

A High-Pressure Brillouin and Raman Scattering Study on $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ Glass: Implications for Pressure-induced Shear Velocity Minima in Silicate Glasses

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Highlights

- High pressure elasticity of a soda-iron silicate glass is investigated to 12 GPa
- Elastic anomalies, including a negative pressure derivative of the shear modulus and an initially nearly pressure-independent Young's modulus, are observed on initial compression.
- Pressure-induced shear velocity minima in silicate glasses are demonstrated to vary systematically with the ratio of network-modifying to network-forming cations.

Abstract

Sound velocities of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass have been measured to 12 GPa by Brillouin spectroscopy. Poisson's ratio and bulk, shear, and Young's moduli are calculated as a function of pressure. The shear and Young's moduli and Poisson's ratio show a shift in the response to compression of the glass at ~ 2.2 GPa, where the pressure dependence of the shear modulus reverses sign. This shift mirrors those of a wide suite of glasses, and further demonstrates that pressure-induced shear velocity minima are general phenomena in silicate glasses containing few network modifiers. Raman spectra have also been collected

39 of the glass up to 6.5 GPa. A relation is proposed between the magnitude of shear velocity
40 depression observed under pressure in silicate glasses and the ratio of the number of
41 network-modifying cations and network-forming cations. This relation can prospectively
42 be deployed to compositionally tailor the pressure dependence of the elastic velocities of
43 silicate glasses.

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46 1. INTRODUCTION

47

48 The elastic constants of silicate glasses under pressure are important for understanding how
49 tetrahedrally-dominated amorphous oxides respond to compaction and differential stress,
50 and are hence a topic of interest within material sciences, condensed matter physics, and
51 the geosciences. For example, the high-pressure behavior of such glasses can be deployed
52 as a constraint on the response to compaction of silicate melts. A range of polymerized
53 silicate glasses have been shown to be elastically anomalous under compression¹⁻¹³.
54 Because of increased resistance to compaction and shear of the structural units, the elastic
55 moduli of most materials increase under pressure. However, a range of silicate glasses have
56 been shown to have elastic moduli (and particularly the shear moduli) that decrease up to
57 pressures of ~ 0.5-3 GPa. For example, both Brillouin and ultrasonic measurements show
58 that the bulk, longitudinal and shear moduli of silica glass soften and exhibit minima near
59 2 GPa [e.g., 1-4]. This anomalous behavior has been observed using either Brillouin
60 spectroscopy or ultrasonic measurements for a diverse suite of compositions, including
61 water-white glass [5], float glass [6], MgSiO₃- and CaMgSi₂O₆-glasses [7,10], a range of
62 alumina-bearing glasses [8,10,13], natural basaltic glasses [9,11] and silica-titania glasses
63 [12]. The connection between such anomalous sound velocity trends under pressure and
64 the structures and compositions of glasses has, however, remained unclear. Moreover,
65 whether the presence of abundant iron affects the anomalous pressure dependences of the
66 sound velocities of glasses is not well-constrained.

67 This study probes the elastic properties of an iron-bearing sodium silicate glass with 95 %
68 enrichment of iron in the ⁵⁷Fe isotope (Na₂⁵⁷FeSi₃O_{8.5}) under high pressure by Brillouin
69 scattering. This particular composition is of interest for a range of reasons. First, it contains
70 trivalent iron almost entirely in tetrahedral coordination [14], and hence it has a ratio of
71 non-bridging oxygens to tetrahedral cations of 0.25. As such, its structure is dominated by
72 network-forming cations, but contains a comparatively small quantity of non-bridging
73 oxygens, as well as network-modifying cations. It is accordingly a model glass for
74 examining systems with limited amounts of depolymerization. Second, it is a transition-
75 metal bearing analogue of sodium aluminosilicate glasses, the properties of which are
76 relevant for both ceramics and earth materials, and it is unclear whether the structural and
77 elastic changes that occur within an iron-bearing silicate glass differ from those in
78 aluminosilicate glasses under compression. Third, the structural and elastic properties of
79 iron-rich glasses are of interest because both naturally-occurring geologic melts and
80 synthetic glasses designed to isolate toxic and radioactive material often have substantial
81 iron contents within them [e.g., 15].

82

83 Brillouin scattering, the inelastic scattering of light induced by acoustic phonons, is
84 commonly deployed to determine sound velocity and elastic constants. This optical
85 technique requires no mechanical contact with the sample and allows experiments on
86 samples of order microns in dimensions, and hence is ideally suited to measure sound
87 velocities under extreme conditions in a diamond anvil cell. These velocity data were
88 initially measured to provide density constraints for high-pressure measurements of the
89 vibrational density of states (DOS) of this glass using nuclear inelastic scattering [14], with
90 a particular focus on the origin of the boson peak in this glass, and its similarity at high
91 pressures to the transverse acoustic singularity of the corresponding crystal. However, a
92 quantitative analysis of these elastic data has not been presented. Here, these data are
93 deployed in conjunction with previous elastic results on silicate glasses under pressure to
94 provide structural/compositional systematics for the anomalous low-pressure elastic
95 properties of polymerized silicate glasses. These data also show the capability of the
96 Brillouin scattering technique to determine the elastic constants for semi-transparent
97 samples using the platelet geometry.

98

99 2. EXPERIMENTAL DETAILS/METHODOLOGY

100

101 The ambient-pressure glass transition temperature of this composition is 723 (± 1) K [14],
102 which is close to that of binary sodium silicate glasses [16,17]. It has a density of 2.714
103 g/cm³, which is larger than those of binary sodium silicate glass and sodium aluminum
104 silicate glass due to the substitution of iron. Due to the presence of iron, the bulk glass
105 aliquot is opaque. After polishing to 30 μm thickness, it is semi-transparent with a dark
106 yellowish color. Permanent densification has also been observed for this glass at 1, 2, and
107 3 GPa at a temperature of 673 K, or $\sim 90\%$ of the glass transition temperature in this material
108 [18], with an emphasis on the vibrational density of states of these densified glasses.

109

110 The Brillouin scattering experiments were conducted by exciting the sample in a diamond
111 anvil cell (DAC) with 480 μm diameter culets and using the 514.5 nm green line of a
112 Spectra Physics Ar-ion laser (the output beam power is 130 mW, and in front of the DAC,
113 the beam power is 60 mW). The semi-transparent glass sample was 80 \times 70 \times 30 μm thick
114 and was mounted in a 170- μm diameter hole in a pre-indented stainless steel gasket of 40
115 μm thickness. Two pieces of ruby $\sim 6 \mu\text{m}$ in size were placed near to the two sides of the
116 sample for pressure calibration. Liquid argon was loaded as a pressure-transmitting
117 medium using a standard liquid nitrogen-based loading technique. The pressure was
118 measured using the shift in the R1 line of the rubies loaded along with the sample. Spectra
119 were recorded in the 70° platelet-scattering geometry with no polarization used for the
120 collected signal. In order to obtain the refractive index, Brillouin measurements were also
121 performed in the backscattering geometry at ambient pressure.

122

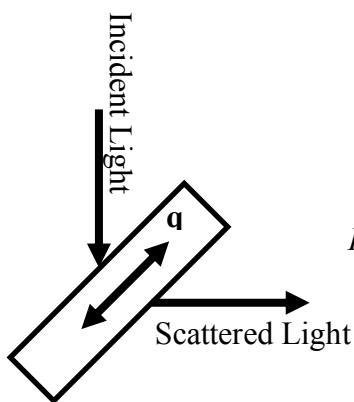
123 Since Doppler-shifted light photons scattered from thermally excited acoustic phonons
124 (sound waves) of the sample produce peaks in the Brillouin spectrum, the sound velocity
125 (V) is expressed as:

126
$$V = \frac{\lambda \nu}{2n \sin(\theta/2)}$$
 (1)
127

128 where ν is the Brillouin frequency shift, λ is the wavelength of the laser, n is the refractive
129 index of the sample, and θ is the scattering angle. We deploy both the back-scattering ($\theta=180^\circ$)
130 and platelet geometries. Whereas the classical backscattering geometry enables only
131 measurements of nV_P (the product of refractive index and longitudinal velocity), the
132 platelet geometry (Figure 1) allows us to measure the sound velocities (V_P and V_S) directly
133 and independently without knowledge of the refractive index according to the relation,

134
$$V = \frac{\lambda \nu}{2 \sin \Theta}$$
 (2)
135

136 where Θ is the angle between the incident light and the normal to the surface of the sample.
137 In this study, the angle is typically chosen as 35° .



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144 *Figure 1. Platelet geometry for Brillouin scattering.*
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148 The two scattering geometries also provide the ability to determine the refractive index n
149 of the glasses directly. Since the velocity of an acoustic phonon in an anisotropic medium
150 is independent of direction, the ratio of the Brillouin shifts, as expressed as the Brillouin
151 shifts in the platelet geometry divided by the Brillouin shifts in the back scattering
152 geometry, is equal to $n/\sin\Theta$. By taking the ratio of the Brillouin shifts at different pressure
153 points, the refractive index n of the glass can be uniquely determined both at ambient
154 conditions, and as a function of pressure. At ambient pressure, the index of refraction of
155 this glass is determined to be 1.627 (± 0.001), which is comparable to those of other
156 transition metal-bearing sodium silicate glasses [19].

157

158 Raman spectra were collected both at ambient and high pressures, and following quenching
159 from high pressures, of this glass. The Raman spectra were recorded using a triple pass
160 Dilor XY spectrometer equipped with a liquid nitrogen-cooled charge-coupled-device

161 (CCD) detector. The 514.5 nm green line of an Ar ion laser was used to excite the sample.
162 The laser light was focused with a Mitutoyo long distance 50 \times objective to a spot of about
163 2 μm diameter in the sample with the laser power being 30 mW on the sample. The
164 spectrometer was calibrated using single-crystal silicon as a reference.

165

166 For the high-pressure Raman measurements in the diamond anvil cell, a finely polished
167 glass fragment 50 \times 40 \times 20 μm in size was mounted in a rhenium gasket in a diamond-
168 anvil cell equipped with 350 μm diameter culets. Two pieces of ruby \sim 5 μm in size were
169 placed on the two sides of the sample for pressure calibration. As with the Brillouin
170 measurements, liquid argon was loaded as a pressure-transmitting medium, and pressures
171 were estimated using the shift in the R1 line of the rubies loaded along with the sample.
172 All spectra were recorded using an unpolarized backscattering geometry.

173 The density of the glass at high pressures is obtained using the velocity data in Table I
174 from:

175

$$176 \rho_p - \rho_o = \int_{P_o}^P \frac{\gamma}{V_B^2} dP \quad (3)$$

177

178

179 Here, ρ_p and ρ_o are the density of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass at pressure P and ambient pressure
180 P_o , respectively, and V_B is the bulk sound velocity of the glass. γ is the ratio of the specific
181 heat at constant volume and pressure (C_p/C_v , which we presume to be 1, in accord with
182 Richet and Bottinga [20], who have shown this parameter to be within 0.2% of unity at 300
183 K for alkali silicate glasses). This latter parameter is present to convert the isentropic
184 velocity/moduli determined in the Brillouin experiment to isothermal conditions. For
185 elastically isotropic materials,

186

$$187 V_B^2 = V_p^2 - 4V_s^2 / 3 \quad (4)$$

188 where V_p and V_s are the longitudinal and transverse sound velocities. The calculated
189 densities as a function of pressure are used to extract the high pressure elastic moduli from
190 the velocity data using $\mu_p = \rho_p V_s^2$ and $K_p = \rho_p V_B^2$, where μ_p is the shear modulus at
191 pressure and K_p is the bulk modulus at pressure.

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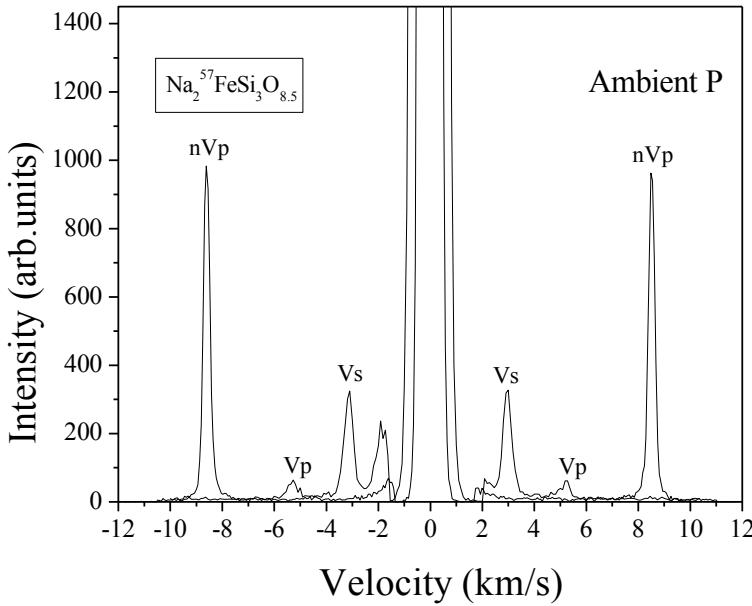
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199 **3. RESULTS AND DISCUSSION**

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203 *Figure 2.* Brillouin spectrum of $\text{Na}_2^{57}\text{FeSi}_3\text{O}_{8.5}$ glass at ambient pressure measured
204 in both platelet geometry and backscattering geometry (the spectrum labeled with
205 nV_p).

206

207 Figure 2 shows representative Brillouin spectra at ambient pressure in both platelet and
208 backscattering geometry: clearly, the amplitude and character of the peaks is sufficient to
209 accurately determine the three parameters of interest: the compressional and shear velocity,
210 and the index of refraction. Importantly, as the peak locations are symmetric across the
211 incident laser line (corresponding to both Stokes and anti-Stokes scattered light), there is
212 redundancy in determining the location of relatively weak peaks, such as the V_p -associated
213 band.

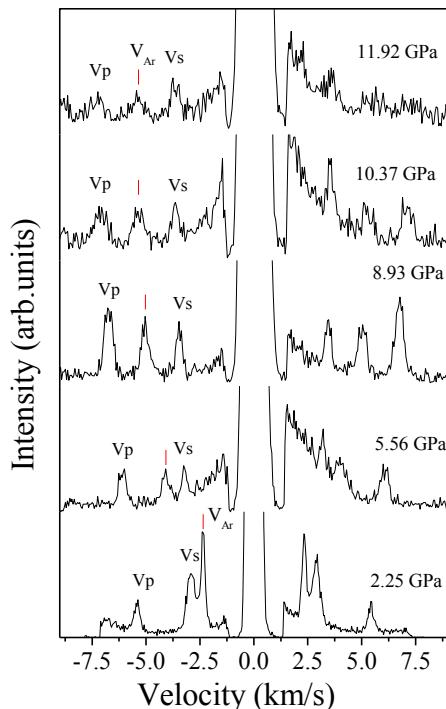
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215 High pressure Brillouin spectra are shown in Figure 3 in the platelet geometry, with the
216 resultant velocities at ambient and high pressure given in Table I. Our ambient pressure
217 velocity determinations agree well with a previous determination [18] Figure 3 shows that
218 the Brillouin peaks associated with both the shear and compressional peaks are well-
219 resolved: the peak from the argon pressure medium crosses over the shear peak near 3 GPa,
220 but it otherwise does not affect the measurement of the sound velocities in the glass. Figure
221 4 shows the shifts in compressional and shear velocities with pressure. The shear velocity
222 initially decreases with pressure up to a minimum value at ~ 2.2 GPa; at higher pressures,
223 it has a positive pressure dependence. Correspondingly, the compressional wave velocity

224 has a small but positive initial pressure shift, which markedly increases above ~ 2.2 GPa.
225 Such anomalous negative (or small, for V_p) shifts are relatively commonly observed within
226 silicate glasses on initial compression [1-13]. The depression in shear velocity in this iron
227 silicate glass is quite close to that observed within a float glass measured using the same
228 techniques and apparatus under compression [6], indicating that two glasses with markedly
229 different compositions, and differing degrees of polymerization, have similar shear
230 velocity behavior under compression.

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Figure 3. Brillouin spectra of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass at various pressures measured in
235 platelet geometry. Red markers denote peaks from Brillouin scattering of the argon
236 pressure medium.

237

238

239 *TABLE I.* Sound velocities of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass at high pressure determined by Brillouin
 240 scattering (upper values are from the first run, lower values from the second run). Errors
 241 are given in parentheses.

242

Pressure (GPa)	V_p (km/s)	V_s (km/s)
1.00E-04	5.261 (16)	3.050 (2)
2.25	5.41 (2)	2.930 (8)
4.04	5.801 (8)	3.11 (1)
5.56	6.07 (1)	3.207 (10)
7.11	6.30 (1)	3.32 (3)
8.93	6.745 (9)	3.45 (1)
10.37	6.96 (2)	3.55 (1)
11.92	7.18 (6)	3.65 (3)
1.16	5.28 (3)	3.012 (9)
1.7	5.34 (2)	2.967 (7)
2.3	5.410 (9)	2.926 (4)
3.2	5.68 (1)	3.013 (7)
4.2	5.92 (2)	3.08 (3)
5.2	5.99 (3)	3.23 (2)
6.8	6.36 (3)	3.32 (2)

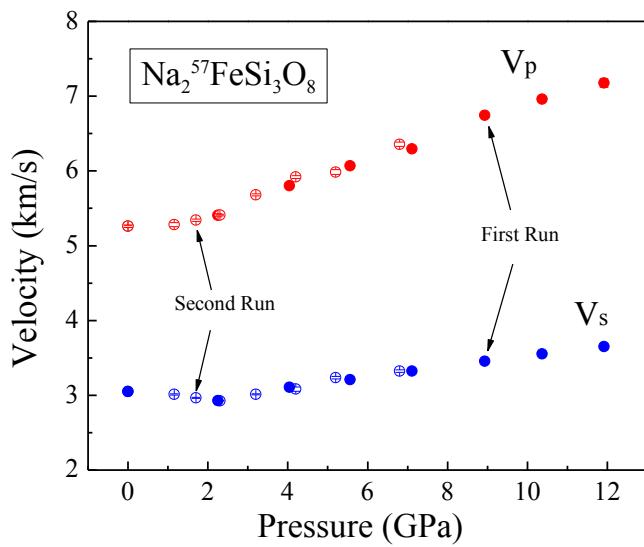
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249 *Figure 4.* Sound velocity of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass at various pressures determined by
250 Brillouin scattering (First and second run). Errors are smaller than the symbols, and
251 are illustrated for the second run (open symbols).

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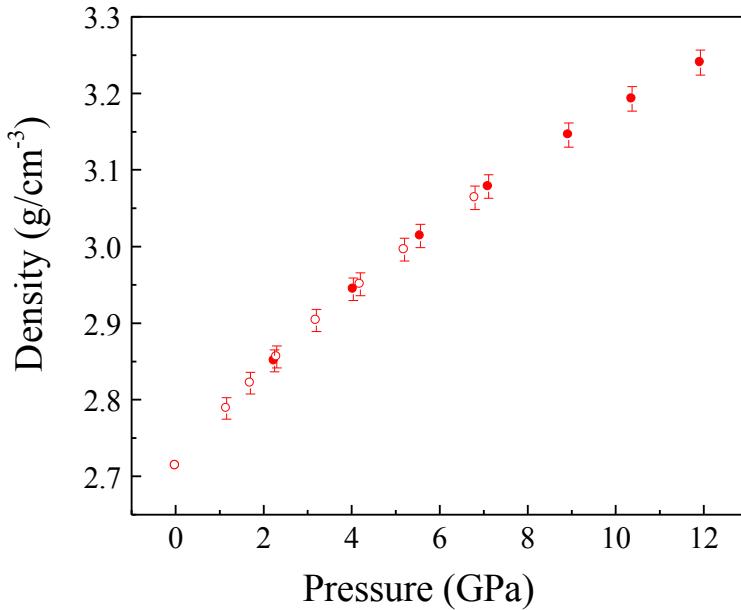
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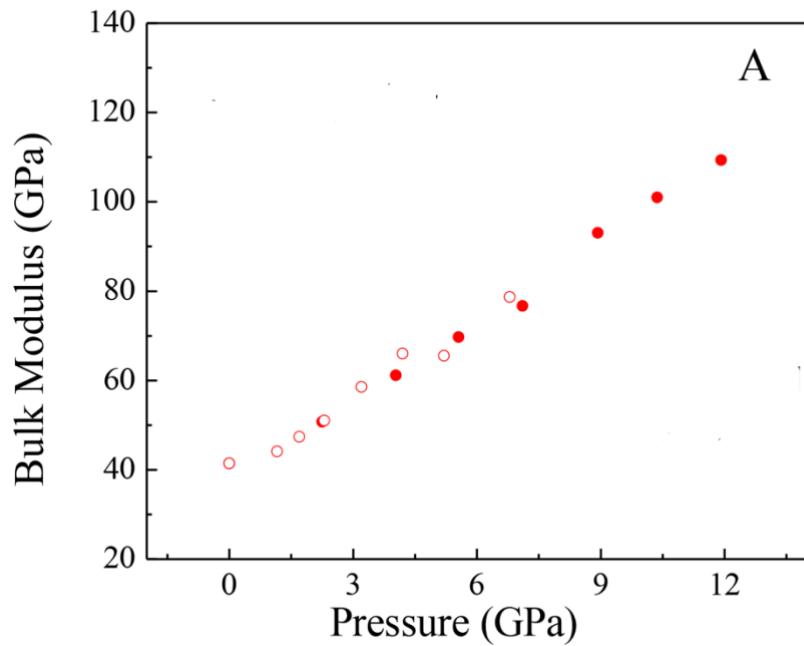
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262 *Figure 5.* Density of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass at various pressures determined by Brillouin
263 scattering in conjunction with Eqn. 3 (solid and open symbols are first and second
264 run, respectively).

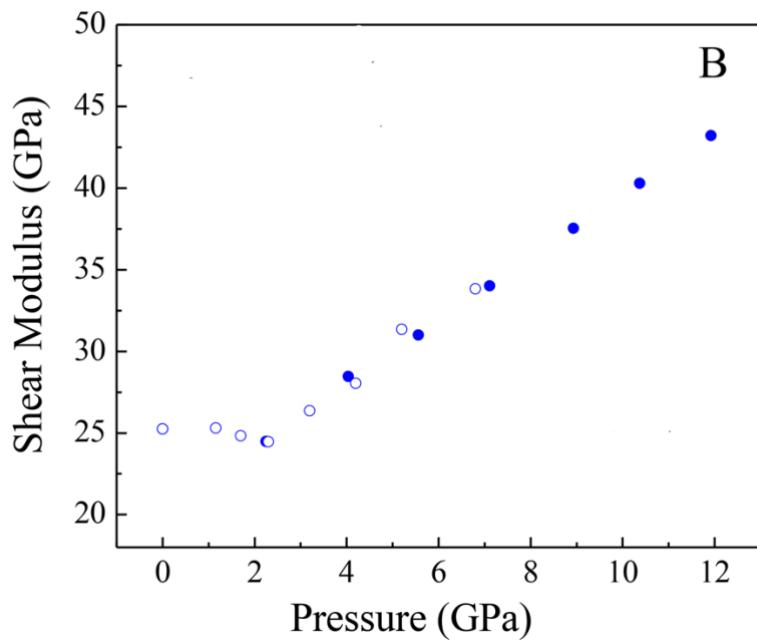
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266 Figure 5 shows the calculated densities of this silicate glass under compression: the
267 magnitude of elastic and irreversible densification (~17%) is somewhat less than that of
268 end-member silica (~23%) [e.g., 2] over the pressure range to 12 GPa, reflecting the
269 slightly higher bulk modulus of this glass relative to pure silica. Figure 6 shows the
270 variation in bulk modulus and shear modulus with pressure. As indicated by the trend in
271 the compressional wave velocity, the bulk modulus appears to initially shift shallowly
272 (Figure 6a), with the trend above ~1.5 GPa being monotonic and nearly linear. Within this
273 higher pressure interval, the derivative of the bulk modulus (dK/dP) with respect to
274 pressure is slightly in excess of 6: this linearity, and comparatively normal value of dK/dP ,
275 indicates that pervasive coordination changes of the silicon cations are unlikely to initiate
276 in this material up to our peak pressure of 12 GPa. The onset, and progressive occurrence,
277 of such coordination changes in glasses has been associated with a markedly enhanced
278 pressure derivative of the bulk modulus [7]. Figure 6b shows that a pronounced minimum
279 in the shear modulus occurs near 2.25 GPa, with higher pressure results again showing a
280 nearly linear increase with compression.

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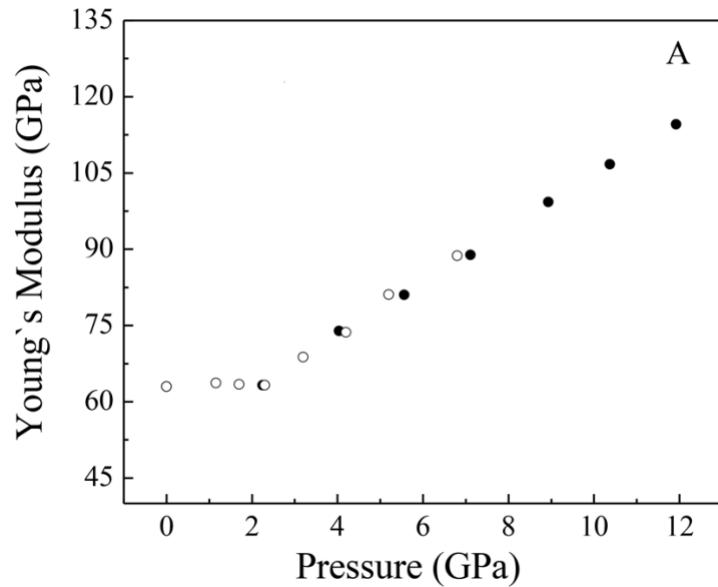
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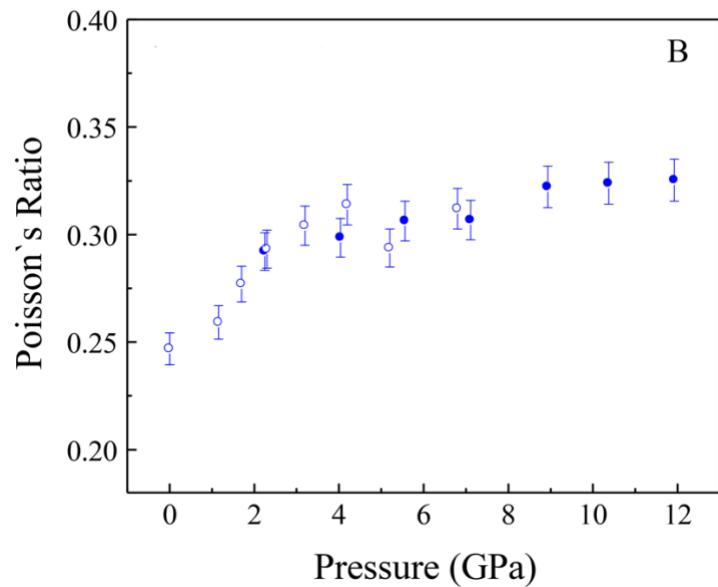
285 *Figure 6.* (A) Bulk moduli of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass at various pressures determined by
286 Brillouin scattering (First and second run). (B) Shear moduli of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass
287 at various pressures determined by Brillouin scattering (first and second run are solid
288 and open symbols, respectively). Error bars are comparable to, or smaller than, the
289 size of the symbols.

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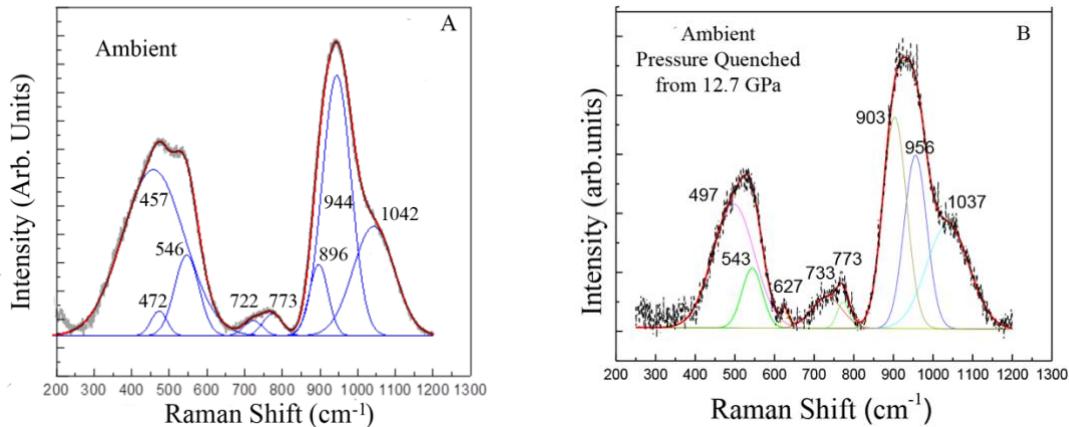
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297 *Figure 7. (A) Young's moduli of Na₂FeSi₃O_{8.5} glass at various pressures determined*
298 *by Brillouin scattering. Error bars are smaller than, or comparable to, the size of the*
299 *symbols. (B) Poisson's ratio of Na₂FeSi₃O_{8.5} glass at high pressures determined by*
300 *Brillouin scattering (closed and open symbols are for the first and second run).*

301

302 This anomalous behavior of the shear modulus is directly reflected in the Young's modulus
 303 and Poisson's ratio of this glass (Figure 7a, b). Young's modulus is nearly constant up to
 304 2.25 GPa (Figure 7a), while Poisson's ratio increases markedly from ~0.25 to ~0.3 in this
 305 pressure range before modestly increasing to the highest pressures of these measurements.
 306 Poisson's ratio is of particular interest, as a general inverse correlation has been drawn
 307 between its value and the 'free volume' fraction in glasses, as manifested by the atomic
 308 packing density [21]. Thus, there is a possible conceptual relationship between the negative
 309 pressure dependence of the shear modulus, the associated comparatively large initial
 310 pressure shift of Poisson's ratio, and an anomalous shift in packing of the glass on its initial
 311 compression.

312



313

314 *Figure 8.* (A) Raman spectrum of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass at ambient pressure, prior to
 315 compression. (B) Raman spectrum of the pressure-quenched sample from the
 316 higher pressure Brillouin experiment.

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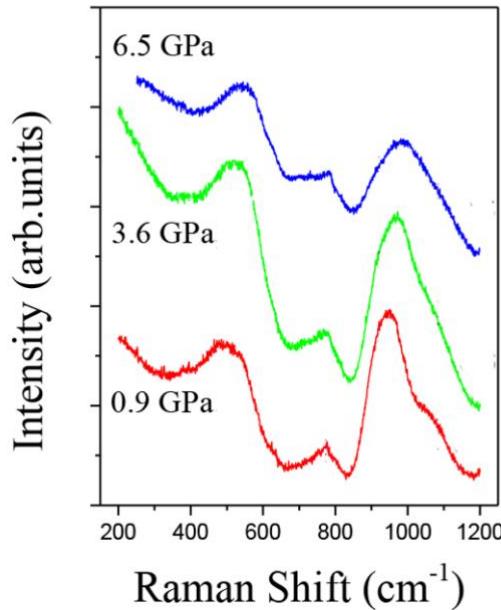
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320 The degree to which the observed changes in elastic properties reflect the elastic
 321 characteristics of the ambient glass, as opposed to irreversible densification, may be
 322 assessed from complementary Raman data on this glass. Figures 8a and 8b show Raman
 323 spectra of the glass before compression, and after compression to 12.7 GPa. Based on the
 324 differing polarization of bands, Wang et al. [22] attributed the bands at 457 and 944 cm^{-1}
 325 to the symmetric stretch of oxygens bridging between fully polymerized tetrahedra, and
 326 the asymmetric stretch of those oxygens, respectively. The shoulders at 549 and 1067 cm^{-1}
 327 are assigned to vibrations of bridging oxygens and non-bridging oxygens associated with
 328 Q^3 species (tetrahedra with three bridging and one non-bridging oxygen). Among the
 329 higher frequency peaks, there has also been a component of an FeO_4 tetrahedral stretching
 330 vibration proposed to be present [e.g., 23]. Following compression to 12.7 GPa, the peak
 331 initially at 457 cm^{-1} has migrated irreversibly to higher frequency: a result compatible with
 332 narrowed T-O-T angles in the decompressed glass [22]. *In situ* Raman spectroscopy of the

333 glass at high pressures (Figure 9) demonstrate that these irreversible changes in
334 morphology of the low frequency peaks do not occur up to at least 6.5 GPa, and hence their
335 onset must lie between 6.5 and 12.7 GPa. Therefore, our observations of the velocity
336 minimum in this material are associated with purely elastic behavior of the glass, and are
337 not expected to be affected by irreversible compaction. Importantly, this loose constraint
338 on the onset of irreversible behavior is compatible with the onset of irreversible compaction
339 in silica glass, which occurs near 8.6 GPa at 300 K [24].

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341

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Figure 9. In situ Raman spectra of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass at high pressures.

343

344

345 A key question emerges from the elastic data: what structural or chemical effects may
346 produce the size of the minima in shear velocity and modulus shown in Figures 4 and 6b,
347 and which are reflected in the dramatic increase in Poisson's ratio in Figure 7b? Such
348 minima are well-known within silicate glasses under pressure and, as in this study, are
349 more often observed in shear velocities than in compressional velocities [2,3,5-7,9,14].
350 Indeed, the magnitude of the shear velocity depression directly reflects the anomalous
351 elasticity of these glasses. Perhaps the most diagnostic experiments to date on the origin of
352 these anomalies have involved the observation that these minima are suppressed in silica
353 glass in a helium medium [25-27]. The mechanism for this suppression has been attributed
354 to He entering into large sites/voids within the structure, and preventing their pressure-
355 induced collapse [25,27]. Our interpretation of these results (which involve incorporation
356 of He at the 1 He per Si atom level or greater) modestly differs: He is well-known to enter
357 large cavities asymmetrically, being attracted to cations [28]. And, the prospective role of
358 He in restricting the Si-O-Si bond angle from bending has long been appreciated from
359 modeling [29]. Therefore, our interpretation of the elastic minima involves non-networking

360 forming atoms (or ions) within the structure playing a pivotal role in restricting the
361 flexibility of Si-O-Si(Al) angles in silicate glasses, and hence narrowing the broad Si-O-Si
362 potential minimum that produces the initial large degree of flexibility of the network (and
363 hence the velocity softening). Subsidiary support for the interpretation that the flexibility
364 of tetrahedral networks is critical in generating these elastic anomalies is derived from the
365 observed negative shear velocity gradients observed in non-silica-containing calcium
366 aluminate glasses [30]: the general phenomenon of anomalous elasticity hence hinges on
367 the presence of a tetrahedral framework, rather than being solely confined to silica-
368 dominated networks.

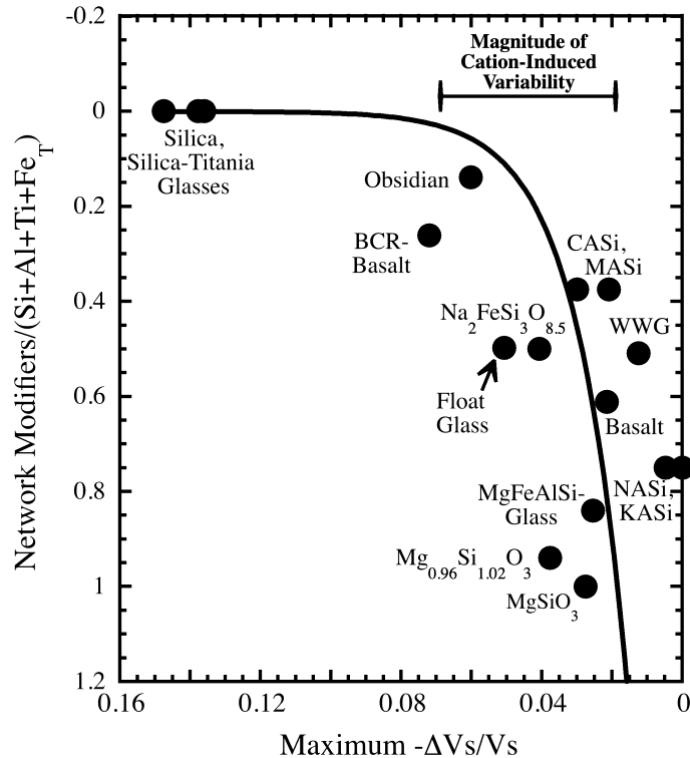
369 The absence of, or reduced amplitude of, a pressure-induced velocity minimum has been
370 correlated with greater degrees of compositionally-induced depolymerization of silicate
371 glasses [8,10,11]. However, the observation that the current $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ composition
372 glass has a maximum shear velocity depression under pressure similar to that of a less
373 polymerized float glass [6] raises the possibility that depolymerization may not be the sole
374 feature that drives these velocity depressions. Indeed, these silicate glasses have NBO/T
375 ratios of 0.25 and 0.7, respectively, and their maximum depressions in shear velocity are -
376 4.1 and -5.1% (which is the opposite trend from that expected from the depolymerization-
377 smaller velocity minima trend). However, the ratio of their number of network-modifying
378 cations to their network-forming cations is essentially identical. Accordingly, we
379 hypothesize that the introduction of network-modifying cations into the glass reduces the
380 flexibility of a subset of the polymerized linkages in the glass. In this regard, it is critical
381 to note that atoms/ions need not lie within the apex of the Si-O-Si(Al) bond angles whose
382 flexibility they impact: even atoms that neighbor the tetrahedra on the O_3 side of the $\text{O}_3\text{Si}-$
383 $\text{O}-\text{SiO}_3$ linkage (or, juxtaposed with non-bridging oxygens) dramatically impact the width
384 of the Si-O-Si potential minimum [29].

385 Figure 10 shows the maximum percentage change in shear velocity depression under
386 pressure within a broad range of non-densified silicate glasses. Clearly, relatively modest
387 amounts of network-forming cations dramatically decrease the amount of maximum shear
388 velocity depression, but a modest velocity anomaly persists even when relatively large
389 amounts of network modifiers are incorporated into the glass. Notably, our interpretation
390 of the structural effect that suppresses the elastic minima differs subtly but significantly
391 from past interpretations: rather than viewing the suppression as an effect of
392 depolymerization, we propose that the suppression is a consequence of interactions
393 between network-modifying cations and the polymerized structures within the glasses. Our
394 analysis implicitly treats the principal effect of network-modifying cations as occupying
395 free volume within the glasses, and hence restricting the deformability of the surrounding
396 polymerized framework. The role of free volume (and its availability/occupancy) is
397 supported by the observation that pressure-densified silica glasses, and particularly those
398 densified at simultaneous high-pressure and -temperature conditions, also have the
399 anomalous elastic effect suppressed [31,32]. A complementary view of the trend in Figure
400 10 is that it reflects a continuum between polymerized systems, with 'floppy' connections
401 between comparatively rigid silica tetrahedra [33], to systems that involve higher average
402 coordination numbers. The higher degrees of connectivity associated with more network-
403 modifying cations [e.g., 34, 35] may lead to progressively less anomalous rigidities as more
404 highly-coordinated cations are incorporated into the glass. This continuum is likely also
405 relevant to tendencies in thermal expansions as well: silicate glasses with larger proportions

406 of network-modifying cations have larger thermal expansions than the low thermal
407 expansion associated with fully polymerized silica, or silica-titania, glasses [36-39]. This
408 interpretation of velocity depression as being induced by a trade-off between connectivity
409 and floppy connections may also illuminate the role of irreversible densification of glasses
410 in eliminating these velocity minima within glasses with notably different degrees of
411 polymerization [2,4,7,8]. For densified glasses, it is possible that 'floppy' connections are
412 suppressed by the narrowed T-O-T angles within the irreversibly densified glasses. In this
413 perspective, irreversible densification may alter not only the bond angles between rigid
414 polymerized units, but also the interactions between the network-modifying cations and
415 the rigid units. Indeed, a shift in the structural role of network-modifying cations in the
416 pressure range in which irreversible densification occurs has been observed in MgSiO_3 and
417 CaSiO_3 glasses [40].

418 One notable aspect of Figure 10 is that it incorporates data on a wide suite of compositions
419 of glasses (spanning from natural alkaline earth-dominated aluminous silicate
420 compositions to multi-component synthetic alkali-rich compositions with silica as the sole
421 network former), and a number of separate high-pressure investigations with variable
422 densities of data sampling (for which coarser data sampling would cause a potential
423 underestimate in the maximum value of velocity depression). As such, despite the broad
424 range of compositions and studies, a gross trend of decreasing velocity anomaly with the
425 relative amount of non-network forming cations is observed. Clearly, the identity of
426 substituted cations also play a role in the magnitude of velocity depression: large alkali
427 cations (such as K) clearly induce smaller velocity depressions than divalent cations [13];
428 yet, this effect appears to be smaller than the normalized number of network modifiers
429 present within the glass. Thus, Figure 10 provides guidelines for, when advantageous,
430 tuning the composition of glasses to produce differing degrees of pressure-induced shear-
431 velocity depression. As an aside, although all the glasses in Figure 10 show decreased
432 compressional velocity slopes at low pressure, only silica, obsidian, and the two chemically
433 complex basalts show clearly negative initial V_p slopes, implying that either complex
434 natural compositions and/or the presence of Al is critical in producing negative shifts in
435 the compressional velocity.

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440 *Figure 10.* Dependence of the maximum amount of shear velocity depression
 441 (normalized to the ambient pressure shear velocity) observed in a range of silicate
 442 glasses under compression on the number of network modifying cations (Mg, Ca,
 443 Na, K) relative to network formers (Si, Al, Fe³⁺) within the glass. Data for silica [2],
 444 silica-titania glasses [12] (2 compositions), obsidian [3], water-white glass (WWG)
 445 [5], a float glass [6], basalt [9], MgSiO₃ [9], and Mg_{0.79}Fe_{0.1}Al_{0.1}Si_{0.96}O₃ [9]
 446 compositions, an oxidized Columbia River (BCR) basalt [11], a Mg_{0.96}Si_{1.02}O₃
 447 composition [7], Mg₃Al₂Si₆O₁₈, Ca₃Al₂Si₆O₁₈, Na₃AlSi₃O₉, K₃AlSi₃O₉ compositions
 448 [13] (MASi, CASi, NASi and KASi) and Na₂FeSi₃O_{8.5} composition (this study) are
 449 shown. The results of [10] are excluded because they did not measure ambient
 450 pressure velocities of their samples, with their measurements initiating between 0.8
 451 and 1.4 GPa. Where present, the oxidized Fe₂O₃ component is treated as a network
 452 former in all cases. The line is an exponential fit to the data and is intended to guide
 453 the eye. The first-order trend is associated with the network-modifier/network-
 454 former ratio, while the horizontal variability in the plot appears to be associated with
 455 the effect of different cations on the depression of the shear velocity [13]: the zero-
 456 velocity depression of K₃AlSi₃O₉ (KAS) relative to the depressions of the Mg-
 457 silicate glasses illustrates this effect. The range at the top illustrates the approximate
 458 magnitude of this cation-induced effect.
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463 **4. CONCLUSIONS**
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465 The elastic properties of $\text{Na}_2\text{FeSi}_3\text{O}_{8.5}$ glass have been measured to pressures of 12 GPa
466 using Brillouin spectroscopy: our ambient pressure results incorporate both
467 measurements in platelet and backscattered geometry, while our high-pressure results
468 deploy the platelet geometry. The results demonstrate that there is a substantial decrease
469 in shear velocity and modulus that reaches a minimum at ~ 2.2 GPa. The corresponding
470 compressional velocity and bulk modulus (as well as the Young's modulus) show
471 anomalously low positive increases up to this pressure. Hence, abundant trivalent iron (as
472 a network former) within this glass does not appear to markedly affect the anomalous
473 elastic behavior of silicate glasses under compression. The systematics of the elasticity of
474 a wide range of silicate glasses under pressure indicate that such anomalous decreases in
475 the shear velocity and modulus are correlated primarily with the ratio of the number of
476 network-modifiers to network-forming cations, but with substantial effects associated
477 with the identity of network modifying cations. This general correlation likely arises from
478 the effect of network-modifying cations in restricting the flexibility of the polymerized
479 $\text{Si}-\text{O}-\text{Si}(\text{Al})$ bond angles within the amorphous structures. Prospectively, this correlation
480 can be deployed as a guide for determining the likely magnitude of anomalous elastic
481 behavior within compositionally variable silicate glasses.

482
483 **ACKNOWLEDGMENTS**
484

485 The glass sample used in this study was kindly provided by Dr. A. Chumakov. This work
486 was supported by the U.S. National Science Foundation (Grants EAR-0074285 and EAR-
487 2017294). We are grateful to the late John Balogh for maintaining the spectroscopic
488 systems at the University of Hawaii, where these measurements were conducted, and to
489 three anonymous reviewers for constructive comments.

490
491 **Author Contributions**

492 **Anwar Hushur:** Formal analysis (lead); Investigation (lead); Writing-original draft
493 (equal). **Murli Manghnani:** Formal Analysis (equal); Conceptualization (lead);
494 Investigation (equal); Resources (lead); Funding Acquisition (equal). **Quentin Williams:**
495 Formal Analysis (equal); Resources (equal); Funding acquisition (equal); Writing-
496 original draft (equal).

497
498 **REFERENCES**
499

500 1) A. Polian and M. Grimsditch, Room temperature densification of a-SiO_2 versus
501 pressure, *Phys. Rev. B*, 47, 13979-13982, 1993.
502 <https://doi.org/10.1103/PhysRevB.47.13979>

503
504 2) C.S. Zha, R.J. Hemley, H.K. Mao, T.S. Duffy and C. Meade, Acoustic velocities and
505 refractive-index of SiO_2 glass to 57.5 GPa by Brillouin scattering, *Phys. Rev. B*, 50,
506 13105-13112, 1994. <https://doi.org/10.1103/PhysRevB.50.13105>

508
509 3) K. Suito, M. Miyoshi, T. Sasakura, and H. Fujisawa, in *High-Pressure*
510 *Research: Application to Earth and Planetary Sciences*, edited by Y. Syono and M. H.
511 Manghnani (TERRAPUB, Tokyo, 1992), pp. 219–225.
512
513 4) A. Yokoyama, M. Matsui, Y. Higo, Y. Kono, T. Irifune and K.-I. Funakoshi, Elastic
514 wave velocities of silica glass at high temperature and high pressures, *J. Appl. Phys.* 107,
515 123530, 2010. <https://doi.org/10.1063/1.3452382>
516
517 5) J.-A. Xu, Brillouin scattering and ultrasonic studies at high temperature and high
518 pressure, *Chem. Geol.*, 128, 17-24, 1996. [https://doi.org/10.1016/0009-2541\(95\)00160-3](https://doi.org/10.1016/0009-2541(95)00160-3)
519
520 6) S.N. Tkachev, M.H. Manghnani and Q. Williams, In situ Brillouin spectroscopy of a
521 pressure-induced apparent second-order transition in a silicate glass, *Phys. Rev. Lett.*, 95,
522 057402, 2005. <https://doi.org/10.1103/PhysRevLett.95.057402>
523
524 7) C. Sanchez-Valle and J.D. Bass, Elasticity and pressure-induced structural changes in
525 vitreous MgSiO₃-enstatite at high pressures, *Earth Planet. Sci. Lett.*, 295, 523-530, 2010.
526 <https://doi.org/10.1016/j.epsl.2010.04.034>
527
528 8) C. Sonneville, D. De Ligny, A. Mermet, B. Champagnon, C. Martinet, G.H.
529 Henderson, T. Deschamps, J. Margueritat and E. Barthel, In situ Brillouin study of
530 sodium alumino silicate glasses under pressure, *J. Chem. Phys.* 139, 074501,
531 2013. <https://doi.org/10.1063/1.4818335>
532
533 9) J. Liu and J.-F. Lin, Abnormal acoustic wave velocities in basaltic and (Fe,Al)-bearing
534 silicate glasses at high pressures, *Geophys. Res. Lett.*, 41, 8832-8839, 2014.
535 <https://doi.org/10.1002/2014GL062053>
536
537 10) T. Sakamaki, Y. Kono, Y. Wang, C. Park, T. Yu, Z. Jing and G. Shen, Contrasting
538 sound velocity and intermediate-range order between polymerized and depolymerized
539 silicate glasses under pressure, *Earth Planet. Sci. Lett.*, 391, 288-295, 2014.
540 <https://doi.org/10.1016/j.epsl.2014.02.008>
541
542 11) A.N. Clark, C.E. Lesher, S.D. Jacobsen and Y. Wang, Anomalous density and elastic
543 properties of basalt at high pressure: Reevaluating of the effect of melt fraction on
544 seismic velocity in the Earth's crust and upper mantle, *J. Geophys. Res.*, 121, 4232-4248,
545 2016. <https://doi.org/10.1002/2016JB012973>
546
547 12) M.H. Manghnani, Q. Williams, T. Matsui, P.C. Schultz and C.R. Kurkjian, Effects of
548 composition, pressure, and temperature on the elastic properties of SiO₂-TiO₂ glasses: An
549 integrated ultrasonic and Brillouin study, *Minerals*, 10, 481, 2020.
550 <https://doi.org/10.3390/min10050481>
551

552 13) K. Aoki, T. Ohashi, O. Ikeda and A. Suzuki, Effects of alkali and alkaline-earth
553 cations on the high-pressure sound velocities of aluminosilicate glasses, *Phys. Chem.*
554 *Minerals*, 47, 28, 2020. <https://doi.org/10.1007/s00269-020-01098-3>

555

556 14) A.I. Chumakov, G. Monaco, A. Monaco, W.A. Crichton, A. Bosak, R. Ruffer, A.
557 Meyer, F. Kargl, L. Comez, D. Fioretto, H. Giefers, S. Roitsch, G. Wortmann, M.H.
558 Manghnani, A. Hushur, Q. Williams, J. Balogh, K. Parlinski, P. Jochym and P. Piekarz,
559 Equivalence of the boson peak in glasses to the transverse acoustic van Hove singularity
560 in crystals, *Phys. Rev. Lett.*, 106, 225501, 2011.
561 <https://doi.org/10.1103/PhysRevLett.106.225501>

562

563 15) P.A. Bingham, J.M. Parker, T.M. Searle and I. Smith, Local structure and medium
564 range ordering of tetrahedrally coordinated Fe^{3+} ions in alkali-alkaline earth-silica
565 glasses, *J. Non-Cryst. Solids*, 353, 2479, 2007.
566 <https://doi.org/10.1016/j.jnoncrysol.2007.03.017>

567

568 16) J.E. Shelby, Thermal expansion of mixed alkali-silicate glasses, *J. Appl. Phys.*, 47,
569 4489-4496, 1976. <https://doi.org/10.1063/1.322418>

570

571 17) R. Knoche, D.B. Dingwell, F.A. Seifert and S.L. Webb, Nonlinear properties of
572 supercooled liquids in the system Na_2O-SiO_2 , *Chem. Geol.*, 116, 1-16, 1994.
573 [https://doi.org/10.1016/0009-2541\(94\)90154-6](https://doi.org/10.1016/0009-2541(94)90154-6)

574

575 18) A. Monaco, A.I. Chumakov, G. Monaco, W.A. Crichton, A. Meyer, L. Comez, D.
576 Fioretto, J. Korecki and R. Ruffer, Effect of densification on the density of vibrational
577 states of glasses, *Phys. Rev. Lett.*, 97, 135501, 2006.
578 <https://doi.org/10.1103/PhysRevLett.97.135501>

579

580 19) E.H. Hamilton, and G.W. Cleek, Properties of sodium titanium silicate glasses, *J.*
581 *Nat. Bur. Stand.*, 61, 89-94, 1958.

582

583 20) P. Richet and Y. Bottinga, Heat capacity of liquid silicates: New measurements on
584 $NaAlSi_3O_8$ and $K_2Si_4O_9$, *Geochim. Cosmochim. Acta*, 44, 1535-1541, 1980.
585 [https://doi.org/10.1016/0016-7037\(80\)90117-9](https://doi.org/10.1016/0016-7037(80)90117-9)

586

587 21) T. Rouxel, H. Ji, T. Hammouda and A. Moreac, Poisson's ratio and the densification
588 of glass under high pressure, *Phys. Rev. Lett.*, 100, 225501, 2008.
589 <https://doi.org/10.1103/PhysRevLett.100.225501>

590

591 22) Z. Wang, T.F. Cooney and S.K. Sharma, In situ structural investigation on iron-
592 containing silicate liquids and glasses, *Geochim. Cosmochim. Acta*, 59, 1571, 1995.
593 [https://doi.org/10.1016/0016-7037\(95\)00063-6](https://doi.org/10.1016/0016-7037(95)00063-6)

594

595 23) D. Di Genova, J. Vasseur, K.-U. Hess, D.R. Neuville and D.B. Dingwell, Effect of
596 oxygen fugacity on the glass transition, viscosity ad structure of silica- and iron-rich

597 magmatic melts, *J. Non-Cryst. Solids*, 470, 78, 2017.
598 <http://dx.doi.org/10.1016/j.jnoncrysol.2017.05.013>

599

600 24) B. Champagnon, C. Martinet, M. Boudeulle, D. Vouagner, C. Coussa, T. Deschamps
601 and L. Grosvalet, High pressure elastic and plastic deformations of silica: In situ diamond
602 anvil cell experiments, *J. Non-Cryst. Solids*, 354, 569, 2008.
603 <https://doi.org/10.1016/j.jnoncrysol.2007.07.079>

604

605 25) T. Sato, N. Funamori and T. Yagi, Helium penetrates into silica glass and reduces its
606 compressibility, *Nat. Comm.*, 2, 345, 2011. <https://doi.org/10.1038/ncomms1343>

607

608 26) G. Shen, Q. Mei, V.B. Prakapenka, P. Lazor, S. Sinogeikin, Y. Meng and C. Park,
609 Effect of helium on structure and compression behavior of SiO_2 glass, *Proc. Natl. Acad.*
610 *Sci.*, 108, 6004-6007, 2011. <https://doi.org/10.1073/pnas.1102361108>

611

612 27) C. Weigel, A. Polian, M. Kint, B. Ruffle, M. Foret and R. Vacher, Vitreous silica
613 distends in helium gas: Acoustic versus static compressibilities, *Phys. Rev. Lett.*, 109,
614 245504, 2012. <https://doi.org/10.1103/PhysRevLett.109.245504>

615

616 28) B.Y. Chen, S.D. Mahanti and M. Yussouff, Helium atoms in zeolite cages: Novel
617 Mott-Hubbard and Bose-Hubbard systems. *Phys. Rev. Lett.*, 75, 473-477, 1995.
618 <https://doi.org/10.1103/PhysRevLett.75.473>

619

620 29) N.L. Ross, and E.P. Meagher, A molecular orbital study of $\text{H}_6\text{Si}_2\text{O}_7$ under simulated
621 compression, *Am. Mineral.*, 69, 1145-1149, 1984.

622

623 30) T.J. Sokolowski and M.H. Manghnani, Adiabatic elastic moduli of vitreous calcium
624 aluminates to 3.5 kbar, *J. Am. Ceram. Soc.*, 52, 539-542, 1969.
625 <https://doi.org/10.1111/j.1151-2916.1969.tb09160.x>

626

627 31) M. Guerette, M.R. Ackerson, J. Thomas, F. Yuan, E.B. Watson, D. Walker and L.
628 Huang, Structure and properties of silica glass densified in cold compression and hot
629 compression, *Sci. Reports* 5, 15343, 2015. <https://doi.org/10.1038/srep15343>

630

631 32) C. Sonneville, A. Mermet, B. Champagnon, C. Martinet, J. Margueritat, D. de Ligny,
632 T. Deschamps and F. Balima, Progressive transformations of silica glass upon
633 densification, *J. Chem. Phys.* 137, 124505, 2012. <https://doi.org/10.1063/1.4754601>

634

635 33) K.O. Trachenko, M.T. Dove, M.J. Harris and V. Heine, Dynamics of silica glass:
636 two-level tunnelling states and low-energy floppy modes, *J. Phys. Cond. Matter*, 12,
637 8041, 2000. <https://doi.org/10.1088/0953-8984/12/37/304>

638

639 34) M. Wyart, H. Liang, A. Kabla and L. Mahadevan, Elasticity of floppy and stiff
640 random networks, *Phys. Rev. Lett.*, 101, 215501, 2008.
641 <https://doi.org/10.1103/PhysRevLett.101.215501>

642

643 35) E. DeGiuli, A. Laversanne-Finot, G. During, E. Lerner and M. Wyart, Effects of
644 coordination and pressure on sound attenuation, boson peak and elasticity in amorphous
645 solids, *Soft Matter*, 10, 5628, 2014. <https://doi.org/10.1039/c4sm00561a>

646

647 36) Y. Kikuchi, H. Sudo and N. Kuzuu, Thermal expansion of vitreous silica:
648 Correspondence between dilatation curve and phase transitions in crystalline silica, *J.*
649 *Appl. Phys.*, 82, 4121, 1997. <https://doi.org/10.1063/1.366279>

650

651 37) J.E. Shelby, Formation and properties of calcium aluminosilicate glasses, *J. Am.*
652 *Ceram. Soc.*, 68, 155, 1985. <https://doi.org/10.1111/j.1151-2916.1985.tb09656.x>

653

654 38) A. Goel, D.U. Tulyaganov, S. Agathopoulos, M.J. Ribeiro and J.M.F. Ferreira,
655 Synthesis and characterization of $MgSiO_3$ -containing glass-ceramics, *Ceram. Internat.*,
656 33, 1481, 2007. <https://doi.org/10.1016/j.ceramint.2006.05.012>

657

658 39) G. Scannell, A. Koike and L. Huang, Structure and thermos-mechanical response of
659 TiO_2 - SiO_2 glasses to temperature, *J. Non-Cryst. Solids*, 447, 238, 2016.
660 <http://dx.doi.org/10.1016/j.jnoncrysol.2016.06.018>

661

662 40) P.S. Salmon et al., Pressure induced structural transformations in amorphous $MgSiO_3$
663 and $CaSiO_3$, *J. Non-Cryst. Solids*, X 3, 100024, 2019.
664 <https://doi.org/10.1016/j.nocx.2019.100024>

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