

Negative Data in Data Sets for Machine Learning Training



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Data-driven chemistry has been described as the “future” of industrial organic synthesis that “will increasingly help guide synthetic chemists through the toughest synthesis problems”, in a recent editorial in *The Journal of Organic Chemistry*.¹ Data-enabled machine learning (ML) methods have been shown to be equal and sometimes superior to human, intuition-driven approaches in common tasks in organic synthesis such as reaction optimization.² This is also reflected by the continued growth of publications discussing ML in organic synthesis over the past eight years (Figure 1).³

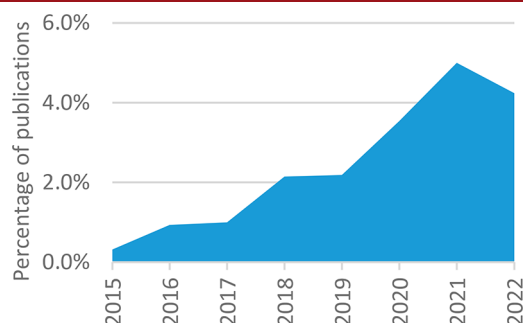


Figure 1. Trend of publications on machine learning and organic synthesis as portion of total publications, 2015–2022.

As in all ML applications, the performance of models for a given task relies heavily on the quality and scope of the training data that cover this task. Some ML models such as yield predictions need to distinguish between successful and unsuccessful reactions, so they require examples of high-, medium-, and low-yielding reactions in the training set. However, reactions with low or no yield (frequently referred to as “negative data”) are rarely included in the published literature. This represents a significant limitation of the literature, which is therefore often insufficient for the purpose of training ML models for these purposes.⁴ Literature-derived databases such as Reaxys or the data set extracted from the U.S. Patent and Trademark Office⁵ suffer from the same selection bias in that often only successful and high-yielding reactions are reported. A recent editorial by Kozlowski touches on the reasons that negative data remain unpublished in the context of the importance of meaningful substrate scopes.⁶ While the original data repositories such as experimental sections of Ph.D. theses or high-throughput experimentation (HTE) data sets do contain the full information on both high- and low-yielding reactions, they are usually not widely accessible or available in a computer-readable form. Electronic laboratory notebooks (ELNs) are an important step to address

this problem and are widely used in industry. Their current inconsistent adoption in academia is likely to improve due to increasing requirements of publishers and funders for FAIR Data usage. Efforts to create open-access databases, such as the Open Reaction Database (ORD),⁷ also try to address this problem by providing a data structure for reporting chemical reactions and by removing the distinction between “positive” and “negative” outcomes.

There has been much interest in minimum information standards in chemistry, including better research data management practices,⁸ and *The Journal of Organic Chemistry* and *Organic Letters* have been active participants in this discussion.⁹ In this Editorial, we argue that the efforts of including wider scope and yield ranges are necessary but not sufficient and need to be complemented by additional information that should be reported. There is much more information in low-yielding reactions than is commonly accepted, and simply stating that a reaction gave 0% yield is insufficient to learn from.

We consider the following scenarios:

- 1. No remaining starting material and no product.** This result implies that the reaction produces a different product than originally intended. The barrier height for the intended reaction need not be prohibitively high, but side reactions have lower barriers. It would of course be desirable if the actual products were characterized and reported, but even without that additional effort, this scenario still needs to be identified.
- 2. Most or all of the starting material remains.** In the case of low conversion, the reaction barrier is prohibitively high, or the reaction is thermodynamically unfavorable. In the case of catalytic reactions, this may also indicate that the catalyst has been deactivated. The latter effect is essentially impossible to distinguish in a single reaction, but it is straightforward to detect utilizing a competition reaction.
- 3. The reaction was not performed as intended.** This could be the result of a variety of factors, including a reaction that was performed using unintended conditions (e.g., contaminations in substrates, reagents, or solvents), an error in physical manipulation (e.g., the

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reaction flask was dropped or product was lost due to a spill), the reaction was performed as a proof of concept and the product never quantified, or changing priorities caused a planned reaction to be abandoned after entry into the ELN but before execution.

When building an ML model for reaction outcome, the implications of the three scenarios would be quite different. In the absence of additional information, a model would not be able to distinguish them. Furthermore, it should be recognized that there are several different ways to report yields (crude, isolated, conversion, etc.). With these possibilities in mind, we propose the following standards for reporting data (negative or not) in ELNs, electronic databases, or traditional laboratory notebooks (LNs):

1. **Isolated and crude reaction yields** should be clearly denoted and reported separately, because problems in the workup or isolation procedure can lead to low reported yields for a high-yielding reaction. For example, crude yields are often preferable for the training of ML models. However, yields determined by chromatography should also provide information on reaction conversion, which can easily be determined by quantifying the amount of starting material by the same method.
2. **Other measures such as conversion are frequently used as proxies of yield**, especially in HTE. It is preferable that both conversion and yield are reported, but if only conversion is available, this should be clearly denoted and not reported as a yield. In this context, analytic problems such as overlapping peaks that can bias the yield readout should be flagged.
3. **A mandatory conclusion before closing the experiment**, which could take the form of a drop-down menu in an ELN, with the following options:
 - A. Significant amount of product was detected (success)
 - B. No significant product was detected, but starting material remains
 - C. Neither starting material nor intended product was detected
 - D. The reaction was not run as intended (incorrect setup, physical error, reaction canceled, other). In this case, a free-text comment describing the observation would be beneficial, but not essential.

It is worth noting that such additional information would be among the first questions raised in discussing low-yielding reactions. The same should be expected when publishing a result as part of a substrate scope or control experiment; however, this is currently not included in commonly used data sources. The proposed standard would help to categorize (E)LN entries as well as further expand the information content of reported negative results by giving the necessary context of what did occur in the reaction. This information is of no extra cost or effort to the chemist, and these data would be invaluable to ML models that are being trained on (E)LN data.

The inclusion of these data would be beneficial in training ML models to predict reaction yields and conditions, which present a long-standing challenge in the application of predictive methods. While reactions that were not run as intended can simply be discarded, the other classifications can be informative for model building and could be used as additional features. We therefore encourage all experimental

chemists and authors to implement procedures that provide the crucial information on “what happened in a low-yielding reaction” in a way that is both easy and standardized according to FAIR Data principles increasingly required by reviewers, editors, and funding agencies.

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Notes

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Biographies

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