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Quantifying erosion rates and weathering pathways that maximize soil organic carbon storage

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Abstract Primary minerals that enter soils through bedrock weathering and atmospheric deposition can generate poorly crystalline minerals (PCM) that preferentially associate with soil organic carbon (SOC). These associations hinder microbial decomposition and the release of CO₂ from soils to the atmosphere, making them a critical geochemical control on terrestrial carbon abundance and persistence. Studies that

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Department of Ecology and Environmental Sciences, Umeå University, 907 36 Umeå, Västerbotten, Sweden explore these relationships are typically derived from soil chronosequences that experience negligible erosion and thus do not readily translate to eroding land-scapes. Here, we propose a theoretical framework to estimate steady-state PCM density and stocks for hilly and mountainous settings by coupling geochemical and geomorphic mass balance equations that account for soil production from bedrock and dust, soil erosion, PCM formation from weathering, and the transformation of PCMs into crystalline phases. We calculate an optimal erosion rate for maximum PCM abundance that arises because PCMs are limited by insufficient weathering at faster erosion rates and loss

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via "ripening" into more crystalline forms at slower erosion rates. The optimal erosion rate for modeled hilltop soil is modulated by reaction rate constants that govern the efficiency of primary mineral weathering and PCM ripening. By comparing our analysis with global compilations of erosion and soil production rates derived from cosmogenic nuclides, we show that landscapes with slow-to-moderate erosion rates may be optimal for harboring abundant PCM stocks that can facilitate SOC sequestration and limit turnover. Given the growing array of erosion-topography metrics and the widespread availability of highresolution topographic data, our framework demonstrates how weathering and critical zone processes can be coupled to inform landscape prioritization for persistent SOC storage potential across a broad range of spatial and temporal scales.

Keywords Weathering · Soil organic carbon · Poorly crystalline minerals · Erosion rate · Soil residence time · Soil organic matter · Geomorphology · Hillslope · Regolith

Introduction

The pedosphere contains more carbon than the atmosphere and biosphere combined (Jobbágy and Jackson 2000). Thus, changes in global soil organic carbon (SOC) stocks can have substantial impacts on atmospheric CO₂, implying that land management strategies for optimizing SOC storage may be an important tool for addressing climate change (Georgiou et al. 2021). Determining the extent to which soils can serve as an effective venue for carbon storage depends on the pace of SOC turnover and SOC persistence (Lavallee et al. 2020). With the key exception of deep organic soils subject to saturation and anoxic conditions (e.g., peatlands), the persistence of SOC is largely determined by the coupling of dissolved and/or microbially-processed organic matter with reactive mineral surfaces to generate relatively long-lived mineral-associated organic carbon (MOC) forms (e.g., Doetterl et al. 2015; Grant et al. 2022). These stabilizing associations with minerals can limit microbial access to SOC, thereby promoting SOC accumulation and persistence. In turn, a multitude of biogeochemical studies have sought to analyze how soil texture and mineralogy modulate the density and net inventory of SOC and MOC in soils (Georgiou et al. 2022; Kramer and Chadwick 2018; Schmidt et al. 2011; Torn et al. 1997). Despite the complex and dynamic nature of organo-mineral interactions (Kleber et al. 2021), studies from a diverse array of settings (as determined by landform age, parent material, and climate) demonstrate a strong correlation between poorly crystalline minerals (PCM) composed of Al and Fe hydroxides (e.g., ferrihydrite, allophane, and imogolite) and SOC (e.g., von Fromm et al. 2021). This association between PCM production and SOC retention is commonly attributed to the high degree of hydration, extensive surface area, and variable charge of PCMs (Chorover et al. 2004), although recent studies also explore the degree to which SOC decomposition drives PCM concentrations (Hall and Thompson 2022). Regardless, carbon dating suggests that MOC turnover rates are substantially slower than those measured for particulate organic carbon, demonstrating the strength of soil mineral controls on SOC persistence and their importance for informing SOC sequestration strategies (Lehmann and Kleber 2015).

In natural landscapes, PCMs originate from the weathering of primary minerals sourced from underlying bedrock (or other substrates) as well as atmospheric deposition (e.g., dust). Weathering rates and pathways are commonly studied using soil chronosequences from dated fluvial, marine, glacial, or volcanic surfaces subject to negligible surface erosion (Lawrence et al. 2015; Masiello et al. 2004; Torn et al. 1997). These studies demonstrate strong linkages between secondary minerals and SOC stocks, including age-dependent trends in mineral reactivity on microbial composition, soil respiration, and SOC stabilization (Doetterl et al. 2018). In particular, seminal work by Torn et al. (1997) in Hawaii shows that SOC fractions and stocks, PCMs, and ¹⁴C ages increase for basalt flows with ages spanning 10² to 10⁵ years old before decreasing as flow ages approach 10^6 years. The ~ 10^5 -ya peak in PCMs, SOC, and 14 C ages occurs because the older surfaces are subject to progressive "ripening" or dehydration of PCMs to crystalline clay phases and oxides, such as gibbsite, goethite, and kaolinite, which have comparably low surface area and charge densities relative to PCMs. This ripening results in a decreased affinity for SOC stabilization. This soil age "sweet spot" for SOC storage has also been observed in other chronosequence studies, although the peak age associated with maximum SOC and PCM density appears to vary, likely



reflecting differences in climate, parent materials, and/or SOC dynamics (Garcia Arredondo et al. 2019; Hunter et al. in press; Lawrence et al. 2015).

Because the vast majority of the Earth surface is subject to erosion, applying PCM-SOC relationships derived from chronosequence datasets requires consideration of both landscape age and erosional history (Slessarev et al. 2019). The geomorphic community has leveraged a vast array of tools, including stream network analysis, hillslope morphometrics, and erosion rates calibrated with cosmogenic nuclides, to show that variability in long-term erosion rates are substantial and pervasive across most hilly and mountainous landscapes (Mudd 2017). As a result, systematic variations in weathering, PCMs, and MOC stocks are likely, but the impact of these variations has not been incorporated into commonly used SOC inventory databases like SoilGrids (Poggio et al. 2021). Studies that bridge soil science and geomorphology often use the soil particle residence time framework, defined as the average length of time that soil particles spend in a soil of thickness, H, between entry (via bedrock-soil conversion) and exit (via erosion, E). This metric serves as a characteristic timescale during which biogeochemical processes occur on actively eroding hillslopes before soils are exported to fluvial networks or depositional zones (Almond et al. 2007; Mudd and Yoo 2010). With one notable exception (Wang et al. 2018), studies that quantify soil mineralogy and organic matter on hillslopes that occupy an erosional gradient are limited. As a result, we lack a generalized theoretical framework to characterize PCM abundance across erosional settings.

Here, we propose a predictive model for PCM concentrations and stocks in soil-mantled, erosional settings by coupling a geomorphic soil production function with a geochemical mass balance for the inputs and losses of primary minerals and PCMs. Our approach takes inspiration from recent work (Slessarev et al. 2022) by accounting for the weathering of primary minerals to PCMs but also incorporates the subsequent ripening of PCMs into less reactive crystalline phases. We demonstrate that the soil age sweet spot of maximum PCM abundance observed in chronosequence studies has an analogy in erosional settings, which corresponds to the erosion rate associated with maximum PCM concentration and stocks that promote MOC formation. This theoretical framework should inform the scaling of global and regional SOC databases and facilitate carbon-based land management decisionmaking across the erosional landscapes that constitute the vast majority of the Earth surface.

Methods

Overview

To calculate how erosion rate modulates the concentration and inventory of PCMs that promote SOC persistence, we modified a mass balance weathering model for ridge crest soils on eroding hillslopes (Ferrier and Kirchner 2008), hereafter referred to as FK08. This model tracks the input and fate of a suite of minerals in a well-mixed, eroding soil, which allows us to test erosion rate controls on soil geochemistry. In our modification depicted in Fig. 1, soils gain mass by the conversion of bedrock (or underlying substrate) and dust incorporation, they lose mass via physical erosion and chemical dissolution, and their composition changes according to the transformation of mineral phases into secondary phases. Our modification enables the model to generate secondary minerals, such as PCMs and clays, at a rate that depends directly on the weathering rate of primary mineral phases (such as plagioclase feldspar). In FK08, by contrast, secondary mineral generation was modeled as a constant value, independent of primary mineral weathering rates.

Theory: soil mineralogy and weathering pathways on eroding hillslopes

We expand the theoretical framework of FK08 to incorporate mineral supply from dust incorporation as well as secondary mineral generation and solute production as independent reactions for any given mineral phase. Most generally, the revised theory accounts for the rate of change of mass for mineral phase *X* in soil according to the balance of inputs and losses of *X*:

$$\frac{d(\rho_s HC_{Xs})}{dt} = \sum Input_X - \sum Loss_X$$
 (1)

where ρ_s is soil density (M L⁻³), H is soil thickness (L), and C_{Xs} is concentration of mineral phase X in soil. Here, we account for inputs of X (M L⁻² T⁻¹) arising from soil production (i.e., the conversion



Soil mass conservation framework Dust deposition, Pd $C_{X,d}=dust$ Physical erosion, E mineral abundances Mineral dissolution, W $C_{X,s}$ =soil Mineral mineral alteration abundances Soil production, $C_{X,r} = rock$ mineral abundances bedrock Primary

Fig. 1 A Schematic of soil-mantled profile subject to mass loss via physical erosion and mineral dissolution, mass gain via dust deposition and soil production, and mineral transformation via secondary mineral generation. B The weathering pathways and mass balance described by our model represent the input of primary minerals from bedrock into soil followed

of bedrock or substrate to mobile soil), dust deposition, and alteration of other mineral phases to X, as follows:

Input rate of X from underlying

$$rock \ or \ substrate = P_r C_{Xr}$$
 (2a)

Input rate of X from atmospheric
or dust deposition =
$$P_d C_{Xd}$$
 (2b)

Input rate of X from alteration of

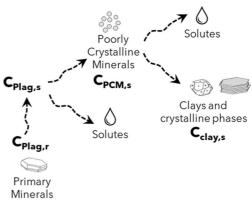
other mineral phases =
$$\sum_{j=1}^{n} \rho_{s} HC_{js} K_{j,X}$$
 (2c)

In addition, we account for losses (M L^{-2} T^{-1}) of mineral phase *X* owing to physical erosion, dissolution, and alteration of *X* to other mineral phases, according to:

Loss rate of X to physical erosion =
$$EC_{X_S}$$
 (3a)

Loss rate of X to solutes =
$$\rho_s HC_{Xs}K_{X,sol}$$
 (3b)

Weathering pathway for poorly crystalline minerals



by the weathering of those minerals into poorly crystalline minerals (or PCMs), which are highly effective at sorbing soil organic carbon (SOC). These PCMs can then subsequently ripen into clays and crystalline pedogenic mineral phases that are less conducive to organic associations

Loss rate of X by alteration to

other mineral phases =
$$\sum_{i=1}^{n} \rho_{s} H C_{Xs} K_{X,j}$$
 (3c)

where P_r and P_d are soil and dust production rate (M L⁻² T⁻¹), respectively, E is physical erosion rate (M L⁻² T⁻¹), C_{Xr} and C_{Xd} are concentrations of mineral phase X in rock and dust (M M⁻¹), respectively, and $K_{X,j}$ is an effective weathering rate constant for the conversion of parent mineral phase X to daughter mineral phase or product j (T⁻¹), where X and j can either serve as parent or daughter phases and solute can only serve as a daughter product. Here, daughter phases are pedogenic minerals, such as PCMs (e.g., allophane) or more intensively weathered clays (e.g., kaolinite). The $K_{X,j}$ term is a bulk weathering parameter that combines the dissolution (or reaction) rate constant and specific surface area terms that are often used to model chemical weathering (Ferrier and West 2017).

Assuming constant soil density, we calculate the rate of change of soil thickness (*H*) by accounting for inputs of soil production and dust and losses to physical erosion and solutes:



$$\frac{dH}{dt} = \frac{1}{\rho_s} \left(P_0 e^{-\alpha H} + P_d - E - \rho_s H \sum_{j=1}^n C_{js} K_{j,sol} \right)$$
(4)

where P_0 is the peak or maximum soil production rate (M L⁻² T⁻¹) and α is a scaling parameter (L⁻¹) associated with an empirical soil production relationship supported by abundant field evidence (Heimsath et al. 2012), which indicates that soil production rates decline exponentially with depth:

$$P_r = P_0 e^{-\alpha H} \tag{5}$$

By applying the product rule for differentiation of Eq. 1 and combining Eqs. 1 to 5, we obtain an expression for the rate of change of the concentration of mineral phase *X* in soil:

$$\frac{dC_{Xs}}{dt} = \frac{P_0 e^{-\alpha H}}{\rho_s H} \left(C_{Xr} - C_{Xs} \right) + \frac{P_d}{\rho_s H} \left(C_{Xd} - C_{Xs} \right)
+ \sum_{j=1}^{n} C_{js} K_{j,X} - C_{Xs} K_{X,sol} - C_{Xs} \sum_{j=1}^{n} K_{X,j}
+ C_{Xs} \sum_{j=1}^{n} C_{js} K_{j,sol}$$
(6)

Taken together, Eqs. 4 and 6 can be used to calculate how the mineral composition of hillslope soils evolves with time and is subject to model assumptions, including negligible chemical alteration of underlying substrate that enters the soil column (e.g., bedrock weathering) and minimal soil transport from upslope locations, such that the model is primarily relevant for ridgecrest or hilltop positions. Most generally, Eqs. 4 and 6 imply that soil mineral concentrations depend on the balance between the supply rate of primary minerals from bedrock and dust and the pace with which those primary minerals are chemically transformed into solutes and pedogenic phases. The former is primarily modulated by the erosion rate imposed by baselevel forcing and hillslope response, and the latter is determined by the relevant weathering rate constants (e.g., $K_{X,i}$) and soil particle residence

The rate constants for primary silicate mineral dissolution have been constrained under field and laboratory conditions and values vary with geologic context and climate (e.g., White and Brantley 2003). By contrast, rate constants relevant for

modeling intermediate weathering pathways (such as alteration of PCMs to kaolinite and/or crystalline oxide phases) are lacking, although a limited number of chronosequence studies that track the formation of crystalline phases imply that the relevant timescales exceed 10^5 yr (Aniku and Singer 1990; Torn et al. 1997). Values of the soil production parameters P_0 and α have been shown to vary with climate and vegetation (Amundson et al. 2015).

Application to poorly crystalline mineral abundance

In order to assess how physical erosion rate, E, modulates the abundance of PCMs, we applied a simplified version of our generalized theoretical framework. First, we assume steady (or time-invariant) soil production, erosion, and weathering, which is relevant at the hillslope and catchment scale in most hilly and mountainous settings. Next, we neglect atmospheric deposition (i.e., $P_d = 0$) given that substantial dust contributions are typically associated with arid settings where PCM production is limited and thus less relevant for our approach. Third, following FK08, we non-dimensionalize Eqs. 4 and 6 using the relationships shown in Table 1. In particular, we define the non-dimensional physical erosion rate, $\hat{E} = E/P_0$, which normalizes E with the peak soil production rate. In addition, we define $\hat{K}_{X,j} = K_{X,j}t_p$, where $t_p = \rho_s/(P_0 \alpha)$, which represents the characteristic soil production time as defined by the soil production function parameters in Eq. 5. Given typical values of P_0 and α (Amundson et al. 2015), t_p is ~ 10^4 year. Although this timescale is closely related to the soil particle residence time, which is typically defined as the ratio of soil thickness to erosion rate (H/E), t_n does not vary with erosion rate and thus maintains a

Table 1 Relationships between dimensional and non-dimensional terms

Parameter	Non-dimensional version
Time, t	$\hat{t} = (P_0 \alpha t) / \rho_s$
Physical erosion rate, E	$\hat{E} = E/P_0$
Soil thickness, H	$\hat{H} = \alpha H$
Effective reaction rate constant, $K_{X,j}$	$\widehat{K}_{X,j} = \left(\rho_s K_{X,j}\right) / \left(P_0 \alpha\right)$



constant value despite simulated variations in erosion rate. In this case, values of $\hat{K}_{X,j}$ reflect the characteristic timescale of soil production relative to that of chemical reaction, which is similar to the Damkohler number often used in chemical weathering models subject to a swap of the numerator and denominator terms (Maher 2010).

Given these assumptions, we obtain the following dimensionless versions of Eqs. 4 and 6, where the $\hat{}$ symbol denotes a dimensionless term (note that C_{Xr} and C_{Xs} are dimensionless as defined here):

$$\frac{d\widehat{H}}{d\widehat{t}} = e^{-\widehat{H}} - \widehat{E} - \widehat{H} \sum_{j=1}^{n} C_{js} \widehat{K}_{j,sol}$$
 (7)

$$\frac{dC_{Xs}}{d\hat{t}} = \frac{e^{-\hat{H}}}{\hat{H}} \left(C_{Xr} - C_{Xs} \right) + \sum_{j=1}^{n} C_{js} \hat{K}_{j,X}
- C_{Xs} \left(\hat{K}_{X,sol} + \sum_{j=1}^{n} \hat{K}_{X,j} - \sum_{j=1}^{n} C_{js} \hat{K}_{j,sol} \right)$$
(8)

Following Ferrier and West (2017), we use a nonlinear equation solver in Matlab (see supplemental materials) to simultaneously solve steady-state versions of Eqs. 7 and 8 (i.e., $dC_{Xs}/dt = 0$, $d\hat{H}/dt = 0$) for all mineral phases. This framework enables us to generate predictions of C_{Xs} and \hat{H} for specified values of the dimensionless physical erosion rate (\hat{E}) , weathering rate constants $(\hat{K}_{X,j})$, and bedrock (or substrate) mineral abundances (C_{Xr}) . In applying this model, we focus on two stages of a weathering pathway that includes: (1) the creation of PCMs via weathering of primary minerals; and (2) their subsequent ripening and transformation into clays and crystalline oxide phases. In particular, we focus on the transformation

of plagioclase into allophane (a common PCM on the aluminum weathering series) and its subsequent alteration into kaolinite and crystalline oxide phases, which are considered collectively for our simulations (Fig. 1). Specifically, we seek to determine whether our formulation generates soil allophane concentration ($C_{allo,s}$) erosion rate (\hat{E}) relationships that yield a maximum value of $C_{allo,s}$, and thus an optimal erosion rate for facilitating MOC abundance.

For these calculations, we use a bedrock composition resembling a plagioclase-rich granite or immature sedimentary unit such as an arkosic sandstone with the following bedrock mineral fractions: quartz (0.25), plagioclase (0.4), potassium feldspar (0.2), and muscovite (0.15). In addition, we use the fieldderived effective chemical solution rate constants listed in Table 2, which are chosen to be within an order of magnitude of values used in Ferrier and Kirchner (2008, supplemental Table C.1). The values used in FK08 account for grain roughness and surface area among other factors that govern the pace of mineral dissolution (Maher et al. 2009). Given our focus on weathering pathways that include the generation and subsequent ripening of allophane, we calculate how different values of $\hat{K}_{plag,allo}$, which characterizes the rate at which plagioclase is transformed into allophane, and $K_{allo,kaol}$, which paces the ripening of allophane into kaolinite and crystalline phases, collectively influence how the concentration of allophane in soil $(C_{allo,s})$ varies with \hat{E} . Put otherwise, we perform a sensitivity analysis to determine how different combinations of the plagioclase → allophane and allophane → kaolinite rate constants, which are poorly constrained by field studies, determine erosion rates that maximize allophane abundance in soil.

Table 2 Values of effective chemical reaction constants $(\hat{K}_{X,p}\hat{K}_{j,X} \text{ and } \hat{K}_{X,sol})$

Parents	Daughters							
	Solutes	Quartz	Plagioclase	K-Feldspar	Muscovite	Kaolinite	Allophane	
Quartz	1×10 ⁻⁴	0	0	0	0	0	0	
Plagioclase	1×10^{-1}	0	0	0	0	0	10^{-2} to 10^{0}	
K-feldspar	1×10^{-2}	0	0	0	0	0	0	
Muscovite	1×10^{0}	0	0	0	0	0	0	
Kaolinite	1×10^{-3}	0	0	0	0	0	0	
Allophane	1×10^{-2}	0	0	0	0	10 ⁻² to 10 ⁰	0	



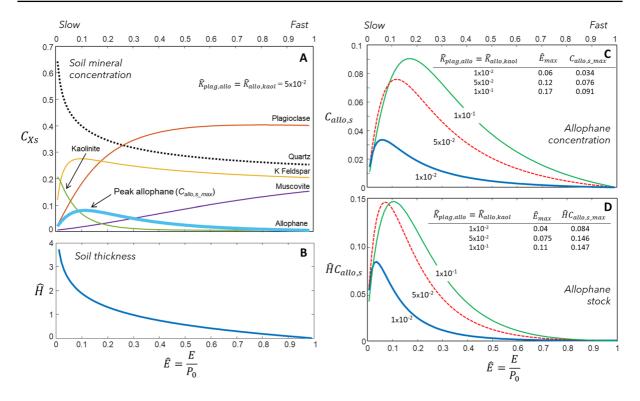


Fig. 2 A Dependence of mineral abundances on steady-state dimensionless erosion rate. B Variation of dimensionless soil thickness with dimensionless erosion rate. C The optimal erosion rate to maximize allophane abundance varies with the chemical reaction parameters for the transformations of plagio-

clase \rightarrow allophane and allophane \rightarrow clays and other crystalline phases. **D** Variation of the optimal erosion rate for peak allophane stocks, calculated as the product of dimensionless soil thickness and allophane abundance

Results

In order to generate steady-state $C_{allo,s}$ - \hat{E} relationships, we simultaneously solve steady-state versions of Eqs. 7 and 8 to simulate how the entire suite of mineral phases varies with \hat{E} . For a given value of \hat{E} , each simulation yields a suite of mineral concentrations that represents the composition of a steadily eroding ridge crest soil (Fig. 2A). For $\hat{K}_{plag,allo} = \hat{K}_{allo,kaol} = 5 \times 10^{-2}$, the soil mineral concentrations for values of \hat{E} slightly below 1.0 are essentially equivalent to the bedrock concentrations listed above. These high \hat{E} values correspond to erosion rates just below the peak soil production rate, P_0 , which reflects a negligibly thin, minimally altered soil with values of $C_{allo,s}$ and $C_{kaol,s}$ equal to zero. With decreasing values of \hat{E} , our simulations show that individual mineral phases have different sensitivity to changes in \hat{E} , reflecting differential weathering as well as the onset of secondary mineral weathering pathways. For the simulation shown in Fig. 2A, these differential changes arise due to the rapid pace of muscovite and plagioclase dissolution relative to other minerals. In addition, with decreasing \hat{E} , the alteration of plagioclase into allophane results in increasing $C_{allo,s}$ values that peak at $0.1 < \hat{E} < 0.15$ (Fig. 2A). As $\widehat{E} \to 0$, $C_{allo,s}$ values decline as long soil residence times allow for extensive transformation of allophane into kaolinite. At these near-zero values of \hat{E} , the modeled soil mineral composition becomes dominated by quartz and kaolinite as the extremely slow erosion rates result in near comprehensive dissolution of non-quartz primary minerals. Our simulations also provide steady-state values of soil thickness (\widehat{H}) that declines rapidly with \widehat{E} (Fig. 2B), which is consistent with field observations (Heimsath et al. 1997).



To explore controls on the erosion rate that optimizes allophane abundance and constrain relevant values of the reaction rate constants for allophane and kaolinite, we generated $C_{allo.s}$ - \hat{E} relationships for parameter combinations within the following bounds: $1 \times 10^{-2} \le \hat{K}_{plag,allo} \le 1 \times 10^{0}$ and $1 \times 10^{-2} \le \hat{K}_{allo,kaol} \le 1 \times 10^{0}$. For example, Fig. 2C shows $C_{allo,s}$ - \hat{E} relationships for three parameter combinations where $\hat{K}_{plag,allo} = \hat{K}_{allo,kaol}$. For these simulations, as the value of $\hat{K}_{allo,kaol}$ (and thus $\hat{K}_{play,allo}$) increases, the value of \hat{E} associated with peak allophane (termed \hat{E}_{max}) also increases, along with the peak concentration of allophane (termed $C_{allo,s\ max}$; Fig. 2C). In order to assess the optimal erosion rate for total allophane abundance, we estimated the allophane stock as the product of \hat{H} and C_{allo} and modeled how the allophane stock varies with \hat{E} (Fig. 2D). According to this analysis, \hat{E}_{max} for allophane stocks is shifted to lower values owing to the nonlinearity of the $\hat{H} - \hat{E}$ relationship shown in Fig. 2B. For the remainder of our analysis, we focus on \hat{E} values that optimize $C_{allo,s}$ (rather than $\hat{H}C_{allo,s}$), while retaining the understanding that consideration of allophane stocks slightly decreases values of \widehat{E}_{max} .

To determine how other combinations of $\widehat{K}_{allo,kaol}$ and $\hat{K}_{play,allo}$ impact the optimal erosion rate and peak allophane concentration, we modeled C_{allo} s- \hat{E} relationships for various parameter combinations of $\widehat{K}_{plag,allo}$ and $\widehat{K}_{allo,kaol}$ and calculated \widehat{E}_{max} (Fig. 3A) as well as $C_{allo,s\ max}$ (Fig. 3B). These simulations demonstrate that \hat{E}_{max} increases with $\hat{K}_{allo,kaol}$ and/or $\widehat{K}_{plag,allo}$ with modeled values of \widehat{E}_{max} varying from 0.08 to 0.5 for $\hat{K}_{plag,allo} = \hat{K}_{allo,kaol} = 1 \times 10^{-2}$ and $\hat{K}_{plag,allo} = \hat{K}_{allo,kaol} = 1 \times 10^0$, respectively (Fig. 3A). By contrast, Fig. 3B shows that C_{allo,s_max} is highest (0.35) for large values of $\hat{K}_{plag,allo}$ and small values of $\hat{K}_{allo,kaol}$, while low values of $C_{allo,s max}$ (<0.05) are associated with low values of $\widehat{K}_{plag,allo}$ and high values of $\hat{K}_{allo,kaol}$. The tradeoff between these two reaction constants reveals how the relative pace of allophane generation from plagioclase and subsequent alteration to kaolinite dictates the steady-state concentration of allophane in eroding soils. Because the concentration of poorly crystalline minerals seldom exceeds 4% (or 0.04) based on a meta-analysis of soil databases (Slesserav et al. 2022), our analysis implies that the parameters defined by the lower right corner of Fig. 3B (i.e., low values of $\hat{K}_{plag,allo}$ and high values of $\hat{K}_{allo,kaol}$) are most likely representative of natural landscapes. This region of the parameter space corresponds with \hat{E} values between 0.1 and 0.25.

Discussion

Our simulations reveal an optimal erosion rate that maximizes the concentration of PCMs: in this case allophane (Fig. 4A). When \hat{E} is high, erosion is rapid and/or soil is thin, which limits the production of PCMs. In essence, soil minerals move through the soil column too rapidly for significant weathering and PCM production to occur. By contrast, PCM ripeninglimited conditions occur when \hat{E} is low, which corresponds to slow erosion with thick, intensely weathered soils, such that PCMs have largely been altered to clays and more crystalline pedogenic phases. At these low values of \hat{E} , soil particle residence time is sufficiently long that PCM ripening has been pervasive, limiting the availability of PCMs available to form long-lived associations with SOC. Our theoretical findings are analogous to datasets derived from chronosequence studies in that PCM concentrations in soils increase with substrate age and attain a maximum value before decreasing with further increases in soil age. In Torn et al. (1997), the sweet spot for PCM concentrations and SOC abundance occurs on ~ 100-kya lava flows, whereas Garcia Arredondo et al. (2019), Hunter et al. (in press), and Lawrence et al. (2015) report soil PCM and SOC peaks at 90, 30, and 24 kya, respectively. While broadly consistent, the differences in these ages likely reflect parent material and climate variables that modulate mineral abundance and the pace of weathering processes, which are encapsulated in the $K_{X,i}$ terms. As elaborated by Mudd and Yoo (2010), these chronosequence studies, which are often used to quantify the mechanisms responsible for timedependent changes in weathering rates (White and Brantley 2003), also do not account for the on-going influxes of primary minerals into the soil from underlying rock and/or dust, which resupply the soil column in erosional settings (Almond et al. 2007). Importantly, the erosion rate associated with peak PCM and



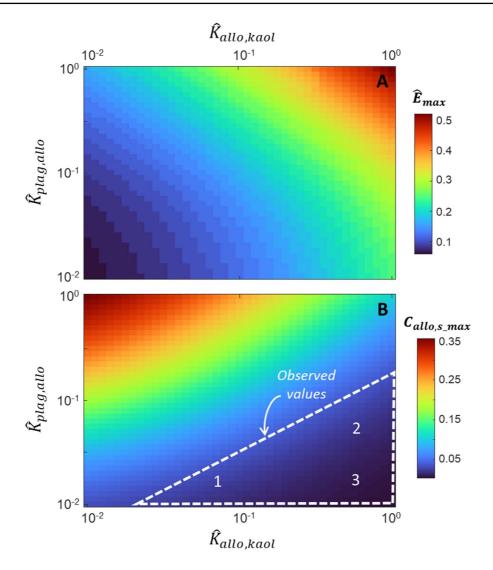


Fig. 3 Influence of reaction rate constants on the optimal erosion rate and peak allophane abundance. **A** Variation of optimal erosion rate (\hat{E}_{max}) for peak allophane abundance (see examples in Fig. 2C) with differing plagioclase \rightarrow allophane and allophane \rightarrow clays/crystalline phases rate constants. Note that increases in either rate constant results in higher values for the dimensionless erosion rate that maximizes allophane abundance. **B** Variation of peak allophane abundance (C_{allo,s_max}) with reaction rate

constants, which shows that the relatively low PCM abundances (<4%) observed in soil geochemical databases (Slesserav et al. 2022) are associated with the slow transformation of plagioclase \rightarrow allophane and relatively rapid transformation of allophane \rightarrow clays/crystalline phases, which is reflected in the lower right corner of the plot and denoted by a dash white triangle. The numbers 1, 2, and 3 refer to the parameter combinations for simulations shown in Fig. 4A and represent the variability of weathering rate constants that are relevant to natural settings

SOC may be decoupled in some landscapes owing to biotic factors that modulate SOM inputs and microbial respiration, which can vary with aspect, slope, and local hydrologic context.

Here, we focus on modeling PCM abundances for predicting SOC sequestration potential across eroding landscapes, which could have meaningful implications for mapping and management. Nonetheless, our analysis is also relevant for assessing soil organic matter and other ecologically-important biogeochemical cycles like soil nitrogen and phosphorus, which also likely vary with the prevalence of PCMs (e.g., Eger et al. 2018). Furthermore, we acknowledge that PCMs are not the only means for generating



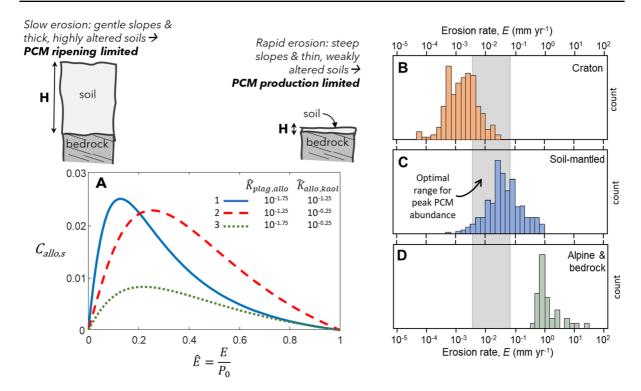


Fig. 4 A Erosion rate controls on soil PCM abundance (in this case allophane) for three simulations using parameter combinations that represent the highlighted region in Fig. 3B, which corresponds to values consistent with field observations. Whereas production limits PCM abundances in rapidly eroding landscapes, PCM alteration to clays and crystalline phases (via ripening) limits PCM abundance in slow-eroding landscapes. According to this analysis, optimal values of \hat{E} for peak PCM

abundance vary from 0.1 to 0.25. **B–D** Erosion rate compilation of Montgomery (2007), which implies that the faster eroding fraction of cratonic settings and slower eroding portion of soil-mantled settings likely harbor optimal PCM abundances, given the estimates of peak soil production rates from Amundson et al. (2015). Although these datasets estimate E and P_0 in dimensions of L T^{-1} , the values can be translated directly to our non-dimensional model framework

substantial MOC pools, given that certain clay minerals are also effective at forming organo-chemical associations—although often with decreased efficacy relative to PCMs. In many soils, the silt and clay fraction correlates with PCM abundance, although key differences have been documented (Rasmussen et al. 2018). In addition, our framework addresses PCM formation independent of the production of organic compounds available to sorb with PCM surfaces, which is requisite for SOC stabilization (Lehmann et al. 2020). Lastly, although our mass balance model assumes well-mixed soil and neglects significant alteration of the underlying substrate that would alter primary mineral abundance and PCM production, it could be adapted to accommodate substrate weathering that precedes soil production (i.e., weathering

of saprolite), as well as incomplete and/or depthdependent soil mixing.

The assumption of steady state erosion, soil thickness, and mineral composition employed here should not be confused with orogen- or even catchment-scale steady denudation (Willett and Brandon 2002), which are frequently violated as evidenced by the presence of knickpoints and other erosional gradients that result from baselevel changes and/or lithologic variations. Rather, our methodology assumes that local (i.e., pedon or solum) soil thickness and properties do not change substantially over timescales less than the soil particle residence time or characteristic soil production time. The steady-state assumption requires careful consideration in slow-eroding settings, which may record a longer history of perturbations related to climatic changes, as previous analyses



have demonstrated the key role of landscape age for predicting weathering intensity and soil mineralogy (Slessarev et al. 2019). As such, a transient analysis of our model predictions under different environmental forcings would be useful to assess the impact of interglacial-glacial fluctuations or other climate variations that determine model parameters (Ferrier and West 2017), particularly, P_0 , α , and the rate constants $\hat{K}_{plag,allo}$ and $\hat{K}_{allo,kaol}$.

Given the strong coupling between erosion rate and tectonic setting, our findings imply that geologic factors may serve as a key bottom-up control on PCM abundance in soils. Montgomery (2007) parsed erosion rate data into cratons, soil-mantled hillslopes, and alpine or bedrock environments associated with rapid tectonic forcing and showed that erosion rates vary by $> 10^5$ amongst these settings (Fig. 4B–D). Importantly, erosion rates greater than 10⁰ mm year⁻¹ tend to occur in alpine or bedrock settings (Fig. 4D) with a patchy or non-existent soil mantle such that these landscapes have negligible potential for SOC storage and low PCM accumulation (Fig. 4A). In this regime, erosion rates equal or exceed peak soil production (P_0) , corresponding with $\hat{E} \ge 1$ in our framework. By contrast, highly-altered soils in slow-eroding cratonic settings with erosion rates lower than 10⁻³ mm year⁻¹ may feature low PCM abundances owing to long soil residence times that facilitate substantial alteration of PCMs into less reactive crystalline phases (Fig. 4B; Doetterl et al. 2018). Our findings imply that maximum PCM abundances are likely associated with erosion rates that occur between these two end-member regimes, and more specifically, in the range of overlap between cratonic and soil-mantled settings (Fig. 4B, C). To generate this prediction, we used our results showing peak PCM concentrations for $0.1 < \hat{E} < 0.25$ (Fig. 4A) and commonly observed P_0 estimates of 0.04 to 0.3 mm yr⁻¹ (Amundson et al. 2015) to calculate optimal erosion rates according to $E = \hat{E}P_0$. From this analysis, we determine that maximum PCM abundance occurs for erosion rates between 0.004 and 0.075 mm yr⁻¹, which is depicted by a vertical gray bar in Fig. 4B-D. Our analysis implies that while steep, rapidly eroding terrain that comprises a small fraction (5–10%) of Earth surface accounts for > 50% of global sediment flux (Larsen et al. 2014), gentle-to-moderate inclined terrain with slow-to-moderate erosion rates constitutes a substantial fraction of the Earth surface and harbors soils with abundant PCMs and elevated SOC sequestration potential. In addition, by combining our predicted PCM stocks with erosion rate data, our framework enables the estimation of erosional PCM export to fluvial systems and depositional zones although these calculations are beyond the scope of this contribution. Importantly, we note that PCM abundance is also attendant to climatic conditions favoring chemical weathering as well as parent materials rich in primary minerals like feldspars (Slessarev et al. 2022), such that our approach also requires consideration of climatic and lithologic constraints.

Our framework for weathering and PCM production across eroding landscapes offers a simplified perspective on coupled and highly nonlinear weathering processes that are often addressed with parameterrich and computationally demanding reactive transport models (Li et al. 2017). These models, which have been successfully applied to chronosequences (Maher et al. 2009), can address the interdependencies of fluid flow, mineral supply, and weathering kinetics, and thus provide an opportunity to assess more sophisticated process linkages than incorporated in our framework. For example, our model does not explicitly account for production of kaolinite directly from plagioclase. Realistically, the rate of plagioclase weathering likely determines the partitioning into secondary phases such as allophane and kaolinite, although the solubility and kinetics of allophane strongly favor allophane production in most settings. A rigorous and mechanistic treatment of these and other relevant weathering pathway controls is possible (Zhu et al. 2010), although it lies beyond the scope of this contribution. Furthermore, our weathering paramterization is not equivalent to compilations of time-dependent rates that emerge from chronosequence field studies that do not account for the supply of fresh minerals via soil production (White and Brantley 2003). Despite these simplifications, our parsimonious approach reveals readily-accessible and testable insights that arise due to first-order process dependencies between erosion rates, soil weathering, PCM abundance, and SOC stabilization potential.

Datasets from soil-mantled settings that include soil mineralogy, SOC, and long-term erosion rate are limited, precluding our ability to rigorously test our framework. Notably, Wang et al. (2018, 2022) analyzed soils above and below a prominent channel



Table 3 Soil thickness, erosion rate, and Fe extraction data for Middle Fork Feather River soil-mantled hillslopes occurring above (POMD) and adjacent (BRC) to a major knickpoint as reported in Wang et al. (2018, 2022)

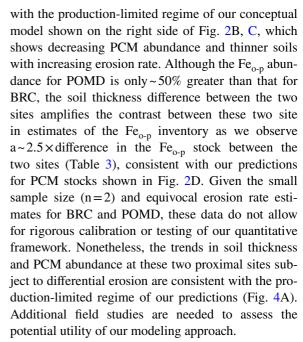
Site	Average soil thickness (m)	Average Fe _{o-p} (mg gm ⁻¹)	Fe _{o-p} inventory (kg m ⁻²)	Topo- graphic cur- vature (m ⁻¹)	Modeled local erosion rate (mm yr ⁻¹)
POMD	1.05	0.682	0.913	0.0046	0.023
BRC	0.5	0.454	0.331	- 0.017	0.087

 Fe_{o-p} is oxalate-extractable iron less pyrophosphate-extractable iron, a proxy for poorly crystalline minerals (PCM). Depth-averaged values of Fe_{o-p} were generated by integrating profiles from several pits and dividing by the average soil thickness. Fe_{o-p} stocks were calculated using soil bulk density values reported in Wang et al. (2018, 2022)

Topographic curvature was estimated using 1-m resolution airborne lidar data (opentopography.org) and the wavelet-based curvature algorithm of Struble and Roering (2021) evaluated at the location of sampling pits. The standard deviation of curvature values at each site is <25%

Local erosion rates modeled using a linear transport model and soil transport coefficient, D, of 0.004 m² yr⁻¹, a typical value for forested settings (Perron 2017)

knickpoint in a tributary of the North Fork Feather River, California, to determine geomorphic controls on soil geochemistry, soil thickness, and SOC. The slow-eroding, above-knickpoint site (POMD) has thicker and more highly altered soils than the fasteroding, below-knickpoint site (BRC). The authors used catchment-averaged erosion rates to define a nearly 10×erosional gradient that distinguishes these two sites (POMD=0.036 mm year⁻¹ and BRC=0.25 mm year⁻¹, Wang et al. 2018), although we interpret the erosional gradient to be weaker owing to delayed hillslope adjustment at BRC (Hurst et al. 2012). Using local hilltop convexity calculated from wavelets and airborne lidar (Struble and Roering 2021) and a soil transport coefficient typical of similar forested settings (Perron 2017), we estimate that BRC erodes 3.5 × faster than POMD. Specifically, we calculate erosion rates of 0.087 and 0.023 mm year⁻¹ for BRC and POMD, respectively (Table 3). With respect to PCM abundance, the Wang et al. (2018, 2022) data show that BRC has less oxalate-extractable minus pyrophosphate-extractable iron (Fe_{o-p}), a commonly used proxy for PCMs, as well as a thinner soil mantle (Table 3). This observation is consistent



At the hillslope and catchment scale, analyses of soil thickness and SOC stocks using topography and geomorphic models have generated promising results (Patton et al. 2019), signaling the potential to leverage coupled process models for SOC prediction. Although our results are developed for a generic hilltop or ridge crest position, probabilistic soil transport simulations imply that soil residence time distributions are increasingly similar across hillslopes in the presence of vigorous soil mixing and absent transient adjustments in response to variations in channel incision (Furbish et al. 2018). From this perspective, our approach should have broad applicability to eroding landscapes in regions like the fertile Pacific Northwest, which contains substantial SOC stocks (Poggio et al. 2021). In western Oregon, for example, longterm erosion rate estimates generated from cosmogenic nuclides vary by orders of magnitude, reflecting the influence of variable rock uplift, knickpoint migration, and river network dynamics (Penserini et al. 2017). The resulting erosional gradients have generated systematic variations in channel steepness, hillslope angle, hilltop curvature, and soil thickness across a range of scales. Fortunately, these spatial patterns can be quantified in high resolution owing to publicly available airborne lidar data, which can be coupled with calibrated geomorphic transport models to predict erosion rate and soil thickness. In particular, erosion metrics based on hilltop convexity and



the steepness of debris flow channel networks have been tested and calibrated across the region (Penserini et al. 2017; Roering et al. 2007). These models are capable of generating erosion rate and soil thickness predictions at the pedon, hillslope, and catchment scales (Hurst et al. 2012; Roering et al. 2007), facilitating our ability to address some of the scale-dependence challenges associated with building and applying SOC inventory models (O'Rourke et al. 2015).

Conclusion

Primary minerals that enter soils through bedrock weathering generate PCMs that associate with SOC exerting a strong geochemical control on terrestrial carbon abundance and persistence. Erosion rate and soil thickness are first-order controls on soil weathering pathways that modify the abundance of PCMs available to associate with SOC. We propose a soil weathering mass balance approach that reveals an optimal erosion rate for potential SOC storage, corresponding to the maximum PCM concentration in a steadily eroding landscape. In rapidly eroding terrain, our results show that PCM abundance is limited by two factors: rapid cycling of primary minerals and thin soils. By contrast, in slow-eroding settings, PCM abundance is limited by the ripening of PCMs into clays and other crystalline phases, which are less conducive to SOC association. In these slow-eroding scenarios, the slow supply of primary minerals from bedrock relative to the pace of weathering processes is a critical control on the abundance of PCMs in soil. Our results provide a framework for assessing erosional controls on weathering and SOC abundance across a range of scales, from plot and hillslope scales up to entire catchments and even tectonic regimes, independent of differences in parent material or biological drivers of SOC dynamics. When coupled with geologic and biological datasets, topographic erosion metrics, and high-resolution topography, our approach enables mapping of PCM abundance across regional scales, facilitating our ability to predict SOC storage potential and forecast SOC response to disturbances.

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Data availability The authors declare that the data supporting the findings of this study are available through the code provided in the supplementary materials.

Declarations

Conflict of interest All of the authors declare that they have no conflicts of interest.

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