

Synergistic photothermal-thermoelectric-photovoltaic energy generation via a transparent spectral modulating solar module

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ABSTRACT

The efficiencies of photovoltaic (PV) and thermoelectric (TE) have been limited by the intrinsic materials properties. In solar applications, PV utilizes the shorter wavelength end of the solar spectrum while the longer portion (IR) can be photothermally (PT) converted to heat by special nano structures for TE electricity generation. We have developed spectral modulating nano systems made of $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{-xS}$ nanoparticles and chlorophyllin capable of synergistic PT-TE-PV solar harvesting and energy generation. In this system, solar light is harvested through the transparent photothermal spectral modulator (TPSM) and segregated into different wavelengths: the IR is absorbed to photothermally heat up the TE hot end for the required thermoelectric temperature span; the UV/visible is directed to PV with reduced IR for significantly reduced heating, thus enhanced power conversion efficiency (PCE). Upon removal of IR, the PV surface temperature can be lowered from 86.9 °C to 65 °C after 120 min solar irradiation, resulting in PCE increase from 16.3 % (dropped from 25.1 % at 0 min) to 19.4 %. A PT-TE-PV solar energy module (PTPSEM) is designed based on TPSM to synergistically generate energy by separately utilizing the full spectrum of solar light. Using the same solar light source, while generating electricity via PV, an additional 2.4 % is gained by a series of commercial thermoelectric generators (TEG), adding up to a total of 21.8 % system efficiency which significantly exceeds the average efficiency of the commercial silicon PV panel used in this study. The spectral modulation mechanism through TPSM is identified based on the optical behaviors of chlorophyllin and $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{-xS}$. Also discussed is the design concept of PTPSEM.

1. Introduction

The power conversion efficiencies of PV and TE have been limited by the intrinsic properties; respectively, ~ 25 % for PV and ~ 5–8 % for TE. The processes driving most photovoltaics utilize solar energy at wavelengths below about 1000 nm in the UV/Vis – NIR range of the solar spectrum. About 58 percent of the energy density of the solar spectrum lies in this wavelength range with the other 42 percent occurring at longer wavelengths. Conventional solar cells typically have substantial absorption in the sub-band-gap wavelength range above approximately 1000 nm [1]. This absorbed long wavelength solar energy contributes to heating of the PV panels resulting in significant efficiency decrease with temperature [2]. A typical temperature coefficient for PV output power is ~ 0.5 percent per °C [3]. A standard PV module, absorbing IR along with UV and visible solar radiation can operate 20 to 40 °C above ambient temperature leading to efficiency losses of up to 20 % [2–3]. Numerous researchers have investigated active cooling of PV modules to improve efficiency [4–7]. However active cooling increases system complexity and imposes energy costs for active elements such as pumps and fans.

The potential of combining PV and a thermoelectric generator (TEG)

has been previously explored [8–9]. In many cases the thermoelectric module is attached directly to the PV panel. Heat generated by PV in the IR band is converted to electrical energy by the thermoelectric. This approach suffers from the completely different temperature dependence of PV and thermoelectric efficiencies. Thermoelectric device output increases with temperature while PV output decreases with temperature. Thus combined PV and thermoelectric devices have an inherent drawback due to inverse temperature dependence.

Although the photothermal (PT) effects of nanoparticles have been extensively studied for medical therapy [10–19], a few studies on the transparent photothermal thin films and bulk materials have been reported for energy applications. We have recently developed transparent, spectral selective, and photothermal thin films based on porphyrins and iron oxides [20–26], capable of raising temperature up to 76 °C under simulated solar light [22]. The optical absorptions of the hybrid films are spectrally tuned with strong absorptions in the UV and IR regions but kept a minimum in the visible band for high average visible transmittance. In this fashion, the solar light can be harvested through multiple layers and converted to thermal energy in a 3D cuboid. The thermal energy generated in a confined volume can be utilized to produce electricity via thermoelectric devices for a variety of applications.

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Furthermore, the transparent hybrids can be spectrally tuned to selectively absorb solar light in the designated frequencies, such as UV and IR as a photonically optimized light source for PV applications.

In this study, we report on the development of the transparent photothermal spectral modulator (TPSM) that is engineered for multiple functions: (1) since the TPSM is transparent, the simulated solar light can pass through it as the light source for a commercial silicon PV panel; (2) meanwhile, the solar light (mainly IR) is absorbed by TPSM and converted to thermal energy via the photothermal effect; (3) the thermal heat is utilized for TEGs to produce electricity through multiple surfaces, and (4) the TPSM can spectrally selectively remove the IR portion for reducing PV surface heating resulting in enhanced photon conversion efficiency (PCE). In this fashion, we combine the PT, TE, and PV effects for synergistically producing thermal and electrical energies by unique materials design of TPSM.

The concept of PT + TE + PV module is schematically illustrated in Fig. 1. As shown in this figure, the TPSM is structurally tailored to specified absorptions from UV to IR for the most efficient 3D solar harvesting and energy generation. As the solar light is harvested by the transparent nano hybrids, most of the IR is absorbed (~42 %) and photothermally (PT) activated to heat up the hot end of TE (figure of merit of TE is proportional to temperature span). The UV/Vis portion (~58%) can pass through the TPSM for PV electricity generation. In this fashion, the entire spectrum of solar light is fully utilized for energy generation via combined effects of PV, PT, and TE. The spectral tunability of the TPSM relies on structural designs for spectral modulated absorptions. Further, TPSM can be rendered transparent enabling white light passing through multilayers, prismatic solids, or a cube, therefore, transforming solar harvesting from 2D to 3D with significantly increased energy density.

We developed a unique TPSM containing either $\text{Fe}_3\text{O}_4 @ \text{Cu}_{2-x}\text{S}$ nanoparticles or chlorophyllin which is highly transparent, spectral selective, strongly photothermal, and capable of absorbing and converting designated photon energies for the most efficient energy generation via the PT, TE, and PV effects. Various PT + TE + PV solar energy modules (PTPSEM) were designed based on TPSM as shown in Fig. 2. The basic design concept is depicted in Fig. 2a. As shown in this figure, the simulated solar light passes through transparent TPSM for two purposes: 1) Photothermally generating thermal energy for heating up the hot end of the thermoelectric generator (TEG). Due to the temperature difference between the hot and cold ends of TEG, electricity is generated mainly by utilizing the IR portion of the solar light. 2) Since TPSM is transparent, simulated solar light can pass through it, reaching the PV panel below for electricity generation. In this fashion, the same solar light source can be spectrally utilized to produce electricity

simultaneously by TE and PV through PT. Furthermore, the IR portion is absorbed by TPSM (converted to heat for TEG), resulting in reduced heating on the PV panel surface.

Based on the concept depicted in Fig. 2a, we designed different PTPSEMs for optimized solar harvesting and energy generation (both thermal and electrical energy). Considering light power attenuation through the TPSM due to its thickness (Fig. 2a), we designed a 5-sided box made of TPSM with a circular opening on the top side. As shown in Fig. 2b, the center beam of the simulated solar light can directly reach the commercial silicon PV panel (Fig. 2c) through the top circle opening for generating electricity with the specified PV PCE. However, the rim of light in fact passes the 2 mm thick top cover (Fig. 2b) with portion of IR removed to reduce the PV surface heating (the so-called spectral modulation). Meanwhile, the 4-side walls of TPSM are photothermally activated to heat up the hot ends of 4 PTGs for generating electricity (Fig. 2d).

The overall structure of PTPSEM is schematically illustrated in Fig. 2d. It can be seen in this figure that the solar light is spectrally utilized via TPSM for efficient energy harvesting and generation. The TPSM is composed of either $\text{Fe}_3\text{O}_4 @ \text{Cu}_{2-x}\text{S}$ or chlorophyllin, both are strongly photothermal and spectral selective with unique optical characteristics. The PTPSEM can synergistically generate energy through the photothermal, thermoelectric, and photovoltaic effects according to spectrally modulated photon frequencies: IR is absorbed by TPSM to heat up TEGs; removal of IR from solar light reduces PV panel heating for enhanced PCE, and the same light source with most of UV + Vis (IR removed) is utilized for PV.

2. Experimental details

2.1. Materials synthesis

The following chemicals were used in this research: iron (III) acetylacetonate ($\text{Fe}(\text{acac})_3$, ≥99.9%), copper (II) acetylacetonate ($\text{Cu}(\text{acac})_2$, ≥99.9%), oleylamine (70%), sulfur (99.998%), N-methyl-2-pyrrolidone (NMP, 99.5%), chloroform (≥99.9%), tetrahydrofuran (THF, 99.9%), isopropyl alcohol (IPA, 99.7%), and polyethylene glycol (PEG); all were purchased from Sigma-Aldrich Inc. Cyclohexane (99.5%) was purchased from Tedia Inc.

$\text{Fe}_3\text{O}_4 @ \text{Cu}_{2-x}\text{S}$ nanoparticles were synthesized by following a previous report [23]. A certain amount of oleylamine was heated to 300 °C in nitrogen gas environment for a period. Meanwhile, amount of $\text{Fe}(\text{acac})_3$ was dissolved in the NMP/oleylamine solution (4:3, v/v) which was added into a preheated oleylamine to form Mixture A. The Mixture A was stirred for 10 min at 300 °C and then cooled down to 70 °C for 10

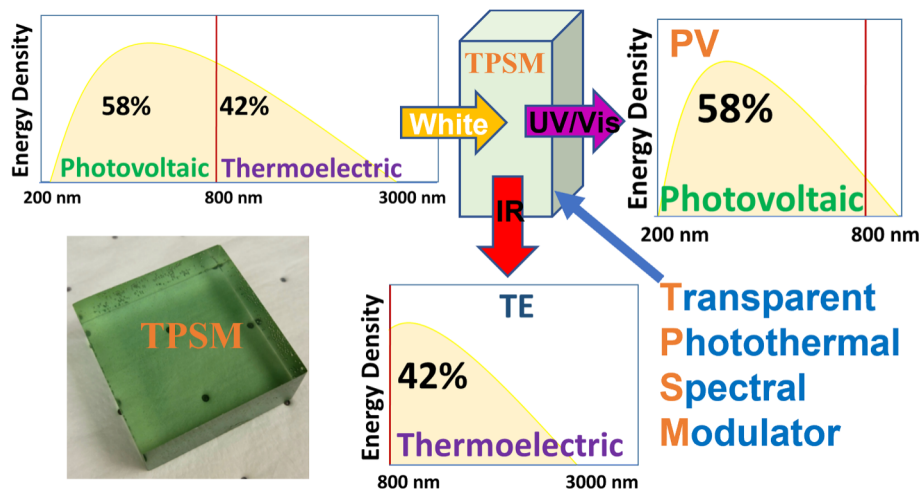


Fig. 1. Schematic diagram showing the spectral tunable nano hybrid as a wavelength segregator and PT heater.

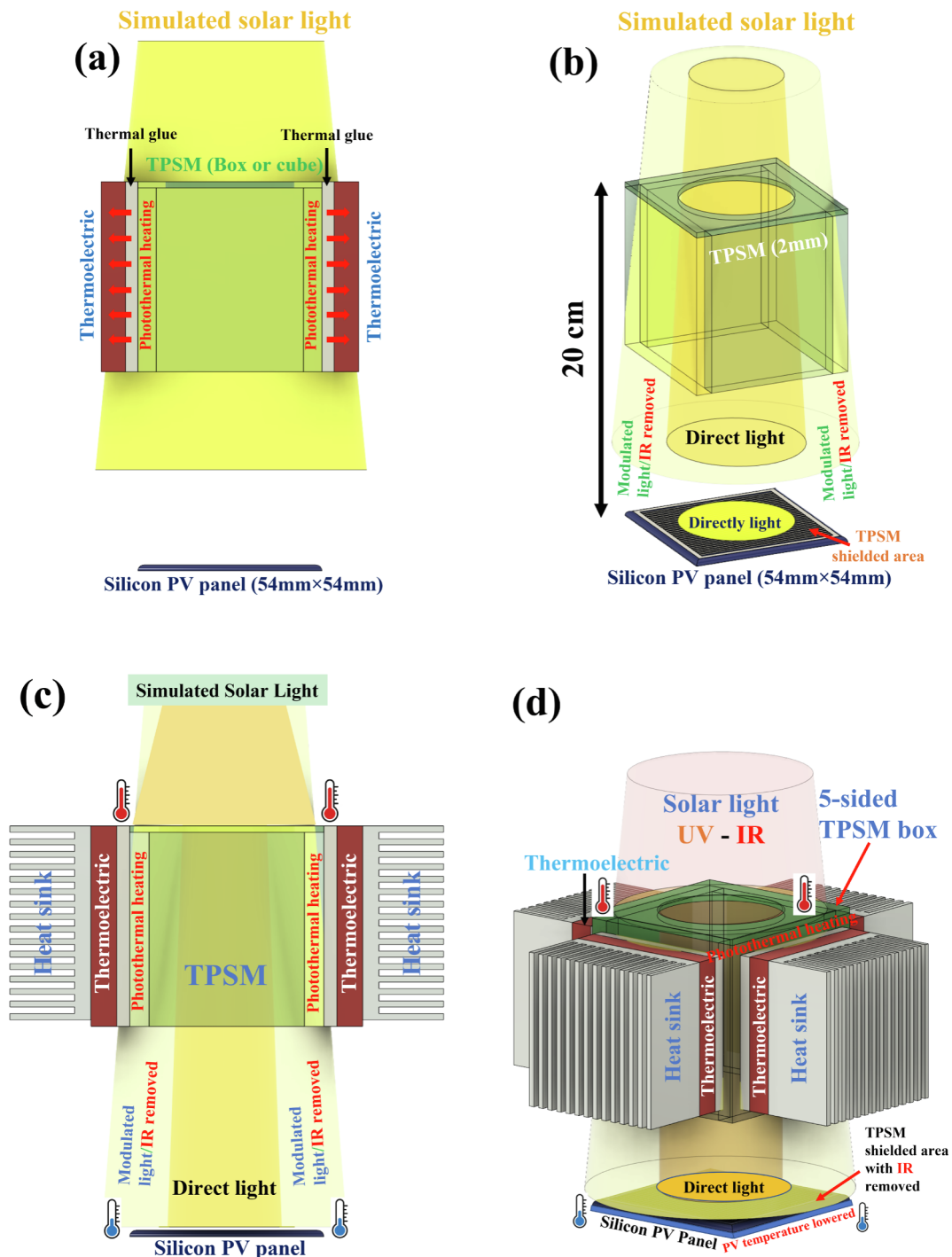


Fig. 2. Schematic diagrams showing (a) the basic concept of PT - TE - PV solar energy module (PTPSEM) consisting of a transparent photothermal spectral modulator (TPSM), the thermoelectric generators (TEG) on the side walls of TPSM, and a commercial silicon PV panel at the bottom which can receive solar light that is spectrally modulated by transparent TPSM. The simulated light first passes the TPSM, generating heat that is transferred to TEGs, then shines on the PV panel; both TEG and PV synergistically generate electricity via utilizing full solar spectrum. (b) To enhance light intensity on the PV surface, a 5-sided box is made of TPSM with wall thickness of 2 mm. A circular opening on the top wall fits to let solar light directly reach PV with high photon intensities. (c) The rim of light will pass through the 2 mm thick TPSM wall on the top to remove portion of IR for reducing PV surface heating. (d) 3D overview of the PTPSEM with four TEGs on the side walls of the 5-sided TPSM box and a commercial PV panel at the bottom, receiving the full spectrum of the solar light in a spectral selective fashion.

min. Amount of sulfur was dissolved in a oleylamine/cyclohexane mixture (6:5, v/v) to form Mixture B. Mixture B was slowly injected into Mixture A to form Mixture C at 70 °C and stirred for 10 min. $\text{Cu}(\text{acac})_2$ was dissolved in the oleylamine/chloroform mixture (1:4, v/v) to form Mixture D which is made of $\text{Cu}(\text{acac})_2$ in oleylamine/cyclohexane mixture (6:5, v/v). Mixture D was then injected into Mixture C. After stirring at 70 °C for 30 min, the $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ nanoparticles were

obtained. The $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ nanoparticles were collected by a strong magnet. The collected nanoparticles were washed with methanol and then dried by oven under vacuum. Finally, $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ nanoparticles were dispersed in THF for later use.

Epoxy resin and silicon molds were used to develop TPSM with various geometries. $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin with different ratios and concentrations were dispersed in epoxy resin and well mixed to

ensure homogeneity. The mixed solution was then poured into a silicon mold and cured for 24 h. The concentrations of the bulk samples were controlled at $200 \mu\text{g}/\text{cm}^3$. Fig. 3 shows the photographs of cubic TPMS made of chlorophyllin (Fig. 3a) and $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ (Fig. 3b). As shown in this figure, the bulk TPMSs of both chlorophyllin and $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ appear to be transparent and homogeneous with respective light green (Fig. 3a) and amber (Fig. 3b) colors. Solar light can pass through it depending on the transparency and thickness. Using $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ or chlorophyllin, the TPMS was made into 5-sided boxes as shown in Fig. 2b. As noted above, the concentration of $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ or chlorophyllin in the TPMS box was controlled at $200 \mu\text{g}/\text{cm}^3$.

2.2. Materials characterization

The TPMS samples were characterized for full-spectrum absorption and transmittance. Maintaining high AVT is critical to ensure significant light power density on the PV cell surface. In this study, we focused on characterization of the porphyrin compounds and iron oxides, namely, chlorophyllin and $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$. These compounds exhibit unique absorption characteristics providing a versatile optical base for making TPMSs in a wide range of photon frequency. To observe their intrinsic optical behaviors, the absorption and transmittance spectra were obtained from the solutions of these compounds to avoid the bulk effects. Fig. 4 shows the absorption spectra of chlorophyllin and $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$. The concentrations of chlorophyllin and chlorophyllin were both fixed at $0.01 \text{ mg}/\text{mL}$.

As shown in Fig. 4a, chlorophyllin has an intensive absorption peak at 404 nm and a small one at 630 nm forming a so-called “saddle-like” spectrum. This is the typical absorption characteristic of the porphyrin compound which is particularly useful for spectral selective solar harvesting and generating photothermal energy, while remaining high AVT. As shown in Fig. 4a, the strong peaks at 404 nm and 630 nm indicate considerable absorptions in the UV and NIR regions for conversion of photons to thermal energies. Low absorptions in the visible region render the porphyrin highly transparent for making TPMS (Fig. 3) in the PT – TE – PV solar energy module (Fig. 2).

The $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ nanoparticles are known for their IR absorptions. Fig. 4b shows the absorption spectra of both Fe_3O_4 and $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ nanoparticles in solutions for comparison. As shown in this figure, Fe_3O_4 typically exhibits a strong absorption in the UV region and quite rapidly subsides over the visible and IR regions. By coating a thin layer of Cu_2S on the Fe_3O_4 nanoparticles, the $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ solution exhibits a broad absorption peak extending to 1400 nm. This strong IR absorption can be used to reduce the PV surface heating from the solar light. The IR

absorption of $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ has been attributed to the localized surface plasmon resonance (LSPR) of Cu_2S in the IR region [27]. The TPMS made with the $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ nanoparticles can spectrally modulate the solar light by absorbing most of IR which contributes to PV surface heating.

2.3. Photothermal, Thermoelectric, and photovoltaic characterization experimental details

An AOSHIKE Micro Solar Cell with $54 \text{ mm} \times 54 \text{ mm}$ (bought from amazon) was used in this study. The measurement setup for I – V characterization is schematically shown in Fig. 5a. A solar simulator (Newport) was used as a light source for a duration of 120 min for each experiment so that the heating effect was recorded. The solar cell surface temperature was monitored by using a thermal camera (FLIR, Inc.). The I – V curve measurement was performed by using Keithley 2400 instrument. A deep thermoelectric generators (size: $40 \text{ mm} \times 40 \text{ mm} \times 3.6 \text{ mm}$ with Bi_2Se_3 as thermoelectric elements, bought from amazon) were used to convert heat generated by the 5-sided TPMS box (Fig. 2d) into electricity. TEGs were attached onto the 4 side surfaces of the TPMS box (Fig. 2d) after 120 min heating. ΔT was calculated by using the temperature difference between the hot and cold ends of TGE which was measured by thermal couples.

The efficiency of solar cell can be calculated using the following equation:

$$\eta_{PV} = \frac{P_{out}}{P_{in}} = \frac{I_{sc} V_{oc} FF}{P_{in}} \quad (1)$$

where P_{out} is output power, P_{in} is input power, I_{sc} is short-circuit current, V_{oc} is open-circuit voltage, I_{MP} is current density at maximum power, V_{MP} is voltage at maximum power, P_{max} is the maximum output power, and FF is fill factor. It should be noted that P_{in} used to determine PCE is the light power density on the PV panel without passing through the TPMS (Fig. 2). Due to varied light transmittances of the TPMS, the light power density on the PV surface also varies.

The thermoelectric measurement setup is schematically shown in Fig. 5b. The efficiency of thermoelectric generator is calculated by

$$\eta = \frac{P_{TE}}{Q_{in}} \quad (2)$$

where P_{TE} (W) is power of TE generator, Q_{in} (J) is the power input of the solar simulator.

P_{TE} is calculated based on the output on a 10Ω resistor, the current I (amp), and voltage V (volt) of the circuit (Fig. 5b). The output of TEG can be written as:

$$P_{TE} = I \times V \quad (3)$$

Seebeck coefficient of the combined TEGs can be calculated as:

$$S = \frac{E}{\Delta T} = - \frac{V}{T_H - T_C} \quad (4)$$

where E is the voltage generated and ΔT is the temperature difference between the hot and cold ends of the TEG.

Figure of merit (ZT) can be calculated by the expression:

$$ZT = \frac{S^2 \sigma}{\kappa} T \quad (5)$$

where κ is thermal conductivity of TEG, σ is electrical conductivity, and T is the absolute temperature.

Power Factor (PF):

$$PF = S^2 \sigma \quad (6)$$

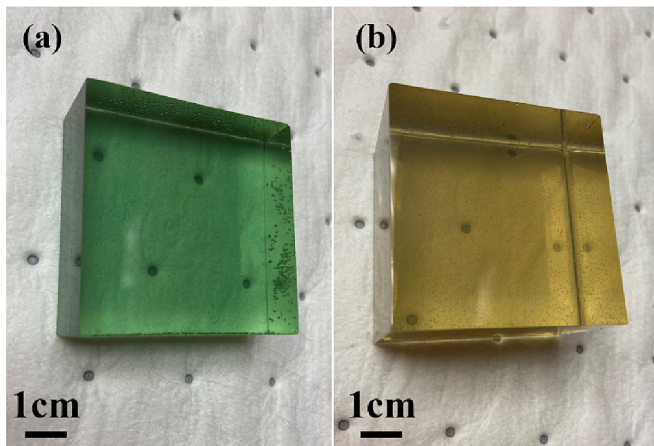


Fig. 3. Photograph of a transparent photothermal spectral modulator (TPSM) with (a) chlorophyllin and (b) $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ nanoparticles in epoxy resin. The concentration of $\text{Fe}_3\text{O}_4/\text{Cu}_2\text{S}$ or chlorophyllin in the TPMS box was controlled at $200 \mu\text{g}/\text{cm}^3$.

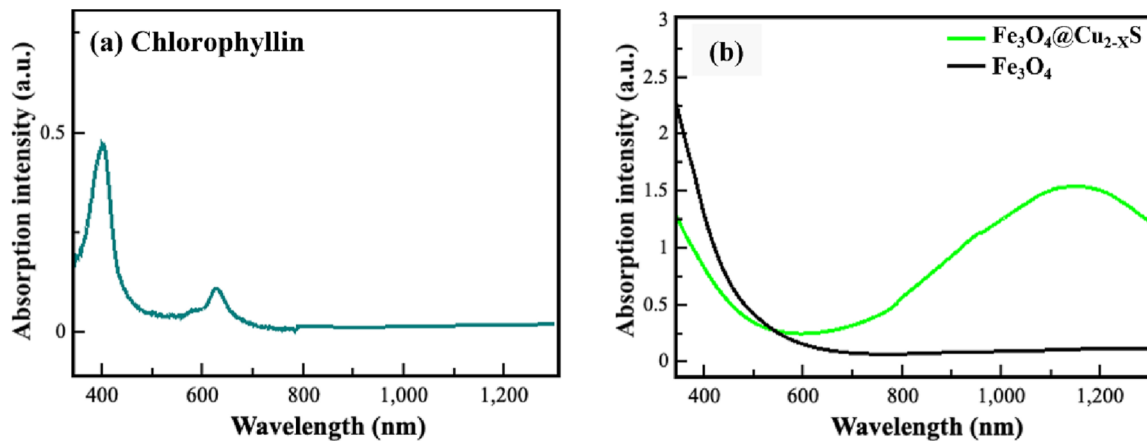


Fig. 4. Absorption spectra of (a) chlorophyllin, and (b) Fe_3O_4 and $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$.

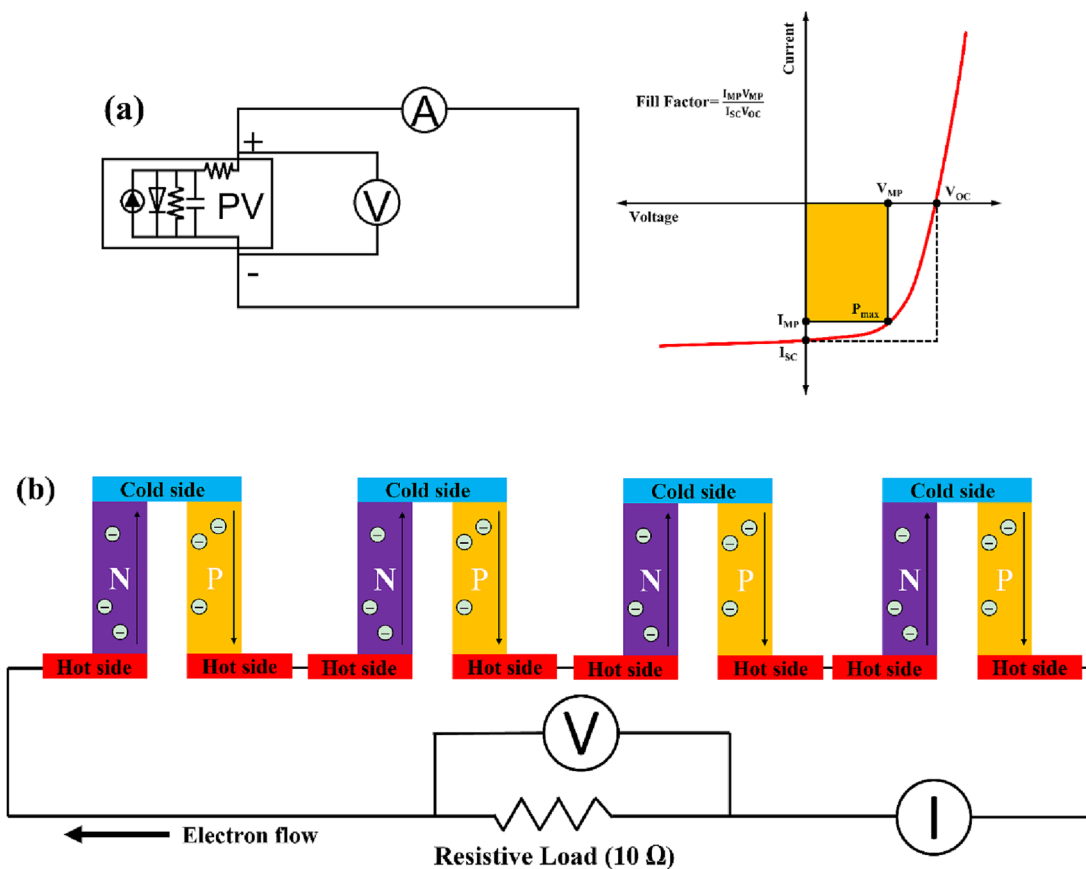


Fig. 5. Schematic diagrams showing (a) I – V curve measurement set up, (b) TEG measurement set up.

2.4. Characterization of the TPSEM cube with $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin

Fig. 6a shows the PTPSEM experimental set up with four commercial TEGs on the four faces of a TPSEM cube (dimensions: $5\text{ cm} \times 5\text{ cm} \times 5\text{ cm}$). The TPSEM cube is composed of $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin dispersed in the matrix of epoxy resin with a concentration of 0.01 mg/ml . Upon simulated solar light irradiation (1.58 W), the TPSEM is photothermally activated to convert incoming photons to thermal energy. As shown in Fig. 6b, the highest temperatures of the side walls can reach 66.6°C and 64.9°C , respectively for the $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin TPSEM cube. As shown in this figure, due to the photothermal effects of $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin in the TPSEM cubes, the

temperatures rise quite rapidly upon solar light irradiation (1.58 W). The solar light is turned off at 120 min; therefore, the temperatures start to drop off thereafter. With these significant temperatures, the four TEGs on the 4-side walls of TPSEM cube generated TE voltages of 1.864 V and 1.845 V (Fig. 6c), and TE powers of 0.189 W and 0.187 W (Fig. 6d), respectively, for $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin TPSEM cubes. However, due to heat transfer between the hot and cold ends of TEGs, the temperature differences, ΔT , are reducing from above 43.6°C to 16.8°C as shown in Fig. 6e. Consequently, both voltages and powers are also reduced. Notice that the drops in voltage and power become much less rapid after 3 min (Fig. 6c and 6d). The TE efficiencies with 4 TEGs are 11.9% and 11.8% respectively for the $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin TPSEM cubes when ΔT is at the maximum as shown in Fig. 6f.

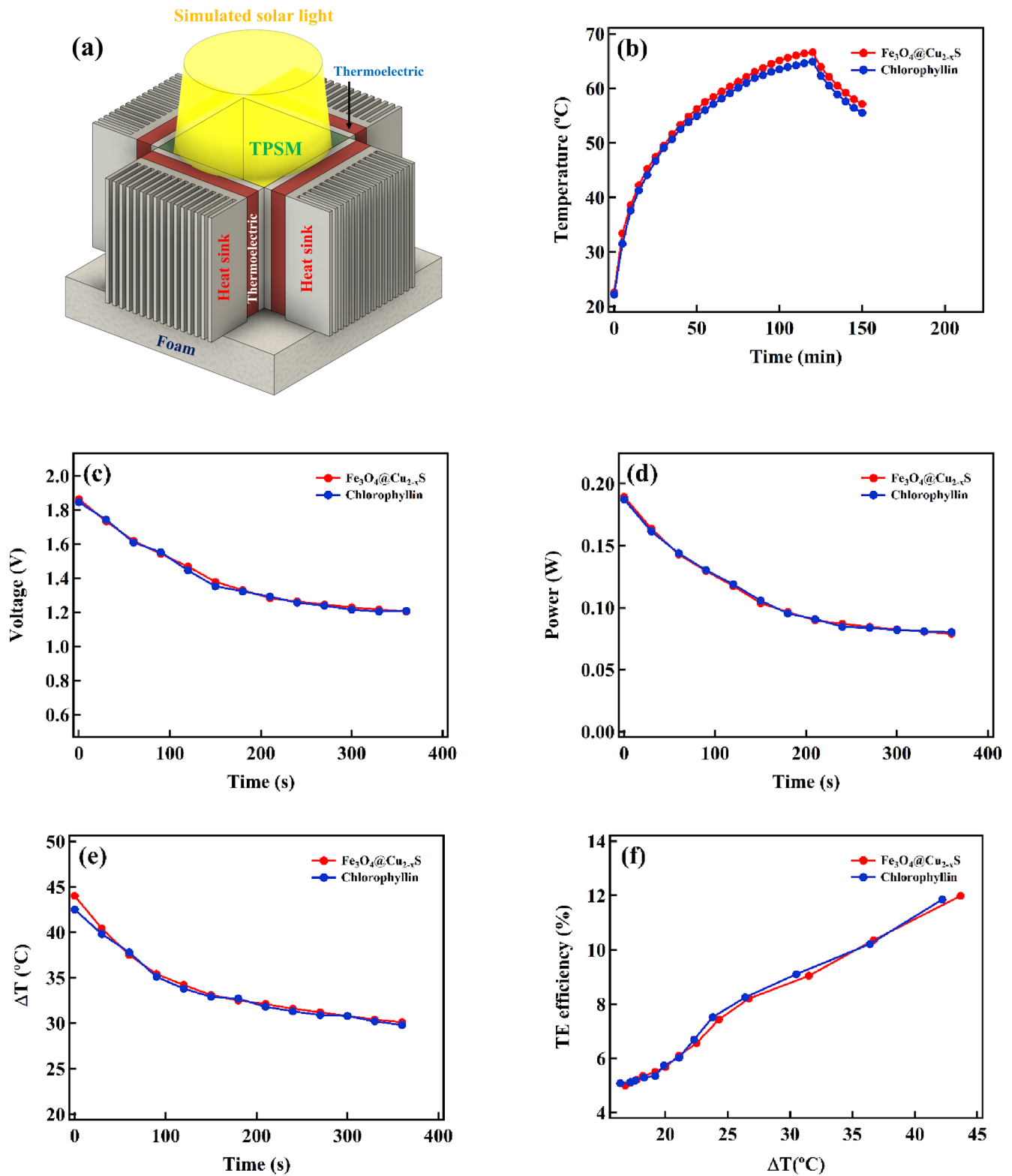


Fig. 6. (a) Schematic diagram showing the PTPSEM experimental set up with four commercial TEGs on the four faces of the TPMS cube. (b) The cube side wall surface temperature vs time (heating curves) for the TPMS with $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin dispersed in the epoxy resin matrices (Fig. 3); (c) The TE voltages generated by TEGs on the 4-side walls of the TPMS cube (in series) with $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin dispersed in the epoxy resin matrix (Fig. 3); (d) TE power outputs generated by TEGs on 4-side walls of the TPMS cube (in series) with $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin dispersed in the epoxy resin matrix (Fig. 3); (e) the temperature differences, (ΔT), between the hot and cold ends of TEGs, and (f) TE efficiencies as function of temperature differences, (ΔT), for the TPMS cubes with $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin dispersed in the epoxy resin matrix.

2.5. TE output power & efficiency of patterned PT box

Fig. 7 shows photothermal and thermoelectric characterization of the 5-sided TPSM boxes (Fig. 2b). As shown in Fig. 7a, the highest temperatures of the side wall surfaces are 41.4 °C and 41.3 °C, respectively for the TPSM made of $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ and chlorophyllin. These values are considerably lower than those of the TPSM cube due to

smaller thickness and overall volume (photothermal energy is an extensive property). The voltages generated by the TEGs are 0.828 V and 0.816 V, respectively for the $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ and chlorophyllin TPSM boxes, which are also lower than those generated by the TPSM cube. The TE powers are 0.038 W and 0.037 W, respectively for the $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ and chlorophyllin TPSM boxes as shown in Fig. 7c. Consistently, both voltage and power decrease with the decreasing of temperature

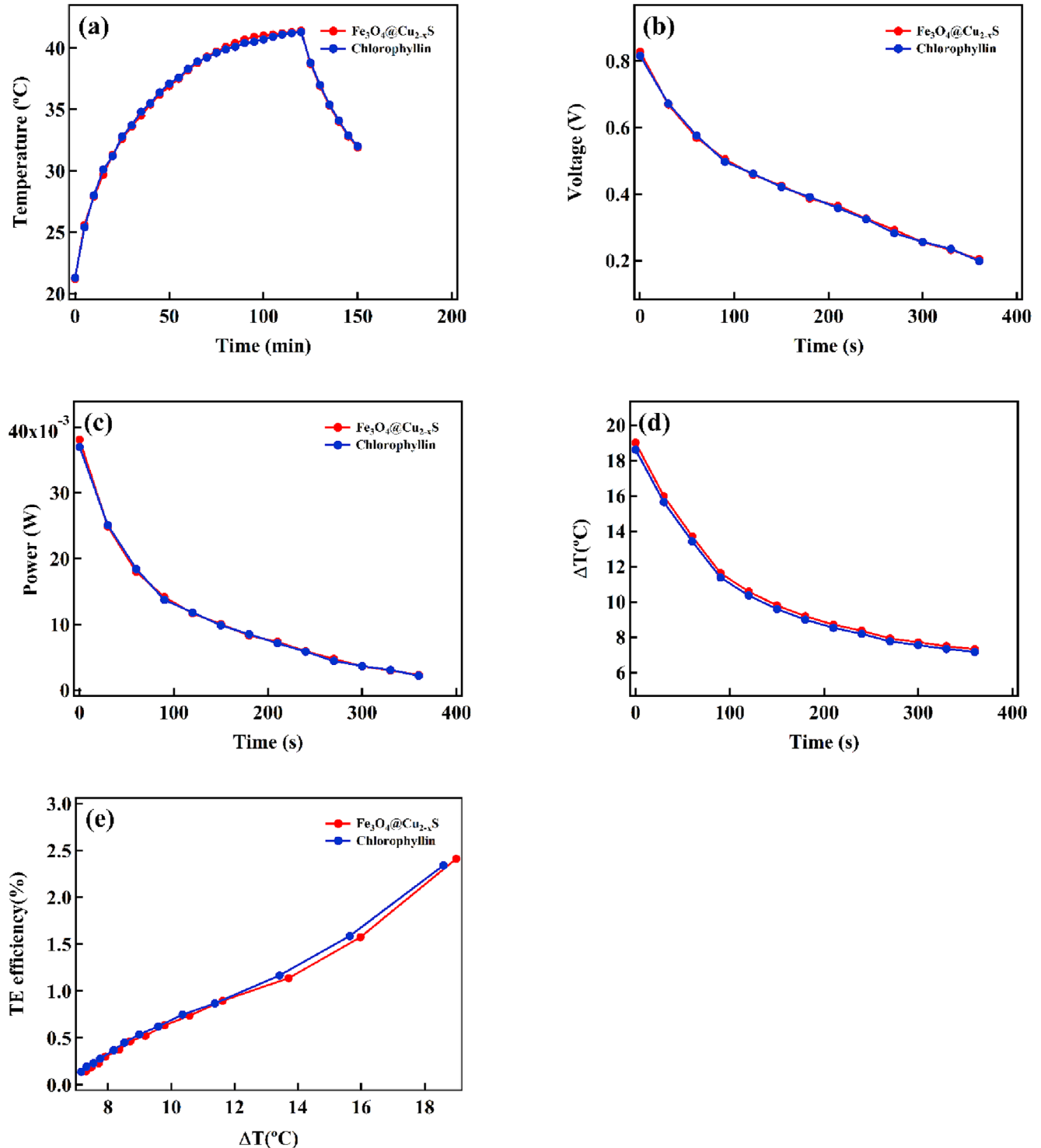


Fig. 7. (a) Temperature vs time for the photothermal heating curves of the side wall surfaces of the 5-sided TPSM box with $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ or chlorophyllin dispersed in the resin matrices (Fig. 2b); (b) The TE voltages generated by TEGs on the 4-side walls of the TPSM box (in series) with $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ or chlorophyllin dispersed in the resin matrix (Fig. 2d); (c) TE power outputs generated by TEGs on the 4-side walls of the TPSM box (in series) with $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ or chlorophyllin dispersed in the resin matrix (Fig. 2d), and (e) The TE efficiencies as function of temperature differences, (ΔT), between the hot and cold ends of TEGs.

difference, ΔT , between hot and cold ends of the TEGs as shown in Fig. 7d. The TE efficiencies are 2.405 % and 2.341 % respectively, for chlorophyllin and $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ boxes when ΔT is the maximum as shown in Fig. 7e.

2.6 I – V curves of silicon PV panel

Fig. 8 shows the I – V curves of the silicon solar panel underneath the 5-sided TPSM box (Fig. 2b) with $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin dispersed in the resin matrices. The inset of Fig. 8a indicates that the center light can directly reach the silicon solar panel through the circular opening on the top wall of the TPSM box with appreciable light power and full spectrum from UV to IR (also see Fig. 2b and 2c). Since the solar light spot has a diameter that is larger than the circular opening, the rim of light can pass through the green area of the top side TPSM wall which is transparent. TPSM contains either $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin, both exhibit some NIR/IR absorptions (Fig. 4). The portion of light passing through the green area will be spectrally modulated with the IR removed, resulting in reduced PV panel surface heating for enhanced PCE.

To investigate the heating effect due to solar irradiation, the I – V curves were taken at 0 min (Fig. 8a) and at 120 min (Fig. 8b). As shown in Fig. 8a, the V_{OC} values of the PV panels do not change significantly with different TPSMs due to insignificant heating at the beginning. However, I_{SC} at 0 min without TPSM is slightly larger (0.197 A) than those with TPSM since the light power is reduced when passing through the top side wall edge (the green area marked in the inset of Fig. 8a). At 120 min (Fig. 8b), I_{SC} without TPSM does not change significantly and remains at 0.193 A. The I_{SC} values at 120 min with different TPSMs are slightly reduced to 0.150 A and 0.148 A, respectively for $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin (Table 1). However, one can see appreciable increases in V_{OC} with $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin TPSMs. As shown in Fig. 8b, while V_{OC} is 1.95 V for the silicon PV panel without TPSM, the V_{OC} value respectively increases to 2.32 V for $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and 2.34 V for chlorophyllin. The pronounced V_{OC} enhancement is attributable to the PV panel surface temperature reduction via IR absorption by TPSM (the green area indicated in the inset of Fig. 8b). The enhanced V_{OC} compensates for the I_{SC} loss associated with the reduced light power by TPSM. As shown in Table 1, the Fill Factor (FF) of the silicon solar panel with TPSMs are respectively 0.850 for $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and 0.846 for chlorophyllin, all significantly higher than that without TPSM (0.659).

2.7 Power conversion efficiency (PCE) of silicon PV panels with TPSMs

Fig. 9 shows the PV panel surface temperature profiles, PCEs, and PV

Table 1

I_{SC} , V_{OC} , I_{max} , V_{max} , FF, P_{max} , and P_{in} of silicon solar panel with different TPSMs at 120 min.

Modulator	I_{SC} (A)	V_{OC} (V)	I_{max} (A)	V_{max} (V)	FF	P_{max} (W)	P_{in} (W)
Without TSM	0.193	1.95	0.156	1.59	0.659	0.248	1.58
$\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ box	0.150	2.32	0.150	1.97	0.850	0.296	1.58
Chlorophyllin box	0.148	2.34	0.151	1.94	0.846	0.293	1.58

power outputs with or without TPSM. The solar light modulation design in the PT – TE – PV module is schematically illustrated in Fig. 2c. As shown in this figure, the simulated solar light can reach the silicon PV panel in two ways: 1) the center light can directly reach the PV panel through the circular opening on the top wall of the 5-sided TPSM box. This is to ensure high light power on the surface of PV panel for generation of electricity. 2) the solar light spot with greater diameter than the circular opening passes the TPSM first (the green area in the inset of Fig. 8a,b), then reaches the PV panel. This part of the light is in fact spectrally modulated by TPSM with some UV and IR portions absorbed due to the optical characteristics of $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin (Fig. 4).

As shown in Fig. 9a, the solar panel surface experiences significant temperature increases upon solar irradiation without TPSM; the temperature rapidly increases to 86.9 °C only after 15 min and remains a plateau thereafter. The light power is turned off at 120 min resulting in sharp fall of temperature. However, with a 2 mm TPSM on the top side of the 5-sided box (inset of Fig. 8a and Fig. 2c), the PV panel surface temperatures are significantly reduced to 65.8 °C and 64.4 °C, respectively with the $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ and chlorophyllin TPSM (Fig. 9a).

As is well known, PCE is negatively dependent on the temperature with a temperature coefficient of 0.25 % per °C [28]. As shown in Fig. 9b, PCE of the silicon panel without PTSM suffers significant decrease with increasing temperature; initially at 25.1 % (before heating starts at 0 min) and drops to 16.3 %, at PV panel surface temperature of 86.9 °C (after 120 min of solar irradiation). This is a typical PCE temperature dependent behavior of most of the PV panels. However, with PTSM of either $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ or chlorophyllin, one can see appreciable gain in PCE up to 19.4 % even at 120 min (Fig. 9b) since the PV panel surface temperature has been reduced to 65.8 °C (Fig. 9a). This pronounced temperature reduction is associated with part of the solar light being spectrally modulated with the IR absorbed by the TPSM, (IR is responsible for surface heating). As can be seen in Fig. 9b, the maximum PCE with TPSM of $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$ has reached 19.4 % after 120 min solar

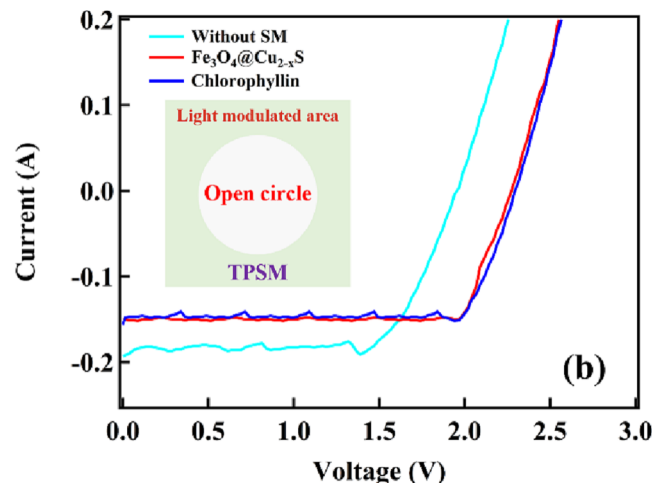
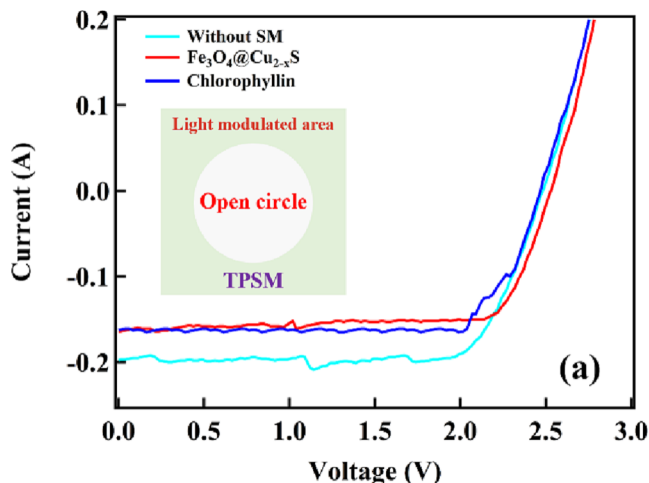


Fig. 8. I – V curves of silicon solar cell at different times (a) 0 min and (b) 120 min.

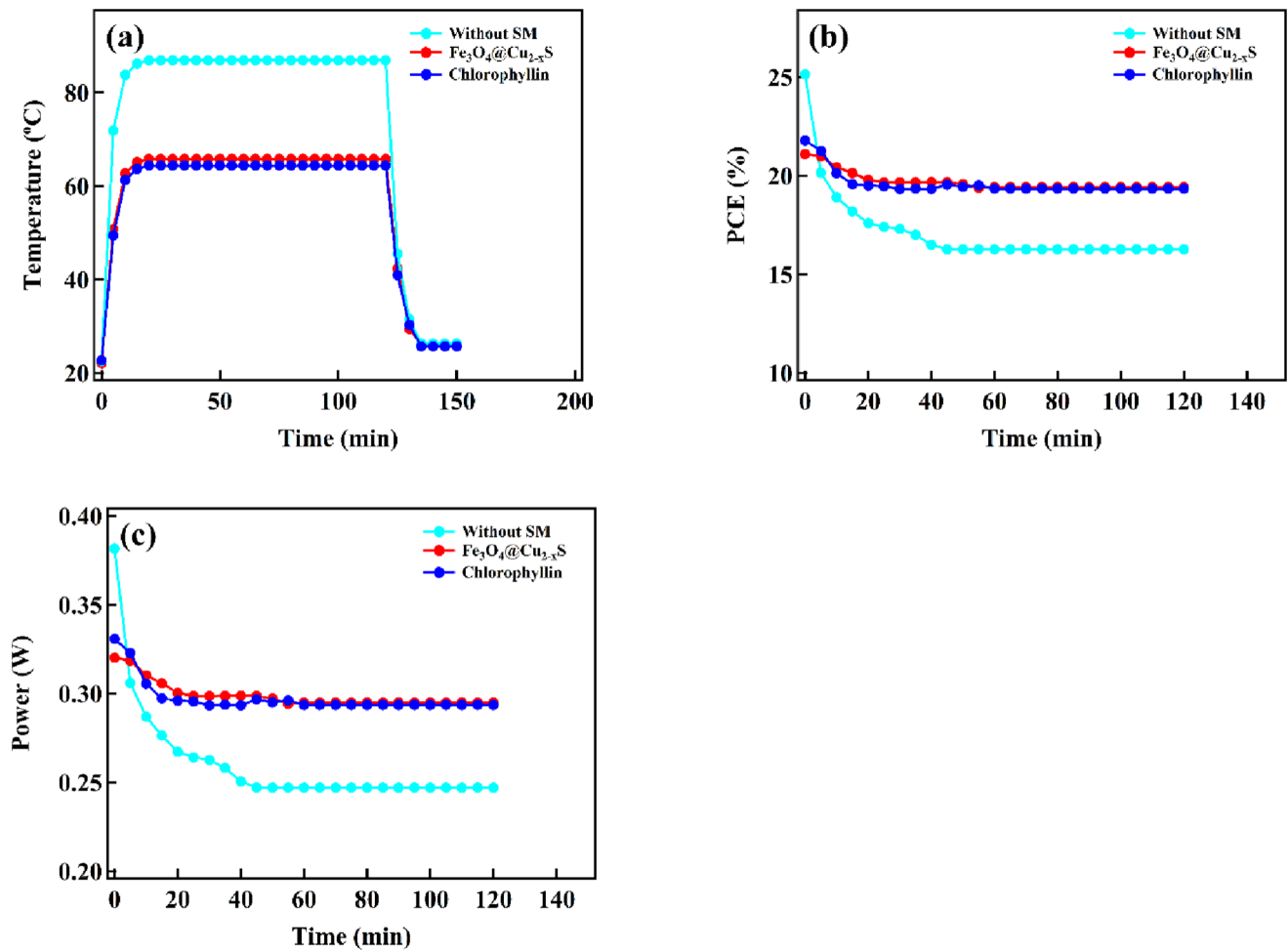


Fig. 9. (a) The commercial silicon PV panel surface temperature vs time with and without TPMS of chlorophyllin and $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$. (b) PCE of the commercial silicon PV panel vs time with and without TPMS of chlorophyllin, and $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$. (c) Power output of the commercial silicon PV panel vs time with and without TPMS of chlorophyllin and $\text{Fe}_3\text{O}_4@\text{Cu}_{2-x}\text{S}$.

irradiation. PCE with chlorophyllin TPMS remains at 19.3 % at 120 min. It is to be noted that the PCEs with TPMSs are all lower than that without TPMS at the beginning due to reduced incoming light power. The PCE values of those with different TPMSs remain almost constant for prolonged solar irradiation (120 min). This is critically important from an application point of view that considerable PCE can retain even under long-term solar irradiation.

The PV output power (OPP) shows similar behaviors as shown in Fig. 9c. As can be seen in this figure, OPP falls sharply to 0.24 W due to temperature increase without TPMS. However, they remain plateau near 0.3 W when portion of the solar light is spectrally modulated by removal of IR. Since all TPMSs exhibit considerable NIR or IR absorptions, the heating effect is reduced, responsible for reduced temperature dependences of the OPPs as compared to the silicon solar panel without TPMS.

3. Discussion

The experimental results from this study indicate high possibilities of fully utilizing solar light for energy generation by the PT-TE-PV solar energy module (PTPSEM) consisting of a transparent photothermal spectral modulator (TPSM), a series of thermoelectric generators (TEG), and a commercial silicon PV panel. The realization of synergistic solar energy generation via PTPSEM critically relies upon optimum distributions of solar light according to the photon frequencies. As described above, the PV efficiencies have been limited by the solar spectrum

having UV and sub-band-gap absorptions with nonideal spectral responses; while the high energy photons contribute to thermalization, the IR portion is responsible for PV heating. Thus, the design concept of this study is to spectrally segregate UV-Vis (for PV) and IR (for TE) through porphyrins and iron oxides with their unique optical properties (saddle-like and U-shaped spectra).

Using the PTPSEM shown in Fig. 2d, the PT-TE efficiency is calculated based on the ratio of incoming light power to the total output electricity of the TEGs. For the set up shown in Fig. 6a with a TPMS cube, the PV PCE is 0 because the cube blocks all the incoming photons due to its large thickness. However, the TPMS cube generates sufficient thermal energy raising its box side surface temperature up to 66.6 °C, resulting in a PT-TE efficiency of 11.9 % and output power of 0.189 W with 4 TEGs (in series) attached onto the 4-side surfaces (Fig. 6a). This implies 11.9 % of the incoming photons is converted into electricity through PT-TE.

To fully utilize solar light via both TE and PV, we designed a 5-sided TPMS box as shown in Fig. 2b and achieved the box side temperature of 41.4 °C (this temperature is lower than that of the TPMS cube due to reduction in PT materials mass and thickness). However, even with this temperature, the PT-TE efficiency reaches up to 2.4% with 4 TEGs (Fig. 2d), generating electric power of 0.038 W; this means 2.4% of the incoming photons are converted to electricity by the TEGs, which is reasonable as compared to the commercial TEGs. In this fashion, the solar light is harvested through two channels of electricity production, namely, TE and PV by photothermally and spectrally utilizing different photon energies through TPSE (IR for PT/TE, and UV-Vis for PV). The

combined PT – TE – PV efficiency reaches up to 21.8 % as shown in Table 2. Note that this efficiency is achieved upon prolonged solar irradiation of 120 min by suppressing PV surface temperature using TPSM, which has significant implications in PV applications.

The factors that influence PCE (and power) are majorly from thermalization associated with photons near the UV region around 400 nm (~ 3 eV) and sub – band – gap absorption above 1127 nm (1.1 eV) which contributes to IR heating under prolonged solar irradiation. If the high-energy photons can be partially absorbed by the spectral modulators, the thermalization effect can be reduced, leading to enhanced PCE under modulated solar light. Both chlorophyllin and $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ have strong absorption peaks near UV; while the former is near 410 nm, the latter occurs at much shorter wavelength near 300 nm. Spectral response decreases at small photon wavelengths [3–4,29–31]. Below 400 nm, those photons with energies above 3.0 eV are not utilized for generating electron – hole pairs. Therefore, photons with energies near 4.1 eV (300 nm) have insignificant effects. However, above 400 nm (3.1 eV), spectral response rapidly increases, contributing the thermalization. Since both chlorophyllin and $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ have strong peaks near UV, considerable photons near 3.0 eV are absorbed, resulting in flowered thermalization.

On the other hand, the increased PCEs of silicon solar panel are also associated with the IR absorptions by chlorophyllin and $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ that are incorporated into the matrices of TPSM. As shown in Fig. 9 and Table 1, although the I_{SC} values are compromised due to flowered incoming solar light power by TPSM, the V_{OC} values at 120 min are enhanced resulting from the reduced PV panel surface temperature. V_{OC} is dependent on temperature in association with dark saturation current I_0 . V_{OC} is logarithmically related to I_0 : $V_{\text{OC}} \sim \ln(I_{\text{SC}}/I_0)$, where I_0 is proportional to the intrinsic carrier concentration n_i which is a thermally activated parameter given by the Boltzmann relationship: $n_i \sim \exp(-E_g/kT)$. At elevated temperatures, V_{OC} will decrease due to increased I_0 and n_i . The bandgap of a semiconductor decreases with increasing temperature due to reduction in bond energy. Smaller energy gap (E_g) will enable more electrons to overcome E_g thus increasing n_i and I_0 resulting in flowered V_{OC} [32–38]. The temperature profiles of the PV panel under various TPSMs indicate consistent temperature reduction. As a result of 22.5 °C temperature drop, the V_{OC} has increased to 2.32 V, responsible for the enhanced PCE and power through an improved FF of 0.850 (Table 1).

The enhanced PCE of silicon solar panel at 120 min by the TPSM are consistent with their optical absorptions as shown in Fig. 4. $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$, due to localized surface plasmonic resonance (LSPR), absorbs IR in a broad range beyond 1300 nm. For silicon, any solar light with wavelength greater than 1127 nm (1.1 eV) does not contribute to the PV effect but to thermal heating. The broad IR absorption by $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ can therefore reduce solar surface heating. Therefore, to further reduce the temperature effect, the IR portion of the solar light should be effectively removed by TPSM. Thermalization is the dominating factor which is associated with the photon energies near the UV region. Chlorophyllin has an absorption peak at 404 nm which is at the high-energy boundary of the spectral response. Spectral response is defined as the ratio of the current generated by the solar panel to the power incident on the solar panel. The spectral response of a silicon solar panel is insignificant below 400 nm. It then increases with wavelength and approaches the

ideal value near 1000 nm where the photon energy is close to the bandgap of silicon (1.12 eV), beyond which the response falls back to zero.

There are two competing factors in maximizing PCE and power via TPSM over the PV panel. Enlarging the open circular area on the top wall of the TESM box will enable more direct solar light exposure, therefore increasing I_{SC} but at the same time introducing various loss mechanisms such as thermalization and sub – band – gap absorption. However, increasing the TPSM covered area (correspondingly reducing the open circle area) will compromise light power density, and hence decrease I_{SC} [39–46]. Therefore, this area ratio will have to be further optimized to achieve well retained PCE and power for prolonged solar irradiation time. The high AVT values of TPSM are associated with their optical characteristics that both chlorophyllin and $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ exhibit the saddle-like or U-shaped spectra with minimums in the visible region which allow for a wider range of solar light to pass the top side wall for enhanced PV effect. Nano hybrids between porphyrins and iron oxides have been developed to spectrally tune the absorptions for ideal solar harvest and energy generation [21,47].

Although this PT – TE – PV solar energy module was experimentally investigated with the model shown in Fig. 2d, it is highly possible to design a practically viable system capable of solar harvesting and generating energy with multiple TEGs and PVs. This concept is schematically depicted in Fig. 10. As shown in this figure, the transparent TPSM box is exposed to sunlight through a considerably large open surface area for efficient solar harvesting. During daytime, the solar light incident angle changes from morning to dusk. Since the TE generator is substantially smaller than the sidewalls of the TPSM box, sunlight can always pass through them for synergistic PT-TE-PV energy generation.

4. Conclusion

In conclusion, a PT – TE – PV solar energy module is designed and developed based on the unique structural and optical characteristics of the porphyrin and iron oxide for spectral selectively harvest solar light and synergistically generating energy via the photothermal, thermoelectric and photovoltaic devices. This novel system has addressed several key issues in solar-based energy harvesting and generation: (1) the PV devices suffer from surface heating and rapid decrease in PCE under prolonged solar irradiation. With TPSM, the IR portion is absorbed resulting in flowered PV heating, leading to enhanced PCE. (2) The IR does not contribute to PCE but PV surface heating, a band of photon energy that is not useful for PV. In the PTPSEM system, the IR is absorbed by TPSM for generating heat as the thermal source of TEG, therefore the full spectrum of solar light (from UV to IR) is well utilized for energy generation. (3) The efficiencies of both TE and PV are limited by various factors; PTPSEM combines the photothermal, thermoelectric, and photovoltaic effects through unique TPSM materials of chlorophyllin and $\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ for synergistic energy generation resulting in significant efficiencies above 21 %. Note that this is the system efficiency after long-term solar heating of 120 min. The TEGs with the photothermal source from TPSM could reach system efficiencies above 11 %. With PTPSEM, the PCE is improved from 16.3% (0 min) to 19.4% (120 min); an additional 2.4 % of the incoming photons are converted into electricity by TEGs, making a total efficiency of 21.8%. Note that both TEG and PV panel used in the PTPSEM are common commercial products with much less efficiency under normal operating conditions such as prolonged solar irradiation. By this novel design with unique materials, the intrinsic efficiency barriers of PV and TE can be lifted by nano hybrids of porphyrins and iron oxides that are structurally enabled to regulate output photons for achieving much greater efficiencies by the synergistic effects of PV, PT, and TE, possibly exceeding all current solar systems.

Table 2

PCE, PT – TE efficiency, and total efficiency of the PT – TE – PV solar energy module (PTPSEM).

	PCE (%)	PT – TE (%)	Total efficiency (%)
$\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ box	19.4	2.4	21.8
Chlorophyllin box	19.3	2.3	21.6
$\text{Fe}_3\text{O}_4@\text{Cu}_2\text{S}$ cube	N/A	11.9	11.8
Chlorophyllin cube	N/A	11.8	11.8
Without TPSM	16.3	N/A	16.3

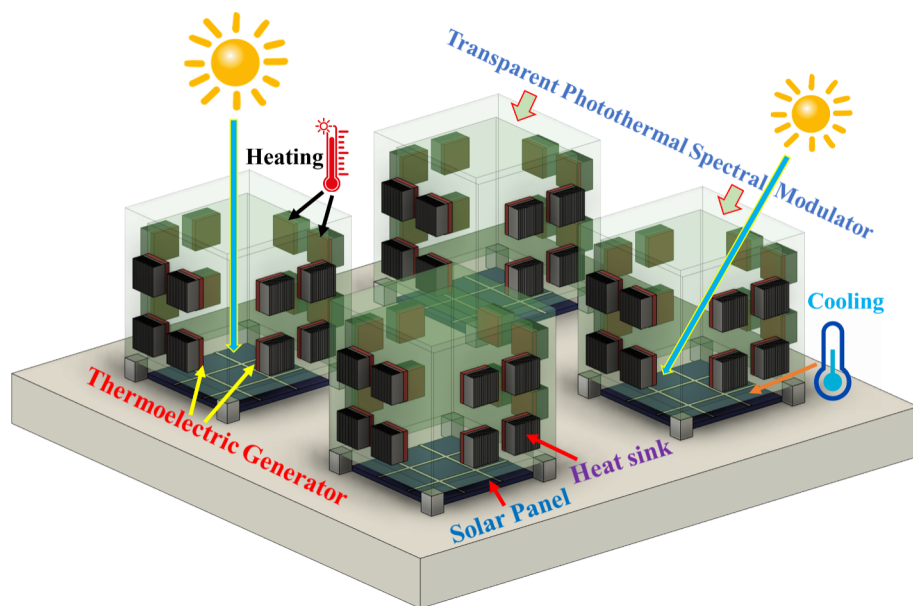


Fig. 10. Schematic diagram showing a practical flexible PT – TE – PV solar energy module.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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