

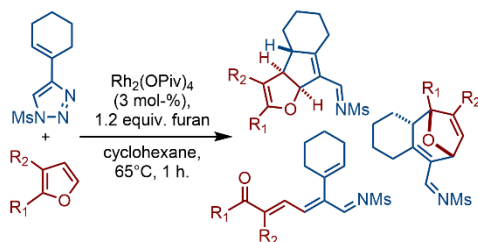
Unearthing the Subtleties of Rhodium(II)-Catalyzed Carbenoid Cycloadditions to Furans with a *N*-Sulfonyl-1,2,3-Triazole Probe

Christian J. Bettencourt,^a Tanja Krainz,^a Sharon Chow,^a Brendan T. Parr,^b William F. Tracy,^b Paul V. Bernhardt,^a Huw M. L. Davies,^{b,*} and Craig M. Williams^{a,*}

^aSchool of Chemistry and Molecular Biosciences, University of Queensland, Brisbane, Queensland 4072, Australia

^bEmory University, Department of Chemistry, 1515 Dickey Drive, Atlanta Georgia, United States

Supporting Information Placeholder

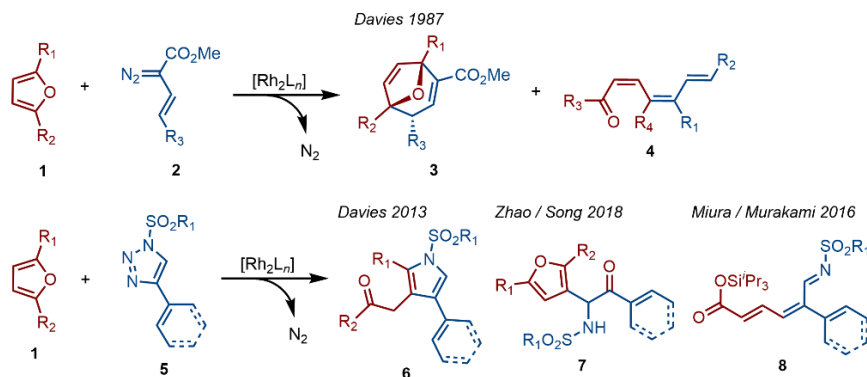


ABSTRACT: The rhodium(II)-catalyzed reaction of a model alkenyl-donor/acceptor *N*-sulfonyl triazole with a wide selection of furans is reported. This investigation unearthed a range of structurally diverse carbocyclic and ring opened products, in good to excellent yields. The products obtained are proposed to arise selectively via cyclopropanation or zwitterionic rearrangement pathways, which are highly dependent on both structural and electronic features of the furan substrate.

The use of vinylogous rhodium metalcarbenes as three-carbon synthons is a proven strategy for the synthesis of complex carbon ring systems via formal [4 + 3] and [3 + 2] cycloadditions.^{1–8} An early example of this type of reactivity is the rhodium-catalyzed reaction of furans **1** with vinyl diazoacetates **2** to form [4+3] cycloadducts **3**, which often competes with furan ring opening to form **4** (Scheme 1). In recent years, considerable interest has been shown in accessing the carbenes from *N*-sulfonyl-1,2,3-triazoles, which act as masked diazo functions, alleviating the need to isolate the decomposition-prone alkenyl-diazo moiety.^{9,10} Despite the high level of interest in *N*-sulfonyl triazole derived metalcarbene chemistry,^{11–19} few methods have been reported of their utilization in cycloaddition reactions for the construction of carbocyclic systems.¹⁰ One explanation for this observation is that donor/acceptor metalcarbenes derived

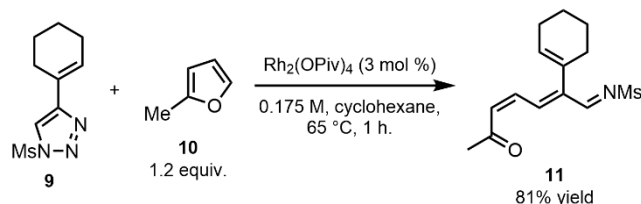
from *N*-sulfonyl triazoles contain a nucleophilic sulfonyl imine, which often becomes incorporated into the reaction to form heterocycles. This is especially true for the reaction of *N*-sulfonyl triazoles **5** with furans **1**, which can generate products derived from sulfonyl imine participation (e.g., **6** and **7**) as well as furan ring opening to form the dienes **8**.^{20–23} Therefore, a combination of competing reaction modes, from both the *N*-sulfonyl triazole and furan has caused the cycloaddition methodology for carbocycle construction to remain underdeveloped.²⁴ In order to build the utility of the reaction between *N*-sulfonyl triazoles and furans to generate carbocyclic products, we have conducted a systematic study to understand the controlling influences for the variety of potential products that can be formed.

Scheme 1. Reactions of vinylogous rhodium(II) carbenes with furans.



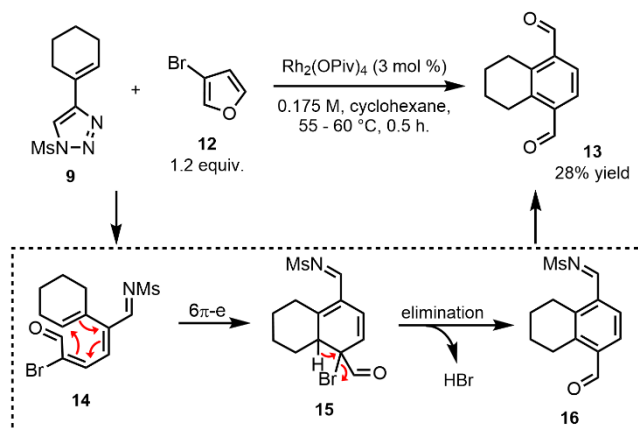
To address this issue, a study to understand the nuances of this reaction was undertaken using 4-(cyclohexenyl)-*N*-(mesyl)-1*H*-1,2,3-triazole (**9**) as a reaction probe (Scheme 2). Previous studies have tended to focus on electron-rich furans, but it was anticipated that carbocycle formation would be more likely to occur when electron-deficient furans were used. As a point of reference, the reaction with 2-methylfuran (**10**) was examined and this resulted in the clean formation of the ring-opened triene **11**. This type of product is considered to be formed by attack at C2 of the furan to form a zwitterionic intermediate, which then undergoes ring opening.¹¹

Scheme 2. Generation of a triene is preferred from 2-methylfuran.



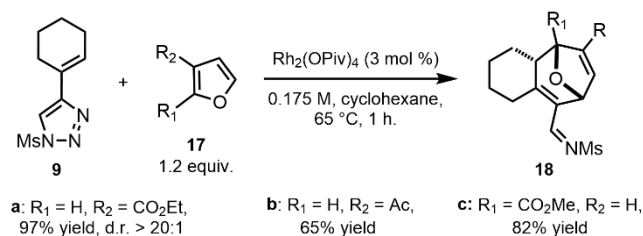
The next experiments examined electron-deficient furans. The reaction with 3-bromofuran (**12**) resulted in an unexpected outcome (Scheme 3). The reaction did not proceed smoothly, but the major product was the dialdehyde **13**, isolated in 28% yield. This product is also proposed to be derived from the initially formed triene **14**, which underwent a 6π electrocyclicization to form **15**, followed by an elimination to form **16** and imine hydrolysis to form **13**.

Scheme 3. Unexpected dialdehyde from 3-bromofuran.



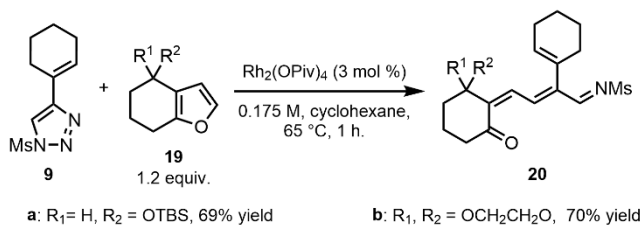
When the reaction was conducted with furans containing a strong electron-withdrawing group (**17a-c**), the chemistry was quite different (Scheme 4). In this case, the [4+3] cycloadducts **18a-c** were cleanly formed as single diastereomers in good yield. The [4+3] cycloadducts are established products from the reactions of vinylcarbenes with dienes and are considered to be generated by a cyclopropanation followed by a Cope rearrangement.²⁵⁻²⁷

Scheme 4. Electron-poor furans prefer carbocycle formation.



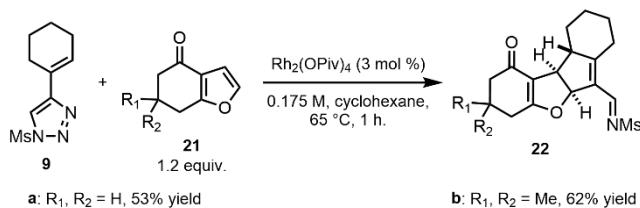
Having established a clear distinction between electron-rich and electron-deficient furans, the study was then extended to fused furan derivatives (Scheme 5). In the case of electron-rich fused furans **19a** and **19b**, trienes were the major product (**20a,b**). This would suggest that the reactions are proceeding through zwitterionic intermediates

Scheme 5. Reaction with electron rich fused furans



Extension of the chemistry to electron-deficient fused furans (i.e. **21**) resulted in the [3+2] cycloaddition product **22** in 53% yield (Scheme 6). Cyclopentene products related to **22**, have previously been accessed under rhodium catalysis from vinyl ethers with vinyl diazoacetates as the carbene source.^{28,29} The formation of **22**, from an electron-deficient furan may suggest a different mechanism to the former, which involves partial zwitterion character, but ultimately results in cyclopentene synthesis.

Scheme 6. Reaction with electron deficient fused furans



The formation of several distinct products from the reaction of the cyclohexyl iminocarbene with furans can be rationalized as shown in Scheme 7. Cyclopropanations of donor/acceptor carbenes like those of form **23** are considered to be a concerted, but highly asynchronous process as illustrated with the hypothetical transition states **24** and **25**. Depending on the functionality on the furan ring, the zwitterionic intermediates **25** and **26** can be preferentially formed instead of the cyclopropane **28**, leading to a variety of different products. Even though in principle, the zwitterionic intermediates could be generated from ring-opening of an initially formed furanocyclopropane, the preponderance of evidence suggests that they are formed in competition to cyclopropanation.³⁰ If electron-donating groups are present on the furan rings, the zwitterionic intermediates are strongly preferred, as this sufficiently stabilizes the build-up of positive charge on the furan ring. The asynchronous cyclopropanation can initiate at either the α or β positions of the furan, which is governed by steric and electronic effects. The α position of the furan is generally preferred (**24**), because it is the position more prone to electrophilic attack, but when the furan is 2,5-disubstituted then attack at the β position is favored (**25**).²⁰ The side products could be derived from fully formed zwitterionic intermediates **26** and **27** or from rhodium-bound zwitterionic structure, analogous to **24** and **25**.

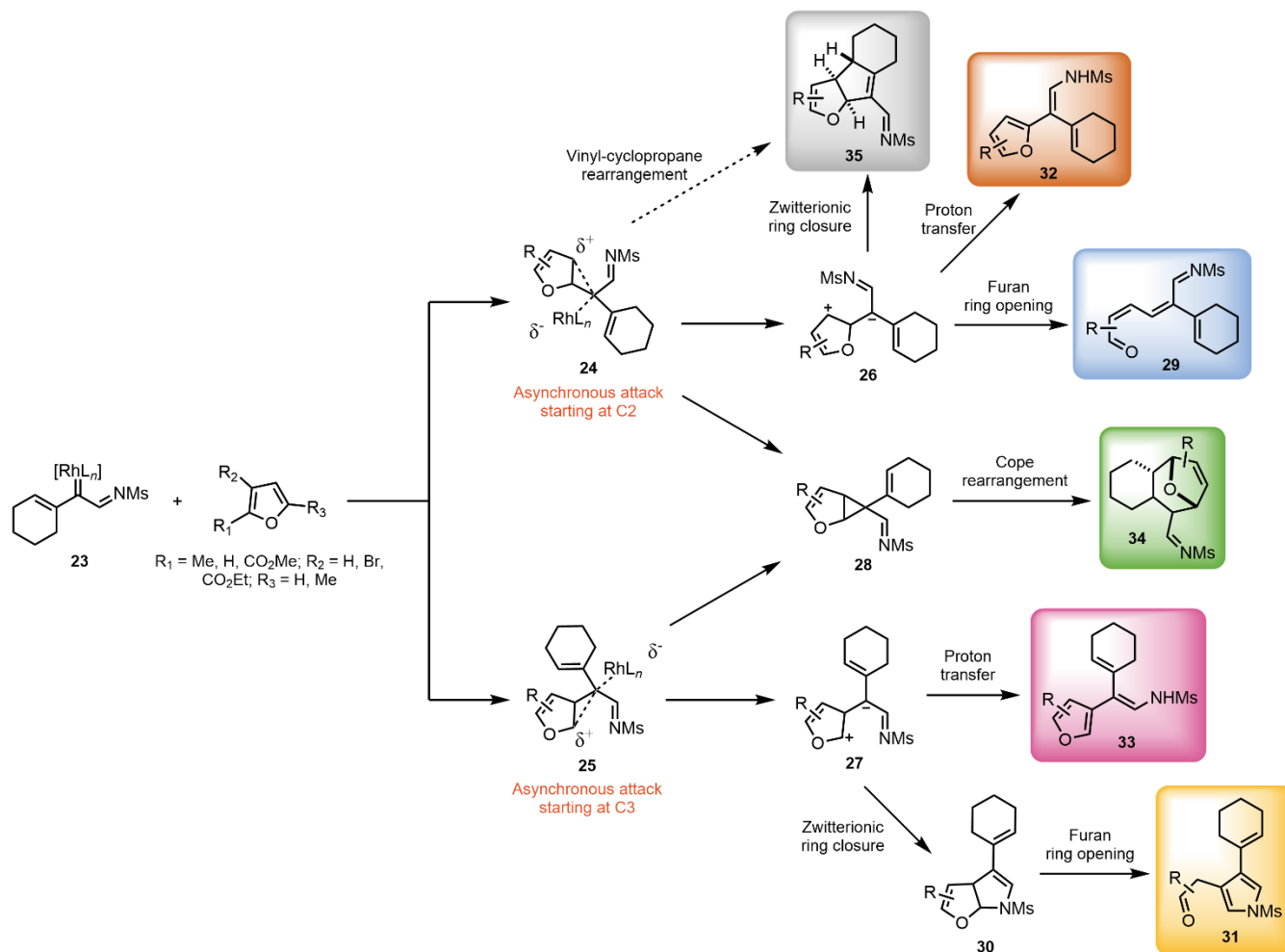
Previous studies on the reaction of *N*-sulfonyltriazoles with furans have tended to use either furan or alkylfurans as substrates, giving products derived from zwitterionic intermediates. In the case of

furan or 2-alkylfurans, the zwitterionic intermediate **26** is favored, and thus the ring-opened diene **29** is preferentially formed (Scheme 7). In the case of 2,5-disubstituted furans, the zwitterionic intermediate **27** is favored, which leads to the formation of the hemiaminal **30**, which then ring-opens to the pyrrole **31**.²⁰ Previous studies have also reported the formation of the formal aromatic substitution products related to **32** or **33**, which are likely formed by a proton transfer reaction from **26** or **27**.³¹⁻³³

In the current study, furans with electron-withdrawing substituents have been used, which lead to other possible reaction outcomes, because the zwitterionic intermediates are not highly favored. The reaction with 3-carboethoxyfuran (**17a**) led to smooth formation of the [4+3] cycloadduct **18a**, by analogy to **34**. The reaction is highly

diastereoselective because it proceeds by a cyclopropanation to form **28** followed by a Cope rearrangement, a well-established reaction for vinyl diazoacetates and *N*-sulfonyltriazoles.^{7,10} In the case of furanocyclohexenone, the initial electrophilic attack occurs at the *alpha*-position, leading to the formation of the [3+2] cycloadduct **35**. When the electron-withdrawing character of the furan is attenuated, then the diene **29** is produced. The formation of three types of products, **29**, **32** and **35**, derived from zwitterionic intermediates involving initial attack at C2 is indicative that these products may not be derived from the fully formed zwitterionic intermediate **26**. Instead, the rhodium bound zwitterionic structure analogous to **24** may be involved, especially in the formation of **35**, which is observed in furan systems with electron-withdrawing groups that would not be ideally suited to stabilize the positive charge in **26**.

Scheme 7. Plausible mechanisms for the formation of the observed reaction product classes.



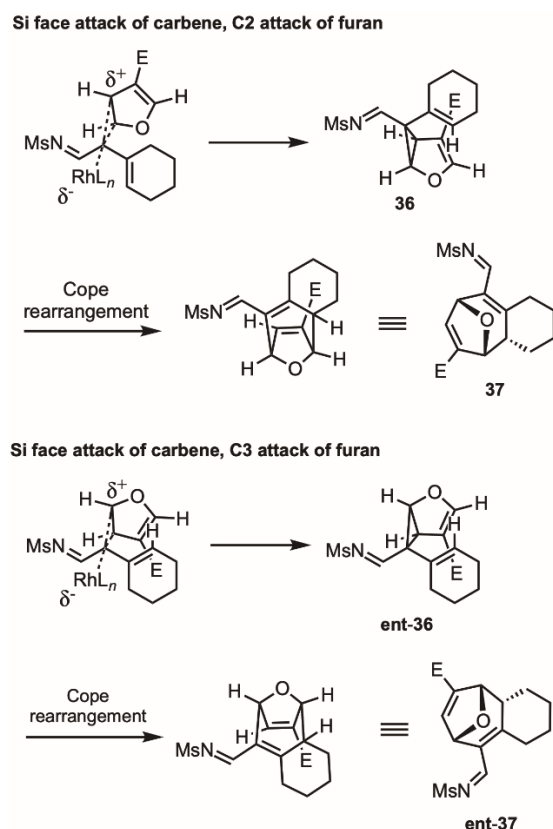
One of the more unexpected outcomes of this study is the clean formation of the [4+3] cycloadduct **18a** from the reaction with 3-carboethoxy furan (**17a**). From a steric and electronic perspective, initial attack at the C2 position of the furan would have been expected, but this typically results in side products derived from the zwitterionic intermediate such as the diene **29**. Either the ester group effectively limited charge build-up during the C2 attack, or C3 attack preferentially occurred. In order to study which process was occurring, the experimental design took advantage of the fact that initial C2- versus C3-attack would be expected to produce opposite asymmetric induction – as illustrated in Scheme 8. This effect has

been previously observed in the reactions of aryl- and vinyl diazoacetates with furans.⁵ Even though the chiral catalyst ensures only one face of the carbene is susceptible to attack (*Si* face in the model illustrated here), either enantiomer of the cyclopropane, **36** or **ent-36**, can be formed depending on the initial position of attack. The subsequent Cope rearrangement of the cyclopropane generates a specific enantiomer of [4+3] cycloadduct, **37** or **ent-37**.

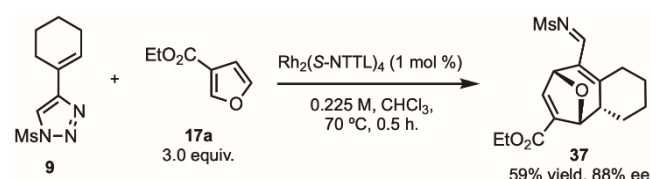
The most well-established chiral catalyst for enantioselective reactions of aryl *N*-sulfonyltriazoles is $\text{Rh}_2(\text{S-NTTL})_4$, which routinely causes the *Si* face of the carbene to be preferentially attacked.^{16,34,35} Therefore, the asymmetric induction generated in the reaction of

carboethoxyfuran **17a** with *N*-sulfonyltriazone **9** using $\text{Rh}_2(\text{S-NTTL})_4$ was examined. The $\text{Rh}_2(\text{S-NTTL})_4$ -catalyzed reaction generated the [4+3] cycloadduct **37** in 59% yield and 88% ee (Scheme 9). The absolute configuration of **37** was established by X-ray crystallography. This result is consistent with a reaction occurring at the *Si* face of the carbene with preferential attack occurring at the *alpha* position of the furan. Therefore, in this case, the electron withdrawing group on the furan disfavors the involvement of zwitterionic intermediates, leading to clean formation of the cyclopropane and subsequent Cope rearrangement to form **37**.

Scheme 8. Analysis of furan asymmetric induction.



Scheme 9. $\text{Rh}_2(\text{S-NTTL})_4$ -induced asymmetric induction.



In conclusion, this study substantially expands and bridges historical work in the area of rhodium(II) metallocarbene transformations of furans, with the use of *N*-sulfonyl-1,2,3-triazole carbene precursors. The results demonstrate that multiple types of products can be formed, and the outcome is strongly dependent on the functionality of the furan. The cyclopropanation step is concerted asynchronous with a considerable build-up of zwitterionic character during the course of these reactions. If the functionality present favors the zwitterionic structure, then the cyclopropanation can be interrupted and side products can be formed. Different types of side product can be generated depending on whether the asynchronous cyclopropanation started at the *alpha*- or *beta*-position of the furan. These studies demonstrate the controlling factors in the reactions of furans with *N*-sulfonyltriazaes, which will enable the selection of appropriate substrates for predictable transformations.

ASSOCIATED CONTENT

Data Availability Statement

The data underlying this study are available in the published article and its Supporting Information.

Supporting Information

Full synthetic procedures, X-ray crystallographic data, ^1H and ^{13}C NMR spectroscopic data, and HRMS reports (PDF).

Accession Codes

CCDC 2216763 (**13**), 2216762 (**20b**), 2216764 (**22**), and 2216761 (**37**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 902 1223 336033.

AUTHOR INFORMATION

Corresponding Authors

Huw M. L. Davies – Department of Chemistry, Emory University, 440 Atwood Hall, 1515 Dickey Drive, Atlanta, GA 30322; orcid.org/0000-0001-6254-9398

Craig M. Williams – School of Chemistry and Molecular Biosciences, University of Queensland, Brisbane, Queensland 4072, Australia; orcid.org/0000-0002-3834-7398; Email: c.williams3@uq.edu.au

Authors

Christian J. Bettencourt – School of Chemistry and Molecular Biosciences, University of Queensland, Brisbane, Queensland 4072, Australia; orcid.org/0000-0001-5855-9198

Tanja Krainz – School of Chemistry and Molecular Biosciences, University of Queensland, Brisbane, Queensland 4072, Australia; orcid.org/0000-0001-5335-9920

Sharon Chow – School of Chemistry and Molecular Biosciences, University of Queensland, Brisbane, Queensland 4072, Australia; orcid.org/0000-0001-8193-3504

Brendan T. Parr – Department of Chemistry, Emory University, 440 Atwood Hall, 1515 Dickey Drive, Atlanta, GA 30322; orcid.org/0000-0001-5774-8122

William F. Tracy – Department of Chemistry, Emory University, 440 Atwood Hall, 1515 Dickey Drive, Atlanta, GA 30322; orcid.org/0000-0003-1157-7684

Paul V. Bernhardt – School of Chemistry and Molecular Biosciences, University of Queensland, Brisbane, Queensland 4072, Australia; orcid.org/0000-0001-6839-1763

Notes

The authors declare no competing financial interest.

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