



1 Distributions of PGE and Re-Os isotope systematics in
2 chromite from the Coobina chromitite deposit in Western
3 Australia: Implications for chromite as a sulfide
4 mineralisation indicator
5

6

7 *Louise Schoneveld¹*, Stephen Barnes¹, Igor S. Puchtel², Svetlana Tessalina³,*
8 *Marek Locmelis⁵*

9

10

11 ¹Discovery, Mineral Resources, CSIRO, Kensington, Western Australia, Australia

12 ²Department of Geology, University of Maryland, College Park, MD 20742, USA

13 ³John de Laeter Centre, Curtin University, Kent St Bentley 6102 WA⁵Department of Geosciences and

14 Geological and Petroleum Engineering, Missouri University of Science & Technology, Rolla, Missouri, USA

15

16 *Corresponding author: louise.schoneveld@csiro.au

17

18

19

20 Revised for *Canadian Mineralogist*

21 Version 2022-07-05

22

23 **Abstract**

24 The abundance of Ru in chromite has been suggested as an indicator of sulfide liquid
25 saturation in komatiites. The komatiite magma-derived Archean Coobina intrusion is known
26 to be barren in terms of sulfide mineralization. Therefore, the Coobina intrusion allows for a
27 useful case study to test the applicability of Ru abundance in chromite as a potential indicator
28 for sulfide mineralization, as well as for better understanding the PGE-chromite association
29 in general.

30 The Coobina intrusion is a highly deformed layered intrusion, interpreted as a flared dyke. It
31 contains multiple massive chromitite seams that have been recently mined for metallurgical
32 grade chromite. In this study, 18 samples from chromitite seams throughout this intrusion are
33 investigated for their whole-rock platinum-group element (PGE) contents which is compared
34 to their chromite mineral chemistry; including PGE content, platinum-group mineral (PGM)
35 mineralogy, and Re-Os isotope systematics. Each sample has a similar chromite major and
36 minor element chemistry, but a unique trace element signature, even within the same seam. In
37 general, there are higher concentrations of Ru (>300 ppb) within chromite in the south-east,
38 (toward the feeder dyke) and lower concentrations (<50 ppb Ru) in the north-west. At a
39 sample scale, Ru in the whole rock and Ru in solid solution in the chromite are inversely
40 correlated, while Ir shows a positive correlation between the whole rock and chromite
41 mineral chemistry, indicating differing partitioning behaviours within the Iridium-group PGE
42 (IPGE = Os, Ir, Ru). The inverse correlation between Ru in solid solution within chromite
43 and Ru in whole-rock chromitite suggests that, for seams with high Ru in whole rock, Ru is
44 occurring within separate PGM phases. This is supported by the observation that the samples
45 with high whole rock Ru also have a high number of visible metal alloy and/or PGM
46 inclusions. Although these inclusions are not necessarily Ru-rich phases, their presence
47 suggests that there is a preference for these samples to form nuggets, which may restrict Ru
48 partitioning into the chromite crystal structure. We suggest that the low Ru values in the
49 Coobina chromite are a result of transient sulfide saturation.

50 If using chromite as a detrital indicator mineral for magmatic sulfide exploration, it must be
51 kept in mind that transient sulfide saturation within chromitite seams may give a false
52 positive signature.

53 **Introduction**

54 Chromite has been associated with economically viable concentrations of platinum
55 group elements (PGE), particularly the more chromite-compatible iridium-subgroup PGE
56 (IPGE = Ir, Ru, Os) and Rh, in locations such as the Merensky Reef of the Bushveld
57 Complex of South Africa (Barnes *et al.* 2004, Smith *et al.* 2021). The part that chromite plays
58 in this PGE accumulation has been the subject of numerous studies (Teigler & Eales 1993,
59 Godel *et al.* 2007). Since the advent of laser ablation ICP-MS and its application to analysis
60 of trace elements at sub-ppm level, it has been possible to investigate to what extent IPGE in
61 chromite-rich rocks are present in solid solution in the chromite lattice, as opposed to
62 occurring as discrete IPGE-rich minerals (Locmelis *et al.* 2011, Pagé *et al.* 2012, Park *et al.*
63 2017). The IPGE contents of chromite and chromite-rich rocks have been used as a constraint
64 in petrogenetic models (Finnigan *et al.* 2008, Naldrett *et al.* 2009a, Park *et al.* 2012, Prichard
65 *et al.* 2017) and provenance studies (Pagé & Barnes 2009, González-Jiménez *et al.* 2015,
66 Pagé & Barnes 2016). In the case of extrusive komatiites, it has been shown that
67 concentrations of IPGE in chromite can exceed hundreds of ppb (Locmelis *et al.* 2011).
68 Fiorentini *et al.* (2004) found that IPGE predominantly occur as micro-inclusions in chromite
69 within komatiitic basalts, but are present in solid solution in chromite within true komatiites.
70 Locmelis and co-authors (Locmelis *et al.* 2011, Locmelis *et al.* 2013, Locmelis *et al.* 2018)
71 proposed that chromite lattice Ru concentration of less than 150 ppb has the potential to
72 indicate equilibration with sulfide liquid and therefore, chromite may be a useful mineral
73 indicator in exploration for magmatic sulfide ores. However, Pagé and Barnes (2016) found
74 that IPGE in chromite tend to be present at much lower concentrations in intrusive rocks
75 relative to extrusive rocks formed from the equivalent magma types, which suggests that
76 intrusive-hosted chromite could represent a false positive signal for this exploration indicator
77 if analysed in soils/surface samples or ex-situ from their host rock.

78 The Coobina chromite deposit in the Archean Sylvania Inlier in Western Australia
79 (Figure 1) is a rare example of an intrusive layered body of komatiitic affinity, containing
80 thick chromitite seams, for which a large amount of whole-rock PGE data is already available
81 (Barnes & Jones 2013). It provides an opportunity to investigate the distribution of IPGE
82 hosted in solid solution in the chromite lattice versus the presence of discrete platinum group
83 minerals (PGM) in samples with no known association with magmatic sulfides. Therefore, a
84 detailed study of the IPGE contents in the Coobina intrusion will provide novel information
85 useful for provenance studies and allow to further evaluate the relationship between Ru in

86 chromite and the presence or absence of magmatic sulfides. In addition, Re-Os isotope data
87 for the Coobina chromitites provide a new data point on the Archean mantle isotope
88 evolution curve for Os, as well as help decipher the source of PGE in these chromitites.

89 The Coobina chromitites have been previously investigated by Barnes and Jones
90 (2013) focussing on the structure of the strongly folded and sheared seams. These authors
91 found that the chromite major element chemistry is indicative of a komatiitic origin and the
92 crystal size distribution is suggestive of in-situ growth rather than mechanical sorting. The
93 chromitite whole-rock PGE contents are low, with values of ~100 ppb for Ru and Ir. The
94 Mg/Fe ratios of chromite within the chromitite seams may have been re-equilibrated with the
95 olivine cumulates they are hosted within. Extensive shearing and folding due to the Capricorn
96 orogen makes it difficult to trace the seams between pits, either by structural or even
97 chemical measurements.

98 In this study we build on the work by Barnes and Jones (2013) and we further
99 investigate the chemistry of chromite using LA-ICP-MS. Eighteen chromitite samples were
100 analysed for trace elements and integrated with the whole-rock and chromite major element
101 data of Barnes and Jones (2013) to infer the controls on the trace element incorporation into
102 chromite and PGM nugget formation. Three chromitite samples, including one replicate, were
103 also investigated for their Re-Os isotope systematics and Re and PGE abundances. A
104 particular area of interest is the partitioning behaviour of Ru, as Ru content of chromite has
105 been proposed to be a potential indicator for sulfide saturation in komatiite magmas and,
106 hence, for magmatic Ni-sulfide prospectivity (Locmelis *et al.* 2011, Locmelis *et al.* 2013,
107 Locmelis *et al.* 2018).

108 Geological setting

109 The Coobina chromitite mine is located ~50 km southeast of Newman (Figure 1),
110 within the Archean Sylvania Inlier in the southern Pilbara Craton of Western Australia. The
111 Coobina mine was owned and mined as a ferrochrome resource by Consolidated Minerals
112 Ltd. from 2002 through 2013.

113 The chromite ore is hosted by a large ultramafic cumulate body, which intruded along
114 the SE contact between the Jimblebar Greenstone Belt and foliated granite of the Sylvania
115 Inlier (Williams & Tyler 1991). The current outcrop at Coobina represents a vertical section
116 (younging to the NW) through an Archean ultramafic intrusion with an interpreted 10 km

117 long feeder conduit toward the east (Figure 1) (Tyler 1991). An anorthositic metagabbro to
118 leucogabbro is found along the northern part of the ultramafic body, in places with a
119 gradational contact. The intrusion is strongly deformed through a series of major east-west
120 trending shears that extended through the region, likely related to the early Proterozoic
121 Capricorn Orogeny associated with the collision of the southern Pilbara and northern Yilgarn
122 Cratons at c. 1830-1780 Ma (Cawood & Tyler 2004, Occhipinti *et al.* 2004).

123 Apart from a broad Archean classification for the Coobina intrusion, no reliable age
124 information is available. Uranium-Pb zircon data available from the Geological Survey of
125 Western Australia GeoView database on the overlying Jimblebar Belt greenstones comprise
126 two dates of 3.193 and 3.185 Ga from felsic metavolcanic rocks (Wingate *et al.* 2019b, a) and
127 maximum deposition ages of 3.565-3.573 Ga for metasedimentary units within the belt. One
128 date of 3.185 Ga from a felsic intrusive rock within the outcrop area of the Coobina intrusion
129 is also available, which can be taken as a minimum age for the intrusion itself. The intrusion,
130 therefore, can be bracketed to have been emplaced between 3.193 and 3.185 Ga. Tyler (1991)
131 suggest that the Inlier forms a separate micro-terrane of meso-Archean age that amalgamated
132 with the main Pilbara Craton between 2.76 and 3.00 Ga.

133 The Coobina intrusion is made up of a serpentinite (formerly dunitic) body, several
134 hundreds of metres thick, which hosts a series of chromite lenses and an underlying
135 peridotite-dominated unit. The intrusion has undergone multiple phases of deformation to the
136 point where the chromitite seams are isoclinally folded, making it difficult to assess whether
137 they were uniformly distributed through the dunite section pre-deformation, but the outcrop
138 distribution suggests that there were probably two main intervals with thick chromitite lenses
139 towards the top of the dunite unit (Fig. 1).

140 Abundant sub-parallel chromite lenses are observed throughout the serpentinite and
141 range from 2 cm to 2 m in width and up to 350 m in length (Figure 2). Primary layering
142 (mm-cm scale) is only observed within and near the chromite lenses. The boundaries of the
143 chromite lenses vary from sharp to diffuse, with massive chromite-rich (~85% chromite)
144 lenses commonly grading into zones of disseminated chromite. The lenticular form of the
145 chromitite bodies is considered to be the result of boudinage developed during extensive tight
146 and locally isoclinal folding of originally laterally continuous seams (Barnes & Jones 2013).

147 The field relationships are consistent with the Coobina intrusion being a composite
148 ultramafic-mafic layered intrusion with Great Dyke-like flared dyke geometry and internal

149 magmatic differentiation. Chemically, the high MgO content of the cumulates and the
150 compositions of the chromites ($\text{Cr}/[\text{Cr}+\text{Al}] = 0.7 - 0.85$, $\text{MgO}/[\text{MgO}+\text{FeO}] = 0.30 - 0.55$ and
151 low Fe^{3+} contents) indicate a likely Al-undepleted komatiitic parental magma (Barnes and
152 Jones 2013).

153 Samples

154 There are a total of 150 samples that were originally analysed using electron
155 microprobe and 46 of these were analysed for whole rock PGE content in the Barnes and
156 Jones (2014) study. In the study covered here, 18 samples with varying whole-rock PGE
157 content were chosen for trace element and isotopic analysis. The sample names and locations
158 are summarised in Figure 1 and Table 1. The samples were taken from the chromite seams
159 and consist of primary euhedral chromite (Figure 3) with minor interstitial serpentine (likely
160 after olivine). The samples were mounted in epoxy and fashioned into 25 mm diameter
161 polished rounds for microanalysis.

162 Three samples (BC1, NC2 and WS2) from seams within different pits were analysed
163 in detail for their texture (Figure 4). These three samples were chosen based on their spatial
164 location and on their whole-rock Ru abundances giving a range of 70 ppb, 87 ppb and 174
165 ppb Ru, respectively. These 3 samples, along with an additional sample (NS3 – 115 ppb Ru)
166 were also examined for their visible inclusion populations and PGE mineralogy. Two
167 samples from the Coobina intrusion were also analysed for the Re-Os isotope systematics.

168 Analytical techniques

169 This study integrates electron microprobe data from Barnes and Jones (2013) with
170 new chromite trace element data obtained on 18 samples via LA ICP-MS. Our study also uses
171 the new data on the PGE content within the chromite to compare with the whole-rock PGE
172 contents published in the Barnes and Jones (2013) study. Additionally, new Re-Os isotopic
173 and Re and PGE abundance data were obtained for 2 samples (and one replicate) of
174 chromitite via negative thermal ionization mass-spectrometry (N-TIMS) and Isotope Dilution
175 Multi-Collector Inductively-Couple Plasma Mass Spectrometry (ID MC-ICP-MS), following
176 the methods described below. Detailed methods for the whole-rock analysis and major
177 element analysis of chromite are detailed in Barnes and Jones (2013). Therefore, only a short
178 summary is presented here.

179 Whole-rock elemental abundances, including the PGE (Ir, Ru, Rh, Pt, and Pd), were
180 measured on 2 kg aliquots from 46 samples (with 4 replicates) at GeoScience laboratories
181 (Ontario Geological Survey, Sudbury, Canada) using a specially customized method of
182 sodium metaphosphate fluxed nickel sulfide fire assay preconcentration followed by ICP-MS
183 (Bédard & Barnes 2004). Of these samples, 41 were also analysed for whole-rock major
184 element concentrations. The results from these analyses were included within the Barnes and
185 Jones (2013) study, and are revisited here to give context to the trace element abundances
186 measured within the chromite grains.

187 The major element analysis of 150 samples of chromitite were conducted using a
188 Cameca SX50 electron microprobe at CSIRO-ARRC (Australian Resources Research Centre;
189 Kensington, Western Australia) using standard wavelength-dispersive spectrometry, 30-nA
190 beam current, 15 kV accelerating voltage, and 100 s counting times. The analysis targeted
191 centres of unaltered grains, away from cracks, with 8-12 points analyzed per sample. All data
192 are available for download from the supplementary material in the Barnes and Jones (2013)
193 study. The ferric iron content of each analysis was estimated in this study by the Barnes
194 (1998) method; i.e. assuming stoichiometry, and an ideal XY_2O_4 where $X = (Fe^{2+}, Ni, Mn,$
195 $Co, Zn)$ and $Y = (Cr^{3+}, Fe^{3+}, Al)$. Titanium was assumed to be present as an ulvöspinel
196 component, and V is assigned to a fictive component. Iron is subdivided into ferrous and
197 ferric to satisfy the conditions $n_y = 2n_x$ where n_y is the total number of atoms of trivalent
198 cations and n_x is the total moles of divalent cations per unit cell.

199 **Os isotopic, Re and PGE bulk chromitite abundance data.**

200 The analytical protocols used in this study to obtain the Re-Os isotopic and Re and
201 PGE abundance data for the chromite samples followed those described in detail by (Puchtel
202 *et al.* 2014) and Puchtel *et al.* . The analyses were performed at the Isotope Geochemistry
203 Laboratory (IGL), University of Maryland, College Park, USA. In summary, 0.015-0.057 g
204 of crushed, unground whole-rock chromitite sample, 5 mL of double Os-purged, triple-
205 distilled concentrated HNO_3 , 4 mL of triple-distilled concentrated HCl , and appropriate
206 amounts of mixed ^{185}Re - ^{190}Os and PGE (^{99}Ru , ^{105}Pd , ^{191}Ir , ^{194}Pt) spikes were sealed in double,
207 internally-cleaned, chilled 25 mL Pyrex™ borosilicate Carius Tubes and heated to 270°C for
208 96 h. Osmium was extracted from the acid solution by CCl_4 solvent extraction (Cohen &
209 Waters 1996), then back-extracted into HBr , followed by purification via microdistillation
210 (Birck *et al.* 1997). Ruthenium, Pd, Re, Ir, and Pt were separated and purified using anion
211 exchange chromatography following the protocol of (Rehkämper & Halliday 1997) with

212 some modifications. Average total analytical blanks (TAB) during the analytical campaign
213 were (in pg): Re 0.33 ± 0.07 , Os 0.40 ± 0.14 , Ir 0.43 ± 0.26 , Ru 3.3 ± 3.2 , Pt 5.1 ± 2.8 , and Pd
214 3.3 ± 1.9 ($\pm 2SE$, $N = 3$). For the three chromite samples analysed, the total analytical blanks
215 for Os, Ir, and Ru constitute less than 0.07% of the total element analysed. Blank
216 contributions for Re varied between 9% and 19%, for Pt – between 1.5% and 4.9%, and for
217 Pd – between 4.1% and 5.3% of the total element analysed.

218 Osmium isotopic measurements were accomplished via negative thermal ionization
219 mass-spectrometry (N-TIMS: (Creaser *et al.* 1991). All samples were analysed using a
220 secondary electron multiplier (SEM) detector of a *ThermoFisher Triton*[®] mass spectrometer
221 at the IGL. The measured isotopic ratios were corrected for mass fractionation using
222 $^{192}\text{Os}/^{188}\text{Os} = 3.083$. The internal precision of measured $^{187}\text{Os}/^{188}\text{Os}$ ratios in all three samples
223 was 0.05-0.06% relative (2SE). The $^{187}\text{Os}/^{188}\text{Os}$ of 300 pg loads of the in-house Johnson-
224 Matthey Os standard measured over the period of 2 years leading up to the current analytical
225 session averaged 0.11376 ± 11 ($\pm 2SD$, $N = 54$). This value characterizes the external
226 precision of the Os isotopic analysis (0.10%) and represents the true uncertainty on the
227 measured $^{187}\text{Os}/^{188}\text{Os}$ ratio for each individual sample. The $^{187}\text{Os}/^{188}\text{Os}$ ratio measured in
228 each sample was also corrected for the instrumental bias relative to the average $^{187}\text{Os}/^{188}\text{Os} =$
229 0.11378 precisely measured in the Johnson-Matthey Os standard on the Faraday cups of the
230 IGL Triton (Puchtel *et al.* 2014).

231 The measurements of Ru, Pd, Re, Ir, and Pt were performed at the IGL via inductively
232 coupled plasma mass-spectrometry (ICP-MS) using a *Nu Plasma* instrument with a triple
233 electron multiplier configuration in a static mode. Isotopic mass fractionation was monitored
234 and corrected for by interspersal of samples with standards. The accuracy of the data was
235 assessed by comparing the results for the reference materials UB-N and GP-13 obtained
236 during the ongoing analytical campaign. Concentrations of all HSE and Os isotopic
237 compositions obtained at the IGL are in good agreement with the results from other
238 laboratories (Table 1 in Puchtel *et al.* (2014)). Diluted spiked aliquots of iron meteorites were
239 run during each analytical session as secondary standards. The results from these runs agreed
240 within 1% for Re and Ir, and within 2% for Ru, Pt, and Pd, with fractionation-corrected
241 values obtained from measurements of undiluted iron meteorite solutions using Faraday cups
242 of the same instrument with signals of >100 mV for the minor isotopes. We, therefore, cite
243 0.1% as uncertainty on the concentration of Os, $\pm 1\%$ on the concentrations of Ir, and $\pm 2\%$
244 relative on the concentrations of Ru in the three samples analysed. The uncertainties on the

245 Re concentrations were between 4.6 and 9.6%, Pt – between 2.0 and 2.4%, and Pd – between
246 2.1 and 2.6% relative, assuming a ~50% variation in the Re and PGE abundances in the
247 average TAB. The uncertainty on the Re concentration was the main source of uncertainty on
248 the Re/Os ratio. For the three samples analysed, this uncertainty was estimated to be between
249 4.7 and 9.7% relative.

250 The initial $\gamma^{187}\text{Os}$ values were calculated as per cent deviation of the isotopic
251 composition at 3.2 Ga (the accepted age of the Coobina intrusion) relative to the chondritic
252 reference of (Shirey & Walker 1998) at that time. The average chondritic Os isotopic
253 composition at 3.2 Ga was calculated using the ^{187}Re decay constant $\lambda = 1.666 \times 10^{-11} \text{ year}^{-1}$
254 (Smoliar *et al.* 1996) and an early Solar System initial $^{187}\text{Os}/^{188}\text{Os} = 0.09531$ at 4.58 Ga and
255 $^{187}\text{Re}/^{188}\text{Os} = 0.40186$ (Smoliar *et al.* 1996, Shirey & Walker 1998).

256 Trace elements in chromite

257 Eighteen chromitite samples were selected for trace element analysis to represent the
258 variable range of PGE in the whole-rock samples. 10-12 150 μm diameter ablations were
259 carried out on each sample for a total of 225 chromite trace element analyses. These analysis
260 were conducted using a Photonmachines, ATLex 300si-x Excite 193nm Excimer ArF laser
261 attached to an Agilent 7700 ICP-MS in the National Geosequestration Laboratory, CSIRO,
262 Kensington, Western Australia. The ablation was set 3 J/cm^2 fluence at a rate of 9 Hz. The
263 plasma conditions were optimized daily, to obtain highest counts with oxide production, with
264 $^{248}\text{ThO}/^{232}\text{Th}$ remaining below 0.4%.

265 Isotopes measured in this study were: ^{27}Al , ^{29}Si , ^{47}Ti , ^{51}V , ^{53}Cr , ^{55}Mn , ^{59}Co ,
266 $^{60,61,62,64}\text{Ni}$, $^{63,65}\text{Cu}$, ^{66}Zn , $^{99,100,101,102,104}\text{Ru}$, ^{103}Rh , $^{105,106,108}\text{Pd}$, ^{107}Ag , ^{111}Cd , $^{191,193}\text{Ir}$, ^{192}Os ,
267 $^{194,195}\text{Pt}$, ^{197}Au . Reference materials USGS GSD-2g, a basaltic glass doped with trace
268 elements, and UQAC-FeS-1 (developed by Universite du Quebec a Chicoutimi), a Ni-Cu-Fe-
269 S pressed power doped with trace elements, were used as bracketing standards during data
270 reduction. UQAC-FeS-1 was used as the primary reference material for the PGEs and
271 precious metals while GSD-2g was used as the bracketing standard for all other elements.
272 Fe 57 was used as the internal standard, using average values of Fe in chromite measured by
273 electron microprobe for individual chromitite samples (and given in Barnes and Jones, 2013).
274 The relative standard deviation of Fe wt.% in each sample ranges from 2% up to 15%.

275 The USGS-MASS-1 and UQAC-FeS-5 reference materials were used to assess the
276 precision and accuracy of the laser ablation ICP-MS analysis. The measured values of these

standards are generally within 10% relative to the published values and have a precision within 6% (1SD). Details of the secondary standard analysis are included in the supplementary material. Time resolved analysis was undertaken with 30 seconds of background with the laser off, then 260 laser pulses were collected which is approximately ~30 seconds of signal. The data was reduced using Iolite v.3.63 (Paton *et al.* 2011), choosing an integration window that avoids any large spikes in the time resolved analysis that would indicate a sub-surface micro-nugget effect. There was a delay of 20 sec. between different analyses to allow for sufficient washout of the signal. The average detection limits for the PGEs in chromite from this method are < 10 ppb (see supplementary material).

Two samples were also analysed by LA-ICP-MS mapping to investigate trace element and PGM distributions in larger areas. The method for this analysis type is included in the supplementary material. Some chromite analyses were duplicated at CCFS/GEMOC, Macquarie University, Sydney, Australia using a Photon Machines Excite Excimer laser ablation system connected to an Agilent 7700cx ICP-MS and give very similar results using Al²⁷ as the internal standard and PGE-A (Alard *et al.* 2000) and NIST610 (Norman *et al.* 1996) as bracketing standards. Details for these analyses are included within the supplementary material.

Results

Textures

The crystal size distribution (CSD) was quantified by measuring individual chromite grain areas on back-scattered electron images using Fiji (Schindelin *et al.* 2012) following the method outlined by Higgins (2000) and Higgins (2006). CSDs are expressed as number densities of grains per unit area per unit length of bin size and are used to understand the number of crystal growth populations or possibly grain resorption or post-cumulous recrystallisation. CSDs were measured on three samples, that vary significantly in their whole rock PGE content to determine if crystallisation history may play a role in the distribution of the PGE content.

Sample BC1 has a measured whole-rock PGE content of 70 ppb and was located in the Blatchfords Main pit. This sample shows a slight reduction in the amount of small grains which suggests some coarsening or sintering (post-cumulus recrystallisation) has occurred in this sample. Sample NC2 from the Newlands Creek pit has a whole rock Ru value of 87 ppb while WS2 from Wrights South pit has a Ru value of 174 ppb. Although these two samples

309 have very different Ru whole rock contents, they show almost identical crystal size
310 distribution patterns.

311 Major and trace element analysis of chromite

312 The major element chemistry of chromite within the Coobina chromitite was
313 discussed in detail by Barnes and Jones (2013). The chromite major element chemistry is
314 broadly similar throughout the entire deposit (Figure 5) and similar to other chromitites
315 worldwide (Barnes and Roeder, 2001). Barnes and Jones (2013) suggested that the chromite
316 within the Coobina deposit has undergone metamorphic exchange of Mg and Fe with the
317 spatially associated olivine based on linear relationships between Mg# (molar Mg/(Mg+Fe²⁺)
318 and Cr# (molar Cr/(Cr+Al)) within individual seams. This chemical exchange gives rise to a
319 relatively greater Fe enrichment in samples with lower modal abundance of chromite, but no
320 evidence was found for significant modification of the proportions of the major trivalent
321 components Cr, Al and Fe³⁺ (Fig. 3A) in these chromites. The proportions of the trivalent
322 cations show some overlap with the field for layered intrusions (Barnes & Roeder 2001), but
323 with its unusually Cr-rich character, the Coobina chromite data array fits more closely to
324 chromite compositions from Al-undepleted komatiites (Barnes 1998). If this chromite were
325 sampled as a resistant indicator mineral, it would be impossible to determine the provenance
326 by major element chemistry alone.

327 The chromite species throughout the Coobina intrusion show minor variations of Cr#
328 between 0.6 and 0.9 and Mg# between 0.2 and 0.6, but on the scale of individual samples,
329 they have previously been shown to display an inverse correlation (*cf.*, Barnes and Jones,
330 2013). The new LA ICP-MS data reported here show that Ti, V, Zn, and Co are positively
331 correlated with the Cr# of the chromite, while Ni shows a weak negative correlation (Figure
332 6A-E).

333 New data are presented here for the IPGE and the PPGE chemistry of chromite. The
334 average PGE contents in chromite for each sample show a weak negative correlation of Ir
335 with Cr# (Figure 6G-L), but no other detectable correlation of PGEs with the Cr or Fe³⁺
336 content (supplementary material Figure 15). Os and Ir concentrations in chromite are
337 positively correlated, ranging from below the detection limit of <10 ppb up to ~100 ppb
338 (Figure 7). Concentrations of Ru and Rh are also strongly correlated (Pagé *et al.* 2012, Park
339 *et al.* 2017), although the concentrations for Ru that are almost one order of magnitude higher
340 than for Rh. Palladium and Os show no correlation with Ru, while Pt contents are mostly

341 below the detection limit in chromite (i.e., <6 ppb). The distribution of Ru in these chromites
342 is not correlated with any of the key major element ratios (Figure 6, supplementary Figure
343 15). The Ru contents in chromite can be separated into 2 groups, those with an average Ru
344 in chromite of >160 ppb (6 samples) and those with <90 ppb (12 samples). In general,
345 chromite with the highest Ru concentrations is located closest to the contact of the chromitite-
346 free peridotite and closer to the suspected feeder dyke of the intrusion (Figure 8) while the
347 lower Ru in chromite values are located closer to the upper contact with the overlying
348 leucogabbro. However, there is significant deformation, folding, and shearing of the
349 chromitite seams, which makes tracing individual seams on a scale greater than a few
350 hundred metres difficult (Figure 8).

351 Whole rock PGE analysis

352 The whole rock chromitite PGE abundance data presented by Barnes and Jones
353 (2013) are compared to and integrated with the new in-situ PGE analysis of chromite to better
354 understand the distribution of these elements on the mineral, sample, and intrusion scales. As
355 illustrated in Figure 9E-F, primitive mantle (PM) normalized PGE patterns are overall similar
356 between the chromite and whole-rock chromitite samples. However, while Ru displays a
357 relatively narrow range in whole rock analysis of 47-174 ppb Ru between 46 samples, Ru
358 contents in chromite vary between 10 and 314 ppb across all chromite grains analysed.
359 Platinum contents in chromite are at or below the detection limit of ~6 ppb Pt, whereas Pt in
360 the whole rock chromitite samples is highly variable, with contents between 3 and 64 ppb.

361 Investigating three replicates of the whole-rock analyses from Barnes and Jones
362 (2013), Ru, Rh, and Pd standard deviations between replicates are between 0.1 and 8%
363 relative of the measured value. Pt variability is much larger with the three replicates having
364 variances between 25 and 62% relative of their measured value (13-51 ppb Pt). One sample
365 (with low whole-rock Pt abundance of 2.6 ppb) has a variance of only 3%. This shows that
366 the Pt abundances may not be reproducible with the whole-rock analysis methodology used
367 due to the tendency of Pt to form nuggets.

368 Presence of nuggets and inclusions of other phases

369 We analysed the visible nugget/inclusion populations of 4 samples (BC1, NC2, NS3
370 and WS2) that represent the two distinct groups; samples of high Ru in chromite and low
371 bulk rock values and vice versa (Figure 10). The whole-rock chromitite PGE contents are not
372 directly reflected by the chromite-lattice bound PGE contents, which suggest the presence of

373 PGE-bearing phases as small inclusions which we investigated using high resolution back-
374 scattered electron (BSE) images that show representative inclusions. These images are
375 included as supplementary material. Due to the small size (<4 μm) of the inclusions and the
376 fast scan EDS technique used, only the major element components can be determined from
377 these inclusions and not the full mineral chemistry. Also, it should be noted that the
378 inclusions observed represent the visible size fraction of the inclusions, and it is possible that
379 much smaller (nm) inclusions exists within these samples.

380 Most of the visible inclusions are near the edge of the grains or within crystal voids
381 (Figure 12). There is no clear correlation between size and type of inclusions and the degree
382 of sintering (post-cumulus recrystallisation) indicated by crystal size distributions (Figure 4).
383 Sample BC1 represents low Ru abundance in the whole-rock (70 ppb) and high values of Ru
384 within chromite (218 ppb). This sample has 11 visible phases that are considered “bright
385 phases” in SEM images, i.e., phases that contain elements with masses much higher than
386 chromium which generally indicates the presence of native metals, sulfides, alloys, or PGMs.
387 The small-sized (< 4 μm) inclusions (28% of the entire inclusion population) are composed
388 mostly of Os-Ir-Pt minerals (Figure 12), but this sample also contains other phases, including
389 sulfide, barite and Bi, Pb, and Sn-minerals. Similarly, sample NC2, which has comparable Ru
390 abundances (87 ppb in the whole-rock, and 282 ppb within chromite) contains only 14 visible
391 inclusions, most of which are less than 5 μm in diameter and are dominated by PGMs (Figure
392 10).

393 Sample NS3 represents high Ru in the whole-rock (115 ppb) and low Ru within
394 chromite (84 ppb). This sample has 30 visible bright inclusions that are all less than 4 μm
395 (Figure 10). The inclusions are composed of Os-Ir-Ru phases (Figure 10), but this sample
396 also contains phases such as native gold, barite and Bi- and W minerals. Sample WS2 has 44
397 bright inclusions, all of which were identified as galena (PbS) by rapid EDS analysis.
398 However, a PGM inclusion was found during LA-ICP-MS mapping (Figure 12), along with
399 many small regions with very high Pb concentrations. This rock has the lowest Ru abundance
400 within the chromite (21 ppb), while the whole rock Ru abundance is high (174 ppb). Of these
401 4 samples examined for bright phases and inclusions, the samples with the highest whole
402 rock Ru values have the largest number of inclusions.

403 **Re-Os isotopic data and Re and PGE abundances in chromitites**

404 Three chromitite samples, including one replicate (Chr1_1, Chr1_2, and Chr2), were
405 analysed for their Re-Os isotopic compositions and Re and PGE abundances (Table 2). The
406 chromitites have low Re abundances, ranging from 0.062 to 0.11 ppb, and relatively high Os
407 concentrations, between 38 and 408 ppb, resulting in very low Re/Os ratios and, thus,
408 minimal age corrections for the radioactive decay of ^{187}Re . The initial $\gamma^{187}\text{Os}$ values of the
409 three samples calculated for the age of the Jimblebar greenstone belt and Coobina ultramafic
410 rock emplacement of ~ 3.189 Ga (the average age obtained using U-Pb zircon dating of the
411 surrounding felsic rocks) vary in a narrow range between -0.53 ± 0.10 and -0.74 ± 0.10 ,
412 indicating evolution of the mantle source of the Coobina intrusion mafic-ultramafic rocks
413 with a long-term near-chondritic $^{187}\text{Re}/^{188}\text{Os}$ ratio. Because of the very narrow range of
414 variations in the $^{187}\text{Re}/^{188}\text{Os}$ ratios between the three chromitites analysed, no reliable
415 isochron can be obtained using these data.

416 Although Pt and Pd abundances are relatively constant and low in the three samples,
417 Os, Ir, and Ru are high and quite variable (Table 2), even between the two separately
418 processed aliquots of the same whole-rock sample (Chr1), indicating non-uniform
419 distribution of Os-Ir alloy and possibly laurite (RuS_2) inclusions within the chromitite or
420 highly localised PGE variations within a sample.

421 **Chromite contribution to the whole rock chromitite PGE content**

422 We calculated the proportion of chromite in the Coobina chromitites after Pagé and
423 Barnes (2016) using Eq. (1):

424
$$\% \text{ Cr-Spinel in sample} = F^{\text{Cr-Spinel}} = \frac{\text{Cr}_2\text{O}_3^{\text{WR}}}{\text{Cr}_2\text{O}_3^{\text{Cr-Spinel}}} \quad (1)$$

425 Where $\text{Cr}_2\text{O}_3^{\text{WR}}$ and $\text{Cr}_2\text{O}_3^{\text{Cr-Spinel}}$ are the Cr content in the whole rock (WR) and chromite in
426 wt. % oxide, respectively. The values of the chromite contribution to the chromitites ranges
427 from 61 to 78% in the Coobina samples, similar to those calculated for the chromitites
428 analysed by Pagé and Barnes (2016) of 40% - 87%. Equation (1) assumes that all chromium
429 in the whole-rock is contained within the chromite, which will result in a slight
430 overestimation, as olivine, the dominant mineral in komatiites, can contain up to ~ 2000 ppm
431 Cr (Pagé *et al.* 2012, Locmelis *et al.* 2019). Using the value calculated using equation (1), we
432 can calculate the proportion of the PGE in the whole rock that are incorporated in the
433 chromite:

434
$$F_i^{Cr-Spinel} = \frac{(C_i^{Cr-Spinel} * F_i^{Cr-Spinel})}{C_i^{WR}} \quad (2)$$

435 Where i is the PGE of interest and C is the concentration in ppb. Equation 2 assumes that all
 436 PGE and Cr are contained in the chromite.

437 Using the PGE contents shown in Table 3, it is evident that the whole-rock budget and
 438 chromite chemistry show some discrepancies, as evidenced by the >200% of the PGE being
 439 attributed to chromite, which results in a negative value for C_{Ru}^{melt} . Consequently, the Ru
 440 abundances in individual grains cannot represent the entire inventory in the whole rock
 441 analysis.

442 Discussion

443 Understanding what controls the PGE budget of chromite has implications for the use
 444 of chromite as an indicator mineral in exploration for nickel sulfide ores in komatiites (e.g.,
 445 Locmelis *et al.* (2018)) and for the use of chromite chemistry in petrogenetic studies
 446 ((Finnigan *et al.* 2008, Naldrett *et al.* 2009a, Park *et al.* 2012, Prichard *et al.* 2017)).

447 Re-Os isotope systematics of the Coobina komatiite source

448 Apart from the “broadly Archean” age, based mainly on field relationships and U-Th
 449 dating of surrounding units, there is no direct age determinations for the Coobina layered
 450 intrusion. However, the U-Pb zircon ages of 3.193 and 3.185 Ga (the average age is 3.189
 451 Ga) for the felsic volcanic rocks associated with the Coobina intrusion (Wingate *et al.* 2019b,
 452 a) provide an indirect constraint on the age of the intrusion.

453 Although no reliable Re-Os isochron can be obtained for the Coobina ultramafic
 454 intrusion, Re-Os model ages can be calculated for each of the three chromitite samples. These
 455 calculated model ages, termed T_{MA} and T_{RD} (Shirey and Walker, 1998), are reported in Table
 456 2. Time of separation from mantle that is growing according to chondritic evolution is
 457 represented by T_{MA} . It is calculated by assuming that the Re/Os of the sample is
 458 representative of its long-term history in the mantle. The time of Re depletion (T_{RD}) age is the
 459 minimum age for Re depletion. It is calculated assuming that a melt-depletion event
 460 previously removed all Re from the sample, and hence, growth of ^{187}Os was completely
 461 terminated at that time. Because the chromitites have very low, magmatic Re abundances,
 462 apparently not disturbed since the time of the emplacement of the intrusion, the calculated
 463 T_{MA} model ages (average $T_{MA} = 3.28 \pm 0.03$ Ga) are very similar to the T_{RD} model ages

464 (average $T_{RD} = 3.25 \pm 0.08$ Ga). These Re-Os model ages are in good agreement with the
465 bracketed age of the Coobina intrusion of between 3.193 and 3.185 Ga .

466 As is evident from Table 2, the analyzed samples are characterized by relatively large
467 variations in Os abundances, indicating presence of Os-rich nuggets in the chromitite.
468 However, such sample heterogeneity does not have an adverse effect on the calculated initial
469 Os isotopic composition as long as the mineral phases analyzed represent primary, magmatic
470 phases. This condition is obviously met for the Coobina chromitite, where the chromite
471 represents a magmatic phase that crystallized from the parental komatiite magma.

472 Using the accepted emplacement age of 3.189 Ga and the average initial Os isotopic
473 composition of the three chromitites ($\gamma^{187}\text{Os}(3.189 \text{ Ga}) = -0.63 \pm 0.21$), the long-term Re/Os
474 ratio with which the source of the Coobina parental komatiite magma evolved can be estimated.
475 In order to model the time-integrated evolution of Re/Os in the Coobina mantle source, the
476 average initial $^{187}\text{Os}/^{188}\text{Os}$ ratio obtained for the Coobina chromitites (0.10442 ± 22) has been
477 used to calculate the minimum $^{187}\text{Re}/^{188}\text{Os}$ ratio required to evolve to this Os isotopic
478 composition by 3.189 Ga, and assuming formation of this mantle domain shortly after the start
479 of the Solar System. Evolution of the Coobina komatiite source from an early Solar System
480 $^{187}\text{Os}/^{188}\text{Os} = 0.09531$ at 4558 Ma (Smoliar *et al.* 1996, Shirey & Walker 1998) to the initial
481 $^{187}\text{Os}/^{188}\text{Os}$ ratio of 0.10442 ± 22 at 3.189 Ga requires $^{187}\text{Re}/^{188}\text{Os}$ ratio of 0.375 ± 9 . This time-
482 integrated $^{187}\text{Re}/^{188}\text{Os}$ ratio for the Coobina komatiite system is within the range of chondritic
483 meteorites (a bulk chondrite average $^{187}\text{Re}/^{188}\text{Os} = 0.410 \pm 51$ ($\pm 2\text{SD}$), as compiled from the
484 data of Walker *et al.* , Brandon *et al.* , and Fischer-Gödde *et al.* , and its calculated $\gamma^{187}\text{Os}(3.189$
485 $\text{Ga})$ value of -0.63 ± 0.21 is also within the range of most Archean and Proterozoic komatiite
486 sources (Figure 10), albeit on the lower end of the range. It is also identical to the $\gamma^{187}\text{Os}(3.18$
487 $\text{Ga})$ value of -0.38 ± 0.43 obtained by Puchtel *et al.* for the contemporaneous Ruth Well
488 komatiites from the Pilbara Craton, whose source was calculated to have evolved with time-
489 integrated $^{187}\text{Re}/^{188}\text{Os} = 0.385 \pm 18$ and was interpreted by Puchtel et al. (2022) to be derived
490 from the convecting upper mantle. In summary, the Re-Os isotopic composition of the Coobina
491 chromitite is consistent with derivation of the Coobina parental komatiite magma from the
492 convecting upper mantle source, providing evidence for the mantle origin of the Coobina PGE
493 inventory.

494

495 **Trace elements in chromite**

496 The commonly measured trace elements in chromite (Zn, V, Mn, Ti, Co) are
497 positively correlated with Cr#. The Cr# is least likely to be altered by postmagmatic
498 processes (Barnes, 2000) and, thus, interpreted to reflect primary magmatic processes. Nickel
499 is not correlated with Cr#, averaging 510 ± 100 ppm throughout all measured samples. As the
500 dominate silicate phase within the chromitite seams is olivine, this consistent Ni value may
501 be due to equilibration with olivine. Comparing the Coobina Ni values to chromite from other
502 komatiitic parent bodies (Figure 13), these Ni contents are low, similar to those measured for
503 komatiites that experienced sulfide saturation and removal of immiscible sulfide liquid
504 (Locmelis *et al.* 2018).

505 Righter *et al.* (2004) and Brenan *et al.* (2012) showed that Ru is compatible within
506 chromite, and that the $D_{(\text{spinel}/\text{melt})}$ value is a strong function of the redox state of the system.
507 This arises from two factors: (1) Ru^{2+} is less soluble in silicate melt than Ru^{3+} (Borisov &
508 Nachtweyh 1998, Brenan *et al.* 2012) and (2) Ru^{3+} has an increasing affinity for more Fe^{3+}
509 rich spinels owing to the shared oxidation state and its octahedral site preference. Therefore,
510 precipitation of discrete Ru phases observed in the Coobina chromitite may be due to
511 reduction of Ru^{3+} to Ru^{2+} over the solubility line (Finnigan *et al.* 2008). Ru^{2+} is the dominant
512 Ru species in silicate melts under highly reduced conditions (in equilibrium with graphite).
513 The Coobina chromitite samples show that Rh concentration is highly correlated with Ru
514 suggesting the possibility that both are existing as dominantly divalent cations.

515 Chromite at Coobina is generally low in ferric iron, with the Fe^{3+} /trivalent cation ratio
516 averaging 0.08 ± 0.04 , typical of komatiites, which are generally regarded as having
517 crystallised near the FMQ buffer (Canil & O'Neill 1996), although some evidence exists for a
518 more reduced early Archean mantle recorded by komatiites (Locmelis *et al.* 2019, Nicklas *et*
519 *al.* 2019).

520 **Chromite in chromitites: partitioning and parental melts**

521 Ruthenium is compatible in chromite; however, the published partition coefficients
522 vary widely. Experimental measurements by Brenan *et al.* (2012) at 1400°C and under redox
523 conditions of $\Delta\text{IW}+7.2$ (calc. $\Delta\text{FQM}+3.74$) gives D_{Ru} of 38, with a predicted value of ~ 30 in
524 typical ferric iron-poor chromite. Studies by Righter *et al.* (2004) produced extremely high
525 partition coefficients, with $D_{\text{Ru}} > 1000$ at 1300°C and $\log \text{fO}_2 -5.07$ (calc. $\Delta\text{FQM}+2.15$), while
526 Pagé *et al.* (2012) calculated partitioning of the PGE into chromite from natural komatiites at

527 $D^{(\text{chromite}/\text{melt})}$ values around 79 for Ru and 9.5 for Ir. Park *et al.* (2012) found that D_{Ru} is
528 dependent on oxygen fugacity, with oxidising environments increasing the partition
529 coefficient of Ru and the other PGE significantly. The latter study showed, in agreement with
530 Pagé *et al.* (2012), that IPGE contents of chromite in intrusive settings are much less than
531 those in volcanic rocks of probably similar magmatic affinity. This implies that cooling rate
532 may play an important role in the final redistribution of Ru, the most likely mechanism being
533 solid state diffusive equilibration with Iridium-group platinum group mineral (IPGM) phases.

534 To evaluate the contribution of chromite to the whole-rock PGE budget, the rocks
535 were normalised to 100% chromite (see Equation 2 and Table 3), removing the small
536 contribution of the olivine (now serpentine) component. These results suggest that the whole
537 rock samples were made of ~60-80% chromite. This calculation, however, assumes that the
538 PGE abundances within the chromite throughout the large (>2kg) sample were
539 homogeneously distributed and/or were accurately represented by a small subsample
540 analysed by LA-ICP-MS. We can see in Table 3 that this is not the case, as when the Ru
541 content in the chromite is high, the chromite makes up >200% of the whole rock PGE
542 budget. This suggests that the high-Ru chromite may be extremely localised, either due to
543 local changes in the chromite partition coefficients and/or a gradual reduction of Ru^{3+} to Ru^{2+}
544 in the melt over time which decreased its compatibility in chromite. When comparing these
545 results with other layered intrusions (Figure 14), such as the Great Dyke, Bushveld, and
546 Black Thor (Pagé & Barnes 2016), Ru abundances in chromite and in whole rock samples are
547 characterized by a weak negative correlation. Notably, the Great Dyke is also calculated to
548 have more than 100% contributions of the PGE to the whole rock concentration (Pagé and
549 Barnes, 2016) from chromite suggesting that either the proportion of chromite is
550 underestimated or the partitioning of Ru into chromite is overestimated in these samples. The
551 concentration of the PGE in the equilibrium/parent melt can also be calculated using partition
552 coefficients.

553 Fiorentini *et al.* (2008) compared whole rock Ru abundances in 2.7 Ga mineralised
554 and unmineralized komatiites. The whole-rock analysis of barren samples shows a strong
555 correlation between Cr (ppm) and Ru (ppb), where the Ru contents range from 2 ppb to 10
556 ppb (Fiorentini *et al.* 2008). The mineralised komatiites have no strong correlation between
557 Ru and Cr, with Ru values between 1 and 20 ppb. Older (3.5–3.2 Ga) barren komatiites
558 generally have lower values of all the PGEs including Ru, which have measured values of 2–5
559 ppb Ru (Maier *et al.* 2009). Using the partitioning coefficient of $D_{\text{Ru}} = 79$ (Page *et al.*, 2012),

560 the calculated silicate component in equilibrium with the Coobina chromite would have a Ru
561 value of between 5-24 ppb. It has been found that intrusive samples tend to have lower
562 partitioning of the PGEs than their extrusive counterparts, suggesting these partition
563 coefficients may be on the higher end of the range. These values are in reasonable agreement
564 with those for barren komatiites from ~3.2 Ga.

565 The general trend of these chromite seams is an inverse correlation between Ru in
566 chromite and the Ru in the whole rock (Figure 9B). As these rocks are chromitites, lacking
567 any evidence for a magmatic sulfide component, the implication is that the Ru in the more
568 Ru-rich samples occurs within separate PGM phases, and that the abundance of these phases
569 is in inverse proportion to the concentration of Ru in the chromite lattice. This is supported
570 by the evidence that samples with high whole rock Ru also have high abundances of PGMs
571 and alloys. The presence of discrete IPGE inclusions suggests that there is a preference for
572 Ru in these samples to form micro-nuggets, rather than partitioning into the chromite lattice
573 such as observed in the NS3 sample Figure 10C and supplementary images.

574 **Metals, nuggets and inclusions.**

575 A common issue in interpreting PGE signatures is the tendency of PGE to form
576 discrete phases (nuggets), i.e. PGMs (Cabri *et al.* 1996, Park *et al.* 2012, Barnes *et al.* 2021).
577 The nugget effect can cause issues in bulk analysis from measuring nuggets that are unevenly
578 distributed at the scale of the analytical aliquot, which can cause large standard deviations in
579 replicate analysis (Barnes *et al.* 2021), as in the Pt analysis presented here.

580 However, it is noted that replicates measured in this study did not show a large
581 variation in Ru abundances, which may reflect that the inclusions, measured in this study,
582 average approximately 4 μm in the longest direction. Micro-nuggets, such as these, can be
583 accounted for by the whole rock analysis, but tend to be missed during LA ICP-MS
584 microanalysis. These micro-inclusions are of various phases, such as galena, PGMs, native
585 gold, and alloys, and are most frequently found within the outer rims of chromite grains
586 (Figure 12) or within voids. In their experiments, Brenan *et al.* (2012) and Finnigan *et al.*
587 (2008) observed the tendency for small alloy and sulfide grains to nucleate on the chromite-
588 melt interface. This is also observed in natural systems, such as the Bushveld Complex,
589 where the chromite and PGM are closely spatially associated (Naldrett *et al.* 2009a).

590 **Ruthenium as an indicator of sulfide saturation in komatiitic chromite**

591 Within a small range of major element chromite compositions, we observe two
592 populations, both >160 ppb Ru and much lower than 100 ppb Ru within the chromite. These
593 populations are similar to the <150 ppb Ru value, which is thought to indicate sulfide
594 saturation in komatiitic systems (Barnes & Fiorentini 2008, Locmelis *et al.* 2018). The major
595 element chemistry of the Coobina chromites are almost indistinguishable to those of known
596 komatiitic basalt parentage, and the trace element chemistry shows both depleted Ni and
597 depleted Ru signatures which could be interpreted to reflect derivation from a sulfide-
598 saturated komatiite magma as proposed by Locmelis *et al.* (2018). However, sulfides are
599 absent in the exposed portion of the intrusion and have not been found despite extensive
600 mining and exploration efforts. Therefore, the low Ru in chromite will yield a false positive
601 result if used as a detrital indicator mineral (i.e., without outcrop context) in the exploration
602 for magmatic Ni-Cu sulfide deposits. The false positive may be related to the intrusive nature
603 of the Coobina body and/or the age of the intrusion. Pagé and Barnes (2016) suggest that due
604 to the long cooling times in chromitites in intrusions, the chromites can recrystallise resulting
605 in reequilibration of the PGE which can diffuse into sulfide phases. However, based on the
606 very similar crystal size distribution (CSD) measurements between all samples (Figure 4), the
607 low-Ru chromite is not correlated to extensive recrystallisation/sintering, which would
608 produce CSDs with a reduction in the small sizes and a flattening of the CSD slope (such as
609 shown in Hunt *et al.* (2021)), suggesting that these changes are not due to local
610 recrystallisation and subsequent redistribution of these elements. Furthermore, Ru is
611 extremely difficult to remobilize (Locmelis *et al.* 2010). Also contributing to the low Ru in
612 chromite signal is the age of these intrusions as older komatiitic bodies have measured Ru
613 values that are lower than the 2.7 Ga komatiites used for the indicator study. If chromite are
614 to be used as a prospectivity indicator for sulfide, without the context of age, low Ru value
615 may be interpreted as a false positive signature for sulfide saturation.

616 The key observation in this data set is the decoupling between the whole-rock Ru
617 abundance and Ru content in chromite. The observed inverse correlation implies that there is
618 an independent control on the whole rock content of Ru in the more Ru-enriched samples,
619 and that this control is accompanied by depletion of Ru hosted in solid solution of the
620 chromite lattice. One possibility is that the Ru-enriched (in whole rock) samples owe their
621 chemistry to accumulation of Ru-enriched PGMs, such as alloys or laurite, from IPGE-
622 saturated melt (Barnes & Fiorentini 2008). However, if this was the case, then the parental

623 magma to both the PGM phases and the chromite must have been saturated in Ru-rich PGM,
624 and hence the chromite should have an Ru content proportionate to saturation levels of Ru. It
625 should, therefore, be enriched in Ru, not depleted relative to chromite that formed from Ru-
626 undersaturated magma.

627 A hint to a possible control may lie in the IPGE-enriched chromitites of the Harold's
628 Grave locality in the Shetland Ophiolite (Prichard *et al.* 2017). In this unusual chromitite
629 occurrence, whole-rock enrichments up to several ppm of IPGE, with only minor Pt and Pd
630 and exceptionally high Ir/Pd ratios of around 50, are associated with large (100-2000 μm)
631 interstitial aggregates of IPGM and minor sulfides between chromite grains. This mode of
632 occurrence is in marked contrast to the typical occurrence of IPGMs in chromitites as
633 inclusions within chromite grains. Prichard *et al.* (2017) postulated that these aggregates were
634 the remnants of partially remelted sulfide droplets generated by transient sulfide saturation
635 during chromite accumulation. As the sulfide droplets remelted, their IPGE contents
636 increased to levels of hundreds of ppm at which point they became saturated in IPGM, which
637 were left behind as the remainder of the sulfide subsequently redissolved. The process is
638 analogous to that proposed by Fonseca *et al.* (2012) to account for growth of IPGM phases
639 during mantle partial melting, and the Shetland chromitites are indeed mantle rocks. The
640 result is a rock with high IPGE contents and little or no sulfide. This process has been
641 invoked by Barnes *et al.* (2021) as a general explanation for the PGE-chromitite association
642 in layered intrusions, accounting for the entire spectrum from low-S IPGE-enriched
643 chromitite to sulfide-bearing chromitite hosted PGE reefs, such as the UG2 chromitite layer
644 of the Bushveld Complex. Furthermore, within the Merensky Reef of the Bushveld Complex,
645 a depletion of PGE contents is observed further from the feeder dyke (Naldrett *et al.* 2009b)
646 attributed to the deposition of sulfide closest to the feeder structures. This is the opposite to
647 what is observed at Coobina, where the whole rock values of Ru apparently decrease (and
648 while the Ru in chromite increase) toward the feeder dyke (Figure 8), suggesting the
649 generation of immiscible sulfide liquid is taking place far from the feeder, while the chromite
650 close to the feeder have higher Ru in chromite values, not affected by the far field transient
651 sulfide saturation, though due to the heavily folded nature of the Coobina seams, it is difficult
652 to determine if this is a true spatial association.

653 Transient sulfide saturation could explain elevated whole-rock Ru (along with Ir and
654 Os) in the more enriched Coobina samples. This would also provide an explanation for the
655 Ru depletion of the chromite in the same rock. Competition for IPGE between sulfide and

656 chromite results in depletion of the chromite in Ru, as observed in komatiitic chromite
657 associated with magmatic sulfide mineralisation (Locmelis *et al.* 2013, Locmelis *et al.* 2018).
658 A recent study by Hunt *et al.* (2021) suggests it is possible (and common) for chromite seams
659 to remain permeable to fresh melt derived from the overlying magma. This would allow fresh
660 melt to percolate through and redissolve any precipitated sulfide. Transient sulfide saturation
661 during the chromite formation and subsequent re-dissolution of the sulfide and re-distribution
662 through percolating melt would provide an explanation for the seemingly paradoxical binary
663 distribution of relatively Ru rich chromitite with Ru poor chromite on one hand, and Ru-poor
664 chromitite with the very localised relatively Ru-rich chromite on the other.

665 Implications for the Ru in chromite mineral indicator

666 If the explanation proposed here is correct, then Ru in chromite in the particular case
667 of chromitite seams is highly susceptible to even at grain scale, transient style sulfide
668 saturation. If sampled in an indicator mineral campaign, chromite grains derived from such
669 an environment would give a false positive result for the exploration of magmatic Ni-Cu-
670 PGE sulfides.

671 Even so, as these chromites are from plutonic settings, they have a high proportion of
672 very coarse grains while as extrusive settings tend to have much smaller grains. Prospective
673 regions could potentially be identified by a combination of chromite grain size and chemistry,
674 along with the integration of additional exploration proxies, such as Ni and Cu in olivine
675 (Locmelis *et al.* 2019, Barnes *et al.* in press), the trace element content of arsenides (Le
676 Vaillant *et al.* 2018), and zoning pattern in pyroxene (Schoneveld *et al.* 2020). Furthermore,
677 chromitite is an extremely rare rock type, even within large ultramafic-dominated intrusions,
678 so its contribution to regional sample populations is likely to be small compared with
679 chromite derived from much more voluminous rocks containing disseminated chromite.

680 Regardless, given the increased demand for battery metals for the green energy
681 transition, the presence of a false positive leading to continued exploration may be more
682 desirable than a false negative, where a potential resource is missed entirely.

683 Conclusions

684 The Coobina Intrusion is a derivation of a komatiite magma from a convecting upper
685 mantle source that evolved with long-term near-chondritic $^{187}\text{Re}/^{188}\text{Os}$ ratio. The calculated

686 average Re-Os model ages for the chromitites, $T_{MA} = 3.28 \pm 0.03$ Ga and $T_{RD} = 3.25 \pm 0.08$ Ga,
687 are consistent with the accepted emplacement age of the Coobina intrusion at 3.189 Ga.

688 Chromite in the Coobina Intrusion shows a wide variability in its Ru content, which
689 shows an inverse relationship with the Ru content in whole-rock. There are two possible
690 explanations for this phenomenon:

691 1) The Ru partition coefficient for chromite is locally lowered via changes in redox
692 conditions, transient sulfide accumulation or a combination of both. The system reaches
693 saturation in Ru-rich PGMs to form micron-scale nuggets that precipitate as a cumulus phase.
694 These cumulate PGMs settle into the lower regions, increasing the Ru content in a rock
695 containing previously crystallised low-Ru chromite.

696 2) The high Ru rocks were formed under conditions of transient sulfide saturation
697 during chromite growth, increasing the whole-rock Ru content, but lowering the Ru content
698 in solid solution in chromite. The sulfide phases then re-dissolve during late-stage melt
699 percolation and are re-distributed as PGMs. This process is envisaged as being similar to that
700 responsible for anomalously high IPGE concentrations in some unusual ophiolitic chromitites
701 and is related more generally to the global PGE-chromite-sulfide association.

702 The ratio of Ru between chromite in solid solution and whole-rock chromitite is not
703 correlated with any crystal chemical component. This favours the transient sulfide saturation
704 mechanism rather than a change in oxidation state, which would be evident in the major
705 element composition of the chromite, particularly its trivalent cation (i.e., Fe^{3+} , Al^{3+} , Cr^{3+})
706 content. Furthermore, as there is no textural evidence of recrystallisation in the studied
707 samples, the process of recrystallisation and redistribution during slow cooling does not
708 appear to have played a major role.

709 Although this intrusion was mined for many years for its chromium resource, there
710 has been no sulfide observed within this intrusion. The chromite within the Coobina
711 intrusions has Ru in solid solution of values of both >150 ppb Ru and < 150 ppb Ru. As low
712 (<150 ppb) Ru content in komatiitic derived chromite has been suggested to indicate sulfide
713 saturation, we suggest that the Coobina intrusion has undergone transient sulfide saturation at
714 small scale. When using chromite as a detrital indicator mineral for the exploration for nickel
715 sulfide, the possibility of transient sulfide saturation should be considered and may indicate a
716 false positive result.

717

718

719 **Tables**

720 *Table 1: Chromitite samples analysed for PGE and trace elements within Chromite with reference to the pits outlined in*
 721 *Figure 1*

Seam	Locality group	Trend-Area	Pit	Easting	Northing
4R	Blatchfords-C5	East	Blatchfords	222015.5	7399163
BM1	Blatchfords-C5	East	Blatchfords Main	222067	7399188
BM2	Blatchfords-C5	East	Blatchfords Main	222059	7399199
BC1	Blatchfords-C5	East	Blatchfords Main	222088	7399339
4K	Blatchfords-C5	East	Central Five	221804.6	7399051
2Q	Falcon-Skywest	Central	Falcon	221432	7399240
SW1	Falcon-Skywest	Central	Sky West	221547	7399402
1G	Finucane	West	Wrights	221218.6	7399865
1T	Finucane East	Central	Finucane	221491.6	7399871
FT3	Finucane East	Central	Finucane East	221540	7399843
NF2	Newlands 246	West	Newlands Four Pit	220945	7399305
NS3	Newlands 246	West	Newlands Six Pit	220945	7399102
3F	Newlands Ck	West	Newlands	221209	7399188
NC2	Newlands Ck	West	Newlands Creek Pit	221085	7398970
NE1	North Eastern Pits	NorthEast	North Eastern Pits	222097	7400448
5B	Tish-Wedgetail	East	Tish Creek	221798.5	7399345
WS2	Wrights	West	Wrights South Pit	221344	7399699
WS4	Wrights	West	Wrights South Pit	221344	7399699

722

723 *Table 2: Re-Os isotopic and Re and PGE abundance data (in ppb) for the Coobina chromites*

Sample No.	Chr1_1	Chr1_2	Chr2
Re	0.06239	0.1134	0.06305
Os	408.2	152.4	38.16
Ir	217.0	111.1	46.45
Ru	830.2	312.3	188.1
Pt	6.939	6.916	4.073
Pd	1.464	5.330	1.105
$^{187}\text{Re}/^{188}\text{Os}$	0.00073 ± 0.0004	0.00358 ± 0.00035	0.00794 ± 0.00037
$^{187}\text{Os}/^{188}\text{Os}$	0.10447 ± 6	0.10450 ± 6	0.10496 ± 6
$\gamma^{187}\text{Os(T)}$	-0.63	-0.74	-0.53
TMA, Ga	3.28	3.30	3.27
TRD, Ga	3.27	3.27	3.20

724 Initial $\gamma^{187}\text{Os}$ values were calculated at $T = 3.189$ Ga.

725 *Table 3: Mass balance Calculations as given by $C_i^{(melt)} = C_i^{WR} - (C_i^{Cr-Spinel} \times F_i^{(Cr-Spinel)})$ (Pagé and Barnes, 2016). C_i^{WR} is the*
 726 *concentration of the PGE within the whole rock and $C_i^{Cr-Spinel}$ is the concentration of the PGE within the chromite lattice,*
 727 *$F_i^{(Cr-Spinel)}$ = % PGE from the whole rock accounted for by the chromite. $C_i^{(melt)}$ is the concentrations of the PGE within the*
 728 *non-chromite component of the rock*

Seam	Ru ppb WR	Ru ppb Chr	% Cr-Spinel in sample	% PGE from the whole rock accounted for by the chromite			$C_{\text{Ir}}^{(melt)}$	$C_{\text{Ru}}^{(melt)}$	$C_{\text{Rh}}^{(melt)}$
				$F_{\text{Ir}}^{(\text{Cr-Spinel})}$	$F_{\text{Ru}}^{(\text{Cr-Spinel})}$	$F_{\text{Rh}}^{(\text{Cr-Spinel})}$			
BC1	70.24	227.84	68%	61%	222%	225%	10.64	-85.35	-7.44
BM1	73.1	240.85	78%	30%	257%	299%	29.88	-114.44	-13.89
NC2	86.96	244.16	77%	48%	216%	61%	30.33	-100.93	11.31
NS3	114.69	88.54	76%	42%	59%	71%	30.77	47.05	4.31
BM2	115.39	82.83	61%	28%	44%	78%	32.41	64.51	2.25
WS4	142.69	14.23	71%	22%	7%		30.65	132.61	
FT3	145.01	85.88	71%	39%	42%	59%	28.59	83.7	3.25
NF2	152.89	30.64	77%	18%	15%	18%	51.44	129.32	11.83
NE1	154.57	28.54	78%	53%	14%	10%	23.46	132.42	12.88
SW1	165.96	50.31	75%	29%	23%	14%	37.96	128.44	15.03
WS2	174.41	26.79	78%	12%	12%	10%	60.46	153.47	11.65

729

730 **Figures**

731

732 *Figure 1: Location map – A) regional geology of the Sylvania Inlier and the Coobina Intrusion, modified from Williams and*
733 *Tyler (1991), B) Local geology of the Jimblebar field area from GSWA online (1:250000 data set) with interpreted >10km*
734 *long feeder dyke Williams and Tyler (1991). C) Detailed geological map showing the Coobina ultramafic sill with pit*
735 *outlines, and Cr sample locations (from Barnes and Jones (2013)). With geological data from Jeffrey (2003; internal*
736 *company report) and S. Jones. GDA1994 MGA Zone 51*

737

738 *Figure 2: A) Large northeast-trending chromite seams in Newlands Creek pit, B) finer scale layering of chromite seams in*
739 *Falcon Pit C) folded chromite lenses from Newlands Pit. Images courtesy of (Barnes & Jones 2013)*

740

741 *Figure 3: Reflected light images of the samples in this study. LA-ICP-MS ablation craters visible in some samples (150µm*
742 *diameter), line in bottom left of each images is 500um. See Table 1 for locations for each sample.*

743

744 *Figure 4: (A) crystal size distribution of chromite, EDS is Equivalent circle diameter in µm. (B to D) SEM images showing*
745 *representative chromitite textures*

746

747 *Figure 5: EPMA data for Coobina (from Barnes and Jones (2013)), Bushveld, Great Dyke, Stillwater (Barnes and Roeder,*
748 *2001) and komatiitic chromite (Locmelis et al. (2018)) and fields of spinel compositions from layered intrusions –*
749 *chromitites and Al-undepleted komatites (AUDK) from Barnes & Roeder (2001)*

750

751 *Figure 6: Trace elements and platinum group elements against Cr/(Cr+Al). Shown are average EPMA and LA-ICP-MS*
752 *analysis for each sample. The error is one standard deviation from analysis on multiple chromite grains in a single sample.*
753 *Colour represents mine pit and shape differentiates seams within this pit*

754

755 *Figure 7: LA-ICP-MS analysis of PGE in chromite from the Coobina intrusion*

756

757 *Figure 8: Distribution of Ruthenium in Chromite. Higher concentrations (>300 ppb) in the south-east, toward the feeder*
758 *dyke and low concentrations (< 50 ppb) in the north-west. Map modified from Barnes and Jones (2013). GDA1994 MGA*
759 *Zone 51.*

760

761 *Figure 9: (A) Average LA-ICP-MS analysis of the PGE in chromite, compared to (B) Whole-rock PGE analysis of the*
762 *corresponding samples described by Barnes and Jones (2013). Values in (A) and (B) are normalised to the composition of*
763 *the primitive mantle (Palme & O'Neill 2014) and plotted with pyrolite (Williams et al. 2020). (C to F) Comparison of the*
764 *whole rock PGE content with the chromite analysis for individual sample averages*

765

766 *Figure 10: Frequency and composition of chromite-hosted micro-nuggets in Coobina samples. Inclusion type given by EDS*
767 *spectra for major content only*

768

769 *Figure 11: Initial $^{187}\text{Os}/^{188}\text{Os}$ isotopic compositions, expressed in $\gamma^{187}\text{Os}$ terms, of the Archean and Proterozoic komatiite*
770 *systems plotted as a function of age. The blue bar for the modern BSE estimate represents the 2SD of the mean from Meisel*
771 *et al. (2001). The sources of the data are as follows. Lapland – Puchtel et al. (2020); Veteny Belt – Puchtel et al. (2016b);*
772 *Belingwe – Puchtel et al. (2009b); Boston Creek – Puchtel et al. (2018); Pyke Hill – Puchtel et al. (2004); Kostomuksha –*
773 *Puchtel & Humayun (2005); Volotsk – Puchtel et al. (2007); Weltevreden and Komati – Puchtel et al. (2014); Schapenburg –*
774 *Puchtel et al. (2009a); Puchtel et al. (2016a); Kelly, Ruth Well, and Regal – Puchtel et al. (2022); Coobina – this study.*
775 *Uncertainties are 2SD. The Os isotopic data for chondritic meteorites are compiled from (Walker et al. 2002), (Brandon et*
776 *al. 2005)), and (Fischer-Gödde et al. 2010) and are plotted as an envelope enclosed between the slanting purple lines and*
777 *corresponding to the entire range of calculated modern $\gamma^{187}\text{Os}$ values projected back to the Solar System initial $^{187}\text{Os}/^{188}\text{Os}$*
778 *ratio. All uncertainties are 2SD of the mean..*

779

780 *Figure 12: A-D) SEM images of bright phases within chromite E) Laser ablation maps (35um square laser spot) of sample*
781 *WS2*

782

783 *Figure 13: Nickel content of chromite from the Coobina intrusion compared to data from Bushveld Complex (Park et al.*
784 *2012), komatiites, komatiitic basalts and ferropicrite (from Locmelis et al. 2018)*

785

786 *Figure 14: Ruthenium contents of chromitites and chromite from Coobina and comparison to stratiform chromitites and*
787 *chromite from the Bushveld Complex, Stillwater, Great Dyke and Black Thor with data from Pagé and Barnes (2016). Whole*
788 *Rock Ru values were calculated by the measured whole rock value divided by the proportion of chromite in the chromitite as*
789 *shown in Table 3*

790 **Supplementary Figure**

791

792 *Figure 15: [Supplementary Figure] Trace elements against $\text{Fe}^{3+}/(\text{Al}+\text{Cr}+\text{Fe}^{3+})$ ratio. Average EPMA and LA-ICP-MS*
793 *analysis for each sample. The error is one standard deviation from analysis on multiple chromite grains in a single sample.*
794 *Colour represents locality and shape differentiates seams within this locality*

795

796 **References**

797 ALARD, O., GRIFFIN, W.L., LORAND, J.P., JACKSON, S.E. & O'REILLY, S.Y. (2000) Non-chondritic
798 distribution of the highly siderophile elements in mantle sulphides. *Nature* **407**, 891-894,
799 <https://doi.org/10.1038/35038049>.

800 BARNES, S.-J., MAIER, W.D. & ASHWAL, L.D. (2004) Platinum-group element distribution in the
801 Main Zone and Upper Zone of the Bushveld Complex, South Africa. *Chemical Geology* **208**,
802 293-317, <https://doi.org/10.1016/j.chemgeo.2004.04.018>.

803 BARNES, S., MAO, Y.-J., YAO, Z.-S., JESUS, A.B., YANG, S., TARANOVIC, V. & MAIER, W.D. (in
804 press) Nickel in olivine as an exploration indicator for magmatic Ni-Cu sulfide deposits: a
805 data review and re-evaluation. <https://doi.org/10.2138/am-2022-8327>,
806 <https://doi.org/10.2138/am-2022-8327>.

807 BARNES, S.J. (1998) Chromite in Komatiites, I. Magmatic Controls on Crystallization and
808 Composition. *Journal of Petrology* **39**, 32, <https://doi.org/10.1093/petroj/39.10.1689>.

809 BARNES, S.J. & ROEDER, P.L. (2001) The Range of Spinel Compositions in Terrestrial Mafic and
810 Ultramafic Rocks. *Journal of Petrology* **42**, 24, <https://doi.org/10.1093/petrology/42.12.2279>.

811 BARNES, S.J. & FIORENTINI, M.L. (2008) Iridium, ruthenium and rhodium in komatiites: Evidence for
812 iridium alloy saturation. *Chemical Geology* **257**, 44-58,
813 <https://doi.org/10.1016/j.chemgeo.2008.08.015>.

814 BARNES, S.J. & JONES, S. (2013) Deformed Chromitite Layers in the Coobina Intrusion, Pilbara
815 Craton, Western Australia*. *Economic Geology* **108**, 337-354,
816 <https://doi.org/10.2113/econgeo.108.2.337>.

817 BARNES, S.J., RYAN, C., MOORHEAD, G., LATYPOV, R., MAIER, W.D., YUDOVSKAYA, M., GODEL, B.,
818 SCHONEVELD, L.E., LE VAILLANT, M. & PEARCE, M.B. (2021) Spatial association between
819 platinum minerals and magmatic sulfides imaged with the Maia Mapper and implications for
820 the origin of the chromite-sulfide-PGE association. *The Canadian Mineralogist* **59**, 1775-
821 1799, <https://doi.org/10.3749/canmin.2000100>.

822 BÉDARD, L.P. & BARNES, S.-J. (2004) Improved Platinum-Group Element Extraction by NiS Fire
823 Assay from Chromitite Ore Samples Using a Flux Containing Sodium Metaphosphate. **28**,
824 311-316, <https://doi.org/10.1111/j.1751-908X.2004.tb00746.x>.

825 BIRCK, J.L., BARMAN, M.R. & CAPMAS, F. (1997) Re-Os Isotopic Measurements at the Femtomole
826 Level in Natural Samples. **21**, 19-27, <https://doi.org/10.1111/j.1751-908X.1997.tb00528.x>.

827 BORISOV, A.V. & NACHTWEYH, K. (1998) Ru Solubility in Silicate Melts: Experimental Results in
828 Oxidizing Region. abstract no. 1320.

829 BRANDON, A.D., HUMAYUN, M., PUCHTEL, I.S. & ZOLENSKY, M.E. (2005) Re-Os isotopic
830 systematics and platinum group element composition of the Tagish Lake carbonaceous
831 chondrite. *Geochimica et Cosmochimica Acta* **69**, 1619-1631,
832 <https://doi.org/10.1016/j.gca.2004.10.005>.

833 BRENAN, J.M., FINNIGAN, C.F., McDONOUGH, W.F. & HOMOLOVA, V. (2012) Experimental
834 constraints on the partitioning of Ru, Rh, Ir, Pt and Pd between chromite and silicate melt:
835 The importance of ferric iron. *Chemical Geology* **302**, 16-32,
836 <https://doi.org/10.1016/j.chemgeo.2011.05.015>.

837 CABRI, L., HARRIS, D.C. & WEISER, T. (1996) Mineralogy and distribution of platinum-group mineral
838 (PGM) placer deposits of the world. *Exploration and Mining Geology* **5**, 73-167.

839 CANIL, D. & O'NEILL, H.S.C. (1996) Distribution of Ferric Iron in some Upper-Mantle Assemblages.
840 *Journal of Petrology* **37**, 609-635, <https://doi.org/10.1093/petrology/37.3.609>.

841 CAWOOD, P.A. & TYLER, I.M. (2004) Assembling and reactivating the Proterozoic Capricorn Orogen:
842 lithotectonic elements, orogenies, and significance. *Precambrian Research* **128**, 201-218,
843 <https://doi.org/10.1016/j.precamres.2003.09.001>.

844 COHEN, A.S. & WATERS, F.G. (1996) Separation of osmium from geological materials by solvent
845 extraction for analysis by thermal ionisation mass spectrometry. *Analytica Chimica Acta* **332**,
846 269-275, [https://doi.org/10.1016/0003-2670\(96\)00226-7](https://doi.org/10.1016/0003-2670(96)00226-7).

847 CREASER, R.A., PAPANASTASSIOU, D.A. & WASSERBURG, G.J. (1991) Negative thermal ion mass
848 spectrometry of osmium, rhenium and iridium. *Geochimica et Cosmochimica Acta* **55**, 397-
849 401, [https://doi.org/10.1016/0016-7037\(91\)90427-7](https://doi.org/10.1016/0016-7037(91)90427-7).

850 FINNIGAN, C.S., BRENAN, J.M., MUNGALL, J.E. & McDONOUGH, W.F. (2008) Experiments and
851 Models Bearing on the Role of Chromite as a Collector of Platinum Group Minerals by Local
852 Reduction. *Journal of Petrology* **49**, 1647-1665, <https://doi.org/10.1093/petrology/egn041>.

853 FIORENTINI, M.L., STONE, W.E., BERESFORD, S.W. & BARLEY, M.E. (2004) Platinum-group element
854 alloy inclusions in chromites from Archaean mafic-ultramafic units: evidence from the
855 Abitibi and the Agnew-Wiluna Greenstone Belts. *Mineralogy and Petrology* **82**, 341-355,
856 10.1007/s00710-004-0044-6.

857 FIORENTINI, M.L., BERESFORD, S.W. & BARLEY, M.E. (2008) Ruthenium–chromium variation: a new
858 lithogeochemical tool in the exploration for komatiite-hosted Ni–Cu–(PGE) deposits.
859 *Economic geology and the bulletin of the Society of Economic Geologists* **103**, 431-437,
860 <http://dx.doi.org/10.2113/gsecongeo.103.2.431>.

861 FISCHER-GÖDDE, M., BECKER, H. & WOMBACHER, F. (2010) Rhodium, gold and other highly
862 siderophile element abundances in chondritic meteorites. *Geochimica et Cosmochimica Acta*
863 **74**, 356-379, <https://doi.org/10.1016/j.gca.2009.09.024>.

864 FONSECA, R.O.C., LAURENZ, V., MALLMANN, G., LUGUET, A., HOEHNE, N. & JOCHUM, K.P. (2012)
865 New constraints on the genesis and long-term stability of Os-rich alloys in the Earth's mantle.
866 *Geochimica et Cosmochimica Acta* **87**, 227-242, <https://doi.org/10.1016/j.gca.2012.04.002>.

867 GODEL, B., BARNES, S.-J. & MAIER, W.D. (2007) Platinum-Group Elements in Sulphide Minerals,
868 Platinum-Group Minerals, and Whole-Rocks of the Merensky Reef (Bushveld Complex,
869 South Africa): Implications for the Formation of the Reef. *Journal of Petrology* **48**, 1569-
870 1604, 10.1093/petrology/egm030 %J Journal of Petrology.

871 GONZÁLEZ-JIMÉNEZ, J.M., LOCMELIS, M., BELOUSOVA, E., GRIFFIN, W.L., GERVILLA, F.,
872 KERESTEDJIAN, T.N., O'REILLY, S.Y., PEARSON, N.J. & SERGEEVA, I. (2015) Genesis and
873 tectonic implications of podiform chromitites in the metamorphosed ultramafic massif of
874 Dobromirtsi (Bulgaria). *Gondwana Research* **27**, 555-574,
875 <https://doi.org/10.1016/j.gr.2013.09.020>.

876 HIGGINS, M.D. (2000) Measurement of crystal size distribution. *American Mineralogist* **85**, 1105-
877 1116.

878 HIGGINS, M.D. (2006) *Quantitative textural measurements in igneous and metamorphic petrology*.
879 Cambridge University Press, Cambridge.

880 HUNT, E.J., O'DRISCOLL, B. & DAY, J.M.D. (2021) Sintering as a key process in the textural evolution
881 of chromitite seams in layered mafic-ultramafic intrusions. *The Canadian Mineralogist* **59**,
882 1661-1692, 10.3749/canmin.2100021 %J The Canadian Mineralogist.

883 LE VAILLANT, M., BARNES, S.J., FIORENTINI, M.L., BARNES, S.-J., BATH, A. & MILLER, J. (2018)
884 Platinum-group element and gold contents of arsenide and sulfarsenide minerals associated
885 with Ni and Au deposits in Archean greenstone belts. *Mineralogical Magazine* **82**, 625-647,
886 10.1180/minmag.2017.081.100.

887 LOCMELIS, M., MELCHER, F. & OBERTHÜR, T. (2010) Platinum-group element distribution in the
888 oxidized Main Sulfide Zone, Great Dyke, Zimbabwe. *Mineralium Deposita* **45**, 93-109,
889 10.1007/s00126-009-0258-y.

890 LOCMELIS, M., PEARSON, N.J., BARNES, S.J. & FIORENTINI, M.L. (2011) Ruthenium in komatiitic
891 chromite. *Geochimica et Cosmochimica Acta* **75**, 3645-3661,
892 <https://doi.org/10.1016/j.gca.2011.03.041>.

893 LOCMELIS, M., FIORENTINI, M.L., BARNES, S.J. & PEARSON, N.J. (2013) Ruthenium variation in
894 chromite from komatiites and komatiitic basalts - a potential mineralogical indicator for
895 nickel sulfide mineralization. *Economic Geology* **108**, 355-364,
896 <https://doi.org/10.2113/econgeo.108.2.355>.

897 LOCMELIS, M., FIORENTINI, M.L., BARNES, S.J., HANSKI, E.J. & KOBUSSEN, A.F. (2018) Ruthenium
898 in chromite as indicator for magmatic sulfide liquid equilibration in mafic-ultramafic systems.
899 *Ore Geology Reviews* **97**, 152-170, <https://doi.org/10.1016/j.oregeorev.2018.05.002>.

900 LOCMELIS, M., AREVALO, R.D., JR., PUCHTEL, I.S., FIORENTINI, M.L. & NISBET, E.G. (2019)
901 Transition metals in komatiitic olivine: Proxies for mantle composition, redox conditions, and
902 sulfide mineralization potential. *American Mineralogist* **104**, 1143-1155,
903 <https://doi.org/10.2138/am-2019-6914>.

904 MAIER, W.D., BARNES, S.J., CAMPBELL, I.H., FIORENTINI, M.L., PELTONEN, P., BARNES, S.-J. &
905 SMITHIES, R.H. (2009) Progressive mixing of meteoritic veneer into the early Earth's deep
906 mantle. *Nature* **460**, 620-623, 10.1038/nature08205.

907 NALDRETT, A.J., KINNAIRD, J., WILSON, A., YUDOVSKAYA, M., MCQUADE, S., CHUNNETT, G. &
908 STANLEY, C. (2009a) Chromite composition and PGE content of Bushveld chromitites: Part 1
909 – the Lower and Middle Groups. *Applied Earth Science* **118**, 131-161,
910 <https://doi.org/10.1179/174327509X12550990458004>.

911 NALDRETT, A.J., WILSON, A., KINNAIRD, J. & CHUNNETT, G. (2009b) PGE Tenor and Metal Ratios
912 within and below the Merensky Reef, Bushveld Complex: Implications for its Genesis.
913 *Journal of Petrology* **50**, 625-659, 10.1093/petrology/egp015.

914 NICKLAS, R.W., PUCHTEL, I.S., ASH, R.D., PICCOLI, P.M., HANSKI, E., NISBET, E.G., WATERTON, P.,
915 PEARSON, D.G. & ANBAR, A.D. (2019) Secular mantle oxidation across the Archean-
916 Proterozoic boundary: Evidence from V partitioning in komatiites and picrites. *Geochimica et*
917 *Cosmochimica Acta* **250**, 49-75, <https://doi.org/10.1016/j.gca.2019.01.037>.

918 NORMAN, M.D., PEARSON, N.J., SHARMA, A. & GRIFFIN, W.L. (1996) Quantitative analysis of trace
919 elements in geological materials by laser ablation ICPMS: instrumental operating conditions
920 and calibration values of NIST glasses. *20*, 247-261, <https://doi.org/10.1111/j.1751-908X.1996.tb00186.x>.

921 OCCHIPINTI, S.A., SHEPPARD, S., PASSCHIER, C., TYLER, I.M. & NELSON, D.R. (2004)
922 Palaeoproterozoic crustal accretion and collision in the southern Capricorn Orogen: the
923 Glenburgh Orogeny. *Precambrian Research* **128**, 237-255,
924 <https://doi.org/10.1016/j.precamres.2003.09.002>.

925 PAGÉ, P., BARNES, S.-J., BÉDARD, J.H. & ZIENTEK, M.L. (2012) In situ determination of Os, Ir, and
926 Ru in chromites formed from komatiite, tholeiite and boninite magmas: Implications for
927 chromite control of Os, Ir and Ru during partial melting and crystal fractionation. *Chemical*
928 *Geology* **302**, 3-15, <https://doi.org/10.1016/j.chemgeo.2011.06.006>.

929 PAGÉ, P. & BARNES, S.-J. (2016) The influence of chromite on osmium, iridium, ruthenium and
930 rhodium distribution during early magmatic processes. *Chemical Geology* **420**, 51-68,
931 <https://doi.org/10.1016/j.chemgeo.2015.11.002>.

932 PAGÉ, P. & BARNES, S.-J. (2009) Using Trace Elements in Chromites to Constrain the Origin of
933 Podiform Chromitites in the Thetford Mines Ophiolite, Québec, Canada. *Economic Geology*
934 **104**, 997-1018, <https://doi.org/10.2113/econgeo.104.7.997>.

935 PARK, J.-W., CAMPBELL, I.H. & EGGIN, S.M. (2012) Enrichment of Rh, Ru, Ir and Os in Cr spinels
936 from oxidized magmas: Evidence from the Ambae volcano, Vanuatu. *Geochimica et*
937 *Cosmochimica Acta* **78**, 28-50, <http://dx.doi.org/10.1016/j.gca.2011.11.018>.

938 PARK, J.-W., KAMENETSKY, V., CAMPBELL, I., PARK, G., HANSKI, E. & PUSHKAREV, E. (2017)
939 Empirical constraints on partitioning of platinum group elements between Cr-spinel and
940 primitive terrestrial magmas. *Geochimica et Cosmochimica Acta* **216**, 393-416,
941 <https://doi.org/10.1016/j.gca.2017.05.039>.

942 PATON, C., HELLSTROM, J., PAUL, B., WOODHEAD, J. & HERGT, J. (2011) Iolite: Freeware for the
943 visualisation and processing of mass spectrometric data. *Journal of Analytical Atomic*
944 *Spectrometry* **26**, 2508-2518, <https://doi.org/10.1039/C1JA10172B>.

945 PRICHARD, H.M., J., B.S., DALE, C.W., GODEL, B., FISHER, P.C. & NOWELL, G.M. (2017) Paragenesis
946 of multiple platinum-group mineral populations in Shetland ophiolite chromitite: 3D X-ray
947 tomography and in situ Os isotopes. *Geochimica et Cosmochimica Acta* **216**, 314-334,
948 <http://dx.doi.org/10.1016/j.gca.2017.03.035>.

949 PUCHTEL, I.S., HUMAYUN, M., CAMPBELL, A.J., SPROULE, R.A. & LESHER, C.M. (2004) Platinum
950 group element geochemistry of komatiites from the Alexo and Pyke Hill areas, Ontario,
951 Canada 11Associate editor: R. J. Walker. *Geochimica et Cosmochimica Acta* **68**, 1361-1383,
952 <https://doi.org/10.1016/j.gca.2003.09.013>.

953

954 PUCHTEL, I.S. & HUMAYUN, M. (2005) Highly siderophile element geochemistry of 187Os-enriched
 955 2.8 Ga Kostomuksha komatiites, Baltic Shield. *Geochimica et Cosmochimica Acta* **69**, 1607-
 956 1618, <https://doi.org/10.1016/j.gca.2004.09.007>.

957 PUCHTEL, I.S., HUMAYUN, M. & WALKER, R.J. (2007) Os–Pb–Nd isotope and highly siderophile and
 958 lithophile trace element systematics of komatiitic rocks from the Volotsk suite, SE Baltic
 959 Shield. *Precambrian Research* **158**, 119-137,
 960 <https://doi.org/10.1016/j.precamres.2007.04.004>.

961 PUCHTEL, I.S., WALKER, R.J., ANHAEUSSER, C.R. & GRUAU, G. (2009a) Re–Os isotope systematics
 962 and HSE abundances of the 3.5 Ga Schapenburg komatiites, South Africa: Hydrous melting
 963 or prolonged survival of primordial heterogeneities in the mantle? *Chemical Geology* **262**,
 964 355-369, <https://doi.org/10.1016/j.chemgeo.2009.02.006>.

965 PUCHTEL, I.S., WALKER, R.J., BRANDON, A.D. & NISBET, E.G. (2009b) Pt–Re–Os and Sm–Nd
 966 isotope and HSE and REE systematics of the 2.7Ga Belingwe and Abitibi komatiites.
 967 *Geochimica et Cosmochimica Acta* **73**, 6367-6389, <https://doi.org/10.1016/j.gca.2009.07.022>.

968 PUCHTEL, I.S., WALKER, R.J., TOUBOUL, M., NISBET, E.G. & BYERLY, G.R. (2014) Insights into early
 969 Earth from the Pt–Re–Os isotope and highly siderophile element abundance systematics of
 970 Barberton komatiites. *Geochimica et Cosmochimica Acta* **125**, 394-413,
 971 <https://doi.org/10.1016/j.gca.2013.10.013>.

972 PUCHTEL, I.S., BLICHERT-TOFT, J., TOUBOUL, M., HORAN, M.F. & WALKER, R.J. (2016a) The
 973 coupled 182W-142Nd record of early terrestrial mantle differentiation. **17**, 2168-2193,
 974 <https://doi.org/10.1002/2016GC006324>.

975 PUCHTEL, I.S., TOUBOUL, M., BLICHERT-TOFT, J., WALKER, R.J., BRANDON, A.D., NICKLAS, R.W.,
 976 KULIKOV, V.S. & SAMSONOV, A.V. (2016b) Lithophile and siderophile element systematics
 977 of Earth's mantle at the Archean–Proterozoic boundary: Evidence from 2.4Gakomatiites.
 978 *Geochimica et Cosmochimica Acta* **180**, 227-255, <https://doi.org/10.1016/j.gca.2016.02.027>.

979 PUCHTEL, I.S., BLICHERT-TOFT, J., TOUBOUL, M. & WALKER, R.J. (2018) 182W and HSE constraints
 980 from 2.7 Ga komatiites on the heterogeneous nature of the Archean mantle. *Geochimica et*
 981 *Cosmochimica Acta* **228**, 1-26, <https://doi.org/10.1016/j.gca.2018.02.030>.

982 PUCHTEL, I.S., MUNDL-PETERMEIER, A., HORAN, M., HANSKI, E.J., BLICHERT-TOFT, J. & WALKER,
 983 R.J. (2020) Ultra-depleted 2.05 Ga komatiites of Finnish Lapland: Products of grainy late
 984 accretion or core–mantle interaction? *Chemical Geology* **554**, 119801,
 985 <https://doi.org/10.1016/j.chemgeo.2020.119801>.

986 PUCHTEL, I.S., NICKLAS, R.W., SLAGLE, J., HORAN, M., WALKER, R.J., NISBET, E.G. & LOCMELIS,
 987 M. (2022) Early global mantle chemical and isotope heterogeneity revealed by the komatiite-
 988 basalt record: The Western Australia connection. *Geochimica et Cosmochimica Acta* **320**,
 989 238-278, <https://doi.org/10.1016/j.gca.2021.11.030>.

990 REHKÄMPER, M. & HALLIDAY, A.N. (1997) Development and application of new ionexchange
 991 techniques for the separation of the platinum group and other siderophile elements from
 992 geological samples. *Talanta* **44**, 663-672, [https://doi.org/10.1016/S0039-9140\(96\)02100-5](https://doi.org/10.1016/S0039-9140(96)02100-5).

993 RIGHTER, K., CAMPBELL, A.J., HUMAYUN, M. & HERVIG, R.L. (2004) Partitioning of Ru, Rh, Pd, Re,
 994 Ir, and Au between Cr-bearing spinel, olivine, pyroxene and silicate melts. *Geochimica et*
 995 *Cosmochimica Acta* **68**, 867-880, <https://doi.org/10.1016/j.gca.2003.07.005>.

996 SCHINDELIN, J., ARGANDA-CARRERAS, I., FRISE, E., KAYNIG, V., LONGAIR, M., PIETZSCH, T.,
 997 PREIBISCH, S., RUEDEN, C., SAALFELD, S., SCHMID, B., TINEVEZ, J.-Y., WHITE, D.J.,
 998 HARTENSTEIN, V., ELICEIRI, K., TOMANCAK, P. & CARDONA, A. (2012) Fiji: an open-source
 999 platform for biological-image analysis. *Nature Methods* **9**, 676, 10.1038/nmeth.2019
 1000 <https://www.nature.com/articles/nmeth.2019#supplementary-information>.

1001 SCHONEVELD, L., BARNES, S.J., MAKKONEN, H.V., LE VAILLANT, M., PATERSON, D.J., TARANOVIC,
 1002 V., WANG, K.-Y. & MAO, Y.-J. (2020) Zoned Pyroxenes as Prospectivity Indicators for
 1003 Magmatic Ni–Cu Sulfide Mineralization. *Frontiers in Earth Science* **8**,
 10.3389/feart.2020.00256.

1004 SHIREY, S.B. & WALKER, R.J. (1998) The Re–Os isotope system in cosmochemistry and high-
 1005 temperature geochemistry. **26**, 423-500, <https://doi.org/10.1146/annurev.earth.26.1.423>.

1007 SMITH, W., MAIER, W., BARNES, S., MOORHEAD, G., REID, D. & KARYKOWSKI, B. (2021) Element
1008 mapping the Merensky Reef of the Bushveld Complex. *Geoscience Frontiers* **12**, 101101,
1009 <https://doi.org/10.1016/j.gsf.2020.11.001>.

1010 SMOLIAR, M.I., WALKER, R.J. & MORGAN, J.W. (1996) Re-Os Ages of Group IIA, IIIA, IVA, and
1011 IVB Iron Meteorites. **271**, 1099-1102, <https://doi.org/10.1126/science.271.5252.1099>.

1012 TEIGLER, B. & EALES, H.V. (1993) Correlation between chromite composition and PGE
1013 mineralization in the Critical Zone of the western Bushveld Complex. *Mineralium Deposita*
1014 **28**, 291-302, 10.1007/BF02739368.

1015 TYLER, I.M. (1991) *The geology of the Sylvania Inlier and the southeast Hamersley Basin*. East Perth,
1016 W.A : Geological Survey of Western Australia, East Perth, W.A.

1017 WALKER, R.J., HORAN, M.F., MORGAN, J.W., BECKER, H., GROSSMAN, J.N. & RUBIN, A.E. (2002)
1018 Comparative 187Re-187Os systematics of chondrites: Implications regarding early solar
1019 system processes. *Geochimica et Cosmochimica Acta* **66**, 4187-4201,
1020 [https://doi.org/10.1016/S0016-7037\(02\)01003-7](https://doi.org/10.1016/S0016-7037(02)01003-7).

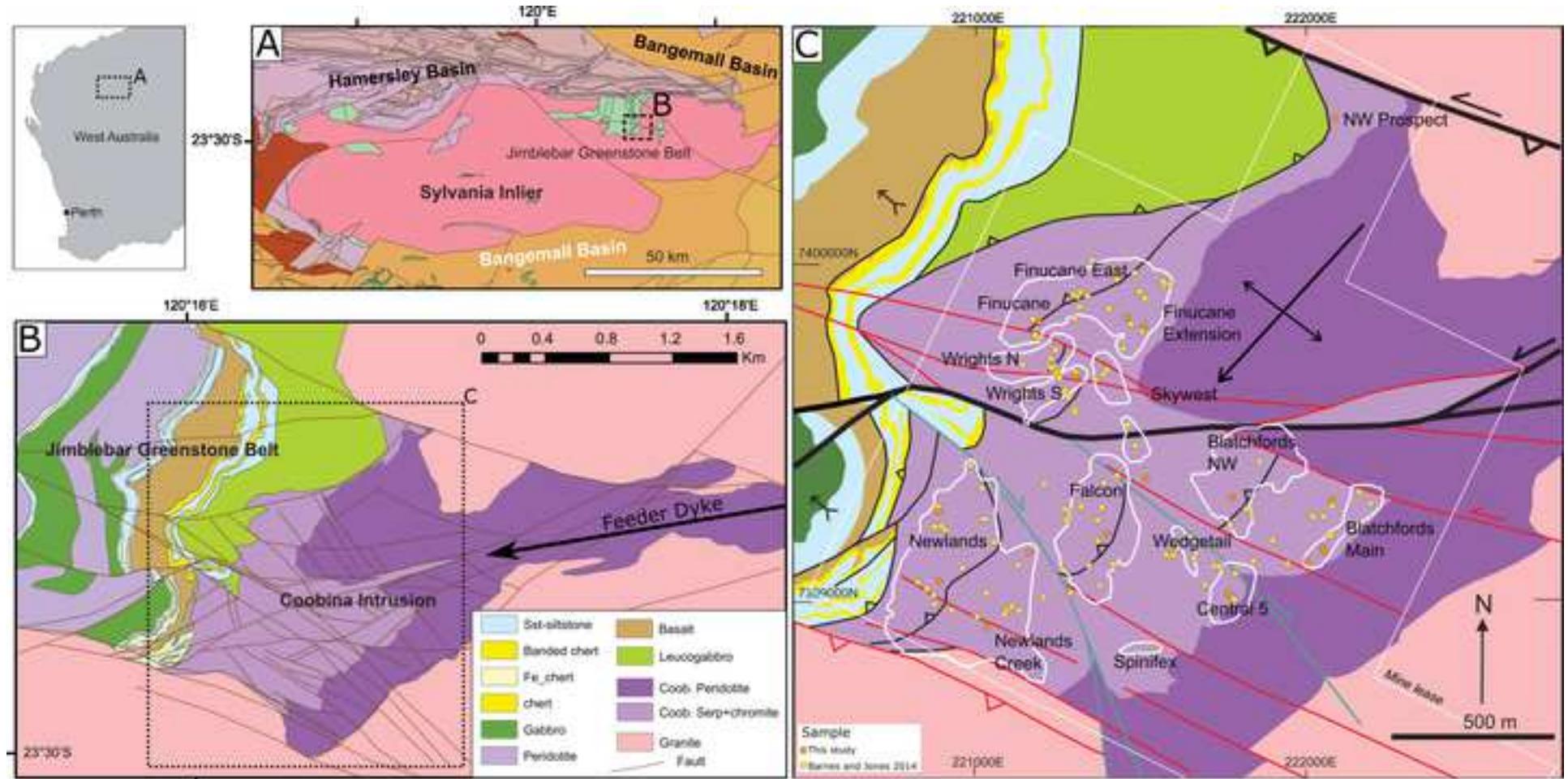
1021 WILLIAMS, I.R. & TYLER, I.M. (1991) Robertson, Western Australia.1:250,000 Geological Series,
1022 Explanatory Notes: Perth (1:250,000). Geological Survey of Western Australia.

1023 WILLIAMS, M.J., SCHONEVELD, L., MAO, Y., KLUMP, J., GOSSES, J., DALTON, H., BATH, A. &
1024 Barnes, S. (2020) pyrolite: Python for geochemistry. *Journal of Open Source Software* **5**,
1025 2314, <https://doi.org/10.21105/joss.02314>.

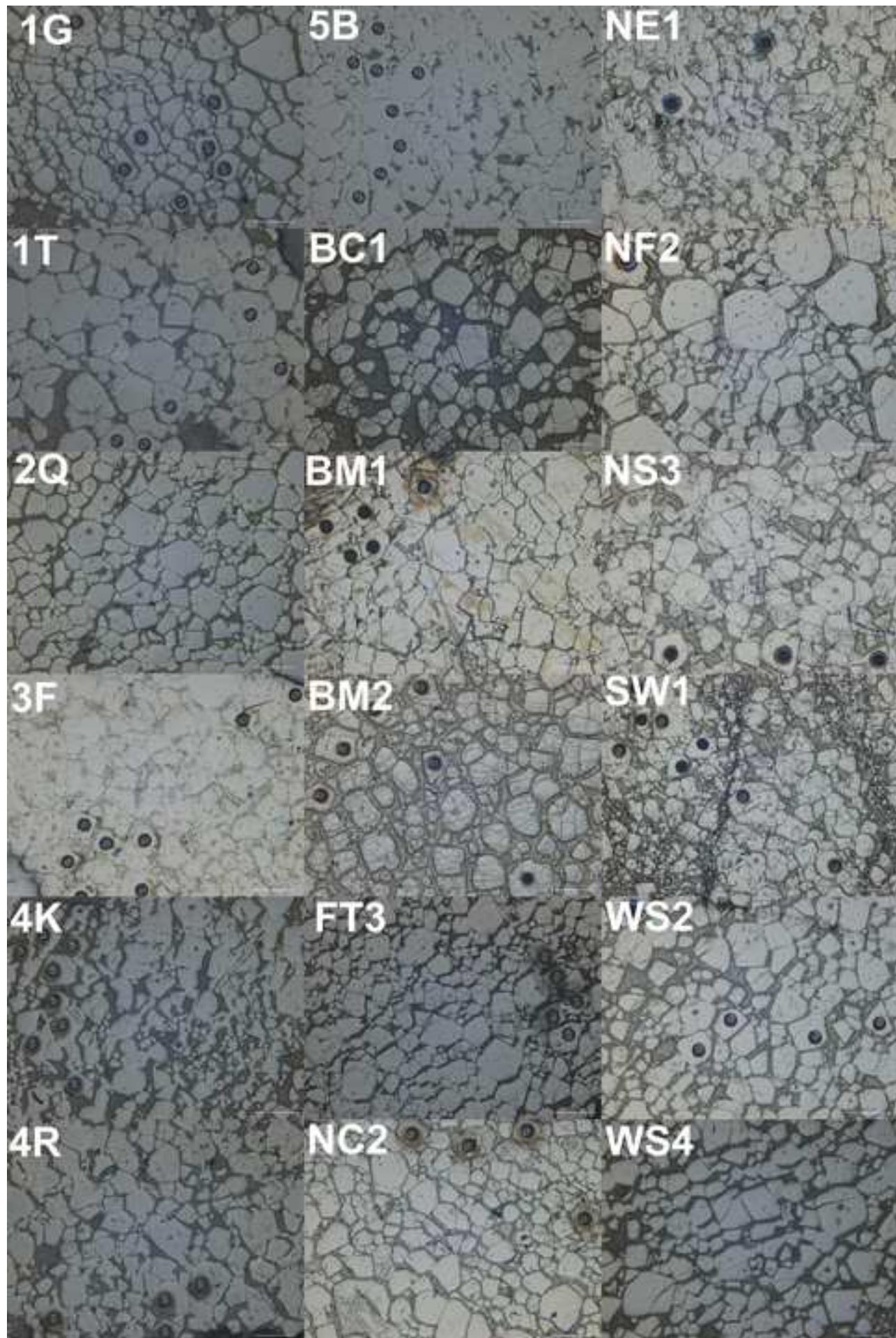
1026 WINGATE, M.T.D., LU, Y. & JOHNSON, S.P. (2019a) 216542: metamonzogranite, Garden Well;
1027 Geochronology Record 1560. *Geological Survey of Western Australia*.

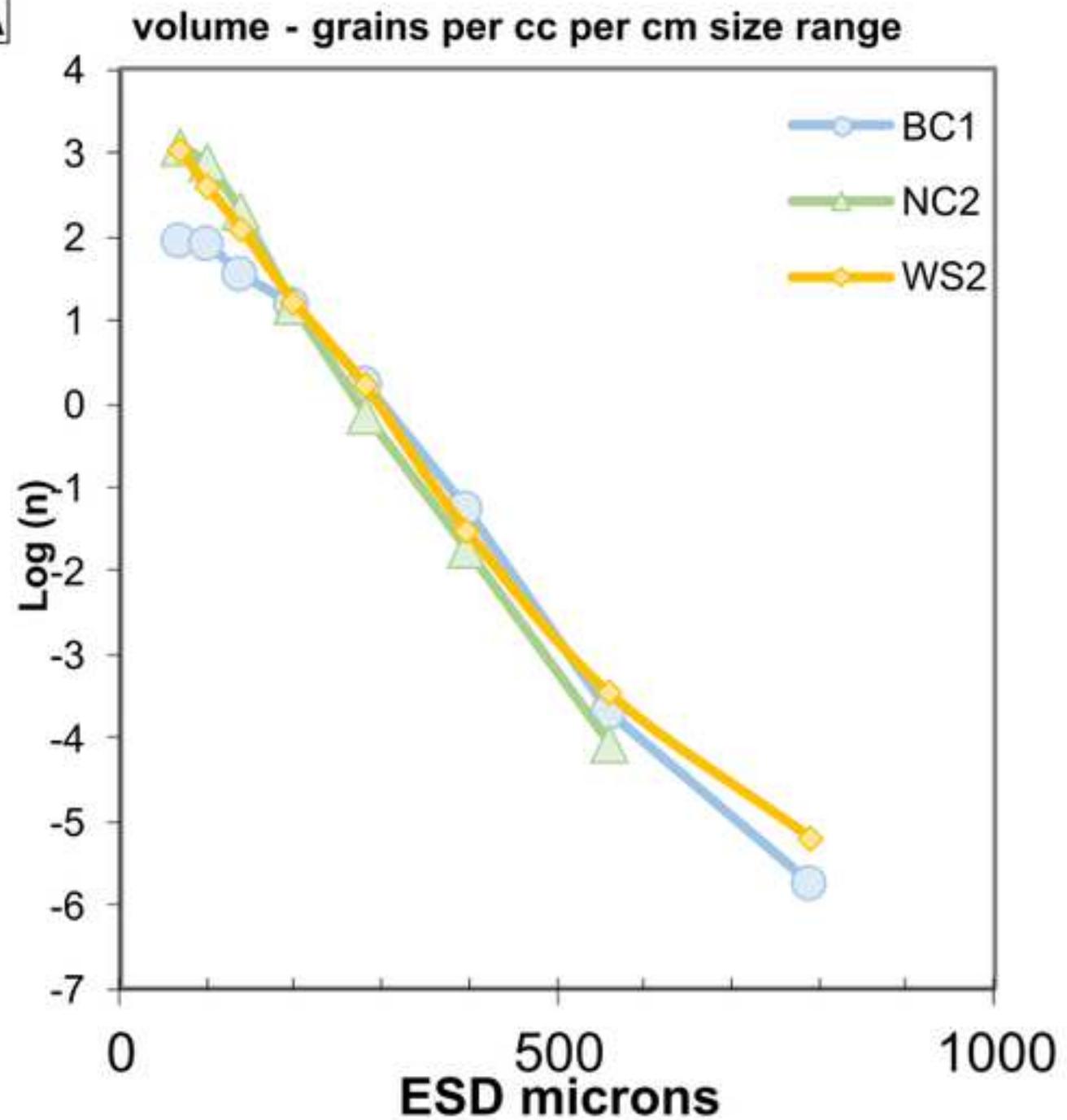
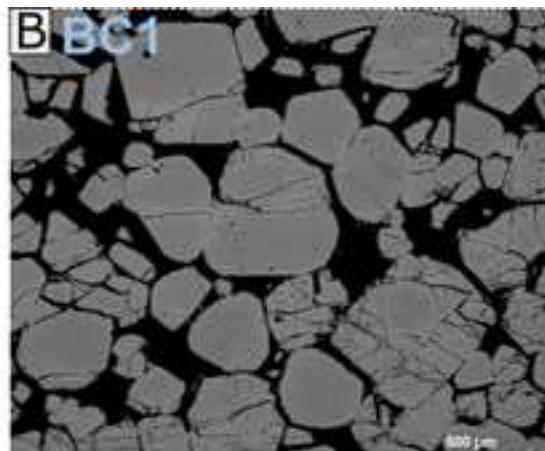
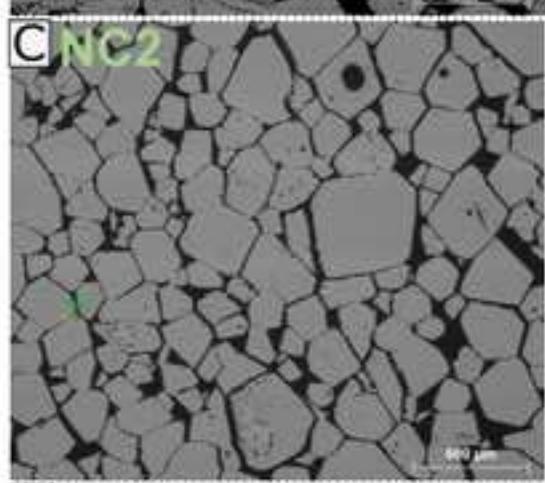
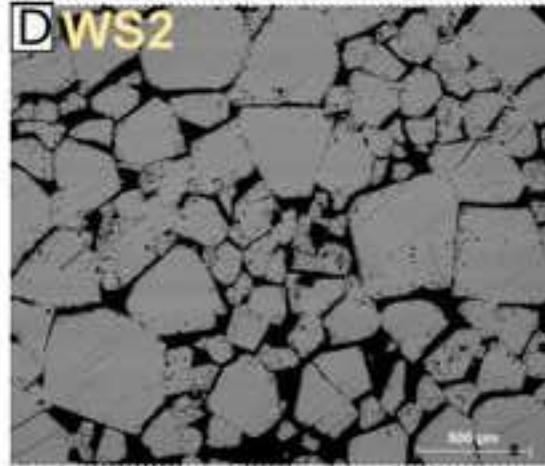
1028 WINGATE, M.T.D., LU, Y. & JOHNSON, S.P. (2019b) 216598: metarhyolite, Cubana Well;
1029 Geochronology Record 1568. *Geological Survey of Western Australia*.

1030







A**B BC1****C NC2****D WS2**

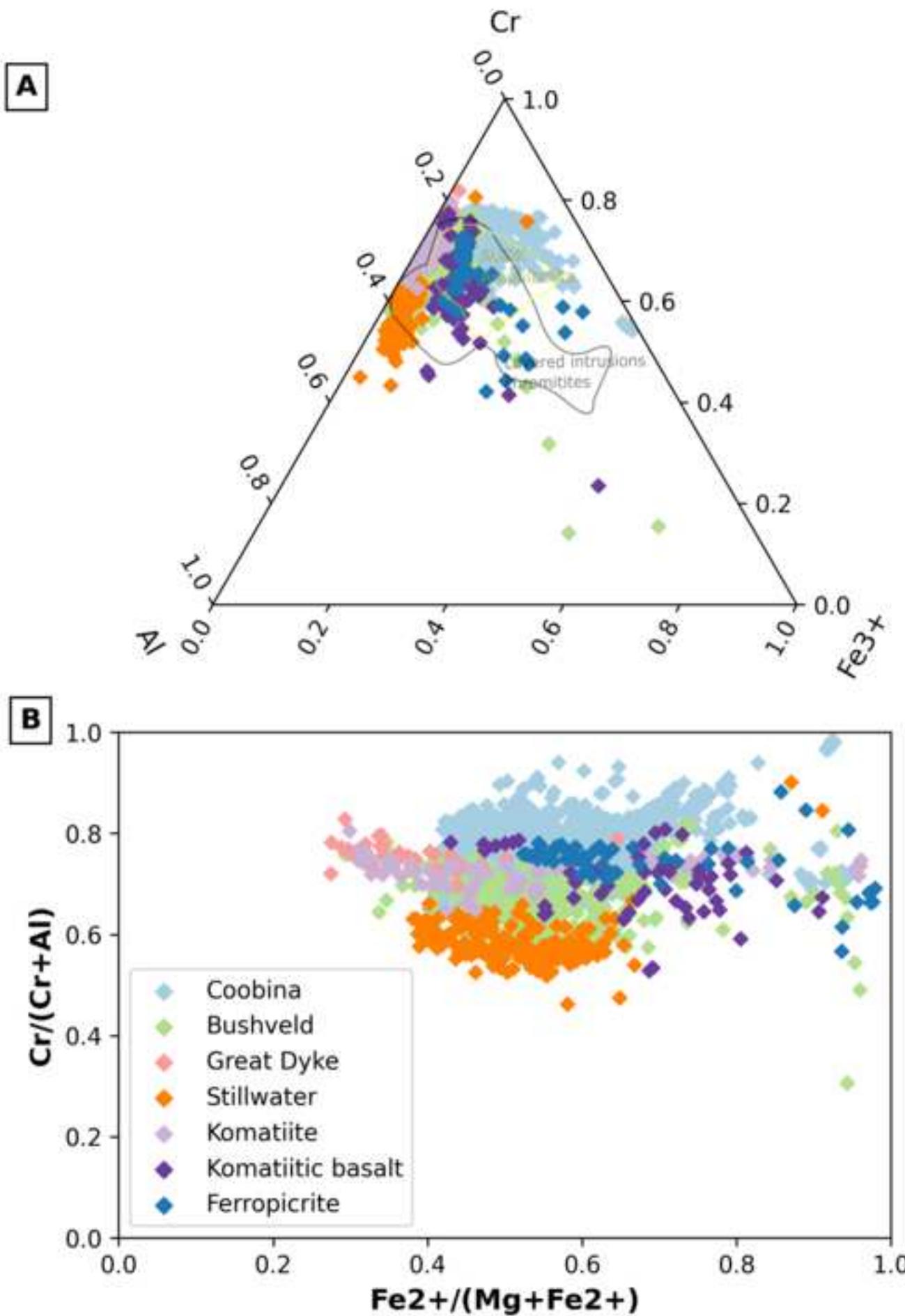
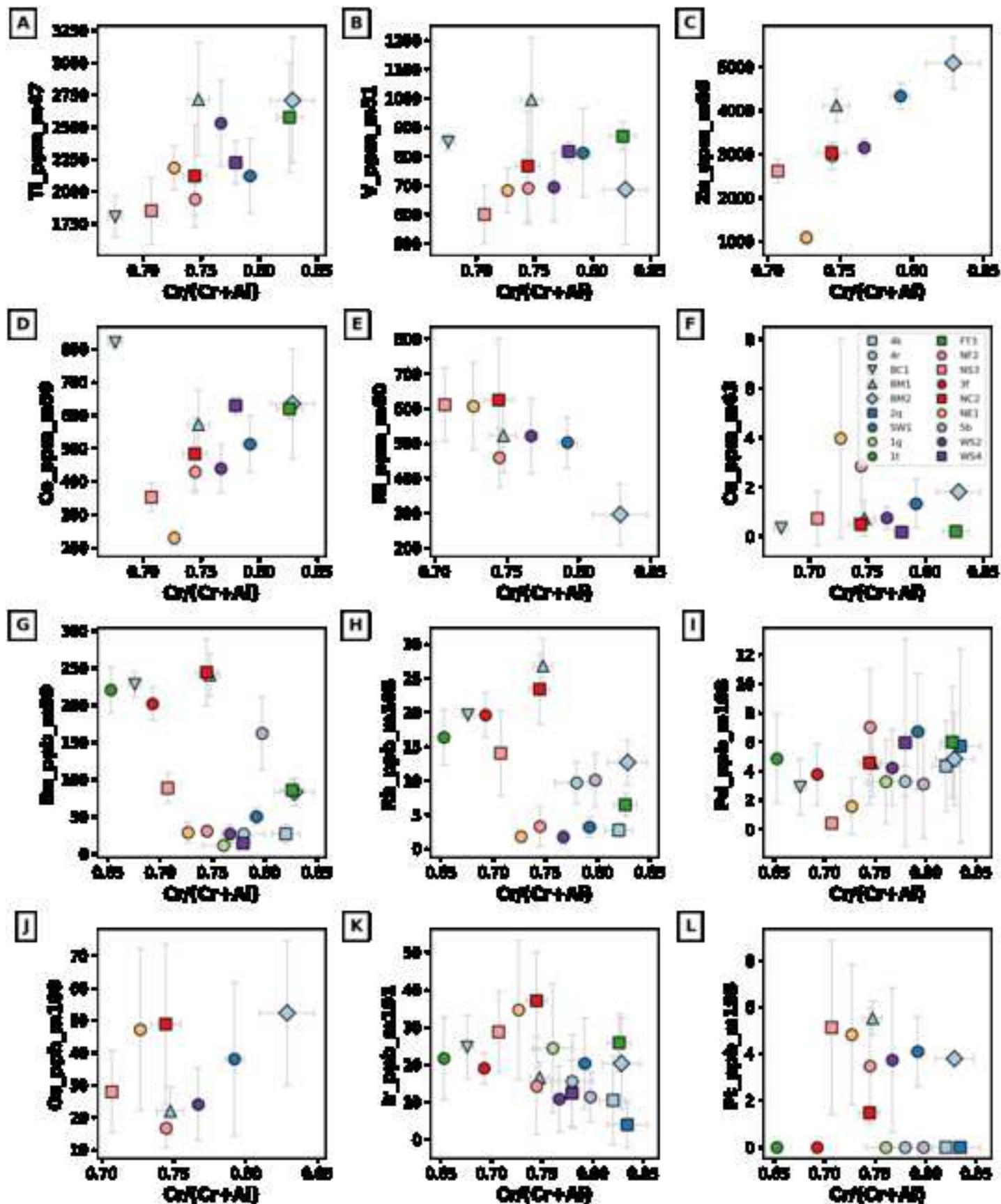


Figure6 TraceElements

[Click here to access/download;Figure;Figure6_chromite_PGE_all_Crnumber.ti](#)



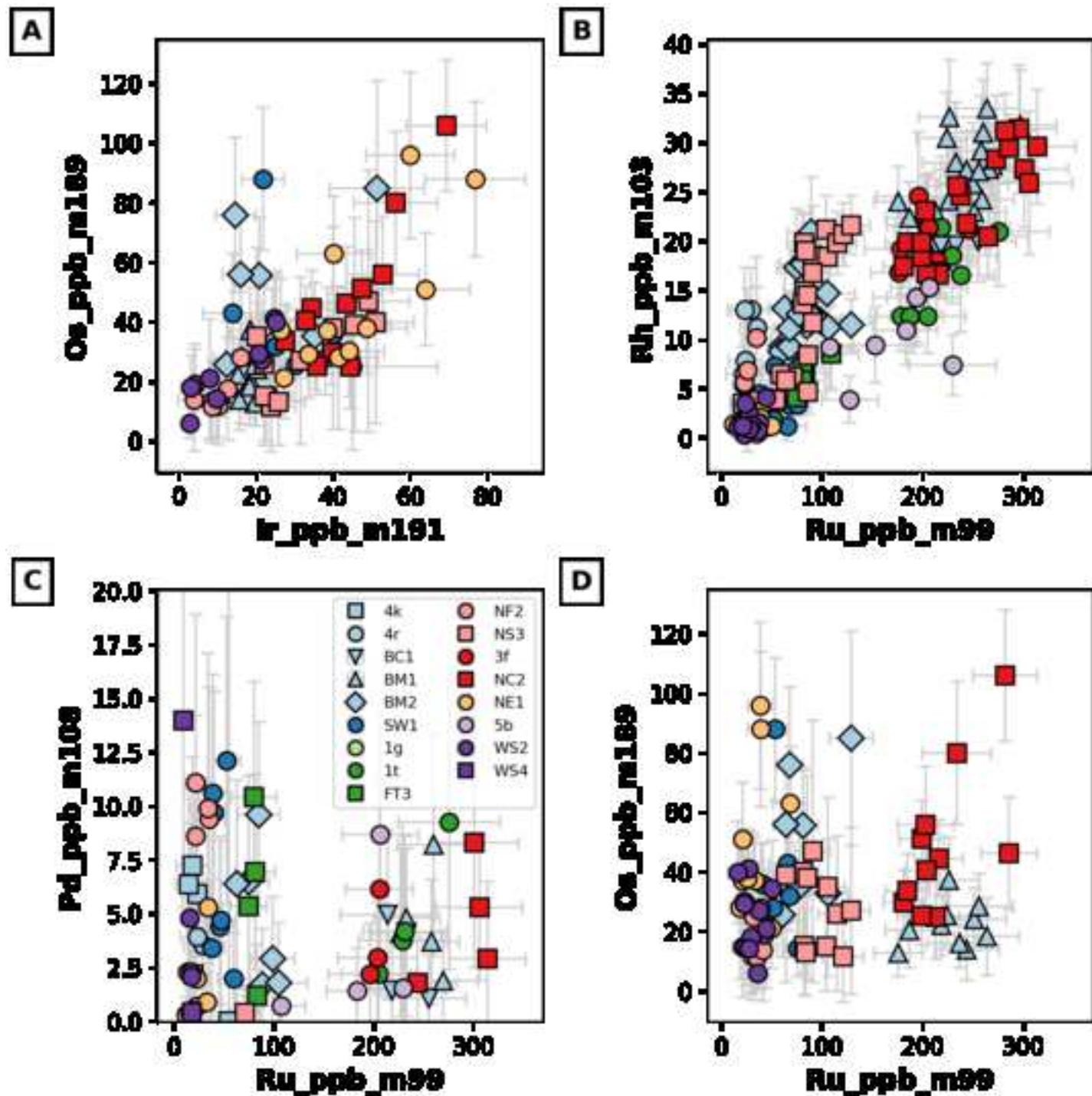
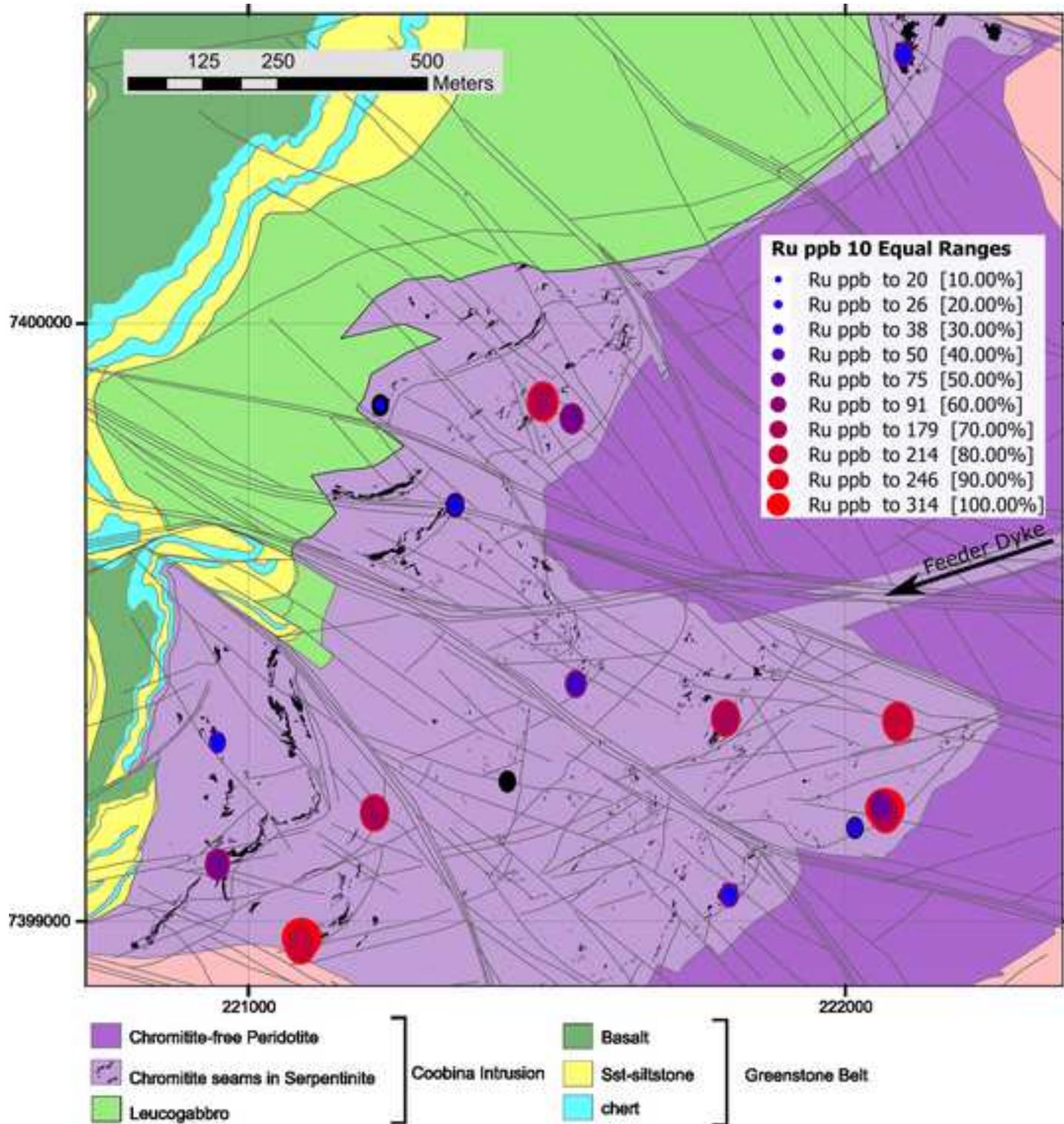


Figure8_MapWithRu

Click here to
access/download;Figure;Figure8_Coobina_map_Ru99ppb_2.tif



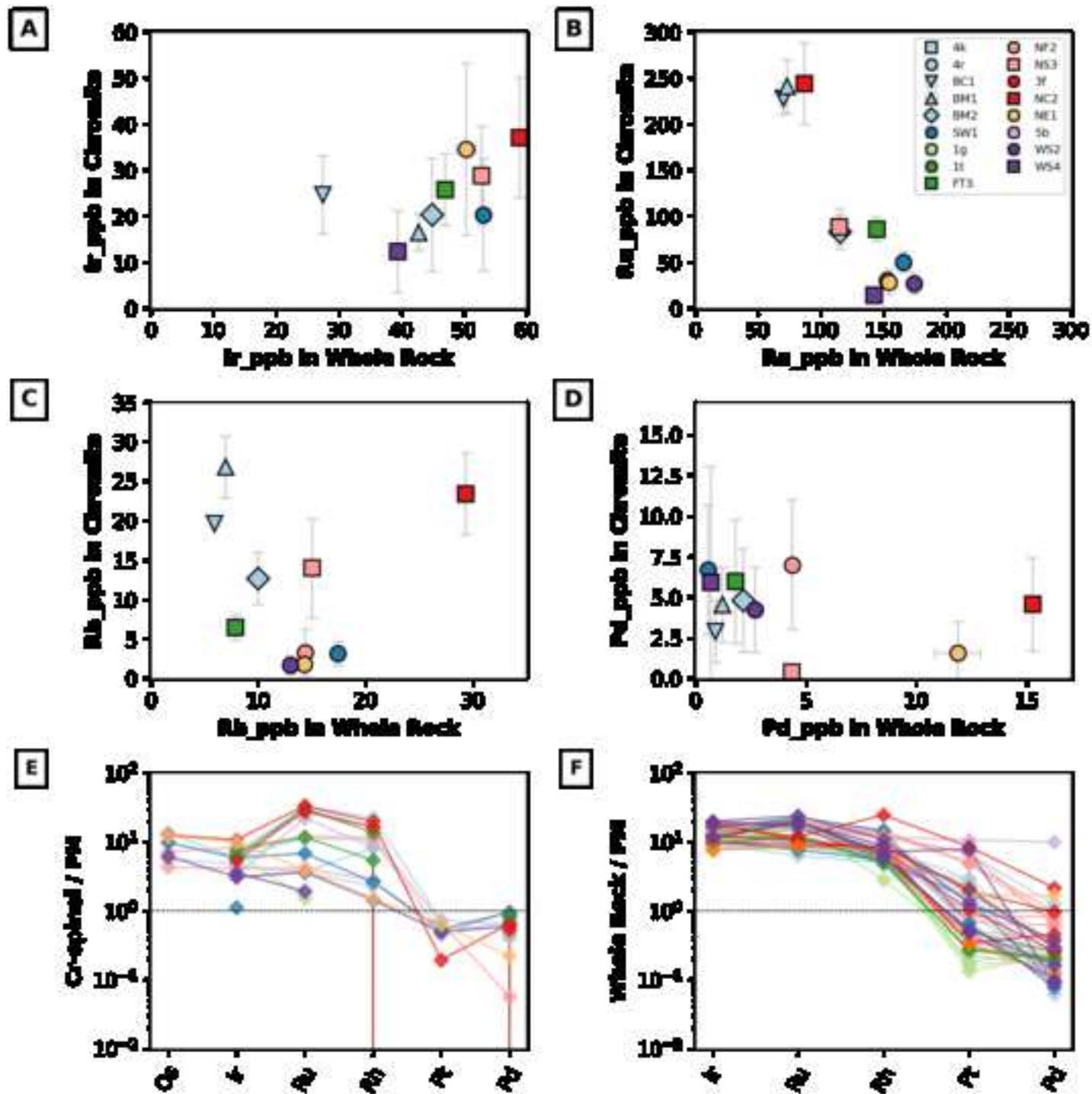


Figure 10.