



Superhydrophobic, stretchable kirigami pencil-on-paper multifunctional device platform

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ABSTRACT

Wearable electronics with applications in healthcare, human-machine interfaces, and robotics often explore complex manufacturing procedures and are not disposable. Although the use of conductive pencil patterns on cellulose paper provides inexpensive, disposable sensors, they have limited stretchability and are easily affected by variations in the ambient environment. This work presents the combination of pencil-on-paper with the hydrophobic fumed SiO₂ (Hf-SiO₂) coating and stretchable kirigami structures from laser cutting to prepare a superhydrophobic, stretchable pencil-on-paper multifunctional sensing platform. The resulting sensor exhibits a large response to NO₂ gas at elevated temperature from self-heating, which is minimally affected by the variations in the ambient temperature and relative humidity, as well as mechanical deformations such as bending and stretching states. The integrated temperature sensor and electrodes with the sensing platform can accurately detect temperature and electrophysiological signals to alert for adverse thermal effects and cardiopulmonary diseases. The thermal therapy and electrical stimulation provided by the platform can also deliver effective means to battle against inflammation/infection and treat chronic wounds. The superhydrophobic pencil-on-paper multifunctional device platform provides a low-cost, disposable solution to disease diagnostic confirmation and early treatment for personal and population health.

1. Introduction

As the core technology in the Internet of Things (IoT) [1,2], flexible and wearable sensors have a broad range of applications such as healthcare, environmental monitoring, energy harvesting, and industrial security [3–7]. In particular, there is a growing interest to collect molecular information such as personal air quality monitoring by wearable gas sensors for generating chemical “big data” in a trillion-sensor society. However, the fabrication of most wearable electronics is expensive and complex due to the use of clean-room-based fabrication techniques on non-biodegradable substrates (e.g., polyimide [PI],

polyethylene terephthalate [PET], and silicone elastomers) [8–15]. On the other hand, paper is flexible, low-cost, lightweight, environmentally friendly, degradable, and renewable, which is promising for disposable sensors. The paper can also be combined with pencil graphite to mechanically exfoliate/form graphene/graphite layers from pencil on paper for varying flexible sensors. Recently reported paper-based sensors [16–20] have been applied for monitoring the gas [21–26], humidity [27–33], and strain [34–44] from the human body and environment. Various green and recyclable sensors can be hand-drawn with a pencil on paper at a low cost. Besides the cracks-based strain sensor to detect bending deformation [45], the gas sensor based on

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graphite powders deposited on polymer substrates by thermal embedding [46] shows high sensitivity for NO_2 and NH_3 , but the limit of detection is only ca. 100 ppb for NO_2 . The use of pencil on paper also goes beyond gas sensors [47] to multi-parameter sensors to detect gas, tensile, temperature, and electrochemistry [48–49]. However, the sensor performance is strongly affected by the complex hydrated conditions from the ambient environment to the human body [50] due to water absorption in the paper. Efforts to address this challenge have led

to the development of hydrophobic paper-based (electrochemical) sensors by the use of wax [51] or hydrophobic barriers [52]. The use of wax treatment reduces the gas permeability and limits the range of operating temperature to avoid melting, whereas the hydrophobic barrier cannot provide protection for the entire device. Meanwhile, most of the previously reported sensors exhibit cross-sensitivity to create challenges in accurately measuring target input signals [53].

Although the corrosion-resistant and superhydrophobic surface can

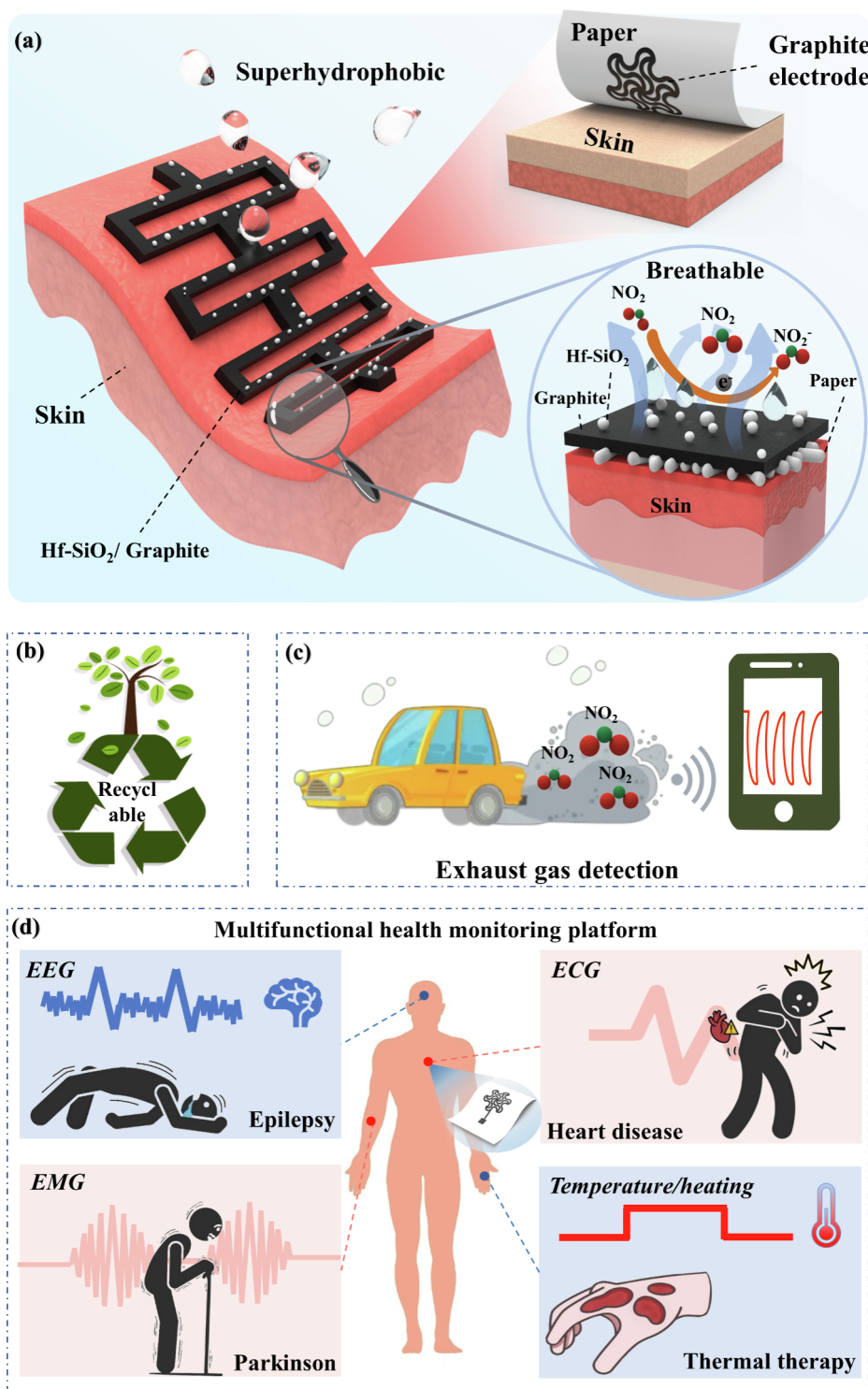


Fig. 1. Design and application of superhydrophobic pencil-paper-based multifunctional sensors. (a) Schematic illustrating the design of the superhydrophobic pencil-paper-based multifunctional sensor (b) that is green and recyclable for (c) detecting automobile exhaust gas and (d) electrophysiological electrical signals (e.g., ECG, EMG, and EEG) and temperature. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

be achieved by treating nanocomposites with fluorine molecules [54], the preparation process is complex and requires 8 h of treatment. Efforts to simplify the process result in the modification of the surface of polymer blend nanofibers with acid-treated carbon nanotubes [55] or an emulsion dip-coating strategy [56], but the use of polymers adhesives or substrates reduces the gas permeability and long-term wearability. Therefore, it is highly desirable to prepare superhydrophobic composite membranes with good air permeability by using a simple and low-cost approach.

With inspiration from the traditional Chinese paper-cutting art, kirigami has recently been applied to provide stretchability to flexible planar electronic devices [57–60], by exploring the 2D to 3D transformation in the planar thin film. Meanwhile, various surface wettability design strategies have been explored for skin-interfaced sensors and devices. Among them, the superhydrophobic coating such as hydrophobic fumed silica (Hf-SiO₂) [61–66] with nanostructures has low surface energy to provide the surface with corrosion resistance, excellent waterproof/self-cleaning property, and biocompatibility. Therefore, it is of high interest to combine the paper-based sensors with kirigami design and superhydrophobic coating to provide the next-generation

disposable wearable electronics for an improved level of comfort and minimized risks against inflammation and infections.

Here, this work reports the use of the graphite mechanically exfoliated from an 8B pencil onto a soft cellulose paper with stretchable kirigami structures and superhydrophobic Hf-SiO₂ for high-fidelity disposable bioelectronics. The resulting sensing platform fabricated on 3D curvilinear skin surfaces can monitor NO₂ gas with high sensitivity, a flow detection limit of several ppb, and fast response/recovery without being affected by the temperature and moisture variations. The integrated temperature sensor can also accurately monitor the skin temperature and reduce inflammation/infection through thermal effects. Additionally, the electrodes integrated with the sensing platform provide continuous, real-time, high-fidelity monitoring of human electrophysiological signals (e.g., ECG, EMG, and EEG) for the diagnosis of human cardiopulmonary diseases. The low-cost, environmentally friendly manufacturing approach from this work opens up opportunities for renewable and green disposable electronics.

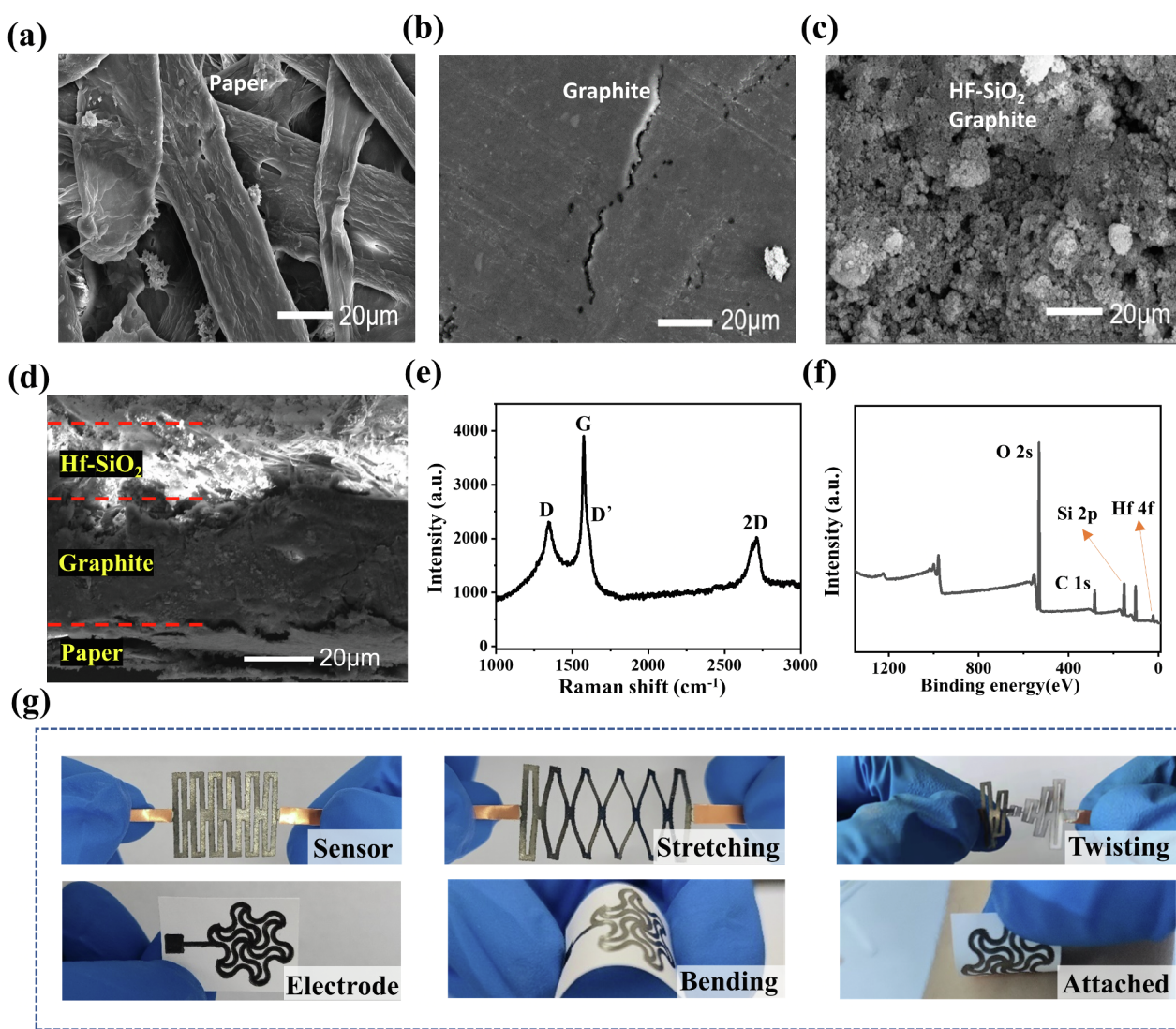


Fig. 2. Characterization of superhydrophobic conductive pencil on paper. Scanning electron microscope (SEM) images showing surface morphologies of (a) cellulose paper, (b) conductive graphite, and (c) Hf-SiO₂/graphite, respectively. (d) Cross-sectional SEM image of the Hf-SiO₂/graphite-coated paper. (e) Raman spectra of the pencil on paper. (f) XPS spectroscopy of the superhydrophobic Hf-SiO₂/graphite coating. (g) Optical images of the sensor platform under varying mechanical deformations.

2. Result and discussion

2.1. Device structure and Characterization

The design of an ultra-thin and ultra-stretchable superhydrophobic pencil-paper-based multifunctional sensing platform relies on the combination of pencil-paper sensors with kirigami architecture and hydrophobic coating (Fig. 1a, Fig. S1). The kirigami structure significantly increases the stretchability of the sensor, whereas the hydrophobic Hf-SiO₂ coating provides the sensor with high water resistance and breathability for an improved level of comfort. The use of pencils and papers that are green and recyclable (Fig. 1b) [67] in the device platform can minimize electronic waste after use while still allowing for high-fidelity measurements of automobile exhaust gas (Fig. 1c) [68,69].

The measurements of electrophysiological signals and temperature (together with heating for thermotherapy) also provide application opportunities in the diagnosis of epilepsy, atrioventricular disorder, and Parkinson's disease (Fig. 1d).

The printing paper composed of cellulose fibers (Fig. 2a) is covered with a uniform graphite film after mechanically exfoliating graphite particles from a commercial 8B pencil (Fig. 2b). Further dipping in the Hf-SiO₂ suspension [70] changes the smooth graphite surface into a rough surface with micro-scale aggregates of nano-SiO₂ (Fig. 2c). The coating of the pencil on paper with Hf-SiO₂ slightly decreases the permeability, but the gap between Hf-SiO₂ nanoparticles helps minimize this issue to still provide reasonably good gas permeability. The dip-coating of Hf-SiO₂ over graphite on paper shifts the vibration peaks of Si-O-Si and Si-O groups from 1114 cm⁻¹ to 1122 cm⁻¹ and 815 cm⁻¹ to

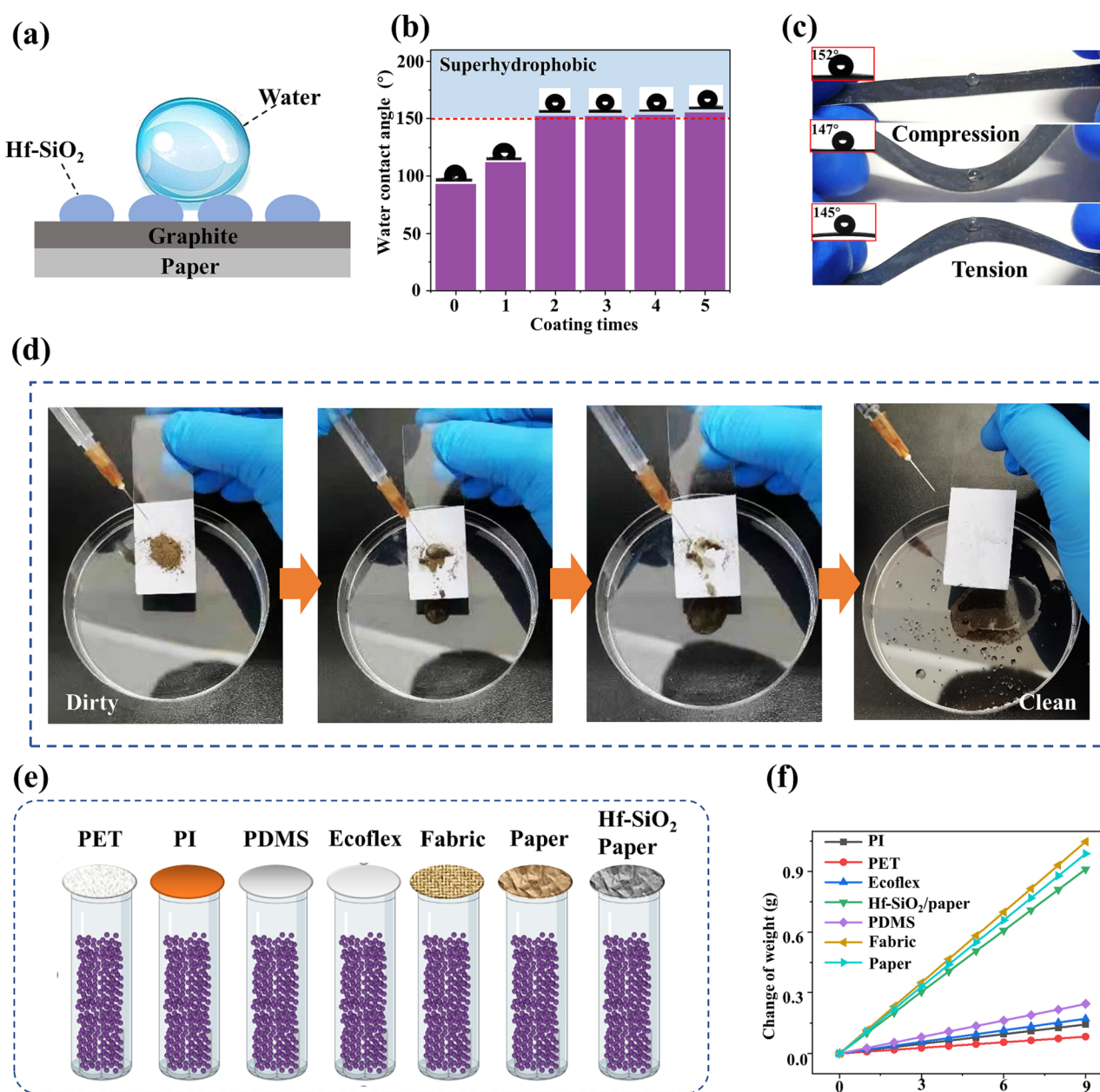


Fig. 3. Evaluation of the superhydrophobic and self-cleaning properties of the Hf-SiO₂ coating. (a) Schematic showing the superhydrophobic Hf-SiO₂ and (b) the effect of the soaking time on the water contact angle. Optical images showing (c) the minimal effect of bending on the superhydrophobic surface, (d) self-cleaning properties. (e) Schematic and (f) measurements of the film permeability revealed by the net weight changes of bottles covered by different materials (paper, fabric, PET, polydimethylsiloxane [PDMS], PI, and Hf-SiO₂ paper) over 9 days.

833 cm^{-1} , respectively (Fig. S2). The strong adhesion between Hf-SiO_2 nanoparticles and graphite is likely attributed to the hydrogen bonding between Hf-SiO_2 and graphite as in [62]. No clear interface between the Hf-SiO_2 and the graphite layer (Fig. 2d) indicates a strong adhesion between the two for enhanced mechanical performance. The slow unfolding of the kirigami structure is followed by complete unfolding to result in larger changes in the mechanical property of the sensor coated with Hf-SiO_2 , with Young's modulus increased from 1.01 to 2.64 kPa (Fig. S3). The finite element analysis in COMSOL reveals the distribution of stress concentration points (red circles) in the kirigami pattern under the uniaxial tensile strain of 100 % (Fig. S4a). The kirigami structure also exhibits good stability under compression and out-of-plane stretching (Fig. S4b, c). The maximum principal peak strain of 1.47 upon stretching of 305 % (fracture strain of the kirigami paper) (Fig. S4d) is also smaller than that of the peak strain of 63.2 for the paper (Fig. S4e). It is worth noting that the stretchability of 305.5 % for the aforementioned linear cutting kirigami pattern is much larger than that of 59.6 % and 52.1 % for the triangular and cruciform kirigami patterns (Fig. S5). The X-ray diffraction analysis (XRD) of pencil traces shows a strong diffraction peak at $2\theta = 26^\circ$, corresponding to the 002 peak of graphite (Fig. S6). The Raman spectrum of the pencil trace (Fig. 2e) shows three prominent peaks at 1353, 1574, and 2709 cm^{-1} , corresponding to the D, G, and 2D bands of multilayered graphene to graphite. The G band due to the in-plane E_{2g} mode arises from the stretching of the C-C bond, whereas the D and 2D bands are attributed to the defects at the graphite edges. The I_{2D}/I_G intensity ratio of 0.49 further confirms the existence of multilayered graphene sheets. The observed C 1s, O 2s, Si 2p (103 eV), and Hf 4f (25 eV) peaks in the XPS analysis of the Hf-SiO_2 /graphite composite (Fig. 2f) confirm the presence of SiO_2 and graphite. The excellent adhesion between layers in the composite allows the pencil-paper-based sensor platform to maintain mechanical stability under various deformations (e.g., stretching and twisting) (Fig. 2g).

2.2. Superhydrophobic self-cleaning performance

As the micro-nano-scaled rough surface of SiO_2 on the graphite/paper (Fig. 3a) is key to superhydrophobic and self-cleaning properties [71], the performance is influenced by the amount of Hf-SiO_2 nanoparticles on the surface, modulated by the number of immersion (Fig. 3b). The water contact angle (WCA) increases from 90° to 112° after one coating and to 152° after another coating. Taken together with the measured rolling angle of 2.86° , the sensor surface after the Hf-SiO_2 coating is confirmed to have changed from hydrophilicity to superhydrophobicity (Fig. S7). The WCA remains almost unchanged with further coating, indicating a stable thickness in the superhydrophobic coating layer. The superhydrophobic surface is not affected by mechanical deformations (Fig. 3c) due to its excellent adhesion. The sensor still exhibits excellent hydrophobicity as evidenced by the negligible changes in the water contact angle on the Hf-SiO_2 coatings after wearing with a 600-mesh abrasive paper over 50 times (Fig. S8). The superhydrophobic coating allows easy removal of dust particles or other contaminations off the surface during washing with water droplets (Fig. 3d). The outstanding self-cleaning property protects the sensor against various liquids commonly seen in daily life (Fig. S9). Meanwhile, measuring the weight change of bottles (with 20 g silica gel desiccant inside) covered by different films according to the ASTM E96-95 test [72] (Fig. 3e) evaluates the film permeability. The Hf-SiO_2 coated paper exhibited similar air permeability (much higher than the commonly used flexible/stretchable films) to untreated paper and fabric (Fig. 3f). The number of immersion times in Hf-SiO_2 shows a negligible small effect on gas permeability of the sensor (Fig. S10). The excellent breathability of the superhydrophobic coating allows a rapid exchange of air and moisture between the human skin and the environment for an improved level of comfort during long-term use.

2.3. Gas sensing performance of the superhydrophobic pencil-on-paper

Softer B-grade pencils (with a larger number) have a lower percentage of clay and a higher percentage of graphite to give a higher conductance (Fig. S11a, b) [73]. 8B pencil with the lowest sheet resistance of 0.33 $\text{k}\Omega/\text{sq}$ is chosen in the following gas sensor study unless otherwise specified. The morphology of the sensor surface is relatively uniform after drawing 3 times, as revealed by the SEM images (Fig. S12a). The response of the sensor after drawing 3 times to 1 ppm NO_2 also becomes almost unchanged (Fig. S12b). The geometric parameters in the kirigami design modulate the effective surface area (Fig. S13a) and the sensitivity with the optimized value achieved at a fine width ratio of 10:1 (Fig. S13b) [74]. The absorption of the oxidizing NO_2 on the P-type graphite surface [75] continuously extracts electrons to increase the number of holes in the graphite and thus reduce its resistance [76–77] (Fig. S14). The resulting sensor exhibits a highly stable response of $\sim 1.05\%$ to 1 ppm NO_2 for five consecutive cycles (Fig. 4a), a fast response/recovery time of 162/534 s (Fig. S15a), and small variations among samples (Fig. S15b). According to the definition of the signal-to-noise ratio (SNR): $\text{SNR} = 20\log_{10}(R/R_{\text{rms}})$, the sensor exhibits an ultra-high SNR of 35 dB at 1 ppm. The dynamic response of the sensor to high (1 to 5 ppm) and low (500 to 900 ppb) NO_2 gas concentrations (Fig. S15c and Fig. 4b) demonstrates the large range of detection. The slope of 1.03 ppm^{-1} in the linear fit ($R^2 = 0.988$) of the response to gas ranging from 500 to 900 ppb (Fig. 4c) helps calculate the theoretical limit of detection (LOD) to be 800 ppt according to $\text{LOD} = 3 \times \text{RMS}_{\text{noise}}/\text{slope}$, with $\text{RMS}_{\text{noise}}$ as the standard deviation in the response baseline. The theoretical estimation is validated by the experimental demonstration of the sensor response of 0.02 % to 6 ppb NO_2 (Fig. 4d). The excellent selectivity of the gas sensor to NO_2 is confirmed by its much higher response to 1 ppm of NO_2 over that to 100 ppm of NH_3 , CO_2 , acetone, ethanol, and methanol (Fig. 4e). The sensor also exhibits long-term stability over 2 weeks (Fig. 4f).

The stretchable kirigami provides the sensor with stable performance over tensile strain from 0 to 40 % (Fig. 4g) and multiple repeated stretching of 10 % over 1500 cycles (Fig. S16a). However, the SNR of the sensor decreases from 35 dB to 30 dB upon 40 % stretching (Fig. S16b). The sensing response only shows a small standard deviation of 2 % as the bending radius of curvature (r) reduces from 35 to 5 mm (Fig. 4h) or over repeated bending of 1500 cycles ($r = 30 \text{ mm}$) (Fig. S16c). For an applied tensile strain of 30 %, the sensor shows fluctuations in the response due to the unfolding of the kirigami pattern (Fig. S17). It stabilizes to be ca. 1 %, which is much smaller than that of 1.05 % for the gas response. Furthermore, strain isolation and decoupling sensing strategies [53,78] can be used to minimize the strain effect. As the Hf-SiO_2 coating on the graphite surface is gas permeable, it does not affect the gas response (Fig. 4fi). Due to the excellent hydrophobicity and self-cleaning property from the Hf-SiO_2 coating, the sweat droplets on the sensor surface do not affect the sensing response after drying, compared to the one without the coating (Fig. 4j and Fig. S18). In addition, the pencil-on-paper-based multifunctional sensor can be easily ignited and carbonized in about 4 s for ease of disposal (Fig. S19a). The deformable gas sensor can be deployed onto curved linear surfaces of 3D objects to detect exhausts from gasoline and diesel vehicles (Fig. S19b). Self-heating of the sensor based on Joule heating [78–79] can elevate the operating temperature to increase the response to 1 ppm NO from 1.08 % at room temperature to 1.20 % at 50°C (Fig. 4k) due to accelerated electron transfer to overcome the barrier [80]. More importantly, the gas sensor operated at 50°C also minimizes the influence of the variations in the environmental temperature and relative humidity (Fig. 4l). As a result, the superhydrophobic, stretchable flow-cost NO_2 gas sensor with an ultra-low LOD and robust mechanical performance outperforms the previous studies based on graphite sensing materials on cellulose substrates (Table S1).

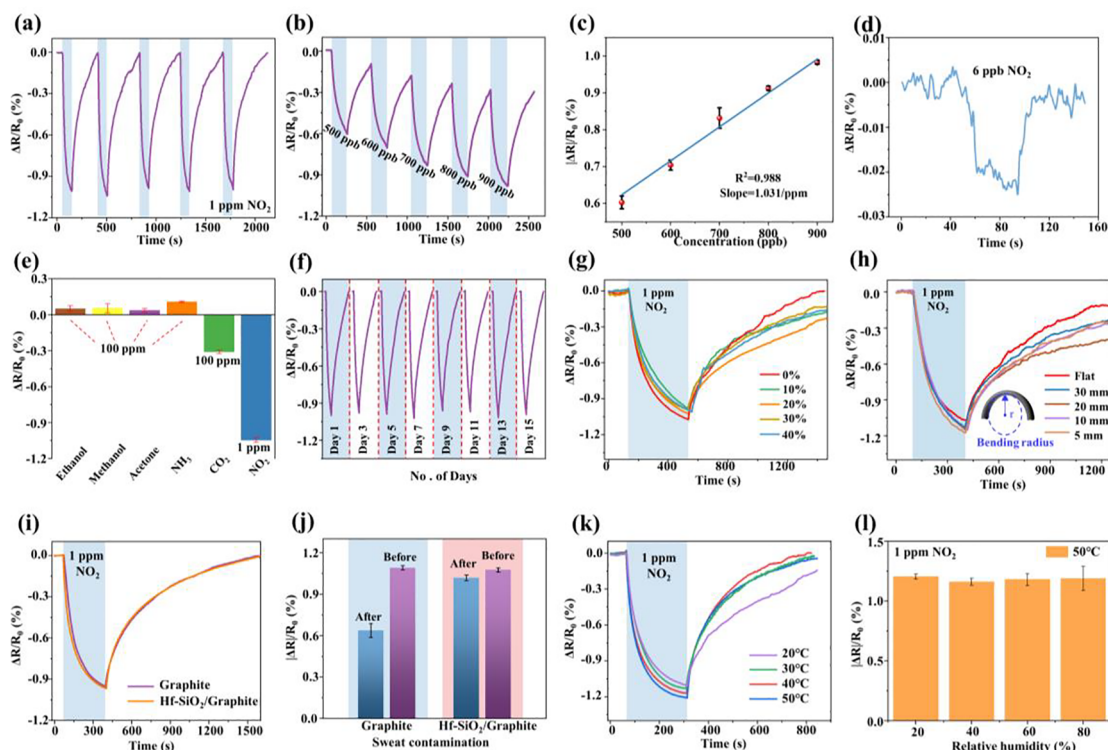


Fig. 4. Gas sensing performance and mechanical stability of the pencil-paper-based sensor. (a) Repeatability test of the sensor to 1 ppm NO₂ over five cycles. (b) Dynamic response of the sensor to NO₂ from 500 to 900 ppb and (c) the linear fit of the response. (d) Experimental demonstration of the gas sensor to detect 6 ppb NO₂. (e) Selectivity of the sensor to NO₂ over ammonia, acetone, ethanol, NH₃, and CO₂. (f) Sensor gas response to 1 ppm NO₂ over 15 days. (g) Gas responses of the sensor to 1 ppm NO₂ under tensile strain from 0 to 40%. (h) Gas responses to 1 ppm NO₂ at different bending radii reduced from 35 to 5 mm. (i) Comparison in the response of the pencil-on-paper before and after coating of Hf-SiO₂. (j) Influence of sweat contamination on the gas response to 1 ppm NO₂ of the sensor without and with the Hf-SiO₂ coating. (k) Comparison of the gas response to 1 ppm NO₂ at an operating temperature from 20 to 50°C. (l) Response of the sensor operated at 50°C to 1 ppm NO₂ at different relative humidity levels.

2.4. Temperature sensing and heating performance of the hydrophobic pencil-on-paper

As one representative transition metal carbide, graphite shows a metallic negative thermal coefficient (NTC), which means that the increased temperature can facilitate charge carrier mobility [81]. As the graphite potential barrier height is very low, thermionic emission becomes dominant to be primarily responsible for the decreased resistance upon temperature increase [82–83]. As shown in the I-V curve (Fig. 5a), the increased sample temperature in the range of 25–100°C indeed leads to decreased resistance. The sensor also exhibits a linear decrease in the resistance of the sensor with increasing temperature ($R^2 = 0.995$) (Fig. 5b), repetitive response to varying temperatures (Fig. 5c, d), a high temperature resolution of 0.5°C (Fig. 5e), relatively fast response to heating/cooling cycles (Fig. 5f). As a result, the sensor can easily detect body temperature ($\sim 36.3^\circ\text{C}$) (Fig. 5g), water temperature in the cup (Fig. 5h), and the temperature of the water flowing through the curved pipes (Fig. 5i). The temperature sensor with good sensitivity and large detection range compares favorably with previous studies using graphite materials on cellulose substrates (Table S2). By modulating the applied voltage, the skin surface temperature can also be effectively increased by the pencil-on-paper heater (aspect ratio of 10:2.5) due to Joule heating (e.g., 15 V for 50°C) (Fig. 5j), which can serve as a thermal therapy to treat joint injuries, improve blood circulation, sterilize medical dressing, and promote wound healing [84–87].

2.5. The superhydrophobic pencil-on-paper to decouple temperature and gas

As the pencil-on-paper sensor sensitively responds to both NO₂ gas

and temperature, it is imperative for the sensor to decouple gas and temperature when the two stimuli are simultaneously present. After introducing a thin PDMS encapsulation that is impermeable to gas (Fig. 6a), the resulting pencil-on-paper sensor exhibits a significantly diminished response to NO₂ (Fig. 6b). The thin PDMS encapsulation only needs to be in the local sensing area to maintain the gas permeability of the sensor. Meanwhile, the rapid heat transport in the thin PDMS film has minimal effect on the temperature sensing, resulting in a negligible difference from the sensor without encapsulation (Fig. 6c). Taken together with the unencapsulated sensor operated at an elevated temperature of 50°C that is insensitive to temperature, the pair of sensors can completely decouple the temperature and NO₂ gas (Fig. S20). In particular, the progressively increased temperature from 22 to 40°C and then decreased back to 22°C (to 3 ppm NO₂) is accurately detected by the encapsulated sensor (top) but does not cause any response in the self-heated sensor (bottom) (yellow shaded region in Fig. 6d). Meanwhile, the variations in the NO₂ gas from 1 to 3 ppm (at room temperature) are captured by the self-heated sensor operated at 50°C (bottom), whereas the encapsulated sensor does not show any response (top) (blue shaded region in Fig. 6d).

2.6. Multifunctional device system for electrophysiological signal monitoring and electrical stimulation

The superhydrophobic pencil-on-paper electrode designed in the open-mesh, serpentine layout (Fig. S21a) provides the real-time, continuous, and high-fidelity monitoring of electrophysiological (EP) signals such as the electrocardiogram (ECG) (Fig. 7). The electrode exhibits negligibly small changes in the resistance upon bending to a radius of 2 mm (Fig. S21b) and over 1000 repeated cycles (bending

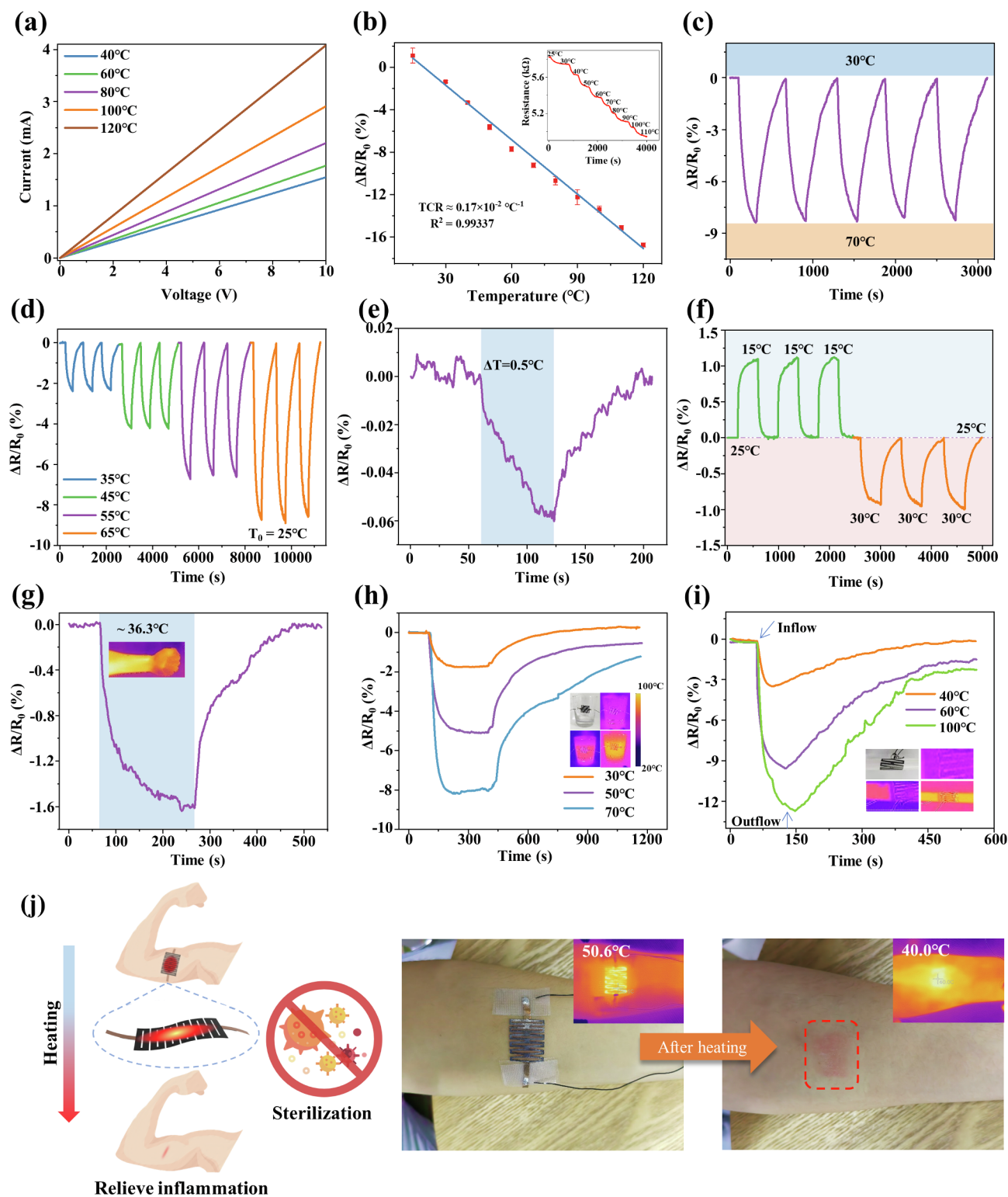


Fig. 5. Application of the hydrophobic pencil-on-paper for temperature sensing and heating. (a) Current-voltage curves of the superhydrophobic pencil-on-paper for the temperature from 40 to 120°C and (b) fits normalized relative resistance change ($\Delta R/R_0$) with temperature. (c) Reusable sensor response cycled between 30 and 70°C. (d) Repeatability sensor response as the temperatures progressively increased from 35 to 65°C. (e) Demonstration of the sensor to detect a small temperature variation of 0.5°C. (f) Reusable sensor response cycled between cold (15°C) and hot (30°C) sources. Demonstrations of the sensor to detect (g) human skin temperature and water temperature (h) in the cup or (i) flowing through the pipe. (j) Pencil-paper sensor as thermal therapy for reduced inflammation and infection (left): human skin temperature during (middle) and after (right) heating.

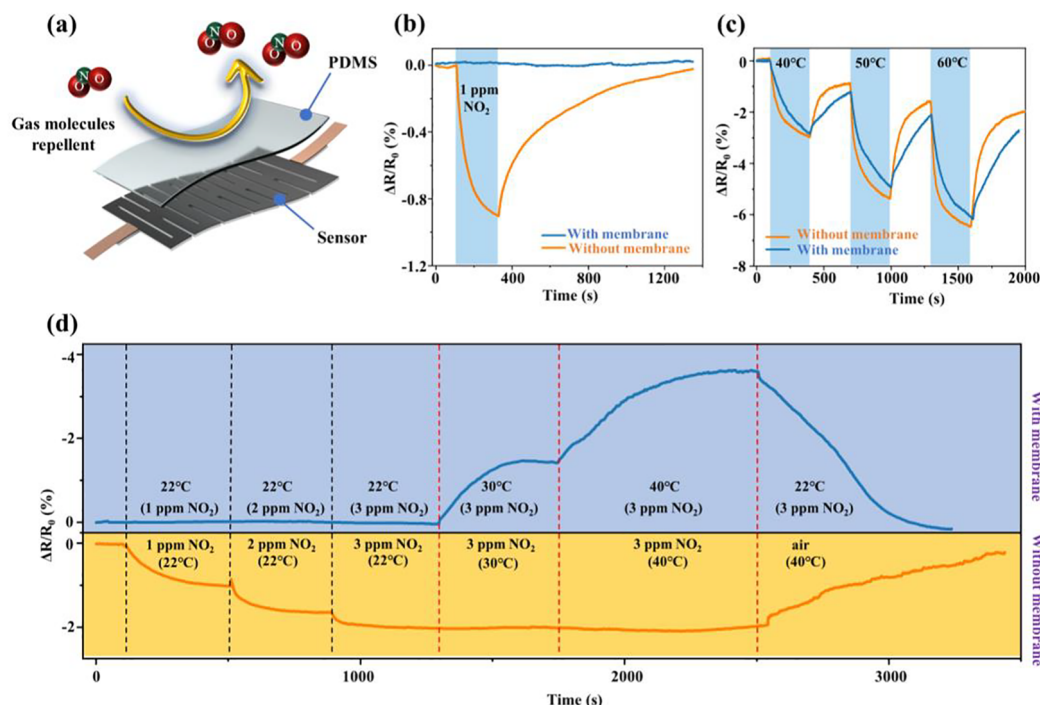


Fig. 6. Demonstration of the superhydrophobic pencil-on-paper sensor to decouple gas and temperature. (a) Schematic showing the encapsulated sensor to block the permeation of the gas molecules. (b) Response of the pencil-on-paper sensor with and without the encapsulation membrane to 1 ppm NO₂. (c) Response of pencil-on-paper sensor with and without encapsulation to the temperature of 40, 50, and 60°C. (d) The combination of the pencil-on-paper sensors with and without encapsulation to completely decouple NO₂ gas and temperature.

radius of 10 mm) (Fig. 7a), as well as high reproducibility among samples (Fig. S22). As the writing cycle increases from 1 to 18, the sheet resistance of the pencil-on-paper electrode (from one-stroke drawing) rapidly decreases from 41.97 kΩ/sq to 0.33 kΩ/sq after 15 cycles and then saturates to 0.30 kΩ/sq (Fig. 7b). As the EP signal quality hinges on the contact quality at the electrode/skin interface, it is essential to choose a biocompatible adhesive with strong adhesion. The use of methylcellulose with an optimal concentration of 20 mg/ml provides the highest signal-to-noise ratio of 39.65 dB (Fig. 7c). Therefore, the electrodes with 15 writing cycles and 20 mg/ml methylcellulose are used in the following studies.

Corresponding to the atrial activation, activation, and depolarization of the ventricles, and repolarization of the ventricles [88], the P wave, QRS complex, and T wave captured by the superhydrophobic pencil-on-paper electrodes are almost identical to those from commercial Ag/AgCl electrodes (Fig. 7d). However, our electrode provides higher SNR (39.28 dB) than the commercial Ag/AgCl electrode (35.82 dB) and exhibits robust performance minimally affected by the environmental temperature from 25 to 45°C (4.2 % change in the SNR from 38.29 to 36.65 dB) (Fig. 7e). Compared with the noisy ECG signals measured from the graphite electrodes without Hf-SiO₂ coating during sweating, the signals from the ones with the coating do not show noticeable changes (Fig. S23). In addition, a high SNR value (>38 dB) is maintained at different skin humidity levels: 50 % RH at rest, 65 % RH from slight sweating, and 85 % RH from significant sweating (Fig. 7f). The electrodes can be used repeatedly over 5 peeling/sticking cycles (Fig. 7g) and over 24 h (Fig. 7h) with outstanding SNR and no noticeable adverse reactions for enhanced comfort and long-term reliability (Fig. 7i). The amplitude of the R peak decreases (or increases) with the increased (or decreased) intrathoracic impedance during inspiration (or exhalation) and remains unchanged during breath holding [89] (Fig. 7j). The respiratory rate (~14 min⁻¹) can also be calculated from the ECG signals (Fig. S24). The heart rate variability (HRV) determined from the R-to-R and P-to-P peak intervals can also be accurately recorded by our electrodes (Fig. 7k) to diagnose autonomic nervous dysfunction that affects

the cardiac sinoatrial node [90]. The HRV can also be used in psychopathology and emotion regulation brain networks to diagnose human depressive disorder and anxiety and analyze the physical health conditions under mental stresses [91–93].

The superhydrophobic pencil-on-paper electrodes can also capture the subtle and rapid skeletal muscle movements produced by recurrent nerve stimulations, including the EMG signals from the human subject during the pull-up exercise (Fig. S25a) with significant sweat production (Fig. S25b). The muscle signals caused by neck rotation (Fig. 8a) and hand motions (e.g., clenching fist, rotating wrist, and unfolding palm) (Fig. S25c) can also be recorded by our electrodes (29.03 dB) and qualitatively compared with those from the commercial electrodes (29.45 dB). The electrodes on the forearm can be used to measure EMG signals for Parkinson's disease such as a simulated resting tremor with a frequency of 3 Hz (Fig. 8b). Besides measurements, the electrodes attached to the skin can stimulate nerve fibers through the generated current to provide opportunities for sensation, electrical massage, and pain relief. By applying square pulse current stimulation (amplitude of 150 μA, pulse duration of 0.2 ms, frequency of 50 Hz) to the forearm (Fig. S25d), the amplitude and RMS of the EMG signals from clenching fist are increased compared with the ones without stimulation (Fig. 8c, d). The slight thermal sensation at the electrical stimulation site is also conducive to applying neural regulation, rehabilitation therapy, and artificial limb motion feedback.

The high-fidelity EEG signals with five characteristic frequency bands (i.e., δ wave of 0.5–3 Hz, θ wave of 4–7 Hz, α wave of 8–13 Hz, β wave of 14–30 Hz, and γ wave of 31–50 Hz) can also be continuously monitored for detecting diverse cerebral activities [94]. As a proof-of-concept example, blinking at a frequency of 0.2 and 1 Hz can be clearly identified in the recorded EEG signals with electrodes on the forehead (Fig. 8e). The cerebral activity in a conscious sedentary state is distributed in the α waveband as revealed by the power spectral density (PSD) analysis (Fig. 8f) and the intensity is low according to the time-frequency spectrogram (Fig. 8g).

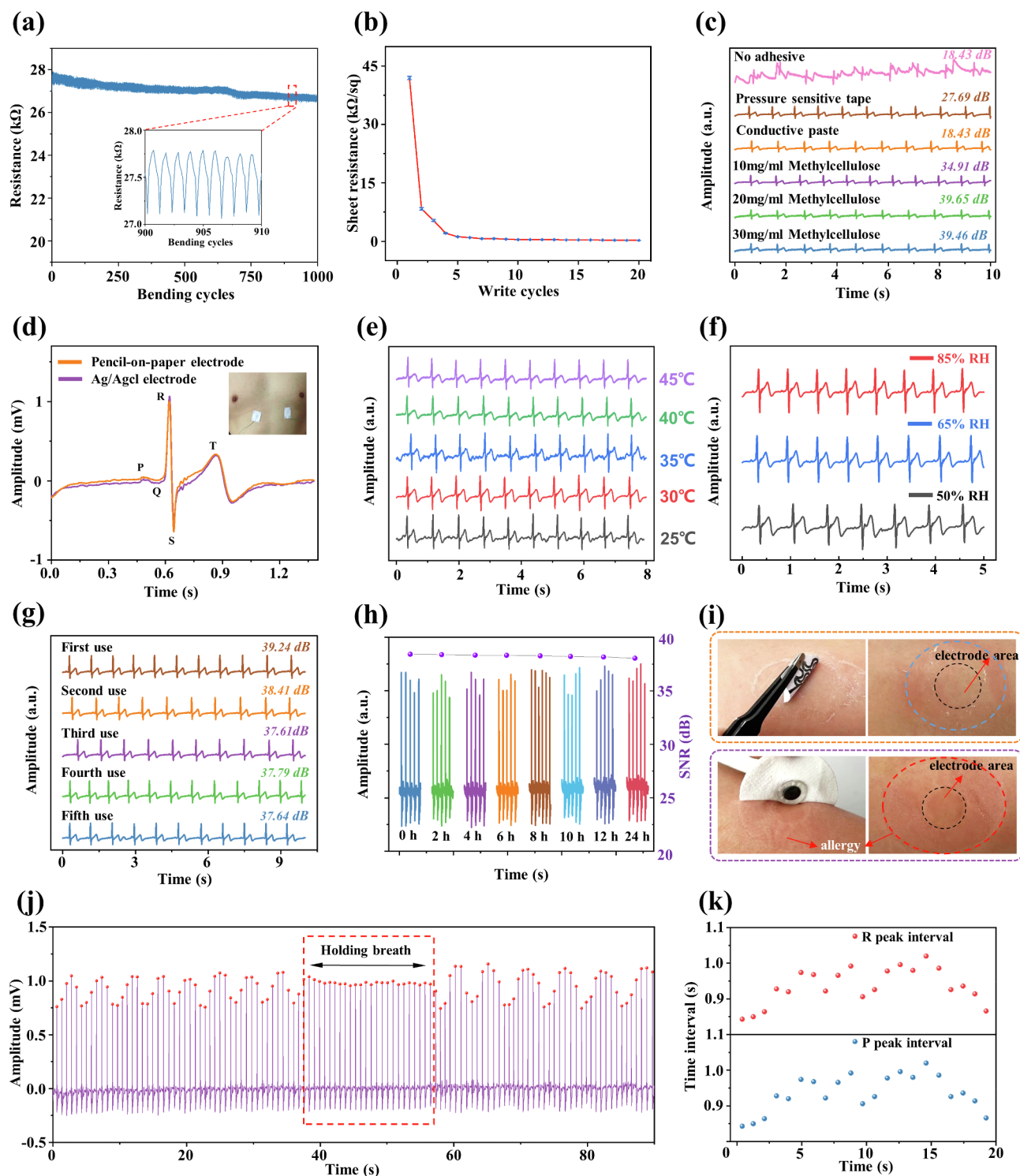


Fig. 7. High-quality ECG detection. (a) The resistance change of the electrode over repeated bending of 1,000 cycles (bending radius of 10 mm). (b) The sheet resistance of the pencil-on-paper as a function of writing cycles (error bars from five samples). (c) Comparison of ECG signals measured by the electrode with different adhesives. (d) Comparison of ECG signals between the commercial Ag/AgCl and the pencil-on-paper electrodes. ECG signals measured at different (e) environmental temperatures (25–45°C) and (f) skin humidity levels (50–85 % RH) before and after sweat. Comparison between ECG signals (g) with different times of use and (h) over 24 h. (i) Comparison of the skin condition between the commercial Ag/AgCl and the pencil-on-paper electrodes placed on the skin for 24 h. (j) Dynamic amplitude responses of the R peaks in the ECG signals while holding breath for ~ 20 s. (k) The heart rate variability determined from the R-to-R and P-to-P peak intervals in the ECG signals.

3. Conclusion

In conclusion, the superhydrophobic, knifem, stretchable pencil-on-paper device platform can simultaneously detect gas, temperature,

and electrophysiological signals, as well as provide treatments via thermal therapy and electrical stimulation. The resulting sensor operated at elevated temperature from self-heating shows a large response and small limit of detection without being affected by variations in the

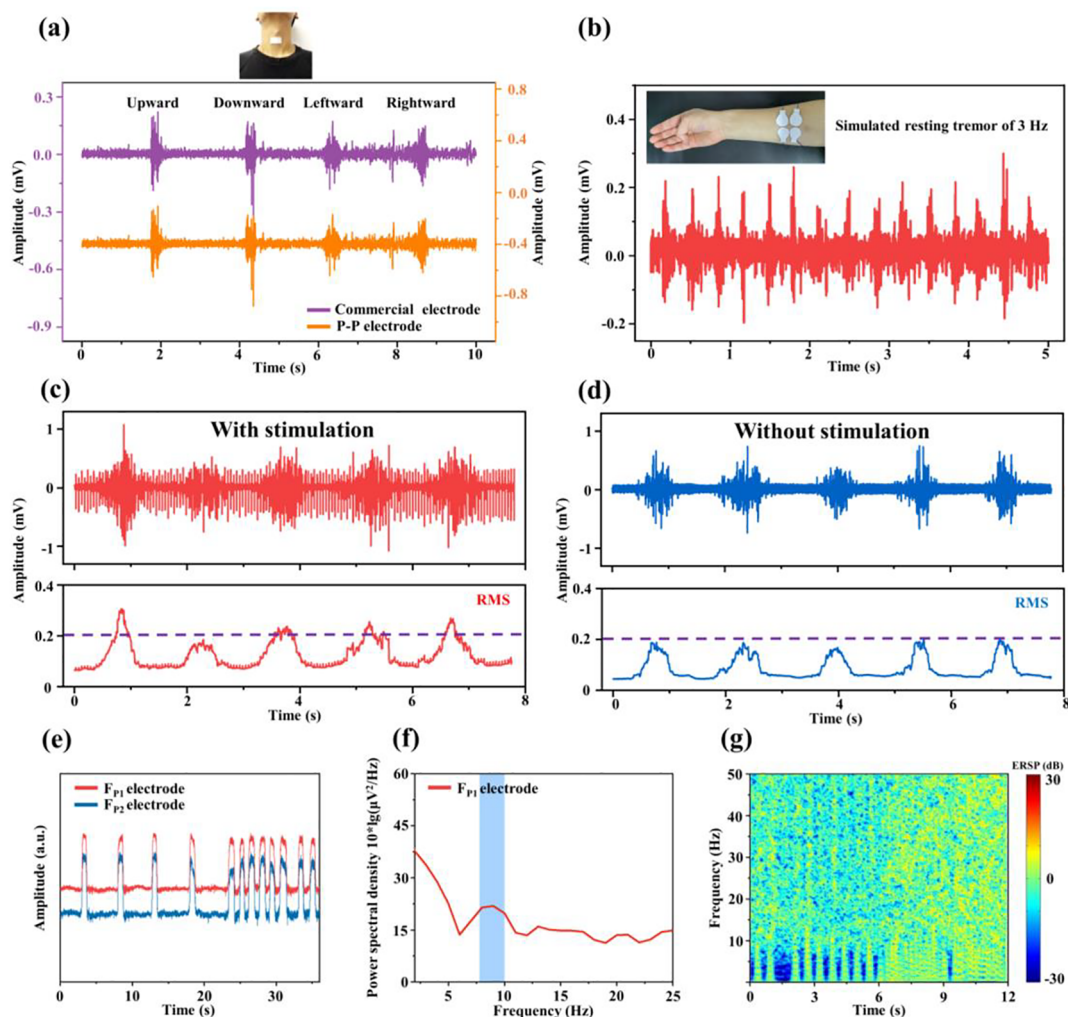


Fig. 8. High-fidelity monitoring of EMG and EEG. (a) EMG signals measured from the human subject during neck rotation (electrodes on the throat). (b) EMG signals of the simulated Parkinson's static tremor with a frequency of 3 Hz measured by electrodes attached to the forearm. Comparison of the EMG signals (c) with and (d) without simultaneous electrical stimulation. (e) EEG signals at a blinking frequency of 0.2 and 1 Hz (electrodes on the forehead). (f) Power spectral density and (g) time-frequency spectrograms of the EEG signals in (e).

ambient environment or mechanical deformations. The temperature sensor with a high precision of 0.5°C can alert for potential adverse thermal effects or inflammation/infection. The electrodes can also be combined to monitor electrophysiological signals for accurate diagnosis of cardiopulmonary diseases. Additionally, the thermal therapy and electrical stimulation provided by the device platform can battle against infection and accelerate chronic wound healing processes. Taken together with onboard processing and communication units, the device platform from this work can be exploited for future wearable electronics with closed-loop control for early disease diagnosis and therapeutics.

4. Experimental section

4.1. Materials

Mitsubishi 9800 Sketch 8B pencil (Japan) was used as received. Hf-SiO_2 with a diameter of 20 nm and a specific surface area of $230 \text{ m}^2/\text{g}$ was purchased from China Taping Metall Materials Co., Ltd. Anhydrous ethanol (99.7 %) was purchased from Jiangsu Qiangsheng Functional Chemical Co., Ltd. Methylcellulose was purchased from Sinopharm Chemical Reagent Company. Commercial ECG electrodes were purchased from Zhejiang XUNDA Brand.

4.2. Fabrication of the pencil-paper-based multifunctional sensor

The A4 printing paper was first cut into rectangular sheets of the target size ($5 \times 7 \text{ cm}^2$). After drawing the pencil on paper, it was cut by a CO_2 laser (VLS 2.30 with a wavelength of 1064 nm and a 2.0 lens from UNIVERSAL, USA) into the desired kirigami pattern. Fabrication of electrophysiological electrodes relied on complete and even filling of the serpentine pattern printed on cellulose paper with the 8B pencil for 15 times. Next, the pencil-on-paper was repeatedly dipped in the Hf-SiO_2 suspension to prepare the superhydrophobic coating, followed by drying at 50°C . The copper wire with silver paste was used to connect the sensor to the data acquisition system. Methylcellulose of 60 mg was added to 3 ml of deionized water and then ultrasound for 10 min in an ultrasonic homogenizer (SCIENTZ-IIID). A stable methylcellulose solution was obtained after 1 h, which was then used to attach the sensor to the skin of human subjects.

4.3. Characterization

Scanning Electron Microscopy (SEM) images were collected by Field Emission Scanning Electron Microscopy (JEOL, JSM 7100F). X-ray photoelectron spectroscopy (XPS) was recorded using an ESCALAB 250 photoelectron spectrometer (Thermo Fisher Scientific, USA). X-ray

photoelectron spectroscopy (XRD) was recorded by a D8 Advance photoelectron spectrometer (Bruker, Germany). A Nexus 870 fourier infrared spectrometer (Nicolet, USA) was used to record FT-IR. Raman spectra were recorded by a laser microscope Raman spectrometer (Renishaw, in Via Reflex). Electromechanical measurements of the sensor were performed with a universal material testing machine (JSV-H1000, Japan). The sheet resistance was measured with a four-point probe. Contact angle images were obtained with a fully automatic contact angle measurement system (DSAHT, KRÜSS GmbH).

4.4. Testing of multifunctional sensors

Different concentrations of NO₂ were prepared by diluting and fully mixing 100 ppm NO₂ calibration gas with air in the testing chamber of 5 L. Different concentrations of NO₂ were prepared by diluting and fully mixing 100 ppm NO₂ calibration gas with air in the testing chamber of 5 L. The gas of the ultra-low concentration used for the validation of the limit of detection was prepared by further diluting 10 ppm NO₂ gas with air at a ratio of 1:9 in the airbag. The gas and temperature sensing performances of the pencil-paper-based sensor were measured and recorded with a source meter (Keithley 2400) at 0.1 V. The temperature distribution was measured by an infrared camera (ONE PRO/LT, FLIR, USA). A Commercial Xifex ECG sensing kit was used to collect and transmit ECG signals to a computer with electrodes placed on the chest in a two-lead configuration. EMG signals were detected by connecting the Trigno Wireless EMG System (DELSYS, USA) with the electrodes on the biceps. The electrodes on both sides of the forehead were connected to the F_{p1} and F_{p2} electrodes of the commercial acquisition system (SynAmps2, Neuroscan) to collect EEG signals.

Data availability

The authors declare that the data supporting the findings of this study are available from the corresponding author upon reasonable request.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

Data will be made available on request.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.cej.2023.142774>.

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