Letter

Observation of flat and weakly dispersing bands in the van der Waals semiconductor Nb₃Br₈ with breathing kagome lattice

Sabin Regmi, Anup Pradhan Sakhya, Tharindu Fernando, Yuzhou Zhao, Dylan Jeff, Milo Sprague, Favian Gonzalez, Milo Sprague, Mazharul Islam Mondal, Nathan Valadez, Damani Jarrett, Alexis Agosto, Jihui Yang, Jiun-Haw Chu, Saiful I. Khondaker, Nathan Valadez, Ting Cao, and Madhab Neupane, Lihui Yang, Iiun-Haw Chu, Saiful I. Khondaker, Nathan Valadez, Ting Cao, and Madhab Neupane, Department of Physics, University of Central Florida, Orlando, Florida 32816, USA

Department of Physics, University of Washington, Seattle, Washington 98195, USA

NanoScience and Technology Center, University of Central Florida, Orlando, Florida 32826, USA

Department of Materials Science and Engineering, University of Washington, Seattle, Washington 98195, USA

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Niobium halides, Nb_3X_8 (X = Cl, Br, I), which are predicted two-dimensional magnets, have recently gotten attention due to their breathing kagome geometry. Here we have studied the electronic structure of Nb_3Br_8 by using angle-resolved photoemission spectroscopy (ARPES) and first-principles calculations. ARPES results depict the presence of multiple flat and weakly dispersing bands. These bands are well explained by the theoretical calculations, which show they have Nbd character indicating their origination from the Nb atoms forming the breathing kagome plane. This van der Waals material can be easily thinned down via mechanical exfoliation to the ultrathin limit and such ultrathin samples are stable as depicted from the time-dependent Raman spectroscopy measurements at room temperature. These results demonstrate that Nb_3Br_8 is an excellent material not only for studying breathing kagome induced flat band physics and its connection with magnetism but also for heterostructure fabrication and for practical application.

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Quantum materials with kagome lattice—a geometry of six triangles sharing the corners to form a hexagon within in their crystal structure have been recently studied as the potential playgrounds for exploring the interplay among parameters such as geometry, topology, electronic correlations, magnetic, and charge density orders [1-12]. From the electronic structure point of view, a kagome lattice may support the presence of flat band, Dirac fermion, and saddle point with van Hove singularity. Angle-resolved photoemission spectroscopy (ARPES) [13-15] has been successfully utilized to experimentally reveal some or all of these features in different kagome materials [7,8,10,16–27]. A majority of the reports have been on kagome systems with conventional kagome geometry, where bond lengths between the ions forming such geometry is equal so that the size of all the triangles is same. Kagome lattice can occur in a different geometry called the breathing kagome, where alternating triangles have different bond lengths between the constituent ions leading to different sizes [28–30]. This difference may induce local electric dipole, resulting in a ferroelectric order [31]. Although it has been theoretically predicted that the breathing kagome systems can host intrinsically robust flat bands [30] and higher-order topology [29], the experimental studies of the breathing kagome systems for their electronic structure have been getting attention only recently [32–35].

Niobium halides Nb_3X_8 (X = Cl, Br, I) [36], which possess breathing kagome plane formed by the Nb atoms, present

themselves as material platforms to investigate the interplay of the breathing kagome geometry with magnetism and electronic correlations in both three- and two-dimensional limits. These compounds exhibit some intriguing attributes that are advantageous for optoelectronic and nanodevice applications [37,38]. They are moderate band-gap semiconductors [33,39] and in the monolayer form, they are predicted ferromagnet candidates [37,40,41]. The introduction of additional layers to the monolayer is predicted to lead to an antiferromagnetic ordering [40]. Importantly, because of very weak van der Waals coupling, they have very low exfoliation energies [38]. Therefore, obtaining the monolayers of these compounds is easily possible with mechanical exfoliation of the bulk crystals [33,34,39,42,43], which is beneficial for the fabrication of heterostructures. Recently, Nb₃Br₈ has been used to fabricate a heterostructure with NbSe₂ to form a Josephson junction that can control the direction of current without the need of magnetic field [44]. Bulk Nb₃Br₈ is in the singlet magnetic ground state at room temperature [45]. Although the density functional theory-(DFT) based computations have been performed to predict the band structure of the monolayer of this compound [37], experimental demonstration of the electronic structure is still lacking.

In this Letter, by means of ARPES measurements and supportive DFT computations, we report the electronic structure of Nb_3Br_8 . The results of the ARPES measurements are consistent with the semiconducting nature of the material and the computed band structures well reproduce the experimental observations. Multiple flat and weakly dispersing bands are observed in the electronic band structure. The orbital-resolved

^{*}Corresponding author: madhab.neupane@ucf.edu

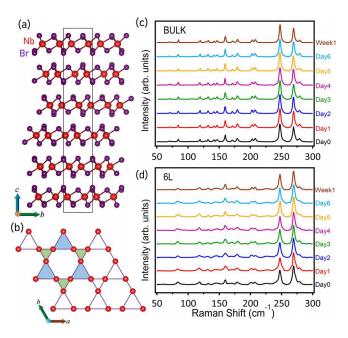


FIG. 1. Crystal structure and Raman spectroscopy measurements. (a) Crystal structure of Nb_3Br_8 . Red and purple balls represent Nb and Br atoms, respectively. (b) Breathing kagome plane of Nb atoms. Up-triangles (light blue) and down-triangles (light green) have different bond lengths. [(c) and (d)] Time-dependent Raman spectra of bulk and $6L\ Nb_3Br_8$, respectively.

calculations suggest these bands to have Nb d character indicative of the origination from the breathing kagome plane of Nb atoms. Moreover, through mechanical exfoliation of bulk crystal, a thin 6L sample of this material has been obtained and the stability of the sample in its ultrathin limit has been demonstrated through time-dependent room-temperature Raman spectroscopy measurements. This study highlights Nb₃Br₈ as an excellent material candidate from both physics and application points of view by revealing the occurrence of flat band physics originating from breathing kagome lattice and by demonstrating an easy exfoliation of ultrathin sample and its stability at room temperature.

High-quality single crystals of $\mathrm{Nb_3Br_8}$ used for this study were grown by using the chemical vapor transport method. The crystal structure and the chemical composition were checked by using x-ray diffraction and energy dispersive x-ray spectroscopy. The ARPES studies on these crystals were carried out at the Stanford Synchrotron Radiation Lightsource endstation 5-2, which is equipped with a DA30 analyzer. DFT-based [46,47] first-principles computational results were implemented in the VASP package with Projector Augmented Wave pseudopotential [48–50]. The mid band gap is set to be the zero-energy level for comparison with the experimental results. Hubbard potential U=1 eV on $\mathrm{Nb}\,d$ orbitals is used to address the effects of on-site Coulomb interactions. Details on the experimental and computational methods have been provided in the Supplemental Material (SM) section I [51].

Nb₃Br₈ is a van der Waals layered material that crystallizes in rhombohedral space group $R\overline{3}m$ (# 166) with lattice parameters a=b=7.080 Å and c=38.975 Å [36]. As shown in Fig. 1(a), a bulk unit cell is composed of six monolay-

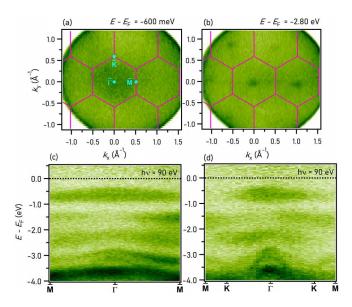


FIG. 2. Energy contours and band dispersion along different high-symmetry directions. [(a) and (b)] Energy contours measured by ARPES at the noted binding energies of $-600\,\mathrm{meV}$ and $-2.8\,\mathrm{eV}$, respectively. [(c) and (d)] Experimental band structures along $\overline{\mathrm{M}}$ - $\overline{\Gamma}$ - $\overline{\mathrm{M}}$ and $\overline{\mathrm{M}}$ - $\overline{\mathrm{K}}$ - $\overline{\mathrm{F}}$ - $\overline{\mathrm{K}}$ - $\overline{\mathrm{M}}$.

ers of Nb₃Br₈, where the neighboring layers are connected through a very weak van der Waals interaction along the crystallographic c axis. In each layer, Nb atoms are arranged in a two-dimensional plane with breathing kagome geometry [Fig. 1(b)] and are sandwiched in between the bilayers of Br atoms on either side. Because of the weak van der Waals coupling, the crystals are easily cleavable along the (0001) direction. The ab-plane orientation of the cleaved surface is indicated by the observation of sharp (0001) peaks in the single-crystal x-ray diffraction pattern (see Fig. S1 in the SM [51]). Importantly, bulk crystals can be easily exfoliated to ultrathin limits mechanically. We have thinned down the bulk crystal by using mechanical exfoliation to obtain a 6L Nb₃Br₈ (see Fig. S2 in the SM [51]). In Fig. 1(c), we present the results of the Raman spectroscopy measurements carried out on bulk and the 6L thin Nb₃Br₈. Tracking the evolution of the Raman spectra over time in Figs. 1(c) and 1(d), it is observed that the Raman modes do not experience shifts in peak frequency, changes in peak intensity, or peak broadening in either the bulk or 6L data. The absence of these changes in the Raman spectra signifies that the crystals are retaining good crystallinity throughout, speaking to the stability of the material in both bulk and thin layers. The fact that the crystal structure of Nb₃Br₈ is stable within a week's time is promising for the future application of the material in technology or further research of its fundamental properties. Recent magnetic susceptibility study reports that the bulk Nb3Br8 undergoes a magnetic transition to a singlet state at around 382 K [45]. Our magnetic susceptibility measurement (see Fig. S1 in the SM [51]) shows an abrupt jump at a temperature of around 394 K, which is slightly off the reported transition temperature.

The results of ARPES and accompanying first-principles band structure calculations are presented in Figs. 2-4, with

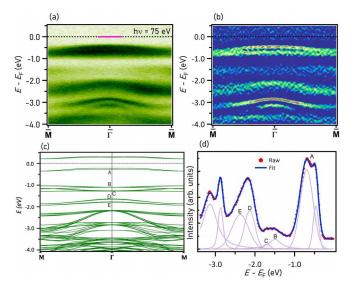


FIG. 3. Experimental and theoretical \overline{M} - $\overline{\Gamma}$ - \overline{M} band dispersion. (a) Band dispersion along \overline{M} - $\overline{\Gamma}$ - \overline{M} measured using incident photon energy of 75 eV and (b) its second derivative plot. (c) Calculated band structure along M- Γ -M. (d) EDC integrated within a momentum window of (-0.1/Å, 0.1/Å) represented by the magneta line in (a) and its Voigt fit.

an aim to unveil the electronic structure and the potential presence of flat bands arising from the breathing kagome geometry in the crystal structure. The energy contours obtained from APRES measurement with 90 eV incident photon energy at binding energies of -0.6 and $-2.8\,\mathrm{eV}$ are presented in Figs. 2(a) and 2(b), respectively. Consistent with the semiconducting nature of the material, no photoemission signal is obtained at the Fermi level. While a clear photoemission intensity can be visualized in the $-0.6\,\mathrm{eV}$ energy contour [Fig. 2(a)], a hexagonal pattern, typical of kagome systems, can be seen in the $-2.8\,\mathrm{eV}$ energy contour [Fig. 2(b)]. In order to reveal the underlying band structure along different high-symmetry directions, we took dispersion maps along the \overline{M} - $\overline{\Gamma}$ - \overline{M} and \overline{M} - \overline{K} - $\overline{\Gamma}$ - \overline{K} - \overline{M} directions. The high-symmetry points are marked in Fig. 2(a). The band structures along these high-symmetry directions seem to reveal similar features. Similar results also have been obtained for different

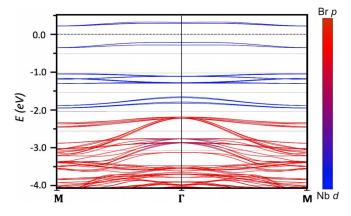


FIG. 4. Computed contribution of Nb d and Br p in the band structure along M- Γ -M.

photon energies of the incident light (see Fig. S3 in the SM [51]).

Figures 3(a) and 3(b) show energy-momentum dispersion along \overline{M} - $\overline{\Gamma}$ - \overline{M} measured with a photon energy of 75 eV and its second derivative plot, respectively. These plots show the presence of multiple flat and weakly dispersing bands within $-2.5 \,\mathrm{eV}$ binding energy. In Fig. 3(c), we present the calculated band structure along the M-Γ-M direction. The calculated band structure seems to reproduce the experimental observations quite well [see Fig. S4 in the SM [51] for comparison of experimental and calculated band dispersion along \overline{M} - \overline{K} - $\overline{\Gamma}$ - \overline{K} - \overline{M}]. For the comparison with the experimental results, the mid band gap in the calculation has been set to zero. We can see that the energy positions of the bands in experimental and calculations results do not match. This discrepancy might arise due to various factors such as the underestimation of the band gap by DFT or the sample not being perfectly stoichiometric. Nevertheless, the calculated band structure shows the presence of bandsets—labeled has two almost flat and parallel bands near the Γ point, among which the upper one seems to weakly disperse away from Γ before merging together near the M point. These bands can be seen in the dispersion map in Fig. 3(a) and better visualized in its second derivative [Fig. 3(b)]. Near the center of the BZ, these bands are almost flat and form a gap, which narrows down going towards the M point at which they seem to meet each other. Similarly, bandsets B and C both consist of two flat bands, one of which seems to separate and become dispersive going away from the Γ point. The bands within the bandsets B and C are hard to distinguish in the experimental dispersion map as they form a wide spectrum of intensity that gives rise to a flat feature centered around 1.6 eV [see Figs. 3(a)]. In Fig. 3(d), we present the energy distribution curve (EDC) integrated within a momentum window of (-0.1 / Å, 0.1 / Å) represented by the magenta line in the raw dispersion map in Fig. 3(a). Several intensity peaks corresponding to the presence of bands can be observed. In particular, a two-peak feature centered around $-0.6\,\mathrm{eV}$ can be observed that corresponds to the pair of almost flat bands (bandset A). By fitting the raw EDC data with Voigt functions, we obtain that these almost flat bands in bandset A are located around -0.46 and -0.70 eV binding energies, respectively. Another broad but much-suppressed intensity feature can be observed centered around 1.6 eV binding energy in the EDC, which corresponds to the observed flat dispersion in Fig. 3(a). A good fitting is obtained by considering presence of two peaks located around -1.5and $-1.7\,\mathrm{eV}$ binding energies. For both bandsets B and C, as the flat band and the flat region of the other band dispersing away from $\overline{\Gamma}$ merge together near the center of the BZ, they appear as these single peaks in the intensity plot within the experimental resolution. Fitting the EDC integrated within $(\overline{M}, \overline{M} - 0.1 \, A^{-1})$, where these bands seem to get separated, shows the presence of a pair of bands in each of the bandsets B and C (see the Fig. S5 in the SM [51]). Below bandset C, there exist a relatively dispersive bandset D and a weakly dispersing bandset E, which are centered around 2.13 and 2.37 eV near the $\overline{\Gamma}$ point in the experimental data [Fig. 3(d)].

The measurements at varying photon energies show that the aforementioned flat and weakly dispersing bands seem to have fairly similar dispersion irrespective of the choice of the photon energy indicative of their origination from twodimensional plane (see the Fig. S3 in the SM [51] for photon energy dependent measurements). To understand if the origination of the flat and dispersing bands is from the breathing kagome plane of Nb atoms, we plot in Fig. 4 the contribution of Nb d and Br p in the electronic structure along M- Γ -M. While the dispersive bands deep below the Fermi level have Br p character, it can be seen that the frontier flat and weakly dispersing bands discussed above have Nb d character (see Fig. S8 in the SM [51] for individual Nb d orbitals' contribution). This confirms that the Nb atoms, which configure themselves in the breathing kagome geometry, give rise to these flat and weakly dispersing bands. Nb₃Br₈ is a predicted ferromagnet in its monolayer form [37]. To see how the band structure in the two-dimensional form will compare with the band structure of the bulk, we computed ferromagnetic band structure calculations for the monolayer (see Fig. S6 in the SM [51]). From the comparison, we anticipate that the frontier electronic band structure for thin samples and bulk will look similar overall.

To conclude, the electronic structure of the niobium halide semiconductor Nb_3Br_8 has been studied by using ARPES and first-principles calculations. In its crystal structure, Nb atoms form a breathing kagome lattice with different Nb-Nb bond lengths in alternate triangles. Experimental dispersion maps reveal the presence of multiple weakly dispersing and flat bands, with the highest set of such bands occurring around 460 and 700 meV below the Fermi level. The comparison

with the first-principles computations show that these flat and weakly dispersing bands have Nb-d character, suggesting their origination from the breathing kagome plane formed by the Nb atoms. Room-temperature Raman measurements on a 6L sample mechanically exfoliated from the bulk crystal depict the stability of the ultrathin sample suggesting Nb₃Br₈ as an excellent ground to study breathing kagome geometry induced flat band physics and its application.

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