Å resolution (PDB ID 7UAZ). Blue and green mesh in panels H and J are $2F_o - F_c$ and polder omit electron density maps contoured at 1σ and 3σ levels, respectively; (G) Inhibition of CYP3A4, CYP1A2, CYP2C9 activity by **4**; (H) Fluorescence spectra of **4** (5 μ M) in the absence and presence of CYP3A4 (3 μ M) bound to different substrates and inhibitors (10–20 μ M) showing ligand-dependent emission yields (0.1 M PBS, pH 7.4, 10% glycerol, $\lambda_{ex} = 433$ nm); (I) CYP3A4 activity with **4** (0.3–10 μ M) determined by P450-Glo CYP3A4 Assay or ketoconazole (1 μ M) as positive control; Concentrations 0.3–10 μ M are statistically significant from control containing vehicle; * $P < 0.05$; (J) Fluorescence microscopy images (GFP filter) of HepG2-CYP3A4 cells treated with **4** (5 μ M), (J) Inset is control fluorescence from vehicle treated cells; (K) Cell viability at different concentrations of **4** (0.05–50 μ M) determined by a cellular viability assay (MTT, 72 h).

To further substantiate the scope of our lead compound **4**, we assessed its inhibitory properties in HepG2 human hepatoma cells, where expression of most drug-metabolizing CYPs is negligible or absent. However, when HepG2 cells are stably engineered with vectors expressing CYPs, protein levels reach those in primary human hepatocytes, which makes this model cell line a convenient *in vitro* tool to mimic drug metabolism in the liver.^{46–48} To determine the CYP3A4 inhibitory activity of **4**, HepG2 cells overexpressing CYP3A4 were used in conjunction with a bioluminescent P450-Glo CYP3A4 assay. Importantly, a strong concentration-dependent decrease in activity was observed, with statistically significant inhibition at 300 nM (Figure 3I, ~20% inhibition, $P < 0.05$ vs control). These data confirm that **4** is able to efficiently penetrate HepG2-CYP3A4 cells and inhibit CYP3A4 activity at nanomolar concentrations.

Finally, to demonstrate that our photosensors can be visualized in cells, we employed fluorescence microscopy. HepG2-CYP3A4 cells were treated with **4** (5 μ M) for 1 h (Figure 3J), then rinsed with PBS (pH 7.0) and imaged using the GFP channel. We found that **4** is cell-permeable and can be visually detected at concentrations as low as 5 μ M. Utilization of metal complexes at such low concentrations limits their cell toxicity. In fact, **4** is well tolerated by HepG2-CYP3A4 cells ($EC_{50} > 50$ μ M), as judged by a cellular viability assay (Figure 3K, MTT, 72 h). This result provides strong evidence that cell toxicity can be avoided or largely minimized when Ir(III) complexes are used as photosensors at low concentrations (<10 μ M).

In summary, Ir(III) compound **4** is a potent and specific inhibitor that serves as a photosensor for CYP3A4 active site occupancy. The luminescence of **4** is quenched upon binding to CYP3A4 and

expected to provide a significant advantage over traditional endpoint assays currently used for detection of drug-drug interactions of CYP3A4 in cells. Another beneficial property of our photosensors is their prolonged luminescence lifetimes, which allows time-resolved fluorescence measurements for excluding autofluorescence, a major problem in bio-imaging that cannot be addressed with current sensors. Studies are now underway in our laboratories to further develop this class of compounds and utilize Ir(III) photosensors for monitoring CYP3A4 active site occupancy *in cellulo*.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Experimental procedures for synthesis and spectral data for **2–5**, procedures for photophysical, pharmacological and biological studies, X-ray crystallographic data

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Notes

The authors declare no competing financial interests.

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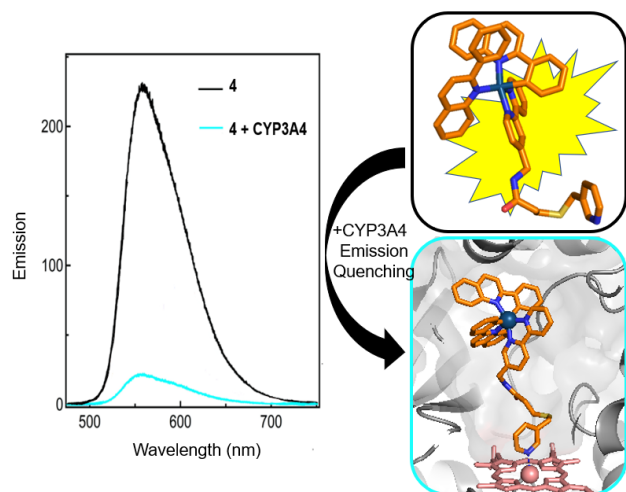
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TOC Graphic



TOC Synopsis: Potent and selective Ir(III)-based CYP3A4 inhibitors were designed and synthesized to act as chemical tools for probing CYP3A4 active site occupancy. Importantly, our Ir(III) probes are highly emissive and have long emission lifetimes. Emission of Ir(III) becomes quenched in the CYP3A4 active site and addition of substrate or inhibitors of CYP3A4 results in emission turn-on.