

Alkynyl Prins Carbocyclization Cascades for the Synthesis of Linear-Fused Heterocyclic Ring Systems

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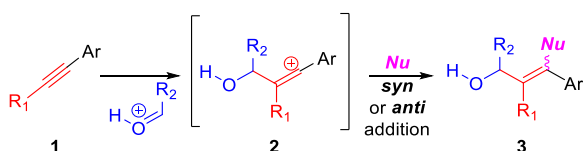
We report a Brønsted acid-catalyzed carbocyclization cascade, featuring condensation of an alcohol/sulfonamide with an aldehyde followed by an intramolecular three-component coupling involving an alkyne, an oxocarbenium/iminium ion, and an arene. A formal cycloaddition is embedded in the cationic cascade, which enables the synthesis of a wide range of fused heterocycles. The diastereoselectivity of the cascade is studied using secondary alcohols/sulfonamides with different carbonyl partners. The described method results in the preparation of synthetically versatile scaffolds with ample opportunity for further derivatization at the resulting tetrasubstituted olefin, or by inclusion of other functionalizable motifs from the starting materials. It is worth noting that this chemistry also facilitates the synthesis of piperidines and 1,4-oxazepanes, as well as the inclusion of indoles and benzofurans, which are privileged motifs for medicinal chemistry. Herein we present the generality of this approach and some chemical transformations that can be achieved with our substrates.

Introduction

The development of methods that enable the synthesis of small molecules is essential for populating high throughput screening libraries used in drug discovery campaigns.¹ In particular, the preparation of synthetically versatile molecules that allow for differential functionalization around a polycyclic scaffold will help maximize diversity among derivatives and provide the ability to tune physicochemical properties.²

Alkyne-carbonyl coupling (Prins) reactions most commonly terminate with the capture of a presumed vinyl cation intermediate **2** with a nucleophile (Scheme 1).³

Scheme 1. Alkyne-Carbonyl Coupling (Prins) Reactions



$Nu = \text{HOR}, \text{HNR}_2, \text{halide}, \text{or arene}$

Without a means to control stereoselectivity or *E/Z* geometry, the synthetic utility of these reactions is limited. The most well-developed versions feature intramolecular alkyne-carbonyl couplings, with a water or a halide acting to capture the intermediate cation **2**.⁴ Our lab has recently been engaged in the development of alkynyl Prins coupling methods: Prins terminated by halide,⁵ *aza*-Prins terminated by halide,⁶ and Prins annulation terminated with

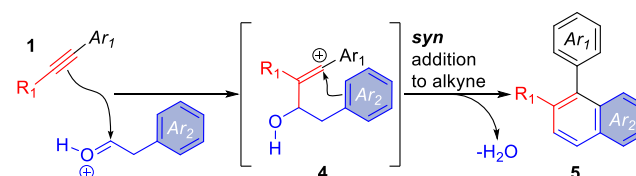
sulfonamide.⁷ More rarely, the vinyl cation intermediate can be trapped with an arene nucleophile, terminating the Prins coupling with C-C rather than C-heteroatom bond formation. Intermolecular C-C bond formation (cf. Scheme 1; $Nu = \text{arene}$) is only successful when the arene reactant is in large excess (as solvent).⁸

Examples of intramolecular arene capture are limited to those shown in Scheme 2. Stepwise [4+2] cycloaddition processes generate functionalized naphthalenes **5** (Scheme 2A),⁹ and cationic cascades developed by Hinkle et al. generate benzo[*f*]isochromenes **8** (Scheme 2B).¹⁰ These two types of carbocyclizations employ simple (phenyl) arenes as the terminating nucleophile and eliminate H_2O (via **4** and **7**) to generate a naphthalene moiety.

In this paper, we describe a carbocyclization cascade that generates fused polycyclic adducts **12** (Scheme 2C). Condensation of readily available building blocks **9** and **10** generates an oxocarbenium or iminium ion **11a**, aligned with an alkyne and an arene such that a formal cycloaddition occurs diastereoselectively. As indicated in Scheme 2C, the broad scope of the one-step process enables variation of ring size, at R_1, R_2, R_3 , and in the arene ring. The scaffolds represent a novel structure class, offering entry into new polycyclic chemical space.

Scheme 2. Cascade Prins Carbocyclizations

A. Stepwise [4+2] cycloaddition with **aromatization**



B. Prins coupling/ cascade carbocyclization with **aromatization**

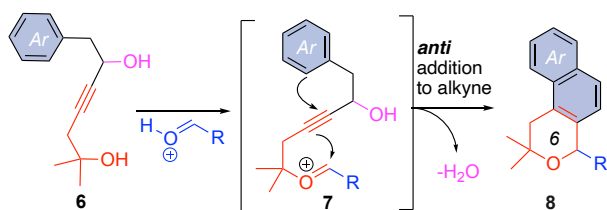
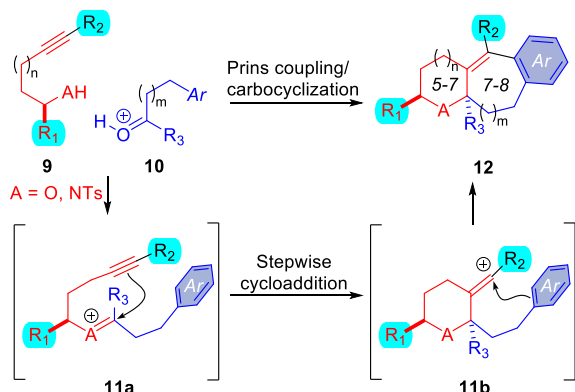
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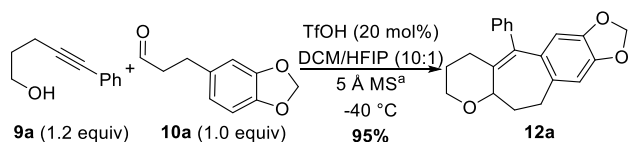
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C. Stepwise Coupling/ Cycloaddition Cascade (*syn* addition to alkyne)

Optimization of the Prins Cascade Carbocyclization

Table 1. Optimization of the Alkynyl Prins Carbocyclization



Entry	Deviations from Standard Conditions	Isolated % Yield
1	no HFIP added	40
2	10 mol% TfOH	87
3	20:1 DCM/HFIP	78
4	Tf ₂ NH (no TfOH)	90
5 ^b	no MS	48 ^c

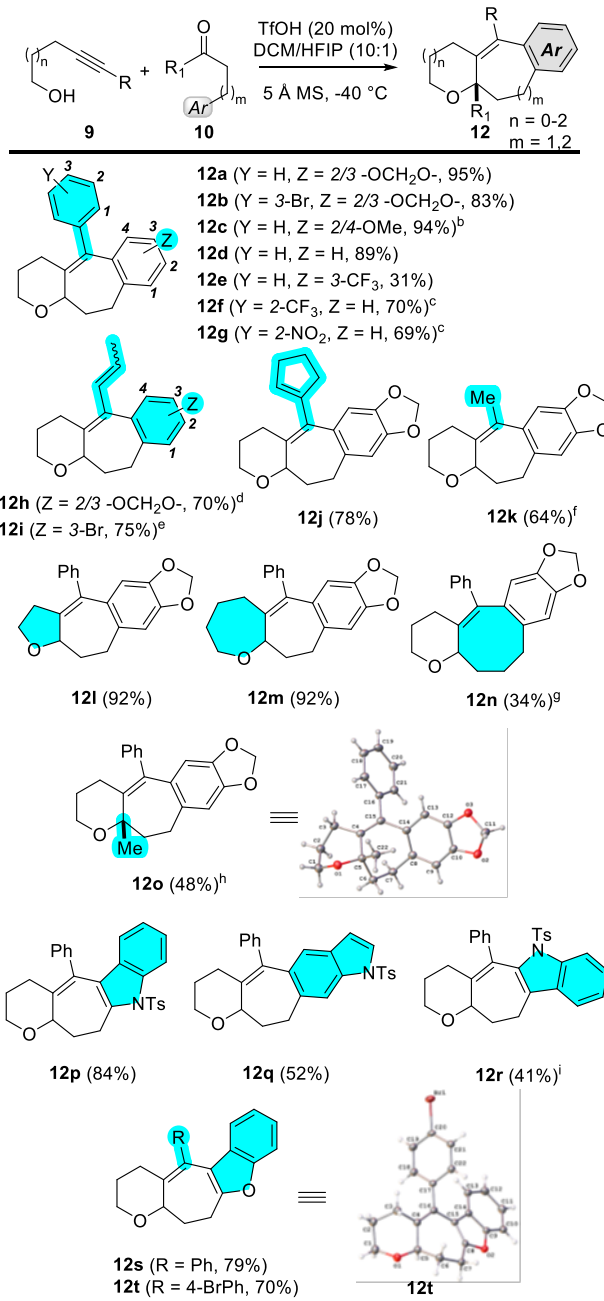
a) 300 mg of 5 Å MS used per 1.0 mmol of limiting reagent **10a** b) No MS added, warmed up to room temperature c) Some starting material recovered.

Optimization experiments for this reaction were done using arenynol **9a** and aldehyde **10a**, with an electron-rich nucleophilic tether (Table 1). Substoichiometric amounts of trifluoromethanesulfonic acid (TfOH) are sufficient to promote the cyclization to afford **12a**. This, in combination with 5 Å molecular sieves (MS) in a solvent mixture of dichloromethane (DCM) and 1,1,1,3,3,3-hexafluoroisopropanol-2-ol (HFIP) results in optimal yields (See Table 1). HFIP was chosen as cosolvent due to its demonstrated ability to stabilize carbocationic intermediates.¹¹ Omission of the HFIP gives the desired product **12a** in 40% yield, as well as unreacted aldehyde, acetal (resulting from condensation of the aldehyde with two molecules of arenynol; see SI for details), and other unidentifiable byproducts (Table 1, Entry 1). Lowering the TfOH loading to 10 mol% gives good yields for **12a** (Entry 2), however, since other substrates required more acid for full consumption of the starting material, we chose 20 mol% TfOH as part of our standard

conditions. Reduction of the amount of HFIP by half gives 78% yield of cyclized product **12a** (Entry 3). Using bis(trifluoromethane)sulfonamide (Tf₂NH) instead of TfOH also affords the desired product in good yields (Entry 4). Finally, exclusion of molecular sieves leads to sluggish reactions where the starting material is never fully consumed, even after warming to room temperature (Entry 5).

Substrate Scope

Scheme 3. Substrate Scope for Alkynyl Oxa-Prins Carbocyclization



a) Conditions: To a mixture of **9**, **10**, and molecular sieves in DCM/HFIP at -40 °C was added TfOH, dropwise. Stir at -40 °C until all starting material is consumed by TLC b) 1.7:1 *p/o*-trapped c) Warmed to 0 °C 1 hour after adding TfOH, then to room temperature d) 6.3:1 E/Z e) 7.9:1 E/Z f) 40 mol% TfOH added at -40 °C, then warmed to

room temperature after an hour **g**) 40 mol% TfOH added at -40°C , then to 0°C after an hour **h**) 1.5 equiv. of corresponding dimethyl ketal **10** ($R_1 = \text{Me}$) used with 1.0 equiv. of alcohol **9a i**) 1:1 mixture of conformers.

Using these optimized conditions, the alkyne and carbonyl scope was explored. Electron-rich, neutral, and deficient arenynes react cleanly to give fused products **12a-12g** in good to excellent yields (Scheme 3). Enyne alcohols work as well to afford products **12h-12j** in good yields. An internal alkyne engages to give product **12k** in 64% yield, although 40 mol% of TfOH and stirring at room temperature is required for the reaction to go to completion, otherwise the acetal of **10a** is isolated (See SI). Although electron deficient arenynes engage well in this chemistry, 4-cyanoarenynes (a more basic functional group) is not tolerated. Similarly, tests with a terminal alkyne, bromoalkyne, alkynylsilane, and diyne also result in the formation of complex mixtures (see SI).

Aldehyde **10** (with Ar = *m*-methoxyphenyl) reacts well to give cycloaddition product **12c** as a 1.7:1 mixture of the *para/ortho*-trapped product in 94% combined yield. Aldehyde **10** (with Ar = Ph) also traps well to give **12d** in 89% yield. Even an electron-deficient arene engages to give **12e**, albeit in 31% yield. Different tether lengths of the alcohol work without any negative impact on the yield, giving access to five-membered oxacycle **12l** in 92% yield, and seven-membered oxacycle **12m** in 82% yields. Eight-membered carbocycle formation is also feasible under the reaction conditions to give product **12n** via a formal [6+2] pathway, albeit in lower yields. Ketone **10** ($R_1 = \text{Me}$) also reacts to afford [5+2] adduct **12o**, although in only 20% yield. Using the dimethyl ketal of **10** ($R_1 = \text{Me}$) instead brings this yield up to 48% (Scheme 3).

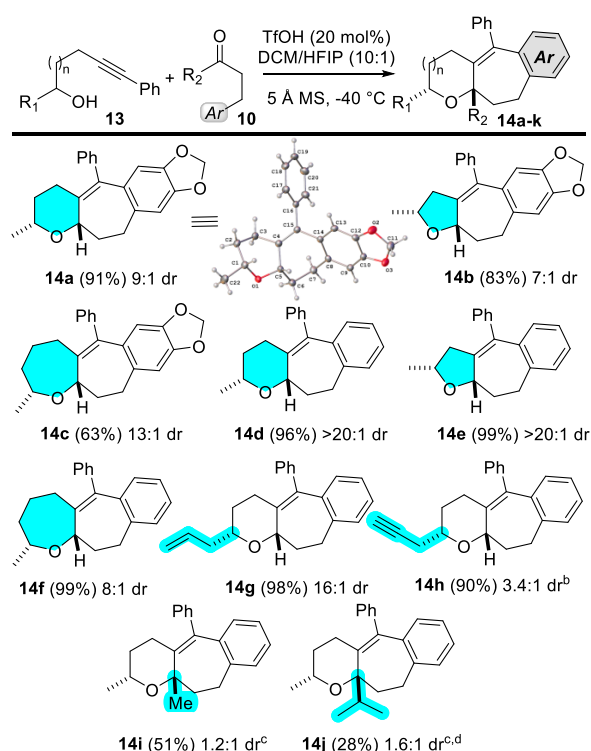
Heteroaromatic nucleophiles also react smoothly under these conditions. Tosyl indole aldehydes engage well to afford products **12p**, **12q**, and **12r** in 84, 52, and 41% yields, respectively. Product **12r** is obtained as a 1:1 mixture of conformers. We hypothesize that these conformers arise from steric interactions between the tosyl protecting group and the phenyl ring on the cycloheptene (Scheme 3). This hypothesis is supported by the fact that, upon detosylation of the indole, a single product is observed (See SI). Benzofuran aldehydes give **12s** (79% yield) and **12t** (70% yield). When *N*-methyl, or deprotected indoles are subjected to the reaction conditions, the aldehyde undergoes decomposition, and no Prins product is formed (See SI).¹²

The alkynyl Prins reaction of secondary alcohols with aldehydes is known to proceed diastereoselectively.^{4g} In a previous paper from our group, we demonstrated that the oxa-Prins reaction with an enantioenriched alcohol like **13** produces enantioenriched Prins adducts.⁷ Scheme 4 shows the diastereoselectivity of our Prins coupling/cycloaddition approach, using secondary alcohols **13** and aldehydes or ketals **10** ($R_2 = \text{Me}$, ^iPr). As evidenced with cases **14a** through **14f**, secondary alcohols **13** react well with aldehydes to give five-, six-, and seven-membered oxacycles in excellent yields and good diastereoselectivities. With a homoallylic alcohol, cycloaddition product **14g** is generated in excellent yield and 16:1 dr. The

diastereoselectivity is lower with a homopropargylic alcohol, which gives **14h** (3.4:1 dr).

The Prins carbocyclization of secondary alcohol **13a** and dimethyl ketal **10** ($R_2 = \text{Me}$) gives Prins product **14i** (51% yield), with virtually no diastereoselectivity. Replacing the methyl with a bulkier isopropyl group makes the reaction sluggish, delivering cycloaddition product **14j** in 28% yield as a 1.6:1 mixture of diastereomers.¹³

Scheme 4. Diastereoselective Prins Carbocyclizations with Secondary Alcohol Reactants



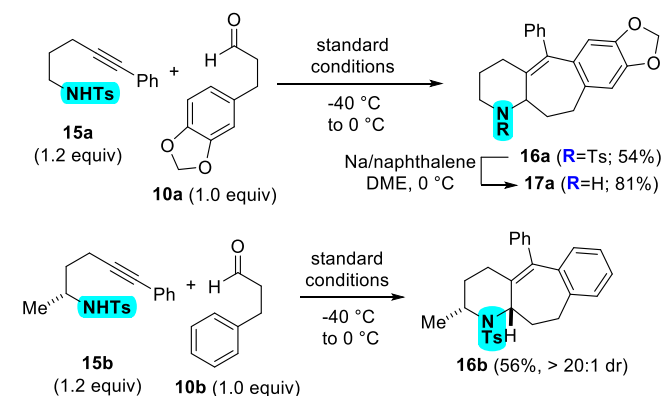
a) Conditions: To a mixture of **13**, **10**, and molecular sieves in DCM/HFIP at -40°C was added TfOH, dropwise. Stir at -40°C until all starting material is consumed by TLC. **b**) Reported dr was observed for 0.2 mmol scale reaction. At 1.0 mmol scale, dr went down to 2.8:1. **c**) Dimethyl ketal of corresponding ketone used **d**) Reaction warmed up to 0°C after 24 hrs until all the alcohol was consumed.

Synthesis of Linear-Fused *N*-Heterocyclic Systems

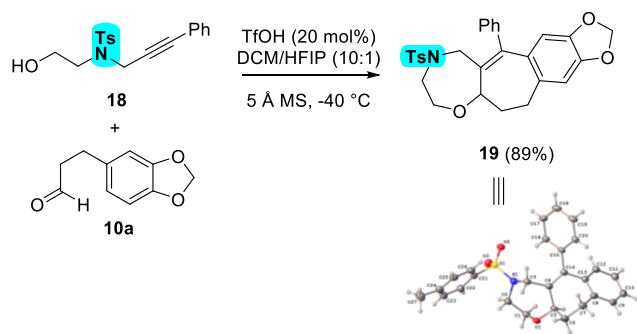
Furthermore, it was exciting to discover that the reaction conditions translate to the preparation of piperidines from sulfonamides **15** (Scheme 5). The reaction performs well for alpha-primary sulfonamide **15a** to give tosyl-protected piperidine **16a** in 54% yield. Starting with alpha-secondary sulfonamide **15b** leads to the diastereoselective formation of piperidine **16b** in 56% yield and >20:1 dr. The tosyl protecting group of **16a** can be removed using sodium naphthalenide to afford piperidine **17a** in 81% yield.

It is also possible to prepare 1,4-oxazepane **19** in 89% yield from alcohol **18**, using the standard reaction conditions (Scheme 6). A single crystal X-ray structure confirmed the identity of the cycloadduct.

Scheme 5. Carbocyclization for Synthesis of Fused Piperidines

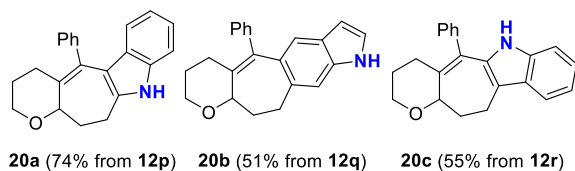


Scheme 6. Prins Carbocyclization for Synthesis of 1,4-oxazepanes



Finally, we focused on unmasking *N*-Ts indoles **12p**, **12q**, and **12r** to access the *N*-H indole motif, which is ubiquitous among bioactive natural products and drug targets.¹⁴ In these cases, detosylation occurred smoothly upon treatment with potassium *tert*-butoxide (3 equiv) in DMSO, providing access to the target scaffolds **20a-c** as shown (Scheme 7).¹⁵

Scheme 7. Deprotection of *N*-Ts Indole Substrates



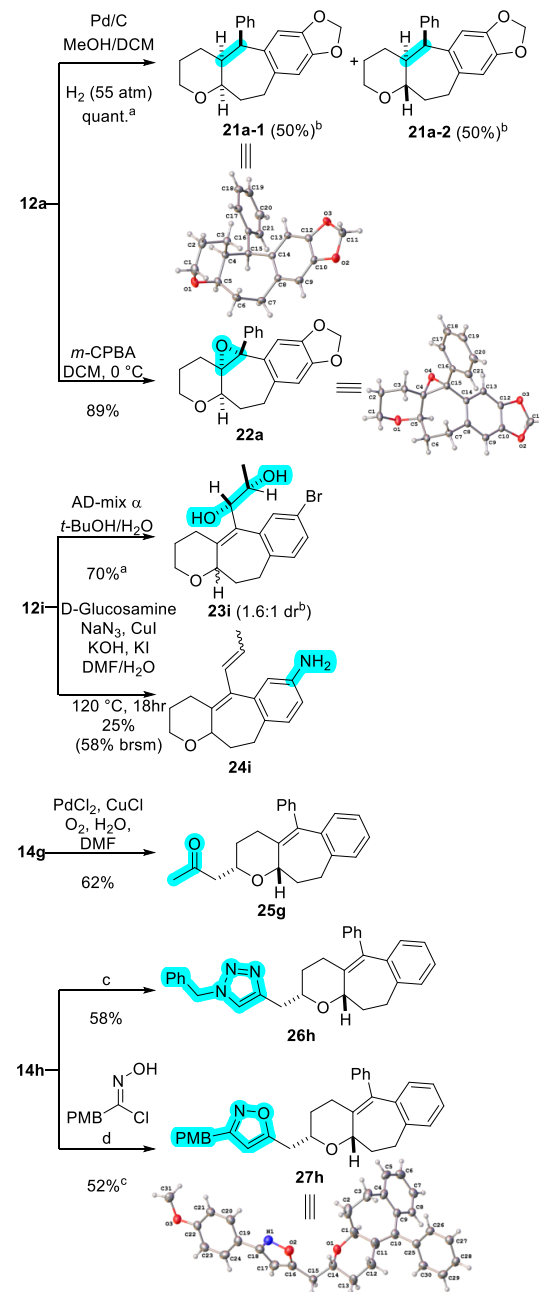
Diversification of Polycyclic Adducts

The synthetic versatility of products **12** and **14** can be exploited to give access to new chemical space. Scheme 8 shows examples of some chemical transformations that can be done to enhance the complexity of these scaffolds. Hydrogenation of the tetrasubstituted olefin in **12a** occurs using palladium on carbon (Pd/C) and gives quantitative yields of **21a** as a 1:1 mixture of diastereomers that can be separated by column chromatography. Hydrogenation of **14a** gives the corresponding hydrogenated product as a 5:1 mixture of inseparable diastereomers (See SI). Oxidation of **14a** with *m*-CPBA delivers epoxide **22a** in 89% yield, with no purification required.

The olefin in **12i** can be oxidized with AD-mix α to give diol **23i** as a 1.6:1 mixture of diastereomers that can be separated by column chromatography. The aryl bromide is a potential functional handle for subsequent cross-coupling operation. Aniline derivative **24i**, for example, was obtained in 25% yield (58% based on recovered starting material) after a copper-mediated amination.¹⁶

Subjection of cycloadduct **14g** (bearing a pendent alkene) to Wacker oxidation conditions gives ketone **25g** in good yields.¹⁷ Finally, applying [3+2] cycloaddition strategies to cycloadduct **14h** (bearing a pendent alkyne) enables efficient introduction of interesting heterocycles such as triazole **26h** and isoxazole **27h**.¹⁸

Scheme 8. Diversification of Cycloadducts



a) Combined yield of two diastereomers **b)** dr determined based on isolated yields of separable diastereoisomers **c)** BnN_3 , Cu powder, $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$, sodium ascorbate, $\text{H}_2\text{O}/t\text{-BuOH}$, rt. **d)** Cu powder, $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$, NaHCO_3 , $\text{H}_2\text{O}/t\text{-BuOH}$, rt.

Conclusions

In summary, alkynyl Prins coupling/carbocyclization sequences have been developed for the diastereoselective preparation of novel

polycyclic systems. The linear-fused ring arrays contain a saturated heterocyclic ring, a seven- or eight-membered carbocycle, and an aromatic or heteroaromatic ring. In the presence of substoichiometric amounts of acid, this method couples two simple building blocks (an aldehyde and an alkynyl alcohol) in a cascade sequence, forging three new bonds (one C-O; two C-C) and two new rings in one pot.

The polycyclic small molecules described here represent a new structure class.¹⁹ The multiple sites available for easy diversification should enable synthesis of derivatives that extend into new areas of chemical space. Thus, the method can offer access to a diverse library of compounds different from any that have been prepared and tested before. If the molecules exhibit biological activity, the simple synthetic procedure will allow for rapid preparation of analogs for follow-up studies. Ongoing work includes the exploration of an asymmetric carbocyclization, expanding the scope of the *aza*-Prins variant of the reaction, and exploring other means of trapping the vinyl cation intermediate.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

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- 12 Subjecting the aldehyde to the reaction conditions (in the absence of alcohol **9**) leads to decomposition of the aldehyde.
- 13 Stereochemical assignment in **14i** and **14j** is tentative, based on aldehyde behavior and x-ray structure for **14a**.
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