





# Rapid Thermal Annealing of p-Type Polysilicon Passivated Contacts Silicon Solar Cells

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**Abstract**—The carrier selective contacts Si solar cells based on tunnel oxide passivated contact (TOPCon) have provided an efficiency of over 25%, and the large-scale production is planned. One of the crucial steps in TOPCon solar cells fabrication is furnace annealing at around 875 °C. We have investigated the rapid thermal annealing (RTA) for TOPCon solar cells. We discovered that the RTA carried out in air showed much fewer hydrogen-induced blisters than in the N<sub>2</sub> atmosphere. Second, the optical radiation and heat generated in RTA had a significant effect on the degradation of  $iV_{oc}$ . The RTA with a SiN<sub>x</sub>:H layer on top of polysilicon generated  $iV_{oc}$  of 706 mV. The mechanism of degradation under air versus nitrogen atmosphere and the effect of high-intensity light during the annealing process were inferred through various characterizations, such as surface morphology, surface passivation, crystallinity determination, chemical compositional changes, electrical sheet resistances, and oxidation state of ultrathin SiO<sub>x</sub>. The TOPCon after air RTA performed better than in nitrogen, optimally at 825 °C. Forming gas annealing further improved the  $iV_{oc}$ . Crystallization and sheet resistance were dependent on the annealing temperature and time.

**Index Terms**—Annealing, blister, polysilicon, rapid thermal annealing (RTA), solar cells, tunnel oxide passivated contact (TOPCon).

## I. INTRODUCTION

THE architecture of high-efficiency c-Si-solar cells based on tunnel oxide passivated contact (TOPCon) generally consists of stack layers of heavily doped (B-doped or P-doped) polycrystalline silicon (poly-Si) layer and ultrathin oxide SiO<sub>2</sub> on the doped c-Si substrate. Although the term TOPCon was coined under the assumption of tunneling current as the charge transport mechanism in these selective contacts, several publications show that the dominating current transport mechanism is through the pinholes formed during the annealing process [1], [2]. There are many parameters and constraints in the fabrication

steps that influence the intrinsic open-circuit voltage ( $iV_{oc}$ ) and efficiency of a solar cell.

The recent research laboratory record-breaking efficiencies of p-TOPCon (based on p-type Si wafer) and n-TOPCon (based on n-type Si wafer) solar cells are 26.1% and 25.8%, respectively [3], [4]. The thermal annealing is a very important fabrication step required for achieving polycrystallization dopant activation and minimizing interfacial defects by improved passivation, thus enhancing the  $iV_{oc}$  and efficiency of a TOPCon solar cell.

Previously, the traditional tube-furnace annealing was used for thermal annealing of n- and p-TOPCon at high temperatures under inert atmospheres, such as N<sub>2</sub> [5], [6], [7], [8], [9], [10], [11]. The thermal processing with heating durations of tens of minutes is widely used in photovoltaic device fabrication and even in some of the industrial processes for POCl<sub>3</sub> diffusion. The firing process of the metallization, which includes heating, dwell, and cooling, is done in tens of seconds. Borden et al. [12] reported a firing process of 30 s for their poly-Si contacts. The blister formation is an important bottleneck in the fabrication of TOPCon architectures as under intense dehydrogenation pressure, the physical rupturing of ultrathin SiO<sub>x</sub> and poly-Si layers occurs. This results in the degradations of the passivation quality [6]. Ingenito et al. [13] described even faster processes while addressing the minimization of the blistering formation issue. Recent reports have shown that the rapid thermal annealing (RTA) has been used for annealing purposes in n- and p-TOPCon fabrication [14], [15], [16]. Rapid annealing is focused on time duration from a few seconds to several minutes. RTA has a fundamental advantage of a short period of annealing time as compared with tube-furnace annealing. Yang et al. [15] reported blister formations during RTA of n-TOPCon solar cells. Although earlier studies have been on suppressing blisters, there is a lack of a comprehensive study on blister formation under the RTA and the role of different atmospheres, such as air versus nitrogen, for p-TOPCon structures [13], [17].

In the RTA chamber, the a-Si:H layer in the TOPCon device faces direct exposure to the intense lamp light and heat for the whole period of rapid annealing. In such circumstances, the effect of intense light intensity under heat during RTA needs to be investigated.

Different dielectric capping layers of SiN<sub>x</sub>:H and Al<sub>2</sub>O<sub>3</sub> and thermal annealing steps have been known to be excellent hydrogenation sources for passivating TOPCon structures using tube furnaces [18], [19], [20], [21], [22]. Moreover, dielectric layers protect TOPCon structures from spiking and damage during

Manuscript received 21 November 2022; accepted 30 January 2023. Date of publication 14 February 2023; date of current version 20 April 2023. This work was supported by the National Science Foundation (NSF) under Grant ECCS-2005098 and Grant IUCRC-1338917. (Corresponding author: Mool C. Gupta.)

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Color versions of one or more figures in this article are available at <https://doi.org/10.1109/JPHOTOV.2023.3241790>.

Digital Object Identifier 10.1109/JPHOTOV.2023.3241790

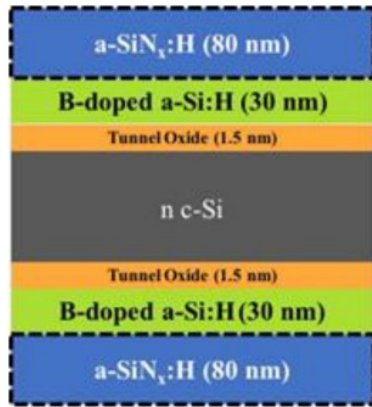


Fig. 1. Schematic diagram of a fabricated TOPCon test structure. The dotted  $a\text{-SiN}_x\text{:H}$  layers are considered when the test structure is TOPCon/ $\text{SiN}_x\text{:H}$ .

screen-printing metallization [23]. Therefore, it is necessary to investigate the utility of using the dielectric  $\text{SiN}_x\text{:H}$  layer.

In this article, we present the results of RTA under air versus  $\text{N}_2$  and the effect of high-intensity light under heat on *in situ* B-doped p-TOPCon solar cells. The mechanism of passivation degradation was investigated through various measurements, such as surface morphology, carrier lifetime, crystallinity, chemical compositional changes, electrical sheet resistances, and oxidation state of  $\text{SiO}_x$ .

## II. EXPERIMENT

### A. Fabrication of the p-TOPCon Test Structure

The p-TOPCon architecture used here is based on the type of poly-Si layer (*in situ* B-doped poly-Si/ $\text{SiO}_x$  stack) on n-type c-Si wafers. The *in situ* boron doping has several advantages over *ex situ* boron doping, such as simplicity, independent of dopant pre-deposition process, higher throughput, higher deposition rate of poly-Si layer, higher boron concentration, higher field-induced passivation, and lower optimal annealing temperature [24], [25], [26]. Fig. 1 shows the schematic diagram of the TOPCon test structure and with  $\text{SiN}_x\text{:H}$  layer (dotted). These test structures were fabricated on 200  $\mu\text{m}$  thick n-type monocrystalline Cz-type  $\langle 100 \rangle$  silicon wafers with a bulk resistivity of 3.2  $\Omega\cdot\text{cm}$ . Saw damage was removed in 9% wt. KOH at 80  $^\circ\text{C}$  for 12 min resulting in a semiplanarized surface. Wafers were cleaned in piranha solution (96% wt.  $\text{H}_2\text{SO}_4$ :  $\text{H}_2\text{O}_2$ : DI :: 1:1:2) and SC-2 (38% wt. HCl:  $\text{H}_2\text{O}_2$ : DI :: 1:1:2) along with oxide removal (5% wt. HF solution) between each step and at the end. The solution temperature was between 50 and 60  $^\circ\text{C}$ . Subsequently, an ultrathin ( $\sim 1.5$  nm) layer of  $\text{SiO}_x$  was grown on both sides of Si in 70% wt. electronics grade  $\text{HNO}_3$  at 100  $^\circ\text{C}$  for 15 min. *In situ* boron-doped polysilicon was then grown on both sides of the wafer in a Tystar tube low pressure chemical vapor deposition (LPCVD) system at 530  $^\circ\text{C}$  for 27 min using precursor gases of silane and diborane resulting in a thickness of approximately 30 nm. A selected sample was annealed in nitrogen at 875  $^\circ\text{C}$  for 30 min in a Centrotherm tube furnace to improve crystallinity and activate dopants. Additionally, some samples were coated with 80 nm of  $\text{SiN}_x\text{:H}$  layer on both sides in a Centrotherm plasma enhanced chemical vapor deposition (PECVD) reactor for improved passivation due to hydrogenation.

The samples were cut into a 1 cm  $\times$  1 cm area from a single 6-in test wafer before the start of any further processing. The quasi-steady-state photoconductance (QSSPC) measurements were done on 1  $\text{in}^2$  test samples. The QSSPC sensor coil diameter was 4 cm. Three samples were prepared for each thermal annealing test. The maximum film thickness variation over 1 cm length was found to be  $\pm 5\%$ . The boron concentration homogeneity was not measured.

Both the RTAs and forming gas anneals (FGAs) were done in the AnnealSys rapid thermal processing (RTP) system under different processing gas environments, such as ultrapure  $\text{N}_2$  and forming gas (5%  $\text{H}_2$ : 95% Ar) from Praxair. In the case of air annealing, the surrounding air was used with no pumping. The humidity recorded in the room was 40%–42%. The temperature regime for breaking Si-H bonds and dehydrogenation is from 300 to 550  $^\circ\text{C}$ , and a slower heating rate is beneficial to avoid blister formations [27]. All the RTAs were done using the three-step process, where the first step of annealing started with a 30-min ramp-up from room temperature to 350  $^\circ\text{C}$  and dwelling for further 30 min, followed by the second step of a 30-min ramp-up to 500  $^\circ\text{C}$  and dwell there for further 30 min. Subsequently, the final step was a 10-min ramp-up to different annealing temperatures, such as 625, 750, 825, and 875  $^\circ\text{C}$ , for a 5-min dwell time. The RTAs were done both in  $\text{N}_2$  and air atmospheres. The three-step method was adapted from Lee et al.'s article [17], but their study was limited only to the tube-furnace annealing. The reference samples were annealed in the tube furnace at 875  $^\circ\text{C}$  for 30 min in  $\text{N}_2$  with a ramp-up from 600  $^\circ\text{C}$  at the rate of  $\sim 10$   $^\circ\text{C}/\text{min}$ . The cooling rate in the RTA was  $\sim 300$   $^\circ\text{C}/\text{min}$  as compared with  $\sim 3$   $^\circ\text{C}/\text{min}$  in the tube furnace. This fast cooling rate emphasizes the rapidity of our RTA processing. The FGA process consisted of annealing treatment at 425  $^\circ\text{C}$  in forming gas for 1 h. To investigate the light-induced damaging effects due to the direct exposure from the tungsten-halogen lamp in the RTA system, a 500  $\mu\text{m}$  thick p-type c-Si wafer cover of 1.05"  $\times$  1.05" area was placed over the test samples during all the annealing, irrespective of the anneal atmospheres. It was ensured that there was no actual physical contact between the Si wafer cover and the test sample by using small Si spacers. Test samples were annealed with direct exposure to the light using the three-step process of RTA at 875  $^\circ\text{C}$  both in  $\text{N}_2$  and air in order to investigate the effect of light-induced degradation. The sample with  $\text{SiN}_x\text{:H}$  layer was directly annealed in RTA at 825  $^\circ\text{C}$  in  $\text{N}_2$ , followed by FGA.

### B. Characterizations

Except for the QSSPC  $iV_{oc}$  measurements, all the characterization studies were accomplished using 1  $\text{cm}^2$  area test samples. The SEM surface morphology was characterized using a 15 $^\circ$  stage tilt in FEI Quanta 650 Field Emission Scanning Electron Microscope. The QSSPC  $iV_{oc}$  measurements were conducted on (1"  $\times$  1") area test samples using WCT-120 Silicon Wafer Lifetime Tester made by Sinton Instruments. The  $iV_{oc}$  data were generated at 1 Sun condition. The photoluminescence (PL) setup for passivation characterization consisted of a 532 nm wavelength and 2 W continuous-wave laser (Melles Griot) as an excitation source with a spot size of  $\sim 1$  mm, a spectrometer Horiba Jobin Yvon iHR320 operated by SynerJY v3.5

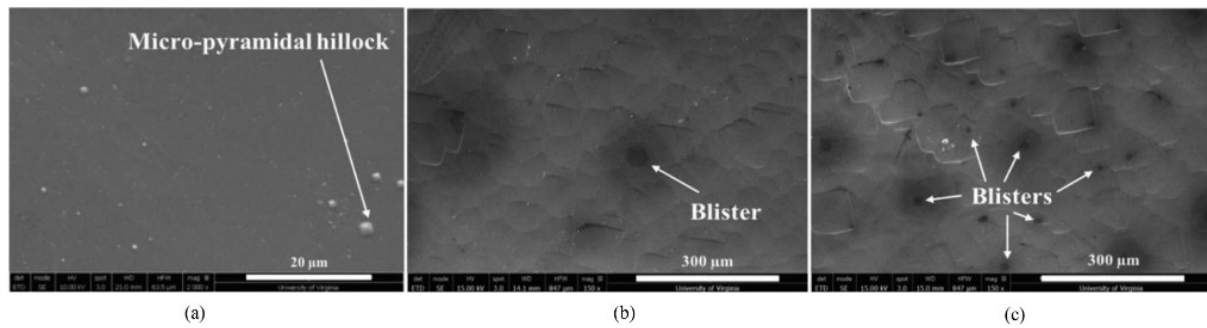


Fig. 2. Top-view SEM images of (a) as-deposited sample, (b) blisters (black areas) after RTA at 875 °C in air, and (c) blisters (black areas) after RTA at 875 °C in  $N_2$ .

software, an InGaAs detector, and an SR830 DSP lock-in amplifier (Stanford Research Systems). The excitation laser power was reduced to 70 mW by neutral density filters and quartz glass reflection so that it does not saturate the PL detector, does not heat the test samples significantly, and provides a high S/N ratio. Each PL spectrum reading consisted of an average of ten scans. The absolute PL intensity values are arbitrary but can be used for a relative comparison of the passivation quality. The PL intensity has a peak value at  $\sim 1150$  nm due to the c-Si bandgap of  $\sim 1.1$  eV [28]. The  $\mu$ -PL measurement (with a spot size of  $\sim 150$   $\mu$ m) was also done to characterize and differentiate the passivation quality on the blisters and outside-the-blisters region. The PL data had an approximate intensity error of  $\sim 2.6\%$ . The Raman spectroscopy for crystallinity measurements was done on Renishaw InVia TM Confocal Raman Microscope with a 405 nm excitation wavelength laser and spot size of 1.8  $\mu$ m. All Raman scattering data had been taken as an average of ten scans with each of 15 s of acquisition time. To calculate the Raman peak “area under the curve,” a reference sample was used every time before making measurements on test samples to account for the changes due to optics, mounting effects, crystallinity, etc. The sheet resistances were measured by Jandel four-point probe station for electrical characterization. The Fourier transform infrared (FTIR) spectroscopy was done in Thermo Scientific Nicolet iS50 FT-IR for chemical compositional characterization. Each FTIR reading was taken as an average of 256 scans, and the background readings were taken as an average of 200 scans. All these characterizations had an error of  $\sim 5\%$ . The X-ray photoelectron spectroscopy (XPS) characterization was done for quantitative surface chemical and elemental composition, depth profiling, and oxidation state determination on PHI Versaprobe III scanning X-ray photoelectron spectrometer with a spot size of  $\sim 200$   $\mu$ m and Al monochromatic X-ray source. The pass energy was 280 eV with a 0.5 eV acquisition step. All the XPS spectra for TOPCon structures had C (1s) peak calibrated to 284.8 eV according to ASTM/NIST [29]. The XPS spectra for the TOPCon/ $SiN_x$ :H samples had N(1s) peak calibrated to 397.86 eV [30]. The error bar of XPS measurements for oxidation states of  $SiO_x$  was approximately 0.2 eV.

### III. RESULTS AND DISCUSSION

We present the results of the RTA of p-TOPCon with and without  $SiN_x$ :H layer under  $N_2$  versus air and the effects of high-intensity light environment.

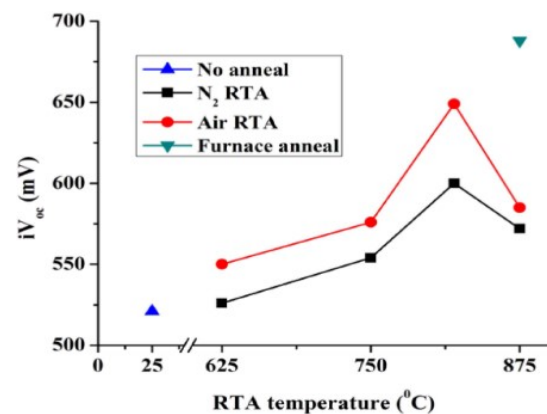


Fig. 3. Graph showing the dependence of  $iV_{oc}$  on the RTA temperatures. The data are given for RTA both in  $N_2$  and air atmospheres. The graphs also show the  $iV_{oc}$  value for the sample with no annealing and after furnace annealing.

#### A. Study of the Effects of RTA Atmospheres

1) *Surface Morphology*: Fig. 2(a) shows the top-view image of as-deposited samples. No blisters were observed except for the octahedral pyramidal features on the surface. The pyramidal structures were caused due to the KOH etching during the semiplanarization process, and they were unaffected by annealing temperatures and time conditions. Schröder et al. [31] have reported the observation and possible mechanism of such pyramidal features during the KOH etching of Si. Fig. 2(b) shows the top-view SEM image of blisters on the air-annealed samples at 875 °C. A higher number density of such similar blisters was observed in the  $N_2$ -annealed test samples at 875 °C, as shown in Fig. 2(c). The blisters covered  $\sim 1\%$  of the total surface area of air-annealed samples, whereas the blisters covered  $\sim 3\%$ – $4\%$  of the total surface area of  $N_2$ -annealed samples. The morphology of samples annealed in the tube furnace is not shown here as no change was noted.

The high-temperature annealing caused the dissociation of Si-H bonds, which resulted in the release of atomic hydrogen in the poly-Si/ $SiO_x$ /c-Si stack. These atomic hydrogens combined to form molecular  $H_2$  under its own immense pressure and consequently escaped while rupturing through the polysilicon layer in the form of blisters [32]. The three-step process of lower temperature annealing minimized the formation of blisters. The furnace-annealed samples showed fewer blisters possibly due to slower heating rates compared with the RTA.



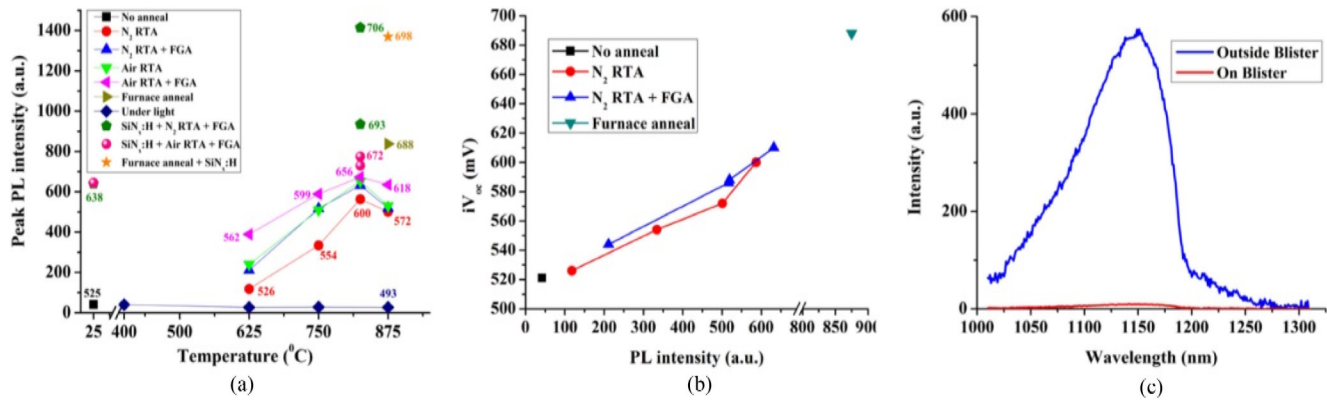


Fig. 4. (a) Qualitative dependence of peak PL intensity on the RTA temperatures for various annealing conditions. The  $iV_{oc}$  (in mV) numbers are provided on the graph and (b) dependence of  $iV_{oc}$  on the peak PL intensity. The data are given for RTA in N<sub>2</sub> and subsequent FGA only, and (c) PL intensity on and outside blister areas.

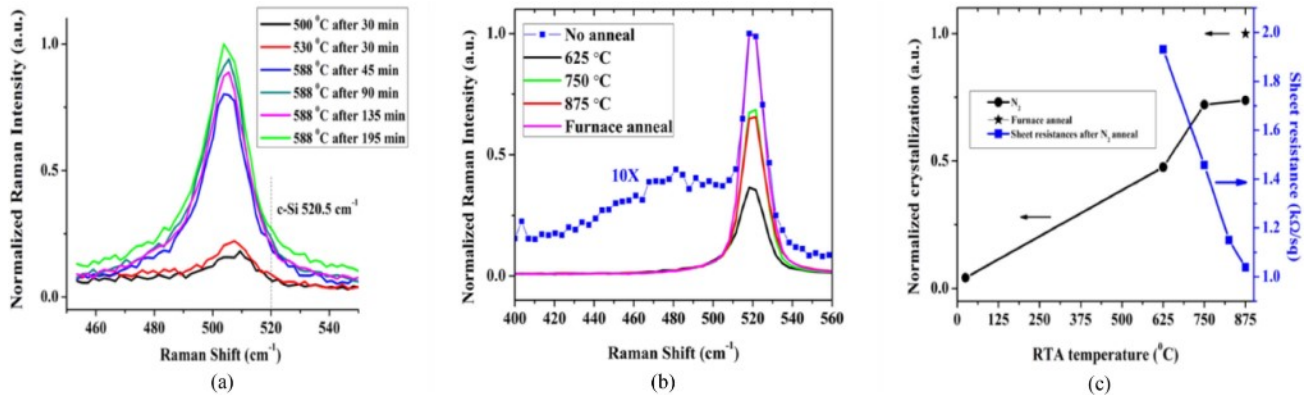


Fig. 5. (a) Raman spectra for the dependence of crystallization of a-Si:H on the anneal time duration while the sample was kept at hot stage at various temperatures. (b) Room temperature Raman spectra for the dependence of crystallization on the RTA temperature, anneal time was 5 min and curves are normalized to furnace anneal. The as-deposited sample curve is magnified ten times due to lower Raman intensity and (c) growth of crystallinity (area under curve) as a function of temperature.

2) *Surface Passivation Quality*: Fig. 3 shows the variation of  $iV_{oc}$  with the RTA temperatures under N<sub>2</sub> and air. The optimal regime of annealing temperature was found to be near 825 °C, and the  $iV_{oc}$  started decreasing at temperatures beyond 825 °C. This could be due to the possible degradation to the SiO<sub>2</sub> passivation layer. The  $iV_{oc}$  peak value was higher in the air (649 mV) than in N<sub>2</sub> (600 mV). Since the *in situ* B-doped poly-Si layers of these test samples were fabricated by the LPCVD process at a lower temperature of 530 °C, there was a significant starting amount of hydrogen, which increased the chances of blister formations. The RTA sample did not reach the tube-furnace-annealed sample  $iV_{oc}$  value of 688 mV, possibly due to the higher density of blisters leading to the passivation degradation.

3) *Photoluminescence*: Fig. 4(a) shows the variation of peak PL intensity w.r.t. RTA temperatures in the N<sub>2</sub> and air atmosphere and FGAs. It was observed that the annealing temperature of 825 °C provided the highest PL intensity, and FGA improved it further. In N<sub>2</sub>-annealed test sample, the highest PL intensity was 563 a.u. and FGA pushed it to 632 a.u. Similarly, in the case of air anneal, the highest PL intensity was 652 a.u. and FGA helped to reach 673 a.u. The furnace-annealed sample had a higher

peak PL intensity. This showed that the passivation quality of air-annealed samples was better than that of N<sub>2</sub>. Fig. 4(b) shows the observed relationship between  $iV_{oc}$  and peak PL intensity. Note when the junction is present, a logarithmic relationship is expected [33]. However, in this case, the PL change occurs due to the oxidation, crystallinity increase, and decrease in defect density. Therefore, it shows a linear relationship.

Fig. 4(c) shows the  $\mu$ -PL intensity between the blister region and outside the blister region. It was found that the peak PL intensity at the blister region was 9 a.u. as compared with ~570 a.u. outside the blister region of an air-annealed test sample at 825 °C. This showed that the blisters caused extensive passivation degradation in the test samples.

4) *Crystallinity Determination*: Fig. 5(a) shows the Raman peak intensity of the polycrystalline phase w.r.t. anneal temperature and time duration in the N<sub>2</sub> atmosphere. Choi et al. [6] reported that the deposition temperature of LPCVD poly-Si should be above 580 °C to ensure lower hydrogen content that prevents blister formation. An increase in crystallinity was observed starting at 588 °C, shown by the Raman peak at 508 cm<sup>-1</sup>, and it increased further as the annealing time was increased. These measurements were carried out in a heated chamber filled

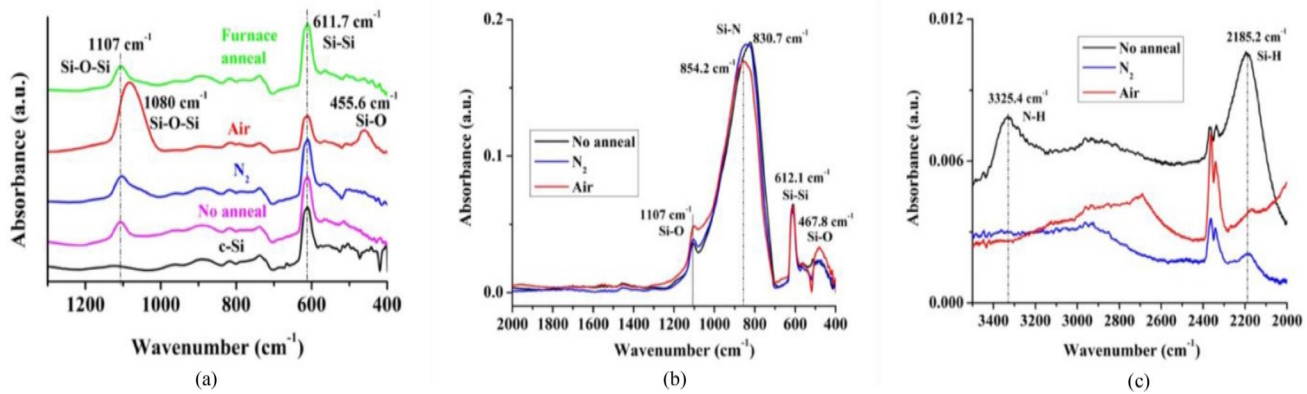


Fig. 6. (a) FTIR spectra for c-Si, TOPCon samples annealed at 825 °C in N<sub>2</sub> and air and results are compared with furnace-annealed sample. (b) FTIR spectra for TOPCon/SiN<sub>x</sub>:H samples as deposited and annealed at 825 °C in N<sub>2</sub> and air. (c) FTIR spectra for TOPCon/SiN<sub>x</sub>:H samples from 3400 to 2000 cm<sup>-1</sup> wavenumber region. All the samples were FGA treated before FTIR.

with N<sub>2</sub>. The standard room temperature Raman peak of c-Si at 520.5 cm<sup>-1</sup> is shifted to 508 cm<sup>-1</sup> due to the measurements done at the high temperature [34]. The as-deposited *in situ* B-doped poly-Si film was a mixture of amorphous and polycrystalline phases. Fig. 5(b) shows the change of a-Si:H peak Raman peak at 480 cm<sup>-1</sup> toward the crystalline phase at 519.1 cm<sup>-1</sup> as the RTA temperature increased from 625 to 875 °C in the N<sub>2</sub> atmosphere. Similar phenomena were observed when the test samples were subjected to RTA under an air atmosphere (not shown here). Fig. 5(c) shows the increase of the “area under the curve” of the Raman curve w.r.t. RTA temperatures in the N<sub>2</sub> atmosphere, indicating the increase in crystallinity. The degree of crystallinity in the *in situ* B-doped TOPCon depends on RTA temperatures and annealing time, not on anneal atmospheres or postanneal FGAs. At room temperature, the Raman peaks for poly-Si and c-Si substrate both occur at 520.5 cm<sup>-1</sup>. The Raman peak of poly-Si is much broader than that of c-Si due its polycrystalline nature. At the Raman excitation wavelength of 405 nm, the transmission through 30 nm poly-Si is only ~12% based on the optical *n* and *k* constants, and therefore, the contribution to the Raman peak from the c-Si substrate was very small.

5) *Chemical Composition*: Fig. 6(a) shows the FTIR absorbance spectra of c-Si, as-deposited film, TOPCon structure annealed at 825 °C in N<sub>2</sub> and air, and furnace annealed in N<sub>2</sub>.

In Fig. 6(a), the absorbance peak at 1107 cm<sup>-1</sup> generally corresponds to the presence of the vibrational stretching mode of Si-O, indicating the presence of interstitial oxygen and native oxide [35], [36]. The shift of this peak toward the 1080 cm<sup>-1</sup> peak corresponds to the stretching mode of Si-O-Si and indicates the formation of typical thermal SiO<sub>2</sub> on the surface after air annealing [37], [38]. The peak at 455.6 cm<sup>-1</sup> for the air-annealed sample corresponds to the rocking-mode vibration of Si-O bonds, which also indicates the formation of SiO<sub>2</sub> [39]. The vibrational state of the Si-Si bond was observed at 611.7 cm<sup>-1</sup> [40].

The different peak positions indicated that the air annealing converted the top part of the poly-Si layer into the SiO<sub>2</sub> layer. Under the N<sub>2</sub> RTA and furnace annealing, there was no significant SiO<sub>2</sub> formation except native oxide growth.

6) *Sheet Resistance*: The conversion of the a-Si:H layer to poly-Si will enhance lateral transport of carriers, dopant

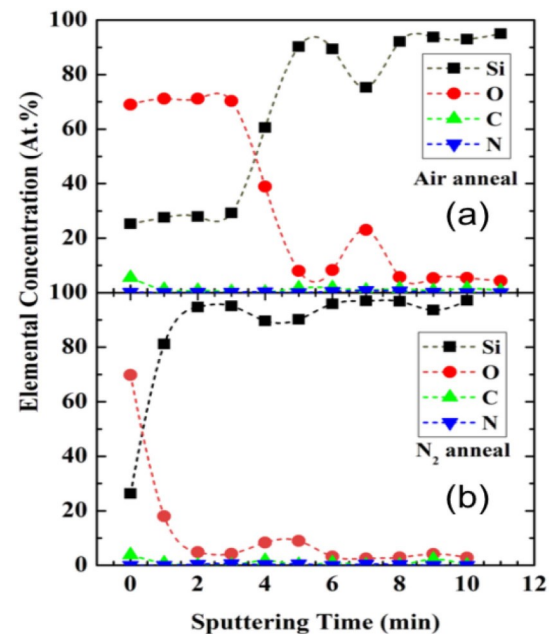


Fig. 7. Depth-profiling XPS spectra showing the atomic percentage profile of (a) air-annealed and (b) N<sub>2</sub>-annealed samples based on the atomic concentrations of Si (2p), O (1s), C (1s), and N (1s).

activation, and passivation. Fig. 5(c) shows the sheet resistances measured at different annealing temperatures in N<sub>2</sub> along with a degree of crystallinity. It was possible to achieve sheet resistances as low as furnace anneal test samples [0.9 kΩ/sq, not shown in Fig. 5(c)]. The sheet resistance of the nonannealed sample was > 1 MΩ/sq. The RTA atmospheres, such as the air, had a negligible effect on the sheet resistance of *in situ* B-doped poly-Si, measured at a constant annealing temperature of 875 °C.

7) *Oxidation State and Elemental Composition*: The depth-profiling XPS composition spectra of test samples after RTA both in air and N<sub>2</sub> at 825 °C are shown in Fig. 7(a) and (b), respectively. The sputtering rate of SiO<sub>2</sub> is expected to be ~8.9 nm/min, and the Si/SiO<sub>2</sub> sputter rate ratio was ~1.1 [41]. Hence, the exact locations of the surface oxide, poly-Si, and buried ultrathin oxide layers are prone to some error due to the

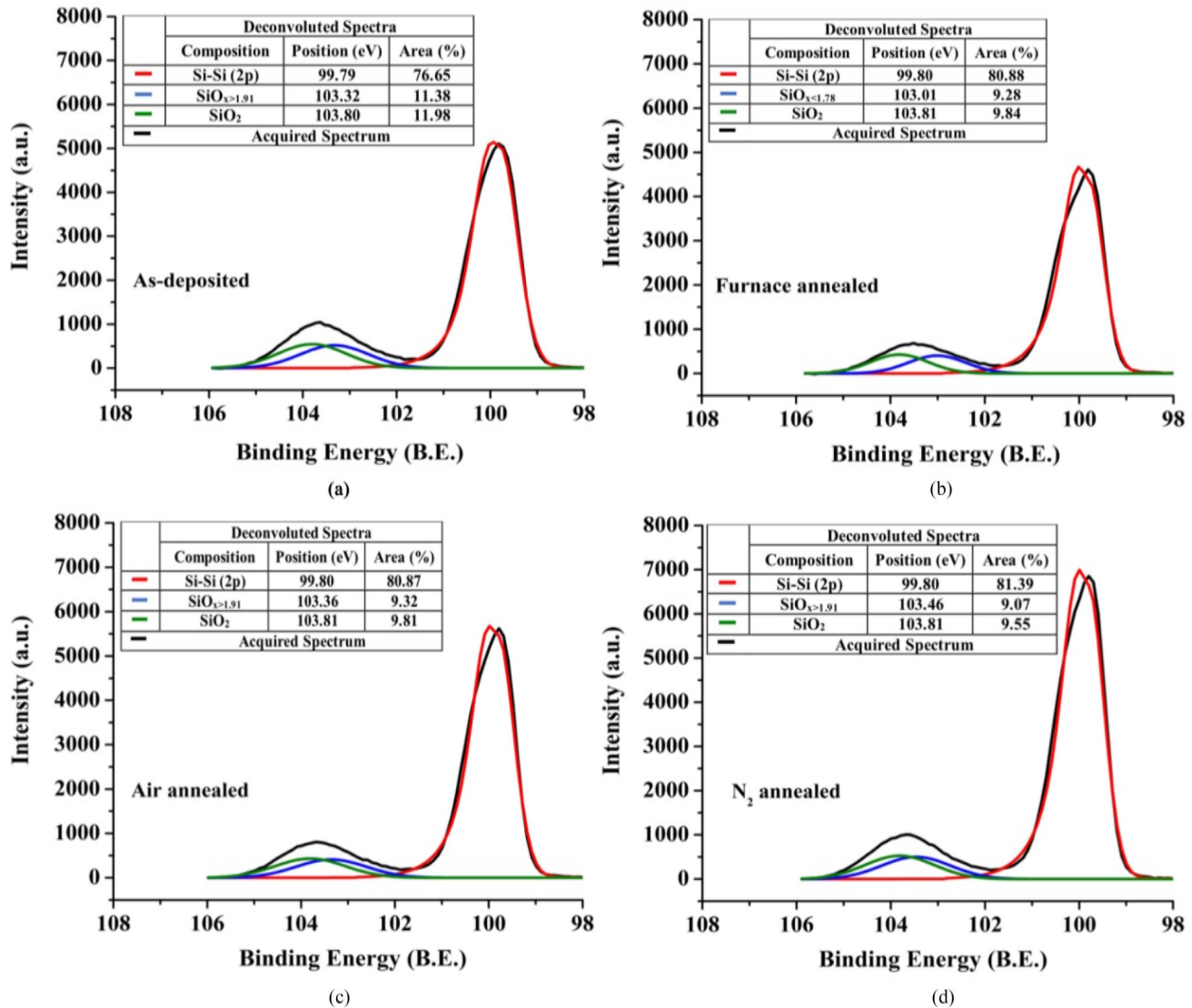


Fig. 8. Surface XPS spectra of ultrathin SiO<sub>2</sub> layer of TOPCon test structures at (a) as-deposited, (b) furnace, (c) air, and (d) N<sub>2</sub> annealing conditions.

undetermined sputter rate through the poly-Si layer. In the case of air annealing, a thicker oxide layer was observed on the top of the remaining poly-Si layer. This proved that the oxygen from air diffused through the surface oxide layer on poly-Si. An increase in oxygen concentration and a decrease in Si concentration under the poly-Si layer is due to the presence of an ultrathin oxide layer. Past the ultrathin oxide/c-Si interface, the Si/O ratio concentration increased. In the case of N<sub>2</sub> annealing, a very thin surface oxide was observed along with the broader peak of Si and oxygen concentrations at the ultrathin oxide region. This shows higher concentration of oxygen present in the ultrathin oxide layer in air-annealed samples. The boron was not able to be detected due to its relatively low concentration. The as-deposited sample had a negligibly thin surface native oxide layer with an intact poly-Si/ ultrathin oxide/c-Si structure (not shown here). This also reiterated our findings from the FTIR study. Fig. 8 shows the XPS spectra of the ultrathin passivating SiO<sub>2</sub> layer after the removal of polysilicon by the chemical etching process for different annealing conditions at 825 °C. XPS spectra show

a peak at 99.8 eV due to the Si-Si bond, and another peak at 103.8 eV is due to the presence of the SiO<sub>x</sub> layer. The Si-O peak was deconvoluted and showed the two major oxidation states of Si<sup>3+</sup> and Si<sup>4+</sup> [42]. The negligible differences in SiO<sub>x</sub>/SiO<sub>2</sub> ratios and their respective peak positions suggested that the RTA atmosphere (air or N<sub>2</sub>) did not significantly affect the chemical composition of the ultrathin silicon oxide. Only for furnace anneal, there was a visible shift in the Si-O peak. In addition, the area of silicon peak was higher after thermal annealing, indicating the increase in elemental Si [43]. The decrease in SiO<sub>2</sub> concentration was noted, possibly which can come from the incorporation of Si into the oxide layer [44].

### B. Effects of High-Intensity Light Environment

Initially, the RTA samples showed poor PL intensity and very low  $iV_{oc}$  value. It was discovered that the source of poor performance is related to the exposure of samples to the presence of high-intensity light and heat in RTA for sample heating. We



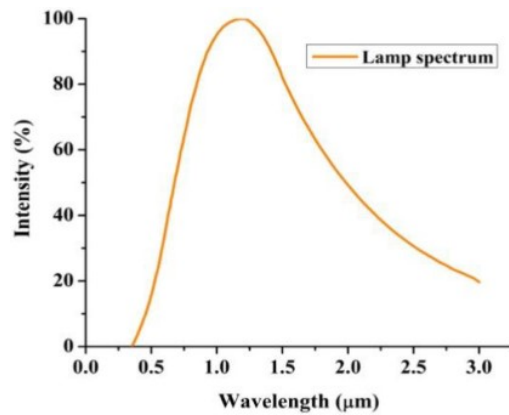


Fig. 9. Spectrum of the RTA lamp used in this study.

TABLE I  
EFFECTS OF HIGH INTENSITY RTA LAMPLIGHT ON THE PL INTENSITY AND  $iV_{OC}$  OF TOPCON STRUCTURES AT 875 °C

Sample		Before FGA		After FGA	
		PL (a.u.)	iV <sub>oc</sub> (mV)	PL (a.u.)	iV <sub>oc</sub> (mV)
As fabricated		41	525	107	537
Without Si wafer cover	N <sub>2</sub>	28	491	37	~520
	Air	27	493	45	~520
With Si wafer cover	N <sub>2</sub>	501	572	517	588
	Air	519	585	642	618

carried out experiments with samples exposed to RTA intense light along with the samples that were covered with Si wafer cover.

Fig. 9 shows the spectrum of the halogen IR lamp used in the RTA studies. The RTA durations were about 5 min at high temperatures, which was long enough to generate the thermal equilibrium. Approximately, 88.9% of the base 4-in Si wafer received the full spectrum of the RTA lamp radiation, as the test samples only occupied 11.1% of the area. Therefore, the base wafer temperature would be expected to be similar to the one without cover. The Si wafer cover absorbed all of the visible light from the lamp and radiated heat to the test sample. In addition, the temperature readout from the thermocouple touching the base Si wafer indicated the sample temperature.

Fig. 4(a) shows the light-induced degradations of peak PL intensity measured at ~1150 nm without Si wafer cover (denoted by “underlight” legend) for both N<sub>2</sub> and air at different annealing temperatures. The rest of the data were recorded using an Si wafer cover without light exposure. The data from Fig. 4(a) have been summarized and tabulated in Table I. It was observed that under high RTA temperature without the Si wafer cover, there were degradations of  $iV_{oc}$ . Later, the post-anneal FGAs recovered from this passivation loss back to the starting passivation quality. It did not improve the  $iV_{oc}$  further from the original value. By using the Si wafer cover during RTA and postanneal FGA, it was possible to get higher  $iV_{oc}$ . This light-induced

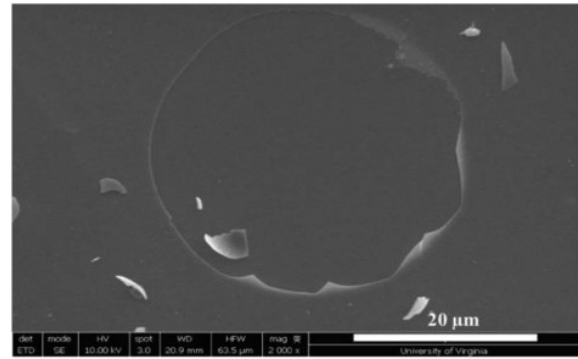
Fig. 10. Top-view SEM image showing a blister formed on the SiN<sub>x</sub>:H layer after RTA at 825 °C in N<sub>2</sub>.

TABLE II  
EFFECTS OF SiN<sub>x</sub>:H COATING ON THE PL INTENSITY AND  $iV_{OC}$  OF TOPCON STRUCTURES UNDER DIFFERENT RTA ATMOSPHERES AT 825 °C

Sample	Before FGA		After FGA	
	PL (a.u.)	$iV_{oc}$ (mV)	PL (a.u.)	$iV_{oc}$ (mV)
As fabricated	638	622	678	634
RTA in N <sub>2</sub>	935	693	1414	706
RTA in Air	730	659	775	672

degradation occurred irrespective of RTA atmospheres at only high temperatures above 625 °C.

### C. Effect of RTA on SiN<sub>x</sub>:H Coated TOPCon Structure

1) *Effect on Surface Morphology*: An SiN<sub>x</sub>:H coated TOPCon sample went through RTA at 825 °C in an N<sub>2</sub> atmosphere. The top-view SEM image in Fig. 10 shows one of the blisters observed on the SiN<sub>x</sub>:H layer due to high hydrogen pressure under RTA. The broken SiN<sub>x</sub>:H fragments can be observed lying surrounding the blister. The area fraction of blisters was measured as 7%–9% of the total surface area. If the blister formation can be minimized, that would lead to higher  $iV_{oc}$  value.

2) *Effect on Passivation Quality*: Fig. 4(a) shows the peak PL intensity of TOPCon/SiN<sub>x</sub>:H test structures annealed by the three-step process at 825 °C under both N<sub>2</sub> and air environments and the subsequent FGAs. All the related data have been summarized and tabulated in Table II. The highest  $iV_{oc}$  of 706 mV was achieved after N<sub>2</sub> annealing. For reference, the  $iV_{oc}$  of the furnace-annealed p-TOPCon structure reached 688 mV without SiN<sub>x</sub>:H and 698 mV with SiN<sub>x</sub>:H coating. This displayed the usefulness of using the SiN<sub>x</sub>:H layer for passivation purposes. The optical  $n$  and  $k$  values for as-deposited SiN<sub>x</sub>:H film were measured by ellipsometry and were 2.108 and 0.00001, respectively. The SiN<sub>x</sub>:H layer would act as an antireflection layer, allowing higher transmission of the excitation laser light and would enhance the PL intensity. However, we are comparing the change in PL intensity between unannealed and annealed samples. The absorption properties of SiN<sub>x</sub>:H are assumed to be unchanged after thermal annealing. Therefore, it does not impact

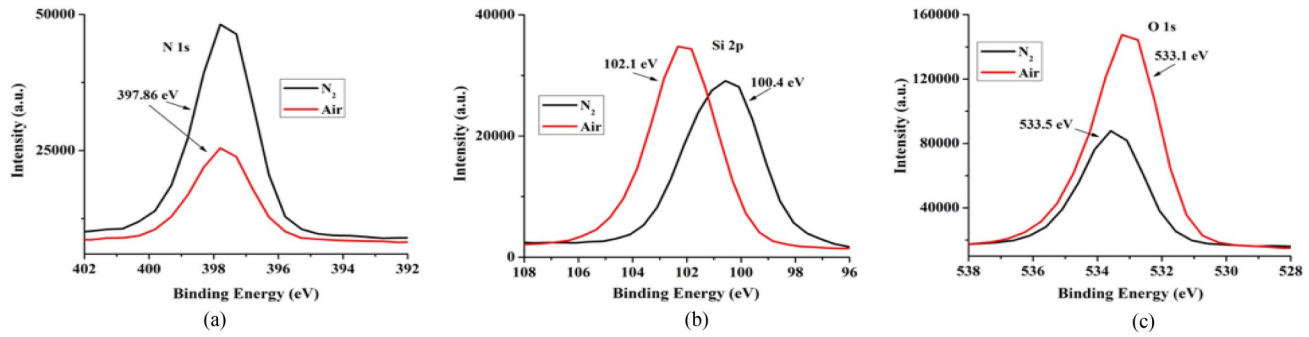


Fig. 11. XPS spectra of (a) N 1s peaks, (b) Si 2p peaks, and (c) O 1s peaks of the TOPCon/SiN<sub>x</sub>:H samples after RTA at 825 °C under N<sub>2</sub> and air atmospheres.

the PL measurement, except in the enhancement of absolute PL signal by  $\sim 30\%$  based on the light interference calculations.

3) *Effect on SiN<sub>x</sub>:H Stoichiometry*: Fig. 6(b) and (c) shows the FTIR absorbance spectra of the TOPCon/SiN<sub>x</sub>:H structures under N<sub>2</sub> and air annealing at 825 °C and compared with as-deposited film. In Fig. 6(b), it was observed that the wagging mode 1107 cm<sup>-1</sup> and bending mode 467.8 cm<sup>-1</sup> peaks of Si-O bond increased after air annealing, which indicated a greater amount of Si-O complexes being formed [45]. The absorbance peak at 612.1 cm<sup>-1</sup> corresponds to the Si-Si bond. The broad peak centered at 840 cm<sup>-1</sup> is generally attributed to the presence of the Si-N bond. The spectra from 830.7 to 854.2 cm<sup>-1</sup> denote the presence of the asymmetric stretching modes of Si-N, which underwent stoichiometric changes likely from Si<sub>3</sub>-Si-N to Si<sub>2</sub>-(H)Si-N [46]. In Fig. 6(c), it was observed that the presence of the stretching modes of N-H and Si-H peaks at 3325.4 cm<sup>-1</sup> and 2185.2 cm<sup>-1</sup>, respectively, in the as-deposited samples showed that they were highly hydrogenated [47]. Later, these peaks disappeared after RTA, which is due to the dehydrogenation.

4) *Effect on Chemical Composition of SiN<sub>x</sub>:H*: Fig. 11 shows the compositional XPS spectra of the TOPCon/SiN<sub>x</sub>:H stack annealed at 825 °C both in N<sub>2</sub> and air. In Fig. 11(a), it was observed that there was a comparatively lower count of the N (1s) peak at 397.86 eV in the air-annealed sample than in N<sub>2</sub> annealed. This suggested that there was a decrease in the Si-N bond in the top 3–5 nm thickness of the SiN<sub>x</sub>:H layer film after air annealing compared with the N<sub>2</sub> annealing. This is due to the possible Si-O bonds formation on the SiN<sub>x</sub>:H surface. In Fig. 11(b), it was observed that there was an Si (2p) peak at 100.4 eV in N<sub>2</sub> annealing and at 102.1 eV in air. The XPS peak at 100.4 eV, observed after N<sub>2</sub> annealing, is attributed to the presence of the Si-Si<sub>3</sub>N<sub>1</sub> complex and the peak at 102.1 eV, observed after air annealing, is attributed to the formation of SiO<sub>x</sub> complexes [48], [49], [50], [51]. The Si-N peak is expected to be located at 101.7 eV, and the SiO<sub>2</sub> peak to be at 103.5 eV, but peak positions are sensitive to stoichiometry [52]. Finally, after air annealing, the intensity of O (1s) peak at 533.1 eV was higher as compared with N<sub>2</sub> annealing, as shown in Fig. 11(c). This proved that there were more surficial SiO<sub>2</sub> and stoichiometric changes that occurred during the air annealing, as previously showed by Miller and Linton [53] and Lin and Hwu [54].

The results on crystallinity property determination by Raman spectroscopy, chemical changes by FTIR and XPS, and

electrical property changes by sheet resistance measurement have been presented. As the annealing temperature increases, the poly-Si crystalline fraction and  $iV_{oc}$  increase, implying lower charge-carrier recombination. The FTIR measurements were not sensitive to detect the passivation quality; however, it detects dehydrogenation in SiN<sub>x</sub>:H capped samples. The depth-profiling XPS measurements show higher concentration of oxygen in the ultrathin oxide after air annealing, which could lead to higher  $iV_{oc}$  compared with that of N<sub>2</sub> for uncapped samples. The surface morphology shows the blister formation affecting the carrier selectivity and, hence, the surface passivation.

#### IV. CONCLUSION

In this article, the effect of RTA atmosphere on surface morphology (blister formation), surface passivation, changes in polysilicon crystallinity, composition, sheet resistance, and determination of oxidation state of ultrathin SiO<sub>2</sub> passivation layer in *in situ* B-doped poly-Si-based p-TOPCon was investigated. Second, it was discovered that the high-intensity light exposure during RTA significantly degraded the surface passivation quality and, hence,  $iV_{oc}$ . Third, first depositing an SiN<sub>x</sub>:H layer on top of polysilicon and carrying out a single step of RTA provide a high  $iV_{oc}$  value of 706 mV. In summary, this article shows the following.

- 1) During annealing, hydrogen-induced blisters must be minimized as confirmed by the PL measurements in which they lead to the loss of surface passivation.
- 2) A higher  $iV_{oc}$  was observed under air RTA compared with nitrogen in SiN<sub>x</sub>:H-uncapped samples,
- 3) UV and visible light radiation during high-temperature RTA induces degradation and its exposure must be avoided during the annealing process.
- 4) The Si-O bond changes from wagging mode to stretching mode under annealing at 875 °C.
- 5) The RTA can provide low sheet resistance value of 0.9 kΩ/sq for a 30 nm *in situ* B-doped polysilicon film.
- 6) Under RTA, the fraction of polycrystalline phase maximizes at around 825 °C.
- 7) RTA in N<sub>2</sub> of SiN<sub>x</sub>:H coated polysilicon film provided the higher  $iV_{oc}$  compared with uncoated film.
- 8) RTA can provide  $iV_{oc}$  value similar to the furnace annealing.



The initial hydrogen content in the as-deposited TOPCon structure should be minimized to avoid blister formation, and the LPCVD of *in situ* B-doped a-Si:H should be done at 588 °C to minimize hydrogen incorporation as well as to obtain a higher poly-crystalline fraction.

#### ACKNOWLEDGMENT

The authors would like to thank the NSF MRI Award #1626201 for using PHI Versaprobe III XPS (UVA NMCF) instrument to obtain the XPS data.

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